

THERMOPHYSICAL PROPERTIES OF MATTER

VOLUME 1

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# THERMAL CONDUCTIVITY

## Metallic Elements and Alloys

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"In this work, when it shall be found that much is omitted, let it not be forgotten that much likewise is performed..."

SAMUEL JOHNSON, A.M.

From last paragraph of Preface to his two-volume *Dictionary of the English Language*.  
Vol. I, page 5, 1755, London, Printed by Strahan.

## Foreword

In 1957, the Thermophysical Properties Research Center (TPRC) of Purdue University, under the leadership of its founder, Professor Y. S. Touloukian, began to develop a coordinated experimental, theoretical, and literature review program covering a set of properties of great importance to science and technology. Over the years, this program has grown steadily, producing bibliographies, data compilations and recommendations, experimental measurements, and other output. The series of volumes for which these remarks constitute a foreword is one of these many important products. These volumes are a monumental accomplishment in themselves, requiring for their production the combined knowledge and skills of dozens of dedicated specialists. The Thermophysical Properties Research Center deserves the gratitude of every scientist and engineer who uses these compiled data.

The individual nontechnical citizen of the United States has a stake in this work also, for much of the science and technology that contributes to his well-being relies on the use of these data. Indeed, recognition of this importance is indicated by a mere reading of the list of the financial sponsors of the Thermophysical Properties Research Center; leaders of the technical industry of the United States and agencies of the Federal Government are well represented.

Experimental measurements made in a laboratory have many potential applications. They might be used, for example, to check a theory, or to help design a chemical manufacturing plant, or to compute the characteristics of a heat exchanger in a nuclear power plant. The progress of science and technology demands that results be published in the open literature so that others may use them. Fortunately for progress, the useful data in any single field are not scattered throughout the tens of thousands of technical journals published throughout the world. In most fields, fifty percent of the useful work appears in no more than thirty or forty journals. However, in the case of TPRC, its field is so broad

that about 100 journals are required to yield fifty percent. But that other fifty percent! It is scattered through more than 3500 journals and other documents, often items not readily identifiable or obtainable. Nearly 50,000 references are now in the files.

Thus, the man who wants to use existing data, rather than make new measurements himself, faces a long and costly task if he wants to assure himself that he has found all the relevant results. More often than not, a search for data stops after one or two results are found--or after the searcher decides he has spent enough time looking. Now with the appearance of these volumes, the scientist or engineer who needs these kinds of data can consider himself very fortunate. He has a single source to turn to; thousands of hours of search time will be saved, innumerable repetitions of measurements will be avoided, and several billions of dollars of investment in research work will have been preserved.

However, the task is not ended with the generation of these volumes. A critical evaluation of much of the data is still needed. Why are discrepant results obtained by different experimentalists? What undetected sources of systematic error may affect some or even all measurements? What value can be derived as a "recommended" figure from the various conflicting values that may be reported? These questions are difficult to answer, requiring the most sophisticated judgment of a specialist in the field. While a number of the volumes in this Series do contain critically evaluated and recommended data, these are still in the minority. The data are now being more intensively evaluated by the staff of TPRC as an integral part of the effort of the National Standard Reference Data System (NSRDS). The task of the National Standard Reference Data System is to organize and operate a comprehensive program to prepare compilations of critically evaluated data on the properties of substances. The NSRDS is administered by the National Bureau of Standards under a directive from the Federal Council for Science

and Technology, augmented by special legislation of the Congress of the United States. TPRC is one of the national resources participating in the National Standard Reference Data System in a united effort to satisfy the needs of the technical community for readily accessible, critically evaluated data.

As a representative of the NBS Office of Standard Reference Data, I want to congratulate Professor Touloukian and his colleagues on the accomplishments represented by this Series of reference data

books. Scientists and engineers the world over are indebted to them. The task ahead is still an awesome one and I urge the nation's private industries and all concerned Federal agencies to participate in fulfilling this national need of assuring the availability of standard numerical reference data for science and technology.

EDWARD L. BRADY  
*Associate Director for Information Programs  
National Bureau of Standards*

## Preface

*Thermophysical Properties of Matter*, the TPRC Data Series, is the culmination of twelve years of pioneering effort in the generation of tables of numerical data for science and technology. It constitutes the restructuring, accompanied by extensive revision and expansion of coverage, of the original *TPRC Data Book*, first released in 1960 in loose-leaf format, 11" x 17" in size, and issued in June and December annually in the form of supplements. The original loose-leaf *Data Book* was organized in three volumes: (1) metallic elements and alloys, (2) nonmetallic elements, compounds, and mixtures which are solid at N.T.P., and (3) nonmetallic elements, compounds, and mixtures which are liquid or gaseous at N.T.P. Within each volume, each property constituted a chapter.

Because of the vast proportions the *Data Book* began to assume over the years of its growth and the greatly increased effort necessary in its maintenance by the user, it was decided in 1967 to change from the loose-leaf format to a conventional publication. Thus, the December 1966 supplement of the original *Data Book* was the last supplement disseminated by TPRC.

While the manifold physical, logistic, and economic advantages of the bound volume over the loose-leaf oversize format are obvious and welcome to all who have used the unwieldy original volumes, the assumption that this work will no longer be kept on a current basis because of its bound format would not be correct. Fully recognizing the need of many important research and development programs which require the latest available information, TPRC has instituted a *Data Update Plan* enabling the subscriber to inquire, by telephone if necessary, for specific information and receive, in many instances, same-day response on any new data processed or revision of published data since the latest edition. In this context, the TPRC Data Series departs drastically from the conventional handbook and giant multivolume classical works, which are no longer adequate media for the dissemination of

numerical data of science and technology without a continuing activity on contemporary coverage. The loose-leaf arrangements of many works fully recognize this fact and attempt to develop a combination of bound volumes and loose-leaf supplement arrangements as the work becomes increasingly large. TPRC's *Data Update Plan* is indeed unique in this sense since it maintains the contents of the TPRC Data Series current and live on a day-to-day basis between editions. In this spirit, I strongly urge all purchasers of these volumes to complete in detail and return the *Volume Registration Certificate* which accompanies each volume in order to assure themselves of the continuous receipt of annual listing of corrigenda during the life of the edition.

The TPRC Data Series consists initially of 13 independent volumes. The initial ten volumes will be published in 1970, and the remaining three by 1972. It is also contemplated that subsequent to the first edition, each volume will be revised, updated, and reissued in a new edition approximately every fifth year. The organization of the TPRC Data Series makes each volume a self-contained entity available individually without the need to purchase the entire Series.

The coverage of the specific thermophysical properties represented by this Series constitutes the most comprehensive and authoritative collection of numerical data of its kind for science and technology.

Whenever possible, a uniform format has been used in all volumes, except when variations in presentation were necessitated by the nature of the property or the physical state concerned. In spite of the wealth of data reported in these volumes, it should be recognized that all volumes are not of the same degree of completeness. However, as additional data are processed at TPRC on a continuing basis, subsequent editions will become increasingly more complete and up to date. Each volume in the Series basically comprises three sections, consisting of a text, the body of numerical data with source references, and a material index.

The aim of the textual material is to provide a complementary or supporting role to the body of numerical data rather than to present a treatise on the subject of the property. The user will find a basic theoretical treatment, a comprehensive presentation of selected works which constitute reviews, or compendia of empirical relations useful in estimation of the property when there exists a paucity of data or when data are completely lacking. Established major experimental techniques are also briefly reviewed.

The body of data is the core of each volume and is presented in both graphical and tabular formats for convenience of the user. Every single point of numerical data is fully referenced as to its original source and no secondary sources of information are used in data extraction. In general, it has not been possible to critically scrutinize all the original data presented in these volumes, except to eliminate perpetuation of gross errors. However, in a significant number of cases, such as for the properties of liquids and gases and the thermal conductivity of all the elements, the task of full evaluation, synthesis, and correlation has been completed. It is hoped that in subsequent editions of this continuing work, not only new information will be reported but the critical evaluation will be extended to increasingly broader classes of materials and properties.

The third and final major section of each volume is the material index. This is the key to the volume, enabling the user to exercise full freedom of access to its contents by any choice of substance name or detailed alloy and mixture composition, trade name, synonym, etc. Of particular interest here is the fact that in the case of those properties which are reported in separate companion volumes, the material index in each of the volumes also reports the contents of the other companion volumes.\* The sets of companion volumes are as follows:

Thermal conductivity:	Volumes 1, 2, 3
Specific heat:	Volumes 4, 5, 6
Radiative properties:	Volumes 7, 8, 9
Thermal expansion:	Volumes 12, 13

The ultimate aims and functions of TPRC's Data Tables Division are to extract, evaluate, reconcile, correlate, and synthesize all available data for the thermophysical properties of materials with

\*For the first edition of the Series, this arrangement was not feasible for Volume 7 due to the sequence and the schedule of its publication. This situation will be resolved in subsequent editions.

the result of obtaining internally consistent sets of property values, termed the "recommended reference values." In such work, gaps in the data often occur, for ranges of temperature, composition, etc. Whenever feasible, various techniques are used to fill in such missing information, ranging from empirical procedures to detailed theoretical calculations. Such studies are resulting in valuable new estimation methods being developed which have made it possible to estimate values for substances and/or physical conditions presently unmeasured or not amenable to laboratory investigation. Depending on the available information for a particular property and substance, the end product may vary from simple tabulations of isolated values to detailed tabulations with generating equations, plots showing the concordance of the different values, and, in some cases, over a range of parameters presently unexplored in the laboratory.

The TPRC Data Series constitutes a permanent and valuable contribution to science and technology. These constantly growing volumes are invaluable sources of data to engineers and scientists, sources in which a wealth of information heretofore unknown or not readily available has been made accessible. We look forward to continued improvement of both format and contents so that TPRC may serve the scientific and technological community with ever-increasing excellence in the years to come. In this connection, the staff of TPRC is most anxious to receive comments, suggestions, and criticisms from all users of these volumes. An increasing number of colleagues are making available at the earliest possible moment reprints of their papers and reports as well as pertinent information on the more obscure publications. I wish to renew my earnest request that this procedure become a universal practice since it will prove to be most helpful in making TPRC's continuing effort more complete and up to date.

It is indeed a pleasure to acknowledge with gratitude the multisource financial assistance received from over fifty of TPRC's sponsors which has made the continued generation of these tables possible. In particular, I wish to single out the sustained major support being received from the Air Force Materials Laboratory-Air Force Systems Command, the Office of Standard Reference Data-National Bureau of Standards, and the Office of Advanced Research and Technology-National Aeronautics and Space Administration. TPRC is indeed proud to have been designated as a National Information Analysis Center for the Department of Defense as well as a component of the National

Standard Reference Data System under the cognizance of the National Bureau of Standards.

While the preparation and continued maintenance of this work is the responsibility of TPRC's Data Tables Division it would not have been possible without the direct input of TPRC's Scientific Documentation Division and, to a lesser degree, the Theoretical and Experimental Research Divisions. The authors of the various volumes are the senior staff members in responsible charge of the work. It should be clearly understood, however, that many have contributed over the years and their contributions are specifically acknowledged in each volume. I wish to take this opportunity to personally

thank those members of the staff, research assistants, graduate research assistants, and supporting graphics and technical typing personnel without whose diligent and painstaking efforts this work could not have materialized.

Y. S. TOULOUKIAN

*Director  
Thermophysical Properties Research Center  
Distinguished Atkins Professor of Engineering*

Purdue University  
Lafayette, Indiana  
July 1969

# Introduction to Volume 1

This volume of *Thermophysical Properties of Matter*, the TPRC Data Series, is perhaps the most comprehensive of all the volumes of the Series. Indeed, it is the result of one of TPRC's oldest data tables programs, initiated in 1959.

The volume comprises three major sections: namely, the front text material together with its bibliography, the main body of numerical data and its references, and the material index.

The text material is intended to assume a role complementary to the main body of numerical data which is the primary purpose of this volume. It is felt that a moderately detailed discussion of the theoretical nature of the property under consideration together with a review of predictive procedures and recognized experimental techniques will be appropriate in a major reference work of this kind. The extensive reference citations given in the text should lead the interested reader to sufficient literature for a detailed study. It is hoped, however, that enough detail is presented for this volume to be self-contained for the practical user.

The main body of the volume consists of the presentation of numerical data compiled over the years in a most comprehensive and meticulous manner. The scope of coverage includes the metallic elements and most metallic alloys and intermetallic compounds of engineering importance. The extraction of all data directly from their original sources ensures freedom from errors of transcription. Furthermore, some gross errors appearing in the original source documents have been corrected. The organization and presentation of the data together with other pertinent information in the use of the tables and figures are discussed in detail in the section entitled *Numerical Data*.

The part of the data tables covering the elements deserves special mention. We wish to point out that the extensive original literature data from near absolute zero to past the melting point have been critically reviewed and analyzed, and "recommended reference values" are presented.

Such recommended values are those that were considered to be the most probable when assessments were made of the information available in late 1968. Their inclusion adds a unique feature that is designed to provide the user with acceptable values. It should be realized, however, that these recommended values are not necessarily the final true values and that changes directed toward this end will often become necessary as more data become available. Future editions will contain these changes and will provide similar recommendations made for an increasing number of materials.

As stated earlier, all data have been obtained from their original sources and each data set is so referenced. TPRC has in its files all documents cited in this volume. Those that cannot readily be obtained elsewhere are available from TPRC in microfiche form.

The material index at the end of the volume covers the contents of all three companion volumes (Volumes 1, 2, and 3) on thermal conductivity. It is hoped that the user will find these comprehensive indices helpful.

This volume has grown out of activities made possible, initially by TPRC's Founder Sponsors, and, since 1960, through the principal support of the Air Force Materials Laboratory—Air Force Systems Command, under the mentorship of Mr. John H. Charlesworth. The effort to make critical assessment of the data for the elements was made possible through the support of the Office of Standard Reference Data—National Bureau of Standards, under the mentorship of Dr. Howard J. White, Jr. Over the past ten years, many graduate students and research assistants have rendered assistance for varying periods under the authors' supervision. We wish to acknowledge in chronological order of their association with TPRC, the contributions of Messrs. K. H. Chu, C. Y. Wang, A. Cezairliyan, K. C. Lin, D. Y. Nee, R. L. Feng, J. J. G. Hsia, M. Mangkornkanok, M. Nalbantyan, G. K. Kirjilian, and Mrs. E. K. C. Lee, and Mr. K. Y. Wu. The

two last mentioned are still at TPRC and participated in the final organization of the tables and figures and the demanding task of checking of details. We wish also to acknowledge the benefit of extensive discussions with Dr. J. Kaspar, Senior Staff Scientist, Materials Sciences Laboratory, Aerospace Corporation, and with Dr. A. Cezairliyan, Physicist, National Bureau of Standards. They are, respectively, Visiting Research Professor and Consultant at TPRC.

Inherent to the character of this work is the fact that in the preparation of this volume, we have drawn most heavily upon the scientific literature and feel a debt of gratitude to the authors of the referenced articles. While their often discordant results have caused us much difficulty in reconciling their findings, we consider this to be our challenge and our contribution to negative entropy of information, as an effort is made to create from the randomly distributed data a condensed, more orderly state.

While this volume is primarily intended as a reference work for the designer, researcher, experimentalist, and theoretician, the teacher at the graduate level may also use it as a teaching tool to

point out to his students the topography of the state of knowledge on the thermal conductivity of metals. We believe there is also much food for reflection by the specialist and the academician concerning the meaning of "original" investigation and its "information content."

The authors and their contributing associates are keenly aware of the possibility of many weaknesses in a work of this scope. We hope that we will not be judged too harshly and that we will receive the benefit of suggestions regarding references omitted, additional material groups needing more detailed treatment, improvements in presentation or in recommended values, and, most important, any inadvertent errors. If the *Volume Registration Certificate* accompanying this volume is returned, the reader will assure himself of receiving annually a list of corrigenda as possible errors come to our attention.

Lafayette, Indiana  
July 1969

Y. S. TOULOUKIAN  
R. W. POWELL  
C. Y. HO  
P. G. KLEMENS



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## GROUPING OF MATERIALS AND LIST OF FIGURES AND TABLES

### 1. ELEMENTS

Figure and/or Table No.	Name	Symbol	Page No.
1*	Aluminum	Al	1
2*	Antimony	Sb	10
3	Arsenic	As	16
4*	Beryllium	Be	18
5*	Bismuth	Bi	25
6*	Boron	B	41
7*	Cadmium	Cd	45
8*	Cerium	Ce	50
9*	Cesium	Cs	54
10*	Chromium	Cr	60
11*	Cobalt	Co	64
12*	Copper	Cu	68
13*	Dysprosium	Dy	82
14*	Erbium	Er	86
15	Europtium	Eu	90
16*	Gadolinium	Gd	93
17*	Gallium	Ga	97
18*	Germanium	Ge	108
19*	Gold	Au	132
20*	Hafnium	Hf	138
21*	Holmium	Ho	142
22*	Indium	In	146
23*	Iridium	Ir	152
24*	Iron	Fe	156
25*	Lanthanum	La	171
26*	Lead	Pb	175
27*	Lithium	Li	192
28*	Lutetium	Lu	198
29*	Magnesium	Mg	202
30*	Manganese	Mn	208
31*	Mercury	Hg	212
32*	Molybdenum	Mo	222
33*	Neodymium	Nd	230
34	Neptunium	Np	234
35*	Nickel	Ni	237
36*	Niobium	Nb	245
37*	Osmium	Os	254
38*	Palladium	Pd	258
39*	Platinum	Pt	262

\*Number marked with an asterisk indicates that recommended values are also reported for this material on separate figure and table of the same number followed by the letter R.

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41*	Potassium	K . . . . .	274
42*	Praseodymium	Pr . . . . .	281
43	Promethium	Pm . . . . .	285
44*	Rhenium	Re . . . . .	288
45*	Rhodium	Rh . . . . .	292
46*	Rubidium	Rb . . . . .	296
47*	Ruthenium	Ru . . . . .	300
48 <sup>b</sup>	Samarium	Sm . . . . .	305
49 <sup>b</sup>	Scandium	Sc . . . . .	309
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58 <sup>c</sup>	Thallium	Tl . . . . .	376
59 <sup>b</sup>	Thorium	Th . . . . .	381
60 <sup>b</sup>	Thulium	Tm . . . . .	385
61 <sup>b</sup>	Tin	Sn . . . . .	389
62 <sup>b</sup>	Titanium	Ti . . . . .	410
63 <sup>b</sup>	Tungsten	W . . . . .	415
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66 <sup>a</sup>	Ytterbium	Yb . . . . .	446
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2. NONFERROUS BINARY ALLOYS

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71	Aluminum + Copper	Al + Cu . . . . .	470
72	Aluminum + Iron	Al + Fe . . . . .	474
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74	Aluminum + Silicon	Al + Si . . . . .	480
75	Aluminum + Tin	Al + Sn . . . . .	483
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\* Number marked with an asterisk indicates that recommended values are also reported for this material on separate figure and table of the same number followed by the letter R.

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81	Antimony + Copper	Sb + Cu	495
82	Antimony + Lead	Sb + Pb	496
83	Antimony + Tin	Sb + Sn	497
84	Beryllium + Aluminum	Be + Al	498
85	Beryllium + Magnesium	Be + Mg	499
86	Bismuth + Antimony	Bi + Sb	502
87	Bismuth + Cadmium	Bi + Cd	505
88	Bismuth + Lead	Bi + Pb	508
89	Bismuth + Tin	Bi + Sn	511
90	Cadmium + Antimony	Cd + Sb	514
91	Cadmium + Bismuth	Cd + Bi	517
92	Cadmium + Thallium	Cd + Tl	520
93	Cadmium + Tin	Cd + Sn	521
94	Cadmium + Zinc	Cd + Zn	524
95	Chromium + Nickel	Cr + Ni	525
96	Cobalt + Carbon	Co + C	526
97	Cobalt + Chromium	Co + Cr	527
98	Cobalt + Nickel	Co + Ni	528
99	Copper + Aluminum	Cu + Al	530
100	Copper + Antimony	Cu + Sb	534
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108	Copper + Lead	Cu + Pb	554
109	Copper + Manganese	Cu + Mn	557
110	Copper + Nickel	Cu + Ni	561
111	Copper + Palladium	Cu + Pd	568
112	Copper + Phosphorus	Cu + P	571
113	Copper + Platinum	Cu + Pt	574
114	Copper + Silicon	Cu + Si	575
115	Copper + Silver	Cu + Ag	578
116	Copper + Tellurium	Cu + Te	581
117	Copper + Tin	Cu + Sn	584
118	Copper + Zinc	Cu + Zn	588
119	Germanium + Silicon	Ge + Si	597
120	Gold + Cadmium	Au + Cd	600
121	Gold + Chromium	Au + Cr	603
122	Gold + Cobalt	Au + Co	606
123	Gold + Copper	Au + Cu	609
124	Gold + Palladium	Au + Pd	614
125	Gold + Platinum	Au + Pt	617
126	Gold + Silver	Au + Ag	620

## 2. NONFERROUS BINARY ALLOYS (continued)

Figure and/or Table No.	Name	Formula	Page No.
127	Gold + Zinc	Au + Zn	623
128	Hafnium + Zirconium	Hf + Zr	624
129	Indium + Lead	In + Pb	627
130	Indium + Thallium	In + Tl	630
131	Indium + Tin	In + Sn	634
132	Lead + Antimony	Pb + Sb	637
133	Lead + Bismuth	Pb + Bi	640
134	Lead + Indium	Pb + In	643
135	Lead + Silver	Pb + Ag	646
136	Lead + Thallium	Pb + Tl	649
137	Lead + Tin	Pb + Sn	652
138	Lithium + Sodium	Li + Na	655
139	Magnesium + Aluminum	Mg + Al	658
140	Magnesium + Cadmium	Mg + Cd	661
141	Magnesium + Calcium	Mg + Ca	662
142	Magnesium + Cerium	Mg + Ce	663
143	Magnesium + Copper	Mg + Cu	666
144	Magnesium + Manganese	Mg + Mn	669
145	Magnesium + Nickel	Mg + Ni	672
146	Magnesium + Silicon	Mg + Si	675
147	Magnesium + Silver	Mg + Ag	678
148	Magnesium + Tin	Mg + Sn	679
149	Magnesium + Zinc	Mg + Zn	680
150	Manganese + Copper	Mn + Cu	683
151	Manganese + Iron	Mn + Fe	684
152	Manganese + Nickel	Mn + Ni	685
153	Mercury + Sodium	Hg + Na	686
154	Molybdenum + Iron	Mo + Fe	690
155	Molybdenum + Titanium	Mo + Ti	691
156	Molybdenum + Tungsten	Mo + W	694
157	Nickel + Chromium	Ni + Cr	697
158	Nickel + Cobalt	Ni + Co	700
159	Nickel + Copper	Ni + Cu	703
160	Nickel + Iron	Ni + Fe	707
161	Nickel + Manganese	Ni + Mn	710
162	Niobium + Uranium	Nb + U	713
163	Niobium + Zirconium	Nb + Zr	716
164	Palladium + Copper	Pd + Cu	720
165	Palladium + Gold	Pd + Au	723
166	Palladium + Platinum	Pd + Pt	726
167	Palladium + Silver	Pd + Ag	727
168	Platinum + Copper	Pt + Cu	730
169	Platinum + Gold	Pt + Au	733
170	Platinum + Iridium	Pt + Ir	734
171	Platinum + Palladium	Pt + Pd	737
172*	Platinum + Rhodium	Pt + Rh	738

\* Number marked with an asterisk indicates that recommended values are also reported for this material on separate figure and table of the same number followed by the letter R.

2. NONFERROUS BINARY ALLOYS (continued)

Figure and/or Table No.	Name	Formula	Page No.
173	Platinum + Ruthenium	Pt + Ru	743
174	Platinum + Silver	Pt + Ag	745
175	Plutonium + Aluminum	Pu + Al	746
176	Plutonium + Iron	Pu + Fe	747
177	Potassium + Sodium	K + Na	748
178	Rubidium + Cesium	Rb + Cs	751
179	Selenium + Bromine	Se + Br	754
180	Selenium + Cadmium	Se + Cd	755
181	Selenium + Chlorine	Se + Cl	756
182	Selenium + Iodine	Se + I	757
183	Selenium + Thallium	Se + Tl	758
184	Silicon + Germanium	Se + Ge	761
185	Silicon + Iron	Si + Fe	764
186	Silver + Antimony	Ag + Sb	767
187	Silver + Cadmium	Ag + Cd	770
188	Silver + Copper	Ag + Cu	773
189	Silver + Gold	Ag + Au	774
190	Silver + Indium	Ag + In	777
191	Silver + Lead	Ag + Pb	780
192	Silver + Manganese	Ag + Mn	783
193	Silver + Palladium	Ag + Pd	786
194	Silver + Platinum	Ag + Pt	790
195	Silver + Tin	Ag + Sn	791
196	Silver + Zinc	Ag + Zn	792
197	Sodium + Mercury	Na + Hg	795
198	Sodium + Potassium	Na + K	798
199	Tantalum + Niobium	Ta + Nb	801
200	Tantalum + Tungsten	Ta + W	802
201	Tellurium + Selenium	Te + Se	805
202	Tellurium + Thallium	Te + Tl	808
203	Thallium + Cadmium	Tl + Cd	811
204	Thallium + Indium	Tl + In	812
205	Thallium + Lead	Tl + Pb	815
206	Thallium + Tellurium	Tl + Te	818
207	Thallium + Tin	Tl + Sn	821
208	Thorium + Uranium	Th + U	822
209	Tin + Aluminum	Sn + Al	823
210	Tin + Antimony	Sn + Sb	824
211	Tin + Bismuth	Sn + Bi	827
212	Tin + Cadmium	Sn + Cd	830
213	Tin + Copper	Sn + Cu	833
214	Tin + Indium	Sn + In	834
215	Tin + Lead	Sn + Pb	839
216	Tin + Mercury	Sn + Hg	842
217	Tin + Silver	Sn + Ag	845
218	Tin + Thallium	Sn + Tl	846

## 2. NONFERROUS BINARY ALLOYS (continued)

Figure and/or Table No.	Name	Formula	Page No.
219	Tin + Zinc	Sn + Zn . . . . .	847
220	Titanium + Aluminum	Ti + Al . . . . .	848
221	Titanium + Manganese	Ti + Mn . . . . .	849
222	Titanium + Oxygen	Ti + O . . . . .	852
223	Tungsten + Rhenium	W + Re . . . . .	855
224	Uranium + Aluminum	U + Al . . . . .	858
225	Uranium + Chromium	U + Cr . . . . .	859
226	Uranium + Iron	U + Fe . . . . .	862
227	Uranium + Magnesium	U + Mg . . . . .	863
228	Uranium + Molybdenum	U + Mo . . . . .	864
229	Uranium + Niobium	U + Nb . . . . .	867
230	Uranium + Silicon	U + Si . . . . .	868
231	Uranium + Zirconium	U + Zr . . . . .	871
232	Vanadium + Iron	V + Fe . . . . .	874
233	Vanadium + Yttrium	V + Y . . . . .	877
234	Zinc + Aluminum	Zn + Al . . . . .	880
235	Zinc + Cadmium	Zn + Cd . . . . .	881
236	Zirconium + Aluminum	Zr + Al . . . . .	882
237	Zirconium + Hafnium	Zr + Hf . . . . .	883
238	Zirconium + Niobium	Zr + Nb . . . . .	886
239	Zirconium + Tin	Zr + Sn . . . . .	887
240	Zirconium + Titanium	Zr + Ti . . . . .	890
241	Zirconium + Uranium	Zr + U . . . . .	891

## 3. NONFERROUS MULTIPLE ALLOYS

242	Aluminum + Copper + $\Sigma X_1$	Al + Cu + $\Sigma X_1$ . . . . .	895
243	Aluminum + Iron + $\Sigma X_1$	Al + Fe + $\Sigma X_1$ . . . . .	905
244	Aluminum + Magnesium + $\Sigma X_1$	Al + Mg + $\Sigma X_1$ . . . . .	908
245	Aluminum + Manganese + $\Sigma X_1$	Al + Mn + $\Sigma X_1$ . . . . .	911
246	Aluminum + Nickel + $\Sigma X_1$	Al + Ni + $\Sigma X_1$ . . . . .	914
247	Aluminum + Silicon + $\Sigma X_1$	Al + Si + $\Sigma X_1$ . . . . .	917
248	Aluminum + Zinc + $\Sigma X_1$	Al + Zn + $\Sigma X_1$ . . . . .	922
249	Aluminum + $\Sigma X_1$	Al + $\Sigma X_1$ . . . . .	925
250	Antimony + Beryllium + $\Sigma X_1$	Sb + Be + $\Sigma X_1$ . . . . .	926
251	Beryllium + Fluorine + $\Sigma X_1$	Be + F + $\Sigma X_1$ . . . . .	929
252	Beryllium + Magnesium + $\Sigma X_1$	Be + Mg + $\Sigma X_1$ . . . . .	932
253	Bismuth + Cadmium + $\Sigma X_1$	Bi + Cd + $\Sigma X_1$ . . . . .	935
254	Bismuth + Lead + $\Sigma X_1$	Bi + Pb + $\Sigma X_1$ . . . . .	938
255	Cadmium + Bismuth + $\Sigma X_1$	Cd + Bi + $\Sigma X_1$ . . . . .	941
256	Chromium + Iron + $\Sigma X_1$	Cr + Fe + $\Sigma X_1$ . . . . .	944
257	Cobalt + Chromium + $\Sigma X_1$	Co + Cr + $\Sigma X_1$ . . . . .	947
258	Cobalt + Iron + $\Sigma X_1$	Co + Fe + $\Sigma X_1$ . . . . .	950
259	Cobalt + Nickel + $\Sigma X_1$	Co + Ni + $\Sigma X_1$ . . . . .	951
260	Copper + Aluminum + $\Sigma X_1$	Cu + Al + $\Sigma X_1$ . . . . .	952



3. NONFERROUS MULTIPLE ALLOYS (continued)

Figure and/or Table No.	Name	Formula	Page No.
261	Copper + Beryllium + $\Sigma X_1$	Cu + Be + $\Sigma X_1$	955
262	Copper + Cadmium + $\Sigma X_1$	Cu + Cd + $\Sigma X_1$	956
263	Copper + Cobalt + $\Sigma X_1$	Cu + Co + $\Sigma X_1$	957
264	Copper + Iron + $\Sigma X_1$	Cu + Fe + $\Sigma X_1$	960
265	Copper + Lead + $\Sigma X_1$	Cu + Pb + $\Sigma X_1$	961
266	Copper + Manganese + $\Sigma X_1$	Cu + Mn + $\Sigma X_1$	964
267	Copper + Nickel + $\Sigma X_1$	Cu + Ni + $\Sigma X_1$	969
268	Copper + Silicon + $\Sigma X_1$	Cu + Si + $\Sigma X_1$	972
269	Copper + Tin + $\Sigma X_1$	Cu + Sn + $\Sigma X_1$	975
270	Copper + Zinc + $\Sigma X_1$	Cu + Zn + $\Sigma X_1$	979
271	Copper + Zirconium + $\Sigma X_1$	Cu + Zr + $\Sigma X_1$	985
272	Lanthanum + Neodymium + $\Sigma X_1$	La + Nd + $\Sigma X_1$	988
273	Lead + Antimony + $\Sigma X_1$	Pb + Sb + $\Sigma X_1$	991
274	Lithium + Boron + $\Sigma X_1$	Li + B + $\Sigma X_1$	992
275	Lithium + Sodium + $\Sigma X_1$	Li + Na + $\Sigma X_1$	995
276	Magnesium + Aluminum + $\Sigma X_1$	Mg + Al + $\Sigma X_1$	998
277	Magnesium + Cerium + $\Sigma X_1$	Mg + Ce + $\Sigma X_1$	1001
278	Magnesium + Cobalt + $\Sigma X_1$	Mg + Co + $\Sigma X_1$	1004
279	Magnesium + Copper + $\Sigma X_1$	Mg + Cu + $\Sigma X_1$	1005
280	Magnesium + Nickel + $\Sigma X_1$	Mg + Ni + $\Sigma X_1$	1008
281	Manganese + Iron + $\Sigma X_1$	Mn + Fe + $\Sigma X_1$	1009
282	Manganese + Silicon + $\Sigma X_1$	Mn + Si + $\Sigma X_1$	1012
283	Molybdenum + Iron + $\Sigma X_1$	Mo + Fe + $\Sigma X_1$	1013
284	Nickel + Aluminum + $\Sigma X_1$	Ni + Al + $\Sigma X_1$	1014
285	Nickel + Chromium + $\Sigma X_1$	Ni + Cr + $\Sigma X_1$	1017
286	Nickel + Cobalt + $\Sigma X_1$	Ni + Co + $\Sigma X_1$	1028
287	Nickel + Copper + $\Sigma X_1$	Ni + Cu + $\Sigma X_1$	1031
288	Nickel + Iron + $\Sigma X_1$	Ni + Fe + $\Sigma X_1$	1035
289	Nickel + Manganese + $\Sigma X_1$	Ni + Mn + $\Sigma X_1$	1038
290	Nickel + Molybdenum + $\Sigma X_1$	Ni + Mo + $\Sigma X_1$	1041
291	Nickel + $\Sigma X_1$	Ni + $\Sigma X_1$	1044
292	Niobium + Molybdenum + $\Sigma X_1$	Nb + Mo + $\Sigma X_1$	1046
293	Niobium + Tantalum + $\Sigma X_1$	Nb + Ta + $\Sigma X_1$	1049
294	Niobium + Titanium + $\Sigma X_1$	Nb + Ti + $\Sigma X_1$	1052
295	Niobium + Tungsten + $\Sigma X_1$	Nb + W + $\Sigma X_1$	1055
296	Silver + Cadmium + $\Sigma X_1$	Ag + Cd + $\Sigma X_1$	1058
297	Silver + $\Sigma X_1$	Ag + $\Sigma X_1$	1061
298	Tantalum + Niobium + $\Sigma X_1$	Ta + Nb + $\Sigma X_1$	1062
299	Tantalum + Tungsten + $\Sigma X_1$	Ta + W + $\Sigma X_1$	1065
300	Tellurium + Arsenic + $\Sigma X_1$	Te + As + $\Sigma X_1$	1068
301	Tin + Antimony + $\Sigma X_1$	Sn + Sb + $\Sigma X_1$	1069
302	Tin + Copper + $\Sigma X_1$	Sn + Cu + $\Sigma X_1$	1072
303	Titanium + Aluminum + $\Sigma X_1$	Ti + Al + $\Sigma X_1$	1073
304	Titanium + Chromium + $\Sigma X_1$	Ti + Cr + $\Sigma X_1$	1077
305	Titanium + Iron + $\Sigma X_1$	Ti + Fe + $\Sigma X_1$	1080
306	Titanium + Manganese + $\Sigma X_1$	Ti + Mn + $\Sigma X_1$	1083

\* Number marked with an asterisk indicates that recommended values are also reported for this material on separate figure and table of the same number followed by the letter R.

3. NONFERROUS MULTIPLE ALLOYS (continued)

Figure and/or Table No.	Name	Formula	Page No.
307	Titanium + Vanadium + $\Sigma X_1$	Ti + V + $\Sigma X_1$	1086
308	Titanium + $\Sigma X_1$	Ti + $\Sigma X_1$	1089
309	Tungsten + Iron + $\Sigma X_1$	W + Fe + $\Sigma X_1$	1090
310	Tungsten + Nickel + $\Sigma X_1$	W + Ni + $\Sigma X_1$	1091
311	Uranium + Molybdenum + $\Sigma X_1$	U + Mo + $\Sigma X_1$	1094
312	Uranium + Zirconium + $\Sigma X_1$	U + Zr + $\Sigma X_1$	1097
313	Zinc + Aluminum + $\Sigma X_1$	Zn + Al + $\Sigma X_1$	1098
314	Zinc + Lead + $\Sigma X_1$	Zn + Pb + $\Sigma X_1$	1099
315	Zirconium + Aluminum + $\Sigma X_1$	Zr + Al + $\Sigma X_1$	1100
316	Zirconium + Hafnium + $\Sigma X_1$	Zr + Hf + $\Sigma X_1$	1101
317	Zirconium + Molybdenum + $\Sigma X_1$	Zr + Mo + $\Sigma X_1$	1104
318	Zirconium + Tantalum + $\Sigma X_1$	Zr + Ta + $\Sigma X_1$	1105
319	Zirconium + Tin + $\Sigma X_1$	Zr + Sn + $\Sigma X_1$	1108
320	Zirconium + Uranium + $\Sigma X_1$	Zr + U + $\Sigma X_1$	1111
321	Zirconium + $\Sigma X_1$	Zr + $\Sigma X_1$	1112

4. FERROUS ALLOYS

A. CARBON STEELS

322	Iron + Carbon + $\Sigma X_1$	Fe + C + $\Sigma X_1$	Group I . . . . .	1113
323	Iron + Carbon + $\Sigma X_1$	Fe + C + $\Sigma X_1$	Group II . . . . .	1124

B. CAST IRONS

324	Iron + Carbon + $\Sigma X_1$	Fe + C + $\Sigma X_1$	Group I . . . . .	1125
325	Iron + Carbon + $\Sigma X_1$	Fe + C + $\Sigma X_1$	Group II . . . . .	1132

C. ALLOY STEELS

326	Iron + Aluminum + $\Sigma X_1$	Fe + Al + $\Sigma X_1$	Group I . . . . .	1142
327	Iron + Aluminum + $\Sigma X_1$	Fe + Al + $\Sigma X_1$	Group II . . . . .	1145
328	Iron + Chromium + $\Sigma X_1$	Fe + Cr + $\Sigma X_1$	Group I . . . . .	1148
329	Iron + Chromium + $\Sigma X_1$	Fe + Cr + $\Sigma X_1$	Group II . . . . .	1152
330	Iron + Chromium + Nickel + $\Sigma X_1$	Fe + Cr + Ni + $\Sigma X_1$	Group I . . . . .	1160
331*	Iron + Chromium + Nickel + $\Sigma X_1$	Fe + Cr + Ni + $\Sigma X_1$	Group II . . . . .	1164
332	Iron + Cobalt + $\Sigma X_1$	Fe + Co + $\Sigma X_1$	Group II . . . . .	1176
333	Iron + Copper + $\Sigma X_1$	Fe + Cu + $\Sigma X_1$	Group I . . . . .	1179
334	Iron + Manganese + $\Sigma X_1$	Fe + Mn + $\Sigma X_1$	Group I . . . . .	1182
335	Iron + Manganese + $\Sigma X_1$	Fe + Mn + $\Sigma X_1$	Group II . . . . .	1191
336	Iron + Molybdenum + $\Sigma X_1$	Fe + Mo + $\Sigma X_1$	Group II . . . . .	1191
337	Iron + Nickel + $\Sigma X_1$	Fe + Ni + $\Sigma X_1$	Group I . . . . .	1197
338	Iron + Nickel + $\Sigma X_1$	Fe + Ni + $\Sigma X_1$	Group II . . . . .	1202
339	Iron + Nickel + Chromium + $\Sigma X_1$	Fe + Ni + Cr + $\Sigma X_1$	Group I . . . . .	1209
340	Iron + Nickel + Chromium + $\Sigma X_1$	Fe + Ni + Cr + $\Sigma X_1$	Group II . . . . .	1212

\* Number marked with an asterisk indicates that recommended values are also reported for this material on separate figure and table of the same number followed by the letter R.

4. FERROUS ALLOYS (continued)

Figure and/or Table No.	Name	Formula	Page No.
C. ALLOY STEELS (continued)			
341	Iron + Phosphorus + $\Sigma X_1$	$Fe + P + \Sigma X_1$	Group I . . . . . 1216
342	Iron + Silicon + $\Sigma X_1$	$Fe + Si + \Sigma X_1$	Group I . . . . . 1217
343	Iron + Silicon + $\Sigma X_1$	$Fe + Si + \Sigma X_1$	Group II . . . . . 1221
344	Iron + Titanium + $\Sigma X_1$	$Fe + Ti + \Sigma X_1$	Group I . . . . . 1225
345	Iron + Tungsten + $\Sigma X_1$	$Fe + W + \Sigma X_1$	Group I . . . . . 1226
346	Iron + Tungsten + $\Sigma X_1$	$Fe + W + \Sigma X_1$	Group II . . . . . 1229

5. INTERMETALLIC COMPOUNDS

Figure and/or Table No.	Formula	Page No.
347	$Sb_2Te_3$	1241
348	$As_2Te_3$	1244
349	$Ba_2Pb$	1245
350	$Ba_2Sn$	1246
351	$Be_xNb_y$	1247
352	$Be_xTa_y$	1250
353	$Be_xU_y$	1253
354	$Be_{13}Zr$	1256
355	$Bi_2Te_3$	1257
356	$B_xSi_y$	1262
357	$CdSb$	1264
358	$CdTe$	1267
359	$Ca_xPb_y$	1270
360	$Ca_2Sn$	1273
361	$CaSi$	1274
362	$CuSbSe_2$	1275
363	$Cu_2Sc_2$	1276
364	$GaAs$	1277
365	$GeTe$	1280
366	$Au_xCu_y$	1281
367	$HfB_2$	1285
368	$InSb$	1287
369	$InAs$	1292
370	$In_2Se_3$	1295
371	$In_2Te_3$	1298
372	$LaSe$	1301
373	$LaTe$	1304
374	$PbTe$	1307
375	$Mg_2Sb_2$	1310
376	$Mg_2Ge$	1311
377	$Mg_2Si$	1314
378	$Mg_2Sn$	1317

## 5. INTERMETALLIC COMPOUNDS (continued)

Figure and/or Table No.	Formula	Page No.
379	HgSe	1320
380	HgTe	1321
381	MoSi <sub>2</sub>	1324
382	NiSb	1327
383	Re <sub>x</sub> As <sub>y</sub>	1330
384	Re <sub>x</sub> Ge <sub>y</sub>	1331
385	RcSe <sub>2</sub>	1332
386	AgSbTe <sub>2</sub>	1335
387	AgCu	1338
388	Ag <sub>2</sub> Se	1339
389	Ag <sub>x</sub> Te <sub>y</sub>	1342
390	Sr <sub>2</sub> Si	1343
391	Sr <sub>2</sub> Sn	1344
392	TaB <sub>2</sub>	1345
393	TaGe <sub>2</sub>	1348
394	Tl <sub>2</sub> Pb	1349
395	SnSe <sub>2</sub>	1352
396	SnTe	1355
397	TiB <sub>2</sub>	1358
398	TiNi	1361
399	W <sub>2</sub> As <sub>1</sub>	1364
400	WB	1365
401	WSe <sub>2</sub>	1368
402	WSi <sub>2</sub>	1369
403	WTe <sub>2</sub>	1370
404	ZnSe	1371
405	ZnSIAe <sub>2</sub>	1374
406	ZrB	1375

## 6. MIXTURES OF INTERMETALLIC COMPOUNDS

407	Sb <sub>2</sub> Se <sub>3</sub> + Ag <sub>2</sub> Se + PbSe	1379
408	Sb <sub>2</sub> Te <sub>3</sub> + Bi <sub>2</sub> Te <sub>3</sub>	1380
409	Sb <sub>2</sub> Te <sub>3</sub> + In <sub>2</sub> Te <sub>3</sub>	1386
410	Bi <sub>2</sub> Te <sub>3</sub> + Sb <sub>2</sub> Te <sub>3</sub>	1388
411	Bi <sub>2</sub> Te <sub>3</sub> + Sb <sub>2</sub> Te <sub>3</sub> + Sb <sub>2</sub> Se <sub>3</sub>	1392
412	Bi <sub>2</sub> Te <sub>3</sub> + Bi <sub>2</sub> Se <sub>3</sub>	1393
413	Cd <sub>2</sub> As <sub>2</sub> + Zn <sub>2</sub> As <sub>2</sub>	1396
414	• CdSb + ZnSb	1397
415	CuSbSe <sub>2</sub> + Cu <sub>3</sub> Se <sub>2</sub>	1400
416	Cu <sub>3</sub> Se <sub>2</sub> + CuSbSe <sub>2</sub>	1401
417	InSb + In <sub>2</sub> Te <sub>3</sub>	1403
418	In <sub>2</sub> Te <sub>3</sub> + Cu <sub>2</sub> Te + Ag <sub>2</sub> Te	1406
419	HgTe + CdTe	1407
420	AgSbTe <sub>2</sub> + SnTe	1410

## 6. MIXTURES OF INTERMETALLIC COMPOUNDS (continued)

Figure and/or Table No.	Formula	Page No.
421	SnTe + AgSbTe <sub>2</sub> . . . . .	1411
422	ZnSb + CdSb . . . . .	1412

## 7. MISCELLANEOUS ALLOYS AND MIXTURES

423	Bi <sub>2</sub> Te <sub>3</sub> + Te . . . . .	1415
424	Be + BeO . . . . .	1416
425	Cr + Al <sub>2</sub> O <sub>3</sub> . . . . .	1419
426	Cu + BeCo . . . . .	1420
427	GaAs + GaP . . . . .	1423
428	InAs + InP . . . . .	1428
429	Mo + ThO <sub>2</sub> . . . . .	1429
430	Na + Na <sub>2</sub> O . . . . .	1432
431	TiNi + Cu . . . . .	1433
432	TiNi + Ni . . . . .	1436
433	W + ThO <sub>2</sub> . . . . .	1439
434	U + UO <sub>2</sub> . . . . .	1442
435	Zr + ZrO <sub>2</sub> . . . . .	1444

# **Theory, Estimation, and Measurement**

# Notation

$A$	Cross-sectional area	$x$	Reduced frequency ( $x = h\nu/\kappa T$ )
$a$	Cube root of the atomic volume; Half the focal length of an ellipsoid; Axis of ellipse	$\Delta x$	Distance difference
$B$	Coefficient of equation (19)	$\alpha$	Parameter
$b$	Axis of ellipse; Numerical constant	$\alpha', \alpha''$	Constants
$C$	Specific heat per unit volume	$\beta$	Parameter
$C(\nu)$	Spectral specific heat	$\gamma$	Anharmonicity coefficient
$c$	Atomic concentration of point defects	$\delta$	Amplitude decrement of temperature wave; Coefficient of equation (11)
$D$	Thermal diffusivity; Coefficient of equation (19)	$\epsilon$	Local thermal strain
$E$	Energy; Voltage drop	$\zeta$	Fermi energy
$e$	Electronic charge	$\theta$	Debye temperature
$G$	Band gap energy	$\kappa$	Boltzmann constant
$h$	Planck constant	$\lambda$	Wavelength
$I$	Electric current	$\lambda_m$	Minimum wavelength
$k$	Thermal conductivity	$\nu$	Frequency
$k_m$	Thermal conductivity maximum	$\nu_m$	Debye limiting cutoff frequency
$k_r$	Thermal conductivity of a reference material	$\pi$	Peltier coefficient; Ratio of the circumference of a circle to its diameter
$L_0$	Lorenz number	$\rho$	Electrical resistivity
$l$	Mean free path; Effective length of a specimen	$\sigma$	Electrical conductivity
$M$	Atomic mass	<i>Subscripts</i>	
$m$	Constant	$d$	Dislocation
$N$	Number of atoms per unit volume	$e$	Electronic component
$N_a$	Number of conduction electrons per atom	$ee$	Electron-electron scattering
$n$	Exponent; Constant	$g$	Lattice component
$P$	Slope	$gd$	Lattice component due to the scattering of phonons by dislocations
$q$	Heat flow per unit time (and length)	$ge$	Lattice component due to the scattering of phonons by electrons
$r$	Radial distance	$gp$	Lattice component due to the scattering of phonons by point defects
$S$	Seebeck coefficient	$i$	Intrinsic
$T$	Temperature	$j$	Type of carrier
$T_c$	Superconducting transition temperature	$n$	Normal state
$T_m$	Temperature corresponding to thermal conductivity maximum	$p$	Phonon
$\Delta T$	Temperature difference	$pe$	Phonon-electron scattering
$t$	Time	$RT$	Room temperature
$U$	Coefficient of equation (24)	$s$	Superconducting state
$V$	Electrical potential	$U$	Umklapp process
$v$	Velocity	$\alpha$	Type of scattering process
$W$	Thermal resistivity	$0$	Residual
$W_\infty$	Theoretical constant thermal conductivity at high temperature	$1, 2, 3, \dots$	Locations, times, or materials

# Theory of Thermal Conductivity of Metallic Materials

## 1. INTRODUCTION

Heat in solids is conducted by various carriers: electrons, lattice waves (or phonons), magnetic excitations, and, in some cases, electromagnetic radiation. The total thermal conductivity is additively composed of contributions from each type of carrier. It can be shown that

$$k = \frac{1}{3} \sum_j C_j v_j l_j \quad (1)$$

where the subscript  $j$  denotes the type of carrier,  $C_j$  is the contribution of carriers of type  $j$  to the specific heat per unit volume,  $v_j$  is the velocity of the carrier (we regard the carrier as a particle; if the carrier is a wave, the appropriate velocity is the group velocity), and  $l_j$  is a suitably defined mean free path.

The theory of thermal conductivity of solids has been the subject of numerous investigations and several review articles and has constituted a large portion of the material in several books [1-15]. It is the purpose of this introductory text to present the major results of the theory only to the extent to which it is needed by the user of these tables: to caution him as to which results are likely to be structure sensitive and thus likely to vary from specimen to specimen, and to help him to judge which materials are likely to have similar properties and thus to guide him in guessing the thermal conductivities of materials which have not been measured.

The occurrence of a mean free path in equation (1) opens up the possibility that in some cases one cannot uniquely define the thermal conductivity of a material. This happens whenever a carrier mean free path becomes comparable to the smallest external dimension of the specimen. It happens particularly in insulators at low temperatures, because of long phonon mean free paths, and in transparent solids at high temperatures, where photons contribute signifi-

cantly to heat transport. In metals it is a relatively rare occurrence, to be considered only in the case of very small particles, very thin wires, or superconductors at extremely low temperatures.

The principal carriers of heat in metals are electrons and lattice waves, leading to an overall thermal conductivity

$$k = k_e + k_l \quad (2)$$

where  $k_e$  is the electronic component and  $k_l$  the lattice component.

Generally  $k_l$  of metals, alloys, and semimetals is of magnitude comparable to the lattice thermal conductivity of insulators of corresponding elastic properties, except at low temperatures (where phonon-electron interaction reduces  $k_l$  in metals). The relative importance of  $k_e$  and  $k_l$  thus depends on the magnitude of  $k_e$ . The electronic component often parallels the electrical conductivity (Wiedemann-Franz law), and the electrical conductivity is highest in pure metals, reduced in the case of alloys, and even lower in semimetals and semiconductors.

We thus have, as a rough rule, that in highly conducting metals, i.e., pure metals with room temperature resistivities of up to say  $5 \mu\Omega \text{ cm}$ ,  $k_e$  is the dominant component over all or almost all temperatures. For poorly conducting metals of higher resistivity,  $k_l$  forms an appreciable component at ordinary temperatures. For alloys of more than 0.5 to 1 percent solute content,  $k_e$  is substantially decreased below the value of the parent metal at low temperatures (i.e., below room temperature) where  $k_l$  becomes significant, but at higher temperatures  $k_e$  approaches the value of pure metals of comparable conductivity. Thus  $k_l$  is an important component of alloys below room temperatures; above room temperatures it is important when the parent metal is a poor conductor.

As we consider materials of increasingly poorer conductivity,  $k_e$  becomes less important relative to



$k_e$ . In semimetals and degenerate semiconductors  $k_e$  and  $k_p$  are frequently comparable except at low temperatures, where  $k_p$  is small. In most semiconductors the electronic thermal conductivity has to be considered only at elevated temperatures.

Since  $k_p$  and  $k_e$  behave differently as functions of temperature and with the introduction of imperfections, it is important to know the relative roles of these two carriers if predictions are to be made; unfortunately, it is not always easy to know what fraction of the measured thermal conductivity can be ascribed to each (see, for example, reference [1]).

For purposes of the theory of conduction properties, we distinguish between three temperature regimes: high, intermediate, and low, with rough divisions at temperatures (on the absolute scale) of  $\theta$  and  $\theta/3$  respectively, where  $\theta$  is the Debye temperature. For our purposes this temperature is related to the upper frequency limit  $\nu_m$  of the spectrum of lattice waves by  $h\nu_m = \kappa\theta$ , where  $h$  and  $\kappa$  are the Planck and Boltzmann constants, respectively. Roughly speaking, at high temperatures each atom vibrates independently of its neighbors, and the theories of lattice vibration simplify. At low temperatures the vibrations are highly correlated and are best described by elastic waves in a continuum with corresponding simplification. The intermediate regime is somewhat awkward, and theoretical results are obtained by interpolation.

All the carriers have mean free paths which are limited in part by the structural imperfections of the solid and in part by the dynamic imperfections produced by thermal vibrations, so that, for each carrier,

$$1/l = \sum_{(\alpha)} 1/l_{(\alpha)} \quad (3)$$

the summation being over all processes  $\alpha$  which scatter the carrier. The theory describing scattering by the thermal vibrations depends on the temperature regime, and this accounts for the different theories to be used for the conduction properties.

Most solids have Debye temperatures  $\theta$  around 300 K, but atomically heavy solids have lower  $\theta$ 's (well below 200 K for gold and lead), while most light-atom solids (diamond, beryllium) have much higher Debye temperatures.

## 2. ELECTRONIC THERMAL CONDUCTIVITY

In metals, where the density of electrons is high, the electron gas is highly degenerate, that is, all electron states of energy  $E < \zeta$  are filled,  $\zeta$  being

the Fermi energy, all states of  $E > \zeta$  are empty, and all conduction properties occur in an energy interval  $\zeta \pm 0(\kappa T)$ , where  $\kappa T \ll \zeta$ . Under these circumstances the electronic component of the specific heat  $C_e \propto T$ ,  $v_e$  is typically  $10^8$  cm sec<sup>-1</sup> and independent of temperature. Thus

$$k_e = \frac{1}{3} C_e v_e l_e \propto T l_e \quad (4)$$

and its temperature dependence is governed by the temperature dependence of  $l_e$ .

Now the electron mean free path  $l_e$  also determines the electrical conductivity. Theories of the electron mean free paths are thus at the same time theories of the electrical conductivity  $\sigma$  of metals, while the thermal and the electrical conductivity are related by the Wiedemann-Franz-Lorenz law

$$k_e/\sigma T = L_0 = \frac{1}{3} \pi^2 (\kappa/e)^2 \quad (5)$$

where  $e$  is the electronic charge. In practical units the Lorenz number  $L_0$  is  $2.443 \times 10^{-8}$  V<sup>2</sup> K<sup>-2</sup>.

The electron mean free path is limited both by electron scattering by defects (chemical impurities and physical defects) and by the thermal vibrations; for in a perfectly periodic crystal lattice  $l_e$  would be infinite. To the extent to which the scattering rates due to different processes are additive,

$$1/l_e = 1/l_0 + 1/l_i(T) \quad (6)$$

where  $l_0$  is the residual mean free path and  $l_i(T)$  the intrinsic mean free path. The first term on the right hand side of equation (6) describes scattering of electrons by defects, and varies from specimen to specimen, but is independent of temperature as long as the nature and concentration of defects are not functions of temperature.\* The second term describes scattering of electrons by lattice vibrations and varies with temperature. In the simple case when the electron gas is isotropic, i.e., the velocity and apparent density of electrons do not depend upon direction relative to the crystal axes, equation (6) leads to an additivity of the corresponding electrical and electronic thermal resistivities, i.e.,

$$1/\sigma = \rho = \rho_0 + \rho_i(T) \quad (7)$$

and

$$1/k_e = W_e = W_0 + W_i \quad (8)$$

\*An important exception would be the vacancy concentration just below the melting point, which in some metals depends on temperature.

Here  $\rho_0$  is the residual electrical resistivity, varying from specimen to specimen,  $\rho_i(T)$  the intrinsic electrical resistivity, and  $W_0, W_i$  are corresponding components of the thermal resistivity.

The Wiedemann-Franz-Lorenz law, equation (5), is based on the following requirements:

- (a) The electron gas is highly degenerate--this appears to be the case in all metals except possibly some transition metals at elevated temperatures.
- (b) The electron mean free path  $l_e$  is the same for electrical as for thermal conduction, so that it cancels in the ratio  $k/\sigma$ .

An analysis of the second requirement shows that this is almost always satisfied for defect scattering [1, 2]; hence

$$\rho_0/W_0 T = L_0 \quad (9)$$

The exceptions are some cases of magnetic impurities, where scattering is simultaneously inelastic and anisotropic; even here the departure from (9) is significant only at low temperature, where  $\kappa T$  is comparable to the Zeeman level splitting of the impurity.

As regards  $\rho_i$  and  $W_i$ , these are related by the Wiedemann-Franz-Lorenz law in the high-temperature limit only, while at lower temperatures

$$\rho_i/W_i T = L_i \quad (10)$$

is generally less than  $L_0$  for reasons given in reference [2]. Generally  $L_i < L_0$  and in the limit of low temperatures

$$L_i = \delta(T/\theta)^2 \quad (11)$$

where the coefficient  $\delta$  depends on the topology of the Fermi surface.

Explicit theoretical expressions for  $\rho_i$  and  $W_i$  have been obtained only for the very simplest of models, with the electron gas similar to a gas of free electrons, with spherical energy contours in momentum space, and the underlying crystal structure replaced by a uniform distribution of positive charge to compensate the charge of the electron gas. This model, the "jellium" model [12], thus foregoes any considerations of effects due to crystal structure and the corresponding electronic band structure of real metals. On such a model, using a Debye model for the spectrum of lattice vibrations, one obtains the following low-temperature limits [2]

$$\rho_i \propto T^0, \quad W_i \propto T^2 \quad (12)$$

and

$$L_i = \rho_i/W_i T = 7.8 N_a^{-2/3} (T/\theta)^2 \quad (13)$$

where  $N_a$  is the number of conduction electrons per atom. The electronic thermal resistivity at low temperatures can thus be written in the form

$$W_e = \alpha T^2 + \beta/T \quad (14)$$

Thus

$$k_e = \frac{1}{\alpha T^2 + \beta/T} \quad (15)$$

where  $\beta = \rho_0/L_0$  depends on the specimen, while  $\alpha$  is an intrinsic property of the metal. Equation (15) implies that  $k_e$ , and thus the total thermal conductivity of pure metals, should pass through a maximum at low temperatures; the purer the specimen, the higher the maximum conductivity and the lower the temperature at which the maximum occurs.

In order to predict theoretically the magnitude of  $\alpha$ , or the magnitude of  $\rho_i$ , one would need to predict the strength of the interaction between electrons and lattice waves and the corresponding electron scattering probabilities. This interaction can only be estimated very roughly.

In a similar manner, the ability of fundamental theory to predict the absolute magnitude of  $\beta$  for a given concentration of impurities or other defects is very limited. Since such calculations also predict  $\rho_0$ , and since electrical resistivities are measured more readily than thermal conductivities, such calculations are usually classified under electrical resistivity calculations. There is an extensive literature on this subject [12], dealing both with theoretical methods and with the vexing question of determining from experiments the specific resistivity of a given number of physical defects of various kinds (i.e., vacancies, interstitials, dislocations, stacking faults) [16]. Even the specific resistivity (or  $\rho_0$  per atomic percent) of a given species of solute atoms is not a trivial determination.

Generally speaking the theory is well able to predict  $\rho_0$  due to point defects which scatter electrons mainly by virtue of a valence difference (vacancies, solute atoms), but does poorly when distortion effects are important (interstitials, dislocations).

The theoretical equation (15) has been extensively compared with low-temperature experimental data for high-purity metals, and disagreements are found [55-57] in that the power of  $T$  for most metals is not 2 but greater, and the coefficient  $\alpha$  is not a constant for a metal. Considering the temperature

dependence of the coefficient  $\alpha$  and the interaction between intrinsic and residual thermal resistivities, equation (15) is modified semiempirically [55-57] to become

$$k_e = \frac{1}{\alpha' T^n + \beta/T} \quad (16)$$

where

$$\alpha' = \alpha \left( \frac{\beta}{n\alpha'} \right)^{(m-n)(m+1)}$$

and  $\alpha'$ ,  $m$ , and  $n$  are constants for a metal. The value of  $n$  lies between 2 and 3 for most metals. Much better agreements are obtained in using equation (16) for fitting experimental data, and this equation has been used extensively for calculating the recommended values presented in this volume for highly conducting elements at temperatures below about  $1.5 T_m$ , with  $T_m$  the temperature corresponding to the maximum conductivity  $k_m$  of the curve.

At high temperatures the theory becomes simpler in form, though it is still difficult to predict absolute values. The theory predicts in the limit, well above the Debye temperature,

$$\begin{aligned} \rho_i &\propto T \\ W_i &= \rho_i/L_0 T = W_\infty \end{aligned} \quad (17)$$

where  $W_\infty$  is a constant. This is essentially a consequence of the fact that, at high temperatures the intrinsic scattering probability, or  $1/l_i(T)$ , varies as  $\langle \epsilon^2 \rangle$ , the mean square thermal strain, which in turn varies as  $T$ .

All this needs qualification if finer details are to be investigated. Thus, for example, thermal expansion as well as high-order interactions might cause deviations from  $l_i \propto T^{-1}$ , and corresponding deviations of  $W_i$  from constancy. There are further deviations of  $L_i$  from  $L_0$  at high temperatures, generally small but not entirely negligible, arising from the transport equations [12]. Thus  $W_i(T)$  tending to a constant value is a good picture only to a first approximation. Finally the assumption of high degeneracy, i.e.,  $\kappa T \ll \zeta$ , which is required so that  $\rho \propto 1/l$  and  $W \propto T/l$ , becomes questionable in some metals at elevated temperatures (i.e., transition metals, actinides), while in the simpler metals we believe it to be a good assumption right up to the melting point.

Finally, the fact that scattering of electrons by thermal vibrations is proportional to  $\langle \epsilon^2 \rangle$  enables us to estimate the order of magnitude of point defect scattering in terms of thermal scattering at

room temperatures, and thus to estimate  $\rho_0$  and  $W_0$  in terms of the room temperature values of  $\rho_i$  and  $W_i$ . At room temperature  $\langle \epsilon^2 \rangle$  is about 0.01 in most solids (depending somewhat on atomic volumes and elastic constants, which are not very variable). If we arbitrarily ascribe scattering by a point defect to be equivalent to unit shear strain over one atomic volume, we find that

$$\begin{aligned} \rho_0 &\simeq (\rho_i)_{RT} c \\ W_0 T &\simeq (W_i)_{RT} c T_{RT} \end{aligned} \quad (18)$$

where  $(\rho_i)_{RT}$ ,  $(W_i)_{RT}$  are the room temperature intrinsic resistivities,  $T_{RT} \simeq 300$  K, and  $c$  is the atomic concentration of point defects in percent. Actual values of  $\rho_0$  and  $W_0 T$  deviate from the estimates of equation (18) by factors of up to about 3 either way, but equation (18) is very useful in estimating resistivities due to unknown point defects, and applies to all metals and semimetals irrespective of the details of the band structure.

In addition to the scattering of electrons by defect and by thermal vibrations, it is also possible for the electrons to scatter each other, producing both an electrical and thermal resistivity. These effects seem important mainly in metals of high density of states such as transition metals, and lead to resistivities of the form

$$\begin{aligned} \rho_{ee} &\simeq BT^2 \\ W_{ee} &\simeq DT \end{aligned} \quad (19)$$

These effects are thus generally important at low temperatures, where they tend to dominate over the respectively  $T^5$  and  $T^2$  variation of  $\rho_i$  and  $W_i$ , provided of course that the samples are pure enough so that the residual resistivities  $\rho_0$  and  $W_0$  do not yet dominate at those low temperatures. As transition metals are obtained in increasingly pure form, the need of additional terms of the form (19) in the electrical and thermal resistivity equations is found in more and more cases.

Major results of the theory of electronic thermal conductivity have been briefly presented as above. For detailed theoretical developments and discussions, the reader is referred to the review papers and books cited before and to other papers on this subject [17-66].

### 3. LATTICE THERMAL CONDUCTIVITY

The thermal vibrations of solids contribute to the thermal conductivity. In insulators this is the

only mechanism of heat transport except at elevated temperatures. The theory as it applies to insulators is described in somewhat greater detail in Volume 2. The lattice thermal conductivity of metals is governed by the same theoretical considerations, but a number of cases which occur in insulators are not relevant to metals, so that the present discussion is not as comprehensive. General reviews of lattice thermal conductivity are given in references [1-4]. Individual research papers on the theory of lattice thermal conductivity include [67-132].

The thermal vibrations of a perfect crystal are described in terms of lattice waves which occupy a spectrum of frequencies from the lowest frequencies to some upper limit,  $\nu_m$ , of the order of  $10^{13}$  Hz. At low frequencies these waves are identical to the elastic waves in the corresponding elastic continuum; at the higher frequencies the atomic structure of the crystal lattice leads to dispersion effects. The corresponding wavelengths range from long waves down to waves of length comparable to the interatomic distances.

These waves are randomly excited in thermal equilibrium and the energy content of the solid is given in terms of the laws of statistical mechanics. The specific heat of solids varies as  $T^3$  at the lowest temperatures and is independent of temperature in the high-temperature regime ( $T > \theta$ , where  $\theta = h\nu_m/\kappa$ ). The spectral distribution of the specific heat per unit volume as given, to a first approximation, by the Debye theory, is of the form

$$C(\nu) d\nu = 9N\kappa \left(\frac{T}{\theta}\right)^3 \frac{x^4 e^x}{(e^x - 1)^2} dx \quad (20)$$

where  $x = h\nu/\kappa T$  is a reduced frequency and  $N$  the number of atoms per unit volume. This holds for  $\nu < \nu_m$ ; for  $\nu > \nu_m$ ,  $C(\nu) = 0$ .

The Debye approximation disregards the dispersion of the high-frequency lattice waves, disregards differences of polarization of different lattice waves, and smears out the crystal structure of the solid. The only concession to the discreteness of the lattice is the choice of the cutoff frequency  $\nu_m$  or the corresponding minimum wavelength  $\lambda_m = v/\nu_m$ , where  $v$  is some average sound velocity. It is chosen so that the total number of waves corresponds to the correct number of normal modes ( $3N$ ) which this assembly of  $N$  atoms ought to have.

In spite of the obvious inadequacy of the Debye approximation, it is frequently chosen as the basis of discussing thermal conductivity, because the inadequacy of that theory is such that small errors in

the theory of the specific heat are usually not too important.

In a perfectly periodic crystal which also obeys perfectly the laws of linear elasticity or Hooke's law (i.e., all restoring forces are linear functions of relative displacements, the elastic energy is a quadratic function of relative displacements), each elastic wave is completely independent of all other elastic waves and maintains forever whatever energy it possesses. Such a crystal could carry a heat current or net flow of elastic energy without a driving force, and would thus have an infinite thermal conductivity.

Real crystals, by virtue of structural defects and deviations from linear elasticity, have lattice waves which continuously interchange energy with each other, so that each lattice wave has a finite mean free path. From equation (1), generalized to take account of the fact that this mean free path  $l(\nu)$  is generally a function of the frequency of the lattice wave,  $\nu$ , the lattice thermal conductivity becomes

$$k_s = \frac{1}{3} \int_0^{\nu_m} C(\nu) v_g l(\nu) d\nu \quad (21)$$

where  $v_g = dv/d(1/\lambda)$  is the group velocity of the waves, and  $C(\nu)$  is the spectral specific heat, given approximately by (20).

The lattice thermal conductivity is thus governed by the mean free path of the lattice waves, in an analogous manner to the electronic conduction properties which are governed by the electron mean free path.

Each individual wave can be regarded as a normal mode (or almost normal mode, if  $l$  is finite) of the crystal, and obeys the dynamical equations of a harmonic oscillator. According to the laws of quantum mechanics, the energy of each oscillator is not continuously variable, but an integral number of quanta, each of energy  $h\nu$ . In fact, this was included in the statistical mechanics leading to (20). Each quantum of energy is called a phonon, and each lattice wave contains an integral number of such phonons. By focusing attention on the phonons, as if they were particles, one can describe the thermal energy of vibration of a crystal as a gas of phonons, and use the concepts of the kinetic theory of gases. This description is completely equivalent to the description in terms of lattice waves, and also leads to (21); it has certain advantages of ease of conceptualization, particularly when we talk of the processes of energy interchange between different waves, which can be regarded as scattering of

phonons from one wave into another, or the breaking up of a phonon into other phonons, etc. The final results are equivalent, and the phonon mean free path  $l(\nu)$  can be defined in an equivalent manner. We shall use either description according to convenience.

At elevated temperatures in good crystals, the principal process limiting the mean free path is the interchange of energy, or scattering of phonons, due to departures from Hooke's law. A local strain  $\epsilon$  introduces a fractional change in local sound velocity of  $\gamma\epsilon$ , where the coefficient  $\gamma$ , a measure of the anharmonicity, is of order unity (frequently  $\gamma \approx 2$ ). At high temperatures the thermal strain at neighboring atomic sites is almost uncorrelated, and scattering is proportional to  $\langle \epsilon^2 \rangle$  and, in turn, proportional to  $T$ . Thus the intrinsic mean free path varies as

$$l_i \propto 1/T \quad (22)$$

analogously to the similar variation of the electron mean free path [see equation (17)]. This has been pointed out by Debye [67].

There is an important difference, however. The electrons contributing to the conductivity all have a short wavelength, but the lattice waves have a continuous spectrum of wavelengths. A perturbed atomic site, acting as an independent scattering source, would scatter long-wave lattice waves very weakly, and  $l(\nu)$  would increase so rapidly with decreasing  $\nu$  that the thermal conductivity integral, equation (21), would diverge at the low-frequency limit. In fact this model would lead to  $l_i(\nu) \propto 1/\nu^4$ , and since  $C(\nu) \propto \nu^2$  at lowest frequencies [see equation (20)], one can readily appreciate the divergence difficulty. This simple model is thus inadequate at low frequencies. To avoid this difficulty, Peierls [68] set up a theory of the anharmonic interaction between lattice waves, which is the basis of all subsequent theoretical work. The theory resolves the thermal vibrations into their proper spectral components of lattice waves, and treats in detail the interchange of energy between groups of three lattice waves, or breaking up of one phonon into two other phonons (or vice versa), satisfying certain interference conditions between the frequencies and the direction and wavelengths involved. The theory leads, with some approximation, to a variation

$$1/l_i \propto \nu^2 T \quad (23)$$

and avoids the divergence difficulties [3].

In detail, however, the theory is quite complicated, involving not only the strength of the anharmonic interaction  $\gamma$ , which cannot be estimated with accuracy, but also the detailed crystal structure. (An elastic continuum would not have a thermal resistivity.) Nevertheless, rough estimates have been made [3, 79] and the intrinsic thermal resistivity of the lattice component is of the form

$$W_U \approx U \left(\frac{h}{\kappa}\right)^3 \frac{\gamma^2 T}{Ma \theta^3} \quad (24)$$

where  $M$  is the atomic mass and  $a^3$  the atomic volume. The numerical coefficient  $U$ , typically of order 1/3, is somewhat uncertain, and depends on the details of the crystal structure. The major factor controlling the intrinsic lattice conductivity, however, is the Debye temperature. Solids of high  $\theta$  will generally have higher values of  $k_t$ .

The subscript  $U$  stands for *Umklapp*, or flip-over; in Peierls' theory the resistive processes are processes in which the phonon interaction is combined with a Bragg reflection of the lattice wave, or Umklapp processes. These are distinct from the "normal" processes, phonon interactions which help establish thermal equilibrium, but do not change the net energy flow associated with the phonon gas. The need to distinguish between these processes and their different role in producing thermal resistivity adds further complexity to the theory, particularly at intermediate temperatures, and makes detailed numerical predictions very difficult.

At low temperatures the thermal resistivity decreases exponentially according to  $W_U \propto e^{-\theta/T}$ , where  $b$  is a numerical constant, which also depends sensitively on the crystal structure and the dispersion of the high-frequency lattice waves. This exponential temperature variation has been observed in perfect insulators, but in many insulators it is overshadowed by the thermal resistance which arises from the scattering of lattice waves by various defects.

In metals, however, another resistive process usually dominates at low temperatures: scattering of the lattice waves or phonons by the free electrons. This is of course the very same process as the scattering of conduction electrons by lattice waves or thermal vibrations which had been invoked earlier to limit the electronic conduction properties and which gave rise to  $\rho_i(T)$  and  $W_i(T)$ . These interactions limit the phonon mean free path; the corre-

sponding reciprocal mean free path for phonon-electron scattering

$$1/l_{pe}(\nu) \propto \nu \quad (25)$$

must be added to the other scattering processes according to equation (3) to obtain  $l(\nu)$  of equation (21).

The result of this scattering mechanism is that  $k_T$  of metals is lower than  $k_T$  of insulating crystals of the same elastic properties, particularly at low temperatures, where  $k_T \propto T^2$ . At high temperatures, however, anharmonic resistive processes dominate, and  $k_T$  tends to equal  $1/W_U$  and thus varies as  $1/T$ . The lattice conductivity will thus have some maximum value at intermediate temperatures. This maximum value is usually considerably smaller than that of insulating crystals, and also smaller than  $k_T$  of most pure metals. Typical values of  $k_T$  at its maximum (which may lie between 20 and 50 K) would not exceed  $0.5 \text{ W cm}^{-1} \text{ K}^{-1}$ , while  $k_T$  of metals and  $k_T$  of insulators may peak at values ranging from tens to hundreds of watt-units.

Equation (25) was based on the assumption that the electrons have such a long mean free path that the lattice wave can interact with individual electrons. If  $l'$  is the electron mean free path, this is equivalent to requiring that

$$l' > \lambda \quad (26)$$

where  $\lambda$  is the wavelength of the lattice wave (or phonon). In alloys, where  $l'$  is finite, this relation breaks down for sufficiently long waves, i.e., at sufficiently low temperatures. With  $l'$  typically 100 $a$  for 1 percent impurity and  $\lambda \approx \frac{1}{2} a(\theta/T)$  for the important thermal waves, where  $a^3$  is the atomic volume, equation (26) is barely satisfied at liquid helium temperatures for a 1 percent alloy (a typical minimum concentration at which  $k_T$  can still be separated from  $k$  and studied).

In the opposite extreme ( $l' \ll \lambda$ ), the lattice wave no longer interacts with individual electrons, but with the electron gas as a whole [83]. The scattering is no longer given by equation (25), but reduced by a factor of order  $(l'/\lambda)$ , so that  $1/l_{pe}(\nu) \propto \nu^2$  and  $k_T \propto T$  instead of  $T^2$ , and appropriately increased. This changed temperature dependence is clearly seen in concentrated alloys of  $\rho_0 > 10 \mu\Omega \text{ cm}$ . But even in more dilute alloys the effect is partially present at liquid helium temperatures and should be more apparent at lower temperatures. A theoretical analysis of these intermediate situations and a comparison with data for the lattice component of

alloys has been given by Lindenfeld and Pennebaker [105].

Finally we must consider the scattering of lattice waves by crystal defects or imperfections.

As a rough rule we can state that extended defects contribute to the lattice thermal resistivity,  $1/k_T$ , most importantly at lowest temperatures (long waves), while point defects make their most important contribution to  $1/k_T$  at intermediate temperatures (at temperatures at or above the maximum in  $k_T$ ).

At low temperatures the frequency dependence of the phonon mean free path is reflected in the temperature dependence of the lattice conductivity. This is readily seen from equation (21), using expression (20) for  $C(\nu)$ . If  $l(\nu) \propto \nu^{-n} \propto T^{-n} \omega^{-n}$ ,

$$k_T \propto T^{3-n} \quad (27)$$

One can show (e.g., reference [3]) that the frequency dependence of  $l(\nu)$  depends on the geometry of the imperfections. For point defects  $n = 4$ , for line defects  $n = 3$ , and for sheets  $n = 2$ . Dislocations have a long-range strain field which is responsible for most of the scattering; for dislocations  $n = 1$ . The additional scattering by the core has a frequency dependence  $n = 3$ , but it is only a minor component.

At lowest temperatures dislocations are generally the most important imperfections scattering phonons. Since the frequency dependence is the same as that for scattering by electrons, these two resistive processes are additive, so that

$$\frac{1}{k_T} = W_T = W_{Te} + W_{Td} \propto T^{-2} \quad (28)$$

where  $W_{Te}$  is the lattice resistivity due to electrons, and  $W_{Td}$  is that due to dislocations. Typically,  $W_{Td}$  becomes comparable to  $W_{Te}$  for dislocation densities of the order of  $10^{10} \text{ per cm}^2$ . The lattice thermal conductivity of alloys is thus sensitive to the state of cold work, even though cold work in alloys produces only small fractional changes in  $\rho_0$  and thus only small changes in  $k_T$ .

Relation (27) does not hold, strictly speaking, for point defects, where  $n = 4$ , for if  $l \propto 1/\nu^4$  the integral (21) diverges at low frequencies, though the resulting resistivity frequently is of the form  $W_{Td} \propto T$ . Point defects must always be considered in conjunction with another resistive process—in conjunction with electron scattering at low temperatures, in conjunction with anharmonic interactions at high

temperatures, and the corresponding integrals (21) can only be evaluated numerically.

At low temperatures point defects lead to a departure from  $k_t \propto T^2$ , which becomes progressively larger as the temperature is increased. Around the maximum of  $k_t$  they depress the maximum and tend to flatten the curve, and at even higher temperatures they depress the conductivity and lead to a temperature dependence slower than  $T^{-1}$ . As a consequence of the properties of equation (21), their resistivity effect increases more slowly than linearly with point defect concentration, and in the limit of high temperatures and high defect concentration  $c$

$$k_t \propto [c(1-c)]^{-1/2} T^{-1/2} \quad (29)$$

This is a result of the fact that in equation (21) point defect scattering leaves the mean free path of the lowest frequencies essentially unchanged.

Quantitative estimates of the strength of the point defect scattering can be made in terms of the difference in mass from that of a normal site and in terms of the volume misfit [3].

Equation (29) assumes absence of any correlation between the position of point defects. Short range order can lead to different frequency dependences and corresponding changes in the temperature dependence of  $k_t$ . An extreme case is presented by systematic spatial variations in the concentration, such as, for example, those due to the strain field of dislocations. These impurity atmospheres lead to scattering that simulates the scattering due to the dislocations which control the atmospheres [129].

#### 4. OTHER CASES

With at least two types of carriers responsible for heat transport, and with each of them being limited by several possible mechanisms, thermal conductivity in metals and alloys [133-138] shows a wide range of differing behavior, which the present review cannot cover comprehensively. To add further to the variety, we have to consider other mechanisms of heat transfer and other mechanisms of resistivity, we also have to take into account cases of non-degeneracy of the electron gas, as are found in semimetals and semiconductors [139-158] of high carrier density, and finally we have to consider the thermal conductivity of superconductors [159-184].

Cooperative effects between the magnetic moments arranged in a regular lattice, leading to the concept of spin waves or magnons, can act both as a

new mechanism of heat transport, and at the same time as a resistive mechanism of electronic and phonon transport [185-198].

In order for the localized magnetic moments arranged in a lattice to act as a carrier of heat, the magnetic dipoles of neighboring atoms must be coupled together, either by direct magnetic forces, or by indirect effects carried by the medium of the free electrons. Both the electronic magnetic moments and the nuclear magnetic moments can be involved in principle, though the latter would become important only at much lower temperature. The exchange energy is probably a rough criterion of the upper limit of temperature at which these effects need be considered. The rare earths, with their wide variety of magnetic effects, form a group of materials where magnetic effects are no doubt important, but their thermal conductivities are not well understood at present. Ferromagnetic and nearly ferromagnetic transition metals are another group where magnetic effects would be important. We are unable at present to give a comprehensive treatment of this field compactly.

The thermal conductivity of semimetals is understood in principle, but in practice we do not always have enough information to interpret it. There are two practical complications: (1) the electronic and lattice components are comparable and (2) the electronic component does not follow the Wiedemann-Franz-Lorenz law except at low temperatures.

The departure from the Wiedemann-Franz-Lorenz law follows from the lack of degeneracy at higher temperatures, i.e., the density of states, the electron velocity, and the electron mean free path vary as a function of electron energy, and the fractional variation over an energy interval of the order of  $kT$  above and below the Fermi energy  $\zeta$  is not negligible, as it is in the case of the degenerate electron gas of good metals. Rules can be stated concerning the departure of the Lorenz number  $k_t/\sigma T$  from  $L_0$  in terms of the functional form of this variation [1]; the trouble is that in many cases we do not have enough information about the electronic band structure of the semimetals to benefit from these rules. Where we have this information, as in the case of graphite, we can make fairly good predictions about the Lorenz number; again in the limit of low temperatures we expect the Lorenz number to tend toward  $L_0$ .

Since in many cases we are not certain of the Lorenz number, we do not always know what

fraction of the total thermal conductivity should be ascribed to  $k_e$  and to  $k_r$ .

The same considerations apply to semiconductors. In many cases  $k_r$  predominates and semiconductors should be classed as insulators for purposes of thermal conductivity. There are, however, a few cases where  $k_e$  is not negligible at higher temperatures. Since in the case of intrinsic semiconductors the Lorenz number can exceed  $L_0$  by a substantial fraction, of order  $(G/\kappa T)^2/12$ , where  $G$  is the band gap energy, the Wiedemann-Franz-Lorenz law is not always a reliable guide in estimating the magnitude of  $k_e$ . When  $G/\kappa T$  is large,  $\sigma$  and  $k_e$  are in any case small, but for intermediate values this enhancement of the Lorenz number, also known as ambipolar diffusion, can be significant [1].

It remains to consider the thermal conductivity of superconductors. Below the transition temperature  $T_c$ , a fraction of the conduction electrons rearrange themselves into an ordered state, of zero entropy, which can carry current without electrical resistance, and also exhibits special magnetic properties. This phenomenon of superconductivity can be quenched by a magnetic field; in many cases the required critical field is quite moderate. It is thus possible to measure the thermal conductivity below  $T_c$ , not only in the superconducting, but also in the normal state.

In the superconducting state, the electronic component  $k_e$  is reduced. The ratio  $k_{es}/k_{en}$  is a function of  $T/T_c$ ; near  $T_c$  it also depends on the degree to which  $W_{en} = 1/k_{en}$  is composed of

intrinsic or defect-induced resistance. At sufficiently low temperatures, where the latter mechanism dominates,  $k_{es}$  decreases exponentially as a function of  $T_c/T$ .

While  $k_e$  is reduced,  $k_r$  is often increased in the superconducting state, because one mechanism of lattice thermal resistance, the scattering of phonons by electrons, is reduced. Well below the transition temperature,  $k_{rs}$  would be similar in character to  $k_r$  of a dielectric solid of corresponding mechanical properties, and be controlled by external boundaries, by lattice imperfections, and, in the case of polycrystalline aggregates, by the grain size.

In the case of pure superconductors, the total thermal conductivity  $k_s$  decreases below  $k_n$ , joining  $k_n$  at  $T_c$ ; with decreasing temperature it first falls rapidly, reaches a minimum (when  $k_r$  becomes appreciable), and then increases again. In that region it is very structure sensitive. Finally  $k_s$  reaches a maximum, and decreases again with decreasing temperature. At this lowest temperature  $k_s$  should depend on external or grain size, and may also be influenced by dislocations.

Many superconductors form an intimate mixture of normal and superconducting regions, either for geometric reasons or because (in superconductors of the second kind) such a mixed state is inherently more stable. The phase boundaries will then also act to limit the carrier mean free paths, and rather complex dependences on the history of the specimen may ensue. These phenomena are at present only partly understood.



# Experimental Determination of Thermal Conductivity

## 1. INTRODUCTION

In the experimental determination of the thermal conductivity of solids, a number of different methods of measurement are required for different ranges of temperature and for various classes of materials having different ranges of thermal conductivity values. A particular method may thus be preferable over the others for a given material and temperature range, and no one method is suitable for all the required conditions of measurement. The appropriateness of a method is further determined by such considerations as the physical nature of the material, the geometry of samples available, the required accuracy of results, the speed of operation, and the time and funds entailed.

The various methods for the measurement of thermal conductivity fall into two categories: the steady-state and the nonsteady-state methods. In the steady-state methods of measurement, the test specimen is subjected to a temperature profile which is time invariant, and the thermal conductivity is determined directly by measuring the rate of heat flow per unit area and temperature gradient after equilibrium has been reached. In the nonsteady-state methods, the temperature distribution in the specimen varies with time, and measurement of the rate of temperature change, which normally determines the thermal diffusivity, replaces the measurement of the rate of heat flow. The thermal conductivity is then calculated from the thermal diffusivity with a further knowledge of the density and specific heat of the test material.

The primary concern in most methods of measurement is to obtain a controlled heat flow in a prescribed direction such that the actual boundary conditions in the experiment agree with those assumed in the theory. Theoretically, the simplest method to obtain a controlled heat flow is to use a specimen in the form of a hollow sphere with a

heater in the center. The heat supplied by the internal heat passes through the specimen in a radial direction without loss. However, in reality it is very difficult to fabricate a spherical heater which produces uniform heat flux in all radial directions. It is also difficult to fabricate spherical specimens and to measure the heat input and the temperature gradient in this experimental arrangement.

A more commonly used method of controlling heat flow in the prescribed direction is the use of guard heaters (combined with thermal insulation in most cases) so adjusted that the temperature gradient is zero in all directions except in the direction of desired heat flow. In most methods of measuring thermal conductivity, a cylindrical specimen geometry ranging from long rod to short disk is utilized, and the heat flow is controlled to be in either the longitudinal (axial) or the radial direction. Thus, most methods can be subdivided into longitudinal and radial heat flow methods, as discussed in more detail later.

Experimental study of the thermal conductivity of solids was started in the eighteenth century. Benjamin Franklin [199] seems first to have pointed out, in 1753, the different ability of different materials "to receive and convey away the heat." He observed materials such as metal and wood to be good or poor conductors of heat by the degree of coldness felt when touched. Fordyce [200] pioneered in 1787 with experiments on the "conducting powers" of pasteboard and iron. The first steady-state comparative method for the measurement of the thermal conductivity of solids was suggested by Franklin and carried out by Ingen-Hausz as reported in 1789 [201]. This method was improved by Despretz as reported in 1822 [202], and Despretz's method was later used by Wiedemann and Franz as reported in 1853 [17] to determine the relative thermal conductivity of a number of metals, leading to the postulation of the Wiedemann-Franz law. Since the

first steady-state *absolute* method was reported in 1851 by Forbes [203, 204] (see also [205, 206]) and the first nonsteady-state *absolute* method was reported in 1861 by Ångström [207], a number of different methods and their variants have been developed over the years. Several general surveys [208–218] are available for the experimental developments of the methods. The mathematical theories of the methods have been reviewed in several books [219–223].

In the sections that follow, the major methods and the extent of their applicability will be briefly described and discussed. For finer details of experimental designs and techniques, the reader is referred to the references given to the individual methods.

In the category of steady-state methods, we will discuss the longitudinal heat flow method, the Forbes' bar method (which is a quasi-longitudinal heat flow method), the radial heat flow method, the direct electrical heating method, the thermoelectrical method, and the thermal comparator method. In the longitudinal and radial heat flow methods, a distinction is made between absolute and comparative methods according to the means of measuring the heat flow. In an absolute method, the rate of heat flow into a specimen is directly determined, usually by measuring the electrical power input to a heater at one end of the specimen. The rate of heat flow out of a specimen may be measured with a flow calorimeter or boil-off calorimeter. With the latter the rate of heat flow is determined by the boil-off rate of liquid, such as water, of known heat of vaporization, while with the former it is determined by the flow rate and temperature rise of a circulating liquid, such as water, of known heat capacity. In a comparative method, the rate of heat flow is usually calculated from the temperature gradient over a reference sample of known thermal conductivity, which is placed in series with the specimen and in which, hopefully, the same heat flow occurs. The methods are further subdivided according to the various specimen geometries.

In the category of nonsteady-state methods, we will discuss the periodic and the transient heat flow methods. According to the direction of heat flow, each of them is also subdivided into longitudinal and radial heat flow methods. Within the transient heat flow methods, we will discuss also the flash method (which is a variant of the longitudinal heat flow method), the line heat source and probe methods (which are variants of the radial heat flow method),

the moving heat source method, and two comparative methods.

It is worth noting that some of the methods discussed below are not suitable for good conductors. They may be suitable for poor conductors such as semiconductors and some for materials such as metallic powders and insulators.

## 2. STEADY-STATE METHODS

### A. Longitudinal Heat Flow Methods

In the longitudinal heat flow methods, the experimental arrangement is so designed that the flow of heat is only in the axial direction of a rod (or disk) specimen. The radial heat loss or gain of the specimen is prevented or minimized and evaluated. Under steady-state conditions and assuming no radial heat loss or gain, the thermal conductivity is determined by the following expression from the one-dimensional Fourier-Biot heat-conduction equation [224, 225]:

$$k = \frac{-q\Delta x}{A\Delta T} \quad (30)$$

where  $k$  is the average thermal conductivity corresponding to the temperature  $(\frac{1}{2})(T_1 + T_2)$ ,  $\Delta T = T_2 - T_1$ ,  $q$  is the rate of heat flow,  $A$  is the cross-sectional area of the specimen, and  $\Delta x$  is the distance between points of temperature measurements for  $T_1$  and  $T_2$ . The different variants of this method are discussed separately below.

#### a. Absolute Methods

(i) *Rod Method.* This method is suitable for good conductors and for all temperatures except for very high temperatures. In fact, this method has been used for almost all measurements below room temperature. The specimen used is in the form of a relatively long rod so as to produce an appreciable temperature drop along the specimen for precise measurement. A source of heat at a constant temperature is supplied at one end of the rod and flows axially through the rod to the other end, where a heat sink at a lower constant temperature is located. The radial heat loss or gain of the rod should be negligible. In order to calculate the thermal conductivity from equation (30), it is necessary to measure the rate of heat flow into and/or out of the rod, the cross-sectional area, the temperatures of at least two points along the rod, and the distance between points of temperature measurements.

For measurements at cryogenic temperatures, radial heat loss does not constitute a serious problem, and thermal insulation and guard heaters are normally not necessary. The measurement is usually made under high vacuum to prevent gas conduction and convection, and a radiation shield surrounding the specimen may be used to minimize radiation losses. The heat is supplied to one end of the specimen by a heating coil of fine resistance wire (which may be wound directly onto the specimen to eliminate contact resistance between heater and specimen) or by a carbon resistor attached to the end. The temperatures may be measured by gas thermometers, vapor-pressure thermometers, thermocouples, resistance thermometers, or magnetic-susceptibility thermometers. General reviews of the low-temperature measurements and experimental techniques have been presented by White [226, 227]. For details of some of the useful low-temperature apparatus the reader may consult references [228-239].

For measurements at high temperatures, heat loss becomes a serious problem because radiant heat transfer increases rapidly with temperature. To prevent radial heat losses, a guard tube surrounding the specimen with controlled guard heaters may be utilized. Insulating powder is usually used to fill the space between the rod specimen and the guard tube, which should have the same temperature distribution along it as does the rod specimen. In fact, as early as 1887, Berget [240, 241] started the use of a guard ring surrounding (and with the same temperature distribution as) the specimen to prevent heat losses.

The rate of heat flow into the specimen may be determined by measuring the power input to a guarded electrical heater at the free end of the rod specimen [242-244], or by measuring the heat flow out of the specimen with a water-flow calorimeter at the low temperature end [245], or by both [246-248]. Temperature measurements are made usually with thermocouples. In order to get correct temperature measurements and to minimize heat conduction along thermocouple leads, the thermocouples should be made of fine wires of low-conductivity alloys, and the leads from the junction should be along isothermal lines.

This method, as used for measurements at high temperatures, has been comprehensively reviewed and discussed by Laubitz [249] and Flynn [250]. Systematic errors in measurements caused by the effects of heat losses, thermal contact resistance, poor thermocouple contacts, and temperature drift

have been analyzed by Bauerle [251].

A variation of this method has been used [252-254] in which the specimen heater is located in a cavity at the center of the rod specimen and a heat sink is at each end. A mean value of the temperature gradient established towards the two ends is used for the thermal conductivity calculation.

(ii) *Plate (or Disk) Method.* This method is suitable for poor conductors such as semiconductors and for low-conductivity materials such as compacted metallic powders and insulators. It is similar to the rod method except for the specimen length to width ratio being greatly reduced to a small fraction. This specimen geometry is favorable for measuring poor conductors, because, the smaller the length to width ratio, the smaller is the ratio of lateral heat losses to the heat flow through the specimen, and the shorter is the equilibrium time. The size of specimen used in various apparatus designed for different kinds of materials varies greatly. For apparatus designed to measure semiconductors, the specimen used may be about 1-cm wide [255], while the apparatus for measuring less homogeneous insulating or refractory materials may require a specimen of over 1 foot in width [256].

In this method, the thermal conductivity is also given by equation (30). The rate of heat flow may be determined by the electrical power input to a guarded heater [256-258], by a guarded water-flow calorimeter [259], by a boil-off calorimeter [260-263], or by a heat flow meter [264]. Temperature measurements are made generally with thermocouples inserted in the specimen or embedded in grooves on the specimen surfaces, depending on the materials tested. Lateral heat losses may be prevented either by utilizing guard heaters or by using a large specimen, of which only a relatively small central area is used for measurement. In the first detailed mathematical analysis of the plate method reported in 1898, Peirce and Willson [265] found already that, if the radius of the specimen is five times larger than that of the central test section whose thickness equals its radius, the temperature at any point within the central test section would not sensibly differ from the temperature at the corresponding point in an infinite disk of the same thickness and same face temperatures. Further mathematical analyses of the errors due to lateral heat loss in guarded hot plate apparatus have been given in [266-268].

Detailed descriptions of recent apparatus for measurements at cryogenic temperatures can be

found in the articles collected in [269], and for measurements at high temperatures in [270]. A comprehensive review of the plate method has been given in [271]. A description of the NBS steam calorimeter apparatus and some useful discussions on this method have also been given in [250].

There are two main kinds of experimental arrangements for the absolute plate (or disk) method: the single-plate system and the twin-plate system. The single-plate system [255, 258-265] requires only one specimen, which is placed between a hot plate and a cold plate, while the twin-plate system [256, 257] requires two similar specimens to be sandwiched between a hot plate in the middle and two cold plates on the outside. The plate method employing the single-plate system was probably first used by Clément, whose experiment on copper was cited by Péclet [272] in 1841. Péclet also used this method to measure the thermal conductivity of copper, and both of them obtained erroneous results. Later improvements on this method have been made by Peirce and Willson [265] and Lees [273] among others. The idea of a twin-plate system was developed by Lees [273] in 1898, but he did not actually adopt the twin-plate system for his plate method in the series of measurements as reported in [273]. However, he used the twin-plate system in his experiments on the effect of pressure on thermal conductivity reported in 1899 [274]. Great improvement on the plate method employing the twin-plate system was made by Poensgen [257] in 1912, who introduced the guard-ring heater to the system as the prototype of the modern guarded hot-plate apparatus.

#### b. Comparative Methods

In the earliest steady-state comparative method suggested by Franklin and carried out by Ingen-Hausz [201] as reported in 1789, rods of various metals were coated with wax and heated at one end to a common temperature in a bath of hot water or oil. The wax melted over a greater distance on a rod of better conducting material, and under steady-state conditions the ratio of the conductivities of the rods is roughly proportional to the squares of these distances. The modern comparative methods are divided-rod (or cut-bar) method and the comparative plate method as discussed below.

(i) *Divided-Rod (or Cut-Bar) method.* The divided-rod method was originated by Lodge [275] in 1878 and later used by Berget [276], Lees [277], and many others. In this method a reference sample

(or samples) of known thermal conductivity is placed in series with the unknown specimen with hopefully the same rate of heat flow through both the reference sample and the specimen. Under such ideal conditions, the thermal conductivity of the specimen is given by

$$k = k_r \frac{A_r(\Delta T/\Delta x)_r}{A(\Delta T/\Delta x)} \quad (31)$$

where the subscript  $r$  designates the reference sample.

This method may be divided into two distinct groups: the "long-specimen" type [276, 278, 279] for measuring the thermal conductivity of good conductors, and the "short specimen" type [275, 277, 280-283] for measuring poor conductors.

Comparative methods have the advantages of simpler apparatus, easier specimen fabrication, and easier operation. Their disadvantages include additional measurement errors due to the required additional measurements of temperatures and thermocouple separations, difficulty in matched guarding, and lower accuracy due to the additional uncertainty in the conductivity of the reference sample, due to the conductivity mismatch between specimen and reference sample, and due to the interfacial thermal contact resistance. These have been carefully analyzed by Laubitz [249] and Flynn [250]. Flynn [250] has pointed out that the ASTM standard cut-bar method C408-58 [282] is not well designed, and the data obtained by using this method can be subject to large errors.

(ii) *Plate (or Disk) Method.* This comparative method is suitable for poor conductors and insulators and is similar to the divided-rod method in principle except that the specimen and the reference samples are now flat plates (or disks) sandwiched between a hot and a cold plate. Christiansen [284] was the first to report in 1881 the use of this type of comparative method in which he compared the thermal conductivity of liquids with that of air. Peirce and Willson [265] used this method to measure the thermal conductivity of marble slabs with glass plates as reference material for comparison. Sieg [285] employed the guard ring in his apparatus to prevent lateral heat loss.

#### c. Combined Method

In using a "combined" method, the apparatus combines the features of both absolute and comparative methods. The rate of heat flow is determined both through a reference sample placed in series with the specimen and simultaneously by a

water-flow calorimeter [286-288] or by measuring the electrical power input to a heater [289]. In the measurements reported in [289], a "dual combined" method was employed in which a heater is located at the center of the divided rod between two short specimens with two longer reference samples at the two ends which are cooled by flowing water.

### B. Forbes' Bar Method

Forbes' original method [203-206] consists of two separate experiments. The first was termed by Forbes the *statical*, and the second the *dynamical*, or cooling experiment. In the *statical* experiment a square wrought iron bar with 1.25-inch side and 8 feet long was heated at one end by molten lead or solder at a fixed high temperature, and the steady-state temperature distribution along the bar was determined with the surface of the bar losing heat by convection and radiation to a constant-temperature environment. In the *dynamical* or cooling experiment, a similar bar but only about 20 inches long was cooled in the same environment from a high uniform temperature, and the rate of heat loss was determined. From these two experiments, the thermal conductivity may be computed as follows.

Replacing  $\Delta x/\Delta T$  in equation (30) by  $dx/dT$ , differentiating the resulting equation with respect to  $x$ , and rearranging gives

$$k = \frac{l}{A} \frac{dq}{dx} \frac{1}{d^2T/dx^2} \quad (32)$$

The statical experiment provides values for  $d^2T/dx^2$ , and the heat loss per unit time per unit length of the bar in the cooling experiment is

$$\frac{dq}{dx} = AC \frac{dT}{dt} \quad (33)$$

where  $dT/dt$  is the measured cooling rate and  $C$  the specific heat per unit volume.

Hogan and Sawyer [290] have improved this method so that it is not necessary to know the specific heat of the material. They used a thin long rod enclosed in an isothermal furnace. Radial heat loss from the specimen was determined by passing an electric current through the specimen and measuring the electric power required to maintain it at a temperature slightly above that of the furnace. This replaces Forbes' cooling experiment, and it is not necessary to know the specific heat since a steady-state condition is prevailing.

Hogan and Sawyer's method was further improved by Laubitz [291]. In his comprehensive

review Laubitz [249] has discussed in detail the generalized Forbes' bar method, including the other major variants currently in use [292-294].

### C. Radial Heat Flow Methods

There are several different types of apparatus all employing radial heat flow. The classification is mainly based upon specimen geometry. In the following we will briefly describe the cylindrical, spherical, ellipsoidal, concentric sphere, concentric cylinder, and plate methods. The reader is referred to the references given for the individual methods for finer details. A comprehensive review of radial heat flow methods has been made by McElroy and Moore [295].

#### a. Absolute Methods

(i) *Cylindrical Method.* The cylindrical method uses a specimen in the form of a right circular cylinder with a coaxial central hole, which contains either a heater or a heat sink, depending on whether the desired heat flow direction is to be radially outward or inward. The use of this method was first reported by Callendar and Nicolson [296] in 1897 for measuring the thermal conductivity of cast iron and mild steel. The cylindrical specimens used were 5 inches in diameter and 2 feet long with 1-inch coaxial holes heated by steam under pressure. The outside of the cylinder was cooled by water circulating rapidly in a spiral tube. Niven [297] in 1905 also used the radial heat flow method for measurements on wood, sand, and sawdust. His method is close to the so-called hot-wire method developed by Andrews [298] in 1840 and Schleiermacher [299] in 1888 for measurements on gases. Kannuliik and Martin [300] used the hot-wire method for measurements on powders as well as on gases.

In the early experiments and also in many later designs [301-304], end guards are not employed. The effect of heat losses from the ends of the specimen is minimized by using a long specimen and monitoring the electric power within only a small section of the specimen away from the ends.

The guarded cylindrical method employing end guards at both ends of the specimen to prevent axial heat losses was developed by Powell [305] and first reported in 1939 for measurements on Armco iron at high temperatures. In the guarded cylindrical method the specimen is generally composed of stacked disks with a coaxial central hole containing either a heater or a heat sink. Temperatures within the specimen are measured either by thermocouples or by an optical pyrometer. For details of some of

the useful apparatus employing the guarded cylindrical method, the reader may consult references [295, 305-310].

The thermal conductivity is calculated from the expression

$$k = \frac{q \ln(r_2/r_1)}{2\pi l(T_1 - T_2)} \quad (34)$$

where  $l$  is the length of the central heater and  $T_1$  and  $T_2$  are temperatures measured at radii  $r_1$  and  $r_2$ , respectively.

Hoch *et al.* [311] have developed a quasi-radial heat flow method in which the specimen in the form of a disk or short cylinder is heated at its convex cylindrical surface in high vacuum by means of high frequency induction and is losing heat from its flat circular end faces by radiation. In this method the inward flow of heat from the cylindrical surface, at which the generation of heat is localized, into the interior of the specimen is, of course, not strictly radial, and the temperature gradient of the flat circular end faces along the radius is related to the thermal conductivity. The theory of this method has recently been revised by J. Vardi and R. Lemlich (to be published in the *Journal of Applied Physics* in 1970).

(ii) *Spherical and Ellipsoidal Methods.* In a spherical method, the heater is completely enclosed inside the specimen which is in the form of a hollow sphere. The heat supplied by the internal heater passes through the specimen radially without loss. Theoretically, this method is ideal. However, there are a number of practical difficulties such as difficult fabrication of a spherical heater which produces uniform heat flux in all radial directions, difficult fabrication of spherical specimens, difficult positioning of thermocouples along spherical isotherms, etc., which have prevented this method from being popular. Laws, Bishop, and McJunkin [312] seem the first to have used this method on solids (not loose-filled materials). A detailed description of a modern design may be found in [301]. The thermal conductivity is calculated from the expression

$$k = \frac{q(1/r_1 - 1/r_2)}{4\pi(T_1 - T_2)} \quad (35)$$

The ellipsoidal method is similar to, but has some advantages over, the spherical method. It was developed by a group of researchers at MIT [313-315]. The major advantage of using a specimen in the form of an ellipsoid instead of a sphere is that

the isothermal surfaces near the plane of the minor axes of an ellipsoid are rather flat so that straight thermocouple wires can be used without ill effect. If  $a$  is half the focal length of the ellipsoid and  $T_1$  and  $T_2$  are temperatures measured at respectively two radii  $r_1$  and  $r_2$  on the minor axis, the thermal conductivity is determined by the expression

$$k = \frac{q}{8\pi a(T_1 - T_2)} \times \ln \left( \frac{\sqrt{(a^2 + r_2^2)} - a}{\sqrt{(a^2 + r_2^2)} + a} \cdot \frac{\sqrt{(a^2 + r_1^2)} + a}{\sqrt{(a^2 + r_1^2)} - a} \right) \quad (36)$$

Despite the aforementioned advantage, the ellipsoidal method is also rarely used due to the other experimental difficulties common to both the ellipsoidal and spherical methods.

(iii) *Concentric Sphere and Concentric Cylinder Methods.* Concentric sphere and concentric cylinder methods are used mainly for measurements on powders, fibers, and other loose-filled materials. The specimen is filled in the space between two concentric spherical (or cylindrical) shells, with the inner sphere (or cylinder) being a heater or a heat sink. In a concentric cylinder apparatus, end guards are usually used to prevent axial heat flow.

A concentric sphere method was first used by Péclet [316] and reported in 1860 with the inner sphere filled with hot water as heater. However, a steady-state condition was not achieved in his pioneering measurements. Later Nusselt [317] succeeded in using this method for measurements on insulating materials with an electric heater installed inside the inner spherical shell. A modern apparatus using a boil-off calorimeter in the inner sphere was described in [318].

A concentric cylinder method was used by Stefan [319] and reported in 1872 for the measurements on gases. It was later adopted for measuring loose-filled materials. Reference [320] describes a modern apparatus employing a guarded boil-off calorimeter inside the inner cylinder. Recently, Flynn and Watson [321] used a concentric cylinder method to measure the high-temperature thermal conductivity of soil.

(iv) *de Sénarmont's Plate Method.* de Sénarmont [322-326] in 1847-48 used a radial heat flow plate method to determine the anisotropy in thermal conductivity of crystalline substances. However, this method does not yield absolute values of thermal conductivity, and furthermore, the axial heat loss is not prevented.

In his method, a thin plate of the sample was coated with a thin film of white wax; heat was applied at a central point by means of a hot, thin silver tube tightly fitted in a hole at the center of the plate. The wax melted around the region where heat was supplied and the bounding line of the melted wax was the visible isotherm, the shape of which indicated the variation of thermal conductivity in the different directions.

If the substance is isotropic, the bounding curve of the melted wax is a circle, whereas for anisotropic substances, this curve is elliptical. In such a case, the ratio of the two thermal conductivities  $k_a$  and  $k_b$  along the two axes  $a$  and  $b$  of the ellipse is given by the expression

$$\frac{k_a}{k_b} = \left(\frac{a}{b}\right)^2 \quad (37)$$

Powell [327] has modified the method in his simple test for anisotropic materials. In testing gallium, he cooled a slice of crystal locally by means of a piece of solid carbon dioxide and observed the contours of the dew and frost areas which formed around the cooled zone. For testing graphite, he followed de Sénarmont's original method but the surface of the plate used was covered with frost by precooling instead of being coated with wax.

#### b. Comparative Methods

(i) *Concentric Cylinder Method.* This method has been used for measurements on some special materials such as those that are radioactive or reactive [328-330] and not for ordinary materials, because it does not have any major advantage over the absolute method. A typical apparatus of this kind consists of a cylindrical specimen which is surrounded by a concentric cylindrical reference sample of known thermal conductivity. A coaxial central hole in the specimen contains a heat source, which produces heat flowing radially through both the specimen and the reference sample. The advantage of using this method for measuring radioactive or reactive materials is that the reference sample which encloses the specimen serves also as a means of containment. The thermal conductivity is determined from the expression

$$k = k_r \frac{(T_3 - T_4) \ln(r_2/r_1)}{(T_1 - T_2) \ln(r_4/r_3)} \quad (38)$$

where  $T_1$  and  $T_2$  are two temperatures measured in the specimen at two radii  $r_1$  and  $r_2$ , respectively, and

$T_3$  and  $T_4$  in the reference sample, at  $r_3$  and  $r_4$ , respectively.

(ii) *Disk Method.* Robinson [331] developed a method, which he termed the "conductive-disk method," for comparative measurements on insulators. This method employs inward radial heat flow from a heater at the circular edge of a disk of suitable conductive reference material sandwiched between two like specimens, which are in turn sandwiched between two circular cold plates at a constant lower temperature. However, the heat flow in this case is not strictly radial, since, as the heat flows radially in the conductive disk toward the center, it flows also from the disk through the specimens to the cold plates. As a result, the steady-state temperature of the disk decreases toward its center, and the rate of decrease depends on the thermal conductivity of the specimens. Robinson obtained an expression for calculating the thermal conductivity of the specimens from the known thermal conductivity and thickness of the disk and from the temperatures of the cold plates and of the disk at its center and at a suitable radius.

#### D. Direct Electrical Heating Methods

In direct electrical heating methods, the specimen is heated directly by passing an electric current through it. These methods are therefore limited to measurements on reasonably good electrical conductors. Furthermore, they usually yield thermal conductivity in terms of electrical conductivity rather than directly. However, direct electrical heating methods have also certain advantages over other methods. They offer a means of easily attaining very high temperatures, use simpler apparatus and experimental techniques than other methods at high temperatures, use relatively small specimens, require relatively short time to reach equilibrium, and also offer the possibility of concurrent determinations of a number of physical properties on the same specimen. According to specimen geometry, these methods fall into two major categories: cylindrical rod and rectangular bar. They will be briefly discussed below. Comprehensive reviews [332-334] on direct electrical heating methods are available.

The thermoelectrical method to be discussed later involves also the direct passage of an electric current through the specimen. However, in that method the specimen is heated (and cooled) by the Peltier effect which is totally different from the Joulean heating responsible for maintaining the

specimen temperature in the direct electrical heating methods discussed here. It is therefore preferable to discuss the thermoelectrical method separately in another section.

#### a. Cylindrical Rod Methods

The direct electrical heating methods in this category involve heating specimens in the form of rods, thin wires, or tubes by the passage of regulated electric current, and measuring potential drops and temperatures for the calculation of thermal conductivity.

There are many different techniques and variants that have been employed over the years since Kohlrausch [335-338] first developed this method. The different variants may be divided into three categories as discussed below.

(i) *Longitudinal Heat Flow Method.* In this method the rod is well insulated or guarded to prevent radial heat losses so that the Joule heat generated in the specimen flows to the two ends. This is the method originally developed by Kohlrausch [335-338]. If the two ends of the rod are held at the same temperature and assuming that in a small temperature range the thermal and electrical conductivities are independent of temperature, the thermal conductivity is given by the simple relation

$$k = \frac{1}{8\rho} \frac{(V_1 - V_3)^2}{(T_2 - T_1)} \quad (39)$$

where  $\rho$  is the electrical resistivity,  $V_1$  and  $V_3$  are the electrical potentials at locations 1 and 3 on the specimen which are at equal and opposite distances from the midpoint 2, and  $T_1$  and  $T_2$  are temperatures at locations 1 and 2. This method was first used for actual measurements by Jaeger and Diesselhorst [339]. A variant of it has been used by Mikryukov [332]. The so-called "necked-down-sample method" [340] may also be considered as a longitudinal heat flow method.

(ii) *Radial Heat Flow Method.* This method uses a thick rod or tube and allows radial heat transfer. Under steady-state conditions, the Joule heat generated in the specimen at regions remote from the ends flows radially to the surface and is then transferred by convection and radiation to the surroundings. This method was first suggested by Mendenhall and applied by Angell [341]. In the case of a cylindrical rod specimen and assuming that in a small temperature range the thermal and electrical conductivities are independent of temperature,

the thermal conductivity is given by the simple relation

$$k = \frac{EI}{4\pi l(T_1 - T_2)} \quad (40)$$

where  $I$  is the electric current,  $E$  is the electrical potential drop over a length  $l$  at the central region of the specimen, and  $T_1$  and  $T_2$  are the temperatures at the axis and surface, respectively, of the rod at the central region. These temperatures were too small for precise measurements on metals, but Powell and Schofield [342] used it for poorer conducting carbon and graphite, and they also took account of the variation of thermal and electrical conductivities with temperature.

(iii) *Thin-Rod-Approximation Method.* The general form of the present method uses a long thin filament heated electrically in vacuum and allows both longitudinal heat conduction and lateral heat transfer by radiation. The "thin-rod approximation" involves the assumption that the temperatures and potentials in all planes normal to the specimen axis are uniform, i.e., their differences in the radial direction are negligible. Worthing [343] first employed this method for measurements on U-shaped filaments at incandescent temperatures. There are many variants [344-358] of this method, all with more or less different experimental designs, mathematical assumptions, and/or computational techniques.

Taylor, Powell, and co-workers [354, 356-358] at TPRC have made improvements and advancements on this method. They have taken the Thomson effect into account, which had never been done before, and have included the temperature dependence of various physical properties. They used the general equation directly and their advanced computational techniques have eliminated the need for mathematical approximations and for matching certain experimental conditions.

It seems appropriate to mention the considerable discrepancies which have resulted from the data obtained by various workers, all of whom used different variants of the direct electrical heating method. One of the most recent of the TPRC papers [358] contains an interesting graphical presentation of all the determinations made on tungsten by these methods for the temperature range 1600 to 2800 K. Six of the fourteen groups of workers obtained results lying well above the recommended curve of Powell, Ho, and Liley [359], and one was well below it, the spread being of the order of 50 percent,



80 percent, and 70 percent at 1800, 2200 and 2600 K, respectively. The other seven had results within about 10 percent of the recommended curve, while the curve fitting the new results of [358] was some 3 to 5 percent below the recommended curve. Earlier reports [333, 356] had contained examples of similar discrepancies for other high-melting-point metals, such as molybdenum, stainless steel, and platinum. The main reasons for these differences include failure to measure accurately small temperature gradients at high temperature, failure to match boundary conditions, errors resulting from simplifying mathematical approximations, and the use of temperature regions in which the thermal conduction term is small compared with the Joulean heating and radiation loss terms.

These have been quoted as examples of current experimental work at the TPRC, which became necessary because of the need to resolve some seriously discordant data and to gain further insight into their causes. The impression must not be given, however, that such discrepancies are confined to metals or to direct electrical heating methods. This is by no means the case, and the literature of heat conduction contains many examples of discordant results for all types of methods used. Titanium carbide, one of the materials dealt with in Volume 2, may be mentioned. The first determinations reported on titanium carbide by Vasilos and Kingery [360] to high temperatures showed the thermal conductivity to decrease from about  $0.2 \text{ W cm}^{-1} \text{ C}^{-1}$  at  $200 \text{ C}$  to  $0.1 \text{ W cm}^{-1} \text{ C}^{-1}$  at  $500 \text{ C}$  and  $0.04 \text{ W cm}^{-1} \text{ C}^{-1}$  at  $1000 \text{ C}$ . Two methods had been used: the divided-rod comparative method for a cube sample up to about  $800 \text{ C}$  and an ellipsoidal radial-flow method from about  $500$  to  $1100 \text{ C}$ . The former method gave results which were greater by from 30 percent to 20 percent over their common temperature range. In 1961 Taylor [361] used a better-substantiated radial heat flow method for cylindrical samples of titanium carbide, and found the thermal conductivity to increase linearly from  $0.38 \text{ W cm}^{-1} \text{ C}^{-1}$  at  $600 \text{ C}$  to  $0.47 \text{ W cm}^{-1} \text{ C}^{-1}$  at  $1600 \text{ C}$ .

These two sets of values, differing at about  $1000 \text{ C}$  by about one order of magnitude and having temperature coefficients of opposite sign, naturally aroused interest, and subsequent contributions by Laubitz [362], Hoch and Vardi [363], and Powell [364, 365] and the nonsteady-state measurements of Taylor and Morreale [366] all supported the higher values of Taylor [361]. It would seem that the higher thermal conductivity of titanium carbide led to

serious errors being associated with the method of Vasilos and Kingery, which were not apparent for substances of lower thermal conductivity. Incidentally, had the much simpler measurement of electrical resistivity been also made, the unusually low resultant Lorenz function should have provided warning that abnormal data were being obtained. It might well be added that the inclusion of electrical resistivity measurements on all possible occasions is a simple extra measurement which also serves to provide very useful information about the properties of the material under test and its behavior on temperature cycling.

The foregoing example also indicates that users of the data tables of these volumes should, in the absence of any analysis that has produced a curve of recommended values, tend to be critical of the values presented, until these are seen to be well supported by independent experiments, correlations, or by additional checks such as that of a reasonable Lorenz function.

An additional outcome of the current TPRC investigation has been the development of a method and of equipment capable of determining a large number of high-temperature physical properties [367]. Their multiple-purpose apparatus is the first operational model that can accurately measure the thermal conductivity, electrical resistivity, total and spectral hemispherical emittance, Thomson coefficient, and Lorenz function on one and the same specimen. This apparatus can also measure the specific heat, enthalpy, thermal diffusivity, thermal expansion, Seebeck coefficient, Peltier coefficient, and Richardson coefficient. The merit of obtaining many different physical properties from one and the same specimen so as to permit meaningful quantitative cross-correlations between properties need not be emphasized here.

#### *b. Rectangular Bar Method*

This method was developed by Longmire [368] and is a geometrically-deformed variant of the radial heat flow method. The specimen used is in the form of a long rectangular bar. This special specimen geometry enables all temperature measurements to be made on the surface of the specimen. As the specimen is heated electrically in vacuum, the heat loss by radiation establishes a radial temperature gradient, and the temperature at the center line of the wider surface of the rectangular bar will be higher than that at the center line of the narrower surface. From measurements of these two

temperatures and the electrical conductivity and total hemispherical emittance of the bar, the thermal conductivity can be calculated using the equation derived by Longmire.

Longmire's method was improved by Pike and Doar [369-371] both in mathematical analysis and in experimental techniques. They further extended this method to the determination of anisotropy in thermal conductivity.

#### E. Thermoelectrical Method

The thermoelectrical method was developed by Borelius [372] and reported in 1917 for the combined measurement of the Peltier heat and thermal conductivity of the same material, and is particularly applicable to the measurements on thermoelectric materials.

In this method, the specimen is held between metallic contacts through which a small direct electric current is passed. Peltier heating thus occurs at one end of the specimen and Peltier cooling at the other end, which establishes a temperature gradient along the specimen. Under steady-state conditions, the rate of Peltier heat generation at the hot end is just balanced by the rate of heat conduction from the hot to the cold end. Thus the thermal conductivity can be calculated from the rate of Peltier heat production  $\pi I$  ( $\pi$  being the Peltier coefficient), the temperature difference between the ends  $\Delta T$ , the cross-sectional area  $A$ , and the length  $l$  by the expression

$$k = \frac{\pi I l}{A \Delta T} \quad (41)$$

Since  $\pi = ST$ ,  $S$  being the Seebeck coefficient,  $\pi$  can be determined by measuring the Seebeck coefficient from the potential difference between the ends after the temperature difference  $\Delta T$  is established.

When the direct electric current is passed through the specimen, Joulean heating will occur, of course. However, the Joulean heating effect can be made negligibly small in a good thermoelectric material by choosing the current small enough, because the Joule heat production is proportional to  $I^2$  while the Peltier heat production is proportional to  $I$ . The Thomson heat effect is generally small.

Borelius' method was used by Sedström [373, 374] for measurements on alloys. Some forty years later, Putley [375] and Harman [376, 377] reinvented this method. A recent apparatus is described in [378].

A transient thermoelectrical method was devel-

oped by Hérinckx and Monfils [379]. In this method a direct electric current is passed through the specimen and the time dependence of the resulting potential drop across the specimen is observed. The thermal conductivity can be derived from the shape and asymptote of this potential drop versus time curve provided that the Seebeck coefficient is known.

#### F. Thermal Comparator Method

The thermal comparator method was developed by Powell [380-383] and is a simple comparative method for the rapid, easy measurement of thermal conductivity.

The essential part of the thermal comparator is an insulated probe with a projecting tip. The probe is integral with a thermal reservoir held at a temperature about 15 to 20 degrees above room temperature. A surface thermocouple is mounted at the tip of the probe and is differentially connected to the thermal reservoir for the measurement of the temperature difference between the reservoir and the tip.

In operation, the probe is gently placed on the surface of the test material. Upon contact of the probe tip of known thermal conductivity  $k_1$  and originally at temperature  $T_1$  with the surface of the test material of thermal conductivity  $k_2$  and at room temperature  $T_2$ , the temperature of the probe tip drops quickly to an intermediate temperature,  $T$ , given by the expression

$$T_1 - T = (T_1 - T_2) \left( \frac{k_2}{k_1 + k_2} \right) \quad (42)$$

This temperature difference is registered by the emf reading of the differential thermocouple after a brief transient period (1 to 2 seconds) has elapsed.

From the emf readings of tests on a series of reference samples of known thermal conductivity, a calibration curve is obtained, and the thermal conductivity of an unknown specimen can thus be determined from the emf reading through the calibration curve.

Powell [384] has made a comprehensive review on this method. Some subsequent developments are discussed in [385]. The *thermal comparator* has been developed by TPRC as an instrument [385] for the rapid determination of the thermal conductivity of solids and liquids and is commercially available from The McClure Park Corp., West Lafayette, Indiana.

### 3. NONSTEADY-STATE METHODS

In nonsteady-state methods, the temperature distribution in the specimen varies with time. The rate of temperature change at certain positions along the specimen is measured in the experiment, and no measurement of the rate of heat flow is required. These methods normally determine the thermal diffusivity, from which the thermal conductivity can be calculated with an additional knowledge of the density and specific heat of the test material. Nonsteady-state methods fall into two major categories, the periodic and the transient heat flow methods, as briefly discussed below. These methods have been comprehensively reviewed by Danielson and Sidles [386], and will be dealt with in Volume 10 of the TPRC Data Series.

#### A. Periodic Heat Flow Methods

In periodic heat flow methods, the heat supplied to the specimen is modulated to have a fixed period. The resulting temperature wave which propagates through the specimen with the same period is attenuated as it moves along. Consequently, the thermal diffusivity can be determined from measurements of the amplitude decrement and/or phase difference of the temperature waves between certain positions in the specimen. In most of the periodic heat flow methods, heat flow is in the longitudinal (axial) direction. However, methods with heat flow in the radial direction have also been used.

##### a. Longitudinal Heat Flow Method

The periodic heat flow method was first developed by Ångström [207, 387] and reported in 1861. In his method a variable heat source capable of producing a sinusoidal temperature variation was attached to the center of a long thin rod specimen, and the temperatures as a function of time at two positions  $l$  apart towards the ends of the rod were measured. From these temperature-time measurements, the velocity,  $v$ , and the amplitude decrement  $\delta$  of the temperature wave can be determined for the calculation of thermal diffusivity. This method has been modified and improved by King [388] and others [389-391]. The thermal diffusivity may be calculated from the expression [390]

$$D = \frac{vl}{2 \ln \delta} \quad (43)$$

The Ångström method, which uses a long rod, has its limitations. In some cases, specimens in the

form of long rods may not be available, and in other cases, such as in the measurements on poor conductors at high temperatures, heat guarding to prevent lateral heat losses from a long rod may be difficult. Consequently, methods using specimens in the form of a small plate or disk have been developed [392-394].

##### b. Radial Heat Flow Method

In this method, the specimen in the form of a cylinder is heated by a heat source capable of producing a periodical temperature variation either at the axis or at the circumference, and the radial temperature variations with time are measured. The thermal diffusivity may be calculated from the phase change of the temperature oscillations, or from the amplitude variation of the oscillations with frequency.

Tanasawa [395] used this method in 1935 for the measurements on humid materials. In his method, a sinusoidal temperature was produced on the surface of a cylindrical specimen, and the temperatures at different radial distances were measured for the calculation of thermal diffusivity.

Filippov and his co-workers have further developed a method of this type [396] and used it for the measurements on metals [397] and molten metals [398, 399] at high temperatures.

The nonsteady-state radial heat flow method has also been employed for measurements on insulators [400, 401].

#### B. Transient Heat Flow Methods

Transient heat flow methods, both longitudinal and radial, were first used by Neumann [402, 403] and reported in 1862. In his method, one end of a bar was heated by a flame until the temperature attained the equilibrium state. The flame was then suddenly removed and temperatures at two positions along the bar were measured as a function of time. Thermal diffusivity can then be calculated from these measurements. For the measurements on poor conductors, he used another method in which a cube or sphere was heated uniformly to a high temperature and then was allowed to cool in the air. The temperatures at the surface and at the center were measured as a function of time.

The modern transient heat flow methods have a wide variety. In the following a number of the major variants are briefly discussed.

*a. Longitudinal Heat Flow Method*

Similar to the longitudinal periodic heat flow method, the longitudinal transient heat flow method can also be subdivided into two major categories, those using a long rod and those using a small plate (or disk).

Methods in which one end of a long rod, which is initially at uniform temperature, is subjected to a short heating pulse have been developed [404, 405]. There are also methods in which steady heating is provided at one end of a rod and the temperatures as a function of time at two or more positions along the rod are observed [406-408].

Transient heat flow methods in which the specimen used is in the form of a small plate or disk have been developed by a number of workers [409-412].

*b. Flash Method*

Although the flash method is a variant of the longitudinal transient heat flow method using a small thin disk specimen geometry, it has a very special feature which makes it a class of its own. In the "flash" method, a flash of thermal energy is supplied to one of the surfaces of a disk specimen within a time interval that is short compared with the time required for the resulting transient flow of heat to propagate through the specimen. This method was developed by Parker, Jenkins, Butler, and Abbott [413] and reported in 1961.

In use, a heat source such as flash tube or laser supplies a flash of energy to the front face of a thin disk specimen, and the temperature as a function of time at the rear face is automatically recorded. The thermal diffusivity is given from the thickness of the specimen,  $l$ , and a specific time,  $t_{1/2}$ , at which the back-face temperature reaches half its maximum value by the expression

$$D = 1.37 l^2 / \pi^2 t_{1/2} \quad (44)$$

Other expressions for the calculation of thermal diffusivity have also been used.

Subsequent improvements on this method have been made [414, 415] by the application of corrections for the finite pulse-time effect and the radiation-loss effect.

*c. Radial Heat Flow Method*

As mentioned before, a radial heat flow method was used by Neumann [402, 403] for measurements on poor conductors. His specimens were of spherical shape.

In modern apparatus, specimens in the form of cylinders are used. A long cylindrical specimen, hollow or solid, which is initially at uniform temperature, is heated either at the axis or at the outer surface and the temperatures as a function of time at different radial distances are measured. In the methods developed by Ginnings [416] and by Cape, Lehman, and Nakata [417], cylindrical specimens were continuously heated at the outer surface.

Specimens in the form of hollow disks stacked on an axial heater with outer disks as end guards have been used by Carter, Maycock, Klein, and Danielson [418].

Although the line heat source and probe methods are also radial transient heat flow methods, they are quite different from other methods and will be discussed in a separate section below.

*d. Line Heat Source and Probe Methods*

The line heat source method was originally developed by Stalhane and Pyk [419] in 1931 and used for measurements on ceramic materials [420]. This method is suitable for the measurements on loose-filled materials such as powders.

In this method, a long thin heater wire which serves as a line heat source was embedded in a large specimen initially at uniform temperature. The heater is then turned on, which produces constant heat,  $q$ , per unit length and time, and the temperature at a point in the specimen is recorded as a function of time. The thermal conductivity is given by the expression

$$k = \frac{q}{4\pi(T_2 - T_1)} \ln \frac{t_2}{t_1} \quad (45)$$

where  $(T_2 - T_1)$  is the temperature difference at two times  $t_1$  and  $t_2$ . Subsequently, this method was also developed by van der Held and his co-workers [421, 422], and others.

The probe method is a more practical line heat source method in which the heat source is enclosed inside a probe for protection and for easy insertion into a specimen. This method was developed by Hooper and his co-workers [423, 424], and others. Blackwell [425, 426] has derived theoretical treatments for practical departures from a true line source, and in the discussion of a paper [427] dealing with the use of a probe method in connection with the routing of electric power cables, he advocated the use of very small thermistors as an alternative to thermocouples.

*e. Moving Heat Source Method*

The moving heat source method was developed by Rosenthal and his co-workers [428-430], and involves the establishment of a quasi-steady-state temperature distribution in a long tubular-shaped specimen heated by a moving localized heat source of constant intensity. As the heat source approaches and moves away, each point in the specimen is subjected to a temperature rise and fall. When the heat source passes over the specimen, the temperature at a point remote from the ends is recorded as a function of time. From this record, a curve of the logarithm of the temperature variation with time is made. The thermal diffusivity is given from the velocity of the heat source,  $v$ , and the slopes  $P_r$  and  $P_f$  on the rising and falling portions of the curve at the same temperature by the expression

$$D = \frac{v^2}{P_r + P_f} \quad (46)$$

*f. Comparative Method*

A comparative method employing transient heat flow was developed by Hsu [431, 432]. In this

method, two identical sets of composite blocks are used. Each set consists of a test specimen and a reference sample whose properties are known. Initially, the two sets are heated separately to uniform but different temperatures, and then they are suddenly brought into contact, with the two test specimens touching each other. The transient temperature at the contact plane between the test specimen and reference sample corresponding to a certain time is measured, and from this the thermal diffusivity of the specimen can be calculated.

Another transient-heat-flow comparative method has been used by Deem *et al.* [433] for the measurements on irradiated materials. The method of measurement is to place the lower ends of a specimen and a reference sample, which are of the same size and initially at room temperature, in molten tin maintained at a constant elevated temperature and then measure the times required for the upper ends to reach a predetermined intermediate temperature. The ratio of the thermal diffusivities is assumed directly proportional to the ratio of the two times measured for the specimen and the reference material.

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# Data Presentation and Related General Information

## 1. SCOPE OF COVERAGE

Presented in this volume are the thermal conductivity data for 69 elements, 172 nonferrous binary alloy systems, 80 nonferrous multiple alloy systems, 25 ferrous alloy systems, 60 intermetallic compounds, 16 mixtures of intermetallic compounds, and 13 miscellaneous alloys and mixtures. These data were obtained by processing over 2150 research documents on the thermal conductivity of metallic materials dated from around 1800 to 1967, of which 1000 contain usable data. Materials within each group are arranged in alphabetical order by name, as listed in the *Grouping of Materials and List of Figures and Tables* in the front of the volume. In all, this volume reports 5539 sets of data on 892 materials, which are listed in the *Material Index* at the end of the volume. The *Material Index* lists also the materials contained in the companion volumes (Volumes 2 and 3) on thermal conductivity.

In addition to metals, semimetals and semiconductors are included in this volume. Although a nonmetal, boron is also included because of its extensive use as an alloying element for metallic alloys. Of course, it is also contained in Volume 2, which covers nonmetals.

The temperature ranges covered by the thermal conductivity data for many materials are from near absolute zero to past the melting point, though for most high-temperature alloys the available data are limited to the solid range.

The data for the elements and a small number of alloys have been critically evaluated, analyzed, and synthesized, and recommended reference values are presented. This procedure involves critical evaluation of the validity of available data and related information, resolution and reconciliation of disagreements of conflicting data, correlation of data in terms of various affecting parameters, and comparison of the resulting values with theoretical

predictions or with results derived from semi-theoretical relationships or from generalized empirical correlations. Besides critical evaluation and analysis of the existing data, thermodynamic principles and semi-empirical techniques are employed to fill in gaps and to extrapolate existing data so that the resulting recommended values are internally consistent and cover as wide a range of the controlling parameters as possible. Future editions of this volume will contain recommended values for an increasing number of materials.

## 2. PRESENTATION OF DATA

The thermal conductivity data and information on test specimens for each material are generally presented in three sections arranged in the following order: Original Data Plot, Specification Table, and Data Table. For the elements and a small number of alloys, a Graph and Table of Recommended Values is added as a fourth section. Furthermore, for a number of materials for which there exists only a small number of data, the Original Data Plot may be omitted.

The Original Data Plot is a full-page log-log-scale graphical presentation of the original thermal conductivity data as a function of temperature. When several sets of data are coincident, some of the data sets may be omitted from the plot for the sake of clarity. They are, however, reported in the Data Table and Specification Table.

The Specification Table provides in a concise form the comprehensive information on the test specimens for which the data are reported. The curve numbers in the Specification Table correspond exactly to the numbers which also appear in the Original Data Plot and in the Data Table. The Specification Table gives for each set of data the reference number which corresponds to the number in the list of References to Data Sources, the year

of publication of the original data, the method of measurement, the temperature range, the reported estimate of error of the data, the specimen designation, and the specimen characterization and test conditions. The information of the last category, which is reported to the extent provided in the original source document, includes the following:

- (1) Purity, chemical composition, carrier concentration;
- (2) Type of crystal, crystal axis orientation, type and concentration of crystal defects;
- (3) Microstructure, grain size, inhomogeneity, and additional phases;
- (4) Specimen shape and dimensions, method and procedure of fabrication;
- (5) Thermal history and cold work history, heat treatment, mechanical, irradiative, and other treatments;
- (6) Manufacturer and supplier, stock number, and catalog number;
- (7) Test environment, degree of vacuum or pressure, heat flow direction, strength and orientation of the applied magnetic field;
- (8) Pertinent physical properties such as density, porosity, hardness, electrical resistivity (residual, ratio, and temperature variations), Lorenz function, etc.
- (9) Reference material <sup>transition temperatures and its property values</sup> for a comparative method of measurement;
- (10) Form in which the extracted data are presented in the original source document other than raw data points;
- (11) Additional information obtained directly from the author.

Unfortunately, in the majority of cases the authors do not report in their research papers all the necessary pertinent information to fully characterize and identify the materials for which their data are reported. This is particularly true for the authors of earlier investigations. Consequently, the amount of information on specimen characterization reported in the Specification Tables varies greatly from specimen to specimen.

In the Data Table, tabular presentation is given for all the data described in the Specification Table and shown or not shown in the Original Data Plot. Many tabular data which are not presented in the original source documents have been obtained directly from the authors through private communications. Attempts have often been made to contact the authors for tabular data whenever the original data

are given in the research paper only in a figure too small to warrant accurate data extraction compatible with the reported accuracy of the measurement. The thermal conductivity data are given in watts per centimeter per degree Kelvin, and the temperatures in degrees Kelvin. For data conversion, the reader is referred to the Conversion Factors for Thermal Conductivity Units given later.

The recommended thermal conductivity values for a material are reported in a separate graph and table following the Data Table. The estimated accuracy of the recommended values and special remarks on material characterization and identification are also noted in the table.

### 3. CLASSIFICATION OF MATERIALS

The classification scheme as shown in the table for the elements and alloys contained in this volume is based strictly upon the chemical composition of the material. This scheme is mainly for the convenience of materials grouping and data organization, and is not intended to be used as basic definitions for the various material groups.

### 4. SYMBOLS AND ABBREVIATIONS USED IN THE FIGURES AND TABLES

In the Specification Tables, the code designations used for the experimental methods are as follows:

C	Comparative method
E	Direct electrical heating method
F	Forbes' bar method
L	Longitudinal heat flow method
P	Periodic or transient heat flow method
R	Radial heat flow method
T	Thermoelectrical method

Other symbols and abbreviations used in the figures and/or tables are as follows:

b.c.c.	Body-centered cubic
c.	Cubic
c.p.h.	Close-packed hexagonal
d	Density
d.	Diamond (crystal structure)
Decomp.	Decomposition
f.c.c.	Face-centered cubic
f.c.t.	Face-centered tetragonal
h.	Hexagonal

Classification of Materials

Classification	Limits of composition (weight percent)*					
	X <sub>1</sub>	X <sub>1</sub> + X <sub>2</sub>	X <sub>2</sub>	X <sub>3</sub>		
1. Metallic elements	>99.5	—	< 0.2	< 0.2		
2. Nonferrous alloys (X <sub>1</sub> ≠ Fe)	A. Binary alloys	—	≥ 99.5	≥ 0.2	≤ 0.2	
		—	≥ 99.5	> 0.2	> 0.2	
	B. Multiple alloys	—	< 99.5	≥ 0.2	≤ 0.2	
		—	< 99.5	> 0.2	> 0.2	
		≤ 99.5	—	< 0.2	< 0.2	
3. Ferrous alloys (X <sub>1</sub> = Fe)	A. Carbon steels	Group I	Fe	C ≤ 2.0	≤ 0.2	≤ 0.6
			Fe	C ≤ 2.0	≤ 0.2	> 0.6
		Group II	Fe	C ≤ 2.0	> 0.2	≤ 0.6
			Fe	C ≤ 2.0	> 0.2	> 0.6
	B. Cast irons	Group I	Fe	C > 2.0	≤ 0.2	≤ 0.6
			Fe	C > 2.0	≤ 0.2	> 0.6
		Group II	Fe	C > 2.0	> 0.2	≤ 0.6
			Fe	C > 2.0	> 0.2	> 0.6
	C. Alloy steels†	Group I	Fe	≠ C	≤ 0.2 and C ≤ 2.0	≤ 0.6
			Fe	≠ C	≤ 0.2	> 0.6
		Group II	Fe	≠ C	> 0.2	≤ 0.6
			Fe	≠ C	> 0.2	> 0.6

\*X<sub>1</sub> ≥ X<sub>2</sub> ≥ X<sub>3</sub> ≥ X<sub>4</sub> . . . .

†In case Mn, P, S, or Si represents X<sub>2</sub>, this particular element is dropped from the last column. Alloy cast irons are also included in Group II of this category.

- I.D. Inside diameter
- k Thermal conductivity
- M.P. Melting point
- monocl. Monoclinic
- NTP Normal temperature and pressure
- O.D. Outside diameter
- orthorh. Orthorhombic
- r. Rhombohedral
- s.c. Superconducting
- Subl. Sublimation
- T Temperature
- t. Tetragonal
- Temp. Temperature
- T.P. Transition point
- Vit. Vitreous
- ρ Electrical resistivity
- μ Micro
- > Greater than
- < Less than
- ~ Approximately

- ③ Curv number
- ④ Single data point number

5. CONVENTION FOR BIBLIOGRAPHIC CITATION

For the following types of documents the bibliographic information is cited in the sequences given below.

Journal Article:

- a. Author(s)—The names and initials of all authors are given. The last name is written first, followed by initials.
- b. Title of article—In this volume, the titles of the journal articles listed in the *References to Text* are given, but not of those listed in the *References to Data Sources*.
- c. Journal title—The abbreviated title of the journal as used in *Chemical Abstracts* is given.

CONVERSION FACTORS FOR UNITS OF THERMAL CONDUCTIVITY

MULTIPLY by appropriate factor to OBTAIN →	$Btu_{IT} hr^{-1} ft^{-1} F^{-1}$	$Btu_{IT} in.hr^{-1} in.^2 F^{-1}$	$Btu_{IT} in.hr^{-1} in.^2 F^{-1}$	$Btu_{IT} in.hr^{-1} in.^2 F^{-1}$	$Btu_{IT} in.hr^{-1} in.^2 F^{-1}$	$Btu_{IT} in.hr^{-1} in.^2 F^{-1}$	$cal_{IT} sec^{-1} cm^{-1} C^{-1}$	$cal_{IT} sec^{-1} cm^{-1} C^{-1}$	$cal_{IT} sec^{-1} cm^{-1} C^{-1}$	$J sec^{-1} cm^{-1} K^{-1}$	$W cm^{-1} K^{-1}$	$W m^{-1} K^{-1}$	$mW cm^{-1} K^{-1}$
$Btu_{IT} hr^{-1} ft^{-1} F^{-1}$	1	12	1.00007	1.00006	$4.18678 \times 10^{-2}$	$4.18678 \times 10^{-2}$	$1.49916$	$1.70071 \times 10^{-2}$	$1.70071 \times 10^{-2}$	$1.70071 \times 10^{-2}$	$1.70071$	$17.0071$	$17.0071$
$Btu_{IT} in. hr^{-1} in.^2 F^{-1}$	$8.4433 \times 10^{-2}$	1	$8.4433 \times 10^{-2}$	$1.00007$	$1.44332 \times 10^{-4}$	$1.44332 \times 10^{-4}$	$6.12007$	$1.44228 \times 10^{-2}$	$1.44228 \times 10^{-2}$	$1.44228 \times 10^{-2}$	$1.44228$	$1.44228$	$1.44228$
$Btu_{IT} hr^{-1} ft^{-1} F^{-1}$	$0.998331$	$11.9920$	1	1	$4.11062 \times 10^{-2}$	$4.11062 \times 10^{-2}$	$1.48816$	$1.72958 \times 10^{-2}$	$1.72958 \times 10^{-2}$	$1.72958 \times 10^{-2}$	$1.72958$	$17.2958$	$17.2958$
$Btu_{IT} in. hr^{-1} in.^2 F^{-1}$	$8.02356 \times 10^{-2}$	$0.999133$	$8.02356 \times 10^{-2}$	1	$1.34252 \times 10^{-4}$	$1.34252 \times 10^{-4}$	$5.124013$	$1.34101 \times 10^{-2}$	$1.34101 \times 10^{-2}$	$1.34101 \times 10^{-2}$	$1.34101$	$1.34101$	$1.34101$
$cal_{IT} sec^{-1} cm^{-1} C^{-1}$	$2.41909 \times 10^2$	$2.90201 \times 10^3$	$2.41909 \times 10^2$	$2.90201 \times 10^3$	1	$1.30997$	$1.00243 \times 10^2$	$4.1965 \times 10^2$	$4.1965 \times 10^2$	$4.1965 \times 10^2$	$4.1965 \times 10^2$	$4.1965 \times 10^2$	$4.1965 \times 10^2$
$cal_{IT} sec^{-1} cm^{-1} C^{-1}$	$2.41747 \times 10^2$	$2.90006 \times 10^3$	$2.41747 \times 10^2$	$2.90006 \times 10^3$	$0.999431$	1	$1.68 \times 10^2$	$4.194$	$4.194$	$4.194$	$4.194 \times 10^2$	$4.194 \times 10^2$	$4.194 \times 10^2$
$kcal_{IT} hr^{-1} m^{-1} C^{-1}$	$0.671729$	$8.06888$	$0.671729$	$8.06888$	$2.77702 \times 10^{-3}$	$2.77702 \times 10^{-3}$	1	$1.16222 \times 10^{-2}$	$1.16222 \times 10^{-2}$	$1.16222 \times 10^{-2}$	$1.16222$	$11.6222$	$11.6222$
$J sec^{-1} cm^{-1} K^{-1}$	$67.7749$	$8.06117 \times 10^2$	$67.7749$	$8.06117 \times 10^2$	$0.238846$	$0.238846$	$86.0421$	1	1	1	$1 \times 10^2$	$1 \times 10^2$	$1 \times 10^2$
$W cm^{-1} K^{-1}$	$67.7749$	$8.06117 \times 10^2$	$67.7749$	$8.06117 \times 10^2$	$0.238846$	$0.238846$	$86.0421$	1	1	1	$1 \times 10^2$	$1 \times 10^2$	$1 \times 10^2$
$W m^{-1} K^{-1}$	$0.677749$	$8.06117$	$0.677749$	$8.06117$	$2.38846 \times 10^{-2}$	$2.38846 \times 10^{-2}$	$0.90421$	$1 \times 10^{-2}$	$1 \times 10^{-2}$	$1 \times 10^{-2}$	1	10	10
$mW cm^{-1} K^{-1}$	$67.7749 \times 10^{-2}$	$8.06117$	$67.7749 \times 10^{-2}$	$8.06117$	$2.38846 \times 10^{-2}$	$2.38846 \times 10^{-2}$	$0.90421 \times 10^{-2}$	$1 \times 10^{-2}$	$1 \times 10^{-2}$	$1 \times 10^{-2}$	$0.1$	1	1

- d. Series, volume, and number—If the series is designated by a letter, no comma is used between the letter for series and the ~~numeral~~ numeral for volume, and they are underlined together. In case series is also designated by a numeral, a comma is used between the numeral for series and the numeral for volume, and only the numeral representing volume is underlined. No comma is used between the numerals representing volume and number. The numeral for number is enclosed in parentheses.
- e. Pages—The inclusive page numbers of the article.
- f. Year—The year of publication.

**Report:**

- a. Author(s).
- b. Title of report—In this volume, the titles of the reports listed in the *References to Text* are given, but not of those listed in the *References to Data Sources*.
- c. Name of the responsible organization.
- d. Report, or bulletin, circular, technical note, etc.
- e. Number
- f. Part
- g. Pages
- h. Year
- i. ASTIA's AD number—This is given in square brackets whenever available.

**Book:**

- a. Author(s)
- b. Title
- c. Volume
- d. Edition
- e. Publisher
- f. Place of publication
- g. Pages
- h. Year

**6. CONVERSION FACTORS FOR THERMAL CONDUCTIVITY UNITS**

The conversion factors given in the table on page 42a are based upon the following basic definitions:

1 in.	= 0.0254 (exactly) m*
1 lb	= 0.45359237 kg*
1 cal <sub>th</sub>	= 4.184 (exactly) J*
1 cal <sub>IT</sub>	= 4.1868 (exactly) J*
1 Btu <sub>th</sub> lb <sup>-1</sup> F <sup>-1</sup>	= 1 cal <sub>th</sub> g <sup>-1</sup> C <sup>-1</sup> †
1 Btu <sub>IT</sub> lb <sup>-1</sup> F <sup>-1</sup>	= 1 cal <sub>IT</sub> g <sup>-1</sup> C <sup>-1</sup> †

The subscripts "th" and "IT" designate "thermochemical" and "International Steam Table," respectively.

**7. CRYSTAL STRUCTURES, TRANSITION TEMPERATURES, AND OTHER PERTINENT PHYSICAL CONSTANTS OF THE ELEMENTS**

The table on the following pages contains information on the crystal structures, transition temperatures, and certain other pertinent physical constants of the elements. This information is very useful in data analysis and synthesis. For example, the thermal conductivity of a material generally changes abruptly when the material undergoes any transformation. One must therefore be extremely cautious in attempting to extrapolate the thermal conductivity values across any phase, state, magnetic, or superconducting transition temperature, as given in the table.

No attempt has been made to critically evaluate the temperatures/constants given in the table and they should not be considered recommended values. This table has an independent series of numbered references which immediately follows the table.

\*National Bureau of Standards, "New Values for the Physical Constants Recommended by NAS-NRC," *NBS Tech. News Bull.*, 47 (10), 175-7, 1963.

†Mueller, E. F. and Rossini, F. D., "The Calory and the Joule in Thermodynamics and Thermochemistry," *Am. J. Phys.*, 12 (1), 1-7, 1944.

CRYSTAL STRUCTURES, TRANSITION TEMPERATURES, AND OTHER PERTINENT PHYSICAL CONSTANTS OF THE ELEMENTS

Name	Atomic Number	Atomic Weight <sup>a</sup>	Density <sup>b</sup> kg m <sup>-3</sup> · 10 <sup>-3</sup>	Crystal Structure	Phase Transition Temp., K	Superconducting Transition Temp., K	Curie Temp., K		Debye Temperature at 0 K, K		Melting Point, K	Boiling Point, K	Critical Temp., K
							K	K	K	K			
Actinium	89	(227)	10.07 <sup>10</sup>	f.c.c. <sup>1</sup>			124 <sup>3</sup>	100 <sup>4</sup> (at -50 K)	1323 <sup>1</sup>	3200 ± 300 <sup>4</sup>			
Aluminum	13	26.9815	2.702 <sup>4</sup>	f.c.c. <sup>1</sup>		1.29 <sup>5</sup> 1.17 <sup>6</sup> 1.18 <sup>9</sup>	423 ± 5 <sup>3</sup>	390 <sup>3</sup>	933.2 <sup>1,19</sup>	2723 <sup>29</sup>	8450 <sup>11</sup>	7740 <sup>100</sup>	
Americium	95	(243)	11.7 <sup>3</sup>	Double c.p.h. <sup>1</sup>					1473 <sup>28</sup>	2880 <sup>28</sup>			
Antimony	51	121.75	6.684 <sup>29</sup>	r. (?) ? (?) ? (?)	367.8 <sup>11</sup> (γ-?) 690 <sup>10</sup> (γ-?) high-pressure modification	2.6 <sup>8</sup> (Sb II)	150 <sup>3</sup>	200 <sup>14</sup>	903.7 <sup>13</sup> 943.65 <sup>27</sup>	1907 ± 10 <sup>3</sup>	2889 <sup>18</sup>		
Argon	18	39.948	0.0017824 <sup>28</sup> (at 273.2 K and 1 atm)	f.c.c. <sup>14</sup>				90 <sup>4</sup> (at -45 K)	83.8 <sup>17</sup>	87.29 <sup>15</sup>	151 <sup>18</sup>		
Arsenic	33	74.9216	5.73 <sup>28</sup> (gray, at 287.2 K) 6.7 <sup>21</sup> (black) 2.0 <sup>23</sup> (yellow)	r. (gray) c. (yellow)			236 <sup>3</sup>	275 <sup>19</sup>	1090 <sup>13</sup> (35.8 atm) (35.8 atm) subl. 886	1090 <sup>13</sup>	1090 <sup>13</sup>		
Astatine	85	(210)								573.2 <sup>19</sup>	650 <sup>28</sup>		
Barium	56	137.34	3.5 <sup>28</sup>	b.c.c. (γ) <sup>10</sup> ? (δ)	648 <sup>11,11</sup> (γ-β)			110.5 ± 1.8 <sup>25</sup> 116 <sup>25</sup>	998.2 <sup>4</sup>	1910 <sup>3</sup>	3663 <sup>14</sup>	3920 <sup>100</sup>	
Berkelium	97	(249)											
Beryllium	4	9.0122	1.83 <sup>13</sup>	c.p.h. (α) <sup>2</sup> b.c.c. (β)	1533 <sup>14</sup> (α-β)	-6 <sup>100</sup> -8.4 <sup>100</sup>	1180 <sup>14</sup>	1001 <sup>3</sup>	1550 <sup>28</sup>	3142 ± 100 <sup>3</sup>	6153 <sup>18</sup>		
Bismuth	83	208.980	9.78 <sup>28</sup>	r. <sup>1</sup>		3.9 (BI I, at 25 lbm <sup>2</sup> ) 7.2 (BI III, at 27 lbm <sup>2</sup> )	119 ± 2 <sup>3</sup>	116 ± 5 <sup>3</sup>	544.525 <sup>1,111</sup>	1824 ± 8 <sup>3</sup>	4630 <sup>21</sup>		
Boron	5	10.811	2.50 <sup>10</sup>	Simple r. (α) <sup>1</sup> r. (β)	1473 <sup>1</sup> (α-β)		1315 <sup>10</sup>	1362 <sup>3</sup>	2573 <sup>6</sup>	4050 ± 100 <sup>28</sup>			
Bromine	35	79.909	3.119 <sup>29</sup>	orthorh. <sup>14</sup>					266.0 <sup>17</sup>	331.93 <sup>28</sup>	584 <sup>16</sup>		

<sup>a</sup> Atomic weights are based on <sup>12</sup>C = 12 as adopted by the International Union of Pure and Applied Chemistry in 1961; those in parentheses are the mass numbers of the isotopes of longest known half-life.

<sup>b</sup> Density values are given at 293.2 K unless otherwise noted.

<sup>c</sup> Superscript numbers designate references listed at the end of the table.

Name	Atomic Number	Atomic Weight	Density, <sup>b</sup> kg m <sup>-3</sup> · 10 <sup>-3</sup>	Crystal Structure	Phase Transition Temp., K	Superconducting Transition Temp., K	Curie Temp., K	Neel Temp., K	Debye Temperature at 0 K, K	Debye Temperature at 298 K, K	Melting Point, K	Boiling Point, K	Critical Temp., K																	
														15	16	17	18	19	20	21	22	23	24	25	26	27	28	29	30	31
Cadmium	48	112.40	8.65 <sup>15</sup>	c. c. h. <sup>2</sup> b. c. c. (?)		0.56 <sup>1</sup> 0.52 <sup>1</sup>			252 ± 48 <sup>3</sup>	221 <sup>3</sup>	594.16 <sup>3,10</sup> Subl. 1123 <sup>10</sup> (at 0.35 mm Hg)	1008 <sup>3</sup>	1903 <sup>10</sup> 5560 <sup>100</sup>																	
Calcium	20	40.08	1.55 <sup>29</sup>	f. c. c. (α) b. c. c. (β)	737 (α-β) <sup>47</sup>				234 ± 5 <sup>3</sup>	220 <sup>3</sup>	1765 <sup>2</sup>		3267 <sup>15</sup>																	
Californium	98	(251)																												
Carbon (amorphous)	6	12.01115	1.8~2.1 <sup>29</sup>																											
Carbon (diamond)	6	12.01115	3.51 <sup>29</sup>	d.					2240 ± 5 <sup>31</sup>	1874 <sup>3</sup>	> 3823 <sup>5</sup>	5100 <sup>5</sup>																		
Carbon (graphite)	6	12.01115	2.26 (α) <sup>29</sup>	h. (α) r. (β)					402 ± 11 <sup>3</sup>	1550 <sup>3</sup>	3925-3970 <sup>5</sup>	4473 <sup>5</sup>																		
Cerium	58	140.12	6.90 <sup>29</sup>	f. c. c. (α) <sup>22</sup> Double c. p. h. (β) <sup>9</sup> f. c. c. (γ) <sup>27</sup> 1003 (γ-δ) <sup>22</sup> b. c. c. (δ) <sup>22</sup>	103 ± 5 (α-β) <sup>21</sup> 283 ± 5 (β-γ) <sup>21</sup>			13 <sup>22</sup>	146 <sup>3</sup>	138 <sup>24</sup>	1077 <sup>28</sup>	3972 <sup>3</sup>	10400 <sup>100</sup>																	
Cesium	55	132.905	1.973 <sup>29</sup>	b. c. c.					40 ± 5 <sup>2</sup>	43 <sup>23</sup>	301.9 <sup>19</sup> Subl. 301.9 <sup>19</sup> (at 1.2 μHg)	939 <sup>19</sup>	2060 <sup>100</sup> 1900 <sup>100</sup>																	
Chlorine	17	35.453	0.003214 <sup>19</sup> (at 273.2 K)	t.																										
Chromium	24	51.996	7.16 <sup>42</sup>	c. p. h. (α) <sup>16</sup> b. c. c. (β)	~239 (α-β) <sup>d</sup>			311 <sup>37</sup>	598 ± 32 <sup>3</sup>	424 <sup>3</sup>	2116 <sup>38</sup>	2918 ± 65 <sup>3</sup>																		
Cobalt	27	58.9332	8.862 <sup>42</sup>	c. p. h. (α) <sup>16</sup> f. c. c. (β)	630 (α-β) <sup>16</sup>				452 ± 17 <sup>3</sup>	386 <sup>3</sup>	1765 <sup>19</sup>	3229 <sup>3</sup>																		
Copper	29	63.54	b. . .	f. c. c.					342 ± 2 <sup>3</sup>	310 <sup>3</sup>	1356 <sup>10</sup>	2811 ± 20 <sup>41</sup>	8500 <sup>100</sup> 8280 <sup>100</sup>																	
Curium	96	(247)	7 <sup>d</sup>	Double c. p. h. <sup>1</sup>																										
Dysprosium	65	162.50	8.555 <sup>d</sup>	c. p. h. (α) <sup>16</sup> b. c. c. (β)	Near m. p. (γ-β)				174 <sup>43</sup>	172 ± 35 <sup>3</sup>	1773 <sup>12</sup>	3011 <sup>44</sup>	7640 <sup>100</sup>																	

<sup>d</sup> Close-packed hexagonal crystalline modification of chromium may be formed by electrodeposition below 283 K under special conditions of deposition process. This c. p. h. form is unstable and will irreversibly transform into b. c. c. form on heating.



Name	Atomic Number	Atomic Weight <sup>a</sup>	Density, <sup>b</sup> kg m <sup>-3</sup> · 10 <sup>-3</sup>	Crystal Structure	Phase Transition Temp., K	Superconducting Transition Temp., K	Curie Temp., K	Melting Temp., at 0 K, K	Debye Temperature at 298 K, K	Melting Point, K	Boiling Point, K	Critical Temp., K
Einsteinium	99	(254)										
Erbium	68	187.26	9.06 <sup>c</sup>	c.p.h. <sup>1</sup> (α) b.c.c. <sup>1</sup> (β)	1643 (α-β)		19 <sup>d</sup>	134 ± 10 <sup>e</sup>	163 <sup>d</sup>	1770 <sup>16</sup>	3000 <sup>1</sup>	7250 <sup>100</sup>
Europium	63	151.96	5.245 <sup>21</sup>	b.c.c. <sup>1</sup>			~90 <sup>d</sup>	117 <sup>2</sup>		1099 <sup>1</sup>	1871 <sup>16</sup>	4600 <sup>100</sup>
Fermium	100	(253)										
Fluorine	9	18.9984	0.001695 <sup>21</sup> (at 273.2 K and 1 atm)	c. <sup>1</sup> (β-F <sub>2</sub> )						53.58 <sup>3</sup>	85.24 <sup>13</sup>	144 <sup>15</sup>
Francium	87	(223)								300.2 <sup>18</sup>	879 <sup>100</sup>	
Cadmium	64	157.25	7.87 <sup>c</sup>	c.p.h. <sup>1</sup> (α) b.c.c. <sup>1</sup> (β)	1535 (α-β)		292 <sup>10</sup>	170 <sup>3</sup>	155 ± 3 <sup>9</sup>	1579 <sup>11</sup>	3540 <sup>3</sup>	8670 <sup>100</sup>
Calcium	31	69.72	5.91 <sup>29</sup>	orthorh. <sup>1</sup> (α) t. <sup>1</sup> (β)	275.6 (α-β) (at 8.86 × 10 <sup>6</sup> mm Hg)	1.691 <sup>1</sup> 7.2 <sup>1</sup> (Ca II, high-pressure modification)		317 <sup>3</sup>	240 <sup>14</sup>	368.83 <sup>5</sup> 275.6 <sup>11</sup> (at 8.86 × 10 <sup>6</sup> mm Hg)	2510 <sup>3</sup>	7820 <sup>27</sup>
Germanium	32	72.59	5.36 <sup>19</sup>	d. <sup>1</sup>		5.5 <sup>17</sup> (at ~118 kbar) 8.4 <sup>100</sup>		378 ± 22 <sup>3</sup>	403 <sup>1</sup>	1210.6 <sup>6</sup>	3100 <sup>3</sup>	5642 <sup>15</sup>
Gold	79	196.967	19.3 <sup>c</sup>	f.c.c. <sup>1</sup>				165 ± 1 <sup>3</sup>	178 ± 8 <sup>3</sup>	1336.2 <sup>10</sup> 1336.15 <sup>23</sup>	3240 <sup>3</sup>	9500 <sup>11</sup> 8060 <sup>100</sup>
Hafnium	72	178.49	13.26 <sup>c</sup>	c.p.h. <sup>1</sup> (α) b.c.c. <sup>1</sup> (β)	2023 ± 20 (α-β)	0.16 <sup>9</sup> 0.35 <sup>100</sup>		256 ± 5 <sup>3</sup>	213 <sup>23</sup>	2495 <sup>11</sup>	4575 ± 150 <sup>10</sup>	
Helium	2	4.0026	0.0001785 <sup>11</sup> (at 273.2 K and 1 atm)	c.p.h. <sup>11</sup>						3.45 <sup>29</sup> 1.8 ± 0.2 <sup>17</sup> (at ~15 K) 4.22 <sup>23</sup> (at 30 atm)	4.216 <sup>11</sup> 4.22 <sup>23</sup>	5.3 <sup>19</sup>
Holmium	67	164.930	8.80 <sup>19</sup>	c.p.h. <sup>1</sup> (α) b.c.c. <sup>1</sup> (β)	Near m.p. (α-β)		20 <sup>d</sup>	114 ± 7 <sup>10</sup>	161 <sup>14</sup>	1734 <sup>19</sup>	3228 <sup>11</sup>	
Hydrogen	1	1.00797	0.00008987 <sup>29</sup> (at 273.2 K and 1 atm)	c.p.h. <sup>11</sup>						116 <sup>17</sup> (para), 13.8 ± 0.1 <sup>17</sup> at ~56 K) 105 <sup>23</sup> (ortho), at ~53 K)	20.39 <sup>13</sup> 20.37 <sup>23</sup>	33.3 <sup>15</sup>
Iodine	49	114.82	7.3 <sup>29</sup>	f.c.c. <sup>1</sup>		3.4035 <sup>1</sup>		108.8 ± 0.3 <sup>14</sup>	129 <sup>14</sup>	429.76 <sup>3,119</sup>	2270 ± 6 <sup>3</sup>	4377 <sup>100</sup> 7060 <sup>100</sup>
Iodine	53	126.9044	4.93 <sup>29</sup>	orthorh. <sup>11</sup>						106 <sup>4</sup> 386.8 <sup>29</sup> (at ~53 K) 441.298, 16 <sup>13</sup> (at 0.31 mm Hg)	457.50 <sup>29</sup>	785 <sup>11</sup>
Iridium	77	192.2	22.5 <sup>c</sup>	f.c.c. <sup>1</sup>		0.14 <sup>1,9</sup>		425 ± 5 <sup>3</sup>	222 <sup>9</sup>	2716 <sup>3,19</sup>	4820 ± 30 <sup>3</sup>	

Name	Atomic Number	Atomic Weight	Density, <sup>b</sup> kg m <sup>-3</sup> · 10 <sup>-3</sup>	Crystal Structure	Phase Transition Temp., K	Superconducting Transition Temp., K	Curie Temp., K	Neel Temp., K	Debye Temperature at 0 K,		Melting Point, K	Boiling Point, K	Critical Temp., K
									K	K			
Iron	26	55.847	7.87 <sup>20</sup>	b.c.c.-ferromag. <sup>1</sup> (α) 1183 (β-γ) b.c.c.-paramag. (β) 1673 (γ-δ) f.c.c. (γ) b.c.c. (δ) f.c.c. <sup>18</sup>		1043 <sup>16</sup>			457 ± 12 <sup>3</sup>	373 <sup>3</sup>	1810 <sup>19</sup>	3160 <sup>20</sup>	6750 <sup>19</sup> 9400 <sup>19</sup>
Krypton	36	83.80	0.003708 <sup>18</sup> (at 273.2 K and 1 atm)							60 <sup>4</sup> (at ~30 K)	116.6 <sup>5</sup>	119.93 <sup>13</sup>	209.4 <sup>14</sup>
Lanthanum	57	138.91	6.18 <sup>12</sup>	Double c.p.h. (α) f.c.c. (β) b.c.c. (γ)	585 <sup>11</sup> (α-β) 1141 <sup>12</sup> (β-γ)	4.9 <sup>6</sup> (α) 6.3 <sup>7</sup> (β)			142 ± 0 <sup>10</sup>	135 ± 5 <sup>10</sup>	1180 <sup>5</sup>	3713 ± 70 <sup>3</sup>	10500 <sup>10</sup>
Lanthanum	103	(257)											
Lead	82	207.19	11.34 <sup>20</sup>	f.c.c. <sup>1</sup>		7.150 <sup>6</sup>			102 ± 5 <sup>3</sup>	87 ± 1 <sup>3</sup>	600.576 <sup>3,11</sup>	2022 ± 10 <sup>41</sup>	5400 <sup>10</sup> 4760 <sup>10</sup>
Lithium	3	6.939	0.534 <sup>18</sup>	b.c.c. <sup>7</sup>	Martensitic transformation at low temp. <sup>14</sup>				352 ± 17 <sup>3</sup>	448 <sup>3</sup>	453.7 <sup>19</sup>	1599 <sup>13</sup>	4150 <sup>10</sup> 3720 <sup>10</sup>
Lutetium	71	174.97	9.85 <sup>20</sup>	c.p.h. (α) b.c.c. (β)	Near m.p. (γ-β) <sup>14</sup>				210 <sup>14</sup>	116 <sup>3</sup>	1823 <sup>19</sup>	4140 <sup>3</sup>	
Magnesium	12	24.312	1.74 <sup>20</sup>	c.p.h. <sup>7</sup>					396 ± 54 <sup>3</sup>	330 <sup>3</sup>	923 <sup>14</sup>	1385 <sup>3</sup>	3530 <sup>10</sup>
Manganese	25	54.9380	7.43(α) <sup>20</sup> 7.29(β) <sup>20</sup> 7.18(γ) <sup>20</sup>	<del>b.c.c. (α)</del> 1000 <sup>12</sup> (α-β) c. (β) <sup>13</sup> <del>b.c.c. (γ)</del> 1374 <sup>12</sup> (β-γ) <del>b.c.c. (δ)</del> 110 <sup>12</sup> (γ-δ) b.c.c. (δ)				95 <sup>5</sup>	418 ± 32 <sup>3</sup>	363 <sup>3</sup>	1517 ± 3 <sup>6</sup>	2360 <sup>13</sup>	6050 <sup>10</sup>
Mendelevium	101	(256)											
Mercury	80	200.59	13.546 <sup>18</sup> 14.19 <sup>20</sup> (at 234.25 K)	r. (α) b.c.t.-pressure induced structure (β)	Martensitic transformation at low temp. <sup>16</sup>	4.153 <sup>6</sup> (α) 3.949 <sup>6</sup> (β)			~ 75 <sup>15</sup>	92 ± 8 <sup>3</sup>	234.28 <sup>1,11</sup>	629.73 <sup>3,16</sup>	1733 <sup>10</sup> 1705 <sup>10</sup>
Molybdenum	42	96.94	10.24 <sup>20</sup>	b.c.c. <sup>7</sup>		0.92 <sup>1,1</sup>			459 ± 11 <sup>3</sup>	377 <sup>3</sup>	2863 <sup>13</sup>	5785 ± 175 <sup>3</sup>	17000 <sup>10</sup> 16800 <sup>10</sup>
Neodymium	60	144.24	7.607 <sup>18</sup>	Double c.p.h. (α) b.c.c. (β)	1135 <sup>12</sup> (α-β)				159 <sup>3</sup>	148 ± 8 <sup>3</sup>	1292 <sup>18</sup>	2956 <sup>10</sup>	7900 <sup>10</sup>
Neon	10	20.183	0.0009002 <sup>20</sup> (at 273.2 K and 1 atm)	f.c.c. <sup>18</sup>						60 <sup>6</sup> (at ~30 K)	24.48 <sup>6</sup>	27.23 <sup>6</sup> 27.06 <sup>23</sup>	44.5 <sup>15</sup>

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Name	Atomic Number	Atomic Weight	Density, <sup>b</sup> kg m <sup>-3</sup> · 10 <sup>-3</sup>	Crystal Structure	Phase Transition Temp., K	Superconducting		Néel Temp., K	Debye Temperature at 0 K, K	Melting Point, K	Boiling Point, K	Critical Temp., K
						Transition Temp., K	Temp., K					
Leptantium	93	(337)	20.46 <sup>c</sup>	orthorh. <sup>2</sup> t. <sup>1</sup> (A) <sup>2</sup>	531 (α-β) 813 (β-γ)		121 <sup>3</sup>	163 <sup>3</sup>	913.2 <sup>5</sup>	4150 <sup>3</sup>		
Nickel	28	58.71	8.90 <sup>c</sup>	b.c.c. <sup>7</sup> f.c.c. <sup>7</sup>	631 <sup>d</sup>		427 ± 14 <sup>3</sup>	345 <sup>3</sup>	1726 <sup>3,10</sup> 1726 ± 4 <sup>61</sup>	3055 <sup>63</sup>	6294 <sup>14</sup> 11750 <sup>100</sup>	
Niobium	41	92.906	8.57 <sup>c</sup>	b.c.c. <sup>7</sup>		9.13 <sup>1</sup> 9.09 <sup>1</sup> 9.1 <sup>1</sup>	241 ± 13 <sup>3</sup>	250 <sup>64</sup>	2741 ± 27 <sup>3</sup> 2688 <sup>14</sup>	4813 <sup>64</sup>	19000 <sup>100</sup>	
Nitrogen	7	14.0067	0.0012506 <sup>18</sup>	c. <sup>11</sup> (α) h. <sup>11</sup> (β)	35.62 <sup>11</sup> (α-β)			70 <sup>6</sup> (at -35 K)	63.29 <sup>5</sup>	77.34 <sup>13,23</sup>	126.2 <sup>15</sup>	
Nobelium	102	(254)										
Osmium	76	190.2	22.48 <sup>29</sup>	c.p.h. <sup>2</sup>		0.655 <sup>1</sup> 0.65 <sup>1</sup>	500 <sup>77</sup>	400 <sup>68</sup>	3283 ± 10 <sup>69</sup>	5300 ± 100 <sup>70</sup>		
Oxygen	8	15.9994	0.001429 <sup>28</sup> (at 273.2 K and 1 atm)	b.c.orthorh. <sup>1</sup> r. <sup>1</sup> (β) c. <sup>1</sup> (γ)	23.876 ± 0.01 <sup>11</sup> (α-β) 43.81 ± 0.01 <sup>12</sup> (δ-γ)			250 <sup>6</sup> (at -225 K) 500 <sup>24</sup> (at -250 K)	54.8 <sup>1</sup>	90.19 <sup>13</sup> 90.18 <sup>25</sup>	154.8 <sup>15</sup>	
Palladium	46	106.4	12.02 <sup>28</sup>	f.c.c. <sup>2</sup>			283 ± 16 <sup>3</sup>	275 <sup>14</sup>	1825 <sup>3,10</sup>	3200 <sup>3</sup>		
Phosphorus	15	30.9738	1.82 <sup>28</sup> (β) 2.22 <sup>28</sup> (γ) 2.69 <sup>28</sup> (δ)	b. <sup>7</sup> (α) b.c.c. <sup>1</sup> (β) c. <sup>1</sup> (γ) f.c.orthorh. <sup>17</sup> (δ)	196 (α-β) <sup>71</sup> 298.16 <sup>13</sup> (β-γ) 298.16 <sup>13</sup> (β-δ)			193 (white) 576 (white) 317.3 (white) 325 (red) 800 (red) 1300 (black)	317.3 (white) 317.3 (white) 317.3 (white) 317.3 (white) 317.3 (white) 317.3 (white)	993.8 <sup>15</sup>		
Platinum	78	195.09	21.45 <sup>28</sup>	f.c.c. <sup>2</sup>			234 ± 1 <sup>3</sup>	225 ± 5 <sup>3</sup>	2042 <sup>3,10</sup>	4100 <sup>3</sup>	8280 <sup>15</sup>	
Plutonium	94	(242)	19.737 <sup>28</sup> (at 298.2 K)	Simple monoc. <sup>2</sup> (α) b.c. monoc. <sup>2</sup> (β) f.c.orthorh. <sup>2</sup> (γ) f.c.c. <sup>2</sup> (δ) b.c.l. <sup>2</sup> (δ') b.c.c. <sup>2</sup> (δ)	396.7 <sup>71</sup> (α-β) 475 <sup>71</sup> (β-γ) 591.4 <sup>71</sup> (γ-δ) 729 <sup>71</sup> (δ,δ') 757 ± 3 <sup>71</sup> (δ'-ε)		171 <sup>74</sup>	176 <sup>74</sup>	912.7 <sup>15</sup>	3727 <sup>15</sup>		
Polonium	84	(210)	9.3 <sup>29</sup> (α) 9.5 <sup>29</sup> (β)	Simple c. <sup>1</sup> (α) r. <sup>1</sup> (β)	327 ± 1.5 <sup>70</sup> (α-β)		81 <sup>3</sup>		527.2 <sup>5</sup>	1235 <sup>70</sup>	2281 <sup>15</sup>	
Potassium	19	39.102	0.86 <sup>29</sup>	b.c.c. <sup>1</sup>			89.4 ± 0.5 <sup>100</sup>	100 <sup>3</sup>	336.6 <sup>5</sup>	1027 <sup>15</sup>	2450 <sup>100</sup> 2140 <sup>100</sup>	
Praseodymium	59	140.907	6.769 <sup>29</sup>	Double c.p.h. <sup>6</sup> (α) b.c.c. <sup>6</sup> (β)	1071 <sup>21</sup> (α-β)		85 ± 1 <sup>68</sup>	138 <sup>71</sup>	1192 ± 2 <sup>19</sup>	3616 <sup>60</sup>	8900 <sup>100</sup>	

Name	Atomic Number	Atomic Weight	Density, <sup>b</sup> g/cm <sup>3</sup> × 10 <sup>-4</sup>	Crystal Structure	Phase Transition Temp., K	Superconducting		Néel Temp., K	Debye Temperature at 0 K, K	Melting Point, K	Boiling Point, K	Critical Temp., K
						Transition Temp., K	Curie Temp., K					
Promethium	61	(145)		<sup>1</sup> h. (γ) <sup>110</sup> b.c.c. (α-β)	1145 <sup>110</sup>					1353 ± 10 <sup>81</sup>	2730 <sup>3</sup>	
Protactinium	91	(231)	15.37 <sup>c</sup>	b.c.c. <sup>3</sup>	1.4 <sup>f</sup>			159 <sup>3</sup>	262 <sup>3</sup>	1503 <sup>6</sup>	4680 <sup>3</sup>	
Radium	88	(226)	5 <sup>29</sup>					89 <sup>3</sup>		973.2 <sup>5</sup>	1900 <sup>3</sup>	
Radon	86	(222)	0.00973 <sup>25</sup> (at 273.2 K and 1 atm)	f.c.c. <sup>7</sup>				400 <sup>4</sup> (at ~200 K)		202.2 <sup>3</sup>	211 <sup>15</sup>	377.16 <sup>15</sup>
Rhenium	75	186.2	21.1 <sup>d</sup>	c.p.h. <sup>7</sup>	1.698 <sup>26</sup>			429 ± 22 <sup>3</sup>	275 <sup>12</sup>	3453 <sup>5</sup>	6035 ± 135 <sup>3</sup>	20060 <sup>11</sup>
Rhodium	45	102.905	12.45 <sup>e</sup>	f.c.c. <sup>7</sup>	possible transformation at 1373-1473 K			480 ± 32 <sup>3</sup>	350 <sup>3</sup>	2230 <sup>3,16,22</sup>	3960 ± 60 <sup>3</sup>	
Rubidium	37	85.47	1.53 <sup>28</sup>	b.c.c. <sup>3</sup>				54 ± 4 <sup>3</sup>	59 <sup>23</sup>	312.04 <sup>5</sup>	969 <sup>28</sup>	2100 <sup>15,16,116</sup> 2030 <sup>49</sup>
Ruthenium	44	101.07	12.2 <sup>23</sup>	c.p.h. (α) ? (β) ? (γ) ? (δ)	13, 11; 13, 11; 13, 11; 13, 11	0.49 <sup>5,8</sup>		600 <sup>67</sup>	415 <sup>3</sup>	2523 ± 10 <sup>19</sup>	4325 ± 25 <sup>3</sup>	
Samarium	62	150.35	7.34 <sup>25</sup>	r. (α) b.c.c. (β)	1190 <sup>22</sup> (α-β)		1.4 <sup>9</sup>	116 <sup>6</sup>	184 ± 4 <sup>3</sup>	1345.2 <sup>20</sup>	2140 <sup>3</sup>	5400 <sup>168</sup>
Scandium	21	44.956	3.00 <sup>d</sup>	c.p.h. (α) b.c.c. (β)	1607 <sup>7</sup> (α-β)			470 ± 80 <sup>13</sup>	476 <sup>3</sup>	1812 <sup>5</sup>	3537 ± 30 <sup>3</sup>	
Selenium	34	78.96	4.56 (α) 4.80 (β)	monocl. (α) h. (β) amorphous <sup>7</sup>	304 <sup>24</sup> (α, 111) nitrication (at ~111 atm) 398 <sup>11</sup> (vt. - β) 423 <sup>13</sup> (α-β)	7.1 <sup>28</sup> (at ~111 atm)		151.7 ± 0.4 <sup>89</sup> (at ~45 K) 150 <sup>4</sup> (at ~75 K)		490.2 <sup>5</sup>	1009 <sup>13</sup> (Sc <sub>1</sub> ) 958.0 <sup>22</sup> (Sc <sub>1</sub> , 31) 1027 <sup>13</sup> (Sc <sub>2</sub> )	1757 <sup>15</sup>
Silicon	14	28.086	2.33 <sup>d</sup>	d. <sup>7</sup>		7.5 <sup>67</sup> (at 118-125 kbar)		647 ± 11 <sup>3</sup>	692 <sup>3</sup>	1685 ± 2 <sup>67</sup>	2753 <sup>26</sup>	5159 <sup>15</sup>
Silver	47	107.870	10.5 <sup>29</sup>	f.c.c. <sup>2</sup>				228 ± 3 <sup>3</sup>	221 <sup>3</sup>	1234.0 <sup>3,13</sup>	2468 ± 15 <sup>41</sup>	7460 <sup>11</sup>
Sodium	11	22.9898	0.9712 <sup>29</sup>	b.c.c. <sup>7</sup>	Marignolle transformation at low temp. <sup>26</sup>			157 ± 1 <sup>3</sup>	155 ± 5 <sup>3</sup>	371.0 <sup>13</sup>	1154 <sup>28</sup>	2800 <sup>11</sup> 2400 <sup>169</sup>
Strontium	38	87.62	2.60 <sup>28</sup>	f.c.c. (α) c.p.h. (β) b.c.c. (γ)	486 <sup>22</sup> (α-β) 878 <sup>22</sup> (β-γ)			747 ± 1 <sup>27</sup>	146 <sup>23</sup>	1042 <sup>5</sup>	1645 <sup>3</sup>	3059 <sup>15</sup> 3810 <sup>169</sup>
Sulfur	16	32.064	2.07 <sup>19</sup> (α) 1.96 <sup>29</sup> (β)	r. (α) monocl. (β)	368.6 <sup>13</sup> (α-β)			200 <sup>3</sup> (β)	527 <sup>20</sup> (α) 250 <sup>6</sup> (α) 392.2 <sup>3</sup> (β) at 40 K) Subl. 368.6 <sup>2</sup> (at 0.0047 mm Hg)	386.0 <sup>3</sup> (α) 717.75 <sup>3,10</sup> (β) 392.2 <sup>3</sup> (β)	717.75 <sup>3,10</sup>	1313 <sup>15</sup>
Tantalum	73	180.948	16.6 <sup>d</sup>	b.c.c. <sup>2</sup>		4.483 <sup>6</sup> 4.48 <sup>9</sup>		247 ± 13 <sup>3</sup>	225 <sup>14</sup>	3289 <sup>6</sup>	5760 ± 60 <sup>3</sup>	22000 <sup>11</sup>

Name	Atomic Number	Atomic Weight	Density, <sup>b</sup> kg m <sup>-3</sup> · 10 <sup>-3</sup>	Crystal Structure	Phase Transition Temp., <sup>c</sup> K	Superconducting Transition Temp., <sup>d</sup> K	Curie Temp., <sup>e</sup> K	Neel Temp., <sup>f</sup> K	Debye Temperature at 0 K, <sup>g</sup> K	Temperature at 298 K, <sup>h</sup> K	Melting Point, <sup>i</sup> K	Boiling Point, <sup>j</sup> K	Critical Temp., <sup>k</sup> K
Technetium	43	(99)	11.50 <sup>18</sup>	c.p.h. <sup>2</sup>		8.22 <sup>6</sup> 11.2 <sup>9</sup>		357 <sup>3</sup>	422 <sup>3</sup>	2473 ± 50 <sup>5</sup>	5300 <sup>3</sup>		
Tellurium	52	127.60	6.24 <sup>19</sup> 6.00 <sup>1</sup> (amorph.)	h. <sup>1</sup> (α) 7 (β) <sup>1</sup> amorph. <sup>1</sup>	621 <sup>12</sup> (α-β)	3.3 <sup>1</sup> (Te II, at 56 kbar)		141 ± 12 <sup>3</sup>		722.7 <sup>5</sup>	1363 ± 1 <sup>3</sup>	2329 <sup>15</sup>	
Terbium	65	158.924	8.25 <sup>20</sup>	c.p.h. <sup>2</sup> (α) b.c.c. <sup>2</sup> (β)	Near m.p. <sup>2</sup> (α-β)		219 <sup>28</sup>	230 <sup>30</sup>	150 <sup>31</sup>	168 <sup>44</sup>	1629 <sup>19</sup>	3810 <sup>3</sup>	
Thallium	81	204.37	11.85 <sup>21</sup>	c.p.h. <sup>2</sup> (α) b.c.c. <sup>2</sup> (β)	508.3 <sup>5</sup> (α-β)	2.39 <sup>3</sup> 2.38 <sup>4</sup> 2.37 <sup>9</sup>		88 ± 1 <sup>3</sup>	96 <sup>14</sup>	576.2 <sup>15</sup>	1939 <sup>22</sup>	3219 <sup>15</sup>	
Thorium	90	232.038	11.7 <sup>22</sup>	f.c.c. <sup>2</sup> (α) b.c.c. <sup>2</sup> (β)	1673 ± 25 <sup>23</sup> (α-β)	1.368 <sup>5</sup> 1.37 <sup>9</sup>		170 <sup>34</sup>	100 <sup>14</sup>	2023 <sup>19</sup>	4500 <sup>26</sup>	14550 <sup>26</sup>	
Thulium	69	168.934	9.32 <sup>23</sup>	c.p.h. <sup>2</sup> (α) b.c.c. <sup>2</sup> (β)	Near m.p. <sup>10</sup> (α-β)		22 <sup>26</sup> (ferro- antiferro.)	53 <sup>36</sup>	177 ± 6 <sup>45</sup>	1818 <sup>1</sup>	2266 <sup>27</sup>	6430 <sup>36</sup>	
Ti	50	118.69	5.750 <sup>24</sup> 7.31 <sup>25</sup> (β)	f.c.c. <sup>1</sup> (α) b.c.t. <sup>1</sup> (β) r. <sup>1</sup> (γ)	286.2 ± 3 <sup>26</sup> (α-β)	3.722 <sup>5</sup> (β)		238 ± 24 <sup>3</sup> (gray) 198 ± 9 <sup>3</sup> (white)	254 <sup>3</sup> (gray) 170 <sup>14</sup> (white)	505.06 <sup>3,18</sup> 2786 ± 14 <sup>3</sup>	6000 ± 200 <sup>3</sup> 9300 <sup>100</sup>	8000 <sup>11</sup>	
Titanium	22	47.90	4.5 <sup>26</sup>	c.p.h. <sup>1</sup> (α) b.c.c. <sup>2</sup> (β)	1155 <sup>13</sup> (α-β)	0.39 <sup>3,9</sup>		428 ± 5 <sup>3</sup>	380 <sup>14</sup>	1983 <sup>29</sup>	3586 <sup>100</sup>		
Tungsten	74	183.85	19.3 <sup>27</sup>	b.c.c. <sup>2</sup>		0.011 <sup>172</sup>		369 ± 17 <sup>3</sup>	312 ± 3 <sup>3</sup>	3653 <sup>3,16,13</sup>	6000 ± 200 <sup>3</sup>	23000 <sup>11</sup>	
Uranium	92	238.03	19.07 <sup>28</sup>	orthorh. <sup>1</sup> (α) t. <sup>1</sup> (β) b.c.c. <sup>2</sup> (γ) b.c.c. <sup>2</sup>	37 ± 2 <sup>114</sup> (α-β) 939 <sup>13</sup> (α-β) 1049 <sup>17</sup> (β-γ)	0.68 <sup>5</sup> (α) 1.80 <sup>5</sup> (γ)		200 <sup>34</sup>	300 <sup>3</sup>	1405.6 ± 0.6 <sup>30,1</sup>	3950 ± 250 <sup>100</sup> 12000 <sup>100</sup>	12500 <sup>27</sup>	
Vanadium	23	50.942	6.1 <sup>29</sup>	b.c.c. <sup>2</sup>		5.3 <sup>5</sup> 5.03 <sup>9</sup>		326 ± 54 <sup>3</sup>	390 <sup>14</sup>	2192 ± 2 <sup>61</sup>	3582 ± 42 <sup>3</sup>	11200 <sup>100</sup>	
Xenon	54	131.30	0.005851 <sup>30</sup> (at 273.2 K and 1 atm)	f.c.c. <sup>18</sup>						161.2 <sup>28</sup>	165.1 <sup>19</sup>	289.75 <sup>15</sup>	
Ytterbium	70	173.04	7.02 <sup>31</sup>	f.c.c. <sup>22</sup> (α) b.c.c. <sup>22</sup> (β)	1071 <sup>2,1</sup> (α-β)			116 <sup>100</sup>		1097 <sup>22</sup>	1970 <sup>3</sup>	4420 <sup>100</sup>	
Yttrium	39	88.905	4.47 <sup>32</sup>	c.p.h. <sup>22</sup> (α) b.c.c. <sup>22</sup> (β)	1753 <sup>118</sup> (α-β)			268 ± 32 <sup>3</sup>	214 <sup>104</sup>	1798 <sup>113</sup>	3670 <sup>100</sup>	8850 <sup>100</sup>	
Zinc	30	65.37	7.140 <sup>33</sup>	c.p.h. <sup>2</sup>		0.875 <sup>5</sup> 0.85 <sup>9</sup>		316 ± 20 <sup>3</sup>	237 ± 3 <sup>1</sup>	692.655 <sup>3,118,100</sup>	1175 <sup>100</sup>	2169 <sup>100</sup> 2910 <sup>100</sup>	
Zirconium	40	91.22	6.57 <sup>34</sup>	c.p.h. <sup>1</sup> (α) b.c.c. <sup>1</sup> (β)	1135 <sup>13</sup> (α-β)	0.546 <sup>5</sup> 0.55 <sup>9</sup>		289 ± 24 <sup>3</sup>	250 <sup>14</sup>	2125 <sup>19</sup>	4650 <sup>24</sup>	12300 <sup>100</sup>	

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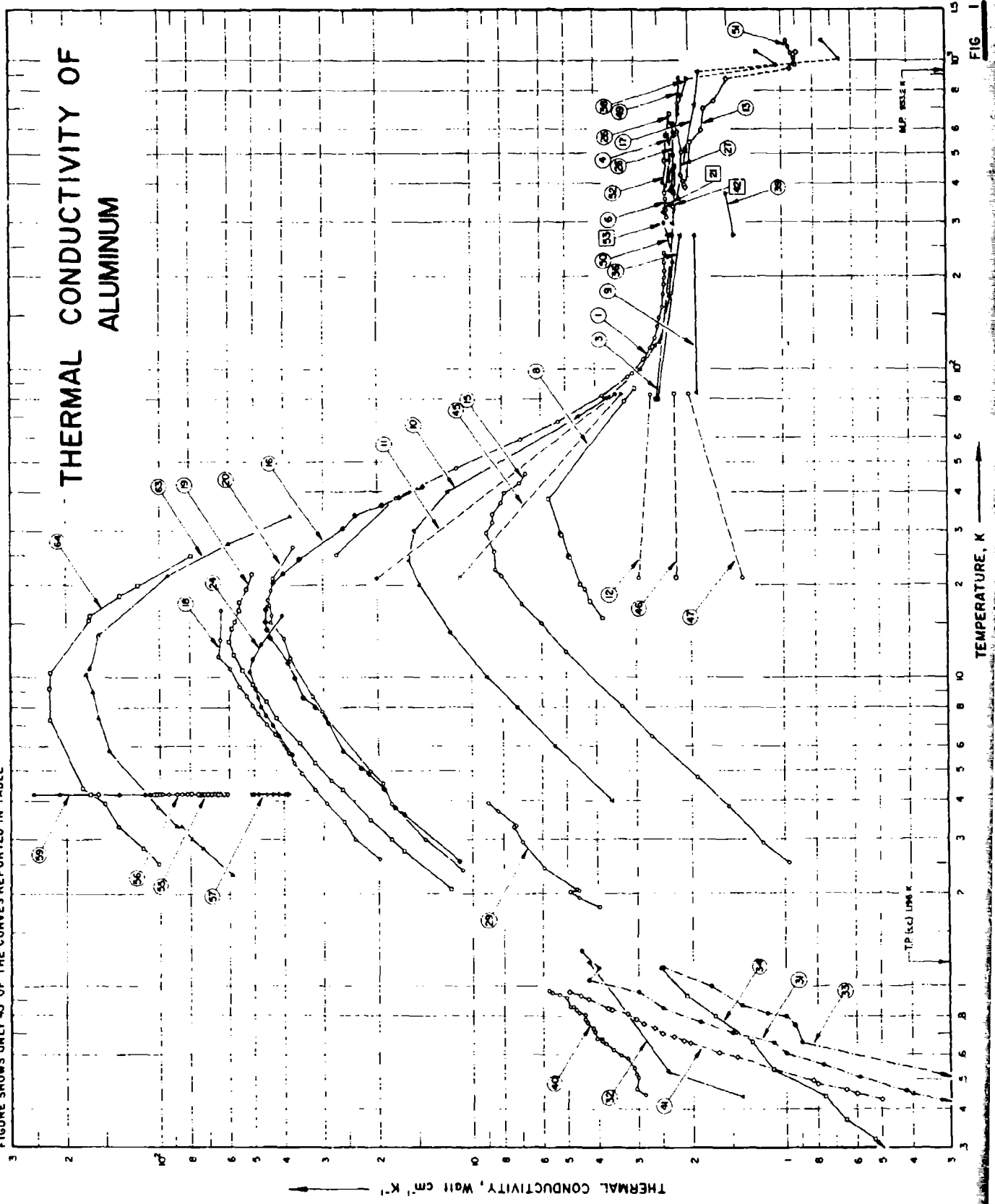
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FIGURE SHOWS ONLY 45 OF THE CURVES REPORTED IN TABLE



# THERMAL CONDUCTIVITY OF ALUMINUM

TEMPERATURE, K

FIG 1

## SPECIFICATION TABLE NO. 1 THERMAL CONDUCTIVITY OF ALUMINUM

(Impurity &lt; 0.20% each; total impurities &lt; 0.50%)

(For Data Reported in Figure and Table No. 1)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
1	114	L	1950	25-238	0.6-1.9		99.99 <sup>+</sup> pure; 0.5 in. dia x 20 in. long; supplied by Aluminum Company of America; cold-drawn with 55% reduction in dia; measured in a vacuum of < 10 <sup>-3</sup> mm Hg.
2	504	P	1961	295-2	±5		Pure; 1.9 x 1.9 x 0.352 cm; thermal conductivity value calculated from measured data of thermal diffusivity and heat capacity and the density value take from Smithsonian Physical Tables (9th ed., 1954).
3	850, 93	L	1929	80-460	3-4		Extremely pure; electrical resistivity reported as 0.725, 2.700, 3.922, and 5.160 uohm cm at -9, 273, 374, and 476 K, respectively.
4	20	L	1951	379-570	1		99.92 Al, 0.04 Si, 0.03 Fe, 0.006 Cu, 0.005 Ti; annealed at 450 C.
5	53	E	1927	353-423	1		High purity.
6	17	L	1955	311-357	1		99.946 Al, 0.0062 Si, 0.0045 Fe, 0.003 Cu, and 0.0001 Mg; 50 mm dia x 70 mm high; manufactured by Metallgesellschaft AG; density 2.691 g cm <sup>-3</sup> at 20 C.
7	491	C	1944	94-147			99.99 <sup>+</sup> pure; supplied by Aluminum Company of America; aluminum used as comparative material.
8	104	L	1951	16-87			High purity; as rolled; measured in a vacuum of < 5 x 10 <sup>-6</sup> mm Hg.
9	619	L	1916	85, 273			Commercial aluminum; 0.5 cm dia x 5 cm long; measured in vacuum.
10	534	L	1957	4, 0-120		JM 340	99.995 pure; single crystal; specimen axis inclined 6°, 40°, and 50° to [100], [011], and [111] direction, respectively; a rod of dia 3.68 mm made by Horizons Inc.; ground down to 3.66 mm in dia, then annealed in vacuum at -400 C for two hours; electrical resistivity reported as 0.025, 0.026, 0.028, 0.065, 0.45, and 2.7 uohm cm at 4, 10, 20, 40, 100, and 300 K, respectively.
11	57	L	1927	21, 83		Al-1	Pure; 7 cm long bar specimen obtained from Aluminum Company of America; annealed in vacuum at 300 C for 2.5 hrs; electrical resistivity reported as 0.0188, 0.3065, and 2.50 uohm cm at -252, -190, and 0 C, respectively.
12	57	L	1927	21, 83		Al-100	Commercial aluminum; annealed in vacuo at 250 C; electrical resistivity reported as 0.1577, 0.458, and 2.65 uohm cm at -252, -190, and 0 C, respectively.
13	45	L	1919	389-1107.3			Pure.
14	127	L	1925	382-645			99.7 pure; supplied by British Aluminum Co., Ltd.; billets 6.75 in. in dia cast from a maximum temperature of 700 C, annealed at 500 C for 2.5 hrs, extruded at 420 C to 0.75 in. dia, then annealed at 450 C for 2.5 hrs; density 2.70 g cm <sup>-3</sup> at 21 C; electrical resistivity reported as 2.83, 2.79, 2.41, 3.45, 3.53, 3.53, 4.47, 4.48, 6.15, 6.23, 6.24, 7.39, 8.77, 8.79, and 9.31 uohm cm at 16.2, 16.8, 15.8, 73.0, 77.0, 79.0, 160.0, 161.0, 302.7, 304.4, 306.2, 399.0, 500.0, 502.4, and 540.7 C, respectively.
15	97	L	1952	2.5-46	2-3	JM 4899	99.994 pure; polycrystalline; 1 ~ 2 mm dia x 5 cm long; supplied by Johnson-Matthey Co.; annealed.

SPECIFICATION TABLE NO. 1 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
16	122		1955	2.6-42	3	JM-4899	99.994 pure; polycrystalline; 0.394 mm dia x 2.97 cm long; supplied by Johnson-Matthey Co.; annealed in vacuum at 600 C for several hrs; electrical resistivity ratio $\rho(293 K)/\rho(20 K) = 279$ .
17	15	F	1947	298-1173	1-5	Al-1	99.95 pure; 2.5 cm dia x 25 cm long.
18	3	L	1951	2.6-17	4	Al-1	99.996 <sup>+</sup> Al, 0.001 Mg, 0.001 Si, 0.0006 Fe, 0.0004 Cu, and 0.0004 Na; single crystal; 0.15 in. dia x 4 in. long; supplied by Aluminum Company of America; residual electrical resistivity $\rho_r = 0.00304 \mu\text{ohm cm}$ ; electrical resistivity ratio $\rho(273 K)/\rho(4.2 K) = 840$ .
19	3	L	1951	2.1-22	4	Al-2	Similar to the above specimen except $\rho_r = 0.00385 \mu\text{ohm cm}$ and $\rho(273 K)/\rho(4.2 K) = 676$ .
20	3	L	1951	2.4-27	4	Al-3	99.995 <sup>+</sup> Al, 0.002 Mg, 0.001 Si, traces of Fe, Cu, and Na; polycrystal, same dimensions and supplier as the above specimen; $\rho_r = 0.0055 \mu\text{ohm cm}$ ; $\rho(273 K)/\rho(4.2 K) = 467$ .
21	276	C	1953	343.2	3	25-Al	Density 2.7 g cm <sup>-3</sup> ; Armco iron used as comparative material.
22	230	L	1925	326			99.97 <sup>+</sup> pure; 1.9 cm dia x 10 cm long; electrical conductivity $33.8 \times 10^4 \text{ ohm}^{-1}\text{cm}^{-1}$ at 23 C.
23	400	L	1955	0.36-0.81			Polycrystalline; in superconducting state.
24	404	L	1950	2.6-16	4	Al-2	99.996 <sup>+</sup> pure; single crystal; supplied by Aluminum Company of America; machined and then etched; crystal slightly damaged by machining.
25	405	E	1940	398-583	$\pm 1$		99.992 Al, 0.0030 Fe, 0.0027 Si, and 0.0024 Cu; cast at 700 C in a mold and cooled to 200 C, rolled to 15 mm dia, drawn to 12.5 mm dia, then reduced to 6.5 mm dia.
26	405	E	1940	382-628	$\pm 1$		99.93 Al, 0.038 Fe, 0.03 Si, and 0.0022 Cu; same fabrication method as above.
27	405	E	1940	399-623	$\pm 1$		99.5 Al, impurities unknown; same fabrication method as above.
29	406	C	1922	313.2	5		99.7 pure; cylindrical specimen of 3 cm long; zinc used as comparative material.
29	498	L	1955	1.8-3.9			99.998 pure; 2.00 mm dia x 9.88 cm long; annealed in vacuum for 5 hrs at 500 C.
30	610	R	1935	373.2	1	Al-1	99.7 pure; electrical conductivity $37.10 \times 10^4 \text{ ohm}^{-1}\text{cm}^{-1}$ at 0 C.
31	409	L	1958	0.13-1.3	10	Al-1	0.01 impurity; with large crystals; annealed in vacuum for 4 hrs at about 600 C; measured in a magnetic field of 0.2 oersted; in superconducting state.
32	409	L	1954	0.44-2.2	10	Al-1	The above specimen in normal state; measured in a longitudinal magnetic field of 115 oersted.
33	409	L	1958	0.16-1.2	10	Al-1	Same specification as the above specimen Al-1; in superconducting state.
34	409	L	1958	0.21-1.2	10	Al-2	The above specimen in normal state.
35	495	I	1949	38-238			99.99 <sup>+</sup> pure; 0.5 in. rod specimen; supplied by Aluminum Company of America; cold-drawn.
36	496	C	1940	80-273	1		Pure; 4.00 mm dia x 60.0 mm long.
37	497	L	1949	25-82			99.98 <sup>+</sup> pure; supplied by Aluminum company of America; cold-drawn.

SPECIFICATION TABLE NO. 1 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
38	480	F	1950	273-1073			99.996 pure; tube specimen of 1.2 in. long with a bore of 0.25 in.; manufactured from Norton's RA 95 material. Electrical conductivity 22.46 and $17.31 \times 10^4 \text{ ohm}^{-1}\text{cm}^{-1}$ at 0 and 100 C, respectively (the author reported 22.46 and $17.31 \times 10^4 \text{ ohm}^{-1}\text{cm}^{-1}$ , probably a typographical error).
39	706	L	1891	273.373			Pure aluminum wire; in normal state.
40	736	L	1965	0.45-0.97			The above specimen measured in superconducting state.
41	736	L	1965	0.43-0.96			0.15 U; 0.500 in. dia x 3 in. long; prepared by dissolution of reactor grade uranium (99.57 pure) in aluminum (99.99 pure) at ~100 C above the alloy liquidus temperature, cast in a graphite mould at 100 C, machined to required dimensions; measured in a vacuum of $< 5 \times 10^{-4}$ mm Hg; copper used as comparative material.
42	591	C	1963	338.2	± 0.4		Similar to above except specimen heat treated at 620 C for 5 days.
43	591	C	1963	338.2		Al-3	99.99 pure; extruded.
44	591	C	1963	338.2	± 1.4		From same cast piece as Al-1 (curve 11); drawn and annealed 2.5% stretched, recrystallized by annealing; grain size 5 to 15 mm long; electrical resistivity reported as 0.0351, 0.319, and 2.52 $\mu\text{ohm cm}$ at -252, -190, and 0 C, respectively.
45	57	L	1927	21.83		Al-191	Same material as Al-100 (curve 12); tempered then 3% stretched, recrystallized by annealing; thermal conductivity measuring length = 2 crystal grams; electrical resistivity reported as 0.219, 0.525, and 2.72 $\mu\text{ohm cm}$ at -252, -190, and 0 C, respectively.
46	57	L	1927	21.83		Al-21	Moderately pure; single crystal; grown by recrystallization; electrical resistivity reported as 0.340, 0.663, and 2.84 $\mu\text{ohm cm}$ at -252, -190, and 0 C, respectively.
47	57	L	1927	21.83		S. P.	S. P. (super pure) aluminum rod from British Aluminum Co.; specimens 2.53 cm in dia and 20.4 cm long; electrical resistivity reported as 2.86 and 7.12 $\mu\text{ohm cm}$ at 40 and 400 C, respectively; Armco iron used as comparative material.
48	659	C	1965	313-673		S. P.	S. P. (super pure) aluminum; 99.993 pure; from British Aluminum Co.; specimen 2.31 cm in dia and 28.0 cm long; electrical resistivity reported as 2.98 and 9.92 $\mu\text{ohm cm}$ at 50 and 600 C, respectively.
49	659	L	1965	323-673		S. P.	S. P. (super pure) aluminum from British Aluminum Co.; specimen 8.0 x 0.44 x 0.44 cm; electrical resistivity reported as 0.74 and 3.02 $\mu\text{ohm cm}$ at -150 and 50 C, respectively.
50	659	L	1965	123-323		S. P.	Electrical resistivity reported as 0.74 and 3.02 $\mu\text{ohm cm}$ ; in molten state; electrical resistivity reported as 26.3 and 30.9 $\mu\text{ohm cm}$ at 700 and 1000 C, respectively; Morgan Crucible Co. grade EY 9 graphite used as comparative material.
51	656	C	1965	973-1273		S. P.	99.99 pure; polycrystal; electrical resistivity reported as 3.15, 4.65, 6.82, and 9.14 $\mu\text{ohm cm}$ at 64.8, 184, 357.8, and 523.4 C, respectively; Lorens function reported as 2.34, 2.40, 2.39, and $2.41 \times 10^3 \text{ V}^2\text{K}^{-2}$ at 64.8, 184, 357.8, and 523.4 C, respectively.
52	617		1967	338-797			

SPECIFICATION TABLE NO. 1 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
53	765	C	1957	298.2			Super-pure; thermal comparator applied on the machined curved surface of the 1 in. dia bar specimen.
54	765	C	1957	298.2			Thermal comparator loaded with 100 gram weight $\alpha$ on the plane lapped surface of the specimen.
55	330	L	1965	4.2	2-3	Sp 1	80 x 5 x 0.066 mm; made from a zone-refined material, cold-rolled and then annealed in air for 24 hrs at 480-500 C; electrical resistivity reported as 0.000538 $\mu\text{ohm cm}$ at 4.2 K; measured in magnetic fields of strength ranging from 0.94 to 12.8 kOe perpendicular to the specimen surface.
56	330	L	1965	4.2	2-3	Sp 1	The above specimen measured in transverse co-planar magnetic fields of strength ranging from 0 to 13.0 kOe.
57	330	L	1965	4.2	2-3	Sp 3	80 x 5 x 0.035 mm; same source and fabrication method as the above specimen; electrical resistivity reported as 0.00103 $\mu\text{ohm cm}$ at 4.2 K; measured in magnetic fields of strength ranging from 0 to 12.7 kOe perpendicular to the specimen surface.
58	330	L	1965	4.2	2-3	Sp 2	The above specimen measured in transverse co-planar magnetic fields of strength ranging from 0 to 12.8 kOe.
59	330	L	1965	4.2	2-3	Sp 3	80 x 5 x 0.129 mm; same source and fabrication method as the above specimen; electrical resistivity reported as 0.000402 $\mu\text{ohm cm}$ at 4.2 K; measured in magnetic fields of strength ranging from 0 to 10.7 kOe perpendicular to the specimen surface.
60	330	L	1965	4.2	2-3	Sp 3	The above specimen measured in transverse co-planar magnetic fields of strength ranging from 0 to 5.98 kOe.
61	330	L	1965	4.2	2-3	Sp 4	80 x 5 x 0.061 mm; same source and fabrication method as the above specimen; bulk electrical resistivity reported as 0.000595 $\mu\text{ohm cm}$ at 4.2 K; measured in magnetic fields of strength ranging from 0 to 13.2 kOe perpendicular to the specimen surface.
62	330	L	1965	4.2	2-3	Sp 4	The above specimen measured in transverse co-planar magnetic fields of strength ranging from 0 to 10.3 kOe.
63	570	L	1963	2-33	1.5	Al <sub>1</sub>	99.9999 pure; specimen made from zone-refined Al, 0.125 in. dia and about 6 cm long; supplied by Consolidated Mining and Smelting Co. of Canada; drawn and etched; residual electrical resistivity $\rho_0$ 0.000903 $\mu\text{ohm cm}$ .
64	570	L	1963	3-25	1.5	Al <sub>1</sub>	Similar to the above specimen except $\rho_0$ 0.000568 $\mu\text{ohm cm}$ .
65	843	P	1966	298.2			Grained ingot supplied by Alcoa Aluminum Co.; mesh size -30 + 45; specimen contained in a 0.75 in. dia x 2 in. long stainless steel cylindrical cell; thermal conductivity measured by using the transient line source method; measured in argon under a pressure of $\sim$ 100 psig.
66	843	P	1966	298.2			Similar to above; measured in nitrogen under a pressure of $\sim$ 100 psig.



DATA TABLE NO. 1 (continued)

T	k	T	k	T	k	T	k	T	k	T	k	T	k	T	k
<u>CURVE 24</u>		<u>CURVE 28<sup>a</sup></u>		<u>CURVE 32</u>		<u>CURVE 35</u>		<u>CURVE 39</u>		<u>CURVE 41 (cont.)</u>		<u>CURVE 42</u>		<u>CURVE 43<sup>b</sup></u>	
2.56	20.0 <sup>c</sup>	313.2	2.14	0.44	1.4	38.4	17.2	273.2	1.437	0.78	3.05	323.2	2.40	338.2	2.23
3.07	21.6 <sup>c</sup>			0.74	2.4	47.8	11.1	373.2	1.514	0.815	3.25	373.2	2.40 <sup>c</sup>		
3.44	26.7 <sup>c</sup>	<u>CURVE 29</u>		1.20	4.3	59.3	7.01			0.835	3.65	473.2	2.375 <sup>a</sup>		
3.49	27.1 <sup>c</sup>	1.81	3.95	<u>CURVE 33</u>		67.9	5.31	<u>CURVE 40</u>		0.845	3.73	573.2	2.335 <sup>a</sup>		
3.90	39.4	1.93	4.39	0.16	0.00083 <sup>d</sup>	82.1	3.82	0.445	2.85	0.91	4.30	673.2	2.275 <sup>c</sup>		
4.00	30.0 <sup>e</sup>	2.03	4.56	0.17	0.00084 <sup>d</sup>	94.3	3.12	0.465	3.05	0.926	4.35	773.2	2.205 <sup>a</sup>		
4.35	31.6 <sup>c</sup>	2.04	4.57	0.17	0.0011 <sup>d</sup>	107.7	2.80	0.505	3.00	0.955	4.35	873.2	2.135		
4.81	34.7 <sup>c</sup>	2.05	4.68	0.17	0.0014 <sup>d</sup>	105.7	2.78	0.505	2.99 <sup>d</sup>	<u>CURVE 44<sup>b</sup></u>		<u>CURVE 45</u>			
4.89	34.7 <sup>c</sup>	2.42	5.96	0.18	0.0013 <sup>d</sup>	117.6	2.66	0.52	3.05	338.2	2.23	123.2	2.485		
5.23	35.7 <sup>c</sup>	2.94	6.95	0.18	0.0018 <sup>d</sup>	126.7	2.57	0.542	3.10	172.2	2.30	172.2	2.30		
5.25	36.7 <sup>c</sup>	3.28	7.82	0.205	0.0024 <sup>d</sup>	138.0	2.51	0.585	3.25	223.2	2.25	223.2	2.25		
5.35	37.2 <sup>c</sup>	3.31	7.35	0.21	0.0044 <sup>d</sup>	147.1	2.47	0.598	3.40	273.2	2.315	273.2	2.315		
5.44	36.8 <sup>c</sup>	3.68	8.35	0.22	0.0048 <sup>d</sup>	159.5	2.42	0.621	3.60	323.2	2.415 <sup>a</sup>	323.2	2.415 <sup>a</sup>		
5.63	37.6	3.92	8.95	0.24	0.0088 <sup>d</sup>	175.0	2.41	0.653	3.80	<u>CURVE 47</u>		<u>CURVE 48<sup>b</sup></u>			
5.86	41.4			0.27	0.021 <sup>d</sup>	187.9	2.40	0.665	3.90	21.2	11.9	1173	0.964		
7.00	41.3	<u>CURVE 30<sup>b</sup></u>		0.32	0.031 <sup>d</sup>	197.6	2.38	0.675	3.90	83.2	3.45	1223	0.95 <sup>b</sup>		
7.55	45.8	273.2	2.27	0.36	0.090 <sup>d</sup>	202.2	2.37	0.71	4.10	<u>CURVE 46</u>		<u>CURVE 49</u>			
8.08	47.5			0.39	0.13 <sup>d</sup>	222.2	2.37	0.73	4.15	338.2	2.26	973.2	0.90		
8.64	48.5	<u>CURVE 31</u>		0.41	0.20	247.6	2.37	0.75	4.15	1023		1023	0.916		
9.21	49.5	0.13	0.0041	0.41	0.20			0.75	4.15	1973		1973	0.932		
10.5	51.5	0.15	0.0040	0.46	0.22	<u>CURVE 36</u>		0.765	4.35	1123		1123	0.948		
11.6	51.1	0.17	0.0014 <sup>d</sup>	0.52	0.32	80.2	2.53	0.81	4.40	1173		1173	0.964		
12.9	47.2	0.21	0.0032	0.55	0.32	273.2	2.13	0.82	4.55	1223		1223	0.95 <sup>b</sup>		
15.9	49.4	0.23	0.0043	0.75	0.94			0.85	4.70	<u>CURVE 51</u>		<u>CURVE 52</u>			
<u>CURVE 25</u>		0.24	0.0032	0.80	1.00	<u>CURVE 37<sup>b</sup></u>		0.855	4.80	338.2	2.25	338.2	2.415 <sup>a</sup>		
394.2	2.29	0.25	0.0105	0.82	1.15	85.0	27.0	0.855	4.80	<u>CURVE 43<sup>b</sup></u>		<u>CURVE 53</u>			
481.2	2.29	0.25	0.012 <sup>d</sup>	0.85	1.40	38.0	17.4	0.915	5.05	21.2	2.13	338.0	2.552 <sup>b</sup>		
581.2	2.25	0.27	0.033 <sup>d</sup>	1.00	1.75	48.5	10.9	0.935	5.75	83.2	3.45	401.6	2.406		
<u>CURVE 26</u>		0.30	0.051	1.15	2.50	59.5	7.11	0.95	5.65	338.2	2.26	457.2	2.377 <sup>a</sup>		
384.2	2.25	0.33	0.090 <sup>d</sup>	<u>CURVE 34</u>		68.0	5.23	0.965	5.75	<u>CURVE 48<sup>b</sup></u>		<u>CURVE 54<sup>b</sup></u>			
481.2	2.25	0.36	0.14	0.205	0.36 <sup>d</sup>	82.0	3.77	<u>CURVE 41</u>		21.2	1.37	631.0	2.188 <sup>a</sup>		
581.2	2.25	0.40	0.26	0.28	0.46 <sup>d</sup>			0.43	0.50	83.2	2.08	730.2	2.155		
<u>CURVE 27</u>		0.42	0.28	0.32	0.53	<u>CURVE 38</u>		0.45	0.66	<u>CURVE 46</u>		<u>CURVE 53</u>			
394.2	2.25	0.45	0.40	0.37	0.65	273.2	2.39	0.465	0.65	338.2	2.38	298.2	2.39		
494.2	2.22	0.46	0.42	0.44	0.76	373.2	2.34	0.485	0.80	473.2	2.39	473.2	2.38		
581.2	2.25	0.51	0.59	0.54	1.10	473.2	2.28	0.495	0.83	573.2	2.34	573.2	2.34		
628.2	2.25	0.56	0.77	0.54	1.70	573.2	2.15	0.61	1.65	673.2	2.29	673.2	2.29		
<u>CURVE 28</u>		0.61	1.00	0.66	1.30	733.2	2.08	0.655	2.05	<u>CURVE 47</u>		<u>CURVE 54<sup>b</sup></u>			
394.2	2.06	0.66	1.10	0.66	1.30	873.2	2.01	0.665	2.15	21.2	1.37	298.2	2.40		
494.2	2.01	0.70	1.50	0.80	1.70	973.2	1.83	0.685	2.30	83.2	2.08	796.6	2.109 <sup>a</sup>		
594.2	2.09	0.96	3.0	1.15	2.50	1073.2	1.21	0.70	2.50	<u>CURVE 48<sup>b</sup></u>		<u>CURVE 53</u>			
691.2	2.16	1.05	4.3	1.15	2.50			0.73	2.65	313.2	2.38	298.2	2.39		
623.2	2.25	1.30	4.5	0.755	2.88			0.755	2.88	373.2	2.39	473.2	2.38		

See Statistical Tables

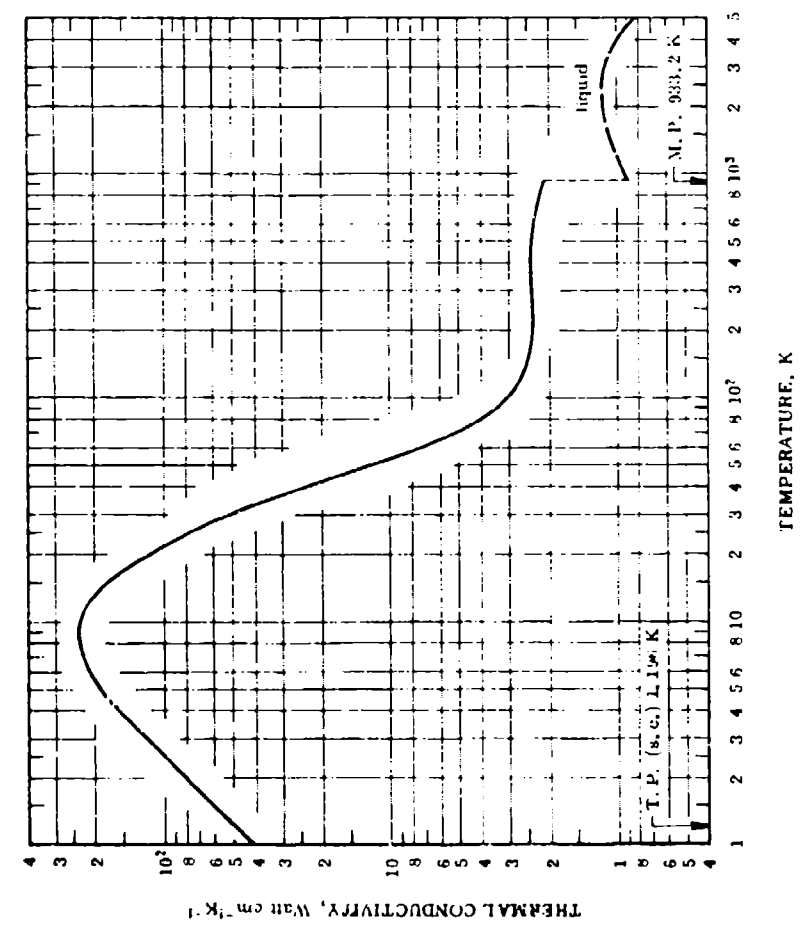
DATA TABLE NO. 1 (continued)

$H \times 10^{-3}$ (Octsted)	k	$H \times 10^{-3}$ (Octsted)	k	$H \times 10^{-3}$ (Octsted)	k	$H \times 10^{-3}$ (Octsted)	k	T	k		
CURVE 55 ( $T = 4.2K$ )											
0	159°	0	104°	1.09	112	2.64	95.0	10.2	172		
0.94	79.1	0.543	61.4°	1.60	108	3.44	87.2	10.8	168		
1.38	70.8°	1.34	49.7°	1.97	104	4.11	90.1	13.5	157		
1.95	67.7°	2.09	41.8°	2.95	111	4.91	89.7	21.6	94		
2.05	66.6°	2.77	39.2	3.47	110	5.70	86.2	27.3	60		
2.64	68.4°	3.49	38.5	3.93	106	6.51	82.2	33.4	38		
3.27	68.2	4.80	38.5	4.54	102	7.36	80.3	CURVE 64			
3.32	66.7	5.63	40.4°	5.03	102	7.95	80.3	2.5	101		
3.98	69.7	6.17	40.6°	5.48	102	8.94	81.3	2.8	114		
4.42	69.7	6.43	41.4	6.04	104	9.53	82.0	3.3	136		
4.61	71.5°	7.43	43.2	6.72	105	10.4	82.4	3.3	136		
5.32	73.5	8.34	45.1	7.12	103	11.4	86.4	3.9	151		
5.62	75.2°	9.03	47.4°	7.51	102	12.3	78.5	4.2	167		
6.12	75.4	9.20	48.0	7.83	101	13.2	76.7	4.4	177		
6.77	72.3	10.2	50.1°	8.22	98.2°	CURVE 62° ( $T = 4.2K$ )					
8.27	64.3°	11.2	50.9	8.74	98.7	0	182	7.3	225		
9.03	62.0°	12.0	59.7°	9.05	100	0.444	118	9.2	226		
9.71	60.9	12.7	49.6	9.55	100	0.728	101	10.4	224		
10.2	61.9°	12.7	49.6	10.3	99.0°	1.14	94.7	15.4	169		
10.8	62.4	CURVE 58° ( $T = 4.2K$ )				1.37	96.0	18.5	134		
12.1	63.8	0	191	10.7	97.6°	1.91	107	19.9	117		
12.2	64.6°	0.143	99.3	CURVE 60° ( $T = 4.2K$ )				24.8	79		
12.8	65.2	0.686	71.1	0	256	CURVE 65°					
CURVE 56 ( $T = 4.2K$ )											
0	153°	0.886	71.1	0.085	224	2.58	119	298.2	0.00255		
0.106	141°	1.36	54.0	0.302	172	3.43	121	298.2	0.00255		
0.273	118°	2.43	51.0	0.450	150	4.25	123	CURVE 66°			
0.879	84.5	2.80	31.8	0.597	138	5.15	124	298.2	0.00364		
1.42	79.8°	3.96	55.9	0.850	108	6.23	127	CURVE 68°			
1.79	81.5	4.71	59.4	1.05	135	8.37	129	298.2	0.00364		
2.50	38.0	5.69	66.4	1.37	134	10.3	125	CURVE 69°			
2.98	33.5	6.57	70.3	1.56	136	CURVE 70°					
3.61	94.0	7.65	71.8	1.91	136	2.3	59	CURVE 71°			
4.34	94.7°	9.20	73.7	2.24	135	2.8	73	CURVE 72°			
5.29	99.6	10.4	74.4	3.86	136	3.3	85	CURVE 73°			
6.91	103	11.5	75.1	5.98	135	3.3	88	CURVE 74°			
8.33	104°	12.8	75.4	CURVE 61° ( $T = 4.2K$ )				3.8	101	CURVE 75°	
9.83	105°	CURVE 59 ( $T = 4.2K$ )				0	185	5.5	145	CURVE 76°	
11.4	104	0	254	0.509	123	7.4	157	7.4	157	CURVE 77°	
12.0	105	0.109	210	0.926	93.6	9.0	164	9.0	164	CURVE 78°	
CURVE 63 (cont.)											
CURVE 64											
CURVE 65°											
CURVE 66°											
CURVE 67°											
CURVE 68°											
CURVE 69°											
CURVE 70°											
CURVE 71°											
CURVE 72°											
CURVE 73°											
CURVE 74°											
CURVE 75°											
CURVE 76°											
CURVE 77°											
CURVE 78°											

\* Not shown on plot



FIGURE AND TABLE NO. 18. RECOMMENDED THERMAL CONDUCTIVITY OF ALUMINUM



T <sub>1</sub>	k <sub>1</sub>	k <sub>2</sub>	T <sub>2</sub>	T <sub>1</sub>	k <sub>1</sub>	k <sub>2</sub>	T <sub>2</sub>
0	0	0	-439.7	500	2.37	137	440.3
1	41.2	2350	-437.9	600	2.32	134	620.3
2	82.0	4740	-436.1	700	2.26	131	800.3
3	121	6990	-434.3	800	2.20	127	980.3
4	157	9070	-432.5	900	2.13	123	1160
5	188	10900	-430.7	933.2	2.11	122	1220
6	213	12300	-428.9	In Liquid State			
7	229	13200	-427.1				
8	238	13800	-425.3	933.2	0.907	52.4	1220
9	240	13900	-423.5	1000	0.950	53.7	1340
10	235	13600	-421.7	1100	0.964	55.7	1520
11	227	13100	-419.9	1200	0.984	57.8	1700
12	215	12400	-418.1	1300	(1.02) <sup>†</sup>	(6.9)	1880
13	202	11700	-416.3	1400	(1.05)	(60.7)	2060
14	189	10900	-414.5	1500	(1.07)	(61.8)	2240
15	176	10200	-412.7	1600	(1.09)	(63.0)	2420
16	163	9420	-410.9	1700	(1.11)	(64.1)	2600
18	138	7970	-407.3	1800	(1.12)	(64.7)	2780
20	117	6760	-403.7	1900	(1.13)	(65.3)	2960
25	77.3	4470	-414.7	2000	(1.14)	(65.9)	3140
30	51.8	2990	-405.7	2200	(1.15)	(66.4)	3500
35	34.9	2020	-396.7	2400	(1.15)	(66.4)	3860
40	23.8	1380	-387.7	2600	(1.15)	(66.4)	4220
45	16.8	971	-378.7	2800	(1.14)	(65.9)	4580
50	12.3	711	-369.7	3000	(1.13)	(65.3)	4940
60	7.54	436	-359.7	3200	(1.11)	(64.1)	5300
70	5.32	307	-348.7	3400	(1.09)	(63.0)	5660
80	4.14	219	-335.7	3600	(1.06)	(61.2)	6020
90	3.44	159	-297.7	3800	(1.03)	(59.5)	6380
100	3.02	114	-279.7	4000	(0.997)	(57.6)	6740
150	2.48	143	-189.7	4500	(0.912)	(52.7)	7640
200	2.37	137	-99.7	5000	(0.818)	(47.3)	8540
250	2.35	136	-9.7	5500	(0.719)	(41.5)	9440
273.2	2.36	136	32.0	6000	(0.614)	(35.5)	10340
300	2.37	137	30.3	7000	(0.392)	(22.6)	12140
350	2.40	139	170.3	8000	(0.156)	(9.01)	13940
400	2.40	139	260.3	8630	(-0)	(-0)	15110

REMARKS

The recommended values are for well-annealed 99.9999% pure aluminum with residual electrical resistivity  $D_0 = 0.000593 \mu\Omega \text{ cm}$  (characterization by  $D_0$  becomes important at temperatures below about 200 K). The values below 1.5 Tm are calculated to fit the experimental data by using  $n = 2.00$ ,  $a = 0.61$ ,  $m = 2.61$ ,  $\alpha = 4.87 \times 10^{-4}$ , and  $\beta = 0.0245$ . The recommended values that are supported by experimental thermal conductivity data are thought to be accurate to within 2 to 3% of the true values near room temperature and 3 to 5% at other temperatures.

<sup>†</sup> T<sub>1</sub> in K, k<sub>1</sub> in Watt cm<sup>-1</sup> K<sup>-1</sup>, T<sub>2</sub> in F, and k<sub>2</sub> in Btu lb<sup>-1</sup> ft<sup>-1</sup> F<sup>-1</sup>.

<sup>‡</sup> Values in parentheses are extrapolated or estimated.

# THERMAL CONDUCTIVITY OF ANTIMONY

FIGURE SHOWS ONLY 16 OF THE CURVES REPORTED IN TABLE

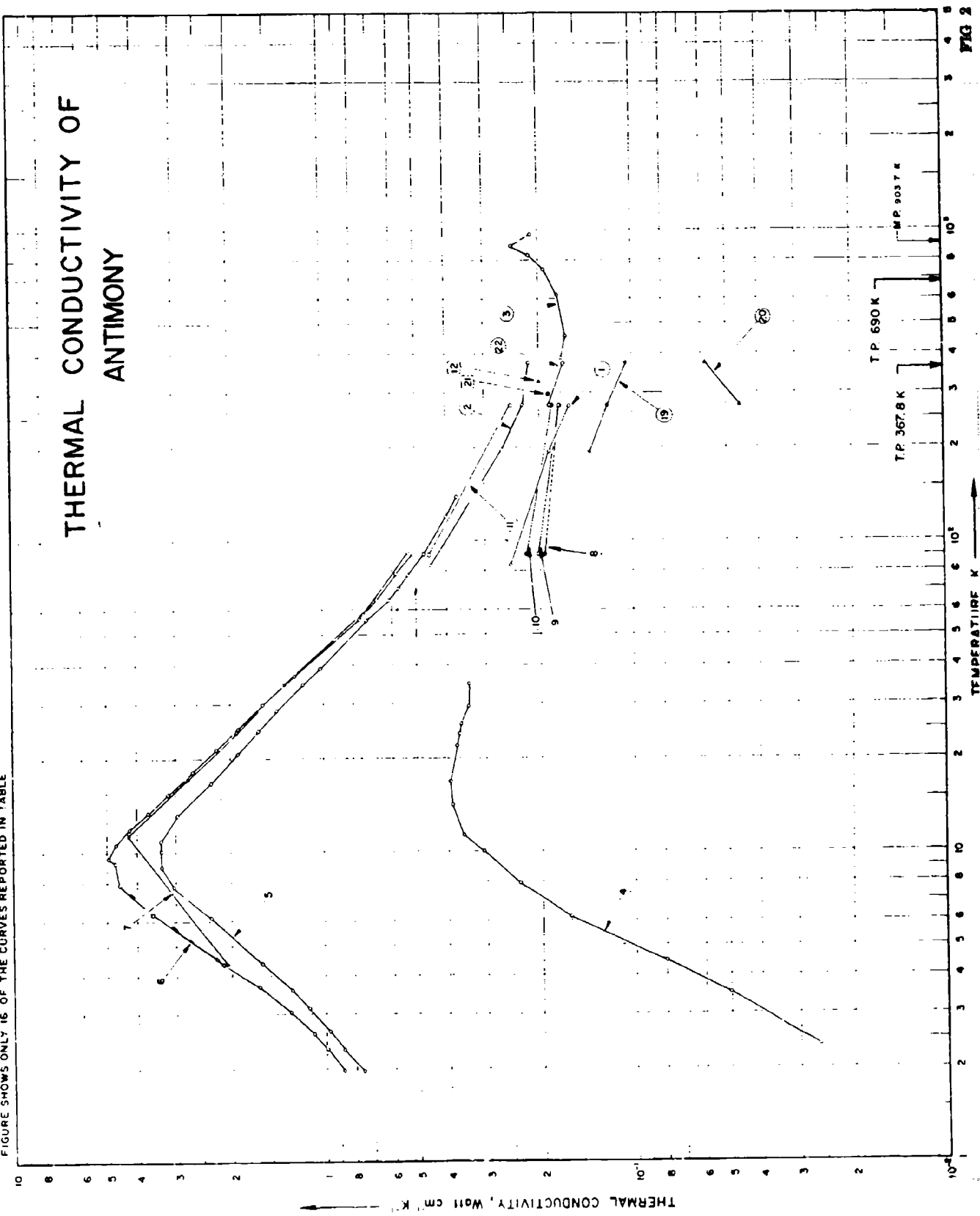


FIG 2

SPECIFICATION TABLE NO. 2 THERMAL CONDUCTIVITY OF ANTIMONY  
(impurity  $\leq$  0.20% each; total impurities  $\leq$  0.50%)

For Data Reported in Figure and Table No. 2

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
1	35	L	1912	83-273			Pure; cold-drawn.
2	49	L	1913	83-373			Pure.
3	85	L	1919	386-965			No details reported.
4	122	L	1955	2.4-35		Sb 1	Polycrystalline; extruded wire; 1.625 cm long, 0.163 cm dia; made from Johnson Matthey Spectrographically Standardized Metals; annealed at 500 C in vacuo for 2 hrs.
5	424	L	1956	2.0-133		Sb 1	High purity polycrystalline specimen; 0.43 x 0.25 x 6 cm; sawn from a lump of extra high purity grade antimony supplied by Bradley Mining Co.; electrical resistivity 47.7 $\mu\text{ohm cm}$ at 295 K; residual electrical resistivity 0.057 $\mu\text{ohm cm}$ .
6	424	L	1958	2.0-91		Sb 2	High purity polycrystalline specimen; 5 mm dia, 6 cm long; crystal width 2 to 5 mm; supplied by Bradley Mining Co.; prepared by zone-refining high purity grade antimony; annealed at 600 C for one wk; electrical resistivity 41.3 $\mu\text{ohm cm}$ at 295 K; residual electrical resistivity 0.054 $\mu\text{ohm cm}$ .
7	424	L	1958	4.4-91		Sb 2a	Second run of the above specimen.
8	425	L	1924	90, 273	5		Polycrystalline specimen with fine grains; cast at 200 C; electrical conductivity at 90 and 273 K being 8.34 and 2.43 x 10 <sup>4</sup> ohm <sup>-1</sup> cm <sup>-1</sup> respectively.
9	425	L	1924	90, 273	5		Polycrystalline specimen with fine grains; electrical conductivity at 90 and 273 K being 8.13 and 2.38 x 10 <sup>4</sup> ohm <sup>-1</sup> cm <sup>-1</sup> respectively.
10	425	L	1924	90, 273	5		Polycrystalline specimen with medium size grains; electrical conductivity at 90 and 273 K being 8.08 and 2.35 x 10 <sup>4</sup> ohm <sup>-1</sup> cm <sup>-1</sup> respectively.
11	425	L	1924	90, 273	5		Polycrystalline specimen with coarse grains; electrical conductivity at 90 and 273 K being 7.89 and 2.32 x 10 <sup>4</sup> ohm <sup>-1</sup> cm <sup>-1</sup> respectively.
12	230	L	1925	327, 2			Total impurity less than .03%; made from Baker's Analyzed Metal.
13	426	L	1947	91, 2		P <sub>1</sub>	Single crystal cylinder specimen; longitudinal axis of the specimen parallel to the z-axis of the crystal; supplied by Kahlbaum; electrical resistivity 26.26 $\mu\text{ohm cm}$ at 0 C; measured at 91.2 K in magnetic fields ranging from 0 to 11.6 kilooersteds.
14	426	L	1947	79, 5		P <sub>1</sub>	The above specimen similarly measured at a temp of 79.5 K.
15	426	L	1947	91, 2		S 14	Similar to the above specimen except longitudinal axis of the specimen perpendicular to z- and x-axis of the crystal; electrical resistivity 37.11 $\mu\text{ohm cm}$ at 0 C; measured at 91.2 K.
16	426	L	1947	79, 5		S 14	The above specimen similarly measured at 79.5 K.
17	426	L	1947	91, 2		S 10	Similar to the above specimen except longitudinal axis perpendicular to z-axis and parallel to x-axis of the crystal; measured at 91.2 K in magnetic fields of 0 to 11.6 kilooersteds.

SPECIFICATION TABLE NO. 2 (continued)

Curve No.	Rel. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
18	426	L	1947	81.2		S 10	<p>The above specimen similarly measured at a temp of 81.2 K.            Specimen pressed from powder at 5000 Kg cm<sup>-2</sup> for 1 hr; anilmony supplied by C. A. F. Kahlbaum.</p> <p>Similar to the above specimen except pressed at 2500 Kg cm<sup>-2</sup> for 1 hr.            Specimen 0.45 cm dia; supplied by Erba; measured under 1.0 atm pressure.            Electrical conductivity at 273 and 373 K being 2.199 and 1.522 x 10<sup>4</sup> ohm<sup>-1</sup> cm<sup>-1</sup> respectively (the paper gives electrical resistivity values as 2.189 and 1.522 x 10<sup>4</sup> ohm<sup>-1</sup> cm<sup>-1</sup>, obviously a typographical error).</p> <p>Molten metal placed in a hole 21 mm in dia drilled in an asbestos cement cylinder 30 mm in height; steel IFC18N9T used as comparative material.</p> <p>Molten specimen contained in a thin-walled stainless steel cylindrical crucible of dimensions 24 mm dia x 100 mm long; electrical resistivity reported as 82.5, 90.2, and 100 μohm cm at 620, 700, and 800 C respectively; thermal conductivity values calculated from measured thermal diffusivity and the specific heat data using the density data taken from Bientas, A. and Sauerwald, F. (Z. Anorg. Chem., 41, 51, 1927).</p>
19	577	L	1913	193-373			
20	577	L	1913	273, 373			
21	511	L	1918	297			
22	706	L	1881	273, 373			
23	838	C	1967	825-1023			
24	319, 320	P	1966	1073.2	8		

DATA TABLE NO. 2 THERMAL CONDUCTIVITY OF ANTIMONY

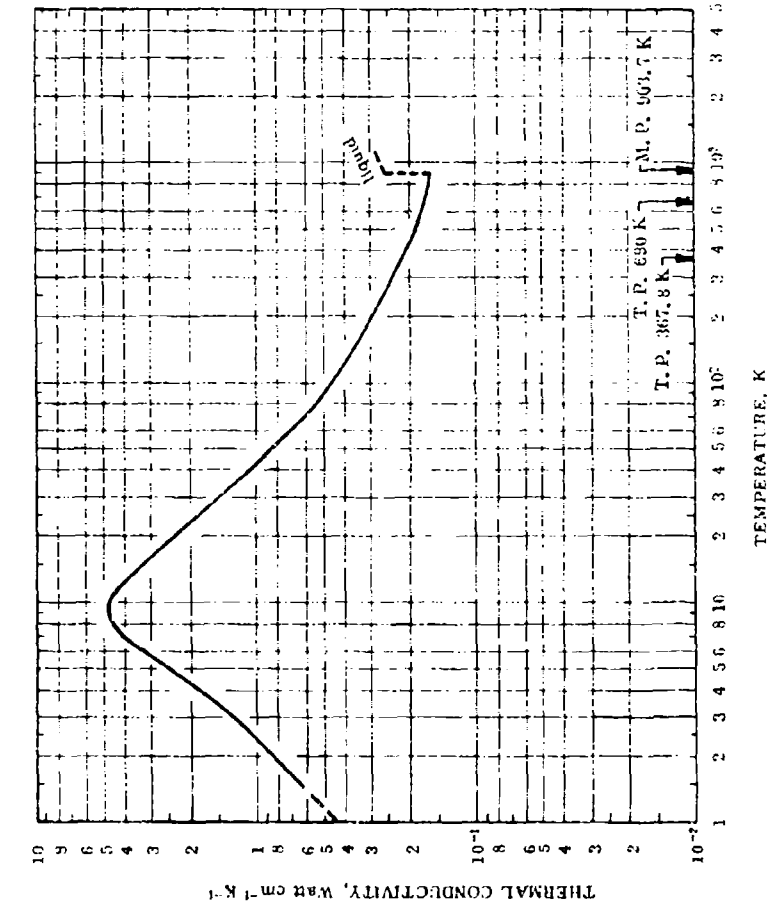
(Impurity &lt; 0.20 each; total impurities &lt; 0.50%)

[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup> K<sup>-1</sup>]

CURVE 1		CURVE 5		CURVE 6 (cont.)		CURVE 12		CURVE 17*		CURVE 23	
T	k	T	k	T	k	T	k	T	k	T	k
83.2	0.248	1.96	0.753*	13.45	3.66*	327.2	0.201	825.2	0.1674	1023.2	0.2721
194.2	0.186	2.30	0.974*	15.3	3.17*	H	k	851.2	0.1674		
273.2	0.159	2.63	0.977*	18.1	2.62*			884.2	0.1716		
		3.14	1.13*	21.4	2.22*			905.2	0.2553		
		3.58	1.24*	24.9	1.88*	CURVE 13*		927.2	0.2637		
		4.36	1.59*	29.7	1.56*	(T = 91.2K)		970.2	0.2679		
		6.13	2.32*	36.7	1.23*	(kilowatts)		1023.2	0.2721		
		7.74	3.04*	55.9	0.76*						
		8.92	3.32*	64.6	0.67*	CURVE 14*		CURVE 18*			
		10.10	3.34*	78.1	0.58	(T = 79.5K)		(T = 81.2K)			
		10.88	3.35*	91.0	0.51	(kilowatts)		(kilowatts)			
		13.17	2.95*					0	0.517		
		16.85	2.30*	CURVE 7				5.65	0.495		
		20.7	1.89*	4.35	2.07*			10.1	0.467		
		24.5	1.61*	11.4	4.26*			11.6	0.461		
		28.5	1.41*	17.2	2.74*						
		34.0	1.16*	24.0	1.93*			CURVE 19			
		38.9	1.02*	34.5	1.34*			193	0.137		
		55.1	0.72*	58.0	0.74*			273	0.121		
		64.3	0.61*	91.2	0.53			373	0.105		
		70.5	0.57	CURVE 8				CURVE 20			
		77.8	0.53	90	0.1925			273	0.0456		
		90.8	0.47	273	0.1716			373	0.0582		
		120.0	0.40	CURVE 15*				CURVE 21			
		138.0	0.37	90	0.2025			297	0.186		
				273	0.1716			CURVE 22			
		CURVE 6		CURVE 9				273	0.185		
		1.97	0.873*	0	0.45*			373	0.166		
		2.31	0.992*	3.43	0.451						
		2.59	1.697*	6.7	0.425						
		3.05	1.30*	10.1	0.419						
		3.68	1.62*	11.6	0.414						
		4.35	2.14*	CURVE 16*							
		4.51	2.22*	(T = 79.5K)							
		6.36	3.59*	0	0.520						
		7.87	4.54*	5.65	0.491						
		9.22	4.69*	8.7	0.472						
		9.56	4.92*	10.1	0.464						
		10.6	4.68*								
		11.9	4.21*								

Not shown on plot

FIGURE AND TABLE NO. 2R RECOMMENDED THERMAL CONDUCTIVITY OF ANTIMONY

RECOMMENDED VALUES<sup>a</sup>  
(For <sup>203</sup>Polycrystalline)

T <sub>1</sub>	k <sub>1</sub>	k <sub>2</sub>	T <sub>2</sub>	T <sub>1</sub>	k <sub>1</sub>	k <sub>2</sub>	T <sub>2</sub>
0	0	0	-459.7	500	0.194	11.2	440.3
1	(0.410) † (24.8)		-457.9	600	0.182	10.5	620.3
2	0.853	49.3	-456.1	700	0.174	10.1	800.3
3	1.28	74.0	-454.3	800	0.168	9.71	980.3
4	1.64	106	-452.5	900	0.167	9.65	1160
5	2.54	149	-450.7	903.7	0.167	9.65	1167
6	3.34	193	-448.9	In Liquid State			
7	4.05	235	-447.1	903.7	0.259	15.0	1167
8	4.63	268	-445.3	1000	0.270	15.6	1340
9	4.89	283	-443.5	1100	(0.280)	(16.2)	1520
10	4.80	277	-441.7				
11	4.50	260	-439.9				
12	4.07	235	-438.1				
13	3.79	219	-436.3				
14	3.51	203	-434.5				
15	3.25	188	-432.7				
16	3.04	176	-430.9				
18	2.67	154	-427.3				
20	2.38	138	-423.7				
25	1.87	108	-414.7				
30	1.54	89.0	-405.7				
33	1.30	75.1	-396.7				
40	1.13	65.3	-387.7				
45	0.994	57.4	-378.7				
50	0.883	51.0	-369.7				
60	0.725	41.9	-351.7				
70	0.620	35.8	-333.7				
80	0.550	31.8	-315.7				
90	0.500	28.9	-297.7				
100	0.464	26.8	-279.7				
150	0.356	20.6	-189.7				
200	0.302	17.4	-99.7				
250	0.267	15.4	-9.7				
273.2	0.255	14.7	32.0				
300	0.243	14.0	80.3				
350	0.226	13.1	170.3				
400	0.212	12.2	260.3				

## REMARKS

The recommended values are for well-annealed high-purity antimony with residual electrical resistivity  $\rho_0 = 0.094 \mu\Omega \text{ cm}$  (characterized by  $\rho_0$  becomes important below room temperature). The values that are supported by experimental thermal conductivity data are thought to be accurate to within 5% of the true values near room temperature and 5 to 10% at other temperatures.

<sup>a</sup> T<sub>1</sub> in K, k<sub>1</sub> in Watt cm<sup>-1</sup> K<sup>-1</sup>, T<sub>2</sub> in F, and k<sub>2</sub> in Btu hr<sup>-1</sup> ft<sup>-1</sup> F<sup>-1</sup>.

† Values in parentheses are extrapolated.

# THERMAL CONDUCTIVITY OF ARSENIC

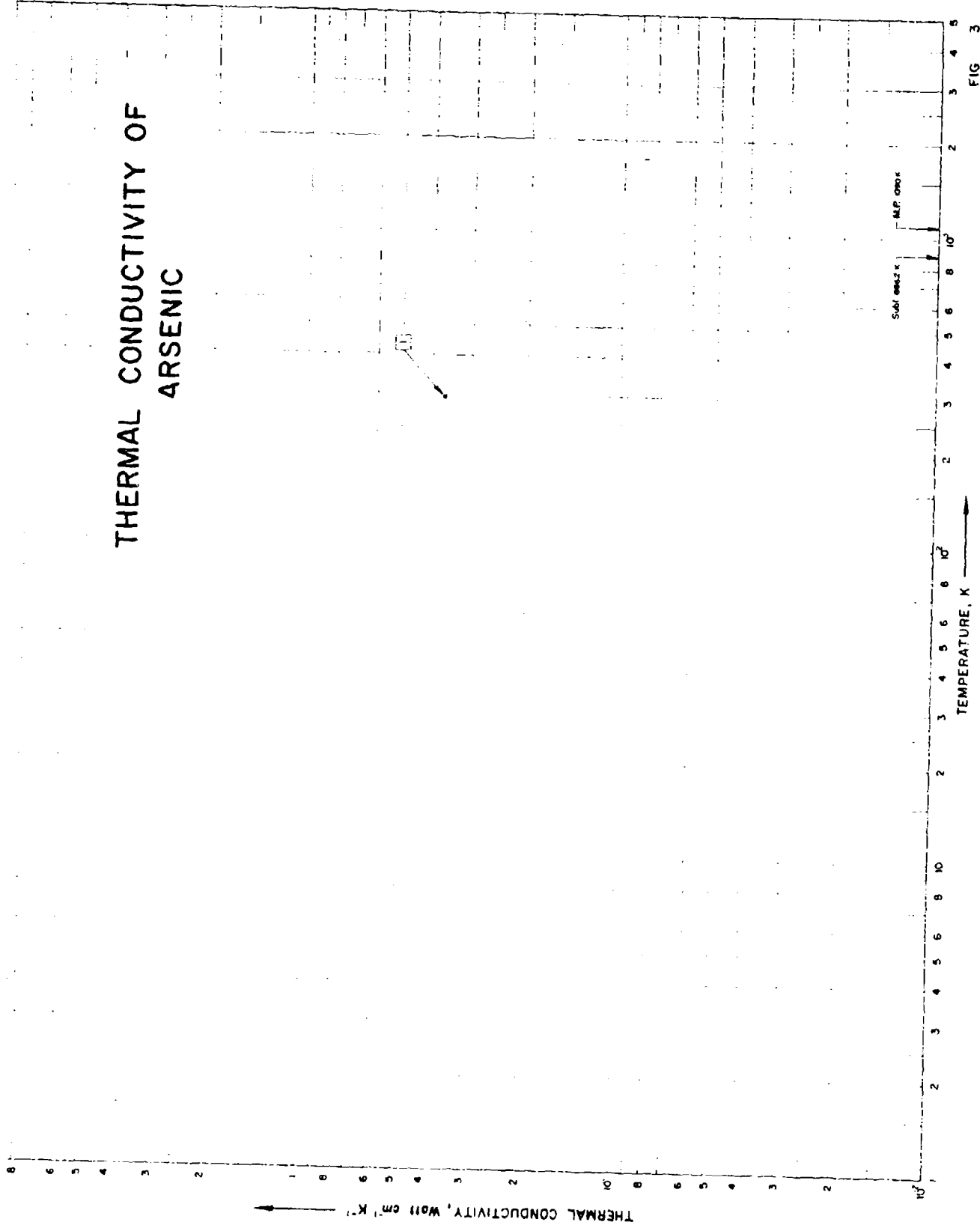


FIG 3 15

## SPECIFICATION TABLE NO. 3 THERMAL CONDUCTIVITY OF ARSENIC

(Impurity &lt; 0.20% each; total impurities &lt; 0.50%)

[ For Data Reported in Figure and Table No. 3 ]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)	Specifications and Remarks
1	813	E	1926	293.2				Polycrystalline; specimen dimensions 2.7 x 1.1 x 0.02 cm obtained by distilling rough crystal in vacuum at about 400 C, the deposit on the containing glass tube polished and smoothed to size; electrical resistivity 46 $\mu\text{ohm cm}$ at 20 C; measured in magnetic fields of 4000 and 8000 gauss which were found to have no effect on the thermal conductivity.



## DATA TABLE NO. 3 THERMAL CONDUCTIVITY OF ARSENIC

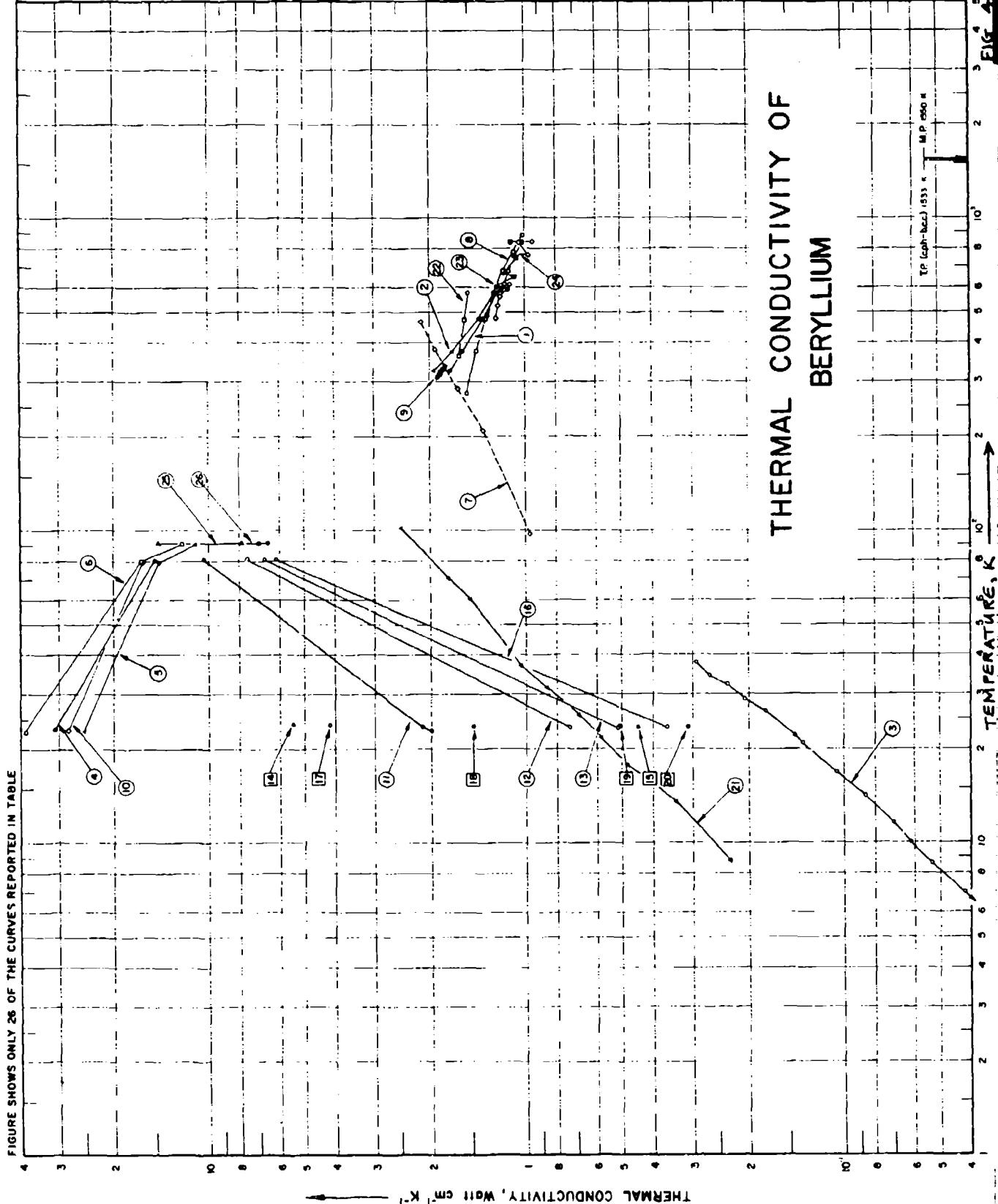
(Impurity &lt; 0.20% each; total impurities &lt; 0.50%)

[Temperature, T, K, Thermal Conductivity, k, Watt cm<sup>-1</sup> K<sup>-1</sup>]

T k

CURVE 1

293.2 0.368



## SPECIFICATION TABLE NO. 4 THERMAL CONDUCTIVITY OF BERYLLIUM

(Impurity &lt; 0.29% each; total impurities &lt; 0.50%)

[ For Data Reported in Figure and Table No. 4 ]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
1	111	C	1953	323-673		Vi-A, R.	Specimen 1 x 1 x 6 cm, spectral analysis showed Mg, Ca, Ba, Si, Fe, Cu, Ti, Al and Mn as impurities; prepared from a block of beryllium by American G. E. C.; sintered; density 1.83 g cm <sup>-3</sup> ; electrical resistivity reported as 5.2, 6.2, 7.7, 10.9, 14.5, 18.2, 22.2, 26.4, and 30.8 μhm cm at 20, 50, 100, 200, 300, 400, 500, 600, and 700 C, respectively.
2	111	C	1953	323-673		Vi-H, T.	The above specimen heat treated at 700 C; electrical resistivity reported as 4.1, 5.0, 6.6, 9.9, 13.5, 17.1, 20.9, 25.2 and 29.9 μhm cm at 20, 50, 100, 200, 300, 400, 500, 600, and 700 C, respectively.
3	122	L	1955	1.8-38	2-3	Be-1	Pure specimen; 2.01 cm long, 0.231 cm dia; made from beryllium powder supplied by Atomic Energy Research Establishment, compressed and sintered at 1100 C in vacuo for several hrs; electrical resistivity ratio $\rho(293K)/\rho(20K) = 352$ .
4	56	L	1940	23-81		Be-3	Single crystal; heat flow perpendicular to hexagonal axis; electrical resistivity reported as 0.0078, 0.0452, and 0.0755 μhm cm at 20, 36, 78, 00, and 90.17 K, respectively.
5	56	L	1940	23-91		Be-4	Single crystal; electrical resistivity reported as 0.0124, 0.0537, and 0.0868 μhm cm at 20, 37, 77, 83, and 90, 29 K, respectively; heat flow perpendicular to hexagonal axis.
6	56	L	1940	23-91		Be-8	Single crystal; electrical resistivity reported as 0.0076, 0.0473, and 0.0770 μhm cm at 20, 34, 77, 95, and 89, 86 K, respectively; heat flow perpendicular to hexagonal axis.
7	278	F	1929	97-464	3-6		Commercially pure specimen; traces of Al, Mg, Cr, Fe, Si, and Mg; ~0.5 total impurities; 21 cm long, 1 cm in dia; supplied by Beryllium Co. of America; electrical resistivity reported as 1.50, 6.45, 14.64, 22.45, 32.45, and 39.00 μhm cm at 84, 294, 496, 674, 880, and 973 K, respectively.
8	753		1959	319.2			Vacuum cast.
9	235	L	1944	307-338		Be 2	Pure; 2.553 cm long, 5.047 cm <sup>2</sup> cross-sectional area.
10	436	L	1938	23, 80		Be 2	Single crystal; hexagonal parallelepiped; supplied by Degussa Co.; length 1.6 cm; hexagonal cross section 0.00648 cm <sup>2</sup> ; electrical resistivity reported as 0.00458, 0.00454, and 3.58 μhm cm at 20, 33, 79, 02, and 273.15 K, respectively; density 1.84 g cm <sup>-3</sup> ; heat flow parallel to the hexagonal axis.
11	436	L	1938	23-81		Be 2	The above specimen measured at H (the transverse magnetic field strength) = 4490 oersts and at θ (angle of rotation of the magnetic field in a plane perpendicular to the specimen axis) = -6°; H perpendicular to one of the binary lateral axes.
12	436	L	1938	23, 81		Be 2	The above specimen measured at H = 8750 oersts and at θ = -6°.
13	436	L	1938	23, 81		Be 2	The above specimen measured at H = 10880 oersts and at θ = -6°.
14	436	L	1938	23, 70		Be 2	The above specimen measured at H = 2280 oersts and at θ = -6°.

SPECIFICATION TABLE NO. 4 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
15	436	L	1938	22, 50		Be 2	The above specimen measured at H = 12200 oersteds and at $\theta = -6^\circ$ .
16	436	L	1938	23, 81		Be 2	The above specimen measured at H = 10880 oersteds and at $\theta = +24^\circ$ at which H is parallel to one of the binary lateral axes.
17	436	L	1938	23, 70		Be 2	The above specimen measured at H = 2280 oersteds and at $\theta = +24^\circ$ .
18	436	L	1938	23, 45		Be 2	The above specimen measured at H = 4490 oersteds and at $\theta = +24^\circ$ .
19	436	L	1938	23, 40		Be 2	The above specimen measured at H = 8750 oersteds and at $\theta = +24^\circ$ .
20	436	L	1938	23, 50		Be 2	The above specimen measured at H = 12200 oersteds and at $\theta = +24^\circ$ .
21	355	L	1955	9-102		Be 2	High purity, < 0.1 Mg and trace of Fe; specimen 4 mm in dia; machined from a sintered rod of high purity beryllium; electrical resistivity at 295 K being 4.95 $\mu\text{ohm cm}$ ; residual resistivity (extrapolated to 0 K) 1.20 $\mu\text{ohm cm}$ .
22	753		1959	363-573			Powder; sintered.
23	753		1959	499-840			Vacuum cast; extruded.
24	753		1959	476-840			Flake, extruded.
25	990	L	1942	92.1		Be 2	Single crystal; hexagonal parallelepiped; supplied by Degussa Co.; length 1.6 cm, hexagonal cross-section 0.0648 cm <sup>2</sup> ; electrical resistivity reported as 0.0450, 0.0763, and 3.58 $\mu\text{ohm cm}$ at 79, 0, 90, 2, and 273, 2 K, respectively; density 1.84 g cm <sup>-3</sup> ; measured in magnetic field of strength 0 to 11.7 kOe at $\theta$ (angle of rotation of magnetic field in a plane perpendicular to the specimen axis) = $-33^\circ$ with the magnetic field perpendicular to one of the binary lateral axes.
26	990	L	1942	92.1		Be 2	The above specimen measured at $\theta = -23^\circ$ and with the magnetic field parallel to one of the binary lateral axes.
27	56	L	1940	23, 81		Be 3	Single crystal; electrical resistivity reported as 0.0835, 0.0789, and 0.1002 $\mu\text{ohm cm}$ at 20.36, 78.00, and 90.17 K, respectively; heat flow perpendicular to the hexagonal axis z; measured in a magnetic field of strength 3.4 kOe perpendicular to z.
28	56	L	1940	23, 81		Be 3	The above specimen measured with the magnetic field of strength 3.4 kOe parallel to z; electrical resistivity reported as 0.1349, 0.0946, and 0.1114 $\mu\text{ohm cm}$ at 20.36, 78.00, and 90.17 K, respectively.
29	56	L	1940	80.6		Be 3	The above specimen measured in a magnetic field of strength 6.8 kOe perpendicular to z; electrical resistivity reported as 0.2037, 0.1465, and 0.1444 $\mu\text{ohm cm}$ at 20.36, 78.00, and 90.17 K, respectively.
30	56	L	1940	23, 81		Be 3	The above specimen measured with the magnetic field of strength 6.8 kOe parallel to z; electrical resistivity reported as 0.3367, 0.1740, and 0.1756 $\mu\text{ohm cm}$ at 20.36, 78.00, and 90.17 K, respectively.

SPECIFICATION TABLE NO. 4 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
31	56	L	1940	23, 81		Be 3	The above specimen measured in a magnetic field of strength 10.1 kOe perpendicular to z; electrical resistivity reported as 0.4829, 0.2422, and 0.2074 $\mu\text{ohm cm}$ at 20.36, 78.00, and 90.17 K, respectively.
32	56	L	1940	23, 81		Be 3	The above specimen measured with the magnetic field of strength 10.1 kOe parallel to z; electrical resistivity reported as 0.6137, 0.2707, and 0.2533 $\mu\text{ohm cm}$ at 20.36, 78.00, and 90.17 K, respectively.
33	56	L	1940	23, 81		Be 3	The above specimen measured with the magnetic field of strength 11.7 kOe perpendicular to z; electrical resistivity reported as 0.6271, 0.2992, and 0.2435 $\mu\text{ohm cm}$ at 20.36, 78.00, and 90.17 K, respectively.
34	56	L	1940	23, 81		Be 3	The above specimen measured with the magnetic field of strength 11.7 kOe parallel to z; electrical resistivity reported as 0.7755, 0.3210, and 0.2968 $\mu\text{ohm cm}$ at 20.36, 78.00, and 90.17 K, respectively.
35	56	L	1940	79.0		Be 4	Single crystal; electrical resistivity reported as 0.0746, 0.0865, and 0.1114 $\mu\text{ohm cm}$ at 20.37, 77.83, and 90.29 K, respectively; heat flow perpendicular to z; measured in a magnetic field of strength 3.4 kOe perpendicular to z.
36	56	L	1940	79.0		Be 4	The above specimen measured with the magnetic field of strength 3.4 kOe parallel to z; electrical resistivity reported as 0.1226, 0.1038, and 0.1240 $\mu\text{ohm cm}$ at 20.37, 77.83, and 90.29 K, respectively.
37	56	L	1940	23, 79		Be 4	The above specimen measured in a magnetic field of strength 6.8 kOe perpendicular to z; electrical resistivity reported as 0.1989 and 0.1508 $\mu\text{ohm cm}$ at 20.37 and 77.83 K, respectively.
38	56	L	1940	23, 79		Be 4	The above specimen measured with the magnetic field of strength 6.8 kOe parallel to z; electrical resistivity reported as 0.2847 and 0.1886 $\mu\text{ohm cm}$ at 20.37 and 77.83 K, respectively.
39	56	L	1940	23-91		Be 4	The above specimen measured in a magnetic field of strength 10.1 kOe perpendicular to z; electrical resistivity reported as 0.3754, 0.2437, and 0.2184 $\mu\text{ohm cm}$ at 20.37, 77.83, and 90.29 K, respectively.
40	56	L	1940	23-91		Be 4	The above specimen measured with the magnetic field of strength 10.1 kOe parallel to z; electrical resistivity reported as 0.4939, 0.2934, and 0.2763 $\mu\text{ohm cm}$ at 20.37, 77.83, and 90.29 K, respectively.
41	56	L	1940	23-91		Be 4	The above specimen measured in a magnetic field of strength 11.7 kOe perpendicular to z; electrical resistivity reported as 0.480, 0.2972, and 0.2548 $\mu\text{ohm cm}$ at 20.37, 77.83, and 90.29 K, respectively.
42	56	L	1940	23-91		Be 4	The above specimen measured with the magnetic field of strength 11.7 kOe parallel to z; electrical resistivity reported as 0.612, 0.345, and 0.3228 $\mu\text{ohm cm}$ at 20.37, 77.83, and 90.29 K, respectively.

SPECIFICATION TABLE NO. 4 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
43	56	L	1940	23-91		Be 8	Single crystal; electrical resistivity reported as 0.0838 and 0.1028 $\mu\text{ohm cm}$ at 78.1 and 89.86 K, respectively; heat flow perpendicular to z; measured in a magnetic field of strength 3.4 kOe perpendicular to z.
44	56	L	1940	23-91		Be 8	The above specimen measured with the magnetic field of strength 3.4 kOe parallel to z; electrical resistivity reported as 0.0962 and 0.1119 $\mu\text{ohm cm}$ at 78.1 and 89.86 K, respectively.
45	56	L	1940	23-91		Be 8	The above specimen measured in a magnetic field of strength 6.8 kOe perpendicular to z; electrical resistivity reported as 0.1566 and 0.1507 $\mu\text{ohm cm}$ at 78.9 and 89.86 K, respectively.
46	56	L	1940	23-91		Be 8	The above specimen measured with the magnetic field of strength 6.8 kOe parallel to z; electrical resistivity reported as 0.1790 and 0.1767 $\mu\text{ohm cm}$ at 78.1 and 89.86 K, respectively.
47	56	L	1940	23-91		Be 8	The above specimen measured in a magnetic field of strength 10.1 kOe perpendicular to z; electrical resistivity reported as 0.2596 and 0.2187 $\mu\text{ohm cm}$ at 78.1 and 89.86 K, respectively.
48	56	L	1940	23-91		Be 8	The above specimen measured with the magnetic field of strength 10.1 kOe parallel to z; electrical resistivity reported as 0.2781 and 0.2572 $\mu\text{ohm cm}$ at 78.1 and 89.86 K, respectively.
49	56	L	1940	23-91		Be 8	The above specimen measured in a magnetic field of strength 11.7 kOe perpendicular to z; electrical resistivity reported as 0.3180 and 0.2582 $\mu\text{ohm cm}$ at 78.1 and 89.86 K, respectively.
50	56	L	1940	23-91		Be 8	The above specimen measured with the magnetic field of strength 11.7 kOe parallel to z; electrical resistivity reported as 0.3312 and 0.3002 $\mu\text{ohm cm}$ at 78.1 and 89.86 K, respectively.

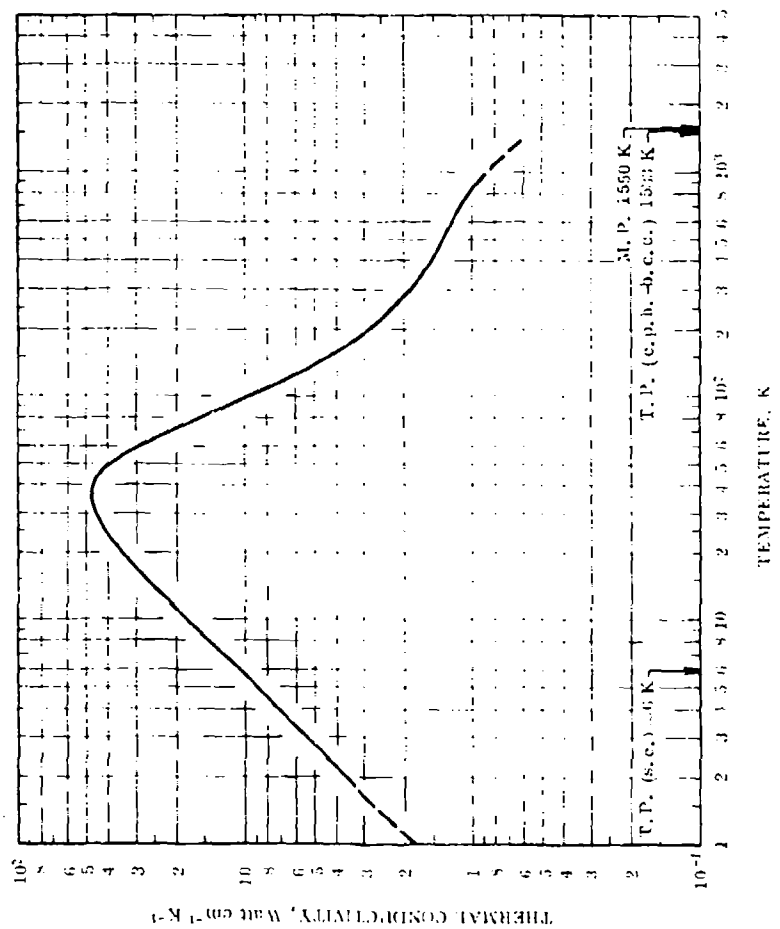
DATA TABLE NO. 4 THERMAL CONDUCTIVITY OF BERYLLIUM

[Impurity < 0.26% each; total impurities < 0.50%]  
 [Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup> K<sup>-1</sup>]

T	k	T	k	T	k	T	k	T	k	T	k	T	k	T	k	T	k	T	k
<u>CURVE 1</u>																			
323.2	1.730	22.7	25.30	22.74	2.00	8.72	0.228	23.2	0.99	22.7	0.90	22.6	0.91	23.2	0.99	22.7	0.90	22.6	0.91
373.2	1.570	79.0	14.68	23.45	2.14	13.5	0.340	80.6	7.79	79.0	6.12	79.7	8.27	80.6	7.79	79.0	6.12	79.7	8.27
473.2	1.360	90.6	11.04	80.7	10.40	17.7	0.480	10.1	7.87	90.6	6.45	90.9	7.92	90.6	6.45	90.9	7.92	90.9	7.92
573.2	1.230			21.7	0.582	21.7	0.582	<u>CURVE 33:</u>				<u>CURVE 42</u>							
673.2	1.130			25.7	0.685	25.7	0.685	<u>CURVE 34:</u>				<u>CURVE 43:</u>							
<u>CURVE 2</u>																			
323.2	1.950	22.6	38.40	23.40	0.736	31.2	0.856	23.2	0.80	23.2	0.80	22.6	0.84	23.2	0.80	22.6	0.84	22.6	0.84
373.2	1.700	79.7	16.42	80.8	7.610	36.7	1.04	80.6	6.56	80.6	6.56	79.7	6.92	80.6	6.56	79.7	6.92	79.7	6.92
473.2	1.400	90.9	12.27	70.0	1.75	70.0	1.75	10.1	6.99	90.9	10.80	90.9	10.80	90.9	10.80	90.9	10.80	90.7	6.91
573.2	1.250			101.5	2.45	101.5	2.45	11.7	6.55			<u>CURVE 44:</u>							
673.2	1.130			23.40	0.517	23.40	0.517	<u>CURVE 35:</u>				<u>CURVE 45:</u>							
<u>CURVE 3</u>																			
1.75	0.008	97.0	0.971	80.9	6.720	363.2	1.607	79.0	11.50	97.0	0.971	22.6	4.36	80.9	6.720	79.7	13.02	79.7	13.02
2.80	0.018	208.2	1.360	473.2	1.556	473.2	1.556	31.2	12.59	208.2	1.360	22.6	4.36	473.2	1.556	90.9	10.59	90.9	10.59
3.40	0.025	372.4	1.912	573.2	1.515	573.2	1.515	31.2	8.01	372.4	1.912	22.6	2.26	573.2	1.515	79.7	11.0	79.7	11.0
4.40	0.032	463.6	2.425	23.70	5.46	494.2	1.226	23.2	4.45	463.6	2.425	80.6	9.44	494.2	1.226	90.9	9.44	90.9	9.44
5.40	0.042	<u>CURVE 8</u>		23.50	9.445	555.2	1.276	23.2	4.45	5.40	0.042	80.6	11.66	555.2	1.276	22.6	2.26	79.7	11.0
6.91	0.053	<u>CURVE 9</u>		23.70	5.46	594.2	1.376	80.6	11.66	6.91	0.053	23.2	4.45	594.2	1.376	22.6	2.26	90.9	9.44
8.56	0.062	<u>CURVE 10</u>		23.40	0.362	631.2	1.474	23.2	4.45	8.56	0.062	80.6	16.31	631.2	1.474	22.7	1.89	22.6	2.26
11.00	0.070	307.1	1.887	80.9	6.15	650.2	1.579	80.6	16.31	11.00	0.070	23.2	4.45	650.2	1.579	79.0	8.32	79.7	8.32
14.13	0.086	313.1	1.874	23.40	0.362	673.2	1.176	23.2	4.45	14.13	0.086	80.6	16.31	673.2	1.176	90.6	7.87	90.9	8.91
16.81	0.106	314.3	1.853	80.9	6.15	746.2	1.067	23.2	4.45	16.81	0.106	23.2	4.45	746.2	1.067	22.6	1.16	79.7	9.71
20.73	0.135	216.1	1.862	23.40	0.362	766.2	1.092	80.6	16.31	20.73	0.135	23.2	4.45	766.2	1.092	22.6	1.16	90.9	8.91
21.96	0.144	319.8	1.858	80.9	6.15	837.2	1.021	23.2	4.45	21.96	0.144	23.2	4.45	837.2	1.021	22.7	1.58	22.6	2.26
26.40	0.178	320.9	1.854	23.70	4.17	840.2	1.113	23.2	4.45	26.40	0.178	23.2	4.45	840.2	1.113	79.0	8.32	79.7	8.32
28.87	0.206	326.9	1.828	23.45	1.47	476.2	1.247	80.6	9.06	28.87	0.206	23.2	4.45	476.2	1.247	90.6	7.87	22.6	1.16
32.06	0.234	327.8	1.828	23.45	1.47	525.2	1.234	23.2	4.45	32.06	0.234	23.2	4.45	525.2	1.234	22.7	1.13	22.6	2.26
34.33	0.266	328.4	1.912	23.40	0.510	563.2	1.205	80.6	8.52	34.33	0.266	23.2	4.45	563.2	1.205	79.0	6.96	22.6	1.16
37.84	0.294	333.3	1.791	23.40	0.510	611.2	1.130	80.6	8.52	37.84	0.294	23.2	4.45	611.2	1.130	90.6	6.95	22.6	1.05
		338.1	1.791	23.40	0.510	613.2	1.172	80.6	8.52			23.2	4.45	613.2	1.172	22.6	1.16	79.7	7.66
<u>CURVE 4</u>																			
23.2	31.00	23.40	28.30	23.40	0.510	733.2	1.142	23.2	4.45	23.2	31.00	23.2	4.45	733.2	1.142	22.7	1.13	22.6	2.26
80.6	14.93	80.3	16.35	23.40	0.313	755.2	1.079	80.6	7.16	80.6	14.93	80.6	7.16	755.2	1.079	79.0	6.96	79.7	7.66
				23.40	0.313	766.2	0.979	23.2	4.45			80.6	7.16	766.2	0.979	90.6	7.87	90.9	7.62
						839.2	1.075	80.6	7.16			23.2	4.45	839.2	1.075	22.7	1.23	22.6	2.26
						840.2	0.946	840.2	0.946			90.6	7.44	840.2	0.946	90.6	7.44	90.6	7.44

\* Not shown on plot

FIGURE AND TABLE NO. 4R RECOMMENDED THERMAL CONDUCTIVITY OF BERYLLIUM



RECOMMENDED VALUES<sup>§</sup>  
(For Polycrystalline)

T <sub>1</sub>	k <sub>1</sub>	k <sub>2</sub>	T <sub>2</sub>	T <sub>1</sub>	k <sub>3</sub>	k <sub>2</sub>	T <sub>2</sub>
0	0	0	-459.7	500	1.39	70.3	440.3
1	(1.81) <sup>‡</sup>	(10.5)	-457.9	600	1.26	72.8	620.3
2	3.02	209	-456.1	700	1.15	66.4	800.3
3	5.42	313	-454.3	800	1.07	51.4	980.3
4	7.23	415	-452.5	900	(0.975)	(56.3)	1160
5	9.04	522	-450.7	1000	(0.887)	(51.3)	1340
6	10.8	624	-448.9	1100	(0.805)	(46.5)	1520
7	12.6	728	-447.1	1200	(0.734)	(42.4)	1700
8	14.4	832	-445.3	1300	(0.674)	(38.9)	1880
9	16.2	936	-443.5	1400	(0.622)	(35.9)	2060
10	18.0	1040	-441.7				
11	19.8	1140	-439.9				
12	21.6	1250	-438.1				
13	23.3	1350	-436.3				
14	25.1	1450	-434.5				
15	26.8	1550	-432.7				
16	28.4	1640	-430.9				
18	31.7	1830	-427.3				
20	34.8	2010	-423.7				
25	41.2	2380	-414.7				
30	45.6	2630	-405.7				
35	47.2	2730	-396.7				
40	46.2	2670	-387.7				
45	44.2	2550	-378.7				
50	40.0	2310	-369.7				
60	29.8	1720	-351.7				
70	21.7	2150	-333.7				
80	16.2	936	-315.7				
90	12.5	722	-297.7				
100	9.90	572	-279.7				
150	4.51	261	-189.7				
200	3.01	174	-99.7				
250	2.36	136	-9.7				
273.2	2.18	126	32.0				
300	2.00	116	80.3				
350	1.78	103	170.3				
400	1.61	93.0	260.3				

REMARKS

The recommended values are for well-annealed high-purity beryllium with residual electrical resistivity  $\rho_0 = 0.0135 \mu\Omega \text{ cm}$  (characterization by  $\rho_0$  becomes important below room temperature). The values below 1.5 Tm are calculated to fit the experimental data by using  $n = 2.90, \delta = 2.56 \times 10^{-3}$ , and  $\beta = 0.553$ . The recommended values that are supported by experimental thermal conductivity data are thought to be accurate to within 5% of the true values near room temperature and 5 to 15% at other temperatures.

<sup>§</sup> T<sub>1</sub> in K, k<sub>1</sub> in Watt cm<sup>-1</sup> K<sup>-1</sup>, T<sub>2</sub> in F, and k<sub>2</sub> in Btu ft<sup>-1</sup> (h<sup>-1</sup> F<sup>-1</sup>). <sup>‡</sup> Values in parentheses are extrapolated.





SPECIFICATION TABLE NO. 5 THERMAL CONDUCTIVITY OF BISMUTH

(Impurity  $\leq 0.20\%$  each; total impurities  $\leq 0.50\%$ )

[For Data Reported in Figure and Table No. 5 ]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition ( weight percent), Specifications and Remarks
1	124	L	1944	2.3-76	1		Highly purified; single crystal; cylindrical specimen of 7.5 cm long and 0.1 cm <sup>2</sup> cross-sectional area; supplied by Hilger Co.; electrical resistivity ratio $\rho(T)/\rho(0C) = 0.320$ , 0.0735, 0.039 and 0.019 at 77.35, 20.4, 14.1 and 4.2 K, respectively.
2	77	E	1900	291-373			Impurities: $\leq 0.03$ <del>total impurities</del> <sup>Fe + Pb</sup> ; cast wire, 8.95 cm long, 1.795 cm dia; density 9.75 g cm <sup>-3</sup> at 18 C; electrical conductivity at 18 and 100 C being 0.840 and 0.628 x 10 <sup>4</sup> ohm <sup>-1</sup> cm <sup>-1</sup> , respectively.
3	126, 324	F	1955	293-373	5		Pure; cast from granular bismuth; electrical conductivity 7360, 6760, 6330, 5920, 5500 ohm <sup>-1</sup> cm <sup>-1</sup> at 293, 313, 333, 353, and 373 K, respectively.
4	113	C	1957	573-623			High purity; molten metal, contained in a cavity 3.5 in. long, 0.94 in. dia; electrical resistivity 128.6, 131.1, 136.0 and 138.5 $\mu$ ohm cm at 300, 350, 400, 450 and 500 C, respectively; stainless steel used as comparative material.
5	49	L	1913	83-373			Pure; electrical conductivity reported as 2.640, 1.190, 0.915, and 0.612 x 10 <sup>4</sup> ohm <sup>-1</sup> cm <sup>-1</sup> at -190, -77, 0, and 100 C, respectively.
6	61	L	1936	19-83	5	PZ 2	99.997 Bi, 0.002 Ag; traces of Pb and Cu; single crystal; length 3 cm, cross-sectional area 0.1 cm <sup>2</sup> ; crystal grown from 'Bi 9506 bismuth' supplied by Adam Hilger Ltd., London; heat flow parallel to trigonal axis; electrical resistivity ratio $\rho(T)/\rho(0C) = 0.9711$ , 0.0634 and 0.0452 at 20.37, 18.47 and 14.15 K, respectively.
7	61	L	1936	19-81	5	PZ 4	99.995 Bi, 0.001 Ag; trace of Pb; single crystal; length 3 cm, cross-sectional area 0.1 cm <sup>2</sup> ; crystal grown from 'Bi 10283 bismuth' supplied by Adam Hilger Ltd., London; heat flow parallel to trigonal axis; $\rho(T)/\rho(0C) = 0.244$ , 0.0540, 0.0474, and 0.0324 at 70.85, 20.37, 18.47 and 14.15 K, respectively.
8	85	L	1919	362-857			Cylindrical specimen.
9	120	L	1914	112-228		No. 1	High purity; single crystal; specimen 1.231 cm in length and cross-section roughly triangular in shape with dimensions $\approx 3$ mm on 2 sides; specimen prepared at California Institute of Technology; heat flow parallel to trigonal axis; measured in vacuum of 10 <sup>-6</sup> mm Hg.
10	120	L	1934	105-208		No. 2	High purity; single crystal; specimen 1.3 cm long and similar in form to No. 1; specimen prepared at California Institute of Technology; heat flow perpendicular to trigonal axis; measured in a vacuum of 10 <sup>-6</sup> mm Hg.
11	1008	L	1934	17-81		P	99.995 Bi; major impurity, Ag; single crystal; specimen consisted of two rods each of size 28 x 5 x 5 mm; grown from H. S. Brandt, Laboratory No. 8016 bismuth, supplied by Adam Hilger Ltd., London; heat flow parallel to trigonal axis (the specimen axis).
12	60	L	1934	17-81		S <sub>1</sub>	99.995 Bi; major impurity, Ag; single crystal; specimen consisted of two rods each of size 28 x 5 x 4 x 4.5 mm; grown from material supplied by Adam Hilger Ltd., London; material melted and pressed into mould to be in contact with a seed crystal, then cooled slowly to crystallize; heat flow parallel to a binary axis.

SPECIFICATION TABLE NO. 5 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
13	60	L	1934	17-81		S <sub>2</sub>	Similar to the above specimen except heat flow parallel to a bisectrix between two binary axes.
14	277	L	1939	298-423		Bi-1	High purity; spectroscopic examination showing traces of Pb and Cu; single crystal; specimen cut in the form of disks 25 mm in dia and 2 mm thick from a large crystal grown by Bridgman's method; supplied by Kahbaum; density 9.74 g cm <sup>-3</sup> at room temp; electrical resistivity reported as 111 μohm cm at 25°C; heat flow perpendicular to trigonal axis. (Data extracted from smooth curve.)
15	277	L	1939	298-423		Bi-2	Similar to the above specimen except electrical resistivity 144 μohm cm at 25°C and heat flow parallel to trigonal axis.
16	424	L	1958	2-80	1	Bi-1	99.97 pure when received; specimen 2 mm dia, 6 cm long; it contained columnar crystals penetrating to the centre of the rod, 16 to 18 crystals being exposed on the circular section; metal supplied by Mining and Chemical Products (London); cast and cooled quickly; residual electrical resistivity $\rho_0 = 104 \mu\text{ohm cm}$ ; $\rho(295\text{K}) = 136 \mu\text{ohm cm}$ .
17	424	L	1958	2-91	1	Bi-2	99.99 pure; 6 columnar crystals per circular section; specimen 3 mm dia, 6 cm long; granular bismuth supplied by the General Chemical Division of Allied Chemical and Dye Corp.; cast in a brass former; cooled slowly; residual electrical resistivity $\rho_0 = 3.9 \mu\text{ohm cm}$ ; $\rho(295\text{K}) = 129 \mu\text{ohm cm}$ .
18	424	L	1958	2-78	1	Bi-3	99.999 pure; crystals about 1 cm long and had the lateral dimensions of the rod; specimen about 3.5 mm dia, 6 cm long; bismuth supplied by Varlacoid Chemical Co. of New York; zone-refined and annealed for several days at a temp just below melting point; residual electrical resistivity $\rho_0 = 2.07 \mu\text{ohm cm}$ ; $\rho(295\text{K}, \text{assumed}) = 118 \mu\text{ohm cm}$ .
19	424	L	1958	2-79	1	Bi-4	99.999 pure; specimen contained about 3 crystals; it had a triangular cross-section of sides 3, 5, and 2.5 mm, 6 cm long, cut from zone-refined bar; supplied by Varlacoid Chemical Co.; residual electrical resistivity $\rho_0 = 1.70 \mu\text{ohm cm}$ ; $\rho(295\text{K}, \text{assumed}) = 118 \mu\text{ohm cm}$ .
20	424	L	1958	2-91	1	Bi-5	Cut from the same bar as the above specimen; contained crystals 2 to 4 mm wide and 1 to 2 cm long; square cross-section 6 x 6 mm; residual electrical resistivity $\rho_0 = 2.4 \mu\text{ohm cm}$ ; $\rho(295\text{K}, \text{assumed}) = 118 \mu\text{ohm cm}$ .
21	469	L	1923	219			0.02 Pb, trace of Fe; single crystal; 1.842 x 1.023 x 0.168 cm; annealed; heat flow parallel to trigonal axis.
22	469	L	1923	219			Similar to above specimen except dimensions 1.843 x 1.022 x 0.167 cm and heat flow perpendicular to trigonal axis.
23	470	L	1949	287.2	1		Pure single crystal; 0.9114 cm cubic specimen; bismuth supplied by Merck; heat flow parallel to trigonal axis.
24	470	L	1949	287.2	1		The above specimen; heat flow perpendicular to the trigonal axis.

SPECIFICATION TABLE NO. 5 (continued)

Curve No.	ReL No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
25	470	L	1949	287.2		2	Similar to the above specimen except heat flow parallel to the trigonal axis.
26	470	L	1949	287.2		2	The above specimen; heat flow perpendicular to the trigonal axis.
27	470	L	1949	287.2		3	Pure, polycrystalline; cubic specimen $0.93 \times 0.93 \times 0.93$ cm; bismuth supplied by Merck.
28	470	L	1949	287.2		4	Similar to the above specimen.
29	230	L	1925	328.2			Total impurities $< 0.03$ ; specimen 10 cm long, 1.9 cm in dia; bismuth from Baker's Analyzed Metal; electrical conductivity $0.94 \times 10^4 \text{ ohm}^{-1}\text{cm}^{-1}$ at 22 C.
30	248	E	1956	313-453	3		99.997 pure.
31	471	L	1903	87-291			Pure; density 9.67, 10.04 and $10.44 \text{ g cm}^{-3}$ at 18, -79, and $-186 \text{ C}$ , respectively; electrical conductivity $0.861$ , $1.196$ and $2.452 \times 10^4 \text{ ohm}^{-1}\text{cm}^{-1}$ at 18, -79, and $-186 \text{ C}$ , respectively. (Note: the paper gives electrical conductivity as $10^5 \text{ ohm}^{-1}\text{cm}^{-1}$ which is probably an error)
32	472	L	1937	91.5		Bi 66	Pure; single crystal; the angle between rod axis and trigonal axis $\phi = 2^\circ$ .
33	472	L	1937	91.5		Bi 66	The above specimen measured in a magnetic field approximately parallel to the z-axis and the xz-plane (z-axis coincident with the trigonal axis; x-axis parallel to a diagonal, which does not intersect with the trigonal axis of one face of the crystal) i. e., parallel to a two-fold secondary axis, with strength H-650 oersteds.
34	472	L	1937	91.6		Bi 66	The above specimen measured at H-650 oersteds approximately parallel to x-axis.
35	472	L	1937	91.4-91.5		Bi 66	The above specimen measured at H-1500 oersteds (H in xz-plane) and at $\psi$ (angle between H and z-axis) ranging from $0$ to $-10^\circ$ .
36	472	L	1937	91.5		Bi 66	The above specimen measured at H-2520 oersteds (H parallel to xz-plane) and at $\psi = 0^\circ$ .
37	472	L	1937	91.7		Bi 66	The above specimen measured at H-2520 oersteds (H parallel to xz-plane) and at $\psi = 90^\circ$ .
38	472	L	1937	91.6		Bi 66	The above specimen measured at H-4850 oersteds (H parallel to xz-plane) and at $\psi = 0^\circ$ .
39	472	L	1937	91.7		Bi 66	The above specimen measured at H-4850 oersteds (H parallel to xz-plane) and at $\psi = 90^\circ$ .
40	472	L	1937	91.5-91.7		Bi 66	The above specimen measured at H-6100 oersteds (H parallel to xz-plane) and at $\psi$ ranging from $0$ to $-20^\circ$ .
41	472	L	1937	91.5		Bi 66	The above specimen measured without magnetic field.
42	472	L	1937	91.6		Bi 66	The above specimen measured in a magnetic field parallel to the yz-plane with H=650 oersteds and $\psi = 10^\circ$ (H approximately parallel to the trigonal axis).
43	472	L	1937	91.8		Bi 66	The above specimen measured at H-650 oersteds (H parallel to yz-plane) and at $\psi = 100^\circ$ (H approximately parallel to y-axis).

SPECIFICATION TABLE NO. 5 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
44	472	L	1937	91.7-92.0		Bi 66	The above specimen measured at H 1500 oersteds (H parallel to yz-plane) and at $\psi$ ranging from $10^\circ$ to $-90^\circ$ .
45	472	L	1937	91.37-91.40		Bi 66	The above specimen measured at H 2520 oersteds (H parallel to yz-plane) and at $\psi$ ranging from $10^\circ$ to $100^\circ$ .
46	472	L	1937	91.8		Bi 66	The above specimen measured at H 4850 oersteds (H parallel to yz-plane) and at $\psi = 10^\circ$ .
47	472	L	1937	92.0		Bi 66	The above specimen measured at H 4850 oersteds (H parallel to yz-plane) and at $\psi = 100^\circ$ .
48	472	L	1937	91.8-92.0		Bi 66	The above specimen measured at H 6100 oersteds (H parallel to yz-plane) and at $\psi$ ranging from $10^\circ$ to $-20^\circ$ .
49	472	L	1937	79.2		Bi 66	The above specimen measured without magnetic field.
50	472	L	1937	60.8		Bi 66	The above specimen measured without magnetic field.
51	472	L	1937	79.2		Bi 66	The above specimen measured at H 650 oersteds (H parallel to yz-plane) and at $\psi = 10^\circ$ .
52	472	L	1937	79.3		Bi 66	The above specimen measured at H 650 oersteds (H parallel to yz-plane) and at $\psi = 100^\circ$ .
53	472	L	1937	79.1		Bi 66	The above specimen measured at H 1500 oersteds (H parallel to yz-plane) and at $\psi = 10^\circ$ .
54	472	L	1937	79.4		Bi 66	The above specimen measured at H 1500 oersteds (H parallel to yz-plane) and at $\psi = 100^\circ$ .
55	472	L	1937	79.3		Bi 66	The above specimen measured at H 2520 oersteds (H parallel to yz-plane) and at $\psi = 10^\circ$ .
56	472	L	1937	79.4		Bi 66	The above specimen measured at H 2520 oersteds (H parallel to yz-plane) and at $\psi = 100^\circ$ .
57	472	L	1937	79.4		Bi 66	The above specimen measured at H 4850 oersteds (H parallel to yz-plane) and at $\psi = 10^\circ$ .
58	472	L	1937	79.4		Bi 66	The above specimen measured at H 4850 oersteds (H parallel to yz-plane) and at $\psi = 100^\circ$ .
59	472	L	1937	79.2		Bi 66	The above specimen measured at H 6100 oersteds (H parallel to yz-plane) and at $\psi = 10^\circ$ .
60	472	L	1937	79.4		Bi 66	The above specimen measured at H 6100 oersteds (H parallel to yz-plane) and at $\psi = 100^\circ$ .
61	472	L	1937	79.4		Bi 66	The above specimen measured at H 6100 oersteds (H parallel to yz-plane) and at $\psi = 100^\circ$ .
62	472	L	1937	91.2		Bi 51	Pure, single crystal; the angle between rod axis and trigonal axis $\psi = 86^\circ$ .
63	472	L	1937	91.2-91.5		Bi 51	The above specimen measured in a magnetic field parallel to the plane containing the trigonal axis (z-axis) and the rod axis with strength H 2520 oersteds and at $\psi'$ (angle between field direction and a line such that at $\psi = 7^\circ$ , H perpendicular z-axis and at $\psi' = 97^\circ$ , H parallel z-axis) ranging from $8^\circ$ to $-10^\circ$ .
64	472	L	1937	91.6-91.8		Bi 51	The above specimen measured at H 6100 oersteds (H parallel to the plane containing z-axis and the rod axis) and at $\psi'$ ranging from $7^\circ$ to $-5^\circ$ .
65	473	L	1934	80-207		Bi 9	Single crystal; 0.55 cm dia x 3.6 cm long; the angle between rod axis and trigonal axis about $80^\circ$ ; electrical resistivity reported as 46, 95, 48, 32, 49, 41, 103, 6 and 112.2 $\mu$ ohm cm at -193, 92, -188, 19, -183, 62, 0, and 21.02 C, respectively.

SPECIFICATION TABLE NO. 5 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)	Specifications and Remarks
66	473	L	1934	81.1		Bi 9		The above specimen measured in a transverse magnetic field (H perpendicular to rod axis) with H (field strength) = 5900 gauss and at $\theta$ (the angle between field and rod axis) ranging from 25 to 170°; electrical resistivity at -194.5 C reported as 601, 620, and 458 $\mu\text{ohm cm}$ at $\theta = 25^\circ, 35^\circ,$ and $170^\circ$ , respectively.
67	473	L	1934	89.4-90.5		Bi 9		The above specimen measured in a transverse magnetic field; H = 5900 gauss and $\theta$ ranging from 35 to 170°; electrical resistivity at -183.5 C reported as 510, 408, and 391 $\mu\text{ohm cm}$ at $\theta = 35^\circ, 95^\circ$ and $170^\circ$ , respectively.
68	425	L	1924	97-103	5			Polycrystal with fine grains; electrical conductivity 2.61 and $9.29 \times 10^3 \text{ ohm}^{-1}\text{cm}^{-1}$ at 90 and 273 K respectively. (Note the paper gives $10^4$ , probably a typographical error.)
69	425	L	1924	90, 273	5			Polycrystal with coarse grains; electrical conductivity 2.55 and $8.98 \times 10^3 \text{ ohm}^{-1}\text{cm}^{-1}$ at 90 and 273 K respectively. (Note the paper gives $10^4$ , probably a typographical error.)
70	368		1954	298.2				Fine-crystalline extruded specimen; concentration of current carriers $8.8 \times 10^{18} \text{ cm}^{-3}$ ; electrical conductivity $6760 \text{ ohm}^{-1}\text{cm}^{-1}$ at 25 C.
71	431	E	1944	372-501				Pure; polycrystal; electrical resistivity 179.20 to 277.00 $\mu\text{ohm cm}$ at 372.3 to 501.1 K.
72	754	L	1963	2.1-4.1				Very pure, single crystal; prepared from spectroscopically pure bismuth; light parallelepiped 24.3 x 6.9 x 2.5 mm; obtained from Johnson Matthey and Co.; crystal resistance ratio $R(300 \text{ K})/R(4.2 \text{ K}) = 40$ .
73	474	L	1950	82-90		Bi-S <sub>4</sub>		Single crystal; rod axis perpendicular to the trigonal axis and approximately parallel to one of the two-fold secondary axes (one side of the base triangle); electrical resistivity reported as 39.93 and 99.4 $\mu\text{ohm cm}$ at -187.5 and 0 C, respectively.
74	474	L	1950	83-89		Bi-S <sub>1</sub>		Single crystal; rod axis perpendicular to trigonal axis and perpendicular to one of the two-fold secondary axes (one side of the base triangle); electrical resistivity reported as 40.18 and 100.7 $\mu\text{ohm cm}$ at -187.5 and 0 C, respectively.
75	475	L	1936	78, 90		Bi 66		Pure; single crystal; the angle between trigonal axis and rod axis $\phi = 2^\circ$ ; 3 mm dia x 4 ~ 5 cm long; electrical resistivity reported as 36.1, 41.0 and 127.4 $\mu\text{ohm cm}$ at -195.39, -182.98 and 0 C, respectively.
76	475	L	1936	78, 90		Bi 13		Pure; single crystal; $\phi = 16^\circ$ ; 3 mm dia x 4 ~ 5 cm long; electrical resistivity reported as 37.4, 42.0, 86.6 and 25.6 $\mu\text{ohm cm}$ at -194.84, -183.20, -78.36 and 0 C, respectively.
77	475	L	1936	78, 90		Bi 51		Pure; single crystal; $\phi = 86^\circ$ ; 7 mm dia x 4 ~ 5 cm long; electrical resistivity reported as 37.0, 40.4 and 99.1 $\mu\text{ohm cm}$ at -194.78, -183.49 and 0 C, respectively.
78	475	L	1936	90.0		Bi 72		Pure; single crystal; $\phi = 85.5^\circ$ ; 3 mm dia x 4 ~ 5 cm long; electrical resistivity reported as 43.0 and 102.9 $\mu\text{ohm cm}$ at -183.13 and 0 C, respectively.
79	475	L	1936	91.8		Bi 66		Pure; single crystal; $\phi = 2^\circ$ ; 0.0898 cm <sup>2</sup> x 2.23 cm long.

SPECIFICATION TABLE NO. 5 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
80	475	L	1936	92.2		Bi 66	The above specimen measured in a transverse magnetic field (H perpendicular to rod axis) with strength H=2520 oersteds and field orientation $\theta$ (the angle between field direction and a line perpendicular to the rod axis such that at $\theta = -12^\circ$ , H is parallel to the x-axis and at $\theta = 78^\circ$ , H is parallel to the y-axis); $\phi = -161^\circ$ ; x-axis parallel to one of the two-fold secondary axes; z-axis coincided with the trigonal axis.
81	475	L	1936	79.6		Bi 66	The above specimen measured without magnetic field.
82	475	L	1936	79.8		Bi 66	The above specimen measured at H=1500 oersteds and at $\theta = 108^\circ$ .
83	475	L	1936	79.9		Bi 66	The above specimen measured at H=2520 oersteds and at $\theta = -41^\circ$ .
84	475	L	1936	79.9		Bi 66	The above specimen measured at H=4850 oersteds and at mean $\theta$ (averaged from values which varied from $15-30^\circ$ ).
85	475	L	1936	79.9		Bi 66	The above specimen measured at H=6100 oersteds and at mean $\theta$ .
86	475	L	1936	91.0		Bi 13	Pure, single crystal; the angle between trigonal axis and rod axis $\phi = 16^\circ$ ; 0.1452 cm <sup>2</sup> x 3.10 cm long.
87	475	L	1936	91.2		Bi 13	The above specimen measured in a transverse magnetic field (H perpendicular to rod axis) with strength H=2460 oersteds and field orientation at $\theta = 4^\circ$ and $38^\circ$ , where $\theta$ is the angle between field direction and a line perpendicular to the rod axis such that at $\theta = 6^\circ$ , H approx. parallel to x-axis.
88	475	L	1936	79, 91		Bi 13	The above specimen measured at H=6100 oersteds and at mean $\theta$ (averaged from values which varied from $15-30^\circ$ ).
89	475	L	1936	78.5		Bi 13	The above specimen measured without magnetic field.
90	475	L	1936	78.7		Bi 13	The above specimen measured at H=2460 oersteds and at $\theta = -4^\circ$ and $38^\circ$ .
91	475	L	1936	78.7		Bi 13	The above specimen measured at H=6100 oersteds and at $\theta = 38^\circ$ .
92	475	L	1936	79, 91		Bi 51	Pure, single crystal; the angle between trigonal axis and rod axis $\phi = 86^\circ$ ; 0.0749 cm <sup>2</sup> x 2.18 cm long.
93	475	L	1936	91.1		Bi 51	The above specimen measured in a transverse magnetic field (H perpendicular to rod axis) with strength H=2520 oersteds and field orientation at $\theta = -152^\circ$ , where $\theta$ is the angle between field direction and a line perpendicular to the rod axis such that at $\theta = 35^\circ$ , H parallel to negative x-axis and at $\theta = -55^\circ$ , H parallel to z-axis.
94	475	L	1936	91.4		Bi 51	The above specimen measured at H=4900 oersteds and at $\theta = -152^\circ$ .
95	475	L	1936	79.3		Bi 51	The above specimen measured at H=2520 oersteds and at mean $\theta$ (averaged from values which varied from $15-30^\circ$ ).

SPECIFICATION TABLE NO. 5 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
96	475	L	1936	91.3		Bi 72	Pure; single crystal; the angle between trigonal axis and rod axis $\phi = 85.5^\circ$ ; $0.0907 \text{ cm}^2 \times 2.51 \text{ cm}$ long.
97	475	L	1936	91.3		Bi 72	The above specimen measured in a transverse magnetic field (H perpendicular to rod axis) with strength H=650 oersteds and field orientation at $\theta = 2^\circ$ , where $\theta$ is the angle between field direction and a line perpendicular to the rod axis such that at $\theta = \theta^\circ$ , H is parallel to negative z-axis and at $\theta = 98^\circ$ , H is perpendicular to z-axis.
98	475	L	1936	91.3		Bi 72	The above specimen measured at H=650 oersteds and at $\theta = -124^\circ$ .
99	475	L	1936	91.2-91.4		Bi 72	The above specimen measured at H=1500 oersteds and at $\theta$ ranging from $2^\circ$ to $-34^\circ$ .
100	475	L	1936	91.2		Bi 72	The above specimen measured at H=2520 oersteds and at $\theta = 2^\circ$ .
101	475	L	1936	91.3		Bi 72	The above specimen measured at H=2520 oersteds and at $\theta = -124^\circ$ .
102	475	L	1936	91.6		Bi 72	The above specimen measured at H=4900 oersteds and at $\theta = 2^\circ$ .
103	475	L	1936	91.6		Bi 72	The above specimen measured at H=4900 oersteds and at $\theta = -124^\circ$ .
104	475	L	1936	91.3-91.5		Bi 72	The above specimen measured at H=6100 oersteds and at $\theta$ ranging from $2^\circ$ to $-34^\circ$ .
105	383		1956	320-500			Pure; electrical conductivity reported as $0.744$ , $0.684$ , $0.602$ , $0.544$ , $0.516$ , $0.509$ , $0.477$ , $0.446$ , $0.421$ and $0.380 \times 10^4 \text{ ohm}^{-1}\text{cm}^{-1}$ at $46.7$ , $69.1$ , $105.8$ , $134.6$ , $148.7$ , $152.1$ , $169.0$ , $188.8$ , $203.8$ and $227.0 \text{ C}$ , respectively.
106	460		1957	373.2			Pure.
107	470	L	1949	287.2			Pure; single crystal; bismuth plate, $2 \times 9.3 \times 9.3 \text{ mm}$ ; supplied by Merck; $\theta$ (the angle between the trigonal axis of the crystal and the direction of heat flow) $= 0^\circ$ .
108	470	L	1949	287.2			Similar to the above specimen except $\theta = 14.7^\circ$ .
109	470	L	1949	287.2			Similar to the above specimen except $\theta = 27.2^\circ$ .
110	470	L	1949	287.2			Similar to the above specimen except $\theta = 50.7^\circ$ .
111	470	L	1949	287.2			Similar to the above specimen except $\theta = 67.9^\circ$ .
112	470	L	1949	287.2			Similar to the above specimen except $\theta = 74.3^\circ$ .
113	470	L	1949	287.2			Similar to the above specimen except $\theta = 99.0^\circ$ .
114	577	L	1913	83-373			Pure; cylindrical specimen; made from bismuth supplied by C.A.F. Kahlebaum; bismuth powder pressed at $5000 \text{ kg cm}^{-2}$ for 1 hour; density $1\%$ less than cast bismuth; electrical conductivity reported as $3.5$ , $3.8$ , and $3.2 \times 10^3 \text{ ohm}^{-1}\text{cm}^{-1}$ at $-190$ , $0$ , and $100 \text{ C}$ , respectively.
115	597, 704	L	1961	313-630			Measurements made on solid specimen and molten specimen (from same source as the solid specimen); $3 \text{ mm}$ in diameter, $64 \text{ mm}$ long used to determine data in liquid state; melting point $544.2 \text{ K}$ .



SPECIFICATION TABLE NO. 5 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
116	592	L	1959	626-945	~6		In liquid state; melting point 271°.
117	755	L	1962	100-300	~6		99.999 pure; single crystal; specimen 2 x 2 x 10 mm; provided by American Smelting and Refining Co.; as received; electrical resistivity 34.2, 42.4, 85.3 and 134.9 $\mu\text{ohm cm}$ at 79, 100, 200 and 300 K, respectively; heat flow parallel to trigonal axis.
118	755	L	1962	100-300	~6		Similar to the above specimen except electrical resistivity 32.5, 39.0, 73.1 and 111.7 $\mu\text{ohm cm}$ at 81, 100, 200 and 300 K, respectively; heat flow perpendicular to trigonal axis.
119	706	L	1881	273, 373			Density 9.74 $\text{g cm}^{-3}$ ; electrical conductivity 0.929 and 0.630 $\times 10^4 \text{ ohm}^{-1} \text{cm}^{-1}$ at 0 and 100 C respectively. (electrical conductivity reported as 0.929 and 0.63 $\times 10^5 \text{ ohm}^{-1} \text{cm}^{-1}$ , probably a typographical error)
120	331	L	1964	1.46			Single crystal with rhombohedral structure; 16.8 x 4.55 x 2.0 mm, with trigonal axis parallel to the small dimension and bisectrix parallel to the large dimension; specimen taken from cylindrical ingot supplied by Texas Instruments Corporation; electrical resistivity reported as 1.1 and 1.2 $\mu\text{ohm cm}$ at 1.12 and 4.2 K, respectively; 0.288, 2/04.2 reported as 91.2; measured with the magnetic field directed along the trigonal axis, and the heat flow along the large dimension; magnetic field ranging from 1.02 to 17.98 kilogauss.
121	333	L	1929	300.2	± .50		Single crystal; disk specimen 25 mm in diameter and 2 mm thick; heat flow perpendicular to the trigonal axis; measured in magnetic fields (H) ranging from 2499 to 10554 gauss perpendicular to the trigonal axis.
122	333	L	1929	300.2	± .50		Similar to the above specimen except measured in magnetic fields parallel to the trigonal axis ranging from 0 to 10773 gauss.
123	333	L	1929	300.2	± .50		Single crystal; disk specimen 25 mm in dia and 2.0 mm thick; heat flow parallel to the trigonal axis; measured in magnetic fields perpendicular to the trigonal axis ranging from 0 to 11161 gauss.
124	333	L	1929	300.2	± .50		Similar to the above specimen except disk 2.02 mm thick and measured in magnetic fields ranging from 5071 to 9847 gauss.
125	333	L	1929	300.2	± .50		The above two specimens combined together (4.02 mm thick); measured in the same conditions as above with magnetic fields ranging from 4997 to 9973 gauss.
126	838	C	1967	453-770			The molten specimen placed in a hole 21 mm in dia drilled in an asbestos cement cylinder of 30 mm height; steel Khl 8N5T used as comparative material.
127	120, 323	L	1934	110-209			Single crystal; specimen length 1.231 cm cross-section roughly triangular in shape with dimensions about 3 mm on a side; trigonal axis parallel to the length; supplied by Prof A. Goetz of CIT; measured in a vacuum of $10^{-6}$ mm Hg, and in a magnetic field of 7806 gauss parallel to one of the binary axes; heat flow along the trigonal axis.

SPECIFICATION TABLE NO. 5 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
128	120, 323	L	1934	117-190		1	The above specimen measured with the magnetic field perpendicular to one of the binary axes.
129	120, 323	L	1934	114-198		2	Single crystal, specimen length 1.3 cm, cross-section roughly triangular in shape with dimensions about 3 mm on a side; trigonal axis perpendicular to the length; measured in a vacuum of $10^{-5}$ mm Hg and in a magnetic field of 7800 gauss perpendicular to the trigonal axis; heat flow perpendicular to the trigonal axis.
130	120, 323	L	1934	108-216		2	The above specimen measured with the magnetic field parallel to the trigonal axis.
131	120, 323	L	1934	122-145		2	The above specimen measured with the magnetic field at $45^\circ$ to the trigonal axis.
132	872	-	1967	1.3-1.9		Sample 3	99.9999 pure; single crystal; specimen 1.57 x 3.1 mm in cross-section; specimen axis along the bisectrix; electrical resistivity ratio $\rho(300)/\rho(4.2) = 140$ ; thermal conductivity values calculated from heat capacity, velocity and effective mean free path.
133	872	-	1967	1.3-2.0		Sample 5	99.9999 pure; single crystal; specimen 3.8 x 3.85 mm in cross-section; specimen axis along the trigonal; electrical resistivity ratio $\rho(300)/\rho(4.2) = 104$ ; thermal conductivity values calculated from heat capacity, velocity and effective mean free path.
134	1002	L	1967	80-301			Single crystal; 0.4 x 0.2 x 0.2 in.; electrical resistivity reported as 0.0349, 0.0472, 0.0680, 0.0854, 0.113, and 0.134 milliohm-cm at 80, 112, 160, 198, 253, and 299 K, respectively; heat flow along the trigonal axis.
135	1002	L	1967	81-303			Single crystal; 0.4 x 0.2 x 0.2 in.; electrical resistivity reported as 0.0324, 0.0445, 0.0602, 0.0727, 0.0950 and 0.116 milliohm-cm at 77, 115, 156, 193, 249 and 298 K, respectively; heat flow along one of the binary axes.

## DATA TABLE NO. 5 THERMAL CONDUCTIVITY OF BISMUTH

(Impurity &lt; 0.20% each; total impurities &lt; 0.50%)

[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup> K<sup>-1</sup>]

T	k	T	k	T	k	T	k	T	k	T	k	T	k
<u>CURVE 1</u>													
2.30	5.847	19.47	0.719	20.01	0.507	3.58	1.79	20.7	0.830	12.3	1.58	20.7	0.830
2.55	3.771	20.44	0.671	61.44	0.166	3.81	2.80	25.4	0.650	15.0	1.24	25.4	0.650
3.25	14.010	21.13	0.613	<u>CURVE 12</u>		4.31	2.33	28.9	0.550	19.4	0.91	33.3	0.480
3.47	15.924	66.71	0.181	16.50	0.770	4.45	2.41	33.3	0.480	24.3	0.71	33.3	0.480
4.05	17.513	67.70	0.179	17.53	0.891	6.08	2.98	59.5	0.252	32.6	0.49	59.5	0.252
14.35	1.239	67.90	0.178	18.53	0.921	6.80	2.72	69.4	0.220	78.5	0.22	69.4	0.220
16.68	1.064	70.63	0.174	19.53	0.837	7.09	2.55	80.0	0.206	<u>CURVE 20</u>			
19.57	0.877	73.17	0.168	81.47	0.259	7.95	2.22	91.0	0.179	2.17	3.66	2.69	6.47
20.16	0.840	75.21	0.164	<u>CURVE 13</u>		8.40	1.97	<u>CURVE 18</u>		3.12	9.60	3.12	9.60
66.80	0.251	76.87	0.159	11.55	1.47	9.70	1.70	1.99	0.83	3.66	14.60	2.41	1.43
69.00	0.240	78.35	0.157	14.02	1.12	17.9	0.918	2.41	1.43	4.22	15.9	2.91	2.39
74.00	0.230	79.67	0.156	17.9	0.918	20.6	0.784	3.69	4.69	4.32	14.0	3.69	4.69
75.50	0.225	83.07	0.152	24.6	0.644	24.6	0.644	4.34	6.30	6.16	7.38	4.46	6.36
<u>CURVE 2</u>													
291.2	0.0810	19.38	0.769	16.51	0.987	28.4	0.544	5.55	5.45	7.49	4.18	5.55	5.45
373.2	0.0674	19.70	0.763	17.52	0.987	33.6	0.465	6.15	4.43	9.06	2.70	6.15	4.43
<u>CURVE 3</u>													
293.2	0.081	19.91	0.757	18.04	0.996	63.6	0.204	7.52	3.22	31.9	0.545	7.52	3.22
313.2	0.079	20.22	0.741	81.48	0.206	68.6	0.204	8.61	2.40	60.0	0.274	8.61	2.40
333.2	0.078	20.57	0.730	<u>CURVE 14</u>		79.6	0.180	10.57	1.73	78.6	0.22	10.57	1.73
353.2	0.077	21.23	0.675	258.2	0.0925	<u>CURVE 17</u>		12.70	1.35	90.7	0.20	12.70	1.35
373.2	0.075	21.93	0.645	323.2	0.0883	2.07	1.14	15.99	1.10	<u>CURVE 21</u>			
<u>CURVE 4</u>													
573.0	0.113	67.02	0.185	423.2	0.0710	2.29	1.54	64.2	0.237 <sup>c</sup>	<u>CURVE 22<sup>d</sup></u>			
623.0	0.118	72.83	0.178	<u>CURVE 15</u>		2.59	2.17	78.2	0.200 <sup>e</sup>	<u>CURVE 23<sup>d</sup></u>			
673.0	0.123	73.83	0.172	298.2	0.0540	2.69	2.42	<u>CURVE 19</u>		<u>CURVE 24</u>			
723.0	0.128	75.68	0.167	323.2	0.0523	3.10	3.35	2.01	1.36	<u>CURVE 25</u>			
773.0	0.134	77.08	0.166	423.2	0.0441	3.22	3.77	2.21	1.77	<u>CURVE 26</u>			
823.0	0.139	77.55	0.165	423.2	0.0419	3.40	4.37	2.57	2.63	<u>CURVE 27</u>			
<u>CURVE 5</u>													
83.2	0.261	152.5	0.132	298.2	0.0540	3.64	4.94	4.48	9.29	<u>CURVE 28</u>			
196.2	0.108	160.0	0.129	323.2	0.0523	4.01	5.95	5.33	8.37	<u>CURVE 29</u>			
273.2	0.102	172.6	0.123	423.2	0.0419	4.59	7.15	5.84	6.92	<u>CURVE 30</u>			
373.2	0.0967	188.8	0.116	<u>CURVE 16</u>		6.18	5.34	6.52	5.36	<u>CURVE 31</u>			
<u>CURVE 6</u>													
83.2	0.261	208.4	0.114	1.93	0.385	7.78	2.96	6.52	5.36	<u>CURVE 32</u>			
196.2	0.108	362.2	0.0757	2.46	0.718	9.09	2.31	7.60	3.72	<u>CURVE 33</u>			
273.2	0.102	433.2	0.0711	2.81	0.924	11.17	1.72	9.14	2.49	<u>CURVE 34</u>			
373.2	0.0967	495.2	0.0741	2.83	1.07	13.8	1.31	10.4	1.98	<u>CURVE 35</u>			
<u>CURVE 7</u>													
506.2	0.0741	506.2	0.0741	3.27	1.30	17.5	0.996	11.36	1.76	<u>CURVE 36</u>			

<sup>c</sup> Not shown on plot

DATA TABLE NO. 5 (continued)

T	k	ψ(deg)	T	k	ψ(deg)	T	k	ψ(deg)	T	k	ψ(deg)
287.2	0.0533		91.77	0.1143		79.37	0.1144		115	0.1590	
											ψ(deg)
											ψ(deg)
											ψ(deg)
287.2	0.0794		91.96	0.0939		79.30	0.1392		-135	0.1575	
											ψ(deg)
											ψ(deg)
											ψ(deg)
287.2	0.0950										ψ(deg)
											ψ(deg)
											ψ(deg)
											ψ(deg)
287.2	0.0794		91.60	0.1245		79.20	0.1327		-112	0.1609	
											ψ(deg)
											ψ(deg)
											ψ(deg)
287.2	0.0908		91.77	0.1088		79.43	0.1123		-95	0.1634	
											ψ(deg)
											ψ(deg)
											ψ(deg)
287.2	0.0795		91.65	0.1045		79.42	0.1167		-80	0.1428	
											ψ(deg)
											ψ(deg)
											ψ(deg)
313.2	0.0460		91.77	0.1088		79.42	0.1167		-65	0.1430	
											ψ(deg)
											ψ(deg)
											ψ(deg)
321.2	0.0460		91.65	0.1045		79.42	0.1167		-56	0.1440	
											ψ(deg)
											ψ(deg)
											ψ(deg)
335.2	0.0444		91.77	0.1088		79.42	0.1167		-35	0.1449	
											ψ(deg)
											ψ(deg)
											ψ(deg)
358.2	0.0444		91.65	0.1045		79.42	0.1167		-5	0.1495	
											ψ(deg)
											ψ(deg)
											ψ(deg)
375.2	0.0439		91.77	0.1088		79.42	0.1167				ψ(deg)
											ψ(deg)
											ψ(deg)
											ψ(deg)
383.2	0.0427		91.65	0.1045		79.42	0.1167				ψ(deg)
											ψ(deg)
											ψ(deg)
											ψ(deg)
409.2	0.0418		91.77	0.1088		79.42	0.1167				ψ(deg)
											ψ(deg)
											ψ(deg)
											ψ(deg)
421.2	0.0427		91.65	0.1045		79.42	0.1167				ψ(deg)
											ψ(deg)
											ψ(deg)
											ψ(deg)
453.2	0.0418		91.77	0.1088		79.42	0.1167				ψ(deg)
											ψ(deg)
											ψ(deg)
											ψ(deg)
87.2	0.233		91.77	0.1088		79.42	0.1167				ψ(deg)
											ψ(deg)
											ψ(deg)
											ψ(deg)
194.2	0.105		91.65	0.1045		79.42	0.1167				ψ(deg)
											ψ(deg)
											ψ(deg)
											ψ(deg)
291.2	0.0907		91.77	0.1088		79.42	0.1167				ψ(deg)
											ψ(deg)
											ψ(deg)
											ψ(deg)
421.2	0.0427		91.65	0.1045		79.42	0.1167				ψ(deg)
											ψ(deg)
											ψ(deg)
											ψ(deg)
453.2	0.0418		91.77	0.1088		79.42	0.1167				ψ(deg)
											ψ(deg)
											ψ(deg)
											ψ(deg)

Not shown on plot

DATA TABLE NO. 5 (continued)

T	k	T	k	T	k	θ(deg)	k	θ(deg)	k	6(deg)	k	T	k	T	k
<u>CURVE 72</u>															
2.05	4.3	91.8	0.1268	78.5	0.1535	<u>CURVE 99<sup>a</sup></u> (T = 91.23-91.36 K)		<u>CURVE 104 (cont.)</u>		<u>CURVE 112<sup>a</sup></u>		<u>CURVE 116 (cont.)</u>			
2.85	10.0	<u>CURVE 80<sup>a</sup></u>		2	0.1617	-124	0.1398	287.2	0.0879	823.2	0.169 <sup>a</sup>				
3.65	15.0	<u>CURVE 81<sup>a</sup></u>		32	0.1541	-94	0.1465	<u>CURVE 113</u>							
4.13	14.5	<u>CURVE 82<sup>a</sup></u>		62	0.1591	-64	0.1443								
<u>CURVE 73<sup>a</sup></u>															
81.95	0.2008	92.2	0.0994	86	0.1492	-34	0.1380	287.2	0.0908	853.2	0.174 <sup>a</sup>				
82.5	0.2008	<u>CURVE 81<sup>a</sup></u>		116	0.1499	<u>CURVE 105</u>		<u>CURVE 114<sup>a</sup></u>							
85.2	0.1969	<u>CURVE 82<sup>a</sup></u>		146	0.1530	319.9	0.0904	83.2	0.208	882.2	0.177 <sup>a</sup>				
85.75	0.1963	<u>CURVE 83<sup>a</sup></u>		-178	0.1625	342.3	0.0904	273.2	0.0812	896.2	0.170 <sup>a</sup>				
86.2	0.1961	<u>CURVE 84<sup>a</sup></u>		-148	0.1562	379.0	0.0891	373.2	0.0506	905.2	0.176 <sup>a</sup>				
88.65	0.1894	<u>CURVE 85<sup>a</sup></u>		-124	0.1536	407.8	0.0874	<u>CURVE 115</u>							
89.75	0.1894	<u>CURVE 86<sup>a</sup></u>		-94	0.1539	421.9	0.0858	83.2	0.208						
<u>CURVE 14<sup>a</sup></u>															
82.5	0.198	79.8	0.1232	78.7	0.1227	425.3	0.0849	313.3	0.0711	100	0.1274				
84.7	0.1951	<u>CURVE 87<sup>a</sup></u>		38	0.1236	442.2	0.0837	397.8	0.0669	119	0.1126				
86.2	0.1932	<u>CURVE 88<sup>a</sup></u>		79.1	0.2031	462.0	0.0824	460.4	0.0689	149	0.0968				
87.6	0.1903	<u>CURVE 89<sup>a</sup></u>		91.1	0.1828	477.0	0.0828	481.2	0.0628	199	0.0791				
88.9	0.1869	<u>CURVE 90<sup>a</sup></u>		91.24	0.1537	500.2	0.0854	537.8	0.0669	249	0.0672				
89.2	0.186	<u>CURVE 91<sup>a</sup></u>		<u>CURVE 101</u>		<u>CURVE 106<sup>a</sup></u>		540.7	0.0753	300	0.0607				
<u>CURVE 15<sup>a</sup></u>															
77.76	0.147	79.9	0.1164	79.9	0.1164	571.2	0.138	336.8	0.142	<u>CURVE 118</u>					
80.17	0.129	<u>CURVE 92<sup>a</sup></u>		91.29	0.1474	373.2	0.0837	598.6	0.142	100	0.1741				
<u>CURVE 16<sup>a</sup></u>															
78.31	0.154	79.9	0.1150	91.1	0.1542	<u>CURVE 107</u>		629.7	0.142	120	0.1646				
90.0	0.137	<u>CURVE 93<sup>a</sup></u>		<u>CURVE 102</u>		<u>CURVE 108</u>		<u>CURVE 116</u>		150	0.1480				
<u>CURVE 17<sup>a</sup></u>															
77.76	0.147	91.4	0.1457	91.55	0.1403	287.2	0.0527	626.2	0.151	200	0.1265				
80.17	0.129	<u>CURVE 94<sup>a</sup></u>		<u>CURVE 103<sup>a</sup></u>		<u>CURVE 109</u>		636.7	0.144 <sup>a</sup>	300	0.0985				
<u>CURVE 18<sup>a</sup></u>															
78.31	0.154	91.0	0.1361	91.56	0.1416	287.2	0.0556	647.2	0.160	<u>CURVE 119<sup>a</sup></u>					
90.0	0.137	<u>CURVE 95<sup>a</sup></u>		<u>CURVE 104<sup>a</sup></u>		<u>CURVE 110</u>		675.2	0.158 <sup>a</sup>	273.2	0.0741				
<u>CURVE 19<sup>a</sup></u>															
78.31	0.154	79.3	0.1723	79.3	0.1723	287.2	0.0594	684.2	0.155	373.2	0.0686				
90.0	0.137	<u>CURVE 96<sup>a</sup></u>		<u>CURVE 105<sup>a</sup></u>		<u>CURVE 111</u>		693.2	0.156 <sup>a</sup>	200	0.1117				
<u>CURVE 20<sup>a</sup></u>															
78.31	0.154	91.25	0.1733	91.25	0.1733	287.2	0.0594	701.7	0.151	<u>CURVE 120</u>					
89.66	0.185	<u>CURVE 97<sup>a</sup></u>		<u>CURVE 106<sup>a</sup></u>		<u>CURVE 112</u>		713.2	0.169 <sup>a</sup>	H(gauss)	k				
90.02	0.175	4	0.1101	2	0.1371	743.2	0.164 <sup>a</sup>	724.2	0.163 <sup>a</sup>	<u>CURVE 120</u>					
<u>CURVE 21<sup>a</sup></u>															
78.37	0.204	36	0.1080	32	0.1323	749.7	0.165 <sup>a</sup>	731.7	0.161	<u>(T = 1.46 K)</u>					
89.66	0.185	<u>CURVE 98<sup>a</sup></u>		56	0.1312	756.2	0.165 <sup>a</sup>	743.2	0.164 <sup>a</sup>	1.02	0.0684				
90.02	0.175	91.26	0.1608	86	0.1337	761.2	0.165 <sup>a</sup>	756.2	0.165 <sup>a</sup>	1.98	0.0684 <sup>a</sup>				
<u>CURVE 22<sup>a</sup></u>															
78.7	0.1220	91.26	0.1608	116	0.1371	768.2	0.160 <sup>a</sup>	761.2	0.160 <sup>a</sup>	2.74	0.0682 <sup>a</sup>				
91.2	0.1042	91.26	0.1608	146	0.1394	772.2	0.160	772.2	0.160	3.54	0.0682 <sup>a</sup>				
<u>CURVE 23<sup>a</sup></u>															
78.7	0.1220	91.26	0.1608	-178	0.1383	782.0	0.169 <sup>a</sup>	782.0	0.169 <sup>a</sup>	4.52	0.0681 <sup>a</sup>				
91.2	0.1042	91.26	0.1608	-148	0.1352	796.7	0.164 <sup>a</sup>	796.7	0.164 <sup>a</sup>	5.31	0.0680 <sup>a</sup>				

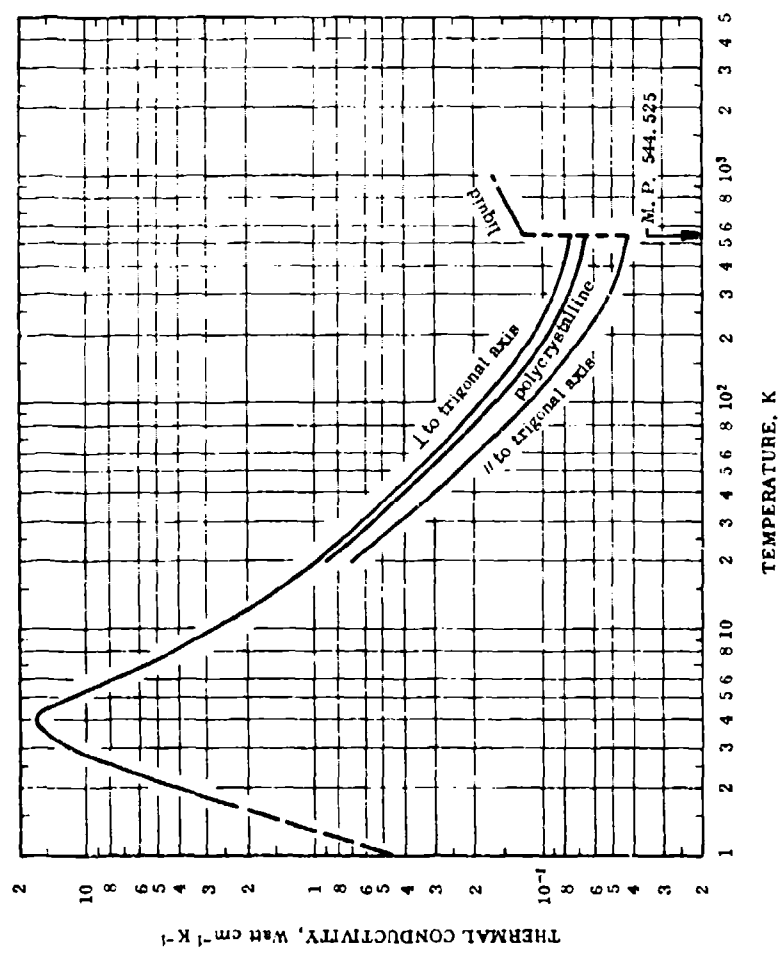
<sup>a</sup> Not shown on plot

DATA TABLE NO. 5 (continued)

H (gauss)	k	H (gauss)	k	T	k	H (gauss)	k	T	k
<b>CURVE 120 (cont.)</b>									
6.91	0.0675*	0000	0.0542	110.2	0.0887	122.1	0.117	197	0.115
8.56	0.0675*	1040	0.0538	119.4	0.0799	133.3	0.111	252	0.100
10.27	0.0671	2049	0.0530	126.1	0.0686	136.3	0.109	303	0.0925
11.47	0.0669*	2592	0.0520	137.5	0.0661	140.1	0.108		
12.15	0.0666*	3038	0.0519	142.1	0.0640	141.4	0.105		
13.25	0.0663	4010	0.0509	146.3	0.0615	145.2	0.103		
14.31	0.0659*	4986	0.0500	149.6	0.0598	<b>CURVE 132</b>			
15.24	0.0655*	5543	0.0493	152.6	0.0586	1.28	0.620		
16.33	0.0652*	5936	0.0491	171.9	0.0536	1.35	0.725		
17.36	0.0649	6878	0.0483	186.5	0.0485	1.40	0.821		
17.98	0.0646*	6981	0.0485	209.1	0.0477	1.42	0.843		
<b>CURVE 121*</b>									
<b>(T = 300.2 K)</b>									
7816	0.0477	8565	0.0471	<b>CURVE 128</b>					
9480	0.0464	9552	0.0465	117.1	0.0782	1.47	0.934		
10640	0.0457	11023	0.0450	129.4	0.0686	1.49	0.973		
11161	0.0456	11161	0.0456	141.4	0.0623	1.54	1.07		
<b>CURVE 124</b>									
<b>(T = 300.2 K)</b>									
5071	0.0497	9847	0.0469	148.9	0.0577	1.61	1.22		
1536	0.0930	1537	0.0930	156.3	0.0544	1.68	1.38		
1820	0.0926	2555	0.0919	162.1	0.0519	1.79	1.68		
2596	0.0919	3477	0.0917	172.7	0.0510	1.91	2.01		
3977	0.0916	5442	0.0905	<b>CURVE 129*</b>					
5992	0.0902	7569	0.0893	113.6	0.129	1.26	0.563		
7892	0.0888	9516	0.0882	130.7	0.114	1.31	0.643		
9818	0.0869	10534	0.0868	144.0	0.105	1.36	0.725		
10724	0.0872	10773	0.0870	158.1	0.0987	1.41	0.813		
<b>CURVE 125*</b>									
<b>(T = 300.2 K)</b>									
4997	0.0502	5034	0.0504	169.8	0.0967	1.48	0.948		
9973	0.0475	9973	0.0475	177.7	0.0941	1.52	1.04		
<b>CURVE 126</b>									
<b>(T = 300.2 K)</b>									
453.2	0.0753	490.2	0.0711	179.7	0.0912	1.61	1.28		
537.2	0.0628	555.2	0.1297	182.0	0.0926	1.72	1.55		
619.2	0.1381	712.2	0.1465	197.7	0.0912	1.82	1.86		
770.2	0.1465	770.2	0.1465	<b>CURVE 130</b>					
<b>(T = 300.2 K)</b>									
108.0	0.144	114.3	0.132	108.0	0.144	<b>CURVE 134*</b>			
128.4	0.121	137.4	0.114	114.3	0.132	80	0.134		
146.9	0.110	158.3	0.103	128.4	0.121	156	0.0753		
174.2	0.0992	215.6	0.0950	146.9	0.110	198	0.0672		
<b>CURVE 135*</b>									
<b>(T = 300.2 K)</b>									
81	0.194	105	0.158	158.3	0.103	252	0.0572		
155	0.134	155	0.134	174.2	0.0992	301	0.0530		
<b>CURVE 135*</b>									
<b>(T = 300.2 K)</b>									
81	0.194	105	0.158	<b>CURVE 135*</b>					
<b>(T = 300.2 K)</b>									

\* Not shown on plot

FIGURE AND TABLE NO. 5R RECOMMENDED THERMAL CONDUCTIVITY OF BISMUTH



REMARKS

The recommended values are for 99.997% pure bismuth. The recommended values that are supported by experimental thermal conductivity data are thought to be accurate to within 4% of the true values near room temperature and 4 to 10% at other temperatures above 10 K. The thermal conductivity near and below the corresponding temperature of its maximum is highly sensitive to small physical and chemical variations of the specimens, and the values below 10 K are intended as typical values for indicating the general trend.

T <sub>1</sub>	Single Crystal (// to c-axis)			Polycrystalline			T <sub>2</sub>
	k <sub>1</sub>	k <sub>2</sub>	k <sub>3</sub>	k <sub>1</sub>	k <sub>2</sub>	k <sub>3</sub>	
0	0	0	0	0	0	0	-459.7
1	(0.452) <sup>‡</sup>	(26.1)					-457.9
2	3.94	228					-456.1
3	11.8	682					-454.3
4	17.1	988					-452.5
5	11.9	688					-450.7
6	7.98	461					-448.9
7	5.77	333					-447.1
8	4.40	254					-445.3
9	3.50	202					-443.5
10	2.88	166					-441.7
11	2.45	142					-439.9
12	2.11	122					-438.1
13	1.85	107					-436.3
14	1.65	95.3					-434.5
15	1.48	85.5					-432.7
16	1.36	78.6					-430.9
18	1.15	66.4					-427.3
20	1.00	57.8					-423.7
25	0.780	45.1	0.538	40.4	0.695	52.0	-414.7
30	0.635	36.7	0.434	25.1	0.568	32.8	-405.7
35	0.536	31.0	0.364	19.6	0.478	27.6	-396.7
40	0.465	26.9	0.311	16.0	0.414	23.9	-387.7
45	0.410	23.7	0.272	15.7	0.365	21.1	-378.7
50	0.367	21.2	0.243	14.0	0.326	18.8	-369.7
60	0.303	17.5	0.199	11.5	0.268	15.5	-351.7
70	0.260	15.0	0.168	9.71	0.231	13.3	-333.7
80	0.230	13.3	0.148	8.55	0.203	11.7	-315.7
90	0.206	11.9	0.131	7.57	0.182	10.5	-297.7
100	0.188	10.9	0.119	6.88	0.165	9.53	-279.7
150	0.136	7.86	0.0826	4.77	0.118	6.82	-189.7
200	0.112	6.47	0.0667	3.85	0.0969	5.60	-99.7
250	0.0995	5.75	0.0581	3.36	0.0857	4.95	-9.7
273.2	0.0953	5.51	0.0554	3.20	0.0822	4.75	32.0
300	0.0915	5.29	0.0528	3.05	0.0786	4.54	80.3
350	0.0860	4.97	0.0491	2.84	0.0737	4.26	170.3

<sup>‡</sup>T<sub>1</sub> in K, k<sub>1</sub> in Watt cm<sup>-1</sup> K<sup>-1</sup>, T<sub>2</sub> in F, and k<sub>2</sub> in Btu lb<sup>-1</sup> ft<sup>-1</sup> F<sup>-1</sup>. <sup>†</sup>Values in parentheses are extrapolated.

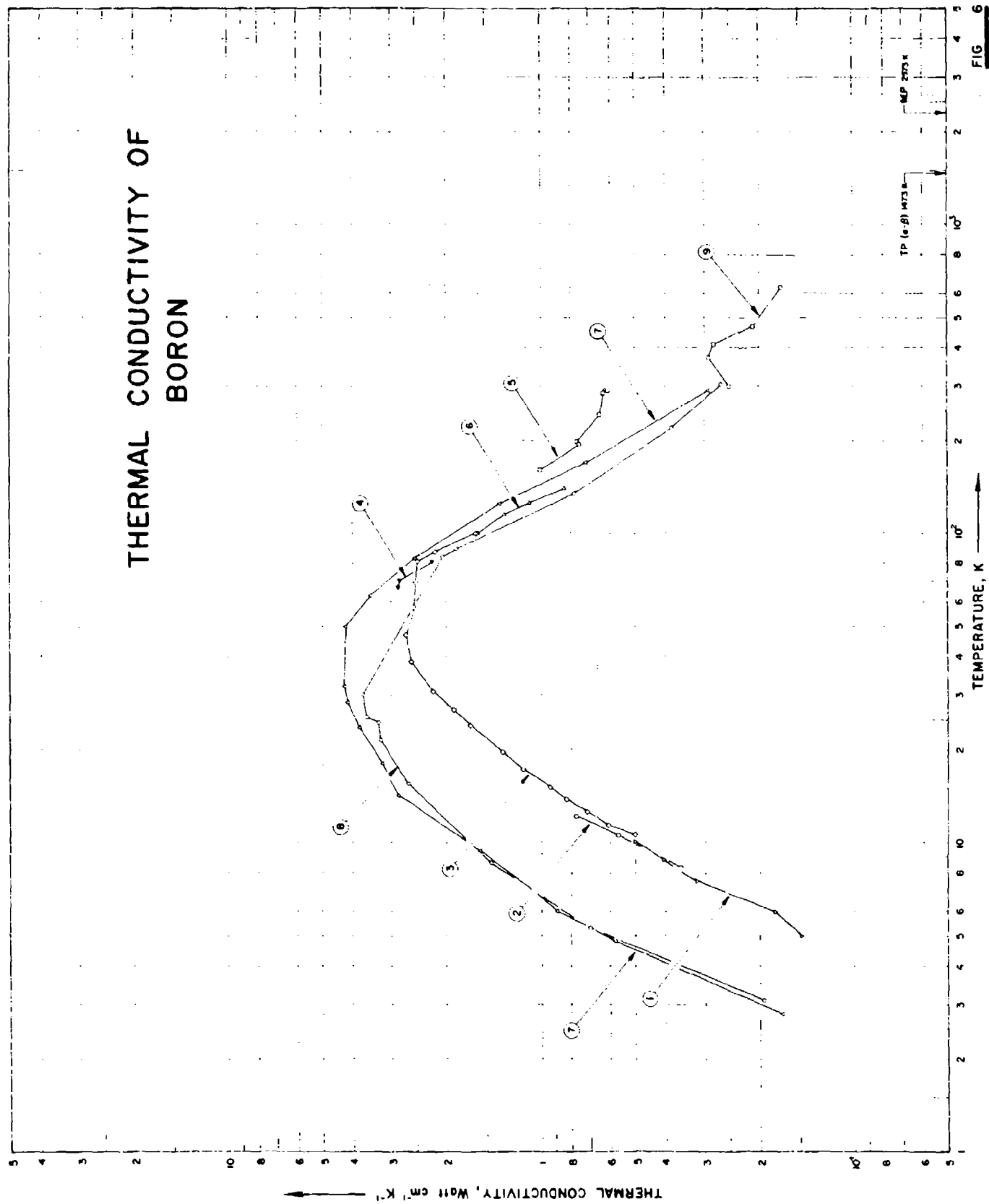
TABLE NO. 5R (continued)

T <sub>1</sub>	Single Crystal (⊥ to c-axis)		Single Crystal (// to c-axis)		Polycrystalline	
	k <sub>1</sub>	k <sub>2</sub>	k <sub>1</sub>	k <sub>2</sub>	k <sub>1</sub>	k <sub>2</sub>
400	0.0422	4.75	0.0469	2.71	0.0704	4.07
500	0.0775	4.48	0.0438	2.53	0.0663	3.83
544.525	0.0761	4.40	0.0429	2.48	0.0650	3.76
			In Liquid State			
			T <sub>1</sub>	k <sub>1</sub>	k <sub>2</sub>	T <sub>2</sub>
			544.525	0.124	7.16	520.457
			600	0.131	7.57	620.3
			700	0.141	8.15	800.3
			800	0.150	8.67	960.3
			900	0.159	9.19	1160
			1000	(0.168)*	(9.71)	1340

\* Values in parentheses are extrapolated.



# THERMAL CONDUCTIVITY OF BORON



## SPECIFICATION TABLE NO. 6 THERMAL CONDUCTIVITY OF BORON

(Impurity - 0.20% each, total impurities - 0.50%)

For Data Reported in Figure and Table No. 6

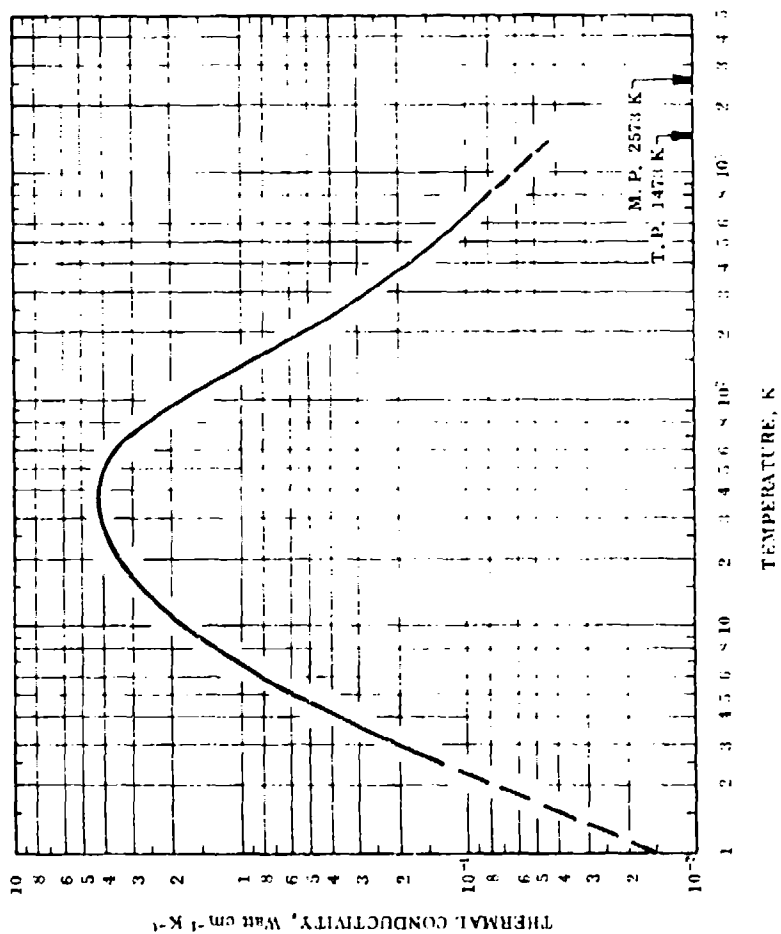
Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
1	790	L	1963	5-10			99.9 B (by difference), 0.1 C, cylindrical specimen 0.25 cm average diameter, 3.8 cm long made from single crystal of the beta-rhombohedral phase, provided by Texaco Experiment Inc., density $2.342 \pm 0.005 \text{ g cm}^{-3}$ ; electrical resistivity $> 5 \times 10^6 \text{ ohm cm}$ at room temperature; Debye temperature 1219 K.
2	790	L	1963	6-12			Rerun of the above specimen.
3	790	L	1963	10-100			Rerun of the above specimen.
4	790	L	1963	67-80			Rerun of the above specimen.
5	790	L	1963	162-290			Rerun of the above specimen.
6	790	L	1963	100-110			Rerun of the above specimen.
7	776	L	1965	2.8-291		R 4	Major impurities: $10 \times 10^{15} \text{ Si}$ , $20 \times 10^{15} \text{ Al}$ , $20 \times 10^{16} \text{ Mn}$ , $6 \times 10^{17} \text{ Ti}$ , and $4 \times 10^{18} \text{ Cu}$ atoms $\text{cm}^{-3}$ ; also about 0.1% (by volume) of precipitated particles 5-50 $\mu$ in diameter (probably of boron nitride, silicon inclusions or small voids); polycrystalline with numerous columnar crystals of $\beta$ -rhombohedral phase 1 cm long 0.3 cm average diameter; specimen 3.8 cm long 0.7 cm average diameter grown by partially purified boron by General Electric Research Lab.; density $2.33 \text{ g cm}^{-3}$ .
8	776	L	1965	3.1-305		R 16	A5 above but composed of columns 2 cm long 0.1 cm average diameter; specimen 2.6 cm long, 0.6 cm average diameter; provided by Eagle-Picher Research Lab., Miami, Okla. (crystal reference No. M6005CP); grown from the melt by floating zone process.
9	335		1965	300-650			No details reported.
10	1009	R	1959	323			99.18 and 0.02 total of Ca, Cu, Fe, Mg, and Si; polycrystalline specimen 1 mm in diameter and several cm long with a 0.025 mm tungsten filament at the center amounting to about 0.7% by weight; prepared by the reduction of boron tribromide by hydrogen near the tungsten filament at about 1250 C; data reported as the average for the range 20 to 30 C.

DATA TABLE NO. 6 THERMAL CONDUCTIVITY OF BORON  
 (Impurity < 0.20% each, total impurities < 0.50%)  
 (Temperature, T, K, Thermal Conductivity, k, Watt cm<sup>-1</sup>K<sup>-1</sup>)

T	k	T	k	T	k
<u>CURVE 1</u>		<u>CURVE 5 (cont.)</u>		<u>CURVE 8 (cont.)</u>	
5.00	0.148	156.6	0.745*	84.0	2.05
5.97	0.1793	197.1	0.763	89.0	1.85
7.50	0.320	242.0	0.648	135.0	0.78
8.80	0.405	285.4	0.630	220.0	0.38
10.09	0.501	290.0	0.608	305.0	0.265
<u>CURVE 2</u>		<u>CURVE 6</u>		<u>CURVE 9</u>	
8.31	0.357	100.6	1.595	300	0.25
10.58	0.570	116.2	1.304	370	0.29
12.21	0.773	126.9	1.095	410	0.28
<u>CURVE 3</u>		<u>CURVE 7</u>		<u>CURVE 10*</u>	
10.61	0.502	2.8	0.17	323	0.0125
11.44	0.613	4.8	0.58		
12.60	0.712	6.0	0.89		
13.82	0.834	9.4	1.57		
15.07	0.936	14.2	2.85		
17.23	1.115	18.0	3.21		
19.57	1.335	23.5	3.80		
23.95	1.674	28.5	4.15		
26.88	1.893	32.0	4.25		
30.75	2.205	50.0	4.20		
38.06	2.595	63.0	3.50		
46.92	2.688	83.0	2.52		
57.53	2.546	125.0	1.35		
79.23	2.515	170.0	0.715		
80.86	2.476	291.0	0.29		
86.86	2.157				
99.74	1.612				
<u>CURVE 4</u>		<u>CURVE 8</u>			
66.70	2.86	3.1	0.195		
70.16	2.80	5.3	0.700		
80.54	2.219	8.6	1.450		
<u>CURVE 5</u>					
161.8	1.019	15.5	2.65		
194.4	0.747	21.5	3.25		
		24.5	3.30		
		25.5	3.60		
		30.0	3.70		

\* Not shown on plot

FIGURE AND TABLE NO. 6R RECOMMENDED THERMAL CONDUCTIVITY OF BORON



RECOMMENDED VALUES\*  
(For Polycrystalline)

T <sub>1</sub>	k <sub>1</sub>	k <sub>2</sub>	T <sub>2</sub>	T <sub>1</sub>	k <sub>1</sub>	k <sub>2</sub>	T <sub>2</sub>
0	0	0	-459.7	500	0.141	8.15	440.3
1	(0.0150)*	(0.467)	-457.9	600	0.113	6.53	620.3
2	(0.0781)	(4.51)	-456.1	700	(0.0941)	(3.44)	800.3
3	0.198	11.4	-454.3	800	(0.0809)	(4.67)	980.3
4	0.375	21.7	-452.5	900	(0.0708)	(4.09)	1160
5	0.588	34.0	-450.7	1000	(0.0629)	(3.63)	1340
6	0.826	47.7	-448.9	1100	(0.0569)	(3.29)	1520
7	1.07	61.8	-447.1	1200	(0.0518)	(2.99)	1700
8	1.31	75.7	-445.3	1300	(0.0472)	(2.73)	1880
9	1.54	89.0	-443.5	1400	(0.0437)	(2.52)	2060
10	1.77	102	-441.7				
11	1.98	114	-439.9				
12	2.19	127	-438.1				
13	2.39	138	-436.3				
14	2.58	149	-434.5				
15	2.76	159	-432.7				
16	2.93	169	-430.9				
18	3.22	186	-427.3				
20	3.46	200	-423.7				
25	3.92	226	-414.7				
30	4.21	243	-405.7				
35	4.30	248	-396.7				
40	4.24	247	-387.7				
45	4.19	242	-378.7				
50	4.04	233	-369.7				
60	3.63	210	-351.7				
70	3.10	179	-333.7				
80	2.63	152	-315.7				
90	2.34	129	-297.7				
100	1.96	110	-279.7				
150	0.910	52.6	-189.7				
200	0.525	30.3	-99.7				
250	0.363	21.0	9.7				
273.2	0.317	18.3	32.0				
300	0.276	15.9	80.0				
350	0.224	12.9	170.3				
400	0.187	10.8	260.3				

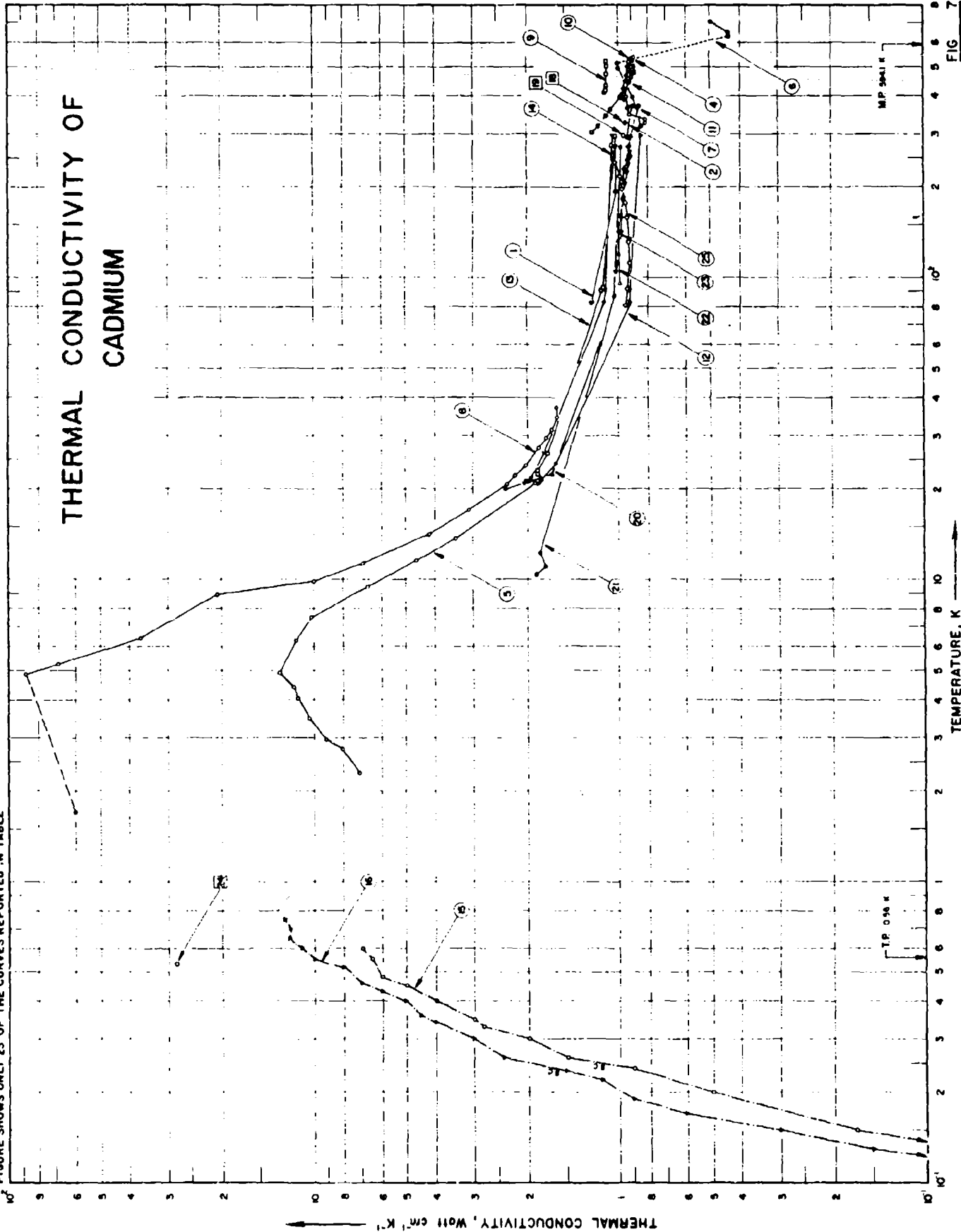
REMARKS

The recommended values are for high-purity boron. The values that are supported by experimental thermal conductivity data are thought to be accurate to within 5% of the true values near room temperature and 5 to 10% at other temperatures above 30 K. The thermal conductivity near and below the corresponding temperature of its maximum is highly sensitive to small physical and chemical variations of the specimens, and the values below 80 K are intended as typical values for indicating the general trend.

\* T<sub>1</sub> in K, k<sub>1</sub> in Watt cm<sup>-1</sup> K<sup>-1</sup>, T<sub>2</sub> in F, and k<sub>2</sub> in Btu hr<sup>-1</sup> ft<sup>-1</sup> F<sup>-1</sup>.

\* Values in parentheses are extrapolated.

FIGURE SHOWS ONLY 23 OF THE CURVES REPORTED IN TABLE



## SPECIFICATION TABLE NO. 7 THERMAL CONDUCTIVITY OF CADMIUM

(Impurity &lt; 0.20% each; total impurities &lt; 0.50%)

[ For Data Reported in Figure and Table No. 7 ]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition ( weight percent), Specifications and Remarks
1	35	L	1912	83-373			Specimen 2-3 cm in dia; electrical conductivity 5.05, 1.835, and $1.289 \times 10^4 \text{ ohm}^{-1}\text{cm}^{-1}$ at -190, -79, and 0 C, respectively.
2	77	E	1900	291.373			< 0.05% <del>total impurities</del> density $8.63 \text{ g cm}^{-3}$ at 16 C; electrical conductivity 13.13 and $9.89 \times 10^4 \text{ ohm}^{-1}\text{cm}^{-1}$ at 18 and 100 C, respectively.
3	77	E	1900	291.373			Similar to the above specimen but drawn into a wire; electrical conductivity 13.25 and $10.18 \times 10^4 \text{ ohm}^{-1}\text{cm}^{-1}$ at 18 and 100 C, respectively.
4	6	L	1931	327-540			Specimen prepared from "pure redistilled Cadmium"; density $8.64 \text{ g cm}^{-3}$ at 21 C; same specimen as used by Lees (Curves 22 and 23).
5	97	L	1952	2-3-21	2-3	Cd 1	99.9999 pure; polycrystalline; cast in glass.
6	19	L	1923	318-708			Specimen 1.5 cm in dia and 12 cm long; melting point 320 C.
7	796	L	1881	273.373			Density $8.62 \text{ g cm}^{-3}$ , electrical conductivity 14.41 and $10.18 \times 10^4 \text{ ohm}^{-1}\text{cm}^{-1}$ at 0 and 100 C, respectively. (The paper reported 14.41 and $10.18 \times 10^4 \text{ ohm}^{-1}\text{cm}^{-1}$ , obviously, a typographical error.)
8	122	L	1955	1-7-37	3	Cd 2	99.995 pure; single crystal; with heat flow at 75° to the hexagonal axis.
9	431	E	1944	414-526			Single crystal; electrical resistivity 10.08, 10.33, 10.90, 12.23, 13.20, and 14.10 $\mu\text{ohm cm}$ at 140.6, 146.9, 162.4, 202.0, 228.6, and 252.4 C, respectively.
10	431	E	1944	393-536			Polycrystal; electrical resistivity 11.84, 13.22, 14.34, 15.22, 16.65, 17.59, and $18.30 \mu\text{ohm cm}$ at 119.6, 152.8, 177.2, 196.9, 228.2, 248.8, and 262.6 C, respectively.
11	383	E	1956	306-506			Pure; electrical conductivity at 12.89, 11.11, 9.51, 8.38, 7.60, and $7.32 \times 10^4 \text{ ohm}^{-1}\text{cm}^{-1}$ at 32.5, 72.2, 122.2, 174.3, 207.5, and 232.6 C, respectively.
12	294	L	1932	21-297		Cd 53	Single crystal; specimen $0.1475 \text{ cm}^2$ in cross-sectional area and 6.70 cm long; angle between rod axis and hexagonal axis $\theta = 14^\circ$ ; electrical resistivity at 0.185, 2.001, 7.65, and $8.27 \mu\text{ohm cm}$ at -252, -190, 0, and 20 C, respectively.
13	294	L	1932	22-295		Cd 47a	Single crystal; specimen $0.1009 \text{ cm}^2$ in cross-sectional area and 4.48 cm long; $\theta = 84^\circ$ ; electrical resistivity 0.1352, 1.63, 6.36, and $6.89 \mu\text{ohm cm}$ at -252, -190, 0, and 20 C, respectively.
14	294	L	1932	21-297		Cd 47b	Similar to the above specimen except $0.0914 \text{ cm}^2$ in cross-sectional area and 6.65 cm long.
15	727	L	1960	0.10-0.60		Cd 1	Single crystal; heat flow along the hexagonal axis; includes superconducting state.
16	727	L	1960	0.10-0.75		Cd 3	Single crystal; heat flow perpendicular to the hexagonal axis; includes superconducting state.
17	230	L	1925	336.2			Impurities < 0.03; specimen in rod form $0.3 \text{ cm}^2$ in cross-sectional area and 5 to 6 cm long; electrical conductivity $13.76 \times 10^4 \text{ ohm}^{-1}\text{cm}^{-1}$ at 23 C.
18	230	L	1925	326.2			Similar to the above specimen except 1.9 cm in dia and 10 cm long.
19	511	L	1918	296.9			Specimen 1.1 cm in dia; supplied by Erba; measured in atmospheric pressure.

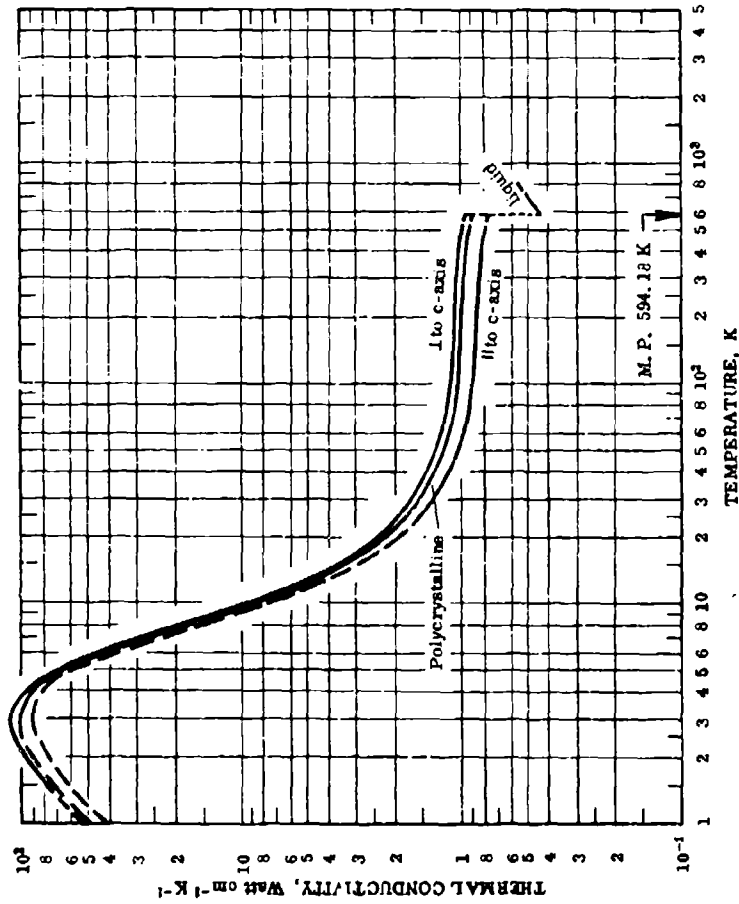
SPECIFICATION TABLE NO. 7 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
20	619	L	1916	20-273			Purified; specimen ~0.5 cm in dia and 5 cm long; electrical conductivity $622.0, 52.5,$ and $14.5 \times 10^4 \text{ ohm}^{-1} \text{cm}^{-1}$ at 20.4, 87.0, and 273 K, respectively.
21	619	L	1916	20-273			Chemically pure (Kahlbaum); specimen ~0.5 cm in dia and 5 cm long; electrical conductivity $53.58,$ and $14.6 \times 10^4 \text{ ohm}^{-1} \text{cm}^{-1}$ at 20.4, 87.0, and 273 K, respectively.
22	88	L	1908	96-297			Turned from a cast stick of "pure Redistilled Cadmium" as used in Cadmium-Cell; specimen 0.585 cm in dia and 7-8 cm long; density $8.64 \text{ g cm}^{-3}$ at 21 C; electrical resistivity 2.22, 2.56, 4.18, 5.05, 5.46, 6.36, 6.96, and $7.78 \mu\text{ohm cm}$ at -178.1, -165.9, -105.8, -75.1, -59.9, -25.2, -5.7, and 22.8 C, respectively; first experiment.
23	88	L	1908	105-295			The above specimen, second experiment.
24	727	L	1960	0.53		Cd 2	Single crystal; heat flow perpendicular to the hexagonal axis, at the transition point.
25	851	L	1960	82-276	10		99.95 pure; specimen 0.1877 in, dia x 2.255 in, long turned from cast stick obtained from A. D. Mackay; data corrected for rise in temperature during measurement.





FIGURE AND TABLE NO. 7R RECOMMENDED THERMAL CONDUCTIVITY OF CADMIUM



REMARKS

The recommended values are for well-annealed 99.9997% pure cadmium with residual electrical resistivity  $\rho_r = 0.000463$ ,  $0.000606$ , and  $0.000502 \mu\Omega \text{ cm}$ , respectively, for single crystal along directions perpendicular and parallel to the c-axis and for polycrystalline cadmium (characterisation by  $\rho_r$  becomes important at temperatures below about 100 K). The values below 1.5  $T_m$  are calculated to fit the experimental data by using  $n = 2.50$ ,  $\alpha' = 1.77 \times 10^{-4}$ , and  $\beta = 0.0188$  for the direction perpendicular to the c-axis; using  $n = 2.50$ ,  $\alpha' = 1.90 \times 10^{-4}$ , and  $\beta = 0.0204$  for the direction parallel to the c-axis; and using  $n = 2.50$ ,  $\alpha' = 1.80 \times 10^{-4}$ , and  $\beta = 0.0204$  for polycrystalline cadmium. The recommended values that are supported by experimental thermal conductivity data are thought to be accurate to within 4% of the true values near room temperature and 4 to 10% at other temperatures.

\*  $T_1$  in K,  $k_1$  in Watt  $\text{cm}^{-1} \text{K}^{-1}$ ,  $T_2$  in  $^{\circ}\text{F}$ , and  $k_2$  in  $\text{Btu hr}^{-1} \text{ft}^{-1} \text{F}^{-1}$ .

† Values in parentheses are extrapolated or estimated.

RECOMMENDED VALUES\*

$T_1$	Single Crystal (//to c-axis)		Polycrystalline	
	$k_1$	$k_2$	$k_1$	$k_2$
0	0	0	0	0
1	(52.8) † (3050)	(2330)	(48.7)	(2910)
2	96.3	(5360)	(89.3)	(5160)
3	111	(6410)	104	6010
4	96.6	(4730)	92.0	5320
5	71.9	(3540)	69.0	3990
6	47.3	(2180)	44.2	2550
8	19.3	(1120)	18.0	1040
10	9.5	(549)	8.87	513
15	3.8	(3.04)	3.55	205
20	2.42	(1.94)	2.26	131
25	1.92	1.54	1.79	103
30	1.67	96.5	1.56	90.1
35	1.51	87.2	1.41	81.5
40	1.41	81.5	1.32	76.3
50	1.28	74.0	1.20	69.3
60	1.21	69.9	1.13	65.3
70	1.16	67.0	1.08	62.4
80	1.13	65.3	1.06	61.2
90	1.11	64.1	1.04	60.1
100	1.10	63.6	1.03	59.5
150	1.06	62.4	1.01	58.4
200	1.06	61.2	0.992	57.4
250	1.05	60.7	0.980	56.6
273.2	1.04	60.1	0.975	56.3
300	1.04	60.1	0.968	55.9
350	1.03	59.5	0.958	55.4
400	1.01	58.4	0.947	54.7
500	0.985	56.9	0.920	53.2
594.18	(0.942)	(54.4)	(0.754)	(50.8)

In Liquid State

$T_1$	$k_1$	$k_2$	$T_2$
594.18	0.416	24.0	609.83
600	0.420	24.3	620.3
700	0.490	28.3	800.3
800	(0.559)	(32.3)	960.3

# THERMAL CONDUCTIVITY OF CERIUM

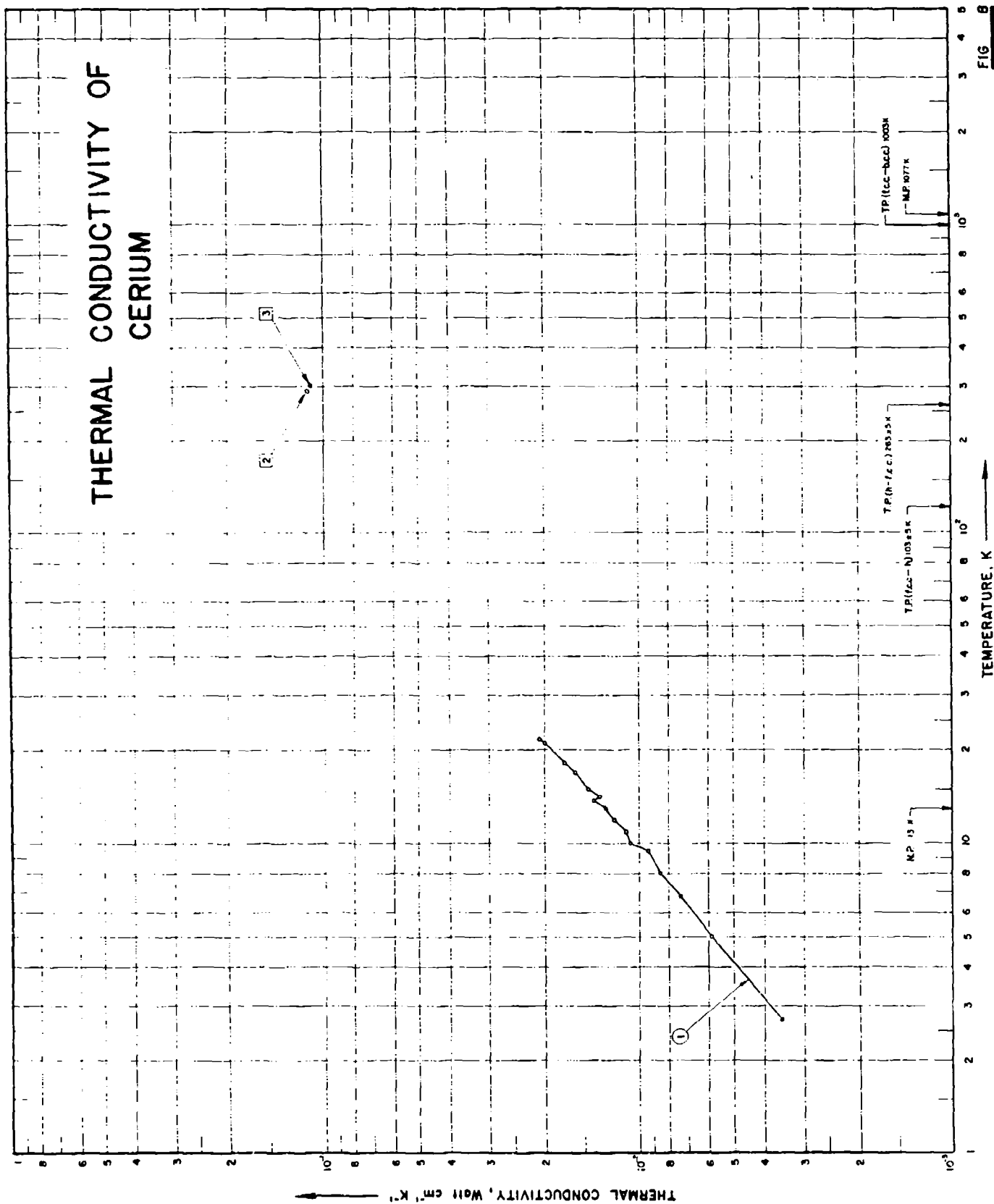


FIG. 8

## SPECIFICATION TABLE NO. 8 THERMAL CONDUCTIVITY OF CERIUM

(Impurity &lt; 0.20% each; total impurities &lt; 0.50%)

(For Data Reported in Figure and Table No. 8 )

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition ( weight percent), Specifications and Remarks
1	122	L	1955	2.7-22	3.0	Ce - 1	99.6 pure; Mg and Ca as major impurities; specimen 1.045 cm long and 0.38 cm square cross section; electrical resistivity ratio $\rho_{293K}/\rho_{200K} = 1.93$ .
2	777	C	1965	291.2	+3		High purity rod of cerium, about 0.25 in. dia and 0.25 in. long obtained from Johnson Matthey and Co., Ltd.; electrical resistivity 74 $\mu\text{ohm cm}$ at $-118^\circ\text{C}$ ; monel metal used as comparative material; measurements made using 2 different comparators.
3	811		1954	301.2	10		No details reported.

## DATA TABLE NO. 8 THERMAL CONDUCTIVITY OF CERIUM

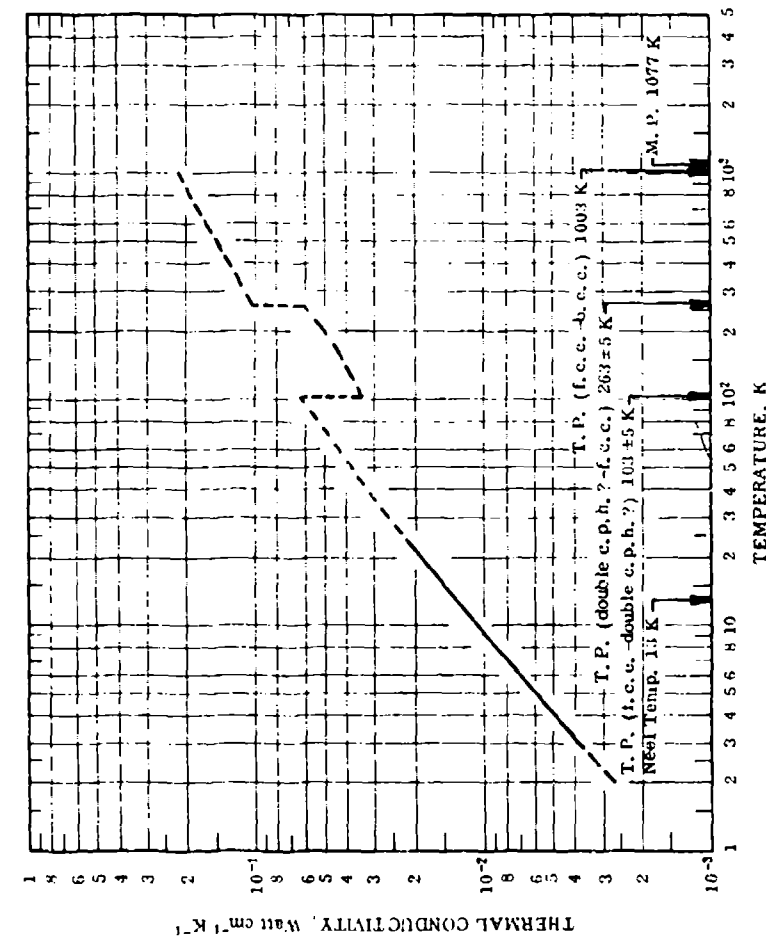
(Impurity &lt; 0.20% each; total impurities &lt; 0.50%)

T, K; Thermal Conductivity,  $\kappa$ , Watt  $\text{cm}^{-1}\text{K}^{-1}$ 

T	$\kappa$
<u>CURVE 1</u>	
2.71	0.00354
5.00	0.00589
6.77	0.00734
8.04	0.00852
9.48	0.00939
16.0	0.0107
10.9	0.0111
11.9	0.0120
13.0	0.0127
13.7	0.0139
14.2	0.0133
15.0	0.0145
17.0	0.0160
18.3	0.0173
21.2	0.0209
21.7	0.0208
<u>CURVE 2</u>	
291.2	0.112
291.2	0.112*
<u>CURVE 3</u>	
301.2	0.109

\* Not shown on plot

FIGURE AND TABLE NO. 8R RECOMMENDED THERMAL CONDUCTIVITY OF CERIUM



RECOMMENDED VALUES (For Polycrystalline)							
T <sub>1</sub>	k <sub>1</sub>	k <sub>2</sub>	T <sub>2</sub>	T <sub>1</sub>	k <sub>1</sub>	k <sub>2</sub>	T <sub>2</sub>
0	0	0	-459.7	300	0.114	6.59	80.3
2	(0.00260) <sup>‡</sup>	(0.150)	-456.1	350	(0.124)	(7.16)	170.3
3	0.00373	0.216	-454.3	400	(0.133)	(7.68)	260.3
4	0.00482	0.278	-452.5	500	(0.150)	(8.67)	440.3
5	0.00584	0.337	-450.7	600	(0.166)	(9.59)	620.3
6	0.00683	0.395	-448.9	700	(0.180)	(10.4)	800.3
7	0.00776	0.448	-447.1	800	(0.193)	(11.2)	980.3
8	0.00868	0.502	-445.3	900	(0.206)	(11.9)	1160
9	0.00959	0.554	-443.5	1000	(0.218)	(12.6)	1340
10	0.0105	0.607	-441.7				
11	0.0113	0.653	-439.9				
12	0.0121	0.705	-438.1				
13	0.0130	0.751	-436.3				
14	0.0138	0.797	-434.5				
15	0.0147	0.849	-432.7				
16	0.0155	0.896	-430.9				
18	0.0171	0.988	-427.3				
20	0.0186	1.07	-423.7				
25	(0.0224)	(1.29)	-414.7				
30	(0.0260)	(1.50)	-405.7				
35	(0.0293)	(1.69)	-396.7				
40	(0.0323)	(1.87)	-387.7				
45	(0.0352)	(2.03)	-378.7				
50	(0.0379)	(2.19)	-369.7				
60	(0.0432)	(2.50)	-351.7				
70	(0.0478)	(2.76)	-333.7				
80	(0.0521)	(3.01)	-315.7				
90	(0.0561)	(3.24)	-297.7				
100	(0.0609)	(3.47)	-279.7				
103±5	(0.0610)	(3.52)	-274.3±9				
103±5	(0.0340)	(1.96)	-274.3±9				
150	(0.0406)	(2.35)	-199.7				
200	(0.0500)	(2.89)	-99.7				
250	(0.0583)	(3.37)	-				
263±5	(0.0602)	(3.48)	13.7±9				
263±5	(0.105)	(6.07)	13.7±9				
273.2	(0.108)	(6.24)	32.0				

REMARKS

The recommended values are for well-annealed 99.6% pure cerium. The values that are supported by experimental thermal conductivity data are thought to be accurate to within 5% of the true values near room temperature and 5 to 15% at other temperatures. Since the thermal conductivity at low temperature is highly sensitive to small physical and chemical variations of the specimens, the values below about 100 K would be higher for specimens of higher purity.

<sup>‡</sup> T<sub>1</sub> in K, k<sub>1</sub> in Watt cm<sup>-1</sup> K<sup>-1</sup>, T<sub>2</sub> in F, and k<sub>2</sub> in Btu hr<sup>-1</sup> ft<sup>-1</sup> F<sup>-1</sup>.

<sup>†</sup> Values in parentheses are extrapolated or estimated.

FIGURE SHOWS ONLY 7 OF THE CURVES REPORTED IN TABLE

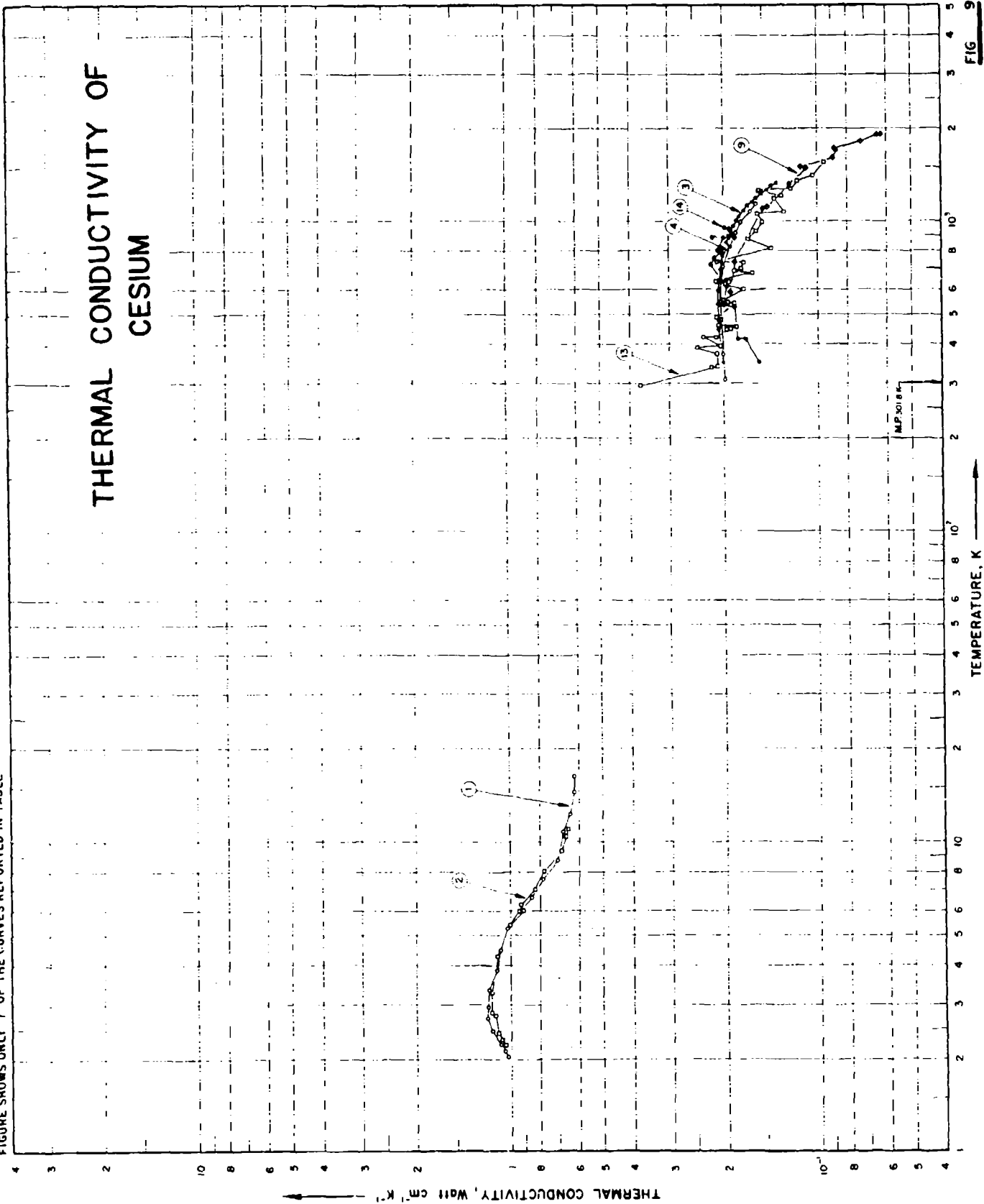


FIG 9



SPECIFICATION TABLE NO. 9 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
10	872, 874	-	1965	589-1910	1.3	Run 2	Similar to the above; second loading of the test capsule.
11	873, 874	-	1965	589-1914	1.5	Run 3	Similar to the above; third loading of the test capsule.
12	875		1962	45-1157	5		Vapor specimen.
13	287	C	1964	35-1556	5		99.994 pure (estimated from freezing point curve and emission spectrography for impurities); freezing point 28.52 C; specimen clad in Nb-1Zr alloy; specimen in liquid state except at 295.2 K where it was mostly solid; electrical resistivity reported as 44, 55, 67, 80, 96, 114, 134, 155, 179, 208, and 246 $\mu\text{ohm cm}$ at 100, 200, 300, 400, 500, 600, 700, 800, 900, 1000, and 1100 C, respectively; Nb-1Zr alloy used as comparative material.
14	987, 988	C	1967	350-950	$\pm 7 \pm 14$		0.05 Na, 0.033 Rb, and 0.0133 K; liquid specimen contained in a hollow cylinder of I.D. 14 mm; prepared from cesium chloride by reduction with calcium and subsequent distillation in a vacuum of $10^{-1}-10^{-3}$ mm Hg; Armco iron used as comparative material.
15	997, 998	C	1967	358-969	$\pm 7 \pm 14$		Similar to the above specimen.
16	997, 998	C	1967	428-904	$\pm 7 \pm 14$		Similar to the above specimen.
17	987, 988	C	1967	460-904	$\pm 7 \pm 14$		Similar to the above specimen.



DATA TABLE NO. 9 THERMAL CONDUCTIVITY OF CESIUM

(Impurity < 0.20% each; total impurities < 0.50%)

[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup> K<sup>-1</sup>]

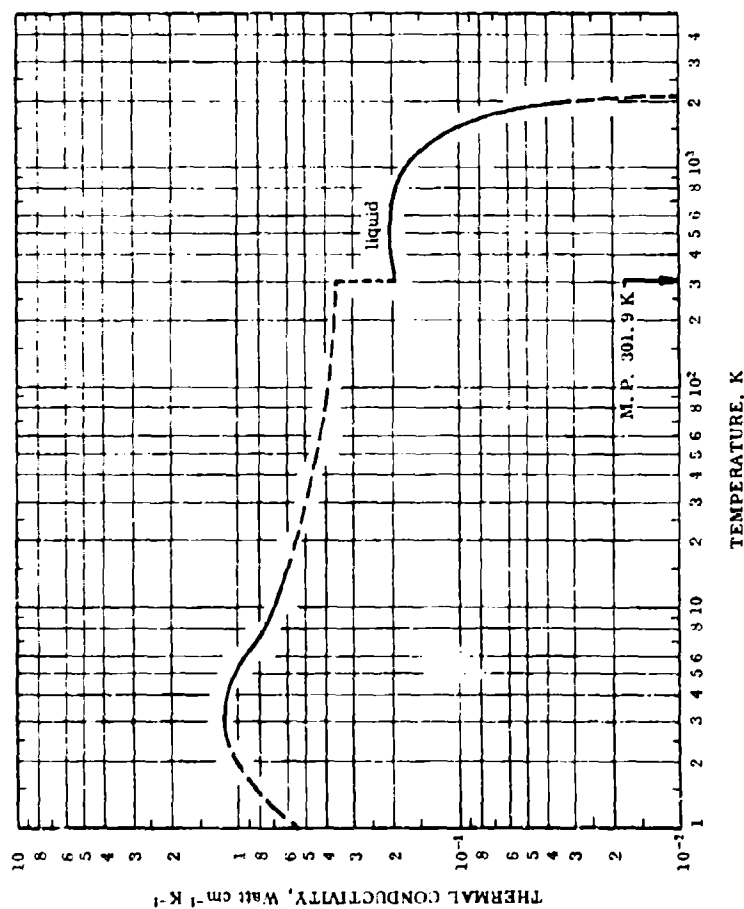
T	k	T	k	T	k	T	k	T	k	T	k	T	k	T	k	T	k	T	k
<b>CURVE 1</b>																			
2.02	1.09	3.16	1.040	4.28	1.066	5.43	1.050	6.57	1.036	7.71	1.022	8.85	1.008	9.99	0.994	11.13	0.980	12.27	0.966
2.19	1.05	2.28	1.066	3.42	1.052	4.56	1.038	5.70	1.024	6.84	1.010	7.98	0.996	9.12	0.982	10.26	0.968	11.40	0.954
2.22	1.08	2.39	1.066	3.53	1.046	4.67	1.032	5.81	1.018	6.95	1.004	8.09	0.990	9.23	0.976	10.37	0.962	11.51	0.948
2.46	1.15	2.53	1.086	3.67	1.066	4.81	1.052	5.95	1.040	7.09	1.026	8.23	1.012	9.37	0.998	10.51	0.984	11.65	0.970
2.70	1.19	2.74	1.086	3.88	1.086	5.02	1.072	6.16	1.058	7.30	1.044	8.44	1.030	9.58	1.016	10.72	1.002	11.86	0.988
2.94	1.18	3.24	1.056	4.38	1.056	5.52	1.032	6.66	1.016	7.80	1.000	8.94	0.984	10.08	0.970	11.22	0.956	12.36	0.942
3.54	1.17	3.86	1.109	4.98	1.109	6.12	1.086	7.26	1.072	8.40	1.058	9.54	1.044	10.68	1.030	11.82	1.016	12.96	1.002
3.85	1.11	4.28	1.102	5.37	1.102	6.51	1.086	7.65	1.072	8.79	1.058	9.93	1.044	11.07	1.030	12.21	1.016	13.35	1.002
4.49	1.08	5.10	1.010	6.12	1.010	7.26	0.996	8.40	0.982	9.54	0.968	10.68	0.954	11.82	0.940	12.96	0.926	14.10	0.912
5.16	1.03	6.10	0.924	7.00	0.924	8.14	0.908	9.28	0.894	10.42	0.880	11.56	0.866	12.70	0.852	13.84	0.838	15.12	0.824
5.99	0.943	7.00	0.828	8.14	0.828	9.28	0.814	10.42	0.800	11.56	0.786	12.70	0.772	13.84	0.758	15.12	0.744	16.40	0.730
6.62	0.856	8.14	0.766	9.28	0.766	10.42	0.752	11.56	0.738	12.70	0.724	13.84	0.710	15.12	0.696	16.40	0.682	17.76	0.668
7.59	0.782	9.28	0.684	10.42	0.684	11.56	0.670	12.70	0.656	13.84	0.642	15.12	0.628	16.40	0.614	17.76	0.600	19.12	0.586
8.76	0.703	10.42	0.624	11.56	0.624	12.70	0.610	13.84	0.596	15.12	0.582	16.40	0.568	17.76	0.554	19.12	0.540	20.52	0.526
10.8	0.676	12.70	0.570	13.84	0.570	15.12	0.556	16.40	0.542	17.76	0.528	19.12	0.514	20.52	0.500	21.88	0.486	23.24	0.472
12.1	0.641	13.84	0.526	15.12	0.526	16.40	0.512	17.76	0.498	19.12	0.484	20.52	0.470	21.88	0.456	23.24	0.442	24.60	0.428
14.5	0.624	15.12	0.480	16.40	0.480	17.76	0.466	19.12	0.452	20.52	0.438	21.88	0.424	23.24	0.410	24.60	0.396	26.04	0.382
16.3	0.624	16.40	0.624	17.76	0.624	19.12	0.610	20.52	0.596	21.88	0.582	23.24	0.568	24.60	0.554	26.04	0.540	27.40	0.526
8.76	0.703	10.42	0.703	11.56	0.703	12.70	0.689	13.84	0.675	15.12	0.661	16.40	0.647	17.76	0.633	19.12	0.619	20.52	0.605
10.8	0.676	12.70	0.676	13.84	0.676	15.12	0.662	16.40	0.648	17.76	0.634	19.12	0.620	20.52	0.606	21.88	0.592	23.24	0.578
12.1	0.641	13.84	0.641	15.12	0.641	16.40	0.627	17.76	0.613	19.12	0.599	20.52	0.585	21.88	0.571	23.24	0.557	24.60	0.543
14.5	0.624	15.12	0.624	16.40	0.624	17.76	0.610	19.12	0.596	20.52	0.582	21.88	0.568	23.24	0.554	24.60	0.540	26.04	0.526
16.3	0.624	16.40	0.624	17.76	0.624	19.12	0.610	20.52	0.596	21.88	0.582	23.24	0.568	24.60	0.554	26.04	0.540	27.40	0.526
<b>CURVE 2</b>																			
2.21	1.04	3.35	1.020	4.50	1.006	5.69	0.992	6.88	0.984	8.07	0.976	9.26	0.968	10.45	0.960	11.64	0.952	12.83	0.944
2.30	1.07	3.46	1.036	4.61	1.022	5.80	1.008	6.99	0.994	8.18	0.986	9.37	0.978	10.56	0.970	11.75	0.962	12.94	0.954
2.43	1.10	3.57	1.052	4.72	1.038	5.91	1.024	7.10	1.014	8.29	1.000	9.48	0.992	10.67	0.984	11.86	0.976	13.05	0.968
2.76	1.12	3.68	1.068	4.83	1.054	6.02	1.040	7.21	1.030	8.40	1.016	9.59	1.002	10.78	0.994	11.97	0.986	13.16	0.978
2.83	1.15	3.79	1.084	4.94	1.070	6.13	1.060	7.32	1.050	8.51	1.036	9.70	1.022	10.89	1.008	12.08	0.994	13.27	0.984
3.28	1.15	3.90	1.096	5.05	1.086	6.24	1.076	7.43	1.066	8.62	1.052	9.81	1.038	11.00	1.024	12.19	1.010	13.38	0.996
4.29	1.10	4.01	1.040	5.16	1.042	6.35	1.022	7.54	1.012	8.73	1.000	9.92	0.986	11.11	0.972	12.30	0.958	13.49	0.944
5.40	1.01	4.12	0.992	5.27	0.998	6.46	0.978	7.65	0.964	8.84	0.950	10.03	0.936	11.22	0.922	12.41	0.908	13.60	0.894
6.00	0.913	4.23	0.944	5.38	0.950	6.57	0.930	7.76	0.916	8.95	0.902	10.14	0.888	11.33	0.874	12.52	0.860	13.71	0.846
6.28	0.928	4.34	0.960	5.49	0.966	6.68	0.946	7.87	0.932	9.06	0.918	10.25	0.904	11.44	0.890	12.63	0.876	13.82	0.862
7.03	0.834	4.45	0.912	5.60	0.918	6.79	0.898	7.98	0.884	9.17	0.870	10.36	0.856	11.55	0.842	12.74	0.828	13.93	0.814
8.04	0.778	4.56	0.864	5.71	0.870	6.90	0.850	8.09	0.836	9.28	0.822	10.47	0.808	11.66	0.794	12.85	0.780	14.04	0.766
9.39	0.683	4.67	0.816	5.82	0.822	7.01	0.802	8.20	0.788	9.39	0.774	10.58	0.760	11.77	0.746	12.96	0.732	14.15	0.718
10.4	0.665	4.78	0.768	5.93	0.774	7.12	0.754	8.31	0.740	9.50	0.726	10.69	0.712	11.88	0.698	13.07	0.684	14.26	0.670
11.0	0.651	4.89	0.720	6.04	0.726	7.23	0.706	8.42	0.692	9.61	0.678	10.80	0.664	12.00	0.650	13.19	0.636	14.37	0.622
<b>CURVE 3</b>																			
349.2	0.201	350.2	0.201	351.2	0.201	352.2	0.201	353.2	0.201	354.2	0.201	355.2	0.201	356.2	0.201	357.2	0.201	358.2	0.201
349.2	0.208	350.2	0.208	351.2	0.208	352.2	0.208	353.2	0.208	354.2	0.208	355.2	0.208	356.2	0.208	357.2	0.208	358.2	0.208
349.2	0.207	350.2	0.207	351.2	0.207	352.2	0.207	353.2	0.207	354.2	0.207	355.2	0.207	356.2	0.207	357.2	0.207	358.2	0.207
349.2	0.208	350.2	0.208	351.2	0.208	352.2	0.208	353.2	0.208	354.2	0.208	355.2	0.208	356.2	0.208	357.2	0.208	358.2	0.208
349.2	0.206	350.2	0.206	351.2	0.206	352.2	0.206	353.2	0.206	354.2	0.206	355.2	0.206	356.2	0.206	357.2	0.206	358.2	0.206
349.2	0.204	350.2	0.204	351.2	0.204	352.2	0.204	353.2	0.204	354.2	0.204	355.2	0.204	356.2	0.204	357.2	0.204	358.2	0.204
349.2	0.203	350.2	0.203	351.2	0.203	352.2	0.203	353.2	0.203	354.2	0.203	355.2	0.203	356.2	0.203	357.2	0.203	358.2	0.203
349.2	0.201	350.2	0.201	351.2	0.201	352.2	0.201	353.2	0.201	354.2	0.201	355.2	0.201	356.2	0.201	357.2	0.201	358.2	0.201
349.2	0.198	350.2	0.198	351.2	0.198	352.2	0.198	353.2	0.198	354.2	0.198	355.2	0.198	356.2	0.198	357.2	0.198	358.2	0.198
349.2	0.196	350.2	0.196	351.2	0.196	352.2	0.196	353.2	0.196	354.2	0.196	355.2	0.196	356.2	0.196	357.2	0.196	358.2	0.196
349.2	0.193	350.2	0.193	351.2	0.193	352.2	0.193	353.2	0.193	354.2	0.193	355.2	0.193	356.2	0.193	357.2	0.193	358.2	0.193
349.2	0.186	350.2	0.186	351.2	0.186	352.2	0.186	353.2	0.186	354.2	0.186	355.2	0.186	356.2	0.186	357.2	0.186	358.2	0.186
349.2	0.188	350.2	0.188	351.2	0.188	352.2	0.188	353.2	0.188	354.2	0.188	355.2	0.188	356.2	0.188	357.2	0.188	358.2	0.188
349.2	0.186	350.2	0.186	351.2	0.186	352.2	0.186	353.2	0.186	354.2	0.186	355.2	0.186	356.2	0.186	357.2	0.186	358.2	0.186
349.2	0.186	350.2	0.186	351.2	0.186	352.2	0.186	353.2	0.186	354.2	0.186	355.2	0.186	356.2	0.186	357.2	0.186	358.2	0.186
349.2	0.186	350.2	0.186	351.2	0.186	352.2	0.186	353.2	0.186	354.2	0.186	355.2	0.186	356.2	0.186	357.2	0.186	358.2	0.186
349.2	0.186	350.2	0.186	351.2	0.186	352.2	0.186	353.2	0.186	354.2	0.186	355.2	0.186	356.2	0.186	357.2	0.186	358.2	0.186
349.2	0.186	350.2	0.186	351.2	0.186	352.2	0.186	353.2	0.186	354.2	0.186	355.2	0.186	356.2	0.186	357.2	0.186	358.2	0.186
349.2	0.186	350.2	0.186	351.2	0.186	352.2	0.186	353.2	0.186	354.2	0.186	355.2	0.186	356.2	0.186	357.2	0.186	358.2	0.186
349.2	0.186	350.2	0.186	351.2	0.186	352.2	0.186	353.2	0.186	354.2	0.186	355.2	0.186	356.2	0.186	357.2	0.186	358.2	0.186
349.2	0.186	350.2	0.186	351.2	0.1														

DATA TABLE NO. 9 (continued)

T	k
<u>CURVE 16 (cont.)</u> *	
705.2	0.214
775.2	0.206
791.2	0.198
857.2	0.224
904.2	0.194
<u>CURVE 17</u> *	
460.2	0.189
468.2	0.185
470.2	0.184
471.2	0.182
477.2	0.196
482.2	0.203
551.2	0.169
556.2	0.193
631.2	0.202
649.2	0.199
650.2	0.207
705.2	0.199
775.2	0.189
776.2	0.188
828.2	0.185
845.2	0.172
904.2	0.159

\* Not shown on plot

FIGURE AND TABLE NO. 9R RECOMMENDED THERMAL CONDUCTIVITY OF CESIUM



RECOMMENDED VALUES\*

$T_1$	$k_1$	$k_2$	$T_2$	$T_1$	$k_1$	$k_2$	$T_2$
0	0	0	-459.7	301.9	0.157	In Liquid State	83.5
1	(0.525) <sup>‡</sup>	(30.3)	-457.9	350	0.200	11.4	170.3
2	(1.00)	(57.8)	-455.1	400	0.203	11.6	250.3
3	1.18	68.2	-454.3	500	0.205	11.8	440.3
4	1.13	65.3	-452.5	600	0.205	11.8	620.3
5	1.04	60.1	-450.7	700	0.201	11.6	800.3
6	0.935	54.0	-448.9	800	0.194	11.2	980.3
7	0.837	48.4	-447.1	900	0.185	10.7	1160
8	0.769	44.4	-445.3	1000	0.175	10.1	1340
9	0.720	41.6	-443.5	1100	0.163	9.42	1520
10	0.689	39.8	-441.7	1200	0.150	8.67	1700
11	0.666	38.5	-439.9	1300	0.136	7.86	1880
12	0.647	37.4	-438.1	1400	0.122	7.05	2060
13	0.630	36.4	-436.3	1500	0.108	6.24	2243
14	0.615	35.5	-434.5	1600	0.094	5.43	2420
15	0.600	34.7	-432.7	1700	0.080	4.62	2600
16	0.590	34.1	-430.9	1800	0.066	3.81	2780
18	(0.572)	(33.1)	-427.3	1900	0.051	2.95	2960
20	(0.554)	(32.0)	-423.7	2000	(0.029)	(1.68)	3140
25	(0.523)	(30.2)	-414.7	2060	(-0)	(-0)	3248
30	(0.500)	(28.9)	-405.7				
35	(0.483)	(27.9)	-396.7				
40	(0.470)	(27.2)	-387.7				
45	(0.457)	(26.4)	-378.7				
50	(0.447)	(25.8)	-369.7				
60	(0.430)	(24.6)	-351.7				
70	(0.420)	(24.3)	-333.7				
80	(0.410)	(23.7)	-315.7				
90	(0.402)	(23.2)	-297.7				
100	(0.397)	(22.9)	-279.7				
150	(0.378)	(21.8)	-189.7				
200	(0.368)	(21.3)	-99.7				
250	(0.363)	(21.0)	-9.7				
273.2	(0.361)	(20.9)	32.0				
300	0.359	20.7	80.3				
301.9	0.359	20.7	83.5				

REMARKS

The recommended values are for high-purity cesium with residual electrical resistivity  $\rho_0 = 0.0465 \mu\Omega$  cm (characterization by  $\rho_0$  becomes important at temperatures below about 50 K). The values that are supported by experimental thermal conductivity data are thought to be accurate to within 5% of the true values near room temperature and 5 to 10% at other temperatures.

\*  $T_1$  in K,  $k_1$  in Watt cm<sup>-1</sup> K<sup>-1</sup>,  $T_2$  in F, and  $k_2$  in Btu hr<sup>-1</sup> ft<sup>-1</sup> F<sup>-1</sup>.

† Values in parentheses are extrapolated, interpolated, or estimated.

# THERMAL CONDUCTIVITY OF CHROMIUM

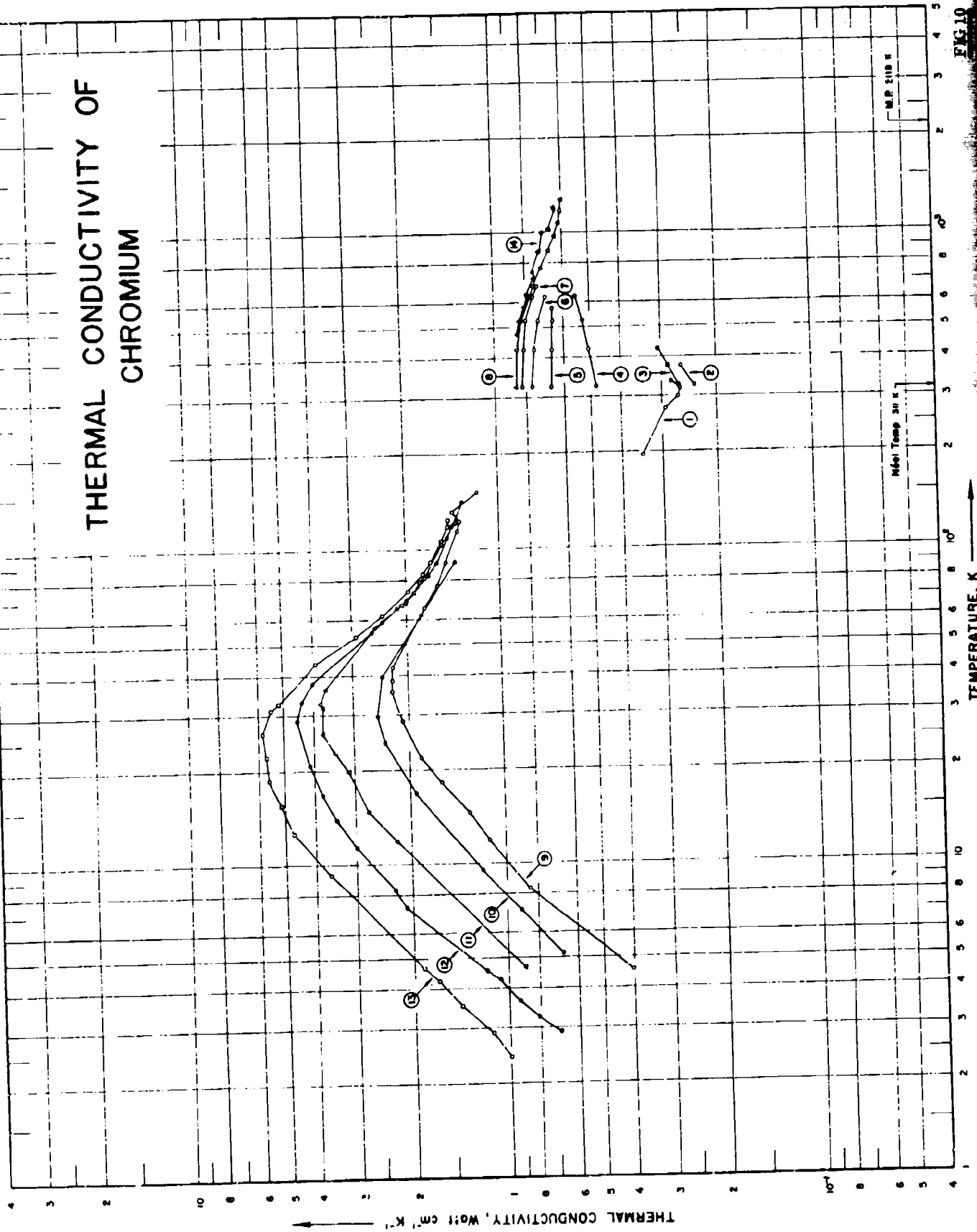


FIG 10

SPECIFICATION TABLE NO. 10 THERMAL CONDUCTIVITY OF CHROMIUM

(Impurity &lt; 0.20% each; total impurities &lt; 0.50%)

[ For Data Reported in Figure and Table No. 10 ]

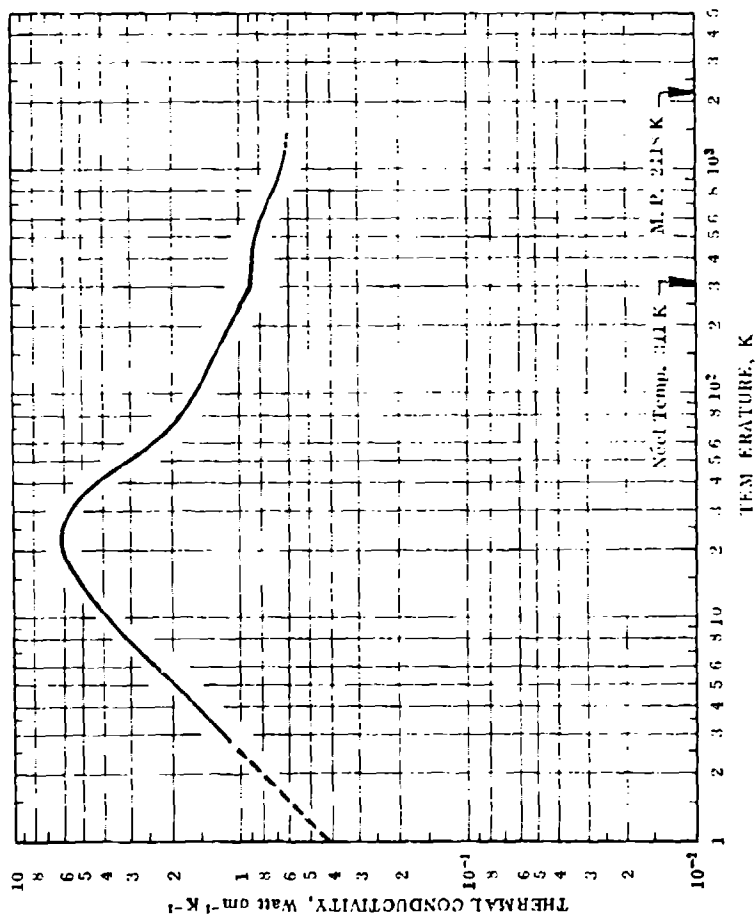
Curve No.	ReL No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition ( weight percent), Specifications and Remarks
1	136	L	1940	196-334		Cr II	Electrolytic; specimen 0.7 x 0.23 x 0.21 cm; annealed at 1000 C for 30 min.
2	112	L.C	1957	323-373			0.43 O; electrodeposited chromium tube, 1.28 cm O.D., 0.63 cm I.D., and 12.05 cm long; as deposited; density 6.975 g cm <sup>-3</sup> , electrical resistivity 40.4, 41.7, 44.0, 45.4, 47.2, 48.9, and 50.3 μohm cm at -144, -112, -31, -44, -8, 32, 73 C, respectively.
3	112	L.C	1957	323-423			The above specimen heat treated at 486 K; electrical resistivity 30.3, 39.7, 42.2, 44.6, and 48.2 μohm cm at -172, 26, 95, 148, and 223 C, respectively.
4	112	L.C	1957	323-623			The above specimen heat treated at 478 K; density increased to 7.08 g cm <sup>-3</sup> ; electrical resistivity 6.2, 25.5, 28.9, 33.2, 35.2, 37.8, 39.9, and 47.2 μohm cm at -173, 21, 92, 174, 215, 265, 313, and 399 C, respectively.
5	112	L.C	1957	323-573			The above specimen heat treated at 818 K; electrical resistivity 4.4, 6.9, 10.5, 13.9, 18.1, 19.9, 25.5, 30.1, 37.2, 44.3, 48.8, and 52.2 μohm cm at -176, -140, -103, -58, -1, 31, 161, 270, 331, 434, 503, and 551 C, respectively.
6	112	L.C	1957	323-623			The above specimen heat treated at 1133 K; electrical resistivity 16.1, 21.7, 32.2, 46.0, and 65.7 μohm cm at 22, 162, 355, 570, and 869 C, respectively.
7	112	L.C	1957	323-673			The above specimen heat treated at 1327 K; electrical resistivity 14.3, 16.2, 19.9, 27.0, 30.4, 34.9, 40.5, 46.8, 53.3, 59.5, 71.8, and 77.4 μohm cm at 0, 62, 147, 293, 350, 435, 528, 636, 730, 816, 994, and 1067 C, respectively.
8	112	L.C	1957	323-1273			The above specimen heat treated at 1683 K; density increased to 7.15 g cm <sup>-3</sup> ; electrical resistivity 1.9, 10.5, 13.8, 17.2, 23.7, 30.2, 35.3, 44.7, 55.2, 65.4, 76.6, 81.6, and 95.2 μohm cm at -179, -46, 26, 120, 282, 406, 505, 669, 841, 999, 1167, 1236, and 1427 C, respectively.
9	68	L	1957	4.5-123		1	99,998 pure; specimen 3 mm in dia and 8 cm long; supplied by the Aeronautical Res. Labs. of the Commonwealth Dept. of Supply; cold worked; residual resistivity 0.255 μohm cm.
10	68	L	1957	5.1-31		2	The above specimen annealed at 1010 C for 4 hrs; residual resistivity 0.181 μohm cm.
11	68	L	1957	4.6-151		3	99,998 pure; partially recrystallized; specimen 3 mm in dia and 8 cm long; supplied by the Aeronautical Res. Labs of the Commonwealth Dept. of Supply; residual resistivity 0.125 μohm cm.
12	68	L	1957	2.9-142		4	The above specimen annealed at 1060 C for 4 hrs; residual resistivity 0.090 μohm cm.
13	68	L	1957	2.4-123		5	99,998 pure; fully recrystallized; specimen 3 mm in dia and 8 cm long; supplied by the Aeronautical Res. Labs of the Commonwealth Dept. of Supply; residual resistivity 0.055 μohm cm ρ(273 K)/ρ(0 K) = 217.
14	69	C	1956	470-1201			Chemically pure; ductile; supplied by the Bureau of Mines, Oregon; density 7.16 g cm <sup>-3</sup> at 24 C.

DATA TABLE NO. 10 THERMAL CONDUCTIVITY OF CHROMIUM

(Impurity  $\leq 0.20\%$  each, total impurities  $\leq 0.50\%$ )Temperature: T, K. Thermal Conductivity, k, Watt  $\text{cm}^{-1}\text{K}^{-1}$ 

T	k	T	k	T	k	T	k	T	k	T	k
<u>CURVE 1</u>											
195.6	0.344	323.2	0.860	4.64	0.887	2.42	0.992				
273.8	0.292	423.2	0.850	11.8	2.22	2.86	1.13				
299.2	0.266	523.2	0.825	14.7	2.72	3.33	1.42				
316.7	0.262	623.2	0.770	19.8	3.14	4.23	1.66				
334.2	0.279	673.2	0.750	22.8	3.41	4.64	1.86				
<u>CURVE 2</u>											
		773.2	0.795	26.2	3.73	5.27	3.60				
		873.2	0.665	31.8	3.71	12.6	4.72				
		973.2	0.635	32.7	3.79	15.5	5.11				
		1073.2	0.615	36.3	3.63	18.7	5.60				
323.2	0.235	1073.2	0.605	56.6	2.52	22.2	5.69				
373.2	0.269	1173.2	0.605	66.9	2.07	26.4	5.81				
<u>CURVE 3</u>											
		1273.2	0.600	67.5	2.01	31.2	5.46				
				72.9	1.89	32.7	5.17				
				78.6	1.82	43.8	3.91				
323.2	0.265			83.0	1.75	53.0	2.90				
373.2	0.285			103.2	1.54	61.9	2.38				
423.2	0.305			121.0	1.39	73.2	1.98				
<u>CURVE 4</u>											
				131.0	1.40	83.1	1.76				
				150.8	1.18	90.9	1.66				
323.2	0.480					105.6	1.54				
423.2	0.505					117.5	1.46				
523.2	0.525					123.4	1.45				
623.2	0.550										
<u>CURVE 5</u>											
				2.88	0.685						
				3.23	0.806						
				3.63	0.927						
				4.23	1.07						
				4.34	1.18						
				7.26	2.10						
				8.27	2.26						
				11.3	2.98						
				13.9	3.41						
				16.7	3.77						
				20.8	4.18						
				28.8	4.53						
				33.1	4.39						
				37.9	4.01						
				54.8	2.59						
				58.9	2.38						
				65.1	2.14						
				69.0	1.99						
				82.3	1.69						
				90.7	1.59						
				108.9	1.47						
				117.5	1.42						
				142.1	1.32						
<u>CURVE 6</u>											
				5.14	0.660						
				7.06	0.903						
				9.48	1.19						
				16.7	1.91						
				24.4	2.38						
				29.6	2.50						
				39.6	2.40						
				61.9	1.80						
				90.7	1.40						
<u>CURVE 7</u>											
				323.2	0.820						
				423.2	0.810						
				523.2	0.795						
				623.2	0.755						
				673.2	0.730						
<u>CURVE 8</u>											
				323.2	0.860						
				423.2	0.850						
				523.2	0.825						
				623.2	0.770						
				673.2	0.750						
				773.2	0.795						
				873.2	0.665						
				973.2	0.635						
				1073.2	0.615						
				1173.2	0.605						
				1273.2	0.600						
<u>CURVE 9</u>											
				4.54	0.403						
				8.27	0.847						
				11.8	1.13						
				14.5	1.31						
				18.2	1.59						
				21.8	1.84						
				28.6	2.10						
				35.3	2.24						
				38.3	2.24						
				42.3	2.23						
				65.1	1.75						
				76.6	1.59						
				90.3	1.48						
				114.3	1.36						
				122.6	1.35						
<u>CURVE 10</u>											
				5.14	0.660						
				7.06	0.903						
				9.48	1.19						
				16.7	1.91						
				24.4	2.38						
				29.6	2.50						
				39.6	2.40						
				61.9	1.80						
				90.7	1.40						
<u>CURVE 11</u>											
				2.42	0.992						
				2.86	1.13						
				3.33	1.42						
				4.23	1.66						
				4.64	1.86						
				5.27	3.60						
				12.6	4.72						
				15.5	5.11						
				18.7	5.60						
				22.2	5.69						
				26.4	5.81						
				31.2	5.46						
				32.7	5.17						
				43.8	3.91						
				53.0	2.90						
				61.9	2.38						
				73.2	1.98						
				83.1	1.76						
				90.9	1.66						
				105.6	1.54						
				117.5	1.46						
				123.4	1.45						
<u>CURVE 12</u>											
				2.88	0.685						
				3.23	0.806						
				3.63	0.927						
				4.23	1.07						
				4.34	1.18						
				7.26	2.10						
				8.27	2.26						
				11.3	2.98						
				13.9	3.41						
				16.7	3.77						
				20.8	4.18						
				28.8	4.53						
				33.1	4.39						
				37.9	4.01						
				54.8	2.59						
				58.9	2.38						
				65.1	2.14						
				69.0	1.99						
				82.3	1.69						
				90.7	1.59						
				108.9	1.47						
				117.5	1.42						
				142.1	1.32						
<u>CURVE 13</u>											
				470.4	0.852						
				524.8	0.839						
				580.9	0.808						
				634.3	0.786						
				638.7	0.787						
				717.1	0.748						
				721.5	0.748						
				750.4	0.756						
				866.0	0.722						
				871.5	0.722						
				872.6	0.711						
				993.3	0.697						
				1024.3	0.659						
				1031.0	0.665						
				1193.2	0.632						
				1201.0	0.637						
<u>CURVE 14</u>											

FIGURE AND TABLE NO. 10R RECOMMENDED THERMAL CONDUCTIVITY OF CHROMIUM



RECOMMENDED VALUES\*

$T_1$	$k_1$	$k_2$	$T_2$	$T_1$	$k_1$	$k_2$	$T_2$
0	0	0	-459.7	300	0.903	52.2	80.3
1	(0.401) <sup>‡</sup>	(23.2)	-457.9	311	0.886	51.2	100.1
2	(0.602)	(41.3)	-456.1	350	0.881	50.9	170.3
3	1.20	69.3	-454.3	400	0.873	50.4	260.3
4	1.66	92.4	-452.5	500	0.848	49.0	440.3
5	1.99	115	-450.7	600	0.805	46.5	620.3
6	2.38	138	-448.9	700	0.757	43.7	800.3
7	2.77	160	-447.1	800	0.713	41.2	980.3
8	3.14	181	-445.3	900	0.678	39.2	1160
9	3.50	202	-443.5	1000	0.653	37.7	1340
10	3.85	222	-441.7	1100	0.636	36.7	1520
11	4.18	242	-439.9	1200	0.624	36.1	1700
12	4.49	259	-438.1	1300	0.616	(35.6)	1880
13	4.78	276	-436.3	1400	0.611	(35.3)	2060
14	5.04	291	-434.5				
15	5.27	305	-432.7				
16	5.48	317	-430.9				
18	5.81	336	-427.3				
20	6.01	347	-423.7				
25	6.07	351	-414.7				
30	5.58	322	-405.7				
35	5.03	291	-396.7				
40	4.30	248	-387.7				
45	3.67	212	-378.7				
50	3.17	183	-369.7				
60	2.48	143	-351.7				
70	2.08	120	-333.7				
90	1.82	105	-315.7				
90	1.68	97.1	-279.7				
100	1.58	91.3	-279.3				
150	1.29	74.5	-189.7				
200	1.11	64.1	-99.7				
250	0.992	57.3	-9.7				
273.2	0.946	54.8	32.0				

REMARKS

The recommended values are for well-annealed 99.998% pure chromium with residual electrical resistivity  $\rho_0 = 0.0609 \mu\Omega \text{ cm}$  (characterization by  $\rho_0$  becomes important below room temperature). The values below  $1.5 \times 10^3 \text{ K}$  are calculated to fit the experimental data by using  $n = 2.00$ ,  $\beta^1 = 1.04 \times 10^{-4}$ , and  $\beta^2 = 2.49$ . The recommended values that are supported by experimental thermal conductivity data are thought to be accurate to within 3% of the true values near room temperature, and 3 to 10% at other temperatures.

\*  $T_1$  in K,  $k_1$  in  $\text{Watt cm}^{-1} \text{K}^{-1}$ ,  $T_2$  in F, and  $k_2$  in  $\text{Btu hr}^{-1} \text{ft}^{-1} \text{F}^{-1}$ .

† Values in parentheses are extrapolated.

# THERMAL CONDUCTIVITY OF COBALT

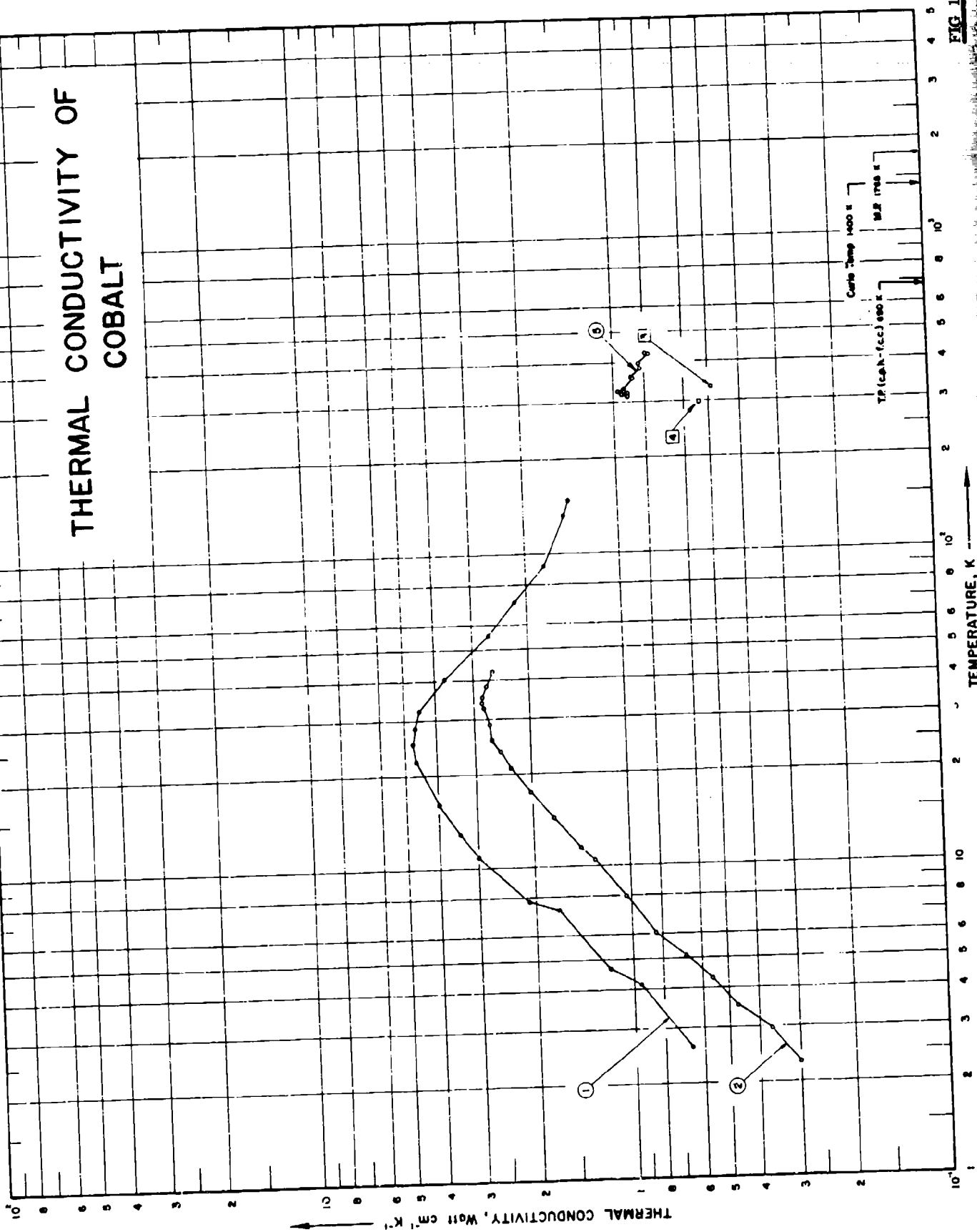


FIG 11



SPECIFICATION TABLE NO. 11 THERMAL CONDUCTIVITY OF COBALT

(Impurity < 0.20% each; total impurities < 0.50%)

[For Data Reported in Figure and Table No. 11 ]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition ( weight percent), Specifications and Remarks
1	150	L	1957	2.6-147		Co 1b	Impurities (by spectrographic analysis) approx. 0.0002 Si, < 0.0005 Fe, approx. 0.0001 Al, Mg and Cu 0.0001 each; specimen 2 mm dia; supplied by Johnson, Matthey and Co., Ltd. (JN644xx); annealed in vacuum for 2 hrs at 700 C; $D_0 = 0.09075 \mu$ ohm cm; $\rho(295 K)/\rho_0 = 64.3$ , $I_0 = 2.55 \times 10^6 W$ ohm $K^{-2}$ .
2	122	L	1955	2.3-43	3.0	Co 1	Polycrystalline rod; 3.03 cm long, 0.204 cm in dia; supplied by Johnson, Matthey and Co., Ltd.; annealed in vacuo for several hrs; electrical resistivity ratio $\rho(293 K)/\rho(20 K) = 29.4$
3	230	L	1925	302-2			Less than 0.03 impurities; supplied by Elmer and Amend; annealed at 900 C for 2 to 3 hrs before machining to size.
4	<del>444</del> 1057		1959	298-2			100 (Nominal) pure; measured at room temp. (assumed to be 25 C)
5	869	C	1964	313-430			99.97 pure; 0.951 cm dia, 4.346 cm long; supplied by Metallurgy Division of the National Physical Laboratory, Armo; iron used as reference; electrical resistivity reported as 6.5, 6.7, 7.4, 7.7, 8.7, 8.9, 10.3, 11.3, 11.4, and 11.6 $\mu$ ohm cm at 20, 22, 51, 55, 82, 87, 126, 151, and 155 C, respectively.

DATA TABLE NO. 11 THERMAL CONDUCTIVITY OF COBALT

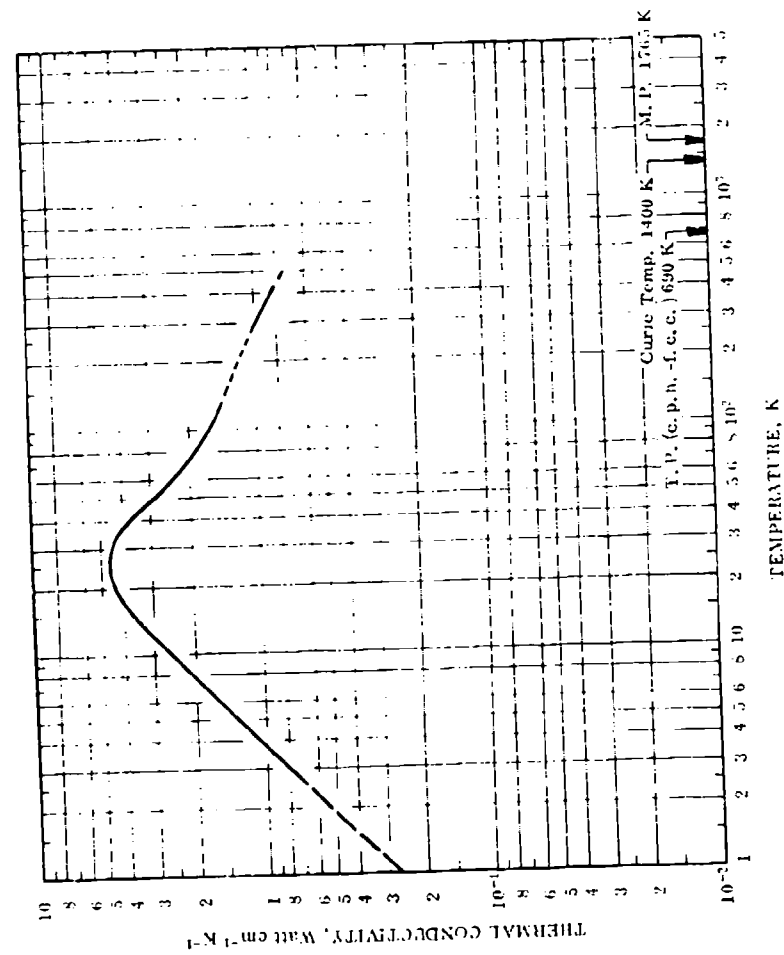
(Impurity - 0.20% each; total impurities - 0.50%)

[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup>K<sup>-1</sup>]

T	k	T	k
	<u>CURVE 1</u>	<u>CURVE 3</u>	
2.39	0.657	332.2	0.490
4.14	0.947		
1.66	1.187	<u>CURVE 4</u>	
7.24	1.697	298.2	0.536
7.76	2.096		
10.86	3.005	<u>CURVE 5</u>	
12.93	3.434	313.2	0.902
16.03	3.990	316.5	0.943
22.24	4.672	320.2	0.908
25.34	4.722	323.2	0.909*
28.45	4.646	324.2	0.930
32.07	4.495	324.2	0.970
40.34	3.737	327.2	0.930
55.34	2.652	356.2	0.872
70.34	2.172	360.2	0.881
91.03	1.742	384.2	0.825
132.90	1.490	397.2	0.837
147.40	1.439	424.2	0.775
	<u>CURVE 2</u>	430.2	0.793
2.30	0.300		
2.95	0.365		
3.53	0.470		
4.30	0.560		
5.10	0.675		
6.02	0.840		
7.97	1.035		
10.05	1.270		
11.50	1.420		
14.40	1.710		
17.40	2.020		
20.92	2.335		
23.53	2.500		
25.82	2.680		
28.75	2.690		
32.55	2.800		
33.75	2.840		
35.40	2.825		
38.30	2.740		
42.60	2.600		

\* Not shown on plot

FIGURE AND TABLE NO. 11K RECOMMENDED THERMAL CONDUCTIVITY OF COBALT



RECOMMENDED VALUES\*  
(For Polycrystalline)

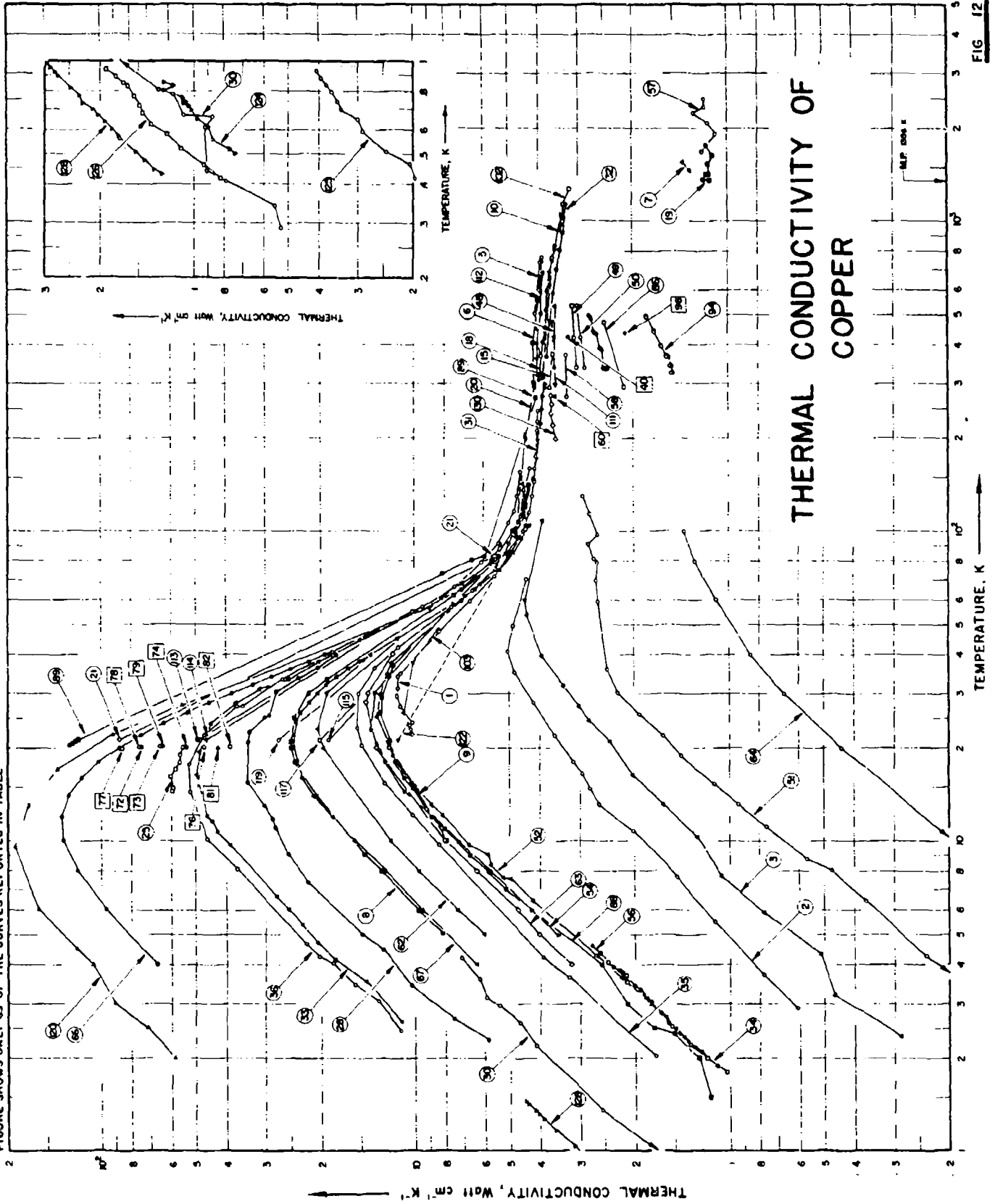
T <sub>1</sub>	k <sub>1</sub>	k <sub>2</sub>	T <sub>2</sub>
0	0	0	-459.7
1	(0.270) <sup>‡</sup>	(15.6)	-457.9
2	(0.539)	(31.1)	-456.1
3	0.808	46.7	-454.3
4	1.08	62.4	-452.5
5	1.34	77.4	-450.7
6	1.61	93.0	-448.9
7	1.87	108	-447.1
8	2.13	123	-445.3
9	2.38	138	-443.5
10	2.63	152	-441.7
11	2.87	166	-439.9
12	3.10	179	-438.1
13	3.31	191	-436.3
14	3.52	203	-434.5
15	3.71	214	-432.7
16	3.89	225	-430.9
18	4.19	242	-427.3
20	4.43	256	-423.7
25	4.70	272	-414.7
30	4.88	285	-405.7
35	4.23	244	-396.7
40	3.77	218	-387.7
45	3.32	192	-378.7
50	2.98	172	-369.7
60	2.48	143	-351.7
70	2.16	125	-333.7
80	1.94	112	-315.7
90	1.78	103	-297.7
100	1.68	97.1	-279.7
150	(1.39)	(80.3)	-199.7
200	(1.22)	(70.5)	-99.7
250	(1.09)	(63.0)	-
273.2	(1.04)	(60.1)	32.0
300	0.992	57.3	80.3
350	0.936	52.9	176.3
400	0.848	49.0	260.3
500	(0.746)	(43.1)	440.3

REMARKS

The recommended values are for well-annealed 99.999% pure cobalt with residual electrical resistivity  $\rho_0 = 0.0995 \mu\Omega \text{ cm}$  (characterization by  $\rho_0$  becomes important below room temperature). The values below 1.5 Tm are calculated to fit the experimental data by using  $n = 2.10 \cdot 10^{-4}$ ,  $0.747 \times 10^{-4}$ , and  $B = 3.71$ . The recommended values that are supported by experimental thermal conductivity data are thought to be accurate to within 4% of the true values near room temperature and 4 to 10% at other temperatures.

\* T<sub>1</sub> in K, k<sub>1</sub> in Watt cm<sup>-1</sup> K<sup>-1</sup>, T<sub>2</sub> in F, and k<sub>2</sub> in Btu lb<sup>-1</sup> ft<sup>-1</sup> F<sup>-1</sup>. <sup>‡</sup> Values in parentheses are extrapolated or interpolated.

FIGURE SHOWS ONLY 69 OF THE CURVES REPORTED IN TABLE



# THERMAL CONDUCTIVITY OF COPPER

FIG. 12

SPECIFICATION TABLE NO. 12 THERMAL CONDUCTIVITY OF COPPER

(Impurity < 0.20% each; total impurities < 0.50%)

[For Data Reported in Figure and Table No. 12.]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
1	114	L	1950	23-245	0.6-1.8	OFHC Cu	Free from oxygen; high conductivity; specimen 0.5 in. in dia and 20 in. long; obtained from American Brass Co.
2	154	L	1956	2.9-70	5	1	0.20 Ni, 0.10 O; 0.05 each of As, Sb, Fe, Pb, and Sn, 0.01 S, 0.003 Bi (composition according to All-Union standard); annealed to 800 C.
3	154	L	1956	2.4-108	5	2	Similar to the above specimen except unannealed.
4	135	L	1935	293, 473		1	99.986 pure; 0.022 O, 0.0016 Fe, 0.0015 S; annealed at 550 C for 1 hr.
5	124	P	1930	368-766		2	Electrolytically pure; specimen ~0.25 cm in dia; annealed for about 10 min at a bright red heat; electrical conductivity 4.47, 3.14, 2.07 and 1.99 x 10 <sup>5</sup> ohm <sup>-1</sup> cm <sup>-1</sup> at 95, 235, 466 and 497 C, respectively; density 8.87 g cm <sup>-3</sup> .
6	124	P	1930	302-744		2	Electrolytically pure; specimen ~0.25 cm in dia; annealed for about 10 min at a bright red heat; electrical conductivity 5.55, 4.0, 2.64 and 2.06 x 10 <sup>5</sup> ohm <sup>-1</sup> cm <sup>-1</sup> at 29, 136, 286 and 471 C, respectively; density 8.87 g cm <sup>-3</sup> .
7	90	L	1957	1456, 1550			0.036 O, 0.02 Ag, 0.002 Al, 0.002 Fe, 0.001 Ni, 0.001 Mg, 0.001 Si, 0.0005 Ca, traces of H and N; electrolytic tough-pitch copper; in molten state.
8	109	L	1957	5.1-142	2-5	Coalesced Copper	High-purity commercial coalesced copper; 0.0013 O, 0.0008 Pb, 0.0007 Ni, < 0.0005 each of Fe, As, and Sb, 0.0002 Sn, < 0.0001 Te, and Ag, < 0.00005 Bi; specimen 0.367 cm in dia and 23.2 cm long; annealed in helium 4 hrs at 400 C, cooled slowly to 200 C, and then kept in helium at 200 C for 8 hrs; density 8.90 g cm <sup>-3</sup> .
9	152	L	1949	10-20			Electrical copper; specimen 0.47 mm in dia and 900 mm long; annealed.
10	108	E	1958	315-1058	3		Commercial electrolytic copper.
11	84	P	1918	349-636			Specimen drawn wire 2.5 mm in dia.
12	95	E	1915	21-374		Cu I	Electrolytically pure; specimen 1 mm in dia; supplied by Siemens and Halske Co.
13	95	E	1915	22-375		Cu II	Electrolytically pure but purity lower than the above specimen.
14	52	L	1952	339-533	5	A	Oxygen-free (0.01 O) high-conductivity copper.
15	77	E	1900	291, 373		Cu II	< 0.05 <del>Fe+Zn</del> , specimen 1.108 cm in dia and 27 cm long; density 8.65 g cm <sup>-3</sup> at 18 C.
16	77	E	1900	291, 373		Cu III	< 0.05 <del>Fe+Zn</del> , specimen 1.107 cm in dia and 27 cm long.
17	77	E	1900	29, 373		Cu III	0.05 Pb, traces of Ni and Fe; specimen 1.107 cm in dia and 27 cm long; drawn; density 8.88 g cm <sup>-3</sup> at 18 C.
18	130	P	1951	309-634			Specimen 0.125 in. in dia and at least 50 cm long.
19	41	C	1956	1362-1761	± 2		Electrolytic tough pitch copper; before measurement, 0.012 O <sub>2</sub> , 0.0048 N <sub>2</sub> , and trace Al, Ca, Mg, Ni, Si and Ti; after measurement, 0.0059 O <sub>2</sub> , 0.0055 N <sub>2</sub> , and all the metallic impurities reduced about ten fold; density 8.83 g cm <sup>-3</sup> ; in molten state.

SPECIFICATION TABLE NO. 12 (continued)

Curve No.	Ref. No.	Method Used*	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
20	80	L	1928	9.5-283		Cu 2b	Single crystal; specimen 0.6 cm in dia and 12 cm long; supplied by General Electric Co.
21	57	L	1927	21, 83		Cu 2b	Very high purity; porous natural crystal from Lake Superior, hammered from 3 mm to 1.3 mm and then annealed 3 hrs at 380 C; electrical resistivity 1.562, 0.235 and 0.00187 $\mu\text{ohm cm}$ at 273, 83 and 21 K, respectively.
22	57	L	1927	21, 83		Cu 3	"Purest" electrolytic copper; fine grains; electrical resistivity 1.552, 0.239 and 0.00424 $\mu\text{ohm cm}$ at 273, 83 and 21 K, respectively.
23	57	L	1927	21, 83		Cu 4a	"Purest" electrolytic copper; with fine grains; annealed 4.5 hrs at 380 C; electrical resistivity 1.56, 0.240, 0.00406 $\mu\text{ohm cm}$ at 273, 83 and 21 K, respectively.
24	57	L	1927	21, 83		Cu 6a	Not very pure; single crystal; annealed 7.5 hrs at 380 C; electrical resistivity 1.58, 0.249, 0.01356 $\mu\text{ohm cm}$ at 273, 83 and 21 K, respectively.
25	18	L	1936	15-20		Cu 2b	Specimen 0.2 mm in dia.
26	58	L	1934	22, 79		Cu 2b	Very high purity; probably somewhat deformed.
27	127	L	1925	369-898		Cu 1	99.9 pure; supplied by Bolton and Sons, Ltd., Oakmoor; density 8.92 $\text{g cm}^{-3}$ at 21 C; electrical resistivity 1.69, 2.60, 3.73, 4.88, and 6.03 $\mu\text{ohm cm}$ at 14, 0, 144, 8, 306, 0, 470, 0, and 630.2 C, respectively.
28	97, 122	L	1952	2.5-41	2-3	Cu 1	99.999 pure; polycrystalline; JM 4234 from Johnson, Matthey and Co.; 3.02 mm dia and 2.99 cm long; annealed for several hrs in vacuo at 800 C; $\rho(293 \text{ K}/\rho(20 \text{ K})) = 85.3$ .
29	23	E	1894	347-440			Electrolytic copper; specimen a prismatic bar with 5 cm x 2 cm cross section.
30	103	L	1953	0.29-4.2			Polycrystalline; commercial grade high purity magnet wire; specimen 0.025 cm in dia and 27.2 cm long; supplied by General Electric Co.
31	88	L	1908	107-299			Turned from soft-drawn high-conductivity copper conductor; specimen 0.585 cm in dia and 7-8 cm long; electrical resistivity 0.375, 0.543, 0.637, 0.909, 0.989, 1.365, 1.506, and 1.750 $\mu\text{ohm cm}$ at -176.8, -151.2, -136.9, -102.0, -91.1, -36.0, -15.5, and 16.9 C, respectively; density 8.84 $\text{g cm}^{-3}$ at 23 C.
32	89	C	1956	367-1144			99.9 pure; electrolytic tough pitch copper; density 8.92 $\text{g cm}^{-3}$ at 24 C.
33	11	L	1952	2.6-91			99.999 pure; about 0.0005 Ag, < 0.0003 Ni, < 0.0004 Pb; JM 4234 from Johnson, Matthey and Co.; drawn and annealed in a helium atmosphere at 450 C for 6 hrs; electrical resistivity for the range 12-15 K given as $\rho = 5.27 \times 10^{-3} + 2.64 \times 10^{-10} T^3$ ( $\mu\text{ohm cm}$ ).
34	2	L	1948	1.8-4.1	2		resistivity for the range 12-15 K given as $\rho = 0.055 + 0.0005 T^3$ CM.
35	145	L	1953	2.0-160	1, but up to 4 between 5 and 15 K	Cu 1	0.003 Ag, 0.003 Ni, 0.003 Pb; approximate composition; free from oxygen; Johnson Matthey and Co. Ltd. No. 1562; annealed in air; $\rho = 0.055 + 0.0005 T^3$ CM. 99.999 pure; JM 4272 from Johnson, Matthey and Co.; about 0.0005 Ag, 0.0004 Pb, and < 0.0003 Ni, and barely visible spectral lines of Ga and Fe; specimen 2 mm in dia rod; as drawn; electrical resistivity for the range 10 to 35 K given as $\rho = 0.0576 \times 10^{-6} + 3.7 \times 10^{-16} T^3$ ( $\text{ohm cm}$ ).

SPECIFICATION TABLE NO. 12 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
36	145	L	1953	2.5-155	4	Cu2	The above specimen annealed in vacuo at 550 C for 3 hrs.
37	116	F	1895	219-260			Electrolytic copper.
38	186	P	1928	305.2			Electrolytically pure; specimen 2.5 mm in dia and 4.69 cm long; electrical conductivity $5.58 \times 10^9$ mho $\text{cm}^{-1}$ at 32 C; density $8.93 \text{ g cm}^{-3}$ at 32 C.
39	224	L	1923	423.2			99.98% pure; annealed.
40	224	L	1923	428.2			99.98% pure; cast.
41	224	L	1923	423.2			99.97% pure; hard-drawn.
42	224	L	1923	430.2			99.76% pure; cast.
43	270	P	1915	308, 333			specimen 0.25 cm in dia and 39 cm long; density $8.93 \text{ g cm}^{-3}$ at room temperature (from Tabellen of Landolt and Bornstein).
44	271	L	1918	273-403	0.11		$\pm 0.079 \text{ O.}$
45	271	L	1918	273-403	0.11		$\pm 0.079 \text{ O.}, 0.106 \text{ Ni.}$
46	271	L	1918	273-403	0.11		$\pm 0.022 \text{ O.}$
47	271	L	1918	273-403	0.11		$\pm 0.022 \text{ O.}, 0.106 \text{ Ni.}$
48	52	L	1952	339-533	5	B	0.015 Fe, 0.011 P; cast.
49	52	L	1952	339-533	5	C	0.061 Fe, 0.016 P; cast.
50	52	L	1952	339-533	5	D	0.089 Fe, 0.015 P; cast.
51	341	L	1954	1.9-130			0.056 Fe; nominal composition; homogenized and annealed; residual electrical resistivity (at helium temp) $0.56 \mu\text{ohm cm.}$
52	341	L	1954	1.9-142			0.0943 Fe; nominal composition; homogenized and annealed; residual electrical resistivity (at helium temp) $0.331 \mu\text{ohm cm.}$
53	225	L	1926	373.2			Electrolytic.
54	145	L	1953	5.0-58	$\leq 4$	Cu 3	Electrolytic.
55	427	L	1960	303.2	1-3	ETP	99.999% pure; JM 4272 from Johnson, Matthey and Co.; about 0.0005 Ag, 0.0004 Pb, 0.0003 Ni, and barely visible spectral lines of Ga and Fe; specimen 1 mm in dia rod; as drawn.
56	404	L	1950	2.5-4.6	4		Electrolytic tough pitch copper; specimen 0.75 in. in dia and 9 in. long.
57	428	R	1957	1673-2500	$\pm 10$		99.998% pure; polycrystal; supplied by Johnson, Matthey and Co.
58	246	T	1919	273, 373			In liquid state.
59	429	R	1937	285.7	$\pm 0.7$		Rolled, drawn, and then heated 0.5 hr at temp close to melting point. Pure.

SPECIFICATION TABLE NO. 12 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
60	430	T	1924	273.2			Pure; rolled and drawn to wire of 1 mm <sup>2</sup> cross section and 3 cm long, and heated at temp close to melting point.
61	431	E	1944	597-1245			Polycrystal.
62	432	L	1957	5.0-40		Coalesced Cu	99.9% pure; 0.0013 O <sub>2</sub> , 0.0007 Ni, 0.0008 Pb, 0.0002 Sn, each of Fe, As, Sb < 0.0005, 0.0001 Te, and Bi < 0.00005; cold rolled, annealed for 1 hr at 650 C, redrawn and reannealed for 17 min at 760 C, followed by grinding to sample size of 0.144 in. in dia; density 8.899 g cm <sup>-3</sup> , porosity 0.5%.
63	432	L	1957	4.0-40		Electrolytic tough pitch	0.01 Fe, 0.001 each Ag and Zn, each of Al, Cr, Pb, Mg, Mn, and Sn < 0.0001; electrolytic tough pitch; density 8.914 g cm <sup>-3</sup> ; ground.
64	432	L	1957	5.0-100		Phosphorus deoxidized Cu	0.027 P, 0.01 each of Fe, Ag, and Zn, 0.001 each Ni and Si, < 0.0001 each of Al, Cr, Pb, Mg, and Mn; density 8.917 g cm <sup>-3</sup> ; ground.
65	433	L	1940	78.2		Electrolytic Cu	0.015 Sb, 0.010 Fe, 0.007 S, trace Pb.
66	434	L	1959	4.0-105			99.999 pure; swaged from about 0.375 in. down to about 0.072 in., cleaned with a 1:1 solution of HCl, and a 1:10 solution of HNO <sub>3</sub> ; annealed in vacuum for 2 hrs at 400 C, drawn through tungsten carbide dies to 0.070 in., cleaned with acids, and finally annealed again in vacuum for 2 hrs at 400 C; slight unavoidable work hardening of the sample during installation in the apparatus.
67	434	L	1959	4.0-105			99.999 pure; swaged from about 0.375 in. down to about 0.0816 in., cleaned with acids, annealed in vacuum for 2 hrs at 400 C, and then drawn through tungsten carbide dies to 0.070 in. in which the cross-section area reduced by 26.4%; not annealed again after drawing.
68	435	L	1900	291.2			Pure.
69	435	L	1900	291.2			Trace As.
70	410	R	1935	273.2	1		Pure; electrical conductivity 62.8 x 10 <sup>4</sup> mho cm <sup>-1</sup> at 273.2 K.
71	436	L	1938	21.17		Cu 12	Natural single crystal; tempered for 3 hrs at 380 C; measured at H (the transverse magnetic field strength) = 0 and $\theta$ (the angle between magnetic field direction and z line perpendicular to rod axis) = 0° at which the electrical resistivity is nearly minimum and H nearly parallel to [100] direction.
72	436	L	1938	21.17		Cu 12	The above specimen measured at H = 2280 oersteds and $\theta = 0^\circ$ .
73	436	L	1938	21.18		Cu 12	The above specimen measured at H = 4490 oersteds and $\theta = 0^\circ$ .
74	436	L	1938	21.21		Cu 12	The above specimen measured at H = 8750 oersteds and $\theta = 0^\circ$ .
75	436	L	1938	21.23		Cu 12	The above specimen measured at H = 10880 oersteds and $\theta = 0^\circ$ .
76	436	L	1938	21.25		Cu 12	The above specimen measured at H = 12200 oersteds and $\theta = 0^\circ$ .



SPECIFICATION TABLE NO. 12 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)	Specifications and Remarks
77	436	L	1936	21-17		Cu 12		The above specimen measured at $H = 0$ oersteds and $\theta = -40^\circ$ at which the electrical resistivity is nearly maximum and $H$ nearly parallel to [110] direction.
78	436	L	1938	21-18		Cu 12		The above specimen measured at $H = 2280$ oersteds and $\theta = -40^\circ$ .
79	436	L	1938	21-19		Cu 12		The above specimen measured at $H = 4490$ oersteds and $\theta = -40^\circ$ .
80	436	L	1938	21-24		Cu 12		The above specimen measured at $H = 8750$ oersteds and $\theta = -40^\circ$ .
81	436	L	1939	21-24		Cu 12		The above specimen measured at $H = 10800$ oersteds and $\theta = -40^\circ$ .
82	436	L	1938	21-30		Cu 12		The above specimen measured at $H = 12200$ oersteds and $\theta = -40^\circ$ .
83	437	E	1902	291.2				Pure.
84	390	P	1956	354.2				Pure.
85	438	E	1914	21-373				Electrolytic copper wire; not annealed and not bent, but heated considerably during soldering.
86	135	L	1935	293-473		Bar 104		99.80 Cu, 0.19 Si, and 0.02 Fe; specimen 0.75 in. in dia and 8 in. long; electrical conductivity $29.58$ and $21.30 \times 10^4$ mho $\text{cm}^{-1}$ at 20 and 200 C, respectively; annealed at 700 C for 2 hrs.
87	439	L	1935	323-848	2			99.9 pure.
88	355	L	1955	1.5-142	1			99.99 pure; 0.02 Ge; specimen 1~2 mm in dia and 6 cm long, drawn; and annealed; electrical resistivity 1.92 $\mu\text{ohm cm}$ at 295 K.
89	619	L	1916	20-273				High purity single crystal natural copper.
90	619	L	1916	22-273				Commercially pure; fine crystalline.
91	440	L, R	1940	78, 273		Electrolytic Cu		Impurities: 0.015 Sb, 0.010 Fe, 0.007 S, 0.0003 Ar; annealed in nitrogen stream for 20 hrs at 380-430 C; electrical conductivity 6.22 and $43.1 \times 10^4$ mho $\text{cm}^{-1}$ at 273 and 78 K, respectively.
92	134	L	1931	337-477	< 2	93		99.94 pure; 0.042 P, 0.04 Fe; annealed at 650 C for 1 hr and cooled in air.
93	134	L	1931	337-494	< 2	82		99.97 pure; 0.075 P, 0.04 Fe; annealed at 650 C for 1 hr and cooled in air.
94	134	L	1931	325-496	< 2	95		99.74 pure; 0.18 P; annealed at 650 C for 1 hr and cooled in air.
95	67	L	1932	438				99.917 pure; 0.083 P; specimen 0.5 in. in dia and 6.5 in. long; annealed.
96	67	L	1932	438				99.865 pure; 0.135 P; specimen 0.5 in. in dia and 6.5 in. long; annealed.
97	67	L	1932	438				99.93 pure; 0.07 As; specimen 0.5 in. in dia and 6.5 in. long; annealed.
98	67	L	1932	438				99.856 pure; 0.144 As; specimen 0.5 in. in dia and 6.5 in. long; annealed.
99	230	L	1955	332.2				Impurity < 0.03; electrical conductivity $50.8 \times 10^4$ mho $\text{cm}^{-1}$ at 296.2 K.

SPECIFICATION TABLE NO. 12 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
100	135	L	1935	293, 473		Bar 114	0.07 Mn, 0.01 Fe, 0.02 Mg; specimen 0.75 in. in dia and 8 in. long, electrical conductivity 52.55 and $32.18 \times 10^4$ mho $\text{cm}^{-1}$ at 20 and 200 C, respectively; annealed at 700 C for 2 hrs.
101	135	L	1935	293, 473		Bar 115	0.14 Mn, 0.01 Fe, 0.01 Mg; specimen 0.75 in. in dia and 8 in. long; electrical conductivity 45.79 and $29.4 \times 10^4$ mho $\text{cm}^{-1}$ at 20 and 200 C, respectively; annealed at 700 C for 2 hrs.
102	40	L	1956	764-1287	5	Electrolytic tough pitch	Electrolytic tough pitch copper meeting Federal Specification QQ-C 576 (minimum 99.9 Cu); density 8.83 $\text{g cm}^{-3}$ ; specimen 7 in. in dia and 1.5 in. thick.
103	562	L	1949	23-245		OFHC	Oxygen-free high conducting (OFHC) copper.
104	496	C	1940	80, 273	1		Pure.
105	504	P	1961	295.2	1.5	OFHC	specimen 1.9 $\text{cm}^2$ in cross-sectional area and 0.312 cm thick.
106	622	P	1960	363.2			Commercial grade; 99.82 pure; density 8.5 $\text{g cm}^{-3}$ .
107	622	P	1960	363.2			The above specimen, second run.
108	622	P	1960	363.2			The above specimen, third run.
109	622	P	1960	363.2			The above specimen, fourth run.
110	579	L	1936	15-20			"Very pure".
111	135	L	1935	293, 473		Bar No. 99	0.07 Al and 0.01 Fe; annealed at 750 C for 2 hrs; electrical conductivity 52.58 and $31.69 \times 10^4$ mho $\text{cm}^{-1}$ at 20 and 200 C, respectively.
112	620		1956	320-773	3		99.99 pure; polycrystalline; electrical resistivity 1.92, 2.50, 3.17, 3.81, 4.43 and 5.03 $\mu\text{ohm cm}$ at 46.9, 134.5, 232.2, 323.3, 411.1 and 499.3 C, respectively.
113	57	L	1927	21, 83		Cu 4	"Purest" electrolyte; with fine grains.
114	57	L	1927	21, 83		Cu 4b	The above specimen hammered, then annealed for 4.5 hrs at 380 C, and recrystallized at 950 C for 5 min.
115	57	L	1927	21, 83		Cu 6	Not very pure; single crystal; sawed from larger block and lathed into rod.
116	57	L	1927	21, 83		Cu 6b	From the same block as the above specimen Cu6; hammered from 6 mm to 2.5 mm dia and annealed for 3 hrs at 380 C.
117	57	L	1927	21, 83		Cu 6c	Similar to the above specimen Cu6b except further annealed in vacuum for 5 min at 950 C; about 25 grain cross-sections per $1 \text{ mm}^2$ .
118	57	L	1927	21, 83		Cu 7	Lathed from the same block as specimen 6; 3 to 4 crystal grains on the measuring length; unannealed.
119	57	L	1927	21, 83		Cu 7a	Similar to the above specimen Cu7 except annealed for 4 hrs at 380 C.

SPECIFICATION TABLE NO. 12 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
120	623	L	1960	2.0-55			99.999 pure copper from the Central Research Lab. of the American Smelting and Refining Co.; less than 0.0001 each of Fe, Sb, Se, and less than 0.0002 each of Te and As; 0.030 in. dia wire, rolled and drawn from a 0.75 in. dia rod then annealed at 530 C in vacuo for some hrs; residual electrical resistivity $0.865 \pm 0.01 \times 10^{-7}$ ohm cm.
121	57	L	1927	21, 83		Cu 2a	Very high purity; porous natural crystal hammered from 3 mm to 1.3 mm dia.
122	57	L	1927	21, 83		Cu 9	Not very pure; single crystal solidified from melt; completely undeformed and unworked.
123	†	P	1968	298.2		No. 1	Disk specimen 1.25 cm in diameter and 0.042 cm thick; obtained from DuPont Detached stock; thermal conductivity value calculated from the measurement of thermal diffusivity with density value taken from Lyman, T. (editor, "Metals Handbook", 8th ed., Vol. 1, 1961) and specific heat capacity value taken from Kelley, K. K. (U. S. Bureau of Mines Bulletin 584, 1960).
124	683		1963	0.50-0.88		Cu 1	Commercial, polycrystalline wire.
125	683		1963	0.42-0.93		Cu 2	99.999 pure, polycrystalline wire.
126	683		1963	0.42-0.94		Cu 3	99.999 pure, polycrystalline wire; annealed at 898.2 K for 3 hrs.
127	706	L	1881	273, 373			Density $8.82 \text{ g cm}^{-3}$ ; electrical conductivity $45.74$ and $33.82 \times 10^4 \text{ ohm}^{-1}\text{cm}^{-1}$ (the author reported $45.74$ and $33.82 \times 10^5$ , probably a typographical error) at 0 and 100 C, respectively.
128	880	L	1965	0.4-1.5	$\pm 1$		Specimen a foil of 0.05 mm thickness, supplied by Chase Brass and Copper Co.; annealed at 530 C for three hrs; residual resistance ratio 270.
129	851	L	1960	80-217	10		Commercial copper; specimen 0.375 in. in dia and 2.975 in. long; cylinder with a part of 1.1 in. long at one end turned down to a dia of 0.125 in.; data corrected for drift rate.
130	851	L	1960	199-275	10		Similar to the above specimen.
131	843	-	1966	298.2			Nearly spherical grains supplied by Belmont Smelting and Refining Co.; mesh size -30 + 35; specimen contained in a 0.75 in. dia and 2 in. long cylindrical cell; thermal conductivity measured by using the transient line source method, the heat source was a 36-gauge constantan wire contained in a 0.025 in. O.D. hypodermic tube soldered along the axis of the cylindrical cell, data calculated from measured line temperatures at two certain times; measured in Freon-12 under a pressure of ~100 psig.
132	843	-	1966	298.2			Similar to the above specimen; measured in argon under a pressure of ~100 psig.
133	843	-	1966	298.2			Similar to the above specimen; measured in nitrogen under a pressure of ~100 psig.
134	843	-	1966	298.2			Similar to the above specimen; measured in methane under a pressure of ~100 psig.
135	843	-	1966	298.2			Similar to the above specimen; measured in helium under a pressure of ~100 psig.

† Larson, K. B. and Koyama, K., J. Appl. Phys., 39 (9), 4408-16, 1968.

SPECIFICATION TABLE NO. 12 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
136	843	-	1966	298.2			Similar to the above specimen; measured in hydrogen under a pressure of ~100 psig. Specimen No. 10 gauge commercial wire 10 cm in length electrical resistivity 1.73 $\mu$ ohm cm at 0 C. The above specimen measured in a longitudinal magnetic field of 10000 gaussess.
137	1005	E	1927	273.2			
138	1005	E	1927	273.2			

DATA TABLE NO. 12 THERMAL CONDUCTIVITY OF COPPER  
(Impurity < 0.20% each; total impurities < 0.50%)  
(Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup>K<sup>-1</sup>)

CURVE 1		CURVE 3 (cont.)		CURVE 5 (cont.)		CURVE 13*		CURVE 19		CURVE 26*		CURVE 29 (cont.)		CURVE 32 (cont.)	
T	k	T	k	T	k	T	k	T	k	T	k	T	k	T	k
22.9	10.2	4.32	0.510	9.08	14.5	21.5	16.9	1362	1.12	21.50	82.0	407.0	4.06	811.0	3.41
23.7	10.3	5.90	0.775	14.0	20.7	91.4	4.62	1373	1.16	78.80	5.78	416.8	4.20	922.0	3.34
24.1	10.1	7.70	1.06	18.3	24.2	273.1	3.85	1430	1.15			427.8	4.20	1033.0	3.30
27.1	11.2	10.4	1.25	20.0	24.8	291.2	3.83	1431	1.13	CURVE 27*		440.0	4.28	1144.0	3.25
29.4	11.4	13.2	1.63	21.0	24.9	298.3	3.83	1548	1.14						
31.1	11.2	16.4	2.00	25.0	24.2	374.6	3.83	1639	1.10	368.2	3.77				
34.0	11.3	21.0	2.38	30.0	21.9			1761	1.16	449.1	3.73				
34.6	11.1	24.5	2.78	33.2	19.1	CURVE 14*				539.0	3.71				
47.5	8.34	27.2	3.01	62.0	7.50					746.0	3.59				
59.7	6.64	31.8	3.34	80.0	5.64	338.7	3.80	CURVE 20		897.9	3.52				
71.7	5.53	39.5	3.91	100.0	4.82	422.1	3.77	95.4	4.73						
85.6	4.90	53.6	4.39	120.0	4.43	533.2	3.66	98.4	4.73	CURVE 28					
99.8	4.49	60.0	4.43	142.0	4.31	199.5	4.41	195.5	4.41	2.5	6.1				
115.1	4.30	108.0	3.91			CURVE 15		292.6	4.13	2.8	7.2				
130.1	4.20			CURVE 9		291.2	3.68	CURVE 21		3.7	10.4				
145.2	4.15			10.0	7.95	373.2	3.50	71.2	88.0	4.7	12.7				
159.6	4.12			15.0	10.0			83.2	5.57	5.2	14.8				
175.5	4.03	293.2	3.93	20.0	12.1	CURVE 16*				7.8	21.9				
211.8	4.05	473.2	3.90			291.2	3.73	CURVE 22*		10.0	25.1				
226.6	4.05			CURVE 10		373.2	3.67	21.2	54.3	11.4	27.4				
245.2	4.04			314.7	3.87			83.2	5.49	13.3	29.8				
		368.2	3.93	623.7	3.66	CURVE 17*				15.7	33.5				
		508.2	4.10	813.5	3.54	291.2	3.84	21.2	54.3	16.2	35.6				
		739.2	3.93	1058.4	3.35	373.2	3.80	83.2	5.49	18.7	33.4				
		766.2	3.89							22.9	33.1				
				CURVE 11*						25.5	29.4				
		10.8	2.00	349.2	3.82	CURVE 18				27.0	26.5				
		13.2	2.59	357.2	3.83	309.2	3.98	2	7.7	30.7	25.1				
		14.8	2.78	409.2	4.18	409.2	3.86	83.2	5	31.5	23.2				
		16.5	2.93	559.2	3.97	519.2	3.77	CURVE 24*		36.8	20.3				
		18.0	3.13	636.2	3.69	596.2	3.74	21.2	25.3	40.7	17.6				
		22.0	3.57			687.2	3.67	83.2	5.28	CURVE 29*					
		28.0	4.18	CURVE 7		749.2	3.58			347.2	3.82				
		35.0	4.81	20.7	17.6	834.2	3.51	14.5	59.5	350.4	3.84				
		49.5	4.85	60.7	4.93			14.8	59.1	335.0	4.02				
		70.0	4.39	273.1	3.92			16.2	60.2	361.4	3.98				
				291.2	3.90			17.1	58.1	370.5	3.82				
				283.7	3.90			18.0	56.5	376.1	3.82				
				374.7	3.90			19.0	56.5	382.7	3.80				
								20.1	54.9	390.1	3.95				
										398.1	3.95				
				CURVE 8											
				5.05	8.16										
				6.00	9.80										
				8.00	12.8										
				CURVE 3											
				2.37	0.285										
				3.20	0.460										

\* Not shown on plot

DATA TABLE NO. 12 (continued)

T	k	I	k	T	k	T	k	T	k	T	k	T	k	T	k		
<u>CURVE 35</u>																	
2.03	1.730	130.10	4.700	273.2	3.874	1.88	0.093*	18.8	11.7	3.84	2.22	20.0	19.6	22	75.0		
3.65	3.250	155.40	4.600	353.2	3.887	2.09	0.102*	20.9	12.2	3.95	2.34	23.0	20.5	24	83.5		
4.25	3.905	<u>CURVE 37*</u>			373.2	3.891	2.32	0.114*	25.2	13.2	4.38	2.58	30.0	19.1	26	51.5	
9.75	8.393	<u>CURVE 38*</u>			403.2	3.895	2.86	0.143*	29.6	12.9	4.61	2.74	40.0	13.8	28	45.25	
12.18	10.170	219.3	3.854	<u>CURVE 39*</u>			3.54	0.183*	34.5	12.3	<u>CURVE 57</u>			30	38.50		
15.42	12.513	224.5	3.862	273.2	3.757	6.45	0.387	37.4	11.9	<u>CURVE 63</u>			32	33.10			
20.30	14.750	229.2	3.940	353.2	3.778	8.08	0.469	59.8	7.05	4.0	3.2	4.0	3.2	34	28.60		
23.15	15.280	233.4	3.958	373.2	3.782	11.21	0.757	67.6	6.20	1673.2	1.20	5.0	4.0	36	24.90		
29.83	15.160	237.2	4.083	403.2	3.791	13.21	0.937	78.4	5.45	1914.3	1.08	6.0	4.7	38	21.75		
36.10	13.480	240.6	4.167	<u>CURVE 46*</u>			15.41	1.11	91.1	4.95	2087.6	1.15	8.0	6.35	40	19.20	
42.20	11.290	243.6	4.180	273.2	3.866	18.9	1.41	112.3	4.48	2243.2	1.27	10.0	8.0	65	7.04		
56.80	7.650	246.3	4.268	373.2	3.866	21.9	1.64	128.9	4.44	2353.7	1.17	20.0	13.3	75	5.7		
64.55	6.510	248.7	4.290	403.2	3.866	25.7	1.92	141.8	4.51	2499.8	1.17	28.0	14.4	85	5.1		
75.30	5.495	250.3	4.330	<u>CURVE 47*</u>			373.2	3.682	<u>CURVE 58</u>			95	5.1				
91.30	4.985	254.2	4.350	273.2	3.866	30.4	2.25	<u>CURVE 54</u>			105	4.7					
101.10	4.780	257.0	4.393	353.2	3.866	36.0	2.44	373.2	3.27	<u>CURVE 64</u>			4.7				
112.00	4.578	259.6	4.430	403.2	3.866	69.4	2.65	<u>CURVE 59*</u>			40.0	11.8					
136.00	4.475	<u>CURVE 38*</u>			273.2	3.761	79.2	2.63	285.7	3.838	<u>CURVE 65*</u>			6.37			
160.30	4.273	305.2	3.845	353.2	3.782	81.2	2.69	5.0	3.50	5.0	0.091*	5.0	0.091*	4	9.45		
<u>CURVE 36 (cont.)</u>																	
2.45	11.09	<u>CURVE 39*</u>			422.1	3.548	90.8	2.80	6.0	4.25	6.0	0.150*	6.0	0.150*	8	12.5	
3.05	13.13	423.2	3.766	373.2	3.786	97.6	2.62	8.0	5.75	10.0	0.193*	8.0	0.193*	10	15.57		
3.45	15.57	<u>CURVE 40</u>			403.2	3.795	114.1	2.76	10.0	7.25	<u>CURVE 60</u>			12	18.4		
4.05	18.31	423.2	3.766	373.2	3.786	130.2	2.93	12.0	8.75	20.0	0.435	20.0	0.435	14	20.85		
4.25	20.25	<u>CURVE 41*</u>			<u>CURVE 48</u>			<u>CURVE 52</u>			30.0	0.865	30.0	0.865	16	22.8	
4.67	22.28	428.2	3.222	338.7	3.548	1.88	1.09	14.5	10.75	273.2	3.55	40.0	0.850	18	23.9		
6.12	36.93	428.2	3.222	422.1	3.548	2.08	1.23	18.0	12.50	<u>CURVE 61*</u>			20	24.41			
10.13	45.88	428.2	3.222	533.2	3.531	2.35	1.39	23.0	13.75	597.1	3.51	60.0	1.10	22	24.44		
14.40	52.19	<u>CURVE 41*</u>			<u>CURVE 49</u>			<u>CURVE 55*</u>			703.1	3.47	100.0	1.27	24	24.00	
17.65	52.60	423.2	3.682	338.7	3.029	58.0	7.50	30.0	13.50	766.1	3.46	100.0	1.39	26	23.09		
21.90	47.09	423.2	3.682	422.1	3.081	3.12	1.84	58.0	7.50	840.1	3.44	<u>CURVE 65*</u>			28	21.87	
27.10	37.13	430.2	3.180	533.2	3.115	3.64	2.19	303.2	3.94	940.1	3.39	78.2	5.51	30	20.45		
27.20	35.60	<u>CURVE 42*</u>			<u>CURVE 50</u>			<u>CURVE 56</u>			1040.1	3.36	<u>CURVE 66</u>			32	19.08
33.30	26.45	308.2	3.800	422.1	3.081	4.29	2.65	303.2	3.94	1115.1	3.32	4	66.9	36	16.32		
40.80	18.31	333.2	3.781	533.2	3.115	6.41	4.18	2.53	1.49	1225.1	3.11	5	96.7	38	15.03		
55.00	10.17	<u>CURVE 43*</u>			<u>CURVE 51</u>			<u>CURVE 53*</u>			1245.1	3.09	8	119.9	40	11.88	
57.00	9.410	338.7	2.856	8.40	5.73	7.62	5.21	2.58	1.56	<u>CURVE 62</u>			45	11.4			
65.95	7.425	422.1	2.942	9.03	5.84	8.40	5.73	2.79	1.64	5.0	6.0	6.0	6.0	65	6.4		
79.15	6.100	533.2	3.781	9.73	6.74	9.03	5.84	3.01	1.79	5.0	6.0	6.0	6.0	75	5.36		
91.10	5.400	<u>CURVE 44*</u>			<u>CURVE 52</u>			<u>CURVE 54</u>			104.1	3.32	12	134.5	85	4.85	
106.55	5.000	338.7	2.856	12.07	8.15	13.07	8.15	3.19	1.81	273.2	3.27	14	127.5	95	4.53		
115.70	4.800	422.1	2.942	13.07	8.87	15.39	10.17	3.31	1.93	373.2	3.29	16	115.7	105	4.3		
<u>CURVE 45*</u>																	
<u>CURVE 46*</u>																	
<u>CURVE 47*</u>																	
<u>CURVE 48</u>																	
<u>CURVE 49</u>																	
<u>CURVE 50</u>																	
<u>CURVE 51</u>																	
<u>CURVE 52</u>																	
<u>CURVE 53*</u>																	
<u>CURVE 54</u>																	
<u>CURVE 55*</u>																	
<u>CURVE 56 (cont.)</u>																	
<u>CURVE 57</u>																	
<u>CURVE 58</u>																	
<u>CURVE 59*</u>																	
<u>CURVE 60</u>																	
<u>CURVE 61*</u>																	
<u>CURVE 62 (cont.)</u>																	
<u>CURVE 63</u>																	
<u>CURVE 64</u>																	
<u>CURVE 65*</u>																	
<u>CURVE 66</u>																	
<u>CURVE 67</u>																	

\* Not shown on plot

DATA TABLE NO. 12 (continued)

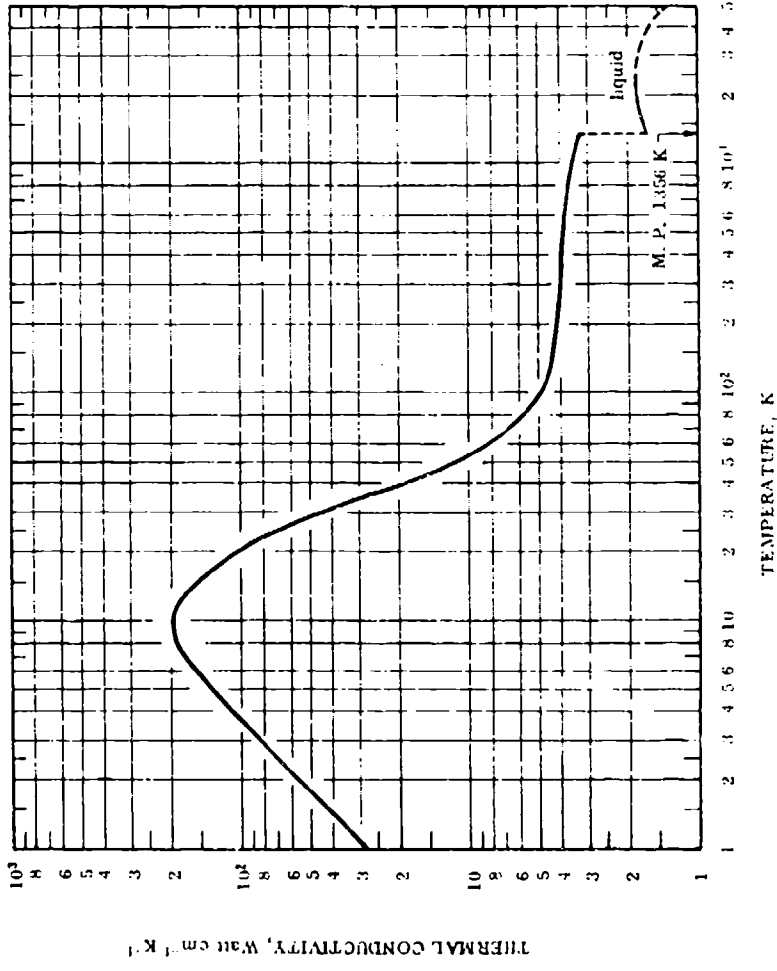
T	k	T	k	T	k	T	k	T	k	T	k	T	k	T	k	T	k	
291.2	3.90	21.19	64.0	1.5	1.15	90.2	4.64	357.2	1.56	763.7	3.59	363.2	3.78	23.0	10.3 <sup>7</sup>	320.1	4.03	
		<u>CURVE 79</u>		<u>CURVE 88</u>		196.0	4.01	357.2	1.56	1005	3.37	<u>CURVE 106*</u>		23.8	10.4*	362.6	4.03	
		<u>CURVE 80*</u>		2.0	1.25	273.0	3.93	358.2	1.56	1146	3.31	<u>CURVE 107*</u>		24.2	10.2*	407.7	4.02	
291.2	1.42	21.24	49.1	2.5	1.50			396.2	1.64	1278	3.17	363.2	3.74	26.0	10.9	443.0	4.02*	
		<u>CURVE 81</u>		3.0	2.10	<u>CURVE 91*</u>		397.2	1.64	<u>CURVE 108*</u>		<u>CURVE 109*</u>		27.3	11.3*	505.4	3.94	
273.2	3.91	21.24	42.3	4.0	2.55	78.2	5.50	443.2	1.72	363.2	3.78	<u>CURVE 110*</u>		29.5	11.5*	596.5	3.91	
		<u>CURVE 82</u>		7.0	5.15	273.2	3.97	444.2	1.72	23.8	10.4*	14.5	59.5	34.1	11.4*	684.3	3.88*	
21.17	86.1	21.30	39.0	8.0	5.70	<u>CURVE 92*</u>		495.2	1.82	24.2	10.2*	14.8	59.2	34.7	11.2*	722.6	3.87*	
		<u>CURVE 83*</u>		9.0	6.70	337.2	3.09	496.2	1.82	26.0	10.9	16.2	60.2	37.6	10.1	772.5	3.86*	
21.17	74.7	291.2	3.92	11.0	8.05	337.2	3.10	498.0	2.96	27.3	11.3*	17.1	58.1	37.6	10.1			
		<u>CURVE 84*</u>		12.0	8.80	385.2	3.18	498.0	2.96	29.5	11.5*	18.0	58.0	37.6	10.1			
21.18	65.4	354.2	3.01	15.0	10.0	386.2	3.17	498.0	2.96	31.3	11.3*	18.0	56.5	37.6	10.1			
		<u>CURVE 85*</u>		16.0	10.8	429.2	3.25	498.0	2.96	31.3	11.3*	18.0	56.5	37.6	10.1			
21.21	53.8	21.0	16.1	18.0	11.6	476.2	3.33	498.0	2.96	34.1	11.4*	18.0	56.5	37.6	10.1			
		<u>CURVE 86</u>		26.0	12.3	477.2	3.31	498.0	2.96	34.1	11.4*	18.0	56.5	37.6	10.1			
21.23	49.4	91.0	5.23	29.0	12.8	<u>CURVE 93*</u>		498.0	2.96	34.1	11.4*	18.0	56.5	37.6	10.1			
		<u>CURVE 87*</u>		34.0	12.1	337.2	2.41	498.0	2.96	34.1	11.4*	18.0	56.5	37.6	10.1			
		<u>CURVE 88</u>		37.0	11.8*	337.2	2.46	498.0	2.96	34.1	11.4*	18.0	56.5	37.6	10.1			
		<u>CURVE 89</u>		60.0	7.00*	337.2	2.47	498.0	2.96	34.1	11.4*	18.0	56.5	37.6	10.1			
		<u>CURVE 90*</u>		67.0	6.05*	337.2	2.53	498.0	2.96	34.1	11.4*	18.0	56.5	37.6	10.1			
		<u>CURVE 91*</u>		78.0	5.40*	337.2	2.54	498.0	2.96	34.1	11.4*	18.0	56.5	37.6	10.1			
		<u>CURVE 92*</u>		91.0	4.90*	337.2	2.56	498.0	2.96	34.1	11.4*	18.0	56.5	37.6	10.1			
		<u>CURVE 93*</u>		112.0	4.40*	337.2	2.62	498.0	2.96	34.1	11.4*	18.0	56.5	37.6	10.1			
		<u>CURVE 94</u>		129.0	4.35*	337.2	2.66	498.0	2.96	34.1	11.4*	18.0	56.5	37.6	10.1			
		<u>CURVE 95*</u>		142.0	4.35*	337.2	2.73	498.0	2.96	34.1	11.4*	18.0	56.5	37.6	10.1			
		<u>CURVE 96</u>		<u>CURVE 89</u>		390.2	2.54	498.0	2.96	34.1	11.4*	18.0	56.5	37.6	10.1			
		<u>CURVE 97*</u>		20.4	127.1	393.2	2.56	498.0	2.96	34.1	11.4*	18.0	56.5	37.6	10.1			
		<u>CURVE 98*</u>		20.6	124.6	436.2	2.62	498.0	2.96	34.1	11.4*	18.0	56.5	37.6	10.1			
		<u>CURVE 99*</u>		20.8	122.2	439.2	2.65	498.0	2.96	34.1	11.4*	18.0	56.5	37.6	10.1			
		<u>CURVE 100*</u>		21.1	120.5	442.2	2.66	498.0	2.96	34.1	11.4*	18.0	56.5	37.6	10.1			
		<u>CURVE 101*</u>		21.1	120.5	488.2	2.73	498.0	2.96	34.1	11.4*	18.0	56.5	37.6	10.1			
		<u>CURVE 102</u>		73.0	8.16	493.2	2.74	498.0	2.96	34.1	11.4*	18.0	56.5	37.6	10.1			
		<u>CURVE 103</u>		80.4	6.56	494.2	2.78	498.0	2.96	34.1	11.4*	18.0	56.5	37.6	10.1			
		<u>CURVE 104*</u>		83.0	5.95	494.2	2.78	498.0	2.96	34.1	11.4*	18.0	56.5	37.6	10.1			
		<u>CURVE 105*</u>		273.0	4.10	494.2	2.78	498.0	2.96	34.1	11.4*	18.0	56.5	37.6	10.1			
		<u>CURVE 106*</u>		<u>CURVE 94</u>		325.2	1.50	498.0	2.96	34.1	11.4*	18.0	56.5	37.6	10.1			
		<u>CURVE 107*</u>		325.2	1.50	325.2	1.51	498.0	2.96	34.1	11.4*	18.0	56.5	37.6	10.1			
		<u>CURVE 108*</u>		341.2	1.53	341.2	1.53	498.0	2.96	34.1	11.4*	18.0	56.5	37.6	10.1			
		<u>CURVE 109*</u>		22.0	13.1	341.2	1.53	498.0	2.96	34.1	11.4*	18.0	56.5	37.6	10.1			
		<u>CURVE 110*</u>		85.0	4.72	341.2	1.53	498.0	2.96	34.1	11.4*	18.0	56.5	37.6	10.1			
		<u>CURVE 111</u>				341.2	1.53	498.0	2.96	34.1	11.4*	18.0	56.5	37.6	10.1			
		<u>CURVE 112</u>				341.2	1.53	498.0	2.96	34.1	11.4*	18.0	56.5	37.6	10.1			

\* Not shown on plot





FIGURE AND TABLE NO. 12R RECOMMENDED THERMAL CONDUCTIVITY OF COPPER



REMARKS

The recommended values are for well-annealed 99.999% pure copper with residual electrical resistivity  $\rho_0 = 0.000551 \mu\Omega$  cm (characterization by  $\rho_0$  becomes important at temperatures below about 200 K). The values below  $1.5 \times 10^3$  m are calculated to fit the experimental data by using  $n = 2.40$ ,  $a = 0.19$ ,  $m = 2.59$ ,  $\alpha' = 4.16 \times 10^{-4}$ , and  $\beta = 0.0348$ . The recommended values that are supported by experimental thermal conductivity data are thought to be accurate to within 3% of the true values near room temperature, and 3 to 5% at other temperatures.

$T_1$	$k_1$	$k_2$	$T_2$	$T_1$	$k_1$	$k_2$	$T_2$
0	0	0	-459.7	1000	3.57	206	1340
1	28.7	1660	-457.9	1100	3.50	202	1520
2	57.3	3310	-456.1	1200	3.42	198	1700
3	85.5	4940	-454.3	1300	3.34	193	1880
4	113	6530	-452.5	1356	3.30	191	1981
5	138	7970	-450.7				
6	159	9190	-448.9				
7	177	10200	-447.1	1356	1.66	95.9	1981
8	189	10900	-445.3	1400	1.67	96.5	2060
9	195	11300	-443.5				
10	196	11300	-441.7	1500	1.71	98.8	2240
11	193	11200	-439.9	1600	1.74	101	2420
12	185	10700	-438.1	1700	1.77	102	2600
13	176	10200	-436.3	1800	1.79	103	2790
14	166	9590	-434.5	1900	1.81	105	2960
15	156	9010	-432.7	2000	1.82	105	3140
16	145	8390	-430.9	2200	1.84	106	3500
18	124	7160	-427.3	2400	1.84	106	3860
20	105	6070	-423.7	2600	(1.84)*	(106)	4200
25	68	3930	-414.7	2800	(1.83)	(106)	4580
30	43	2480	-405.7	3000	(1.80)	(104)	4940
35	29	1690	-396.7	3200	(1.78)	(103)	5300
40	20.5	1180	-387.7	3400	(1.74)	(101)	5660
45	15.3	884	-378.7	3600	(1.70)	(98.2)	6020
50	12.2	705	-369.7	3800	(1.66)	(95.9)	6380
60	8.5	491	-351.7	4000	(1.61)	(93.0)	6740
70	6.7	387	-333.7	4500	(1.48)	(85.5)	7640
80	5.7	329	-315.7	5000	(1.33)	(76.8)	8540
90	5.14	297	-297.7	5500	(1.17)	(67.6)	9440
100	4.83	279	-279.7	6000	(0.989)	(57.1)	10340
150	4.28	247	-189.7	7000	(0.611)	(35.3)	12140
200	4.13	239	-99.7	8000	(0.212)	(12.2)	13940
250	4.04	233	-9.7	8500	(0.036)	(2.08)	14840
273.2	4.01	232	32.0				
300	3.98	230	80.3				
350	3.94	228	170.3				
400	3.92	226	260.3				
500	3.88	224	440.3				
600	3.83	221	620.3				
700	3.77	218	800.3				
800	3.71	214	980.3				
900	3.64	210	1160				

\*  $T_1$  in K,  $k_1$  in Watt cm<sup>-1</sup>K<sup>-1</sup>,  $T_2$  in F, and  $k_2$  in Btu hr<sup>-1</sup>F<sup>-1</sup>. † Values in parentheses are estimated.

# THERMAL CONDUCTIVITY OF DYSPROSIUM

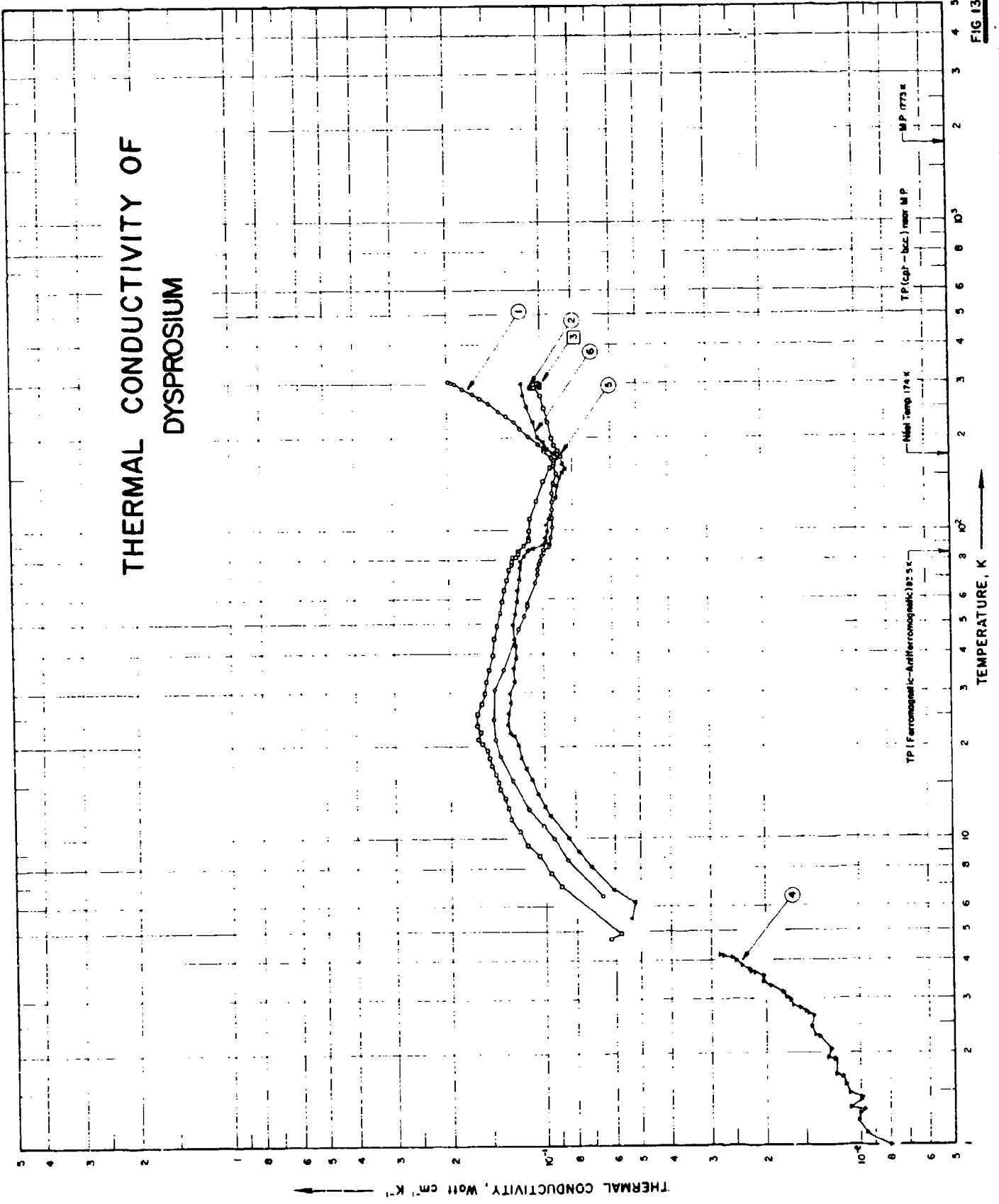


FIG 13

SPECIFICATION TABLE NO. 13 THERMAL CONDUCTIVITY OF DYSPROSIUM

(Impurity < 0.20% each; total impurities < 0.50%)

[For Data Reported in Figure and Table No. 13]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
1	807	L	1964	6.5-306	2		0.2 Ta, 0.1 Tb, 0.05 Ca, 0.05 Ho, 0.02 Er, 0.02 Si, 0.02 Y, 0.01 Fe, 0.01 Mg and traces of Cu and La; polycrystalline; 0.476 cm dia, 5 cm long; supplied by St. Elol Corp.; electrical resistivity 9.55 $\mu\text{ohm cm}$ at 4.2 K; measured in a vacuum of $\sim 6 \times 10^{-6}$ mm Hg; T. P. (ferromagnetic - antiferromagnetic) 85 K; Néel temperature 180 K.
2	777	C	1965	291.2	3		High purity; polycrystalline; 0.25 in. long, 0.25 in. dia; supplied by Johnson Matthey and Co. Ltd.; electrical resistivity 105 $\mu\text{ohm cm}$ at 18 C; measurements made using two different therm. comparators; Monel metal used as comparative material.
3	811		1954	301.2	10		No information given.
4	261	L	1967	1.0-4.2	1		Pure polycrystalline specimen 1.5 mm in dia supplied by Johnson Matthey and Co.
5	4	L	1968	4.7-300	6		0.0500 Er, 0.0500 Tb, 0.0400 Ta, < 0.0200 Cd, 0.0157 O, 0.0100 Fe, < 0.0100 Ho, < 0.0100 Si, < 0.0050 Al, < 0.0050 Cr, 0.0029 H, 0.0020 Ca, 0.0010 Mg, 0.0010 N, and < 0.0010 Y; single crystal; 9.48 x 2.30 x 2.12 mm; grown from arc-melted buttons using the strain anneal method; < 1120> direction (a-axis) along the specimen axis; electrical resistivity reported as 4.60, 4.60, 4.62, 4.74, 5.47, 5.32, 8.20, 13.4, 20.7, 31.2, 35.1, 36.5, 38.1, 42.3, 52.7, 71.8, 83.2, 88.4, 93.4, 94.9, 99.4, and 111.5 $\mu\text{ohm cm}$ at 4.2, 6.9, 9.0, 12.0, 18.0, 22.1, 28.2, 40.1, 55.3, 75.0, 83.5, 86.0, 88.9, 94.4, 114.0, 143.8, 160.2, 167.6, 178.3, 188.6, 213.1, and 299.4 K, respectively; electrical resistivity ratio $\rho(300 \text{ K})/\rho(4.2 \text{ K}) = 24.2$ ; residual electrical resistivity 4.59 $\mu\text{ohm cm}$ ; Lorenz function reported as 5.80, 4.99, 4.54, 4.76, 4.83, 5.14, 5.32, 5.16, 4.99, 4.97, 5.01, 4.96, 4.85, 4.64, 4.64, 4.26, 3.96, and 3.83 x $10^{-4} \text{V}^2 \text{K}^{-2}$ at 6.9, 11.6, 17.8, 25.7, 33.8, 45.5, 61.1, 78.8, 87.5, 94.2, 115.9, 146.2, 163.4, 175.6, 180.8, 227.8, 273.8, and 300.0 K, respectively; heat flow along the a-axis.
6	4	L	1968	5.8-300	6		Single crystal; 12.79 x 2.21 x 2.19 mm; grown from arc-melted buttons using the strain anneal method; <0001> direction (c-axis) along the specimen axis; electrical resistivity reported as 5.79, 5.79, 5.88, 6.32, 7.01, 8.83, 11.7, 17.3, 26.7, 31.8, 33.8, 35.0, 36.1, 41.4, 42.0, 47.2, 63.1, 83.1, 83.2, 83.0, 77.0, 70.6, 70.3, 70.3, 70.9, 73.7, and 77.2 $\mu\text{ohm cm}$ at 4.2, 8.0, 11.1, 16.0, 20.2, 26.0, 35.2, 49.6, 71.0, 81.1, 84.9, 87.0, 88.1, 88.9, 89.9, 98.1, 124.0, 150.1, 160.2, 163.2, 165.9, 174.3, 186.6, 194.6, 199.9, 219.1, 259.8, and 299.4 K, respectively; electrical resistivity ratio $\rho(300 \text{ K})/\rho(4.2 \text{ K}) = 13.4$ ; residual electrical resistivity 5.77 $\mu\text{ohm cm}$ ; Lorenz function reported as 5.26, 4.53, 4.24, 4.28, 4.15, 4.46, 4.58, 4.56, 4.79, 4.64, 4.75, 4.72, 4.42, 3.91, 3.80, 3.59, 3.61, 3.37, 3.16, and 3.03 x $10^{-4} \text{V}^2 \text{K}^{-2}$ at 8.0, 13.7, 19.5, 23.9, 33.9, 58.7, 78.7, 86.1, 88.5, 91.4, 125.4, 141.5, 159.5, 174.9, 184.5, 192.9, 198.1, 228.3, 262.9, and 300.0 K, respectively; heat flow along the c-axis.

#Boys, D.W. and Legvold, S., Phys. Rev., 174, 377-84, 1968; also USABC IS-T-185, 1967.

DATA TABLE NO. 13 THERMAL CONDUCTIVITY OF DYSPROSIUM

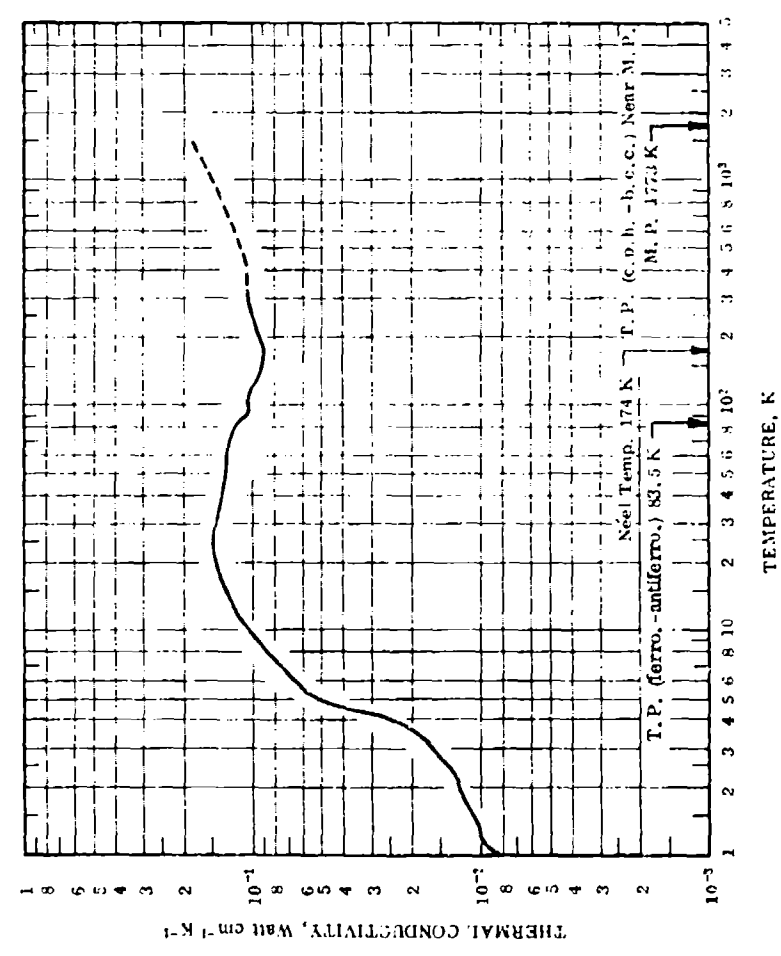
(Impurity &lt; 0.20% each; total impurities &lt; 0.50%)

(Temperature, T, K; Thermal Conductivity, k, Wat cm<sup>-1</sup> K<sup>-1</sup>)

CURVE 1			CURVE 4 (cont.)			CURVE 5 (cont.)			CURVE 6 (cont.)		
T	k		T	k		T	k		T	k	
6.5	0.066		2.638	0.01440		17.9	0.148		200.0	0.093	
8.5	0.085		2.634	0.01462		18.9	0.151		225.0	0.095	
10.0	0.094		2.742	0.01499		19.9	0.153		249.9	0.099	
11.0	0.101		2.781	0.01569		20.9	0.159		275.1	0.101	
12.5	0.113		2.869	0.01641		21.7	0.163		299.5	0.103	
15.5	0.126		2.970	0.01673		22.6	0.161		CURVE 6		
18.5	0.139		3.045	0.01725		24.0	0.165		5.8	0.053	
21.0	0.144		3.161	0.01762		25.9	0.164		1.0	0.052	
24.5	0.146		3.165	0.01793*		28.0	0.159		6.9	0.060	
30.5	0.145		3.165	0.01810*		29.9	0.157		7.9	0.071	
35.5	0.135		3.299	0.01927		33.0	0.154		9.0	0.078	
48.0	0.120		3.411	0.02042		36.2	0.150		140.2	0.090	
53.0	0.115		3.463	0.02004*		39.8	0.147		149.9	0.088	
57.0	0.112		3.577	0.02029		45.2	0.146		155.0	0.086	
58.0	0.112		3.577	0.02077*		50.0	0.142		160.2	0.084	
68.0	0.106		3.652	0.02162		55.0	0.138		165.0	0.085	
72.0	0.104		3.692	0.02232		59.9	0.136		170.0	0.086	
75.5	0.104		3.762	0.02238*		65.0	0.134		172.5	0.088	
78.5	0.103		3.776	0.02256		69.9	0.131		175.0	0.088*	
80.5	0.102		3.861	0.02385		75.2	0.128		177.5	0.091	
83.5	0.101		3.950	0.02397*		78.5	0.126		180.1	0.096	
85.5	0.099		4.016	0.02484		80.0	0.126		182.5	0.096*	
86.5	0.099*		4.097	0.02561		82.5	0.125		184.9	0.099	
87.5	0.099		4.125	0.02641*		82.5	0.121		187.6	0.099	
89.5	0.095		4.136	0.02737		85.0	0.119		189.9	0.098	
96.3	0.095		4.218	0.02791		86.5	0.119		195.0	0.099	
90.5	0.095*		CURVE 5			90.0	0.115		200.0	0.103	
95.0	0.094		4.7	0.062		92.8	0.111		225.1	0.107	
103.0	0.093		4.9	0.059		95.0	0.111		249.8	0.112	
110.5	0.093		1.146	0.00959		100.1	6.110		275.2	0.114	
117.5	0.093		1.216	0.01031		110.2	0.110		299.8	0.116	
124.5	0.093		1.277	0.01007		125.4	0.104				
133.5	0.093		1.308	0.00979		125.4	0.104				
143.5	0.092		1.406	0.01005		145.2	0.099				
153.0	0.090		1.432	0.00991		160.4	0.094				
162.0	0.092		1.476	0.01092		165.3	0.092				
172.0	0.093		1.576	0.01139		170.5	0.092				
181.5	0.098		1.673	0.01158		175.0	0.088				
182.5	0.099*		1.705	0.01194		177.8	0.089				
203.0	0.110		1.914	0.01210		180.4	0.090				
214.5	0.117*		1.937	0.01265		182.4	0.069				
			2.057	0.01236		185.1	0.090				
			2.270	0.01343		190.1	0.091				
			2.297	0.01388							
			2.437	0.01438							
			2.466	0.01448*							

\*Not shown on plot

FIGURE AND TABLE NO. 13R RECOMMENDED THERMAL CONDUCTIVITY OF DYSPROSIUM



RECOMMENDED VALUES\*  
(For Polycrystalline)

T <sub>1</sub>	k <sub>1</sub>	k <sub>2</sub>	T <sub>2</sub>	T <sub>1</sub>	k <sub>1</sub>	k <sub>2</sub>	T <sub>2</sub>
0	0	0	-459.7	60	0.130	7.51	-351.7
1	0.00834	0.482	-457.9	70	0.127	7.34	-333.7
1.10	0.00834	0.540	-457.69	80	0.122	7.05	-315.7
1.20	0.0101	0.584	-457.51	85	0.117	6.76	-306.7
1.25	0.0102	0.589	-457.42	90	0.109	6.30	-297.7
1.30	0.0102	0.589	-457.33	95	0.105	6.07	-288.7
1.50	0.0109	0.630	-456.97	100	0.105	6.07	-279.7
1.70	0.0117	0.676	-456.61	105	0.105	6.07	-270.7
2	0.0125	0.722	-456.3	110	0.105	6.07	-261.7
2.20	0.0129	0.745	-455.71	120	0.102	5.84	-243.7
2.50	0.0140	0.809	-455.17	150	0.0838	5.42	-189.7
3	0.0166	0.959	-454.3	170	0.0886	5.12	-153.7
4	0.0249	1.44	-452.5	200	0.0959	5.51	-89.7
4.30	0.0302	1.74	-451.93	250	0.102	5.89	9.7
4.40	0.0335	1.94	-451.75	273.2	0.105	6.07	32.0
4.50	0.0387	2.24	-451.57	300	0.107	6.18	80.3
4.70	0.0448	2.59	-451.21	350	(0.108) <sup>‡</sup>	(6.24)	170.3
5	0.0518	2.99	-450.7	400	(0.109)	(6.30)	260.3
6	0.0667	3.85	-448.9	500	(0.114)	(6.59)	440.3
7	0.0781	4.51	-447.1	600	(0.122)	(7.05)	620.3
8	0.0982	5.10	-445.3	700	(0.129)	(7.45)	800.3
9	0.0970	5.60	-443.5	800	(0.137)	(7.92)	980.3
10	0.105	6.07	-441.7	900	(0.144)	(8.32)	1160
11	0.112	6.47	-439.9	1000	(0.152)	(8.78)	1340
12	0.118	6.82	-438.1	1100	(0.159)	(9.19)	1520
13	0.123	7.11	-436.3	1200	(0.167)	(9.65)	1700
14	0.127	7.34	-434.5	1300	(0.174)	(10.1)	1880
15	0.131	7.57	-432.7	1400	(0.181)	(10.5)	2060
16	0.135	7.80	-430.9	1500	(0.188)	(10.9)	2240
18	0.141	8.15	-427.3				
20	0.146	8.44	-423.7				
25	0.151	8.72	-414.7				
30	0.147	8.49	-405.7				
35	0.142	8.20	-396.7				
40	0.139	8.03	-387.7				
45	0.137	7.92	-378.7				
50	0.135	7.80	-369.7				

REMARKS

The recommended values are for well-annealed 99.99% pure dysprosium with residual electrical resistivity  $\rho_0 = 4.7 \mu\Omega \text{ cm}$  (characterization by  $\rho_0$  becomes important at temperatures below about 200 K). The values that are supported by experimental thermal conductivity data are thought to be accurate to within 3% of the true values near room temperature and 5 to 15% at other temperatures.

\* T<sub>1</sub> in K, k<sub>1</sub> in Watt cm<sup>-1</sup> K<sup>-1</sup>, T<sub>2</sub> in F, and k<sub>2</sub> in Btu lb<sup>-1</sup> ft<sup>-1</sup> F<sup>-1</sup>. † Values in parentheses are estimated.

# THERMAL CONDUCTIVITY OF ERBIUM

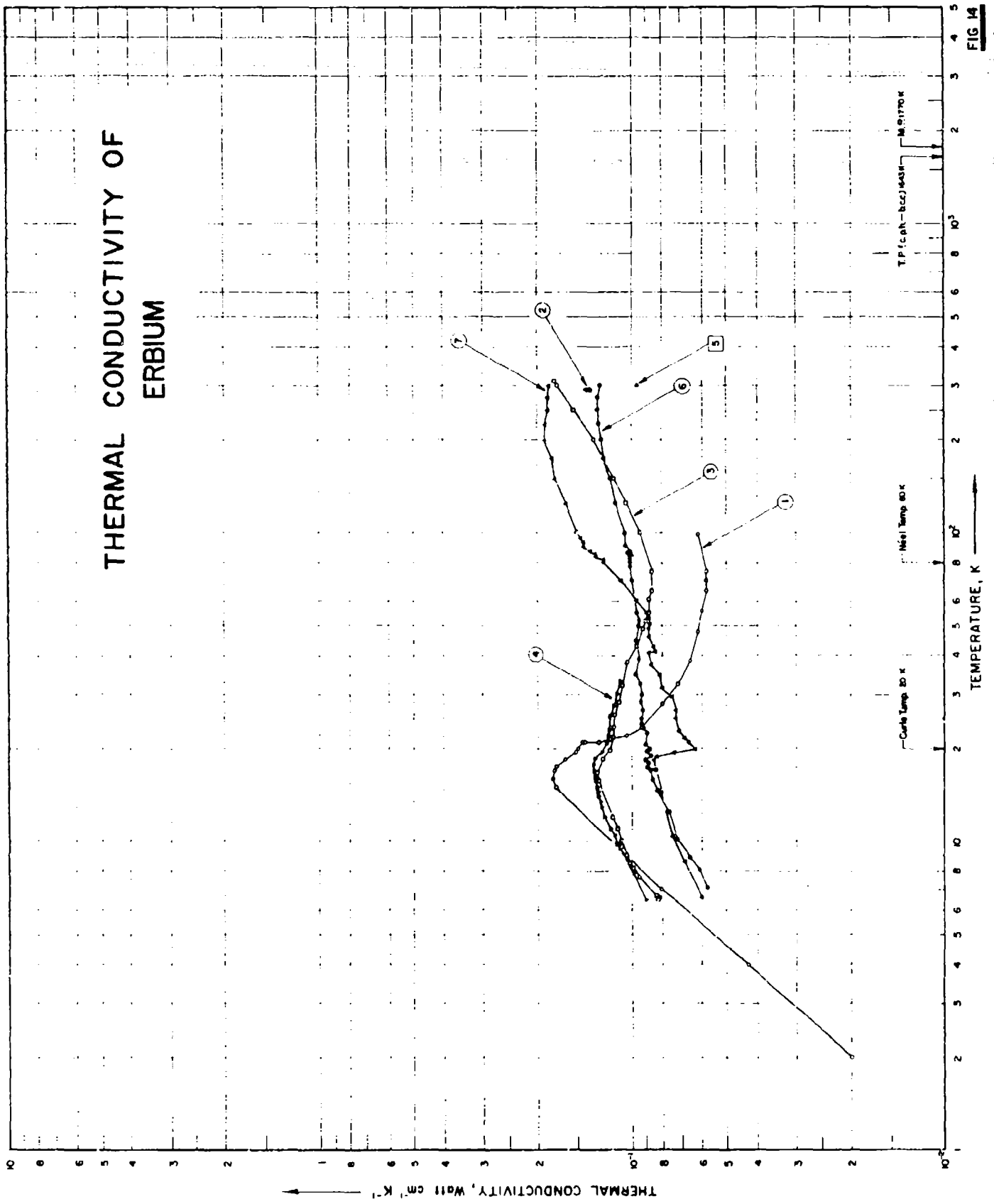


FIG 14

## SPECIFICATION TABLE NO. 14 THERMAL CONDUCTIVITY OF ERBIUM

(Impurity &lt; 0.20% each; total impurities &lt; 0.50%)

i For Data Reported in Figure and Table No. 14j

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
1	322, 808	L	1965	2-99			99.9 pure; polycrystalline; 3 x 0.2 x 0.025 cm; annealed at 850 C. for 2 hr in vacuum of 10 <sup>-4</sup> mm Hg; electrical resistivity 8.44 $\mu\text{ohm cm}$ ; electrical resistivity ratio $\rho(298 \text{ K})/\rho(4.2 \text{ K}) = 10.2$ ; T. P. (ferromagnetic - antiferromagnetic) 20 K; Neel temperature 80 K; data taken from smoothed curve.
2	777	C	1965	231.2	3		High purity; polycrystalline; 0.25 in. dia. 0.25 in. thick; supplied by Johnson Matthey and Co. Ltd.; electrical resistivity 79 $\mu\text{ohm cm}$ at 18 C; measurements made using 2 different thermal comparators; Monel used as comparative material.
3	809	L	1965	6.5-310			0.01 Ca, 0.02 Ho, 0.005 Mg, 0.07 O, 0.01 Si, and trace Tm; polycrystalline; 0.476 cm dia, 6 cm long; supplied by Research Chemicals; arc-melted, machined, swaged, and annealed in vacuum ( $\sim 10^{-5}$ torr) at 900 K for 50 hrs; T. P. (ferromagnetic - antiferromagnetic) 19 K; Neel temperature 86 K; data above 75 K extracted from smooth curve; electrical resistivity 3.79 $\mu\text{ohm cm}$ at 4.16 K; measured in vacuum.
4	809	L	1965	6.5-33			The second run of the above specimen; measured during cooling.
5	811		1954	301.2	10		No information given.
6	†	L	1963	6.9-300	6		0.0500 Ta, 0.0235 O, 0.0260 Mg, < 0.0200 Ca, 0.0150 Cr, 0.0150 Fe, < 0.0100 Dy, < 0.0100 Ho, < 0.0050 Si, < 0.0030 Y, 0.0017 H, 0.0011 N, < 0.0010 Tm, < 0.0010 Yb, and traces of Cu and W; single crystal; 5.81 x 1.88 x 1.86 mm; grown from arc-melted buttons using the strain anneal method; < 10 <sup>10</sup> > direction (b-axis) along the specimen axis; electrical resistivity ratio $\rho(300 \text{ K})/\rho(4.2 \text{ K}) = 17.4$ ; Lorenz function reported as: 4.42, 3.91, 3.71, 3.99, 4.52, 4.90, 5.17, 5.24, 5.08, 4.73, 4.42, 3.94, and 3.75 x 10 <sup>-8</sup> V <sup>2</sup> /K <sup>2</sup> at 6.4, 10.3, 15.9, 22.0, 31.3, 43.8, 65.1, 80.0, 101.8, 126.4, 179.2, 260.1, and 295.9 K, respectively; heat flow along the b-axis.
7	†	L	1968	4.9-300	6		0.0900 Fe, < 0.0500 Ta, 0.0280 O, 0.0200 Ca, 0.0200 Cr, 0.0200 Mg, 0.0130 Y, < 0.0100 Dy, < 0.0100 Ho, < 0.0050 Si, 0.0014 H, < 0.0010 Tm, < 0.0010 Yb, 0.0068 N, and traces of Cu, Ni, and W; single crystal; 6.13 x 1.51 x 1.31 mm; grown from arc-melted buttons using the strain anneal method; < 0001> direction (c-axis) along the specimen axis; electrical resistivity ratio $\rho(300 \text{ K})/\rho(4.2 \text{ K}) = 10.1$ ; Lorenz function reported as 4.45, 4.03, 3.88, 3.82, 3.88, 4.68, 4.68, 5.30, 5.76, 6.48, 6.93, 5.31, 4.20, 3.83, 3.32, and 2.90 x 10 <sup>-8</sup> V <sup>2</sup> /K <sup>2</sup> at 4.2, 8.5, 11.6, 15.2, 19.4, 19.8, 32.1, 43.9, 53.0, 64.5, 73.3, 81.8, 131.6, 220.0, and 300.0 K, respectively; heat flow along the c-axis.

†Boys, D. W. and Legvold, S., Phys. Rev., 174, 377-84, 1968; also USAEC IS-T-185, 1967.

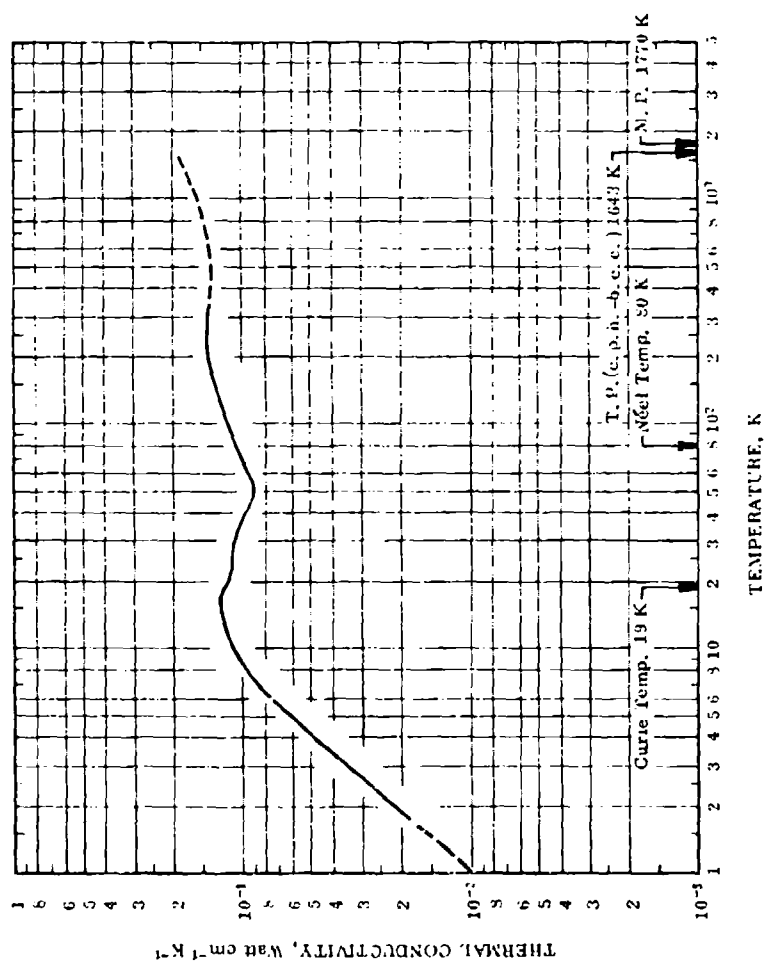




FIGURE AND TABLE NO. 14R RECOMMENDED THERMAL CONDUCTIVITY OF ERBIUM

RECOMMENDED VALUES\*  
(For Polycrystalline)

$T_1$	$k_1$	$k_2$	$T_2$	$T_1$	$k_1$	$k_2$	$T_2$
0	0	0	-459.7	40	0.101	5.84	-387.7
1	0.0106*	0.612	-457.9	45	0.0948	5.48	-378.7
2	0.0227	1.31	-456.1	50	0.0938	5.36	-369.7
3	0.0356	2.06	-454.3	55	0.0935	5.40	-360.7
4	0.0488	2.82	-452.5	60	0.0967	5.59	-351.7
5	0.0625	3.61	-450.7	70	0.104	6.01	-333.7
6	0.0765	4.42	-448.9	80	0.110	6.36	-315.7
7	0.0890	5.14	-447.1	90	0.115	6.64	-297.7
8	0.0989	5.71	-445.3	100	0.119	6.88	-279.7
9	0.107	6.18	-443.5	150	0.134	7.74	-189.7
10	0.112	6.47	-441.7	200	0.143	8.26	-99.7
11	0.117	6.76	-439.9	250	0.145	8.38	-9.7
12	0.120	6.93	-438.1	273.2	0.144	8.32	32.0
13	0.123	7.11	-436.3	300	0.143	8.26	80.3
14	0.125	7.22	-434.5	350	0.140	8.09	170.3
15	0.126	7.40	-432.7	400	0.139	8.03	260.3
16	0.130	7.51	-430.9	500	0.140	8.09	4440.3
18	0.129	7.45	-427.3	600	0.143	8.26	620.3
20	0.120	6.93	-423.7	700	0.146	8.44	900.3
21	0.118	6.82	-421.9	800	0.150	8.67	980.3
22	0.118	6.82	-420.1	900	0.154	8.90	1160
24	0.117	6.76	-416.5	1000	0.158	9.13	1340
25	0.116	6.70	-414.7	1100	0.163	9.42	1520
30	0.111	6.41	-405.7	1200	0.170	9.82	1700
35	0.107	6.18	-396.7	1300	0.177	10.2	1880
				1400	0.183	10.6	2060
				1500	0.190	11.0	2240



REMARKS

The recommended values are for well-annealed 99.99% pure erbium with residual electrical resistivity  $\rho_0 = 3.79 \mu\text{hm cm}$  (characterization by  $\rho_0$  becomes important at temperatures below about 200 K). The values that are supported by experimental thermal conductivity data are thought to be accurate to within 5% of the true values near room temperature and 5 to 15% at other temperatures.

\*  $T_1$  in K,  $k_1$  in Watt cm<sup>-1</sup> K<sup>-1</sup>,  $T_2$  in F, and  $k_2$  in Btu hr<sup>-1</sup> ft<sup>-1</sup> F<sup>-1</sup>. † Values in parentheses are extrapolated or estimated.

# THERMAL CONDUCTIVITY OF EUROPIUM

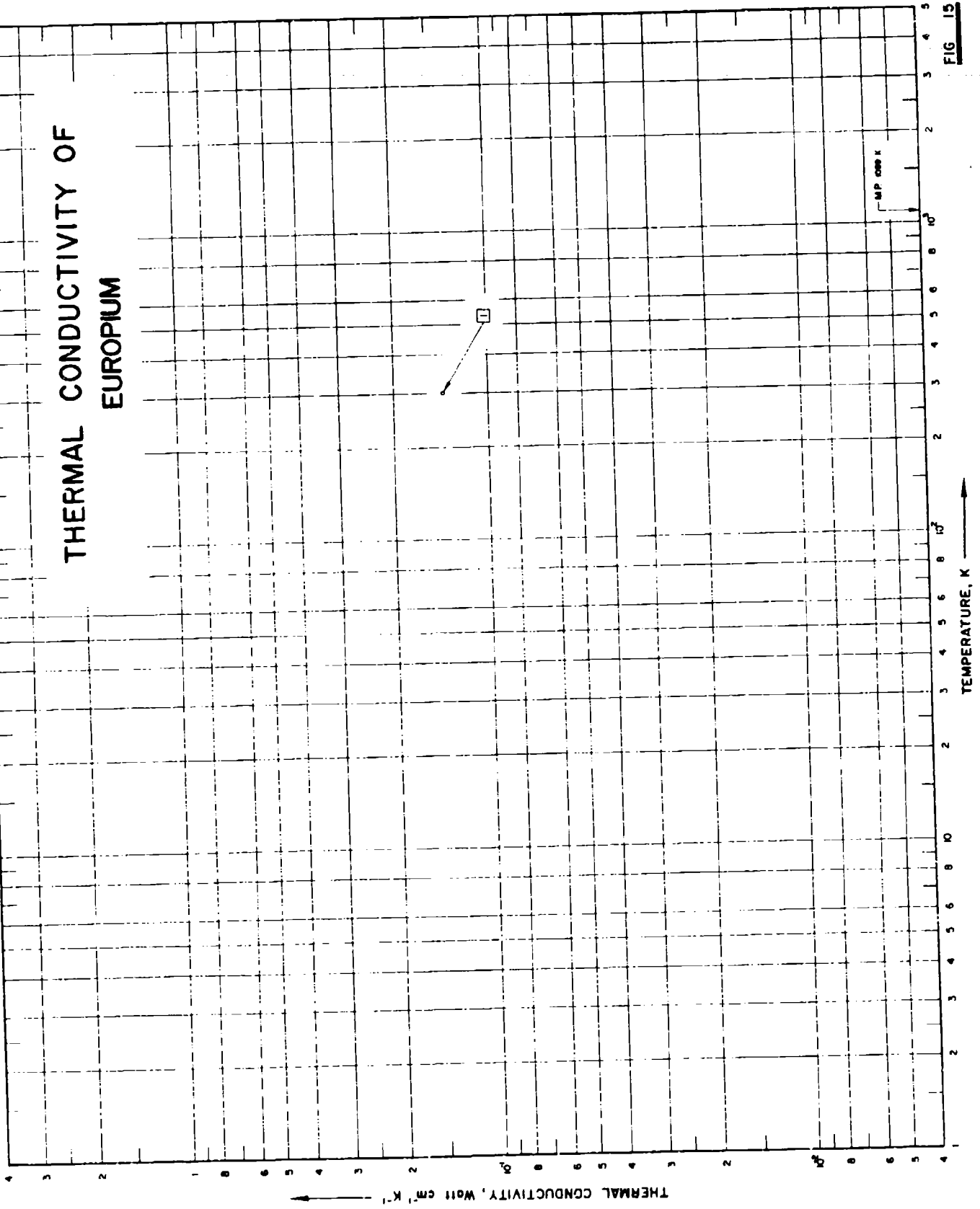


FIG 15

## SPECIFICATION TABLE NO. 15 THERMAL CONDUCTIVITY OF EUROPIUM

(Impurity &lt; 0.20% each; total impurities &lt; 0.50%)

[ For Data Reported in Figure and Table No. 15 ]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition ( weight percent), Specifications and Remarks
1	256	-	1966	300			Predicted value calculated from electrical resistivity value averaged from data of Spedding, F. H., et al. (Trans. ADME, 212, 379, 1958) and Colvin, R. V., et al. (Phys. Rev. 120, 741, 1960), and the Lorenz number $4.29 \times 10^{-8} \text{ V}^2/\text{K}^2$ based on the smoothed curve of Lorenz number vs. atomic number given by the authors.

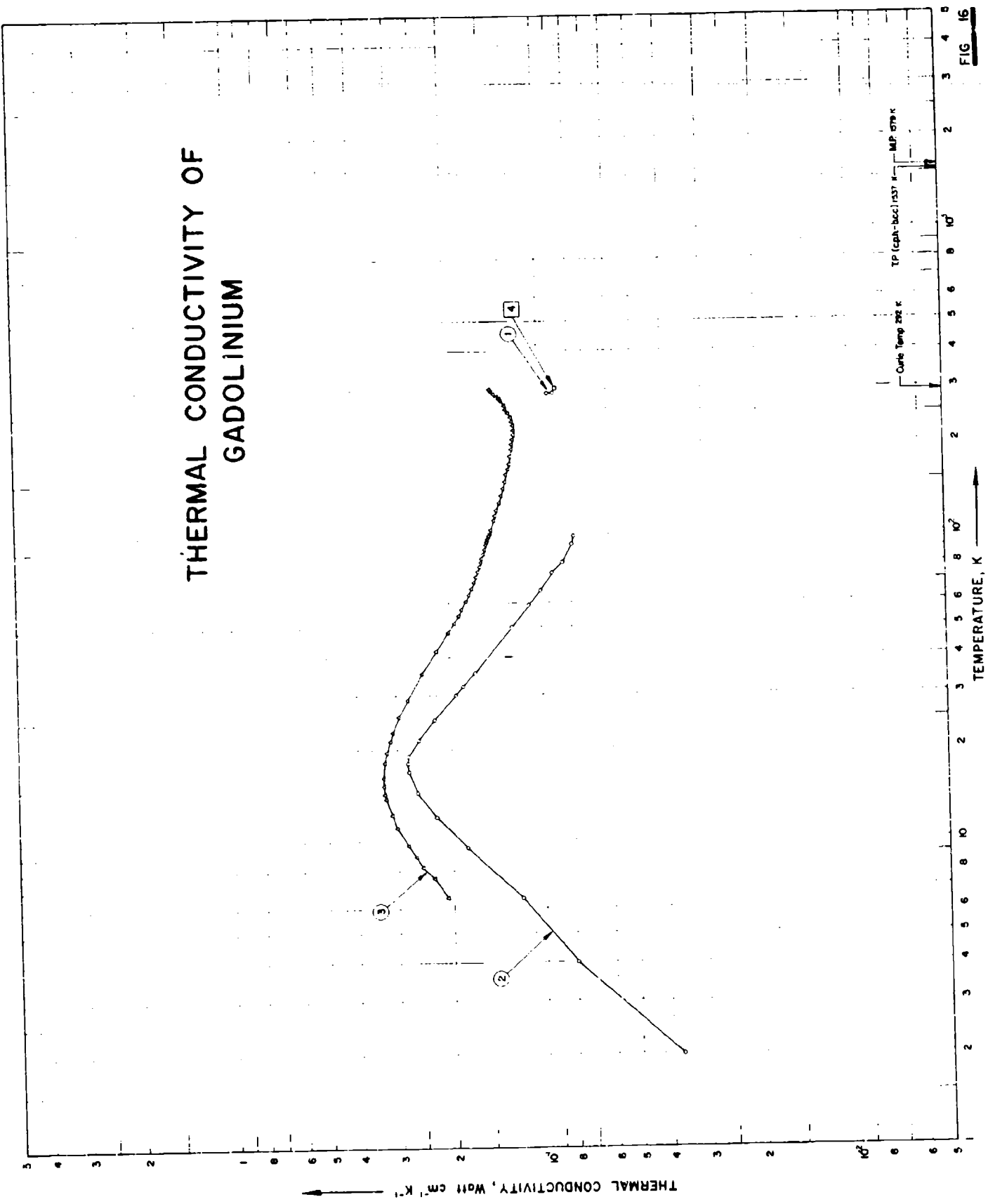
## DATA TABLE NO. 15 THERMAL CONDUCTIVITY OF EUROPIUM

(Impurity &lt; 0.20% each; total impurities &lt; 0.50%)

[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup>K<sup>-1</sup>]

T	k
<u>CURVE 1</u>	
300	0.14

# THERMAL CONDUCTIVITY OF GADOLINIUM



## SPECIFICATION TABLE NO. 16 THERMAL CONDUCTIVITY OF GADOLINIUM

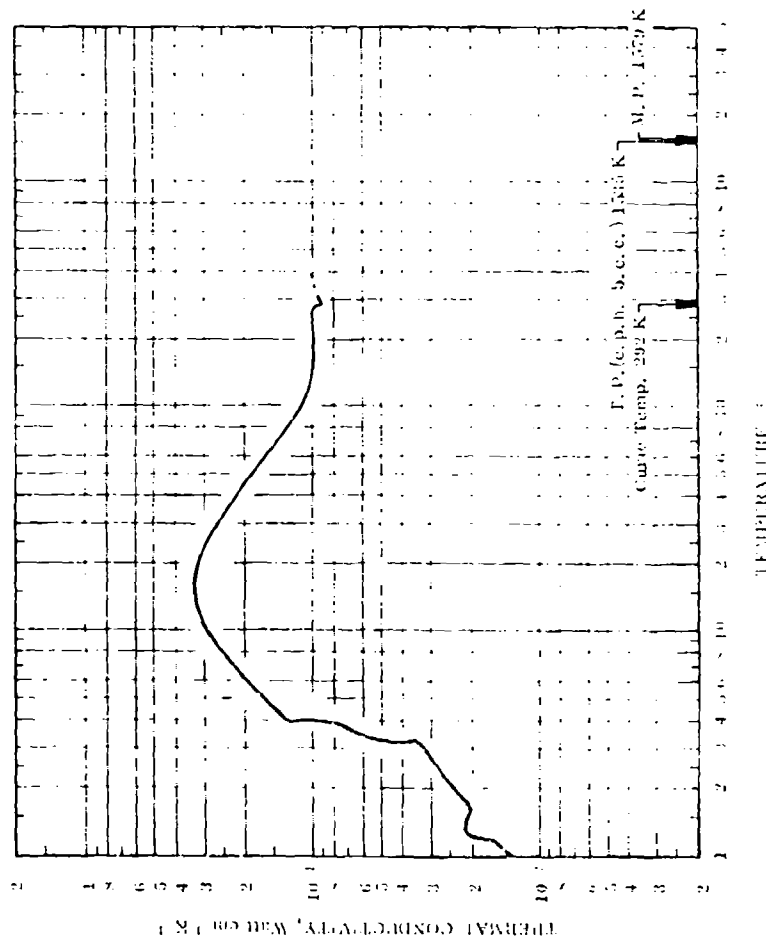
(impurity &lt; 0.20% each; total impurities &lt; 0.50%)

[For Data Reported in Figure and Table No. 16.]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
1	777	C	1965	291.2	± 3.0		High purity, polycrystalline, specimen 0.25 in. in diameter and 0.25 in. long, supplied by Johnson Matthey Co.; electrical resistivity reported at about 18 C as 134 $\mu$ ohm cm; Monel metal used as comparative material; measurements made using 2 different thermal comparators.
2	814	L	1966	2.0-99			99.9 pure, strip specimen 0.25 mm thick; baked for 1 1/2 hours at 650 C; measured in helium atmosphere; electrical resistivity reported at 4.2 K as 3.00 $\mu$ ohm cm; electrical resistivity ratio $\rho(293 K) / \rho(4.2 K) = 47.4$ .
3	815	L	1964	6.5-300			Polycrystalline gadolinium, measured in vacuum at about $6 \times 10^{-6}$ mm Hg; electrical resistivity reported at 4.18 K as 2.41 $\mu$ ohm cm; antiferromagnetic-paramagnetic transition occurred at $\sim 270$ K.
4	811		1954	301.2	10		No information given.



FIGURE AND TABLE NO. 16R RECOMMENDED THERMAL CONDUCTIVITY OF GADOLINIUM



## REMARKS:

The recommended values are for well-annealed high-purity gadolinium with residual electrical resistivity  $\rho_r = 2.41 \mu\Omega \text{ cm}$  (characterization by  $\rho_r$  becomes important at temperatures below about 200 K). The values that are supported by experimental thermal conductivity data are thought to be accurate to within 3% of the true values near room temperature and 5 to 15% at other temperatures.

RECOMMENDED VALUES<sup>a</sup>  
(For Polycrystalline)

$T_1$	$k_1$	$k_2$	$T_2$	$T_3$	$k_3$	$T_4$
0	0	0	-459.67	15	0.335	$T_4$
1	(0.6118) <sup>b</sup>	(0.797)	-457.47	16	0.335	-430.9
1.15	0.0133	0.307	-457.60	18	0.332	-427.3
1.20	0.0167	0.365	-457.51	20	0.325	-423.7
1.22	0.0201	1.16	-457.47	25	0.297	-414.7
1.25	0.0212	1.22	-457.42	30	0.267	-405.7
1.30	0.0216	1.25	-457.43	35	0.245	-396.7
1.50	0.0211	1.22	-456.97	40	0.224	-387.7
1.60	0.0207	1.20	-456.79	45	0.206	-378.7
1.70	0.0213	1.23	-456.61	50	0.190	-369.7
2	0.0219	1.38	-456.07	60	0.164	-351.7
3	0.0319	1.84	-454.27	70	0.145	-333.7
3.10	0.0325	1.88	-454.09	80	0.132	-315.7
3.20	0.0333	1.94	-454.91	90	0.122	-297.7
3.22	0.0339	1.96	-454.87	100	0.115	-279.7
3.25	0.0416	2.40	-453.82	150	0.101	-189.7
3.30	0.0492	2.84	-453.73	200	0.0998	-99.7
3.50	0.0641	3.70	-453.19	250	0.101	-9.7
3.90	0.0724	4.18	-452.83	273.2	0.101	32.0
3.90	0.0755	4.39	-452.65	280	0.0998	44.9
4	0.101	5.95	-452.47	285	0.0923	53.3
4.02	0.120	6.93	-452.43	290	0.0916	62.3
4.10	0.131	7.57	-452.29	295	0.0921	71.3
4.20	0.135	7.80	-452.11	300	0.0928	80.3
4.30	0.145	8.38	-451.57	330	(0.0980)	170.3
5	0.161	9.39	-450.77	400	(0.101)	260.3
6	0.193	11.2	-448.9			
7	0.223	12.9	-447.1			
8	0.249	14.4	-445.3			
9	0.272	15.7	-441.7			
10	0.290	16.8	-441.7			
11	0.309	17.6	-439.9			
12	0.317	18.3	-438.1			
13	0.325	18.8	-436.3			
14	0.331	19.1	-434.5			

<sup>a</sup>  $T_1$  in K,  $k_1$  in Watt cm<sup>-1</sup> K<sup>-1</sup>,  $T_2$  in °F, and  $k_2$  in Btu ft<sup>-1</sup> h<sup>-1</sup> F<sup>-1</sup>. <sup>b</sup> Values in parentheses are extrapolated.



# THERMAL CONDUCTIVITY OF GALLIUM

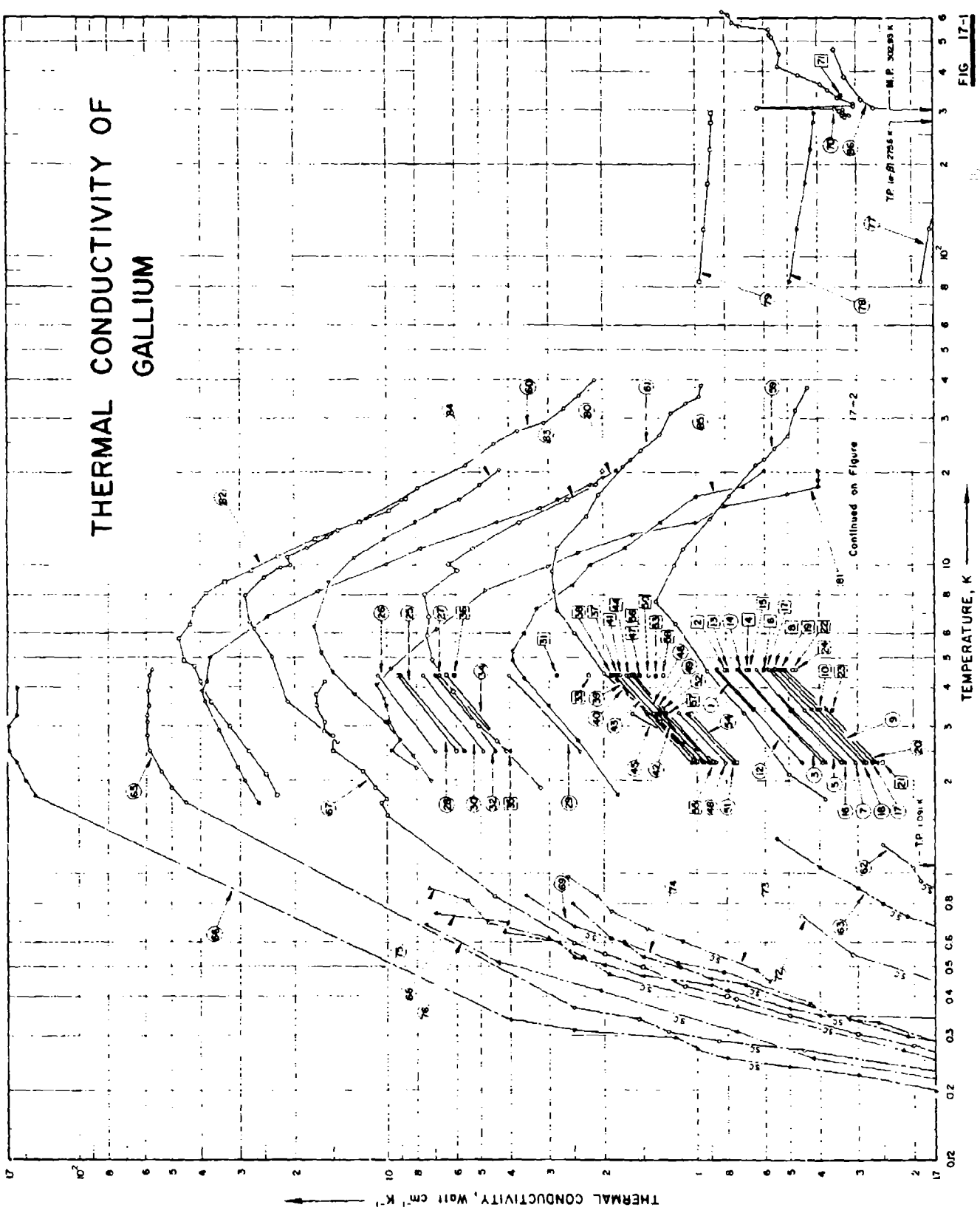
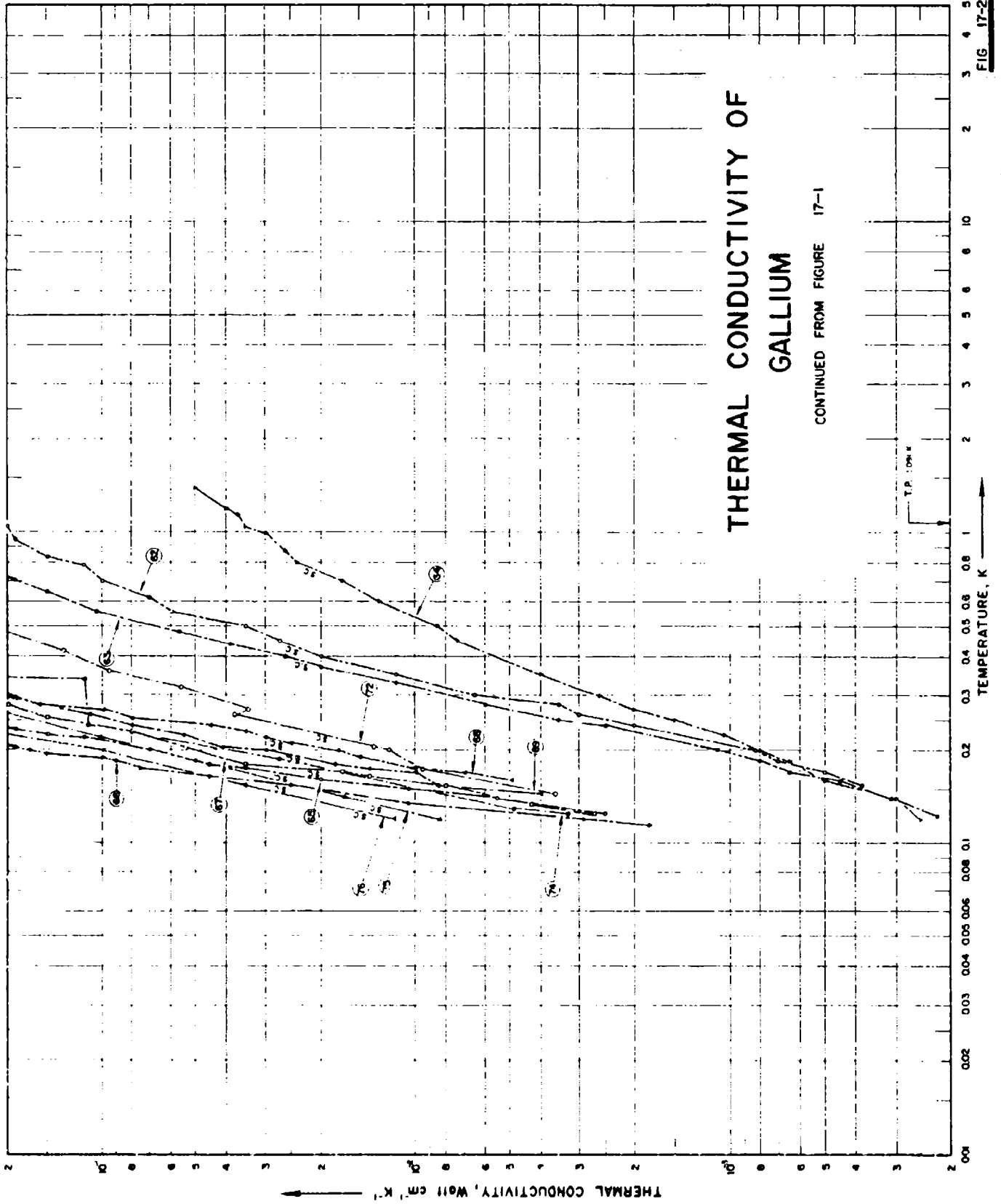


FIG 17-1

TEMPERATURE, K

THERMAL CONDUCTIVITY, Watt cm<sup>-1</sup> K<sup>-1</sup>



SPECIFICATION TABLE NO. 17 THERMAL CONDUCTIVITY OF GALLIUM

(Impurity - 0.20% each; total impurities - 0.50%)

(For Data Reported in Figure and Table No. 17.)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)	Specifications and Remarks
1	342	L	1953	2.3-4.6		Ga 42-1		Single crystal; 2.92 cm long, 0.223 cm dia; supplied by National Physical Lab; rod axis parallel to the high electrical resistance direction of the crystal; measured in a longitudinal field of 0.36 KOe (kiloborsted).
2	342	L	1953	4.6		Ga 42-1		The above specimen measured in a longitudinal field of 0.73 KOe.
3	342	L	1953	2.3-4.6		Ga 42-1		The above specimen measured in a longitudinal field of 1.08 KOe.
4	342	L	1953	4.6		Ga 42-1		The above specimen measured in a longitudinal field of 1.47 KOe.
5	342	L	1953	2.3-4.6		Ga 42-1		The above specimen measured in a longitudinal field of 1.81 KOe.
6	342	L	1953	4.6		Ga 42-1		The above specimen measured in a longitudinal field of 2.15 KOe.
7	342	L	1953	2.3-4.6		Ga 42-1		The above specimen measured in a longitudinal field of 2.51 KOe.
8	342	L	1953	4.6		Ga 42-1		The above specimen measured in a longitudinal field of 2.98 KOe.
9	342	L	1953	2.3-4.6		Ga 42-1		The above specimen measured in a longitudinal field of 3.24 KOe.
10	342	L	1953	3.4		Ga 42-1		The above specimen measured in a longitudinal field of 3.62 KOe.
11	342	L	1953	2.3-4.6		Ga 42-1		The above specimen measured in a longitudinal field of 3.65 KOe.
12	342	L	1953	2.3-4.6		Ga 42-1		The above specimen measured in a transverse field of 0.36 KOe.
13	342	L	1953	4.6		Ga 42-1		The above specimen measured in a transverse field of 0.73 KOe.
14	342	L	1953	2.3-4.6		Ga 42-1		The above specimen measured in a transverse field of 1.10 KOe.
15	342	L	1953	4.6		Ga 42-1		The above specimen measured in a transverse field of 1.42 KOe.
16	342	L	1953	2.3-4.6		Ga 42-1		The above specimen measured in a transverse field of 1.78 KOe.
17	342	L	1953	4.6		Ga 42-1		The above specimen measured in a transverse field of 2.17 KOe.
18	342	L	1953	2.3-4.6		Ga 42-1		The above specimen measured in a transverse field of 2.53 KOe.
19	342	L	1953	4.6		Ga 42-1		The above specimen measured in a transverse field of 2.90 KOe.
20	342	L	1953	2.3-4.6		Ga 42-1		The above specimen measured in a transverse field of 3.26 KOe.
21	342	L	1953	2.3		Ga 42-1		The above specimen measured in a transverse field of 3.54 KOe.
22	342	L	1953	4.6		Ga 42-1		The above specimen measured in a transverse field of 3.62 KOe.
23	342	L	1953	3.4		Ga 42-1		The above specimen measured in a transverse field of 3.66 KOe.
24	342	L	1953	4.6		Ga 42-1		The above specimen measured in a transverse field of 3.88 KOe.
25	342	L	1953	2.5-4.4		Ga 42-2		Single crystal; 2.45 cm long, 0.218 cm dia; supplied by National Physical Lab; rod axis parallel to the low electrical resistance direction; measured in a transverse field of 0.35 KOe.

SPECIFICATION TABLE NO. 17 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
26	342	L	1953	2.5, 4.4		Ca 42-2	The above specimen measured in a longitudinal field of 0.35 KOe.
27	342	L	1953	2.5, 4.4		Ca 42-2	The above specimen measured in a transverse field of 1.05 KOe.
28	342	L	1953	2.5, 4.4		Ca 42-2	The above specimen measured in a longitudinal field of 1.05 KOe.
29	342	L	1953	2.5, 4.4		Ca 42-2	The above specimen measured in a transverse field of 1.80 KOe.
30	342	L	1953	2.5, 4.4		Ca 42-2	The above specimen measured in a longitudinal field of 1.80 KOe.
31	342	L	1953	4.4		Ca 42-2	The above specimen measured in a transverse field of 2.55 KOe.
32	342	L	1953	2.5, 4.4		Ca 42-2	The above specimen measured in a longitudinal field of 2.55 KOe.
33	342	L	1953	4.4		Ca 42-2	The above specimen measured in a transverse field of 3.24 KOe.
34	342	L	1953	2.5, 4.4		Ca 42-2	The above specimen measured in a longitudinal field of 3.24 KOe.
35	342	L	1953	2.5		Ca 42-2	The above specimen measured in a longitudinal field of 3.76 KOe.
36	342	L	1953	4.4		Ca 42-2	The above specimen measured in a longitudinal field of 3.91 KOe.
37	342	L	1953	4.4		Ca 42-2	The above specimen measured in a transverse field of 3.98 KOe.
38	342	L	1953	4.4		Ca 42-3	Single crystal; supplied by National Physical Lab; rod axis parallel to the intermediate electrical resistance direction; measured in a transverse field of 0.2 KOe.
39	342	L	1953	2.3, 4.4		Ca 42-3	The above specimen measured in a transverse field of 0.38 KOe.
40	342	L	1953	2.3, 4.4		Ca 42-3	The above specimen measured in a longitudinal field of 0.38 KOe.
41	342	L	1953	4.4		Ca 42-3	The above specimen measured in a transverse field of 0.75 KOe.
42	342	L	1953	2.3, 3.3		Ca 42-3	The above specimen measured in a transverse field of 1.17 KOe.
43	342	L	1953	2.3, 4.4		Ca 42-3	The above specimen measured in a longitudinal field of 1.17 KOe.
44	342	L	1953	4.4		Ca 42-3	The above specimen measured in a transverse field of 1.43 KOe.
45	342	L	1953	2.3, 3.3		Ca 42-3	The above specimen measured in a transverse field of 1.80 KOe.
46	342	L	1953	2.3, 4.4		Ca 42-3	The above specimen measured in a longitudinal field of 1.80 KOe.
47	342	L	1953	4.4		Ca 42-3	The above specimen measured in a transverse field of 2.16 KOe.
48	342	L	1953	2.3, 3.3		Ca 42-3	The above specimen measured in a transverse field of 2.53 KOe.
49	342	L	1953	2.3, 4.4		Ca 42-3	The above specimen measured in a longitudinal field of 2.53 KOe.
50	342	L	1953	4.4		Ca 42-3	The above specimen measured in a transverse field of 2.90 KOe.
51	342	L	1953	2.3, 3.3		Ca 42-3	The above specimen measured in a transverse field of 3.22 KOe.
52	342	L	1953	2.3, 4.4		Ca 42-3	The above specimen measured in a longitudinal field of 3.22 KOe.

SPECIFICATION TABLE NO. 17 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
53	342	L	1953	4.4		Ga 42-3	The above specimen measured in a transverse field of 3.60 KOb.
54	342	L	1953	2.3-3.3		Ga 42-3	The above specimen measured in a transverse field of 3.72 KOb.
55	342	L	1953	2.3		Ga 42-3	The above specimen measured in a longitudinal field of 3.7
56	342	L	1953	4.4		Ga 42-3	The above specimen measured in a longitudinal field of 3.84 KOb.
57	342	L	1953	3.3		Ga 42-3	The above specimen measured in a longitudinal field of 3.86 KOb.
58	342	L	1953	4.4		Ga 42-3	The above specimen measured in a transverse field of 4.09 KOb.
59	122	L	1955	1.8-38		Ga 42-1	Single crystal; 2.92 cm long, 0.223 cm in dia; supplied by National Physical Lab; rod axis parallel to the high electrical resistance direction of the crystal; electrical resistivity ratio $\rho_{293K} / \rho_{0K} = 92.11$ .
60	122	L	1955	2.7-40		Ga 42-2	Single crystal; 2.43 cm long, 0.218 cm in dia; supplied by National Physical Lab; rod axis parallel to the low electrical resistance direction of the crystal; electrical resistivity ratio $\rho_{293K} / \rho_{0K} = 111$ .
61	122	L	1955	2.3-79		Ga 42-3	Single crystal supplied by National Physical Lab; rod axis parallel to the intermediate electrical resistance direction of the crystal; electrical resistivity ratio $\rho_{293K} / \rho_{0K} = 106.5$ .
62	608	L	1960	0.10-1.3	10	3 Da	0.1 impurities (mainly Si, P, K, Ca, Al, Ti, and V); single crystalline rod; 3 mm dia; rod axis parallel to the crystallographic a-direction ( $a = 4.5258 \text{ \AA}$ ); electrical resistivity ratio $\rho_{293K} / \rho_{0K} = 1.28 \times 10^4$ ; in superconducting state.
63	608	L	1960	0.12-1.3	10	3 Db	0.1 impurities (mainly Si, P, K, Ca, Al, Ti, and V); single crystalline rod; rod axis parallel to the crystallographic b-direction ( $b = 4.5198 \text{ \AA}$ ); electrical resistivity ratio $\rho_{293K} / \rho_{0K} = 1.18 \times 10^4$ ; in superconducting state.
64	608	L	1960	0.12-1.4	10	3 Dc	0.1 impurities (mainly Si, P, K, Ca, Al, Ti, and V); single crystalline rod; rod axis parallel to the crystallographic c-direction ( $c = 7.6602 \text{ \AA}$ ); electrical resistivity ratio $\rho_{293K} / \rho_{0K} = 0.67 \times 10^4$ ; in superconducting state.
65	608	L	1960	0.13-4.6	10	2 Pa	0.001 impurities (mainly Si, P, Ca, Al, Ti, and V); single crystalline rod; rod axis parallel to the crystallographic a-direction ( $a = 4.5258 \text{ \AA}$ ); electrical resistivity ratio $\rho_{293K} / \rho_{0K} = 2.08 \times 10^4$ ; in normal and superconducting state.
66	608	L	1960	0.12-4.0	10	2 Pb	0.002 impurities (mainly Si, P, Ca, Al, Ti, and V); single crystalline rod; rod axis parallel to the crystallographic b-direction ( $b = 4.5198 \text{ \AA}$ ); electrical resistivity ratio $\rho_{293K} / \rho_{0K} = 2.33 \times 10^4$ ; in normal and superconducting state.
67	608	L	1960	0.13-4.2	10	2 Pc	0.001 impurities (mainly Si, P, Ca, Al, Ti, and V); single crystalline rod; rod axis parallel to the crystallographic c-direction ( $c = 7.6602 \text{ \AA}$ ); electrical resistivity ratio $\rho_{293K} / \rho_{0K} = 1.35 \times 10^4$ ; in normal and superconducting state.
68	608	L	1960	0.16-9.75	10	b-3P	0.001 impurities (mainly Si, P, Ca, Al, Ti, and V); single crystalline rod; rod axis parallel to the crystallographic b-direction ( $b = 4.5198 \text{ \AA}$ ); specimen 1.7 mm in dia; electrical resistivity ratio $\rho_{293K} / \rho_{0K} = 2.44 \times 10^4$ ; in superconducting state.

SPECIFICATION TABLE NO. 17 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)	Specifications and Remarks
69	608	L	1960	0.15-0.85	10	c-4P		0.001 impurities (mainly Si, P, Ca, Al, Ti and V); single crystalline rod; rod axis parallel to the crystallographic c-direction ( $C = 7.6602 \text{ \AA}$ ); dia 0.7 mm; electrical resistivity ratio $\rho(293 \text{ K})/\rho_0 = 10^4$ , in superconducting state.
70	597	L	1961	283-621	5			99.999 pure; measured in solid state and liquid state; 3 mm dia, 64 mm long; melting point 30 C.
71	757	C	1957	333				99.95 pure liquid gallium; supplied by Aluminum Company of America; mercury of 0.0001 impurity used as comparative material.
72	682	L	1960	0.2-0.7		c-6		0.0077 impurity; single crystalline; specimen 0.12 cm dia, approx 50 cm long; heated above the critical temperature after each measuring cycle and brought back into the superconducting state in a magnetic field compensated to approx 0.2 oersted; electrical resistivity ratio $\rho(293 \text{ K})/\rho_0 = 1.85 \times 10^4$ .
73	682	L	1960	0.5-1.0		c-7		Similar to the above specimen except 0.0023 impurity; 0.083 cm dia; $\rho(293 \text{ K})/\rho_0 = 6.25 \times 10^3$ .
74	682	L	1960	0.1-0.8		c-8		Similar to the above specimen except 0.0017 impurity; 0.16 cm dia; $\rho(293 \text{ K})/\rho_0 = 8.33 \times 10^3$ .
75	682	L	1960	0.1-0.9		c-9		Similar to the above specimen except 0.00086 impurity; 0.115 cm dia; $\rho(293 \text{ K})/\rho_0 = 1.67 \times 10^4$ .
76	682	L	1960	0.1-0.7		c-10		Similar to the above specimen except 0.0005 impurity; 0.11 cm dia; $\rho(293 \text{ K})/\rho_0 = 2.78 \times 10^4$ .
77	759	L	1963	83-293		Ga 14-2		Single crystalline rod; approx 4 mm in dia; supplied by National Chemical Laboratory; electrical resistivity reported as 12.0, 20.3, 30.8, 40.6, 50.3, and 54.3 $\mu\text{ohm cm}$ at 83, 123, 173, 223, 273, and 293 K, respectively; electrical resistivity ratio $\rho(293 \text{ K})/\rho(20.4 \text{ K}) = 136$ ; heat flow parallel to the c-axis.
78	758	L	1963	83-293		Ga 14-5		Similar to the above specimen except electrical resistivity reported as 3.52, 6.18, 9.42, 12.7, 16.05, and 17.40 $\mu\text{ohm cm}$ at 83, 123, 173, 223, 273, and 293 K, respectively; electrical resistivity ratio $\rho(293 \text{ K})/\rho(20.4 \text{ K}) = 159$ ; heat flow parallel to the a-axis.
79	758	L	1963	83-293		Ga 14-4		Similar to the above specimen except electrical resistivity reported as 1.72, 2.92, 4.44, 5.96, 7.49, and 8.10 $\mu\text{ohm cm}$ at 83, 123, 173, 223, 273, 293 K, respectively; electrical resistivity ratio $\rho(293 \text{ K})/\rho(20.4 \text{ K}) = 155$ ; heat flow parallel to the b-axis.
80	759	L	1954	1.7-20		a <sub>1</sub>		Impurities: 0.01 Hg, 0.001 Ca, 0.001 Fe, 0.001 Si, 0.0901-0.001 Pb, 0.0001 Mg, and 0.00001 Cu; single crystal; Ga supplied by Aluminum Company of America; electrical resistivity $\rho(273 \text{ K}) = 16.1 \mu\text{ohm cm}$ ; electrical resistivity ratio $\rho(273 \text{ K})/\rho(14 \text{ K}) = 588$ ; heat flow parallel to a-axis.
81	759	L	1954	2.2-20		c <sub>1</sub>		Impurities: 0.01 Hg, 0.001 Ca, 0.001 Fe, 0.0001-0.001 Cu, 0.0001 Mg, and trace of Pb; $\rho(273 \text{ K}) = 52.0 \mu\text{ohm cm}$ ; $\rho(273 \text{ K})/\rho(14 \text{ K}) = 455$ ; heat flow parallel to c-axis.

SPECIFICATION TABLE NO. 17 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
82	759	L	1954	1.8-10		b <sub>1</sub>	Impurities: 0.01 Hg, 0.001 Cr, 0.001 Fe, 0.0001-0.001 Cu, 0.0001-0.003 Pb, trace of Pb; single crystal; Ga supplied by Aluminum Company of America; $\rho(273\text{ K}) = 7.6\ \mu\text{ohm cm}$ ; $\rho(14\text{ K}) = 625$ ; heat flow parallel to b-axis.
83	759	L	1954	1.9-20		a <sub>2</sub>	Impurities: 0.05-0.5 Hg, 0.001 Ca, 0.001 Fe, 0.001 Si, 0.0002-0.003 Pb, 0.0001-0.001 Cu, and 0.0001 Mg; single crystal; Ga supplied by Aluminum Company of America; $\rho(273\text{ K}) = 19.8\ \mu\text{ohm cm}$ ; $\rho(14\text{ K}) = 455$ ; heat flow parallel to b-axis.
84	759	L	1954	2.0-20		b <sub>1</sub>	Impurities: 0.01-0.1 Hg, 0.001 Ca, 0.001 Fe, 0.0001-0.001 Cu, 0.0001-0.001 Pb, and 0.0001 Mg; $\rho(273\text{ K}) = 7.5\ \mu\text{ohm cm}$ ; $\rho(14\text{ K}) = 476$ ; heat flow parallel to b-axis.
85	759	I	1954	1.8-20		c <sub>1</sub>	Impurities: 0.02-0.2 Hg, 0.001 Ca, 0.001 Fe, 0.0001-0.001 Cu, 0.0001 Mg, and trace of Pb; $\rho(273\text{ K}) = 32.8\ \mu\text{ohm cm}$ ; $\rho(14\text{ K}) = 370$ heat flow parallel to c-axis.
86	838	C	1966	278-473			The molten metal placed in a hole 21 mm in dia drilled in an asbestos cement cylinder of 30 mm height; steel 1Kh18N9T used as comparative material.





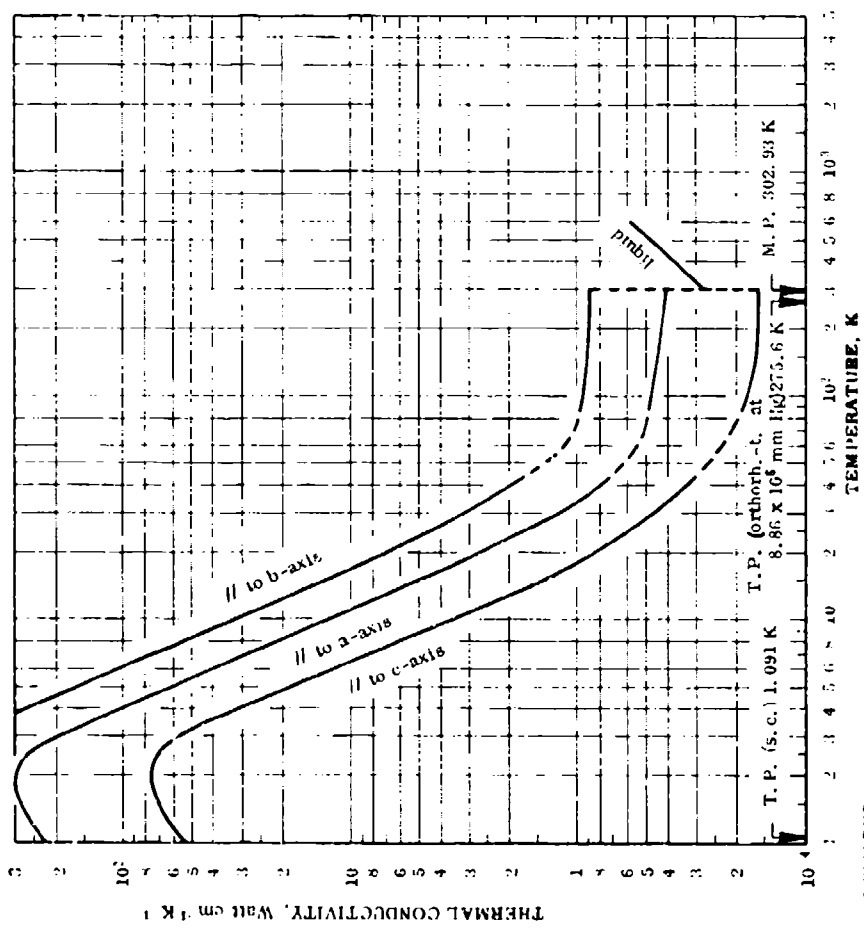


DATA TABLE NO. 17 (continued)

T	k	T	k	T	k
<u>CURVE 81</u>					
2.2	8.0	1.5	3.2	6.0	3.6
2.5	9.6	2.7	4.4	7.2	3.3
2.7	9.0	3.7	5.1	8.6	2.5
3.1	9.8	3.9	6.1	10.0	2.2
3.5	10.0	4.3	6.5*	11.4	1.7
3.7	10.2	4.9	7.1	13.8	1.3
4.1	10.8	5.9	7.4	16.7	1.0
4.6	9.6	6.8	7.3	18.0	0.7
6.2	6.8	8.0	7.5	20.3	0.6
8.3	4.8	9.6	5.9	<u>CURVE 86</u>	
9.9	3.0	10.0	6.3	278.2	0.167*
11.0	2.4	11.3	5.3	288.2	0.163*
12.6	1.6	13.8	3.7	300.2	0.151*
13.8	1.0	16.3	2.6	305.2	0.264
15.6	0.8	18.3	2.1	325.2	0.289
17.0	0.5	20.1	2.0	382.2	0.327
18.0	0.4	<u>CURVE 84</u>		473.2	0.352
19.9	0.4	2.0	7.2		
20.0	0.4°	2.4	8.2		
20.2	0.7	2.7	9.1*		
<u>CURVE 82</u>					
1.8	22.6	3.1	10.1		
2.1	24.6	3.8	12.7		
2.5	28.0	4.5	14.9		
3.0	32.4	5.2	16.3		
3.6	36.8	6.3	17.0		
4.2	39.8	8.6	15.4		
4.7	41.8	10.6	12.7		
4.9	45.2	12.2	10.1		
5.8	47.0	13.8	8.0		
6.4	43.2	15.0	6.9		
7.2	42.0	16.4	5.8		
8.1	38.2	18.2	5.0		
9.6	33.2	20.3	4.3		
<u>CURVE 85</u>					
10.7	22.4	1.8	1.8		
12.1	17.0	2.7	2.4		
13.0	14.4	3.5	3.0		
13.8	12.2	4.3	3.6		
14.9	9.8	4.9	3.9		
16.4	8.6	5.3	3.9		

\* Not shown on plot

FIGURE AND TABLE NO. 17R RECOMMENDED THERMAL CONDUCTIVITY OF GALLIUM



REMARKS

The recommended values are for 99.999% pure gallium with residual electrical resistivity  $\rho_0 = 0.000100$ ,  $0.0000341$ , and  $0.000424 \mu\Omega$  cm along directions parallel to a-, b-, and c-axis, respectively (characterization by  $\rho_0$  becomes important at temperatures below about 150 K). The values below 1.5 T<sub>m</sub> are calculated to fit the experimental data by using  $n = 2.00$ ,  $\sigma' = 3.28 \times 10^{-4}$ , and  $\beta = 0.00409$  for the direction parallel to a-axis; using  $n = 2.00$ ,  $\sigma' = 1.26 \times 10^{-4}$ , and  $\beta = 0.00140$  for the direction parallel to b-axis; and using  $n = 2.00$ ,  $\sigma' = 11.2 \times 10^{-4}$ , and  $\beta = 0.0174$  for the direction parallel to c-axis. The recommended values that are supported by experimental thermal conductivity data are thought to be accurate to within 4% of the true values near room temperature and 4 to 15% at other temperatures.

<sup>†</sup>T<sub>1</sub> in K, k<sub>1</sub> in Watt cm<sup>-1</sup> K<sup>-1</sup>, T<sub>2</sub> in F, and k<sub>2</sub> in Btu lb<sup>-1</sup> ft<sup>-1</sup> F<sup>-1</sup>. <sup>‡</sup>Values in parentheses are interpolated. <sup>§</sup>The values recommended for gallium single crystal in the direction parallel to the a-axis are also approximately good for polycrystalline gallium.

RECOMMENDED VALUES<sup>§</sup>

T <sub>1</sub>	(/ / to a-axis)		(/ / to b-axis)		(/ / to c-axis)		T <sub>2</sub>
	k <sub>1</sub>	k <sub>2</sub>	k <sub>1</sub>	k <sub>2</sub>	k <sub>1</sub>	k <sub>2</sub>	
0	0	0	0	0	0	0	-459.7
1	226	13100	657	38000	54.1	3130	-457.9
2	298	17200	832	48100	76.0	4390	-456.1
3	194	10100	524	30300	58.2	3360	-454.3
4	99.3	5740	272	15700	31.1	1900	-452.5
5	59.2	3420	163	9420	18.2	1050	-450.7
6	38.8	2240	107	6180	11.8	682	-448.9
7	27.2	1570	74.0	4280	8.15	471	-447.1
8	19.9	1150	54.2	3130	5.93	343	-445.3
9	15.2	878	41.4	2390	4.47	258	-443.5
10	11.9	688	32.5	1880	3.49	202	-441.7
12	7.98	461	21.6	1259	2.33	135	-438.1
15	4.88	282	13.2	763	1.44	83.2	-432.7
20	2.65	153	7.13	412	0.835	48.2	-423.7
25	1.71	94.8	4.54	262	0.588	34.0	-414.7
30	1.23	71.1	3.22	186	0.462	26.7	-405.7
35	0.952	55.0	2.46	142	0.385	22.2	-396.7
40	0.795	45.9	1.98	114	0.333	19.2	-387.7
50	(0.634) <sup>†</sup>	(36.6)	(1.42)	(82.0)	(0.269)	(15.5)	(-369.7)
60	(0.555)	(32.1)	(1.15)	(66.4)	(0.233)	(13.5)	(-351.7)
70	(0.516)	(29.8)	(1.02)	(58.9)	(0.210)	(12.1)	(-333.7)
80	(0.494)	(28.5)	(0.983)	(56.8)	(0.196)	(11.3)	(-315.7)
90	0.483	27.9	0.960	55.5	0.187	10.8	-297.7
100	0.474	27.4	0.951	54.9	0.181	10.5	-279.7
150	0.443	25.6	0.918	51.0	0.167	9.65	-189.7
200	0.424	24.5	0.896	51.8	0.163	9.42	-99.7
250	0.414	23.9	0.885	51.1	0.160	9.24	-9.7
273.2	0.410	23.7	0.884	51.1	0.160	9.24	32.0
300	0.406	23.5	0.883	51.0	0.159	9.19	80.3
302.93	0.406	23.5	0.883	51.0	0.159	9.19	85.60

In Liquid State

T <sub>1</sub>	k <sub>1</sub>	k <sub>2</sub>	T <sub>2</sub>
302.93	0.281	16.2	85.60
400	0.378	21.8	260.3
500	0.482	27.8	440.3
600	0.586	33.9	620.3

# THERMAL CONDUCTIVITY OF GERMANIUM

FIGURE SHOWS ONLY 80 OF THE CURVES REPORTED IN TABLE

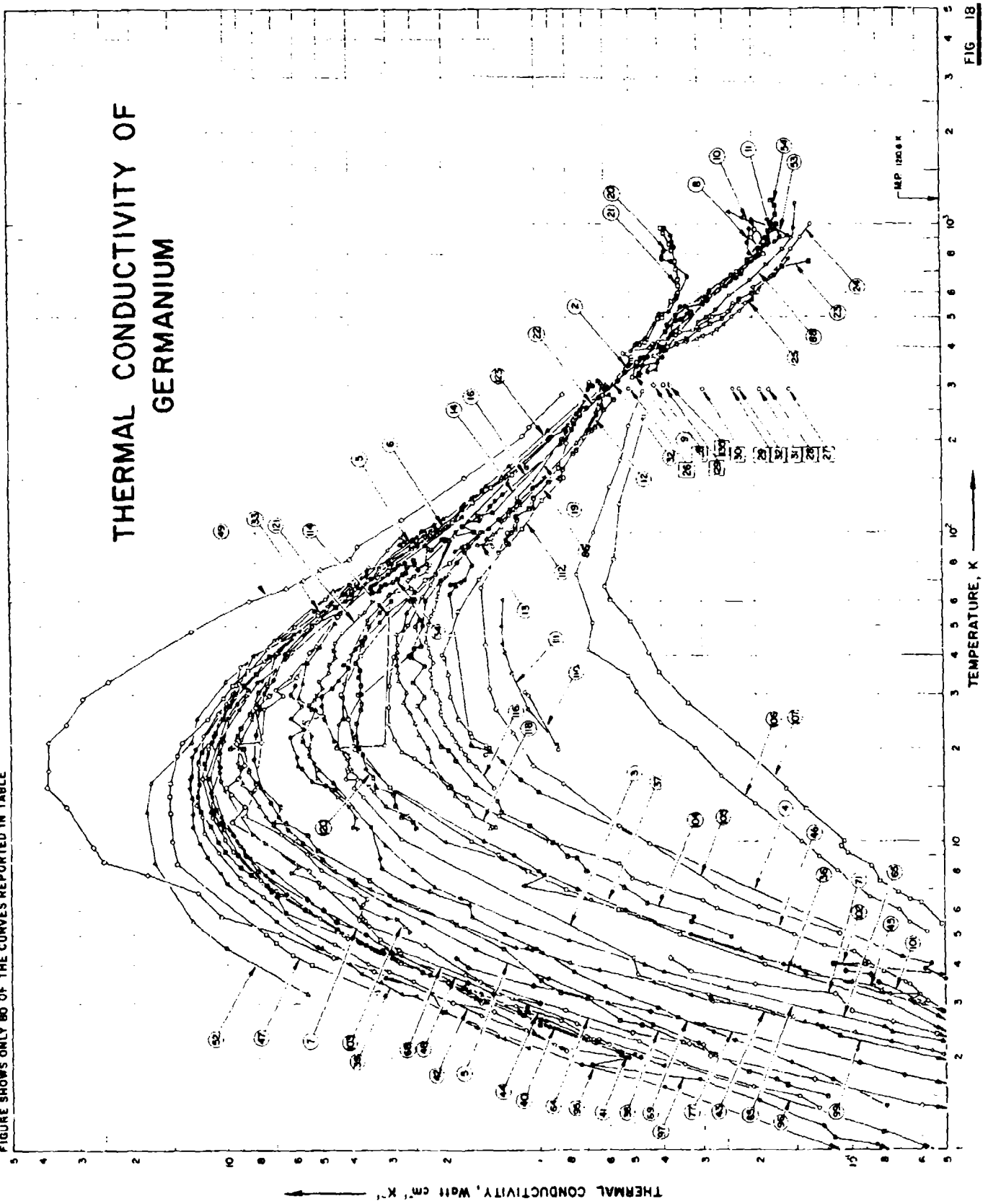


FIG 18

SPECIFICATION TABLE NO. 15 THERMAL CONDUCTIVITY OF GERMANIUM

(Impurity < 0.20% each; total impurities < 0.50%)

[For Data Reported in Figure and Table No. 4-1]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
1	264	C	1952	298-373	10		High-purity; n-type single crystal; 0.313 x 0.313 x 0.75 in.; heat flow parallel to [100] crystalline axis; Ni and Zn used as comparative materials.
2	267	C	1955	278-365			High-purity; n-type single crystal; supplied by Westinghouse Research Laboratories; heat flow parallel to the [100] crystal direction; cast Zn and cast Ni used as comparative materials.
3	265	L	1951	2.7-79	20		High-purity; single crystal; 0.25 in. in dia and ~1.5 in. long; prepared by melting in a graphite crucible, solidified slowly by lowering the crucible through the furnace at a rate of 3 in. hr <sup>-1</sup> ; electrical resistivity 0.30 ohm cm at room temperature.
4	155	L	1951	2.7-86	<20		0.0022 Al; single crystal; 0.125 in. in dia and ~1.5 in. long; same preparation method as above; electrical resistivity 0.0021 ohm cm at room temperature.
5	274	L	1954	2.2-95		Ge 1	High-purity; single crystal; 1 cm long and 1.2 x 0.6 mm cross section; electrical resistivity at room temperature 10 ohm cm.
6	343	L	1956	2.8-137		Ge 3a	High-purity; p-type; polycrystalline.
7	343	L	1956	1.9-111		Ge 3b	The above specimen cleaned, annealed at 550 C for 3 hrs in helium, then cooled slowly.
8	344	L	1959	330-962		1	n-type single crystal; 0.4 cm <sup>2</sup> in cross sectional area and 0.3 cm long; electrical resistivity 3 ohm cm at 293 K.
9	344	L	1959	319-962		2	Similar to the above specimen except electrical resistivity 0.05 ohm cm at 293 K.
10	344	L	1959	331-1094		3	Similar to the above specimen except 0.5 cm long and electrical resistivity 0.03 ohm cm at 293 K.
11	344	L	1959	324-1040		4	Similar to the above specimen except 0.3 cm long and electrical resistivity 0.001 ohm cm at 293 K.
12	345	L	1957	87-293	3-8	1	Sb-doped; n-type single crystal; 2.035 x 0.231 x 0.417 cm; electrical resistivity 2.84 ohm cm at room temperature.
13	345	L	1957	91-307	3-8	2	Sb-doped; n-type single crystal; 1.800 x 0.228 x 0.294 cm; electrical resistivity 6 x 10 <sup>-3</sup> ohm cm at room temperature.
14	345	L	1957	88-300	3-8	8	Sb-doped; p-type single crystal; thermally converted from the above specimen 2; 1.675 x 0.212 x 0.250 cm; electrical resistivity 1.6 ohm cm at room temperature.
15	345	L	1957	89-311	3-8	3	Sb-doped; n-type single crystal; dimensions 1.575 x 0.262 x 0.329 cm; electrical resistivity 40 ohm cm at room temperature.
16	345	L	1957	94-311	3-8	4	Ga-doped; p-type single crystal; dimensions 2.000 x 0.356 x 0.285 cm; electrical resistivity 68 ohm cm at room temperature.

SPECIFICATION TABLE NO. 18 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
17	345	L	1957	96-305	3-8	5	Ga-doped p-type single crystal; dimensions 1.450 x 0.215 x 0.330 cm; electrical resistivity 51 ohm cm at room temperature.
18	345	L	1957	208-260	3-8	6	Ga-doped p-type single crystal; dimensions 2.205 x 0.295 x 0.503 cm; electrical resistivity 5 ohm cm at room temperature.
19	345	L	1957	90-306	3-5	7	Ga-doped p-type single crystal; 2.13 cm long, 0.062 cm <sup>2</sup> cross sectional area; electrical resistivity $3 \times 10^{-3}$ ohm cm at room temperature.
20	346	L	1959	388-971			Single crystal; intrinsic; 18 mm dia x 20 mm long.
21	346	L	1959	405-971			p-type; single crystal; 18 mm in dia x 20 mm long; impurity concentration $1.1 \times 10^{16}$ cm <sup>-3</sup> .
22	347		1958	76-370			p-type; single crystal; crystallographic orientation [111]; heat treated at 500 C for 32 hrs.
23	348	C	1959	368-758			n-type; single crystal; oriented in the [100] direction; 30 mils dia x 100 mils long; measured in He atmosphere; Ni used as comparative material (data from Honda, Simidv, 1917).
24	348	C	1959	354-1000			Similar to the above specimen but oriented in the [110] direction.
25	348	C	1959	370-775			Similar to the above specimen.
26	349	L	1954	293	4		Single crystal; impurity concentration $1.4 \times 10^{15}$ cm <sup>-3</sup> ; approx 15 mm long and 16 mm in dia.
27	349	L	1958	293	4		Ga-doped p-type single crystal; impurity concentration $7.4 \times 10^{17}$ atom cm <sup>-3</sup> ; approx 15 mm long and 16 mm in dia.
28	349	L	1958	293	4		Fe-doped n-type single crystal; impurity concentration $4.1 \times 10^{17}$ atom cm <sup>-3</sup> ; approx 15 mm long and 16 mm in dia.
29	349	L	1958	293	4		Ga-doped p-type single crystal; impurity concentration $4.1 \times 10^{16}$ atom cm <sup>-3</sup> ; approx 15 mm long and 16 mm in dia.
30	349	L	1958	293	4		Fe-doped n-type single crystal; impurity concentration $2.2 \times 10^{16}$ atom cm <sup>-3</sup> ; approx 15 mm long and 16 mm in dia.
31	349	L	1958	293	4		Ga-doped p-type single crystal; impurity concentration $8.8 \times 10^{16}$ atom cm <sup>-3</sup> ; approx 15 mm long and 16 mm in dia.
32	349	L	1958	293	4		Fe-doped n-type single crystal; impurity concentration $7.5 \times 10^{16}$ atom cm <sup>-3</sup> ; approx 15 mm long and 16 mm in dia.
33	350	L	1953	50-91		A18-4-1	Al-doped p-type single crystal; cut transverse to the axis of crystal growth; electrical resistivity 0.2 ohm cm at room temperature.

SPECIFICATION TABLE NO. 1<sup>a</sup> (continued)

Curve No.	Rel. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
34	350	I	1955	45-81		Sb-7	Sb-doped n-type single crystal; specimen parallel to the [110] direction; cut transverse to the axis of crystal growth; electrical resistivity 0.013 ohm cm at room temperature.
35	351		1954	2-83	3-5		Pure crystal.
36	352	I	1957	2.0-100		Ge-2	Ga-doped p-type single crystal; 2.2 x 3.6 x 2.5 mm; cut transverse to the crystal growth axis; electrical resistivity 0.9 and 7.84 milliohm cm at 10 K and room temperature, respectively.
37	352	I	1957	2.0-100		Ge-2	The above specimen irradiated in Argonne CP-5 reactor with fast neutrons (total flux $5 \times 10^{16}$ neutrons $\text{cm}^{-2}$ ) then kept at room temperature for about 5 months; electrical resistivity 0.4 and 7.65 milliohm cm at room temperature, respectively.
38	353	I	1958	1.5-120		PN-3E	Ga-Sb-doped p-type single crystal; data derived from the measurement of three specimens.
39	354	I	1957	2.2-280		Ge-2	n-type single crystal; carrier concentration $10^{15} \text{ cm}^{-3}$ ; specimen cross section 2.53 x 2.68 mm; zone grown; electrical resistivity at room temperature approx 20 ohm cm; heat flow parallel to the [100] direction.
40	354	I	1957	2.1-31		Ge-3	In-doped n-type single crystal; carrier concentration $10^{15} \text{ cm}^{-3}$ ; specimen cross section 2.56 x 2.65 mm; grown by zone melting; electrical resistivity 35 ohm cm at room temperature; heat flow parallel to [111] direction.
41	354	I	1957	2.9-84		Ge-4	In-doped p-type single crystal; carrier concentration $1.9 \times 10^{14} \text{ cm}^{-3}$ ; specimen cross section 2.65 x 2.66 mm; grown by zone melting; electrical resistivity 21 ohm cm at room temperature; heat flow parallel to [111] direction.
42	354	I	1957	2.2-75		Ge-5	n-type single crystal; carrier concentration $10^{12} \text{ cm}^{-3}$ ; specimen cross section 2.19 x 2.10 mm; pulled from melt, cut parallel to crystal growth direction; electrical resistivity approx 41 ohm cm at room temperature; heat flow parallel to [100] direction.
43	354	I	1957	2.0-97		Ge-7	In-doped p-type single crystal; carrier concentration $2.3 \times 10^{16} \text{ cm}^{-3}$ ; specimen cross section 2.70 x 2.69 mm; grown by zone melting; electrical resistivity approx 0.19 ohm cm at room temperature; heat flow parallel to the [111] direction.
44	354	I	1957	2.2-80		Ge-19	In-doped p-type single crystal; carrier concentration $10^{15} \text{ cm}^{-3}$ ; specimen cross section 2.56 x 2.69 mm; grown by zone melting; electrical resistivity approx 2.75 ohm cm at room temperature; heat flow parallel to the [111] direction.
45	354	I	1957	2.0-100		Ge-11	Ga-doped p-type single crystal; carrier concentration $2 \times 10^{18} \text{ cm}^{-3}$ ; specimen cross section 2.13 x 1.86 mm; pulled from melt, cut parallel to crystal growth direction; electrical resistivity approx 0.009 ohm cm at room temperature; heat flow parallel to the [100] direction.

SPECIFICATIONS TABLE NO. 14 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
46	354	L	1957	2.2-96		Ge 12	Ge-doped p-type single crystal; carrier concentration $10^{19}$ cm $^{-3}$ ; specimen cross section 2.06 x 1.95 mm; pelted from melt, cut parallel to crystal growth direction; electrical resistivity, approx 0.0027 ohm cm at room temperature; heat flow parallel to the [100] direction.
47	355	L	1955	3.3-155	±5	Ge 1	n-type single crystal; specimen 2 x 4 mm cross section, 5 mm long; electrical resistivity 31 ohm cm at 22°C.
48	779	L	1958	3.0-25		Normal Ge	n-type single crystal.
49	779	L	1958	2.1-240		Enriched Ge $^{13}$	Specimen composed of the following isotopes: 95.74 Ge $^{74}$ , 0.728 Ge $^{76}$ , 1.10 Ge $^{77}$ , 1.54 Ge $^{78}$ and 0.440 Ge $^{80}$ ; supplied by Union Carbide Nuclear Co.; $1.2 \times 10^{18}$ excess donor atoms cm $^{-3}$ ; 2.54 cm long, 0.13 x 0.157 cm cross section; zone refined, grown using a modified Teal little crystal puller; heat flow parallel to the [100] direction.
50	578	C	1960	311-683			n-type; electrical resistivity approx 5 ohm cm (inhomogeneous, electrical resistivity especially high at center); F. H. stainless steel used as the reference (data on F. H. stainless steel from R. W. Powell 1936).
51	263	L	1958	300		T-1097	Sh-doped n-type; polycrystalline; specimen dimensions 5 x 5 x 1.5 mm; electrical resistivity 10.0 ohm cm at 300 K; measured in a vacuum of approx $10^{-6}$ mm Hg.
52	790, 684	L	1960	3.2-300	+5		Cu-doped p-type single crystal; (approx $10^{18}$ atoms cm $^{-3}$ ); dimensions 0.94 x 0.94 x 3.2 cm; dislocation density $3 \times 10^5$ cm $^{-2}$ .
53	790	R	1960	300-1020	+5		Very pure; n-type polycrystalline; crystal size approx 0.2 cm; specimen 1.27 cm in dia 6.1 cm long; zone refined, ground and cut to desired size; electrical resistivity 3.0 ohm cm at 300 K; average crystallite size 0.2 cm.
54	781, 684	R	1963	398-1194	+5		Intrinsic Ge; carrier concentration $2 \times 10^{15}$ cm $^{-3}$ ; doped by copper during measurement to give $5 \times 10^{15}$ acceptors cm $^{-3}$ ; cylindrical specimen approx 2.6 cm dia x 13 cm long made from single crystal grown by Czochralski's method from zone refined germanium of the G. E. Co.; specimen aligned in the [100] crystalline direction; electrical resistivity before thermal conductivity measurement 46.6 ohm cm at room temperature changed to 4.6 ohm cm after the measurement.
55	626	P	1960	308-1073	±2		As-doped n-type single crystal; 2 in. long; cross section 0.3 x 0.3 in.; heat flow and rod axis parallel to [111] direction; electrical resistivity at room temp, 0.3 ohm cm; thermal conductivity values calculated from measured thermal diffusivity data and the specific heat value of $1.83$ J cm $^{-2}$ K $^{-1}$ (derived from Dulong-Petit law).
56	782	L	1957	16			Very pure; single crystal; 14.80 mm $^2$ cross section; polished.
57	783	L	1960	112-429		Sample 1	p-type; crystal obtained by Czochralski method; cross sectional area, 0.53 x 0.32 cm $^2$ ; sand blasted; electrical resistivity $\rho$ at room temperature 60 ohm cm; data corrected for radiation.



SPECIFICATION TABLE NO. 18 (continued)

Curve No.	Rel. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
58	783	L	1960	83-455		Sample 1A	Similar to the above specimen except cross sectional area $0.45 \times 0.25 \text{ cm}^2$ , produced from specimen 1 by grinding away a side.
59	783	L	1960	108-460		Sample 2	Similar to the above specimen except cross sectional area $0.74 \times 0.55 \text{ cm}^2$ , $\rho = 21 \text{ ohm cm}$ at room temperature.
60	783	L	1960	109-461		Sample 3	n-type; crystal obtained by Czochralskii method; cross sectional area $0.39 \times 0.63 \text{ cm}^2$ , and blasted $\rho = 2 \text{ ohm cm}$ at room temperature.
61	783	L	1960	106-451		Sample 4	Ga-doped p-type; obtained from zone melting; cross sectional area $0.72 \times 0.87 \text{ cm}^2$ , $\rho = 0.049 \text{ ohm cm}$ at room temperature.
62	783	L	1960	107-460		Sample 5	Sb-doped n-type; obtained from zone melting; cross sectional area $0.82 \times 0.53 \text{ cm}^2$ , $\rho = 0.043 \text{ ohm cm}$ at room temperature.
63	784	L	1962	1.2-98	6	Sb 30	Sb-doped single crystal; $12.217 \times 3.211 \times 1.5055 \text{ mm}$ ; from ingot grown by the Czochralskii technique with the growth axis in [110] direction; cut transversely to the ingot axis, ground to size with specimen axis approx. [100] direction; carrier concentration $n = 2.5 \times 10^{18} \text{ cm}^{-3}$ , electrical resistivity reported as 2.691, 2.70, 2.67, 2.63, 2.6, 2.846, 2.855, 2.871, 2.877, 2.904, 2.919, 3.319, 3.386, 3.413, 3.441, 3.443, and $3.524 \text{ m}\Omega$ ; $n = 1.234, 1.664, 2.004, 2.658, 4.128, 14.00, 14.78, 16.11, 17.98, 18.11, 20.22, 55.60, 68.44, 74.14, 84.93, 90.46, \text{ and } 296 \text{ K}$ , respectively.
64	784	L	1962	1.3-87	6	Sb 172	Sb-doped single crystal; $15.918 \times 3.8221 \times 3.7129 \text{ mm}$ ; same fabrication method as above; specimen axis approx in [111] direction; $n = 6.1 \times 10^{17} \text{ cm}^{-3}$ , electrical resistivity reported as $2.15 \times 10^6, 3.20 \times 10^4, 3.22 \times 10^3, 5.66 \times 10^2, 63.22, 21.46, 4.311, 2.277, 2.056, 1.608, 1.115, 0.9610, 0.6570, 0.5448, 0.4587, 0.3685, 0.3039, 0.2309, 0.1557, 0.1075, 0.09557, 0.09040, 0.09343, 0.09512, 0.09708, 0.09631, 0.1052, 0.1252, 0.1812, 0.248, \text{ and } 0.3988 \text{ ohm cm}$ at 4.214, 5.139, 5.947, 6.763, 8.422, 9.690, 12.54, 14.22, 14.64, 15.45, 17.03, 17.82, 20.22, 21.76, 23.39, 26.02, 28.65, 34.06, 39.36, 47.13, 52.74, 62.13, 77.22, 81.32, 82.93, 97.20, 120.2, 166.6, 211.1, and 297.5 K, respectively.
65	784	L	1962	1.3-145	6	Sb 187	Sb-doped single crystal; $16.422 \times 3.8906 \times 4.0601 \text{ mm}$ ; same fabrication method as above; specimen axis approx in [100] direction; $n = 1.2 \times 10^{17} \text{ cm}^{-3}$ , electrical resistivity reported as 192.4, 178.3, 172.8, 168.7, 163.6, 158.8, 155.8, 154.2, 152.8, 153.3, 154.1, 155.2, 151.0, 141.8, 132.4, 113.5, 98.69, 87.6, 79.16, 68.39, 55.54, 43.86, 37.13, 31.25, 24.55, 23.36, 22.51, 20.32, 18.81, 18.51, and 24.4 milliohm cm at 1.314, 1.436, 1.639, 1.835, 2.137, 2.582, 3.042, 3.451, 4.209, 5.481, 6.587, 8.208, 12.16, 14.12, 16.84, 20.23, 23.39, 25.71, 28.32, 31.46, 37.49, 45.16, 51.71, 60.13, 77.24, 82.12, 86.62, 102.6, 126.2, 141.5, and 296.9 K, respectively.

SPECIFICATION TABLE NO. 14 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
66	794	I	1962	1.2-98	6	Sb 207	Sb-doped single crystal: $14.73 \times 3.975 \times 3.9063$ mm; same fabrication method and specimen axis orientation as above; $n = 2.4 \times 10^{16} \text{ cm}^{-3}$ ; electrical resistivity reported as $7.2 \times 10^3$ , $1.63 \times 10^4$ , $7.15 \times 10^4$ , $3.40 \times 10^5$ , $9.072 \times 10^5$ , $1.31 \times 10^6$ , $4.31$ , $179.2$ , $35.9$ , $12.49$ , $2.847$ , $1.331$ , $0.7779$ , $0.4813$ , $0.3191$ , $0.2570$ , $0.2143$ , $0.1702$ , $0.1400$ , $0.1049$ , $0.07895$ , $0.06426$ , $0.05369$ , $0.04743$ , $0.04187$ , $0.04091$ , $0.04086$ , $0.04180$ , $0.04372$ , $0.04692$ , and $0.09460$ ohm cm at $2.140$ , $2.584$ , $3.016$ , $3.467$ , $4.214$ , $5.222$ , $5.935$ , $6.609$ , $8.194$ , $9.563$ , $12.21$ , $14.04$ , $15.84$ , $17.91$ , $20.21$ , $21.75$ , $23.28$ , $25.48$ , $27.72$ , $29.01$ , $34.06$ , $44.33$ , $52.17$ , $60.21$ , $77.36$ , $84.40$ , $99.69$ , $115.1$ , $128.9$ , $146.1$ , and $296.4$ K, respectively.
67	794	I	1962	1.5-137	6	Sb 222	Sb-doped single crystal: $18.128 \times 4.0751 \times 4.0667$ mm; same fabrication method and specimen axis orientation as above; $n = 1.1 \times 10^{19} \text{ cm}^{-3}$ ; electrical resistivity reported as $4.394$ , $4.372$ , $4.385$ , $4.404$ , $4.418$ , $4.430$ , $4.447$ , $4.466$ , $5.486$ , $4.523$ , $4.585$ , $4.642$ , $4.728$ , $4.786$ , $4.849$ , $4.891$ , $4.906$ , $5.020$ , $5.141$ , $5.268$ , $5.373$ , $5.453$ , $5.517$ , $5.479$ , $5.436$ , $5.375$ , $5.262$ , $5.075$ , $4.940$ , and $4.951$ milliohm cm at $1.320$ , $1.579$ , $1.925$ , $2.587$ , $3.032$ , $3.422$ , $4.607$ , $5.624$ , $6.971$ , $8.256$ , $10.87$ , $13.14$ , $15.90$ , $17.87$ , $20.20$ , $21.75$ , $23.24$ , $26.15$ , $30.76$ , $36.27$ , $42.04$ , $48.21$ , $54.31$ , $57.28$ , $83.58$ , $92.17$ , $105.7$ , $128.9$ , $149.7$ , and $299.8$ K, respectively.
68	794	I	1962	1.4-137	6	As 223 I	As-doped single crystal: $15.682 \times 4.0709 \times 4.0645$ mm; same fabrication method and specimen axis orientation as above; $n = 2.1 \times 10^{16} \text{ cm}^{-3}$ ; electrical resistivity reported as $7.4 \times 10^3$ , $2.7 \times 10^4$ , $1.99 \times 10^5$ , $343.1$ , $28.34$ , $6.263$ , $3.467$ , $1.267$ , $0.4056$ , $0.4490$ , $0.3821$ , $0.2957$ , $0.2079$ , $0.1640$ , $0.1160$ , $0.0947$ , $0.07229$ , $0.06290$ , $0.05499$ , $0.04620$ , $0.04479$ , $0.04337$ , $0.04303$ , $0.04250$ , $0.04193$ , and $0.09447$ ohm cm at $4.208$ , $5.029$ , $7.378$ , $8.443$ , $10.77$ , $12.95$ , $13.96$ , $15.71$ , $18.04$ , $20.20$ , $21.59$ , $23.25$ , $25.93$ , $29.34$ , $32.92$ , $36.80$ , $43.40$ , $48.46$ , $55.44$ , $71.09$ , $77.17$ , $91.16$ , $104.7$ , $122.3$ , $142.0$ , and $293.9$ K, respectively.
69	794	I	1962	1.5-125	6	As 225 II	As-doped single crystal: $15.334 \times 3.803 \times 4.0645$ mm; same fabrication method and specimen axis orientation as above; $n = 5.3 \times 10^{16} \text{ cm}^{-3}$ ; electrical resistivity $0.04175$ ohm cm at room temperature.
70	794	I	1962	1.4-134	6	As 226	As-doped single crystal: $17.482 \times 4.0612 \times 3.9648$ mm; same fabrication method and specimen axis orientation as above; $n = 8.5 \times 10^{16} \text{ cm}^{-3}$ ; electrical resistivity reported as $5.370$ , $5.377$ , $5.544$ , $5.592$ , $5.617$ , $5.642$ , $8.631$ , $8.800$ , $8.794$ , $8.764$ , $8.709$ , $8.587$ , $8.625$ , $8.749$ , $8.943$ , $9.170$ , $9.394$ , $9.608$ , $9.730$ , $9.872$ , $9.880$ , $9.730$ , $9.522$ , $9.018$ , $8.075$ , $8.224$ , $7.907$ , $7.590$ , and $6.268$ milliohm cm at $1.346$ , $1.564$ , $1.751$ , $1.946$ , $2.384$ , $3.020$ , $3.507$ , $4.191$ , $4.927$ , $5.394$ , $5.855$ , $7.015$ , $7.996$ , $10.65$ , $14.32$ , $17.61$ , $23.29$ , $28.20$ , $32.11$ , $38.92$ , $49.48$ , $58.49$ , $77.15$ , $85.36$ , $95.98$ , $121.2$ , $123.5$ , $136.4$ , and $293.5$ K, respectively.

SPECIFICATION TABLE NO. 1 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K.	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
71	74	1	1952	1.4-4.2	5	As 232	As-doped single crystal; 19.812 x 3.5479, 4.034 mm; same fabrication method and specimen axis orientation as above; $n = 3.1 \times 10^{15} \text{ cm}^{-3}$ ; electrical resistivity reported as 0.8294, 0.7764, 0.7216, 0.6667, 0.6131, 0.5307, 0.4573, 0.3341, 0.2719, 0.2043, and 0.01467 ohm cm at 1.349, 1.533, 1.635, 1.764, 1.883, 2.205, 2.526, 3.471, 4.198, 77.2, and 296.5 K, respectively.
72	74	1	1952	1.3-129	5	As 233 I	As-doped single crystal; 17.305 x 4.0731 x 3.9978 mm; same fabrication method and specimen axis orientation as above; $n = 1.1 \times 10^{15} \text{ cm}^{-3}$ ; electrical resistivity reported as $1.97 \times 10^4$ , $4.69 \times 10^3$ , $1.69 \times 10^3$ , $7.04 \times 10^2$ , $2.238 \times 10^4$ , $2.222 \times 10^4$ , $2.189 \times 10^4$ , $3.108$ , $3.15.6$ , $104.2$ , $44.19$ , $16.65$ , $5.344$ , $2.647$ , $1.246$ , $0.7792$ , $0.4944$ , $0.2371$ , $0.2784$ , $0.2309$ , $0.1822$ , $0.1471$ , $0.1213$ , $0.0825$ , $0.06402$ , $0.0244$ , $0.04378$ , $0.03024$ , $0.02782$ , $0.02584$ , $0.02424$ , $0.02318$ , $0.02277$ , and $0.02129$ ohm cm at 2.318, 2.598, 3.046, 3.513, 4.204, 4.296, 4.188, 5.132, 6.487, 7.315, 8.048, 9.144, 10.80, 12.18, 14.30, 15.93, 17.90, 20.28, 21.64, 23.12, 25.31, 27.69, 30.22, 35.90, 42.60, 48.36, 55.08, 76.52, 84.12, 94.16, 106.1, 122.9, 141.4, and 295.4 K, respectively.
73	74	1	1952	1.3-129		As 233 II	As-doped single crystal; 17.053 x 3.9504 x 4.0951 mm; same fabrication method and specimen axis orientation as above; $n = 1.7 \times 10^{15} \text{ cm}^{-3}$ ; electrical resistivity reported as $2.81 \times 10^4$ , $1.826 \times 10^4$ , $9742$ , $2845$ , $1045$ , $373.7$ , $154.4$ , $36.90$ , $14.15$ , $5.802$ , $1.956$ , $1.188$ , $0.8430$ , $0.5392$ , $0.3085$ , $0.2690$ , $0.2149$ , $0.1757$ , $0.1483$ , $0.1247$ , $0.1216$ , $0.1040$ , $0.09275$ , $0.06963$ , $0.05547$ , $0.04247$ , $0.03694$ , $0.03077$ , $0.02317$ , $0.02172$ , $0.02090$ , $0.01982$ , $0.01838$ , $0.01778$ , $0.01703$ , $0.01618$ , and $0.01977$ ohm cm at 1.445, 1.578, 1.786, 2.171, 2.526, 2.946, 3.338, 4.202, 5.091, 5.914, 7.468, 8.479, 9.291, 10.84, 13.57, 14.34, 16.14, 18.16, 20.22, 21.61, 23.16, 25.74, 27.78, 33.47, 38.80, 46.49, 51.32, 59.03, 77.25, 82.08, 96.98, 93.10, 103.8, 109.3, 118.6, 122.7, and 297.0 K, respectively.
74	74	1	1952	1.3-76		SiGa 120	Doped with antimony and gallium; single crystal; 19.413 x 4.2278 x 3.7717 mm; same fabrication method and specimen axis orientation as above; $n = 5.4 \times 10^{16} \text{ cm}^{-3}$ ; electrical resistivity reported as $2.13 \times 10^4$ , $1.65 \times 10^4$ , $1.29 \times 10^4$ , $1.83 \times 10^4$ , $891$ , $685$ , $421$ , $346$ , $191.4$ , $124.0$ , $86.78$ , $56.26$ , $55.66$ , $41.60$ , $34.63$ , $29.32$ , $20.81$ , $15.59$ , $10.58$ , $8.182$ , $6.502$ , $4.978$ , $4.420$ , $4.125$ , $3.534$ , $3.075$ , $2.425$ , $1.884$ , $1.599$ , $1.162$ , $0.8203$ , $0.5925$ , and $0.1304$ ohm cm at 1.303, 1.539, 1.496, 1.569, 1.637, 1.756, 2.017, 2.141, 2.589, 3.023, 3.475, 4.210, 4.229, 4.884, 5.354, 6.841, 7.091, 8.388, 11.23, 13.66, 16.44, 20.41, 22.42, 23.64, 26.52, 29.17, 34.26, 39.97, 43.96, 52.48, 63.04, 81.42, and 296.4 K, respectively.

SPECIFICATION TABLE NO. 18 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)	Specifications and Remarks
75	784	L	1962	1.3-14.8	6	SbGa 183		Doped with antimony and gallium; single crystal; 15.65 x 3.7364 x 3.6070 mm; same fabrication method and specimen axis orientation as above; $n = 1.5 \times 10^{17} \text{ cm}^{-3}$ ; electrical resistivity reported as 25.7, 21.24, 17.78, 14.54, 12.14, 9.619, 6.192, 4.578, 3.638, 2.835, 2.149, 1.750, 1.537, 1.245, 1.038, 0.9525, 0.866, 0.7876, 0.7180, 0.6489, 0.6039, 0.5136, 0.4704, 0.3877, 0.3183, 0.2497, 0.2110, 0.1684, 0.1397, 0.1291, 0.1177, and 0.06187 ohm cm at 1, 269, 1, 368, 1, 487, 1, 634, 1, 751, 2, 022, 2, 596, 3, 511, 4, 206, 5, 212, 6, 792, 8, 398, 9, 787, 12, 53, 15, 94, 17, 94, 20, 32, 24, 16, 26, 21, 29, 85, 32, 83, 39, 78, 47, 47, 52, 55, 62, 58, 77, 18, 87, 21, 109, 8, 123, 9, 141, 9, 161, 6, and 296, 2 K, respectively.
76	784	L	1962	1.4-151	6	SbGa 204		Doped with antimony and gallium; single crystal; 16.822 x 3.913 x 3.924 mm; same fabrication method and specimen axis orientation as above; $n = 2.6 \times 10^{16} \text{ cm}^{-3}$ ; electrical resistivity reported as $1.19 \times 10^5$ , $1.04 \times 10^4$ , $6.82 \times 10^4$ , $6.82 \times 10^4$ , $2.66 \times 10^4$ , $1.27 \times 10^4$ , $7.53 \times 10^3$ , 3857, 3055, 2229, 1712, 1204, 682, 2, 457, 241, 171, 120, 80, 09, 70, 41, 61, 69, 51, 53, 40, 21, 31, 79, 21, 69, 15, 13, 11, 37, 7, 597, 4, 137, 3, 775, 3, 374, 2, 822, 2, 260, 1, 854, 1, 428, 1, 161, 0, 9402, and 0.4279 ohm cm at 1, 944, 1, 946, 2, 111, 2, 589, 3, 062, 3, 509, 4, 201, 4, 251, 4, 919, 5, 365, 6, 019, 7, 404, 8, 655, 11, 51, 13, 67, 16, 27, 20, 21, 21, 71, 23, 32, 25, 69, 28, 94, 32, 41, 35, 52, 45, 12, 50, 76, 50, 71, 77, 08, 78, 86, 84, 113, 90, 66, 99, 63, 108, 7, 122, 7, 130, 9, 132, 1, and 293, 8 K, respectively.
77	785	L	1962	1.3-4.2		1 N		Sb-doped p-type single crystal; dimensions approx 1 x 2 x 25 mm; specimen axis in [110] direction; lapped and etched; impurity concentration, $n = 2 \times 10^{16} \text{ cm}^{-3}$ ; electrical resistivity 0.13 ohm cm at 300 K.
78	785	L	1962	1.7-3.9		2 N		Similar to the above specimen except specimen axis in [111] direction and electrical resistivity 0.12 ohm cm at 300 K.
79	785	L	1962	1.7-4.1		3 N		Similar to the above specimen except specimen axis in [100] direction, $n = 1 \times 10^{16} \text{ cm}^{-3}$ , and electrical resistivity 0.20 ohm cm at 300 K.
80	785	L	1962	2.0-4.0		4 N		As-doped single crystal; dimensions approx 1 x 2 x 25 mm; specimen axis in [110] direction; lapped and etched; $n = 2 \times 10^{16} \text{ cm}^{-3}$ ; electrical resistivity 0.12 ohm cm at 300 K.
81	785	L	1952	2.1-4.0		5 N		Pure; dimensions approx 1 x 2 x 25 mm; specimen axis in [111] direction; lapped and etched; electrical resistivity 31 ohm cm at 300 K.
82	785	L	1942	2.1-3.9		1 P		p-type; similar to the above specimen except specimen axis in [110] direction and electrical resistivity 47 ohm cm at 300 K.
83	786	L	1958	2.1-67	10-30			Pure germanium crystal; size approx 0.125 x 0.125 x 0.625 in.; provided by the Radio Corporation of America.
84	787, 751		1963	90-300				Germanium crystal; before neutron bombardment.

SPECIFICATION TABLE NO. 13 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
85	757, 751		1953	90-300			The above specimen after a bombardment of $6 \times 10^{17}$ neutron $\text{cm}^{-2}$ .
86	757, 751		1963	85-270			The above specimen after a bombardment of $1.2 \times 10^{18}$ neutron $\text{cm}^{-2}$ .
87	752	P	1962	300-10 <sup>5</sup>		Ge-1810	As-doped n-type single crystal; carrier concentration $5 \times 10^{15}$ atoms $\text{cm}^{-3}$ .
88	752	P	1962	317-1075		Ge-1796	Similar to the above specimen except carrier concentration $3 \times 10^{15}$ atoms $\text{cm}^{-3}$ .
89	752	P	1962	562-1136		Ge-5	Cd-doped p-type single crystal; carrier concentration $2.4 \times 10^{19}$ atoms $\text{cm}^{-3}$ .
90	1010	L	1957	18-94	2		n-type single crystal; specimen dimensions 1.5 x 1.8 x 15 mm; electron concentration approx $10^{16}$ $\text{cm}^{-3}$ ; heat flow parallel to [100] direction.
91	788	L	1963	1.3-82	30	CS	Sb-doped single crystal; supplied by Bell Telephone Laboratories; cut into an "L" shape; dimensions of legs approx 5 mm long, 1.2 x 1.2 mm cross section; one leg connected to heat sink (S leg), another leg connected to heater (H leg); S leg perpendicular to H leg; S leg axis aligned in the [112] direction; measurements made on S leg.
92	788	L	1963	1.3-85	30	CH	Data from measurements made on H leg of the above specimen.
93	788	L	1963	1.4-84	30	DS	Similar to the above specimen but with the H leg bent to a circular curvature of radius 3.35 cm; measurements made on the S leg.
94	788	L	1963	1.4-88	30	DH	Data from measurements made on the H leg
95	789	L	1962	0.2-3.6	5	Ge 2	Same specimen as the one for curve 39.
96	789	L	1962	0.2-3.2	5	Ge 3	Same specimen as the one for curve 40.
97	789	L	1962	0.3-2.9	5	Ge 4	Same specimen as the one for curve 41.
98	789	L	1962	0.2-0.7	5	Ge 5	Same specimen as the one for curve 42.
99	789	L	1962	0.3-3.1	5	Ge 7	Same specimen as the one for curve 43.
100	789	L	1962	0.3-0.8	5	Ge 10	Same specimen as the one for curve 44.
101	789	L	1962	0.2-3.8	5	Ge 11	Same specimen as the one for curve 45.
102	789	L	1962	0.3-4.0	5	Ge 12	Same specimen as the one for curve 45.
103	339, 883	L	1955	5.1-291	<10 in 4 <sup>1</sup> - 20 K and 100 <sup>1</sup> -300 <sup>1</sup> K -15 in 20 <sup>1</sup> - 100 K.		Sb-doped n-type single crystal, 2 x 4 x 15 mm; long dimension in the [112] direction; obtained by Czochralski technique; electrical resistivity 0.10 ohm $\text{cm}$ .

SPECIFICATION TABLE NO. 14 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
104	339, 883, 988	L	1965	4.9-268	<10 in 4°-20 K and 100°-300°K, <15 in 20°-100 K.		Similar to the above specimen; irradiated at 30 C with a fast-neutron integrated flux of $1.1 \times 10^{17} \text{ n cm}^{-2}$ .
105	339, 883	L	1965	5.0-275	<10 in 4°-20 K and 100°-300°K, <15 in 20°-100 K.		Similar to the above specimen; irradiated at 30 C with a fast-neutron integrated flux of $2.5 \times 10^{16} \text{ n cm}^{-2}$ .
106	883	L	1957	5.1-287	<10 in 4°-20°K and 100°-300°K, <15 in 20°-100 K.		Similar to the above specimen; irradiated at 30 C with a fast neutron integrated flux of $1.7 \times 10^{16} \text{ n cm}^{-2}$ .
107	883	L	1957	5.4-500	<10 in 4°-100°K and 100°-300°K, <15 in 100°-300°K.		Similar to the above specimen; irradiated at 30 C with a fast-neutron integrated flux of $3.4 \times 10^{16} \text{ n cm}^{-2}$ .
108	884	L	1965	47-136		Ge II	High purity n-type single crystal; obtained from Eagle-Picher Company; bar shaped, 0.153 cm wide, 0.048 cm thick; long dimension in the <110> direction; irradiated (base temperature near 47 K, tip temperature 70 K) for a length of 1 cm in <111> direction with a total electron flux of $1.01 \times 10^{18} \text{ 2-Mev e cm}^{-2}$ ; annealed for 15 min at 175 K; electrical resistivity 50 ohm cm, carrier concentration approx $10^{16} \text{ cm}^{-3}$ ; measured on warming in the dark from 47 K after the electron traps were filled by a white-light illumination at 47 K.
109	884	L	1965	18-310		Ge I	High purity n-type single crystal; obtained from Eagle-Picher Company; bar shaped, 0.159 cm wide, 0.043 cm thick; long dimension was the <110> direction; electrical resistivity 50 ohm cm; carrier concentration approx $10^{14} \text{ cm}^{-3}$ .
110	884	L	1965	20-31		Ge I	Same specimen as above; irradiated (base temperature near 20 K, tip temperature 50 K) for a length of 1 cm in <111> direction with a total electron flux of $3.4 \times 10^{18} \text{ 2-Mev e cm}^{-2}$ ; at 30 K for 15 min.
111	884	L	1965	20-61		Ge I	The above specimen annealed for the second time at 70 K for 15 min.
112	884	L	1965	19-150		Ge I	The above specimen annealed for the third time at 175 K for 15 min.
113	884	L	1965	18-65		Ge I	The above specimen annealed for the fourth time at 325 K for 15 min.

SPECIFICATION TABLE NO. 18 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
114	884	L	1965	20-230		Ge I	The above specimen annealed for the fifth time at 405 K for 15 min.
115	884	L	1965	11-298		Ge II	High purity n-type single crystal; obtained from Eagle-Picher Company; bar shaped, 0.133 cm wide, 0.048 cm thick; long dimension in the <110> direction; electrical resistivity 50 ohm cm; carrier concentration approx $10^{16}$ cm <sup>-3</sup> .
116	884	L	1965	11-74		Ge II	Same specimen as above; irradiated (base temperature near 47 K, tip temperature 70 K) for a length of 1 cm in <111> direction with a total electron flux of $1.01 \times 10^{18}$ 2-Mev e cm <sup>-2</sup> ; annealed at 77 K for 15 min.
117	884	L	1965	11-118		Ge II	The above specimen annealed for the second time at 125 K for 15 min.
118	884	L	1965	11-121		Ge II	The above specimen annealed for the third time at 140 K for 15 min.
119	884	L	1965	11-171		Ge II	The above specimen annealed for the fourth time at 175 K for 15 min.
120	884	L	1965	11-150		Ge II	The above specimen annealed for the fifth time at 209 K for 15 min.
121	884	L	1965	11-202		Ge II	The above specimen annealed for the sixth time at 230 K for 15 min.
122	884	L	1965	10-303		Ge II	The above specimen annealed for the seventh time at 405 K for 15 min.
123	885		1964	92-300		A	Sample type A.
124	885		1964	89-301		B	Single crystal; irradiated at 70 C for a fast neutron flux of $6 \times 10^{16}$ n cm <sup>-2</sup> .
125	885		1964	90-271		C	Sample type C; the above specimen except irradiated at 70 C by a fast neutron flux of $1.2 \times 10^{16}$ n cm <sup>-2</sup> .
126	887, 886		1964	300			Ga-doped p-type single crystal; carrier concentration $1.29 \times 10^{14}$ cm <sup>-3</sup> .
127	887, 886		1964	300			Ga-doped p-type single crystal; carrier concentration $1.70 \times 10^{13}$ cm <sup>-3</sup> .
128	887, 886		1964	300			Ga-doped p-type single crystal; carrier concentration $7.76 \times 10^{13}$ cm <sup>-3</sup> .
129	887, 886		1964	300			Ga-doped p-type single crystal; carrier concentration $1.12 \times 10^{13}$ cm <sup>-3</sup> .
130	887, 886		1964	303			As-doped n-type single crystal; carrier concentration $4.07 \times 10^{15}$ cm <sup>-3</sup> .
131	887, 886		1964	300			As-doped n-type single crystal; carrier concentration $4.37 \times 10^{16}$ cm <sup>-3</sup> .
132	887, 886		1964	300			As-doped n-type single crystal; carrier concentration $9.77 \times 10^{16}$ cm <sup>-3</sup> .

SPECIFICATION TABLE NO. 18 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition ( weight percent), Specifications and Remarks
133	887, 886		1964	300			As-doped n-type single crystal; carrier concentration $3.24 \times 10^{19} \text{ cm}^{-3}$ .
134	887, 886		1964	300			As-doped n-type single crystal; carrier concentration $1.02 \times 10^{19} \text{ cm}^{-3}$ .
135	887, 886		1964	300			As-doped n-type single crystal; carrier concentration $1.51 \times 10^{19} \text{ cm}^{-3}$ .
136	887, 886		1964	300			As-doped n-type single crystal; carrier concentration $6.03 \times 10^{19} \text{ cm}^{-3}$ .
137	888, 889	L	1962	110-317		I 1	Sb-doped n-type single crystal; electrical resistivity 0.0205 - 0.0227 ohm cm at 300 K.
138	888, 889	L	1962	108-304		I 2	Similar to the above specimen.
139	888, 889	L	1962	142-308		I 3	Similar to the above specimen.
140	888, 889	L	1962	121-306		I 4	Ga-doped p-type single crystal; electrical resistivity 0.154-0.155 ohm cm at 300 K.
141	888, 889	L	1962	98-316		I 5	Similar to the above specimen.
142	888, 889	L	1962	122-312		I 6	Similar to the above specimen.
143	888, 889	L	1962	109-307		II 1	Sb-doped n-type single crystal; electrical resistivity 13.9-15.1 ohm cm at 300 K.
144	888, 889	L	1962	118-302		II 2	Similar to the above specimen.
145	888, 889	L	1962	114-304		II 3	Ga-doped p-type single crystal; electrical resistivity 25.4-26.0 ohm cm at 300 K.
146	888, 889	L	1962	131-310		II 4	Similar to the above specimen.
147	888, 889	L	1962	128-306		II 5	Similar to the above specimen.



DATA TABLE NO. 18 THERMAL CONDUCTIVITY OF GERMANIUM

(Impurity &lt; 0.20% each; total impurities &lt; 0.50%)

[Temperature, T, K. Thermal Conductivity, k, Watt cm<sup>-1</sup> K<sup>-1</sup>]

T	k	T	k	T	k	T	k	T	k	T	k	T	k	T	k	T	k
<u>CURVE 1*</u>		<u>CURVE 4 (cont.)</u>		<u>CURVE 7 (cont.)</u>		<u>CURVE 10 (cont.)</u>		<u>CURVE 13 (cont.)</u>		<u>CURVE 15 (cont.)</u>		<u>CURVE 18</u>					
298	0.586	4.21	0.0688	16.1	11.4	900.9	0.184	112	1.26*	120	1.18	208	0.820				
373	0.469	14.5	0.998	21.7	10.4	952.4	0.189	119	1.17	124	1.13	220	0.778				
<u>CURVE 2</u>		<u>CURVE 6</u>		<u>CURVE 8</u>		<u>CURVE 11</u>		<u>CURVE 14</u>		<u>CURVE 16</u>		<u>CURVE 19*</u>					
278.3	0.605*	3.26	1.68	328.9	0.464	323.6	0.464	94	1.97	146	1.03	90	1.39				
294.4	0.586	3.78	2.39	347.2	0.455	389.1	0.376	101	1.82	159	1.03	93	1.41				
308.7	0.566*	15.8	1.140	354.6	0.403	515.5	0.323	88	2.18	108	1.65	114	1.23*				
325.8	0.538*	18.1	1.300	396.8	0.384	854.7	0.178	95	2.01	108	1.65	116	1.23*				
346.5	0.515*	19.7	1.44	407.6	0.364	1000.0	0.171	100	1.59	122	1.42	130	1.15				
364.6	0.496	20.4	1.65	420.0	0.344	1039.5	0.178	105	1.72	127	1.36	146	1.05				
<u>CURVE 3</u>		<u>CURVE 5</u>		<u>CURVE 9</u>		<u>CURVE 12</u>		<u>CURVE 15</u>		<u>CURVE 17*</u>		<u>CURVE 20</u>					
2.70	0.154	2.23	0.314	318.5	0.489	87	1.82	89	1.63	89	1.63	387.6	0.494				
3.20	0.290	2.56	0.433	354.6	0.474	90	1.73	92	1.56*	92	1.52	404.9	0.481				
3.70	0.487	2.88	0.514	398.4	0.375	99	1.48	96	1.47	96	1.47	409.6	0.473				
4.20	0.590	3.00	0.610	420.0	0.355	103	1.43	98	1.46	100	1.42	427.4	0.444				
10.20	2.76	3.22	0.714	448.4	0.336	111	1.33	104	1.37	104	1.37	510.2	0.398				
14.0	3.66	3.60	0.942	469.6	0.320	119	1.21	106	1.38	106	1.38	520.8	0.371				
15.0	3.95	4.48	1.620	489.6	0.304	127	1.13	109	1.40	112	1.45	574.7	0.348				
15.9	3.85	5.48	2.375	518.8	0.288	140	1.02	113	1.42	114	1.45	675.7	0.324				
16.0	4.20	6.21	3.02	548.4	0.272	140	1.02	115	1.42	116	1.45	757.6	0.304				
16.9	4.10	7.23	3.91	581.5	0.256	155	0.92	122	1.42	122	1.42	789.2	0.288				
17.3	4.14	8.75	4.74	618.8	0.240	155	0.92	127	1.42	127	1.42	840.3	0.288				
19.1	4.07*	9.90	5.70	654.6	0.224	155	0.92	133	1.42	133	1.42	862.1	0.288				
20.2	4.45*	11.0	6.39	699.3	0.208	155	0.92	133	1.42	133	1.42	897.9	0.288				
20.3	3.91	12.1	6.85	750.0	0.192	155	0.92	133	1.42	133	1.42	917.4	0.288				
20.5	3.95	13.5	7.29	800.0	0.176	155	0.92	133	1.42	133	1.42	970.9	0.288				
20.6	3.11	15.0	7.85	848.4	0.160	155	0.92	133	1.42	133	1.42	997.9	0.288				
20.6	3.11	15.0	7.85	897.9	0.144	155	0.92	133	1.42	133	1.42	1024.9	0.288				
63.5	2.97	18.5	8.21	948.4	0.128	155	0.92	133	1.42	133	1.42	1051.9	0.288				
66.0	2.55	22.3	8.11	1000.0	0.112	155	0.92	133	1.42	133	1.42	1078.9	0.288				
70.0	2.60	26.8	7.85	1051.9	0.096	155	0.92	133	1.42	133	1.42	1105.9	0.288				
75.0	2.27	30.4	7.60	1105.9	0.080	155	0.92	133	1.42	133	1.42	1132.9	0.288				
79.0	2.21	33.5	7.12	1157.9	0.064	155	0.92	133	1.42	133	1.42	1159.9	0.288				
<u>CURVE 4</u>		<u>CURVE 10</u>		<u>CURVE 14</u>		<u>CURVE 18</u>		<u>CURVE 22*</u>		<u>CURVE 26*</u>		<u>CURVE 30*</u>					
2.72	0.9265*	2.86	0.654	331.1	0.436	91	1.55	96	1.77	96	1.77	387.6	0.494				
3.59	0.0457	2.86	1.09	336.7	0.410	94	1.46	96	1.77	96	1.77	404.9	0.481				
3.87	0.0581	3.39	1.76	348.1	0.390	95	1.46	98	1.77	98	1.77	427.4	0.444				
4.00	0.0553	4.28	2.79	361.1	0.374	98	1.38	100	1.77	100	1.77	451.9	0.408				
		7.88	7.56	389.1	0.358	101	1.36	104	1.77	104	1.77	476.9	0.372				
		8.83	8.48	448.4	0.342	104	1.30	106	1.77	106	1.77	501.9	0.346				
		12.03	10.4	481.1	0.326	109	1.27	109	1.77	109	1.77	526.9	0.320				
				516.9	0.310	113	1.27	113	1.77	113	1.77	551.9	0.294				
				541.9	0.294	117	1.27	117	1.77	117	1.77	576.9	0.268				
				566.9	0.278	121	1.27	121	1.77	121	1.77	601.9	0.242				
				591.9	0.262	125	1.27	125	1.77	125	1.77	626.9	0.216				
				616.9	0.246	129	1.27	129	1.77	129	1.77	651.9	0.190				
				641.9	0.230	133	1.27	133	1.77	133	1.77	676.9	0.164				
				666.9	0.214	137	1.27	137	1.77	137	1.77	701.9	0.138				
				691.9	0.198	141	1.27	141	1.77	141	1.77	726.9	0.112				
				716.9	0.182	145	1.27	145	1.77	145	1.77	751.9	0.086				
				741.9	0.166	149	1.27	149	1.77	149	1.77	776.9	0.060				
				766.9	0.150	153	1.27	153	1.77	153	1.77	801.9	0.034				
				791.9	0.134	157	1.27	157	1.77	157	1.77	826.9	0.008				

\* Not shown on plot

DATA TABLE NO. 14 (continued)

T	k	T	k	T	k	T	k	T	k	T	k	T	k
<u>CURVE 21</u>													
404.9	0.463	353.6	0.497	293.2	0.155	67.2	2.97	25.0	4.10	4.50	1.30	9.80	12.30
411.5	0.444	357.1	0.493	<u>CURVE 28</u>		68.2	3.07	30.0	4.30	5.40	1.85	11.20	12.90
421.9	0.435	370.4	0.492	69.0	3.01	69.0	3.01	40.0	3.75	6.70	3.00	12.90	12.90
435.0	0.400	381.6	0.525	293.2	0.180	69.4	2.89	70.0	2.30	7.50	3.50	13.90	13.00
437.5	0.387	393.7	0.430	<u>CURVE 29</u>		70.4	2.83	80.0	1.90	9.35	5.10	14.90	13.20
451.9	0.363	408.2	0.421	71.1	2.78	71.1	2.78	90.0	1.70	10.5	5.50	16.00	13.20
461.8	0.349	416.7	0.401	72.1	2.73	72.1	2.73	100.0	1.52	11.5	6.20	16.40	13.00
468.8	0.345	425.5	0.381	72.6	2.80	72.6	2.80	<u>CURVE 37</u>		11.5	7.30	18.80	13.90
482.9	0.345	438.6	0.352	72.9	2.70	72.9	2.70	2.0	0.0305	13.0	6.75	19.80	12.70
482.3	0.346	446.4	0.342	73.2	2.79	73.2	2.79	3.0	0.063	15.5	8.10	30.00	9.30
484.3	0.350	454.5	0.332	73.5	2.68	73.5	2.68	4.0	0.165	17.0	8.50	40.00	8.20
489.7	0.361	465.1	0.302	74.0	2.70	74.0	2.70	5.0	0.330	20.0	9.80	47.00	6.00
482.1	0.367	520.8	0.281	74.4	2.64	74.4	2.64	6.0	0.530	21.4	9.99	48.00	5.80
484.6	0.385	552.5	0.231	75.1	2.62	75.1	2.62	7.0	0.750	26.0	8.40	63.0	4.25
497.9	0.392	552.5	0.210	75.6	2.61	75.6	2.61	8.0	1.000	31.0	7.90	75.0	3.55
<u>CURVE 22</u>													
76	2.420	833.3	0.151	76.0	2.69	76.0	2.69	9.0	1.20	40.0	6.40	130.0	1.50
200	0.912	909.1	0.140	76.3	2.67	76.3	2.67	10.0	1.50	50.0	5.25	157.0	1.18
254	0.686	1000.0	0.130	76.6	2.57	76.6	2.57	15.0	3.00	55.0	4.30	204.0	0.90
283	0.582	<u>CURVE 25</u>		77.3	2.60	77.3	2.60	20.0	3.8	77.0	3.20	225.0	0.807
312	0.549	370.4	0.433	77.7	2.58	77.7	2.58	25.0	4.10	83.0	2.52	265.0	0.70
346	0.473	377.3	0.401	78.6	2.47	78.6	2.47	30.0	3.70	100.0	2.30	280.0	0.58
370	0.439	377.3	0.371	79.0	2.47	79.0	2.47	40.0	2.60	120.0	1.70	<u>CURVE 42</u>	
<u>CURVE 23</u>													
307.6	0.461	370.4	0.433	80.4	2.51	80.4	2.51	60.0	2.30	<u>CURVE 39</u>			
400.0	0.399	377.3	0.401	81.3	2.48	81.3	2.48	70.0	2.05	2.15	1.09	2.05	0.498
425.5	0.326	408.1	0.351	80.4	2.51	80.4	2.51	80.0	1.80	2.30	1.30	2.10	0.570
434.5	0.298	416.7	0.327	81.3	2.48	81.3	2.48	90.9	1.65	2.35	1.37	2.52	0.935
476.2	0.276	425.5	0.304	283.2	0.586	283.2	0.586	100.0	1.50	2.50	1.60	3.05	1.51
485.0	0.252	434.8	0.291	<u>CURVE 35</u>		<u>CURVE 35</u>		<u>CURVE 35</u>		2.80	2.00	5.23	4.50
529.8	0.238	444.4	0.273	283.2	0.586	283.2	0.586	100.0	1.50	6.70	6.40	4.65	4.55
571.4	0.229	454.5	0.263	<u>CURVE 36</u>		<u>CURVE 36</u>		<u>CURVE 36</u>		8.00	8.00	4.65	4.55
598.8	0.200	476.2	0.248	2.0	0.035	2.0	0.035	1.50	0.120	9.10	9.60	6.00	6.00
684.9	0.171	500.0	0.234	1.63	0.076	1.63	0.076	1.63	0.160	3.20	2.78	6.30	7.25
729.9	0.162	526.3	0.220	4.0	0.190	4.0	0.190	1.70	0.175	3.50	3.24	6.60	7.70
737.6	0.132	571.4	0.198	5.0	0.353	5.0	0.353	1.80	0.215	3.75	3.75	6.60	7.80
<u>CURVE 24</u>													
307.6	0.461	64.6	3.39	2.05	0.275	2.05	0.275	2.05	0.275	4.50	5.00	9.20	10.70
400.0	0.399	65.2	3.20	7.0	0.82	7.0	0.82	2.25	0.350	4.55	5.15	25.00	9.98
425.5	0.326	65.6	3.18	8.0	1.09	8.0	1.09	2.55	0.470	5.50	7.00	28.00	9.20
476.2	0.276	66.4	3.16	9.0	1.35	9.0	1.35	2.65	0.555	6.40	8.35	31.00	8.50
485.0	0.252	67.2	3.17	10.0	1.60	10.0	1.60	3.10	0.700	7.30	10.00	31.00	8.50
529.8	0.238	67.2	3.17	15.6	2.00	15.6	2.00	3.25	0.855	7.90	10.50	22.00	11.90
571.4	0.229	67.2	3.17	20.0	3.90	20.0	3.90	3.55	0.990	8.85	11.60	26.00	10.60
598.8	0.200	266.2	0.502	20.0	3.90	20.0	3.90	3.55	0.990	8.85	11.60	30.00	9.40
684.9	0.171	<u>CURVE 26</u>		<u>CURVE 26</u>		<u>CURVE 26</u>		<u>CURVE 26</u>		<u>CURVE 41</u>		<u>CURVE 41</u>	
729.9	0.162	266.2	0.502	2.00	0.475	2.00	0.475	2.00	0.475	2.00	0.475	2.00	0.475
737.6	0.132	<u>CURVE 26</u>		<u>CURVE 26</u>		<u>CURVE 26</u>		<u>CURVE 26</u>		<u>CURVE 41</u>		<u>CURVE 41</u>	
266.2	0.502	2.00	0.475	2.00	0.475	2.00	0.475	2.00	0.475	2.00	0.475	2.00	0.475

Not shown on plot

















DATA TABLE NO. 18 (continued)

CURVE 117 (cont.) <sup>*</sup>			CURVE 118 (cont.) <sup>*</sup>			CURVE 119 (cont.) <sup>*</sup>			CURVE 120 (cont.) <sup>*</sup>			CURVE 121 (cont.) <sup>*</sup>			CURVE 122 (cont.) <sup>*</sup>			CURVE 124 (cont.) <sup>*</sup>			
T	k	T	T	k	T	T	k	T	T	k	T	k	T	T	k	T	k	T	k	T	k
21.9	2.17	39.5	2.85	55.3	2.89	85.1	2.17 <sup>*</sup>	135.5	10.1	3.79	10.1	3.79	120.8	1.56	217.8	0.813	217.8	0.813	217.8	0.813	
23.9	2.25	41.3	2.83	60.7	2.74	91.0	2.01 <sup>*</sup>	151.7	10.3	3.86	10.3	3.86	136.5	1.38	243.8	0.719	243.8	0.719	243.8	0.719	
25.1	2.38	43.5	2.82	65.3	2.59	110.7	1.63 <sup>*</sup>	178.2	10.6	3.98	10.6	3.98	151.7	1.22	271.6	0.637	271.6	0.637	271.6	0.637	
27.0	2.44	46.9	2.83	70.6	2.41	129.7	1.30 <sup>*</sup>	202.3	10.6	3.86	10.6	3.86	166.7	1.12	301.3	0.564	301.3	0.564	301.3	0.564	
28.7	2.37	48.4	2.70	75.5	2.28	150.3	1.16 <sup>*</sup>		12.2	3.98	12.2	3.98	181.6	1.02							
32.0	2.38	50.4	2.63 <sup>*</sup>	80.5	2.17				13.0	4.30	13.0	4.30	201.8	0.925							
34.4	2.48	55.0	2.63	85.5	2.02				14.1	4.70	14.1	4.70	225.9	0.838							
37.0	2.52	60.8	2.47	95.7	1.83				15.0	5.16	15.0	5.16	252.3	0.775							
40.3	2.47	65.6	2.39 <sup>*</sup>	110.7	1.58				16.1	5.10 <sup>*</sup>	16.1	5.10 <sup>*</sup>	278.0	0.723							
43.4	2.50	70.3	2.28 <sup>*</sup>	120.5	1.38				17.1	5.42	17.1	5.42	302.7	0.681							
46.7	2.46	75.3	2.15 <sup>*</sup>	135.2	1.24				18.1	5.21	18.1	5.21									
47.3	2.46	80.5	2.07 <sup>*</sup>	170.6	1.02				19.0	5.41	19.0	5.41									
50.1	2.46	85.5	1.95 <sup>*</sup>						20.1	5.23	20.1	5.23									
55.6	2.58	90.6	1.92 <sup>*</sup>						22.1	5.41	22.1	5.41									
60.3	2.31	90.3	1.79 <sup>*</sup>						24.2	5.11	24.2	5.11									
64.9	2.21	100.5	1.67 <sup>*</sup>						25.6	4.86	25.6	4.86									
70.5	2.07	110.9	1.53 <sup>*</sup>						26.8	5.41	26.8	5.41									
75.2	2.03	131.2	1.23 <sup>*</sup>						27.7	5.41	27.7	5.41									
									28.8	5.08	28.8	5.08									
									29.8	5.10	29.8	5.10									
									31.5	4.89	31.5	4.89									
									33.1	4.99	33.1	4.99									
									35.2	4.67	35.2	4.67									
									37.5	4.66	37.5	4.66									
									39.5	4.48	39.5	4.48									
									41.7	4.47	41.7	4.47									
									44.1	4.45	44.1	4.45									
									47.4	4.22	47.4	4.22									
									54.1	3.73	54.1	3.73									
									55.5	3.59	55.5	3.59									
									60.4	3.38	60.4	3.38									
									68.1	3.11 <sup>*</sup>	68.1	3.11 <sup>*</sup>									
									70.6	2.84 <sup>*</sup>	70.6	2.84 <sup>*</sup>									
									75.5	2.63 <sup>*</sup>	75.5	2.63 <sup>*</sup>									
									80.5	2.51 <sup>*</sup>	80.5	2.51 <sup>*</sup>									
									85.1	2.43 <sup>*</sup>	85.1	2.43 <sup>*</sup>									
									90.4	2.31 <sup>*</sup>	90.4	2.31 <sup>*</sup>									
									100.2	2.18 <sup>*</sup>	100.2	2.18 <sup>*</sup>									
									110.7	1.93 <sup>*</sup>	110.7	1.93 <sup>*</sup>									
									120.8	1.68 <sup>*</sup>	120.8	1.68 <sup>*</sup>									
										1.51 <sup>*</sup>		1.51 <sup>*</sup>									
										2.23 <sup>*</sup>		2.23 <sup>*</sup>									
										3.06		3.06									

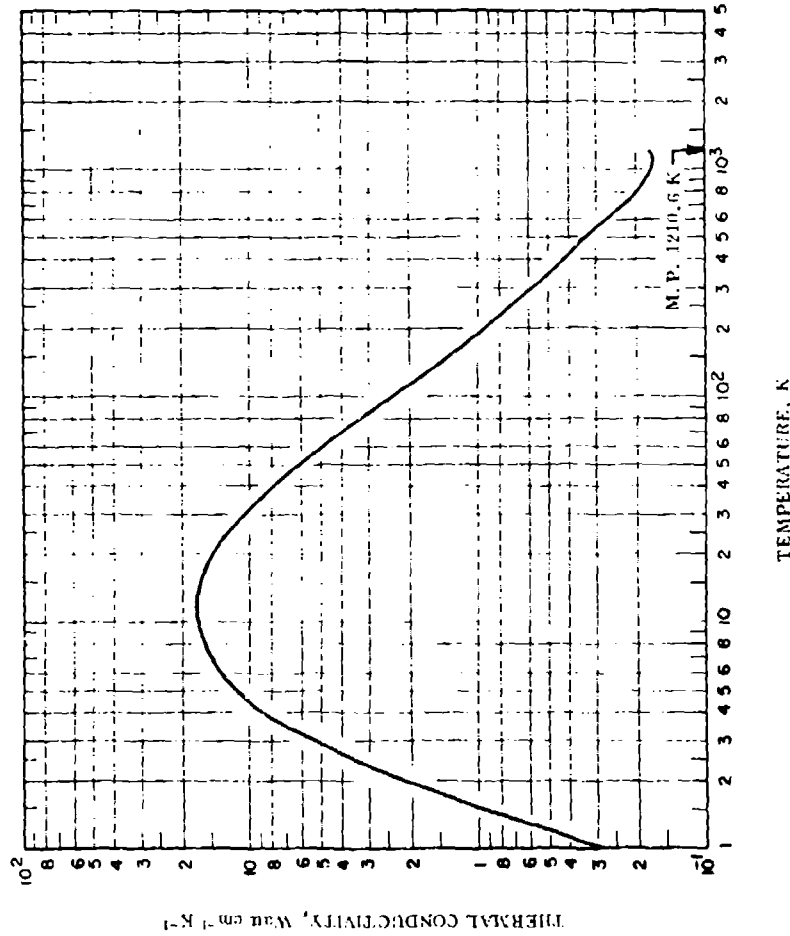
\* Not shown on plot

DATA TABLE NO. 18 (continued)

T	k	T	k	T	k	T	k	T	k
<u>CURVE 131*</u>		<u>CURVE 138 (cont.)*</u>		<u>CURVE 143 (cont.)*</u>		<u>CURVE 147 (cont.)*</u>			
300	0.598	300.5	0.598	126.2	1.52	221.5	0.841		
		305.7	0.590	127.2	1.55	229.6	0.803		
<u>CURVE 132*</u>		313.5	0.565	129.6	1.53	236.2	0.774		
300	0.582	<u>CURVE 139*</u>		129.6	1.55	305.5	0.598		
		142.0	1.36	136.6	1.36				
<u>CURVE 133*</u>		152.6	1.27	302.0	0.573				
300	0.509	227.0	0.799	307.4	0.573				
		308.2	0.590	<u>CURVE 144*</u>					
<u>CURVE 134*</u>		<u>CURVE 140*</u>		118.1	1.70				
300	0.346	121.4	1.67	122.2	1.61				
<u>CURVE 135</u>		132.7	1.48	145.1	1.30				
300	0.272	143.6	1.36	203.4	0.891				
		151.6	1.23	215.8	0.841				
<u>CURVE 136*</u>		206.9	0.904	223.4	0.791				
300	0.202	220.3	0.849	229.1	0.770				
		227.0	0.828	301.7	0.598				
<u>CURVE 137*</u>		301.4	0.619	<u>CURVE 145*</u>					
300	0.202	305.5	0.619	113.8	1.77				
		<u>CURVE 141*</u>		125.9	1.54				
110.3	1.87	98.2	2.23	140.9	1.36				
117.1	1.70	108.8	1.90	155.0	1.19				
127.7	1.49	201.3	0.908	170.9	1.09				
317.1	0.556	207.0	0.879	206.4	0.874				
<u>CURVE 138*</u>		214.0	0.874	213.5	0.849				
107.7	1.88	299.0	0.615	221.6	0.816				
116.5	1.78	310.8	0.590	231.3	0.778				
118.2	1.65	316.0	0.569	304.3	0.577				
128.1	1.54	<u>CURVE 142*</u>		<u>CURVE 146*</u>					
136.4	1.39	121.9	1.62	130.6	1.54				
140.6	1.35	123.4	1.56	148.1	1.30				
153.1	1.20	142.4	1.33	310.2	0.573				
156.9	1.18	146.4	1.28	<u>CURVE 147*</u>					
168.8	1.10	162.5	1.14	128.4	1.57				
210.5	0.900	311.8	0.561	131.9	1.51				
210.5	0.866	<u>CURVE 143*</u>		135.6	1.45				
217.5	0.849	109.4	1.91	143.8	1.35				
217.5	0.824	125.4	1.50	166.4	1.17				
221.0	0.799			209.9	0.874				
229.0	0.774			218.4	0.854				
241.5	0.741								

Not shown on plot

FIGURE AND TABLE NO. 18B RECOMMENDED THERMAL CONDUCTIVITY OF GERMANIUM



RECOMMENDED VALUES*					
$T_1$	$k_1$	$k_2$	$T_2$	$k_1$	$T_2$
0	0	0	-459.7	500	0.338
1	0.274	15.8	-457.9	600	0.273
2	2.06	119	-456.1	700	0.227
3	5.35	309	-454.3	800	0.198
4	8.77	507	-452.5	900	0.182
5	11.6	670	-450.7	1000	0.174
6	13.9	803	-448.9	1100	0.170
7	15.5	896	-447.1	1200	0.174
8	16.6	959	-445.3		
9	17.3	1000	-443.5		
10	17.7	1020	-441.7		
11	17.9	1030	-439.9		
12	18.0	1040	-438.1		
13	17.9	1030	-436.3		
14	17.7	1020	-434.5		
15	17.3	1000	-432.7		
16	16.9	976	-430.9		
18	15.9	919	-427.3		
20	14.9	861	-423.7		
25	12.7	734	-414.7		
30	10.8	624	-405.7		
35	9.20	532	-396.7		
40	7.98	461	-387.7		
45	6.95	402	-378.7		
50	6.15	355	-369.7		
60	4.87	281	-351.7		
70	3.93	227	-333.7		
80	3.25	188	-315.7		
90	2.70	156	-297.7		
100	2.32	134	-279.7		
150	1.32	76.3	-189.7		
200	0.968	55.9	-99.7		
250	0.749	43.3	-9.7		
273.2	0.667	38.5	32.0		
300	0.594	34.6	80.3		
350	0.495	28.6	170.3		
400	0.432	25.0	260.3		

REMARKS

The recommended values are for high-purity germanium. The values are thought to be accurate to within 4% of the true values near room temperature and 4 to 10% at other temperatures above 40 K. The thermal conductivity near and below the corresponding temperature of its maximum is highly sensitive to small physical and chemical variations of the specimens, and the values below 40 K are intended as typical values for indicating the general trend.

\*  $T_1$  in K,  $k_1$  in Watt cm<sup>-1</sup> K<sup>-1</sup>,  $T_2$  in F, and  $k_2$  in Btu hr<sup>-1</sup> ft<sup>-1</sup> F<sup>-1</sup>.

# THERMAL CONDUCTIVITY OF GOLD

FIGURE SHOWS ONLY 19 OF THE CURVES REPORTED IN TABLE

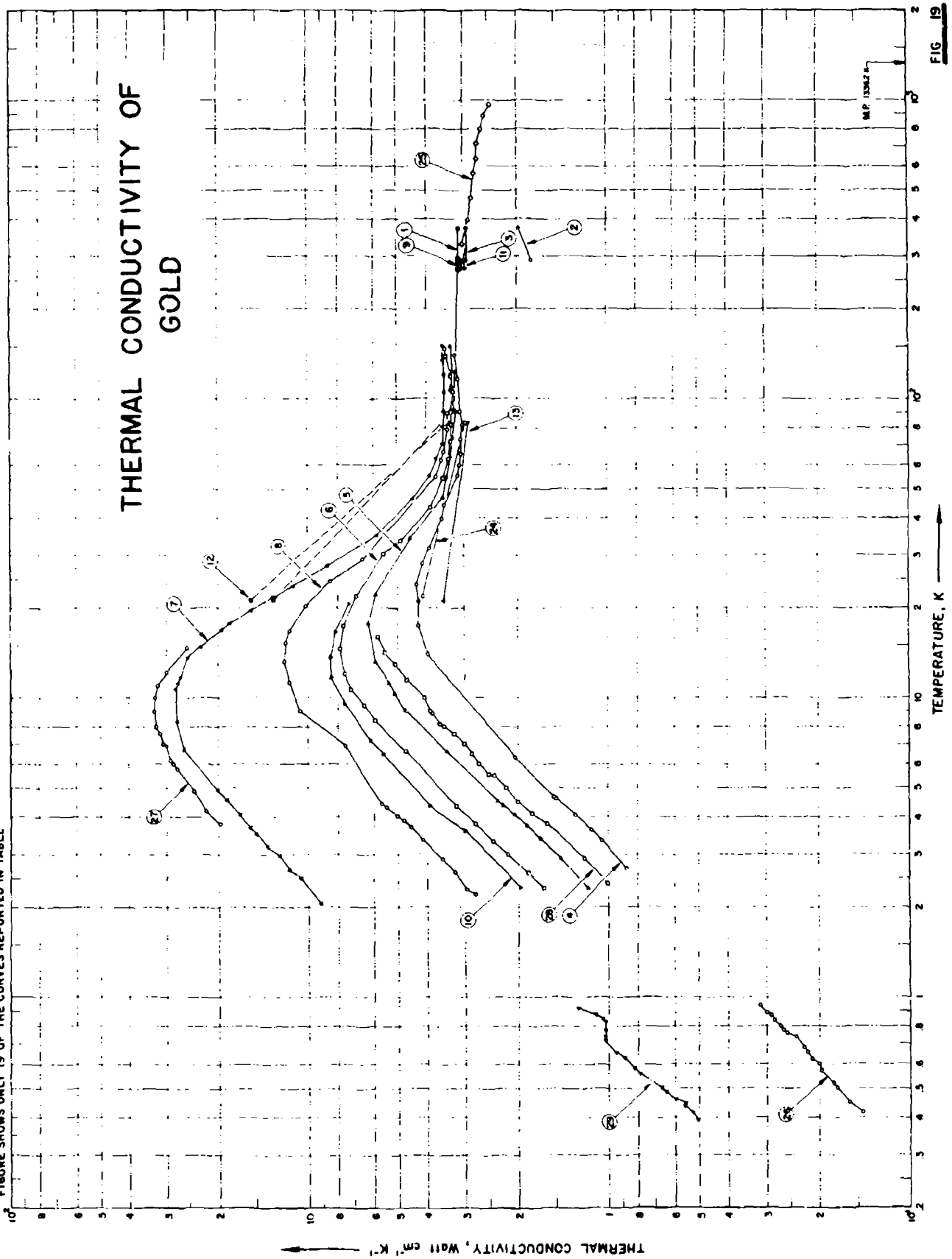


FIG. 19

SPECIFICATION TABLE NO. 19 THERMAL CONDUCTIVITY OF GOLD

(Impurity < 0.20% each, total impurities < 0.50%)

(For Data Reported in Figure and Table No. 19 J)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
1	95	E	1915	22-374			99.999 pure; electrical conductivity $48.4$ and $45.1 \times 10^4 \text{ ohm}^{-1}\text{cm}^{-1}$ at $273.1$ and $291 \text{ K}$ , respectively.
2	77	E	1900	291, 373		Au I	99.8 pure; 0.1 Fe, 0.1 Cu.
3	77	E	1900	291, 373		Au II	High purity.
4	146	L	1953	2.7-141	1	Au I	99.9 pure; major impurity Ag, trace Pt, faint traces of Fe, Cu, and Sn; specimen 2 mm in dia obtained from Garrett, Davidson and Matthey of Sydney.
5	146	L	1953	2.3-150	1	Au 2	The above specimen annealed at $700 \text{ C}$ in vacuo and slowly cooled.
6	146	L	1953	2.3-148	1	Au 3	99.999+ pure; spectral analysis showed lines of Ag, and Cu and faint lines of Cd, Fe, Mg, and Na, and very faint lines of Ca and Zn; specimen 1.5 mm dia rod; obtained from Garrett, Davidson and Matthey of Sydney.
7	146	L	1953	2.1-151	1	Au 4	The above specimen annealed at $700 \text{ C}$ in vacuo for about 3 hrs and slowly cooled to $200 \text{ C}$ in 6 hrs.
8	146	L	1953	2.2-90	1	Au 5	The above annealed specimen cold drawn to 1.3 mm dia.
9	78	E	1931	273-292			99.99 pure; specimen 0.07960 cm in dia and 20.12 cm long.
10	97	L	1952	2.3-21	2-3	Au 1	99.999 pure; polycrystalline wire, obtained from Johnson Matthey (JM 1916a).
11	8	L	1914	273-373		Au 12	Specimen 0.1014 cm in dia; specific gravity 19.49.
12	57	L	1927	21, 83		Au II	High purity; single crystal; unstrained; electrical resistivity reported as 0.0142, 0.488 and $2.04 \mu\text{ohm cm}$ at $-252$ , $-190$ , and $0 \text{ C}$ , respectively.
13	57	L	1927	21, 83		Au II	Commercially pure; cold-worked and annealed; electrical resistivity reported as 0.1174, 0.599 and $2.16 \mu\text{ohm cm}$ at $-252$ , $-190$ , and $0 \text{ C}$ , respectively.
14	172	E	1927	297.2		1a	Specimen 4 m in dia and 20 cm long; made from forged material and machined to shape; electrical resistivity $2.44 \mu\text{ohm cm}$ at $24 \text{ C}$ .
15	172	E	1927	297.2		1b	The above specimen measured after being annealed at $800 \text{ C}$ for 1 hr.
16	487	L	1894	326.2			Specimen 2.0 mm dia.
17	246	T	1919	273, 373			Rolled and drawn; heated 0.5 hr close to melting point.
18	430	T	1924	273.2			Pure; rolled and drawn to a wire, specimen 3 cm long and 1 mm <sup>2</sup> cross-section, and then heated close to melting point.
19	451		1930	291.2			Pure; tempered at $800 \text{ C}$ , quenched, rolled, and drawn.
20	399	L	1925	290, 373			Pure.
21	241		1911	298.2			Pure.

SPECIFICATION TABLE NO. 19 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
22	57	L	1927	21.83		Au 14	High purity; single crystal; unstrained; electrical resistivity reported as 0.0160, 0.490, and 2.04 $\mu$ ohm cm at -252, -190, and 0 C, respectively.
23	57	L	1927	21.83		Au 13	Originally single crystal, hammered to 2 mm dia; annealed 5.5 hrs at 360 C; electrical resistivity reported as 0.0117, 0.483, and 2.04 $\mu$ ohm cm at -250, -190, and 0 C, respectively.
24	57	L	1927	21.83		Au 12a	Commercially pure; remelted and hammered to 2 mm dia; annealed, tempered 3 hrs at 390 C; electrical resistivity reported as 0.0941, 0.575, and 2.14 $\mu$ ohm cm at -250, -190, and 0 C, respectively.
25	617		1957	3.1-964			99.99 pure polycrystal.
26	683	L	1963	0.418-0.94			99.97 pure polycrystalline wire.
27	768	L	1964	3.8-15		Oxidized	40 $\mu$ thick foils rolled from spec-pure Johnson-Matthey material; annealed at 1223 K for 24 hrs in air; electrical resistance ratio $(R_{273}R_{4.2})/(R_{4.2})$ 610; electrical resistivity reported as 25.369, 25.379, 25.390, 25.404, 25.442, 25.496, and 25.529 $\mu$ ohm cm at 2.29, 2.79, 3.20, 3.61, 4.11, 4.63, and 4.90 K, respectively; $L_1$ (2.46 $\pm$ 0.02) $\times 10^{-8}$ watt ohm $K^{-2}$ .
24	764	L	1964	2.4-16		Vacuum annealed	40 $\mu$ thick foils rolled from spec-pure Johnson-Matthey material, annealed at 1223 K for 24 hrs in vacuum (pressure $10^{-5}$ mm Hg); electrical resistance ratio $(R_{273}R_{4.2})/(R_{4.2})$ 32.2; electrical resistivity reported as 467.9, 463.8, 461.2, 458.7, 456.9, 455.0, 453.3, 451.9, 450.7, 449.8, 449.0, and 448.2 $\mu$ ohm cm at 2.32, 2.79, 3.20, 3.62, 4.0, 4.40, 4.90, 5.38, 5.88, 6.33, 6.80, 7.45, 8.03, and 8.68 K, respectively.
29	746	L	1965	0.39-0.92			Very pure (higher purity than the specimen used by Davey and Mendelsohn 1963); polycrystalline.
30	1005	E	1927	273-2			99.9 pure; specimen 0.125 in. in dia and 10 cm long; obtained from Baker and Co.; electrical resistivity 2.214 $\mu$ ohm cm at 0 C.

DATA TABLE NO. 19 THERMAL CONDUCTIVITY OF GOLD

(Impurity: 0.20% each, total impurities = 0.50%)

Temperature: T K; Thermal Conductivity: k Watt cm<sup>-1</sup> K<sup>-1</sup>.

T	k	T	k	T	k	T	k	T	k	T	k	T	k	T	k
<u>CURVE 1</u>															
21.5	13.062	2.32	1.173	43.30	3.885	50.80	3.195	2.31	1.965	37	2.93	0.418	0.144		
91.5	3.234	2.89	1.420	54.45	3.525	106.00	3.480	3.59	2.991			0.450	0.160		
273.1	3.110	3.38	1.692	65.15	3.375	120.30	3.480	4.36	2.942	<u>CURVE 19</u>		0.505	0.176		
291.2	3.110	3.72	1.865	72.50	3.315	135.00	3.510	6.30	3.598			0.524	0.180		
294.6	3.110	4.06	2.060	82.90	3.315	150.70	3.540	7.26	6.195	291.2	3.09	0.575	0.197		
373.7	3.110	4.38	2.255	90.80	3.315	<u>CURVE 8</u>		9.57	7.610	<u>CURVE 20</u>		0.600	0.200		
		4.53	2.433	106.80	3.250	2.20	2.170	11.71	8.462			0.625	0.211		
<u>CURVE 2</u>															
291.2	1.790	6.62	3.430	120.30	3.400	2.29	2.970	15.76	8.462	290.2	2.930	0.656	0.220		
373.2	1.970	10.38	5.120	140.20	3.435	2.62	3.250	16.58	8.120	373.2	2.933	0.680	0.225		
		11.28	5.330	148.30	3.465	2.89	3.565	20.66	7.350			0.740	0.240		
<u>CURVE 3</u>															
		13.21	5.995	2.89	3.565	<u>CURVE 11</u>		<u>CURVE 21</u>				0.780	0.264		
		17.75	6.265	3.71	4.105	273.2	2.950	298.2	3.18			0.800	0.270		
		22.25	5.934	3.90	4.810	290.2	2.950					0.840	0.283		
		34.00	4.578	2.96	11.565	373.2	2.933					0.870	0.291		
<u>CURVE 4</u>															
291.2	2.930	2.05	9.120	2.95	9.120	<u>CURVE 12</u>		21.2	13.4			0.890	0.300		
373.2	2.940	2.50	10.640	2.50	10.640	3.62	7.590	31.2	3.37	<u>CURVE 27</u>		0.940	0.316		
		2.96	12.460	3.32	5.450	4.12	5.670	21.2	13.4						
		64.35	3.313	3.19	12.830	6.32	7.590	31.2	3.37						
		74.30	3.253	3.65	15.805	9.63	10.580	21.2	13.700						
		90.70	3.193	4.01	17.040	11.25	11.520	31.2	3.320						
		108.50	3.313	4.53	18.850	13.23	11.900	<u>CURVE 23</u>							
		121.40	3.283	4.95	20.060	15.35	11.877	21.2	13.5						
		150.40	3.343	6.70	26.030	16.85	11.500	31.2	3.32						
<u>CURVE 5</u>															
		2.32	1.654	8.35	27.730	20.30	10.150	<u>CURVE 24</u>							
		2.60	1.857	10.70	28.000	21.70	8.495	21.2	4.16						
		2.98	2.180	11.25	27.600	29.20	6.535	31.2	2.96						
		3.32	2.445	11.80	26.950	53.60	3.733	<u>CURVE 25</u>							
		3.79	2.782	13.70	25.300	62.60	3.555	230.7	3.03						
		3.79	2.782	14.90	22.950	67.00	3.495	396.2	2.88						
		4.36	3.203	17.45	18.400	79.40	3.390	471.5	2.82						
		6.62	4.760	19.60	15.750	89.90	3.385	636.5	2.78						
		8.42	5.935	21.80	13.150	<u>CURVE 9</u>		716.5	2.70						
		9.47	6.490	23.80	11.550	273.20	3.061	797.8	2.62						
		10.68	7.230	27.40	8.625	278.48	3.067	888.8	2.58						
		12.05	7.620	35.10	5.810	279.64	3.054	963.8	2.46						
		14.60	7.869	46.20	4.460	287.06	3.059								
		17.45	7.680	53.20	3.915	291.60	3.067								
		21.80	6.930	63.20	3.705										
		30.40	5.840	71.30	3.540										
		33.70	4.880	81.50	3.490										

Not shown on plot



DATA TABLE NO. 19 (continued)

T k

CURVE 28

2.4	1.0
2.9	1.2
3.8	1.6
4.1	1.8
4.5	2.0
5.0	2.2
5.5	2.4
5.55	2.5
6.0	2.7
6.5	2.85
7.0	3.0
7.6	3.25
8.05	3.5
8.2	3.6
8.9	3.84
9.0	3.9
10.0	4.1
11.5	4.7
13.0	5.1
14.25	5.5
16.0	5.8

CURVE 29

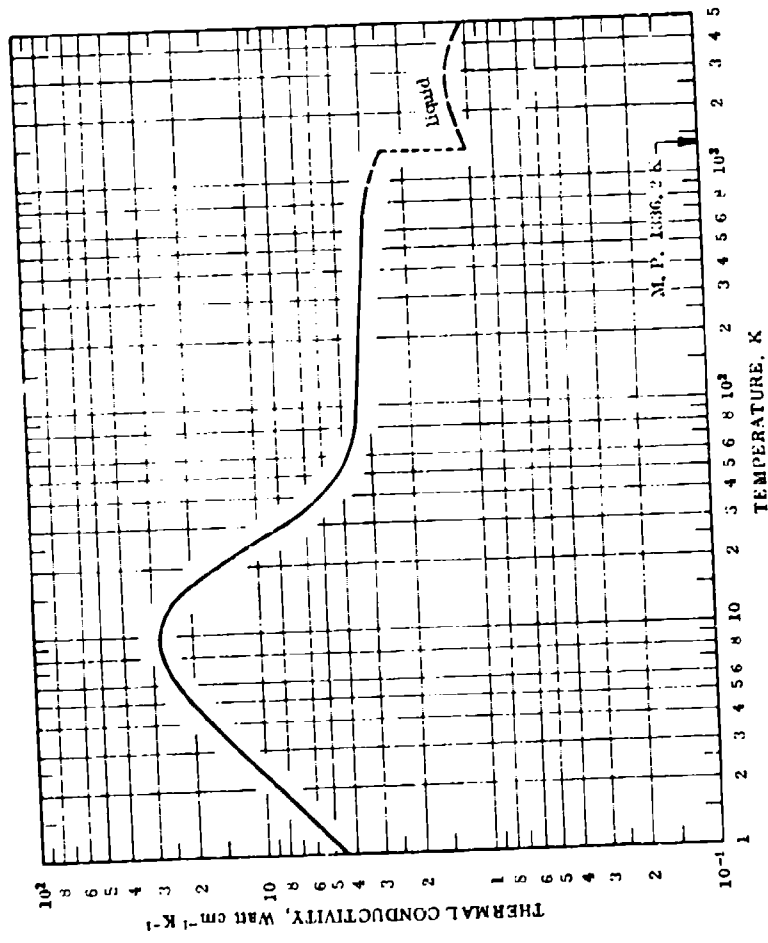
0.393	0.505
0.418	0.530
0.44	0.560
0.445	0.555
0.46	0.600
0.488	0.645
0.503	0.668
0.56	0.790
0.585	0.828
0.63	0.890
0.655	0.948
0.715	1.03
0.735	1.03
0.775	1.03
0.83	1.03
0.845	1.05
0.87	1.11
0.92	1.27

CURVE 30

273.2	3.01
-------	------

Not shown on plot

FIGURE AND TABLE NO. 19R RECOMMENDED THERMAL CONDUCTIVITY OF GOLD



RECOMMENDED VALUES*							
$T_1$	$k_1$	$k_2$	$T_2$	$T_1$	$k_1$	$k_2$	$T_2$
0	0	0	-459.7	1000	(2.70)*	(161)	1340
1	4.44	25	-457.9	1100	(2.71)	(157)	1520
2	8.85	511	-456.1	1200	(2.62)	(151)	1709
3	13.1	757	-454.3	1300	(2.51)	(145)	1880
4	17.1	988	-452.5	1336.2	(2.47)	(143)	1945
In Liquid State							
5	20.7	1200	-450.7				
6	23.7	1370	-448.9				
7	26.0	1500	-447.1	1336.2	(1.05)	(60.7)	1945
8	27.5	1590	-445.3	1400	(1.06)	(61.2)	2060
9	28.2	1630	-443.5				
10	28.2	1630	-441.7	1500	(1.09)	(63.0)	2240
11	27.7	1600	-439.9	1600	(1.12)	(64.7)	2420
12	26.7	1540	-438.1	1700	(1.14)	(65.9)	2600
13	25.5	1470	-436.3	1800	(1.16)	(67.0)	2780
14	24.1	1390	-434.5	1900	(1.18)	(68.2)	2960
15	22.6	1310	-432.7	2000	(1.20)	(69.3)	3140
16	20.9	1210	-430.9	2200	(1.22)	(70.5)	3500
18	17.7	1020	-427.3	2400	(1.24)	(71.6)	3860
20	15.0	867	-423.7	2600	(1.25)	(72.2)	4220
25	10.2	589	-414.7	2800	(1.25)	(72.2)	4580
30	7.6	439	-405.7	3000	(1.24)	(72.2)	4940
35	6.1	352	-396.7	3200	(1.24)	(71.6)	5300
40	5.2	300	-387.7	3400	(1.23)	(71.1)	5660
45	4.6	266	-378.7	3600	(1.21)	(69.9)	6020
50	4.2	243	-369.7	3800	(1.19)	(68.8)	6380
60	3.8	220	-351.7	4000	(1.17)	(67.6)	6740
70	3.58	207	-333.7	4500	(1.11)	(64.1)	7640
80	3.52	203	-315.7	5000	(1.02)	(58.9)	8540
90	3.48	201	-297.7	5500	(0.933)	(53.9)	9440
100	3.45	199	-279.7	6000	(0.839)	(48.5)	10340
150	3.35	194	-189.7	7000	(0.620)	(35.8)	12140
200	3.27	189	-99.7	8000	(0.384)	(22.2)	13940
250	3.20	185	-	9000	(0.131)	(7.57)	15740
273.2	3.18	184	-	9500	(-0)	(-0)	16640
300	3.15	182	86.3				
350	3.13	181	170.3				
400	3.12	180	260.3				
500	3.09	179	440.3				
600	3.04	176	620.3				
700	2.98	172	800.3				
800	2.92	169	980.3				
900	2.85	165	1160				

REMARKS

The recommended values are for well-annealed 99.999% pure gold with residual electrical resistivity  $\rho_0 = 0.0050 \mu\Omega$  cm (characterization by  $\rho_0$  becomes important at temperatures below about 100 K). The values below  $1.5 T_m$  are calculated to fit the experimental data by using  $n = 2.00$ ,  $a = 0.46$ ,  $m = 2.46$ ,  $\sigma' = 4.60 \times 10^{-5}$ , and  $\beta = 0.225$ . The recommended values that are supported by experimental thermal conductivity data are thought to be accurate to within 3% of the true values near room temperature, and 3 to 6% at other temperatures.

\*Values in parentheses are extrapolated or estimated.

$T_1$  in K,  $k_1$  in Watt  $cm^{-1}K^{-1}$ ,  $T_2$  in F, and  $k_2$  in Btu  $hr^{-1}ft^{-1}F^{-1}$ .

# THERMAL CONDUCTIVITY OF HAFNIUM

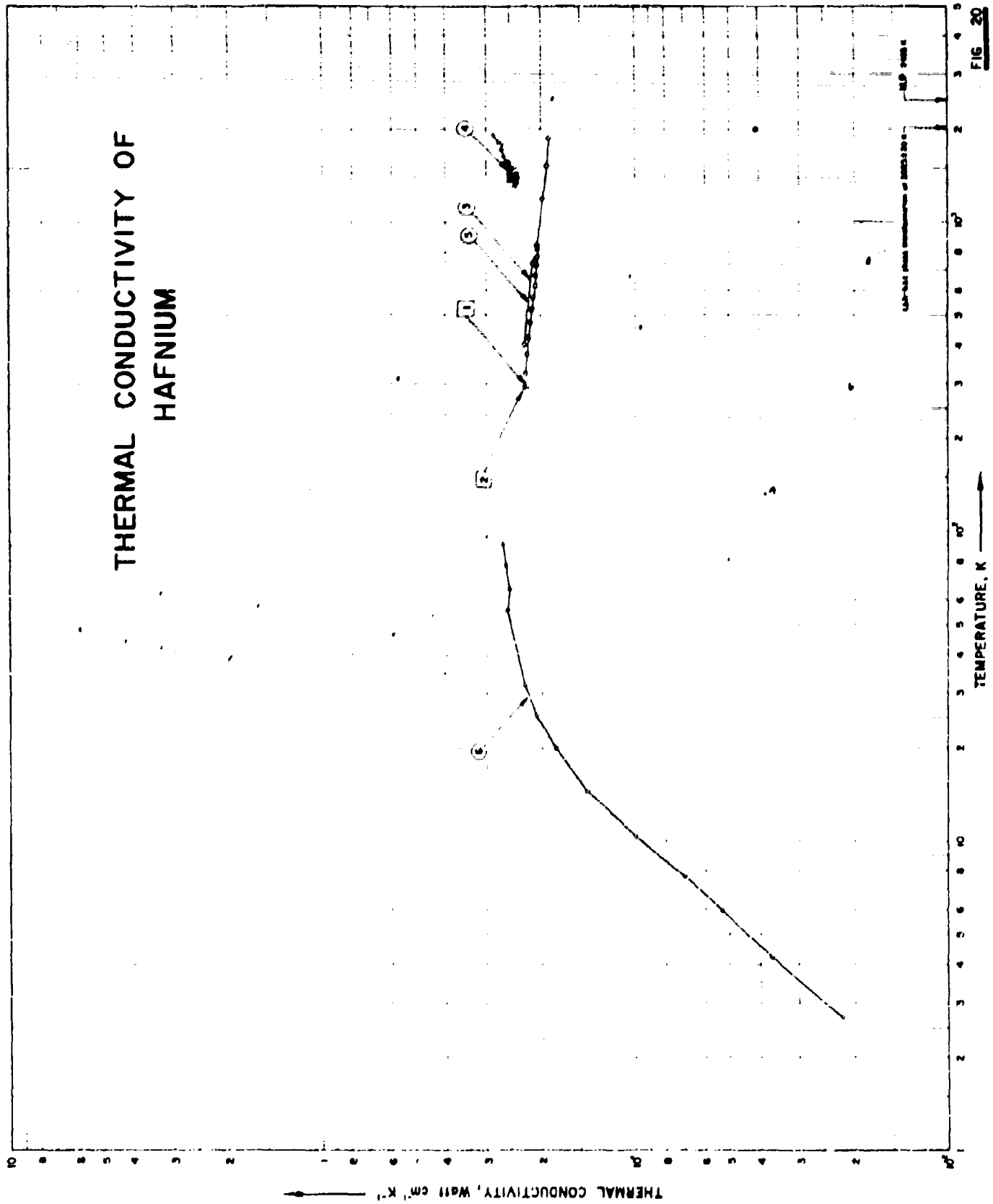


FIG. 20

SPECIFICATION TABLE NO. 20 THERMAL CONDUCTIVITY OF HAFNIUM

[ For Data Reported in Figure and Table No. 20 ]

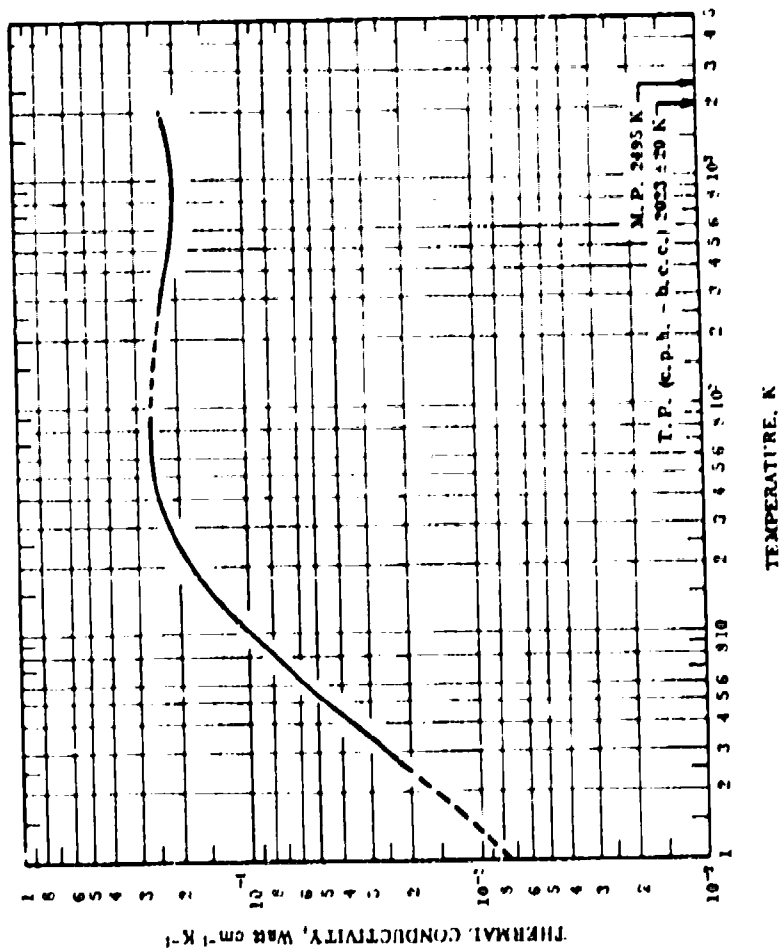
Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)		Composition (continued), Specifications and Remarks
							Hf	Zr	
1	821		1959	293.2					Density 13.36 g cm <sup>-3</sup> ; electrical resistivity 30 μohm cm at 20 C; data probably not original.
2	822		1961	293.2					Hexagonal close packed; density 13.1 g cm <sup>-3</sup> ; electrical resistivity 35.1 μohm cm at ~20 C.
3	614	R	1961	401-187K	± 5.0				99 Hafnium, 1.0 max Zr, 0.1 max Ti and Si each, 0.01 max Fe, V and Zn each, 0.001 max Mn, Ni and Cu each, 0.0001 max Mg.
4	890, 891	T	1966	1301-1908		Iodide Hf			12 mm dia x 65 mm long; density 13.06 g cm <sup>-3</sup> ; measured in a vacuum of 5 x 10 <sup>-5</sup> mm Hg.
5	336	C	1953	323-823			~ 97.96	2.0	0.008 Pb, 0.007 Al, 0.006 W, 0.005 Fe, 0.004 Cu, < 0.003 Zn, 0.002 each of Si, Ti and Mo, trace Sn, U, Co, Ni, Mg, Cr and Mn; specimen 2 cm in dia and 15 cm long; supplied by Westinghouse Atomic Power Division; electrical resistivity 34, 1, 40, 6, 47, 1, 53, 6, 60, 1 and 66.6 μohm cm at 0, 50, 100, 150, 200 and 250 C, respectively; measured in vacuum of ~1 x 10 <sup>-5</sup> mm Hg; Armeo iron used as comparative material.
6	151	L	1957	2 7-9i		HF 1	99.5-99	0.5-1.0	Specimen 5 x 1.52 mm and ~6 cm long; supplied by Foote Mineral Co.; as received; A <sub>0</sub> = 4.23 μohm cm; electrical resistivity ratio p (295 K)/p <sub>0</sub> = 8.58.

## DATA TABLE NO. 20 THERMAL CONDUCTIVITY OF HAFNIUM

[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup> K<sup>-1</sup>]

T	k	T	k
<u>CURVE 1</u>			
293.2	0.221	<u>CURVE 5 (cont.)</u>	
<u>CURVE 2</u>			
293.2	0.224	723.2	0.206
<u>CURVE 3</u>			
400.9	0.226	773.2	0.205
737.1	0.212	823.2	0.205
845.4	0.206	<u>CURVE 6</u>	
1190.4	0.198	2.70	0.0221
1527.6	0.191	4.23	0.0370
1877.6	0.189	5.96	0.0532
<u>CURVE 4</u>			
1301	0.240	7.70	0.0700
1339	0.236	10.40	0.100
1352	0.250	14.50	0.142
1386	0.234	20.00	0.180
1400	0.249	25.60	0.208
1432	0.236	31.90	0.226
1442	0.250	55.80	0.255
1491	0.239	65.00	0.251
1501	0.250	77.90	0.257
1557	0.250	90.80	0.262
1565	0.254	<u>CURVE 5</u>	
1538	0.260	323.2	0.223
1709	0.264	373.2	0.250
1781	0.265	423.2	0.218
1807	0.271	473.2	0.215
1908	0.283	523.2	0.213
<u>CURVE 5</u>			
323.2	0.223	573.2	0.210
373.2	0.250	623.2	0.208
423.2	0.218	673.2	0.207
473.2	0.215		
523.2	0.213		
573.2	0.210		
623.2	0.208		
673.2	0.207		

FIGURE AND TABLE NO. 23R RECOMMENDED THERMAL CONDUCTIVITY OF HAFNIUM



RECOMMENDED VALUES\*  
(For Polycrystalline)

T <sub>1</sub>	k <sub>1</sub>	k <sub>2</sub>	T <sub>2</sub>	T <sub>1</sub>	k <sub>1</sub>	k <sub>2</sub>	T <sub>2</sub>
0	0	0	-459.7	500	0.217	12.5	440.3
1	(0.00764)†	(0.441)	-457.9	600	0.213	12.3	620.3
2	(0.0163)	(0.942)	-456.1	700	0.210	12.1	800.3
3	0.0255	1.47	-454.3	800	0.208	12.0	980.3
4	0.0349	2.02	-452.5	900	0.207	12.0	1160
5	0.0445	2.57	-450.7	1000	0.207	12.0	1340
6	0.0544	3.14	-448.9	1100	0.208	12.0	1520
7	0.0645	3.73	-447.1	1200	0.209	12.1	1700
8	0.0746	4.31	-445.3	1300	0.210	12.1	1880
9	0.0848	4.90	-443.5	1400	0.211	12.2	2060
10	0.0952	5.50	-441.7	1500	0.213	12.3	2240
11	0.106	6.12	-439.9	1600	0.215	12.4	2420
12	0.116	6.70	-438.1	1700	0.218	12.6	2600
13	0.126	7.28	-436.3	1800	0.220	12.7	2780
14	0.135	7.80	-434.5	1900	0.223	12.9	2960
15	0.144	8.32	-432.7	2000	(0.226)	(13.1)	3140
16	0.152	8.78	-430.9				
18	0.167	9.65	-427.3				
20	0.180	10.4	-423.7				
25	0.205	11.9	-414.7				
30	0.221	12.8	-405.7				
35	0.237	13.5	-396.7				
40	0.241	13.9	-387.7				
45	0.247	14.3	-378.7				
50	0.251	14.5	-369.7				
60	0.256	14.8	-351.7				
70	0.259	15.0	-333.7				
80	0.260	15.0	-315.7				
90	0.260	15.0	-297.7				
100	(0.260)	(15.0)	-279.7				
150	(0.251)	(14.5)	-189.7				
200	(0.244)	(14.1)	-99.7				
250	(0.236)	(13.6)	-	9.7			
273.2	(0.233)	(13.5)		30.2			
300	0.230	13.3		80.3			
350	0.226	12.1		170.3			
400	0.223	12.9		260.3			

REMARKS

The recommended values are for well-annealed 99% pure hafnium with residual electrical resistivity  $\rho_0 = 4.23 \mu\Omega \text{ cm}$  (characterization by  $\rho_0$  becomes important at low room temperatures). The values that are supported by experimental thermal conductivity data are thought to be accurate to within 5% of the true values near room temperature and 5 to 10% at other temperatures.

\* T<sub>1</sub> in K, k<sub>1</sub> in Watt cm<sup>-1</sup> K<sup>-1</sup>, T<sub>2</sub> in F, and k<sub>2</sub> in Btu hr<sup>-1</sup> ft<sup>-1</sup> F<sup>-1</sup>. † Values in parentheses are extrapolated or interpolated.

# THERMAL CONDUCTIVITY OF HOLMIUM

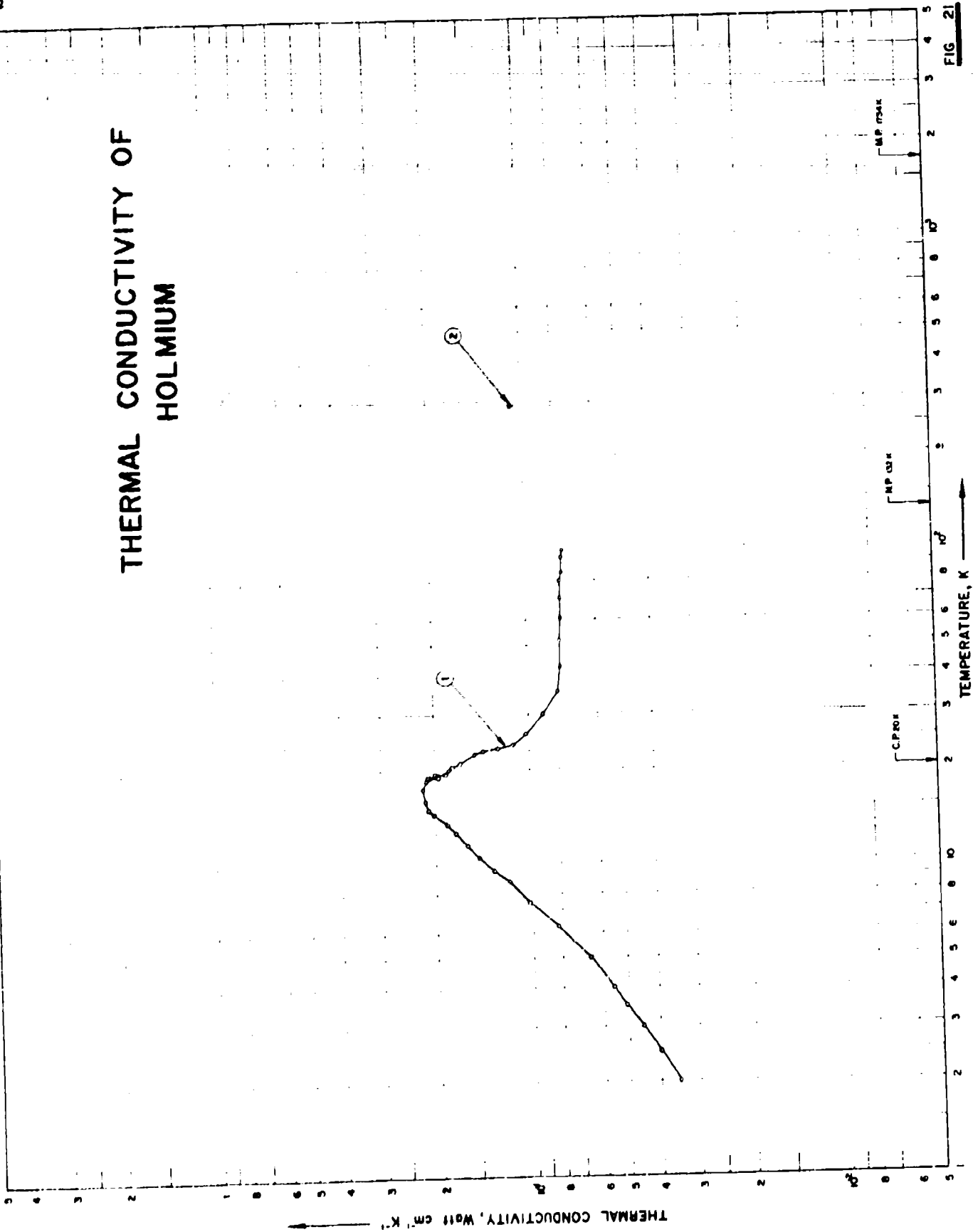


FIG. 21

## SPECIFICATION TABLE NO. 21 THERMAL CONDUCTIVITY OF HOLMIUM

(Impurity &lt; 0.20% each; total impurities &lt; 0.50%)

[ For Data Reported in Figure and Table No. 21 ]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition ( weight percent), Specifications and Remarks
1	808, 322	L	1965	2.0-100			99.9 pure; polycrystalline; dimensions 3 x 0.2 x 0.025 cm; annealed at 700 C for 3 hrs in vacuum of 10 <sup>-4</sup> mm Hg; electrical resistivity 7.32 μohm cm at 4.2 K; electrical resistivity ratio $(\rho_{291K})/(\rho_{4.2K}) = 10.9$ ; ferromagnetic below 20 K; antiferromagnetic between 20 and 133 K.
2	777	C	1965	291	3		High purity; polycrystalline; 0.25 in. dia., 0.25 in. long; supplied by Johnson Matthey and Co. Ltd.; electrical resistivity 108 μohm cm at 18 C; measurements made using 2 different thermal comparators; Monel metal used as comparative material.



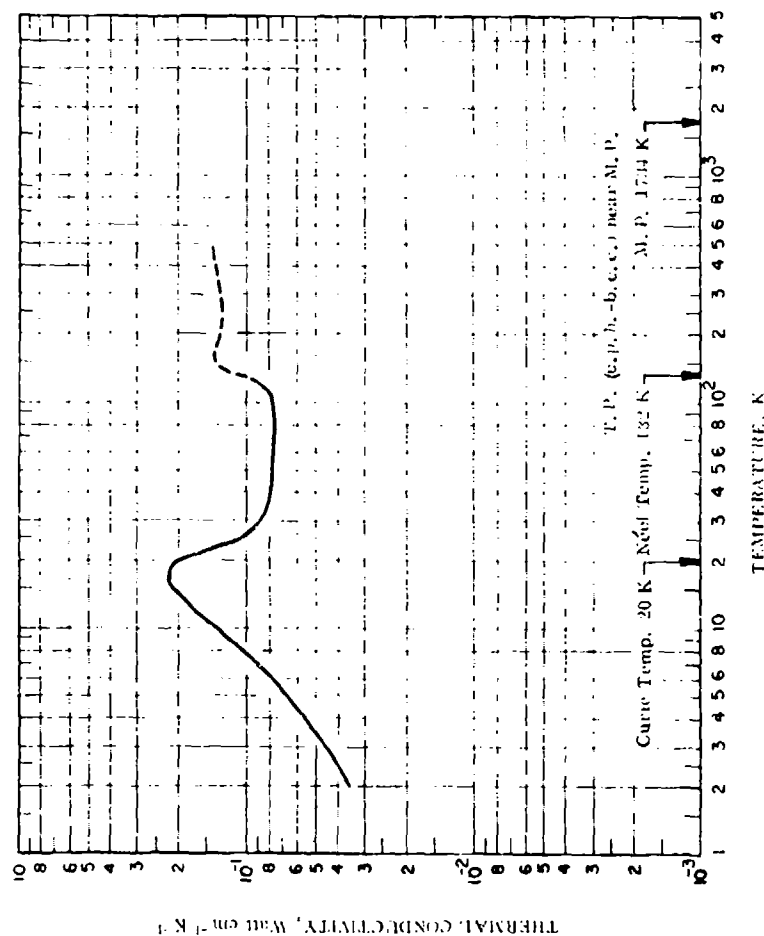
## DATA TABLE NO. 21 THERMAL CONDUCTIVITY OF HOLMIUM

(Impurity &lt; 0.20% each, total impurities &lt; 0.50%)

Temperature, T, K. Thermal Conductivity,  $k$ , Watt  $\text{cm}^{-1}\text{K}^{-1}$ 

T	k	T	k
CURVE 1		CURVE 2	
2.0	0.035	291.2	0.105
2.5	0.040	291.2	0.107
3.0	0.045		
3.5	0.051		
4.0	0.056		
5.0	0.066		
6.3	0.083		
7.5	0.103		
8.8	0.117		
9.5	0.132		
10.5	0.147		
11.5	0.160		
12.5	0.175		
13.5	0.185		
14.5	0.202		
15.0	0.210		
16.0	0.215		
17.5	0.217		
18.5	0.213		
19.0	0.211		
19.0	0.207		
19.5	0.200		
19.0	0.195		
19.5	0.186		
20.0	0.182		
20.5	0.177		
21.0	0.167		
22.5	0.150		
23.0	0.140		
23.5	0.125		
24.0	0.112		
26.0	0.102		
30.0	0.090		
35.5	0.080		
42.5	0.078		
51.5	0.078		
61.0	0.077		
70.5	0.077		
80.0	0.077		
85.0	0.076		
95.0	0.076		
100.0	0.075		

FIGURE AND TABLE NO. 21R RECOMMENDED THERMAL CONDUCTIVITY OF HOLMIUM



REMARKS

The recommended values are for well-annealed 99.9% pure holmium with residual electrical resistivity  $\rho_0 = 7.32 \mu\Omega \text{ cm}$  (characterization by  $\rho_0$  becomes important at temperatures below about 200 K). The recommended values that are supported by experimental thermal conductivity data are thought to be accurate to within 10 to 15% of the true values.

RECOMMENDED VALUES<sup>a</sup>  
(For Polycrystalline)

$T_1$	$k_1$	$k_2$	$T_2$
0	0	0	-459.7
2	0.0350	2.02	-456.1
3	0.0454	2.02	-454.3
4	0.0559	3.23	-452.5
5	0.0669	3.87	-450.7
6	0.0796	4.60	-448.9
7	0.0930	5.37	-447.1
8	0.107	6.18	-445.3
9	0.122	7.05	-443.5
10	0.137	7.92	-441.7
11	0.152	8.78	-439.9
12	0.167	9.65	-438.1
13	0.182	10.5	-436.3
14	0.195	11.3	-434.5
15	0.208	12.0	-432.7
16	0.216	12.5	-430.9
18	0.215	12.4	-427.3
20	0.191	11.0	-423.7
25	0.107	6.18	-414.7
30	0.0896	5.18	-405.7
35	0.0803	4.67	-396.7
40	0.0784	4.53	-387.7
45	0.0780	4.51	-378.7
50	0.0777	4.49	-369.7
60	0.0772	4.46	-351.7
70	0.0767	4.43	-333.7
80	0.0762	4.40	-315.7
90	0.0759	4.39	-297.7
100	0.0762	4.40	-279.7
130	(0.0985) <sup>b</sup>	(5.69)	-225.7
135	(0.115)	(6.64)	-216.7
150	(0.137)	(7.92)	-189.7
170	(0.135)	(7.80)	-153.7
200	(0.130)	(7.51)	-99.7
250	(0.129)	(7.45)	-9.7
273.2	(0.129)	(7.45)	32.0
300	(0.130)	(7.51)	80.3
350	(0.132)	(7.63)	170.3
400	(0.134)	(7.74)	260.3
500	(0.141)	(8.15)	440.3

<sup>a</sup>  $T_1$  in K,  $k_1$  in Watt cm<sup>-1</sup> K<sup>-1</sup>,  $T_2$  in F, and  $k_2$  in Btu hr<sup>-1</sup> ft<sup>-1</sup> F<sup>-1</sup>. <sup>b</sup> Values in parentheses are extrapolated or estimated.

# THERMAL CONDUCTIVITY OF INDIUM

FIGURE SHOWS ONLY 20 OF THE CURVES REPORTED IN TABLE

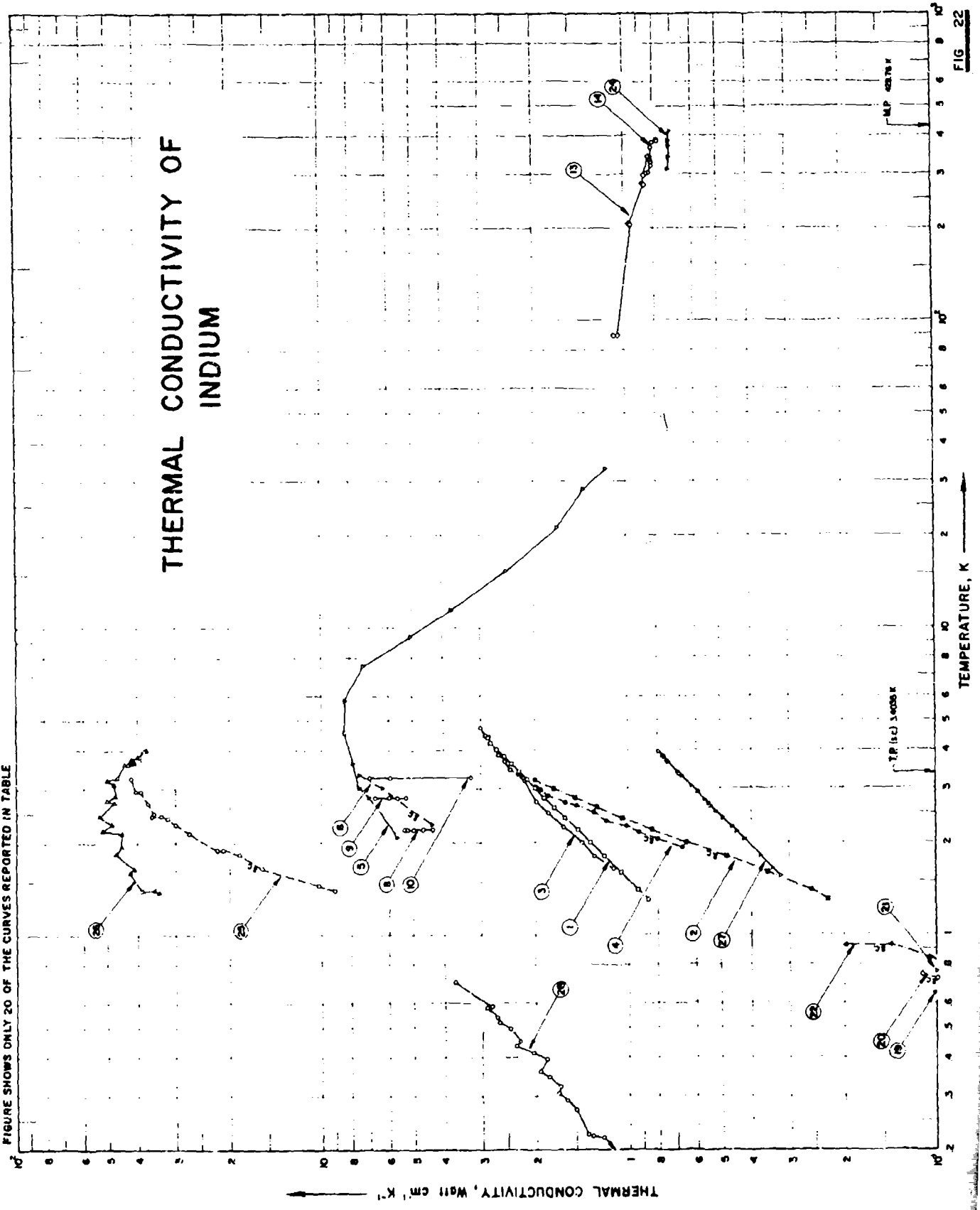


FIG 22

## SPECIFICATION TABLE NO. 22 THERMAL CONDUCTIVITY OF INDIUM

(Impurity &lt; 0. 20% each; total impurities &lt; 0. 50%)

[ For Data Reported in Figure and Table No. 22 ]

Curve No.	Rel. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition ( weight percent).	Specifications and Remarks
1	132	L	1955	1.3-4.2	±2.5		99.95 pure; single crystal; cylindrical specimen; obtained from Stout, J. W. and Gutman, L.; residual electrical resistivity 0.0371 $\mu\text{ohm cm}$ ; measured in a vacuum of $< 5 \times 10^{-7}$ mm Hg and in a longitudinal magnetic field of approx 1000 oersteds; in normal state; data from smoothed curve.	
2	132	L	1955	1.3-3.2	±2.5		The above specimen measured in superconducting state; data from smoothed curve.	
3	76	L	1952	1.7-5			99.9 pure; single crystal; supplied by Stout, J. W. and Gutman, L.; transition temp 3.40 K; measured in a magnetic field of ~100 gauss; in normal state.	
4	76	L	1952	1.9-3.4			The above specimen measured in superconducting state.	
5	97	L	1952	2.1-33	2-3	JM 4398; In 1	99.993 pure; polycrystalline; 1-2 mm dia x 5 cm long; supplied by Johnson Matthey; in normal state.	
6	97	L	1952	2.3-3.3		JM 4398; In 1	The above specimen measured in superconducting state.	
7	794, 400	L	1953	0.46-0.87		JM 4398; In 2	99.993 pure; single crystal; in superconducting state; preliminary result.	
8	342	L	1953	2.2		JM 4398; In 1	99.993 pure; polycrystalline; 1-2 mm dia x 5 cm long; supplied by Johnson Matthey; annealed in vacuo; measured in transverse magnetic fields of strength ranging from 0.34 to 2.87 Kiloersters kOe.	
9	342	L	1953	2.8		JM 4398; In 1	The above specimen measured in magnetic fields of strength ranging from 0.34 to 2.87 kOe.	
10	342	L	1953	3.25		JM 4398; In 1	The above specimen measured in H ranging from 1.10 to 3.94 kOe.	
11	290, 285	L	1952	2.13		JM 3249; In	Pure; single crystal; 2.8 mm dia; spectroscopically standardized indium supplied by Johnson Matthey; cast; somewhat strained in mounting; electrical resistivity ratio $\rho(273 \text{ K})/\rho(4.2 \text{ K}) = 5500$ ; measured in transverse magnetic fields of strength ranging from 0 to 190 gauss.	
12	290, 285	L	1952	2.13		JM 3249; In	The above specimen measured in transverse fields of decreasing strength ranging from 184 to 0 gauss.	
13	795	L	1962	89-342	~2		Total impurity < 0.03 (probably Sn and Pb); ~0.85 cm dia x 6.5 cm long; supplied by Johnson Matthey; electrical resistivity reported as 1.65, 3.08, 4.60, 6.22, 8.0, 10.0, 12.15, and 13.0 $\mu\text{ohm cm}$ at 73, 123, 173, 223, 273, 323, 373, and 393 K, respectively; Lorenz function reported as 2.46, 2.60, 2.62, 2.60, 2.57, 2.59, 2.61, and 2.61 x $10^{-4}$ Wobm $\text{K}^{-2}$ at the above temps, respectively; density 7.334 $\text{g cm}^{-3}$ .	
14	795	C	1962	303-390	~2		The above specimen measured in another apparatus; Armco iron used as comparative material.	

SPECIFICATION TABLE NO. 22 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)	Specifications and Remarks
15	454	L	1958	0.26-0.56	2	JM 10281; A0		Spectroscopically pure; single crystal; 4.79 mm dia; indium supplied by Johnson Matthey; cast, crystallized, annealed in air at 120 C and electropolished; residual electrical resistivity 0.00125 $\mu\text{ohm cm}$ ; in superconducting state.
16	454	L	1958	0.27-0.58	2	JM 10281; A1		The above specimen etched with 25% $\text{H}_2\text{SO}_4$ solution; in superconducting state.
17	454	L	1958	0.36-0.59	2	JM 10281; A2		The above specimen etched with 25% $\text{H}_2\text{SO}_4$ solution and annealed at 120 C for 9 days; in superconducting state.
18	454	L	1958	0.32-0.65	2	JM 10281; A3		The above specimen electropolished (dia reduced to 1.86 mm); damaged by bending during reduction; in superconducting state.
19	454	L	1958	0.34-0.65	2	JM 10281; A4		The above specimen etched with 25% $\text{H}_2\text{SO}_4$ solution; in superconducting state.
20	454	L	1958	0.30-0.74	2	JM "chem pure"; B0		99.9 pure (by difference); single crystal; 5.13 mm dia; supplied by Johnson Matthey; cast, crystallized, electropolished, and annealed in air at 120 C; in superconducting state.
21	454	L	1958	0.30-0.76	2	JM "chem pure"; B1		The above specimen etched with 25% $\text{H}_2\text{SO}_4$ solution; in superconducting state.
22	412	L	1955	0.25-0.93		JM 4938		99.993 pure; single crystal; $\sim 1$ mm dia $\times$ 4 cm long; made from Johnson Matthey metal; in superconducting state.
23	412	L	1955	0.29-0.79		JM 4938		99.993 pure; polycrystal; 1.6 mm dia $\times$ 4 cm long; made from Johnson Matthey metal; in superconducting state.
24	796	L	1964	313-417	7-8			99.997 pure.
25	797, 799	L	1960	1.4-3.3				Spectroscopically pure; polycrystalline; $\sim 0.5$ mm in dia; extruded; annealed at room temp for several months; electrical resistivity ratio $\rho(\text{room temp})/\rho(0 \text{ K})$ estimated to be 11000; in superconducting state.
26	797, 798	L	1960, 1961	1.4-4.0				The above specimen measured in a longitudinal magnetic field; in normal state (data corrected to zero field).
27	799	L	1965	1.6-4.0	2	In 1		As major impurity; specimen $\sim 3$ mm dia; in supplied by American Smelting and Refining Co.; cast in vacuum; annealed near melting point for at least 2 wks; residual electrical resistivity 0.122 $\mu\text{ohm cm}$ ; transition temp 3.39 K; measured in a magnetic field; in normal state.
28	732	L	1965	0.023-0.70				99.999 pure (nominal); 0.5 mm dia $\times$ 5 cm long; supplied by Koch-Light Laboratories Ltd. (Colbrook, England); measured in a longitudinal magnetic field of 550 gauss; in normal state.
29	892, 893	L	1963	312-415	7-8			99.997 pure.

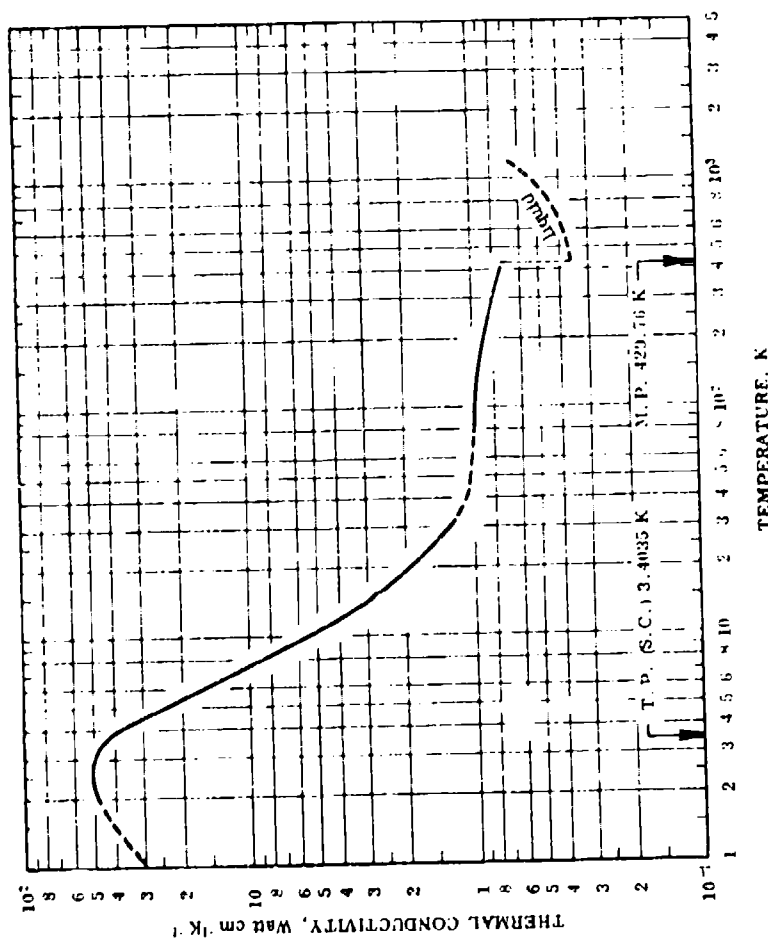


DATA TABLE NO. 22 (continued)

CURVE 22		CURVE 25		CURVE 27		CURVE 28 (cont.)		CURVE 28 (cont.)		CURVE 29 <sup>r</sup>	
T	k	T	k	T	k	T	k	T	k	T	k
0.248	0.00355	1.40	9.0	1.55	0.320	0.046	0.315 <sup>r</sup>	0.520	2.65		
0.255	0.00375	1.45	10.3	1.65	0.341	0.048	0.28	0.540	2.70		
0.280	0.00425	1.65	15.5	1.80	0.372	0.0555	0.34	0.560	2.90		
0.295	0.00470	1.82	18.5	2.05	0.423	0.060	0.385	0.585	2.80		
0.310	0.00450	1.90	21.0	2.17	0.447	0.061	0.36	0.70	3.70		
0.315	0.00525	1.90	22.0	2.25	0.463	0.070	0.40				
0.315	0.00600	2.15	27.0	2.40	0.493	0.072	0.43				
0.330	0.00600	2.30	30.0	2.53	0.519	0.078	0.405				
0.400	0.00860	2.40	32.0	2.63	0.540	0.0795	0.425				
0.410	0.00840	2.45	33.5	2.67	0.548	0.084	0.45	312.2	0.720		
0.450	0.00910	2.45	36.0	2.75	0.564	0.085	0.49	343.2	0.711		
0.450	0.00980	2.50	35.5	2.92	0.596	0.091	0.56	373.2	0.720		
0.475	0.0124	2.70	37.0	3.05	0.621	0.093	0.50	388.2	0.724		
0.490	0.0125	2.95	39.0	3.30	0.671	0.099	0.56	415.2	0.716		
0.510	0.0120	2.95	40.5	3.35	0.681	0.103	0.60				
0.530	0.0115	3.25	42.0	3.40	0.690	0.110	0.62				
0.595	0.0178			3.43	0.696	0.113	0.61				
0.706	0.0230			3.65	0.737	0.118	0.63				
0.705	0.0320			3.70	0.747	0.118	0.67				
0.748	0.0345	1.38	34.0	3.70	0.746	0.128	0.70				
0.860	0.0740	1.40	35.0	3.80	0.764	0.133	0.74				
0.810	0.0930	1.60	42.5	3.80	0.765	0.140	0.75				
0.930	0.132	1.65	41.0	3.85	0.774	0.156	0.81				
0.930	0.138	1.85	47.0	3.95	0.793	0.155	0.84				
		1.95	45.0			0.155	0.89				
		2.15	45.0			0.160	0.91				
		2.20	52.0			0.165	0.95				
0.285	0.0021	2.30	48.5	0.025	0.17	0.180	1.01				
0.301	0.0026	2.40	48.5	0.024	0.143	0.190	1.08				
0.390	0.0049	2.45	53.0	0.028	0.173	0.220	1.23				
0.402	0.0059	2.70	47.5	0.0285	0.212	0.225	1.23				
0.475	0.0109	2.75	50.0	0.029	0.205	0.250	1.28				
0.550	0.019	2.85	47.0	0.032	0.195	0.270	1.50				
0.648	0.022	3.10	47.5	0.0335	0.21	0.290	1.60				
0.790	0.041	3.20	50.0	0.034	0.215	0.305	1.70				
		3.25	46.5	0.034	0.23	0.320	1.68				
		3.60	44.0	0.036	0.23	0.345	1.81				
		3.60	43.0	0.039	0.245	0.360	1.95				
312.2	0.715	3.65	42.0	0.041	0.26	0.395	1.85				
342.2	0.707	3.65	41.0	0.041	0.23	0.415	2.05				
372.2	0.711	3.75	42.0	0.043	0.245	0.435	2.31				
388.2	0.714	3.80	40.0	0.044	0.265	0.450	2.27				
417.2	0.705	4.00	38.0	0.044	0.31	0.495	2.45				
		4.00	37.5	0.045							

Not shown on plot

FIGURE AND TABLE NO. 22R RECOMMENDED THERMAL CONDUCTIVITY OF HIGH-PURITY OF INDIUM



RECOMMENDED VALUES\*

T, °K	Polycrystalline		T <sub>1</sub>	T <sub>2</sub>	In Liquid State		T <sub>2</sub>
	k <sub>1</sub>	k <sub>2</sub>			k <sub>1</sub>	k <sub>2</sub>	
0	0	0	429.76		(20.9)		313.90
1	(29.5)‡	(1700)	500		(0.372)		440.3
2	47.8	2760	600		(0.389)		620.3
3	48.9	2840	700		(0.414)		800.3
4	37.3	2160	800		(0.444)		980.3
5	24.4	1410	900		(0.478)		1160
6	16.4	948	1000		(0.524)		1340
7	11.7	676	1100		(0.585)		1520
8	8.70	503	1200		(0.674)		1700
9	6.85	396					
10	5.58	322					
11	4.72	271					
12	4.09	236					
13	3.60	208					
14	3.20	185					
15	2.88	166					
16	2.62	151					
18	2.22	128					
20	1.94	112					
25	1.51	87.2					
30	1.28	74.0					
35	(1.15)	(66.4)					
40	(1.09)	(63.0)					
45	(1.06)	(61.2)					
50	(1.04)	(60.1)					
60	(1.02)	(58.9)					
70	(1.00)	(57.8)					
80	(0.982)	(57.3)					
90	0.983	56.8					
100	0.976	56.4					
150	0.939	54.3					
200	0.897	51.8					
250	0.855	49.4					
273.2	0.837	48.4					
300	0.817	47.2					
350	0.777	44.9					
400	0.745	43.0					
429.76	(0.729)	(42.1)					

REMARKS

The recommended values are for well-annealed high-purity indium with residual electrical resistivity  $\rho_0 = 0.000828 \mu\Omega$  cm (characterized by  $\rho_0$  becomes important at temperatures below about 80 K). The recommended values that are supported by experimental thermal conductivity data are thought to be accurate to within 4% of the true values near room temperature and 4 to 10% at other temperatures.

\* T<sub>1</sub> in K, v<sub>1</sub> in Watt cm<sup>-1</sup> K<sup>-1</sup>, T<sub>2</sub> in F, and k<sub>2</sub> in Btu lb<sup>-1</sup> ft<sup>-1</sup> F<sup>-1</sup>. ‡ Values in parentheses are extrapolated, interpolated, or estimated.



# THERMAL CONDUCTIVITY OF IRIDIUM

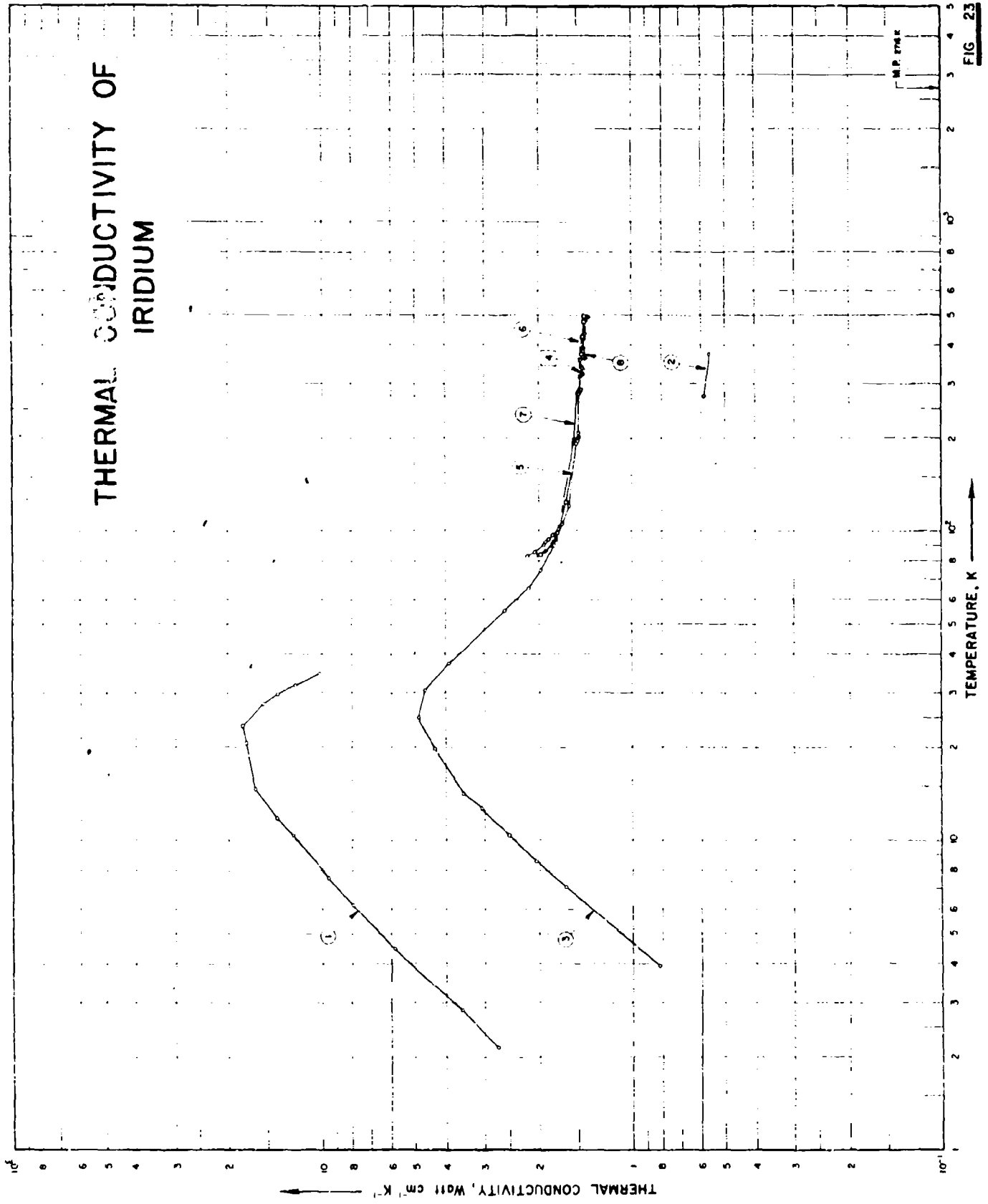


FIG. 23

SPECIFICATION TABLE NO. 23 THERMAL CONDUCTIVITY OF IRIIDIUM

(Impurity < 0.20% each; total impurities < 0.50%)

[For Data Reported in Figure and Table No. 23]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
1	97	L	1952	2.2-35	2.0-3.0	JM 3441; Ir 1	99.995 pure; 1-2 mm dia x 5 cm long; supplied by Johnson Matthey; annealed. Pure; square cross section 0.103 x 0.101 cm, 10.0 cm long; specific gravity 22.33; electrical resistivity reported as 8,190, 8,480, 10,521, and 10,59 $\mu\text{ohm cm}$ at 0, 12.11, 97.20, and 100 C, respectively.
2	8	F	1914	290-373			
3	149	L	1957	3.9-90		JM 10371; Ir 2	99.98% pure; approx impurities: 0.01% Pt, 0.002 Pd, and 0.001 Cu; 2 mm dia x 5-7 cm long; supplied by Johnson Matthey; annealed at 1300 C; residual electrical resistivity ( $\rho_0$ ) = 0.1014 $\mu\text{ohm cm}$ ; electrical resistivity ratio $\rho(\rho_0 K)/\rho(0 K) = 49.5$ ; Lorenz function $2.50 \times 10^{-6} \text{V}^2 \text{K}^{-2}$ at 0 K.
4	411	C	1955	322.2			Impurities estimated: 0.02-0.05 Rh, 0.002-0.005 Ru, and 0.001 Pd; 5 cm long, 0.3182 cm in dia; supplied by Johnson Matthey; annealed at 1313 C; density 22.49 g cm <sup>-3</sup> .
5	411	L	1955	83-289	<3.1		The above specimen; electrical resistivity reported as 0.79, 0.85, 0.95, 1.00, 1.02, 1.10, 1.20, 1.29, 1.59, 3.05, 3.17, 3.31, 3.43, 4.82, 4.92, and 5.07 $\mu\text{ohm cm}$ at -190.5, -187.5, -183.2, -180.8, -179.6, -175.8, -171.2, -166.9, -152.7, -82.2, -77.0, -70.0, -64.1, 4.3, 8.8, and 16.1 C, respectively.
6	665	L, C	1962	83-493			0.02-0.05 Rh, 0.002-0.005 Ru, and 0.001 Pd; 5.0 cm long, 0.318 cm in dia; supplied by Johnson Matthey; heated to 1310 C; density 22.43 g cm <sup>-3</sup> ; electrical resistivity 4.71 $\mu\text{ohm cm}$ at 273 K; residual electrical resistivity 0.055 $\mu\text{ohm cm}$ ; data obtained by using two methods of measurement for low and moderate temp ranges; Armco iron used as comparative material for the moderate temp measurement.
7	249	L	1967	83-386			0.02-0.05 Rh, 0.002-0.005 Ru, and 0.001 Pd; specimen 0.318 cm in dia and 5 cm long; supplied by Johnson Matthey; annealed at 1590 K; density 22.43 g cm <sup>-3</sup> ; electrical resistivity ratio $\rho(273 K)/\rho(4.2 K) = 85.7$ ; electrical resistivity reported as 1.16, 3.25, 5.33, 7.39, and 9.42 $\mu\text{ohm cm}$ at 190, 200, 300, 400, and 500 K, respectively.
8	249	C	1967	315-492			The above specimen measured in a comparative apparatus using Armco iron as reference material.

DATA TABLE NO. 23 THERMAL CONDUCTIVITY OF IRIIDIUM

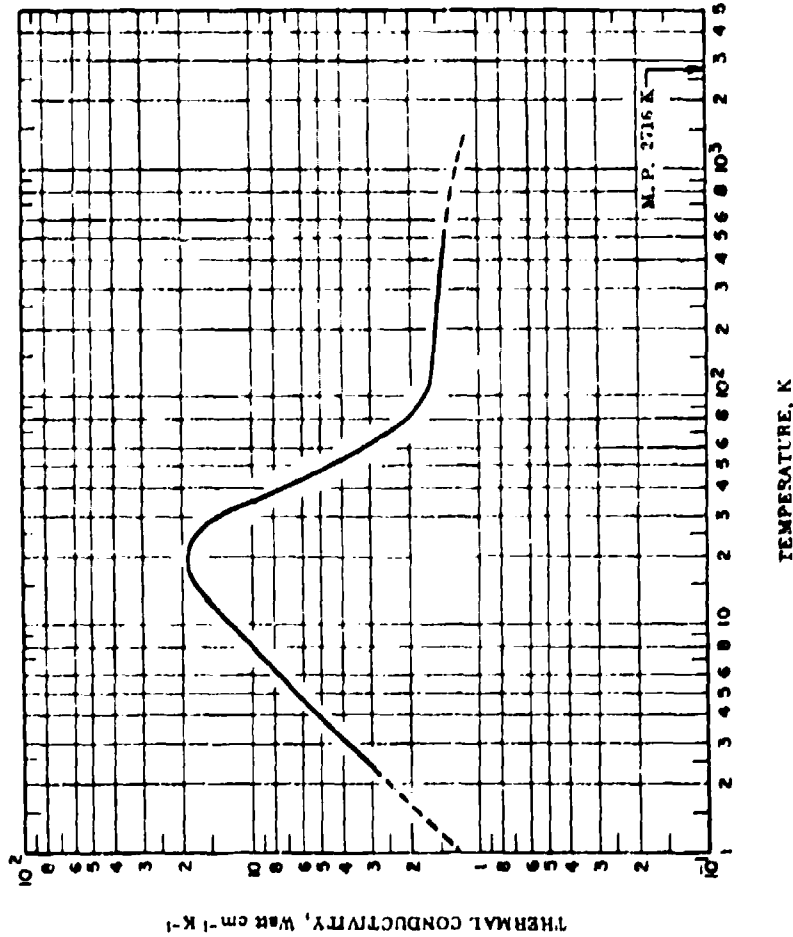
(Impurity &lt;0.20% each; total impurities &lt;0.50%)

(Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup>K<sup>-1</sup>)

T	k	T	k	T	k
<u>CURVE 1</u>					
2.15	2.710	82.7	2.20	102.6	1.72
2.84	3.570	85.7	2.07	107.1	1.685*
4.47	5.849	90.0	1.94	190.9	1.54
6.19	7.914	92.4	1.91	196.1	1.535
7.57	9.591	93.6	1.87	203.1	1.505*
10.49	12.473	97.4	1.91	209.0	1.49*
11.79	13.978	102.0	1.74	277.4	1.49
14.62	16.559	106.3	1.66	281.9	1.48
20.65	17.677	120.5	1.60	289.2	1.475
23.50	18.150	191.0	1.51	386.1	1.45
27.53	15.613	196.2	1.50		
29.76	13.892	203.2	1.49		
31.83	12.129	209.1	1.49		
34.58	10.237	277.5	1.48		
		282.0	1.47		
		289.3	1.46		
<u>CURVE 2</u>					
290.2	0.590				
373.2	0.565				
<u>CURVE 3</u>					
3.92	0.821	83.2	1.98		
7.09	1.642	91.2	1.75		
8.56	2.052	98.2	1.62		
10.45	2.500	108.2	1.56*		
12.69	3.060	173.2	1.54*		
14.18	3.545	223.2	1.50		
19.78	4.328	273.2	1.48*		
25.06	4.888	323.2	1.47*		
30.60	4.664	373.2	1.45		
37.31	3.918	423.2	1.44		
55.20	2.575	473.2	1.425		
65.30	2.164	493.2	1.42		
74.60	1.978				
89.90	1.791				
<u>CURVE 4</u>					
323.2	1.45				
<u>CURVE 5</u>					
82.7	2.01				
85.2	1.905				
90.0	1.805				
92.5	1.79				
93.5	1.765				
98.1	1.75*				
<u>CURVE 6</u>					
315.1	1.45				
316.6	1.475*				
327.7	1.475*				
333.8	1.445				
341.4	1.455				
356.5	1.465				
361.9	1.40				
372.3	1.445*				
399.3	1.45				
408.5	1.43*				
414.8	1.42*				
416.8	1.425*				
418.8	1.425				
429.5	1.435*				
433.1	1.42				
477.2	1.405				
492.2	1.385				
<u>CURVE 7</u>					
82.7	2.01				
85.2	1.905				
90.0	1.805				
92.5	1.79				
93.5	1.765				
98.1	1.75*				
<u>CURVE 8</u>					
315.1	1.45				
316.6	1.475*				
327.7	1.475*				
333.8	1.445				
341.4	1.455				
356.5	1.465				
361.9	1.40				
372.3	1.445*				
399.3	1.45				
408.5	1.43*				
414.8	1.42*				
416.8	1.425*				
418.8	1.425				
429.5	1.435*				
433.1	1.42				
477.2	1.405				
492.2	1.385				

\* Not shown on Plot

FIGURE AND TABLE NO. 218 RECOMMENDED THERMAL CONDUCTIVITY OF IRIIDIUM



RECOMMENDED VALUES\*

T <sub>1</sub>	k <sub>1</sub>	k <sub>t</sub>	T <sub>2</sub>	k <sub>t</sub>	T <sub>2</sub>	k <sub>2</sub>	T <sub>1</sub>
0	0	0	-459.7	0	500	81.5	440.3
1	(1.29) <sup>b</sup>	(75.1)	-457.9	(150)	600	79.7	620.3
2	(2.60)	(150)	-456.1		700	78.0	800.3
3	3.90	225	-454.3		800	76.3	980.3
4	5.19	300	-452.5		900	74.5	1160
5	6.49	374	-450.7		1000	72.8	1340
6	7.77	449	-448.9		1100	71.1	1520
7	9.04	523	-447.1		1200	69.3	1700
8	10.3	595	-445.3		1300	67.6	1880
9	11.5	664	-443.5		1400	65.9	2060
10	12.7	734	-441.7		1500	64.1	2240
11	13.8	797	-439.9				
12	14.9	861	-438.1				
13	15.9	919	-436.3				
14	16.7	965	-434.5				
15	17.5	1010	-432.7				
16	18.1	1050	-430.9				
18	18.9	1090	-427.3				
20	19.0	1100	-423.7				
25	17.2	994	-414.7				
30	13.7	792	-405.7				
35	10.1	594	-396.7				
40	7.50	433	-387.7				
45	5.89	340	-378.7				
50	4.72	273	-369.7				
60	3.31	191	-351.7				
70	2.54	147	-333.7				
80	2.09	121	-315.7				
90	1.94	106	-297.7				
100	1.72	99.4	-279.7				
150	1.58	91.9	-189.7				
200	1.53	88.4	-99.7				
250	1.49	86.1	-9.7				
273.2	1.48	85.5	32.6				
300	1.47	84.9	60.3				
350	1.46	84.4	170.3				
400	1.44	83.2	280.3				

REMARKS

The recommended values are for well-annealed 99.995% pure iridium with residual electrical resistivity  $\rho_0 = 0.0188 \mu\Omega \text{ cm}$  (characterization by  $\rho_0$  becomes important at temperatures below about 150 K). The values below 1.5 Tm are calculated to fit the experimental data by using  $n = 3.00$ ,  $\sigma^* = 0.0175 \times 10^{-7}$ , and  $\beta = 0.770$ . The recommended values that are supported by experimental thermal conductivity data are thought to be accurate to within 4% of the true values near room temperature and 4 to 10% at other temperatures.

\* T<sub>1</sub> in K, k<sub>1</sub> in Watt cm<sup>-1</sup> K<sup>-1</sup>, T<sub>2</sub> in F, and k<sub>2</sub> in Btu B<sup>-1</sup> ft<sup>-1</sup> F<sup>-1</sup>. Values in parentheses are extrapolated or estimated.

# THERMAL CONDUCTIVITY OF IRON

FIGURE SHOWS ONLY 27 OF THE CURVES REPORTED IN TABLE

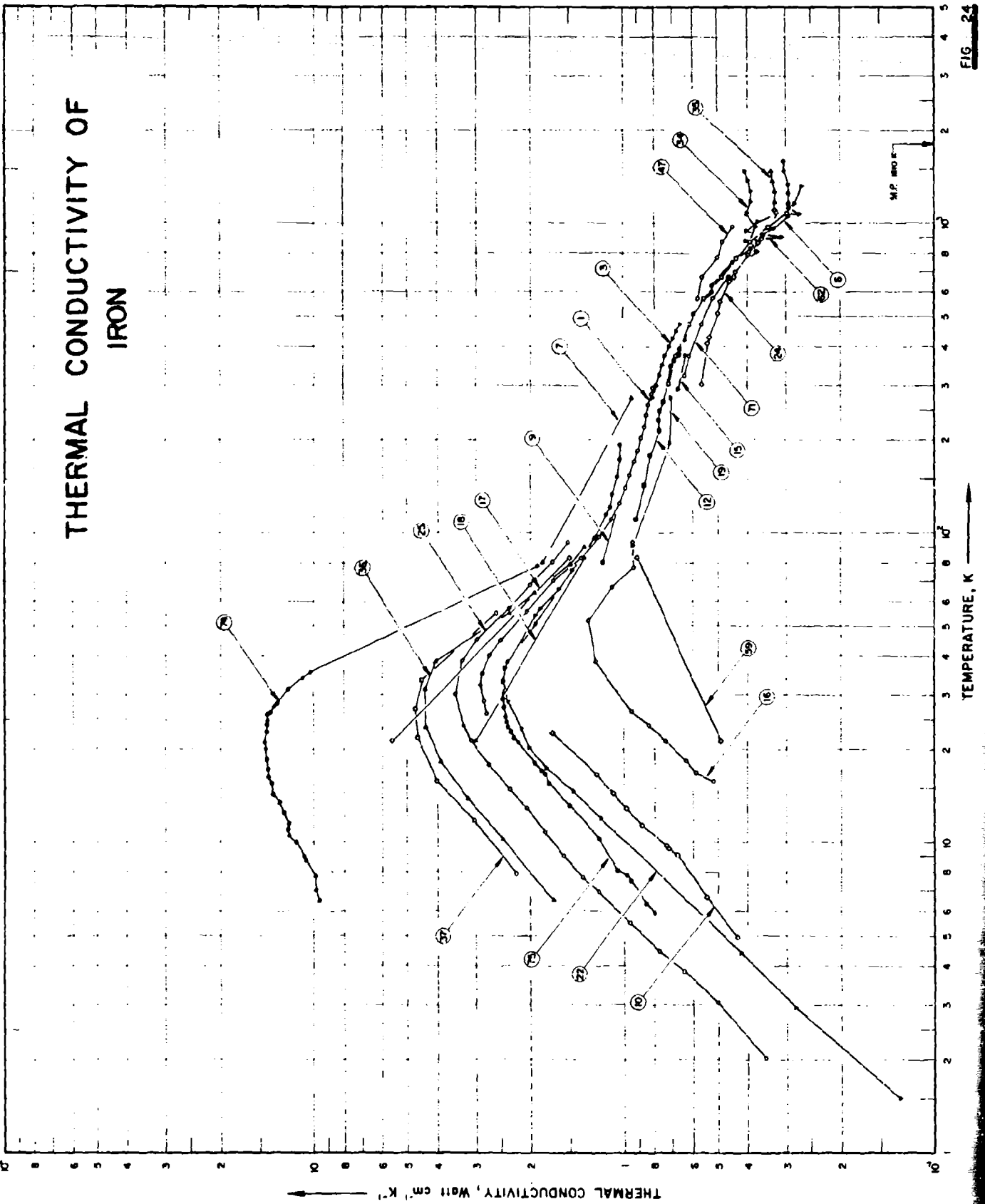


FIG. 24

SPECIFICATION TABLE NO. 24 THERMAL CONDUCTIVITY OF IRON

(Impurity &lt; 0.20% each; total impurities &lt; 0.50%)

(For Data Reported in Figure and Table No. 24)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
1	115	L	1951	26-292			99.99 pure; supplied by Johnston Mackay Ltd.
2	<del>1014</del> 1014	L	1968	82-373	1.6	IS AF3	99.96 Fe, 0.007 Cu, 0.007 Ni, 0.0059 C, 0.004 Mn, 0.004 Si, 0.003 S, 0.0023 N, 0.002 Cr, 0.001 Al, 0.001 P, and 0.0008 O; 1.247 cm dia x 10.44 cm long; prepared by National Physical Laboratory, England; as received; density 7.872 g cm <sup>-3</sup> at 24 C; electrical resistivity reported as 0.925, 5.26, 8.97, and 10.33 μohm cm at 77.78, 194.8, 273.3, and 298.8 K, respectively; Lorenz function reported as 1.625, 1.885, 2.420, 2.704, and 2.884 x 10 <sup>-6</sup> V <sup>2</sup> K <sup>-2</sup> at 70.2, 100, 200, 300, and 389.8 K, respectively; measured in a vacuum of 5 x 10 <sup>-4</sup> torr.
3	1	L	1935	273-473	2		0.0045 C, 0.002 Mn, 0.0015 S, 0.001 P, 0.0006 Ni, 0.0002 Si, traces of Al and Mg; 1 cm dia x 15 cm long; density 7.871 ± 0.002 g cm <sup>-3</sup> ; electrical resistivity reported as 11.5, 14.5, and 17.8 μohm cm at 50, 100, and 150 C, respectively; data taken from smoothed curve.
4	129, 161	C	1933	373-772	5	Basic open hearth Iron	0.042 P, 0.03 Mn, 0.02 C, and 0.005 S; hot-rolled to 1 in. bar; high purity lead used as comparative material; data taken from smoothed curve.
5	5	L	1947	320-1016	1.95	Armco Iron	99.808 Fe (by difference), 0.067 Cu, 0.039 S, 0.035 Mn, 0.028 C, and 0.024 Ni; 2.50 cm dia x 8.00 cm long; machined from a hot-rolled 1.5 in. rod.
6	39	R	1958	371-1594	2	Armco Iron	Specimen consisted of three annular rings each of 0.625 in. I.D., 3.0 in. O.D., and 1 in. thick.
7	34	L	1927	80,273		Electrolyte iron; 1	Coarse-grained; electrical conductivity reported as 121 and 10.4 x 10 <sup>4</sup> ohm <sup>-1</sup> cm <sup>-1</sup> at 80 and 273 C, respectively.
8	34	L	1927	80,273		Electrolyte iron; 2	Fine-grained; electrical conductivity reported as 121 and 10.4 x 10 <sup>4</sup> ohm <sup>-1</sup> cm <sup>-1</sup> at 80 and 273 C, respectively.
9	34	L	1927	80,273		Electrolyte iron; 3	Obtained from Firma Heraeus; electrical conductivity reported as 61.2 and 9.4 x 10 <sup>4</sup> ohm <sup>-1</sup> cm <sup>-1</sup> at 80 and 273 C, respectively.
10	81	L	1939	4,9-23		Electrolyte iron	Extremely pure; 2.545 mm dia x 12.32 cm long; electrical resistivity ratio ρ(273 K)/ρ <sub>0</sub> = 29.4.
11	21	R	1951	718-1008	± 1	Armco Iron	Annular cylindrical specimen of 1.5 in. I.D., 6 in. O.D., and 2.25 in. thick.
12	91	C	1951	111-394		Armco Iron	0.035 Cu, 0.026 S, 0.015 Mn, 0.014 C, and 0.004 P; 2 cm dia x 15 cm long; Armco iron used as comparative material.
13	69	L	1937	323-961	1	Armco Iron; 1	0.032 Mn, 0.03 S, 0.015 C, 0.013 Cu, 0.01 Si, and 0.003 P; 1.4 cm dia x 10 cm long; annealed at 950 C.
14	69	L	1937	304-943	1	Armco Iron; 2	Similar to the above specimen.
15	77	E	1900	291,373		Fe I	0.1 C; 1.3007 cm dia x 27.0 cm long; density 7.84 g cm <sup>-3</sup> at 18 C; electrical conductivity reported as 8.357 and 5.950 x 10 <sup>4</sup> ohm <sup>-1</sup> cm <sup>-1</sup> at 18 and 100 C, respectively.

SPECIFICATION TABLE NO. 24 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
16	104	E	1951	16-93		6036	99.93 pure; forged measured in a vacuum of $5 \times 10^{-6}$ mm Hg.
17	57	L	1927	21, 83		Fe 1	Electrolytically refined; cold-worked and annealed; electrical resistivity reported as 0.681, 0.778, and 8.71 $\mu\text{hm cm}$ at -252, -190, and 0 C, respectively.
18	57	L	1927	21, 83		Fe 2	Technically pure; polycrystalline; electrolytically precipitated; electrical resistivity reported as 0.1437, 0.929, and 9.11 $\mu\text{hm cm}$ at -252, -190, and 0 C, respectively.
19	79	E	1933	90-273		Armco Iron	0.056 Cu, 0.026 S, 0.017 Mn, 0.011 C, 0.006 P, and 0.002 Si; 0.3924 cm dia x 14.53 cm long; electrical resistivity reported as 1.531, 5.74, 9.57, and 15.49 $\mu\text{hm cm}$ at -183.00, -78.50, 0, and 100 C, respectively.
20	699	L	1965	88-300			0.064 O, 0.0027 C, 0.002 S, 0.001 Mn, 0.001 Ni, 0.001 Si, and trace Cr; electrical resistivity as 1.22, 5.60, 6.50, 7.66, 9.96, 10.0, 11.3, 12.5, and 13.6 $\mu\text{hm cm}$ at 83, 203, 223, 243, 273, 293, 313, 333, and 353 K, respectively.
21	89	C	1956	273-1273			99.906 Fe (by difference), 0.035 Cu, 0.026 S, 0.015 Mn, 0.014 C, and 0.004 P; measured in a vacuum of $\sim 2 \times 10^{-5}$ mm Hg; a section of the specimen used as comparative material.
22	83	L	1956	1.5-128		JM 5092	99.99 Fe; 0.005 Ni, 0.0002 Cu, 0.0001 Ag; traces of Mn and Mg; 2 mm dia rod supplied by Johnson Matthey and Co.; annealed at 750 C for 4 hrs in vacuo; electrical resistivity reported as 0.248 and 10.0 $\mu\text{hm cm}$ at 4.2 and 293 K, respectively.
23	131	C	1953	323-1073	2	Swedish Iron	0.028 Si, 0.026 C, 0.021 P, 0.02 Mn, and 0.011 S; annealed in vacuum at 900 C; advance (55 Cu-45 Ni) used as comparative material.
24	71	L	1917	303-1107		Swedish Iron	Pure; cylindrical specimen of 1 cm dia x 8 cm long with a cavity of 6 mm dia x 16 mm long in its middle portion; electrical resistivity reported as 15.3, 20.0, 23.1, 26.3, 32.0, 36.4, 43.6, 45.3, 53.1, 61.6, 76.4, 80.7, 90.2, 92.1, 98.7, 100.0, 100.7, 102.2, 105.8, 108.6, 110.3, 111.3, 112.9, 113.8, 114.5, 115.1, 115.8, and 117.0 $\mu\text{hm cm}$ at 30, 94, 140, 183, 254, 302, 375, 390, 462, 527, 627, 660, 700, 709, 742, 746, 751, 760, 772, 789, 810, 817, 851, 862, 888, 890, and 901 C, respectively.
25	122	L	1955	2.0-93	$\pm 3$	JM 4975, Fe 1	99.99 pure; polycrystal; 0.202 cm dia x 2.89 cm long; supplied by Johnson-Matthey; annealed in vacuo for several hours at two thirds the melting temperature; electrical resistivity ratio $(\rho_{293K}/\rho_{20K}) = 67.4$ .
26	110, 476	L	1934	273-1073	2	Armco Iron	99.918 Fe, 0.025 Mn, 0.023 C, 0.020 S, 0.007 P, and 0.007 Si; 2.895 in. dia x 84 in. long; made from two similar rods of Armco ingot iron each 3 in. dia x 42 in. long; electrical resistivity reported as 9.6, 15.0, 22.6, 31.4, 43.1, 55.3, 69.8, 87.0, and 105.5 $\mu\text{hm cm}$ at 0, 100, 260, 300, 400, 500, 600, 700, and 800 C, respectively.
27	251, 260	P	1959	300-1298		Armco Iron; V	0.1875 in. dia x 2 in. long; machined from Armco stock supplied by Mapes and Sprowl Steel Co.; Curie point 770 C; transition point ( $\mu$ - $\gamma$ ) 910 C; measured in a vacuum of $\sim 5 \times 10^{-4}$ mm Hg; thermal conductivity values calculated from measured thermal diffusivity data and specific heat values taken from literature.
28	251	P	1959	295, 1256		Armco Iron; IV	Similar to above.

SPECIFICATION TABLE NO. 24 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Ranc. K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
29	1004		1966	629-973		Armco Iron	No details reported.
30	1005	E	1927	273.2		Armco Iron	0.125 in. dia x 10 cm long; electrical resistivity 15.8 $\mu$ ohm cm at 0 C.
31	444	C	1939	512-1046		Armco Iron	99.9:8 Fe, 0.025 Mn, 0.023 C, 0.020 S, 0.007 P, and 0.007 Si; 1.28 cm dia x 14 cm long; machined; lower section of the specimen used as comparative material.
32	444	C	1939	730-1138		Armco Iron	Similar to above except specimen of 1 in. dia.
33	444	R	1939	369-1278		Armco Iron	Specimen consisted of two superimposed disks of 1.1 cm I.D., 6.3 cm O.D., and 2.54 cm thick, machined from the similar original rods as above.
34	163	L	1936	303-1473		D	0.05 Cu, 0.040 S, 0.02 Mn, 0.011 P, 0.01 C, and trace of Si.
35	163	L	1936	303-1473		PD 00	0.08 Mn, 0.022 S, 0.02 C, 0.016 P, and 0.01 Si.
36	253	L	1959	6.5-90		Electrolytic Iron; 1	28 x 2.5 x 2.5 mm; doubly refined; cut out of a precipitated plate, annealed at 950 C, compressed, and reannealed in vacuo at 950 C; electrical resistivity ratio $\rho(273 \text{ K})/\rho_0 = 84.7-104.2$ ; residual electrical resistivity 0.09-0.10 $\mu$ ohm cm; specimen believed to be from the same material as that of Grunisen, E. and Goens, E. (see curve No. 17).
37	253	L	1959	7.9-90		2	Similar to the above specimen except dimensions 30 x 2.4 x 1.7 mm, and electrical resistivity reported as 0.092, 0.097, 0.100, 0.106, 0.120, 0.269, 0.368, 0.631, 0.744, 1.06, and 10.3 $\mu$ ohm cm at 4.2, 15.2, 20.8, 26.1, 32.5, 54.4, 61.2, 74.2, 79.1, 90.2, and 293.0 K, respectively.
38	276	C	1953	343	3	Armco Iron	Pure; density (25 C) = 7.9 g cm <sup>-3</sup> .
39	217	C	1959	410-1057	4	Armco Iron	99.745 Fe, 0.16 Cu, 0.03 C, 0.03 S, 0.015 Si, 0.01 Mn, and 0.01 P; obtained from commercial source in wrought form; iron used as comparative material (data of H. W. Deem).
40	64	L	1900	301.331		Armco Iron	99.93 Fe, 0.059 C; density 7.785 g cm <sup>-3</sup> .
41	203	L	1957	300	$\pm 1.5$	Armco Iron	Commercial Armco iron; electrical conductivity $9.09 \times 10^4$ ohm <sup>-1</sup> cm <sup>-1</sup> at 300 K.
42	340	L	1956	355-800	5	Armco Iron	0.75 in. dia rod.
43	97	L	1952	2.3-32	2-3	JM 4975; Fe 1	99.99 pure; 1-2 mm dia x 5 cm long; supplied by Johnson Matthey; annealed.
44	445	L	1960	303-1273	$\pm 2.5$	Armco Iron	0.083 Cu, 0.030 Mn, 0.023 S, 0.006 P, 0.004 Si, 0.02 C; specimen in two halves each of length 7.156 cm and dia 2.324 cm; supplied by BMI; annealed for 0.5 hr at 850 C; electrical resistivity reported as 10.4, 15.6, 22.9, 32.0, 42.9, 55.7, 70.3, 87.9, 106.0, 112.0, and 115.3 $\mu$ ohm cm at 0, 100, 200, 300, 400, 500, 600, 700, 800, 900, and 1000 C, respectively; measured in a vacuum of 0.2-3 x 10 <sup>-5</sup> mm Hg.
45	446	R	1959	855-1194		Armco Iron	Specimen consisted of three stacked hollow cylinders each of 2.5 in. O.D. and 2.5 in. high.



SPECIFICATION TABLE NO. 21 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)	Specifications and Remarks
46	356	R	1956	372-1172	± 7		Pure.	
47	439	L	1945	373-973	2		0.034 Mn, 0.03 C, 0.012 S, 0.005 P, and 0.003 Si.	
48	635	R	1957	373-1173	0.5		Pure.	
49	625		1960	904-1104	± 14	Armco Iron	No details reported.	1.9 x 1.9 x 0.100 cm; thermal conductivity values calculated from measured data of thermal diffusivity and specific heat, and density 7.87 g cm <sup>-3</sup> taken from Smithsonian Physical Tables (9th ed., 1954).
50	504	P	1961	295	± 5	Armco Iron		Wire specimen of 200 mm long.
51	673	E	1932	289.6	± 1.3	Electrolytic iron		
52	591	C	1963	338.2	± 1.8			99.82 pure; 0.500 in. dia x 3 in. long; hot-rolled; measured in a vacuum of $< 5 \times 10^{-4}$ mm Hg; nickel-plated copper used as comparative material.
53	626	P	1960	995-1294	± 2	Armco Iron		0.1475 in. dia x 2 in. long; machined from rod stock Armco iron obtained from Mapes and Sprowl; thermal conductivity values calculated from measured thermal diffusivity data and specific heat values of Darken, L.S. and Smith, R.F. (Ind. Eng. Chem., 43, 1415, 1951).
54	627	R	1964	385-1092	± 4.9	Armco Iron		0.1 Cu, 0.1 Ni, 0.086 O, 0.05 Mn, < 0.05 Al, < 0.05 Cr, < 0.05 Mo, 0.023 S, < 0.02 Si, < 0.02 V, 0.013 C, < 0.01 Ti, 0.006 P, 0.0050 N, and < 0.0003 H; grain size 20-40 $\mu$ ; electrical resistivity reported as 3.2, 5.3, 7.6, 10.2, 12.6, 15.9, 19.4, 23.3, 27.4, 32.7, 38.2, 44.0, 50.3, 56.7, 64.1, 72.0, 80.7, 90.2, 101.2, 104.7, 112.4, 114.5, 116.2, and 117.7 $\mu$ ohm cm at -130, -100, -50, 0, 50, 100, 150, 200, 250, 300, 350, 400, 450, 500, 550, 600, 650, 700, 750, 800, 850, 900, 950, and 1000 C, respectively; run no. 1.
55	627	R	1964	494-1194	± 4.9	Armco Iron		The above specimen; run No. 2A.
56	627	R	1964	973-1206	± 4.9	Armco Iron		The above specimen; run No. 2B.
57	627	R	1964	1206-1273	± 4.9	Armco Iron		The above specimen; run No. 2C.
58	627	R	1964	1025-1198	± 4.9	Armco Iron		The above specimen; run No. 2D.
59	57	L	1927	21.83		Fe 3		Polycrystalline; made by electrolytic method; hammered; tempered for 1 hr at 500 C; electrical conductivity reported as 1.060, 1.917, and 9.95 $\mu$ ohm cm at -252, -100, and 0 C, respectively.
60	664	E	1957	385-870	± 4	Armco Iron		0.045 S, 0.04 C, and 0.005 P; electrical resistivity reported as 16.36, 22.87, 32.13, 41.00, 53.48, 54.10, and 67.15 $\mu$ ohm cm at 111.7, 203, 301, 390, 493, 499, and 597 C, respectively.
61	634	R	1962	648-1263	± 10	Armco Iron		Specimen size 2 in. O.D., 3 in. long with a 0.5 in. center hole.
62	634	R	1962	753-1323	± 10	Armco Iron		The above specimen measured by using different heat sink.

SPECIFICATION TABLE NO. 21 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
63	661	R	1964	97-1273			0.012 O, 0.008 P, 0.007 C, 0.007 Al, 0.004 S, and 0.002 N; test disks annealed for several hrs at 900 C; electrical resistivity reported as 1.1, 1.7, 2.4, 3.1, 4.0, 5.0, 5.9, 6.9, 7.9, 8.8, 9.87, 11.6, 14.7, 18.1, 21.4, 26.0, 30.1, 35.0, 40.3, 46.8, 53.3, 60.1, 68.0, 76.0, 84.6, 94.2, 102.1, 106.3, 109.0, 111.1, and 113.1 $\mu\text{ohm cm}$ at -189, -160, -140, -120, -100, -80, -60, -40, -20, 0, 20, 50, 100, 150, 200, 250, 300, 350, 400, 450, 500, 550, 600, 650, 700, 750, 800, 850, 900, 950, and 1000 C, respectively.
64	660	P	1964	273-1373		Armco Iron	Armco iron obtained from BMI; thermal conductivity values calculated from thermal diffusivity and specific heat using density value $7.874 \text{ g cm}^{-3}$ given by Cleaves and Thompson measured data of (The Metal-Iron, McGraw-Hill, p. 271, 1935).
65	636	L	1954	316.2		Armco Iron	0.01 Si, traces of Ni, Cu, Al, Mn, Mo, and Ti (in order of decreasing amounts); 0.4 cm dia x 3.5 cm long; prepared by repeated electrolyse, machined; density $7.874 \text{ g cm}^{-3}$ . Disk specimen. 6.75 in. dia x 1.5 in. thick. Measured by Robinson, H. E., NBS.
66	629	R	1956	803-1049		Armco Iron	99.834 (by difference) Fe, 0.083 Cu, 0.030 Mn, 0.023 S, 0.02 C, 0.006 P, and 0.004 Si; chemical and spectrographic analysis at NPL showed 0.083 Ti in addition to the above; 1 in. rod from American Rolling Mill Co. in hot-rolled condition; annealed for 30 min at 1600 F (871 C) in air, followed by furnace cooling; electrical resistivity reported as 1.5, 3.3, 5.3, 7.5, 10.0, 12.5, 15.6, 19.1, 23.0, 27.0, 31.2, 36.3, 41.5, 47.4, 54.0, 61.0, 68.8, 77.2, 86.2, 96.8, 105.2, 109.8, 112.4, 114.1, and 115.8 $\mu\text{ohm cm}$ at -200, -150, -100, -50, 0, 50, 100, 150, 200, 250, 300, 350, 400, 450, 500, 550, 600, 650, 700, 750, 800, 850, 900, 950, and 1000 C, respectively; data taken from smoothed curve.
67	40, 43	L	1956	808-1153		Armco Iron	
68	217	L	1959	373-773		Armco Iron	
69	631	L, C	1962	73-1273		Armco Iron	
70	632, 900	L	1963	113-913		Armco Iron	99.834 Fe (by difference), 0.083 Cu, 0.030 Mn, 0.023 S, 0.02 C, 0.006 P, and 0.004 Si; specimen 37 cm in length and 2.386 cm dia; obtained from the American Rolling Mill Co. by Battelle Memorial Institute in the form of 1 in. rod; annealed for 1/2 hr at 1123 K; data taken from smoothed curve. Polycrystalline.
71	624		1959	323-673		Armco Iron	Corrected values for the temperature variation on the data (Curve 44) of Laubitz, 1960.
72	633	L	1963	373-1273		Armco Iron	99.995 Fe, < 0.0020 O, < 0.0006 N, 0.0004 C, 0.00015 Co, 0.00011 Cu, 0.000065 Cr, 0.00002 Ti, 0.000019 Ge, and 0.000018 V; polycrystalline; specimen 0.305 cm in dia made from commercial electrolytic iron; fabricated by swaging to 0.483 cm dia with intermediate annealing treatments in Pd-purified hydrogen, and after the final annealing at 650 C for 1/2 hr, a 0.305 cm dia gauge section was chemically polished into the specimen; final equiaxed grain size about 0.1 mm; electrical resistivity ratio $\rho(297\text{K})/\rho(4.2\text{K}) = 302$ .
73	670	L	1965	6.5-198		A-1	

SPECIFICATION TABLE NO. 24 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
74	670	L	1965	6.5-78		A-II	The second run of the above specimen.
75	670	L	1965	6.0-193		B	99.925 Fe, 0.0230 C, 0.0140 O, 0.0116 S, 0.0100 Si, 0.0040 P, 0.0023 Cu, 0.0017 Ti, 0.0016 Zr, 0.0013 Ce, 0.0010 Ni, 0.0009 Cr, 0.0009 Mg, 0.0007 Mn, 0.0005 As, 0.0004 Co, and 0.0003 Ca; polycrystalline; specimen made by vacuum melting commercial electrolytic iron in the conventional fashion; annealed; electrical resistivity ratio, $\rho(297K)/\rho(4.2K) = 27.1$ .
76	671	L	1961	100-280			"Very pure"; manufactured by Philips Research Labs, Eindhoven, Holland; wire 2.5 mm in dia, annealed at about 500 C for 10 hrs; electrical resistivity 9.8 $\mu\text{ohm cm}$ at 20 C.
77	671	L	1961	100-280			Spectroscopically standardized iron from Johnson, Matthey and Co.; rod 5.0 mm in dia; annealed at about 500 C for 10 hrs; electrical resistivity 9.9 $\mu\text{ohm cm}$ at 20 C.
78	695, 696	R	1965	425-773		Armco Iron	99.865 Fe (by difference), 0.07 Mn, 0.04 Si, 0.015 C, and 0.01 Cu; specimen consisted of two disks each with a length of 20 mm.
79	683		1963	0.42-0.95			99.97 pure; polycrystalline wire.
80	706	L	1981	273.373			Electrical conductivity 10.37 and 6.628 $\times 10^4 \text{ ohm}^{-1}\text{cm}^{-1}$ at 0 and 100° C (author reported 10.37 and 6.628 $\times 10^4$ , probably a typographical error).
81	841	L	1967	316-483		Pure iron No. 1	0.0250 Ni, 0.0100 Cu, 0.0100 Mn, 0.0070 Cr, 0.0050 C, 0.0040 O, 0.0040 S, 0.0040 V, 0.0030 P, 0.0010 Mn, < 0.0010 Si, 0.0006 N, and 0.000048 H; 1.27 cm dia x 15 cm long; supplied by Metals Research; electrical resistivity reported as 11.7, 14.7, 17.9, 21.6, and 25.6 $\mu\text{ohm cm}$ at 50, 100, 150, 200, and 250 C, respectively.
82	841	L	1967	307-499		Pure iron No. 2	0.0055 Ni, 0.0053 Si, 0.0038 Al, 0.0035 S, 0.0020 Co, 0.0017 P, 0.0014 C, 0.0010 Cr, < 0.0010 Mn, 0.0008 O, 0.0007 N, and 0.000016 H; short rod of 1.27 cm dia; prepared by Metallurgy Division of National Physical Laboratory; machined from disk; electrical resistivity reported as 11.9, 14.5, 18.2, 21.8, 25.8, 30.3, 41.0, 53.3, 67.9, 85.2, and 104.2 $\mu\text{ohm cm}$ at 50, 100, 150, 200, 250, 300, 400, 500, 600, 700, and 800 C, respectively.
83	841	L	1967	323-573		Pure free iron	0.0800 Si, 0.0300 C, 0.0150 P, 0.0100 Mn, and 0.0100 S; 2.54 cm dia x 20 cm long; supplied by Low Moor Best Yorkshire Iron Limited; electrical resistivity reported as 15.8, 18.7, 22.0, 25.9, 30.0, 34.6, 45.0, 57.1, 71.0, 87.5, and 107.2 $\mu\text{ohm cm}$ at 50, 100, 150, 200, 250, 300, 400, 500, 600, 700, and 800 C, respectively.
84	894	R	1964	332-1173	1	High purity iron	0.001 ~ 0.01 Ni, 0.001 ~ 0.01 Si, 0.003 C, 0.003 S, 0.0025 O, 0.0011 P, 0.0001 ~ 0.001 Al, 0.0001 ~ 0.001 Ca, 0.0001 ~ 0.001 Cu, 0.0005 N, 0.0001 H; prepared by arc-melting Armco iron stock in pure inert atmosphere to produce pancake shaped billets, rolled into sheets and cut to make feed stock for electron-beam melting, then cast into 4 in. dia x 6 in. long billet, trimmed off outside edges and machined from center portion two disks of dimensions 3.25 in. dia x 1.130 in. thick and 3.25 in. dia x 1.450 in. thick, four Armco iron disks added as end backup disks to form specimen column of 9 in. high, consisted of six disks in total; electrical resistivity reported as 1.037, 5.17, 9.04, 10.35, 11.06, 14.74, 21.92, 30.67, 41.07, 53.98, 66.33, 85.85, 105.48, 109.45, 112.35, 112.37, 113.92, and 115.30 $\mu\text{ohm cm}$ at -200, -84, 0, 24.37, 100, 200, 300, 398, 501, 598, 700, 801, 850, 900, 925, 964, and 1000 C, respectively.

SPECIFICATION TABLE NO. 24 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
85	894	R	1964	1193-1273	1	High purity iron	2nd run of the above specimen.
86	894	R	1964	1166-1272	1	High purity iron	3rd run of the above specimen; after 2nd run, temperature raised to 1065 C and left overnight to stabilize thermocouples.
87	312	C	1963	349-926		Armco Iron	1 in. dia x 0.250 in. thick; Armco iron used as comparative material.
88	895	R	1964	373-1273		Armco Iron	99.97 pure; disc shaped specimen 10.45 cm in dia with an axial hole 1.27 cm in dia; measured in a vacuum of $2 \times 10^{-6}$ mm Hg.
89	477	L	1962	397-636		Armco Iron	Specimen supplied by Abeles, B.
90	321	P	1956	336-1259		Armco Iron	Thermal conductivity values calculated from the measured data of thermal diffusivity specific heat data of Darkin, L. S. and Smith, R. P., (Ind. Engr. Chem., 43, p. 1815, 1951) and the density of $7.867 \text{ g cm}^{-3}$ .
91	916	R	1966	323-1273	$\pm 1.5$	High purity iron	99.95 Fe (approx); 0.002-0.02 Si, 0.014 C, 0.00095-0.0095 Ni, 0.0088 O, <0.0056 H, 0.0052 S, 0.00021-0.0021 Al, 0.002 P, 0.002 N, 0.00014-0.0014 Ca, and 0.00009-0.0009 Cu, obtained by electron-beam melting Armco iron; electrical resistivity reported as 0.40, 1.01, 5.31, 9.04, 11.72, 14.70, 18.06, 21.84, 26.10, 30.72, 35.90, 41.51, 47.53, 54.12, 61.22, 68.89, 77.10, 86.22, 96.46, 105.53, 109.58, 112.56, 113.09, 112.54, 113.66, and 115.49 $\mu\text{ohm cm}$ at -269, -195.7, -79.1, 0, 50, 100, 150, 200, 250, 300, 350, 400, 450, 500, 550, 600, 650, 700, 750, 800, 850, 900, 910, 920, 950, and 1000 C, respectively.
92	1004	1966	636-1102				Pure.
93	1005	E	1927	273.2			0.125 in. dia x 10 cm long; electrical resistivity $15.8 \mu\text{ohm cm}$ at 0 C; measured in a longitudinal magnetic field of 10000 gauss.
94	1005	E	1927	273.2			The above specimen measured in a transverse magnetic field of 4000 gauss.



DATA TABLE NO. 21 (continued)

T	k	T	k	T	k	T	k	T	k	T	k	T	k								
<u>CURVE 25 (cont.)</u>																					
773.2	0.440	629.2	0.314	791.2	0.416	323.2	0.527*	343.2	0.686	395.4	0.658	473.2	0.584								
873.2	0.398*	773.2	0.381	792.2	0.428	373.2	0.527*	414.3	0.639	414.3	0.639	573.2	0.528								
973.2	0.356	846.2	0.384	846.2	0.397	473.2	0.515*	416.5	0.630	416.5	0.630	673.2	0.474								
1073.2	0.314	883.2	0.375	903.2	0.393	573.2	0.485*	421.4	0.674	421.4	0.674	773.2	0.427								
<u>CURVE 27</u>																					
309.3	0.711	595.2	0.331	906.2	0.374	673.2	0.452*	409.5	0.632	421.4	0.704	873.2	0.383								
300.4	0.715	596.2	0.331	983.2	0.410*	773.2	0.410*	437.0	0.644	474.3	0.668	973.2	0.340								
307.9	0.702	600.2	0.331	1017.2	0.335	873.2	0.377*	463.2	0.615	474.3	0.680	1073.2	0.303								
325.3	0.614	601.2	0.335	1035.2	0.321	973.2	0.347*	473.1	0.607	504.3	0.599	1173.2	0.291								
318.0	0.530	602.2	0.335	1037.2	0.312	1073.2	0.331*	506.6	0.586	504.3	0.623	1273.2	0.303								
<u>CURVE 28*</u>																					
626.2	0.459	1038.2	0.320	1046.2	0.305	1173.2	0.326	511.0	0.590	520.9	0.633										
712.2	0.462	1082.2	0.300	1261.2	0.298	1373.2	0.326	515.4	0.594	577.1	0.645										
817.8	0.431	1082.2	0.299	1473.2	0.335	1473.2	0.335	529.4	0.577	539.2	0.561										
826.2	0.416	<u>CURVE 29*</u>																			
894.4	0.394	<u>CURVE 30</u>																			
932.1	0.365	512.2	0.585	1096.2	0.292	<u>CURVE 31*</u>															
994.9	0.349	724.2	0.462	1136.2	0.277	1096.2	0.292	1126.2	0.277	<u>CURVE 32*</u>											
1010.2	0.349	847.2	0.406	1159.2	0.277	1136.2	0.293	1156.2	0.277	6.5	1.6										
1036.9	0.279	885.2	0.385	1168.2	0.277	10.3	2.46	1182.2	0.271	13.8	3.17										
1047.1	0.295	893.2	0.369	1186.2	0.277	18.2	3.91	1192.2	0.271	16.2	4.36										
1069.0	0.290	951.2	0.345	1226.2	0.274	31.3	4.37	1226.2	0.274	740.9	0.448										
1094.2	0.292	1001.2	0.326	1278.2	0.277	36.3	4.05	1306.2	0.414	750.6	0.414										
1114.6	0.294	1046.2	0.308	<u>CURVE 33</u>																	
1140.7	0.294	<u>CURVE 34</u>																			
1144.7	0.291	303.2	0.724*	323.2	0.711*	373.2	0.674*	473.2	0.611*	573.2	0.548*	673.2	0.490*								
1162.2	0.298	730.2	0.452	773.2	0.435*	873.2	0.398*	973.2	0.377	1073.2	0.402	1173.2	0.389								
1171.2	0.305	856.2	0.384	1082.2	0.2925	1138.2	0.280	1173.2	0.398*	1273.2	0.389	1373.2	0.398								
1187.7	0.279	992.2	0.332	1138.2	0.280	1187.2	0.280	1278.2	0.277	1306.2	0.414	1473.2	0.406								
1190.9	0.286	1058.2	0.293	<u>CURVE 35</u>																	
1223.2	0.277	1082.2	0.2925	303.2	0.724*	323.2	0.711*	373.2	0.674*	473.2	0.611*	573.2	0.548*								
1259.8	0.276	1138.2	0.280	730.2	0.452	773.2	0.435*	873.2	0.398*	973.2	0.377	1073.2	0.402								
1281.0	0.288	1187.7	0.279	856.2	0.384	1082.2	0.2925	1138.2	0.280	1187.2	0.280	1278.2	0.277								
1285.2	0.289	1190.9	0.286	992.2	0.332	1138.2	0.280	1187.2	0.280	1278.2	0.277	1306.2	0.414								
1297.7	0.284	1223.2	0.277	1058.2	0.293	1082.2	0.2925	1138.2	0.280	1187.2	0.280	1278.2	0.277								
<u>CURVE 36*</u>																					
294.7	0.720	369.2	0.683	604.2	0.538	607.2	0.521	610.2	0.512	691.2	0.486	794.7	0.684								
1255.9	0.278	411.2	0.649	604.2	0.538	607.2	0.521	610.2	0.512	691.2	0.486	794.7	0.684								
<u>CURVE 37</u>																					
<u>CURVE 38*</u>																					
<u>CURVE 39*</u>																					
<u>CURVE 40</u>																					
<u>CURVE 41*</u>																					
<u>CURVE 42 (cont.)</u>																					
303.2	0.527*	303.2	0.527*	303.2	0.527*	303.2	0.527*	303.2	0.527*	303.2	0.527*	303.2	0.527*								
<u>CURVE 43</u>																					
<u>CURVE 44*</u>																					
<u>CURVE 45*</u>																					
<u>CURVE 46*</u>																					
<u>CURVE 47</u>																					
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<u>CURVE 95*</u>																					
<u>CURVE 96*</u>																					
<u>CURVE 97*</u>																					
<u>CURVE 98*</u>																					
<u>CURVE 99*</u>																					
<u>CURVE 100*</u>																					

\* Not shown on plot



DATA TABLE NO. 24 (continued)

CURVE 73 (cont.)		CURVE 74 (cont.)		CURVE 75 (cont.)		CURVE 76*		CURVE 77*		CURVE 78*		CURVE 79*		CURVE 74	
T	k	T	k	T	k	T	k	T	k	T	k	T	k	T	k
43.3	6.75	12.5	12.55	38.3	2.39	273.2	0.697	312.2	0.7367	336.2	0.2966	306.7	0.756	349.2	0.678
47.3	5.47	12.8	12.6*	50.6	1.92	373.2	0.681	394.2	0.6898	1193.2	0.2837	311.1	0.758	462.2	0.649
50.7	4.64	13.5	13.0	73.6	1.52*	470.2	0.623	470.2	0.6223	1210.2	0.2822	318.9	0.754	536.2	0.569
53.8	4.07	14.3	13.6	88.7	1.32*	563.2	0.5633	563.2	0.5633	1232.2	0.2866	345.6	0.735	447.2	0.630
57.7	3.48	15.5	13.75	96.1	1.26	643.2	0.5072	643.2	0.5072	1272.2	0.3000	366.8	0.719	490.2	0.603
62.3	2.95	16.2	14.1	115.4	1.14	723.2	0.4603	723.2	0.4603	1166.2	0.2966	416.3	0.684	496.2	0.616
70.7	2.33	17.3	14.1	121.2	1.11	804.2	0.4201	804.2	0.4201	1193.2	0.2837	451.8	0.651	539.2	0.582
81.7	1.87	18.6	14.3	134.4	1.09	887.2	0.3782	887.2	0.3782	1210.2	0.2822	471.0	0.631	599.2	0.546
86.5	1.74	20.0	14.4	153.6	1.05	973.2	0.3382	973.2	0.3382	1232.2	0.2866	487.1	0.609	645.2	0.520
91.5	1.64	21.0	14.5	173.0	1.03	1027.2	0.3121	1027.2	0.3121	1272.2	0.3000	498.5	0.629	669.2	0.521
96.5	1.53	21.5	14.4*	192.5	1.02	1069.2	0.2973	1069.2	0.2973	1166.2	0.2966	536.2	0.635	692.2	0.517
101.3	1.45	22.8	14.35	372.0	0.735	1173.2	0.2964	1173.2	0.2964	1193.2	0.2837	592.2	0.621	691.2	0.511
106.1	1.42*	23.9	14.3	372.0	0.735	1173.2	0.2964	1173.2	0.2964	1210.2	0.2822	691.2	0.621	691.2	0.511
110.9	1.37	24.1	14.15	385.3	0.716	385.3	0.716	385.3	0.716	1232.2	0.2866	710.2	0.621	691.2	0.511
115.7	1.33*	25.9	14.2	399.7	0.708	399.7	0.708	399.7	0.708	1272.2	0.3000	710.2	0.621	691.2	0.511
120.5	1.30	26.3	13.9	405.4	0.706	405.4	0.706	405.4	0.706	1166.2	0.2966	710.2	0.621	691.2	0.511
125.2	1.26*	27.5	13.6	426.2	0.685	426.2	0.685	426.2	0.685	1193.2	0.2837	710.2	0.621	691.2	0.511
130.0	1.24	28.1	13.55	429.3	0.675	429.3	0.675	429.3	0.675	1210.2	0.2822	710.2	0.621	691.2	0.511
134.0	1.22*	31.2	12.3	433.8	0.678	433.8	0.678	433.8	0.678	1232.2	0.2866	710.2	0.621	691.2	0.511
139.6	1.20	33.9	10.9	443.4	0.666	443.4	0.666	443.4	0.666	1272.2	0.3000	710.2	0.621	691.2	0.511
144.5	1.18*	35.4	10.2	465.7	0.651	465.7	0.651	465.7	0.651	1166.2	0.2966	710.2	0.621	691.2	0.511
149.3	1.17	38.4	1.91	483.3	0.638	483.3	0.638	483.3	0.638	1193.2	0.2837	710.2	0.621	691.2	0.511
154.1	1.15*	280		280	0.808	280	0.808	280	0.808	1210.2	0.2822	710.2	0.621	691.2	0.511
158.9	1.14									1232.2	0.2866	710.2	0.621	691.2	0.511
163.6	1.13*									1272.2	0.3000	710.2	0.621	691.2	0.511
168.1	1.12	5.9	0.8	425.2	0.651	425.2	0.651	425.2	0.651	1166.2	0.2966	710.2	0.621	691.2	0.511
173.6	1.12*	6.3	0.85	485.2	0.588	485.2	0.588	485.2	0.588	1193.2	0.2837	710.2	0.621	691.2	0.511
178.5	1.11	7.5	0.95	537.2	0.568	537.2	0.568	537.2	0.568	1210.2	0.2822	710.2	0.621	691.2	0.511
183.3	1.11*	7.8	0.98	597.2	0.522	597.2	0.522	597.2	0.522	1232.2	0.2866	710.2	0.621	691.2	0.511
188.0	1.11*	8.1	1.05	697.2	0.447	697.2	0.447	697.2	0.447	1272.2	0.3000	710.2	0.621	691.2	0.511
192.9	1.10*	10.2	1.2	773.2	0.438	773.2	0.438	773.2	0.438	1166.2	0.2966	710.2	0.621	691.2	0.511
197.8	1.10	13.1	1.5							1193.2	0.2837	710.2	0.621	691.2	0.511
		15.5	1.75							1210.2	0.2822	710.2	0.621	691.2	0.511
		16.6	1.8							1232.2	0.2866	710.2	0.621	691.2	0.511
		17.0	1.85							1272.2	0.3000	710.2	0.621	691.2	0.511
		17.5	1.87*							1166.2	0.2966	710.2	0.621	691.2	0.511
		18.0	1.95							1193.2	0.2837	710.2	0.621	691.2	0.511
		18.5	1.95*							1210.2	0.2822	710.2	0.621	691.2	0.511
		8.8	10.65							1232.2	0.2866	710.2	0.621	691.2	0.511
		9.0	10.7							1272.2	0.3000	710.2	0.621	691.2	0.511
		10.0	11.5							1166.2	0.2966	710.2	0.621	691.2	0.511
		10.5	12.1							1193.2	0.2837	710.2	0.621	691.2	0.511
		10.7	12.15*							1210.2	0.2822	710.2	0.621	691.2	0.511
		11.0	12.2							1232.2	0.2866	710.2	0.621	691.2	0.511
		11.5	12.15							1272.2	0.3000	710.2	0.621	691.2	0.511
		11.8	12.6*							1166.2	0.2966	710.2	0.621	691.2	0.511

\* Not shown on plot



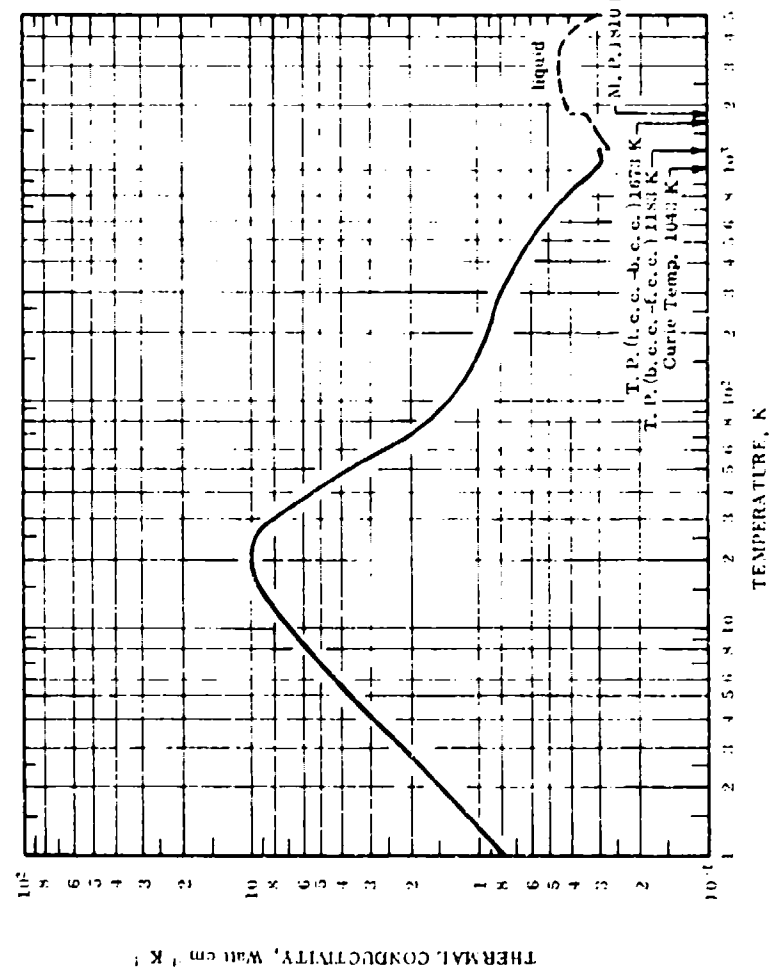
DATA TABLE NO. 24 (continued)

T	k
<u>CURVE 91 (cont.)<sup>e</sup></u>	
773.2	0.436
873.2	0.386
973.2	0.319
1023.2	0.313
1073.2	0.297
1123.2	0.296
1173.2	0.296
1183.2	0.297
1193.2	0.240
1223.2	0.247
1273.2	0.298
<u>CURVE 92<sup>c</sup></u>	
636.2	0.497
733.2	0.467
833.2	0.409
919.2	0.368
930.2	0.370
1008	0.331
1096	0.303
1102	0.310
<u>CURVE 93<sup>c</sup></u>	
273.2	0.765
<u>CURVE 94<sup>f</sup></u>	
273.2	0.771

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<sup>e</sup> Not shown on plot

FIGURE AND TABLE NO. 24R-1 RECOMMENDED THERMAL CONDUCTIVITY OF IRON



T <sub>1</sub>	k <sub>1</sub>	k <sub>2</sub>	T <sub>2</sub>	T <sub>1</sub>	k <sub>1</sub>	k <sub>2</sub>	T <sub>2</sub>
0	0	0	-459.7	500	0.613	35.4	440.3
1	(0.75) †	(43.3)	-457.9	600	0.547	31.6	620.3
2	1.49	86.1	-456.1	700	0.487	28.1	800.3
3	2.24	129	-454.3	800	0.433	25.0	960.3
4	2.97	172	-452.5	900	0.390	22.0	1160
5	3.71	214	-450.7	1000	0.326	18.8	1340
6	4.42	255	-448.9	1059	0.296	17.1	1447
7	5.13	296	-447.1	1100	0.297	17.2	1520
8	5.80	335	-445.3	1183	0.299	17.3	1670
9	6.45	373	-443.5	1183	0.279	16.1	1670
10	7.05	407	-441.7	1200	0.282	16.3	1700
11	7.62	440	-439.9	1300	0.299	17.3	1860
12	8.13	470	-438.1	1400	0.309	17.9	2060
13	8.58	496	-436.3	1500	0.318	18.4	2240
14	8.97	518	-434.5	1600	(0.327)	(18.9)	2420
15	9.30	537	-432.7	1673	(0.334)	(19.3)	2552
16	9.56	552	-430.9	1700	(0.336)	(19.4)	2600
18	9.88	571	-427.3	1800	(0.345)	(19.9)	2780
20	9.97	575	-423.7	1810	(0.346)	(20.0)	2798
25	9.36	541	-414.7				
30	8.14	470	-405.7				
35	6.81	393	-396.7				
40	5.55	321	-387.7				
45	4.50	250	-378.7				
50	3.72	215	-369.7				
60	2.65	153	-351.7				
70	2.04	118	-333.7				
80	1.68	97.1	-315.7				
90	1.46	84.4	-297.7				
100	1.32	76.3	-279.7				
150	1.04	60.1	-189.7				
200	0.94	54.3	-99.7				
250	0.865	50.0	-				
273.2	0.825	48.2	32.0				
300	0.803	46.4	80.3				
350	0.744	43.0	170.3				
400	0.694	40.1	260.3				
6000	(0.147)	( 8.49)	10340				
6500	(0.051)	( 2.95)	11240				
6750	( ~0 )	( ~0 )	11690				

REMARKS

The recommended values are for well-annealed 99.99% pure iron with residual electrical resistivity  $\rho_0 = 0.0327 \mu\Omega$  cm (characterization by  $\rho_0$  becomes important below room temperature). The values below 1.5 T<sub>m</sub> are calculated to fit the experimental data by using  $n = 2.10$ ,  $a = 0.37$ ,  $m = 2.47$ ,  $\alpha^0 = 2.05 \times 10^{-4}$ , and  $\beta = 1.34$ . The recommended values that are supported by experimental thermal conductivity data are thought to be accurate to within 3% of the true values near room temperature, and 3 to 8% at other temperatures.

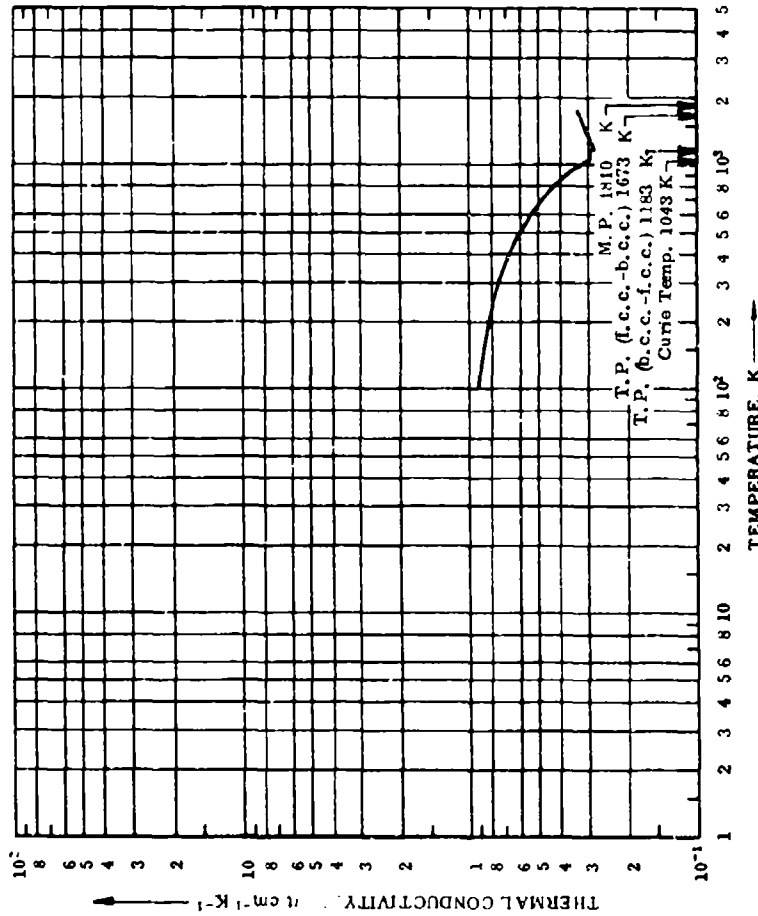
\* T<sub>1</sub> in K, k<sub>1</sub> in Watt cm<sup>-1</sup> K<sup>-1</sup>, T<sub>2</sub> in F, k<sub>2</sub> in Btu hr<sup>-1</sup> ft<sup>-1</sup> F<sup>-1</sup>.

† Values in parentheses are extrapolated or estimated.

In Liquid State

1810	(0.403)	(23.3)	2798
1900	(0.415)	(24.0)	2960
2000	(0.426)	(24.6)	3140
2200	(0.441)	(25.5)	3500
2400	(0.450)	(26.0)	3860
2600	(0.455)	(26.3)	4220
2800	(0.458)	(26.5)	4580
3000	(0.458)	(26.5)	4940
3200	(0.456)	(26.3)	5300
3400	(0.451)	(26.1)	5660
3600	(0.442)	(25.5)	6020
3800	(0.430)	(24.8)	6380
4000	(0.415)	(24.0)	6740
4500	(0.368)	(21.3)	7640
5000	(0.308)	(17.8)	8540
5500	(0.233)	(13.5)	9440
6000	(0.147)	( 8.49)	10340
6500	(0.051)	( 2.95)	11240
6750	( ~0 )	( ~0 )	11690

FIGURE AND TABLE NO. 24R-2 RECOMMENDED THERMAL CONDUCTIVITY OF ARMCO IRON  
 [Typical composition: 0.09 O; 0.08 Cu, Ni each; <0.05 Al, Cr, Mn, Mo each;  
 0.015 C, S, Si, Ti, V each; 0.005 N, P each; 0.0001 H]



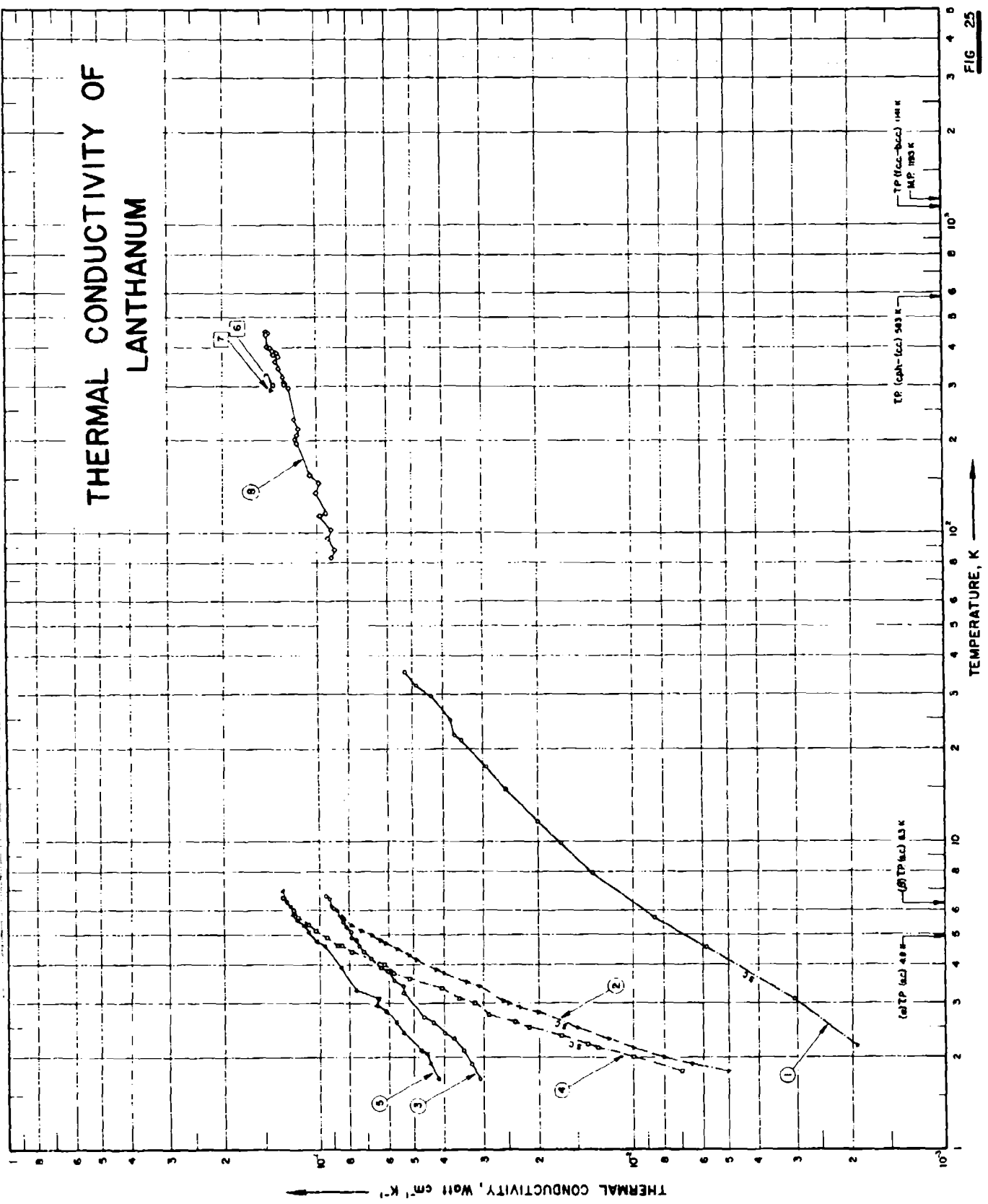
RECOMMENDED VALUES*		
$T_1$	$k_1$	$T_2$
100	0.913	-279.7
150	0.854	-189.7
200	0.804	-99.7
250	0.764	-9.7
273.2	0.747	32.0
300	0.727	80.3
350	0.691	170.3
400	0.657	260.3
500	0.593	440.3
600	0.531	620.3
700	0.473	800.3
800	0.422	980.3
900	0.372	1160
1000	0.323	1340
1059	0.293	1446
1100	0.294	1520
1180	0.286	1664
1190	0.286	1682
1200	0.287	1700
1300	0.296	1880
1400	0.305	2080
1500	0.314	2240
1600	(0.322)*	2420
1700	(0.330)	2600
1800	(0.338)	2780

REMARKS

The recommended values are thought to be accurate to within 3% of the true values below room temperature, 2% from room temperature to about 1000 K, and 3 to 8% from 1000 to 1600 K.

\*  $T_1$  in K,  $k_1$  in  $\text{Watt cm}^{-1} \text{K}^{-1}$ ,  $T_2$  in F, and  $k_2$  in  $\text{Btu hr}^{-1} (\text{ft}^{-1} \text{F}^{-1})$ . † Values in parentheses are extrapolated.

# THERMAL CONDUCTIVITY OF LANTHANUM



## SPECIFICATION TABLE NO. 25 THERMAL CONDUCTIVITY OF LANTHANUM

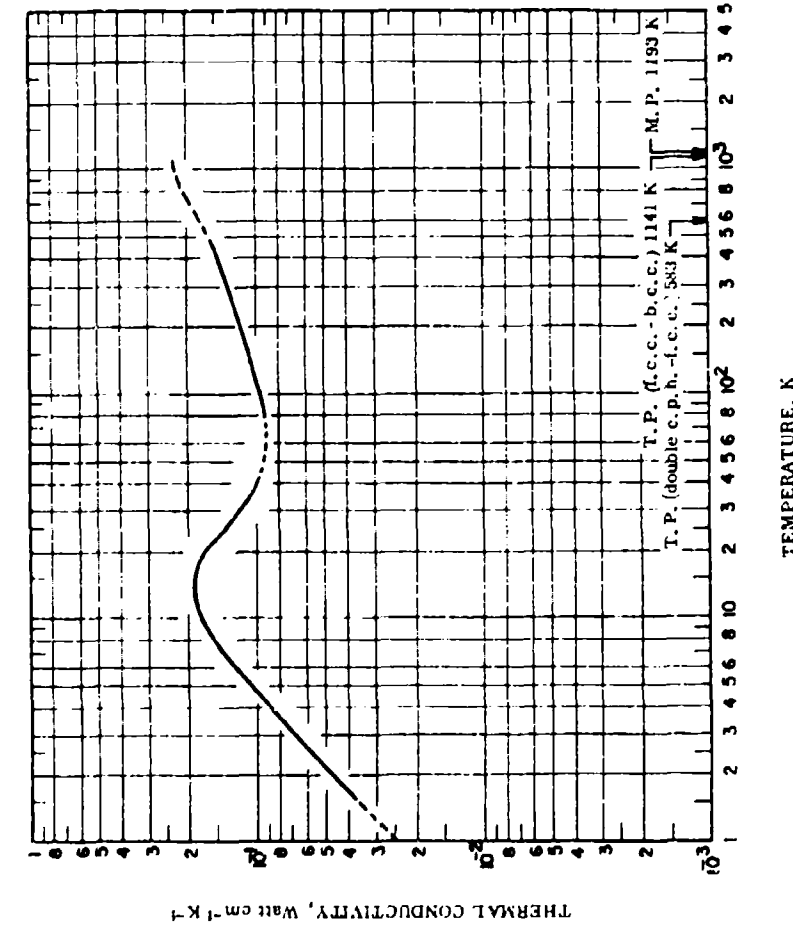
(Impurity &lt; 0.20% each; total impurities &lt; 0.50%)

[ For Data Reported in Figure and Table No. 25 ]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition ( weight percent), Specifications and Remarks
1	122	L	1955	2.2-36		La I	99.94 pure; Ca and Be as major impurities; polycrystalline; superconducting below 4.7 K.
2	762	L	1965	1.8-6.6		La I	99.99 nominal purity; polycrystalline rod, of f. c. c. form; 5 cm long, 0.4 cm in dia; supplied by H. Fleishman Ltd.; annealed at 600 C for 24 hrs; residual electrical resistivity $\rho$ (4.2 K) 1.72 $\mu$ ohm cm; electrical resistivity ratio $\rho$ (293 K)/ $\rho$ (4.2 K) = 32.4; specimen in superconducting state below the transition point of 6.04 K (determined magnetically); x-ray analysis showed a trace of h. c. p. phase dispersed in the f. c. c. phase.
3	762	L	1965	1.7-6.7		La I	The above specimen measured in a magnetic field of 6600 gauss; specimen in normal conducting state; Lorenz number, $L_0 = 2.79 \times 10^{-4} \text{ V}^2 \text{ K}^{-2}$ .
4	762	L	1965	1.8-6.6		La II	Similar to the above except annealed at 600 C for 106 hrs; residual electrical resistivity $\rho$ (4.2 K) = 1.29 $\mu$ ohm cm; electrical resistivity ratio $\rho$ (293 K)/ $\rho$ (4.2 K) = 44.0; specimen in superconducting state below the transition point of 6.04 K (determined magnetically); x-ray analysis showed a trace of h. c. p. phase dispersed in the f. c. c. phase.
5	762	L	1965	1.7-7.0		La II	The above specimen measured in a magnetic field of 6600 gauss; specimen in normal conducting state; Lorenz number, $L_0 = 2.83 \times 10^{-4} \text{ V}^2 \text{ K}^{-2}$ .
6	811		1954	301.2	10		No details given.
7	256	C	1966	291	$\pm 6$		< 0.01 rare earth metals, ~ 0.02 base metals; polycrystalline specimen 1 cm in dia and 1.2 cm long; electrical resistivity 61 $\mu$ ohm cm at 291 K; data point derived by the authors from measurements by 2 different thermal comparators.
8	932, 933	L	1966	83-450	$\pm 3$ to $\pm 5$	La I	6.1 O, 0.01 Ce, 0.005 Fe, 0.005 Cu, 0.005 Nd, and 0.005 Pr; hexagonal polycrystalline; electron-beam refined; electrical resistivity reported as 28.5, 37.3, 46.6, 53.9, 60.6, 66.1, 69.4, and 73.1 $\mu$ ohm cm at 98, 151, 199, 251, 300, 351, 397, and 447 K, respectively; measured in a vacuum of $10^{-4}$ to $10^{-5}$ mm Hg.



FIGURE AND TABLE NO. 25R RECOMMENDED THERMAL CONDUCTIVITY OF LANTHANUM



RECOMMENDED VALUES\*  
(For Polycrystalline)

T <sub>1</sub>	k <sub>1</sub>	k <sub>2</sub>	T <sub>1</sub>	T <sub>2</sub>	k <sub>1</sub>	k <sub>2</sub>	T <sub>1</sub>	T <sub>2</sub>
0	0	0	459.7	500	(0.162)	(9.36)	440.3	440.3
1	(0.0250)*	(1.44)	-457.9	600	(0.178)	(10.3)	620.3	620.3
2	0.0468	2.70	-456.1	700	(0.196)	(11.3)	800.3	800.3
3	0.0674	3.89	-454.3	800	(0.211)	(12.2)	980.3	980.3
4	0.0875	5.06	-452.5	900	(0.222)	(12.8)	1160	1160
5	0.107	6.18	-450.7	1000	(0.229)	(13.2)	1340	1340
6	0.124	7.16	-448.9	1100	(0.232)	(13.4)	1520	1520
7	0.140	8.09	-447.1					
8	0.154	8.90	-445.3					
9	0.166	9.59	-443.5					
10	0.176	10.2	-441.7					
11	0.183	10.6	-439.9					
12	0.188	10.9	-438.1					
13	0.191	11.0	-436.3					
14	0.192	11.1	-434.5					
15	0.191	11.0	-432.7					
16	0.188	10.9	-430.9					
18	0.179	10.3	-427.3					
20	0.168	9.71	-423.7					
25	0.141	8.15	-414.7					
30	0.121	6.99	-405.7					
35	0.108	6.24	-396.7					
40	(0.101)	(5.84)	-387.7					
45	(0.0969)	(5.60)	-378.7					
50	(0.0943)	(5.45)	-369.7					
60	(0.0927)	(5.36)	-351.7					
70	(0.0929)	(5.37)	-333.7					
80	(0.0941)	(5.44)	-315.7					
90	0.0958	5.54	-297.7					
100	0.0978	5.65	-279.7					
150	0.109	6.30	-189.7					
200	0.118	6.82	-99.7					
250	0.127	7.34	-9.7					
273.2	0.131	7.57	32.0					
300	0.135	7.80	80.3					
350	0.142	8.20	170.3					
400	0.149	8.61	260.3					

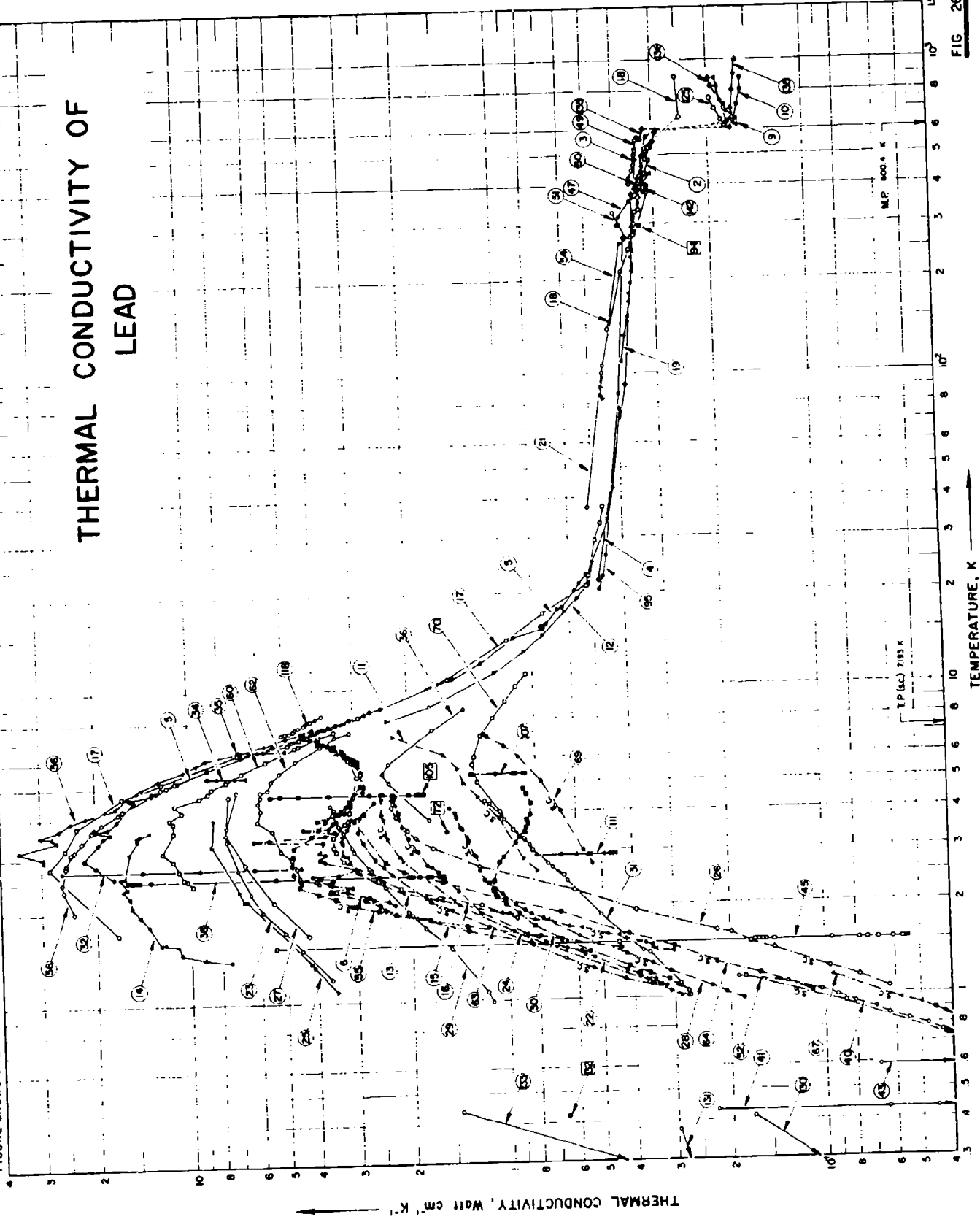
REMARKS

The recommended values are for well-annealed 99.99% pure lanthanum with residual electrical resistivity  $\rho_0 = 1.29 \mu\Omega \text{ cm}$  (characterization by  $\rho_0$  becomes important below about 150 K). The values that are supported by experimental thermal conductivity data are thought to be accurate to within 5% of the true values near room temperature and 5 to 15% at other temperatures.

\* T<sub>1</sub> in K, k<sub>1</sub> in Watt cm<sup>-1</sup> K<sup>-1</sup>, T<sub>2</sub> in F, and k<sub>2</sub> in Btu lb<sup>-1</sup> ft<sup>-1</sup> F<sup>-1</sup>. † Values in parentheses are extrapolated, interpolated, or estimated.

# THERMAL CONDUCTIVITY OF LEAD

FIGURE SHOWS ONLY 65 OF THE CURVES REPORTED IN TABLE





## SPECIFICATION TABLE NO. 26 THERMAL CONDUCTIVITY OF LEAD

(Impurity: 0.20% each total impurities: 0.50%)

[For Data Reported in Figure and Table No. 26 I

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)	Specifications and Remarks
1	48	C	1954	326-497				Specimen of 1 in. cube, cut and machined from a bar of melting-point lead supplied by NBS (sample No. 49 C); all surfaces carefully lapped; nickel used as comparative material.
2	129, 852	C	1933	273-510	3	L.S.		Bureau of Standard melting point standard lead; purity indicated by freezing point of 327.4 C; specimen 15 cm long, 2 cm in dia; melted in graphite and cast in bottom feed cast iron mold; all data referred to the value 0.352 Watt cm <sup>2</sup> K <sup>-1</sup> at 0 C taken from International Critical Tables, Volume II, p. 218.
3	84	P	1918	363-483				Pure, "squeezed" wire, 3.1 mm in dia; thermal conductivity values calculated from measured data of thermal diffusivity and the specific heat values taken from literature.
4	95	E	1915	22-374				99.99% pure; specimen 6.24 cm long, 0.2996 cm in dia; electrical resistivity reported as 19.26 and 20.64 $\mu$ ohm cm at 0 and 18 C, respectively.
5	63	L	1940	2.6-23				Single crystal, pure lead obtained from Adam Hilger Ltd. (H.S. brand); melted in high vacuum; filtered through a narrow glass opening, pressed in nitrogen into a glass tube of the desired shape then cooled slowly to make a specimen of 15 cm long, 2.5 mm in dia; transition point = 7.13 K; thermal conductivity data in normal state below transition point obtained by applying a transversal magnetic field of strength 472-510 gauss.
6	63	L	1940	2.0-7.1				The above specimen in superconducting state.
7	77	E	1900	291, 273				99.95 Pb (by difference), 0.05 total Cu, Bi, Fe, and Ni; 1.8028 cm dia x 27.0 cm long; density 11.32 g cm <sup>-3</sup> at 18 C; electrical conductivity reported as 4.84 and 3.64 x 10 <sup>4</sup> ohm <sup>-1</sup> cm <sup>-1</sup> at 18 and 100 C, respectively.
8	77	L	1900	291, 273		Lab No. 5973		Similar to the above specimen.
9	113	C	1957	623-873				99.995% pure; molten specimen contained in a thin-walled tube; electrical resistivity reported as 95.0, 97.2, 99.5, 102.0, 104.4, and 106.8 $\mu$ ohm cm at 350, 400, 450, 500, 550, and 600 C, respectively; 0.5% carbon steel used as comparative material.
10	85	L	1919	381-874				Cylindrical specimen.
11	18	L	1936	2.6-7.1				In superconducting state.
12	18	L	1936	7.9-77				No details reported.
13	117	L	1949	1.4-3.8		Pb II		High purity; single crystal; specimen 3.8 mm in dia obtained from Adam Hilger Ltd. (H.S. brand); in superconducting state.
14	117	L	1949	1.4-3.9		Pb II		The above specimen in normal state; measured in a longitudinal magnetic field of 850 oersteds.
15	117	L	1949	1.4-2.5		Pb III		Similar to the above specimen but 4.0 mm in dia; in superconducting state.

SPECIFICATION TABLE NO. 26 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)	Specifications and Remarks
16	97	L	1952	1.8-6.7	± 3	Pb I	99.998 pure; Tadanac lead; single crystal; in superconducting state.	
17	97	L	1952	1.7-38	± 3	Pb I	The above specimen in normal state.	
18	14	L	1940	138-887			No details reported.	
19	88	L	1908	109-299			Turned from a bar of pure lead supplied by Messrs. Hazendale, Manchester; density 11.29 g cm <sup>-3</sup> at 25 C; electrical resistivity reported as 6.71, 9.71, 12.9, 15.7, 18.5, and 20.9 $\mu$ ohm cm at -170, -129.4, -89.2, -51.8, -14.0, and 17.4 C, respectively.	
20	144	L	1952	317.2			Specimen cut from melting point standard lead supplied by NBS; 1.75 x 9.1875 x 0.1875 in.; measured in vacuo.	
21	16	F	1929	37-378			No details reported.	
22	257, 379	L	1958	1.1-4.6	± 2	Pb I	99.99 pure; monocrystal; obtained from Johnson & Matthey Co. Ltd. (No. 560); specimen ~7 cm long; 3 mm in dia; annealed in vacuo for several days at a few degrees below the melting point; residual electrical resistivity 0.008 $\mu$ ohm cm; in superconducting state.	
23	257, 379	L	1958	1.2-4.8	± 2	Pb I	The above specimen in normal state; measured in a transversal magnetic field of 1000 gauss.	
24	257	L	1958	1.0-4.6	± 2	Pb 2	99.99 pure; polycrystal; obtained from Johnson and Matthey Co. Ltd. (No. 560); grain size 0.5 mm; specimen ~7 cm long; 3 mm in dia; annealed in vacuo for several hrs at a few degrees below the melting point; residual electrical resistivity 0.008 $\mu$ ohm cm; in superconducting state.	
25	257	L	1958	1.1-3.9	± 2	Pb 2	The above specimen in normal state; measured in a transversal magnetic field of 1000 gauss.	
26	257	L	1958	1.1-4.6	± 2	Scroil	Pure; hollow cylindrical specimen 3 cm in dia made from lead foil 0.070 mm thick; annealed in vacuo for 5 days at a few degrees below the melting point; in superconducting state.	
27	257	L	1958	1.0-4.6	± 2	Scroil	The above specimen in normal state; measured in a magnetic field of 1000 gauss.	
28	257	L	1958	0.98-4.2	± 2	PbBi 0.02	99.98 Pb, 0.02 Bi, polycrystal with long crystals; specimen ~7 cm long; 3 mm in dia; annealed in vacuo for several hrs at a few degrees below the melting point; residual electrical resistivity 0.021 $\mu$ ohm cm; in superconducting state.	
29	257	L	1958	0.98-4.3	± 2	PbBi 0.02	The above specimen in normal state; measured in a magnetic field of 1000 gauss.	
30	257	L	1958	1.1-4.8	± 2	PbBi 0.1	99.899 Pb (by difference), 0.101 Bi; polycrystal; grain size 0.3 mm; specimen ~7 cm long; 3 mm in dia; annealed in vacuo for several hrs at a few degrees below the melting point; residual electrical resistivity 0.092 $\mu$ ohm cm; in superconducting state.	
31	257	L	1958	1.0-4.4		Pb Bi 0.1	The above specimen in normal state; measured in a magnetic field of 1000 gauss.	

SPECIFICATION TABLE NO. 26 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
32	342	L	1953	2.7		Pb 1	99.99% pure; single crystal; measured in transverse magnetic fields of strength ranging from 0.70 to 3.90 kilosterds.
33	342	L	1953	2.7		Pb 1	The above specimen measured in longitudinal magnetic fields of strength ranging from 0.87 to 3.94 kilosterds.
34	342	L	1953	5.3		Pb 1	The above specimen measured in transverse magnetic fields of strength ranging from 1.86 to 3.94 kilosterds.
35	342	L	1953	6.4		Pb 1	The above specimen measured in transverse magnetic fields of strength ranging from 0.52 to 3.94 kilosterds.
36	18	L	1936	3.8-8.6			Measured in a magnetic field of 764 gauss.
37	18	L	1936	6.39			The above specimen measured in a magnetic field of 765 gauss.
38	461	L	1951	2.5			99.99% pure; 0.5 cm dia x 10 cm long, measured in transverse magnetic fields of strength ranging from 0 to 921 gauss.
39	461	L	1951	2.5			The above specimen measured in transverse magnetic fields of decreasing strength ranging from 685 to 0 gauss.
40	462	L	1952	0.40-1.2			Single crystal; in superconducting state.
41	462	L	1952	0.43			The above specimen measured in magnetic fields with increasing strength ranging from 0 to 100% of the critical magnetic field.
42	462	L	1952	0.43			The above specimen measured in magnetic fields with decreasing strength ranging from 82 to 0% of the critical magnetic field.
43	462	L	1952	0.59			The above specimen measured in magnetic fields with increasing strength ranging from 0 to 100% of the critical magnetic field.
44	462	L	1952	0.59			The above specimen measured in magnetic fields with decreasing strength ranging from 72 to 37% of the critical magnetic field.
45	462	L	1952	1.5			The above specimen measured in magnetic fields with increasing strength ranging from 0 to 46% of the critical magnetic field.
46	462	L	1952	1.5			The above specimen measured in magnetic fields with decreasing strength ranging from 69 to 0% of the critical magnetic field.
47	214	F	1956	299-437	± 3		Nominally pure; electrical conductivity reported as 4.4, 4.25, 3.6, 3.05, 2.65, and 2.45 x 10 <sup>4</sup> ohm <sup>-1</sup> cm <sup>-1</sup> at 37, 50, 100, 150, 200, and 230 C, respectively.
48	466	C	1922	313.2	5		Pure lead specimen 3 cm long and 3 cm in dia; zinc used as a comparative material.
49	171	F	1944	405-570			Pure; single crystal; electrical resistivity reported as 29.67, 34.01, 39.66, 42.01, and 47.16 μohm cm at 405.1, 445.1, 499.1, 521.1, and 570.1 K, respectively.

SPECIFICATION TABLE NO. 26 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)	Specifications and Remarks
30	431	F	1944	350-540				Pure polycrystal; electrical resistivity reported as 30, 1, 38, 16, 42, 44, and 46, $\mu\Omega/\text{cm}$ at 190, 1, 467, 5, 489, 0, and 489, 9 K, respectively.
31	463	E	1924	273-326				Pure (supposed to be Kahlbaum's); 25 cm long, cross-sectional area 0.439 $\text{cm}^2$ .
32	482	L	1960	0.16-1.2		I		99.999 pure; single crystal, specimen 0.13 cm in dia $\sim$ 5.0 cm long; in superconducting state.
33	464	R	1917	255-310	1.2			Commercially pure (major impurity probably tin); specimen composed of 2 hollow hemispheres of 3.65 cm internal radius and 7 cm external radius.
34	463	C	1907	43-300				Pure specimen (Kahlbaum lead) 32.85 mm in dia and 7 cm long; copper used as comparative material.
35	466	L	1961	1.4-7.5		E		99.9% Pb (by difference), $\sim$ 0.1 metallic impurities; single crystal; enriched in isotopes of lead; specimen 1.54 cm long, 0.186 cm in dia; cast in high vacuum ( $10^{-5}$ mm Hg); annealed in vacuum for 5 hrs at 260 C; in superconducting state.
36	468	L	1961	2.9-7.3		E		The above specimen measured in a longitudinal magnetic field at 900 gauss; in normal state.
37	466	L	1961	1.5-7.7		D		99.95 Fe (by difference), 0.05 metallic impurities; specimen 2.40 cm long, 0.123 cm in dia; same fabrication method as the above specimen; in superconducting state.
38	466	L	1961	2.0-7.6		D		The above specimen measured in a longitudinal magnetic field of 900 gauss; in normal state.
39	466	L	1961	2.4-7.6		B		Similar to the above specimen but 2.26 cm long and 0.123 cm in dia; in superconducting state.
40	466	L	1961	2.4-7.3		B		The above specimen measured in a longitudinal magnetic field of 900 gauss; in normal state.
41	466	L	1961	2.4-7.7		C		Similar to the above specimen but 2.05 cm long and 0.123 cm in dia; in superconducting state.
42	466	L	1961	2.4-7.3		C		The above specimen measured in a longitudinal magnetic field of 900 gauss; in normal state.
43	349, 676	L	1958	1.0-4.4				99.99 pure; single crystal; straight wire; annealed at 270 C for 3 days; in superconducting state.
44	349, 676	L	1958	1.0-4.0				The above specimen bent at 4.2 K and annealed at 90 K; in superconducting state.
45	349, 676	L	1958	1.1-4.4				The above specimen annealed at 280 K; in superconducting state.
46	467	C	1953	313-429	$\pm 5$	55 NI - 1		NBS melting point standard lead; inconel used as comparative material.
47	412	L	1955	0.41-1.2	$\pm 5$			99.998 pure Tinbase lead; single crystal; measured without magnetic shielding; in superconducting state.

SPECIFICATION TABLE NO. 26 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
68	412	L	1955	0.30-0.87	± 5		The above specimen measured with magnetic shielding; in superconducting state.
69	508, 468	L	1950	2.7-7.2			99.98 Pb (by difference), 0.02 Bi; in superconducting state.
70	508, 468	L	1950	2.5-11			The above specimen in normal state; measured in a magnetic field.
71	230	L	1925	327.2			Baker's analyzed metal; total impurities < 0.03; rod 1.9 cm in dia and 10 cm long; electrical conductivity $4.76 \text{ ohm}^{-1}\text{cm}^{-1}$ at 22 C.
72	18	L	1936	3.47			Measured in a magnetic field of 1006 gauss.
73	18	L	1936	4.4, 4.6			The above specimen measured in a magnetic field of 956 gauss.
74	506	C	1953	322-414		55 M - 1	NBS melt point standard lead; specimen 0.350 in. in dia and 0.510 in. long; copper used as the comparative material.
75	506	C	1953	319-419		55 P - 1	Similar to the above specimen but 0.450 in. in dia and 0.509 in. long.
76	506	C	1953	319-385		55 J - 1	Similar to the above specimen but 0.250 in. in dia and 0.265 in. long.
77	506	C	1953	321-416		55 K - 1	Similar to the above specimen but 0.250 in. in dia and 0.528 in. long.
78	506	C	1953	316-398		55 L - 1	Similar to the above specimen but 0.300 in. in dia and 0.502 in. long.
79	506	C	1953	319-400		55 N - 1	Similar to the above specimen but 0.410 in. in dia and 0.489 in. long.
80	506	C	1953	316-436		55 N - 2	Similar to the above specimen but 0.410 in. in dia and 0.487 in. long.
81	506	C	1953	322-401		55 Q - 1	Similar to the above specimen but 0.500 in. in dia and 0.500 in. long.
82	506	C	1953	314-405		55 Q - 2	Similar to the above specimen but 0.500 in. in dia and 0.476 in. long.
83	96, 468	L	1950	2.6-0.4			99.9 Pb, 0.1 Bi; in normal state; measured in a magnetic field.
84	96, 468	L	1950	2.7-6.4			The above specimen in superconducting state.
85	509	C	1954	314-381	± 3	55 B - 1	Accurately ground specimen $0.500 \pm 0.001$ in. in dia and $0.500 \pm 0.005$ in. long; electrolytic deposited pure copper used as a comparative material; reference data of copper taken from International Critical Tables, Vol. 5, McGraw-Hill, p. 221, 1929.
86	509	C	1954	324-401	± 3	55 B - 2	Second run of the above specimen.
87	509	C	1954	314-414	± 3	55 B - 3	Third run of the above specimen.
88	735, 839	P	1965	850-1250	6-8		Molten specimen in a tantalum crucible made from 2 coaxial tubes with dia of 23.8 and 8 mm, each tube 0.12 mm thick; data calculated from measured data of thermal diffusivity and specific heat, and values of density taken from Slavinskii, M. P. [Physicochemical Properties of elements (in Russian), 1952].
89	510	L	1956	373-473		H	99.997+ pure electrolytic lead; specimen 20 mm in dia and 40 mm long.
90	510	L	1956	328-523		B	Similar to the above specimen.

SPECIFICATION TABLE NO. 26 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
91	510	L	1956	328-523		P	Rectangular specimen of the same purity as the above specimen; size 22 x 22 x 40 mm.
92	510	L	1956	380-510		A	Similar to the above specimen but 20 mm in dia and 59 mm long.
93	510	L	1956	373-473		L	Similar to the above specimen but only 40 mm long.
94	511	L	1918	298.0			Specimen radius 0.675 cm; furnished by "Erba".
95	619	L	1916	20-273			Lead (technical) specimen 0.5 cm in dia and 5 cm long; electrical conductivity reported as $173.57$ and $5.09 \times 10^4 \text{ ohm}^{-1}\text{cm}^{-1}$ at 20.4 and 273 K, respectively.
96	619	L	1916	21-273			Pure Kahlbaum lead specimen 0.5 cm in dia and 5 cm long.
97	592	L	1959	700-1130			In liquid state; melting point 327.4 C; measured in a vacuum of $5 \times 10^{-4}$ mm Hg.
98	702	E	1961	293-347	1		99.99 pure; size 0.184 x 2 x 6 in.; specimen cut from a prefabricated sheet.
99	703	C	1955	319-411			NBS melting point standard lead; data obtained by using 28 gauge iron-constantan thermocouples with OFHC copper used as comparative material.
100	703	C	1955	328-405			The above specimen measured by using 30 gauge copper-constantan thermocouples.
101	703	C	1955	317-376			The above specimen measured by using 24 gauge copper-constantan thermocouples.
102	706	L	1881	273, 373			23.7 cm long; electrical conductivity reported as $5.141$ and $3.602 \times 10^4 \text{ ohm}^{-1}\text{cm}^{-1}$ at 0 and 100 C, respectively. (The author reported as 5.141 and $3.602 \times 10^5 \text{ ohm}^{-1}\text{cm}^{-1}$ , obviously a typographical error.)
103	237	L	1950	2.7			99.998% pure (by difference), impurity < 0.002; cylindrical specimen prepared from Johnson Matthey H. S. lead; measured in longitudinal magnetic fields of increasing strength ranging from 0 to 1000 gauss.
104	237	L	1950	2.7			The above specimen measured in magnetic fields of decreasing strength ranging from 1000 to 0 gauss.
105	237	L	1950	4.6			The above specimen measured in magnetic fields of increasing strength ranging from 0 to 1000 gauss.
106	237	L	1950	4.6			The above specimen measured in magnetic fields of decreasing strength ranging from 610 to 33 gauss.
107	237	L	1950	5.29			About 99.98 Pb (by difference), 0.02 Bi; cylindrical specimen prepared from Johnson Matthey H. S. lead (impurity < 0.002%); measured in longitudinal magnetic fields of increasing strength 0 to 1000 gauss.
108	237	L	1950	5.29			The above specimen measured in longitudinal magnetic fields of decreasing strength ranging from 1000 to 45 gauss.
109	237	L	1950	5.40			The above specimen measured in transverse magnetic fields of increasing strength ranging from 0 to 1000 gauss.

SPECIFICATION TABLE NO. 26 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
110	237	L	1950	5.40			The above specimen measured in transverse magnetic fields of decreasing strength ranging from 996 to 0 gauss.
111	237	L	1950	2.89			The above specimen measured in transverse magnetic fields of increasing strength ranging from 0 to 1000 gauss.
112	237	L	1950	2.89			The above specimen measured in transverse magnetic fields of decreasing strength ranging from 1000 to 0 gauss.
113	237	L	1950	2.92			The above specimen measured in longitudinal magnetic fields of increasing strength ranging from 0 to 1000 gauss.
114	237	L	1950	2.92			The above specimen measured in longitudinal magnetic fields of decreasing strength ranging from 1000 to 0 gauss.
115	237	L	1950	2.6-21			The above specimen measured in a magnetic field greater than the critical field, in normal state.
116	237	L	1950	2.8-7.2			The above specimen measured before applying the magnetic field; in superconducting state.
117	237	L	1950	2.7-3.9			The above specimen measured after applying the magnetic field; in superconducting state.
118	693, 739	L	1963	7.2-8.3			Lead specimen grade 69 of the Consolidated Mining and Smelting Co.; single crystal; 0.25 in. in dia and 3 in. long; zone refined.
119	694, 720	L	1963	6.3-8.3			The above specimen measured in a magnetic field of 600 gauss; in normal state; data corrected to zero field.
120	694, 720	L	1963	6.3-8.3			The above specimen measured in a magnetic field of 680 gauss; in normal state; data corrected to zero field.
121	694, 720	L	1963	6.2-7.3			The above specimen measured in a magnetic field at 800 gauss; in normal state; data corrected to zero field.
122	694, 720	L	1963	5.5-7.2			The above specimen in superconducting state.
123	710	L	1945	291-333	> 2	NBS sample 49b	Calibration specimen for freezing point determination.
124	710	L	1945	302-330	> 2	NBS sample 49b	The above specimen remeasured with a slightly different method of balancing thermocouples to avoid radial heat losses.
125	707	C	1953	617-755			0.10 foreign non-volatile matter; 0.001 Ag, and 0.001 other foreign metals; in normal state; produced by Mallinckrodt Chemical Works; test cylinder 3 in. in dia; also iron used as comparative material; data taken from smoothed curve, corrected for the effect of transients.

SPECIFICATION TABLE NO. 26 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
126	731	L	1952	0.13-0.29			99.998 <sup>4</sup> pure; provided by Johnson Matthey & Co. Ltd., London, (Batch No. 3620), specimen size 3 x 2.1 x 25 mm; warm up number 1; in superconducting state.
127	731	L	1952	0.18-0.36			The above specimen, warm up number 2.
128	731	L	1952	0.19-0.35			The above specimen, warm up number 3.
129	732	L	1965	0.015-0.23			99.999 pure (nominal); supplied by Koch-Light Laboratories Ltd. (Colnbrook, England); wire 5 cm long and 0.5 mm in dia; measured in a longitudinal magnetic field of 1000 gauss; in normal state.
130	733	L	1962	0.11-0.41			99.995 pure; polycrystalline; material obtained from Central Research Laboratories, American Smelting and Refining Co.; ratio of specimen cross sectional area to length 3.47 x 10 <sup>-3</sup> cm; cut and rolled from a lead bar of the mentioned purity; annealed at room temperature for many weeks; measured in a longitudinal magnetic field of 900 gauss; in normal state.
131	733	L	1962	0.20-0.33			The above specimen measured in a transverse magnetic field of 3000 gauss.
132	733	L	1962	0.42			The above specimen measured in a transverse magnetic field of 2000 gauss.
133	733	L	1962	0.14-0.41			The above specimen measured in a transverse magnetic field of 1000 gauss.
134	734	R	1926	323.2			Specimen in the form of a long hollow cylinder.
135	744	P	1966	560-1355	7-8		Molten lead filled the space between two coaxial thin-walled tubes of tantalum of 24 and 8 mm dia, respectively; thermal conductivity values calculated from measured data of thermal diffusivity and specific heat.
136	838	C	1967	474-870			Molten specimen placed in a hole 21 mm in dia drilled in an asbestos cement cylinder 30 mm in height; ICH18N9T steel used as comparative material.
137	840	L	1966	223-573	< 1		Lead specimen cut from bar of 99.999 pure or better; supplied by Dept. of Mines and Technical Surveys, Ottawa; smoothed values (experimental point deviations less than 1.5%).
138	840	L	1966	223-573	< 1		Lead specimen cut from the same bar as above and measured by another apparatus with modifications of the thermal shielding.
139	841	L	1967	335-602	13		99.995 <sup>4</sup> Pb, 0.001 Cd, 0.0005 Ag, 0.0005 Cu, 0.0005 Bi; 7 mm dia x 15 cm long; supplied by Johnson Matthey & Co.; electrical resistivity reported as 19.3, 23.4, 27.5, 31.8, 36.3, 40.8, and 45.7 μ ohm at 0, 50, 100, 150, 200, 250, and 300 C, respectively.
140	842	C	1967	316-420		Pyrometric standard lead 49 c	0.03 Bi, 0.002 Ag, 0.002 Cd, 0.001 Fe, 0.001 Ni, 0.001 Si, 0.001 Te, 0.0005 Cu, 0.0005 Sn, and 0.0001 Mg; electrical resistivity reported as 0.394, 0.735, 4.84, and 21.31 μ ohm cm at 20, 25, 77, and 298 K, respectively; M. P. 327.3 C; Armco iron used as comparative material.



SPECIFICATION TABLE NO. 26 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
141	842	C	1967	323-434		Pyrometric standard lead 49 e	0.001 Fe, 0.001 Ni, 0.001 Si, < 0.001 Te, 0.0005 Ag, 0.0005 Bi, 0.0005 Cu, < 0.0005 Cd, < 0.0005 Sn, and 0.0001 Mg; electrical resistivity reported as 0.366, 0.685, 4.83, and 21.25 $\mu$ ohm cm at 20, 25, 77, and 298 K, respectively. M.P. 327.417 C, Armco Iron used as comparative material.
142	870	L	1949	338-399			No details reported.

DATA TABLE NO. 26 THERMAL CONDUCTIVITY OF LEAD

(Impurity < 0.20% each; total impurities < 0.50%)

[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup>K<sup>-1</sup>]

CURVE 1		CURVE 2		CURVE 5 (cont.)		CURVE 6 (cont.)		CURVE 7		CURVE 12		CURVE 13 (cont.)		CURVE 15		CURVE 16		CURVE 17		CURVE 18	
T	k	T	k	T	k	T	k	T	k	T	k	T	k	T	k	T	k	T	k	T	k
325.7	0.342	273.2	0.352	4.84	13.2	2.38	3.57	291.2	0.346	7.90	2.326	3.59	3.17	1.41	0.55	1.82	0.937	1.69	10.710	3.38	3.485
325.9	0.347	329.2	0.3412	4.85	13.3	2.42	3.64	373.2	0.341	8.77	1.695	3.74	3.13	1.52	0.85	2.01	1.201	2.78	28.225	3.72	3.228
326.2	0.351	333.2	0.3399	5.19	10.9	2.54	4.10			11.10	1.953	3.75	3.13	1.67	1.33	2.30	2.152	3.20	25.000	3.74	3.102
328.2	0.348	395.2	0.3278	5.67	8.70	2.56	3.39			13.00	0.847	3.84	3.10	1.80	1.76	2.84	3.274	5.14	12.270	3.75	3.102
328.2	0.342	436.2	0.3170	6.19	6.76	2.57	2.99			14.60	0.735	3.84	3.10	1.80	1.76	3.04	3.459	6.28	6.967	3.76	3.102
347.2	0.347	439.2	0.3218*	6.68	5.35	2.83	3.75	291.2	0.343	17.30	0.621	1.37	7.50	1.93	14.50	3.14	3.498	4.65	16.285	3.77	3.102
347.7	0.341	481.2	0.3060	7.15	4.22	2.87	3.88	373.2	0.339	19.10	0.565	1.45	10.80	2.06	15.40	3.36	3.485	6.71	4.818	3.78	3.102
358.2	0.347	483.2	0.3140	7.17	4.22	2.96	3.72			21.00	0.524	1.55	11.00	2.21	16.60	3.72	3.228	8.18	22.820	3.78	3.102
364.2	0.342	526.2	0.3070	7.17	4.13	3.09	3.88			25.00	0.500	1.67	12.20	2.39	15.90	4.04	3.102	10.71	1.492	3.79	3.102
364.2	0.342	550.2	0.3037	7.31	4.07	3.26	3.77			34.00	0.442	1.79	13.70	2.56	16.70	6.73	3.868	12.70	3.54	3.80	3.102
365.2	0.339			7.42	3.85	3.34	6.21	623.2	0.160	43.00	0.326	1.93	14.50	2.58	16.10	6.96	4.422	14.10	0.951	3.81	3.102
381.7	0.339			7.43	3.83	3.35	5.59	627.2	0.164			2.06	15.40	2.58	16.10	7.16	4.186	17.16	0.731	3.82	3.102
383.2	0.342			7.52	3.75	3.44	3.72	647.2	0.167			2.21	15.60	2.58	16.10	8.18	4.222	22.06	0.527	3.83	3.102
394.7	0.338	363.2	0.346	7.63	3.56	3.53	3.46	648.2	0.159			2.39	15.90	2.56	16.70	8.28	6.967	22.51	0.516	3.84	3.102
395.2	0.336	365.2	0.345*	7.76	3.38	3.61	3.61	673.2	0.166			2.56	16.70	2.56	16.70	8.28	6.967	29.28	0.491	3.85	3.102
404.2	0.335	483.2	0.337	8.01	3.06	3.73	3.52	683.2	0.173			2.56	16.70	2.56	16.70	8.28	6.967	33.06	0.470	3.86	3.102
404.2	0.332	483.2	0.340	8.05	3.06*	3.80	3.42	718.2	0.175			2.56	16.70	2.56	16.70	8.28	6.967	37.50	0.461	3.87	3.102
413.2	0.336			8.32	2.75	3.82	3.94	783.2	0.184			2.56	16.70	2.56	16.70	8.28	6.967	37.50	0.461	3.88	3.102
413.2	0.332			8.43	2.65	3.82	3.94	823.2	0.184			2.56	16.70	2.56	16.70	8.28	6.967	37.50	0.461	3.88	3.102
427.8	0.331			8.69	2.43	4.03	3.34	873.2	0.187			2.56	16.70	2.56	16.70	8.28	6.967	37.50	0.461	3.88	3.102
428.2	0.332			10.74	1.45	4.05	3.21					2.56	16.70	2.56	16.70	8.28	6.967	37.50	0.461	3.88	3.102
436.7	0.331	21.8	0.480	11.98	1.16	4.25	3.19					2.56	16.70	2.56	16.70	8.28	6.967	37.50	0.461	3.88	3.102
439.4	0.326	91.7	0.378	14.38	0.913	4.63	3.03					2.56	16.70	2.56	16.70	8.28	6.967	37.50	0.461	3.88	3.102
445.7	0.329	273.1	0.350*	15.47	0.730	4.63	2.99					2.56	16.70	2.56	16.70	8.28	6.967	37.50	0.461	3.88	3.102
446.2	0.327	295.1	0.349	15.57	0.746	4.73	2.82					2.56	16.70	2.56	16.70	8.28	6.967	37.50	0.461	3.88	3.102
455.7	0.330	373.8	0.349	15.85	0.714	4.84	2.79					2.56	16.70	2.56	16.70	8.28	6.967	37.50	0.461	3.88	3.102
456.0	0.323			17.63	0.658	5.22	2.82					2.56	16.70	2.56	16.70	8.28	6.967	37.50	0.461	3.88	3.102
464.2	0.322			17.92	0.633	5.25	2.85					2.56	16.70	2.56	16.70	8.28	6.967	37.50	0.461	3.88	3.102
465.2	0.326	2.58	17.5	17.92	0.633	5.25	2.85					2.56	16.70	2.56	16.70	8.28	6.967	37.50	0.461	3.88	3.102
476.2	0.323	2.81	20.0	20.19	0.575	5.28	2.87					2.56	16.70	2.56	16.70	8.28	6.967	37.50	0.461	3.88	3.102
477.2	0.322	2.97	21.7	21.31	0.552	5.68	2.91					2.56	16.70	2.56	16.70	8.28	6.967	37.50	0.461	3.88	3.102
490.2	0.322	3.33	19.6	22.70	0.526	5.80	2.99					2.56	16.70	2.56	16.70	8.28	6.967	37.50	0.461	3.88	3.102
490.7	0.323	3.49	18.9			5.80	2.99					2.56	16.70	2.56	16.70	8.28	6.967	37.50	0.461	3.88	3.102
496.2	0.318	3.59	17.9			5.92	2.99					2.56	16.70	2.56	16.70	8.28	6.967	37.50	0.461	3.88	3.102
497.2	0.314	3.71	18.2			6.02	3.05					2.56	16.70	2.56	16.70	8.28	6.967	37.50	0.461	3.88	3.102
		3.71	18.2			6.14	3.12					2.56	16.70	2.56	16.70	8.28	6.967	37.50	0.461	3.88	3.102
		3.94	16.9			6.14	3.15					2.56	16.70	2.56	16.70	8.28	6.967	37.50	0.461	3.88	3.102
		4.55	14.9			6.24	3.23					2.56	16.70	2.56	16.70	8.28	6.967	37.50	0.461	3.88	3.102
		4.60	15.4			6.41	3.38*					2.56	16.70	2.56	16.70	8.28	6.967	37.50	0.461	3.88	3.102
		4.75	13.3			6.44	3.40*					2.56	16.70	2.56	16.70	8.28	6.967	37.50	0.461	3.88	3.102
		4.81	12.7			6.51	3.47*					2.56	16.70	2.56	16.70	8.28	6.967	37.50	0.461	3.88	3.102
						6.51	3.47*					2.56	16.70	2.56	16.70	8.28	6.967	37.50	0.461	3.88	3.102
						6.76	4.10					2.56	16.70	2.56	16.70	8.28	6.967	37.50	0.461	3.88	3.102
						6.96	4.26					2.56	16.70	2.56	16.70	8.28	6.967	37.50	0.461	3.88	3.102

\* Not shown on plot



DATA TABLE NO. 36 (continued)

$H_2O(H_2O)$	$R$	$H_2O(H_2O)$	$k$	$T$	$k$	$T$	$k$	$T$	$k$	$T$	$k$	$T$	$k$	$T$	$k$
<u>CURVE 43</u> (T = 0.59K)		<u>CURVE 46<sup>o</sup></u>		<u>CURVE 51</u>		<u>CURVE 55 (cont.)</u>		<u>CURVE 56 (cont.)</u>		<u>CURVE 57 (cont.)</u>		<u>CURVE 59<sup>o</sup> (cont.)</u>		<u>CURVE 60 (cont.)</u>	
0	0.0112 <sup>o</sup>	69	1.35	273.2	0.367	2.35	4.50	5.82	9.4	6.98	3.72	3.21	2.41	6.57	4.57
14	0.0197 <sup>o</sup>	58	0.699	299.7	0.385	2.43	4.45	6.06	8.8	7.09	4.00	3.31	2.47	6.61	4.46
27	0.0227 <sup>o</sup>	53	0.469	326.3	0.430	2.59	4.70	6.39	6.75	7.18	4.20	3.48	2.53	7.05	3.83
44	0.0256 <sup>o</sup>	47	0.294			2.76	4.77	6.67	5.70	7.31	3.95	3.74	2.59	7.34	3.44
55	0.0239	44	0.169	<u>CURVE 52</u>		2.94	4.74	7.00	4.90	7.49	3.67	3.86	2.53	<u>CURVE 61<sup>o</sup></u>	
72	0.0223 <sup>o</sup>	42	0.126	0.157	0.0004 <sup>o</sup>	3.14	4.70	7.32	4.27	7.68	3.42	4.08	2.44		
83	0.0136 <sup>o</sup>	18	0.123	0.184	0.0008 <sup>o</sup>	3.40	4.10					4.22	2.46		
93	0.0139 <sup>o</sup>	0	0.120	0.225	0.0012 <sup>o</sup>	3.61	3.90	<u>CURVE 57<sup>o</sup></u>		<u>CURVE 58</u>		4.24	2.41		
80	0.0206 <sup>o</sup>	$T$	$k$	0.275	0.00222 <sup>o</sup>	3.83	3.62 <sup>o</sup>	1.51	0.435	2.00	23.6	4.45	2.35		
89	0.0348 <sup>o</sup>	<u>CURVE 47</u>		0.37	0.0040 <sup>o</sup>	4.00	3.40 <sup>o</sup>	1.70	0.81	2.35	25.5	4.53	2.35		
100	0.0680	298.2	0.393	0.45	0.0090 <sup>o</sup>	4.04	3.44 <sup>o</sup>	1.85	1.28	2.49	25.6	5.37	2.35		
<u>CURVE 44<sup>o</sup></u> (T = 0.59K)		409.2	0.356	0.47	0.0098 <sup>o</sup>	4.13	3.30	2.05	1.98	2.80	25.0	5.44	2.36		
72	0.170	437.2	0.305	0.56	0.0183 <sup>o</sup>	4.40	3.10	2.20	2.17	2.86	24.8	5.55	2.36		
63	0.194	<u>CURVE 48<sup>o</sup></u>		0.72	0.039 <sup>o</sup>	4.55	2.91	2.27	2.25	3.19	24.0	5.87	2.42		
54	0.102	284.5	0.375	0.94	0.086	4.81	2.96 <sup>o</sup>	2.37	3.14	3.16	23.4	5.98	2.47		
45	0.0439	409.2	0.322	1.15	0.178	4.90	2.86	2.45	3.41	3.36	22.1	6.09	2.52		
37	0.0177	437.2	0.305	<u>CURVE 53<sup>o</sup></u>		5.42	2.96	2.55	3.60	3.62	20.5	6.26	2.62		
<u>CURVE 45</u> (T = 1.5K)		313.2	0.356	284.5	0.375	5.88	2.96	2.93	4.00	3.69	20.0	6.60	2.90		
0	0.156	<u>CURVE 49</u>		295.2	0.361	6.19	3.12 <sup>o</sup>	3.11	3.95	3.85	19.0	6.68	2.94		
6.2	0.154	405.1	0.356	309.9	0.346	6.35	3.22	3.25	3.87	4.15	16.7	6.77	3.05		
11	0.172	421.1	0.354	<u>CURVE 54</u>		6.57	3.45	3.30	3.61	4.21	16.4	7.08	3.52		
23	0.166	445.1	0.346	83.2	0.448	6.65	3.60	3.49	3.61	4.92	12.2	7.14	3.45		
29	0.161	467.1	0.343	85.2	0.456	6.74	3.74 <sup>o</sup>	3.75	3.39	5.00	11.8	7.19	3.61		
34	0.156 <sup>o</sup>	499.1	0.339	90.2	0.452	6.89	3.95	3.97	3.20	5.15	11.0	7.23	3.52		
41	0.154 <sup>o</sup>	499.1	0.337	261.2	0.385	6.99	4.05 <sup>o</sup>	4.07	3.12	5.22	10.7 <sup>o</sup>	7.62	3.52		
47	0.0775	515.1	0.337	273.2	0.346 <sup>o</sup>	7.14	4.23 <sup>o</sup>	4.16	3.09	5.45	9.5 <sup>o</sup>	<u>CURVE 60</u>			
50	0.0676	521.1	0.336	295.2	0.342 <sup>o</sup>	7.25	4.35	4.25	3.00	5.57	9.06 <sup>o</sup>	2.41	9.8		
53	0.0565	555.1	0.336	298.2	0.345 <sup>o</sup>	7.45	4.00	4.25	3.00	5.73	8.2 <sup>o</sup>	2.48	10.2		
55	0.0556	566.1	0.333	299.7	0.353 <sup>o</sup>			4.48	2.88	5.82	8.0 <sup>o</sup>	2.52	10.5		
59	0.0546	570.1	0.331	<u>CURVE 55</u>				4.65	2.81	6.00	7.2 <sup>o</sup>	2.69	10.8		
61	0.0559			2.92	29.5			4.97	2.72	6.14	6.75 <sup>o</sup>	2.90	11.3		
64	0.0602	1.44	0.44	3.00	29.5			5.08	2.72	6.58	5.40 <sup>o</sup>	2.90	11.3		
70	0.0730	1.59	0.79	3.16	35.0			5.14	2.72	6.87	4.65 <sup>o</sup>	3.08	12.3		
76	0.0943	1.72	1.34	3.39	26.3			5.25	2.72	7.24	4.05 <sup>o</sup>	3.17	11.3		
81	0.145	1.90	2.25	3.49	27.7			5.48	2.77	7.58	3.50 <sup>o</sup>	3.83	11.3		
86	0.433	2.03	3.06	3.52	30.0			5.65	2.77			3.91	11.6		
<u>CURVE 49</u>		2.22	3.93	3.69	26.3			5.69	2.70	<u>CURVE 59<sup>o</sup></u>					
<u>CURVE 50</u>		1.59	0.79	3.89	24.1			6.03	2.85	2.40	1.58				
<u>CURVE 51</u>		1.72	1.34	3.94	21.5			6.13	2.94	2.49	1.70				
<u>CURVE 52</u>		1.90	2.25	4.02	17.1			6.18	3.00	2.70	1.98				
<u>CURVE 53</u>		2.03	3.06	4.32	15.7			6.42	3.13	2.89	2.23				
<u>CURVE 54</u>		2.14	3.48	4.68	14.1			6.57	3.25	3.05	2.35				
<u>CURVE 55</u>		2.22	3.93	4.91	12.6			6.75	3.50	3.25	2.60				
<u>CURVE 56</u>				5.21	10.4			6.86	3.60	3.40	2.70				
<u>CURVE 57</u>				5.57											
<u>CURVE 58</u>															
<u>CURVE 59</u>															
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<u>CURVE 93</u>															
<u>CURVE 94</u>															
<u>CURVE 95</u>															
<u>CURVE 96</u>															
<u>CURVE 97</u>															
<u>CURVE 98</u>															
<u>CURVE 99</u>															
<u>CURVE 100</u>															

\* Not shown on plot

DATA TABLE NO. 26 (continued)

T	k	T	k	T	k	T	k	T	k	T	k	T	k	T	k	T	k
<u>CURVE 62 (cont.)</u>		<u>CURVE 64 (cont.)</u>		<u>CURVE 68</u>		<u>CURVE 73</u>		<u>CURVE 79 (cont.)</u>		<u>CURVE 84 (cont.)</u>		<u>CURVE 90</u>		<u>CURVE 95</u>			
3.07	5.15	2.80	2.40	0.295	0.0619	4.37	2.22	382.0	0.318	3.01	0.290	328.2	0.356	20.4	0.476		
3.57	5.45	3.05	2.70 <sup>a</sup>	0.37	0.0032	4.64	2.33	399.5	0.318	3.15	0.325	346.2	0.351	22.4	0.465		
3.84	6.05	3.80	3.00	0.41	0.0055					3.25	0.260	383.2	0.345	26.2	0.451		
4.15	5.90	4.0	2.80	0.475	0.0060	<u>CURVE 74</u>		<u>CURVE 80</u>		3.51	0.358	393.2	0.344	85.7	0.397		
4.38	6.00			0.50	0.0075					3.62	0.368	426.2	0.339	273.0	0.372		
4.58	5.90	<u>CURVE 65</u>		0.54	0.0114	322.0	0.321	316.4	0.327	4.45	0.426	436.2	0.337	<u>CURVE 96</u>			
4.75	5.90	1.08	0.180	0.675	0.0215	340.8	0.320	341.5	0.327	4.78	0.470	465.2	0.333	<u>CURVE 96</u>			
4.81	5.60	1.10	0.205	0.675	0.0225	358.0	0.319	359.9	0.325	4.80	0.450	483.2	0.331	<u>CURVE 96</u>			
5.15	5.45	1.18	0.350	0.75	0.0310	376.3	0.317	378.7	0.324	6.35	0.596	503.2	0.326	<u>CURVE 96</u>			
5.44	5.15	1.30	0.33	0.805	0.0315	413.5	0.310	436.1	0.321	<u>CURVE 85</u>		523.2	0.324	<u>CURVE 91</u>			
5.96	4.53	1.48	0.51	0.87	0.0402	<u>CURVE 75</u>		<u>CURVE 81</u>		<u>CURVE 85</u>		523.2	0.324	<u>CURVE 91</u>			
6.59	3.78	1.68	0.70	<u>CURVE 69</u>		319.0	0.331	321.9	0.335	314.1	0.322	328.2	0.356	<u>CURVE 97</u>			
7.04	3.40	1.90	1.00	2.7	0.535	341.7	0.319	345.5	0.322	337.6	0.326	348.2	0.351	<u>CURVE 97</u>			
7.30	3.06	2.10	1.25	3.0	0.570	383.4	0.320	385.0	0.322	361.4	0.318	383.2	0.345	<u>CURVE 97</u>			
<u>CURVE 63</u>		2.35	1.70	4.0	0.670	383.2	0.312	385.0	0.318	<u>CURVE 86</u>		393.2	0.344	<u>CURVE 97</u>			
1.00	0.27	2.80	2.30	5.0	0.765	419.2	0.308	401.1	0.318	323.5	0.326	426.2	0.339	<u>CURVE 97</u>			
1.10	0.355	3.05	2.45	6.0	0.907	<u>CURVE 76</u>		<u>CURVE 82</u>		341.6	0.326	436.2	0.337	<u>CURVE 97</u>			
1.15	0.43	3.30	2.60	6.0	1.00	313.7	0.332	316.1	0.336	361.4	0.322	465.2	0.333	<u>CURVE 97</u>			
1.25	0.54	3.70	2.65	7.0	1.13	341.2	0.335	346.1	0.331	361.7	0.322	483.2	0.326	<u>CURVE 97</u>			
1.35	0.66	3.95	2.60	7.15	1.16	361.0	0.330	355.2	0.331	381.7	0.322	503.2	0.326	<u>CURVE 97</u>			
1.50	0.84	4.40	2.50	8.0	1.18	384.7	0.328	380.2	0.328	401.3	0.318	523.2	0.324	<u>CURVE 97</u>			
1.95	1.50	<u>CURVE 66</u>		2.5	0.65	<u>CURVE 77</u>		<u>CURVE 83</u>		<u>CURVE 87</u>		523.2	0.324	<u>CURVE 97</u>			
2.15	1.80	313.4	0.322	3.0	0.825	320.9	0.338	314.3	0.331	314.3	0.331	380.2	0.343	<u>CURVE 92</u>			
2.55	2.30	336.2	0.320	3.5	0.95	338.1	0.341	352.6	0.326	352.6	0.326	398.2	0.339	<u>CURVE 92</u>			
2.70	2.45	362.2	0.303	4.0	1.04	360.8	0.337	364.9	0.331	380.2	0.318	426.2	0.335	<u>CURVE 92</u>			
2.85	2.60	381.6	0.310	5.0	1.22	382.1	0.332	2.75	0.305	426.2	0.335	426.2	0.335	<u>CURVE 92</u>			
3.25	2.85	428.7	0.311	5.7	1.26	415.9	0.309	2.70	0.295	453.2	0.331	453.2	0.331	<u>CURVE 92</u>			
3.50	2.95	<u>CURVE 67</u>		6.0	1.25	<u>CURVE 78</u>		2.75	0.315	468.2	0.328	468.2	0.328	<u>CURVE 92</u>			
3.90	2.90	0.405	0.00375 <sup>b</sup>	7.0	1.18	320.9	0.338	3.19	0.345	498.2	0.324	498.2	0.324	<u>CURVE 92</u>			
4.35	2.60	0.43	0.005 <sup>b</sup>	8.0	1.08	320.9	0.338	3.44	0.431	510.2	0.322	510.2	0.322	<u>CURVE 92</u>			
<u>CURVE 64</u>		0.585	0.011 <sup>a</sup>	9.0	0.985	<u>CURVE 79</u>		3.60	0.465	<u>CURVE 88</u>		<u>CURVE 93</u>		<u>CURVE 92</u>			
1.03	0.110	0.715	0.024 <sup>b</sup>	10.0	0.91	315.5	0.322	3.90	0.505	850	0.156	840.0	0.155	<u>CURVE 93</u>			
1.18	0.160	0.74	0.027 <sup>b</sup>	11.0	0.84	335.2	0.331	4.35	0.528	900	0.157	868.2	0.163	<u>CURVE 93</u>			
1.28	0.220	0.79	0.035 <sup>a</sup>	327.2	0.347	355.4	0.322	4.85	0.575	990	0.158	890.2	0.167	<u>CURVE 93</u>			
1.45	0.300	0.815	0.038 <sup>a</sup>	327.2	0.347	378.5	0.322	8.50	0.650	1000	0.158	890.2	0.167	<u>CURVE 93</u>			
1.60	0.415	0.88	0.045	398.2	0.318	398.2	0.318	9.00	0.675	1100	0.159	892.2	0.161	<u>CURVE 93</u>			
1.90	0.68	0.92	0.050	<u>CURVE 72</u>		398.2	0.318	9.40	0.650	1200	0.160	897.7	0.166	<u>CURVE 93</u>			
2.10	1.02	0.955	0.0578	3.47	1.59	<u>CURVE 79</u>		<u>CURVE 84</u>		1250	0.160	898.7	0.171	<u>CURVE 93</u>			
2.40	1.60	1.00	0.065	319.2	0.339	319.2	0.339	2.70	0.250	<u>CURVE 89</u>		903.2	0.165	<u>CURVE 93</u>			
2.60	2.10	1.10	0.0725 <sup>a</sup>	338.8	0.331	338.8	0.331	2.70	0.263	373.2	0.356	916.7	0.162	<u>CURVE 93</u>			
		1.15	0.095	362.5	0.322	362.5	0.322	2.90	0.283	423.2	0.347	923.7	0.164	<u>CURVE 93</u>			
										473.2	0.339	931.2	0.180	<u>CURVE 93</u>			
												937.2	0.168	<u>CURVE 93</u>			

<sup>a</sup> Not shown on plot

Table 1012

DATA TABLE NO. 26 (continued)

T	k	T	k	H(gauss)	k	H(gauss)	k	H(gauss)	k	H(gauss)	k	H(gauss)	k	T	k
<b>CURVE 97 (cont.)</b>															
941.2	0.169	317.2	0.324	625	0.86	0	0.835	0	0.540	338	0.560	20.36	0.509	<b>CURVE 115 (cont.)</b>	
941.2	0.179	353.1	0.324	605	0.80	120	0.836	145	0.550	578	0.560	21.03	0.330		
955.7	0.180	376.2	0.320	566	0.76	155	0.868	165	0.550	613	0.568	<b>CURVE 116</b>			
966.2	0.180			529	0.70	183	0.869	182	0.850	633	0.600				
971.7	0.183	<b>CURVE 102</b>				485	0.70	205	0.882	652	0.698	2.82	0.545		
980.0	0.173	273.2	0.350	410	0.68	220	0.869	225	0.910	686	0.849	3.32	0.618		
983.2	0.179	373.2	0.320	0	0.61	277	0.974	259	0.951	728	0.849	3.81	0.653		
986.7	0.186	<b>CURVE 105</b>				292	0.878	265	0.981	925	0.851	4.34	0.697		
998.2	0.182			309	0.880	305	1.070	305	1.070	1000	0.852	4.58	0.733		
1000.2	0.178	<b>CURVE 106</b>				312	0.922	345	1.189	500	0.490	5.42	0.844		
1002.7	0.189			328	0.944	360	1.210	360	1.220	539	0.535	5.82	0.905		
1020.2	0.188	<b>CURVE 103*</b>				340	1.062	380	1.220	580	0.602	6.50	0.990		
1023.2	0.182	0	0.62	312	1.40	340	1.81	417	1.261	614	0.675	6.72	1.065		
1045.7	0.192	225	0.63	362	1.82	342	1.275	539	1.220	653	0.788	7.21	1.160		
1059.2	0.200	376	0.63	376	1.83	402	1.283	1000	1.230	692	0.802	<b>CURVE 117</b>			
1061.2	0.196	225	0.65	389	1.84	462	1.285	729	0.803	688	0.856	2.67	0.308		
1063.2	0.202	340	0.64	400	1.89	711	1.275	1000	0.805	660	0.855	3.03	0.411		
1066.2	0.194	415	0.65	415	2.00	1000	1.265	996	1.240	645	0.785	3.03	0.425		
1069.2	0.189	437	0.65	437	2.45	<b>CURVE 110</b>				600	0.560	3.42	0.500		
1099.2	0.198	441	0.65	441	2.68	<b>CURVE 112</b>				600	0.560	3.92	0.583		
1099.2	0.212	464	0.65	464	2.80	996	1.230	725	1.230	645	0.855	<b>CURVE 118</b>			
1105.2	0.197	464	0.65	464	3.60	610	1.235	610	1.235	1000	0.852	7.215	5.043		
1130.2	0.209	464	0.65	464	4.30	490	1.230	490	1.230	928	0.806	7.226	5.009		
<b>CURVE 98*</b>															
293.1	0.351	630	0.70	494	5.15	809	1.265	450	0.812	800	0.812	7.297	4.889		
303.6	0.350	650	0.72	724	5.50	495	1.260	430	0.808	725	0.808	7.387	4.756		
315.0	0.346	630	0.73	1000	5.50	461	1.260	410	1.232	685	0.808	7.455	4.661		
327.2	0.343	<b>CURVE 100</b>				438	1.259	390	1.235	642	0.508	7.491	4.609		
346.5	0.344	33	1.90	375	1.247	420	1.252	378	1.235	602	0.698	7.244	4.983		
<b>CURVE 99*</b>															
318.9	0.324	180	1.90	355	1.240	320	1.247	320	1.148	530	0.558	7.270	4.976		
349.7	0.331	262	1.91	320	0.980	355	1.240	300	1.112	490	0.506	7.283	4.910		
366.8	0.318	336	2.00	309	0.937	265	1.020	226	0.950	412	0.440	7.297	4.889		
410.8	0.312	364	2.05	285	0.910	226	0.922	205	0.910	370	0.420	7.387	4.756		
<b>CURVE 100*</b>															
327.5	0.324	382	2.10	262	0.894	190	0.896	300	0.408	725	0.408	7.455	4.661		
355.4	0.317	406	4.49	225	0.884	164	0.885	262	0.406	538	1.260	7.505	4.595		
378.7	0.317	428	4.55	45	0.873	150	0.876	225	0.404	580	1.250	7.629	4.431		
404.7	0.313	454	4.55	113	0.865	113	0.865	0	0.403	609	1.240	7.662	4.403		
<b>CURVE 101*</b>															
327.5	0.324	798	4.54	463	4.65	40	0.860	7.35	1.150	671	1.205	7.765	4.277		
355.4	0.317	750	4.55	466	5.35	40	0.855	7.35	1.150	7.35	1.150	7.797	4.239		
378.7	0.317	722	4.54	490	5.55	7.63	1.137	7.63	1.137	7.63	1.137	7.835	4.206		
404.7	0.313	683	2.02	524	5.55	8.11	1.073	8.11	1.073	8.11	1.073	7.964	4.057		
<b>CURVE 102*</b>															
327.5	0.324	665	1.35	524	5.55	9.54	0.931	9.54	0.931	10.60	0.870	7.980	4.043		
355.4	0.317	665	1.35	610	5.55	11.44	0.809	11.44	0.809	11.44	0.809	8.113	3.905		
378.7	0.317	610	5.55	610	5.55	14.96	0.637	14.96	0.637	14.96	0.637	8.270	3.754		

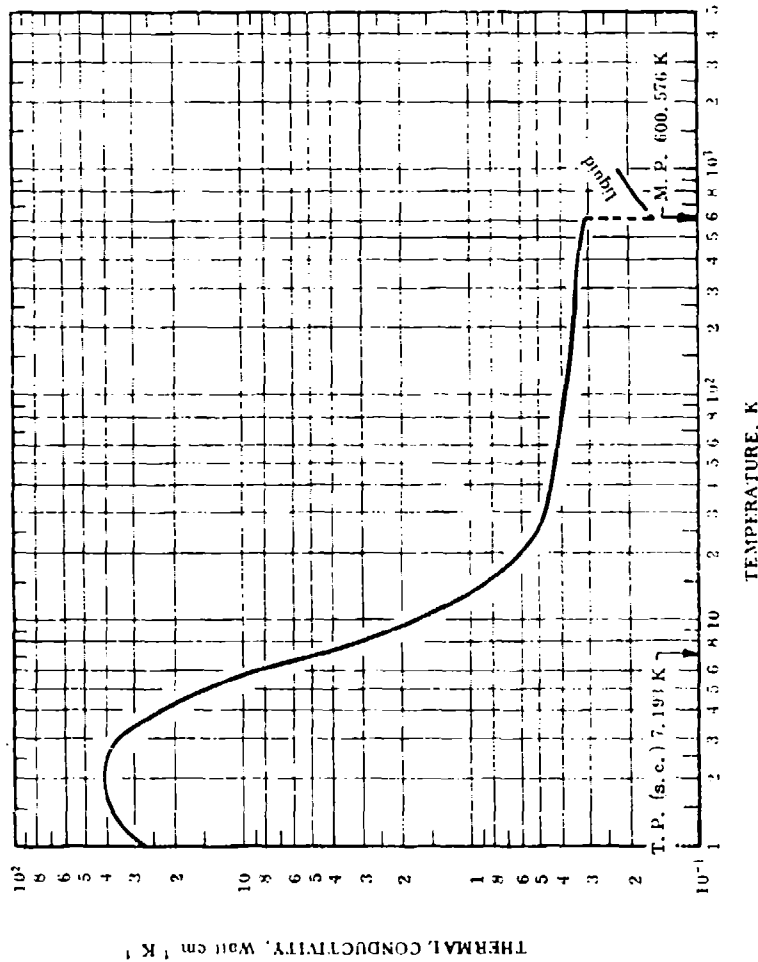
\* Not shown or plot

DATA TABLE NO. 26 (continued)

CURVE 119 <sup>f</sup>			CURVE 121 (cont.) <sup>f</sup>			CURVE 129 (cont.) <sup>f</sup>			CURVE 132			CURVE 141		
T	k		T	k		T	k		T	k		T	k	
6.409	6.790	6.756	6.044	0.356	0.0168 <sup>h</sup>	0.057	0.42	0.462	223	0.360	323.2	0.350		
6.519	6.506	6.813	5.873	0.356	0.0175	0.0455	CURVE 131			273	0.353	340.2	0.345	
6.683	6.143	7.27	4.937	0.360	0.020	0.0575	CURVE 131			323	0.346	362.2	0.339	
6.932	5.617	317.6	0.368	0.0205	0.067	0.068	0.14	0.136 <sup>h</sup>	373	0.339	380.2	0.338		
7.009	5.451	323.9	0.368	0.025	0.068	0.023	0.058	0.14	423	0.332	389.2	0.343		
7.110	5.238	333.2	0.372	0.024	0.057	0.024	0.057	0.285	473	0.324	395.2	0.335		
7.164	5.159	CURVE 122 <sup>f</sup>			0.024	0.057	0.44	1.44	523	0.317	402.2	0.343		
7.219	5.055	5.484	2.524	0.025	0.0705	CURVE 131			573	0.310	414.2	0.335		
7.351	4.811	5.407	2.542	0.025	0.075	CURVE 131			323.2	0.343	434.2	0.334		
7.412	4.721	5.026	2.585	0.028	0.0725	CURVE 139 <sup>f</sup>			334.8	0.3581	CURVE 142			
7.591	4.491	315.7	0.356	0.029	0.086	CURVE 135			336.7	0.3590	337.6	0.331 <sup>h</sup>		
7.898	4.143	329.8	0.356	0.031	0.083	CURVE 135			367.3	0.3562	380.2	0.321		
8.044	3.982	616.5	0.175	0.034	0.086	0.034	0.086	0.034	370.7	0.3541	398.9	0.314		
8.208	3.813	644.3	0.180	0.0425	0.110	0.046	0.142	0.054	412.2	0.3514				
8.290	3.733	699.8	0.189	0.046	0.142	0.054	0.155	0.061	416.0	0.3501				
CURVE 120 <sup>f</sup>			755.4	0.196	0.054	0.155	0.061	0.195	417.9	0.3522				
6.279	7.063	6.130	2.839	0.175	0.066	0.200	0.066	0.161	436.5	0.3503				
6.371	6.909	6.219	2.962	0.180	0.073	0.215	0.073	0.159	442.3	0.3500				
6.504	6.418	6.272	2.974	0.189	0.083	0.215	0.083	0.157	468.9	0.3443				
6.623	6.299	6.272	2.974	0.189	0.093	0.285	0.093	0.156 <sup>h</sup>	601.5	0.3390				
6.872	5.752	6.336	3.140	0.196	0.103	0.320	0.103	0.156 <sup>h</sup>	CURVE 140 <sup>f</sup>					
6.955	5.570	6.376	3.211	0.196	0.110	0.350	0.110	0.155	316.2	0.341				
7.195	5.096	6.414	3.246	0.196	0.125	0.385	0.125	0.155	335.2	0.341				
7.225	5.024	6.491	3.424	0.196	0.140	0.435	0.140	0.155	336.2	0.341				
7.250	4.989	6.583	3.607	0.196	0.165	0.50	0.165	0.155	339.2	0.339				
7.292	4.901	6.670	3.803	0.196	0.185	0.565	0.185	0.155	349.2	0.340				
7.329	4.839	6.689	3.886	0.196	0.215	0.65	0.215	0.155	347.2	0.341				
7.518	4.574	6.816	4.187	0.196	0.23	0.71	0.23	0.155	349.2	0.339				
7.561	4.515	6.880	4.263	0.196	CURVE 130			357.2	0.334					
7.695	4.354	6.932	4.348	0.196	0.105	0.0490 <sup>h</sup>	0.105	0.1925	358.2	0.332				
7.695	4.354	6.955	4.450	0.196	0.155	0.0691 <sup>h</sup>	0.155	0.1925	372.2	0.333				
7.967	4.159	6.985	4.474	0.196	0.26	0.0864 <sup>h</sup>	0.26	0.1925	377.2	0.329				
7.967	4.159	7.005	4.474	0.196	0.41	0.170	0.41	0.1925	381.2	0.330				
8.058	3.970	7.051	4.588	0.196	CURVE 131			389.2	0.328					
8.150	3.865	7.066	4.628	0.196	0.198	0.150 <sup>h</sup>	0.198	0.150 <sup>h</sup>	391.2	0.332				
8.157	3.863	7.117	4.697	0.196	0.240	0.259 <sup>h</sup>	0.240	0.259 <sup>h</sup>	401.2	0.325				
8.167	3.851	7.118	4.761	0.196	0.375	0.295	0.375	0.295	420.2	0.321				
8.310	3.721	7.161	4.836	0.196	0.015	0.0415	0.015	0.0415	420.2	0.321				
CURVE 121 <sup>f</sup>			7.198	4.919	0.0155	0.032	0.0155	0.032	573	0.315				
6.225	7.264	7.190	4.960	0.196	CURVE 120 <sup>f</sup>			CURVE 131						
6.273	7.134	7.215	4.964	0.196	0.146	0.0000182	0.146	0.0000182	223	0.362				
6.332	6.968	7.211	4.988	0.196	0.219	0.0000229	0.219	0.0000229	273	0.355				
			7.211	4.988	0.309	0.0000410	0.309	0.0000410	323	0.349				
			7.211	4.988	0.378	0.0000583	0.378	0.0000583	373	0.342				
			7.211	4.988	0.415	0.0000583	0.415	0.0000583	423	0.335				
			7.211	4.988	0.455	0.0000583	0.455	0.0000583	473	0.329				
			7.211	4.988	0.495	0.0000583	0.495	0.0000583	523	0.321				
			7.211	4.988	0.535	0.0000583	0.535	0.0000583	573	0.315				

Not shown on plot

FIGURE AND TABLE NO. 26R RECOMMENDED THERMAL CONDUCTIVITY OF LEAD



REMARKS

The recommended values are for well-annealed 99.99% pure lead with residual electrical resistivity  $\rho_0 = 0.000880 \mu\Omega$  cm (characterization by  $\rho_0$  becomes important at temperatures below about 90 K). The values below 1.5 T<sub>m</sub> are calculated to fit the experimental data by using  $n = 3.00$ ,  $\alpha^1 = 7.40 \times 10^{-4}$ , and  $\beta = 0.0333$ . The recommended values that are supported by experimental thermal conductivity data are thought to be accurate to within 3% of the true values near room temperature, and 3 to 10% at other temperatures.

RECOMMENDED VALUES<sup>1</sup>

T <sub>1</sub>	k <sub>1</sub>	k <sub>2</sub>	T <sub>2</sub>	T <sub>1</sub>	k <sub>1</sub>	k <sub>2</sub>	T <sub>2</sub>
0	0	0	-459.7	600.6	0.155 <sup>†</sup>	(8.96)	621.4
1	27.7	1600	-457.9	700	0.174	10.1	800.3
2	42.4	2450	-456.1	800	0.190	11.0	930.3
3	34.0	1960	-454.3	900	0.203	11.7	1150
4	22.4	1290	-452.5	1000	0.215	12.4	1340
5	13.8	750	-450.7				
6	8.2	474	-448.9				
7	4.9	283	-447.1				
8	3.2	185	-445.3				
9	2.3	133	-443.5				
10	1.78	103	-441.7				
11	1.46	84.4	-439.9				
12	1.23	71.1	-438.1				
13	1.07	61.8	-436.3				
14	0.94	54.3	-434.5				
15	0.84	48.5	-432.7				
16	0.77	44.5	-430.9				
18	0.66	38.1	-427.3				
20	0.59	34.1	-423.7				
25	0.507	29.5	-414.7				
30	0.477	27.0	-405.7				
35	0.462	26.7	-396.7				
40	0.451	26.1	-387.7				
45	0.442	25.5	-378.7				
50	0.435	25.1	-369.7				
60	0.424	24.5	-351.7				
70	0.415	24.0	-333.7				
90	0.407	23.5	-315.7				
90	0.401	23.2	-297.7				
100	0.396	22.9	-279.7				
150	0.377	21.8	-189.7				
200	0.366	21.1	-99.7				
250	0.358	20.7	-9.7				
273.2	0.355	20.5	32.0				
300	0.352	20.3	80.3				
350	0.348	20.1	170.3				
400	0.338	19.5	260.3				
500	0.325	18.8	440.3				
600	0.312	18.0	620.3				
600.5	0.312	18.0	621.2				

<sup>1</sup>T<sub>1</sub> in K, k<sub>1</sub> in Watt cm<sup>-1</sup> K<sup>-1</sup>, T<sub>2</sub> in F, and k<sub>2</sub> in Btu hr<sup>-1</sup> ft<sup>-1</sup> F<sup>-1</sup>. † Values in parentheses are extrapolated.



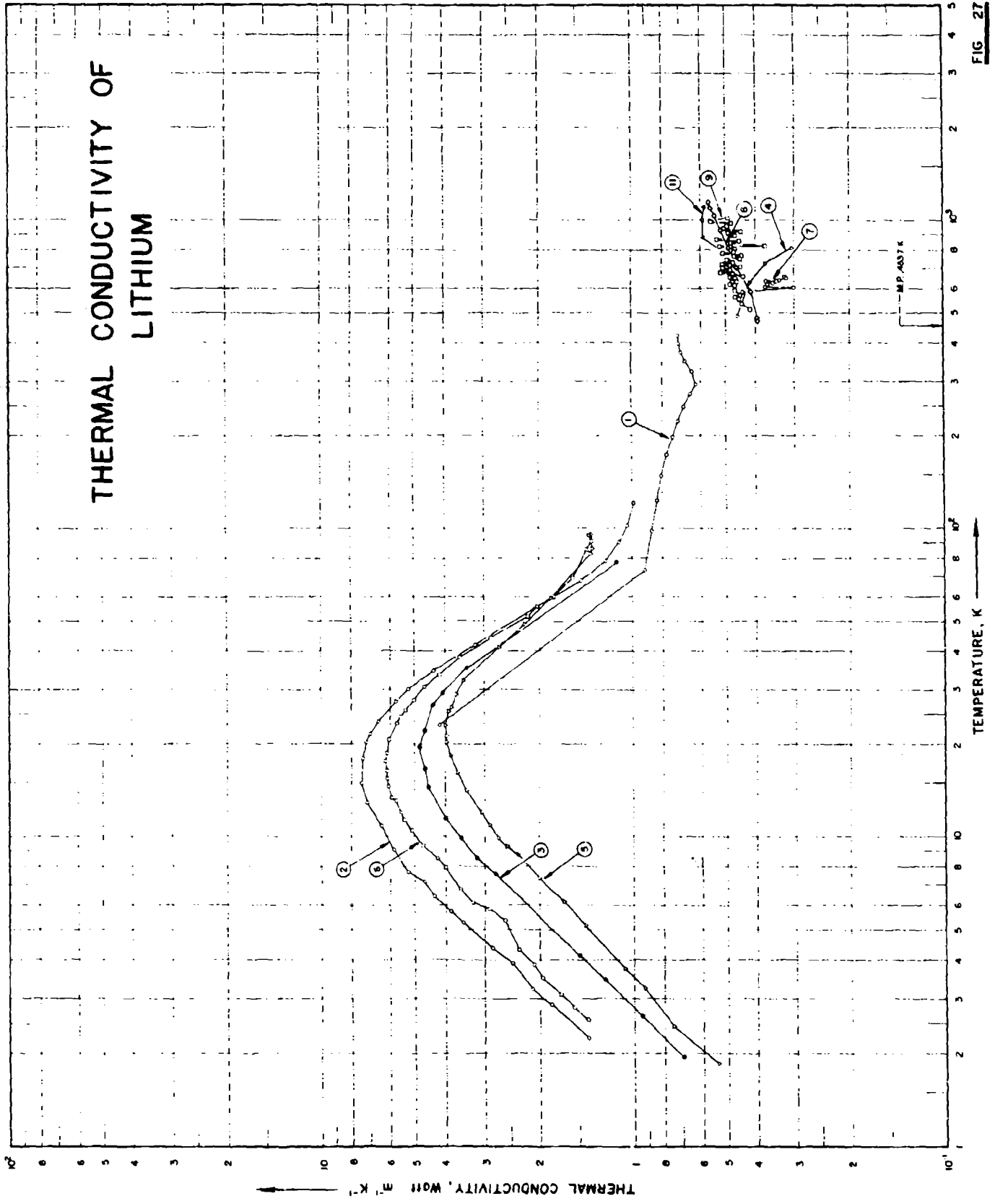


FIG. 27

SPECIFICATION TABLE NO. 5. THERMAL CONDUCTIVITY OF LITHIUM

See Data Reported in Figure and Table No. 5.

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K.	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
1	15	F	1954	23-422			Pure; 1.19 cm dia and 25 cm long; extruded; electrical conductivity reported as 1059, 115, 57, 36, 1, 26, 9, 21, 1, 17, 5, 1, 10, 12, 45, 11, 55, 10, 35, 9, 44, 8, 70, 4, 13, 2, 37, 20, 1, 14 x 10 <sup>6</sup> ohm <sup>-1</sup> cm <sup>-1</sup> at 21, 23, 98, 123, 148, 173, 198, 223, 244, 273, 293, 323, 348, 373, 398, and 423 K, respectively.
2	32	L	1966	2.2-121		LI 2	High purity; 1.4 mm dia, supplied by A. D. Mac Kay (New York); extruded; electrical resistivity ratio $\rho(295 \text{ K})/\rho(0 \text{ K}) = 254$ (using Guntz and Broniewski's value $\rho(295 \text{ K}) = 9.44 \mu\text{ ohm cm}$ ).
3	37	L	1964	2.0-74		LI 2	High purity; 1.4 mm dia; supplied by New Metals and Chemical Ltd (London); $\rho(295 \text{ K})/\rho(0 \text{ K}) = 157$ (using Guntz and Broniewski's value $\rho(295 \text{ K}) = 9.44 \mu\text{ ohm cm}$ ).
4	142	C	1965	4.9-812	10		99.8 Li, 0.042 Cl, 0.03 N, 0.09 heavy metals, 0.02 Na, 0.06 Ca, 0.03 Fe and Al, 0.015 Sr; specimen in liquid state; supplied by Maywood Chemical Works; 1.999 in. dia; Armerium used as comparative material.
5	223	L	1956	1.9-96		LI 1	High purity; possibly contaminated with Cu; 0.83 mm dia; distilled; electrical resistivity $1.9 \mu\text{ ohm cm}$ at room temperature and $0.084 \mu\text{ ohm cm}$ at 0 K; electrical resistivity ratio (room temp)/ $\rho(0 \text{ K}) = 190$ .
6	223	L	1956	2.6-95		LI 2	Similar to the above specimen but no Cu contamination; electrical resistivity $9.17 \mu\text{ ohm cm}$ at room temperature and $0.1475 \mu\text{ ohm cm}$ at 0 K; $\rho(\text{room temp})/\rho(0 \text{ K}) = 200$ .
7	243	L	1960	5.0-652	10		Composition before test: 99.52 <sup>1</sup> Li (by difference), 0.1 Ca, 0.1 Si, <0.1 Hg, <0.1 P, 0.01 Al, 0.01 B, 0.01 Cr, 0.01 Cu, 0.01 Fe, 0.01 K, <0.01 Na, and 0.01 Ni; after test: 99.22 <sup>1</sup> Li (by difference), 0.1 Ca, <0.1 Hg, <0.1 P, 0.1 Al, 0.2 B, 0.02 Cr, 0.04 Cu, 0.01 K, 0.1 Na and <0.01 Ni; specimen in liquid state; 0.684 in. dia; supplied by Maywood Chemical Company; M.P. 146 C.
8	565, 462	-	1961	467-1128			(Nominal specification): 99.9 Li (min), 0.03 Na, 0.01 K, 0.01 Ca, 0.002 Cl, 0.01 N, 0.005 Fe, 0.01 Ni, and 0.002 Cr; pretest 0.04 Na, 0.23 N, 0.012 Fe, 0.004 Ni, and 0.013 Cr; specimen in liquid state; 0.59 in. dia; electrical resistivity reported as 23.6, 25.8, 29.7, 33.1, 35.9, 38.5, 41.2, 42.5, and 43.8 $\mu\text{ ohm cm}$ at 193.6, 199.2, 209.4, 422.2, 505.0, 614.9, 753.6, 808.3, and 864.4 C, respectively; data calculated from Wickraman Franz-Lorenz relationship using $L = 2.16 \times 10^4 \text{ } ^\circ\text{K}^2$ , this value being based on unpublished thermal conductivity values of 0.416, 0.428, and 0.496 at 283.3, 262.8, and 522.8 C that had been obtained from C. T. Ewing of the Naval Research Laboratory, Washington.
9	919, 592	L	1958	511-1012			M.P. 146 C; data in liquid state; measured in vacuum ( $\sim 4 \times 10^{-4}$ mm Hg).
10	744, 763, 671	C	1964	596-1052	± 4 - 15		Composition before test: 99.82 Li (by difference), 0.015 Na, 0.06 K, 0.0001 Ca, 0.0005 Al, 0.001 Si, 0.04 Cl, 0.02 N, <0.0015 Ni, <0.0012 Cr, <0.0010 Ti, 0.0062 N, 0.0003 O, 0.0027 Fe, and 0.025 others; after test: values assumed to remain the same except 0.0058 Si, 0.0022 Ni, 0.001 Cr, 0.002 Mn, and 0.0024 Fe; 99.81 Li (by difference), specimen in liquid state; supplied by the Foote Mineral Company; measured in vacuum ( $3 \times 10^{-5}$ mm Hg); type 317 stainless steel used as comparative material; data calculated by comparing to the top reference material (between heater and specimen).

SPECIFICATION TABLE NO. 27 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)	Specifications and Remarks
11	764, 765, 766	C	1965	631-1100	± 0.15			Data of the above specimen calculated by comparing with the bottom reference material (Armco iron, between specimen and heat sink).  Density reported as 0.4992, 0.4989, 0.4724, 0.4654, 0.4581, 0.4440, and 0.4289 g cm <sup>-3</sup> at 615, 7, 714, 3, 862, 7, 910, 4, 1054, 1157 and 1311 K, respectively; electrical resistivity reported as 12.16, 13.16, 14.73, 15.54, 26.54, 28.30, 30.39, 31.02, 32.10, 33.40, 34.90, 35.99, 37.53, 38.61, 39.97, 41.51, 42.68, 46.05, 48.21, 48.96, and 49.67 μ ohm cm at 360, 394, 432, 451, 487, 536, 597, 604, 655, 696, 764, 815, 878, 918, 943, 1045, 1104, 1246, 1319, 1342, and 1372 K, respectively; thermal conductivity data calculated from measured electrical resistivity values and the Lorenz number $2.45 \times 10^{-8}$ V <sup>2</sup> K <sup>-2</sup> .
12	959	-	1965	549-1723	-			0.0490 O, 0.0130 Na, 0.0010 Fe, 0.0030 Ni, < 0.0010 N, < 0.0005 Nb, and < 0.0005 Zr; positive impurities 0.24 O, 0.01 Fe, 0.01 Nb, < 0.0010 N, < 0.0010 Ni, < 0.0010 Zr, and 0.0002 C; molten specimen contained in a Ni-17r alloy capsule; electrical resistivity reported as 27.2, 29.9, 32.7, 35.4, 38.1, 40.8, 43.5, 46.1, 48.6, 51.2, 53.7, 56.1, and 59.1 μ ohm cm at 200, 300, 400, 500, 600, 700, 800, 900, 1000, 1100, 1200, 1300, 1400, and 1430 C, respectively; data calculated from the measured electrical resistivity data and the Lorenz number $2.29 \times 10^{-8}$ V <sup>2</sup> K <sup>-2</sup> , this value being based on measured thermal conductivity data of Cooke, J. W. (J. Chem. Phys., 40 (7), 1962-9, 1964).

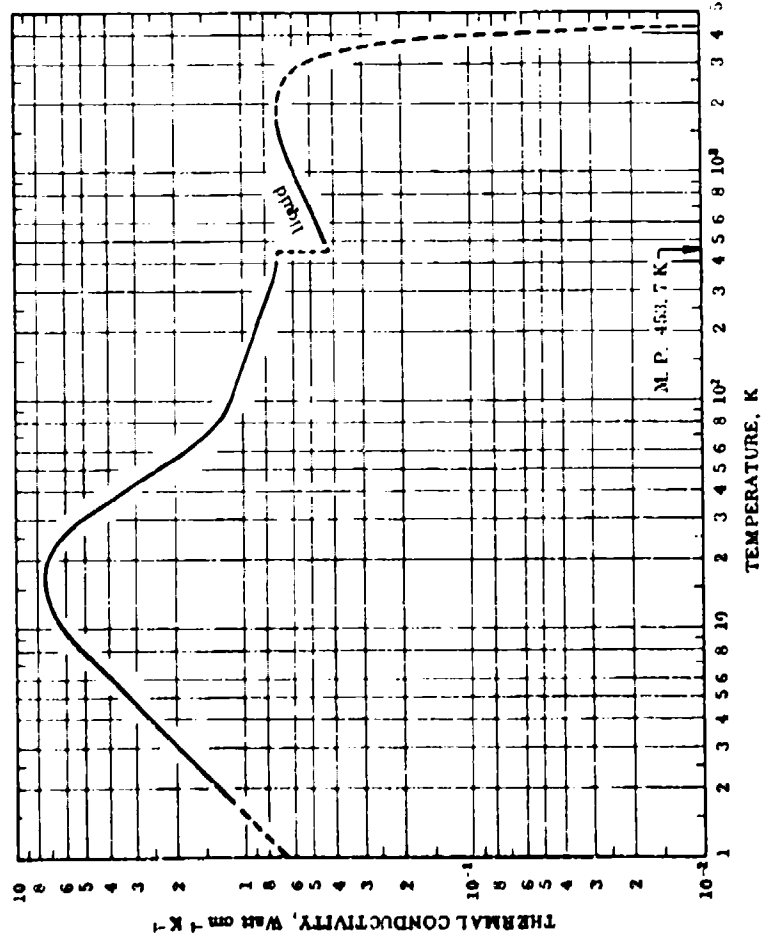


DATA TABLE NO. 27 (continued)

$\tau$	$k$
	<u>CURVE 13</u>
568.9	0.445*
673.2	0.471*
773.2	0.502*
873.2	0.525*
973.2	0.547*
1073	0.566
1173	0.583
1273	0.599
1373	0.614
1473	0.628
1573	0.643
1723	0.661

\*Not shown on plot

FIGURE AND TABLE NO. 27R RECOMMENDED THERMAL CONDUCTIVITY OF LITHIUM



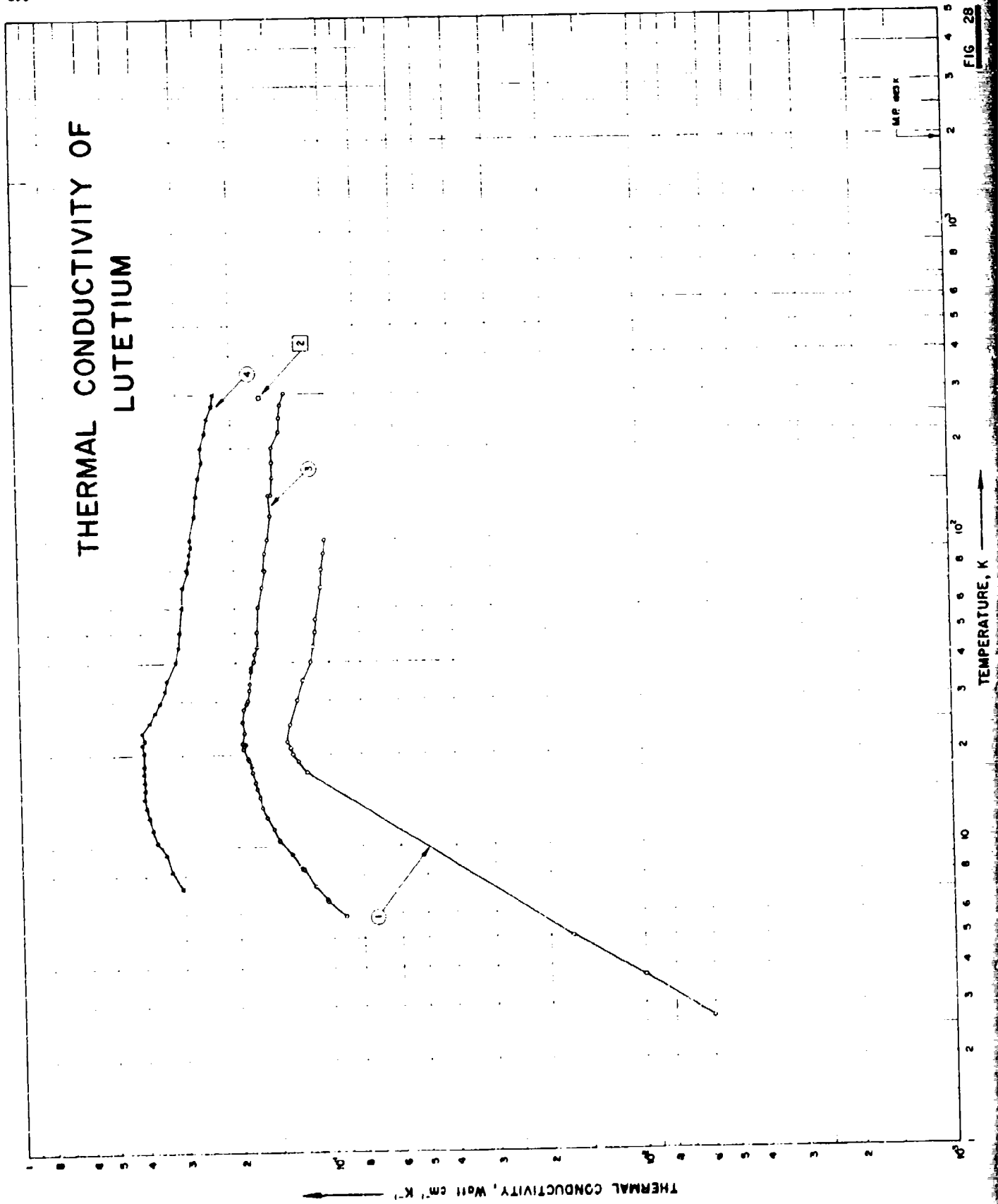
RECOMMENDED VALUES\*

T <sub>1</sub>	k <sub>1</sub>	k <sub>2</sub>	T <sub>2</sub>	T <sub>1</sub>	k <sub>1</sub>	k <sub>2</sub>	T <sub>2</sub>
0	0	0	-459.7	453.7	(0.428)	(24.7)	357.0
1	(0.658)†	(38.0)	-457.9	500	0.443	25.6	440.2
2	1.32	76.3	-456.1	600	0.475	27.5	620.3
3	1.97	114	-454.3	700	0.509	29.4	900.3
4	2.62	151	-452.5	800	0.541	31.3	960.3
5	3.29	190	-450.7	900	0.572	33.1	1160
6	3.86	223	-448.9	1000	0.600	34.7	1340
7	4.36	263	-447.1	1100	0.623	36.1	1520
8	4.83	298	-445.3	1200	0.647	37.4	1700
9	5.27	328	-443.5	1300	0.665	38.4	1880
10	5.73	354	-441.7	1400	0.680	39.3	2060
11	6.13	376	-439.9	1500	0.691	39.9	2240
12	6.52	394	-438.1	1600	0.699	40.4	2420
13	6.93	410	-436.3	1700	0.704	40.7	2600
14	7.25	419	-434.5	1800	(0.707)	(40.9)	2780
15	7.58	426	-432.7	1900	(0.707)	(40.9)	2960
16	7.90	428	-430.9	2000	(0.705)	(40.7)	3140
17	8.21	427	-429.1	2200	(0.696)	(40.2)	3500
18	8.51	427	-427.3	2400	(0.676)	(40.2)	3860
19	8.80	427	-425.5	2600	(0.645)	(37.3)	4220
20	9.08	416	-423.7	2800	(0.602)	(34.7)	4580
21	9.35	364	-414.7	3000	(0.543)	(31.4)	4940
22	9.60	300	-405.7	3200	(0.467)	(27.0)	5300
23	9.83	244	-396.7	3400	(0.383)	(22.1)	5660
24	10.05	198	-387.7	3600	(0.293)	(16.9)	6020
25	10.25	162	-378.7	3800	(0.193)	(11.2)	6380
26	10.43	136	-369.7	4000	(0.086)	(4.97)	6740
27	10.59	103	-361.7	4150	(0.002)	(0.12)	7010
28	10.73	86.7	-353.7				
29	10.85	76.3	-345.7				
30	10.96	69.9	-337.7				
31	11.05	65.9	-329.7				
32	11.13	63.9	-321.7				
33	11.20	62.5	-313.7				
34	11.25	61.7	-305.7				
35	11.29	61.3	-297.7				
36	11.32	61.0	-289.7				
37	11.34	60.7	-281.7				
38	11.35	60.5	-273.7				
39	11.36	60.3	-265.7				
40	11.36	60.1	-257.7				
41	11.35	59.9	-249.7				
42	11.33	59.7	-241.7				
43	11.30	59.5	-233.7				
44	11.26	59.2	-225.7				
45	11.21	58.9	-217.7				
46	11.15	58.6	-209.7				
47	11.08	58.3	-201.7				
48	11.00	58.0	-193.7				
49	10.91	57.7	-185.7				
50	10.81	57.4	-177.7				
51	10.70	57.1	-169.7				
52	10.58	56.8	-161.7				
53	10.45	56.5	-153.7				
54	10.31	56.2	-145.7				
55	10.17	55.9	-137.7				
56	10.02	55.6	-129.7				
57	9.87	55.3	-121.7				
58	9.71	55.0	-113.7				
59	9.54	54.7	-105.7				
60	9.37	54.4	-97.7				
61	9.19	54.1	-89.7				
62	9.01	53.8	-81.7				
63	8.82	53.5	-73.7				
64	8.63	53.2	-65.7				
65	8.43	52.9	-57.7				
66	8.23	52.6	-49.7				
67	8.03	52.3	-41.7				
68	7.82	52.0	-33.7				
69	7.61	51.7	-25.7				
70	7.39	51.4	-17.7				
71	7.17	51.1	-9.7				
72	6.95	50.8	-1.7				
73	6.72	50.5	6.3				
74	6.49	50.2	14.3				
75	6.25	50.0	22.3				
76	6.01	49.8	30.3				
77	5.76	49.6	38.3				
78	5.51	49.4	46.3				
79	5.25	49.2	54.3				
80	5.00	49.0	62.3				
81	4.73	48.8	70.3				
82	4.46	48.6	78.3				
83	4.19	48.4	86.3				
84	3.91	48.2	94.3				
85	3.63	48.0	102.3				
86	3.35	47.8	110.3				
87	3.07	47.6	118.3				
88	2.78	47.4	126.3				
89	2.49	47.2	134.3				
90	2.20	47.0	142.3				
91	1.91	46.8	150.3				
92	1.62	46.6	158.3				
93	1.33	46.4	166.3				
94	1.04	46.2	174.3				
95	0.75	46.0	182.3				
96	0.46	45.8	190.3				
97	0.17	45.6	198.3				
98		45.4	206.3				
99		45.2	214.3				
100		45.0	222.3				

REMARKS

The recommended values are for high-purity lithium with residual electrical resistivity  $\rho_0 = 0.0371 \mu\Omega \text{ cm}$  (characterization by  $\rho_0$  becomes important at temperatures below about 200 K). The values below 1.5 Tm are calculated to fit the experimental data by using  $n = 2.00$ ,  $\alpha' = 1.57 \times 10^{-4}$ , and  $\beta = 1.52$ . The recommended values that are supported by experimental thermal conductivity data are thought to be accurate to within 5% of the true values near room temperature and 5 to 10% at other temperatures.

\* T<sub>1</sub> in K, k<sub>1</sub> in Watt cm<sup>-1</sup>K<sup>-1</sup>, T<sub>2</sub> in F, and k<sub>2</sub> in Btu hr<sup>-1</sup>ft<sup>-1</sup>. † Values in parentheses are extrapolated or estimated.



TEMPERATURE, K

FIG 28

SPECIFICATION TABLE NO. 28 THERMAL CONDUCTIVITY OF LUTETIUM

(Impurity &lt; 0.20% each; total impurities &lt; 0.50%)

[For Data Reported in Figure and Table No. 28.]

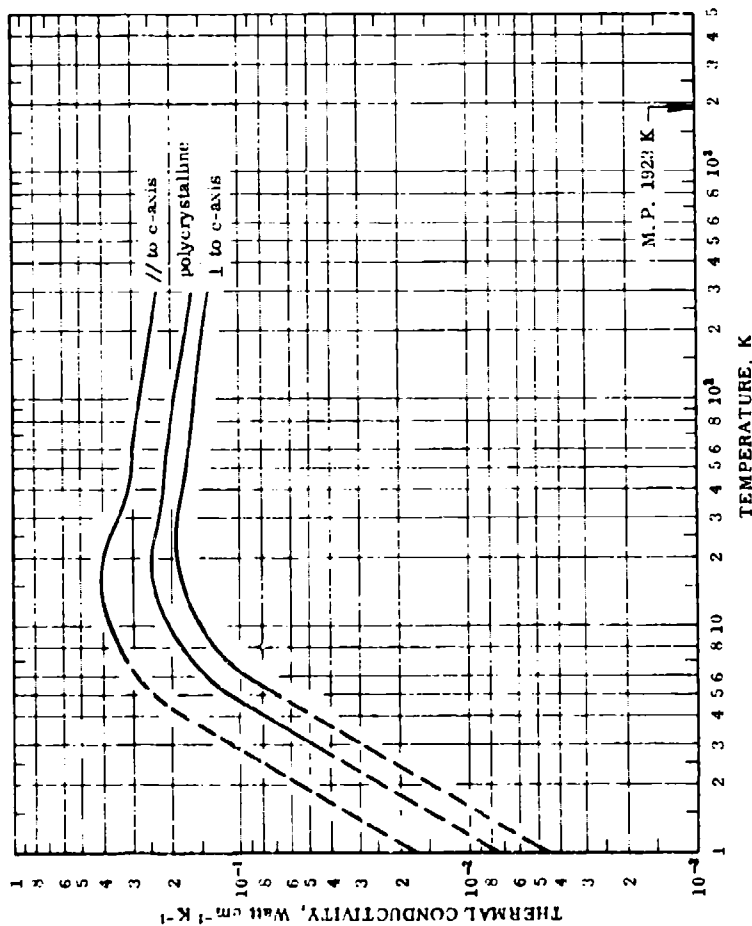
Curve No.	Ref. No.	Method Used*	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)	Specifications and Remarks
1	994, 820	L	1965	2.7 5-100				99.99 pure; polycrystalline; strip specimen 0.25 mm thick; annealed in stream of helium vapor at 600 C for 3 hrs; electrical resistivity reported at 4.2 and 293 K, respectively as 12.3 and 79 $\mu\text{ohm cm}$ ; data from: smoothed curve; Lorenz function reported as $3.40 \times 10^{-8} \text{ V}^2 \text{ K}^{-2}$ in the residual resistance region.
2	256	C	1966	291	$\pm 4$			< 0.1 rare earth metal; ~0.5 Ta, ~0.05 base metals; polycrystalline specimen $1.2 \times 1.2 \times 0.31 \text{ cm}$ ; electrical resistivity 59 $\mu\text{ohm cm}$ at 291 K; data proposed by the author from measurements of 2 different thermal comparators.
3	†	L	1967	5.5-300	5-6			0.0600 Yb, < 0.0200 Ta, 0.012% O, < 0.0100 Er, < 0.0100 Y, < 0.0030 Fe, 0.0025 N, < 0.0020 Ca, < 0.0010 each of Al, Cr, Cu, Mg, Ni, Si, and Tm, and < 0.0005 Se; single crystal; $5.1 \times 1.53 \times 1.40 \text{ cm}$ ; grown from arc-melted buttons using the strain anneal method; $\langle 1010 \rangle$ direction (b-axis) along the specimen axis; electrical resistivity reported as 2.650, 2.650, 2.652, 2.652, 2.663, 2.703, 2.924, 3.313, 4.103, 5.818, 9.089, 12.981, 18.884, 22.779, 28.478, 36.422, 44.358, 52.232, 62.593, and 76.517 $\mu\text{ohm cm}$ at 1.3, 2.4, 3.3, 4.2, 7.0, 9.9, 14.6, 18.8, 24.0, 32.4, 44.4, 57.5, 77.5, 90.5, 111.0, 159.7, 169.6, 190.9, 240.3, and 297.5 K, respectively; electrical resistivity ratio $\rho_{300\text{K}}/\rho_{4.2\text{K}} = 28.9$ ; residual electrical resistivity 2.65 $\mu\text{ohm cm}$ ; Lorenz function reported as $4.31 \times 10^{-8} \text{ V}^2 \text{ K}^{-2}$ , 3.44, 3.70, 3.89, 4.00, 4.00, 3.93, 3.77, 3.60, and $3.52 \times 10^{-8} \text{ V}^2 \text{ K}^{-2}$ at 5.5, 14.7, 21.4, 30.1, 45.4, 67.4, 97.8, 114.6, 150.2, 207.3, 267.1, and 300.0 K, respectively; heat flow along b-axis.
4	†	L	1967	7.3-299				0.0200 Tm, < 0.0200 Ta, < 0.0200 Ti, < 0.0100 W, < 0.0100 Y, 0.0099 O, < 0.0030 Fe, < 0.0010 each of Al, Ca, Cr, Cu, Er, Mg, Ni, and Se, 0.0008 N, < 0.0005 Sc, and < 0.0005 Yb; single crystal; $17.96 \times 2.18 \times 2.14 \text{ mm}$ ; grown from arc-melted buttons using the strain anneal method; < 0001 > direction along the specimen axis; electrical resistivity reported as 0.759, 0.759, 0.761, 0.761, 0.770, 0.798, 0.852, 0.941, 1.069, 1.174, 1.842, 2.861, 4.709, 6.253, 8.630, 11.811, 15.178, 18.849, 21.581, 26.250, and 34.793 $\mu\text{ohm cm}$ at 1.2, 2.4, 3.3, 4.2, 7.0, 10.0, 13.0, 16.0, 19.0, 24.0, 30.0, 40.6, 56.1, 70.3, 92.6, 120.8, 150.0, 180.0, 200.8, 240.0, and 298.6 K, respectively; electrical resistivity ratio $\rho_{300\text{K}}/\rho_{4.2\text{K}} = 45.7$ ; residual electrical resistivity 0.76 $\mu\text{ohm cm}$ ; Lorenz function reported as $3.37 \times 10^{-8} \text{ V}^2 \text{ K}^{-2}$ , 2.33, 2.22, 2.19, 2.26, 2.46, 2.58, 2.61, and $2.61 \times 10^{-8} \text{ V}^2 \text{ K}^{-2}$ at 5.5, 12.2, 18.4, 24.8, 30.7, 45.2, 58.9, 68.1, 77.6, 158.3, and 300.0 K, respectively; heat flow along c-axis.

\*Boyd, D. W. and Legvold, S., "Thermal Conductivities and Lorenz Functions of Dy, Er, and Lu Single Crystals," to be published in Physical Review; also USAEC IS-T-185, 1967.





FIGURE AND TABLE NO. 28R RECOMMENDED THERMAL CONDUCTIVITY OF LUTETIUM



RECOMMENDED VALUES\*

T <sub>1</sub>	(/ / to c-axis)		(⊥ to c-axis)		(Polycrystalline)		T <sub>2</sub>
	k <sub>1</sub>	k <sub>2</sub>	k <sub>1</sub>	k <sub>2</sub>	k <sub>1</sub>	k <sub>2</sub>	
0	0	0	0	0	0	0	-459.7
1	(0.0156)†	(0.901)	(0.00515)	(0.298)	(0.00746)	(0.431)	-457.9
2	(0.0500)	(2.99)	(0.0167)	(0.965)	(0.0240)	(1.39)	-456.1
3	(0.0993)	(5.74)	(0.0330)	(1.91)	0.0476	2.75	-454.3
4	(0.162)	(9.36)	(0.0539)	(3.11)	0.0772	4.46	-452.5
5	(0.218)	(12.6)	(0.0748)	(4.32)	0.107	6.18	-450.7
6	(0.263)	(15.2)	0.0940	5.43	0.133	7.68	-448.9
7	(0.296)	(17.1)	0.111	6.41	0.154	8.90	-447.1
8	0.323	18.7	0.125	7.22	0.172	9.94	-445.3
9	0.344	19.9	0.136	7.86	0.186	10.7	-443.5
10	0.360	20.8	0.146	8.44	0.198	11.4	-441.7
11	0.373	21.6	0.154	8.90	0.207	12.0	-439.9
12	0.383	22.1	0.161	9.30	0.216	12.5	-438.1
13	0.391	22.6	0.167	9.65	0.222	12.8	-436.3
14	0.398	23.0	0.172	9.94	0.228	13.2	-434.5
15	0.403	23.3	0.176	10.2	0.233	13.5	-432.7
16	0.406	23.5	0.179	10.3	0.236	13.6	-430.9
18	0.408	23.6	0.185	10.7	0.241	13.9	-427.3
20	0.405	23.4	0.188	10.9	0.243	14.0	-423.7
25	0.382	22.1	0.191	11.0	0.238	13.8	-414.7
30	0.351	20.3	0.188	10.9	0.229	13.2	-405.7
35	0.329	19.0	0.182	10.5	0.222	12.8	-386.7
40	0.317	18.3	0.178	10.3	0.217	12.5	-387.7
45	0.306	17.8	0.175	10.1	0.213	12.3	-378.7
50	0.303	17.5	0.173	10.0	0.209	12.1	-369.7
60	0.296	17.1	0.169	9.76	0.204	11.8	-351.7
70	0.290	16.8	0.166	9.59	0.199	11.5	-333.7
80	0.285	16.5	0.163	9.42	0.196	11.3	-315.7
90	0.280	16.2	0.161	9.30	0.192	11.1	-297.7
100	0.276	15.9	0.159	9.19	0.189	10.9	-279.7
150	0.261	15.1	0.152	8.78	0.179	10.3	-189.7
200	0.249	14.4	0.147	8.49	0.172	9.94	-99.7
250	0.240	13.9	0.142	8.20	0.166	9.59	-
273.2	0.236	13.6	0.140	8.09	0.164	9.48	32.0
300	0.232	13.4	0.138	7.97	0.162	9.36	80.3

REMARKS

The recommended values are for well-annealed 99.96% pure lutetium with residual electrical resistivity  $\rho_0 = 0.8$  and  $2.7 \mu\Omega$  cm respectively in the direction parallel and perpendicular to the c-axis (characterization by  $\rho_0$  becomes important at temperatures below about 200 K). The recommended values that are supported by experimental data are thought to be accurate to within 5% of the true values near room temperature and 5 to 15% at other temperatures.

\* T<sub>1</sub> in K, k<sub>1</sub> in Watt cm<sup>-1</sup> K<sup>-1</sup>, T<sub>2</sub> in F, and k<sub>2</sub> in Btu lb<sup>-1</sup> ft<sup>-1</sup> F<sup>-1</sup>. † Values in parentheses are extrapolated.

# THERMAL CONDUCTIVITY OF MAGNESIUM

FIGURE SHOWS ONLY 22 OF THE CURVES REPORTED IN TABLE

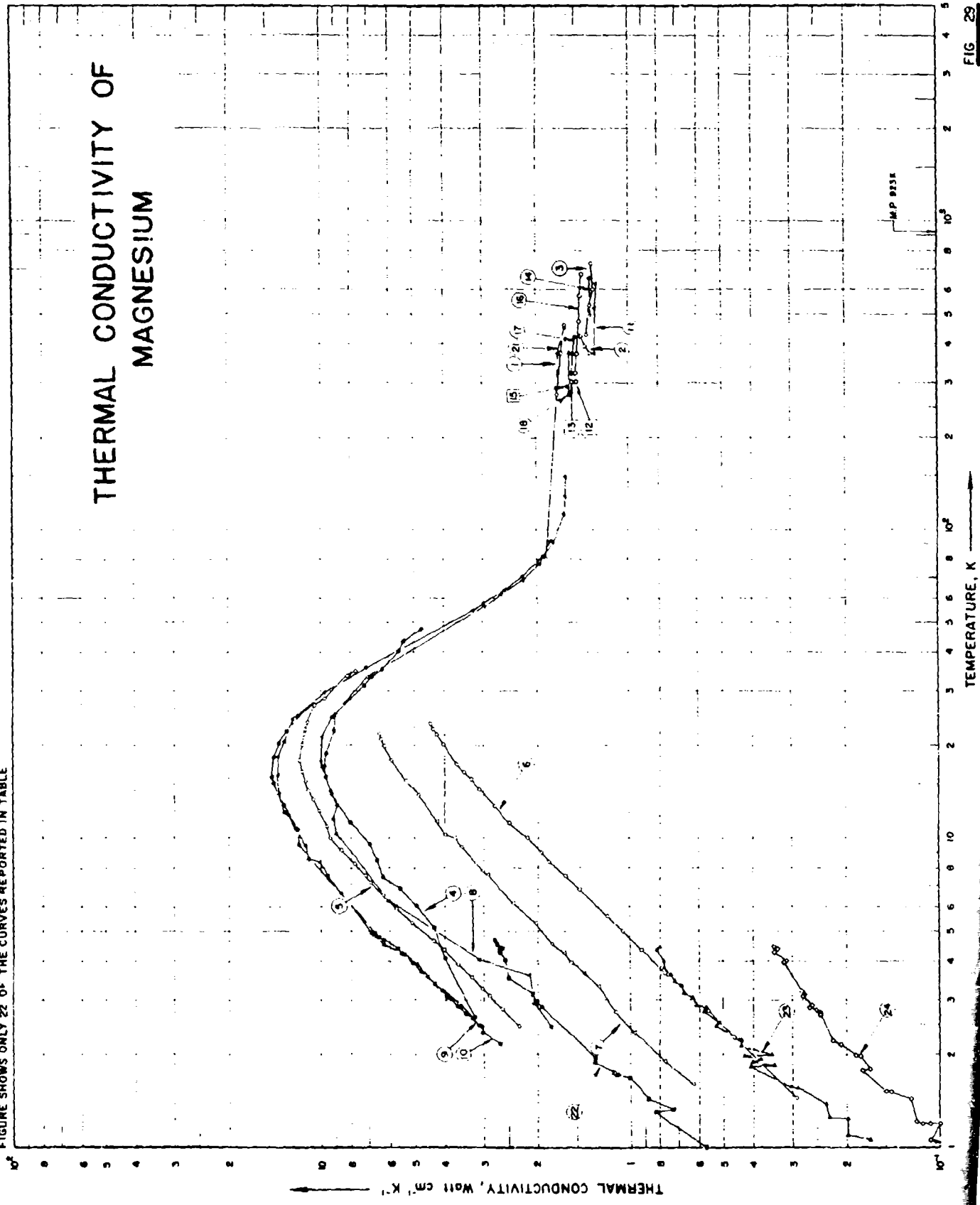


FIG 29

## SPECIFICATION TABLE NO. 29 THERMAL CONDUCTIVITY OF MAGNESIUM

(Impurity &lt; 0.20% each; total impurities &lt; 0.50%)

[For Data Reported in Figure and Table No. 29 ]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)	Specifications and Remarks
1	850, 93	L	1929	80-460	3.0-4.0			Extremely pure; 3 cm x 1.23 cm <sup>2</sup> ; electrical resistivity reported as 0.32, 3.91, 5.56, and 7.27 $\mu$ ohm cm at 80, 273, 373, and 460 K, respectively.
2	51	E	1927	373-423	1.0			Commercially pure.
3	127	L	1925	430-729				99.6 pure; 0.75 in. rod, obtained from Magnesium Co., Ltd.; extruded, then annealed for 6 hrs at 360 C; density at 21 C) 1.75 g cm <sup>-3</sup> ; electrical resistivity reported as 4.59, 6.19, 8.13, 10.35, 11.13, and 13.74 $\mu$ ohm cm at 20.0, 101.2, 199.4, 314.0, 348.3, and 480.1 C, respectively.
4	97	L	1952	2.6-47	2.0-3.0	JM 1703; Mg 1		99.95 pure; polycrystalline; 1-2 mm dia x 5 cm long; obtained from Johnson-Matthey.
5	121	L	1954	2.5-35	1.0	JM 1703; Mg 2		99.95 <sup>+</sup> Mg, 0.03 Mn, 0.0075 Fe, and 0.004 Al; 15 mm dia specimen made from Johnson Matthey standardized rod; annealed in vacuo at 500 C for 6 hrs.
6	137	L	1957	1.5-24	1.0	Mg (Mn)		99.95 <sup>+</sup> Mg, 0.043 Mn, 0.0048 Zn, 0.0012 Ca, 0.0011 Pb, 0.0011 Sn, 0.0010 Fe, 0.001 Si, 0.0002 Al, 0.0001 Cu, and 0.0001 Ni; polycrystalline; 3.2 mm dia x 9 cm long; prepared by Dow Chemical Co.; annealed; electrical resistivity reported as 0.1479, 0.1279, 0.1196, 0.1147, and 0.1217 $\mu$ ohm cm at 1, 5, 10, 14.5, and 20 K, respectively.
7	137	L	1957	1.6-22	1.0	Mg (Fe)		99.94 <sup>+</sup> Mg, 0.013 Fe, 0.0023 Mn, 0.0013 Pb, traces of Al, Ca, Cu, Si, Ag, and Na; polycrystalline; 3.2 mm dia x 9 cm long; prepared from a Johnson Matthey spectrographic rod; electrical resistivity reported as 0.06624, 0.06456, 0.0655, 0.0679, and 0.0727 $\mu$ ohm cm at 1, 5, 10, 15, and 20 K, respectively.
8	275	L	1953	2.5-91	0.5-1.0	JM 1848; Mg 1		99.94 <sup>+</sup> Mg, 0.013 Fe, 0.0023 Mn, 0.0013 Pb, traces of Ca, Cu, Si, Ag, and Na; 3 mm dia rod drawn by Johnson-Matthey from a sample JM 1848.
9	275	L	1953	2.5-149	0.5-1.0	JM 1848; Mg 2		The above specimen annealed in vacuo for 3 hrs at 350 C.
10	275	L	1953	2.2-27	0.5-1.0	JM 1848; Mg 3		Similar to the above specimen Mg 2.
11	225	L	1928	373-623				J, 0.01 Si, 0.012 Cu, and 0.014 total Fe and Al; 1 in. dia x 12 in. long; annealed for 5 hrs at 530 C before machining.
12	408	E	1925	302.2	< 0.5	Mg		0.175 Si, 0.052 Al, and 0.014 Fe; 3 mm dia x 20 cm long; chill-cast; electrical resistivity 4.32 $\mu$ ohm cm at 29 C.
13	408	E	1925	301.2	< 0.5	Mg		The above specimen annealed for 30 min at 450 C; electrical resistivity 4.42 $\mu$ ohm cm at 29 C.
14	295		1952	493, 553				Extruded powder specimen; density 98-100% of theoretical value.
15	673	E	1932	291.3	$\pm$ 1.3			Pure; electrical conductivity $2.31 \times 10^8$ ohm <sup>-1</sup> cm <sup>-1</sup> at 18.1 C.
16	674	L, C	1964	323-673		Mg 1		99.95 Mg, 0.033 Al, and 0.012 Zn; 1.9 cm in dia and 30 cm long; supplied by the Metallurgy Division of the National Physical Laboratory; forged and stabilizing heat treated; electrical resistivity reported as 4.5, 5.01, 5.85, 7.57, 9.30, and 11.04 $\mu$ ohm cm at 293, 323, 373, 473, 573, and 673 K, respectively.

SPECIFICATION TABLE NO. 29 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
17	674	L, C	1964	323-423		Mg II	99.96 Mg, 0.017 Al, and 0.004 Zn; 0.635 cm in dia and 10 cm long; supplied by Messrs. Johnson, Matthey & Co., Ltd.; electrical resistivity reported as 4.34, 4.85, 5.70, and 6.51 $\mu\text{ohm cm}$ at 293, 323, 373, and 423 K, respectively. Electrical conductivity reported as 24.47 and 17.5 $\times 10^4 \text{ ohm}^{-1}\text{cm}^{-1}$ at 0 and 100 C, respectively.
18	706	L	1981	273, 373			Spectroscopically pure; specimen 1.27 cm long; thermal conductivity values calculated from measured thermal diffusivity data.
19	719	P	1965	293.2			Similar to the above specimen but 0.635 cm long.
20	719	P	1965	293.2			Specimen 4.025 in. in dia and 1.015 in. thick.
21	722	L	1963	307-324		Sample # 765	99.98 <sup>+</sup> Mg, 0.01 Mn, 0.003 Zn, 0.0012 Pb, 0.001 Ca, <0.001 Si, <0.001 Sn, 0.0008 Fe, 0.0002 Al, <0.0001 Cu, and <0.0001 Ni; specimen 9.03 cm long and 0.319 cm in dia; electrical resistivity reported as 0.048, 0.042, 0.04, 0.037, 0.04, 0.042, 0.047, and 0.058 $\mu\text{ohm cm}$ at 1.0, 3.0, 5.0, 10.0, 15.0, 20.0, 25.0, and 30 K, respectively.
22	721	L	1953	1.0-4.7			and 0.058 $\mu\text{ohm cm}$ at 1.0, 3.0, 5.0, 10.0, 15.0, 20, and 25 K, respectively.
23	721	L	1953	1.1-4.4		Sample # 767	99.95 Mg, 0.043 Mn, 0.0048 Zn, 0.0012 Ca, 0.0011 Pb, 0.0011 Sn, 0.0010 Fe, <0.001 Si, 0.0002 Al, <0.0001 Cu, and <0.0001 Ni; specimen 8.93 cm long and 0.307 cm in dia; electrical resistivity reported as 0.153, 0.144, 0.136, 0.123, 0.120, 0.127, and 0.137 $\mu\text{ohm cm}$ at 1.0, 3.0, 5.0, 10, 15, 20, and 25 K, respectively.
24	721	L	1953	1.0-4.5		Sample # 370	99.87 <sup>+</sup> Mg, 0.12 Mn, 0.0036 Zn, 0.0014 Pb, 0.0011 Fe, <0.001 Si, <0.001 Sn, 0.0006 Ca, 0.0002 Al, <0.0002 Ni, and 0.0001 Cu; specimen 9.35 cm long and 0.305 cm in dia; electrical resistivity reported as 0.365, 0.34, 0.32, 0.29, 0.275, 0.30, and 0.37 $\mu\text{ohm cm}$ at 1.0, 3.0, 5.0, 10, 20, 30, and 40 K, respectively.
25	943	P	1966	298.2			Spherical grains supplied by Valley Metallurgical Processing Co.; specimen contained in a 0.75 in. dia $\times$ 2 in. long cylindrical cell; mesh size -100 + 200; thermal conductivity measured by the transient line source method; measured in Freon - 12 under a pressure of $\sim$ 100 psig.
26	843	P	1966	298.2			Similar to above; measured in argon under a pressure of $\sim$ 100 psig.
27	843	P	1966	298.2			Similar to above; measured in nitrogen under a pressure of $\sim$ 100 psig.
28	843	P	1966	298.2			Similar to above; measured in methane under a pressure of $\sim$ 100 psig.
29	843	P	1966	298.2			Similar to above; measured in helium under a pressure of $\sim$ 100 psig.
30	843	P	1966	298.2			Similar to above; measured in hydrogen under a pressure of $\sim$ 100 psig.

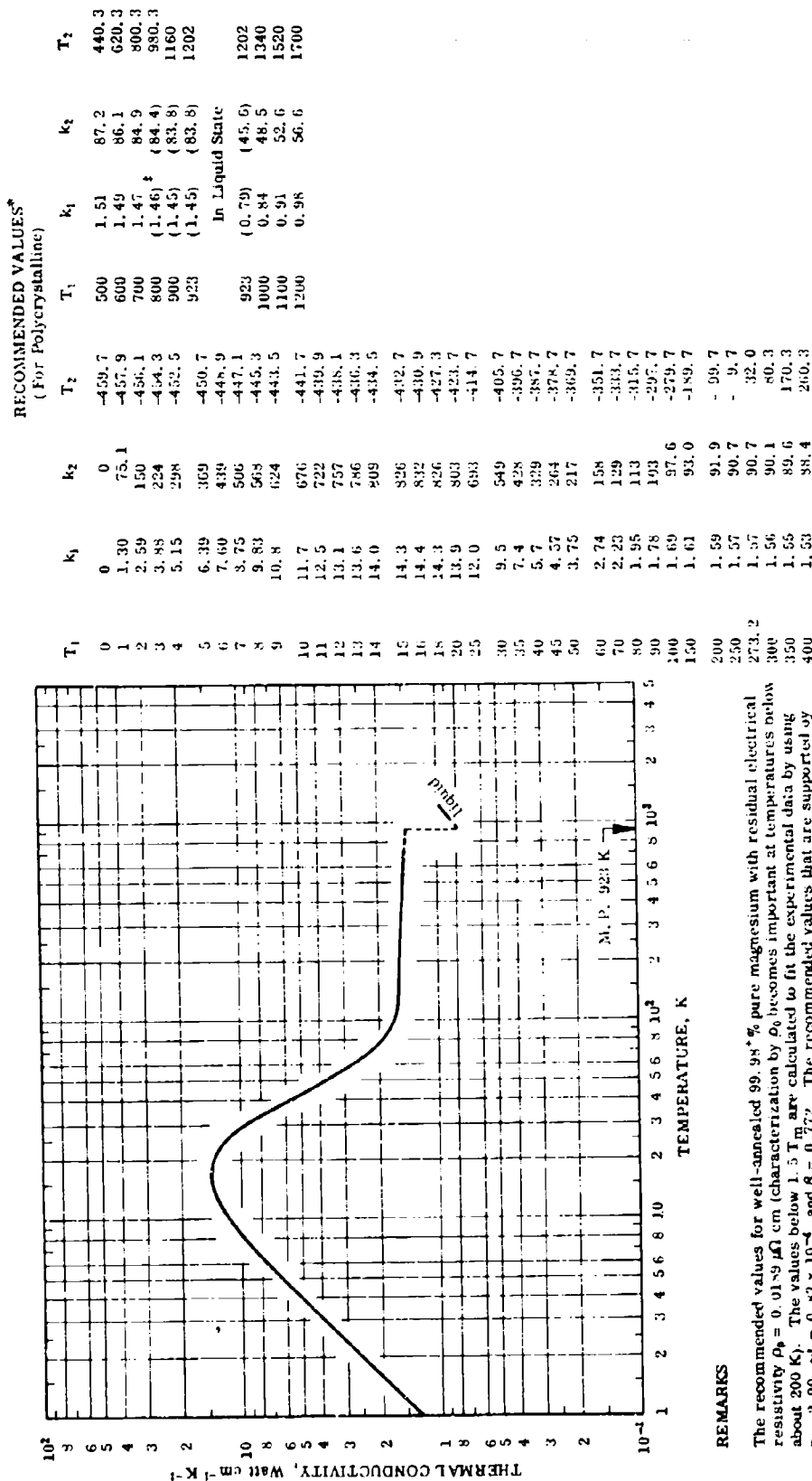


DATA TABLE NO. 29 (continued)

T	k	T	k
<u>CURVE 23 (cont.)</u>			
4.34	0.82	298.2	0.00272
4.44	0.805		
<u>CURVE 24</u>			
1.02	0.092 <sup>c</sup>	298.2	0.00406
1.04	0.096		
1.06	0.108		
1.20	0.100	298.2	0.00112
1.20	0.108		
1.20	0.115		
1.22	0.120		
1.45	0.125		
1.52	0.145	298.2	0.0155
1.53	0.150		
1.78	0.180		
1.80	0.170		
1.97	0.182		
1.99	0.188		
2.16	0.210		
2.22	0.223		
2.70	0.245		
2.74	0.245		
2.77	0.252		
2.85	0.265		
2.90	0.260		
3.10	0.278		
3.13	0.275		
3.97	0.320		
4.02	0.315		
4.22	0.33		
4.30	0.345		
4.39	0.335		
4.48	0.345		
<u>CURVE 25<sup>a</sup></u>			
298.2	0.00100		
<u>CURVE 26<sup>b</sup></u>			
298.2	0.00269		

Not shown on plot

FIGURE AND TABLE NO. 29R RECOMMENDED THERMAL CONDUCTIVITY OF MAGNESIUM



## REMARKS

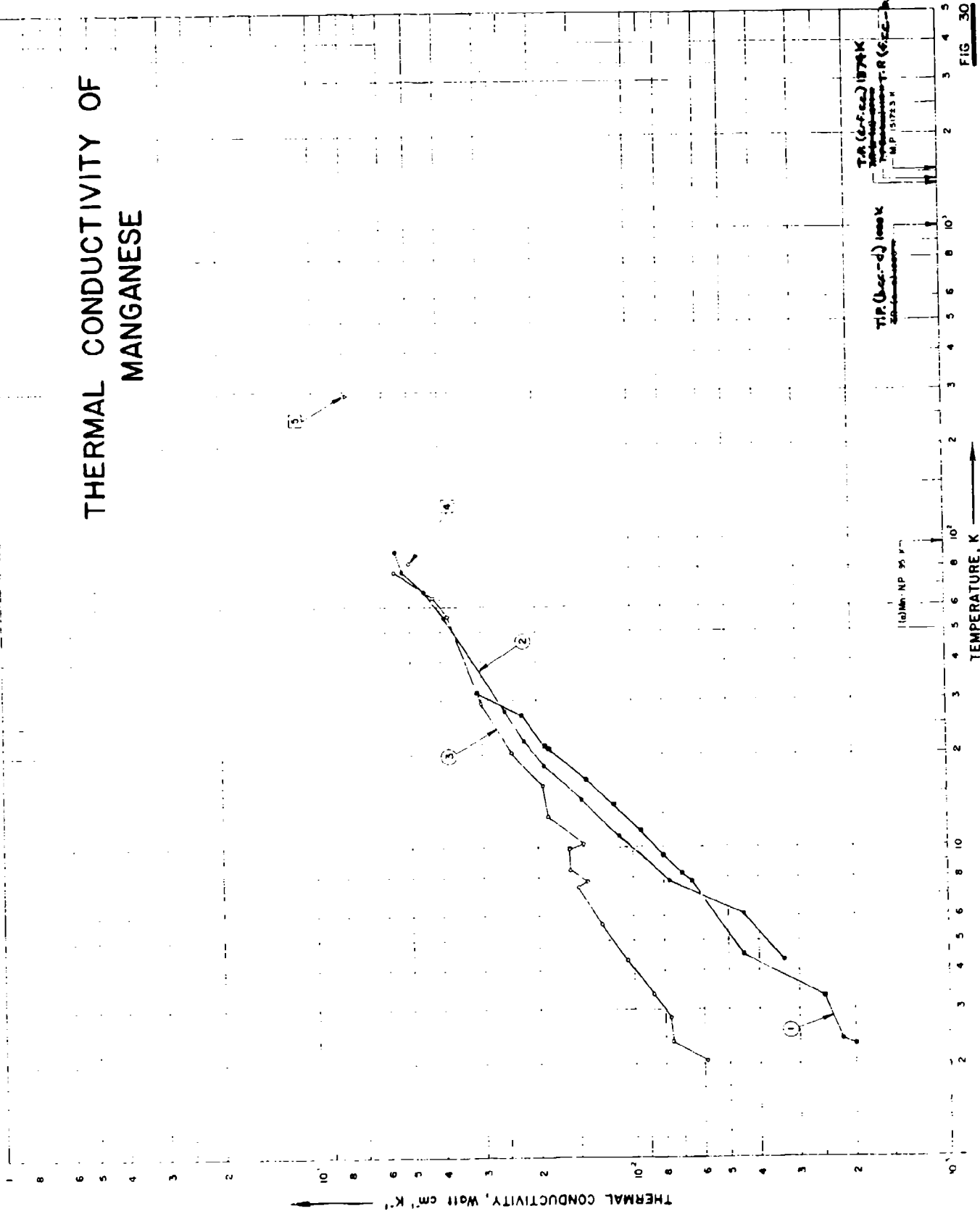
The recommended values for well-annealed 99.98% pure magnesium with residual electrical resistivity  $\rho_0 = 0.019 \mu\Omega$  cm (characterization by  $\rho_0$  becomes important at temperatures below about 200 K). The values below 1.5 Tm are calculated to fit the experimental data by using  $n = 2.00$ ,  $\alpha^2 = 0.83 \times 10^{-4}$ , and  $\beta = 0.772$ . The recommended values that are supported by experimental thermal conductivity data are thought to be accurate to within 3% of the true values near room temperature, and 3 to 10% at other temperatures.

\*  $T_1$  in K,  $k_1$  in Watt  $\text{cm}^{-1} \text{K}^{-1}$ ,  $T_2$  in F, and  $k_2$  in Btu  $\text{hr}^{-1} \text{ft}^{-1} \text{F}^{-1}$ .

† Values in parentheses are extrapolated.



# THERMAL CONDUCTIVITY OF MANGANESE



## SPECIFICATION TABLE NO. 30 THERMAL CONDUCTIVITY OF MANGANESE

(Impurity &lt; 0.20% each; total impurities &lt; 0.50%)

[ For Data Reported in Figure and Table No. 30 ]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition ( weight percent).	Specifications and Remarks
1	97	L	1952	2.3-32	2-3	JM 2472; Mn 1	99.99 pure; polycrystalline; supplied by Johnson Matthey and Co. Ltd.; annealed; electrical resistivity ratio $\rho(273K)/\rho(20K) = 1.47$ .	
2	372	L	1957	4.3-9.		Mn 2	99.99 Mn, 0.001 Mg; $\alpha$ - manganese; cross section 3 x 1.1 mm; JM 10792 from Johnson, Matthey Co. Ltd.; electrical resistivity reported as: 330 and 378 $\mu$ ohm cm at 4.2 K and room temperature, respectively.	
3	372	L	1957	2.1-78		Mn 3	Similar to the above specimen except annealed in vacuum at 600 C; electrical resistivity 150 $\mu$ ohm cm at room temperature; residual electrical resistivity 11.3 $\mu$ ohm cm.	
4	697	L	1955	83.2			$\beta$ - manganese; approx 16 mm long, 5 mm dia; electrical resistivity 110 $\mu$ ohm cm at -19 $^{\circ}$ C.	
5	255	C	1966	293		JM 810	Pure $\alpha$ - manganese with impurities Mg, Ca, and < 0.01 S; some gaseous impurity expected; specimen a small irregular-shaped flake ~ 0.1 cm thick from Johnson Matthey and Co. Ltd.; electrolytically prepared; high-alloy steel, titanium alloy, and an alumina based ceramic used as reference materials.	

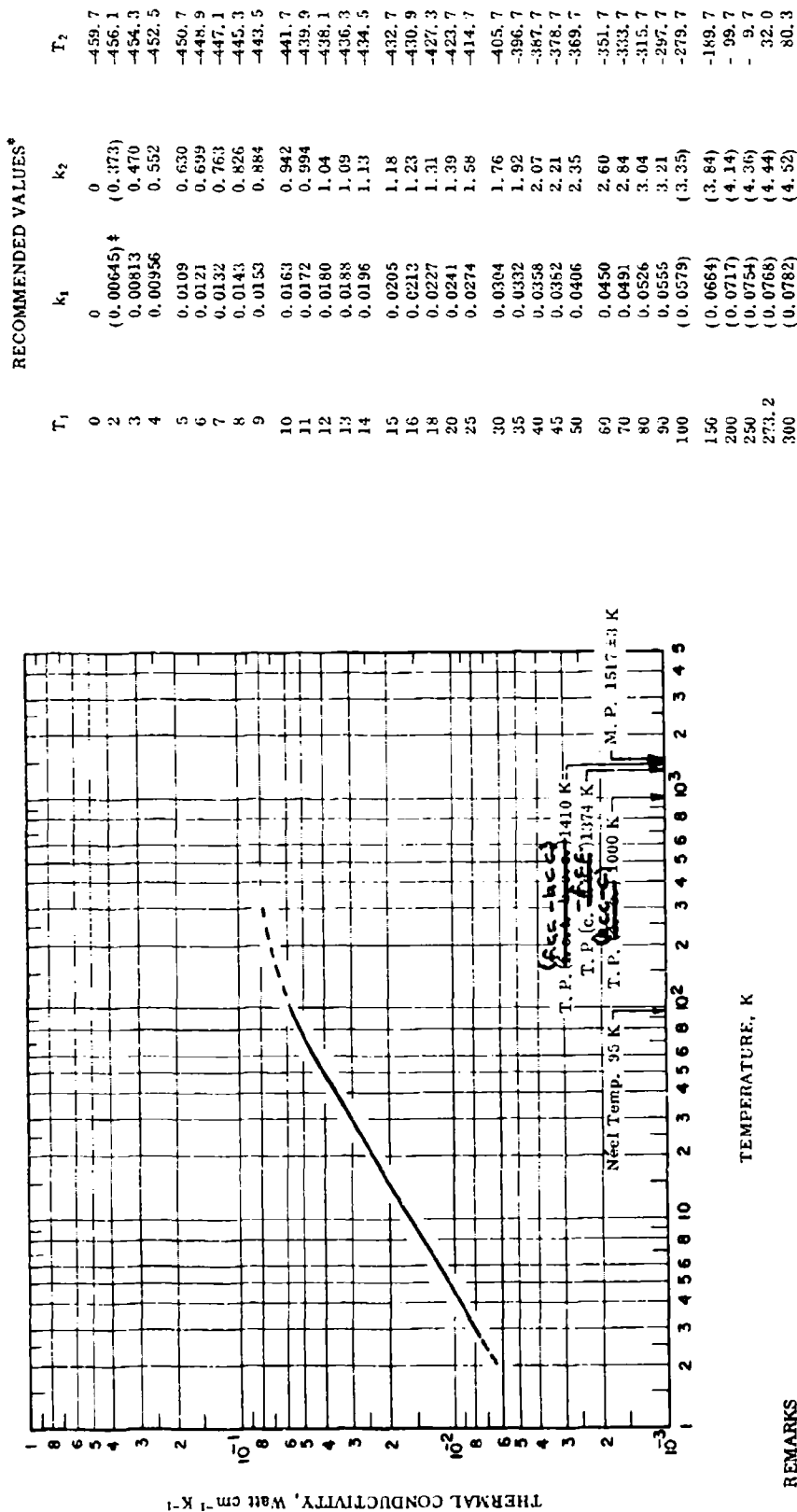
DATA TABLE NO. 30 THERMAL CONDUCTIVITY OF MANGANESE

(impurity &lt; 0.20% each; total impurities &lt; 0.50%)

[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup> K<sup>-1</sup>]

T	k	T	k
<u>CURVE 1</u>			
2.3	0.0020	12.71	0.0186
2.4	0.0022	15.9	0.0193
3.3	0.0025	20.3	0.0241
4.5	0.0045	29.3	0.0302
7.7	0.0055	55.9	0.038
8.2	0.007	64.2	0.042
9.4	0.009	78.0	0.056
11.3	0.0095	<u>CURVE 4</u>	
13.8	0.0115	83.2	0.05
16.7	0.014	<u>CURVE 5</u>	
20.9	0.0185	293.0	0.078
21.4	0.019	<u>CURVE 2</u>	
26.9	0.022	4.31	0.00335
31.8	0.031	6.09	0.00448
<u>CURVE 3</u>			
7.97	0.00772	10.9	0.0111
14.3	0.0147	14.3	0.0147
18.4	0.0191	18.4	0.0191
22.2	0.022	22.2	0.022
27.6	0.0253	27.6	0.0253
55.7	0.039	55.7	0.039
67.5	0.045	67.5	0.045
78.0	0.053	78.0	0.053
90.5	0.055	90.5	0.055
<u>CURVE 3</u>			
2.05	0.0059	2.37	0.0076
2.82	0.0077	3.37	0.00876
4.32	0.0106	5.66	0.0126
7.42	0.0151	7.77	0.0141
8.43	0.0158	9.85	0.0160
10.28	0.0144		

FIGURE AND TABLE NO. 30R RECOMMENDED THERMAL CONDUCTIVITY OF MANGANESE



REMARKS

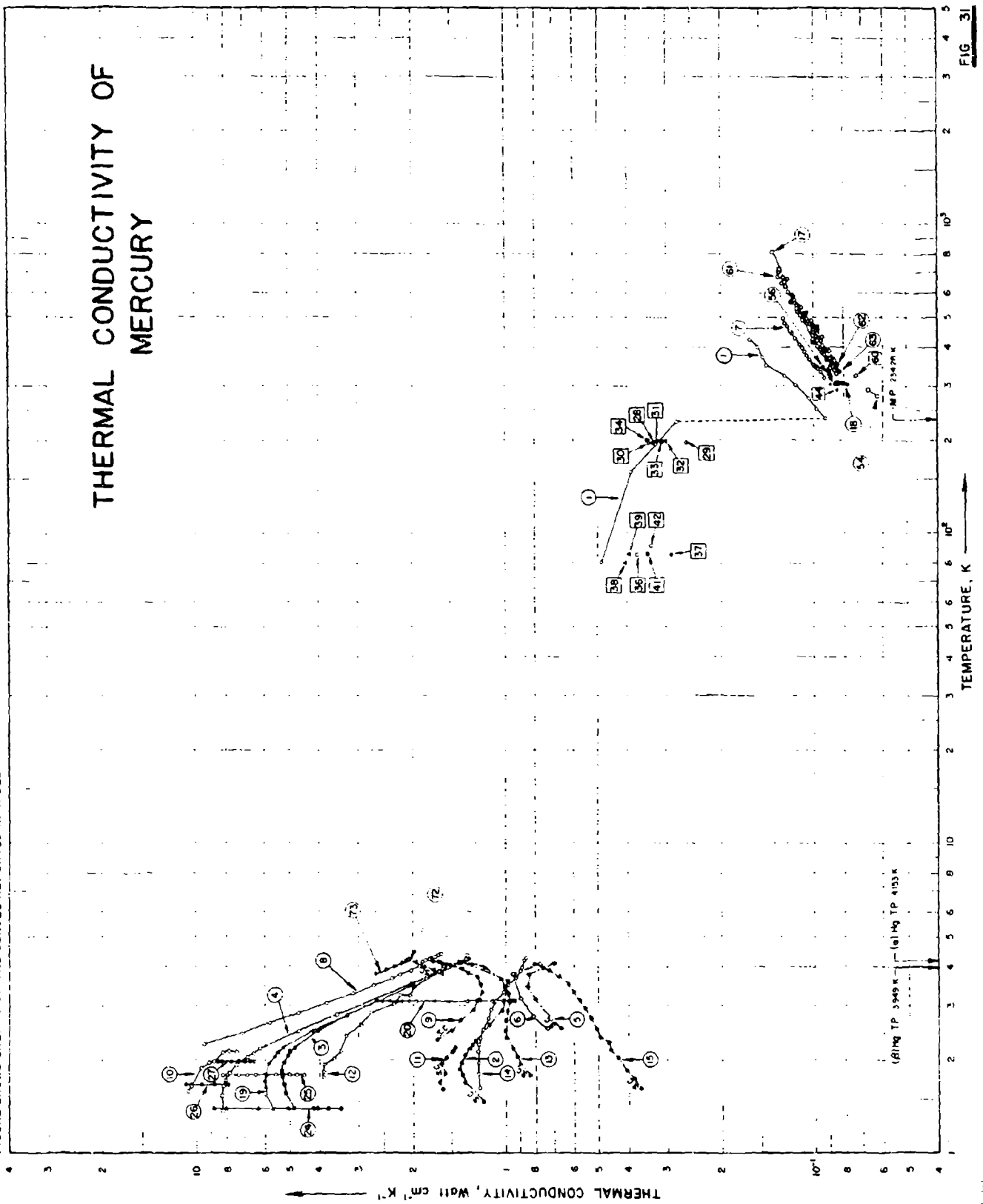
The recommended values are for well-annealed 99.99% pure manganese with residual electrical resistivity  $\rho_0 \approx 11.3 \mu\Omega$  cm (characterization by  $\rho_0$  becomes important at temperatures below about 200 K). The recommended values that are supported by experimental thermal conductivity data are thought to be accurate to within 5% of the true values near room temperature and 5 to 15% at other temperatures.

\* T<sub>1</sub> in K, k<sub>1</sub> in Watt cm<sup>-1</sup> K<sup>-1</sup>, T<sub>2</sub> in F, and k<sub>2</sub> in Btu hr<sup>-1</sup> ft<sup>-1</sup> F<sup>-1</sup>.

† Values in parentheses are extrapolated or interpolated.

# THERMAL CONDUCTIVITY OF MERCURY

FIGURE SHOWS ONLY 46 OF THE CURVES REPORTED IN TABLE



## SPECIFICATION TABLE NO. 31 THERMAL CONDUCTIVITY OF MERCURY

(Impurity: 0.20% each, total impurities: 0.50%)

(For Data Reported in Figure and Table No. 31)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
1	50	L	1919	80-423			Pure; specimen filled in an iron cylinder; electrical conductivity reported as 13.77, 7.292, 5.327, 4.287, 1.070, and $1.055 \times 10^4 \text{ ohm}^{-1} \text{cm}^{-1}$ at -193, -115, -75, 1, -44.7, -37, and -20.5 C., respectively.
2	143	L	1957	1.5-4.1		Hg 3	99.9 Hg, 0.05 Cu, 0.05 Ag, and trace of other base elements; polycrystalline; commercially available cp reagent from Elmer and Amend, Cat. No. M-141, in superconducting state.
3	143	L	1957	1.4-3.5		Hg 3	The above specimen measured in a transverse magnetic field of 459 gauss; in normal state.
4	143	L	1957	1.4-4.4		Hg 3	The above specimen measured in a transverse magnetic field of 491 gauss; in normal state.
5	59	L	1936	2.5-4.1			High purity; specimen contained in an U-shaped tube; in superconducting state.
6	59	L	1936	2.5-4.1			The above specimen measured in a magnetic field of 436 gauss; in normal state.
7	316,65	L	1936	318-492			Pure; specimen contained in a 4.9 cm dia asbestos cylinder.
8	74	L	1950	2.3-4.4		Hg 1	99.99 <sup>1</sup> pure; in normal state.
9	74	L	1950	2.3-4.2		Hg 1	99.99 <sup>1</sup> pure; in superconducting state.
10	74	L	1950	1.6-2.1		Hg 2	0.002 Cd; in normal state.
11	74	L	1950	1.6-2.2		Hg 2	0.002 Cd; in superconducting state.
12	74	L	1950	1.8-4.2		Hg 3	0.007 Cd; in normal state.
13	74	L	1950	1.8-4.0		Hg 3	0.007 Cd; in superconducting state.
14	74	L	1950	1.6-4.3		Hg 6	0.10 In; in normal state.
15	74	L	1950	1.6-4.1		Hg 6	0.10 In; in superconducting state.
16	657	P	1926	298			Liquid specimen contained in a cylindrical tube of 4 cm dia and 20 cm long; thermal conductivity value calculated from measured thermal diffusivity.
17	265	L	1955	426-810	1.4		99.999 <sup>4</sup> Hg, 0.0001-0.001 Mg; chemical analysis after experiment showed 0.0004 Fe, 0.0002 Cr, and 0.0001 Ni; Lorentz function reported as 2.64, 2.59, 2.61, 2.63, and $2.64 \times 10^{-8} \text{ V}^2 \text{K}^{-2}$ at 100, 184, 256, 288, and 297 C., respectively.
18	258	L	1903	303-308			Pure; specimen filled in a container of cross sectional area 315 cm <sup>2</sup> and thickness 0.955 cm.
19	143	L	1957	1.4-4.2		Hg 3	99.9 Hg, 0.05 Cu, 0.05 Ag, and trace of other base elements; commercially available cp reagent from Elmer and Amend, Cat. No. M-141; measured in a magnetic field of 737 gauss; in normal state.
20	143	L	1957	3.1		Hg 1	99.99 <sup>1</sup> Hg, 0.005 Ag, and trace Cu; commercially available cp reagent from Elmer and Amend, Cat. No. M-141; measured in transverse magnetic fields with strength H ranging from 8.4 to 190 gauss; in superconducting state.
21	143	L	1957	3.1		Hg 1	The above specimen measured in transverse magnetic fields with strength H ranging from 247 to 974 gauss; in normal state.

SPECIFICATION TABLE NO. 31 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)	Specifications and Remarks
22	143	L	1957	3.1		Hg 1		The above specimen measured in longitudinal magnetic fields with strength H ranging from 139 to 198 gauss, in superconducting state.
23	143	L	1957	3.1		Hg 1		The above specimen measured in longitudinal magnetic fields with strength H ranging from 281 to 727 gauss, in normal state.
24	143	L	1957	1.1		Hg 2		99.995% Hg, trace Ag; commercially available ep reagent from Eimer and Amend, Cat. No. M-141; measured in transverse magnetic fields with strength H ranging from 455 to 965 gauss, in normal state.
25	143	L	1957	1.8		Hg 2		The above specimen measured in transverse magnetic fields with strength H ranging from 460 to 1060 gauss, in normal state.
26	143	L	1957	1.67		Hg 2		The above specimen measured in longitudinal magnetic fields with strength H ranging from 455 to 965 gauss, in normal state.
27	143	L	1957	1.98		Hg 2		The above specimen measured in longitudinal magnetic fields with strength H ranging from 440 to 943 gauss, in normal state.
28	254	L	1932	196.2		7		Single crystal; the angle between principal crystallographic axis and rod axis $9 - 21^\circ$ , grown in a glass tube; electrical resistivity $15.15 \mu\text{ohm cm}$ at $196.2 \text{ K}$ .
29	254	L	1932	196.4		22		Similar to above but $\theta = 90^\circ$ and electrical resistivity $19.30 \mu\text{ohm cm}$ at $196.4 \text{ K}$ .
30	254	L	1932	196.8		23		Similar to above but $\theta = 90^\circ$ and electrical resistivity $14.58 \mu\text{ohm cm}$ at $196.8 \text{ K}$ .
31	254	L	1932	197.5		26		Similar to above but $\theta = 25^\circ$ and electrical resistivity $15.72 \mu\text{ohm cm}$ at $197.5 \text{ K}$ .
32	254	L	1932	197.6		27		Similar to above but $\theta = 46^\circ$ and electrical resistivity $17.15 \mu\text{ohm cm}$ at $197.6 \text{ K}$ .
33	254	L	1932	197.1		28		Similar to above but $\theta = 38^\circ$ and electrical resistivity $16.41 \mu\text{ohm cm}$ at $197.1 \text{ K}$ .
34	254	L	1932	198.4		29		Similar to above but $\theta = 0^\circ$ and electrical resistivity $14.74 \mu\text{ohm cm}$ at $198.4 \text{ K}$ .
35	254	L	1932	197.3		30		Similar to above but $\theta = 0^\circ$ and electrical resistivity $14.63 \mu\text{ohm cm}$ at $197.3 \text{ K}$ .
36	254	L	1932	85.2		3		Similar to above but $\theta = 25^\circ$ and electrical resistivity $5.95 \mu\text{ohm cm}$ at $85.2 \text{ K}$ .
37	254	L	1932	85.4		22		Similar to above but $\theta = 90^\circ$ and electrical resistivity $7.45 \mu\text{ohm cm}$ at $85.4 \text{ K}$ .
38	254	L	1932	80.2		23		Similar to above but $\theta = 0^\circ$ and electrical resistivity $5.31 \mu\text{ohm cm}$ at $80.2 \text{ K}$ .
39	254	L	1932	85.5		24		Similar to above but $\theta = 8^\circ$ and electrical resistivity $5.75 \mu\text{ohm cm}$ at $85.5 \text{ K}$ .
40	254	L	1932	85.5		26		Similar to above but $\theta = 24^\circ$ and electrical resistivity $6.05 \mu\text{ohm cm}$ at $85.5 \text{ K}$ .
41	254	L	1932	86.6		27		Similar to above but $\theta = 46^\circ$ and electrical resistivity $6.65 \mu\text{ohm cm}$ at $86.6 \text{ K}$ .
42	254	L	1932	90.6		28		Similar to above but $\theta = 46^\circ$ and electrical resistivity $7.02 \mu\text{ohm cm}$ at $90.6 \text{ K}$ .
43	254	L	1932	86.2		30		Similar to above but $\theta = 0^\circ$ and electrical resistivity $5.73 \mu\text{ohm cm}$ at $86.2 \text{ K}$ .

SPECIFICATION TABLE NO. 31 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
44	594	L	1913	290.5			In liquid state; measurement made on a flowing specimen at a rate of 871 g per 15 min.
45	593	L	1913	248.2			In liquid state; measurement made on a flowing specimen at a rate of 1003 g per 15 min.
46	598	L	1913	249.0			In liquid state; measurement made on a flowing specimen at a rate of 1079 g per 15 min.
47	594	L	1913	290.7			In liquid state; measurement made on a flowing specimen at a rate of 1099 g per 15 min.
48	598	L	1913	248.2			In liquid state; measurement made on a flowing specimen at a rate of 1159 g per 15 min.
49	598	L	1913	248.2			In liquid state; measurement made on a flowing specimen at a rate of 1199 g per 15 min.
50	594	L	1913	287.2			In liquid state; measurement made on a flowing specimen at a rate of 1296 g per 15 min.
51	598	L	1913	288.7			In liquid state; measurement made on a flowing specimen at a rate of 1301 g per 15 min.
52	598	L	1913	288.7			In liquid state; measurement made on a flowing specimen at a rate of 1361 g per 15 min.
53	598	L	1913	288.7			In liquid state; measurement made on a flowing specimen at a rate of 1422 g per 15 min.
54	644	P	1880	279.290			In liquid state; specimen filled in a cylindrical container; thermal conductivity values calculated from measured data of thermal diffusivity, specific heat, and density.
55	638	R	1959	587-717			Chemically pure mercury vapor.
56	639	C	1961	304-343			Triple distilled liquid mercury, contained in a thin-walled reservoir of 3.5 in. deep; electrical resistivity reported as 96.7, 98.6, 101.1, 103.5, 109.1 and 115.0 $\mu\text{ohm cm}$ at 30, 50, 75, 100, 150, and 200 C, respectively; first experiment with thermocouples welded on the steel wall of mercury reservoir; stainless steel No. 15 used as comparative material.
57	640	R	1889	476.2			In vapor state; contained in a glass tube of dia. 18.2 mm; measured at pressures ranging from 3.0 to 10.3 mm Hg; measured by hot-wire method.
58	641	L	1915	313			In liquid state; contained in a cylindrical vessel of 4.9 cm dia x 40 cm long.
59	642	L	1887	323			In liquid state; contained in a tube of 13.2 mm dia x 20 cm long.
60	643	P	1864	323.2			In liquid state; thermal conductivity calculated from measured thermal diffusivity.
61	637, 592	L	1958	328-700		1	In liquid state.
62	637, 592	L	1958	353-556		2	In liquid state.
63	637, 592	L	1958	333-560		3	In liquid state.
64	639	C	1961	307-351			Triple distilled liquid mercury; contained in a thin-walled reservoir of 3.5 in. deep; electrical resistivity reported as 96.7, 98.6, 101.1, 103.5, 109.1, and 115.0 $\mu\text{ohm cm}$ at 30, 50, 75, 100, 150, and 200 C, respectively; first experiment with thermocouple immersed in the mercury; stainless steel No. 15 used as comparative material.
65	639	C	1961	334.7			Second experiment of the above specimen with thermocouples welded on the steel wall of mercury reservoir; mercury and stainless steel in poor electrical contact.



SPECIFICATION TABLE NO. 31 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
66	639	C	1961	319-383			Second experiment of the above specimen with thermocouples immersed in the mercury.
67	639	C	1961	318-317			Third experiment of the above specimen with a honeycomb of mica inserted to subdivide the mercury column and with thermocouples welded on the steel wall of mercury reservoir.
68	639	C	1961	332-416			Third experiment of the above specimen with thermocouples immersed in the mercury.
69	639	C	1961	307-368			Fourth experiment of the above specimen with the mercury column still further subdivided by the insertion of 35 cellophane drinking straws and with thermocouples welded on the steel wall of mercury reservoir.
70	639	C	1961	326-514			Measurement made on new mercury specimen after having the test apparatus thoroughly cleaned with new thermocouples welded on the steel wall of the mercury reservoir, a honeycomb of mica also being inserted to subdivide mercury column.
71	639	L	1961	304, 333			Similar to the above specimen but measured by guarded hot-plate method.
72	633	L	1963	3, 8-1.2			Very pure; crystallized from triple distilled Hg; cylindrical specimen, cast in liquid air; in super conducting state.
73	633	L	1963	3, 8-1.2			The above specimen measured in a magnetic field of 600 gauss; in normal state.
74	633	L	1963	3, 8-1.2			The above specimen measured in a magnetic field of 800 gauss; in normal state.
75	316	L	1936	373-412	± 3	Amalgam: 1	0.104 Na; liquid specimen contained in a hollow asbestos cylinder; prepared by melting (under paraffin) appropriate amounts of certified pure Hg (from Mallinckrodt Chemical Co.) and Na (from Chemistry Department of Univ. of Kansas) in furnace, the liquid then kept at 150° C for 12 hrs.

DATA TABLE NO. 51 THERMAL CONDUCTIVITY OF MERCURY  
 Purity - 0.20% each; total impurities - 0.50%

(Temperature, T, K; Thermal Conductivity, k, Watts cm<sup>-1</sup>K<sup>-1</sup>.)

T	k	T	k	T	k	T	k	T	k	T	k	T	k			
<u>CURVE 1</u>																
80.2	0.485	1.37	4.28	4.12	1.65	1.78	0.884	2.51	0.505	307.54	0.08319	<u>CURVE 18 (cont.)</u>				
137.7	0.389	1.53	8.32	4.15	1.72	1.93	0.855	2.72	0.525	307.62	0.08494					
194.8	0.325	1.65	8.26	4.18		2.05	0.915	2.92	0.550	307.65	0.08017					
229.0	0.278	1.72	8.1	0.115		2.18	0.985	3.10	0.575	307.82	0.08301					
252.5	0.0912	1.82	7.84	0.123		2.36	0.992	3.30	0.600	307.97	0.08023*					
252.5	0.0975	1.93	7.42	0.125		2.48	1.009	3.50	0.624	<u>CURVE 19</u>						
273.2	0.104	2.05	6.86	0.126		2.68	0.995	3.70	0.675	1.38	5.68					
300.2	0.115	2.10	6.54			2.87	0.987	3.87	0.723	1.55	5.96					
323.6	0.125	2.18	6.25			3.06	0.979	3.96	0.759	1.64	5.99					
349.8	0.143	2.48	4.68			3.26	0.975	4.06	0.803	1.72	6.01					
373.1	0.148	2.80	3.50			3.45	1.00	<u>CURVE 16</u>		1.81	5.99					
399.1	0.152	3.50	2.00			3.65	1.03			1.96	5.86					
422.6	0.161	4.27	1.34			3.95	1.17			2.06	5.59					
4.35	1.32	3.05	3.80			1.04	1.28	298	0.094	2.12	5.47					
<u>CURVE 2</u>																
1.46	1.18	3.27	3.12			<u>CURVE 14</u>		<u>CURVE 15</u>		<u>CURVE 17</u>						
1.69	1.35	3.47	2.68			1.61	1.22	426.4	0.1014	489.4	0.1096					
1.78	1.37	3.57	2.35			1.70	1.64	489.4	0.1110	507.9	0.1111					
1.86	1.40	3.77	2.20			1.82	1.57	507.9	0.1139	532.9	0.1144					
1.96	1.39	3.88	2.03			1.95	1.24	600.3	0.1250	670.0	0.1262					
1.95	1.39	4.00	1.91			2.03	1.24	670.0	0.1259	869.5	0.1354					
2.06	1.35	4.14	1.87			2.12	1.25	<u>CURVE 12</u>		<u>CURVE 20</u>						
2.17	1.31	4.18	1.83			2.14	1.48	1.76	3.89	<u>CURVE 3, 1K</u>						
2.23	1.28	4.22	1.77			2.19	1.47	1.83	3.88	8.4	0.938					
2.28	1.26	4.25	1.74			2.22	1.88	1.93	3.85	74.4	0.943					
2.59	1.14	4.29	1.69			2.25	2.00	2.02	3.59	87.4	0.948*					
2.89	1.03	4.34	1.65			2.28	2.14	3.67	3.57	92.0	0.952					
2.59	0.98	4.38	1.61			2.34	1.22	3.86	0.895	100.0	0.956*					
4.17	1.40	4.67	0.943			2.49	1.19	4.16	0.887	106	0.950					
<u>CURVE 9</u>																
2.91	1.67	2.39	3.28			2.70	1.14	302.50	0.08468	114	1.04					
2.48	1.50	2.60	2.95			2.90	1.12	302.57	0.07820	125	1.24					
2.68	1.40	2.84	2.74			3.09	1.08	302.71	0.08092	136	1.35					
2.88	1.28	3.04	2.50			3.30	1.02	302.72	0.08117	148	1.65					
3.09	1.25	3.23	2.05			3.48	0.982	302.75	0.08586	160	1.53					
3.29	1.20	3.23	2.05			3.67	0.957	302.78	0.08021	171	1.72					
345.8	0.0991	3.45	3.99			3.86	0.895	302.82	0.08389*	176	1.96					
345.4	0.0992	3.69	3.45			4.16	0.870	302.82	0.08389*	179	2.08					
347.9	0.0999	3.65	3.81			4.26	0.853	303.12	0.07971	183	2.27					
352.1	0.101	3.84	3.84			1.61	0.365	307.50	0.08146	186	2.40					
352.1	0.101	3.86	3.86			1.73	0.383									
364.4	0.104	3.91	3.59			1.85	0.405									
364.4	0.104	3.91	3.59			2.02	0.430									
375.8	0.106	4.01	3.51			2.15	0.433									
386.1	0.108	4.05	3.51			2.26	0.460									
398.0	0.110	4.09	3.51			2.37	0.493									
<u>CURVE 7</u>																
1.40	4.85	318.1	0.0927			Not shown on plot										
1.56	5.18	330.9	0.0950													
1.67	5.26	346.1	0.0973													
1.77	5.31	345.8	0.0991													
1.85	5.32	345.4	0.0992													
1.96	5.24	347.9	0.0999													
2.06	5.07	352.1	0.101													
2.13	5.00	364.4	0.104													
2.19	4.81	375.8	0.106													
2.48	4.05	386.1	0.108													
2.79	3.26	398.0	0.110													
3.49	2.00	398.0	0.110													

\* Not shown on plot

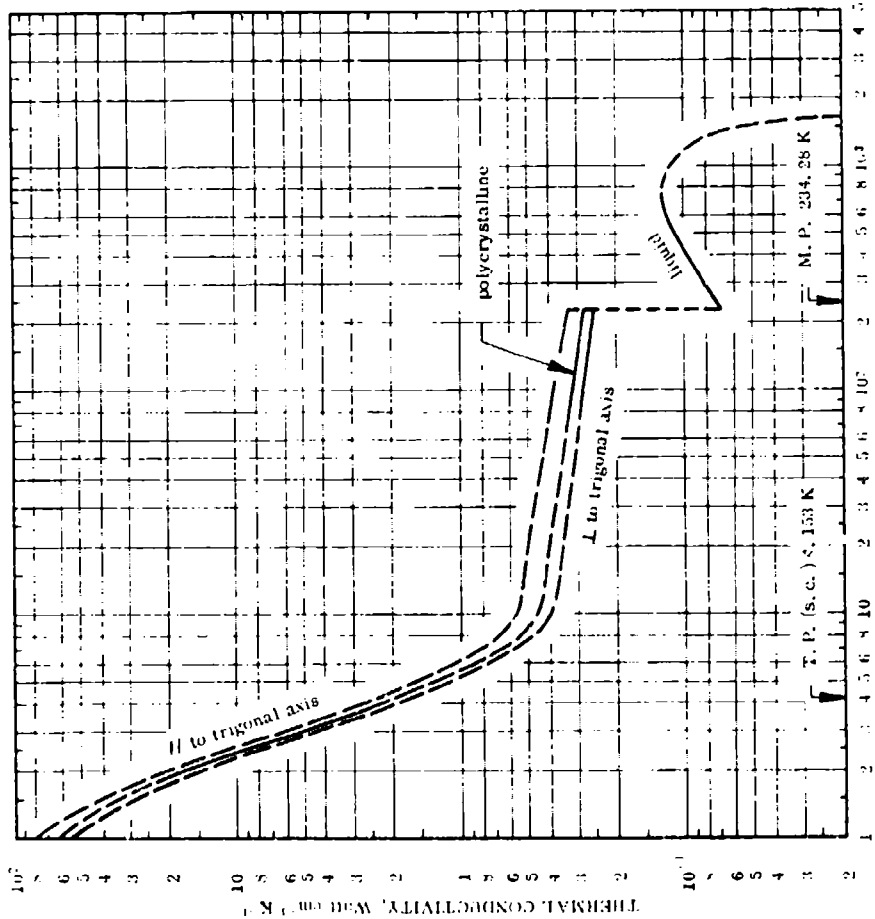
DATA TABLE NO. 11 (continued)

H (gauss)	k	H (gauss)	k	T	k	T	k	T	k	T	k	T	k
CURVE 20 (cont.)													
188	2.54	CURVE 25 (T = 1.50K)		196.2	0.330	85.55	0.373	288.7	0.042	323	0.0443	580.2	0.117
190	2.63	460	8.20	CURVE 41		CURVE 52		CURVE 60		CURVE 61 (cont.)		596.2	0.118
CURVE 21 <sup>6</sup>													
(T = 3.1K)													
247	2.61	545	7.94	196.4	0.259	86.6	0.345	288.7	0.045	323.2	0.0746	608.2	0.121
483	2.54	610	6.85	CURVE 42		CURVE 54		CURVE 61		613.2	0.122	633.2	0.123
730	2.50	653	6.49	196.8	0.341	90.6	0.334	277.7	0.0634	633.2	0.127	650.2	0.131
852	2.48	730	5.95	CURVE 43 <sup>2</sup>		CURVE 55		290.2	0.0778	676.2	0.123	700.2	0.130
974	2.42	825	5.35	CURVE 31		CURVE 32		CURVE 55		CURVE 62			
CURVE 22 <sup>2</sup>													
(T = 3.1K)													
139	0.935	197.5	0.324	86.2	0.359	567.2	0.000905	368.2	0.045	353.2	0.084	368.2	0.084
167	0.943	CURVE 32		CURVE 44		608.7	0.000984	378.2	0.053	368.2	0.087	393.2	0.092
177	0.990	455	10.6	CURVE 33		CURVE 45		635.9	0.000107	383.2	0.094	396.2	0.096
181	1.23	500	10.4	197.6	0.303	290.5	0.0843	685.3	0.000110	396.2	0.097	398.2	0.092
186	1.55	745	8.77	CURVE 34		CURVE 35		690.8	0.000114	398.2	0.099	413.2	0.101
189	2.02	825	8.77	197.1	0.312	288.2	0.0832	717.3	0.000116	408.2	0.094	416.2	0.093
193	2.38	890	8.06	CURVE 35		CURVE 36		CURVE 56		408.2	0.095	418.2	0.099
193	2.63	943	7.94	CURVE 36		CURVE 37		420.2	0.098	423.2	0.098	423.2	0.093
198	2.60	965	7.94	198.4	0.345	289.0	0.0846	303.7	0.088	426.2	0.094	432.2	0.098
CURVE 23 <sup>6</sup>													
(T = 3.1K)													
284	2.63	198.4	0.345	CURVE 37		CURVE 38		319.7	0.085	443.2	0.101	443.2	0.098
361	2.63	CURVE 35		CURVE 36		CURVE 37		323.7	0.089	450.2	0.099	443.2	0.103
500	2.62	197.3	0.340	CURVE 38		CURVE 39		329.7	0.092	450.2	0.101	446.2	0.104
727	2.59	CURVE 36		CURVE 40		CURVE 41		329.2	0.092	450.2	0.104	450.2	0.103
CURVE 24													
(T = 1.40K)													
455	8.85	85.2	0.372	CURVE 37		CURVE 38		331.2	0.095	456.2	0.106	460.2	0.100
500	8.06	CURVE 37		CURVE 39		CURVE 40		334.7	0.095	468.2	0.107	468.2	0.106
610	6.33	85.4	0.290	CURVE 39		CURVE 40		336.7	0.091	473.2	0.110	473.2	0.101
720	5.08	CURVE 38		CURVE 41		CURVE 42		341.2	0.094	478.2	0.109	478.2	0.106
825	4.20	85.4	0.407	CURVE 40		CURVE 41		342.7	0.094	483.2	0.103	483.2	0.103
852	4.07	CURVE 39		CURVE 42		CURVE 43		CURVE 57		483.2	0.106	483.2	0.107
890	3.76	80.2	0.407	CURVE 41		CURVE 42		476.2	0.090772	488.2	0.113	488.2	0.107
965	3.44	CURVE 40		CURVE 43		CURVE 44		476.2	0.090772	496.2	0.114	496.2	0.105
CURVE 25													
(T = 1.57K)													
139	0.935	197.1	0.312	CURVE 42		CURVE 43		CURVE 58		500.2	0.108	506.2	0.112
167	0.943	CURVE 42		CURVE 44		CURVE 45		543.2	0.115	513.2	0.106	513.2	0.106
177	0.990	455	10.6	CURVE 43		CURVE 44		548.2	0.116	513.2	0.112	513.2	0.112
181	1.23	500	10.4	CURVE 44		CURVE 45		566.2	0.116	566.2	0.116	566.2	0.116
186	1.55	745	8.77	CURVE 45		CURVE 46		CURVE 59		575.2	0.117	575.2	0.117
189	2.02	825	8.77	CURVE 46		CURVE 47		CURVE 60		CURVE 61 (cont.)			
193	2.38	890	8.06	CURVE 47		CURVE 48		CURVE 61		323	0.0443	323	0.0443
193	2.63	CURVE 47		CURVE 49		CURVE 50		CURVE 61		323.2	0.0746	323.2	0.0746
198	2.60	965	7.94	CURVE 48		CURVE 49		CURVE 61		323.2	0.0746	323.2	0.0746
CURVE 26													
(T = 1.57K)													
284	2.63	410	8.55	CURVE 49		CURVE 50		CURVE 61		323.2	0.0746	323.2	0.0746
361	2.63	CURVE 49		CURVE 51		CURVE 52		CURVE 61		323.2	0.0746	323.2	0.0746
500	2.62	410	8.33	CURVE 50		CURVE 51		CURVE 61		323.2	0.0746	323.2	0.0746
727	2.59	625	7.75	CURVE 51		CURVE 52		CURVE 61		323.2	0.0746	323.2	0.0746
CURVE 27													
(T = 1.57K)													
284	2.63	410	8.55	CURVE 52		CURVE 53		CURVE 61		323.2	0.0746	323.2	0.0746
361	2.63	CURVE 52		CURVE 54		CURVE 55		CURVE 61		323.2	0.0746	323.2	0.0746
500	2.62	625	7.75	CURVE 53		CURVE 54		CURVE 61		323.2	0.0746	323.2	0.0746
727	2.59	720	7.30	CURVE 54		CURVE 55		CURVE 61		323.2	0.0746	323.2	0.0746
CURVE 28													
(T = 1.40K)													
455	8.85	85.2	0.372	CURVE 55		CURVE 56		CURVE 61		323.2	0.0746	323.2	0.0746
500	8.06	CURVE 55		CURVE 57		CURVE 58		CURVE 61		323.2	0.0746	323.2	0.0746
610	6.33	85.4	0.290	CURVE 56		CURVE 57		CURVE 61		323.2	0.0746	323.2	0.0746
720	5.08	CURVE 56		CURVE 58		CURVE 59		CURVE 61		323.2	0.0746	323.2	0.0746
825	4.20	80.2	0.407	CURVE 57		CURVE 58		CURVE 61		323.2	0.0746	323.2	0.0746
852	4.07	CURVE 57		CURVE 59		CURVE 60		CURVE 61		323.2	0.0746	323.2	0.0746
890	3.76	85.4	0.407	CURVE 58		CURVE 59		CURVE 61		323.2	0.0746	323.2	0.0746
965	3.44	CURVE 58		CURVE 60		CURVE 61		CURVE 61		323.2	0.0746	323.2	0.0746

\* Not shown on plot



FIGURE AND TABLE NO. 31R RECOMMENDED THERMAL CONDUCTIVITY OF MERCURY



REMARKS  
 TEMPERATURE, K  
 The recommended values are for 99.999% pure mercury with residual electrical resistivity  $\rho_0 = 0.000236$ ,  $0.000427$ , and  $0.00072$   $\Omega$  cm, respectively, for single crystal along directions parallel and perpendicular to trigonal axis and for polycrystalline mercury (characterization by  $\rho_0$  becomes important at temperatures below about 40 K). The recommended values that are supported by experimental thermal conductivity data are thought to be accurate to within 5% of the true values near room temperature, and 5 to 10% at other temperatures.

RECOMMENDED VALUES\*

T <sub>1</sub>	Single Crystal (// to trig. axis)			Polycrystalline			T <sub>2</sub>
	k <sub>1</sub>	k <sub>2</sub>	k <sub>3</sub>	k <sub>1</sub>	k <sub>2</sub>	k <sub>3</sub>	
0	0	0	0	0	0	0	-459.7
1	(82.4) <sup>‡</sup>	(4760)	(57.2)	(65.6)	(3790)	(3790)	-457.9
2	(21.5)	(1240)	(14.9)	17.1	985	985	-456.1
3	(6.34)	(366)	(4.50)	5.05	292	292	-454.3
4	(2.84)	(164)	(1.97)	2.26	131	131	-452.5
5	(1.66)	(95.9)	(1.15)	(1.32)	(56.3)	(56.3)	-450.7
6	(1.11)	(64.1)	(0.770)	(0.882)	(44.5)	(44.5)	-448.9
7	(0.834)	(48.2)	(0.581)	(0.665)	(38.4)	(38.4)	-447.1
8	(0.691)	(39.9)	(0.481)	(0.551)	(31.8)	(31.8)	-445.3
9	(0.613)	(35.5)	(0.429)	(0.491)	(28.8)	(28.8)	-443.5
10	(0.576)	(33.3)	(0.400)	(0.460)	(26.6)	(26.6)	-441.7
11	(0.559)	(32.3)	(0.387)	(0.445)	(25.7)	(25.7)	-439.9
12	(0.547)	(31.6)	(0.382)	(0.437)	(25.2)	(25.2)	-438.1
13	(0.535)	(31.1)	(0.377)	(0.432)	(25.0)	(25.0)	-436.3
14	(0.532)	(30.7)	(0.373)	(0.427)	(24.7)	(24.7)	-434.5
15	(0.527)	(30.5)	(0.369)	(0.422)	(24.4)	(24.4)	-432.7
16	(0.522)	(30.2)	(0.366)	(0.419)	(24.2)	(24.2)	-430.9
18	(0.512)	(29.6)	(0.360)	(0.410)	(23.7)	(23.7)	-427.3
20	(0.504)	(29.1)	(0.354)	(0.404)	(23.3)	(23.3)	-425.7
25	(0.488)	(28.2)	(0.343)	(0.392)	(22.6)	(22.6)	-414.7
30	(0.474)	(27.4)	(0.334)	(0.382)	(22.1)	(22.1)	-405.7
35	(0.462)	(26.7)	(0.327)	(0.373)	(21.6)	(21.6)	-396.7
40	(0.452)	(26.1)	(0.320)	(0.365)	(21.1)	(21.1)	-387.7
45	(0.444)	(25.7)	(0.315)	(0.359)	(20.7)	(20.7)	-378.7
50	(0.437)	(25.2)	(0.311)	(0.354)	(20.5)	(20.5)	-369.7
70	(0.424)	(24.5)	(0.304)	(0.345)	(19.9)	(19.9)	-351.7
80	(0.404)	(23.3)	(0.293)	(0.337)	(19.5)	(19.5)	-333.7
90	0.396	22.9	0.288	0.330	19.1	19.1	-315.7
100	0.390	22.5	0.285	0.324	18.7	18.7	-297.7
150	0.360	20.8	0.271	0.320	18.5	18.5	-189.7
200	0.340	19.6	0.264	0.289	16.7	16.7	-99.7
234.28	0.329	19.0	0.260	0.283	16.4	16.4	-37.97

\*T<sub>1</sub> in K, k<sub>1</sub> in Watt cm<sup>-1</sup> K<sup>-1</sup>, T<sub>2</sub> in F, and k<sub>2</sub> in Btu ft<sup>-1</sup> h<sup>-1</sup> F<sup>-1</sup>. <sup>‡</sup>Values in parentheses are extrapolated or estimated.

TABLE NO. 31R (continued)

In Liquid State				
$T_1$	$k_1$	$k_2$	$T_2$	
234.28	0.0697	4.03	-37.97	
250	0.0732	4.23	-9.7	
273.2	0.0782	4.52	32.0	
300	0.0834	4.82	80.3	
350	0.0915	5.29	170.3	
400	0.0994	5.69	250.3	
500	0.110	6.36	440.3	
600	0.120	6.93	620.3	
700	0.127	7.34	800.3	
800	0.128	7.40	990.3	
900	(0.124)*	(7.16)	1160	
1000	(0.117)	(6.76)	1340	
1100	(0.108)	(6.24)	1520	
1200	(0.0984)	(5.69)	1700	
1300	(0.0872)	(5.04)	1880	
1400	(0.0732)	(4.23)	2060	
1500	(0.0559)	(3.25)	2240	
1600	(0.0345)	(1.99)	2420	
1700	(0.0094)	(0.543)	2600	
1733	(0.00045)	(0.026)	2660	

\*Values in parentheses are extrapolated or estimated.

# THERMAL CONDUCTIVITY OF MOLYBDENUM

FIGURE SHOWS ONLY 31 OF THE CURVES REPORTED IN TABLE

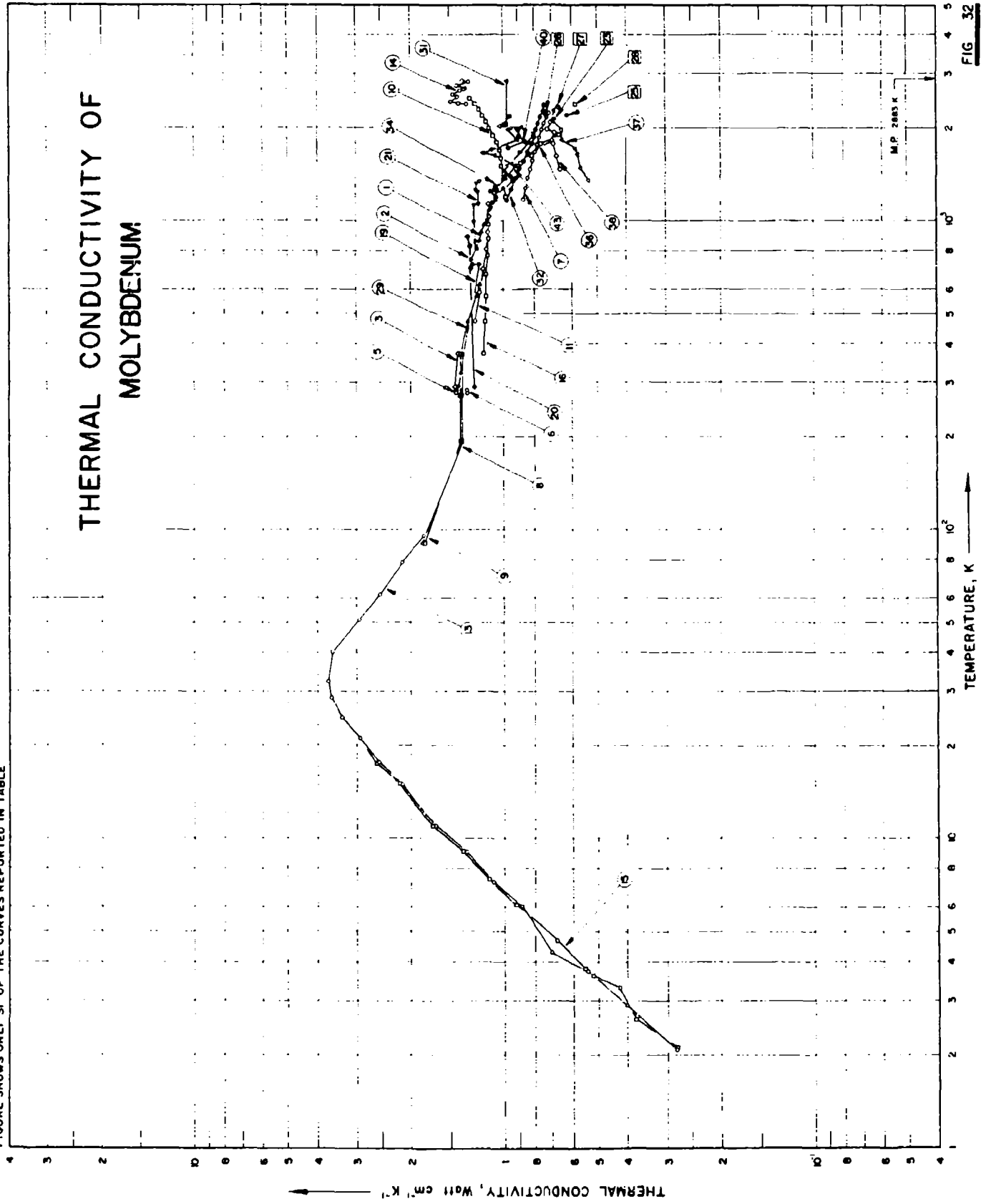


FIG. 32

SPECIFICATION TABLE NO. 32 THERMAL CONDUCTIVITY OF MOLYBDENUM

(impurity < 0.20% each; total impurities < 0.50%)

[ For Data Reported In Figure and Table No. 32 ]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition ( weight percent), Specifications and Remarks
1	100	R	1952	811-1422	5		99.9 Mo, < 0.005 Fe, and < 0.003 C; test specimen consisted of 5 stacked disks each of ~5 in. O.D., 0.75 in. I.D., and 0.50 in. thick; obtained from Fansteel Metallurgical Corp.; prepared by powder-metallurgy techniques; pressed, sintered, then rolled to finished size at just below the recrystallization temperature.
2	40	L	1956	749-1915	5		Pure; 7 in. dia x 1.5 in. thick; arc-melted.
3	8	F	1914	290, 373			Pure; 0.0520 cm dia x 45.0 cm long; density 9.933 g cm <sup>-3</sup> ; electrical resistivity reported as 5.806 and 8.516 μohm cm at 0 and 100 C, respectively.
4	78	E	1931	277, 283	2	Mo I	Pure; 0.09983 cm dia x 9.915 cm long; annealed at 220 C; electrical resistivity reported as 5.10 and 5.51 μohm cm at 0 and 18 C, respectively.
5	78	E	1931	278, 283	2	Mo I	The above specimen annealed at 900 C; electrical resistivity 5.07 μohm cm at 0 C.
6	78	E	1931	271, 282	2	Mo II	Less pure than the above sample; 0.1069 cm dia x 10.14 cm long; annealed at 220 C; electrical resistivity reported as 5.81 and 6.22 μohm cm at 0 and 18 C, respectively.
7	723	R	1965	1173-2248			0.0269 C, < 0.01 Ca, < 0.01 Cu, < 0.01 Fe, < 0.01 Mg, < 0.01 Si, 0.0006 H, 0.0006 O, and 0.00019 N; as received; specimen 1.985 ± 0.015 in. in dia, arc-cast; electrical resistivity reported as 5.6, 17.8, 32.2, 48.0, 63.2, and 73.3 μohm cm at 0, 500, 1000, 1500, 2000, and 2300 C, respectively; density 99% of theoretical value; measurements made on 7 heating and cooling cycles, mean values taken from data of 4th to 7th cycles reported.
8	79	E	1933	90-373		Mo I	99.836+ Mo, 0.05 Bi, 0.05 Cd, 0.01 Al, 0.01 Ge, 0.01 Sn, 0.01 Ti, 0.01 V, 0.01 W, 0.001 Co, 0.001 Cu, 0.001 Pt, 0.001 Rh, and trace of C; 0.09979 cm dia x 12.627 cm long; electrical resistivity reported as 0.952, 3.39, 5.25, and 7.67 μohm cm at -183.00, -78.50, 0, and 100 C, respectively.
9	79	E	1933	90-373		Mo 2	Cut from the same wire as the above specimen; 0.09980 cm dia x 9.859 cm long; electrical resistivity reported as 0.882, 3.33, 5.17, 7.56, and 10.05 μohm cm at -183.00, -78.50, 0, 100, and 217.96 C, respectively.
10	156	E	1927	1200-2500			Pure; electrical resistivity reported as 29.2, 32.2, 35.2, 38.2, 41.2, 44.3, 47.3, 50.4, 53.5, 56.6, 59.7, 62.8, 66.0, and 69.2 μohm cm at 1200, 1300, 1400, 1500, 1600, 1700, 1800, 1900, 2000, 2100, 2200, 2300, 2400, and 2500 K, respectively.
11	89	C	1956	478-1144			Pure; supplied by Climax Molybdenum Co.; 2 cm dia x 15 cm long; arc melted; density 10.24 g cm <sup>-3</sup> ; Armco iron used as comparative material.
12	996	L	1967	86-377			Spectroscopically standardized molybdenum; 5 mm dia x 10 cm long; supplied by Johnson, Matthey & Co.; electrical resistivity reported as 0.18, 0.57, 0.80, 3.20, 5.55, and 7.31 μohm cm at 4, 76, 91, 194, 297, and 374 K, respectively.



SPECIFICATION TABLE NO. 32 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
13	122	L	1955	2.1-95	± 3	JM 2131; Mo 1	99.95 pure; polycrystalline; 0.52 cm dia x 2.85 cm long; obtained from Johnson, Matthey; electrical resistivity reported as 1.41, 1.44, 1.45, 1.49, 1.51, 1.80, 2.26, and 2.84 $\mu$ ohm cm at 20.4, 25.2, 29.3, 37.2, 48.6, 58.8, 75.3, and 90.1 K, respectively.
14	255	E	1960	2384-2849			0.18 Fe, 0.073 Si, 0.04 C, 0.036 Mn, 0.005 O, and 0.01 others; arc-melted and cast; under inert gas, hot-worked, and hot rolled, polished; 0.125 in. dia x 10 in. long; obtained from Fansteel Corp.
15	97	L	1952	2.1-21	2-3	JM 2331; Mo 1	99.95 pure; 1-2 mm dia x 5 cm long; supplied by Johnson-Matthey.
16	414	L	1954	373-973	± 2		Coarse grain structure on the outside and fine grain structure in the interior; with a large number of inclusions; 1 in. bar, forged and machined.
17	503	E	1961	1068-1183			99.98 pure; 1 mm wire; obtained from Radium-Elektrizitätsgesellschaft, Wipperfurth; polished; annealed in vacuo for 12 hrs at about 1000 C; electrical resistivity reported as 24.4, 25.5, 26.9, 28.3, and 29.7 $\mu$ ohm cm at 761, 800, 850, 900, and 950 C, respectively.
18	543	C	1955	473-1173	± 5	Heat No. 990	Recrystallized at 1505 C; measured in a vacuum of $2 \times 10^{-5}$ mm Hg; Armco iron used as comparative material; data taken from smoothed curve.
19	599	E	1961	290-871			99.9 pure; received from Fansteel Metallurgical Corp; electrical resistivity 5.98 $\mu$ ohm cm at 23 C.
20	599	E	1961	290-890			2nd run of the above specimen.
21	599	E	1961	290-1325			3rd run of the above specimen.
22	1011	E	1961	1122.1727			Tubular specimen 8 mm O.D., 5 mm I.D., and 100 mm long.
23	601		1962	2129			Heated in high vacuum ( $10^{-5}$ mm Hg) by high frequency induction to 1000 to 3000 C; localized heating within 0.003 in. of the surface at current frequencies of 500000 cps; specimen 0.4923 in. in dia and 0.863 in. in length; measured with the cylindrical axis parallel to the magnetic field; run G-2.
24	601		1962	2161			The above specimen; run G-3.
25	601		1962	2200			The above specimen; run G-5.
26	601		1962	2216.5			The above specimen; run G-1.
27	601		1962	2351.5			The above specimen; run M-1.
28	601		1962	2382			The above specimen; run M-3.
29	652	L, C	1961	323-623		JM 720	Spectrographically standardized molybdenum; obtained from Johnson, Matthey and Co.; rod of about 5 mm in dia and 15 cm in length; electrical resistivity reported as 5.65, 6.25, 7.45, 9.9, 12.45, 13.75, 15.1, 17.85, 20.6, 23.3, 26, 28.7, 31.5, 34.4, 37.2, 40.1, 43, and 44.7 $\mu$ ohm cm at 20, 50, 100, 200, 300, 350, 400, 500, 600, 700, 800, 900, 1000, 1100, 1200, 1300, 1400, and 1450 C, respectively; Armco iron used as comparative material.

SPECIFICATION TABLE NO. 12 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
30	653	E	1963	1075-1640			<b>Lorenz</b> Single crystal; Lorenz function reported as 2.97, 2.76, 2.72, 2.71, 2.66, 2.57, and $2.33 \times 10^{-8} \text{V}^2/\text{K}^2$ at 1080, 1200, 1310, 1320, 1420, 1540, and 1650 K, respectively.
31	667	E	1962	1520-2085	$\pm 10$		Spectrographically pure; wire specimen 0.910 in. in dia; electrical resistivity reported as 41.3, 50.6, 58.1, and 59.1 $\mu\text{ohm cm}$ at 1550, 1830, 2040, and 2110 K, respectively; measured in a vacuum of $< 10^{-5}$ mm Hg.
32	600, 709	E	1960	1173-2473			Wire; 1 mm in dia, 30 mm long; electrical resistivity reported as 27.8, 35.6, 45.2, 56.4, and 74.8 $\mu\text{ohm cm}$ at 1173, 1473, 1773, 2073, and 2473 K, respectively; data taken from smoothed curve.
33	654	F	1965	1340-2510	$\pm 5$		99.99 Mo (by difference), $< 0.01$ Fe, and traces of other elements; 0.04 in. thick sheet; obtained from Murex Co.; sintered and hot rolled; average grain size (after test) 110 $\mu$ ; density 10.3 $\text{g cm}^{-3}$ ; thermal conductivity values calculated from measured thermal diffusivity data using the specific heat data compiled by Kubaschewski, O., and Evans, L. L. (Metallurgical Thermochimistry, Pergamon, London, 1956).
34	710	E	1938	1207-1400			Very pure; 20-mil wire; polished; aged at about 2200 K for 15 min; electrical resistivity reported as 26.4, 32.3, 37.8, 42.8, and 47.4 $\mu\text{ohm cm}$ at 1110, 1325, 1515, 1685, and 1840 K, respectively.
35	710	E	1938	1315-1647			Same as the above specimen.
36	710	E	1938	1545-1905			Same as the above specimen.
37	711	R	1965	1362-2252	6	Sample 1	99.98 <sup>+</sup> Mo, 0.005 Fe, 0.004 Si, 0.003 Ni, 0.0923 O, 0.0021 C, 0.001 V, $< 0.0005$ N, and 0.0023 H; specimen 2.118 cm in dia and 0.225 cm thick; prepared by powder metallurgy techniques; polished with No. 4/0 emery paper; average grain dia 34 $\mu$ ; density 9.104 $\text{g cm}^{-3}$ ; experiment performed in high vacuum ( $10^{-5}$ mm Hg); specimen heated by high frequency induction current; specimen axis parallel to the axis of magnetic field; data calculated from total emittance measurements using specific heat data from an empirical formula whose agreement with those of Kirilin, V.A. et al was within 2%; run No. 1.
38	711	R	1965	1476-2100	6	Sample 1	99.964 <sup>+</sup> Mo, 0.028 C, 0.0021 O, 0.002 Si, 0.001 Cu, 0.001 Fe, 0.001 V, $< 0.0005$ N, and 0.00015 H; specimen 1.905 cm in dia, 0.206 cm thick; prepared by arc-melting technique; average grain dia 706 $\mu$ ; density 10.119 $\text{g cm}^{-3}$ ; same measuring conditions and method as above; run No. 1.
39	711	R	1965	1687-2272	6	Sample 2	The above specimen; run No. 2.
40	711	R	1965	1740-2194	6	Sample 3	99.948 <sup>+</sup> Mo, 0.011 C, $< 0.01$ Fe, $< 0.01$ Si, $< 0.01$ Ti, $< 0.01$ V, 0.003 O, 0.0006 N, and 0.0602 H; specimen 1.910 cm in dia, 0.195 cm thick; prepared by arc-melting and heated to 2500 K for very long times in hydrogen such that it underwent grain growth; average grain dia 4850 $\mu$ ; density 10.163 $\text{g cm}^{-3}$ ; same measuring condition and method as above; run No. 1.
41	711	R	1965	1592-1974	6	Sample 3	The above specimen; run No. 2.

SPECIFICATION TABLE NO. 32 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
42	744	E	1966	1609-2355	7	Specimen 1	No details reported.
43	845, 844	P	1966	1140-1816			99.9 Mo, 0.01 MoO <sub>3</sub> , 0.001 Ni, 0.001 SiO <sub>2</sub> , traces <sup>MnO</sup> and CaO; cylindrical specimen 10 mm in dia and 70 mm long; density 10.2 g cm <sup>-3</sup> at room temp; electrical resistivity reported as 5.78 μohm cm at 23°C; thermal conductivity values calculated from measured data of thermal diffusivity, specific heat, and density; reported values taken from smoothed curve.



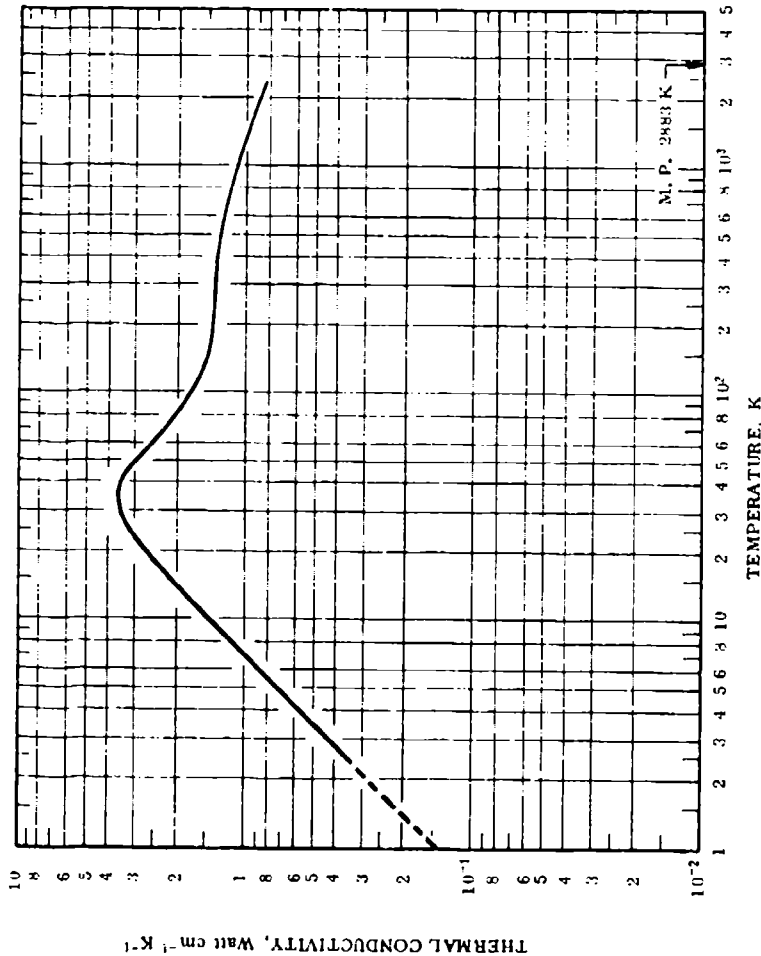
DATA TABLE NO. 32 (continued)

T	k
<u>CURVE 40 (cont.)</u>	
1880.0	0.891
1970.5	0.895
2062.5	0.992
2194.5	0.950
<u>CURVE 41<sup>e</sup></u>	
1592.0	0.854
1733.0	0.824
1889.5	0.826
1933.5	0.916
1974.0	0.929
<u>CURVE 42<sup>e</sup></u>	
1609	0.960
1765	0.950
1875	0.920
2005	0.940
2105	0.960
2275	0.960
2355	0.880
<u>CURVE 43</u>	
1140	1.125 <sup>e</sup>
1253	1.061
1380	0.983
1500	0.913
1635	0.824
1816	0.704

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Not shown on plot

FIGURE AND TABLE NO. 32R RECOMMENDED THERMAL CONDUCTIVITY OF MOLYBDENUM



RECOMMENDED VALUES*										
$T_1$	$k_1$	$k_2$	$T_2$	$T_1$	$k_1$	$k_2$	$T_2$	$T_1$	$k_1$	$T_2$
0	0	0	-459.7	500	1.30	75.1	440.3	500	1.30	440.3
1	(0.146)†	(3.44)	-457.9	600	1.26	72.8	620.3	600	1.26	620.3
2	(0.292)	(16.9)	-456.1	700	1.22	70.5	800.3	700	1.22	800.3
3	0.438	25.3	-454.3	800	1.18	68.2	990.3	800	1.18	990.3
4	0.584	33.7	-452.5	900	1.15	66.4	1160	900	1.15	1160
5	0.730	42.2	-450.7	1000	1.12	64.7	1340	1000	1.12	1340
6	0.876	50.6	-448.9	1100	1.08	62.4	1520	1100	1.08	1520
7	1.02	58.9	-447.1	1200	1.05	60.7	1700	1200	1.05	1700
8	1.17	67.6	-445.3	1300	1.02	58.9	1890	1300	1.02	1890
9	1.31	75.7	-443.5	1400	0.996	57.5	2060	1400	0.996	2060
10	1.45	83.8	-441.7	1500	0.970	56.0	2240	1500	0.970	2240
11	1.60	92.4	-439.9	1600	0.946	54.7	2420	1600	0.946	2420
12	1.74	101	-438.1	1703	0.925	53.4	2600	1703	0.925	2600
13	1.88	109	-436.3	1800	0.907	52.4	2780	1800	0.907	2780
14	2.01	116	-434.5	1900	0.893	51.6	2960	1900	0.893	2960
15	2.15	124	-432.7	2000	0.880	50.8	3140	2000	0.880	3140
16	2.28	132	-430.9	2200	0.858	49.6	3500	2200	0.858	3500
18	2.53	146	-427.3	2400	0.840	48.5	3860	2400	0.840	3860
20	2.77	160	-423.7	2600	0.825	47.7	4220	2600	0.825	4220
25	3.25	188	-414.7	2800	0.813	47.0	4580	2800	0.813	4580
30	3.55	205	-405.7							
35	3.62	209	-396.7							
40	3.51	203	-387.7							
45	3.26	188	-378.7							
50	3.00	173	-369.7							
60	2.69	150	-351.7							
70	2.30	133	-333.7							
80	2.09	121	-315.7							
90	1.92	111	-297.7							
100	1.79	103	-279.7							
150	1.49	86.1	-189.7							
200	1.43	82.6	-99.7							
250	1.40	80.9	-9.7							
273.2	1.39	80.3	32.0							
300	1.38	79.7	80.3							
350	1.36	78.6	170.3							
400	1.34	77.4	260.3							

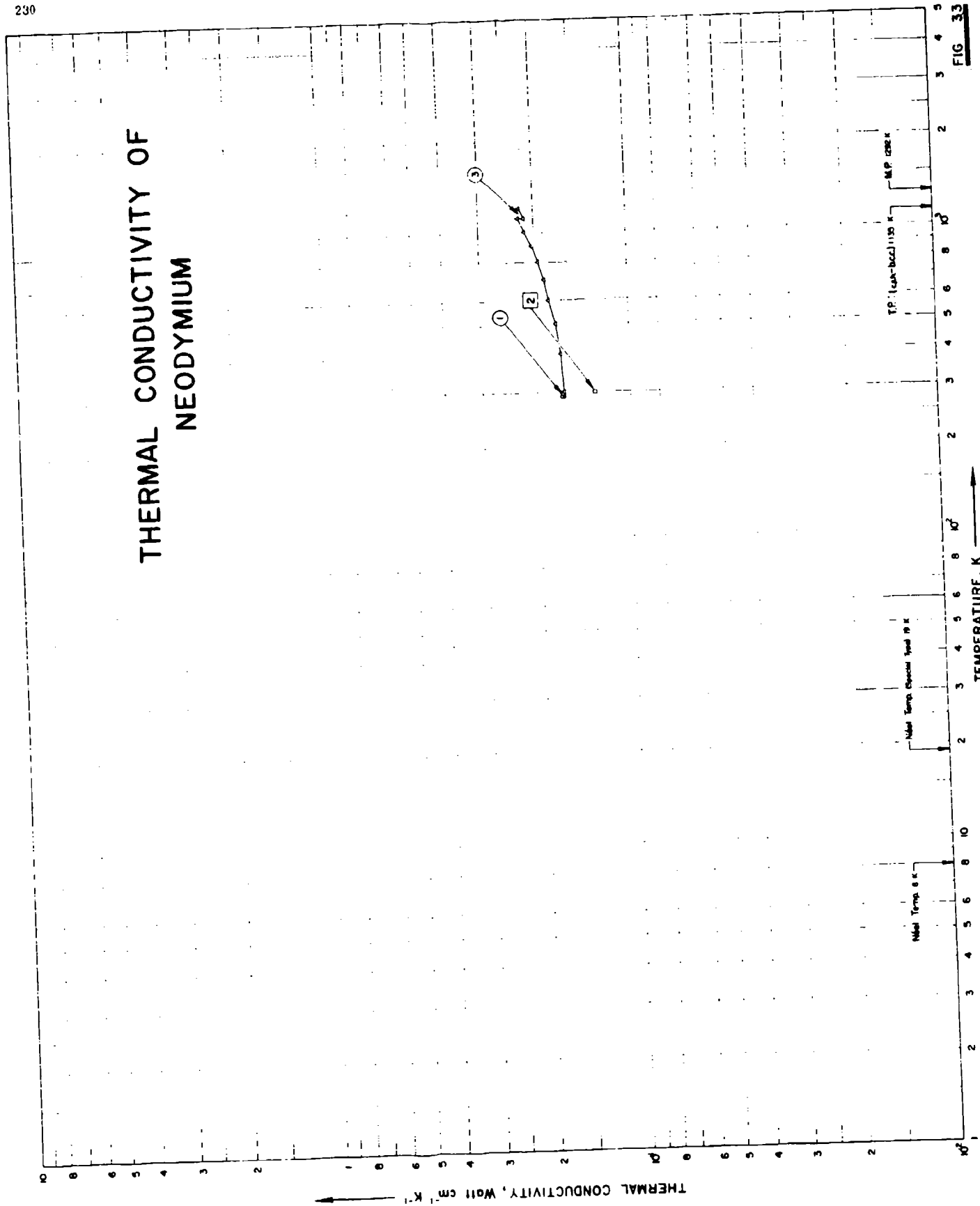
REMARKS

The recommended values are for well-annealed 99.95% pure molybdenum with residual electrical resistivity  $\rho_0 = 0.167 \mu\Omega \text{ cm}$  (characterization by  $\rho_0$  becomes important at temperatures below about 250 K). The values below 1.5  $T_m$  are calculated to fit the experimental data by using  $n = 2.60$ ,  $\alpha' = 7.76 \times 10^{-6}$ , and  $\beta = 6.85$ . The recommended values that are supported by experimental thermal conductivity data are thought to be accurate to within 4% of the true values near room temperature, and 4 to 10% at other temperatures.

\*  $T_1$  in K,  $k_1$  in Watt  $\text{cm}^{-1} \text{K}^{-1}$ ,  $T_2$  in F, and  $k_2$  in Btu  $\text{hr}^{-1} \text{ft}^{-1} \text{F}^{-1}$ .

† Values in parentheses are extrapolated.

# THERMAL CONDUCTIVITY OF NEODYMIUM



## SPECIFICATION TABLE NO. 33: THERMAL CONDUCTIVITY OF NEODYMIUM

(Impurity  $\pm$  0.20% each, total impurities  $\pm$  0.50%)

[ For Data Reported in Figure and Table No. 33 ]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition ( weight percent), Specifications and Remarks
1	777	C	1965	291.2	$\pm$ 3.0		High purity polycrystalline, specimen 0.25 in. in diameter and 0.25 in. long; supplied by Johnson Matthey Co.; electrical resistivity $65 \mu$ ohm cm at 18 C.; Monel metal used as comparative material; measurements made with 2 different thermal comparators.
2	811		1954	301	$\pm$ 10.0		No details reported.
3	999	-	1966	300-1200			Estimated values given as the sum of electronic thermal conductivity and the lattice thermal conductivity where electronic thermal conductivity values calculated from the theoretical Lorenz number $L_0 = 2.443 \times 10^{-8} \text{ V}^2 \text{ K}^{-2}$ and the estimated electrical resistivity reported as 65, 76, 86, 95, 104, 112, 118, 123, 125, 130 ( $\alpha$ ), 136 ( $\beta$ ), and 136 $\mu$ ohm cm at 300, 400, 500, 600, 700, 800, 900, 1000, 1100, 1120, 1120, and 1200 K, respectively; whereas lattice thermal conductivity values calculated from the empirical equation $k_L = 15.6 T^{-1}$ .



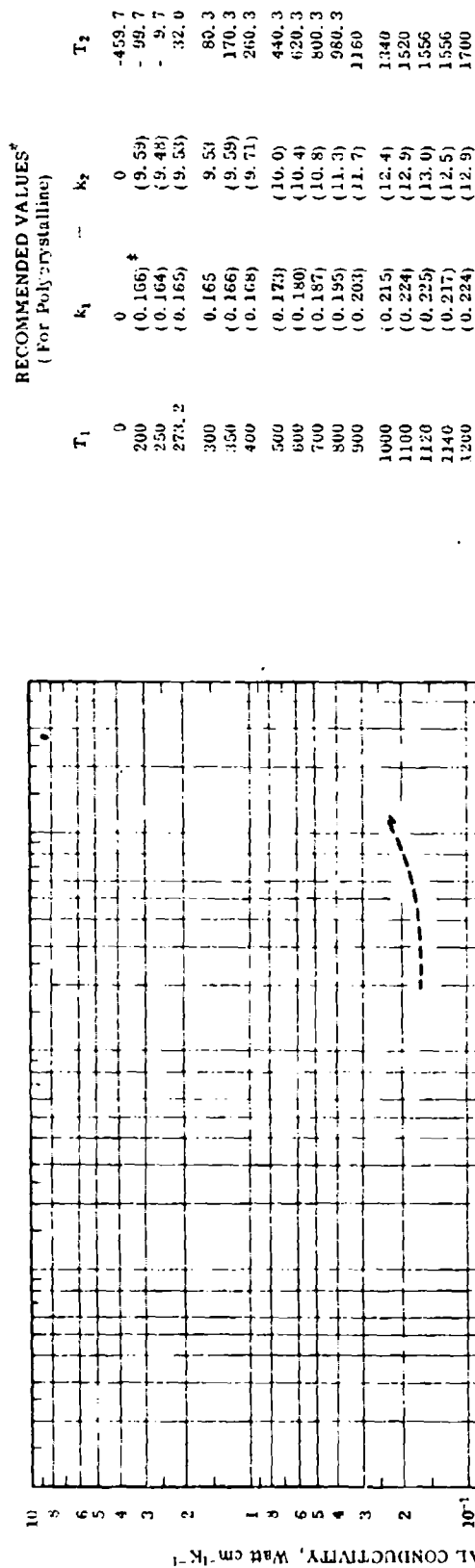
## DATA TABLE NO. 3: THERMAL CONDUCTIVITY OF NEODYMIUM

(Impurity &lt;math&gt;9.29\%&lt;/math&gt; each; total impurities &lt;math&gt;0.36\%&lt;/math&gt;)

T, Temperature,  $^{\circ}\text{K}$ .  $k$ , Thermal Conductivity,  $\text{k. Watt cm}^{-1}\text{K}^{-1}$ .

T	k
<u>CURVE 1</u>	
291.2	0.165
291.2	0.166
<u>CURVE 2</u>	
301	0.130
<u>CURVE 3</u>	
300	0.165
400	0.158
500	0.173
600	0.181
700	0.187
800	0.195
900	0.203
1000	0.215
1100	0.224
1120	0.225
1120	0.215
1200	0.224

FIGURE AND TABLE NO. 33R RECOMMENDED-THERMAL CONDUCTIVITY OF NEODYMIUM



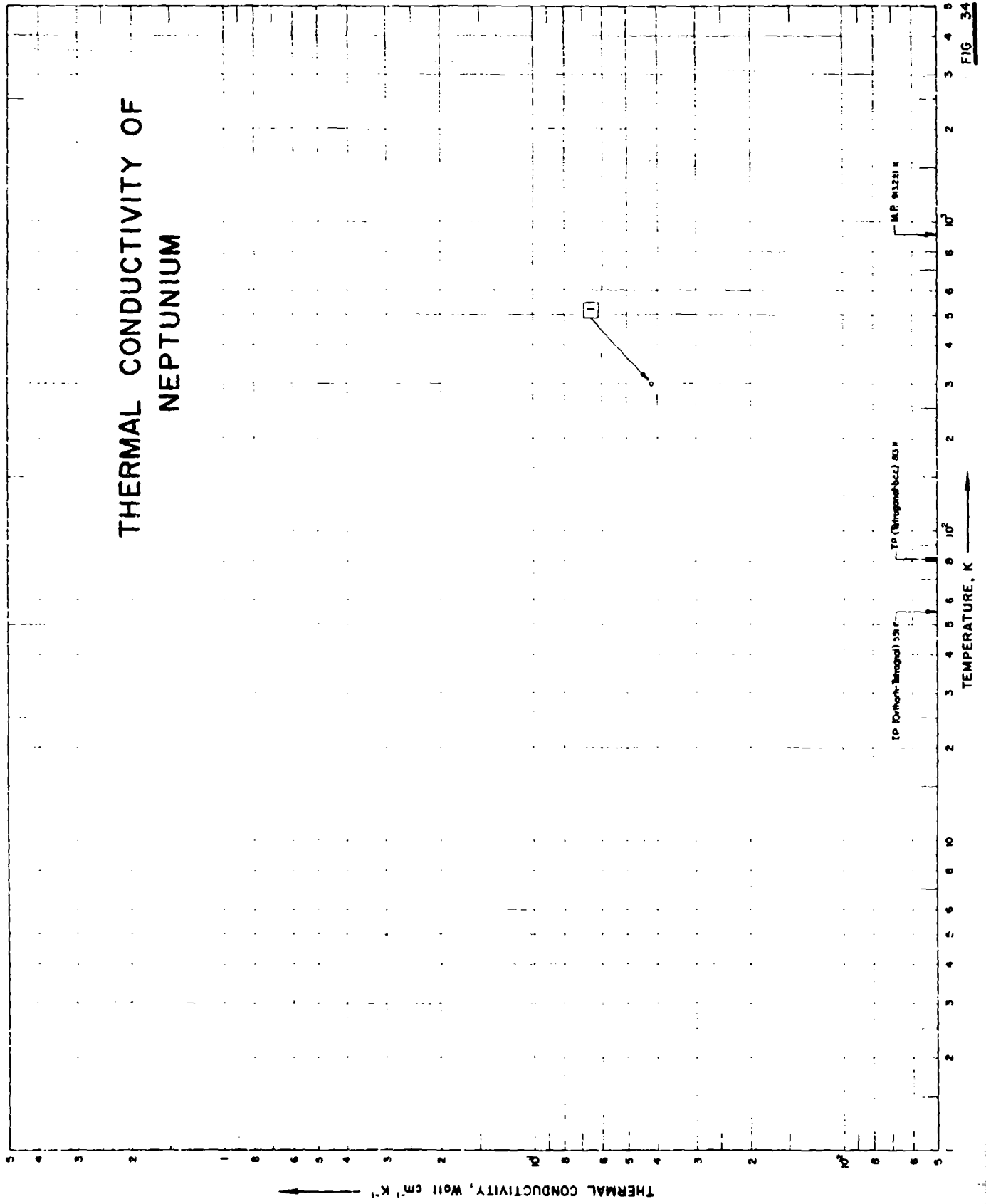
REMARKS

The recommended values are for high-purity neodymium. The recommended values are thought to be accurate to within 5% of the true values near room temperature.

\* T<sub>1</sub> in K, k<sub>1</sub> in Watt cm<sup>-1</sup> K<sup>-1</sup>, T<sub>2</sub> in F, and k<sub>2</sub> in Btu (h<sup>-1</sup> ft<sup>-1</sup>) F<sup>-1</sup>.

‡ Values in parentheses are estimated.

# THERMAL CONDUCTIVITY OF NEPTUNIUM



## SPECIFICATION TABLE NO. 34 THERMAL CONDUCTIVITY OF NEPTUNIUM

(Impurity &lt; 0.20% each; total impurities &lt; 0.50%)

[ For Data Reported in Figure and Table No. 34 ]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
1	819		1961	300			Neptunium in $\alpha$ -phase; data determined from resistivity measurements using Kanalluk method; electrical resistivity reported as 116.4, 116.1, 117.7, 117.8, 119.1, 119.3, 120.5, 120.7, 120.9, 120.8, and 121.3 $\mu\text{ohm cm}$ at 310, 314, 334, 347, 370, 373, 425, 433, 472, 512, and 538 C, respectively.

## DATA TABLE NO. 34 THERMAL CONDUCTIVITY OF NEPTUNIUM

(Impurity &lt; 0.24% each; total impurities &lt; 0.50%)

[Temperature, T, K. Thermal Conductivity,  $k$ , Watt  $\text{cm}^{-1}\text{K}^{-1}$ ]

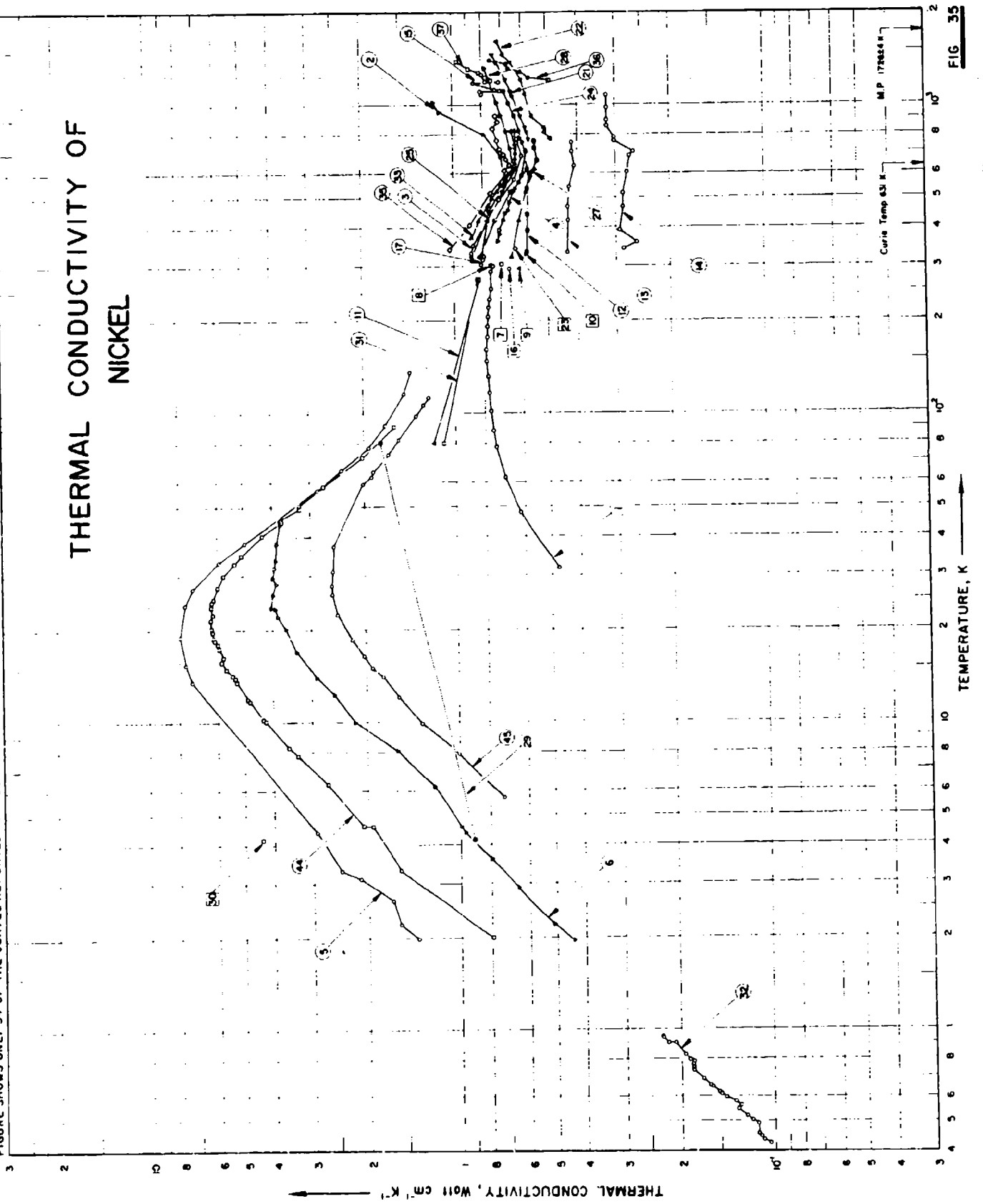
T k

CURVE 1

300 0.0418

# THERMAL CONDUCTIVITY OF NICKEL

FIGURE SHOWS ONLY 34 OF THE CURVES REPORTED IN TABLE



## SPECIFICATION TABLE NO. 35 THERMAL CONDUCTIVITY OF NICKEL

(Impurity &lt; 0.20% each; total impurities &lt; 0.50%)

(For Data Reported in Figure and Table No. 35)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)	Specifications and Remarks
1	114	L	1950	32-300	1.2-2.0	"L" Nickel	Commercially pure; 0.5 in. dia x 20 in. long; supplied by International Nickel Co.	
2	124	P	1950	327-1016			Pure nickel, electrolyzed from Mond anodes; wire, about 0.2 cm in dia; vacuum melted under a pressure of 0.3 mm Hg using an Arsen furnace and an aluminum crucible; chill cast, forged, and cold drawn to the above dimensions, annealed twice at about 750 C for several hrs; electrical conductivity reported as 9.60, 5.95, 4.10, 3.03, 2.74, 2.47, and $2.32 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 54, 179, 285, 407, 530, 676, and 743 C, respectively; density $8.74 \text{ g cm}^{-3}$ ; thermal conductivity values calculated from measured data of thermal diffusivity, specific heat, and density.	
3	129	C	1933	330-775	5	N <sub>1</sub>	99.94 Ni, 0.03 Fe, 0.016 Co, 0.006 Cu, 0.006 Si, 0.005 C, and 0.004 S; specimen 2 cm in dia and 15 cm long; melted in Arsen furnace and furnace cooled; lead used as comparative material, reference value taken from International Critical Tables <del>Vol. V, p. 221</del> (Vol. V, p. 221).	
4	101	L	1955	363-780	3		99.65 pure (by difference), 0.094 Si, 0.082 Cu, 0.056 Fe, 0.027 C, 0.025 Co, 0.008 S, and 0.007 Al; specimen 7.938 in. long and 0.787 in. in dia; prepared in a zircon crucible from high purity electrolytic nickel shot, hot rolled at 1000 C to a bar 1 in. square, machined and ground to size; annealed for 45 min at 1000 C in hydrogen atmosphere.	
5	83	L	1956	2.0-136		JM 4497	99.99 Ni, traces of Al, Ca, Cu, Si, Ag, and very faint traces of Li, Mg, and Na; material obtained from Johnson Matthey Co.; specimen 2 mm in dia; annealed for 4 hrs in vacuo at 750 C; electrical resistivity $7.22 \mu\text{ohm cm}$ at 293 K; residual electrical resistivity $9.0347 \mu\text{ohm cm}$ .	
6	122	L	1955	2.0-44	3	JM 4884, Ni 1	99.997 pure; polycrystalline; specimen 2.92 cm long, 0.305 cm in dia; obtained from Johnson Matthey Co. (JM 4884); annealed at 1150 C for several hrs in vacuo; electrical resistivity ratio $\rho(293\text{K})/\rho(20\text{K}) = 80.9$ .	
7	166	P	1928	305.2		R-12	Wire about 35 cm long, 0.32 cm in dia; electrical conductivity $9.66 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$ at approx. 32 C; thermal conductivity value calculated from measured data of thermal diffusivity and specific heat.	
8	238	F	1927	303.2			0.10 Fe, 0.037 C, 0.019 S, 0.013 Cu, 0.006 Si, traces of Al, Co, Mn, and P; specimen 20 cm long, 5 mm in dia; obtained from Mond and Co.; cast and machined, annealed for 40 min at 800 C; electrical resistivity $8.58 \mu\text{ohm cm}$ at 30 C.	
9	499	P	1937	298.2	0.66		99.98 pure; annealed in hydrogen at 870 C; density $8.79 \text{ g cm}^{-3}$ ; electrical resistivity $7.21 \mu\text{ohm cm}$ at 22 C; thermal conductivity value calculated from measured data of thermal diffusivity, specific heat, and density.	
10	230	L	1925	329.2		Electrolytic nickel	99.75-99.85 pure; supplied by International Nickel Co. of America; electrical conductivity $8.24 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$ .	
11	496	C	1940	80, 273	<1		Specimen 60 mm long, 4 mm in dia; copper used as comparative material.	
12	500	C	1954	333-763			Extremely pure; specimen 1 in. cube; supplied by the Vacuum Metal Corp; vacuum cast.	
13	300	C	1954	333-753			Similar to the above specimen but with cylindrical pores of 0.246 cm in dia; porosity 9.8%.	

SPECIFICATION TABLE NO. 35 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
14	500	C	1954	343-1073		Nickel O	Similar to the above specimen but porosity 19.5%.
15	503	E	1961	1093-1263			99.95 pure; wire 1 mm in dia.; vacuum melted and cast; polished; annealed for 12 hrs at 1000 C.
16	504	P	1961	295.2	± 5		1.26 cm dia x 0.100 cm thick; thermal conductivity value calculated from measured data of thermal diffusivity and specific heat, and density value taken from Smithsonian Physical Tables (9th ed., 1954).
17	618	C	1960	303-323	± 3		Specimen 20 mm in dia and 18 mm long; steel used as comparative material.
18	618	C	1960	303-317	± 3		The above specimen using pure Ni as comparative material.
19	618	C	1960	302-320	± 3		The above specimen using yellow brass as comparative material.
20	618	C	1960	305-321	± 3		The above specimen using Al as comparative material.
21	594	R	1961	778-1462		"1," nickel	Specimen consisted of 5 vertically stacked hollow cylinders, each 2.625 in. O. D. and 1 in. high, and having a 0.25 in. bore concentric with the axis.
22	40	L	1956	778-1615	5	"A," nickel	Disk, 7 in. in dia and 1.5 in. thick; density 8.844 g cm <sup>-3</sup> .
23	276	C	1953	343.2	3	"A," nickel	Cylinder 1.75 in. long and 0.22 in. in dia.; density 8.8 g cm <sup>-3</sup> ; Armco iron used as comparative material.
24	675	L, C	1965	323-1123		Electrolytic nickel; Sample 1	0.03 Fe, 0.01 each of Al, Cr, Co, Cu, Mg, Mn, Mo, Si, Sn, Ti, Zn, and Zr, 0.095 Pb, and 0.002 B; supplied by The Castner Kellner Alkali Co.; tube of 1.272 cm I. D., 1.908 cm O. D., and 20 cm long; density 8.61 g cm <sup>-3</sup> ; electrical resistivity reported as 7.1, 8.3, 13.0, 19.4, 28.0, 32.8, 36.1, 39.3, 42.4, and 45.2 μohm cm at 293, 323, 423, 523, 623, 723, 823, 923, 1023, and 1123 K, respectively; Armco iron used as comparative material; data taken from smoothed curve.
25	675	L, C	1965	323-823		Electrolytic nickel; Sample 2	Very high purity; supplied by the National Engineering Lab.; tube with 0.634 cm I. D., 2.801 cm O. D., and 19 cm long; density 8.90 g cm <sup>-3</sup> ; electrical resistivity reported as 10.6, 14.5, 23.7, and 33.2 μohm cm at 100, 260, 290, and 500 C, respectively; Armco iron used as comparative material; data taken from smoothed curve.
26	675	L, C	1965	323-823		Electrolytic nickel; Sample 3	99.5 ± 0.1 Ni, 0.1-0.2 Co, 0.1-0.2 Si, 0.04 Fe, 0.03 Mg, and 0.01 Cr; supplied by the Atomic Energy Research Establishment in the form of 3 tubes of 1.589 cm O. D., 1.538 cm I. D., and about 43 cm long. 32 strips each 0.35 cm wide and 14 cm long were cut from the tubes and pressed together to form a compact specimen; density 8.9 g cm <sup>-3</sup> ; electrical resistivity reported as 8.3, 9.6, 14.3, 20.6, 29.7, 34.1, and 37.3 μohm cm at 293, 324, 423, 523, 623, 723, and 923 K, respectively; Armco iron used as comparative material; data taken from smoothed curve.
27	675	L, C	1965	323-623		Electrolytic nickel; Sample 4	Commercial nickel; rod 2.54 cm in dia. about 20 cm long; supplied by the Explosives Research and Development Establishment; electrical resistivity reported as 10.1, 11.3, 16.3, 22.8, and 31.5 μohm cm at 293, 323, 423, 523, and 623 K, respectively; Armco iron used as comparative material; data taken from smoothed curve.



SPECIFICATION TABLE NO. 35 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
28	675	L, C	1965	323-1323		Electrolytic nickel; Sample 5	High spectrographic purity; very small impurities of Al, Ca, Cu, Li, Si, Ag, Mg, and Na; supplied by Johnson Matthey Co. (Laboratory No. 4457); rod 0.5 cm in dia and 15 cm long; density 8.91 g cm <sup>-3</sup> ; electrical resistivity reported as 7.1, 8.3, 13.1, 19.4, 28.3, 33.2, 36.4, 39.2, 42.1, 44.7, 47.5, and 49.8 μohm cm at 293, 323, 423, 523, 623, 723, 823, 923, 1023, 1123, 1223, and 1323 K, respectively; Armco iron used as comparative material; data taken from smoothed curve.
29	716	L	1962	4.2, 81		Ni 5011 (I)	Specimen 0.15 cm in dia turned from a cylindrical sample 5.2 cm long; supplied by Johnson Matthey Co.; annealed for 4 hrs at 1273 K in vacuum of 10 <sup>-4</sup> mm Hg, then furnace cooled at a rate of 150 K per hr; electrical resistivity reported as 0.11, 0.676, and 7.16 μohm cm at 4.18, 80.5, and 292 K, respectively; electrical resistivity ratio $\rho(273K)/\rho(4.2K) = 60$ .
30	716	L	1962	4.18	14	Ni 5011 (II)	Specimen 0.19 cm in dia drawn from a cylindrical sample 5.0 cm long; supplied by Johnson Matthey Co.; annealed for 10 hrs at 1573 K in hydrogen and left at 1573 K in a vacuum of 10 <sup>-2</sup> mm Hg for 2 hrs; electrical resistivity reported as 0.0213, 0.60, and 6.35 μohm cm at 4.18, 80.5, and 273.15 K, respectively; $\rho(273K)/\rho(4.2K) = 298$ .
31	3 <sup>a</sup>	L	1927	80, 273		Electrolytic nickel	Electrical conductivity reported as 90.2 and 13.05 x 10 <sup>4</sup> ohm <sup>-1</sup> cm <sup>-1</sup> at 80 and 273 K, respectively.
32	736	L	1965	0.42-0.95	1		Pure.
33	737, 877	E	1965	373-773	± 2	No. 1	99.87 Ni + Co; tube: 8.51 mm O. D. and 8.025 mm I. D.; electrical resistivity reported as 7.90, 9.30, 11.50, 17.24, 24.70, 31.84, and 35.36 μohm cm at 20, 50, 100, 200, 300, 400, and 500 C, respectively.
34	737	E	1965	373-748	± 2	No. 2	Tube 12.96 mm O. D. and 11.025 mm I. D.; electrical resistivity reported as 11.60, 17.29, 24.74, 32.01, and 33.98 μohm cm at 100, 200, 300, 400, and 475 C, respectively.
35	743	E	1964	340-920			99.999 pure; specimen 30 cm long and 0.3 cm in dia, annealed in vacuum for 48 hrs at 1173 K; electrical resistivity reported as 8.0, 11.5, 17.0, 24.5, 29.0, 35.0, 38.5, 42.0, 44.0, and 46.0 μohm cm at 40, 105, 210, 305, 375, 485, 590, 720, 780, and 900 C, respectively.
36	846	E	1967	1201-1393			99.95 pure; 14 cm x 1 cm x 0.05 cm; obtained from Johnson, Matthey and Co., London; data obtained without heating the ends of the specimen.
37	846	E	1967	1202-1396			99.95 pure; 14 cm x 1 cm x 0.05 cm; obtained from Johnson, Matthey and Co., London; measuring technique improved by heating the ends of the specimen.
38	843	P	1966	298.2			Spherical granular specimen supplied by Linde Co. contained in a 0.75 in. dia x 2 in. long cylindrical cell; mesh size -230 +325; thermal conductivity measured by the transient line source method; measured in Freon-12 under a pressure of ~100 psig.
39	843	P	1966	298.2			Similar to above; measured in argon under a pressure of ~100 psig.

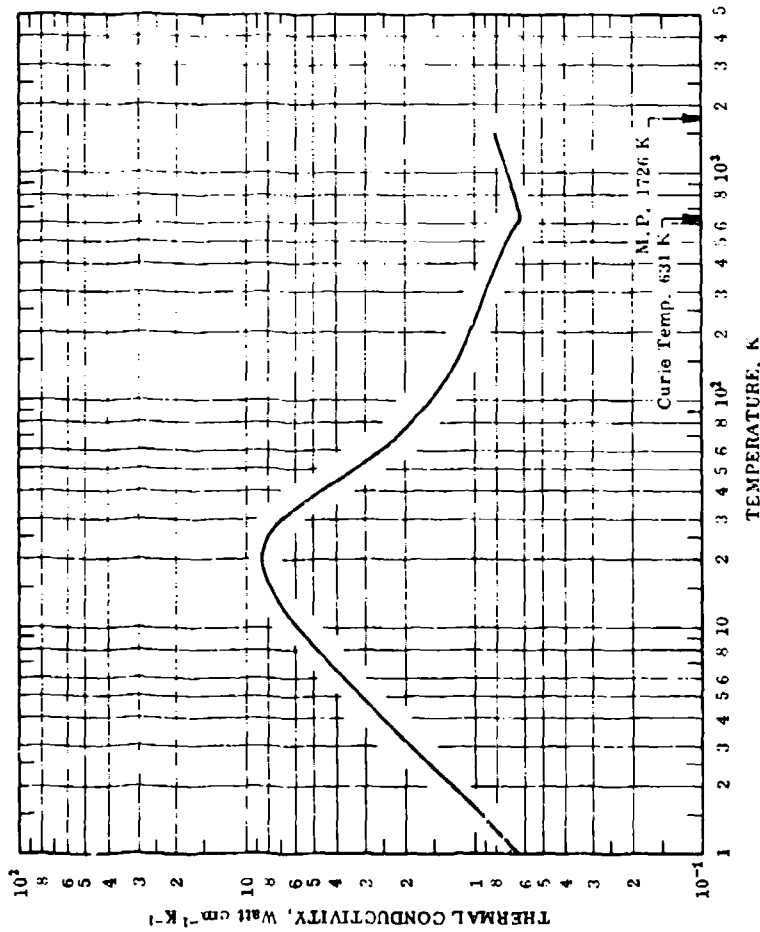
SPECIFICATION TABLE NO. 35 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)	Specifications and Remarks
40	843	P	1966	298.2				Similar to above; measured in nitrogen under a pressure of ~100 psig.
41	843	P	1966	298.2				Similar to above; measured in methane under a pressure of ~100 psig.
42	843	P	1966	298.2				Similar to above; measured in helium under a pressure of ~100 psig.
43	843	P	1966	298.2				Similar to above; measured in hydrogen under a pressure of ~100 psig.
44	917	E	1965	2.0-90	0.5-5	A		0.0016 (impurities (mostly Fe and Si); polycrystalline; 2 mm in dia; obtained from Johnson Matthey and Co.; annealed for 12 hrs at 850 C; electrical resistivity reported as 0.0005998, 0.001197, 0.003741, 0.005248, 0.009616, 0.01675, 0.04624, and 0.0719 $\mu\text{ohm cm}$ at 5, 8, 1, 13, 4, 16, 1, 20, 1, 24, 7, 34, 4, and 40.0 K, respectively.
45	917	E	1965	5.7-114	0.5-5	B		0.13 Cu; specimen 4 mm in dia; supplied by Johnson Matthey and Co.; chill cast from J. M. 890 Ni and J. M. 30 Cu; annealed for 12 hrs at 850 C.
46	186	P	1928	305.2		A nickel		0.25 cm dia x 35 cm long; density 8.90 $\text{g cm}^{-3}$ ; thermal conductivity value calculated from measured data of thermal diffusivity, specific heat, and density.





FIGURE AND TABLE NO. 35R RECOMMENDED THERMAL CONDUCTIVITY OF NICKEL



## REMARKS

The recommended values are for well-annealed 99.99% pure nickel with residual electrical resistivity  $\rho_0 = 0.0384 \mu\Omega \text{ cm}$  (characterization by  $\rho_0$  becomes important below room temperature). The values below  $1.5 T_m$  are calculated to fit the experimental data by using  $n = 2.00$ ,  $\alpha' = 9.57 \times 10^{-4}$ , and  $\beta = 1.57$ . The recommended values that are supported by experimental thermal conductivity data are thought to be accurate to within 5% of the true values near room temperature, and 5 to 10% at other temperatures.

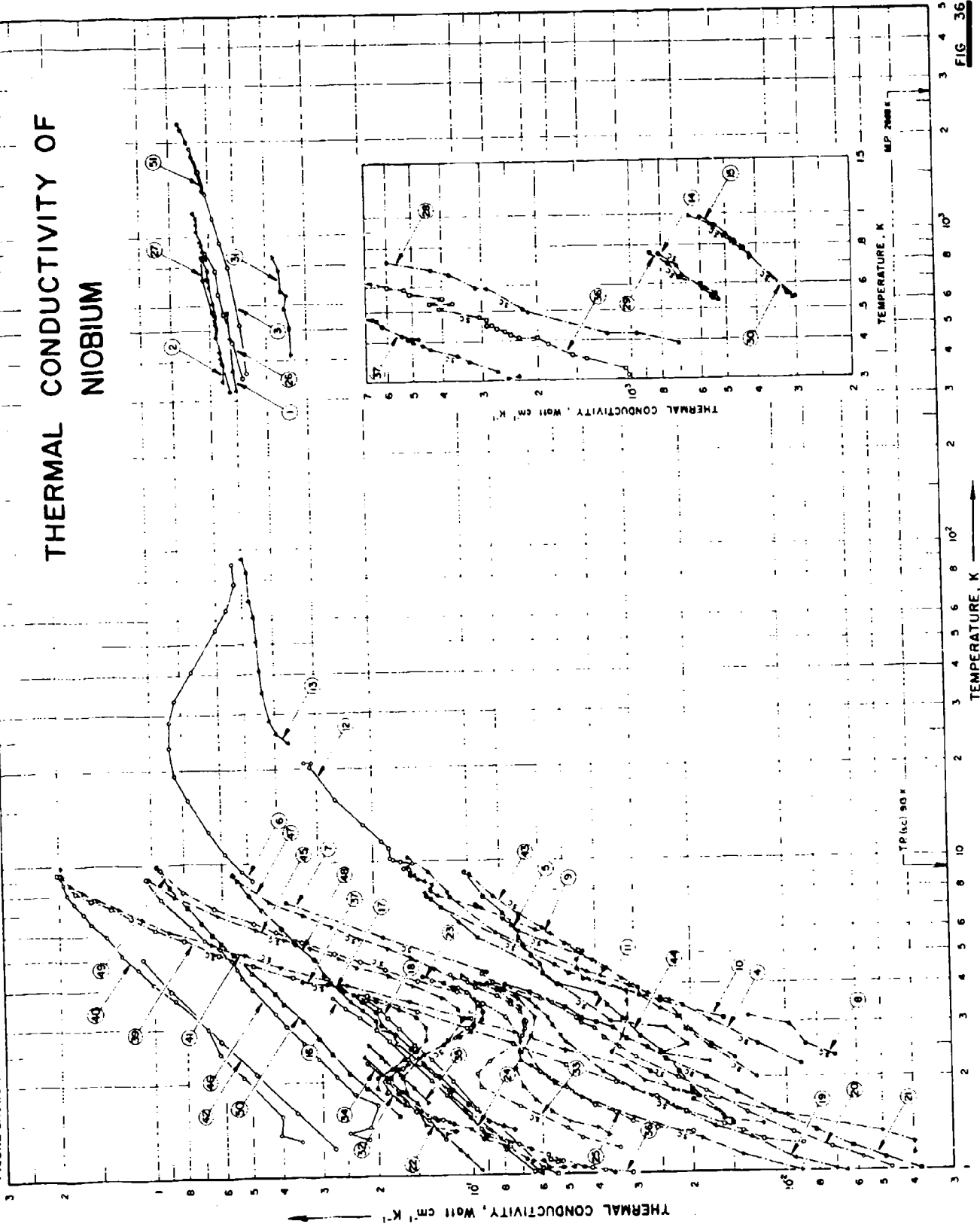
		RECOMMENDED VALUES*						
		$T_1$	$k_1$	$T_2$	$k_2$	$T_1$	$k_2$	$T_2$
0	0	600	0.655	-459.7	37.8	620.3		
1	(0.64) †	631	0.638	-457.9	36.9	676.1		
2	1.27	700	0.653	-456.1	37.7	800.3		
3	1.91	800	0.674	-454.3	38.9	980.3		
4	2.54	900	0.696	-452.5	40.2	1160		
5	3.16	1000	0.718	-450.7	41.5	1340		
6	3.77	1100	0.739	-448.9	42.7	1520		
7	4.36	1200	0.761	-447.1	44.0	1700		
8	4.94	1300	0.782	-445.3	45.2	1880		
9	5.49	1400	0.804	-443.5	46.5	2060		
10	6.00	1500	0.825	-441.7	47.7	2240		
11	6.48			-439.9				
12	6.91			-438.1				
13	7.30			-436.3				
14	7.64			-434.5				
15	7.92			-432.7				
16	8.15			-430.9				
18	8.45			-427.3				
20	8.56			-423.7				
25	8.15			-414.7				
30	6.95			-405.7				
35	5.62			-396.7				
40	4.63			-387.7				
45	3.91			-378.7				
50	3.36			-369.7				
60	2.63			-351.7				
70	2.21			-333.7				
80	1.93			-315.7				
90	1.72			-297.7				
100	1.58			-279.7				
150	1.21			-189.7				
200	1.06			-99.7				
250	0.97			-				
273.2	0.94			32.0				
300	0.905			60.3				
350	0.850			170.3				
400	0.801			260.3				
500	0.721			440.3				

\*  $T_1$  in K,  $k_1$  in Watt  $\text{cm}^{-1} \text{K}^{-1}$ ,  $T_2$  in F, and  $k_2$  in Btu  $\text{hr}^{-1} \text{ft}^{-1} \text{F}^{-1}$ .

† Values in parentheses are extrapolated.

# THERMAL CONDUCTIVITY OF NIOBIUM

FIGURE SHOWS ONLY 49 OF THE CURVES REPORTED IN TABLE



## SPECIFICATION TABLE NO. 36 THERMAL CONDUCTIVITY OF NIOBIUM

(Impurity - 0.20% each; total impurities - 0.50%)

For Data Reported in Figure and Table No. 36

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)	Specifications and Remarks
1	652	C	1961	323-573			0.1 Ta, 0.915 Ti, 0.01 C, 0.01 Fe, 0.01 N, 0.01 O, and 0.01 Si; ~6 mm dia x 10 cm long; manufactured by Murex Ltd; sintered above 2000 C and cold swaged; electrical resistivity reported as 15.0, 16.5, 18.7, 23.2, and 27.7 $\mu\text{ohm cm}$ at 233, 323, 373, 473, and 573 K, respectively; Armco iron used as comparative material.	
2	724	P	1965	345-1195			99.95% Nb, 0.011 O, 0.005 C, 0.0027 N, and 0.0006 H; specimen 0.25 in. in dia and 2 in. long; obtained from Kawecki Chemical Co.; refined by electron beam melting, annealed and machined to size; density 8.61 g cm <sup>-3</sup> ; electrical resistivity reported as 15.2, 24.2, 39.1, and 49.3 $\mu\text{ohm cm}$ at 300, 600, 900, and 1200 K, respectively; thermal conductivity values calculated from measured data of thermal diffusivity and specific heat data taken from Jaeger, F. M. and Veenstra, W. A. (Rec. Trav. Chim., 53: 677, 1934).	
3	39	R	1958	365-1911			High purity; specimen consisted three stacked disks each of 0.625 in. I. D., 3.0 in. O. D., and 1 in. thick.	
4	96	L	1950	2.2-10	2.0-5.0		High purity; in normal state; measured in a magnetic field.	
5	96	L	1950	2.3-7.3	2.0-5.0		High purity; in superconducting state.	
6	151	L	1957	8.9-90		Nb 5	99.9% pure; wire drawn from a rod of ductile niobium 1.59 mm dia obtained from Fansteel Metallurgical Corp; ideal electrical resistivity reported as 0.018, 0.084, 0.25, 0.53, 0.97, 2.36, 3.90, 7.0, 9.8, 12.3, 13.5, and 14.5 $\mu\text{ohm cm}$ at 15, 20, 30, 40, 50, 75, 100, 150, 200, 250, 273, and 295 K, respectively; residual electrical resistivity <del>0.47</del> $\mu\text{ohm cm}$ ; transition temp 9.25 K; in normal state.	
7	151	L	1957	4.4-7.5		Nb 5	The above specimen measured in superconducting state.	
8	97	L	1952	2.3-3.1	2.0-3.0	JM 4526; Nb 1	99.99 pure; polycrystalline; magnetic field "frozen in"; in superconducting state; measured after removing the applied magnetic field.	
9	97	L	1952	2.0-9.2	2.0-3.0	Nb 1	The above specimen in superconducting state; measured before applying any magnetic field.	
10	97	L	1952	3.1-7.8	2.0-3.0	Nb 1	The above specimen measured in a field of 2300 gauss; assumed in superconducting state below 6 K and in normal state above 6 K.	
11	97	L	1952	2.3-7.9	2.0-3.0	Nb 1	The above specimen measured in a field of 3300 gauss; assumed in superconducting state below 5 K and in normal state above 5 K.	
12	97	L	1952	9.5-21	2.0-3.0	Nb 1	The above specimen in normal state.	
13	122	L	1955	24-94	3.0	JM 4526; Nb 1	99.99 pure; polycrystalline; 0.470 cm dia x 3.03 cm long; electrical resistivity reported as 0.145, 0.149, 0.153, 0.166, 0.183, 0.220, 0.276, 0.319, 0.416, and 0.462 $\mu\text{ohm cm}$ at 20, 6, 23.4, 26.2, 31.7, 38.9, 47.7, 59.2, 67.1, 82.9, and 90.0 K, respectively.	
14	400	L	1955	0.54-0.75		Nb 1	99.99 pure; polycrystalline; magnetic field "frozen in"; in superconducting state; measured after removing the applied magnetic field.	

SPECIFICATION TABLE NO. 36 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
15	400	L	1953	0.55-0.97		Nb I	99.99 pure; polycrystalline; in superconducting state; measured before applying the magnetic field.
16	389	L	1958	1.0-3.7			Rod specimen; in normal state.
17	389	L	1958	1.0-4.2			The above specimen bent to 5-13.4% strained; in normal state.
18	389	L	1958	1.0-3.7			The above specimen bent to 19.5% strained; in normal state.
19	389, 676	L	1958	0.97-4.2			Single crystal; rod specimen; zone refined; not intentionally annealed; bent to 5.1% strained; in superconducting state.
20	389, 676	L	1958	1.0-4.4			The above specimen bent to 11.4% strained; in superconducting state.
21	389, 676	L	1958	0.87-4.1			The above specimen bent to 19.5% strained; in superconducting state.
22	389, 501	L	1958	0-2.4		Nb II	0.0063 Cu and 0.0003 Mg; single crystal; specimen made by floating zone melting of polycrystalline rod; in normal state (data reported in Ref. 389 are 10 times higher than those reported in Ref. 501 and the latter are used).
23	389, 501	L	1958	1.0-4.3		Nb II	The above specimen in superconducting state (data reported in Ref. 389 are 10 times higher than those reported in Ref. 501 and the latter are used).
24	389, 501	L	58	1.0-3.0		Nb I	Similar to the above specimen; not intentionally annealed; in normal state (data reported in Ref. 389 are 10 times higher than those reported in Ref. 501 and the latter are used).
25	389, 501, 676	L	1958	1.0-4.3		Nb I	The above specimen in superconducting state (data reported in Ref. 389 are 10 times higher than those reported in Ref. 501 and the latter are used).
26	413, 138	L	1955	353-888			99.95 Nb and 0.05 O; rectangular specimen; density $8.38 \text{ g cm}^{-3}$ ; electrical resistivity reported as 16.41, 20.85, 25.22, 29.74, 34.19, 38.63, 43.07, and $45.30 \text{ } \mu\text{hm cm}$ at 0, 100, 200, 300, 400, 500, 600, and 650 C, respectively.
27	413, 138	L	1955	323-856			99.95 Nb and 0.05 O; cylindrical specimen; density $8.65 \text{ g cm}^{-3}$ ; electrical resistivity reported as 15.22, 19.18, 23.13, 27.09, 31.04, 35.00, and $38.96 \text{ } \mu\text{hm cm}$ at 0, 100, 200, 300, 400, 500, and 600 C, respectively.
28	412	L	1955	0.39-0.72		JM 4526	99.99 pure; polycrystalline; measured with magnetic shielding; in superconducting state.
29	412	L	1955	0.54-0.76			The above specimen measured without magnetic shielding; in superconducting state.
30	412	L	1955	0.54-0.99			Same as above, 2nd run.
31	606	L	1954	417-853			Density $7.73 \text{ g cm}^{-3}$ ; electrical resistivity reported as 31.25, 35.78, 40.30, 44.83, 49.35, 53.88, and $58.40 \text{ } \mu\text{hm cm}$ at 0, 100, 200, 300, 400, 500, and 600 C, respectively.



SPECIFICATION TABLE NO. 36 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
32	677	L	1960	1.1-4.2			Single crystal; 4 mm in dia and 50 mm long; prepared by the "floating zone" technique in an electron bombardment furnace; in superconducting state.
33	677	L	1960	1.2-4.4			The above specimen irradiated by a dose of $10^{18}$ fast neutrons $\text{cm}^{-2}$ at $30 \pm 5$ C, then allowed a few weeks for radioactivity to decay; in superconducting state.
34	677	L	1960	1.3-2.3			The above specimen before irradiation; in normal state; measured in a magnetic field.
35	677	L	1960	1.7-2.6			The above irradiated specimen (curve 33) in normal state; measured in a magnetic field.
35	705	L	1962	0.26-1.2	5	Nb II	0.003 Cu and 0.0003 Mg; single crystal; dia 4.0 mm; ratio of length to cross sectional area $25.7 \text{ cm}^{-1}$ ; obtained by the floating zone melting of polycrystalline rod of niobium in vacuum; electrical resistivity ratio $\rho(298\text{K})/\rho_0 = 60.5$ ; in superconducting state.
37	705	L	1962	0.25-4.2	5	Nb III	Dia 2.2 mm; ratio of length to cross sectional area $89.6 \text{ cm}^{-1}$ ; electrical resistivity ratio $\rho(298\text{K})/\rho_0 = 120.0$ ; in superconducting state.
38	724	L, C	1957	373, 473		Sample B	0.1 Ta; electrical resistivity reported as 16.2, 19.5, and $23.5 \mu\text{ohm cm}$ at 20, 100, and 200 C, respectively; Armco iron used as comparative material.
39	847	L	1966	1.2-9.4	10	Nb I	0.03 Al, 0.03 Fe, 0.02 Si, 0.01 C, <0.01 Cl, <0.01 Cr, <0.01 Pb, and <0.01 Mn; single crystal; 2.34 mm in dia and 20 mm long; supplied by Johnson Matthey and Co.; obtained by fusion in a floating zone by electronic bombardment; residual electrical resistivity $0.09 \mu\text{ohm cm}$ ; transition temp $9.5 \text{ K}$ ; in superconducting state.
40	847	L	1966	1.3-9.4	10	Nb I	The above specimen in normal state.
41	847	L	1966	1.1-9.9	10	Nb I	The above specimen irradiated by $5.6 \times 10^{17}$ fast neutrons $\text{cm}^{-2}$ ; residual electrical resistivity $0.11 \mu\text{ohm cm}$ ; in superconducting state.
42	847	L	1966	1.2-5.0	10	Nb I	The above specimen in normal state.
43	847	L	1966	1.4-9.0	10	Nb I	The above specimen annealed at 1870 C in a vacuum of $5 \times 10^{-4}$ torr for 63 hrs; residual electrical resistivity $2.48 \mu\text{ohm cm}$ ; in superconducting state.
44	847	L	1966	1.4-9.2	10	Nb I	The above specimen in normal state.
45	847	L	1966	1.3-9.0	10	Nb III DA	0.1 Ta, 0.01 Ti, 0.007 Fe, 0.005 Cu, 0.005 N, 0.005 O, 0.003 Na, 0.002 Al, 0.002 C, 0.002 Si, and 0.901 H; single crystal; 5.10 mm in dia and 21 mm long; made from polycrystalline sample of Pechiney; annealed at 1350 C in a vacuum of $<10^{-4}$ torr for 3 min; residual electrical resistivity $0.21 \mu\text{ohm cm}$ ; transition temp $9.25 \text{ K}$ ; in superconducting state.
46	847	L	1966	1.3-9.0	10	Nb III DA	The above specimen in normal state.

SPECIFICATION TABLE NO. 36 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent). Specifications and Remarks
47	847	L	1966	1.3-8.8	10	Nb IV AA	Single crystal; 2.96 mm in dia and 21 $\mu$ m long; supplied by Kuhlmann; annealed at 1320 C by electron bombardment for 15 min in a vacuum of $<10^{-6}$ torr; residual electrical resistivity $\rho_0 = 0.38 \mu\text{ohm cm}$ ; transition temp 9.25 K; in superconducting state.
48	847	L	1966	1.4-9.2	10	Nb IV AA	The above specimen in normal state.
49	847	L	1966	1.3-9.6	10	Nb IV B	Single crystal; 2.96 mm in dia and 21 mm long; supplied by Kuhlmann; obtained by fusion in a floating zone; residual electrical resistivity 0.22 $\mu\text{ohm cm}$ ; transition temp 9.25 K; in superconducting state.
50	847	L	1966	1.5-8.9	10	Nb IV B	The above specimen in normal state.
51	848	E	1966	1400-2300	$\pm 2$		99.7 Nb + Ta, 0.17 Ta, 0.06 Si, 0.03 Fe, and 0.025 Ti; cylindrical specimen 65 mm long and 14 mm in dia finished to an "eighth-class" surface (max height of asperities 2.2 $\mu$ ); preheated at 2000 to 2200 K for 4 hrs; density 8.56 g $\text{cm}^{-3}$ ; measured in a vacuum of $5 \times 10^{-6}$ mm Hg.

## DATA TABLE NO. 36 THERMAL CONDUCTIVITY OF NIOBIUM

(Impurity  $\leq$  0.20% each; total impurities  $<$  0.50%)[Temperature, T, K; Thermal Conductivity, k, Watt  $\text{cm}^{-1}\text{K}^{-1}$ ]

T	k	T	k	T	k	T	k	T	k	T	k	T	k	T	k	T	k
<u>CURVE 1</u>		<u>CURVE 4</u>		<u>CURVE 6 (cont.)</u>		<u>CURVE 10</u>		<u>CURVE 13 (cont.)</u>		<u>CURVE 17</u>		<u>CURVE 19 (cont.)</u>		<u>CURVE 20</u>		<u>CURVE 21</u>	
323.2	0.50	2.20	0.0090	55.80	0.608	3.08	0.0158	34.90	0.419	1.00	0.061	3.90	0.0790	1.02	0.00470	1.10	0.00580
373.2	0.51	2.55	0.0124	64.40	0.560	3.08	0.0250	41.04	0.448	1.298	0.082	4.09	0.0890	1.19	0.00725	1.10	0.00580
473.2	0.52	3.05	0.0175	77.90	0.528	4.35	0.0483	51.04	0.454	1.52	0.101	4.23	0.102*	1.33	0.00960	1.10	0.00580
573.2	0.53	3.40	0.0212	90.00	0.534	4.63	0.0538	60.62	0.461	1.86	0.119			1.55	0.0135	1.10	0.00580
<u>CURVE 2</u>		<u>CURVE 5</u>		<u>CURVE 7</u>		<u>CURVE 11</u>		<u>CURVE 14</u>		<u>CURVE 18</u>		<u>CURVE 20</u>		<u>CURVE 21</u>		<u>CURVE 21</u>	
345	0.545	4.40	0.0427	4.42	0.089	2.25	0.0175	0.535	0.00053	0.999	0.054	2.00	0.0220	2.70	0.0380	2.90	0.0408
390	0.545	4.90	0.0830	3.77	0.189	2.38	0.0253	0.560	0.00054	1.04	0.060	2.34	0.0300	2.70	0.0380	2.90	0.0408
395	0.550*	7.70	0.115	7.50	0.180	2.55	0.0205	0.580	0.00054	1.62	0.097	2.51	0.0340	2.70	0.0380	2.90	0.0408
410	0.555	8.00	0.117			2.88	0.0240	0.590	0.00054	1.98	0.115	2.34	0.0300	2.70	0.0380	2.90	0.0408
430	0.560	9.27	0.135			3.05	0.0360	0.640	0.00069	1.62	0.097	2.34	0.0300	2.70	0.0380	2.90	0.0408
510	0.570	10.35	0.157			4.65	0.0640	0.685	0.00072	1.98	0.115	2.34	0.0300	2.70	0.0380	2.90	0.0408
525	0.575*					5.68	0.0938	0.748	0.00382	1.62	0.097	2.34	0.0300	2.70	0.0380	2.90	0.0408
555	0.575					6.38	0.107			1.98	0.115	2.34	0.0300	2.70	0.0380	2.90	0.0408
570	0.580*					7.88	0.136*			1.62	0.097	2.34	0.0300	2.70	0.0380	2.90	0.0408
690	0.609					7.88	0.136*			1.98	0.115	2.34	0.0300	2.70	0.0380	2.90	0.0408
725	0.605									1.62	0.097	2.34	0.0300	2.70	0.0380	2.90	0.0408
840	0.630									1.98	0.115	2.34	0.0300	2.70	0.0380	2.90	0.0408
865	0.615									1.62	0.097	2.34	0.0300	2.70	0.0380	2.90	0.0408
895	0.630									1.98	0.115	2.34	0.0300	2.70	0.0380	2.90	0.0408
935	0.635*									1.62	0.097	2.34	0.0300	2.70	0.0380	2.90	0.0408
965	0.640									1.98	0.115	2.34	0.0300	2.70	0.0380	2.90	0.0408
1040	0.660									1.62	0.097	2.34	0.0300	2.70	0.0380	2.90	0.0408
1095	0.660									1.98	0.115	2.34	0.0300	2.70	0.0380	2.90	0.0408
1140	0.660									1.62	0.097	2.34	0.0300	2.70	0.0380	2.90	0.0408
1195	0.675									1.98	0.115	2.34	0.0300	2.70	0.0380	2.90	0.0408
<u>CURVE 3</u>		<u>CURVE 6</u>		<u>CURVE 9</u>		<u>CURVE 12</u>		<u>CURVE 15</u>		<u>CURVE 19</u>		<u>CURVE 21</u>		<u>CURVE 21</u>		<u>CURVE 21</u>	
364.8	0.466	8.85	0.483	2.00	0.0125	9.48	0.154	0.545	0.000302	0.970	0.00600*	0.870	0.00337*	1.00	0.00380	1.10	0.00490
321.1	0.496	9.42	0.520	2.50	0.0175	2.86	0.0206	0.555	0.00032	1.00	0.00650	1.10	0.00620	1.10	0.00380	1.10	0.00490
798.1	0.521	10.77	0.582	2.86	0.0218	3.15	0.0218	0.728	0.00043	1.02	0.00700	1.19	0.00815	1.10	0.00380	1.10	0.00490
958.2	0.532	12.70	0.660	3.60	0.0248	3.60	0.0248	0.78	0.00044	1.60	0.0225	2.30	0.0240	1.10	0.00380	1.10	0.00490
1134.7	0.583	16.00	0.767	4.25	0.0388	9.83	0.150	0.82	0.00047	1.85	0.0308	2.50	0.0275	1.10	0.00380	1.10	0.00490
1374.3	0.614	18.00	0.842	4.45	0.0375	10.0	0.164	0.83	0.00050	2.10	0.0385	2.67	0.0320	1.10	0.00380	1.10	0.00490
1617.2	0.642	19.23	0.942	5.00	0.0445	10.1	0.172	0.86	0.00051	2.63	0.0500	2.90	0.0355	1.10	0.00380	1.10	0.00490
1910.7	0.689	23.65	0.870	5.08	0.046	10.3	0.177	0.88	0.00051	2.76	0.0520	3.02	0.0400	1.10	0.00380	1.10	0.00490
		28.50	0.865	5.85	0.0555	11.1	0.179	0.88	0.00054	2.98	0.0550	3.13	0.0445	1.10	0.00380	1.10	0.00490
		33.30	0.835	6.00	0.0618	11.6	0.188	0.92	0.00054	3.15	0.0570	3.40	0.0500	1.10	0.00380	1.10	0.00490
		41.00	0.736	7.13	0.0945	13.2	0.214	0.97	0.00060	3.18	0.184	4.10	0.0850	1.10	0.00380	1.10	0.00490
				7.80	0.126	15.6	0.261	1.02	0.00600*	3.35	0.195	4.39	0.0920	1.10	0.00380	1.10	0.00490
				8.00	0.133	20.3	0.314	1.02	0.00600*	3.65	0.215			1.10	0.00380	1.10	0.00490
				8.18	0.133	20.6	0.301	1.22	0.062					1.10	0.00380	1.10	0.00490
				8.48	0.130	20.9	0.327	1.41	0.069					1.10	0.00380	1.10	0.00490
				8.63	0.138			1.60	0.077					1.10	0.00380	1.10	0.00490
				8.80	0.145			1.80	0.089					1.10	0.00380	1.10	0.00490
				9.03	0.148			2.04	0.131					1.10	0.00380	1.10	0.00490
				9.03	0.152*			2.35	0.160					1.10	0.00380	1.10	0.00490
				9.18	0.152			2.69	0.182					1.10	0.00380	1.10	0.00490
								24.17	0.365					1.10	0.00380	1.10	0.00490
								25.83	0.397					1.10	0.00380	1.10	0.00490
								28.33	0.418					1.10	0.00380	1.10	0.00490
								3.43	0.250					1.10	0.00380	1.10	0.00490
								3.70	0.275					1.10	0.00380	1.10	0.00490

Not shown on plot

DATA TABLE NO. 36 (continued)

T	k	T	k	T	k	T	k	T	k	T	k
<u>CURVE 22</u>		<u>CURVE 29</u>		<u>CURVE 33</u>		<u>CURVE 36 (cont.)</u>		<u>CURVE 37 (cont.)</u>		<u>CURVE 37 (cont.)</u>	
1.33	0.0940	0.54	0.00055	1.20	0.0340	0.420	0.00245	0.360	0.00360	1.60	0.148*
1.15	0.105	0.55	0.00055*	1.30	0.0450	0.430	0.00255*	0.380	0.00440	1.68	0.165
1.20	0.110	0.58	0.00059	1.40	0.0500	0.44	0.00270	0.390	0.00465	1.90	0.200
1.40	0.130	0.595	0.00060*	1.64	0.0710	0.45	0.00290	0.400	0.00500	2.05	0.200
1.69	0.150	0.630	0.000705	1.90	0.1040	0.45	0.00280	0.410	0.00520	2.20	0.180
1.85	0.165	0.695	0.00076	2.10	0.1340	0.47	0.00280	0.410	0.00480	2.35	0.165
2.05	0.187	0.755	0.00087	2.30	0.0880	0.48	0.00305	0.420	0.00540	2.50	0.150
2.39	0.216	<u>CURVE 30</u>		2.68	0.0700	0.51	0.00410	0.450	0.00625	2.70	0.140
<u>CURVE 23</u>		<u>CURVE 30</u>		2.90	0.0680*	0.53	0.00370	0.470	0.00650	2.95	0.140
0.997	0.0355	0.540	0.000305	3.06	0.0670	0.53	0.00440	0.480	0.00725	3.20	0.130
1.05	0.0445	0.555	0.000315	3.30	0.0700	0.55	0.00400	0.490	0.00775	3.40	0.170
1.20	0.0660	0.725	0.00042	3.48	0.0750	0.57	0.0051	0.490	0.00800*	3.60	0.200
1.32	0.0906	0.825	0.000425	3.70	0.0830	0.59	0.0053	0.510	0.00850	3.80	0.235
1.61	0.146	0.780	0.000445	3.95	0.0960	0.60	0.0060	0.540	0.0100	3.90	0.250
1.82	0.173	0.810	0.00047	4.11	0.102	0.66	0.0095	0.540	0.0105	4.10	0.285
2.05	0.181	0.820	0.00040	4.37	0.116	0.68	0.0105	0.550	0.0110	4.20	0.340
2.19	0.174	0.850	0.000505	<u>CURVE 34</u>		0.73	0.0120	0.560	0.0102	<u>CURVE 38*</u>	
2.42	0.141	0.930	0.00056	1.30	0.122	0.76	0.0125	0.590	0.0130	373.2	0.506
2.60	0.122	0.99	0.00065	1.50	0.138	0.77	0.0138	0.600	0.0145	473.2	0.527
2.75	0.109	<u>CURVE 31</u>		1.70	0.163	0.82	0.0165	0.650	0.0160	<u>CURVE 39</u>	
2.85	0.100	417.2	0.334	2.00	0.185*	0.84	0.0180	0.67	0.0165	1.22	0.009
2.96	0.0981	505.2	0.338	2.20	0.215	0.85	0.0190	0.68	0.0160	1.26	0.012
3.10	0.0915	583.2	0.349	<u>CURVE 35</u>		0.89	0.0215	0.71	0.0210	1.35	0.014
3.46	0.0935	643.2	0.343	1.70	0.138	0.90	0.0230	0.73	0.0220	1.41	0.016
3.61	0.0971	689.2	0.357	1.90	0.153	0.92	0.0270	0.82	0.0340	1.55	0.021
3.99	0.113	778.2	0.376	2.10	0.170	0.95	0.0285	0.83	0.0320	1.74	0.026
4.05	0.121	853.2	0.376	2.32	0.189	0.98	0.040	0.86	0.0330	1.94	0.035
4.30	0.141	<u>CURVE 32</u>		2.40	0.195	1.00	0.031	0.87	0.0350	2.21	0.050
<u>CURVE 24</u>		1.10	0.0520	2.61	0.212	1.01	0.038	0.88	0.0370	2.39	0.074
1.02	0.0640	1.10	0.0520	<u>CURVE 36</u>		1.08	0.054	0.90	0.0350*	2.59	0.074
1.20	0.0790	1.24	0.0720	0.260	0.06079	1.10	0.057	0.95	0.0430	2.93	0.103
1.40	0.0940	1.50	0.124	0.265	0.06083	1.11	0.057*	0.97	0.0460	3.33	0.143
1.80	0.119	1.60	0.151	0.255	0.06086	1.15	0.058	0.97	0.0430	3.75	0.212
2.01	0.135	1.86	0.185	0.255	0.06165*	<u>CURVE 37</u>		1.05	0.0480	4.31	0.362
2.36	0.150	2.00	0.189	0.255	0.06165*	0.260	0.06079	1.05	0.0520	4.73	0.485
2.70	0.182	2.20	0.166*	0.275	0.06086	0.300	0.06101	1.05	0.0580	5.13	0.615
2.95	0.200	2.60	0.122	0.300	0.06103	0.310	0.06103	1.10	0.0601	5.72	0.803
<u>CURVE 25</u>		2.60	0.098*	0.325	0.06105	0.325	0.06105	1.10	0.0650	6.20	0.981
1.00	0.00900	3.00	0.098*	0.350	0.06140	0.350	0.06140	1.12	0.0680	6.76	1.176
1.14	0.0140	3.53	0.095*	0.360	0.06155	0.360	0.06155	1.23	0.0780*	7.26	1.375
		3.80	0.102	0.390	0.06230	0.390	0.06230	1.30	0.0925	7.75	1.579
		3.94	0.109	0.410	0.06200	0.410	0.06200	1.40	0.115	8.21	1.758
		4.18	0.132	0.410	0.06230	0.410	0.06230	1.45	0.115	8.80	1.919
				0.410	0.06230	0.410	0.06230	1.50	0.130	9.39	2.017

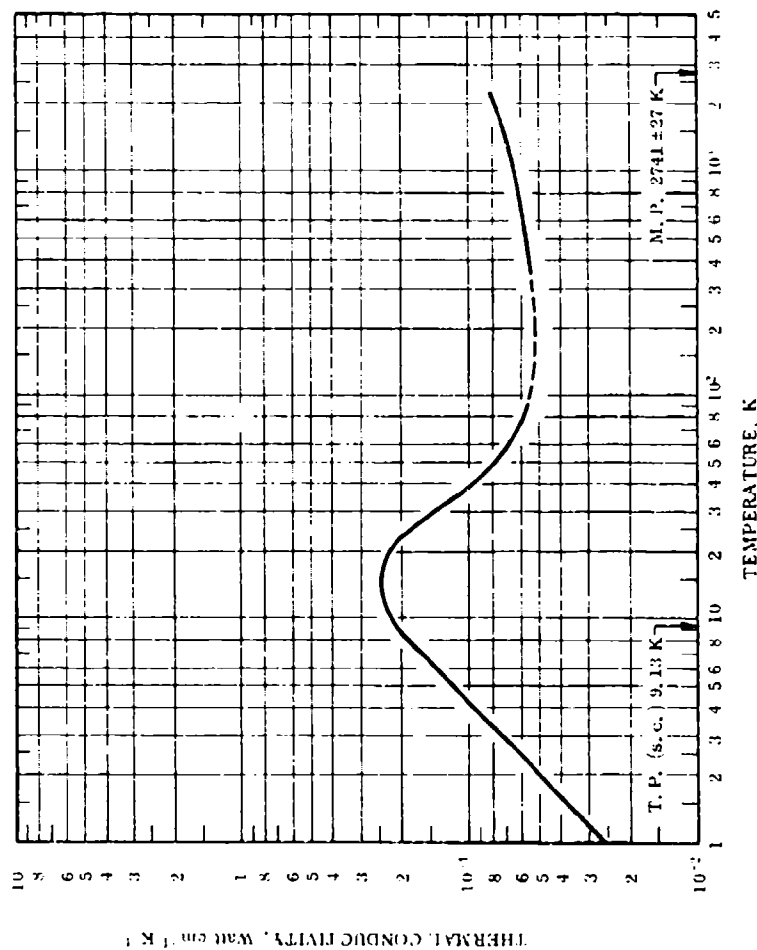
\* Not shown on plot

DATA TABLE NO. 36 (continued)

T	k	T	k	T	k	T	k	T	k	T	k
<u>CURVE 40</u>											
1.30	0.347	1.23	0.273	4.95	0.064	4.95	0.565 <sup>o</sup>	4.38	0.290	5.95	0.648
1.36	0.402	1.60	0.357	5.34	0.068	5.37	0.613	4.91	0.310	6.49	0.702
1.56	0.397	2.13	0.480	5.66	0.070	5.73	0.690	5.52	0.355	7.27	0.782
2.08	0.529	2.73	0.613	6.13	0.073	6.30	0.764	6.17	0.392	8.46	0.880
2.48	0.620	3.95	0.885	6.75	0.079	7.77	0.944	7.06	0.454	9.87	0.962
2.97	0.678	5.01	1.087	7.67	0.090	9.03	1.053	7.87	0.496		
3.67	0.861			9.15	0.103			9.19	0.555		
4.16	1.000										
4.67	1.126										
5.17	1.285										
5.93	1.435										
6.51	1.596										
7.05	1.670										
7.61	1.782										
8.11	1.877										
8.76	1.938										
9.36	1.981										
<u>CURVE 41</u>											
1.12	0.004	3.50	0.032	3.36	0.109	3.61	0.071	3.16	0.078	1400	0.628
1.22	0.004	3.78	0.032	3.66	0.122	3.86	0.081	3.50	0.092	1600	0.654
1.32	0.005	4.14	0.044	4.00	0.152	4.22	0.111	3.82	0.111	1800	0.680
1.50	0.010	4.47	0.046	4.34	0.181	4.54	0.129	4.15	0.148	2000	0.706
1.76	0.020	4.79	0.046	4.63	0.228	4.92	0.154	4.53	0.186	2200	0.732
1.95	0.028	5.06	0.044	4.93	0.259	5.29	0.182	4.83	0.222	2300	0.745
2.16	0.032	5.40	0.050	5.31	0.318	5.73	0.223	5.14	0.267		
2.45	0.050	6.05	0.059	5.70	0.382	6.25	0.273	5.56	0.337		
2.82	0.073	6.72	0.070	6.13	0.472	6.50	0.335	6.01	0.409		
3.15	0.104	7.68	0.094	6.85	0.619	7.53	0.429	6.42	0.483		
3.57	0.156	9.00	0.099	7.89	0.851	8.81	0.536	7.23	0.642		
4.05	0.242			8.99	1.021			8.15	0.808		
4.67	0.352							9.64	0.941		
5.54	0.513										
6.12	0.886										
6.72	1.090										
7.21	1.295										
7.77	1.502										
8.23	1.660										
8.73	1.920										
9.31	1.931										
9.91	1.958										
<u>CURVE 42</u>											
1.41	0.012	1.42	0.015	1.31	0.212	1.38	0.074	1.54	0.171		
1.47	0.016	1.49	0.016 <sup>o</sup>	1.37	0.245	1.48	0.089	1.89	0.215		
1.57	0.017	1.60	0.017 <sup>o</sup>	1.43	0.211	1.55	0.101 <sup>o</sup>	2.13	0.243		
1.69	0.020	1.70	0.019 <sup>o</sup>	1.67	0.198	1.67	0.103	2.45	0.275		
1.73	0.019	1.83	0.020	1.83	0.235	1.86	0.116	2.75	0.303		
1.83	0.023	2.08	0.022	2.07	0.267	2.05	0.127	3.12	0.339		
2.08	0.027	2.41	0.025	2.38	0.305	2.26	0.142	3.55	0.388		
2.41	0.036	2.71	0.029	2.66	0.337	2.52	0.156	3.84	0.412		
2.72	0.033	3.09	0.033	3.03	0.322	2.82	0.174	4.14	0.446		
3.10	0.032	3.41	0.037	3.32	0.414	3.00	0.192	4.47	0.491		
3.50	0.032	3.71	0.040	3.97	0.491	3.30	0.199	5.15	0.556		
3.78	0.032	4.01	0.053	4.33	0.527	3.59	0.220	5.57	0.605		
4.14	0.044	4.65	0.060	4.64	0.544	3.99	0.259				
<u>CURVE 43</u>											
1.34	0.121	1.34	0.037	1.33	0.037	1.33	0.037	1.34	0.020	1400	0.628
1.37	0.113	1.46	0.049	1.46	0.049	1.46	0.049	1.38	0.026	1600	0.654
1.44	0.141	1.57	0.056	1.57	0.056	1.57	0.056	1.46	0.031	1800	0.680
1.66	0.152 <sup>o</sup>	1.71	0.063	1.71	0.063	1.56	0.035	1.56	0.035	2000	0.706
1.84	0.162	1.89	0.069	1.89	0.069	1.64	0.041	1.64	0.041	2200	0.732
2.06	0.166	2.17	0.073	2.17	0.073	1.87	0.050	1.87	0.050	2300	0.745
2.35	0.148	2.43	0.070	2.43	0.070	2.11	0.061	2.11	0.061		
2.67	0.119	2.74	0.065	2.74	0.065	2.43	0.069 <sup>o</sup>	2.43	0.069 <sup>o</sup>		
3.09	0.106	3.15	0.063	3.15	0.063	2.75	0.070 <sup>o</sup>	2.75	0.070 <sup>o</sup>		
3.36	0.109	3.61	0.071	3.61	0.071	3.16	0.078	3.16	0.078		
3.66	0.122	3.86	0.081	3.86	0.081	3.50	0.092	3.50	0.092		
4.00	0.152	4.22	0.111	4.22	0.111	3.82	0.111	3.82	0.111		
4.34	0.181	4.54	0.129	4.54	0.129	4.15	0.148	4.15	0.148		
4.63	0.228	4.92	0.154	4.92	0.154	4.53	0.186	4.53	0.186		
4.93	0.259	5.29	0.182	5.29	0.182	4.83	0.222	4.83	0.222		
5.31	0.318	5.73	0.223	5.73	0.223	5.14	0.267	5.14	0.267		
5.70	0.382	6.25	0.273	6.25	0.273	5.56	0.337	5.56	0.337		
6.13	0.472	6.50	0.335	6.50	0.335	6.01	0.409	6.01	0.409		
6.85	0.619	7.53	0.429	7.53	0.429	6.42	0.483	6.42	0.483		
7.89	0.851	8.81	0.536	8.81	0.536	7.23	0.642	7.23	0.642		
8.99	1.021					8.15	0.808	8.15	0.808		
<u>CURVE 44</u>											
1.42	0.015	1.42	0.015	1.38	0.074	1.38	0.074	1.54	0.171		
1.49	0.016 <sup>o</sup>	1.49	0.016 <sup>o</sup>	1.42	0.090	1.42	0.090	1.89	0.215		
1.60	0.017 <sup>o</sup>	1.60	0.017 <sup>o</sup>	1.48	0.089	1.48	0.089	2.13	0.243		
1.70	0.019 <sup>o</sup>	1.70	0.019 <sup>o</sup>	1.55	0.101 <sup>o</sup>	1.55	0.101 <sup>o</sup>	2.45	0.275		
1.83	0.020	1.83	0.020	1.67	0.103	1.67	0.103	2.75	0.303		
2.08	0.022	2.08	0.022	1.86	0.116	1.86	0.116	3.12	0.339		
2.41	0.025	2.41	0.025	2.05	0.127	2.05	0.127	3.55	0.388		
2.71	0.029	2.71	0.029	2.26	0.142	2.26	0.142	3.84	0.412		
3.09	0.033	3.09	0.033	2.52	0.156	2.52	0.156	4.14	0.446		
3.41	0.037	3.41	0.037	2.82	0.174	2.82	0.174	4.47	0.491		
3.71	0.040	3.71	0.040	3.00	0.192	3.00	0.192	5.15	0.556		
4.01	0.053	4.01	0.053	3.30	0.199	3.30	0.199	5.57	0.605		
4.33	0.059	4.33	0.059	3.59	0.220	3.59	0.220				
4.65	0.060	4.65	0.060	3.99	0.259	3.99	0.259				

Not shown on plot

FIGURE AND TABLE NO. 36R RECOMMENDED THERMAL CONDUCTIVITY OF NIOBIUM



## REMARKS

The recommended values are for well-annealed 99.9% pure niobium with residual electrical resistivity  $\rho_0 = 0.0975 \mu\Omega/\text{cm}$  (characterization by  $\rho_0$  becomes important at temperatures below about 150 K). The values below  $1.5 T_m$  are calculated to fit the experimental data by using  $n = 2.00$ ,  $\alpha' = 5.92 \times 10^{-4}$ , and  $\beta = 3.99$ . The recommended values that are supported by experimental thermal conductivity data are thought to be accurate to within 5% of the true values near room temperature, and 5 to 10% at other temperatures.

		RECOMMENDED VALUES*							
$T_1$	$T_2$	$k_1$	$k_2$	$T_1$	$T_2$	$k_1$	$k_2$	$T_1$	$T_2$
0		0	0	500	-459.7	0.567	32.8	440.3	440.3
1		0.251	14.5	600	-457.9	0.582	33.6	620.3	620.3
2		0.501	28.9	700	-456.1	0.598	34.6	800.3	800.3
3		0.749	43.3	800	-454.3	0.613	35.4	980.3	980.3
4		0.993	57.4	900	-452.5	0.629	36.3	1160	1160
5		1.23	71.1	1000	-450.7	0.644	37.2	1340	1340
6		1.46	84.4	1100	-448.9	0.659	38.1	1520	1520
7		1.67	96.5	1200	-447.1	0.675	39.0	1700	1700
8		1.86	107	1300	-445.3	0.690	39.9	1880	1880
9		2.04	118	1400	-443.5	0.705	40.7	2060	2060
10		2.18	126	1500	-441.7	0.721	41.7	2240	2240
11		2.30	133	1600	-439.9	0.735	42.5	2420	2420
12		2.39	138	1700	-438.1	0.750	43.3	2600	2600
13		2.46	142	1800	-436.3	0.764	44.1	2780	2780
14		2.49	144	1900	-434.5	0.778	45.0	2960	2960
15		2.50	144	2000	-432.7	0.791	45.7	3140	3140
16		2.49	144	2200	-430.9	0.815	47.1	3500	3500
18		2.42	140		-427.3				
20		2.25	132		-423.7				
25		1.87	108		-414.7				
30		1.45	83.8		-405.7				
35		1.16	67.0		-396.7				
40		0.97	56.0		-387.7				
45		0.84	48.5		-378.7				
50		0.76	43.9		-369.7				
60		0.66	38.1		-351.7				
70		0.61	35.2		-333.7				
80		0.58	33.5		-315.7				
90		0.563	32.5		-297.7				
100		(0.552) †	(31.9)		-279.7				
150		(0.530)	(30.6)		-183.7				
200		(0.526)	(30.4)		-99.7				
250		(0.530)	(30.6)		-9.7				
273.2		(0.533)	(30.8)		32.0				
300		(0.537)	(31.0)		80.3				
350		0.544	31.4		170.3				
400		0.552	31.9		260.3				

\*  $T_1$  in K,  $k_1$  in Watt  $\text{cm}^{-1} \text{K}^{-1}$ ,  $T_2$  in F, and  $k_2$  in Btu  $\text{hr}^{-1} \text{ft}^{-1} \text{F}^{-1}$ .

† Values in parentheses are interpolated.

# THERMAL CONDUCTIVITY OF OSMIUM

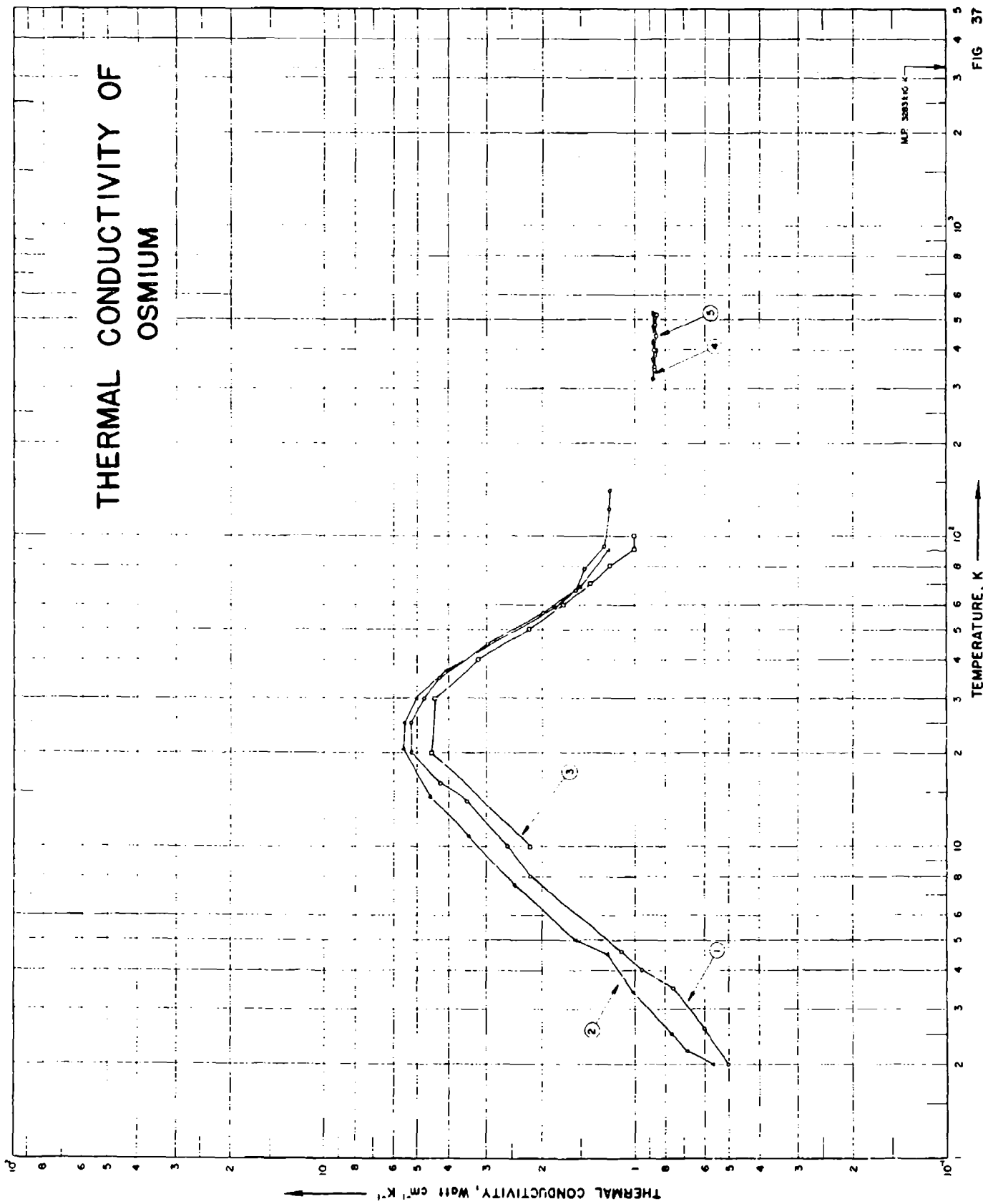


FIG. 37

## SPECIFICATION TABLE NO. 37 THERMAL CONDUCTIVITY OF OSMIUM

(Impurity &lt; 0.20% each; total impurities 0.50%)

For Data Reported in Figure and Table No. 37 :

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)	Specifications and Remarks
1	384	L	1958	2.0-140		Os 2	99.995 pure; specimen 0.6 cm in dia and 6 cm long; powder supplied by Johnson Matthey and Mallory Ltd.; specimen prepared by arc-melting of pressed powder; residual electrical resistivity $0.10 \mu\text{ohm cm}$ ; electrical resistivity ratio $\rho(293\text{K})/\rho_0 = 92.6$ .	
2	384	L	1958	2.0-91		Os 3	99.99 pure; specimen 0.188 cm in dia and 5 cm long; powder supplied by Baker Platinum Co.; specimen prepared by arc-melting of pressed powder; residual electrical resistivity $0.0872 \mu\text{ohm cm}$ ; electrical resistivity ratio $\rho(295\text{K})/\rho_0 = 105.7$ .	
3	512	L	1957	10-100		Os 1	99.995 pure; powder supplied by Johnson Matthey Co.; specimen prepared by arc-melting of pressed powder in helium atmosphere; electrical resistivity ratio $\rho(293\text{K})/\rho(4.2\text{K}) = 20.41$ .	
4	665	C	1962	323-523			0.0091 Ag, 0.0002 Cu, 0.0005 Fe, 0.002 Rh, and 0.03 Ru; specimen 0.489 cm in dia and 2.7 cm long; supplied by Johnson Matthey Co.; prepared by argon-arc melting and ground to size; density $22.45 \text{ g cm}^{-3}$ ; electrical resistivity $8.532$ and $0.272 \mu\text{ohm cm}$ at 273 and 4.2 K, respectively; electrical resistivity ratio $\rho(273\text{K})/\rho(4.2\text{K}) = 31.4$ ; data extracted from smooth curve.	
5	249	C	1967	337-519			0.0001 Ag, 0.0002 Cu, 0.0005 Fe, 0.002 Rh, 0.03 Ru; polycrystalline; specimen 0.459 cm in dia and 2.7 cm long; supplied by Johnson Matthey Co.; arc-melted and ground; annealed at 1820 K; density $22.45 \text{ g cm}^{-3}$ ; electrical resistivity ratio $\rho(273\text{K})/\rho(4.2\text{K}) = 33.3$ (the paper reported density as $12.45 \text{ g cm}^{-3}$ , and the latter ratio as 22.45, apparently a typographical error. This has been confirmed by the author).	



## DATA TABLE NO. 37 THERMAL CONDUCTIVITY OF OSMIUM

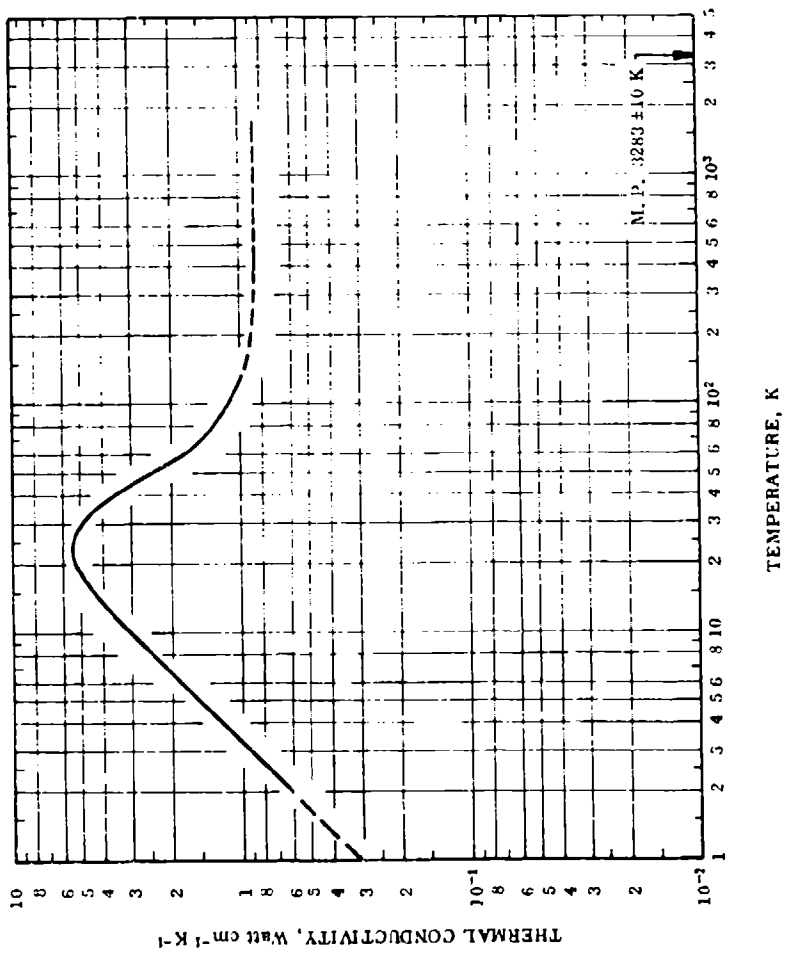
(Impurity &lt; 0.20% each; total impurities &lt; 0.50%)

[Temperature, T, K; Thermal Conductivity,  $k$ , Watt  $\text{cm}^{-1}\text{K}^{-1}$ ]

T	k	T	k
<u>CURVE 1</u>			
2.0	0.50	10	2.2
2.6	0.60	20	4.5
3.5	0.75	30	4.4
4.0	0.95	40	3.2
4.6	1.10	50	2.2
8.0	2.18	60	1.7
10.0	2.58	70	1.4
14.0	3.50	80	1.2
16.0	4.23	90	1
20.0	5.23	100	1
25.0	5.25	<u>CURVE 4</u>	
30.0	4.74	323	0.87
35.0	4.24	373	0.87
45.0	2.98	423	0.87
56.5	1.98	473	0.87
67.0	1.55	523	0.87
78.5	1.46	<u>CURVE 5</u>	
92.5	1.25	336.7	0.866*
122.0	1.21	343.0	0.862
140.0	1.20	343.1	0.866*
<u>CURVE 2</u>			
2.0	0.56	353.5	0.870
2.2	0.68	398.4	0.858
2.5	0.76	399.0	0.870
3.4	1.01	444.5	0.859
4.5	1.22	483.1	0.865
5.0	1.55	517.5	0.854
7.5	2.45	<u>CURVE 3</u>	
10.8	3.46	10	2.2
15.5	4.55	20	4.5
20.6	5.54	30	4.4
25.0	5.50	40	3.2
30.0	5.00	50	2.2
37.0	4.01	60	1.7
59.0	1.40	70	1.4
69.0	1.50	80	1.2
90.8	1.21	90	1

Not shown on plot

FIGURE AND TABLE NO. 37R RECOMMENDED THERMAL CONDUCTIVITY OF OSMIUM



REMARKS

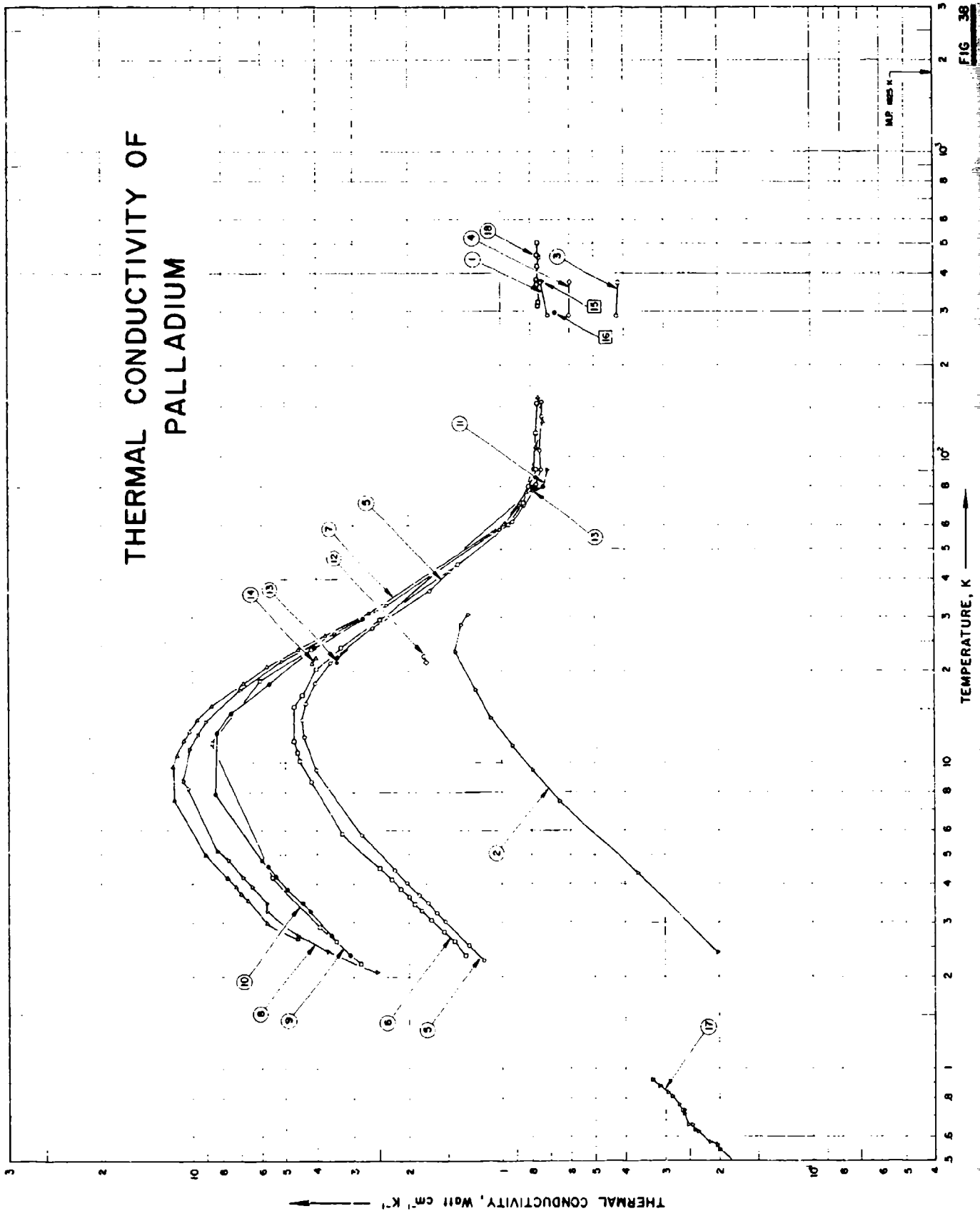
The recommended values are for well-annealed 99.99% pure osmium with residual electrical resistivity  $\rho_0 = 0.0804 \mu\Omega \text{ cm}$  (characterization by  $\rho_0$  becomes important at temperature below about 250 K). The values below  $1.5 \text{ Tm}$  are calculated to fit the experimental data by using  $n = 3.00$ ,  $\sigma' = 3.35 \times 10^{-4}$ , and  $\beta = 3.29$ . The recommended values that are supported by experimental thermal conductivity data are thought to be accurate to within 5% of the true values at room and moderate temperatures and 5 to 10% at other temperatures.

RECOMMENDED VALUES\*  
(For Polycrystalline)

$T_1$	$k_1$	$k_2$	$T_2$
0	0	0	-459.7
1	(0.304) †	(17.6)	-457.9
2	0.608	35.1	-456.1
3	0.912	52.7	-454.3
4	1.22	70.5	-452.5
5	1.52	87.8	-450.7
6	1.82	105	-448.9
7	2.12	122	-447.1
8	2.42	140	-445.3
9	2.72	157	-443.5
10	3.01	174	-441.7
11	3.29	190	-439.9
12	3.57	206	-438.1
13	3.84	222	-436.3
14	4.09	236	-434.5
15	4.34	251	-432.7
16	4.56	263	-430.9
18	4.94	285	-427.3
20	5.23	302	-423.7
25	5.44	314	-414.7
30	5.00	289	-405.7
35	4.30	244	-396.7
40	3.58	207	-387.7
45	2.95	170	-378.7
50	2.45	142	-369.7
60	1.79	103	-351.7
70	1.50	86.7	-333.7
80	1.33	76.8	-315.7
90	1.21	69.9	-297.7
100	1.13	65.3	-279.7
150	(0.962)	(55.6)	-189.7
200	(0.908)	(52.5)	-99.7
250	(0.886)	(51.2)	-9.7
273.2	(0.880)	(50.8)	32.0
300	(0.876)	(50.6)	80.3
350	0.870	50.3	170.3
400	0.869	50.2	260.3
500	0.869	50.2	440.3
600	(0.869)	(50.2)	620.3
1000	(0.869)	(50.2)	1340
1673	(0.869)	(50.2)	2552

\*  $T_1$  in K,  $k_1$  in Watt  $\text{cm}^{-1} \text{K}^{-1}$ ,  $T_2$  in F, and  $k_2$  in Btu  $\text{lb}^{-1} \text{ft}^{-1} \text{F}^{-1}$ . † Values in parentheses are extrapolated or interpolated.

# THERMAL CONDUCTIVITY OF PALLADIUM



SPECIFICATION TABLE NO. 38 THERMAL CONDUCTIVITY OF PALLADIUM

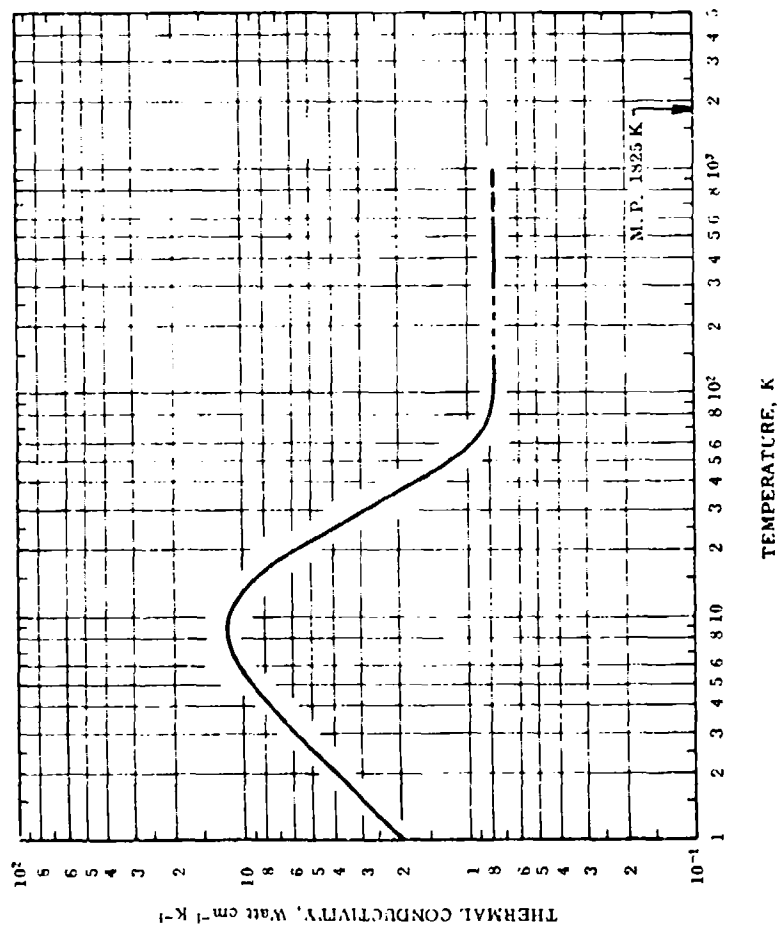
(Impurity &lt; 0.20% each; total impurities &lt; 0.50%)

[ For Data Reported in Figure and Table No. 38 ]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition ( weight percent), Specifications and Remarks
1	77	F	1900	291-373		Pd 1	Pure; specimen 1.610 cm in dia and 27.0 cm long; cast, density 11.96 g cm <sup>-3</sup> at 18 C.
2	97, 122	L	1952	2.4-31	2.0-3.0	Pd 1	99.995 pure; annealed polycrystal; specimen 0.152 cm in dia and 2.82 cm long; supplied by Johnson Matthey (No. 2134); electrical resistivity ratio $\rho(293K)/\rho(20K) = 34.1$ .
3	8	F	1914	290-373			Commercially pure; specimen 0.1010 cm in dia and 35.1 cm long; supplied by Messrs. Isenthal and Co.; electrical resistivity 17.815 and 16.532 $\mu\text{ohm cm}$ at 4.96 and 93.57 C, respectively.
4	8	L	1914	290-373			Pure; specimen 0.0905 cm in dia and 35.2 cm long; electrical resistivity 10.334 and 13.497 $\mu\text{ohm cm}$ at 13.26 and 99.14 C, respectively.
5	82	L	1955	2.3-154		Pd 1	99.995 pure; traces of Ag, Ca, Cu, Si, and Mg; specimen 3 mm in dia; supplied by Johnson Matthey (JM2928); strained.
6	82	L	1955	2.3-150		Pd 2	The above specimen annealed in vacuo for about 4 hrs at 250 C.
7	82	L	1955	2.7-157		Pd 3	The above specimen annealed at 450 C for about 4 hrs.
8	82	L	1955	2.1-131		Pd 4	The above specimen annealed at 650 C for about 4 hrs.
9	82	L	1955	2.4-91		Pd 5	The above specimen annealed at 1000 C for about 4 hrs.
10	82	L	1955	2.2-24		Pd 6	The above specimen drawn to 2 mm dia, annealed at 450 C for about 4 hrs; electrical resistivity $1.82 \times 10^{-8} + 2.12 \times 10^{-8} \text{ ohm cm}$ (the last term should have a factor 10 <sup>-12</sup> ).
11	58	L	1934	79-91		Pd I	Medium pure; unannealed; residual electrical resistivity $\rho_{res} = 9.98 \mu\text{ohm cm}$ .
12	58	L	1934	21-22		Pd I	The above specimen measured after 5.5 months; residual electrical resistivity $\rho_{res} = 9.93 \mu\text{ohm cm}$ .
13	58	L	1934	21-91		Pd II	Very pure; drawn and unannealed; residual electrical resistivity $\rho_{res} = 9.81 \mu\text{ohm cm}$ .
14	58	L	1934	21-81		Pd II	The above specimen annealed for 2 hrs at 360 C; residual electrical resistivity $\rho_{res} = 9.77 \mu\text{ohm cm}$ .
15	390	P	1956	375.2			Pure.
16	241	E	1911	298.2			Pure.
17	736	L	1965	0.42-0.92			Pure palladium.
18	665, 249	C	1962	314-502			0.005 Rh, 0.0005 Au, 0.0005 Fe, 0.0002 Pt, 0.0001 Cu, and 0.0001 Ag; polycrystalline; specimen 0.636 cm in dia and 6.1 cm long supplied by Johnson Matthey Co.; density 12.02 g cm <sup>-3</sup> ; electrical resistivity 2.72, 7.05, 10.9, 14.5, and 17.9 $\mu\text{ohm cm}$ at 100, 200, 300, 400, and 500 K, respectively; $\rho(273K)/\rho(4.2K) = 69$ ; Armco iron used as comparative material.



FIGURE AND TABLE NO. 388 RECOMMENDED THERMAL CONDUCTIVITY OF PALLADIUM



RECOMMENDED VALUES\*

T <sub>1</sub>	k <sub>1</sub>	k <sub>2</sub>	T <sub>2</sub>	T <sub>3</sub>	k <sub>3</sub>	T <sub>4</sub>	k <sub>4</sub>	T <sub>5</sub>
0	0	0	-459.7	500	0.755	440.3	42.6	440.3
1	1.99	11.5	-457.9	600	(0.755)	620.3	(43.6)	620.3
2	3.96	22.9	-456.1	700	(0.755)	800.3	(43.6)	800.3
3	5.86	33.9	-454.	800	(0.755)	980.3	(43.6)	980.3
4	7.61	44.0	-452.5	900	(0.755)	1160	(43.6)	1160
5	9.13	52.8	-450.7	1000	(0.755)	1340	(43.6)	1340
6	10.3	59.5	-448.9					
7	11.1	64.1	-447.1					
8	11.6	67.0	-445.3					
9	11.7	67.6	-443.5					
10	11.5	66.4	-441.7					
11	11.2	64.7	-439.9					
12	10.7	61.8	-438.1					
13	10.1	58.4	-436.3					
14	9.49	54.8	-434.5					
15	8.88	51.3	-432.7					
16	8.28	47.8	-430.9					
18	7.08	40.9	-427.3					
20	5.98	34.6	-423.7					
25	4.04	23.3	-414.7					
30	2.85	16.5	-405.7					
35	2.15	12.4	-396.7					
40	1.72	9.4	-387.7					
45	1.43	8.2	-378.7					
50	1.23	7.1	-369.7					
60	0.982	5.6	-351.7					
70	0.868	5.0	-333.7					
80	0.811	4.6	-315.7					
90	0.786	4.5	-297.7					
100	0.773	4.4	-279.7					
150	0.755	4.3	-189.7					
200	(0.755)†	(43.6)	-99.7					
250	(0.755)	(43.6)	-9.7					
273.2	(0.755)	(43.6)	32.0					
300	0.755	43.6	60.3					
350	0.755	43.6	170.3					
400	0.755	43.6	260.3					

REMARKS

The recommended values are for well-annealed 99.995% pure palladium with residual electrical resistivity  $\rho_0 = 0.0123 \mu\Omega \text{ cm}$  (characterization by  $A_0$  becomes important at temperatures below about 150 K). The values below 1.5 T<sub>m</sub> are calculated to fit the experimental data by using  $n = 2.00$ ,  $a = 0.40$ ,  $m = 2.40$ ,  $\alpha' = 1.54 \times 10^{-4}$ , and  $\beta = 0.502$ . The recommended values that are supported by experimental thermal conductivity data are thought to be accurate to within 4% of the true values near room temperature and 4 to 10% at other temperatures.

\* T<sub>1</sub> in K, k<sub>1</sub> in Watt cm<sup>-1</sup> K<sup>-1</sup>, T<sub>3</sub> in F, and k<sub>3</sub> in Btu lb<sup>-1</sup> ft<sup>-1</sup> F<sup>-1</sup>. † Values in parentheses are extrapolated or interpolated.

# THERMAL CONDUCTIVITY OF PLATINUM

FIGURE SHOWS ONLY 29 OF THE CURVES REPORTED IN TABLE

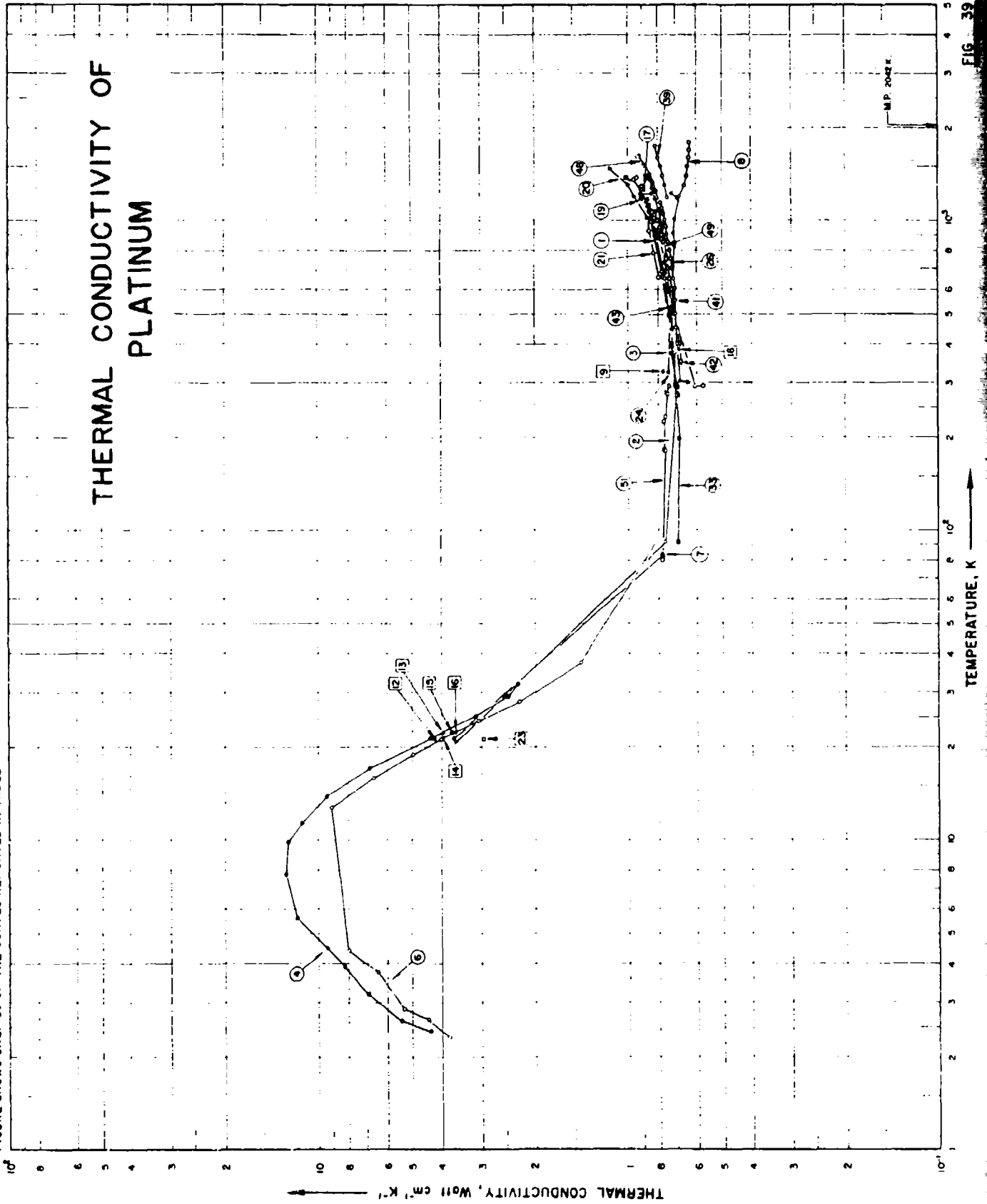


FIG. 39

## SPECIFICATION TABLE NO. 39 THERMAL CONDUCTIVITY OF PLATINUM

(impurity &lt; 0.20% each; total impurities &lt; 0.50%)

[For Data Reported in Figure and Table No. 39]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)	Specifications and Remarks
1	70	E	1930	293-1293	3			99.95 pure; electrical resistivity data fitted into the equation $\rho = 10.48 [1 + 3.695 \times 10^{-5} \times (T-15) - 5.98 \times 10^{-11} \times (T-15)^2 + 5.25 \times 10^{-11} \times (T-15)^3] \mu\text{ohm cm}$ , T in C.
2	95	E	1915	21-374				Very high purity; drawn and electrically annealed; electrical conductivity 10.2 and 9.5 $\times 10^4$ mho $\text{cm}^{-1}$ at 278.1 and 291 K, respectively.
3	77	E	1900	291.373		Pt II		Pure; specimen 1.614 cm in dia and 27.0 cm long; density 21.39 $\text{g cm}^{-3}$ at 18 C.
4	97, 122	L	1952	2.4-32	2-3	Pt I		99.999 pure; supplied by Johnson Matthey and Co. (JM 2157b); annealed wire; $\rho(293 \text{ K})/\rho(20 \text{ K}) = 202$ .
5	8	F	1914	273-373				Pure; electrical conductivity 10.24 and 7.35 $\times 10^4$ mho $\text{cm}^{-1}$ at 273 and 373 K, respectively.
6	149	L	1957	2.3-91				99.99% pure; specimen 1.5 mm in dia; supplied by Baker Platinum Co.; annealed at 1050 C; $\rho(295 \text{ K})/\rho_0 = 833$ ; residual electrical resistivity 0.0125 $\mu\text{ohm cm}$ .
7	57	L	1927	21.63		Pt III		Very pure; polycrystal; drawn and electrically annealed; electrical resistivity 0.0650, 2.10, and 9.81 $\mu\text{ohm cm}$ at -252, -190 and 0 C, respectively.
8	273	E	1954	1200-1800				Spectroscopically pure wire; obtained from Johnson Matthey and Co.
9	487	L	1894	326.2				Pure; specimen 2.0 mm in dia.
10	451	C	1930	291.2				Pure; tempered at 800 C and quenched, rolled and drawn; gold used as comparative material [k data 3.09 $\text{W cm}^{-1}\text{K}^{-1}$ ].
11	399	L	1925	290, 373				Pure.
12	436	L	1938	21.17		Pt IV 33		Quasi-isotropic; electrical resistivity 0.0416 and 9.81 $\mu\text{ohm cm}$ at 21.38 and 273.2 K, respectively.
13	436	L	1938	22.01		Pt IV 33		The above specimen, second measurement.
14	436	L	1939	21.21		Pt IV 33		The above specimen measured at H = 8750 oersteds; electrical resistivity 0.04578 $\mu\text{ohm cm}$ at 21.38 K.
15	436	L	1938	22.10		Pt IV 33		The above specimen measured at H = 8750 oersteds.
16	436	L	1938	22.15		Pt IV 33		The above specimen measured at H = 12200 oersteds; electrical resistivity 0.04820 $\mu\text{ohm cm}$ at 21.38 K.
17	488	E	1929	293-1293	0.7-2.0			99.95 pure; electrical resistivity 10.65, 24.90, 35.01 and 43.61 $\mu\text{ohm cm}$ at 20, 412, 725, and 1020 C, respectively.
18	390	P	1956	364.2				Pure.
19	503	E	1961	1073-1223				99.9 chemically pure; specimen in the form of 0.1 mm dia wire stretched between two heaters; wire surface polished with Vienneuse chalk or Paris red (crocus, polishing powder); annealed at about 1000 C for 12 hrs.
20	599	E	1961	301-1473				99.9 pure; electrical resistivity 10.6 $\mu\text{ohm cm}$ at 23 C.
21	599	E	1961	292-1376				Similar to the above specimen.
22	241	L	1911	298.2				Less than 0.03 impurity.



SPECIFICATION TABLE NO. 39 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
23	57	L	1927	21.2		Pt IV	Pure; polycrystal; annealed; electrical resistivity 0.0899 and 9.83 $\mu\text{hm cm}$ at 21.2 and 273.2 K, respectively.
24	665	C	1962	323-523			0.0001 Cu, 0.0001 Fe, and <0.0001 Pd; specimen 0.635 cm in dia and 6.1 cm long; supplied by Johnson Matthey and Co.; annealed at approx 1000 C; density 21.51 g $\text{cm}^{-3}$ ; electrical resistivity 0.013 and 9.85 $\mu\text{hm cm}$ at 4.2 and 273 K, respectively; 0.321 cm dia Armco iron rod used as comparative material; heat outflow also measured by water-flow calorimeter.
25	645	C	1963	315-503		Pt 1	0.0001 Cu, 0.0001 Fe, and <0.0001 Pd; specimen 0.62 cm in dia and 6.1 cm in length; density 21.5 g $\text{cm}^{-3}$ ; machined; annealed at about 1000 C; 0.321 cm dia Armco iron rod used as comparative material; heat outflow also measured by water-flow calorimeter.
26	645	C	1963	445-1220		Pt 1	The above specimen; 0.371 cm dia Armco iron rod used as comparative material.
27	645	C	1963	787-1153		Pt 1	The above specimen; 1.273 cm dia Armco iron rod used as comparative material.
28	645	C	1963	760-1070		Pt 1	Same as above; reassembled.
29	645	C	1963	335-467		Pt 2	0.0001 Si, <0.0001 Ag, <0.0001 Ca, <0.0001 Cu, 0.0001 Fe, <0.0001 Mg, and <0.0001 Pd; specimen 1.269 cm in dia and 10.16 cm in length; annealed at ~1000 C; density 21.5 g $\text{cm}^{-3}$ ; electrical resistivity (measured after all other tests) 9.9, 13.8, 17.4, 21.0, 24.5, 27.9, 31.1, 34.3, 37.3, 40.2, and 43.0 $\mu\text{hm cm}$ at 0, 100, 200, 300, 400, 500, 600, 700, 800, 900, and 1000 C, respectively; Lorenz function at these temperatures being respectively 2.66, 2.70, 2.68, 2.67, 2.65, 2.64, 2.60, 2.57, 2.54, and 2.47 $\times 10^{-5}\text{V}\mu\text{K}^{-2}$ ; 1.222 cm dia Armco iron rod used as comparative material; heat outflow also measured by water-flow calorimeter.
30	645	C	1963	357-800		Pt 2	Same as above but 1.9 cm dia Armco iron rod used as comparative material.
31	645	C	1963	575-1141		Pt 2	The above specimen; 0.371 cm dia Armco iron rod used as comparative material.
32	736	L	1965	0.43-0.82		Pt 2	Pure platinum wire.
33	847	E	1952	90-579			Wire 11.6 cm long and 1.5 mm dia.
34	662	L	1964	298-358	0.5		99.98 pure; $\approx 0.0030$ Ir, 0.0021-0.0023 <sup>Se</sup> 0.0021-0.0023 Rh, 0.0015-0.0017 Al, 0.0015-0.0017 Pd, 0.0011 Au, 0.0007-0.0009 Mg, 0.0007-0.0009 Ag, and 0.0004-0.0005 Fe; specimen 5.0 cm in dia and 7.0 cm long; cast, cold-pressed and machined; density 21.32 g $\text{cm}^{-3}$ at 20 C; held at 600 C for 2 hrs; first run. <b>0.004 - 0.006 CH</b>
35	662	L	1964	294-349	0.5		The above specimen; second run.
36	662	L	1964	294-365	0.5		The above specimen; third run.
37	662	L	1964	296-363	0.5		The above specimen; fourth run.
38	†	L	1967	0.42-0.81	<1.0		99.999 pure; polycrystalline wire specimen; form factor $L/a = 7.74 \times 10^3 \text{cm}^{-1}$ ; obtained from Johnson and Matthey Co.; electrical resistivity 0.08004 $\mu\text{hm cm}$ at 1.5 K; $\rho(293 \text{ K})/\rho(1.5 \text{ K}) = 148$ .

SPECIFICATION TABLE NO. 39 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
39	654	P	1965	1180-1750			99.95 Pt sheet of 1 mm thickness; average grain size after test 1000 $\mu$ ; density 21.5 g cm <sup>-3</sup> ; data calculated from thermal diffusivity measurements using the specific heat data of Kubaschewski, O. and Evans, L. I. (Metallurgical Thermochemistry, Pergamon Press, 1956).
40	624	E	1959	323-773			99.99 pure; polycrystal.
41	648, 993	P	1964	300-1150	$\pm 3$	I	99.999 Pt, impurities (atomic %): 0.002 Pd, 0.001 Ir, 0.001 Ag, 0.001 Zn, 0.0006 Mo, 0.0005 Os, 0.0006 Ru, 0.0004 In, 0.0002 Re, 0.0002 W, 0.0001 Cu, 0.00007 Rh, and 0.00005 Ta; supplied by J. Bishop and Co.; $D_{T_1}K/D_2K = 900$ , annealed at 1200 K for at least 1 hr; data calculated from thermal diffusivity data using a constant density of 21.37 g cm <sup>-3</sup> and the specific heat data of Jaeger, F. M. and Rosenbohm, E. (Physica, 6, 1123-5, 1939).
42	648, 993	P	1964	300-1250	$\pm 3$	II	99.9 Pt, impurities (atomic %): 0.35 Rh, 0.24 Ir, 0.05 Pd, 0.04 Ag, 0.034 Ru, 0.015 Cu, 0.007 Zn, 0.001 W, 0.0006 Ta, 0.0005 In, 0.0004 Re, 0.0001 Os, and < 0.0001 Pt; supplied by J. Bishop and Co.; $D_{T_1}K/D_2K = 12$ ; annealed at 1200 K for at least one hr; data calculated from thermal diffusivity data using a constant density of 21.37 g cm <sup>-3</sup> and the specific heat data of Jaeger, F. M. and Rosenbohm, E. (Physica, 6, 1123-5, 1939).
43	689, 690	E	1965	273-1383	< 5		Platinum wire 0.3 mm in dia.
44	700, 993	P	1965	400-1200	$\pm 3$	B	99.999 Pt (nominal), impurities (atomic %): 0.2 Pd, 0.06 Cu, 0.057 Rh, 0.01 Ag, 0.004 Zn, 0.001 Ir, 0.001 Ru, 0.0006 Os, 0.0002 Re, 0.0002 W, < 0.0001 Mo, 0.00007 Ta, and 0.00006 In; rod 0.1875 in. in dia and about 10 in. long; supplied by Engelhard Industries; annealed at 1200 K for at least one hr; electrical resistivity, 10.9, 14.75, 18.45, 22.10, 25.64, 29.00, 32.20, 35.35, and 38.45 $\mu$ ohm cm at 300, 400, 500, 600, 700, 800, 900, 1000, and 1100 K respectively and electrical resistivity ratio $D_{T_1}K/D_2K = 100$ determined upon completion of the thermal diffusivity measurements; thermal conductivity values calculated from the thermal diffusivity measurements using a constant density of 21.37 g cm <sup>-3</sup> from Smithsonian Physical Tables (1954) and also specific heat data of Jaeger and Rosenbohm (1939).
45	700, 993	P	1965	400-1200	$\pm 3$	C	Data from a similar specimen having a nominal purity of 99.9 impurities (atomic %): 0.009 Rh, 0.006 Pd, 0.004 Ag, 0.003 Zn, 0.002 Cu, 0.001 Ir, 0.0006 W, 0.0005 In, 0.0004 Re, 0.0003 Os, < 0.0003 Mo, 0.0002 Ta, and < 0.0002 Ru; and electrical resistivity 11.30, 15.13, 18.90, 22.60, 26.14, 29.51, 32.76, 35.86, and 38.89 $\mu$ ohm cm at 300, 400, 500, 600, 700, 800, 900, 1000, and 1100 K respectively; $D_{T_1}K/D_2K = 34$ .

SPECIFICATION TABLE NO. 39 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
46	700, 993	P	1965	400-1200	±3	D	Data from another specimen having a nominal purity of 99.999 impurities (atomic %): 0.002 Zn, 0.001 Cu, 0.001 In, 0.001 Ir, 0.001 Pd, 0.001 Ag, 0.0006 Mo, 0.0004 Os, 0.0006 Ru, 0.0002 Re, 0.0002 W, 0.0007 Rh, and 0.00006 Ta; from Sigmund Cohn Corp. with a corresponding electrical resistivity of 10.90, 14.68, 18.40, 21.98, 25.45, 29.82, 32.04, 35.10, and 38.13 $\mu\text{ohm cm}$ at 300, 400, 500, 600, 700, 800, 900, 1000, and 1100 K, respectively; $\rho_{293K}/\rho_{4.2K} = 5000$ . 99.999 Pt, 0.0004 Rh, 0.0003 Fe; specimen 1.2 cm in dia and 10 cm long; supplied by Engelhard Industries, Inc., Newark, New Jersey; machined from a special lot of platinum.
47	507	L	1966	300-1000	±1.7		Nominal purity 99.9, following percentages are upper limits: 0.1 Ag, 0.2 Au, 0.01 Cr, 0.1 Cu, 0.1 Fe, 0.01 Mn, 0.01 Ni, 0.1 Pd, 0.2 Rh, 0.2 Ru, 0.1 Si; specimen 0.479 cm in dia and 1.5 cm long; supplied by Engelhard Inc.; electrical resistivities from smoothed curve of author's measurements: 34.54, 35.68, 37.90, 40.06, 42.60, 45.01, 46.89, 49.75, and 52.28 $\mu\text{ohm cm}$ at 970, 1006, 1078, 1153, 1243, 1331, 1402, 1511, and 1611 K respectively; thermal conductivity values calculated from the thermal diffusivity measurements using a constant density 21.37 $\text{g cm}^{-3}$ from Forsythe, W. E. (Smithsonian Physical Tables, 9th revised ed.) and the specific heat data of Jaeger, F. M. and Rosenbohm, E. (Physical J. 1123, 1939).
48	878	P	1966	970-1611			99.987 pure; density 21.384 ± 0.002 $\text{g cm}^{-3}$ at 21 C, ice point resistivity 9.847 ± 0.01 $\mu\text{ohm cm}$ (corrected to 0 C dimensions); Vickers Hardness Number subsequent to annealing 37.
49	879, 991, 1000	L	1966	373-1373			The above specimen measured by different method.
50	879, 991	E	1966	373-1373			
51	249	C	1967	80-294		Platinum I	0.0001 Cu, 0.0001 Fe, 0.0001 Pd; polycrystalline specimen 0.635 cm in dia and 6.1 cm long; supplied by Johnson, Matthey and Co., annealed at 1273 K; density 21.51 $\text{g cm}^{-3}$ ; electrical resistivity ratio $\rho_{293K}/\rho_{4.2K} = 740$ .
52	249	C	1967	300-500		Platinum II	0.0001 each of Fe and Si, < 0.0001 each of Hg, Cu, Pd, Ca, and Mg; polycrystalline; specimen 1.269 cm in dia and 10.16 cm, long; supplied by Johnson, Matthey and Co., annealed at 1250 K; density 21.5 $\text{g cm}^{-3}$ .
53	989	E	1947	195-373			Wire specimen 1.438 ± 0.003 mm in dia; measured in a vacuum of $<10^{-5}$ mm Hg.

## DATA TABLE NO. 39 THERMAL CONDUCTIVITY OF PLATINUM

(Impurity &lt; 0.20% each; total impurities &lt; 0.50%)

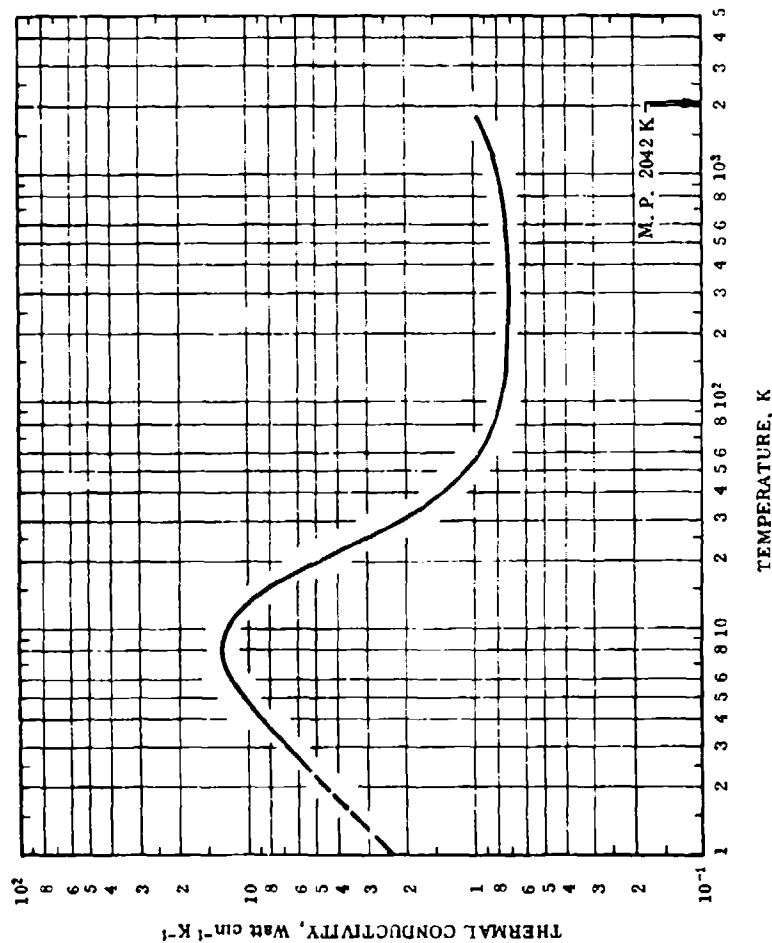
[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup> K<sup>-1</sup>]

T	k	T	k	T	k	T	k	T	k	T	k	T	k	T	k	T	k
<u>CURVE 1</u>		<u>CURVE 6</u>		<u>CURVE 12</u>		<u>CURVE 20</u>		<u>CURVE 25<sup>b</sup> (cont.)</u>		<u>CURVE 29<sup>a</sup></u>		<u>CURVE 32</u>					
292.7	0.699	2.30	3.810	21.17	4.25	301	0.640	327	0.730	335	0.725	345	0.115	0.425	0.115	0.425	0.115
689.2	0.779	2.63	4.459	<u>CURVE 13</u>		363	0.725	330	0.725	346	0.725	370	0.135	0.470	0.135	0.470	0.135
998.2	0.842	2.85	5.325	3.75	6.494	590	0.738	337	0.732	370	0.738	371	0.138	0.480	0.138	0.480	0.138
1293.2	0.900	3.75	6.494	22.01	3.97	730	0.761	343	0.724	371	0.750	376	0.148	0.500	0.148	0.500	0.148
<u>CURVE 2</u>		4.38	8.052	<u>CURVE 14</u>		860	0.794	343	0.738	376	0.734	382	0.163	0.560	0.163	0.560	0.163
20.7	3.635	12.72	9.134	9.21	0.805	921	0.875	360	0.722	382	0.729	401	0.175	0.575	0.175	0.575	0.175
91.4	0.762	15.79	6.667	10.37	0.875	1037	0.960	373	0.735	401	0.732	434	0.178	0.590	0.178	0.590	0.178
273.1	0.699	18.86	4.978	12.30	0.960	1230	1.00	373	0.736	434	0.729	437	0.185	0.600	0.185	0.600	0.185
291.1	0.701	21.05	4.026	13.31	1.00	1331	1.14	385	0.723	437	0.724	440	0.198	0.638	0.198	0.638	0.198
294.2	0.701 <sup>a</sup>	24.12	3.074	14.73	1.14	1473		395	0.732	440	0.733	467	0.207	0.645	0.207	0.645	0.207
373.3	0.706	27.85	2.251	<u>CURVE 15</u>		429	0.727	429	0.727	467	0.733	<u>CURVE 30<sup>b</sup></u>		0.680	0.213	0.680	0.213
<u>CURVE 3</u>		37.28	1.429	22.10	3.74	430	0.715	437	0.715	470	0.727	0.710	0.223	0.710	0.223	0.710	0.223
291.2	0.696	292	0.577	29.15	3.63	292	0.577	453	0.722	480	0.712	0.745	0.231	0.745	0.231	0.745	0.231
373.2	0.725	293	0.609	29.15	3.63	293	0.609	480	0.722	503	0.723	0.770	0.245	0.770	0.245	0.770	0.245
<u>CURVE 4</u>		83.2	0.780	654	0.798	654	0.798	503	0.723	<u>CURVE 26</u>		0.773	0.248	0.773	0.248	0.773	0.248
2.41	4.387	784	0.830	784	0.830	784	0.830	445	0.725	512	0.722	0.815	0.258	0.815	0.258	0.815	0.258
2.60	5.450	925	0.857	925	0.857	925	0.857	445	0.725	512	0.722	<u>CURVE 33</u>		0.790	0.258	0.790	0.258
3.18	6.968	1085	0.860	1085	0.860	1085	0.860	445	0.725	512	0.722	0.815	0.265	0.815	0.265	0.815	0.265
3.90	8.315	1240	0.929	1240	0.929	1240	0.929	445	0.725	512	0.722						
4.47	9.462	1240	0.929	1240	0.929	1240	0.929	445	0.725	512	0.722						
5.59	11.785	1240	0.929	1240	0.929	1240	0.929	445	0.725	512	0.722						
7.74	12.820	1240	0.929	1240	0.929	1240	0.929	445	0.725	512	0.722						
9.81	12.560	1240	0.929	1240	0.929	1240	0.929	445	0.725	512	0.722						
11.35	11.400	1240	0.929	1240	0.929	1240	0.929	445	0.725	512	0.722						
13.76	9.462	1240	0.929	1240	0.929	1240	0.929	445	0.725	512	0.722						
16.95	6.882	1240	0.929	1240	0.929	1240	0.929	445	0.725	512	0.722						
21.25	4.387	1240	0.929	1240	0.929	1240	0.929	445	0.725	512	0.722						
24.95	2.140	1240	0.929	1240	0.929	1240	0.929	445	0.725	512	0.722						
29.00	2.452	1240	0.929	1240	0.929	1240	0.929	445	0.725	512	0.722						
31.74	2.280	1240	0.929	1240	0.929	1240	0.929	445	0.725	512	0.722						
<u>CURVE 5<sup>a</sup></u>		<u>CURVE 11<sup>a</sup></u>		<u>CURVE 18</u>		<u>CURVE 22</u>		<u>CURVE 28</u>		<u>CURVE 34</u>		<u>CURVE 35<sup>a</sup></u>					
273.2	0.691	326.2	0.778	384.2	0.690	298.2	0.700	760	0.707	787	0.719	1085	0.734	294.3	0.7041	294.3	0.7041
290.2	0.690	373.2	0.711	1073.2	0.82	21.2	2.96	810	0.736	833	0.723	1109	0.738	297.7	0.7049	297.7	0.7049
373.2	0.711	1118.2	0.87	1118.2	0.87	1073.2	0.82	810	0.736	977	0.723	1141	0.732	305.2	0.7054	305.2	0.7054
		1128.2	0.86	1128.2	0.86	1128.2	0.86	846	0.735	1027	0.730			318.7	0.7051	318.7	0.7051
		1148.2	0.88	1148.2	0.88	1148.2	0.88	846	0.735	1140	0.730						
		1173.2	0.87	1173.2	0.87	1173.2	0.87	846	0.735	1153	0.723						
		1173.2	0.91	1173.2	0.91	1173.2	0.91	846	0.735	<u>CURVE 31</u>							
		1223.2	0.91	1223.2	0.91	1223.2	0.91	846	0.735	755	0.729						
		1223.2	0.84	1223.2	0.84	1223.2	0.84	846	0.735	800	0.733						
		290.2	0.690	326.2	0.778	326.2	0.778	846	0.735	<u>CURVE 31</u>							
		373.2	0.711	373.2	0.711	373.2	0.711	846	0.735	575	0.738						
		373.2	0.711	373.2	0.711	373.2	0.711	846	0.735	648	0.732						
		373.2	0.711	373.2	0.711	373.2	0.711	846	0.735	726	0.735						
		373.2	0.711	373.2	0.711	373.2	0.711	846	0.735	846	0.731						
		373.2	0.711	373.2	0.711	373.2	0.711	846	0.735	944	0.721						
		373.2	0.711	373.2	0.711	373.2	0.711	846	0.735	1085	0.734						
		373.2	0.711	373.2	0.711	373.2	0.711	846	0.735	1109	0.738						
		373.2	0.711	373.2	0.711	373.2	0.711	846	0.735	1141	0.732						

<sup>a</sup> Not shown on Plot



FIGURE AND TABLE NO. 39R RECOMMENDED THERMAL CONDUCTIVITY OF PLATINUM



## RECOMMENDED VALUES\*

$T_1$	$k_1$	$k_2$	$T_2$	$T_1$	$k_1$	$k_2$	$T_2$
0	0	0	-459.7	350	0.715	41.3	170.3
1	(2.31)†	(134)	-457.9	400	0.716	41.4	260.3
2	(4.60)	(266)	-456.1	500	0.722	41.7	440.3
3	6.79	392	-454.3	600	0.730	42.2	620.3
4	8.80	509	-452.5	700	0.741	42.8	800.3
5	10.5	607	-450.7	800	0.755	43.6	980.3
6	11.8	682	-448.9	900	0.770	44.5	1160
7	12.6	728	-447.1	1000	0.786	45.4	1340
8	12.9	745	-445.3	1100	0.805	46.5	1520
9	12.8	740	-443.5	1200	0.826	47.7	1700
10	12.3	711	-441.7	1300	0.848	49.0	1880
11	11.7	676	-439.9	1400	0.871	50.3	2060
12	10.9	630	-438.1	1500	0.896	51.8	2240
13	10.1	584	-436.3	1600	0.921	53.2	2420
14	9.30	537	-434.5	1700	0.947	54.7	2600
15	8.41	486	-432.7	1800	0.973	56.2	2780
16	7.59	439	-430.9				
18	6.12	354	-427.3				
20	4.95	286	-423.7				
25	3.13	181	-414.7				
30	2.15	124	-405.7				
35	1.68	97.1	-396.7				
40	1.39	80.3	-387.7				
45	1.22	70.5	-378.7				
50	1.09	63.0	-369.7				
60	0.947	54.7	-351.7				
70	0.862	49.8	-333.7				
80	0.815	47.1	-315.7				
90	0.789	45.6	-297.7				
100	0.775	44.8	-279.7				
150	0.740	42.8	-189.7				
200	0.724	41.8	-99.7				
250	0.717	41.4	-9.7				
273.2	0.715	41.3	32.0				
300	0.714	41.3	80.3				

## REMARKS

The recommended values are for well-annealed 99.999% pure platinum with residual electrical resistivity  $\rho_0 = 0.0106 \mu\Omega \text{ cm}$  (characterization by  $\rho_0$  becomes important at temperatures below about 150 K). The values below 1.5  $T_m$  are calculated to fit the experimental data by using  $\beta = 2.10$ ,  $\alpha' = 3.01 \times 10^{-4}$ , and  $\beta = 0.433$ . The recommended values that are supported by experimental thermal conductivity data are thought to be accurate to within 5% of the true values near room temperature and 5 to 10% at other temperatures.

\*  $T_1$  in K,  $k_1$  in  $\text{Watt cm}^{-1} \text{K}^{-1}$ ,  $T_2$  in F, and  $k_2$  in  $\text{Btu hr}^{-1} \text{ft}^{-1} \text{F}^{-1}$ .

† Values in parentheses are extrapolated.

# THERMAL CONDUCTIVITY OF PLUTONIUM

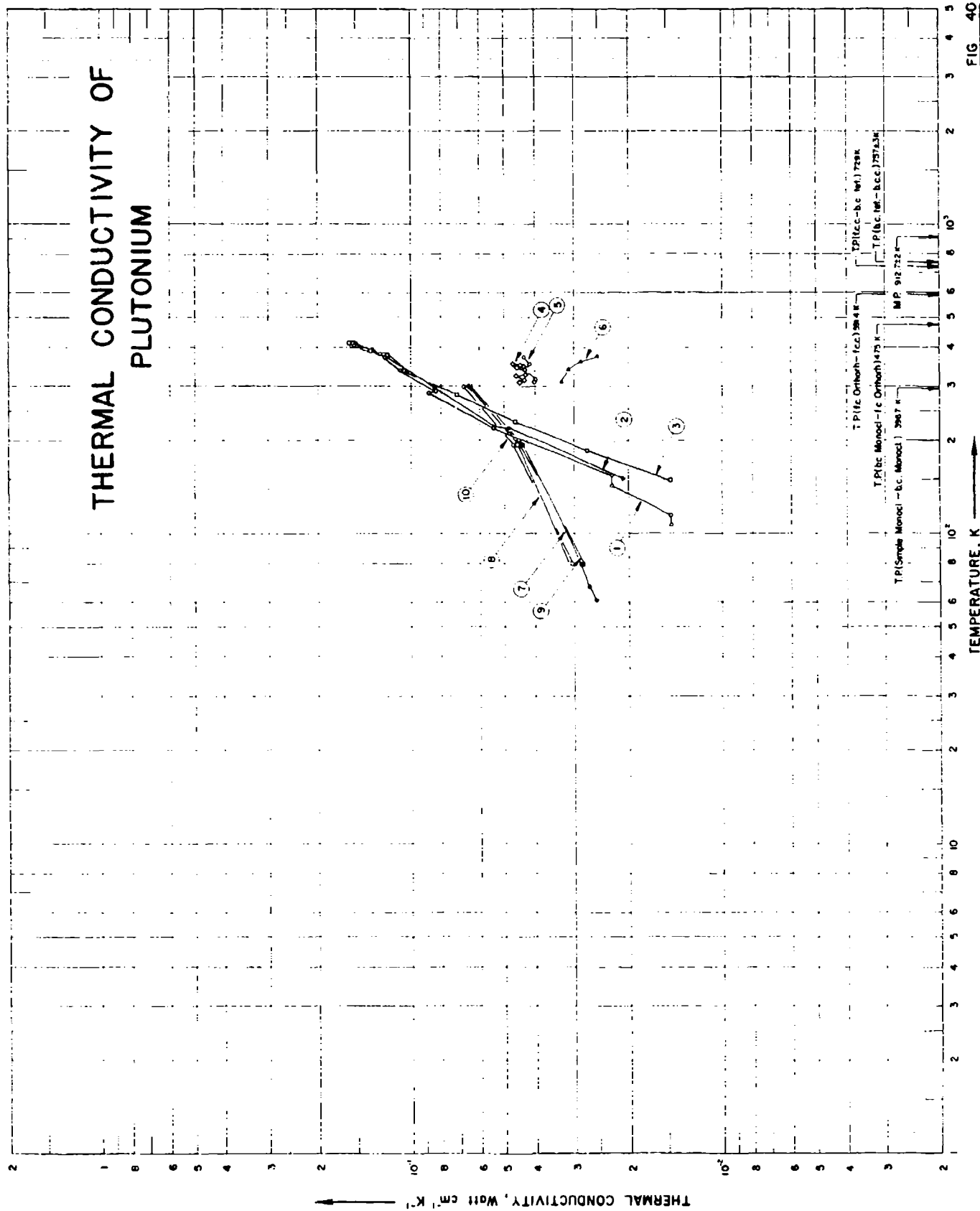


FIG. 40

SPECIFICATION TABLE NO. 40 THERMAL CONDUCTIVITY OF PLUTONIUM

(Impurity &lt; 0.20% each; total impurities &lt; 0.50%)

[ For Data Reported in Figure and Table No. 40 ]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition ( weight percent), Specifications and Remarks
1	921, 373	L	1957	108-413		1	99.95 pure; isotopic content: 94.76 Pu-239, 4.51 Pu-240, 0.29 Pu-241 and 0.04 Pu-242; specimen 0.231 in. in dia and 3.50 in. long; heat generation 0.01067 cal sec <sup>-1</sup> cm <sup>-2</sup> ; density 19.58 g cm <sup>-3</sup> ; phase transformation ( $\alpha$ to $\beta$ ) at 397.0 $\pm$ 0.2 K; electrical resistivity 61.1, 128.0, 157.0, 153.5, 146.3, 141.0, 107.5, 107.0, 106.2, 105.0, 97.7, 96.5, 99.4, and 107.1 $\mu$ ohm cm at 27.2, 50, 100, 150, 273, 380, 420, 475, 505, 590, 625, 725, 735, and 774 K, respectively.
2	921, 373	L	1957	152-413		2	99.97 pure; isotopic content: 95.35 Pu-239, 4.37 Pu-240, 0.26 Pu-241 and 0.03 Pu-242; specimen 0.250 in. in dia and 3.50 in. long; heat generation 0.01034 cal sec <sup>-1</sup> cm <sup>-2</sup> ; density 19.56 g cm <sup>-3</sup> ; phase transformation ( $\alpha$ to $\beta$ ) at 396.4 $\pm$ 0.1 K; electrical resistivity 68.5, 128.0, 156.8, 153.5, 146.6, 141.8, 109.5, 110.0, 109.5, 109.3, 103.0, 104.0, 104.8, 114.0, and 114.1 $\mu$ ohm cm at 25.8, 50, 100, 150, 273, 380, 420, 475, 505, 590, 625, 725, 735, 774, and 787 K, respectively.
3	921, 373	L	1957	148-413		3	99.93 pure; isotopic content: 95.33 Pu-239, 4.39 Pu-240, 0.28 Pu-241 and 0.06 Pu-242; specimen 0.227 in. in dia and 3.75 in. long; heat generation 0.01034 (assumed) cal sec <sup>-1</sup> cm <sup>-2</sup> ; density 19.53 g cm <sup>-3</sup> .
4	767	L	1958	309-357	5		Impurities 0.025; $\alpha$ -phase; specimen 1 in. in dia and 5 in. long; zone refined; self heating used as source of power during measurement; data extracted from two runs.
5	767	E	1958	309-374	1.5		Impurities 0.025; $\alpha$ -phase; 1.6 in. long; 0.08 in. dia; zone refined; cast.
6	767	E	1958	309-374			The above specimen, data corrected for emissivity (assumed to be 0.3).
7	281	E	1967	62-300	1	$\alpha$ -Plutonium	99.98 <sup>+</sup> pure; monoclinic crystalline; specimen 0.25 in. in dia and 1.81 in. long; arc-melted and induction cast into an MgO mold; density 19.62 g cm <sup>-3</sup> ; specimen had randomly oriented grains and a large number of microcracks.
8	281	E	1967	80-300	1	$\alpha$ -Plutonium	99.98 <sup>+</sup> pure; monoclinic crystalline; specimen 0.25 in. in dia and 1.81 in. long; arc-melted and cast into a mold at -40 C, then annealed at 110 C; density 19.77 g cm <sup>-3</sup> ; specimen had randomly oriented grains and very few microcracks.
9	281	E	1967	80-300	1	$\alpha$ -Plutonium	99.98 <sup>+</sup> pure; specimen 0.25 in. in dia and 1.80 in. long with long axis parallel to preferential alignment of the (020) plane in the monoclinic crystals; prepared by heating the cast ingot into the beta-phase temperature range and then cooling it to room temperature under a compressive load of 60,000 psi; density 19.77 g cm <sup>-3</sup> .
10	281	E	1967	80-300	1	$\alpha$ -Plutonium	Similar to the above specimen except the long axis aligned perpendicular to the (020) plane of the monoclinic crystal.



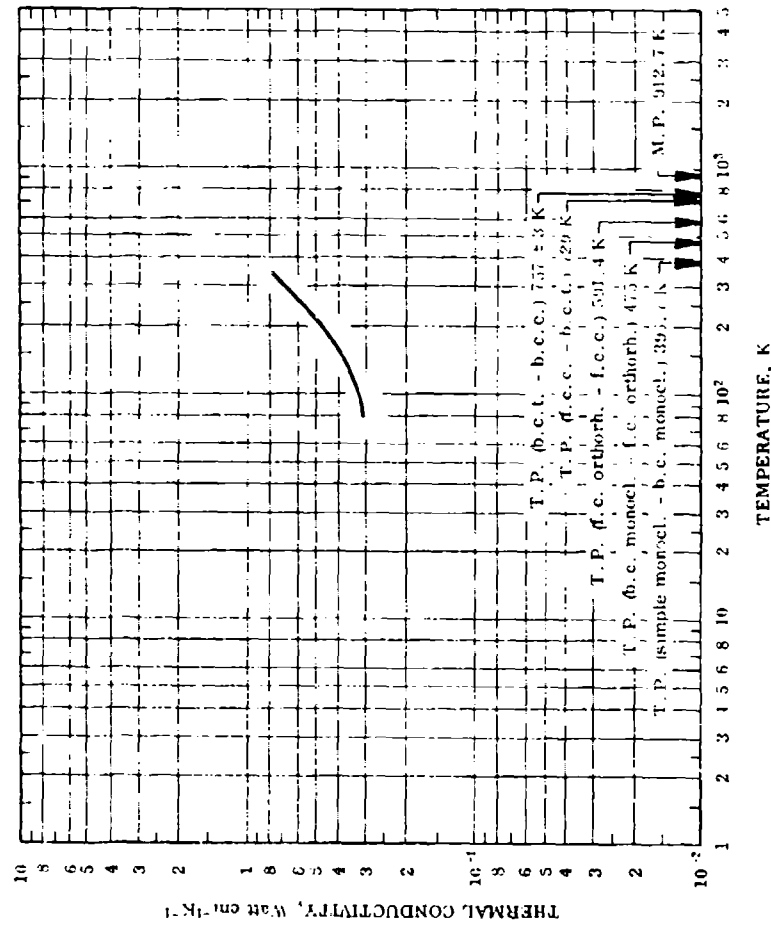
DATA TABLE NO. 40 THERMAL CONDUCTIVITY OF PLUTONIUM

(Impurity < 0.2%; each; total impurities < 0.50%)  
 (Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup>K<sup>-1</sup>)

T	k	T	k	T	k
<u>CURVE 1</u>					
107.5	0.0146	339.2	0.0431	80.0	0.0306
115.0	0.0146	346.2	0.0456	195.0	0.0467
143.0	0.0226	347.2	0.0435	300.0	0.0674
155.0	0.0226	349.2	0.0448		
220.0	0.0544	349.7	0.0433		
286.0	0.0879	351.2	0.0444		
298.0	0.0837	356.7	0.0469		
337.0	0.1088	<u>CURVE 5</u>			
373.0	0.1213	309.2	0.0404		
383.0	0.1255	316.7	0.0400		
405.0	0.1527	345.7	0.0433		
413.0	0.1548	354.2	0.0416		
		374.7	0.0433		
<u>CURVE 2</u>					
152.0	0.0209	<u>CURVE 6</u>			
220.0	0.0481	309.2	0.0331		
222.5	0.0536	340.2	0.0312		
295.0	0.0849	360.7	0.0285		
337.0	0.1021	374.7	0.0251		
373.0	0.1192	<u>CURVE 7</u>			
390.0	0.1320	61.5	0.0256		
405.0	0.1464	66.0	0.0268		
413.0	0.1506	80.0	0.0285		
		195.0	0.0436		
		300.0	0.0645		
<u>CURVE 3</u>					
148.0	0.0146	<u>CURVE 8</u>			
187.0	0.0272	80.0	0.0313		
230.0	0.0460	195.0	0.0452		
280.0	0.0711	300.0	0.0648		
337.0	0.1050	<u>CURVE 9</u>			
375.0	0.1192	80.0	0.0281		
390.0	0.1339	195.0	0.0439		
405.0	0.1548	300.0	0.0651		
413.0	0.1590	<u>CURVE 4</u>			
		309.2	0.0448		
		314.7	0.0433		
		325.2	0.0460		
		328.2	0.0429		

Not shown on plot

FIGURE AND TABLE NO. 40R RECOMMENDED THERMAL CONDUCTIVITY OF PLUTONIUM



RECOMMENDED VALUES<sup>a</sup>  
(For Polycrystalline)

T <sub>1</sub>	k <sub>1</sub>	k <sub>2</sub>	T <sub>2</sub>
0	0	0	-459.7
30	0.0306	1.77	-315.7
90	0.0318	1.84	-297.7
100	0.0331	1.91	-279.7
150	0.0399	2.31	-189.7
200	0.0476	2.75	-99.7
250	0.0564	3.28	-9.7
273.2	0.0616	3.56	32.0
300	0.0674	3.89	80.3
350	0.0790	4.56	170.3

REMARKS

The recommended values are for well-annealed 99.98% pure plutonium. The recommended values are thought to be accurate to within 10% of the true values near room temperature and 10 to 20% at other temperatures.

<sup>a</sup> T<sub>1</sub> in K, k<sub>1</sub> in Watt cm<sup>-1</sup> K<sup>-1</sup>, T<sub>2</sub> in F, and k<sub>2</sub> in Btu hr<sup>-1</sup> ft<sup>-1</sup> F<sup>-1</sup>.

# THERMAL CONDUCTIVITY OF POTASSIUM

FIGURE SHOWS ONLY 21 OF THE CURVES REPORTED IN TABLE

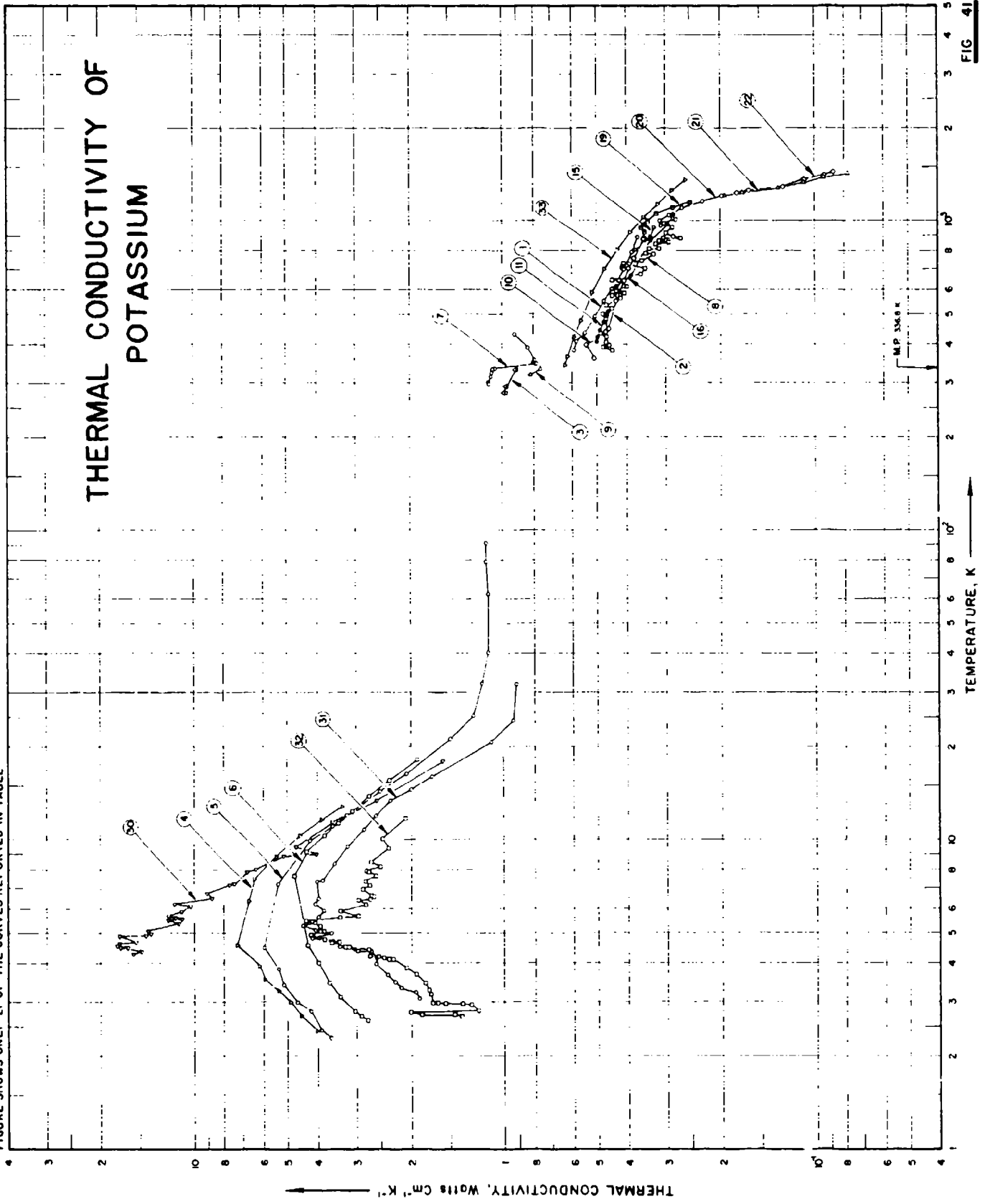


FIG 41

SPECIFICATION TABLE NO. 41 THERMAL CONDUCTIVITY OF POTASSIUM

(Impurity is 0.20% each; total impurities is 0.5%).

[ For Data Ref. in Figure and Table No. 41 ]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition ( weight percent), Specifications and Remarks
1	105	P	1956	384-891			Pure; in liquid state; thermal conductivity values calculated from measured (in argon) thermal diffusivity data using the specific heat and density values given in Liquid Metals Handbook (Lyon, R., Editor), 2nd Edition, 1952.
2	502, 38	L	1951	448-883	1		< 0.001 O, < 0.00001 each of Na, Ca, Al, 3b, and Li; distilled; in liquid state.
3	72	E	1913	278-331			Pure; trace of N; supplied by Eimer and Amend; electrical resistivity reported as 6.492, 6.442, 7.015, 7.035, 6.980, 8.353, and 8.338 $\mu$ ohm cm at 5.0, 5.0, 20.6, 20.7, 20.9, 57.4, and 57.8 C, respectively.
4	92	L	1956	2.4-13		K 1	Very pure; 1.3 mm dia; electrical resistivity ratio $\rho(295 \text{ K})/\rho(0 \text{ K}) = 532$ (using Hackspill's value $\rho(295 \text{ K}) = 7.08 \mu$ ohm cm); Lorenz function L (0 K) = $2.55 \times 10^{-6} \text{ } \mu\text{K}^{-1}$ .
5	92	L	1956	2.3-91		K 2	Very pure; 2.1 mm dia; electrical resistivity ratio $\rho(295 \text{ K})/\rho(0 \text{ K}) = 513$ (using Hackspill's value $\rho(295 \text{ K}) = 7.98 \mu$ ohm cm); Lorenz function L (0 K) = $2.31 \times 10^{-6} \text{ } \mu\text{K}^{-1}$ .
6	92	L	1956	2.6-18		K 4	Very pure; 1.3 mm dia; electrical resistivity ratio $\rho(295 \text{ K})/\rho(0 \text{ K}) = 325$ (using Hackspill's value $\rho(295 \text{ K}) = 7.02 \mu$ ohm cm); Lorenz function L (0 K) = $2.49 \times 10^{-6} \text{ } \mu\text{K}^{-2}$ .
7	355	L	1940	298-433	2-4		Doubly distilled; measured across melting point (approx 62 C).
8	919, 592, 920	L	1958	382-1007			M.P. 63.7 C; specimen in liquid state; measured in vacuum of $\sim 4 \times 10^{-4}$ mm Hg.
9	763	C	1963	319, 333			0.1 Na, 0.0050 Rb, 0.0035 O, 0.0030 Li, < 0.0010 each of Cs, Zr, Fe, Co, and Ni; Nb-1 Zr alloy used as comparative material.
10	766, 854, 855, 856	C	1963	360-449			Liquid state; same pretest impurities as the above specimen; additional impurities after test: 0.00105 Nb and 0.00015 Zr (contaminated from specimen container, made from Nb-1 Zr alloy); electrical resistivity reported as 15.4, 21.5, 28.4, 35.8, 44.4, 54.7, 66.4, 79.5, 93.8, 110, 131, 145, and 153 $\mu$ ohm cm at 373, 473, 573, 673, 773, 873, 973, 1073, 1173, 1273, 1373, 1423, and 1448 K, respectively; Nb-1 Zr alloy used as comparative material; run A, equilibrium 1.
11	766, 854, 855, 856	C	1963	408-511			Run A, equilibrium 2 of the above specimen.
12	766, 854, 855, 856	C	1963	412-501			Run A, equilibrium 3 of the above specimen.
13	766, 854, 855, 856	C	1963	491-658			Run A, equilibrium 4 of the above specimen.
14	766, 854, 855, 856	C	1963	529-787			Run A, equilibrium 5 of the above specimen.
15	766, 854, 855, 856	C	1963	584-953			Run A, equilibrium 6 of the above specimen.
16	766, 854, 855, 856	C	1963	611-1054			Run A, equilibrium 7 of the above specimen.

SPECIFICATION TABLE NO. 41 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
17	766, 854, 855, 856	C	1963	604-822			Same specimen as above; run B, equilibrium 1.
18	766, 854, 855, 856	C	1963	745-897			Run B, equilibrium 2 of the above specimen.
19	766, 854, 855, 856	C	1963	923-1154			Run B, data set 3, reading 1 of the above specimen.
20	766, 854, 855, 856	C	1963	970-1210			Run B, data set 3, reading 2 of the above specimen.
21	766, 854, 855, 856	C	1963	1219-1435			Run B, data set 3, reading 14 of the above specimen.
22	766, 854, 855, 856	C	1963	1234-1449			Run B, data set 3, reading 15 of the above specimen.
23	148, 857		1966	700-1500	± 30		Vapor; measured in the 1 mm gap between concentric cylinders 900 mm long, vapor pressure = 0.01 kg cm <sup>-2</sup> .
24	148, 857		1966	800-1500	± 20		Similar to the above except vapor pressure = 0.05 kg cm <sup>-2</sup> .
25	148, 857		1966	900-1500	± 20		Similar to the above except vapor pressure = 0.1 kg cm <sup>-2</sup> .
26	148, 857		1966	1000-1500	± 20		Similar to the above except vapor pressure = 0.5 kg cm <sup>-2</sup> .
27	148, 857		1966	1100-1500	± 20		Similar to the above except vapor pressure = 1.0 kg cm <sup>-2</sup> .
28	148, 857		1966	1100-1500	± 20		Similar to the above except vapor pressure = 2.0 kg cm <sup>-2</sup> .
29	148, 857		1966	700-1100	± 20		Similar to the above except measured on saturate curve.
30	858	L	1967	4.3-18	7	K-9	99.97 pure; single crystal; rectangular specimen with length/cross-sectional area = 12; material obtained from Mine Safety Appliances Corp.; prepared by growing from the bulk material using the Bridgman technique.
31	858	L	1967	3.1-32	7	ZK-1	Originally 99.97 pure; single crystal; rectangular specimen with length to cross-sectional area ratio = 12; material obtained from Mine Safety Appliances Corp.; prepared by melting in a stainless-steel boat, zone refined at a rate of 1 in. hr <sup>-1</sup> for 16 passes, grown by using the Bridgman technique. The zone refining technique is believed to have introduced impurities not present in the original bulk material.
32	858	L	1967	2.7-12	7	ZK-2	Similar to the above specimen
33	859, 870, 871	-	1965	341-1366			Specimen in liquid state; density reported as 0.7851, 0.7434, 0.7161, 0.6887, 0.6664, 0.6270, 0.6024, and 0.5861 g cm <sup>-3</sup> at 520.5, 701.3, 827.7, 944.3, 1048, 1206, 1302, and 1374 K, respectively; electrical resistivity reported as 7.02, 7.32, 7.54, 8.05, 15.05, 17.96, 20.31, 24.83, 28.34, 32.64, 37.84, 41.43, 47.70, 51.81, 58.51, 65.94, 71.44, 81.14, 87.82, 98.61, 106.63, 119.87, and 130.61 μ ohm at 296, 309, 314, 329, 376, 431, 476, 541, 591, 648, 712, 755, 822, 863, 926, 988, 1031, 1102, 1144, 1210, 1253, 1319 and 1365 K, respectively; thermal conductivity values calculated from measured electrical resistivity data and the Lorenz number $2.45 \times 10^{-8} \text{ V}^2 \text{ K}^{-2}$ .

SPECIFICATION TABLE NO. 41 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)	Specifications and Remarks
34	786, 555	-	1963	473-1423	< E 2	Run No. 1		Thermal conductivity values calculated from the measurements of electrical resistivity, reported as 15.4, 21.5, 28.4, 35.8, 44.4, 54.7, 56.4, 74.2, 79.5, 93.8, 110, 121, and 145 $\mu$ ohm cm at 100, 200, 300, 400, 500, 600, 700, 800, 900, 1000, 1100, and 1150 C, respectively; Lorenz Number assumed to be $2.14 \times 10^{-8}$ V $^2$ K $^{-2}$ based upon experimental information.
35	563	-	1966	845.2		Run No. 1		Vapor specimen filled in a test cell <sup>8</sup> in. long connected with a boiler; boiler pressure -- 10 mm Hg; thermal conductivity measured by using the dynamic hot-wire method.
36	563	-	1966	847.2		Run No. 2		Same as above except boiler pressure 47 mm Hg.
37	563	-	1966	914.2		Run No. 3		Same as above except boiler pressure 73 mm Hg.
38	563	-	1966	978.2		Run No. 4		Same as above except boiler pressure 80 mm Hg.
39	563	-	1966	983.2		Run No. 5		Same as above; in different run.
40	563	-	1966	1034		Run No. 6		Same as above except boiler pressure 78 mm Hg.
41	563	-	1966	1035		Run No. 7		Same as above except boiler pressure 1417 mm Hg.
42	563	-	1966	1036		Run No. 8		Same as above; in different run.
43	563	-	1966	1116		Run No. 9		Same as above except boiler pressure 1144 mm Hg.
44	756	-	1962	473-1073				0.52 Na, 0.02 Fe, and 0.004 O (poorest); molten specimen contained in a type 347 stainless-steel tube; supplied by Fisher Scientific Co.; electrical resistivity 8.07, 8.24, 8.59, 8.82, 9.47, 9.81, 14.77, 15.48, 17.90, 21.86, 26.06, 29.57, 34.11, 38.32, 43.30, 48.47, 54.04, 59.47, 60.02, 66.75, and 74.30 $\mu$ ohm cm at 25.3, 29.4, 38.3, 51.4, 58.3, 59.2, 79.4, 91.9, 140.8, 205.0, 256.4, 313.9, 373.6, 429.2, 481.9, 542.8, 583.3, 646.9, 651.4, 706.7, and 764.2 C, respectively; thermal conductivity values calculated from measured electrical resistivity data and the Lorenz function of 2.07, 2.11, 2.14, 2.17, 2.21, 2.29, and $2.34 \times 10^{-8}$ V $^2$ K $^{-2}$ at 200, 300, 400, 500, 600, 700, and 750 C, respectively; the first five values being derived from the thermal conductivity measurements of Ewing, C. T., and Grand, J. A. (NRL Report 3835, 1951) and the authors' own electrical resistivity data.

DATA TABLE NO. 41 THERMAL CONDUCTIVITY OF POTASSIUM

(Impurity 0.02% each; total impurities 0.50%)

(Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup> K<sup>-1</sup>)

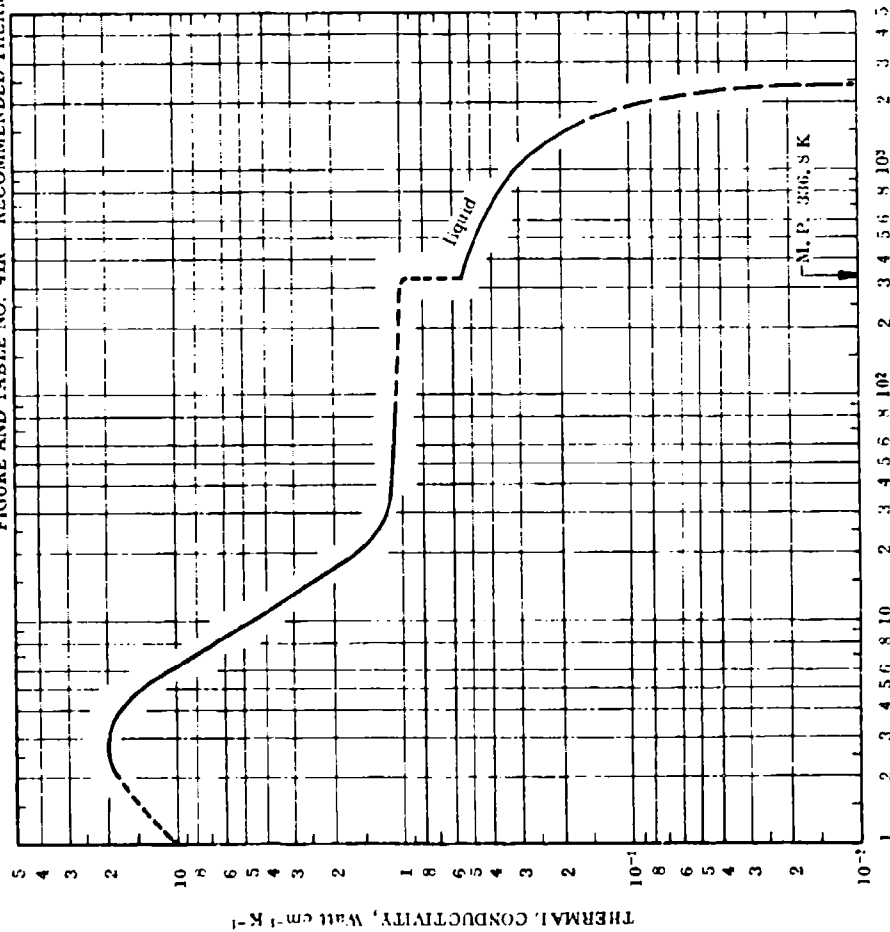
T	k	T	k	T	k	T	k	T	k	T	k	T	k
<u>CURVE 1</u>													
383.8	0.591	8.84	5.407	12.43	3.680	466.2	0.477	673.2	0.360	311.2	0.917	737.2	0.373
408.5	0.581	10.30	4.552	13.82	2.713	473.7	0.488	676.7	0.413	333.2	0.760	787.2	0.375
435.6	0.545	11.70	3.802	15.00	2.345	474.2	0.457	677.2	0.389	<u>CURVE 15</u>			
490.8	0.506	12.80	3.333	18.13	1.908	480.2	0.459	659.2	0.403	360.2	0.508	594.2	0.451 <sup>o</sup>
554.4	0.472	<u>CURVE 5</u>											
606.1	0.444	2.29	3.632	297.7	1.103	483.2	0.465	709.7	0.349	398.2	0.541	632.2	0.412 <sup>o</sup>
697.9	0.407	2.43	3.908	298.7	1.102	489.2	0.456	710.6	0.384	427.2	0.469	707.2	0.411
794.4	0.384	2.80	4.205	314.0	1.09	497.2	0.451	713.2	0.404	439.2	0.477	809.2	0.375
890.8	0.368	2.99	4.667	323.2	1.98	506.2	0.451	726.2	0.378	449.2	0.477	893.2	0.336
<u>CURVE 2</u>													
448.0	0.455	3.40	5.126	336.2	1.97	509.2	0.465	730.2	0.372	498.2	0.499	954.2	0.326
549.6	0.431	3.83	5.333	341.2	1.062	518.2	0.464	730.7	0.407	<u>CURVE 11</u>			
641.9	0.408	4.49	5.954	353.2	1.06	522.2	0.442	735.2	0.365	422.2	0.496	<u>CURVE 16</u>	
700.0	0.394	7.20	5.345	346.2	0.785	533.7	0.439	745.7	0.366	611.2	0.434		
757.2	0.377	9.91	4.210	347.2	0.795	534.2	0.463	746.2	0.386	667.2	0.386		
875.2	0.353	11.05	3.586	348.2	0.798	538.7	0.446	759.2	0.348	734.2	0.381		
883.1	0.352	12.08	3.218	354.7	0.792	543.2	0.460	768.2	0.358	777.2	0.336		
<u>CURVE 3</u>													
274.2	0.971	14.40	2.506	355.7	0.792	553.2	0.430	785.2	0.328	829.2	0.295		
278.2	0.992	16.39	2.069	360.7	0.835	553.7	0.452	787.7	0.349	979.2	0.285		
293.8	0.979	21.20	1.494	391.2	0.835	561.2	0.418	790.7	0.374	1054.2	0.282		
294.1	0.971	25.23	1.264	403.2	0.910	568.2	0.430	799.2	0.341	<u>CURVE 17</u>			
330.6	0.904	32.06	1.180	571.2	0.442	571.2	0.442	799.7	0.321	412.2	0.500		
<u>CURVE 4</u>													
2.42	4.046	40.10	1.126	587.2	0.407	587.2	0.407	814.2	0.339	449.2	0.490	604.2	0.431
2.71	4.529	62.30	1.126	595.2	0.397	595.2	0.397	818.2	0.314	633.2	0.500	633.2	0.423
2.99	4.897	79.00	1.149	598.7	0.430	598.7	0.430	819.5	0.339	478.2	0.407	678.2	0.407
3.27	5.333	91.00	1.149	604.2	0.430	604.2	0.430	826.2	0.308	501.2	0.435	740.2	0.384
3.55	5.908	382.2	0.451	610.2	0.444	610.2	0.444	845.2	0.323	<u>CURVE 13</u>			
3.91	6.200	394.2	0.477	613.7	0.399	613.7	0.399	855.2	0.322	491.2	0.473	788.2	0.363
4.54	7.287	403.2	0.465	619.7	0.455	619.7	0.455	856.7	0.291	519.2	0.449	822.2	0.338
6.35	6.690	409.2	0.465	624.2	0.418	624.2	0.418	863.7	0.291	<u>CURVE 18</u>			
7.46	6.437	413.2	0.460	629.2	0.404	629.2	0.404	864.2	0.294	519.2	0.449	840.2	0.361
<u>CURVE 6</u>													
2.62	2.759	421.2	0.404	633.2	0.414	633.2	0.414	884.2	0.305	561.2	0.464	857.2	0.358
2.70	2.897	428.2	0.464	638.2	0.401	638.2	0.401	888.2	0.267	615.2	0.465	874.2	0.338
2.99	4.897	434.2	0.485	641.2	0.420	641.2	0.420	893.2	0.281	554.2	0.407	745.2	0.389
3.27	5.333	440.7	0.465	646.2	0.442	646.2	0.442	916.2	0.298	706.2	0.389	797.2	0.399
3.55	5.908	447.2	0.477	657.2	0.414	657.2	0.414	924.2	0.291	757.2	0.399	840.2	0.361
3.91	6.200	453.2	0.475	660.2	0.416	660.2	0.416	944.2	0.289	797.2	0.399	874.2	0.338
4.54	7.287	455.2	0.451	664.2	0.418	664.2	0.418	958.2	0.285	829.2	0.422	897.2	0.358
6.35	6.690	459.2	0.453	669.2	0.395	669.2	0.395	983.2	0.303	857.2	0.430		
7.46	6.437	463.2	0.466	669.2	0.395	669.2	0.395	1007.2	0.277	897.2	0.402		

Not shown on plot.





FIGURE AND TABLE NO. 41R RECOMMENDED THERMAL CONDUCTIVITY OF POTASSIUM



RECOMMENDED VALUES*					
(In Liquid State)					
$T_1$	$k_1$	$k_2$	$T_2$	$k_3$	$T_3$
0	0	0	-459.7	0.479	500
1	(10.1) <sup>†</sup>	(384)	-457.9	0.548	336.8
2	(17.6)	(1020)	-456.1	0.642	300
3	19.6	1130	-454.3	0.520	400
4	17.5	1010	-452.5		
5	14.0	809	-450.7		
6	10.8	624	-448.9		
7	8.33	481	-447.1		
8	6.62	383	-445.3		
9	5.42	313	-443.5		
10	4.58	265	-441.7		
11	3.97	229	-439.9		
12	3.47	200	-438.1		
13	3.06	177	-436.3		
14	2.74	158	-434.5		
15	2.48	143	-432.7		
16	2.26	131	-430.9		
18	1.91	110	-427.3		
20	1.67	96.5	-423.7		
25	1.33	76.8	-414.7		
30	1.23	71.1	-405.7		
35	1.18	68.2	-396.7		
40	1.15	66.4	-387.7		
45	1.13	65.3	-378.7		
50	1.12	64.7	-369.7		
60	1.10	63.6	-351.7		
70	1.09	63.0	-333.7		
80	1.08	62.4	-315.7		
90	1.08	62.4	-297.7		
100	(1.07)	(61.8)	-279.7		
150	(1.05)	(60.7)	-189.7		
200	(1.04)	(60.1)	-99.7		
250	(1.04)	(60.1)	-9.7		
273.2	(1.04)	(60.1)	32.0		
300	1.02	58.9	80.3		
336.8	0.985	56.9	146.6		

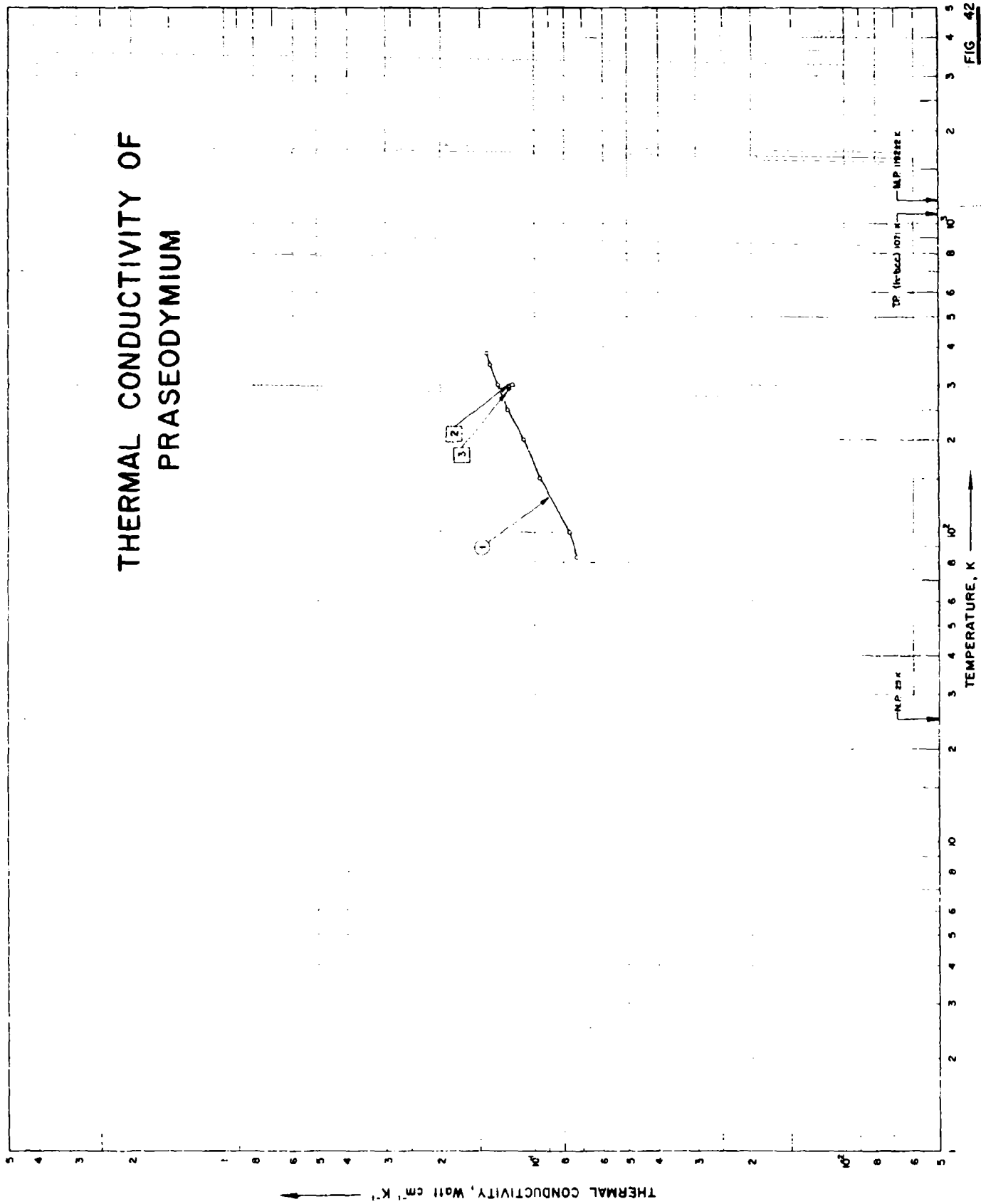
REMARKS

The recommended values are for 99.97% pure potassium with residual electrical resistivity  $\rho_0 = 0.00237 \mu\Omega$  cm (characterization by  $\rho_0$  becomes important at temperatures below about 80 K). The values below 1.5 T m are calculated to fit the experimental data by using  $n = 2.00$ ,  $\alpha' = 2.06 \times 10^{-3}$ , and  $\beta = 0.0973$ . The recommended values that are supported by experimental thermal conductivity data are thought to be accurate to within 5% of the true values near room temperature and 5 to 10% at other temperatures.

\*  $T_1$  in K,  $k_1$  in Watt  $\text{cm}^{-1} \text{K}^{-1}$ ,  $T_2$  in F, and  $k_2$  in  $\text{Btu lb}^{-1} \text{ft}^{-1} \text{F}^{-1}$ .

<sup>†</sup> Values in parentheses are extrapolated, interpolated, or estimated.

# THERMAL CONDUCTIVITY OF PRASEODYMIUM



## SPECIFICATION TABLE NO. 42 THERMAL CONDUCTIVITY OF PRASEODYMIUM

[ For Data Reported in Figure and Table No. 42 ]

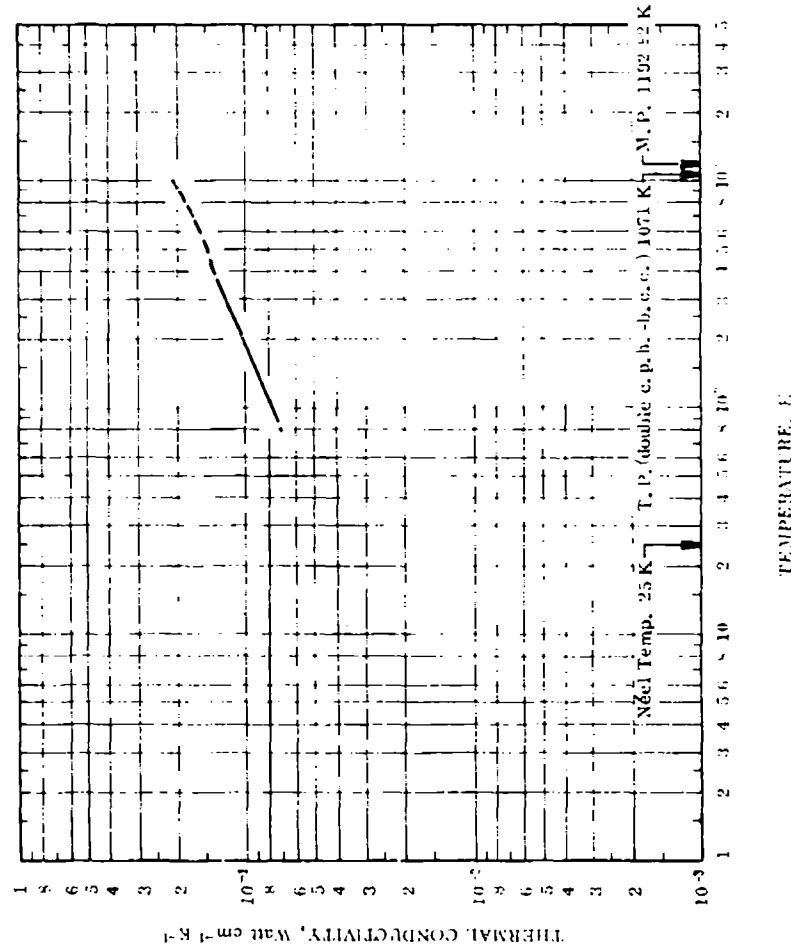
Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition ( weight percent), Specifications and Remarks
1	810	I	1964	83-350			Impurities: 0.5 La, 0.04 Cu, 0.03 Fe, and 0.01 Ca; prepared by briquetting powder under a pressure of approx 8000 Kg cm <sup>-2</sup> and annealing in vacuo (~1 x 10 <sup>-4</sup> mm Hg) for 1-2 hrs at 1600-1800 C; measured in vacuo of ~5 x 10 <sup>-6</sup> mm Hg; data taken from smoothed curve of measurements on several specimens.
2	811		1954	301	10		No details reported.
3	256	C	1966	291	± 4		~0.1 Ta, 0.1 other rare earth metals, and ~0.03 other base metals; high purity polycrystalline specimen 1 cm in dia, 1.2 cm long; electrical resistivity 66 μohm cm at 291 K; data proposed by the author from measurements of 2 different thermal comparators.

DATA TABLE NO. 42 THERMAL CONDUCTIVITY OF PRASEODYMIUM

[Temperature, T, K, Thermal Conductivity, k, Watt cm<sup>-1</sup>K<sup>-1</sup>]

T	k
<u>CURVE 1</u>	
83.0	0.0728
100.0	0.0768
150.0	0.0962
200.0	0.108
250.0	0.121
300.0	0.131
350.0	0.139
380.0	0.143
<u>CURVE 2</u>	
301.0	0.117
<u>CURVE 3</u>	
291.0	0.120

FIGURE AND TABLE NO. 42R RECOMMENDED THERMAL CONDUCTIVITY OF PRASEODYMIUM



RECOMMENDED VALUES\*  
(For Polycrystalline)

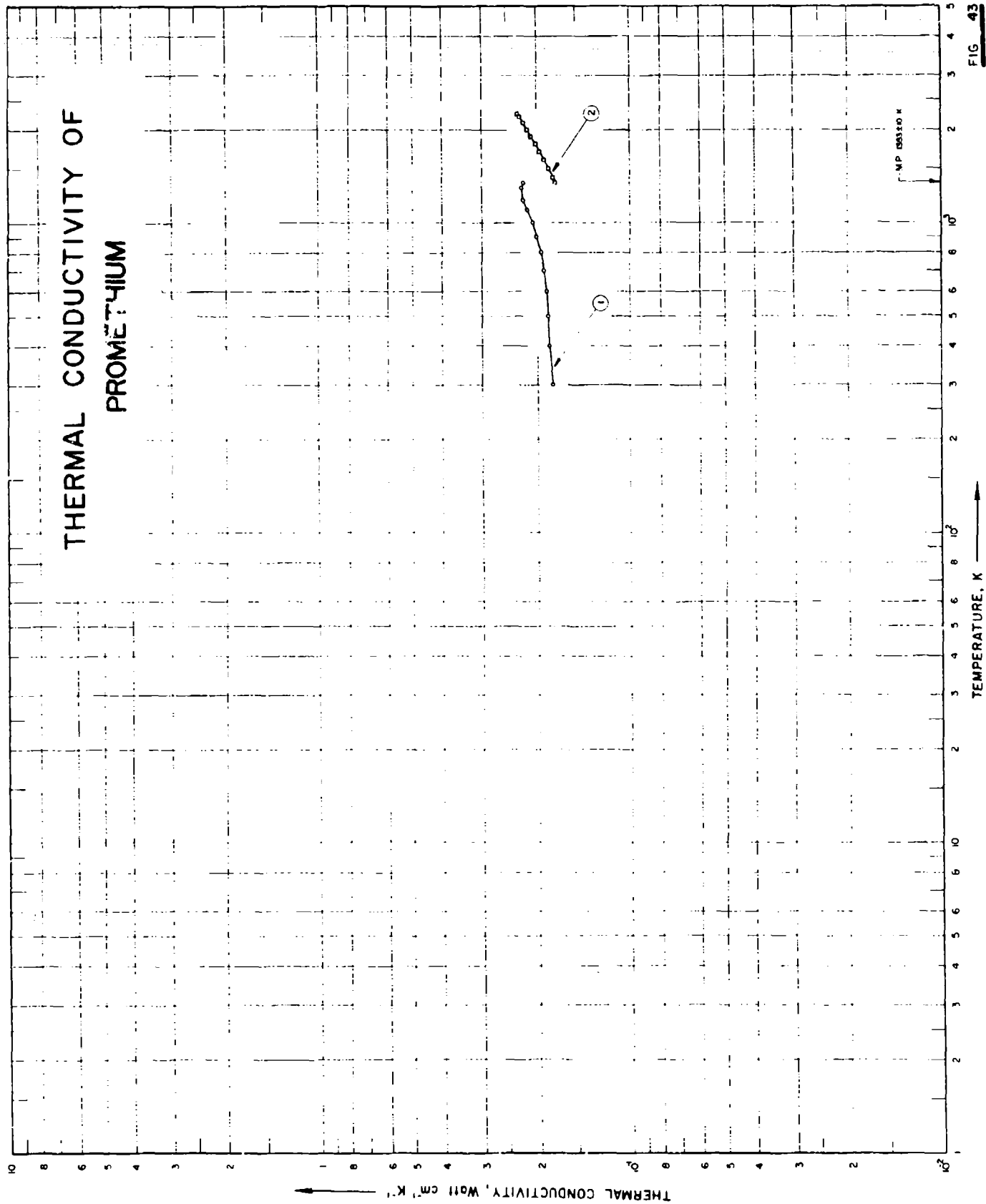
$T_1$	$k_1$	$k_2$	$T_2$
0	0	0	-459.7
80	(0.0592) <sup>‡</sup>	(4.00)	-315.7
90	0.0732	4.23	-297.7
100	0.0769	4.44	-279.7
150	0.0926	5.35	-189.7
200	0.106	6.12	-99.7
250	0.116	6.70	-9.7
273.2	0.120	6.93	32.0
300	0.125	7.22	90.3
350	0.132	7.63	170.3
400	(0.136)	(7.86)	260.3
500	(0.147)	(8.49)	440.3
600	(0.157)	(9.07)	620.3
700	(0.169)	(9.76)	800.3
800	(0.184)	(10.5)	980.3
900	(0.200)	(11.6)	1160
1000	(0.216)	(12.5)	1340

REMARKS

The recommended values are for well-annealed 99.4% praseodymium. The recommended values that are supported by experimental thermal conductivity data are thought to be accurate to within 5% of the true values near room temperature and 5 to 15% at other temperatures.

\* $T_1$  in K,  $k_1$  in  $\text{Watt cm}^{-1} \text{K}^{-1}$ ,  $T_2$  in F, and  $k_2$  in  $\text{Btu lb}^{-1} \text{ft}^{-1} \text{F}^{-1}$ . <sup>‡</sup>Values in parentheses are extrapolated or estimated.

# THERMAL CONDUCTIVITY OF PROMETHIUM



## SPECIFICATION TABLE NO. 43 THERMAL CONDUCTIVITY OF PROMETHIUM

(Impurity &lt; 0.20% each; total impurities &lt; 0.50%)

[ For Data Reported in Figure and Table No. 43 ]

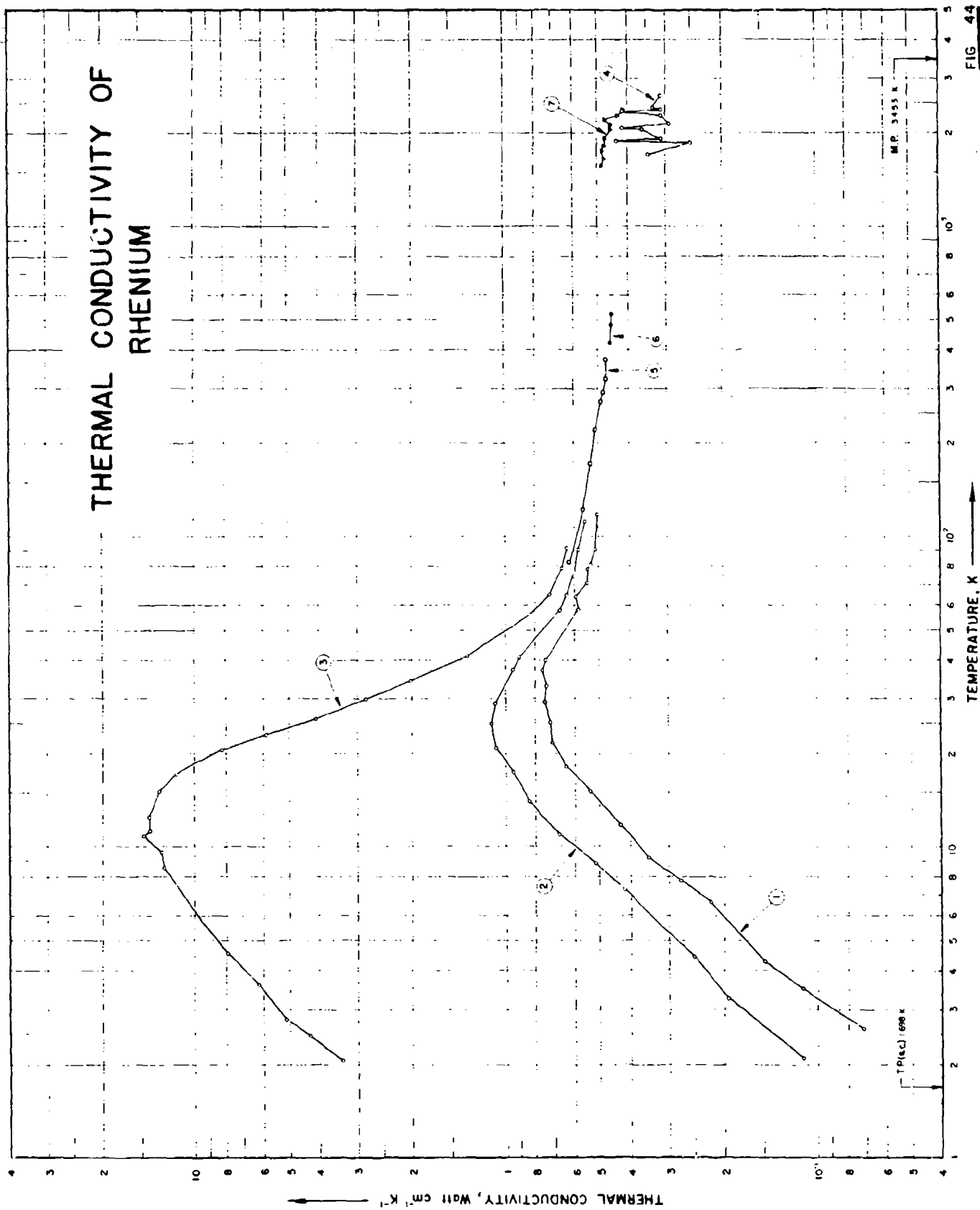
Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition ( weight percent).	Specifications and Remarks
1	999	-	1966	300-1353				Solid state; estimated thermal conductivity values given as the sum of electronic thermal conductivity and lattice thermal conductivity where electronic thermal conductivity values calculated from the theoretical Lorenz number $L_0 = 2.443 \times 10^{-8} \text{ V}^2 \text{ K}^{-2}$ and the estimated electrical resistivity reported as 54.0, 68.0, 80.0, 92.0, 102.0, 111.5, 120.5, 128.0, 134.5, 139.0 ( $\alpha$ ), 146.0 ( $\beta$ ), 149.0, and 152.0 $\mu\text{hm cm}$ at 300, 400, 500, 600, 700, 800, 900, 1000, 1100, 1185, 1185, 1300, and 1353 K, respectively, and lattice thermal conductivity calculated from the empirical equation $k_L = 16 \text{ T}^{-1}$ ; c. p. h. to b. c. c. transformation temp estimated to be 1185 K; Neel temp estimated to be 6 K.
2	999	-	1966	1353-2253				Liquid state; estimated thermal conductivity values given as the sum of electronic thermal conductivity and phonon conductivity where electronic thermal conductivity values calculated from estimated values of Lorenz number, and electrical resistivity, whereas phonon conductivity values ranging from 0.004 to 0.012 $\text{W cm}^{-1} \text{ C}^{-1}$ being based on predictions due to Rao, M. R. (Phys. Rev., 59, 212, 1941), Turnbull, A. G. (Aust. J. Appl. Sci., 12, 324-29, 1961) and Powell, R. W. (Amer. Soc. Mech. Engrs., 249-95, 1965); approx mean conductivity values given by $k = 0.110 + 6.0 \times 10^{-4} \text{ T}(\text{C}) \text{ W cm}^{-1} \text{ C}^{-1}$ .

DATA TABLE NO. 43 THERMAL CONDUCTIVITY OF PROMETHIUM  
(Impurity of 0.20% each; total impurities of 0.56%)

[Temperature, T, K; Thermal Conductivity,  $k$ , Watt cm<sup>-1</sup>K<sup>-1</sup>]

T	k
<u>CURVE 1</u>	
300	0.179
400	0.184
500	0.185
600	0.187
700	0.191
800	0.195
900	0.201
1000	0.207
1100	0.215
1185	0.222
1185	0.213
1300	0.223
1353	0.230
<u>CURVE 2</u>	
1353	0.175
1400	0.178
1500	0.184
1600	0.190
1700	0.196
1800	0.202
1900	0.208
2000	0.214
2100	0.220
2200	0.226
2253	0.229





## SPECIFICATION TABLE NO. 44 THERMAL CONDUCTIVITY OF RHENIUM

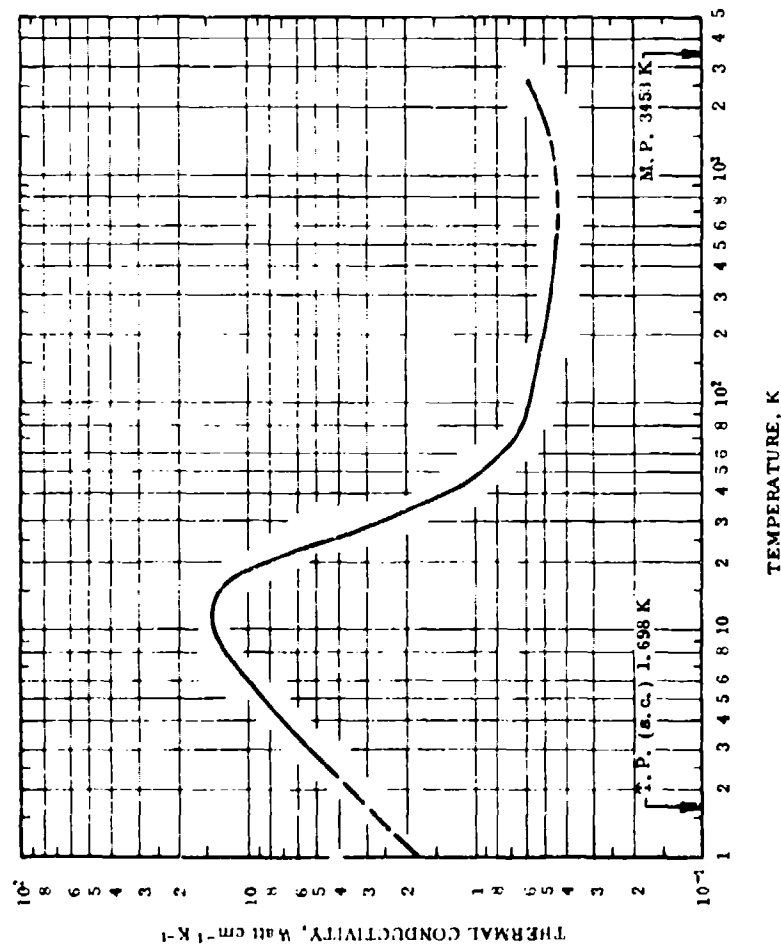
(Impurity &lt; 0.20% each; total impurities &lt; 0.5%)

For Data Reported in Figure and Table No. 44

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)	Specifications and Remarks
1	150	L	1957	2.6-118		Re 1	99.5% Re, total metallic impurities < 0.1, consisting of Cu, Fe, and Mo; cut from a rolled sheet 0.75 mm thick supplied by A. D. Mackay Inc.; density 21.3 g cm <sup>-3</sup> at room temp; residual electrical resistivity = 87 μohm cm; electrical resistivity ratio $\rho(295K)/\rho_0 = 24.9$ .	
2	150	L	1957	2.1-112		Re 2	Cut from the same sheet as the above specimen; annealed in vacuo at 700 C for 2 hrs; residual electrical resistivity 0.369 μohm cm; electrical resistivity ratio $\rho(295K)/\rho_0 = 40.7$ .	
3	150	L	1957	2.1-92		Re 4	Total impurities < 0.01; prepared by zone-melting rhenium powder in an argon arc furnace; 6 mm dia x 3-5 cm long; residual electrical resistivity 0.0139 μohm cm; electrical resistivity ratio $\rho(295K)/\rho_0 = 1337$ .	
4	667	E	1962	1700-2650	10		Spectrographically pure; 0.10 in. dia; electrical resistivity reported as 75.2, 84.9, 92.0, 95.2, 100.0, 103.9, 106.5, 108.3, 109.5, and 110.1 μohm cm at 1130, 1410, 1630, 1815, 1975, 2115, 2250, 2370, 2495, and 2605 K, respectively; measured in a vacuum of $\sim 10^{-5}$ mm Hg.	
5	610	L	1963	83-373			High purity; traces of noble metals; 7.0 cm long, 0.486 cm in dia; supplied by Johnson Matthey Co.; heat treated at 1390 C; electrical resistivity reported as 2.9, 5.8, 9.6, 13.3, 17.2, 18.8, 21.2, 24.9, 28.8, 32.6, and 36.1 μohm cm at 83, 123, 173, 223, 273, 293, 323, 373, 423, 473, and 523 K; residual electrical resistivity 0.078 μohm cm; density 20.99 g cm <sup>-3</sup> ; data taken from smooth curve.	
6	610	C	1963	423-523			The above specimen measured by comparative method using Armco iron as comparative material; data taken from smoothed curve.	
7	849	-	1966	1577-2397			0.0047 C and 0.001116 O; hexagonal; specimen 1.0711 cm in dia and 0.1582 cm thick; density 20.97 g cm <sup>-3</sup> ; thermal conductivity derived from the temp distribution on the flat surface of the cylindrical disc specimen heated in high vacuum ( $10^{-5}$ mm Hg) by high frequency induction generating localized heating within 0.005 in. of the surface at current frequency of 500,000 cps with heat lost only by radiation; the cylindrical surface being assumed isothermal, and the temp gradient along the radius was analytically correlated to the thermal conductivity.	



FIGURE AND TABLE NO. 44R RECOMMENDED THERMAL CONDUCTIVITY OF RHENIUM



## REMARKS

The recommended values are for well-annealed 99.99% pure rhenium with residual electrical resistivity  $\rho_0 = 0.0140 \mu\Omega \text{ cm}$  (characterization by  $\rho_0$  becomes important at temperatures below about 200 K). The values below 1.5  $T_m$  are calculated to fit the experimental data by using  $n = 2.50$ ,  $\alpha = 4.56 \times 10^{-5}$ , and  $\beta = 0.570$ . The recommended values that are supported by experimental thermal conductivity data are thought to be accurate to within 4% of the true values near room temperature and 4 to 10% at other temperatures.

RECOMMENDED VALUES\*  
(For Polycrystalline)

$T_1$	$k_1$	$k_2$	$T_2$	$T_1$	$k_1$	$k_2$	$T_2$
0	0	0	-459.7	500	0.448	25.9	440.3
1	(1.75) <sup>‡</sup>	(101)	-457.9	600	(0.442)	(25.5)	620.3
2	(3.51)	(203)	-456.1	700	(0.440)	(25.4)	800.3
3	5.25	303	-454.3	800	(0.441)	(25.5)	980.3
4	6.95	402	-452.5	900	(0.443)	(25.6)	1160
5	8.55	456	-450.7	1000	(0.446)	(25.8)	1340
6	10.1	584	-448.9	1100	(0.451)	(26.1)	1520
7	11.5	664	-447.1	1200	(0.457)	(26.4)	1700
8	12.6	728	-445.3	1300	(0.464)	(26.8)	1860
9	13.4	774	-443.5	1400	(0.471)	(27.2)	2060
10	14.0	809	-441.7	1500	(0.478)	(27.6)	2240
11	14.3	826	-439.9	1600	0.486	28.1	2420
12	14.2	820	-438.1	1700	0.493	28.5	2600
13	14.0	809	-436.3	1800	0.500	28.9	2780
14	13.5	790	-434.5	1900	0.509	29.4	2960
15	12.9	745	-432.7	2000	0.519	30.0	3140
16	12.1	699	-430.9	2200	0.539	31.1	3500
18	10.3	595	-427.3	2400	0.563	32.5	3860
20	8.36	483	-423.7	2600	0.592	34.2	4220
25	4.55 <sup>‡</sup>	263	-414.7				
30	2.7 <sup>‡</sup>	161	-405.7				
35	1.90	110	-396.7				
40	1.41	81.5	-387.7				
45	1.13	65.3	-378.7				
50	0.962	55.6	-369.7				
60	0.774	44.7	-351.7				
70	0.678	39.2	-333.7				
80	0.629	36.3	-315.7				
90	0.606	35.0	-297.7				
100	0.589	34.0	-279.7				
150	0.538	31.1	-189.7				
200	0.510	29.5	-99.7				
250	0.492	28.4	-9.7				
273.2	0.486	28.1	32.0				
300	0.479	27.7	80.3				
350	0.470	27.2	170.3				
400	0.461	26.6	260.3				

\*  $T_1$  in K,  $k_1$  in  $\text{Watt cm}^{-1} \text{K}^{-1}$ ,  $T_2$  in F, and  $k_2$  in  $\text{Btu hr}^{-1} \text{ft}^{-1} \text{F}^{-1}$ . <sup>‡</sup>Values in parentheses are extrapolated or interpolated.

# THERMAL CONDUCTIVITY OF RHODIUM

FIGURE SHOWS ONLY 11 OF THE CURVES REPORTED IN TABLE

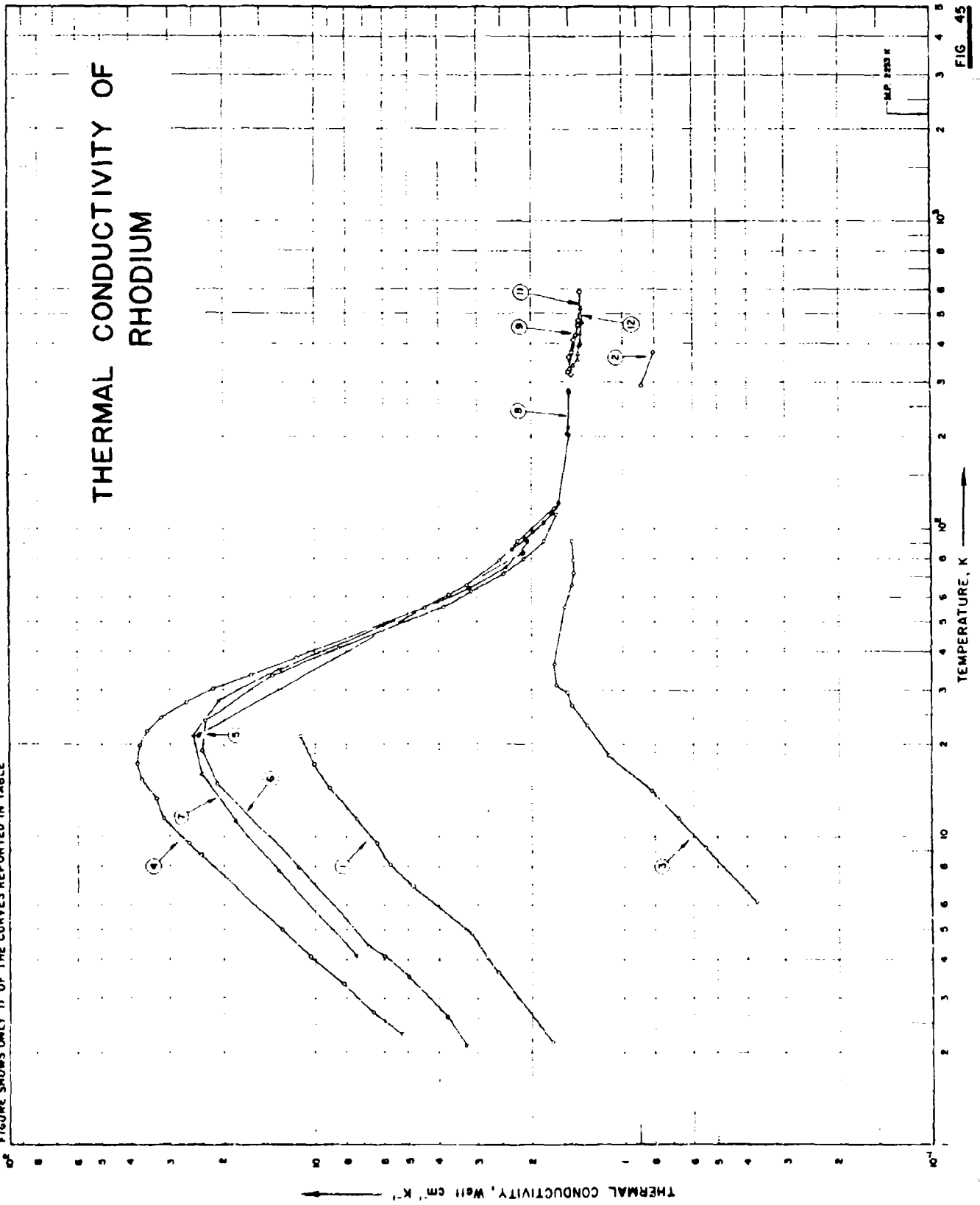


FIG. 45

## SPECIFICATION TABLE NO. 45 THERMAL CONDUCTIVITY OF RHODIUM

(Impurity &lt; 0.20% each, total impurities &lt; 0.50%)

[ For Data Reported in Figure and Table No. 45 ]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition ( weight percent), Specifications and Remarks
1	97	L	1952	2.2-21		JM 2357; Rh 1	99.995 pure; 1-2 mm dia x 5 cm long; supplied by Johnson Matthey.
2	8	F	1914	290-373			Pure; 0.1030 x 0.1030 x 10.0 cm; electrical resistivity reported as 4.811 and 6.211 $\mu\text{ohm cm}$ at 0 and 100 C, respectively; specific gravity 12.505.
3	149	L	1937	6.1-91		Rh 1	99.9 pure; 1.5 mm dia; supplied by Baker Platinum Co.; annealed at 1050 C; residual electrical resistivity 0.44 $\mu\text{ohm cm}$ ; electrical resistivity ratio $\rho(295\text{K})/\rho(0\text{K}) = 12$ .
4	149	L	1937	2.3-116		JM 8308; Rh 2	99.997 Rh, 0.002 Fe, and 0.0005 Cu; 1.5 mm dia; supplied by Johnson Matthey; annealed at 1300 C; ideal electrical resistivity reported as 0.00075, 0.0022, 0.0124, 0.048, 0.107, 0.42, 0.87, 1.91, 2.93, 4.33, 4.78, and 5.8 $\mu\text{ohm cm}$ at 15, 20, 30, 40, 50, 75, 100, 150, 200, 273, 295, and 350 K, respectively; residual electrical resistivity 0.0084 $\mu\text{ohm cm}$ ; electrical resistivity ratio $\rho(295\text{K})/\rho(0\text{K}) = 570$ .
5	57	L	1927	21, 83			Pure; supplied by Heraeus; rolled into square wire; annealed in vacuum at 1030 C for 10 min; electrical resistivity reported as 0.01635, 0.595, and 4.58 $\mu\text{ohm cm}$ at 21.2, 83.2, and 273.2 K, respectively.
6	253	L	1959	2.1-111		1	Specimen 7.5 x 0.15 x 0.15 cm; made from the specimen used by Gruneisen and Goene, 1927; electrical resistivity reported as 0.0153, 0.0160, 0.0167, 0.0188, 0.0318, 0.0541, 0.170, 0.350, 0.524, 0.751, 1.178, and 5.043 $\mu\text{ohm cm}$ at 4.2, 14.3, 18.7, 23.2, 32.9, 40.8, 54.7, 70.1, 78.8, 90.2, 110.6, and 292.3 K, respectively.
7	253	L	1959	4.1-90		2	Same source as the above specimen; dimensions 3.4 x 0.15 x 0.15 cm; annealed in vacuum at 1409 C for 5 hrs and cooled slowly; residual electrical resistivity 0.0148 $\mu\text{ohm cm}$ .
8	411	L	1955	86-282			0.01-0.1 Ir, 0.005 Fe, 0.002-0.005 Ag, and 0.001-0.003 Pd; 5 cm long, 0.349 cm dia; supplied by Johnson Matthey Co.; annealed in vacuum at 1326 C; density 12.45 $\text{g cm}^{-3}$ ; electrical resistivity reported as 0.63, 0.765, 0.86, 1.02, 1.16, 1.265, 1.34, 2.95, 2.99, 3.155, 4.44, and 4.51 $\mu\text{ohm cm}$ at 85.8, 92.4, 97.1, 104.9, 112.2, 117.6, 121.6, 202.6, 204.8, 213.8, 277.8, and 281.6 K, respectively.
9	665	C	1962	321-458			The above specimen measured by comparative method using Armco iron as comparative material; residual electrical resistivity 0.024 $\mu\text{ohm cm}$ ; electrical resistivity ratio $\rho(273\text{K})/\rho_0 = 182$ .
10	249	L	1967	84-282		Rh 1	0.01-0.1 Ir, 0.005 Fe, 0.002-0.005 Ag, and 0.001-0.003 Pd; specimen 0.348 cm in dia and 5 cm long supplied by Johnson Matthey Co.; annealed at 1610 K; density 12.44 $\text{g cm}^{-3}$ ; electrical resistivity reported as 0.92, 2.9, 4.90, 6.95, and 9.15 $\mu\text{ohm cm}$ at 100, 200, 300, 400, and 500 K, respectively; electrical resistivity ratio $\rho(273\text{K})/\rho(4.2\text{K}) = 180$ .
11	249	C	1967	318-591		Rh 1	The above specimen measured by comparative method using Armco iron as comparative material.
12	249	C	1967	310-521		Rh 2	0.001 Fe, 0.0002 Ag, 0.0001 Cu, and 0.0001 Pd; specimen 0.6 cm in dia, 6 cm long; supplied by Johnson Matthey and Co.; density 12.22 $\text{g cm}^{-3}$ ; electrical resistivity reported as 0.3, 2.95, 4.95, 7.05, and 9.22 $\mu\text{ohm cm}$ at 100, 200, 300, 400, and 500 K, respectively; electrical resistivity ratio $\rho(273\text{K})/\rho(4.2\text{K}) = 233$ .

## DATA TABLE NO. 45 THERMAL CONDUCTIVITY OF RHODIUM

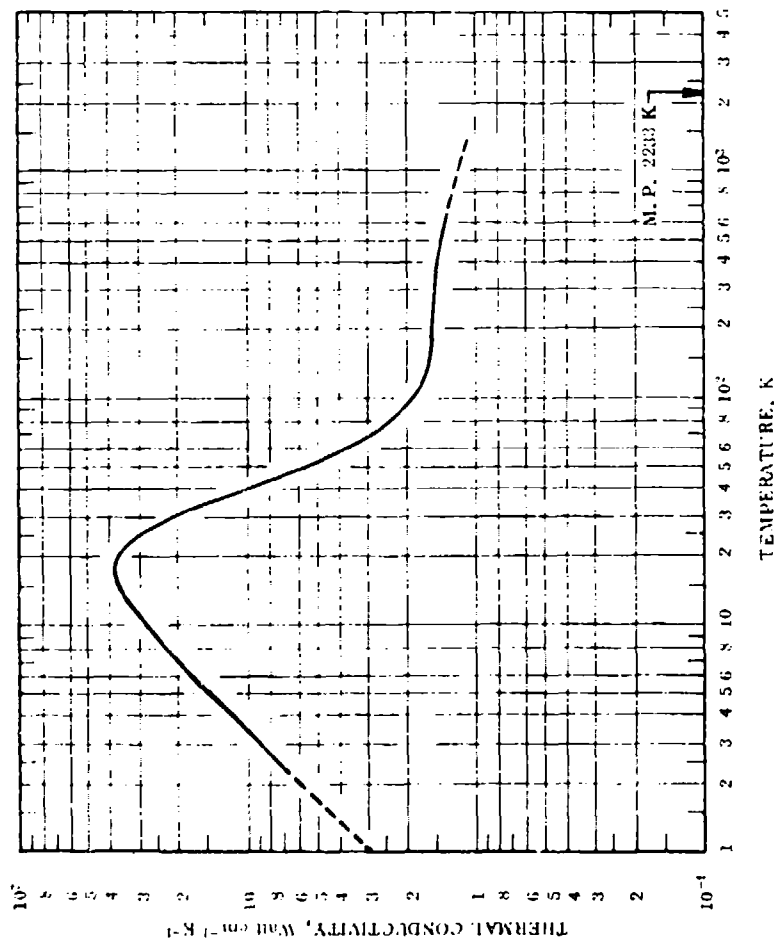
(Impurity - 0.20% each; total impurities &lt; 0.50%)

[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup> K<sup>-1</sup>]

T	k	T	k	T	k	T	k	T	k	T	k		
<u>CURVE 1</u>													
2.15	1.720	8.67	23.651	90.8	1.82	81.3	2.11	349.1	1.403 <sup>a</sup>				
2.16	2.581	9.44	25.893	111.2	1.66	85.4	2.06	395.8	1.395				
4.82	3.140	11.48	31.696	<u>CURVE 7</u>									
6.88	4.860	13.27	33.482	4.1								7.43	
8.05	5.740	15.30	37.054	7.7	13.21	97.7	1.855	432.4	1.378				
9.46	6.965	17.09	39.244	11.2	18.09	105.3	1.755	466.1	1.357				
11.35	7.741	19.64	37.649	15.9	23.37	112.9	1.70	514.1	1.32				
14.37	9.032	21.94	35.714	21.2	24.79	118.5	1.69	520.9	1.37				
17.12	10.150	24.23	32.292	27.6	20.61	122.3	1.66						
21.07	11.230	27.30	26.468	34.9	13.03	202.6	1.54						
<u>CURVE 2</u>													
33.40	16.220	30.10	21.726	37.8	5.20	204.9	1.545						
38.00	11.584	35.00	11.584	63.6	2.41	213.8	1.515						
55.10	4.431	55.10	4.431	74.8	2.41	277.7	1.53						
372.2	0.879	65.30	3.267	90.3	2.06	277.9	1.55						
<u>CURVE 3</u>													
90.80	2.228	78.30	2.550	<u>CURVE 8</u>									
6.12	0.371	90.80	2.228	85.8	2.31	281.5	1.52						
5.18	0.545	115.60	1.683	<u>CURVE 11</u>									
11.49	0.669	92.4										2.10	
14.03	0.917	97.1										1.98	
18.37	1.114	104.9										1.82	
22.96	1.312	112.2										1.71	
26.53	1.485	117.6										1.67	
29.34	1.535	121.6										1.63	
30.87	1.658	202.6										1.51	
36.22	1.683	204.8										1.53	
55.36	1.559	213.8										1.52	
61.22	1.510	277.8										1.51	
65.05	1.485	281.6										1.51	
71.43	1.460	317.5										1.48	
78.80	1.460	322.8										1.51	
30.80	1.485	324.6										1.52	
<u>CURVE 4</u>													
2.27	5.272	330.1										1.49	
2.68	6.559	331.1										1.445 <sup>a</sup>	
3.32	8.242	363.6										1.49	
4.08	10.470	370.3										1.46 <sup>a</sup>	
4.98	13.000	414.6										1.43	
<u>CURVE 5</u>													
21.20	23.800	452.5										1.405	
83.20	2.150	475.1										1.39	
<u>CURVE 6</u>													
2.1	3.27	591.2										1.385	
2.6	3.77	<u>CURVE 12</u>											
3.5	5.00	310.0										1.445 <sup>a</sup>	
4.1	5.96	316.1										1.47	
4.4	6.73	330.0										1.453 <sup>a</sup>	
7.9	11.28	348.6										1.45	
14.8	20.99	355.9										1.41	
19.0	23.38	360.7										1.453	
23.7	22.60	370.8										1.41	
33.2	13.92												
40.9	8.43												
53.5	3.85												
62.2	3.15												
71.0	2.47												
79.5	2.12												

<sup>a</sup> Not shown on plot

FIGURE AND TABLE NO. 45R RECOMMENDED THERMAL CONDUCTIVITY OF RHODIUM



RECOMMENDED VALUES\*

T <sub>1</sub>	k <sub>1</sub>	k <sub>2</sub>	T <sub>2</sub>	T <sub>1</sub>	k <sub>1</sub>	k <sub>2</sub>	T <sub>2</sub>
0	0	0	-459.7	500	1.40	80.9	440.3
1	(2.91) <sup>†</sup>	(168)	-457.9	600	(1.36)	(78.6)	620.3
2	(5.81)	(316)	-456.1	700	(1.31)	(75.7)	800.3
3	8.72	504	-454.3	800	(1.27)	(73.4)	980.3
4	11.6	670	-452.5	900	(1.24)	(71.6)	1160
5	14.5	838	-450.7	1000	(1.21)	(69.9)	1340
6	17.3	1000	-448.9	1100	(1.18)	(68.2)	1520
7	20.1	1160	-447.1	1200	(1.15)	(66.4)	1700
8	22.8	1320	-445.3	1300	(1.13)	(65.3)	1880
9	25.4	1470	-443.5	1400	(1.11)	(64.1)	2060
10	27.8	1610	-441.7				
11	30.0	1750	-439.9				
12	32.0	1890	-438.1				
13	33.7	1970	-436.3				
14	35.1	2030	-434.5				
15	36.1	2050	-432.7				
16	36.9	2130	-430.9				
18	37.2	2150	-427.3				
20	36.4	2100	-423.7				
25	30.7	1770	-414.7				
30	21.6	1250	-405.7				
35	14.5	838	-396.7				
40	10.2	589	-387.7				
45	7.47	432	-378.7				
50	5.70	329	-369.7				
60	3.75	218	-351.7				
70	2.89	167	-333.7				
80	2.38	138	-315.7				
90	2.06	119	-297.7				
100	1.86	107	-279.7				
150	1.58	91.3	-189.7				
200	1.54	89.0	-99.7				
250	1.52	87.8	-9.7				
273.2	1.51	87.2	32.0				
300	1.50	86.7	80.3				
350	1.48	85.5	170.3				
400	1.46	84.4	260.3				

REMARKS

The recommended values are for well-annealed 99.997% pure rhodium with residual electrical resistivity  $D_0 = 0.00840 \mu\Omega \text{ cm}$  (characterized by  $D_0$  becomes important at temperatures below about 200 K). The values below 1.5 T<sub>1</sub> are calculated to fit the experimental data by using  $n = 2.70$ ,  $\sigma = 3.16 \times 10^{-5}$ , and  $\beta = 0.344$ . The recommended values that are supported by experimental thermal conductivity data are thought to be accurate to within 4% of the true values near room temperature and 4 to 10% at other temperatures.

\* T<sub>1</sub> in K, k<sub>1</sub> in Watt cm<sup>-1</sup> K<sup>-1</sup>, T<sub>2</sub> in F, and k<sub>2</sub> in Btu hr<sup>-1</sup> ft<sup>-1</sup> F<sup>-1</sup>.

<sup>†</sup> Values in parentheses are extrapolated or estimated.



# THERMAL CONDUCTIVITY OF RUBIDIUM

FIGURE SHOWS ONLY 3 OF THE CURVES REPORTED IN TABLE

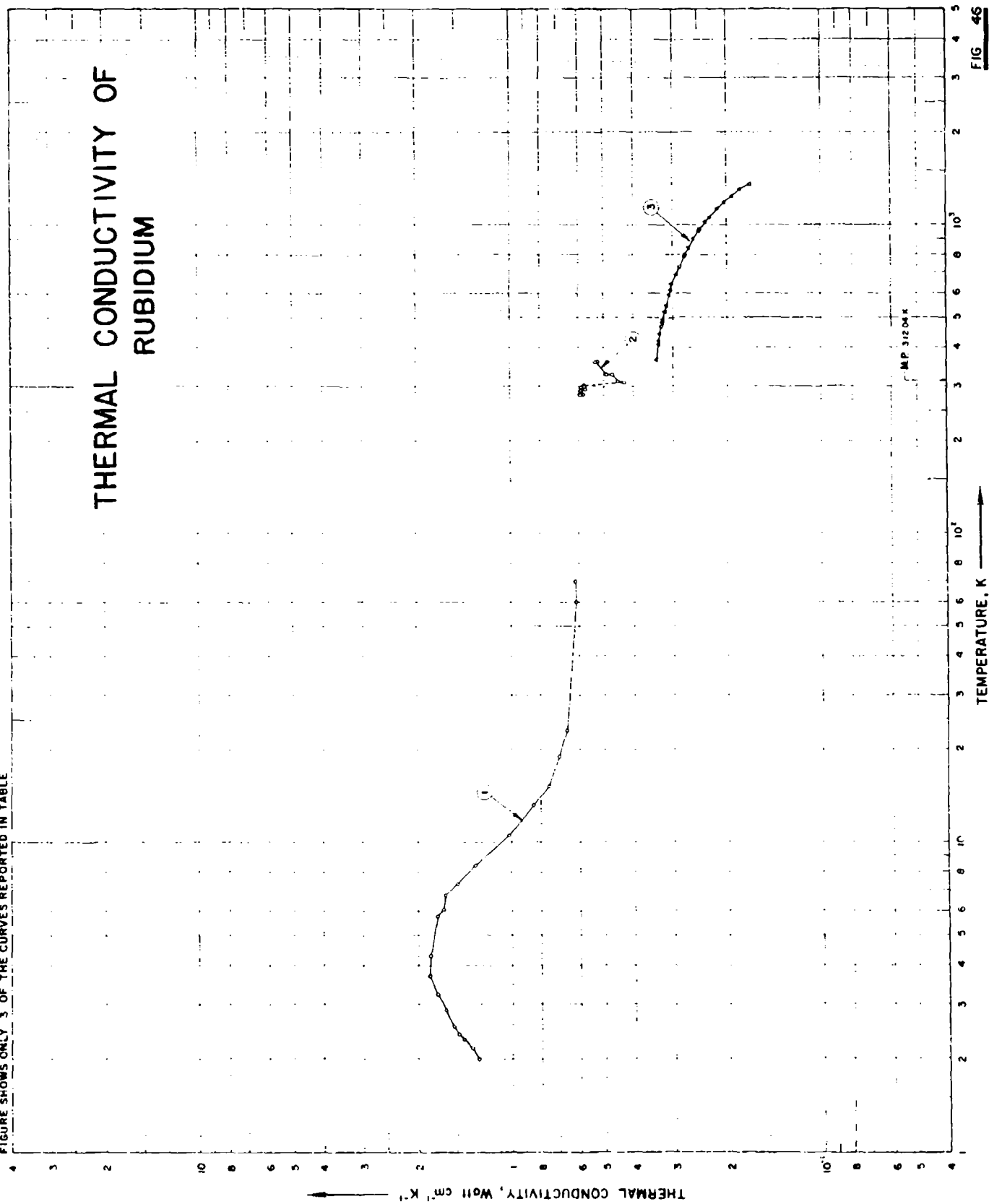


FIG 46

## SPECIFICATION TABLE NO. 46 THERMAL CONDUCTIVITY OF RUBIDIUM

(Impurity  $\leq 0.20\%$  each; total impurities  $\leq 0.50\%$ )

[For Data Reported in Figure and Table No. 46]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
1	92	L	1956	2.0-70		Rb 1	High purity: 1.65 mm dia; supplied by A. D. Mackay (New York); electrical resistivity ratio $\rho(295 \text{ K})/\rho(4 \text{ K}) = 380$ (using electrical resistivity $\rho(295 \text{ K}) = 14.6 \mu\text{ohm cm}$ ). Supplied by Ukraine Chemical Institute of Odessa; specific resistance 35% greater than that of pure Rb; M. P. S. C. lower than that of pure Rb.
2	385	L	1940	281-361			99.5 pure; electrical resistivity measured in argon and reported as 11.28, 12.51, 26.55, 30.12, 30.88, 32.82, 35.39, 35.90, 40.56, 42.70, 47.03, 47.48, 54.49, 58.01, 60.70, 69.82, 77.47, 85.30, 86.05, 96.39, 105.82, 117.59, 129.96, 144.17, 144.27, 159.55, 174.96, 196.97, and 197.36 $\mu\text{ohm cm}$ at 273.2, 302.6, 366.5, 414.3, 418.7, 442.1, 469.3, 482.6, 524.3, 547.1, 592.1, 595.4, 643.7, 691.5, 734.8, 790.9, 848.7, 901.5, 903.7, 969.3, 1020.9, 1076.5, 1133.4, 1190.9, 1195.4, 1249.8, 1303.7, 1367.6, and 1369.8 K, respectively; thermal conductivity values calculated from electrical resistivity data using theoretical Lorenz number $2.45 \times 10^{-8} \text{ V}^2 \text{ K}^{-2}$ .
4	882	-	1967	822-969			Specimen in vapor state; measuring method based on the study of laminar flow in a long tube with a constant wall temperature; combined with the heat exchange theory; data calculated from the measured temperature and flow rate and the specific heat of the vapor which was obtained from Achenner, P. Y., et al., (USAFEC Rept. AGN-8192, 1967).
5	881	-	1962	944-1121			Vapor Rubidium specimen; atomic dia $\sigma = 12.7 \text{ \AA}$ ; thermal conductivity data estimated.
6	756	-	1962	312-1025			0.32 Cs, 0.06 K, and 0.62 Na; composition after testing, 0.39 Cs, 0.13 Na, 0.11 K, 0.03 Ca, 0.008 Fe, 0.005 O, 0.002 Ni, $\pm 0.001$ each of Cr and Li; molten specimen contained in a type 347 stainless-steel tube; supplied by American Potash and Chemical Corp.; electrical resistivity reported as 13.85, 14.67, 22.84, 22.93, 23.35, 25.96, 31.62, 37.06, 42.30, 46.59, 46.61, 52.45, 58.01, 59.37, 64.61, 71.48, 72.49, 81.06, 91.29, 99.05, and 109.31 $\mu\text{ohm cm}$ at 25.6, 37.5, 55.2, 41.7, 46.4, 91.7, 146.7, 204.2, 260.3, 308.3, 309.4, 361.7, 412.5, 426.1, 463.1, 520.3, 528.9, 581.7, 650.1, 697.2, and 751.7 C, respectively; thermal conductivity values calculated from measured electrical resistivity data and the theoretical Lorenz number $2.45 \times 10^{-8} \text{ V}^2 \text{ K}^{-2}$ .

DATA TABLE NO. 46 THERMAL CONDUCTIVITY OF RUBIDIUM

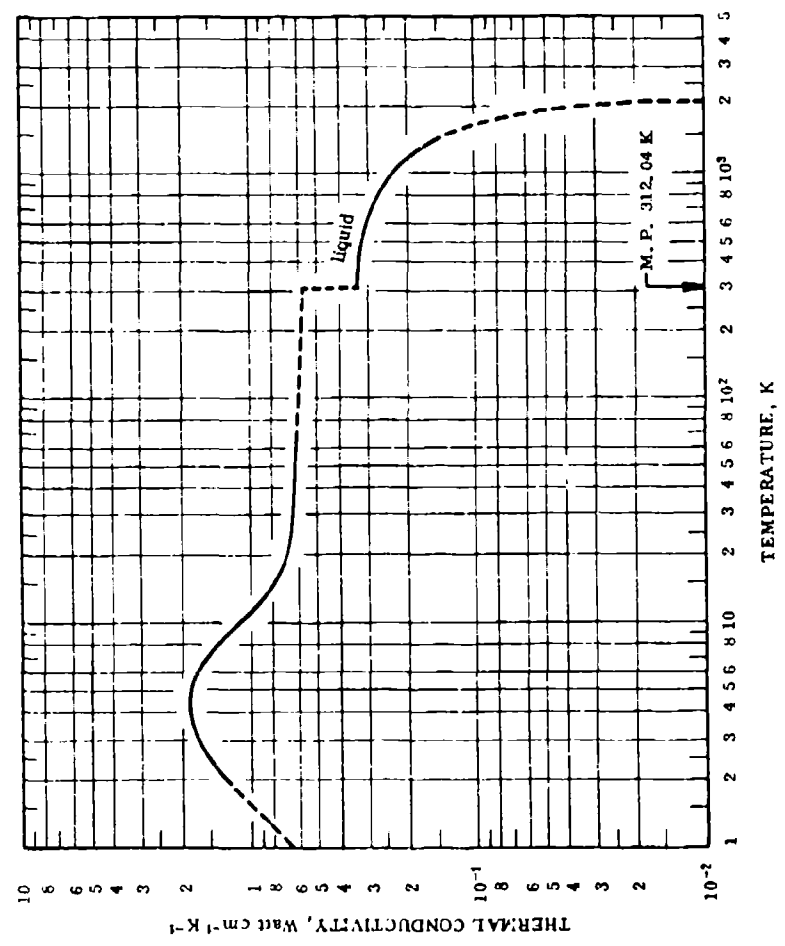
(Impurity &lt; 0.20% each; total impurities &lt; 0.50%)

[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup>K<sup>-1</sup>]

CURVE 1		CURVE 3		CURVE 4*	
T	k	T	k	T	k
2.00	1.272	366.5	0.3387	822.1	0.0000486
2.19	1.335	414.3	0.3337	874.3	0.0000537
2.32	1.422	418.7	0.3321*	969.3	0.0000625
2.41	1.480	426.5	0.3311		
2.55	1.543	442.1	0.3291	CURVE 5*	
2.88	1.646	469.3	0.3254	544	0.0000116
3.25	1.744	482.6	0.3236	1000	0.0000113
3.71	1.848	492.1	0.3223	1060	0.0000114
4.32	1.866	524.3	0.3179	1121	0.0006111
5.80	1.744	528.7	0.3172*		
6.12	1.670	547.1	0.3146	CURVE 6*	
6.77	1.640	592.1	0.3081	312.4	0.335
7.39	1.495	595.4	0.3076*	419.9	0.325
8.45	1.309	618.2	0.3042	533.5	0.309
10.60	1.024	643.7	0.3003	634.9	0.297
13.30	0.846	691.5	0.2929	699.3	0.289
15.31	0.753	733.7	0.2861	854.9	0.258
19.02	0.700	734.8	0.2859*	923.3	0.248
23.20	0.661	790.9	0.2767	1025	0.230
60.0	0.618	803.2	0.2747		
70.0	0.618	840.9	0.2683		
		848.7	0.2670*		
		901.5	0.2578		
		903.7	0.2574*		
281.2	0.59	957.1	0.2480		
282.2	0.58	969.3	0.2458		
290.7	0.59	1020.9	0.2363		
293.2	0.57	1068.7	0.2275		
296.7	0.59	1076.5	0.2260*		
298.7	0.575	1135.4	0.2149		
300.7	0.575	1190.4	0.2043		
302.2	0.575	1190.9	0.2042*		
309.2	0.42	1195.4	0.2033*		
310.5	0.45	1249.8	0.1927		
327.8	0.47	1303.7	0.1820		
328.2	0.49	1307.1	0.1814*		
360.2	0.51	1367.6	0.1692		
361.2	0.52	1368.8	0.1688*		

\* Not shown on plot

FIGURE AND TABLE NO. 46R RECOMMENDED THERMAL CONDUCTIVITY OF RUBIDIUM



RECOMMENDED VALUES\*

T <sub>1</sub>	k <sub>1</sub>	k <sub>2</sub>	T <sub>2</sub>	T <sub>1</sub>	k <sub>1</sub>	k <sub>2</sub>	T <sub>2</sub>
0	0	0	-459.7	312.04	0.339	19.6	102.0
1	(0.663) <sup>‡</sup> (38.3)	0	-457.9	350	0.336	19.4	170.3
2	1.27	78.4	-456.1	400	0.331	19.1	260.3
3	1.71	98.8	-451.3	500	0.318	18.4	440.3
4	1.91	106	-452.5	600	0.304	17.6	620.3
5	1.88	109	-448.9	700	0.289	16.7	900.3
6	1.73	100	-447.1	800	0.273	15.8	980.3
7	1.56	90.1	-443.5	900	0.255	14.7	1160
8	1.38	79.7	-441.7	1000	0.237	13.7	1340
9	1.22	70.5	-439.9	1100	0.219	12.7	1520
10	1.09	63.0	-438.1	1200	0.200	11.6	1700
11	0.951	57.3	-436.3	1300	0.180	10.4	1880
12	0.919	53.1	-434.5	1400	(0.159)	( 9.19)	2060
13	0.859	49.6	-432.7	1500	(0.139)	(8.03)	2240
14	0.810	46.8	-430.9	1600	(0.118)	(6.82)	2420
15	0.772	44.6	-427.7	1700	(0.096)	(5.55)	2600
16	0.746	43.1	-423.7	1800	(0.074)	(4.28)	2780
18	0.710	41.0	-414.7	1900	(0.052)	(3.00)	2960
20	0.685	39.6	-405.7	2000	(0.027)	(1.56)	3140
25	0.657	38.0	-396.7	2100	( ~ 0 )	( ~ 0 )	3320
30	0.647	37.4	-387.7				
35	0.640	37.0	-378.7				
40	0.635	36.7	-369.7				
45	0.630	36.4	-351.7				
50	0.627	36.2	-333.7				
60	0.620	35.8	-315.7				
70	0.615	35.5	-297.7				
80	(0.611)	(35.3)	-279.7				
90	(0.607)	(35.1)	-189.7				
100	(0.603)	(34.8)	-99.7				
150	(0.594)	(34.3)	-9.7				
200	(0.589)	(34.0)	32.0				
250	(0.586)	(33.9)	80.3				
273.2	(0.583)	(33.7)	102.0				
300	0.582	33.6					
312.04	0.581	33.6					

REMARKS

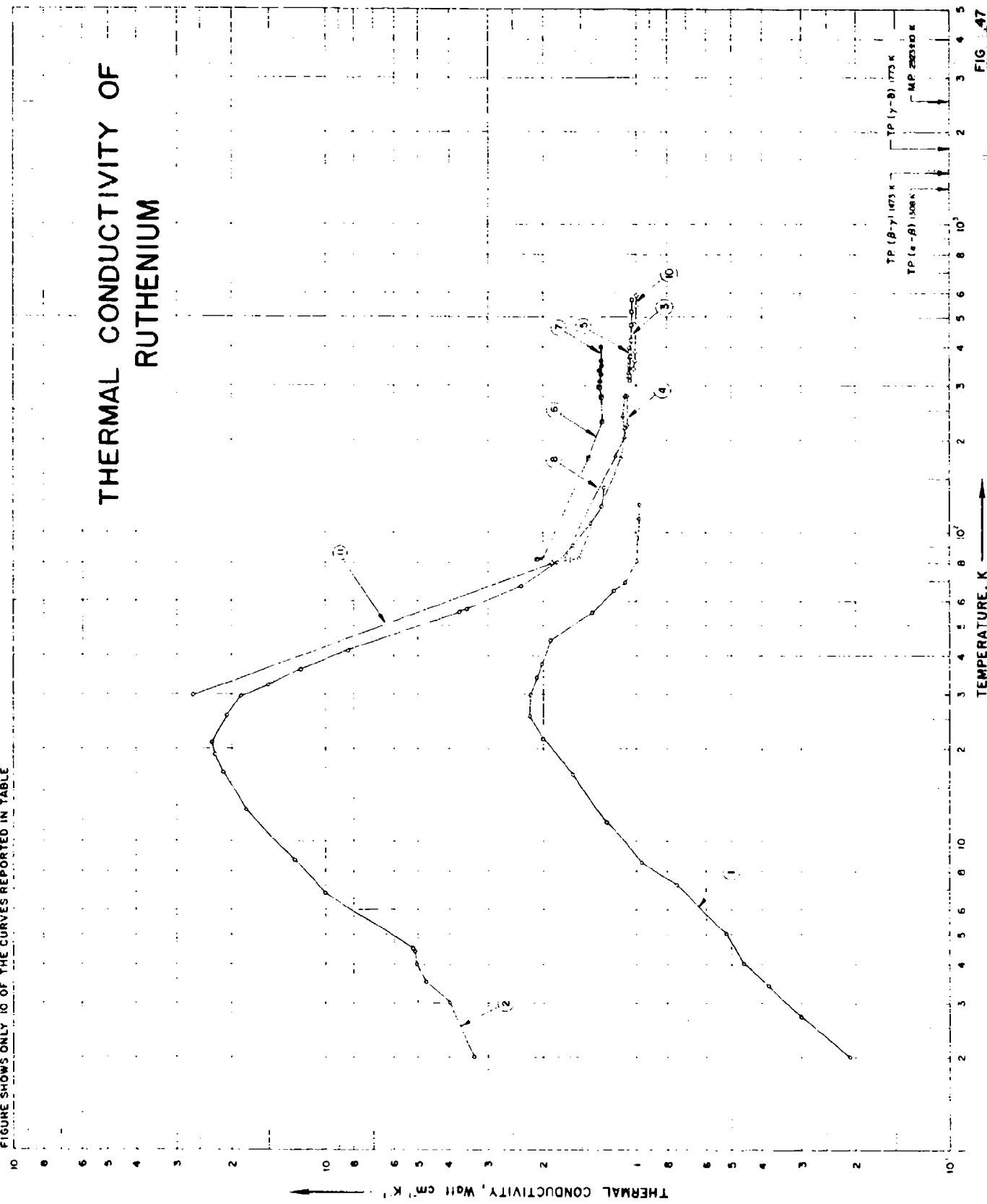
The recommended values are for high-purity rubidium with residual electrical resistivity  $\rho_0 = 0.0968 \mu\Omega \text{ cm}$  (characterization by  $\rho_0$  becomes important at temperatures below about 70 K). The values below 1.5 Tm are calculated to fit the experimental data by using  $n = 2.00$ ,  $\alpha' = 0.0093$ , and  $\beta = 1.56$ . The recommended values that are supported by experimental thermal conductivity data are thought to be accurate to within 5% of the true values near room temperature and 5 to 10% at other temperatures.

\* T<sub>1</sub> in K, k<sub>1</sub> in W m<sup>-1</sup> cm<sup>-1</sup> K<sup>-1</sup>, T<sub>2</sub> in F, and k<sub>2</sub> in Btu hr<sup>-1</sup> ft<sup>-1</sup> F<sup>-1</sup>.

‡ Values in parentheses are extrapolated, interpolated, or estimated.

# THERMAL CONDUCTIVITY OF RUTHENIUM

FIGURE SHOWS ONLY 10 OF THE CURVES REPORTED IN TABLE



TEMPERATURE, K →

## SPECIFICATION TABLE NO. 47 THERMAL CONDUCTIVITY OF RUTHENIUM

(Impurity  $\leq$  0.20% each; total impurities  $\leq$  0.50%)

[ For Data Reported in Figure and Table No. 47 ]

Curve No.	R.L. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition ( weight percent), Specifications and Remarks
1	384	L	1958	2.0-124	11	Ru 2	99.995% pure; polycrystalline; grain size 1 $\mu$ m; approx 6 mm in dia and 7 cm long; Ru powder supplied by Baker Platinum Co.; specimen prepared by air-melting pressed pellets of powder in an inert atmosphere; specific gravity 12.9 at 22 C; ideal electrical resistivity reported as 0.034, 0.067, 0.19, 0.38, 1.07, 1.90, 3.61, 5.26, 6.85, 7.50, and 8.27 $\mu$ ohm cm at 25, 30, 40, 50, 75, 100, 150, 200, 250, 273, and 295 K, respectively; residual electrical resistivity 0.235 $\mu$ ohm cm; $\rho(273K)/\rho_0 = 36.1$ ; Lorenz function $2.40 \times 10^{-8} \text{ V}^2 \text{ K}^{-2}$ at 0 K.
2	384	L	1958	2.0-140	9	Ru 3	Similar to the above specimen except dimensions approx 5 mm in dia and 6 cm long; specific gravity 12.25 at 22 C; ideal electrical resistivity reported as 0.005, 0.010, 0.037, 0.11, 0.54, 1.25, 2.80, 4.38, 5.76, 6.69, and 7.37 $\mu$ ohm cm at 25, 30, 40, 50, 75, 100, 150, 200, 250, 273, and 295 K, respectively; residual electrical resistivity 0.0158 $\mu$ ohm cm; $\rho(295K)/\rho_0 = 467$ ; Lorenz function $2.46 \times 10^{-8} \text{ V}^2 \text{ K}^{-2}$ at 0 K.
3	665	C	1962	313-573	2		0.1 Fe, 0.03 Rh, 0.002 Pt, 0.001 Cu, 0.001 Ni, and 0.0005 Pd; 2.5 cm long; 0.660 cm in dia; supplied by Johnson Matthey Co.; argon-arc melted and ground; density 12.36 $\text{g cm}^{-3}$ ; electrical resistivity reported as 0.566 and 7.13 $\mu$ ohm cm at liquid helium and ice temp, respectively; Armco iron used as comparative material.
4	249	L	1967	83-280		a	99.96 Ru (by difference), 0.03 Os, 0.006 Fe, 0.003 Ni, and 0.001 Pd; single crystal; specimen 0.65 cm in dia, 10 cm long; axis of specimen perpendicular to prism axis of crystal; supplied by the International Nickel Co. Ltd. (Mond); as received; density 12.38 $\text{g cm}^{-3}$ ; electrical resistivity reported as 1.42, 4.58, 7.62, 10.5, and 13.3 $\mu$ ohm cm at 100, 200, 300, 400, and 500 K, respectively; electrical resistivity ratio $\rho(273K)/\rho(4.2K) = 94$ .
5	249	C	1967	322-476		a	The above specimen measured by comparative method using Armco iron as comparative material.
6	249	L	1967	83-298		b	Same purity and supplier as the above specimen; 0.68 cm in dia, 10 cm long; single crystal; axis of the specimen parallel to prism axis of crystal; as received; density 12.38 $\text{g cm}^{-3}$ ; electrical resistivity reported as 1.07, 3.46, 5.82, 8.15, and 10.4 $\mu$ ohm cm at 100, 200, 300, 400, and 500 K, respectively; electrical resistivity ratio $\rho(273K)/\rho(4.2K) = 76.5$ .
7	249	C	1967	310-404		b	The above specimen measured by comparative method using Armco iron as comparative material.
8	249	L	1967	83-277		c	0.03 Os, 0.006 Fe, 0.003 Ni, and 0.001 Pd; polycrystalline bar 0.635 cm in dia, 10 cm long; supplied by the International Nickel Co.; pressed at 20 ton in. $^{-2}$ , sintered in vacuo at 1920 K and hot forged; as received; density 12.24 $\text{g cm}^{-3}$ ; electrical resistivity reported as 1.30, 4.38, 7.43, 10.4, and 13.2 $\mu$ ohm cm at 100, 200, 300, 400, and 500 K, respectively; electrical resistivity ratio $\rho(273K)/\rho(4.2K) = 388$ .
9	249	C	1967	365-510		c	The above specimen measured by comparative method using Armco iron as comparative material.

SPECIFICATION TABLE NO. 47 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
10	249	C	1967	341-592		d	0.03 Rh, 0.03 Fe, 0.002 Pt, 0.001 Cu, 0.001 Ni, and 0.0005 Bi; polycrystalline; 0.66 cm in dia., 2.5 cm long; supplied by Johnson Matthey and Co.; arc-melted and ground; as received; density $12.36 \text{ g cm}^{-3}$ ; electrical resistivity reported as 1.83, 4.83, 7.85, 10.74, and $13.4 \mu\text{ohm cm}$ at 100, 200, 300, 400, and 500 K, respectively; Armco iron used as comparative material.

DATA TABLE NO. 17 THERMAL CONDUCTIVITY OF RUBIDIUM  
(Impurity <math>\le 0.20\%</math> each; total impurities <math>\le 0.50\%</math>)

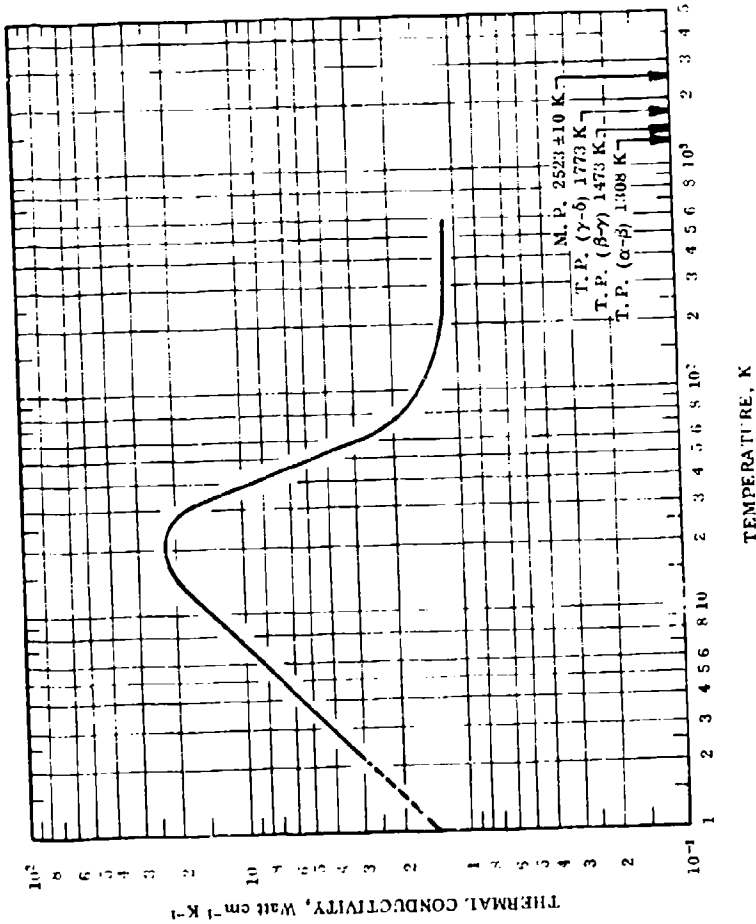
(Temperature, T, K; Thermal Conductivity,  $k$ , Watt  $\text{cm}^{-1}\text{K}^{-1}$ .)

T	k	T	k	T	k	T	k	T	k	
<u>CURVE 1</u>										
2.0	0.21	55.2	3.7	82.6	2.09	170.1	1.01	<u>CURVE 9 (cont.)</u>		
2.7	0.30	56.5	3.5	82.1	2.04	510.1	1.03			
3.4	0.38	67.0	2.35	83.1	2.01	<u>CURVE 10</u>				
4.0	0.46	80.0	1.80	173.3	1.42	341.1	1.03			
5.0	0.52	90.8	1.60	177.1	1.41	346.1	1.025			
7.2	0.75	107.6	1.40	230.6	1.29	346.1	1.025			
8.5	0.96	122.0	1.39	233.1	1.28	362.1	1.025			
11.5	1.26	140.0	1.28	276.6	1.39	388.6	1.02			
16.5	1.6	<u>CURVE 3</u>			279.1	1.31	390.1	1.02		
21.5	2.0	313	1.060	297.6	1.32	404.1	1.015			
25.5	2.21	323	1.060	<u>CURVE 7</u>			544.1	1.015		
29.7	2.2	373	1.050	310.1	1.32	591.6	1.01			
34.0	2.1	423	1.045	327.1	1.31					
37.5	2.02	473	1.045	338.1	1.325					
44.7	1.9	523	1.040	350.1	1.305					
55.0	1.38	573	1.035	356.1	1.31					
64.8	1.2	<u>CURVE 4</u>			362.1	1.305				
69.0	1.1	82.9	1.62	366.1	1.30					
81.0	1.0	82.1	1.58	404.1	1.305					
111	0.98	87.1	1.53	<u>CURVE 8</u>						
124	0.98	173.3	1.14	82.6	1.70					
<u>CURVE 2</u>										
2.0	3.33	177.1	1.14	82.1	1.67					
3.0	4.0	223.9	1.08	83.1	1.66					
3.5	4.75	225.1	1.06	177.8	1.18					
4.0	5.08	277.9	1.08	205.1	1.11					
4.4	5.15	280.1	1.09	219.6	1.11					
4.5	5.22	<u>CURVE 5</u>			221.6	1.09				
6.8	10.05	322.1	1.07	227.6	1.085					
8.7	12.5	331.3	1.065	227.6	1.11					
12.7	18.0	336.9	1.07	239.5	1.13					
16.8	21.3	349.7	1.065	277.4	1.10					
19.2	22.5	355.1	1.055	<u>CURVE 9</u>						
21.0	23.0	364.2	1.07	365.1	1.05					
25.6	20.8	373.8	1.055	385.1	1.06					
29.5	18.8	399.9	1.065	389.1	1.04					
32.1	15.08	412.1	1.05	393.1	1.05					
36.0	12.0	476.1	1.03	441.1	1.04					
41.9	8.34									

Not shown on plot



FIGURE AND TABLE NO. 47R RECOMMENDED THERMAL CONDUCTIVITY OF RUTHENIUM



REMARKS

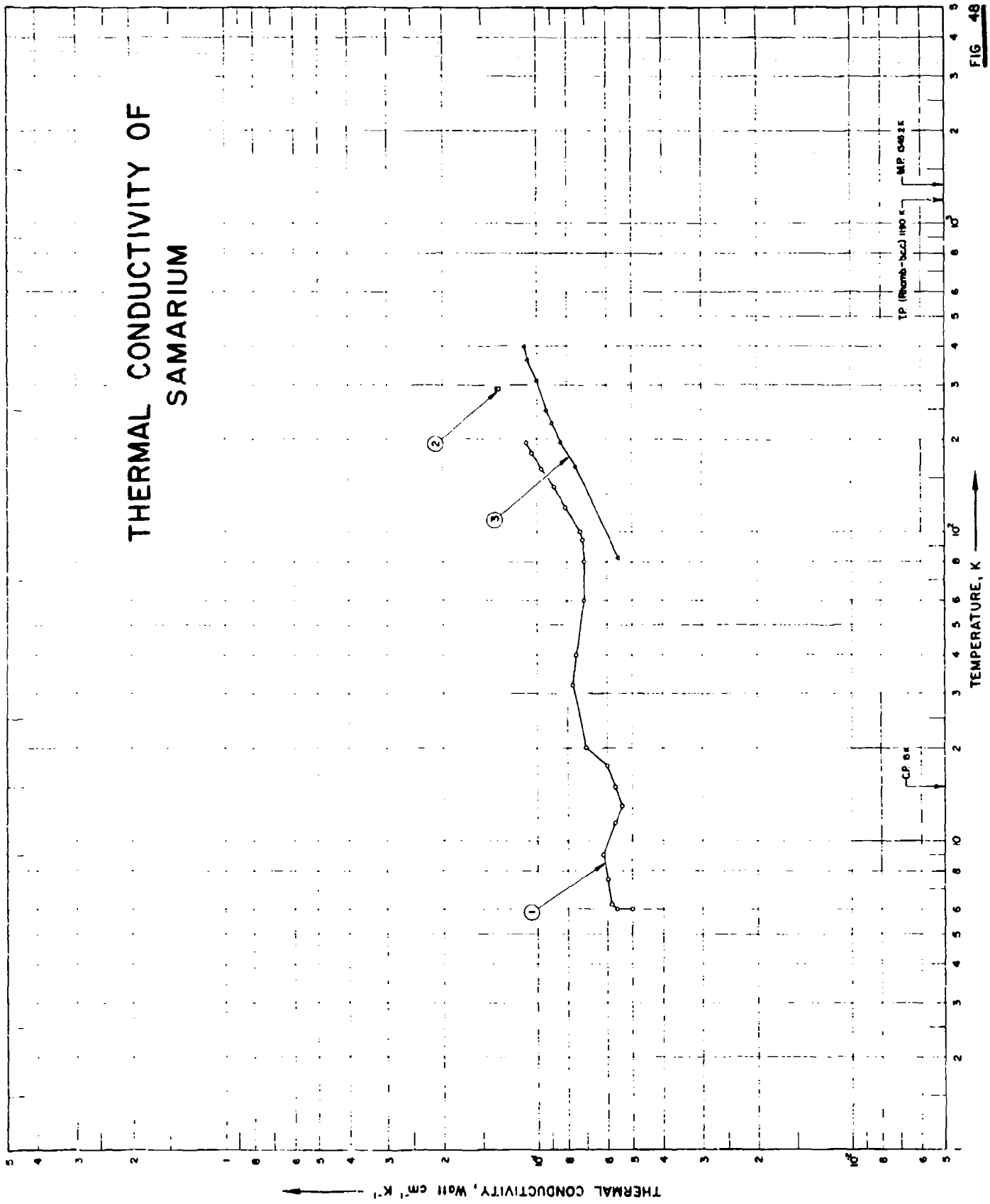
The recommended values are for well-annealed 99.995% pure ruthenium with residual electrical resistivity  $\rho_0 = 0.00860 \mu\Omega \text{ cm}$  (characterization by  $\rho_0$  becomes important at temperatures below 250 K). The values below 1.5 Tm are calculated to fit the experimental data by using  $n = 3.10$ ,  $a = 2.5$ ,  $m = 5.60$ ,  $\alpha' = 3.91 \times 10^9$ , and  $\beta = 0.705$ . The recommended values are thought to be accurate to within 4% of the true values near room temperature and 4 to 10% at other temperatures.

\*  $T_1$  in K,  $k_1$  in Watt  $\text{cm}^{-1} \text{K}^{-1}$ ,  $T_2$  in F, and  $k_2$  in Btu  $\text{hr}^{-1} \text{ft}^{-1} \text{F}^{-1}$ .

† Values in parentheses are extrapolated.

RECOMMENDED VALUES* (For Polycrystalline)			
$T_1$	$k_1$	$k_2$	$T_2$
0	0	0	-459.7
1	(1.42)†	(82.0)	-457.9
2	2.84	164	-456.1
3	4.26	246	-454.3
4	5.67	328	-452.5
5	7.09	410	-450.7
6	8.50	491	-448.9
7	9.90	572	-447.1
8	11.3	65	-445.3
9	12.7	734	-443.5
10	14.0	809	-441.7
11	15.3	884	-439.9
12	16.5	953	-438.1
13	17.7	1020	-436.3
14	18.8	1090	-434.5
15	19.8	1140	-432.7
16	20.6	1190	-430.9
18	22.0	1270	-427.3
20	22.7	1310	-423.7
25	21.8	1260	-414.7
30	17.7	1020	-405.7
35	13.1	757	-396.7
40	9.33	539	-387.7
45	6.72	386	-378.7
50	4.95	286	-369.7
60	3.04	176	-351.7
70	2.23	129	-333.7
80	1.84	106	-315.7
90	1.65	95.3	-297.7
100	1.54	89.0	-279.7
150	1.28	74.0	-189.7
200	1.18	68.2	-99.7
250	1.17	67.6	-9.7
273.2	1.17	67.6	32.0
300	1.17	67.6	80.3
350	1.16	67.0	170.3
400	1.15	66.4	260.3
500	1.13	65.3	440.3
600	1.11	64.1	620.3

# THERMAL CONDUCTIVITY OF SAMARIUM



## SPECIFICATION TABLE NO. 48 THERMAL CONDUCTIVITY OF SAMARIUM

(Impurity &lt; 0.20% each; total impurities &lt; 0.50%)

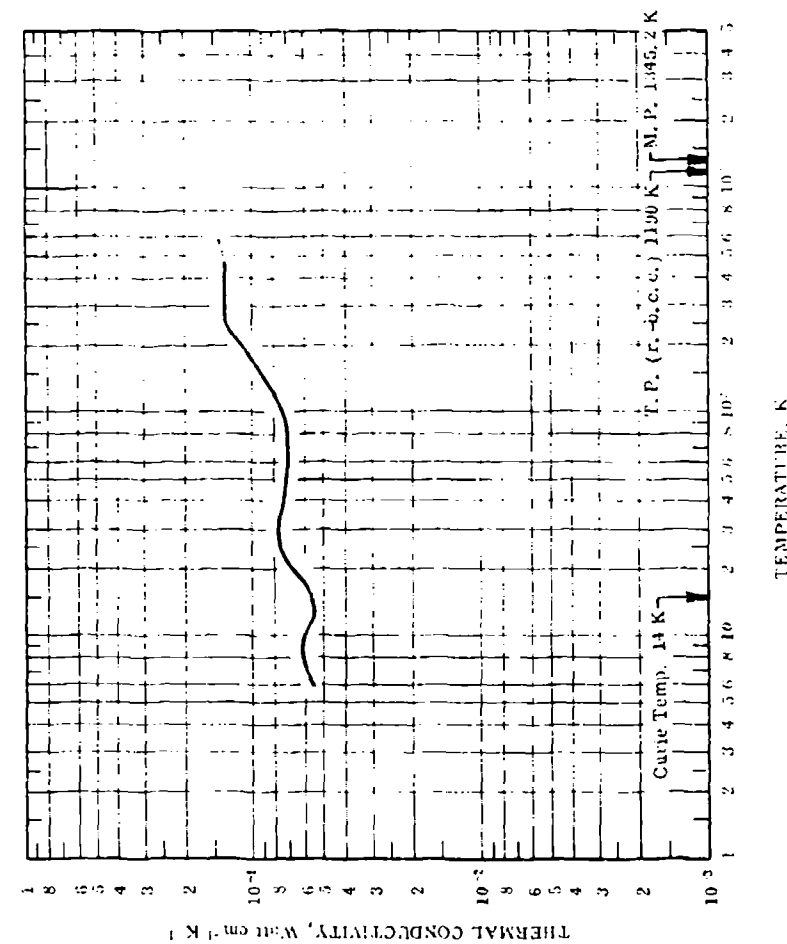
[ For Data Reported in Figure and Table No. 48 ]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition ( weight percent), Specifications and Remarks
1	812	L	1965	6.0-196			0.05 Eu, 0.02 Ca, 0.01 Cd, 0.01 Mg, and 0.005 Si; polycrystalline; 0.479 cm dia, 6 cm long, Sm supplied by Research Chemicals; arc-melted in 100 torr argon atmosphere and machined; electrical resistivity reported as 6.73, 7.90, 12.1, 15.1, 16.3, 18.2, 28.2, 39.6, 51.3, 61.8, 62.3, 63.0, 64.7, 70.5, 82.5, and 93.4 $\mu\text{ohm cm}$ at 4.17, 8.12, 13.8, 16, 20, 40, 60, 80, 100, 105, 110, 120, 160, 240, and 308 K, respectively; data taken from smoothed curve.
2	777	C	1965	291.2	3		High purity; polycrystalline; 0.25 in. in dia, 0.25 in. long; supplied by Johnson Matthey and Co.; electrical resistivity 94 $\mu\text{ohm cm}$ at 18 C; measurements made using 2 different thermal comparators. Monel metal used as comparative material.
3	810	L	1964	83-397			Impurities: 0.5 Eu, 0.18 Ca, 0.02 Cd, 0.01 Nd, and 0.01 Y; prepared by briquetting powder under a pressure of $\sim 8000 \text{ Kg cm}^{-2}$ and annealing in vacuo ( $\sim 1 \times 10^{-4} \text{ mm Hg}$ ) for 1-2 hrs at 1500-1800 C; measured in vacuo of $\sim 5 \times 10^{-4} \text{ mm Hg}$ ; data taken from smoothed curve of measurements on several specimens.

DATA TABLE NO. 48 THERMAL CONDUCTIVITY OF SAMARIUM  
 (Temperature, T, K. Thermal Conductivity,  $k$ , Watt  $\text{cm}^{-1}\text{K}^{-1}$ )

T	k
<u>CURVE 1</u>	
6.0	0.050
6.0	0.056
6.25	0.0585
7.5	0.060
9.9	0.062
11.5	0.057
13	0.054
15	0.057
17.5	0.060
20	0.070
32	0.077
40	0.075
60	0.071
80	0.071
94	0.072
100	0.073
120	0.081
140	0.088
160	0.097
180	0.104
196	0.108
<u>CURVE 2</u>	
291.2	0.113
291.2	0.114
<u>CURVE 3</u>	
82.5	0.0552
163	0.0753
195	0.0836
225	0.0895
247	0.0933
308	0.100
360	0.107
397	0.110

FIGURE AND TABLE NO. 48R RECOMMENDED THERMAL CONDUCTIVITY OF SAMARIUM



RECOMMENDED VALUES\*  
(For Polycrystalline)

T <sub>1</sub>	k <sub>1</sub>	k <sub>2</sub>	T <sub>2</sub>
0	0	0	-459.7
6	0.0555	3.21	-448.9
7	0.0591	3.41	-447.1
8	0.0611	3.53	-445.3
9	0.0618	3.57	-443.5
10	0.0607	3.51	-441.7
11	0.0586	3.39	-439.9
12	0.0557	3.22	-438.1
13	0.0542	3.13	-436.3
14	0.0551	3.18	-434.5
15	0.0566	3.27	-432.7
16	0.0580	3.35	-430.9
17	0.0598	3.46	-429.1
18	0.0623	3.60	-427.3
19	0.0655	3.81	-425.5
20	0.0692	4.00	-423.7
21	0.0703	4.06	-421.9
22	0.0727	4.20	-420.1
23	0.0738	4.26	-418.3
25	0.0754	4.36	-414.7
30	0.0770	4.45	-405.7
35	0.0768	4.44	-396.7
40	0.0754	4.36	-387.7
45	0.0742	4.29	-378.7
50	0.0732	4.23	-369.7
60	0.0714	4.13	-351.7
70	0.0708	4.09	-333.7
80	0.0709	4.10	-315.7
90	0.0714	4.13	-297.7
100	0.0735	4.25	-279.7
150	0.0924	5.34	-189.7
200	0.113	6.53	-99.7
220	0.123	7.11	-63.7
250	0.132	7.63	-9.7
273.2	0.133	7.68	32.0
300	0.133	7.68	80.3
350	0.133	7.68	170.3
400	0.133	7.68	260.3
500	(0.135) <sup>‡</sup>	(7.80)	440.3
600	(0.141)	(8.15)	620.3

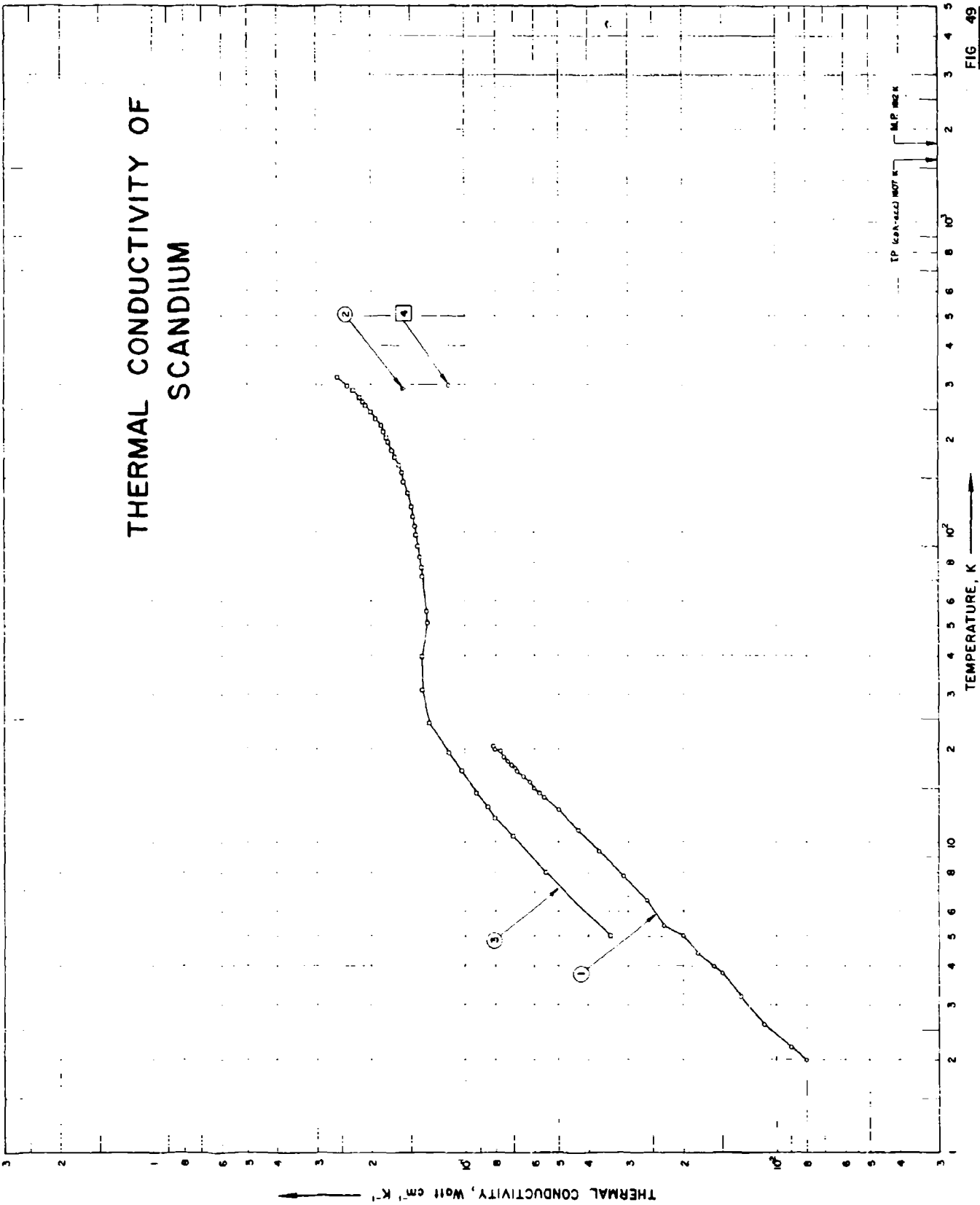
REMARKS

The recommended values are for well-annealed 99.99% pure samarium with residual electrical resistivity  $\rho_0 = 6.73 \mu\Omega \text{ cm}$  (characterization by  $\rho_0$  becomes important at temperatures below about 200 K). The recommended values that are supported by experimental thermal conductivity data are thought to be accurate to within 5% of the true values near room temperature and 5 to 15% at other temperatures.

\* T<sub>1</sub> in K, k<sub>1</sub> in Watt cm<sup>-1</sup> K<sup>-1</sup>, T<sub>2</sub> in F, and k<sub>2</sub> in Btu hr<sup>-1</sup> ft<sup>-1</sup> F<sup>-1</sup>.

<sup>‡</sup> Values in parentheses are extrapolated.

# THERMAL CONDUCTIVITY OF SCANDIUM



## SPECIFICATION TABLE NO. 49 THERMAL CONDUCTIVITY OF SCANDIUM

(Impurity: 0.20% each; total impurities: 0.50%)

[For Data Reported in Figure and Table No. 49]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)	Specifications and Remarks
1	817-597	L	1965	2.0-21				Approx. 99.9 pure; flat specimen 0.25 mm thick; electrical resistivity $71 \mu\text{ohm cm}$ at 293 K; electrical resistivity ratio $\rho(293\text{K})/\rho(4.2\text{K}) = 9.59$ ; Lorenz number $2.96 \times 10^{-8} \text{ V}^2 \text{ K}^{-2}$ ; 9, 1, 2
2	777	C	1965	293.2	±3.0			High purity (polycrystalline); specimen 0.25 in. in dia and 0.25 in. long; supplied by Johnson Matthey Co.; electrical resistivity $52 \mu\text{ohm cm}$ at 18 C; Monel metal used as comparison material; measurements made using 2 different thermal comparators.
3	818	L	1964	5-316				High purity (pieces of Ti, Cu, and Fe; polycrystalline; specimen 0.486 cm in dia and 6.33 cm long; supplied by St. Eloi Corp. electrical resistivity reported as 10.7, 12.0, 25.0, 35.0, 32.8, 36.1, 47.9, 54.8, and $61.8 \mu\text{ohm cm}$ at 6, 2, 40, 80, 120, 160, 200, 240, 280, and 320 K, respectively; residual electrical resistivity $\sim 10.6 \mu\text{ohm cm}$ ; measured in a vacuum of $6 \times 10^{-6}$ mm Hg.
4	1003	-	1961	298.2				99.3 Ti, 0.03 Cu, 0.61 Ag, 0.002 Fe, and 0.01 other rare earth metals; melting point 1522-5 C; electrical resistivity reported as 67.91, 112, 131, 146, 159, 172, 183, 193, 203, 212, and $215 \mu\text{ohm cm}$ at 0, 100, 200, 300, 400, 500, 600, 700, 800, 900, 1000, and 1360 C, respectively; thermal conductivity value calculated from the measured datum of electrical resistivity and the Lorenz function taken as $2.7 \times 10^{-4} \text{ V}^2 \text{ K}^{-2}$ .

DATA TABLE NO. 49 THERMAL CONDUCTIVITY OF SCANDIUM

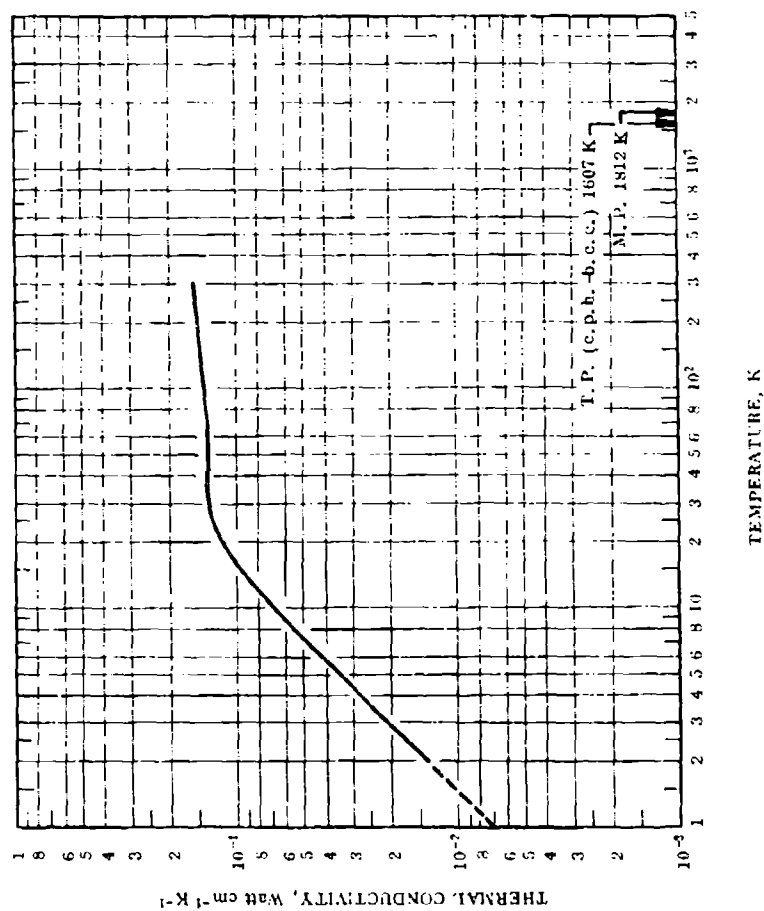
(Impurity &lt; 0.2%, each; total impurities &lt; 0.50%)

Temperature, T, K. Thermal Conductivity,  $\kappa$ , Watt  $\text{cm}^{-1}\text{K}^{-1}$ .

T	$\kappa$	T	$\kappa$
<u>CURVE 1</u>			
2.0	0.008	19.5	0.113
2.2	0.009	24.3	0.130
2.6	0.011	31.0	0.137
3.2	0.013	39.8	0.137
3.8	0.015	51.9	0.132
4.0	0.016	55.5	0.133
4.4	0.016	72.0	0.137
5.0	0.020	77.0	0.138
5.4	0.023	83.0	0.140
6.5	0.026	90.0	0.142
7.8	0.031	98.0	0.144
9.4	0.037	105.0	0.145
11.0	0.043	112	0.147
12.8	0.050	121	0.149
14.0	0.056	134	0.152
14.5	0.058	146	0.157
15.0	0.060	156	0.159
15.7	0.062	165	0.163
16.4	0.065	175	0.168
17.0	0.068	184	0.171
17.4	0.069	196	0.176
17.8	0.071	201	0.178
18.3	0.073	211	0.182
18.8	0.075	221	0.185
19.7	0.077	232	0.192
20.0	0.080	245	0.199
20.5	0.081	256	0.207
<u>CURVE 2</u>			
291.2	0.155	264	0.211
291.2	0.159	273	0.218
<u>CURVE 3</u>			
5.0	0.0343	287	0.229
8.2	0.055	298.2	0.239
10.5	0.070	316	0.257
12.0	0.080		
13.0	0.085		
14.5	0.0920		
17.0	0.103		
<u>CURVE 3 (cont.)</u>			
<u>CURVE 4</u>			
298.2 0.113			



FIGURE AND TABLE NO. 49R RECOMMENDED THERMAL CONDUCTIVITY OF SCANDIUM



## REMARKS

The recommended values are for well-annealed high-purity scandium with residual electrical resistivity  $\rho_0 = 10.6 \mu\Omega$  cm (characterization by  $\rho_0$  becomes important at temperatures below about 250 K). The recommended values that are supported by experimental thermal conductivity data are thought to be accurate to within 5% of the true values near room temperature and 5 to 15% at other temperatures.

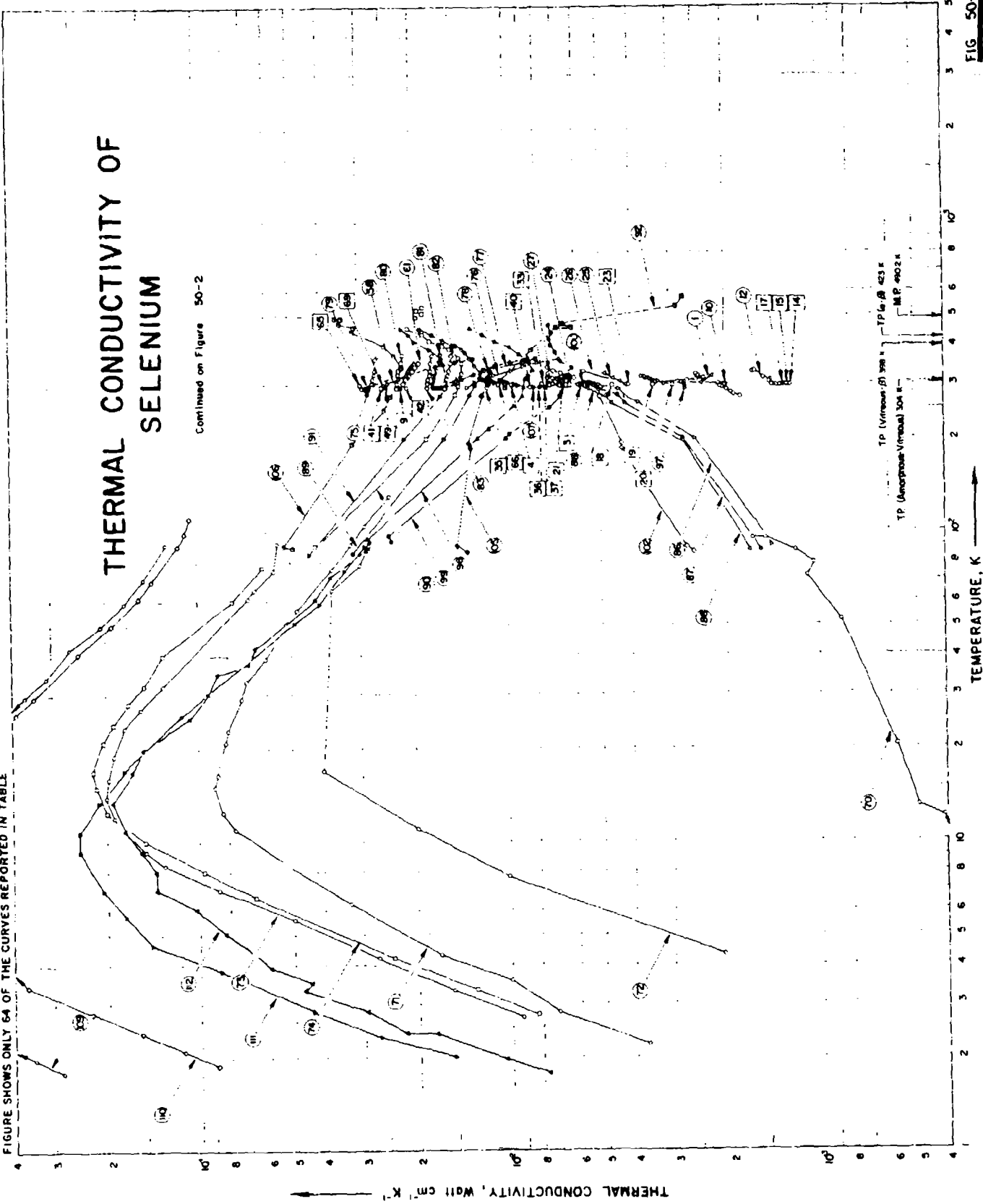
RECOMMENDED VALUES <sup>a</sup> (for Polycrystalline)			
$T_1$	$k_1$	$k_2$	$T_2$
0	0	0	-456.7
1	(0.00706) <sup>b</sup>	(6.408)	-457.9
2	0.0140	0.909	-456.1
3	0.0208	1.20	-454.3
4	0.0276	1.55	-452.5
5	0.0344	1.99	-450.7
6	0.0412	2.38	-448.9
7	0.0479	2.77	-447.1
8	0.0545	3.15	-445.3
9	0.0612	3.54	-443.5
10	0.0678	3.92	-441.7
11	0.0738	4.26	-439.9
12	0.0797	4.60	-438.1
13	0.0855	4.94	-436.3
14	0.0910	5.26	-434.5
15	0.0960	5.55	-432.7
16	0.101	5.84	-430.9
18	0.109	6.30	-427.3
20	0.117	6.76	-423.7
25	0.130	7.51	-414.7
30	0.137	7.92	-405.7
35	0.138	7.97	-396.7
40	0.136	7.97	-387.7
45	0.136	7.86	-378.7
50	0.135	7.80	-369.7
60	0.136	7.86	-351.7
70	0.138	7.97	-333.7
80	0.139	8.03	-315.7
90	0.141	8.15	-297.7
100	0.143	8.26	-279.7
150	0.149	8.61	-189.7
200	0.153	8.84	-99.7
250	0.156	9.01	-9.7
273.2	0.157	9.07	32.0
300	0.158	9.13	80.3

<sup>a</sup>  $T_1$  in K,  $k_1$  in Watt  $\text{cm}^{-1} \text{K}^{-1}$ ,  $T_2$  in F, and  $k_2$  in  $\text{Btu hr}^{-1} \text{ft}^{-1} \text{F}^{-1}$ . Values in parentheses are extrapolated.

# THERMAL CONDUCTIVITY OF SELENIUM

Continued on Figure 50-2

FIGURE SHOWS ONLY 64 OF THE CURVES REPORTED IN TABLE



TP (Vireon) 398 K TP (B) 423 K  
 TP (Acropose-Vireon) 304 K M.P. 402.4 K

# THERMAL CONDUCTIVITY OF SELENIUM

CONTINUED FROM FIGURE 50-1

FIGURE SHOWS ONLY 64 OF THE CURVES REPORTED IN TABLE

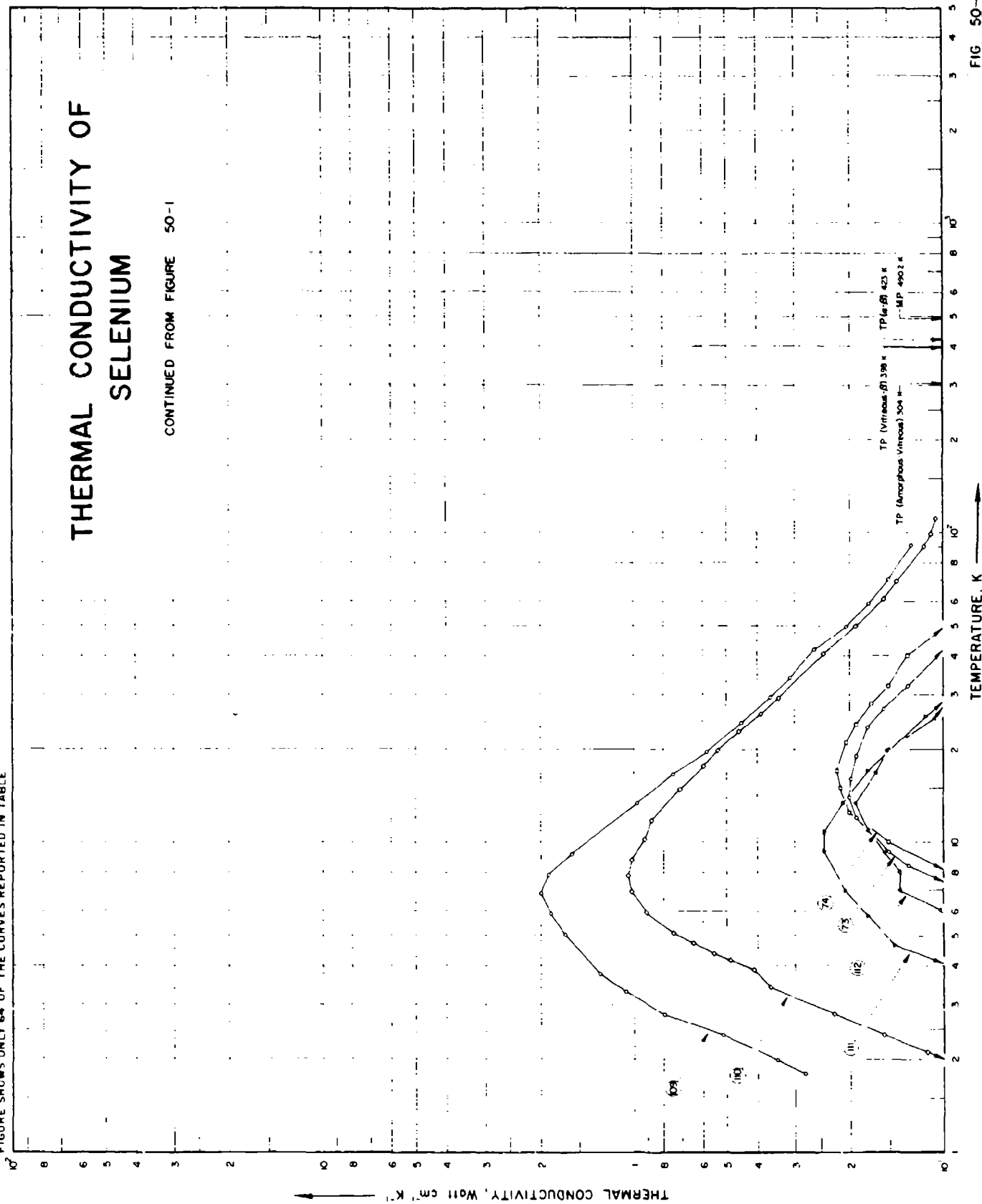


FIG 50-2

## SPECIFICATION TABLE NO. 50 THERMAL CONDUCTIVITY OF SELENIUM

(Impurity &lt; 0.20% each; total impurities &lt; 0.50%)

[ For Data Reported in Figure and Table No. 50 ]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition ( weight percent), Specifications and Remarks
1	357	L	1943	301-319	< 1.8		Pure.
2	358	L	1957	300.7	< 4		99.996 pure; vitreous and amorphous; specimen 20 mm in dia; polished.
3	358	L	1957	300.7	< 4		0.065 Br; same structure and dimensions as the above specimen; prepared by fusion in a molybdenum crucible, solidified and polished.
4	358	L	1957	300.7	< 4		0.13 Br; same structure, dimensions, and fabrication method as above.
5	358	L	1957	300.7	< 4		0.16 Br; same structure, dimensions, and fabrication method as above.
6	358	L	1957	293.2	< 4		99.996 pure; hexagonal crystalline; specimen 18 mm in dia; polished.
7	358	L	1957	293.2	< 4		0.065 Br; same structure and dimensions as the above specimen; prepared by melting vitreous selenium containing bromine in a ceramic crucible, pouring into a molybdenum beaker, first crystallization at 130 C for 30 min, then second crystallization at 200 C for 25 min, polished.
8	358	L	1957	293.2	< 4		0.13 Br; same structure, dimensions, and fabrication method as above.
9	358	L	1957	293.2	< 4		0.032 Br; same structure, dimensions, and fabrication method as above.
10	359	P	1956	273-323			Pure selenium from Merck; thermal conductivity values calculated from measured data of thermal diffusivity, specific volume, and the specific heat data taken from Tamnana, G. and Von Gronow, H. E. (Z. Anorg. Allg. Chemie, 192, 193, 1930).
11	360	L	1917	298.2		Disc-1	Vitreous selenium; 6.5 cm in dia and about 0.5 cm thick; cast in a hot iron mould, aged for 7 yrs.
12	360	L	1917	297-331		Disc-2	Vitreous selenium; 6.5 cm dia x 0.7523 cm thick; cast in a hot iron mould, aged for 1 to 8 days.
13	360	L	1917	298.2		Disc-2	The above specimen re-tested after being aged for 1 yr.
14	360	L	1917	298.2		Disc-3	Vitreous selenium; 6.5 cm dia x ~0.5 cm thick; cast in a hot iron mould, aged for 10 days.
15	360	L	1917	298.2		Disc-3	The above specimen re-tested after being aged for 1 yr.
16	360	L	1917	298.2		Disc-4	Similar to above but prepared from highly purified selenium and aged for 10 days.
17	360	L	1917	298.2		Disc-5	Similar to above but aged for 2 days.
18	360	L	1917	298.2		Disc A-1	Crystalline specimen 6.5 cm dia x ~0.5 cm thick; prepared by heating the vitreous disk in an oil oven to 160 C for 1 hr, cooled slowly, ground and polished; aged for 11 days.
19	360	L	1917	298.2		Disc A-II	The above specimen aged for 164 days.
20	360	L	1917	298.2		Disc A-III	The above specimen aged for 1 yr.
21	360	L	1917	298.2		Disc B-I	Similar to the above specimen but prepared by heating at 170 C and aged for 16 days.
22	360	L	1917	298.2		Disc B-II	The above specimen aged for 134 days.

SPECIFICATION TABLE NO. 50 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
23	360	L	1917	298.2		Disc B-III	The above specimen aged for 1 yr.
24	360	L	1917	298-326		Disc C-I	Crystalline specimen; 6.5 cm dia x 0.5774 cm thick; prepared by heating at 180 C and aged for 38 days.
25	360	L	1917	298-326		Disc C-II	The above specimen re-tested after being aged for 95 days.
26	360	L	1917	298-330		Disc C-III	The above specimen re-tested after being aged for 1 yr.
27	360	L	1917	298-325		Disc D-I	Similar to the above specimen but prepared by heating at 192 C and aged for 28 days.
28	360	L	1917	298.2		Disc D-II	The above specimen re-tested after being aged for 148 days.
29	360	L	1917	298.2		Disc D-III	The above specimen re-tested after being aged for 1 yr.
30	360	L	1917	298.2		Disc E-I	Similar to the above specimen but prepared by heating at 200 C and aged for 9 days.
31	360	L	1917	298.2		Disc E-II	The above specimen re-tested after 156 days.
32	360	L	1917	298.2		Disc E-III	The above specimen re-tested after 1 yr.
33	360	L	1917	298.2		Disc F-I	Similar to the above specimen but prepared by heating at 214 C and aged for 42 days.
34	360	L	1917	298.2		Disc F-II	Similar to the above specimen but prepared by heating at 214 C and aged for 42 days.
35	361	L	1957	294		Disc F-III	The above specimen re-tested after being aged for 1 yr.
36	361	L	1957	294		1	The above specimen re-tested after being aged for 1 yr.
37	361	L	1957	294		1	99.994 pure; amorphous; 20 mm dia cylindrical specimen.
38	361	L	1957	294		1	0.0015 Cl; amorphous; 20 mm dia cylindrical specimen.
39	361	L	1957	294		1	Similar to the above specimen but doped with 0.015 Cl.
40	361	L	1957	294		1	Similar to the above specimen but doped with 0.03 Cl.
41	361	L	1957	294		2	Similar to the above specimen but doped with 0.06 Cl.
42	361	L	1957	294		2	Similar to the above specimen but doped with 0.125 Cl.
43	361	L	1957	294		2	Similar to the above specimen but doped with 0.03 Cl.
44	361	L	1957	294		2	Similar to the above specimen but doped with 0.06 Cl.
45	361	L	1957	294		3	Similar to the above specimen but doped with 0.125 Cl.
46	361	L	1957	294		3	99.994 pure; crystalline; 20 mm dia cylindrical specimen; prepared from vitreous form by heating at 130 C for 40 min.
47	361	L	1957	294		3	Similar to above but doped with 0.015 Cl.
48	361	L	1957	294		3	Similar to above specimen but doped with 0.03 Cl.
49	361	L	1957	294		3	Similar to above specimen but doped with 0.06 Cl.
50	361	L	1957	294		3	Similar to above specimen but doped with 0.125 Cl.

SPECIFICATION TABLE NO. 50 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
51	362	L	1957	291.7		1	99.996 pure; amorphous.
52	362	L	1957	291.7		2	Amorphous selenium doped with 0.0035 Cl.
53	362	L	1957	291.7		3	Similar to the above specimen but doped with 0.007 Cl.
54	362	L	1957	291.7		4	Similar to the above specimen but doped with 0.015 Cl.
55	362	L	1957	291.7		5	Similar to the above specimen but doped with 0.03 Cl.
56	362	L	1957	291.7		6	Similar to the above specimen but doped with 0.06 Cl.
57	362	L	1957	291.7		7	Similar to the above specimen but doped with 0.125 Cl.
58	363	L	1958	363-353		1	99.996 pure; crystalline.
59	363	L	1958	303-353		2	Crystalline selenium doped with 0.0035 Cl.
60	363	L	1958	303-353		3	Crystalline selenium doped with 0.007 Cl.
61	363	L	1958	303-353		4	Crystalline selenium doped with 0.015 Cl.
62	363	L	1958	303-353		5	Crystalline selenium doped with 0.06 Cl.
63	363	L	1958	303-353		6	Crystalline selenium doped with 0.03 Cl.
64	364	L	1957	299.2			99.994 pure; amorphous.
65	364	L	1957	299.2			99.994 pure; crystalline specimen formed by heating vitreous selenium at 214 C.
66	364	L	1957	299.2			0.009 I; amorphous specimen.
67	364	L	1957	299.2			0.009 I; crystalline specimen.
68	364	L	1957	299.2			0.103 I; amorphous specimen.
69	364	L	1957	299.2			0.103 I; crystalline specimen.
70	365	L	1958	1.9-95		Se-1	Glassy specimen ~3 cm long and 1 cm dia; prepared by melting selenium powder (of probable 99.9% purity) in a split brass mold at about 250 C and quenching rapidly in ice water.
71	365	L	1958	2.2-130		Se-2	Polycrystalline; same dimensions and preparation method as above.
72	365	L	1958	4.3-90		Se-3	Crystalline; 6 cm long, 1 cm in dia; supplied by Fairmount Chemical Co. (Newark, N.J.).
73	365	L	1958	2.7-92		Se-4	Polycrystalline specimen ~4 cm long and 4 mm in dia; average grain dimensions ~20 $\mu$ ; produced by melting 99.999 pure selenium powder (from Canadian Copper Refiners Ltd) under vacuum in a glass tube, chilling rapidly to produce solid rod of glassy selenium, and annealing in vacuo at about 210 C for 50-60 hrs.
74	365	L	1958	2.8-77		Se-5	Similar to the above specimen but having a regular triangular cross section of about 0.26 cm <sup>2</sup> .

SPECIFICATION TABLE NO. 50 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
75	522	L	1961	293-363			99.996 pure; polycrystalline; specimen about 10 mm thick and 16 mm in dia; annealed at 110 C and 210 C for 1 hr.
76	522	L	1961	293-363			0.05 Tl; polycrystalline selenium specimen of similar dimensions as above; prepared by melting together 99.996 pure selenium and Tl <sub>2</sub> Se in a vacuum of 10 <sup>-4</sup> mm Hg; annealed at 110 C and 210 C for 1 hr.
77	522	L	1961	293-363			Similar to the above specimen but doped with 0.0125 Tl.
78	522	L	1961	293-363			Similar to the above specimen but doped with 0.1 Tl.
79	791	L	1963	293-373		1	99.996 pure; crystalline; ~10 mm dia cylindrical specimen; heated at 215 C for 8 hrs.
80	791	L	1963	293-373		2	Similar to the above specimen but doped with 0.01 Bi.
81	791	L	1963	293-373		3	Similar to the above specimen but doped with 0.02 Bi.
82	791	L	1963	293-373		4	Similar to the above specimen but doped with 0.04 Bi.
83	791	L	1963	293-373		5	Similar to the above specimen but doped with 0.06 Bi.
84	791	L	1963	293-373		6	Similar to the above specimen but doped with 0.08 Bi.
85	791	L	1963	293-373		7	Similar to the above specimen but doped with 0.1 Bi.
86	792, 1012	L	1963	90-300	3-5	V-3	Amorphous; about 10 mm in dia; prepared from the melt of 99.999 pure selenium by rapid cooling in vacuum.
87	792, 1012	L	1963	87-300	3-5	V-4	Similar to above but prepared from 99.9999 pure selenium.
88	792, 1012	L	1963	87-300	3-5	V-5	Similar to above but prepared from 99.99999 pure selenium.
89	792, 1012	L	1963	85-453	3-5	V-3	Crystalline; about 10 mm in dia; prepared from the amorphous specimen V-3 by annealing in vacuum at 210 C for 50 hrs.
90	792, 1012	L	1963	90-453	3-5	V-4	Similar to above but prepared from the amorphous specimen V-4.
91	792, 1012	L	1963	85-455	3-5	V-5	Similar to above but prepared from the amorphous specimen V-5.
92	793, 934	P	1964	293-573	±10		Data cover both solid and liquid state.
93	805	L	1966	294-313	3	1	Amorphous selenium, glass-formation temp ~31 C.
94	805	L	1966	294-313	3	3	Amorphous selenium irradiated by an electron beam with an energy of 5 MeV for 30 min.
95	805	L	1966	294-313	3	2	Similar to above but irradiated for only 10 min.
96	805	L	1966	288-318	3	1	Amorphous selenium.
97	805	L	1966	288-318	3	3	0.197 P; amorphous.
98	806	L	1966	87-455			Hexagonal single crystal grown out of a melt of grade B5 selenium (99.99999 pure); each crystal being 15 x 2 x 2 mm in size; specimen dimensions 7 x 6 x 4 mm; measurement carried out in darkness under a vacuum of 10 <sup>-4</sup> mm Hg; heat flow parallel to crystal axis.

SPECIFICATION TABLE NO. 50 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
99	806	L	1966	93-465			Similar to above but measured perpendicular to crystal axis.
100	898	L	1966	293-315	3-5	B-5	99.99999 pure; prepared from the melt by rapid cooling in vacuum; vitrification temp 31 C.
101	898	L	1966	293-315	3-5		Doped with 0.05 Cd; prepared from the melt by rapid cooling in vacuum; vitrification temp 32.5 C.
102	898	L	1966	89-341	3-5	B-4	99.9999 pure; amorphous; prepared from the melt by rapid cooling in vacuum.
103	898	L	1966	86-338	3-5	B-4	Similar to the above specimen except annealed at 373 K for 0.5 hr.
104	898	L	1966	88-335	3-5	B-4	Similar to the above specimen except annealed at 373 K for 2 hrs.
105	898	L	1966	86-338	3-5	B-4	Similar to the above specimen except annealed at 373 K for 10 hrs.
106	898	L	1966	89-533	3-5		Crystalline specimen prepared from the amorphous phase (specimen B-4) by annealing in vacuum at 210 C for 60 hrs; includes the liquid phase.
107	898	L	1966	290-317	3-5		Doped with 0.05 Tl; amorphous specimen, prepared from the melt of 99.99999 pure selenium with admixture of thallium by rapid cooling in vacuum.
108	898	L	1966	291-317	3-5		Similar to the above specimen except doped with 0.125 Tl.
109	961	L	1967	1.8-94		A	Single crystal; specimen 1.46 mm <sup>2</sup> in cross section and 1.21 mm long; grown from the vapor phase; heat flow parallel to the c-axis (additional information and the tabulated data obtained from author).
110	961	L	1967	1.9-112		B	Single crystal; specimen 0.973 mm <sup>2</sup> in cross section and 0.98 mm long; grown from the melt; heat flow parallel to the c-axis (additional information and the tabulated data obtained from author).
111	961	L	1967	2.0-90		C	Cut from the same crystal as the above specimen. In the form of an almost circular platelet 12.1 mm in dia and 1.2 mm thick, with c-axis parallel to the flat faces; measured in the direction perpendicular to both the thickness and the c-axis, in the central portion of the platelet across a length of 2.5 mm with effective cross-section 15 mm <sup>2</sup> (additional information and the tabulated data obtained from author).
112	961	L	1967	1.8-89		D	Cut from the same crystal as the above specimen. In the form of an almost circular platelet 12.1 mm in dia and 1.60 mm thick, with c-axis parallel to the flat faces; measured in the direction perpendicular to both the thickness and the c-axis, in the central portion of the platelet across a length of 3.66 mm with effective cross-section 19.4 mm <sup>2</sup> (additional information and the tabulated data obtained from author).





DATA TABLE NO. 50 (continued)

T	k	T	k	T	k	T	k	T	k	T	k	T	k	T	k
<u>CURVE 61 (cont.)</u>		<u>CURVE 70</u>		<u>CURVE 75</u>		<u>CURVE 80</u>		<u>CURVE 85</u>		<u>CURVE 89 (cont.)</u>		<u>CURVE 94</u>		<u>CURVE 98</u>	
343.2	0.0195	1.9	0.00023	293.2	0.0240	293	0.0223	353	0.0190	315	0.0121	353	0.0190	315	0.0121
353.2	0.0190	2.4	0.00025	303.2	0.0264	303	0.0222	373	0.0189	328	0.0121	373	0.0189	328	0.0121
<u>CURVE 62</u>		<u>CURVE 73</u>		<u>CURVE 76</u>		<u>CURVE 81</u>		<u>CURVE 86</u>		<u>CURVE 90</u>		<u>CURVE 91</u>		<u>CURVE 92</u>	
343.2	0.0197	3.1	0.00027	323.2	0.0209	293	0.0172	313	0.0210	390	0.0138	390	0.0210	390	0.0138
380	0.0134	4.1	0.00036	343.2	0.0146	303.2	0.0126	363.2	0.0216	400	0.0146	400	0.0210	400	0.0146
380	0.0134	4.4	0.00035	353.2	0.0126	313.2	0.0117	373	0.0213	415	0.0159	415	0.0209	415	0.0159
390	0.0134	6.8	0.00037	363.2	0.0126	323.2	0.0109	383	0.0207	435	0.0172	435	0.0207	435	0.0172
390	0.0134	8.8	0.000375	<u>CURVE 74</u>		333	0.0170	393	0.0205	453	0.0188	<u>CURVE 93</u>		390	0.0134
390	0.0134	12.0	0.00041	2.7	0.00093	343.2	0.0096	393	0.0172	90	0.0418	90	0.0418	200	0.0180
390	0.0134	13.0	0.00050	3.3	0.0155	343.2	0.0088	393	0.0157	200	0.0142	200	0.0180	277	0.0146
390	0.0134	20.5	0.00058	4.2	0.027	353.2	0.0088	393	0.0155	277	0.0146	277	0.0146	295	0.0142
390	0.0134	20.5	0.00066	5.6	0.050	363.2	0.0079	393	0.0177	295	0.0142	295	0.0142	325	0.0138
390	0.0134	72.0	0.00110	7.0	0.087	363.2	0.0067	393	0.0175	325	0.0138	325	0.0138	350	0.0138
390	0.0134	79.0	0.00105	8.4	0.130	363.2	0.0067	393	0.0173	350	0.0138	350	0.0138	365	0.0142
390	0.0134	87.0	0.00120	9.3	0.150	363.2	0.0067	393	0.0151	365	0.0142	365	0.0142	390	0.0138
390	0.0134	95.0	0.00150	12.0	0.190	363.2	0.0067	393	0.0149	390	0.0138	390	0.0138	400	0.0159
390	0.0134	95.0	0.00165	14.0	0.200	363.2	0.0067	393	0.0149	400	0.0159	400	0.0159	415	0.0176
<u>CURVE 63</u>		<u>CURVE 71</u>		<u>CURVE 77</u>		<u>CURVE 82</u>		<u>CURVE 87</u>		<u>CURVE 91</u>		<u>CURVE 91</u>		<u>CURVE 92</u>	
303.2	0.0210	2.2	0.0037	293.2	0.0167	293	0.0157	293	0.0157	85	0.0435	85	0.0435	85	0.0435
313.2	0.0205	2.8	0.0071	313.2	0.0142	303	0.0157	293	0.0157	200	0.0142	200	0.0142	200	0.0142
323.2	0.0201	3.55	0.0101	333.2	0.0117	313	0.0155	303	0.0155	277	0.00439	277	0.00439	277	0.00439
343.2	0.0196	4.3	0.017	353.2	0.0088	323	0.0177	313	0.0177	290	0.00502	290	0.00502	290	0.00502
343.2	0.0192	6.3	0.033	363.2	0.0079	333	0.0153	323	0.0176	300	0.00544	300	0.00544	300	0.00544
353.2	0.0188	8.3	0.0515	363.2	0.0079	353	0.0151	323	0.0175	325	0.0155	325	0.0155	325	0.0155
<u>CURVE 64</u>		<u>CURVE 72</u>		<u>CURVE 78</u>		<u>CURVE 83</u>		<u>CURVE 88</u>		<u>CURVE 91</u>		<u>CURVE 91</u>		<u>CURVE 92</u>	
299.2	0.0128	11.0	0.077	293.2	0.0159	293	0.0157	293	0.0179	87	0.00155	87	0.00155	87	0.00155
299.2	0.0128	12.5	0.084	303.2	0.0146	303.2	0.0157	293	0.0179	200	0.00272	200	0.00272	200	0.00272
299.2	0.0128	15.0	0.089	313.2	0.0138	313.2	0.0157	303	0.0178	260	0.00418	260	0.00418	260	0.00418
299.2	0.0128	16.5	0.087	323.2	0.0126	323.2	0.0157	303	0.0178	277	0.00418	277	0.00418	277	0.00418
299.2	0.0128	23.0	0.080	333.2	0.0117	333.2	0.0155	313	0.0177	300	0.00544	300	0.00544	300	0.00544
299.2	0.0128	29.0	0.073	343.2	0.0108	353.2	0.0088	313	0.0177	325	0.0155	325	0.0155	325	0.0155
299.2	0.0128	33.0	0.070	363.2	0.0096	363.2	0.0079	323	0.0176	350	0.0138	350	0.0138	350	0.0138
299.2	0.0128	39.0	0.061	363.2	0.0084	363.2	0.0067	323	0.0175	365	0.0142	365	0.0142	365	0.0142
299.2	0.0128	56.0	0.048	363.2	0.0084	363.2	0.0067	353	0.0175	390	0.0176	390	0.0176	390	0.0176
299.2	0.0128	76.0	0.037	363.2	0.0084	363.2	0.0067	373	0.0173	415	0.0192	415	0.0192	415	0.0192
299.2	0.0128	91.0	0.032	363.2	0.0084	363.2	0.0067	393	0.0173	435	0.0209	435	0.0209	435	0.0209
299.2	0.0128	105.0	0.029	363.2	0.0084	363.2	0.0067	393	0.0173	455	0.0230	455	0.0230	455	0.0230
299.2	0.0128	130.0	0.024	363.2	0.0084	363.2	0.0067	393	0.0173	455	0.0230	455	0.0230	455	0.0230
<u>CURVE 65</u>		<u>CURVE 73</u>		<u>CURVE 79</u>		<u>CURVE 84</u>		<u>CURVE 89</u>		<u>CURVE 91</u>		<u>CURVE 91</u>		<u>CURVE 92</u>	
299.2	0.0295	4.3	0.0021	293	0.0291	293	0.0195	293	0.0195	85	0.0314	85	0.0314	85	0.0314
299.2	0.0295	7.7	0.0102	303	0.0269	303	0.0195	303	0.0195	200	0.0159	200	0.0159	200	0.0159
299.2	0.0295	7.7	0.0102	313	0.0267	313	0.0193	313	0.0193	277	0.0130	277	0.0130	277	0.0130
299.2	0.0295	7.7	0.0102	323	0.0266	323	0.0192	323	0.0192	388	0.00825	388	0.00825	388	0.00825
299.2	0.0295	7.7	0.0102	333	0.0262	333	0.0191	333	0.0191	455	0.00682	455	0.00682	455	0.00682
299.2	0.0295	7.7	0.0102	353	0.0262	353	0.0191	353	0.0191						
299.2	0.0295	7.7	0.0102	373	0.0258	373	0.0191	373	0.0191						

\* Not shown on plot

DATA TABLE NO. 50 (continued)

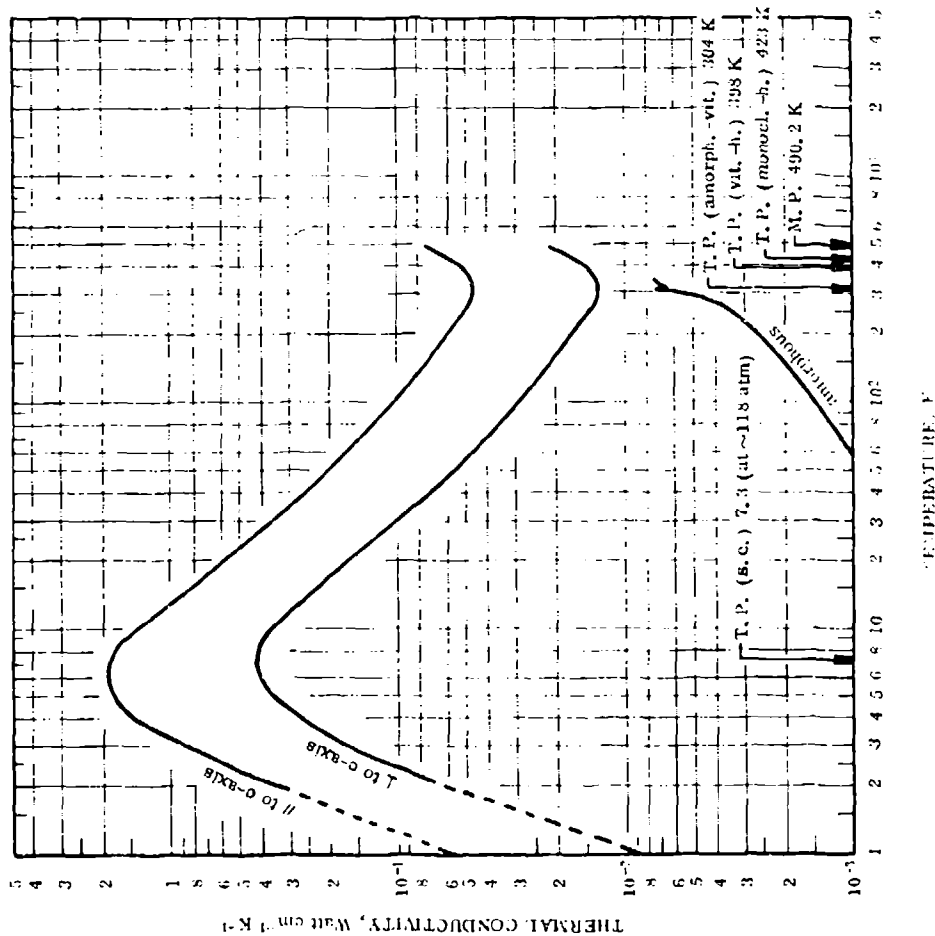
CURVE 92 (cont.)		CURVE 96		CURVE 98 (cont.)		CURVE 100 (cont.)		CURVE 103 (cont.)*		CURVE 106 (cont.)*		CURVE 109		CURVE 110 (cont.)	
T	k	T	k	T	k	T	k	T	k	T	k	T	k	T	k
455	0.00610	288.2	0.00502	215	0.0114	311.1	0.00671*	313	0.00757	471	0.0337	1.77	0.272	62.5	0.146
471	0.00855	290.0	0.00502	255	0.00941	312.2	0.00679	321	0.00771	487	0.0335	1.98	0.338	70.8	0.133
535	0.00295	292.0	0.00502	275	0.00879	313.2	0.00695*	328	0.00775	492	0.0350	2.36	0.51	92.1	0.109
573	0.00270	294.2	0.00502	293	0.00870	315.0	0.00727	332	0.00786	494	0.0192	2.80	0.78	100.2	0.104
<u>CURVE 93</u>		295.7	0.00502	313	0.00858	<u>CURVE 101*</u>		335	0.00786	508	0.0184	3.35	1.05	112.0	0.101
294.2	0.00494	297.5	0.00523	333	0.00816	<u>CURVE 104*</u>		524	0.0191	524	0.0184	5.10	1.63	<u>CURVE 111</u>	
297.3	0.00494	299.5	0.00544	355	0.00837	292.6	0.00411	<u>CURVE 107</u>		533	0.0184	6.08	1.92	1.96	0.0154
300.0	0.00502	301.2	0.00523	375	0.00900	296.5	0.00416	88	0.00808	<u>CURVE 108*</u>		7.02	1.96	2.34	0.0266
302.8	0.00519	303.2	0.00502	395	0.00962	299.2	0.00413	185	0.00834	290.8	0.00766*	8.04	1.84	2.78	0.043
303.2	0.00552	303.7	0.00606	413	0.0107	301.6	0.00428	190	0.00874	293.4	0.00803	9.50	1.52	3.84	0.084
303.2	0.00552	304.2	0.00732	435	0.0117	303.8	0.00418	274	0.00874	293.4	0.00874*	17.1	0.72	4.59	0.141
303.7	0.00628	305.5	0.00669	455	0.0129	304.6	0.00421	295	0.00890	296.9	0.00874*	20.2	0.555	5.92	0.169
304.2	0.00707	306.2	0.00649	308	0.00886	305.5	0.00743	303	0.00870	298.9	0.00933	25.6	0.428	7.04	0.202
305.4	0.00669	308.5	0.00669	315	0.00890	306.7	0.00680	308	0.00886	300.5	0.00933	30.1	0.345	9.45	0.236
306.4	0.00644	309.7	0.00690	93	0.0236	306.9	0.00675	315	0.00890	302.0	0.0108*	35.2	0.295	11.0	0.226
307.2	0.00628	311.7	0.00732	97	0.0241	308.8	0.00663	323	0.00896	302.9	0.0108*	43.1	0.247	13.8	0.203
308.4	0.00644	313.2	0.00753	97	0.0238*	311.7	0.00711	330	0.00911	303.2	0.0102*	51.6	0.194	17.2	0.166
310.0	0.00644	315.0	0.00774	200	0.010	313.2	0.00745	335	0.00928	304.2	0.0101	60.7	0.166	25.8	0.109
310.0	0.00653	316.7	0.00816	207	0.00975	315.1	0.00774	<u>CURVE 105</u>		305.9	0.0107	72.7	0.142	51.2	0.047
311.4	0.00653	318.0	0.00837	253	0.00724	<u>CURVE 102</u>		86	0.0134	308.9	0.0110	93.8	0.120	59.1	0.0388
313.4	0.00665	320	0.00640*	275	0.00669*	86	0.00252	90	0.0144	312.4	0.0112*	75.5	0.0322	75.5	0.0322
<u>CURVE 94</u>		340	0.00266	320	0.00640	89	0.00266	186	0.0129	314.4	0.0113*	89.6	0.0285	<u>CURVE 112</u>	
294.2	0.00707	288.2	0.00266	340	0.00607	89	0.00266	186	0.0129	317.0	0.0115*	1.88	0.087	1.76	0.0077
297.2	0.00690	291.5	0.00264*	360	0.00669	186	0.00428	192	0.0131	<u>CURVE 109*</u>		2.09	0.112	1.96	0.0108
297.5	0.00686	293.2	0.00268*	380	0.00690	193	0.00428	275	0.0118*	290.9	0.0141	2.38	0.152	2.36	0.0172
297.7	0.00686	295.5	0.00268*	400	0.00711	275	0.00496	294	0.0115*	292.0	0.0141	3.38	0.355	2.79	0.0292
299.6	0.00686	297.3	0.00268*	420	0.00724	293	0.00496	300	0.0113	294.0	0.0138	4.40	0.472	2.36	0.0216
301.8	0.00669	299.3	0.00268*	440	0.00711	299	0.00510*	307	0.0114	297.0	0.0140	4.38	0.472	3.29	0.0461
304.2	0.00661	299.7	0.00268	465	0.00724	303	0.00556*	316	0.0113*	299.1	0.0139	4.38	0.535	3.48	0.0438
305.8	0.00563	300.7	0.00310*	304	0.00724	304	0.00556*	331	0.0112*	301.3	0.0141	4.75	0.63	3.91	0.0588
309.1	0.00644	304.7	0.00320*	314	0.00559	314	0.00559	338	0.0111*	302.3	0.0141	5.16	0.725	5.01	0.081
313.2	0.00619	306.7	0.00326	319	0.00564*	319	0.00564*	<u>CURVE 106</u>		303.3	0.0145	6.1	0.88	6.03	0.10
<u>CURVE 95</u>		308.5	0.00331*	324	0.00509*	324	0.00509*	89	0.0489	304.4	0.0147	7.0	0.98	7.08	0.134
294.2	0.00655	297.2	0.00339	297.2	0.00509*	331	0.00704	90	0.0520	304.4	0.0147	7.88	1.02	8.02	0.133
295.4	0.00659	309.7	0.00341	299.7	0.00521	336	0.00720*	90	0.0520	307.0	0.0147	8.88	0.98	8.02	0.133
300.2	0.00703	311.0	0.00341	311.0	0.00531	341	0.00727	193	0.0313	308.3	0.0147	10.5	0.88	9.38	0.149
303.2	0.00711	314.0	0.00354	302.9	0.00544*	341	0.00727	248	0.0245	309.6	0.0147	12.1	0.83	11.1	0.169
306.0	0.00828	316.0	0.00358	303.3	0.00545*	341	0.00727	274	0.0233	311.1	0.0147	15.2	0.67	13.6	0.185
294.2	0.00655	318.0	0.00368	303.6	0.00551	86	0.00363	300	0.0216*	312.8	0.0147	17.3	0.57	17.2	0.158
295.4	0.00659	301.8	0.00368	303.6	0.00551	183	0.00532	318	0.0216*	314.3	0.0148	20.1	0.505	25.6	0.102
300.2	0.00703	303.2	0.00368	304.1	0.00599*	188	0.00665	339	0.0227	316.7	0.0147	23.4	0.43	30.3	0.0885
303.2	0.00711	304.9	0.00368	304.8	0.00665	273	0.00630	371	0.0227	<u>CURVE 103*</u>		26.6	0.367	35.2	0.083
306.0	0.00828	307.4	0.00368	306.3	0.00665	297	0.00620*	390	0.0252	87	0.0285	30.1	0.322	37.8	0.0665
307.4	0.00828	307.4	0.00620*	307.1	0.00620*	297	0.00620*	300	0.00634*	93	0.0277	40.9	0.229	42.6	0.0635
310.9	0.00841	308.3	0.00634*	308.3	0.00634*	300	0.00634*	309.9	0.00656	95	0.0282	40.9	0.229	<u>CURVE 110 (cont.)</u>	
313.1	0.00849	309.9	0.00656	309.9	0.00656	304	0.00656	304	0.00746	200	0.0123	50.9	0.182		

Not shown on plot

## DATA TABLE NO. 50 (continued)

T	k
51.3	0.0494
60.9	0.0402
72.6	0.036
88.7	0.0266

FIGURE AND TABLE NO. 50R RECOMMENDED THERMAL CONDUCTIVITY OF SELENIUM



REMARKS

The recommended values are for high-purity selenium. The recommended values for selenium single crystals that are supported by experimental thermal conductivity data are thought to be accurate to within 5% of the true values near room temperature and 5 to 15% at other temperatures above 20 K. Below 20 K the values are intended only for indicating the general trend, since the thermal conductivity near and below the corresponding temperature of its maximum is highly sensitive to small physical and chemical variations of the specimens. The recommended values for amorphous selenium are thought to be accurate to within 10%.

<sup>a</sup> T<sub>1</sub> in K, k<sub>1</sub> in Watt cm<sup>-1</sup> K<sup>-1</sup>, T<sub>2</sub> in °F, and k<sub>2</sub> in Btu hr<sup>-1</sup> ft<sup>-1</sup> F<sup>-1</sup>.

<sup>b</sup> Values in parentheses are extrapolated or estimated.

RECOMMENDED VALUES<sup>a</sup>

T <sub>1</sub>	Single Crystal (// to c-axis)			Single Crystal (⊥ to c-axis)			T <sub>2</sub>
	k <sub>1</sub>	k <sub>2</sub>	k <sub>3</sub>	k <sub>1</sub>	k <sub>2</sub>	k <sub>3</sub>	
0	0	0	0	0	0	0	-459.7
1	(0.0563) <sup>b</sup>	(3.25)	(0.00865)	(0.00865)	(0.500)	(0.500)	-457.9
2	0.353	20.4	(0.0600)	(0.0600)	(3.47)	(3.47)	-456.1
3	0.855	49.4	0.160	0.160	9.24	9.24	-454.3
4	1.41	81.5	0.268	0.268	15.5	15.5	-452.5
5	1.75	101	0.348	0.348	20.1	20.1	-450.7
6	1.92	111	0.395	0.395	22.8	22.8	-448.9
7	1.90	110	0.410	0.410	23.7	23.7	-447.1
8	1.78	103	0.406	0.406	23.5	23.5	-445.3
9	1.61	93.0	0.350	0.350	22.5	22.5	-443.5
10	1.42	82.0	0.359	0.359	20.7	20.7	-441.7
11	1.26	72.8	0.329	0.329	19.0	19.0	-439.9
12	1.12	64.7	0.301	0.301	17.4	17.4	-438.1
13	1.01	58.4	0.276	0.276	15.9	15.9	-436.3
14	0.919	53.1	0.253	0.253	14.6	14.6	-434.5
15	0.844	48.8	0.234	0.234	13.5	13.5	-432.7
16	0.778	45.0	0.217	0.217	12.5	12.5	-430.9
18	0.672	36.8	0.189	0.189	10.9	10.9	-427.3
20	0.588	34.0	0.166	0.166	9.59	9.59	-423.7
25	0.448	25.9	0.127	0.127	7.34	7.34	-414.7
30	0.362	20.9	0.103	0.103	5.95	5.95	-405.7
35	0.303	17.5	0.0866	0.0866	5.00	5.00	-396.7
40	0.260	15.0	0.0743	0.0743	4.29	4.29	-387.7
45	0.228	13.2	0.0651	0.0651	3.76	3.76	-378.7
50	0.203	11.7	0.0580	0.0580	3.35	3.35	-369.7
60	0.168	9.71	0.0480	0.0480	2.77	2.77	-351.7
70	0.144	8.32	0.0411	0.0411	2.37	2.37	-333.7
80	0.126	7.29	0.0360	0.0360	2.08	2.08	-315.7
90	0.112	6.47	0.0320	0.0320	1.85	1.85	-297.7
100	0.103	5.95	0.0294	0.0294	1.70	1.70	-279.7
150	0.0762	4.40	0.0218	0.0218	1.26	1.26	-169.7
200	0.0608	3.51	0.0174	0.0174	1.01	1.01	-99.7
250	0.0513	2.96	0.0147	0.0147	0.849	0.849	-
273.2	0.0481	2.78	0.0137	0.0137	0.792	0.792	32.0
300	0.0452	2.61	0.0130	0.0130	0.751	0.751	80.3
350	0.0461	2.66	0.0132	0.0132	0.763	0.763	170.3
400	0.0538	3.11	0.0154	0.0154	0.890	0.890	260.3
490.2	0.0747	4.32	0.0213	0.0213	1.23	1.23	422.7

TABLE NO. 50R (continued)

RECOMMENDED VALUES*		Amorphous					
$T_1$	$k_1$	$k_2$	$T_2$	$T_1$	$k_1$	$k_2$	$T_1$
0	0	0	-459.7	304	0.00732	0.423	87.5
1	(0.000130)*	(0.00751)	-457.9	304.5	0.00681	0.393	88.4
2	0.000236	0.0136	-456.1	305	0.00657	0.380	89.3
3	0.000290	0.0168	-454.3	305.5	0.00640	0.370	90.2
4	0.000323	0.0197	-452.5				
5	0.000342	0.0198	-450.7	306	0.00627	0.362	91.1
6	0.000358	0.0207	-448.9	306.5	0.00619	0.358	92.0
7	0.000374	0.0216	-447.1	307	0.00619	0.358	92.9
8	0.000390	0.0225	-445.3	307.5	0.00625	0.361	93.8
9	0.000405	0.0234	-443.5				
10	0.000420	0.0243	-441.7	308	0.00631	0.365	94.7
11	0.000435	0.0251	-439.9	309	0.00644	0.372	96.5
12	0.000450	0.0260	-438.1	310	0.00656	0.379	98.3
13	0.000465	0.0269	-436.3	320	0.00782	0.452	116.3
14	0.000480	0.0277	-434.5				
15	0.000494	0.0285	-432.7				
16	0.000508	0.0294	-430.9				
18	0.000534	0.0309	-427.3				
20	0.000560	0.0324	-423.7				
25	0.000619	0.0358	-414.7				
30	0.000675	0.0390	-405.7				
35	0.000730	0.0422	-396.7				
40	0.000788	0.0455	-387.7				
45	0.000843	0.0487	-378.7				
50	0.000920	0.0520	-369.7				
60	0.00102	0.0589	-351.7				
70	0.00113	0.0653	-333.7				
80	0.00125	0.0722	-315.7				
90	0.00136	0.0786	-297.7				
100	0.00148	0.0855	-279.7				
150	0.00204	0.118	-189.7				
200	0.00263	0.152	-99.7				
250	0.00360	0.208	-				
273.2	0.00428	0.247	32.0				
290	0.00484	0.280	62.3				
295	0.00504	0.291	71.3				
300	0.00528	0.305	80.3				
301	0.00533	0.308	82.1				
302	0.00538	0.311	83.9				
303	0.00544	0.314	85.7				
303.5	0.00547	0.316	86.6				

\*  $T_1$  in K,  $k_1$  in Watt cm<sup>-1</sup> K<sup>-1</sup>,  $T_2$  in F, and  $k_2$  in Btu hr<sup>-1</sup> ft<sup>-1</sup> F<sup>-1</sup>.

† Values in parentheses are extrapolated.

# THERMAL CONDUCTIVITY OF SILICON

FIGURE SHOWS ONLY 55 OF THE CURVES REPORTED IN TABLE

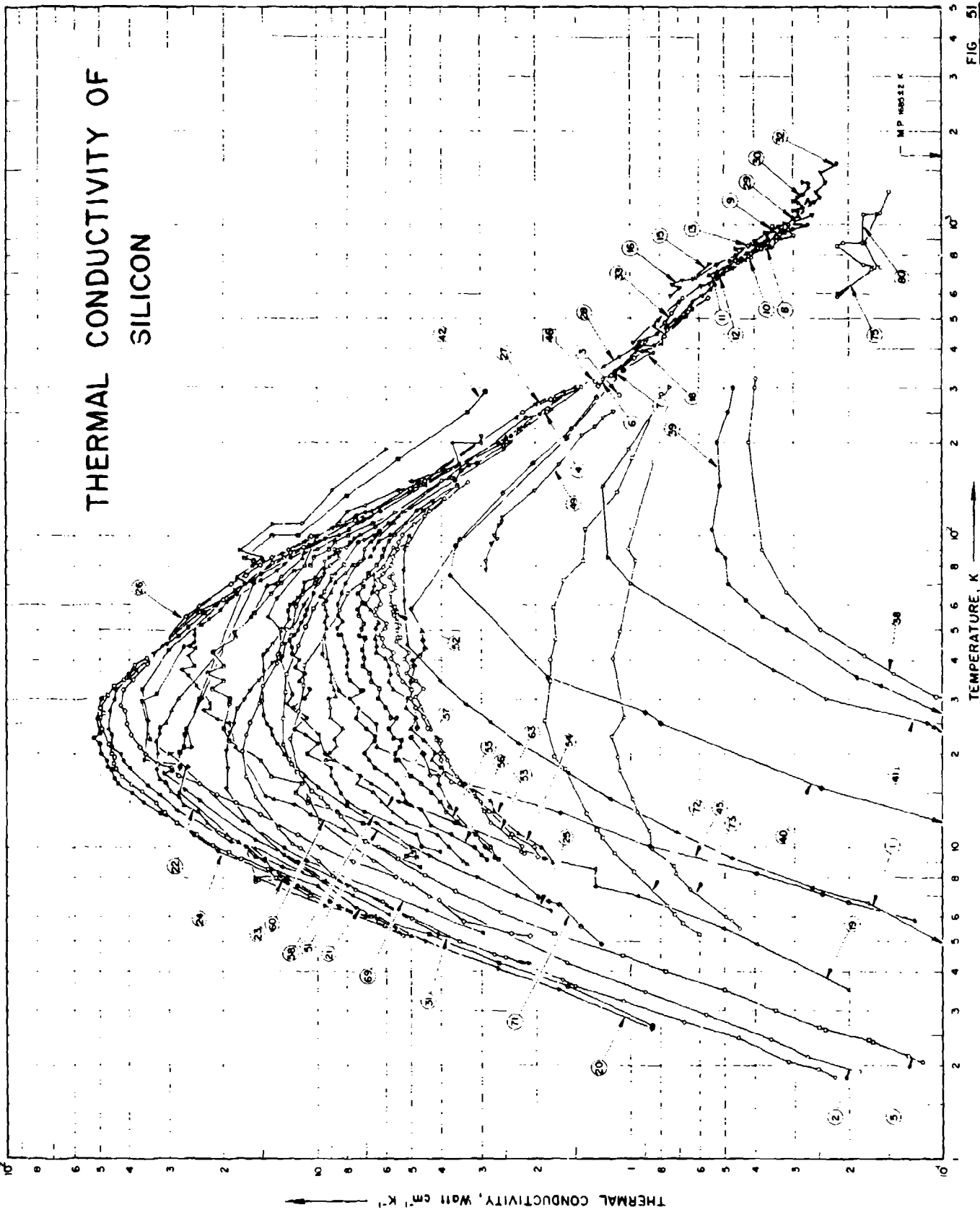


FIG. 51

## SPECIFICATION TABLE NO. 51 THERMAL CONDUCTIVITY OF SILICON

(Impurity  $\leq 0.20\%$  each; total impurities  $\leq 0.50\%$ )

[For Data Reported in Figure and Table No. 51.]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
1	274	L	1954	1.7-100		Si 1	High purity; polycrystalline, composed of fairly large crystallites about 0.2 mm in size; specimen 1 cm long, 1.5 mm <sup>2</sup> cross section; supplied by Messrs. Johnson Matthey Co.; nickel plated to solder thermal contacts, excess nickel dissolved with acid.
2	343	L	1956	1.9-149	1-5	Si 1	Pure single crystal; n-type; specimen cross sectional area 1.75 x 1.5 mm <sup>2</sup> ; electrical resistivity 6.7 ohm cm at 295 K.
3	351	L	1954	283	5-8	Si 2	Pure crystal.
4	354	L	1957	1.9-300		Si 2	Pure single crystal; n-type; axis of specimen along [100] direction; 5 x 0.215 x 0.198 cm; carrier concentration $5 \times 10^{14}$ cm <sup>-3</sup> ; electrical resistivity 19-25 ohm cm at room temp.
5	354	L	1957	2.1-80		Si 3	Gold-doped single crystal; p-type; axis of specimen [100] direction; 5 x 0.236 x 0.221 cm; carrier concentration $10^{15}$ cm <sup>-3</sup> ; electrical resistivity 18-26 ohm cm at room temp.
6	578	C	1960	303-579	$\pm 5$		Pure single crystal; p-type; electrical resistivity 3 ohm cm at room temp; FH stainless steel used as comparative material.
7	578	C	1960	328-533	$\pm 5$		Pure single crystal; n-type; electrical resistivity 3 ohm cm at room temp. FH stainless steel used as comparative material.
8	692, 745	C	1962	767, 846	$\pm 20$	KA-1 (Knappic)	p-type single crystal; 23 mm dia x 8 mm thick; specimen axis in (111) orientation; supplied by Knappic Electro-Physics; carrier concentration $10^{18}$ cm <sup>-3</sup> ; Armco iron used as comparative material.
9	692, 745	C	1962	769-989	$\pm 20$	KA-1 (Knappic)	Second run of the above specimen.
10	692, 745	C	1962	756-997	$\pm 20$	KA-1 (Knappic)	Third run of the above specimen.
11	692, 745	C	1962	687, 826	$\pm 20$	KA-1 (Knappic)	Fourth run of the above specimen.
12	692, 745	C	1962	679, 778	$\pm 20$	KB-1 (Knappic)	n-type single crystal; 23 mm dia x 8 mm thick; specimen axis in (111) orientation; supplied by Knappic Electro-Physics; carrier concentration $5 \times 10^{16}$ cm <sup>-3</sup> ; Armco iron used as comparative material.
13	692, 745	C	1962	857, 906	$\pm 20$	KB-2 (Knappic)	Similar to above.
14	692, 745	C	1962	748	$\pm 20$	KB-2 (Knappic)	Second run of above specimen.
15	692, 745	C	1962	669-1092	$\pm 20$	KB-2 (Knappic)	Third run of above specimen.



SPECIFICATION TABLE NO. 51 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
16	678	E	1962	589-1073			p-type single crystal; 0.25 in. dia x 1.5 in. long; corrected for isothermal conditions and shield.
17	678	E	1962	593-883			The above specimen held for some time at 1073 K; measured in the cooling-down period.
18	678	E	1962	334-473			The above specimen measured with the length of the thermocouple wire between junctions and shield increased by a factor of approx two to three.
19	679	L	1961	3, 5-210	10	Q-20	p-type single crystal; boron doped ( $1 \times 10^{15}$ atoms $\text{cm}^{-3}$ ); O concentration, $2 \times 10^{17}$ atoms $\text{cm}^{-3}$ ; dimensions $3 \times 3 \times 20$ mm; supplied by H. L. Taylor, Texas Instruments, Inc.; electrical resistivity reported as 7.2 ohm cm at 0 C.
20	679	L	1961	2, 6-190	10	M-1	p-type single crystal; boron doped ( $2 \times 10^{15}$ atoms $\text{cm}^{-3}$ ); oxygen concentration, $10^{17}$ $\text{cm}^{-3}$ ; dimensions $3 \times 3 \times 20$ mm; supplied by F. J. Bourassa, Electronics Chemical Div., Merck and Co., Inc.; electrical resistivity reported as 2600 ohm cm at 0 C.
21	680	L	1961	5, 5-246		K4	n-type single crystal; oxygen concentration, $1.4 \times 10^{18}$ $\text{cm}^{-3}$ ; specimen cross section $0.625 \times 0.625$ cm; carrier concentration $3.5 \times 10^{14}$ $\text{cm}^{-3}$ ; dislocation density of the order of $10^4$ $\text{cm}^{-2}$ ; electrical resistivity reported as 12 ohm cm at room temp.
22	680	L	1961	5, 2-208		K5	n-type single crystal; oxygen concentration, $6 \times 10^{17}$ $\text{cm}^{-3}$ ; specimen cross section $0.622 \times 0.622$ cm; carrier concentration $3.5 \times 10^{15}$ $\text{cm}^{-3}$ ; dislocation density of the order of $10^4$ $\text{cm}^{-2}$ ; electrical resistivity reported as 110 ohm cm at room temp.
23	680	L	1961	5, 3-200		M6	n-type single crystal; phosphorus doped; oxygen concentration, $10^{18}$ $\text{cm}^{-3}$ ; carrier concentration $1.1 \times 10^{15}$ $\text{cm}^{-3}$ ; specimen cross section, $0.634 \times 0.640$ cm; dislocation density of the order of $10^4$ $\text{cm}^{-2}$ ; electrical resistivity reported as 5 ohm cm at room temp.
24	689	L	1961	6-120		M4	n-type single crystal; phosphorus doped; oxygen concentration, $10^{18}$ $\text{cm}^{-3}$ ; carrier concentration $4 \times 10^{15}$ $\text{cm}^{-3}$ ; cross section $0.637 \times 0.629$ cm; dislocation density of the order of $10^4$ $\text{cm}^{-2}$ ; electrical resistivity reported as 260 ohm cm at room temp.
25	680	L	1961	6, 3-298		SA-1	p-type single crystal; boron doped; oxygen concentration, $7 \times 10^{17}$ $\text{cm}^{-3}$ ; cross section $0.616 \times 0.623$ cm; carrier concentration $4.8 \times 10^{15}$ $\text{cm}^{-3}$ ; electrical resistivity reported as 3, 0 ohm cm at room temp.
26	680	L	1961	5, 2-275		M3	p-type single crystal; boron doped; oxygen concentration, $10^{18}$ $\text{cm}^{-3}$ ; cross section $0.635 \times 0.632$ cm; carrier concentration $4.0 \times 10^{15}$ $\text{cm}^{-3}$ ; dislocation density of the order of $10^4$ $\text{cm}^{-2}$ ; electrical resistivity reported as 4, 5 ohm cm at room temp.
27	680	L	1961	5, 9-300		M2	p-type single crystal; boron doped; oxygen concentration, $10^{18}$ $\text{cm}^{-3}$ ; specimen cross section $0.630 \times 0.640$ cm; carrier concentration $4.0 \times 10^{14}$ $\text{cm}^{-3}$ ; dislocation density of the order of $10^4$ $\text{cm}^{-2}$ ; electrical resistivity reported as 45, 5 ohm cm at room temp.

SPECIFICATION TABLE NO. 51 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)	Specifications and Remarks
28	681, 660	P	1963	300-850		4A		p-type single crystal; specimen axis in [111] orientation; 0.9 cm in dia, 6 cm long; electrical resistivity reported as 107 ohm cm at 300 K; thermal conductivity values calculated from measured data of thermal diffusivity using the specific heat data taken from Dennison, D. H. (Institute for Atomic Research, Ames, Iowa) and density (determined by Smakula, A. and Sils, V.) $2.32502 \pm 3 \times 10^{-4} \text{ g cm}^{-3}$ at 298 K.
29	681, 660	P	1963	775-1200		1F		n-type single crystal; 0.9 cm in dia, 6 cm long; specimen axis in [111] orientation; electrical resistivity reported as 33.9, 50.0, 58.0, 42.0, 6.2, and 0.1 ohm cm at 300, 400, 460, 500, 650, and 1000 K, respectively; thermal conductivity values calculated by the same method as above.
30	681, 660	P	1963	1115-1370		3C		n-type single crystal; 0.9 cm in dia, 6 cm long; specimen axis in [100] orientation; electrical resistivity reported as 1010, 2000, 1700, 125, 6.2, and 0.1 ohm cm at 300, 375, 400, 500, 650, and 1000 K, respectively; thermal conductivity values calculated by the same method as above.
31	684	L	1964	4.3-304	$\pm 5$			High purity; p-type single crystal; specimen axis in [111] orientation; 2 cm long and average dia 0.44 cm; vacancy clusters $< 1 \mu$ in dia; electrical resistivity reported as $\sim 2000$ ohm cm at room temp; measured in helium atmosphere.
32	684	R	1964	418-1577	$\pm 5$			n-type single crystal; 2.6 cm dia $\times$ 13 cm long; axis of cylinder in [111] direction; produced by floating zone process in argon atmosphere; dislocation density of the order of $10^6 \text{ cm}^{-2}$ ; carrier concentration, $1.27 \times 10^{15} \text{ cm}^{-3}$ ; electrical resistivity reported as 440 ohm cm at room temp; measured in helium atmosphere; after the measurement, room temp resistivity dropped to 177 ohm cm, carrier concentration rose to $2.46 \times 10^{15} \text{ cm}^{-3}$ .
33	747, 746	C	1961	320-578		S-B-1		Single crystal; p-type; impurity concentration $2 \times 10^{15}$ atoms $\text{cm}^{-3}$ , supplied by Battelle Memorial Institute; ground to a dia of 11.8 mm and sliced to 7 mm thick; measured in a vacuum of $10^{-4}$ mm Hg; Armco iron (99.9% Fe) used as comparative material.
34	747, 746	C	1961	300-495		S-B-1		Second run of the above specimen.
35	747, 746	C	1961	404, 307		S-B-1		Third run of the above specimen.
36	747, 746	C	1961	302-467		S-B-2		Similar to above except impurity concentration $6 \times 10^{14}$ atoms $\text{cm}^{-3}$ .
37	747, 746	C	1961	336-594		S-B-2		Second run of the above specimen.
38	748	L	1964	10-320	$\pm 5$	R-3		Polycrystalline, p-type; major impurity boron, $5 \times 10^{16}$ atoms $\text{cm}^{-3}$ ; 1.24 cm effective dia, 3.2 cm long; electrical conductivity reported as $3.8 \times 10^3 \text{ ohm}^{-1} \text{ cm}$ at 300 K.
39	748	L	1964	2.1-300	$\pm 5$	R-5		Synthetic single crystal, p-type; major impurity boron, $3 \times 10^{16}$ atoms $\text{cm}^{-3}$ ; 0.56 cm effective dia, 2.6 cm long; electrical conductivity reported as $2.2 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 300 K.

SPECIFICATION TABLE NO. 51 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
40	748	L	1964	3.2-340	±5	R-6	Synthetic single crystal, n-type; major impurity phosphorus, $2.0 \times 10^{19}$ atoms $\text{cm}^{-3}$ ; 1.20 cm effective dia, 3.2 cm long; electrical conductivity reported as $3.6 \times 10^2$ ohm $^{-1}$ cm at 300 K.
41	748	L	1964	3.8-300	±5	R-55	Synthetic single crystal, n-type; major impurity phosphorus, $1.7 \times 10^{19}$ atoms $\text{cm}^{-3}$ ; 0.55 cm effective dia, 1.7 cm long; electrical conductivity reported as $1.3 \times 10^3$ ohm $^{-1}$ cm at 300 K.
42	749	L	1961	2.7-290	3	Si-1	High purity; 0.2 x 0.4 x 2 cm; supplied by Texas Instruments Inc.; electrical resistivity $3.4 \times 10^3$ ohm cm at <del>300 K</del> <b>0°C</b> .
43	749	L	1961	3.1-106	3	Si-2	Similar to the above specimen but with more impurities; electrical resistivity 18 ohm cm at <del>300 K</del> <b>0°C</b> .
44	749	L	1961	3.6-230	3	Si-3	Similar to the above specimen but with more impurities; electrical resistivity 7.2 ohm cm at <del>300 K</del> <b>0°C</b> .
45	749	L	1961	5.8-210	3	Si-4	Similar to the above specimen but with more impurities; electrical resistivity 0.57 ohm cm at max thermal conductivity.
46	586	C	1956	313		Si	p-type; electrical resistivity 2 to 3 ohm cm at 293 K; Firth Brown F. H. steel used as comparative material.
47	750	P	1961	311-1013	2	Si-142	Solid specimen; electrical resistivity 100 ohm cm at <del>300 K</del> <b>room temperature</b> ; measured in vacuo; thermal conductivity values calculated from measured data of thermal diffusivity using specific heat data taken from Amer. Inst. Physics Handbook (McGraw Hill Book Co., New York, p. 4-42, 1957).
48	751		1963	94-295			Virgin specimen.
49	751		1963	79-250			Similar to the above specimen except irradiated with $1.2 \times 10^{18}$ fast neutrons $\text{cm}^{-2}$ .
50	752	P	1962	310-1220		Si-1142	Pure; intrinsic; single crystal; thermal conductivity values calculated from measured data of thermal diffusivity using specific heat data taken from Amer. Inst. Physics Handbook (McGraw Hill Book Co., New York, 1957).
51	899	L	1965	9.3-399		1	Prepared from high-purity vacuum-floating-zone single crystal o-type (residual boron) material obtained from Merck and Co.; 0.152 cm wide x 0.046 cm thick; long dimension in the <111> direction; electrical resistivity 5000 ohm cm, carrier concentration $\sim 3 \times 10^{17}$ $\text{cm}^{-3}$ .
52	899	L	1965	47-59		1	The above specimen irradiated in <110> direction with a total time-integrated flux of $8.0 \times 10^{18}$ 2-Mev $\text{e}^{-}$ $\text{cm}^{-2}$ on a length of 1.0 cm; annealed at 60 K for 15 min.
53	899	L	1965	9.5-76		1	The above specimen annealed again for 15 min at 77 K.
54	899	L	1965	9.3-132		1	The above specimen annealed again for 15 min at 135 K.
55	899	L	1965	9.1-146		1	The above specimen annealed again for 15 min at 150 K.
56	899	L	1965	9.1-176		1	The above specimen annealed again for 15 min at 180 K.
57	899	L	1965	9.8-227		1	The above specimen annealed again for 15 min at 230 K.

SPECIFICATION TABLE NO. 51 (continued)

Curve No.	Expt. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)	Specifications and Remarks
58	899	L	1965	9.7-172		1		The above specimen annealed again for 15 min at 290 K.
59	899	L	1965	9.4-309		1		The above specimen annealed again for 15 min at 410 K.
60	899	L	1965	8.6-303		2		Prepared from high-purity vacuum-floating-zone single crystal p-type (residual boron) material obtained from Merck and Co.; 0.153 cm wide x 0.048 cm thick; long dimension was the 111 direction; electrical resistivity 5000 ohm cm; carrier concentration $\sim 3 \times 10^{16} \text{ cm}^{-3}$ .
61	899	L	1965	47-59		2		The above specimen irradiated in 110 direction with a total time-integrated flux of $8.0 \times 10^{16} \text{ 2-Mev } n \text{ cm}^{-2}$ on a length of 1.0 cm; annealed at 60 K for 15 min.
62	899	L	1965	8.9-76		2		The above specimen annealed again for 15 min at 77 K.
63	899	L	1965	8.9-132		2		The above specimen annealed again for 15 min at 135 K.
64	899	L	1965	8.8-146		2		The above specimen annealed again for 15 min at 150 K.
65	899	L	1965	8.7-176		2		The above specimen annealed again for 15 min at 180 K.
66	899	L	1965	8.4-226		2		The above specimen annealed again for 15 min at 230 K.
67	899	L	1965	8.5-81		2		The above specimen annealed again for 15 min at 280 K.
68	899	L	1965	8.9-301		2		The above specimen annealed again for 15 min at 410 K.
69	339, 883	L	1965	5.4-285	10 in 4-20K and 100-100K 15 in 20-100K	2		The above specimen irradiated in 110 direction with a total time-integrated flux of $1.1 \times 10^{16} \text{ n cm}^{-2}$ .
70	339, 883, 949	L	1965	5.5-288	Same as above	2		Similar to the above specimen; irradiated at 30 C with a fast-neutron integrated flux $2.5 \times 10^{16} \text{ n cm}^{-2}$ .
71	339, 883	L	1965	4.9-277	Same as above	2		Similar to the above specimen; irradiated at 30 C with a fast-neutron integrated flux $1.7 \times 10^{16} \text{ n cm}^{-2}$ .
72	883	L	1965	5.5-173	Same as above	2		Similar to the above specimen; irradiated at 30 C with a fast-neutron integrated flux $3.4 \times 10^{16} \text{ n cm}^{-2}$ .
73	883	L	1965	5.2-286	Same as above	2		n-type single crystal; carrier concentration $N_D = 5 \times 10^{15} \text{ cm}^{-3}$ ; high purity silicon used as comparative material.
74	335	C	1965	567-1072		S-1		n-type single crystal; $N_D = 5 \times 10^{15} \text{ cm}^{-3}$ ; Arco iron used as comparative material.
75	335	C	1965	596-1073		S-2		n-type single crystal; supplied by Knapp Electro-Physics; 23 mm dia x 8 mm thick; circular cross-section perpendicular to 111 direction; $N_D \sim 5 \times 10^{15} \text{ cm}^{-3}$ ; Arco iron used as comparative material.
76	335	C	1965	98		2		2nd run of the above specimen.
77	335	C	1965	179-256		2		3rd run of the above specimen.
78	335	C	1965	143, 217		2		

SPECIFICATION TABLE NO. 51 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
79	901	C	1965	98-255	10		n-type single crystal; supplied by Knapp Electro-Physics; 23 mm dia x 8 mm thick; circular cross-section perpendicular to $111$ direction; $N_D \sim 5 \times 10^{18} \text{ cm}^{-3}$ ; Arcco iron used as comparative material.
80	902	C	1967	588-1276	20-25	Si-E4	0.1 P-doped; n-type single crystal; measured in a vacuum of $5 \times 10^{-6}$ to $10^{-4}$ torr; Arcco iron used as comparative material.



DATA TABLE NO. 51 (continued)

CURVE 20 (cont.)			CURVE 21			CURVE 22 (cont.)			CURVE 23 (cont.)			CURVE 24 (cont.)			CURVE 25 (cont.)			CURVE 26 (cont.)			CURVE 27 (cont.)			CURVE 28			CURVE 29 (cont.)																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																													
T	k		T	k		T	k		T	k		T	k		T	k		T	k		T	k		T	k		T	k		T	k		T	k																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																						
24.0	26.5		6.30	8.00		16.9	42.3		45.0	30.5		66.3	11.1		55.0	26.5		116.0	7.50		1115	0.30		1030	0.30 <sup>o</sup>	30.0	16.9		47.5	28.0		71.0	11.0		59.6	24.0		119.0	7.00		1050	0.30	350	6.10		375	1.08		375	1.08		400	0.97		1220	0.29	420	7.00		445	0.94		445	0.94		470	0.76		1270	0.28	495	8.70		520	0.82		520	0.82		545	0.64		1340	0.27 <sup>o</sup>	565	10.5		565	0.63		565	0.63		590	0.59		1370	0.28	620	13.2		620	0.55		620	0.55		670	0.50		1420	0.28	695	15.9		695	0.53		695	0.53		720	0.47		1470	0.28	745	18.0		745	0.50		745	0.50		770	0.45		1520	0.28	795	21.0		795	0.43		795	0.43		825	0.41		1570	0.28	850	24.5		850	0.40		850	0.40		880	0.36		1620	0.28	900	28.0		900	0.36		900	0.36		930	0.34		1670	0.28	975	33.0		975	0.33		975	0.33		1010	0.31		1720	0.28	995	35.0		995	0.31		995	0.31		1040	0.31		1770	0.28	1010	38.5		1010	0.31		1010	0.31		1090	0.31		1820	0.28																																																																																																																																																																																																																																																																																																																																																																																																																																																								
30.0	23.5		7.50	10.0		17.7	44.8		47.5	28.0		71.0	11.0		59.6	24.0		119.0	7.00		1160	0.28	37.0	18.0		9.20	14.8		19.5	45.0		56.5	23.0		81.5	9.50		75.0	17.0		1070	0.29	43.0	15.0		10.5	18.0		21.0	46.0		60.0	21.0		85.0	14.0		80.0	15.4		1090	0.29 <sup>o</sup>	49.0	12.0		11.0	21.5		24.0	49.3		65.0	18.0		87.5	8.80		85.0	14.0		1115	0.29	55.0	9.0		11.4	25.0		26.0	50.0		70.0	17.0		90.0	8.50 <sup>o</sup>		280.0	2.00		1140	0.27	61.0	7.0		12.0	28.0		27.0	49.0		75.0	15.2		97.0	7.50		290.0	1.64		1200	0.27	67.0	5.0		13.0	31.0		28.0	49.9		80.0	12.1		100.0	7.25		299.5	1.50				73.0	3.0		13.7	34.0		30.8	47.5		85.0	12.1		107.0	6.50								79.0	1.0		14.5	37.0		33.0	45.0		90.0	10.0		113.0	5.65									85.0	0.5		15.1	40.0		34.5	42.0		95.0	8.50		120.0	5.00										91.0	0.2		16.2	43.0		38.5	38.5		100.0	7.40		127.0	4.42											97.0	0.1		17.0	46.0		40.5	35.0		105.0	7.90		134.0	3.88											103.0	0.0		18.0	49.0		42.0	32.0		110.0	7.40		141.0	3.42											109.0	0.0		19.5	52.0		45.0	24.0		113.0	7.00		148.0	3.04											115.0	0.0		21.0	55.0		48.0	20.0		120.0	6.25		155.0	2.75											121.0	0.0		22.0	58.0		50.0	14.0		125.0	5.50		162.0	2.50											127.0	0.0		23.0	61.0		52.0	10.0		130.0	4.80		169.0	2.25											133.0	0.0		24.0	64.0		54.0	7.0		135.0	4.20		176.0	2.05											139.0	0.0		25.0	67.0		56.0	5.0		140.0	3.70		183.0	1.90											145.0	0.0		26.0	70.0		58.0	3.0		145.0	3.20		190.0	1.80											151.0	0.0		27.0	73.0		60.0	2.0		150.0	2.80		197.0	1.70											157.0	0.0		28.0	76.0		62.0	1.5		155.0	2.50		204.0	1.64											163.0	0.0		29.0	79.0		64.0	1.0		160.0	2.25		211.0	1.58											169.0	0.0		30.0	82.0		66.0	0.8		165.0	2.05		218.0	1.52											175.0	0.0		31.0	85.0		68.0	0.6		170.0	1.90		225.0	1.47											181.0	0.0		32.0	88.0		70.0	0.5		175.0	1.75		232.0	1.42											187.0	0.0		33.0	91.0		72.0	0.4		180.0	1.65		239.0	1.38											193.0	0.0		34.0	94.0		74.0	0.3		185.0	1.58		246.0	1.34											199.0	0.0		35.0	97.0		76.0	0.2		190.0	1.52		253.0	1.30										

Not shown in table



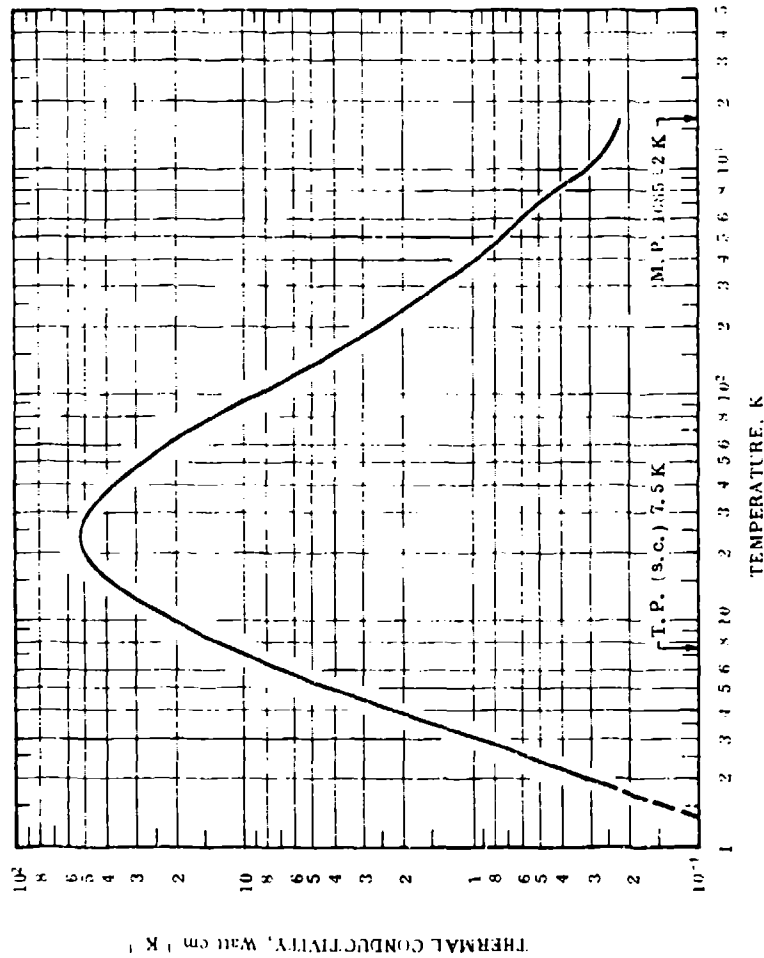








FIGURE AND TABLE NO. 51R RECOMMENDED THERMAL CONDUCTIVITY OF SILICON



RECOMMENDED VALUES\*

T <sub>1</sub>	k <sub>1</sub>	k <sub>2</sub>	T <sub>2</sub>	T <sub>1</sub>	k <sub>1</sub>	k <sub>2</sub>	T <sub>2</sub>
0	0	0	-459.7	500	0.762	44.0	440.3
1	(0.0448)†	(2.59)	-457.9	600	0.619	35.8	620.3
2	0.317	18.2	-456.1	700	0.508	29.4	800.3
3	0.998	57.7	-454.3	800	0.422	24.4	980.3
4	2.26	131	-452.5	900	0.359	20.7	1160
5	4.24	245	-450.7	1000	0.312	18.0	1340
6	6.86	396	-448.9	1100	0.279	16.1	1520
7	9.91	573	-447.1	1200	0.257	14.8	1700
8	13.4	774	-445.3	1300	0.244	14.1	1880
9	17.2	994	-443.5	1400	0.235	13.6	2060
10	21.1	1220	-441.7	1500	0.227	13.1	2240
11	24.8	1430	-439.9	1600	0.221	12.8	2420
12	28.7	1660	-438.1	1685	0.220	12.7	2573
13	32.5	1880	-436.3				
14	36.0	2080	-434.5				
15	39.3	2270	-432.7				
16	42.2	2440	-430.9				
18	46.7	2700	-427.3				
20	49.4	2850	-423.7				
25	51.4	2970	-414.7				
30	48.1	2780	-405.7				
35	41.3	2390	-396.7				
40	35.3	2040	-387.7				
45	30.6	1770	-378.7				
50	26.8	1550	-369.7				
60	21.1	1220	-351.7				
70	16.8	971	-333.7				
80	13.4	774	-315.7				
90	10.8	624	-297.7				
100	8.84	511	-279.7				
150	4.09	236	-189.7				
200	2.64	153	-99.7				
250	1.77	110	-				
273.2	1.68	97.1	32.0				
300	1.48	85.5	80.3				
350	1.19	68.8	170.3				
400	0.989	57.1	260.3				

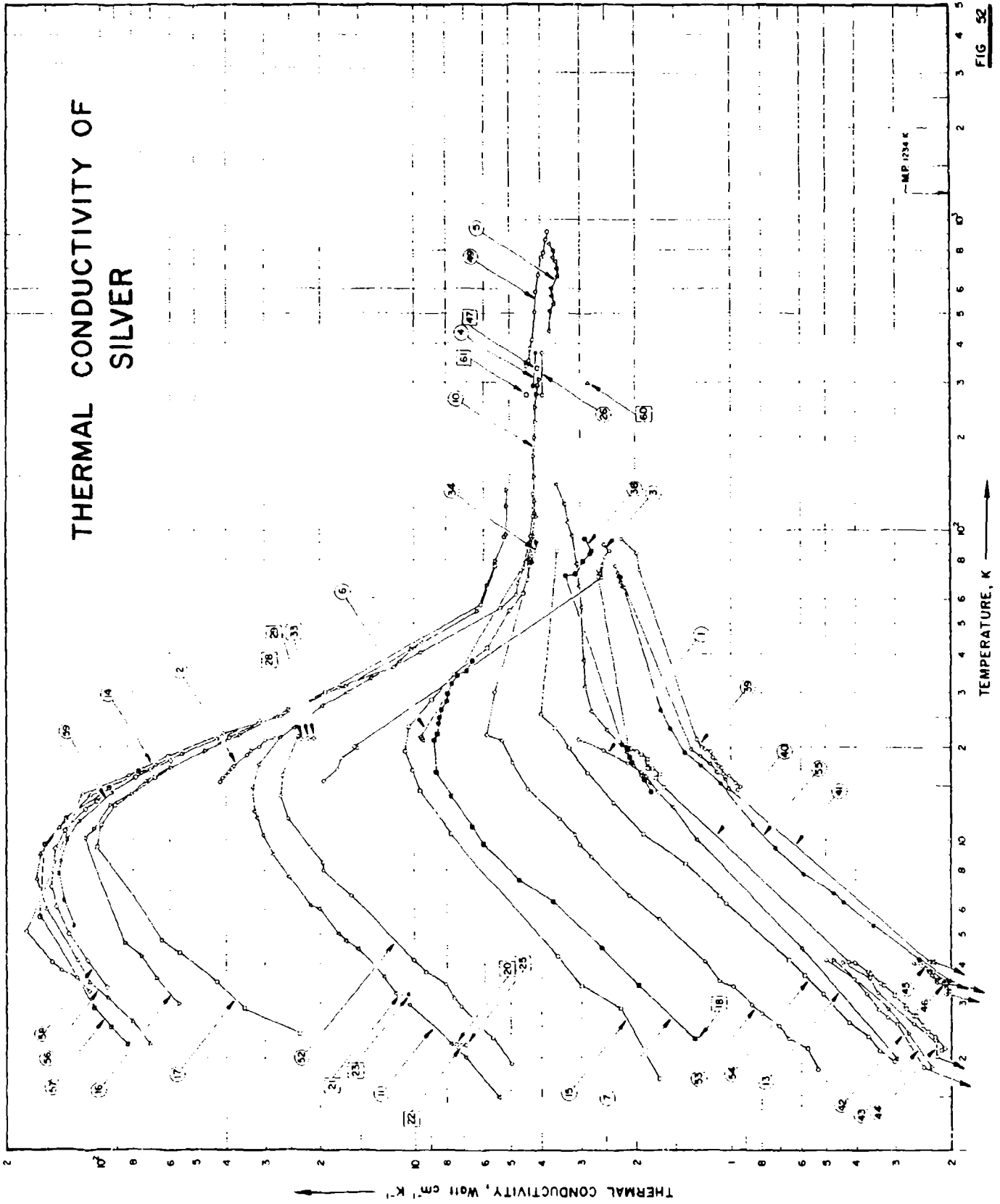
REMARKS

The recommended values are for high-purity silicon. The recommended values that are supported by experimental thermal conductivity data are thought to be accurate to within 4% of the true values at room and moderate temperatures and 4 to 10% at other temperatures above 50 K. The thermal conductivity near and below the corresponding temperature of its maximum is highly sensitive to small physical and chemical variations of the specimens, and the values below 50 K are intended as typical values for indicating the general trend.

\* T<sub>1</sub> in K, k<sub>1</sub> in Watt cm<sup>-1</sup> K<sup>-1</sup>m, T<sub>2</sub> in F, and k<sub>2</sub> in Btu hr<sup>-1</sup> ft<sup>-1</sup> F<sup>-1</sup>. † Values in parentheses are extrapolated.

# THERMAL CONDUCTIVITY OF SILVER

FIGURE SHOWS ONLY 46 OF THE CURVES REPORTED IN TABLE



TEMPERATURE, K  $\longrightarrow$

FIG. 52

## SPECIFICATION TABLE NO. 52 THERMAL CONDUCTIVITY OF SILVER

(Impurity &lt; 0.20% each; total impurities &lt; 0.50%)

[For Data Reported in Figure and Table No. 52]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)	Specifications and Remarks
1	51	L	1956	14-21		Ag 2	Commercially pure; supplied by Nordiska Affineriet, Halsingborg; cold-worked; electrical resistivity ratio $\rho(273K)/\rho_0 = 38.4$ .	
2	51	L	1956	15-21		Ag 2t	The above specimen etched and annealed at 740 K; electrical resistivity reported as 0.00447, 0.00519, 0.00619, 0.00743, 0.238, 0.297, 0.359, and 1.47 $\mu\text{ohm cm}$ at 14, 16, 18, 20, 70, 80, 90, and 273 K, respectively; electrical resistivity ratio $\rho(273K)/\rho_0 = 417$ .	
3	51	L	1956	16-90		Ag 4t	Similar to the above specimen but annealed at 750 K; electrical resistivity reported as 0.00694, 0.00758, 0.00852, 0.00983, 0.235, 0.299, 0.363, and 1.48 $\mu\text{ohm cm}$ at 14, 16, 18, 20, 70, 80, 90, and 273 K, respectively; electrical resistivity ratio $\rho(273K)/\rho_0 = 250$ .	
4	77	E	1900	291, 373			99.98 pure; 1.1086 cm dia x 25.2 cm long; density 10.53 $\text{g cm}^{-3}$ at 18 C; electrical conductivity reported as 61.4 and 46.9 $\times 10^4 \text{ohm}^{-1} \text{cm}^{-1}$ at 18 and 100 C, respectively.	
5	6	L	1931	437-838			99.9 pure; 0.585 cm dia x 7-8 cm long; melting point 961 C.	
6	58	L	1934	21-91		Ag 1	Pure; cold-worked and annealed at 350 C for 2 hrs.	
7	97, 122	L	1952	2.3-38	2-3	JM 1722, Ag 1	99.99 pure; polycrystalline wire; 1.22 mm dia x 2.85 cm long; supplied by Johnson Matthey; $\rho(273K)/\rho(20K) = 30.9$ .	
8	78	E	1931	283-291	2	Ag 1	Commercially pure electrolytic silver; 0.05286 cm dia x 8.82 cm long; electrical conductivity 64.6 $\times 10^4 \text{ohm}^{-1} \text{cm}^{-1}$ at 273 K; Lorenz function 2.32 $\times 10^{-8} \text{V}^2 \text{K}^{-2}$ at 273 K.	
9	78	E	1931	278-284	2	Ag II	Spectroscopically pure; 0.05059 cm dia x 8.74 cm long; electrical conductivity 61.2 $\times 10^4 \text{ohm}^{-1} \text{cm}^{-1}$ at 273 K; Lorenz function 2.41 $\times 10^{-8} \text{V}^2 \text{K}^{-2}$ at 273 K.	
10	88	L	1908	110-306			99.9 pure; 0.585 cm dia x 7-8 cm long; density 10.47 $\text{g cm}^{-3}$ at 21 C; electrical resistivity reported as 0.460, 0.470, 0.456, 0.609, 0.660, 0.693, 0.890, 0.923, 0.942, 1.236, 1.239, 1.245, 1.468, 1.471, 1.675, and 1.684 $\mu\text{ohm cm}$ at -178.3, -177.7, -176.1, -151.2, -144.9, -139.2, -109.0, -102.5, -98.6, -54.9, -52.8, -50.6, -13.0, -12.0, -21.0, and <del>21.5</del> 21.5 C, respectively.	
11	122	L	1955	1.5-44	3	JM 3351, Ag 2	99.99 <sup>+</sup> pure; polycrystalline; 1.33 mm dia x 2.8 cm long; supplied by Johnson Matthey; prepared from a 5 mm rod by rolling and drawing; annealed in vacuo at 750 C for several hrs.	
12	79	E	1933	90-373			Traces of Bi, Cd, Cu, Pb, Mg, Si, and Na; 0.06095 cm dia x 9.770 cm long; drawn from a rod of H.S. brand silver supplied by A. Hülgner, Ltd.; annealed at 500 C; electrical resistivity reported as 0.341, 1.035, 1.510, 2.123, and 2.863 $\mu\text{ohm cm}$ at -183.00, -78.50, 0.100, and 217.96 C, respectively; measured in a vacuum of $10^{-4}$ mm Hg.	
13	147	L	1953	2.0-140	1-3	JM 4606, Ag 1	99.999 <sup>+</sup> pure; polycrystalline; 2 mm dia rod supplied by Johnson Matthey.	
14	147	L	1953	3.3-131	1-3	JM 4606, Ag 2	The above specimen annealed at 650 C; grain size ~0.1 mm.	

SPECIFICATION TABLE NO. 52 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
15	147	L	1953	1.7-134	1-3	JM 4606; Ag 3	1.16 mm dia rod drawn from the above specimen.
16	147	L	1953	3.0-135	1-3	JM 4606; Ag 4	The above specimen annealed at 650 C.
17	147	L	1953	2.4-95	1-3	JM 4606; Ag 5	The above specimen, Ag 4, after being removed and replaced in cryostat.
18	342	L	1953	2.3		Ag 1	99.99 pure; polycrystal; annealed; measured in a transverse field of 1.09 kilooersteds.
19	342	L	1953	2.2		Ag 2	99.999 pure; polycrystal; annealed; measured in a transverse field of 1.09 kilooersteds.
20	342	L	1953	2.2		Ag 2	The above specimen measured in a transverse field of 1.75 kilooersteds.
21	342	L	1953	3.2		Ag 2	The above specimen measured in a transverse field of 1.97 kilooersteds.
22	342	L	1953	2.2		Ag 2	The above specimen measured in a transverse field of 2.7 kilooersteds.
23	342	L	1953	3.2		Ag 2	The above specimen measured in a transverse field of 3.6 kilooersteds.
24	342	L	1953	2.2		Ag 2	The above specimen measured in a longitudinal field of 3.6 kilooersteds.
25	342	L	1953	2.2		Ag 2	The above specimen measured in a transverse field of 3.7 kilooersteds.
26	246	T	1919	273.373			1 mm dia wire; rolled and drawn; heated 0.5 hr at temp close to melting point; electrical conductivity reported as 57.0 and $41.3 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 0 and 100 C, respectively.
27	436	L	1938	21.18		Ag 37	Single crystal.
28	436	L	1938	21.24		Ag 37	The above specimen measured at H (the transverse magnetic field strength) = 8810 oersteds and at $\theta$ (the angle of rotation of the magnetic field in a plane perpendicular to the specimen axis) = $1^\circ$ at which the dependence of $k$ on H is maximum.
29	436	L	1938	21.26		Ag 37	The preceding specimen measured at H = 10850 oersteds and at $\theta = +1^\circ$ .
30	436	L	1938	21.18		Ag 37	The above specimen measured without magnetic field.
31	436	L	1938	21.20		Ag 37	The above specimen measured at H = 4580 oersteds and at $\theta = +45^\circ$ at which the dependence of $k$ on H is minimum.
32	436	L	1938	21.26		Ag 37	The above specimen measured at H = 8810 oersteds and at $\theta = +45^\circ$ .
33	436	L	1938	21.27		Ag 37	The above specimen measured at H = 10850 oersteds and at $\theta = +45^\circ$ .
34	58	L	1934	79.91		Ag e4	Pure; single crystal; deformed; electrical resistivity 1.50 $\mu\text{ohm cm}$ at 0 C.
35	58	L	1934	90		Ag e4	The above specimen annealed for 2 hrs at 350 C; electrical resistivity 1.49 $\mu\text{ohm cm}$ at 0 C.
36	58		1934	80.91		Ag e5	Pure; single crystal.
37	504	P	1961	295.2	±5		Pure; 1.9 x 1.9 x 0.322 cm; thermal conductivity value calculated from measured datum of thermal diffusivity using specific heat and density values taken from Smithsonian Physical Tables (9th ed., 1954).

SPECIFICATION TABLE NO. 52 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
38	51	L	1955	14-94		Ag-Mn 3	0.14 at. % Mn; polycrystal; rectangular rod specimen of square cross section 2.5 x 2.5 mm; annealed at 720 K for several hrs in vacuo; electrical resistivity reported as 0.233, 0.214, 0.235, 0.237, 0.473, 0.534, 0.595, and 1.69 $\mu\text{ohm cm}$ at 14, 16, 19, 20, 70, 80, 90, and 273 K, respectively; electrical resistivity ratio $\rho(273\text{K})/\rho_0 = 7.26$ .
39	51	L	1956	15-94		Ag-Mn 2	0.32 at. % Mn; polycrystal; rectangular rod specimen of square cross section 2.5 x 2.5 mm; annealed at 720 K for several hrs in vacuo; electrical resistivity reported as 0.523, 0.521, 0.523, 0.526, 0.772, 0.830, 0.881, and 1.98 $\mu\text{ohm cm}$ at 14, 16, 18, 20, 70, 80, 90, and 273 K, respectively; electrical resistivity ratio $\rho(273\text{K})/\rho_0 = 3.77$ .
40	22	L	1959	1.6-74			0.14 at. % Mn; polycrystal; prepared from pure silver and from manganese of 99.995% pure; melted, rolled, and cut into rods of cross sectional area $\sim 2.5 \text{ mm}^2$ ; residual electrical resistivity 0.27 $\mu\text{ohm cm}$ .
41	22	L	1959	1.5-76			0.32 at. % Mn; polycrystal with fine grains; same fabrication method as above; residual electrical resistivity 0.54 $\mu\text{ohm cm}$ .
42	643	L	1956	1.5-4.1			0.14 at. % Mn; measured in a magnetic field of 25.5 kilooersteds.
43	649	L	1956	1.9-4.1			0.14 at. % Mn; measured in a magnetic field of 19 kilooersteds.
44	649	L	1956	1.4-4.0			0.14 at. % Mn; measured in a magnetic field of 12 kilooersteds.
45	649	L	1956	3.0-4.0			0.32 at. % Mn; measured in a magnetic field of 19 kilooersteds.
46	649	L	1956	1.5-4.0			0.32 at. % Mn; measured in a magnetic field of 25.5 kilooersteds.
47	230	L	1925	333			99.9 pure; electrical conductivity $58.8 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 25 C.
48	241	E	1911	298			Impurities < 0.3; electrical conductivity $57.35 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 25 C.
49	617	E	1957	338-917			99.99 pure; polycrystal; electrical resistivity reported as 1.89, 2.21, 2.97, 4.62, and 5.91 $\mu\text{ohm cm}$ at 338, 2, 503, 2, 753, 2, and 917.2 K, respectively; Lorenz function reported as 2.45, 2.46, 2.45, 2.42, and $2.44 \times 10^{-3} \text{ V}^2 \text{ K}^{-2}$ at the above temps, respectively.
50	650	E	1960	306			99.99 pure.
51	736	L	1965	0.43-0.93			Pure silver wire.
52	382	L	1966	1.9-22	2.0	63 grade	0.005 at. % Mn; prepared from 99.9999 pure silver supplied by Cominco and 99.95 pure manganese supplied by Johnson Matthey and Mallory; melted in argon; chill cast, rolled to 1 mm thick, rectangular wire cut from the ingot; annealed at 750 C for 4 hrs in a vacuum of $< 2 \times 10^{-4}$ torr.
53	382	L	1966	1.8-82	2.0	59 grade	0.067 at. % Mn; same sources of materials and fabrication method as above.
54	382	L	1966	2.0-85	2.0	59 grade	0.11 at. % Mn; same sources of materials and fabrication method as above.
55	382	L	1966	1.8-86	2.0	59 grade	0.31 at. % Mn; same sources of materials and fabrication method as above.



SPECIFICATION TABLE NO. 52 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
56	570	L	1963	2, 2-10	1.5	1	Pure; specimen 3 mm in dia and about 6 cm long; supplied by Engelhard Industries in Toronto; rolled and drawn, etched with nitric acid; annealed at 450 C in vacuo for 3 hrs; residual electrical resistivity 799 $\mu\Omega$ cm.
57	570	L	1963	2, 2-17	1.5	2	Similar to the above specimen except residual electrical resistivity 670 $\mu\Omega$ cm.
58	903	L	1966	3, 4-29		HFN735, 1	99.9999 Ag, 0.00001 Fe, 0.00001 Si, 0.00001 Ca, and 0.00001 Mg; fine grain polycrystalline; obtained from the Consolidated Mining and Smelting Co.; remelted and outgassed; annealed at 450 C for 24 hrs; residual electrical resistivity 0.00081 $\mu\Omega$ cm.
59	903	L	1966	5, 3-14		HFN735, 2	Similar to the above specimen except annealed at 530 C for 24 hrs and residual electrical resistivity 0.00088 $\mu\Omega$ cm.
60	904	C	1962	298, 2			0.05 In, cast; copper used as comparative material.
61	1005	E	1927	273, 2			99.9 pure; 0.125 in. dia x 10 cm long; obtained from Baker and Co.; electrical resistivity 1.491 $\mu\Omega$ cm at 0 C.
62	79	F	1945	90-373			Traces of Bi, Cd, Cu, Pb, Mg, Si, and Na; 0.06095 cm dia x 9.770 cm long; drawn from a rod of H. S. brand silver supplied by A. Hilger Ltd.; electrical resistivity reported as 0.377, 1.036, 1.509, and 2.121 $\mu\Omega$ cm at -183.00, -78.50, 0, and 100 C, respectively; measured in a vacuum of $10^{-4}$ mm Hg.

DATA TABLE NO. 53 THERMAL CONDUCTIVITY OF SILVER  
(Impurity: 0.20% each; total impurities < 0.50%)

[Temperature, T, K; Thermal Conductivity, k, Watts cm <sup>-1</sup> K <sup>-1</sup> ]	
T	k
<u>CURVE 1</u>	
14.4	1.710
16.0	1.860
16.6	1.920
17.7	2.130
19.5	2.540
21.3	3.050
<u>CURVE 2</u>	
15.4	42.00
15.7	41.20
16.1	40.30
16.6	39.40
17.0	38.40
17.3	37.70
18.7	34.20
19.4	33.20
20.0	31.10
21.0	30.00
21.4	23.10
<u>CURVE 3</u>	
15.5	19.50
17.0	17.80
18.5	17.10
19.9	15.30
20.0	15.60
20.2	2.57
20.8	2.56
81.4	2.50
85.2	2.42
89.5	2.54
<u>CURVE 4</u>	
291.2	4.210
373.2	4.150
<u>CURVE 5</u>	
437.2	3.749
508.2	3.711
538.2	3.615
<u>CURVE 6</u>	
21.0	9.46
21.4	9.48
21.6	9.52
21.7	4.44
21.8	4.38
21.9	4.30
<u>CURVE 7</u>	
2.30	1.311
3.40	1.967
4.50	2.568
6.33	3.661
7.43	4.700
9.72	6.011
11.20	6.557
13.92	7.760
16.71	8.634
20.97	8.743
22.11	8.525
23.81	8.470
24.79	8.415
26.32	8.306
28.12	7.980
29.70	7.923
31.67	7.650
33.74	7.380
35.22	6.830
37.90	6.557
<u>CURVE 8*</u>	
293.2	4.105
<u>CURVE 8 (cont.)</u>	
570.2	4.100
290.4	4.105
<u>CURVE 9*</u>	
27.2	4.033
279.0	4.029
281.9	4.038
284.0	4.038
<u>CURVE 10</u>	
110.2	4.150
113.2	4.176
123.2	4.185
148.2	4.205
173.2	4.218
198.2	4.200
223.2	4.172
248.2	4.170
273.2	4.105
291.2	4.075
306.2	4.020
<u>CURVE 11</u>	
1.47	5.45
2.00	6.90
2.21	7.79
2.95	10.45
3.63	12.55
4.47	15.34
5.00	17.45
6.00	20.10
6.16	21.50
7.58	25.36
8.95	28.20
10.43	30.35
11.79	31.66
12.42	32.35
14.74	32.75
17.63	31.25
<u>CURVE 11 (cont.)</u>	
21.40	26.95
23.10	24.26
27.10	19.65
30.00	16.35
33.16	13.65
37.58	11.16
45.58	9.45
<u>CURVE 12*</u>	
90.2	4.259
194.7	4.205
273.2	4.180
373.2	4.176
<u>CURVE 13</u>	
1.98	0.298
4.47	0.596
10.00	1.277
12.75	1.532
16.17	1.915
19.15	2.130
22.55	2.468
25.95	2.765
31.95	2.895
38.00	2.935
56.60	2.980
65.95	3.010
77.87	3.060
95.75	3.190
106.60	3.275
120.60	3.360
139.60	3.575
<u>CURVE 14</u>	
3.27	104.50
3.62	120.09
3.93	134.50
4.04	145.50
5.10	174.00
14.05	115.00
16.15	82.10
<u>CURVE 14 (cont.)</u>	
19.70	60.85
22.13	40.85
25.32	26.80
31.06	17.50
36.20	11.70
40.45	9.560
55.97	5.380
61.90	4.800
67.25	4.725
75.55	4.530
95.75	4.320
110.85	4.255
130.60	4.255
<u>CURVE 15</u>	
1.70	1.702
2.87	2.235
3.40	2.980
4.25	3.550
10.43	7.745
14.50	9.615
16.80	10.130
19.35	10.760
23.20	10.380
28.50	8.830
34.05	7.553
41.70	5.870
54.90	5.020
62.55	4.550
76.60	4.340
95.50	4.300
115.50	4.210*
134.05	4.210*
<u>CURVE 16</u>	
2.98	56.38
3.60	66.80
4.25	74.90
4.58	84.70
10.20	111.50
10.95	105.30
12.98	93.60
<u>CURVE 16 (cont.)</u>	
14.05	80.85
15.53	72.35
17.23	59.58
19.35	46.80
21.70	36.17
26.38	25.53
31.50	16.60
41.05	10.12
54.68	6.380
66.17	5.900
79.15	5.530
96.60	5.105
118.95	5.105
135.10	5.105
<u>CURVE 17</u>	
2.38	23.4
2.88	34.9
3.50	43.0
4.38	55.8
4.75	63.9
9.56	102.5
11.3	99.6
12.9	90.2
15.9	67.6
21.4	39.1
29.8	19.1
29.8	19.1
57.5	6.15
76.6	5.53
94.8	5.16
<u>CURVE 18</u>	
2.3	1.29
<u>CURVE 19*</u>	
2.2	7.75
<u>CURVE 20</u>	
2.2	7.46
<u>CURVE 21</u>	
3.2	11.14
<u>CURVE 22</u>	
2.2	7.14
<u>CURVE 23</u>	
3.2	10.50
<u>CURVE 24*</u>	
2.2	7.41
<u>CURVE 25</u>	
2.2	6.90
<u>CURVE 26</u>	
273.2	3.96
373.2	3.95
<u>CURVE 27*</u>	
21.19	31.4
<u>CURVE 28</u>	
21.24	23.3
<u>CURVE 29</u>	
21.26	22.0
<u>CURVE 30*</u>	
21.18	31.4
<u>CURVE 31*</u>	
21.20	27.3

\* Not shown on plot

DATA TABLE NO. 52 (continued)

T	k	T	k	T	k	T	k	T	k	T	k	T	k
<u>CURVE 32*</u>		<u>CURVE 39</u>		<u>CURVE 40 (cont.)</u>		<u>CURVE 41 (cont.)</u>		<u>CURVE 45</u>		<u>CURVE 48*</u>		<u>CURVE 51 (cont.)</u>	
21.26	23.0	14.6	0.954	4.05	0.422	14.84	1.025	3.00	0.178 <sup>c</sup>	293	4.68	0.890	0.120
<u>CURVE 33</u>		14.9	0.938	14.87	1.648	15.70	1.046	3.20	0.190 <sup>c</sup>	<u>CURVE 45</u>		0.933	0.129
<u>CURVE 34</u>		15.4	0.970	15.97	1.704	16.67	1.112	3.32	0.194 <sup>c</sup>	<u>CURVE 48*</u>		<u>CURVE 51</u>	
21.27	21.1	16.3	1.020	16.04	1.688	17.74	1.173	3.36	0.200 <sup>c</sup>	<u>CURVE 45</u>		<u>CURVE 51</u>	
<u>CURVE 35*</u>		16.4	1.070	17.07	1.847	18.77	1.231	3.49	0.207	338.2	4.41	1.91	4.97
78.6	4.27	17.2	1.075	17.10	1.765	19.70	1.343	3.52	0.207	352.2	4.36	2.31	5.70
90.8	4.12	17.4	1.100	17.96	1.842	65.33	2.190	3.57	0.214	385.2	4.31	2.93	7.18
<u>CURVE 36*</u>		18.0	1.120	18.08	1.845	65.35	2.192	3.67	0.220	412.2	4.25*	3.10	7.55
80.0	4.54	18.5	1.150	18.90	1.922	67.96	2.227	3.78	0.229	435.2	4.23	3.44	8.07
<u>CURVE 37*</u>		19.4	1.160	19.03	1.981	70.86	2.262	3.78	0.229	503.2	4.18	3.78	8.07
79.8	4.25	19.7	1.220	19.77	2.121	73.34	2.274	4.02	0.250	587.2	4.12	3.78	9.27
91.1	4.13	2.	1.240	19.86	2.082	76.24	2.333	<u>CURVE 46</u>		667.2	4.05	4.13	10.1
<u>CURVE 38*</u>		2.	1.280	19.86	2.082	76.24	2.333	1.51	0.095 <sup>c</sup>	753.2	3.96	6.65	16.0
295.2	3.975	74.0	1.950	19.97	2.148	<u>CURVE 42</u>		1.88	0.124 <sup>c</sup>	786.2	3.91	7.96	19.5
<u>CURVE 39</u>		84.0	1.980	70.32	2.616	1.52	0.197 <sup>c</sup>	2.01	0.136 <sup>c</sup>	853.2	3.85	8.47	19.6
14.40	1.800	93.5	2.240	73.84	2.618	1.87	0.245	2.19	0.150 <sup>c</sup>	917.2	3.80	9.48	21.6
<u>CURVE 40</u>		<u>CURVE 41</u>		<u>CURVE 42</u>		2.21	0.272	2.41	0.161 <sup>c</sup>	<u>CURVE 48*</u>		11.7	25.5
14.40	1.800	1.56	0.152 <sup>c</sup>	1.51	0.0755 <sup>c</sup>	2.56	0.259	2.60	0.164 <sup>c</sup>	<u>CURVE 45</u>		13.9	26.8
14.80	1.790*	1.84	0.178 <sup>c</sup>	1.56	0.0805 <sup>c</sup>	2.70	0.320	2.81	0.172 <sup>c</sup>	<u>CURVE 48*</u>		16.3	26.5
15.70	1.870	1.90	0.190 <sup>c</sup>	1.78	0.0943 <sup>c</sup>	2.85	0.340	3.00	0.187 <sup>c</sup>	<u>CURVE 45</u>		21.6	23.6
16.90	1.900	1.92	0.189 <sup>c</sup>	1.93	0.104 <sup>c</sup>	3.20	0.372	3.01	0.183 <sup>c</sup>	<u>CURVE 50*</u>		<u>CURVE 53</u>	
17.90	2.050	2.11	0.215	2.12	0.119 <sup>c</sup>	3.59	0.409	3.20	0.197 <sup>c</sup>	<u>CURVE 51*</u>		1.82	0.532
18.50	2.100*	2.14	0.210	2.33	0.132 <sup>c</sup>	4.11	0.500	3.20	0.209	0.425	0.655	1.82	0.579
19.70	2.170*	2.34	0.225	2.72	0.138 <sup>c</sup>	<u>CURVE 43</u>		3.29	0.201	0.450	0.660	2.14	0.661
20.40	2.220	2.42	0.238	2.85	0.149 <sup>c</sup>	1.86	0.231	3.37	0.206 <sup>c</sup>	0.463	0.6613	2.29	0.708
21.00	2.350	2.44	0.244	2.91	0.148 <sup>c</sup>	2.40	0.274	3.39	0.201 <sup>c</sup>	0.510	0.6675	2.52	0.800
71.00	3.320	2.54	0.246	3.03	0.162 <sup>c</sup>	2.87	0.319	3.47	0.211 <sup>c</sup>	0.525	0.6700	2.77	0.865
72.20	3.100	2.73	0.258	3.19	0.171 <sup>c</sup>	2.87	0.319	3.50	0.221 <sup>c</sup>	0.540	0.6725	2.95	0.865
79.20	2.950	2.75	0.263	3.31	0.179 <sup>c</sup>	4.12	0.474	3.55	0.220 <sup>c</sup>	0.575	0.6800	3.39	0.993
84.00	2.800	2.89	0.277	3.40	0.183 <sup>c</sup>	<u>CURVE 44</u>		3.67	0.230	0.628	0.6888	3.51	1.10
85.00	2.880	2.94	0.277	3.50	0.186 <sup>c</sup>	1.40	0.132 <sup>c</sup>	3.76	0.234	0.653	0.690	4.06	1.23
93.60	2.900	2.95	0.278	3.51	0.190 <sup>c</sup>	2.02	0.222	3.77	0.240	0.670	0.6925	4.06	1.70
<u>CURVE 41</u>		3.16	0.304	3.53	0.187 <sup>c</sup>	2.24	0.240	4.02	0.261	0.698	0.6938	6.75	2.11
<u>CURVE 42</u>		3.19	0.303	3.60	0.199 <sup>c</sup>	2.24	0.240	4.02	0.261	0.720	0.6962	8.81	2.77
<u>CURVE 43</u>		3.44	0.329	3.70	0.197 <sup>c</sup>	2.59	0.262	4.02	0.261	0.745	0.6975	9.66	3.03
<u>CURVE 44</u>		3.47	0.335	3.80	0.208 <sup>c</sup>	3.24	0.349	4.02	0.261	0.745	0.6950	10.5	3.13
<u>CURVE 45</u>		3.74	0.363	4.05	0.228 <sup>c</sup>	3.61	0.365	4.02	0.261	0.790	0.1025	11.7	3.51
<u>CURVE 46</u>		3.76	0.367	4.08	0.235	4.03	0.444	4.02	0.261	0.840	0.111	14.6	4.41
<u>CURVE 47</u>		4.03	0.402	4.08	0.234	4.03	0.444	4.02	0.261	0.840	0.111	17.8	4.93
<u>CURVE 48</u>		4.03	0.402	4.08	0.234	4.03	0.444	4.02	0.261	0.840	0.111	20.9	5.32

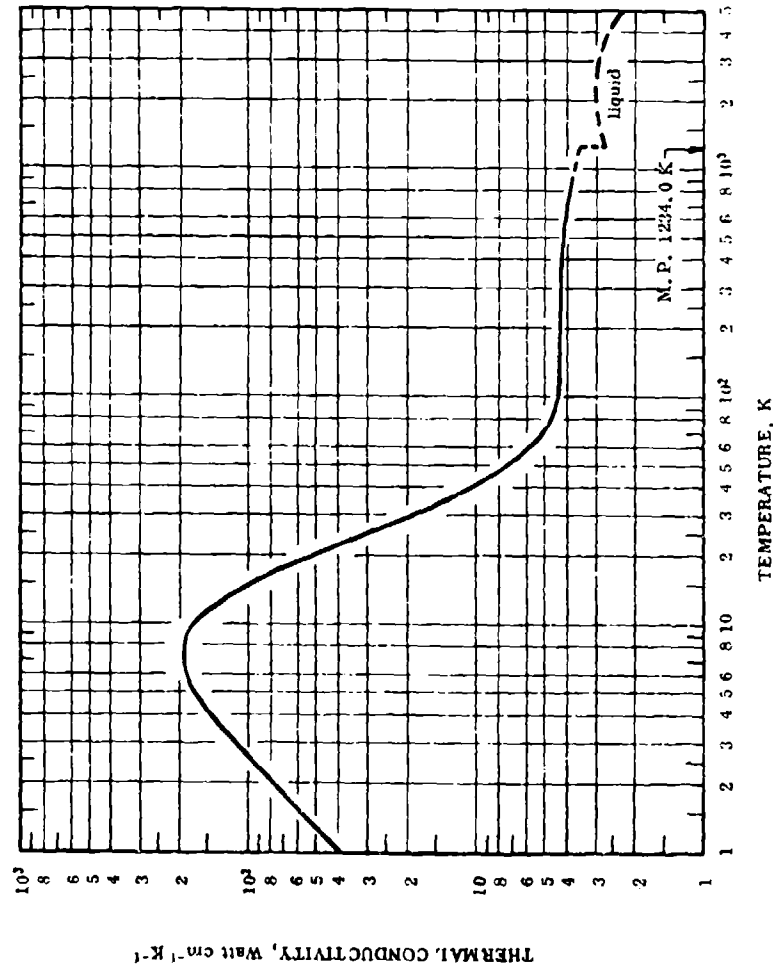
\* Not shown on plot

DATA TABLE NO. 52 (continued)

T	k	T	k	T	k	T	k
<u>CURVE 53 (cont.)</u>							
21.5	5.92	2.2	70	5.33	122.2		
30.1	5.57	2.6	80	6.41	132.8		
69.3	4.43	3.1	95	7.83	137.4		
82.4	4.25	3.6	108	8.55	136.2		
<u>CURVE 54</u>							
1.95	0.304	6.0	151	10.05	129.0		
2.10	0.337	7.5	161	11.59	117.6		
2.34	0.368	9.0	157	14.30	92.3		
2.59	0.424	10.9	138	<u>CURVE 60</u>			
3.20	0.512	14.0	107	298.2	2.84		
3.67	0.588	17.5	67	<u>CURVE 61</u>			
4.14	0.658	19.1	55	273.2	4.44		
6.27	1.04	21.3	43	<u>CURVE 62*</u>			
6.62	1.09	24.3	31	90.2	4.247		
8.41	1.39	26.5	25*	194.7	4.201		
10.4	1.90	30.3	19*	273.2	4.180		
<u>CURVE 57</u>							
13.2	2.36	2.2	83	373.2	4.167		
16.5	2.86	2.5	90				
19.9	3.25	2.9	105				
22.9	3.61	5.7	158				
25.5	4.00	9.7	152				
85.3	3.55	11.7	130				
<u>CURVE 55</u>							
1.80	0.106*	14.8	94				
1.91	0.106*	16.5	77				
2.03	0.110*	<u>CURVE 58</u>					
2.52	0.140*	3.37	95.6				
3.00	0.163*	4.11	110.1				
3.54	0.202	4.97	127.7				
4.16	0.252	6.12	140.4				
5.33	0.353	7.14	146.6				
6.31	0.436	9.40	141.2				
6.79	0.473	10.73	131.2				
7.76	0.589	12.62	112.5				
9.44	0.723	13.47	103.5				
11.3	0.851	15.92	78.2				
15.3	1.08	16.86	70.2				
17.5	1.25	18.16	59.6				
19.3	1.40	21.01	43.2*				
22.9	1.56	22.79	35.3*				
26.4	1.68	23.77	31.8*				
70.8	2.23	25.75	26.5*				
86.3	2.41*	26.60	24.0*				
		29.35	18.7*				

\* Not shown on plot

FIGURE AND TABLE NO. 52R RECOMMENDED THERMAL CONDUCTIVITY OF SILVER



RECOMMENDED VALUES\*

$T_1$	$k_1$	$k_2$	$T_2$	$T_1$	$k_1$	$k_2$	$T_2$
0	0	0	-459.7	500	4.13	239	440.3
1	39.4	2280	-457.9	600	4.05	234	620.3
2	78.3	4520	-456.1	700	3.97	229	800.3
3	115	6640	-454.3	800	3.89	225	980.3
4	147	8490	-452.5	900	3.82	221	1160
5	172	9940	-450.7	1000	(3.74) <sup>†</sup>	(216)	1340
6	187	10800	-448.9	1100	(3.66)	(211)	1520
7	193	11200	-447.1	1200	(3.58)	(207)	1700
8	190	11000	-445.3	1234	(3.55)	(205)	1762
9	181	10500	-443.5		In Liquid State		
10	168	9710	-441.7				
11	154	8900	-439.9	1234	(1.75)	(101)	1762
12	139	8030	-438.1	1300	(1.78)	(103)	1880
13	124	7160	-436.3	1400	(1.83)	(106)	2060
14	109	6300	-434.5	1500	(1.87)	(108)	2240
15	96	5550	-432.7	1600	(1.90)	(110)	2420
16	85	4910	-430.9	1700	(1.93)	(112)	2600
18	66	3810	-427.3	1800	(1.95)	(113)	2786
20	51	2950	-423.7	1900	(1.96)	(113)	2960
25	29.5	1700	-414.7	2000	(1.97)	(114)	3140
30	19.3	1120	-405.7	2200	(1.98)	(114)	3500
35	13.7	792	-396.7	2400	(1.98)	(114)	3860
40	10.5	607	-387.7	2600	(1.97)	(114)	4220
45	8.4	485	-378.7	2800	(1.95)	(113)	4586
50	7.0	404	-369.7	3000	(1.91)	(110)	4940
60	5.5	318	-351.7	3200	(1.87)	(108)	5300
70	4.97	287	-333.7	3400	(1.82)	(105)	5660
80	4.71	272	-315.7	3600	(1.76)	(102)	6020
90	4.60	266	-297.7	3800	(1.70)	(98.2)	6380
100	4.50	260	-279.7	4000	(1.63)	(94.2)	6740
150	4.32	250	-189.7	5000	(1.45)	(83.8)	7640
200	4.30	248	-98.7	5000	(1.23)	(71.1)	8540
250	4.28	247	-9.7	5500	(1.01)	(58.4)	9440
273.2	4.28	247		6000	(0.764)	(44.1)	10340
300	4.27	247		6500	(0.514)	(29.7)	11240
350	4.24	245	170.3	7000	(0.250)	(14.4)	12140
400	4.20	243	260.3	7460	(~0)	(~0)	12968

REMARKS

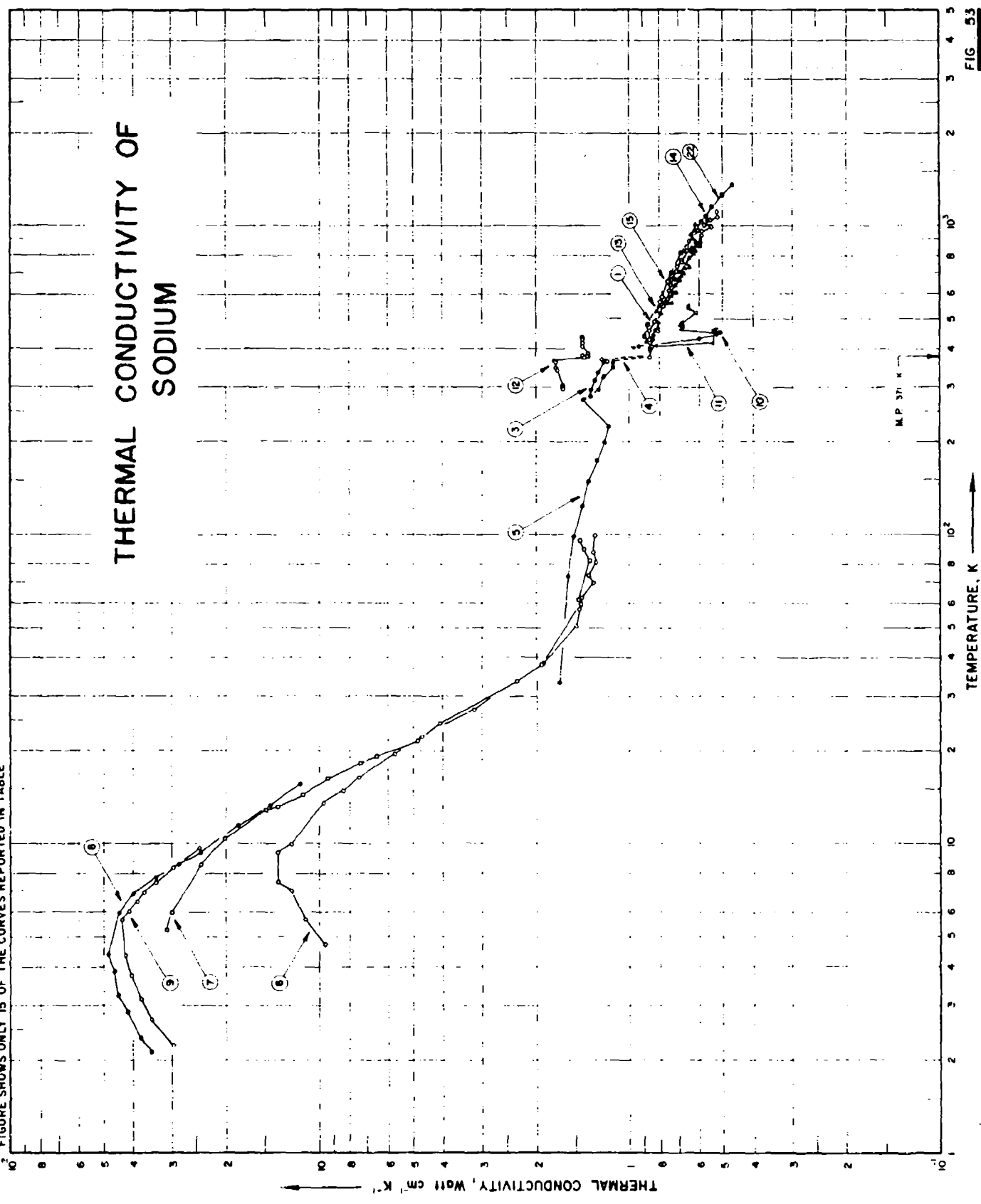
The recommended values are for well-annealed 99.999% pure silver with residual electrical resistivity  $\rho_0 = 0.000620 \mu\Omega \text{ cm}$  (characterization by  $\rho_0$  becomes important at temperatures below about 150 K). The values below 1.5  $T_m$  are calculated to fit the experimental data by using  $n = 2.20$ ,  $a = 0.55$ ,  $m = 2.75$ ,  $\alpha^1 = 7.30 \times 10^{-4}$ , and  $\beta = 0.0254$ . The recommended values that are supported by experimental thermal conductivity data are thought to be accurate to within 2% of the true values near room temperature, and 2 to 5% at other temperatures.

\*  $T_1$  in K,  $k_1$  in Watt  $\text{cm}^{-1} \text{K}^{-1}$ ,  $T_2$  in F, and  $k_2$  in Btu  $\text{hr}^{-1} \text{ft}^{-1} \text{F}^{-1}$ .

† Values in parentheses are extrapolated or estimated.

# THERMAL CONDUCTIVITY OF SODIUM

FIGURE SHOWS ONLY 15 OF THE CURVES REPORTED IN TABLE



## SPECIFICATION TABLE NO. 53 THERMAL CONDUCTIVITY OF SODIUM

(Impurity &lt; 0.20% each; total impurities &lt; 0.50%)

[For Data Reported in Figure and Table No. 53]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
1	105	P	1956	374-961			Pure; thermal conductivity values calculated from measured (in argon) thermal diffusivity data using specific heat data of Ginnings, D.C., et al (J. Res., NBS, 45, 1950) and density data of Miller, R.R. (Liquid Metals Handbook, 2nd ed, 1952).
2	502, 38	L	1951	455-786	1		Impurities: 0.0001 to 0.001 Ag, < 0.00001 K, < 0.001 O, and negligible amounts of Li, Si, Be, Cs, Rb, Ca, Al, Mg, Fe, Cr, Ni, Sn, and Pb; distilled.
3	72	E	1913	279-361			Pure, supplied by Eimer and Amend; electrical resistivity reported as 4.66, 5.06, 5.63, 6.04, and 6.63 $\mu$ ohm cm at 5.7, 21.5, 42.1, 61.4, and 88.1 C, respectively.
4	65	L	1938	358-485			Pure; measured across melting point (97.5 C); electrical conductivity 1.02 and $0.73 \times 10^6$ ohm <sup>-1</sup> cm <sup>-1</sup> at 100 and 200 C, respectively; extrapolation of the thermal conductivity data for the solid and the liquid state to the melting point gives the ratio 1.33.
5	12	F	1926	33-348			Pure; 1.10 cm in dia., 25 cm long; extruded; electrical resistivity 4.26 $\mu$ ohm cm at 0 C.
6	10	L	1951	4.7-99	2-3	Na I	Approx 0.01 to 0.1 Ca and Al; supplied by British-Thomson-Houston Research Lab.; cast under vacuum in soft glass tubes; electrical conductivity ranging from $106$ to $3.15 \times 10^6$ ohm <sup>-1</sup> cm <sup>-1</sup> at 2 to 46.7 K.
7	10	L	1951	5.3-96	2-3	Na II	Trace of Ag; supplied by Messrs. Phillips Ltd, Mitcham; cast under vacuum in soft glass tubes; electrical conductivity ranging from $756$ to $1.0 \times 10^6$ ohm <sup>-1</sup> cm <sup>-1</sup> at 2 to 90 K.
8	92	L	1956	2.1-16		Na 2	High purity; 0.5 mm in dia; electrical resistivity ratio $\rho(295 K)/\rho(0 K) = 3420$ (using Hackspill's value $\rho(295 K) = 4.75 \mu$ ohm cm).
9	92	L	1956	2.2-9.6		Na 3	High purity; 0.13 mm in dia; electrical resistivity ratio $\rho(295 K)/\rho(0 K) = 2860$ (using Hackspill's value $\rho(295 K) = 4.75 \mu$ ohm cm).
10	243	L	1950	407-462			Commercial grade (high purity); supplied by Mine Safety Appliance Co.; 0.694 in. dia; M.P. 97.9 C; specimen in liquid state; apparatus in open air.
11	243	L	1950	403-549			The above specimen; apparatus in heated oven.
12	385	L	1940	296-433	2		Distilled; measured across melting point (approx. 97 C).
13	592	L	1959	402-1011			M.P. 97.5 C; specimen in liquid state; measured in vacuum of approx. $4 \times 10^{-4}$ mm Hg.
14	866, 769	P	1961	623-1153	$\pm 2.5$		Impurities (after test): 0.049 Cr, 0.041 K, 0.016 Fe, 0.016 O, 0.014 Ni, < 0.002 Pb, 0.0017 Mn, 0.0011 Ti, < 0.001 Al, 0.00045 Cu, 0.0003 Ca, 0.00027 Mg, and 0.0002 Ag; distilled; 8.4 mm dia, approx. 230 mm long; in liquid state; measured in vacuum; data calculated from an empirical equation derived from experimental data.
15	770, 853	C	1965	363-1103	5.5		Melting point 97.81 C; specimen in liquid state; AISI 304 stainless steel used as comparative material.
16	148, 857		1966	900-1500	$\pm 20$		Vapor; measured in the 1 mm gap between concentric cylinders 900 mm long; vapor pressure = 0.01 kg cm <sup>-2</sup> .

SPECIFICATION TABLE NO. 53 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
17	148. 857		1966	1000-1500	± 20		Similar to the above except vapor pressure = 0.05 kg cm <sup>-2</sup> .
18	148. 857		1966	1000-1500	± 20		Similar to the above except vapor pressure = 0.1 kg cm <sup>-2</sup> .
19	148. 857		1966	1100-1500	± 20		Similar to the above except vapor pressure = 0.5 kg cm <sup>-2</sup> .
20	148. 857		1966	1200-1500	± 20		Similar to the above except vapor pressure = 1.0 kg cm <sup>-2</sup> .
21	148. 857		1966	800-1200	± 20		Similar to the above except measured on saturation curve.
22	861. 859. 860		1964	437-1366			Density reported as 0.8977, 0.8255, 0.8119, 0.7881, 0.7640, 0.7391, and 0.6967 g cm <sup>-3</sup> at 433.8, 804.1, 873.1, 972.7, 1085, 1149, and 1284 K, respectively; electrical resistivity reported as 5.23, 5.72, 6.54, 6.70, 6.82, 11.04, 11.10, 11.99, 12.42, 14.54, 15.61, 16.25, 18.01, 20.09, 21.86, 24.76, 28.01, 31.54, 35.31, 37.35, 41.76, 46.14, 48.91, 54.16, 60.10, 66.17, 69.59, and 72.48 μohm cm at 302, 324, 356, 365, 370, 406, 413, 431, 444, 501, 525, 542, 554, 630, 668, 726, 790, 850, 913, 945, 1009, 1079, 1108, 1171, 1238, 1300, 1354, and 1360 K, respectively; thermal conductivity values calculated from measured electrical resistivity data and the Lorenz number $2.45 \times 10^{-8} \text{ V}^2 \text{ K}^{-2}$ .
23	756. 862		1962	473-1173			Composition (pretest): < 0.0775 Cs, < 0.0175 K, < 0.0150 Li, 0.0066 Fe, 0.0048 N, 0.0032 O, 0.0022 Ni, and < 0.0010 Cr; composition (posttest): < 0.0375 Cs, < 0.0375 K, 0.0215 C, < 0.0150 Li, 0.0055 O, 0.0049 N, 0.0045 Fe, < 0.0010 Cr, and < 0.0009 Ni; purchased from U.S. Industrial Chemicals Co.; purified by melting and forcing the molten liquid through a 20 micron stainless steel filter under purified argon; electrical resistivity reported as 9.64, 11.44, 13.78, 17.98, 23.16, 28.68, 34.91, 41.86, 46.40, and 51.60 μohm cm at 371.2, 424.5, 482.5, 585.7, 693.9, 804.4, 908.7, 1012.8, 1072.8, and 1126.0 K, respectively; thermal conductivity values calculated from electrical resistivity data using Lorenz function of 2.31, 2.31, 2.33, 2.33, 2.36, 2.41, 2.46, and 2.52 x 10 <sup>-8</sup> V <sup>2</sup> K <sup>-2</sup> at 473, 573, 673, 873, 973, 1073, and 1123 K, respectively; the first four values being derived from the thermal conductivity measurements of Ewing, C. T. and Grand, J. A. (NRL Rept. 3835, 1951) and the authors' own electrical resistivity data.
24	868. 867	L	1965	328	< 15		0.13 Na <sub>2</sub> O.



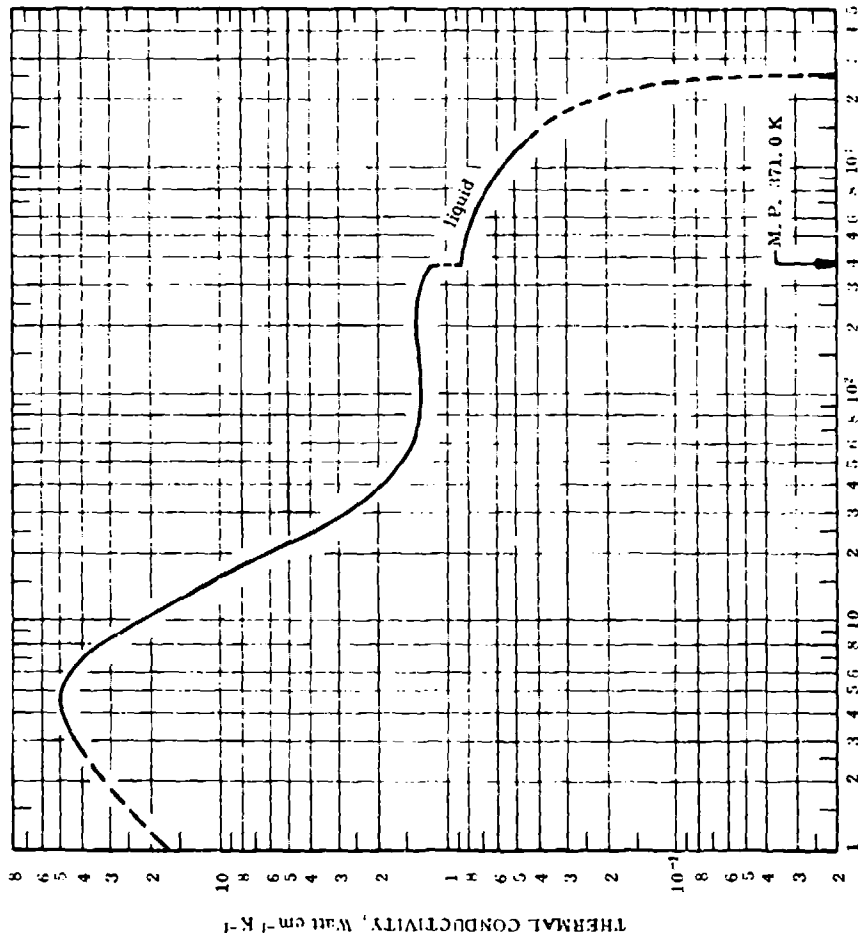


DATA TABLE NO. 53 (continued)

CURVE 14 (cont.)		CURVE 17 (cont.)		CURVE 22 (cont.)	
T	k	T	k	T	k
673.2	0.711 <sup>*</sup>	1200	0.000405	1033	0.585
773.2	0.665 <sup>*</sup>	1300	0.000429	1144	0.540 <sup>*</sup>
873.2	0.628 <sup>*</sup>	1400	0.000459	1255	0.498
973.2	0.590	1500	0.000492	1366	0.460
1073.2	0.561				
1153.2	0.540				
<u>CURVE 15</u>					
363	1.180	1000	0.000497	973.2	0.515
358	1.224	1100	0.000430	573.2	0.768
378	0.860	1200	0.000422	673.2	0.711
413	0.836 <sup>*</sup>	1300	0.000438	773.2	0.668
458	0.808	1400	0.000464	873.2	0.631
518	0.790 <sup>*</sup>	1500	0.000495	973.2	0.600
583	0.783			1073.2	0.573
588	0.780			1173.2	0.549
658	0.740 <sup>*</sup>				
660	0.748	1100	0.000609		
701	0.700 <sup>*</sup>	1200	0.000529		
743	0.676 <sup>*</sup>	1300	0.000499	328	1.236
783	0.662 <sup>*</sup>	1400	0.000499		
838	0.632 <sup>*</sup>	1500	0.000516		
883	0.614 <sup>*</sup>				
928	0.584				
958	0.580				
990	0.542	1200	0.000612		
1000	0.572	1300	0.000556		
1046	0.548	1400	0.000535		
1063	0.518	1500	0.000541		
1103	0.520				
<u>CURVE 16</u>					
900	0.000358	800	0.000485		
1000	0.000341	900	0.000545		
1100	0.000360	1000	0.000593		
1200	0.000391	1100	0.000628		
1300	0.000422	1200	0.000658		
1400	0.000455				
1500	0.000489				
<u>CURVE 17</u>					
1000	0.000436	437	0.883		
1100	0.000394	478	0.860		
		589	0.793 <sup>*</sup>		
		700	0.732		
		812	0.680		
		922	0.629 <sup>*</sup>		

<sup>\*</sup> Not shown on plot

FIGURE AND TABLE NO. 51R RECOMMENDED THERMAL CONDUCTIVITY OF SODIUM



RECOMMENDED VALUES\*

T <sub>1</sub>	k <sub>1</sub>	k <sub>2</sub>	T <sub>2</sub>	T <sub>1</sub>	k <sub>1</sub>	k <sub>2</sub>	T <sub>2</sub>
0	0	0	-459.7	371.0	In Liquid State		
1	(16.6) <sup>‡</sup>	(.959)	-457.9	400	0.88	50.8	208.1
2	(31.8)	(1.940)	-456.1		0.865	50.0	280.3
3	43.2	2.500	-454.3	500	0.815	47.1	440.3
4	48.5	2.840	-452.5	600	0.764	44.1	620.3
5	48.2	2.780	-450.7	700	0.715	41.3	800.3
6	44.2	2.550	-449.9	800	0.668	38.6	980.3
7	38.4	2.220	-447.1	900	0.625	36.1	1160
8	31.7	1.830	-445.3	1000	0.583	33.7	1340
9	26.3	1.520	-443.5	1100	0.543	31.4	1520
10	22.0	1.270	-441.7	1200	0.503	29.1	1700
11	18.8	1.090	-439.9	1300	0.465	26.9	1880
12	16.1	930	-438.1	1400	0.428	24.7	2060
13	14.0	809	-436.3	1500	0.393	22.7	2240
14	12.2	705	-434.5	1600	(0.358)	(20.7)	2420
15	10.7	618	-432.7	1700	(0.325)	(18.8)	2600
16	9.40	543	-430.9	1800	(0.292)	(16.9)	2780
18	7.48	432	-427.3	1900	(0.260)	(15.0)	2960
20	6.09	352	-423.7	2000	(0.229)	(13.2)	3140
25	3.94	228	-414.7	2200	(0.170)	(9.82)	3500
30	2.83	164	-405.7	2400	(0.112)	(6.47)	3860
35	2.22	128	-396.7	2600	(0.056)	(3.24)	4220
40	1.85	109	-387.7	2800	(0.0013)	(0.0751)	4580
45	1.71	98.8	-378.7				
50	1.58	91.3	-369.7				
60	1.45	83.8	-351.7				
70	1.38	79.7	-333.7				
80	1.34	77.4	-315.7				
90	1.33	76.8	-297.7				
100	1.32	76.3	-279.7				
150	1.33	76.8	-189.7				
200	1.38	79.7	-99.7				
250	1.37	79.2	-9.7				
273.2	1.35	78.0	32.0				
300	1.32	76.3	80.3				
350	1.23	71.1	170.3				
371.0	1.20	69.3	208.1				

REMARKS

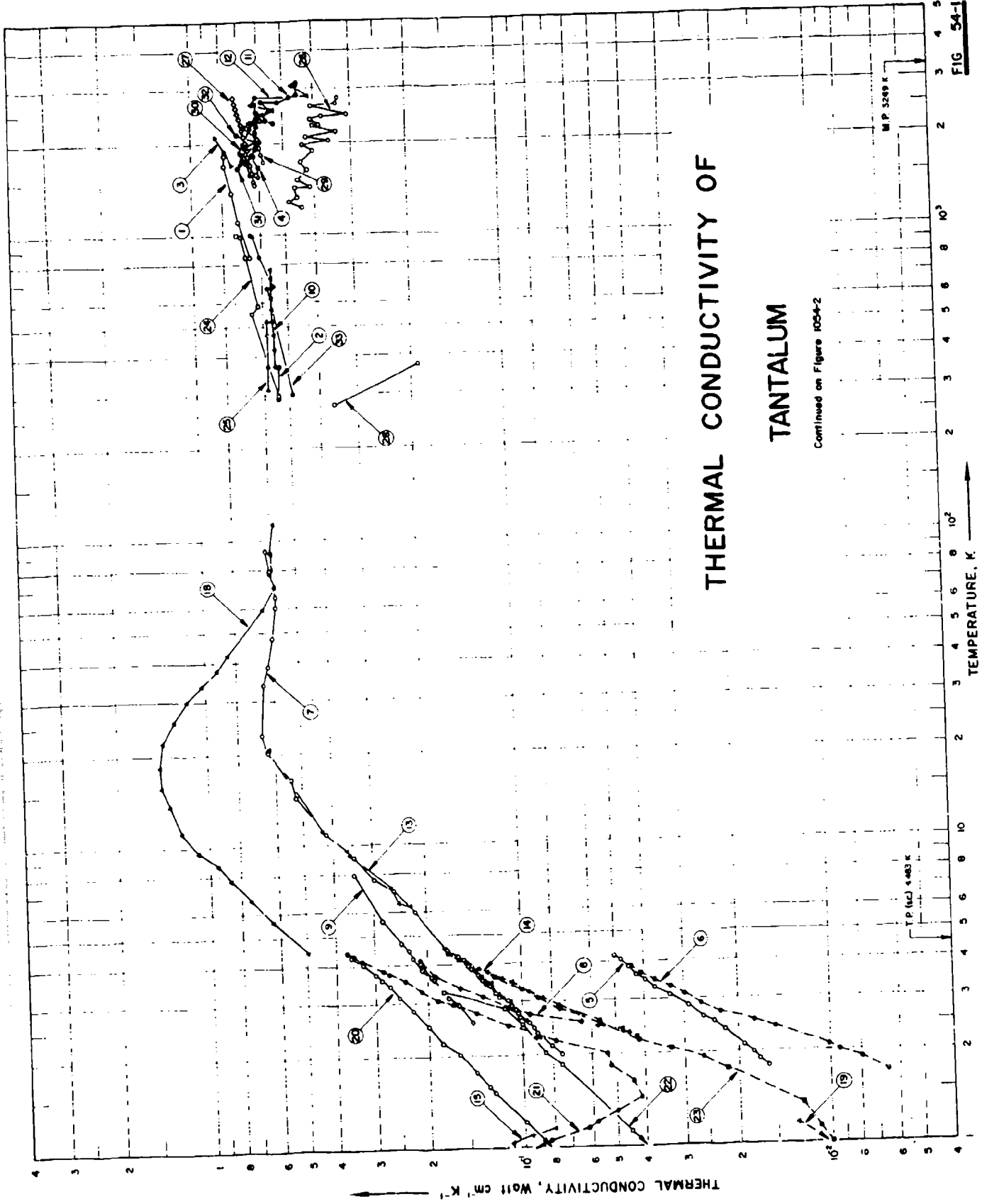
The recommended values are for high-purity sodium with residual electrical resistivity  $\rho_0 = 0.00147 \mu\Omega \text{ cm}$  (characterization by  $\rho_0$  becomes important at temperatures below about 100 K). The values below 1.5 Tm are calculated to fit the experimental data by using  $n = 2.00$ ,  $\sigma^* = 3.50 \times 10^{-6}$ , and  $\beta^* = 0.0600$ . The recommended values that are supported by experimental thermal conductivity data are thought to be accurate to within 5% of the true values near room temperature and 5 to 10% at other temperatures.

\* T<sub>1</sub> in K, k<sub>1</sub> in Watt cm<sup>-1</sup> K<sup>-1</sup>, T<sub>2</sub> in F, and k<sub>2</sub> in Btu hr<sup>-1</sup> ft<sup>-1</sup> F<sup>-1</sup>.

‡ Values in parentheses are extrapolated or estimated.

# THERMAL CONDUCTIVITY OF TANTALUM

Continued on Figure 1054-2



T.P. (C) 3,483 K

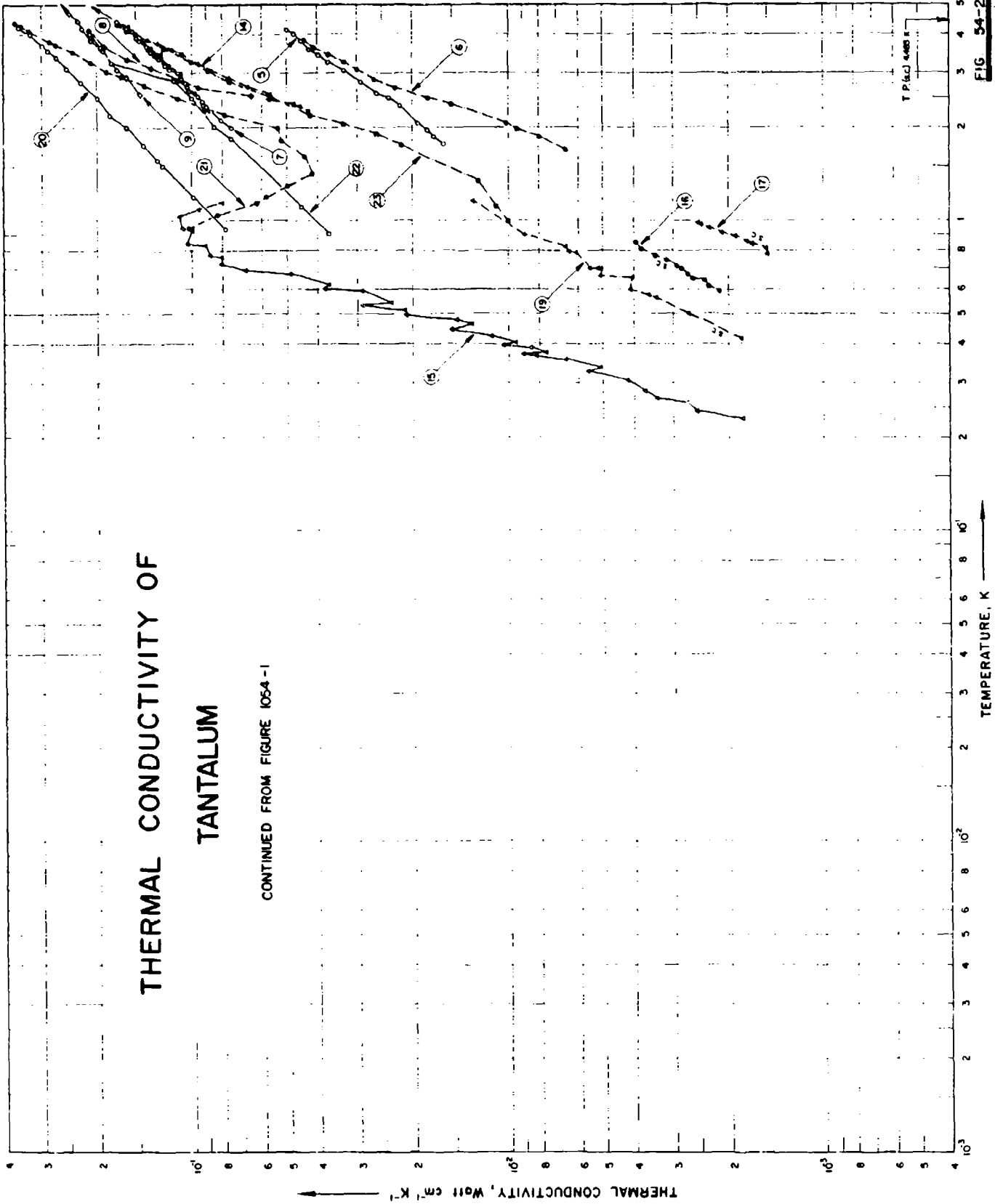
M.P. 3299 K

TEMPERATURE, K. →

THERMAL CONDUCTIVITY, Watt cm<sup>-1</sup> K<sup>-1</sup> ↑

# THERMAL CONDUCTIVITY OF TANTALUM

CONTINUED FROM FIGURE 1054-1



TEMPERATURE, K

FIG 54-2

## SPECIFICATION TABLE NO. 54 THERMAL CONDUCTIVITY OF TANTALUM

(Impurity  $\leq 0.20\%$  each; total impurities  $\leq 0.50\%$ )

(For Data Reported in Figure and Table No. 54)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
1	42	L	1956	842-1820	5		Impurities: (pre-test): 0.052 N, traces of Ca, Cu, and Mg; impurities (after test): 0.12 O, 0.044 N, 0.0061 H, traces of Al, Ca, Cu, Fe, and Mg; sintered; density 16.48 g cm <sup>-3</sup> .
2	8	F	1914	290, 373			Pure; 0.075 cm dia x 28.14 cm long; specific gravity 16.67; electrical resistivity reported as 14.452 and 19.178 $\mu\text{ohm cm}$ at 0 and 100 C, respectively.
3	153	E	1914	1700-2100		No. 5	Pure; filament.
4	849		1966	1665-2671			0.0036 O, 0.0018 N, 0.00009 H, and 0.00005 C; 1.9662 cm dia x 0.2273 cm thick; machined from a 1 in. rod supplied by Fansteel Metallurgical Corp; average grain size 1.86 mm; density 16.60 g cm <sup>-3</sup> ; thermal conductivity derived from the temp distribution on the flat surface of the cylindrical disc heated in vacuum by induction.
5	74	L	1950	1.7-4.2	3	Hilger 8017, Ta1	99.9 pure; polycrystalline; superconducting transition point 4.38 K; measured in a magnetic field; in normal state.
6	74	L	1950	1.7-3.9	3	Hilger 8017, Ta1	The above specimen in superconducting state.
7	122	L	1955	2.0-92	3	JM3804, Ta1	99.98 pure; polycrystalline; specimen 0.225 cm in dia, 3 cm long; Johnson Matthey's unannealed rod; electrical resistivity ratio $\rho(293K)/\rho(4K) = 19.7$ ; electrical resistivity reported as 0.62, 0.63, 0.67, 0.90, 1.05, 1.46, 2.07, 2.35, 3.04, and 3.51 $\mu\text{ohm cm}$ at 11.3, 16.1, 20.5, 32.2, 37.2, 46.9, 59.6, 65.1, 78.4, and 89.4 K, respectively; superconducting transition temp 4.38 K; measured in a magnetic field; in normal state.
8	96	L	1950	2.6-4.2	2-5		Very pure; in superconducting state.
9	96	L	1950	2.6-7.9	2-5		Very pure; measured in a magnetic field; in normal state.
10	182	L	1959	373-773	5		99.886 Ta, 0.0100 Nb, 0.0140 O, 0.0100 W, 0.0100 Zn, 0.0100 Cu, 0.0050 N, 0.0050 Mo, 0.0025 C, 0.0025 H, 0.0010 each of Pb, Sn, and Zr, 0.0020 each of Co, Sr, and V, 0.0010 each of Al, Ba, Bi, Cr, Fe, and Ni, 0.0005 each of Ag and Ti, 0.0003 each of B, Mn, Si, and Na, 0.0002 Be, and 0.0001 each of Ca, Cu, and Mg; specimen bar machined from a rod obtained from Fansteel Metallurgical Corp; data taken from smoothed curve.
11	255	E	1960	2343-3148		1	0.02 Si, 0.005 Fe, 0.001 Mo, 0.0008 C, and 0.052 others; prepared by pressing and sintering tantalum powder, then hot and cold rolled.
12	255	E	1960	2326-3071		2	0.0032-0.005 O, 0.0035 Nb, 0.0028 Fe, 0.0016 C, $\leq 0.001$ N, and 0.0175 others; cast in vacuum, cold rolled, swaged, and cold drawn.
13	97	L	1952	2.3-21	2-3	JM3804, Ta1	99.98 pure; 1-2 mm dia x 5 cm long; obtained from Johnson Matthey Co.; measured in a magnetic field; in normal state.
14	97	L	1952	2.3-3.9	2-3	Ta 1	The above specimen in superconducting state.
15	705	L	1962	0.23-1.2	5	Ta II	Single crystal; specimen dia 6.1 mm; ratio of length to cross sectional area 16.6 cm <sup>-1</sup> ; obtained by floating zone melting polycrystalline rod in a vacuum; electrical resistivity ratio $\rho(298K)/\rho_0 = 47.0$ .

SPECIFICATION TABLE NO. 54 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
16	400	L	1953	0.60-0.86		Ta 1	99.98 pure; polycrystalline; effected by "frozen in" magnetic field; in superconducting state.
17	400	L	1953	0.79-1.0		Ta 1	Separate run of the above specimen; in superconducting state. 99.9 pure; specimen consisted of four 1.5 mm wires supplied by Fansteel Metallurgy Corp.; annealed in vacuum at 2500 C; ideal electrical resistivity reported as 0.0032, 0.017, 0.051, 0.12, 0.23, 0.54, 0.95, 1.43, 1.96, 2.50, 3.03, 3.65, 4.6, 5.6, 6.65, 7.65, 8.6, 9.6, 11.0, 12.1, and 13.1 $\mu\text{hm cm}$ at 10, 15, 20, 25, 30, 40, 50, 60, 70, 80, 90, 100, 120, 140, 160, 180, 200, 220, 250, 273, and 295 K, respectively; electrical resistivity ratio $\rho(295\text{K})/\rho_0 = 62.1$ .
18	401	L	1959	4.4-114		Ta 3	
19	412	L	1955	0.42-1.2		JM3804	
20	501	L	1961	0.95-4.3			0.005 Fe, 0.003 Si, 0.0003 O, and 0.00025 H; single crystal; specimen obtained by floating-zone melting polycrystalline rod; electrical resistivity ratio $\rho(298\text{K})/\rho_0 = 63$ ; measured in a magnetic field; in normal state.
21	501	L	1961	0.95-4.4			The above specimen in superconducting state.
22	501	L	1961	0.92-4.0			0.1 Nb, 0.01 C, 0.01 Fe, 0.01 Mo, 0.01 W, 0.001 O, 0.00075 N, and 0.00045 H; polycrystalline; electrical resistivity ratio $\rho(298\text{K})/\rho_0 = 31$ ; measured in a magnetic field; in normal state; specimen same as that used by Rosenberg in 1955 (curve 7).
23	501	L	1961	1.0-4.3		No. 9	The above specimen in superconducting state.
24	599	E	1961	299-1000			99.9 pure; obtained from Fansteel Metallurgical Corp.; density 16.4 $\text{g cm}^{-3}$ ; electrical resistivity reported as 15.1, 34.3, 35.0, 41.3, 48.2, and 48.8 $\mu\text{hm cm}$ at 297, 670, 685, 840, 980, and 1000 K, respectively.
25	652	L, C	1961	323-523		JM 615	Spectrographically standardized tantalum; obtained from John Mathey and Co.; about 4.5 mm in dia and 10 cm long; electrical resistivity reported as 14.5, 15.45, 17.72, 22.25, and 24.4 $\mu\text{hm cm}$ at 293, 323, 373, 473, and 523 K, respectively; Armco iron used as the comparative material.
26	709	E	1962	1233-2793			1 mm in dia, 30 mm long; electrical resistivity reported as 50, 73, 89, and 109 $\mu\text{hm cm}$ at 900, 1500, 2000, and 2500 C, respectively.
27	654	P	1965	1460-2820 $\pm 5$			~99.89 Ta (by difference), <0.1 Nb, <0.01 C, and traces of other elements; 0.040 in. thick sheet; obtained from Murex Co.; vacuum beam melted; average grain size after testing 140 $\mu$ ; density 16.6 $\text{g cm}^{-3}$ ; thermal conductivity values calculated from measured data of thermal diffusivity using the specific heat data of Kubaschewski, O. and Evans, L. L., (Metallurgical Thermochemistry, London, Pergamon, 1956).
28	24	E	1943	273, 373	10-90		99.9 pure; wire 0.01 in. in dia and ~15.7 in. long (40 cm); obtained from Fansteel Corp; electrical resistivity reported as 2.46, 12.41, and 17.18 $\mu\text{hm cm}$ at 77, 33, 273, 2, and 373.4 K, respectively; measured in a vacuum of $10^{-4}$ mm Hg.

SPECIFICATION TABLE NO. 54 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
29	849	-	1966	1578-2007		No. 1	0.0019 C, 0.0017 H, 0.0017 N, and 0.0017 O; specimen 2.4892 cm in dia and 0.3927 cm thick; avg. grain size 0.26 mm; density 16.65 g cm <sup>-3</sup> ; thermal conductivity derived from the temp distribution on the flat surface of the cylindrical disc specimen heated in high vacuum (10 <sup>-4</sup> mm Hg) by high frequency induction.
30	849	-	1966	1700-2398		No. 2	0.0655 O, 0.0137 C, 0.0016 N, and 0.00027 H; machined from the same bar as the above specimen; 2.2232 cm in dia and 0.2125 cm thick; density 16.63 g cm <sup>-3</sup> ; measuring method same as above.
31	849	-	1966	1660-2490		No. 3	0.0114 O, 0.0016 N, 0.003 C, and 0.00027 H; machined from the same bar as the above specimen; 2.2232 cm in dia and 0.2018 cm thick; avg. grain size 1.04 mm dia; density 16.62 g cm <sup>-3</sup> ; measuring method same as above.
32	849	-	1966	1563-2142		No. 4	0.0036 O, 0.0018 N, 0.0009 H, and 0.00005 C; machined from the same bar as the above specimen; 1.9075 cm in dia and 0.2316 cm thick; avg. grain size 1.23 mm dia; density 16.63 g cm <sup>-3</sup> ; measuring method same as above.
33	313	T	1962	300-995			No details given for the specimen; thermal conductivity measured by the "small area contact method".



## DATA TABLE NO. 54 THERMAL CONDUCTIVITY OF TANTALUM

(Impurity &lt; 0.20% each; total impurities &lt; 0.50%)

[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup>K<sup>-1</sup>]

T	k	T	k	T	k	T	k	T	k	T	k	T	k	T	k
<b>CURVE 1</b>															
841.5	0.676	3.84	0.0450	7.00	0.259	473.2	0.559	3.85	0.146	0.38	0.0076	0.79	0.0015		
1101.5	0.713	4.03	0.0477	7.63	0.297	523.2	0.582	4.12	0.157	0.395	0.0085	0.82	0.00152		
1378.7	0.746	4.18	0.0500	9.07	0.344	573.2	0.563	4.23	0.162	0.40	0.0104	0.85	0.00168		
1688.7	0.786	<b>CURVE 6</b>													
1819.8	0.782	10.93	0.412	10.93	0.412	623.2	0.563	6.25	0.227	0.41	0.0095	0.865	0.00175		
<b>CURVE 2</b>															
1.70	0.0066	14.43	0.525	16.50	0.590	673.2	0.563	6.35	0.248	0.45	0.0151	0.90	0.00190		
1.88	0.0080	20.20	0.637	20.20	0.637	723.2	0.563	7.15	0.264	0.47	0.0132	0.925	0.00210		
1.98	0.0094	23.10	0.655	23.10	0.655	773.2	0.563	7.20	0.264	0.485	0.0146	0.952	0.00232		
2.07	0.0102	33.80	0.650	33.80	0.650	<b>CURVE 11</b>									
2.40	0.0152	38.80	0.625	38.80	0.625	2343	0.533	11.03	0.432	0.52	0.0215	0.98	0.00245		
2.52	0.0178	48.05	0.600	48.05	0.600	2427	0.602	20.5	0.635	0.54	0.029	9.985	0.0025		
2.70	0.0227	60.40	0.583	60.40	0.583	2587	0.543	20.8	0.63	0.55	0.0235	<b>CURVE 18</b>			
2.88	0.0265	65.60	0.583	65.60	0.583	2742	0.582	2.26	0.0426	0.60	0.038	4.41	0.495		
3.10	0.0300	79.20	0.608	79.20	0.608	2783	0.473	2.35	0.0453	0.61	0.037	5.64	0.632		
3.28	0.0330	92.40	0.622	92.40	0.622	2839	0.473	2.37	0.0474	0.68	0.049	6.65	0.743		
3.46	0.0370	<b>CURVE 3</b>													
3.65	0.0410	2.55	0.065	2.55	0.065	3093	0.452	2.38	0.0474	0.70	0.068	7.71	0.862		
3.85	0.0440	2.70	0.095	2.70	0.095	3148	0.450	2.47	0.0518	0.73	0.081	8.62	0.947		
<b>CURVE 4</b>															
1665	0.601	2.86	0.110	2.86	0.110	<b>CURVE 14</b>									
1816	0.630	3.12	0.134	3.12	0.134	2.26	0.0426	2.56	0.0567	0.77	0.0805	11.3	1.24		
1923	0.616	3.34	0.160	3.34	0.160	2.35	0.0453	2.72	0.0663	0.84	0.091	13.6	1.35		
2043	0.597	3.68	0.190	3.68	0.190	2.47	0.0474	2.72	0.0765	0.85	0.104	15.7	1.44		
2163	0.615	4.15	0.213	4.15	0.213	2.56	0.0518	2.90	0.0865	0.88	0.108	18.4	1.45		
2260	0.610	<b>CURVE 5</b>													
2468	0.597	2.00	0.075	2.00	0.075	2326	0.649	3.07	0.0862	0.94	0.10	22.0	1.41		
2524	0.605	2.13	0.081	2.13	0.081	2563	0.536	3.21	0.0952	0.95	0.108	25.9	1.28		
2671	0.616	2.35	0.090	2.35	0.090	2678	0.632	3.48	0.107	1.04	0.11	30.0	1.16		
<b>CURVE 7</b>															
2.44	0.0925	2.44	0.0925	2.44	0.0925	2811	0.613	3.57	0.116	1.08	0.096	33.6	1.03		
2.65	0.0955	2.54	0.0955	2.54	0.0955	2844	0.438	3.77	0.129	1.15	0.080	37.7	0.919		
2.74	0.102	2.65	0.102	2.65	0.102	3071	0.467	3.88	0.138	1.15	0.080	42.2	0.848		
2.80	0.105	2.74	0.103	2.74	0.103	<b>CURVE 12</b>									
2.90	0.108	2.94	0.162	2.94	0.162	2.26	0.049	2.72	0.0765	0.85	0.104	15.7	1.44		
3.02	0.111	3.10	0.174	3.10	0.174	2.35	0.0453	2.90	0.0865	0.88	0.108	18.4	1.45		
3.10	0.118	3.34	0.190	3.34	0.190	2.47	0.0474	3.07	0.0952	0.94	0.10	22.0	1.41		
3.20	0.122	3.56	0.192	3.56	0.192	2.56	0.0518	3.21	0.0952	0.95	0.108	25.9	1.28		
3.37	0.131	3.85	0.210	3.85	0.210	2.72	0.0663	3.48	0.107	1.04	0.11	30.0	1.16		
3.54	0.139	4.20	0.225	4.20	0.225	2.86	0.0765	3.77	0.129	1.15	0.080	37.7	0.919		
3.68	0.147	4.46	0.232	4.46	0.232	2.90	0.0865	3.88	0.138	1.15	0.080	42.2	0.848		
3.85	0.150	4.70	0.246	4.70	0.246	3.00	0.109	3.57	0.129	1.15	0.080	59.4	0.646		
3.92	0.155	5.00	0.264	5.00	0.264	3.17	0.119	3.77	0.129	1.15	0.080	70.7	0.588		
4.13	0.162	7.94	0.345	7.94	0.345	3.35	0.125	3.88	0.138	1.15	0.080	78.0	0.605		
4.25	0.174	<b>CURVE 13</b>													
4.48	0.196	2.26	0.049	2.26	0.049	2.35	0.0453	2.90	0.0865	0.60	0.00215	80.0	0.600		
4.62	0.219	2.56	0.0567	2.56	0.0567	2.47	0.0474	3.07	0.0952	0.62	0.00232	90.6	0.595		
4.85	0.238	2.72	0.0663	2.72	0.0663	2.56	0.0518	3.21	0.0952	0.66	0.00260	91.8	0.600		
5.08	0.258	2.90	0.0765	2.90	0.0765	2.72	0.0765	3.48	0.107	0.68	0.00270	114.1	0.587		
5.31	0.278	3.07	0.0862	3.07	0.0862	2.86	0.0765	3.77	0.129	0.71	0.00282	<b>CURVE 19</b>			
5.54	0.298	3.21	0.0952	3.21	0.0952	2.90	0.0865	3.88	0.138	0.725	0.00294	0.42	0.00183		
5.77	0.318	3.48	0.107	3.48	0.107	3.00	0.109	3.57	0.129	0.75	0.00315	0.505	0.00268		
6.00	0.338	3.57	0.116	3.57	0.116	3.17	0.119	3.77	0.129	0.78	0.00342	0.57	0.0034		
6.23	0.358	3.77	0.129	3.77	0.129	3.35	0.125	3.88	0.138	0.82	0.00380	0.58	0.0036		
6.46	0.378	3.88	0.138	3.88	0.138	3.46	0.125	3.57	0.129	0.82	0.00395	0.58	0.0036		
6.69	0.425	4.23	0.155	4.23	0.155	3.54	0.133	3.77	0.129	0.82	0.00395	0.58	0.0036		
6.92	0.445	4.48	0.174	4.48	0.174	3.60	0.137	3.88	0.138	0.82	0.00395	0.58	0.0036		
7.15	0.465	4.73	0.196	4.73	0.196	3.62	0.135	3.88	0.138	0.82	0.00395	0.58	0.0036		
7.38	0.485	5.08	0.219	5.08	0.219	3.62	0.135	3.88	0.138	0.82	0.00395	0.58	0.0036		

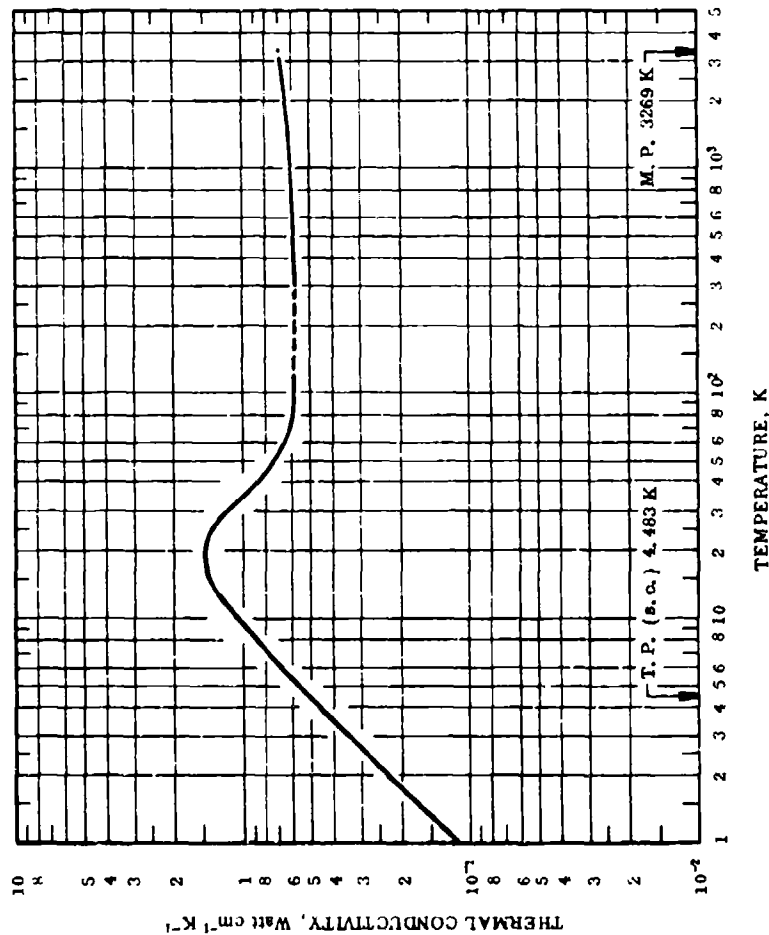
Not shown on plot

DATA TABLE NO. 54 (continued)

CURVE 19 (cont.)		CURVE 21 (cont.)		CURVE 25		CURVE 26		CURVE 28		CURVE 33 (cont.)	
T	k	T	k	T	k	T	k	T	k	T	k
0.665	0.00405	3.51	0.240	323.2	0.582	1578	0.617	273.2	0.36	845	0.613
0.670	0.0051	3.72	0.270	373.2	0.585	1746	0.594	373.4	0.19	988	0.643
0.705	0.00505	3.84	0.283	473.2	0.585	1841	0.590			995	0.652
0.710	0.0052	4.12	0.330	523.2	0.582	1942	0.597				
0.710	0.0055	4.30	0.350			2007	0.606				
0.796	0.0461	4.40	0.368								
0.80	0.00645										
0.83	0.0066										
0.91	0.00885										
1.16	0.0130										
<u>CURVE 20</u>		<u>CURVE 22</u>		<u>CURVE 27</u>		<u>CURVE 30</u>		<u>CURVE 31</u>		<u>CURVE 32</u>	
0.95	0.0783	0.92	0.0370	1460	0.627	1700	0.691	1660	0.703	1563	0.603
1.20	0.0995	1.12	0.0450	1500	0.630	1759	0.678	1828	0.680	1585	0.640
1.51	0.124	1.85	0.0750	1600	0.637	1869	0.693	2043	0.598	1712	0.665
1.58	0.129	2.02	0.0856	1700	0.644	1985	0.675	2269	0.680	1714	0.668
1.76	0.143	2.50	0.100	1800	0.651	2033.2	0.643	2391	0.567	1811	0.621
2.02	0.162	2.85	0.114	1900	0.659	2093.2	0.423	2490	0.574	1907	0.667
2.20	0.182	3.25	0.180	2000	0.666	2153.2	0.339			2020	0.613
2.50	0.200	3.65	0.195	2200	0.681	2273.2	0.402			2142	0.707
2.82	0.227	4.01	0.212	2300	0.688	2373.2	0.385				
3.12	0.252			2400	0.695	2443.2	0.410				
3.40	0.270			2500	0.703	2493.2	0.379				
3.57	0.287			2600	0.710	2618.2	0.310				
3.71	0.300			2700	0.717	2688.2	0.410				
4.01	0.328			2820	0.728	2793.2	0.335				
4.26	0.357						0.331				
<u>CURVE 21</u>		<u>CURVE 23</u>		<u>CURVE 24</u>		<u>CURVE 29</u>		<u>CURVE 32</u>		<u>CURVE 33</u>	
0.95	0.102	1.00	0.0100	299	0.542	1578	0.517	1660	0.703	845	0.613
1.05	0.0832	1.13	0.0110	670	0.661	1746	0.594	1828	0.680	988	0.643
1.15	0.0628	1.35	0.0125	686	0.621	1841	0.590	2043	0.598	995	0.652
1.20	0.0581	1.76	0.0217	845	0.680	1942	0.597	2269	0.680		
1.30	0.0500	1.91	0.0266	1000	0.709	2007	0.606	2391	0.567		
1.44	0.0418	2.06	0.0331	1600	0.728			2490	0.574		
1.63	0.0440	2.20	0.0410	2200	0.728						
1.83	0.0521	2.52	0.0562	2820	0.728						
2.00	0.0535	2.82	0.0755								
2.22	0.0783	3.06	0.0890								
2.50	0.111	3.28	0.100								
2.75	0.142	3.61	0.122								
2.95	0.168	4.05	0.152								
3.05	0.187	4.16	0.160								
3.28	0.210	4.34	0.174								

Not shown on plot

FIGURE AND TABLE NO. 54R RECOMMENDED THERMAL CONDUCTIVITY OF TANTALUM



T <sub>1</sub>	k <sub>1</sub>	k <sub>2</sub>	T <sub>2</sub>	T <sub>1</sub>	k <sub>1</sub>	k <sub>2</sub>	T <sub>2</sub>
0	0	0	-459.7	500	0.582	33.6	440.3
1	0.115	6.64	-457.9	600	0.586	33.9	620.3
2	0.230	13.3	-456.1	700	0.590	34.1	800.3
3	0.345	19.9	-454.3	800	0.594	34.3	980.3
4	0.459	26.5	-452.5	900	0.598	34.6	1160
5	0.571	33.0	-450.7	1000	0.602	34.8	1340
6	0.681	39.3	-448.9	1100	0.606	35.0	1520
7	0.788	45.5	-447.1	1200	0.610	35.2	1700
8	0.891	51.5	-445.3	1300	0.614	35.5	1880
9	0.989	57.1	-443.5	1400	0.618	35.7	2060
10	1.08	62.4	-441.7	1500	0.622	35.9	2240
11	1.16	67.0	-439.7	1600	0.626	36.2	2420
12	1.24	71.6	-438.1	1700	0.630	36.4	2600
13	1.30	75.1	-436.3	1800	0.634	36.6	2780
14	1.36	78.6	-434.5	1900	0.637	36.8	2960
15	1.40	80.9	-432.7	2000	0.640	37.0	3140
16	1.44	83.2	-430.9	2200	0.647	37.4	3500
18	1.47	84.9	-427.3	2400	0.653	37.7	3860
20	1.47	84.9	-423.7	2600	0.658	38.0	4220
25	1.36	78.6	-414.7	2800	0.663	38.3	4580
30	1.16	67.0	-405.7	3000	0.665	38.4	4940
35	0.99	57.2	-396.7	3200	(0.666)	(38.5)	5300
40	0.87	50.3	-387.7				
45	0.78	45.1	-378.7				
50	0.72	41.6	-369.7				
60	0.651	37.6	-351.7				
70	0.616	35.6	-333.7				
80	0.603	34.8	-315.7				
90	0.596	34.4	-297.7				
100	0.592	34.2	-279.7				
150	(0.580)*	(33.5)	-189.7				
200	(0.575)	(33.2)	-99.7				
250	(0.574)	(33.2)	-9.7				
273.2	(0.574)	(33.2)	32.0				
300	0.575	33.2	80.3				
350	0.576	33.3	170.3				
400	0.578	33.4	260.3				

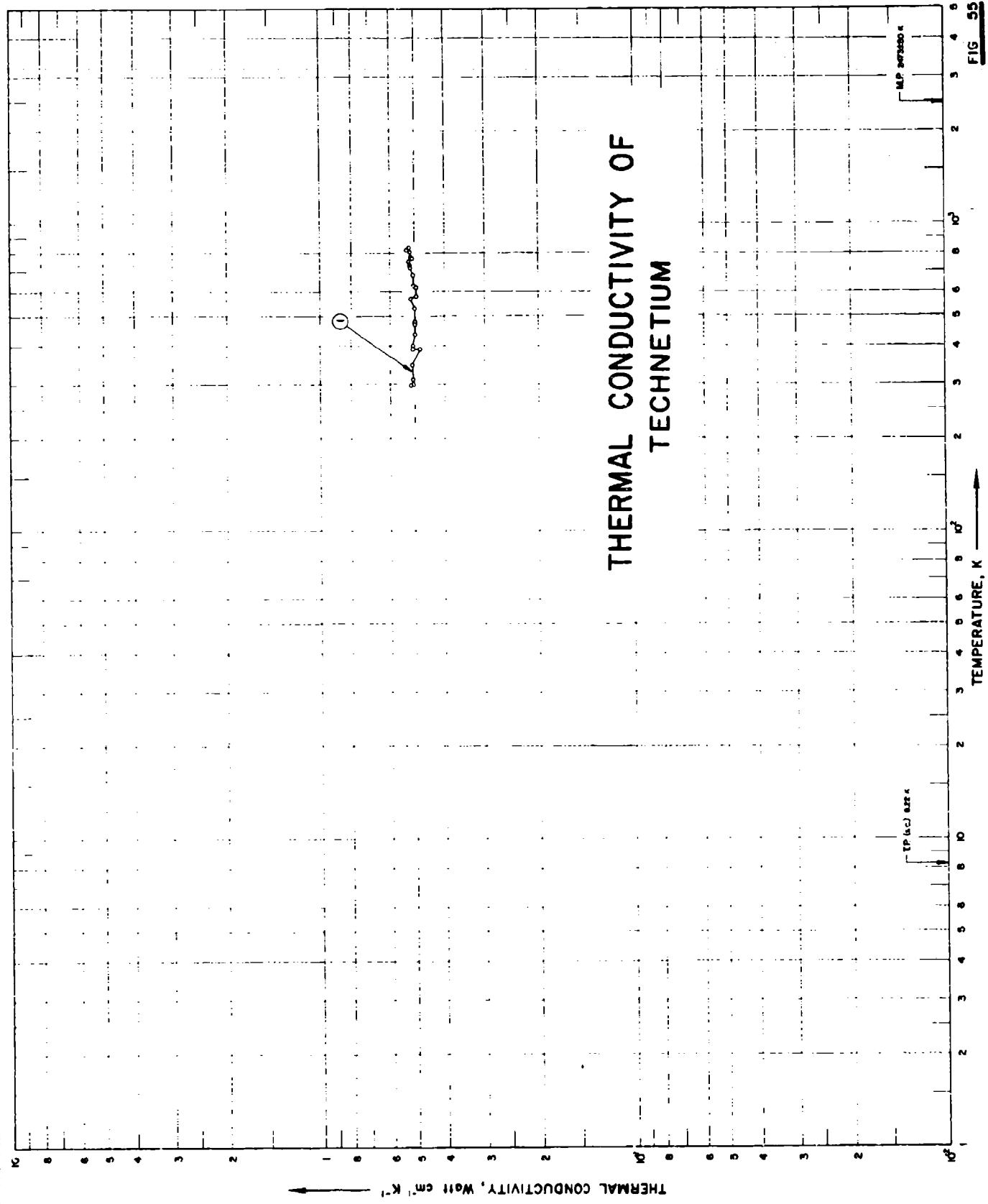
REMARKS

The recommended values are for well-annealed 99.9% pure tantalum with residual electrical resistivity  $\rho_0 = 0.212 \mu\Omega \text{ cm}$  (characterization by  $\rho_0$  becomes important at temperatures below about 150 K). The values below  $1.5 T_m$  are calculated to fit the experimental data by using  $n = 2.10$ ,  $\alpha' = 4.52 \times 10^{-4}$ , and  $\beta = 6.69$ . The recommended values that are supported by experimental thermal conductivity data are thought to be accurate to within 5% of the true values near room temperature, and 5 to 10% at other temperatures.

\* T<sub>1</sub> in K, k<sub>1</sub> in Watt cm<sup>-1</sup> K<sup>-1</sup>, T<sub>2</sub> in F, and k<sub>2</sub> in Btu hr<sup>-1</sup> ft<sup>-1</sup> F<sup>-1</sup>.

† Values in parentheses are extrapolated or interpolated.

FIG 55



# THERMAL CONDUCTIVITY OF TECHNETIUM

THERMAL CONDUCTIVITY, Watt cm<sup>-1</sup> K<sup>-1</sup> ←

TEMPERATURE, K →

## SPECIFICATION TABLE NO. 55 THERMAL CONDUCTIVITY OF TECHNETIUM

(Impurity &lt; 0.20% each; total impurities &lt; 0.50%)

[ For Data Reported in Figure and Table No. 55 ]

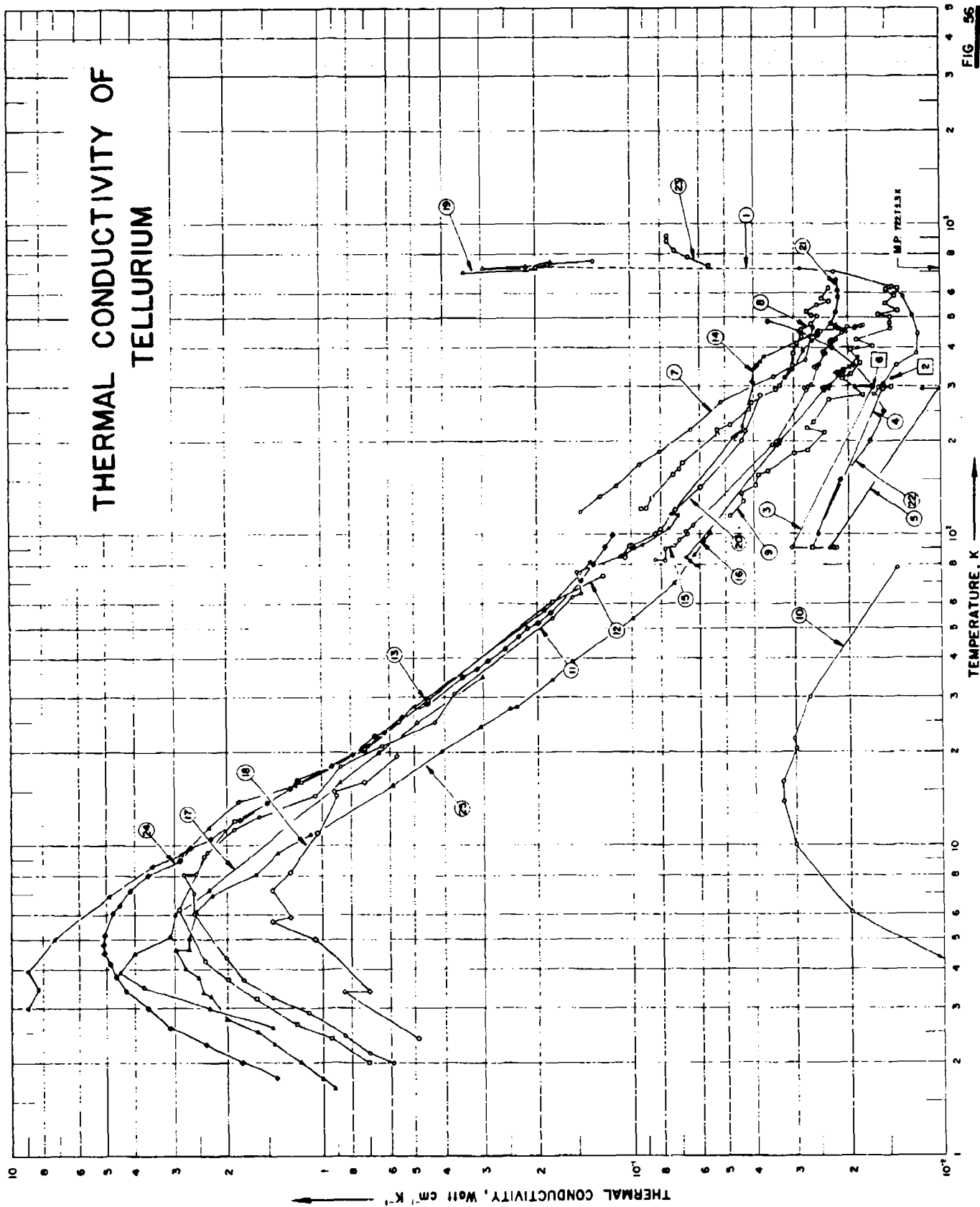
Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition ( weight percent), Specifications and Remarks
1	830	P	1965	298-838			Total impurities ~0.0150; specimen ~0.12 cm thick and 2.3 cm in dia; prepared from reduced metal recovered from fission product wastes; the material melted in an electron beam evaporator, heated to 1540 C by induction, press forged, and ground to final shape; thermal conductivity values calculated from measured data of thermal diffusivity, the measured density (11.492 g cm <sup>-3</sup> ), and the heat capacity data taken from Stull, D. R. and Sinke, G. C. (Thermodynamic Properties of the Elements, American Chemical Soc., Washington, D. C., pp. 198-9, 1956).

DATA TABLE NO. 55 THERMAL CONDUCTIVITY OF TECHNETIUM

(Impurity 0.20% each; total impurities = 0.50%)  
 Temperature T, K. Thermal Conductivity,  $k$ , Watt  $\text{cm}^{-1} \text{K}^{-1}$ .

T, K	k	CURVE 1
298	6.515	
301	6.502	
311	6.504	
318	6.510	
331	6.481	
339	6.506	
401	6.506	
438	6.498	
471	6.498	
481	6.498	
531	6.498	
571	6.515	
581	6.494	
621	6.494	
628	6.502	
678	6.510	
683	6.502	
718	6.519	
733	6.519	
751	6.523	
771	6.506	
781	6.515	
808	6.519	
823	6.531	
833	6.523	
838	6.523	

Not shown on plot



SPECIFICATION TABLE NO. 56 THERMAL CONDUCTIVITY OF TELLURIUM

(Impurity - 0.20% each; total impurities - 0.50%)

[ For Data Reported in Figure and Table No. 56 ]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition ( weight percent)	Specifications and Remarks
1	297	L	1957	285-763	1-5			Spectrally pure; polycrystalline; bar specimen prepared by triple fractional distillation in a vacuum of $10^{-4}$ mm Hg, cold pressing under 4000 kg $cm^{-2}$ and hot pressing at 673 K under 380 kg $cm^{-2}$ for 6 hrs; melting point 452 C; measured in both solid and liquid states.
2	366	P	1916	318.5				Material supplied by Eimer and Amend, melted and cast in a hydrogen atmosphere; density 6.25 g $cm^{-3}$ ; thermal conductivity value calculated from measured thermal diffusivity using specific heat taken from literature.
3	367	L	1933	90-298				<0.01 impurities, single crystal; electrical resistivity reported as 0.021 and 0.035 ohm cm at 90 and 298 K, respectively; heat flow parallel to crystal axis.
4	367	L	1933	90-299				<0.01 impurities; polycrystalline; cast; electrical resistivity 0.030 ohm cm at 296 K.
5	367	L	1933	90-296				<0.01 impurities; polycrystalline; cast; electrical resistivity 0.200 ohm cm at 296 K.
6	368	L,C	1954	298.2	0.2-3			15 mm dia disk; electrical resistivity reported as 0.235 ohm cm at 298 K; concentration of current carriers = $3.46 \times 10^{18} cm^{-3}$ .
7	369		1959	118-462		1		Single crystal; tempered in vacuum for 24 hrs at 673 K; heat flow parallel to principal crystal axis.
8	369		1959	122-622		2		Single crystal; heat flow parallel to principal crystal axis.
9	369		1959	115-622		3		Single crystal; heat flow perpendicular to principal crystal axis.
10	370	I	1957	2.0-76		Te 1		~99.5 pure; polycrystalline; 5 mm dia, 1.5 cm long; broken from a longer rod; supplied by Messrs. A. D. Mackay, Inc.
11	370	L	1957	2.0-92		Te 2		99.99 pure; polycrystalline; individual crystals 1 or 2 mm wide and up to 1 cm long; specimen 3 mm in dia, ~5 cm long, fabricated from pure crystalline lump supplied by Messrs. A. D. Mackay, Inc., zone refined, etched, melted under vacuum in Pyrex tube and allowed to recrystallize.
12	370	L	1957	2.0-74		Te 3		Similar to above but the specimen composed of only 5 or 6 crystals of larger size.
13	370	L	1957	3.0-82		Te 5		The above specimen annealed for about 5 days at a temp just below the melting point, then cooled slowly for 24 hrs to produce a single crystal; crystallographic axis at about 80° to the axis of cylindrical specimen.
14	371	L	1959	85-472		I		Single crystal; 0.72 x 1.06 x 1.95 cm; hole concentration $1 \times 10^{18} cm^{-3}$ ; heat flow in direction of main crystallographic axis.
15	371	L	1959	83-471		II		Single crystal; 0.48 x 0.84 x 1.67 cm; prepared by recrystallization of Te distilled two or three times, slow cooling in a sealed evacuated ampule made of high melting-point glass; hole concentration $9 \times 10^{18} cm^{-3}$ ; heat flow in direction of main crystallographic axis.
16	371	L	1959	80-375		III		Sb-doped; single crystal; 0.77 x 0.80 x 2.03 cm; prepared in same manner as the above specimen, except some antimony added to the twice-distilled tellurium; hole concentration $5 \times 10^{18} cm^{-3}$ .



SPECIFICATION TABLE NO. 5C (continued)

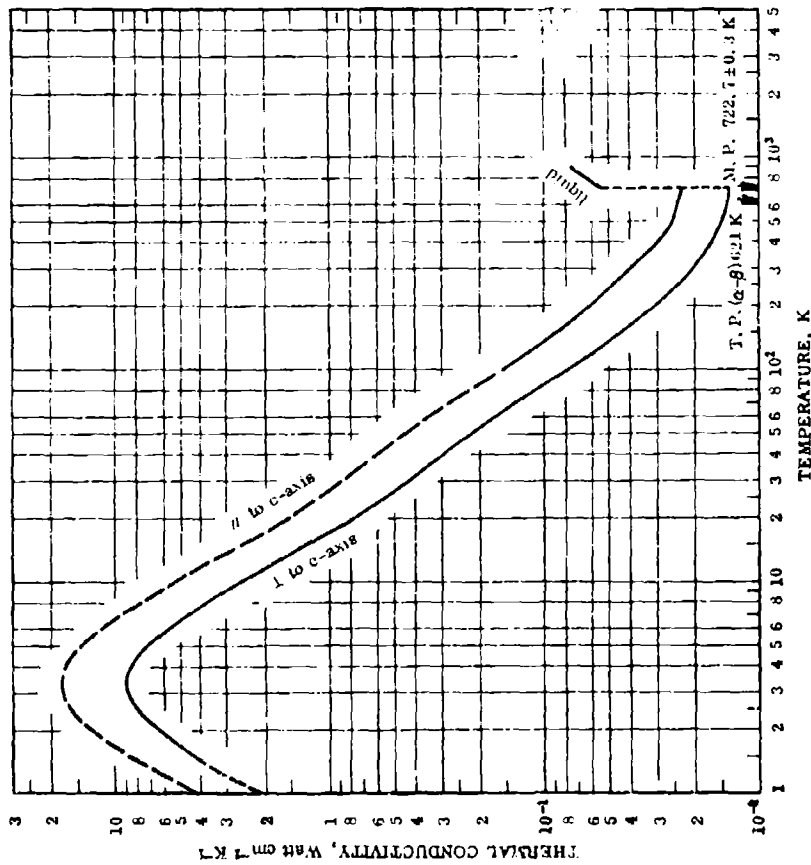
Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
17	801	L	1962	2.6-35		No. 1	Single crystal; 28 x 0.31 x 0.3 cm; specimen axis along principal crystal axis; prepared by zone melting; annealed for 70 hrs at 593 K, etched in SR-4; carrier concentration $3 \times 10^{14} \text{ cm}^{-3}$ at 77 K.
18	801	L	1962	2.4-20		No. 2	Single crystal; 17 x 0.36 x 0.33 cm; specimen axis along principal crystal axis; prepared by zone melting; etched in HNO <sub>3</sub> ; carrier concentration $7 \times 10^{14} \text{ cm}^{-3}$ at 77 K.
19	802	-	1962	696-750			Pure; liquid specimen contained in a sealed Pyrex glass vessel; electrical resistivity reported as 0.77, 0.68, 0.58, 0.49, 0.48, 0.45, 0.44, and 0.41 milliohm cm at 394, 404, 433, 469, 472, 496, 502, and 528 C, respectively; data corrected for conduction of heat through the current lead.
20	803	L	1963	84-280	5-8	No. 1	Single crystal; specimen cut from single crystal ingot obtained by slow cooling of molten tellurium in a sealed evacuated ampoule; annealed in sealed ampoule for 90 hrs at 613 K; hole concentration $\sim 2 \times 10^{18} \text{ cm}^{-3}$ at 80 K; measured under atm. of argon, with heat flow along the c-axis.
21	803	C	1963	320-660	5-8	No. 1	The above specimen measured by a comparative method using fused quartz as comparative material.
22	804	C	1961	100-485			99.6 pure; polycrystalline; specimen 0.3 cm long and 0.5 cm in dia; hole concentration $3 \times 10^{18} \text{ cm}^{-3}$ (as calculated from Hall effect); brass (38.5 Zn, 61.5 Cu) used as comparative material; data taken from smoothed curve of 4 measurements.
23	914	L	1966	743-823			99.995 pure; molten specimen contained in a short cylindrical cell.
24	961	L	1967	1.8-98			Single crystal; specimen 2.09 mm <sup>2</sup> in cross section and 1.45 mm long; heat flow parallel to the c-axis; (additional information and the tabulated data obtained from author).
25	961	L	1967	1.7-101			Single crystal; specimen 2.418 mm <sup>2</sup> in cross section and 1.55 mm long; heat flow perpendicular to the c-axis; (additional information and the tabulated data obtained from author).



DATA TABLE NO. 56 (continued)

T	k	T	k	T	k
<u>CURVE 19</u>					
696.2	0.340	743	0.056	79.9	0.133
710.2	0.2150	793	0.065	90.8	0.119
723.2	0.200	833	0.072	98.3	0.114
723.2	0.295	892	0.076	<u>CURVE 25</u>	
728.2	0.215	923	0.076	1.66	0.918
750.2	0.180	<u>CURVE 24</u>		1.82	1.04
<u>CURVE 20</u>					
84	0.105	1.78	1.42	1.99	1.19
92	0.0996	1.98	1.82	2.28	1.45
100	0.0837	2.27	2.38	2.5	1.65
104	0.0805	2.58	3.10	2.74	2.0
120	0.0721	2.97	3.64	3.25	2.3
142	0.0598	3.40	4.35	3.38	2.45
200	0.0440	3.75	4.68	3.75	2.55
215	0.0427	4.20	4.89	4.05	2.8
230	0.0380	4.60	5.07	4.60	2.75
<u>CURVE 21</u>					
320	0.0322	4.89	5.12	4.65	3.05
340	0.0305	5.10	5.08	5.1	2.65
390	0.0279	5.25	5.02	6.1	2.55
420	0.0262	6.02	4.78	6.95	2.22
440	0.0249	6.09	4.79	8.1	1.63
480	0.0226	6.52	4.51	9.55	1.38
520	0.0218	7.12	4.14	10.1	1.06
560	0.0215	8.10	3.62	13.3	0.77
610	0.0215	9.04	2.88	15.6	0.58
660	0.0218	10.8	2.25	18.0	0.480
<u>CURVE 22</u>					
100	0.0251	12.1	1.81	20.3	0.405
150	0.0213	13.6	1.48	24.1	0.295
200	0.0169	15.1	1.26	28.2	0.235
250	0.0153	16.2	1.16	28.2	0.245
300	0.0167	18.1	0.93	34.3	0.178
350	0.0192	20.8	0.78	39.1	0.153
400	0.0226	22.5	0.65	45.4	0.123
450	0.0285	22.8	0.62	52.5	0.095
485	0.0359	23.3	0.54	70.3	0.073
		28.5	0.445	84.9	0.062
		34.3	0.337	95.0	0.059
		36.6	0.31	101.0	0.057
		39.3	0.287		
		42.8	0.255		
		47.1	0.225		
		49.3	0.216		
		52.5	0.196		
		55.6	0.177		
		71.2	0.143		

FIGURE AND TABLE NO. 56R RECOMMENDED THERMAL CONDUCTIVITY OF TELLURIUM



REMARKS

The recommended values are for 99.99% pure tellurium. The recommended values that are supported by experimental thermal conductivity data are thought to be accurate to within 5% of the true values near room temperature and 5 to 10% at other temperatures above 10 K. The thermal conductivity near and below the corresponding temperature of its maximum is highly sensitive to small physical and chemical variations of the specimens, and the values below 10 K are intended as typical values for indicating the general trend.

\*  $T_1$  in K,  $k_1$  in  $\text{Watt cm}^{-1} \text{K}^{-1}$ ,  $T_2$  in F, and  $k_2$  in  $\text{Btu hr}^{-1} \text{ft}^{-1} \text{F}^{-1}$ .

† Values in parentheses are extrapolated or estimated.

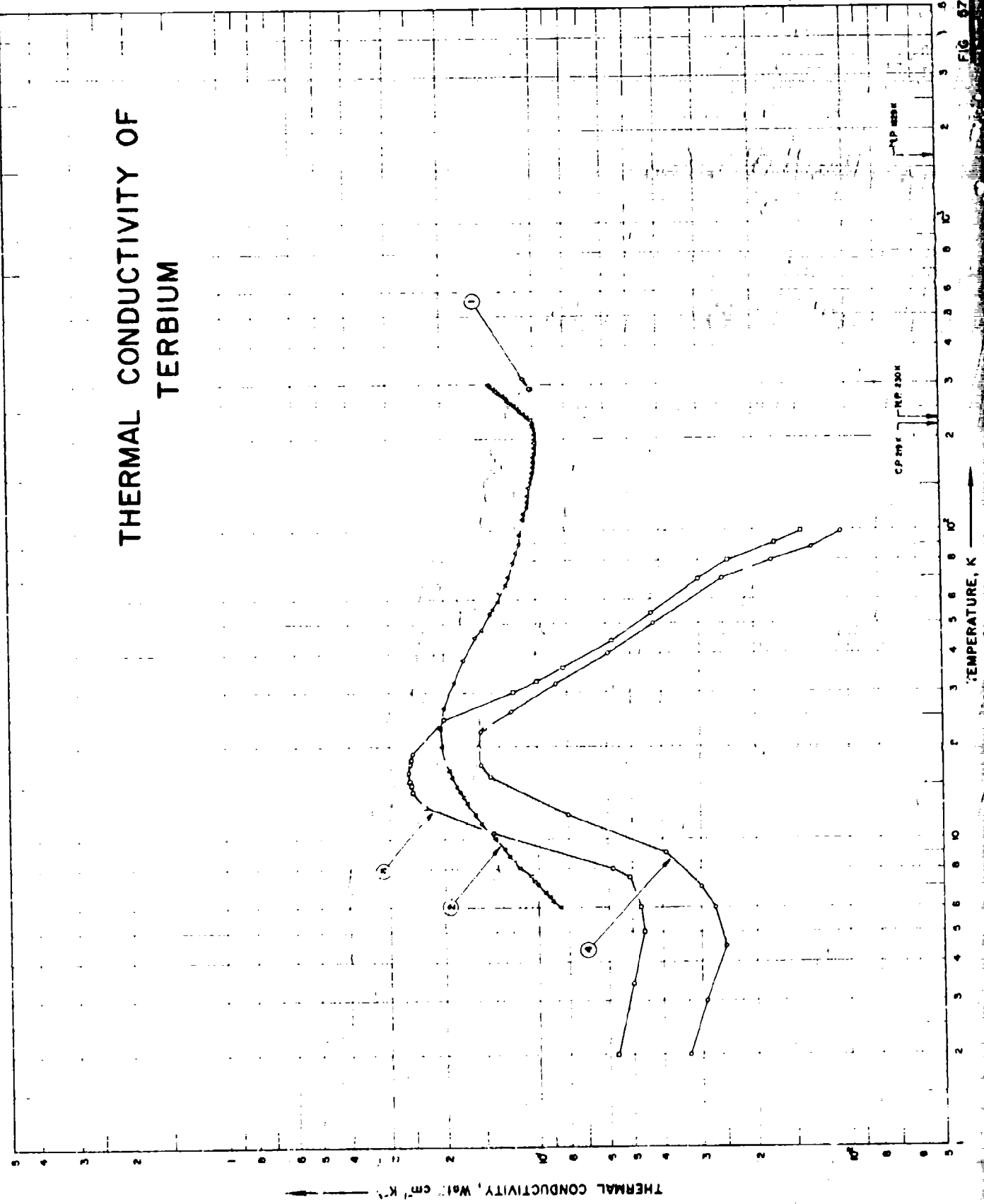
RECOMMENDED VALUES\*

$T_1$	(// to c-axis)		Single Crystal		$T_2$
	$k_1$	$k_2$	$k_1$	$k_2$	
0	0	0	0	0	-459.7
1	(4.02) †	(232)	(2.01)	(116)	-457.9
2	(12.7)	(734)	6.33	366	-456.1
3	(17.5)	(1010)	8.76	506	-454.3
4	(17.2)	(994)	8.62	498	-452.5
5	(14.4)	(832)	7.18	415	-450.7
6	(11.7)	(676)	5.83	337	-448.9
8	(7.76)	(448)	3.88	224	-445.3
10	(5.36)	(310)	2.68	155	-441.7
15	(2.96)	(198)	1.28	74.0	-432.7
20	(1.55)	(89.6)	0.776	44.8	-423.7
25	(1.13)	(63.1)	0.566	32.7	-414.7
30	(0.886)	(51.2)	0.443	25.6	-405.7
35	(0.732)	(41.9)	0.363	21.0	-396.7
40	(0.610)	(35.2)	0.305	17.6	-387.7
50	(0.454)	(26.2)	0.227	13.1	-369.7
60	(0.346)	(20.1)	0.174	10.1	-351.7
70	(0.276)	(15.9)	0.138	7.97	-333.7
80	(0.224)	(12.9)	0.112	6.47	-315.7
90	(0.187)	(10.8)	0.0937	5.41	-297.7
100	(0.159)	(9.19)	0.0795	4.59	-279.7
150	0.0893	5.16	0.0448	2.59	-189.7
200	0.0615	3.55	0.0311	1.80	-99.7
250	0.0491	2.84	0.0246	1.42	-9.7
273.2	0.0442	2.55	0.0226	1.31	32.0
300	0.0396	2.29	0.0208	1.20	80.3
350	0.0336	1.94	0.0185	1.07	170.3
400	0.0294	1.70	0.0169	0.976	260.3
500	0.0251	1.45	0.0151	0.872	440.3
600	0.0240	1.39	0.0143	0.826	620.3
700	0.0230	1.33	0.0140	0.809	800.3
722.7	0.0229	1.32	0.0140	0.809	841.2

In Liquid State

$T_1$	$k_1$	$T_2$
722.7	0.0545	841.2
900	0.0662	980.3
900	0.0759	1160

# THERMAL CONDUCTIVITY OF TERBIUM



## SPECIFICATION TABLE NO. 57 THERMAL CONDUCTIVITY OF TERBIUM

(Impurity &lt; 0.20% each; total impurities &lt; 0.50%)

[ For Data Reported in Figure and Table No. 57 ]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent). Specifications and Remarks
1	777	C	1965	291.2	±3.0		High purity; polycrystalline; specimen 0.25 in. in dia and 0.25 in. long; supplied by Johnson Matthey Co.; electrical resistivity reported as 119 $\mu\text{ohm cm}$ at about 18 C; monel metal used as comparative material; measurement made using 2 different thermal comparators.
2	816	L	1964	6.0-300			0.08 O, 0.06 Y, 0.01 Cu, 0.01 Si, and 0.003 Mg; polycrystalline; specimen 0.476 cm in dia and 6 cm long; supplied by Research Chemicals; arc-melted for 12 min, machined, swaged, heated in vacuum of $10^{-5}$ mm Hg at 790 K for 40 hrs and cooled to room temp in about 3 hrs; measured in vacuum of $6 \times 10^{-4}$ mm Hg; electrical resistivity reported as 4,851, 5,006, 5,843, 7,041, 11,196, 14,998, 19,637, 23,504, 28,815, 41,320, 57,922, 85,109, 105,708, 112,480, 116,077, 117,000, 119,308, 121,942, and 123,686 $\mu\text{ohm cm}$ at 4.18, 12.62, 20.09, 25.37, 37.21, 45.71, 55.37, 63.13, 73.61, 97.72, 128.29, 175.61, 212.73, 223.27, 231.13, 241.45, 264.35, 287.39, and 303.59 K, respectively; ferromagnetic-antiferromagnetic and antiferromagnetic-paramagnetic transition occurred at 219 K and 230 K respectively.
3	814		1966	2.0-99		Tb 1	99.9 pure; specimen 0.25 mm in dia; baked for 1.5 hrs at 650 C; measured in helium atmosphere; electrical resistivity 4.13 $\mu\text{ohm cm}$ at 4.2 K; electrical resistivity ratio $\rho(293\text{K})/\rho(4.2\text{K}) = 30$ ; data taken from smoothed curve.
4	814		1966	2.0-99		Tb 2	99.9 pure; specimen 0.25 mm in dia; baked for 1.5 hrs at 650 C; measured in helium atmosphere; electrical resistivity 7.90 $\mu\text{ohm cm}$ at 4.2 K; electrical resistivity ratio $\rho(293\text{K})/\rho(4.2\text{K}) = 15.6$ ; data taken from smoothed curve.

DATA TABLE NO. 57 THERMAL CONDUCTIVITY OF TERBREM

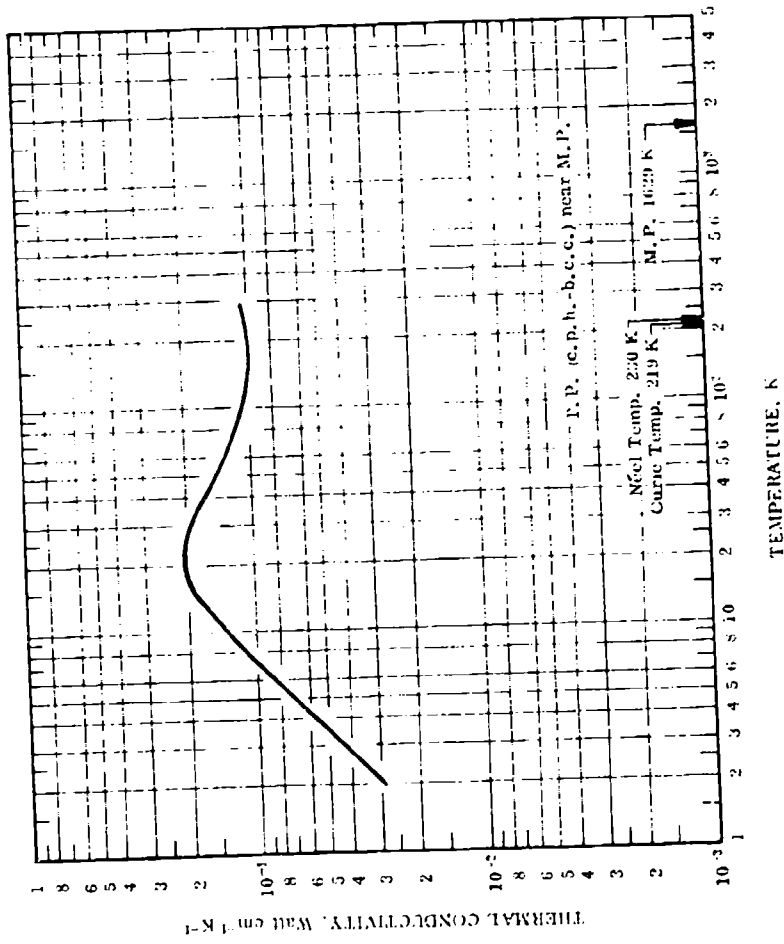
(Impurity - 0.20% each; total impurities - 0.50%)

(Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup>K<sup>-1</sup>)

T	k	T	k	T	k	T	k		
<u>CURVE 1</u>									
291.2	0.102	79.7	0.118	235.2	0.105	69.5	0.093		
291.2	0.105	83.8	0.116	236.0	0.105	80.0	0.024		
<u>CURVE 2</u>									
6.02	0.0854	91.0	0.113	240.2	0.107	91.0	0.017		
6.28	0.0900	97.7	0.112	245.1	0.110	99.4	0.014		
6.48	0.0912	103.5	0.111	250.1	0.112	<u>CURVE 4</u>			
6.68	0.0949	109.0	0.109	255.0	0.115	2.0	0.033		
7.11	0.101	111.6	0.108	255.7	0.115	3.0	0.029		
7.25	0.103	114.5	0.108	260.6	0.118	4.5	0.025		
7.56	0.106	120.6	0.105	265.6	0.121	6.1	0.027		
7.66	0.107	131.9	0.105	273.5	0.123	7.9	0.030		
8.02	0.114	138.2	0.104	275.5	0.126	9.0	0.039		
8.36	0.119	144.1	0.103	280.4	0.129	12.0	0.080		
8.79	0.124	149.7	0.102	285.3	0.132	16.7	0.143		
9.26	0.128	156.1	0.101	293.2	0.134	17.5	0.153		
10.0	0.137	162.2	0.100	295.1	0.137	20.0	0.155		
10.6	0.145	168.2	0.100	300.0	0.139	22.5	0.152		
11.3	0.153	174.2	0.100	<u>CURVE 3</u>				23.0	0.148
12.0	0.159	180.4	0.0994	2.0	0.0565	26	0.121		
13.2	0.169	181.9	0.0998	3.4	0.050	32	0.087		
13.7	0.173	184.3	0.0997	5.0	0.046	40	0.059		
14.2	0.177	185.7	0.0994	6.0	0.047	50	0.042		
14.8	0.182	187.4	0.0997	7.5	0.051	70	0.025		
15.1	0.183	194.0	0.0994	8.0	0.048	80	0.0175		
15.9	0.189	198.9	0.0999	10.5	0.140	88	0.013		
16.7	0.193	206.3	0.0996	12.7	0.230	99	0.0105		
18.0	0.200	212.5	0.0999	14.3	0.250				
20.1	0.203	213.7	0.100	15.0	0.253				
22.9	0.205	214.0	0.100	15.5	0.255				
26.6	0.200	215.2	0.0999	16.6	0.257				
32.4	0.185	217.5	0.100	17.6	0.255				
38.3	0.173	219.2	0.100	18.2	0.253				
43.2	0.158	220.9	0.100	19.0	0.250				
48.0	0.150	222.7	0.100	24.5	0.20				
54.2	0.141	223.1	0.100	30.0	0.119				
56.0	0.137	224.4	0.100	32.5	0.10				
59.3	0.133	225.1	0.101	36.0	0.0825				
63.1	0.130	226.1	0.100	44.0	0.0575				
67.3	0.125	228.6	0.101	54.0	0.0425				
71.2	0.122	230.1	0.102						

Not shown on plot

FIGURE AND TABLE NO. 57R RECOMMENDED THERMAL CONDUCTIVITY OF TERBIUM



RECOMMENDED VALUES\*  
(For Polycrystalline)

T <sub>1</sub>	k <sub>1</sub>	k <sub>2</sub>	T <sub>2</sub>
0	0	0	-459.7
2	0.0290	1.68	-456.1
3	0.0431	2.49	-454.3
4	0.0572	3.31	-452.5
5	0.0712	4.11	-450.7
6	0.0852	4.92	-448.9
7	0.0988	5.71	-447.1
8	0.113	6.53	-445.3
9	0.126	7.28	-443.5
10	0.138	7.97	-441.7
11	0.149	8.61	-439.9
12	0.159	9.19	-438.1
13	0.168	9.71	-436.3
14	0.177	10.2	-434.5
15	0.183	10.6	-432.7
16	0.189	10.9	-430.9
18	0.198	11.4	-427.3
20	0.203	11.7	-423.7
25	0.203	11.7	-414.7
30	0.193	11.2	-405.7
35	0.180	10.4	-396.7
40	0.168	9.71	-387.7
45	0.157	9.07	-378.7
50	0.147	8.49	-369.7
60	0.133	7.68	-351.7
70	0.124	7.16	-333.7
80	0.117	6.76	-315.7
90	0.113	6.53	-297.7
100	0.110	6.36	-279.7
150	0.102	5.89	-189.7
200	0.099H	5.77	- 99.7
250	0.102	5.89	- 9.7
273.2	0.103	5.95	32.0
300	0.104	6.01	80.3

REMARKS

The recommended values are for well-annealed 99.84% pure terbium with residual electrical resistivity  $\rho_0 = 4.85 \mu \Omega$  cm (characterization by  $\rho_0$  becomes important at temperatures below about 200 K). The recommended values that are supported by experimental thermal conductivity data are thought to be accurate to within 5% of the true values near room temperature and 5 to 15% at other temperatures.

\* T<sub>1</sub> in K, k<sub>1</sub> in Watt cm<sup>-1</sup> K<sup>-1</sup>, T<sub>2</sub> in F, and k<sub>2</sub> in Btu hr<sup>-1</sup> ft<sup>-1</sup> F<sup>-1</sup>.



# THERMAL CONDUCTIVITY OF THALLIUM

FIGURE SHOWS ONLY 27 OF THE CURVES REPORTED IN TABLE

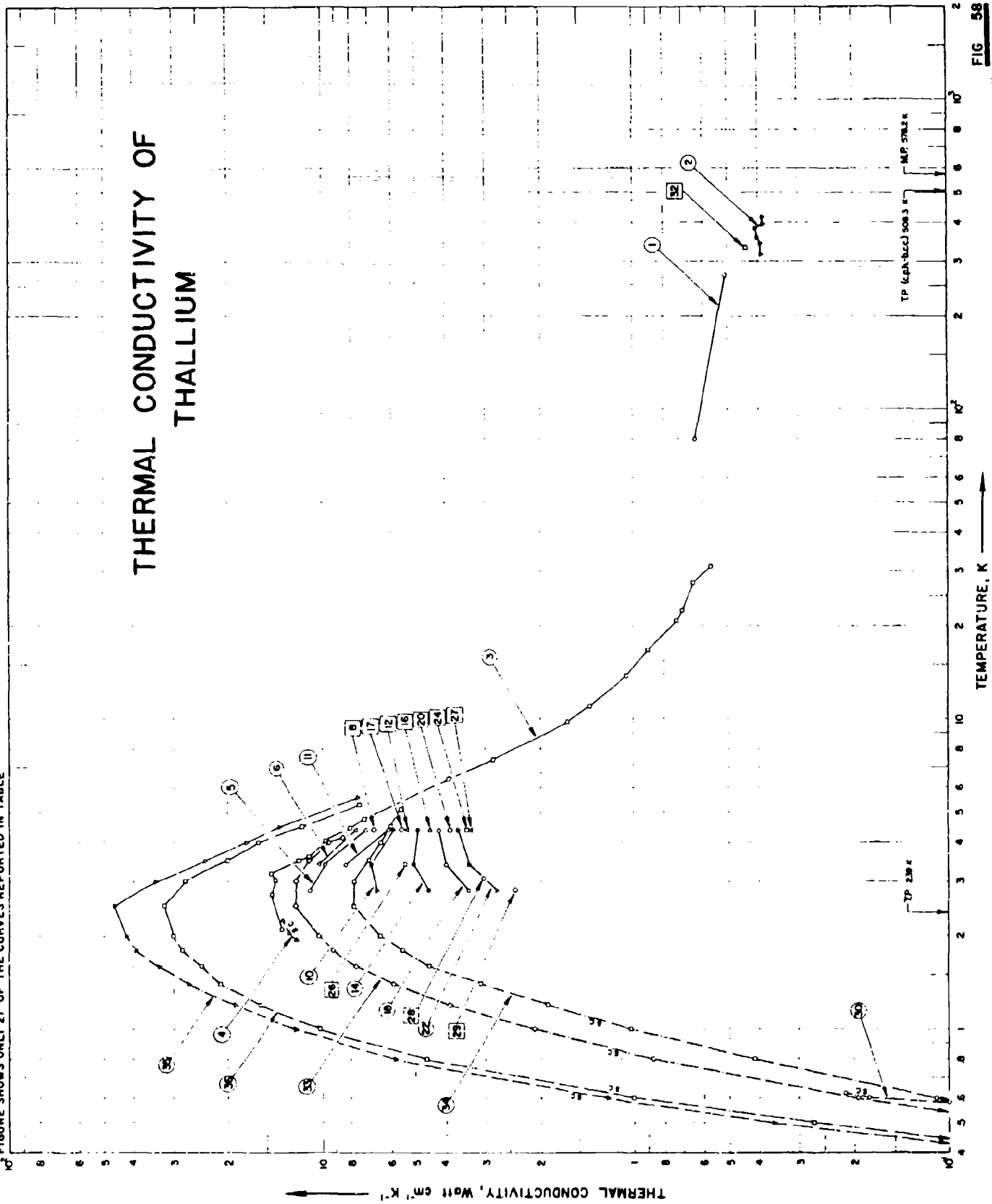


FIG 58

SPECIFICATION TABLE NO. 58 THERMAL CONDUCTIVITY OF THALLIUM

(Impurity < 0.20% each; total impurities < 0.50%)

[ For Data Reported in Figure and Table No. 58 ]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition ( weight percent),	Specifications and Remarks
1	34	L	1927	80, 273				Pure thallium (electrolytic); electrical conductivity reported as 27.8 and 6.73 x 10 <sup>4</sup> ohm <sup>-1</sup> cm <sup>-1</sup> at 80 and 273 K respectively.
2	19	L	1923	318-422				Cylindrical specimen 1.5 cm in dia and 12 cm long; melting point 302 C.
3	122	L	1955	2.1-31		JM 2544; T11		99.99 pure; polycrystalline; 2.99 cm long and 0.16 cm in dia; supplied by Johnson Mathey and Co.; annealed in vacuo for several hrs and coated with celluloid varnish to prevent oxidation; measured in a magnetic field; in normal state.
4	122	L	1955	2.0-2.3		JM 2544; T11		The above specimen in superconducting state.
5	342	L	1953	2.8-4.4		JM 2544; T11		99.99 pure; polycrystalline; 5 cm long and ~0.2 cm in dia; supplied by Johnson Mathey and Co.; measured in a transverse magnetic field of 0.34 kOe.
6	342	L	1953	3.4, 4.4		T11		The above specimen measured in a longitudinal magnetic field of 0.34 kOe.
7	342	L	1953	4.4		T11		The above specimen measured in a transverse magnetic field of 0.51 kOe.
8	342	L	1953	4.4		T11		The above specimen measured in a transverse magnetic field of 0.71 kOe.
9	342	L	1953	4.4		T11		The above specimen measured in a longitudinal magnetic field of 0.71 kOe.
10	342	L	1953	2.8-4.4		T11		The above specimen measured in a transverse magnetic field of 1.09 kOe.
11	342	L	1953	3.4, 4.4		T11		The above specimen measured in a longitudinal magnetic field of 1.09 kOe.
12	342	L	1953	4.4		T11		The above specimen measured in a transverse magnetic field of 1.42 kOe.
13	342	L	1953	4.4		T11		The above specimen measured in a longitudinal magnetic field of 1.42 kOe.
14	342	L	1953	2.8-4.4		T11		The above specimen measured in a transverse magnetic field of 1.79 kOe.
15	342	L	1953	3.4, 4.4		T11		The above specimen measured in a longitudinal magnetic field of 1.79 kOe.
16	342	L	1953	4.4		T11		The above specimen measured in a transverse magnetic field of 2.14 kOe.
17	342	L	1953	4.4		T11		The above specimen measured in a longitudinal magnetic field of 2.14 kOe.
18	342	L	1953	2.8-4.4		T11		The above specimen measured in a transverse magnetic field of 2.5 kOe.
19	342	L	1953	3.4, 4.4		T11		The above specimen measured in a longitudinal magnetic field of 2.5 kOe.
20	342	L	1953	4.4		T11		The above specimen measured in a transverse magnetic field of 2.85 kOe.
21	342	L	1953	4.4		T11		The above specimen measured in a longitudinal magnetic field of 2.85 kOe.
22	342	L	1953	2.8-4.4		T11		The above specimen measured in a transverse magnetic field of 3.22 kOe.
23	342	L	1953	3.4, 4.4		T11		The above specimen measured in a longitudinal magnetic field of 3.22 kOe.
24	342	L	1953	4.4		T11		The above specimen measured in a transverse magnetic field of 3.59 kOe.
25	342	L	1953	4.4		T11		The above specimen measured in a longitudinal magnetic field of 3.59 kOe.
26	342	L	1953	3.4		T11		The above specimen measured in a longitudinal magnetic field of 3.70 kOe.
27	342	L	1953	4.4		T11		The above specimen measured in a transverse magnetic field of 3.79 kOe.

SPECIFICATION TABLE NO. 5A (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
28	342	L	1953	3.4		Tl 1	The above specimen measured in a transverse magnetic field of 3.82 kOe.
29	342	L	1953	2.8		Tl 1	The above specimen measured in a transverse magnetic field of 3.91 kOe.
30	400	L	1953	0.29-0.63		Tl 1	The above specimen measured at low temperatures; in superconducting state; preliminary results reported.
31	412	L	1955	0.26-0.84	±5	Tl 1	More complete results from the same thallium batch (JM 2544) as the above specimen; in superconducting state.
32	230	L	1925	333			Specimen 1.9 cm in dia., 10 cm long; made from pure thallium from Eimer and Amend; electrical conductivity $5.88 \times 10^6 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 25 C.
33	712	L	1961	0.41-4.2		Tl-3	Pure; specimen 1 mm in dia; made from single crystal; specimen axis at 20 degrees with the crystal hexagonal axis; residual electrical resistivity 0.0026 $\mu\text{ohm cm}$ .
34	712	L	1951	0.19-5.2		Tl-4	Pure; specimen 1.1 mm in dia; made from single crystal; specimen axis at 80 degrees with the crystal hexagonal axis; residual electrical resistivity 0.0050 $\mu\text{ohm cm}$ .
35	712	L	1961	0.21-5.6		Tl-7	Pure; specimen 1.6 mm in dia; specimen axis at 30 degrees with the crystal hexagonal axis; residual electrical resistivity $0.00024 \pm 0.00003 \mu\text{ohm cm}$ .
36	712	L	1961	0.34-5.3		Tl-8	0.9 mm in dia; obtained by etching the above specimen; residual electrical resistivity 0.00045 $\mu\text{ohm cm}$ .

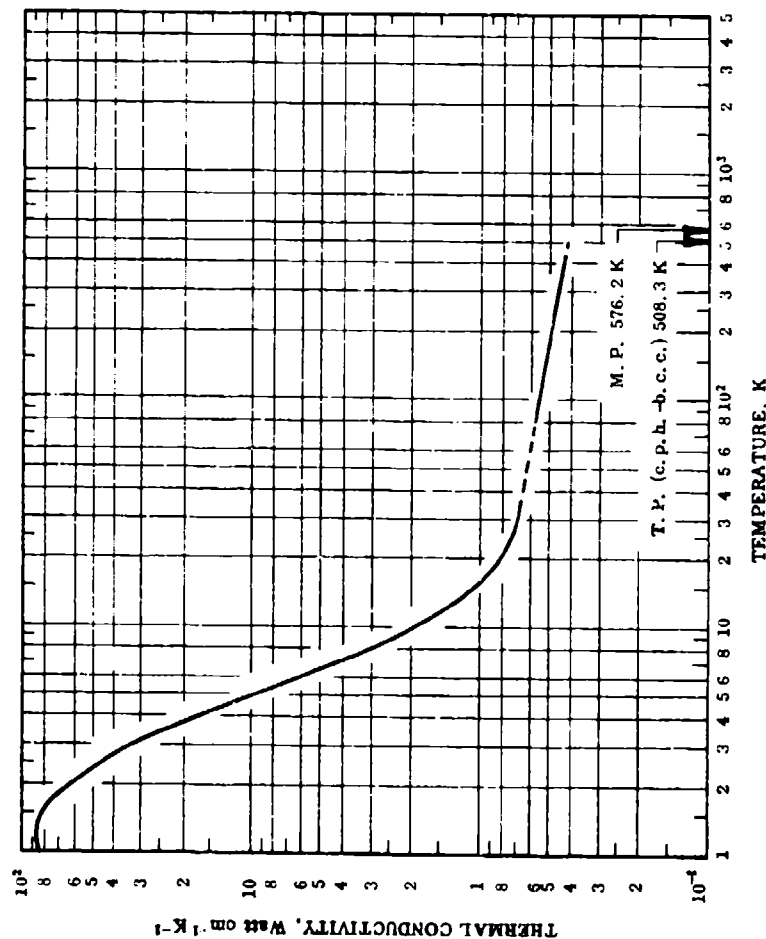
DATA TABLE NO. 58 THERMAL CONDUCTIVITY OF THALLIUM

(Impurity < 0.20% each; total impurities < 0.50%)  
 [Temperature, T, K. Thermal Conductivity, k, Watt cm<sup>-1</sup>K<sup>-1</sup>]

CURVE 1		CURVE 5		CURVE 14		CURVE 23 <sup>a</sup>		CURVE 31 <sup>a</sup>		CURVE 34		CURVE 35(cont.)	
T	k	T	k	T	k	T	k	T	k	T	k	T	k
80.0	0.635	2.8	11.0	2.8	4.55	3.4	5.58	0.255	0.00110	0.185	0.00034 <sup>a</sup>	4.5	13.5
273.0	0.506	3.4	9.82	3.4	5.09	4.4	4.86	0.275	0.00135	0.20	0.00046 <sup>a</sup>	5.6	7.60
		4.4	7.16	4.4	4.92			0.330	0.00248	0.25	0.00102		
CURVE 2		CURVE 6		CURVE 15		CURVE 24		CURVE 36		CURVE 36			
318.2	0.391	4.4	7.03	4.4	4.47	4.4	3.45	0.183	0.00378	0.30	0.00195	0.34	0.0080
345.2	0.392	3.4	10.2	3.4	7.01			0.402	0.00570	0.35	0.00430	0.40	0.0095
361.2	0.400	4.4	7.72	4.4	5.88			0.425	0.0075	0.40	0.0095	0.40	0.044
363.2	0.401 <sup>b</sup>					CURVE 25 <sup>b</sup>		0.465	0.0108	0.50	9.042	0.50	0.270
386.2	0.408					4.4	4.80	0.495	0.0127	0.60	0.110	0.50	0.270
396.2	0.385							0.507	0.0154	0.80	0.415	0.60	1.00
422.2	0.388							0.56	0.0202	1.0	1.02	0.80	4.60
								0.59	0.0215	1.2	1.90	1.0	10.2
CURVE 3		CURVE 8		CURVE 17		CURVE 26		CURVE 32		CURVE 35			
2.13	13.3	4.4	7.03	4.4	4.47	3.4	5.40	0.605	0.0205	1.4	3.10	1.2	15.7
2.73	14.4	4.4	6.79	4.4	6.55			0.645	0.0285	1.6	4.5	1.4	21.0
3.04	13.9					3.4	5.40	0.70	0.0320	1.8	5.5	1.6	24.0
3.18	14.4							0.74	0.0340	2.0	6.5	1.8	28.0
3.50	11.7					4.4	3.34	0.74	0.0495	2.5	7.8	2.0	30.0
3.60	11.0							0.84	0.0574	3.0	7.8	2.5	32.0
4.02	9.75									3.5	7.0	3.0	27.5
4.48	8.00									4.0	6.5	3.5	20.0
4.75	7.25									4.5	6.0	4.0	16.0
6.43	3.90									5.16	5.4	4.5	11.5
7.38	2.81											5.3	7.50
9.80	1.62												
11.0	1.38												
13.8	1.06												
16.7	0.895												
20.8	0.730												
22.4	0.699												
27.6	0.649												
31.0	0.568												
CURVE 4		CURVE 12		CURVE 21 <sup>b</sup>		CURVE 30 <sup>c</sup>		CURVE 33		CURVE 35			
1.95	12.0	4.4	5.32	4.4	5.05	0.290	0.0013	0.41	0.014	0.205	0.00065 <sup>a</sup>		
2.05	12.6	3.4	6.15	3.4	3.88	0.305	0.0014	0.50	0.060	0.25	0.0016		
2.25	13.1	4.4	5.92	4.4	5.32	0.323	0.00154	0.60	0.195	0.30	0.0016		
						0.340	0.0017	0.80	0.870	0.325	0.0060		
						0.363	0.00265	1.0	2.10	0.35	0.0145		
						0.390	0.0048	1.2	3.90	0.40	0.056		
						0.466	0.010	1.4	5.90	0.50	0.360		
						0.484	0.0175	1.6	7.70	0.60	1.22		
						0.512	0.033	1.8	9.20	0.80	5.8		
						0.468	0.0063	2.0	10.2	1.0	12.0		
						0.408	0.0048	2.5	12.0	1.2	19.0		
						0.466	0.010	3.0	12.0	1.4	26.5		
						0.484	0.0175	3.5	11.0	1.6	33.0		
						0.512	0.033	4.0	9.5	1.8	39.0		
						0.468	0.0063	4.15	8.5	2.0	42.0		
						0.484	0.0175	4.5	6.0	2.5	46.0		
						0.512	0.033	333.0	1.439	3.0	34.0		
						0.468	0.0063			3.5	23.5		
						0.484	0.0175			4.0	17.5		
						0.512	0.033						

<sup>a</sup> Not shown on plot

FIGURE AND TABLE NO. 568 RECOMMENDED THERMAL CONDUCTIVITY OF THALLIUM



## REMARKS

The recommended values are for well-annealed high-purity thallium with residual electrical resistivity  $\rho_0 = 0.00289 \mu\Omega/\text{cm}$  (characterization by  $\rho_0$  becomes important at temperatures below about 60 K). The values below 1.5 Tm are calculated to fit the experimental data by using  $n = 2.25$ ,  $\alpha' = 2.274 \times 10^{-3}$ , and  $\beta = 0.00982$ . The recommended values that are supported by experimental thermal conductivity data are thought to be accurate to within 5% of the true values near room temperature and 5 to 10% at other temperatures.

\*  $T_1$  in K,  $k_1$  in  $\text{Watt cm}^{-1} \text{K}^{-1}$ ,  $T_2$  in F, and  $k_2$  in  $\text{Btu hr}^{-1} \text{ft}^{-1} \text{F}^{-1}$ .

† Values in parentheses are extrapolated or interpolated.

RECOMMENDED VALUES* (For Polycrystalline)				
$T_1$	$k_1$	$k_2$	$T_2$	
0	0	0	-459.7	
1.0	82.7	4780	-457.9	
1.2	86.0	4970	-457.5	
1.5	91.9	4730	-456.9	
1.8	71.2	4110	-456.5	
2.0	63.4	3660	-456.1	
2.5	45.9	2650	-455.2	
3.0	33.2	1920	-454.3	
3.5	23.8	1380	-453.4	
4.0	17.6	1020	-452.5	
5	10.2	589	-450.7	
6	6.19	358	-448.9	
7	4.04	233	-447.1	
8	2.95	170	-445.3	
9	2.30	133	-443.5	
10	1.87	108	-441.7	
15	1.07	61.8	-432.7	
20	0.811	46.9	-423.7	
30	0.682	39.4	-405.7	
40	(0.650) †	(37.6)	-387.7	
50	(0.626)	(36.2)	-369.7	
60	(0.607)	(35.1)	-351.7	
70	(0.590)	(34.1)	-333.7	
80	0.578	33.4	-315.7	
90	0.567	32.8	-297.7	
100	0.556	32.1	-279.7	
150	0.519	30.0	-189.7	
200	0.494	28.5	-98.7	
250	0.476	27.5	-9.7	
273.2	0.469	27.1	32.0	
300	0.461	26.6	80.3	
400	0.438	25.3	260.3	
500	(0.421)	(24.3)	440.3	

# THERMAL CONDUCTIVITY OF THORIUM

FIGURE SHOWS ONLY 5 OF THE CURVES REPORTED IN TABLE

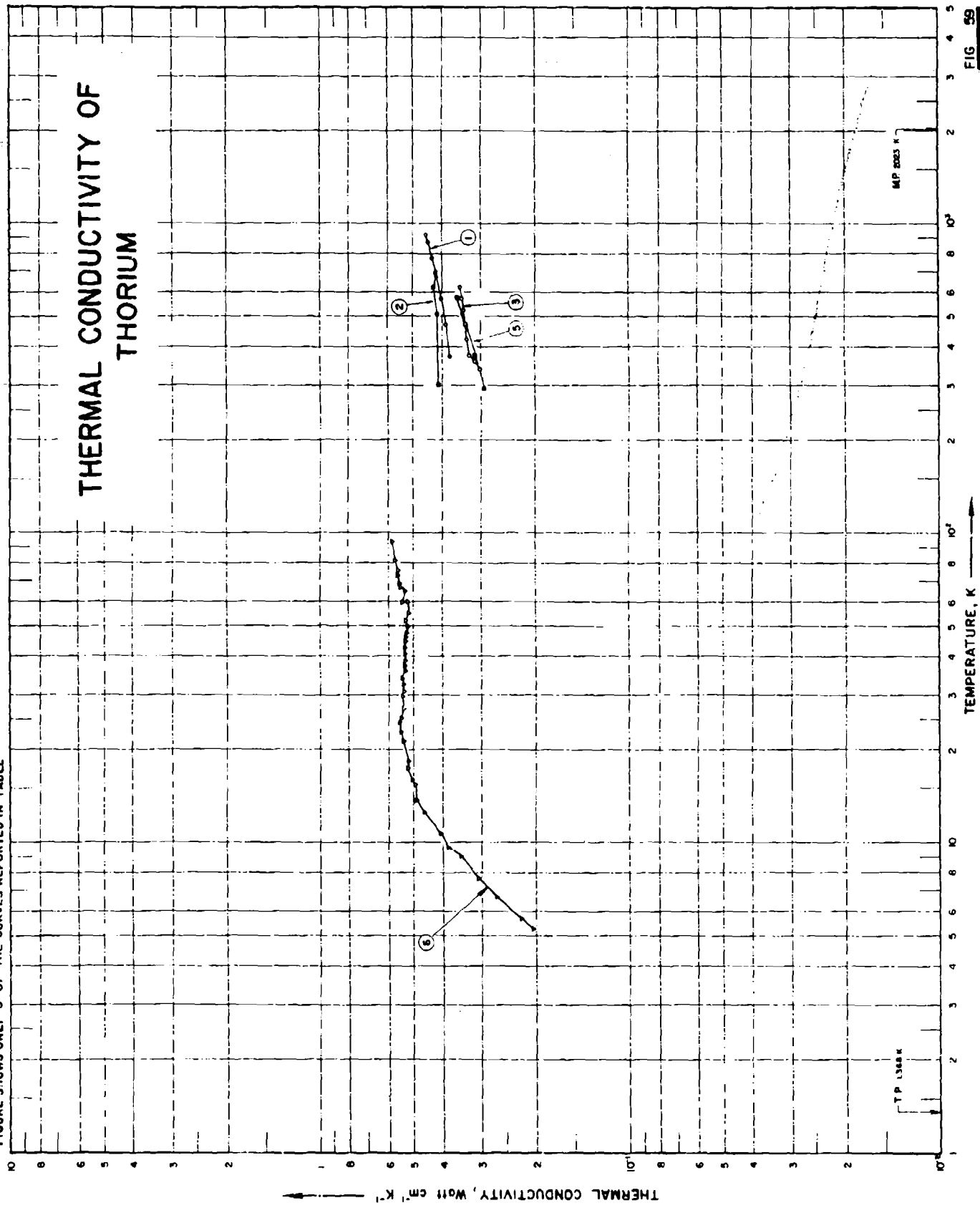


FIG 59

## SPECIFICATION TABLE NO. 59 THERMAL CONDUCTIVITY OF THORIUM

(Impurity &lt; 0.20% each; total impurities &lt; 0.50%)

[ For Data Reported in Figure and Table No. 59 ]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition ( weight percent), Specifications and Remarks
1	94, 266		1951	373-923		Ames thorium	Specimen hot rolled at 733 C and air cooled; density 11.6 g cm <sup>-3</sup> ; melting point 1680 ± 25 C; data determined in an atmosphere of purified argon.
2	130, 778	P	1951	301-697			99.85 pure; specimen 0.125 in. in dia, ~50 cm long; thermal conductivity values calculated from measured data of thermal diffusivity using density 11.558 g cm <sup>-3</sup> , and specific heat of 0.1188 joules g <sup>-1</sup> C <sup>-1</sup> (from C. F. Miller).
3	422, 239		1945	338-623			Specimen made from Ames extruded thorium.
4	235, 239	L	1944	335-367			Pure; specimen 1.618 cm long, cross sectional area 5.042 cm <sup>2</sup> .
5	423		1945	293-573			Specimen ~6 cm long, 1 cm in dia; manufactured by Westinghouse Lamp Co.; x-ray analysis after test showed the presence of thorium oxide (probably formed during the test); electrical resistivity reported as 27.5 and 32 μ ohm cm at 20 and 100 C, respectively; Lorenz function 2.76 and 2.68 x 10 <sup>-8</sup> V <sup>2</sup> K <sup>-1</sup> at 20 and 100 C, respectively.
6	935	L	1965	5.3-94	5		Cylindrical specimen 4 mm in dia and 30 mm long; supplied by Dr. J. A. Lee, A. E. R. F. Harwell; machined from an ingot of argon-arc-melted van Arkel metal of high purity; electrical resistivity ratio ρ(273 K)/ρ(4.2 K) = 20.49; electrical resistivity reported as 0.75, 0.91, 1.26, 1.82, 2.39, 2.97, 4.06, 5.15, 14.7, and 15.3 μ ohm cm at 10, 20, 30, 40, 50, 60, 80, 100, 273.15, and 295 K, respectively; residual electrical resistivity 0.72 μ ohm cm; Lorenz function reported as 2.85, 2.87, 2.45, 2.28, 2.24, 2.49, 2.82, and 3.02 x 10 <sup>-8</sup> V <sup>2</sup> K <sup>-1</sup> at 5, 15, 18, 25, 30, 50, 75, and 100 K, respectively; thermal conductivity data averaged from the results of several separate runs.

DATA TABLE NO. 59 THERMAL CONDUCTIVITY OF THORIUM

(Impurity - 0.20% each; total impurities - 0.50%)

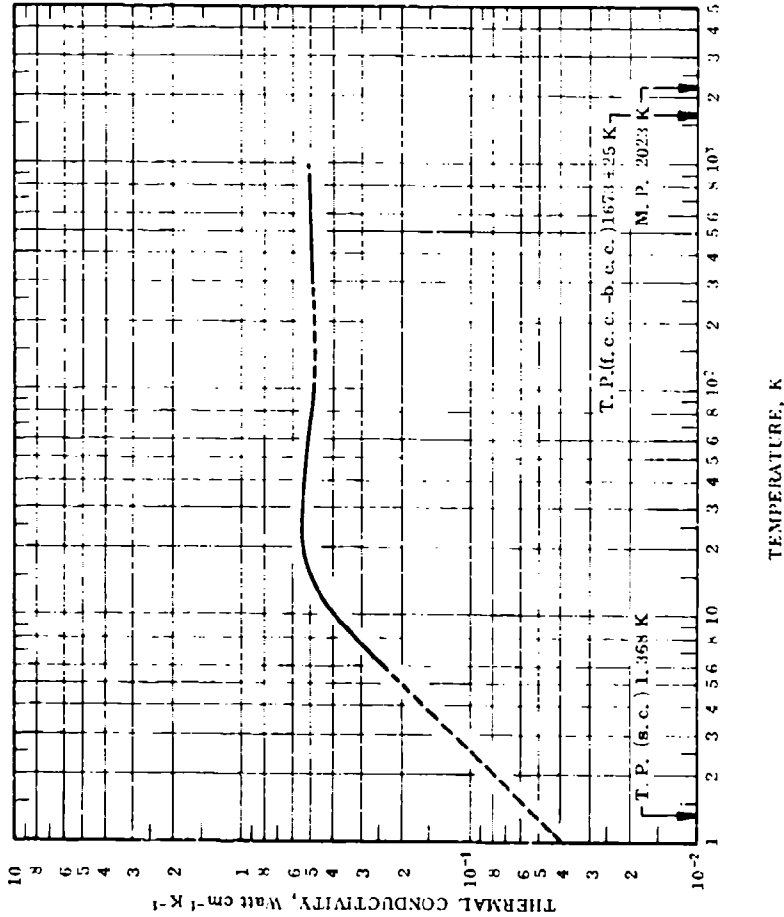
[Temperature: T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup> K<sup>-1</sup>]

CURVE 1		CURVE 5		CURVE 6 (cont.)	
T	k	T	k	T	k
373.2	0.377	5.3	0.205	73.1	0.562
473.2	0.389	5.7	0.223	76.0	0.564
573.2	0.402	6.7	0.263	81.8	0.570
673.2	0.418	7.7	0.308	82.6	0.574
773.2	0.431	9.1	0.349	94.0	0.584
873.2	0.444	9.7	0.382		
923.2	0.452	10.7	0.405		
		12.6	0.459		
		13.8	0.483		
		13.9	0.491		
		15.5	0.493		
301.2	0.412	16.1	0.502		
510.7	0.416	17.5	0.523		
623.7	0.428	18.4	0.518		
697.0	0.420	18.4	0.530		
		21.3	0.538		
		21.3	0.542		
		22.8	0.548		
338.2	0.301	24.5	0.551		
356.7	0.314	25.2	0.545		
373.2	0.326	27.0	0.539		
423.2	0.331	29.7	0.542		
473.2	0.336	30.9	0.535		
523.2	0.340	32.5	0.538		
573.2	0.345	34.0	0.542		
623.2	0.350	35.9	0.528		
		37.4	0.532		
		38.7	0.531		
		40.8	0.531		
331.5	0.307	42.6	0.531		
344.0	0.313	44.6	0.529		
347.3	0.309	46.0	0.529		
351.0	0.311	47.5	0.524		
352.7	0.313	50.0	0.522		
358.2	0.318	50.0	0.517		
366.5	0.318	52.3	0.529		
		56.2	0.517		
		59.9	0.510		
293.2	0.293	59.9	0.543		
373.2	0.311	65.2	0.531		
473.2	0.335	66.7	0.547		
573.2	0.356	69.0	0.553		

\* Not shown on plot



FIGURE AND TABLE NO. S9R RECOMMENDED THERMAL CONDUCTIVITY OF THORIUM



RECOMMENDED VALUES\*

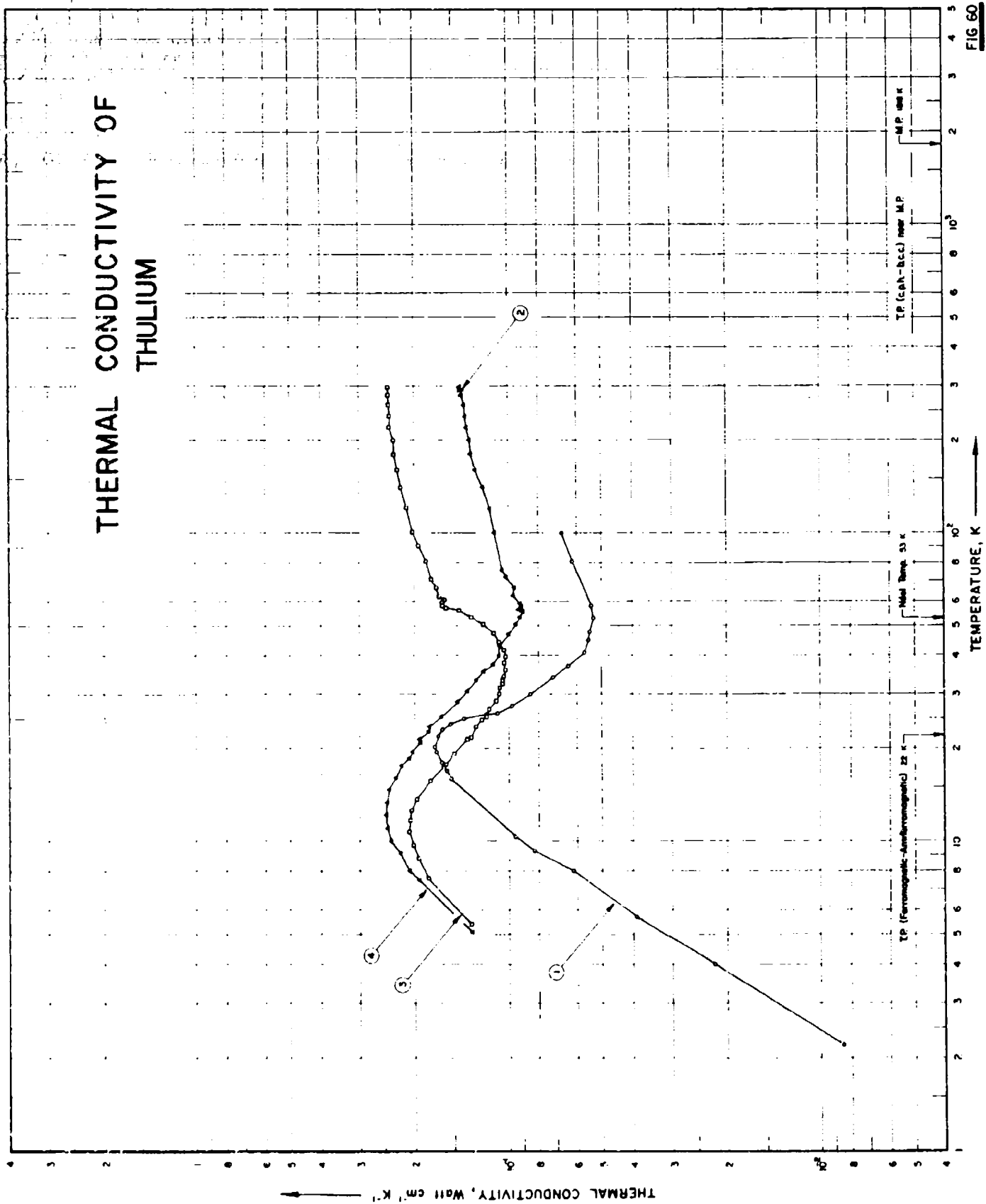
T <sub>1</sub>	k <sub>1</sub>	k <sub>2</sub>	T <sub>2</sub>	T <sub>1</sub>	k <sub>1</sub>	k <sub>2</sub>	T <sub>2</sub>
0	0	0	-459.7	500	0.498	28.8	440.3
1	(0.0304) †	(2.28)	-457.9	600	0.501	28.9	620.3
2	(0.0745)	(4.54)	-456.1	700	0.504	29.1	800.3
3	(0.118)	(6.82)	-454.3	800	0.508	29.4	980.3
4	(0.156)	(9.13)	-452.5	900	0.511	29.5	1160
5	(0.197)	(11.4)	-450.7	1000	(0.515)	(29.8)	1340
6	0.237	13.7	-448.9				
7	0.276	15.9	-447.1				
8	0.314	18.1	-445.3				
9	0.352	20.3	-443.5				
10	0.387	22.4	-441.7				
11	0.419	24.2	-439.9				
12	0.447	25.8	-438.1				
13	0.470	27.2	-436.3				
14	0.488	28.2	-434.5				
15	0.499	28.8	-432.7				
16	0.511	29.5	-430.9				
18	0.526	30.4	-427.3				
20	0.538	31.1	-423.7				
25	0.548	31.7	-414.7				
30	0.547	31.6	-405.7				
35	0.542	31.3	-396.7				
40	0.537	31.0	-387.7				
45	0.532	30.7	-378.7				
50	0.527	30.5	-369.7				
60	0.515	29.8	-351.7				
70	0.504	29.1	-333.7				
80	0.497	28.7	-315.7				
90	0.492	28.4	-297.7				
100	(0.488)	(28.2)	-279.7				
150	(0.486)	(28.1)	-189.7				
200	(0.483)	(28.2)	-99.7				
250	(0.483)	(28.3)	-9.7				
273.2	(0.450)	(28.3)	32.0				
300	0.491	28.4	80.3				
350	0.493	28.5	170.3				
410	0.495	28.6	260.3				

REMARKS

The recommended values are for well-annealed high-purity thorium with residual electrical resistivity  $\rho_0 = 0.72 \mu\Omega \text{ cm}$  (characterized by  $\rho_0$  becomes important below room temperature). The recommended values that are supported by experimental thermal conductivity data are thought to be accurate to within 10 to 15% of the true values.

\* T<sub>1</sub> in K, k<sub>1</sub> in Watt cm<sup>-1</sup> K<sup>-1</sup>, T<sub>2</sub> in F, and k<sub>2</sub> in Btu hr<sup>-1</sup> ft<sup>-1</sup> F<sup>-1</sup>. † values in parentheses are extrapolated or estimated.

# THERMAL CONDUCTIVITY OF THULIUM



## SPECIFICATION TABLE NO. 60 THERMAL CONDUCTIVITY OF THULIUM

(Impurity &lt; 0.20% each; total impurities &lt; 0.50%)

[For Data Reported in Figure and Table No. 60]

Curve No.	Ref. No.	Method Used	Year	T Range	Reported Error, %	Name and Specimen Designation	Composition (weight percent)	Specifications and Remarks
1	994	L	1965	2.2-100			95.99 pure; polycrystalline; strip specimen 0.25 mm thick; annealed in a stream of helium vapor at 650 C for 3 hrs; electrical resistivity reported as 12.7 and 79 $\mu\text{ohm cm}$ at 4.2 and 293 K, respectively; antiferromagnetic-paramagnetic transition at 53 K; Lorenz function reported as $7.30 \times 10^{-8} \text{ V}^2 \text{ K}^{-2}$ in the residual resistance region.	
2	256	C	1966	291	±4		< 0.1 rare earth metal and ~0.01 base metals; polycrystalline; $1.2 \times 1.2 \times 0.31 \text{ cm}$ ; electrical resistivity 72 $\mu\text{ohm cm}$ at 291 K; measurements made using 2 different thermal comparators.	
3	3	L	1967	5.1-294	6-4		< 0.0200 Hg, < 0.0200 Lu, < 0.0200 Y, 0.0065 O, < 0.0060 Al, < 0.0060 Si, < 0.0050 Fe, < 0.0050 Ni, < 0.0030 Er, < 0.0020 Ca, < 0.0020 Cr, < 0.0010 Mg, < 0.0010 Zr, 0.0001 H, trace of Mn, and faint trace of Dy; single crystal; $7.23 \times 1.250 \times 1.224 \text{ mm}$ ; arc-melted ingot suspended and sealed in helium at 0.5 atm in a tantalum bomb, fired in an annealing furnace consisting of a temperature gradient region ( $25 \text{ C cm}^{-1}$ ) and a constant temperature region, annealed at 1200 C for 12 hrs and at 1300 C for 72 hrs in the gradient region, then annealed at 1425 C for 16 hrs in the constant temperature region, cut and hand-lapped to size; electrical resistivity reported as 1.730, 1.750, 1.733, 1.746, 1.798, 2.461, 3.633, 5.522, 7.703, 10.35, 15.98, 21.97, 29.97, 31.11, 32.76, 36.23, 39.72, 49.44, 53.25, 76.20, and 88.12 $\mu\text{ohm cm}$ at 2.7, 4.2, 6.0, 8.0, 10.0, 16.0, 20.2, 24.5, 28.3, 32.1, 39.3, 46.3, 54.0, 56.5, 60.3, 70.9, 82.7, 120.3, 180.7, 241.6, and 298.9 K, respectively; electrical resistivity ratio $\rho_{300\text{K}}/\rho_{4.2\text{K}} = 51.0$ ; residual electrical resistivity 1.73 $\mu\text{ohm cm}$ ; Lorenz function reported as 4.08, 4.38, 3.45, 3.70, 4.22, 4.66, 4.88, 5.14, 5.20, 5.12, 4.78, 4.49, 4.26, and $4.17 \times 10^{-8} \text{ V}^2 \text{ K}^{-2}$ at 5.1, 9.1, 17.4, 22.6, 31.1, 43.3, 55.2, 59.2, 63.9, 78.4, 135.0, 200.5, 262.2, and 300.0 K, respectively; heat flow along b-axis.	
4	4	L	1967	5.4-299			< 0.0200 Hg, < 0.0200 Lu, < 0.0200 Y, 0.0100 O, < 0.0060 Al, < 0.0060 Si, < 0.0050 Fe, < 0.0050 Ni, < 0.0030 Er, < 0.0020 Ca, < 0.0020 Cr, < 0.0010 Mg, < 0.0010 Yb, 0.0001 H, < 0.0003 N, traces of Cu and Dy, and faint traces of Mn and W; single crystal; $7.24 \times 1.450 \times 1.166 \text{ mm}$ ; same fabrication method as above; electrical resistivity reported as 3.641, 3.647, 3.708, 4.161, 5.538, 8.218, 13.94, 19.37, 24.02, 25.81, 25.85, 25.72, 24.35, 21.24, 19.90, 18.60, 17.80, 17.81, 18.21, 19.68, 20.51, 25.33, 33.93, and 47.05 $\mu\text{ohm cm}$ at 1.4, 4.2, 8.0, 12.0, 16.0, 20.0, 26.0, 31.1, 37.6, 44.1, 45.4, 47.1, 51.2, 54.9, 56.0, 56.8, 57.5, 57.9, 61.9, 73.6, 90.1, 120.8, 195.0, and 300.7 K, respectively; electrical resistivity ratio $\rho_{300\text{K}}/\rho_{4.2\text{K}} = 12.9$ ; residual electrical resistivity 3.65 $\mu\text{ohm cm}$ ; Lorenz function reported as 8.87, 8.98, 7.44, 5.98, 5.96, 6.04, 6.75, 6.20, 5.66, 4.81, 4.96, 4.83, 4.68, 4.21, 3.99, and $3.77 \times 10^{-8} \text{ V}^2 \text{ K}^{-2}$ at 4.5, 6.4, 10.7, 15.6, 18.0, 21.8, 31.8, 44.9, 52.3, 56.6, 59.7, 67.4, 86.7, 160.6, 220.2, and 300.1 K, respectively; heat flow along c-axis.	

Edwards, D. W. and Legyold, S., "Transport Properties of Thulium Single Crystal," to be published in Physical Review; also USAFC IS-T-173, 1967.

DATA TABLE NO. 60 THERMAL CONDUCTIVITY OF THULIUM

(purity < 0.20% each; total impurities < 0.50%)

[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup> K<sup>-1</sup>]

CURVE 1			CURVE 3 (cont.)			CURVE 4			CURVE 4 (cont.)		
T	k		T	k		T	k		T	k	
2.2	0.0085		10.0	0.238		5.4	0.132		121.0	0.211	
4.0	0.022		11.1	0.245		7.6	0.181		140.7	0.220	
5.7	0.039		12.2	0.247		8.8	0.185		160.6	0.226	
8.0	0.052		13.4	0.246		9.7	0.203		180.9	0.231	
9.3	0.083		14.7	0.242		10.7	0.209		201.0	0.232	
10.3	0.096		16.1	0.230		11.7	0.207		221.1	0.239	
16.0	0.153		17.6	0.221		12.6	0.206		240.9	0.239	
17.0	0.158		18.6	0.209		13.7	0.197		261.0	0.240	
17.5	0.162		19.6	0.204		15.7	0.179		280.8	0.241	
18.0	0.164		20.9	0.192		17.8	0.159		298.6	0.241	
19.5	0.170		21.5	0.194		19.4	0.149				
20.3	0.172		22.7	0.181		21.5	0.136				
22.0	0.168		23.6	0.180		21.7	0.132				
23.0	0.163		25.5	0.165		23.6	0.128				
24.0	0.154		28.4	0.146		24.7	0.123				
24.5	0.147		30.7	0.136		25.5	0.118				
26.0	0.139		33.4	0.127		26.8	0.116				
25.8	0.118		35.7	0.121		29.5	0.110				
26.0	0.109		37.5	0.113		30.0	0.107				
27.5	0.098		40.1	0.108		31.4	0.107				
30.0	0.085		43.3	0.108		32.4	0.105				
34.0	0.072		47.1	0.100		33.2	0.105				
37.0	0.064		50.6	0.095		34.2	0.104				
41.0	0.057		53.5	0.092		36.1	0.103				
45.0	0.055		55.8	0.090		37.9	0.104				
48.0	0.0545		56.5	0.092		39.7	0.103				
53.3	0.0530		57.8	0.091		41.7	0.104				
58.3	0.0540		59.2	0.092		44.4	0.107				
81.0	0.0620		62.8	0.097		47.3	0.112				
100.0	0.0670		66.8	0.096		50.7	0.121				
			71.8	0.102		53.6	0.132				
			75.7	0.105		56.1	0.144				
			101.2	0.111		57.4	0.158				
			120.9	0.115		58.5	0.163				
			141.1	0.121		59.6	0.163				
			161.1	0.128		61.0	0.160				
			191.1	0.132		62.4	0.165				
			200.9	0.133		66.6	0.170				
			220.8	0.136		70.5	0.176				
			240.9	0.137		86.5	0.194				
			260.8	0.138		90.7	0.194				
			280.8	0.141		100.7	0.202				
			298.5	0.143							

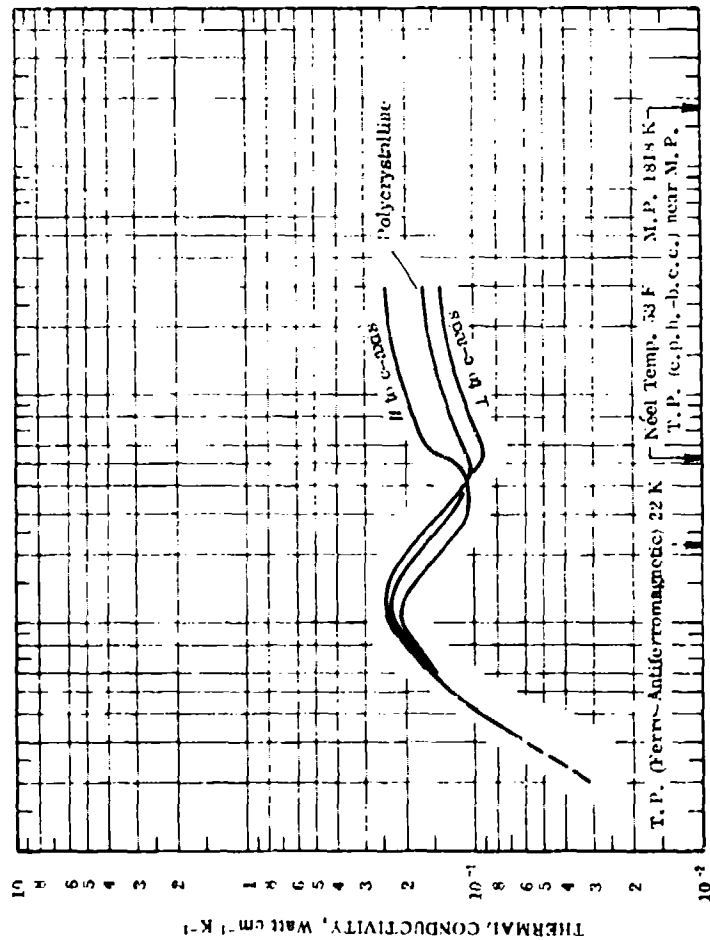
CURVE 2

291	0.140
291	0.141

CURVE 3

5.1	0.133
7.5	0.135
8.0	0.208
9.1	0.223

FIGURE AND TABLE NO. 60R RECOMMENDED THERMAL CONDUCTIVITY OF THULIUM



RECOMMENDED VALUES\*

T <sub>1</sub>	Single Crystal			Polycrystalline		
	(// to c-axis)	(⊥ to c-axis)	(⊥ to c-axis)	k	k <sub>1</sub>	k <sub>2</sub>
0	0	0	0	0	0	0
2	0.147	8.49	0.159	0.127	7.34	459.7
3	0.169	9.76	0.184	0.155	8.96	454.3
4	0.186	10.7	0.205	0.178	10.3	462.5
5	0.194	11.4	0.223	0.198	11.4	443.5
6	0.206	11.9	0.236	0.214	12.4	441.7
7	0.208	12.1	0.244	0.226	13.0	439.9
8	0.203	12.0	0.246	0.232	13.4	438.1
9	0.194	11.7	0.245	0.230	13.3	436.3
10	0.185	10.7	0.238	0.226	13.1	434.5
15	0.175	10.1	0.232	0.219	12.7	432.7
20	0.164	9.13	0.218	0.211	12.2	430.9
25	0.150	8.30	0.202	0.194	11.2	427.3
30	0.136	6.95	0.167	0.179	10.3	423.7
35	0.123	6.19	0.141	0.149	8.61	414.7
40	0.105	6.07	0.123	0.127	7.34	405.7
42.4	0.106	6.12	0.106	0.116	6.70	396.7
45	0.109	6.30	0.102	0.108	6.24	387.7
50	0.119	6.88	0.0948	0.106	6.12	383.4
55	0.141	8.15	0.0912	0.102	6.01	378.7
58	0.160	9.24	0.0911	0.103	5.95	369.7
60	0.164	9.48	0.0920	0.107	6.18	360.7
70	0.175	10.1	0.0977	0.111	6.41	355.3
80	0.185	10.7	0.105	0.113	6.53	351.7
90	0.193	11.2	0.108	0.120	6.93	333.7
100	0.200	11.6	0.111	0.126	7.28	315.7
150	0.224	12.9	0.126	0.131	7.57	297.7
200	0.235	13.6	0.134	0.135	7.80	278.7
250	0.241	13.9	0.138	0.152	8.78	189.7
273.2	0.242	14.0	0.140	0.162	9.35	99.7
300	0.242	14.0	0.141	0.167	9.65	9.7
				0.168	9.71	32.0
				0.179	9.71	80.3

REMARKS

The recommended values are for well-annealed 99.9% pure thulium with residual electrical resistivity  $\rho_0 = 3.5$  and  $1.7 \mu\Omega$  cm along the directions parallel and perpendicular to the c-axis of the single crystal, respectively (characterization by  $\rho_0$  becomes important at temperatures below about 200 K). The recommended values that are supported by experimental thermal conductivity data are thought to be accurate to within 3% of the true values near room temperature and 5 to 15% at other temperatures.

\* T in K, k<sub>1</sub> in Watt cm<sup>-1</sup> K<sup>-1</sup>, T<sub>2</sub> in F, and k<sub>2</sub> in Btu hr<sup>-1</sup> ft<sup>-1</sup> F<sup>-1</sup>.

† Values in parentheses are extrapolated.

# THERMAL CONDUCTIVITY OF TIN

Continued on Figure 61-2

FIGURE SHOWS ONLY TO OF THE CURVES REPORTED IN TABLE

FIGURE SHOWS ONLY TO OF THE CURVES REPORTED IN TABLE

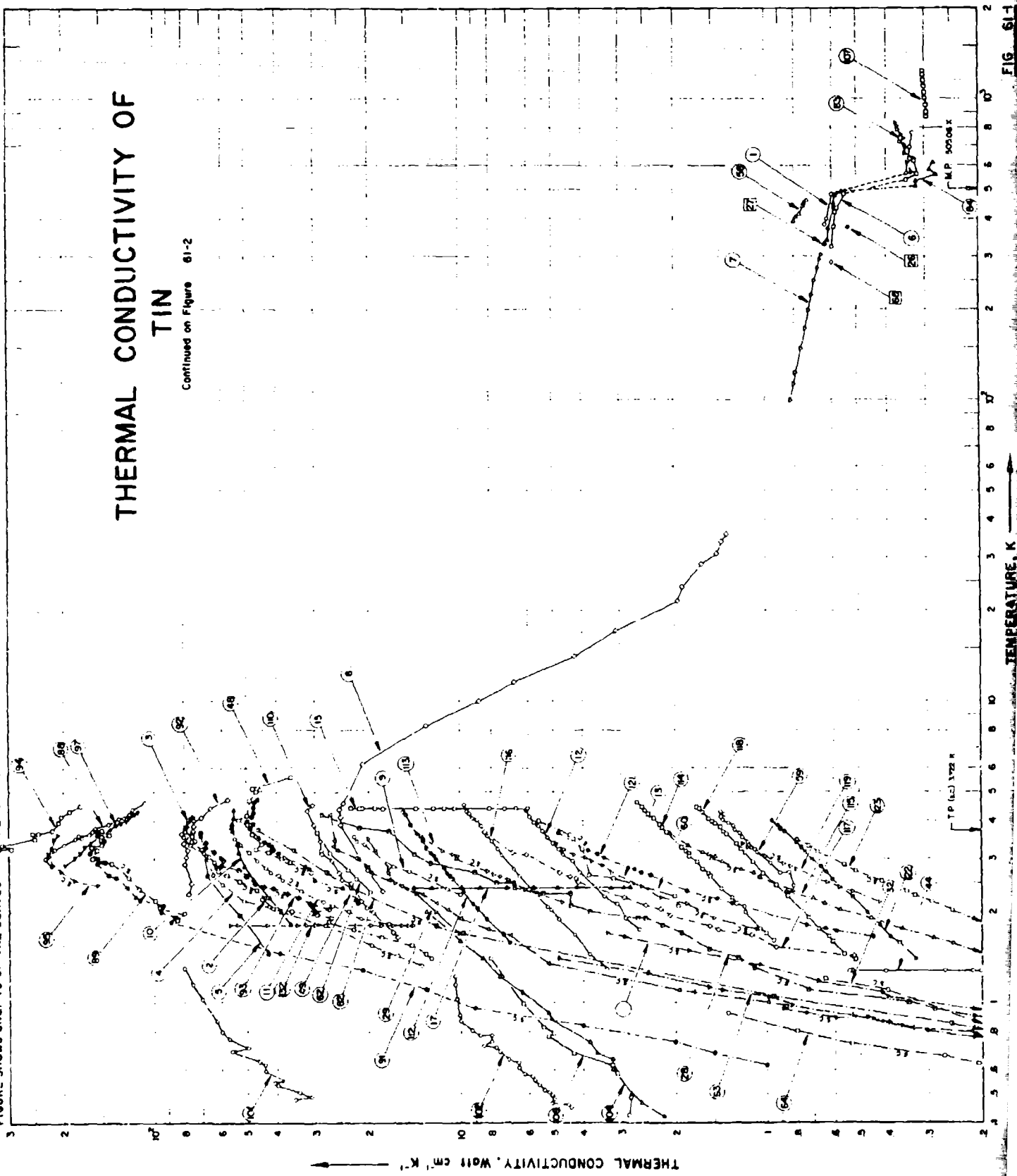
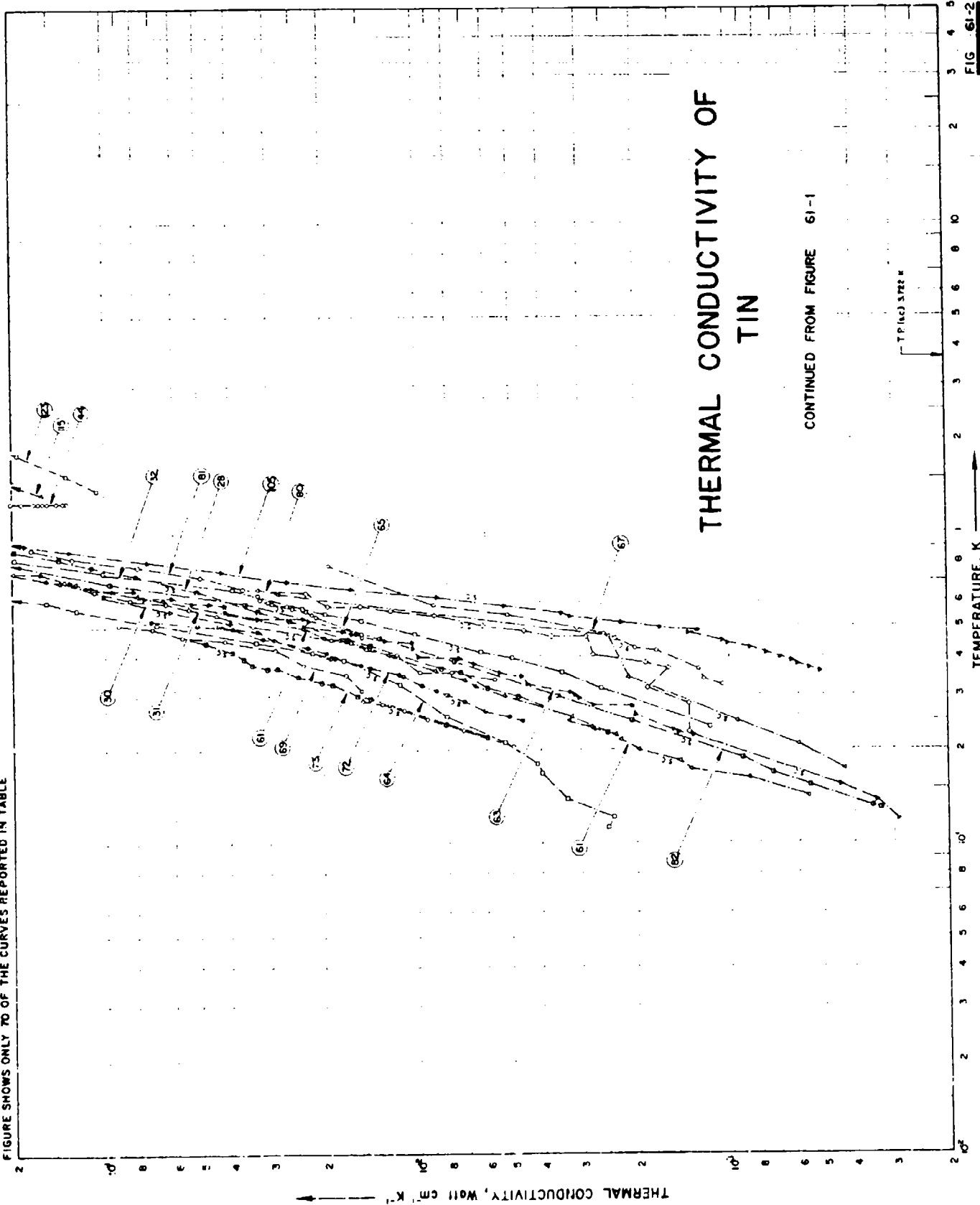


FIG 61-1

FIGURE SHOWS ONLY 70 OF THE CURVES REPORTED IN TABLE



# THERMAL CONDUCTIVITY OF TIN

CONTINUED FROM FIGURE 61-1

T.P. (IN) 3722 K

TEMPERATURE, K

SPECIFICATION TABLE NO. 61 THERMAL CONDUCTIVITY OF TlN

(Impurity  $< 0.20\%$  each; total impurities  $< 0.50\%$ )

[ For Data Reported in Figure and Table No. 61 ]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition ( weight percent).	Specifications and Remarks
1	85	L	1919	381-771		Sn I	Pure; in both solid and liquid states.	
2	117	L	1949	1.4-3.7		Sn I	99.992 <sup>+</sup> pure; single crystal; 2.3 mm dia; made of Chempur tin (99.992 pure) purified further by several times melting in vacuo, crystallizing and etching; measured with heat flow at 95 degrees to the tetragonal axis; in superconducting state.	
3	117	L	1949	3.8-4.1		Sn I	The above specimen in normal state.	
4	117	L	1949	1.3-3.6		Sn II	Single crystal; 0.8 mm dia x 70 mm long; made of Chempur tin purified further by several times melting in vacuo, crystallizing and etching; electrical resistivity ratio $\rho(273K)/\rho(4.2K) = 16700$ ; measured with heat flow at 85 degrees to the tetragonal axis; in superconducting state.	
5	117	L	1949	1.5-3.7		Sn II	The above specimen measured in a magnetic field of strength 510 Oe; in normal state.	
6	19	L	1923	323-620			Specimen in both solid and liquid states; 12 cm long and 1.5 cm in dia; melting point 212 C.	
7	94	L	1908	99-303			Pure; from Kahlbaum; density 7.28 g cm <sup>-3</sup> at 21 C; electrical resistivity reported as 3.00 and 10.65 $\mu$ ohm cm at -170.4 and 11.6 C, respectively.	
8	122	L	1955	2.3-3.6	3	Sn 1	99.997 pure; single crystal; 2.95 cm long, 0.389 cm in dia; supplied by Johnson, Matthey Co. Ltd.; measured in a magnetic field; in normal state.	
9	122	L	1955	2.3-3.6	3	Sn 1	The above specimen in superconducting state.	
10	457	L	1949	1.8-4.4	3	Sn 2	99.997 Sn (by difference), 0.003 impurities; polycrystalline; measured in a longitudinal magnetic field; in normal state.	
11	457	L	1949	1.8-3.5	3	Sn 2	The above specimen in superconducting state.	
12	457	L	1949	1.8-4.4	3	Sn 3	99.967 Sn (by difference), 0.033 Hg; polycrystalline; measured in a longitudinal magnetic field; in normal state.	
13	457	L	1949	1.8-3.4	3	Sn 3	The above specimen in superconducting state.	
14	270	P	1915	308.2			Specimen 25 cm long, 0.25 cm in dia; thermal conductivity value calculated from measured thermal diffusivity using the values of density and specific heat taken from the Tabellen of Landolt and Bornstein.	
15	342	L	1953	4.4		Sn 1	99.987 pure; single crystal; supplied by Johnson, Matthey Co. Ltd.; measured in transverse magnetic fields with strength H ranging from 0.19 to 3.57 kOe.	
16	342	L	1953	3.0		Sn 1	The above specimen measured in transverse magnetic fields with strength H ranging from 0.29 to 3.57 kOe.	
17	342	L	1953	2.4		Sn 1	The above specimen measured in transverse magnetic fields with strength H ranging from 0.35 to 3.75 kOe.	



SPECIFICATION TABLE NO. 61 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)	Specifications and Remarks
18	342	L	1953	2.4		Sn 1		The above specimen measured in longitudinal magnetic fields with strength H ranging from 0.29 to 3.75 kOe.
19	342	L	1953	3.0		Sn 1		The above specimen measured in longitudinal magnetic fields with strength H ranging from 0.35 to 3.66 kOe.
20	342	L	1953	4.4		Sn 1		The above specimen measured in longitudinal magnetic fields with strength H ranging from 0.35 to 3.75 kOe.
21	74	L	1950	2.21	3	Sn 2		99.996 pure; homogeneous solid solution with few large crystals; superconducting transition point 3.71 K; measured in magnetic fields with strength H ranging from 62 to 1453 gauss.
22	74	L	1950	4.29	3	Sn 2		The above specimen measured in magnetic fields with strength H ranging from 62 to 1453 gauss.
23	74	L	1950	2.42	3	Sn 3		99.967 pure; homogeneous solid solution with few large crystals; superconducting transition point 3.68 K; measured in magnetic fields with strength H ranging from 123 to 1213 gauss.
24	412	L	1955	0.39-0.65				99.997 pure; single crystal; supplied by Johnson, Matthey Co. Ltd.; in superconducting state (same specimen as used for curve No. 8).
25	412	L	1955	0.25-0.80				99.997 pure; polycrystalline; supplied by Johnson, Matthey Co. Ltd.; in superconducting state.
26	460		1957	373.2				Electrical conductivity $6.6 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 100 C.
27	230	L	1925	327				0.43 total impurities; specimen 10 cm long and 1.9 cm in dia; electrical conductivity $8.96 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 22 C.
28	452	L	1955	0.2-4.2	2-4	JM 4600; Sn2		Spectroscopically pure; single crystal with tetragonal axis parallel to rod axis; 2.530 mm dia rod supplied by Johnson, Matthey Co. Ltd.; cast and recrystallized; superconducting state.
29	452	L	1955	1.7-3.5	2-4	JM 4600; Sn2		The above specimen measured in a longitudinal field of 400 gauss; in normal state; data corrected for magneto-conductivity.
30	452	L	1955	0.40-0.64	2.4	JM 4600; Sn3		Similar to the above specimen but the dia 5.11 mm; in superconducting state.
31	452	L	1955	0.34-0.71	2-4	JM 4600; Sn4		Similar to the above specimen but with tetragonal axis at $88^\circ$ to the rod axis, and rod dia 2.135 mm; in superconducting state.
32	452	L	1955	0.24-1.2	2-4	JM 4600; Sn5		Pure; polycrystalline; specimen dia 2.315 mm; cast, recrystallized, and strained grain size 0.50 mm; in superconducting state.
33	290	L	1952	1.59		Sn II		99.996 pure; polycrystal with several large crystals; 4.1 mm dia rod; prepared from Johnson-Matthey tin, J.M. Lab No. 2356; electrical resistivity ratio $\rho(273K)/\rho(4.2K) = 8000$ ; measured in increasing transverse magnetic fields with strength H ranging from zero to 303 gauss.

SPECIFICATION TABLE NO. 61 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
34	290	L	1952	1.59		Sn II	The above specimen measured in decreasing transverse magnetic fields with strength H ranging from 246 to 120 gauss.
35	290	L	1952	2.21		Sn II	The above specimen measured in increasing transverse magnetic fields with strength H ranging from zero to 273 gauss.
36	290	L	1952	2.21		Sn II	The above specimen measured in decreasing transverse magnetic fields with strength H ranging from 255 to zero gauss.
37	290	L	1952	2.53		Sn II	The above specimen measured in increasing transverse magnetic fields with strength H ranging from zero to 240 gauss.
38	290	L	1952	2.53		Sn II	The above specimen measured in decreasing transverse magnetic fields with strength H ranging from 149 to zero gauss.
39	290	L	1952	2.18		Sn IV	99.996 pure; single crystal with its tetragonal axis at about 70° from the geometric axis; 4.3 mm dia rod; prepared from Johnson-Matthey tin, J. M. Lab No. 2356; electrical resistivity ratio $\rho(273K)/\rho(4.2K) = 11000$ ; measured in increasing transverse magnetic fields with strength H ranging from zero to 272 gauss and field direction nearly parallel to the tetragonal axis.
40	290	L	1952	2.18		Sn IV	The above specimen measured in decreasing transverse magnetic fields nearly parallel to the tetragonal axis with strength H ranging from 181 to zero gauss.
41	290	L	1952	2.86		Sn IV	The above specimen measured in increasing transverse magnetic fields with strength H ranging from zero to 204 gauss and field direction at about 20° with the tetragonal axis.
42	290	L	1952	2.86		Sn IV	The above specimen measured in decreasing transverse magnetic fields with strength H ranging from 115 to zero gauss and field direction at about 20° with the tetragonal axis.
43	290	L	1952	3.62		Sn IV	The above specimen measured in transverse magnetic fields with strength H ranging from zero to 2620 gauss and field direction at about 20° with the tetragonal axis.
44	290	L	1952	1.27			99.866 pure; 0.134 Bi; polycrystalline; specimen 5.0 mm in dia; prepared from Johnson-Matthey materials; measured in increasing transverse magnetic fields with strength H ranging from zero to 304 gauss.
45	290	L	1952	1.27			The above specimen measured in decreasing transverse magnetic fields with strength H ranging from 237 to zero gauss.
46	285	L	1952	2.5			Spectroscopically pure; polycrystal with a few large crystals; 3-4 mm dia x 10 cm long; prepared from Johnson-Matthey tin; measured in increasing transverse magnetic fields with strength H ranging from zero to 238.2 gauss.
47	285	L	1952	2.5			The above specimen measured in decreasing transverse magnetic fields with strength H ranging from 149.7 to zero gauss.
48	404, 738	L	1950	2.7-5.6	4	Sn I	99.986 <sup>+</sup> pure; single crystal; supplied by Johnson-Matthey Co. Ltd.; specimen 8 cm long, 4 mm in dia, in superconducting state below transition temperature.

SPECIFICATION TABLE NO. 61 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
49	404, 738	L	1950	2.1-4.7	4	Sn II	Similar to the above specimen but with a dia of 2 mm; in superconducting state below transition temperature.
50	404, 738	L	1950	2.18	4	Sn II	The above specimen measured in longitudinal magnetic fields with strength H ranging from zero to 455.2 gauss.
51	404, 738	L	1950	3.19	4	Sn II	The above specimen measured in longitudinal magnetic fields with strength H ranging from zero to 258.7 gauss.
52	404, 738	L	1950	3.32	4	Sn II	The above specimen measured in longitudinal magnetic fields with strength H ranging from zero to 376.1 gauss.
53	404, 738	L	1950	3.77	4	Sn II	The above specimen measured in longitudinal magnetic fields with strength H ranging from zero to 375.2 gauss.
54	404, 738	L	1950	4.38	4	Sn II	The above specimen measured in longitudinal magnetic fields with strength H ranging from zero to 532 gauss.
55	404, 738	L	1950	5.02	4	Sn II	The above specimen measured in longitudinal magnetic fields with strength H ranging from zero to 434.5 gauss.
56	404, 738	L	1950	2.2-3.7	4	Sn II	The above specimen; superconducting transition point 3.69 K; measured in a magnetic field, in normal state.
57	406	C	1922	313.2	5		Pure; specimen 3 cm long and 3 cm in dia; zinc used as comparative material.
58	421	E	1941	390-450			Single crystal; electrical resistivity reported as 14.45, 15.30, 15.80, 17.55, 18.03 and 18.94 $\mu$ ohm cm at 390, 2, 404, 4, 412, 0, 439, 5, 447, 6, and 460.2 K, respectively.
59	317, 457	L	1957	3.4-4.3	2	1	0.197 Bi; 4 mm dia rod; annealed for several months; electrical resistivity reported as 0.0721 and 11.69 $\mu$ ohm cm, at 4.2 and 300 K, respectively; measured in a magnetic field of 560 gauss; in normal state.
60	458	L	1958	2.1-4.0	5	Sn 1	99.9 pure, monocrystalline; 1.89 mm dia rod; polished; in superconducting state.
61	458	L	1958	0.11-4.2	5	Sn 2	99.998 pure; monocrystalline; 1.72 mm dia rod with rough surface; angle between specimen axis and [001] direction = 30°; in superconducting state.
62	458	L	1958	2.3-3.7	5	Sn 2	The above specimen in normal state.
63	458	L	1958	0.12-1.4	5	Sn 3	99.998 pure; monocrystalline; 1.49 mm dia rod with rough surface; angle between specimen axis and [001] direction = 70°; in superconducting state.
64	458	L	1958	0.11-0.92	5	Sn 4	99.998 pure; monocrystalline; 1.81 mm dia rod with polished surface; angle between specimen axis and [001] direction = 45°; in superconducting state.
65	499	L	1953	0.17-0.61		Sn 2	99.997 pure; polycrystalline; effected by "frozen in" magnetic field, in superconducting state.
66	499	L	1953	0.37-0.71		Sn 2	99.997 pure; polycrystalline; in superconducting state.
67	499	L	1953	0.33-0.80		Sn 1	99.997 pure; single crystal; in superconducting state.

SPECIFICATION TABLE NO. 61 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)	Specifications and Remarks
64	459	P	1905	286.7				Chemically pure; specimen in ring form; cast and turned; density $7.31 \text{ g cm}^{-3}$ at $14 \text{ C}$ ; electrical resistivity $11.82 \text{ } \mu\text{ ohm cm}$ at $13.5 \text{ C}$ ; thermal conductivity value calculated from measured data of thermal diffusivity and specific heat.
69	454	L	1958	0.21-0.52	2	E 0		Spectroscopically pure; single crystal; 4.91 mm dia rod with specimen axis at $85^\circ$ to the tetrad axis; provided by Johnson Matthey Co. Ltd.; cast, crystallized, and electro-polished to $0.05 \text{ } \mu$ in surface roughness; in superconducting state.
70	454	L	1958	0.21-0.52	2	E 1		The above specimen 50% clouding etched by exposure to HCl fumes; surface roughness $0.3 \text{ } \mu$ ; in superconducting state.
71	454	L	1958	0.26-0.51	2	E 2		The above specimen lightly etched to $0.7 \text{ } \mu$ in surface roughness; in superconducting state.
72	454	L	1958	0.35-0.53	2	E 3		The above specimen etched to $1.1 \text{ } \mu$ in surface roughness; in superconducting state.
73	454	L	1958	0.21-0.47	2	E 4		The above specimen electro-polished $0.05 \text{ } \mu$ in surface roughness and $4.75 \text{ mm}$ in dia; in superconducting state.
74	454	L	1958	0.21-0.47	2	E 5		The above specimen 25% clouding etched by exposure to HCl fumes; surface roughness $0.12 \text{ } \mu$ ; in superconducting state.
75	454	L	1958	0.25-0.52	2	E 6		The above specimen electro-polished to $0.10 \text{ } \mu$ in surface roughness and $3.15 \text{ mm}$ in dia; in superconducting state.
76	454	L	1958	0.24-0.53	2	E 7		The above specimen etched to $1.0 \text{ } \mu$ in surface roughness; in superconducting state.
77	454	L	1958	0.27-0.65	2	E 10		The above specimen electro-polished, etched, annealed at $220 \text{ C}$ , then again etched to $1.0 \text{ } \mu$ in surface roughness and $1.96 \text{ mm}$ in dia; residual electrical resistivity $0.00377 \text{ } \mu\text{ ohm cm}$ ; in superconducting state.
78	454	L	1958	0.24-0.66	2	D 0		Spectroscopically pure; 2.82 mm dia rod consisted of 3 large crystals; as cast; surface roughness $0.10 \text{ } \mu$ ; residual electrical resistivity $0.0014 \text{ } \mu\text{ ohm cm}$ ; in superconducting state.
79	454	L	1958	0.26-0.53	2	D 1		The above specimen etched to $0.7 \text{ } \mu$ in surface roughness; in superconducting state.
80	455	L	1953	0.18-0.67	10-25	Sn II		Spectroscopically pure; polycrystalline; 1.3 mm dia rod made up of crystals of the order of the dia; cast in tube; in superconducting state.
81	455	L	1953	0.23-0.90	10-25	Sn III		Similar to the above specimen but with dia $0.7 \text{ mm}$ ; in superconducting state.
82	682	L	1960	0.13-4.0				$0.002$ impurity; single crystal; $0.175 \text{ cm}$ in dia and $\sim 50 \text{ mm}$ long; cast in vacuo in thin-walled glass capillary, in which crystallization took place immediately after casting; electrical resistivity ratio $\rho(293\text{K})/\rho - 6250$ ; in superconducting state below transition point.
83	592	L	1959	570-833				In molten state; melting point $231.9 \text{ C}$ .
84	597, 708	L	1961	337-610	2-5			99.94 pure; in both solid and liquid states.

SPECIFICATION TABLE NO. 61 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
85	509	C	1954	306-378	± 3	56 B-1	Specimen made from NBS freezing-point tin No. 42b (freezing point 231.9 C); 0.500 in. dia x 0.500 in. long; electrolytically deposited pure copper used as comparative material; reference data of copper taken from International Critical Tables (Vol. 5, McGraw Hill Book Co., New York, p. 221, 1929).
86	511	L	1958	298.0			Specimen with radius of 0.7 cm furnished by the manufacturer Erba; measured in atmospheric pressure.
87	504	P	1961	295.2	± 5		Rectangular plate 1.9 x 1.9 x 0.306 cm; supplied by Armco; thermal conductivity value calculated from measured thermal diffusivity using values of density and specific heat taken from Smithsonian Physical Tables (9th ed., 1954).
88	712	L	1961	2.9-4.6		Sn-1	High purity; single crystal; 2.6 mm in dia; specimen axis in the [001] orientation; residual electrical resistivity $\rho_f(7 \pm 0.5) \times 10^{-4} \mu\text{ ohm cm}$ ; superconducting transition point 3.72 K; measured in a longitudinal magnetic field; in normal state; data corrected to zero field.
89	712	L	1961	1.9-3.7		Sn-1	The above specimen in superconducting state.
90	712	L	1961	2.6-4.4		Sn-2	Similar to the above specimen but 1.1 mm dia; and residual electrical resistivity $\rho_f(1.65 \pm 0.2) \times 10^{-4} \mu\text{ ohm cm}$ ; measured in a longitudinal magnetic field; in normal state; data corrected to zero field.
91	712	L	1961	0.6-3.7		Sn-2	The above specimen in superconducting state.
92	712	L	1961	2.3-4.7		Sn-3	Similar to the above specimen but 1.5 mm in dia and residual electrical resistivity $\rho_f(6.1 \times 10^{-4} \mu\text{ ohm cm}$ ; measured in a longitudinal magnetic field; in normal state; data corrected to zero field.
93	712	L	1961	2.0-3.7		Sn-3	The above specimen in superconducting state.
94	712	L	1961	2.8-4.5		Sn-4	High purity; single crystal; 2.1 mm in dia; specimen axis in the [110] orientation; residual electrical resistivity $\rho_f(1.2 \pm 0.5) \times 10^{-4} \mu\text{ ohm cm}$ ; measured in a longitudinal magnetic field.
95	712	L	1961	2.5-3.7		Sn-4	The above specimen in superconducting state.
96	706	L	1981	273, 373			Density 7.27 g cm <sup>-3</sup> ; electrical conductivity reported as 9.346 and 13.524 x 10 <sup>4</sup> ohm <sup>-1</sup> cm <sup>-1</sup> at 0 and 100 C, respectively.
97	739	L	1961	2.7-4.3		Sn 1	High purity; single crystal; rod specimen about 14 cm long made from 2 mm dia extruded wire; nominal orientation [001]; rod along the tetrad axis; specimen crystallized by slow cooling, etched in concentrated HCl; electrical resistivity ratio $\rho(293K)/\rho_0 = 80000$ ; measured in a magnetic field; in normal state.
98	739	L	1961	2.8-4.2		Sn 2	Similar to the above specimen but made from 5 parts of Johnson, Matthey Specpure and 8 parts of high purity tin from Vulcan De-luming Co.; electrical resistivity ratio $\rho(293K)/\rho_0 = 23000$ ; measured in a magnetic field; in normal state.

SPECIFICATION TABLE NO. 61 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)	Specifications and Remarks
99	739	L	1961	2.7-4.2		Sn 4		Similar to the above specimen but made from equal parts of Johnson, Matthey Specpure and Vulcan De-tinning Co. high purity tin; electrical resistivity ratio $\rho(293K)/\rho_0 = 4500$ ; measured in a magnetic field; in normal state.
100	740	L	1965	0.34-1.3	< 10			99.99 <sup>+</sup> pure; single crystal; specimen 2.2 mm in dia and 100 mm long; residual electrical resistivity $\rho(4.2K) = 1.7 \times 10^{19}$ ohm cm in superconducting state.
101	740	L	1965	0.48-1.3	< 10			The above specimen measured in a longitudinal magnetic field of 500 oersteds; in normal state.
102	740	L	1965	0.45-1.2	< 10			The above specimen measured in a transverse magnetic field of 500 oersteds; in normal state.
103	740	L	1965	0.39-1.5	< 10			0.05 mm tin foil produced by rolling tin having the same purity as the above specimen; in superconducting state.
104	740	L	1965	0.42-2.0	< 10			The above specimen measured in a longitudinal magnetic field of 330 oersteds; in normal state.
105	740	L	1965	0.36-1.7	< 10			Similar to the above specimen but the foil had been preliminarily etched; in superconducting state.
106	740	L	1965	0.43-1.4	< 10			The above specimen measured in a longitudinal magnetic field of 330 oersteds; in normal state.
107	735, 839	P	1965	870-1230	6-8			Molten specimen filled in the space between 2 coaxial tantalum tubes of dia 23.8 and 8 mm, each tube 0.12 mm thick; thermal conductivity values calculated from measured data of thermal diffusivity and specific heat using data of density taken from M. P. Slavinskii, (Physicochemical Properties of Elements [Russian] 1952).
108	744	P	1966	465-1365	7			Molten specimen filled in the space between two coaxial thin-walled tantalum tubes of 24 and 8 mm dia, respectively; thermal conductivity values calculated from measured data of thermal diffusivity and specific heat.
109	838	C	1966	429-773				Molten specimen placed in a hole of 21 mm in dia drilled in an asbestos cement cylinder of 30 mm height; IKTIN9T steel used as comparative material.
110	837	L	1967	1.6-4.5	1	Sn 1		99.999 pure; supplied by Johnson-Matthey; extruded into 1.5 mm dia wire; electrical resistivity reported as 0.00213 and 13.06 $\mu$ ohm cm at 4.2 and 273 K, respectively; superconducting transition temperature 3.720 K; below the transition temperature, a longitudinal magnetic field was applied to the specimen; in normal state.
111	837	L	1967	1.6-3.6	1	Sn 1		The above specimen measured with the magnetic field removed; in superconducting state.
112	837	L	1967	1.6-4.5	1	Pb 1		0.019 Pb; prepared by vacuum-melting appropriate amounts of Johnson-Matthey 99.999 pure Sn and Pb, extruding into 1.5 mm dia wire; annealed at -200 C for several days; electrical resistivity reported as 0.00564 and 12.71 $\mu$ ohm cm at 4.2 and 273 K, respectively; superconducting transition point 3.716 K; measured in a longitudinal magnetic field; in normal state.

SPECIFICATION TABLE NO. 61 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)	Specifications and Remarks
113	837	L	1967	2.2-3.6	1	Pb 1		The above specimen measured without the magnetic field; in superconducting state.
114	837	L	1967	1.6-4.6	1	Pb 2		0.174 Pb, prepared by vacuum-melting appropriate amounts of Johnson-Matthey 99.999 pure Sn and Pb, extruding into 1.5 mm dia wire; annealed at ~200 C for several days; electrical resistivity reported as 0.0500 and 13.19 $\mu$ ohm cm at 4.2 and 273 K, respectively; superconducting transition point 3.713 K; measured in longitudinal magnetic field; in normal state.
115	837	L	1967	1.4-3.7	1	Pb 2		The above specimen measured without the magnetic field; in superconducting state.
116	837	L	1967	1.1-4.5	1	Bi 1		0.012 Bi, prepared by vacuum-melting appropriate amounts of Johnson-Matthey 99.999 pure Sn and Bi, extruding into 1.5 mm dia wire; annealed at ~200 C for several days; electrical resistivity reported as 0.00578 and 12.61 $\mu$ ohm cm at 4.2 and 273 K, respectively; superconducting transition point 3.725 K; measured in a longitudinal magnetic field; in normal state.
117	837	L	1967	1.4-3.7	1	Bi 1		The above specimen measured without the magnetic field; in superconducting state.
118	837	L	1967	1.4-4.4	1	Bi 2		0.140 Bi, prepared by vacuum-melting appropriate amounts of Johnson-Matthey 99.999 pure Sn and Bi, extruding into 1.5 mm dia wire; annealed at ~200 C for several days; electrical resistivity reported as 0.0721 and 11.91 $\mu$ ohm cm at 4.2 and 273 K, respectively; superconducting transition point 3.709 K; measured in a longitudinal magnetic field; in normal state.
119	837	L	1967	2.2-3.7	1	Bi 2		The above specimen measured without the magnetic field; in superconducting state.
120	837	L	1967	1.6-4.7	1	Hg 1		0.018 Hg, prepared by vacuum-melting appropriate amounts of Johnson-Matthey 99.999 pure Sn and Hg, extruding into 1.5 mm dia wire; annealed at ~200 C for several days; electrical resistivity reported as 0.0201 and 12.87 $\mu$ ohm cm at 4.2 and 273 K, respectively; superconducting transition point 3.718 K; measured in a longitudinal magnetic field; in normal state.
121	837	L	1967	1.7-3.7	1	Hg 1		The above specimen measured without the magnetic field; in superconducting state.
122	837	L	1967	1.4-4.0	1	Hg 2		0.168 Hg, prepared by vacuum-melting appropriate amounts of Johnson-Matthey 99.999 pure Sn and Hg, extruding into 1.5 mm dia wire; annealed at ~200 C for several days; electrical resistivity reported as 0.113 and 11.28 $\mu$ ohm cm at 4.2 and 273 K, respectively; superconducting transition point 3.686 K; measured in a longitudinal magnetic field; in normal state.
123	837	L	1967	1.4-3.6	1	Hg 2		The above specimen measured without the magnetic field; in superconducting state.
124	436	L	1958	1.6-4.6	4-5	3		0.047 Hg, prepared by vacuum-melting appropriate amounts of Johnson-Matthey 99.999 pure Sn and Hg, casting into 1 mm dia x 12 cm long wire in a pyrex capillary; <b>Residual</b> electrical resistivity 0.014 $\mu$ ohm cm; measured in a magnetic field; in normal state.
125	836	L	1958	1.6-4.2	4-5	3		The above specimen measured without the magnetic field; in superconducting state.

SPECIFICATION TABLE NO. 51 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
126	836	L	1958	1.6-4.8	4-5	S	0.049 Bi; prepared by vacuum-melting appropriate amounts of Johnson-Matthey 99.999 pure Sn and Bi, casting into 1 mm dia x 1.2 cm long wire in a pyrex capillary; <b>residual</b> electrical resistivity 0.020 $\mu$ ohm cm; measured in a magnetic field; in normal state.
127	836	L	1958	1.5-2.8	4-5	S	The above specimen measured without the magnetic field; in superconducting state.
128	836	L	1958	1.6-4.9	4-5	S	0.040 Pb; prepared by vacuum-melting appropriate amounts of Johnson-Matthey 99.999 pure Sn and Pb, casting into 1 mm dia x 1.2 cm long wire in a pyrex capillary; <b>residual</b> electrical resistivity 0.010 $\mu$ ohm cm; measured in a magnetic field; in normal state.
129	836	L	1958	1.6-2.6	4-5	S	The above specimen measured without the magnetic field; in superconducting state.
130	835	L	1966	1.65	$\pm$ 4	Sn 5.0	Polycrystalline; 2.25 mm dia x 10 cm long; supplied by Johnson-Matthey and Co.; specimen recrystallized in an alumina packing; annealed at a temperature just below the melting point for several hrs; electrical resistivity ratio $\rho(290K)/\rho(4.2K) = 5000$ ; measured in a transverse static magnetic field with strength ranging from 72 to 277 gauss.
131	835	L	1966	1.65	$\pm$ 4	Sn 5.0	The above specimen measured with the field rotated about the specimen axis 10 times at 5 sec rev <sup>-1</sup> between points of measurement.
132	835	L	1966	1.80	$\pm$ 4	Sn 47	Single crystal with tetrad axis at 3 degrees to the specimen axis; 2.15 mm dia x 10 cm long; same preparation procedure as the above specimen; electrical resistivity ratio $\rho(290K)/\rho(4.2K) = 47000$ ; measured in a transverse static magnetic field with strength ranging from 76 to 227 gauss.
133	835	L	1966	1.80	$\pm$ 4	Sn 47	The above specimen measured with the field rotated about the specimen axis 5 times at 8 sec rev <sup>-1</sup> between points of measurement.









DATA TABLE NO. 61 (continued)

T	k	T	k	T	k	T	k	T	k	T	k		
<u>CURVE 62</u>		<u>CURVE 63 (cont.)</u>		<u>CURVE 65 (cont.)</u>		<u>CURVE 69 (cont.)</u>		<u>CURVE 72</u>		<u>CURVE 74* (cont.)</u>		<u>CURVE 77*</u>	
2.32	18.8	0.680	0.110	0.58	0.0250	0.428	0.0285	0.252	0.00455	0.350	0.0195	0.265	0.00216
2.58	22.0	0.710	0.130	0.59	0.0260	0.455	0.0410	0.260	0.0050	0.370	0.0238	0.278	0.00246
2.74	23.5	0.755	0.160	0.60	0.0290	0.472	0.0560	0.270	0.0060	0.385	0.0265	0.292	0.00285
3.12	26.7	0.770	0.210	0.61	0.0297	0.500	0.070	0.290	0.0070	0.405	0.0316	0.310	0.00326
3.65	29.4	0.805	0.300	0.623	0.0315	0.515	0.090	0.305	0.00795	0.410	0.0360	0.380	0.00380
<u>CURVE 63</u>		<u>CURVE 66*</u>		<u>CURVE 70*</u>		<u>CURVE 75*</u>		<u>CURVE 75*</u>		<u>CURVE 76*</u>		<u>CURVE 76*</u>	
0.120	0.00029	0.890	0.470	0.21	0.00345	0.350	0.0109	0.350	0.0112	0.423	0.0318	0.620	0.00830
0.135	0.00034	0.890	0.590	0.237	0.0050	0.372	0.0140	0.372	0.0140	0.445	0.0380	0.645	0.0101
0.156	0.00044	0.920	0.740	0.24	0.0054	0.380	0.0150	0.380	0.0150	0.465	0.0460	0.665	0.0130
0.225	0.00130	0.950	0.880	0.25	0.0056	0.400	0.0180	0.400	0.0180	0.475	0.0498	0.685	0.0160
0.230	0.00145	0.980	0.830	0.27	0.00876	0.420	0.0192	0.420	0.0192	0.490	0.0540	0.705	0.0190
0.235	0.00150	1.05	0.980	0.275	0.00925	0.435	0.0222	0.435	0.0222	0.515	0.0630	0.725	0.0220
0.263	0.00200	1.10	1.410	0.305	0.0120	0.465	0.0308	0.465	0.0308	0.530	0.0705	0.745	0.0250
0.279	0.00205	1.12	1.55	0.325	0.0140	0.500	0.0385	0.500	0.0385	0.580	0.0880	0.765	0.0280
0.282	0.00270	1.25	2.45	0.360	0.0220	0.515	0.0498	0.515	0.0498	0.620	0.1125	0.785	0.0310
0.298	0.00300	1.40	4.80	0.380	0.0230	0.530	0.0705	0.530	0.0705	0.660	0.125	0.805	0.0340
0.305	0.00310	<u>CURVE 64</u>		<u>CURVE 67</u>		<u>CURVE 71*</u>		<u>CURVE 73</u>		<u>CURVE 76*</u>		<u>CURVE 76*</u>	
0.310	0.00320	0.114	0.00245	0.327	0.00105	0.390	0.0234	0.297	0.014	0.241	0.00295	0.24	0.00375
0.330	0.00440	0.123	0.00235	0.342	0.0012	0.425	0.0305	0.303	0.0154	0.258	0.00360	0.255	0.00425
0.350	0.00460	0.141	0.00330	0.368	0.00125	0.450	0.0350	0.330	0.0184	0.276	0.00472	0.28	0.00595
0.360	0.00540	0.170	0.00395	0.422	0.0017	0.470	0.0415	0.335	0.020	0.286	0.00495	0.298	0.00690
0.385	0.00650	0.183	0.00410	0.431	0.0020	0.500	0.0540	0.350	0.024	0.300	0.00590	0.32	0.0090
0.400	0.00730	0.214	0.00520	0.464	0.0023	0.515	0.0655	0.372	0.028	0.345	0.0079	0.34	0.0102
0.408	0.00980	0.260	0.00800	0.479	0.0025	0.530	0.0880	0.382	0.030	0.350	0.0089	0.35	0.0108
0.450	0.0102	0.330	0.0113	0.590	0.00305	0.580	0.118	0.390	0.0335	0.380	0.012	0.377	0.0150
0.455	0.0125	0.395	0.0170	0.594	0.00506	0.620	0.148	0.400	0.0355	0.392	0.0133	0.422	0.0209
0.480	0.0155	0.419	0.0215	0.595	0.00875	0.665	0.205	0.418	0.0388	0.410	0.0147	0.445	0.0242
0.480	0.0160	0.450	0.025	0.799	0.0188	0.680	0.286.7	0.448	0.0473	0.420	0.0153	0.47	0.0305
0.490	0.0170	0.515	0.0680	<u>CURVE 68</u>		<u>CURVE 72*</u>		<u>CURVE 74*</u>		<u>CURVE 79*</u>		<u>CURVE 79*</u>	
0.510	0.0230	0.575	0.125	286.7	0.596	0.680	0.207	0.466	0.0550*	0.435	0.0184	0.655	0.102
0.530	0.0220	0.610	0.155	<u>CURVE 69</u>		<u>CURVE 73*</u>		<u>CURVE 75*</u>		<u>CURVE 78*</u>		<u>CURVE 78*</u>	
0.530	0.0245*	0.630	0.205	0.207	0.0049	0.700	0.235	0.48	0.070	0.452	0.0285	0.26	0.02225
0.540	0.0230	0.665	0.265	0.254	0.0071	0.720	0.254	0.50	0.0940	0.452	0.0285	0.273	0.0031
0.540	0.0265*	0.730	0.530	0.273	0.0109	0.740	0.273	0.51	0.0515	0.459	0.0335	0.315	0.00425
0.540	0.0320	0.800	0.800	0.285	0.0128	0.760	0.285	0.51	0.0515	0.510	0.0450	0.349	0.0062
0.565	0.0320*	0.920	1.350	0.293	0.0148	0.780	0.293	0.51	0.0515	0.530	0.0600	0.36	0.0071
0.570	0.0410	0.920	1.350	0.315	0.0149	0.800	0.315	0.51	0.0515	0.580	0.0880	0.372	0.0080
0.595	0.0410	0.920	1.350	0.354	0.0168	0.820	0.354	0.51	0.0515	0.620	0.1125	0.39	0.00975
0.600	0.0460	0.920	1.350	0.380	0.0230	0.840	0.380	0.51	0.0515	0.660	0.125	0.39	0.00975
0.604	0.0520	0.47	0.0148	0.47	0.0148	0.860	0.47	0.51	0.0515	0.700	0.150	0.39	0.00975
0.630	0.0600	0.48	0.0149	0.48	0.0148	0.880	0.48	0.51	0.0515	0.740	0.170	0.39	0.00975
0.630	0.0740	0.545	0.0208	0.545	0.0149	0.900	0.545	0.51	0.0515	0.780	0.190	0.39	0.00975
0.665	0.0800	0.555	0.0215	0.555	0.0168	0.920	0.555	0.51	0.0515	0.820	0.210	0.39	0.00975
0.665	0.0940	0.57	0.0225	0.57	0.0230	0.940	0.57	0.51	0.0515	0.860	0.230	0.39	0.00975

\* Not shown on plot







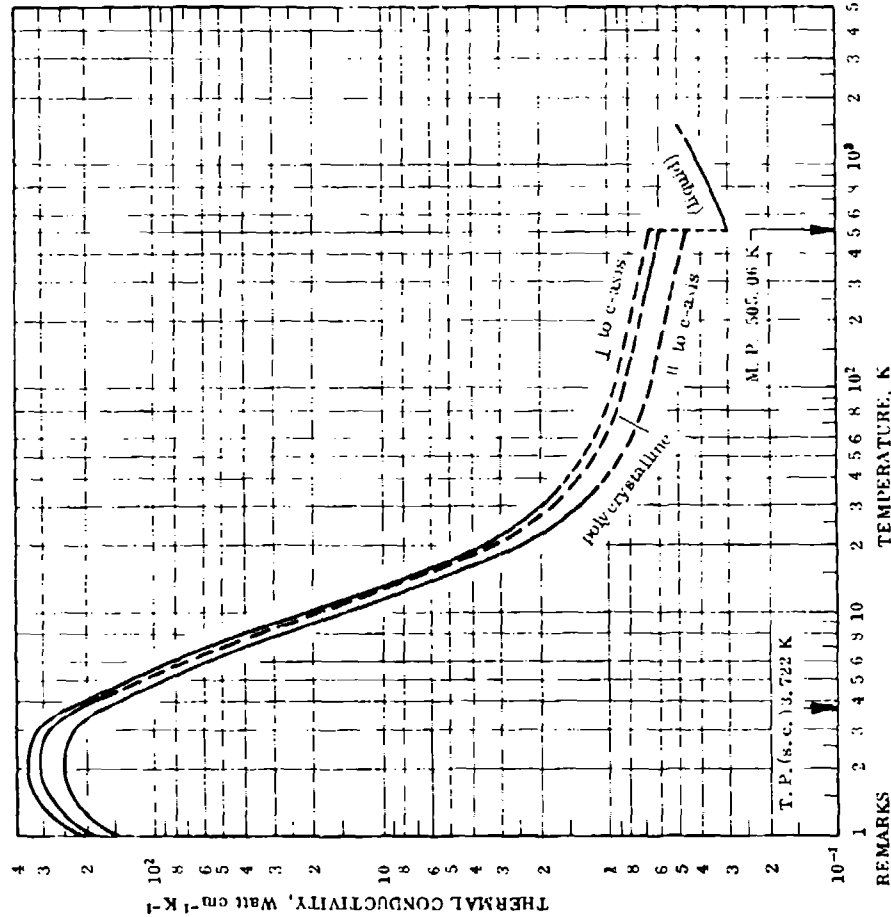
DATA TABLE NO. 61 (continued)

CURVE 129 <sup>a</sup>			CURVE 129 (cont.) <sup>a</sup>			CURVE 131 (cont.) <sup>a</sup>			CURVE 132 (cont.) <sup>a</sup>			CURVE 133 (cont.) <sup>a</sup>		
T	k	H (gauss)	T	k	H (gauss)	H (gauss)	k	H (gauss)	H (gauss)	k	H (gauss)	k	H (gauss)	k
1.616	1.59		2.738	7.41	156	9.99	210	33.0 <sup>b</sup>	159	15.9				
1.675	1.38		2.828	7.11	169	10.1	207	35.2 <sup>b</sup>	150	17.0				
1.743	1.94				177	10.5	204	17.6 <sup>b</sup>	139	19.0				
1.810	2.27				185	11.7	195	17.6 <sup>b</sup>	130	21.8				
1.885	2.44				194	11.7	192	16.8 <sup>b</sup>	121	28.5				
2.046	3.39				204	13.6	186	16.1 <sup>b</sup>	111	54.6				
2.115	3.53				204	27.5	177	15.8 <sup>b</sup>	103	56.2				
2.146	3.71				231	30.9	164	15.9 <sup>b</sup>	81					
2.220	4.04				267	31.5	158	16.4 <sup>b</sup>						
2.304	4.18				240	31.2	149	18.2 <sup>b</sup>						
2.541	5.55				223	28.7	140	19.9 <sup>b</sup>						
2.605	6.31				214	18.6	130	22.6 <sup>b</sup>						
2.687	5.94				206	14.6	121	27.4 <sup>b</sup>						
2.769	6.00				186	11.0	111	42.6						
2.864	7.05				176	10.4	104	52.9						
3.059	6.91				171	10.3	96	54.1 <sup>b</sup>						
3.145	8.38				159	10.0								
3.265	9.00				146	10.1								
3.353	9.71				129	10.7								
3.499	9.03				96	10.8								
3.565	9.01				73	10.8								
3.690	9.61													
3.823	10.60													
3.945	10.93													
4.064	11.09													
4.348	9.42													
4.470	9.46													
4.589	11.92													
4.709	11.78													
4.855	12.01													
CURVE 129 <sup>b</sup>														
1.592	4.86				76	55.9	119	33.0						
1.615	4.51				92	56.2 <sup>c</sup>	126	25.9						
1.643	4.63				111	56.2 <sup>c</sup>	134	21.9						
1.685	4.44				116	56.2 <sup>c</sup>	141	19.1						
1.720	4.53				119	36.1 <sup>c</sup>	149	17.2						
2.019	4.92				126	24.8 <sup>c</sup>	159	15.6						
2.061	5.35				134	21.8 <sup>c</sup>	167	14.5						
2.080	5.75				142	19.6	177	13.7						
2.142	5.35				149	18.0	185	13.1						
2.212	5.63				158	16.6	195	12.8						
2.534	7.64				167	15.5	205	12.8						
2.656	7.02				176	14.7	211	14.3						
					186	14.3	213	33.9						
					195	14.2 <sup>c</sup>	210	32.5						
					204	14.7 <sup>c</sup>	205	210						
					210	19.1 <sup>c</sup>	201	19.4						
					213	29.9	198	17.5						
					227	34.1	186	15.2						
					222	33.2 <sup>c</sup>	177	15.0						
					132	34.6 <sup>c</sup>	167	15.4						
CURVE 131 <sup>b</sup>														
					85	10.8								
					110	10.7								
					132	10.3								
CURVE 132 <sup>b</sup>														
					76	55.9								
					92	56.2 <sup>c</sup>								
					111	56.2 <sup>c</sup>								
					116	56.2 <sup>c</sup>								
					119	36.1 <sup>c</sup>								
					126	24.8 <sup>c</sup>								
					134	21.8 <sup>c</sup>								
					142	19.6								
					149	18.0								
					158	16.6								
					167	15.5								
					176	14.7								
					186	14.3								
					195	14.2 <sup>c</sup>								
					204	14.7 <sup>c</sup>								
					210	19.1 <sup>c</sup>								
					213	29.9								
					227	34.1								
					222	33.2 <sup>c</sup>								
					132	34.6 <sup>c</sup>								

<sup>a</sup> Not shown on plot



FIGURE AND TABLE NO. 61R RECOMMENDED THERMAL CONDUCTIVITY OF TIN



RECOMMENDED VALUES\*

T <sub>1</sub>	Single Crystal (// to c-axis)				Polycrystalline			
	k <sub>1</sub>	k <sub>2</sub>	k <sub>3</sub>	k <sub>4</sub>	k <sub>1</sub>	k <sub>2</sub>	k <sub>3</sub>	k <sub>4</sub>
0	0	0	0	0	0	0	0	0
1	204	11900	142	8200	182	10600	182	10600
2	360	20900	250	14400	323	18700	323	18700
3	331	19100	230	13000	297	17200	297	17200
4	202	11700	146	8090	181	10500	181	10500
5	130	7310	90	5200	(117)	(6760)	(117)	(6760)
6	85	4910	59	3410	(76)	(4390)	(76)	(4390)
7	58	3350	40	2310	(52)	(3000)	(52)	(3000)
8	40	2310	28	1620	(36)	(2040)	(36)	(2040)
9	29	1680	20.1	1160	(26)	(1500)	(26)	(1500)
10	21.5	1240	14.9	861	(19.3)	(1120)	(19.3)	(1120)
11	16.5	933	11.4	659	(14.6)	(855)	(14.6)	(855)
12	12.9	745	9.0	520	(11.6)	(670)	(11.6)	(670)
13	10.4	601	7.2	416	(9.3)	(537)	(9.3)	(537)
14	8.5	491	5.9	341	(7.6)	(439)	(7.6)	(439)
15	7.0	404	4.9	293	(6.3)	(364)	(6.3)	(364)
16	5.9	341	4.1	237	(5.3)	(306)	(5.3)	(306)
18	4.5	260	3.1	179	(4.0)	(231)	(4.0)	(231)
20	3.6	208	2.5	144	(3.2)	(185)	(3.2)	(185)
25	2.5	144	1.72	99.4	(2.22)	(128)	(2.22)	(128)
30	2.0	116	1.36	78.6	(1.76)	(102)	(1.76)	(102)
35	1.67	96.5	1.16	67.0	(1.50)	(86.7)	(1.50)	(86.7)
40	(1.50) <sup>‡</sup>	(86.7)	(1.04)	(60.1)	(1.33)	(76.9)	(1.33)	(76.9)
45	(1.37)	(79.2)	(0.95)	(54.9)	(1.23)	(71.1)	(1.23)	(71.1)
50	(1.24)	(74.0)	(0.89)	(51.4)	(1.15)	(66.4)	(1.15)	(66.4)
60	(1.16)	(67.0)	(0.80)	(46.2)	(1.04)	(60.1)	(1.04)	(60.1)
70	(1.07)	(61.8)	(0.74)	(42.8)	(0.96)	(55.5)	(0.96)	(55.5)
80	(1.02)	(58.9)	(0.71)	(41.0)	(0.91)	(52.6)	(0.91)	(52.6)
90	(0.98)	(56.6)	(0.68)	(39.3)	(0.86)	(50.8)	(0.86)	(50.8)
100	(0.95)	(54.9)	(0.66)	(38.1)	(0.85)	(49.1)	(0.85)	(49.1)
150	(0.867)	(50.1)	(0.602)	(34.8)	(0.779)	(45.0)	(0.779)	(45.0)
200	(0.816)	(47.1)	(0.567)	(32.8)	(0.733)	(42.4)	(0.733)	(42.4)
250	(0.775)	(44.8)	(0.538)	(31.1)	(0.696)	(40.2)	(0.696)	(40.2)
273.2	(0.759)	(43.9)	(0.527)	(30.5)	(0.682)	39.4	(0.682)	39.4
300	(0.742)	(42.9)	(0.515)	(29.8)	0.666	38.5	0.666	38.5
350	(0.715)	(41.3)	(0.496)	(28.7)	0.642	37.1	0.642	37.1
400	0.693	40.0	0.481	27.8	0.622	35.9	0.622	35.9
500	0.664	38.4	0.461	26.6	0.596	34.4	0.596	34.4
505.06	0.662	38.3	0.460	26.6	0.595	34.4	0.595	34.4

REMARKS  
 The recommended values are for well-annealed 99.999% pure white tin with residual electrical resistivity  $\rho_0 = 0.000120$ ,  $0.000172$ , and  $0.000133 \mu\Omega \text{ cm}$ , respectively, for single crystal along directions perpendicular and parallel to the c-axis and for polycrystalline tin (characterization by  $\rho_0$  becomes important at temperatures below about 150 K). The recommended values that are supported by experimental thermal conductivity data are thought to be accurate to within 3% of the true values near room temperature and 3 to 1.5% at other temperatures.

\* T<sub>1</sub> in K, k<sub>1</sub> in Watt cm<sup>-1</sup> K<sup>-1</sup>, T<sub>2</sub> in F, and k<sub>2</sub> in Btu hr<sup>-1</sup> ft<sup>-1</sup> F<sup>-1</sup>.  
<sup>‡</sup> Values in parentheses are extrapolated, interpolated, or estimated.

TABLE NO. 61R (continued)

In Liquid State			
$T_1$	$k_1$	$k_2$	$T_2$
505.06	0.303	17.5	449.44
600	0.323	18.7	620.3
700	0.343	19.8	800.3
800	0.364	21.0	980.3
900	0.384	22.2	1160
1000	0.405	23.4	1340
1100	0.425	24.6	1520
1200	0.446	25.8	1700
1300	0.466	26.9	1880
1400	(0.487) <sup>‡</sup>	(28.1)	2060
1500	(0.507)	(29.3)	2240

<sup>‡</sup>  $T_1$  in K,  $k_1$  in Watt  $\text{cm}^{-1} \text{K}^{-1}$ ,  $T_2$  in F, and  $k_2$  in Btu  $\text{hr}^{-1} \text{ft}^{-1} \text{F}^{-1}$ .  
<sup>‡</sup> Values in parentheses are extrapolated.

# THERMAL CONDUCTIVITY OF TITANIUM

FIGURE SHOWS ONLY 18 OF THE CURVES REPORTED IN TABLE

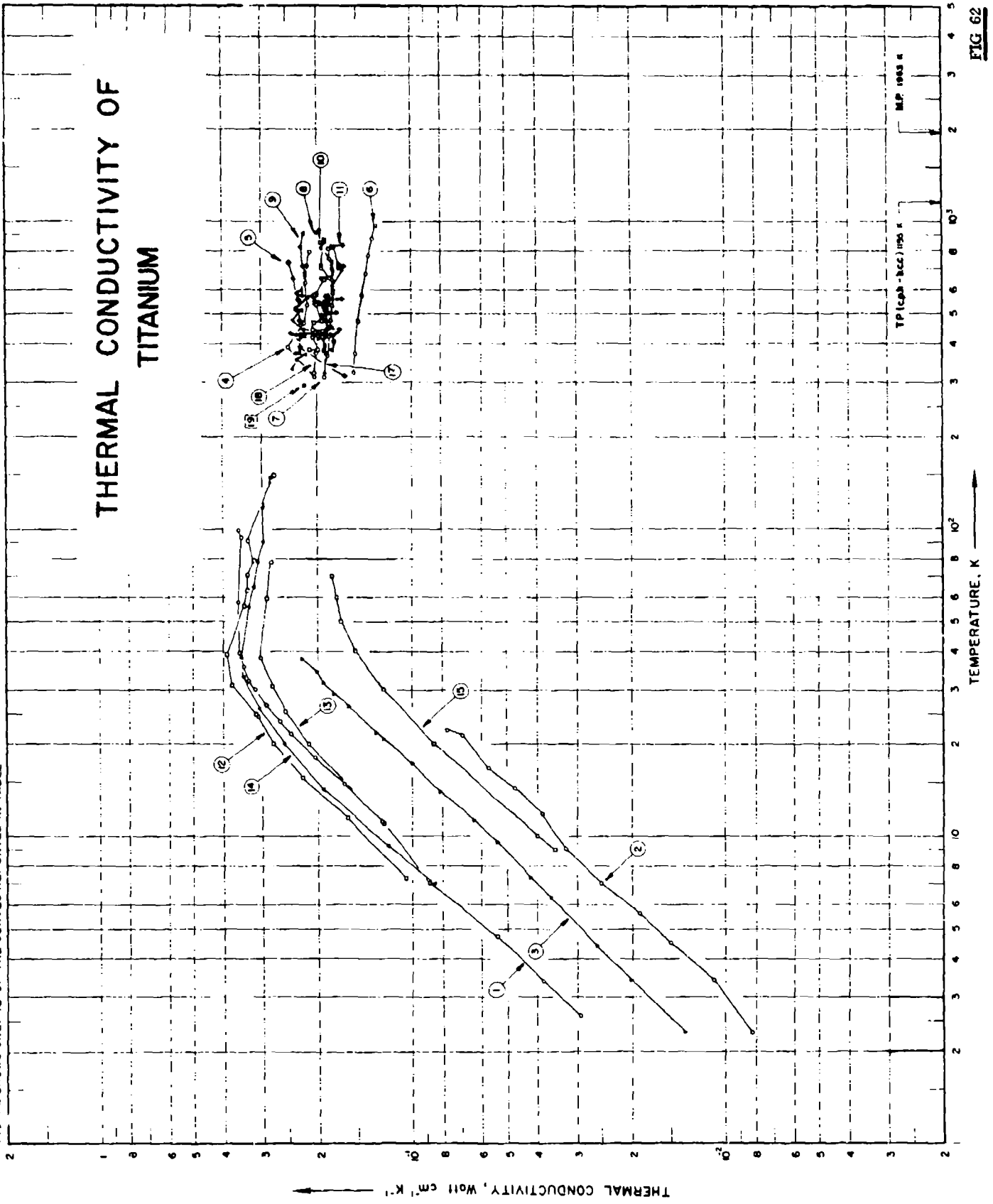


FIG 62

SPECIFICATION TABLE NO. 62 THERMAL CONDUCTIVITY OF TITANIUM

(purity 0.29% each; total impurities 0.50%)

(For Data Reported in Figure and Table No. 62)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
1	122	L	1955	2.6-99	5-10	T13	99.99 pure; single crystal.
2	97	L	1952	2.3-22	2-3	T11	99.9 pure; polycrystalline; supplied by Associated Electrical Industries Research Laboratories.
3	97	L	1952	2.3-38	2-3	T12	99.99 pure; polycrystalline; same supplier as above; annealed.
4	441		1957	312-799	3	Iodide Titanium	99.9 pure; annealed in vacuum at 700 C for 5 hrs; electrical resistivity reported as 24.8, 50.3, 67.2, 79.0, 100.2, and 115.0 $\mu\text{ohm cm}$ at 38.9, 116.0, 196.0, 263.0, 359.3, 445.7, and 526.0 C, respectively.
5	441		1957	315-732			Forged titanium specimen 99.6 pure; annealed in vacuum at 700 C for 5 hrs; electrical resistivity reported as 65.8, 58.8, 73.5, 84.0, 98.2, and 107.5 $\mu\text{ohm cm}$ at 41.5, 136.6, 201.6, 278.3, 376.0, and 459.0 C, respectively.
6	131	C	1953	323-973	2		0.1 Mn, 0.04 Fe, 0.083 C, and 0.01 Mg; annealed at 700 C; Advance (55 Cu-45 Ni) used as comparative material.
7	231, 304	C	1958	311-811	5	A-55 (RC-55)	Commercially pure; in a mill-annealed condition; electrical resistivity reported as 52.63, 72, 83, 92, 101, 110, 118, 125, and 132 $\mu\text{ohm cm}$ at 211, 366, 422, 477, 533, 589, 644, 700, 755, and 811 K, respectively; measured in a vacuum of $5 \times 10^{-5}$ mm Hg.
8	742	E	1964	388-923			99.6 pure (Russian commercial titanium); obtained from the Central Boiler and Turbine Institute; specimen 5 mm in dia and 100 mm long; experiment carried out in vacuum ( $10^{-4}$ - $10^{-5}$ mm Hg); electrical resistivity reported as 47, 64, 82, 99, 117, 133, 143 and 145 $\mu\text{ohm cm}$ at 0, 100, 200, 300, 400, 500, 600 and 650 C, respectively.
9	340	L	1956	332-915	10	T175 A(1)	Commercially pure; 0.75 in. dia rod.
10	340	L	1956	383-858	10	T175 A(2)	99.75 Ti, 0.131 O, 0.07 Fe, 0.06 C, 0.048 N, and 0.0068 H; 0.75 in. dia rod.
11	340	L	1956	375-838	10	RC-55	99.64 Ti, 0.123 O, and 0.12 Fe, 0.0073 H, 0.08 C, and 0.028 N; 0.75 in. dia rod.
12	401	L	1959	7.3-150		T13	99.99 pure; specimen cross section 3.1 x 1.6 mm; supplied by Wingard; annealed in vacuum for 60 hrs at 800 C; ideal electrical resistivity reported as 0.020, 0.075, 0.20, 0.65, 1.4, 2.3, 3.5, 4.85, 6.35, 7.9, 11.2, 14.8, 18.5, 22.1, 25.7, 29.3, 34.8, 39.0, and 43.1 $\mu\text{ohm cm}$ at 20, 25, 30, 40, 50, 60, 70, 80, 90, 100, 120, 140, 160, 180, 200, 220, 250, 273, and 295 K, respectively; electrical resistivity ratio $\rho_{393}/\rho_0 = 21.9$ ; Lorenz function $2.74 \times 10^6 \text{ V}^2 \text{ K}^{-2}$ near 0 K.
13	401	L	1959	11-78		T14	99.99 pure; as rolled; electrical resistivity ratio $\rho_{393}/\rho_0 = 16.4$ ; ideal electrical resistivity 43.8 $\mu\text{ohm cm}$ at 295 K; Lorenz function $2.81 \times 10^6 \text{ V}^2 \text{ K}^{-2}$ near 0 K.
14	401	L	1959	7.0-147		T15	99.99 pure; annealed in vacuum for 60 hrs at 800 C; electrical resistivity ratio $\rho_{393}/\rho_0 = 18.3$ ; ideal electrical resistivity 43.2 $\mu\text{ohm cm}$ at 295 K; Lorenz function $3.14 \times 10^6 \text{ V}^2 \text{ K}^{-2}$ near 0 K.
15	672	±P	1964	9-70			99.9 pure; data taken from smoothed curve.

SPECIFICATION TABLE NO. 62 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition ( weight percent). Specifications and Remarks
16	683	L	1963	0.3-0.9			Single crystal.
17	698	L, C	1961	323-573		Sample A	Normal commercial grade; electrical resistivity reported as 56.0, 65.0, 73.5, 82.5, 90.5 and 98.5 $\mu\text{ohm cm}$ at 50, 100, 150, 200, 250, and 300 C, respectively; energy flow measured both calorimetrically and by using Armco iron as a comparative material.
18	698	L, C	1961	323-573		Sample B	High purity grade; electrical resistivity reported as 51.8, 60.8, 70.0, 79.2, 88.4, and 97.5 $\mu\text{ohm cm}$ at 50, 100, 150, 200, 250, and 300 C, respectively; energy flow measured both calorimetrically and by using Armco iron as a comparative material.
19	698	L, C	1961	293.2		Sample C	Very high purity grade; DPN (Diamond Pyramid Hardness Number) 58-62; electrical resistivity 42.7 $\mu\text{ohm cm}$ at 20 C; energy flow measured both calorimetrically and by using Armco iron as a comparative material.

DATA TABLE NO. 62 THERMAL CONDUCTIVITY OF TITANIUM

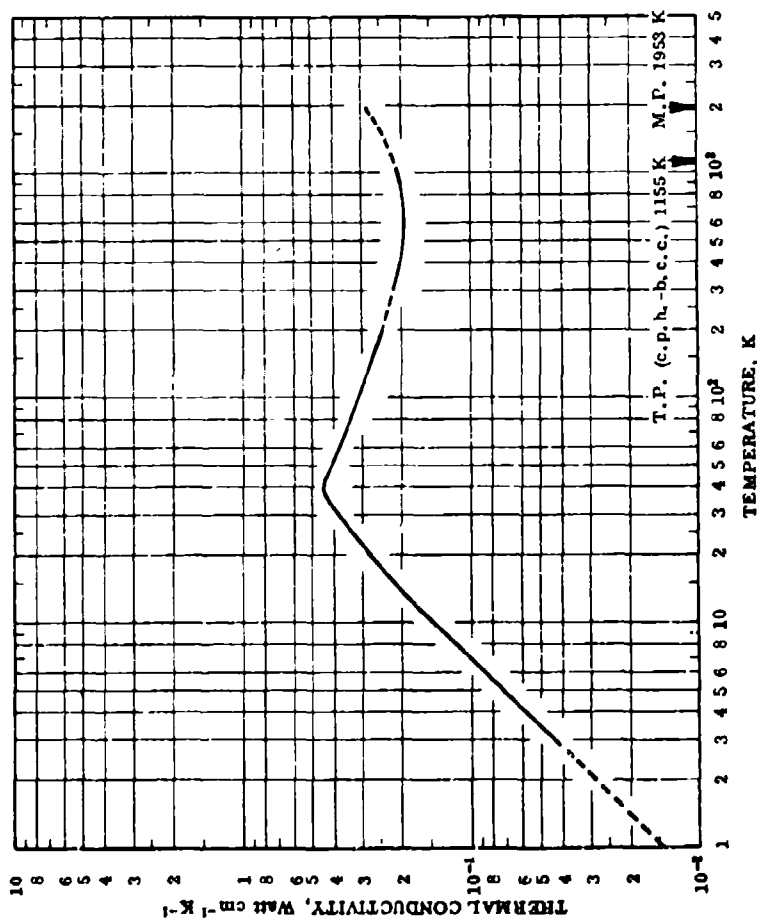
(Impurity <0.20% each; total impurities <0.50%)

(Temperature, T, K; Thermal Conductivity k, Watt cm<sup>-1</sup>K<sup>-1</sup>)

T	k	T	k	T	k	T	k	T	k	T	k
<u>CURVE 1</u>											
2.50	0.0293	20.90	0.124	396.2	0.225	504.6	0.173	7.0	0.085	423.2	0.1865
3.35	0.0344	21.76	0.130	669.6	0.219	504.6	0.186	9.3	0.120	473.2	0.186
4.70	0.0336	26.60	0.160	679.0	0.219	537.9	0.183	14.3	0.193	523.2	0.185
7.01	0.0480	29.12	0.178	715.4	0.222	540.7	0.195	20.0	0.257	573.2	0.184
11.17	0.125	31.60	0.192	826.2	0.225	546.2	0.187	26.0	0.308	<u>CURVE 18</u>	
14.80	0.167	34.40	0.201	915.1	0.221	559.6	0.166	33.0	0.347	323.2	0.205
18.20	0.205	37.70	0.225	915.1	0.221	559.6	0.182	36.0	0.351	373.2	0.201
21.56	0.245	<u>CURVE 10</u>									
23.64	0.268	<u>CURVE 11</u>									
26.75	0.293	383.2	0.214	609.6	0.177	616.5	0.177	65.0	0.320	423.2	0.197
30.10	0.317	384.5	0.197	693.2	0.166	693.2	0.166	78.5	0.310	473.2	0.1925
32.20	0.333	408.2	0.177	420.7	0.206	701.2	0.170	91.0	0.299	523.2	0.189
35.70	0.347	456.2	0.183	420.7	0.191	715.7	0.161	147.0	0.281	<u>CURVE 19</u>	
39.50	0.356	537.2	0.215	440.1	0.190	724.0	0.170	9	0.035	293.2	0.220
57.70	0.361	588.2	0.178	442.9	0.206	821.2	0.176	10	0.04	373.2	0.201
93.80	0.351	738.2	0.178	470.7	0.205	830.4	0.178	20	0.085	423.2	0.197
98.70	0.358	868.2	0.193	475.7	0.193	837.9	0.164	30	0.123	473.2	0.1925
<u>CURVE 2</u>											
2.30	0.0083	314.7	0.164	540.1	0.200	547.1	0.196	9	0.035	373.2	0.201
3.41	0.0109	389.8	0.227	547.1	0.200	547.1	0.200	10	0.04	423.2	0.197
4.51	0.0150	474.8	0.227	579.0	0.204	547.1	0.204	20	0.085	473.2	0.1925
5.61	0.0188	551.5	0.229	581.2	0.200	581.2	0.200	30	0.123	523.2	0.189
7.03	0.0250	649.8	0.238	648.4	0.188	648.4	0.188	40	0.153	573.2	0.184
9.10	0.0326	732.2	0.245	653.4	0.192	653.4	0.192	50	0.17	<u>CURVE 15</u>	
11.87	0.0386	<u>CURVE 12</u>									
14.40	0.0475	332.3	0.240	655.4	0.182	655.4	0.182	25.0	0.315	9	0.035
16.76	0.0571	356.2	0.231	709.0	0.193	709.0	0.193	31.0	0.378	10	0.04
21.21	0.0695	369.3	0.217	717.3	0.193	717.3	0.193	39.0	0.390	20	0.085
22.36	0.0774	417.3	0.228	858.2	0.188	858.2	0.188	56.0	0.345	30	0.123
<u>CURVE 3</u>											
2.31	0.0126	423.4	0.219	858.2	0.195	858.2	0.195	20.0	0.277	50	0.17
3.41	0.0202	473.2	0.148	858.2	0.195	858.2	0.195	24.5	0.310	60	0.175
4.40	0.0259	573.2	0.144	858.2	0.195	858.2	0.195	25.0	0.315	70	0.18
6.32	0.0362	673.2	0.140	858.2	0.195	858.2	0.195	25.0	0.315	<u>CURVE 16</u>	
8.58	0.0533	773.2	0.137	858.2	0.195	858.2	0.195	31.0	0.378	0.378	0.00970
11.32	0.0639	873.2	0.133	858.2	0.195	858.2	0.195	31.0	0.378	0.398	0.00105
13.94	0.0815	973.2	0.129	858.2	0.195	858.2	0.195	56.0	0.345	0.410	0.00111
17.27	0.100	858.2	0.182	858.2	0.195	858.2	0.195	63.0	0.338	0.442	0.00122
<u>CURVE 4</u>											
2.31	0.0126	374.8	0.189	374.8	0.189	374.8	0.189	0.478	0.00139	0.535	0.00160
3.41	0.0202	412.9	0.190	412.9	0.190	412.9	0.190	0.580	0.00168	0.580	0.00168
4.40	0.0259	422.9	0.186	422.9	0.186	422.9	0.186	0.620	0.00173	0.620	0.00173
6.32	0.0362	426.2	0.177	426.2	0.177	426.2	0.177	0.680	0.00186	0.680	0.00186
8.58	0.0533	445.1	0.169	445.1	0.169	445.1	0.169	0.730	0.00202	0.730	0.00202
11.32	0.0639	445.7	0.165	445.7	0.165	445.7	0.165	0.780	0.00222	0.780	0.00222
13.94	0.0815	458.2	0.178	458.2	0.178	458.2	0.178	0.830	0.00263	0.830	0.00263
17.27	0.100	482.3	0.185	482.3	0.185	482.3	0.185	0.880	0.00263	0.880	0.00263
<u>CURVE 7</u>											
3.11	0.189	374.8	0.189	374.8	0.189	374.8	0.189	323.2	0.1865	373.2	0.1875
3.66	0.185	412.9	0.190	412.9	0.190	412.9	0.190	373.2	0.1865		
4.22	0.182	422.9	0.186	422.9	0.186	422.9	0.186				

\* Not shown on plot

FIGURE AND TABLE NO. 62R RECOMMENDED THERMAL CONDUCTIVITY OF TITANIUM



## REMARKS

The recommended values are for well-annealed 99.99% pure titanium with residual electrical resistivity  $\rho_0 = 1.70 \mu\Omega \text{ cm}$  (characterization by  $\rho_0$  becomes important below room temperature). The values below 1.5  $T_m$  are calculated to fit the experimental data by using  $n = 2.6$ ,  $\alpha = 4.32 \times 10^{-5}$ , and  $\beta = 69.5$ . The recommended values that are supported by experimental thermal conductivity data are thought to be accurate to within 5% of the true values near room temperature and 5 to 15% at other temperatures.

RECOMMENDED VALUES\*  
(For Polycrystalline)

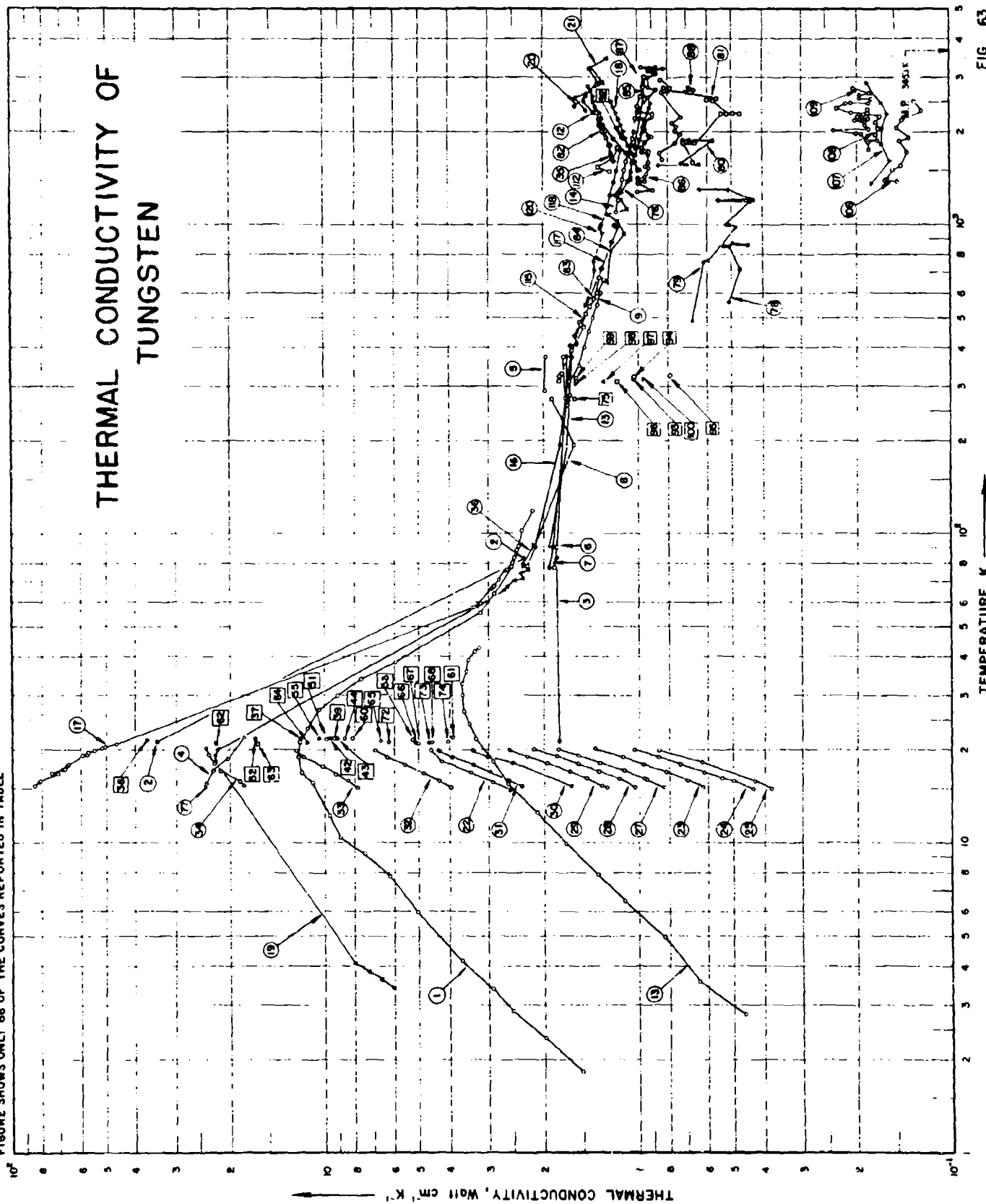
T, °C	$k_1$	$k_2$	$T_2$	$T_1$	$k_1$	$k_2$	$T_2$
0	0	0	-459.7	500	0.197	11.4	440.3
1	(0.0144)†	(0.832)	-457.9	600	0.194	11.2	620.3
2	(0.0288)	(1.66)	-456.1	700	0.194	11.2	800.3
3	0.0432	2.50	-454.3	800	0.197	11.4	980.3
4	0.0576	3.33	-452.5	900	0.202	11.7	1160
5	0.0719	4.15	-450.7	1000	(0.207)	(12.0)	1340
6	0.0863	4.99	-448.9	1100	(0.213)	(12.3)	1520
7	0.101	5.84	-447.1	1200	(0.220)	(12.7)	1700
8	0.115	6.64	-445.3	1300	(0.228)	(13.2)	1880
9	0.129	7.45	-443.5	1400	(0.236)	(13.6)	2060
10	0.144	8.32	-441.7	1500	(0.245)	(14.2)	2240
11	0.158	9.13	-439.9	1600	(0.253)	(14.6)	2420
12	0.172	9.94	-438.1	1700	(0.262)	(15.1)	2600
13	0.186	10.7	-436.3	1800	(0.271)	(15.7)	2780
14	0.200	11.6	-434.5	1900	(0.280)	(16.2)	2960
15	0.214	12.4	-432.7				
16	0.227	13.1	-430.9				
18	0.254	14.7	-427.3				
20	0.279	16.1	-423.7				
25	0.337	19.5	-414.7				
30	0.382	22.1	-405.7				
35	0.411	23.7	-396.7				
40	0.422	24.4	-387.7				
45	0.416	24.0	-378.7				
50	0.401	23.2	-369.7				
60	0.377	21.8	-351.7				
70	0.356	20.6	-333.7				
80	0.339	19.6	-315.7				
90	0.324	18.7	-297.7				
100	0.312	18.0	-279.7				
150	0.270	15.6	-189.7				
200	(0.245)	(14.2)	-99.7				
250	(0.229)	(13.2)	-9.7				
273.2	(0.224)	(12.9)	32.0				
300	0.219	12.7	80.3				
350	0.210	12.1	170.3				
400	0.204	11.8	260.3				

\*  $T_1$  in K,  $k_1$  in  $\text{Watt cm}^{-1} \text{K}^{-1}$ ,  $T_2$  in F, and  $k_2$  in  $\text{Btu hr}^{-1} \text{ft}^{-1} \text{F}^{-1}$ .

† Values in parentheses are extrapolated, interpolated, or estimated.

# THERMAL CONDUCTIVITY OF TUNGSTEN

FIGURE SHOWS ONLY 66 OF THE CURVES REPORTED IN TABLE





## SPECIFICATION TABLE NO. 63 THERMAL CONDUCTIVITY OF TUNGSTEN

(Impurity &lt; 0.20% each, total impurities &lt; 0.50%)

[ For Data Reported in Figure and Table No. 63 ]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition ( weight percent), Specifications and Remarks
1	150	L	1957	1.8-1119		W 1b	0.01 Mo, traces of Fe, Si and Cu; 4 mm dia rod; annealed in vacuum at 1350 C; electrical resistivity ratio $\rho(295K)/\rho_0 = 169$ ; residual electrical resistivity 0.0315 $\mu\text{ohm cm}$ .
2	57	L	1927	21, 83		W 1	High purity; single crystal; electrical resistivity reported as 0.00589 and 0.681 $\mu\text{ohm cm}$ at -252 and -190 C, respectively.
3	57	L	1927	21, 83		W 2	Less pure than the above specimen; single crystal; electrical resistivity reported as 0.266, 1.024, and 5.29 $\mu\text{ohm cm}$ at -252, -190, and 0 C, respectively.
4	16	L	1936	16-22			Very pure; electrical resistivity ratio $\rho(273K)/\rho_0 = 2.18 \times 10^3$ .
5	8	F	1914	290, 373		Pladuram	Pure; 0.0600 cm dia x 28.5 cm long; electrical resistivity reported as 5.206 and 7.562 $\mu\text{ohm cm}$ at 0 and 100 C, respectively.
6	24	E	1943	77-373		2	High purity; 0.0250 cm dia x 40.85 cm long; drawn; aged at 2400 and 2600 C; electrical resistivity reported as 0.6736, 0.9132, 3.18, 5.034, and 7.392 $\mu\text{ohm cm}$ at 77.4, 90.2, 193, 273.2, and 372.8 K, respectively.
7	24	E	1943	77-373		8	High purity; 0.0250 cm dia x 40.00 cm long; drawn; aged at 2300 C; electrical resistivity reported as 0.6135, 0.8558, 5.035, and 7.429 $\mu\text{ohm cm}$ at 77.36, 90.2, 273.2, and 373.1 K, respectively.
8	99	E	1936	78-273			Commercially pure; 0.00254 cm dia x 14.8 cm long; aged at white heat for several hrs in vacuum, etched; Lorenz function reported as 2.12, 2.68, and 3.48 $\text{V}^2 \text{K}^{-2}$ at 78, 194, and 273 K, respectively; measured in a vacuum of $< 10^{-4}$ mm Hg.
9	87	E	1936	240-600			Pure; 0.00499 cm dia wire; annealed at 2400 K; data taken from smoothed curve.
10	78	E	1931	276-280	2		Commercially pure; 0.1022 cm dia x 17.63 cm long; supplied by General Electric Co.; annealed at 220 C; electrical conductivity $16.7 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 273 K; measured in a vacuum of $< 10^{-4}$ mm Hg.
11	78	E	1931	276-286	2		Commercially pure; 0.1022 cm dia x 19.96 cm long; supplied by General Electric Co.; annealed at 1300 C; electrical conductivity $17.7 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 273 K; measured in a vacuum of $< 10^{-4}$ mm Hg.
12	151	E	1914	1500-2500			Pure; 0.0209 cm dia filament; data taken from smoothed curve.
13	97	L	1952	2.8-43	2-3	JM2260, W1	99.99 pure; polycrystalline; annealed.
14	106	E	1941	1100-2000			Traces of metallic impurities; aged at 2700 K for 2 hrs; electrical resistivity reported as 30.0, 35.5, 42.3, 49.4, 58.2, and 66.6 $\mu\text{ohm cm}$ at 1180, 1350, 1570, 1800, 2050, and 2300 K, respectively; data taken from smoothed curve.
15	79	E	1933	90-373		W:	99.96% W, traces of Si, Ta and V; single crystal; 7.846 cm x 0.01053 cm <sup>2</sup> ; electrical resistivity reported as 0.892, 3.22, 4.98, and 7.35 $\mu\text{ohm cm}$ at -183.00, -78.50, 0, and 100 C, respectively; heat flow parallel to crystal axis.

SPECIFICATION TABLE NO. 63 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
16	79	E	1933	90-373		W2	99.96% W, traces of Si, Ta and V; single crystal; 7.940 cm x 0.01022 cm <sup>2</sup> ; electrical resistivity reported as 0.843, 3.17, 4.94, and 7.29 μhm cm at -183.00, -78.50, 0, and 100 C, respectively; heat flow at 45 degrees to the crystal axis.
17	62	L	1938	15-88			High purity; single crystal; specimen axis in [111] direction; electrical resistivity reported as 0.00236, 0.00315, 0.00417, 0.00422, 0.1425, 0.3230, 0.3475, 0.3565, 0.3945, 0.4230, 0.4460, 0.4990, 0.5110, 0.5595, 0.6065, 0.7040, and 0.8070 μhm cm at 14.14, 17.55, 20.36, 20.42, 50.55, 53.50, 65.20, 65.80, 68.20, 69.80, 71.30, 74.30, 74.95, 77.40, 80.10, 85.05, and 90.15 K, respectively.
18	44	E	1925	1500-2500			Pure; density 19.3 g cm <sup>-3</sup> at room temp; electrical resistivity reported as 5.64, 8.06, 13.54, 19.47, 25.70, 32.02, 38.52, 45.22, 52.08, 59.10, 66.25, 73.55, 81.0, 88.5, 96.2, 103.8, 111.7, and 115.7 μhm cm at 300, 400, 600, 800, 1000, 1200, 1400, 1600, 1800, 2000, 2200, 2400, 2600, 2800, 3000, 3200, 3400, and 3500 K, respectively.
19	272	L	1957	3.4-76		1-38	Pure single crystal; specimen axis in [100] direction; electrical resistivity reported as 0.0123, 0.0141, 0.2155, 0.3551, 0.4297, 0.5087, and 0.5921 μhm cm at 14.50, 20.42, 55.35, 63.95, 68.51, 72.97, and 77.35 K, respectively; heat flow parallel to (±5°) the crystal axis.
20	255	E	1960	2400-3194		1	0.04 Mo, 0.006 O, 0.005 Ti, 0.005 Ni, 0.004 Fe, and 0.027 others; prepared by pressing and sintering metal powder; hot-worked.
21	255	E	1960	2344-3451		2	99.95 W, 0.04 Mo, 0.002 Cu, and 0.008 others; prepared by pressing and sintering metal powder; hot-worked.
22	398	L	1949	15-20			High purity; single crystal; specimen axis in [111] direction; electrical resistivity reported as 0.319, 0.281, 0.236, and 0.187 μhm cm at 14.20, 15.98, 18.05, and 20.48 K, respectively; heat flow at 45 degrees to crystal axis; measured in a field of 10.3 kilosterds perpendicular to specimen axis.
23	398	L	1949	15-20			As above but measured in a field of 26.39 kilosterds; electrical resistivity reported as 1.932, 1.810, 1.677, 1.527, 1.382, 1.239, and 1.060 μhm cm at 14.21, 15.07, 15.96, 17.02, 18.06, 19.11, and 20.48 K, respectively.
24	398	L	1949	15-20			As above but measured in a field of 32.65 kilosterds; electrical resistivity reported as 2.926, 2.737, 2.541, 2.317, 2.099, 1.882, and 1.594 μhm cm at 14.20, 15.07, 15.98, 17.02, 18.04, 19.08, and 20.51 K, respectively.
25	398	L	1949	15-20			As above but measured in a field of 36.27 kilosterds; electrical resistivity reported as 3.572, 3.347, 3.106, 2.565, and 1.967 μhm cm at 14.21, 15.07, 15.99, 18.05, and 20.45 K, respectively.
26	62	L	1938	15-20			High purity; single crystal; specimen axis parallel to (1, 1, 1) direction; measured in a transverse magnetic field of 23.85 kilogauss perpendicular to specimen axis.
27	62	L	1938	15-20			The above specimen measured in a transverse magnetic field of 21.83 kilogauss.
28	62	L	1938	15-20			The above specimen measured in a transverse magnetic field of 18.96 kilogauss.

SPECIFICATION TABLE NO. 63 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
29	62	L	1938	15-20			The above specimen measured in a transverse magnetic field of 16.69 kilogauss.
30	62	L	1938	15-20			The above specimen measured in a transverse magnetic field of 13.82 kilogauss.
31	62	L	1938	15-20			The above specimen measured in a transverse magnetic field of 11.44 kilogauss.
32	62	L	1938	15-20			The above specimen measured in a transverse magnetic field of 8.18 kilogauss.
33	62	L	1938	15-20			The above specimen measured in a transverse magnetic field of 5.22 kilogauss.
34	62	L	1938	15-20			The above specimen measured in a transverse magnetic field of 2.61 kilogauss.
35	253	E	1959	1600-2700			Spectrographically pure; 0.010 in. dia; electrical resistivity reported as 46.4, 53.3, 63.3, 71.4, and 83.7 $\mu\text{ohm cm}$ at 1622, 1925, 2230, 2471, and 2965 K, respectively.
36	448	L	1937	22-31		W 1	7 cm x 0.106 cm <sup>2</sup> specimen axis at 8 degrees to the [110] direction; measured in a vacuum of $10^{-4}$ - $10^{-5}$ mm Hg.
37	448	L	1937	21.7		W 1	Measured at H (the transverse magnetic field strength) = 4850 oersteds and $\theta$ (the angle between the magnetic field direction and a line perpendicular to the rod axis) = $-90^\circ$ at which H parallel to [111] direction.
38	448	L	1937	27.8		W 1	The above specimen measured at H = 6730 oersteds and $\theta = -90^\circ$ .
39	448	L	1937	21.8		W 1	The above specimen measured at H = 6100 oersteds and $\theta = -90^\circ$ .
40	448	L	1937	21.7		W 1	The above specimen measured at H = 6100 oersteds and $\theta = -60^\circ$ .
41	448	L	1937	21.7		W 1	The above specimen measured at H = 6100 oersteds and $\theta = -50^\circ$ .
42	448	L	1937	21.7		W 1	The above specimen measured at H = 6100 oersteds and $\theta = -40^\circ$ .
43	448	L	1937	21.8		W 1	The above specimen measured at H = 6100 oersteds and $\theta = -20^\circ$ .
44	448	L	1937	21.5		W 1	The above specimen measured at H = 6100 oersteds and $\theta = 0^\circ$ at which H perpendicular to [111] direction.
45	448	L	1937	21.8		W 1	The above specimen measured at H = 6100 oersteds and $\theta = +20^\circ$ .
46	448	L	1937	21.8		W 1	The above specimen measured at H = 6100 oersteds and $\theta = +40^\circ$ .
47	448	L	1937	21.8		W 1	The above specimen measured at H = 6100 oersteds and $\theta = +60^\circ$ .
48	448	L	1937	21.8		W 1	The above specimen measured at H = 6100 oersteds and $\theta = +70^\circ$ .
49	448	L	1937	21.8		W 1	The above specimen measured at H = 6100 oersteds and $\theta = +80^\circ$ .
50	448	L	1937	21.8		W 1	The above specimen measured at H = 6100 oersteds and $\theta = +90^\circ$ .
51	448	L	1937	21.7		W 1	The above specimen measured at H = 4850 oersteds and $\theta = +70^\circ$ .
52	448	L	1937	21.6		W 1	The above specimen measured at H = 2520 oersteds and $\theta = +70^\circ$ .
53	448	L	1937	21.5		W 1	The above specimen with the magnetic field removed.

SPECIFICATION TABLE NO. 63 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
54	436	L	1938	22.2		W 1	Specimen axis in [110] direction; electrical resistivity 0.00463 $\mu\text{ohm cm}$ at $-252.83\text{ C}$ ; measured at H (the transverse magnetic field strength) = 0 and $\theta$ (the angle between the magnetic field direction and a line perpendicular to the rod axis) = $-56^\circ$ at which H parallel to [110] direction.
55	436	L	1938	21.7		W 1	The above specimen measured at H = 6100 oersteds and $\theta = -56^\circ$ .
56	436	L	1938	21.5		W 1	The above specimen measured at H = 12200 oersteds and $\theta = -56^\circ$ .
57	436	L	1938	21.5		W 1	The above specimen measured at H = 0 oersteds and $\theta = +70^\circ$ at which H perpendicular to [111] direction.
58	436	L	1938	21.6		W 1	The above specimen measured at H = 2520 oersteds and $\theta = +70^\circ$ .
59	436	L	1938	21.7		W 1	The above specimen measured at H = 4850 oersteds and $\theta = +70^\circ$ .
60	436	L	1938	21.8		W 1	The above specimen measured at H = 5100 oersteds and $\theta = +70^\circ$ .
61	436	L	1938	21.8		W 1	The above specimen measured at H = 12200 oersteds and $\theta = +70^\circ$ .
62	436	L	1938	21.0		W 13a	Specimen axis in [100] direction; electrical resistivity 0.01054 $\mu\text{ohm cm}$ at $-252.82\text{ C}$ ; measured at H (the transverse magnetic field strength) = 0 and $\theta$ (the angle between the magnetic field direction and a line perpendicular to the rod axis) = $-5^\circ$ at which H nearly parallel to [100] direction.
63	436	L	1938	20.8		W 13a	The above specimen measured at H = 2280 oersteds and $\theta = -5^\circ$ ; electrical resistivity 0.01980 $\mu\text{ohm cm}$ at $-252.82\text{ C}$ .
64	436	L	1938	21.0		W 13a	The above specimen measured at H = 4490 oersteds and $\theta = -5^\circ$ ; electrical resistivity 0.0346 $\mu\text{ohm cm}$ at $-252.82\text{ C}$ .
65	436	L	1938	21.4		W 13a	The above specimen measured at H = 8750 oersteds and $\theta = -5^\circ$ ; electrical resistivity 0.0760 $\mu\text{ohm cm}$ at $-252.82\text{ C}$ .
66	436	L	1938	21.0		W 13a	The above specimen measured at H = 10880 oersteds and $\theta = -5^\circ$ ; electrical resistivity 0.1044 $\mu\text{ohm cm}$ at $-252.82\text{ C}$ .
67	436	L	1938	21.0		W 13a	The above specimen measured at H = 11080 oersteds and $\theta = -5^\circ$ .
68	436	L	1938	21.1		W 13a	The above specimen measured at H = 12200 oersteds and $\theta = -5^\circ$ .
69	436	L	1938	21.0		W 13a	The above specimen measured at H = 0 oersteds and $\theta = -50^\circ$ at which H parallel to [110] direction.
70	436	L	1938	20.8		W 13a	The above specimen measured at H = 2280 oersteds and $\theta = -50^\circ$ .
71	436	L	1938	21.0		W 13a	The above specimen measured at H = 4490 oersteds and $\theta = -50^\circ$ .
72	436	L	1938	21.1		W 13a	The above specimen measured at H = 8750 oersteds and $\theta = -50^\circ$ .
73	436	L	1938	21.0		W 13a	The above specimen measured at H = 10880 oersteds and $\theta = -50^\circ$ .
74	436	L	1938	21.1		W 13a	The above specimen measured at H = 12200 oersteds and $\theta = -50^\circ$ .

SPECIFICATION TABLE NO. 63 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)	Specifications and Remarks
75	449	L	1917	274.2				0.1 mm dia; tempered for 20 hrs at 225 C; electrical resistivity 0.204 $\mu\text{hm cm}$ at 1 C; Lorenz function $2.88 \times 10^{-8} \text{ V}^2 \text{ K}^{-1}$ at 1 C.
76	602	E	1961	1173-2473	<6			Spectrally pure; two 2.2 mm dia wires used as the test materials; etched and primarily annealed in a high vacuum at 1700 C for 1 hr; electrical resistivity reported as 29.8, 33.1, 38.4, 45.3, 52.0, 59.5, 67.3, and 76.7 $\mu\text{hm cm}$ at 900, 1000, 1200, 1400, 1600, 1800, 2000, and 2200 C, respectively; measured in a high vacuum.
77	579	L	1936	16-22		C-54		No details reported.
78	603	R	1962	559-860	2-4	C-54		0.003 Fe, 0.0026 Si, 0.0020 O, 0.0010 S, 0.0010 P and Ni, Cu, H and N; specimen 0.75 in. O.D., 0.25 in. I.D., and 0.75 in. long; arc-cast; maximum exposure temp 2255 C; density 18.87 g $\text{cm}^{-3}$ (98.4% of theoretical).
79	603	R	1962	484-1287	2-4	C-55		The above specimen; 2nd run.
80	603	R	1962	1555-1872	2-4	C-85		Similar to the above specimen.
81	603	R	1962	1571-2939	2-4	C-86		Similar to the above specimen; the specimen melted at temp beyond 2255 C (probably because of the carbon eutectic formation, the carbon might come from furnace vapor).
82	651	E	1925	1800-2800		JM 740		Pure wire.
83	652	C	1961	323-673		JM 740		Spectrographically standardized tungsten; JM 740 of Johnson Matthey and Co.; about 4 mm in dia and 10 cm in length; electrical resistivity reported as 5.45, 6.1, 7.3, 9.8, 12.45, 15.2, 18.1, 21.4, 24.6, 27.8, 30.9, 34.3, 37.7, 41.4, 45.1, 49.7, and 51.8 $\mu\text{hm cm}$ at 20, 50, 100, 200, 300, 400, 500, 600, 700, 800, 900, 1000, 1100, 1200, 1300, 1400, and 1450 C, respectively; Armco iron used as comparative material.
84	653	E	1963	645-1660				Single crystal.
85	654	P	1965	1300-2900				99.5 pure; impurities: Fe, Mo and traces of other elements; 1.5 mm thick disc cut from a s-veged rod; from General Electric Co. Osram Lamp Works; average grain size (after testing) 46 $\mu$ ; density 19.3 g $\text{cm}^{-3}$ ; thermal conductivity values calculated from thermal diffusivity measurements using specific heat data of Kubaschewski, O. and Evans, L. L. (Metallurgical Thermochemistry, Pergamon, 1956).
86	669	L	1964	1283-3223				Short rod; electrical resistivity reported as 33.0, 36.0, 39.6, 42.0, 45.2, 48.6, 52.1, 56.0, 59.4, 63.2, 67.0, 71.0, 74.4, 78.0, 81.6, 85.2, 89.0, 92.6, 96.2, and 103.6 $\mu\text{hm cm}$ at 1000, 1100, 1200, 1300, 1400, 1500, 1600, 1700, 1800, 1900, 2000, 2100, 2200, 2300, 2400, 2500, 2600, 2700, 2800, 2900, and 3000 C, respectively.
87	656	-	1963	1200-3000				<0.1% impurities; cylindrical specimen; thermal conductivity values calculated from measured heat flow and specific radiation loss.
88	667	E	1962	1615-2780	$\pm 10$			Spectrographically pure; 0.10 in. dia; electrical resistivity reported as 40.0, 50.0, 56.6, 66.6, and 80.6 $\mu\text{hm cm}$ at 1545, 1812, 2087, <del>2087</del> and 2618 K, respectively; measured in a vacuum of $<10^{-6}$ mm Hg.

SPECIFICATION TABLE NO. 62 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
89	668	E	1964	1900			Foil of 60 $\mu$ thick; wire rider of 0.2 mm in dia placed on that part of the foil where temp was constant; circular diaphragms used in optical pyrometer system.
90	668	E	1964	1900			Foil of 60 $\mu$ thick; rider dia 0.3 mm, circular diaphragms in system.
91	668	E	1964	1900			Foil of 60 $\mu$ thick; rider dia 0.2 mm, slit diaphragms in system.
92	668	E	1964	1900			Wire of 6.2 mm in dia; rider dia 0.2 mm, slit diaphragms in system.
93	688	L	1964	317.2	5	LASL; set No. 1, sample 1	An oxide layer on the surface and $< 0.0050$ oxide inside the specimen; porous right cylinder prepared from tungsten powder obtained from Powder Metallurgical Group of Los Alamos Scientific Lab; material hydrostatically pressed in a plastic sack with 30,000 psi initial pressure, machined and sintered at 1500 C for 2 hrs in a hydrogen reducing atm; particle size 0.8 micron; 72.3% theo. density; electrical resistivity 10.4 $\mu$ hm cm at 20 C.
94	688	L	1964	323.2	5	LASL; set No. 1, sample 2	Similar to the above specimen except 72.1% theo. density, and 10.6 $\mu$ hm cm electrical resistivity at 20 C.
95	686	L	1964	326.2	5	LASL; set No. 1, sample 3	Similar to the above specimen except 1350 C sintering temp, 63.2% theo density, and 13.3 $\mu$ hm cm electrical resistivity at 20 C.
96	688	L	1964	311.2	5	LASL; set No. 1, sample 6	Similar to the above specimen except 1575 C sintering temp, 78.1% theo. density, and 9.1 $\mu$ hm cm electrical resistivity at 20 C.
97	688	L	1964	311.2	5	LASL; set No. 1, sample 7	Similar to the above specimen except 1625 C sintering temp, 83.6% theo. density, and 8.2 $\mu$ hm cm electrical resistivity at 20 C.
98	688	L	1964	308.2	5	LASL; set No. 1, sample 11	Similar to the above specimen except sintered at 1700 C for 9 hrs, 95.3% theo. density, 6.2 $\mu$ hm cm electrical resistivity at 20 C, and the ratio of isolated pores to total pores $\approx 0.9$ .
99	688	L	1964	320.2	5	LASL; set No. 1 sample 12	Similar to the above specimen except 95.5% theo. density, and 6.3 $\mu$ hm cm electrical resistivity at 20 C.
100	688	L	1964	319.2	5	LASL; set No. II, sample 2	Similar to the above specimen except sintered at 1700 C for 3 hrs, particle size 2-4.5 microns, 74.4% theo. density, and 10.5 $\mu$ hm cm electrical resistivity at 20 C.
101	775	-	1966	1930-2933	$\sim 15$	S 1	99.8 $\mu$ pure; cylindrical specimen 1.52 in. in dia, 0.502 in. thick; polished; thermal conductivity determined by equating the axial heat flux within the specimen to the radiation flux at the center of the top surface.
102	775	-	1966	2005-2983	$\sim 15$	S 2	Similar to above; except dimensions 1.006 in. dia, 0.504 in. thick.
103	775	-	1966	2138-2978	$\sim 15$	S 3	Similar to above except dimensions 1.0066 in. dia, 0.356 in. thick.
104	775	-	1966	2086-3075	$\sim 15$	S 4	Similar to above except dimensions 0.804 in. dia, 0.284 in. thick.
105	775	-	1966	1506-2150	$\sim 15$	P 1	Unknown purity; cylindrical specimen; 1.52 in. dia, 0.538 in. thick; fabricated by gravity sintering tungsten particles 0.006 to 0.01 in. in size; fired for a long duration at $> 2478$ K; porosity 55%; thermal conductivity data determined by the same method as above.

SPECIFICATION TABLE NO. 63 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
106	775	-	1966	1384-2569	~15	P 2	Similar to above except dimensions 1.45 in. dia, 0.507 in. thick, and 45% porosity.
107	775	-	1966	1366-2894	~15	P 3	Similar to above except dimensions 0.975 in. dia, 0.379 in. thick, and 46% porosity.
108	775	-	1966	1958-2366	~15	P 4	Similar to above except dimensions 1.03 in. dia, 0.302 in. thick, and 42% porosity.
109	775	-	1966	1755-2753	~15	P 5	Similar to above except dimensions 0.80 in. dia, 0.25 in. thick, and 42% porosity.
110	709	E	1962	1173-2473			1 mm in dia, 30 mm long; electrical resistivity reported as 0.295, 0.23, 0.258, and 0.293 $\mu\text{ohm cm}$ at 1000, 1400, 1800, and 2200 C, respectively.
111	905, 906	E	1965	1618-2081		Sample No. 1	Foil strip; 2 mm x 60 $\mu$ x 20 mm.
112	905, 906	E	1965	1490-1779		Sample No. 2	Foil strip; 3 mm x 60 $\mu$ x 20 mm.
113	905, 906	E	1965	1979-2999		Sample No. 3	Specimen 0.3 mm in dia and 20 mm long.
114	905, 906	E	1965	1963-2384		Sample No. 4	Specimen 0.2 mm in dia and 20 mm long.
115	841	L	1967	313-664		I	99.99 W, 0.01 Mo, trace Si and Cu; 0.4 cm dia x 10 cm long; supplied by Johnson Matthey Co.; measured in the intermediate-temp apparatus.
116	841	C	1967	451-751		I	The above specimen measured in the high-temp apparatus; Armco iron used as comparative material.
117	841	C	1967	405-992		D	Similar to above.
118	907, 908	P	1966	1000-2000			99.95 W and 0.035 Mo; forged rod specimen 10 mm in dia and 80 mm long; density 19.17 $\text{g cm}^{-3}$ ; thermal conductivity values calculated from measured data of thermal diffusivity using specific heat data taken from Hoch, M. and Johnston, H. L. (J. Phys. Chem., 65, 855, 1961).
119	909	C	1966	285-500	$\pm 5$	I	0.026 O, 0.030 Mo, <0.005 Si, 0.001 each of Cu and Ag, <0.001 each of Al, Ca, Fe, Mg, Mn, and Ni, and <0.0005 N; tungsten sheets of ~0.060 in. thick supplied by Fansteel Metallurgical Corp; specimen dimensions 1.000 in. dia x 1.250 in. long; squares cut from the sheets clamped together to form cubes, single welds perpendicular to the sheets made at opposite ends with an inert-gas arc welder, machined to size with the sheets parallel to the cylinder axis; thermocouple holes drilled at 75 degrees to the sheets; density 19.21 $\text{g cm}^{-3}$ at 26.3 C; Armco iron used as comparative material; measured in a helium atm with diatomaceous insulation.
120	909	C	1966	549-972	$\pm 5$	I	Second run of the above specimen with thermatomic carbon insulation.
121	909	C	1966	823-1042	$\pm 5$	I	Same as above, third run.
122	909	R	1966	1266-1997	$\pm 7$	2	Prepared from the same material as the above specimen; consisted of 32 one-inch dia discs with 0.25 in. holes in their centers, the central 16 discs used as test specimen.
123	909	R	1966	1451-2033	$\pm 7$	2	Second run of the above specimen.

SPECIFICATION TABLE NO. 63 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
124	849	-	1966	1513-1930		No. 1	0.0007 C, 0.0005 each of N and O, and 0.00005 H; specimen 2.5339 cm in dia and 0.27 cm long; density 18.89 g cm <sup>-3</sup> , thermal conductivity derived from the temp distribution on the flat surface of the cylindrical disc specimen heated in high vacuum (10 <sup>-6</sup> mm Hg) by high frequency induction.
125	849	-	1966	1572-1905		No. 2	Similar to the above specimen except specimen 2.4785 cm in dia and 0.2714 cm thick with density 19.03 g cm <sup>-3</sup> .
126	849	-	1966	1836-2608		No. 3	Similar to the above specimen except specimen 2.0901 cm in dia and 0.27 cm thick with avg grain size 0.035 mm dia and density 19.23 g cm <sup>-3</sup> .
127	996	L	1967	87-377			Spectroscopically standardized; 4 mm dia x 10 cm long; supplied by Johnson Matthey Co.; electrical resistivity reported as 0.24, 0.83, 1.08, 3.32, 5.61, and 733 μohm cm at 5, 78, 91, 195, 298, and 374 K, respectively. (Tabulated data received from author.)





DATA TABLE NO. 63 (continued)

T	k	T	k	T	k	T	k	T	k	T	k	T	k
1600	1.19	<u>CURVE 43</u>	21.76	9.21	<u>CURVE 55</u>	21.72	10.5	<u>CURVE 67</u>	20.97	5.02	<u>CURVE 76 (cont.)</u>	1571.0	0.663
1700	1.22	<u>CURVE 44</u>			<u>CURVE 56</u>			<u>CURVE 68</u>			<u>CURVE 77</u>	1844.3	0.711
1800	1.24	<u>CURVE 45</u>	21.80	8.8	<u>CURVE 57</u>	21.54	5.27					1944.3	0.656
1900	1.25	<u>CURVE 46</u>			<u>CURVE 58</u>							2263.7	0.538
2000	1.27	<u>CURVE 47</u>	21.75	9.06								2263.7	0.541
2100	1.30	<u>CURVE 48</u>										2263.7	0.493
2200	1.30	<u>CURVE 49</u>	21.75	8.3	<u>CURVE 60</u>	21.50	36.7	<u>CURVE 70</u>	21.02	22.2		1500	1.05
2300	1.31	<u>CURVE 50</u>			<u>CURVE 61</u>						<u>CURVE 85</u>	1660	1.00
2400	1.37	<u>CURVE 51</u>	21.75	9.2	<u>CURVE 62</u>	21.55	16.6	<u>CURVE 71</u>	20.80	16.4			
2500	1.39	<u>CURVE 52</u>			<u>CURVE 63</u>			<u>CURVE 72</u>					
2600	1.39	<u>CURVE 53</u>	21.75	8.2	<u>CURVE 64</u>	21.79	8.2	<u>CURVE 73</u>	20.97	11.2			
2700	1.41	<u>CURVE 54</u>			<u>CURVE 65</u>			<u>CURVE 74</u>					
		<u>CURVE 55</u>	21.79	8.2	<u>CURVE 66</u>	21.79	8.2	<u>CURVE 75</u>	21.14	6.30			
		<u>CURVE 56</u>			<u>CURVE 67</u>			<u>CURVE 76</u>					
		<u>CURVE 57</u>	21.76	8.7	<u>CURVE 68</u>	21.76	8.7						
		<u>CURVE 58</u>			<u>CURVE 69</u>								
		<u>CURVE 59</u>	21.75	9.16	<u>CURVE 70</u>	21.75	9.16						
		<u>CURVE 60</u>			<u>CURVE 71</u>								
		<u>CURVE 61</u>	21.79	9.4	<u>CURVE 72</u>	21.71	10.0						
		<u>CURVE 62</u>			<u>CURVE 73</u>								
		<u>CURVE 63</u>	21.74	10.4	<u>CURVE 74</u>	21.74	10.4						
		<u>CURVE 64</u>			<u>CURVE 75</u>								
		<u>CURVE 65</u>	21.55	16.6	<u>CURVE 76</u>	21.55	16.6						
		<u>CURVE 66</u>											
		<u>CURVE 67</u>	21.72	10.3	<u>CURVE 68</u>	21.72	10.3						
		<u>CURVE 68</u>			<u>CURVE 69</u>								
		<u>CURVE 69</u>	21.5	36.7	<u>CURVE 70</u>	21.5	36.7						
		<u>CURVE 70</u>			<u>CURVE 71</u>								
		<u>CURVE 71</u>	21.74	9.7	<u>CURVE 72</u>	21.74	9.7						
		<u>CURVE 72</u>			<u>CURVE 73</u>								
		<u>CURVE 73</u>	21.5	36.7	<u>CURVE 74</u>	21.5	36.7						
		<u>CURVE 74</u>			<u>CURVE 75</u>								
		<u>CURVE 75</u>	21.5	36.7	<u>CURVE 76</u>	21.5	36.7						
		<u>CURVE 76</u>			<u>CURVE 77</u>								
		<u>CURVE 77</u>	21.5	36.7	<u>CURVE 78</u>	21.5	36.7						
		<u>CURVE 78</u>			<u>CURVE 79</u>								
		<u>CURVE 79</u>	21.5	36.7	<u>CURVE 80</u>	21.5	36.7						
		<u>CURVE 80</u>			<u>CURVE 81</u>								
		<u>CURVE 81</u>	21.5	36.7	<u>CURVE 82</u>	21.5	36.7						
		<u>CURVE 82</u>			<u>CURVE 83</u>								
		<u>CURVE 83</u>	21.5	36.7	<u>CURVE 84</u>	21.5	36.7						
		<u>CURVE 84</u>			<u>CURVE 85</u>								
		<u>CURVE 85</u>	21.5	36.7	<u>CURVE 86</u>	21.5	36.7						
		<u>CURVE 86</u>			<u>CURVE 87</u>								
		<u>CURVE 87</u>	21.5	36.7	<u>CURVE 88</u>	21.5	36.7						
		<u>CURVE 88</u>			<u>CURVE 89</u>								
		<u>CURVE 89</u>	21.5	36.7	<u>CURVE 90</u>	21.5	36.7						
		<u>CURVE 90</u>			<u>CURVE 91</u>								
		<u>CURVE 91</u>	21.5	36.7	<u>CURVE 92</u>	21.5	36.7						
		<u>CURVE 92</u>			<u>CURVE 93</u>								
		<u>CURVE 93</u>	21.5	36.7	<u>CURVE 94</u>	21.5	36.7						
		<u>CURVE 94</u>			<u>CURVE 95</u>								
		<u>CURVE 95</u>	21.5	36.7	<u>CURVE 96</u>	21.5	36.7						
		<u>CURVE 96</u>			<u>CURVE 97</u>								
		<u>CURVE 97</u>	21.5	36.7	<u>CURVE 98</u>	21.5	36.7						
		<u>CURVE 98</u>			<u>CURVE 99</u>								
		<u>CURVE 99</u>	21.5	36.7	<u>CURVE 100</u>	21.5	36.7						
		<u>CURVE 100</u>			<u>CURVE 101</u>								

Not shown on plot

DATA TABLE NO. 63 (continued)

T	k	T	k	T	k	T	k	T	k	T	k	T	k
<u>CURVE 89</u>		<u>CURVE 90</u>		<u>CURVE 91</u>		<u>CURVE 92</u>		<u>CURVE 93</u>		<u>CURVE 94</u>		<u>CURVE 95</u>	
1900	1.21	1930.4	1.191	2427.6	0.969	2488.7	0.166	1218.2	1.125	1979	1.205	405.2	1.657
		2050.9	1.142	2452.6	0.999	2699.3	0.177	1248.2	1.118	2060	1.116	414.2	1.587
<u>CURVE 96</u>		<u>CURVE 97</u>		<u>CURVE 98</u>		<u>CURVE 99</u>		<u>CURVE 100</u>		<u>CURVE 101</u>		<u>CURVE 102</u>	
1900	1.17	2191.5	1.203	2566.5	1.237	2677.6	1.350	1338.2	1.088	2217	1.072	494.2	1.600
		2294.3	1.186	2677.6	1.237	2744.3	1.263	1348.2	1.079	2263	1.129	497.2	1.508*
<u>CURVE 103</u>		<u>CURVE 104</u>		<u>CURVE 105</u>		<u>CURVE 106</u>		<u>CURVE 107</u>		<u>CURVE 108</u>		<u>CURVE 109</u>	
1900	1.18	2652.6	1.056	2744.3	1.263	2841.5	1.324	1958.2	0.173	2367	1.055	511.2	1.489*
		2824.8	1.054	2841.5	1.324	2883.2	0.938	2041.5	0.170	2471	1.035	720.2	1.305
<u>CURVE 110</u>		<u>CURVE 111</u>		<u>CURVE 112</u>		<u>CURVE 113</u>		<u>CURVE 114</u>		<u>CURVE 115</u>		<u>CURVE 116</u>	
1900	1.18	3022.1	1.064	3074.8	1.038	1505.9	0.0857	1423.2	1.061	1963	1.113	1000	1.31
		2933.2	1.018	1710.9	0.0415	1591.5	0.0848	1493.2	1.040	2084	1.143	1150	1.25
<u>CURVE 117</u>		<u>CURVE 118</u>		<u>CURVE 119</u>		<u>CURVE 120</u>		<u>CURVE 121</u>		<u>CURVE 122</u>		<u>CURVE 123</u>	
1900	1.10	2149.8	0.1099	1710.9	0.0415	1874.8	0.170	1758.2	0.992	1963	1.113	1249	1.22
		2149.8	0.1099	1874.8	0.170	1877.6	0.186	1783.2	0.990	2084	1.143	1375	1.19
<u>CURVE 124</u>		<u>CURVE 125</u>		<u>CURVE 126</u>		<u>CURVE 127</u>		<u>CURVE 128</u>		<u>CURVE 129</u>		<u>CURVE 130</u>	
317.2	1.046	2285.4	1.385	1702.6	0.137	1966.5	0.200	1873.2	0.975	2084	1.143	1500	1.17
		2139.7	1.246	1869.3	0.145	2030.4	0.235	2018.2	0.956	2384	1.124	1750	1.14
<u>CURVE 131</u>		<u>CURVE 132</u>		<u>CURVE 133</u>		<u>CURVE 134</u>		<u>CURVE 135</u>		<u>CURVE 136</u>		<u>CURVE 137</u>	
323.2	1.046	2235.9	1.220	1384.3	0.148	1966.5	0.200	2078.2	0.952	2113.2	0.943	2000	1.10*
		2353.3	1.211	1413.2	0.157	2102.6	0.187	2193.2	0.937	313.2	1.799	285.1	1.867
<u>CURVE 138</u>		<u>CURVE 139</u>		<u>CURVE 140</u>		<u>CURVE 141</u>		<u>CURVE 142</u>		<u>CURVE 143</u>		<u>CURVE 144</u>	
326.2	0.795	2441.5	1.186	1505.4	0.153	2158.2	0.187	2268.2	0.929	314.2	1.786*	284.1	1.83
		2447.1	1.317	1552.6	0.144	2181.5	0.202	2348.2	0.923	329.2	1.767	293.7	1.831
<u>CURVE 145</u>		<u>CURVE 146</u>		<u>CURVE 147</u>		<u>CURVE 148</u>		<u>CURVE 149</u>		<u>CURVE 150</u>		<u>CURVE 151</u>	
311.2	1.172	2555.4	1.385	1588.7	0.145	2205.4	0.183	2403.2	0.914	337.2	1.678	373.7	1.67
		2583.2	1.056	1588.7	0.145	2233.2	0.198	2473.2	0.908	343.2	1.714*	384.8	1.627
<u>CURVE 152</u>		<u>CURVE 153</u>		<u>CURVE 154</u>		<u>CURVE 155</u>		<u>CURVE 156</u>		<u>CURVE 157</u>		<u>CURVE 158</u>	
311.2	1.172	2655.4	1.385	1702.6	0.137	2285.9	0.188	2473.2	0.908	343.2	1.643	397.0	1.627
		2705.4	1.038	1869.3	0.145	2310.9	0.208	2518.2	0.898	404.2	1.498	457.0	1.520
<u>CURVE 159</u>		<u>CURVE 160</u>		<u>CURVE 161</u>		<u>CURVE 162</u>		<u>CURVE 163</u>		<u>CURVE 164</u>		<u>CURVE 165</u>	
311.2	1.297	2772.1	1.419	1902.6	0.136	2374.8	0.186	1638	1.159	484.2	1.547	499.8	1.496
		2833.7	0.969	2058.2	0.148	2391.5	0.210	1691	1.219	542.2	1.423		
<u>CURVE 166</u>		<u>CURVE 167</u>		<u>CURVE 168</u>		<u>CURVE 169</u>		<u>CURVE 170</u>		<u>CURVE 171</u>		<u>CURVE 172</u>	
311.2	1.185	2863.7	1.411	2191.5	0.142	2472.1	0.216	1691	1.310	592.2	1.353*		
		2919.3	1.185	2334.8	0.124	2477.6	0.208	1749	1.224	600.2	1.352		
<u>CURVE 173</u>		<u>CURVE 174</u>		<u>CURVE 175</u>		<u>CURVE 176</u>		<u>CURVE 177</u>		<u>CURVE 178</u>		<u>CURVE 179</u>	
308.2	1.590	2510.9	0.131	2510.9	0.131	2477.6	0.182	1863	1.224	604.2	1.286		
		2647.1	0.138*	2647.1	0.138*	2557.1	0.177	1946	1.202				
<u>CURVE 180</u>		<u>CURVE 181</u>		<u>CURVE 182</u>		<u>CURVE 183</u>		<u>CURVE 184</u>		<u>CURVE 185</u>		<u>CURVE 186</u>	
308.2	1.590	2569.4	0.133*	2569.4	0.133*	2635.9	0.196	1964	1.154				
		2672.1	0.181	2672.1	0.181	2635.9	0.196	2081	1.199				
<u>CURVE 187</u>		<u>CURVE 188</u>		<u>CURVE 189</u>		<u>CURVE 190</u>		<u>CURVE 191</u>		<u>CURVE 192</u>		<u>CURVE 193</u>	
320.2	1.590	1366.5	0.179	1366.5	0.179	2752.6	0.205						
		1602.6	0.156	1602.6	0.156								
<u>CURVE 194</u>		<u>CURVE 195</u>		<u>CURVE 196</u>		<u>CURVE 197</u>		<u>CURVE 198</u>		<u>CURVE 199</u>		<u>CURVE 200</u>	
319.2	0.962	1835.9	0.166	1835.9	0.166	1173.2	1.151	1499	1.219				
		2047.1	0.165	2047.1	0.165	1198.2	1.140	1549	1.343				
<u>CURVE 201</u>		<u>CURVE 202</u>		<u>CURVE 203</u>		<u>CURVE 204</u>		<u>CURVE 205</u>		<u>CURVE 206</u>		<u>CURVE 207</u>	
		2285.9	0.158	2285.9	0.158			1650	1.220				
								1779	1.161				

Not shown on plot

DATA TABLE NO. 63 (continued)

T	k	T	k
<u>CURVE 122*</u>			
1206	1.10	1896	0.950
1264	1.116	1946	0.946
1264	1.070	2206	0.911
1267	1.089	2350	0.902
1613	1.096	2463	0.866
1615	1.002	2608	0.905
1678	0.997	<u>CURVE 127*</u>	
1681	1.007	87.3	2.003
1847	1.024	95.7	1.952
1847	1.005	99.6	1.927
1997	0.986	199.5	1.733
1997	0.995	201.7	1.729
<u>CURVE 123*</u>			
1451	1.096	204.3	1.727
1452	1.093	278.2	1.660
1453	1.112	281.5	1.659
1454	1.086	334.7	1.603
2005	0.982	337.8	1.606
2005	1.028	373.8	1.556
2005	0.975	377.3	1.556
2005	0.979		
2033	1.002		
2033	0.997		
2033	0.991		

CURVE 124\*

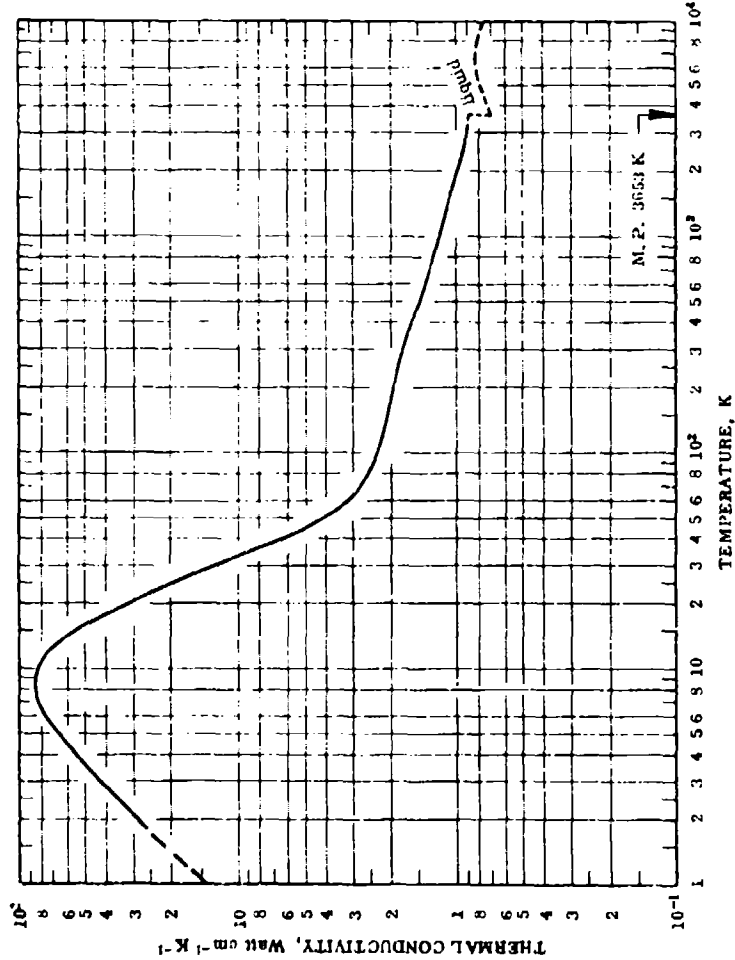
1513	1.183
1536	1.197
1578	1.169
1678	1.072
1748	1.122
1835	1.088
1898	1.092
1930	1.008

CURVE 125\*

1571.5	1.182
1640	1.144
1675	1.094
1719.5	1.055
1745.5	1.036
1835.5	1.048
1905	1.017

\* Not shown on plot

FIGURE AND TABLE NO. 63R RECOMMENDED THERMAL CONDUCTIVITY OF TUNGSTEN



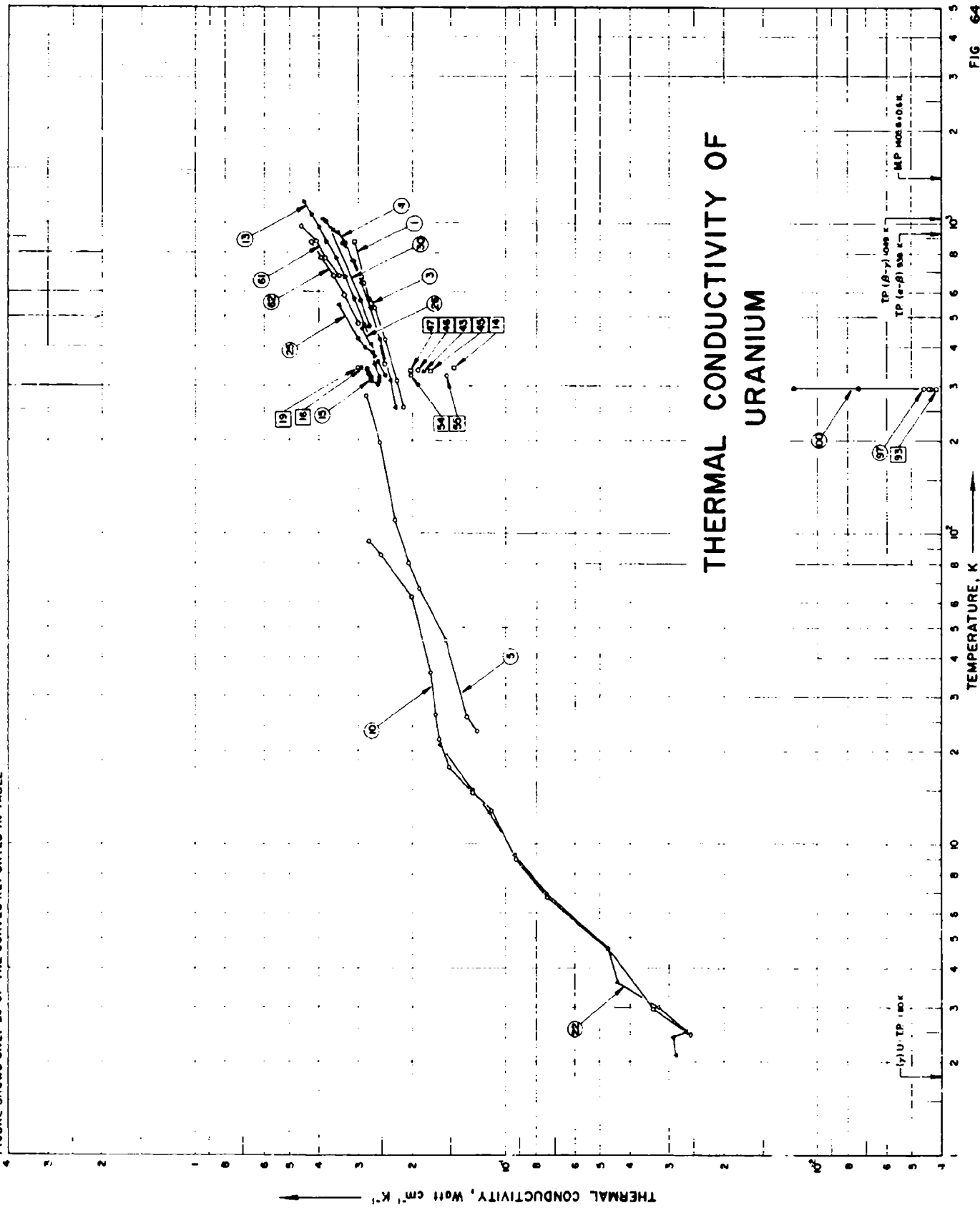
## REMARKS

The recommended values are for well-annealed 99.99% pure tungsten with residual electrical resistivity  $\rho = 0.00170 \mu\Omega \text{ cm}$  (characterized by  $\rho_0$  becomes important at temperatures below about 200 K). The values below  $1.5 T_m$  are calculated to fit the experimental data by using  $n = 2.40$ ,  $\alpha' = 2.06 \times 10^{-5}$ , and  $\beta = 0.0696$ . The recommended values that are supported by experimental thermal conductivity data are thought to be accurate to within 3% of the true values near room temperature, and 3 to 5% at other temperatures.

		RECOMMENDED VALUES*					
$T_1$	$T_2$	$k_1$	$k_2$	$T_1$	$k_1$	$k_2$	$T_2$
0	0	0	0	1000	1.21	69.9	1340
1	(14.4)†	(83.2)		1200	1.15	66.4	1700
2	28.7	1660		1400	1.11	64.1	2060
3	42.6	2460		1600	1.07	61.8	2420
4	55.6	3210		1800	1.03	59.5	2780
5	67.1	3880		2000	1.00	57.8	3140
6	76.2	4400		2200	0.977	56.5	3500
7	82.4	4760		2400	0.957	55.3	3860
8	85.3	4930		2600	0.938	54.2	4220
9	85.1	4920		2800	0.924	53.4	4580
10	82.4	4760		3000	0.913	52.8	4940
12	72.4	4180		3500	(0.898)	(51.9)	5840
14	60.4	3490		3653	(0.895)	(51.7)	6115
16	49.3	2850					
18	40.0	2310					
20	32.6	1880		3653	(0.705)	(40.7)	6115
25	20.4	1180		4000	(0.730)	(42.2)	6740
30	13.1	757		4500	(0.761)	(44.0)	7640
35	8.9	514		5000	(0.785)	(45.4)	8540
40	6.5	376		6000	(0.811)	(46.9)	10340
50	4.17	241		7000	(0.819)	(47.3)	12140
60	3.18	184		8000	(0.816)	(47.1)	13940
70	2.76	159		9000	(0.799)	(46.2)	15740
80	2.56	148					
90	2.44	141		10000	(0.759)	(44.4)	17540
100	2.35	136		11000	(0.732)	(42.3)	19340
150	2.20	121		12000	(0.694)	(40.1)	21140
200	1.97	114		13000	(0.646)	(37.3)	22940
250	1.86	107		14000	(0.594)	(34.3)	24740
273.2	1.82	105		15000	(0.538)	(31.1)	26540
				16000	(0.478)	(27.6)	28340
300	1.78	103		17000	(0.416)	(24.0)	30140
350	1.70	98.2		18000	(0.352)	(20.3)	31940
400	1.62	93.6		19000	(0.286)	(16.5)	33740
500	1.49	86.1		20000	(0.217)	(12.5)	35540
600	1.39	80.3		21000	(0.146)	(8.44)	37340
700	1.33	76.8		22000	(0.0736)	(4.25)	39140
800	1.28	74.0		23000	(-0)	(-0)	40940
900	1.24	71.7					

\*  $T_1$  in K,  $k_1$  in  $\text{Watt cm}^{-1} \text{K}^{-1}$ ,  $T_2$  in F, and  $k_2$  in  $\text{Btu hr}^{-1} \text{ft}^{-1} \text{F}^{-1}$ . † Values in parentheses are extrapolated or estimated.

FIGURE SHOWS ONLY 25 OF THE CURVES REPORTED IN TABLE



## SPECIFICATION TABLE NO. 64 THERMAL CONDUCTIVITY OF URANIUM

(Impurity &lt; 0.20% each; total impurities &lt; 0.50%)

[ For Data Reported in Figure and Table No. 64 ]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition ( weight percent), Specifications and Remarks
1	46		1958	353, 873			Mg-reduced.
2	46		1958	423, 873			Ca-reduced.
3	32	R	1954	255-867			Pure; cylindrical disc specimen; rolled in the $\alpha$ -phase; as received; data taken from smoothed curve.
4	32	R	1954	255-1033			Pure; cylindrical disc specimen; rolled in the $\alpha$ -phase; heated to 750 C for 10 min in the $\beta$ -phase, then water quenched; data taken from smoothed curve.
5	139	L	1952	23-278	$\pm 5.0$		Approx 99.8 pure; specimen in a rod form 0.5 in. in dia and 2.5 in. long; heated at 700 C in a lead bath for 10 min then quenched in water at 50 C; electrical resistivity reported as 4.07, 9.79, 10.4, 9.96, 9.72, 13.3, 13.4, 13.2, 21.6, 21.6, 21.6, and 28.4 $\mu\text{ohm cm}$ at 21.14, 77.61, 77.61, 77.61, 77.61, 109.7, 109.7, 109.7, 194.5, 194.5, and 277.2 K, respectively; $\rho(293\text{K})/\rho(20\text{K}) = 10.6$ .
6	29	C	1952	313, 373		695	Pure; 1.445 in. dia x 2.008 in. long; supplied by Sylvania Electric Products Corp; prepared by hot-pressing uranium powder; density 18.86 g $\text{cm}^{-3}$ ; Armco Iron used as comparative material.
7	29	C	1952	313, 373		BMI	Pure; rolled from 1.625 in. to 0.875 in. in dia at 500 C, heat treated at 725 C for 0.5 hr, water quenched from 725 C, and $\alpha$ -phase annealed at 525 C for 1 hr; density 18.86 g $\text{cm}^{-3}$ ; Armco Iron used as comparative material.
8	29	C	1952	313, 373		688	Pure; 1.449 in. dia x 1.98 in. long; supplied by Sylvania Electric Products Corp; powder-compact prepared by decomposition of $\text{UH}_3$ under hot pressing; grain size 0.25-0.50 mm in dia; density 18.84 g $\text{cm}^{-3}$ ; Armco Iron used as comparative material.
9	29	C	1952	313, 373		690	Pure; 1.445 in. dia x 2.06 in. long; supplied by Sylvania Electric Products Corp; powder-compact specimen prepared by decomposition of $\text{UH}_3$ under hot pressing; fine grained; density 18.905 g $\text{cm}^{-3}$ ; Armco Iron used as comparative material.
10	122	L	1955	2.5-94	$\pm 3.0$		Highly pure; specimen 2.95 cm long and 0.203 cm in dia; supplied by Atomic Energy Research Establishment; electrical resistivity reported as 2.2, 2.4, 2.5, 2.7, 3.0, 4.2, 6.7, 9.4, and 11.1 $\mu\text{ohm cm}$ at 9.4, 13.5, 17.8, 20.6, 26.1, 35.1, 56.3, 79.5, and 90.0 K, respectively.
11	86	C	1945	407-534		U1	0.068 C, 0.004 Si, 0.0035 Fe, 0.002 Ni, 0.0009 N, 0.0002 Cr, 0.0002 Ag, and 0.00014 B; 1 in. dia x 10.75 in. long; extruded; thermal conductivity measured in the direction of extrusion; brass used as comparative material.
12	86	C	1945	340-711		U2	0.0720 C, 0.0150 Fe, 0.0100 Ag, 0.0020 Ni, 0.0020 N, 0.00175 Si, 0.0005 Cu, 0.0003 Cr, and 0.00012 B; 1 in. dia x 10.75 in. long; extruded; thermal conductivity measured in the direction of extrusion; brass used as comparative material.

SPECIFICATION TABLE NO. 61 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
13	33	E	1955	371-1105	±5.0		Pure; 0.86 cm dia x 14 cm long; density 18.6 g cm <sup>-3</sup> , electrical resistivity reported as 39.6, 41.6, 41.8, 43.4, 43.9, 45.4, 46.3, 47.2, 48.8, 50.9, 51.2, 52.4, 54.1, 56.5, 56.1, 57.8, 58.2, 56.9, 57.8, 58.4, 57.9, 59.5, 59.3, 58.9, 57.7, 58.5, 56.0, 50.5, 50.4, 54.3, and 54.3 $\mu\text{ohm cm}$ at 97, 137, 137, 185, 188, 192, 234, 236, 277, 316, 323, 353, 366, 432, 440, 458, 463, 466, 475, 532, 532, 597, 623, 640, 645, 652, 696, 711, 720, 829, and 935 C, respectively; thermal conductivity data show no discontinuous change at the transition points, whereas the electrical resistivity data show sudden changes.
14	276	C	1951	343	3.0		Pure; specimen 0.1875 x 0.1875 x 1.75 in.; density 18.8 g cm <sup>-3</sup> ; Armco iron used as comparative material.
15	245	L	1945	304-319	2.0	C-241-7B	From an "as rolled" rod, heated 2 hrs at 850 C and water quenched.
16	394	C	1955	343.2	±3.0	1 B 2	0.03 C; supplied by Argonne National Laboratory; as-rolled; Armco iron used as comparative material.
17	394	C	1955	343.2	±3.0	2 B 3	0.08 C; supplied by Argonne National Laboratory; as-rolled; Armco iron used as comparative material.
18	394	C	1955	343.2	±3.0	A-6	High purity; supplied by Argonne National Laboratory; quenched from 1000 C; Armco iron used as comparative material.
19	394	C	1955	343.2	±3.0	A-8	High purity; supplied by Argonne National Laboratory; quenched from 900 C; Armco iron used as comparative material.
20	394	C	1955	343.2	±3.0	35	Supplied by Argonne National Laboratory; prepared from hot-pressed UH <sub>3</sub> ; Armco iron used as comparative material.
21	394	C	1955	343.2	±3.0	5 A 3	0.1 Cr; supplied by Argonne National Laboratory; as-rolled; Armco iron used as comparative material.
22	97	L	1952	2.1-21	2.0-3.0	U 1	Supplied by Atomic Energy Research Establishment.
23	414	L	1954	373-923	±2.0	U 1	Bar specimen; cast.
24	414	L	1954	373-923	±2.0	U 2	The above specimen heated to 690 C, maintained for several hrs in the $\beta$ -phase then cooled to room temp at a rate of 4.2 C per min to change from $\beta$ to $\alpha$ phase.
25	415	E	1953	311-548	±5.0		Cylindrical bar specimen; cast.
26	415	E	1953	323-458	±5.0		Cylindrical bar specimen; cast; irradiated to 190 M.W.D./Tonne at 300 C; "cooling time" > 1 yr.
27	395	C	1958	293-1073	< ±5.0		Measured in vacuum; Zircaloy-2 was used as comparative material.
28	416	L	1949	293-473		Canadian extruded No. 1	0.026 Si, 0.0188 Fe, 0.0036 Ni, 0.0030 Mn, 0.001 Cu, 0.0009 Cr, 0.0001 Co, and 0.00005 Ag; specimen approx 2.5 cm in dia and 8.0 cm in length; taken from a bar of metal refined in Canada, extruded in the $\gamma$ -phase at 800 C to 900 C by Bureau of Mines; heated at 250 C for an hr for tinning; measured in vacuum.



SPECIFICATION TABLE NO. 61 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
29	416	L	1949	291-422		Canadian extruded No. 1	The above specimen measured in a hotter furnace with different temp gradient in guard sleeve.
30	416	L	1949	470-1003		Canadian extruded No. 1	The above specimen heated at 800 C for 0.5 hr before measurement.
31	416	L	1949	420, 667		Canadian extruded No. 1	The above specimen heated to 400 C three times before measurement.
32	416	L	1949	470, 572		Canadian extruded No. 1	The above specimen heated to 500 C; measured by a new main heater.
33	416	L	1949	573, 9		Canadian extruded No. 1	The above specimen heated to 400 C for 40 min prior to measurement.
34	416	L	1949	574-872		Canadian extruded No. 1	The above specimen tempered for 6 hrs at 600 C during the measurement.
35	416	L	1949	304-878		Canadian extruded No. 1	The above specimen completely remounted and heated to 600 C to set bonds.
36	416	L	1949	477-877		Canadian extruded No. 1	The above specimen heated to 600 C for 3 hrs and cooled down before measurement.
37	416	L	1949	299-878		Canadian extruded No. 1	The above specimen heated for 1.5 hrs at 700 C.
38	416	L	1949	477-878		Canadian extruded No. 1	The above specimen heated for 2 hrs at 700 C.
39	416	L	1949	475-878		Canadian extruded No. 1	The above specimen heated for 2.5 hrs at 600 C.
40	417, 701	C	1943	323-573			Natural uranium; 1 in. in dia; extruded; measured along the direction of extrusion; brass used as comparative material (assumed thermal conductivity 0.23 cal-sec <sup>-1</sup> cm <sup>-1</sup> C <sup>-1</sup> ).
41	413	L	1955	489-884			Cast uranium; specimen heated from $\alpha$ -phase to $\beta$ -phase and cooled again to $\alpha$ -phase; electrical resistivity reported as 44, 7, 46, 2, 54, 2, 56, 3, and 58, 1 $\mu$ hm cm at 202, 296, 405, 509, and 599 C, respectively.
42	418	L	1942	313, 2		Tuballoy	Sintered; density 14.79 g cm <sup>-3</sup> at about 25 C; the sintered uranium specimen contained possibly some uranium carbide.
43	418	L	1942	333, 2		Tuballoy	Sintered and cold-pressed with 200 tons; density 17.22 g cm <sup>-3</sup> at about 25 C; the sintered uranium specimen contained possibly some uranium carbide.
44	418	L	1942	334, 2		Tuballoy	Pure; sintered; density 16.29 g cm <sup>-3</sup> at about 25 C.
45	418	L	1942	334, 2		Tuballoy	Pure; sintered; density 15.72 g cm <sup>-3</sup> at about 25 C.
46	418	L	1942	336, 2		Tuballoy	Fused; density 18.06 g cm <sup>-3</sup> at about 25 C.
47	418	L	1942	335, 2		Tuballoy	Fused; density 18.52 g cm <sup>-3</sup> at about 25 C.

SPECIFICATION TABLE NO. 61 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
48	420	C	1957	407-932	± 7.0		Pure; measured in a vacuum of $10^{-6}$ mm Hg; Armco iron used as comparative material.
49	421	C	1961	423-1023		Casting No. 747	0.005 each of K, P, Ti and Zn, 0.002 Si, 0.002 each of Ca and Mo, 0.001-0.002 C, 0.001 each of As and Na, 0.0003-0.0005 Fe, 0.0005 each of Al, Fe, Co, Ni, and Sn, 0.0004 Sb, 0.0003 Mg, 0.0001 Cu, 0.0001 each of Ag, Bi, Cr, Li, and Pb, and 0.00001 B; specimen 2 cm in dia and 15 cm long; machined from cast ingot; Armco iron used as comparative material.
50	422		1945	398-573			Pure; $\gamma$ -extruded; measured in argon.
51	396		1954	293-1173			Average values for two specimens measured.
52	504	R	1943	357-514			Hollow cylinder; extruded.
53	605	P	1955	323.2			Specimen in the shape of a sphere; cast by Westinghouse Co.; thermal conductivity value calculated from measured thermal diffusivity using the specific heat value of 0.026 cal g <sup>-1</sup> C <sup>-1</sup> and the density 18.6 g cm <sup>-3</sup> .
54	605	C	1955	323.2	10.0		Specimen 2 in. in dia and 1.31 in. long; cast by Westinghouse Co.; cold rolled iron used as comparative material (reference value $0.91 \text{ cal cm}^{-1} \text{ sec}^{-1} \text{ C}^{-1}$ ).
55	605	L, C	1955	323.2	10.0		Porous specimen 2.5 cm x 2 cm x 2.05 cm; prepared from sintered metal powder by Metal Hydrides Corp; copper used as comparative material (reference value $0.91 \text{ cal cm}^{-1} \text{ sec}^{-1} \text{ C}^{-1}$ ).
56	538	L	1956	293-973	5.0		Pure; measured in a vacuum of about $1 \times 10^{-4}$ mm Hg.
57	606	L	1954	334-873			In $\alpha$ -phase; cast; electrical resistivity reported as 34, 48, 36, 62, 42, 76, 46, 90, 51, 04, 55, 18, 59, 32, 61, 39, and 56, 74 $\mu\text{ohm cm}$ at 0, 100, 200, 300, 400, 500, 600, 650, and 690 C, respectively.
58	607	P	1954	323-1048			Specimen 0.125 in. in dia; Ames uranium; thermal conductivity values calculated from measured data of thermal diffusivity.
59	416	L	1956	285-776		Canadian extruded No. 2	0.026 Si, 0.0188 Fe, 0.0036 Ni, 0.0030 Mn, 0.0010 Cu, 0.0009 Cr, 0.0001 Co, and 0.00005 Ag; prepared from a bar of metal refined in Canada and extruded in the $\gamma$ -phase at a temp between 800 and 900 C by Bureau of Mines; specimen approx 2.5 cm in dia and 8.0 cm in length; ends plated with Ni and Cu, linned to the main heater and the heat sink at approx 250 C for 1 hr, then cooled to room temp; heated to 600 C and held for 45 min for bonding; measured in vacuum.
60	416	L	1956	471.2		Canadian extruded No. 2	The above specimen remounted and heated to 750 C for 15 min to set bonds.
61	416	L	1956	677-979		Canadian extruded No. 2	The above specimen heated to 700 C and cooled to room temp prior to the measurement; measured with decreasing temp.
62	416	L	1956	480-875		Canadian extruded No. 2	The above specimen heated to 300 C and cooled to room temp prior to the measurement.

SPECIFIC HEAT TABLE NO. 64 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)	Specifications and Remarks
63	416	L	1956	293-781		Chalk River No. 1		Specimen approx 2.5 cm in dia and 8.0 cm in length; ends-plated with Ni and Cu, tuned to the main heater and heat sink at approx 250 C for 1 hr, then cooled to room temp; measured in vacuum.
64	416	L	1956	432-687		Chalk River No. 2		Similar to above.
65	416	L	1956	497-738		Chalk River No. 2		The above specimen remounted.
66	416	L	1956	298-683		Chalk River No. 3		Similar to the above specimen.
67	416	L	1956	474-726		Chalk River No. 3		The above specimen remounted.
68	730	L	1945	301-333	2	C-241-7A		Pure; prepared from a rolled rod; heated 2 hrs at 850 C and water quenched.
69	730	L	1945	304-333	2	C-245-1		Pure; prepared from a rolled rod.
70	800	C	1966	422-819	5.0	13		Specimen 2.5 cm in dia and 17.7 cm long; Springfield uranium; $\beta$ -quenched and $\alpha$ -annealed; Armco iron used as comparative material.
71	910	P	1953	337-588				Specimen 0.125 in. in dia and 30 cm long; swaged from a Hanford uranium slug and annealed; thermal conductivity values calculated from the measured data of thermal diffusivity using the density and specific heat data of Katz, J. J. and Rabinowitch, E. ("The Chemistry of Uranium", McGraw-Hill, pp. 144-8, 158).
72	910	P	1953	481-693				Another run of the above specimen.
73	911	C	1965	333.2				0.05-0.12 Al, 0.1 C, 0.02-0.05 Fe, 0.01 total of N, O and Si; Springfields standard adjusted uranium; specimen 2.9 cm in dia and 7.5 cm long; cast; heat treated by traverse water quenching in the beta phase (666-760 C) followed by an anneal for 1 hr at 550 C; Armco iron used as comparative material.
74	911	C	1965	333.2				Similar to the above specimen.
75	911	C	1965	333.2				Similar to the above specimen.
76	911	C	1965	333.2				Similar to the above specimen.
77	911	C	1965	333.2				Similar to the above specimen.
78	911	C	1965	333.2				Similar to the above specimen.
79	911	C	1965	333.2				Similar to the above specimen.
80	911	C	1965	333.2				Similar to the above specimen.
81	911	C	1965	333.2				Similar to the above specimen except annealed at 450 C for 3000 hrs.
82	911	C	1965	333.2				Similar to the above specimen.
83	911	C	1965	333.2				Similar to the above specimen.
84	911	C	1965	333.2				Similar to the above specimen except annealed at 450 C for 10000 hrs.

SPECIFICATION TABLE NO. 64 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
85	911	C	1965	333.2			Similar to the above specimen.
86	911	C	1965	333.2			Similar to the above specimen.
87	911	C	1965	333.2			Similar to the above specimen.
88	911	C	1965	333.2			0.05-0.12 Al, 0.1 C, 0.02-0.05 Fe, 0.01 total of N, O and Si; irradiated Springfield standard adjusted uranium; specimen 2.9 cm in dia and 7.5 cm long; cast; heat treated by traverse water quenching in the beta phase (666-760 C) followed by an anneal for 1 hr at 550 C; irradiated in the Calder Hall reactors at an estimated mean temp of 250 C with doses ranging from 1152 to 1932 MWD/te; Armco iron used as comparative material.
89	911	C	1965	333.2			Similar to the above specimens except irradiated at an estimated mean temp of 320 C with doses ranging from 3000 to 3298 MWD/te.
90	911	C	1965	333.2			Similar to the above specimen except irradiated at an estimated mean temp of 370 C with doses ranging from 659 to 1750 MWD/te.
91	911	C	1965	333.2			Similar to the above specimens except irradiated at an estimated mean temp of 395 to 415 C with doses ranging from 55 to 4680 MWD/te.
92	911	C	1965	333.2			Similar to the above specimens except irradiated at estimated mean temp 420 to 450 C with doses ranging from 985 to 3157 MWD/te.
93	843	-	1966	297.7			0.05 Fe, 0.01 Mg, 0.008 Mo, 0.005 Si, 0.005 P, 0.005 K, 0.005 Ti, 0.005 Zn, 0.002 Ca, 0.001 As, 0.001 Na, 0.0005 Ni, 0.0005 Al, 0.0005 Co, 0.0005 Sn, 0.0004 Mn, 0.0002 Cu, 0.0001 Pb, traces of Ag, Bi, Cr, Li, Sb, Be, and B; spherical powder obtained from National Lead Co., contained in a 0.75 in. dia x 2 in. long stainless steel cylindrical cell; mesh size -16 + 20; grain size 1000 $\mu$ ; thermal conductivity measured by the transient line source method; measured in nitrogen at atmospheric pressure.
94	843	-	1966	297-553			Same impurities, source, and measuring method as above; mesh size -70 + 80; grain size 190 $\mu$ ; measured in nitrogen at atmospheric pressure.
95	843	-	1966	297-553			Same impurities, source, and measuring method as above; mesh size -230 + 325; grain size 50 $\mu$ ; measured in nitrogen at atmospheric pressure.
96	843	-	1966	298.2			Same impurities, source, and measuring method as above; mesh size -16 + 20; average grain size 1000 $\mu$ ; measured in nitrogen under pressures ranging from 2.85 x 10 <sup>-5</sup> to 8.71 x 10 <sup>-5</sup> mm Hg.
97	843	-	1966	298.2			Same impurities, source, and measuring method as above; mesh size -40 + 50; average grain size 350 $\mu$ ; measured in nitrogen under pressures ranging from 5.13 x 10 <sup>-5</sup> to 6.166 x 10 <sup>-5</sup> mm Hg.
98	843	-	1966	298.2			Same impurities, source, and measuring method as above; mesh size -70 + 80; measured in nitrogen at 1 atm.
99	843	-	1966	298.2			Similar to above; measured in nitrogen under pressures ranging from 10 <sup>-2</sup> to 5.495 x 10 <sup>-5</sup> mm Hg.

SPECIFICATION TABLE NO. 64 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
100	843	-	1966	298.2			Similar to above; measured in helium under pressures ranging from $10^{-2}$ to $3.589 \times 10^3$ mm Hg.
101	843	-	1966	298.2			Similar to above; measured in methane under pressures ranging from $10^{-2}$ to $3.715 \times 10^3$ mm Hg.
102	843	-	1966	298.2			Similar to above; measured in argon under pressures ranging from $10^{-2}$ to $5.370 \times 10^3$ mm Hg.
103	843	-	1966	298.2			Similar to above; measured in nitrogen under pressures ranging from $2.92 \times 10^{-4}$ to $5.188 \times 10^3$ mm Hg.
104	843	-	1966	298.2			Same impurities, source, and measuring method as above; mesh size $-170 + 200$ ; average grain size $80 \mu$ ; measured in nitrogen under pressures ranging from $2.34 \times 10^{-5}$ to $4.955 \times 10^3$ mm Hg.
105	843	-	1966	298.2			Same impurities, source, and measuring method as above; mesh size $-230 + 375$ ; average grain size $53 \mu$ ; measured in nitrogen under pressures ranging from $1.05 \times 10^{-2}$ to $2.483 \times 10^3$ mm Hg.
106	843	-	1966	298.2			Same impurities, source, and measuring method as above; mesh size $-230 + 325$ ; measured in nitrogen at 1 atm.
107	843	-	1966	298.2			Similar to above; measured in nitrogen under pressures ranging from $0.01$ to $3890$ mm Hg.
108	912	C	1959	293-973			Specimen $0.5$ in. in dia and $5.625$ in. long; unirradiated, unclad national uranium; measured in a vacuum of $2 \times 10^{-4}$ mm Hg; Armooc Iron used as comparative material; data reported here are ten times lower than the original data, which are believed to be wrong as the results of typographical error.
109	992	-	1966	354-1059			Total impurity content $< 0.03$ ; melted and cast in an alumina-coated graphite crucible; cooled and machined to desired dimensions; grain size $\sim 0.25$ mm; electrical resistivity reported as 33.8, 38.3, 44.3, 48.9, 53.8, 55.9, 57.5, 58.2, 56.0, 56.0, and $54.5 \mu\text{ohm cm}$ at 81, 156, 234, 339, 440, 526, 570, 630, 662, 724, and 786 C respectively; Lorenz function reported as 2.17, 1.93, 2.82, 3.44, 3.51, 3.24, 3.37, and $3.30 \times 10^{-9} \text{ V}^2 \text{ K}^{-2}$ at 96, 251, 467, 668, 670, 768, 773, and 801 C, respectively; thermal conductivity values originally used by the authors when deriving these Lorenz function values had been calculated from their measured data for specific heat and thermal diffusivity, present thermal conductivity values calculated (by TPRC) from reported electrical resistivity data and Lorenz function values.



DATA TABLE NO. 61 (continued)

T	k	T	k	T	k	T	k	T	k	T	k
<u>CURVE 39 (cont.)</u>		<u>CURVE 48</u>		<u>CURVE 51</u>		<u>CURVE 57</u>		<u>CURVE 60</u>		<u>CURVE 64 (cont.)</u>	
877.3	0.485	407.0	0.291	293.2	0.26	334.2	0.277	471.2	0.292	436.2	0.279
877.6	0.482	540.7	0.296	373.2	0.27	420.2	0.291	587.2	0.302	490.4	0.287
877.8	0.505	687.2	0.296	473.2	0.29	479.2	0.292	676.3	0.309	533.4	0.285
<u>CURVE 40</u>		0.31		571.2	0.31	483.2	0.289	778.2	0.325	586.9	0.289
323.2	0.238	735.0	0.315	673.2	0.33	484.2	0.294	677.4	0.346	535.4	0.299
373.2	0.249	798.5	0.352	773.2	0.36	491.2	0.300	776.9	0.380	586.4	0.310
423.2	0.259	798.7	0.364	873.2	0.39	595.2	0.324	878.2	0.409	636.9	0.323
473.2	0.269	801.9	0.341	973.2	0.43	663.2	0.323	979.3	0.459	686.9	0.333
523.2	0.279	836.9	0.359	1073.2	0.48	672.2	0.321	<u>CURVE 62</u>		<u>CURVE 65</u>	
573.2	0.290	865.6	0.328	1173.2	0.52	674.2	0.326	479.5	0.300	496.8	0.290
<u>CURVE 41</u>		<u>CURVE 52</u>		<u>CURVE 58</u>		<u>CURVE 66</u>		<u>CURVE 70</u>		<u>CURVE 75</u>	
489.2	0.286	884.7	0.356	357.2	0.261	323.2	0.237	587.2	0.302	333.2	0.276
534.2	0.298	907.0	0.400	400.2	0.262	343.2	0.222	676.3	0.358	333.2	0.285
584.2	0.325	907.0	0.397	419.2	0.263	398.2	0.232	778.2	0.423	333.2	0.268
634.2	0.344	916.5	0.406	456.2	0.275	428.2	0.237	875.2	0.423	333.2	0.276
684.2	0.353	926.1	0.342	476.2	0.278	488.2	0.252	979.3	0.459	333.2	0.285
<u>CURVE 42</u>		<u>CURVE 53</u>		<u>CURVE 59</u>		<u>CURVE 67</u>		<u>CURVE 72</u>		<u>CURVE 76</u>	
333.2	0.240	932.4	0.397	323.2	0.247	1048.2	0.447	474.2	0.283	333.2	0.266
<u>CURVE 43</u>		<u>CURVE 49</u>		<u>CURVE 54</u>		<u>CURVE 68</u>		<u>CURVE 73</u>		<u>CURVE 77</u>	
423.2	0.285	423.2	0.285	323.2	0.247	663.2	0.320	474.2	0.283	333.2	0.266
473.2	0.293	473.2	0.293	323.2	0.247	768.2	0.360	522.2	0.292	333.2	0.276
523.2	0.303	523.2	0.303	323.2	0.247	853.2	0.382	577.2	0.322	333.2	0.276
573.2	0.315	573.2	0.315	323.2	0.247	923.2	0.415	599.8	0.310	333.2	0.276
623.2	0.323	623.2	0.323	323.2	0.247	943.2	0.420	676.2	0.335	333.2	0.276
673.2	0.333	673.2	0.333	323.2	0.247	978.2	0.440	723.2	0.347	333.2	0.276
723.2	0.346	723.2	0.346	323.2	0.247	1018.2	0.460	780.7	0.360	333.2	0.276
773.2	0.362	773.2	0.362	323.2	0.247	1048.2	0.447	801.4	0.371	333.2	0.276
823.2	0.377	823.2	0.377	323.2	0.247	1048.2	0.447	830.4	0.386	333.2	0.276
873.2	0.391	873.2	0.391	323.2	0.247	1048.2	0.447	859.8	0.401	333.2	0.276
923.2	0.410	923.2	0.410	323.2	0.247	1048.2	0.447	889.7	0.414	333.2	0.276
973.2	0.439	973.2	0.439	323.2	0.247	1048.2	0.447	919.7	0.429	333.2	0.276
1023.2	0.469	1023.2	0.469	323.2	0.247	1048.2	0.447	949.7	0.444	333.2	0.276
<u>CURVE 44</u>		<u>CURVE 50</u>		<u>CURVE 55</u>		<u>CURVE 69</u>		<u>CURVE 74</u>		<u>CURVE 78</u>	
334.2	0.142	334.2	0.142	323.2	0.247	1048.2	0.447	474.2	0.283	333.2	0.276
<u>CURVE 45</u>		<u>CURVE 46</u>		<u>CURVE 56</u>		<u>CURVE 71</u>		<u>CURVE 79</u>		<u>CURVE 82</u>	
334.2	0.175	334.2	0.175	323.2	0.247	1048.2	0.447	474.2	0.283	333.2	0.276
<u>CURVE 46</u>		<u>CURVE 47</u>		<u>CURVE 57</u>		<u>CURVE 72</u>		<u>CURVE 80</u>		<u>CURVE 83</u>	
336.2	0.191	336.2	0.191	323.2	0.247	1048.2	0.447	474.2	0.283	333.2	0.276
<u>CURVE 47</u>		<u>CURVE 48</u>		<u>CURVE 58</u>		<u>CURVE 73</u>		<u>CURVE 81</u>		<u>CURVE 84</u>	
335.2	0.202	335.2	0.202	323.2	0.247	1048.2	0.447	474.2	0.283	333.2	0.276
335.2	0.202	335.2	0.202	323.2	0.247	1048.2	0.447	474.2	0.283	333.2	0.276

Not shown on plot

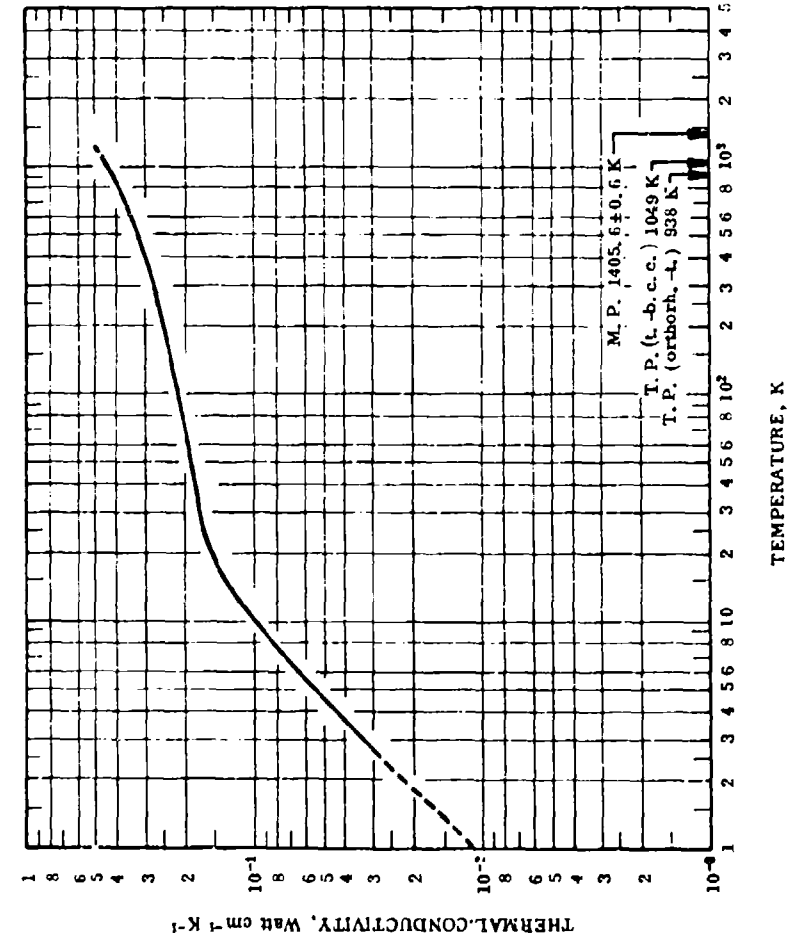
DATA TABLE NO. 64 (continued)

T	k	Irradiation Level (MWD/te)	k	T	k	p (mm Hg)	k	p (mm Hg)	k
<u>CURVE 87*</u>									
T = 333.2 K									
333.2	0.279			298.2	0.00439				
<u>CURVE 92 (cont.)*</u>									
T = 298.2 K									
2801	0.249			0.010	0.000079*				
3157	0.259			0.556	0.00208*				
<u>CURVE 93</u>									
T = 298.2 K									
297.7	0.00420			0.912	0.00235*				
<u>CURVE 94*</u>									
T = 298.2 K									
1152	0.231			22.9	0.00132				
1152	0.232			143	0.00246				
1152	0.235			794	0.00325*				
1163	0			3236	0.00340				
1889	0.240			5495	0.00344*				
1832	0.237			<u>CURVE 100</u>					
T = 298.2 K									
<u>CURVE 95*</u>									
T = 333.2 K									
3000	0.245			0.010	0.000106*				
3060	0.248			0.813	0.000300*				
3298	0.246			1.41	0.000404*				
<u>CURVE 96*</u>									
T = 298.2 K									
659	0.254			28.2	0.00296				
680	0.251			155	0.00745*				
1750	0.253			794	0.0119				
<u>CURVE 97</u>									
T = 298.2 K									
55	0.257			1950	0.0123				
1801	0.259			3989	0.0129*				
3330	0.250			<u>CURVE 101*</u>					
3337	0.251			T = 298.2 K					
3570	0.249			0.010	0.0000866				
4660	0.257			1.43	0.000415				
<u>CURVE 98*</u>									
T = 333.2 K									
965	0.259			23.7	0.00203				
1090	0.254			148	0.00382				
1330	0.252			785	0.00586				
1670	0.246			3715	0.00720				
<u>CURVE 99*</u>									
T = 298.2 K									
0.0000513	0.0000812*			0.010	0.0000316				
0.000197	0.0000874*			1.53	0.000373				
0.00631	0.000178*			30.6	0.00111				
1.38	0.000816*			140	0.00170				
14.8	0.00208*			832	0.00227				
130	0.00427*			3055	0.00252				
3162	0.00460			5370	0.00255				
4519	0.00435			<u>CURVE 102*</u>					
6166	0.00444			T = 298.2 K					
<u>CURVE 103*</u>									
T = 298.2 K									
0.0000292	0.0000348			0.0000234	0.0000238				
0.000507	0.0000398			0.00102	0.0000245				
0.00646	0.0000941			0.00282	0.0000310				
0.0741	0.000244			0.0610	0.0000732				
0.724	0.000411			0.367	0.000153				
22.7	0.00136			10.5	0.000728				
150	0.00244			157	0.00205				
564	0.00315			804	0.00302				
3388	0.00336			2931	0.00310				
5188	0.00345			4955	0.00310				
<u>CURVE 104*</u>									
T = 298.2 K									
0.0000234	0.0000238			0.0000234	0.0000238				
0.00102	0.0000245			0.00102	0.0000245				
0.00282	0.0000310			0.0610	0.0000732				
0.367	0.000153			0.367	0.000153				
10.5	0.000728			10.5	0.000728				
157	0.00205			157	0.00205				
804	0.00302			804	0.00302				
2931	0.00310			2931	0.00310				
4955	0.00310			4955	0.00310				
<u>CURVE 105*</u>									
T = 298.2 K									
0.0105	0.0000409			0.0105	0.0000409				
1.06	0.000248			1.06	0.000248				
10.4	0.000551			10.4	0.000551				
162	0.00136			162	0.00136				
1072	0.00233			1072	0.00233				
2483	0.00264			2483	0.00264				
<u>CURVE 106*</u>									
T = 298.2 K									
298.2	0.00339			298.2	0.00339				

\* Not shown on plot



FIGURE AND TABLE NO. 64R RECOMMENDED THERMAL CONDUCTIVITY OF URANIUM



RECOMMENDED VALUES\*  
(For Polycrystalline)

T <sub>1</sub>	k <sub>1</sub>	k <sub>2</sub>	T <sub>1</sub>	T <sub>2</sub>	k <sub>1</sub>	k <sub>2</sub>	T <sub>2</sub>
0	0	0	500	440.3	0.317	18.3	440.3
1	(0.0114) †	(0.659)	600	620.3	0.340	19.6	620.3
2	(0.0228)	(1.32)	700	800.3	0.364	21.0	800.3
3	0.0338	1.95	800	980.3	0.388	22.4	980.3
4	0.0442	2.55	900	1160	0.413	23.9	1160
5	0.0541	3.13	1000	1340	0.439	25.4	1340
6	0.0638	3.69	1100	1520	0.463	26.8	1520
7	0.0731	4.22	1200	1700	(0.490)	(28.3)	1700
8	0.0818	4.73					
9	0.0898	5.19					
10	0.0960	5.66					
11	0.106	6.12					
12	0.113	6.53					
13	0.120	6.93					
14	0.126	7.28					
15	0.132	7.63					
16	0.138	7.97					
18	0.149	8.61					
20	0.158	9.13					
25	0.167	9.65					
30	0.173	10.0					
35	0.178	10.3					
40	0.182	10.5					
45	0.186	10.7					
50	0.189	10.9					
60	0.196	11.2					
70	0.202	11.7					
80	0.208	12.0					
90	0.212	12.2					
100	0.217	12.5					
150	0.236	13.6					
200	0.251	14.5					
250	0.264	15.3					
273.2	0.270	15.6					
300	0.276	15.9					
350	0.286	16.5					
400	0.296	17.1					

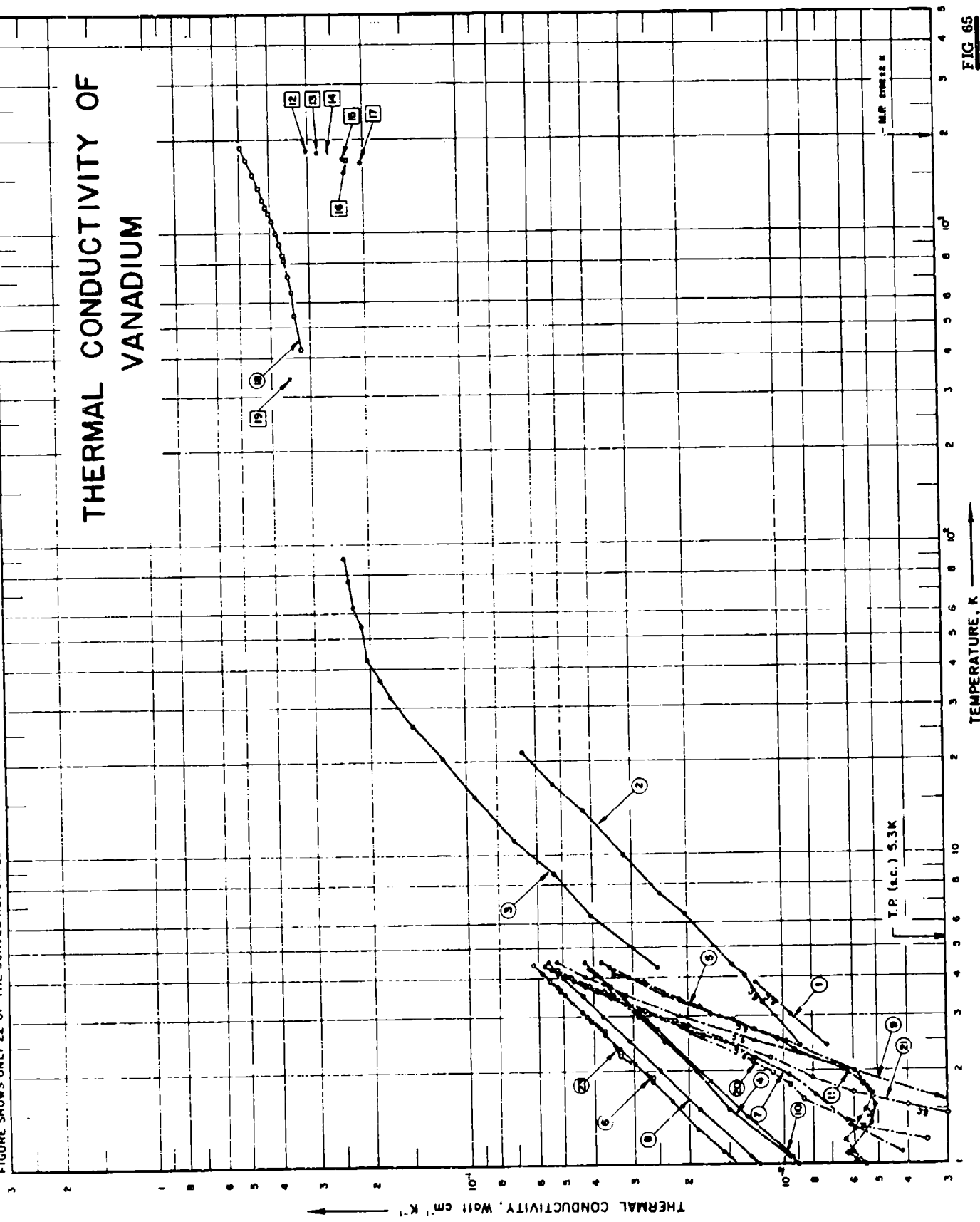
REMARKS

The recommended values are for well-annealed high-purity uranium with residual electrical resistivity  $\rho_0 = 2.2 \mu\Omega \text{ cm}$  (characterization by  $\rho_0$  becomes important at temperatures below about 200 K). The recommended values that are supported by experimental thermal conductivity data are thought to be accurate to within 5% of the true values near room temperature and 5 to 10% at other temperatures.

\* T<sub>1</sub> in K, k<sub>1</sub> in Watt cm<sup>-1</sup> K<sup>-1</sup>, T<sub>2</sub> in F, k<sub>2</sub> in Btu hr<sup>-1</sup> ft<sup>-1</sup> F<sup>-1</sup>. † Values in parentheses are extrapolated.

# THERMAL CONDUCTIVITY OF VANADIUM

FIGURE SHOWS ONLY 22 OF THE CURVES REPORTED IN TABLE



## SPECIFICATION TABLE NO. 65 THERMAL CONDUCTIVITY OF VANADIUM

(Impurity &lt; 0.20% each; total impurities &lt; 0.50%)

[ For Data Reported in Figure and Table No. 65 ]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition ( weight percent), Specifications and Remarks
1	122	L	1955	2.4-3.84	3	V I	Polycrystalline; specimen 0.725 cm long, 0.0995 cm in dia; made from spectrographically standardized metal from Johnson, Matthey and Co.; in superconducting state.
2	122	L	1955	2.4-21		V I	The above specimen measured in a magnetic field; in normal state.
3	151	L	1957	4.3-90		V 4	Approx. 99.9 pure; obtained from Electrometallurgical Co.; specimen 0.55 mm in dia; annealed in vacuo at 1300 C; residual electrical resistivity $\rho_0 = 4.83 \mu\text{ohm cm}$ ; ideal electrical resistivity reported as 0.014, 0.036, 0.14, 0.38, 0.74, 2.3, 4.25, 8.7, 12.95, 16.65, 16.3, and 19.9 $\mu\text{ohm cm}$ at 15, 20, 30, 40, 50, 75, 100, 150, 200, 250, 273, and 295 K, respectively.
4	389	L	1958	0.5-4.5		V I	Single crystal; in normal state.
5	389	L	1958	1.0-4.5		V I	The above specimen in superconducting state.
6	389	L	1958	0.18-4.4		V II	Single crystal; in normal state.
7	389	L	1958	1.1-4.4		V II	The above specimen in superconducting state.
8	389	L	1958	0.2-4.5		V II	Polycrystalline; in normal state.
9	389	L	1958	1.5-4.5		V II	The above specimen in superconducting state.
10	501	L	1961	1.1-4.3		V II	0.05 Fe, 0.01 Si, 0.005 Mo, 0.0005 Mn, and 0.0003 Cu; single crystal; specimen obtained by floating-zone melting of polycrystalline rods; measured in magnetic field of 6200 oersted; in normal state.
11	501	L	1961	0.92-4.3		V II	The above specimen measured with the magnetic field removed; in superconducting state.
12	601	-	1962	1840			Specimen 0.50 in. in dia and 0.442 in. thick; heated in high vacuum ( $10^{-5}$ mm Hg) by high frequency induction to 1000-3000 C; localized heating within 0.003 in. of the surface at current frequency of 500000 cps; heat lost only by radiation, cylindrical surface assumed isothermal, and the temperature gradient along the radius analytically correlated to the thermal conductivity, run No. 1.
13	601	-	1962	1807.5			The above specimen, run No. 3.
14	601	-	1962	1801.5			The above specimen, run No. 4.
15	601	-	1962	1729			The above specimen, run No. 6.
16	601	-	1962	1707.5			The above specimen, run No. 7.
17	601	-	1962	1674.5			The above specimen, run No. 8.
18	614	R	1961	423-1876	< 5		99.74 V, 0.073 O, 0.048 Fe, 0.043 N, and 0.042 C; specimen composed of 5 one-inch dia disks; hot rolled and annealed; density 6.05 g cm <sup>-3</sup> .

SPECIFICATION TABLE NO. 65 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
19	760	C	1955	343	± 3		99.6% pure, calcium reduced vanadium from the Electrometallurgical Co.; Armco iron used as comparative material.
20	677	L	1960	1.2-4.4			Single crystal; ~50 mm long, 4 mm in dia; prepared by 'floating zone' technique; in superconducting state.
21	677	L	1960	1.5-4.3			The above specimen irradiated to a dose of $10^{18}$ fast neutrons $\text{cm}^{-2}$ ; in superconducting state.
22	677	L	1960	1.4-3.2			The above specimen measured before irradiation; in normal state.
23	677	L	1960	1.9-4.4			The above specimen measured after irradiated to a dose of $10^{18}$ fast neutrons $\text{cm}^{-2}$ ; in normal state.

## DATA TABLE NO. 65 THERMAL CONDUCTIVITY OF VANADIUM

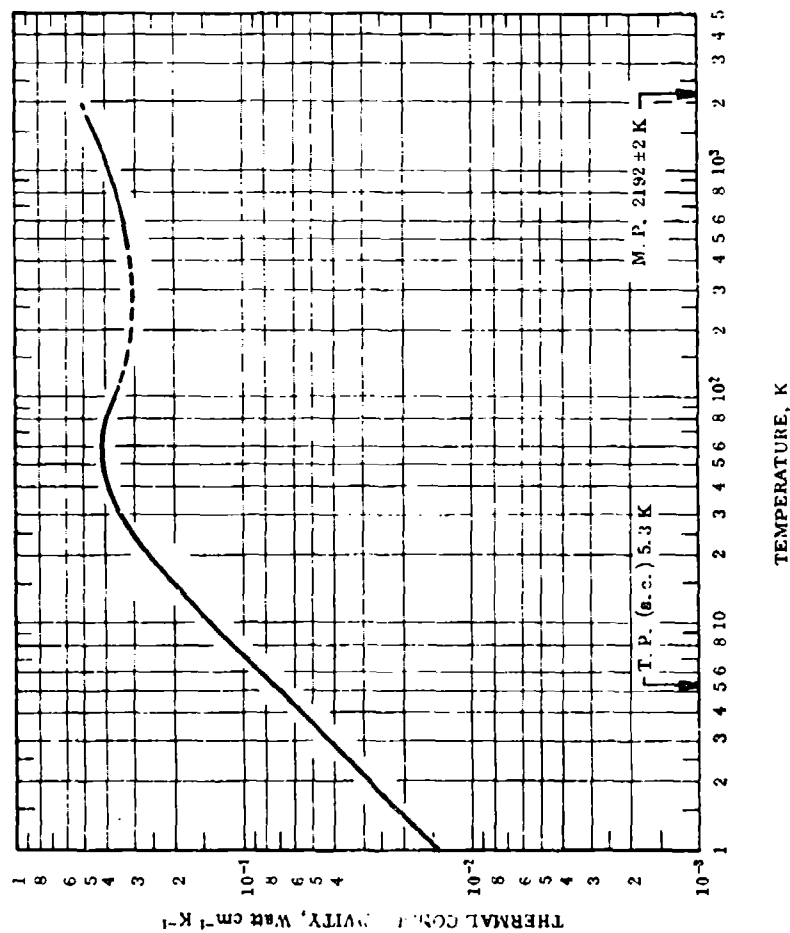
(Impurity &lt; 0.20% each; total impurities &lt; 0.50%)

[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup> K<sup>-1</sup>]

T	k	T	k	T	k	T	k	T	k	T	k	T	k	T	k	T	k
<u>CURVE 1</u>		<u>CURVE 4 (cont.)</u>		<u>CURVE 7</u>		<u>CURVE 10</u>		<u>CURVE 15</u>		<u>CURVE 20 (cont.)</u>		<u>CURVE 23 (cont.)</u>					
2.40	0.0072	3.0	0.029	1.10	0.0042	1.05	0.0095	1729	0.232	2.92	0.0235	3.90	0.055				
3.06	0.0094	3.5	0.0346	1.35	0.0062	1.42	0.0151	<u>CURVE 16</u>		3.28	0.0316*	4.40	0.062*				
3.84	0.0122	4.0	0.038	1.68	0.0078	1.85	0.0173	<u>CURVE 17</u>		3.51	0.0360*						
<u>CURVE 2</u>		4.5	0.043	1.94	0.0096	2.72	0.0255	<u>CURVE 18</u>		3.81	0.0435*						
2.40	0.0088	2.45	0.0245	2.20	0.0123	2.90	0.0272	1674.5	0.204	4.10	0.0495*						
3.31	0.0116	2.60	0.0180	2.45	0.0153	3.00	0.0285	<u>CURVE 19</u>		4.20	0.0535*						
4.05	0.0132	2.76	0.0208	3.14	0.0294	3.42	0.0323	343	0.351	4.35	0.0575*						
4.38	0.0145	2.95	0.0245	3.75	0.0356	3.75	0.0356	<u>CURVE 20</u>		1.45	0.003						
6.40	0.0204	3.14	0.0285	4.10	0.0394	4.10	0.0394	<u>CURVE 21</u>		1.55	0.004						
7.44	0.0245	3.28	0.0318	4.30	0.0414	4.30	0.0414	423.2	0.324	1.70	0.006						
9.92	0.0318	3.45	0.0358	<u>CURVE 11</u>		423.2	0.324	643.2	0.346	1.90	0.008						
13.84	0.0428	3.60	0.0390	0.92	0.0051*	724.8	0.357	818.2	0.366	2.20	0.01						
16.78	0.0531	3.74	0.0426	1.08	0.0063	847.1	0.369	918.7	0.377	2.48	0.015*						
21.32	0.0663	3.90	0.0462	1.36	0.0053	993.7	0.385	1088.7	0.396	2.80	0.020						
<u>CURVE 3</u>		2.0	0.0060	4.04	0.0495	1.55	0.0051	1162.6	0.404	2.90	0.022						
4.30	0.0250	4.15	0.0520	4.27	0.0550	1.83	0.0056	1211.5	0.414	3.13	0.0275						
6.35	0.0405	4.30	0.0575	4.38	0.0575	2.03	0.0065	1392.1	0.434	3.38	0.032*						
8.65	0.0530	4.40	0.061	<u>CURVE 8</u>		2.33	0.0092	1539.8	0.452	3.60	0.037						
11.15	0.0710	4.5	0.038	0.20	0.0020*	2.54	0.0105	1716.5	0.475	3.78	0.041						
15.40	0.0943	<u>CURVE 5</u>		0.50	0.0060*	2.73	0.0124	1875.4	0.495	4.0	0.047						
20.40	0.118	0.18	0.0022*	1.0	0.0120	2.94	0.0145	<u>CURVE 22*</u>		4.25	0.052						
26.00	0.147	0.38	0.0053*	1.5	0.0185	3.05	0.0170	1.35	0.019								
32.30	0.173	0.60	0.0085*	2.0	0.0248	3.22	0.0185	1.70	0.024								
36.55	0.186	0.86	0.0120*	2.5	0.0310	3.40	0.0214	1.85	0.027								
42.70	0.205	1.10	0.0156	3.0	0.0375	3.88	0.0276	2.25	0.032								
54.60	0.214	1.30	0.0188	3.5	0.0435	4.12	0.0345	2.65	0.039								
63.10	0.227	1.60	0.0226	4.0	0.050	4.32	0.0353	3.15	0.045								
76.55	0.234	2.12	0.0298	4.5	0.056	<u>CURVE 12</u>		<u>CURVE 23</u>									
90.20	0.241	2.36	0.0333	<u>CURVE 9</u>		1840	0.303	1.90	0.026								
<u>CURVE 4</u>		2.61	0.0370	2.90	0.0410	<u>CURVE 13</u>		1.90	0.033								
0.5	0.004*	3.10	0.0435	3.10	0.0435	1807.5	0.279	2.25	0.037								
1.0	0.009	3.30	0.0468	2.5	0.0123	<u>CURVE 14</u>		2.70	0.042								
1.5	0.014	3.53	0.0497	3.0	0.0215	1801.5	0.260	3.00	0.046*								
2.0	0.019	3.74	0.0523	3.5	0.0316			3.30	0.051								
2.5	0.024	3.97	0.0558	4.0	0.0425			3.65									
		4.16	0.0589	4.5	0.0526												
		4.41	0.0622														

\* Not shown on plot

FIGURE AND TABLE NO. 65R RECOMMENDED THERMAL CONDUCTIVITY OF VANADIUM



REMARKS

The recommended values are for well-annealed high-purity vanadium with residual electrical resistivity  $\rho_0 = 1.72 \mu\Omega \text{ cm}$  (characterization by  $\rho_0$  becomes important at temperatures below about 150 K). The recommended values that are supported by experimental thermal conductivity data are thought to be accurate to within 5% of the true values near room temperature and 5 to 15% at other temperatures.

TEMPERATURE, K

$T_1$	$k_1$	$k_2$	$T_2$	$T_1$	$k_1$	$k_2$	$T_2$
0	0	0	-459.7	500	0.331	19.1	440.3
1	0.0142	0.820	-457.9	600	0.342	19.8	620.3
2	0.0282	1.63	-456.1	700	0.352	20.3	800.3
3	0.0422	2.44	-454.3	800	0.363	21.0	980.3
4	0.0561	3.24	-452.5	900	0.374	21.6	1160
5	0.0697	4.03	-450.7	1000	0.386	22.3	1340
6	0.0835	4.82	-448.9	1100	0.396	22.9	1520
7	0.0971	5.61	-447.1	1200	0.412	23.8	1700
8	0.111	6.41	-445.3	1300	0.424	24.5	1880
9	0.125	7.22	-443.5	1400	0.436	25.2	2060
10	0.138	7.97	-441.7	1500	0.447	25.8	2240
11	0.151	8.72	-439.9	1600	0.460	26.6	2420
12	0.165	9.53	-438.1	1700	0.473	27.3	2600
13	0.177	10.2	-436.3	1800	0.484	28.0	2780
14	0.190	11.0	-434.5	1900	0.497	(28.7)	2960
15	0.202	11.7	-432.7	2000	(0.509)	(29.4)	3140
16	0.214	12.4	-430.9				
18	0.237	13.7	-427.3				
20	0.258	14.9	-423.7				
25	0.305	17.6	-414.7				
30	0.342	19.8	-405.7				
35	0.369	21.3	-396.7				
40	0.389	22.5	-387.7				
45	0.401	23.2	-378.7				
50	0.405	23.4	-369.7				
60	0.406	23.5	-351.7				
70	0.402	23.2	-333.7				
80	0.390	22.5	-315.7				
90	0.373	21.6	-297.7				
100	(0.357) <sup>†</sup>	(20.6)	-279.7				
150	(0.324)	(18.7)	-189.7				
200	(0.315)	(18.2)	-99.7				
250	(0.312)	(18.0)	-				
273.2	(0.313)	(18.1)	32.0				
300	(0.315)	(18.2)	80.3				
350	(0.318)	(18.4)	170.3				
400	(0.321)	(18.5)	260.3				

<sup>†</sup>  $T_1$  in K,  $k_1$  in  $\text{Watt cm}^{-1} \text{K}^{-1}$ ,  $T_2$  in F, and  $k_2$  in  $\text{Btu hr}^{-1} \text{ft}^{-1} \text{F}^{-1}$ .

<sup>‡</sup> Values in parentheses are extrapolated or interpolated.

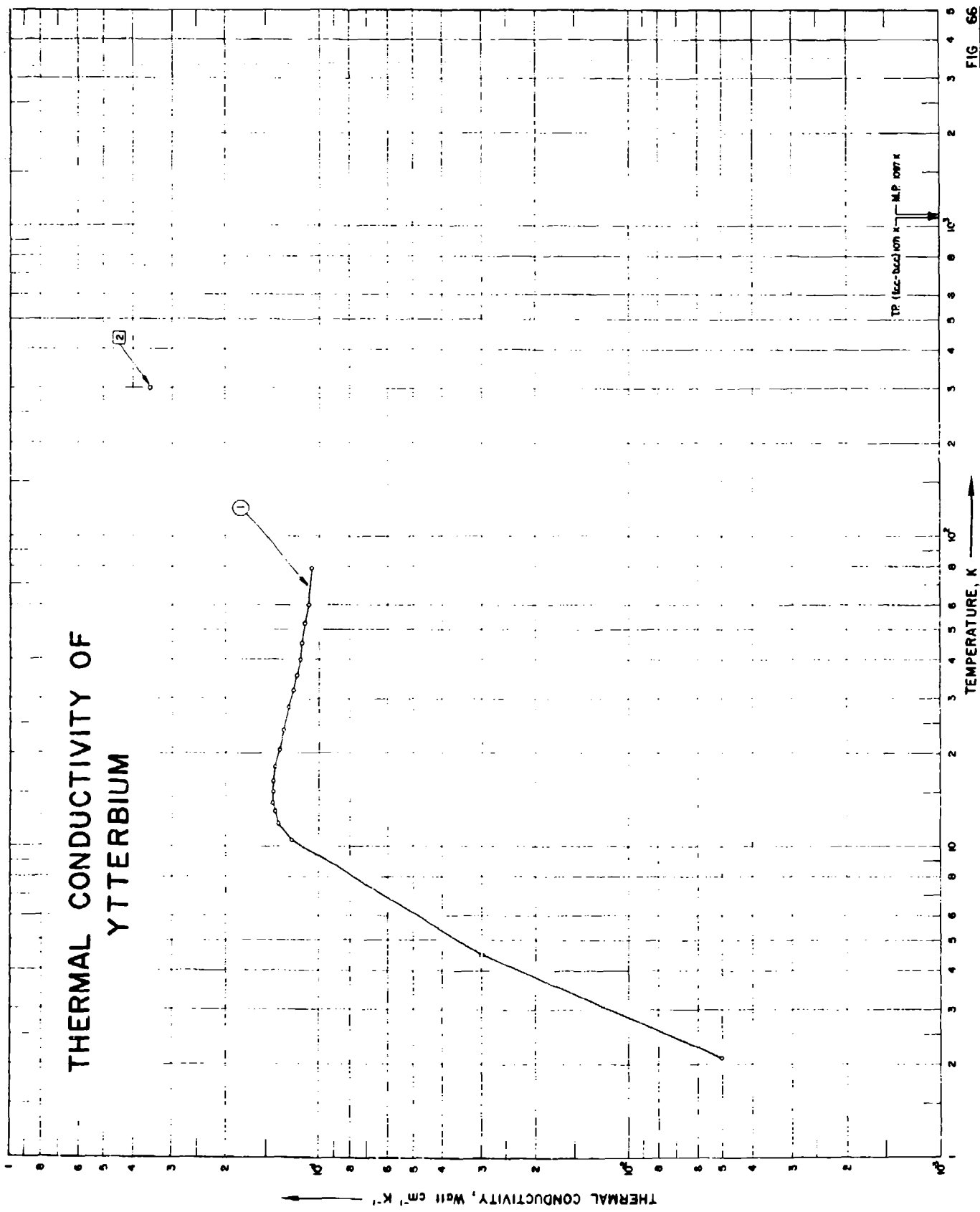


FIG. 66

## SPECIFICATION TABLE NO. 66 THERMAL CONDUCTIVITY OF YTTERBIUM

(Impurity 0.20% each; total impurities 0.50%)

[ For Data Reported in Figure and Table No. 66 ]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)	Specifications and Remarks
1	984 820	L	1965	2.1-79				99.99 pure polycrystalline; strip specimen 0.25 mm thick; annealed in helium vapor at 450 C for 2.5 hrs; electrical resistivity reported as 5.56 and 27 $\mu$ ohm cm at 4.2 and 293 K, respectively; Lorenz function in the residual resistance region found to be $3.17 \times 10^{-8} \text{ K}^{-2}$ ; data taken from smoothed curve.
2	256	-	1966	300				Predicted value-calculated from electrical resistivity value averaged from data of Spedding, F. H., et al. (Trans. AIME, 213, 379, 1958) and Curry, M. A., et al. (Phys. Rev., 117, 953, 1960), using the Lorenz number $3.36 \times 10^{-8} \text{ K}^{-2}$ based on the smoothed curve of Lorenz number as atomic number given by the authors.

## DATA TABLE NO. 66 THERMAL CONDUCTIVITY OF YTTERBIUM

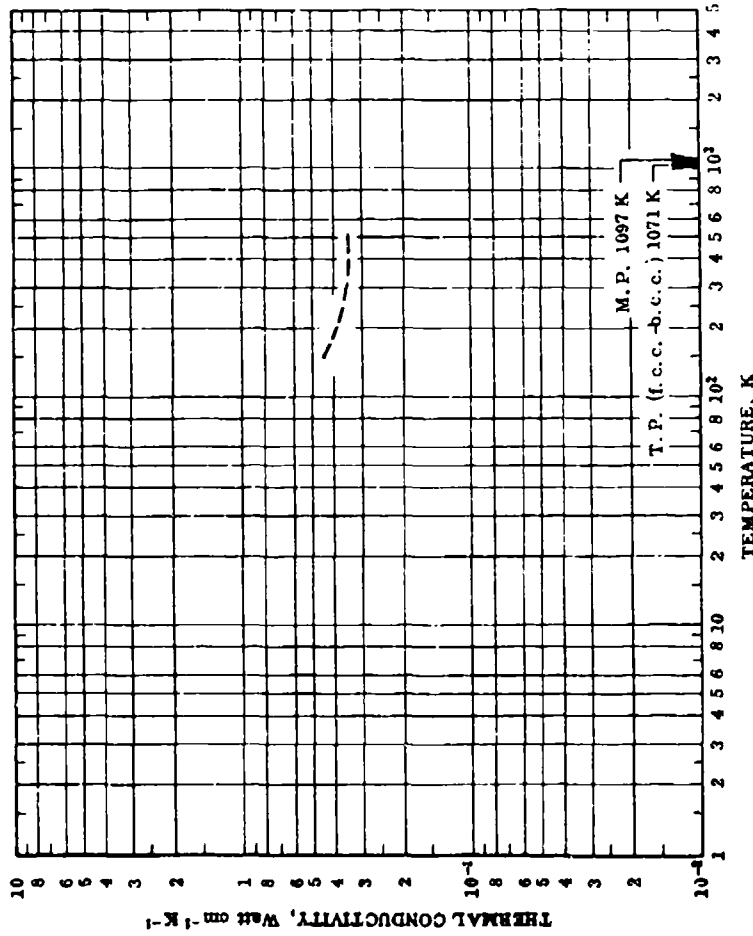
(Impurity 0.20% each; total impurities 0.50%)

[ Temperature, T, K. Thermal Conductivity, k, Watt cm<sup>-1</sup>K<sup>-1</sup> ]

T	k	
	CURVE 1	CURVE 2
2.10	0.005	
4.50	0.03	40.0 0.114
10.5	0.123	45.3 0.113
11.9	0.136	52.5 0.110
13.1	0.140	60.0 0.107
13.8	0.143	78.8 0.105
15.0	0.143	
16.3	0.142	
18.1	0.140	
20.5	0.135	
23.8	0.130	
28.1	0.125	
31.9	0.120	
35.6	0.117	



FIGURE AND TABLE NO. 66R RECOMMENDED THERMAL CONDUCTIVITY OF YTTERBIUM



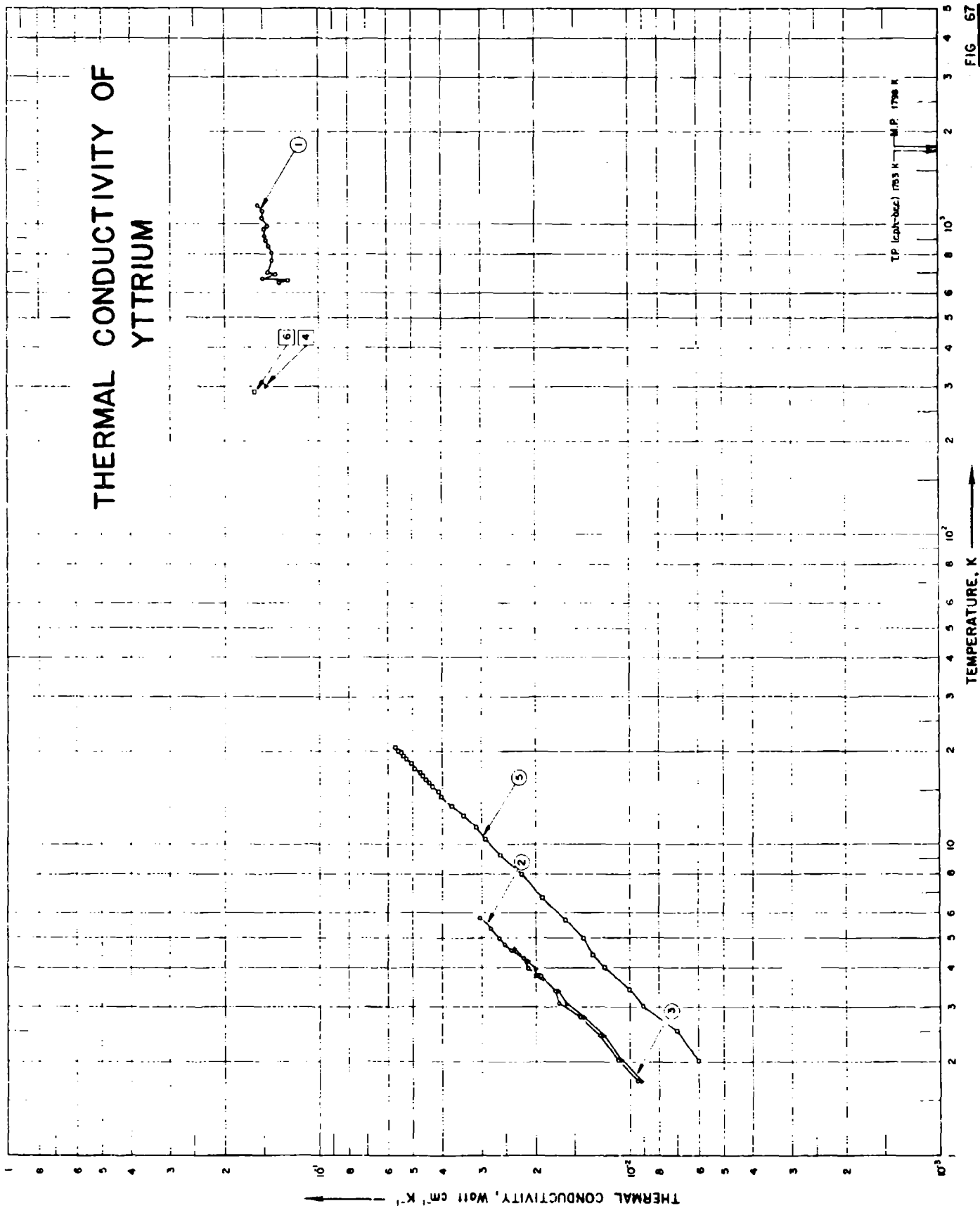
Recommended Values*			
$T_1$	$k_1$	$k_2$	$T_2$
0	0	0	-459.7
150	(0.423)†	(24.4)	-169.7
200	(0.384)	(22.2)	-99.7
250	(0.361)	(20.9)	-9.7
273.2	(0.354)	(20.5)	32.0
300	(0.349)	(20.2)	80.3
350	(0.345)	(19.9)	170.3
400	(0.341)	(19.7)	260.3
500	(0.337)	(19.5)	440.3

REMARKS

The recommended values are for high-purity ytterbium. The recommended values are obtained by estimation and their accuracy is uncertain.

\*  $T_1$  in K,  $k_1$  in  $\text{Watt cm}^{-1} \text{K}^{-1}$ ,  $T_2$  in F, and  $k_2$  in  $\text{Btu hr}^{-1} \text{ft}^{-1} \text{F}^{-1}$ . † Values in parentheses are estimated.

# THERMAL CONDUCTIVITY OF YTTRIUM



## SPECIFICATION TABLE NO. 67 THERMAL CONDUCTIVITY OF YTTRIUM

(Impurity 0.20% each; total impurities 0.50%)

[ For Data Reported in Figure and Table No. 67 ]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)	Specifications and Remarks
1	761		1959	653-1153	3-5			No details reported.
2	762	L	1965	1.8-5.9				99.99 nominal purity; polycrystalline; machined from zone-refined ingot; annealed at 1150 C for 75 hrs; residual electrical resistivity 3.10 $\mu\text{ohm cm}$ ; electrical resistivity ratio $\rho(293 \text{ K})/\rho(4.2 \text{ K}) = 13.0$ ; Lorenz number $L_0 = 2.65 \times 10^{-8} \text{ V}^2 \text{ K}^{-2}$ .
3	762	L	1965	1.8-4.7				The above specimen measured in a magnetic field of 6600 gauss.
4	811		1954	301.2	10			No details given.
5	817 897		1965	2.0-21				Approx 99.9 pure; flat specimen 0.25 mm thick; electrical resistivity 80 $\mu\text{ohm cm}$ at 293 K; electrical resistivity ratio $\rho(293 \text{ K})/\rho(4.2 \text{ K}) = 7.3$ ; Lorenz number $3.00 \times 10^{-8} \text{ V}^2 \text{ K}^{-2}$ at 4.2 K.
6	256	C	1966	291	$\pm 4$			$\sim 0.1 \text{ Ta}$ , $< 0.1$ other rare earth metal, and $\sim 0.03$ other base metals; polycrystalline; specimen 0.63 cm in dia 0.63 cm long; electrical resistivity 53 $\mu\text{ohm cm}$ at 291 K; data derived by the authors from measurements by 2 different thermal comparators.

DATA TABLE NO. 67 THERMAL CONDUCTIVITY OF YTTRIUM

(Impurity &lt; 0.20%, total impurities &lt; 0.50%)

[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup>K<sup>-1</sup>]

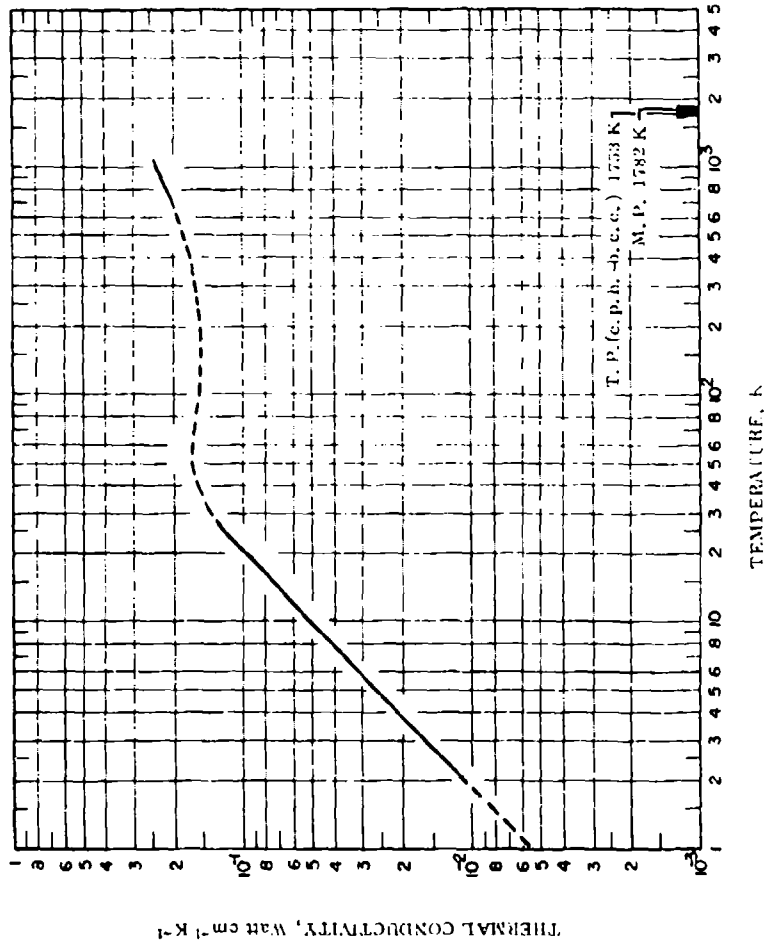
T	k	T	k	T	k
<u>CURVE 1</u>					
652.6	0.134	1.75	0.0092	19.7	0.054
652.6	0.131*	2.05	0.0107	20.0	0.055
655.4	0.134*	2.45	0.0122	20.5	0.056
663.7	0.125	2.40	0.0140	<u>CURVE 5</u>	
674.8	0.151	3.10	0.0160	291.0	0.159
697.1	0.137	3.40	0.0170	<u>CURVE 6</u>	
705.4	0.145	3.80	0.0200		
774.8	0.141	4.00	0.0200		
785.9	0.142*	4.25	0.0212		
819.3	0.141	4.65	0.0235		
844.3	0.142*	<u>CURVE 4</u>			
860.9	0.145	301.2	0.146		
891.5	0.147				
891.5	0.144*	<u>CURVE 5</u>			
928.7	0.148	2.0	0.006		
977.6	0.149	2.5	0.007		
980.4	0.146*	3.0	0.009		
997.1	0.146	3.4	0.010		
1058.2	0.152	4.0	0.012		
1063.7	0.152*	4.4	0.013		
1072.1	0.150*	5.0	0.014		
1083.2	0.152*	5.7	0.016		
1088.7	0.150	6.7	0.019		
1102.6	0.150	8.0	0.022		
1135.9	0.151*	9.2	0.026		
1152.6	0.156	10.4	0.029		
<u>CURVE 2</u>					
1.75	0.0095	11.4	0.031		
2.05	0.0110	12.4	0.034		
2.45	0.0125	13.3	0.037		
2.80	0.0145	14.2	0.040		
3.10	0.0170	14.8	0.041		
3.40	0.0175	15.4	0.043		
3.80	0.0195	15.7	0.044		
4.00	0.0213	16.1	0.045		
4.35	0.0220	16.6	0.046		
4.60	0.0243	17.0	0.047		
4.80	0.0255	17.5	0.049		
5.00	0.0265	18.2	0.050		
5.40	0.0283	18.8	0.052		
5.85	0.0305	19.2	0.053		

\* Not shown on plot

FIGURE AND TABLE NO 67R RECOMMENDED THERMAL CONDUCTIVITY OF YTTRIUM

RECOMMENDED VALUES\*  
(For Polycrystalline)

T <sub>1</sub>	k <sub>1</sub>	k <sub>2</sub>	T <sub>2</sub>	T <sub>1</sub>	k <sub>1</sub>	k <sub>2</sub>	T <sub>2</sub>
0	0	0	-459.7	500	(0.180)	(10.4)	440.3
1	(0.06552) <sup>†</sup>	(0.319)	-457.3	600	(0.192)	(11.1)	620.3
2	0.0198	0.624	-456.1	700	0.203	11.7	800.3
3	0.0129	0.919	-454.3	800	0.215	12.4	980.3
4	0.0210	1.21	-452.5	900	0.226	13.1	1160
5	0.0261	1.51	-450.7	1000	0.235	13.6	1340
6	0.0311	1.80	-448.9	1100	0.238	13.8	1520
7	0.0360	2.08	-447.1				
8	0.0410	2.37	-445.3				
9	0.0460	2.66	-443.5				
10	0.0510	2.95	-441.7				
11	0.0553	3.21	-439.9				
12	0.0601	3.48	-438.1				
13	0.0652	3.77	-436.3				
14	0.0698	4.03	-434.5				
15	0.0746	4.31	-432.7				
16	0.0792	4.58	-430.9				
18	0.0887	5.13	-427.3				
20	0.0982	5.67	-423.7				
25	(0.121)	(6.99)	-414.7				
30	(0.138)	(7.97)	-405.7				
35	(0.149)	(8.61)	-396.7				
40	(0.157)	(9.07)	-387.7				
45	(0.163)	(9.42)	-378.7				
50	(0.166)	(9.59)	-369.7				
60	(0.167)	(9.65)	-351.7				
70	(0.164)	(9.48)	-333.7				
80	(0.160)	(9.24)	-315.7				
90	(0.158)	(9.13)	-297.7				
100	(0.157)	(9.07)	-279.7				
150	(0.156)	(9.01)	-189.7				
200	(0.158)	(9.13)	-99.7				
250	(0.160)	(9.24)	-9.7				
273.2	(0.160)	(9.24)	32.0				
300	0.162	9.36	80.3				
350	(0.165)	(9.53)	170.3				
400	(0.169)	(9.76)	260.3				

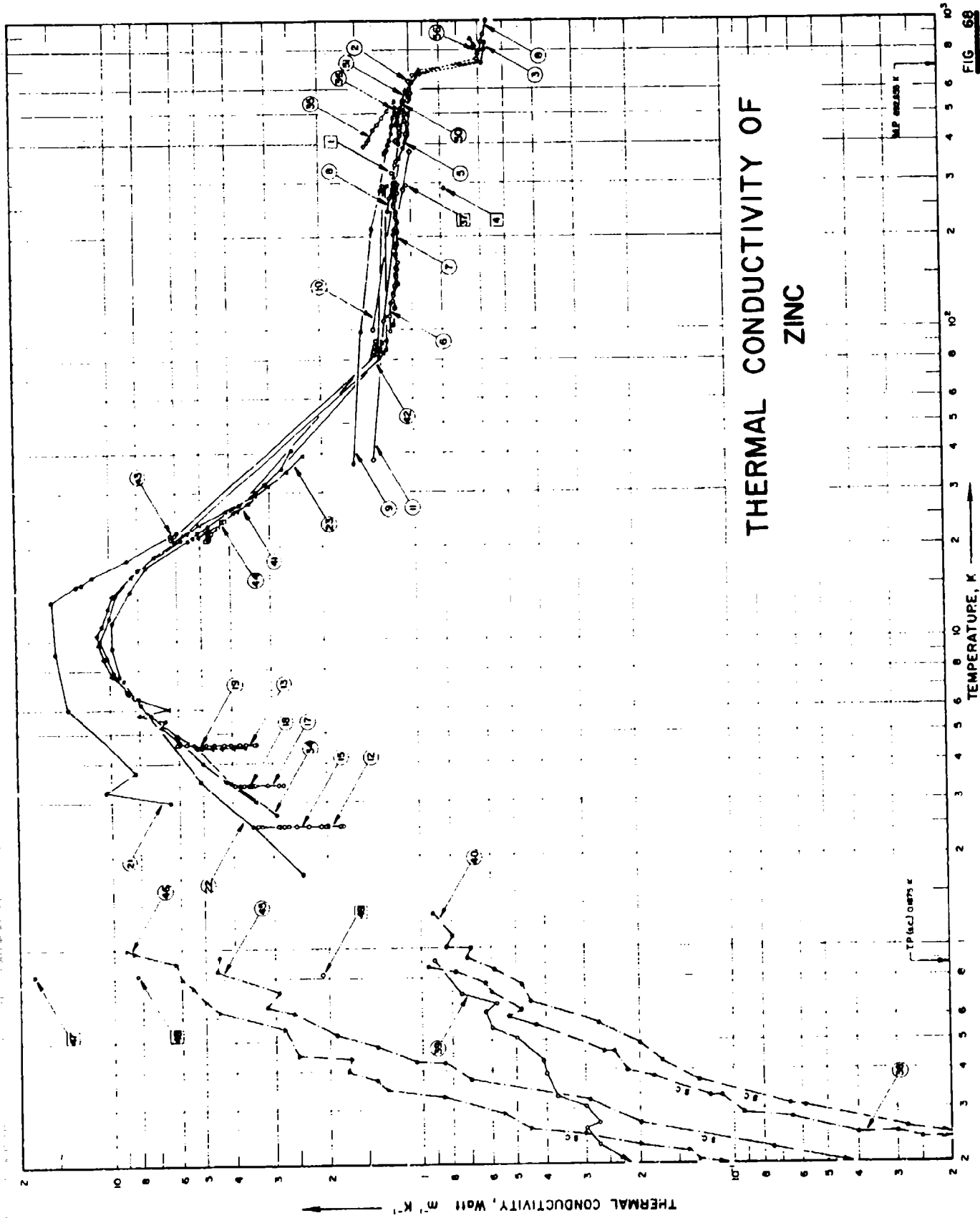


REMARKS

The recommended values are for well-annealed 99.99% pure yttrium with residual electrical resistivity  $\rho_1 = 5.10 \mu\Omega \text{ cm}$  (characterization by  $\rho_1$  becomes important below room temperature). The recommended values that are supported by experimental thermal conductivity data are thought to be accurate to within 5% of the true values near room temperature and 5 to 15% at other temperatures.

\* T<sub>1</sub> in K, k<sub>1</sub> in Watt cm<sup>-1</sup> K<sup>-1</sup>, T<sub>2</sub> in F, and k<sub>2</sub> in Btu ft<sup>-1</sup> ft<sup>-1</sup> F<sup>-1</sup>.

<sup>†</sup> Values in parentheses are extrapolated or interpolated.



THERMAL CONDUCTIVITY OF  
ZINC

## SPECIFICATION TABLE NO. 64 THERMAL CONDUCTIVITY OF ZINC

(impurity &lt; 0.20% each; total impurities &lt; 0.50%)

[ For Data Reported in Figure and Table No. 64 ]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition ( weight percent), Specifications and Remarks
1	230	L	1925	323			99.97% pure; Baker's analyzed metal; cylindrical specimen 10 cm long, 1.9 cm in dia; electrical conductivity at 22 C being $17.0 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$ .
2	6	L	1931	409-640			Pure redistilled zinc; cylindrical specimen of 0.585 cm dia; fracture crystalline with crystals radiating from axis of rod; density $7.10 \text{ g cm}^{-3}$ at 21 C; the same specimen used by Lees in 1908 (curve 6).
3	95	L	1919	402-851			Measured in both solid and liquid states.
4	511	L	1918	288.4			Specimen radius 0.685 cm; furnished by the manufacturer Erba.
5	427	L	1925	361-562			99.8% pure; obtained from London Zinc Mills; cast from billets, rolled at 200 C, sawed into strips and drawn cold; density $7.13 \text{ g cm}^{-3}$ at 21 C; electrical resistivity reported as 6.08, 8.09, 10.48, and $14.50 \mu\text{ohm cm}$ at 35, 105, 200, and 350.2 C, respectively.
6	88	L	1908	99-297			Pure; turned from a cast stick of "pure redistilled zinc"; fracture crystalline with crystals radiating from the axis of rod; cylinder about 7 cm long and 0.585 cm in dia; density $7.15 \text{ g cm}^{-3}$ at 21 C; electrical resistivity reported as 1.699, 1.96, 3.26, 3.69, 4.32, 5.36, 6.30, 6.99, 7.14, and $8.01 \mu\text{ohm cm}$ at -180.3, -168.4, -116.3, -99.7, -70.1, -24.7, 16.7, 47.8, 54.3, and 90.3 C, respectively; first experiment.
7	88	L	1908	104-300			Second experiment of the above specimen.
8	13, 14	F	1939	243-1003			Specimens 4-5 cm in dia, 20-25 cm long used to find data in the solid state; for the liquid state molten zinc contained in a graphite cylinder to form a specimen 25 cm long and 4 cm in dia.
9	16	F	1929	37-382			99.993 Zn, 0.005 Fe, and 0.0018 Cd; single crystal; obtained from the Bureau of Standards; melted in an evacuated glass tube, lowered from the furnace at the rate of $1 \text{ cm h}^{-1}$ ; heat flow parallel to the basal plane.
10	16	F	1929	98-434		No. 2	Same compositions and supplier as the above specimen; polycrystalline; cast in vacuo in a graphite mold.
11	16	F	1929	38-380		No. 1	Similar to above but cast in open air.
12	342	L	1953	2.5		Zn 2	99.997% pure; single crystal; 1-2 mm dia x 5 cm long; obtained from Imperial Smelting Corp; specimen axis at $80^\circ$ with the hexagonal axis; measured in transverse magnetic fields with strength H ranging from 0.17 to 3.73 kilooersteds.
13	342	L	1953	4.6		Zn 2	The above specimen measured in transverse magnetic fields with strength H ranging from 0.17 to 3.73 kilooersteds.
14	342	L	1953	4.6		Zn 2	The above specimen measured in longitudinal magnetic fields with strength H ranging from 0.17 to 3.73 kilooersteds.
15	342	L	1953	2.5		Zn 4	Similar to the above specimen but rod axis at $13^\circ$ with the hexagonal axis; measured in transverse magnetic fields with strength ranging from 0.36 to 3.59 kilooersteds.
16	342	L	1953	2.5		Zn 4	The above specimen measured in longitudinal magnetic fields with strength ranging from 0.36 to 3.50 kilooersteds.

SPECIFICATION TABLE NO. 68 (continued)

Curie No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
17	342	L	1953	3.4		Zn 4	The above specimen measured in transverse magnetic fields with strength ranging from 0.36 to 3.59 kilooersteds.
18	342	L	1953	3.4		Zn 4	The above specimen measured in longitudinal magnetic fields with strength ranging from 0.36 to 3.75 kilooersteds.
19	342	L	1953	4.5		Zn 4	The above specimen measured in transverse magnetic fields with strength ranging from 0.36 to 3.90 kilooersteds.
20	342	L	1953	4.5		Zn 4	The above specimen measured in longitudinal magnetic fields with strength ranging from 0.36 to 3.85 kilooersteds.
21	97	L	1952	3.0-23	2-3	Zn 1	99.9995 pure; polycrystalline; 1-2 mm dia x 5 cm long; provided by Hilger H. S. Braud (MS 8392); annealed in evacuated quartz tube for several hrs at two-thirds the melting point.
22	97	L	1952	1.8-41	2-3	Zn 2	99.997 pure; single crystal; 1-2 mm dia x 5 cm long; provided by Imperial Smelting Corp; hexagonal axis at 80° to the specimen axis; annealed as the above specimen.
23	97	L	1952	3.0-40	2-3	Zn 3	Similar to the above specimen but hexagonal axis at 13° to the specimen axis.
24	280	L	1934	330.2	<0.8		99.99% Zn, 0.0047 Pb, 0.0008 Cd, 0.0004 Fe, and 0.0002 Cu; unstrained single crystals; grown from a single 50 lb slab of "Evanwall" zinc; specimens 30 cm long, area of cross section 1.24 cm <sup>2</sup> ; measured on specimens having various orientations with values of $\cos^2 \theta$ ranging from zero to 0.990 where $\theta$ is the angle between the normal to the basal cleavage plane and the axis of the rod.
25	280	L	1934	330.2	<0.8		Some of the above specimens strained by bending and straightening in both directions of the midpoint; measured with $\cos^2 \theta$ ranging from zero to 0.51.
26	280	L	1934	330.2	<0.8		The above specimens annealed at 380 C for 11 hrs.
27	280	L	1934	330.2	<0.8		"Optically mosaic" crystals; two of the specimens grown from the same allotment of zinc as the above specimens; the other two ( $\cos^2 \theta = 0.93$ and 0.935, respectively) grown from a different lot under different conditions; measured on the four specimens with $\cos^2 \theta$ ranging from 0.13 to 0.935.
28	280	L	1934	330.2	<0.8		The second one of the above specimens strained by bending and straightening.
29	280	L	1934	330.2	<0.8		The first specimen ( $\cos^2 \theta = 0.13$ ) and the above specimen annealed.
30	279	R	1951	331.4			Powdered; apparent density 2.454 g cm <sup>-3</sup> .
31	279	R	1951	330.8			Powdered; apparent density 2.443 g cm <sup>-3</sup> .
32	279	R	1951	331.0			Powdered; apparent density 2.465 g cm <sup>-3</sup> .
33	279	R	1951	331.1			Powdered; apparent density 2.456 g cm <sup>-3</sup> .
34	122	L	1955	2.7-21	3	Zn 4	99.997 pure; single crystal with hexagonal axis at 13° to rod axis; obtained from Imperial Smelting Corp; 2.75 cm x 0.0234 cm <sup>2</sup> .



SPECIFICATION TABLE NO. 6<sup>a</sup> (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)	Specifications and Remarks
35	431	E	1944	394-550				Single crystal; electrical resistivity reported as 8.32, 8.88, 9.59, 10.10, 10.85, 11.54, and 12.66 $\mu\text{ohm cm}$ at 121.1, 143.9, 169.6, 188.4, 213.8, 238.9, and 277.1 C, respectively. <sup>2, 65</sup>
36	431	E	1944	371-529				Polycrystal; electrical resistivity reported as 8.22, <sup>9, 09</sup> 9.09, 9.45, 10.19, 11.04, 11.85, 12.70, and 13.19 $\mu\text{ohm cm}$ at 97.9, 111.2, <sup>137, 6</sup> 137.6, 159.4, 186.4, 213.6, 247.1, and 255.8 C, respectively. <sup>128, 9</sup>
37	504	P	1961	295.2	± 5			Pure; specimen size 1.9 x 1.16 x 0.282 cm; thermal conductivity value calculated from measured thermal diffusivity using data of specific heat and density taken from Smithsonian Physical Tables (9th ed., 1954).
38	409	L	1958	0.14-0.87	10	Zn 1		0.001 impurity; single crystal; ~1.5 mm in dia and 100 mm long; specimen axis at an angle of 30 degrees to the [001] direction; measured in a magnetic field of 0.2 oersted; in superconducting state.
39	409	L	1958	0.18-0.91	10	Zn 1		The above specimen measured in a longitudinal magnetic field of 60 oersteds; in normal state.
40	409	L	1958	0.22-1.3	10	Zn 2		Similar to the above specimen but measured in a magnetic field of 0.2 oersted; in superconducting state.
41	294	L	1932	22-230		Zn 61		Single crystal; specimen 5.27 cm long, area of cross section 0.0552 cm <sup>2</sup> ; angle between specimen axis and hexagonal axis $\theta = 3.6^\circ$ ; electrical resistivity reported as 0.0674, 1.331, 5.69, and 6.16 $\mu\text{ohm cm}$ at -252, -190, 0, and 20 C, respectively.
42	294	L	1932	21-92		Zn 100		Single crystal; $\theta = 4.9^\circ$ ; electrical resistivity reported as 0.056, 1.333, 5.72, and 6.20 $\mu\text{ohm cm}$ at -252, -190, 0, and 20 C, respectively.
43	294	L	1932	21-233		Zn 72		Single crystal; specimen 6.13 cm long, area of cross section 0.0614 cm <sup>2</sup> ; $\theta = 8.7^\circ$ ; electrical resistivity reported as 0.0522, 1.3, 5.58, and 6.05 $\mu\text{ohm cm}$ at -252, -190, 0, and 20 C, respectively.
44	294	L	1932	21-296		Zn 101		Single crystal; specimen 4.94 cm long, area of cross section 0.0623 cm <sup>2</sup> ; $\theta = 79.7^\circ$ ; electrical resistivity reported as 0.0524, 1.179, 5.43, and 5.88 $\mu\text{ohm cm}$ at -252, -190, 0, and 20 C, respectively.
45	727, 291	L	1960	0.10-0.94		Zn 4		Single crystal; grown along the principal crystallographic direction by Kapitza's method; superconducting transition point 0.825 K; heat flow parallel to the hexagonal axis; in superconducting state.
46	727, 291	L	1960	0.10-1.09		Zn 7		Similar to above but heat flow perpendicular to the hexagonal axis; in superconducting state.
47	727, 291	L	1960	0.825		Zn 1		Similar to above.
48	727, 291	L	1960	0.825		Zn 2		Similar to above.
49	727, 291	L	1960	0.825		Zn 5		Similar to above but heat flow parallel to the hexagonal axis.

SPECIFICATION TABLE NO. 64 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
50	129	C	1933	462-553	5	Z. S.	99.9% Zn, 0.04 Pb, and 0.02 Fe; 2 cm dia x 15 cm long; specimen prepared by remelting commercially pure zinc and casting in graphite mold; lead used as comparative material (reference value taken from International Critical Table Vol. II: 0.352 Watt cm <sup>-1</sup> C <sup>-1</sup> at 0 C).
51	129	C	1933	313-596	5	Z. S.	Similar to the above specimen except commercial malleable nickel used as the indirect comparative material (based on the data of lead).
52	129	C	1933	342-602	5	Z. S.	Similar to that of the above specimen except zinc used as the indirect comparative material (based on the data of lead).
53	77	L	1900	291, 373		Zinc II	99.97 Zn (by difference), 0.01 Cd, 0.01 Fe, and 0.01 Pb; specimen 27 cm long and 1.805 cm in dia; density: 7.11 g cm <sup>-3</sup> ; electrical conductivity reported as 16.51 and 12.59 x 10 <sup>4</sup> ohm <sup>-1</sup> cm <sup>-1</sup> at 18 and 100 C, respectively.
54	77	L	1900	291, 373		Zinc II, wire	Similar to the above specimen but electrical conductivity reported as 15.98 and 12.42 x 10 <sup>4</sup> ohm <sup>-1</sup> cm <sup>-1</sup> at 18 and 100 C, respectively.
55	734	R	1926	323, 673			Specimen in the form of a hollow cylinder.
56	838	C	1966	576-874			Molten specimen placed in a hole 21 mm in dia drilled in an asbestos cement cylinder of 30 mm height; 1Kh18N9T steel used as comparative material.
57	850	L	1929	83-476			Prepared by melting pure zinc supplied by Firma Kahbaum in a quartz tube, then quickly solidified in cool water; electrical resistivity reported as 1.658, 5.663, 7.837, and 10.36 μhm cm at 83.2, 273, 374, and 476 K, respectively.

DATA TABLE NO. 68 THERMAL CONDUCTIVITY OF ZINC

(Impurity &lt; 0.20% each; total impurities &lt; 0.50%)

[Temperature, T, K; Thermal Conductivity,  $k$ , Watt  $\text{cm}^{-1}\text{K}^{-1}$ ]

T	k	T	k	T	k	T	k	T	k	T	k	T	k	T	k	cos $\theta$	k								
<b>CURVE 1</b>																									
323	1.159	129.2	1.15	428.2	1.07	608.2	0.996	2.15	2.26*	0.36	2.86	5.38	3.1	32.0	3.1	<b>CURVE 24* (cont.)</b>									
<b>CURVE 2</b>																									
409.2	1.075*	139.2	1.14	673.2	0.975	758.2	0.607	2.86	2.02	1.08	2.77	5.35	2.7	35.8	2.7	0.26	1.05								
426.2	1.075*	149.2	1.13	758.2	0.607	803.2	0.594	3.57	1.81	1.80	2.69	5.21	2.5	41.0	2.5	0.34	1.04								
436.2	1.071	158.2	1.12	858.2	0.586	908.2	0.577	3.73	1.77	2.52	2.60	4.88	<b>CURVE 23</b>												
443.2	1.075	166.2	1.11	1003.2	0.565	1003.2	0.565	<b>CURVE 13 (T = 4.6 K)</b>									3.0	3.4	0.815	1.02					
479.2	1.042	186.2	1.13	<b>CURVE 17 (T = 3.4 K)</b>																		3.5	4.2	0.990	1.01
552.2	1.021*	214.2	1.13	<b>CURVE 18 (T = 3.4 K)</b>																		4.0	5.0	<b>CURVE 25*</b>	
555.2	1.008*	234.2	1.14	<b>CURVE 19 (T = 4.5 K)</b>																		4.75	6.0	(T = 330.2 K)	
585.2	0.992	251.2	1.14	<b>CURVE 20* (T = 4.5 K)</b>																		5.25	6.8		
586.2	0.992	264.2	1.13	<b>CURVE 21</b>																		5.5	6.6		
640.2	0.996	277.2	1.11	<b>CURVE 22</b>																		5.75	7.4		
<b>CURVE 3</b>																		5.0	6.4						
402.2	1.096	296.2	1.11	<b>CURVE 26*</b>																		6.5	8.0		
515.2	1.046	297.2	1.12	<b>CURVE 27*</b>																		6.75	8.7		
586.2	1.008	297.2	1.11	<b>CURVE 28*</b>																		7.75	9.9		
635.2	0.975	381.2	1.10	<b>CURVE 29*</b>																		8.7	9.9		
673.2	0.920	434.2	1.03	<b>CURVE 30*</b>																		9.8	10.6		
733.2	0.586	434.2	1.03	<b>CURVE 31*</b>																		10.6	11.0		
810.2	0.577	434.2	1.03	<b>CURVE 32*</b>																		10.6	11.0		
831.2	0.573	434.2	1.03	<b>CURVE 33*</b>																		10.6	11.0		
<b>CURVE 4</b>																		11.3	10.6						
288.4	0.785	104.2	1.15	<b>CURVE 34*</b>																		12.8	10.1		
<b>CURVE 5</b>																		14.0	9.8						
360.8	1.08	118.2	1.14	<b>CURVE 35*</b>																		16.3	8.4		
390.8	1.06	131.2	1.15	<b>CURVE 36*</b>																		16.3	8.4		
433.5	1.03	141.2	1.11	<b>CURVE 37*</b>																		16.3	8.4		
499.1	1.01	141.2	1.11	<b>CURVE 38*</b>																		16.3	8.4		
562.0	0.996	159.2	1.12	<b>CURVE 39*</b>																		16.3	8.4		
<b>CURVE 6</b>																		17.5	2.4						
99.2	1.18	208.2	1.11	<b>CURVE 40*</b>																		22.3	5.5		
112.2	1.18	224.2	1.11	<b>CURVE 41*</b>																		22.3	5.5		
123.2	1.17	240.2	1.12	<b>CURVE 42*</b>																		23.8	5.0		
<b>CURVE 7</b>																		26.3	4.1						
499.2	1.075*	266.2	1.13	<b>CURVE 43*</b>																		29.5	3.3		
562.0	0.996	277.2	1.12	<b>CURVE 44*</b>																		31.5	3.0		
<b>CURVE 8</b>																		35.0	2.6						
360.8	1.08	277.2	1.14	<b>CURVE 45*</b>																		39.5	2.3		
390.8	1.06	284.2	1.14	<b>CURVE 46*</b>																		39.5	2.3		
433.5	1.03	293.2	1.13	<b>CURVE 47*</b>																		39.5	2.3		
499.1	1.01	297.2	1.11	<b>CURVE 48*</b>																		39.5	2.3		
562.0	0.996	300.2	1.12	<b>CURVE 49*</b>																		39.5	2.3		
<b>CURVE 9</b>																		cos $\theta$	k						
99.2	1.18	243.2	1.19	<b>CURVE 50*</b>																		cos $\theta$	k		
112.2	1.18	268.2	1.17	<b>CURVE 51*</b>																		cos $\theta$	k		
123.2	1.17	288.2	1.12	<b>CURVE 52*</b>																		cos $\theta$	k		
<b>CURVE 10</b>																		0.91	1.00						
402.2	1.096	37.2	1.57	<b>CURVE 53*</b>																		0.91	1.00		
515.2	1.046	99.2	1.47	<b>CURVE 54*</b>																		0.91	1.00		
586.2	1.008	213.2	1.35	<b>CURVE 55*</b>																		0.91	1.00		
635.2	0.975	283.2	1.26	<b>CURVE 56*</b>																		0.91	1.00		
673.2	0.920	381.2	1.19	<b>CURVE 57*</b>																		0.91	1.00		
733.2	0.586	381.2	1.19	<b>CURVE 58*</b>																		0.91	1.00		
810.2	0.577	381.2	1.19	<b>CURVE 59*</b>																		0.91	1.00		
831.2	0.573	381.2	1.19	<b>CURVE 60*</b>																		0.91	1.00		

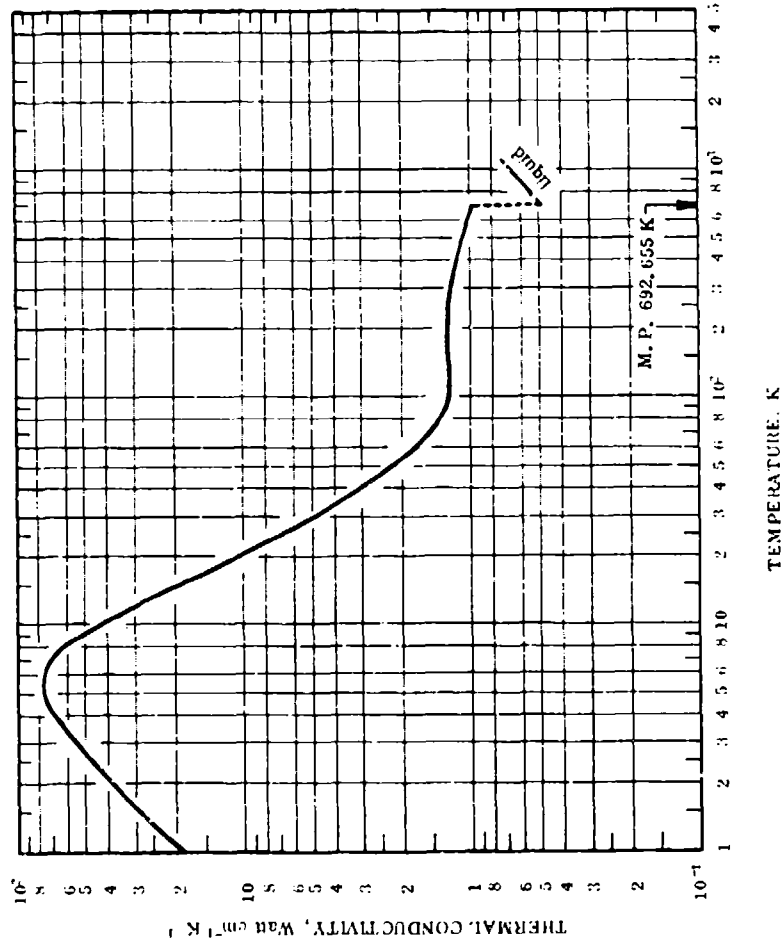
Not shown on plot

DATA TABLE NO. 68 (continued)

T	k	T	k	T	k	T	k	T	k	T	k
<u>CURVE 30*</u>		<u>CURVE 35 (cont.)</u>		<u>CURVE 42</u>		<u>CURVE 45 (cont.)</u>		<u>CURVE 46 (cont.)</u>		<u>CURVE 53*</u>	
331.4	0.00436	410.7	1.17	21.0	6.04	0.132	0.00475 <sup>c</sup>	3.800	6.000	291.2	1.11
<u>CURVE 31*</u>		432.5	1.15	21.0	6.02 <sup>b</sup>	0.140	0.00750 <sup>c</sup>	0.900	6.200	373.2	1.096
459.5	1.15	459.5	1.15	21.4	6.04*	0.170	0.0180 <sup>b</sup>	0.975	8.500	<u>CURVE 54*</u>	
486.7	1.13	486.7	1.13	22.0	5.70	0.225	0.0750	1.000	9.000		
330.8	0.00412	527.510 <sup>a</sup>	1.11	80.2	1.289	0.270	0.2000	<u>CURVE 47</u>			
<u>CURVE 32*</u>		528.9	1.11	82.8	1.275	0.320	0.290	291.2	1.07		
331.0	0.00502	<u>CURVE 37</u>		85.8	1.271	0.375	0.700	373.2	1.08		
		295.2	1.046	87.4	1.277	0.425	0.850	<u>CURVE 55*</u>			
<u>CURVE 33*</u>		71.1	0.75	90.9	1.264	0.430	1.050	323.2	1.11		
331.1	0.00483	91.7	1.261	91.7	1.261	0.480	1.400	673.2	0.900		
<u>CURVE 34</u>		<u>CURVE 40</u>		<u>CURVE 43</u>		<u>CURVE 48</u>		<u>CURVE 56</u>			
2.72	2.91	0.22	0.016 <sup>c</sup>	21.3	6.22	0.620	2.600	0.825	8.3		
3.46	4.07	0.23	0.013 <sup>b</sup>	21.8	6.25	0.650	3.200	<u>CURVE 49</u>			
4.58	5.28	0.26	0.028	21.9	6.03	0.725	2.900	0.825	2.1		
4.92	6.07	0.305	0.059	84.0	1.306	0.850	4.600	<u>CURVE 50</u>			
5.76	7.39	0.31	0.066	292.5	1.231	0.940	4.500	462.2	1.106		
8.84	8.78	0.37	0.13	292.8	1.248	0.100	0.005 <sup>b</sup>	493.2	1.087		
7.42	8.98	0.43	0.17	<u>CURVE 44</u>		0.115	0.008 <sup>b</sup>	692.2	0.9252		
7.90	9.72	0.49	0.20	21.0	4.80	0.130	0.010 <sup>b</sup>	713.2	0.5861*		
8.84	10.20	0.57	0.27	21.0	4.77	0.140	0.025 <sup>b</sup>	734.2	0.5903		
10.0	10.80	0.67	0.45	21.3	4.69	0.145	0.033 <sup>b</sup>	786.2	0.5987		
11.9	10.00	0.85	0.59	21.7	4.68	0.170	0.050 <sup>b</sup>	832.2	0.6154		
14.1	9.53	0.93	0.72	21.9	4.64	0.185	0.075 <sup>b</sup>	874.2	0.6322		
17.0	7.97	1.00	0.70	22.1	4.66	0.185	0.085 <sup>b</sup>	<u>CURVE 57*</u>			
21.2	5.80	1.10	0.80	22.2	4.59	0.205	0.100 <sup>b</sup>	83.2	1.256		
<u>CURVE 35</u>		1.30	0.92	23.2	4.38	0.220	0.140	273	1.252		
394.2	1.41	24.2	4.21	24.2	4.21	0.205	0.130	374	1.248		
406.0	1.38	24.4	4.14	24.4	4.14	0.220	0.140	476	1.239		
417.0	1.36	83.5	1.35	83.5	1.35	0.230	0.200	<u>CURVE 52*</u>			
442.7	1.32	85.4	1.345	85.4	1.345	0.250	0.300	342.2	1.138		
461.5	1.28	88.4	1.359	88.4	1.359	0.260	0.450	358.2	1.130		
486.9	1.23	92.0	1.324 <sup>c</sup>	92.0	1.324 <sup>c</sup>	0.290	0.650	398.2	1.123		
512.0	1.18	293.1	1.249 <sup>c</sup>	293.1	1.249 <sup>c</sup>	0.330	0.850	430.2	1.096		
550.2	1.12	295.2	1.230 <sup>b</sup>	295.2	1.230 <sup>b</sup>	0.350	1.250	438.2	1.096		
<u>CURVE 36</u>		27.7	3.49	296.1	1.234 <sup>b</sup>	0.375	1.400	467.2	1.097		
371.0	1.21	84.7	1.25	<u>CURVE 45</u>		0.400	1.750	492.2	1.078		
384.3	1.20	86.8	1.236	0.100	0.00055 <sup>a</sup>	0.440	1.700	504.2	1.080		
397.0	1.18	92.6	1.229	0.110	0.00110 <sup>a</sup>	0.450	2.500	544.2	1.059		
		291.3	1.206	0.115	0.00110 <sup>a</sup>	0.550	2.800	602.2	1.029		
		292.5	1.196	0.120	0.00170 <sup>a</sup>	0.625	4.500				
		0.27	0.27	0.125	0.00325 <sup>a</sup>	0.675	5.000				
		0.27	0.27	0.125	0.00325 <sup>a</sup>	0.750	5.500				

\* Not shown on plot

FIGURE AND TABLE NO. 68K RECOMMENDED THERMAL CONDUCTIVITY OF ZINC



RECOMMENDED VALUES\*  
(For Polycrystalline)

T <sub>1</sub>	k <sub>1</sub>	k <sub>2</sub>	T <sub>2</sub>	T <sub>1</sub>	k <sub>1</sub>	k <sub>2</sub>	T <sub>2</sub>
0	0	0	-459.7	500	1.11	64.1	440.3
1	19.0	1100	-457.9	600	1.05	60.7	620.3
2	37.9	2190	-456.1	700	0.92, 655	57.8	787.091
3	55.5	3210	-454.3	In Liquid State			
4	69.7	4030	-452.5				
5	77.8	4500	-450.7	692.655	(0.495)	†(28.6)	787.091
6	78.0	4510	-448.9	700	(0.499)	(28.6)	800.3
7	71.7	4140	-447.1	800	0.557	32.2	980.3
8	61.8	3570	-445.3	900	0.615	35.5	1160
9	51.9	3000	-443.5	1000	0.673	38.9	1340
10	43.2	2500	-441.7	1100	(0.730)	(42.2)	1520
11	36.4	2100	-439.9				
12	30.8	1780	-438.1				
13	26.1	1510	-436.3				
14	22.4	1290	-434.5				
15	19.4	1120	-432.7				
16	16.9	976	-430.9				
18	13.3	768	-427.3				
20	10.7	618	-423.7				
25	6.7	399	-414.7				
30	4.5	232	-405.7				
35	3.5	155	-396.7				
40	2.8	102	-387.7				
45	2.4	72	-378.7				
50	2.1	53	-369.7				
60	1.71	38.8	-351.7				
70	1.48	28.5	-333.7				
80	1.35	21.7	-315.7				
90	1.27	17.4	-297.7				
100	1.24	14.3	-279.7				
150	1.22	74.0	-189.7				
200	1.23	72.8	-99.7				
250	1.22	71.6	-9.7				
273.2	1.22	70.5	32.0				
300	1.21	69.9	80.3				
350	1.18	68.2	170.3				
400	1.16	67.0	260.3				

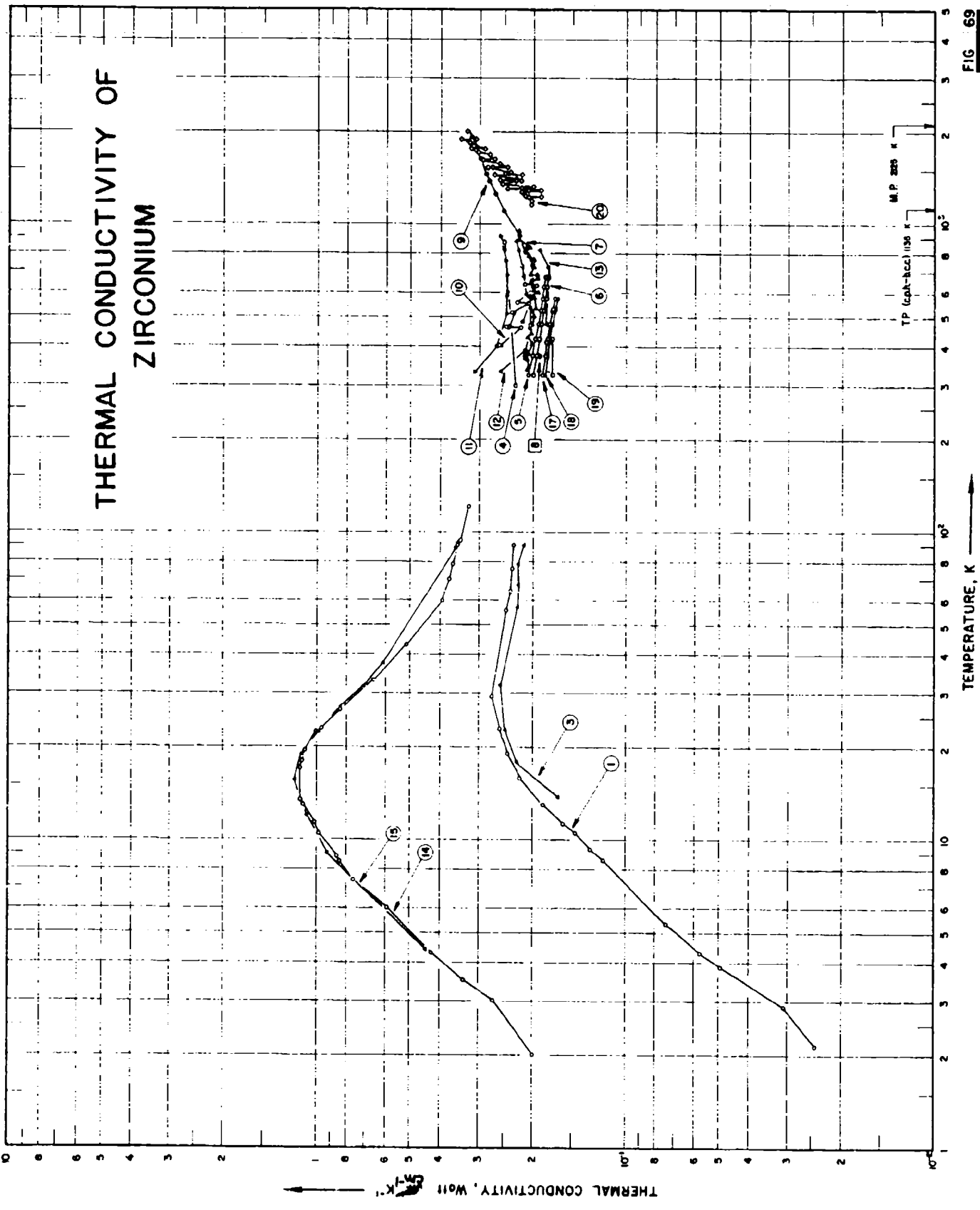
TEMPERATURE, K

REMARKS

The recommended values are for well-annealed 99.999% pure zinc with residual electrical resistivity  $\rho_0 = 0.00128 \mu\Omega \text{ cm}$  (characterization by  $\rho_0$  becomes important at temperatures below about 150 K). The values below 1.5 T are calculated to fit experimental data by using  $n = 3.00$ ,  $\sigma' = 1.83 \times 10^{-4}$ , and  $\beta = 0.0525$ . The recommended values that are supported by experimental thermal conductivity data are thought to be accurate to within 3% of the true values near room temperature and 3 to 10% at other temperatures.

\* T<sub>1</sub> in K, k<sub>1</sub> in Watt cm<sup>-1</sup> K<sup>-1</sup>, T<sub>2</sub> in F, and k<sub>2</sub> in Btu hr<sup>-1</sup> ft<sup>-1</sup> p<sup>-1</sup>. † Values in parentheses are extrapolated.

# THERMAL CONDUCTIVITY OF ZIRCONIUM



## SPECIFICATION TABLE NO. 69 THERMAL CONDUCTIVITY OF ZIRCONIUM

(Impurity &lt; 0.20% each; total impurities &lt; 0.50%)

[ For Data Reported in Figure and Table No. 69 ]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition ( weight percent), Specifications and Remarks
1	83	L	1956	2.2-91		Zr 1a	99.99 Zr; spectroanalysis shows Fe (all sensitive lines), Hf and Ni (all sensitive lines faintly), Si and Ti (some sensitive lines), and Al, Cr, Cu, and Mg (faintly visible); JMS5000 from Johnson, Matthey and Co.; 3 mm dia rod annealed at 950 C for 5 hrs in vacuum; electrical resistivity 48 $\mu\text{ohm cm}$ at 293 K, residual electrical resistivity 1.98 $\mu\text{ohm cm}$ ; mounted in the cryostat with a push fit into copper fitting; measured with the current lead (for the measurements of electrical resistivity) attached.
2	83	L	1956	3.3-90		Zr 1b	The above specimen measured with the current lead removed.
3	83	L	1956	14-90		Zr 1c	The above specimen unintentionally strained by drilling and tapping to insert the connectors for re-mounting.
4	292	P	1954	298, 873		2632 A	Preliminary results.
5	27	C	1953	323-673			Pure; 2 cm dia x 15 cm long; arc-melted from WAPD grade 1 crystal bar; Armco iron used as comparative material; data taken from smoothed curve.
6	27	C	1953	323-673		495	Pure; 2 cm dia x 15 cm long; arc-melted from Bureau of Mines sponge Zr; Armco iron used as comparative material; data taken from smoothed curve.
7	101	L	1955	336-950	2.0		Nominally pure; cylindrical specimen 7.938 in. long, 0.787 in. dia; obtained from Westinghouse; prepared from Foote Grade 1 crystal bar ingot; the ingot melted in tungsten arc furnace, forged at 845 C in argon to the size 10 x 1 x 1 in., annealed in vacuum for 0.5 hr at 1000 C; machined to final shape.
8	555	C	1956	373.2			Hafnium-containing crystal bar.
9	614	R	1961	484-1925	5		99.95 Zr, 0.029 Fe, 0.017 C, 0.0045 Hf, and < 0.031 other elements; specimen consisted of 5 one-in. dia disks; density 6.49 g $\text{cm}^{-3}$ .
10	194	L	1951	402-639	$\pm 5.0$	D-151	Assumed to be pure; 0.626 in. dia crystal bar; lot No. D-151; obtained from Argonne National Laboratory.
11	441		1957	331-917		Iodide Zirconium	99.9 pure; annealed in vacuum for 8 hrs at 700 C; electrical resistivity at 58.0, 124.1, 239.8, 321.0, 415.6, 499.6, 558.8, and 644.0 C being respectively, 36.1, 47.6, 66.6, 75.8, 87.0, 94.4, 100.0, and 106 $\mu\text{ohm cm}$ ; Lorenz number reported at these temperatures were 3.38, 3.33, 3.18, 3.11, 3.08, 3.04, 3.03, and 2.92 x 10 <sup>-8</sup> V <sup>2</sup> K <sup>-2</sup> , respectively.
12	441		1957	332-879			99.78 Zr, 0.14 Hf, and 0.08 C; electrical resistivity reported as 53.76, 64.93, 78.74, 87.71, 95.25, 105.26, 111.11, 120.48, and 125.00 $\mu\text{ohm cm}$ at 59.0, 117.0, 202.0, 262.0, 318.0, 402.0, 456.0, 548.0, and 606.0 C, respectively; Lorenz numbers reported at these temperatures were 3.46, 3.44, 3.54, 3.36, 3.37, 3.34, 3.37, 3.28 and 3.29 x 10 <sup>-8</sup> V <sup>2</sup> K <sup>-2</sup> , respectively.

SPECIFICATION TABLE NO. 69 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
13	295	L,R,C	1952	473-823			Pure; 98-100% of theoretical density.
14	401	L	1959	2.0-121		Zr 4	99.95 Zr, 0.0132 Hf, <0.0100 each of P and Zn, 0.0079 C, 0.0021-0.0050 O, 0.0003-0.0050 N, 0.0024 Fe, 0.0011 Ni, 0.0002-0.0007 each of Ca, Cr, H, Mo, and Si, and <0.0010 other elements; arc cast, annealed at 1100 C for 4 hrs, swaged at room temperature, annealed at 1000 C for 15 min and at 800 C in a vacuum of $1-2 \times 10^{-6}$ mm Hg for 15 min, cut to four lengths and clamped together.
15	401	L	1959	4.4-89		Zr 4a	The above specimen unclamped and retightened.
16	442	C	1951	323-573	± 3.0	Zr 1	0.04 Hf, 0.04 Fe, 0.02 Ni, 0.007 Ti, 0.001 Al, and 0.001 Sn; Westinghouse ingot D-216 forged at 950 C, and machined; electrical resistivity reported as 44.1 and 81.3 $\mu\text{ohm cm}$ at 298 and 533 K, respectively.
17	442	C	1951	323-573	± 3.0	SA 1568; Zr 7	0.1 Fe, 0.07 Ta, 0.07 C, 0.02 Al, 0.007 Ti, and 0.0055 N; obtained from ANI; annealed; electrical resistivity reported as 50.5, 68.2, and 85.1 $\mu\text{ohm cm}$ at 298, 415, and 533 K, respectively.
18	442	C	1951	323-573	± 3.0	SA 1576; Zr 8	0.16 Ta, 0.10 Fe, 0.06 Al, 0.02 C, 0.015 N, and 0.005 Ti; obtained from ANI; electrical resistivity reported as 52.4, 70.1, and 86.6 $\mu\text{ohm cm}$ at 298, 415, and 533 K, respectively.
19	715	L,C	1961	323, 423	± 5	050	99.827 Zr (by difference), 0.110, 0.045 Fe, 0.01 C, and 0.008 N; as-extruded rod 10 cm long, 1.27 cm in dia; arc-melted; electrical resistivity reported as 59.5 and 75 $\mu\text{ohm cm}$ at 323 and 423 K, respectively; Armco iron used as comparative material; energy flow also measured calorimetrically.
20	741	I	1965	1160-2000		Iodide Zirconium	99.5% pure; 14 mm dia x 65 mm long; vacuum annealed; density $6.45 \text{ g cm}^{-3}$ at room temperature.
21	843	-	1966	298.2			Powder specimen contained in a 0.75 in. dia x 2 in. long cylindrical cell; average grain size 36.9 $\mu$ ; thermal conductivity measured by using the transient line source method, measured in nitrogen at limiting high pressure.
22	843	-	1966	298.2			Similar to above except average grain size 48.0 $\mu$ .
23	843	-	1966	298.2			Similar to above except average grain size 57.5 $\mu$ .
24	843	-	1966	298.2			Similar to above except average grain size 67.8 $\mu$ .
25	843	-	1966	298.2			Similar to above except average grain size 84.5 $\mu$ .
26	843	-	1966	298.2			Similar to above except average grain size 95.3 $\mu$ .
27	843	-	1966	298.2			Similar to above except average grain size 137 $\mu$ .
28	843	-	1966	298.2			Similar to above except average grain size 164 $\mu$ .
29	843	-	1966	298.2			Similar to above except average grain size 199 $\mu$ .



SPECIFICATION TABLE NO. 69 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
30	843	-	1966	298.2			Similar to above except average grain size 228 $\mu$ .
31	843	-	1966	298.2			Similar to above except average grain size 318 $\mu$ .
32	843	-	1966	298.2			Similar to above; mesh size -70 +80; measured in nitrogen at 1 atm.
33	843	-	1966	298.2			Similar to above; mesh size -70 +80; measured in nitrogen under pressure in the range $1.06 \times 10^{-2} \sim 0.89 \times 10^3$ mm Hg.
34	843	-	1966	298.2			Similar to above; measured in helium under pressure in the range $1.00 \times 10^{-2} \sim 3.467 \times 10^3$ mm Hg.
35	843	-	1966	298.2			Similar to above; measured in argon under pressure in the range $1.00 \times 10^{-2} \sim 4.786 \times 10^3$ mm Hg.

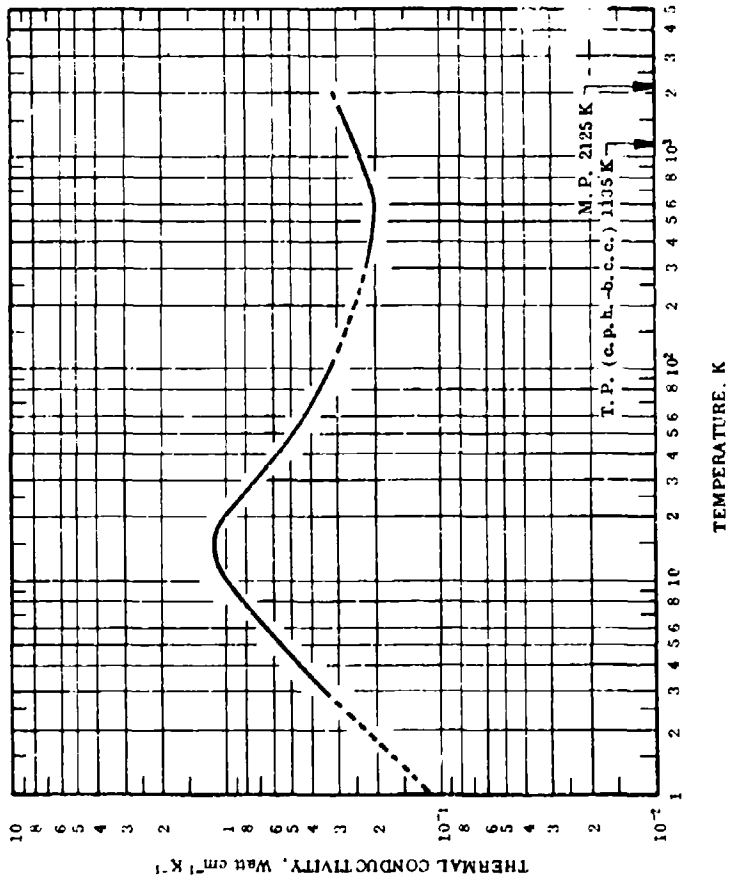


DATA TABLE NO. 69 (continued)

p(mm Hg)	k
CURVE 35*	
T = 298.7	
0.0100	0.0000469
1.36	0.000233
11.0	0.000787
135	0.00187
724	0.00230
2818	0.00264
4786	0.00264

\* Not shown on plot

FIGURE AND TABLE NO. 698 RECOMMENDED THERMAL CONDUCTIVITY OF ZIRCONIUM



RECOMMENDED VALUES\*  
(For Polycrystalline)

T <sub>1</sub>	k <sub>1</sub>	k <sub>2</sub>	T <sub>1</sub>	T <sub>2</sub>	k <sub>1</sub>	k <sub>2</sub>	T <sub>2</sub>
0	0	0	499.7	440.3	0.210	12.1	440.3
1	(0.111)†	(6.41)	457.9	620.3	0.207	12.0	620.3
2	(0.223)	(12.9)	456.1	800.3	0.209	12.1	800.3
3	0.333	19.2	454.3	980.3	0.216	12.5	980.3
4	0.442	25.5	452.5	1160	0.226	13.1	1160
5	0.549	31.7	450.7	1340	0.237	13.7	1340
6	0.652	37.7	448.9	1520	0.248	14.3	1520
7	0.748	43.2	447.1	1700	0.257	14.8	1700
8	0.837	48.4	445.3	1880	0.266	15.4	1880
9	0.916	52.9	443.5	2060	0.275	15.9	2060
10	0.984	56.9	441.7	2240	0.282	16.3	2240
11	1.04	60.1	439.9	2420	0.290	16.8	2420
12	1.08	62.4	438.1	2600	0.297	17.2	2600
13	1.11	64.1	437.3	2780	0.302	17.4	2780
14	1.13	65.3	434.5	2960	0.308	17.8	2960
15	1.13	65.3	432.7	3140	0.313	(18.1)	3140
16	1.12	64.7	430.9				
18	1.08	62.4	427.3				
20	1.01	58.4	423.7				
25	0.85	49.1	414.7				
30	0.74	42.8	405.7				
35	0.65	37.6	396.7				
40	0.58	33.5	387.7				
45	0.535	30.9	378.7				
50	0.497	28.7	369.7				
60	0.442	25.5	351.7				
70	0.403	23.3	333.7				
80	0.373	21.6	315.7				
90	0.350	20.2	297.7				
100	0.332	19.2	279.7				
150	(0.278)	(16.1)	189.7				
200	(0.252)	(14.6)	99.7				
250	(0.237)	(13.7)	9.7				
273.2	(0.232)	(13.4)	32.0				
300	0.227	13.1	80.3				
350	0.221	12.8	170.3				
400	0.216	12.5	260.3				

REMARKS

The recommended values are for well-annealed 99.95% pure zirconium with residual electrical resistivity  $\rho_0 = 0.219 \mu\Omega \text{ cm}$  (characterization by  $\rho_0$  becomes important below room temperature). The values below  $1.5 T_m$  are calculated to fit the experimental data by using  $n = 2.2$ ,  $c_1 = 7.45 \times 10^{-4}$ , and  $\beta = 8.98$ . The recommended values that are supported by experimental thermal conductivity data are thought to be accurate to within 5% of the true values near room temperature and 5 to 15% at other temperatures.

\* T<sub>1</sub> in K, k<sub>1</sub> in Watt cm<sup>-1</sup> K<sup>-1</sup>, T<sub>2</sub> in F, and k<sub>2</sub> in Btu hr<sup>-1</sup> ft<sup>-1</sup> F<sup>-1</sup>.

† Values in parentheses are extrapolated or interpolated.

SPECIFICATION TABLE NO. 70 THERMAL CONDUCTIVITY OF [ALUMINUM + ANTIMONY] ALLOYS

Al - Sb 99.50% impurity 0.20% each

Curve No.	Rel. Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) Al	Composition (weight percent) Sb	Composition (continued), Specifications and Remarks
1	L	1925	325.2			99.0	10.0	Approx composition, total impurity 0.03 in each metal; specimen 1.9 cm in dia and 10 cm long, supplied by Bakers; electrical conductivity $24.1 \times 10^4 \text{ ohm}^{-1}\text{cm}^{-1}$ at 23 C.
2	L	1925	325.2			89.0	20.0	Similar to above specimen except electrical conductivity, $17.9 \times 10^4 \text{ ohm}^{-1}\text{cm}^{-1}$ at 23 C.
3	L	1925	325.2			79.0	30.0	Similar to above specimen except electrical conductivity, $13.3 \times 10^4 \text{ ohm}^{-1}\text{cm}^{-1}$ at 23 C.
4	L	1925	325.2			69.0	40.0	Similar to above specimen except electrical conductivity, $7.90 \times 10^4 \text{ ohm}^{-1}\text{cm}^{-1}$ at 23 C.
5	L	1925	325.2			59.0	50.0	Similar to above specimen except electrical conductivity, $5.14 \times 10^4 \text{ ohm}^{-1}\text{cm}^{-1}$ at 23 C.

DATA TABLE NO. 70 THERMAL CONDUCTIVITY OF [ALUMINUM + ANTIMONY] ALLOYS

Al - Sb 99.50% impurity 0.20% each

Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup>K<sup>-1</sup>

T	k	T	k
CURVE 1			
325.2	1.833	325.2	1.004
CURVE 2			
325.2	1.594	325.2	0.808
CURVE 3			
325.2	1.406		

No graphical presentation

FIGURE SHOWS ONLY II OF THE CURVES REPORTED IN TABLE

# THERMAL CONDUCTIVITY OF ALUMINUM + COPPER ALLOYS

[Al + Cu ≥ 99.80%; impurity ≤ 0.20% each]

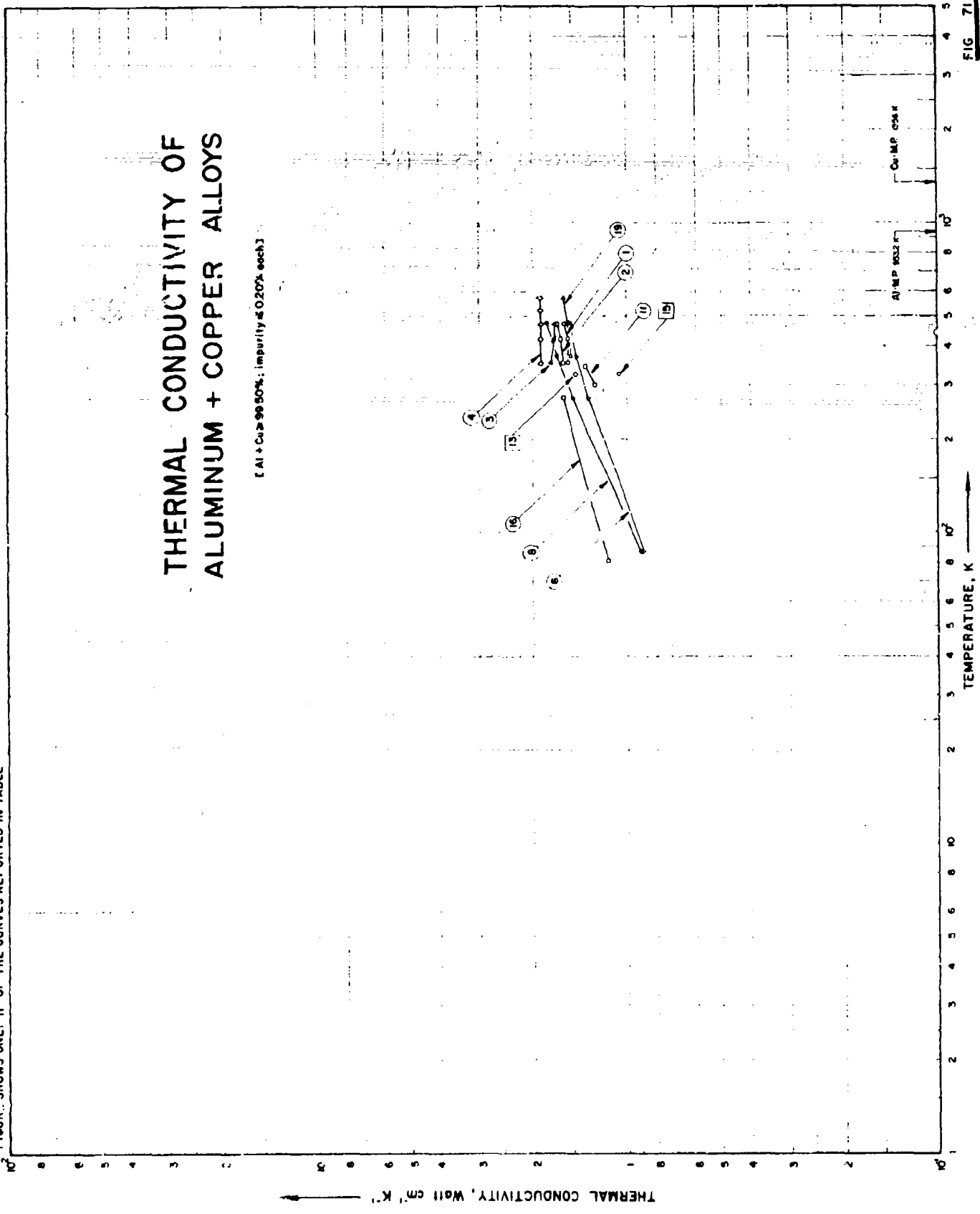


FIG 71

## SPECIFICATION TABLE NO. 71 THERMAL CONDUCTIVITY OF [ALUMINUM + COPPER] ALLOYS

(Al + Cu = 99.50%; impurity = 0.20% each)

(For Data Reported in Figure and Table No. 71)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) Al	Cu	Composition (continued), Specifications and Remarks
1	55	⊕L	1928	353-473		No. 655	86.0	14.0	Approx. composition: specimen 2.53 cm in diameter and 38 cm long; cast; electrical resistivity reported at 353, 423, 473, 523, 573, and 623 K respectively as 5.24, 6.25, 6.97, 7.63, 8.40 and 9.14 $\mu$ ohm cm.
2	55	⊕L	1928	353-473		No. 671	88.0	12.0	Specimen 2.53 cm in diameter and 38 cm long; cast; electrical resistivity reported at 353, 423, 473, 523, 573, and 623 K respectively as 3.20, 5.96, 6.51, 7.03, 7.57 and 8.11 $\mu$ ohm cm.
3	55	⊕L	1928	353-473		No. 921	88.0	12.0	Trace Fe; specimen 2.53 cm in diameter and 38 cm long; cast; electrical resistivity reported at 353, 423, 473, 523, 573, and 623 K respectively as 4.64, 5.61, 6.34, 7.12, 7.95, 8.92 $\mu$ ohm cm.
4	55	⊕L	1928	353-573		No. 2313	92.0	8.0	Specimen 2.53 cm in diameter and 38 cm long; cast; electrical resistivity reported at 353, 423, 473, 523, 573 and 623 K respectively as 4.06, 4.77, 5.40, 6.16, 7.03 and 8.08 $\mu$ ohm cm.
5	55	⊕L	1928	353-573		No. 2312	95.5	4.5	Specimen 2.53 cm in diameter and 38 cm long; cast; electrical resistivity reported at 353, 423, 473, 523, 573 and 623 K respectively as 4.04, 4.96, 5.61, 6.26, 6.92 and 7.58 $\mu$ ohm cm.
6	93	L	1931	87-476	3-4		92.0	8.0	Approx. composition; cast; electrical conductivity reported at 87, 273, 373 and 476 K respectively as 65.1, 29.3, 20.2 and 14.6 x 10 <sup>6</sup> ohm <sup>-1</sup> cm <sup>-1</sup> .
7	93	L	1931	87-476	3-4		92.0	8.0	Approx. composition; cast; electrical conductivity reported at 87, 273, 373 and 476 K respectively as 65.1, 29.3, 20.2, and 14.6 x 10 <sup>6</sup> ohm <sup>-1</sup> cm <sup>-1</sup> .
8	93	L	1931	87-476	3-4		85.0	15.0	Approx. composition; cast; electrical conductivity reported at 87, 273, 373 and 476 K respectively as 59.6, 22.3, 16.0 and 14.2 x 10 <sup>6</sup> ohm <sup>-1</sup> cm <sup>-1</sup> .
9	53	E	1927	353-423	1.0		96.0	4.0	Approx. composition, cast.
10	53	E	1927	373.2			88.0	12.0	Approx. composition; cast; density 2.95 g cm <sup>-3</sup> ; electrical conductivity 0.16 x 10 <sup>6</sup> ohm <sup>-1</sup> cm <sup>-1</sup> at 100 C.
11	25		1921	301, 346			~92.0	~8.0	Trace Si.
12	230	L	1925	326.2			90.0	10.0	Approx. composition, total impurity < 0.03 in each metal; prepared by fusing; specimen 1.9 cm in diameter and 10 cm long; supplied by Baker; electrical conductivity 26.0 x 10 <sup>6</sup> ohm <sup>-1</sup> cm <sup>-1</sup> at 23 C.
13	230	L	1925	326.2			80.0	20.0	Similar to the above; specimen except electrical conductivity, 20.9 x 10 <sup>6</sup> ohm <sup>-1</sup> cm <sup>-1</sup> at 23 C.

SPECIFICATION TABLE NO. 71 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) Al	Composition (weight percent) Cu	Composition (continued), Specifications and Remarks
14	230	L	1925	426.2			70.0	30.0	Similar to above except electrical conductivity $14.5 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 23°C.
15	230	L	1925	426.2			50.0	50.0	Similar to above except electrical conductivity $15.3 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 23°C.
16	36	R	1935	81.273	1.0		96.0	4.0	Approx. composition; cast sheet; annealed at 510°C for 45 min. and quenched in ice water.
17	826		1934	293.473			88.0	12.0	Density $3.0 \text{ g cm}^{-3}$ ; Brinell hardness number 80.
18	98	L	1922	373-573		V 671 A	88.0	12.0	Approx. composition; specimen 15 in. long and 1 in. in diameter, supplied by Metallurgical Dept. of National Physical Laboratory (England); chill cast.
19	98	L	1922	373-573		V 671 D	88.0	12.0	Approx. composition; commercially pure aluminum; specimen 15 in. long and 1 in. in diameter, supplied by Metallurgical Dept. of National Physical Lab.; annealed at 450°C.
20	98	L	1922	373-573		V 671 C	88.0	12.0	Similar to above specimen except sand cast.



DATA TABLE NO. 71 THERMAL CONDUCTIVITY OF ALUMINUM-COPPER ALLOYS

(Al-Cu 99.50% impurity 0.20% each)

Temperature, T, K. Thermal Conductivity, k, Watt cm<sup>-1</sup> K<sup>-1</sup>.

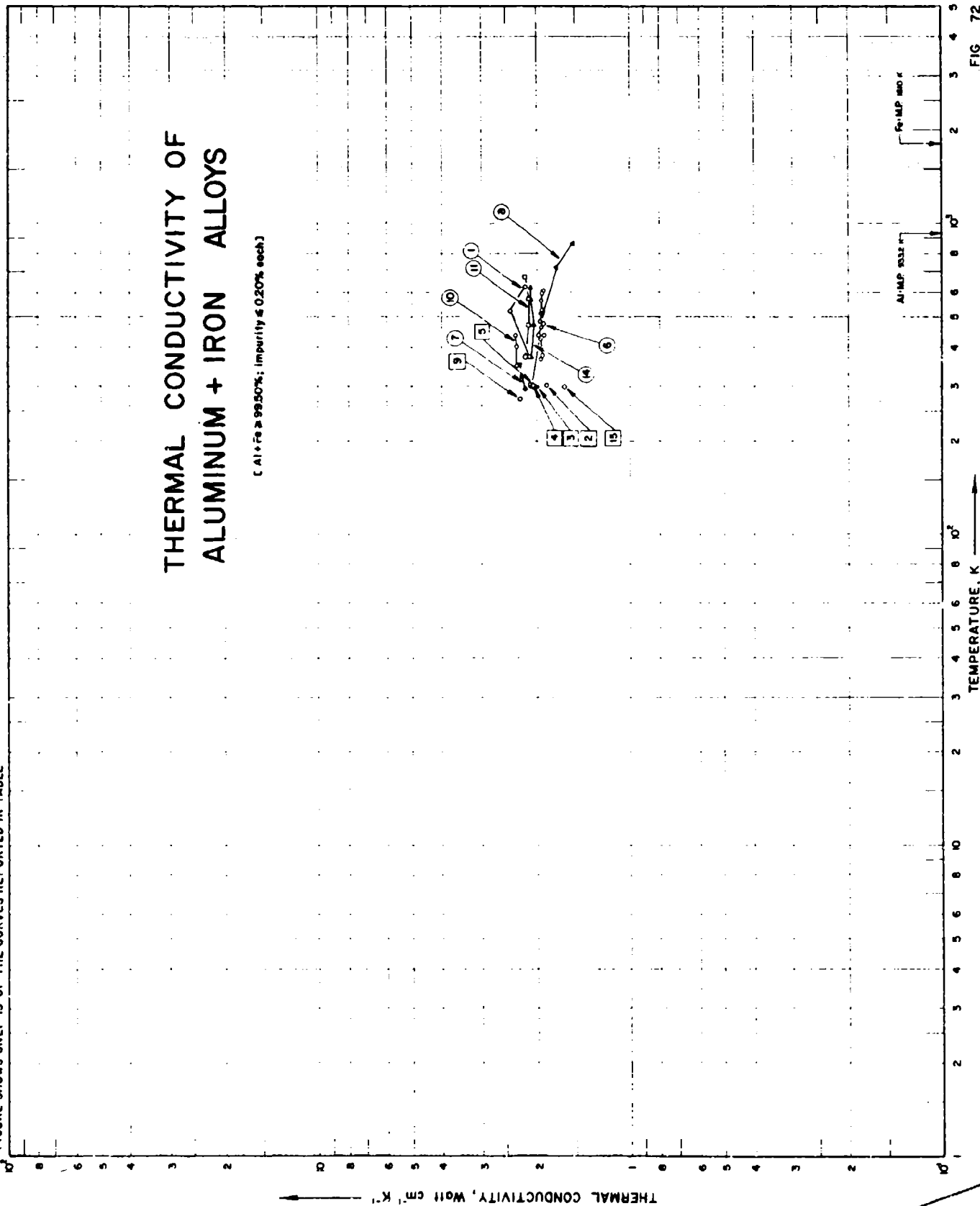
T	k	T	k	T	k	T	k
<u>CURVE 1</u>							
353.00	1.548	87	0.895	326.2	1.06		
423.00	1.548	273	1.32				
473.00	1.550	373	1.45				
476	1.52	476	1.52				
<u>CURVE 2</u>							
353.00	1.550			81.20	1.155		
423.00	1.612			273.20	1.596		
473.00	1.674						
<u>CURVE 3</u>							
353.00	1.757	87	0.904				
423.00	1.715	273	1.48				
473.00	1.715	373	1.67	293	1.46		
		476	1.81	473	1.67		
<u>CURVE 4</u>							
353.00	1.883						
423.00	1.883	353.0	1.674	373	1.80		
473.00	1.883	373.0	1.674	473	1.64		
523.00	1.883	393.0	1.674	573	1.69		
573.00	1.883	423.0	1.674				
<u>CURVE 5</u>							
353.00	1.883						
423.00	1.883			373	1.53		
473.00	1.883			473	1.55		
523.00	1.883	373.2	1.42	573	1.59		
573.00	1.883						
<u>CURVE 6</u>							
353.00	1.883	301.00	1.268				
423.00	1.883	345.80	1.547	373	1.42		
473.00	1.883			473	1.51		
523.00	1.883			573	1.55		
573.00	1.925	336.2	1.61				
<u>CURVE 7</u>							
87	0.887						
273	1.32	326.2	1.46				
373	1.44						
476	1.52						
		326.2	1.30				

Not shown on plot

FIGURE SHOWS ONLY 13 OF THE CURVES REPORTED IN TABLE

# THERMAL CONDUCTIVITY OF ALUMINUM + IRON ALLOYS

(Al + Fe a 99.50%; Impurity  $\leq$  0.20% each)



SPECIFICATION TABLE NO. 72 THERMAL CONDUCTIVITY OF [ALUMINUM + IRON] ALLOYS

(Al + Fe > 99.50%; impurity < 0.20% each)

[ For Data Reported in Figure and Table No. 72 ]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) Al	Composition (weight percent) Fe	Composition (continued), Specifications, and Remarks
1	225	L	1928	373-623			99.57	0.24	0.19 Si, trace Cu; specimen 12 in. long and 1 in. in diameter; annealed at 530 C for 5 hrs. before machining.
2	403	E	1925	302.2	< 0.5	AL	99.354	0.509	0.137 Si; specimen 5 mm thick and 29 cm long; chill-cast; specific gravity 2.70; specific resistance $0.354 \times 10^{-5}$ ohm $cm^{-1}$ .
3	408	E	1925	293.2	< 0.5	AL	99.354	0.509	0.137 Si; the above specimen annealed at 450 C for 30 min.; specific resistance $0.331 \times 10^{-5}$ ohm $cm^{-1}$ .
4	408	E	1925	301.2	< 0.5	AL	99.354	0.509	0.137 Si; specimen about 3 mm thick and 20 cm long; forged and cold-drawn; specific resistance $0.316 \times 10^{-5}$ ohm $cm^{-1}$ .
5	408	E	1925	304.2	< 0.5	AL	99.354	0.509	0.137 Si; the above specimen annealed at 500 C for 30 min.; specific resistance $0.313 \times 10^{-5}$ ohm $cm^{-1}$ .
6	30	L	1925	364-609	< 2.0		99.5	0.43	0.07 Si, trace Cu; commercially pure Al; specimen 15.5 in. long and 0.73 in. in diameter, cast.
7	17	L	1958	293-353	± 1.0	DIN 712	99.5	0.32	0.019 Cu, 0.034 Zn, 0.019 Mg, 0.021 Mn, 0.012 Ti, 0.16 Si; commercial pure aluminum; density $2.703 \text{ g cm}^{-3}$ at 20 C.
8	15	F	1947	294-863	1.0-5.0	AL-1	99.20	0.67	0.10 Si, 0.01 Cu, < 0.01 Mn and Mg; specimen 25 cm long and 2.5 cm in diameter.
9	410	R	1935	273.2	1.0	AL-2	99.66	0.2	0.14 Si.
10	54	L	1927	350-437			99.6	0.2	0.2 Si, single crystal; specimen 2.54 cm in diameter and 38 cm long, supplied by National Physical Lab.; specific resistivity $2.89 \times 10^{-5}$ ohm $cm^{-1}$ at 18 C.
11	98	L	1922	373-673		1	99.48	0.38	0.14 Si, trace Cu supplied by National Physical Lab.; chill-cast.
12	98	L	1922	373-673		2	99.48	0.38	0.14 Si, trace Cu; supplied by National Physical Lab.; chill-cast.
13	98	L	1922	373-623		1	99.48	0.38	0.14 Si, trace Cu; supplied by National Physical Lab.; sand-cast.
14	98	L	1922	373-623		2	99.48	0.38	0.14 Si, trace Cu; supplied by National Physical Lab.; sand-cast.
15	827	L	1958	298.2			92.4	7.6	Produced by powder metallurgical process; extruded rod 3/4 in. in diameter; heated at 800 F for 100 hrs.; density $2.89 \text{ g cm}^{-3}$ .

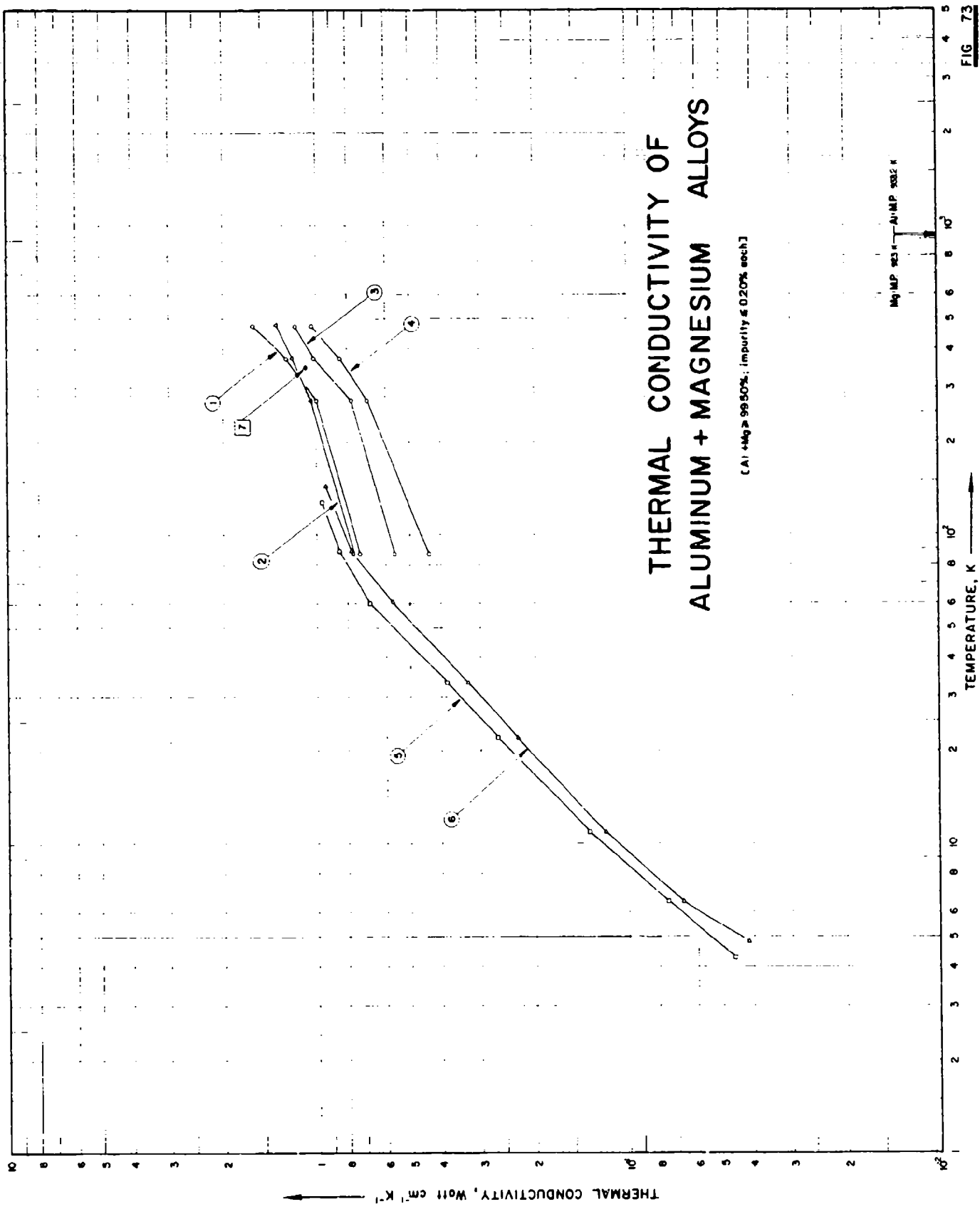
DATA TABLE NO. 72 THERMAL CONDUCTIVITY OF [ALUMINUM + IRON] ALLOYS

(Al + Fe) 99.50%; impurity = 0.20% each

Temperature, T, K. Thermal Conductivity, k, Watt cm<sup>-1</sup> K<sup>-1</sup>

T	k	T	k	T	k
<u>CURVE 1</u>					
373.2	2.117	298.2	2.090	298.2	1.63
523.2	2.410	518.2	1.910		
623.2	2.184	723.2	1.730		
		863.2	1.530		
<u>CURVE 2</u>					
302.2	1.858				
<u>CURVE 3</u>					
299.2	2.000	273.2	2.259		
<u>CURVE 4</u>					
301.2	2.059	349.6	2.305		
		400.6	2.310		
		437.1	2.330		
<u>CURVE 5</u>					
304.2	2.071	373.2	2.18		
		473.2	2.13		
		573.2	2.13		
		673.2	2.18		
<u>CURVE 6</u>					
364.2	1.95				
367.2	1.95				
375.2	1.92				
423.7	1.92	373.2	2.13		
425.2	1.95	473.2	2.13		
427.7	1.95	573.2	2.18		
438.2	1.88	673.2	2.22		
438.7	1.97				
461.2	1.91				
475.7	1.90				
485.2	1.96	373.2	2.09		
512.7	1.95	473.2	2.09		
527.7	1.93	573.2	2.18		
565.2	1.94	623.2	2.18		
594.2	1.92				
606.7	1.90				
<u>CURVE 7</u>					
293.2	2.145	373.2	2.09		
323.2	2.220	473.2	2.05		
353.2	2.255	573.2	2.09		
		623.2	2.09		

Not shown on plot



## SPECIFICATION TABLE NO. 73 THERMAL CONDUCTIVITY OF [ALUMINUM + MAGNESIUM] ALLOYS

(Al + Mg - 99.50%; impurity - 0.20% each)

[For Data Reported in Figure and Table No. 73.]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)		Composition (continued), Specifications and Remarks
							Al	Mg	
1	93	L	1931	87-476	3.0-4.0		92.0	8.0	Approx. composition; cast; electrical conductivity reported at 87, 273, 373 and 476 K respectively as 20.02, 13.21, 10.5 and $8.8 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$
2	93	L	1931	87-476	3.0-4.0		92.0	8.0	Approx. composition; annealed; electrical conductivity reported at 87, 273, 373 and 476 K respectively as 24.5, 15.05, 12.25 and $10.25 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$
3	93	L	1931	87-476	3.0-4.0		88.0	12.0	Approx. composition; cast; electrical conductivity reported at 87, 273, 373 and 476 K respectively as 19.6, 11.95, 9.4, $7.85 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$
4	93	L	1931	87-476	3.0-4.0		86.0	14.0	Approx. composition; annealed; electrical conductivity reported at 87, 273, 373 and 476 K respectively as 12.7, 8.96, 8.05 and $7.6 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$
5	828		1960	4.3-128		5052	97.7-97.1	2.2-2.8	0.10 Mn; annealed.
6	828		1960	4.8-144		5154	96.8-96.0	3.1-3.9	0.10 Mn; annealed.
7	829		1940	348.2		Magnalium	93.0	7.0	Approx. composition; specimen 15 mm in diameter and 72 mm long; density $2.63 \text{ g cm}^{-3}$ .

## DATA TABLE NO. 73 THERMAL CONDUCTIVITY OF (ALUMINUM + MAGNESIUM) ALLOYS

(Al + Mg : 99.50%, impurity : 0.20% each)

(Temperature, T, K. Thermal Conductivity, k, Watt cm<sup>-1</sup> K<sup>-1</sup>)

T	k	T	k
<u>CURVE 1</u>			
87	0.728	22.1	0.228
273	1.000	33.2	0.327
373	1.264	60.9	0.571
476	1.598	88.7	0.779
		144.3	0.935
<u>CURVE 2</u>			
87	0.766		
273	1.046		
373	1.201		
476	1.356		
<u>CURVE 3</u>			
87	0.561		
273	0.774		
373	1.013		
476	1.184		
<u>CURVE 4</u>			
87	0.435		
273	0.690		
373	0.845		
476	1.042		
<u>CURVE 5</u>			
4.28	0.0462		
6.50	0.0753		
10.9	0.134		
22.1	0.265		
33.2	0.384		
60.9	0.675		
88.7	0.846		
127.6	0.961		
<u>CURVE 6</u>			
4.83	0.0415		
6.50	0.0675		
10.9	0.119		
<u>CURVE 6 (cont.)</u>			
<u>CURVE 7</u>			
348.2 1.087			

# THERMAL CONDUCTIVITY OF ALUMINUM + SILICON ALLOYS

[Al + Si ≥ 99.50%; impurity ≤ 0.20% each]

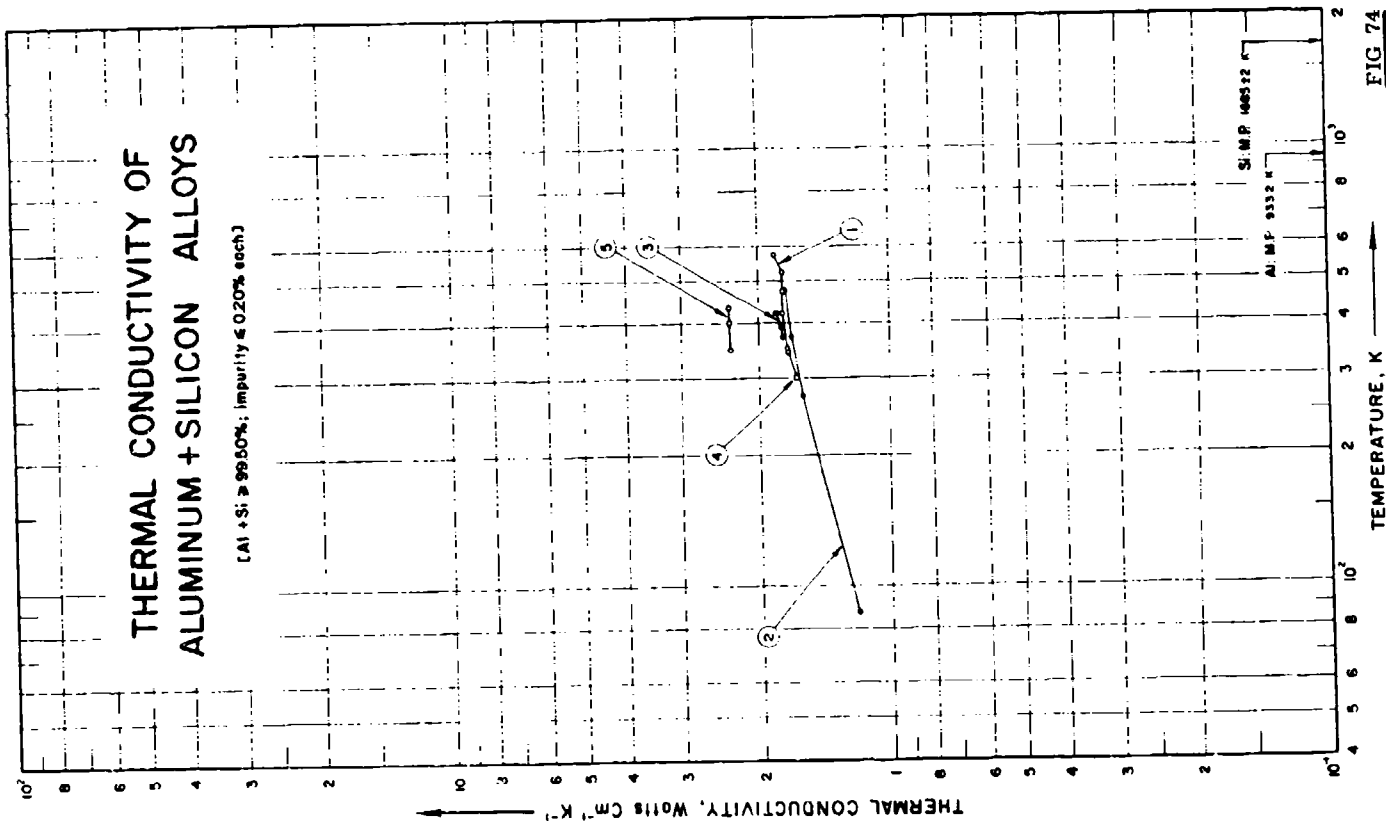


FIG 74



## SPECIFICATION TABLE NO. 74 THERMAL CONDUCTIVITY OF ALUMINUM + SILICON ALLOYS

(Al + Si : 99.50%; impurity &lt; 0.20% each)

[ For Data Reported in Figure and Table No. 74 ]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)		Composition (continued), Specifications and Remarks
							Al	Si	
1	55	±L	1928	353-573		2358	87.0	13.0	Nominal composition; fine structure cause by adding trace Na at high temp.; specimen 1 1/4 in. in diameter and 15 in. long; vertically sand cast; annealed at 450 C for 1 hr.; electrical resistivity reported at 353, 423, 473, 523, 573 and 623 K respectively as 5.36, 6.48, 7.31, 8.18, 9.07 and 10.0 μ ohm cm.
2	93	L	1931	87-476	3.0-4.0	Alusil	80.0	20.0	Nominal composition; as cast; electrical conductivity reported at 87, 273, 373 and 476 K respectively as 41.10, 17.23, 13.40 and 10.6 x 10 <sup>4</sup> ohm <sup>-1</sup> cm <sup>-1</sup> .
3	53	E	1927	373-423	1.0	Alpax	87.0	13.0	Approx. composition; as cast; density 2.65 g cm <sup>-3</sup> .
4	25		1921	301, 346			89.0-86.0	11.0-14.0	No details reported.
5	54	L	1927	350-437			99.6	0.2	0.2 Fe, single crystal; specimen 2.54 cm in diameter and 38 cm long; supplied by National Physical Lab.; specific resistivity 2.89 x 10 <sup>-6</sup> ohms cm <sup>-1</sup> at 18 C.

DATA TABLE NO. 74 THERMAL CONDUCTIVITY OF (ALUMINUM + SILICON) ALLOYS  
(Al + Si : 99.50%, impurity : 0.2% each)

(Temperature, T. K. Thermal Conductivity,  $k$ , Watt  $\text{cm}^{-1} \text{K}^{-1}$ )

T	k
<u>CURVE 1</u>	
353.0	1.715
423.0	1.757
473.0	1.757
523.0	1.757
573.0	1.799
<u>CURVE 2</u>	
87.0	1.205
273.0	1.586
373.0	1.686
476.0	1.745
<u>CURVE 3</u>	
373.0	1.757
393.0	1.774
423.0	1.820
<u>CURVE 4</u>	
501.0	1.632
546.2	1.703
<u>CURVE 5</u>	
349.6	2.305
400.6	2.310
437.1	2.330

SPECIFICATION TABLE NO. 75 THERMAL CONDUCTIVITY OF [ALUMINUM + TIN] ALLOYS  
(Al + Sn > 99.50%; impurity < 0.20% each)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) Al Sn	Composition (continued), Specifications and Remarks
1	230	L	1925	324.2			90.0 10.0	Approx composition; total impurity < 0.03 in each metal; specimen 1.9 cm dia and 10 cm long; supplied by Baker; electrical conductivity, $28.9 \times 10^4 \text{ ohm}^{-1}\text{cm}^{-1}$ at 22 C.
2	230	L	1925	324.2			70.0 30.0	Similar to above specimen except electrical conductivity, $22.8 \times 10^4 \text{ ohm}^{-1}\text{cm}^{-1}$ at 22 C.
3	230	L	1925	324.2			50.0 50.0	Similar to above specimen except electrical conductivity, $19.1 \times 10^4 \text{ ohm}^{-1}\text{cm}^{-1}$ at 22 C.

DATA TABLE NO. 75 THERMAL CONDUCTIVITY OF [ALUMINUM + TIN] ALLOYS

(Al + Sn > 99.50%; impurity < 0.20% each)

[Temperature, T, K; Thermal Conductivity, k,  $\text{Watt cm}^{-1}\text{K}^{-1}$ ]

T k

CURVE 1<sup>o</sup>

324.2 1.862

CURVE 2<sup>o</sup>

324.2 1.732

CURVE 3<sup>o</sup>

324.2 1.393

No graphical presentation

FIGURE SHOWS ONLY 13 OF THE CURVES REPORTED IN TABLE

# THERMAL CONDUCTIVITY OF ALUMINUM + URANIUM ALLOYS

(Al + U ≥ 99.50%; impurity ≤ 0.20% each)

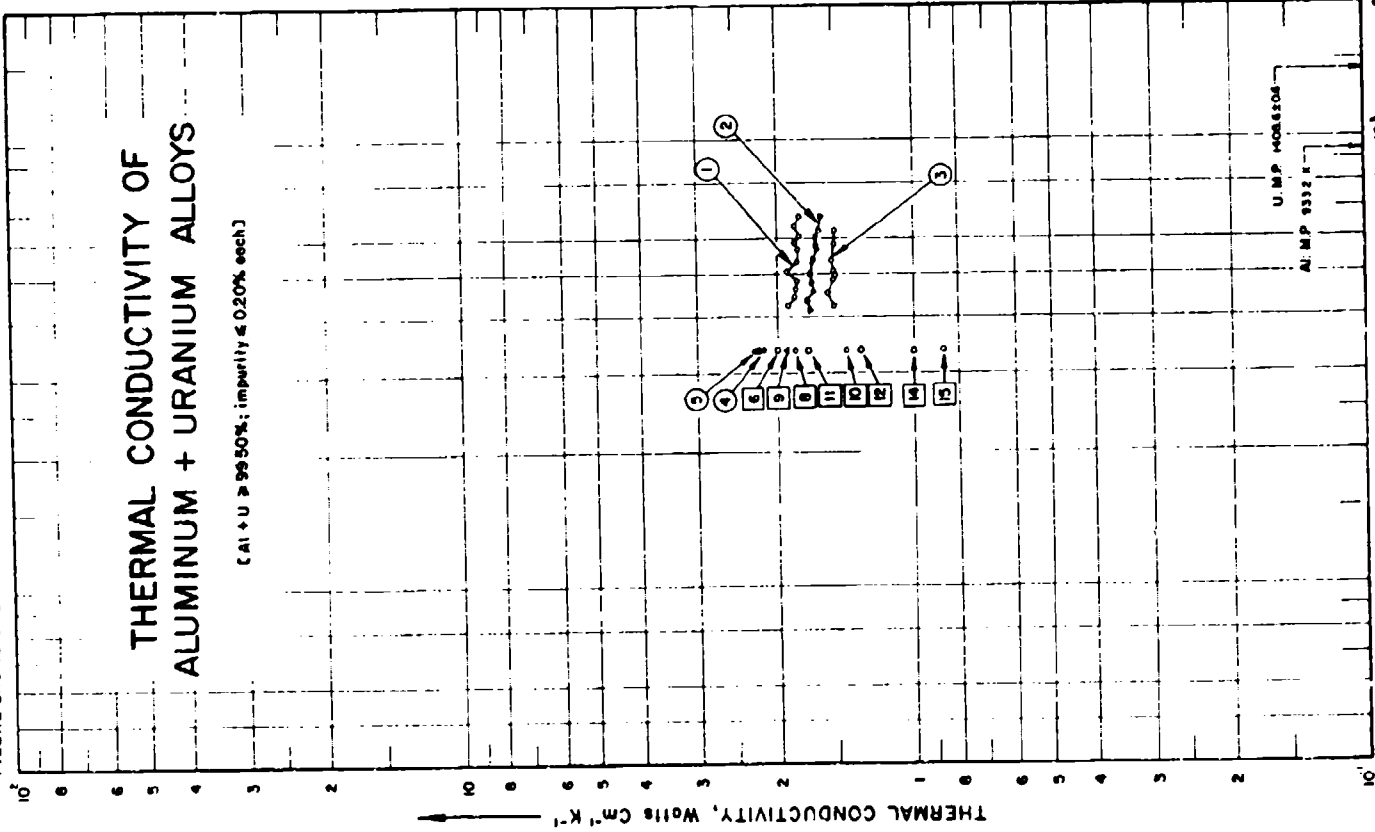


FIG 76

SPECIFICATION TABLE NO. 76 THERMAL CONDUCTIVITY OF ALUMINUM + URANIUM ALLOYS

(Al + U = 99.50%; Impurity = 0.20% each)

[For Data Reported in Figure and Table No. 76]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) U	Composition (continued), Specifications and marks
1	125	C	1955	424-671			87.5	Nominal composition; forged bar; annealed for 0.5 hr. at 370 C; pure lead used as standard.
2	125	C	1955	415-671			77.3	Similar to the above specimen.
3	125	C	1955	426-625			39.5	Similar to the above specimen.
4	591	C	1963	338.2			99.5	Nominal composition; cast from reactor-grade uranium (99.5% pure) and aluminum (99.99% pure); specimen 3 in. long, 0.500 in. diameter; measured in a vacuum of $< 5 \times 10^{-4}$ mm Hg; copper used as standard.
5	591	C	1963	338.2			99.5	The above specimen heat treated 5 days at 620 C.
6	591	C	1963	338.2			93.03	Nominal composition; cast from reactor-grade uranium (99.5% pure) and aluminum (99.99% pure); specimen 3 in. long, 0.500 in. dia.; measured in a vacuum of $< 5 \times 10^{-4}$ mm Hg; copper used as standard.
7	591	C	1963	338.2			93.03	The above specimen heat treated 5 days at 620 C.
8	591	C	1963	338.2			87.09	Calculated composition; cast from reactor-grade uranium (99.5% pure) and aluminum (99.99% pure); specimen 3 in. long, 0.500 in. dia.; measured in a vacuum of $< 5 \times 10^{-4}$ mm Hg; copper used as standard.
9	591	C	1963	338.2			87.09	The above specimen heat treated 5 days at 620 C.
10	591	C	1963	338.2			78.57	Calculated composition; cast from reactor-grade uranium (99.5% pure) and aluminum (99.99% pure); 3 in. long, 0.500 in. dia.; measured in a vacuum of $< 5 \times 10^{-4}$ mm Hg; copper used as standard.
11	591	C	1963	338.2			78.57	The above specimen heat treated for 5 days at 620 C.
12	591	C	1963	338.2			67.06	Approx. composition (impurities by spectrographic analysis: $< 0.10$ Fe, $< 0.07$ Si, $< 0.04$ Ca; and $< 0.02$ B); cast from reactor-grade uranium (99.5% pure) and aluminum (99.99% pure); 3 in. long, 0.500 in. dia.; measured in a vacuum of $< 5 \times 10^{-4}$ mm Hg; copper used as standard.
13	591	C	1963	338.2			67.06	The above specimen heat treated for 5 days at 620 C.
14	591	C	1963	338.2			57.51	Cast from reactor-grade uranium (99.5% pure) and aluminum (99.99% pure); specimen 3 in. long, 0.600 in. dia.; measured in a vacuum of $< 5 \times 10^{-4}$ mm Hg; copper used as standard.
15	591	C	1963	338.2			57.51	The above specimen heat treated for 5 days at 620 C.

## DATA TABLE NO. 76 THERMAL CONDUCTIVITY OF [ALUMINUM + URANIUM] ALLOYS

(Al + U &gt; 99.50%; impurity &lt; 0.20% each)

[Temperature, T, K; Thermal Conductivity, k, Watts  $\text{cm}^{-1}\text{K}^{-1}$ ]

T	k	CURVE 1
423.5	1.88	
441.7	1.81	
462.3	1.82	
481.7	1.80	
503.5	1.90	
531.4	1.79	
556.5	1.75	
598.4	1.82	
604.1	1.75	
638.3	1.81	
670.9	1.76	

T	k	CURVE 2
415.0	1.69	
436.8	1.71	
455.0	1.66	
478.0	1.65	
498.7	1.69	
532.6	1.66	
537.4	1.66	
567.8	1.62	
579.9	1.63	
605.3	1.64	
624.7	1.60	
670.8	1.59	

T	k	CURVE 3
425.9	1.49	
452.6	1.55	
485.3	1.50	
496.2	1.46	
503.5	1.48	
537.4	1.50	
581.1	1.49	
624.7	1.48	

T	k	CURVE 4
338.2	2.19	
338.2	2.20	

T	k	CURVE 5
338.2	2.25 <sub>p</sub>	
338.2	2.24	
338.2	2.23	

T	k	CURVE 6
338.2	1.99	

T	k	CURVE 7*
338.2	2.04	

T	k	CURVE 8
338.2	1.82	

T	k	CURVE 9
338.2	1.90	

T	k	CURVE 10
338.2	1.41	

T	k	CURVE 11
338.2	1.71	

T	k	CURVE 12
338.2	1.32	

T	k	CURVE 13*
338.2	1.31	

T	k	CURVE 14
338.2	1.00	

T	k	CURVE 15
338.2	0.866	

\* Not shown on plot

SPECIFICATION TABLE NO. 77 THERMAL CONDUCTIVITY OF [ALUMINUM + ZINC] ALLOYS

(Al + Zn ~ 99.50%; impurity < 0.20% each)

Curve No.	Rel. Method No. Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)	Composition (continued), Specifications and Remarks
						Al Zn	
1	230 L	1925	323.2			90 10	Approx composition; prepared by using aluminum and zinc, each containing < 0.03 total impurities; supplied by Baker; specimen 10 cm long, 1.9 cm dia; electrical conductivity $25.0 \times 10^4 \text{ ohm}^{-1}\text{cm}^{-1}$ at 23 C.
2	230 L	1925	323.2			80 20	Similar to the above specimen except electrical conductivity, $19.6 \times 10^4 \text{ ohm}^{-1}\text{cm}^{-1}$ at 23 C.
3	230 L	1925	323.2			70 30	Similar to the above specimen except electrical conductivity, $13.7 \times 10^4 \text{ ohm}^{-1}\text{cm}^{-1}$ at 23 C.

DATA TABLE NO. 77 THERMAL CONDUCTIVITY OF [ALUMINUM + ZINC] ALLOYS

(Al + Zn ~ 99.50%; impurity < 0.20% each)

[Temperature, T, K; Thermal Conductivity, k, Watt  $\text{cm}^{-1}\text{K}^{-1}$ ]

T	k
<u>CURVE 1<sup>st</sup></u>	
323.2	1.619
<u>CURVE 2<sup>nd</sup></u>	
323.2	1.360
<u>CURVE 3<sup>rd</sup></u>	
323.2	1.126

No graphical presentation

SPECIFICATION TABLE NO. 78 THERMAL CONDUCTIVITY OF (ANTIMONY + ALUMINUM) ALLOYS

(Sb + Al = 99.50%; impurity = 0.20% each)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)		Composition (continued), Specifications and Remarks
							Sb	Al	
1	230	L	1925	325.2			50	50	Approx composition; prepared by fusing aluminum and antimony, each containing < 0.03 impurities; supplied by Baker; 10 cm long, 1.9 cm dia, electrical conductivity $5.44 \times 10^4 \text{ ohm}^{-1}\text{cm}^{-1}$ at 23 C.
2	230	L	1925	325.2			60	40	Similar to the above specimen except electrical conductivity, $1.79 \times 10^4 \text{ ohm}^{-1}\text{cm}^{-1}$ at 23 C.
3	230	L	1925	325.2			70	30	Similar to the above specimen except electrical conductivity, $0.74 \times 10^4 \text{ ohm}^{-1}\text{cm}^{-1}$ at 23 C.
4	230	L	1925	325.2			80	20	Similar to the above specimen except electrical conductivity, $0.014 \times 10^4 \text{ ohm}^{-1}\text{cm}^{-1}$ at 23 C.
5	240	L	1925	325.2			90	10	Similar to the above specimen except electrical conductivity, $0.119 \times 10^4 \text{ ohm}^{-1}\text{cm}^{-1}$ at 23 C.

DATA TABLE NO. 78 THERMAL CONDUCTIVITY OF (ANTIMONY + ALUMINUM) ALLOYS

(Sb + Al = 99.50%; impurity = 0.20% each)

[Temperature, T, K; Thermal Conductivity, k, Watt  $\text{cm}^{-1}\text{K}^{-1}$ ]

T	k	T	k
CURVE 1			
325.2	0.508	325.2	0.219
CURVE 2			
325.2	0.177	325.2	0.243
CURVE 3			
325.2	0.414		

No graphical presentation



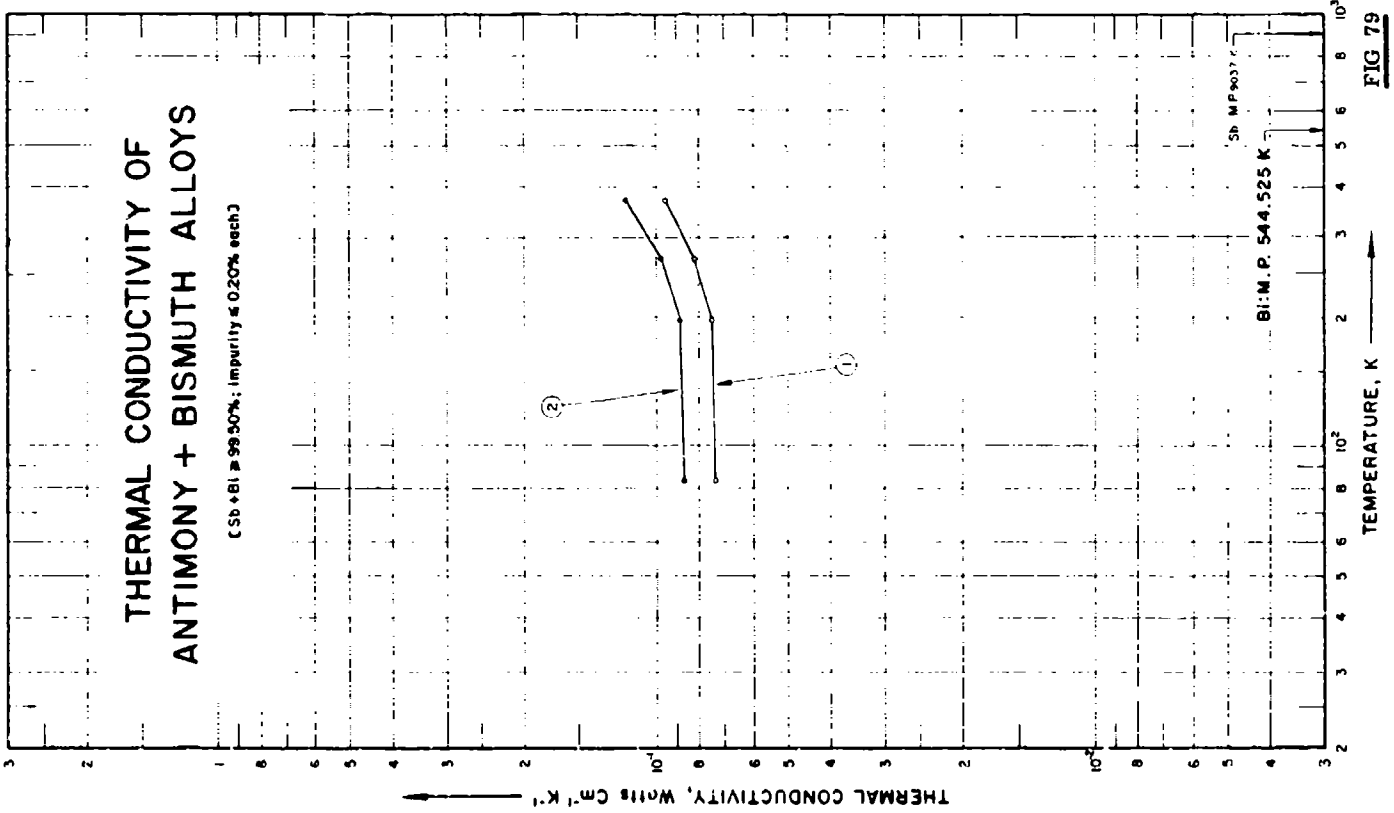


FIG 79

## SPECIFICATION TABLE NO. 79 THERMAL CONDUCTIVITY OF [ANTIMONY + BISMUTH] ALLOYS

(Sb + Bi > 99.50%; Impurity  $\leq$  0.20% each)

[ For Data Reported In Figure and Table No. 79 ]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, $^{\circ}$ F	Reported Error, %	Name and Specimen Designation	Composition (weight percent)		Composition (continued), Specifications and Remarks
							Sb	Bi	
1	49	L	1913	83-373			50	50	Approx. composition.
2	49	L	1913	83-373			70	30	Approx. composition.

## DATA TABLE NO. 79 THERMAL CONDUCTIVITY OF [ANTIMONY + BISMUTH] ALLOYS

(Sb + Bi > 99.50%; Impurity  $\leq$  0.20% each)[Temperature, T, K, Thermal Conductivity, k, Watt cm<sup>-1</sup> K<sup>-1</sup>]

T	k
<u>CURVE 1</u>	
83.2	0.0732
196.2	0.0745
273.2	0.0820
373.2	0.0958
<u>CURVE 2</u>	
83.2	0.0862
196.2	0.0083
273.2	0.0979
373.2	0.118

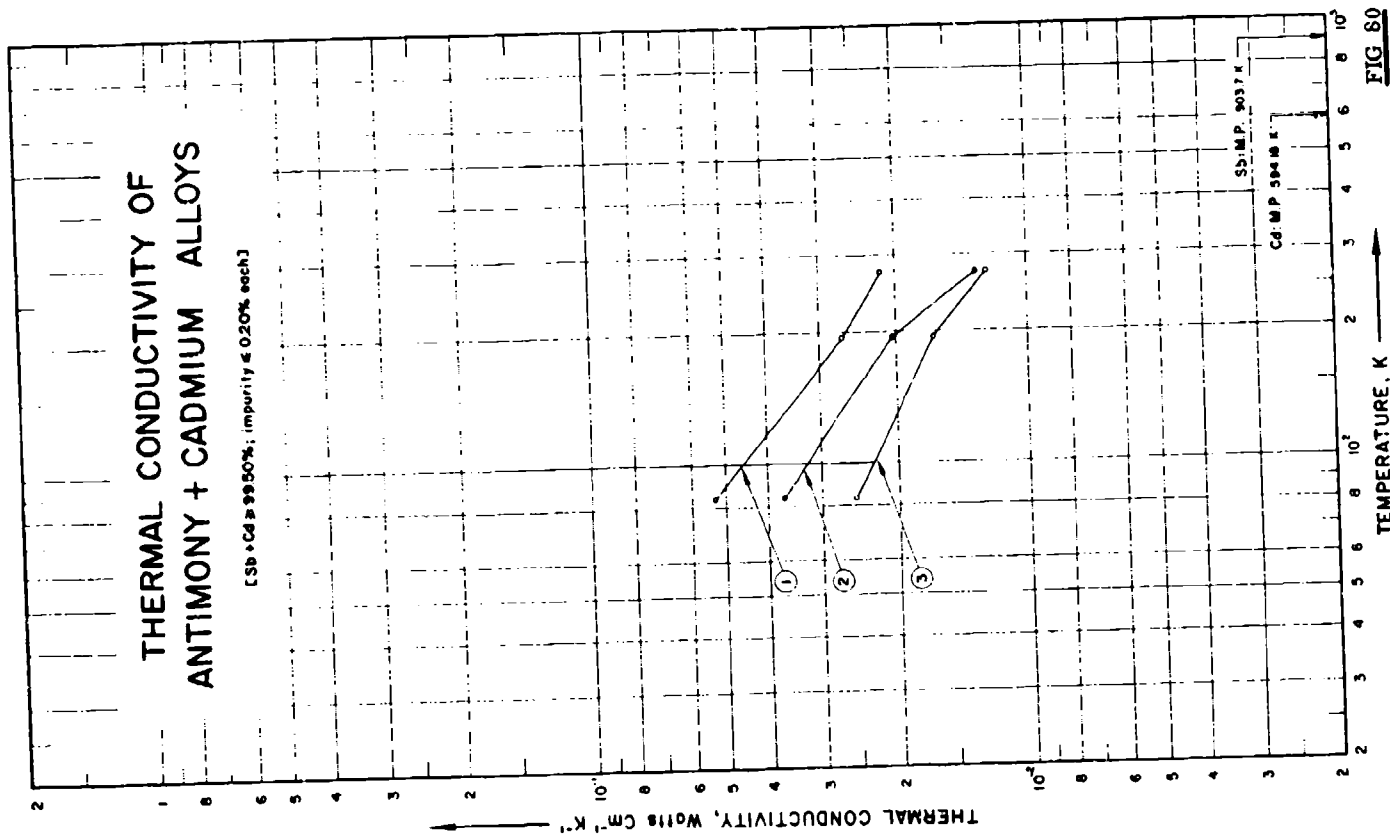


FIG 80

## SPECIFICATION TABLE NO. 80 THERMAL CONDUCTIVITY OF (ANTIMONY + CADMIUM) ALLOYS

(Sb + Cd - 99.50%, impurity  $\pm$  0.20% each)

For Data Reported in Figure and Table No. 80 C

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)		Composition (continued). Specifications and Remarks
							Sb	Cd	
1	35	E	1912	83-273			50	50	Approx. composition: test specimens 2-3 cm in dia; electrical conductivity $0.588 \times 10^4$ , $0.795 \times 10^4$ , and $1.37 \times 10^4$ ohm <sup>-1</sup> cm <sup>-1</sup> at 0, -79, and -190 C respectively.
2	35	E	1912	83-273			51.7	48.3	Calculated composition, electrical conductivity 19.9, 31.45 and 62.6 ohm <sup>-1</sup> cm <sup>-1</sup> at 0, -79, and -190 C respectively.
3	35	E	1912	83-273			66.7	33.3	Calculated composition: electrical conductivity 247, 272, and 202.5 ohm <sup>-1</sup> cm <sup>-1</sup> at 0, -79, and -190 C respectively.

DATA TABLE NO. 80 THERMAL CONDUCTIVITY OF [ANTIMONY + CADMIUM] ALLOYS

(Sb + Cd ± 99.50%, impurity ± 0.20% each)

[ Temperature, T, K. Thermal Conductivity, k, Watt cm<sup>-1</sup> K<sup>-1</sup> ]

T	k
<u>CURVE 1</u>	
83.2	0.0530
194.2	0.0265
273.2	0.0217
<u>CURVE 2</u>	
83.2	0.0369
194.2	0.0205
273.2	0.0132
<u>CURVE 3</u>	
83.2	0.0230
194.2	0.0165
273.2	0.0125

SPECIFICATION TABLE NO. 81 THERMAL CONDUCTIVITY OF [ANTIMONY + COPPER] ALLOYS

(Sb + Cu = 99.50%; impurity &lt; 0.26% each)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)		Composition (continued), Specifications and Remarks
							Sb	Cu	
1	319, 320	P	1966	1073.2	8		50	50	Molten specimen contained in a thin-walled stainless steel cylindrical crucible of dimensions 24 mm dia x 100 mm long; electrical resistivity reported as 138 and 133 $\mu\text{ohm cm}$ at 700 and 800 C, respectively; thermal conductivity values calculated from the thermal diffusivity and the specific heat measurements using the density data from Biernias, A. and Sauerwald, F., Z. anorg. chem., <u>41</u> , 51, 1927.
2	319, 320	P	1966	1073.2	8		60	40	Similar to the above specimen except electrical resistivity, reported as 108, 119, and 132 $\mu\text{ohm cm}$ at 620, 700, and 800 C, respectively.
3	319, 320	P	1966	1073.2	8		76.7	23.3	Similar to the above specimen except electrical resistivity, reported as 75.0, 85.0, and 95.3 $\mu\text{ohm cm}$ at 620, 700, and 800 C, respectively.
4	319, 320	P	1966	1073.2	8		80	20	Similar to the above specimen except electrical resistivity, reported as 76.3, 94.8, and 95.3 $\mu\text{ohm cm}$ at 620, 700, and 800 C, respectively.
5	319, 320	P	1966	1073.2	8		99	10	Similar to the above specimen except electrical resistivity, reported as 80.2, 90.2, and 100 $\mu\text{ohm cm}$ at 620, 700, and 800 C, respectively.

DATA TABLE NO. 81 THERMAL CONDUCTIVITY OF [ANTIMONY + COPPER] ALLOYS

(Sb + Cu = 99.50%; impurity  $\leq$  0.26% each)[Temperature, T, K; Thermal Conductivity, k, Watt  $\text{cm}^{-1}\text{K}^{-1}$ ]

T	k	T	k	T	k	T	k
1073.2	0.236	1073.2	0.253	1073.2	0.288	1073.2	0.301
	<u>CURVE 1<sup>a</sup></u>		<u>CURVE 2<sup>a</sup></u>		<u>CURVE 3<sup>a</sup></u>		<u>CURVE 4<sup>a</sup></u>
							<u>CURVE 5<sup>a</sup></u>

<sup>a</sup> No graphical presentation

SPECIFICATION TABLE NO. 82 THERMAL CONDUCTIVITY OF [ANTIMONY + LEAD] ALLOYS

(Sb + Pb ≥ 99.50%; impurity ≤ 0.20% each)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) Sb	Composition (weight percent) Pb	Composition (continued), Specifications and Remarks
1	230	L	1925	327.2			90	10	Approx composition; prepared by using antimony and lead, each containing < 0.03 impurities; supplied by Baker; specimen 10 cm long, 1.9 cm dia; electrical conductivity, $2.44 \times 10^6 \text{ ohm}^{-1}\text{cm}^{-1}$ at 22 C. Similar to the above specimen except electrical conductivity, $2.29 \times 10^6 \text{ ohm}^{-1}\text{cm}^{-1}$ at 22 C. Similar to the above specimen except electrical conductivity, $2.32 \times 10^6 \text{ ohm}^{-1}\text{cm}^{-1}$ at 22 C. Similar to the above specimen except electrical conductivity, $2.33 \times 10^6 \text{ ohm}^{-1}\text{cm}^{-1}$ at 22 C. Similar to the above specimen except electrical conductivity, $2.46 \times 10^6 \text{ ohm}^{-1}\text{cm}^{-1}$ at 22 C.
2	230	L	1925	327.2			80	20	
3	230	L	1925	327.2			70	30	
4	230	L	1925	327.2			60	40	
5	230	L	1925	327.2			50	50	

DATA TABLE NO. 82 THERMAL CONDUCTIVITY OF [ANTIMONY + LEAD] ALLOYS

(Sb + Pb ≥ 99.50%; impurity ≤ 0.20% each)  
[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup>K<sup>-1</sup>]

T	k	T	k	T	k
<u>CURVE 1*</u>		<u>CURVE 3*</u>		<u>CURVE 5*</u>	
327.2	0.201	327.2	0.197	327.2	0.201
<u>CURVE 2*</u>		<u>CURVE 4*</u>			
327.2	0.188	327.2	0.201		

\* No graphical presentation



## SPECIFICATION TABLE NO. 83 THERMAL CONDUCTIVITY OF [ANTIMONY + TIN] ALLOYS

(Sb + Sn &gt; 99.50%; impurity ≤ 0.20% each)

Curve No.	Ref. Method No.	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) Sb	Composition (weight percent) Sn	Composition (continued), Specifications and Remarks
1	230	L	1925	330.2		90	10	Approx composition; prepared by fusing antimony and tin, each containing < 0.03 total impurity; supplied by Baker; specimen 10 cm long, 1.9 cm dia; electrical conductivity, $1.97 \times 10^4 \text{ ohm}^{-1}\text{cm}^{-1}$ at 22 C.
2	230	L	1925	330.2		80	20	Similar to the above specimen except electrical conductivity, $1.43 \times 10^4 \text{ ohm}^{-1}\text{cm}^{-1}$ at 22 C.
3	230	L	1925	330.2		70	30	Similar to the above specimen except electrical conductivity, $2.29 \times 10^4 \text{ ohm}^{-1}\text{cm}^{-1}$ at 22 C.
4	230	L	1925	330.2		60	40	Similar to the above specimen except electrical conductivity, $2.71 \times 10^4 \text{ ohm}^{-1}\text{cm}^{-1}$ at 22 C.
5	230	L	1925	330.2		50	50	Similar to the above specimen except electrical conductivity, $3.46 \times 10^4 \text{ ohm}^{-1}\text{cm}^{-1}$ at 22 C.

## DATA TABLE NO. 83 THERMAL CONDUCTIVITY OF [ANTIMONY + TIN] ALLOYS

(Sb + Sn &gt; 99.50%; impurity ≤ 0.20% each)

[Temperature, T, K; Thermal Conductivity, k, Watt  $\text{cm}^{-1}\text{K}^{-1}$ ]

T	k	T	k
<u>CURVE 1*</u>			
330.2	0.188	330.2	0.213
<u>CURVE 2*</u>			
330.2	0.176	330.2	0.268
<u>CURVE 3*</u>			
330.2	0.197		

\* No graphical presentation

## SPECIFICATION TABLE NO. 84 THERMAL CONDUCTIVITY OF [BERYLLIUM + ALUMINUM] ALLOYS

(Be + Al)  $\geq$  99.50%; Impurity  $\leq$  0.20% each)

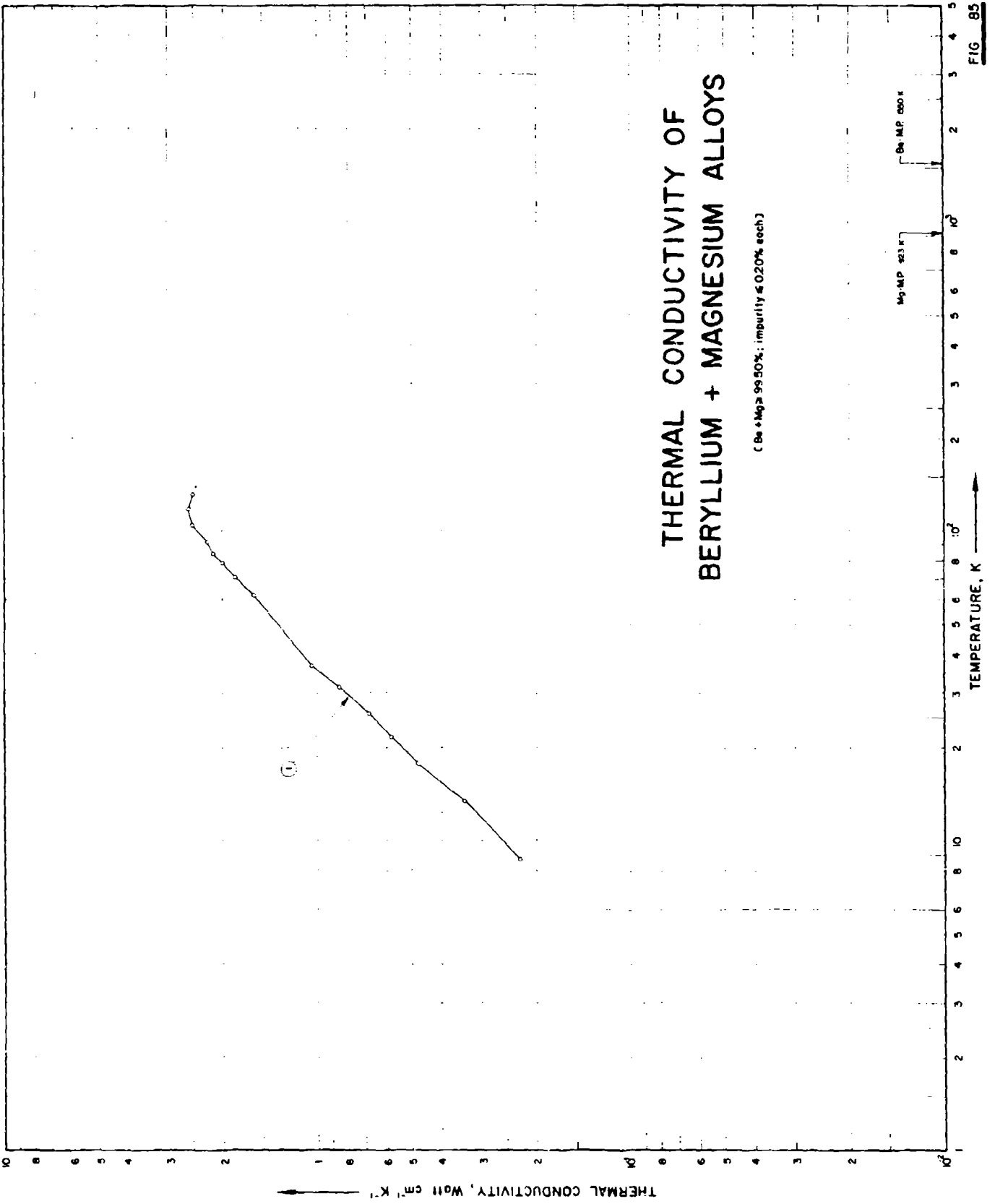
Curve No.	Ref. Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) Be	Al	Composition (continued), Specifications and Remarks
1	235 L	1944	303-338			97	3	Cylindrical specimen 2.542 cm long, 5.070 cm <sup>2</sup> cross sectional area.
2	918	1964	297.1			67	33	Density 2.05 gm cm <sup>-3</sup> .
3	918	1964	297.1			64	36	Density 2.07 gm cm <sup>-3</sup> .
4	916	1964	297.1			57	43	Density 2.14 gm cm <sup>-3</sup> .

## DATA TABLE NO. 84 THERMAL CONDUCTIVITY OF [BERYLLIUM + ALUMINUM] ALLOYS

(Be + Al)  $\geq$  99.50%; impurity  $\leq$  0.20% each)[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup>K<sup>-1</sup>]

T	k	T	k
<u>CURVE 1*</u>			
303.0	1.569		
313.7	1.548		
319.0	1.544		
324.7	1.527		
328.0	1.515		
330.3	1.523		
337.6	1.515		
<u>CURVE 2*</u>			
297.1	1.92		
<u>CURVE 3*</u>			
297.1	1.65		
<u>CURVE 4*</u>			
297.1	1.55		

No graphical presentation



## SPECIFICATION TABLE NO. 85 THERMAL CONDUCTIVITY OF [BERYLLIUM + MAGNESIUM] ALLOYS

(Be + Mg  $\geq$  99.50%, impurity  $\leq$  0.20% each)

[ For Data Reported in Figure and Table No. 85 ]

Curve No.	Ref. No.	Method Use	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) Be	Mg	Composition (continued), Specifications and Remarks
1	355	L	1965	8.7-132			98	2	Composition estimated from spectrographic analysis; trace of Fe; cylindrical specimen 5 mm dia.; sintered; supplied by Brush Co.; electrical resistivity ( $\rho$ minimum) = 1.11 $\mu$ ohm cm and $\rho$ (295 K) = 5.08 $\mu$ ohm cm.

## DATA TABLE NO. 85 THERMAL CONDUCTIVITY OF (BERYLLIUM + MAGNESIUM) ALLOYS

(Be + Mg: 99.50%, impurity: 0.20% each)

[Temperature, T, K; Thermal Conductivity,  $k$ , Watt  $\text{cm}^{-1} \text{K}^{-1}$ ]

T	k	CURVE 1
8.72	0.225	
13.5	0.340	
17.7	0.486	
21.7	0.582	
25.7	0.685	
31.2	0.836	
36.7	1.04	
61.7	1.60	
71.1	1.83	
78.7	1.99	
84.4	2.15	
91.5	2.26	
104.6	2.53	
117.8	2.60	
131.5	2.51	

# THERMAL CONDUCTIVITY OF BISMUTH + ANTIMONY ALLOYS

(Bi + Sb  $\geq$  99.50%; impurity  $\leq$  0.20% each)

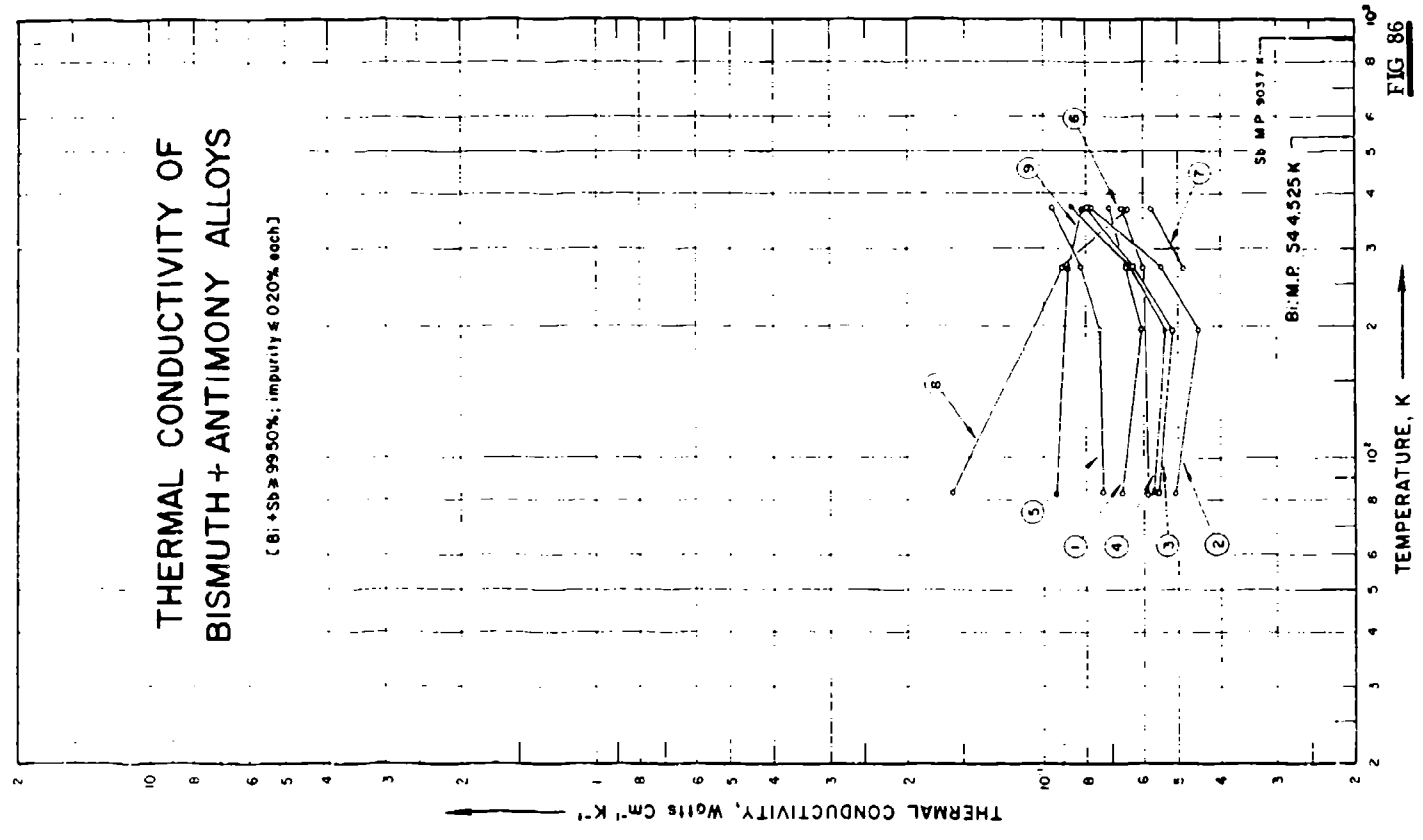


FIG. 86

SPECIFICATION TABLE NO. 86 THERMAL CONDUCTIVITY OF BISMUTH - ANTIMONY ALLOYS

(Bi - Sb 99.99% purity - 0.20% each)

For Data Reported in Figure and Table No. 86

Curve No.	Ret. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) Bi Sb	Composition (continued), Specifications and Remarks	
1	49	L	1913	83-573			91	9	Cast, electrical conductivity 530, 750, 6920, and 5210 ohm <sup>-1</sup> cm <sup>-1</sup> at 150, 277, 0, and 100 C respectively.
2	49	L	1913	83-573			89	11	Cast, electrical conductivity 5300, 6010, 5310, and 4250 ohm <sup>-1</sup> cm <sup>-1</sup> at 150, 277, 0, and 100 C respectively.
3	49	L	1913	83-573			87	13	Cast, electrical conductivity 6570, 6430, 6270, and 4870 ohm <sup>-1</sup> cm <sup>-1</sup> at 150, 277, 0, and 100 C respectively.
4	49	L	1913	83-573			80	20	Cast, electrical conductivity 5410, 6300, 6030, and 5030 ohm <sup>-1</sup> cm <sup>-1</sup> at 150, 277, 0, and 100 C respectively.
5	49	L	1913	83-573			50	50	Cast, electrical conductivity 8050, 5950, 5990, and 4340 ohm <sup>-1</sup> cm <sup>-1</sup> at 150, 277, 0, 100 C respectively.
6	49	L	1913	83-573			95	5	Prepared by pressing chemically pure Bi and <del>95</del> powder for one hr. at 5000 kg/cm <sup>2</sup> . <b>Sb</b>
7	577	L	1913	273-373			93	7	Similar to the above specimen; electrical conductivity 3000, 2800 and 1100 ohm <sup>-1</sup> cm <sup>-1</sup> at 150, 0, and 100 C respectively.
8	577	L	1913	83-573			89	11	Similar to the above specimen; electrical conductivity 1800, 2000, and 2100 ohm <sup>-1</sup> cm <sup>-1</sup> at 150, 0, and 100 C respectively.
9	577	L	1913	83-573			87	13	Similar to the above specimen; electrical conductivity 2400, 2600 and 2400 ohm <sup>-1</sup> cm <sup>-1</sup> at 150, 0, and 100 C respectively.

DATA TABLE NO. 46 THERMAL CONDUCTIVITY OF [BISMUTH + ANTIMONY] ALLOYS

(Bi + Sb  $\geq$  99.50%; Impurity  $\leq$  0.20% each)[Temperature, T, K; Thermal Conductivity, k, Wata  $\text{cm}^{-1}\text{K}^{-1}$ ]

T	k	T	k
<u>CURVE 1</u>			
83.2	0.0669	83.2	0.160
196.2	0.0602	273.2	0.0904
273.2	0.0657	373.2	0.0649
373.2	0.0711	<u>CURVE 9.</u>	
<u>CURVE 2</u>			
83.2	0.0506	83.2	0.0933
196.2	0.0448	273.2	0.0679
273.2	0.0548	373.2	0.0908
373.2	0.0778	<u>CURVE 3</u>	
<u>CURVE 4</u>			
83.2	0.0552	83.2	0.0561
196.2	0.0515	196.2	0.0531
273.2	0.0632	273.2	0.0636
373.2	0.0799	273.2	0.0858
<u>CURVE 5</u>			
83.2	0.0732	83.2	0.0732
196.2	0.0745	196.2	0.0745
273.2	0.0820	273.2	0.0820
373.2	0.0958	373.2	0.0958
<u>CURVE 6</u>			
83.2	0.0582	83.2	0.0582
273.2	0.0598	273.2	0.0598
373.2	0.0669	373.2	0.0669
<u>CURVE 7</u>			
273.2	0.0485	273.2	0.0485
373.2	0.0573	373.2	0.0573



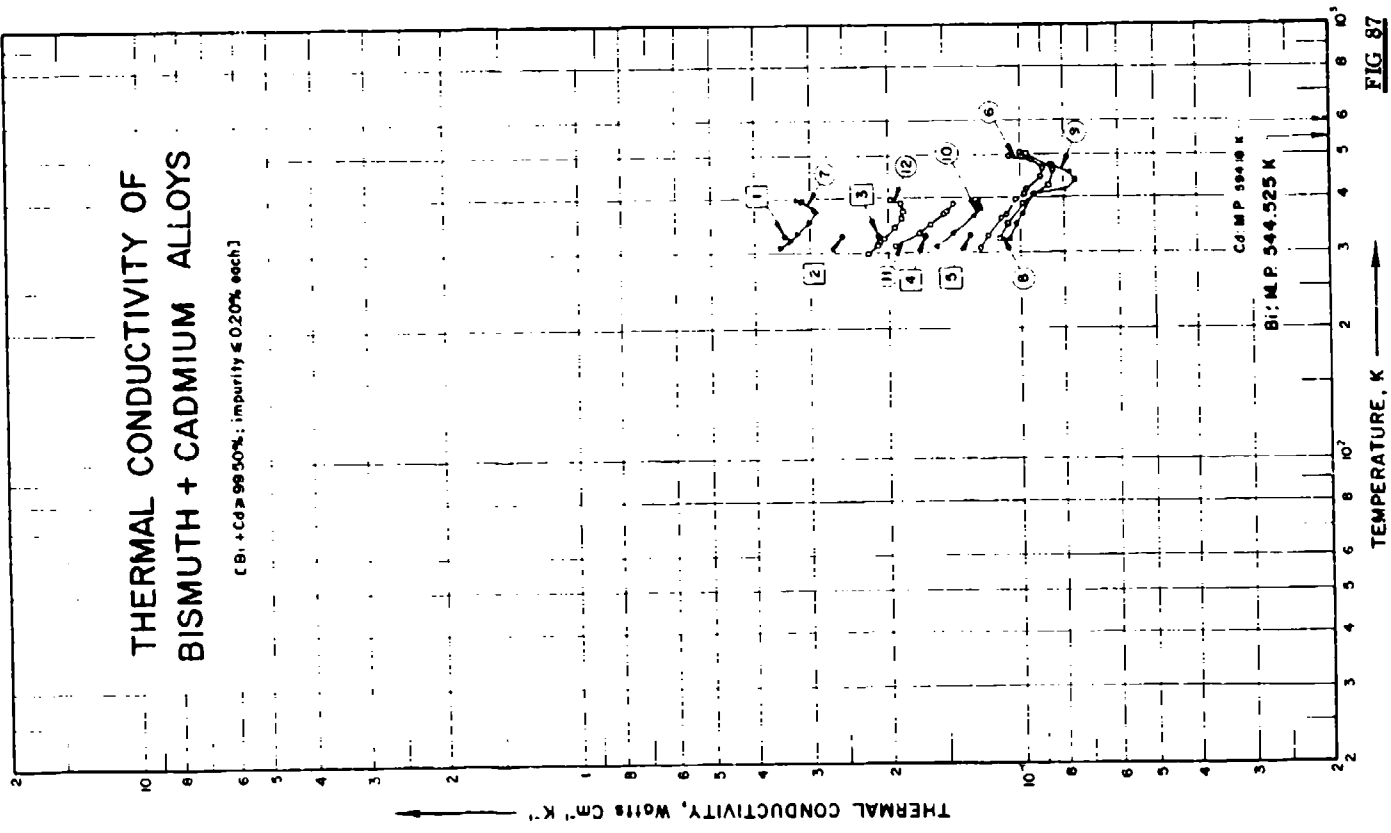


FIG. 87

## SPECIFICATION TABLE NO. 57 THERMAL CONDUCTIVITY OF BISMUTH-CADMIUM ALLOYS

(Bi-Cd - 99.50%, impurity 0.20% each)

(For Data Reported in Figure and Table No. 87)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) Bi	Composition (weight percent) Cd	Composition (continued), Specifications and Remarks
1	230	L	1925	328			50	50	Approximate composition, prepared from Bi (0.63 impurity supplied by Baker) and Cd (no details reported). Specimen cast and machined to 10 cm long, 1.9 cm dia.; electrical conductivity $3.51 \times 10^6 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 22 C.
2	230	L	1925	328			60	40	Similar to the above specimen except electrical conductivity: $2.55 \times 10^6 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 22 C.
3	230	L	1925	328			70	30	Similar to the above specimen except electrical conductivity: $2.13 \times 10^6 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 22 C.
4	230	L	1925	328			80	20	Similar to the above specimen except electrical conductivity: $1.75 \times 10^6 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 22 C.
5	230	L	1925	328			90	10	Similar to the above specimen except electrical conductivity: $1.56 \times 10^6 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 22 C.
6	383		1956	307-338			99.6	0.4	Electrical conductivity ranging from $0.591$ to $0.435 \times 10^6 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 17 to 224.8 C respectively.
7	383		1956	310-357			63.2	36.8	Electrical conductivity ranging from $3.49$ to $2.68 \times 10^6 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 36.4 to 125.9 C respectively.
8	383		1956	324-366			91.1	0.9	Electrical conductivity ranging from $0.841$ to $0.458 \times 10^6 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 50.6 to 232.5 C respectively.
9	383		1956	321-375			93.4	1.6	Electrical conductivity ranging from $1.010$ to $0.511 \times 10^6 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 47.8 to 202.3 C respectively.
10	383		1956	309-394			93.0	10.0	Electrical conductivity ranging from $1.229$ to $0.607 \times 10^6 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 36.1 to 121.0 C respectively.
11	383		1956	310-387			85.3	14.7	Electrical conductivity ranging from $1.61$ to $1.25 \times 10^6 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 47.0 to 114.2 C respectively.
12	383		1956	300-397			80.7	19.3	Electrical conductivity ranging from $1.96$ to $1.54 \times 10^6 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 42.3 to 202.6 C respectively.

DATA TABLE NO. 57 THERMAL CONDUCTIVITY OF [BISMUTH + CADMIUM] ALLOYS

(Bi + Cd) - 95.50% impurity - 0.20% each

[Temperature, T, K. Thermal Conductivity, k, Watts cm<sup>-1</sup>K<sup>-1</sup>]

T	k	T	k	T	k	T	k
CURVE 1							
328.0	0.339	323.8	0.1110	299.8	0.220		
		349.4	0.1060	312.4	0.210		
CURVE 2							
		351.4	0.1060	318.9	0.207		
		387.7	0.0987	324.9	0.203		
CURVE 3							
328.0	0.251	425.5	0.0865	342.0	0.192		
		465.6	0.0845	359.5	0.185		
		474.0	0.0849	373.8	0.183		
CURVE 4							
328.0	0.209	501.9	0.0967	388.2	0.186		
		508.1	0.1010	397.1	0.197		
CURVE 5							
328.0	0.163	321.0	0.106				
		349.2	0.102				
		368.5	0.0983				
CURVE 6							
328.0	0.136	372.2	0.0979				
		408.3	0.0957				
		418.0	0.0795				
CURVE 7							
306.9	0.1240	437.0	0.0749				
327.9	0.1180	456.5	0.0774				
339.1	0.1100	472.6	0.0845				
365.0	0.1070	478.5	0.0855				
395.9	0.1020	CURVE 10					
407.0	0.0975	309.3	0.1518				
419.1	0.0967	332.0	0.1418				
446.1	0.0900	363.5	0.1276				
468.8	0.0891	379.7	0.1234				
485.1	0.0916	386.2	0.1237				
498.0	0.1060	394.2	0.1247				
CURVE 8							
309.6	0.0346	310.2	0.192				
320.3	0.331	332.2	0.168				
331.0	0.319	349.0	0.159				
351.0	0.300	366.7	0.149				
371.8	0.289	371.9	0.147				
377.9	0.292	387.5	0.142				
394.6	0.311						
397.1	0.320						

# THERMAL CONDUCTIVITY OF BISMUTH + LEAD ALLOYS

(B) + Pb ≥ 59.50%; impurity ≤ 0.20% each

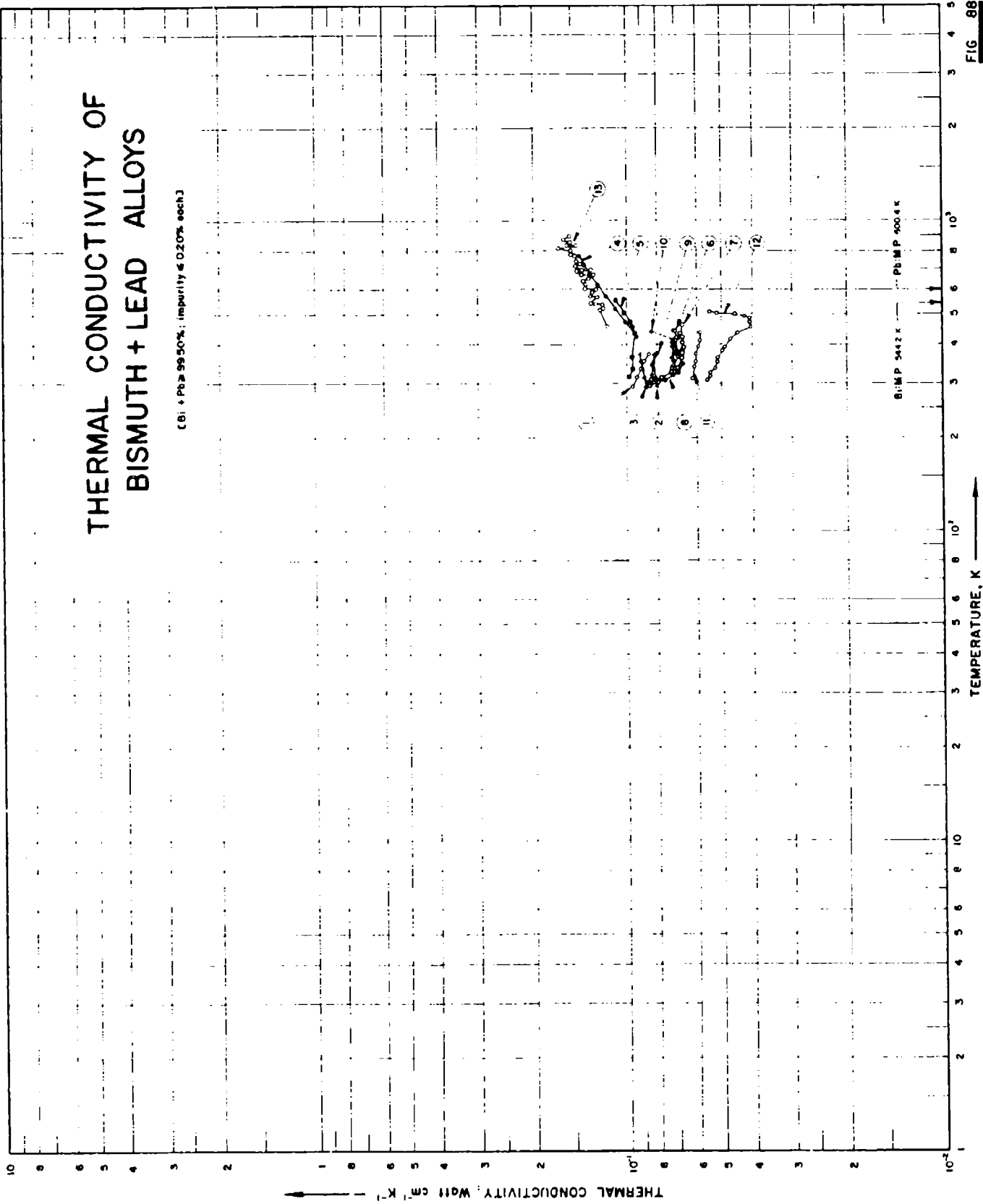


FIG. 88

## SPECIFICATION TABLE NO. 88 THERMAL CONDUCTIVITY OF BISMUTH-LEAD ALLOYS

(Bi + Pb = 99.50%; impurity = 0.20% each)

For Data Reported in Figure and Table No. 88

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) Bi	Composition (weight percent) Pb	Composition (continued), Specifications and Remarks
1	126, 324	F	1955	293-373	< 7	6 B	99.15	0.85	Specimen composed of a few large crystals, prepared by fusing granular Pb and Bi in a N <sub>2</sub> atmosphere, casting in Pyrex tubing, and slowly cooling; electrical conductivity ranging from 4370 to 4450 ohm <sup>-1</sup> cm <sup>-1</sup> at 293.2 to 373.2 K respectively.
2	126, 324	F	1955	293-373	< 8	8 B	98.5	1.5	Similar to the above specimen except electrical conductivity ranging from 3420 to 3666 ohm <sup>-1</sup> cm <sup>-1</sup> at 293.2 to 373.2 K respectively.
3	126, 324	F	1955	293-373	< 6	9	97.0	3.0	Similar to the above specimen except electrical conductivity ranging from 2830 to 3040 ohm <sup>-1</sup> cm <sup>-1</sup> at 293.2 to 373.2 K respectively.
4	113	C	1957	423-773		Bi-Pb eutectic	55.5	44.5	Eutectic composition, prepared from Pb of 99.99% purity and Bi of high purity; molten state contained in a cavity 3.5 in. deep and 0.97 in. in dia.; stainless steel used as standard and as container; electrical resistivity ranging from 112.5 to 126.6 μ ohm cm at 150 to 500 C respectively.
5	19	L	1923	315-558		Bi-Pb eutectic	54.0	46.0	Eutectic alloy; specimen 1.5 cm dia., 12 cm long; melting point 403 K
6	249	E	1956	295-443	± 3		99.0	1.0	
7	248	E	1956	308-466	± 3		98.5	1.5	Annealed for 48 hr. at 120 C.
8	248	E	1956	308-446	± 3		98.0	2.0	Annealed for 72 hr. at 120 C.
9	248	E	1956	308-376	± 3		66.22	33.67	
10	248	E	1956	319-441	± 3		54.56	45.44	
11	248	E	1956	311-440	± 3		95.18	4.82	
12	383		1956	307-511			98.0	2.0	Electrical conductivity ranging from 0.325 to 0.421 x 10 <sup>4</sup> ohm <sup>-1</sup> cm <sup>-1</sup> at 307.2 to 511.2 K respectively.
13	592	L	1959	464-902		Bi-Pb eutectic	55.5	44.5	Molten state contained in a cavity 14.0 to 13.5 mm dia. and 140 to 210 mm long; measured in a vacuum of 1.4 x 10 <sup>-4</sup> mm Hg; melting point 123.5 C; density ranging from 10.370 to 9.876 g cm <sup>-3</sup> at 130 to 700 C respectively.

DATA TABLE NO. 18 THERMAL CONDUCTIVITY OF BISMUTH-LEAD ALLOYS

( $\rho = 10^{-10}$  ohm-cm, impurity = 0.20% each)  
 Temperature, T, K; Thermal Conductivity,  $\kappa$ , Watt/cm<sup>2</sup>K<sup>-1</sup>

CURVE 1		CURVE 6		CURVE 10		CURVE 13 (room)	
T	$\kappa$	T	$\kappa$	T	$\kappa$	T	$\kappa$
295.2	0.0966	295.2	0.0871	319.2	0.0711	325.2	0.117
313.2	0.093	296.2	0.0837	337.2	0.0711	343.2	0.119
331.2	0.089	303.2	0.0775	352.2	0.0711	346.2	0.126
349.2	0.085	329.2	0.0724	363.2	0.0720	347.2	0.130
367.2	0.083	335.2	0.0707	375.2	0.0711	354.2	0.124
		342.2	0.0711	386.2	0.0715	372.2	0.131
		363.2	0.0690	403.2	0.0720	383.2	0.128
		371.2	0.0680	415.2	0.0715	397.2	0.127
		374.2	0.0693	431.2	0.0827	404.2	0.123
		379.2	0.0699	443.2	0.0827	407.2	0.127
		383.2	0.0703			417.2	0.136
						431.2	0.126
						435.2	0.133
						443.2	0.137
						446.2	0.139
						454.2	0.144
						460.2	0.147
						463.2	0.149
						466.2	0.151
						472.2	0.153
						475.2	0.155
						478.2	0.157
						483.2	0.159
						486.2	0.161
						491.2	0.163
						494.2	0.165
						497.2	0.167
						500.2	0.169
						503.2	0.171
						506.2	0.173
						509.2	0.175
						512.2	0.177
						515.2	0.179
						518.2	0.181
						521.2	0.183
						524.2	0.185
						527.2	0.187
						530.2	0.189
						533.2	0.191
						536.2	0.193
						539.2	0.195
						542.2	0.197
						545.2	0.199
						548.2	0.201
						551.2	0.203
						554.2	0.205
						557.2	0.207
						560.2	0.209
						563.2	0.211
						566.2	0.213
						569.2	0.215
						572.2	0.217
						575.2	0.219
						578.2	0.221
						581.2	0.223
						584.2	0.225
						587.2	0.227
						590.2	0.229
						593.2	0.231
						596.2	0.233
						599.2	0.235
						602.2	0.237
						605.2	0.239
						608.2	0.241
						611.2	0.243
						614.2	0.245
						617.2	0.247
						620.2	0.249
						623.2	0.251
						626.2	0.253
						629.2	0.255
						632.2	0.257
						635.2	0.259
						638.2	0.261
						641.2	0.263
						644.2	0.265
						647.2	0.267
						650.2	0.269
						653.2	0.271
						656.2	0.273
						659.2	0.275
						662.2	0.277
						665.2	0.279
						668.2	0.281
						671.2	0.283
						674.2	0.285
						677.2	0.287
						680.2	0.289
						683.2	0.291
						686.2	0.293
						689.2	0.295
						692.2	0.297
						695.2	0.299
						698.2	0.301
						701.2	0.303
						704.2	0.305
						707.2	0.307
						710.2	0.309
						713.2	0.311
						716.2	0.313
						719.2	0.315
						722.2	0.317
						725.2	0.319
						728.2	0.321
						731.2	0.323
						734.2	0.325
						737.2	0.327
						740.2	0.329
						743.2	0.331
						746.2	0.333
						749.2	0.335
						752.2	0.337
						755.2	0.339
						758.2	0.341
						761.2	0.343
						764.2	0.345
						767.2	0.347
						770.2	0.349
						773.2	0.351
						776.2	0.353
						779.2	0.355
						782.2	0.357
						785.2	0.359
						788.2	0.361
						791.2	0.363
						794.2	0.365
						797.2	0.367
						800.2	0.369
						803.2	0.371
						806.2	0.373
						809.2	0.375
						812.2	0.377
						815.2	0.379
						818.2	0.381
						821.2	0.383
						824.2	0.385
						827.2	0.387
						830.2	0.389
						833.2	0.391
						836.2	0.393
						839.2	0.395
						842.2	0.397
						845.2	0.399
						848.2	0.401
						851.2	0.403
						854.2	0.405
						857.2	0.407
						860.2	0.409
						863.2	0.411
						866.2	0.413
						869.2	0.415
						872.2	0.417
						875.2	0.419
						878.2	0.421
						881.2	0.423
						884.2	0.425
						887.2	0.427
						890.2	0.429
						893.2	0.431
						896.2	0.433
						899.2	0.435
						902.2	0.437
						905.2	0.439
						908.2	0.441
						911.2	0.443
						914.2	0.445
						917.2	0.447
						920.2	0.449
						923.2	0.451
						926.2	0.453
						929.2	0.455
						932.2	0.457
						935.2	0.459
						938.2	0.461
						941.2	0.463
						944.2	0.465
						947.2	0.467
						950.2	0.469
						953.2	0.471
						956.2	0.473
						959.2	0.475
						962.2	0.477
						965.2	0.479
						968.2	0.481
						971.2	0.483
						974.2	0.485
						977.2	0.487
						980.2	0.489
						983.2	0.491
						986.2	0.493
						989.2	0.495
						992.2	0.497
						995.2	0.499
						998.2	0.501
						1001.2	0.503

Not shown on plot

# THERMAL CONDUCTIVITY OF BISMUTH + TIN ALLOYS

(Bi + Sn is 99.90%; impurity  $\leq$  0.20% each)

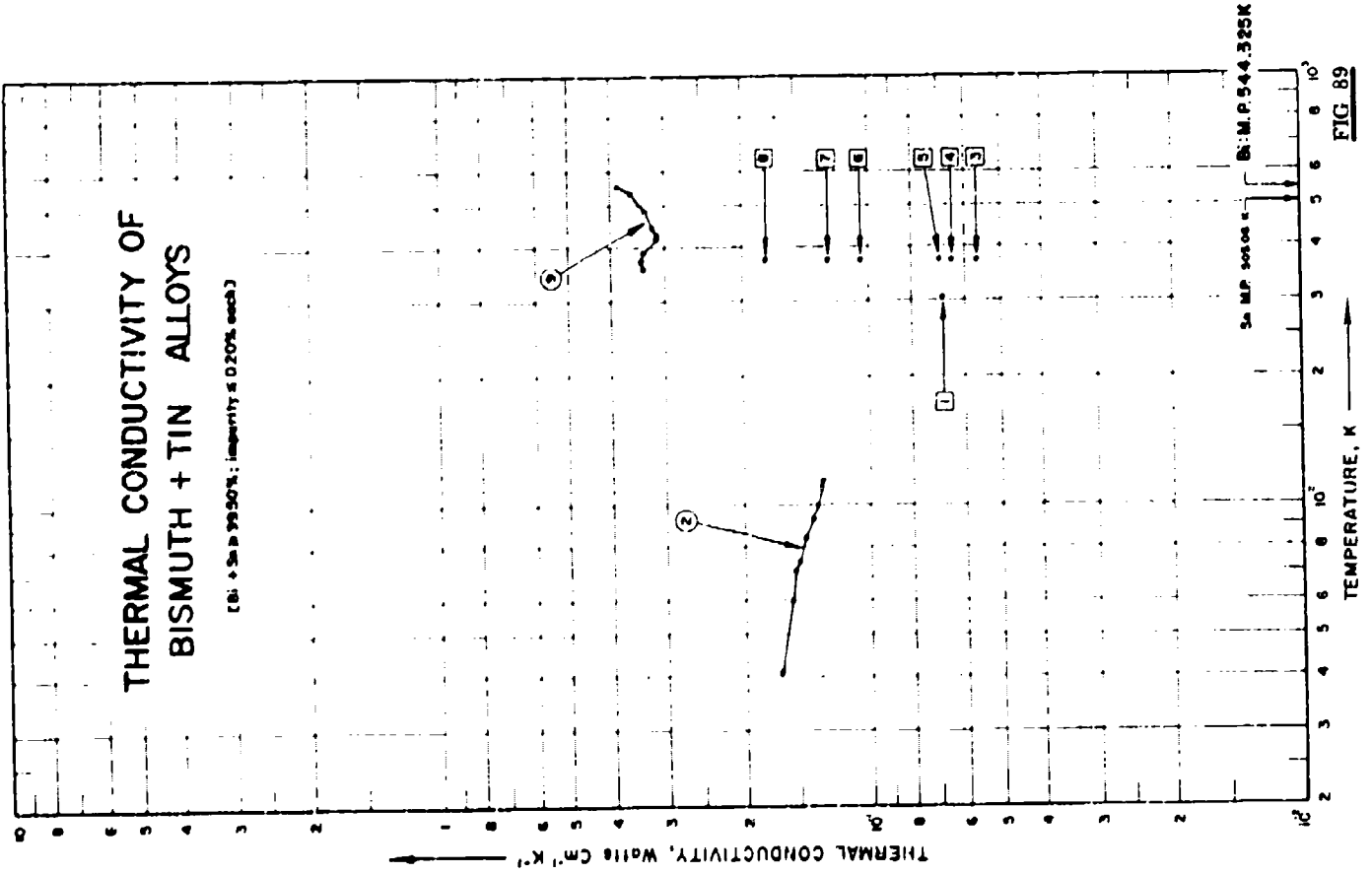


FIG 89

Bi: M.P. 544.385K

SPECIFICATION TABLE NO. 89 THERMAL CONDUCTIVITY OF (BISMUTH - TIN) ALLOYS  
(Bi - Sn 99.50% impurity - 0.20% each)

(For Data Reported in Figure and Table No. 89)

Curve No.	Ref. No. Used	Method	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) Bi	Sn	Composition (continued), Specifications and Remarks
1	357	L	1943	302.2	±1.0	Hutchins' alloy	95	5	No details reported.
2	460		1957	41-114			70	30	No details reported.
3	460		1957	373.2			99.14	0.86	No details reported.
4	460		1957	373.2			97.0	3.0	No details reported.
5	460		1957	373.2			93	5	No details reported.
6	460		1957	373.2			85	15	No details reported.
7	460		1957	373.2			71	29	No details reported.
8	460		1957	373.2			59	41	No details reported.
9	514	L	1962	358-553	±5.0		50	50	Specimen in liquid state at temperatures above 145 C.



## DATA TABLE NO. 89 THERMAL CONDUCTIVITY OF [BISMUTH + TIN] ALLOYS

(Bi + Sn  $\geq$  99.50%, impurity  $\leq$  0.20% each)[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup>K<sup>-1</sup>]

T	k	$\bar{T}$	k
<u>CURVE 1</u>			
302.2	0.0685	388.2	0.339
		418.2	0.314
<u>CURVE 2</u>			
41.0	0.165	428.2	0.314
60.0	0.155	443.2	0.322
70.5	0.152	483.2	0.335
74.0	0.149	533.2	0.360
84.0	0.144	553.2	0.385
92.5	0.138		
100.0	0.135		
110.0	0.132		
114.0	0.131		
<u>CURVE 3</u>			
373.2	0.0565		
<u>CURVE 4</u>			
373.2	0.0649		
<u>CURVE 5</u>			
373.2	0.0690		
<u>CURVE 6</u>			
373.2	0.105		
<u>CURVE 7</u>			
373.2	0.126		
<u>CURVE 8</u>			
373.2	0.176		
<u>CURVE 9</u>			
358.2	0.339		
373.2	0.343		

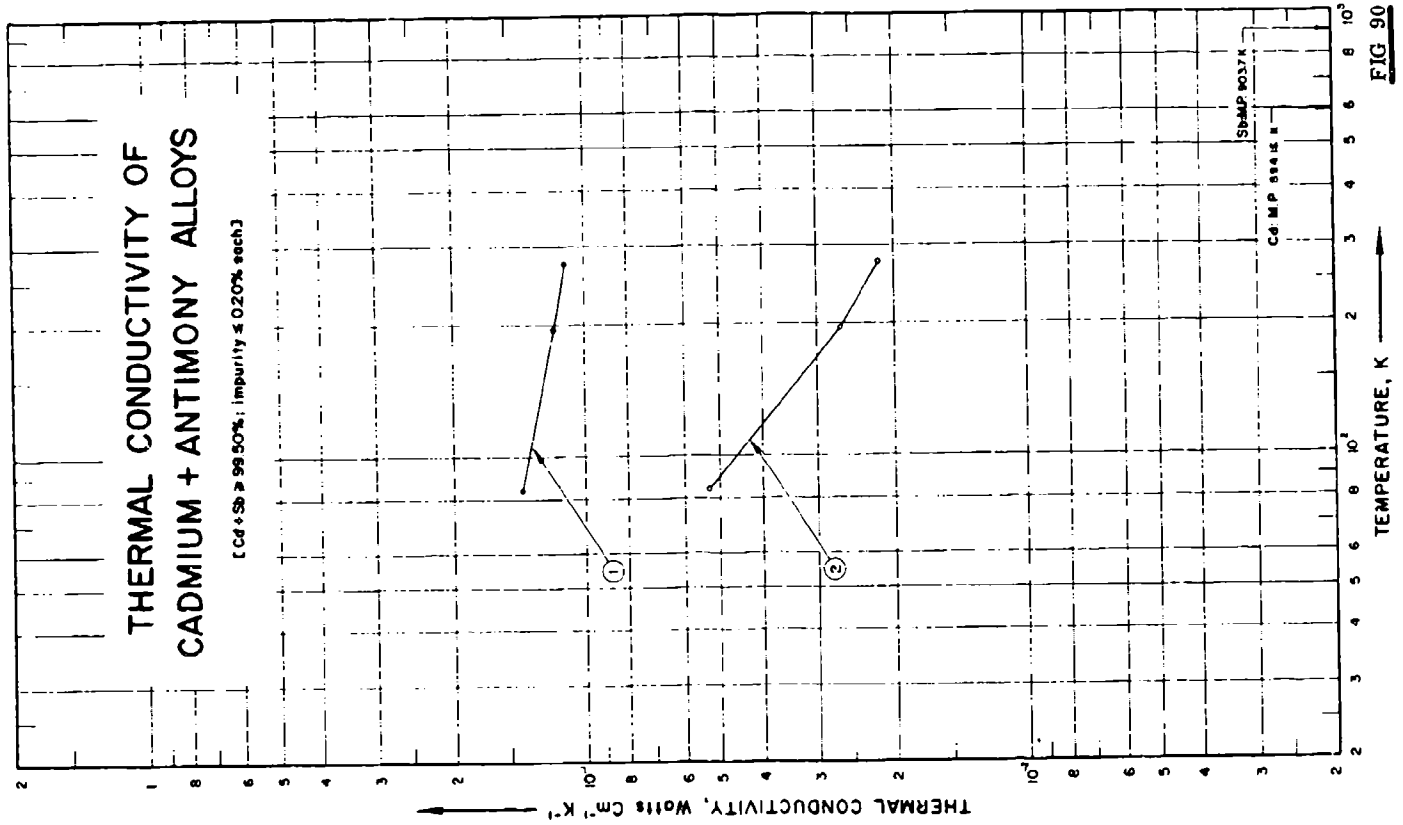


FIG 90

## SPECIFICATION TABLE NO. 90 THERMAL CONDUCTIVITY OF [CADMIUM + ANTIMONY] ALLOYS

(Cd + Sb = 99.50%; impurity = 0.20% each)

[ For Data Reported in Figure and Table No. 90 ]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) Cd	Sb	Composition (continued), Specifications and Remarks
1	35	F	1912	83-273			66.7	33.3	Calculated composition; test specimen 2-3 cm in dia.; electrical conductivity $2.69 \times 10^3$ , $3.425 \times 10^3$ , and $6.78 \times 10^3$ ohm <sup>-1</sup> cm <sup>-1</sup> at 0, -79, and -190 C respectively.
2	35	E	1912	83-273			50	50	Approx. composition, electrical conductivity $0.588 \times 10^3$ , $0.795 \times 10^3$ , and $1.37 \times 10^3$ ohm <sup>-1</sup> cm <sup>-1</sup> at 0, -79, and -190 C respectively.

## DATA TABLE NO. 90 THERMAL CONDUCTIVITY OF [CADMIUM + ANTIMONY] ALLOYS

(Cd + Sb  $\geq$  99.50%, impurity  $<$  0.20% each)[Temperature, T, K. Thermal Conductivity, k, Watt cm<sup>-1</sup> K<sup>-1</sup>]

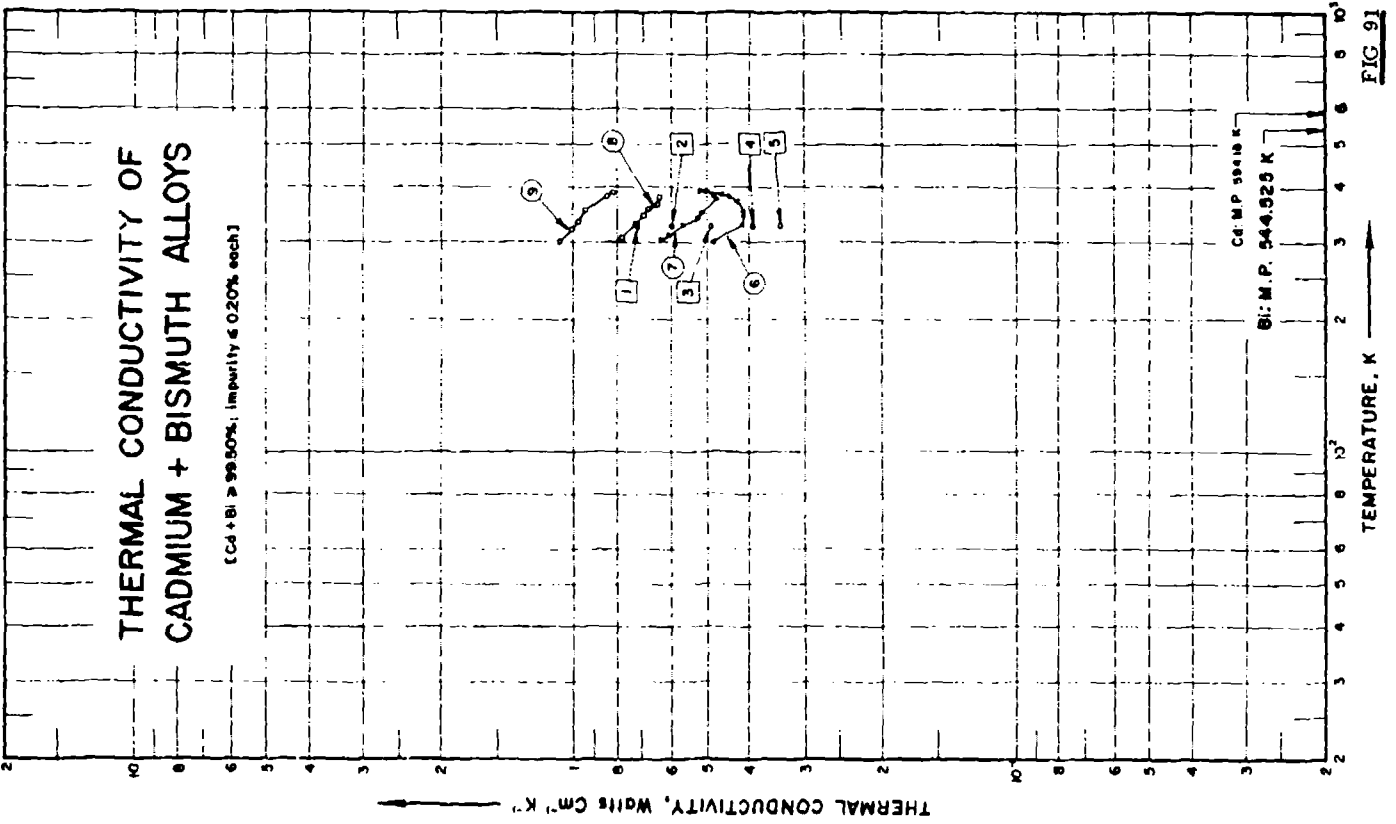
T k

CURVE 1

83.2	0.140
194.2	0.116
273.2	0.112

CURVE 2

83.2	0.0530
194.2	0.0265
273.2	0.0217



## SPECIFICATION TABLE NO. 91 THERMAL CONDUCTIVITY OF (CADMIUM + BISMUTH) ALLOYS

(Cd + Bi = 99.50%; impurity : 0.20%, each)

[ For Data Reported in Figure and Table No. 91 ]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)		Composition (continued), Specifications and Remarks
							Cd	Bi	
1	230	L	1925	328.2			90.0	10.0	Approx. composition; Bi metal contained 0.03 of total impurity, supplied by Baker; specimen 1.9 cm in diameter and 10 cm long, electrical conductivity $8.92 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 22 C.
2	230	L	1925	328.2			80.0	20.0	Similar to above specimen except electrical conductivity, $6.47 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 22 C.
3	230	L	1925	328.2			70.0	30.0	Similar to above specimen except electrical conductivity, $5.24 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 22 C.
4	230	L	1925	328.2			60.0	40.0	Similar to above specimen except electrical conductivity, $3.86 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 22 C.
5	230	L	1925	328.2			50.0	50.0	Similar to above specimen except electrical conductivity, $3.51 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 22 C.
6	383		1956	303-394			50.6	49.4	Approx. composition; electrical conductivity 4.84, 4.42, 4.255, 4.2, 4.04, 3.94, 3.905 and $3.82 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 29.3, 57.4, 71.7, 79.9, 97.9, 107.9, 112.2 and 120.7 C respectively.
7	383		1956	304-396			67.7	32.3	Approx. composition; electrical conductivity 5.8, 5.68, 5.47, 5.24, 5.15, 4.85, 4.77 and $4.67 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 30.7, 39.2, 55.7, 67.4, 78.1, 103.1, 113.5 and 122.9 C respectively.
8	383		1956	307-383			89.8	10.2	Approx. composition; electrical conductivity 8.45, 7.85, 7.52, 7.26, 6.95, 6.83 and $6.81 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 33.8, 57.5, 72.4, 65.1, 94.2, 106.6, 109.6 C respectively.
9	383		1956	302-388			90.0	9.8	0.2 Pb; electrical conductivity 10.8, 10.11, 9.74, 8.97, 8.44 and $8.17 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 28.8, 47.7, 61.4, 83.3, 105.8 and 114.9 C respectively.

DATA TABLE NO. 91 THERMAL CONDUCTIVITY OF [CADMIUM + BISMUTH] ALLOYS

(Cd + Bi : 99.50%; impurity : 0.20% each)

(Temperature, T, K. Thermal Conductivity, k, Watt cm<sup>-1</sup> K<sup>-1</sup>)

T	k	T	k
<u>CURVE 1</u>			
328.2	0.729	307.0	0.778
<u>CURVE 2</u>			
		310.7	0.7259
		345.6	0.699
		358.3	0.678
328.2	0.598	367.4	0.649
<u>CURVE 3</u>			
		379.8	0.640*
		382.8	0.636
328.2	0.490	<u>CURVE 5</u>	
<u>CURVE 4</u>			
		302.0	1.08
		320.9	1.01
328.2	0.393	334.6	0.975
		356.5	0.941
		379.0	0.845
328.2	0.339	388.1	0.820
<u>CURVE 6</u>			
302.5	0.485		
330.6	0.416		
344.9	0.413		
353.1	0.412		
371.0	0.428		
381.1	0.449		
365.4	0.464		
393.9	0.504		
<u>CURVE 7</u>			
303.9	0.632		
312.4	0.607		
328.9	0.565		
340.6	0.527		
351.3	0.515		
375.3	0.477		
386.7	0.485		
396.1	0.519		

\* Not shown on plot

## SPECIFICATION TABLE NO. 92 THERMAL CONDUCTIVITY OF [CADMIUM + THALLIUM] ALLOYS

(Cd + Tl) ≥ 99.50%; impurity ≤ 0.20% each)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) Cd	Composition (weight percent) Tl	Composition (continued), Specifications and Remarks
1	230	L	1925	336.2			50.0	50.0	Specimen 5 or 6 cm long with a cross-section ~ 0.3 cm <sup>2</sup> ; electrical conductivity $0.877 \times 10^6 \text{ ohm}^{-1}\text{cm}^{-1}$ at 23 C.
2	230	L	1925	336.2			60.0	40.0	Similar to above specimen except electrical conductivity, $0.926 \times 10^6 \text{ ohm}^{-1}\text{cm}^{-1}$ at 23 C.
3	230	L	1925	336.2			70.0	30.0	Similar to above specimen except electrical conductivity, $1.02 \times 10^6 \text{ ohm}^{-1}\text{cm}^{-1}$ at 23 C.
4	230	L	1925	336.2			80.0	20.0	Similar to above specimen except electrical conductivity, $1.11 \times 10^6 \text{ ohm}^{-1}\text{cm}^{-1}$ at 23 C.
5	230	L	1925	336.2			90.0	10.0	Similar to above specimen except electrical conductivity, $1.22 \times 10^6 \text{ ohm}^{-1}\text{cm}^{-1}$ at 23 C.

## DATA TABLE NO. 92 THERMAL CONDUCTIVITY OF [CADMIUM + THALLIUM] ALLOYS

(Cd + Tl) ≥ 99.50%; impurity ≤ 0.20% each)

[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup>K<sup>-1</sup>]

T	k	T	k
<u>CURVE 1*</u>		<u>CURVE 4*</u>	
336.2	0.661	336.2	0.799
<u>CURVE 2*</u>		<u>CURVE 5*</u>	
336.2	0.703	336.2	0.866
<u>CURVE 3*</u>			
336.2	0.753		

\* No graphical presentation



# THERMAL CONDUCTIVITY OF CADMIUM + TIN ALLOYS

[Cd + Sn > 99.50%; impurity < 0.20% each]

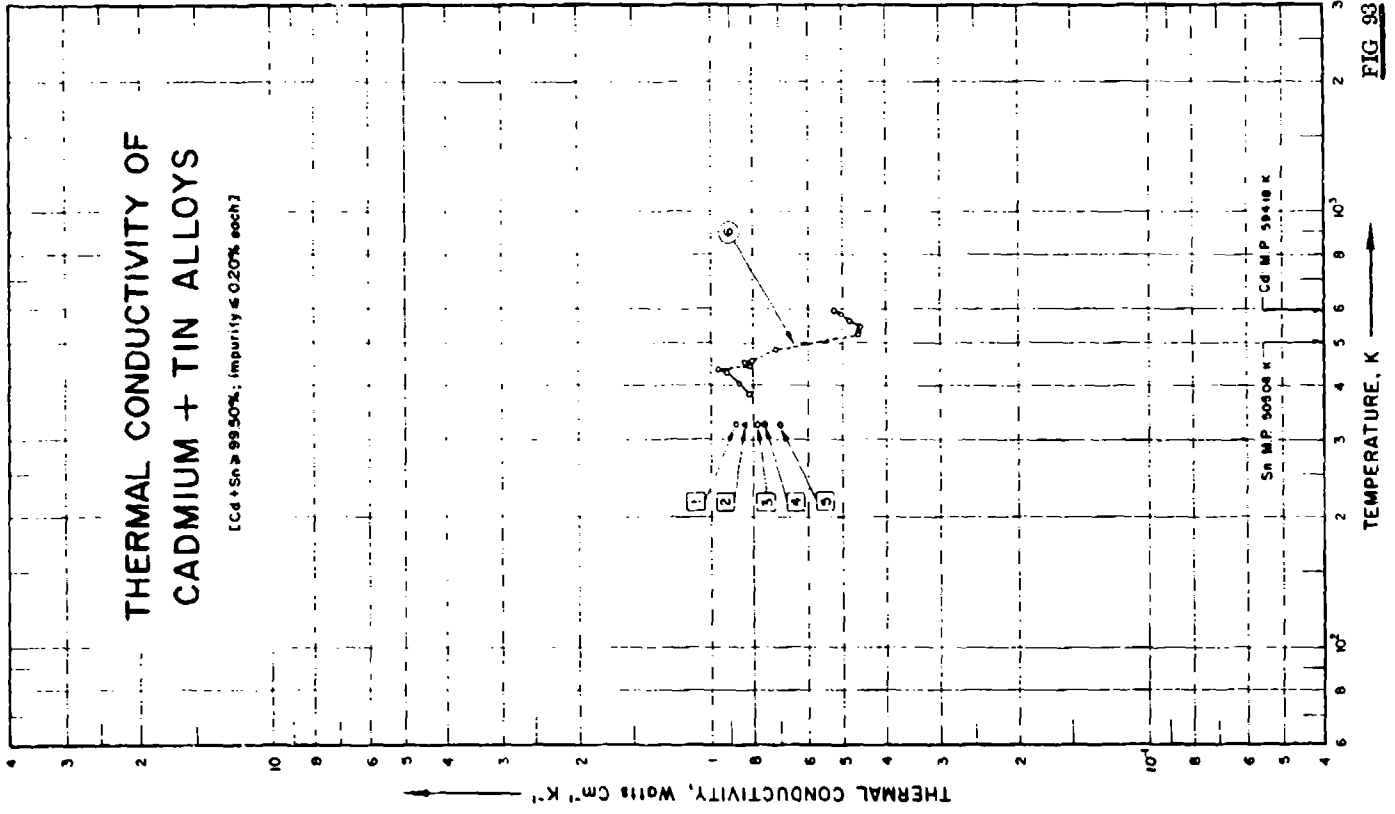


FIG 93

## SPECIFICATION TABLE NO. 93 THERMAL CONDUCTIVITY OF (CADMIUM - TIN) ALLOYS

(Cd - Sn &lt; 99.50%; impurity &lt; 0.30% each)

For Data Reported in Figure and Table No. 93.2

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)		Composition (continued), Specifications and Remarks
							Cd	Sn	
1	230	L	1925	326.2			90.0	10.0	Approx. composition; < 0.03 of total impurity in Baker's analyzed tin; specimen 1.9 cm in diameter and 10 cm long; electrical conductivity $12.7 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 22 C.
2	230	L	1925	326.2			80.0	20.0	Similar to above specimen except electrical conductivity, $12.3 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 22 C.
3	230	L	1925	326.2			70.0	30.0	Similar to above specimen except electrical conductivity, $11.4 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 22 C.
4	230	L	1925	326.2			60.0	40.0	Similar to above specimen except electrical conductivity, $10.7 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 22 C.
5	230	L	1925	326.2			50.0	50.0	Similar to above specimen except electrical conductivity, $9.98 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 22 C.
6	514	L	1962	383-591	< 5.0		70.0	29.7	Specimen in liquid state at temperatures > 548 K.

## DATA TABLE NO. 93 THERMAL CONDUCTIVITY OF [CADMIUM + TIN] ALLOYS

(Cd + Sn = 99.50%; impurity &lt; 0.20% each)

[Temperature, T, K. Thermal Conductivity, k, Watt cm<sup>-1</sup>K<sup>-1</sup>]

T	k
<u>CURVE 1</u>	
326.2	0.875
<u>CURVE 2</u>	
326.2	0.837
<u>CURVE 3</u>	
326.2	0.782
<u>CURVE 4</u>	
326.2	0.753
<u>CURVE 5</u>	
326.2	0.699
<u>CURVE 6</u>	
383.2	0.816
403.2	0.856
428.2	0.912
433.2	0.962
443.2	0.816
453.2	0.837
457.2	0.803
483.2	0.711
523.2	0.460
533.2	0.460
548.2	0.455
563.2	0.481
583.2	0.502
591.2	0.523

SPECIFICATION TABLE NO. 94 THERMAL CONDUCTIVITY OF [CADMIUM + ZINC] ALLOYS  
(Cd + Zn ~ 99.50%; impurity  $\leq 0.20\%$ , each)

Curve No.	Ref. Method No. Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Compositor (weight percent) Cd	Zn	Composition (continued), Specifications and Remarks
1	230 L	1925	326.2			90.0	10.0	Approx composition: $< 0.07$ of total impurity in Baker's analyzed Zn; specimen 1.5 cm in dia and 10 cm long; electrical conductivity $13.91 \times 10^6 \text{ ohm}^{-1}\text{cm}^{-1}$ at 22 C.
2	230 L	1925	326.2			80.0	20.0	Similar to above specimen except electrical conductivity, $14.1 \times 10^6 \text{ ohm}^{-1}\text{cm}^{-1}$ at 22 C.
3	230 L	1925	326.2			70.0	30.0	Similar to above specimen except electrical conductivity, $14.4 \times 10^6 \text{ ohm}^{-1}\text{cm}^{-1}$ at 22 C.
4	230 L	1925	326.2			60.0	40.0	Similar to above specimen except electrical conductivity, $14.7 \times 10^6 \text{ ohm}^{-1}\text{cm}^{-1}$ at 22 C.

DATA TABLE NO. 94 THERMAL CONDUCTIVITY OF [CADMIUM + ZINC] ALLOYS

(Cd + Zn ~ 99.50%; impurity  $\leq 0.20\%$ , each)  
[Temperature, T, K; Thermal Conductivity,  $\kappa$ , Watt  $\text{cm}^{-1}\text{K}^{-1}$ ]

T	$\kappa$
<u>CURVE 1<sup>o</sup></u>	
326.2	0.954
<u>CURVE 2<sup>o</sup></u>	
326.2	0.967
<u>CURVE 3<sup>o</sup></u>	
326.2	0.996
<u>CURVE 4<sup>o</sup></u>	
326.2	1.021

No. graphical presentation

SPECIFICATION TABLE NO. 95 THERMAL CONDUCTIVITY OF [CHROMIUM + NICKEL] ALLOYS  
 (Cr + Ni ~ 99.50%; impurity  $\leq 0.20\%$  each)

Curve No.	Ref. Method No.	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)		Composition (continued), Specifications and Remarks
						Cr	Ni	
1	230	L	1925	329.2		50.0	50.0	Approx. composition: specimen ~ 5 cm long with cross-sections 0.3 cm; made from nickel 99.75 to 99.85 pure including cobalt, trace Fe and Cu, supplied by International Nickel Co. of America, fused with Cr supplied by Elmer and Amend; electrical conductivity $0.83 \times 10^7$ ohm <sup>-1</sup> cm <sup>-1</sup> at 22 C.

DATA TABLE NO. 95 THERMAL CONDUCTIVITY OF [CHROMIUM + NICKEL] ALLOYS

(Cr + Ni ~ 99.50%; impurity  $\leq 0.20\%$  each)

[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup>K<sup>-1</sup>]

T k  
 CURVE #  
 329.2 0.117

\* No graphical presentation

## SPECIFICATION TABLE NO. 96 THERMAL CONDUCTIVITY OF [COBALT + CARBON] ALLOYS

(Co + C = 99.50%; impurity  $\leq$  0.20%, each)

Curve No.	Ref. Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) Co	Composition (weight percent) C	Composition (continued), Specifications and Remarks
1	238	E	1927	303.2		~99.46	0.22	0.20 Fe, 0.003 P, 0.034 S, 0.032 Si, 0.05 Al, trace Mn and Ni; Co supplied by Sugiyama & Co.; specimen 5 mm in dia and 20 cm long; cast and machined; heated 40 min at 800 C and slowly cooled.

## DATA TABLE NO. 96 THERMAL CONDUCTIVITY OF [COBALT + CARBON] ALLOYS

(Co + C = 99.50%; impurity  $\leq$  0.20%, each)(Temperature, T, K; Thermal Conductivity,  $k$ , Watt  $\text{cm}^{-1}\text{K}^{-1}$ )

T k

CURVE 1<sup>st</sup>

303.2 0.544

No graphical presentation

SPECIFICATION TABLE NO. 97 THERMAL CONDUCTIVITY OF [COBALT + CHROMIUM] ALLOYS

(Co + Cr > 99.50%; impurity ≤ 0.20% each)

Curve No.	Ref. Method No. Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)		Composition (continued), Specifications and Remarks
						Co	Cr	
1	230	L	1955	332.2		90.0	10.0	Alloy made from Cr (pure and free from C) and Co both supplied by Eimer and Amend; specimen ~ 5 cm long with a cross section of 0.3 cm <sup>2</sup> ; electrical conductivity 1.78 x 10 <sup>6</sup> ohm <sup>-1</sup> cm <sup>-1</sup> at 22 C.
2	230	L	1955	332.2		70.0	30.0	Similar to above specimen except electrical resistivity, 1.26 x 10 <sup>6</sup> ohm <sup>-1</sup> cm <sup>-1</sup> at 22 C.
3	230	L	1955	332.2		60.0	40.0	Similar to above specimen except electrical resistivity, 1.09 x 10 <sup>6</sup> ohm <sup>-1</sup> cm <sup>-1</sup> at 22 C.

DATA TABLE NO. 97 THERMAL CONDUCTIVITY OF [COBALT + CHROMIUM] ALLOYS

(Co + Cr > 99.50%; impurity ≤ 0.20% each)

(Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup>K<sup>-1</sup>)

T	k
<u>CURVE 1*</u>	
332.2	0.142
<u>CURVE 2*</u>	
332.2	0.130
<u>CURVE 3*</u>	
332.2	0.105

\* No graphical presentation

SPECIFICATION TABLE NO. 94 THERMAL CONDUCTIVITY OF (COBALT + NICKEL) ALLOYS

(Co + Ni) ~ 99.50%; impurity &lt; 0.20% each

Curve No.	Ref. Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) Co	Composition (weight percent) Ni	Composition (continued), Specifications and Remarks
1	238	E	1927	303.2		50	50	The alloy made from Ni (containing impurities 0.1 Fe, 0.037 C, 0.013 S, 0.006 Si, 0.013 Cu, trace P, Al and Mn) and Cobalt (containing 0.2 Fe, 0.22 C, 0.003 P, 0.004 S, 0.032 Si, 0.05 Al, trace Ni, Mn and Cu); specimen 5 mm in dia and 20 cm long; cast and machined; then heated for 40 min at 800 C and slowly cooled; electrical resistivity $1.373 \times 10^{-5}$ ohm cm at 30 C.
2	238	E	1927	303.2		60	40	The alloy made from the same materials as above; specimen similarly prepared; electrical resistivity $1.333 \times 10^{-5}$ ohm cm at 30 C.
3	238	E	1927	303.2		70	30	The alloy made from the same materials as above; specimen similarly prepared; electrical resistivity $1.233 \times 10^{-5}$ ohm cm at 30 C.
4	238	E	1927	303.2		75	25	The alloy made from the same materials as above; specimen similarly prepared; electrical resistivity $1.162 \times 10^{-5}$ ohm cm at 30 C.
5	238	E	1927	303.2		80	20	The alloy made from the same materials as above; specimen similarly prepared; electrical resistivity $1.296 \times 10^{-5}$ ohm cm at 30 C.
6	238	E	1927	303.2		85	15	The alloy made from the same materials as above; specimen similarly prepared; electrical resistivity $1.38 \times 10^{-5}$ ohm cm at 30 C.
7	238	E	1927	303.2		90	10	The alloy made from the same materials as above; specimen similarly prepared; electrical resistivity $1.331 \times 10^{-5}$ ohm cm at 30 C.
8	238	E	1927	303.2		95	5	The alloy made from the same materials as above; specimen similarly prepared; electrical resistivity $1.294 \times 10^{-5}$ ohm cm at 30 C.



## DATA TABLE NO. 98 THERMAL CONDUCTIVITY OF [COBALT + NICKEL] ALLOYS

(Co + Ni ≥ 99.50%; Impurity ≤ 0.20% each)

[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup>K<sup>-1</sup>]

T	k
	<u>CURVE 1*</u>
303.2	0.506
	<u>CURVE 2*</u>
303.2	0.523
	<u>CURVE 3*</u>
303.2	0.556
	<u>CURVE 4*</u>
303.2	0.573
	<u>CURVE 5*</u>
303.2	0.556
	<u>CURVE 6*</u>
303.2	0.527
	<u>CURVE 7*</u>
303.2	0.510
	<u>CURVE 8*</u>
303.2	0.523

\* No graphical presentation

# THERMAL CONDUCTIVITY OF COPPER + ALUMINUM ALLOYS

[Cu + Al 99.50%; Impurity ≤ 0.20% each]

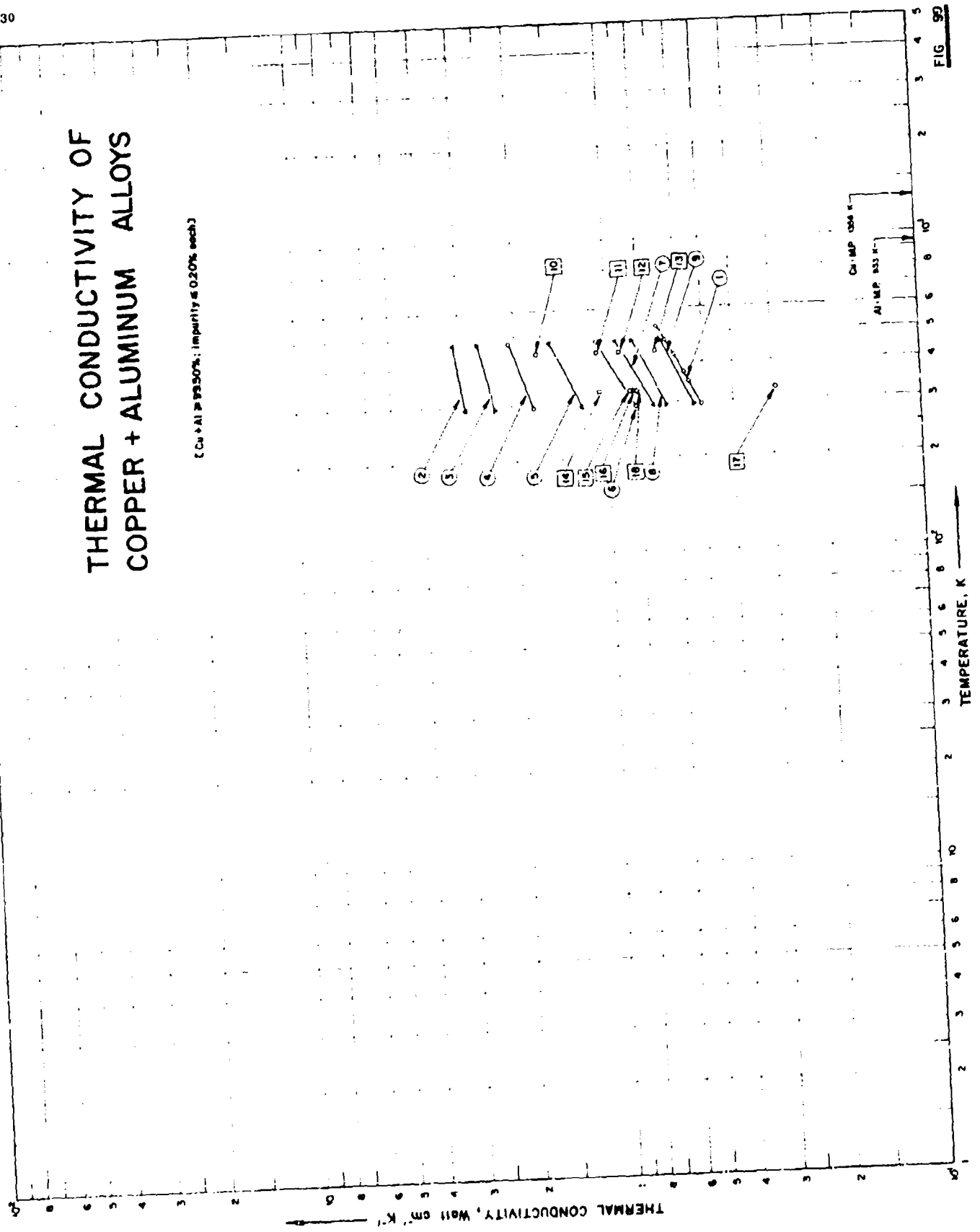


FIG 90

SPECIFICATION TABLE NO. 99 THERMAL CONDUCTIVITY OF (COPPER + ALUMINUM) ALLOYS  
(Cu + Al - 99.50%; impurity  $\leq 0.20\%$  each)

(For Data Reported in Figure and Table No. 99 )

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) Cu	Al	Composition (continued), Specifications and Remarks
1	55	L	1928	293-523		Aluminum bronze, 6	90.0	10.0	Approx. composition: specimen 2.53 cm in diameter and 38 cm long; chill-cast and annealed; electrical resistivity reported at 293, 348, 373, 423, 473 and 523 C respectively as 14.7, 15.6, 16.0, 16.7, 17.5 and 18.3 $\mu$ ohm cm.
2	135	L	1935	293, 473		100	99.77	0.22	0.01 Fe; specimen 0.75 in. in diameter and 8 in. long; rolled and annealed at 750 C for 2 hrs.; electrical conductivity reported at 20 and 200 C respectively as 41.91 $\times 10^4$ and 27.59 $\times 10^4$ ohm $^{-1}$ cm $^{-1}$ .
3	135	L	1935	293, 473		101	99.47	0.47	0.02 Fe; similar to above specimen; electrical conductivity reported at 20 and 200 C respectively as 32.10 $\times 10^4$ and 22.91 $\times 10^4$ ohm $^{-1}$ cm $^{-1}$ .
4	135	L	1935	293, 473		76	93.20	0.71	0.09 Fe; similar to above specimen; electrical conductivity reported at 20 and 200 C respectively as 23.40 $\times 10^4$ and 17.95 $\times 10^4$ ohm $^{-1}$ cm $^{-1}$ .
5	135	L	1935	293, 473		77	98.08	1.89	0.03 Fe; similar to above specimen except annealed at 700 C; electrical conductivity reported at 20, 200 C respectively as 15.91 $\times 10^4$ and 13.00 $\times 10^4$ ohm $^{-1}$ cm $^{-1}$ .
6	135	L	1935	293, 473		45	95.25	4.61	0.14 Fe; similar to above specimen; annealed at 700 C for 2 hrs.; electrical conductivity reported at 20, 200 C respectively as 10.26, 8.824 $\times 10^4$ ohm $^{-1}$ cm $^{-1}$ .
7	135	L	1935	293, 473		46	92.15	7.72	0.13 Fe; specimen 0.75 in. in diameter and 8 in. long; rolled and annealed at 750 C for 3 1/2 hrs.; slowly cooled in furnace; electrical conductivity reported at 20 and 200 C respectively as 8.934 and 7.65 $\times 10^4$ ohm $^{-1}$ cm $^{-1}$ .
8	135	L	1935	293, 473		102	90.56	9.37	0.07 Fe; similar to the above specimen; annealed at 750 C for 2 hrs. and very slowly cooled in furnace at 450 C for 18 hrs.; electrical conductivity reported at 20 and 200 C respectively as 6.24 and 7.056 $\times 10^4$ ohm $^{-1}$ cm $^{-1}$ .
9	135	L	1935	293, 473		130	87.76	12.15	0.09 Fe; similar to the above specimen except electrical conductivity reported at 20 and 200 C respectively as 6.925 and 5.738 $\times 10^4$ ohm $^{-1}$ cm $^{-1}$ .
10	67	L	1932	438, 2			98.25	1.75	Specimen prepared from Al (containing 0.21 Fe, 0.29 Si) and high grade Cu; cast in iron mould 7 in. long and 9/16 in. in diameter; specimen 6 1/2 in. long and 1/2 in. in diameter; annealed at 500 C.

SPECIFICATION TABLE NO. 99 (continued)

Curve No.	Ref. Method No. Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent): C <sub>1</sub>	Composition (weight percent): Al	Composition (continued), Specifications and Remarks
11	67	L	438.2			94.90	5.10	Similar to the above specimen.
12	67	L	438.2			91.55	8.45	Similar to the above specimen.
13	67	L	438.2			87.22	12.78	Similar to the above specimen.
14	230	L	326.2			50.0	50.0	Approx. composition; total impurity < 0.03 in each metal specimen 1.9 cm in diameter and 10 cm long; supplied by Baker; electrical conductivity $15.3 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 23 C.
15	230	L	326.2			60.0	40.0	Similar to the above specimen; except electrical conductivity $10.6 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 23 C.
16	230	L	326.2			70.0	30.0	Similar to the above specimen; except electrical conductivity $9.76 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 23 C.
17	230	L	326.2			80.0	20.0	Similar to the above specimen except electrical conductivity $3.60 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 23 C.
18	230	L	326.2		Aluminum bronze	90.0	10.0	Similar to the above specimen except electrical conductivity $9.98 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 23 C.

DATA TABLE NO. 99 THERMAL CONDUCTIVITY OF COPPER + ALUMINUM ALLOYS

(Cu + Al : 99.50%; impurity  $\pm 0.20\%$  each)

(Temperature, T, K; Thermal Conductivity, k, Wat: cm<sup>-1</sup>K<sup>-1</sup>)

T	k	T	k
<u>CURVE 1</u>			
293.2	0.510	293.2	0.536
473.2	0.532	473.2	0.674
<u>CURVE 10</u>			
423.2	0.415	438.2	1.695
473.2	0.633	<u>CURVE 11</u>	
523.2	0.695	438.2	1.071
<u>CURVE 2</u>			
293.2	2.912	<u>CURVE 12</u>	
473.2	3.113	438.2	0.916
<u>CURVE 3</u>			
293.2	2.347	<u>CURVE 13</u>	
473.2	2.607	438.2	0.707
<u>CURVE 4</u>			
293.2	1.749	<u>CURVE 14</u>	
473.2	2.084	326.2	1.059
<u>CURVE 5</u>			
293.2	1.226	<u>CURVE 15</u>	
473.2	1.544	326.2	0.753
<u>CURVE 6</u>			
293.2	0.828	<u>CURVE 16</u>	
473.2	1.071	326.2	0.745
<u>CURVE 7</u>			
293.2	0.724	<u>CURVE 17</u>	
473.2	0.937	326.2	0.293
<u>CURVE 8</u>			
293.2	0.653	<u>CURVE 18</u>	
473.2	0.837	326.2	0.816

## SPECIFICATION TABLE NO. 100 THERMAL CONDUCTIVITY OF COPPER + ANTIMONY ALLOYS

(Cu + Sb &gt; 99.50%; Impurity ≤ 0.20% each)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)	Composition (continued), Specifications and Remarks
							Cu Sb	
1	319 320	P	1966	1073.2	8		50 50	Molten specimen contained in a thin-walled stainless steel cylindrical crucible of dimensions 24 mm dia x 100 mm long; electrical resistivity reported as 138 and 153 $\mu$ ohm cm at 700 and 800 C, respectively; thermal conductivity values calculated from the thermal diffusivity and the specific heat measurements using the density data from Bienias, A and Sauerwald, F., Z. anorg. chem., 41, 51, 1927.

## DATA TABLE NO. 100 THERMAL CONDUCTIVITY OF COPPER + ANTIMONY ALLOYS

(Cu + Sb &gt; 99.50%; Impurity ≤ 0.20% each)

[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup>K<sup>-1</sup>]

T k

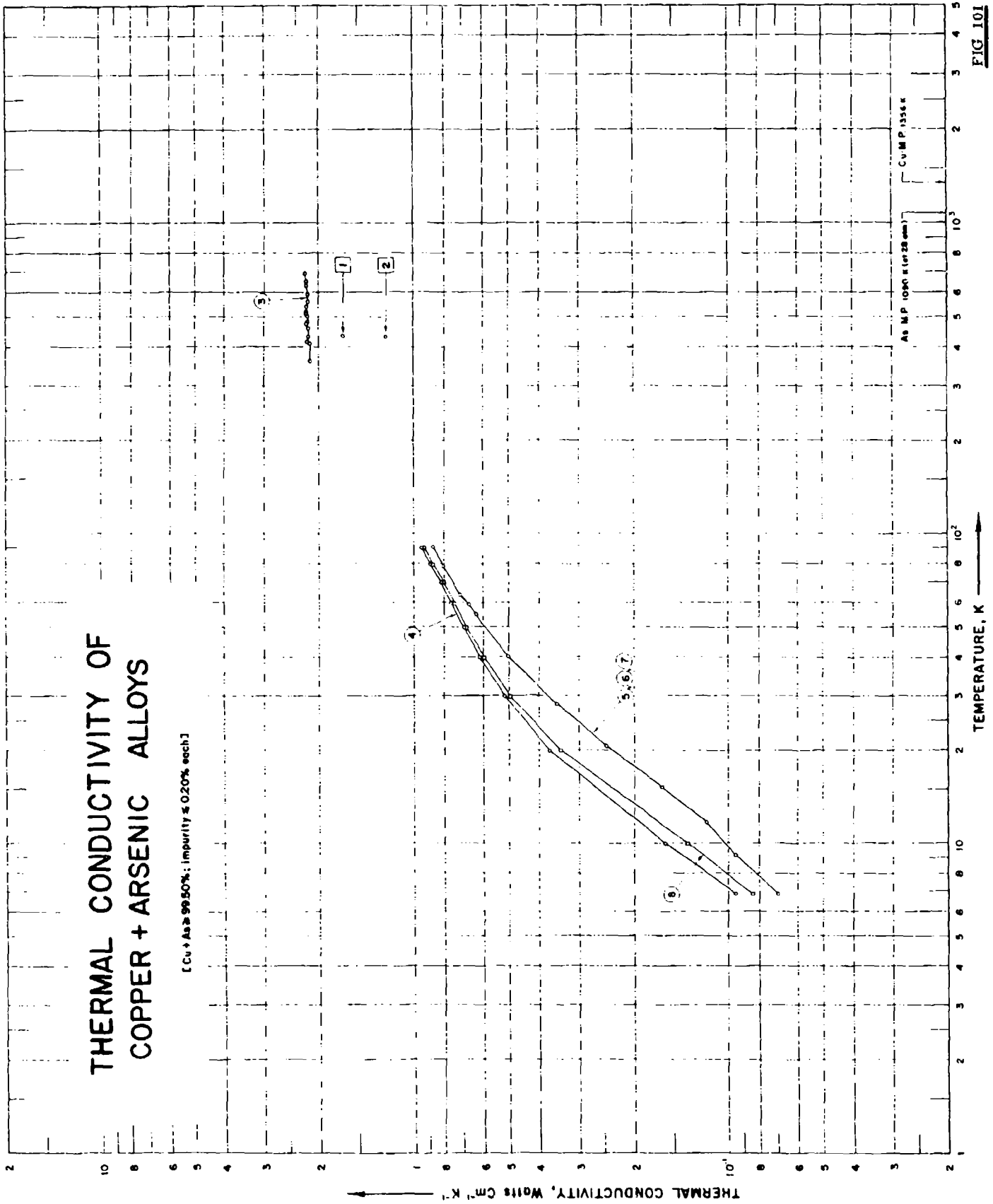
CURVE 1

1073.2 0.236

No graphical presentation

# THERMAL CONDUCTIVITY OF COPPER + ARSENIC ALLOYS

[Cu + As = 99.50%; Impurity ≤ 0.20% each]



## SPECIFICATION TABLE No. 101 THERMAL CONDUCTIVITY OF COPPER-ARSENIC ALLOYS

100 Cu, As 99.500% impurity 0.200% each)

For Data Reported in Figure and Table No. 101.

Curve No.	Ref. No. Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Cu, %	As, %	Composition (continued), Specifications and Remarks
1	67 L	1952	478-2			99.608	0.392	High grade copper bar with traces impurities; specimen 6.5 in. long and 0.5 in. in diameter; cast and machined; thermal conductivity data obtained from the mean value of 16 readings.
2	67 L	1952	478-2			99.565	0.435	Similar to the above specimen.
3	30 L	1923	303-694	-2.6		99.53	0.470	0.022 Ni, 0.003 Sb, Nil Sn and Pb, specimen 15.5 in. long and 0.75 in. in diameter; supplied by Birmingham factory and Metal Co.; rolled, drawn and annealed.
4	236 L	1959	6.9-90		No. 0	99.55	0.45	0.05 P; specimen 3.25 mm in diameter and 8 cm long, drawn and then prolongedly annealed at 450 C.
5	236 L	1959	6.9-91		No. 1	99.55	0.45	0.05 P; similar to above specimen; drawn, prolongedly annealed at 450 C and then severely deformed torsionally.
6	236 L	1959	6.9-91		No. 2	99.55	0.45	0.05 P; the above specimen annealed in helium as temp. increased up to 175 C at a rate of 6 C per min.
7	236 L	1959	6.9-91		No. 3	99.55	0.45	0.05 P; the above specimen again annealed in helium as temp. increased up to 275 C at a rate of 6 C per min.
8	236 L	1959	6.9-90		No. 4	99.55	0.45	0.05 P; the above specimen again annealed in helium as temp. increased up to 450 C at a rate of 6 C per min.



## DATA TABLE NO. 101 THERMAL CONDUCTIVITY OF [COPPER + ARSENIC] ALLOYS

(Cu + As  $\geq$  99.50%; Impurity  $\leq$  0.20% each)[Temperature, T, K; Thermal Conductivity, k, Watts  $\text{cm}^{-1}\text{K}^{-1}$ ]

T	k	T	k	T	k
<u>CURVE 1</u>					
433.2	1.665	20.7	0.245	30.0	0.497
<u>CURVE 2</u>					
438.2	1.222	28.3	0.351	40.0	0.597
<u>CURVE 3</u>					
363.2	2.142	55.1	0.638	50.0	0.684
413.2	2.142	59.1	0.670	60.0	0.744
415.2	2.167	64.1	0.712	70.0	0.809
434.2	2.155	79.3	0.805	80.0	0.871
462.7	2.163	91.3	0.872	90.0	0.927
478.2	2.184	<u>CURVE 6</u>			
485.2	2.167	6.9	0.0700		
504.7	2.172	9.2	0.0947		
515.7	2.192	11.7	0.118		
521.2	2.192	15.3	0.163		
542.2	2.188	20.7	0.245		
565.7	2.163	28.3	0.351		
594.7	2.167	40.5	0.502		
633.2	2.188	55.1	0.638		
657.2	2.192	59.1	0.670		
693.7	2.205	64.1	0.712		
<u>CURVE 4</u>					
6.9	0.0947	6.9	0.0700		
10.0	0.155	9.2	0.0947		
20.0	0.371	11.7	0.118		
30.0	0.517	15.3	0.163		
40.0	0.618	20.7	0.245		
50.0	0.695	28.3	0.351		
60.0	0.762	40.5	0.502		
70.0	0.820	55.1	0.638		
80.0	0.889	59.1	0.670		
90.0	0.947	64.1	0.712		
<u>CURVE 5</u>					
6.9	0.0700	6.9	0.0700		
9.2	0.0947	9.2	0.0947		
11.7	0.118	11.7	0.118		
15.3	0.163	15.3	0.163		
<u>CURVE 7</u>					
6.9	0.0700	6.9	0.0700		
9.2	0.0947	9.2	0.0947		
11.7	0.118	11.7	0.118		
15.3	0.163	15.3	0.163		
<u>CURVE 8</u>					
6.9	0.0700	6.9	0.0700		
9.2	0.0947	9.2	0.0947		
11.7	0.118	11.7	0.118		
15.3	0.163	15.3	0.163		
20.0	0.245	20.0	0.245		
28.3	0.351	28.3	0.351		
40.5	0.502	40.5	0.502		
55.1	0.638	55.1	0.638		
59.1	0.670	59.1	0.670		
64.1	0.712	64.1	0.712		
79.3	0.805	79.3	0.805		
91.3	0.872	91.3	0.872		
<u>CURVE 8</u>					
6.9	0.0700	6.9	0.0700		
9.2	0.0947	9.2	0.0947		
11.7	0.118	11.7	0.118		
15.3	0.163	15.3	0.163		
20.0	0.245	20.0	0.245		
28.3	0.351	28.3	0.351		
40.5	0.502	40.5	0.502		
55.1	0.638	55.1	0.638		
59.1	0.670	59.1	0.670		
64.1	0.712	64.1	0.712		
79.3	0.805	79.3	0.805		
91.3	0.872	91.3	0.872		

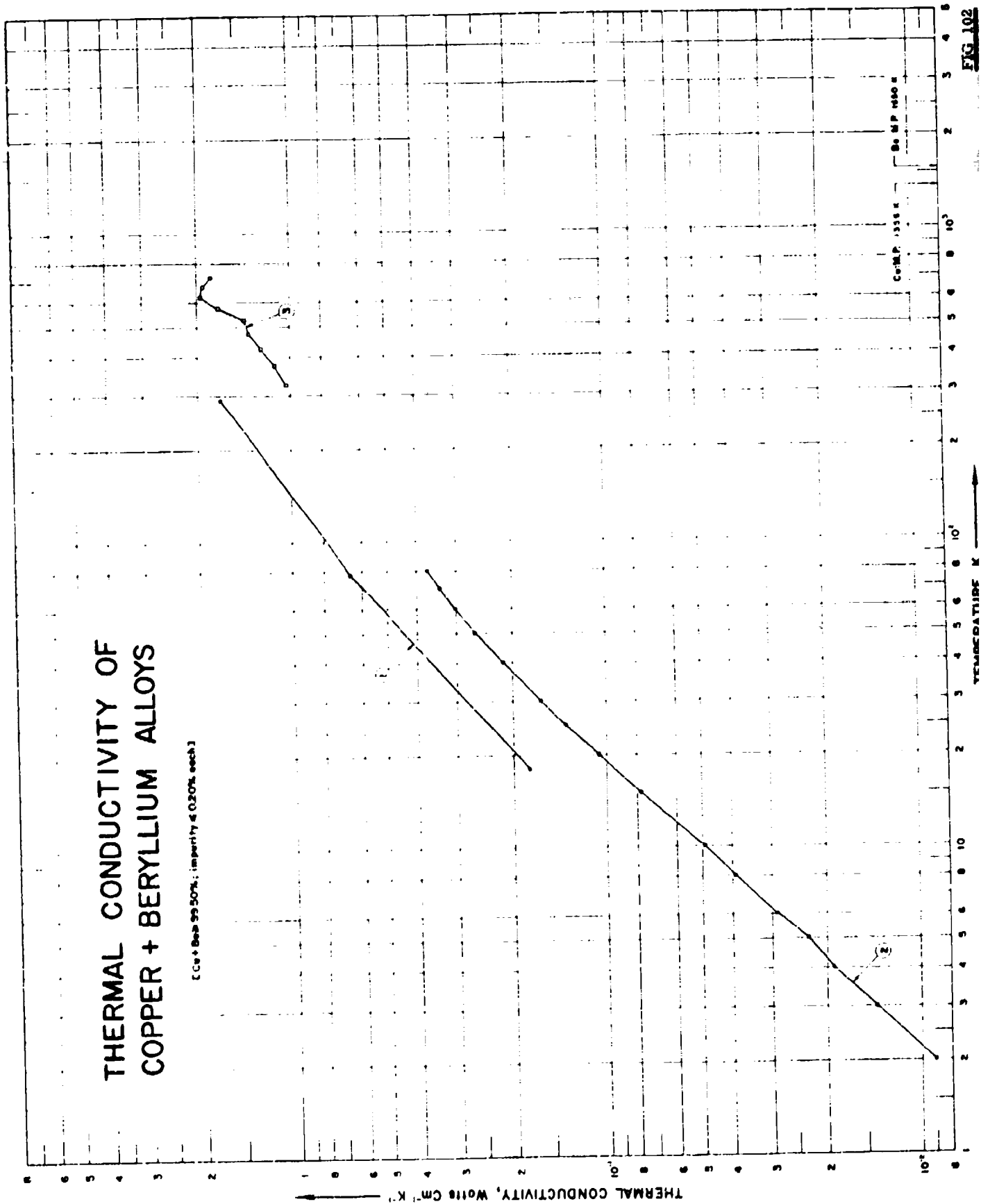


FIG 102

## SPECIFICATION TABLE NO. 102 THERMAL CONDUCTIVITY OF (COPPER-BERYLLIUM) ALLOYS

(Cu + Be 99.50%, impurity 0.20% each)

For Data Reported in Figure and Table No. 102

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (Cu)	Composition (weight percent) Be	Composition (continued), Specifications and Remarks
1	193	L	1939	18-290		Beryllium Bronze	98.49	1.50	0.01 Fe; specimen ~4.97 mm in diameter and 92 mm long; heated at 400 C for 3 hrs. and cooled slowly.
2	229	L	1955	2.0-90		Beryllium Copper	98.0	2.0	Approx. composition: heated at 300 C for 2 hrs.; electrical resistivity 8.25, 6.2 and 5.54 x 10 <sup>-6</sup> ohm cm at room temp. 77 and 4.2 K respectively.
3	338	L	1933	323-723			97.55	2.45	Approx. composition; specimen machined from a cast bar; heated to 1450 F for 1 hour and quenched in cold water; hardened at 570 F and air cooled.

## DATA TABLE NO. 102 THERMAL CONDUCTIVITY OF [COPPER + BERYLLIUM] ALLOYS

(Cu + Be  $\approx$  99.50%, Impurity  $\leq$  0.20% each)[Temperature, T, K. Thermal Conductivity, k, Watt cm<sup>-1</sup>K<sup>-1</sup>]

T	k
<u>CURVE 1</u>	
18	0.177
78	0.653
290	1.700
<u>CURVE 2</u>	
2	0.009
3	0.014
4	0.019
5	0.023
6	0.029
8	0.039
10	0.049
15	0.078
20	0.107
25	0.135
30	0.162
40	0.215
50	0.262
60	0.304
70	0.340
80	0.371
<u>CURVE 3</u>	
323.2	1.046
373.2	1.130
423.2	1.255
473.2	1.380
523.2	1.420
573.2	1.715
623.2	1.966
673.2	1.927
723.2	1.897

SPECIFICATION TABLE NO. 103 THERMAL CONDUCTIVITY OF [COPPER + CADMIUM] ALLOYS

(Cu + Cd &gt; 99.50%; impurity &lt; 0.20% each)

Curve No.	Ref. Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)		Composition (continued), Specifications and Remarks
						Cu	Cd	
1	L	1935	293, 473		Bar 134	99.21	0.85	0.009 Si, 0.007 Fe; specimen 0.75 in. in dia and 8 in. long; supplied by American Brass Co.; rolled, annealed and cold drawn then heated at 700 C for 2 hrs; electrical conductivity 50.561 and 31.19 x 10 <sup>4</sup> ohm <sup>-1</sup> cm <sup>-1</sup> at 20 and 200 C respectively.
2	L	1935	293, 473		Bar 69	99.04	0.90	0.07 Fe; similar to above specimen except heated at 750 C for 1.50 hrs; electrical conductivity 36.87 and 26.56 x 10 <sup>4</sup> ohm <sup>-1</sup> cm <sup>-1</sup> at 20 and 200 C respectively.

DATA TABLE NO. 103 THERMAL CONDUCTIVITY OF [COPPER + CADMIUM] ALLOYS

(Cu + Cd &gt; 99.50%; impurity &lt; 0.20% each)

[ Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup>K<sup>-1</sup> ]

T	k
<u>CURVE 1<sup>c</sup></u>	
293.2	3.448
473.2	3.519
<u>CURVE 2<sup>c</sup></u>	
293.2	2.761
473.2	3.117

\* No graphical presentation

# THERMAL CONDUCTIVITY OF COPPER + CHROMIUM ALLOYS

[Cu+Cr ≥ 99.90%; Impurity ≤ 0.20% each]

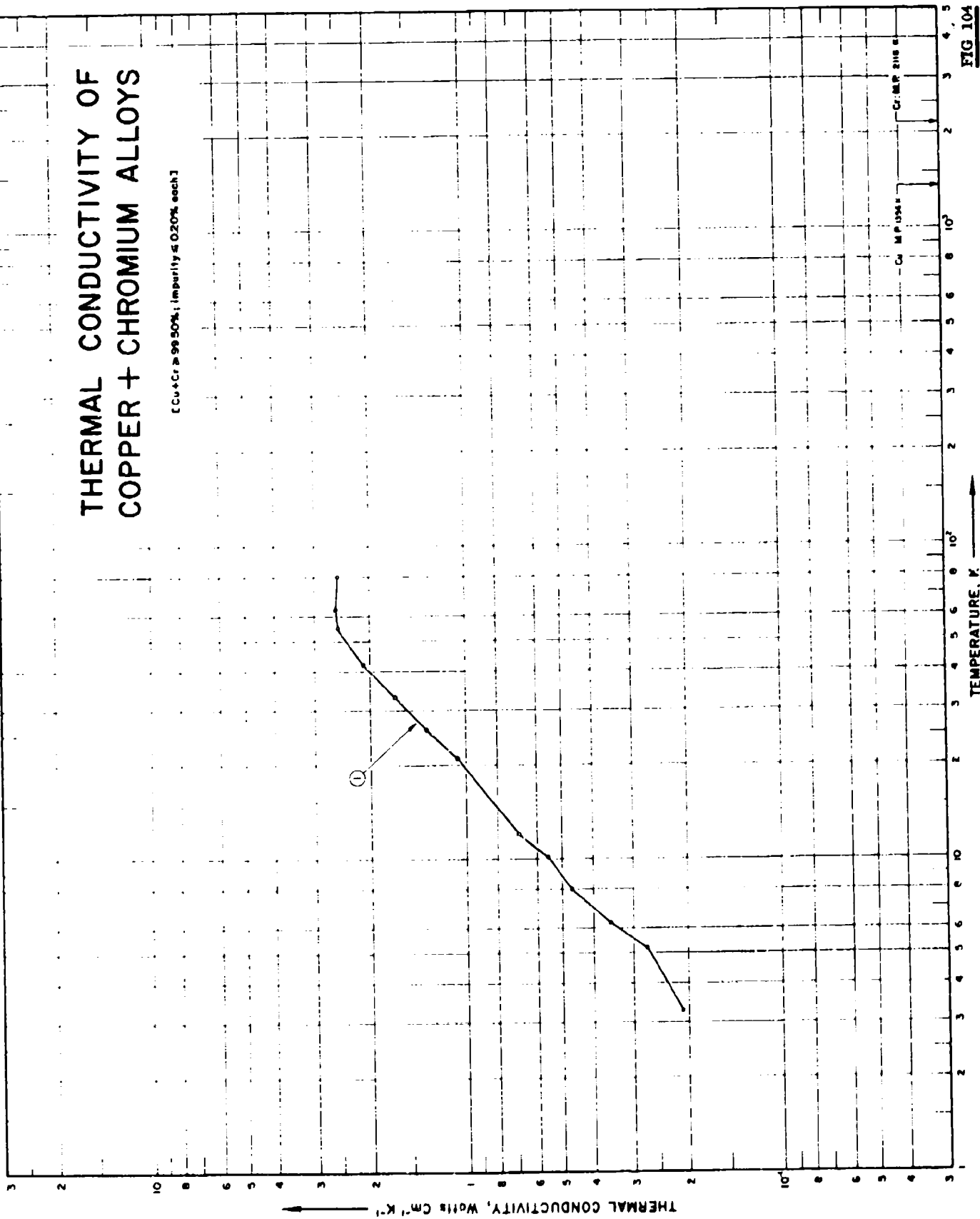


FIG 104

## SPECIFICATION TABLE NO. 104 THERMAL CONDUCTIVITY OF COPPER - CHROMIUM ALLOYS

(Cu - Cr: 99.50%, impurity 0.20% each)

(For Data Reported in Figure and Table No. 104)

Curve No.	Ref. Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) Cu Cr	Composition (continued), Specifications and Remarks
1	L	1956	3.0-40	±5.0	Russian cupralloy type 5	99.2 0.61	0.1% Ag; unannealed.

## DATA TABLE NO. 104 THERMAL CONDUCTIVITY OF [COPPER + NIROMIUM] ALLOYS

(Cu + Cr : 99.50%; Impurity : 0.20% each)

(Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup>K<sup>-1</sup>)

T	k
<u>CURVE 1</u>	
3.28	0.209
5.20	0.272
6.23	0.351
8.00	0.469
10.2	0.556
12.0	0.686
21.0	1.06
26.0	1.33
33.0	1.67
42.0	2.09
55.0	2.51
63.2	2.55
80.0	2.51



# THERMAL CONDUCTIVITY OF COPPER + COBALT ALLOYS

[Cu + Co > 99.50%; impurity < 0.20% each]

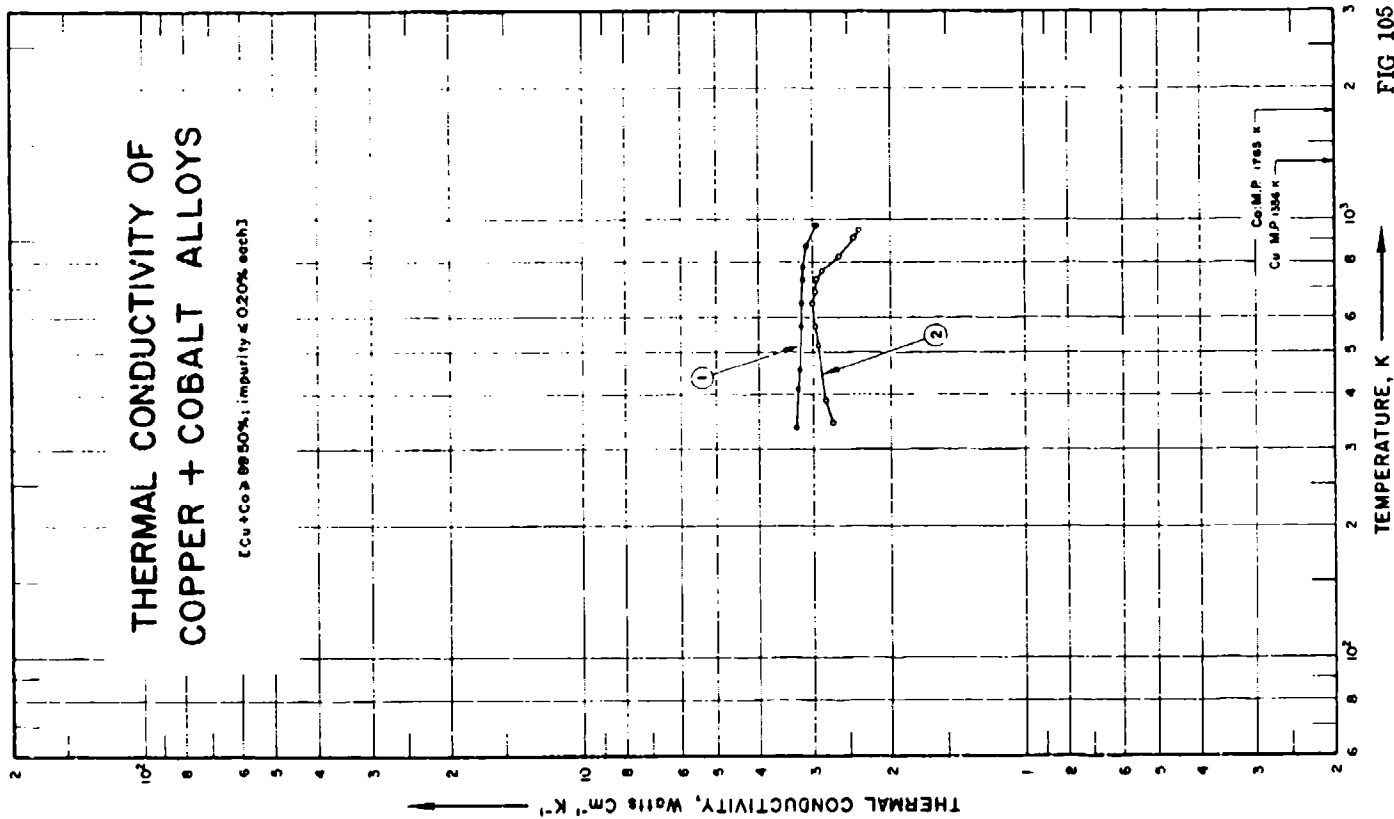


FIG 105

## SPECIFICATION TABLE NO. 105 THERMAL CONDUCTIVITY OF [COPPER + COBALT] ALLOYS

(Cu + Co: 99.50%; impurity: 0.20% each)

[For Data Reported in Figure and Table No. 105]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) Cu	Composition (weight percent) Co	Composition (continued), Specifications and Remarks
1	377		1957	337-973			99.23	0.60	0.10 Zr, 0.03 F; electrical conductivity reported at 63.3, 138, 181, 302, 376, 457, 508, 600 and 700 C respectively as 39, 8, 32, 0, 28, 90, 22, 61, 19, 8, 17, 37, 16, 14, 14, 20 and $12.05 \times 10^4$ ohm <sup>-1</sup> cm <sup>-1</sup> .
2	541	E	1959	345-948			99.25	0.65	0.1 Bc

DATA TABLE NO. 105 THERMAL CONDUCTIVITY OF (COPPER + COBALT) ALLOYS

(Cu + Co) - 5.50%; Impurity - 0.20% each

[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup> K<sup>-1</sup>]

T	k
<u>CURVE 1</u>	
336.5	3.31
411.2	3.28
455.0	3.25
575.2	3.22
649.5	3.22
730.7	3.19
781.5	3.18
871.2	3.12
973.2	2.95
<u>CURVE 2</u>	
345.0	2.720
388.2	2.829
518.7	2.929
573.2	2.983
648.2	3.025
686.2	2.979
733.2	2.954
766.2	2.870
829.2	2.636
913.2	2.452
948.2	2.377

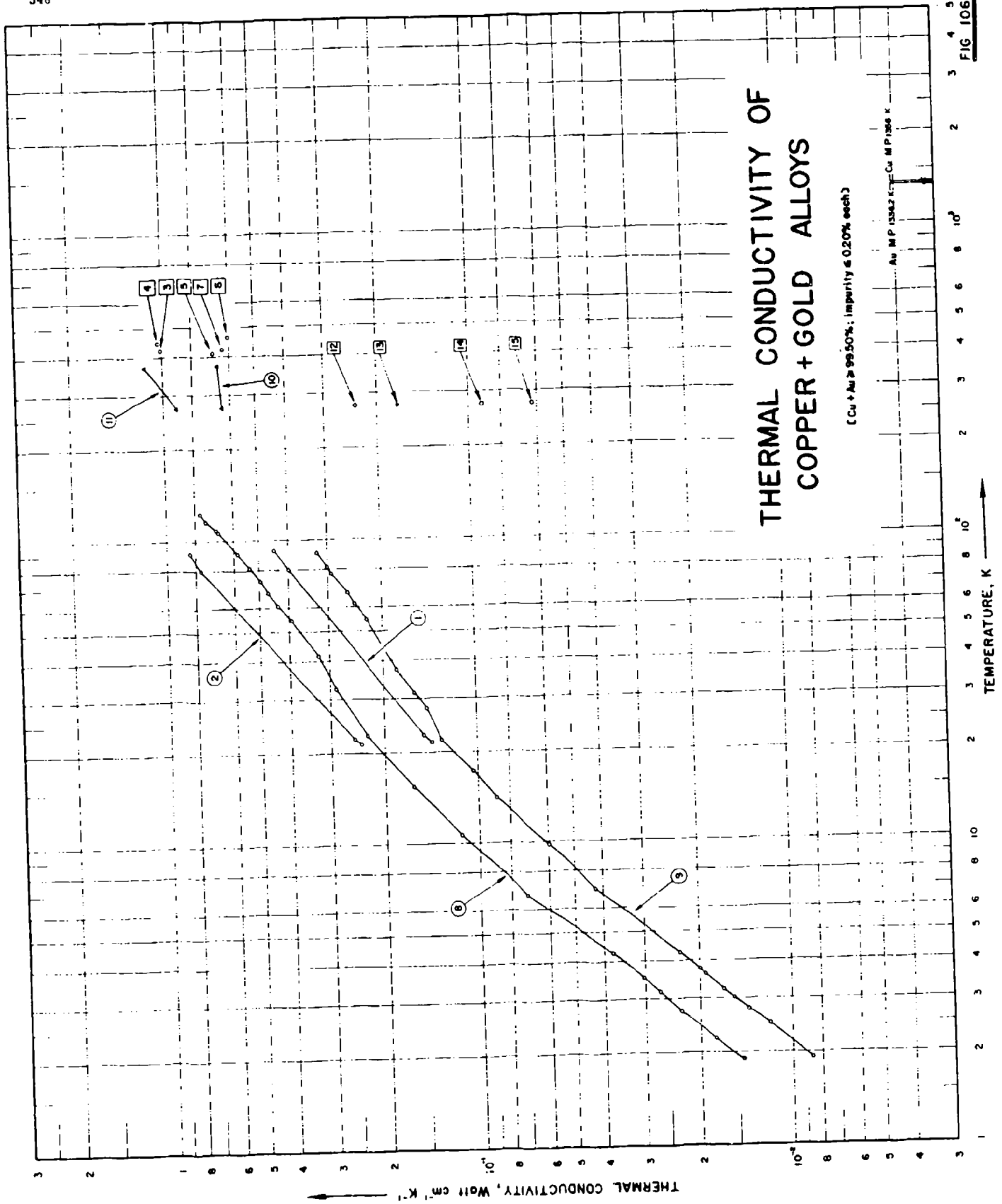


FIG 106

## SPECIFICATION TABLE NO. 106 THERMAL CONDUCTIVITY OF [COPPER + GOLD] ALLOYS

(Cu + Au - 99.50%; impurity - 0.20% each)

[ For Data Reported in Figure and Table No. 106 ]

Curve No.	Rel. Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)		Composition (continued), Specifications and Remarks
						Cu	Au	
1	54	L	1934	22-93		75.2	24.8	Calculated composition: polycrystalline; unannealed; <del>electrical resistivity 5.09 and 4.71 <math>\mu</math>ohm cm at <del>0-190</del> and -251 C, respectively.</del> <del>6.54</del> , <del>0-190</del> , <del>3.40-156 ohm-cm</del> electrical resistivity 2.487 and 2.172 $\mu$ ohm cm at <del>0-190</del> and -251 C, respectively. <del>3.08</del> , <del>0-190</del>
2	58	L	1934	21-91		87.6	12.6	Similar to above specimen except <del>3.40-156 ohm-cm</del> electrical resistivity 2.487 and 2.172 $\mu$ ohm cm at <del>0-190</del> and -251 C, respectively. <del>3.08</del> , <del>0-190</del>
3	232	L	1957	422.7		56.33	43.67	Calculated composition: specimen 1.43 cm long with cross section 0.63 cm <sup>2</sup> ; density 14.3 g cm <sup>-3</sup> ; cast.
4	232	L	1957	448.2		56.33	43.67	The above specimen; annealed for 10 hrs.
5	232	L	1957	411.2		56.33	43.67	Similar to above specimen; annealed for 20 hrs.
6	232	L	1957	467.2		56.33	43.67	Similar to above specimen; annealed for 30 hrs.
7	232	L	1957	422.2		56.33	43.67	Similar to above specimen; annealed for 40 hrs.
8	233	L	1957	1.9-124		<del>79.9</del> <del>40.7</del>	<del>2.1</del>	Specimen 8 cm long and 0.5 cm in dia; annealed at 750 C for 1 hr; $\rho_0 = 3.53 \mu$ ohm cm.
9	233	L	1957	1.9-91		<del>62.0</del> <del>40.7</del>	<del>38.0</del>	Similar to above specimen except $\rho_0 = 7.04 \mu$ ohm cm.
10	246	T	1919	273.373		55.24	44.76	Calculated composition; specimen rolled and drawn to wire 1 mm dia; heated to near melting point for 0.5 hr; electrical conductivity $5.7 \times 10^4$ and $5.5 \times 10^4$ ohm <sup>-1</sup> cm <sup>-1</sup> at 0 and 100 C, respectively.
11	246	T	1919	273.373		73.52	26.48	Similar to above specimen; electrical conductivity $10.7 \times 10^4$ and $9.1 \times 10^4$ ohm <sup>-1</sup> cm <sup>-1</sup> at 0 and 100 C, respectively.
12	430	T	1924	273.2		94.6	5.4	Calculated composition; specimen rolled and drawn to a wire of 3 cm in length and 1 mm <sup>2</sup> cross-section, then heated to the melting point; electrical resistivity 8.2 $\mu$ ohm cm at 0 C.
13	430	T	1924	273.2		87.6	12.4	Similar to above specimen except electrical resistivity 4.7 $\mu$ ohm cm at 0 C.
14	430	T	1924	273.2		72.7	27.3	Similar to above specimen except electrical resistivity 7.3 $\mu$ ohm cm at 0 C.
15	430	T	1924	273.2		55.0	45.0	Similar to above specimen except electrical resistivity 10.4 $\mu$ ohm cm at 0 C.

DATA TABLE NO. 106 THERMAL CONDUCTIVITY OF COPPER-GOLD ALLOYS

(Cu + Au 99.50% Impurity 0.20% each)

[Temperature, T, K; Thermal Conductivity,  $k$ , Watt cm<sup>-1</sup> K<sup>-1</sup>]

T	k	T	k	T	k
<u>CURVE 1</u>					
21.50	0.141	15.46	0.1627	273.20	0.64
22.70	0.151	22.75	0.229	273.20	0.66
29.20	0.400	32.14	0.289	<u>CURVE 11</u>	
32.70	0.446	41.87	0.325	273.20	0.91
<u>CURVE 2</u>					
		54.77	0.395	273.20	1.14
		60.94	0.438	<u>CURVE 12</u>	
		67.16	0.471	273.2	2.37
21.30	0.239	71.41	0.497	<u>CURVE 13</u>	
22.10	0.253	81.10	0.519	273.2	1.71
29.00	0.776	90.50	0.587	<u>CURVE 14</u>	
31.00	0.841	104.50	0.676	273.2	0.91
<u>CURVE 3</u>					
		107.50	0.687	<u>CURVE 15</u>	
		116.00	0.740	273.2	0.62
422.70	1.021	124.00	0.771	<u>CURVE 16</u>	
<u>CURVE 9</u>					
<u>CURVE 4</u>					
448.20	1.046	1.92	0.00863	<u>CURVE 17</u>	
<u>CURVE 5</u>					
		2.48	0.0119	<u>CURVE 18</u>	
		2.77	0.0138	<u>CURVE 19</u>	
		3.94	0.0152	<u>CURVE 20</u>	
		3.23	0.0167	<u>CURVE 21</u>	
411.20	0.686	3.614	0.0191	<u>CURVE 22</u>	
<u>CURVE 6</u>					
		3.75	0.0200	<u>CURVE 23</u>	
		4.29	0.0231	<u>CURVE 24</u>	
		6.93	0.0427	<u>CURVE 25</u>	
467.20	0.607	9.78	0.0594	<u>CURVE 26</u>	
<u>CURVE 7</u>					
		14.08	0.0886	<u>CURVE 27</u>	
		17.25	0.1036	<u>CURVE 28</u>	
422.20	0.632	21.91	0.1308	<u>CURVE 29</u>	
<u>CURVE 8</u>					
		27.82	0.146	<u>CURVE 30</u>	
		31.29	0.1604	<u>CURVE 31</u>	
		37.32	0.182	<u>CURVE 32</u>	
		54.75	0.225	<u>CURVE 33</u>	
1.90	0.0145	61.45	0.247	<u>CURVE 34</u>	
2.23	0.0179	66.94	0.260	<u>CURVE 35</u>	
2.73	0.0230	78.70	0.294	<u>CURVE 36</u>	
3.17	0.0267	81.00	0.302	<u>CURVE 37</u>	
4.26	0.0364	90.70	0.325	<u>CURVE 38</u>	
6.88	0.0777			<u>CURVE 39</u>	
10.69	0.1147			<u>CURVE 40</u>	

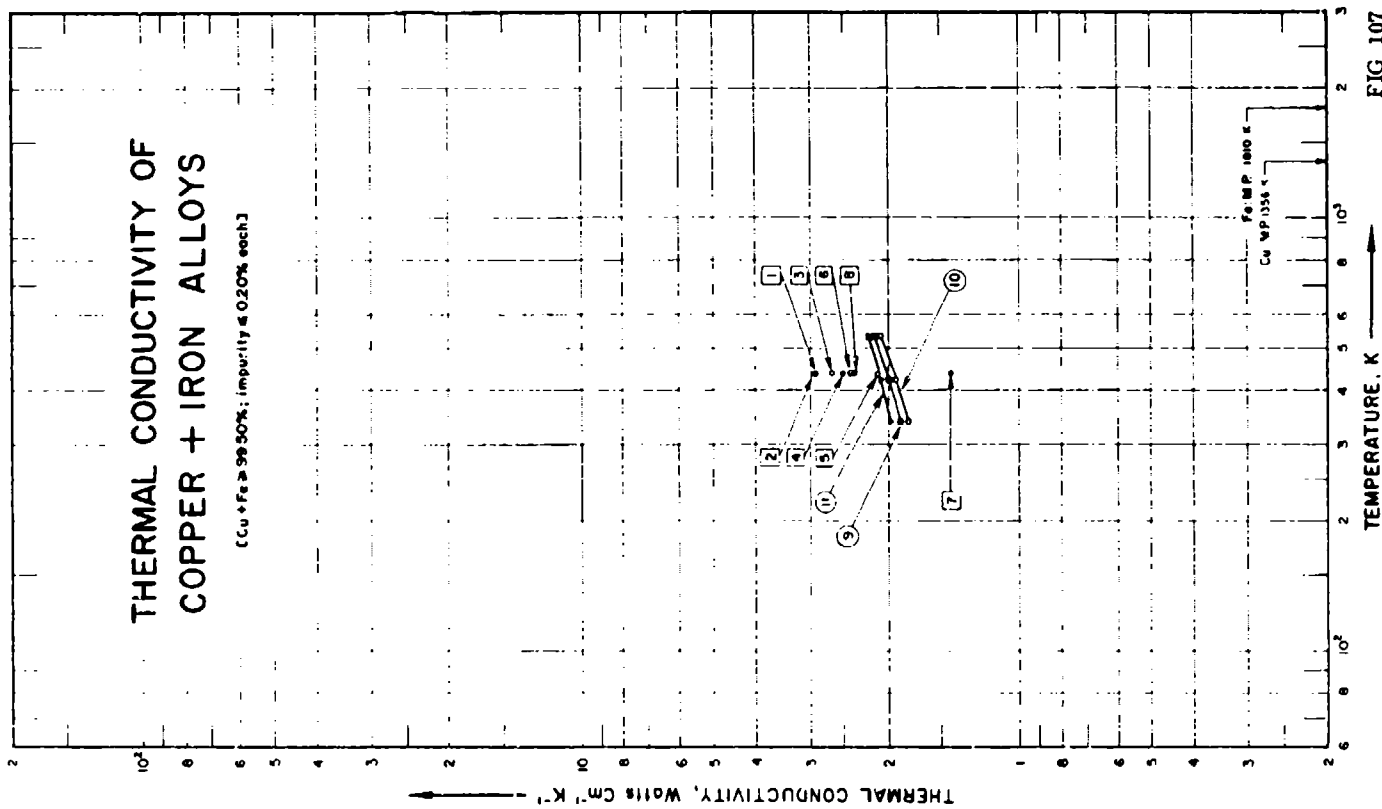


FIG 107

## SPECIFICATION TABLE NO. 107 THERMAL CONDUCTIVITY OF (COPPER + IRON) ALLOYS

(Cu + Fe: 99.50% impurity: 0.20% each)

For Data Reported in Figure and Table No. 107]

Curve No.	Pat. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) Cu	Composition (weight percent) Fe	Composition (continued), Specifications and Remarks
1	67	L	1932	438.2			99.80	0.20	High grade electrolytic Cu with traces of impurities: specimen 6.5 in. long and 0.5 in. in diameter; heated at 1000 C for 1 hr. then quenched.
2	67	L	1932	438.2			99.80	0.20	Similar to the above specimen; heated at 650 C for 1 hr. then quenched.
3	67	L	1932	438.2			99.71	0.29	Similar to the above specimen; heated at 1000 C for 1 hr. then quenched.
4	67	L	1932	438.2			99.71	0.29	Similar to the above specimen; heated at 650 C for 1 hr. then quenched.
5	67	L	1932	438.2			99.50	0.50	Similar to the above specimen; heated at 1000 C for 1 hr. then quenched.
6	67	L	1932	438.2			99.50	0.50	Similar to the above specimen; heated at 650 C for 1 hr. then quenched.
7	67	L	1932	438.2			98.93	1.07	Similar to the above specimen; heated at 1000 C for 1 hr. then quenched.
8	67	L	1932	438.2			98.93	1.07	Similar to the above specimen; heated at 650 C for 1 hr. then quenched.
9	52	L	1932	339-533	± 5.0	F	98.736	1.25	0.014 P; specimen 0.5 in. in diameter and 6 in. long.
10	52	L	1932	339-533	± 5.0	G	95.822	4.16	0.018 P; specimen 0.5 in. in diameter and 6 in. long.
11	52	L	1932	339-533		E	90.568	0.42	0.012 P; similar to above specimen.



DATA TABLE NO. 107 THERMAL CONDUCTIVITY OF COPPER-IRON ALLOYS

(Cu - Fe - 99.50% impurity 0.20% each)

Temperature, T, K. Thermal Conductivity, k, Watt cm<sup>-1</sup> K<sup>-1</sup>.

T	k	T	k
<u>CURVE 1</u>			
438.2	2.367	438.7	1.999
		422.1	2.077
<u>CURVE 2</u>			
438.2	2.215	533.2	2.215
<u>CURVE 3</u>			
438.2	2.983		
<u>CURVE 4</u>			
438.2	2.678		
<u>CURVE 5</u>			
438.2	2.523		
<u>CURVE 6</u>			
438.2	2.113		
<u>CURVE 7</u>			
438.2	2.437		
<u>CURVE 8</u>			
438.2	1.431		
<u>CURVE 9</u>			
438.2	2.385		
<u>CURVE 10</u>			
338.7	1.886		
422.1	2.005		
533.2	2.163		
<u>CURVE 11</u>			
338.7	1.817		
422.1	1.916		
533.2	2.094		

# THERMAL CONDUCTIVITY OF COPPER + LEAD ALLOYS

[Cu + Pb 99.50%; impurity  $\leq$  0.20% each]

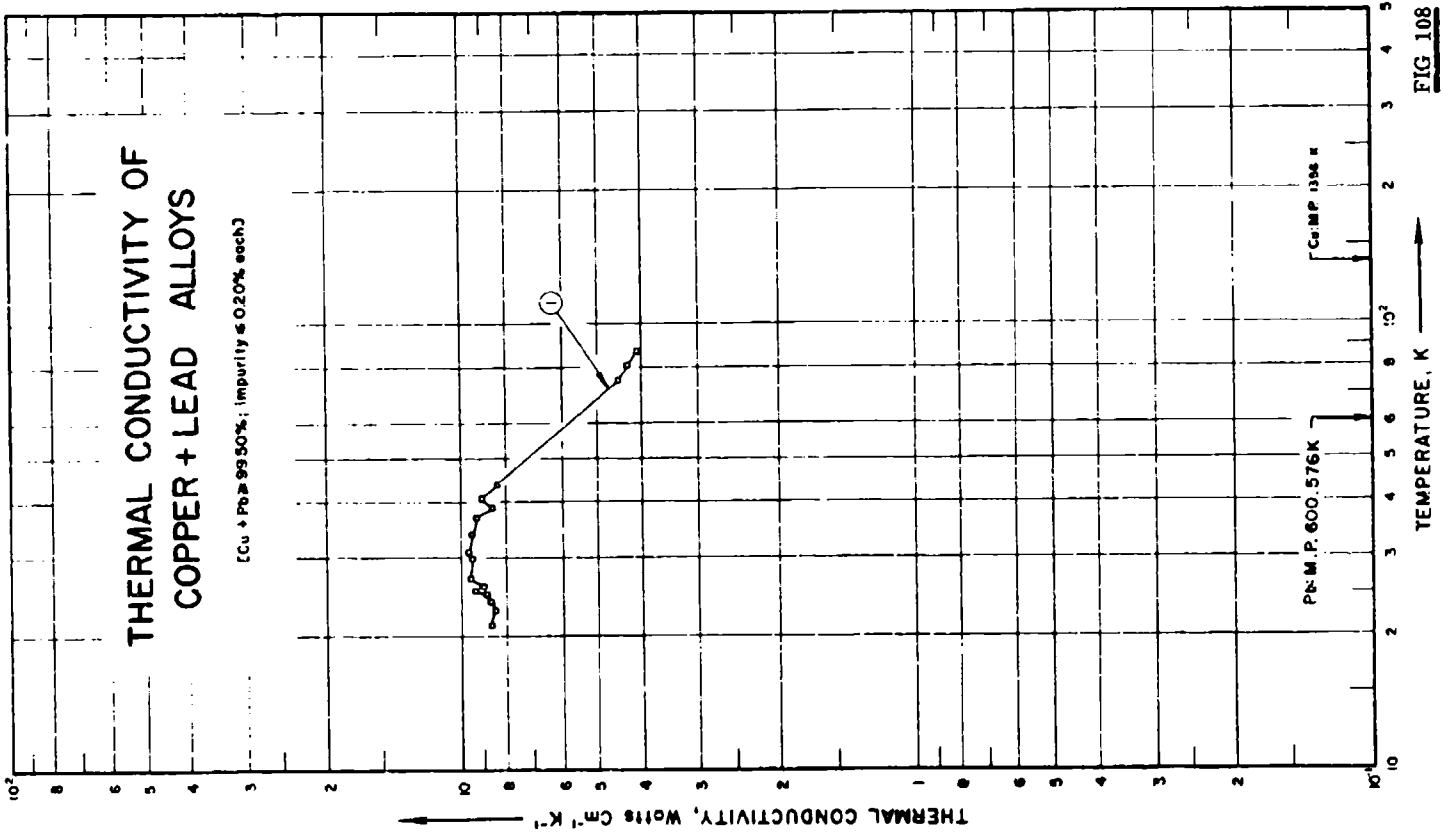


FIG. 108

SPECIFICATION TABLE NO. 108 THERMAL CONDUCTIVITY OF [COPPER + LEAD] ALLOYS

(Cu + Pb ≥ 99.50%; impurity ≤ 0.20% each)

[For Data Reported in Figure and Table No. 108]

Curve No.	Ref. Method No. Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)		Composition (continued), Specifications and Remarks
						Cu	Pb	
1	582	L	1955	21-87	Lead Cu-126	98.94	1.04	Specimen 0.25 in. in dia.; supplied by Kenosha plant of the American Brass Co.; commercial hard-drawn temper rod; grain size 0.004 cm in transverse section and about 0.005 cm long by 0.004 cm wide in the longitudinal section.

## DATA TABLE NO. 1CS THERMAL CONDUCTIVITY OF [COPPER + LEAD] ALLOYS

(Cu + Pb : 99.50%, impurity  $\leq 0.20\%$  each)[Temperature, T, K; Thermal Conductivity, k, Watt  $\text{cm}^{-1} \text{K}^{-1}$ ]

T	k
<u>CURVE 1</u>	
21.3	8.70
23.0	9.50
24.0	8.70
25.0	8.90
25.5	9.40
26.0	9.00
27.0	9.60
30.0	9.50
31.0	9.70
34.0	9.50
37.0	9.30
39.0	8.60
41.0	9.90
44.0	8.40
75.0	4.50
81.0	4.30
87.0	4.10

# THERMAL CONDUCTIVITY OF COPPER + MANGANESE ALLOYS

[Cu + Mn > 99.50%; impurity < 0.20% each]

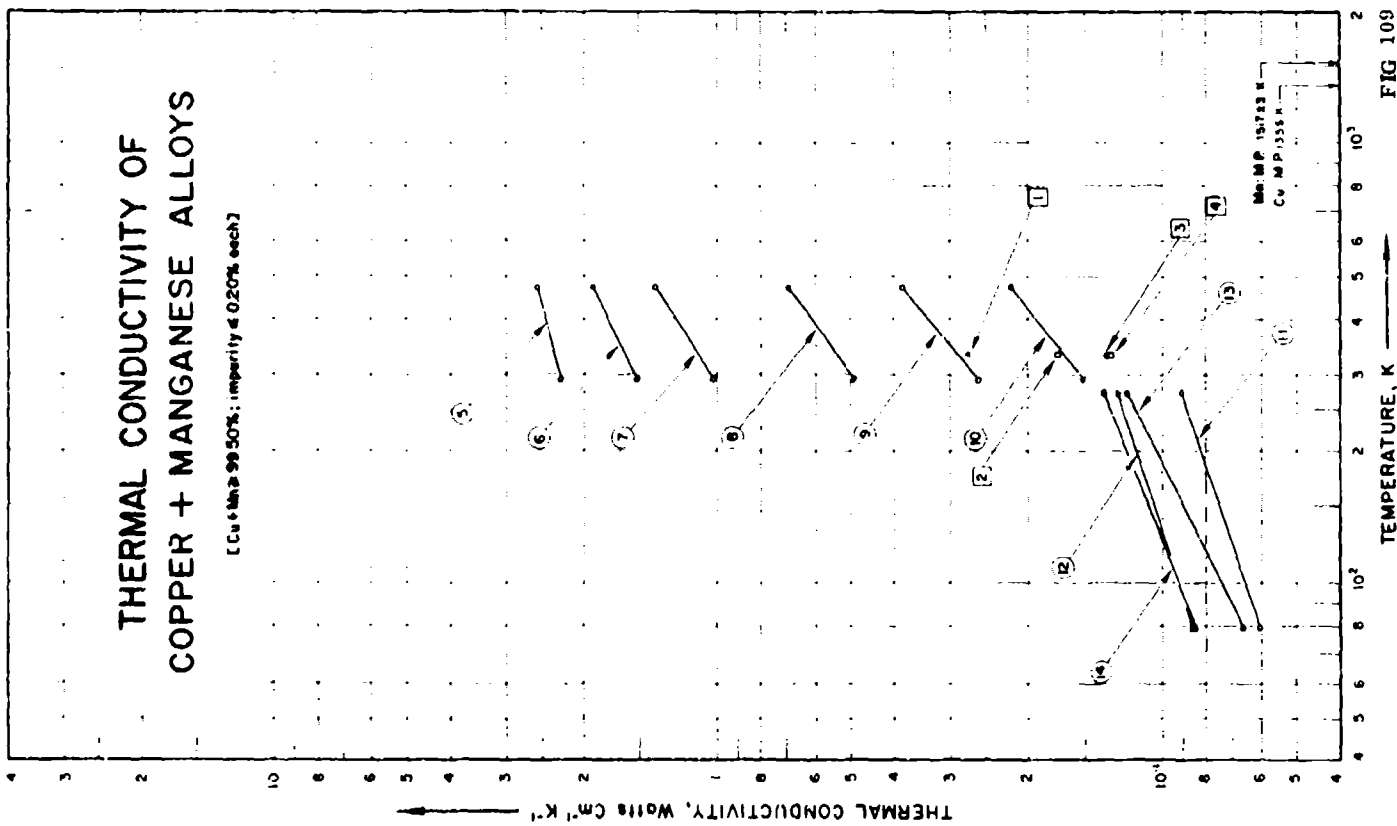


FIG 109

## SPECIFICATION TABLE NO. 109 THERMAL CONDUCTIVITY OF [COPPER + MANGANESE] ALLOYS

(Cu + Mn > 99.50%; impurity  $\leq$  0.20% each)

[ For Data Reported in Figure and Table No. 109 ]

Curve No.	Ref. No. Used	Method	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)		Composition (continued), Specifications and Remarks
							Cu	Mn	
1	230	L	1925	332.2			90.0	10.0	Approx. composition; specimen ~5 cm long with cross section 0.3 cm <sup>2</sup> ; made from Cu (<0.03 of total impurity) supplied by Baker; fused with Mn supplied by Elmer and Amend; electrical conductivity $2.76 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 23 C.
2	230	L	1925	332.2			80	20	Similar to the above specimen except electrical conductivity $1.59 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 23 C.
3	230	L	1925	332.2			70	30	Similar to the above specimen except electrical conductivity $1.11 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 23 C.
4	230	L	1925	332.2			60	40	Similar to the above specimen except electrical conductivity $0.916 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 23 C.
5	135	L	1935	293.473		Bar 116	99.55	0.43	0.01 Fe, 0.01 Mg; specimen 0.75 in. in diameter and 8 in. long; supplied by American Brass Co.; annealed at 700 C for 2 hrs.; electrical conductivity 29.56 and $22.00 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 20 and 200 C respectively.
6	135	L	1935	293.473		Bar 117	99.05	1.05	0.01 Fe, 0.01 Mg; similar to the above specimen except electrical conductivity 19.07 and $15.84 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 20 and 200 C respectively.
7	135	L	1935	293.473		Bar 118	98.27	1.77	0.03 Fe, 0.01 Mg; similar to the above specimen except electrical conductivity 12.54 and $11.22 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 20 and 200 C respectively.
8	135	L	1935	293.473		Bar 119	95.34	4.55	0.06 Fe, 0.02 Mg; similar to the above specimen except electrical conductivity 5.567 and $5.446 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 20 and 200 C respectively.
9	135	L	1935	293.473		Bar 120	90.25	9.53	0.18 Fe, 0.02 Mg, 0.021 C; similar to the above specimen except electrical conductivity 2.814 and $2.829 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 20 and 200 C respectively.
10	135	L	1935	293.473		Bar 121	80.03	19.82	0.09 Fe, 0.02 Mg, 0.035 C; similar to the above specimen except electrical conductivity 1.453 and $1.474 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 20 and 200 C respectively.
11	34		1927	80.273		1	70	30	Approx. composition; coarse grain; electrical conductivity 0.784 and $0.743 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 80 and 273 K respectively.

SPECIFICATION TABLE NO. 109 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (Cu)	Composition (weight percent) Mn	Composition (continued), Specifications and Remarks
12	34		1927	80, 273		2	70	30	Approx. composition; medium grain; electrical conductivity $1.09$ and $1.054 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 80 and 273 K respectively.
13	34		1927	80, 273		3	70	30	Approx. composition; medium grain; electrical conductivity $0.902$ and $0.863 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 80 and 273 K respectively.
14	34		1927	80, 273		4	70	30	Approx. composition; fine grain; electrical conductivity $1.034$ and $0.953 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 80 and 200 C respectively.

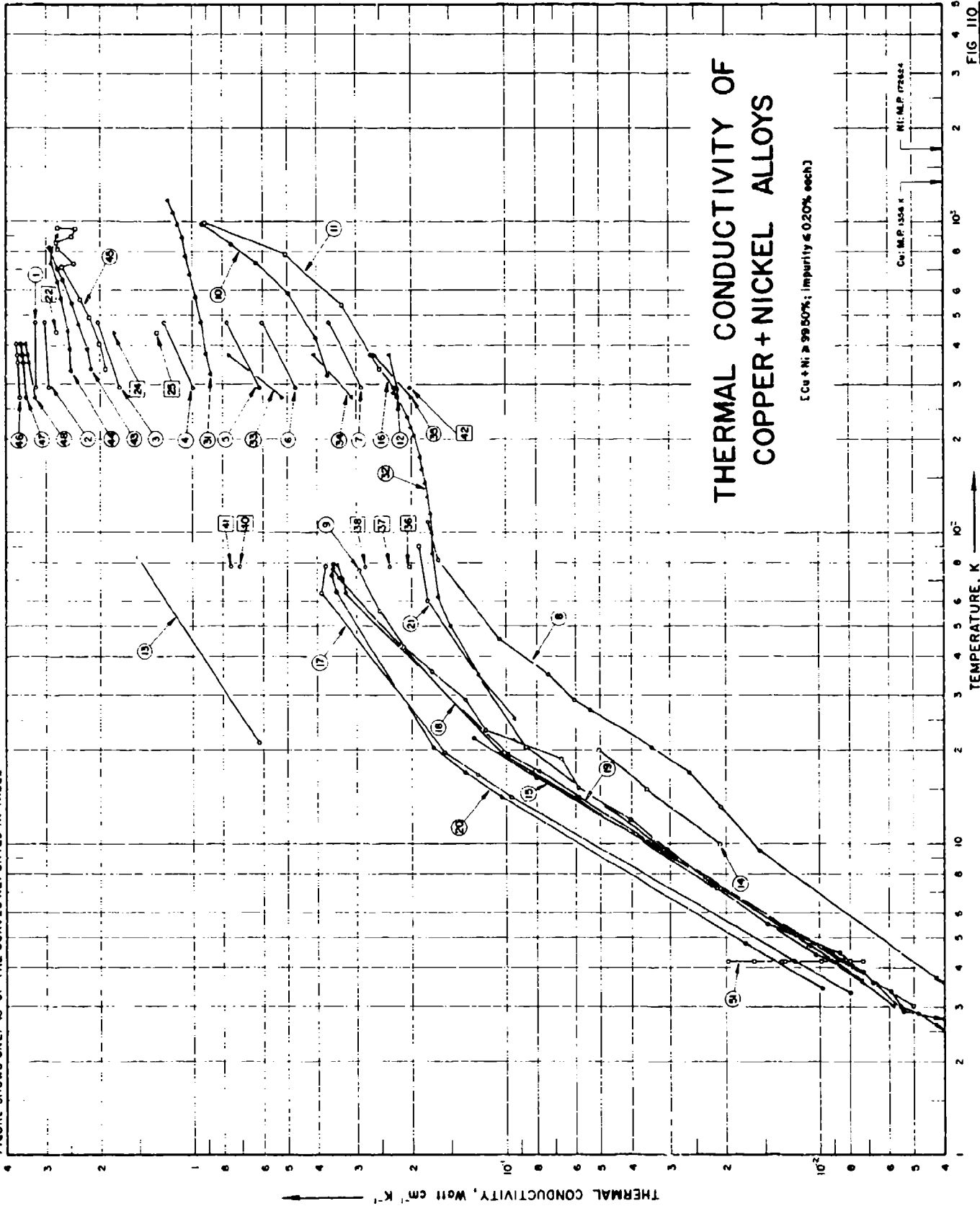
DATA TABLE NO. 109 THERMAL CONDUCTIVITY OF [COPPER + MANGANESE] ALLOYS

(Cu + Mn  $\geq$  99.50%; Impurity  $\leq$  0.20% each)[ Temperature, T, K; Thermal Conductivity, k, Watts  $\text{cm}^{-1} \text{K}^{-1}$  ]

T	k	T	k
<u>CURVE 1</u>			
332.2	0.272	293.2	0.259
		473.2	0.381
<u>CURVE 2</u>			
332.2	0.172	<u>CURVE 10</u>	
		293.2	0.151
		473.2	0.218
<u>CURVE 3</u>			
332.2	0.134	<u>CURVE 11</u>	
		80	0.0605
		273	0.0905
<u>CURVE 4</u>			
332.2	0.130	<u>CURVE 12</u>	
		80	0.0856
		273	0.126
<u>CURVE 5</u>			
293.2	2.259	<u>CURVE 13</u>	
473.2	2.561	80	0.066
		273	0.120
<u>CURVE 6</u>			
293.2	1.502	<u>CURVE 14</u>	
473.2	1.900	80	0.0849
		273	0.136
<u>CURVE 7</u>			
293.2	1.021		
473.2	1.389		
<u>CURVE 8</u>			
293.2	0.490		
473.2	0.690		



FIGURE SHOWS ONLY 43 OF THE CURVES REPORTED IN TABLE



## SPECIFICATION TABLE NO. 110 THERMAL CONDUCTIVITY OF COPPER-NICKEL ALLOYS

(Cu - Ni - 99.50%; impurity - 0.20% each)

(For Data Reported in Figure and Table No. 110)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) Cu	Composition (weight percent) Ni	Composition (continued), Specifications and Remarks
1	135	L	1935	293-473		Bar 107	99.73	0.27	0.11 Fe, 0.03 Mg; specimen 0.75 in. in diameter and 8 in. long; supplied by American Brass Co.; annealed at 800 C for 2 hrs.; electrical conductivity $45.76 \text{ and } 29.11 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 20 C and 200 C respectively.
2	135	L	1935	293-473		Bar 108	99.47	0.54	0.02 Fe, 0.04 Mg; similar to the above specimen except annealed at 800 C for 3 hrs.; electrical conductivity $39.94 \text{ and } 25.86 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 20 and 200 C respectively.
3	135	L	1935	293-473		Bar 109	97.94	1.97	0.02 Fe, 0.04 Mg; similar to the above specimen except annealed at 800 C for 4 hrs.; electrical conductivity $22.71 \text{ and } 17.58 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 20 and 200 C respectively.
4	135	L	1935	293-473		Bar 110	94.92	5.09	0.01 Fe, 0.03 Mg; similar to the above specimen except electrical conductivity $12.39 \text{ and } 10.54 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 20 and 200 C respectively.
5	135	L	1935	293-473		Bar 111	85.90	10.07	0.02 Fe, 0.03 Mg, 0.024 C; similar to the above specimen except electrical conductivity $7.07 \text{ and } 6.46 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 20 and 200 C respectively.
6	135	L	1935	293-473		Bar 125	84.83	15.07	0.05 Fe, 0.01 Mg, 0.03 Mn; similar to the above specimen except electrical conductivity $5.094 \text{ and } 4.755 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 20 and 200 C respectively.
7	135	L	1935	293-473		Bar 124	69.51	30.23	0.05 Fe, 0.05 Mg, 0.13 Mn; similar to the above specimen except electrical conductivity $2.754 \text{ and } 2.730 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 20 and 200 C respectively.
8	154	L	1956	233-108		Russian cupro nickel NI-81.7	81.0	19.0	Specimen in strip form cut from a 6 x 5 mm tube; measured in helium, unannealed.
9	154	L	1956	233-76		Russian cupro nickel NI-81.6	81.0	19.0	The above specimen annealed at 800 C; measured in helium.
10	124	P	1940	324-584			79.8	20.0	0.2 Mn; face Mg; specimen 0.25 cm in diameter and 35 cm long; chill cast, hot rolled and cold drawn then annealed at 700 C for 12 hrs.; electrical conductivity $3.54, 3.46, 3.33, 3.21, 3.12 \text{ and } 3.02 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 48, 130, 315, 462, 575 and 711 C respectively.
11	124	P	1940	335-990			59.8	40.0	Similar to the above specimen except electrical conductivity $1.99, 1.96 \text{ and } 1.92 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 62, 266, 510 and 717 C respectively.

SPECIFICATION TABLE NO. 110 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) Cu	Composition (weight percent) Ni	Composition (continued), Specifications and Remarks
12	8	F	1914	273-373		Eureka	60.0	40.0	Approx. composition; specimen 6 mm in diameter and 9 mm long.
13	57	L	1927	21, 83			99.0	1.0	Specimen 7 cm long and 0.1 to 0.3 cm wide, drawn; electrical resistivity 2.97, 1.60 and 1.295 $\mu$ ohm cm at 0, -150 and -252 C respectively.
14	152	L	1949	10-20			70.0	30.0	Approx. composition; specimen 4.1 mm in diameter and 21 mm long, supplied by Yorkshire Copper Works Ltd., cold-worked.
15	75	L	1951	1, 9-22			80.0	20.0	Approx. composition; average grain size 0.011 mm.
16	77	E	1900	291, 373			60.0	40.0	Density $8.92 \text{ g cm}^{-3}$ at 18 C; electrical conductivity 2.04 and $2.037 \times 10^8 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 18 and 100 C respectively.
17	155	L	1951	3, 3-78		CN 1	90.0	10.0	Specimen 1/8 in. in diameter, machined from an annealed bar; electrical resistivity 12.5, 12.72 and 14.68 $\mu$ ohm cm at 19, 7, 78, 9 and 286 K respectively.
18	155	L	1951	3, 0-76		CN 2	90.0	10.0	Specimen 1/8 in. in diameter, cold-worked by rolling from 0.25 in. thick to 0.14 in. before being machined to size; electrical resistivity 12.65 and 14.69 $\mu$ ohm cm at 76.2 and 296 K respectively.
19	155	L	1951	3, 6-79		CN 3	90.0	10.0	Specimen 1/8 in. in diameter, severely cold-worked, rolled from 0.5 in. cross section to 0.22 x 0.24 in. before machining; electrical resistivity 12.63 and 14.65 $\mu$ ohm cm at 78, 7 and 298 K respectively.
20	155	L	1951	3, 4-79		CN 4	90.0	10.0	Single crystal; specimen 1/8 in. in diameter; electrical resistivity 13.0, 13.10 and 15.04 $\mu$ ohm cm at 29.5, 79, 3 and 298 K respectively.
21	9	L	1951	3, 0-91			60.0	40.0	36 gauge wire bound and soldered together.
22	67	L	1932	438, 2			99.22	0.78	High grade electrolytic Cu with traces of impurities; specimen 6.5 in. long and 0.5 in. in diameter; annealed at 900 C.
23	67	L	1932	438, 2			98.43	1.57	Similar to the above specimen.
24	67	L	1932	438, 2			97.24	2.76	Similar to the above specimen.
25	67	L	1932	438, 2			95.1	4.9	Similar to the above specimen.

SPECIFICATION TABLE NO. 110 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) Cu Ni	Composition (continued), Specifications and Remarks
26	230	L	1925	330.2			50 50	Approx composition; specimen ~5 cm long with cross section 0.3 cm <sup>2</sup> ; made from Cu K 0.03 of total impurity supplied by Baker, fused with Ni (99.75 to 99.95 pure including cobalt) supplied by International Nickel Co. of America; electrical conductivity $1.98 \times 10^6 \text{ ohm}^{-1}\text{cm}^{-1}$ at 25 C.
27	230	L	1925	330.2			60.0 40.0	Approx composition; similar to the above specimen except electrical conductivity $2.04 \times 10^6 \text{ ohm}^{-1}\text{cm}^{-1}$ at 25 C.
28	230	L	1925	330.2			70 30	Similar to the above specimen except electrical conductivity $2.48 \times 10^6 \text{ ohm}^{-1}\text{cm}^{-1}$ at 25 C.
29	230	L	1925	330.2			90 10	Similar to the above specimen except electrical conductivity $3.49 \times 10^6 \text{ ohm}^{-1}\text{cm}^{-1}$ at 25 C.
30	186	P	1928	305.2		<b>Advance</b>	45.0 45.0	Approx composition; impurity < 0.03.
31	131	C	1953	323-1173	± 2.0	Lohm	93.4 6.05	0.01 Mn, 0.01 Si; annealed at 900 C; lead used as comparative material.
32	219	L	1951	26-295	< 2.3	Constantan	55.0 45.0	Calculated composition; specimen rolled and drawn to 1 mm thick; heated 0.5 hr close to melting point; electrical conductivity $6.2$ and $6.1 \times 10^4 \text{ ohm}^{-1}\text{cm}^{-1}$ at 0 and 100 C, respectively.
33	245	T	1919	273, 373			49.94 10.06	
34	246	T	1919	273, 373			71.00 20.10	Similar to the above specimen except electrical conductivity $3.5$ and $3.3 \times 10^4 \text{ ohm}^{-1}\text{cm}^{-1}$ at 0 and 100 C, respectively.
35	246	T	1919	273, 373			60.0 39.98	Similar to the above specimen except electrical conductivity $2.0$ and $2.0 \times 10^4 \text{ ohm}^{-1}\text{cm}^{-1}$ at 0 and 100 C, respectively.
36	433	L	1940	78.2		S	69.70.05 29.89	Cu-Ni alloy containing 0.03 Mn, 0.03 Fe, and traces of other impurities, made from electrolytic Ni (containing 0.53 Co, 0.05 Fe, 0.02 Al) and electrolytic Cu (containing 0.015 Sb, 0.01 Fe, 0.007 S and trace of P); specimen 4.0 mm in dia and 6 mm long; electrical resistivity $40.3 \mu\text{ohm cm}$ at -195 C.
37	433	L	1940	78.2		S	69.80.11 29.83	The alloy containing 0.04 Mn, 0.02 Fe and traces of other impurities made from the same materials as above; electrical resistivity $27.1 \mu\text{ohm cm}$ at -195 C.

SPECIFICATION TABLE NO. 110 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) Cu	Composition (weight percent) Ni	Composition (continued), Specifications and Remarks
38	433	L	1940	78.2		10	86.05	13.24	0.11 Fe and trace Mn; made from the same material as above; electrical resistivity $17.6 \mu\text{ohm cm}$ at $-195^\circ\text{C}$ .
39	433	L	1940	78.2		11	90.39	9.47	0.14 Fe, traces of Mn and other impurities; made from the same materials as above; electrical resistivity $11.9 \mu\text{ohm cm}$ at $-195^\circ\text{C}$ .
40	433	L	1940	78.2		12	96.24	3.67	0.09 Fe and traces of other impurities; as above but electrical resistivity $3.43 \mu\text{ohm cm}$ at $-195^\circ\text{C}$ .
41	433	L	1940	78.2		13	98.94	1.03	0.03 Fe and traces of other impurities; as above but electrical resistivity $1.039 \mu\text{ohm cm}$ at $-195^\circ\text{C}$ .
42	435	L	1900	291.2			54.0	46.6	Approx. composition: density $8.89 \text{ g cm}^{-3}$ ; electrical conductivity $1.99 \times 10^5 \text{ ohm}^{-1} \text{ cm}^{-1}$ at $18^\circ\text{C}$ .
43	377		1957	316-825			91.05	0.50	0.1 Be, 0.15 Co; electrical conductivity $25.8, 23.1, 26.4, 18.25, 16.5, 15.67$ and $14.33 \times 10^5 \text{ ohm}^{-1} \text{ cm}^{-1}$ at $63.0, 114.6, 195, 273, 373, 483.5$ and $551.3^\circ\text{C}$ respectively.
44	378		1957	333-990			94.90	0.90	0.10 Be, 0.10 Zr; electrical resistivity $3.34, 3.85, 4.33, 5.21, 5.78, 6.33, 7.05$ and $8.14 \mu\text{ohm cm}$ at $59.4, 105.6, 171.5, 291.6, 365.6, 457, 534.5$ and $626.5^\circ\text{C}$ respectively.
45	378		1957	336-947			99.0	0.80	0.20 Fe; electrical resistivity $4.25, 4.88, 5.56, 6.01, 6.40, 6.57, 7.18, 7.28, 7.66, 8.93$ and $9.78 \mu\text{ohm cm}$ at $62.8, 130.9, 217.5, 290.6, 462.5, 440.3, 540.3, 536.3, 574.3, 618.0$ and $673.6^\circ\text{C}$ respectively.
46	271	L	1938	273-403	0.11	3	99.717	0.204	0.079 O; specimen 50.6 cm long.
47	271	L	1938	273-403	0.11	4	99.618	0.303	0.079 O; specimen 50.6 cm long.
48	271	L	1938	273-403	0.11	5	99.413	0.508	0.079 O; specimen 50.6 cm long.
49	271	L	1938	273-403	0.11	5	99.961	0.303	0.0042 Fe, 0.0014 Pb, trace Sn and Zn; specimen 50.6 cm long.
50	271	L	1938	273-403	0.11	10	99.47	0.508	0.022 O; specimen 50.6 cm long.
51	612	F	1962	4.2			60.0	40.0	Approx. composition; specimen about 1 to 3 mm in diameter and $\sim 100 \text{ mm}$ long; measured in different strain conditions.



DATA TABLE NO. 110 (continued)

T	K	ε	T	K	ε
<u>CURVE 41</u>					
332.6	2.46		273.2	3.456	
388.8	2.48		353.2	3.520	
444.7	2.52		373.2	3.535	
564.8	2.66		403.2	3.560	
638.8	2.73				
730.2	2.86				
807.7	2.86				
893.7	2.74				
<u>CURVE 45</u>					
336.0	1.91		273.2	3.318	
404.1	1.99		353.2	3.380	
490.7	2.16		373.2	3.406	
563.8	2.31		403.2	3.431	
735.7	2.42				
713.5	2.66				
833.5	2.75				
811.5	2.73				
947.5	2.73				
891.2	2.45				
946.8	2.38				
<u>CURVE 46</u>					
273.2	3.610				
353.2	3.657				
373.2	3.670				
403.2	3.686				
<u>CURVE 47</u>					
273.2	3.464				
353.2	3.527				
373.2	3.544				
403.2	3.563				
<u>CURVE 48</u>					
273.2	3.251				
353.2	3.337				
373.2	3.435				
403.2	3.490				
<u>CURVE 49<sup>a</sup></u>					
273.2	3.456				
353.2	3.520				
373.2	3.535				
403.2	3.560				
<u>CURVE 50<sup>a</sup></u>					
273.2	3.318				
353.2	3.380				
373.2	3.406				
403.2	3.431				
<u>CURVE 51</u>					
(T = 4.2°K)					
Strain K(Watt cm <sup>-1</sup> K <sup>-1</sup> )					
0	0.0197				
0.069	0.0165				
0.019	0.0121				
0.024	0.0130				
0.023	0.0099				
0.125	0.008				
0.193	0.0073				

<sup>a</sup> Not shown on plot

# THERMAL CONDUCTIVITY OF COPPER + PALLADIUM ALLOYS

(Cu + Pd ≥ 99.50%; impurity ≤ 0.20% each)

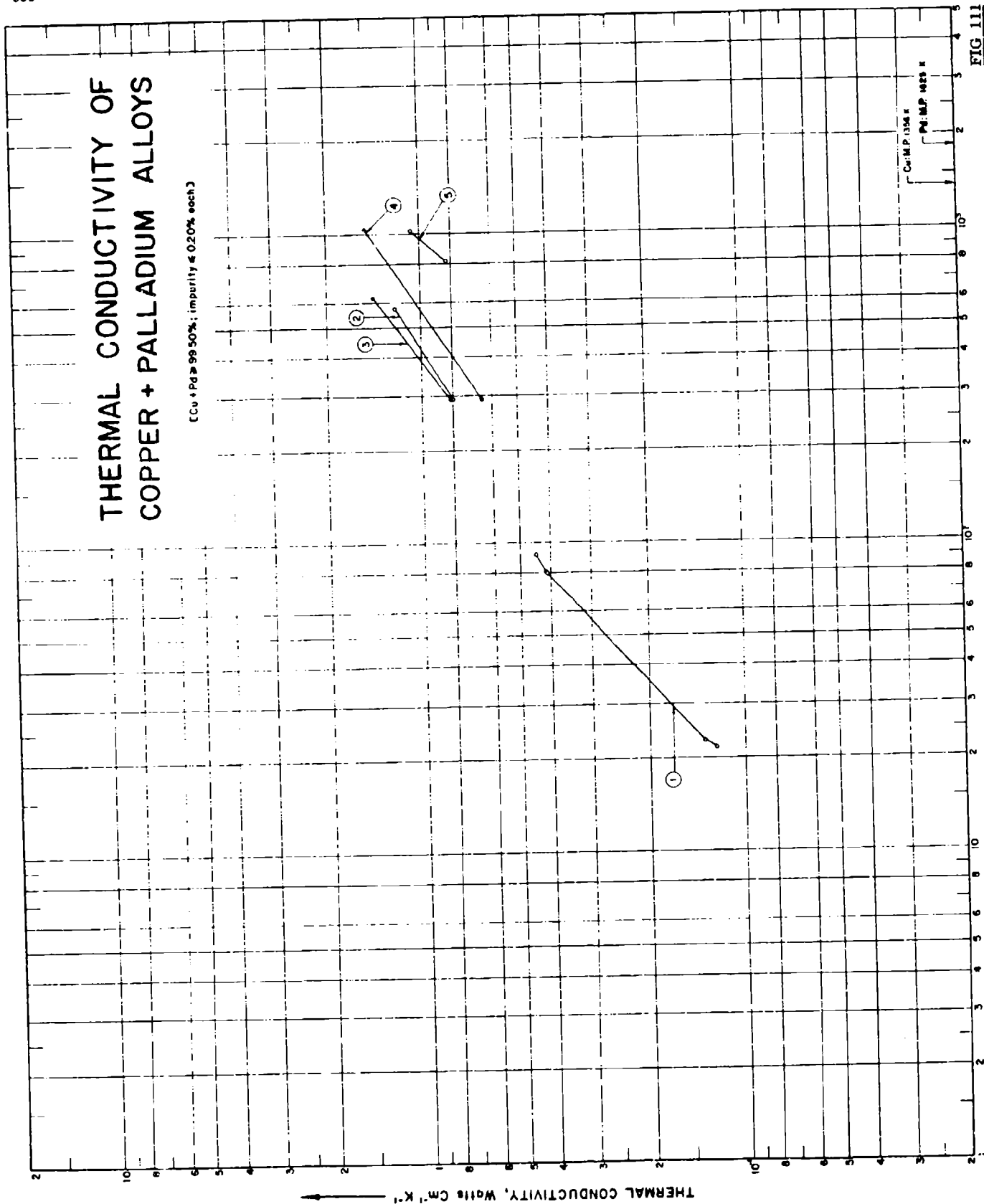


FIG. 111

TEMPERATURE, K →

← THERMAL CONDUCTIVITY, Watts Cm<sup>-1</sup> K<sup>-1</sup>

Cu: M.P. 1356 K  
Pd: M.P. 1825 K



SPECIFICATION TABLE NO. 111 THERMAL CONDUCTIVITY OF COPPER-PALLADIUM ALLOYS

(Cu-Pd 99.50% Impurity 0.20% each)

For Data Reported in Figure and Table No. 111

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) Cu	Pd	Composition (continued), Specifications and Remarks
1	58	L	1914	22-91			80.7	19.3	Calculated composition, polycrystal, unannealed; <del>electrical resistivity, 5.98 and 5.184 <math>\mu</math> ohm cm at 0-140 and -251 C, respectively. 6.82,</del>
2	391	L	1958	293, 577	3.0		75.82	24.18	Calculated composition, annealed at 600 to 700 C for 2 hrs. with ordered atomic arrangement.
3	391	L	1958	293, 623	3.0		64.18	35.82	Similar to the above specimen.
4	391	L	1958	293, 1048	3.0		75.82	24.18	Similar to the above specimen.
5	391	L	1958	618, 1028	3.0		64.18	35.82	Similar to the above specimen.

## DATA TABLE NO. III THERMAL CONDUCTIVITY OF [COPPER + PALLADIUM] ALLOYS

(Cu + Pd : 99.50%, impurity  $\leq$  0.20% each)[ Temperature, T, K. Thermal Conductivity, k, Watt cm<sup>-1</sup> K<sup>-1</sup> ]

T	k
<u>CURVE 1</u>	
21.6	0.124
22.7	0.135
79.9	0.411
80.3	0.415
91.1	0.448
<u>CURVE 2</u>	
293.2	0.80
573.2	1.22
<u>CURVE 3</u>	
293.2	0.815
623.2	1.41
<u>CURVE 4</u>	
293.2	0.55
1049.2	1.50
<u>CURVE 5</u>	
818.2	0.83
1928.2	1.07

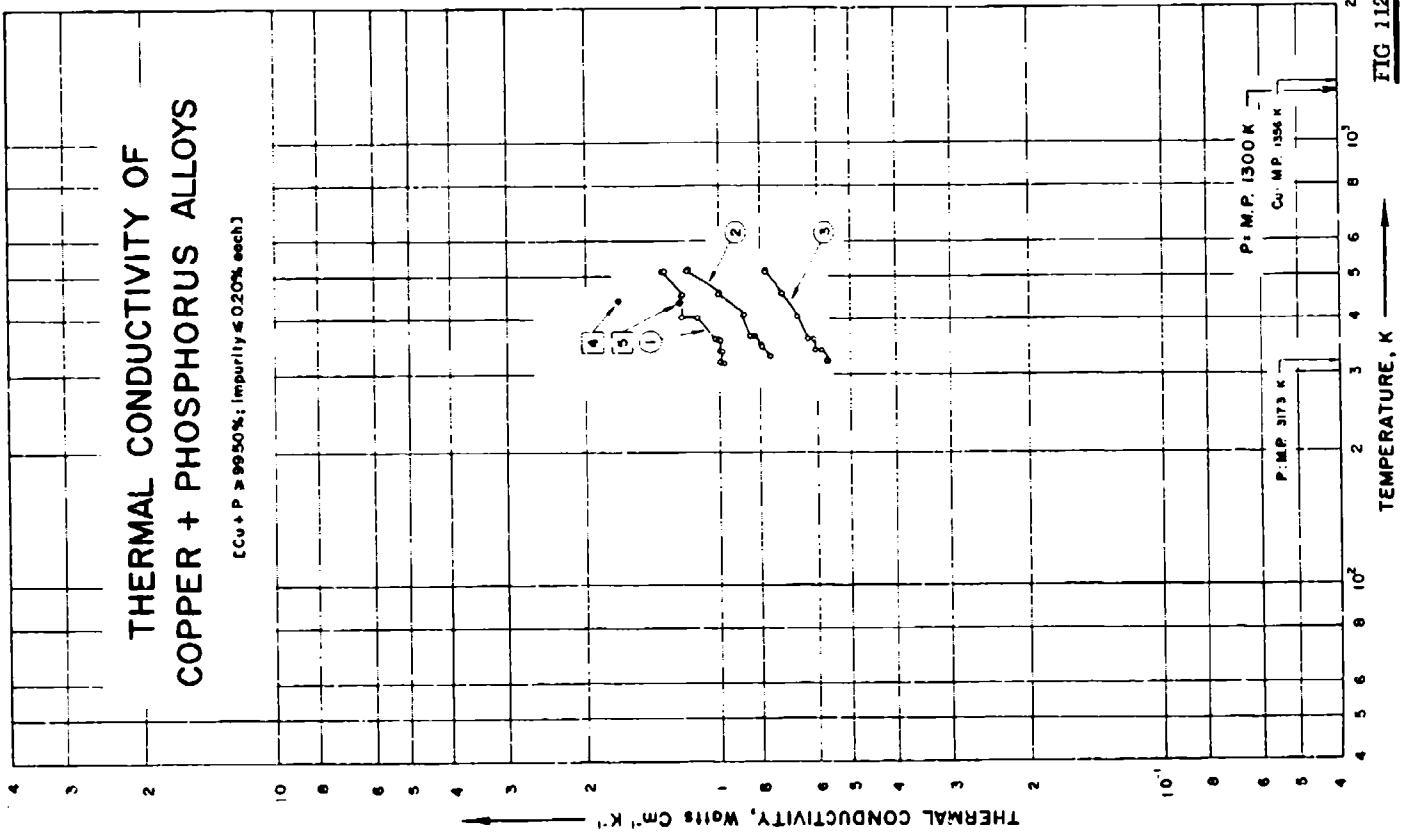


FIG 112

## SPECIFICATION TABLE NO. 112 THERMAL CONDUCTIVITY OF (COPPER + PHOSPHORUS) ALLOYS

(Cu + P = 99.50%; impurity = 0.20% each)

[For Data Reported in Figure and Table No. 112.]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) Cu	P	Composition (continued), Specifications and Remarks
1	134	L	1931	318-512	<2.0	83	99.53	0.460	0.05 Fe; specimen 7/8 in. in diameter; cast, cold-rolled and annealed at 650 C for 1 hr. before machining; electrical conductivity $11.842 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 20 C.
2	134	L	1931	329-516	<2.0	94	99.31	0.677	0.02 Fe; similar to the above specimen except electrical conductivity $8.614 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 20 C.
3	134	L	1931	322-516	<2.0	84	99.12	0.330	0.06 Fe; similar to the above specimen except electrical conductivity $6.5488 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 20 C.
4	67	L	1932	438.2			99.771	0.229	High grade electrolytic Cu with traces of impurities; specimen 6.5 in. long and 0.5 in. in diameter; cast and machined; thermal conductivity data obtained from the mean value of 16 readings.
5	67	L	1932	438.2			99.594	0.406	Similar to the above specimen.

DATA TABLE NO. 112 THERMAL CONDUCTIVITY OF [COPPER + PHOSPHORUS] ALLOYS

(Cu + P - 99.50%, Impurity - 0.20% each)

Temperature, T, K. Thermal Conductivity, k, Watt cm<sup>-1</sup> K<sup>-1</sup>

T	k	T	k
<u>CURVE 1</u>			
318.2	0.971	402.2	0.669
320.2	0.996	454.2	0.720
337.2	0.983	455.2	0.720
338.2	0.987	515.2	0.782
339.2	1.004	516.2	0.782
357.2	0.992	<u>CURVE 4</u>	
358.2	1.004		
360.2	1.017	438	1.665
401.2	1.130	<u>CURVE 5</u>	
404.2	1.121		
452.2	1.218	438	1.222
454.2	1.205		
511.2	1.326		
512.2	1.305		

T	k
<u>CURVE 2</u>	
329.2	0.753
329.2	0.757
346.2	0.791
346.2	0.795
347.2	0.799
366.2	0.828
366.2	0.833
366.2	0.841
408.2	0.874
409.2	0.870
454.2	1.004
455.2	1.000
515.2	1.180
516.2	1.176

T	k
<u>CURVE 3</u>	
322.2	0.569
322.2	0.561
340.2	0.586
340.2	0.603
361.2	0.611
361.2	0.636
401.2	0.669

SPECIFICATION TABLE NO. 113 THERMAL CONDUCTIVITY OF [COPPER + PLATINUM] ALLOYS

(Cu + Pt > 99.50%; impurity  $\leq$  0.20% each)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)		Composition (continued), Specifications and Remarks
							Cu	Pt	
1	232	L	1957	465.2			56.57	43.43	Calculated composition; cast.
2	232	L	1957	483.2			56.57	43.43	Calculated composition; after 10 hrs annealing.
3	232	L	1957	481.7			56.57	43.43	Calculated composition; after 20 hrs annealing.
4	232	L	1957	482.7			56.57	43.43	Calculated composition; after 40 hrs annealing.
5	232	L	1957	500.7			56.57	43.43	Calculated composition; after 50 hrs annealing.

DATA TABLE NO. 113 THERMAL CONDUCTIVITY OF [COPPER + PLATINUM] ALLOYS

(Cu + Pt > 99.50%; impurity  $\leq$  0.20% each)[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup>K<sup>-1</sup>]

T	k	T	k
<u>CURVE 1*</u>			
465.2	0.339	500.7	0.732
<u>CURVE 2*</u>			
483.2	0.523		
<u>CURVE 3*</u>			
481.7	0.565		
<u>CURVE 4*</u>			
482.7	0.644		

No graphical presentation

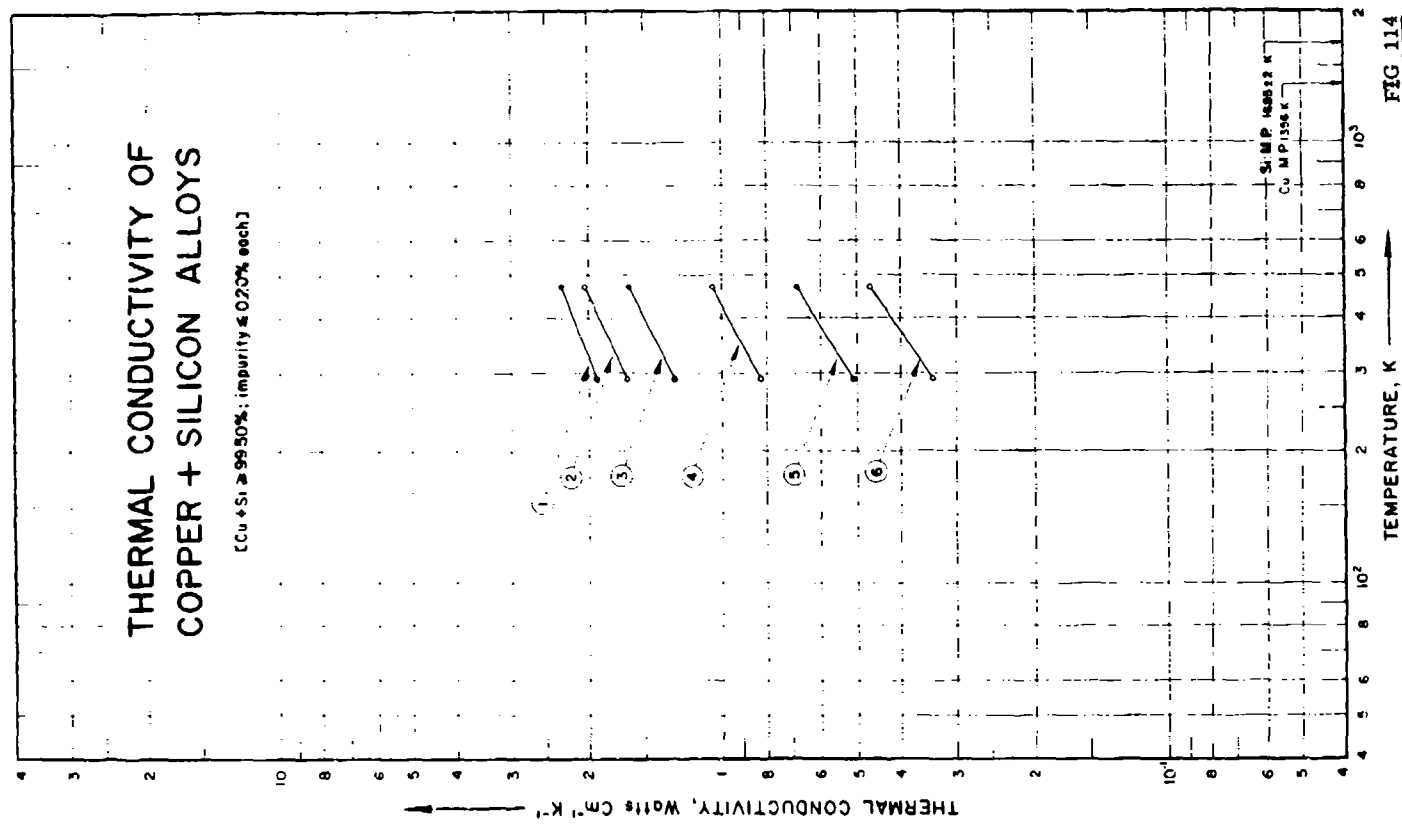


FIG. 114

## SPECIFICATION TABLE NO. 114 THERMAL CONDUCTIVITY OF [COPPER + SILICON] ALLOYS

(Cu + Si &gt; 99.50%; impurity &lt; 0.20% each)

[ For Data Reported in Figure and Table No. 114 ]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) Cu	Composition (weight percent) Si	Composition (continued), Specifications and Remarks
1	135	L	1935	293, 473		105	99.76	0.23	0.02 Fe; specimen 0.75 in. in diameter and 8 in. long; supplied by American Brass Co.; hot rolled, cast and cold drawn to 7/8 in. in diameter; annealed at 700 C for 2 hrs.; electrical conductivity 26.6 and 19.8 x 10 <sup>4</sup> ohm <sup>-1</sup> cm <sup>-1</sup> at 20 and 200 C respectively.
2	135	L	1935	293, 473		96	99.65	0.32	0.032 Fe, trace Pb; similar to the above specimen except annealed at 700 C for 1 hr.; electrical conductivity 21.71 and 16.96 x 10 <sup>4</sup> ohm <sup>-1</sup> cm <sup>-1</sup> at 20 and 200 C respectively.
3	135	L	1935	293, 473		106	99.53	0.45	0.03 Fe; similar to the above specimen except annealed at 700 C for 2 hrs.; electrical conductivity 17.19 and 13.96 x 10 <sup>4</sup> ohm <sup>-1</sup> cm <sup>-1</sup> at 20 and 200 C respectively.
4	135	L	1935	293, 473		78	99.06	1.00	0.03 Fe; similar to the above specimen except electrical conductivity 10.5 and 9.13 x 10 <sup>4</sup> ohm <sup>-1</sup> cm <sup>-1</sup> at 20 and 200 C respectively.
5	135	L	1935	293, 473		79	98.09	1.98	0.05 Fe; similar to the above specimen except electrical conductivity 6.235 and 5.659 x 10 <sup>4</sup> ohm <sup>-1</sup> cm <sup>-1</sup> at 20 and 200 C respectively.
6	135	L	1935	293, 473		80	96.0	3.91	0.02 Fe; similar to the above specimen except electrical conductivity 3.945 and 3.655 x 10 <sup>4</sup> ohm <sup>-1</sup> cm <sup>-1</sup> at 20 and 200 C respectively.



## DATA TABLE NO. 114 THERMAL CONDUCTIVITY OF (COPPER + SILICON) ALLOYS

(Cu + Si - 99.54%, Impurity - 0.21%, each)

(Temperature, T - K; Thermal Conductivity, k, Watt cm<sup>-1</sup>K<sup>-1</sup>)

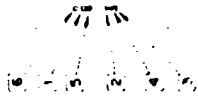
T	k
<u>CURVE 1</u>	
293.2	1.920
473.2	2.301
<u>CURVE 2</u>	
293.2	1.648
473.2	2.038
<u>CURVE 3</u>	
293.2	1.293
473.2	1.623
<u>CURVE 4</u>	
293.2	0.824
473.2	1.059
<u>CURVE 5</u>	
293.2	0.510
473.2	0.696
<u>CURVE 6</u>	
293.2	0.339
473.2	0.469

# THERMAL CONDUCTIVITY OF COPPER + SILVER ALLOYS

[ Cu + Ag in 99-50% ; impurity < 0.20% each ]

THERMAL CONDUCTIVITY,  $\text{Watt cm}^{-1} \text{K}^{-1}$

TEMPERATURE, K



Ag M.P. 1234.6 K  
Cu M.P. 1356.4 K

FIG. 115

SPECIFICATION TABLE NO. 115 THERMAL CONDUCTIVITY OF (COPPER-SILVER) ALLOYS

(Cu + Ag = 99.50% impurity = 0.20% each)

For Data Reported in Figure and Table No. 115\*

Curve No.	Rel. Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) Cu Ag	Composition (continued), Specifications and Remarks
1	230	1	1925	335.2		50 50	Approx. composition; specimen ~5 cm long with cross section 0.3 cm <sup>2</sup> ; made from Cu (< 0.03 of total impurity) supplied by Baker <sup>2</sup> ; fused with Ag (99.9 pure); electrical conductivity $4.36 \times 10^5 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 25 C.
2	230	1	1925	335.2		60 40	Similar to the above specimen except electrical conductivity $3.75 \times 10^5 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 25 C.
3	230	1	1925	335.2		70 30	Similar to the above specimen except electrical conductivity $3.59 \times 10^5 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 25 C.
4	230	1	1925	335.2		80 20	Similar to the above specimen except electrical conductivity $3.72 \times 10^5 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 25 C.
5	230	1	1925	335.2		90 10	Similar to the above specimen except electrical conductivity $4.2 \times 10^5 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 25 C.
6	230	1	1925	335.2	Over bronze	95 5	Similar to the above specimen except electrical conductivity $4.70 \times 10^5 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 25 C.
7	57	1	1927	21.83	Silver bronze: Cu 10	97.0 3.0	Specimen 7 cm long and 0.3 to 0.3 cm wide; unannealed; electrical resistivity 1.79, 0.429 and 0.130 $\mu \text{ ohm cm}$ at 0, -190 and -252 C respectively.
8	57	1	1927	21.83	Silver bronze: Cu 20a	97.0 3.0	Similar to the above specimen except annealed at 390 C for 3 hrs.; electrical resistivity 1.724, 0.362 and 0.0697 $\mu \text{ ohm cm}$ at 0, -190 and -252 C respectively.

## DATA TABLE NO. 115 THERMAL CONDUCTIVITY OF COPPER - SILVER ALLOYS

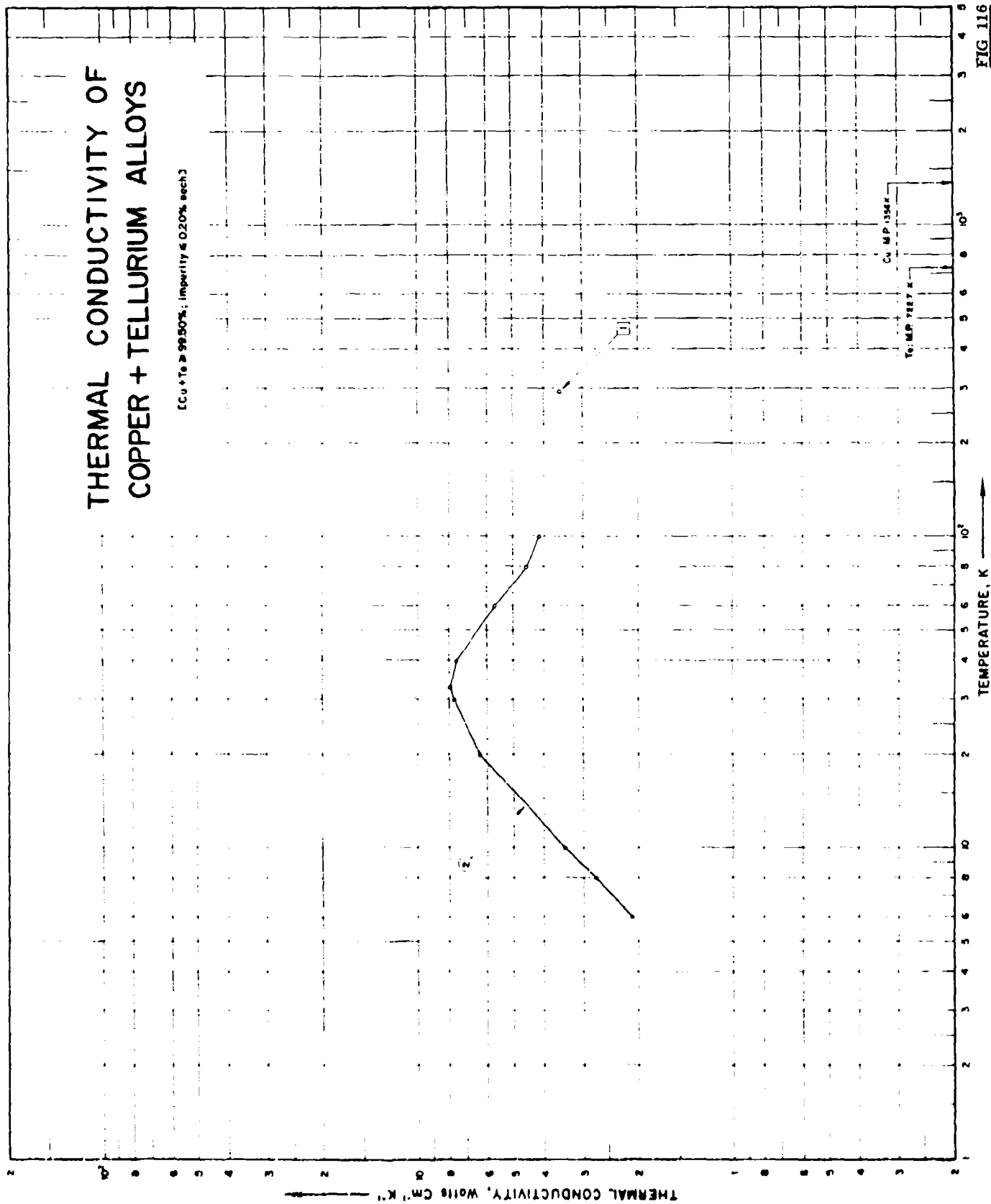
(Cu - Ag: 99.50%; impurity: 0.20%, each)

(Temperature: T, K; Thermal Conductivity,  $k$ , Watt cm<sup>-1</sup> K<sup>-1</sup>)

T	$k$
<u>CURVE 1</u>	
315.2	3.126
<u>CURVE 2</u>	
335.2	2.734
<u>CURVE 3</u>	
335.2	2.628
<u>CURVE 4</u>	
335.2	2.674
<u>CURVE 5</u>	
335.2	3.023
<u>CURVE 6</u>	
335.2	3.251
<u>CURVE 7</u>	
21.2	3.57
83.2	3.57
<u>CURVE 8</u>	
21.2	6.19
83.2	4.06

# THERMAL CONDUCTIVITY OF COPPER + TELLURIUM ALLOYS

{Cu + Te ≥ 99.50%; impurity ≤ 0.20% each}



## SPECIFICATION TABLE NO. 116 THERMAL CONDUCTIVITY OF (COPPER - TELLURUM) ALLOYS

(Cu + Te = 99.50%; impurity = 0.20% each)

[For Data Reported in Figure and Table No. 116.]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)		Composition (continued), Specifications and Remarks
							Cu	Te	
1	546		1960	293.2		ASTM B701-58T	99.5	0.5	No other details reported.
2	432	L	1957	6.0-100		Free-cutting (Te)Cu	99.424	0.56	0.007 P, 0.001 Fe, 0.001 Si, 0.001 Ag, 0.001 Zn, <0.001 Al, Pb, Mg, Mn, and Sn each; commercial hard tempered rod; specimen ground down from a 0.25 in. dia rod; average grain size 0.1 x 0.016 mm longitudinally and 0.016 x 0.016 mm transversely; density 8.909 g cm <sup>-3</sup> .

## DATA TABLE NO. 116 THERMAL CONDUCTIVITY OF COPPER-TELLURIUM ALLOYS

(Cu - Te) (50% compacts - 0.20" each)

Temperatures: T. K. Thermal Conductivity: k, Watt/cm<sup>2</sup>K<sup>2</sup>.

T	k
<u>CURVE 1</u>	
293.2	3.49
<u>CURVE 2</u>	
300	3.41
310	3.37
320	3.41
330	3.35
340	3.33
350	3.31
360	3.28
370	3.25
380	3.22
390	3.19
400	3.16

# THERMAL CONDUCTIVITY OF COPPER + TIN ALLOYS

(Cu + Sn = 99.50%; impurity < 0.20% each)

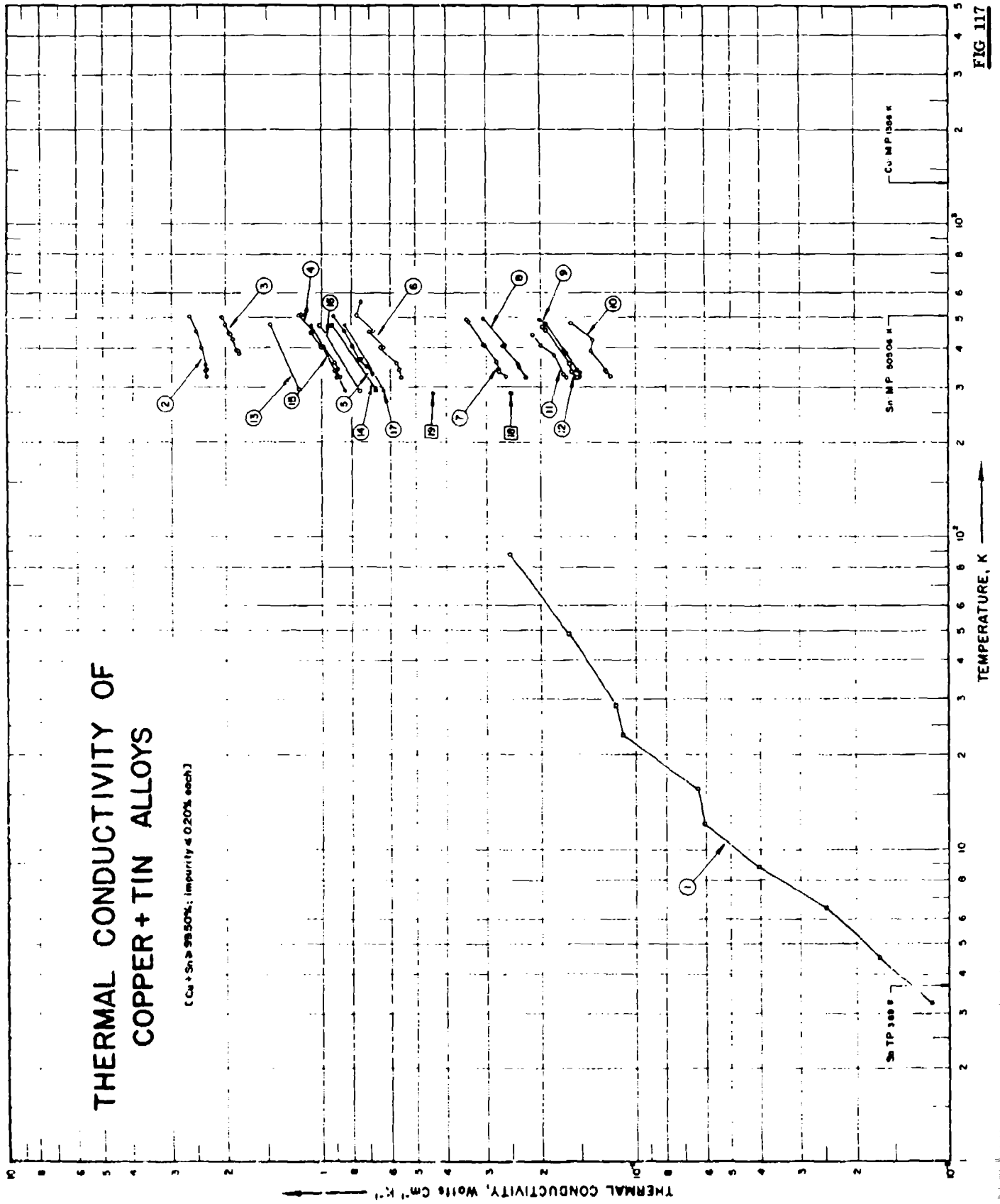


FIG 117



SPECIFICATION TABLE NO. 117 THERMAL CONDUCTIVITY OF COPPER-TIN ALLOYS

(Cu + Sn 99.5%; impurity 0.20% each)

For Data Reported in Figure and Table No. 117.

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Cu	Sn	Composition (weight percent)	Composition (continued), Specifications and Remarks
1	134	L	1956	323-588	5.0	Phosphor bronze 75	93.30	6.46	0.2 P, unannealed.	
2	134	L	1961	323-507	2.0	74	99.00	0.99	0.002 P, 0.01 Fe, trace Pb; cold-rolled to 7/8 in. in diameter and annealed at 630 C then slowly cooled; electrical conductivity $31.364 \times 10^9 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 20 C.	
3	134	L	1961	323-503	2.0	75	98.06	1.92	0.01 P, 0.01 Fe, trace Pb; specimen supplied by American Brass Co.; high grade commercial alloy, cast in a mold of 24/4 in. in diameter; cold-rolled and annealing to 7/8 in. in diameter at 630 C for 30 min then slowly cooled; electrical conductivity $21.154 \times 10^9 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 20 C.	
4	334	L	1931	323-511	2.0	41	94.96	4.92	0.06 P, 0.05 Fe, 0.61 Pb; similar to the above specimen except electrical conductivity $10.674 \times 10^9 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 20 C.	
5	374	L	1931	331-509		43	92.45	7.48	0.04 P, 0.02 Fe, 0.01 Pb; similar to the above specimen except electrical conductivity $7.945 \times 10^9 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 20 C.	
6	134	L	1931	323-562		44	89.51	10.40	0.03 P, 0.05 Fe, 0.01 Pb; similar to the above specimen except electrical conductivity $6.4362 \times 10^9 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 20 C.	
7	515	C	1952	323-492		Bronze: A <sub>1</sub>	82.0	11.0	Specimen mean thickness 0.982 cm and 8.87 cm long; prepared by powder metallurgy method from powder of mean particle size of 0.00133 cm diameter; supplied by Messrs. Sintered Products; density 6.45 g cm <sup>-3</sup> ; electrical resistivity 33.0 μohm cm at 20 C, porosity 25.8%.	
8	515	C	1952	323-493		Bronze: B <sub>1</sub>	89.0	11.0	Specimen mean thickness 0.889 cm and 8.89 cm long; prepared by powder metallurgy method; mean particle size 0.00493 cm; supplied by Messrs. Sintered Products; density 6.30 g cm <sup>-3</sup> ; electrical resistivity 38.5 μohm cm at 20 C; porosity 28.2%.	
9	515	C	1952	323-491		Bronze: P <sub>1</sub>	89.0	11.0	Specimen 0.85 cm <sup>2</sup> cross section and 7.6 cm long; prepared by powder metallurgy method; mean particle size 0.00493 cm; supplied by Messrs. Sintered Products; density 5.85 g cm <sup>-3</sup> ; electrical resistivity 37.1 μohm cm at 20 C; porosity 32.7%.	

SPECIFICATION TABLE NO. 117 (continued)

Curve No.	Ref. No.	Method Usec.	Year	Temp. Range, °	Reported Error, %	Name and Specimen Designation	Composition (weight percent)		Composition (continued), Specifications and Remarks
							Cu	Sn	
10	515	C	1952	323-469		Bronze; C <sub>1</sub>	89.0	11.0	Specimen mean thickness 0.948 cm and 8.88 cm long; prepared by powder metallurgy method; mean particle size; 0.01275 cm; supplied by Messrs. Sintered Products; density 5.55 g cm <sup>-3</sup> ; electrical resistivity 81.5 μohm cm at 20 C; porosity 36.2%.
11	515	C	1952	323-438		Bronze; D <sub>1</sub>	89.0	11.0	Specimen mean thickness 0.932 cm and 8.89 cm long; prepared by powder metallurgy method; mean particle size 0.02113 cm; supplied by Messrs. Sintered Products; density 5.75 g cm <sup>-3</sup> ; electrical resistivity 52.0 μohm cm at 20 C; porosity 34.5%.
12	515	C	1952	323-473		Bronze; E <sub>1</sub>	89.0	11.0	Specimen mean thickness 0.945 cm and 8.89 cm long; prepared by powder metallurgy method; mean particle size 0.04 cm; supplied by Messrs. Sintered Products; density 5.50 g cm <sup>-3</sup> ; electrical resistivity 60.1 μohm cm at 20 C; porosity 36.8%.
13	516	L	1941	293, 473		Phosphor bronze; 1	96.84	3.11	0.02 P, <0.01 Fe, <0.01 Ni, <0.005 Pb, <0.005 Sb; cast, after air cooling annealed at 625 C then cold-rolled and machined; electrical resistivity 6.37 μohm cm at 20 C.
14	516	L	1941	293, 473		Phosphor bronze; 3	92.6	7.31	0.02 P, <0.01 Fe, <0.005 Pb, <0.005 Sb; cast after air cooling, annealed at 625 C, hot-rolled at 300 C and annealed at 625 C for 2-1/2 hrs, again hot-rolled at 300 C and annealed at 625 C for 2-1/2 hrs then cold-rolled and machined; electrical resistivity 12.31 μohm cm at 20 C.
15	516	L	1941	293, 473		Phosphor bronze; 6	94.6	5.27	0.09 P, <0.001 Fe, <0.005 Pb, <0.005 Sb; similar to the above specimen except electrical resistivity 10.25 μohm cm at 20 C.
16	516	L	1941	293, 473		Phosphor bronze; 7	93.9	6.65	0.12 P, <0.01 Fe, <0.01 Zn, <0.005 Pb, <0.005 Sb; similar to the above specimen except electrical resistivity 12.83 μohm cm at 20 C.
17	516	L	1941	293, 473		Phosphor bronze; 5	96.16	3.71	0.12 P, <0.01 Fe, <0.01 Zn, <0.005 Pb, <0.005 Sb; cast, after air cooling; annealed at 625 C then cold-rolled and machined.
18	459	R	1905	286.7		Sn <sub>2</sub> Cu <sub>3</sub>	75.55	24.45	Cast and turned; density 8.89 g cm <sup>-3</sup> at 14 C.
19	459	R	1905	286.7		Sn <sub>10</sub> Cu <sub>9</sub>	90.1	9.9	Cast and turned; density 8.475 g cm <sup>-3</sup> at 14 C.

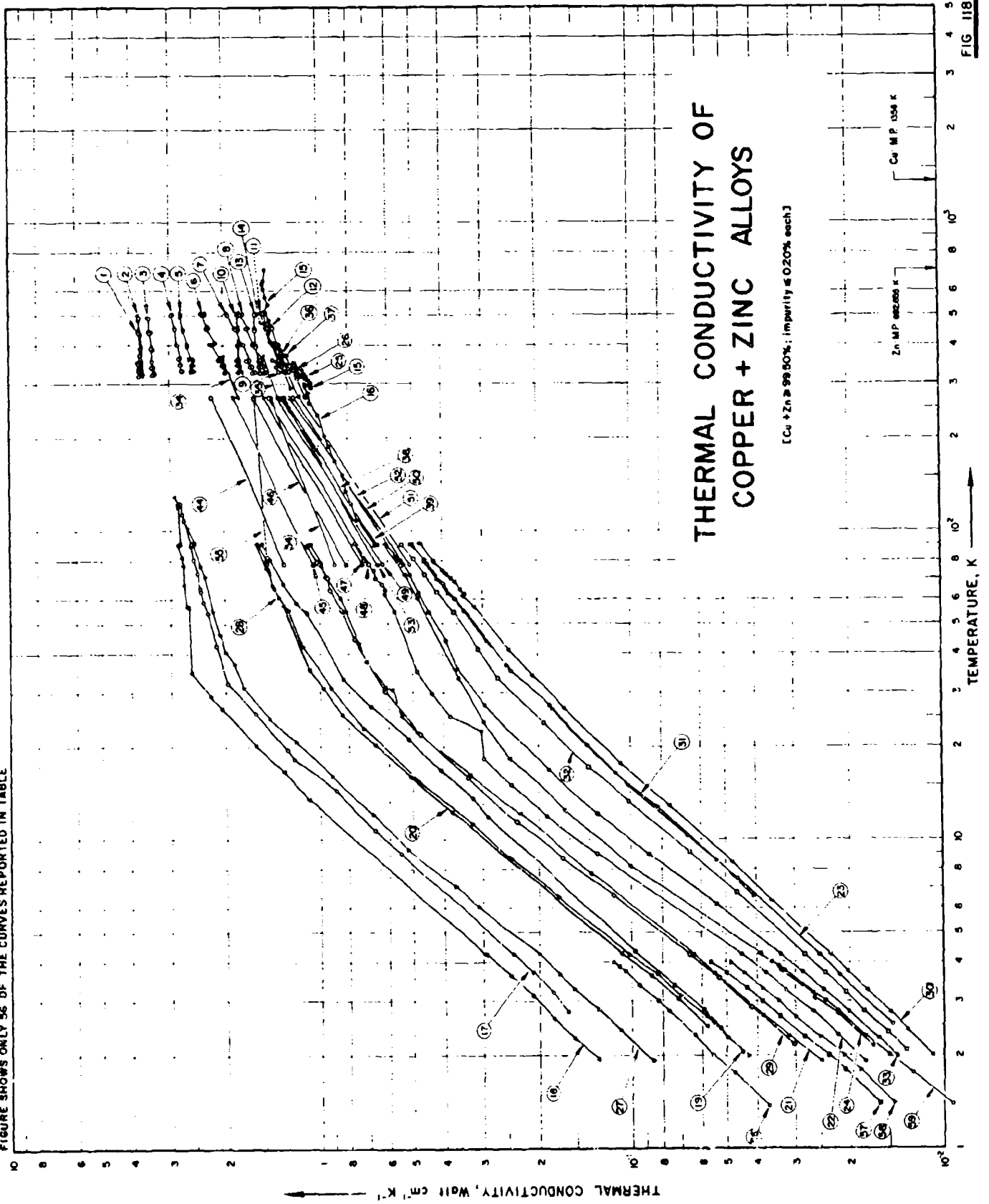
DATA TABLE NO. 117 THERMAL CONDUCTIVITY OF COPPER-TIN ALLOYS

Cu-30, 39, 50% impurity @ 0.20% each

[Temperature, T, K; Thermal Conductivity, k, W/m<sup>2</sup>·K<sup>-1</sup>]

T	k	T	k	T	k	T	k	T	k
<u>CURVE 1</u>									
323.2	0.0111	323.2	0.891	323.2	0.297	323.2	0.165	286.7	0.247
330.2	0.0167	331.2	0.970	333.7	0.270	328.2	0.167	<u>CURVE 10</u>	
337.2	0.0217	341.2	0.904	340.7	0.272	371.2	0.180	286.7	0.180
344.2	0.0301	342.2	0.883	360.2	0.276	403.2	0.199	<u>CURVE 11</u>	
351.2	0.0403	360.2	0.910	402.2	0.300	435.2	0.211	286.7	0.180
358.2	0.0536	402.2	1.00	407.2	0.305	<u>CURVE 12</u>			
365.2	0.070	403.2	0.987	431.2	0.317	323.2	0.151	335.7	0.157
372.2	0.089	432.2	1.08	432.2	0.342	335.7	0.161	455.7	0.193
379.2	0.116	453.2	1.06	<u>CURVE 8</u>				463.2	0.197
386.2	0.153	510.2	1.18	323.2	0.222	<u>CURVE 13</u>			
388.0	0.251	511.2	1.15	335.7	0.234	403.2	0.264	293.2	1.17
<u>CURVE 2</u>									
324.2	2.33	<u>CURVE 5</u>		403.2	0.762	<u>CURVE 14</u>			
324.2	2.34	331.2	0.678	403.2	0.762	<u>CURVE 15</u>			
330.2	2.34	332.2	0.686	403.2	0.716	293.2	0.669	473.2	1.16
330.2	2.35	350.2	0.716	403.2	0.762	<u>CURVE 16</u>			
336.2	2.36	369.2	0.762	<u>CURVE 9</u>					
336.2	2.36	370.2	0.749	323.2	0.149	333.2	0.149	293.2	0.920
340.2	2.44	405.2	0.795	333.2	0.149	339.2	0.153	473.2	0.920
401.2	2.43	453.2	0.858	350.2	0.849	385.2	0.165	<u>CURVE 17</u>	
430.2	2.51	454.2	0.849	454.2	0.849	390.7	0.167	293.2	0.628
431.2	2.53	455.2	0.849	509.2	0.916	476.2	0.192	473.2	1.08
505.2	2.65	<u>CURVE 6</u>		490.7	0.201	<u>CURVE 10</u>			
507.2	2.63	333.2	0.552	341.2	0.561	323.2	0.119	293.2	0.753
<u>CURVE 3</u>									
383.2	1.85	360.2	0.573	401.2	0.649	473.2	1.00	<u>CURVE 18</u>	
395.2	1.84	401.2	0.649	403.2	0.628	473.2	1.00	293.2	0.628
396.2	1.87	403.2	0.628	452.2	0.703	473.2	1.00	473.2	0.628
429.2	1.92	452.2	0.703	453.2	0.682	473.2	1.00	<u>CURVE 19</u>	
443.2	1.95	453.2	0.682	473.2	1.00	473.2	1.00	293.2	0.628
447.2	1.96	473.2	1.00	473.2	1.00	473.2	1.00	473.2	0.628
479.2	2.01	512.2	0.770	473.2	1.00	473.2	1.00	473.2	0.628
500.2	2.05	563.2	0.749	469.2	0.159	473.2	1.08	473.2	0.837
503.2	2.08								

FIGURE SHOWS ONLY 56 OF THE CURVES REPORTED IN TABLE



SPECIFICATION TABLE NO. 11\* THERMAL CONDUCTIVITY OF [COPPER + ZINC] ALLOYS

(Cu + Zn - 99.50%, impurity - 0.20% each)

[For Data Reported in Figure and Table No. 115]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) Cu	Zn	Composition (continued), Specifications and Remarks
1	133	L	1930	319-494		90	99.64	0.35	0.02 Fe, 0.01 Pb; polycrystalline grain dia 0.070 mm; specimen ~13.25 in. long, 0.750 in. dia; annealed at 650 C for 1 hr; electrical conductivity $55.264 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 20 C.
2	133	L	1930	323-501	2	89	99.45	0.51	0.01 Fe, 0.01 Pb; polycrystalline, grain dia 0.110 mm; specimen ~13.25 in. long, 0.750 in. dia; annealed at 650 C for 1 hr; electrical conductivity $53.325 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 20 C.
3	133	L	1930	324-493	2	73	88.93	0.95	0.02 Fe; polycrystalline grain dia 0.120 mm; specimen ~13.25 in. long, 0.750 in. dia; annealed at 700 C for 1 hr; electrical conductivity $47.685 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 20 C.
4	133	L	1930	332-505	2	12	96.94	3.04	0.02 Fe; polycrystalline, grain dia 0.100 mm; specimen ~13.25 in. long, 0.750 in. dia; annealed at 760 C for 0.75 hr; electrical conductivity $36.607 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 20 C.
5	133	L	1930	329-506	2	13	95.21	4.77	Similar to the above specimen except grain dia 0.085 mm; electrical conductivity $33.562 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 20 C.
6	133	L	1930	328-509	2	14	90.07	9.91	0.01 Fe, 0.01 Pb; polycrystalline, grain dia 0.095 mm; specimen ~13.25 in. long, 0.750 in. dia; annealed at 700 C for 0.75 hr; electrical conductivity $25.293 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 20 C.
7	133	L	1930	326-504	2	15	83.20	16.76	0.03 Fe, 0.01 Pb; polycrystalline, grain dia 0.125 mm; specimen ~13.25 in. long, 0.750 in. dia; annealed at 700 C for 0.75 hr; electrical conductivity $20.108 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 20 C.
8	133	L	1930	327-509	2	16	79.62	20.35	0.01 Fe, 0.02 Pb; polycrystalline, grain dia 0.190 mm; specimen ~13.25 in. long, 0.750 in. dia; annealed at 700 C for 0.75 hr; electrical conductivity $18.459 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 20 C.
9	133	L	1930	325-512	2	22	59.20	40.75	0.02 Fe, 0.03 Pb; polycrystalline, grain dia 0.070 mm; specimen ~13.25 in. long, 0.750 in. dia; annealed at 650 C for 3 hr; electrical conductivity $16.700 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 20 C.

SPECIFICATION TABLE NO. 11<sup>a</sup> (continued)

Curve No.	Ref. Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) Cu	Zn	Composition (continued), Specifications and Remarks
10	133	L	1930	329-508	2	55	49.49	0.01 Fe, 0.04 Pb; polycrystalline, grain dia 16 mm; specimen -13.25 in. long, 0.750 in. dia; annealed at 650 C for 2 hr; electrical conductivity $25.852 \times 10^4$ ohm <sup>-1</sup> cm <sup>-1</sup> at 20 C.
11	133	L	1930	329-512	2	15	30.81	0.03 Fe, 0.02 Pb; polycrystalline, grain dia 0.675 mm; specimen -13.25 in. long, 0.750 in. dia; annealed at 650 C for 0.75 hr; electrical conductivity $15.857 \times 10^4$ ohm <sup>-1</sup> cm <sup>-1</sup> at 20 C.
12	133	L	1930	329-511	2	21	34.53	0.01 Fe, 0.03 Pb; polycrystalline; grain dia 0.480 mm; specimen -13.25 in. long, 0.750 in. dia; annealed at 650 C for 0.75 hr; electrical conductivity $15.325 \times 10^4$ ohm <sup>-1</sup> cm <sup>-1</sup> at 20 C.
13	133	L	1930	326-505	2	88	45.02	0.01 Fe; polycrystalline; grain dia 0.010 mm; specimen -13.25 in. long, 0.750 in. dia; annealed at 650 C for 2 hr; electrical conductivity $20.466 \times 10^4$ ohm <sup>-1</sup> cm <sup>-1</sup> at 20 C.
14	135	L	1935	293-473	19		33.72	0.01 Fe, 0.03 Pb; annealed at 700 C.
15	6	L	1931	351-763	Brass 70/30	66.24 70	39	Approx. composition, specimen -7.5 cm long, 0.585 cm dia; density 8.44 g cm <sup>-3</sup> at 22 C.
16	85		1908	108-298	Brass 70/30	70	30	Specific gravity 8.44 at 22 C.
17	234	L	1957	2.8-123	25	98.35	1.63	Specimen -5 cm long, 0.5 cm dia; supplied by Johnson, Matthey and Co. Ltd; drawn; $\rho_{10}$ = 0.425 $\mu$ ohm cm.
18	234	L	1957	2.0-130	2	98.37	1.63	The above specimen annealed at 500 C for 4 hr in a He atmosphere; $\rho_{10}$ = 0.38 $\mu$ ohm cm.
19	234	L	1957	2.1-91	55	94.63	5.37	Specimen -8 cm long, 0.5 cm dia; supplied by Johnson, Matthey and Co. Ltd; drawn; $\rho_{10}$ = 1.22 $\mu$ ohm cm.
20	234	L	1957	2.5-91	5	94.61	5.37	The above specimen annealed at 500 C for 4 hr in a He atmosphere; $\rho_{10}$ = 1.12 $\mu$ ohm cm.
21	234	L	1957	1.9-91	10	90.02	9.97	Specimen -8 cm long, 0.5 cm dia; supplied by Johnson, Matthey and Co. Ltd; drawn; annealed in a He atmosphere at 500 C for 4 hr; $\rho_{10}$ = 1.88 $\mu$ ohm cm.
22	234	L	1957	1.9-91	20	80.52	19.47	Similar to the above specimen except $L_0$ = 2.97 $\mu$ ohm cm.

SPECIFICATION TABLE NO. 118 (continued)

Curve No.	Red. Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Cu	Zn	Composition (continued), Specifications and Remarks	
23	234	L	1957	2.5-91	1	30S	68.13	31.87	Specimen 8 cm long, 0.5 cm dia; supplied by Johnson, Matthey and Co. Ltd.; drawn; $\rho_0 = 4.3$ $\mu\text{ohm cm}$ .
24	234	L	1957	2.2-91	1	30	68.13	31.87	The above specimen annealed in a He atmosphere at 500 C for 4 hr; $\rho_0 = 3.60$ $\mu\text{ohm cm}$ .
25	235	L	1944	302-335		Brass			Specimen 2.565 cm long, cross sectional area 5.017 cm <sup>2</sup> .
26	235	L	1944	314-344		Brass			Specimen 2.570 cm long, cross sectional area 3.417 cm <sup>2</sup> .
27	233	L	1957	1.9-121			<del>68.13</del> 74	<del>31.87</del> 26	Approx. composition specimen 8 cm long, 0.5 cm dia; drawn, annealed at 550 C for 4 hr; $\rho_0 = 0.56$ $\mu\text{ohm cm}$ .
28	233	L	1957	2.0-91			<del>68.13</del> 76	<del>31.87</del> 24	Similar to the above specimen except $\rho_0 = 1.20$ $\mu\text{ohm cm}$ .
29	233	L	1957	2.2-91		Commercial bronze	<del>68.13</del> 76	<del>31.87</del> 24	Similar to the above specimen except $\rho_0 = 1.94$ $\mu\text{ohm cm}$ .
30	236	L	1959	2.0-91		1	68	32	Approx. composition: $\alpha$ brass; machined from an annealed and torsionally deformed bar.
31	236	L	1959	6.5-91		2	68	32	Similar to the above specimen but annealed (after machining) up to 250 C at a rate of 6 C min <sup>-1</sup> .
32	236	L	1959	2.1-91		3	68	32	Similar to the above specimen but annealed (after machining) up to 290 C at a rate of 6 C min <sup>-1</sup> .
33	236	L	1959	2.0-91		4	68	32	Similar to the above specimen but annealed (after machining) up to 400 C at a rate of 6 C min <sup>-1</sup> .
34	246	T	1919	273, 373			92.65	7.35	Rolled and drawn; annealed close to the melting point for 1/2 hr.
35	246	T	1919	273, 373			85.65	14.35	Similar to the above specimen.
36	246	T	1919	273, 373			72.11	27.89	Similar to the above specimen.
37	246	T	1919	273, 373			66.97	33.03	Similar to the above specimen.
38	425	L	1924	90, 273		Red brass	82.	18	Polycrystalline; fine grained.
39	425	L	1924	90, 273		Red brass	82	18	Polycrystalline; coarse grained.
40	517	L	1959	2.3-4.4		Brass	85	15	$\alpha$ brass; prepared from Johnson, Matthey spectrographically standardized metals; specimen 2.5 mm dia, 4 cm long; annealed just below melting point for 40 hr.
41	517	L	1959	2.3-4.5		Brass	85	15	The above specimen drawn to produce 4.6% strain.

SPECIFICATION TABLE NO. 11\* (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (Cu) %	Composition (weight percent) Zn	Composition (continued), Specifications and Remarks
42	317	L	1959	2.4-4.5		Brass	86	15	The above specimen drawn to produce 10.4% strain.
43	317	L	1959	2.3-4.4		Brass	86	15	The above specimen drawn to produce 19.8% strain.
44	440	L+R	1940	78,273		1	95.429	4.540	$\alpha$ -brass with impurities of: 0.014 Sb, 0.010 Fe, 0.007 S, and trace As; calculated composition; annealed in N <sub>2</sub> for 20 hr at 380-400 C.
45	440	L+R	1940	78,273		2	92.791	7.180	Similar to the above specimen except 0.014 Sb, 0.009 Fe, 0.006 S and trace As.
46	440	L+R	1940	78,273		3	86.842	13.150	Similar to the above specimen except 0.013 Sb, 0.009 Fe, 0.006 S, and trace As.
47	440	L+R	1940	78,273		4	82.551	17.420	Similar to the above specimen except 0.012 Sb, 0.008 Fe, 0.006 S, and trace As.
48	440	L+R	1940	78,273		5	79.704	20.270	Similar to the above specimen.
49	440	L+R	1940	78,273		6	75.716	24.560	Similar to the above specimen except 0.011 Sb, 0.008 Fe, 0.005 S, and trace As.
50	440	L+R	1940	78,273		7	69.978	30.000	Similar to the above specimen except 0.011 Sb, 0.007 Fe, 0.005 S, and trace As.
51	440	L+R	1940	78,273		8	64.050	35.95	Similar to the above specimen except 0.010 Sb, 0.006 Fe, 0.004 S, and trace As.
52	440	L+R	1940	78,273		9	62.281	37.700	$\alpha\beta$ brass; impurities: 0.009 Sb, 0.006 Fe, 0.004 S, and trace As; calculated composition; annealed in N <sub>2</sub> at 380-400 C for 20 hr.
53	440	L+R	1940	78,273		10	59.911	40.070	Similar to the above specimen.
54	440	L+R	1940	78,273		11	55.602	44.38	Similar to the above specimen except 0.008 Sb, 0.006 Fe, 0.004 S, and trace As.
55	440	L+R	1940	78,273		12	51.073	48.910	$\beta$ brass; impurities: 0.008 Sb, 0.005 Fe, 0.004 S, and trace As; calculated composition; annealed in N <sub>2</sub> at 380-400 C for 20 hr.
56	51*	L	1960	1.4-4.0	±5	Z4	95.4	4.59	0.01 Zn; annealed for 21 hr at 540 C; electrical resistivity 1.13 and 1.08 $\mu$ ohm cm at 1.05 and 4.2 K, respectively.
57	51*	L	1960	1.4-4.0	±5	Z15	84.53	15.43	0.02 Fe, 0.02 Pb; annealed for 21 hr at 540 C; electrical resistivity 2.53 and 2.36 $\mu$ ohm cm at 1.05 and 4.2 K, respectively.



SPECIFICATION TABLE NO. 11\* (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)		Composition (continued), Specifications and Remarks
							Cu	Zn	
37	518	L	1960	1.4-4.0	±5	Z20	86.56	13.43	0.01 Fe, coldworked; annealed at 500 C for 17 hr; electrical resistivity 2.73 and 2.58 $\mu\text{ohm cm}$ at 1.05 and 4.2 K, respectively.
39	518	L	1960	1.4-4.0	±5	Z30	69.95	30.02	0.02 Fe, 0.01 Pb; annealed for 21 hr at 540 C; electrical resistivity 4.22 and 4.10 $\mu\text{ohm cm}$ at 1.5 and 4.2 K, respectively.

## DATA TABLE NO. 118 THERMAL CONDUCTIVITY OF (COPPER + ZINC) ALLOYS

(Cu + Zn = 99.50%; impurity = 0.20% each)

[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup>K<sup>-1</sup>]

CURVE 1		CURVE 3 (cont.)		CURVE 5 (cont.)		CURVE 7 (cont.)		CURVE 10 (cont.)		CURVE 14		CURVE 17 (cont.)		CURVE 18 (cont.)	
T	k	T	k	T	k	T	k	T	k	T	k	T	k	T	k
319.4	3.779	357.2	3.360	453.2	2.684	503.9	1.916	359.6	1.770	293.2	1.201	4.279	0.2463	91.20	2.900
325.0	3.690	358.2	3.360*	495.5	2.749	504.0	1.920	360.1	1.782	473.2	1.460	9.01	0.5507	91.30	2.780
332.4	3.715	392.2	2.389	505.2	2.757			399.2	1.782			10.69	0.6738	123.40	2.840
333.0	3.705	393.2	3.372	506.2	2.753	CURVE 9		403.4	1.766	CURVE 15		14.51	0.890	129.80	2.910
341.0	3.706	394.2	3.385			CURVE 11		452.8	1.766	CURVE 16		CURVE 19			
350.4	3.707	441.2	3.423			CURVE 12		452.8	1.766	351.2	1.276	18.25	1.295		
354.4	3.728	442.2	3.406			CURVE 9		452.8	1.766	367.2	1.280	25.84	1.625		
355.0	3.699	443.2	3.423	326.2	1.958	337.4	1.481	505.4	1.749	372.2	1.326	32.20	1.980	2.05	0.0443
373.5	3.699	443.2	3.454	332.5	1.946	337.8	1.448	508.0	1.736	386.2	1.351	42.96	2.120	2.45	0.0523
393.1	3.690	493.2	3.464	344.2	1.970	344.0	1.469			394.2	1.351	54.60	2.270	2.83	0.0591
394.9	3.690	493.2	3.464	346.4	1.992	360.3	1.522			413.2	1.406	59.00	2.350	3.37	0.0733
441.4	3.715	493.2	3.464	346.5	1.983	405.7	1.494	329.4	1.259	437.2	1.423	64.00	2.380	4.33	0.0872
443.9	3.711			360.2	1.986	406.2	1.607			443.2	1.410	65.40	2.400	11.76	0.287
493.5	3.732			362.6	2.042	454.5	1.690			493.2	1.431	73.20	2.430	13.55	0.325
				362.1	2.033	456.6	1.669	405.0	1.377	511.2	1.494	81.30	2.510	16.71	0.409
				404.1	2.135	507.9	1.757	456.1	1.439	541.2	1.431	90.90	2.560	21.13	0.518
				406.2	2.127	509.4	1.741	511.5	1.510	571.2	1.456	97.00	2.560	27.01	0.681
				407.8	2.121					634.2	1.494	108.30	2.710	33.10	0.823
				453.9	2.234					646.2	1.477	114.30	2.760	54.40	1.098
				454.8	2.234					703.2	1.464	122.60	2.790	58.60	1.170
				455.2	2.192									67.40	1.280
				456.2	2.192*									80.90	1.424
				456.5	2.213									91.00	1.505
				507.0	2.322										
				507.2	2.264										
				508.0	2.326										
				508.2	2.285										
				509.5	2.310										
						CURVE 10									
				326.1	1.615	326.1	1.519	326.1	1.519	108.2	0.745	1.95	0.129		
				328.2	1.565	341.7	1.536	341.7	1.536	123.2	0.778	2.277	0.152		
				342.0	1.653	341.8	1.510	341.8	1.510	135.2	0.808	2.65	0.178	2.48	0.0579
				343.6	1.607	357.2	1.521	357.2	1.521	152.2	0.833	3.135	0.210	3.04	0.0717
				358.7	1.690	359.4	1.557	359.4	1.557	170.2	0.887	3.63	0.245	3.61	0.0867
				359.5	1.690	399.1	1.582	399.1	1.582	188.2	0.925	4.266	0.293	4.29	0.1065
				402.8	1.728	448.5	1.586	448.5	1.586	204.2	0.954	4.29	0.297	4.29	0.1065
				403.7	1.736	451.9	1.586	451.9	1.586	238.2	1.004	13.55	1.090	12.25	0.377
				430.5	1.816	502.2	1.586	502.2	1.586	260.2	1.054	16.70	1.200	16.00	0.510
				451.1	1.854	359.4	1.795	359.4	1.795	281.2	1.059	20.30	1.610	20.25	0.665
										296.2	1.096	26.60	2.060	25.50	0.842
										298.2	1.092	29.36	2.340	31.00	0.977
												33.00	2.560	33.60	1.073
												56.99	2.620	57.65	1.200
												57.02	2.660	65.35	1.374
												67.12	2.520	75.90	1.458
												77.45	2.737	91.00	1.594
												82.19	2.760		
												91.10	2.854		

Not shown on plot



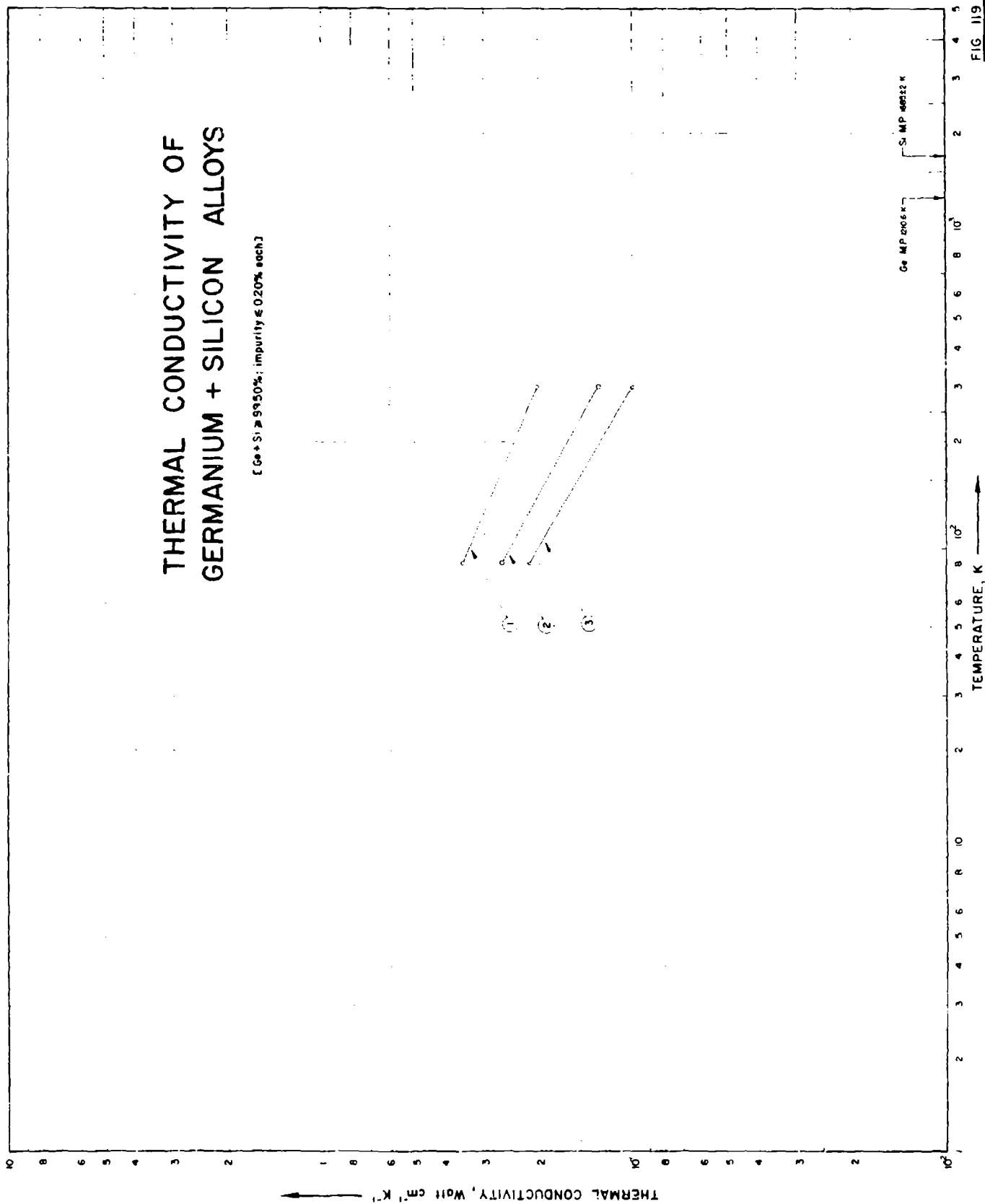
DATA TABLE NO. 11s (continued)

T	k	T	k	T	k	T	k
<u>CURVE 43</u>							
2.3	0.00240	78.2	0.510				
2.5	0.00260	273.2	1.438	273.2	0.0560		
2.75	0.00280						
3.0	0.00318	<u>CURVE 52</u>					
3.2	0.00342	78.2	0.548	1.4	0.0145		
3.5	0.00371	273.2	1.458	1.8	0.0184		
3.6	0.00383						
3.9	0.00417	<u>CURVE 53</u>					
4.2	0.00454	78.2	0.646	3.0	0.0340		
4.4	0.00477	273.2	1.276	3.3	0.0385		
<u>CURVE 44</u>							
78.2	1.284						
273.2	2.188	<u>CURVE 54</u>					
<u>CURVE 45</u>							
78.2	1.046	78.2	0.887				
273.2	1.874	273.2	1.411				
<u>CURVE 46</u>							
78.2	0.808						
273.2	1.602	<u>CURVE 55</u>					
<u>CURVE 47</u>							
78.2	0.714	78.2	1.369				
273.2	1.481	273.2	1.607				
<u>CURVE 48</u>							
78.2	0.689						
273.2	1.459	<u>CURVE 56</u>					
<u>CURVE 49</u>							
78.2	0.629	1.37	0.0366				
273.2	1.333	1.76	0.0475	1.4	0.0140		
<u>CURVE 50</u>							
78.2	0.4	2.00	0.0551	1.8	0.0208		
273.2	1.213	2.31	0.0635	2.0	0.0245		
<u>CURVE 51</u>							
78.2	0.4	2.59	0.0770	2.3	0.0280		
273.2	1.213	3.00	0.0851	2.5	0.0335		
<u>CURVE 57</u>							
78.2	0.689	3.38	0.0941	3.0	0.0385		
273.2	1.459	3.55	0.105	3.3	0.0430		
<u>CURVE 58</u>							
78.2	0.689	4.00	0.115				
273.2	1.459						

Not shown on plot

# THERMAL CONDUCTIVITY OF GERMANIUM + SILICON ALLOYS

{ Ge + Si ≥ 99.50%; impurity ≤ 0.20% each }



## SPECIFICATION TABLE NO. 119 THERMAL CONDUCTIVITY OF GERMANIUM SILICON ALLOYS

66Ge / Si 99.50% impurity 0.20% each

For Data Reported in Figure and Table No. 119

Curve No.	Rel. Method No. Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition weight percent Ge	Composition weight percent Si	Composition (continued), Specifications and Remarks
1	263	L	1958	80-300		95.11	4.76	$\rho$ -type; calculated composition; In doped lattice constant table; specimen 5x5x15 mm; grown by zone leveling technique; grain size 1 $\mu$ or dia; measured in a vacuum of 10 <sup>-3</sup> mm Hg.
2	263	L	1958	80-300		82.76	17.21	Similar to the above specimen except lattice constant 5.5687.
3	263	L	1958	80-300		65.00	34.90	Similar to the above specimen except lattice constant 5.5577.

DATA TABLE NO. 119 THERMAL CONDUCTIVITY OF GERMANIUM + SILICON ALLOYS

(Ge + Si - 95, 50% purity,  $\pm 0.20\%$  each)

Temperature, T, K; Thermal Conductivity k, Watt cm<sup>-1</sup>K<sup>-1</sup>

T k

CURVE 1

80 0.350  
300 0.201

CURVE 2

80 0.264  
300 0.128

CURVE 3

80 0.214  
300 0.100

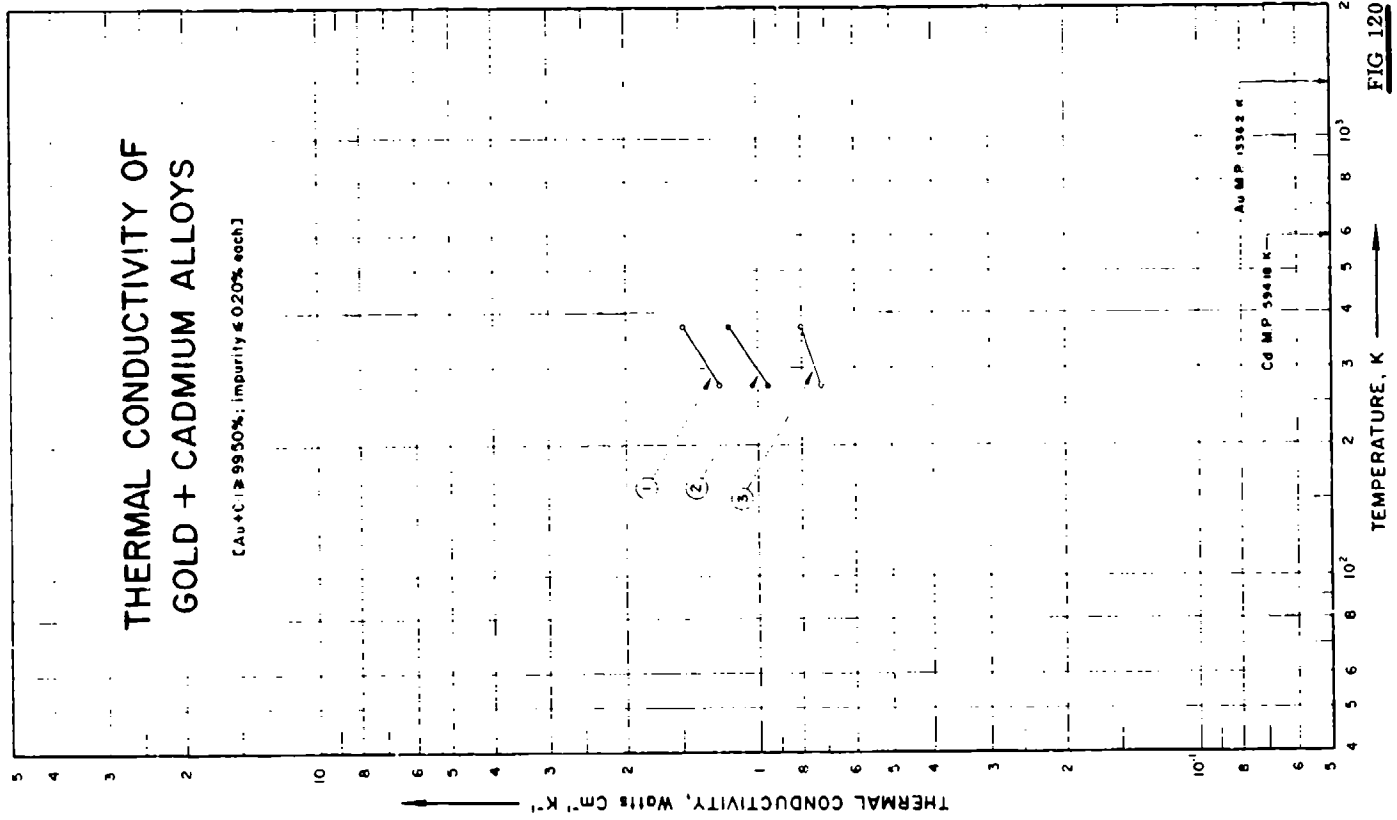


FIG 120



SPECIFICATION TABLE NO. 120 THERMAL CONDUCTIVITY OF [GOLD - CADMIUM] ALLOYS

(Au + Cd) 99.56% impurity 0.24% each

[For Data Reported in Figure and Table No. 120]

Curve No.	Reg. Method No.	Year	Temp. Range, K.	Reported Error, %	Name and Specimen Designation	Composition (weight percent) Au Cd	Composition (continued), Specifications and Remarks
1	246	T	1919	273, 373		96.85 3.15	Calculated composition, rolled and drawn to 1 mm dia wire annealed close to melting point for 9.5 hr. electrical conductivity $17.4$ and $16.2 \times 10^8 \text{ ohm}^{-1} \text{cm}^{-1}$ at 0 and 100 C., respectively.
2	246	T	1919	273, 373		94.85 5.15	Similar to the above specimen except electrical conductivity $12.9$ and $12.3 \times 10^8 \text{ ohm}^{-1} \text{cm}^{-1}$ at 0 and 100 C., respectively.
3	246	T	1919	273, 373		80.41 19.59	Similar to the above specimen except electrical conductivity $5.9$ and $5.2 \times 10^8 \text{ ohm}^{-1} \text{cm}^{-1}$ at 0 and 100 C., respectively.

## DATA TABLE NO. 120 THERMAL CONDUCTIVITY OF GOLD - CADMIUM ALLOYS

(Au + Cd - 99.50%; impurity - 0.20%; each)

Temperature, T, K; Thermal Conductivity k, Watt cm<sup>-1</sup>K<sup>-1</sup>.

T	k
<u>CURVE 1</u>	
273.2	1.23
373.2	1.48
<u>CURVE 2</u>	
273.2	0.95
373.2	1.17
<u>CURVE 3</u>	
273.2	0.72
373.2	0.90

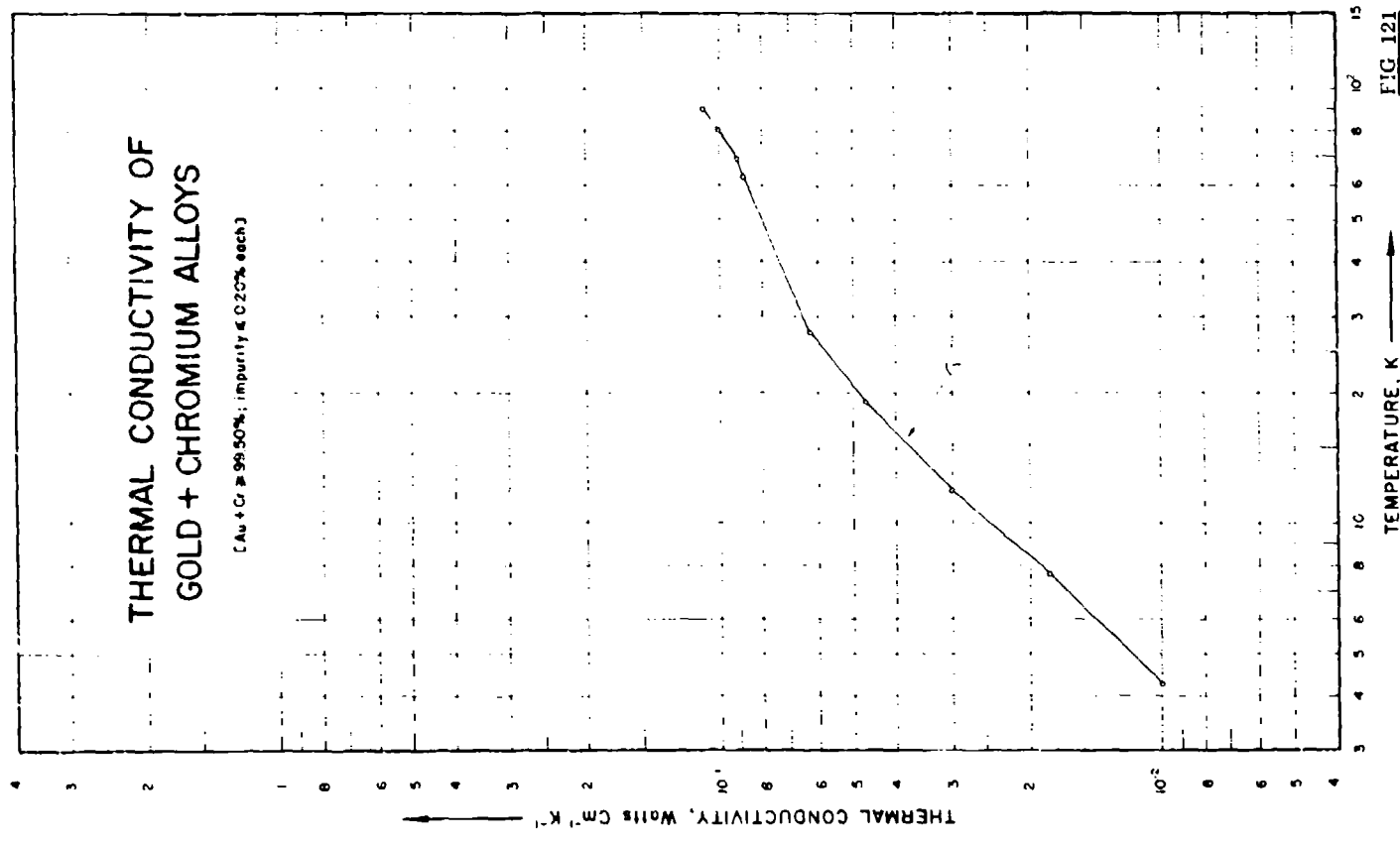


FIG 121

## SPECIFICATION TABLE NO. 121 THERMAL CONDUCTIVITY OF (GOLD + CHROMIUM) ALLOYS

(Au + Cr: 99.50% impurity: 0.20% each)

[For Data Reported in Figure and Table No. 121]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) Au      Cr	Composition (continued), Specifications and Remarks
1	450	L	1959	4.3-90			98.29      1.71	Calculated composition; 2 mm dia; drawn, annealed in He at 500 C for 12 hrs; $L_p = 27.9 \mu\text{ohm cm}$ .

## DATA TABLE NO. 121 THERMAL CONDUCTIVITY OF [GOLD - CHROMIUM] ALLOYS

(Au + Cr) 99.50%; impurity  $\pm 0.20\%$  each)[Temperature, T, K; Thermal Conductivity, k, Watt  $\text{cm}^{-1}\text{K}^{-1}$ ]

T	k	CURVE 1
4.3	0.01	
7.7	0.018	
12.0	0.03	
19.2	0.047	
27.6	0.063	
63.2	0.089	
69.6	0.092	
80.4	0.101	
90.2	0.109	

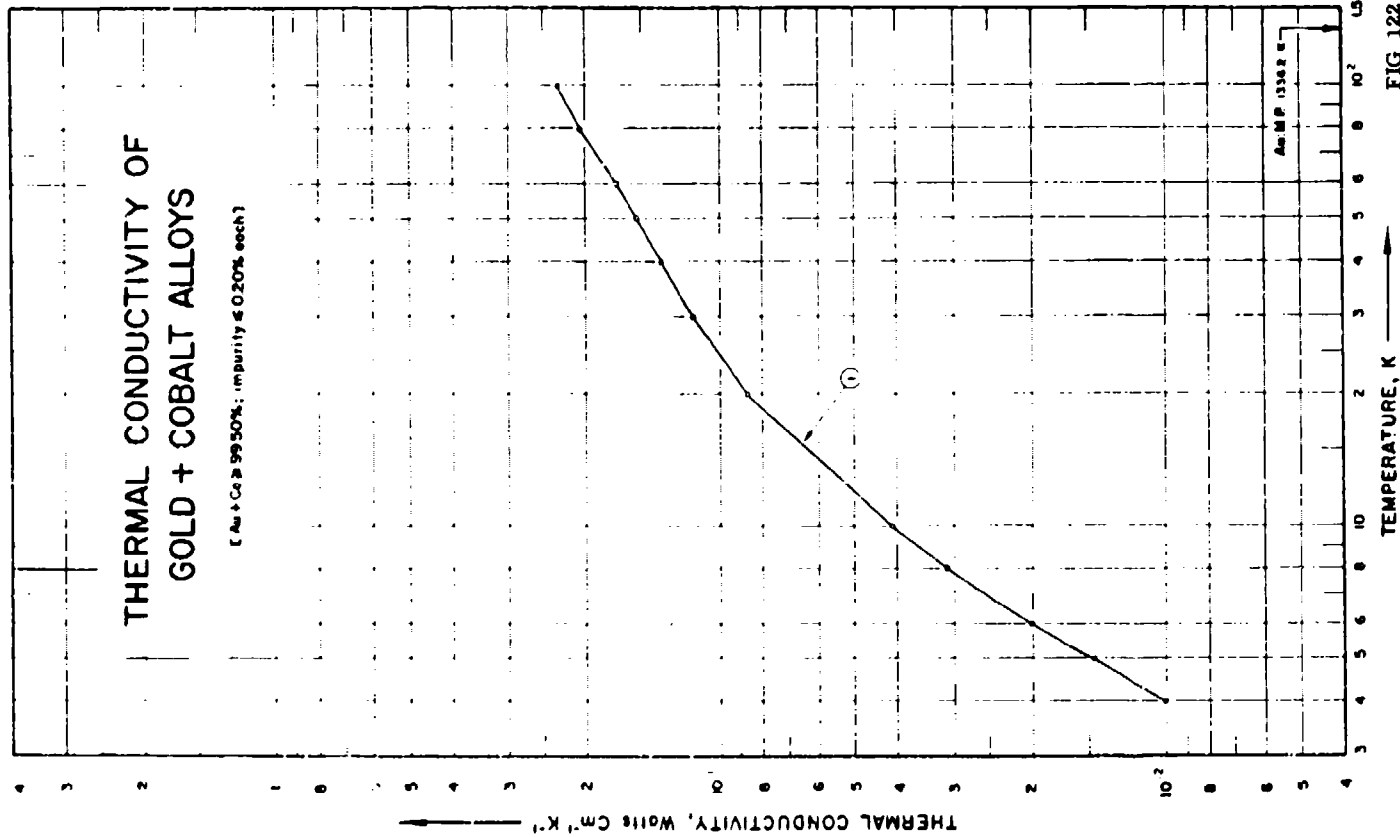


FIG 122

SPECIFICATION TABLE NO. 122 THERMAL CONDUCTIVITY OF GOLD-COBALT ALLOYS

(Au-Cu, 99.56% purity, 0.20" each)

For Data Reported in Figure and Table No. 122

Curve No.	Rel. Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)		Composition (continued), Specifications and Remarks
						Au	Cu	
1	333	1960	4-100			99.561	0.439	Calculated composition: hard drawn; supplied by Sigmund Cohn Corp.; electrical resistivity $1.20 \times 10^{-9}$ ohm cm at 30 K.

## DATA TABLE NO. 122 THERMAL CONDUCTIVITY OF [GOLD : COBALT] ALLOYS

(Au : Co : 99.50% : impurity  $\pm$  0.20% each)[Temperature, T, K: Thermal Conductivity, k, Watt cm<sup>-1</sup> K<sup>-1</sup>]

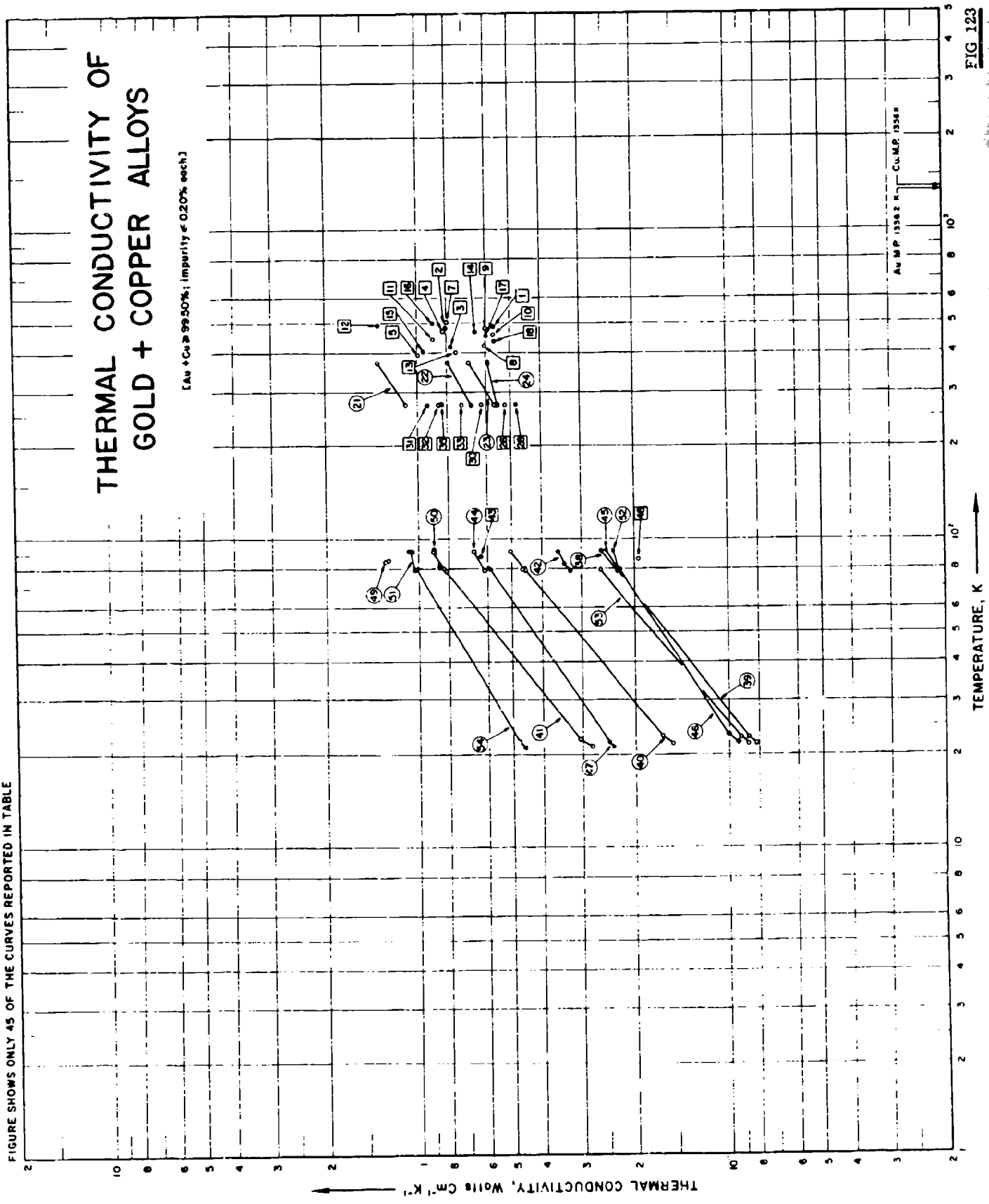
T	k	CURVE 1
4.0	0.01	
5.0	0.0146	
6.0	0.020	
8.0	0.031	
10.0	0.041	
20.0	0.086	
30.0	0.115	
40.0	0.135	
50.0	0.154	
60.0	0.171	
80.0	0.205	
100.0	0.23	



FIGURE SHOWS ONLY 45 OF THE CURVES REPORTED IN TABLE

# THERMAL CONDUCTIVITY OF GOLD + COPPER ALLOYS

[Au + Cu ≥ 99.50%; Impurity ≤ 0.20% each]



TEMPERATURE, K →

## SPECIFICATION TABLE NO. 123 THERMAL CONDUCTIVITY OF [GOLD-COPPER] ALLOYS

(Au + Cu - 99.50%; impurity - 0.50%)

[For Data Reported in Figure and Table No. 123]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition: (weight percent) Au Cu	Composition (continued), Specifications and Remarks
1	232	L	1957	488.7		IV	75.61 24.39	Calculated composition, cast; specimen 1.30 cm long, 0.63 cm <sup>2</sup> cross sectional area; density 19.34 g cm <sup>-3</sup> .
2	232	L	1957	483.2		IV		The above specimen annealed 10 hrs at 200 C.
3	232	L	1957	420.7		IV		The above specimen annealed 20 hrs at 200 C.
4	232	L	1957	473.7		IV		The above specimen annealed 30 hrs at 200 C.
5	232	L	1957	395.2		IV		The above specimen annealed 40 hrs at 200 C.
6	232	L	1957	406.2		V	85.20 14.80	Calculated composition, cast; specimen 1.30 cm long, 0.63 cm <sup>2</sup> cross sectional area; density 19.40 g cm <sup>-3</sup> .
7	232	L	1957	504.7		V		The above specimen annealed 10 hrs at 200 C.
8	232	L	1957	436.2		V		The above specimen annealed 20 hrs at 200 C.
9	232	L	1957	451.7		V		The above specimen annealed 30 hrs at 200 C.
10	232	L	1957	460.7		V		The above specimen annealed 40 hrs at 200 C.
11	232	L	1957	445.7		II	50.82 49.18	Calculated composition, cast; specimen 1.49 cm long, 0.63 cm <sup>2</sup> cross sectional area; density 15.05 g cm <sup>-3</sup> .
12	232	L	1957	483.2		II		The above specimen annealed 10 hrs at 200 C.
13	232	L	1957	401.7		II		The above specimen annealed 20 hrs at 200 C.
14	232	L	1957	470.2		II		The above specimen annealed 30 hrs at 200 C.
15	232	L	1957	403.7		II		The above specimen annealed 40 hrs at 200 C.
16	232	L	1957	497.7		III	62.54 37.46	Calculated composition, cast; specimen 1.45 cm long, 0.63 cm <sup>2</sup> cross sectional area; density 16.70 g cm <sup>-3</sup> .
17	232	L	1957	455.7		III		The above specimen annealed 10 hrs at 200 C.
18	232	L	1957	437.7		III		The above specimen annealed 20 hrs at 200 C.
19	232	L	1957	457.7		III		The above specimen annealed 30 hrs at 200 C.
20	232	L	1957	444.7		III		The above specimen annealed 40 hrs at 200 C.
21	246	T	1919	273, 373			96.73 3.27	Calculated composition; rolled and drawn to 1 mm dia wire; annealed close to melting point for 0.5 hr; electrical conductivity 14.3 and 13.4 x 10 <sup>6</sup> ohm <sup>-1</sup> cm <sup>-1</sup> at 0 and 100 C respectively.
22	246	T	1919	273, 373			92.55 7.45	Similar to the above specimen except electrical conductivity 8.5 and 8.2 x 10 <sup>6</sup> ohm <sup>-1</sup> cm <sup>-1</sup> at 0 and 100 C respectively.

SPECIFICATION TABLE NO. 123 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) Au Cu	Composition (continued), Specifications and Remarks
23	246	T	1919	273.273			87.77 12.23	Similar to the above specimen except electrical conductivity 6.3 and $5.9 \times 10^4 \text{ ohm}^{-1}\text{cm}^{-1}$ at 0 and 100 C, respectively.
24	246	T	1919	273.273			59.25 40.75	Similar to the above specimen except electrical conductivity 5.0 and $4.6 \times 10^4 \text{ ohm}^{-1}\text{cm}^{-1}$ at 0 and 100 C, respectively.
25	430	T	1924	273.2			50.8 49.2	Specimen rolled and drawn to 1 mm <sup>2</sup> cross-sectional area; 3 cm long; annealed close to melting point for 0.5 hr; electrical resistivity at 273 K, $\rho(273) = 10.8 \mu\text{ohm cm}$ .
26	430	T	1924	273.2			54.0 46.0	Similar to the above specimen except $\rho(273) = 11.4 \mu\text{ohm cm}$ .
27	430	T	1924	273.2			57.0 43.0	Similar to the above specimen except $\rho(273) = 11.8 \mu\text{ohm cm}$ .
28	430	T	1924	273.2			62.6 37.4	Similar to the above specimen except $\rho(273) = 13.0 \mu\text{ohm cm}$ .
29	430	T	1924	273.2			67.2 32.8	Similar to the above specimen except $\rho(273) = 13.6 \mu\text{ohm cm}$ .
30	430	T	1924	273.2			71.9 28.1	Similar to the above specimen except $\rho(273) = 10.5 \mu\text{ohm cm}$ .
31	430	T	1924	273.2			78.1 21.9	Similar to the above specimen except $\rho(273) = 7.6 \mu\text{ohm cm}$ .
32	430	T	1924	273.2			78.2 21.8	Similar to the above specimen except $\rho(273) = 7.6 \mu\text{ohm cm}$ .
33	430	T	1924	273.2			78.9 21.1	Similar to the above specimen except $\rho(273) = 8.4 \mu\text{ohm cm}$ .
34	430	T	1924	273.2			82.1 17.9	Similar to the above specimen except $\rho(273) = 11.6 \mu\text{ohm cm}$ .
35	430	T	1924	273.2			82.4 17.6	Similar to the above specimen except $\rho(273) = 11.6 \mu\text{ohm cm}$ .
36	430	T	1924	273.2			87.5 12.5	Similar to the above specimen except $\rho(273) = 11.6 \mu\text{ohm cm}$ .
37	430	T	1924	273.2			94.1 5.9	Similar to the above specimen except $\rho(273) = 8.0 \mu\text{ohm cm}$ .
38	58	L	1934	80.92		11	89.6 10.4	Polycrystalline specimen; cast; electrical resistivity at 33 K, $\rho(33) = 927 \mu\text{ohm cm}$ .
39	58	L	1934	22-80		11a	89.6 10.4	The above specimen annealed 40 hrs at 365 C in vacuo; $\rho(83) = 927 \mu\text{ohm cm}$ .
40	58	L	1934	22-91		12	96.9 3.10	Polycrystalline specimen; cast; $\rho(83) = 434.5 \mu\text{ohm cm}$ .
41	58	L	1934	21-91		13	98.43 1.57	Polycrystalline specimen; cast; $\rho(83) = 435.3 \mu\text{ohm cm}$ .
42	58	L	1934	19-91		14a	50.1 49.9	Polycrystalline specimen; cast; quenched from 800 C; $\rho(83) = 664 \mu\text{ohm cm}$ .
43	58	L	1934	87.4		14b	50.1 49.1	The above specimen annealed at -400 C for 20 hrs; $\rho(83) = 323 \mu\text{ohm cm}$ .
44	58	L	1934	79.92		14c	50.1 49.1	The above specimen annealed at -350 C for 32 hrs; $\rho(83) = 312.6 \mu\text{ohm cm}$ .

SPECIFICATION TABLE NO. 123 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)		Composition (continued), Specifications and Remarks
							Au	Cu	
45	54	L	1934	80-92		14J	50.1	49.1	The above specimen annealed at ~420 C for 2 hrs then quenched.
46	58	L	1934	22-80		14C	50.1	49.1	The above specimen measured after 5 months; $\rho(83) = 988 \mu\text{ohm cm}$ .
47	58	L	1934	21-81		14I	50.1	49.1	The above specimen annealed at ~325 C for .30 hrs; $\rho(83) = 341 \mu\text{ohm cm}$ .
48	58	L	1934	46-9		15a	75.6	24.4	Polycrystalline specimen, cast; quenched from 400 C; $\rho(83) = 1157 \mu\text{ohm cm}$ .
49	58	L	1934	85-85		15b	75.6	24.4	The above specimen annealed at 360 C for 22 hrs; $\rho(83) = 175.3 \mu\text{ohm cm}$ .
50	58	L	1934	81-92		15c	75.6	24.4	The above specimen annealed at 345 C for .30 hrs; $\rho(83) = 222.8 \mu\text{ohm cm}$ .
51	58	L	1934	79-91		15d	75.6	24.4	The above specimen annealed at 325 C for .30 hrs; $\rho(83) = 179.7 \mu\text{ohm cm}$ .
52	58	L	1934	79-91		15e	75.6	24.4	The above specimen annealed at 800 C for 2 hrs; quenched; $\rho(83) = 917.0 \mu\text{ohm cm}$ .
53	58	L	1934	22-79		15f	75.6	24.4	The above specimen measured after 4 months; $\rho(83) = 790.0 \mu\text{ohm cm}$ .
54	58	L	1934	21-80		15g	75.6	24.4	The above specimen annealed at ~325 C for .30 hrs; $\rho(83) = 182.6 \mu\text{ohm cm}$ .

DATA TABLE NO. 123 THERMAL CONDUCTIVITY OF GOLD-COPPER ALLOYS

(Au-Cu 99.50% Impurity 0.20% each)

Temperature, T, K. Thermal Conductivity, k, Watts cm<sup>-1</sup>K<sup>-1</sup>

T	k	T	k	T	k	T	k	T	k	T	k
CURVE 1	CURVE 12	CURVE 22	CURVE 32	CURVE 41	CURVE 49						
188.7	0.561	493.2	1.34	273.2	0.67	273.2	0.55	21.2	0.278	85.9	0.193
				373.2	0.79			22.4	0.305		
CURVE 2	CURVE 13	CURVE 23	CURVE 33	CURVE 42	CURVE 49			78.8	0.314		
483.2	0.803	481.7	0.745			273.2	0.72	80.3	0.328		
				CURVE 31				91.4	0.393	84.8	1.25
CURVE 3	CURVE 14	CURVE 24	CURVE 34	CURVE 43	CURVE 50					85.3	1.27
439.7	0.774	470.2	0.649	273.2	0.64	273.2	0.51	78.6	0.325		
								83.2	0.339	80.9	0.849
CURVE 4	CURVE 15	CURVE 25	CURVE 35	CURVE 44	CURVE 51			90.8	0.355	91.8	0.894
473.7	0.816	403.7	0.946	273.2	0.55	273.2	0.49				
				373.2	0.59						
CURVE 5	CURVE 16	CURVE 26	CURVE 36	CURVE 45							
395.2	0.687	497.7	0.879	273.2	0.50	273.2	0.50	87.4	0.629	79.1	1.03
										91.3	1.07
CURVE 6	CURVE 17	CURVE 27	CURVE 37	CURVE 46						91.4	1.08
466.2	0.812	453.7	0.594	273.2	0.52	273.2	0.73	79.3	0.616		
								91.7	0.662		
CURVE 7	CURVE 18	CURVE 28	CURVE 38	CURVE 47						79.3	0.227
504.7	0.795	437.7	0.561	273.2	0.48	273.2	0.73	80.2	0.226	91.4	0.235
								91.6	0.248		
CURVE 8	CURVE 19	CURVE 29	CURVE 39	CURVE 48						21.6	0.0861
426.2	0.602	457.7	0.552	273.2	0.48	273.2	0.73	21.9	0.0936	22.7	0.0916
								23.2	0.100	79.1	0.259
CURVE 9	CURVE 20	CURVE 30	CURVE 40	CURVE 49				80.0	0.227		
481.7	0.598	444.7	0.569	273.2	0.47	273.2	0.229	21.7	0.0517		
								22.8	0.0861		
CURVE 10	CURVE 21	CURVE 31	CURVE 41	CURVE 50				21.3	0.236		
450.7	0.565	273.2	1.09	273.2	0.62	273.2	0.229	22.1	0.245		
		373.2	1.34					80.3	0.582		
CURVE 11								90.7	0.591		
445.7	0.879	273.2	0.92	273.2	0.92	273.2	0.92	79.3	0.461		
								80.1	0.459		
								91.3	0.505		

T<sub>1</sub> Not shown on plot

# THERMAL CONDUCTIVITY OF GOLD + PALLADIUM ALLOYS

[Au + Pd > 99.90%; impurity < 0.20% each]

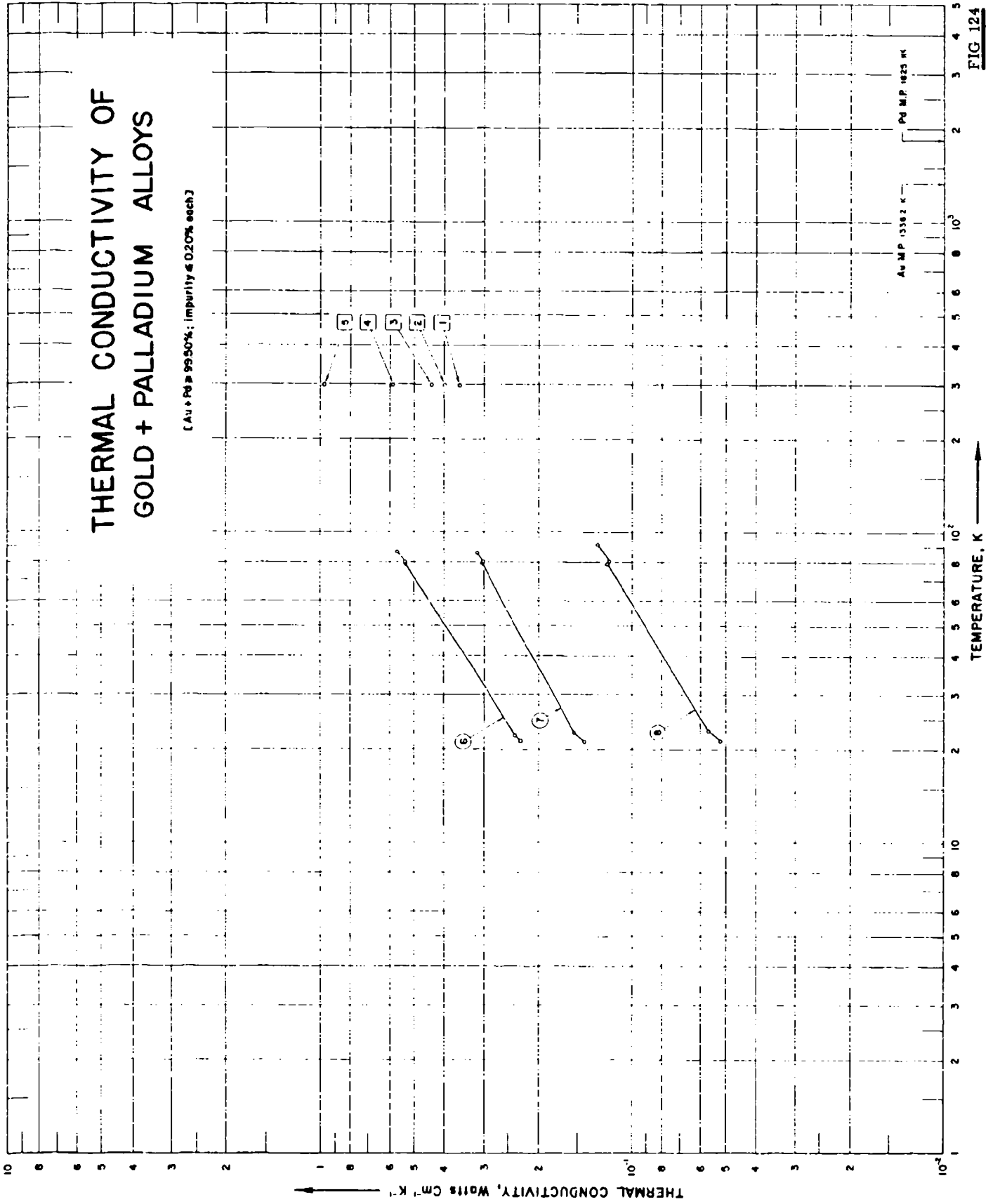


FIG 124

## SPECIFICATION TABLE NO. 124 THERMAL CONDUCTIVITY OF (GOLD + PALLADIUM) ALLOYS

(Au + Pd) 99.50%; impurity 0.20% each

[For Data Reported in Figure and Table No. 124]

Curve No.	Ref. No. Used	Method	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) Au Pd	Composition (continued), Specifications and Remarks
1	241	E	1911	298.2			50 50	Approx. composition; electrical conductivity $3.7 \times 10^4$ ohm <sup>-1</sup> cm <sup>-1</sup> at 25 C.
2	241	E	1911	298.2			60 40	Approx. composition; electrical conductivity $4.02 \times 10^4$ ohm <sup>-1</sup> cm <sup>-1</sup> at 25 C.
3	241	E	1911	298.2			70 30	Approx. composition; electrical conductivity $5.45 \times 10^4$ ohm <sup>-1</sup> cm <sup>-1</sup> at 25 C.
4	241	F	1911	298.2			80 20	Approx. composition; electrical conductivity $7.82 \times 10^4$ ohm <sup>-1</sup> cm <sup>-1</sup> at 25 C.
5	241	E	1911	298.2			90 10	Approx. composition; electrical conductivity $13.27 \times 10^4$ ohm <sup>-1</sup> cm <sup>-1</sup> at 25 C.
6	58	L	1934	21-87		22	95 5	Calculated composition; heated at 800 C for 2 hrs; <del>5.44</del> electrical resistivity $3.939$ and $3.479$ $\mu$ ohm cm at <del>25</del> and -251 C, respectively.
7	58	L	1934	21-86		23	90 10	Calculated composition; heated at 800 C for 2 hrs; <del>9.10</del> electrical resistivity $9.605$ and $7.175$ $\mu$ ohm cm at <del>25</del> and -251 C, respectively.
8	55	L	1934	21-92		24	60.1 39.9	Calculated composition; heated at 800 C for 2 hrs; <del>27.5</del> electrical resistivity $24.48$ and $23.66$ $\mu$ ohm cm at <del>25</del> and -251 C, respectively.

9-190,

DATA TABLE NO. 121 THERMAL CONDUCTIVITY OF GOLD-PALLADIUM ALLOYS

Au-Pd 39.30% (molar) 0.20% each

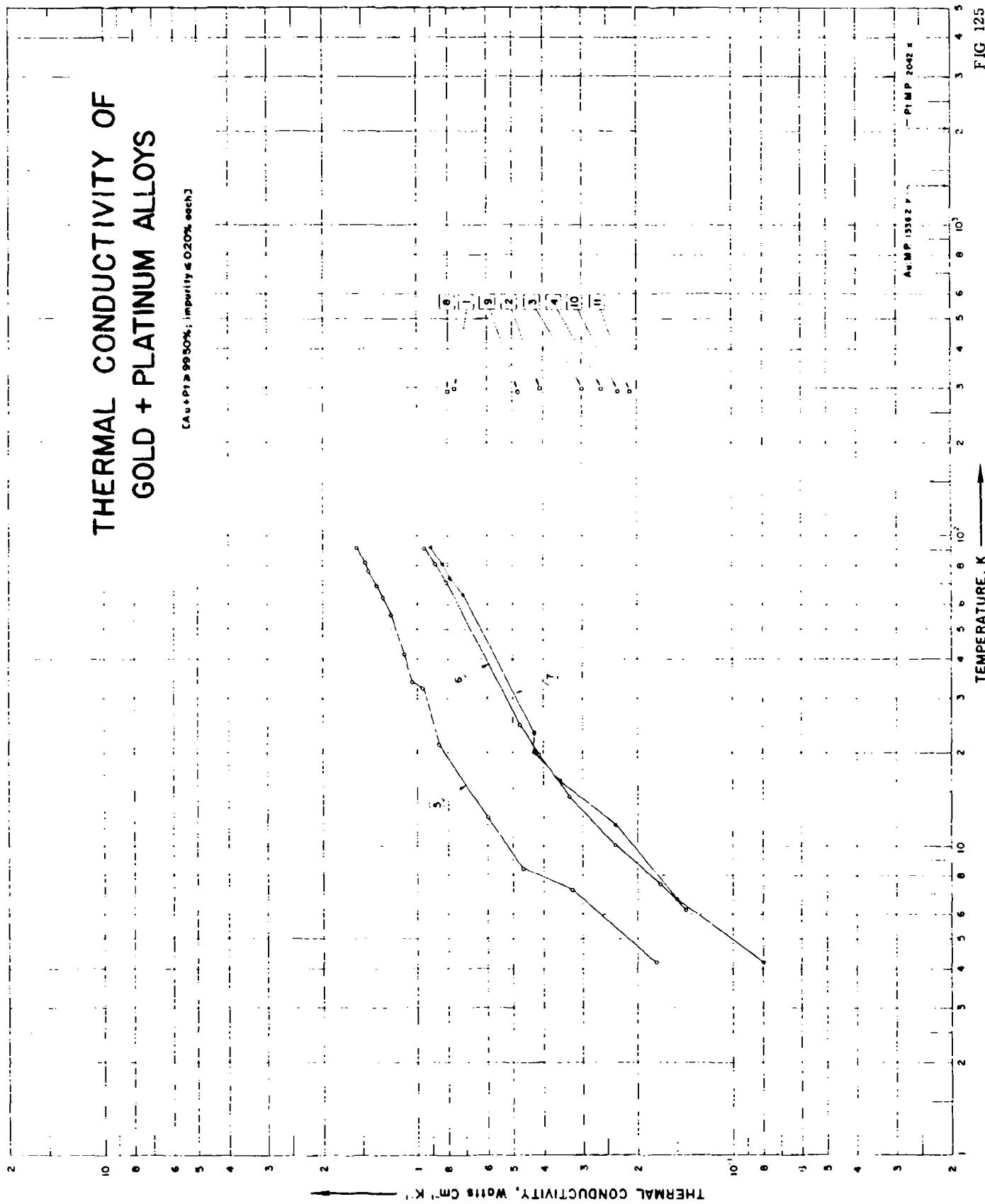
Temperature, T, K; Thermal Conductivity,  $k$ , Watt cm<sup>-1</sup>K<sup>-1</sup>

T	k
<u>CURVE 1</u>	
298.2	0.360
<u>CURVE 2</u>	
298.2	0.400
<u>CURVE 3</u>	
298.2	0.440
<u>CURVE 4</u>	
298.2	0.500
<u>CURVE 5</u>	
298.2	0.580
<u>CURVE 6</u>	
31.3	0.227
32.2	0.232
33.1	0.237
34.0	0.242
34.9	0.247
35.8	0.252
<u>CURVE 7</u>	
31.3	0.442
32.6	0.439
33.6	0.435
34.5	0.430
35.5	0.425
36.0	0.417
<u>CURVE 8</u>	
31.3	0.629
32.0	0.626
32.5	0.620
33.0	0.615
33.5	0.610



# THERMAL CONDUCTIVITY OF GOLD + PLATINUM ALLOYS

(Au + Pt ≥ 99.50%; impurity ≤ 0.20% each)



## SPECIFICATION TABLE NO. 125 THERMAL CONDUCTIVITY OF GOLD-PLATINUM ALLOYS

(Au + Pt = 99.50%; Impurity = 0.20% each)

(For Data Reported in Figure and Table No. 125)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) Au Pt	Composition (continued), Specifications and Remarks
1	241	E	1911	298.2			90 10	Approx. composition; electrical conductivity $9.61 \times 10^4$ ohm <sup>-1</sup> cm <sup>-1</sup> at 25 C.
2	241	E	1911	298.2			80 20	Approx. composition; electrical conductivity $5.49 \times 10^4$ ohm <sup>-1</sup> cm <sup>-1</sup> at 25 C.
3	241	E	1911	298.2			70 30	Approx. composition; electrical conductivity $5.10 \times 10^4$ ohm <sup>-1</sup> cm <sup>-1</sup> at 25 C.
4	241	F	1911	298.2			60 40	Approx. composition; electrical conductivity $3.03 \times 10^4$ ohm <sup>-1</sup> cm <sup>-1</sup> at 25 C.
5	450	L	1959	4.2-92			99.3 0.7	Specimen 2: 6 mm in dia; homogenized at 1050 C then drawn (reducing the cross section by 90%) and then annealed in vacuum for 4 hrs at 1050 C; $\rho_0 = 0.82 \mu\text{ohm cm}$ .
6	450	L	1959	6.2-91			98.2 1.8	Specimen 3: 3 mm in dia; received as drawn; annealed in vacuum at 1052 C for 4 hrs; $\rho_0 = 1.95 \mu\text{ohm cm}$ .
7	450	L	1959	4.2-91			95.4 4.6	Specimen 3: 3 mm dia; received as drawn; annealed in vacuum at 1050 C for 4 hrs; $\rho_0 = 2.04 \mu\text{ohm cm}$ .
8	451	T	1930	291.2			92.07 7.93	Specimen 0: 79 cm in dia and 25 mm long; supplied by Heraeus, W. C.; rolled and drawn from a piece that had been tempered at 800 C and quenched.
9	451	T	1930	291.2			84.13 15.87	Similar to the above specimen.
10	451	T	1930	291.2			68.22 31.78	Similar to the above specimen.
11	451	T	1930	291.2			55.25 44.75	Similar to the above specimen.

## DATA TABLE NO. 125 THERMAL CONDUCTIVITY OF [COLD + PLATINUM] ALLOYS

(Au + Pt ≥ 99.50%; Impurity ≤ 0.20% each)

[Temperature, T, K; Thermal Conductivity, k, Watts cm<sup>-1</sup>K<sup>-1</sup>]

T	k	T	
<u>CURVE 1</u>			
298	0.770	<u>CURVE 7</u>	
<u>CURVE 2</u>			
296	0.410	4.2	0.080
<u>CURVE 3</u>			
296	0.300	6.7	0.150
<u>CURVE 4</u>			
298	0.260	11.7	0.235
<u>CURVE 5</u>			
4.2	0.175	16.2	0.355
7.2	0.325	19.2	0.43
8.4	0.465	23.1	0.43
12.3	0.600	64.2	0.715
21.2	0.855	72.7	0.79
32.2	0.960	80.8	0.835
33.7	1.04	91.2	0.91
41.5	1.10	<u>CURVE 8</u>	
55.5	1.21	291.2	0.80
63.0	1.28	<u>CURVE 9</u>	
76.4	1.43	291.2	0.48
81.3	1.46	<u>CURVE 10</u>	
91.5	1.56	291.2	0.23
<u>CURVE 6</u>			
6.2	0.14	<u>CURVE 11</u>	
7.5	0.17	291.2	0.21
10.0	0.235		
14.4	0.33		
19.7	0.415		
24.5	0.475		
70.5	0.81		
80.6	0.88		
91.0	0.95		

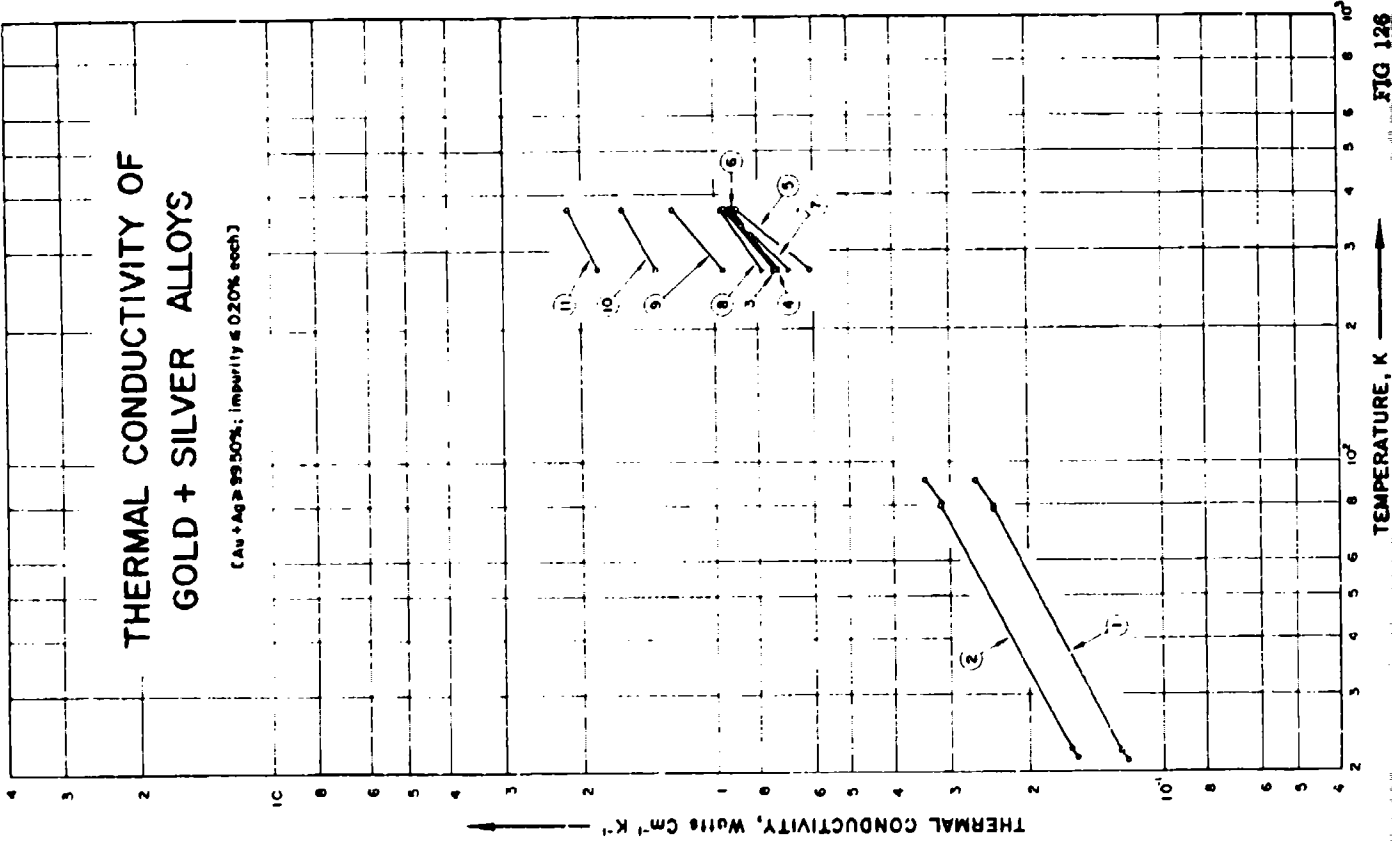


FIG. 126

TEMPERATURE, K

## SPECIFICATION TABLE NO. 126 THERMAL CONDUCTIVITY OF [GOLD - SILVER] ALLOYS

(Au + Ag = 99.50% impurity 0.20% each)

[ For Data Reported in Figure and Table No. 126 ]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) Au Ag	Composition (continued), Specifications and Remarks
1	58	L	1934	21-91		6	64.6 35.4	Calculated composition: single crystal; <del>electrical resistivity 10.2 and 8.85 <math>\mu\text{ohm cm}</math> at 0 and -251 C, respectively.</del> <b>10.2, 9-190,</b>
2	58	L	1934	22-92		7	84.5 15.5	Calculated composition: single crystal; <del>electrical resistivity 11.6 and 6.69 <math>\mu\text{ohm cm}</math> at 0 and -251 C, respectively.</del> <b>8.69, 0-190,</b>
3	246	T	1919	273.373			54.62 45.38	Calculated composition: specimen rolled and drawn to 1 mm thick; heated 0.5 hr at temp near the melting point; electrical conductivity $9.1$ and $8.4 \times 10^4 \text{ ohm}^{-1}\text{cm}^{-1}$ at 0 and 100 C, respectively.
4	246	T	1919	273.373			60.32 39.68	Similar to the above specimen except electrical conductivity $9.1$ and $8.5 \times 10^4 \text{ ohm}^{-1}\text{cm}^{-1}$ at 0 and 100 C, respectively.
5	246	T	1919	273.373			65.46 34.54	Similar to the above specimen except electrical conductivity $7.2$ and $7.2 \times 10^4 \text{ ohm}^{-1}\text{cm}^{-1}$ at 0 and 100 C, respectively.
6	246	T	1919	273.373			69.17 30.83	Similar to the above specimen except electrical conductivity $5.9$ and $5.4 \times 10^4 \text{ ohm}^{-1}\text{cm}^{-1}$ at 0 and 100 C, respectively.
7	246	T	1919	273.373			73.19 26.81	Similar to the above specimen except electrical conductivity $9.1$ and $8.5 \times 10^4 \text{ ohm}^{-1}\text{cm}^{-1}$ at 0 and 100 C, respectively.
8	246	T	1919	273.373			81.23 18.77	Similar to the above specimen except electrical conductivity $10.2$ and $9.6 \times 10^4 \text{ ohm}^{-1}\text{cm}^{-1}$ at 0 and 100 C, respectively.
9	246	T	1919	273.373			88.82 11.18	Similar to the above specimen except electrical conductivity $13.2$ and $12.4 \times 10^4 \text{ ohm}^{-1}\text{cm}^{-1}$ at 0 and 100 C, respectively.
10	246	T	1919	273.373			93.84 6.16	Similar to the above specimen except electrical conductivity $18.1$ and $15.9 \times 10^4 \text{ ohm}^{-1}\text{cm}^{-1}$ at 0 and 100 C, respectively.
11	246	T	1919	273.373			97.26 2.74	Similar to the above specimen except electrical conductivity $25.1$ and $22.0 \times 10^4 \text{ ohm}^{-1}\text{cm}^{-1}$ at 0 and 100 C, respectively.

## DATA TABLE NO. 126 THERMAL CONDUCTIVITY OF [GOLD + SILVER] ALLOYS

(Au + Ag ≥ 89.50%; Impurity ≤ 0.20% each)

[Temperature, T, K; Thermal Conductivity, k, Watts cm<sup>-1</sup>·K<sup>-1</sup>]

T	k	T	k
<u>CURVE 1</u>			
21.4	0.120	273.2	0.96
22.5	0.125	373.2	1.25
79.0	0.238		
79.6	0.238	<u>CURVE 10</u>	
91.2	0.262	273.2	1.36
		373.2	1.61
<u>CURVE 2</u>			
21.7	0.156	<u>CURVE 11</u>	
22.7	0.162	273.2	1.84
80.0	0.314	373.2	2.14
81.0	0.312		
91.6	0.335		
<u>CURVE 3</u>			
273.2	0.73		
373.2	0.95		
<u>CURVE 4</u>			
273.2	0.72		
373.2	0.93		
<u>CURVE 5</u>			
273.2	0.61		
373.2	0.89		
<u>CURVE 6</u>			
273.2	0.73		
373.2	0.91		
<u>CURVE 7</u>			
273.2	0.68		
373.2	0.93		
<u>CURVE 8</u>			
273.2	0.78		
373.2	0.96		

## SPECIFICATION TABLE NO. 127 THERMAL CONDUCTIVITY OF [GOLD + ZINC] ALLOYS

(Au + Zn : 99.50%; impurity  $\leq 0.20\%$  each)

Curve No.	Rel. Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)		Composition (continued), Specifications and Remarks
						Au	Zn	
1	T	1919	273.373			98.32	1.68	Calculated composition: specimen rolled and drawn to 1 mm thick; heated .50 hr at temp near the melting point; electrical conductivity $15.6$ and $14.4 \times 10^6 \text{ ohm}^{-1}\text{cm}^{-1}$ at 0 and 100 C, respectively. Similar to the above specimen except electrical conductivity $5.5$ and $5.6 \times 10^7 \text{ ohm}^{-1}\text{cm}^{-1}$ at 0 and 100 C, respectively.
2	T	1919	273.373			96.41	3.59	

## DATA TABLE NO. 127 THERMAL CONDUCTIVITY OF [GOLD + ZINC] ALLOYS

(Au + Zn : 99.50%; impurity  $\leq 0.20\%$  each)[Temperature, T, K; Thermal Conductivity, k, Watt  $\text{cm}^{-1}\text{K}^{-1}$ ]

T	k
<u>CURVE 1<sup>a</sup></u>	
273.2	1.17
373.2	1.33
<u>CURVE 2<sup>a</sup></u>	
273.2	0.67
373.2	0.81

<sup>a</sup> No graphical presentation

# THERMAL CONDUCTIVITY OF HAFNIUM + ZIRCONIUM ALLOYS

(Hf + Zr = 99.50%, Impurity  $\leq$  0.20% each)

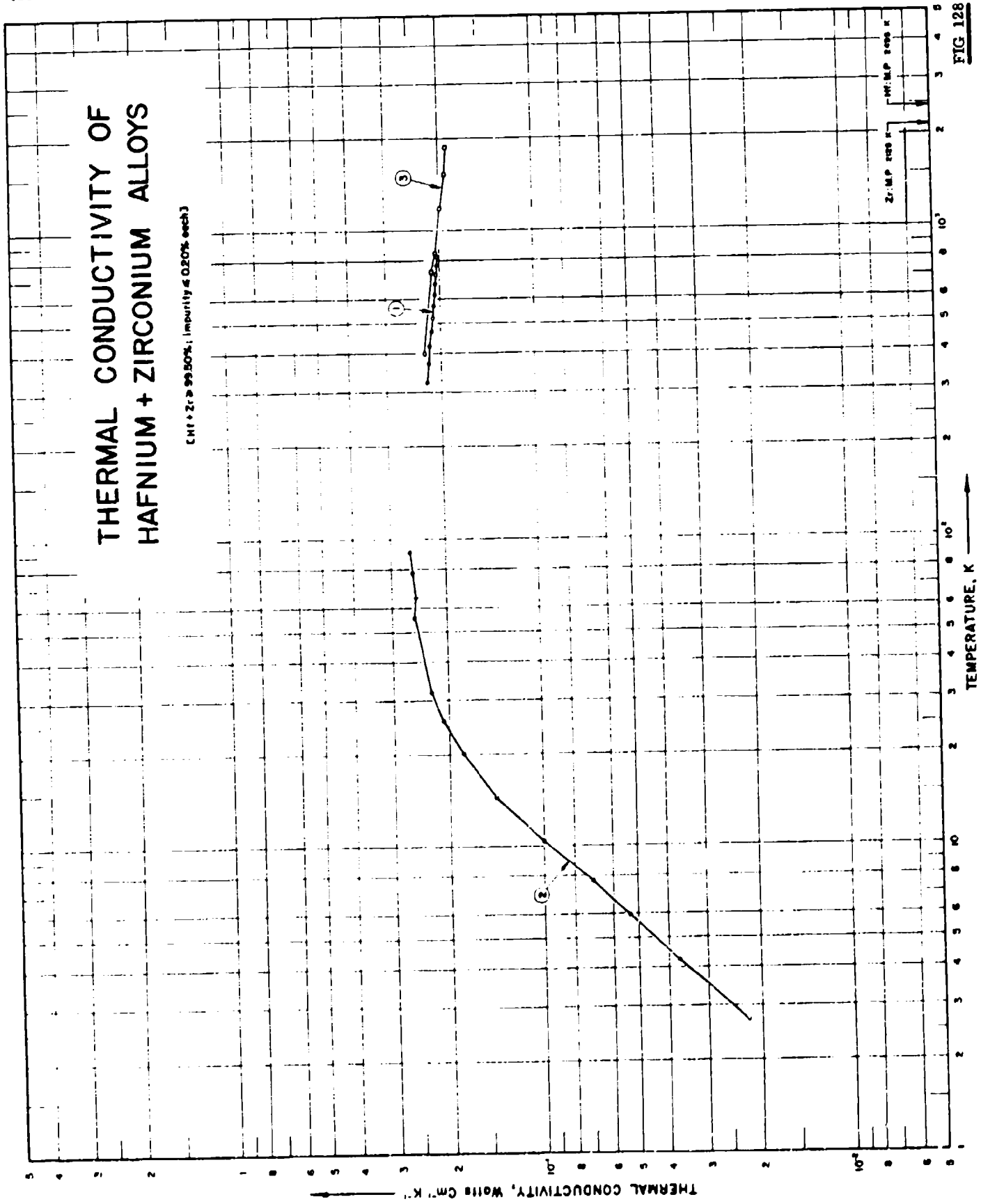


FIG 128



## SPECIFICATION TABLE NO. 12\* THERMAL CONDUCTIVITY OF HAFNIUM - ZIRCONIUM ALLOYS

(Hf - Zr - 99.50%; impurity - 0.50% each)

For Data Reported in Figure and Table No. 12\*

Curve No.	Ref. No. Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)	Composition (continued), Specifications and Remarks	
1	336	C	1953	0.23-0.23		97.96	2.0	0.008 Pb, 0.007 Al, 0.006 W, 0.005 Fe, 0.001 Cu, 0.003 Zn, 0.002 each of Si, Ti and Mo, trace Sn, U, Co, Ni, Mg, Cr and Mn; specimen 2 cm in dia and 15 cm long; supplied by Westinghouse Atomic Power Division; electrical resistivity 34.1, 40.6, 47.1, 53.6, 60.1 and 66.6 $\mu\text{ohm cm}$ at 0, 50, 100, 150, 200 and 250 C, respectively; measured in vacuum of $\sim 1 \times 10^{-5}$ mm Hg; Arrico iron used as comparative material.
2	151	L	1957	2.7-91	HFI	99.5-99	0.5-1.0	Specimen 5 x 1.52 mm and $\sim 6$ cm long; supplied by Fodor Mineral Co.; as received; $\rho_s = 4.23 \mu\text{ohm cm}$ ; electrical resistivity ratio $\rho_{295}/\rho_{100} = 0.1165$ .
3	614	R	1961	401-1578		99	1 Max	0.1 Max Ti and Si, 0.01 Max Fe, V and Zn, 0.001 Max Mn, Ni and Cu, 0.0001 Max Mg; specimen contained 5 one-inch dia disks.

DATA TABLE NO. 12<sup>a</sup> THERMAL CONDUCTIVITY OF NIOBIUM + ZIRCONIUM ALLOYS

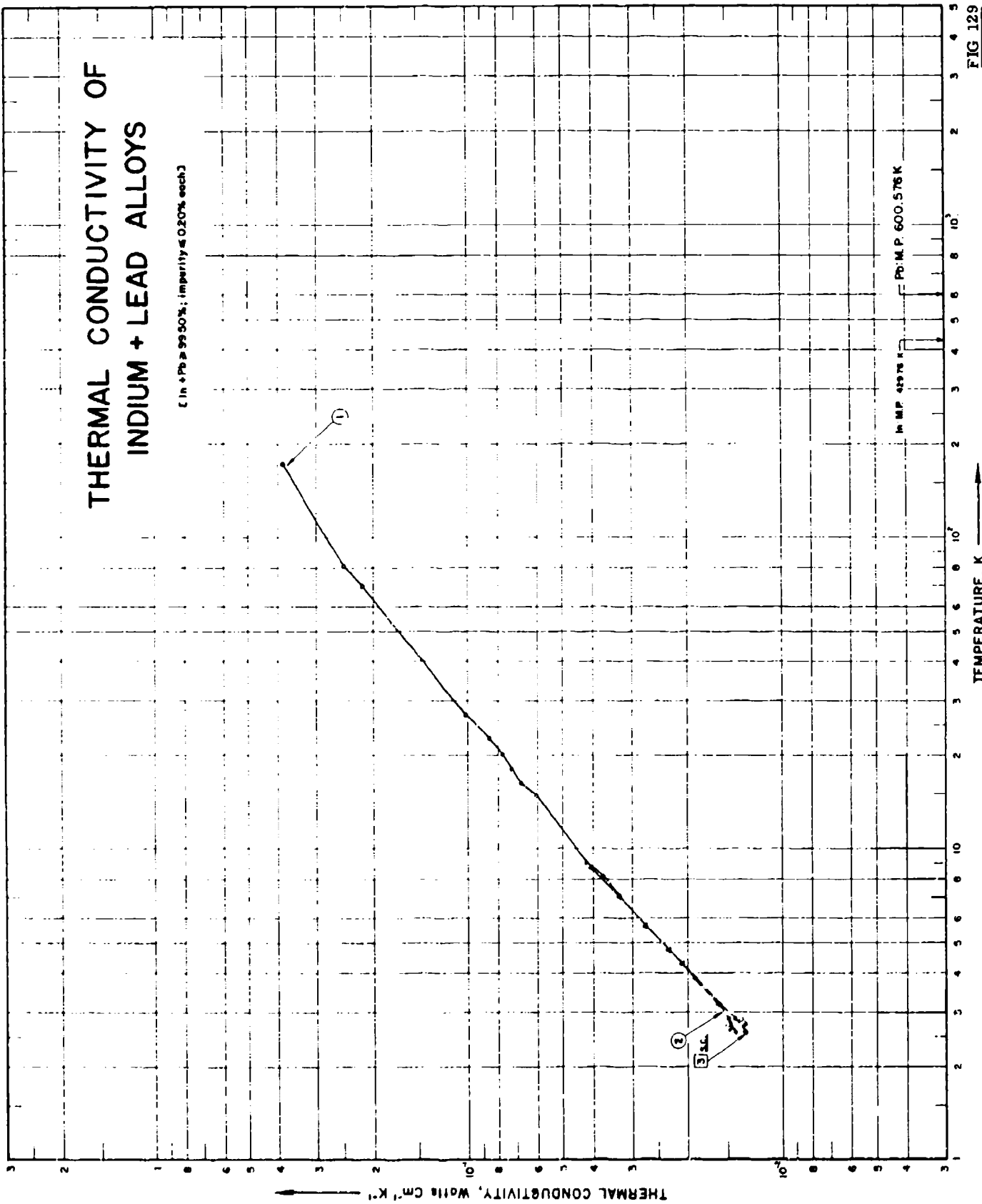
(Hf + Zr = 99.50%; impurity = 0.20% each)

[Temperature, T, K; Thermal Conductivity  $k$ , Watt cm<sup>-1</sup>K<sup>-1</sup>]

T	k
<u>CURVE 1</u>	
323.2	0.223
373.2	0.220
423.2	0.218
473.2	0.215
523.2	0.213
573.2	0.210
623.2	0.208
673.2	0.207
723.2	0.206
773.2	0.205
823.2	0.205
<u>CURVE 2</u>	
2.70	0.0221
4.23	0.0370
5.96	0.0532
7.70	0.0700
10.40	0.100
14.50	0.142
20.00	0.180
25.60	0.208
31.90	0.225
38.80	0.235
45.00	0.251
51.90	0.257
59.80	0.262
<u>CURVE 3</u>	
400.9	0.226
737.1	0.212
948.2	0.206
1190.4	0.198
1527.6	0.191
1877.6	0.189

# THERMAL CONDUCTIVITY OF INDIUM + LEAD ALLOYS

(In + Pb ≥ 99.50%; impurity ≤ 0.20% each)



## SPECIFICATION TABLE NO. 129 THERMAL CONDUCTIVITY OF [INDIUM - LEAD] ALLOYS

(In + Pb = 99.50%; Impurities = 0.20%)

(For Data Reported in Figure and Table No. 129)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) In	Pb	Composition (continued), Specifications and Remarks
1	228	L	1936	2.6-174			85.48	14.52	Without magnetic field.
2	228	L	1936	2.6-8.8			85.48	14.52	Measured in a magnetic field of 642 gauss; less than the threshold field intensity.
3	228	L	1936	2.6-3.2			85.48	14.52	Specimen in superconducting state; measured in a magnetic field of 214 gauss.

DATA TABLE NO. 129 THERMAL CONDUCTIVITY OF (INDIUM + LEAD) ALLOYS

(In + Pb) 95.50%; impurities 0.20%

Temperature, T, K; Thermal Conductivity,  $k$ , Watt/cm<sup>2</sup>·K<sup>-1</sup>.

L	K
<u>CURVE 1</u>	
2.00	0.0132
2.87	0.0142
3.19	0.0155
3.85	0.0181
4.30	0.0209
4.76	0.0240
5.09	0.0273
7.09	0.0328
8.15	0.0370
9.01	0.0420
14.9	0.0602
16.3	0.0676
18.1	0.0725
20.1	0.0775
22.7	0.0835
27.0	0.101
70.0	0.216
81.0	0.248
174.0	0.388
<u>CURVE 2</u>	
2.59	0.0141
2.85	0.0148
3.19	0.0160
3.85	0.0191
4.30	0.0209
4.76	0.0229
5.09	0.0273
7.06	0.0328
8.75	0.0407
<u>CURVE 3</u>	
2.59	0.0131
2.87	0.0140
3.19	0.0157

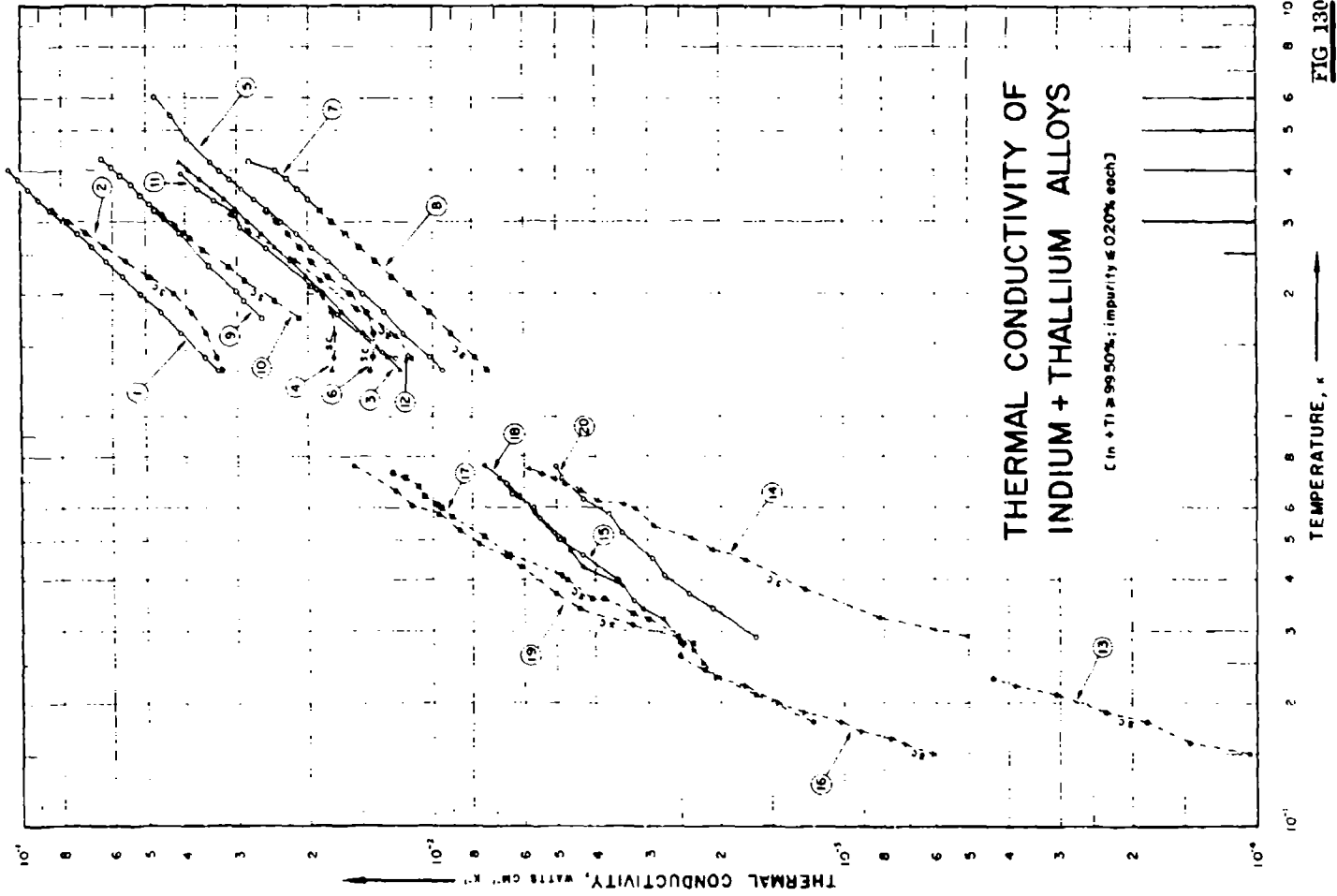


FIG 130

## SPECIFICATION TABLE NO. 130 THERMAL CONDUCTIVITY OF INDIUM-TITANIUM ALLOYS

(In Ti - 99.50%; impurity 0.50% each)

 For Data Reported in Figure and Table No. 130<sup>7</sup>

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error,	Name and Specimen Designation	Composition (weight percent) In	Composition (continued), Specifications and Remarks	
1	132	L	1955	1.3-1.0	2.5		91.43	8.57	Calculated composition; single crystal - 0.05 impurities; annealed; critical temp. $T_C = 3.28$ K; critical magnetic field ( $H_C$ ) 276.5 oersteds; in normal state; data extracted from smoothed curve.
2	132	L	1955	1.3-3.2	2.5		91.43	8.57	The above specimen in superconducting state.
3	132	L	1955	1.3-1.2	2.5		73.10	26.90	Similar to the above specimen except $T_C = 3.252$ K; $H_C = 281.1$ oersteds; in normal state.
4	132	L	1955	1.3-1.2	2.5		76.10	23.90	The above specimen in superconducting state.
5	132	L	1955	1.3-0.9	2.5		69.20	30.80	Similar to the above specimen except $T_C = 3.223$ K; $H_C = 282.3$ oersteds; in normal state.
6	132	L	1955	1.3-1.2	2.5		61.20	38.80	The above specimen in superconducting state.
7	132	L	1955	1.3-1.2	2.5		56.72	43.28	Calculated composition; polycrystal - 0.05 impurities; annealed; $T_C = 3.305$ K; data extracted from smoothed curve; in normal state.
8	132	L	1955	1.3-3.2	2.5		56.72	43.28	The above specimen in superconducting state.
9	76	L	1952	1.7-1.3			53.49	16.51	Calculated composition; - 0.05 impurities; single crystal; annealed; in normal state.
10	76	L	1952	1.7-3.1			46.49	16.51	The above specimen in superconducting state.
11	107	L	1955	1.4-2.9	3-10		76.10	23.90	Calculated composition; polycrystal; error 3% below 0.4 K; in normal state.
12	107	L	1955	1.4-5.1	3-10		76.10	23.90	The above specimen in superconducting state.
13	107	L	1955	0.15-0.23	3		76.10	23.90	Calculated composition; polycrystal; copper potassium Tutton salt used as cooling agent; in superconducting state.
14	107	L	1955	0.29-0.75	3		76.10	23.90	The above specimen measured with chromium potassium alum as cooling agent; in superconducting state.
15	107	L	1955	0.35-0.69	3		76.10	23.90	The above specimen in normal state.
16	107	L	1955	0.15-0.27	3		76.10	23.90	Calculated composition; single crystal; copper potassium Tutton salt used as cooling agent; in superconducting state.
17	107	L	1955	0.28-0.73	3		76.10	23.90	The above specimen measured with chromium potassium alum as cooling agent; in superconducting state.

SPECIFICATION TABLE NO. 130 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition in	Composition (weight percent) II	Composition (continued), Specifications and Remarks
18	107	I	1955	0.25-0.76	±		70.10	75.90	The above specimen in normal state.
19	107	I	1955	0.18-0.6	±		69.20	80.80	Calculated composition; single crystal; in superconducting state.
20	107	I	1955	0.20-0.76	±		69.20	80.80	The above specimen in normal state.



DATA TABLE NO. 130 THERMAL CONDUCTIVITY OF [INDIUM - THALLIUM] ALLOYS  
 ( $\ln + Tl \geq 99.50\%$ , impurity  $\leq 0.20\%$  each)

[Temperature, T, K; Thermal Conductivity, k, Watts $cm^{-1}K^{-1}$ ]	
T	k
<u>CURVE 1</u>	
1.30	0.0333
1.40	0.0356
1.60	0.0410
1.80	0.0461
2.00	0.0515
2.20	0.0570
2.40	0.0625
2.60	0.0680
2.80	0.0735
3.00	0.0790
3.20	0.0845
3.40	0.0900
2.60	0.0960
3.80	0.102
4.00	0.108
<u>CURVE 2</u>	
1.30	0.0327
1.40	0.0335
1.60	0.0386
2.00	0.0430
2.20	0.0490
2.40	0.0560
2.60	0.0630
2.80	0.0700
3.00	0.0770
3.20	0.0840
<u>CURVE 3</u>	
1.30	0.0129
1.60	0.0148
1.80	0.0166
2.00	0.0184
2.20	0.0203
2.40	0.0222
2.60	0.0241
2.80	0.0260
3.00	0.0281
3.20	0.0302
<u>CURVE 4</u>	
1.30	0.0176
1.40	0.0174
1.60	0.0173
1.80	0.0176
2.00	0.0185
2.20	0.0203
2.40	0.0222
2.60	0.0241
2.80	0.0260
3.00	0.0281
3.20	0.0302
<u>CURVE 5</u>	
1.30	0.0095
1.40	0.0102
1.60	0.0117
1.80	0.0131
2.00	0.0147
2.20	0.0163
2.40	0.0179
2.60	0.0196
2.80	0.0214
3.00	0.0233
<u>CURVE 6</u>	
1.30	0.0142
1.40	0.0140
1.60	0.0139
<u>CURVE 7</u>	
1.30	0.00755
1.40	0.00790
1.60	0.00905
1.80	0.0102
2.00	0.0114
2.20	0.0126
2.40	0.0138
2.60	0.0150
2.80	0.0162
3.00	0.0175
3.20	0.0187
<u>CURVE 8</u>	
1.30	0.00733
1.40	0.00790
1.60	0.00935
1.80	0.0102
2.00	0.0114
2.20	0.0126
2.40	0.0138
2.60	0.0150
2.80	0.0162
3.00	0.0175
3.20	0.0187
<u>CURVE 9</u>	
1.30	0.0123
1.44	0.0133
1.77	0.0169
2.11	0.0193
2.58	0.0231
2.90	0.0295
3.13	0.0302
3.39	0.0340
3.63	0.0374
3.91	0.0410
<u>CURVE 10</u>	
1.75	0.0211
1.93	0.0241
2.16	0.0289
2.33	0.0315
2.57	0.0362
2.75	0.0390
2.82	0.0403
2.91	0.0425
3.10	0.0455
<u>CURVE 11</u>	
1.39	0.0123
1.44	0.0133
1.77	0.0169
2.11	0.0193
2.58	0.0231
2.90	0.0295
3.13	0.0302
3.39	0.0340
3.63	0.0374
3.91	0.0410
<u>CURVE 12</u>	
1.39	0.0113
1.57	0.0123
1.84	0.0153
2.16	0.0187
2.42	0.0216
2.85	0.0280
3.12	0.0310
<u>CURVE 13</u>	
1.30	0.00755
1.40	0.00790
1.60	0.00905
1.80	0.0102
2.00	0.0114
2.20	0.0126
2.40	0.0138
2.60	0.0150
2.80	0.0162
3.00	0.0175
3.20	0.0187
<u>CURVE 14</u>	
0.291	0.000500
0.323	0.000818
0.377	0.00123
0.446	0.00173
0.473	0.00209
0.509	0.00235
0.540	0.00290
0.596	0.00323
0.610	0.00345
0.625	0.00395
0.660	0.00432
0.683	0.00473
0.705	0.00505
0.725	0.00545
0.745	0.00586
<u>CURVE 15</u>	
0.355	0.00325
0.400	0.00355
0.460	0.00430
0.500	0.00490
0.565	0.00530
0.600	0.00570
0.630	0.00610
0.690	0.00669
<u>CURVE 16 (cont.)</u>	
0.230	0.00205
0.250	0.00220
0.270	0.00231
<u>CURVE 17</u>	
0.28	0.00232
0.32	0.00300
0.33	0.00327
0.36	0.00382
0.36	0.00409
0.40	0.00468
0.41	0.00482
0.46	0.00650
0.51	0.00745
0.57	0.00900
0.61	0.00955
0.61	0.00960
0.64	0.0105
0.68	0.0107
0.71	0.0116
0.73	0.0125
<u>CURVE 18</u>	
0.28	0.00250
0.32	0.00275
0.34	0.00310
0.396	0.00355
0.430	0.00430
0.47	0.00460
0.50	0.00470
0.54	0.00525
0.58	0.00565
0.61	0.00615
0.64	0.00655
0.71	0.00685
0.76	0.00715
<u>CURVE 19</u>	
0.31	0.00328
0.34	0.00437
0.37	0.00500
0.43	0.00610
0.46	0.00660
0.49	0.00770
0.53	0.00860
0.58	0.00970
0.61	0.0112
0.66	0.0122
0.76	0.0155
<u>CURVE 20</u>	
0.29	0.00164
0.34	0.00209
0.37	0.00237
0.41	0.00273
0.45	0.00293
0.52	0.00346
0.58	0.00373
0.63	0.00428
0.71	0.00482
0.76	0.00500

# THERMAL CONDUCTIVITY OF INDIUM+TIN ALLOYS

(In 4 Sn 99.50%; Impurity 0.20% each)

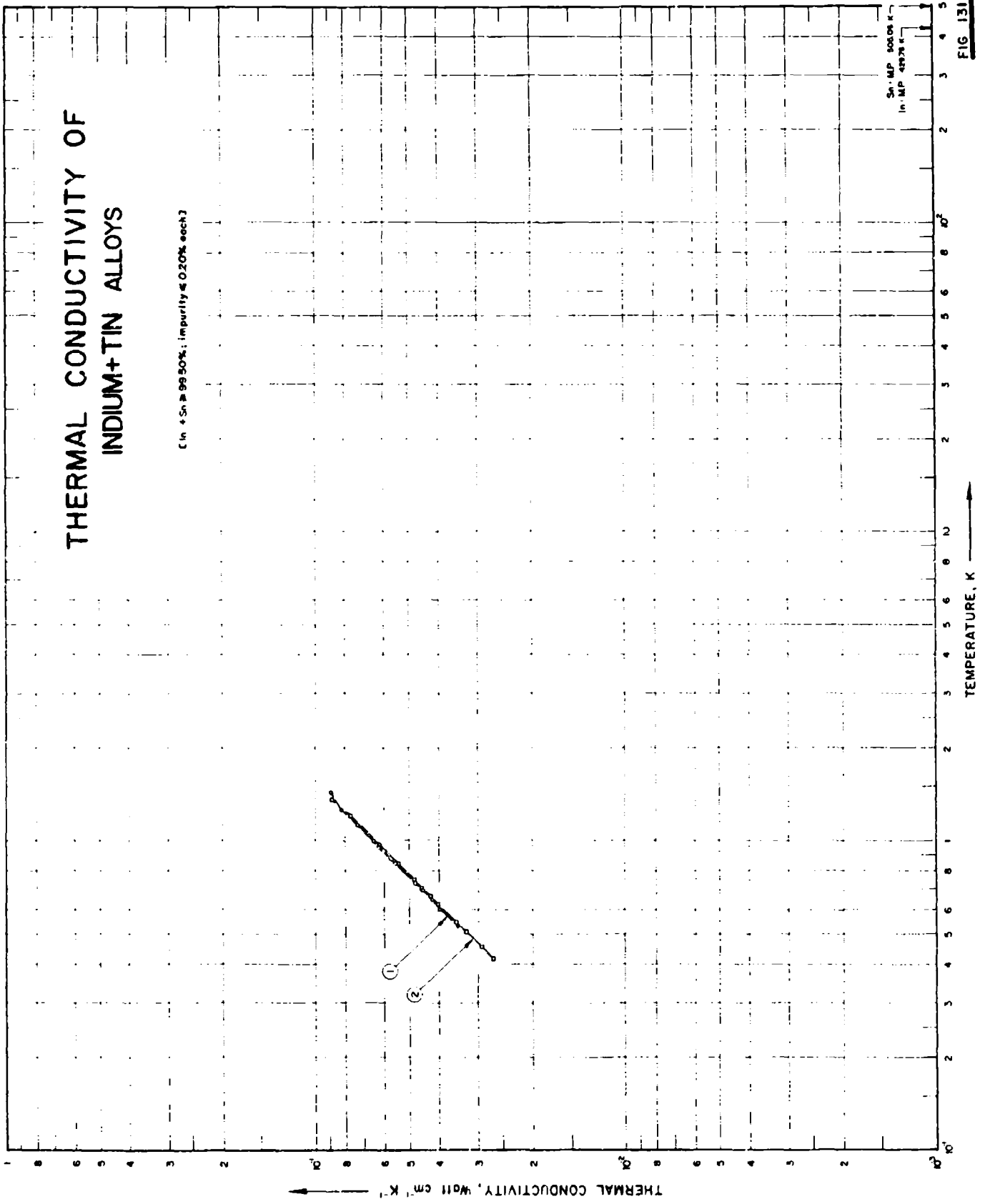


FIG. 131

## SPECIFICATION TABLE NO. 131 THERMAL CONDUCTIVITY OF INDIUM - TIN ALLOYS

(In 1 Sn 99.50% impurities 0.20%)

For Data Reported in Figure and Table No. 131

Curve No.	Rel. Method No. Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)	Composition (continued), Specifications and Remarks
						In Sn	
1	SS0 L	1965	0.5-1.3	±1	In 1	99 Sn 1	Supplied by American Smelting and Refining Co.; a total of 3 mm dia; annealed at 130°C for six months; electrical resistivity 0.37 x 10 <sup>-8</sup> ohm cm at 4.2 K; normal-state conductivity was measured in a longitudinal magnetic field of 900 gauss.
2	SS0 L	1965	0.4-1.4	±1	In 2		The above specimen heat repeatedly at room temperature before the measurement.

## DATA TABLE NO. 131 THERMAL CONDUCTIVITY OF [INDIUM + TIN] ALLOYS

(In + Sn = 99.50%; impurity = 0.20% each)

[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup>K<sup>-1</sup>]

T	k
<u>CURVE 1</u>	
0.532	0.0349
0.600	0.0398
0.646	0.0421
0.691	0.0454
0.731	0.0470
0.845	0.0552
0.950	0.0622
0.992	0.0649
1.103	0.0731
1.260	0.0822
1.342	0.0888
<u>CURVE 2</u>	
0.418	0.0268
0.455	0.0292
0.509	0.0328
0.548	0.0354
0.625	0.0404
0.661	0.0428
0.703	0.0456
0.744	0.0473
0.845	0.0543
0.971	0.0622
1.012	0.0648
1.125	0.0722
1.202	0.0772
1.365	0.0882

Not shown on plot

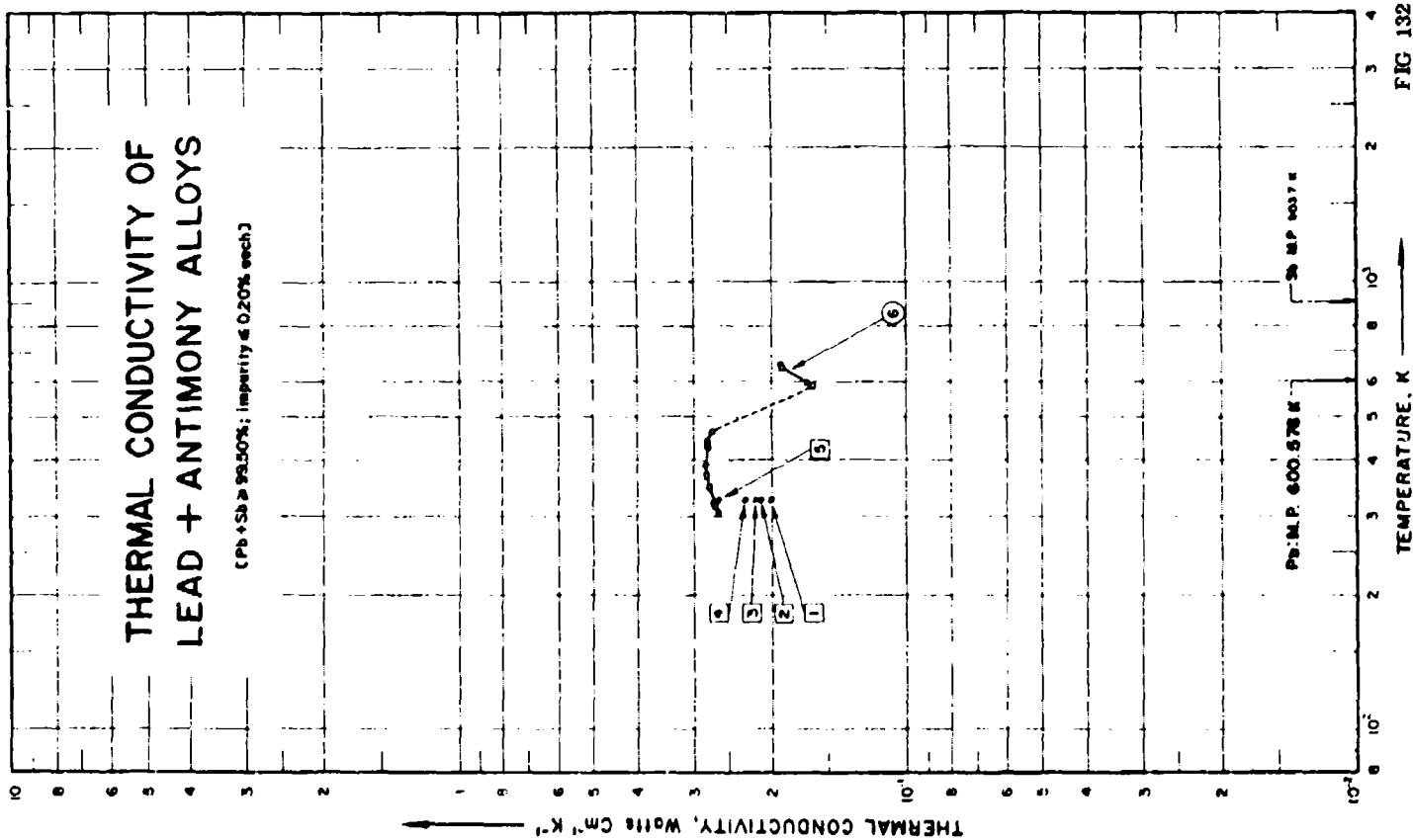


FIG 132

## SPECIFICATION TABLE NO. 12. THERMAL CONDUCTIVITY OF LEAD-ANTIMONY ALLOYS

Pb - Sb - 99.50%; impurities - 0.20% each

For Data Reported in Figure and Table No. 132.

Curve No.	Rel. Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) Pb	Composition (weight percent) Sb	Composition (continued), Specifications and Remarks
1	230	L	1925	±2%		50	50	Prepared by fusing Pb and Sb, each containing 0.03 impurities; supplied by Baker; specimen 10 cm long 1.9 cm dia; electrical conductivity $2.46 \times 10^5 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 22 C.
2	230	L	1925	±2%		60	40	Similar to the above specimen except electrical conductivity $2.66 \times 10^5 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 22 C.
3	230	L	1925	±2%		70	30	Similar to the above specimen except electrical conductivity $2.87 \times 10^5 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 22 C.
4	230	L	1925	±2%		80	20	Similar to the above specimen except electrical conductivity $3.10 \times 10^5 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 22 C.
5	230	L	1925	±2%		90	10	Similar to the above specimen except electrical conductivity $3.60 \times 10^5 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 22 C.
6	19	L	1923	307-650		87	13	Eutectic alloy; melting point 248 C; specimen 1.5 cm dia. 12 cm long.

## DATA TABLE NO. 132 THERMAL CONDUCTIVITY OF LEAD + ANTIMONY ALLOYS

(Pb + Sb 99.50%; impurity 0.20% each)

Temperature, T, K; Thermal Conductivity  $k$ , Watt cm<sup>-1</sup>K<sup>-1</sup>

T	k
<u>CURVE 1</u>	
327	0.261
<u>CURVE 2</u>	
327	0.212
<u>CURVE 3</u>	
327	0.218
<u>CURVE 4</u>	
327	0.230
<u>CURVE 5</u>	
327	0.264
<u>CURVE 6</u>	
307.2	0.264
316.2	0.267
324.2	0.269
349.2	0.277
370.2	0.280
392.2	0.281
428.2	0.278
441.2	0.279
463.2	0.272
589.2	0.162
598.2	0.168
645.2	0.190
655.2	0.191

# THERMAL CONDUCTIVITY OF LEAD + BISMUTH ALLOYS

(Pb wt% > 99.50%; impurity < 0.20% each)

FIGURE SHOWS ONLY 18 OF THE CURVES REPORTED IN TABLE

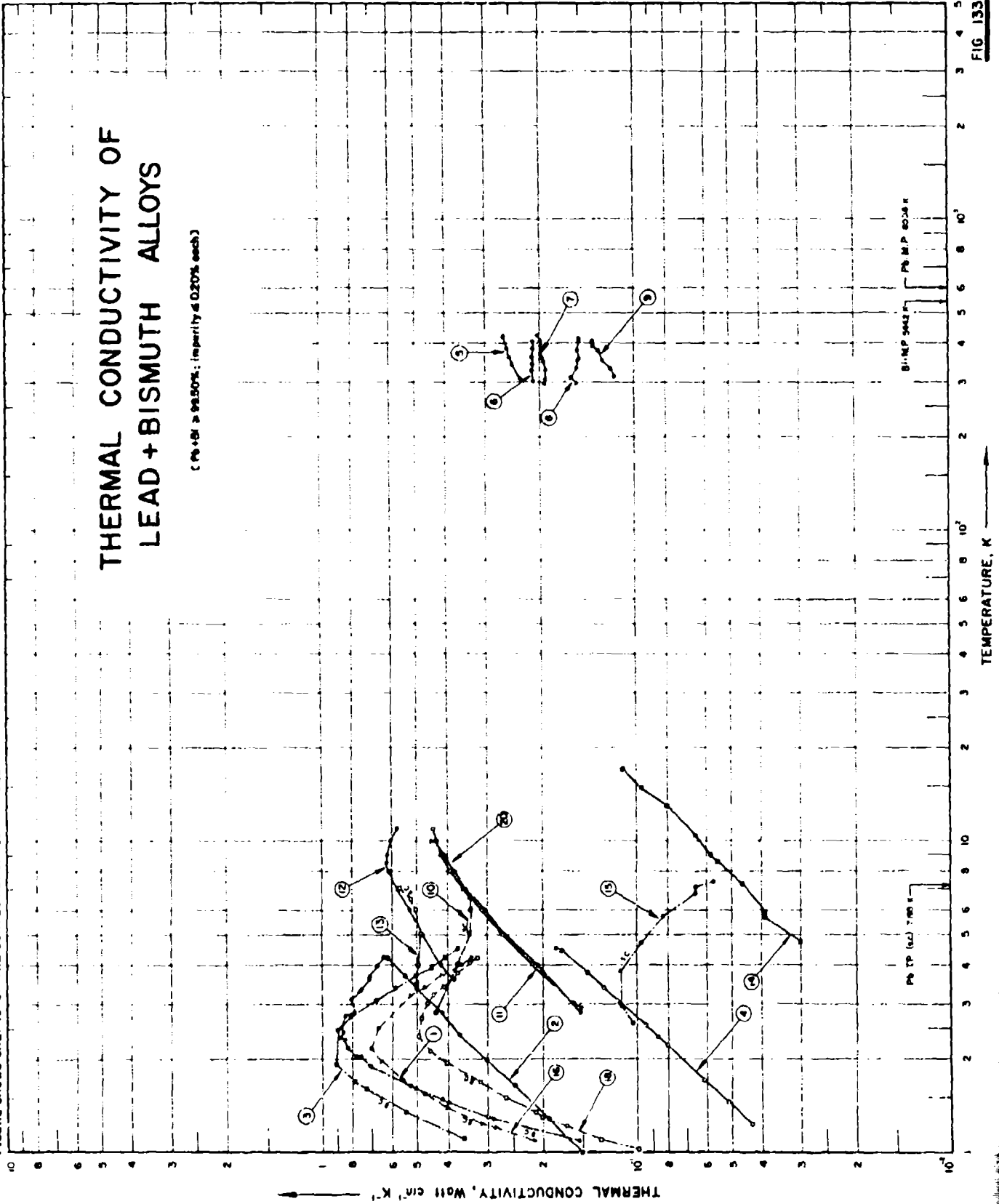


FIG 133



## SPECIFICATION TABLE NO. 133 THERMAL CONDUCTIVITY OF LEAD-BISMUTH ALLOYS

(Pb + Bi) 99.50%; impurity 0.20% each

(For Data Reported in Figure and Table No. 133)

Curve No.	Rel. Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) Pb	Composition (weight percent) Bi	Composition (continued), Specifications and Remarks	
1	257	L	1958	1.1-4.3	±2	Pb-Bi 0.2	99.794	0.202	Polycrystal; grain size 0.5 μm; annealed in vacuo for several hr. close to the melting point; in superconducting state.
2	257	L	1958	1.0-4.2	±2	Pb-Bi 0.2	99.794	0.202	The above specimen measured in a magnetic field of 1000 gauss; in normal state.
3	257	L	1958	1.1-4.5	±2	Pb-Bi 0.7	99.284	0.716	Single crystal; annealed in vacuo for several days close to the melting point; in superconducting state.
4	257	L	1958	1.2-4.5	±2	Pb-Bi 0.7	99.284	0.716	The above specimen measured in a magnetic field of 1000 gauss; in normal state.
5	248	E	1956	308-421	±3		89.80	10.20	
6	248	E	1956	302-406	±3		85.88	14.12	
7	248	E	1956	296-425	±3		78.34	21.66	
8	248	E	1956	298-415	±3		72.74	27.26	
9	248	E	1956	311-405	±3		61.15	38.85	
10	468	L	1952	2.8-6.6			99.5	0.5	In superconducting state.
11	468	L	1952	2.9-11			99.5	0.5	The above specimen in normal state.
12	468	L	1952	3.6-11			99.8	0.2	In normal state.
13	468	L	1952	3.5-7.6			99.8	0.2	The above specimen in superconducting state.
14	96	L	1950	4.8-17			90	10	In normal state.
15	96	L	1950	2.6-7.4			90	10	The above specimen in superconducting state.
16	389	L	1958	1.0-4.3			99	1	Specimen straight; annealed; measured in superconducting state.
17	389	L	1958	1.0-4.2			99	1	The above specimen bent; measured in superconducting state.
18	389	L	1958	1.0-4.2			99	1	The above specimen annealed at room temp; measured in superconducting state.

DATA TABLE NO. 133 THERMAL CONDUCTIVITY OF (LEAD + BISMUTH) ALLOYS

(Pb + Bi: 99.50%; impurity 0.24% each)

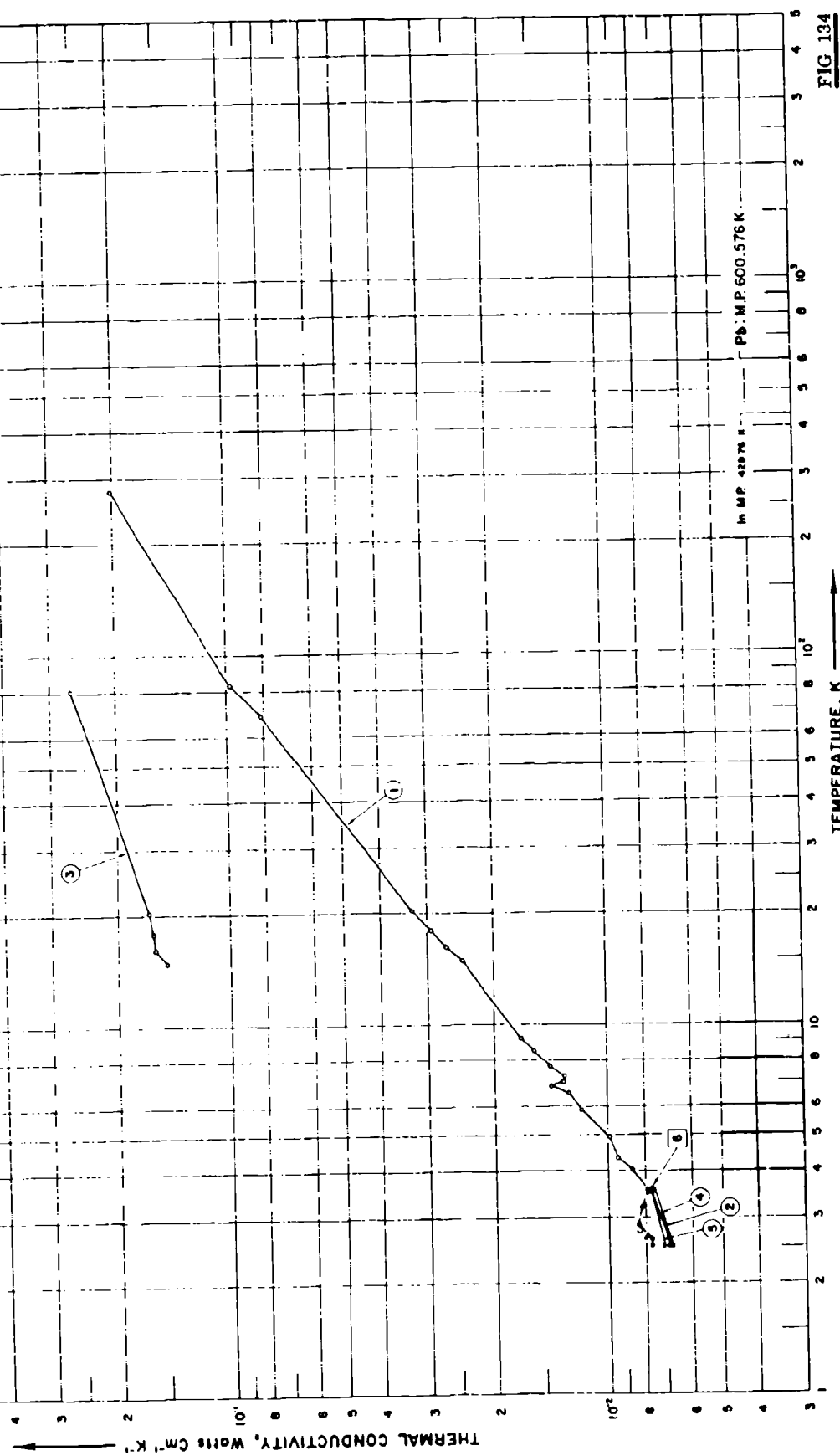
[Temperature, T, K; Thermal Conductivity k, Watt cm<sup>-1</sup>K<sup>-1</sup>]

CURVE 1		CURVE 2		CURVE 3 (cont.)		CURVE 4		CURVE 5		CURVE 6		CURVE 7		CURVE 8		CURVE 9		CURVE 10		CURVE 11		CURVE 12		CURVE 13		CURVE 14		CURVE 15		CURVE 16		CURVE 17 (cont.)			
T	k	T	k	T	k	T	k	T	k	T	k	T	k	T	k	T	k	T	k	T	k	T	k	T	k	T	k	T	k	T	k	T	k	T	k
1.09	0.154	1.00	0.130	1.69	0.782	298.2	0.151	298.2	0.151	303.2	0.228	298.2	0.192	4.0	0.215	2.60	0.103	3.35	0.750	390.2	0.135	3.0	0.423	4.0	0.215	2.60	0.103	4.10	0.355	4.23	0.348	4.10	0.355	4.23	0.348
1.28	0.286	1.28	0.191	1.93	0.900	310.2	0.150	310.2	0.150	343.4	0.242	306.2	0.190	5.0	0.235	3.00	0.112	3.61	0.704	352.2	0.126	4.0	0.372	5.0	0.235	3.00	0.112	4.23	0.348	4.23	0.348	4.23	0.348	4.23	0.348
1.30	0.298	1.64	0.255	2.47	0.890	332.2	0.151	332.2	0.151	363.4	0.247	326.2	0.190	6.0	0.310	3.80	0.113	3.75	0.702	377.2	0.130	5.0	0.397	6.0	0.310	3.80	0.113	4.08	0.334	4.08	0.334	4.08	0.334	4.08	0.334
1.45	0.397	1.95	0.300	2.74	0.806	352.2	0.150	352.2	0.150	396.2	0.253	349.2	0.192	7.0	0.355	4.70	0.097	4.44	0.663	399.2	0.135	6.0	0.480	7.0	0.355	4.70	0.097	4.20	0.320	4.20	0.320	4.20	0.320	4.20	0.320
1.49	0.418	2.39	0.327	3.09	0.675	366.2	0.149	366.2	0.149	420.7	0.359	358.2	0.196	8.0	0.389	5.70	0.083	3.98	0.653	405.2	0.135	6.5	0.525	8.0	0.389	5.70	0.083	4.00	0.320	4.00	0.320	4.00	0.320	4.00	0.320
1.61	0.503	2.99	0.341	3.36	0.586	380.2	0.151	380.2	0.151	455.2	0.419	394.2	0.197	9.0	0.415	6.80	0.065	3.29	0.754	405.2	0.135	7.0	0.600	9.0	0.415	6.80	0.065	4.00	0.320	4.00	0.320	4.00	0.320	4.00	0.320
1.64	0.522	3.35	0.335	3.69	0.500	409.2	0.149	409.2	0.149	485.2	0.485	408.2	0.197	10.0	0.430	7.10	0.065	3.67	0.702	415.2	0.149	7.0	0.660	10.0	0.430	7.10	0.065	4.00	0.320	4.00	0.320	4.00	0.320	4.00	0.320
1.90	0.700	3.98	0.663	3.94	0.449	425.2	0.203	425.2	0.203	501.7	0.507	425.2	0.203	11.0	0.440	7.10	0.057	4.25	0.638	415.2	0.149	11.0	0.575	11.0	0.440	7.10	0.057	4.00	0.320	4.00	0.320	4.00	0.320	4.00	0.320
2.04	0.754	4.25	0.638	4.23	0.403	450	0.371	450	0.371	542.7	0.59	425.2	0.203	11.0	0.440	7.10	0.057	4.25	0.638	415.2	0.149	11.0	0.575	11.0	0.440	7.10	0.057	4.00	0.320	4.00	0.320	4.00	0.320	4.00	0.320
2.04	0.774	4.25	0.638	4.50	0.371	450	0.371	450	0.371	585.7	0.710	425.2	0.203	11.0	0.440	7.10	0.057	4.25	0.638	415.2	0.149	11.0	0.575	11.0	0.440	7.10	0.057	4.00	0.320	4.00	0.320	4.00	0.320	4.00	0.320
2.17	0.825	4.25	0.638	4.50	0.371	450	0.371	450	0.371	601.7	0.707	425.2	0.203	11.0	0.440	7.10	0.057	4.25	0.638	415.2	0.149	11.0	0.575	11.0	0.440	7.10	0.057	4.00	0.320	4.00	0.320	4.00	0.320	4.00	0.320
2.30	0.841	4.25	0.638	4.50	0.371	450	0.371	450	0.371	626.7	0.710	425.2	0.203	11.0	0.440	7.10	0.057	4.25	0.638	415.2	0.149	11.0	0.575	11.0	0.440	7.10	0.057	4.00	0.320	4.00	0.320	4.00	0.320	4.00	0.320
2.35	0.876	4.25	0.638	4.50	0.371	450	0.371	450	0.371	651.7	0.710	425.2	0.203	11.0	0.440	7.10	0.057	4.25	0.638	415.2	0.149	11.0	0.575	11.0	0.440	7.10	0.057	4.00	0.320	4.00	0.320	4.00	0.320	4.00	0.320
2.45	0.851	4.25	0.638	4.50	0.371	450	0.371	450	0.371	676.7	0.710	425.2	0.203	11.0	0.440	7.10	0.057	4.25	0.638	415.2	0.149	11.0	0.575	11.0	0.440	7.10	0.057	4.00	0.320	4.00	0.320	4.00	0.320	4.00	0.320
2.53	0.866	4.25	0.638	4.50	0.371	450	0.371	450	0.371	701.7	0.710	425.2	0.203	11.0	0.440	7.10	0.057	4.25	0.638	415.2	0.149	11.0	0.575	11.0	0.440	7.10	0.057	4.00	0.320	4.00	0.320	4.00	0.320	4.00	0.320
2.76	0.841	4.25	0.638	4.50	0.371	450	0.371	450	0.371	726.7	0.710	425.2	0.203	11.0	0.440	7.10	0.057	4.25	0.638	415.2	0.149	11.0	0.575	11.0	0.440	7.10	0.057	4.00	0.320	4.00	0.320	4.00	0.320	4.00	0.320
2.83	0.781	4.25	0.638	4.50	0.371	450	0.371	450	0.371	751.7	0.710	425.2	0.203	11.0	0.440	7.10	0.057	4.25	0.638	415.2	0.149	11.0	0.575	11.0	0.440	7.10	0.057	4.00	0.320	4.00	0.320	4.00	0.320	4.00	0.320
3.11	0.806	4.25	0.638	4.50	0.371	450	0.371	450	0.371	776.7	0.710	425.2	0.203	11.0	0.440	7.10	0.057	4.25	0.638	415.2	0.149	11.0	0.575	11.0	0.440	7.10	0.057	4.00	0.320	4.00	0.320	4.00	0.320	4.00	0.320
3.29	0.754	4.25	0.638	4.50	0.371	450	0.371	450	0.371	801.7	0.710	425.2	0.203	11.0	0.440	7.10	0.057	4.25	0.638	415.2	0.149	11.0	0.575	11.0	0.440	7.10	0.057	4.00	0.320	4.00	0.320	4.00	0.320	4.00	0.320
3.35	0.750	4.25	0.638	4.50	0.371	450	0.371	450	0.371	826.7	0.710	425.2	0.203	11.0	0.440	7.10	0.057	4.25	0.638	415.2	0.149	11.0	0.575	11.0	0.440	7.10	0.057	4.00	0.320	4.00	0.320	4.00	0.320	4.00	0.320
3.61	0.704	4.25	0.638	4.50	0.371	450	0.371	450	0.371	851.7	0.710	425.2	0.203	11.0	0.440	7.10	0.057	4.25	0.638	415.2	0.149	11.0	0.575	11.0	0.440	7.10	0.057	4.00	0.320	4.00	0.320	4.00	0.320	4.00	0.320
3.67	0.702	4.25	0.638	4.50	0.371	450	0.371	450	0.371	876.7	0.710	425.2	0.203	11.0	0.440	7.10	0.057	4.25	0.638	415.2	0.149	11.0	0.575	11.0	0.440	7.10	0.057	4.00	0.320	4.00	0.320	4.00	0.320	4.00	0.320
3.98	0.663	4.25	0.638	4.50	0.371	450	0.371	450	0.371	901.7	0.710	425.2	0.203	11.0	0.440	7.10	0.057	4.25	0.638	415.2	0.149	11.0	0.575	11.0	0.440	7.10	0.057	4.00	0.320	4.00	0.320	4.00	0.320	4.00	0.320
4.25	0.638	4.25	0.638	4.50	0.371	450	0.371	450	0.371	926.7	0.710	425.2	0.203	11.0	0.440	7.10	0.057	4.25	0.638	415.2	0.149	11.0	0.575	11.0	0.440	7.10	0.057	4.00	0.320	4.00	0.320	4.00	0.320	4.00	0.320

Not shown on plot

# THERMAL CONDUCTIVITY OF LEAD + INDIUM ALLOYS

[Pb+In > 99.50%; Impurity < 0.20% each]



## SPECIFICATION TABLE NO. 134 THERMAL CONDUCTIVITY OF [LEAD + INDIUM] ALLOYS

(Pb + In - 99.50%; impurity &lt; 0.20% each)

[For Data Reported in Figure and Table No. 134]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition Pb	Composition (weight percent) In	Composition (continued), Specifications and Remarks
1	228	L	1936	2.5-278			64.35	35.65	Calculated composition; specimen in superconducting state at lower temp.
2	228	L	1936	2.5, 3.6			64.35	35.65	The above specimen measured in a magnetic field of 721 gauss (below critical field strength).
3	228	L	1936	15-81			99.444	0.556	Calculated composition.
4	228	L	1936	2.5, 3.5			64.35	35.65	Calculated composition; measured in a magnetic field of 481 gauss.
5	226	L	1936	2.5, 3.6			64.35	35.65	The above specimen measured in a magnetic field of 240 gauss.
6	228	L	1936	3.56			64.35	35.65	The above specimen measured in a magnetic field of 601 gauss.

## DATA TABLE NO. 134 THERMAL CONDUCTIVITY OF [LEAD + INDIUM] ALLOYS

(Pb + In : 99.50%; impurity &lt; 0.20% each)

[Temperature, T, K; Thermal Conductivity k, Watt cm<sup>-1</sup>K<sup>-1</sup>]

T	k	T	k
<u>CURVE 1</u>			
2.51	0.00769	<u>CURVE 5</u>	
2.64	0.00769	2.52	0.00709
2.87	0.00803	3.56	0.00769
3.20	0.00813	<u>CURVE 6</u>	
3.56	0.00787	3.56	0.00763
4.02	0.00862		
4.36	0.00943		
4.94	0.00990		
5.87	0.0117		
6.52	0.0127		
6.81	0.0141		
7.02	0.0132		
7.25	0.0131		
7.69	0.0143		
8.45	0.0157		
9.16	0.0170		
14.90	0.0242		
16.30	0.0267		
18.00	0.0293		
20.40	0.0326		
69.00	0.0813		
83.00	0.0980		
278.00	0.202		
<u>CURVE 2</u>			
2.53	0.00676		
3.56	0.00752		
<u>CURVE 3</u>			
14.80	0.149		
16.10	0.158		
17.90	0.160		
20.40	0.165		
81.00	0.263		
<u>CURVE 4</u>			
2.53	0.00680		
3.56	0.00769		

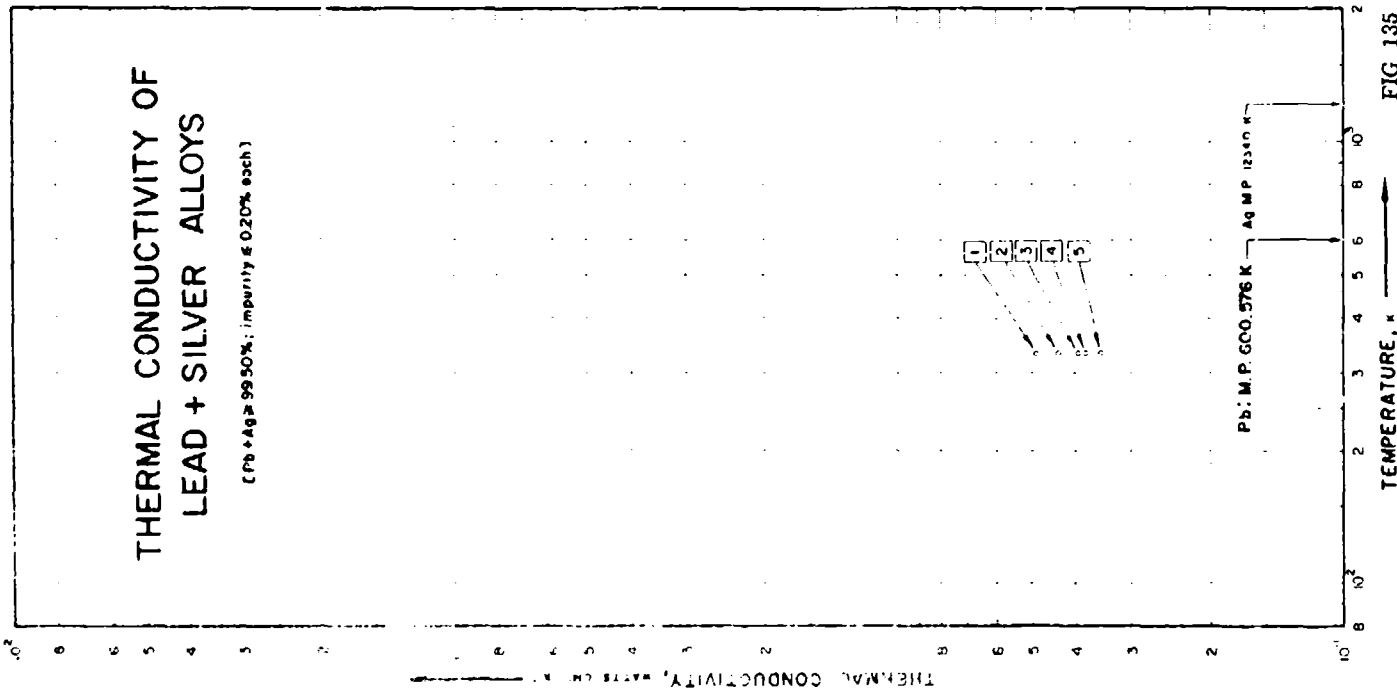


FIG 135

SPECIFICATION TABLE NO. 135 THERMAL CONDUCTIVITY OF LEAD-SILVER ALLOYS

(Pb - Ag - 99.50%; impurities - 0.20% each)

For Data Reported in Figure and Table No. 135

Curve No.	Ref. No. Used	Method	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)	Composition (atom %)	Composition (contaminated), Specifications and Remarks
1	230	L	1925	333			50	50	Prepared by fusing Pb (0.02 impurities, supplied by Baker) and Ag (99.9 pure); specimen 3.5 cm long, 0.3 cm <sup>2</sup> cross-sectional area; electrical conductivity $6.15 \times 10^5 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 22 C.
2	230	L	1925	333			60	40	Similar to the above specimen except electrical conductivity $6.21 \times 10^5 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 22 C.
3	230	L	1925	333			70	30	Similar to the above specimen except electrical conductivity $4.95 \times 10^5 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 22 C.
4	230	L	1925	333			80	20	Similar to the above specimen except electrical conductivity $1.88 \times 10^5 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 22 C.
5	230	L	1925	333			90	10	Similar to the above specimen except electrical conductivity $4.57 \times 10^5 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 22 C.

## DATA TABLE NO. 135 THERMAL CONDUCTIVITY OF LEAD + SILVER ALLOYS

(Pb + Ag 99.50%; impurity = 0.20% each)

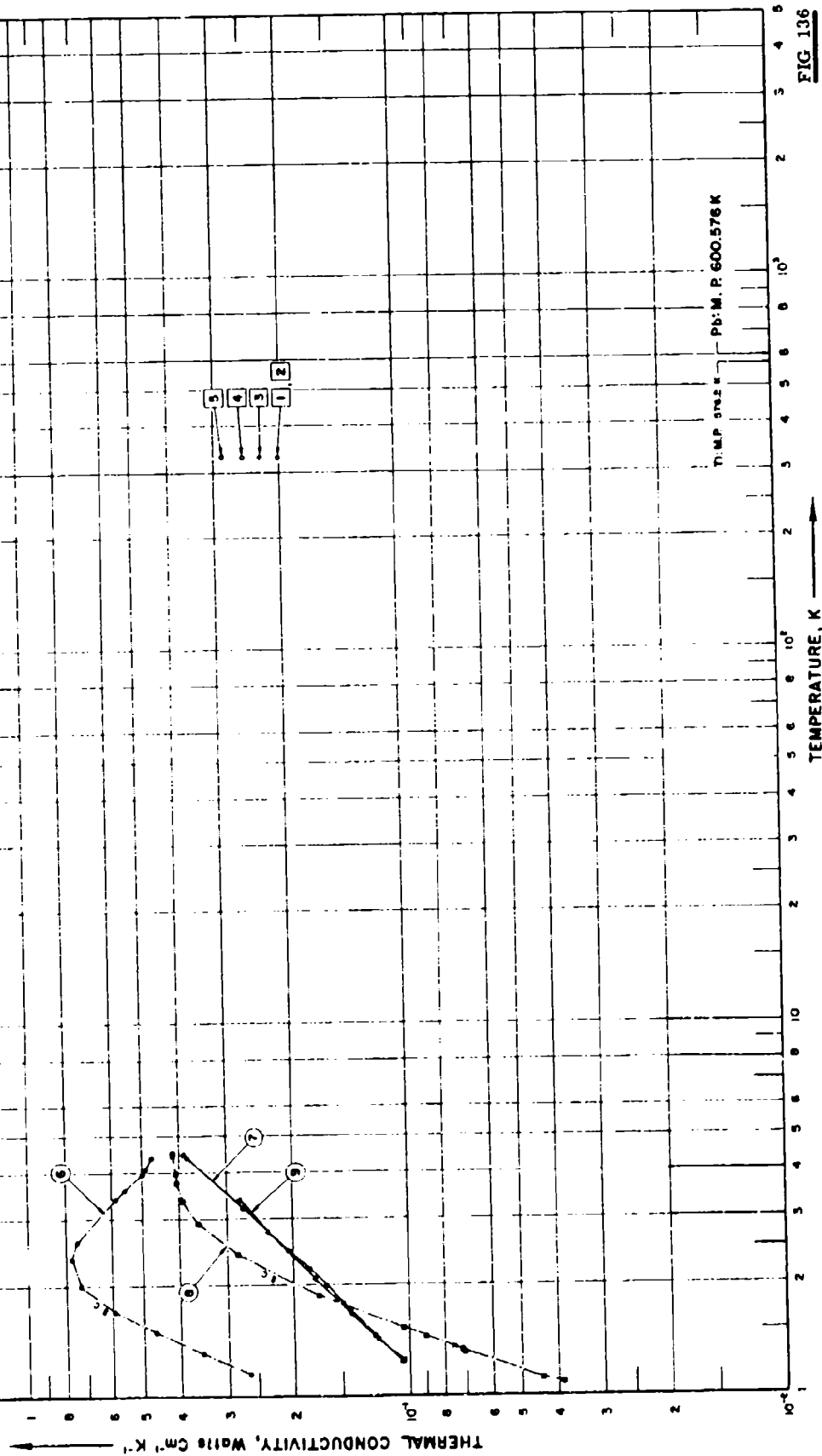
[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup>K<sup>-1</sup>]

T	k
<u>CURVE 1</u>	
333	0.490
<u>CURVE 2</u>	
333	0.439
<u>CURVE 3</u>	
333	0.395
<u>CURVE 4</u>	
333	0.381
<u>CURVE 5</u>	
333	0.351



# THERMAL CONDUCTIVITY OF LEAD + THALLIUM ALLOYS

[Pb+Tl] ≥ 99.50%, impurity ≤ 0.20% each



## SPECIFICATION TABLE NO. 136 THERMAL CONDUCTIVITY OF (LEAD-THALLIUM) ALLOYS

(Pb + Tl - 99.50%; impurity - 0.20% each)

(For Data Reported in Figure and Table No. 136)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)		Composition (continued), Specifications and Remarks
							Pb	Tl	
1	230	L	1925	303			50	50	Prepared by fusing Pb (0.03 impurities, supplied by Baker) and Tl (technically pure, supplied by Emmer and Amend); specimen ~5.5 cm long, 0.3 cm <sup>2</sup> cross sectional area; electrical conductivity at 25 C $2.54 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$ .
2	230	L	1925	303			60	40	Similar to the above specimen except electrical conductivity $2.62 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 25 C.
3	230	L	1925	303			70	30	Similar to the above specimen except electrical conductivity $2.74 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 25 C.
4	230	L	1925	303			80	20	Similar to the above specimen except electrical conductivity $2.98 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 25 C.
5	230	L	1925	303			90	10	Similar to the above specimen except electrical conductivity $3.54 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 25 C.
6	379, 257	L	1956	1.2-4.4	± 2	PbTl 0.6	99.428	0.572	Polycrystal (long crystals); annealed in vacuo close to melting point for several hrs; in superconducting state.
7	379, 257	L	1956	1.2-4.5	± 2	PbTl 0.6	99.428	0.572	The above specimen measured in a magnetic field of 1000 gauss, in normal state.
8	379, 257	L	1956	1.1-4.5	± 2	PbTl 0.6	99.428	0.572	Similar to the above specimen except strained; in superconducting state.
9	379, 257	L	1956	1.3-4.4	± 2	PbTl 0.6	99.428	0.572	The above specimen measured in a magnetic field of 1000 gauss, in normal state.

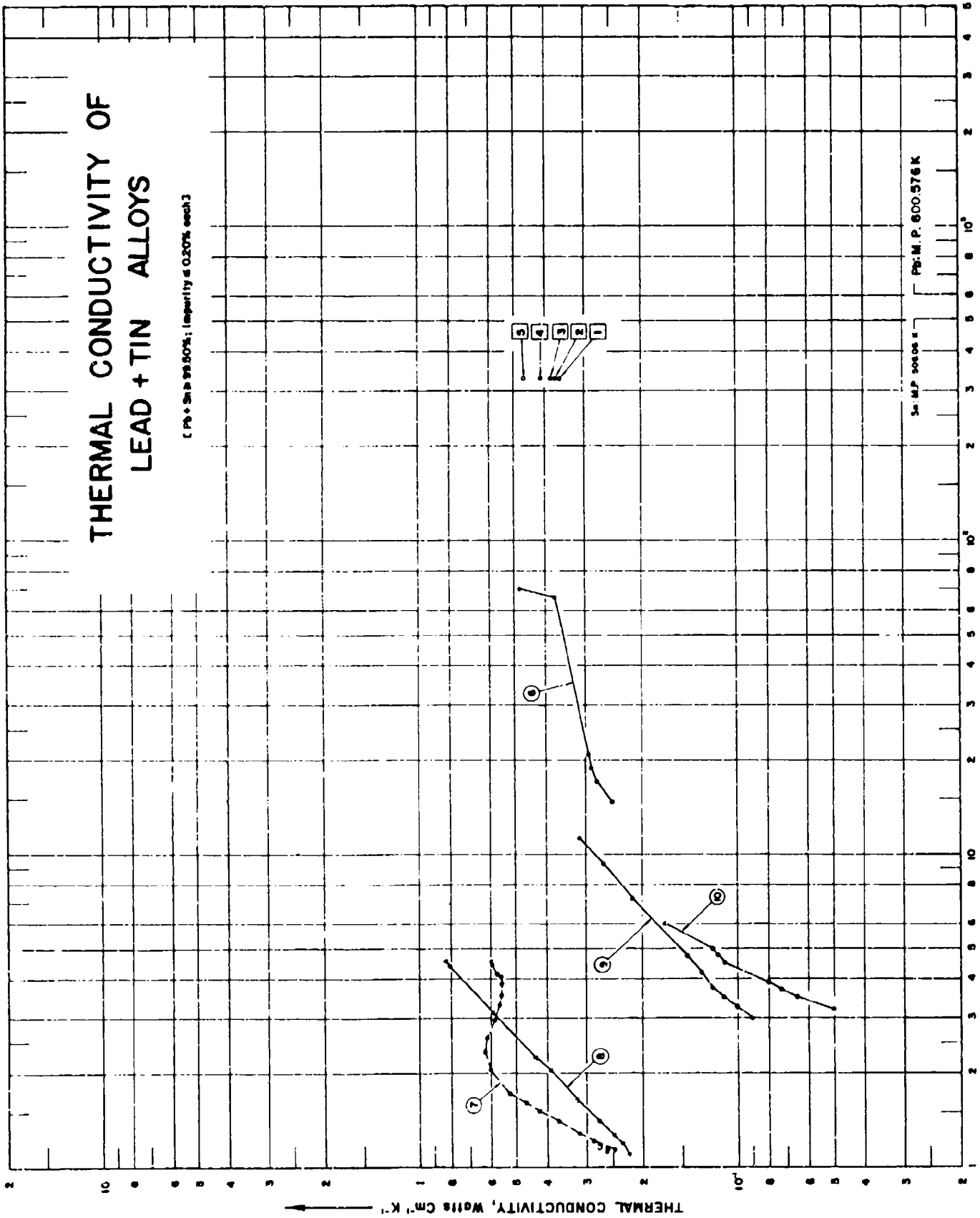
DATA TABLE NO. 136 THERMAL CONDUCTIVITY OF [LEAD + THALLIUM] ALLOYS

(Pb + Tl > 99.50%, impurity  $\leq$  0.20% each)[Temperature, T, K; Thermal Conductivity, k, Watts  $\text{cm}^{-1}\text{K}^{-1}$ ]

T	k	T	k
<u>CURVE 1</u>			
333	0.201	1.08	0.0389
		1.12	0.0440
<u>CURVE 2</u>			
		1.32	0.0707
		1.34	0.0715
333	0.201	1.36	0.0754
		1.45	0.0900
<u>CURVE 3</u>			
		1.53	0.1025
		1.86	0.172
333	0.226	2.42	0.282
		2.93	0.356
<u>CURVE 4</u>			
		3.36	0.390
		3.77	0.403
333	0.251	4.00	0.407
		4.48	0.416
<u>CURVE 5</u>			
		4.48	0.420
<u>CURVE 6</u>			
333	0.263	1.25	0.1038
		1.42	0.118
		1.67	0.141
		2.08	0.176
		2.77	0.233
		4.38	0.377
<u>CURVE 7</u>			
		1.15	0.264
		1.31	0.349
		1.49	0.468
		1.70	0.593
		1.99	0.735
		2.36	0.778
		2.62	0.750
		3.41	0.586
		3.60	0.553
		3.97	0.499
		4.11	0.490
		4.40	0.471
<u>CURVE 8</u>			
		1.24	0.104
		1.46	0.121
		1.71	0.142
		1.78	0.149
		1.98	0.164
		2.31	0.182
		2.47	0.205
		3.22	0.273
		4.40	0.380
		4.49	0.388

# THERMAL CONDUCTIVITY OF LEAD + TIN ALLOYS

(Pb + Sn is 99.80%; impurity  $\leq$  0.20% each)



Sn: MP 504.08 K Pb: M.P. 600.576 K

TEMPERATURE, K

SPECIFICATION TABLE NO. 137 THERMAL CONDUCTIVITY OF [LEAD + TIN] ALLOYS

(Pb + Sn : 99.50%; impurity : 0.20% each)

[For Data Reported in Figure and Table No. 137]

Curve No.	Ref. No. Used	Method	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition Pb	Composition Sn	Composition (continued), Specifications and Remarks
1	230	L	1925	327.2			99	10	Approx. composition; <0.03 total impurity in each metal; specimen 1.9 cm in dia and 10 cm long; supplied by Baker; electrical conductivity $4.95 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 22 C.
2	230	L	1925	327.2			89	20	Similar to the above specimen except electrical conductivity $5.29 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 22 C.
3	230	L	1925	327.2			79	30	Similar to the above specimen except electrical conductivity $5.65 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 22 C.
4	230	L	1925	327.2			60	40	Similar to the above specimen except electrical conductivity $5.99 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 22 C.
5	230	L	1925	327.2			50	50	Similar to the above specimen except electrical conductivity $6.47 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 22 C.
6	228	L	1936	15-70			56.0	44.0	No other details reported.
7	257	L	1958	1.1-4.5	$\pm 2.0$	PbSn 0.5	99.73	0.27	79 (99.99 pure) supplied by Johnson Matthey; specimen 7 cm long, 3 mm in dia; polycrystal with grain size 0.1 mm; annealed in vacuo for several hrs at a temp a few degrees above the melting point; in superconducting state; $D_0 = 0.124 \mu \text{ ohm cm}$ .
8	257	L	1958	1.1-4.6	$\pm 2.0$	PbSn 0.5	99.73	0.27	The above specimen in normal state at 1000 gauss.
9	96	L	1950	3.0-11			70.0	30	Approx. composition; measured in normal state.
10	96	L	1950	3.2-6.0			70	30	Approx. composition; measured in superconducting state.

## DATA TABLE NO. 137 THERMAL CONDUCTIVITY OF [LEAD + TIN] ALLOYS

(Pb + Sn &gt; 99.50% impurity ≤ 0.20% each)

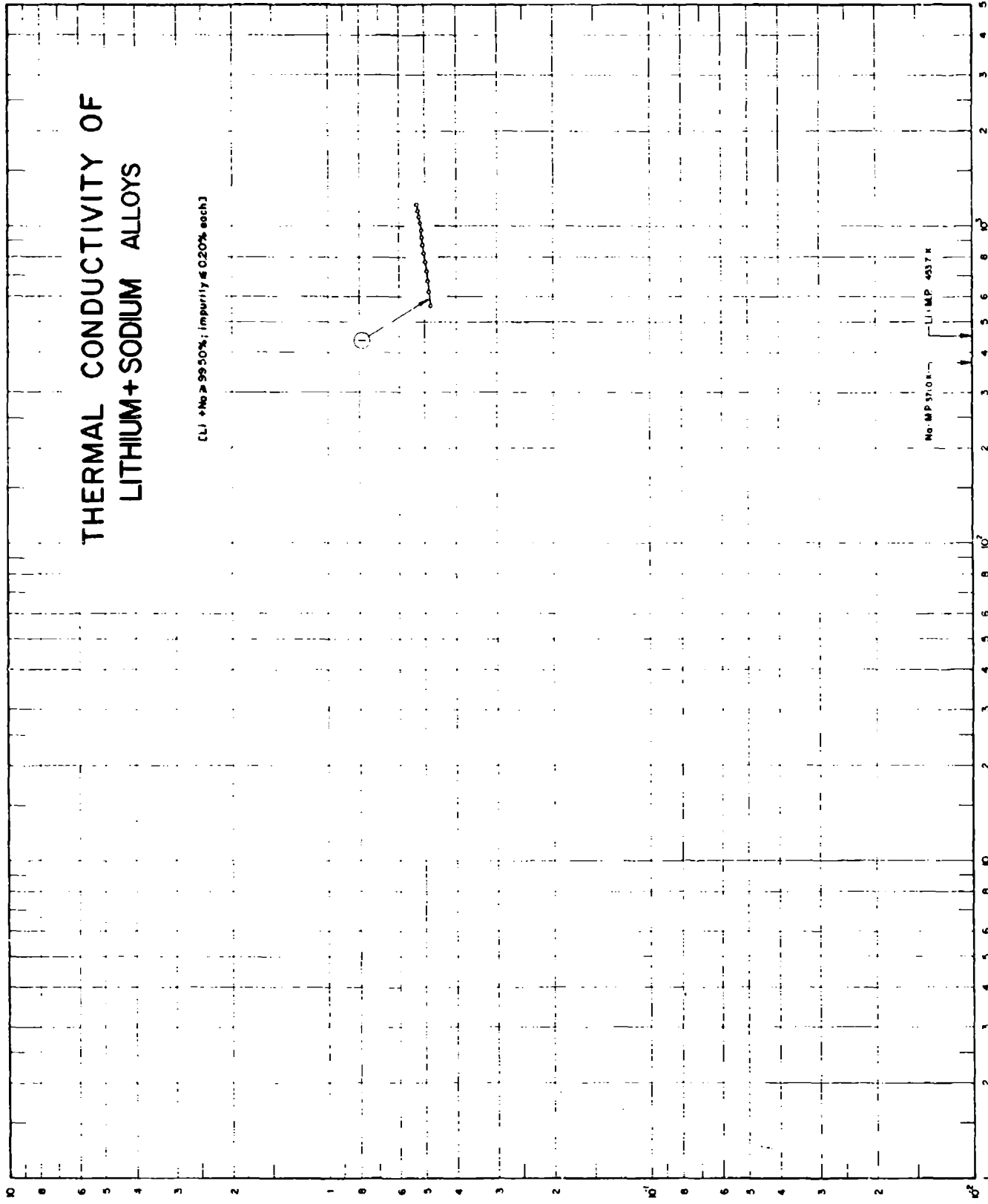
[Temperature, T. K; Thermal Conductivity, k, Watts cm<sup>-1</sup> K<sup>-1</sup>]

T	k	T	k
<u>CURVE 1</u>			
327	0.380	<u>CURVE 7 (cont.)</u>	
		4.05	0.558
		4.14	0.575
<u>CURVE 2</u>			
		4.42	0.595
		4.53	0.596
327	0.372	<u>CURVE 8</u>	
		1.10	0.222
		1.19	0.233
		1.27	0.248
		1.40	0.275
		1.64	0.322
		2.04	0.392
		2.23	0.436
		3.10	0.595
		4.38	0.810
		4.56	0.835
327	0.464	<u>CURVE 9</u>	
		3.0	0.090
		3.2	0.100
		3.5	0.110
		3.7	0.120
		4.2	0.130
		4.7	0.145
		7.2	0.215
		9.3	0.265
		11.3	0.315
		<u>CURVE 10</u>	
		3.2	0.050
		3.5	0.065
		3.7	0.072
		3.9	0.080
		4.5	0.110
		4.8	0.115
		5.0	0.120
		6.0	0.170
<u>CURVE 3</u>			
327	0.385	<u>CURVE 9</u>	
		14.7	0.249
		17.1	0.276
		18.9	0.287
		20.8	0.294
		66.0	0.377
		76.0	0.483
<u>CURVE 4</u>			
327	0.414	<u>CURVE 7</u>	
		1.14	0.246
		1.22	0.286
		1.28	0.316
		1.40	0.368
		1.51	0.424
		1.61	0.469
		1.72	0.525
		2.05	0.602
		2.33	0.630
		2.59	0.620
		2.95	0.584
		3.30	0.562
		3.55	0.550
		3.87	0.553

# THERMAL CONDUCTIVITY OF LITHIUM + SODIUM ALLOYS

[Li] + Na ≥ 99.50%; impurity ≤ 0.20% each.]

THERMAL CONDUCTIVITY,  $\text{Watt cm}^{-1} \text{K}^{-1}$



TEMPERATURE, K

## SPECIFICATION TABLE NO. 138 THERMAL CONDUCTIVITY OF LITHIUM - SODIUM ALLOYS

(Li + Na = 99.50%; Impurity = 0.20%)

[ For Data Reported in Figure and Table No. 138 ]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)		Composition (continued), Specifications and Remarks
							Li	Na	
1	922, 923	C	1965	563-1193	±8	Bal.	9.26	0.0072 N, 0.0063 Ca, 0.0011 K, 0.96 other impurities; Specimen tilted in a type 1Kb 18 NBT steel container 14-15 mm in dia and 220 mm long; Armo iron used as comparative material; thermal conductivity data calculated from smooth curve.	



## DATA TABLE NO. 138 THERMAL CONDUCTIVITY OF [LITHIUM + SODIUM] ALLOYS

(Li + Na = 99.50%; Impurity &lt; 0.20% each)

[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup>K<sup>-1</sup>]

T	k
<u>CURVE 1</u>	
563.2	0.480
573.2	0.481*
623.2	0.485
673.2	0.490
723.2	0.493
773.2	0.497
823.2	0.501
873.2	0.506
923.2	0.510
973.2	0.514
1023	0.518
1073	0.522
1123	0.526
1173	0.530
1193	0.532*

\* Not shown on plot

# THERMAL CONDUCTIVITY OF MAGNESIUM + ALUMINUM ALLOYS

(Mg+Al = 99.50%; impurity ≤ 0.20% each)

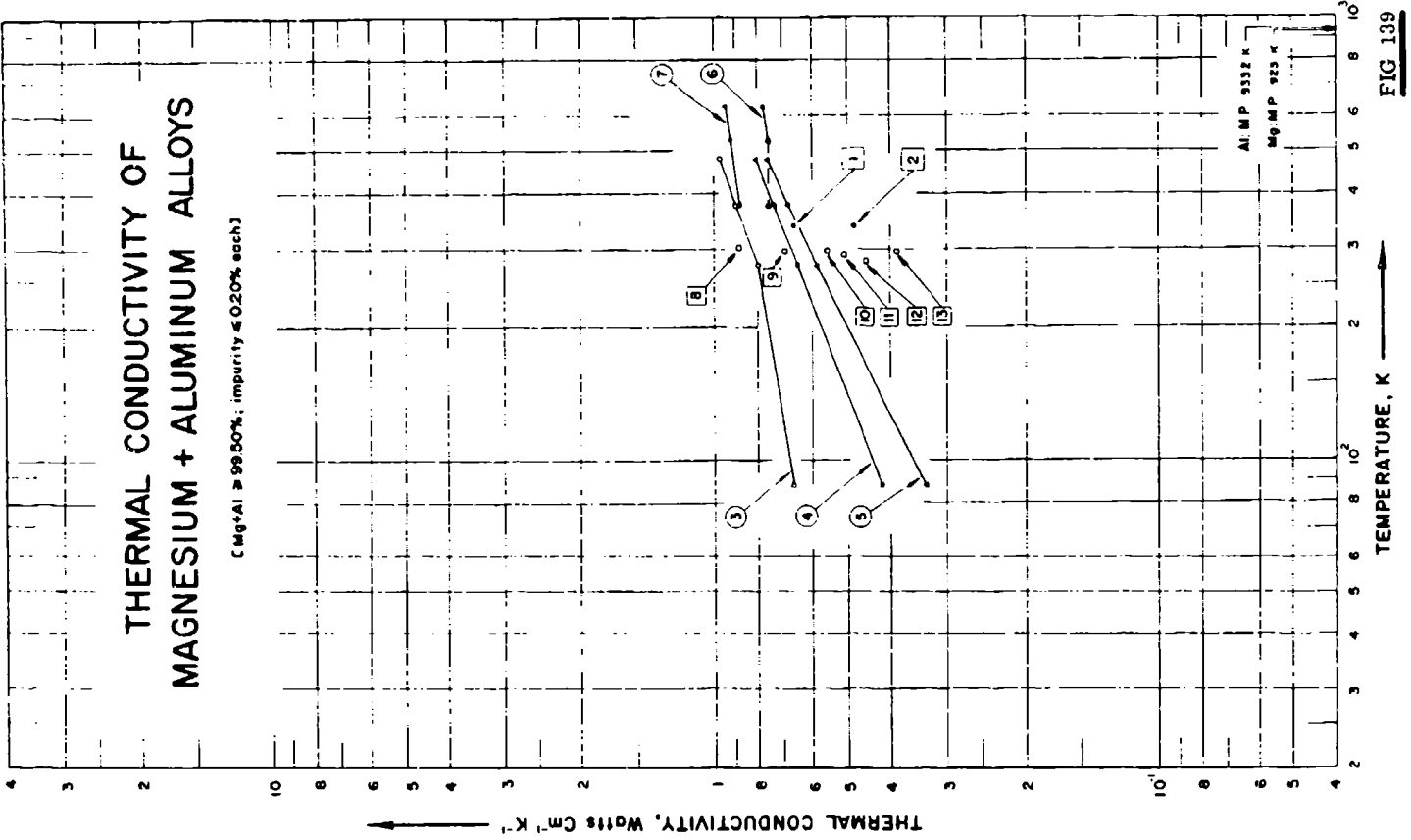


FIG 139

## SPECIFICATION TABLE NO. 139 THERMAL CONDUCTIVITY OF [MAGNESIUM + ALUMINIUM] ALLOYS

(Mg + Al - 99.50%; impurity &lt; 0.50% each)

For Data Reported in Figure and Table No. 139.

Curve No.	Ref. No. Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) Mg Al	Composition (continued), Specifications and Remarks
1	270	L	336.2			95.82 4.12	0.028 Fe, 0.019 Si; specimen ~5 cm long with cross-section 0.3 cm <sup>2</sup> ; supplied by Aluminum Co. of America; electrical conductivity $9.06 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 63 C.
2	230	L	536.2			89.82 10.12	0.023 Si, 0.028 Fe; similar to the above specimen except electrical conductivity $6.00 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 63 C.
3	830, 93	L	87-476	3-4		94.0 6.0	Specimen 1.23 cm <sup>2</sup> in cross section and 3 cm long; cast; electrical conductivity 14.7, 8.04, 6.47 and $5.99 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 87, 273, 373 and 476 K, respectively.
4	850, 93	L	87-476	3-4		92.0 8.0	Similar to the above specimen except electrical conductivity 13.32, 7.31, 5.95 and $5.55 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 87, 273, 373 and 476 K, respectively.
5	850, 93	L	87-476	3-4		88 12	Similar to the above specimen except electrical conductivity 9.65, 5.99, 5.27 and $4.90 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 87, 273, 373 and 476 K, respectively.
6	225	L	373-623			94.0 6	Specimen 12 in long and 1 in. in dia; annealed at 300 C for 3 hrs.
7	225	L	373-623			89 11	Similar to the above specimen.
8	673	E	300.2	1.3		97.9 2.1	Specimen 200 mm long; electrical conductivity $11.9 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 27 C.
9	673	E	295.5	1.3		95.8 4.2	Specimen 200 mm long; electrical conductivity $6.9 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 22.3 C.
10	673	E	295.1	1.3		93.8 6.2	Specimen 200 mm long; electrical conductivity $6.9 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 21.9 C.
11	673	E	291.5	1.3		91.8 8.2	Specimen 200 mm long; electrical conductivity $5.9 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 18.3 C.
12	673	E	281.5	1.3		89.7 10.3	Specimen 200 mm long; electrical conductivity $5.5 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 19.3 C.
13	673	E	296.5	1.3		87.8 12.2	Specimen 200 mm long; electrical conductivity $5.1 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 23.1 C.

## DATA TABLE NO. 139 THERMAL CONDUCTIVITY OF (MAGNESIUM + ALUMINUM) ALLOYS

(Mg + Al) : 99.50% ; impurity  $\leq$  0.20% each )(Temperature, T, K; Thermal Conductivity, k, Watts  $\text{cm}^{-1}\text{K}^{-1}$ )

T	k	T	k
<u>CURVE 1</u>			
336.0	0.665	300.2	0.597
<u>CURVE 2</u>			
336.0	0.485	295.5	0.690
<u>CURVE 3</u>			
87.0	0.603	295.1	0.556
273.0	0.799	<u>CURVE 11</u>	
373.0	0.895	291.5	0.510
476.0	0.971	<u>CURVE 12</u>	
<u>CURVE 4</u>			
87.0	0.419	231.5	0.452
273.0	0.649	<u>CURVE 13</u>	
373.0	0.736	296.3	0.385
476.0	0.808	<u>CURVE 5</u>	
<u>CURVE 5</u>			
87.0	0.335	373.2	0.753
273.0	0.546	523.2	0.753
373.0	0.682	523.2	0.774
476.0	0.761	<u>CURVE 6</u>	
<u>CURVE 6</u>			
373.2	0.753	373.2	0.879
523.2	0.753	523.2	0.920
523.2	0.774	623.2	0.941
<u>CURVE 7</u>			
<u>CURVE 7</u>			

## SPECIFICATION TABLE NO. 140 THERMAL CONDUCTIVITY OF [MAGNESIUM + CADMIUM] ALLOYS

(Mg + Cd - 99.50%; impurity  $\leq$  0.20% each)

Curve No.	Rel. Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) Mg Cd	Composition (continued), Specifications and Remarks
1	93, 1, 450	1929	87-476	3-4		92.0 8.0	Specimen 1.23 cm <sup>2</sup> in cross section and 3 cm long; as cast; electrical conductivity 85.0, 18.43, 14.85 and 11.33 x 10 <sup>4</sup> ohm <sup>-1</sup> cm <sup>-1</sup> at 87, 273, 373 and 476 K respectively.

## DATA TABLE NO. 140 THERMAL CONDUCTIVITY OF [MAGNESIUM + CADMIUM] ALLOYS

(Mg + Cd - 99.50%; impurity  $\leq$  0.20% each)(Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup>K<sup>-1</sup>)

T	k
<u>CURVE 1</u>	
87.00	1.301
273.00	1.414
373.00	1.477
476.00	1.544

No graphical presentation

SPECIFICATION TABLE NO. 141 THERMAL CONDUCTIVITY OF (MAGNESIUM + CALCIUM) ALLOYS  
(Mg + Ca - 99.5% impurity  $\leq 0.26\%$  each)

Curve No.	Ref. Method No.	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) Mg	Composition (weight percent) Ca	Composition (continued), Specifications and Remarks
1	397	L	1979	323-523	±3	W 1641	2.82	0.07 impurities; specimen ~ 29 cm long and 1.4 cm in dia; forged at elevated temperature; density 1.74 g cm <sup>-3</sup> ; electrical resistivity 4.65, 5.35, 7.05 and 8.7 $\mu\text{ohm cm}$ at 20, 50, 150 and 250 C respectively.

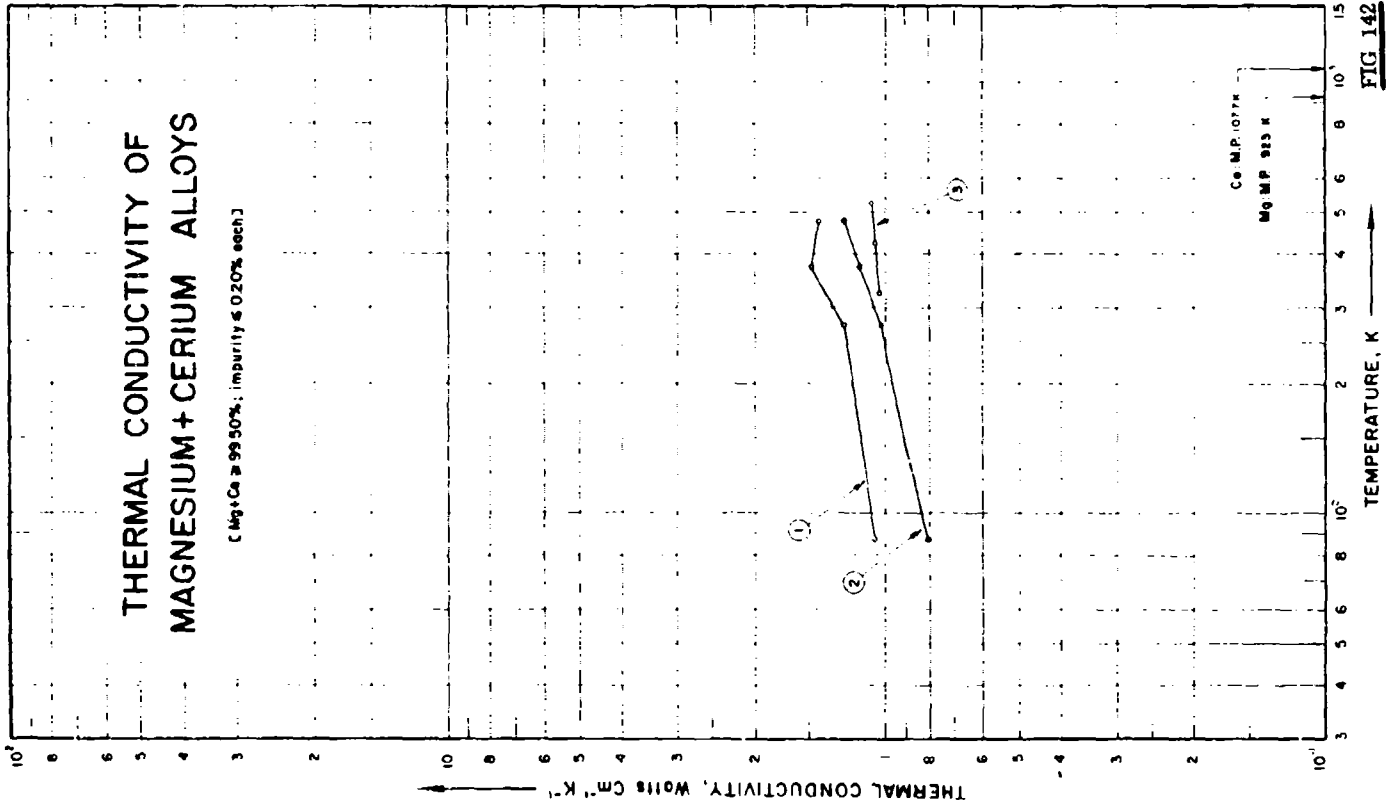
DATA TABLE NO. 141 THERMAL CONDUCTIVITY OF (MAGNESIUM + CALCIUM) ALLOYS

(Mg + Ca - 99.50% impurity  $\leq 0.20\%$  each)

[ Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup>K<sup>-1</sup> ]

T	k
	<u>CURVE 1<sup>a</sup></u>
323.2	1.402
423.2	1.423
523.2	1.423

<sup>a</sup> No graphical presentation



## SPECIFICATION TABLE NO. 142 THERMAL CONDUCTIVITY OF [MAGNESIUM + CERIUM] ALLOYS

(Mg + Ce - 99.50% impurity 0.20% each)

[For Data Reported in Figure and Table No. 142.]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) Mg	Composition (weight percent) Ce	Composition (continued), Specifications and Remarks
1	850, 93	I.	1929	87-476	3-4		92.0	8.0	Specimen 1.23 cm <sup>2</sup> in cross section and 3 cm long, as cast; electrical conductivity 74, 02, 14, 0, 10, 07 and 8, 22 x 10 <sup>4</sup> ohm <sup>-1</sup> cm <sup>-1</sup> at 87, 273, 373 and 476 K, respectively.
2	850, 93	I.	1929	87-476	3-4		88.0	12.0	Similar to the above specimen except electrical conductivity 45, 3, 11, 5, 8, 19 and 6, 32 x 10 <sup>4</sup> ohm <sup>-1</sup> cm <sup>-1</sup> at 87, 273, 373 and 476 K, respectively.
3	397	L	1939	323-523	3	W 1630	89.84	10.1	Specimen ~29 cm long and 1.4 cm in dia; 0.06 impurities, forged at elevated temperature; density 1.86 g cm <sup>-3</sup> ; Ce in the form of mischmetal; electrical resistivity 6, 7, 7, 4, 9, 7, and 11, 9 μohm cm at 20, 50, 150 and 250 C, respectively.



## DATA TABLE NO. 142 THERMAL CONDUCTIVITY OF [MAGNESIUM + CERIUM] ALLOYS

(Mg + Ce : 99.56%, impurity : 0.20% each)

[Temperature, T, K; Thermal Conductivity,  $k$ , Watt  $\text{cm}^{-1} \text{K}^{-1}$ ]

T	k
<u>CURVE 1</u>	
87.0	1.059
273.0	1.251
373.0	1.473
476.0	1.423
<u>CURVE 2</u>	
87.0	0.804
273.0	1.029
373.0	1.159
476.0	1.255
<u>CURVE 3</u>	
323.2	1.046
423.2	1.067
523.2	1.088

# THERMAL CONDUCTIVITY OF MAGNESIUM + COPPER ALLOYS

(Mg + Cu ≥ 99.90%; impurity ≤ 0.20% each)

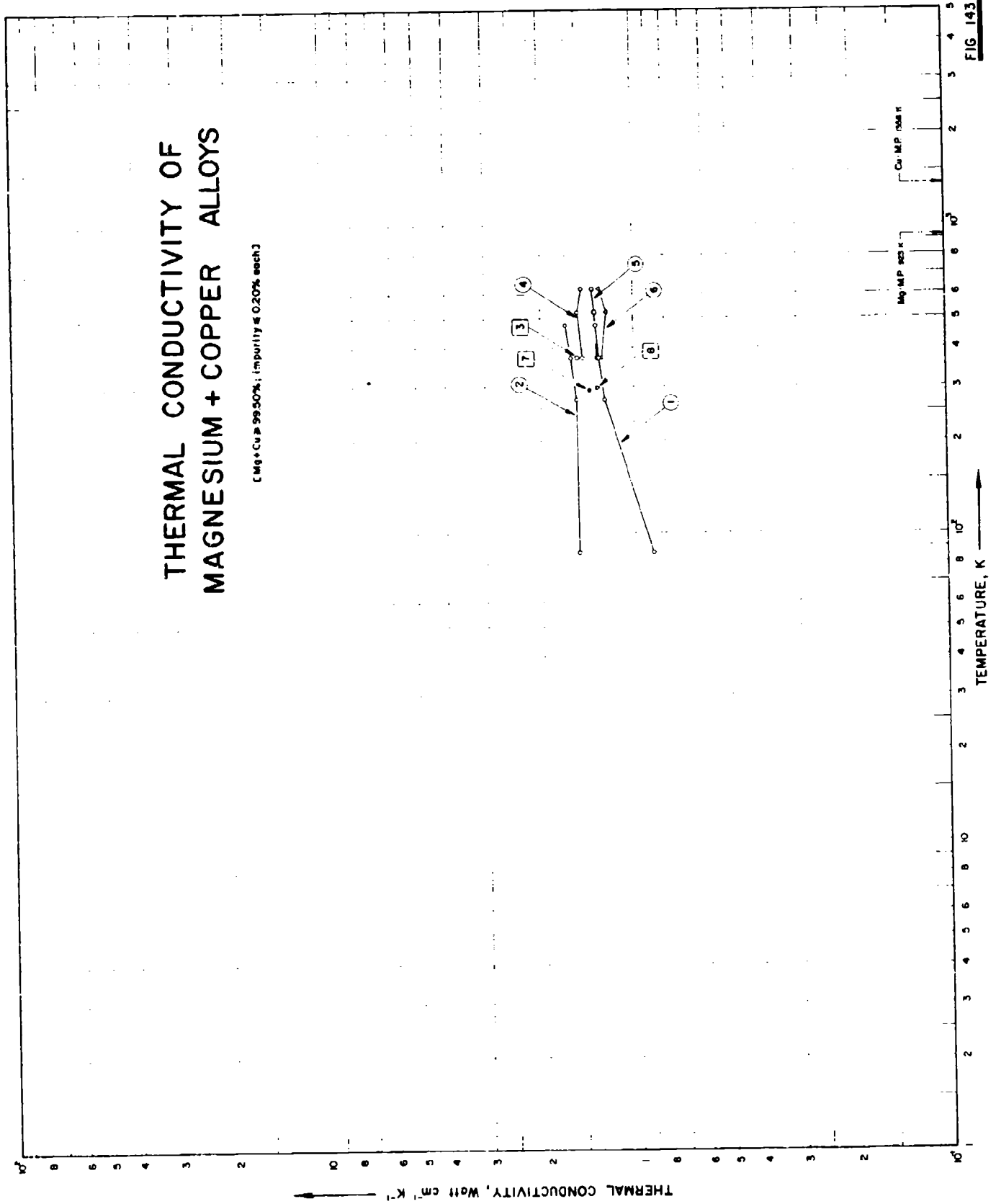


FIG 143

SPIFICATION TABLE NO. 13 THERMAL CONDUCTIVITY OF MAGNESIUM-COPPER ALLOYS  
(Mg-Cu 99.50% purity, 0.20% each)

For Data Reported in Figure and Table No. 143.

Curve No.	Rel. Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) Mg	Composition (weight percent) Cu	Composition (continued), Specifications and Remarks
1	S50 93	1929	87-476	3-4		92.0	8.0	Specimen 1: 24 cm <sup>2</sup> in cross-section and 3 cm long; cast; electrical conductivity 43.0, 17.57, 14.95 and 10.7 x 10 <sup>6</sup> ohm <sup>-1</sup> cm <sup>-1</sup> at 87, 273, 373 and 476 K, respectively.
2	S50 93	1929	87-476	3-4		85.0	15.0	Similar to the above specimen except electrical conductivity 92.6, 20.6, 17.5 and 12.5 x 10 <sup>6</sup> ohm <sup>-1</sup> cm <sup>-1</sup> at 87, 273, 373 and 476 K, respectively.
3	S3	1927	373	1.0		96.0	4.0	Forged.
4	225	1928	373-623			97.0	3.0	Specimen 12 in. long and 1 in. in dia; annealed at 300 C for 3 hrs.
5	225	1928	373-623			8.0	14.0	Similar to the above specimen.
6	225	1928	373-623			91.0	1.0	Similar to the above specimen.
7	673	1932	293.5	1.3		97.0	2.4	Specimen 200 mm long; electrical conductivity 22.0 x 10 <sup>6</sup> ohm <sup>-1</sup> cm <sup>-1</sup> at 20.3 C.
8	673	1932	297.4	1.3		93.5	6.5	Specimen 200 mm long; electrical conductivity 20.8 x 10 <sup>6</sup> ohm <sup>-1</sup> cm <sup>-1</sup> at 24.2 C.

## DATA TABLE NO. 143 THERMAL CONDUCTIVITY OF MAGNESIUM-COPPER ALLOYS

(Mg-Cu 99.50% purity 0.20% each)

Temperature, T, K; Thermal Conductivity,  $k$ , Watt cm<sup>-1</sup> K<sup>-1</sup>.

T	k
<u>CURVE 1</u>	
87	0.879
273	1.247
373	1.297
476	1.326
<u>CURVE 2</u>	
87	1.506
273	1.544
373	1.639
476	1.633
<u>CURVE 3</u>	
373	1.506
<u>CURVE 4</u>	
373.2	1.464
523.2	1.506
623.2	1.464
<u>CURVE 5</u>	
373.2	1.297
523.2	1.339
623.2	1.360
<u>CURVE 6</u>	
373.2	1.276
523.2	1.234
623.2	1.297
<u>CURVE 7</u>	
293.5	1.389
<u>CURVE 8</u>	
297.4	1.310

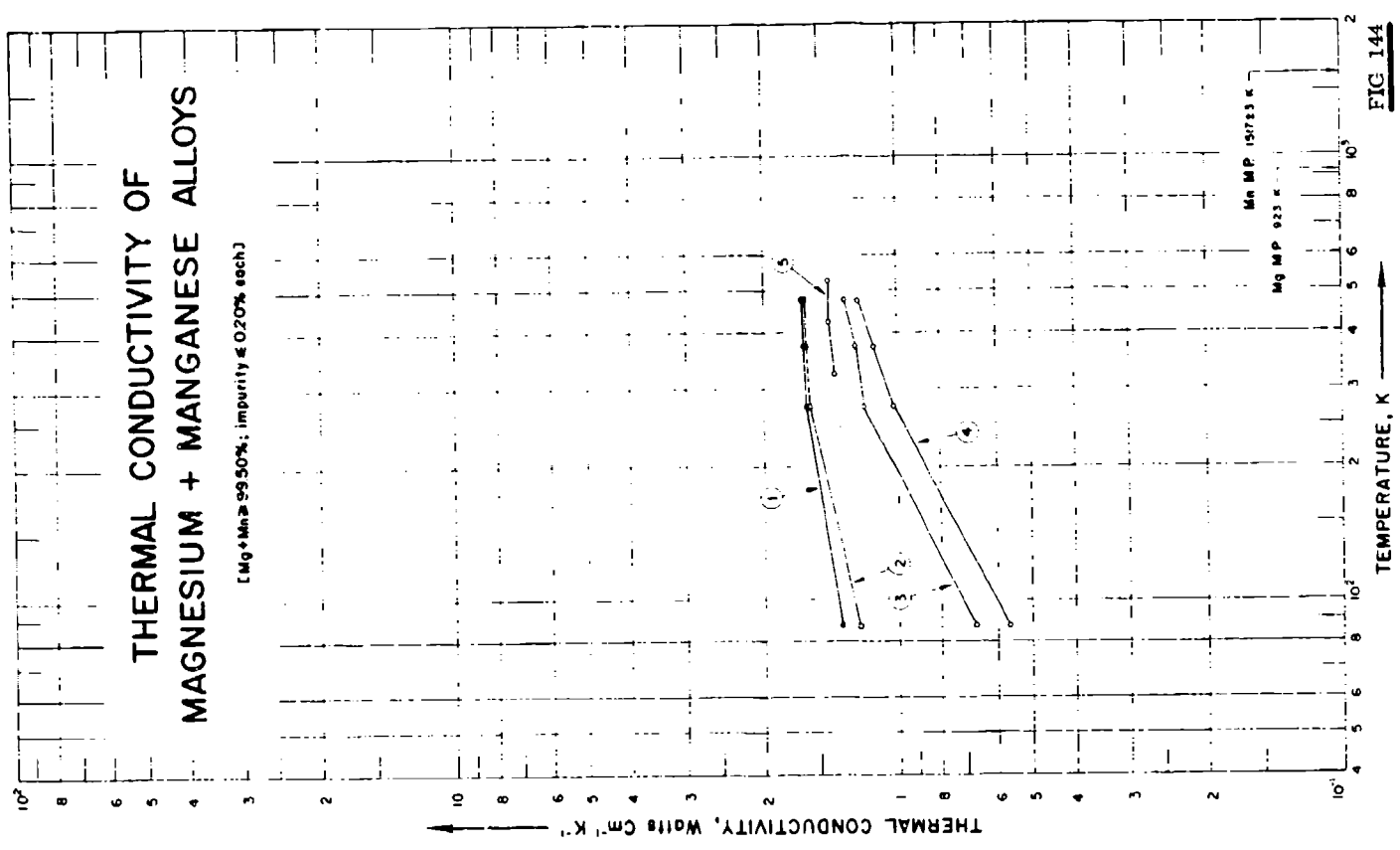


FIG 144

## SPECIFICATION TABLE NO. 144 THERMAL CONDUCTIVITY OF (MAGNESIUM - MANGANESE) ALLOYS

(Mg + Mn - 99.50%; impurity - 0.20% each)

[ For Data Reported in Figure and Table No. 144 ]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) Mg	Mn	Composition (continued), Specifications and Remarks
1	850, 93	L	1929	87-476	3-4		99.5	0.5	Specimen 1.23 cm <sup>2</sup> in cross-section and 1.4 cm long; cast; electrical conductivity 93, 90, 19, 05, 13, 26 and 10.6 x 10 <sup>4</sup> ohm <sup>-1</sup> cm <sup>-1</sup> at 87, 273, 373 and 476 K, respectively.
2	850, 93	L	1929	87-476	3-4		99.2	0.8	Specimen 1.23 cm <sup>2</sup> in cross-section and 3 cm long; cast; electrical conductivity 64, 46, 17, 88, 12, 30 and 10.0 x 10 <sup>4</sup> ohm <sup>-1</sup> cm <sup>-1</sup> at 87, 273, 373 and 476 K, respectively.
3	850, 93	L	1929	87-476	3-4		95.0	2.0	Similar to the above specimen except electrical conductivity 38, 50, 14, 37, 11.0 and 8.70 x 10 <sup>4</sup> ohm <sup>-1</sup> cm <sup>-1</sup> at 87, 273, 373 and 476 K, respectively.
4	850, 93	L	1929	87-476	3-4		96.46	3.54	Similar to the above specimen except electrical conductivity 29.5, 12, 66, 9, 8 and 4.0 x 10 <sup>4</sup> ohm <sup>-1</sup> cm <sup>-1</sup> at 87, 273, 373 and 476 K, respectively.
5	397	L	1939	323-523	3.0	W 1567	97.29	2.64	0.07 total impurities; specimen .29 cm long and 1.4 cm in dia; forged at elevated temperature; density 1.77 g cm <sup>-3</sup> ; electrical resistivity 4.9, 5.1, 7.15 and 8.9 uohm cm at 20, 50, 130 and 250 C, respectively.

## DATA TABLE NO. 144 THERMAL CONDUCTIVITY OF (MAGNESIUM + MANGANESE) ALLOYS

(Mg + Mn .99.50%; impurity .0.20% each)

[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup> K<sup>-1</sup>]

T	k
<u>CURVE 1</u>	
87	1.339
273	1.598
373	1.632
476	1.644
<u>CURVE 2</u>	
87	1.222
273	1.577
373	1.603
476	1.615
<u>CURVE 3</u>	
87	0.674
273	1.176
373	1.243
476	1.310
<u>CURVE 4</u>	
87	0.565
273	1.021
373	1.130
476	1.234
<u>CURVE 5</u>	
323.2	1.381
423.2	1.423
523.2	1.423

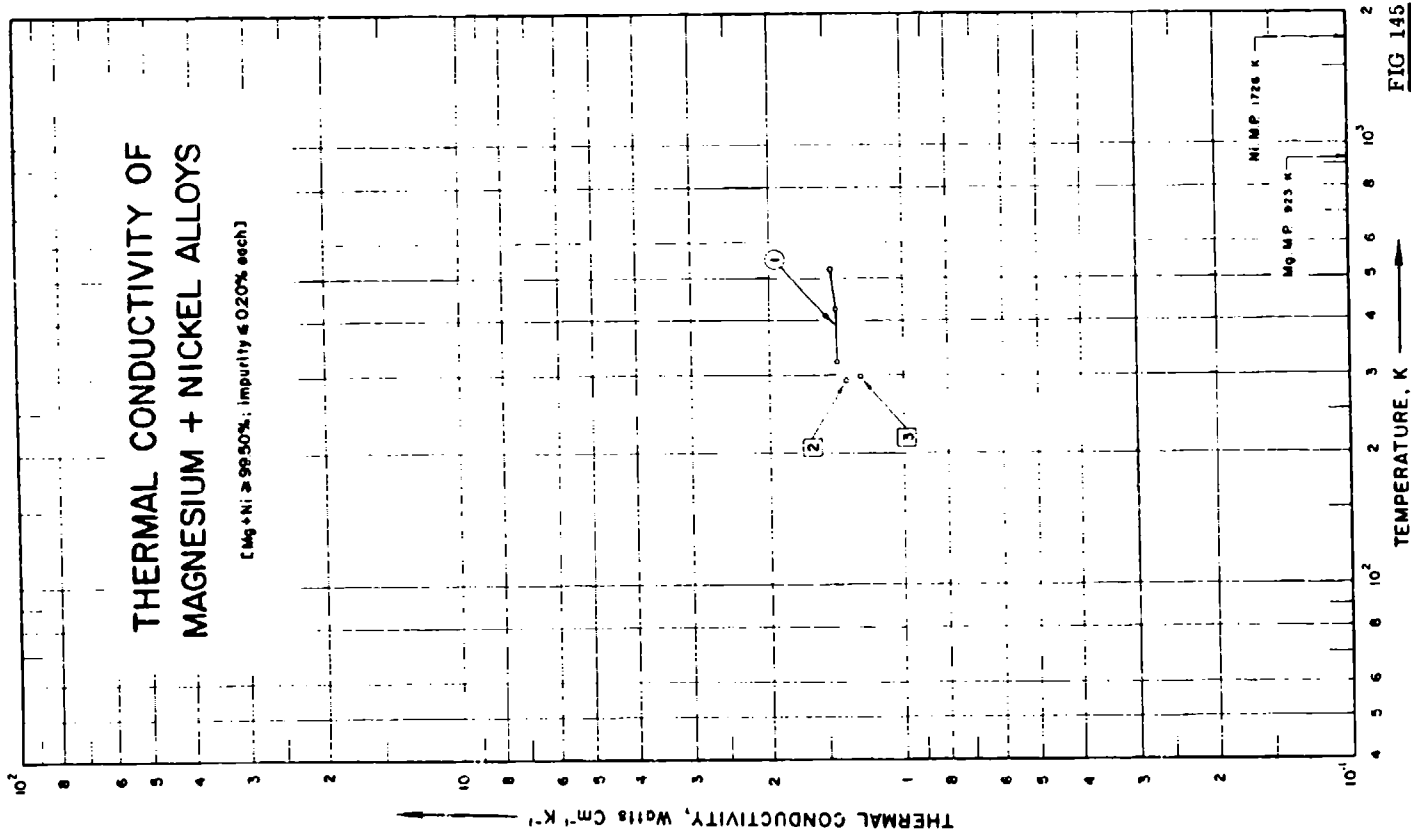


FIG 145



SPECIFICATION TABLE NO. 145 THERMAL CONDUCTIVITY OF (MAGNESIUM + NICKEL) ALLOYS

(Mg - Ni - 99.50%, impurity - 0.20% each)

[For Data Reported in Figure and Table No. 145]

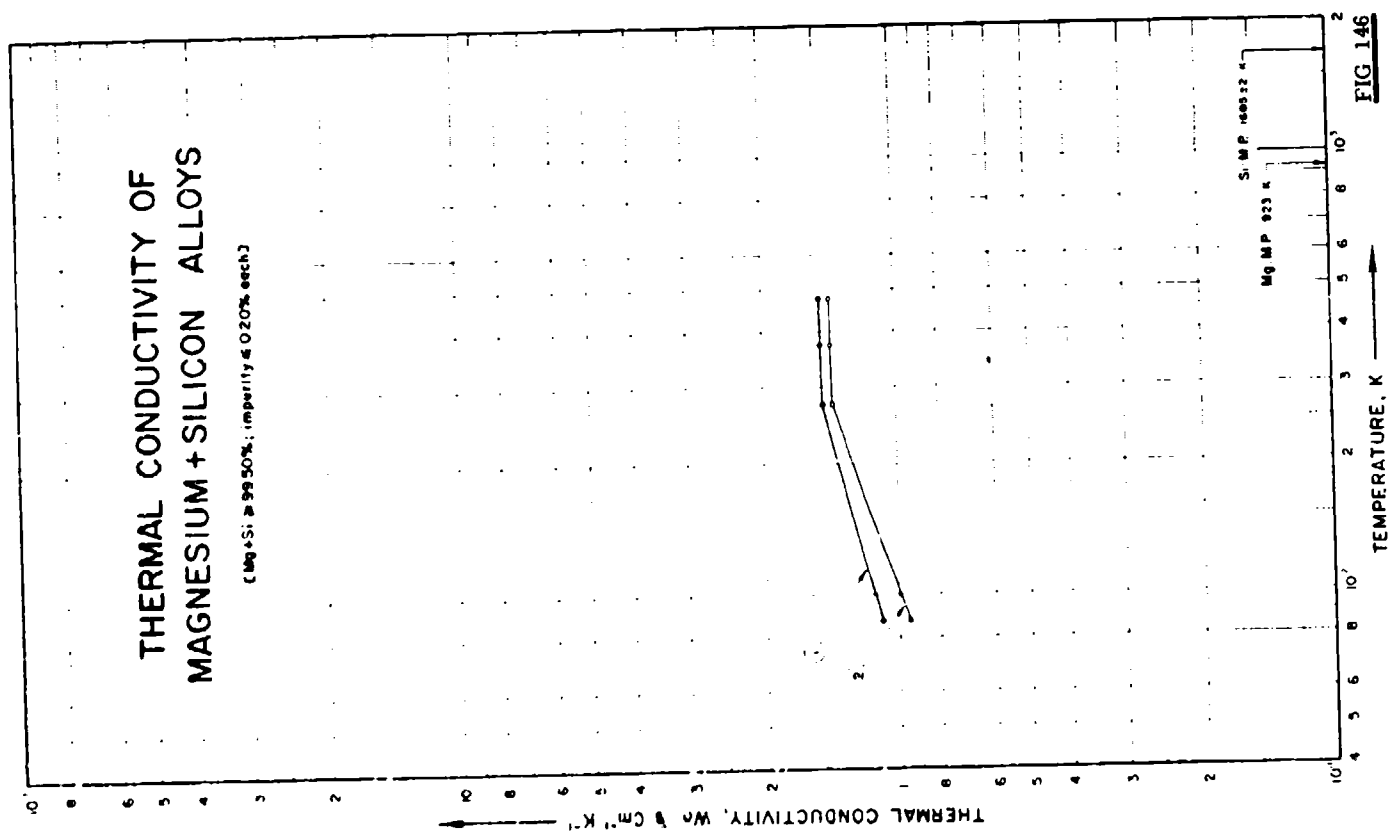
Curve No.	Rel. Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)		Composition (continued), Specifications and Remarks	
						Mg	Ni		
1	397	1	1939	323-323	± 3.0	W 1635	94.37	5.56	0.07 impurities; specimen ~29 cm long and 1.4 cm in dia; density 1.84 g cm <sup>-3</sup> ; electrical resistivity 4.7, 5.2, 7.0, and 8.75 μohm cm at 20, 50, 150 and 250 C, respectively.
2	673	F	1932	293.4	1.3		98.1	1.9	Specimen 200 mm long; electrical conductivity 21.4 x 10 <sup>4</sup> ohm <sup>-1</sup> cm <sup>-1</sup> at 20.2 C.
3	673	E	1932	297.7	1.3		94.2	5.8	Specimen 200 mm long; electrical conductivity 20.3 x 10 <sup>4</sup> ohm <sup>-1</sup> cm <sup>-1</sup> at 24.5 C.

## DATA TABLE NO. 117 THERMAL CONDUCTIVITY OF MAGNESIUM-NICKEL ALLOYS

(Mg-Ni alloys, 20% impurities, 0.20" each)

Temperature: T, K Thermal Conductivity:  $\lambda$ , W/cm<sup>2</sup>°K

1	2
<u>CURVE 1</u>	
2000	1.42
1500	1.42
1000	1.40
<u>CURVE 2</u>	
2000	1.36
<u>CURVE 3</u>	
2000	1.26



## SPECIFICATION TABLE NO. 146 THERMAL CONDUCTIVITY OF [MAGNESIUM + SILICON] ALLOYS

(Mg + Si) 99.50%, impurity  $\pm 0.20\%$  each

[For Data Reported in Figure and Table No. 146]

Curve No.	Ref. No. Used	Method	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) Mg	Si	Composition (continued), Specifications and Remarks
1	450, 93	L	1929	87-476	3-4		99.3	0.7	Specimen 1.23 cm <sup>2</sup> in cross-section and 3 cm long, cast, electrical conductivity 113.0, 18.76, 12.96 and 10.04 x 10 <sup>4</sup> ohm <sup>-1</sup> cm <sup>-1</sup> at 87, 273, 373 and 476 K, respectively.
2	450, 93	L	1929	87-476	3-4		98.5	1.5	Similar to the above specimen except electrical conductivity 109.0, 17.37, 11.72 and 8.93 x 10 <sup>4</sup> ohm <sup>-1</sup> cm <sup>-1</sup> at 87, 273, 373 and 476 K, respectively.

## DATA TABLE NO. 146 THERMAL CONDUCTIVITY OF [MAGNESIUM + SILICON] ALLOYS

(Mg + Si) = 99.50%, impurity = 0.20%, each)

[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup> K<sup>-1</sup>]

T	k
<u>CURVE 1</u>	
47	1.096
273	1.473
373	1.482
476	1.486
<u>CURVE 2</u>	
47	0.950
273	1.397
373	1.402
476	1.406

SPECIFICATION TABLE NO. 147 THERMAL CONDUCTIVITY OF (MAGNESIUM + SILVER) ALLOYS  
(Mg + Ag - 99.50%; impurity < 0.20% each)

Curve No.	Ref. Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) Mg	Ag	Composition (continued), Specifications and Remarks
1	673 F	1972	295.7	1.3		97.8	2.2	Specimen 200 mm long; electrical conductivity $19.2 \times 10^4$ ohm <sup>-1</sup> cm <sup>-1</sup> at 25.5 C.
2	673 F	1972	300.2	1.3		94.0	6.0	Specimen 200 mm long; electrical conductivity $17.3 \times 10^4$ ohm <sup>-1</sup> cm <sup>-1</sup> at 27 C.

DATA TABLE NO. 147 THERMAL CONDUCTIVITY OF (MAGNESIUM + SILVER) ALLOYS

(Mg + Ag - 99.50%; impurity < 0.20% each)

(Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup>K<sup>-1</sup>)

T k

CURVE 1

295.7 1.310

CURVE 2

300.2 1.155

No graphical presentation

SPECIFICATION TABLE NO. 14<sup>s</sup> THERMAL CONDUCTIVITY OF [MAGNESIUM + TIN] ALLOYS

(Mg + Sn = 99.50%; impurity &lt; 0.20% each)

Curve No.	Ref. Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) Mg	Composition (weight percent) Sn	Composition (continued), Specifications and Remarks
1	673 E	1932	294.2	1.3		97.8	2.2	Specimen 200 mm long; electrical conductivity $15.6 \times 10^4$ ohm <sup>-1</sup> cm <sup>-1</sup> at 21 C.
2	673 E	1932	294.7	1.3		93.6	6.4	Specimen 200 mm long; electrical conductivity $10.2 \times 10^4$ ohm <sup>-1</sup> cm <sup>-1</sup> at 21.5 C.

DATA TABLE NO. 14<sup>s</sup> THERMAL CONDUCTIVITY OF [MAGNESIUM + TIN] ALLOYS

(Mg + Sn = 99.50%; impurity &lt; 0.20% each)

(Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup>K<sup>-1</sup>)

T	k
<u>CURVE 1</u>	
294.2	1.059
<u>CURVE 2</u>	
294.7	0.741

No graphical presentation

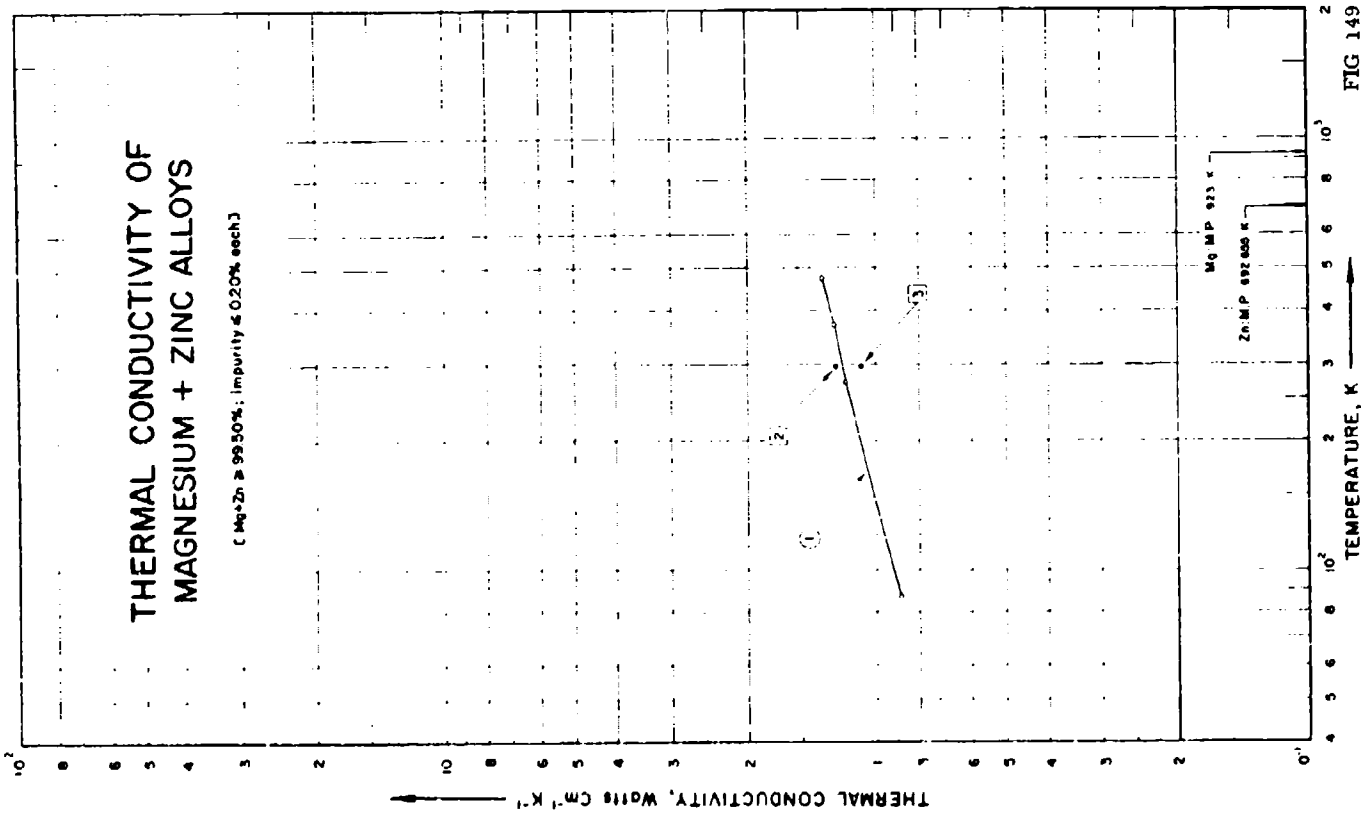


FIG. 149



## SPECIFICATION TABLE NO. 149 THERMAL CONDUCTIVITY OF (MAGNESIUM - ZINC) ALLOYS

(Mg - Zn 99.50%; impurity  $\pm 0.20\%$  each.)

[For Data Reported in Figure and Table No. 149.]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) Mg                      Zn	Composition (continued), Specifications and Remarks
1	530 93	L	1929	87-476	3-4		92.0                      8.0	Specimen 1.23 cm <sup>2</sup> in cross-section and 3 cm long, electrical conductivity 34.0, 15.37, 11.50 and 8.50 x 10 <sup>4</sup> ohm <sup>-1</sup> cm <sup>-1</sup> at 87, 273, 373 and 476 K, respectively.
2	673	E	1932	299.0	1.3		99.9                      2.1	Specimen 200 mm long; electrical conductivity 18.7 x 10 <sup>4</sup> ohm <sup>-1</sup> cm <sup>-1</sup> at 25.5 C.
3	673	E	1932	298.7	1.3		93.9                      6.1	Specimen 200 mm long; electrical conductivity 16.6 x 10 <sup>4</sup> ohm <sup>-1</sup> cm <sup>-1</sup> at 25.5 C.

## DATA TABLE NO. 149 THERMAL CONDUCTIVITY OF [MAGNESIUM - ZINC] ALLOYS

(Mg + Zn  $\geq$  99.50%; impurity  $\leq$  0.20% each)[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup> K<sup>-1</sup>]

T	k
<u>CURVE 1</u>	
87	0.887
273	1.19
373	1.27
476	1.35
<u>CURVE 2</u>	
299	1.26
<u>CURVE 3</u>	
299.7	1.09

SPECIFICATION TABLE NO. 150 THERMAL CONDUCTIVITY OF [MANGANESE + COPPER] ALLOYS

(Mn + Cu = 99.50%; impurity ≤ 0.20% each)

Curve No.	Ref. Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)		Composition (continued), Specifications and Remarks
						Mn	Cu	
1	230	L	1927	332.2		60.0	40.0	Approx composition; the alloy made from Cu (< 0.01 impurity) supplied by Baker and fused with Mn, supplied by Eimer and Amend; specimen ~ 5 cm long with 0.3 cm <sup>2</sup> in cross-section; electrical conductivity $0.52 \times 10^6 \text{ ohm}^{-1}\text{cm}^{-1}$ at 23 C.
2	230	L	1925	332.2		80.0	20.0	Similar to the above specimen except electrical conductivity $0.687 \times 10^6 \text{ ohm}^{-1}\text{cm}^{-1}$ at 23 C.

DATA TABLE NO. 150 THERMAL CONDUCTIVITY OF [MANGANESE + COPPER] ALLOYS

(Mn + Cu = 99.50%; impurity < 0.20% each)

[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup>K<sup>-1</sup>]

T	k
<u>CURVE 1</u>	
332.2	0.113
<u>CURVE 2</u>	
332.2	0.105

No graphical presentation

SPECIFICATION TABLE NO. 131 THERMAL CONDUCTIVITY OF MANGANESE + IRON ALLOYS  
(Mn + Fe - 99.50%; impurity - 0.20% each)

Curve No.	Ref. Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) Mn	Composition (weight percent) Fe	Composition (continue), Specifications and Remarks
1	204	L	1937	317.4	Ferromanganese, 23	96.79	14.21	No detail reported.

DATA TABLE NO. 131 THERMAL CONDUCTIVITY OF MANGANESE + IRON ALLOYS  
(Mn + Fe - 99.50%; impurity - 0.20% each)

Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup> K<sup>-1</sup>.

T k

CURVE J

317.4 0.715

No graphical presentation

SPECIFICATION TABLE NO. 152 THERMAL CONDUCTIVITY OF (MANGANESE + NICKEL) ALLOYS

(Mn + Ni = 99.50%; impurity &lt; 0.20% each)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) Mn	Composition (weight percent) Ni	Composition (continued), Specifications and Remarks
1	230	L	1925	333.2			50	50	Approx composition; specimen made from Ni (99.75 to 99.85 pure including cobalt) supplied by International Nickel Co. of America and fused with Mn, supplied by Elmer and Amend; specimen ~5 cm long with 0.3 cm <sup>2</sup> in cross-section; electrical conductivity $3.56 \times 10^3 \text{ ohm}^{-1}\text{cm}^{-1}$ at 25 C.
2	230	L	1925	333.2			60	40	Similar to the above specimen except electrical conductivity $4.59 \times 10^3 \text{ ohm}^{-1}\text{cm}^{-1}$ at 25 C.
3	230	L	1925	333.2			70	30	Similar to the above specimen except electrical conductivity $5.12 \times 10^3 \text{ ohm}^{-1}\text{cm}^{-1}$ at 25 C.
4	230	L	1925	333.2			90	10	Similar to the above specimen except electrical conductivity $5.59 \times 10^3 \text{ ohm}^{-1}\text{cm}^{-1}$ at 25 C.

DATA TABLE NO. 152 THERMAL CONDUCTIVITY OF (MANGANESE + NICKEL) ALLOYS

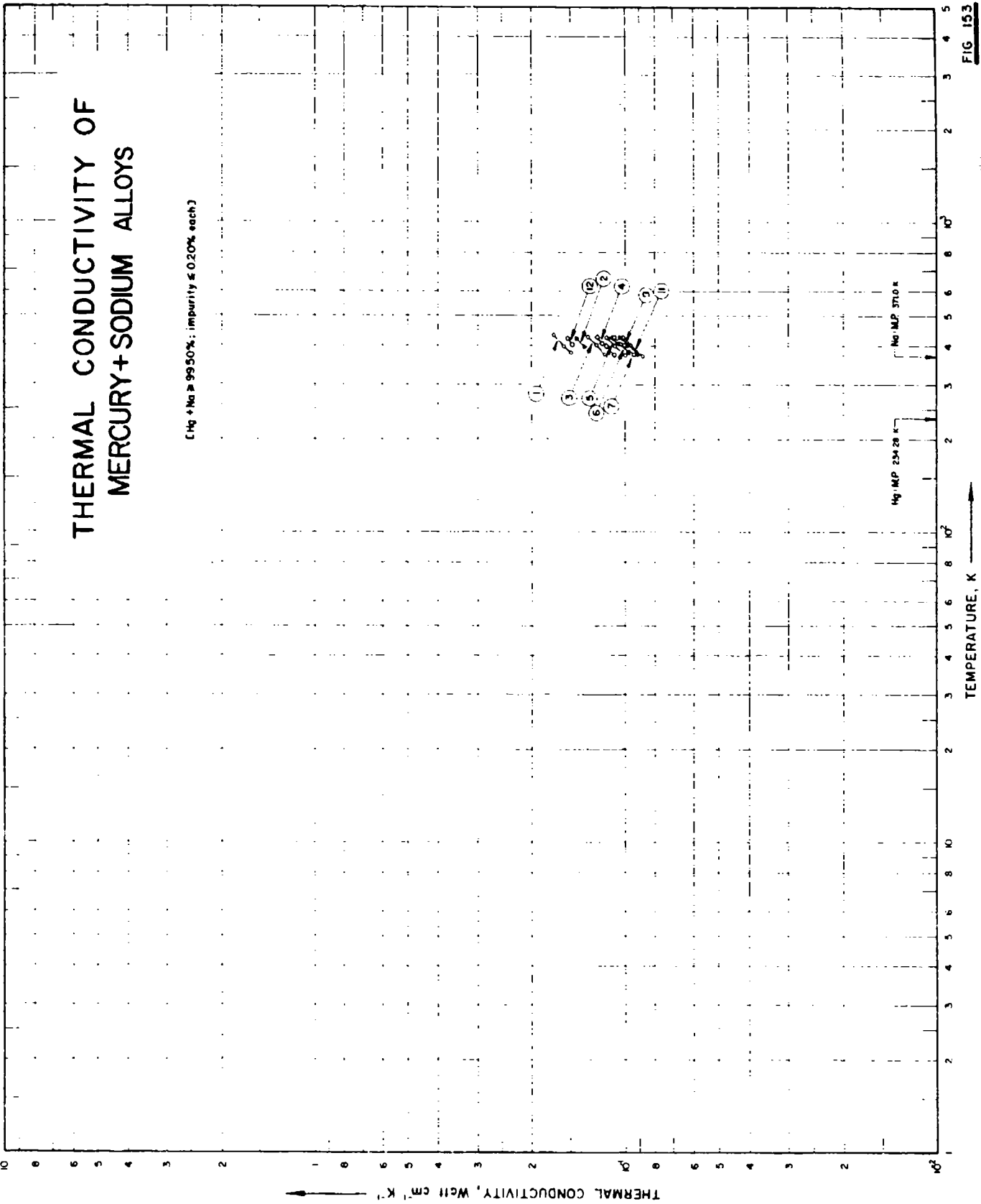
(Mn + Ni = 99.50%; impurity &lt; 0.20% each)

Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup>K<sup>-1</sup>

T	k	T	k
<u>CURVE 1'</u>	0.092	<u>CURVE 1'</u>	
333.2		333.2	0.092
<u>CURVE 2'</u>			
333.2	0.096		
<u>CURVE 3'</u>			
333.2	0.105		

\* No graphical presentation

FIGURE SHOWS ONLY 10 OF THE CURVES REPORTED IN TABLE



SPECIFICATION TABLE NO. 153 THERMAL CONDUCTIVITY OF MERCURY-SODIUM ALLOYS

Hg - Na 49.50% (mass) 0.30%

For Data Reported in Figure and Table No. 153

Curve No.	Rel. Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Hg, %	Na, %	Composition (continued), Specifications and Remarks
1	316, 65	L	1936 350-424	±3.0	18	55.57	44.43	Molten specimen contained in a cylinder, prepared by melting under paraffin the appropriate amount of certified pure Hg from Matheson Chemical Co. and Na from Chemists Dept. of Univ. of Kansas.
2	316, 65	L	1936 397-427	±3.0	16	58.95	41.05	Similar to the above specimen.
3	316, 65	L	1936 376-429	±3.0	15	62.69	37.30	Similar to the above specimen.
4	316, 65	L	1936 375-429	±3.0	14	65.54	34.46	Similar to the above specimen.
5	316, 65	L	1936 374-426	±3.0	12	68.56	31.44	Similar to the above specimen.
6	316, 65	L	1936 379-426	±3.0	19	71.57	28.43	Similar to the above specimen.
7	316, 65	L	1936 375-427	±3.0	9	73.26	26.74	Similar to the above specimen.
8	316, 65	L	1936 376-427	±3.0	8	74.89	25.10	Similar to the above specimen.
9	316, 65	L	1936 376-424	±3.0	6	76.15	23.85	Similar to the above specimen.
10	316, 65	L	1936 373-424	±3.0	5	77.55	22.45	Similar to the above specimen.
11	316, 65	L	1936 371-421	±3.0	4	78.89	21.11	Similar to the above specimen.
12	316, 65	L	1936 405-425	±3.0	17	56.8	43.2	Similar to the above specimen.
13	316, 65	L	1936 377-430	±3.0	13	68.0	32.0	Similar to the above specimen.
14	316, 65	L	1936 378-425	±3.0	11	69.7	30.3	Similar to the above specimen.
15	316, 65	L	1936 377-423	±3.0	7	75.6	24.4	Similar to the above specimen.

SPECIFICATION TABLE NO. 153 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) Hg	Ni	Composition (continued), Specifications and Remarks
16	316	L	1936	372-403	+3.0	3	98.71	1.29	Similar to the above specimen.
17	316	L	1936	380-410	+3.0	2	99.767	0.233	Similar to the above specimen.



DATA TABLE NO. 153 THERMAL CONDUCTIVITY OF [MERCURY + SODIUM] ALLOYS

(Hg + Na = 99.50%; Impurity &lt; 0.20% each)

(Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup>K<sup>-1</sup>)

T	k	T	k	T	k
<u>CURVE 1</u>					
330.4	0.149	375.8	0.0928	377.0	0.0929
398.8	0.157	400.0	0.0998	400.9	0.0990
433.7	0.170	427.3	0.107	422.6	0.1052
<u>CURVE 2</u>					
397.2	0.135	375.9	0.0911	372.3	0.1046
423.3	0.144	400.4	0.0877	401.5	0.1110
		424.4	0.104	402.5	0.1115
<u>CURVE 3</u>					
376.4	0.116	<u>CURVE 10*</u>			
401.7	0.124	373.4	0.0910	379.5	0.1063
429.0	0.132	599.0	0.0972	403.5	0.1111
		400.3	0.0972	409.7	0.1131
<u>CURVE 4</u>					
375.4	0.108	421.8	0.103	<u>CURVE 17*</u>	
399.0	0.115	423.7	0.104		
409.3	0.118	<u>CURVE 11</u>			
429.4	0.122	570.6	0.0879		
		400.8	0.0967		
<u>CURVE 5</u>					
374.4	0.0999	420.8	0.102		
399.5	0.107	<u>CURVE 12</u>			
426.4	0.115	405.0	0.1474		
		414.0	0.1537		
<u>CURVE 6</u>					
378.7	0.0975	<u>CURVE 13*</u>			
379.2	0.0978	376.9	0.1030		
405.0	0.106	399.1	0.1101		
426.4	0.111	430.1	0.1186		
<u>CURVE 7</u>					
374.8	0.0940	<u>CURVE 14*</u>			
400.4	0.101	377.7	0.0995		
426.9	0.109	400.5	0.1058		
		425.2	0.1126		

\* Not shown on plot

## SPECIFICATION TABLE NO. 154 THERMAL CONDUCTIVITY OF [MOLYBDENUM + IRON] ALLOYS

(Mo + Fe : 99.50%; impurity  $\leq$  0.20% each)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) Mo	Composition (weight percent) Fe	Composition (continued), Specifications and Remarks
1	204	I.	1937	330.4		Russian ferromolybdenum	62	Bal	0.1 C; specimen 20 x 20 mm cross sectional area.

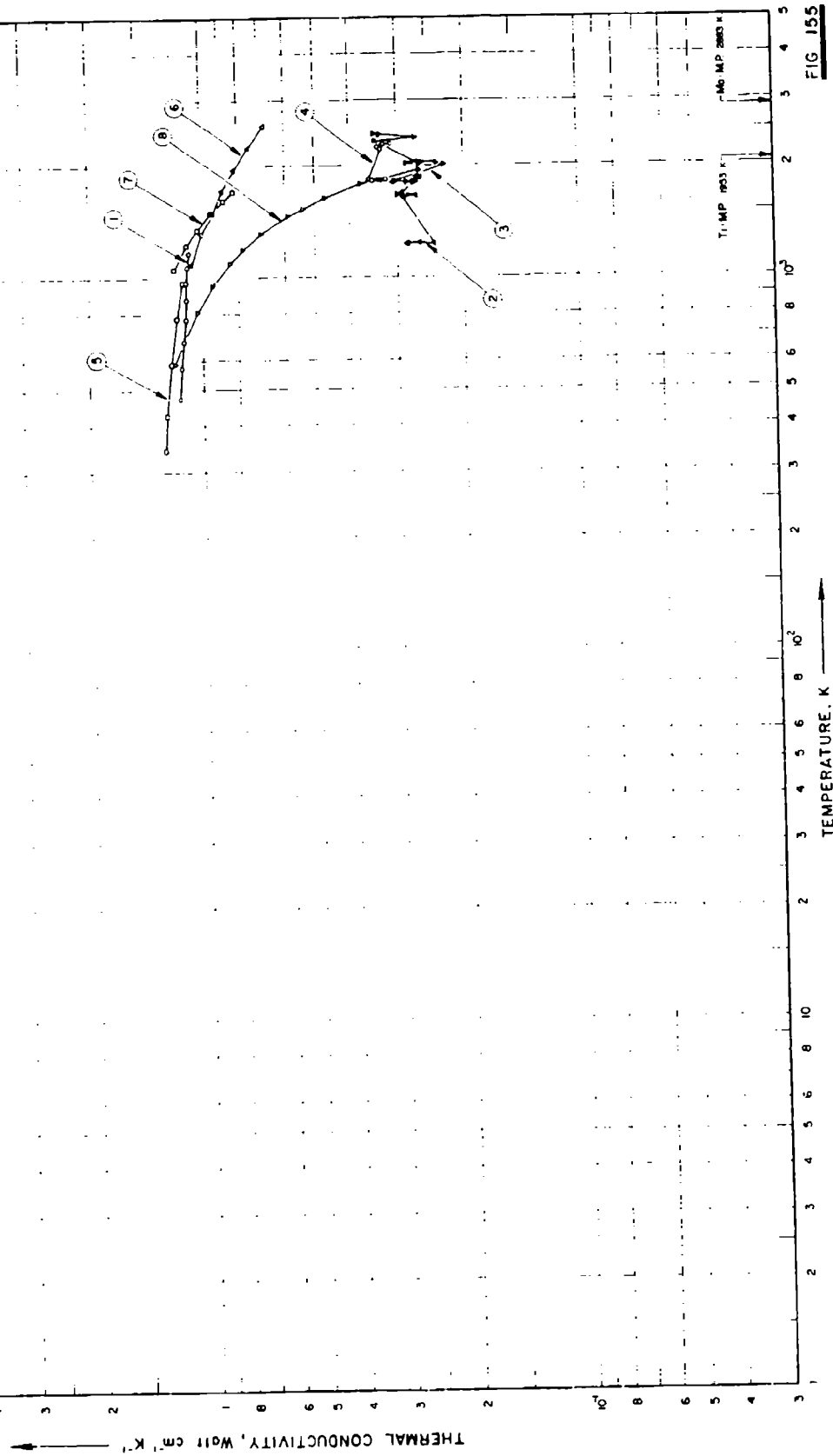
## DATA TABLE NO. 154 THERMAL CONDUCTIVITY OF [MOLYBDENUM + IRON] ALLOYS

(Mo + Fe : 99.50%; impurity  $\leq$  0.20% each)[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup>K<sup>-1</sup>]

T k  
 CURVE 1'  
 330.4 0.57k

# THERMAL CONDUCTIVITY OF MOLYBDENUM+TITANIUM ALLOYS

(Mo + Ti ≥ 99.50%; impurity ≤ 0.20% each)



## SPECIFICATION TABLE NO. 155 THERMAL CONDUCTIVITY OF [MOLYBDENUM + TITANIUM] ALLOYS

(Mo + Ti) 99.50%, impurity  $\pm 0.20\%$  each

[For Data Reported in Figure and Table No. 155]

Curve No.	Ref. Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition Mo	Composition (weight percent) Ti	Composition (continued), Specifications and Remarks
1	543 C	1955	473-1173	5	Heat No. 1132	99.5	0.5	Recrystallized at 2700 F; measured in a vacuum of $2 \times 10^{-6}$ mm Hg.
2	544 R	1963	1233-1863	$\sim 6$	1	99.5	0.5	Specimen 3/4 in. long, 3/4 in. O. D., 1/4 in. I. D.; surface scratches eliminated.
3	544 R	1963	1660-2432	$\sim 6$	1	99.5	0.5	The above specimen re-measured; partially melted after measurement.
4	544 R	1963	1400-2247	$\sim 6$		99.5	0.5	Specimen 3 in. long, 2.5 in. O. D., 3/4 in. I. D.; surface scratches eliminated; discolored during measurement.
5	583 C	1963	344-975	$\pm 4$		99.51	0.49	Impurities: 0.026 C, < 0.001 Fe, < 0.0001 Ni, < 0.001 Si, 0.07 Zr, 0.0007 O <sub>2</sub> , 0.0001 H <sub>2</sub> , and 0.0001 N <sub>2</sub> ; specimen 2 in. dia., 1 in. thick; supplied by Climax Molybdenum; density 622 lb ft <sup>-3</sup> ; measured in a He atmosphere; Armco iron used as standard.
6	583 P	1963	1100-2578	$\pm 4$		99.51	0.49	The above specimen measured using different method.
7	845 P	1965	1057-1707		Specimen 2	Bal.	0.28	0.18 Zr, commercial Mo; cylindrical specimen 15 mm in dia and 70 mm long; density $10.17 \text{ g cm}^{-3}$ ; electrical resistivity $6.52 \mu\text{ohm cm}$ at 23 C; thermal conductivity data obtained from the smooth curve calculated from measurements of thermal diffusivity, specific heat and density.
8	924 R	1961	589-1947			99.5	0.5	Specimen composed of 15 rings, 3 of which were 1 in. thick and 12 were 0.5 in. thick; measured in helium atmosphere; data extracted from the smooth curve.

DATA TABLE NO. 155 THERMAL CONDUCTIVITY OF [MOLYBDENUM + TITANIUM] ALLOYS

(Mo + Ti = 99.5%; impurity  $\leq 0.20\%$  each)[Temperature, T, K; Thermal Conductivity, k, Watt  $\text{cm}^{-1}\text{K}^{-1}$ ]

T	k	T	k	T	k
<u>CURVE 1</u>					
473.2	1.15	1799.8	0.307	588.7	1.19
573.2	1.14	1825.9	0.349	810.4	1.03
673.2	1.12	1858.7	0.322	947.6	0.938
773.2	1.10	1877.1	0.356	1098	0.838
873.2	1.10	2205.4	0.332	1189	0.774
973.2	1.10	2244.3	0.338	1316	0.689
1073.2	1.09	2277.6	0.327	1465	0.587
1173.2	1.08	2286.5	0.312	1539	0.537
				1647	0.466
				1780	0.372
				1947	0.269
<u>CURVE 2</u>					
1233.2	0.281	344.3	1.27		
1233.2	0.284	427.6	1.25		
1241.5	0.261	588.7	1.21		
1241.5	0.238	780.4	1.17		
1645.9	0.288	974.8	1.11		
1652.1	0.267				
1668.7	0.290				
1859.3	0.266				
1860.4	0.262				
1863.2	0.258				
<u>CURVE 3</u>					
1679.8	0.297	1099.8	1.07		
1659.8	0.283	1319.3	1.00		
1659.8	0.295	1494.3	0.935		
1798.2	0.270	1702.6	0.883		
1799.8	0.266	1958.2	0.815		
1800.9	0.282	2238.7	0.746		
1814.8	0.304	2577.6	0.678		
2008.2	0.224				
2020.4	0.261				
2025.9	0.283				
2030.4	0.258				
2032.6	0.234				
2035.9	0.263				
2328.7	0.341				
2379.8	0.266				
2405.9	0.233				
2431.5	0.245				
<u>CURVE 4</u>					
		1799.8	0.307		
		1825.9	0.349		
		1858.7	0.322		
		1877.1	0.356		
		2205.4	0.332		
		2244.3	0.338		
		2277.6	0.327		
		2286.5	0.312		
<u>CURVE 5</u>					
		344.3	1.27		
		427.6	1.25		
		588.7	1.21		
		780.4	1.17		
		974.8	1.11		
<u>CURVE 6</u>					
		1099.8	1.07		
		1319.3	1.00		
		1494.3	0.935		
		1702.6	0.883		
		1958.2	0.815		
		2238.7	0.746		
		2577.6	0.678		
<u>CURVE 7</u>					
		1057	1.188		
		1220	1.095		
		1349	1.018		
		1500	0.941		
		1609	0.878		
		1707	0.821		

\* Not shown on plot

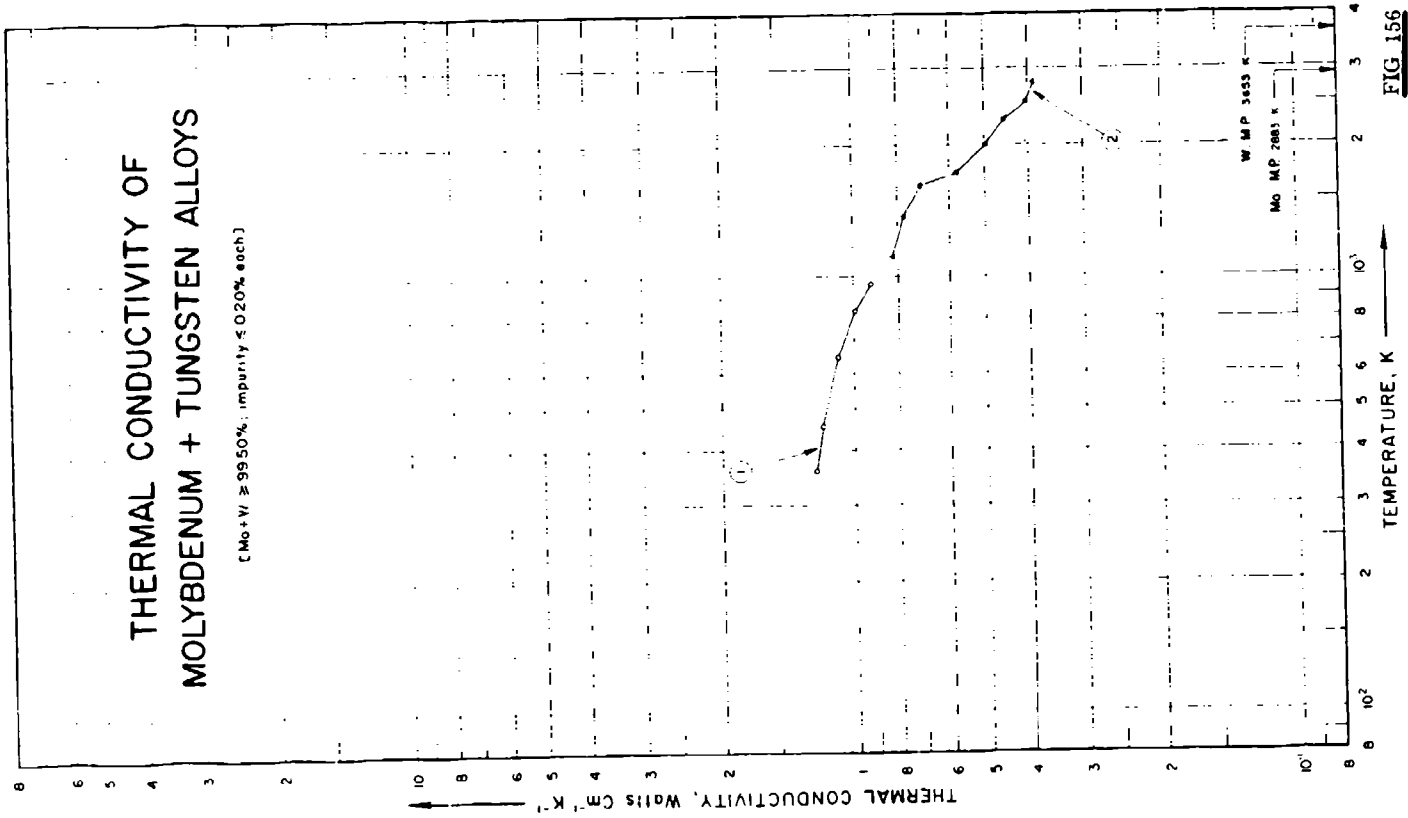


FIG 156

## SPECIFICATION TABLE NO. 156 THERMAL CONDUCTIVITY OF (MOLYBDENUM - TUNGSTEN) ALLOYS

(Mo - W = 99.50%; impurity  $\pm 0.20\%$  each)

[For Data Reported in Figure and Table No. 156]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (Weight percent) Mo	W	Composition (continued), Specifications and Remarks
1	583	C	1963	558-964	$\pm 4$		70.17	29.83	Impurities: 0.07 Zr and 0.012 C; specimen 2 in. dia., 1 in. thick, supplied by Climax Molybdenum, density 9.33 g/cm <sup>3</sup> , measured in a He atmosphere; Armco iron used as standard
2	583	P	1963	1110-2772	$\pm 4$		70.17	29.83	The above specimen measured using different method.

DATA TABLE NO. 156 THERMAL CONDUCTIVITY OF MOLYBDENUM - TUNGSTEN ALLOYS

(Mo + W - 99.50% impurity 0.20% each)

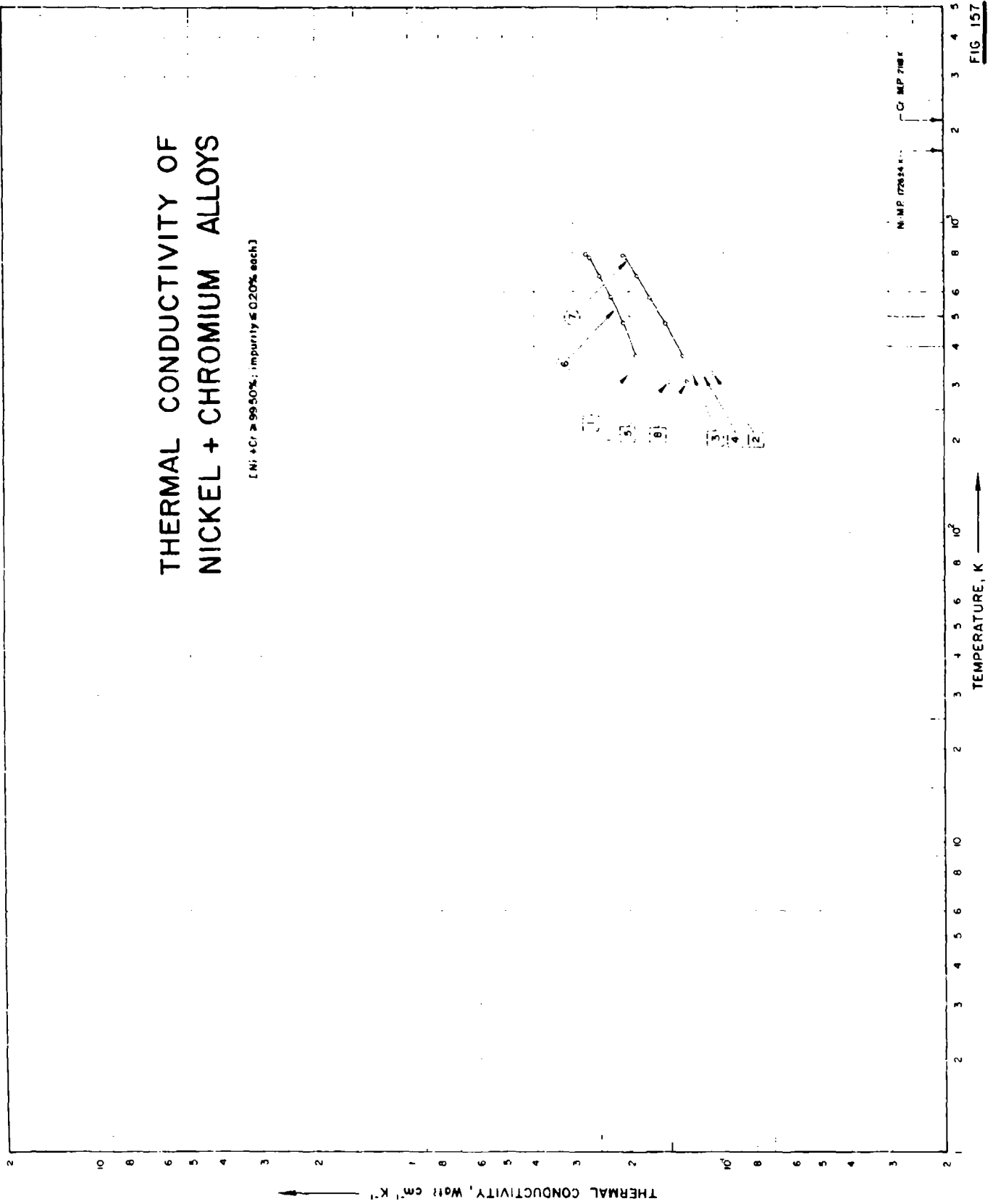
[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup>K<sup>-1</sup>]

T	k
<u>CURVE 1</u>	
358.1	1.23
452.6	1.19
652.6	1.09
838.7	0.993
963.7	0.919
<u>CURVE 2</u>	
1110.9	0.820
1376.5	0.775
1613.7	0.704
1732.1	0.575
2016.5	0.497
2291.5	0.452
2519.3	0.405
2772.1	0.389



# THERMAL CONDUCTIVITY OF NICKEL + CHROMIUM ALLOYS

[Ni + Cr ≥ 99.50%; impurity ≤ 0.20% each]



## SPECIFICATION TABLE NO. 157 THERMAL CONDUCTIVITY OF [NICKEL-CHROMIUM] ALLOYS

(Ni + Cr ± 99.50%; impurity ± 0.20% each)

[For Data Reported in Figure and Table No. 157.]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) Ni Cr	Composition (continued), Specifications and Remarks
1	230	L	1925	329			90 10	Prepared by fusing Ni (99.75 L) 99.85 pure including Co. supplied by International Nickel Co.) and Cr (supplied by Emet and Amend.); specimen ~5.5 cm long, 0.3 cm <sup>2</sup> cross-sectional area; electrical conductivity 1403 ohm <sup>-1</sup> cm <sup>-1</sup> at 25 C.
2	230	L	1925	329			70 30	Similar to the above specimen except electrical conductivity 945 ohm <sup>-1</sup> cm <sup>-1</sup> at 25 C.
3	230	L	1925	329			60 40	Similar to the above specimen except electrical conductivity 813 ohm <sup>-1</sup> cm <sup>-1</sup> at 25 C.
4	230	L	1925	329			50 50	Similar to the above specimen except electrical conductivity 830 ohm <sup>-1</sup> cm <sup>-1</sup> at 25 C.
5	186	P	1928	305		Nachrome N	80 20	Specimen 0.25 cm dia., 8.40 g cm <sup>-3</sup> density.
6	129	C	1933	373-795	3-5	Chromel P	90 10	Specimen 2 cm dia., 15 cm long; lead used as standard (0.352 watt cm <sup>-1</sup> deg <sup>-1</sup> at 6 C. assumed value).
7	129	C	1933	373-788	3-5	Chromel A	80 20	Similar to the above specimen.
8	673	E	1952	309.1			80 20	Specimen 200 mm long; electrical conductivity 9520 ohm <sup>-1</sup> cm <sup>-1</sup> at 35.9 C.

DATA TABLE NO. 157 THERMAL CONDUCTIVITY OF NICKEL - CHROMIUM ALLOYS

(Ni + Cr: 99.50%, purity 0.20% each)

(Temperature: T, K; Thermal Conductivity,  $k$ , Watt  $\text{cm}^{-1} \text{K}^{-1}$ )

T	k
<u>CURVE 1</u>	
329.2	0.197
<u>CURVE 2</u>	
329.2	0.109
<u>CURVE 3</u>	
329.2	0.126
<u>CURVE 4</u>	
329.2	0.117
<u>CURVE 5</u>	
305	0.150
<u>CURVE 6</u>	
373.2	0.190
473.2	0.209
573.2	0.228
673.2	0.247
773.2	0.266
794.7	0.271
<u>CURVE 7</u>	
373.2	0.136
473.2	0.154
573.2	0.172
673.2	0.189
773.2	0.206
788.4	0.209
<u>CURVE 8</u>	
309.1	0.131

# THERMAL CONDUCTIVITY OF NICKEL + COBALT ALLOYS

[Ni + Co ≥ 99.50%; impurity ≤ 0.20% each.]

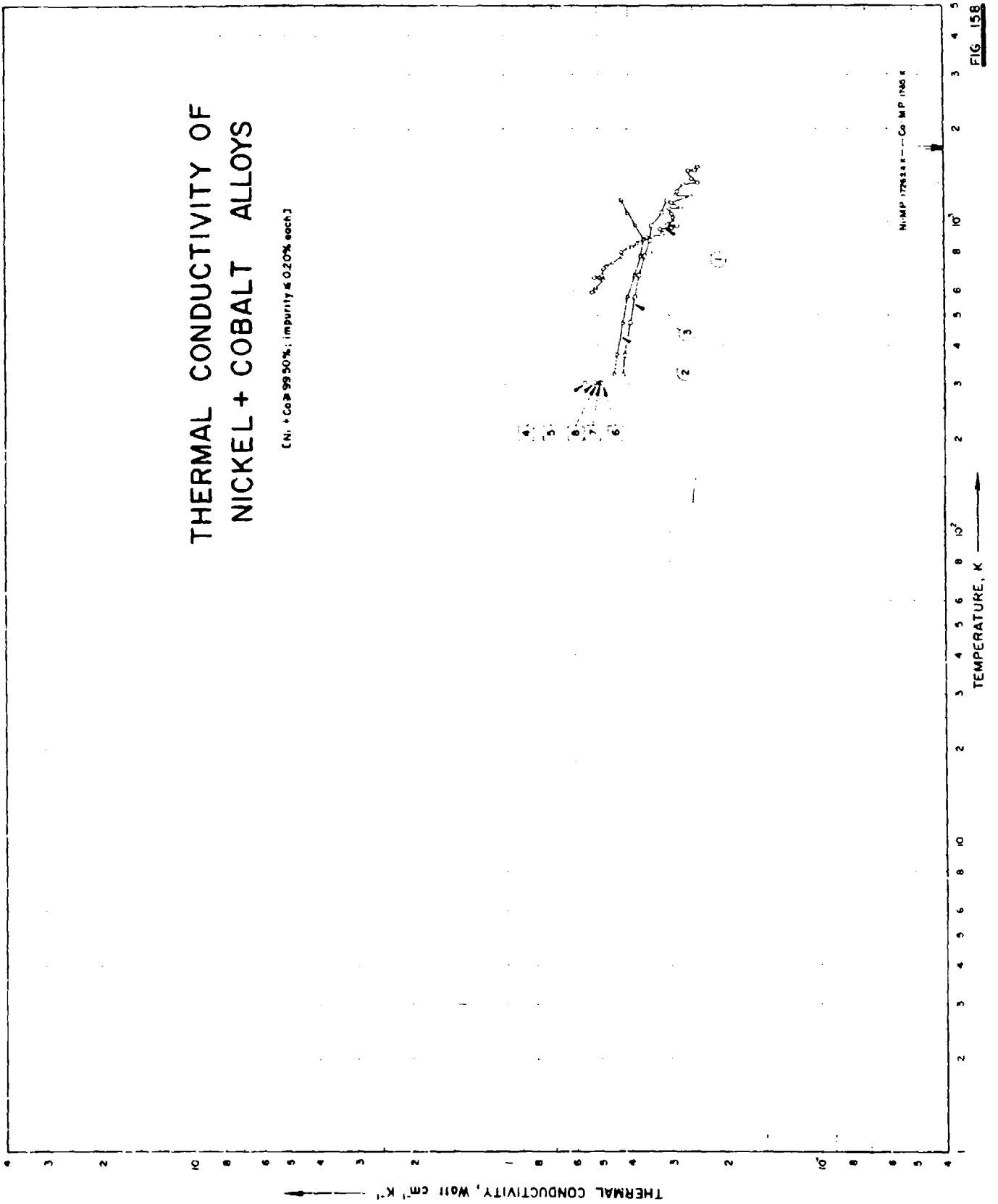


FIG. 158

## SPECIFICATION TABLE NO. 154 THERMAL CONDUCTIVITY OF (NICKEL-COBALT) ALLOYS

(Ni + Co) 99.50%; Impurity &lt; 0.20% each

[For Data Reported in Figure and Table No. 158.]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) Ni	Composition (weight percent) Co	Composition (continued), Specifications and Remarks
1	4	E	1911	593-1508			~97.5	~2.5	Approximate composition: hollow cylindrical rod 6 in. long, 1.2 cm O. D. and 0.168 cm I. D.
2	131	C	1953	323-1173	2	RCA N91	78.1	21.6	0.185 Mn, 0.115 C, and 0.01 Mg; annealed at 1000 C.; lead used as primary standard; advance (55 Cu, 45 Ni) used as working standard
3	131	C	1953	323-1173	2	RCA N97	59.5	40.0	0.175 Mn, 0.132 C, 0.19 Si, and 0.01 Mg; annealed at 1000 C.
4	218	E	1927	303			~90	~10	Impurities: 0.11 Fe, 0.06 C, trace P, 0.02 S, 0.009 Si, 0.005 Al, trace Mn, and 0.01 Cu; cast machined; annealed 40 min at 800 C.; slowly cooled.
5	238	E	1927	303			~80	~20	Impurities: 0.12 Fe, 0.07 C, trace P, 0.02 S, 0.01 Si, 0.01 Al, trace Mn, and 0.01 Cu; cast, machined; annealed 40 min at 800 C.; slowly cooled.
6	238	E	1927	303			~70	~30	Impurities: 0.13 Fe, 0.09 C, trace P, 0.002 S, 0.01 Si, 0.02 Al, trace Mn, and 0.009 Cu; cast, machined; annealed 40 min at 800 C.; cooled slowly.
7	238	E	1927	303			~60	~40	Impurities: 0.14 Fe, 0.11 C, 0.001 P, 0.04 Si, 0.02 Al, trace Mn, and 0.008 Cu; cast, machined; annealed 40 min at 800 C.; cooled slowly.
8	238	E	1927	303			~56	~59	Impurities: 0.15 Fe, 0.13 C, 0.002 P, 0.03 S, 0.02 Si, 0.03 Al, trace Mn, and 0.007 Cu; cast, machined; annealed for 40 min at 800 C. cooled slowly.

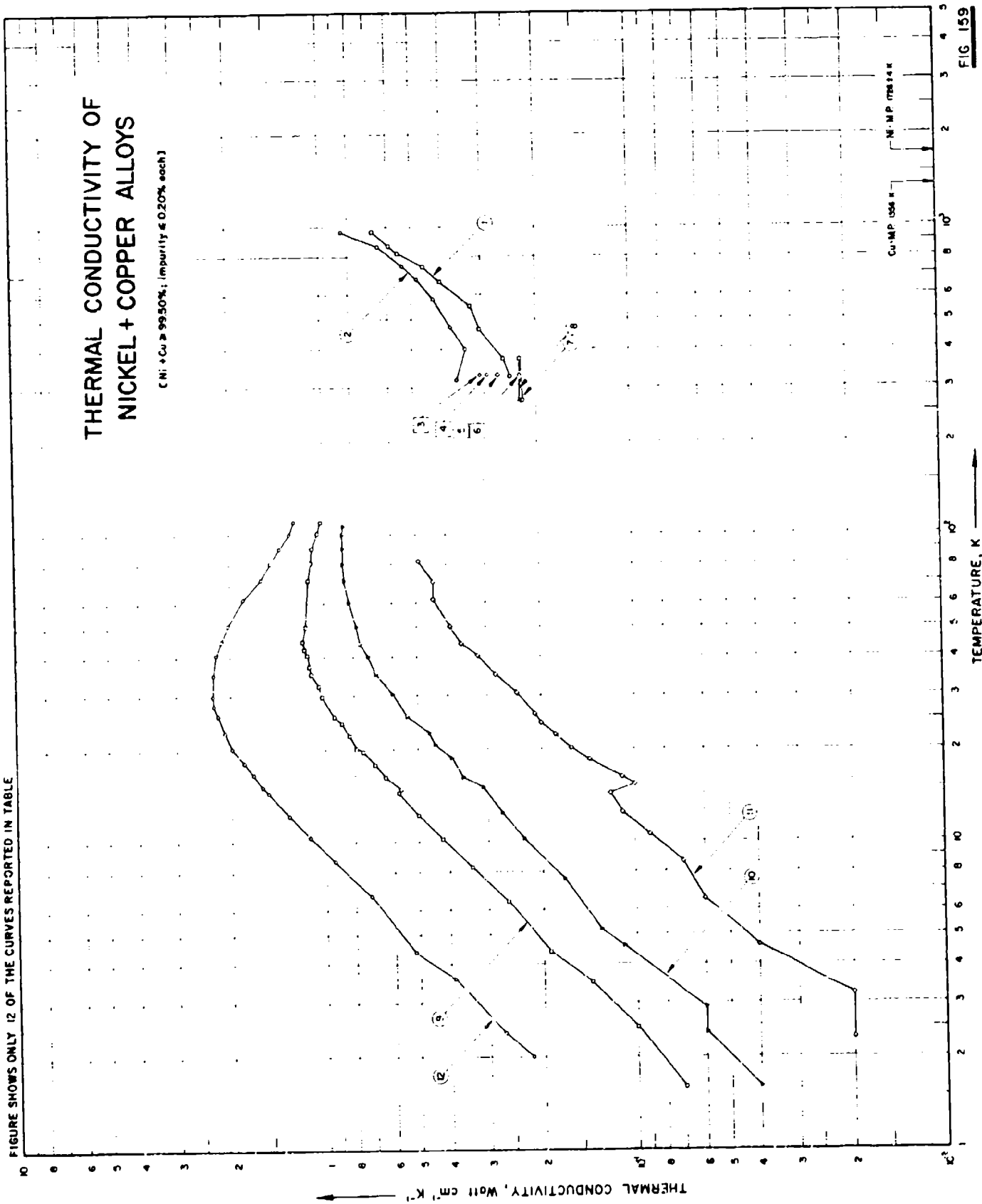
DATA TABLE NO. 15. THERMAL CONDUCTIVITY OF NICKEL-COBALT ALLOYS

(Ni + Co = 99.50%, impurity = 0.20% each)  
(Temperature, T, K; Thermal Conductivity, k, Watt/cm<sup>2</sup>K<sup>-1</sup>)

T	k	T	k	T	k
<u>CURVE 1</u>					
593.2	0.523	1223.2	0.259		
593.2	0.510	1263.2	0.280		
598.2	0.510	1323.2	0.268		
608.2	0.519	1343.2	0.243		
613.2	0.510	1378.2	0.253		
653.2	0.481	1473.2	0.259		
663.2	0.515	1473.2	0.247		
673.2	0.485	1483.2	0.247		
708.2	0.481	1508.2	0.243		
718.2	0.473				
743.2	0.448				
773.2	0.418				
798.2	0.423	323.20	0.446		
808.2	0.423	353.2	0.437		
828.2	0.397	473.2	0.419		
828.2	0.389	573.2	0.401		
838.2	0.393	673.2	0.383		
838.2	0.359	773.2	0.366		
853.2	0.372	873.2	0.359		
863.2	0.377	973.2	0.380		
883.2	0.339	1073.2	0.401		
893.2	0.351	1173.2	0.421		
898.2	0.360				
898.2	0.351				
903.2	0.314				
923.2	0.310				
948.2	0.319	323.2	0.418		
958.2	0.293	373.2	0.411		
972.2	0.280	473.2	0.396		
978.2	0.276	573.2	0.384		
983.2	0.301	673.2	0.370		
1023.2	0.285	773.2	0.356		
1033.2	0.289	873.2	0.343		
1043.2	0.289	973.2	0.339		
1068.2	0.293	1073.2	0.315		
1103.2	0.297	1173.2	0.301		
1113.2	0.272				
1123.2	0.272				
1163.2	0.293				
1213.2	0.255				
1223.2	0.285				
<u>CURVE 2</u>					
<u>CURVE 3</u>					
<u>CURVE 4</u>					
		303.2	0.515		
				303.2	0.498
				303.2	0.506
				303.2	0.552

Not shown on plot

FIGURE SHOWS ONLY 12 OF THE CURVES REPORTED IN TABLE



## SPECIFICATION TABLE NO. 159 THERMAL CONDUCTIVITY OF (NICKEL-COPPER) ALLOYS

(Ni + Cu = 99.50%, impurity  $\leq 0.20\%$  each)

[For Data Reported in Figure and Table No. 159]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) Ni Cu	Composition (continued), Specifications and Remarks
1	124	P	1930	325-970			~60 ~40	0.2 Mn, trace Mg; annealed at 700 C, density 8.61 g cm <sup>-3</sup> , electrical conductivity ranging from 1.88 to 1.74 x 10 <sup>4</sup> ohm <sup>-1</sup> cm <sup>-1</sup> at 52 to 697 C.
2	124	P	1930	317-966			~80 ~20	0.2 Mn, trace Mg; annealed at 700 C, density 8.62 g cm <sup>-3</sup> , electrical conductivity ranging from 2.36 to 1.96 x 10 <sup>4</sup> ohm <sup>-1</sup> cm <sup>-1</sup> at 44 to 693 C.
3	230	L	1925	330			80 20	Prepared by fusing Ni (99.75 to 99.85 pure, supplied by International Nickel Co.) and Cu (<0.03 impurities, supplied by Baker); specimen ~5.5 cm long, 0.3 cm <sup>2</sup> cross sectional area; electrical conductivity 3.00 x 10 <sup>4</sup> ohm <sup>-1</sup> cm <sup>-1</sup> at 25 C.
4	230	L	1925	330			70 30	Similar to the above specimen except electrical conductivity 2.17 x 10 <sup>4</sup> ohm <sup>-1</sup> cm <sup>-1</sup> at 25 C.
5	230	L	1925	330			60 40	Similar to the above specimen except electrical conductivity 2.02 x 10 <sup>4</sup> ohm <sup>-1</sup> cm <sup>-1</sup> at 25 C.
6	230	L	1925	330			50 50	Similar to the above specimen except electrical conductivity 1.98 x 10 <sup>4</sup> ohm <sup>-1</sup> cm <sup>-1</sup> at 25 C.
7	246	T	1919	273, 373			60, 90 39, 07	Rolled and drawn; annealed close to melting point for 1/2 hr. Similar to the above specimen.
8	246	T	1919	273, 373			81, 63 18, 37	0.649 ± 0.104 Cu from x-ray microanalysis (0.965 Cu from residue resistivity measurement); specimen 4 mm in dia; supplied by Johnson Matthey and Co.; chill cast from J. M. 890 Ni and J. M. 30 Cu; annealed for 12 hrs at 850 C; electrical resistivity 1.0 x 10 <sup>-3</sup> , 4.99 x 10 <sup>-3</sup> , 1.422 x 10 <sup>-2</sup> , 0.16 x 10 <sup>-2</sup> , 4.27 x 10 <sup>-2</sup> , 7.48 x 10 <sup>-2</sup> and 8.79 x 10 <sup>-2</sup> ohm cm at 4, 1, 10, 1, 16, 0, 21, 8, 25, 2, 31, 8 and 34, 1 K respectively.
9	917	E	1965	1.6-111	0.5-5	C		
10	917	E	1965	1.6-107	0.5-5	D		1.65 ± 0.22 Cu from x-ray microanalysis (1.73 Cu from residue resistivity measurement); specimen 4 mm in dia; supplied by Johnson Matthey and Co.; chill cast from J. M. 890 Ni and J. M. 30 Cu; annealed for 12 hrs at 850 C; electrical resistivity 1.01 x 10 <sup>-3</sup> , 5.12 x 10 <sup>-3</sup> , 2.12 x 10 <sup>-2</sup> , 3.57 x 10 <sup>-2</sup> , 5.63 x 10 <sup>-2</sup> , and 8.185 x 10 <sup>-2</sup> ohm cm at 3, 4, 7, 4, 15, 8, 20, 1, 24, 8 and 29, 6 K, respectively.



SPECIFICATION TABLE NO. 159 (continued)

Curve No.	Rel. Method No.	Year	Temp. Range, K	Reported Error	Name and Specimen Designation	Composition (weight percent)		Composition (continued), Specifications and Remarks
						Ni	Cu	
11	917	E	1965	2.5-22.1	0.3-5	E		4.52 ± 0.65 Cu from X-ray microanalysis (5.28 Cu from residue resistivity measurement); specimen 3 mm in dia; supplied by Johnson Matthey and Co.; chill cast from J.M. 890 Ni and J.M. 30 Cu; annealed for 12 hrs at 550°C; electrical resistivity $9.02 \times 10^{-7}$ , $2.01 \times 10^{-7}$ , $1.67 \times 10^{-7}$ and $8.71 \times 10^{-7}$ ohm cm at 8.5, 12.2, 21.7 and 29.3 K, respectively.
12	917	E	1965	2.0-111	0.5-5	F		0.116 ± 0.051 Cu from X-ray microanalysis (0.172 Cu from residue resistivity measurement); specimen 1 mm in dia; supplied by Johnson Matthey and Co.; chill cast from J.M. 890 Ni and J.M. 30 Cu; annealed for 12 hrs at 550°C; electrical resistivity $1.1 \times 10^{-7}$ , $4.64 \times 10^{-7}$ , $1.05 \times 10^{-7}$ , $2.4 \times 10^{-7}$ , $4.63 \times 10^{-7}$ and $7.15 \times 10^{-7}$ ohm cm at 0.3, 1.2, 1.6, 21.9, 28.9 and 44.6 K, respectively.

DATA TABLE NO. 159 THERMAL CONDUCTIVITY OF NICKEL-COPPER ALLOYS

(Ni - Cu 99.50% purity - 0.20% each)

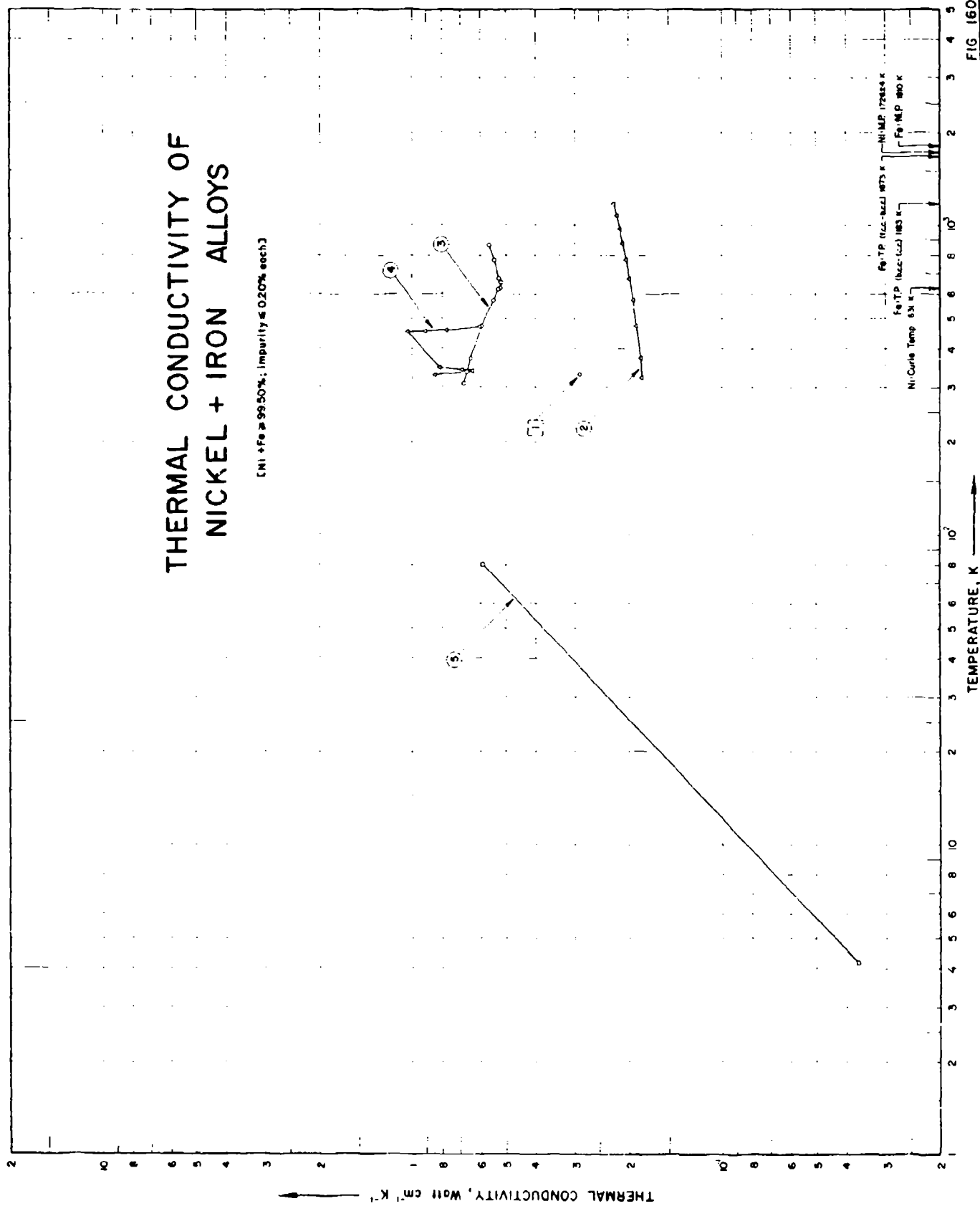
(Temperature, T, K; Thermal Conductivity,  $k$ , Watt/cm<sup>2</sup>K<sup>2</sup>)

T	k	T	k	T	k	T	k	T	k
<u>CURVE 1</u>									
325.2	0.243	273.2	0.26	7.6	0.17	50.2	0.39		
374.2	0.255	373.2	0.26	10.2	0.23	61.7	0.44		
407.2	0.305	<u>CURVE 9</u>							
592.2	0.326	1.6	0.07	15.1	0.31	70.1	0.44		
607.2	0.406	16.2	0.36	17.2	0.39	82.1	0.49		
747.2	0.456	18.9	0.39	20.7	0.44	<u>CURVE 12</u>			
827.2	0.552	22.8	0.46	22.8	0.46	2.0	0.22		
868.2	0.590	25.7	0.54	25.7	0.54	2.4	0.27		
970.2	0.669	30.5	0.60	30.5	0.60	3.6	0.39		
<u>CURVE 2</u>									
317.2	0.360	6.4	0.26	35.3	0.68	4.4	0.52		
401.2	0.339	8.3	0.34	40.2	0.72	6.7	0.72		
471.2	0.377	10.3	0.42	45.4	0.76	8.7	0.94		
584.2	0.427	12.3	0.50	50.4	0.78	10.4	1.14		
679.2	0.481	14.5	0.58	60.3	0.82	12.3	1.31		
747.2	0.536	14.9	0.57	71.4	0.85	14.6	1.52		
807.2	0.643	16.3	0.64	80.4	0.86	15.2	1.60		
900.2	0.849	17.9	0.69	90.2	0.86	16.7	1.71		
<u>CURVE 3</u>									
330.0	0.305	20.3	0.79	100.4	0.86	18.2	1.83		
<u>CURVE 4</u>									
330.0	0.289	22.2	0.83	101.7	0.85	20.3	1.99		
<u>CURVE 5</u>									
330.0	0.266	24.4	0.88	107.3	0.85	22.9	2.10		
<u>CURVE 6</u>									
330.0	0.225	25.6	0.93	<u>CURVE 11</u>					
<u>CURVE 7</u>									
273.2	0.22	29.3	1.02	2.3	0.02	26.0	2.21		
373.2	0.26	32.2	1.05	3.2	0.02	27.9	2.27		
<u>CURVE 8</u>									
373.2	0.26	35.1	1.11	4.6	0.04	30.1	2.29		
<u>CURVE 10</u>									
330.0	0.225	37.4	1.12	6.5	0.06	35.2	2.28		
<u>CURVE 11 (cont.)</u>									
330.0	0.225	40.6	1.14	8.6	0.07	40.9	2.23		
<u>CURVE 12 (cont.)</u>									
330.0	0.225	42.5	1.16	10.6	0.09	43.5	2.14		
330.0	0.225	45.3	1.18	12.5	0.11	51.0	2.03		
330.0	0.225	50.2	1.15	14.4	0.12	62.1	1.81		
330.0	0.225	51.1	1.10	15.4	0.10	71.8	1.59		
330.0	0.225	60.4	1.09	16.3	0.11	81.3	1.48		
330.0	0.225	101.0	1.05	18.5	0.14	96.3	1.37		
330.0	0.225	110.6	1.02	20.1	0.16	100.8	1.28		
<u>CURVE 10 (cont.)</u>									
330.0	0.225	22.4	0.18	22.4	0.18	111.4	1.24		
330.0	0.225	24.4	0.20	24.4	0.20				
330.0	0.225	26.1	0.21	26.1	0.21				
330.0	0.225	30.6	0.24	30.6	0.24				
330.0	0.225	35.0	0.28	35.0	0.28				
330.0	0.225	40.4	0.32	40.4	0.32				
330.0	0.225	45.2	0.36	45.2	0.36				

\* Not shown on plot

# THERMAL CONDUCTIVITY OF NICKEL + IRON ALLOYS

(Ni + Fe ≥ 99.50%; Impurity ≤ 0.20% each)



## SPECIFICATION TABLE NO. 100 THERMAL CONDUCTIVITY OF [NICKEL-IRON] ALLOYS

(Ni + Fe = 99.50%; impurity &lt; 0.20% each)

[For Data Reported in Figure and Table No. 100]

Curve No.	Ref. No. Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition Ni	Composition (weight percent) Fe	Composition (continued), Specifications and Remarks
1	206	1920	330-2			75.1	24.9	Prepared from Fe (99.97 pure), Ni (high purity), 0.1 C. specimen ~5.8 cm long, 0.96 cm dia.
2	131	C	323-1173	2		50.85	48.50	0.12 Mn, 0.24 C, and 0.003 S, annealed at 950 C; lead used as primary standard; Advance (55 Cu, 45 Ni) used as working standard.
3	129	C	309-865	5	N.S. nickel, commercial	99*	0.6	0.14 Cu, 0.05 Mn, and 0.014 S, specimen 2 cm dia, 15 cm long (0.352 watt cm <sup>-1</sup> deg <sup>-1</sup> at 0 C, assumed value)
4	305	L	328-472		commercial nickel	99.4	0.2	0.1 Mg, 0.05 Co, 0.01 Sn, 0.02 Si, 0.02% C, 0.01 Cf, 0.01 Mn, 0.003 Ti, 0.002 each of Al and Pb, and 0.005 S; experimental method inaccurate.
5	716	L	4.2-81			85.2	14.8	Specimen 0.2 cm dia, 5.2 cm long; fused in an induction furnace under vacuum of 10 <sup>-3</sup> torr, from Ni and Fe supplied by Johnson-Matthey; cold rolled, annealed at 1173 K for 2 hrs, slowly cooled.

DATA TABLE NO. 160 THERMAL CONDUCTIVITY OF [NICKEL + IRON] ALLOYS

(Ni + Fe ± 99.30%; impurity ± 0.20% each)

(Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup> K<sup>-1</sup>)

T	k	T	k
<u>CURVE 1</u>			
330.2	0.289	4.18	0.0307
<u>CURVE 2</u>			
323.2	0.181	80.5	0.589
373.2	0.183		
473.2	0.189		
573.2	0.194		
673.2	0.199		
773.2	0.204		
873.2	0.209		
973.2	0.214		
1073.2	0.219		
1173.2	0.224		
<u>CURVE 3</u>			
309.2	0.682		
336.6	0.668		
373.2	0.649		
473.2	0.599		
573.2	0.549		
622.5	0.527		
627.2	0.522		
653.8	0.519		
673.2	0.524		
773.2	0.546		
864.6	0.566		
<u>CURVE 4</u>			
325.2	0.845		
340.2	0.840		
342.2	0.830		
347.2	0.815		
452.2	1.038		
457.2	0.908		
457.2	0.774		
472.2	0.598		

FIGURE SHOWS ONLY 7 OF THE CURVES REPORTED IN TABLE

# THERMAL CONDUCTIVITY OF NICKEL + MANGANESE ALLOYS

(Ni + Mn ≥ 99.50%; impurity ≤ 0.20% each)

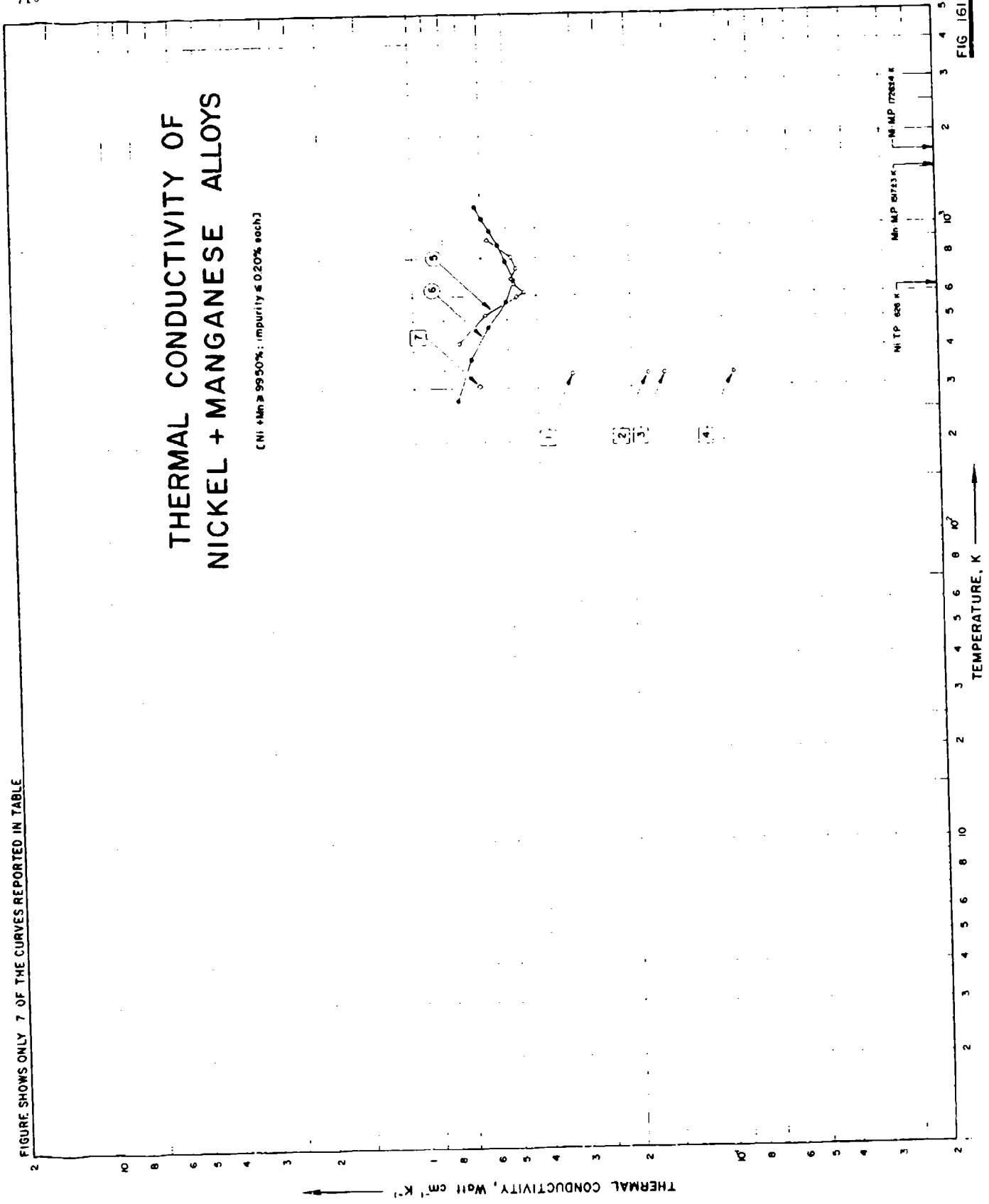


FIG 161

SPECIFICATION TABLE NO. 161 THERMAL CONDUCTIVITY OF NICKEL-MANGANESE ALLOYS  
(Ni-Mn 99.50%, impurity 0.20% each)

(For Data Reported in Figure and Table No. 161)

Curve No.	Rel. Method Used	Year	Temp. Range, K	Temp. Reported Range, K	Specimen Designation	Composition (weight percent)		Composition (continued), Specifications and Remarks
						Ni	Mn	
1	L	1925	333			99	10	Prepared from Ni (99.75 to 99.85 pure including Co, supplied by International Nickel Co.) and Mn (pure); specimen 5.5 cm long, 0.3 cm <sup>2</sup> cross sectional area; electrical conductivity 27.7 x 10 <sup>8</sup> ohm <sup>-1</sup> cm <sup>-1</sup> at 25 C.
2	L	1925	333			80	20	Similar to the above specimen except electrical conductivity 14.2 x 10 <sup>8</sup> ohm <sup>-1</sup> cm <sup>-1</sup> at 25 C.
3	L	1925	333			70	30	Similar to the above specimen except electrical conductivity 10.4 x 10 <sup>8</sup> ohm <sup>-1</sup> cm <sup>-1</sup> at 25 C.
4	L	1925	333			50	50	Similar to the above specimen except electrical conductivity 3.56 x 10 <sup>8</sup> ohm <sup>-1</sup> cm <sup>-1</sup> at 25 C.
5	C	1957	420-914		A-Nickel	99.54	0.25	0.07 Fe, 0.03 Co, 0.03 Si, 0.03 Mg and traces of other metals; Arisco iron used as standard.
6	F	1950	273-1173		A-Nickel	99.48	0.22	0.06 C, 0.02 S, 0.005 Si, 0.65 Cu, and 0.14 Fe; specimen 0.625 cm dia, 39 cm long; electrical conductivity reported as 11.5, 5.12, 2.87, 2.40, 2.10, and 2.02 x 10 <sup>8</sup> ohm <sup>-1</sup> cm <sup>-1</sup> at 0, 200, 400, 600, 800, and 900 C, respectively.
7	P	1928	305		Grade A	99.4	0.20	0.1 Cu, 0.15 Fe, 0.05 Si, 0.1 C, 0.005 S.
8	C	1954	422-910	5-19	A-Nickel	99.512	0.25	0.068 Fe, 0.034 each of Co and Mg, 0.03 Si, 0.02 Ti, 0.006 Al, 0.001 B, 0.014 Cu, 0.0005 each of Ca and Cr; specimen 2 cm dia and 15.25 cm long; Arisco iron used as comparative material.

DATA TABLE NO. 161 THERMAL CONDUCTIVITY OF NICKEL - MANGANESE ALLOYS

(Ni + Mn &lt;math&gt;\leq 99.50\%&lt;/math&gt;; impurity &lt;math&gt;\le 0.20\%&lt;/math&gt; each)

(Temperature, T, K; Thermal Conductivity, k, Watt/cm<sup>2</sup> K<sup>-1</sup>)

T	k	T	k
<u>CURVE 1</u>			
333	0.310	422.65	0.7130
		521.31	0.5909
<u>CURVE 2</u>			
		597.39	0.4636
333	0.176	618.55	0.4349
		650.22	0.4731
<u>CURVE 3</u>			
		678.87	0.4790
		739.15	0.4676
		799.59	0.4009
333	0.155	969.53	0.5777
<u>CURVE 4</u>			
333	0.092		
<u>CURVE 5</u>			
420.4	0.715		
520.0	0.590		
594.8	0.464		
617.4	0.439		
655.9	0.473		
680.8	0.481		
739.7	0.469		
800.8	0.485		
914.1	0.577		
<u>CURVE 6</u>			
273.2	0.728		
373.2	0.653		
473.2	0.577		
573.2	0.502		
673.2	0.477		
773.2	0.506		
873.2	0.536		
973.2	0.569		
1073.2	0.598		
1173.2	0.628		
<u>CURVE 7</u>			
305.2	0.615		

Not shown on plot



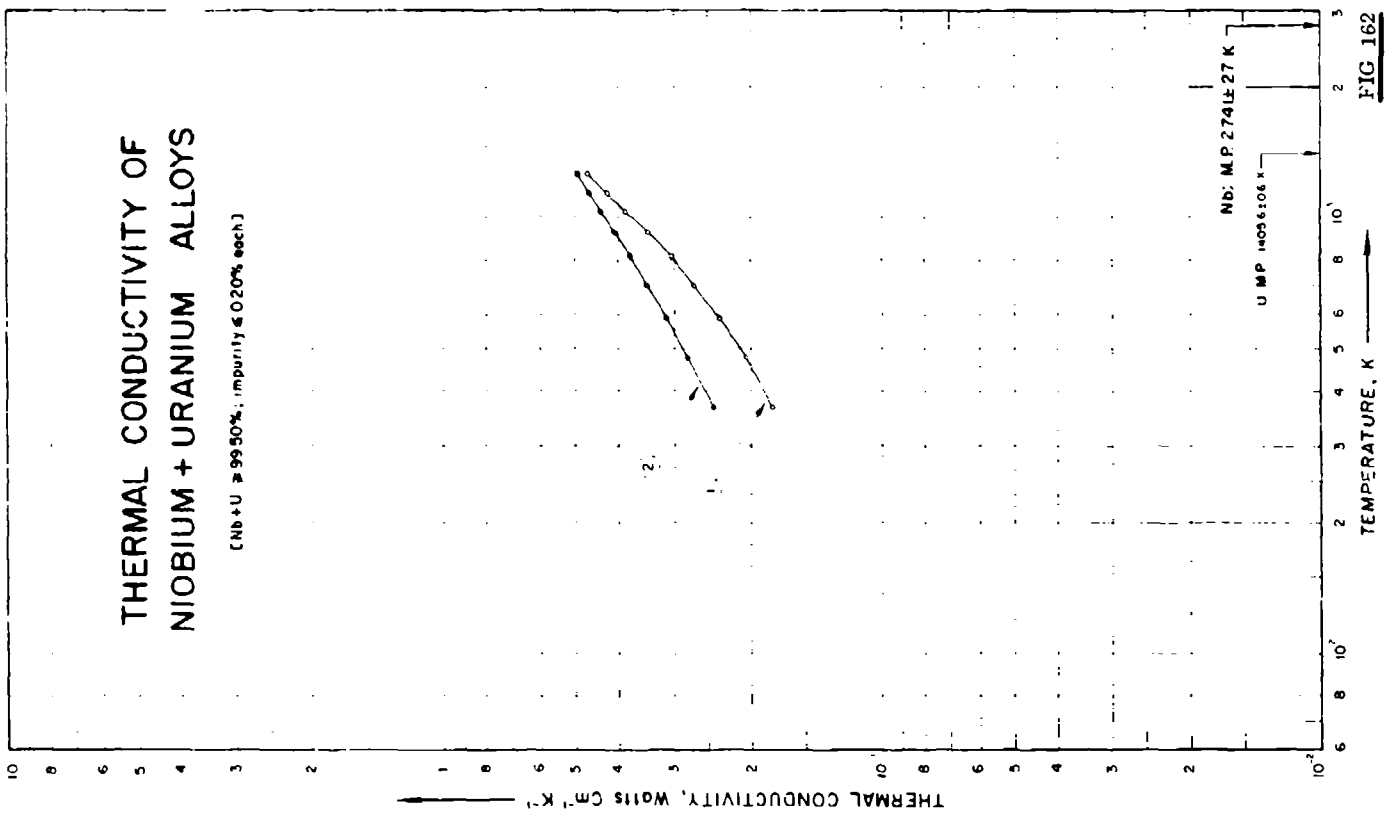


FIG. 162

## SPECIFICATION TABLE NO. 162 THERMAL CONDUCTIVITY OF (NIOBIA + URANIUM) ALLOYS

(Nb + U : 99.99%, impurities : 0.20% each)

[For Data Reported in Figure and Table No. 162]

Curve No.	Rel. Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) Nb	Composition (weight percent) U	Composition (continued), Specifications and Remarks
1	489 C	1961	367-1255			90	10	Prepared by arc casting Nb (99.9 pure) and U (pure), measured in vacuo of $2 \times 10^{-5}$ mm Hg. Arneo iron used as standard.
2	489 C	1961	367-1255			80	20	Similar to the above specimen.

DATA TABLE NO. 162 THERMAL CONDUCTIVITY OF (NIODIUM + URANIUM) ALLOYS

(Ni + U - 99.50%; impurity - 0.20% each)

[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup> K<sup>-1</sup>]

T	k	CURVE 1	CURVE 2
366.5	0.177		0.242
477.6	0.205		0.277
588.7	0.235		0.312
699.9	0.264		0.343
811.0	0.304		0.374
922.2	0.341		0.406
1033.2	0.384		0.439
1144.3	0.425		0.466
1255.4	0.470		0.493

FIGURE SHOWS ONLY IS OF THE CURVES REPORTED IN TABLE

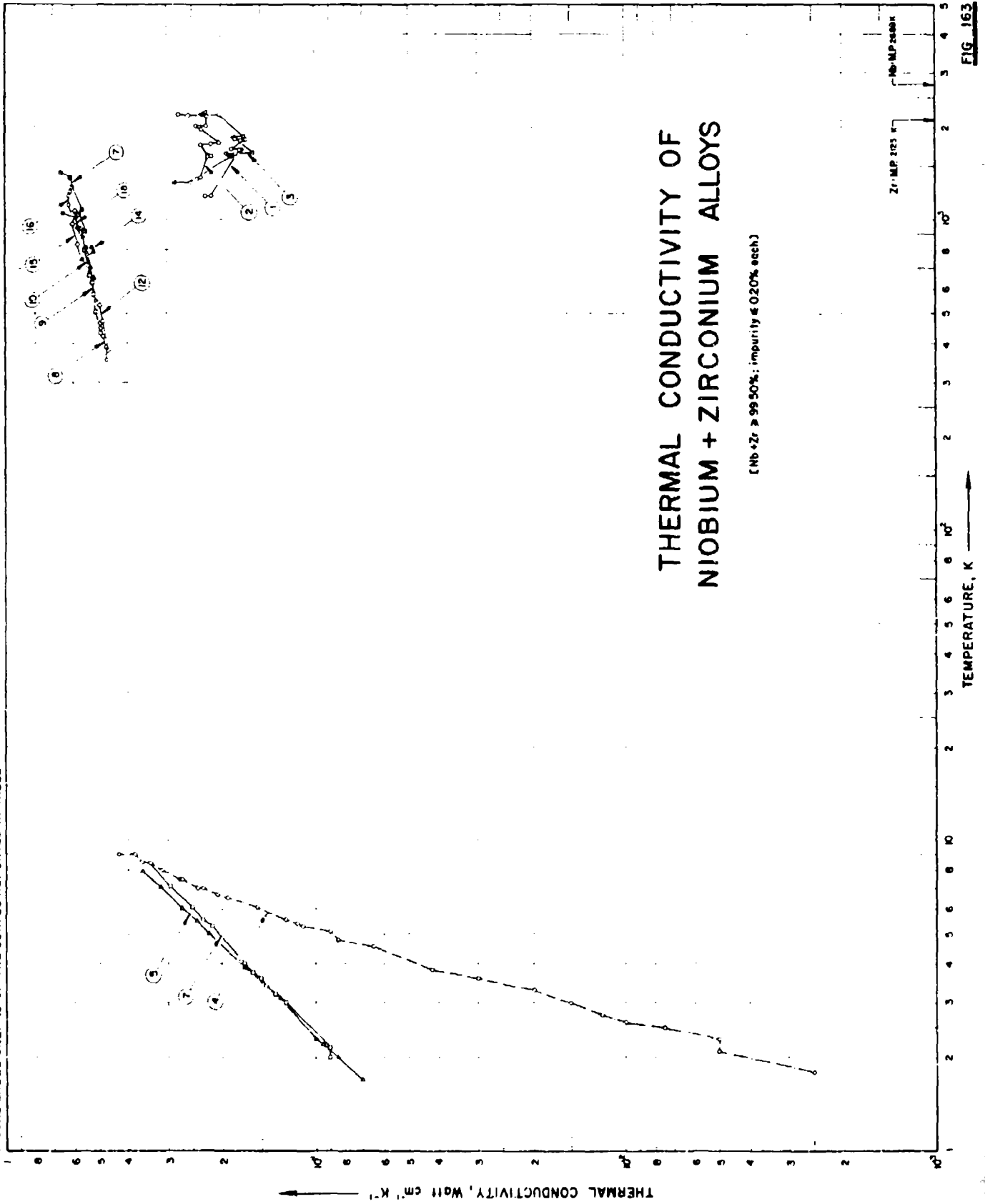


FIG. 163

SPECIFICATION TABLE NO. 163 THERMAL CONDUCTIVITY OF NIOBIUM-ZIRCONIUM ALLOYS

Nb-Zr 99.50% niobium, 0.20% zirconium

For Data Reported in Figure and Table No. 163

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) Nb	Composition (weight percent) Zr	Composition (continued), Specifications and Remarks
1	544	R	1963	1200-1835	±6	Nb-0.5 Zr	99.5	0.5	Specimen 0.75 m. long, 0.75 m. O.D., and 0.25 in. I.D.; ground; surface scratches eliminated; annealed at 3350 F. The above specimen remeasured; partially melted during test.
2	544	R	1963	1302-2175	±6	Nb-0.5 Zr	99.5	0.5	
3	544	R	1963	1322-2225	±6	Nb-0.5 Zr	99.5	0.5	
4	725	L	1963	138-910			99.8	0.2	Specimen 1 m. long, 2.5 m. O.D., 0.75 in. I.D.; ground; surface scratches eliminated; discolored during test.
5	725	L	1963	137-810			99.8	0.2	Specimen 1 mm wire; annealed in vacuo; in superconducting state.
6	725	L	1963	139-815			98.0	2.0	The above specimen measured in a magnetic field of 10,000 gauss; in normal state.
7	725	L	1963	230-815			98.0	2.0	Similar to the above specimen but in superconducting state.
8	766,925, 855	C	1963	355-465	±4		99.0	1.0	The above specimen measured in a magnetic field of 10,000 gauss; in normal state.
9	766,925, 855	C	1963	433-664	±4				Armed iron used as comparative material; interpolated electrical resistivity 19.5, 23.7, 27.7, 31.4, 35.1, 38.7, 42.1 and 45.4 μhm cm at 100, 200, 300, 400, 500, 600, 700 and 800 C, respectively; run No. 1, equilibrium No. 1. The above specimen run No. 1, equilibrium No. 2.
10	766,925, 855	C	1963	549-941	±4				The above specimen run No. 1, equilibrium No. 3.
11	766,925, 855	C	1963	616-1057	±4				The above specimen run No. 1, equilibrium No. 4.
12	766,925, 855	C	1963	370-531	±4				The above specimen run No. 1, equilibrium No. 5.
13	766,925, 855	C	1963	420-545	±4				The above specimen; run No. 2, equilibrium No. 1.
14	766,925, 855	C	1963	650-919	±4				The above specimen run No. 2, equilibrium No. 2.
15	766,925, 855	C	1963	722-1077	±4				The above specimen run No. 2, equilibrium No. 3.
16	766,925, 855	C	1963	918-1213	±4				The above specimen run No. 2, equilibrium No. 4.

SPECIFICATION TABLE NO. 163 (continued)

Curve No.	Ref. Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition Ni, Zr	Composition (weight percent) Zr	Composition (continued), Specifications and Remarks
17	766-925, C S35	1963	922-1421	±4				The above specimen run No. 2, equilibrium No. 5.
18	766-925, C S35	1963	534-1031	±4				The above specimen run No. 2, equilibrium No. 6.

## DATA TABLE NO. 163 THERMAL CONDUCTIVITY OF NIOBIUM + ZIRCONIUM ALLOYS

(Nb + Zr -99.50%, impurity 0.20% each)

Temperature, T, K; Thermal Conductivity,  $k$ , Watt cm<sup>-1</sup> K<sup>-1</sup>.

T	k	T	k	T	k	T	k	T	k	T	k	T	k	T	k
<u>CURVE 1</u>				<u>CURVE 4</u>				<u>CURVE 6</u>				<u>CURVE 8 (cont.)</u>			
1199.8	0.225	1.8	0.0025	1.9	0.0025	424.2	0.488	433.2	0.492	424.2	0.488	650.2	0.518	424.2	0.488
1199.8	0.215	2.1	0.005	2.1	0.0025	449.2	0.494	306.2	0.510	449.2	0.494	745.2	0.564	449.2	0.494
1199.8	0.214	2.3	0.005	2.5	0.010	465.2	0.495	577.2	0.520	465.2	0.495	819.2	0.526	465.2	0.495
1693.7	0.174	2.5	0.0075	2.6	0.012	<u>CURVE 9</u>				819.2	0.526	<u>CURVE 15</u>			
1697.1	0.172	2.6	0.010	2.9	0.015	433.2	0.492	433.2	0.492	433.2	0.492	722.2	0.549 <sup>†</sup>	433.2	0.492
1697.1	0.185	2.75	0.012	3.05	0.017	506.2	0.510	506.2	0.510	506.2	0.510	835.2	0.580	506.2	0.510
1833.7	0.167	3.0	0.015	3.2	0.020	630.2	0.524	630.2	0.524	630.2	0.524	971.2	0.601	630.2	0.524
1835.4	0.180	3.3	0.020	3.4	0.025	664.2	0.534	664.2	0.534	664.2	0.534	1077	0.593	664.2	0.534
<u>CURVE 2</u>				<u>CURVE 5</u>				<u>CURVE 10</u>				<u>CURVE 16</u>			
1332.1	0.240	4.8	0.085	4.3	0.055	549.2	0.512	549.2	0.512	549.2	0.512	818.2	0.573 <sup>†</sup>	549.2	0.512
1370.9	0.237	5.1	0.090	4.5	0.060	676.2	0.533	676.2	0.533	676.2	0.533	955.2	0.554 <sup>†</sup>	676.2	0.533
1610.9	0.221	5.3	0.110	5.05	0.080	797.2	0.553	797.2	0.553	797.2	0.553	1119	0.619	797.2	0.553
1611.5	0.214	5.4	0.115	5.5	0.120	885.2	0.558	885.2	0.558	885.2	0.558	1292	0.620	885.2	0.558
1618.7	0.220	5.55	0.125	6.5	0.155	941.2	0.568	941.2	0.568	941.2	0.568	1243	0.612	941.2	0.568
1760.4	0.231	6.1	0.155	7.05	0.230	<u>CURVE 11<sup>a</sup></u>				1243	0.612	<u>CURVE 17<sup>b</sup></u>			
1773.7	0.215	6.55	0.192	7.5	0.275	616.2	0.519	616.2	0.519	616.2	0.519	922.2	0.548	616.2	0.519
1777.1	0.202	6.7	0.207	7.5	0.320	770.2	0.543	770.2	0.543	770.2	0.543	1042	0.575	770.2	0.543
1976.5	0.232	7.0	0.230	7.95	0.320	915.2	0.570	915.2	0.570	915.2	0.570	1083	0.563	915.2	0.570
2014.8	0.240	7.05	0.240	8.45	0.367	1020	0.557	1020	0.557	1020	0.557	1276	0.604	1020	0.557
2014.8	0.224	7.5	0.270	3.2	0.135	1087	0.573	1087	0.573	1087	0.573	1377	0.613	1087	0.573
2158.2	0.223	7.5	0.275	3.4	0.145	<u>CURVE 12</u>				1421	0.658	<u>CURVE 18</u>			
2174.3	0.225	8.0	0.315	3.6	0.150	379.2	0.458	379.2	0.458	379.2	0.458	704.2	0.530	379.2	0.458
2175.9	0.255	3.45	0.340	3.75	0.160	427.2	0.462 <sup>†</sup>	427.2	0.462 <sup>†</sup>	427.2	0.462 <sup>†</sup>	816.2	0.548	427.2	0.462 <sup>†</sup>
2178.2	0.276	8.5	0.362	4.0	0.170	509.2	0.511 <sup>†</sup>	509.2	0.511 <sup>†</sup>	509.2	0.511 <sup>†</sup>	949.2	0.581	509.2	0.511 <sup>†</sup>
<u>CURVE 3</u>				<u>CURVE 7</u>				<u>CURVE 13<sup>b</sup></u>				<u>CURVE 19</u>			
1321.5	0.281	9.0	0.380	2.0	0.090	531.2	0.495	531.2	0.495	531.2	0.495	1020	0.588	531.2	0.495
1328.2	0.252	9.05	0.430	3.0	0.125	<u>CURVE 14<sup>a</sup></u>				1051	0.647	<u>CURVE 20</u>			
1616.5	0.185	1.7	0.070	3.2	0.135	420.2	0.480	420.2	0.480	420.2	0.480	460.2	0.492	420.2	0.480
1640.4	0.183	2.0	0.085	3.6	0.150	508.2	0.548	508.2	0.548	508.2	0.548	535.2	0.491	508.2	0.548
1641.5	0.192	2.2	0.095	3.75	0.160	545.2	0.490	545.2	0.490	545.2	0.490	390.2	0.475	545.2	0.490
1647.1	0.159	2.3	0.100	4.0	0.170	<u>CURVE 8</u>				390.2	0.475	<u>CURVE 14<sup>a</sup></u>			
1647.1	0.159	2.3	0.100	5.5	0.230	355.2	0.473	355.2	0.473	355.2	0.473	<u>CURVE 14<sup>a</sup></u>			
1854.8	0.183	3.1	0.130	3.0	0.125	390.2	0.475	390.2	0.475	390.2	0.475	<u>CURVE 14<sup>a</sup></u>			
1856.5	0.180 <sup>*</sup>	3.9	0.170	3.2	0.135	<u>CURVE 8</u>				390.2	0.475	<u>CURVE 14<sup>a</sup></u>			
1870.9	0.169	5.0	0.220	3.4	0.145	420.2	0.480	420.2	0.480	420.2	0.480	<u>CURVE 14<sup>a</sup></u>			
2188.2	0.203	5.5	0.240	3.6	0.150	460.2	0.492	460.2	0.492	460.2	0.492	<u>CURVE 14<sup>a</sup></u>			
2198.2	0.234	6.05	0.270	3.75	0.160	508.2	0.548	508.2	0.548	508.2	0.548	<u>CURVE 14<sup>a</sup></u>			
2225.4	0.225	7.05	0.315	3.9	0.170	545.2	0.490	545.2	0.490	545.2	0.490	<u>CURVE 14<sup>a</sup></u>			
		7.95	0.360	8.5	0.355 <sup>†</sup>	<u>CURVE 8</u>				390.2	0.475	<u>CURVE 14<sup>a</sup></u>			
				8.5	0.355 <sup>†</sup>	<u>CURVE 8</u>				390.2	0.475	<u>CURVE 14<sup>a</sup></u>			

\* Not shown on plot

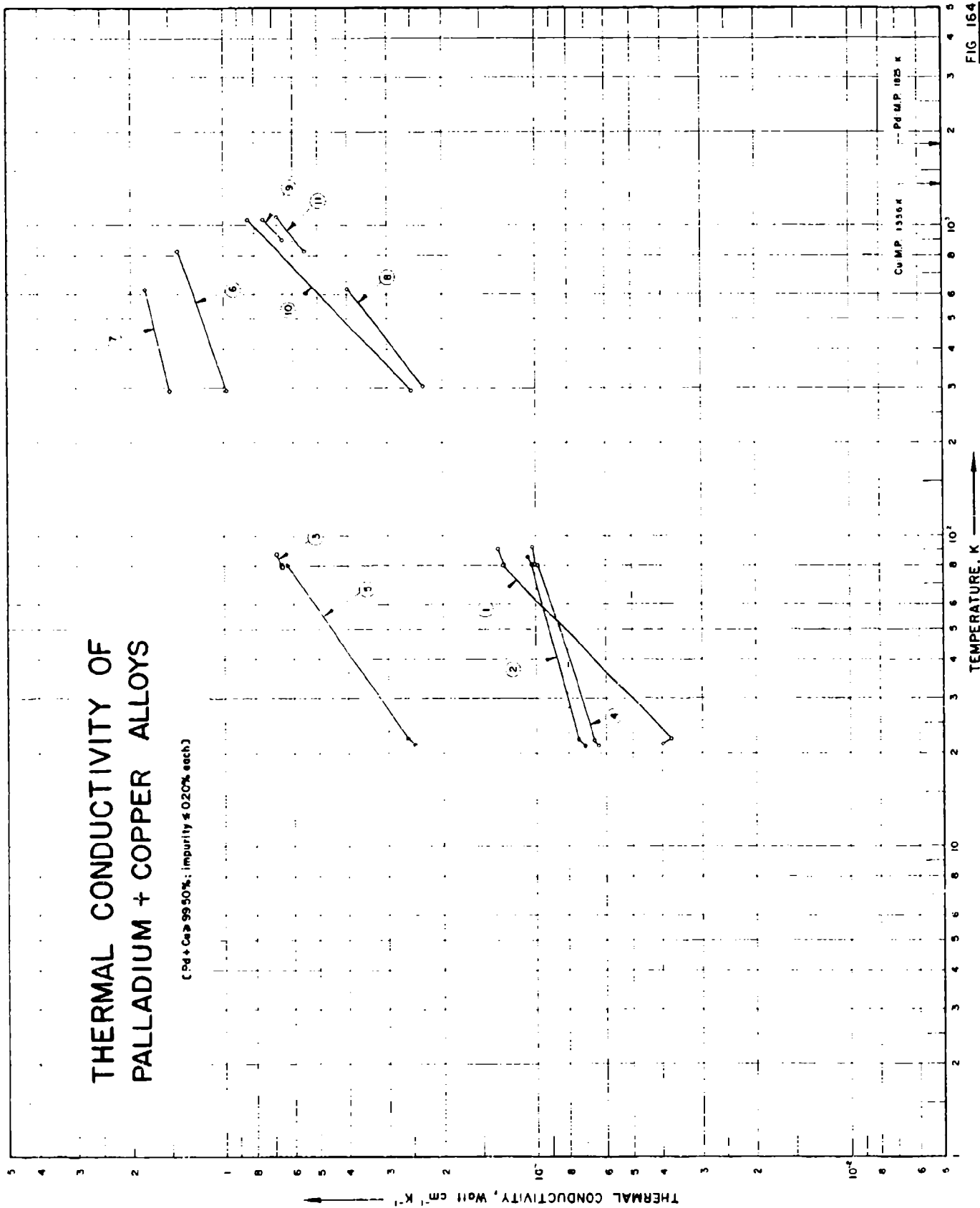


FIG 164



## SPECIFICATION TABLE NO. 164 THERMAL CONDUCTIVITY OF [PALLADIUM - COPPER ALLOYS

(Pd - Cu &gt; 99.50%; impurity &lt; 0.20%, each)

(For Data Reported in Figure and Table No. 164)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition Pd	Composition (weight percent) Cu	Composition (continued), Specifications and Remarks
1	58	L	1934	21-91		18	90.8	9.2	Unannealed polycrystal; $\rho = 28.65 \mu\text{hm cm}$ .
2	58	L	1934	21-95		19	62.7	37.3	Annealed, <del>26.5 mm cm</del> electrical resistivity <b>36.8</b> and <del>37.15 mm cm</del> at <del>275.15</del> <b>275.15</b> and <b>291.60 K, respectively.</b>
3	58	L	1934	75-97		21a	57.8	42.2	Annealed, <del>26.5 mm cm</del> electrical resistivity <b>5.1</b> and <del>5.32 mm cm</del> at <del>275.2</del> <b>275.2</b> and <b>292.6 K, respectively.</b>
4	58	L	1934	21-92		21b	57.8	42.2	The above specimen annealed in vacuo for 2 hrs at $\sim 550 \text{ C}$ .
5	58	L	1934	21-80		21c	57.8	42.2	The above specimen annealed at $\sim 325 \text{ C}$ for 30 hrs.
6	391	L	1958	293, 823	3		52.75	47.25	Calculated composition; specimen from a 0.2 mm thick sheet; cold rolled, annealed for 2 hrs at $\sim 550 \text{ C}$ ; ordered atomic arrangement.
7	391	L	1958	293, 623	3		57.81	42.19	Similar to the above specimen.
8	391	L	1958	303, 623	3		70.57	29.33	Similar to the above specimen.
9	391	L	1958	893, 1048	3		52.75	47.25	Similar to the above specimen except disordered atomic arrangement.
10	391	L	1958	293, 1048	3		57.81	42.19	Similar to the above specimen.
11	391	L	1958	821, 1073	3		70.67	29.33	Similar to the above specimen.

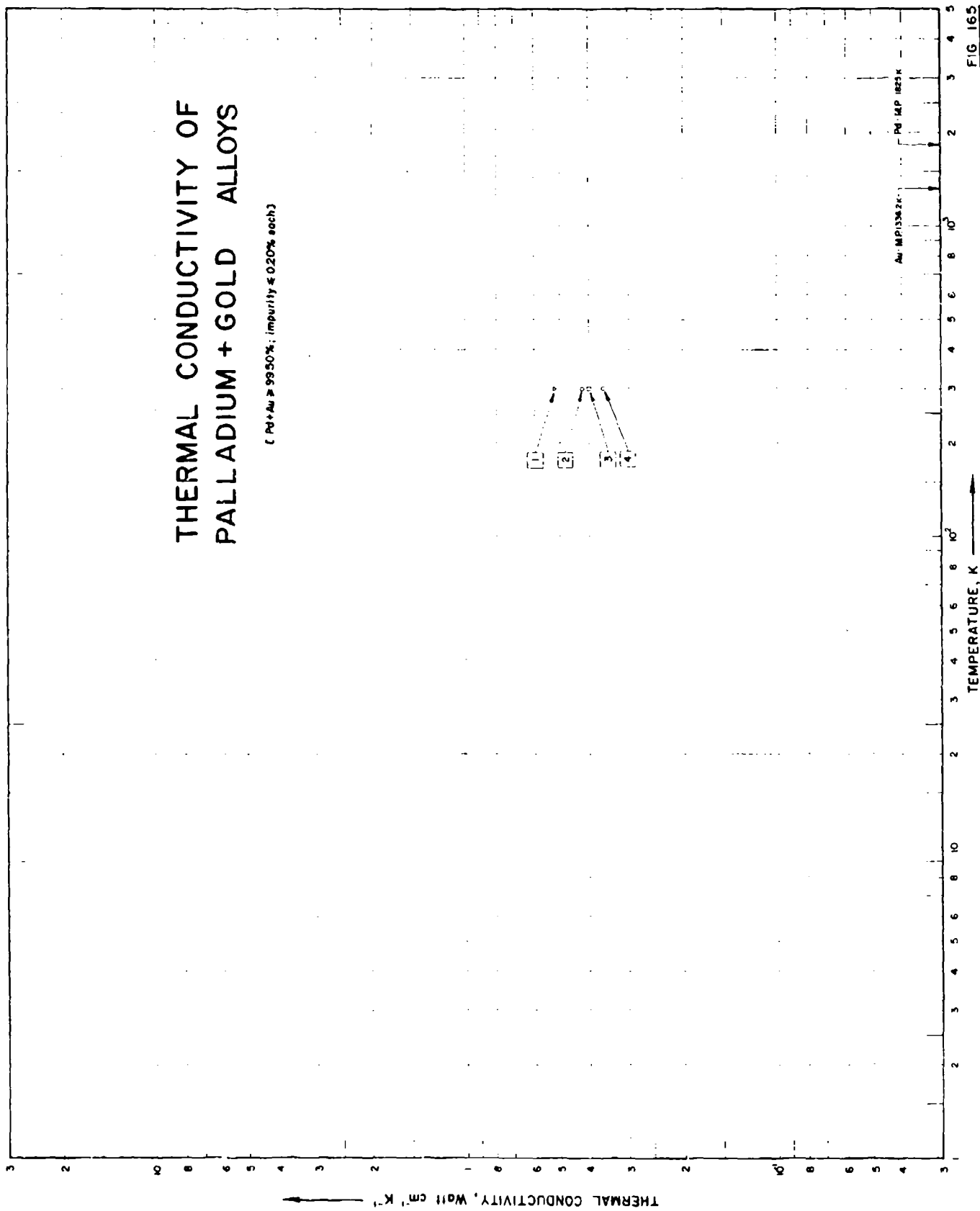
## DATA TABLE NO. 164 THERMAL CONDUCTIVITY OF [PALLADIUM + COPPER] ALLOYS

(Pd + Cu  $\geq$  99.50%; impurity  $\leq$  0.20% each)[Temperature, T, K. Thermal Conductivity, k, Watts cm<sup>-1</sup>K<sup>-1</sup>]

T	k	T	k
<u>CURVE 1</u>			
21.4	0.0396	303.2	0.23
22.2	0.0372	523.2	0.40
80.0	0.129	<u>CURVE 9</u>	
80.5	0.128	893.2	0.65
91.3	0.134	1048.2	0.75
<u>CURVE 2</u>			
21.0	0.0706	<u>CURVE 10</u>	
22.1	0.0742	293.2	0.25
80.4	0.103	1048.2	0.84
85.3	0.107	<u>CURVE 11</u>	
<u>CURVE 3</u>			
79.3	0.652	821.2	0.55
80.9	0.658	1073.2	0.68
86.6	0.683	<u>CURVE 4</u>	
<u>CURVE 5</u>			
21.1	0.0634	<u>CURVE 6</u>	
21.9	0.0658	21.3	0.246
80.2	0.0988	22.3	0.259
80.7	0.101	80.2	0.628
92.0	0.103	<u>CURVE 7</u>	
<u>CURVE 6</u>			
293.2	0.98	<u>CURVE 7</u>	
823.2	1.42	293.2	1.50
<u>CURVE 7</u>			
623.2	1.80	<u>CURVE 8</u>	

# THERMAL CONDUCTIVITY OF PALLADIUM + GOLD ALLOYS

(Pd+Au ≥ 99.50%; impurity ≤ 0.20% each)



## SPECIFICATION TABLE NO. 165 THERMAL CONDUCTIVITY OF [PALLADIUM + GOLD] ALLOYS

(Pd + Au  $\geq$  99.50%; impurity  $\leq$  0.20% each)

[ For Data Reported in Figure and Table No. 165 ]

Curve No.	Ref. Method No. Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition Pd	Composition (weight percent) Au	Composition (continued), Specifications and Remarks
1	241	E	1911	298.2		90	10	Approx. composition, electrical conductivity $6.65 \times 10^4$ ohm <sup>-1</sup> cm <sup>-1</sup> at 25 C.
2	241	E	1911	298.2		80	20	Approx. composition, electrical conductivity $5.33 \times 10^4$ ohm <sup>-1</sup> cm <sup>-1</sup> at 25 C.
3	241	E	1911	298.2		70	30	Approx. composition, electrical conductivity $4.72 \times 10^4$ ohm <sup>-1</sup> cm <sup>-1</sup> at 25 C.
4	241	E	1911	298.2		60	40	Approx. composition, electrical conductivity $3.89 \times 10^4$ ohm <sup>-1</sup> cm <sup>-1</sup> at 25 C.
5	241	E	1911	298.2		50	50	Approx. composition, electrical conductivity $3.74 \times 10^4$ ohm <sup>-1</sup> cm <sup>-1</sup> at 25 C.

## DATA TABLE NO. 165 THERMAL CONDUCTIVITY OF [PALLADIUM + GOLD] ALLOYS

(Pd + Au + 99.50%; Impurity  $\leq 0.20\%$  each)[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup> K<sup>-1</sup>]

T	k
<u>CURVE 1</u>	
298.2	0.52
<u>CURVE 2</u>	
298.2	0.42
<u>CURVE 3</u>	
298.2	0.40
<u>CURVE 4</u>	
298.2	0.36
<u>CURVE 5*</u>	
298.2	0.36

\*Not shown on plot

## SPECIFICATION TABLE NO. 166 THERMAL CONDUCTIVITY OF [PALLADIUM + PLATINUM] ALLOYS

(Pd + Pt : 99.50%; impurity  $\leq$  0.20% each)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) Pd	Composition (weight percent) Pt	Composition (continued), Specifications and Remarks
1	241	E	1911	298.2			90	10	Approx composition; electrical conductivity $6.56 \times 10^4$ ohm <sup>-1</sup> cm <sup>-1</sup> at 25 C.
2	241	E	1911	298.2			80	20	Approx composition; electrical conductivity $5.07 \times 10^4$ ohm <sup>-1</sup> cm <sup>-1</sup> at 25 C.
3	241	E	1911	298.2			70	30	Approx composition; electrical conductivity $4.43 \times 10^4$ ohm <sup>-1</sup> cm <sup>-1</sup> at 25 C.
4	241	E	1911	298.2			60	40	Approx composition; electrical conductivity $4.02 \times 10^4$ ohm <sup>-1</sup> cm <sup>-1</sup> at 25 C.
5	241	F	1911	298.2			50	50	Approx composition; electrical conductivity $3.79 \times 10^4$ ohm <sup>-1</sup> cm <sup>-1</sup> at 25 C.

## DATA TABLE NO. 166 THERMAL CONDUCTIVITY OF [PALLADIUM + PLATINUM] ALLOYS

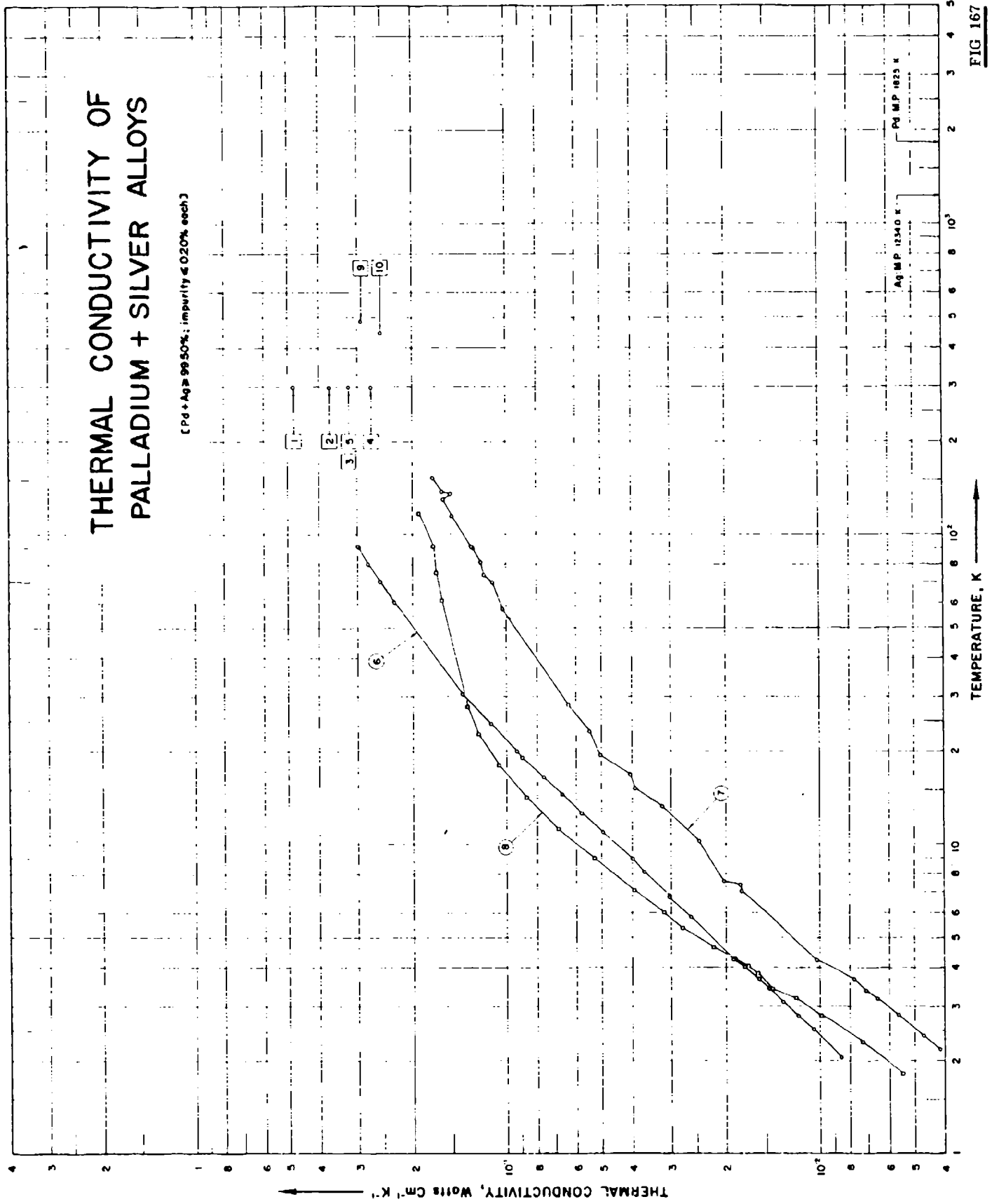
(Pd + Pt : 99.50%; impurity  $\leq$  0.20% each)[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup>K<sup>-1</sup>]

T	k	T	k
<u>CURVE 1<sup>a</sup></u>	<u>CURVE 4<sup>b</sup></u>		
298.2	0.560	298.2	0.380
<u>CURVE 2<sup>c</sup></u>	<u>CURVE 5<sup>d</sup></u>		
298.2	0.410	298.2	0.370
<u>CURVE 3<sup>e</sup></u>			
298.2	0.400		

No graphical presentation

# THERMAL CONDUCTIVITY OF PALLADIUM + SILVER ALLOYS

(Pd + Ag ≥ 99.50%; impurity ≤ 0.20% each)



SPECIFICATION TABLE NO. 167 THERMAL CONDUCTIVITY OF PALLADIUM - SILVER ALLOYS  
(Pd : Ag 99.50%: impurity 0.26% each)

[ For Data Reported in Figure and Table No. 167 ]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition Pd	Composition (weight percent) Ag	Composition (continued), Specifications and Remarks
1	241	E	1911	298.2			90	10	Approx. composition; electrical conductivity $4.71 \times 10^4$ ohm <sup>-1</sup> cm <sup>-1</sup> at 25 C.
2	241	E	1911	298.2			80	20	Approx. composition; electrical conductivity $3.21 \times 10^4$ ohm <sup>-1</sup> cm <sup>-1</sup> at 25 C.
3	241	E	1911	298.2			70	30	Approx. composition; electrical conductivity $2.56 \times 10^4$ ohm <sup>-1</sup> cm <sup>-1</sup> at 25 C.
4	241	E	1911	298.2			60	40	Approx. composition; electrical conductivity $2.38 \times 10^4$ ohm <sup>-1</sup> cm <sup>-1</sup> at 25 C.
5	241	E	1911	298.2			50	50	Approx. composition; electrical conductivity $3.03 \times 10^4$ ohm <sup>-1</sup> cm <sup>-1</sup> at 25 C.
6	240	L	1956	2.1-92			95	5	Specimen supplied by Johnson-Matthey and Co., Ltd.; annealed at 860 C; $\rho_0 = 5.81$ $\mu$ ohm cm; electrical resistivity 16.8 $\mu$ ohm cm at 293 K.
7	240	L	1956	2.2-152			70	30	Similar to the above specimen except $\rho_0 = 35.6$ $\mu$ ohm cm; electrical resistivity 40.9 $\mu$ ohm cm at 293 K.
8	240	L	1956	1.8-117			50	50	Similar to the above specimen except $\rho_0 = 27.7$ $\mu$ ohm cm; electrical resistivity 30.5 $\mu$ ohm cm at 293 K.
9	390	P	1956	486.7			75	25	
10	390	P	1956	448.2			50	50	



DATA TABLE NO. 167 THERMAL CONDUCTIVITY OF [PALLADIUM + SILVER] ALLOYS

(Pd + Ag  $\geq$  99.50%, impurity  $\leq$  0.20% each)[Temperature, T, K; Thermal Conductivity, k, Watts  $\text{cm}^{-1}\text{K}^{-1}$ ]

T	k	T	k	T	k
<u>CURVE 1</u>		<u>CURVE 6(cont.)</u>		<u>CURVE 8(cont.)</u>	
298.0	0.480	90.99	0.237	4.65	0.0220
<u>CURVE 2</u>		91.50	0.259	5.34	0.0276
<u>CURVE 3</u>		<u>CURVE 7</u>		5.99	0.0315
298.0	0.370	2.181	0.00413	7.08	0.0395
<u>CURVE 4</u>		2.418	0.00470	8.99	0.0528
298.0	0.320	2.805	0.00563	11.22	0.0684
<u>CURVE 5</u>		3.169	0.00655	14.17	0.0864
298.0	0.270	3.374	0.00708	18.0	0.1056
<u>CURVE 6</u>		3.659	0.00775	22.64	0.124
298.0	0.320	4.256	0.0102	27.65	0.134
<u>CURVE 7</u>		7.06	0.0178	61.48	0.162
298.0	0.270	7.41	0.0180	75.6	0.168
<u>CURVE 8</u>		7.59	0.0233	91.3	0.172
298.0	0.320	10.15	0.0245	117.2	0.191
<u>CURVE 9</u>		13.27	0.0320	<u>CURVE 9</u>	
298.0	0.270	15.07	0.0392	486.7	0.293
<u>CURVE 10</u>		16.81	0.0405	<u>CURVE 10</u>	
298.0	0.270	19.35	0.0503	448.2	0.251
<u>CURVE 11</u>		23.09	0.0544	<u>CURVE 11</u>	
298.0	0.270	28.24	0.0635	<u>CURVE 12</u>	
<u>CURVE 12</u>		57.45	0.103	<u>CURVE 12</u>	
298.0	0.270	70.04	0.111	<u>CURVE 13</u>	
<u>CURVE 13</u>		81.29	0.121	<u>CURVE 13</u>	
298.0	0.270	73.94	0.118	<u>CURVE 14</u>	
<u>CURVE 14</u>		91.23	0.130	<u>CURVE 14</u>	
298.0	0.270	90.8	0.124	<u>CURVE 15</u>	
<u>CURVE 15</u>		114.5	0.150	<u>CURVE 15</u>	
298.0	0.270	137.1	0.161	<u>CURVE 16</u>	
<u>CURVE 16</u>		129.7	0.160	<u>CURVE 16</u>	
298.0	0.270	136.2	0.151	<u>CURVE 17</u>	
<u>CURVE 17</u>		151.6	0.173	<u>CURVE 17</u>	
298.0	0.270	<u>CURVE 8</u>		<u>CURVE 18</u>	
<u>CURVE 18</u>		1.811	0.00546	<u>CURVE 18</u>	
298.0	0.270	2.315	0.00725	<u>CURVE 19</u>	
<u>CURVE 19</u>		2.823	0.00991	<u>CURVE 19</u>	
298.0	0.270	3.179	0.0119	<u>CURVE 20</u>	
<u>CURVE 20</u>		3.434	0.0134	<u>CURVE 20</u>	
298.0	0.270	3.833	0.0159	<u>CURVE 21</u>	
<u>CURVE 21</u>		4.017	0.0171	<u>CURVE 21</u>	
298.0	0.270	4.277	0.0186	<u>CURVE 22</u>	
<u>CURVE 22</u>		<u>CURVE 22</u>		<u>CURVE 22</u>	

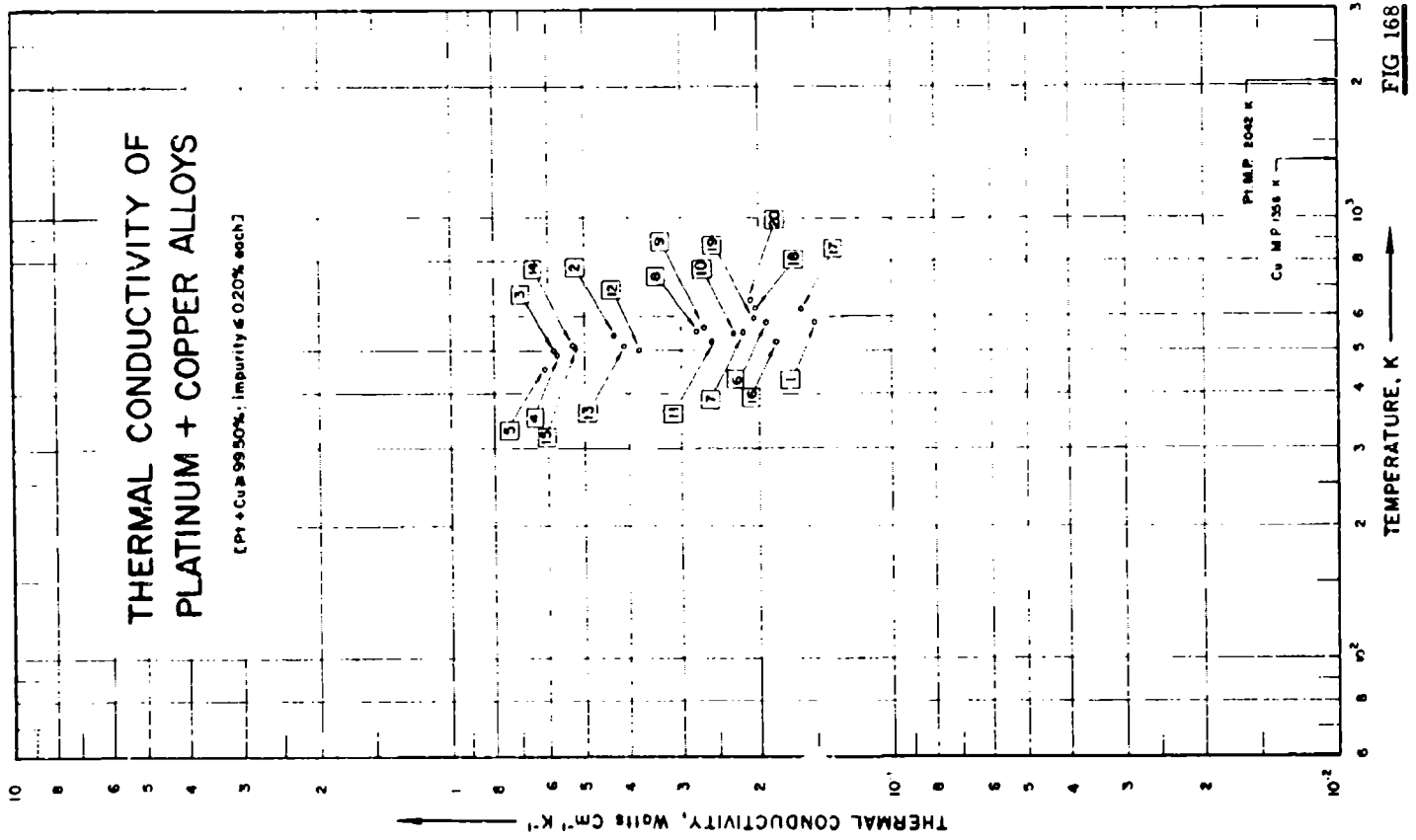


FIG 168

SPECIFICATION TABLE NO. 16<sup>s</sup> THERMAL CONDUCTIVITY OF [PLATINUM + COPPER] ALLOYS(Pt + Cu  $\geq$  99.50%; Impurity  $\leq$  0.20% each)(For Data Reported in Figure and Table No. 16<sup>s</sup>)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)		Composition (continued), Specifications and Remarks
							Pt	Cu	
1	232	L	1957	586.2			75.43	24.57	Cast.
2	232	L	1957	544.2			75.43	24.57	The above specimen after 10 hrs annealing.
3	232	L	1957	502.7			75.43	24.57	The above specimen after 15 hrs annealing.
4	232	L	1957	450.2			75.43	24.57	The above specimen after 50 hrs annealing.
5	232	L	1957	458.2			75.43	24.57	The above specimen after 100 hrs annealing.
6	232	L	1957	583.7			85.08	14.92	Cast.
7	232	L	1957	532.7			85.08	14.92	The above specimen after 10 hrs annealing.
8	232	L	1957	556.2			85.08	14.92	The above specimen after 20 hrs annealing.
9	232	L	1957	570.2			85.08	14.92	The above specimen after 50 hrs annealing.
10	232	L	1957	549.2			85.08	14.92	The above specimen after 100 hrs annealing.
11	232	L	1957	529.7			50.58	49.42	Cast.
12	232	L	1957	505.2			50.58	49.42	The above specimen after 5 hrs annealing.
13	232	L	1957	518.2			50.58	49.42	The above specimen after 10 hrs annealing.
14	232	L	1957	519.2			50.58	49.42	The above specimen after 20 hrs annealing.
15	232	L	1957	511.2			50.58	49.42	The above specimen after 100 hrs annealing.
16	232	L	1957	529.2			62.31	37.69	Cast.
17	232	L	1957	626.2			62.31	37.69	The above specimen after 10 hrs annealing.
18	232	L	1957	623.7			62.31	37.69	The above specimen after 20 hrs annealing.
19	232	L	1957	594.2			62.31	37.69	The above specimen after 40 hrs annealing.
20	232	L	1957	654.2			62.31	37.69	The above specimen after 50 hrs annealing.

DATA TABLE NO. 16<sup>s</sup> THERMAL CONDUCTIVITY OF (PLATINUM + COPPER) ALLOYS(Pt + Cu  $\geq$  99.50%; Impurity  $\leq$  0.20% each)(Temperature, T, K; Thermal Conductivity, k, Watts cm<sup>-1</sup>K<sup>-1</sup>)

T	k	T	k
	<u>CURVE 1</u>	<u>CURVE 11</u>	
586.2	0.151	529.7	0.259
	<u>CURVE 2</u>	<u>CURVE 12</u>	
544.2	0.431	505.2	0.377
	<u>CURVE 3</u>	<u>CURVE 13</u>	
502.7	0.586	518.2	0.410
	<u>CURVE 4</u>	<u>CURVE 14</u>	
490.2	0.573	519.2	0.531
	<u>CURVE 5</u>	<u>CURVE 15</u>	
458.2	0.615	511.2	0.527
	<u>CURVE 6</u>	<u>CURVE 16</u>	
582.7	0.192	529.2	0.184
	<u>CURVE 7</u>	<u>CURVE 17</u>	
552.7	0.218	626.2	0.163
	<u>CURVE 8</u>	<u>CURVE 18</u>	
556.2	0.280	629.7	0.205
	<u>CURVE 9</u>	<u>CURVE 19</u>	
570.2	0.272	594.2	0.205
	<u>CURVE 10</u>	<u>CURVE 20</u>	
549.2	0.230	654.2	0.209

SPECIFICATION TABLE NO. 169 THERMAL CONDUCTIVITY OF [PLATINUM + GOLD] ALLOYS  
(Pt + Au 99.50%; impurity = 0.20% each)

Curve No.	Rel. Method Used	Year	Temp. Range, K	Reported Error	Name and Specimen Designation	Composition (weight percent)		Composition (continued), Specifications and Remarks
						Pt	Au	
1	151	1930	291.2			54.75	44.25	Calculated composition; tempered at 800 C and quenched, then rolled and drawn.
2	151	1930	291.2			74.81	25.19	Similar to the above specimen.
3	151	1930	291.2			89.90	10.10	Similar to the above specimen.
4	151	1930	291.2			95.96	4.04	Similar to the above specimen.

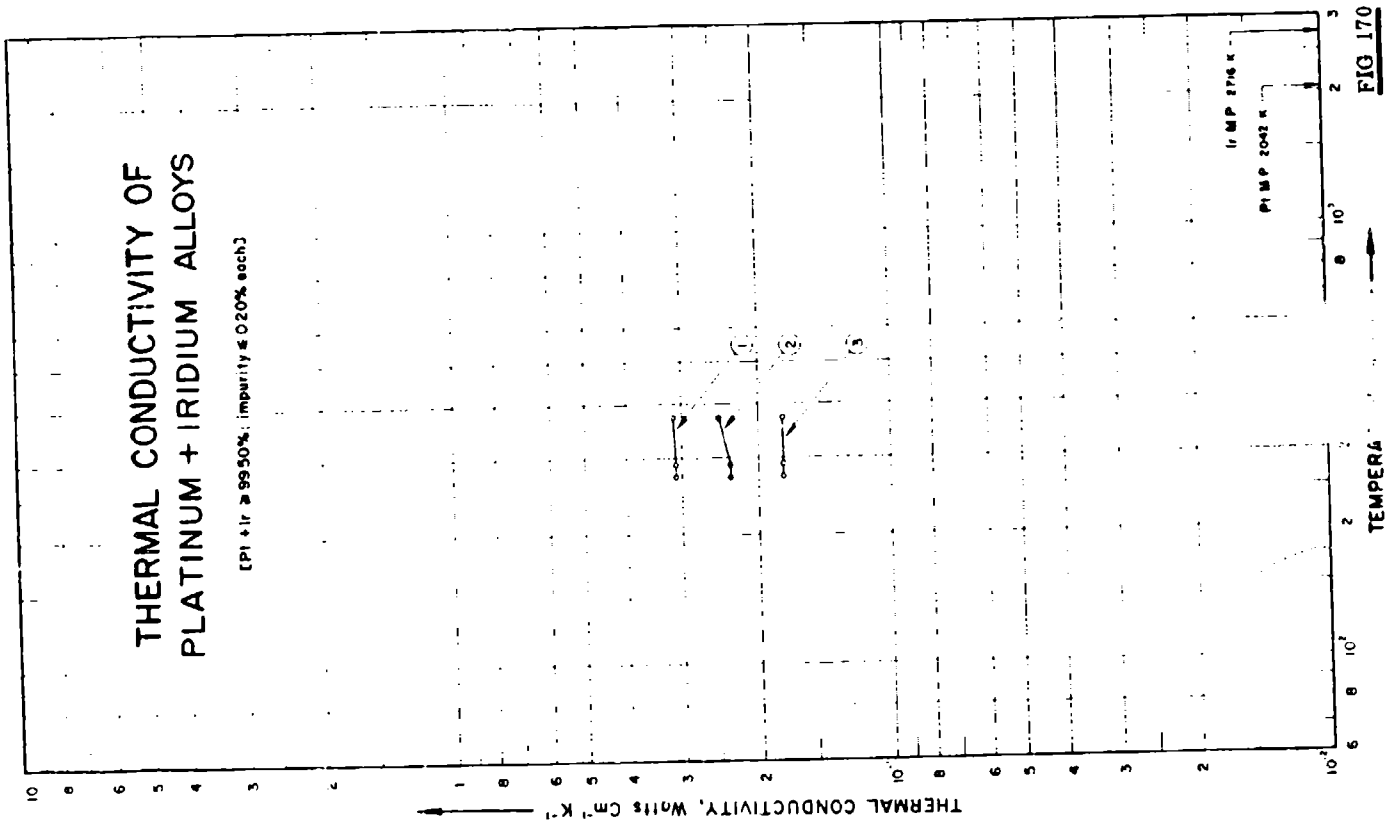
DATA TABLE NO. 169 THERMAL CONDUCTIVITY OF [PLATINUM + GOLD] ALLOYS

(Pt + Au 99.50%; impurity = 0.20% each)

(Temperature, T, K; Thermal Conductivity,  $k$ , Watt  $cm^{-1}K^{-1}$ )

T	k
<u>CURVE 1</u>	
291.2	0.21
<u>CURVE 2</u>	
291.2	0.24
<u>CURVE 3</u>	
291.2	0.35
<u>CURVE 4</u>	
291.2	0.46

No graphical presentation



## SPECIFICATION TABLE NO. 170 THERMAL CONDUCTIVITY OF PLATINUM IRIUM ALLOYS

(Pt - Ir - 99.50% impurity - 0.20% each)

(For Data Reported in Figure and Table No. 170.)

Curve No.	Rel. Method No. Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)		Composition (continued), Specifications and Remarks
						Pt	Ir	
1	S	1914	273-373		1	96	10	Approx. composition; electrical conductivity $3.58 \times 10^6$ ohm <sup>-1</sup> cm <sup>-1</sup> at 273 and 373 C. respectively.
2	S	1914	273-373		2	85	15	Approx. composition; electrical conductivity $3.49 \times 10^6$ ohm <sup>-1</sup> cm <sup>-1</sup> at 273 and 373 C. respectively.
3	S	1914	273-373		3	50	20	Approx. composition; electrical conductivity $3.02 \times 10^6$ ohm <sup>-1</sup> cm <sup>-1</sup> at 273 and 373 C. respectively.

## DATA TABLE NO. 170 THERMAL CONDUCTIVITY OF PLATINUM-IRIDIUM ALLOYS

(Pt + Ir 99.50% impurity - 0.20% each)

[ Temperature, T, K; Thermal Conductivity,  $k$ , Watt cm<sup>-1</sup> K<sup>-1</sup> ]

T	k
<u>CURVE 1</u>	
273.2	0.310
290.2	0.310
373.2	0.314
<u>CURVE 2</u>	
273.2	0.234
290.2	0.244
373.2	0.247
<u>CURVE 3</u>	
273.2	0.176
290.2	0.176
373.2	0.176



SPECIFICATION TABLE NO. 171 THERMAL CONDUCTIVITY OF [PLATINUM + PALLADIUM] ALLOYS

(Pt + Pd 99.50%; impurity  $\leq$  0.20% each)

Curve No.	Ref. Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) Pt	Composition (weight percent) Pd	Composition (continued), Specifications and Remarks
1	241 E	1911	298.2			50	50	Approx composition; electrical conductivity $3.75 \times 10^4$ ohm <sup>-1</sup> cm <sup>-1</sup> at 25 C.
2	241 E	1911	298.2			60	40	Approx composition; electrical conductivity $3.65 \times 10^4$ ohm <sup>-1</sup> cm <sup>-1</sup> at 25 C.
3	241 E	1911	298.2			70	30	Approx composition; electrical conductivity $3.80 \times 10^4$ ohm <sup>-1</sup> cm <sup>-1</sup> at 25 C.
4	241 E	1911	298.2			80	20	Approx composition; electrical conductivity $4.17 \times 10^4$ ohm <sup>-1</sup> cm <sup>-1</sup> at 25 C.
5	241 E	1911	298.2			90	10	Approx composition; electrical conductivity $5.20 \times 10^4$ ohm <sup>-1</sup> cm <sup>-1</sup> at 25 C.

DATA TABLE NO. 171 THERMAL CONDUCTIVITY OF [PLATINUM + PALLADIUM] ALLOYS

(Pt + Pd 99.50%; impurity  $\leq$  0.20% each)(Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup>K<sup>-1</sup>)

T	k	T	k
<u>CURVE 1*</u>			
298.2	0.370	298.2	0.42
<u>CURVE 2*</u>			
298.2	0.340	298.2	0.43
<u>CURVE 3*</u>			
298.2	0.360		

\* No graphical presentation

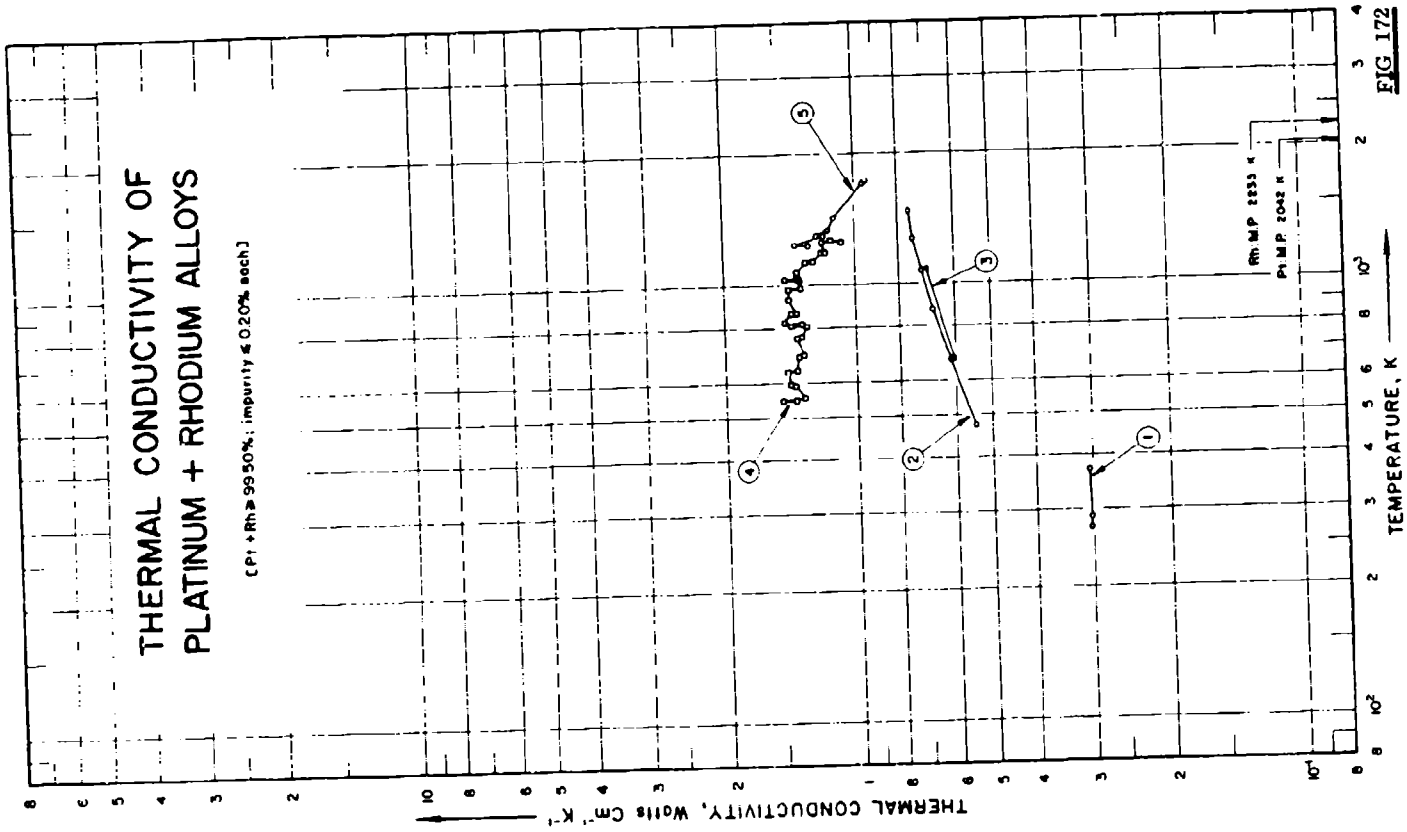


FIG. 172

## SPECIFICATION TABLE NO. 172 THERMAL CONDUCTIVITY OF PLATINUM RHOIDIUM ALLOYS

(R Rh 99.50; purity 0.20% each)

For Data Reported in Figure and Table No. 172:

Curve No.	Ref. Method Used	Year	Temp. Range, K	Revised Form	Name and Specimen Designation	Composition (weight percent) Pt	Composition (weight percent) Rh	Composition (continued), Specifications and Remarks
1	S	1914	273-277			90	10	Specimen 0.1018 cm in dia and 35.1 cm long; <b>22.45 and 25.49 <math>\mu\text{ohm cm}</math> electrical resistivity at <del>two</del> 0 and 100 C, respectively.</b>
2	S	1962	473-1473			60	40	0.01 - 0.1 Pt, 0.001 - 0.01 Cu, Ir, Pd, Si and Zr each, 0.001 each B and Ca; machined and then annealed at 1000 C. Specimen in the form of right cylinder of 2.539 cm in dia and 7.5 cm long; the solid portion of the cylinder 6.49 cm in length.
3	S	1962	673-1073			60	40	Similar to the above specimen; measured as the temperature decreases.
4	R	1962	548-1254			60	40	1 in. outside dia and 0.25 in. inside dia discs punched from 0.04 in. thick sheet and stacked to 1 in. high; annealed at 982 C for 30 min; grain size increase by 2 to 3 times after the measurement.
5	R	1962	1240-1747			60	40	The second run of the above specimen.

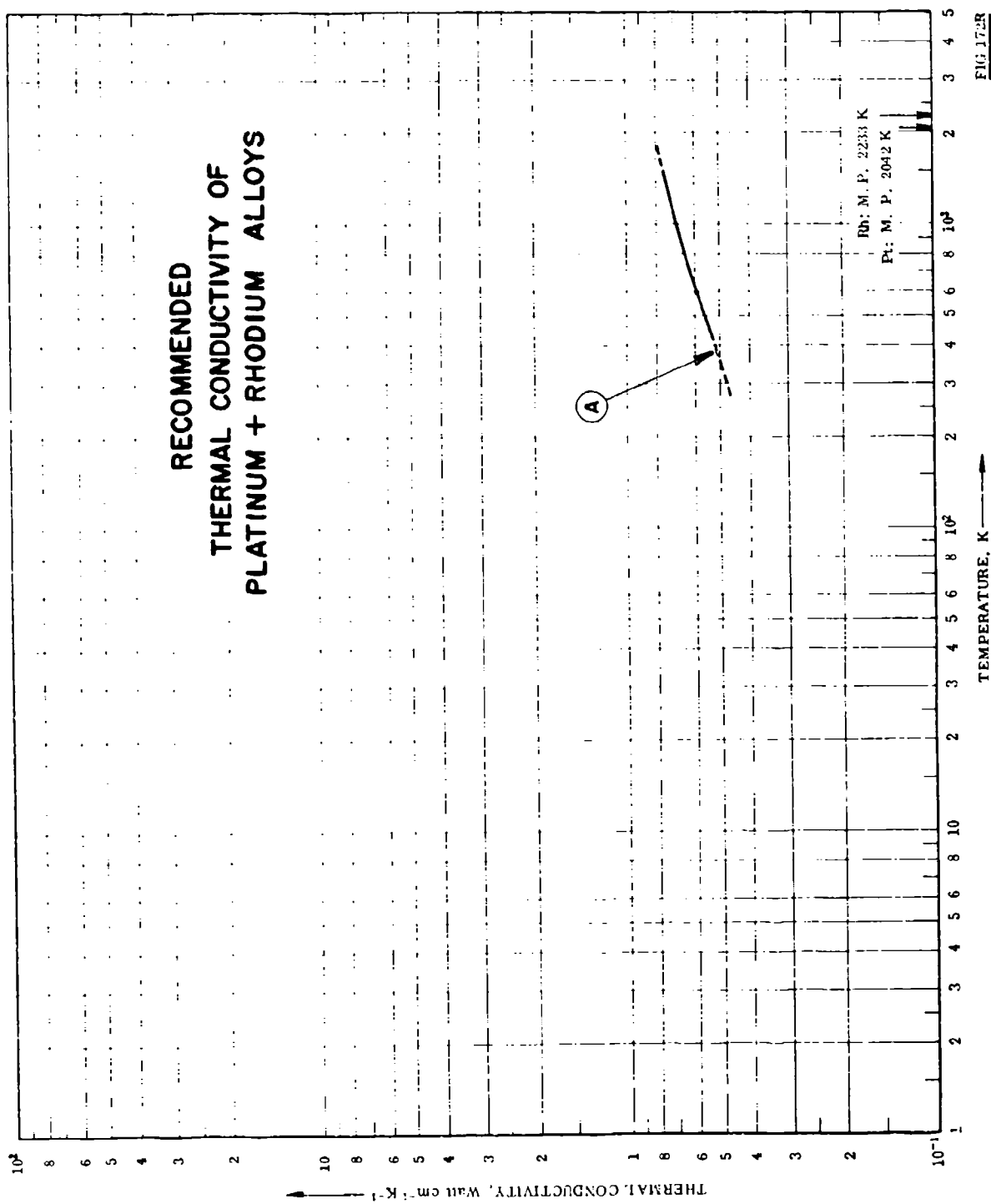
DATA TABLE NO. 172 THERMAL CONDUCTIVITY OF [PLATINUM - RHODIUM] ALLOYS

(Pt - Rh: 99.50% impurity 0.20% each)

[Temperature, T, K; Thermal Conductivity, k, Watts cm<sup>-1</sup>K<sup>-1</sup>]

T	k	T	k
<u>CURVE 1</u>			
273.2	0.301	927.1	1.40 <sup>1</sup>
290.2	0.301	984.3	1.42
373.2	0.305	985.4	1.44
		985.9	1.34
<u>CURVE 2</u>			
		1023.7	1.35
		1026.5	1.15
		1027.1	1.36
473.2	0.543	1079.3	1.36
673.2	0.615	1090.4	1.35 <sup>2</sup>
873.2	0.672	1082.1	1.35 <sup>2</sup>
1073.2	0.710	1332.1	1.30
1273.2	0.745	1334.3	1.25
1473.2	0.755	1498.2	1.20
		1498.2	1.17
<u>CURVE 3</u>			
673.2	0.609	1200.9	1.20 <sup>2</sup>
1073.2	0.693	1253.7	1.19
		1253.7	1.09
		1254.3	1.14
<u>CURVE 4</u>			
547.6	1.38	<u>CURVE 5</u>	
547.6	1.48	1240.4	1.24
549.8	1.32	1241.5	1.37
590.9	1.39	1300.4	1.18
591.5	1.41	1309.4	1.22
638.2	1.44	1309.4	1.19 <sup>3</sup>
638.7	1.37	1340.9	1.15
690.4	1.36 <sup>4</sup>	1245.3	1.16 <sup>2</sup>
690.4	1.39	1349.3	1.15 <sup>4</sup>
691.5	1.32	1437.6	1.12
760.4	1.37	1437.6	1.14 <sup>4</sup>
763.7	1.34	1405.9	1.11 <sup>4</sup>
763.7	1.36 <sup>5</sup>	1451.5	1.13 <sup>4</sup>
808.7	1.30	1710.9	0.952
809.8	1.41	1733.2	0.940 <sup>4</sup>
809.8	1.33	1747.1	0.966 <sup>4</sup>
825.9	1.46		
866.5	1.40		
866.5	1.37		
925.4	1.42		

Not shown on plot



## SPECIFICATION TABLE NO. 172R RECOMMENDED THERMAL CONDUCTIVITY OF [PLATINUM + RHODIUM] ALLOYS

[ For Data Reported in Figure and Data Table No. 172R ]

Curve No.	Name and Designation	Nominal Composition (weight percent) and Remarks	Estimated Error
A		60 Pt and 40 Rh, well annealed	± 5% from 500 to 1000 K and ± 5 to ± 10% below 500 K and above 1000 K.

## DATA TABLE NO. 172R RECOMMENDED THERMAL CONDUCTIVITY OF [PLATINUM + RHODIUM] ALLOYS

[ Temperature,  $T_1$  in K and  $T_2$  in F; Thermal Conductivity,  $k_1$  in Watt  $\text{cm}^{-1} \text{K}^{-1}$  and  $k_2$  in Btu  $\text{hr}^{-1} \text{ft}^{-1} \text{F}^{-1}$  ]

$T_1$	$k_1$	$k_2$	$T_2$	$T_1$	$k_1$	$k_2$	$T_2$
273.2	(0.464)†	(26.8)	32.0	1500	(0.758)	(43.8)	2240
300	(0.467)	(27.0)	80.3	1600	(0.768)	(44.4)	2420
350	(0.497)	(28.7)	170.3	1700	(0.778)	(45.0)	2600
400	(0.516)	(29.6)	260.3	1800	(0.787)	(45.5)	2780
500	0.555	32.1	440.3				
600	0.589	34.0	620.3				
700	0.623	36.0	800.3				
800	0.650	37.6	980.3				
900	0.672	38.8	1160				
1000	0.692	40.0	1340				
1100	0.711	41.1	1520				
1200	0.727	42.0	1700				
1300	0.738	42.6	1880				
1400	0.748	43.2	2060				

† Values in parentheses are extrapolated.

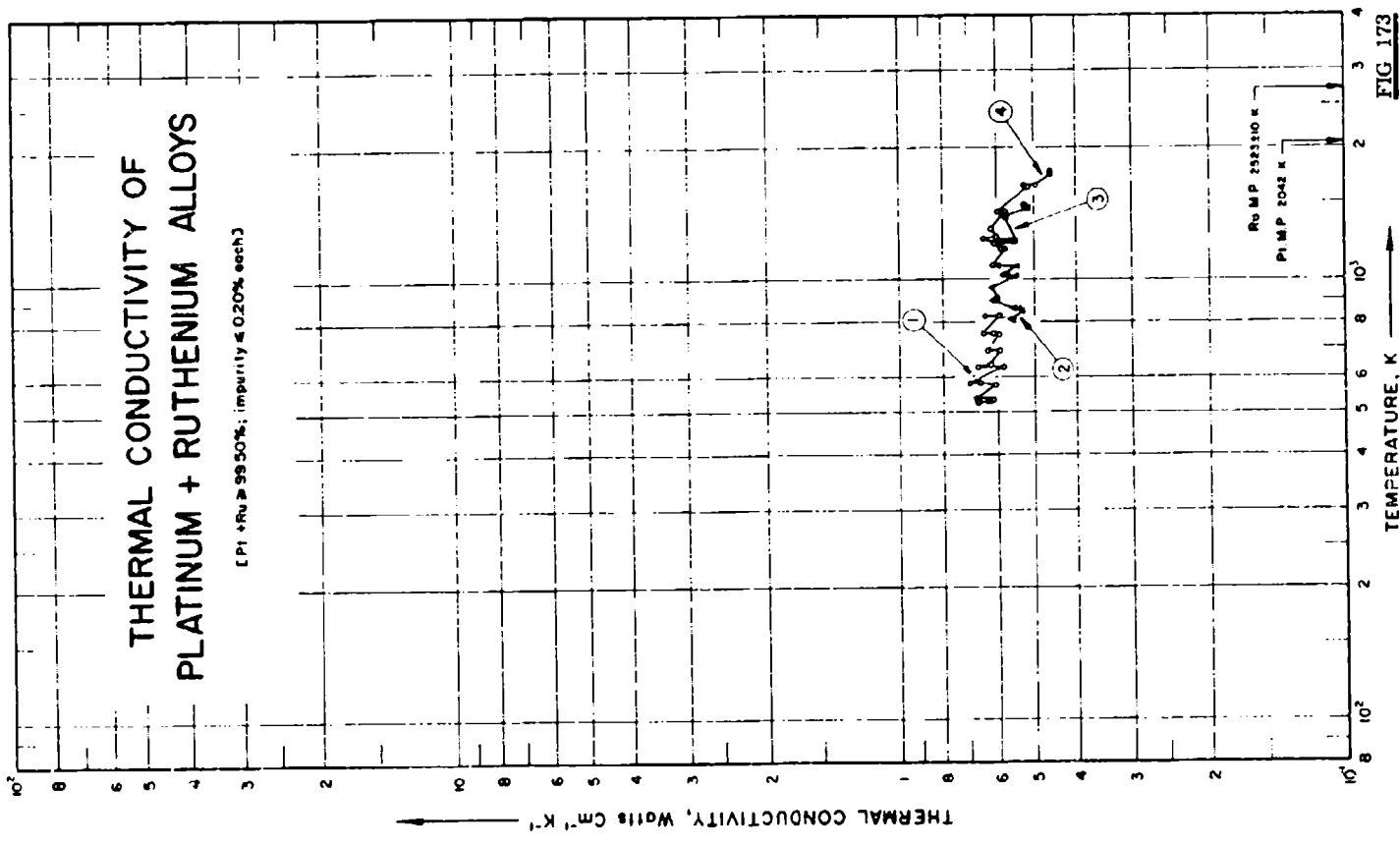


FIG 173

SPECIFICATION TABLE NO. 173 THERMAL CONDUCTIVITY OF (PLATINUM + RUTHENIUM) ALLOYS  
(Pt + Ru - 99.50% impurity - 0.20% each)

For Data Reported in Figure and Table No. 173

Curve No.	Ref. Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)		Composition (continued), Specifications and Remarks
						Pt	Ru	
1	589 R	1962	528-826			95	5	Specimen prepared by stacking 1 in. O. D. and 0.25 in. I. D. discs punched from a 0.040 in. thick sheet to a thickness of 1 in.; annealed at 952 C for 30 min; grain size increased by 2 to 3 times after the measurement. The second run of the above specimen. The third run of the above specimen. The fourth run of the above specimen.
2	589 R	1962	909-1089			95	5	
3	589 R	1962	1221-1472			95	5	
4	589 R	1962	1072-1763			95	5	

DATA TABLE NO. 173 THERMAL CONDUCTIVITY OF (PLATINUM + RUTHENIUM) ALLOYS  
(Pt + Ru - 99.50% impurity - 0.20% each)

(Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup>K<sup>-1</sup>)

T	k	CURVE 1 (cont.)		T	k	CURVE 2 (cont.)		T	k	CURVE 3		T	k	CURVE 4 (cont.)	
		T	k			T	k			T	k			T	k
527.6	0.653			857.6	0.551	1221.1	0.565	1189.3	0.588						
530.4	0.626	754.9	0.649	859.8	0.613	1222.1	0.590	1211.5	0.606						
531.5	0.619	824.3	0.587	900.9	0.614	1222.1	0.587	1227.6	0.584						
533.7	0.765	824.8	0.590	903.2	0.600	1222.1	0.550	1247.1	0.599						
580.4	0.610	825.9	0.601	959.8	0.616	1395.4	0.574	1249.8	0.643						
581.5	0.659	825.9	0.640	961.5	0.623	1395.4	0.568	1249.8	0.639						
582.1	0.694			962.1	0.627	1395.4	0.583	1252.2	0.603						
634.3	0.584			1015.4	0.544	1363.2	0.511	1314.8	0.619						
634.3	0.665			1018.2	0.548	1471.5	0.524	1314.8	0.614						
634.8	0.622			1018.7	0.571			1427.6	0.594						
690.4	0.594	808.2	0.557	1021.5	0.573			1653.7	0.513						
690.9	0.590	810.4	0.562	1023.7	0.584			1635.9	0.524						
690.9	0.630	813.7	0.571	1078.7	0.544			1635.9	0.492						
759.9	0.597	870.4	0.531	1083.2	0.547	1072.1	0.610	1738.2	0.479						
752.1	0.614	852.1	0.522	1088.7	0.600	1072.1	0.593	1738.2	0.479						
		854.3	0.537			1082.1	0.610	1763.2	0.459						
		855.9	0.521			1183.2	0.577								

Not shown on plot



SPECIFICATION TABLE NO. 174 THERMAL CONDUCTIVITY OF (PLATINUM + SILVER) ALLOYS  
(Pt + Ag = 99.50%; impurity = 0.20% each)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)		Composition (continued), Specifications and Remarks
							Pt	Ag	
1	390	PL	1956	553.2			80	20	No detail reported.
2	390	PL	1956	506.7			75	25	No detail reported.
3	390	PL	1956	521.2			50	50	No detail reported.

DATA TABLE NO. 174 THERMAL CONDUCTIVITY OF (PLATINUM + SILVER) ALLOYS

(Pt + Ag = 99.50%; impurity = 0.20% each)

[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup>K<sup>-1</sup>]

T k

CURVE 1\*  
553.2 0.363

CURVE 2\*  
506.7 0.249

CURVE 3\*  
521.2 0.272

\* No graphical presentation

SPECIFICATION TABLE NO. 175 THERMAL CONDUCTIVITY OF [PLUTONIUM - ALUMINUM] ALLOYS

Pu - Al 99.50% impurity 0.20% each

Curve No.	Ref. Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)	Composition (continued), Specifications and Remarks
1	926	C	1960	5-10	DeBe-stabilized	Pu 99.0 Al 1.0	Specimen 2 cm in dia and 5 cm long; Inconel and Armco Iron used as the comparative material.

DATA TABLE NO. 175 THERMAL CONDUCTIVITY OF [PLUTONIUM - ALUMINUM] ALLOYS

(Pu - Al 99.50% impurity 0.20% each)

[ Temperature, T, K; Thermal Conductivity,  $k$ , Watt cm<sup>-1</sup>K<sup>-1</sup> ]

T	k
CURVE 1	
373.2	0.092
473.2	0.115
573.2	0.138
673.2	0.159
773.2	0.188
873.2	0.209

No graphical presentation

## SPECIFICATION TABLE NO. 176 THERMAL CONDUCTIVITY OF [PLUTONIUM + IRON] ALLOYS

(Pu + Fe - 99.50%; impurity  $\leq$  0.20% each)

Curve No.	Ref. Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) Pu	Composition (weight percent) Fe	Composition (continued), Specifications and Remarks
1	182	L	1959	473-823	5.0	90.5	9.5	Measured in vacuum of $10^{-4}$ mm Hg; melting point at 410 C.

## DATA TABLE NO. 176 THERMAL CONDUCTIVITY OF [PLUTONIUM + IRON] ALLOYS

(Pu + Fe - 99.50%; impurity  $\leq$  0.20% each)[Temperature, T, K; Thermal Conductivity, k, Watt  $\text{cm}^{-1}\text{K}^{-1}$ ]

T	k
473.20	0.138
523.20	0.148
573.20	0.156
623.20	0.164
673.20	0.173
693.20	0.179
723.20	0.184
773.20	0.191
823.20	0.197

No graphical presentation

# THERMAL CONDUCTIVITY OF POTASSIUM + SODIUM ALLOYS

(K + Na = 99.50%; impurity  $\leq$  0.20% each)

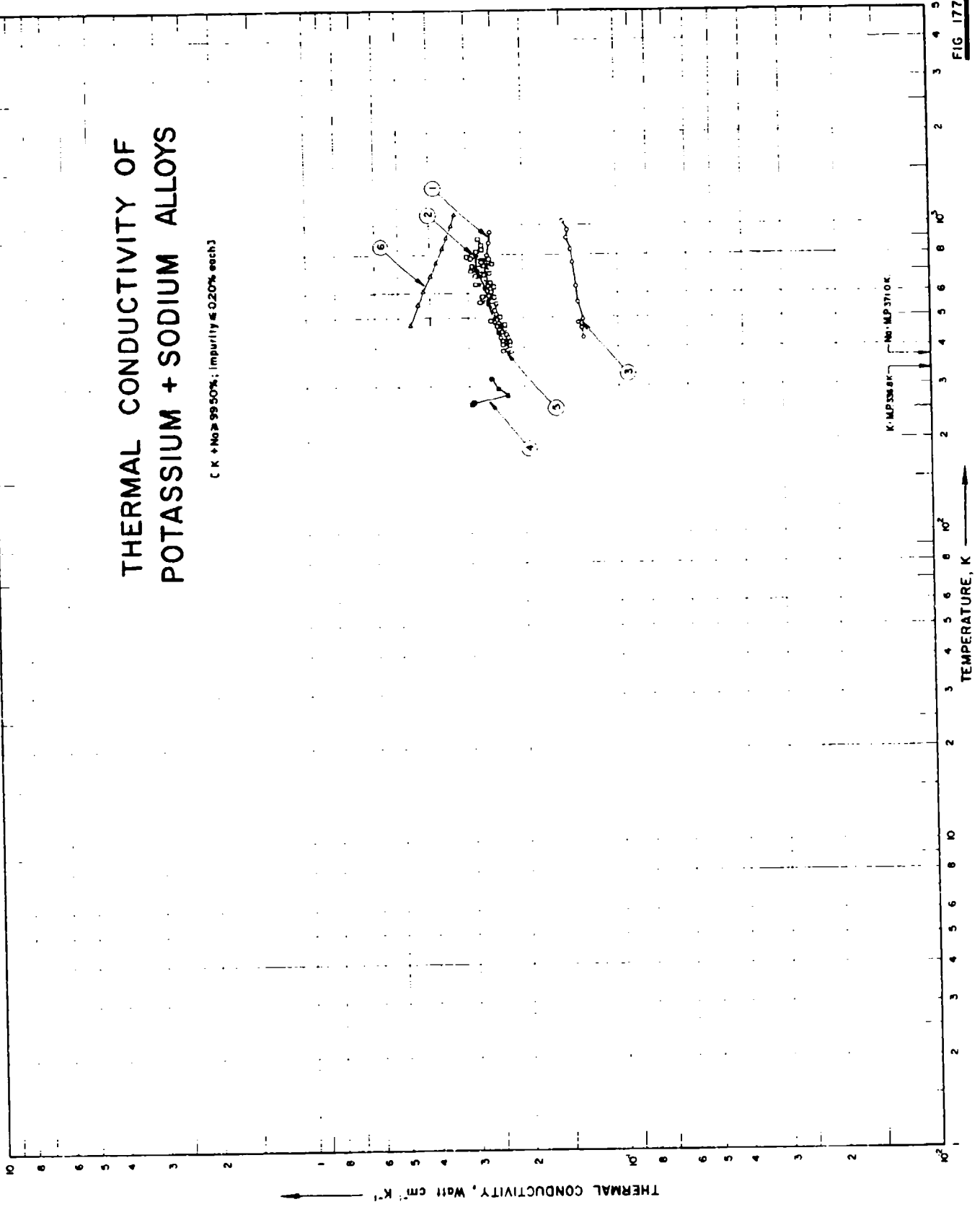


FIG. 177

SPECIFICATION TABLE NO. 177 THERMAL CONDUCTIVITY OF POTASSIUM - SODIUM ALLOYS

(K : Na = 99.50%; impurity 0.20%)

For Data Reported in Figure and Table No. 177

Curve No.	Ret. Method Used	Year	Temp. Range, K	Reported Error, %	Name of Specimen Designation	Composition (weight percent) K	Composition (weight percent) Na	Composition (continued), Specifications and Remarks
1	L	1955	430-950	1.0		77.7	22.3	Highly purified; specimen in liquid state; electrical resistivity 41.6, 44.4, 47.3, 50.8, 67.3, 75.3, and 89.2 $\mu\text{ohm cm}$ at 150, 200, 300, 400, 500, 600, and 700 C respectively.
2	L	1955	449-794	1.0		56.5	43.5	Highly purified; specimen in liquid state; electrical resistivity 47.23, 54.33, 62.21, and 69.37 $\mu\text{ohm cm}$ at 200, 300, 400, and 500 C respectively.
3	P	1956	432-1030			78.0	22.0	In liquid state.
4	E	1913	262-316			63.0	37.0	Very pure; sodium free from Fe, Ca, Mg, Al, and K; potassium free from Fe, Ca, Mg, and Al; supplied by Kaiser and Amend; specimen 1.562 cm I.D. and 20 cm long; electrical resistivity 21.93, 22.10, 23.04, 24.11, 25.13, 26.17, 27.23, 28.33, and 30.40 $\mu\text{ohm cm}$ at -10, 0, +5, 10, 15, 20, 25, 30, 35, 40, 45, 50, 55, 60, 65, 70, 75, 80, 85, 90, 95, 100, 105, 110, 115, 120, 125, 130, 135, 140, 145, 150, 155, 160, 165, 170, 175, 180, 185, 190, 195, 200, 205, 210, 215, 220, 225, 230, 235, 240, 245, 250, 255, 260, 265, 270, 275, 280, 285, 290, 295, 300, 305, 310, 315, 320, 325, 330, 335, 340, 345, 350, 355, 360, 365, 370, 375, 380, 385, 390, 395, 400, 405, 410, 415, 420, 425, 430, 435, 440, 445, 450, 455, 460, 465, 470, 475, 480, 485, 490, 495, 500, 505, 510, 515, 520, 525, 530, 535, 540, 545, 550, 555, 560, 565, 570, 575, 580, 585, 590, 595, 600, 605, 610, 615, 620, 625, 630, 635, 640, 645, 650, 655, 660, 665, 670, 675, 680, 685, 690, 695, 700, 705, 710, 715, 720, 725, 730, 735, 740, 745, 750, 755, 760, 765, 770, 775, 780, 785, 790, 795, 800, 805, 810, 815, 820, 825, 830, 835, 840, 845, 850, 855, 860, 865, 870, 875, 880, 885, 890, 895, 900, 905, 910, 915, 920, 925, 930, 935, 940, 945, 950, 955, 960, 965, 970, 975, 980, 985, 990, 995, 1000 C respectively; specimen in liquid state.
5	L	1959	388-899			75	25	In liquid state.
6	758, 862	1962	473-1073			Bal.	0.32	0.02 Fe and 0.004 O (post test); molten specimen contained in a type 347 stainless steel tube; supplied by Fisher Scientific Co.; electrical resistivity reported as 8.07, 8.24, 8.39, 8.53, 9.47, 9.83, 14.75, 15.48, 17.00, 21.80, 26.06, 29.57, 34.11, 38.32, 43.10, 48.47, 54.04, 59.47, 66.02, 66.75, and 74.30 $\mu\text{ohm cm}$ at 25, 30, 35, 40, 45, 50, 55, 60, 65, 70, 75, 80, 85, 90, 95, 100, 105, 110, 115, 120, 125, 130, 135, 140, 145, 150, 155, 160, 165, 170, 175, 180, 185, 190, 195, 200, 205, 210, 215, 220, 225, 230, 235, 240, 245, 250, 255, 260, 265, 270, 275, 280, 285, 290, 295, 300, 305, 310, 315, 320, 325, 330, 335, 340, 345, 350, 355, 360, 365, 370, 375, 380, 385, 390, 395, 400, 405, 410, 415, 420, 425, 430, 435, 440, 445, 450, 455, 460, 465, 470, 475, 480, 485, 490, 495, 500, 505, 510, 515, 520, 525, 530, 535, 540, 545, 550, 555, 560, 565, 570, 575, 580, 585, 590, 595, 600, 605, 610, 615, 620, 625, 630, 635, 640, 645, 650, 655, 660, 665, 670, 675, 680, 685, 690, 695, 700, 705, 710, 715, 720, 725, 730, 735, 740, 745, 750, 755, 760, 765, 770, 775, 780, 785, 790, 795, 800, 805, 810, 815, 820, 825, 830, 835, 840, 845, 850, 855, 860, 865, 870, 875, 880, 885, 890, 895, 900, 905, 910, 915, 920, 925, 930, 935, 940, 945, 950, 955, 960, 965, 970, 975, 980, 985, 990, 995, 1000 C respectively; thermal conductivity data calculated from measured electrical resistivity values and the Lorenz number based on thermal conductivity data of Ewing, C. T. and Grand, J. A. (AI R. port 3833, 1954).

DATA TABLE NO. 177 THERMAL CONDUCTIVITY OF [ POTASSIUM + SODIUM ] ALLOYS

(K + Na <99.50%; impurity <0.20% each)  
 [ Temperature, T, K, Thermal Conductivity, k, Watt cm<sup>-1</sup> K<sup>-1</sup> ]

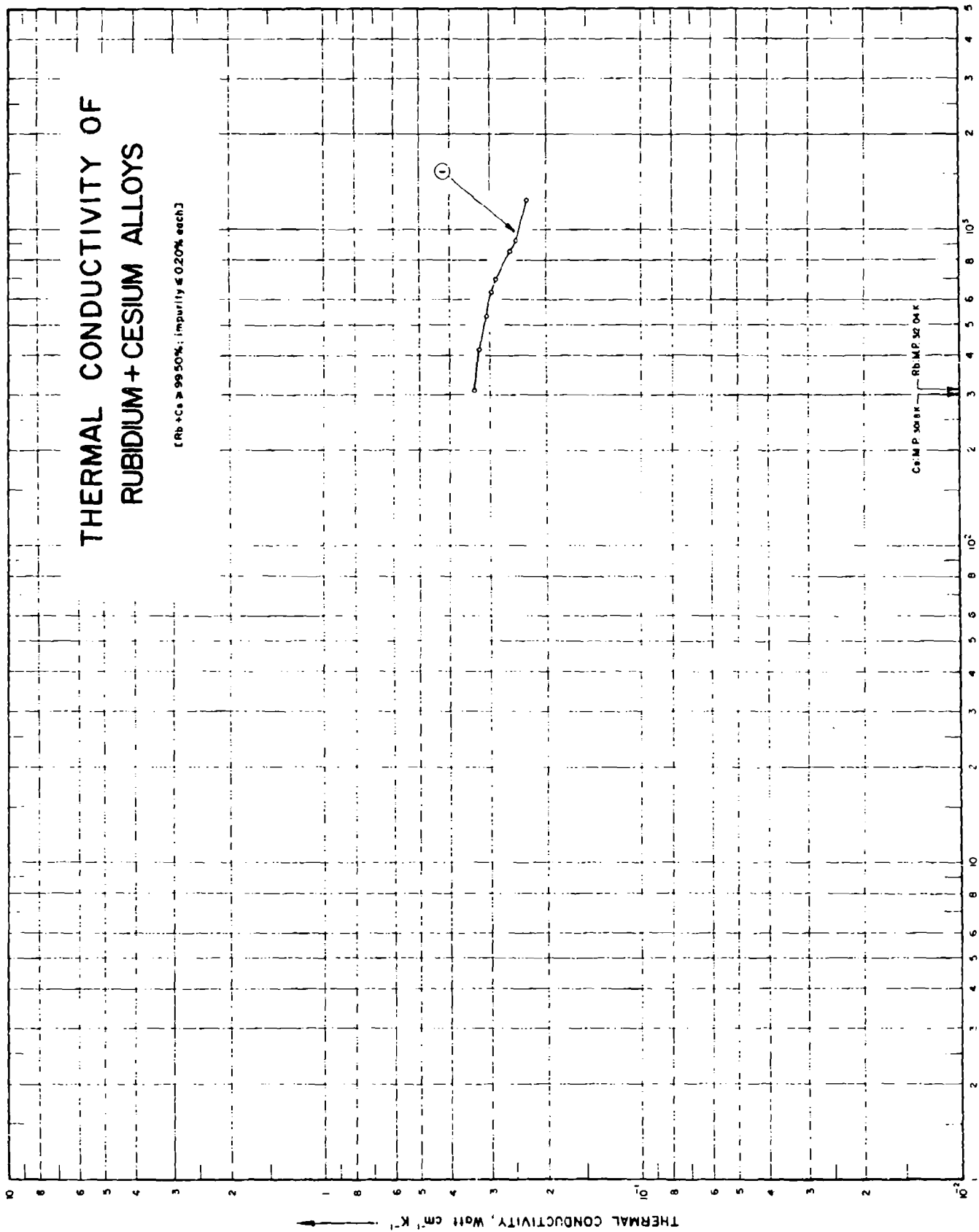
T	k	T	k	T	k	T	k	T	k
<u>CURVE 1</u>									
430.3	0.239	388.2	0.229	588.2	0.265 <sup>6</sup>	793.2	0.277 <sup>6</sup>		
484.6	0.249	390.7	0.235	600.7	0.260 <sup>6</sup>	797.2	0.284 <sup>6</sup>		
589.4	0.260	393.2	0.220	604.7	0.266 <sup>6</sup>	802.7	0.280 <sup>6</sup>		
637.2	0.261	406.2	0.221	605.2	0.255	808.2	0.263 <sup>6</sup>		
718.4	0.262	408.2	0.235	608.2	0.261 <sup>6</sup>	827.7	0.282		
798.7	0.261	418.2	0.221	613.2	0.250	834.2	0.273		
876.7	0.258	419.2	0.235	628.2	0.261 <sup>6</sup>	856.2	0.273		
949.5	0.256	425.2	0.232	630.2	0.250	899.2	0.277		
		430.2	0.225	633.2	0.263 <sup>6</sup>				
<u>CURVE 2</u>									
		430.7	0.244 <sup>6</sup>	641.2	0.262				
		439.7	0.228	644.7	0.252 <sup>6</sup>				
		442.2	0.239 <sup>6</sup>	650.0	0.264 <sup>6</sup>	473.2	0.458		
449.0	0.244	447.2	0.235	658.2	0.256	549.2	0.434		
558.6	0.260	451.2	0.236	664.2	0.261 <sup>6</sup>	609.2	0.417		
627.7	0.266	458.2	0.232	667.7	0.276 <sup>6</sup>	683.2	0.398		
716.3	0.271	458.7	0.244	671.2	0.260 <sup>6</sup>	750.2	0.380		
794.2	0.271	463.7	0.236	681.2	0.258 <sup>6</sup>	836.2	0.362		
		467.7	0.246	681.2	0.282	905.2	0.350		
		469.7	0.232 <sup>6</sup>	685.2	0.282	990.2	0.338		
		474.2	0.232	687.0	0.268	1073	0.328		
486.1	0.131	476.2	0.242	689.2	0.277				
482.6	0.134	479.2	0.232	695.0	0.258				
495.3	0.129	484.2	0.242	703.0	0.274				
561.4	0.134	488.2	0.256	703.7	0.295				
632.0	0.136	493.2	0.245	707.2	0.261 <sup>6</sup>				
761.4	0.138	500.2	0.250	707.2	0.291				
832.0	0.143	504.2	0.235	709.2	0.277				
901.4	0.147	508.2	0.244	713.2	0.284				
960.3	0.145	514.2	0.246	715.0	0.266				
1029.7	0.150	520.7	0.257	723.2	0.274				
		526.2	0.249	733.2	0.281 <sup>6</sup>				
		522.2	0.252	736.2	0.289				
		541.7	0.246	738.2	0.268				
		545.2	0.257	747.2	0.273 <sup>6</sup>				
262.6	0.295	555.2	0.271	748.2	0.253				
264.3	0.291	556.7	0.245	758.2	0.286 <sup>6</sup>				
279.0	0.229	558.2	0.236	759.2	0.272				
279.4	0.231	568.2	0.279	767.7	0.291				
233.9	0.245	568.7	0.274	777.2	0.258				
295.2	0.244	573.2	0.261	780.0	0.292				
295.3	0.244	575.2	0.271	787.7	0.300				
316.1	0.259	578.2	0.271						
		584.2	0.499						
		585.2	0.270						

Not shown on plot

FIG. 178

# THERMAL CONDUCTIVITY OF RUBIDIUM + CESIUM ALLOYS

[Rb + Cs ≥ 99.50%; impurity ≤ 0.20% each]



TEMPERATURE, K

THERMAL CONDUCTIVITY, Wolt cm<sup>-1</sup> K<sup>-1</sup>

Cs: M.P. 301.8 K  
Rb: M.P. 30.04 K

①

## SPECIFICATION TABLE NO. 175 THERMAL CONDUCTIVITY OF RUBIDIUM-CESIUM ALLOYS

80 Cs - 99.50% impurity 0.20% each

For Data Reported in Figure and Table No. 175

Curve No.	Ref. No. Used	Method	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) Rb, Cs	Composition (continued), Specifications and Remarks
1	756		1967	312-1025			Rb 0.32 Cs	0.06 K, 0.05 Na; composition after testing, 0.25 Cs, 0.13 Na, 0.11 K, 0.63 Ca, 0.005 Fe, 0.005 O, 0.002 Nb, 0.001 each of Cr and La; liquid state; contained in a type 347 stainless steel tube; obtained from American Potash and Chemical Corp.; electrical resistivity reported as 11, 25, 13, 16, 13, 83, 14, 67, 22, 84, 22, 53, 23, 35, 25, 96, 31, 62, 37, 06, 42, 30, 46, 59, 46, 61, 52, 45, 58, 01, 59, 37, 64, 61, 71, 48, 72, 49, 81, 06, 91, 29, 99, 05, and 109, 31 $\mu$ ohm cm at 0, 25, 0, 25, 0, 37, 5, 39, 2, 41, 7, 46, 4, 91, 7, 146, 7, 204, 2, 260, 3, 309, 3, 309, 4, 361, 7, 412, 5, 426, 1, 463, 1, 520, 3, 528, 9, 581, 7, 650, 1, 697, 2, and 751, 7 C, respectively; thermal conductivity data calculated from measured electrical resistivity values and the Lorenz number $2.45 \times 10^{-8}$ W ohm K <sup>-2</sup> .



## DATA TABLE NO. 175 THERMAL CONDUCTIVITY OF [RUBIDIUM + CESIUM] ALLOYS

(Rb + Cs  $\pm$  99.50%; impurity  $\leq$  0.20% each)[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup> K<sup>-1</sup>]

T	k
312.4	0.335
419.9	0.325
533.5	0.309
634.9	0.297
699.3	0.289
854.9	0.258
923.3	0.248
1025	0.230

## SPECIFICATION TABLE NO. 179 THERMAL CONDUCTIVITY OF [SELENIUM + BROMINE] ALLOYS

(Se + Br = 99.50%; impurity &lt; 0.20% each)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) Se	Composition (weight percent) Br	Composition (continued), Specifications and Remarks
1	358	L	1957	300.7	< 4.0		99.5	0.5	Vitreous and amorphous.
2	358	L	1957	293.2	< 4.0		99.5	0.5	Hexagonal crystal.
3	358	L	1957	293.2	< 4.0		99.75	0.25	Hexagonal crystal.

## DATA TABLE NO. 179 THERMAL CONDUCTIVITY OF [SELENIUM + BROMINE] ALLOYS

(Se + Br = 99.50%; impurity &lt; 0.20% each)

[ Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup>K<sup>-1</sup> ]

T	k
<u>CURVE 1</u>	
300.7	0.0101
<u>CURVE 2</u>	
293.2	0.0217
<u>CURVE 3</u>	
293.2	0.0220

No graphical presentation

SPECIFICATION TABLE NO. 180 THERMAL CONDUCTIVITY OF [SELENIUM + CADMIUM] ALLOYS

(Se + Cd - 99.50%; impurity - 0.20% each)

Curve No.	Rel. Method No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) Se	Composition (weight percent) Cd	Composition (continued), Specifications and Remarks
1	898	L	1966	293-315	3-5	B-5	99.5	0.5	Amorphous specimen prepared from the melt of 99.99999 pure selenium by rapid cooling in vacuum; doped with cadmium.
2	898	L	1966	293-315	3-5	B-5	99.0	1.0	Similar to the above specimen.

DATA TABLE NO. 180 THERMAL CONDUCTIVITY OF [SELENIUM + CADMIUM] ALLOYS

(Se + Cd - 99.50%; impurity - 0.20% each)

[ Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup>K<sup>-1</sup> ]

T	k	T	k
CURVE 1		CURVE 2	
292.6	0.00437	292.6	0.00456
294.5	0.00444	294.6	0.00466
296.2	0.00456	296.4	0.00474
298.6	0.00470	299.4	0.00493
300.1	0.00484	303.4	0.00498
301.6	0.00490	299.4	0.00509
303.6	0.00505	303.4	0.00520
305.3	0.00516	304.8	0.00531
306.0	0.00525	306.4	0.00566
307.2	0.00628	306.8	0.00493
307.4	0.00558	308.3	0.00523
308.9	0.00594	309.8	0.00536
310.3	0.00616	311.8	0.00608
311.2	0.00637	315.1	0.00686
313.0	0.00667		
315.0	0.00706		

No graphical presentation

SPECIFICATION TABLE NO. 1-1 THERMAL CONDUCTIVITY OF [SELENIUM + CHLORINE] ALLOYS

(Se + Cl) 99.50% impurity = 0.20% each

Curve No.	Ref. Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) Se	Composition (weight percent) Cl	Composition (continued), Specifications and Remarks
1	361	1	1957	294.2	No. 1	99.75	0.25	Amorphous.
2	361	1	1957	294.2	No. 2	99.75	0.25	Crystalline form prepared from vitreous form by heating at 150 C for 40 min.
3	361	1	1957	294.2	No. 3	99.75	0.25	Crystalline form prepared from vitreous form by heating at 200 C for 40 min.
4	362	1	1957	294.7	No. 8	99.75	0.25	Amorphous.
5	362	1	1957	294.7	No. 9	99.50	0.50	Amorphous.
6	362	1	1957	294.7	No. 10	99.0	1.0	Amorphous.

DATA TABLE NO. 1-1 THERMAL CONDUCTIVITY OF [SELENIUM + CHLORINE] ALLOYS

(Se + Cl) 99.50% impurity = 0.20% each

[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup> K<sup>-1</sup>]

T	k	T	k
<u>CURVE 1</u>		<u>CURVE 4</u>	
294.2	0.00854	291.7	0.00845
<u>CURVE 2</u>		<u>CURVE 5</u>	
294.2	0.0254	291.7	0.00858
<u>CURVE 3</u>		<u>CURVE 6</u>	
294.2	0.0254	291.7	0.00987

No graphical presentation

SPECIFICATION TABLE NO. 1-2 THERMAL CONDUCTIVITY OF [SELENIUM + IODINE] ALLOYS

(Se + I 99.50% purity 0.20% each)

Curve No.	Ref. Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) Se	Composition (continued), Specifications and Remarks
1	364	L	1957	299.2		99.557	Amorphous.
2	365	L	1957	299.2		99.557	Crystallized at 214 C from vitreous form.
3	364	L	1957	299.2		99.466	Amorphous.
4	364	L	1957	299.2		99.466	Crystallized at 211 C from vitreous form.
5	364	L	1957	299.2		99.323	Amorphous.
6	364	L	1957	299.2		99.223	Crystallized at 211 C from vitreous form.
7	364	L	1957	299.2		99.077	Amorphous.
8	364	L	1957	299.2		99.077	Crystallized at 214 C from vitreous form.
9	364	L	1957	299.2		98.743	Amorphous.
10	364	L	1957	299.2		98.743	Crystallized at 214 C from vitreous form.
11	364	L	1957	299.2		98.615	Amorphous.
12	364	L	1957	299.2		98.615	Crystallized at 214 C from vitreous form.

DATA TABLE NO. 1-2 THERMAL CONDUCTIVITY OF [SELENIUM + IODINE] ALLOYS

(Se + I 99.50% in purity 0.20% each)

[Temperature, T, K; Thermal Conductivity, k, Watt/cm<sup>2</sup>/K<sup>1</sup>]

T	k	T	k	T	k	T	k	T	k
<u>CURVE 1</u>	<u>CURVE 3</u>	<u>CURVE 5</u>	<u>CURVE 7</u>	<u>CURVE 9</u>	<u>CURVE 11</u>				
299.2	0.00883	299.2	0.00916	299.2	0.00941	299.2	0.00979	299.2	0.0111
								299.2	0.0115
<u>CURVE 2</u>	<u>CURVE 4</u>	<u>CURVE 6</u>	<u>CURVE 8</u>	<u>CURVE 10</u>	<u>CURVE 12</u>				
299.2	0.0229	299.2	0.0234	299.2	0.0244	299.2	0.0253	299.2	0.0290
								299.2	0.0298

No graphical presentation

# THERMAL CONDUCTIVITY OF SELENIUM + THALLIUM ALLOYS

(Se + Tl ≥ 99.50%; impurity ≤ 0.20% each)

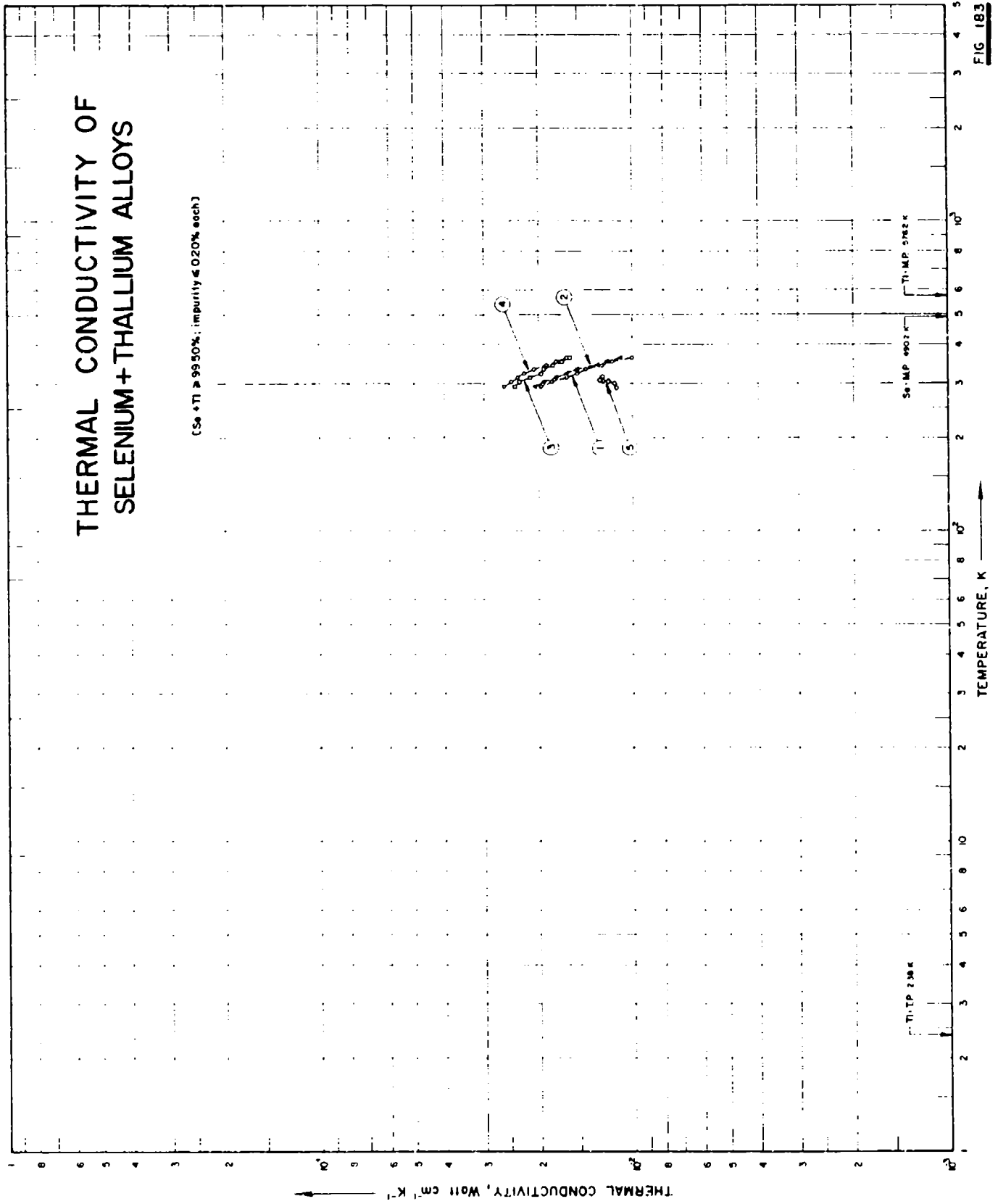


FIG. 183

## SPECIFICATION TABLE NO. 183 THERMAL CONDUCTIVITY OF [SELENIUM + THALLIUM] ALLOYS

(Se + Tl = 99.50%; impurity  $\leq 0.30\%$ )

[For Data Reported in Figure and Table No. 183]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)	Composition (continued), Specifications and Remarks
1	522	L	1961	293-363			99.5      0.5	Polycrystal; prepared by fusing Se (99.996 pure) and Tl <sub>2</sub> Se in 10-4 mm Hg vacuo; specimen 16 mm dia, 8-10 mm thick disk; annealed at 110 and 210 C for 1 hr.
2	522	L	1961	293-363			99.25      0.75	Similar to the above specimen.
3	522	L	1961	293-363			99.0      1.0	Similar to the above specimen.
4	522	L	1961	293-363			98.5      1.5	Similar to the above specimen.
5	898	L	1961	291-317	3-5	B-5	99.0      1.0	Amorphous specimen prepared from the melt of 99.9999 pure selenium by rapid cooling in vacuum.

DATA TABLE NO. 183 THERMAL CONDUCTIVITY OF (SELENIUM + THALLIUM) ALLOYS

(Se + Tl = 99.50% impurity = 0.20%)

(Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup>K<sup>-1</sup>)

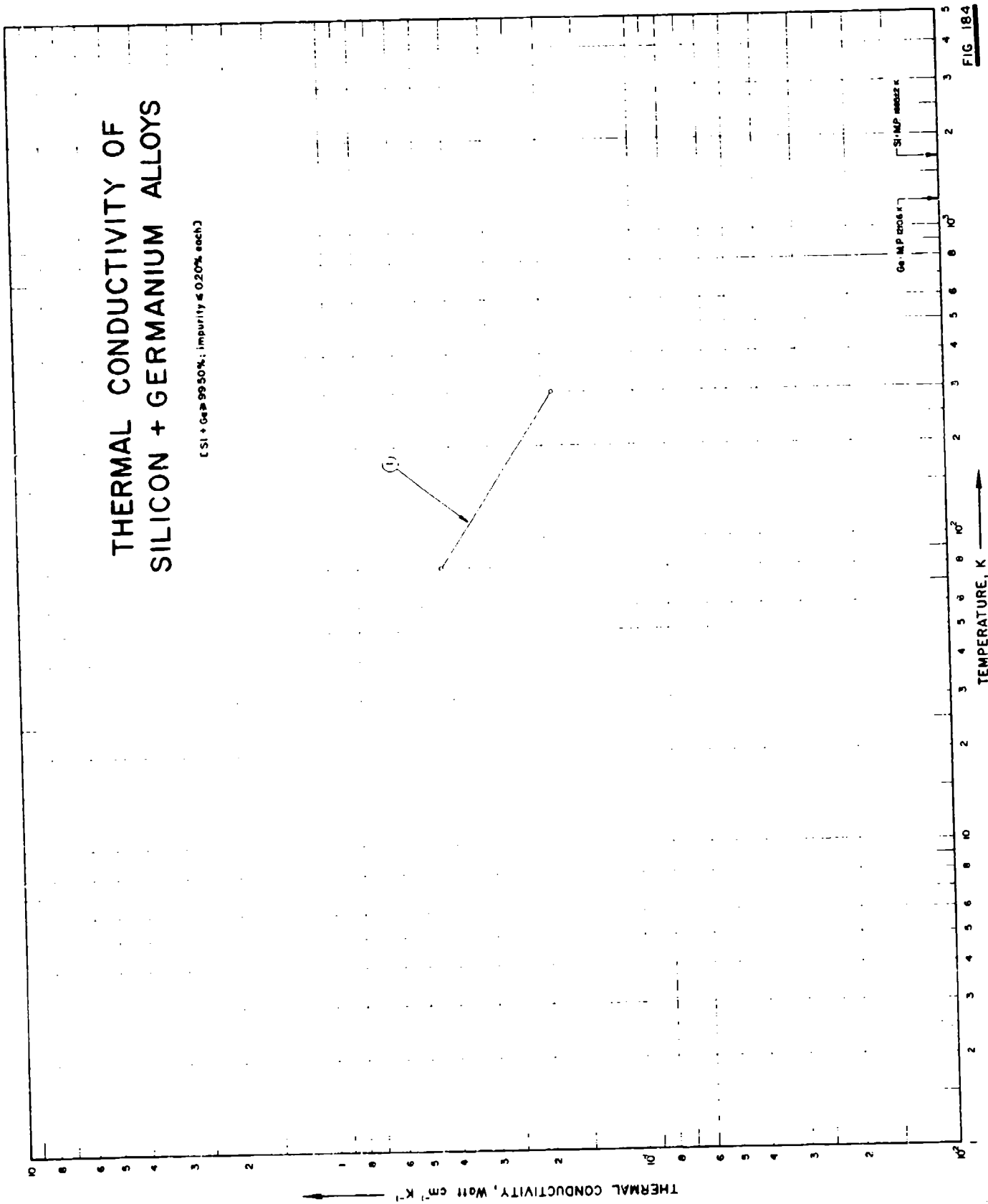
T	k	T	k
<b>CURVE 1</b>			
293.2	0.0197	290.9	0.0113
303.2	0.0182	292.3	0.0113 <sup>a</sup>
313.2	0.0163	294.3	0.0113 <sup>a</sup>
323.2	0.0151	296.9	0.0113 <sup>a</sup>
333.2	0.0142	299.0	0.0115
343.2	0.0126	300.9	0.0116 <sup>a</sup>
353.2	0.0117	302.6	0.0116 <sup>a</sup>
363.2	0.0100	305.3	0.0120
<b>CURVE 2</b>			
293.2	0.0205	306.8	0.0126
303.2	0.0192	307.1	0.0125 <sup>a</sup>
313.2	0.0176	307.8	0.0128 <sup>a</sup>
323.2	0.0163	311.0	0.0128 <sup>a</sup>
333.2	0.0151	313.9	0.0126 <sup>a</sup>
343.2	0.0130	316.9	0.0126
353.2	0.0121		
363.2	0.0109		
<b>CURVE 3</b>			
293.2	0.0238		
303.2	0.0230		
313.2	0.0213		
323.2	0.0197		
333.2	0.0192		
343.2	0.0180		
353.2	0.0169		
363.2	0.0159		
<b>CURVE 4</b>			
293.2	0.0250		
303.2	0.0245		
313.2	0.0234		
323.2	0.0222		
333.2	0.0207		
343.2	0.0195		
353.2	0.0176		
363.2	0.0163		

Not shown on plot



# THERMAL CONDUCTIVITY OF SILICON + GERMANIUM ALLOYS

(SI + Ga 99.50%; impurity  $\leq$  0.20% each)



## SPECIFICATION TABLE NO. 154 THERMAL CONDUCTIVITY OF [SILICON + GERMANIUM] ALLOYS

(Si + Ge + 99.50%; impurity = 0.20% each)

(For Data Reported in Figure and Table No. 154)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) Si	Gr.	Composition (continued), Specifications and Remarks
1	263	L	1954	80-300			77.69	22.31	n-type; calculated composition; 5b-doped; lattice constant 5.448; specimen 5 x 5 x 15 mm, grown by zone leveling technique; grain size < 1 cm dia.; measured in vacuo of ~10 <sup>-6</sup> mm Hg; electrical resistivity 18 ohm cm at 300 K.

## DATA TABLE NO. 14 THERMAL CONDUCTIVITY OF [SILICON + GERMANIUM] ALLOYS

(Si + Ge .99.50%, impurity .0.20% each)

[Temperature, T, K; Thermal Conductivity,  $k$ , Watt  $\text{cm}^{-1} \text{K}^{-1}$ ]

T	k
80	0.431
300	0.185

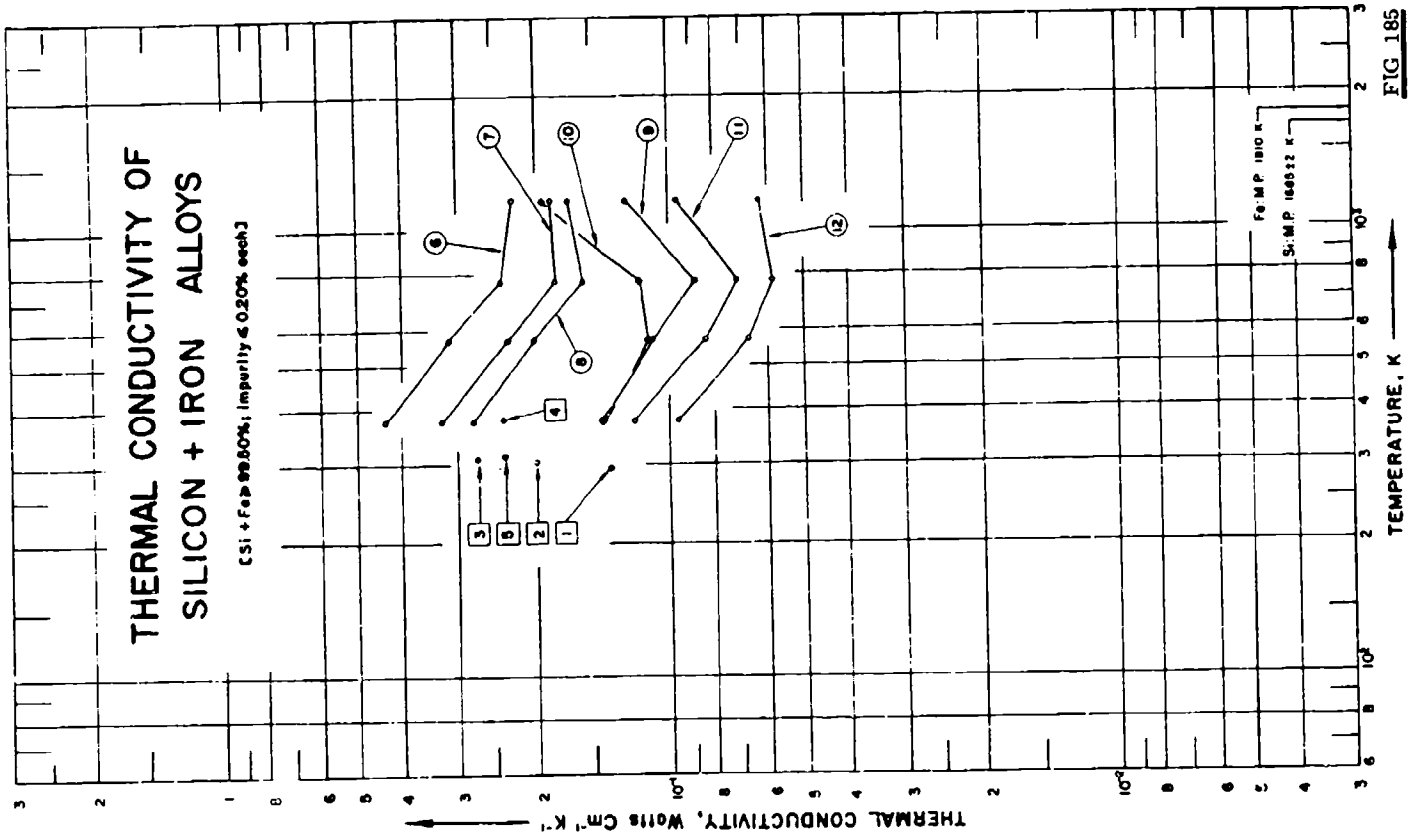


FIG 185

## SPECIFICATION TABLE NO. 185 THERMAL CONDUCTIVITY OF (SILICON - IRON) ALLOYS

(Si - Fe: 99.50%; impurity - 0.20%; each)

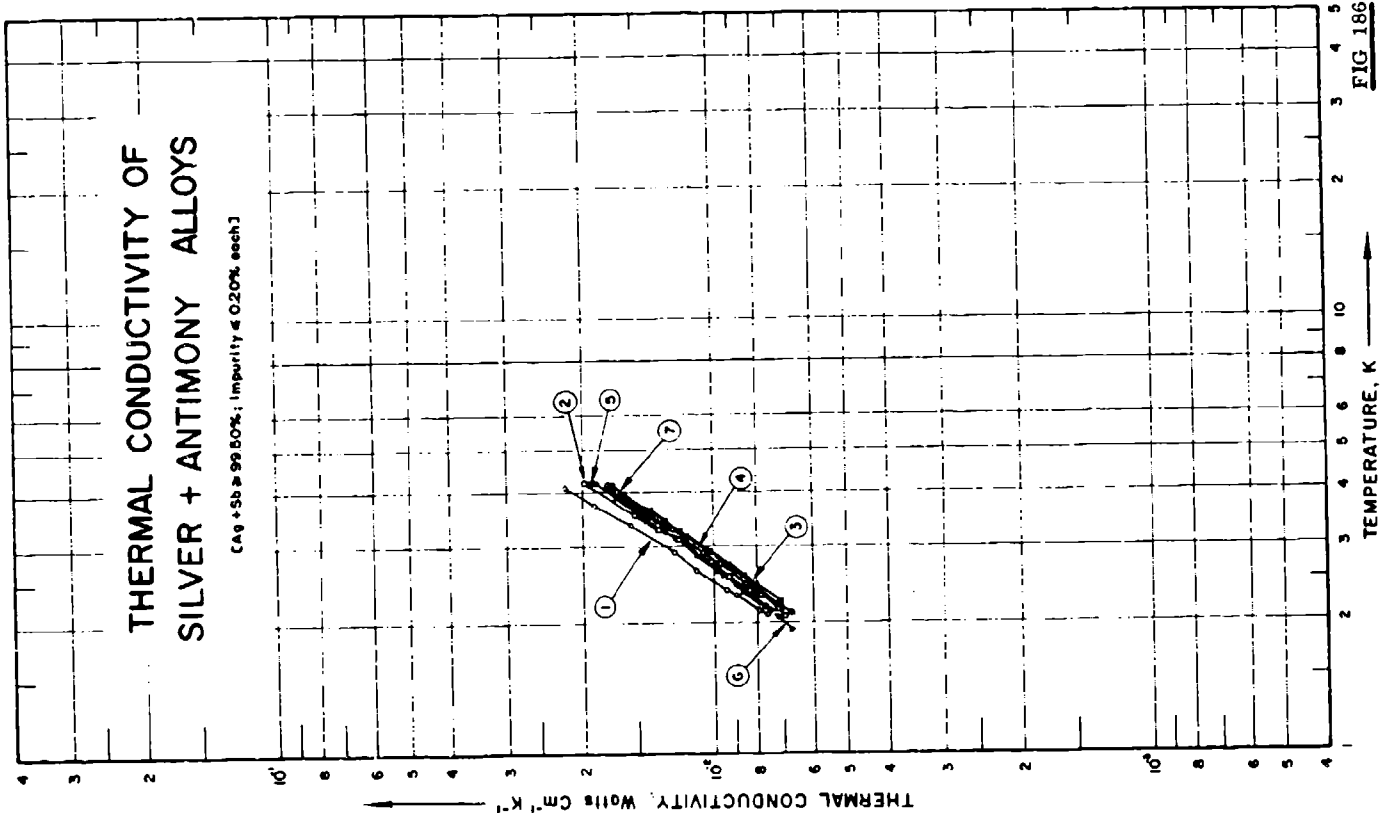
[For Data Reported in Figure and Table No. 185]

Curve No.	Ref. No. Used	Method	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) Si	Composition (weight percent) Fe	Composition (continued), Specifications and Remarks
1	204	L	1937	295-5		Russian ferrosilicon, 1	76.8	23.2	Porous specimen; cross-section 20 x 20 mm.
2	204	L	1937	304.7		Russian ferrosilicon, 2	71.7	28.3	Specimen cross section 20 x 20 mm.
3	204	L	1937	309.2		Russian ferrosilicon, 3	73.15	26.85	Specimen cross section 26 x 20 mm; heat flow perpendicular to thickness.
4	204	L	1937	381.1		Russian ferrosilicon, 4	78.8	21.2	Specimen cross section 20 x 20 mm.
5	204	L	1937	315.6		Russian ferrosilicon, 5	76.8	23.2	Specimen cross section 20 x 20 mm.
6	356	R	1956	373-1173	±7		90.0	10.0	No details reported.
7	356	R	1956	373-1173	±7		80.0	20.0	No details reported.
8	356	R	1956	373-1173	±7		75.0	25.0	No details reported.
9	356	R	1956	373-1173	±7		62.0	38.0	No details reported.
10	356	R	1956	373-1173	±7		59.0	41.0	No details reported.
11	356	R	1956	373-1173	±7		55.5	44.5	No details reported.
12	356	R	1956	373-1173	±7		52.5	47.5	No details reported.

## DATA TABLE NO. 143 THERMAL CONDUCTIVITY OF [SILICON + IRON] ALLOYS

(Si + Fe  $\pm$  99.50%, Impurity  $\leq$  0.10% each)[Temperature, T, K. Thermal Conductivity, k, Watts cm<sup>-1</sup>K<sup>-1</sup>]

T	k	T	k
<u>CURVE 1</u>			
295.5	0.136	773.2	0.0879
		1173.2	0.126
<u>CURVE 2</u>			
304.7	0.200	<u>CURVE 10</u>	
<u>CURVE 3</u>			
		373.2	0.142
		573.2	0.113
		773.2	0.117
309.2	0.271	1173.2	0.192
<u>CURVE 4</u>			
<u>CURVE 11</u>			
381.1	0.236	373.2	0.121
		573.2	0.0837
<u>CURVE 5</u>			
		773.2	0.0711
315.6	0.236	1173.2	0.0962
<u>CURVE 6</u>			
<u>CURVE 12</u>			
373.2	0.431	373.2	0.0962
573.2	0.314	573.2	0.0669
773.2	0.238	773.2	0.0506
1173.2	0.224	1173.2	0.0629
<u>CURVE 7</u>			
373.2	0.326		
573.2	0.230		
773.2	0.180		
1173.2	0.184		
<u>CURVE 8</u>			
373.2	0.276		
573.2	0.201		
773.2	0.155		
1173.2	0.167		
<u>CURVE 9</u>			
373.2	0.142		
573.2	0.109		



SPECIFICATION TABLE NO. 186 THERMAL CONDUCTIVITY OF SILVER - ANTIMONY ALLOYS  
(Ag - Sb - 59.50% impurity - 0.20% each)

† For Data Reported in Figure and Table No. 186†

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)	Ag	Sb	Composition (continued), Specifications and Remarks
1	523	L	1959	2.1-4.1		1	97.94	2.06		Impurities: 0.1 cast, machined; specimen 0.100 X 0.150 X 3 mm.; annealed for 20 hrs. at 50-100 C below liquidus; grain size a few tenths of a millimeter $\rho_{\perp}$ 12.4 $\mu\Omega\text{m}$ cm.
2	525	L	1959	2.1-4.2		2	96.94	3.06		Similar to the above specimen except $\rho_{\perp}$ 17.4 $\mu\Omega\text{m}$ cm.
3	524	L	1959	2.2-4.1		3	95.53	4.47		Similar to the above specimen except $\rho_{\perp}$ 24.7 $\mu\Omega\text{m}$ cm.
4	523	L	1959	2.4-4.1		4	95.53	4.47		Similar to the above specimen except $\rho_{\perp}$ 28.3 $\mu\Omega\text{m}$ cm.
5	523	L	1959	2.1-4.2		5	94.72	5.28		Similar to the above specimen except $\rho_{\perp}$ 33.2 $\mu\Omega\text{m}$ cm.
6	523	L	1959	1.9-4.2		6	93.64	6.36		Similar to the above specimen except $\rho_{\perp}$ 39.5 $\mu\Omega\text{m}$ cm.
7	523	L	1959	2.1-4.1		7				Prepared by melting cutting chips from specimens 4 and 6 (above); $\rho_{\perp}$ 39.6 $\mu\Omega\text{m}$ cm.

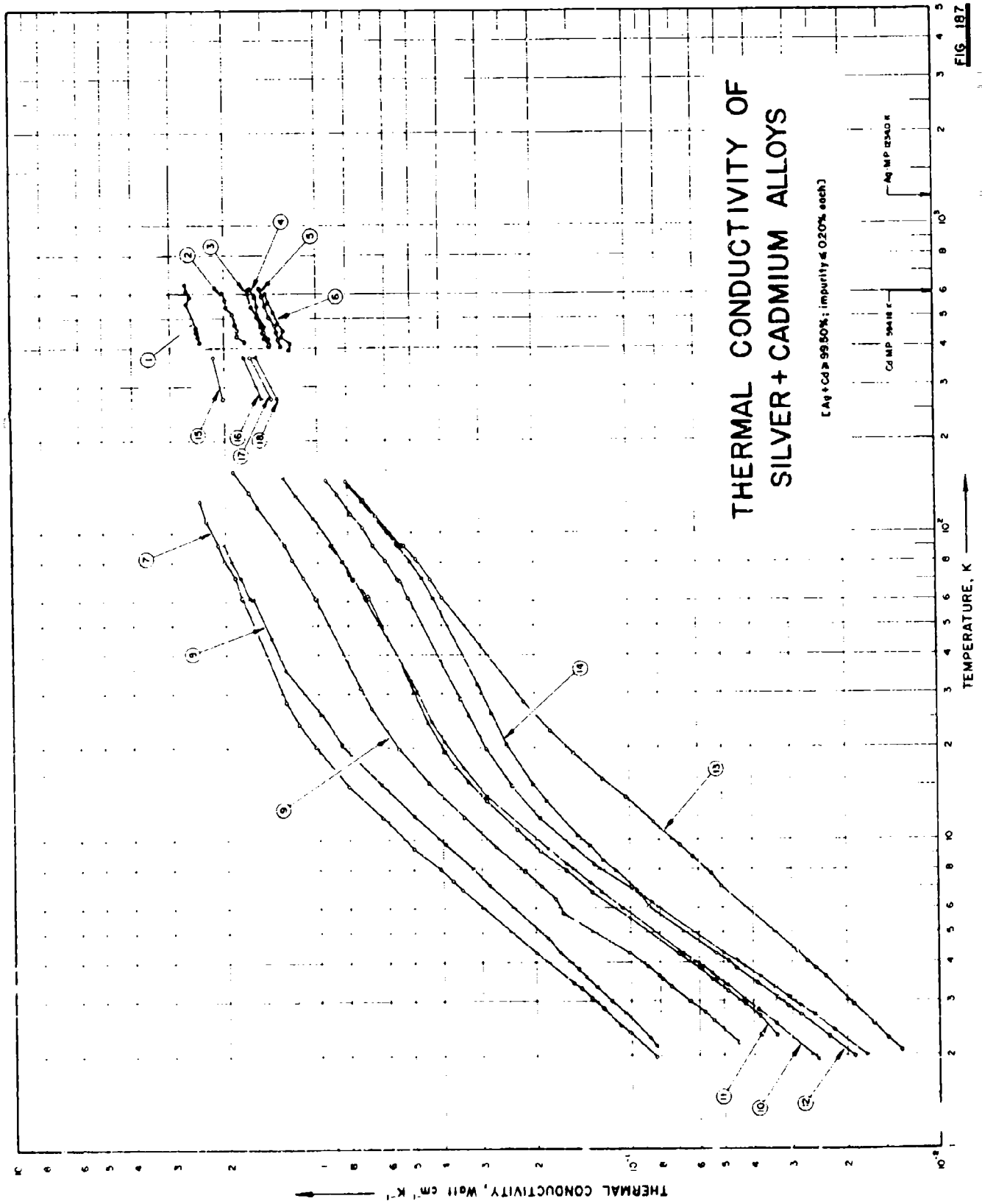


## DATA TABLE NO. 1-6 THERMAL CONDUCTIVITY OF [SILVER + ANTIMONY] ALLOYS

(Ag + Sb : 99.50%; impurity &lt; 0.50% each)

[Temperature, T, K; Thermal Conductivity, k, Watts cm<sup>-1</sup> K<sup>-1</sup>]

T	k	T	k
<u>CURVE 1.</u>			
2.08	0.00734	3.50	0.0139
2.13	0.00790	3.70	0.0149
2.30	0.00890	3.92	0.0162
2.37	0.00936	4.13	0.0175
2.63	0.0110	<u>CURVE 5.</u>	
2.90	0.0124	2.13	0.00744
3.35	0.0154	2.17	0.00765
3.72	0.0189	2.29	0.00814
4.10	0.0218	2.41	0.00886
		2.63	0.00982
		3.00	0.0115
		3.32	0.0132
		3.68	0.0151
		3.88	0.0161
		4.05	0.0174
		4.20	0.0182
<u>CURVE 2.</u>			
2.07	0.00690	<u>CURVE 6.</u>	
2.15	0.00718	1.93	0.00563
2.31	0.00784	2.05	0.00711
2.37	0.00832	2.35	0.00851
2.55	0.00924	2.46	0.00897
2.75	0.0101	2.57	0.00950
3.10	0.0121	2.85	0.0109
3.27	0.0134	3.13	0.0123
3.55	0.0151	3.45	0.0142
4.20	0.0198	3.88	0.0167
		4.20	0.0186
<u>CURVE 3.</u>			
2.23	0.00704	<u>CURVE 7.</u>	
2.41	0.00800	2.10	0.00666
2.55	0.00857	2.17	0.00702
2.70	0.00929	2.39	0.00801
2.90	0.0102	2.68	0.00920
3.15	0.0116	3.23	0.0120
3.42	0.0130	3.62	0.0141
3.62	0.0139	4.13	0.0171
3.82	0.0155		
4.10	0.0173		
<u>CURVE 4.</u>			
2.38	0.00816		
2.45	0.00848		
2.71	0.00967		
2.89	0.0106		
3.27	0.0125		



SPICHERATION TABLE NO. 187 THERMAL CONDUCTIVITY OF SILVER-CADMIUM ALLOYS

(Ag-Cd 99.50% impurity 0.20% each)

For Data Reported in Figure and Table No. 187

Curve No.	Rel. Method Used	Year	Temp. Range, K.	Reported Error, %	Name and Specimen Designation	Composition (weight percent) Ag	Composition (weight percent) Cd	Composition (continued), Specifications and Remarks
1	C	1951	412-648		5A	94.65	5.35	Cast and machined; cylindrical specimen slight porosity on ends; Armco iron used as standard.
2	C	1951	420-608		12B	87.1	12.9	Similar to the above with 25 internal voids 1/8 to 1/32 in. in dia.
3	C	1951	412-648		20A	77.8	22.0	Cast, cold-rolled (67% reduction), machined; Armco iron used as standard.
4	C	1951	408-678		25A	75.2	24.5	Similar to the above specimen.
5	C	1951	405-62		30A	68.1	31.8	Similar to the above specimen.
6	C	1951	390-604		35A	61.35	35.65	Cast and machined; cylindrical specimen; small voids visible throughout; Armco iron used as standard.
7	L	1956	2-0-127			98.04	1.96	Supplied by Johnson, Matthey and Co., Ltd.; annealed at 530 C. $\rho_0 = 0.62 \mu\text{ohm cm}$ ; electrical resistivity at 293 K; $\rho(293) = 2.26 \mu\text{ohm cm}$ .
8	L	1956	2-1-91			98.04	1.96	The above specimen before annealing.
9	L	1956	2-1-159			95.09	4.91	Supplied by Johnson, Matthey and Co., Ltd.; annealed at 530 C; $\rho_0 = 1.38 \mu\text{ohm cm}$ ; $\rho(293) = 3.13 \mu\text{ohm cm}$ .
10	L	1956	1-9-177			90.49	9.51	Supplied by Johnson, Matthey and Co., Ltd.; annealed at 500 C.
11	L	1956	2-3-51			90.49	9.51	Supplied by Johnson, Matthey and Co., Ltd.; annealed at 610 C. $\rho_0 = 2.26 \mu\text{ohm cm}$ ; $\rho(293) = 4.09 \mu\text{ohm cm}$ .
12	L	1956	2-0-149			80.79	19.21	Supplied by Johnson, Matthey and Co., Ltd.; annealed at 500 C; $\rho_0 = 3.30 \mu\text{ohm cm}$ ; $\rho(293) = 5.82 \mu\text{ohm cm}$ .
13	L	1956	2-0-148			80.79	19.21	The above specimen before annealing.
14	L	1956	2-0-145			70.03	29.97	Supplied by Johnson, Matthey and Co., Ltd.; annealed at 500 C; $\rho_0 = 3.67 \mu\text{ohm cm}$ ; $\rho(293) = 6.22 \mu\text{ohm cm}$ .
15	T	1919	273-373			95.84	4.16	Calculated composition; wire, 1 mm dia; rolled and drawn; annealed for 1/2 hr close to melting point; electrical conductivity, $29.3$ and $24.2 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 0 and 100 C, respectively.
16	T	1919	273-373			90.86	9.14	Similar to the above specimen except electrical conductivity $21.0$ and $19.2 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 0 and 100 C, respectively.
17	T	1919	273-373			87.6	12.34	Similar to the above specimen except electrical conductivity $19.2$ and $18.0 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 0 and 100 C, respectively.
18	T	1919	273-373			83.17	16.87	Similar to the above specimen except electrical conductivity $17.1$ and $14.9 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 0 and 100 C, respectively.



SPECIFICATION TABLE NO. 188 THERMAL CONDUCTIVITY OF (SILVER + COPPER) ALLOYS

Ag + Cu 99.50% ; impurity 0.20% each

Curve No.	Year	Year	Range, K	Reported Value, k	Name and Specimen Designation	Composition (weight percent) Ag	Composition (weight percent) Cu	Composition (continues), Specifications and Remarks
1	1925	1	335.2	335.2		95.0	5.0	Impurities 0.03.
2	1925	1	335.2	335.2		85.0	15.0	Impurities 0.03.
3	1925	1	335.2	335.2		75.0	25.0	Impurities 0.03.
4	1925	1	335.2	335.2		65.0	35.0	Impurities 0.03.
5	1925	1	335.2	335.2		60.0	40.0	Impurities 0.03.
6	1925	1	335.2	335.2		55.0	45.0	Impurities 0.03.
7	1925	1	335.2	335.2		50.0	50.0	Impurities 0.03.
8	1925	1	335.2	335.2		50.0	50.0	Impurities 0.03.

DATA TABLE NO. 188 THERMAL CONDUCTIVITY OF (SILVER + COPPER) ALLOYS

Ag + Cu 99.50% ; impurity 0.20% each

[ Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup> K<sup>-1</sup> ]

T	k	T	k	T	k
<u>CURVE 1</u>		<u>CURVE 4</u>		<u>CURVE 7</u>	
335.2	3.519	335.2	3.192	335.2	3.109
<u>CURVE 2</u>		<u>CURVE 5</u>		<u>CURVE 8</u>	
335.2	3.431	335.2	3.113	335.2	3.126
<u>CURVE 3</u>		<u>CURVE 6</u>			
335.2	3.301	335.2	3.134		

No graphical presentation

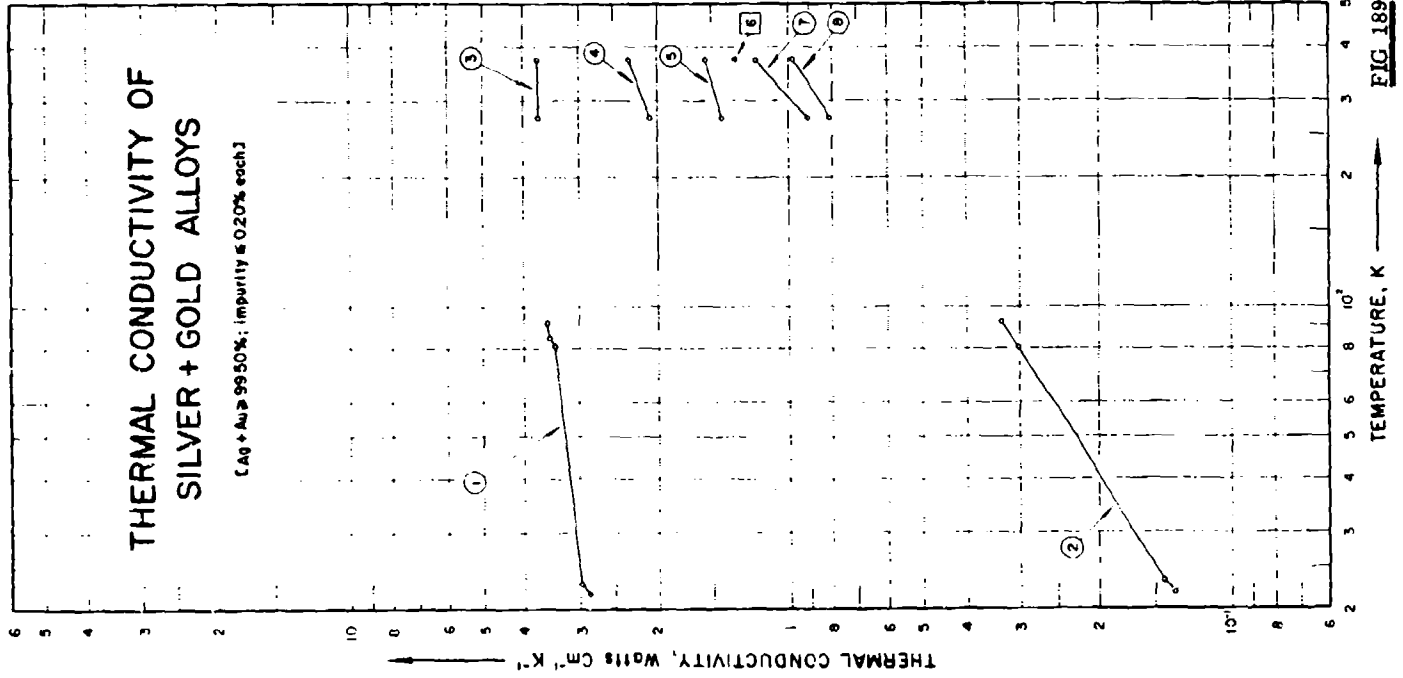


FIG 189

SPECIFICATION TABLE NO. 189 THERMAL CONDUCTIVITY OF [SILVER - GOLD] ALLOYS

(Ag + Au 99.96% - 9.26% each)

(For Data Reported in Figure and Table No. 189)

Curve No.	Rct. Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)		Composition (continued), Specifications and Remarks
						Ag	Au	
1	58	1	1931	±2-0.2		99.33	0.67	Calculated composition, wire specimen: electrical resistivity <del>0.473, and 1.63 ohm cm at 0°C, -190°C, and -251°C, respectively.</del> <b>0.473, and 1.63 ohm cm at 0°C, -190°C, and -251°C, respectively.</b>
2	58	1	1931	±2-0.2		92.22	7.78	Calculated composition; single et (SCL); electrical resistivity <del>7.25, and 8.87 ohm cm at 0°C, -190°C, and -251°C, respectively.</del> <b>7.25, and 8.87 ohm cm at 0°C, -190°C, and -251°C, respectively.</b>
3	246	T	1919	±73-373		95.88	4.12	Calculated composition; wire; 1 mm dia. Polished and drawn; annealed close to melting point for 9.5 hr. electrical resistivity <del>0.190, and -251°C, respectively.</del> <b>0.190, and -251°C, respectively.</b>
4	246	T	1919	±73-373		91.22	8.78	Similar to the above specimen, electrical conductivity 25.3 and $24.2 \times 10^8 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 0 and 100°C, respectively.
5	246	T	1919	±73-373		80.74	19.26	Similar to the above specimen except electrical conductivity 19.5 and $16.9 \times 10^8 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 0 and 100°C, respectively.
6	246	T	1919	±73-2		76.34	23.66	Similar to the above specimen except electrical conductivity 14.7 and $13.5 \times 10^8 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 0 and 100°C, respectively.
7	246	T	1919	±73-373		68.63	31.37	Similar to the above specimen except electrical conductivity 12.5 to $11.5 \times 10^8 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 0 and 100°C, respectively.
8	246	T	1919	±73-373		55.81	44.16	Similar to the above specimen except electrical conductivity 10.3 and $9.7 \times 10^8 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 0 and 100°C, respectively.

DATA TABLE NO. 1-9 THERMAL CONDUCTIVITY OF SILVER-GOLD ALLOYS

Ag-Au 50-50% (molar) - 0.26" each

Temperature: 1 K Thermal Conductivity:  $k$  Watt cm<sup>-1</sup> K<sup>-1</sup>

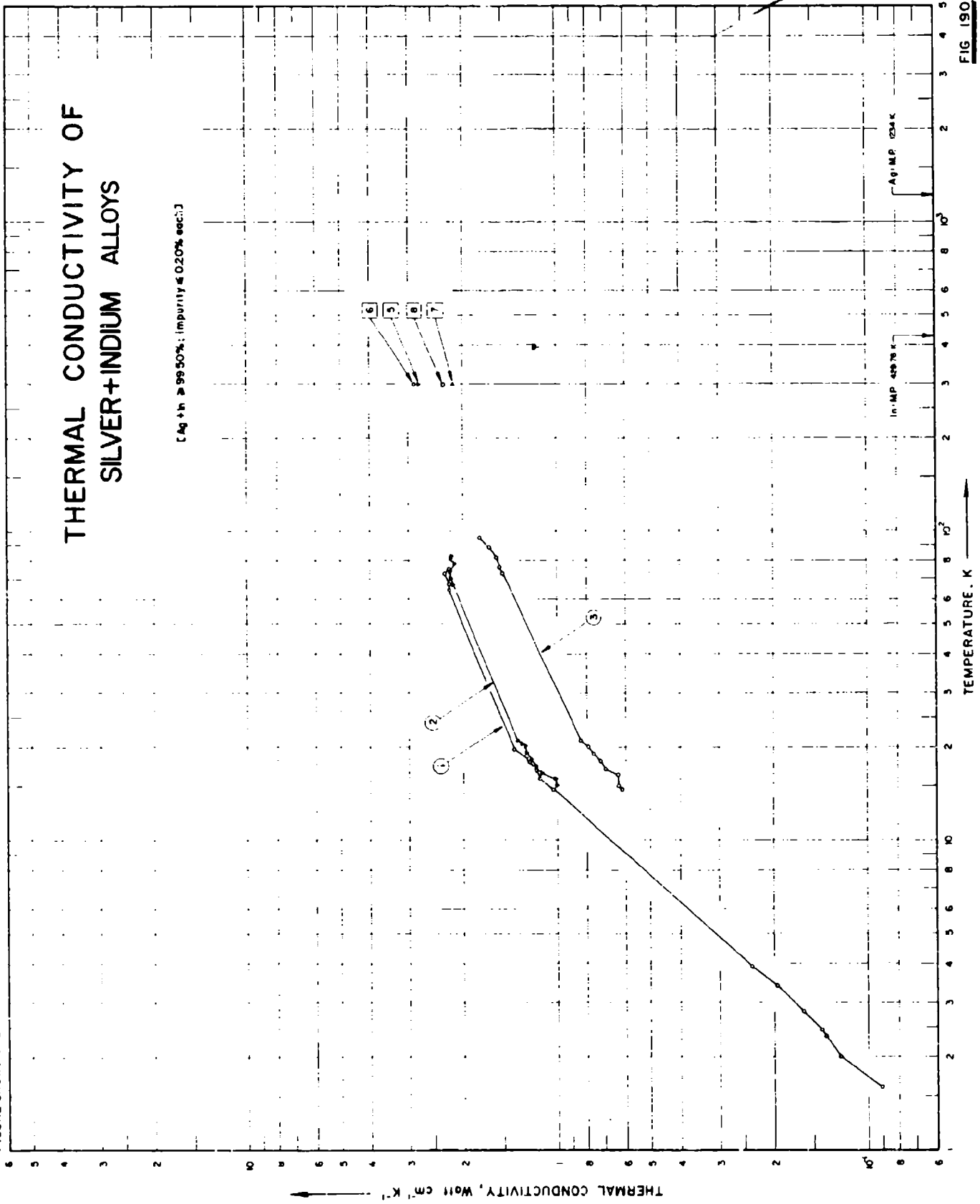
T	k
<u>CURVE 1</u>	
23.1	2.87
24.8	3.01
30.7	3.45
34.3	3.53
31.7	3.56
<u>CURVE 2</u>	
21.8	0.136
23.2	0.144
30.2	0.334
30.2	0.302
32.0	0.334
<u>CURVE 3</u>	
373.2	3.71
373.2	3.72
<u>CURVE 4</u>	
373.2	2.08
373.2	2.30
<u>CURVE 5</u>	
373.2	1.42
373.2	1.55
<u>CURVE 6</u>	
373.2	1.32
<u>CURVE 7</u>	
373.2	0.91
373.2	1.19
<u>CURVE 8</u>	
373.2	0.81
373.2	0.98



FIGURE SHOWS ONLY 7 OF THE CURVES REPORTED IN TABLE

# THERMAL CONDUCTIVITY OF SILVER+INDIUM ALLOYS

[Ag + In  $\geq$  99.50%; impurity  $\leq$  0.20% each]



SPECIFICATION TABLE NO. 190 THERMAL CONDUCTIVITY OF SILVER-INDIUM ALLOYS

(Ag-In 49.56% impurity - 0.29% each)

For Data Reported in Figure and Table No. 190

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) Ag	In	Composition (continued), Specifications and Remarks
1	22	L	1959	1.9-76			99.75	0.25	Calculated composition: polycrystalline specimen ~6 cm long, $\rho_0$ 0.45 $\mu\text{hm cm}$ (same as the succeeding specimen, Ag-In 1)
2	51	L	1956	14-87		Ag-In 1	99.75	0.25	Calculated composition, prepared from Ag (pure, supplied by Nordiska Affäret) and In (0.00065 impurity), annealed in vacuo at 720 C; electrical resistivity ranging from 0.425 to 1.89 $\mu\text{hm cm}$ at 14 to 273 K, respectively.
3	51	L	1956	14-95		Ag-In 2	99.45	0.55	Similar to the above specimen except electrical resistivity ranging from 0.895 to 2.395 $\mu\text{hm cm}$ at 14 to 273 K, respectively.
4	504	C	1962	298.2			99.0	1.0	Cast; copper used as comparative material.
5	504	C	1962	298.2			98.0	2.0	Cast; copper used as comparative material.
6	504	C	1962	298.2			97.0	3.0	Cast; copper used as comparative material.
7	504	C	1962	298.2			96.0	4.0	Cast; copper used as comparative material.
8	504	C	1962	298.2			95.0	5.0	Cast; copper used as comparative material.

## DATA TABLE NO. 190 THERMAL CONDUCTIVITY OF [SILVER + INDIUM] ALLOYS

(Ag + In : 99.50%, impurity : 0.20%, each)

(Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup> K<sup>-1</sup>)

T	k	T	k
<u>CURVE 1</u>			
1.59	0.091	18.0	0.722
2.00	0.133	19.0	0.760
2.33	0.157	20.0	0.796
2.45	0.142	21.0	0.844
2.80	0.162	75.0	1.509
3.40	0.196	76.8	1.560
3.93	0.236	82.0	1.570
14.69	1.035	88.2	1.650
15.89	1.114	94.8	1.770
16.87	1.158	<u>CURVE 4*</u>	
17.99	1.228	298.2	2.74
18.93	1.297	<u>CURVE 5</u>	
19.73	1.385	298.2	2.75
19.88	1.395	<u>CURVE 6</u>	
64.46	2.242	298.2	2.85
67.27	2.232	<u>CURVE 7</u>	
72.72	2.304	298.2	2.15
75.47	2.236	<u>CURVE 8</u>	
<u>CURVE 2</u>			
15.0	1.00	298.2	2.31
15.7	1.01	<u>CURVE 9</u>	
16.5	1.12	298.2	2.15
17.4	1.17	<u>CURVE 10</u>	
18.3	1.22	298.2	2.31
19.1	1.25	<u>CURVE 11</u>	
20.2	1.26	298.2	2.31
20.5	1.30	<u>CURVE 12</u>	
20.9	1.35	298.2	2.31
67.0	2.17	<u>CURVE 13</u>	
70.0	2.20	298.2	2.31
74.4	2.20	<u>CURVE 14</u>	
78.0	2.12	298.2	2.31
81.2	2.20	<u>CURVE 15</u>	
83.2	2.19	298.2	2.31
<u>CURVE 3</u>			
14.7	0.619	<u>CURVE 16</u>	
15.0	0.632	298.2	2.31
16.3	0.632	<u>CURVE 17</u>	
17.0	0.696	298.2	2.31

\* Not shown on plot

# THERMAL CONDUCTIVITY OF SILVER + LEAD ALLOYS

[ Ag + Pb ≥ 99.50%, impurity ≤ 0.20% each ]

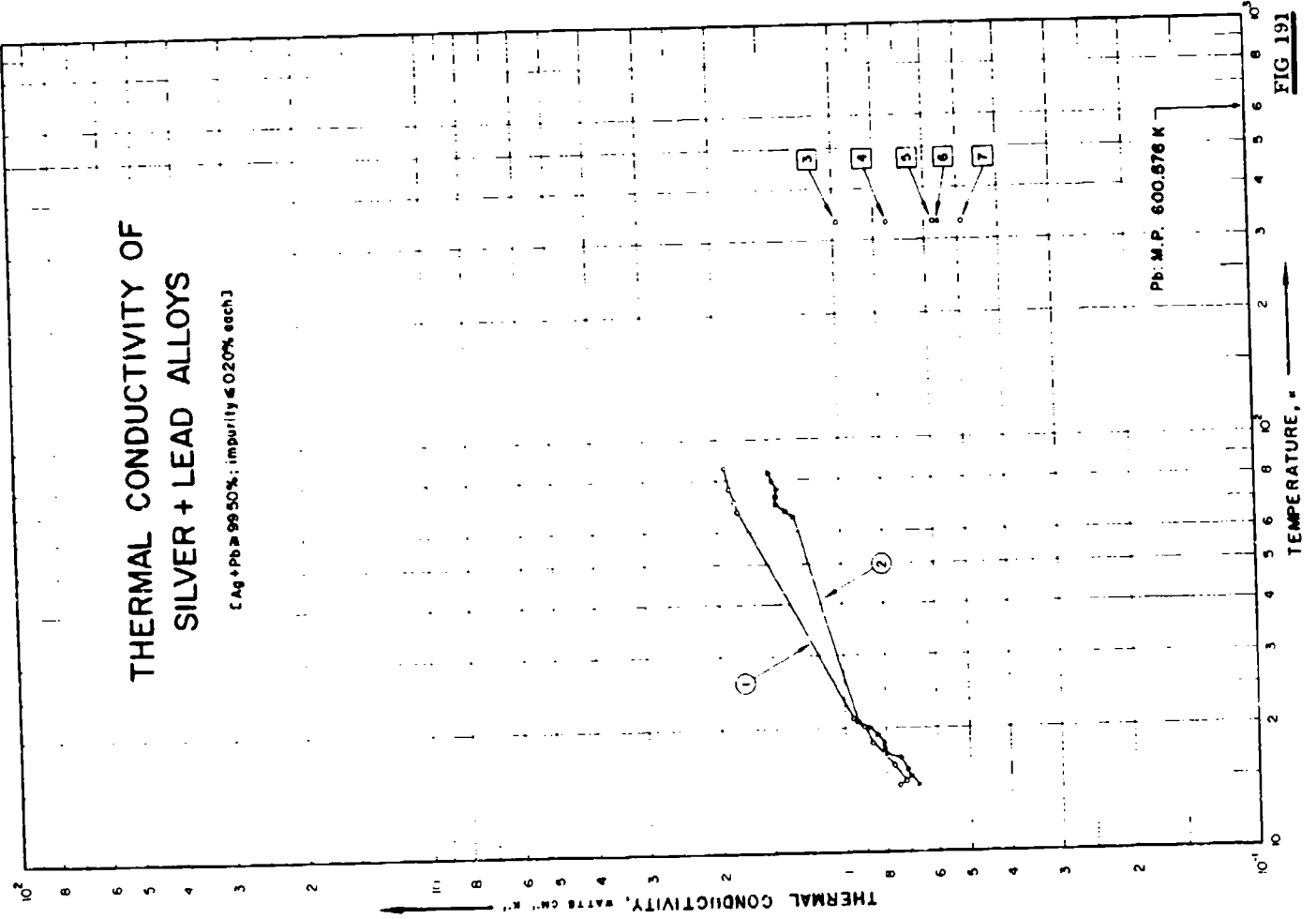


FIG 191

## SPECIFICATION TABLE NO. 191 THERMAL CONDUCTIVITY OF SILVER-LEAD ALLOYS

(Ag-Pb 99.50% impurity 0.20% each)

For Data Reported in Figure and Table No. 191

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) Ag	Pb	Composition (continued), Specifications and Remarks
1	51	L	1956	15-80		Ag-Pb 1	99.75	0.25	Calculated composition, prepared from Ag (pure, supplied by Nordiska Alfvieriet) and Pb (0.0005 impurity), annealed in vacuo at 720 C., electrical resistivity ranging from 0.637 to 2.07 $\mu\text{ohm cm}$ at 14 to 273 K., respectively.
2	51	L	1956	15-83		Ag-Pb 2	99.62	0.38	Similar to the above specimen but quenched in water; electrical resistivity 0.983 to 2.47 $\mu\text{ohm cm}$ at 14 to 273 K., respectively.
3	230	L	1925	333			99	10	Prepared by fusing Ag (99.9 pure) and Pb (0.03 impurities, supplied by Baker); specimen ~ 5.5 cm long, 0.3 $\text{cm}^2$ cross-sectional area; electrical conductivity $12.17 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 25 C.
4	230	L	1925	333			80	20	Similar to the above specimen except electrical conductivity $9.43 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 25 C.
5	230	L	1925	333			70	30	Similar to the above specimen except electrical conductivity $7.14 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 25 C.
6	230	L	1925	333			60	40	Similar to the above specimen except electrical conductivity $6.90 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 25 C.
7	230	L	1925	333			50	50	Similar to the above specimen except electrical conductivity $6.15 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 25 C.

DATA TABLE NO. 191 THERMAL CONDUCTIVITY OF (SILVER + LEAD) ALLOYS

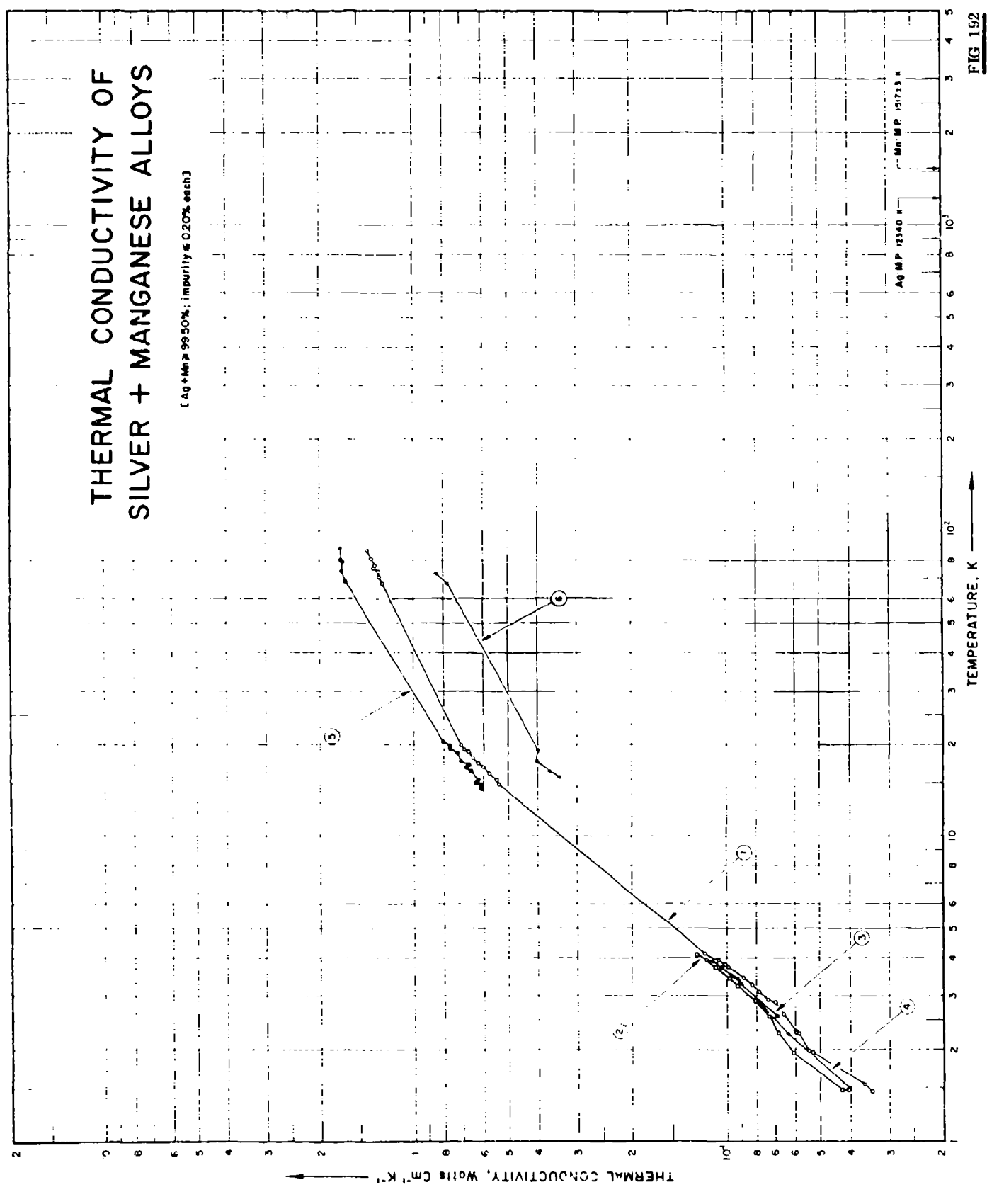
(Ag - Pb - 99.50%, impurity 0.20% each)

[Temperature: T, K. Thermal Conductivity: k, Watt cm<sup>-1</sup> K<sup>-1</sup>]

T	k	T	k
<u>CURVE 1</u>			
14.7	0.740	<u>CURVE 5</u>	
14.9	0.724	333	0.577
16.3	0.763	<u>CURVE 6</u>	
18.5	0.854	333	0.561
20.3	0.900	<u>CURVE 7</u>	
21.2	0.936	333	0.490
67.6	1.790	<u>CURVE 2</u>	
71.0	1.820	14.6	0.665
76.4	1.860	15.3	0.696
86.4	1.926	15.6	0.704
		15.9	0.710
		16.9	0.732
		17.4	0.738
		18.0	0.805
		18.5	0.810
		19.3	0.840
		20.1	0.876
		20.7	0.932
		65.2	1.320
		67.6	1.370
		69.7	1.430
		72.8	1.440
		76.0	1.430
		79.5	1.470
		81.0	1.490
		83.20	1.490
		<u>CURVE 3</u>	
		333	0.987
		<u>CURVE 4</u>	
		333	0.745

# THERMAL CONDUCTIVITY OF SILVER + MANGANESE ALLOYS

[Ag + Mn = 99.50%; impurity ≤ 0.20% each]



## SPECIFICATION TABLE NO. 192 THERMAL CONDUCTIVITY OF SILVER-MANGANESE ALLOYS

(Ag-Mn 99.50%; impurity = 0.20% each)

[For Data Reported in Figure and Table No. 192]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)		Composition (continued), Specifications and Remarks
							Ag	Mn	
1	22	L	1959	1.5-57			99.72	0.28	Calculated composition; prepared from pure Ag and Mn with 0.00005 impurities, uniform polycrystalline; specimen cross-section 2.5 mm <sup>2</sup> ; rolled; $\rho_0 = 1.17 \mu\text{ohm cm}$ .
2	649	L	1956	1.5-4.1			99.72	0.28	Calculated composition; measured in a magnetic field of 25.5 kilooersteds.
3	649	L	1956	2.6, 3.7			99.72	0.28	Calculated composition; measured in a magnetic field of 19 kilooersteds.
4	649	L	1956	1.5-3.9		1	99.72	0.28	Calculated composition; measured in a magnetic field of 12 kilooersteds.
5	51	L	1956	14-88			99.72	0.28	Calculated composition; prepared from pure Ag and Mn with 0.00005 impurities; polycrystalline; supplied by Nordiska AIFvarett, Halsingborg; specimen 2.5 x 2.5 mm <sup>2</sup> in cross-section, rolled and cut; annealed at 730 K.
6	51	L	1956	16-72		4	87.31	12.69	Similar to the above specimen.



DATA TABLE NO. 192 THERMAL CONDUCTIVITY OF [SILVER + MANGANESE] ALLOYS

(Ag + Mn = 99.50%; impurity &lt; 0.20% each)

[Temperature, T, K; Thermal Conductivity, k, Watts cm<sup>-1</sup>K<sup>-1</sup>]

f	k	T	k	T	k
CURVE 1					
1.47	0.0340	1.48	0.0405	17.7	0.702
1.55	0.0350	1.48	0.0425	19.8	0.724
1.96	0.0524	1.95	0.0610	19.4	0.764
1.98	0.0541	2.27	0.0680	19.8	0.764
2.27	0.0584	2.57	0.0727	20.4	0.800
2.28	0.0597	2.88	0.0805	68.4	1.660
2.31	0.0598	3.23	0.0920	74.0	1.700
2.61	0.0657	3.41	0.0980	79.2	1.690
2.86	0.0637	3.70	0.108	80.5	1.710
2.91	0.0734	3.92	0.115	88.0	1.720
3.08	0.0783	4.10	0.123		
3.26	0.0827	CURVE 2			
3.43	0.0880	CURVE 3			
3.74	0.0990	2.58	0.0688	15.6	0.340
3.81	0.101	3.72	0.1038	16.4	0.365
3.92	0.106	CURVE 4			
4.12	0.117	1.50	0.0402	17.6	0.492
14.83	0.532	2.25	0.0635	19.2	0.398
15.31	0.544	3.41	0.0926	67.6	0.780
16.98	0.599	3.91	0.111	72.4	0.940
17.48	0.620	CURVE 5			
18.03	0.643	14.4	0.610		
18.96	0.668	14.8	0.610		
19.38	0.689	15.0	0.632		
19.96	0.704	15.4	0.622		
67.46	1.260	16.3	0.654		
70.98	1.285	16.8	0.678		
73.98	1.286	17.2	0.664		
75.48	1.348				
77.39	1.328				
81.58	1.361				
86.51	1.406				

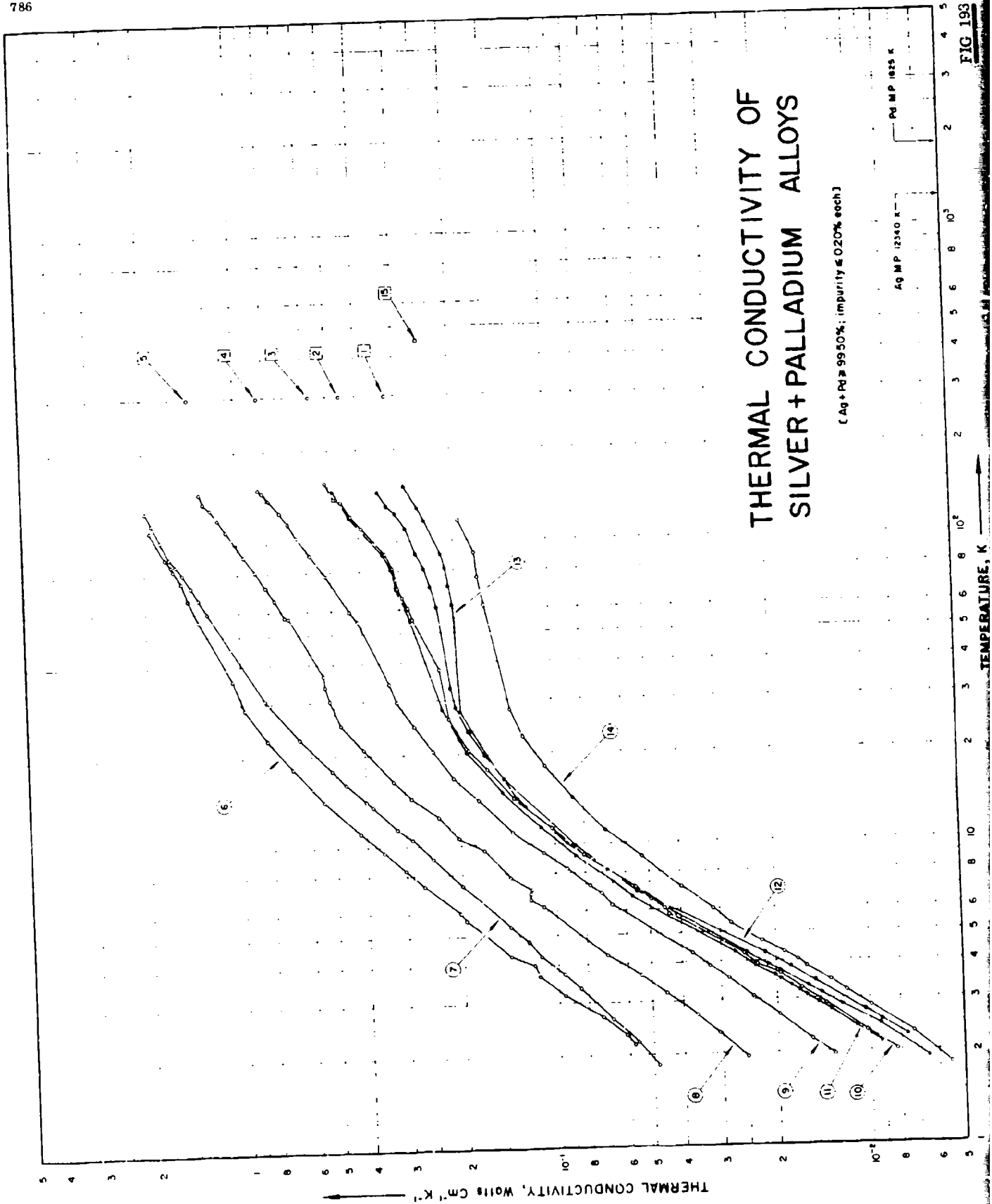


FIG 193

## SPECIFICATION TABLE NO. 193 THERMAL CONDUCTIVITY OF (SILVER-PALLADIUM) ALLOYS

(Ag + Pd = 99.50%; impurity = 0.20% each)

[For Data Reported in Figure and Table No. 193]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) Ag Pd	Composition (continued), Specifications and Remarks
1	241	E	1911	298.2			50 50	Approx. composition; electrical conductivity $3.03 \times 10^4$ ohm <sup>-1</sup> cm <sup>-1</sup> at 25 C.
2	241	E	1911	298.2			60 40	Approx. composition; electrical conductivity $3.56 \times 10^4$ ohm <sup>-1</sup> cm <sup>-1</sup> at 25 C.
3	241	E	1911	298.2			70 30	Approx. composition; electrical conductivity $6.43 \times 10^4$ ohm <sup>-1</sup> cm <sup>-1</sup> at 25 C.
4	241	E	1911	298.2			80 20	Approx. composition; electrical conductivity $9.47 \times 10^4$ ohm <sup>-1</sup> cm <sup>-1</sup> at 25 C.
5	241	E	1911	298.2			50 50	Approx. composition; electrical conductivity $16.14 \times 10^4$ ohm <sup>-1</sup> cm <sup>-1</sup> at 25 C.
6	240	L	1956	2.2-112			97.92 2.08	Specimen supplied by Johnson, Matthey and Co., Ltd.; annealed at 610 C. $\rho_0 = 0.99 \mu\text{ohm cm}$ ; electrical resistivity $2.52 \mu\text{ohm cm}$ at 293 K.
7	240	L	1956	1.8-123			97.92 2.08	Specimen supplied by Johnson, Matthey and Co., Ltd.; strained; $\rho_0 = 0.14 \mu\text{ohm cm}$ ; electrical resistivity $2.54 \mu\text{ohm cm}$ at 293 K.
8	240	L	1956	1.9-147			94.94 5.06	Specimen supplied by Johnson, Matthey and Co., Ltd.; annealed at 610 C. $\rho_0 = 2.20 \mu\text{ohm cm}$ ; electrical resistivity $3.91 \mu\text{ohm cm}$ at 293 K.
9	240	L	1956	2.0-150			90.1 9.9	Specimen supplied by Johnson, Matthey and Co., Ltd.; annealed at 650 C. $\rho_0 = 4.15 \mu\text{ohm cm}$ ; electrical resistivity $6.0 \mu\text{ohm cm}$ at 293 K.
10	240	L	1956	2.3-157			79.92 20.08	Specimen supplied by Johnson, Matthey and Co., Ltd.; annealed at 650 C.
11	240	L	1956	2.1-147			79.92 20.08	Specimen supplied by Johnson Matthey and Co., Ltd.; annealed at 800 C. $\rho_0 = 8.45 \mu\text{ohm cm}$ ; electrical resistivity $10.0 \mu\text{ohm cm}$ at 293 K.
12	240	L	1956	2.2-145			70.38 29.62	Specimen supplied by Johnson, Matthey and Co., Ltd.; annealed at 800 C. $\rho_0 = 12.78 \mu\text{ohm cm}$ ; electrical resistivity $14.66 \mu\text{ohm cm}$ at 293 K.
13	240	L	1956	1.9-151			60 40	Specimen supplied by Johnson, Matthey and Co., Ltd.; annealed at 880 C. $\rho_0 = 18.10 \mu\text{ohm cm}$ ; electrical resistivity $21.1 \mu\text{ohm cm}$ at 293 K.

SPECIFICATION TABLE NO. 193 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) Ag	Composition (weight percent) Pd	Composition (continued), Specifications and Remarks
14	240	L	1956	1.8-117			50	50	Specimen supplied by Johnson, Matthey and Co., Ltd.; annealed at 880 C; $\rho_{27.7}$ = 27.7 $\mu\text{hm cm}$ , electrical resistivity 27.7 $\mu\text{hm cm}$ at 293 K.
15	390	P	1956	448.2			50	50	



SPECIFICATION TABLE NO. 191 THERMAL CONDUCTIVITY OF (SILVER + PLATINUM) ALLOYS

(Ag + Pt - 99.50% impurity - 0.20% each)

Curve No.	Ref. Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) Ag	Composition (weight percent) Pt	Composition (continued), Specifications and Remarks
1	241 E	1911	298			90.0	10.0	
2	241 E	1911	298			75.0	25.0	
3	241 E	1911	298			70.0	30.0	
4	241 E	1911	298			67.0	33.0	
5	390 P	1956	523.2			59.0	50.0	

DATA TABLE NO. 191 THERMAL CONDUCTIVITY OF (SILVER + PLATINUM) ALLOYS

(Ag + Pt - 99.50% impurity - 0.20% each)

[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup>K<sup>-1</sup>]

T k

CURVE 1

298 0.98

CURVE 2

298 0.38

CURVE 3

298 0.31

CURVE 4

298 0.30

CURVE 5

523.2 0.372

† No graphical presentation

SPECIFICATION LABEL NO. 195 THERMAL CONDUCTIVITY OF (SILVER + TIN) ALLOYS

(Ag + Sn 99.50% impurity ± 0.20 each)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error,	Name and Specimen Designation	Composition (weight percent)		Composition (continued), Specifications and Remarks
							Ag	Sn	
1	2-30	L	1925	333.2			96.0	10.0	
2	2-30	L	1925	333.2			86.0	20.0	
3	2-30	L	1925	333.2			70.0	30.0	
4	2-30	L	1925	333.2			60.0	40.0	
5	2-30	L	1925	333.2			50.0	50.0	

DATA TABLE NO. 195 THERMAL CONDUCTIVITY OF (SILVER + TIN) ALLOYS

(Ag + Sn 99.50% impurity ± 0.20% each)

(Temperature, T, K. Thermal Conductivity,  $k$ , W/m<sup>2</sup>°K<sup>-1</sup>)

T	k	T	k
<u>CURVE 1</u>			
333.2	0.297	333.2	0.577
<u>CURVE 2</u>			
333.2	0.197		
<u>CURVE 3</u>			
333.2	0.393		
<u>CURVE 4</u>			
333.2	0.490		

No graphical presentation

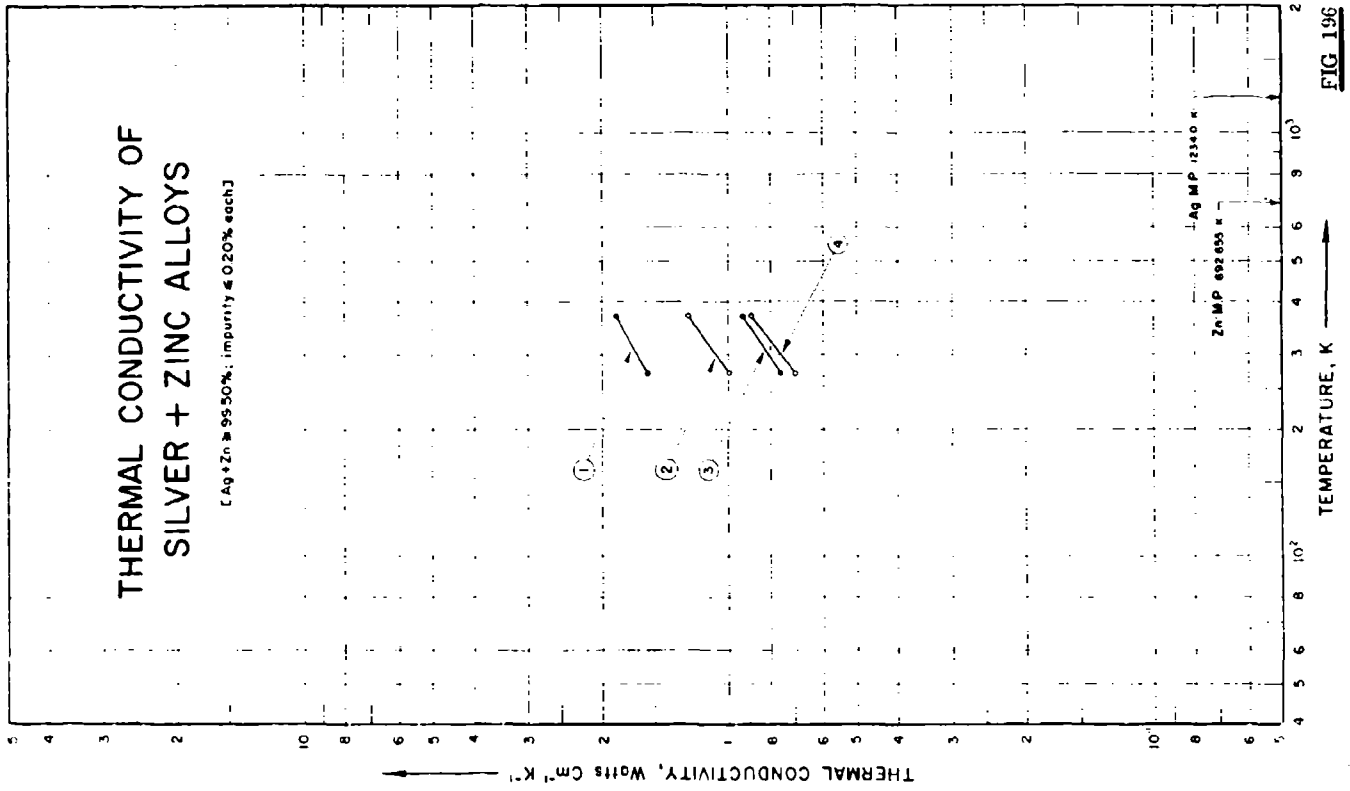


FIG 196



SPECIFICATION TABLE NO. 196 THERMAL CONDUCTIVITY OF [SILVER (ZINC) ALLOYS

(Ag + Zn 99.50% purity 0.50% each)

[For Data Reported in Figure and Table No. 196]

Curve No.	Rel. Method No. Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) Ag Zn	Composition (continued), Specifications and Remarks
1	246	T	1919	273.373		96.47 3.53	Calculated composition; specimen rolled and drawn to wire 1 mm thick; heated to near melting point for 0.5 hr; electrical conductivity $21.4$ and $19.5 \times 10^4 \text{ ohm}^{-1}\text{cm}^{-1}$ at 0 and 100 C., respectively.
2	246	T	1919	273.373		92.63 7.37	Similar to the above specimen except electrical conductivity $13.5$ and $13.0 \times 10^4 \text{ ohm}^{-1}\text{cm}^{-1}$ at 0 and 100 C., respectively.
3	246	T	1919	273.373		86.93 13.07	Similar to the above specimen except electrical conductivity $9.3$ and $9.2 \times 10^4 \text{ ohm}^{-1}\text{cm}^{-1}$ at 0 and 100 C., respectively.
4	246	T	1919	273.373		81.06 18.94	Similar to the above specimen except electrical conductivity $8.1$ and $8.2 \times 10^4 \text{ ohm}^{-1}\text{cm}^{-1}$ at 0 and 100 C., respectively.

## DATA TABLE NO. 196 THERMAL CONDUCTIVITY OF SILVER-ZINC ALLOYS

(Ag-Zn 99.50%; impurity 0.20% each)

[Temperature, T, K; Thermal Conductivity,  $k$ , Watt cm<sup>-1</sup> K<sup>-1</sup>]

T	k
<u>CURVE 1</u>	
273.2	1.56
373.2	1.43
<u>CURVE 2</u>	
273.2	1.00
373.2	1.27
<u>CURVE 3</u>	
273.2	0.76
373.2	0.93
<u>CURVE 4</u>	
273.2	0.70
373.2	0.49

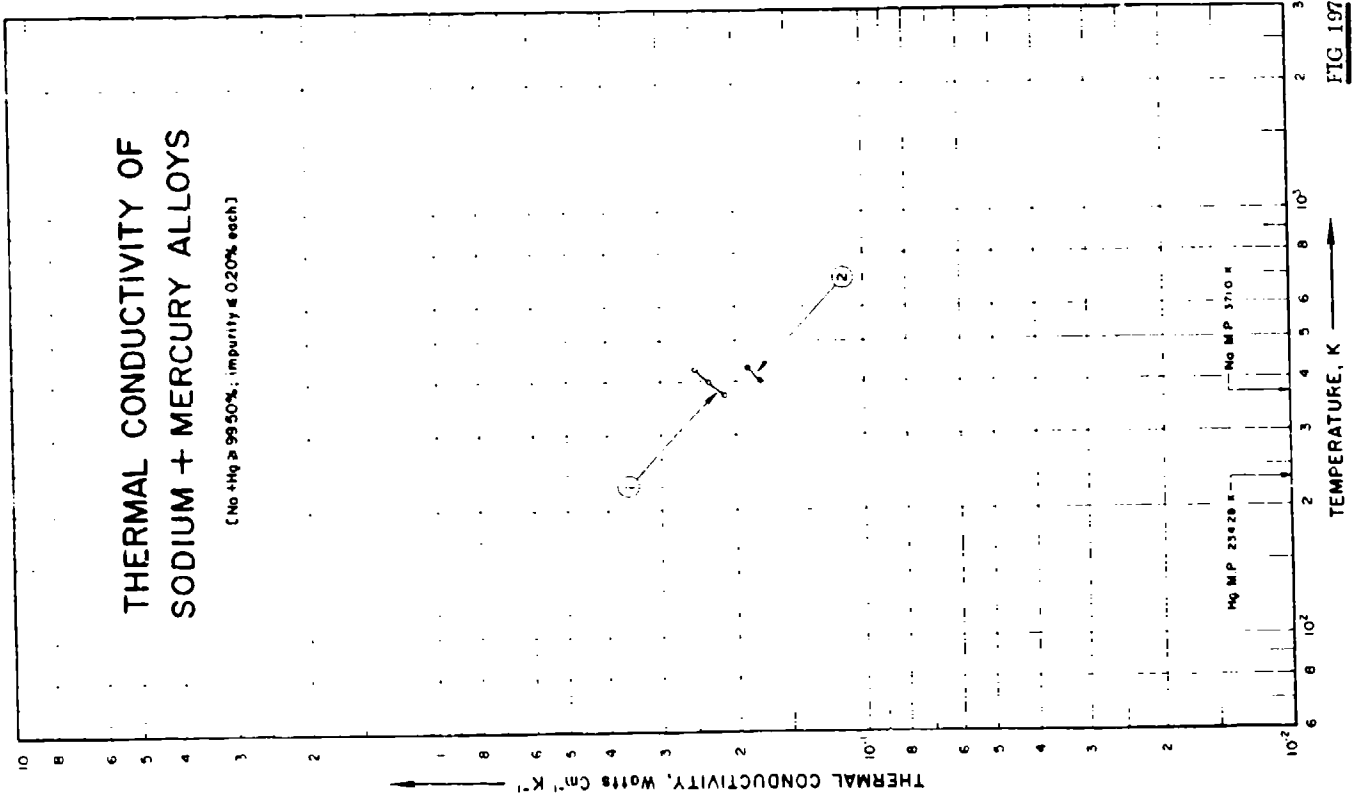


FIG 197

SPECIFICATION TABLE NO. 197 THERMAL CONDUCTIVITY OF (SODIUM + MERCURY) ALLOYS  
(Na + Hg - 99.50%; impurity 0.20%)

(For Data Reported in Figure and Table No. 197)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition Na	Composition (weight percent) Hg	Composition (continued), Specifications and Remarks
1	65	L	1938	373-429	± 3.0		53.63	36.97	Calculated composition; pure; supplied by Mallinckrodt Chemical Co.; in liquid state.
2	65	L	1938	405-434	± 3.0		52.50	47.50	Similar to the above specimen.

DATA TABLE NO. 197 THERMAL CONDUCTIVITY OF [SODIUM + MERCURY] ALLOYS  
 (Na + Hg = 99.50%; impurity = 0.50%)

[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup> K<sup>-1</sup>]

T	k
<u>CURVE 1</u>	
373.4	0.213
399.8	0.232
429.0	0.249
<u>CURVE 2</u>	
464.6	0.175
433.5	0.187

# THERMAL CONDUCTIVITY OF SODIUM + POTASSIUM ALLOYS

(Na + K ≥ 99.50%; impurity ≤ 0.20% each)

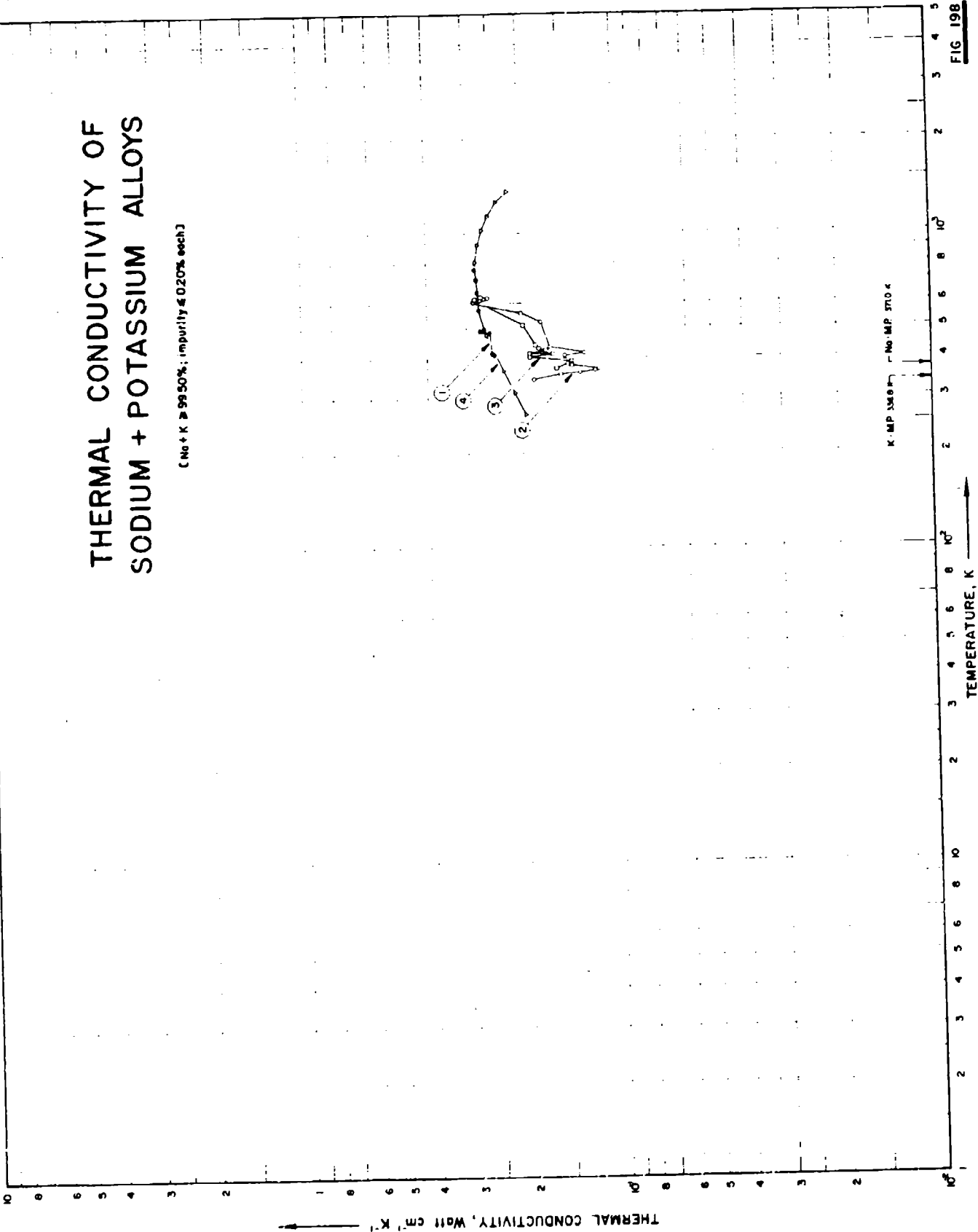


FIG 198

SPECIFICATION TABLE NO. 194 THERMAL CONDUCTIVITY OF [SODIUM - POTASSIUM] ALLOYS  
(Na - K - 99.50%, impurity  $\pm 0.20\%$  each)

(For Data Reported in Figure and Table No. 198)

Curve No.	Ref. No. Used	Method	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) Na	Composition (weight percent) K	Composition (continued), Specifications and Remarks
1	242	C	1946	412-770			51.7	48.3	Commercially pure; in liquid state; stainless steel used as comparative material (thermal conductivity $\sim 0.2$ watt cm <sup>-1</sup> C <sup>-1</sup> ).
2	243	L	1950	347-621		Na K			Eutectic composition; prepared from commercially pure Na and K, in liquid state (melting point $-12.5$ C).
3	243	L	1950	373-622		Na K			The above specimen solidified and then remelted.
4	859, 861	-	1965	266-1366		Na K			Eutectic composition; electrical resistivity reported as $36.26 \sim 151.28$ $\mu\text{ohm cm}$ in the range $310 \sim 1363$ K; thermal conductivity data calculated from measured electrical resistivity values and the Lorenz number $2.45 \times 10^{-8}$ W ohm K <sup>-2</sup> .

## DATA TABLE NO. 19s THERMAL CONDUCTIVITY OF [SODIUM + POTASSIUM] ALLOYS

(Na + K - 59, 50%, impurity  $\pm 0.20\%$  each)[Temperature, T, K; Thermal Conductivity, k, Watt  $\text{cm}^{-1} \text{K}^{-1}$ ]

T	K	T	k
<u>CURVE 1</u>			
412.20	0.245	431.20	0.176
415.80	0.249	441.20	0.180
484.50	0.254	512.20	0.197
489.30	0.271	512.20	0.197*
495.80	0.263	612.20	0.240
497.00	0.261 <sup>c</sup>	622.20	0.268
570.90	0.274		
653.00	0.276		
714.80	0.278		
769.70	0.282		
<u>CURVE 2</u>			
346.70	0.184	266	0.154
355.20	0.151	311	0.210
360.20	0.130	366	0.227
363.20	0.130 <sup>b</sup>	478	0.257
366.50	0.117	589	0.274*
381.20	0.134	700	0.279*
397.20	0.146	811	0.278
409.80	0.146	922	0.273
417.20	0.126	1033	0.264
435.20	0.163	1144	0.252
524.20	0.172	1255	0.236
559.20	0.201	1366	0.218
560.20	0.192 <sup>c</sup>		
505.20	0.285		
611.20	0.272		
618.20	0.264		
621.20	0.255		
<u>CURVE 3</u>			
373.20	0.155		
395.20	0.142		
392.20	0.138		
406.20	0.188		
413.20	0.159		
414.20	0.188		
423.20	0.167		

\* Not shown on plot



199 THERMAL CONDUCTIVITY OF [TANTALUM + NIOBIUM] ALLOYS

(Ta + Nb = 99.50%; impurity < 0.20% each)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) Ta	Composition (weight percent) Nb	Composition (continued), Specifications and Remarks
1	304	L	1966	1208-2900 ± 12			99.61	0.33	0.02 Mo, 0.014 W, < 0.01 each of Fe, Si, and Ti; specimen 7.28 mm in dia and 65.8 mm long; prepared from a bar produced by electron-beam melting in vacuum; density 16.57 g/cm <sup>3</sup> at 20 C; electrical resistivity reported as 54.8, 63.3, 64.5, 72.4, 80.7, 90.5, 100.4 and 105.2 $\mu$ ohm cm at 1243, 1488, 1512, 1750, 2010, 2350, 2623, and 2782 K, respectively.

199 THERMAL CONDUCTIVITY OF [TANTALUM + NIOBIUM] ALLOYS

(Ta + Nb = 99.50%; impurity < 0.20% each)

[ Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup>K<sup>-1</sup> ]

T	k	T	k	CURVE I (cont.) <sup>a</sup>	
				CURVE I <sup>a</sup>	CURVE I (cont.) <sup>a</sup>
1208	0.557	2160	0.604		
1250	0.510	2275	0.604		
1280	0.522	2360	0.621		
1306	0.572	3393	0.629		
1350	0.547	2506	0.635		
1410	0.545	2695	0.631		
1460	0.606	2906	0.663		
1490	0.549	2900	0.684		
1505	0.572				
1545	0.596				
1560	0.621				
1606	0.594				
1670	0.586				
1862	0.586				
2037	0.602				
2060	0.602				

<sup>a</sup> No graphical presentation

# THERMAL CONDUCTIVITY OF TANTALUM + TUNGSTEN ALLOYS

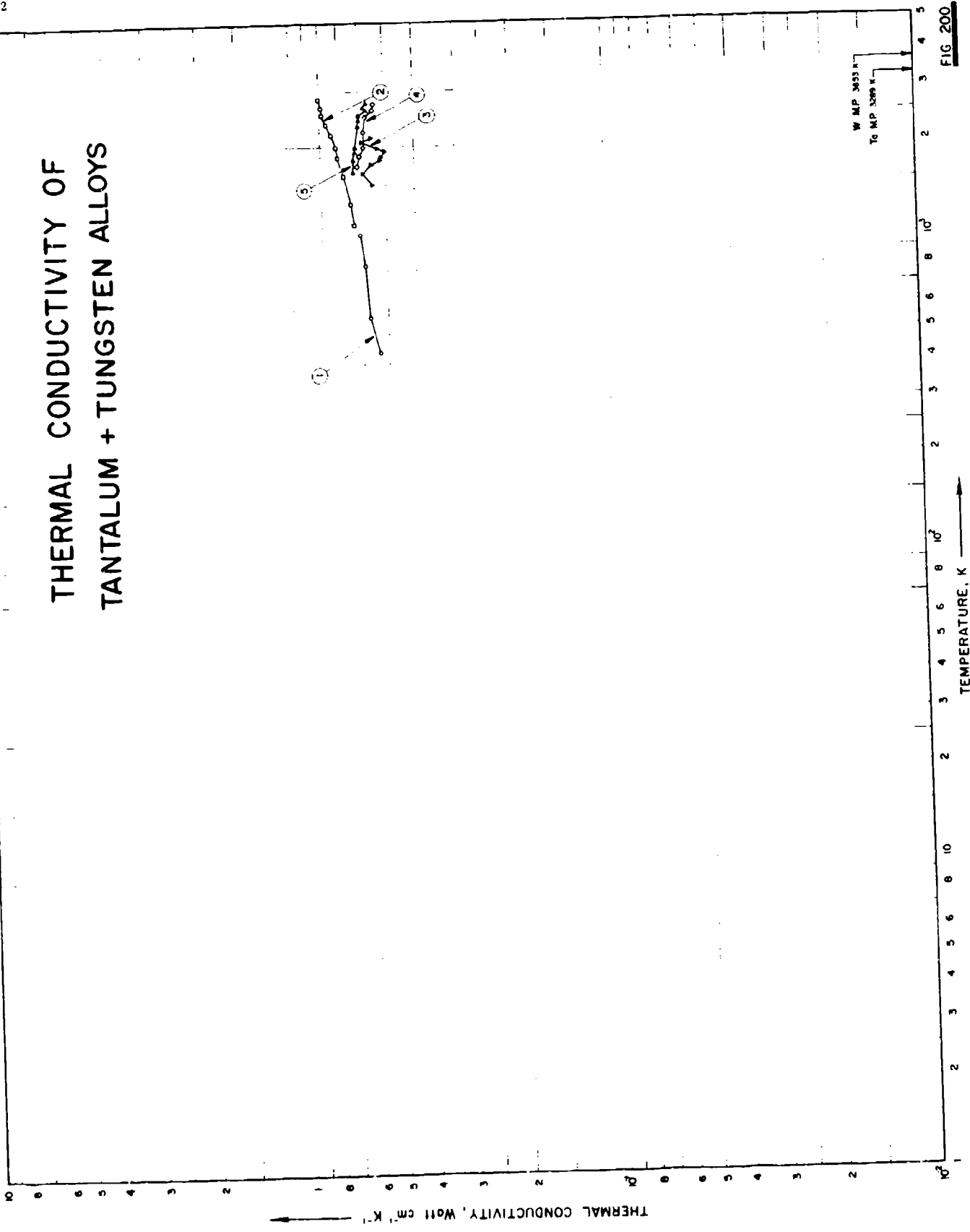


FIG. 200

## SPECIFICATION TABLE NO. 200 THERMAL CONDUCTIVITY OF TANTALUM-TUNGSTEN ALLOYS

(Ta-W 99.50%; impurity 0.20%)

For Data Reported in Figure and Table No. 200

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) Ta	Composition (weight percent) W	Composition (continued), Specifications and Remarks
1	583	C	1963	433-1049	±4		99.43	9.40	Impurity: 0.10 Nb, 0.02 Si, 0.02 Ti, 0.01 Mo, 0.01 Ni, 0.005 Fe, and 0.004 C; specimen 2 in. dia, 1 in. long; measured in an He atmosphere, Armo Iron as standard. The above specimen measured by another method.
2	583	P	1963	1122-2830	±4		99.43	9.40	
3	849	-	1966	1514-2110		Ta-10 W; No. 1	Bal.	9.0	0.0025 C and 0.002 O; specimen 2, 126.5 cm in dia and 0.3254 cm long; density 16.91 g cm <sup>-3</sup> ; thermal conductivity was derived from the temp distribution on the flat surface of the cylindrical disc specimen heated in high vacuum (10 <sup>-6</sup> mm Hg) by high frequency induction generating localized heating within 0.003 in. of the surface at current frequency of 500,000 cps with heat lost only by radiation; the cylindrical surface being assumed isothermal, and the temp gradient along the radius was analytically correlated to the thermal conductivity.
4	849	-	1966	1731-2742		Ta-10 W; No. 2	Bal.	9.9	0.0025 C and 0.002 O; similar to the above specimen except specimen 2, 125.1 cm in dia and 0.3145 cm long.
5	849	-	1966	1652-2756		T 111; No. 1	Bal.	7.8	2.0 H, 0.0034 O, 0.0023 N, 0.0022 C and 0.0005 H; cut from the same bar as the above specimen; specimen 2, 5476 cm in dia and 0.2504 cm long; density 16.73 g cm <sup>-3</sup> ; measuring method same as that for the above specimen.
6	849	-	1966	1324-2082		T 111; No. 2	Bal.	7.8	2.0 H, 0.0034 O, 0.0023 N, 0.0022 C, and 0.0005 H; similar to the above specimen.

## DATA TABLE NO. 200 THERMAL CONDUCTIVITY OF [TANTALUM + TUNGSTEN] ALLOYS

(Ta + W .99.50%; impurity = 0.20%)

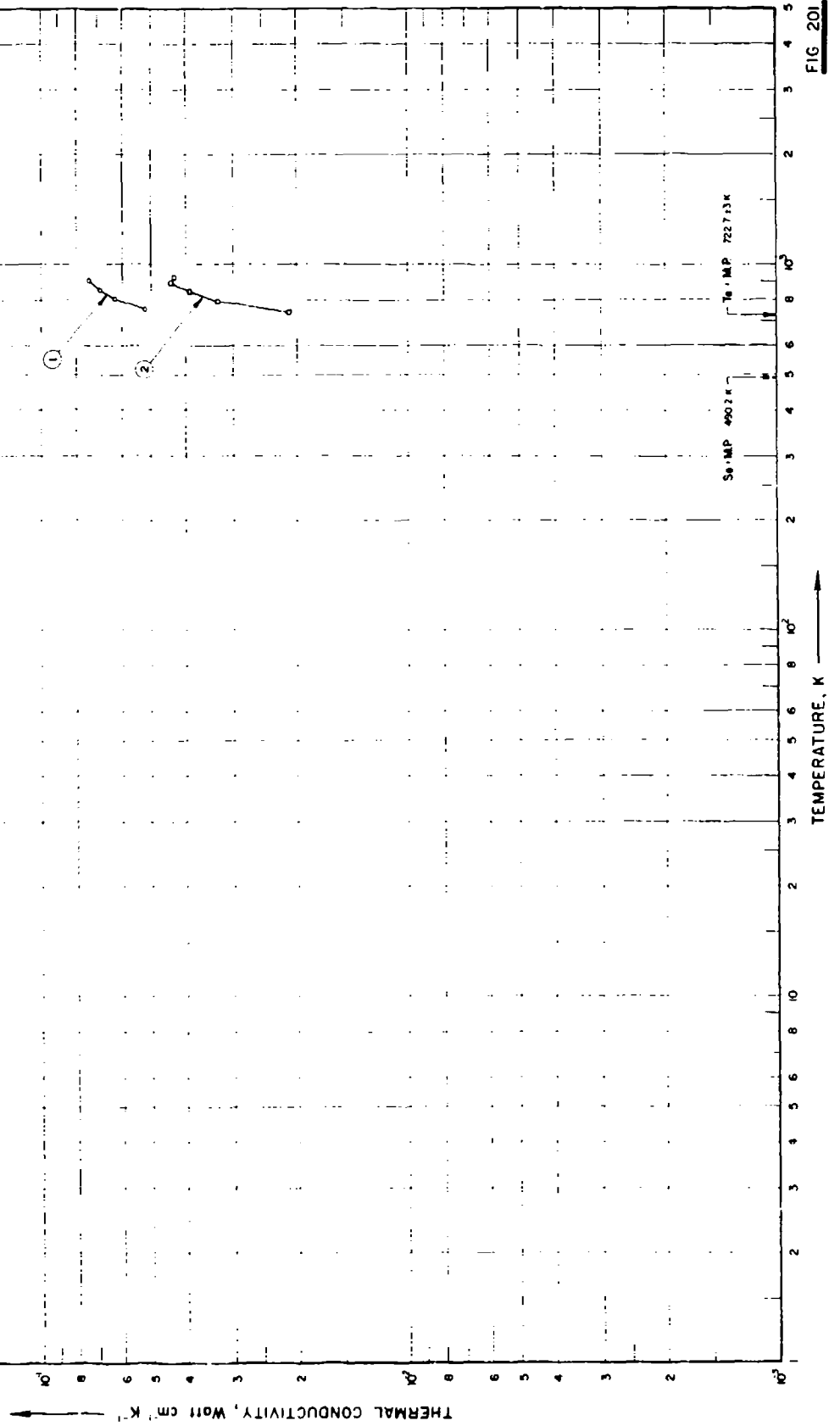
Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup> K<sup>-1</sup>

T	k	T	k
<u>CURVE 1</u>			
435.4	0.537	1652	0.628
565.9	0.571	1816	0.627
828.2	0.588	1912	0.617
1048.7	0.604	1990.5	0.617
<u>CURVE 2</u>			
1122.1	0.630	2103.5	0.596
1313.7	0.644	2324.5	0.600
1608.2	0.677	2430	0.596
1849.8	0.701	2520	0.596
1997.1	0.713	2604	0.570
2188.7	0.736	2671	0.579
2369.3	0.762	2756	0.565
2538.7	0.781	<u>CURVE 5</u>	
2677.6	0.787	1524	0.675
2849.8	0.800	1585	0.627
<u>CURVE 3</u>			
1514	0.546	1685	0.686
1648	0.585	1900	0.626
1753	0.562	1808	0.622
1907	0.519	1915	0.647
1867	0.516	2022	0.557
1938	0.496	2082	0.596
1973	0.526	<u>CURVE 6</u>	
2072	0.586	1524	0.675
2130	0.547	1585	0.627
<u>CURVE 4</u>			
1731	0.610	1685	0.686
1868	0.599	1900	0.626
1990.5	0.581	1808	0.622
2075	0.582	1915	0.647
2238	0.580	2022	0.557
2496	0.571	2082	0.596
2619	0.542	<u>CURVE 6</u>	
2742	0.538	1524	0.675
		1585	0.627
		1685	0.686
		1900	0.626
		1808	0.622
		1915	0.647
		2022	0.557
		2082	0.596

Not shown on plot

# THERMAL CONDUCTIVITY OF TELLURIUM+SELENIUM ALLOYS

(Te + Se ≥ 99.50%, impurity ≤ 0.20% each)



TEMPERATURE, K →

FIG 201

SPECIFICATION TABLE NO. 201 THERMAL CONDUCTIVITY OF (TELLURIUM + SELENIUM) ALLOYS

(Te + Se (99.50% impurity (0.50%))

[For Data Reported in Figure and Table No. 201]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) Te	Composition (weight percent) Se	Composition (continued), Specifications and Remarks
1	914	L	1966	753-904			96.85	3.15	Prepared from 99.995 pure Te and 99.9985 pure Se; molten specimen contained in a short cylindrical cell.
2	914	L	1966	742-923			86.60	13.40	Similar to the above specimen.

## DATA TABLE NO. 201 THERMAL CONDUCTIVITY OF [TELLURIUM + SELENIUM] ALLOYS

(Te + Se - 99.50%, impurity - 0.20% each)

[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup>K<sup>-1</sup>]

T k

CURVE 1

753	0.052
805	0.063
850	0.065
904	0.074

CURVE 2

742	0.021
794	0.033
843	0.030
893	0.044
923	0.043

# THERMAL CONDUCTIVITY OF TELLURIUM+THALLIUM ALLOYS

[Te+Tl: 99.50%; purity  $\leq$  0.20% each]

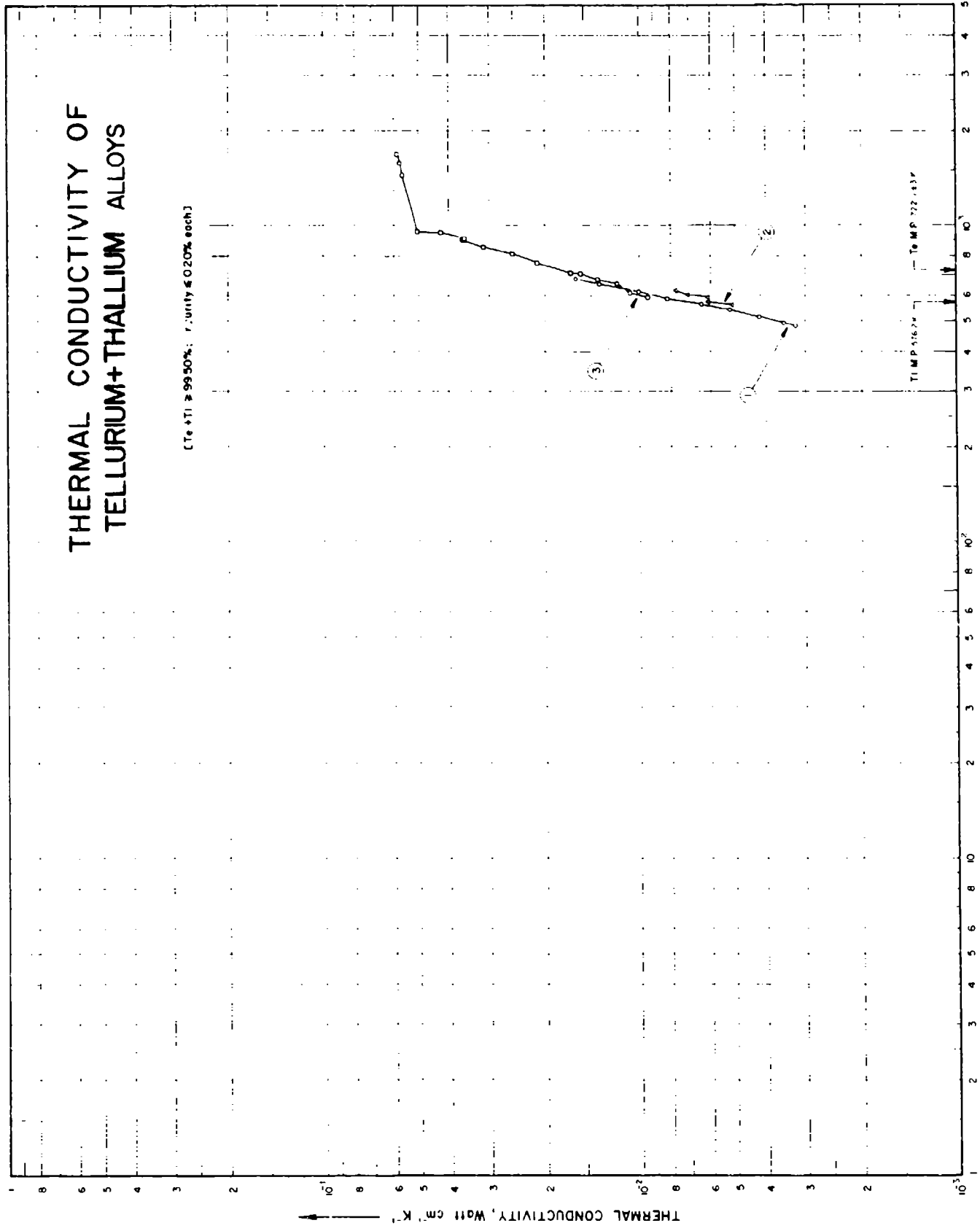


FIG. 202



SPECIFICATION TABLE NO. 202 THERMAL CONDUCTIVITY OF TELLURIUM + TITANIUM ALLOYS

(Te + Ti = 99.56%; impurity = 0.20%)

(For Data Reported in Figure and Table No. 202)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) Fe Ti	Composition (continued), Specifications and Remarks
1	927	E	1965	483-680			69 31	p-type; electrical resistivity 9.79, 2.80, 1.60, and 1.41 x 10 <sup>-5</sup> ohm cm at 205, 302, 401, and 400 C respectively; measured by the neck-down sample method.
2	927	E	1965	561-627			55 42	p-type; electrical resistivity 4.33, 3.61, and 2.69 x 10 <sup>-5</sup> ohm cm at 256, 310, and 378 C respectively; measured by the neck-down sample method.
3	927	E	1965	594-1008			52 48	p-type; electrical resistivity 4.24, 2.79, 1.93, 1.62, and 1.26 x 10 <sup>-5</sup> ohm cm at 321, 424, 527, 620, and 796 C respectively; measured by the neck-down sample method.

## DATA TABLE NO. 202 THERMAL CONDUCTIVITY OF TELLURIUM + THALLIUM ALLOYS

(Te + Tl = 99.50%; impurity &lt; 0.20%, each)

T, Temperature, °K. k, Thermal Conductivity, K. Watt cm<sup>-1</sup>K<sup>-1</sup>

T	k
<u>CURVE 1</u>	
483.2	0.0032
492.2	0.0035
515.2	0.0042
541.2	0.0052
569.2	0.0064
598.2	0.0082
620.2	0.0100
653.2	0.0135
690.2	0.0161
<u>CURVE 2</u>	
561.2	0.0051
576.2	0.0061
592.2	0.0065
607.2	0.0071
627.2	0.0077
<u>CURVE 3</u>	
594.2	0.0094
617.2	0.0107
657.2	0.0118
672.2	0.0136
701.2	0.0154
709.2	0.0166
762.2	0.0202
814.2	0.0252
855.2	0.0314
902.2	0.0364
910.2	0.0361
930.2	0.0441
959.2	0.0505
1045.2	0.0568
1058.2	0.0576
1068.2	0.0583

## SPECIFICATION TABLE NO. 203 THERMAL CONDUCTIVITY OF [THALLIUM + CADMIUM] ALLOYS

(Ti + Cd = 99.50%; impurity &lt; 0.20% each)

Curve No.	Ref. Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)		Composition (continued), Specifications and Remarks
						Te	Cd	
1	230 L	1925	336.2			90.0	10.0	Impurities < 0.03.
2	230 L	1925	336.2			80.0	20.0	Impurities < 0.03.
3	230 L	1925	336.2			70.0	30.0	Impurities < 0.03.
4	230 L	1925	336.2			60.0	40.0	Impurities < 0.03.
5	230 L	1925	336.2			50.0	50.0	Impurities < 0.03.

## DATA TABLE NO. 203 THERMAL CONDUCTIVITY OF [THALLIUM + CADMIUM] ALLOYS

(Ti + Cd) = 99.50%; impurity &lt; 0.20% each)

[ Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup>K<sup>-1</sup> ]

T	k	T	k
CURVE 1*			
336.2	0.444	336.2	0.661
CURVE 2*			
336.2	0.494		
CURVE 3*			
336.2	0.536		
CURVE 4*			
336.2	0.582		

\* No graphical presentation

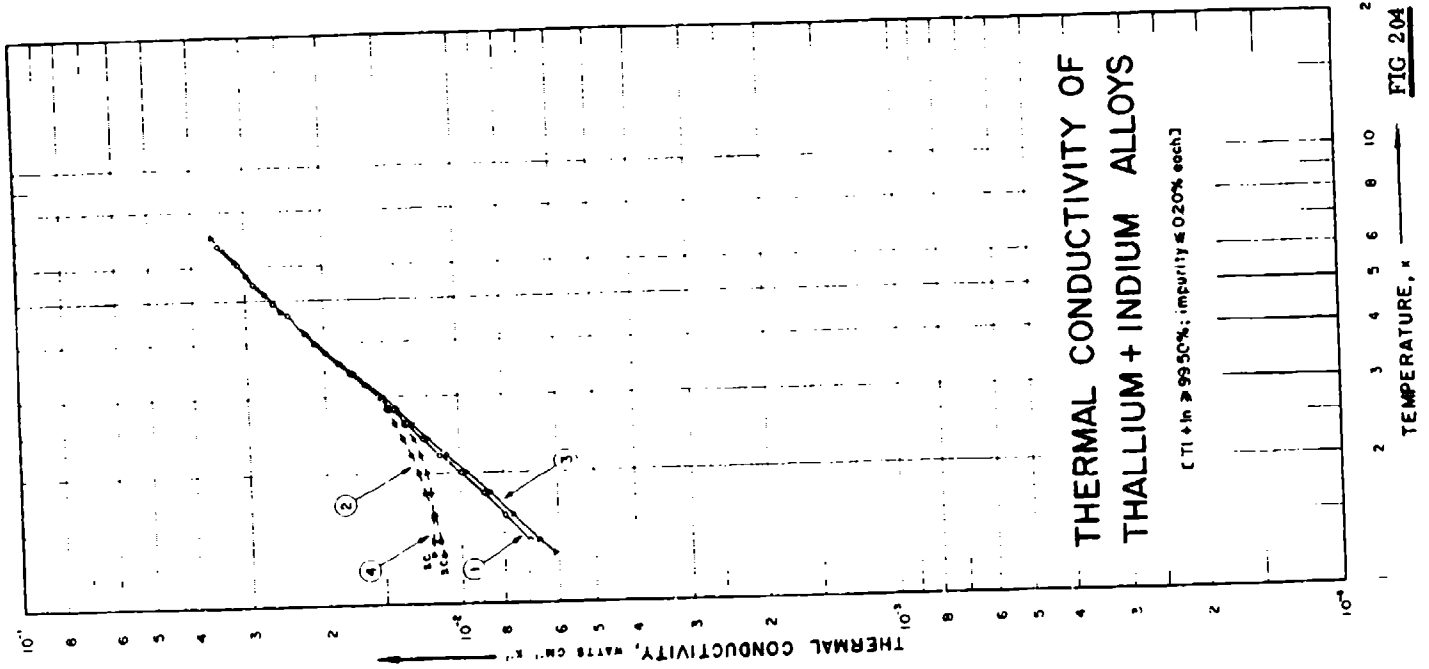


FIG. 204

SPECIFICATION TABLE NO. 204 THERMAL CONDUCTIVITY OF [THALLIUM - INDIUM] ALLOYS  
(Tl + In 99.50%; impurity 0.20%)

For Data Reported in Figure and Table No. 204

Curve No.	Ref. No. Used	Method	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition weight percent Tl	Composition weight percent In	Composition (continued), Specifications and Remarks
1	132	L	1955	1.3-6.7	2.5		52.17	47.83	Calculated composition; 0.05 impurities; annealed polycrystal; measured under vacuum of $\sim 5 \times 10^{-5}$ mm Hg and in a longitudinal magnetic field; in normal state.
2	132	L	1955	1.3-2.8	2.5		52.17	47.83	The above specimen in superconducting state.
3	132	L	1955	1.3-7.0	2.5		64.03	35.97	Calculated composition; 0.05 impurities; annealed polycrystal; measured in a longitudinal magnetic field and under vacuum of $\sim 5 \times 10^{-5}$ mm Hg; in normal state.
4	132	L	1955	1.3-2.6	2.5		64.03	35.97	The above specimen in superconducting state.

## DATA TABLE NO. 204 THERMAL CONDUCTIVITY OF [THALLIUM + INDIUM] ALLOYS

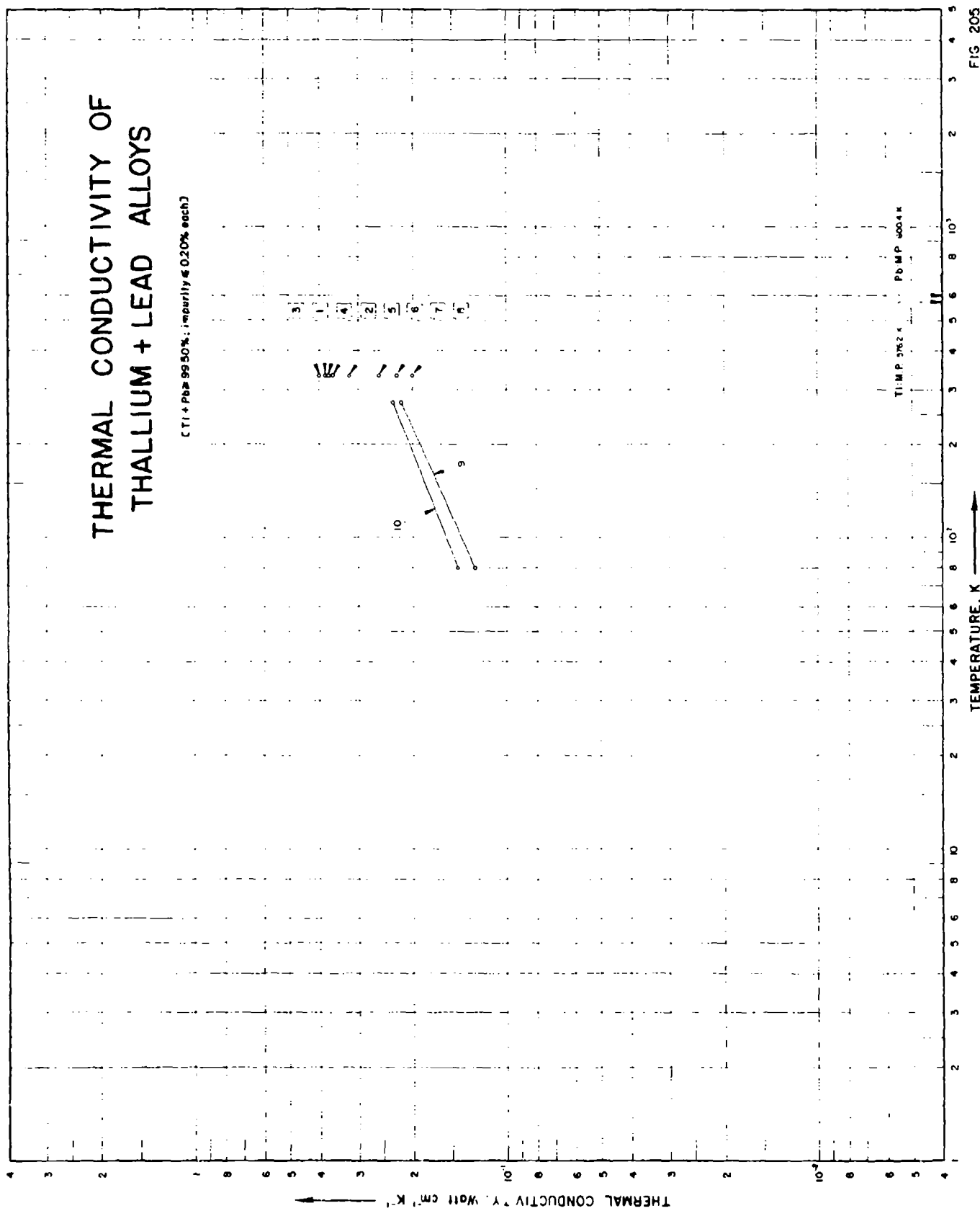
(Tl + In : 99.50%; impurity &lt; 0.20%)

[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup>K<sup>-1</sup>]

T	k	T	k
<u>CURVE 1</u>			
1.30	0.0064	2.40	0.0118
1.40	0.0069	2.60	0.0128
1.60	0.0079	2.80	0.0139
1.80	0.0089	3.00	0.0151
2.00	0.0099	3.20	0.0162
2.20	0.0110	3.40	0.0174
2.40	0.0120	3.60	0.0186
2.60	0.0131	3.80	0.0198
2.80	0.0141	4.00	0.0211
3.00	0.0152	4.20	0.0223
3.20	0.0164	4.74	0.0251
3.40	0.0175	5.21	0.0272
3.60	0.0187	5.74	0.0298
3.80	0.0200	6.19	0.0318
4.00	0.0212	7.01	0.0359
4.20	0.0224		
4.67	0.0244	<u>CURVE 4</u>	
4.95	0.0261	1.30	0.0115
5.48	0.0289	1.40	0.0115
6.04	0.0314	1.60	0.0115
6.71	0.0346	1.80	0.0116
		2.00	0.0118
		2.20	0.0121
		2.40	0.0125
		2.60	0.0132
		<u>CURVE 2</u>	
1.30	0.0109		
1.40	0.0111		
1.60	0.0114		
1.80	0.0119		
2.00	0.0123		
2.20	0.0128		
2.40	0.0134		
2.60	0.0140		
2.80	0.0146		
		<u>CURVE 3</u>	
1.30	0.0061		
1.40	0.0066		
1.60	0.0076		
1.80	0.0086		
2.00	0.0097		
2.20	0.0107		

# THERMAL CONDUCTIVITY OF THALLIUM + LEAD ALLOYS

CTI + Pb ≥ 99.50%; impurity ≤ 0.20% each



## SPECIFICATION TABLE NO. 205 THERMAL CONDUCTIVITY OF [THALLIUM + LEAD] ALLOYS

(Tl + Pb : 99.50%; impurity &lt; 0.20% each)

[ For Data Reported in Figure and Table No. 205 ]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)		Composition (continued), Specifications and Remarks
							Tl	Pb	
1	230	L	1925	333			98	2	Prepared from Pb (< 0.03 impurity, supplied by Baker) and Tl (technically pure, supplied by Elmer and Amend); specimen -5.5 cm long, 0.3 cm <sup>2</sup> cross-sectional area; electrical conductivity at 25 C, $\sigma(25\text{ C})$ $4.95 \times 10^4$ ohm <sup>-1</sup> cm <sup>-1</sup> .
2	230	L	1925	333			96	4	Similar to the above specimen except $\sigma(25\text{ C})$ $4.72 \times 10^4$ ohm <sup>-1</sup> cm <sup>-1</sup> .
3	230	L	1925	333			94	6	Similar to the above specimen except $\sigma(25\text{ C})$ $5.16 \times 10^4$ ohm <sup>-1</sup> cm <sup>-1</sup> .
4	230	L	1925	333			90	10	Similar to the above specimen except $\sigma$ (25 C) $4.90 \times 10^4$ ohm <sup>-1</sup> cm <sup>-1</sup> .
5	230	L	1925	333			80	20	Similar to the above specimen except $\sigma$ (25 C) $4.02 \times 10^4$ ohm <sup>-1</sup> cm <sup>-1</sup> .
6	230	L	1925	333			70	30	Similar to the above specimen except $\sigma$ (25 C) $3.04 \times 10^4$ ohm <sup>-1</sup> cm <sup>-1</sup> .
7	230	L	1925	333			60	40	Similar to the above specimen except $\sigma$ (25 C) $2.63 \times 10^4$ ohm <sup>-1</sup> cm <sup>-1</sup> .
8	230	L	1925	333			50	50	Similar to the above specimen except $\sigma$ (25 C) $2.63 \times 10^4$ ohm <sup>-1</sup> cm <sup>-1</sup> .
9	34	L	1927	80, 273		PbTl; I	66.0	34.0	Coarse grained; electrical conductivity 2.817 and $4.2 \times 10^4$ ohm <sup>-1</sup> cm <sup>-1</sup> at 273 and 80 K, respectively.
10	34	L	1927	80, 273		PbTl; II	66.0	34.0	Fine grained; electrical conductivity 2.672 and $3.93 \times 10^4$ ohm <sup>-1</sup> cm <sup>-1</sup> at 273 and 80 K, respectively.



## DATA TABLE NO. 205 THERMAL CONDUCTIVITY OF [THALLIUM + LEAD] ALLOYS

(Tl + Pb : 99.50%; impurity : 0.20% each)

[Temperature, T, K; Thermal Conductivity, k, watt cm<sup>-1</sup>K<sup>-1</sup>]

T	k
<u>CURVE 1</u>	
333	0.385
<u>CURVE 2</u>	
333	0.364
<u>CURVE 3</u>	
533	0.402
<u>CURVE 4</u>	
333	0.377
<u>CURVE 5</u>	
333	0.322
<u>CURVE 6</u>	
333	0.259
<u>CURVE 7</u>	
333	0.226
<u>CURVE 8</u>	
333	0.201
<u>CURVE 9</u>	
80	0.127
273	0.219
<u>CURVE 10</u>	
80	0.14*
273	0.232

# THERMAL CONDUCTIVITY OF THALLIUM+TELLURIUM ALLOYS

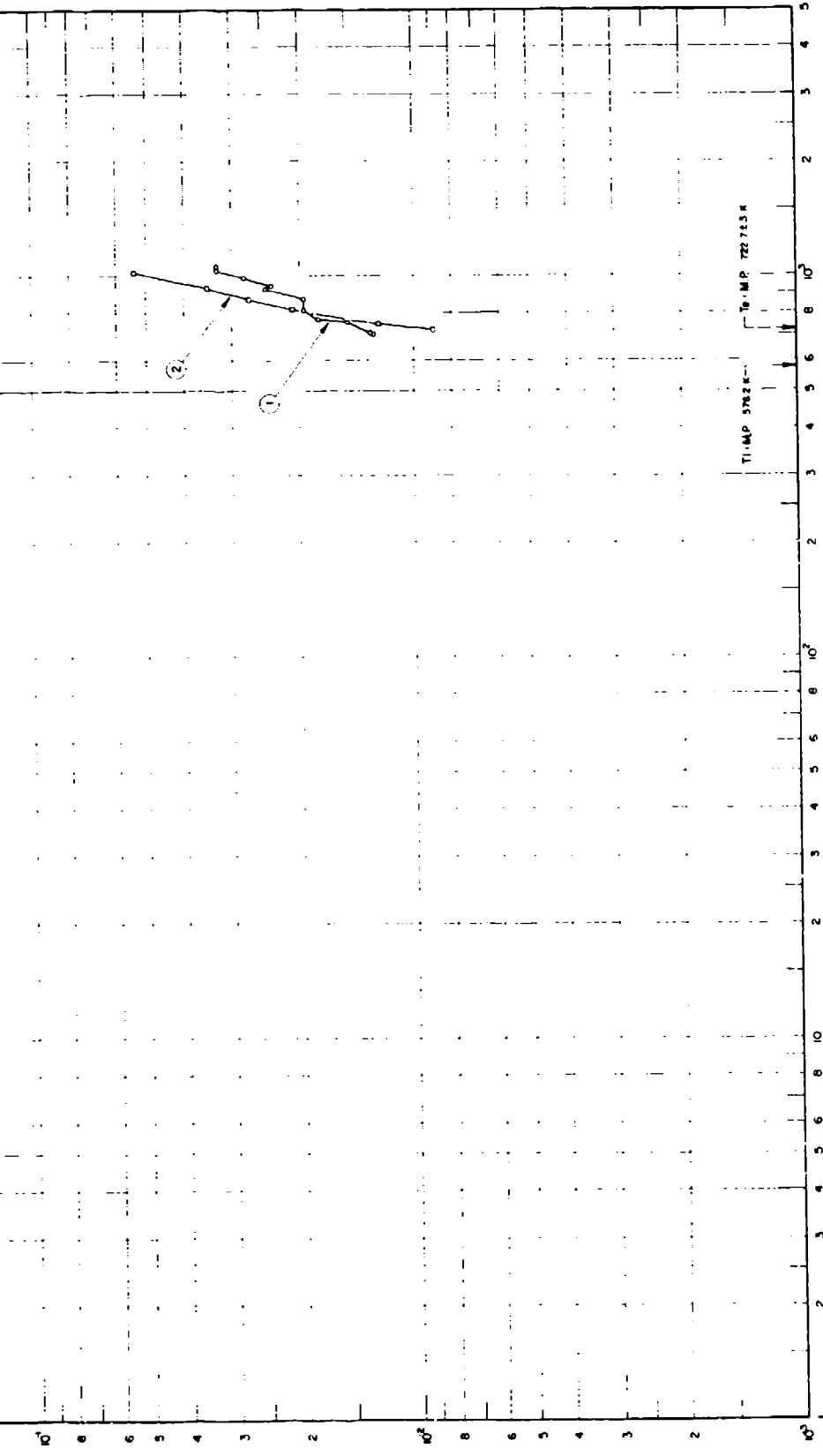
(TI + Te = 99.50%; impurity  $\leq$  0.20% each)

THERMAL CONDUCTIVITY,  $\text{Watt cm}^{-1} \text{K}^{-1}$

TEMPERATURE, K

TI M.P. 578.2 K  
Te M.P. 722.745 K

FIG 206



## SPECIFICATION TABLE NO. 206 THERMAL CONDUCTIVITY OF THALLIUM - TELLURIUM ALLOYS

(Tl - Te 50, 50%; impurity 0.20%)

For Data Reported in Figure and Table No. 206\*

Curve No.	Rel. Method No. Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) Tl	Composition (weight percent) Te	Composition (continued), Specifications and Remarks
1	E	1965	701-1067			54	46	p-type; electrical resistivity 2.95, 2.29, 1.95, 1.71, and 1.52 x 10 <sup>-3</sup> ohm cm at 431, 533, 656, 713, and 769 C respectively; measured by the neck-down sample method.
2	E	1965	722-1024			68	32	n-type; electrical resistivity 1.51, 1.49, 1.42, and 1.21 x 10 <sup>-3</sup> ohm cm at 478, 540, 648, and 768 C respectively; measured by the neck-down sample method.

## DATA TABLE NO. 206 THERMAL CONDUCTIVITY OF [THALLIUM + TELLURIUM] ALLOYS

(Tl + Te) 99.50%; impurity 0.20% each)

Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup>K<sup>-1</sup>

T	k
<u>CURVE 1</u>	
701.2	0.0128
710.2	0.0131
737.2	0.0149
769.2	0.0178
811.2	0.0195
868.2	0.0195
925.2	0.0246
942.2	0.0237
985.2	0.0277
1041.2	0.0329
1067.2	0.0328
<u>CURVE 2</u>	
722.2	0.0089
748.2	0.0124
813.2	0.0209
869.2	0.0170
928.2	0.0347
1024.2	0.0337

## SPECIFICATION TABLE NO. 207 THERMAL CONDUCTIVITY OF [ THALLIUM + TIN ] ALLOYS

(Tl + Sn - 99.50%; impurity &lt; 0.20% each)

Curve No.	Ref. Method No. Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) Tl	Composition (weight percent) Sn	Composition (continued), Specifications and Remarks
1	230 L	1925	336.2			99.0	10.0	Impurities < 0.03.
2	230 L	1925	336.2			80.0	20.0	Impurities < 0.03.
3	230 L	1925	336.2			70.0	30.0	Impurities < 0.03.
4	230 L	1925	336.2			60.0	40.0	Impurities < 0.03.
5	230 L	1925	336.2			53.4	46.6	Impurities < 0.03.
6	230 L	1925	336.2			50.0	50.0	Impurities < 0.03.

## DATA TABLE NO. 207 THERMAL CONDUCTIVITY OF [ THALLIUM + TIN ] ALLOYS

(Tl + Sn - 99.50%; impurity < 0.20% each)  
[ Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup>K<sup>-1</sup> ]

T	k	T	k
<u>CURVE 1<sup>a</sup></u>		<u>CURVE 5<sup>a</sup></u>	
336.2	0.301	336.2	0.331
<u>CURVE 2<sup>a</sup></u>		<u>CURVE 6<sup>a</sup></u>	
336.2	0.255	336.2	0.372
<u>CURVE 3<sup>a</sup></u>			
336.2	0.259		
<u>CURVE 4<sup>a</sup></u>			
336.2	0.289		

<sup>a</sup> No graphical presentation

SPECIFICATION TABLE NO. 298 THERMAL CONDUCTIVITY OF [ THORIUM + URANIUM ] ALLOYS

(Th + U - 99.50%; impurity - 0.20% each)

Curve No.	Ref. Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)		Composition (continued), Specifications and Remarks
						Th	U	
1	396	C	1954	293-673		97	3	

DATA TABLE NO. 298 THERMAL CONDUCTIVITY OF [ THORIUM + URANIUM ] ALLOYS

(Th + U - 99.50%; impurity - 0.20% each)

[ Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup>K<sup>-1</sup> ]

T	k
293.2	0.39
373.2	0.39
473.2	0.39
573.2	0.40
673.2	0.40

CURVE 1

No graphical presentation

SPECIFICATION TABLE NO. 209 THERMAL CONDUCTIVITY OF [TIN + ALUMINUM] ALLOYS

(Sn + Al) = 99.50%; impurity 0.50% each

Curve No.	M.L. Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)		Composition (continued), Specifications and Remarks
						Sn	Al	
1	L	1925	324.2			50.0	50.0	Impurities less than 0.03.
2	L	1925	324.2			60.0	40.0	Impurities less than 0.03.
3	L	1925	324.2			70.0	30.0	Impurities less than 0.03.
4	L	1925	324.2			80.0	20.0	Impurities less than 0.03.
5	L	1925	324.2			90.0	10.0	Impurities less than 0.03.

DATA TABLE NO. 209 THERMAL CONDUCTIVITY OF [TIN + ALUMINUM] ALLOYS

(Sn + Al) = 99.50%; impurity 0.50% each

[Temperature, T, K Thermal Conductivity, k, Watt cm<sup>-1</sup>K<sup>-1</sup>]

T	k	T	k
<u>CURVE 1</u> <sup>c</sup>	<u>CURVE 5</u> <sup>c</sup>		
324.2	1.383	324.2	0.912
<u>CURVE 2</u> <sup>c</sup>			
324.2	1.255		
<u>CURVE 3</u> <sup>c</sup>			
324.2	1.142		
<u>CURVE 4</u> <sup>c</sup>			
324.2	0.950		

<sup>c</sup> No graphical presentation

# THERMAL CONDUCTIVITY OF TIN + ANTIMONY ALLOYS

(Sn + Sb ≥ 99.50%; impurity ≤ 0.20% each)

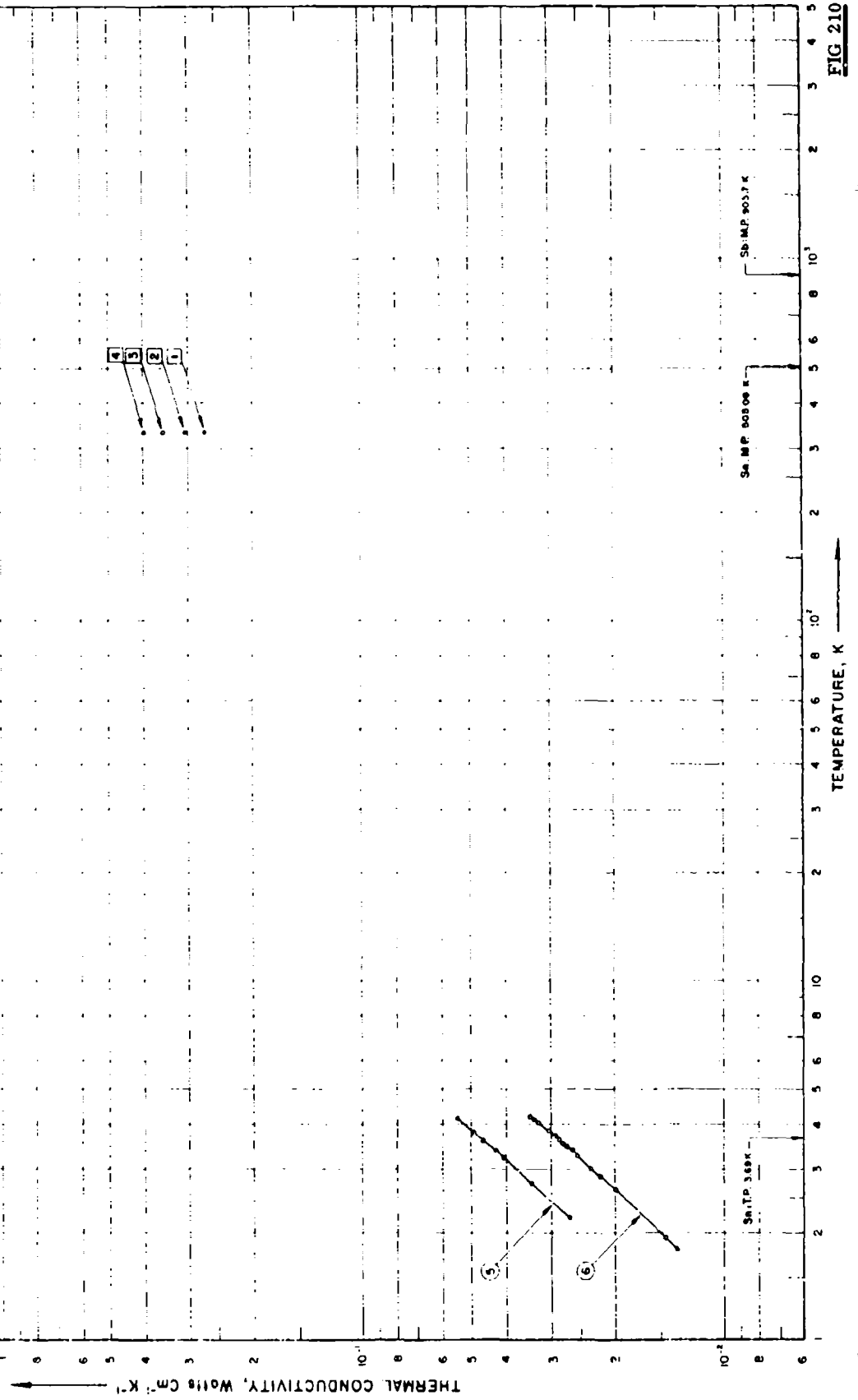


FIG 210



## SPECIFICATION TABLE NO. 210 THERMAL CONDUCTIVITY OF TIN - ANTIMONY ALLOYS

(Sn - Sb 99.50% impurity 0.20% each)

For Data Reported in Figure and Table No. 210\*

Curve No.	Rel. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) Sn	Composition (weight percent) Sb	Composition (continued), Specifications and Remarks
1	230	L	1925	330.2			50	50	Prepared from Sn and Sb both containing 0.03 impurities; supplied by Baker; specimen 10 cm long, 1.9 cm dia; electrical conductivity at 22°C: $\sigma$ (22°C) = $3.46 \times 10^4$ ohm <sup>-1</sup> cm <sup>-1</sup> .
2	230	L	1925	330.2			60	40	Similar to the above specimen except $\sigma$ (22°C) = $4.00 \times 10^4$ ohm <sup>-1</sup> cm <sup>-1</sup> .
3	230	L	1925	330.2			70	30	Similar to the above specimen except $\sigma$ (22°C) = $4.59 \times 10^4$ ohm <sup>-1</sup> cm <sup>-1</sup> .
4	230	L	1925	330.2			80	20	Similar to the above specimen except $\sigma$ (22°C) = $5.23 \times 10^4$ ohm <sup>-1</sup> cm <sup>-1</sup> .
5	315-456	L	1957	2.2-4.2	2		97.03	2.97	Specimen ~4 mm dia, ~10 cm long; annealed for several months; electrical resistivity 2.142 and 1.605 $\mu$ hm cm at 4.2 and 300 K, respectively; measured in a 560 gauss magnetic field; in normal state.
6	315-456	L	1957	1.4-4.2	2		93.85	6.15	Similar to the above specimen except electrical resistivity 3.483 and 1.491 $\mu$ hm cm at 4.2 and 300 K, respectively.

## DATA TABLE NO. 210 THERMAL CONDUCTIVITY OF [TIN + ANTIMONY] ALLOYS

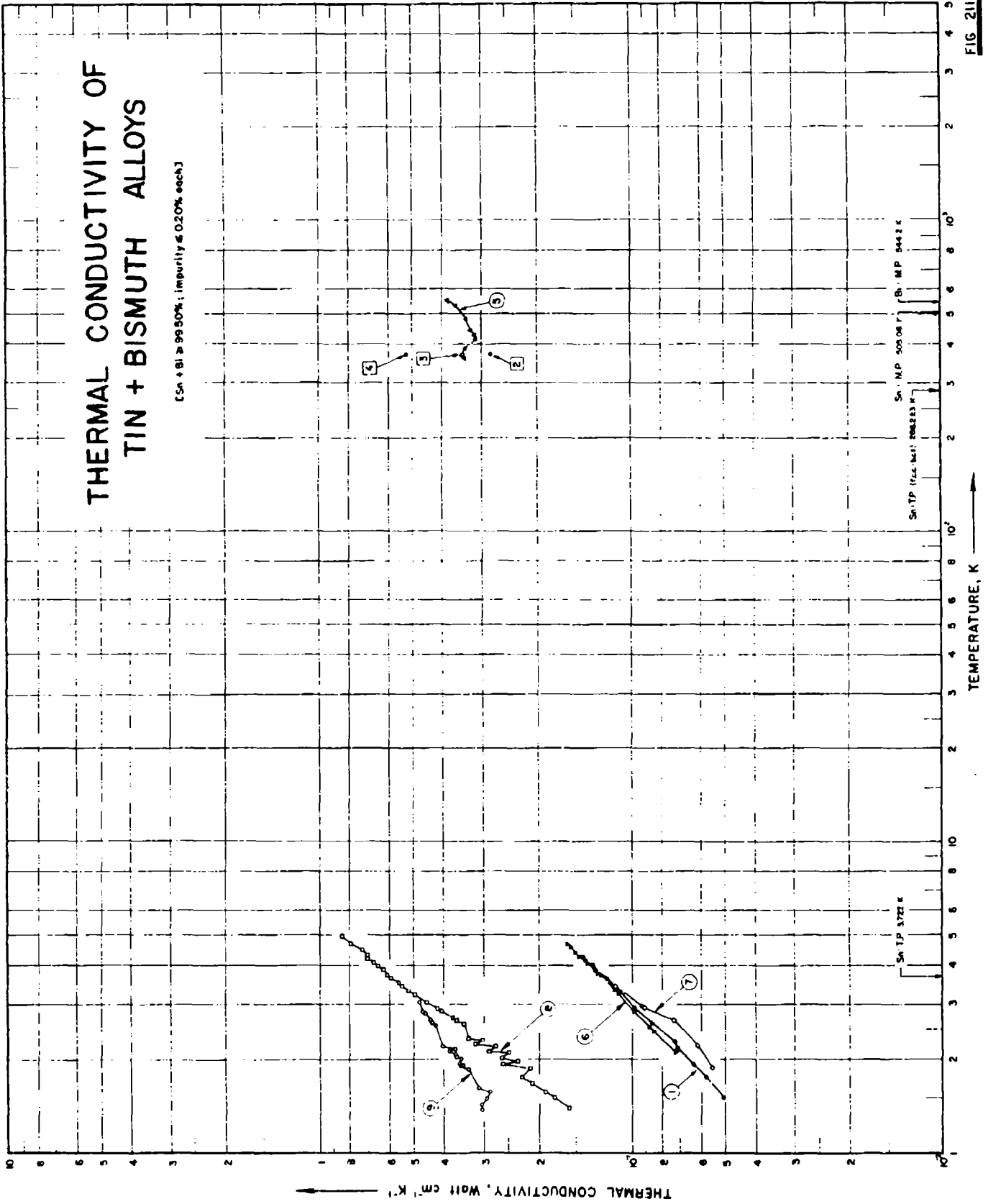
(Sn + Sb ± 99.50%: impurity ± 0.20% each)

[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup> K<sup>-1</sup>]

T	k
<u>CURVE 1</u>	
330.2	0.368
<u>CURVE 2</u>	
330.2	0.305
<u>CURVE 3</u>	
330.2	0.352
<u>CURVE 4</u>	
330.2	0.398
<u>CURVE 5</u>	
2.20	0.0368
2.74	0.0419
3.23	0.0406
3.37	0.0427
3.60	0.0462
3.80	0.0491
4.17	0.0544
<u>CURVE 6</u>	
1.40	0.0134
1.94	0.0145
2.63	0.0198
2.86	0.0219
3.00	0.0233
3.26	0.0254
3.40	0.0261
3.46	0.0272
3.54	0.0279
3.63	0.0286
3.70	0.0293
3.83	0.0304
4.03	0.0325
4.11	0.0332
4.20	0.0342

# THERMAL CONDUCTIVITY OF TIN + BISMUTH ALLOYS

{Sn + Bi ≥ 99.50%; impurity ≤ 0.20% each}



## SPECIFICATION TABLE NO. 211 THERMAL CONDUCTIVITY OF [TIN + BISMUTH] ALLOYS

(Sn + Bi : 99.50%, impurity : 0.20% each)

[For Data Reported in Figure and Table No. 211]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) Sn Bi	Composition (continued), Specifications and Remarks	
1	456	L	1958	1.5-4.2	2		98.08	1.92	Specimen ~4 mm dia. ~10 cm long; annealed for several months; measured in a 560 gauss magnetic field; in normal state.
2	460		1957	373.2			61	39	
3	460		1957	373.2			80	20	
4	460		1957	373.2			95	5	
5	514	L	1962	358-553	5		50	50	Prepared from 99.94 pure Sn; measured across melting point.
6	837	L	1967	3.1-4.7	1	Bi:3	98.46	1.54	Prepared by vacuum-melting appropriate amounts of Johnson-Matthey 99.999 pure Sn and Bi, extruding into 1.5 mm dia wire annealed at ~200 C for several days; electrical resistivity 0.796 and 13.31 $\mu\text{ohm cm}$ at 4.2 and 273 K respectively; Tc 3700 K; normal-state data were taken at temperatures below Tc with a longitudinal magnetic field applied to the sample.
7	837	L	1967	1.9-3.4	1	Bi:3	98.46	1.54	Same as the above specimen except the magnetic field was removed so the superconducting-state data were taken.
8	836	L	1958	1.4-5.0	4-5	4	99.47	0.53	Prepared by vacuum-melting appropriate amounts of Johnson-Matthey 99.999 pure Sn and Bi, casting into 1 mm dia x 12 cm long wire; electrical resistivity 0.250 $\mu\text{ohm cm}$ ; a magnetic field was applied when taking normal-state data at temperatures below Tc.
9	836	L	1958	1.4-3.1	4-5	4	99.47	0.53	Same as the above specimen except the magnetic field was removed so the superconducting-state data were taken.



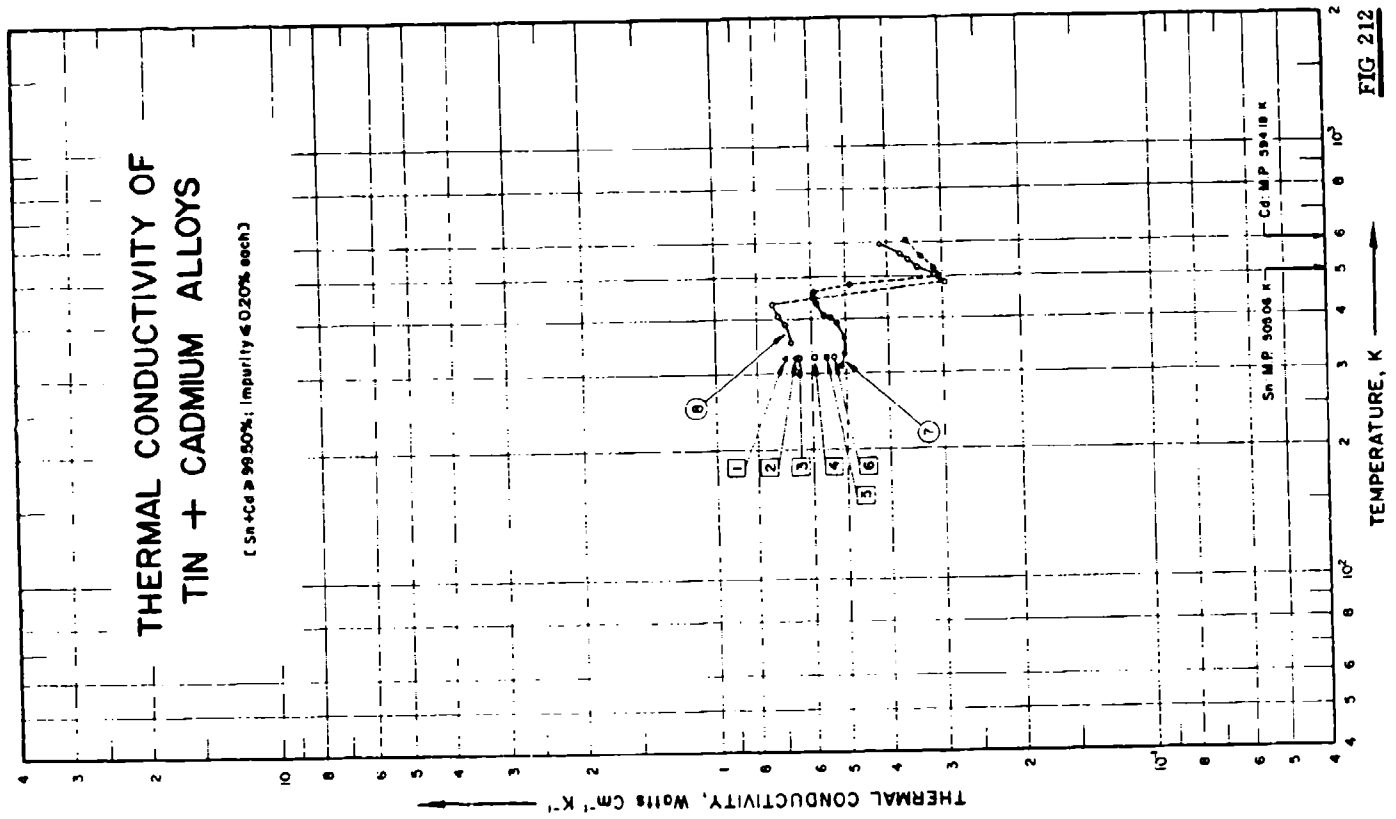


FIG 212

SPECIFICATION TABLE NO. 212 THERMAL CONDUCTIVITY OF TIN-CADMIUM ALLOYS

(Sn-Cd 99.50% purity - 0.20% each)

For Data Reported in Figure and Table No. 212

Curve No.	Rel. Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) Sn	Composition (weight percent) Cd	Composition (continued), Specifications and Remarks
1	230	L	1925	±26		50	50	Prepared from Sn containing 0.05 impurities; supplied by Baker; specimen 10 cm long, 1.9 cm dia.; electrical conductivity at 22°C: $\sigma (22^\circ\text{C}) = 9.98 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$ .
2	230	L	1925	±26		60	40	Similar to the above specimen except $\sigma (22^\circ\text{C}) = 9.11 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$ .
3	230	L	1925	±26		70	30	Similar to the above specimen except $\sigma (22^\circ\text{C}) = 9.15 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$ .
4	230	L	1925	±26		80	20	Similar to the above specimen except $\sigma (22^\circ\text{C}) = 8.39 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$ .
5	230	L	1925	±26		90	10	Similar to the above specimen except $\sigma (22^\circ\text{C}) = 7.73 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$ .
6	230	L	1925	±26		95	5	Similar to the above specimen except $\sigma (22^\circ\text{C}) = 7.54 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$ .
7	514	L	1962	314-603		99.94	0.06	Prepared from 99.94 pure Sn; in liquid state above 225°C.
8	514	L	1962	353-536		99.94	0.06	Prepared from 99.94 pure Sn; in liquid state above 215°C.

DATA TABLE NO. 212 THERMAL CONDUCTIVITY OF [TIN + CADMIUM] ALLOYS

(Sn + Cd : 99.50%, impurity : 0.20% each)

[Temperature, T. K. Thermal Conductivity, k, Watt cm<sup>-1</sup> K<sup>-1</sup>]

T	k	T	k
<u>CURVE 1</u>			
326	0.699	353.2	0.669
		388.2	0.690
<u>CURVE 2</u>			
		408.2	0.711
		433.2	0.732
326	0.653	488.2	0.297
		503.2	0.314
<u>CURVE 3</u>			
		528.2	0.343
		548.2	0.360
326	0.644	568.2	0.377
		593.2	0.418
<u>CURVE 4</u>			
326	0.594		
<u>CURVE 5</u>			
326	0.557		
<u>CURVE 6</u>			
326	0.536		
<u>CURVE 7</u>			
		313.2	0.515
		333.2	0.502
		363.2	0.506
		393.2	0.527
		403.2	0.544
		408.2	0.545
		433.2	0.586
		438.2	0.582
		448.2	0.598
		463.2	0.594
		478.2	0.490
		498.2	0.305
		523.2	0.314
		553.2	0.315
		603.2	0.364

Not shown on plot



SPECIFICATION TABLE NO. 213 THERMAL CONDUCTIVITY OF [TIN + COPPER] ALLOYS  
(Sn + Cu 99.50% impurity ± 0.20% each)

Curve No.	Ref. Method No. Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)		Composition (continued), Specifications and Remarks
						Sn	Cu	
1	459 R	1905	287		Sn <sub>99.25</sub> Cu <sub>0.75</sub>	90.25	9.75	Cast and turned.
2	459 R	1905	287		Sn <sub>95</sub> Cu <sub>5</sub>	75.05	24.95	Cast and turned.

DATA TABLE NO. 213 THERMAL CONDUCTIVITY OF [TIN + COPPER] ALLOYS

(Sn + Cu 99.50% impurity ± 0.20% each)  
[ Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup>K<sup>-1</sup> ]

T k

CURVE 1

286.7 0.548

CURVE 2

286.7 0.584

No graphical presentation

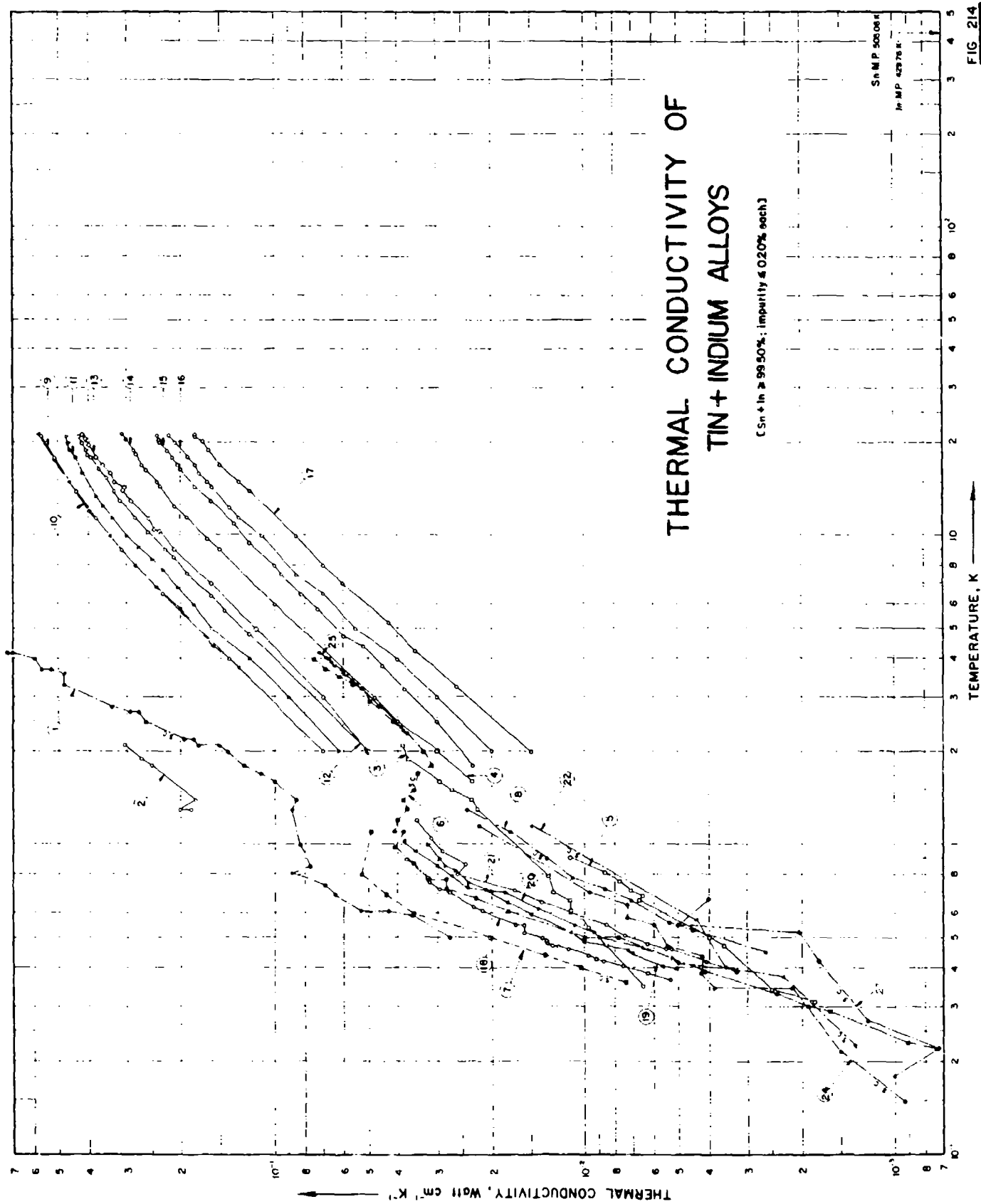


FIG. 214

## SPECIFICATION TABLE NO. 214 THERMAL CONDUCTIVITY OF (TN · INDIUM) ALLOYS

(Sn · In · 99.50%; impurity 0.20% each)

(For Data Reported in Figure and Table No. 214)

Curve No.	Rel. Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) Sn	In	Composition (continued), Specifications and Remarks	
1	452	L	1955	0.5-4.2	2-4	Sn 7 JM 4600 · In	99.65	0.35	Cast, single crystal with tetragonal axis 82° to rod axis; in superconducting state.
2	452	L	1955	1.3-2.1	2-4	Sn 7 JM 4600 · In	99.65	0.35	The above specimen measured in a longitudinal field of 400 gauss; in normal state.
3	452	L	1955	0.35-2.1	2-4	Sn 8 JM 4600 · In	96.9	3.1	Cast, single crystal with tetragonal axis 73° to rod axis; measured in a longitudinal field of 450 gauss; in normal state.
4	452	L	1955	1.6-2.0	2-4	Sn 8 JM 4600 · In	96.9	3.1	The above specimen measured after switching off the magnetic field.
5	452	L	1955	0.31-0.91	2-4	Sn 8 JM 4600 · In	96.9	3.1	The above specimen measured in a longitudinal field of 500 gauss; in normal state.
6	452	L	1955	0.22-4.0	2-4	Sn 8 JM 4600 · In	96.9	3.1	The above specimen in superconducting state.
7	452	L	1955	0.36-1.1	2-4	Sn 8 JM 4600 · In	96.9	3.1	The above specimen annealed at 210 C for 14 days; in superconducting state.
8	452	L	1955	0.45-1.3	2-4	Sn 9 JM 4600 · In	96.9	3.1	Polycrystal; grain size about 0.3 mm; cast, strained and annealed for 17 days; in superconducting state.
9	453	L	1958	2.0-21	<1.0	Sn 2	97.98	2.02	Prepared from spectroscopically pure Sn and 99.9 pure In; single crystal; angle between tetragonal axis and specimen axis (orientation) = 90°.
10	453	L	1958	3.8-21	<1.0	Sn 2'	97.98	2.02	Similar to the above specimen except orientation = 77°; in normal state.
11	453	L	1958	2.0-21	<1.0	Sn 2.1	97.89	2.11	Similar to the above specimen except orientation = 70°; in normal state.
12	453	L	1958	2.0-21	<1.0	Sn 2.5	97.55	2.45	Similar to the above specimen except orientation = 85°; in normal state.
13	453	L	1958	2.0-21	<1.0	Sn 2.8	97.16	2.84	Prepared from spectroscopically pure Sn and 99.9 pure In; single crystal with about 5% of volume being inclusions of foreign orientation; orientation = 78°; in normal state.
14	453	L	1958	2.0-22	<1.0	Sn 4	96.04	3.96	Prepared from spectroscopically pure Sn and 99.9 pure In; coarse polycrystal with grain size about 2-3 mm; in normal state.
15	453	L	1958	1.8-21	<1.0	Sn 5	91.03	4.97	Similar to the above specimen.

SPECIFICATION TABLE NO. 214 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) Sn In	Composition (continued), Specifications and Remarks
16	453	L	1958	2.0-21	±1.0	Sn 5.7	94.35	Similar to the above specimen.
17	453	L	1958	2.0-21	±1.0	Sn 5.2	91.8	Prepared from spectroscopically pure Sn and 99.9 pure In; fine-grained polycrystal, in normal state.
18	454	L	1958	0.37-0.90	2.0	B 0	97.0	Specimen 1.48 mm in diameter; cast; homogenized by annealing in vacuo for 2 weeks at 320 C.; electro-polished; angle between tetrad axis and specimen axis = 80°; surface roughness 0.09 μ.
19	454	L	1958	0.40-1.0	2.0	B 4	97.0	Similar to the above specimen except the diameter 0.875 mm and surface roughness 0.15 μ.
20	454	L	1958	0.39-1.0	2.0	B 1	97.0	Similar to the above specimen except the diameter 1.48 mm, surface roughness 1.0 μ, etched.
21	454	L	1958	0.39-1.2	2.0	B 5	97.0	Similar to the above specimen except the diameter 0.875 mm.
22	455	L	1954	0.23-1.2	10-25	Sn IX	97	Specimen 2.8 mm in diameter; polycrystal; cast in tube; crystals size of the order of the diameter; in superconducting state.
23	455	L	1953	0.18-0.67	10-25	Sn N	97	Similar to the above specimen except the size of crystals being a fraction of the diameter.
24	455	L	1954	0.15-1.2	10-25	Sn IV	99.7	Similar to the above specimen except crystals size of the order of the diameter.
25	456	L	1958	2.3-4.2	2	∴	97.0	Specimen annealed for several months; measured in a longitudinal magnetic field of about 560 gauss; in normal state.

DATA TABLE NO. 214 THERMAL CONDUCTIVITY OF [TIN + INDIUM] ALLOYS

(Sn + In = 99.50%, Impurity = 0.20% each)

[Temperature, T, K. Thermal Conductivity, k, Watts cm<sup>-1</sup>K<sup>-1</sup>]

T	k	<u>CURVE 1</u>	T	k	<u>CURVE 2</u>	T	k	<u>CURVE 3</u> (cont.)	T	k	<u>CURVE 4</u>	T	k	<u>CURVE 5</u>	T	k	<u>CURVE 6</u> (cont.)	T	k	<u>CURVE 7</u>	T	k	<u>CURVE 8</u>	T	k	<u>CURVE 9</u>	T	k	<u>CURVE 10</u>	T	k	<u>CURVE 11</u>	T	k	<u>CURVE 12</u>	T	k	<u>CURVE 13</u>	T	k	<u>CURVE 14</u>	T	k	<u>CURVE 15</u>	T	k	<u>CURVE 16</u>	T	k	<u>CURVE 17</u>	T	k	<u>CURVE 18</u>	T	k	<u>CURVE 19</u>	T	k	<u>CURVE 20</u>
0.50	0.0274	<u>CURVE 1</u>	0.56	0.0101	<u>CURVE 3</u> (cont.)	0.77	0.032	<u>CURVE 6</u> (cont.)	2.0	0.070	<u>CURVE 9</u>	2.00	0.050	<u>CURVE 12</u>	2.00	0.048	<u>CURVE 15</u>	2.00	0.048	<u>CURVE 18</u>	2.00	0.051	<u>CURVE 19</u>	2.00	0.050	<u>CURVE 20</u>	0.487	0.0134	<u>CURVE 16</u> (cont.)	0.487	0.0134	<u>CURVE 19</u>	0.396	0.0051	<u>CURVE 17</u>	0.396	0.0051	<u>CURVE 18</u>	0.396	0.0051	<u>CURVE 19</u>	0.396	0.0051	<u>CURVE 20</u>	0.396	0.0051	<u>CURVE 20</u>	0.396	0.0051	<u>CURVE 20</u>									
0.59	0.0357	<u>CURVE 1</u>	0.61	0.0111	<u>CURVE 3</u> (cont.)	0.87	0.0355	<u>CURVE 6</u> (cont.)	4.0	0.110	<u>CURVE 9</u>	4.80	0.120	<u>CURVE 12</u>	4.00	0.100	<u>CURVE 15</u>	4.00	0.100	<u>CURVE 18</u>	4.00	0.100	<u>CURVE 19</u>	4.00	0.100	<u>CURVE 20</u>	0.496	0.0134	<u>CURVE 16</u> (cont.)	0.496	0.0134	<u>CURVE 19</u>	0.402	0.0056	<u>CURVE 17</u>	0.402	0.0056	<u>CURVE 18</u>	0.402	0.0056	<u>CURVE 20</u>	0.402	0.0056	<u>CURVE 20</u>	0.402	0.0056	<u>CURVE 20</u>												
0.61	0.0430	<u>CURVE 1</u>	0.66	0.0112	<u>CURVE 3</u> (cont.)	0.98	0.041	<u>CURVE 6</u> (cont.)	6.5	0.230	<u>CURVE 9</u>	6.40	0.230	<u>CURVE 12</u>	6.00	0.220	<u>CURVE 15</u>	6.00	0.220	<u>CURVE 18</u>	6.00	0.220	<u>CURVE 19</u>	6.00	0.220	<u>CURVE 20</u>	0.532	0.0137	<u>CURVE 16</u> (cont.)	0.532	0.0137	<u>CURVE 19</u>	0.455	0.0072	<u>CURVE 17</u>	0.455	0.0072	<u>CURVE 18</u>	0.455	0.0072	<u>CURVE 20</u>	0.455	0.0072	<u>CURVE 20</u>	0.455	0.0072	<u>CURVE 20</u>												
0.61	0.0530	<u>CURVE 1</u>	0.70	0.0127	<u>CURVE 3</u> (cont.)	1.1	0.0381	<u>CURVE 6</u> (cont.)	9.0	0.310	<u>CURVE 9</u>	7.50	0.310	<u>CURVE 12</u>	8.00	0.310	<u>CURVE 15</u>	8.00	0.310	<u>CURVE 18</u>	8.00	0.310	<u>CURVE 19</u>	8.00	0.310	<u>CURVE 20</u>	0.518	0.0157	<u>CURVE 16</u> (cont.)	0.518	0.0157	<u>CURVE 19</u>	0.468	0.0085	<u>CURVE 17</u>	0.468	0.0085	<u>CURVE 18</u>	0.468	0.0085	<u>CURVE 20</u>	0.468	0.0085	<u>CURVE 20</u>	0.468	0.0085	<u>CURVE 20</u>												
0.69	0.064	<u>CURVE 1</u>	0.79	0.0131	<u>CURVE 3</u> (cont.)	1.1	0.041	<u>CURVE 6</u> (cont.)	11.5	0.380	<u>CURVE 9</u>	10.00	0.380	<u>CURVE 12</u>	11.50	0.380	<u>CURVE 15</u>	11.50	0.380	<u>CURVE 18</u>	11.50	0.380	<u>CURVE 19</u>	11.50	0.380	<u>CURVE 20</u>	0.548	0.0166	<u>CURVE 16</u> (cont.)	0.548	0.0166	<u>CURVE 19</u>	0.520	0.0111	<u>CURVE 17</u>	0.520	0.0111	<u>CURVE 18</u>	0.520	0.0111	<u>CURVE 20</u>	0.520	0.0111	<u>CURVE 20</u>	0.520	0.0111	<u>CURVE 20</u>												
0.74	0.069	<u>CURVE 1</u>	1.3	0.0222	<u>CURVE 3</u> (cont.)	1.2	0.040	<u>CURVE 6</u> (cont.)	14.0	0.440	<u>CURVE 9</u>	13.00	0.440	<u>CURVE 12</u>	14.00	0.440	<u>CURVE 15</u>	14.00	0.440	<u>CURVE 18</u>	14.00	0.440	<u>CURVE 19</u>	14.00	0.440	<u>CURVE 20</u>	0.61	0.0216	<u>CURVE 16</u> (cont.)	0.61	0.0216	<u>CURVE 19</u>	0.628	0.0238	<u>CURVE 17</u>	0.628	0.0238	<u>CURVE 18</u>	0.628	0.0238	<u>CURVE 20</u>	0.628	0.0238	<u>CURVE 20</u>	0.628	0.0238	<u>CURVE 20</u>												
0.81	0.086	<u>CURVE 1</u>	1.4	0.0232	<u>CURVE 3</u> (cont.)	1.3	0.037	<u>CURVE 6</u> (cont.)	18.0	0.515	<u>CURVE 9</u>	17.00	0.515	<u>CURVE 12</u>	18.00	0.515	<u>CURVE 15</u>	18.00	0.515	<u>CURVE 18</u>	18.00	0.515	<u>CURVE 19</u>	18.00	0.515	<u>CURVE 20</u>	0.628	0.0230	<u>CURVE 16</u> (cont.)	0.628	0.0230	<u>CURVE 19</u>	0.65	0.0238	<u>CURVE 17</u>	0.65	0.0238	<u>CURVE 18</u>	0.65	0.0238	<u>CURVE 20</u>	0.65	0.0238	<u>CURVE 20</u>	0.65	0.0238	<u>CURVE 20</u>												
0.85	0.077	<u>CURVE 1</u>	1.5	0.0270	<u>CURVE 3</u> (cont.)	1.4	0.0381	<u>CURVE 6</u> (cont.)	21.0	0.570	<u>CURVE 9</u>	19.00	0.570	<u>CURVE 12</u>	21.00	0.570	<u>CURVE 15</u>	21.00	0.570	<u>CURVE 18</u>	21.00	0.570	<u>CURVE 19</u>	21.00	0.570	<u>CURVE 20</u>	0.65	0.0236	<u>CURVE 16</u> (cont.)	0.65	0.0236	<u>CURVE 19</u>	0.68	0.0275	<u>CURVE 17</u>	0.68	0.0275	<u>CURVE 18</u>	0.68	0.0275	<u>CURVE 20</u>	0.68	0.0275	<u>CURVE 20</u>	0.68	0.0275	<u>CURVE 20</u>												
1.0	0.083	<u>CURVE 1</u>	1.6	0.0295	<u>CURVE 3</u> (cont.)	1.5	0.0355	<u>CURVE 6</u> (cont.)			<u>CURVE 9</u>			<u>CURVE 12</u>			<u>CURVE 15</u>			<u>CURVE 18</u>			<u>CURVE 19</u>			<u>CURVE 20</u>	0.718	0.0296	<u>CURVE 16</u> (cont.)	0.718	0.0296	<u>CURVE 19</u>	0.718	0.0296	<u>CURVE 17</u>	0.718	0.0296	<u>CURVE 18</u>	0.718	0.0296	<u>CURVE 20</u>	0.718	0.0296	<u>CURVE 20</u>	0.718	0.0296	<u>CURVE 20</u>												
1.3	0.088	<u>CURVE 1</u>	1.9	0.037	<u>CURVE 3</u> (cont.)	1.7	0.0345	<u>CURVE 6</u> (cont.)			<u>CURVE 9</u>			<u>CURVE 12</u>			<u>CURVE 15</u>			<u>CURVE 18</u>			<u>CURVE 19</u>			<u>CURVE 20</u>	0.718	0.0296	<u>CURVE 16</u> (cont.)	0.718	0.0296	<u>CURVE 19</u>	0.718	0.0296	<u>CURVE 17</u>	0.718	0.0296	<u>CURVE 18</u>	0.718	0.0296	<u>CURVE 20</u>	0.718	0.0296	<u>CURVE 20</u>	0.718	0.0296	<u>CURVE 20</u>												
1.4	0.085	<u>CURVE 1</u>	2.1	0.0385	<u>CURVE 3</u> (cont.)	1.8	0.0311	<u>CURVE 6</u> (cont.)			<u>CURVE 9</u>			<u>CURVE 12</u>			<u>CURVE 15</u>			<u>CURVE 18</u>			<u>CURVE 19</u>			<u>CURVE 20</u>	0.718	0.0296	<u>CURVE 16</u> (cont.)	0.718	0.0296	<u>CURVE 19</u>	0.718	0.0296	<u>CURVE 17</u>	0.718	0.0296	<u>CURVE 18</u>	0.718	0.0296	<u>CURVE 20</u>	0.718	0.0296	<u>CURVE 20</u>	0.718	0.0296	<u>CURVE 20</u>												
1.6	0.100	<u>CURVE 1</u>			<u>CURVE 3</u> (cont.)	2.0	0.033	<u>CURVE 6</u> (cont.)			<u>CURVE 9</u>			<u>CURVE 12</u>			<u>CURVE 15</u>			<u>CURVE 18</u>			<u>CURVE 19</u>			<u>CURVE 20</u>	0.718	0.0296	<u>CURVE 16</u> (cont.)	0.718	0.0296	<u>CURVE 19</u>	0.718	0.0296	<u>CURVE 17</u>	0.718	0.0296	<u>CURVE 18</u>	0.718	0.0296	<u>CURVE 20</u>	0.718	0.0296	<u>CURVE 20</u>	0.718	0.0296	<u>CURVE 20</u>												
1.7	0.111	<u>CURVE 1</u>			<u>CURVE 3</u> (cont.)	2.3	0.037	<u>CURVE 6</u> (cont.)			<u>CURVE 9</u>			<u>CURVE 12</u>			<u>CURVE 15</u>			<u>CURVE 18</u>			<u>CURVE 19</u>			<u>CURVE 20</u>	0.718	0.0296	<u>CURVE 16</u> (cont.)	0.718	0.0296	<u>CURVE 19</u>	0.718	0.0296	<u>CURVE 17</u>	0.718	0.0296	<u>CURVE 18</u>	0.718	0.0296	<u>CURVE 20</u>	0.718	0.0296	<u>CURVE 20</u>	0.718	0.0296	<u>CURVE 20</u>												
1.8	0.125	<u>CURVE 1</u>			<u>CURVE 3</u> (cont.)	2.5	0.0415	<u>CURVE 6</u> (cont.)			<u>CURVE 9</u>			<u>CURVE 12</u>			<u>CURVE 15</u>			<u>CURVE 18</u>			<u>CURVE 19</u>			<u>CURVE 20</u>	0.718	0.0296	<u>CURVE 16</u> (cont.)	0.718	0.0296	<u>CURVE 19</u>	0.718	0.0296	<u>CURVE 17</u>	0.718	0.0296	<u>CURVE 18</u>	0.718	0.0296	<u>CURVE 20</u>	0.718	0.0296	<u>CURVE 20</u>	0.718	0.0296	<u>CURVE 20</u>												
2.0	0.142	<u>CURVE 1</u>			<u>CURVE 3</u> (cont.)	2.9	0.0455	<u>CURVE 6</u> (cont.)			<u>CURVE 9</u>			<u>CURVE 12</u>			<u>CURVE 15</u>			<u>CURVE 18</u>			<u>CURVE 19</u>			<u>CURVE 20</u>	0.718	0.0296	<u>CURVE 16</u> (cont.)	0.718	0.0296	<u>CURVE 19</u>	0.718	0.0296	<u>CURVE 17</u>	0.718	0.0296	<u>CURVE 18</u>	0.718	0.0296	<u>CURVE 20</u>	0.718	0.0296	<u>CURVE 20</u>	0.718	0.0296	<u>CURVE 20</u>												
2.1	0.151	<u>CURVE 1</u>			<u>CURVE 3</u> (cont.)	3.3	0.056	<u>CURVE 6</u> (cont.)			<u>CURVE 9</u>			<u>CURVE 12</u>			<u>CURVE 15</u>			<u>CURVE 18</u>			<u>CURVE 19</u>			<u>CURVE 20</u>	0.718	0.0296	<u>CURVE 16</u> (cont.)	0.718	0.0296	<u>CURVE 19</u>	0.718	0.0296	<u>CURVE 17</u>	0.718	0.0296	<u>CURVE 18</u>	0.718	0.0296	<u>CURVE 20</u>	0.718	0.0296	<u>CURVE 20</u>	0.718	0.0296	<u>CURVE 20</u>												
2.2	0.176	<u>CURVE 1</u>			<u>CURVE 3</u> (cont.)	3.7	0.069	<u>CURVE 6</u> (cont.)			<u>CURVE 9</u>			<u>CURVE 12</u>			<u>CURVE 15</u>			<u>CURVE 18</u>			<u>CURVE 19</u>			<u>CURVE 20</u>	0.718	0.0296	<u>CURVE 16</u> (cont.)	0.718	0.0296	<u>CURVE 19</u>	0.718	0.0296	<u>CURVE 17</u>	0.718	0.0296	<u>CURVE 18</u>	0.718	0.0296	<u>CURVE 20</u>	0.718	0.0296	<u>CURVE 20</u>	0.718	0.0296	<u>CURVE 20</u>												
2.2	0.195	<u>CURVE 1</u>			<u>CURVE 3</u> (cont.)	4.0	0.075	<u>CURVE 6</u> (cont.)			<u>CURVE 9</u>			<u>CURVE 12</u>			<u>CURVE 15</u>			<u>CURVE 18</u>			<u>CURVE 19</u>			<u>CURVE 20</u>	0.718	0.0296	<u>CURVE 16</u> (cont.)	0.718	0.0296	<u>CURVE 19</u>	0.718	0.0296	<u>CURVE 17</u>	0.718	0.0296	<u>CURVE 18</u>	0.718	0.0296	<u>CURVE 20</u>	0.718	0.0296	<u>CURVE 20</u>	0.718	0.0296	<u>CURVE 20</u>												
2.3	0.263	<u>CURVE 1</u>			<u>CURVE 3</u> (cont.)	4.0	0.075	<u>CURVE 6</u> (cont.)			<u>CURVE 9</u>			<u>CURVE 12</u>			<u>CURVE 15</u>			<u>CURVE 18</u>			<u>CURVE 19</u>			<u>CURVE 20</u>	0.718	0.0296	<u>CURVE 16</u> (cont.)	0.718	0.0296	<u>CURVE 19</u>	0.718	0.0296	<u>CURVE 17</u>	0.718	0.0296	<u>CURVE 18</u>	0.718	0.0296	<u>CURVE 20</u>	0.718	0.0296	<u>CURVE 20</u>	0.718	0.0296	<u>CURVE 20</u>												
2.7	0.275	<u>CURVE 1</u>			<u>CURVE 3</u> (cont.)	4.0	0.075	<u>CURVE 6</u> (cont.)			<u>CURVE 9</u>			<u>CURVE 12</u>			<u>CURVE 15</u>			<u>CURVE 18</u>			<u>CURVE 19</u>			<u>CURVE 20</u>	0.718	0.0296	<u>CURVE 16</u> (cont.)	0.718	0.0296	<u>CURVE 19</u>	0.718	0.0296	<u>CURVE 17</u>	0.718	0.0296	<u>CURVE 18</u>	0.718	0.0296	<u>CURVE 20</u>	0.718	0.0296	<u>CURVE 20</u>	0.718	0.0296	<u>CURVE 20</u>												
2.7	0.292	<u>CURVE 1</u>			<u>CURVE 3</u> (cont.)	4.0	0.075	<u>CURVE 6</u> (cont.)			<u>CURVE 9</u>			<u>CURVE 12</u>			<u>CURVE 15</u>			<u>CURVE 18</u>			<u>CURVE 19</u>			<u>CURVE 20</u>	0.718	0.0296	<u>CURVE 16</u> (cont.)	0.718	0.0296	<u>CURVE 19</u>	0.718	0.0296	<u>CURVE 17</u>	0.718	0.0296	<u>CURVE 18</u>	0.718	0.0296	<u>CURVE 20</u>	0.718	0.0296	<u>CURVE 20</u>	0.718	0.0296	<u>CURVE 20</u>												
2.8	0.335	<u>CURVE 1</u>			<u>CURVE 3</u> (cont.)	4.0	0.075	<u>CURVE 6</u> (cont.)			<u>CURVE 9</u>			<u>CURVE 12</u>			<u>CURVE 15</u>			<u>CURVE 18</u>			<u>CURVE 19</u>			<u>CURVE 20</u>	0.718	0.0296	<u>CURVE 16</u> (cont.)	0.718	0.0296	<u>CURVE 19</u>	0.718	0.0296	<u>CURVE 17</u>	0.718	0.0296	<u>CURVE 18</u>	0.718	0.0296	<u>CURVE 20</u>	0.718	0.0296	<u>CURVE 20</u>	0.718	0.0296	<u>CURVE 20</u>												
3.3	0.48	<u>CURVE 1</u>			<u>CURVE 3</u> (cont.)	4.0	0.075	<u>CURVE 6</u> (cont.)			<u>CURVE 9</u>			<u>CURVE 12</u>			<u>CURVE 15</u>			<u>CURVE 18</u>			<u>CURVE 19</u>			<u>CURVE 20</u>	0.718	0.0296	<u>CURVE 16</u> (cont.)	0.718	0.0296	<u>CURVE 19</u>	0.718	0.0296	<u>CURVE 17</u>	0.718	0.0296	<u>CURVE 18</u>	0.718	0.0296	<u>CURVE 20</u>	0.718	0.0296	<u>CURVE</u>															

DATA TABLE NO. 21 (continued)

T	k	T	k
<u>CURVE 20 (cont.)</u>			
0.705	0.02	0.148	0.000930
0.75	0.0238	0.215	0.00150
0.822	0.0263	0.345	0.00385
0.852	0.0284	0.345	0.00215
0.90	0.0295	0.385	0.00425
0.998	0.0322	0.435	0.00420
<u>CURVE 21</u>			
0.39	0.00325	0.550	0.00600
0.42	0.0041	0.580	0.00730
0.46	0.00525	0.640	0.00730
0.46	0.00545	0.700	0.00960
0.45	0.00630	1.150	0.0220
0.508	0.00745	<u>CURVE 25</u>	
0.547	0.0085	2.34	0.0381
0.602	0.0111	2.40	0.0586
0.652	0.0135	2.80	0.0452
0.71	0.0163	2.83	0.0466
0.80	0.0260	3.29	0.0544
0.86	0.0244	3.40	0.0551
0.948	0.0290	3.54	0.0586
1.05	0.0315	3.60	0.0600
1.20	0.035	3.69	0.0607
<u>CURVE 22</u>			
0.225	0.00135	3.80	0.0642
0.375	0.00230	3.89	0.0657
0.500	0.00555	3.97	0.0667
0.570	0.00490	4.03	0.0681
1.150	0.0150	4.23	0.0726
<u>CURVE 23</u>			
0.150	0.00100		
0.220	0.00073		
0.270	0.00123		
0.320	0.00175		
0.370	0.00205		
0.420	0.00250		
0.470	0.00300		
0.520	0.00400		

# THERMAL CONDUCTIVITY OF TIN + LEAD ALLOYS

(Sn + Pb ≥ 99.50%; impurity ≤ 0.20% each)

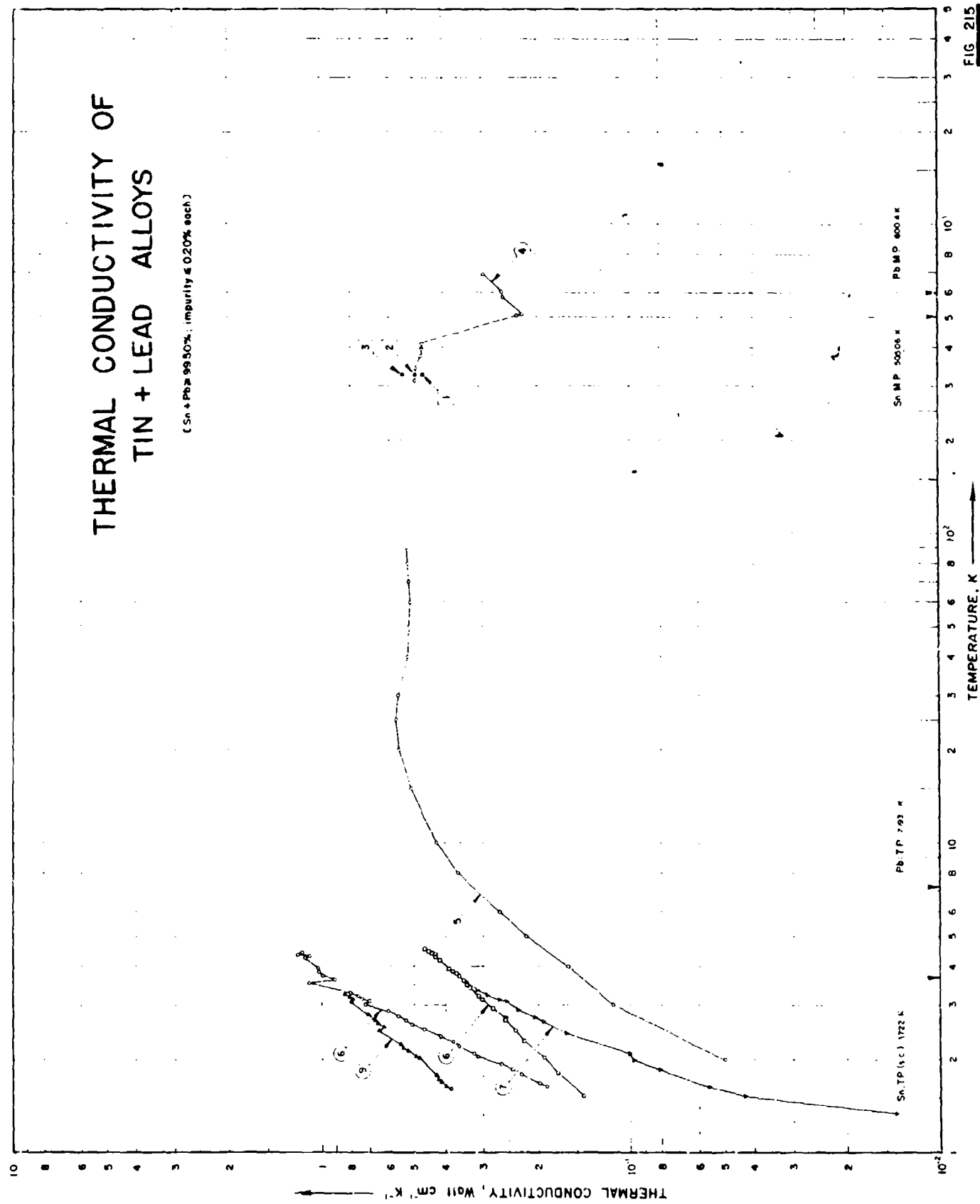


FIG. 215

## SPECIFICATION TABLE NO. 215 THERMAL CONDUCTIVITY OF TIN-LEAD ALLOYS

(Sn + Pb 99.50%; impurity 0.20% each)

[For Data Reported in Figure and Table No. 215.]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) Sn	Pb	Composition (continued), Specifications and Remarks
1	230	L	1925	327.2			50	50	Approx. composition: 0.03 total impurity in each metal; specimen 1.9 cm in dia and 10 cm long; supplied by Baker; electrical conductivity $6.47 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 22 C.
2	230	L	1925	327.2			60	40	Similar to the above specimen except electrical conductivity $6.92 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 22 C.
3	230	L	1925	327.2			80	20	Similar to the above specimen except electrical conductivity $7.62 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 22 C.
4	19	L	1923	313-693			62	38	Specimen 1.5 cm in dia and 12 cm long; melting point 180 C.
5	229	L	1955	2.0-90		Soft solder	60	40	No other details reported.
6	837	L	1967	1.5-4.6	1	Pb 3	98.50	1.70	Prepared by vacuum-melting appropriate amounts of Johnson-Matthey 99.999 pure Sn and Pb, extruding into 1.5 mm dia wire, annealed at ~200 C for several days; electrical resistivity 0.00478 and 13.55 $\mu\text{ohm cm}$ at 4.2 and 273 K, respectively. $T_c$ 3.752 K; normal-state data were taken at temperatures below $T_c$ with a longitudinal magnetic field applied to the sample.
7	837	L	1967	1.3-3.6	1	Pb 3	98.30	1.70	Same as the above specimen, except the magnetic field was removed so the superconducting-state data were taken.
8	936	L	1958	1.6-4.4	4-5	7	99.54	0.46	Prepared by vacuum-melting appropriate amounts of Johnson-Matthey 99.999 pure Sn and Pb, casting into 1 mm dia x 12 cm long wire with pyrex capillary; electrical resistivity 0.112 $\mu\text{ohm cm}$ ; a magnetic field was applied when taking normal-state data at temperatures below $T_c$ .
9	856	L	1958	1.6-3.3	4-5	7	99.54	0.46	Same as the above specimen except the magnetic field was removed so the superconducting-state data were taken.



DATA TABLE NO. 215 THERMAL CONDUCTIVITY OF (TN + LEAD) ALLOYS

(Sn + Pb 89.50%; impurity : 0.20% each)

(Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup>K<sup>-1</sup>)

T	k	T	k	T	k	T	k	T	k
<u>CURVE 1</u>		<u>CURVE 6</u>		<u>CURVE 7 (cont.)</u>		<u>CURVE 9 (cont.)</u>			
327.2	0.464	1.537	0.142	3.124	0.265	1.690	0.413		
<u>CURVE 2</u>		1.815	0.172	3.242	0.289	1.723	0.420		
2.030	0.131	2.030	0.131	3.367	0.312	1.770	0.425		
2.299	0.222	2.299	0.222	3.518	0.340*	2.021	0.479		
2.488	0.237	2.488	0.237	3.565	0.340	2.050	0.494		
2.677	0.255	2.677	0.255	<u>CURVE 8</u>					
2.716	0.255	2.716	0.255	1.635	0.188	2.184	0.522		
2.923	0.278	2.923	0.278	1.683	0.198	2.248	0.544		
3.116	0.300	3.116	0.300	1.794	0.227	2.470	0.652		
3.201	0.308	3.201	0.308	1.851	0.242	2.547	0.632		
3.439	0.331	3.439	0.331	1.934	0.263	2.616	0.658		
3.491	0.339	3.491	0.339	2.043	0.311	2.682	0.672		
3.601	0.347	3.601	0.347	2.094	0.320	2.778	0.702		
3.728	0.360	3.728	0.360	2.211	0.360	3.008	0.726*		
3.795	0.367	3.795	0.367	2.285	0.378	3.127	0.791		
3.868	0.379	3.868	0.379	2.373	0.415	3.186	0.807		
3.972	0.468	3.972	0.468	2.501	0.464	3.265	0.844		
4.122	0.467	4.122	0.467	2.593	0.507				
509.2	0.231	4.193	0.415	2.677	0.532				
512.2	0.223	4.264	0.415*	2.762	0.567				
583.2	0.255	4.292	0.426	2.872	0.613				
609.2	0.257	4.337	0.423*	3.016	0.719				
630.2	0.297	4.372	0.427	3.080	0.698				
<u>CURVE 5</u>		4.419	0.437	3.137	0.736				
2	0.050	4.482	0.449	3.203	0.757				
3	0.115	4.537	0.454*	3.283	0.800				
4	0.160	4.578	0.462	3.545	1.009				
5	0.220	4.619	0.469*	3.633	0.908				
6	0.265	4.620	0.457*	3.729	0.992				
8	0.365	<u>CURVE 7</u>							
10	0.425	1.346	0.014	3.842	1.016				
15	0.510	1.526	0.043	3.967	1.039				
20	0.560	1.636	0.056	4.266	1.130				
25	0.575	1.856	0.081	4.322	1.098				
30	0.565	1.997	0.098	4.358	1.191				
40	0.525	2.093	0.101	4.406	1.176*				
50	0.520	2.445	0.162	4.440	1.151				
60	0.515	2.657	0.152	<u>CURVE 9</u>					
70	0.520	2.745	0.205	1.604	0.383				
80	0.525	2.894	0.233	1.630	0.397				
90	0.530	3.076	0.252						

\* Not shown on plot

# THERMAL CONDUCTIVITY OF TIN+MERCURY ALLOYS

[Sn + Hg > 99.50%, impurity ≤ 0.20% each]

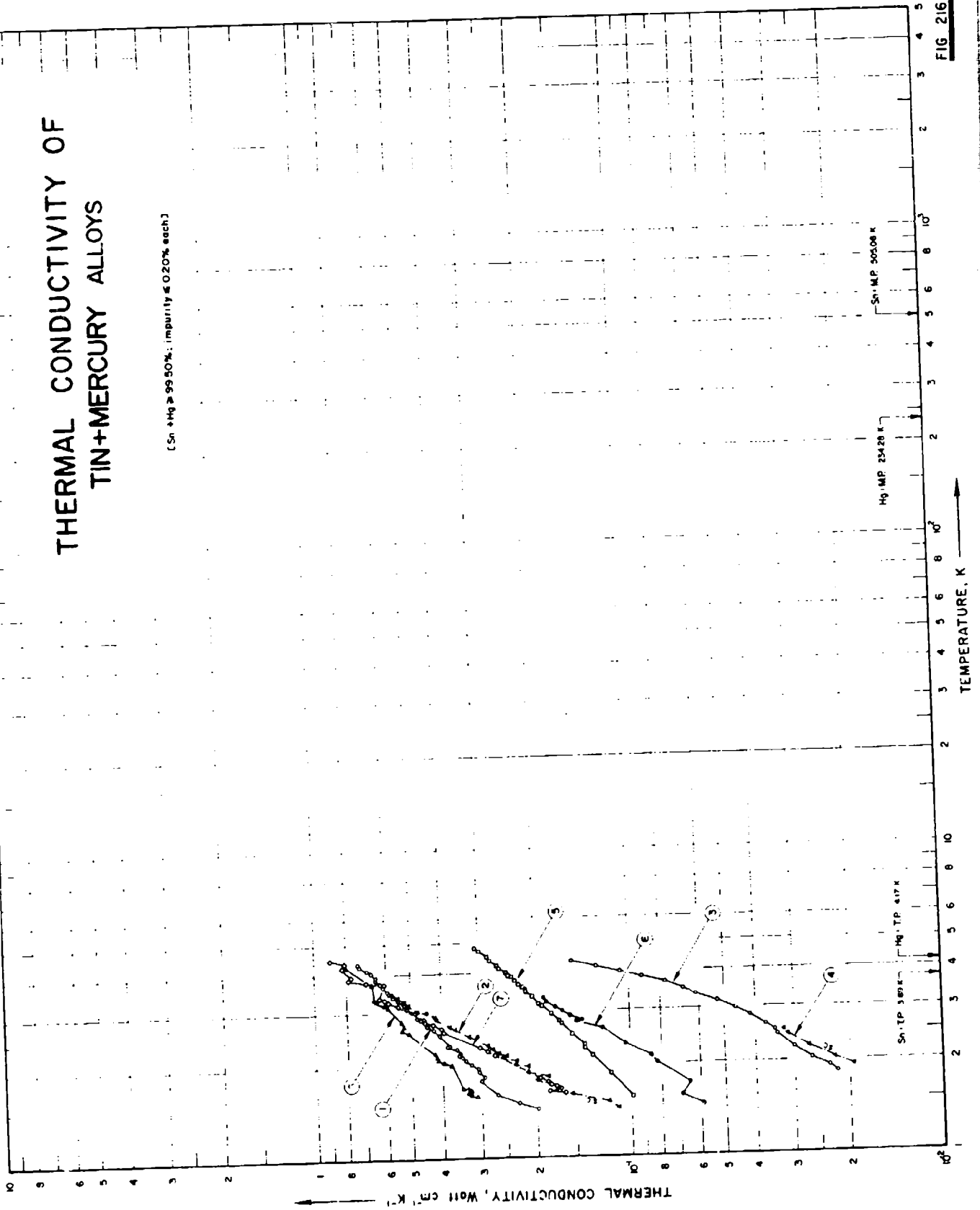


FIG 216

## SPECIFICATION TABLE NO. 216 THERMAL CONDUCTIVITY OF [TIN + MERCURY] ALLOYS

(Sn + Hg = 99.50%; impurity &lt; 0.20% each)

[ For Data Reported in Figure and Table No. 216 ]

Curve No.	Rel. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) Sn	Hg	Composition (continued), Specifications and Remarks
1	74	L	1950	1.4-4.3	3.0	Sn 6	99.67	0.33	Homogeneous solid solution with few large crystals supplied by Johnson-Matthey (J. 2356); in normal state, measured in a longitudinal magnetic field.
2	74	L	1950	1.4-3.5	3.0	Sn 6	99.67	0.33	The above specimen in superconducting state.
3	74	L	1950	1.8-4.3	3.0	Sn 9	95.9	4.1	Specimen in two phase state with few large crystals; supplied by Johnson-Matthey (J. 2356); in normal state; measured in a longitudinal magnetic field.
4	74	L	1950	1.9-2.5	3.0	Sn 9	95.9	4.1	The above specimen in superconducting state.
5	837	L	1967	1.6-4.8	1	Hg 3	98.15	1.85	Prepared by vacuum-melting appropriate amounts of Johnson-Matthey 99.999 pure Sn and Hg, extruding into 1.5 mm dia wire, annealed at -200 C for several days; electrical resistivity 0.475 and 13.35 $\mu\text{ohm cm}$ at 4.2 and 273 K, respectively, $T_c$ 3.646 K; normal-state data were taken at temperatures below $T_c$ with a longitudinal magnetic field applied at the sample.
6	837	L	1967	1.5-3.5	1	Hg 3	98.15	1.85	Same as the above specimen, except the magnetic field was removed so the superconducting-state data were taken.
7	836	L	1958	1.6-4.4	4-5	1	99.47	0.53	Prepared by vacuum-melting appropriate amounts of Johnson-Matthey 99.999 pure Sn and Hg, casting into 1 mm dia x 12 cm long wire with pyrex capillary; electrical resistivity 0.17 $\mu\text{ohm cm}$ ; a magnetic field was applied when taking normal-state data at temperatures below $T_c$ .
8	836	L	1958	1.6-4.3	4-5	1	99.47	0.53	Same as the above specimen except the magnetic field was removed so the superconducting-state data were taken.

## DATA TABLE NO. 216 THERMAL CONDUCTIVITY OF [TIN + MERCURY] ALLOYS

(Sn + Hg = 99.50%; Impurity  $\pm 0.20\%$  each)[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup>K<sup>-1</sup>]

T	k	T	k	T	k	T	k	T	k
<u>CURVE 1</u>		<u>CURVE 2 (cont.)</u>		<u>CURVE 4</u>		<u>CURVE 6</u>		<u>CURVE 7 (cont.)</u>	
1.44	0.198	1.84	0.183	1.92	0.0195	1.46	0.059	3.30	0.809
1.51	0.229	1.90	0.213	2.04	0.0223	1.58	0.069	3.35	0.628
1.61	0.267	1.98	0.212	2.21	0.0270	1.72	0.065	3.71	0.509
1.79	0.300	2.01	0.233	2.44	0.0315	2.02	0.083	3.78	0.590
1.84	0.291	2.08	0.231	2.51	0.0325	2.14	0.086	3.86	0.784
1.91	0.305	2.12	0.255			2.33	0.104	3.94	0.770
1.97	0.306	2.16	0.249	<u>CURVE 5</u>		2.36	0.105*	4.23	0.921
2.02	0.322	2.20	0.264	1.56	0.099	2.61	0.122*	4.35	0.805
2.08	0.335	2.27	0.275	1.86	0.116	2.62	0.128*	4.45	0.394
2.13	0.343	2.34	0.285	1.86	0.116	2.77	0.148	<u>CURVE 8</u>	
2.16	0.352	2.50	0.327	2.15	0.132	2.79	0.142	1.58	0.309
2.20	0.348	2.63	0.363	2.26	0.140	2.89	0.155	1.61	0.324
2.27	0.359	2.69	0.376	2.32	0.140	3.00	0.164	1.65	0.331
2.32	0.355	2.83	0.415	2.51	0.153	3.06	0.172	1.67	0.324
2.33	0.376	2.82	0.420	2.74	0.165	3.21	0.184	1.69	0.331
2.51	0.403	3.00	0.446	2.77	0.168	3.21	0.186*	1.69	0.344
2.63	0.424	3.08	0.475	2.94	0.178	3.31	0.187	1.72	0.344*
2.70	0.445	3.18	0.508	3.11	0.190	3.54	0.215*	2.02	0.372
2.83	0.463	3.27	0.528	3.17	0.195			2.06	0.396
2.88	0.478	3.37	0.550	3.35	0.204			2.10	0.414
3.00	0.483	3.43	0.571	3.43	0.214	<u>CURVE 7</u>		2.15	0.415
3.08	0.516	3.52	0.516	3.52	0.215	1.62	0.163	2.20	0.421
3.18	0.525	3.54	0.218*	3.54	0.218*	1.65	0.182	2.57	0.507
3.27	0.550	3.58	0.219	3.58	0.219	1.67	0.167	2.63	0.535
3.37	0.560	3.63	0.222	3.63	0.222	1.70	0.174	2.70	0.527
3.46	0.588	3.66	0.226	3.66	0.226	1.72	0.180	2.75	0.522*
3.54	0.593	3.77	0.231	3.77	0.231	1.81	0.195	2.79	0.538
3.65	0.612	3.87	0.238	3.87	0.238	1.84	0.192*	3.11	0.592
3.73	0.622	3.94	0.244	3.94	0.244	2.05	0.234*	3.17	0.608
3.84	0.648	4.00	0.248	4.00	0.248	2.11	0.253*	3.25	0.654*
3.92	0.646	4.15	0.259	4.15	0.259	2.24	0.285	3.28	0.645
4.03	0.664	4.24	0.263	4.24	0.263	2.30	0.301	3.32	0.654
4.12	0.682	4.28	0.266	4.28	0.266	2.58	0.395	3.72	0.669
4.24	0.721	4.39	0.276	4.39	0.276	2.66	0.400	4.26	0.800
4.32	0.730	4.46	0.281	4.46	0.281	2.75	0.425		
		4.55	0.289*	4.55	0.289*	2.79	0.450		
<u>CURVE 2</u>		4.71	0.298	4.71	0.298	3.00	0.301		
1.44	0.110	3.67	0.0770	4.71	0.298	2.30	0.301		
1.51	0.118	3.82	0.0910	4.75	0.299*	2.58	0.395		
1.60	0.155	3.98	0.107	4.83	0.307	2.66	0.400		
1.70	0.190	4.15	0.127			2.75	0.425		
		4.32	0.152			2.79	0.450		

Not shown on plot

SPECIFICATION TABLE NO. 217 THERMAL CONDUCTIVITY OF (TIN + SILVER) ALLOYS

(Sn + Ag : 99.50%; impurity  $\leq$  0.20% each)

Curve No.	Ref. Method No. Used	Year	Temp Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)		Composition (continued), Specifications and Remarks
						Sn	Ag	
1	230	L	1925	333.2		60.0	40.0	Impurities $\leq$ 0.03.
2	230	L	1925	333.2		70.0	30.0	Impurities $\leq$ 0.03.
3	230	L	1925	333.2		80.0	20.0	Impurities $\leq$ 0.03.
4	230	L	1925	333.2		90.0	10.0	Impurities $\leq$ 0.03.

DATA TABLE NO. 217 THERMAL CONDUCTIVITY OF (TIN + SILVER) ALLOYS

(Sn + Ag : 99.50%; impurity  $\leq$  0.20% each){ Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup>K<sup>-1</sup>}

T	k
<u>CURVE 1*</u>	
333.2	0.611
<u>CURVE 2*</u>	
333.2	0.611
<u>CURVE 3*</u>	
333.2	0.611
<u>CURVE 4*</u>	
333.2	0.603

\* No graphical presentation

SPECIFICATION TABLE NO. 214 THERMAL CONDUCTIVITY OF [TIN + THALLIUM] ALLOYS

(Sn + Tl = 99.50%; impurity  $\leq$  0.20% each)

Curve No.	Ref. Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)		Composition (continued), Specifications and Remarks
						Sn	Tl	
1	230 L	1925	336.2			53.8	46.2	Impurities $\leq$ 0.03.
2	230 L	1925	336.2			60.0	40.0	Impurities $\leq$ 0.03.
3	230 L	1925	336.2			70.0	30.0	Impurities $\leq$ 0.03.
4	230 L	1925	336.2			80.0	20.0	Impurities $\leq$ 0.03.
5	230 L	1925	336.2			90.0	10.0	Impurities $\leq$ 0.03.

DATA TABLE NO. 215 THERMAL CONDUCTIVITY OF [TIN + THALLIUM] ALLOYS

(Sn + Tl = 99.50%; impurity  $\leq$  0.20% each)[Temperature, T, K; Thermal Conductivity, k, Watt  $\text{cm}^{-1}\text{K}^{-1}$ ]

T	k
<u>CURVE 1*</u>	
336.2	0.385
<u>CURVE 2*</u>	
336.2	0.418
<u>CURVE 3*</u>	
336.2	0.435
<u>CURVE 4*</u>	
336.2	0.485
<u>CURVE 5*</u>	
336.2	0.557

\* No graphical presentation

SPECIFICATION TABLE NO. 219 THERMAL CONDUCTIVITY OF [TIN + ZINC] ALLOYS

(Sn + Zn = 99.50%; impurity  $\leq$  0.20% each)

Curve No.	Ref. Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) Sn	Zn	Composition (continued), Specifications and Remarks
1	19	1	1923	508-706		92.0	8.0	

DATA TABLE NO. 219 THERMAL CONDUCTIVITY OF [TIN + ZINC] ALLOYS

(Sn + Zn = 99.50%; impurity  $\leq$  0.20% each)

[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup>K<sup>-1</sup>]

T	k
308.2	0.596
337.2	0.615
364.2	0.617
388.2	0.626
421.2	0.630
488.2	0.238
616.2	0.306
706.2	0.367

CURVE 1<sup>0</sup>

No graphical presentation

SPECIFICATION TABLE NO. 220 THERMAL CONDUCTIVITY OF [ TITANIUM + ALUMINUM ] ALLOYS

(Ti + Al ~ 99.50%; impurity  $\leq$  0.2 % each)

Curve No.	Ref. Method No. Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)		Composition (continued), Specifications and Remarks
						Ti	Al	
1	554	1954	317.4			92	8	Melted in a nonconsumable arc furnace.
2	555 C	1956	403.2			93	7	

DATA TABLE NO. 220 THERMAL CONDUCTIVITY OF [ TITANIUM + ALUMINUM ] ALLOYS

(Ti + Al ~ 99.50%; impurity  $\leq$  0.20% each)

[ Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup>K<sup>-1</sup> ]

T	k
<u>CURVE 1</u>	
317.4	0.092
<u>CURVE 2</u>	
403.2	0.0893

No graphical presentation



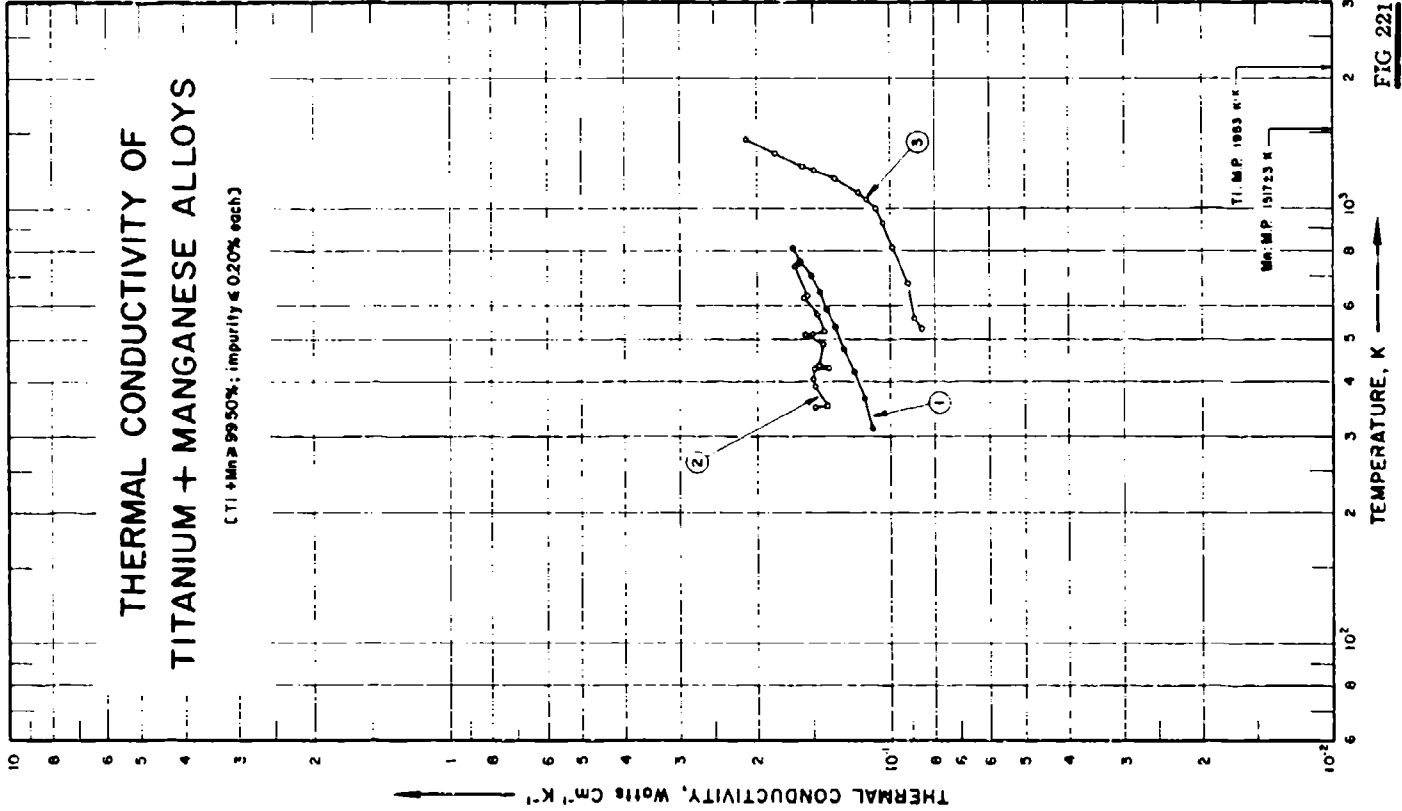


FIG 221

## SPECIFICATION TABLE NO. 221 THERMAL CONDUCTIVITY OF TITANIUM-MANGANESE ALLOYS

(Ti - Mn - 99.50%; impurity 0.20% each)

For Data Reported in Figure and Table No. 221.

Curve No.	Ref. No. Used	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)		Composition (continued), Specifications and Remarks
							Ti	Mn	
1	231	C	1958	311-811	<5	Ti-8Mn	92	8	Nominal composition; formerly designated as RC-130 A, in a mill-annealed condition; measured in vacuum of $\sim 2 \times 10^{-5}$ mm Hg; electrical resistivity 93, 100, 158, 115, 122, 128, 135, 141, 146 and 151 $\mu$ ohm cm at 311, 366, 422, 477, 533, 589, 644, 700, 755 and 811 C respectively. Armco ingot iron used as comparative material. 0.20 Fe, 0.177 O, 0.0063 H, 0.05 C and 0.034 N; specimen 3/4 in. in diameter.
2	340	L	1956	350-746	10	Ti-130A	93.21	6.5	
3	614	R	1961	534-1446	<5	Ti-8Mn	91.81	7.9	

## DATA TABLE NO. 221 THERMAL CONDUCTIVITY OF [TITANIUM - MANGANESE] ALLOYS

(Ti - Mn : 99.50%, impurity : 0.20%, each)

Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup>K<sup>-1</sup>;

T	k	T	k
<u>CURVE 1</u>			
<u>CURVE 3 (cont.)</u>			
311	0.110	1222.1	0.149
366	0.115	1254.3	0.158
422	0.122	1345.4	0.183
477	0.128	1445.9	0.211
531	0.134		
589	0.140		
644	0.145		
700	0.152		
755	0.160		
811	0.167		
<u>CURVE 2</u>			
349.8	0.148		
351.6	0.140		
390.1	0.148		
405.4	0.150		
429.0	0.149		
429.0	0.138		
436.8	0.146		
487.9	0.143		
517.3	0.156		
517.9	0.151		
521.2	0.142		
574.0	0.146		
626.2	0.158		
631.8	0.153		
737.9	0.165		
745.7	0.162		
<u>CURVE 3</u>			
533.7	0.0855		
560.9	0.0888		
674.8	0.0917		
813.7	0.0990		
928.2	0.104		
999.8	0.108		
1050.9	0.114		
1086.5	0.118		
1167.1	0.134		

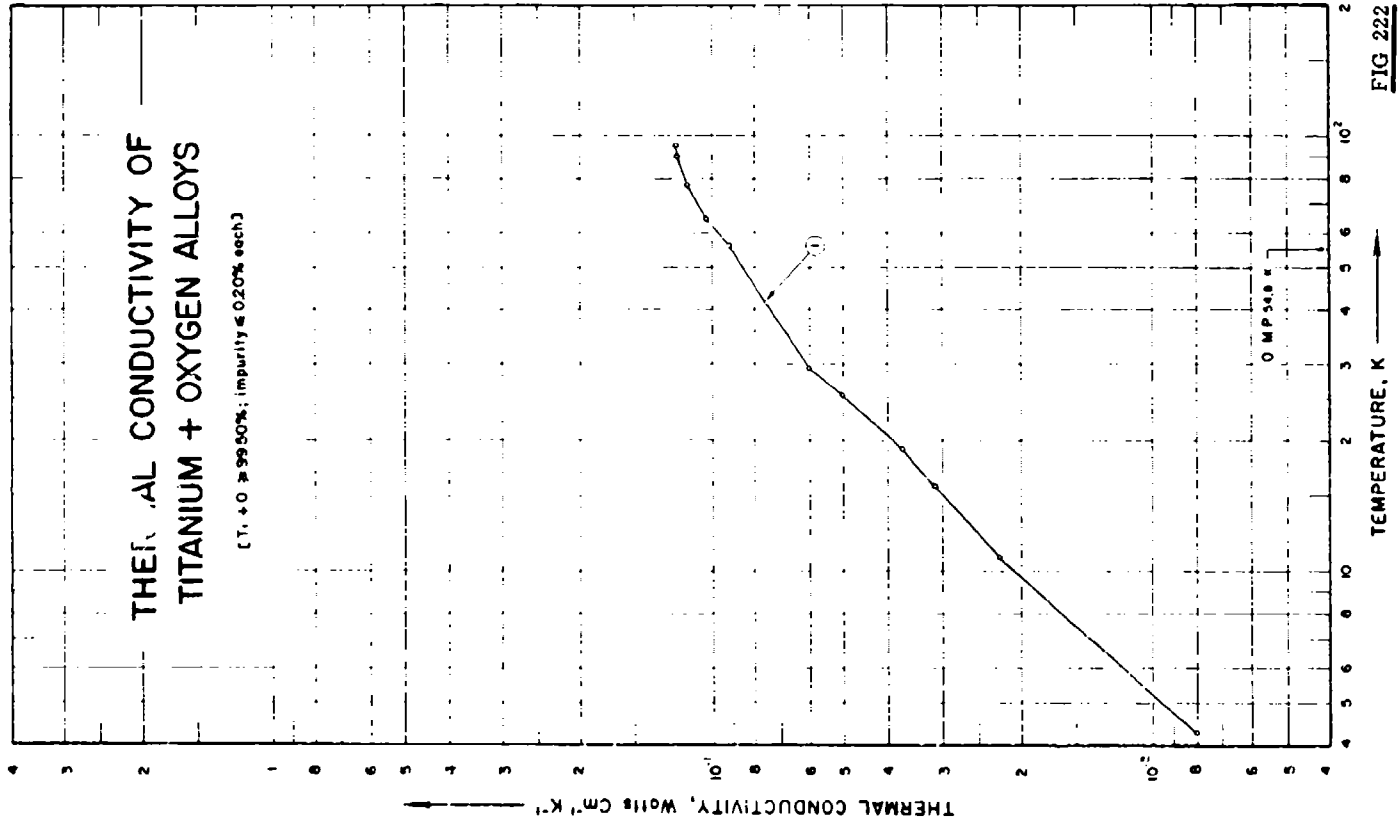


FIG 222

## SPECIFICATION TABLE NO. 222 THERMAL CONDUCTIVITY OF (TITANIUM - OXYGEN) ALLOYS

(Ti + O = 99.50 % ; impurity = 0.20 % each)

(For Data Reported in Figure and Table No. 222)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) Ti	Composition (weight percent) O	Composition (continued), Specifications and Remarks
1	83	L	1936	4.3-96		JM 4233	98	1.63	9.024 Mg, 0.13 Si, 0.05 Fe, 0.081 Ni, 0.14 C; Specimen 3 mm in dia, supplied by Messrs. Johnson-Matthey and Co., Ltd; annealed at 920 C for 5 hrs in vacuo; $\rho = 21.6 \mu\text{ohm cm}$ , electrical resistivity $70 \mu\text{ohm cm}$ at 293 K.

DATA TABLE NO. 222 THERMAL CONDUCTIVITY OF {TITANIUM - OXYGEN} ALLOYS

(Ti + O = 99.50%; impurity = 0.50% each)

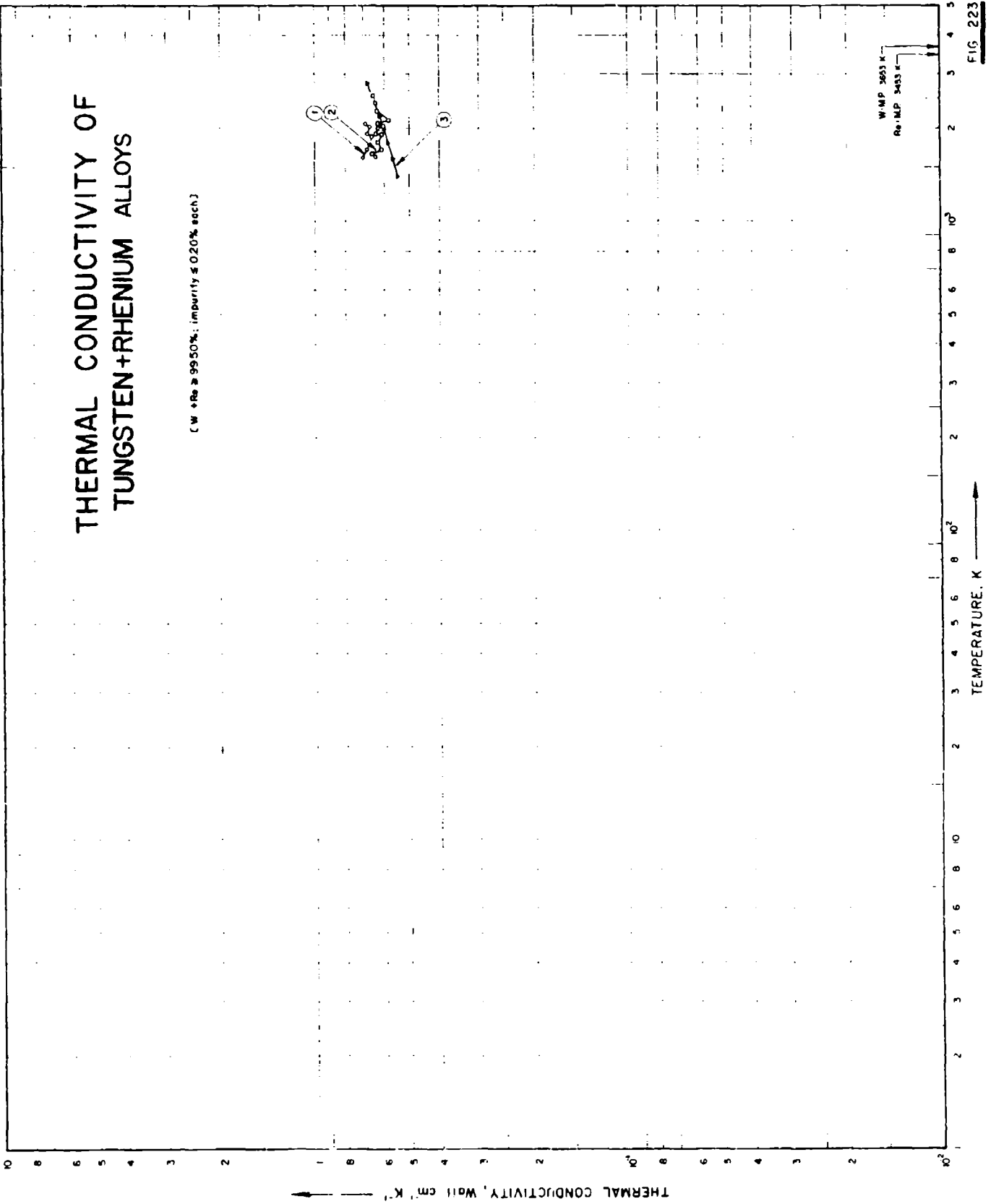
[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup>K<sup>-1</sup>]

T	k
4.30	0.0080
10.81	0.0225
15.75	0.0314
19.25	0.0370
25.65	0.0504
29.50	0.0606
56.40	0.0920
65.00	0.104
77.50	0.114
90.50	0.120
95.25	0.121

CURVE 1

# THERMAL CONDUCTIVITY OF TUNGSTEN+RHENIUM ALLOYS

(W + Re ≥ 99.50%; impurity ≤ 0.20% each)



SPECIFICATION TABLE NO. 223 THERMAL CONDUCTIVITY OF TUNGSTEN - RHENIUM ALLOYS

(W = Re = 99.50%; impurity = 0.50% each)

[For Data Reported in Figure and Table No. 223.]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) W	Re	Composition (continued), Specifications and Remarks
1	849	-	1966	1617-2074			Bal	25.04	0.0023 C, 0.0014 O, 0.001 Fe and 0.0065 N impurities; specimen 1.9126 cm in dia and 0.0766 cm long; avg grain size 0.216 mm dia; density 19.73 g cm <sup>-3</sup> ; thermal conductivity was derived from the temp distribution on the flat surface of the cylindrical disc specimen heated in high vacuum (10 <sup>-5</sup> mm Hg) by high frequency induction generating localized heating within 0.003 in. of the surface; at current frequency of 50000 cps with heat lost only by radiation; the cylindrical surface being assumed isothermal, and the temp gradient along the radius was analytically correlated to the thermal conductivity.
2	849	-	1966	1625-2553			Bal	24.5	0.0025 C, 0.0022 O, and 0.0002 B; cut from the same bar as the above specimen; specimen 1.9208 cm in dia and 0.2703 cm long; avg grain size 0.041 mm dia; density 19.19 g cm <sup>-3</sup> ; measuring method same as that for the above specimen.
3	928, 929	-	1966	1400-2800	± 12	VR-27-VT	73.0	27.0	Specimen made from an ingot subjected to rotary swaging 7.8 mm in dia and 65 mm long; ground to a surface-finish class of 8; annealed in vacuum at 2200 K for 2 hrs before the measurements; melting point 3300 K; measured in vacuum with electronic heating at 1-5 x 10 <sup>-5</sup> mm Hg; electrical resistivity 60.4, 67.3, 73.9, 79.9, 85.7, 91.1, 96.8, 102.8 and 109.2 x 10 <sup>-6</sup> ohm cm at 1200, 1400, 1600, 1800, 2000, 2200, 2400, 2600, and 2800 K, respectively.



## DATA TABLE NO. 223 THERMAL CONDUCTIVITY OF [TUNGSTEN + RHENIUM] ALLOYS

(W + Re) 99.56%; Impurity  $\leq 0.20\%$ Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup>K<sup>-1</sup>

T k

CURVE 1

1616.5	0.702
1710	0.682
1837	0.660
1928	0.680
2034	0.668
2074	0.657

CURVE 2

1625	0.640
1669	0.657
1720	0.612
1802.5	0.627
1804.5	0.637 <sup>a</sup>
1917	0.610
1925.5	0.638
2002.5	0.626
2043.5	0.603
2088	0.626
2097	0.623 <sup>a</sup>
2128	0.590
2283	0.631
2415	0.639
2553	0.648

CURVE 3

1490	0.544
1600	0.563
1900	0.582
2000	0.601
2100	0.620
2400	0.639 <sup>a</sup>
2600	0.658 <sup>a</sup>
2800	0.677

<sup>a</sup> Not shown on plot

## SPECIFICATION TABLE NO. 224 THERMAL CONDUCTIVITY OF [ URANIUM + ALUMINUM ] ALLOYS

(U + Al) ~ 99.50%; impurity  $\leq$  0.20% each)

Curve No.	Ref. Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)		Composition (continued), Specifications and Remarks
						U	Al	
1	591 C	1963	318.2	$\leq$ 1.6		58.4	Bal	Fe $<$ 0.1, Si $<$ 0.07, Ca $<$ 0.04, and B $<$ 0.02; prepared by the dissolution of reactor-grade Uranium ( $>$ 99.5 purity) in aluminum (99.99 purity) at approximately 100 C above the alloy liquidus temperatures and then cast in a graphite mould at 100 C.
2	591 C	1963	318.2	$\leq$ 4.0		58.4	Bal	Fe $<$ 0.1, Si $<$ 0.07, Ca $<$ 0.04, and B $<$ 0.02; same as the above; specimen except measured after heat-treated at 620 C for 5 days.

## DATA TABLE NO. 224 THERMAL CONDUCTIVITY OF [ URANIUM + ALUMINUM ] ALLOYS

(U + Al) ~ 99.50%; impurity  $\leq$  0.20% each)[ Temperature, T, K; Thermal Conductivities, k, Watt cm<sup>-1</sup>K<sup>-1</sup> ]

T	k	CURVE 1 <sup>a</sup>		CURVE 2 <sup>b</sup>	
		318.2	0.561	318.2	0.343
318.2	0.561			318.2	0.335
318.2	0.565				

No graphical presentation

# THERMAL CONDUCTIVITY OF URANIUM + CHROMIUM ALLOYS

[ U + Cr = 99.50% ; impurity ≤ 0.20% each ]

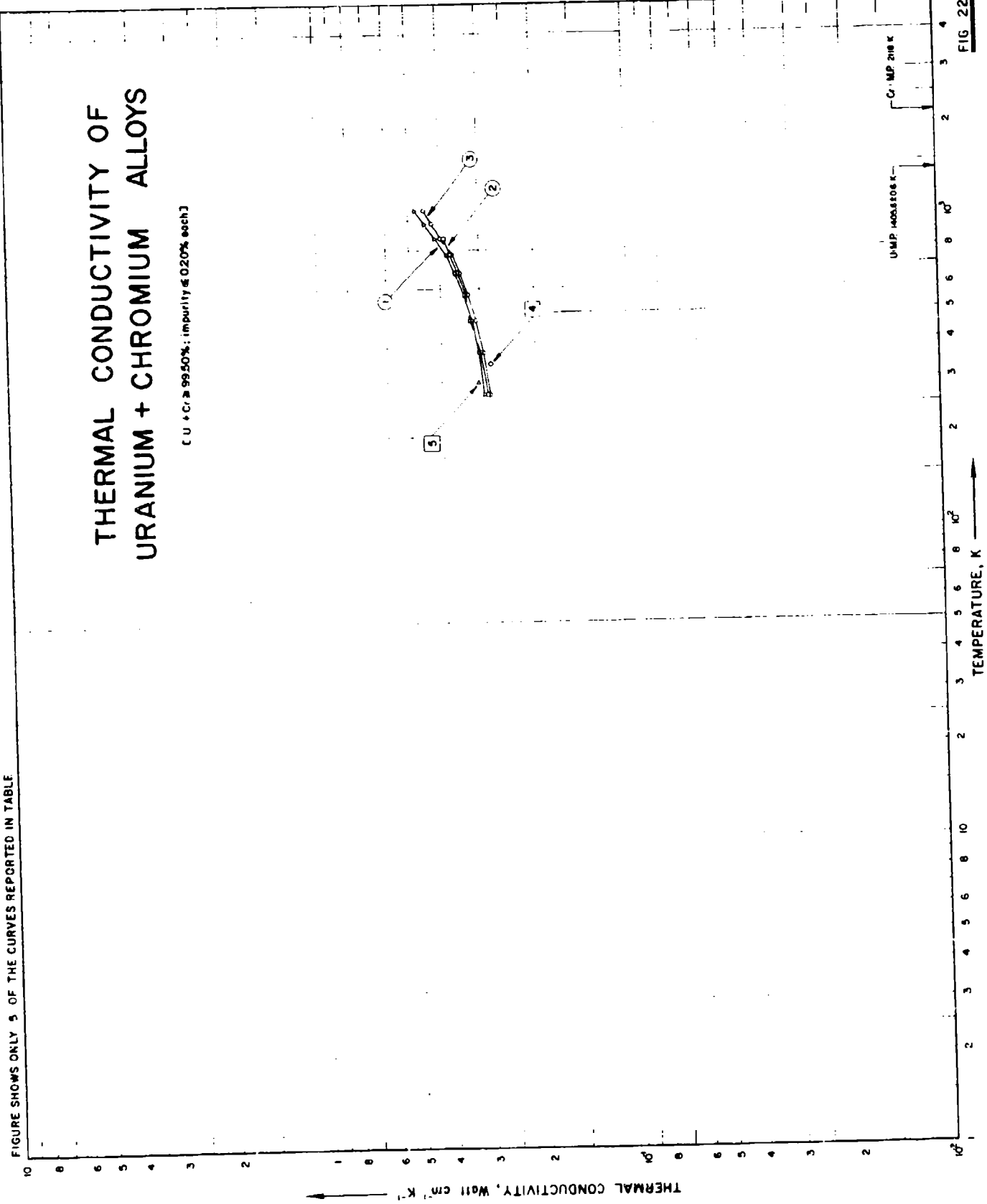


FIGURE SHOWS ONLY 5 OF THE CURVES REPORTED IN TABLE

SPECIFICATION TABLE NO. 225 THERMAL CONDUCTIVITY OF [URANIUM + CHROMIUM] ALLOYS  
(U + Cr : 99.50%, impurity : 0.20% each)

[ For Data Reported in Figure and Table No. 225 ]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition U	Composition (weight percent) Cr	Composition (continued), Specifications and Remarks
1	269	C	1954	273-1073	± 5	86	94.8	5.2	Biscuit U, specimen 2 cm in diameter and 15 cm long; eutectic; cast in cold graphite; measured in vacuum ~ 5 x 10 <sup>-5</sup> mm Hg; Arinco ingot iron used as comparative material
2	269	C	1954	273-873	± 5	90-2	94.4	5.2	Similar to the above specimen, except specimen cast in copper.
3	269	C	1954	273-1073	± 5	774	94.9	5.2	Similar to the above specimen except specimen cast in warm graphite.
4	394	C	1955	343.2	± 3	7B1	99.5	0.5	Specimen supplied by Argonne National Lab.; as rolled.
5	557	P	1953	295.2			94.71	5.29	Eutectic.
6	356	C	1954	293-1073			95.0	5.0	Thermal conductivity data obtained from the average values of 4 specimens.

## DATA TABLE NO. 225 THERMAL CONDUCTIVITY OF [URANIUM + CHROMIUM] ALLOYS

(U + Cr = 99.50%, impurity  $\leq$  0.20% each)[Temperature, T, K, Thermal Conductivity,  $\kappa$ , Watt  $\text{cm}^{-1} \text{K}^{-1}$ ]

T	$\kappa$	T	$\kappa$
<u>CURVE 1</u>			
273.2	0.293	293.2	0.29
373.2	0.304	373.2	0.30
473.2	0.319	473.2	0.32
573.2	0.338	573.2	0.34
673.2	0.360	673.2	0.36
773.2	0.383	773.2	0.38
873.2	0.415	873.2	0.40
973.2	0.444	973.2	0.43
1073.2	0.477	1073.2	0.46

CURVE 2

273.2	0.285
373.2	0.301
473.2	0.321
573.2	0.338
673.2	0.356
773.2	0.372
873.2	0.389

CURVE 3

273.2	0.282
373.2	0.295
473.2	0.310
573.2	0.330
673.2	0.350
773.2	0.371
873.2	0.396
973.2	0.423
1073.2	0.446

CURVE 4

343.2	0.277
-------	-------

CURVE 5

298.2	0.306
-------	-------

<sup>2</sup> Not shown on plot

## SPECIFICATION TABLE NO. 226 THERMAL CONDUCTIVITY OF [ URANIUM + IRON ] ALLOYS

( U + Fe : 99.50%; impurity  $\leq$  0.20% each )

Curve No.	Ref. Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)		Composition (continued), Specifications and Remarks
						U	Fe	
1	557	P	1953	298.2		89.9	10.1	Eutectic.

## DATA TABLE NO. 226 THERMAL CONDUCTIVITY OF [ URANIUM + IRON ] ALLOYS

( U + Fe : 99.50%; impurity  $\leq$  0.20% each )[ Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup>K<sup>-1</sup> ]

T k

CURVE 1<sup>st</sup>

298.2 0.0917

No graphical presentation

SPECIFICATION TABLE NO. 227 THERMAL CONDUCTIVITY OF [ URANIUM + MAGNESIUM ] ALLOYS  
 (U + Mg : 99.50%; impurity  $\leq$  0.20% each)

Curve No.	Ref. Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) U	Mg	Composition (continued), Specifications and Remarks
1	295	1952	523, 653			73.5	26.5	Extruded powder specimen.

DATA TABLE NO. 227 THERMAL CONDUCTIVITY OF [ URANIUM + MAGNESIUM ] ALLOYS  
 (U + Mg : 99.50%; impurity  $\leq$  0.20% each)

[ Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup>K<sup>-1</sup> ]

T	k
523.2	1.130
653.2	1.100

CURVE 1<sup>st</sup>

No graphical presentation

FIGURE SHOWS ONLY 6 OF THE CURVES REPORTED IN TABLE

# THERMAL CONDUCTIVITY OF URANIUM + MOLYBDENUM ALLOYS

(U + Mo 99.90%; impurity  $\leq$  0.20% each)

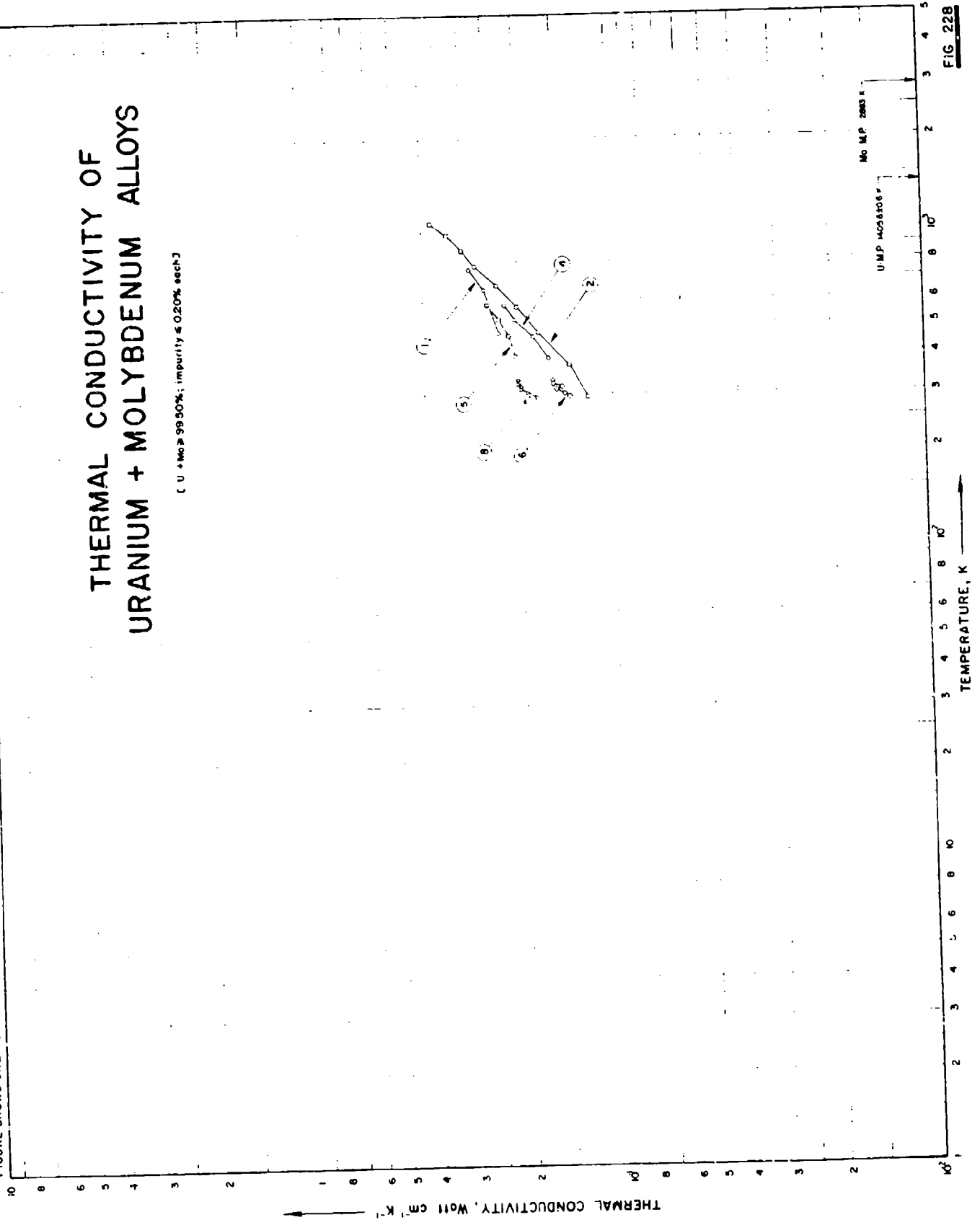


FIG 228



SPECIFICATION TABLE NO. 228 THERMAL CONDUCTIVITY OF URANIUM + MOLYBDENUM ALLOYS  
(U + Mo = 99.50%, impurity = 0.20% each)

[ For Data Reported in Figure and Table No. 228 ]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition U	Composition Mo	Composition (weight percent)	Composition (continued), Specifications and Remarks
1	46		1958	473-710			94.6	5.4		No details reported.
2	538	L	1956	293-1073			90	10		Measured under vacuo of $\sim 1 \times 10^{-3}$ mm Hg.
3	392		1945	398-585			95	5		Metastable gamma state.
4	392		1945	393-583			95	5		The above specimen annealed for 40 hrs. at 520 C; alpha state.
5	392		1945	398-583			95	5		The above specimen water quenched from 700 C; gamma state.
6	730	L	1945	295-332	2	C-270-5	95	5		Specimen as rolled rod.
7	730	L	1945	302-334	2	C-270-10 B	95	5		Similar to the above specimen but annealed 2 hr. at 550 C. then water quenched.
8	730	L	1945	294-333	2	C-270-10 A	95	5		Similar to the above specimen but annealed 2 hr. at 850 C. furnace cooled to 200 C then water quenched.

## DATA TABLE NO. 228 THERMAL CONDUCTIVITY OF URANIUM + MOLYBDENUM ALLOYS

(U + Mo = 99.50%, Impurity = 0.20% each)

(Temperature, T, K; Thermal Conductivity,  $k$ , Watt cm<sup>-1</sup> K<sup>-1</sup>)

T	k	T	k
<u>CURVE 1</u>			
471.2	0.230	318.7	0.145
634.2	0.215	324.4	0.155
799.2	0.255	332.9	0.155
<u>CURVE 2</u>			
291.2	0.191	301.6	0.138
353.2	0.138	303.7	0.146
471.2	0.174	311.0	0.151
571.2	0.201	324.5	0.159
674.2	0.235	334.4	0.163
771.2	0.272		
873.2	0.311		
973.2	0.335		
1073.2	0.375		
<u>CURVE 3</u>			
398.2	0.160	294.2	0.176
461.2	0.180	304.7	0.188
523.2	0.205	311.7	0.197
583.2	0.220	318.9	0.197
		324.5	0.201
		332.6	0.201
<u>CURVE 4</u>			
361.2	0.169		
461.2	0.180		
523.2	0.205		
583.2	0.220		
<u>CURVE 5</u>			
398.2	0.205		
463.2	0.215		
523.2	0.230		
583.2	0.250		
<u>CURVE 6</u>			
286.3	0.118		
302.2	0.142		
311.5	0.151		

Not shown on plot

## SPECIFICATION TABLE NO. 229 THERMAL CONDUCTIVITY OF [URANIUM + NIOBIUM] ALLOYS

(U + Nb - 99.50% impurity  $\leq$  0.20% each)

Curve No.	Ref. Method Used	Year	Temp. Range, K	Reports: Error, %	Name and Specimen Designation	Composition (weight percent)		Composition (continued), Specifications and Remarks
						U	Nb	
422		1945	400-619			96	4	

## DATA TABLE NO. 229 THERMAL CONDUCTIVITY OF [URANIUM + NIOBIUM] ALLOYS

(U + Nb - 99.50% impurity  $\leq$  0.20% each)[ Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup>K<sup>-1</sup> ]

T	k
399.7	0.220
474.2	0.215
548.7	0.204
618.7	0.215

CURVE 1\*

\* No graphical presentation

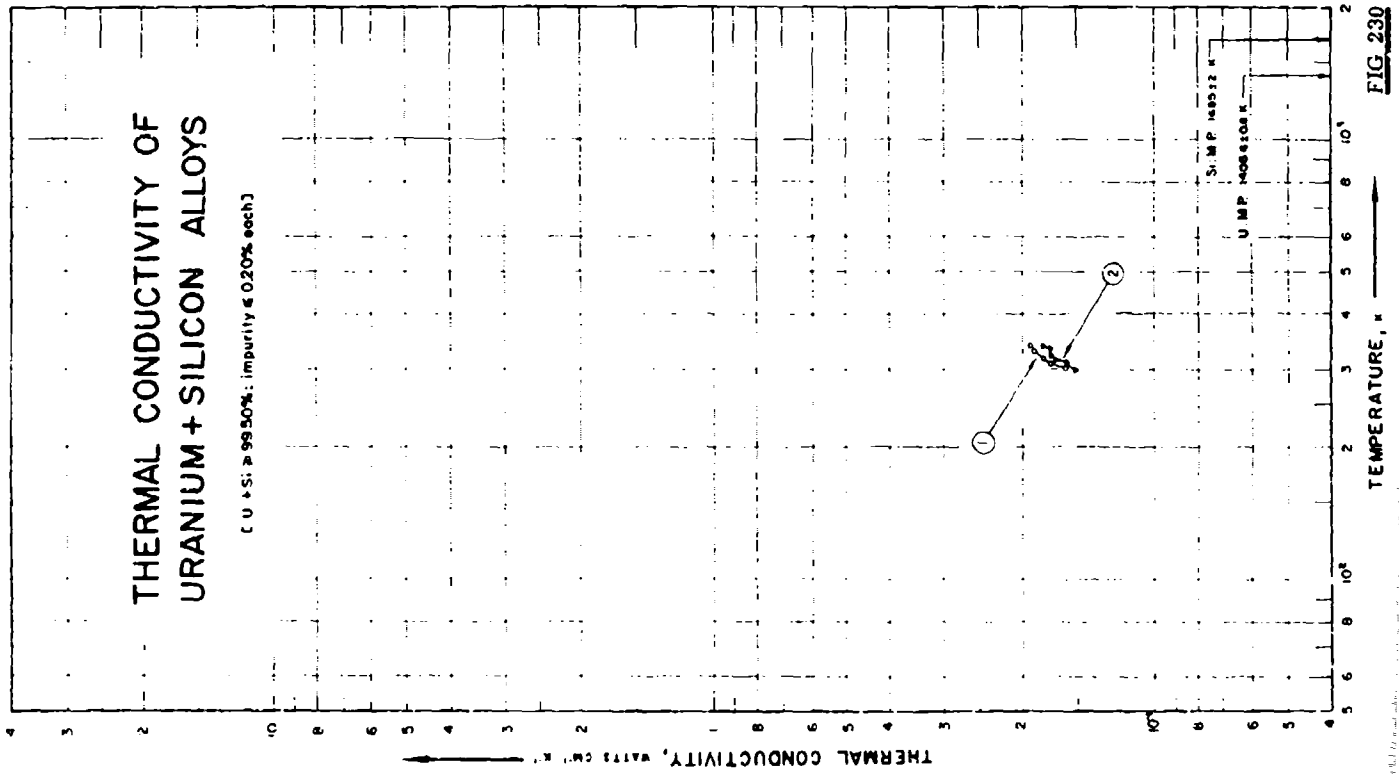


FIG. 230

SPECIFICATION TABLE NO. 230 THERMAL CONDUCTIVITY OF URANIUM-SILICON ALLOYS

(U = Si = 99.50%; impurity = 0.20% each)

[ For Data Reported in Figure and Table No. 230 ]

Curve No.	Rel. Method No.	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)		Composition (continued), Specifications and Remarks
						U	Si	
1	530	1945	301-310	2	E-19	81	19	Heat treated: specimen homogeneous in the $\alpha$ -phase. Similar to the above specimen.
2	530	1945	298-339	2	E-20	77	23	

## DATA TABLE NO. 230 THERMAL CONDUCTIVITY OF [ URANIUM + SILICON ] ALLOYS

(U + Si - 99.50%; impurity - 0.20% each)

(Temperature, T, K; Thermal Conductivity,  $k$ , Watt cm<sup>-1</sup> K<sup>-1</sup>)

T	k
<u>CURVE 1</u>	
301.0	0.159
308.8	0.172
319.2	0.180
330.9	0.188
339.6	0.192
<u>CURVE 2</u>	
298.1	0.151
310.5	0.159
315.7	0.167
324.4	0.172
333.9	0.172
339.2	0.180

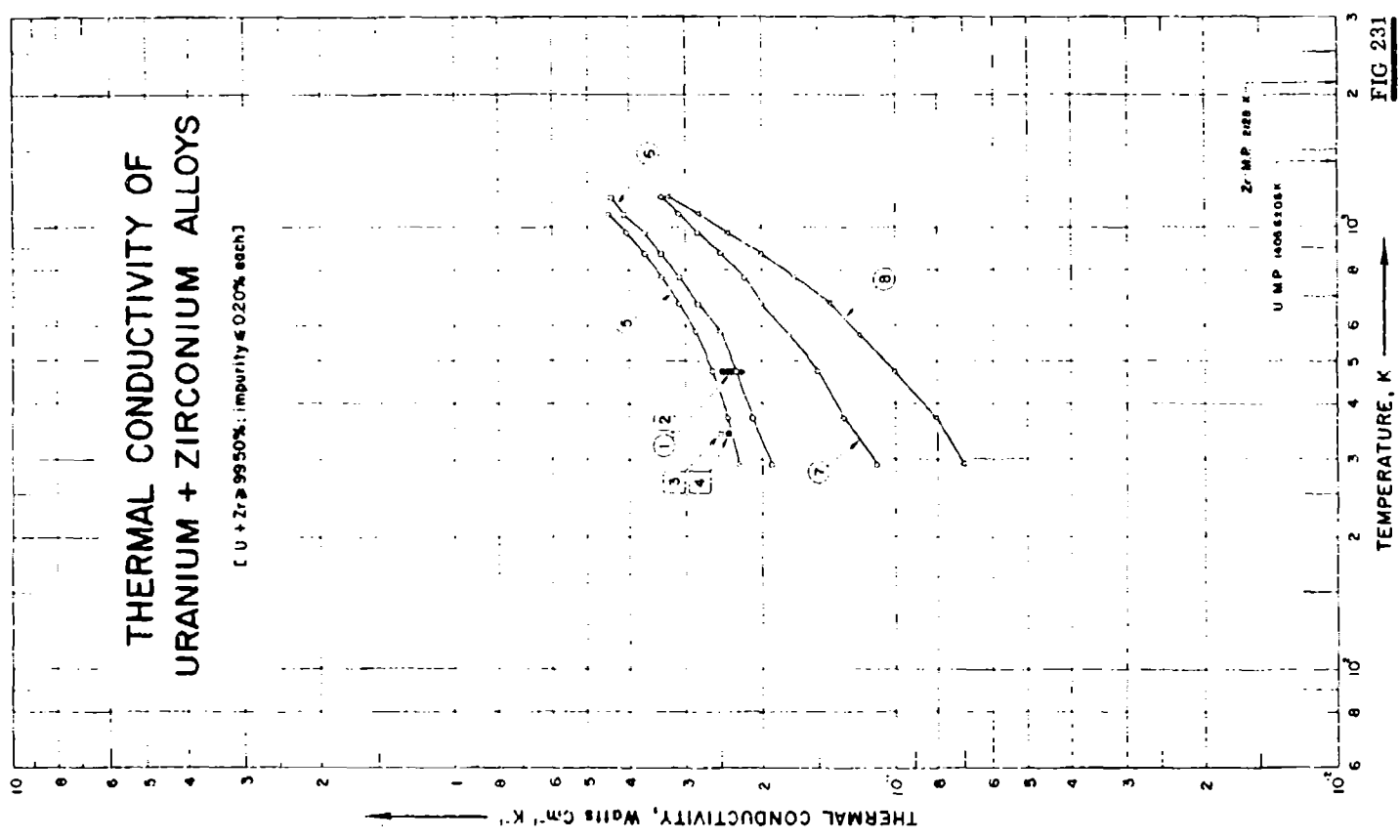


FIG 231

## SPECIFICATION TABLE NO. 231 THERMAL CONDUCTIVITY OF URANIUM-ZIRCONIUM ALLOYS

(U-Zr 99.50% impurity 0.20% each)

For Data Reported in Figure and Table No. 231

Curve No.	Ref. No. Used	Method	Year	Temp. Range, K	Temp. Reported Error, %	Name and Specimen Designation	Composition (weight percent) U	Composition (weight percent) Zr	Composition (continued), Specifications and Remarks
1	393	C	1955	473.2	±10	ANL-AA-20(ho0)	98.4	1.6	10.0 U-235 enriched; irradiated. 0.75 (at. %) burn up.
2	394	C	1955	473.2	±10	ANL-AA-22(ho6)	98.4	1.6	Similar to the above specimen except 0.21 (at. %) burn up.
3	394	C	1955	343.2	±3	3 B 1	98.9	1.1	As rolled.
4	394	C	1955	343.2	±3	4 B 4	98.5	1.5	As rolled.
5	395	C	1958	293-1073	±5		98.5	1.5	Measured in vacuo; zircaloy -2 used as standard.
6	396		1954	293-1173			95.0	5.0	No details reported.
7	396		1954	293-1173			80.0	20.0	No details reported.
8	396		1954	293-1173			60.0	40.0	No details reported.



DATA TABLE NO. 231 THERMAL CONDUCTIVITY OF [URANIUM + ZIRCONIUM] ALLOYS

(U + Zr = 99.50%; impurity = 0.26% each)

[Temperature, T, K; Thermal Conductivity,  $k$ , Watt  $\text{cm}^{-1}\text{K}^{-1}$ ]

T	k	T	k
<u>CURVE 1</u>			
473.2	0.222	873.2	0.34
473.2	0.243	973.2	0.37
473.2	0.234	1073.2	0.41
473.2	0.247	1173.2	0.44
473.2	0.238		
<u>CURVE 2</u>			
473.2	0.226	253.2	0.11
473.2	0.247	373.2	0.15
473.2	0.243	473.2	0.15
473.2	0.231	573.2	0.17
473.2	0.247	673.2	0.20
		773.2	0.22
		873.2	0.25
		973.2	0.28
		1073.2	0.31
		1173.2	0.34
<u>CURVE 3</u>			
343.2	0.249		
<u>CURVE 4</u>			
343.2	0.239	293.2	0.07
		373.2	0.08
<u>CURVE 5</u>			
281.2	0.226	473.2	0.10
373.2	0.240	573.2	0.12
473.2	0.260	673.2	0.14
573.2	0.285	773.2	0.17
673.2	0.310	873.2	0.20
773.2	0.340	973.2	0.24
873.2	0.370	1073.2	0.28
973.2	0.405	1173.2	0.33
1073.2	0.445		
<u>CURVE 6</u>			
293.2	0.19		
373.2	0.21		
473.2	0.23		
573.2	0.25		
673.2	0.28		
773.2	0.31		

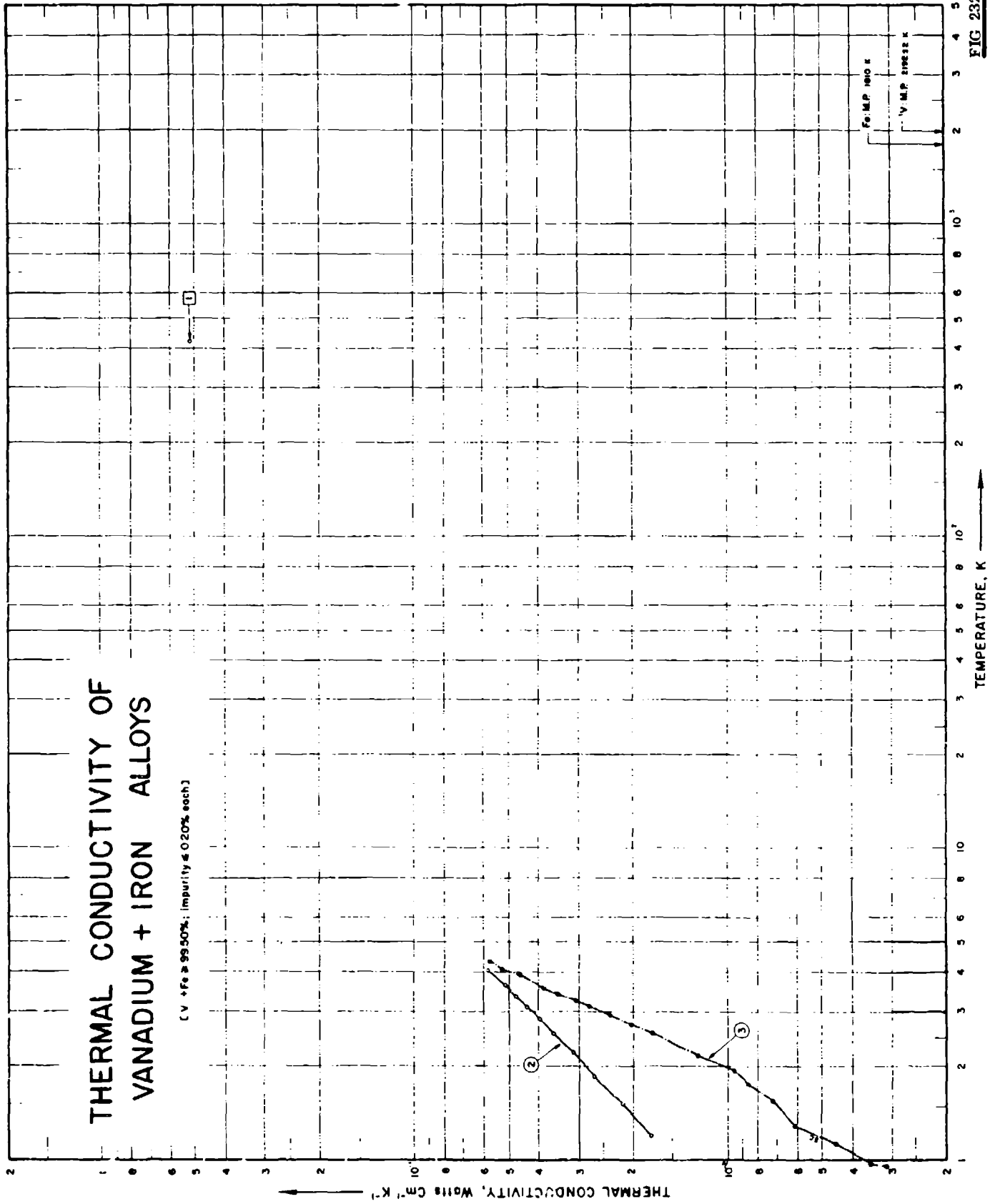


FIG 232

## SPECIFICATION TABLE NO. 232 THERMAL CONDUCTIVITY OF VANADIUM-IRON ALLOYS

(V - Fe 99.50%; impurity 0.50% each)

For Data Reported in Figure and Table No. 232

Curve No.	Rec. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition V	Composition (weight percent) Fe	Composition (continued), Specifications and Remarks
1	204	1	1957	423.2		Russian ferrovanadium 36	83	16.56	0.11 C.
2	501		1961	1.2-4.1		VI	99.761	0.2	0.02 Si, 0.005 Cu, 0.002 Ti, 0.002 Mo, 0.01 Mn; single crystal; specimen obtained by the floating-zone melting of polycrystalline rod; in normal state in a field of 6200 oersteds.
3	502		1961	0.95-4.4		VI	99.761	0.2	0.02 Si, 0.005 Cu, 0.002 Ti, 0.002 Mo, 0.01 Mn; the above specimen in superconducting state.

## DATA TABLE NO. 232 THERMAL CONDUCTIVITY OF [VANADIUM + IRON] ALLOYS

(V + Fe = 99.50%; impurity = 0.20% each)

[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup>K<sup>-1</sup>]

T k

CURVE 1

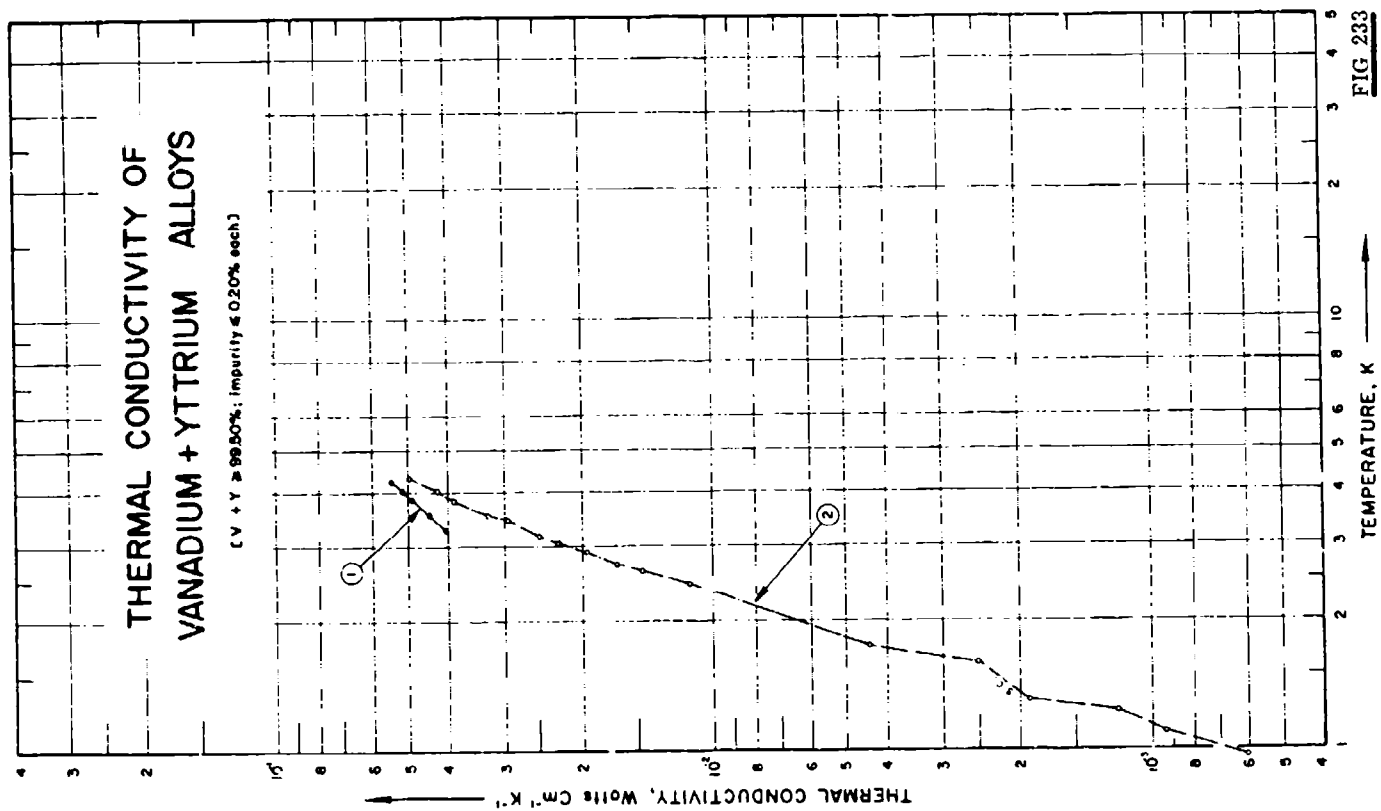
423.2 0.516

CURVE 2

1.20 0.0175  
 1.52 0.0215  
 1.86 0.0266  
 2.21 0.0311  
 2.55 0.0359  
 2.85 0.0355  
 3.10 0.0434  
 3.35 0.0475  
 3.65 0.0511  
 4.05 0.0580

CURVE 3

0.95 0.0031  
 0.97 0.0035  
 1.13 0.0045  
 1.29 0.0061  
 1.55 0.0071  
 1.75 0.0085  
 1.94 0.0095  
 2.16 0.0124  
 2.56 0.0174  
 2.73 0.0201  
 2.93 0.0236  
 3.11 0.0276  
 3.25 0.0305  
 3.44 0.0347  
 3.59 0.0385  
 3.93 0.045~  
 4.13 0.0521  
 4.55 0.0572



## SPECIFICATION TABLE NO. 233 THERMAL CONDUCTIVITY OF VANADIUM-YTTRIUM ALLOYS

(V - Y 99.50% purity 0.20% each)

For Data Reported in Figure and Table No. 233

Curve No.	Rel. Method No. Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) %	Composition (continued), Specifications and Remarks
1	501	1961	3.3-4.3			99.35	0.09 Fe, 0.01 Si, 0.05 N, 0.02 C, and 0.01 poly-crystal in normal state in a field of 4000 oersteds.
2	501	1961	0.97-4.3			0.11	The above specimen in superconducting state.

## DATA TABLE NO. 233 THERMAL CONDUCTIVITY OF [VANADIUM + YITTRIUM] ALLOYS

(V + Y = 99.50%; impurity = 0.20% each)

[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup>K<sup>-1</sup>]

T	k
<u>CURVE 1</u>	
3.25	0.0402
3.55	0.0411
3.85	0.0487
4.05	0.0514
4.25	0.0544
<u>CURVE 2</u>	
0.97	0.00060
1.10	0.00093
1.24	0.0012
1.31	0.0019
1.60	0.0025
1.75	0.0044
2.45	0.0113
2.62	0.0143
2.72	0.0165
2.90	0.0195
3.04	0.0224
3.16	0.0248
3.45	0.0294
3.53	0.0325
3.74	0.0387
4.05	0.0422
4.31	0.0495

## SPECIFICATION TABLE NO. 231 THERMAL CONDUCTIVITY OF [ZINC + ALUMINUM] ALLOYS

[Zn + Al - 99.50% impurity - 0.50% each]

Curve No.	Rel. Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) Zn	Composition (weight percent) Al	Composition (continued), Specifications and Remarks
1	17	R	1958	293-353	± 1.0	Zavak Nr 100	4.15	0.025 Cu, 0.032 Si, 0.007 Mg, 0.007 Fe, trace Bi, Cd, Pb, Sn, and Ti.
2	230	L	1925	323.2		69.0	19.0	Impurities - 0.07.

## DATA TABLE NO. 231 THERMAL CONDUCTIVITY OF [ZINC + ALUMINUM] ALLOYS

[Zn + Al - 99.50% impurity - 0.50% each]

[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup>K<sup>-1</sup>]

T	k
<u>CURVE 1</u>	
293.2	1.17
323.2	1.17
353.2	1.17
<u>CURVE 2</u>	
323.2	1.19

No graphical presentation



SPECIFICATION TABLE NO. 235 THERMAL CONDUCTIVITY OF ZINC + CADMIUM ALLOYS

(Zn + Cd - 99.50%; impurity < 0.50% each)

Curve No.	Rel. Method No. Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) Zn	Composition (weight percent) Cd	Composition (continued), Specifications and Remarks
1	230	L	1925	326.2		60.0	40.6	Impurities < 0.03,
2	230	L	1925	326.2		70.6	31.0	Impurities < 0.03,
3	230	L	1925	326.2		80.0	20.0	Impurities < 0.03,
4	230	L	1925	326.2		95.0	5.0	Impurities < 0.03,

DATA TABLE NO. 235 THERMAL CONDUCTIVITY OF ZINC + CADMIUM ALLOYS

(Zn + Cd - 99.50%; impurity < 0.20% each)

[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup>K<sup>-1</sup>]

γ k

CURVE 1\*

326.2 1.042

CURVE 2\*

326.2 1.067

CURVE 3\*

326.2 1.088

CURVE 4\*

326.2 1.126

\* No graphical presentation

## SPECIFICATION TABLE NO. 236 THERMAL CONDUCTIVITY OF [ZIRCONIUM + ALUMINUM] ALLOYS

(Zr + Al) ~ 99.50%; impurity &lt; 0.20% each)

Curve No.	Ref. Method No. Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)		Composition (continued), Specifications and Remarks
						Zr	Al	
1	554	1954	318.7			96	4	Melted in a nonconsumable electrode arc furnace.
2	555	1956	406.2			97	3	

## DATA TABLE NO. 236 THERMAL CONDUCTIVITY OF [ZIRCONIUM + ALUMINUM] ALLOYS

(Zr + Al) ~ 99.50%; impurity &lt; 0.20% each)

{ Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup> K<sup>-1</sup> }

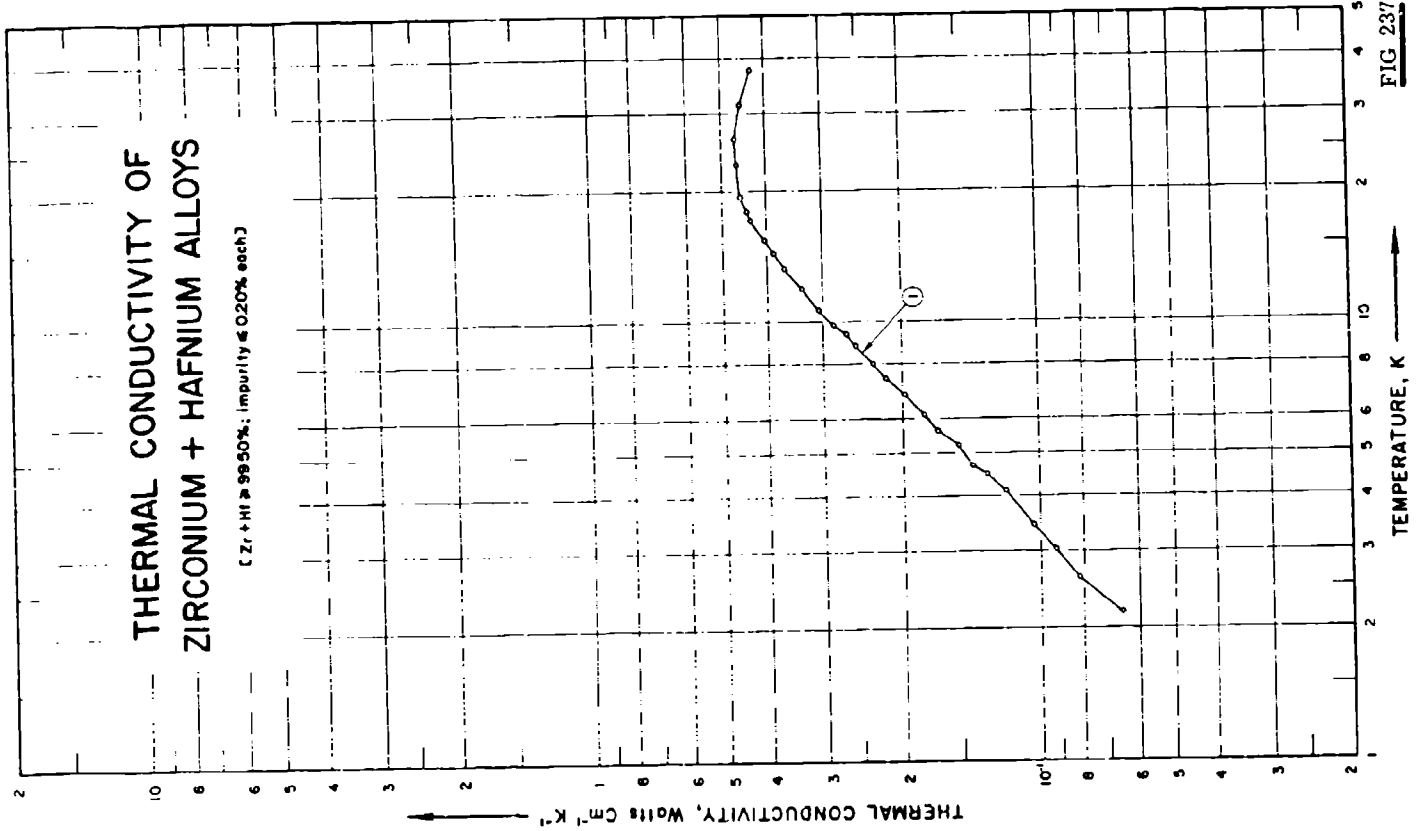
T k

CURVE 1

318.7 0.0837

CURVE 2

406.2 0.0828



## SPECIFICATION TABLE NO. 237 THERMAL CONDUCTIVITY OF [ ZIRCONIUM - HAFNIUM ] ALLOYS

(Zr + Hf = 99.50%, impurity = 0.20% each)

! For Data Reported in Figure and Table No. 237

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) Zr	Composition (weight percent) Hf	Composition (continued), Specifications and Remarks
1	122	L	1955	2.2-3 s	3	Zr 2	~55	~2	Polycrystalline specimen (large crystals), 3.07 cm long, 0.44 cm dia., annealed, supplied by Metro-Vickers.

## DATA TABLE NO. 237 THERMAL CONDUCTIVITY OF ZIRCONIUM-HAFNIUM ALLOYS

(Zr + Hf = 99.50%, impurity = 0.20% each)

[Temperature, T, K. Thermal Conductivity, k, Watt cm<sup>-1</sup> K<sup>-1</sup>]

T	k
2.16	0.0658
2.62	0.0821
3.04	0.0927
3.45	0.104
4.07	0.119
4.50	0.131
4.71	0.141
5.25	0.152
5.65	0.165
6.14	0.180
6.83	0.198
7.42	0.219
8.04	0.238
8.53	0.255
9.42	0.266
9.90	0.285
10.50	0.307
12.00	0.335
13.34	0.366
14.59	0.387
15.50	0.405
17.20	0.433
18.00	0.444
19.59	0.459
23.00	0.469
26.42	0.471
31.50	0.456
37.90	0.431

SPECIFICATION TABLE NO. 27 THERMAL CONDUCTIVITY OF [ZIRCONIUM + NIOBIUM] ALLOYS  
(Zr + Nb = 99.50%; impurity  $\leq$  0.20% each)

Curve No.	Rel. Method No. Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) Zr Nb	Composition (continued), Specifications and Remarks
1	441	1957	342-899			98.26 1.52	0.14 Hf, 0.08 C.

DATA TABLE NO. 28 THERMAL CONDUCTIVITY OF [ZIRCONIUM + NIOBIUM] ALLOYS

(Zr + Nb = 99.50%; impurity  $\leq$  0.20% each)

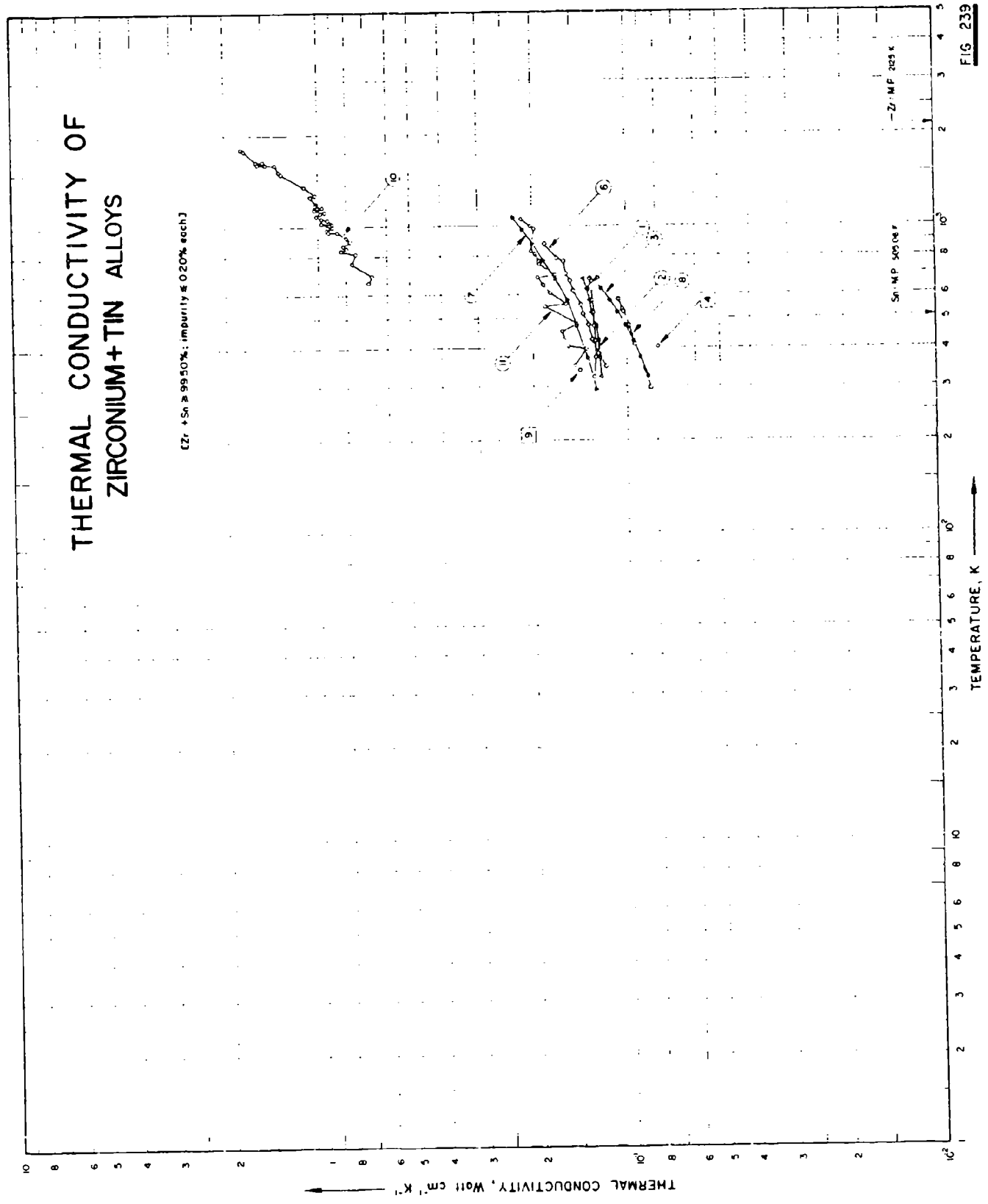
[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup>K<sup>-1</sup>]

T	k
342.2	6.213
391.2	6.213
479.1	6.210
514.9	6.210
562.0	6.213
632.2	6.216
680.9	6.219
753.8	6.227
834.2	6.245
898.7	6.242

No graphical presentation

# THERMAL CONDUCTIVITY OF ZIRCONIUM+TIN ALLOYS

[Zr + Sn ≥ 99.50%; impurity ≤ 0.20% each]



SPECIFICATION TABLE NO. 259 THERMAL CONDUCTIVITY OF ZIRCONIUM-TIN ALLOYS

(Zr = Sn 99.50%; impurity = 0.20% each)

For Data Reported in Figure and Table No. 259

Curve No.	Rel. Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) Zr	Composition (weight percent) Sn	Composition (continued), Specifications and Remarks
1	27	C	323-673		315	95.45	2.510	0.005 Ni; specimen 2 cm dia., 15 cm long; arc-melted and forged at 1700 F; Arnico iron used as standard.
2	27	C	323-673		370	97.005	2.300	0.002 Ni; specimen 2 cm dia., 15 cm long; arc-melted and forged at 1600 F; Arnico iron used as standard.
3	27	C	323-673		1009	93.0	7.0	Prepared from <i>Evade</i> grade <i>Terstal</i> bar Zr and c. p. Sn; arc-melted; Arnico iron as standard.
4	595	C	404.2			90	10	Melted in a nonconsumable electrode arc furnace.
5	555	C	407.2			95	5	Melted in a nonconsumable electrode arc furnace.
6	101	L	348-873	3		97.0	3.0	Specimen 7.935 in. long, 0.787 in. dia; supplied by Westinghouse; melted in vacuum furnace, forged at 1450 C in argon; annealed 0.5 hr at 1000 C in vacuum.
7	395	C	293-1053	1.5	Zircaloy - 2	98.2	1.5	0.15 Ti, 0.10 Cr, 0.05 Ni (nominal composition from Metals Handbook).
8	442	C	298-573	1.3	335	95	5	Induction melted from low-bathum Bureau of Mines; sponge in graphite mold; forged in 1500 F in air.
9	591	C	338.2	1.5	Zircaloy - 2	99.5	1.5	Hot rolled; copper used as standard.
10	930	R	629-1793		Zircaloy - 4	Bd	0.61	Typical composition with impurities as received: 0.23 Fe, 0.16 Mg, 0.086 Cr, 0.0176 Cu, 0.91 each of H and W, 0.0043 Ni, 0.0035 C, 0.0024 each of Al, Mn and Si, 0.0018 Ti, 0.0012 Nb, 0.001 Co, 0.0001 each of B and Cd; specimen exposed 100 hrs at 400 to 500 C with impurities 2.5 Mg, 1.16 Sn, 0.24 Fe, 0.057 Cr, 0.0216 Cu, 0.013 H, 0.01 W, 0.012 Ni, 0.0037 C, 0.0022 Mn, 0.002 Al, 0.0025 Si, 0.0019 Ti, trace Co, Ni, B and Cd; exposed 200 hrs at 400 to 500 C with impurities of 0.5 Sn, 0.19 Fe, 0.0515 Cr, 0.021 Ni, 0.0132 Cu, 0.01 each of H and W, 0.0087 Al, 0.0032 C, 0.0046 Si, trace Mg, Ti, Co, Ni, B and Cd.
11	931	L, C	351-1058	1.5	Zircaloy - 4	Bd	1.2-1.7	0.17-0.21 Fe, 0.07-0.13 Cr, 0.28-0.35 total in parities of Fe, Ni, Cr; nickel used as comparative material.





SPECIFICATION TABLE NO. 240 THERMAL CONDUCTIVITY OF [ZIRCONIUM + TITANIUM] ALLOYS  
(Zr + Ti) - 99.50%; impurity  $\leq 0.20\%$  each)

Curve No.	Ref. Method Used	Year	Temp. Range, K	Reported Zr, %	Name and Specimen Designation	Composition (weight percent) Zr	Ti	Composition (continued), Specifications and Remarks
1	554	1954	319.7			93	7	

DATA TABLE NO. 240 THERMAL CONDUCTIVITY OF [ZIRCONIUM + TITANIUM] ALLOYS  
(Zr + Ti) - 99.50%; impurity  $\leq 0.20\%$  each)

[ Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup>K<sup>-1</sup> ]

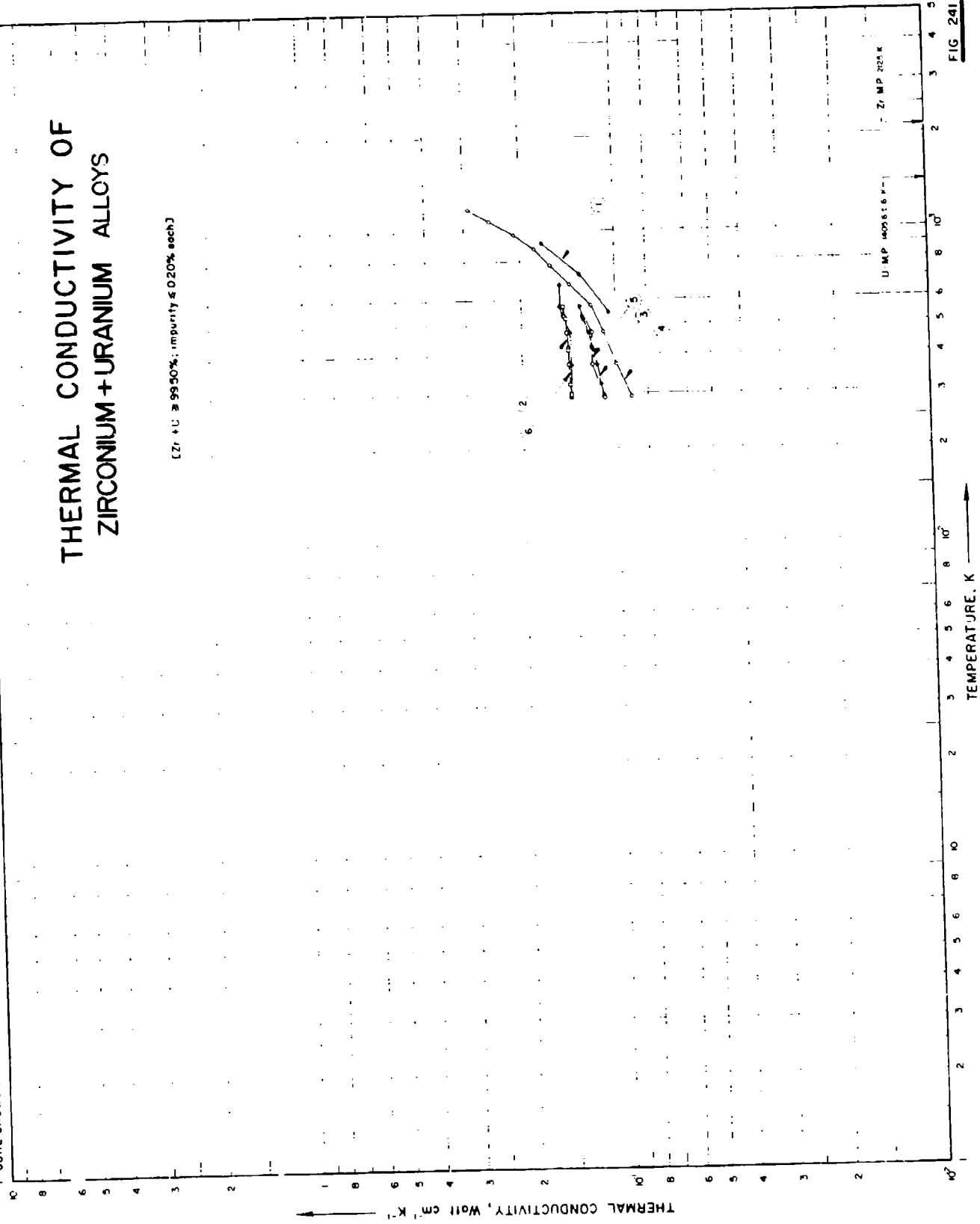
T k  
CURVE 1  
319.7 0.130

No graphical presentation.

FIGURE SHOWS ONLY 6 OF THE CURVES REPORTED IN TABLE

# THERMAL CONDUCTIVITY OF ZIRCONIUM + URANIUM ALLOYS

(Zr + U  $\geq$  99.90%; impurity  $\leq$  0.20% each)



SPECIFICATION TABLE NO. 241 THERMAL CONDUCTIVITY OF ZIRCONIUM-URANIUM ALLOYS

(Zr = U - 99.50%; impurity < 0.20% each)

(For Data Reported in Figures and Table No. 241)

Curve No.	Ref. No. Used	Method	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) Zr	Composition (continued), Specifications and Remarks
1	295	C	1952	543-913	± 5		70.3	Extruded; specimen 0.75 in. dia; 9 in. long.
2	442	C	1951	293-573	± 3	5	97	Electrical resistivity 1.1 and 102.5 $\mu\text{ohm cm}$ at 25, and 260 C, respectively.
3	442		1951	293-573		6	96	Electrical resistivity 85.5 and 116.0 $\mu\text{ohm cm}$ at 25, and 260 C, respectively.
4	396		1954	293-1173			70	No details reported.
5	396		1954	293-573			86	No details reported.
6	396		1954	293-673			97	No details reported.

DATA TABLE NO. 241 THERMAL CONDUCTIVITY OF [ZIRCONIUM + URANIUM] ALLOYS

(Zr + U = 99.50%; impurity  $\leq$  0.20% each)

[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup> K<sup>-1</sup>]

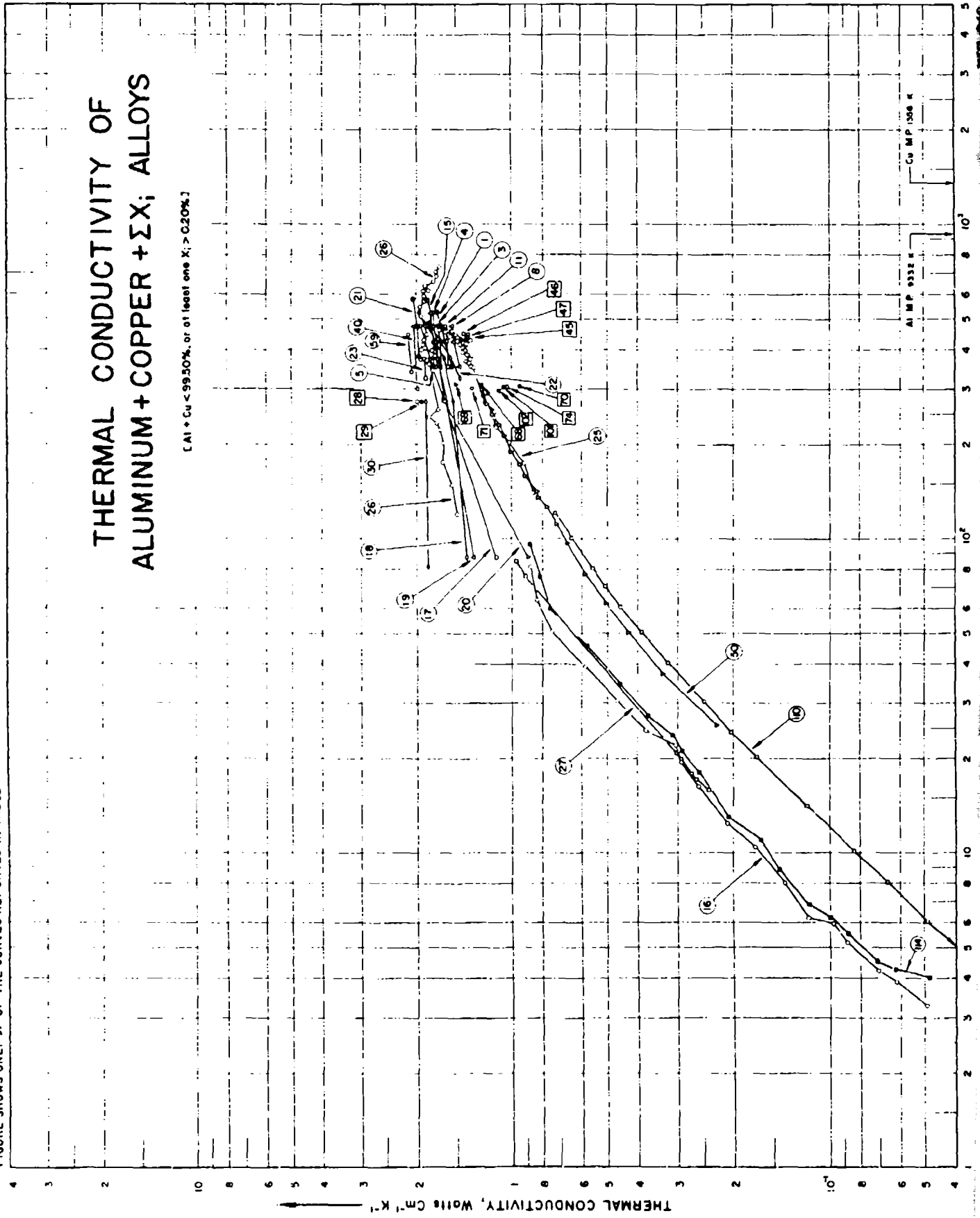
T	k	T	k
<u>CURVE 1</u>			
548.2	0.106	293.2	0.11
723.2	0.130	373.2	0.12
913.2	0.170	473.2	0.12
		573.2	0.13*
<u>CURVE 2</u>			
298.2	0.140	298.2	0.14
323.2	0.141	373.2	0.14
373.2	0.142	473.2	0.14
415.2	0.143	573.2	0.15
423.2	0.143	673.2	0.15
473.2	0.144		
523.2	0.145		
533.2	0.146		
573.2	0.147		
<u>CURVE 3</u>			
298.2	0.111		
323.2	0.113		
373.2	0.116		
415.2	0.119		
423.2	0.120		
473.2	0.123		
523.2	0.127		
533.2	0.128		
573.2	0.130		
<u>CURVE 4</u>			
293.2	0.09		
373.2	0.10		
473.2	0.11		
573.2	0.12		
673.2	0.14		
773.2	0.16		
873.2	0.18		
973.2	0.21		
1073.2	0.25		
1173.2	0.29		

\* Not shown on plot

FIGURE SHOWS ONLY 37 OF THE CURVES REPORTED IN TABLE

# THERMAL CONDUCTIVITY OF ALUMINUM + COPPER + ΣX; ALLOYS

[Al + Cu < 99.50%, or of least one X; > 0.20%.]



SPECIFICATION TABLE NO. 242 THERMAL CONDUCTIVITY OF [ALUMINUM + COPPER + EX<sub>1</sub>] ALLOYS(Al + Cu < 99.50% or at least one X<sub>1</sub> > 0.20%)

[For Data Reported in Figure and Table No. 242.]

Curve No.	Ref. Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Cu	Composition (weight per cent)					Zn	Composition (continued), Specifications and Remarks
							Fe	Mg	Mn	Ni	Si		
1	55	P-L	1928	353-523	Bar 779	12.0				3.0			Cast and annealed.
2	55	P-L	1928	353-473	Bar 792	12.0				2.0			Cast and annealed.
3	55	P-L	1928	353-473	Bar 766	9.0				2.0			Cast and annealed.
4	55	P-L	1926	353-523	Bar 762	8.0				1.0			Cast and annealed.
5	55	P-L	1928	353-473	Bar 795	8.0				3.0			Cast and annealed.
6	55	P-L	1928	353-473	Bar 829	9.0	1.0			2.0			Cast and annealed.
7	55	P-L	1928	353-473	Bar 888	8.0	1.5			2.0			Cast and annealed.
8	55	P-L	1928	353-473	Bar 810	8.0	2.0			2.0			Cast and annealed.
9	55	P-L	1928	353-473	Bar 906	4.0	1.5			2.0			Cast and annealed.
10	55	P-L	1928	353-473	Bar 785	8.0	1.0						Cast and annealed.
11	55	P-L	1928	353-473	Bar 789	8.0	2.0						Cast and annealed.
12	55	P-L	1928	353-473	Bar 886	4.0	2.0	0.5					Cast and annealed.
13	55	P-L	1928	353-473	Bar 656	8.0		1.0					Cast and annealed.
14	55	P-L	1928	353-473	Bar 798	7.0							1.0 Ag. cast and annealed.
15	55	P-L	1928	353-573	Bar 2311	4.0				3.0			Cast and annealed.
16	154, 587	L	1956	3.3-81		4.4	1.5	0.6					Unannealed.
17	93	L	1931	87-476	Y-Alloy	4.0	1.3			2.0			As cast.
18	93	L	1931	87-476	Y-Alloy	4.0	1.5			2.0			Annealed.
19	93	L	1931	87-476	Nelson-Kebkenleg 10	4.0	1.5	2.0					Cast.
20	93	L	1931	87-476	Duralumin	3.0/5.0	0.5						
21	20	L	1951	380-575	2	4.1	0.1	1.63		0.06	0.05		0.004 Ti; annealed at 300 - 500 C.
22	20	L	1951	324-526	3	4.33	0.06	1.37		0.02	0.10		0.005 Ti; heated to 500 C and quenched in water.
23	20	L	1951	324-374	3a	4.33	0.06	1.37		0.02	0.10		0.005 Ti; water-quenched from 500 C; drawn at 550 C.
24	20	L	1951	343-626	4	4.25	0.36	1.59	0.01	0.16	0.02		0.007 Ti; heated at about 300 C.
25	91	C	1951	140-532	245 - T4	4.5	1.5	0.6					As received.
26	91	C	1951	119-731	245 - T4	4.5	1.5	0.6					After heated to 300 C.
27	104	L	1951	15 - 85	Duralumin	4.10	0.42	0.57					94.0 Al; as stamped.
28	36	L	1935	273		4.0	0.5						Aged at 215 C.
29	36	L	1935	273		5.0	0.5						Aged at 215 C.
30	36	L	1935	81, 273		7.0	0.5						Aged at 215 C.

SPECIFICATION TABLE NO. 242 (continued)

Curve No.	Ref. Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight per cent)							Composition (continued), Specifications and Remarks	
						Cu	Fe	Mg	Mn	Ni	Si	Zn		
31	223	L	1937	298, 473	+ 3.0	7543	10.40	1.40	0.29			0.59		Chill-cast.
32	223	L	1937	298, 473	+ 3.0	7543	10.4	1.4	0.29			0.59		The above specimen annealed for 2 hrs at 371 C, then cooled to 316 C at 14 C per hr, and then cooled in furnace.
33	223	L	1937	298, 473	+ 3.0	7544	3.94	0.63	1.52		2.14	0.55		Chill-cast.
34	223	L	1937	298, 473	+ 3.0	7544	3.94	0.63	1.52		2.14	0.55		The above specimen annealed for 2 hrs at 371 C, then cooled to 316 C at 14 C per hr, and then cooled in furnace.
35	223	L	1937	298, 473	+ 3.0	7626	4.39	0.73				0.66		Cast in green sand; quenched from high-temperature solution treatment; aged at room temperature.
36	223	L	1937	298, 473	+ 3.0	7626	4.39	0.73				0.66		The above specimen annealed for 2 hrs at 371 C, then cooled to 316 C at 14 C per hr, and then cooled in furnace.
37	223	L	1937	298, 473	+ 3.0	7640	3.84	0.80	1.29		1.96	0.55		Forged and quenched from high-temperature solution treatment, and then given a low-temperature precipitation treatment.
38	223	L	1937	298, 473	+ 3.0	7640	3.84	0.80	1.29		1.96	0.55		The above specimen annealed at 371 C for 2 hrs, cooled to 316 C at 14 C per hr, and cooled in furnace.
39	223	L	1937	298, 473	+ 3.0	7643	4.45	0.50		0.76		0.85		Forged and quenched from high-temperature solution treatment, and then given a low-temperature precipitation treatment.
40	223	L	1937	298, 473	+ 3.0	7643	4.45	0.50		0.76		0.85		The above specimen annealed at 371 C for 2 hrs, cooled to 316 C at 14 C per hr, and then cooled in furnace.
41	223	L	1937	298, 473	+ 3.0	7644	3.79	0.96	0.49	0.59		0.15		Forged and quenched from high-temperature solution treatment.
42	223	L	1937	298, 473	+ 3.0	7644	3.7	0.96	0.49	0.54		0.15		The above specimen annealed at 371 C for 2 hrs, cooled to 316 C at 14 C per hr, and then cooled in furnace.
43	223	L	1937	298, 473	+ 3.0	7678	7.06	1.21				0.75	2.22	Sand-cast.
44	223	L	1937	298, 473	+ 3.0	7678	7.06	1.21				0.75	2.22	The above specimen annealed at 371 C for 2 hrs, cooled to 316 C at 14 C per hr, and cooled in furnace.
45	224	L	1923	436		12	7.61	0.56		0.35		0.31		Cast.
46	224	L	1923	447		12	6.78	0.58		0.36		0.23		Cast.
47	224	L	1923	445		12	7.30	0.59		0.41		0.23	0.03	Cast.
48	224	L	1923	428		12	8.04	0.63						Cast.



SPECIFICATION TABLE NO. 242 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Al	Cu	Fe	Mg	Mn	Ni	Si	Zn	Composition (continued), Specifications and Remarks
49	225	L	1928	373-623		"Y" Alloy	4		1.5		2				Annealed for 3 hrs at 300 C.
50	226	L	1951	25-296	<2.0	24S	4.49	0.34	1.47	0.66			0.13	0.01	0.02 Ti, 0.01 Cr; as reported by ALCOA.
51	227	L	1949	293-573		RR 59	2.31	1.23	1.46		1.20		0.88		0.07 Ti; wrought; heated at 525 C for 2 hrs and quenched, and heated at 170 C for 16 hrs and quenched, and again heat-treated at 300 C.
52	227	L	1949	293-473		"Y" Alloy	3.76	0.40	1.33		1.85		0.45		Wrought; heated at 511 C and quenched in fairly hot water, then aged at room temperature. The above specimen again heat-treated at 300 C.
53	227	L	1949	293-573		"Y" Alloy	3.76	0.40	1.33		1.85		0.45		Original composition reported as 98.99 Al (containing 0.21 Fe and 0.29 Si); as cast.
54	67	L	1932	338, 438		3	98.49		1.01	0.209			0.287		Original composition reported as 94.94 Al (containing 0.21 Fe and 0.29 Si); as cast.
55	67	L	1932	338, 438		5	94.47	5.06	0.199				0.275		Original composition reported as 94.94 Al (containing 0.21 Fe and 0.29 Si); as cast.
56	67	L	1932	338, 438		6	92.34	7.20	0.195				0.269		Original composition reported as 92.80 Al (containing 0.21 Fe and 0.29 Si); as cast.
57	67	L	1932	338, 438		8	88.05	11.51	0.186				0.257		Original composition reported as 88.45 Al (containing 0.21 Fe and 0.29 Si); as cast.
58	67	L	1932	338, 438		9	79.52	15.46	0.175				0.245		Original composition reported as 84.54 Al (containing 0.21 Fe and 0.29 Si); as cast.
59	67	L	1932	338, 438		3A	98.49	1.01	0.209				0.287		Original composition reported as 98.99 Al (containing 0.21 Fe and 0.29 Si); annealed at 500 C.
60	67	L	1932	338, 438		5A	94.47	5.06	0.199				0.275		Original composition reported as 94.94 Al (containing 0.21 Fe and 0.29 Si); annealed at 500 C.

SPECIFICATION TABLE NO. 242 (continued)

Curve No.	ReL No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)							Composition (continued), Specifications and Remarks	
							Al	Cu	Fe	Mg	Mn	Ni	Si		Zn
61	67	L	1932	338, 438		6A	92.34	7.20	0.195				0.259		Original composition reported as 92.80 Al (containing 0.21 Fe and 0.29 Si); annealed at 500 C.
62	67	L	1932	338, 438		8A	88.05	11.51	0.186				0.257		Original composition reported as 88.49 Al (containing 0.21 Fe and 0.29 Si); annealed at 500 C.
63	67	L	1932	338, 438		9A	84.12	15.46	0.178				0.245		Original composition reported as 84.54 Al (containing 0.21 Fe and 0.29 Si); annealed at 500 C.
64	67	L	1932	338, 438		10A	79.52	20.98	0.168				0.232		Original composition reported as 79.92 Al (containing 0.21 Fe and 0.29 Si); annealed at 500 C.
65	67	L	1932	338, 438		11A	74.03	25.60	0.156				0.216		Original composition reported as 74.40 Al (containing 0.21 Fe and 0.29 Si); annealed at 500 C.
66	67	L	1932	338, 438		12A	69.17	30.46	0.146				0.202		Original composition reported as 69.54 Al (containing 0.21 Fe and 0.29 Si); annealed at 500 C.
67	67	L	1932	338		10	79.52	20.08	0.168				0.232		Original composition reported as 79.92 Al (containing 0.21 Fe and 0.29 Si); as cast.
68	408	E	1925	305.2	<0.5	British L-8		12.21	0.62				0.30		Chill-cast.
69	408	E	1925	307.2	<0.5	British L-8		12.21	0.62				0.30		The above specimen annealed for 30 min at 450 C.
70	408	E	1925	304.2	<0.5	Japanese 2E-8		12.17	0.64				0.22		Chill-cast.
71	408	E	1925	298.2	<0.5	Japanese 2E-8		12.17	0.64				0.22		The above specimen annealed for 30 min at 450 C.
72	408	E	1925	302.2	<0.5	F		10.52	0.84			1.04	0.34		3.29 Sn; chill-cast.
73	408	E	1925	304.2	<0.5	F		10.52	0.84			1.04	0.34		3.29 Sn; the above specimen annealed for 30 min at 450 C.
74	408	E	1925	302.2	<0.5	Japanese M-1		8.42	0.70				0.27		Chill-cast.
75	408	E	1925	296.2	<0.5	Japanese M-1		8.42	0.70				0.27		The above specimen annealed for 30 min at 450 C.
75	408	E	1925	305.2	<0.5	No. 12		8.07	0.63				0.38		Chill-cast.

SPECIFICATION TABLE NO. 242 (continued)

Curve No.	Ref. Method No. Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight per cent)							Zn	Composition (continued), Specifications and Remarks
						Cu	Fe	Mg	Mn	Ni	Si			
77	408	E	1925	302.2	<0.5	No. 12	8.07	0.63					0.38	The preceding specimen annealed for 30 min at 450 C. The above annealed specimen heated for 30 min at 500 C, then quenched in water at about 8 C, and then measured after 4 to 5 hrs.
78	408	E	1925	299.2	<0.5	No. 12	8.07	0.63					0.38	
79	408	E	1925	303.2	<0.5	No. 12	8.07	0.63					0.38	The above quenched specimen measured after aging for 2 weeks. 1.24 Sn; chill-cast. 1.24 Sn, the above specimen annealed for 30 min at 450 C.
80	408	E	1925	306.2	<0.5	British 21-11	6.56	0.65					0.32	
81	408	E	1925	304.2	<0.5	British 21-11	6.86	0.65					0.32	
82	408	E	1925	302.2	<0.5	D	5.27	0.82	1.15	0.50			0.45	Chill-cast. The above specimen annealed for 30 min at 450 C.
83	408	E	1925	299.2	<0.5	D	5.27	0.82	1.15	0.50			0.45	
84	408	E	1925	298.2	<0.5	D	5.27	0.82	1.15	0.50			0.45	The above annealed specimen heated for 30 min at 490 C, then quenched in water at about 8 C, and then measured after 4 to 5 hrs. The above quenched specimen measured after aging for 2 weeks.
85	408	E	1925	301.2	<0.5	D	5.27	0.82	1.15	0.50			0.45	
86	408	E	1925	308.2	<0.5	D	5.27	0.82	1.15	0.50			0.45	Forged and cold-drawn. The above specimen annealed for 30 min at 500 C.
87	408	E	1925	306.2	<0.5	D	5.27	0.82	1.15	0.50			0.45	
88	408	E	1925	301.2	<0.5	Dizeppelin	4.32	0.87	0.42	0.55			0.38	Chill-cast. The above specimen annealed for 30 min at 450 C.
89	408	E	1925	300.2	<0.5	Dizeppelin	4.32	0.87	0.42	0.55			0.38	
90	408	E	1925	303.2	<0.5	Dizeppelin	4.32	0.87	0.42	0.55			0.38	Forged and cold-drawn. The above specimen annealed for 30 min at 500 C.
91	408	E	1925	304.2	<0.5	Dizeppelin	4.32	0.87	0.42	0.55			0.38	
92	408	E	1925	305.2	<0.5	British Y-1	2.53	0.87	0.90		1.75		0.36	Chill-cast. The above specimen annealed for 30 min at 450 C.
93	408	E	1925	302.2	<0.5	British Y-1	2.53	0.87	0.90		1.75		0.36	
94	408	E	1925	299.2	<0.5	British Y-1	2.53	0.87	0.90		1.75		0.36	The above annealed specimen heated for 30 min at 530 C, then quenched in water at about 8 C, and then measured after 4 to 5 hrs. The above quenched specimen measured after aging for 2 weeks.
95	408	E	1925	308.2	<0.5	British Y-1	2.53	0.87	0.90		1.75		0.36	
96	408	E	1925	309.2	<0.5	British Y-2	4.44	0.65	0.86		2.06		0.49	The above specimen annealed for 30 min at 450 C. Chill-cast.
97	408	E	1925	311.2	<0.5	British Y-2	4.44	0.66	0.86		2.06		0.49	
98	408	E	1925	306.2	<0.5	N	1.85	0.95	1.48				0.12	The above specimen annealed for 30 min at 450 C.
99	408	E	1925	301.2	<0.5	N	1.85	0.95	1.48				0.12	

SPECIFICATION TABLE NO. 212 (continued)

Curve No.	ReL No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)							Composition (continued), Specifications and Remarks		
							Al	Cu	Fe	Mg	Mn	Ni	Si		Zn	
100	408	E	1925	301.2	<0.5	N	1.85	0.95	1.48				0.12			The preceding annealed specimens heated for 30 min at 500 C, then quenched in water at about 8 C, and then measured after 4 to 5 hrs.
101	408	E	1925	300.2	<0.5	R	1.78	0.92					0.38			0.97 Cr; chill-cast.
102	408	E	1925	294.2	<0.5	R	1.78	0.92					0.38			0.87 Cr; the above specimen annealed for 30 min at 450 C.
103	408	E	1925	306.2	<0.5	X	1.78	0.55	1.56		1.01		0.29			Chill-cast.
104	408	E	1925	299.2	<0.5	X	1.78	0.55	1.56		1.01		0.29			The above specimen annealed for 30 min at 450 C.
105	408	E	1925	303.2	<0.5	X	1.78	0.55	1.56		1.01		0.28			The above annealed specimen heated for 30 min at 500 C, then quenched in water at about 8 C, and then measured after 4 to 5 hrs.
106	525	L	1938	323-623	7.0	1	92.4	4.01	0.22	1.57	0.01	1.94	0.21			Stamped and annealed at 180-210 C for several hrs.
107	525	L	1938	323-623	7.0	2	91.5	4.36	0.25	1.45	0.10	1.84	0.30			Cast and annealed at 180-210 C for several hrs.
108	525	L	1938	323-623	7.0	3	87.71	10.73	0.01	0.30		1.10	0.12	0.03		Trace Co; cast and annealed at 180-210 C for several hrs.
109	526	L	1958	311-645		2014-T6	3.9/5.0	1.0 Max	0.2/6.8	0.4/1.2			0.5/1.2	0.25 Max		0.10 Cr, 0.15 (Max) Ti; 0.15 total others; heat treated.
110	524	L	1960	4.0-120		2024-T4	4.58	0.1	1.7	0.1			0.1	0.1		0.1 Ga, 0.1 V, 0.05 Cr, 0.01 Sn, 0.001 Ca, 0.001 Zr, 0.001 Ag, 0.052 mm (longitudinal) and 0.048 mm (transverse).
111	527	L	1935	398,488		RR 53	2.25	1.4	1.6			1.3	1.25			0.1 Ti; as cast.
112	527	L	1935	398,488		RR 53	2.25	1.4	1.6			1.3	1.25			0.1 Ti; annealed at 175 C for 24 hrs.
113	527	L	1935	398,488		RR 53	2.25	1.4	1.6			1.3	1.25			0.1 Ti; annealed at 250 C for 24 hrs.
114	913	L	1965	20-573		Al-2014-T6	92.61	4.57	0.44	0.45	0.93		0.88			0.06 Zn, 0.04 Ti, 0.02 Cr.

SPECIFICATION TABLE NO. 242 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)							Composition (continued), Specifications and Remarks		
							Al	Cu	Fe	Mg	Mn	Ni	Si		Zn	
115	93	L	1931	87-476	3-4	K-S Alloy special	82.9	15.0	0.6	0.3	0.6	0.6	0.6	0.6	0.05	0.01 Cr, 0.02 Ti, 0.10 V, 0.16 Zr.
116	913	L	1964	48-573		Al-2219-T31	93.12	5.91	.21	0.01	0.28	0.13	0.05			0.01 Cr, 0.02 Ti, 0.10 V, 0.16 Zr.

Nominal composition; cast;  
 electrical conductivity  $48.1 \times 10^4$   
 $\text{ohm}^{-1}\text{cm}^{-1}$  at 87, 273, 373  
 and 476 K, respectively.

DATA TABLE NO. 242 THERMAL CONDUCTIVITY OF [ALUMINUM + COPPER + EX.] ALLOYS

(Al + Cu < 99.50% or at least one X<sub>i</sub> > 0.20%)

[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup>K<sup>-1</sup>]

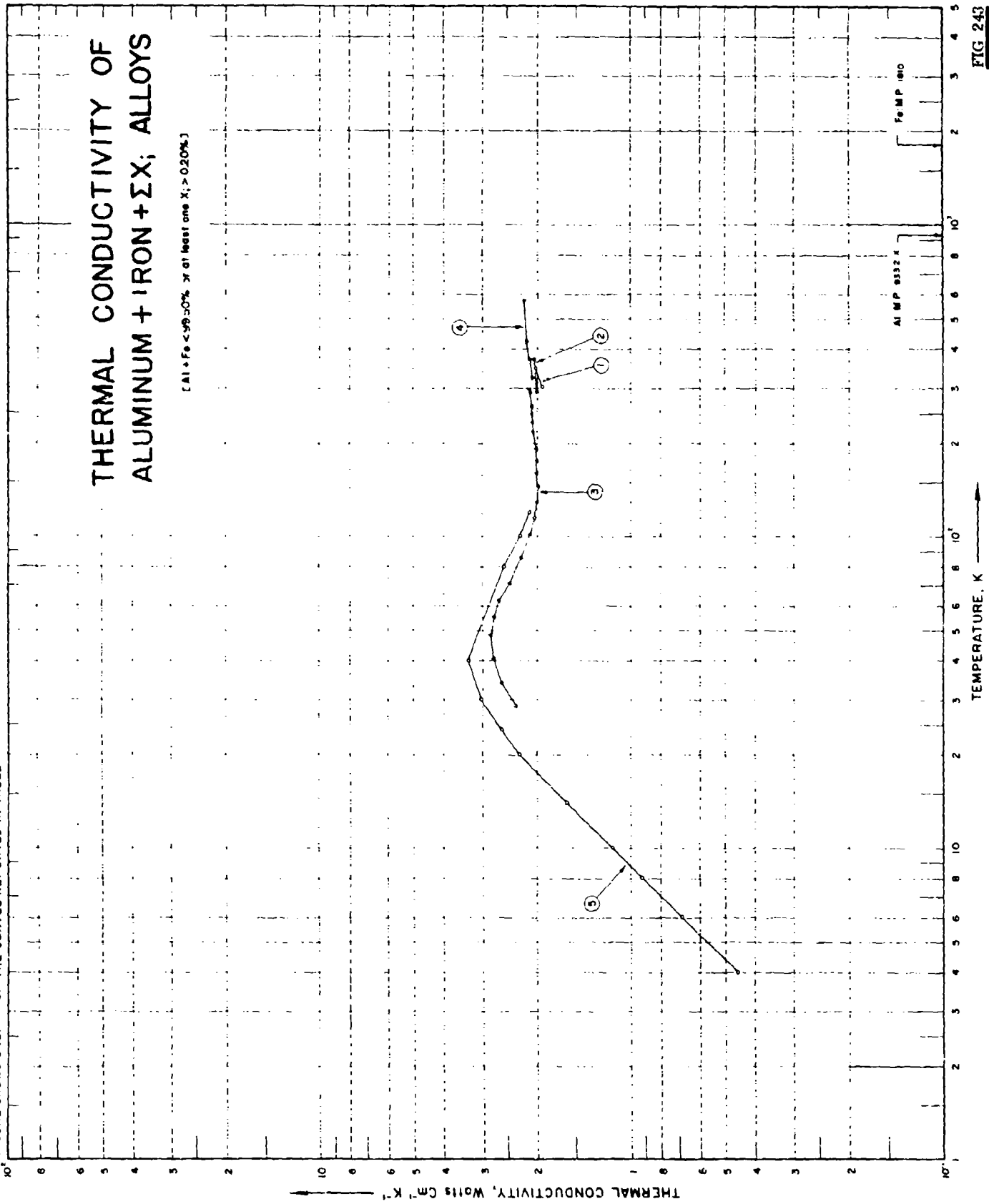
T	k	T	k	T	k	T	k	T	k	T	k	T	k	T	k	T	k	T	k	T	k
<u>CURVE 1</u>																					
353.00	1.590	423.00	1.632	473.00	1.674	523.00	1.715	573.00	1.757	623.00	1.799	673.00	1.841	723.00	1.883	773.00	1.925	823.00	1.967	873.00	2.009
<u>CURVE 2*</u>																					
353.00	1.590	423.00	1.632	473.00	1.674	523.00	1.715	573.00	1.757	623.00	1.799	673.00	1.841	723.00	1.883	773.00	1.925	823.00	1.967	873.00	2.009
<u>CURVE 3</u>																					
353.00	1.590	423.00	1.632	473.00	1.674	523.00	1.715	573.00	1.757	623.00	1.799	673.00	1.841	723.00	1.883	773.00	1.925	823.00	1.967	873.00	2.009
<u>CURVE 4</u>																					
353.00	1.590	423.00	1.632	473.00	1.674	523.00	1.715	573.00	1.757	623.00	1.799	673.00	1.841	723.00	1.883	773.00	1.925	823.00	1.967	873.00	2.009
<u>CURVE 5</u>																					
353.00	1.590	423.00	1.632	473.00	1.674	523.00	1.715	573.00	1.757	623.00	1.799	673.00	1.841	723.00	1.883	773.00	1.925	823.00	1.967	873.00	2.009
<u>CURVE 6*</u>																					
353.00	1.590	423.00	1.632	473.00	1.674	523.00	1.715	573.00	1.757	623.00	1.799	673.00	1.841	723.00	1.883	773.00	1.925	823.00	1.967	873.00	2.009
<u>CURVE 7*</u>																					
353.00	1.590	423.00	1.632	473.00	1.674	523.00	1.715	573.00	1.757	623.00	1.799	673.00	1.841	723.00	1.883	773.00	1.925	823.00	1.967	873.00	2.009
<u>CURVE 8</u>																					
353.00	1.590	423.00	1.632	473.00	1.674	523.00	1.715	573.00	1.757	623.00	1.799	673.00	1.841	723.00	1.883	773.00	1.925	823.00	1.967	873.00	2.009
<u>CURVE 9*</u>																					
353.00	1.674	423.00	1.715	473.00	1.757	523.00	1.799	573.00	1.841	623.00	1.883	673.00	1.925	723.00	1.967	773.00	2.009	823.00	2.051	873.00	2.093
<u>CURVE 10*</u>																					
353.00	1.674	423.00	1.715	473.00	1.757	523.00	1.799	573.00	1.841	623.00	1.883	673.00	1.925	723.00	1.967	773.00	2.009	823.00	2.051	873.00	2.093
<u>CURVE 11</u>																					
353.00	1.590	423.00	1.590	473.00	1.590	523.00	1.590	573.00	1.590	623.00	1.590	673.00	1.590	723.00	1.590	773.00	1.590	823.00	1.590	873.00	1.590
<u>CURVE 12*</u>																					
353.00	1.590	423.00	1.590	473.00	1.590	523.00	1.590	573.00	1.590	623.00	1.590	673.00	1.590	723.00	1.590	773.00	1.590	823.00	1.590	873.00	1.590
<u>CURVE 13*</u>																					
353.00	1.590	423.00	1.590	473.00	1.590	523.00	1.590	573.00	1.590	623.00	1.590	673.00	1.590	723.00	1.590	773.00	1.590	823.00	1.590	873.00	1.590
<u>CURVE 14*</u>																					
353.00	1.590	423.00	1.590	473.00	1.590	523.00	1.590	573.00	1.590	623.00	1.590	673.00	1.590	723.00	1.590	773.00	1.590	823.00	1.590	873.00	1.590
<u>CURVE 15</u>																					
353.00	1.632	423.00	1.715	473.00	1.757	523.00	1.799	573.00	1.841	623.00	1.883	673.00	1.925	723.00	1.967	773.00	2.009	823.00	2.051	873.00	2.093
<u>CURVE 16</u>																					
353.00	1.464	423.00	1.506	473.00	1.548	523.00	1.590	573.00	1.632	623.00	1.674	673.00	1.715	723.00	1.757	773.00	1.799	823.00	1.841	873.00	1.883
<u>CURVE 17</u>																					
353.00	1.117	423.00	1.169	473.00	1.221	523.00	1.273	573.00	1.325	623.00	1.377	673.00	1.429	723.00	1.481	773.00	1.533	823.00	1.585	873.00	1.637
<u>CURVE 18</u>																					
353.00	1.674	423.00	1.715	473.00	1.757	523.00	1.799	573.00	1.841	623.00	1.883	673.00	1.925	723.00	1.967	773.00	2.009	823.00	2.051	873.00	2.093
<u>CURVE 19</u>																					
353.00	1.632	423.00	1.715	473.00	1.757	523.00	1.799	573.00	1.841	623.00	1.883	673.00	1.925	723.00	1.967	773.00	2.009	823.00	2.051	873.00	2.093
<u>CURVE 20</u>																					
353.00	1.715	423.00	1.799	473.00	1.841	523.00	1.883	573.00	1.925	623.00	1.967	673.00	2.009	723.00	2.051	773.00	2.093	823.00	2.135	873.00	2.177
<u>CURVE 21</u>																					
380.20	1.962	472.20	1.996	575.20	2.050	678.20	2.104	781.20	2.158	884.20	2.212	987.20	2.266	1090.20	2.320	1193.20	2.374	1296.20	2.428	1399.20	2.482
<u>CURVE 22</u>																					
402.20	1.423	408.30	1.423	414.10	1.423	419.90	1.423	425.70	1.423	431.50	1.423	437.30	1.423	443.10	1.423	448.90	1.423	454.70	1.423	460.50	1.423
<u>CURVE 23</u>																					
427.50	1.435	434.60	1.485	441.70	1.485	448.80	1.485	455.90	1.485	463.00	1.485	470.10	1.485	477.20	1.485	484.30	1.485	491.40	1.485	498.50	1.485
<u>CURVE 24*</u>																					
431.40	1.607	455.40	1.596	479.40	1.585	503.40	1.574	527.40	1.563	551.40	1.552	575.40	1.541	599.40	1.530	623.40	1.519	647.40	1.508	671.40	1.497
<u>CURVE 25</u>																					
140.80	0.854	172.90	0.912	202.50	1.008	229.30	1.109	254.70	1.125	287.90	1.205	349.00	1.318	354.00	1.347	361.80	1.360	366.90	1.369	367.80	1.369
<u>CURVE 25</u>																					
140.80	0.854	172.90	0.912	202.50	1.008	229.30	1.109	254.70	1.125	287.90	1.205	349.00	1.318	354.00	1.347	361.80	1.360	366.90	1.369	367.80	1.369
<u>CURVE 25</u>																					
140.80	0.854	172.90	0.912	202.50	1.008	229.30	1.109	254.70	1.125	287.90	1.205	349.00	1.318	354.00	1.347	361.80	1.360	366.90	1.369	367.80	1.369
<u>CURVE 26</u>																					
172.90	0.912	202.50	1.008	229.30	1.109	254.70	1.125	287.90	1.205	349.00	1.318	354.00	1.347	361.80	1.360	366.90	1.369	367.80	1.369	374.50	1.351
<u>CURVE 26</u>																					
172.90	0.912	202.50	1.008	229.30	1.109	254.70	1.125	287.90	1.205	349.00	1.318	354.00	1.347	361.80	1.360	366.90	1.369	367.80	1.369	374.50	1.351
<u>CURVE 26</u>																					
172.90	0.912	202.50	1.008	229.30	1.109	254.70	1.125	287.90	1.205	349.00	1.318	354.00	1.347	361.80	1.360	366.90	1.369	367.80	1.369	374.50	1.351
<u>CURVE 26</u>																					
172.90	0.912	202.50	1.008	229.30	1.109	254.70	1.125	287.90	1.205	349.00	1.318	354.00	1.347	361.80	1.360	366.90	1.369	367.80	1.369	374.50	1.351
<u>CURVE 26</u>																					
172.90	0.912	202.50	1.008	229.30	1.109	254.70	1.125	287.90	1.205	349.00	1.318	354.00	1.347	361.80	1.360	366.90	1.369	367.80	1.369	374.50	1.351
<u>CURVE 26</u>																					
172.90	0.912	202.50	1.008	229.30	1.109	254.70	1.125	287.90	1.205	349.00	1.318	354.00	1.347	361.80	1.360	366.90	1.369	367.80	1.369	374.50	1.351
<u>CURVE 26</u>																					
172.90	0.912	202.50	1.008	229.30	1.109	254.70	1.125	287.90	1.205	349.00	1.318	354.00	1.347	361.80	1.360	366.90	1.369	367.80	1.369	374.50	1.351
<u>CURVE 26</u>																					
172.90	0.912	202.50	1.008	229.30	1.109	254.70	1.125	287.90	1.205	349.00	1.318	354.00	1.347	361.80	1.360	366.90	1.369	367.80	1.369	374.50	1.351
<u>CURVE 26</u>																					
172.90	0.912	202.50	1.008	229.30	1.109	254.70	1.125	287.90	1.205	349.00	1.318	354.00	1.347	361.80	1.360	366.90	1.369	367.80	1.369	374.50	1.351
<u>CURVE 26</u>																					
172.90	0.912	202.50	1.008	229.30	1.109	254.70	1.125	287.90	1.205	349.00	1.318	354.00	1.347	361.80	1.360	366.90	1.369	367.80	1.369	374.50	1.351
<u>CURVE 26</u>																					
172.90	0.912	202.50	1.008	229.30	1.109	254.70	1.125	287.90	1.205	349.00	1.318	354.00	1.347	361.80	1.360	366.90	1.369	367.80	1.369	374.50	1.351
<u>CURVE 26</u>																					
172.90	0.912	202.50	1.008	229.30	1.109	254.70	1.125	287.90	1.205	349.00	1.318	354.00	1.347	361.80	1.360	366.90	1.369	367.80	1.369	374.50	1.351
<u>CURVE 26</u>																					
172.90	0.912	202.50	1.008	229.30	1.109	254.70	1.125	287.90	1.205	349.00	1.318	354.00	1.347	361.80	1.360	366.90	1.369	367.80	1.369	374.50	1.351
<u>CURVE 26</u>																					
172.90	0.912	202.50	1.008	229.30	1.109	254.70	1.125	287.90	1.205	349.00	1.318	354.00	1.347	361.80	1.360	366.90	1.369	367.80	1.369	374.50	1.351
<u>CURVE 26</u>																					
172.90	0.912	202.50	1.008	229.30	1.109	254.70	1.125	287.90	1.205	349.00	1.318	354.00	1.347	361.80	1.360	366.90	1.369	367.80	1.369	374.50	1.351
<u>CURVE 26</u>																					
172.90	0.912	202.50	1.008	229.30	1.109	254.70	1.125	287.90	1.205	349.00	1.318	354.00	1.347	361.80	1.360	366.90	1.369	367.80	1.369	374.50	1.351
<u>CURVE 26</u>																					
172.90	0.912	202.50	1.008	229.30	1.109	254.70	1.125	287.90	1.205	349.00	1.318	354.00	1.347	361.80	1.360	366.90	1.369	367.80	1.369	374.50	1.351
<u>CURVE 26</u>																					
172.90	0.912	202.50	1.008																		

DATA TABLE NO. 242 (continued)

CURVE 50 (cont.)		CURVE 55		CURVE 65		CURVE 75		CURVE 85		CURVE 95		CURVE 105		CURVE 110		CURVE 114 (cont.)	
T	k	T	k	T	k	T	k	T	k	T	k	T	k	T	k	T	k
157.1	0.904	338.2	1.51	338.2	1.44	303.2	1.32	304.2	1.73	306.2	1.47	4	0.0315	473.2	1.88		
171.0	0.937	438.2	1.69									6	0.0489	523.2	1.94		
189.0	1.00											8	0.0661	573.2	1.98		
212.1	1.05	CURVE 58		CURVE 68		CURVE 78		CURVE 88		CURVE 98		CURVE 108		CURVE 115			
230.3	1.09	305.2	1.23	305.2	1.47	305.2	1.44	299.2	1.64	303.2	1.45	14	0.0882				
245.1	1.14	338.2	1.51									20	0.170				
267.5	1.19	438.2	1.56									24	0.203	87.0	1.138		
286.1	1.22											30	0.249	273.0	1.393		
296.1	1.24	CURVE 59		CURVE 69		CURVE 79		CURVE 89		CURVE 99		CURVE 109		CURVE 116			
		338.2	2.05	307.2	1.45	304.2	1.65	302.2	1.63	303.2	1.45	40	0.322	373.0	1.502		
		438.2	2.13									50	0.390	474.0	1.590		
293.2	1.69											60	0.452				
373.2	1.75	CURVE 60		CURVE 70		CURVE 80		CURVE 90		CURVE 100		CURVE 110		CURVE 116			
473.2	1.82	304.2	0.933	304.2	0.933	302.2	1.18	299.2	1.38	323.2	1.59	70	0.508	48.2	0.44		
573.2	1.94											80	0.557	73.2	0.56		
		338.2	1.93									100	0.645	98.2	0.71		
		438.2	1.97									120	0.727	123.2	0.77		
		CURVE 61		CURVE 71		CURVE 81		CURVE 91		CURVE 101		CURVE 111		CURVE 116			
293.2	1.35	298.2	1.33	298.2	1.33	299.2	1.51	308.2	1.32	323.2	1.46	398.2	1.32	148.2	0.93		
373.2	1.50											488.2	1.44	198.2	1.00		
473.2	1.74	CURVE 62		CURVE 72		CURVE 82		CURVE 92		CURVE 102		CURVE 112		CURVE 116			
		304.2	1.34	302.2	1.34	298.2	1.23	309.2	1.29	309.2	1.29	273.2	1.17	223.2	1.07		
												398.2	1.37	248.2	1.13		
		CURVE 63		CURVE 73		CURVE 83		CURVE 93		CURVE 103		CURVE 113		CURVE 116			
293.2	1.82	304.2	1.59	301.2	1.22	301.2	1.22	311.2	1.46	373.2	1.51	488.2	1.38	323.2	1.28		
373.2	1.88													373.2	1.33		
473.2	1.94													423.1	1.39		
573.2	1.94	CURVE 64		CURVE 74		CURVE 84		CURVE 94		CURVE 104		CURVE 114		CURVE 116			
		302.2	1.02	305.2	1.56	305.2	1.56	306.2	1.57	323.2	1.63	398.2	1.39	473.2	1.44		
												488.2	1.41	523.2	1.49		
		338.2	1.82											573.2	1.54		
		438.2	1.75														
		CURVE 65		CURVE 75		CURVE 85		CURVE 95		CURVE 105		CURVE 115		CURVE 116			
338.2	1.81	296.2	1.35	306.2	1.59	306.2	1.59	301.2	1.65	323.2	1.63	20.2	0.50				
438.2	2.10											48.2	0.82				
		CURVE 66		CURVE 76		CURVE 86		CURVE 96		CURVE 106		CURVE 116		CURVE 116			
		305.2	1.39	301.2	1.22	301.2	1.22	301.2	1.59	423.2	1.76	73.2	0.96				
338.2	1.63											98.2	1.07				
438.2	1.88	CURVE 67		CURVE 77		CURVE 87		CURVE 97		CURVE 107		CURVE 117		CURVE 116			
		302.2	1.67	300.2	1.52	300.2	1.52	300.2	1.04	323.2	1.84	123.2	1.16				
												148.2	1.25				
		338.2	1.64									173.2	1.32				
		438.2	1.69									198.2	1.35				
		CURVE 68		CURVE 78		CURVE 88		CURVE 98		CURVE 108		CURVE 118		CURVE 116			
338.2	1.69	302.2	1.67	300.2	1.52	300.2	1.52	300.2	1.04	311.0	1.54	223.2	1.45				
438.2	1.81									333.2	1.55	248.2	1.50				
		CURVE 69		CURVE 79		CURVE 89		CURVE 99		CURVE 109		CURVE 119		CURVE 116			
		299.2	1.32	303.2	1.48	303.2	1.48	291.2	1.09	422.2	1.64	273.2	1.56				
										519.5	1.48	323.2	1.66				
		338.2	1.62							616.8	1.73	373.2	1.74				
		438.2	1.59							644.6	1.74	423.1	1.81				

\* Not shown on plot

FIGURE SHOWS ONLY 5 OF THE CURVES REPORTED IN TABLE





## SPECIFICATION TABLE NO. 243 THERMAL CONDUCTIVITY OF [ALUMINUM - IRON - ZN] ALLOYS

(Al - Fe 99.50 or at least one Ni 0.20-9)

[For Data Reported in Figure and Table No. 243]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)							Composition (continued), Specifications and Remarks			
							Al	Fe	Cr	Cu	Mg	Mn	Si		Ti		
1	483	L	1922	302-347	+1.5	Al	98.87	0.75				0.34					
2	77	E	1900	281-373			99.4	0.7	0.4								
3	226	L	1951	29-297	±2.0	J 51	98.17	0.56	0.03	0.29	0.56	0.02	0.38	0.01			
4	528		1953	323-573		Cond-Al	Bal	0.53			0.32		0.10				Cast, hot-rolled above 750 F, annealed at 840 F for 4 hrs, cold-rolled by a 10% reduction, and then aged at 350 F for 3 hrs.
5	524	L	1960	4.0-120		1100-0	Bal	0.41	0.30	0.1	0.1	0.1	0.22	0.01			0.1 Ga, 0.01 Pb, 0.01 V, 0.001 Cu, 0.001 Zr, 0.0001 Bi, 0.01 Zn; average grain size 0.040 mm x 0.032 mm (longitudinal) x 0.036 mm (transverse).
6	224	L	1923	410.2		Al-1	99.49	0.3		0.07			0.3				0.07 Zn; hard-drawn.

DATA TABLE NO. 243 THERMAL CONDUCTIVITY OF ALUMINUM-IRON-EX<sub>12</sub> ALLOYS(Al + Fe) 99.50% or at least one X<sub>i</sub> > 0.20%Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup>K<sup>-1</sup>

T	k	T	k
<u>CURVE 1</u>			
301.8	1.94		
347.2	2.04		
<u>CURVE 2</u>			
291.2	2.010		
373.2	2.060		
<u>CURVE 3</u>			
281.66	2.351		
33.72	2.607		
40.46	2.774		
48.22	2.812		
55.09	2.733		
62.02	2.661		
70.48	2.469		
85.51	2.264		
102.26	2.105		
115.15	2.050		
129.51	2.004		
145.09	1.946		
160.18	1.894		
175.17	1.844		
190.17	1.801		
216.55	2.063		
233.05	2.067		
248.07	2.084		
262.05	2.092		
290.98	2.117		
296.96	2.125		
<u>CURVE 4</u>			
323.2	2.09		
373.2	2.13		
423.2	2.17		
573.2	2.21		
<u>CURVE 5</u>			
4	0.459		
6	0.638		
8	0.918		
10	1.145		
14	1.61		
20	2.28		
24	2.62		
30	3.04		
40	3.32		
80	2.57		
100	2.27		
120	2.12		
<u>CURVE 6*</u>			
410.2	1.906		

\* Not shown on plot

# THERMAL CONDUCTIVITY OF ALUMINUM + MAGNESIUM + ΣX; ALLOYS

(Al + Mg < 99.50%, or of least one X; > 0.20%)

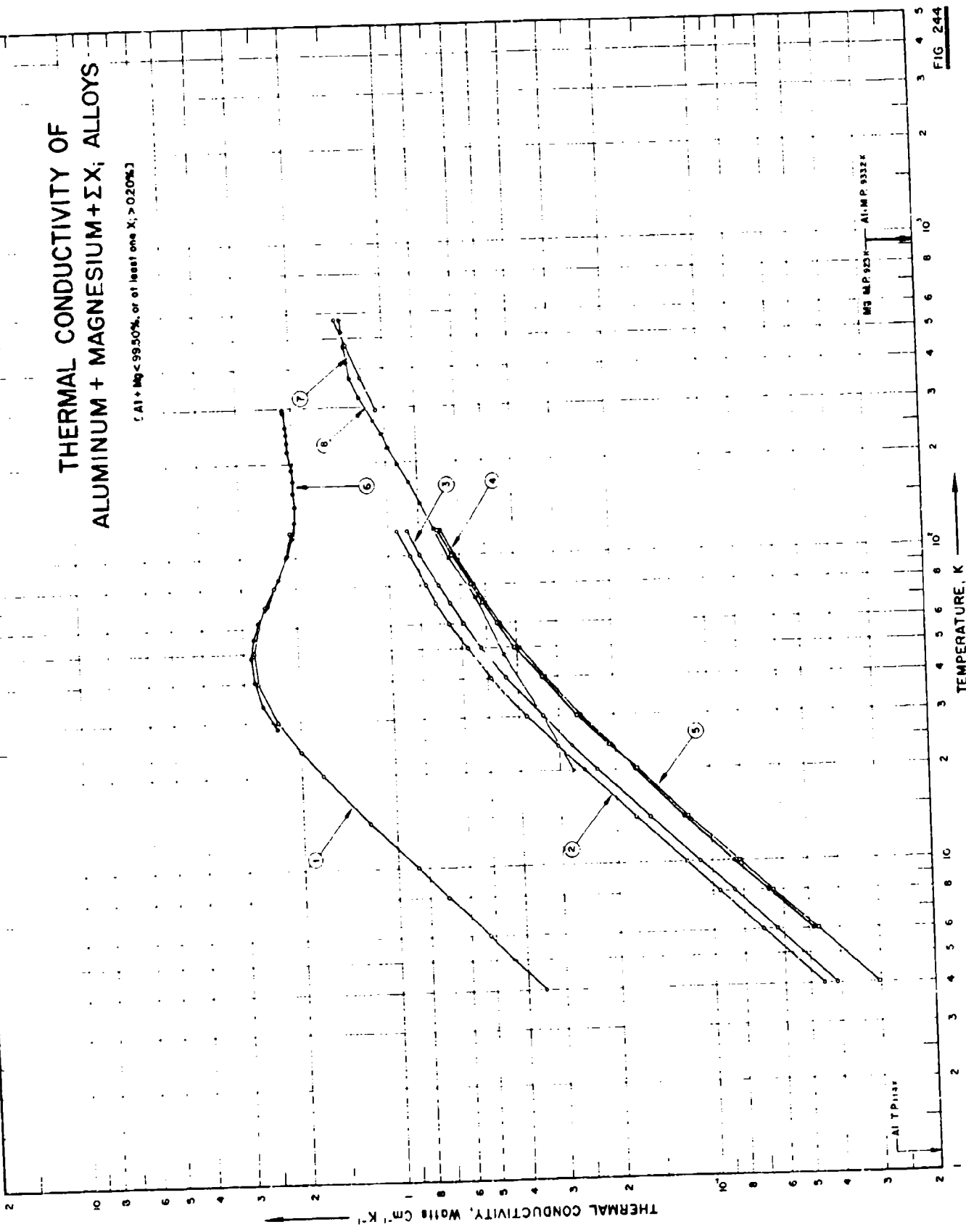


FIG. 244

## SPECIFICATION TABLE NO. 244 THERMAL CONDUCTIVITY OF ALUMINUM - MAGNESIUM - [5%] ALLOYS

(Al - Mg 99.50% or at least one  $N_1$  - 0.20%)

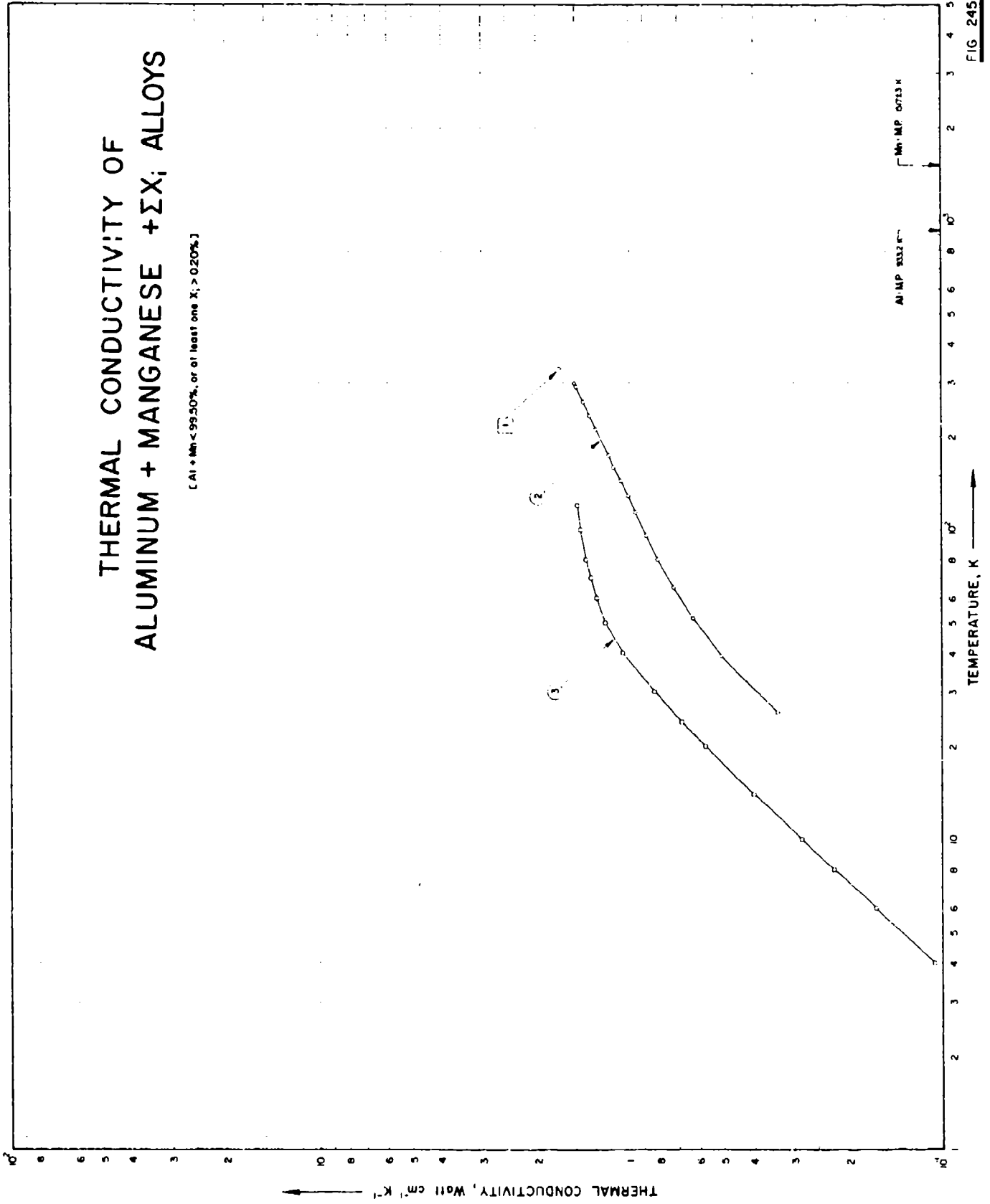
[ For Data Reported in Figure and Table No. 244 ]

Curve No.	ReL No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)							Composition (continued), Specifications and Remarks			
							Al	Mg	Cr	Cu	Fe	Mn	Si		Ti	Zn	
1	524	L	1960	4-120		6063- <del>T5</del> T5	Bal	0.65	0.01	0.01	0.1	0.1	0.36	0.01	0.01	0.01	0.1 Ga, 0.01 V, 0.001 Ca, 0.001 Pb; grain size 0.052 mm x 0.048 mm (longitudinal) and 0.052 mm (transverse); precipitation heat-treated.
2	524	L	1960	4-120		5052-O	Bal	2.46	0.22	0.1	0.1	0.1	0.1	0.01	0.01	0.1	0.1 Ga, 0.01 V, 0.001 Ca, 0.001 Zr; grain size 0.056 mm x 0.032 mm (longitudinal) and 0.040 mm (transverse); annealed in vacuum for 1 hr at 350 C.
3	524	L	1960	4-120		5154-O	Bal	3.32	0.21	0.1	0.1	0.1	0.1	0.01	0.01	0.01	0.01 V, 0.01 Zr, 0.001 Ca, 0.001 Pb; grain size 0.036 mm x 0.028 mm (longitudinal) and 0.032 mm (transverse); annealed in vacuum for 1 hr at 350 C.
4	524	L	1960	6-120		5083-O	Bal	4.44	0.1	0.04	0.1	0.7	0.1				Average crystal grain size 0.74 mm x 0.21 mm (longitudinal) and 0.54 mm x 0.14 mm (transverse); annealed in vacuum for 1 hr at 350 C.
5	524	L	1960	4-120		5086-F	Bal	4.10	0.1	0.07	0.28	0.51	0.1	0.02	0.1		Average crystal grain size 0.061 mm x 0.022 mm (longitudinal) and 0.096 mm x 0.020 mm (transverse).
6	226	L	1951	59-297	-2.0	J 51	98.17	0.56	0.01	6.29	0.56	0.02	0.38	0.01			1.20 Ni, 0.25 Co; cast; heated for 10 hrs at 160-170 C and cooled in air.
7	227	L	1949	293-573		RR 131D	94.87	1.33	0.18	0.30	0.30	0.44	0.50	0.12	0.45		
8	913	L	1964	20-573		Al 5456-H343	92.87	5.1	0.13	0.10		0.9		0.20	0.25		0.40 Fe and Si, 0.15 others.



# THERMAL CONDUCTIVITY OF ALUMINUM + MANGANESE + ΣX; ALLOYS

[Al + Mn < 99.50%, or of least one X; > 0.20%]



SPECIFICATION TABLE NO. 245 THERMAL CONDUCTIVITY OF [ALUMINUM + MANGANESE + EX.] ALLOYS  
(Al + Mn < 99.50% or at least one X<sub>1</sub> > 0.20%)

[For Data Reported in Figure and Table No. 245.]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)				Mg	Composition (continued), Specifications and Remarks
							Mn	Fe	Cu	Si		
1	230	L	1925	336.2			1.07	0.66	0.48	0.27		Specimen ~6 cm long with cross-section 0.3 cm <sup>2</sup> ; supplied by Aluminum Co. of America; electrical conductivity 23.30 x 10 <sup>4</sup> ohm <sup>-1</sup> cm <sup>-1</sup> at 23 C.
2	226	L	1951	26-300	< 2	3004	1.2	0.52	0.16	0.13	1.02	0.1 Ga, 0.01 each Ca, Ti, V and Zn, 0.001 each Bi, Pb and Zr; specimen drawn to 3.66 mm in dia; supplied by Aluminum Co. of America; "as fabricated" condition; average grain size 0.016 x 0.008 mm in longitudinal and 0.012 mm in the transverse directions.
3	524	L	1960	4.0-120		3003-F	1.23	0.48	0.1	0.15	0.1	

DATA TABLE NO. 245 THERMAL CONDUCTIVITY OF [ALUMINUM + MANGANESE +  $\Sigma X_i$ ] ALLOYS(Al + Mn < 99.50% or at least one  $X_i > 0.20\%$ )[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup> K<sup>-1</sup>]

T	k
<u>CURVE 1</u>	
336.0	1.695
<u>CURVE 2</u>	
25.74	0.339
39.17	0.610
51.91	0.628
65.18	0.724
80.22	0.812
96.02	0.883
114.88	0.958
129.98	1.013
145.04	1.067
160.04	1.125
174.96	1.167
212.26	1.280
234.93	1.343
260.22	1.406
291.44	1.477
299.87	1.490
<u>CURVE 3</u>	
4	0.107
6	0.163
8	0.222
10	0.282
14	0.402
20	0.575
24	0.682
30	0.833
40	1.06
50	1.20
60	1.28
70	1.33
80	1.38
100	1.44
120	1.47



FIGURE SHOWS ONLY 9 OF THE CURVES REPORTED IN TABLE

**THERMAL CONDUCTIVITY OF  
ALUMINUM + NICKEL +  $\Sigma X_i$  ALLOYS**

[ Al + Ni < 99.50%, or of least one  $X_i$  > 0.20% ]

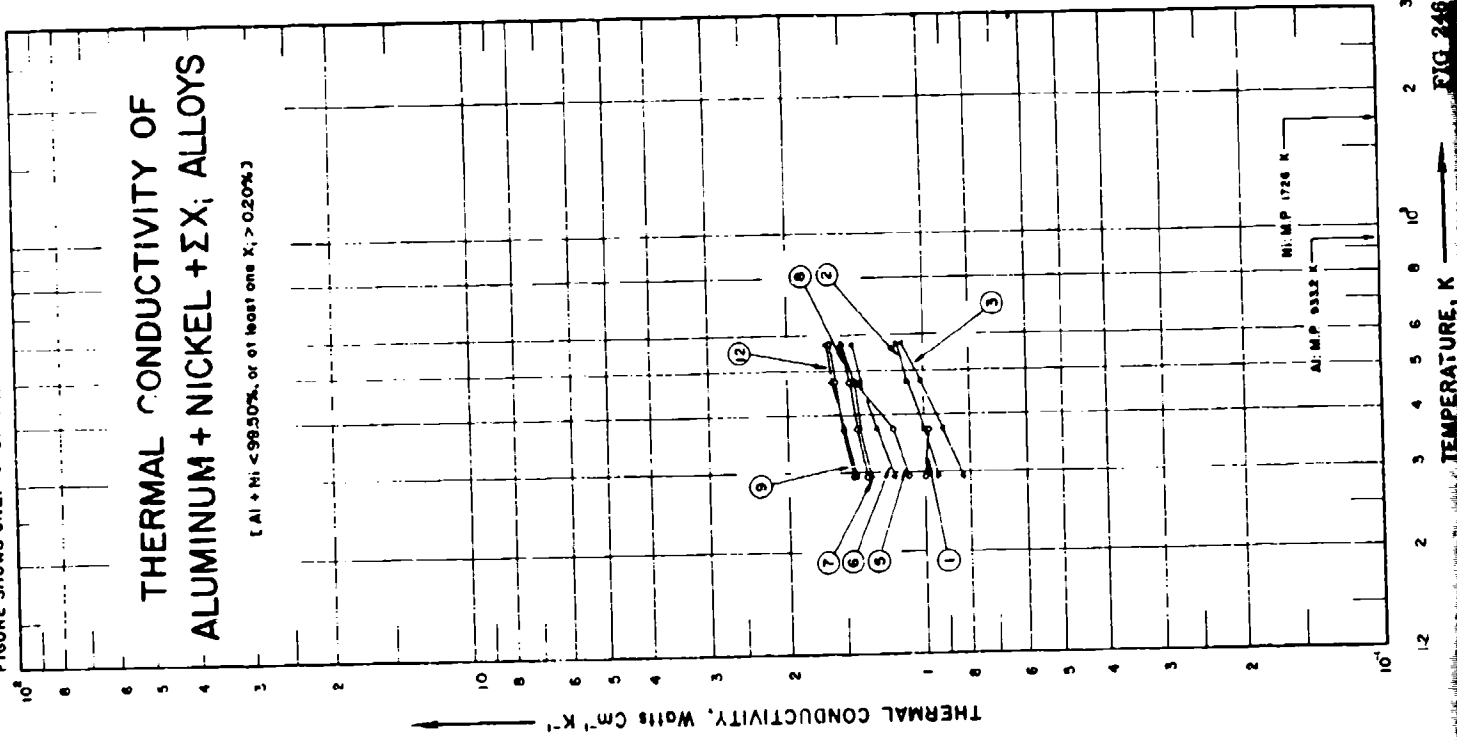


FIG. 246

SPECIFICATION TABLE NO. 246 THERMAL CONDUCTIVITY OF [ALUMINUM + NICKEL +  $\Sigma X_i$ ] ALLOYS(Al + Ni < 99.50% or at least one  $X_i$  > 0.20%)

[For Data Reported in Figure and Table No. 246]

Curve No.	Ref. Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight per cent)										Composition (continued), Specifications and Remarks
						Ni	Cr	Cu	Fe	Mg	Mn	Si	Ti	Composition (continued), Specifications and Remarks		
1	227	L	1949	293-373	RAE 40 C	5.0	0.5	2.0	0.5	0.5	3.0	0.3			0.4 Be, sand-cast; heated 6 hrs at 570 + 5 C and quenched in cold water, and then heated 20 hrs at 150 C and cooled in air.	
2	227	L	1949	293-573	RAE 40 C	5.0	0.5	2.0	0.5	0.5	3.0	0.3			0.4 Be, the above specimen again heated at 300 C.	
3	227	L	1949	293-573	RAE 47 $\Phi$ D	4.0	1.0	1.0	0.5	0.5	3.0	0.2	0.2		Sand-cast; heated at 300 C.	
4	227	L	1949	293-573	RAE 47 $\Phi$ D	4.0	1.0	1.0	0.5	0.5	3.0	0.2	0.2		Chill-cast; heated at 300 C.	
5	227	L	1949	293-473	RAE 55	2.90	0.15	1.89	0.43	0.56	1.55	0.21	0.07		Chill-cast; heated in solution 4 hrs at 570 C and quenched in boiling water, and then heated 12 hrs at 260 C and cooled in air.	
6	227	L	1949	293-573	RAE 55	2.90	0.15	1.89	0.43	0.56	1.55	0.21	0.07		The above specimen again heat-treated at 300 C.	
7	227	L	1949	293-573	RAE 55	2.90	0.15	1.89	0.43	0.56	1.55	0.21	0.07		The above specimen again heat-treated at 400 C.	
8	227	L	1949	293-573	RAE 40 C	5.0	0.5	2.0	0.5	0.5	3.0	0.3			Wrought; heated 6 hrs at 570 + 5 C and quenched in cold water, then heated 20 hrs at 150 C and cooled in air, and again heat-treated at 300 C.	
9	227	L	1949	293-573	RAE 47 D	4.0	1.0	1.0	0.5	0.5	3.0	0.3			0.4 Be; wrought; heated 6 hrs at 570 + 5 C and quenched in cold water, then heated 20 hrs at 160 C and cooled in air, and again heat-treated at 300 C.	
10	227	L	1949	293-473	RAE 55(Bar 39 A)	2.85	0.49	1.67	0.41		2.02	0.17	0.07		Wrought; heated in solution 4 hrs at 570 C and quenched in boiling water, and then aged 40 hrs at 160 C and cooled in air.	
11	227	L	1949	293-573	RAE 55(Bar 33 A)	2.85	0.49	1.67	0.41		2.02	0.17	0.07		The above specimen again heat-treated at 300 C.	
12	227	L	1949	293-573	RAE 55(Bar 39 A)	3.01	0.17	1.69	0.40	0.49	1.41	0.15	0.03		Wrought; heated in solution 4 hrs at 570 C and quenched in boiling water, then aged 40 hrs at 160 C and cooled in air, and again heat-treated at 300 C.	

DATA TABLE NO. 246 THERMAL CONDUCTIVITY OF [ALUMINUM + NICKEL +  $\Sigma X_i$ ] ALLOYS(Al + Ni < 99.50% or at least one  $X_i > 0.20\%$ )[Temperature, T, K. Thermal Conductivity, k, Watts  $\text{cm}^{-1}\text{K}^{-1}$ ]

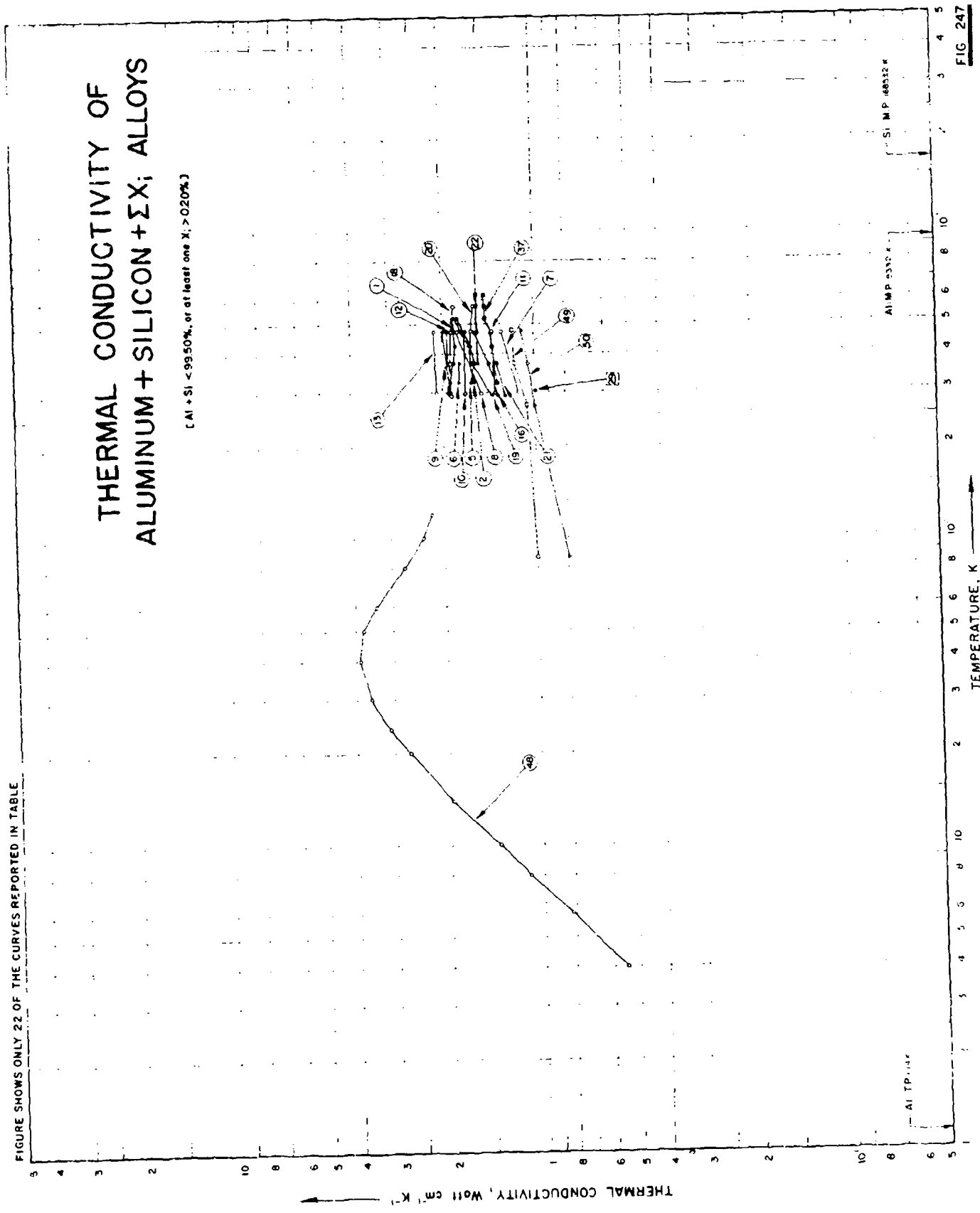
T	k	T	k
<u>CURVE 1</u>			
293.20	1.004	293.20	1.339
373.20	0.983	373.20	1.402
		473.20	1.464
		573.20	1.506
<u>CURVE 2</u>			
293.20	0.941	293.20	1.443
373.20	1.004	373.20	1.506
473.20	1.088	473.20	1.569
573.20	1.151	573.20	1.611
<u>CURVE 3</u>			
293.20	0.837	293.20	1.172
373.20	0.920	373.20	1.297
473.20	1.025	473.20	1.423
573.20	1.130		
<u>CURVE 4*</u>			
293.20	0.837	293.20	1.297
273.20	0.941	373.20	1.381
473.20	1.025	473.20	1.464
573.20	1.109	573.20	1.548
<u>CURVE 5</u>			
293.20	1.086	293.20	1.423
373.20	1.172	373.20	1.506
473.20	1.423	473.20	1.590
		573.20	1.632
<u>CURVE 6</u>			
293.20	1.172		
373.20	1.276		
473.20	1.381		
573.20	1.443		
<u>CURVE 7</u>			
293.20	1.318		
373.20	1.381		
473.20	1.423		
573.20	1.506		

\* Not shown on plot

FIGURE SHOWS ONLY 22 OF THE CURVES REPORTED IN TABLE

# THERMAL CONDUCTIVITY OF ALUMINUM + SILICON + $\Sigma$ X; ALLOYS

[Al + Si < 99.50%, or at least one X; > 0.20%]



## SPECIFICATION TABLE NO. 247 THERMAL CONDUCTIVITY OF [ALUMINUM + SILICON + EX.] ALLOYS

(Al + Si < 99.50% or at least one X<sub>i</sub> > 0.20%)

[For Data Reported in Figure and Table No. 247.]

Curve No.	Ref. Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight per cent)							Composition (continued), Specifications and Remarks
						Si	Cu	Fe	Mn	Ni	Ti	Zn	
1	20	L	324-523	<3.0	5	5.5	1.43	0.41	0.42	0.27	0.04	0.14	Annealed at 300 - 500 C; molten metal not fluxed with gases.
2	20	L	325-524	<3.0	6	5.5	1.43	0.41	0.42	0.27	0.04	0.14	Water-quenched from 520 C; molten metal not fluxed with gases.
3	20	L	373	<3.0	6a	5.5	1.43	0.41	0.42	0.27	0.04	0.14	Water-quenched from 520 C, then drawn at 550 C; molten metal not fluxed with gases.
4	20	L	373-426	<3.0	7	5.5	1.43	0.41	0.42	0.27	0.04	0.14	Annealed at 300 - 500 C; molten metal strongly fluxed with gases.
5	20	L	323-523	<3.0	8	5.5	1.43	0.41	0.42	0.27	0.04	0.14	Water-quenched from 520 C; molten metal strongly fluxed with gases.
6	20	L	323-373	<3.0	8a	5.5	1.43	0.41	0.42	0.27	0.04	0.14	Water-quenched from 520 C, then drawn at 550 C; molten metal strongly fluxed with gases.
7	223	L	298, 473	+4.0	7542	13.8	0.75	1.09	1.18	2.45			Chill-cast.
8	223	L	298, 473	+4.0	7628	5.04	0.95	0.36	0.34				Cast, solution-treated, and aged.
9	223	L	298, 473	+4.0	7642	0.91	0.5	0.58	0.5				Forged, solution-treated, and precipitation-treated.
10	223	L	298, 473	+4.0	7679	11.78	0.84	0.76	1.06	0.92			Forged, solution-treated, and precipitation-treated.
11	223	L	298, 473	+4.0	7542a	13.8	0.75	1.09	1.18	2.45			Annealed at 700 F.
12	223	L	298, 473	+4.0	7628a	5.04	0.95	0.36	0.34				Annealed at 700 F.
13	223	L	298, 473	+4.0	7642a	0.91	0.5	0.58	0.5				Annealed at 700 F.
14	223	L	298, 473	+4.0	7679a	11.78	0.84	0.76	1.06	0.92			Annealed at 700 F.
15	227	L	293-573		RR50	2.25	1.40	1.18	0.12	0.90	0.19		Cast; heated 10 hrs at 160 - 170 C and air-cooled; and then heat-treated at 300 C.
16	227	L	293-473		RR53c	2.42	1.33	1.12	0.50	0.87	0.15		Cast; heated 2 hrs at 530 C and water-quenched, and again heated 15 hrs at 160 - 170 C.
17	227	L	293-573		RR53c	2.42	1.33	1.12	0.50	0.87	0.15		The above specimen again heat-treated at 300 C.
18	227	L	293-573		Alpax Gamma	12.0		0.28	0.35	0.29			Cast; heated 4 hrs at 510 - 518 C and quenched in cold water and then heated 16 hrs at 150 - 165 C, and again heat-treated at 300 C.
19	227	L	293, 373	+7%	SA 1	11.0	5.0	0.5	0.6	0.05			0.2 Co; chill-cast; heated 3 hrs at 495 - 500 C and quenched in cold water, and then aged 16 hrs at 165 C and air-cooled.
20	227	L	293-573	+7%	SA 1	11.0	5.0	0.5	0.6	0.05			0.2 Co; the above specimen again heat-treated at 300 C.
21	227	L	293, 373	+4%	SA 44	11.9	5.0	0.5	0.5	0.1			0.3 Co; chill-cast; heated 3 hrs at 495 - 500 C and quenched in cold water and then aged 16 hrs at 165 C and air-cooled.

SPECIFICATION TABLE NO. 247 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight per cent)							Composition (continued), Specifications and Remarks
							Si	Cu	Fe	Mg	Mn	Ni	Ti	
22	227	L	1949	293-573	+4%	SA 44	11.0	5.0	0.5	0.5	0.4	0.1	0.1	0.3 Co; the above specimen again heat-treated at 300 C.
23	227	L	1949	293-473		Lo Ex	11.80	1.03	0.50	0.91	0.03	1.02	0.02	Wrought; heated 12 hrs at 522 C and aged 4 hrs at 135 C and cooled in air, and again aged at 200 C and cooled in air.
24	227	L	1949	293-473		Lo Ex	11.80	1.03	0.50	0.90	0.03	1.02	0.02	Wrought; heated 12 hrs at 522 C and aged 4 hrs at 135 C and cooled in air, and then aged at 200 C and cooled in air, and again heat-treated at 300 C.
25	227	L	1949	293-372	+4%	SA 1	11.0	5.0	0.5	0.6		0.05		0.2 Co; wrought; heated 3 hrs at 495-500 C and quenched in cold water, and then aged 16 hrs at 165 C and cooled in air.
26	227	L	1949	293-573	+4%	SA 1	11.0	5.0	0.5	0.6		0.05		0.2 Co; the above specimen again heat-treated at 300 C.
27	227	L	1949	293-273	+8%	SA 44	11.0	5.0	0.5	0.5	0.4	0.1		0.3 Co; wrought; heated 3 hrs at 495-500 C and quenched in cold water, and then aged 16 hrs at 165 C and cooled in air.
28	227	L	1949	293-573	+8%	SA 44	11.0	5.0	0.5	0.5	0.4	0.1		0.3 Co; the above specimen again heat-treated at 300 C.
29	408	E	1925	304.2	<0.5	K	6.13	3.8	0.92	1.58	0.58			Chill-cast.
30	408	E	1925	296.2	<0.5	K	6.13	3.8	0.92	1.58	0.58			The above specimen annealed for 30 min at 450 C.
31	408	E	1925	296.2	<0.5	K	6.13	3.8	0.92	1.58	0.58			The above annealed specimen heated for 30 min at 500 C, then quenched in water at about 8 C, and then measured after 4 to 5 hrs.
32	408	E	1925	311.2	<0.5	K	6.13	3.8	0.92	1.58	0.58			The above quenched specimen measured after aging for 2 weeks.
33	408	E	1925	302.2	<0.5	S	11.88		0.80					Chill-cast.
34	408	E	1925	301.2	<0.5	S	11.88		0.80					The above specimen annealed for 30 min at 450 C.
35	408	E	1925	300.2	<0.5	S	11.88		0.80					The above annealed specimen heated for 30 min at 500 C, then quenched in water at about 8 C, and then measured after 4 to 5 hrs.
36	408	E	1925	304.2	<0.5	S	11.88		0.80					The above quenched specimen measured after aging for 2 weeks.
37	525	L	1937	323-623	7.0	4	11.0	0.95	0.25	0.55	0.04	0.93		85.87 Al, 0.01 Co; stamped and annealed at 180-210 C for several hrs.
38	525	L	1937	323-623	7.0	5	12.45	1.21	0.35	0.74	0.05	0.94		84.26 Al; cast and annealed at 180-210 C for several hrs.
39	525	L	1937	323-623	7.0	6	17.36	1.47	0.61	0.67	1.73	1.48		0.05 75.44 Al, 1.18 Co; cast and annealed at 460 C.
40	527	L	1935	398, 488		RR 50	2.2	1.3	1.0	0.1	1.3	0.18		Cast.
41	527	L	1935	398, 488		RR 50	2.2	1.3	1.0	0.1	1.3	0.18		Annealed at 175 C for 24 hrs.

SPECIFICATION TABLE NO. 21E (continued)

Curve No.	Ref. No.	Method Used	Year	Tensile Range, K	Rept'd Ref.	Name and Specimen Designation	Composition (weight percent)							Composition (continued), Specifications and Remarks	
							Si	Cu	Fe	Mg	Mn	Ni	Ti		Zn
42	527	L	1935	398, 488		RR 59	2.2	1.1	1.0	0.1	1.3	0.18			Annealed at 250 C for 24 hrs.
43	527	L	1935	398, 488		$\gamma$ -Sulamin modified	12.0	1.2	0.5	0.5					Tempered.
44	527	L	1935	398, 488		$\gamma$ -Sulamin modified	12.0	1.2	0.5	0.5					Tempered and annealed at 175 C for 24 hrs.
45	527	L	1935	398, 488		$\gamma$ -Sulamin modified	12.0	1.2	0.5	0.5					Tempered and annealed at 250 C for 24 hrs.
46	527	L	1935	398, 488		Silumin sodium modified	12.0	1.2	0.5	0.5					0.06 Na; sand-cast; porous rod; specific weight 2.57.
47	527	L	1935	398, 488		Silumin sodium modified	12.0	1.2	0.5	0.5					0.06 Na; sand-cast; porous rod; specific weight 2.67.
48	524	L	1960	4.0-120		1100 F	0.13	0.1	0.1	0.1	0.99	0.01			0.1 Ga, 0.1 V, 0.02 Cr, 0.01 Pb, 0.01 Sn, 0.001 Ca, 0.001 Zr; drawn specimen 3.66 mm dia; avg grain size 0.024 mm x 0.098 mm (longitudinal) and 0.012 mm (transverse).
49	53	L	1931	87-476	3-4	K-S alloy 245	14	4.5	0.7	1.0	1.5				Nominal composition, cast; electrical conductivity 27.5, 12.5, 9.9 and $8.0 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 87, 273, 373 and 476 K respectively.
50	9.	L	1931	87-476	3-4	K-S alloy 240	21	1.5	0.5	0.7					1.2 Co; similar to the above specimens except electrical conductivity 27.8, 9.61, 6.0 and $5.54 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 87, 273, 373, and 476 K respectively.

DATA TABLE NO. 247 THERMAL CONDUCTIVITY OF [ALUMINUM + SILICON +  $\Sigma X_i$ ] ALLOYS

(Al + Si < 99.50% or at least one  $X_i > 0.20\%$ )

[Temperature, T, K; Thermal Conductivity, k, Watts  $cm^{-1}K^{-1}$ ]

T	k	T	k	T	k	T	k	T	k	T	k	T	k		
<u>CURVE 1</u>															
324.2	1.863	298.2	1.865	473.2	1.736	293.2	1.297	300.2	1.732	398.2	1.615	87.0	0.996		
374.2	1.837	473.2	1.987	573.2	1.736	372.2	1.381	486.2	1.644	486.2	1.644	273.0	1.067		
423.2	1.812	<u>CURVE 17 (COBL)*</u>													
473.2	1.833	<u>CURVE 18</u>													
523.2	1.841	293.2	1.883	<u>CURVE 25*</u>											
<u>CURVE 2</u>															
298.2	1.674	293.2	1.799	<u>CURVE 35*</u>											
473.2	1.690	373.2	1.757	<u>CURVE 36*</u>											
<u>CURVE 11</u>															
473.2	1.862	473.2	1.757	<u>CURVE 37</u>											
573.2	1.841	573.2	1.757	323.2	1.34	<u>CURVE 42*</u>									
<u>CURVE 19</u>															
293.2	1.381	293.2	1.339	373.2	1.34*	<u>CURVE 43*</u>									
473.2	1.389	373.2	1.423	423.2	1.38	<u>CURVE 44*</u>									
524.2	1.824	473.2	1.388	473.2	1.38	<u>CURVE 45*</u>									
<u>CURVE 12</u>															
293.2	1.900	293.2	1.423	523.2	1.46	<u>CURVE 46*</u>									
473.2	1.946	373.2	1.423	623.2	1.46	<u>CURVE 47*</u>									
<u>CURVE 3*</u>															
373.2	1.791	<u>CURVE 28*</u>													
<u>CURVE 4*</u>															
293.2	1.632*	<u>CURVE 38*</u>													
373.2	1.632*	293.2	1.632	<u>CURVE 48</u>											
473.2	1.632	373.2	1.674	323.2	1.34	<u>CURVE 46*</u>									
573.2	1.590	473.2	1.715	373.2	1.38	<u>CURVE 46*</u>									
<u>CURVE 21</u>															
293.2	1.255	293.2	1.255	423.2	1.38	<u>CURVE 46*</u>									
373.2	1.339	373.2	1.339	473.2	1.36	<u>CURVE 46*</u>									
<u>CURVE 22</u>															
298.2	1.778	298.2	1.356	473.2	1.30	<u>CURVE 47*</u>									
473.2	1.749	304.2	0.996	523.2	1.42	<u>CURVE 47*</u>									
<u>CURVE 14*</u>															
298.2	1.778	304.2	1.42	573.2	1.42	<u>CURVE 47*</u>									
473.2	1.749	623.2	1.46	623.2	1.46	<u>CURVE 47*</u>									
<u>CURVE 15*</u>															
293.2	2.611	<u>CURVE 39*</u>													
373.2	1.674	323.2	1.17	<u>CURVE 48</u>											
473.2	1.715	373.2	1.21	4	0.554	<u>CURVE 48</u>									
573.2	1.757	423.2	1.26	6	0.823	<u>CURVE 48</u>									
<u>CURVE 16</u>															
293.2	1.590	473.2	1.548	8	1.107	<u>CURVE 48</u>									
373.2	1.339	573.2	1.548	10	1.395	<u>CURVE 48</u>									
473.2	1.569	623.2	1.38	14	1.96	<u>CURVE 48</u>									
<u>CURVE 17*</u>															
293.2	1.632	<u>CURVE 40*</u>													
373.2	1.611	311.2	1.139	20	2.70	<u>CURVE 40*</u>									
473.2	1.423	373.2	1.611	24	3.12	<u>CURVE 40*</u>									
573.2	1.569	473.2	1.674	30	3.58	<u>CURVE 40*</u>									
<u>CURVE 7</u>															
298.2	1.138	398.2	1.477	40	3.89	<u>CURVE 40*</u>									
473.2	1.285	486.2	1.636	50	3.78	<u>CURVE 40*</u>									
<u>CURVE 9</u>															
298.2	1.498	302.2	1.310	60	3.40	<u>CURVE 40*</u>									
473.2	1.778	<u>CURVE 41*</u>													
<u>CURVE 5</u>															
323.2	1.615	398.2	1.410	80	2.74	<u>CURVE 41*</u>									
373.2	1.565	486.2	1.464	100	2.38	<u>CURVE 41*</u>									
424.2	1.619	<u>CURVE 41*</u>													
472.2	1.699	398.2	1.583	120	2.23	<u>CURVE 41*</u>									
523.2	1.782	486.2	1.586	<u>CURVE 41*</u>											
<u>CURVE 6</u>															
323.2	1.774	<u>CURVE 41*</u>													
373.2	1.761	<u>CURVE 41*</u>													
<u>CURVE 8</u>															
298.2	1.498	<u>CURVE 41*</u>													
473.2	1.778	<u>CURVE 41*</u>													

\* Not shown on plot



FIGURE SHOWS ONLY 10 OF THE CURVES REPORTED IN TABLE

# THERMAL CONDUCTIVITY OF ALUMINUM + ZINC + ΣX; ALLOYS

[Al + Zn < 99.50%, or at least one X; > 0.20%]

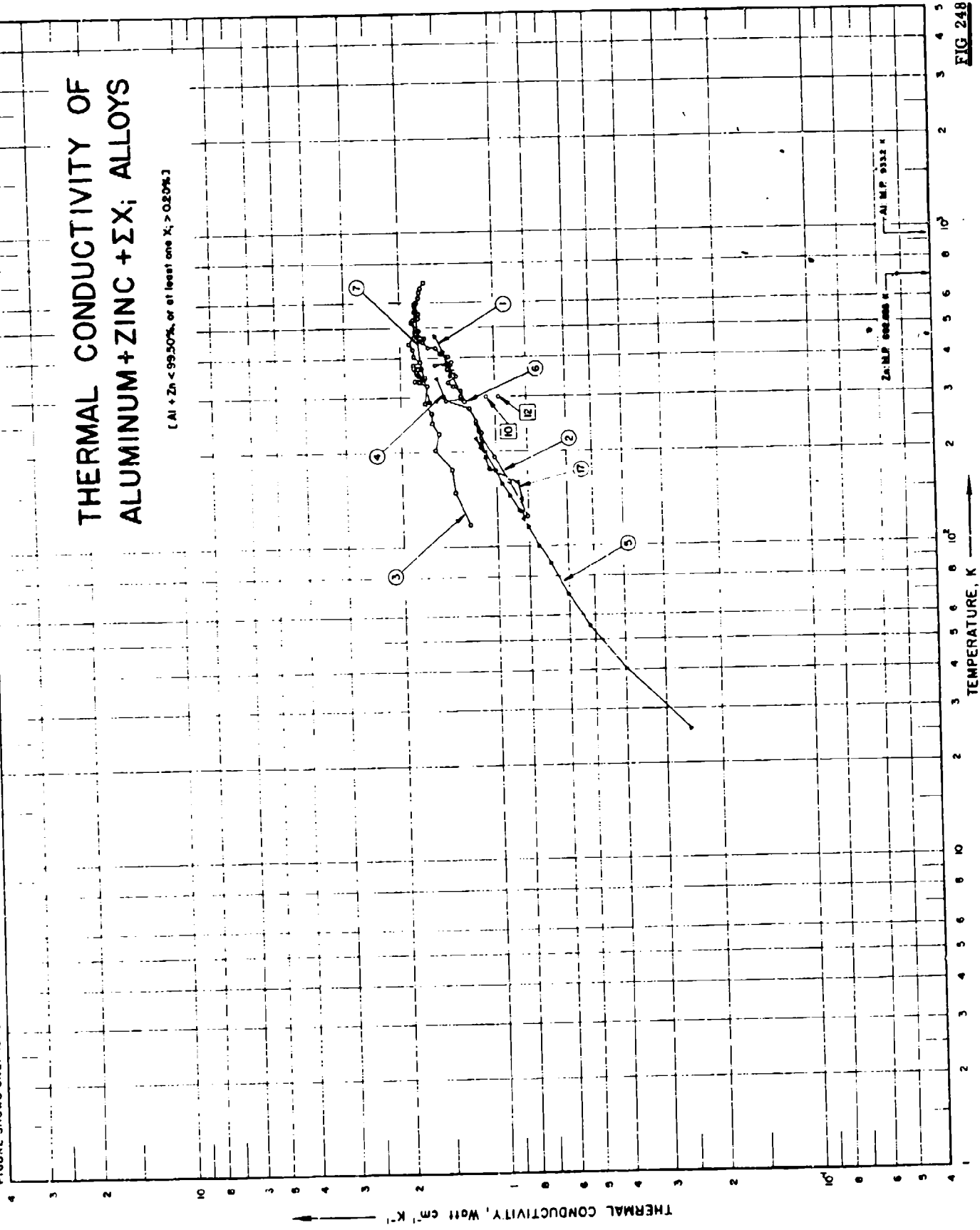


FIG. 248

SPECIFICATION TABLE NO. 248 THERMAL CONDUCTIVITY OF [ALUMINUM + ZINC +  $\Sigma X_i$ ] ALLOYS

(Al + Zn < 99.50% or at least one  $X_i > 0.20\%$ )

[For Data Reported in Figure and Table No. 248]

Curve No.	Ref. Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)							Ti	Composition (continued), Specifications and Remarks		
						Zn	Cr	Cu	Fe	Mg	Mn	Si				
1	55	P L	1928	353-473		Bar 661	13.0		0.3						Cast and annealed.	
2	91	C	1951	125-609		7075	5.6	0.3	1.6		2.5				As received.	
3	91	C	1951	117-702		7075	5.6	0.3	1.6		2.5				After heated at about 275 C.	
4	25		1921	301, 346			10.0		2.0						Nominal composition.	
5	226	L	1951	26-250	<2.0	7075	5.5	0.3	1.5		2.5	0.2			Wrought; 2 hrs solution heat treatment at 450 C and quenched in water at 70 C, and then aged 4 hrs at 135 C and cooled in air.	
6	227	L	1949	293, 373		RR 77	4.96		2.20	0.31	2.54	0.54	0.26		Trace	
7	227	L	1949	293-373		RR 77	4.96		2.20	0.31	2.54	0.54	0.26		Trace	The above specimen again heated at 300 C.
9	408	E	1925	307	<0.5	British L-5	12.02		2.70	0.57			0.39			Chill-cast.
9	408	E	1925	301	<0.5	British L-5	12.02		2.70	0.57			0.39			The above specimen annealed for 30 min at 450 C.
10	408	E	1925	301	<0.5	A	20.32		2.57	0.57			0.37			Chill-cast.
11	408	E	1925	300	<0.5	A	20.32		2.57	0.57			0.37			The above specimen annealed for 30 min at 450 C.
12	408	E	1925	300	<0.5	A	20.32		2.57	0.57			0.37			The above annealed specimen heated for 30 min at 500 C, quenched in water at about 8 C, and then measured after 4 to 5 hrs.
13	408	E	1925	305	<0.5	A	20.32		2.57	0.57			0.37			The above quenched specimen measured after aging for 2 weeks.
14	408	E	1925	304	<0.5	G	2.55		2.53	0.84	Trace	0.46	0.32			Chill-cast.
15	408	E	1925	301	<0.5	G	2.55		2.53	0.84	Trace	0.46	0.32			The above specimen annealed for 30 min at 450 C.
16	408	E	1925	302	<0.5	G	2.55		2.53	0.84	Trace	0.46	0.32			The above annealed specimen heated for 30 min at 500 C, quenched in water at about 8 C, and then measured after 4 to 5 hrs.
17	685	L	1963	122-221		7075-T6	5.1	0.18	1.2	0.7	2.1	0.30	0.50	0.20		Nominal composition; cross sectional area 0.105 cm <sup>2</sup> and 2.55 cm long.
							6.1	0.40	2.0		2.9					

DATA TABLE NO. 248 THERMAL CONDUCTIVITY OF (ALUMINUM + ZINC + X<sub>1</sub>) ALLOYS(Al + Zn < 99.50% or at least one X<sub>1</sub> > 0.20%)[Temperature, T, K; Thermal Conductivity, K, Watt cm<sup>-1</sup>K<sup>-1</sup>]

T	k	T	k	T	k	T	k	T	k
<u>CURVE 1</u>		<u>CURVE 3</u>		<u>CURVE 5</u>		<u>CURVE 11*</u>			
353.00	1.339	117.30	1.250	25.76	0.251	300.2	1.08		
423.00	1.464	148.30	1.356	40.22	0.397	<u>CURVE 12</u>			
473.00	1.548	177.30	1.389	55.18	0.519				
<u>CURVE 2</u>		204.50	1.561	70.24	0.607				
125.10	0.803	230.00	1.519	88.03	0.682	300.2	0.579		
161.50	0.916	249.90	1.590	100.18	0.745	<u>CURVE 13*</u>			
193.90	1.008	269.30	1.598	115.10	0.803				
232.40	1.109	288.30	1.665	130.40	0.858				
254.90	1.155	327.20	1.636	145.63	0.912				
276.50	1.218	337.60	1.741	159.83	0.962				
294.70	1.443	337.60	1.117	175.31	1.004	<u>CURVE 14*</u>			
297.10	1.272	347.00	1.695	202.39	1.088				
305.10	1.297	349.30	1.686	215.30	1.113				
316.00	1.297	360.40	1.770	235.06	1.138	<u>CURVE 15*</u>			
326.60	1.360	364.10	1.770	249.86	1.163				
337.50	1.406	371.10	1.753	<u>CURVE 6</u>		301.2	1.45		
352.20	1.331	373.60	1.807	293.20	1.255	<u>CURVE 16*</u>			
364.40	1.356	381.70	1.820	373.20	1.423				
380.50	1.406	390.10	1.715	<u>CURVE 7</u>		302.2	1.32		
381.10	1.389	408.90	1.807	293.20	1.632	<u>CURVE 17</u>			
381.50	1.548	427.10	1.820	373.20	1.715				
390.50	1.372	445.10	1.870	473.20	1.757				
397.40	1.423	489.50	1.778	573.20	1.757				
408.90	1.410	425.50	1.837	<u>CURVE 8*</u>		122	0.83		
412.80	1.469	538.10	1.745	307.2	1.32	141	0.84		
413.10	1.485	557.10	1.736	<u>CURVE 9*</u>		160	0.86		
433.90	1.544	561.40	1.795	301.2	1.33	177	1.06		
434.70	1.615	583.30	1.782	<u>CURVE 10</u>		192	1.08		
458.00	1.707	597.70	1.774	301.2	1.33	207	1.12		
466.60	1.661	628.60	1.736	<u>CURVE 4</u>		221	1.17		
480.20	1.770	652.40	1.715	301.00	1.452				
491.50	1.736	674.90	1.703	346.20	1.527				
500.70	1.778	701.80	1.665						
533.20	1.807								
533.50	1.787								
566.20	1.795								
571.00	1.770								
609.30	1.766								

\* Not shown on plot

SPECIFICATION TABLE NO. 240 THERMAL CONDUCTIVITY OF ALUMINUM +  $\Sigma X_1$  ALLOYS Al +  $\Sigma X_1$ 

Curve No.	Ref. Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) Al	Composition (continued), Specifications and Remarks
1	4	1911	373-873			95	Commercial purity.
2	6	L	350-827			99	Sp density (29°C) = 2.70.
3	88	L	113-291			99	Turned from a rod supplied by Johnson, Matthey and Co.; density 2.70 g cm <sup>-3</sup> at 20°C; electrical resistivity 2.72 ohm cm <sup>-2</sup> at 0°C.
4	405	L	399-623	±1.0		99.5	Cast at 700°C and molded at 200°C; drawn to 9.5 mm dia.

DATA TABLE NO. 249 THERMAL CONDUCTIVITY OF ALUMINUM +  $\Sigma X_1$  ALLOYS Al +  $\Sigma X_1$ (Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup> K<sup>-1</sup>)

T	k	T	k	T	k
<u>CURVE 1<sup>a</sup></u>					
373.20	2.050	595.20	2.210	198.20	2.983
473.20	2.343	638.20	2.180	222.20	2.075
573.20	2.720	665.20	2.167	248.20	2.088
673.20	3.130	676.20	2.170	273.20	2.109
773.20	3.682	720.20	2.120	291.20	2.110
873.20	4.225	771.20	2.095	<u>CURVE 4</u>	
<u>CURVE 2<sup>a</sup></u>					
356.20	2.205	771.20	2.063	399.2	2.057
373.20	2.125	792.20	2.075	410.2	2.068
393.20	2.137	827.20	2.105	428.2	2.105
452.20	2.218	<u>CURVE 3<sup>a</sup></u>		508.2	2.032
477.20	2.247	113.20	2.150	591.2	2.161
511.20	2.260	123.20	2.125	623.2	2.219
569.20	2.214	148.20	2.058		
		173.20	2.058		

No graphical presentation

# THERMAL CONDUCTIVITY OF ANTIMONY + BERYLLIUM + $\Sigma X_i$ ALLOYS

(Sb + Be < 99.50%, or at least one  $X_i$  > 0.20%)

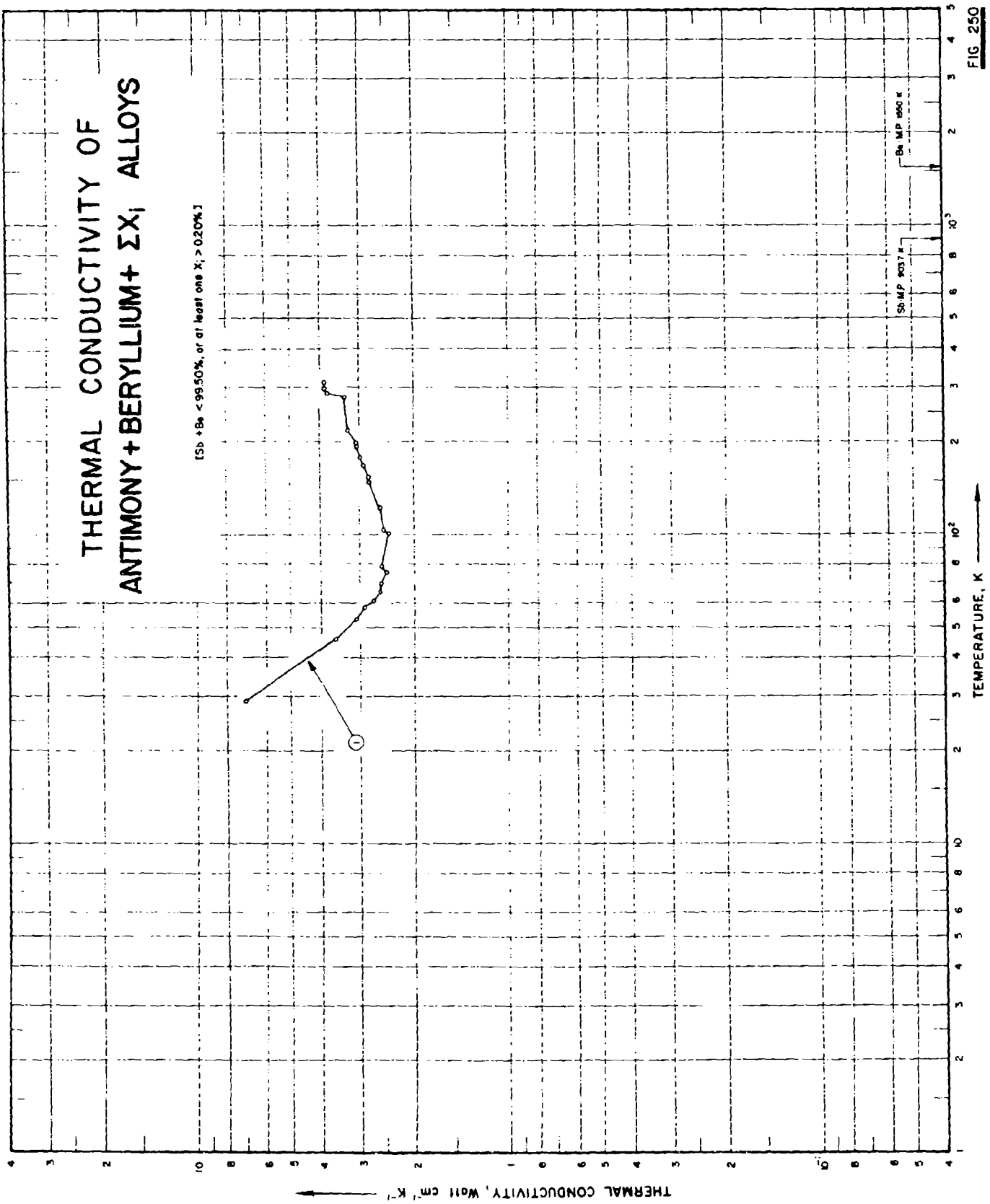


FIG. 250

SPECIFICATION TABLE NO. 250 THERMAL CONDUCTIVITY OF [ANTIMONY + BERYLLIUM +  $\Sigma X_i$ ] ALLOYS

(Sb + Be < 99.50% or at least one  $X_i > 0.20\%$ )

[ For Data Reported in Figure and Table No. 250 ]

Curve No.	Ref. Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)			Composition (continued), Specifications and Remarks
						Sb	Be	Pb	
1	1001	L	1964	29-313	64.5	35	0.5	Single crystal; specimen ~9 mm long with cross section ~0.25 cm <sup>2</sup> ; heat flow measured parallel to c-axis.	

DATA TABLE NO. 250 THERMAL CONDUCTIVITY OF [ANTIMONY + BERYLLIUM +  $\Sigma X_i$ ] Sb + Be +  $\Sigma X_i$   
 [Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup> K<sup>-1</sup>]

T	k
28.9	7.03
45.9	3.63
53.1	3.10
58.4	2.90
61.6	2.72
65.4	2.59
69.3	2.53
75.7	2.46
79.7	2.56
101.8	2.43
104.8	2.52
123.0	2.59
148.5	2.80
153.5	2.80
168.1	2.92
178.7	3.00
192.0	3.09
198.7	3.09
219.4	3.30
278.3	3.39
288.0	3.81
297.2	3.91
312.5	3.90

CURVE 1

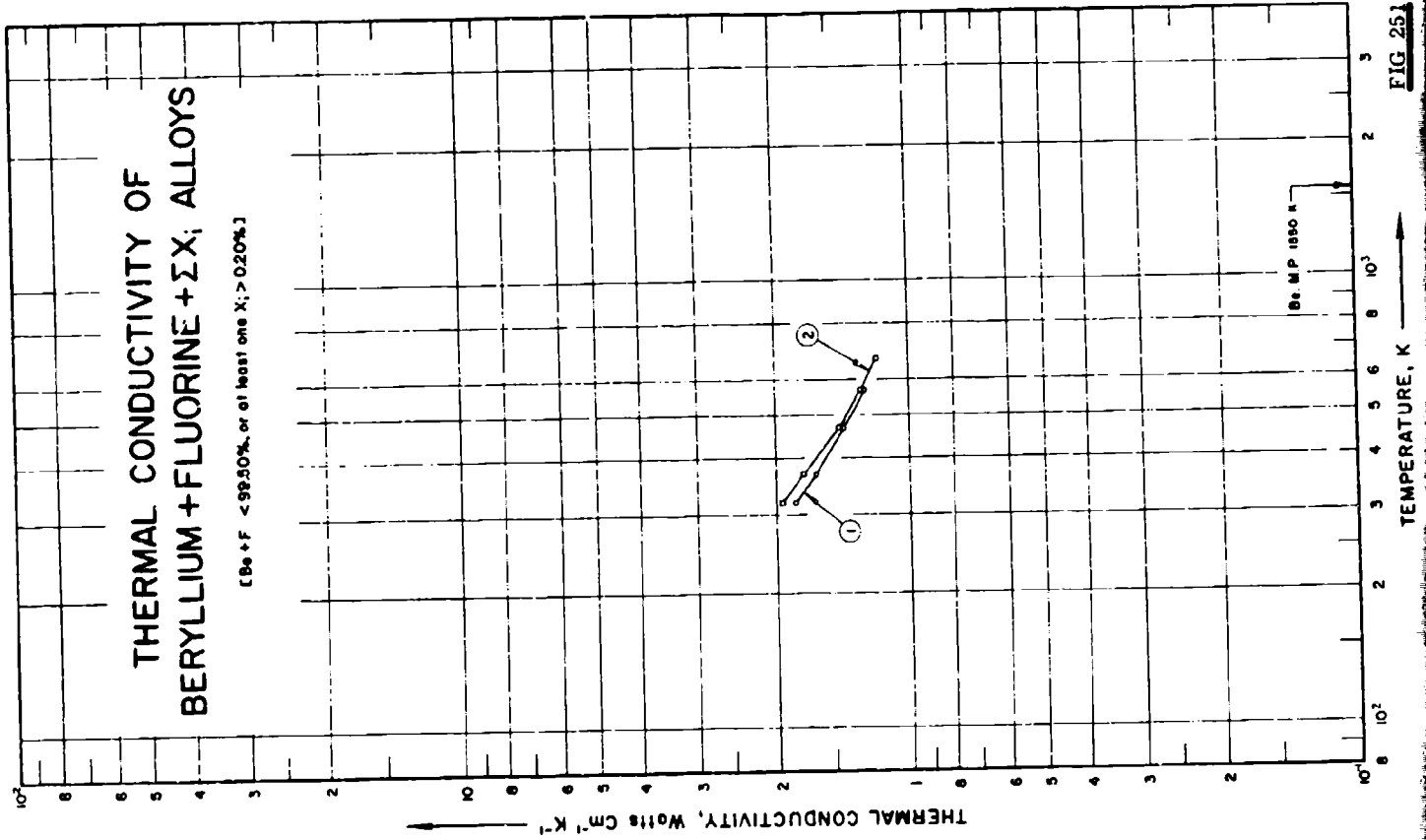


FIG 251



SPECIFICATION TABLE NO. 251 THERMAL CONDUCTIVITY OF [BERYLLIUM + FLUORINE + EX<sub>1</sub>] ALLOYS(Be + F < 99.50% or at least one X<sub>1</sub> > 0.20%)

[ For Data Reported in Figure and Table No. 251 ]

Curve No.	Ref. Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)				Composition (continued), Specifications and Remarks	
						Be	Fe	Al	Mg		Mn
1	111 C	1953	323-573		Xi A. R.	Bal	0.53		0.34		0.53 Fe and 0.34 Mg in the form of MgF <sub>2</sub> ; other impurities C, Ca, Al, and Mn.
2	111 C	1953	323-673		Xi H. T.	Bal	0.53		0.34		The above specimen after heat treatment at 700 C.

DATA TABLE NO. 251 THERMAL CONDUCTIVITY OF [BERYLLIUM + FLUORINE +  $\Sigma X_i$ ] ALLOYS

(Be + F < 99.50% or at least one  $X_i > 0.20\%$ )

[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup>K<sup>-1</sup>]

T	k
<u>CURVE 1</u>	
323.2	1.81
373.2	1.63
473.2	1.40
573.2	1.25
<u>CURVE 2</u>	
323.2	1.93
373.2	1.70
473.2	1.43
573.2	1.27
673.2	1.17

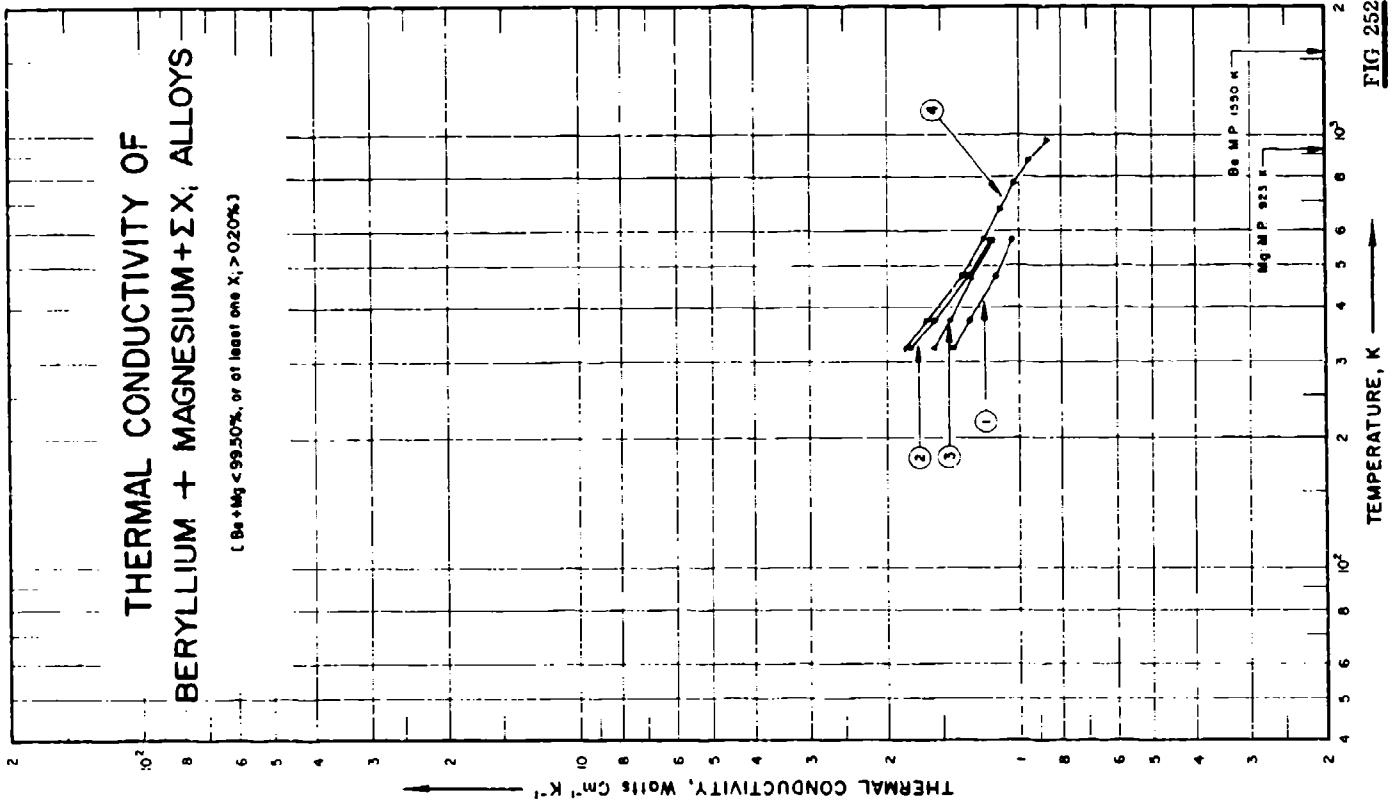


FIG. 252

SPECIFICATION TABLE NO. 252 THERMAL CONDUCTIVITY OF [BERYLLIUM + MAGNESIUM +  $\Sigma X_i$ ] ALLOYS(Be + Mg < 99.50% or at least one  $X_i > 0.20\%$ )

[ For Data Reported in Figure and Table No. 252 ]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)					Composition (continued), Specifications and Remarks		
							Be	Mg	Al	Ca	C		F	Fe
1	111	C	1953	323-573		II A. R.	96.5	1.81	0.06	0.035	0.032	1.52	0.55	Traces of Cu and Mn; chill-cast and machined.
2	111	C	1953	323-573		II H. T.	96.5	1.81	0.06	0.035	0.032	1.52	0.55	Traces of Cu and Mn; the above specimen after heat treatment at about 700 C.
3	111	C	1953	323-573		V A. R.								Approx. the same as the specimen II A. R.
4	111	C	1953	323-973		V H. T.								The above specimen after heat treatment at 700 C.

DATA TABLE NO. 252 THERMAL CONDUCTIVITY OF [BERYLLIUM + MAGNESIUM + EX<sub>1</sub>] ALLOYS(Be + Mg < 99.50% or at least one X<sub>1</sub> > 0.20%)[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup>K<sup>-1</sup>]

T	k
<u>CURVE 1</u>	
323.2	1.41
373.2	1.30
473.2	1.13
573.2	1.04
<u>CURVE 2</u>	
323.2	1.76
373.2	1.55
473.2	1.31
573.2	1.17
<u>CURVE 3</u>	
323.2	1.56
373.2	1.44
473.2	1.28
573.2	1.16
<u>CURVE 4</u>	
323.2	1.80
373.2	1.61
473.2	1.35
573.2	1.21
673.2	1.11
773.2	1.03
873.2	0.95
973.2	0.86

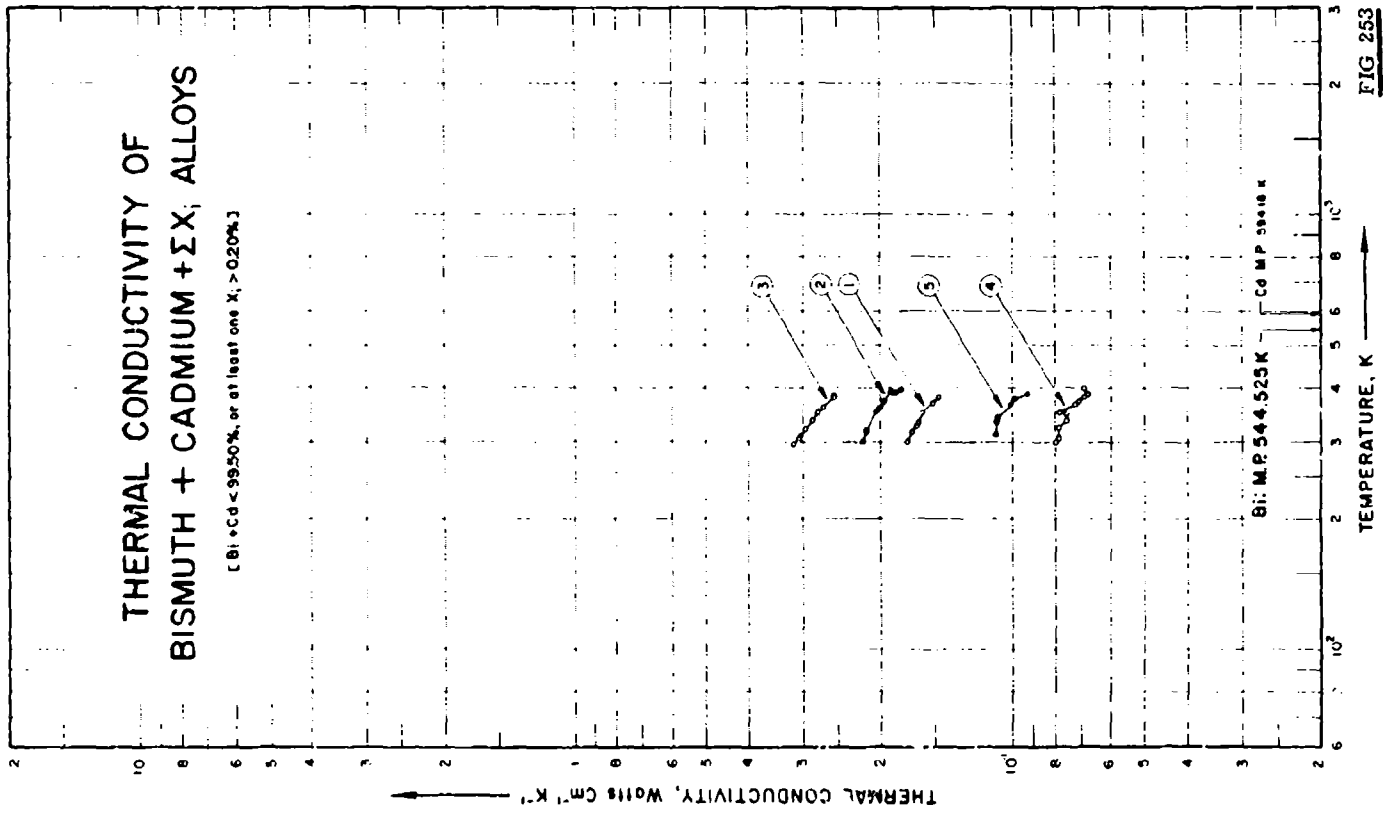


FIG 253

SPECIFICATION TABLE NO. 253 THERMAL CONDUCTIVITY OF [BISMUTH + CADMIUM +  $\Sigma X_i$ ] ALLOYS(Bi + Cd < 99.50% or at least one  $X_i$  > 0.20%)

[ For Data Reported in Figure and Table No. 253 ]

Curve No.	Ref. Method No. Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)			Composition (continued), Specifications and Remarks
						Bi	Cd	Pb	
1	383	1956	304-382			77.03	21.4	1.57	Calculated composition; electrical conductivity 1.449, 1.399, 1.376, 1.357, 1.325, 1.261 and $1.222 \times 10^4$ ohm <sup>-1</sup> cm <sup>-1</sup> at 30.5, 44.8, 55.0, 62.1, 77.9, 96.3 and 108.4 C, respectively.
2	383	1956	308-398			73.7	21.8	1.5	Calculated composition; electrical conductivity 1.507, 1.479, 1.468, 1.391, 1.361, 1.328, 1.298, 1.273 and $1.233 \times 10^4$ ohm <sup>-1</sup> cm <sup>-1</sup> at 34.5, 46.5, 48.0, 82.6, 89.6, 101.7, 119.0, 120, 125.2 C, respectively.
3	383	1956	298-383			59.49	39.3	1.21	Calculated composition; electrical conductivity 2.923, 2.835, 2.797, 2.722, 2.638, 2.601, 2.556, 2.478, 2.351 and $2.352 \times 10^4$ ohm <sup>-1</sup> cm <sup>-1</sup> at 24.7, 35.0, 39.9, 50.4, 64.9, 70.6, 80.0, 91.1, 108.9 and 109.8 C, respectively.
4	383	1956	300-399			94.86	3.2	1.94	Calculated composition; electrical conductivity 0.494, 0.486, 0.478, 0.467, 0.467, 0.464, 0.449, 0.441, 0.424, 0.420 and $0.423 \times 10^4$ ohm <sup>-1</sup> cm <sup>-1</sup> at 27.1, 34.2, 53.4, 63.3, 76.3, 80.1, 94.5, 99.6, 109.4, 113.8 and 125.8 C, respectively.
5	383	1956	314-388			88.59	9.6	1.81	Calculated composition; electrical conductivity 0.736, 0.724, 0.721, 0.682, 0.666 and $0.623 \times 10^4$ ohm <sup>-1</sup> cm <sup>-1</sup> at 40.3, 62.9, 70.5, 194.8, 103.6 and 14.6 C, respectively.

DATA TABLE NO. 253 THERMAL CONDUCTIVITY OF [BISMUTH + CADMIUM +  $\Sigma X_i$ ] ALLOYS

(Bi + Cd < 99.50% or at least one  $X_i > 0.20\%$ )

[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup> K<sup>-1</sup>]

T	k	T	k
<u>CURVE 1</u>			
303.7	0.174	<u>CURVE 4 (cont.)</u>	
318.0	0.169	382.6	0.0682
328.2	0.166	387.0	0.0674
335.3	0.164	399.0	0.0692
351.1	0.160	<u>CURVE 5</u>	
369.5	0.152	313.5	0.109
381.5	0.147	336.1	0.108
<u>CURVE 2</u>			
307.7	0.219	343.7	0.107
319.7	0.215	368.0	0.101
321.2	0.214	376.8	0.0987
355.8	0.203	387.8	0.0925
362.8	0.200	<u>CURVE 3</u>	
374.9	0.196	297.9	0.317
392.2	0.189	308.2	0.308
393.2	0.185	313.1	0.303
398.4	0.179	323.6	0.295
<u>CURVE 3</u>			
338.1	0.285	338.1	0.285
343.9	0.282	343.9	0.282
353.2	0.277	353.2	0.277
364.3	0.269	364.3	0.269
382.1	0.254	382.1	0.254
383.6	0.254	383.6	0.254
<u>CURVE 4</u>			
300.3	0.0795	300.3	0.0795
307.4	0.0782	307.4	0.0782
326.6	0.0782	326.6	0.0782
336.5	0.0749	336.5	0.0749
349.5	0.0749	349.5	0.0749
353.3	0.0778	353.3	0.0778
367.7	0.0729	367.7	0.0729
372.8	0.0707	372.8	0.0707



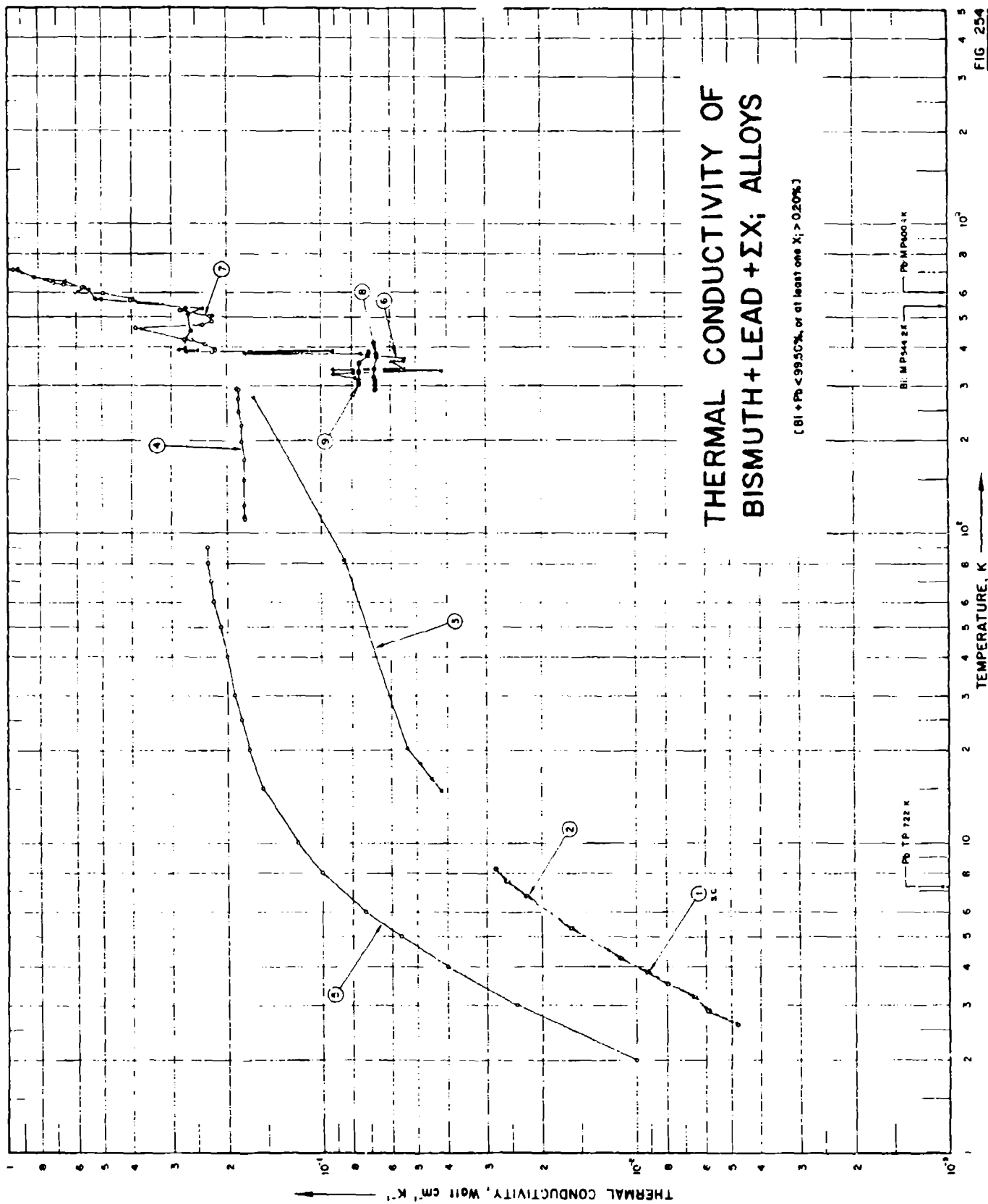


FIG 254

SPECIFICATION TABLE NO. 254 THERMAL CONDUCTIVITY OF [BISMUTH + LEAD +  $\Sigma X_i$ ] ALLOYS(Bi + Pb < 99.50% or at least one  $X_i > 0.20\%$ )

[For Data Reported in Figure and Table No. 254]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Bi	Composition (weight percent)				Composition (continued), Specifications and Remarks
								Pb	Cd	Sn		
1	228	L	1936	2.6-8.3		Rose Metal	50	25		25	Near eutectic; in superconducting state at zero gauss.	
2	228	L	1936	2.6-8.3		Rose Metal	50	25		25	Near eutectic; at 721 gauss (also superconducting).	
3	228	L	1936	15-276		Rose Metal	50	25		25	Near eutectic; in normal state at zero gauss.	
4	88	L	1908	111-295		Lipowitz alloy	50	25	11	14	Nominal composition; state unspecified.	
5	229	L	1955	2.0-90		Wood's Metal	48	26	13	13	Data cover both solid and liquid states.	
6	247	E	1950	319-714	±3.0		≈50.35	31.20	9.70	9.70	8.80	Data cover both solid and liquid states.
7	247	E	1950	367-822	±3.0		≈50.35	31.20	9.70	9.70	8.80	Data cover both solid and liquid states.
8	383	E	1956	292-381			97.62	1.98		1.00		
9	383	E	1956	305-396			96.53	1.97		1.50		

DATA TABLE NO. 254 THERMAL CONDUCTIVITY OF [BISMUTH + LEAD +  $\Sigma X_i$ ] ALLOYS(Bi + Pb < 99.50% or at least one  $X_i > 0.20\%$ )[Temperature, T, K; Thermal Conductivity, k, Watt  $\text{cm}^{-1}\text{K}^{-1}$ ]

T	k	T	k	T	k	T	k	T	k
<u>CURVE 1</u>		<u>CURVE 4 (cont.)</u>		<u>CURVE 6 (cont.)</u>		<u>CURVE 8</u>			
2.60	0.00476	273.20	0.184	426.40	0.268	292.3	0.0674		
2.88	0.00581	291.20	0.184	452.70	0.260	301.7	0.0674		
3.20	0.00658	295.20	0.186	513.90	0.264	311.7	0.0674		
3.53	0.00794	<u>CURVE 5</u>		535.40	0.238	321.9	0.0678		
3.85	0.00917	2.00	0.010	571.80	0.502	341.6	0.0678		
4.26	0.0113	3.00	0.024	572.50	0.523	372.0	0.0669		
5.31	0.0160	4.00	0.040	618.70	0.548	380.7	0.0669		
6.74	0.0225	5.00	0.056	651.50	0.653	<u>CURVE 9</u>			
7.52	0.0260	6.00	0.073	678.80	0.820	304.5	0.0766		
8.26	0.0280	8.00	0.100	714.40	0.929	312.3	0.0766		
<u>CURVE 2</u>		10.00	0.120	<u>CURVE 7</u>		331.1	0.0761		
2.60	0.00474	15.00	0.155	367.20	0.222	355.0	0.0757		
2.88	0.00588	20.00	0.170	390.50	0.218	374.5	0.0715		
3.20	0.00654	25.00	0.180	393.00	0.218	386.3	0.0707		
3.53	0.00794	30.00	0.190	410.60	0.234				
3.85	0.00926	40.00	0.200	421.50	0.259				
4.26	0.0112	50.00	0.210	425.30	0.272				
5.31	0.0161	60.00	0.220	461.80	0.389				
6.73	0.0223	70.00	0.225	475.50	0.238				
7.52	0.0259	80.00	0.230	485.30	0.222				
8.26	0.0280	90.00	0.230	507.20	0.226				
<u>CURVE 3</u>		<u>CURVE 6</u>		530.10	0.280				
14.70	0.0418	319.10	0.079	531.70	0.268				
16.10	0.0448	327.40	0.092	533.60	0.268				
17.90	0.0485	331.20	0.0795	568.40	0.506				
20.10	0.0532	338.50	0.0418	571.20	0.498				
82.00	0.0847	339.80	0.092	599.70	0.598				
276.00	0.164	342.40	0.0544	602.70	0.607				
<u>CURVE 4</u>		358.80	0.0588	626.40	0.565				
111.20	0.175	363.00	0.0544	629.60	0.573				
113.20	0.176	370.50	0.0544	639.40	0.661				
123.20	0.176	380.10	0.0753	649.40	0.720				
148.20	0.176	381.80	0.176	712.50	0.958*				
173.20	0.176	386.10	0.092	739.40	1.092*				
198.20	0.180	388.10	0.092	785.80	1.218*				
223.20	0.180	388.20	0.092	821.70	1.858*				
246.20	0.184	391.30	0.247						
		393.60	0.285						
		399.90	0.268						

\* Not shown on plot

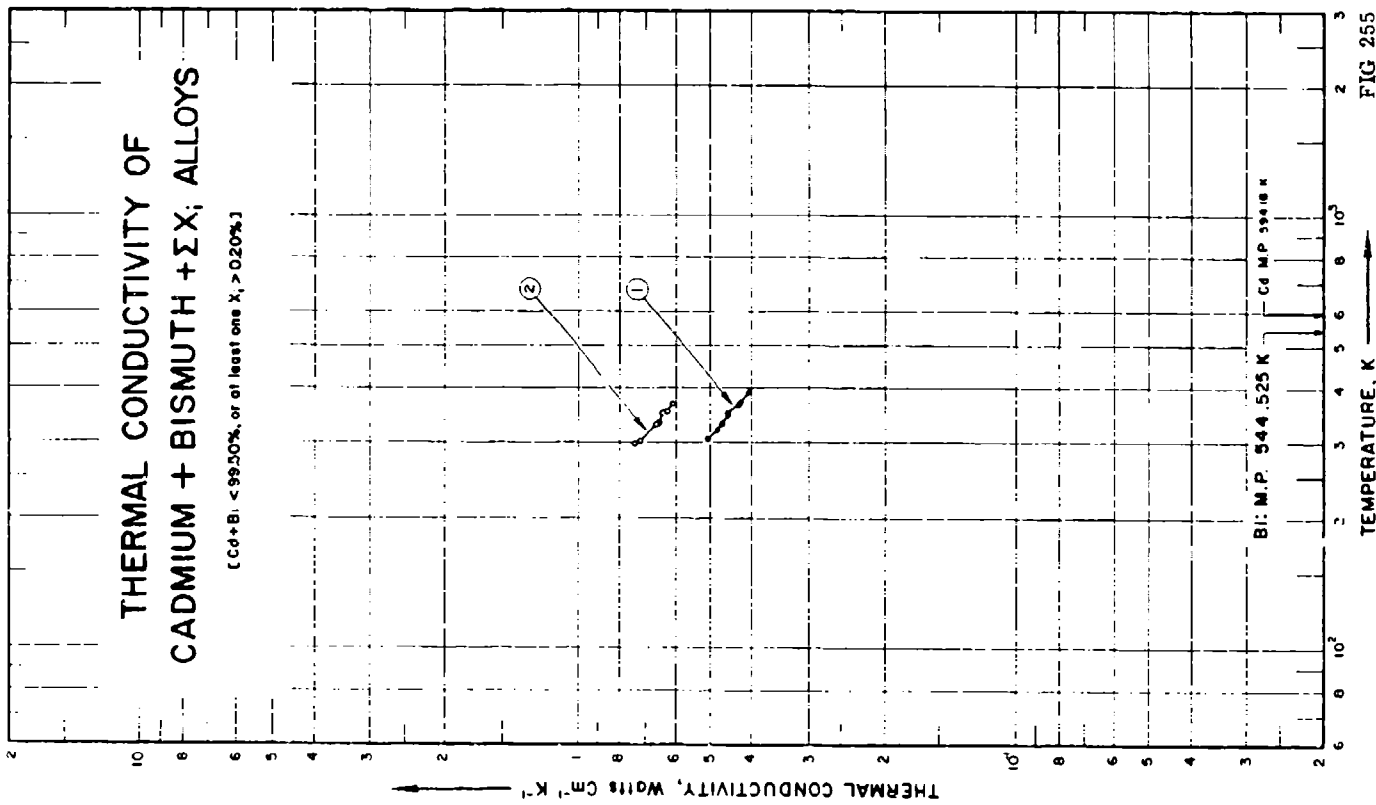


FIG. 255

SPECIFICATION TABLE NO. 255 THERMAL CONDUCTIVITY OF [CADMIUM + BISMUTH +  $\Sigma X_i$ ] ALLOYS  
(Cd + Bi < 99.50% or at least one  $X_i > 0.20\%$ )

[For Data Reported in Figure and Table No. 255]

Curve No.	Ref. Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) Cd Bi Pb	Composition (continued), Specifications and Remarks
1	383	1956	306-393			61.3 37.93 0.77	Calculated composition; electrical conductivity 4.98, 4.69, 4.59, 4.44, 4.42, 4.18, 4.14 and $3.97 \times 10^4$ ohm <sup>-1</sup> cm <sup>-1</sup> at 32.8, 46.8, 60.3, 73.8, 77.0, 95.0, 98.1 and 120.0 C, respectively.
2	383	1956	299-370			80.0 19.6 0.4	Calculated composition; electrical conductivity 7.38, 7.19, 6.62, 6.51, 6.42, 6.26 and $6.14 \times 10^4$ ohm <sup>-1</sup> cm <sup>-1</sup> at 25.6, 31.5, 58.9, 63.6, 77.5, 80.8 and 96.3 C, respectively.

DATA TABLE NO. 255 THERMAL CONDUCTIVITY OF [CADMIUM + BISMUTH +  $\Sigma X_i$ ] ALLOYS(Cd + Bi < 99.50% or at least one  $X_i > 0.20\%$ )[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup> K<sup>-1</sup>]

T k

CURVE 1

306.0	0.510
320.0	0.481
333.5	0.469
347.0	0.452
350.2	0.452
368.2	0.427
371.3	0.423
393.2	0.406

CURVE 2

298.8	0.741
308.7	0.720
332.1	0.661
336.8	0.653
350.7	0.644
354.0	0.628
369.5	0.615

FIGURE SHOWS ONLY 7 OF THE CURVES REPORTED IN TABLE

**THERMAL CONDUCTIVITY OF  
CHROMIUM + IRON +  $\Sigma X$ ; ALLOYS**

(Cr + Fe < 99.50%, or at least one X; > 0.20%)

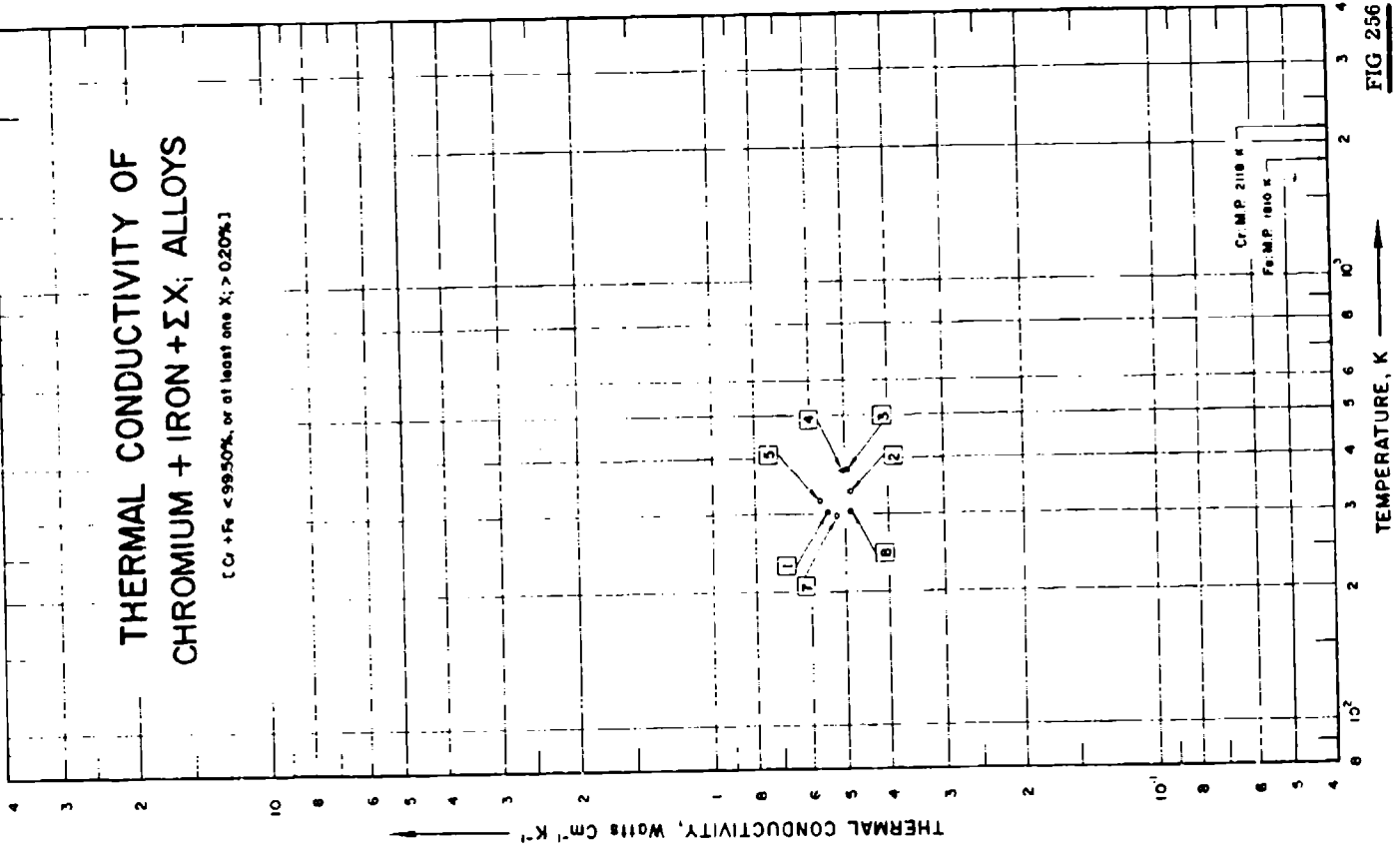


FIG 256

SPECIFICATION TABLE NO. 256 THERMAL CONDUCTIVITY OF [CHROMIUM + IRON + EX<sub>1</sub>] ALLOYS

(Cr + Fe < 99.50% or at least one X<sub>i</sub> > 0.20%)

[ For Data Reported in Figure and Table No. 256 ]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)				Composition (continued), Specifications and Remarks
							Cr	Fe	C	Si	
1	204	L	1937	302.9		Ferrochromium, 24	53.37	45.37	0.18	1.08	
2	204	L	1937	338.0		Ferrochromium, 25	62.94	34.39	0.11	2.56	
3	204	L	1937	389.5		Ferrochromium, 26	50.4	49.0	0.20	0.40	
4	204	L	1937	378.8		Ferrochromium, 27	51.4	45.65	0.45	2.30	
5	204	L	1937	323.2		Ferrochromium, 28	53.73	44.91	0.24	1.12	Obtained from fusion of iron with ferrochromium.
6	204	L	1937	379.4		Ferrochromium, 29	56.12	41.31	1.64	0.93	
7	204	L	1937	299.6		Ferrochromium, 30	53.8	41.3	4.45	0.45	
8	204	L	1937	305.7		Ferrochromium, 31	62.7	30.6	6.25	0.45	



DATA TABLE NO. 256 THERMAL CONDUCTIVITY OF [CHROMIUM + IRON +  $\Sigma X_i$ ] ALLOYS(Cr + Fe < 99.50% or at least one  $X_i > 0.20\%$ )[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup>K<sup>-1</sup>]

T	k
<u>CURVE 1</u>	
302.9	0.545
<u>CURVE 2</u>	
338.0	0.490
<u>CURVE 3</u>	
390.5	0.499
<u>CURVE 4</u>	
378.8	0.508
<u>CURVE 5</u>	
323.2	0.573
<u>CURVE 6*</u>	
379.4	0.501
<u>CURVE 7</u>	
299.6	0.521
<u>CURVE 8</u>	
305.7	0.489

\* Not shown on plot

FIGURE SHOWS ONLY 9 OF THE CURVES REPORTED IN TABLE

THERMAL CONDUCTIVITY OF  
COBALT + CHROMIUM + ΣX; ALLOYS

[Co + Cr < 99.50%, or at least one X; > 0.20%]

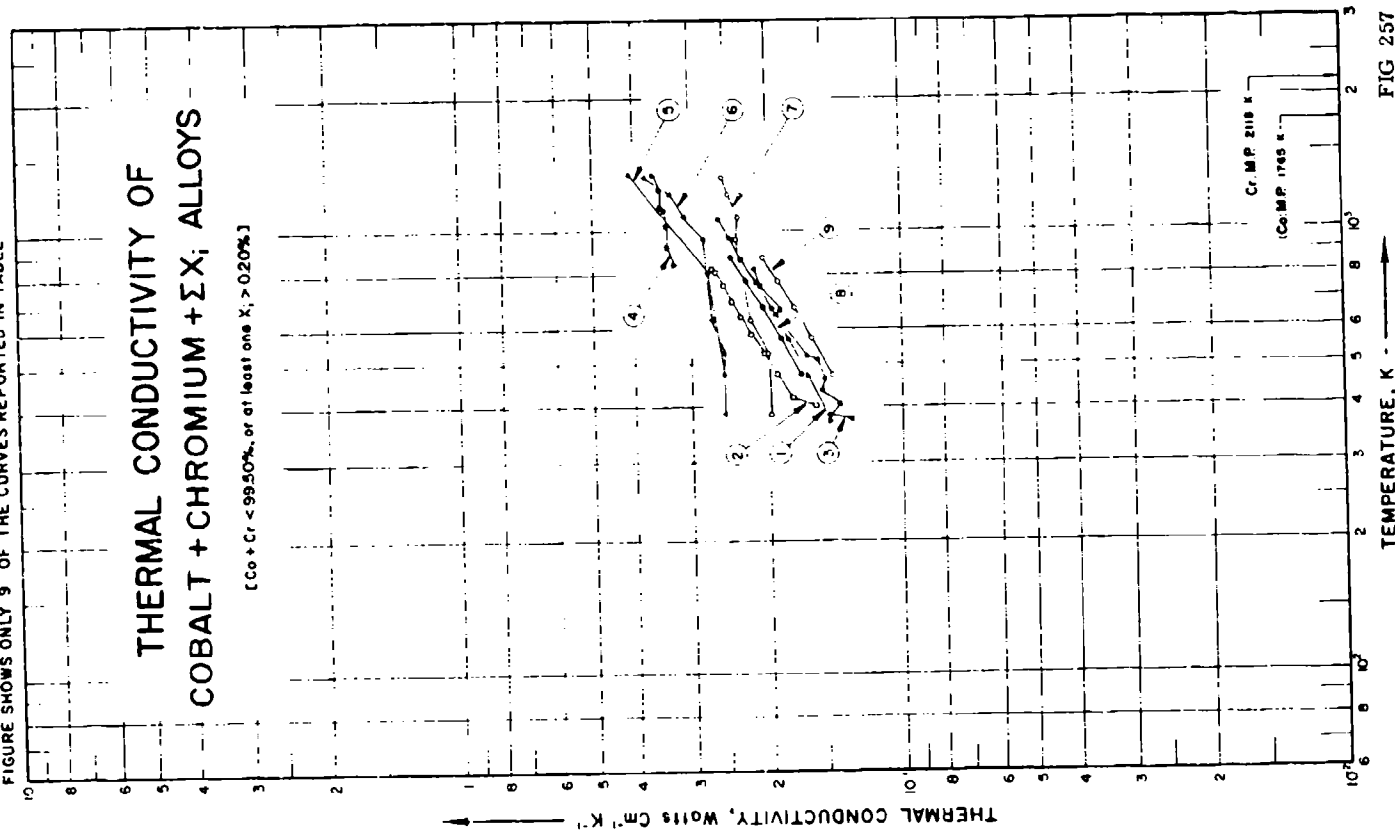


FIG 257

SPECIFICATION TABLE NO. 257 THERMAL CONDUCTIVITY OF [COBALT + CHROMIUM +  $\Sigma X_i$ ] ALLOYS(Co + Cr < 99.50% or at least one  $X_i > 0.20\%$ )

[For Data Reported in Figure and Table No. 257]

Curve No.	ReL No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Co	Cr	Fe	Ni	W	C	Mo	Mn	Composition (continued), Specifications and Remarks
1	181	$\pm$ C	1953	373-1073		British G-32	45.6	19.1	15.64	10.5	0.27	2.2	2.2	0.77	0.52 Si, 1.4 Nb, 3.0 V.
2	37	C	1951	404-628	4.0	S 816	45.0	20.0	15.0	20.0					Lead used as comparative material.
3	37	C	1951	377-826	4.0	X-40	45.0	25.5	2.0	10.5	7.5	0.53			Lead used as comparative material.
4	43	L	1958	849-1356	5.0	Haynes stellite alloy	60.49	26.69	1.54	2.39	0.253	5.42			Specimen 6.75 in. in dia and 1.50 in. thick; as received.
5	376		1960	478-1366		WI-52	Bal.	20/22	1.5/2.5	1.0/Max	10/12	0.4/0.5		0.5	1.5-2.0 Nb + Ta, 0.04 P, 0.04 S, 0.5 Si.
6	376	C	1960	388-1341		WI-52	Bal.	20/22	1.5/2.5	1.0/Max	10/12	0.4/0.5		0.5	1.5-2.0 Nb + Ta, 0.04 P, 0.04 S, 0.5 Si; Armco iron used as comparative material.
7	376	C	1960	388-1341		WI-52	Bal.	20/22	1.5/2.5	1.0/Max	10/12	0.4/0.5		0.5	1.5-2.0 Nb + Ta, 0.04 P, 0.04 S, 0.5 Si, specimen coated with chromium; Armco iron used as comparative material.
8	616		1947	473-873		Haynes stellite 21	Bal.	25/30	2.0/Max	1.5/3.5	0.2/0.35	4.5/6.5			Wrought; density 8.3 g cm <sup>-3</sup> .
9	616		1947	473-873		H. S. No. 21	Bal.	25/30	2.0/Max	1.5/3.5	0.2/0.35	4.5/6.5			Density 8.3 g cm <sup>-3</sup> .
10	616		1947	473-873		H. S. No. 23	Bal.	23/29	2.0/Max	1.5/Max	4/7	0.35/0.5			Density 8.54 g cm <sup>-3</sup> .
11	616		1947	473-873		H. S. No. 31	Bal.	23/28	1.5/Max	9/12	6/9	0.45/0.6			Density 8.61 g cm <sup>-3</sup> .

DATA TABLE NO. 257 THERMAL CONDUCTIVITY OF [COBALT + CHROMIUM + ΣX<sub>i</sub>] ALLOYS

(Co + Cr < 99.50% or at least one X<sub>i</sub> > 0.20%)

[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup>K<sup>-1</sup>]

T	k	T	k	T	k
<u>CURVE 1</u>					
373.2	0.146	843.8	0.325	473.2	0.145
473.2	0.163	930.2	0.336	573.2	0.160
573.2	0.130	1007.7	0.336	673.2	0.175
673.2	0.197	1127.7	0.341	773.2	0.190
773.2	0.213	1149.8	0.347	873.2	0.205
873.2	0.230	1284.1	0.348	<u>CURVE 10*</u>	
973.2	0.243	1355.6	0.358	473.2	0.154
1073.2	0.259			573.2	0.176
<u>CURVE 2</u>					
404.0	0.157	477.6	0.252	673.2	0.180
422.0	0.177	310.5	0.274	773.2	0.189
477.6	0.192	1366.5	0.404	873.2	0.212
533.2	0.205	<u>CURVE 6</u>		<u>CURVE 11*</u>	
588.7	0.218	388.2	0.251	473.2	0.148
644.3	0.230	533.7	0.252	573.2	0.175
699.8	0.242	632.1	0.268	673.2	0.183
755.4	0.253	964.8	0.280	773.2	0.194
810.9	0.263	1088.2	0.309	873.2	0.219
827.6	0.267	1225.4	0.330	<u>CURVE 3</u>	
<u>CURVE 4</u>					
377.2	0.130	1341.4	0.378	<u>CURVE 7</u>	
386.3	0.146	<u>CURVE 5</u>		388.2	0.198
408.4	0.138	477.6	0.252	533.7	0.200
433.1	0.152	310.5	0.274	632.1	0.218
464.8	0.149	1366.5	0.404	964.8	0.235
514.2	0.156	<u>CURVE 6</u>		1088.2	0.232
524.3	0.164	388.2	0.251	1225.4	0.245
651.2	0.192	533.7	0.252	1341.5	0.252
670.1	0.188	632.1	0.268	<u>CURVE 8</u>	
732.2	0.208	964.8	0.280	473.2	0.170
826.2	0.215	1088.2	0.309	573.2	0.188
<u>CURVE 5</u>					
473.2	0.154	477.6	0.252	673.2	0.206
573.2	0.176	310.5	0.274	773.2	0.224
673.2	0.180	1366.5	0.404	873.2	0.242
773.2	0.189	<u>CURVE 6</u>		<u>CURVE 9</u>	
873.2	0.212	388.2	0.251	473.2	0.145
<u>CURVE 6</u>					
388.2	0.251	533.7	0.252	573.2	0.160
533.7	0.252	632.1	0.268	673.2	0.175
632.1	0.268	964.8	0.280	773.2	0.190
964.8	0.280	1088.2	0.309	873.2	0.205
1088.2	0.309	1225.4	0.330	<u>CURVE 10*</u>	
1225.4	0.330	1341.4	0.378	473.2	0.154
1341.4	0.378	<u>CURVE 7</u>		573.2	0.176
<u>CURVE 7</u>					
377.2	0.130	388.2	0.198	673.2	0.180
386.3	0.146	533.7	0.200	773.2	0.189
408.4	0.138	632.1	0.218	873.2	0.212
433.1	0.152	964.8	0.235	<u>CURVE 11*</u>	
464.8	0.149	1088.2	0.232	473.2	0.148
514.2	0.156	1225.4	0.245	573.2	0.175
524.3	0.164	1341.5	0.252	673.2	0.183
651.2	0.192	<u>CURVE 8</u>		773.2	0.194
670.1	0.188	473.2	0.170	873.2	0.219
732.2	0.208	573.2	0.188	<u>CURVE 12*</u>	
826.2	0.215	673.2	0.206	<u>CURVE 13*</u>	
<u>CURVE 8</u>					
473.2	0.170	773.2	0.224	<u>CURVE 14*</u>	
573.2	0.188	873.2	0.242	<u>CURVE 15*</u>	
673.2	0.206	<u>CURVE 9</u>		<u>CURVE 16*</u>	
773.2	0.224	<u>CURVE 10*</u>		<u>CURVE 17*</u>	
873.2	0.242	<u>CURVE 11*</u>		<u>CURVE 18*</u>	

\* Not shown on plot

SPECIFICATION TABLE NO. 258 THERMAL CONDUCTIVITY OF [COBALT + IRON +  $\Sigma X_i$ ] ALLOYS Co + Fe +  $\Sigma X_i$   
 (Co + Fe < 99.50% or at least one  $X_i > 0.20\%$ )

Curve No.	Rel. Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)			Composition (continued), Specifications and Remarks	
						Co	Fe	Cu		
1	208	E	1919	303.2	Co-Fe 9	67.988	30.72	0.195	0.086	0.093 Mn, 0.77 Ni, 0.009 P, 0.008 S, 0.131 Si.
2	208	E	1919	303.2	Co-Fe 10	77.695	20.95	0.21	0.054	0.062 Mn, 0.88 Ni, 0.006 P, 0.005 S, 0.134 Si.
3	208	E	1919	303.2	Co-Fe 11	78.402	11.18	0.225	0.029	0.031 Mn, 0.99 Ni, 0.003 P, 0.003 S, 0.137 Si.
4	208	E	1919	303.2	Co-Fe-Ni 12	97.12	1.4	0.24		1.1 Ni, 0.14 Si.

DATA TABLE NO. 258 THERMAL CONDUCTIVITY OF [COBALT + IRON +  $\Sigma X_i$ ] ALLOYS Co + Fe +  $\Sigma X_i$

(Co + Fe < 99.50% or at least one  $X_i > 0.20\%$ )

[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup> K<sup>-1</sup>]

T k

CURVE 1\*

303.2 0.720

CURVE 2\*

303.2 0.712

CURVE 3\*

303.2 0.402

CURVE 4\*

303.2 0.692

\* No graphical presentation

SPECIFICATION TABLE NO. 259 THERMAL CONDUCTIVITY OF [COBALT + NICKEL +  $\Sigma X_i$ ] ALLOYS Co + Ni +  $\Sigma X_i$

(Co + Ni > 99.50% or at least one  $X_i$  > 0.20%)

Curve No.	Ref. Method No. Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) Co Ni	Composition (continued), Specifications and Remarks
1	238	E	1927	303.2		95	Nickel obtained from Mond and Co. containing 0.1 Fe, 0.037 C, 0.019 S, 0.013 Cu, 0.006 Si, trace Cr, P, Al, and Mn impurities; cobalt obtained from Sigsbyasi and Co. containing 0.20 Fe, 0.220 C, 0.05 Al, 0.034 S, 0.032 Si, 0.003 P, trace Ni, and Mn impurities; cast and machined; heated at 800 C for 46 min and slowly cooled.

DATA TABLE NO. 259 THERMAL CONDUCTIVITY OF [COBALT + NICKEL +  $\Sigma X_i$ ] ALLOYS Co + Ni +  $\Sigma X_i$

(Co + Ni > 99.50% or at least one  $X_i$  > 0.20%)

[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup> K<sup>-1</sup>]

T k

CURVE 1\*

303.2 0.523

\* No graphical presentation

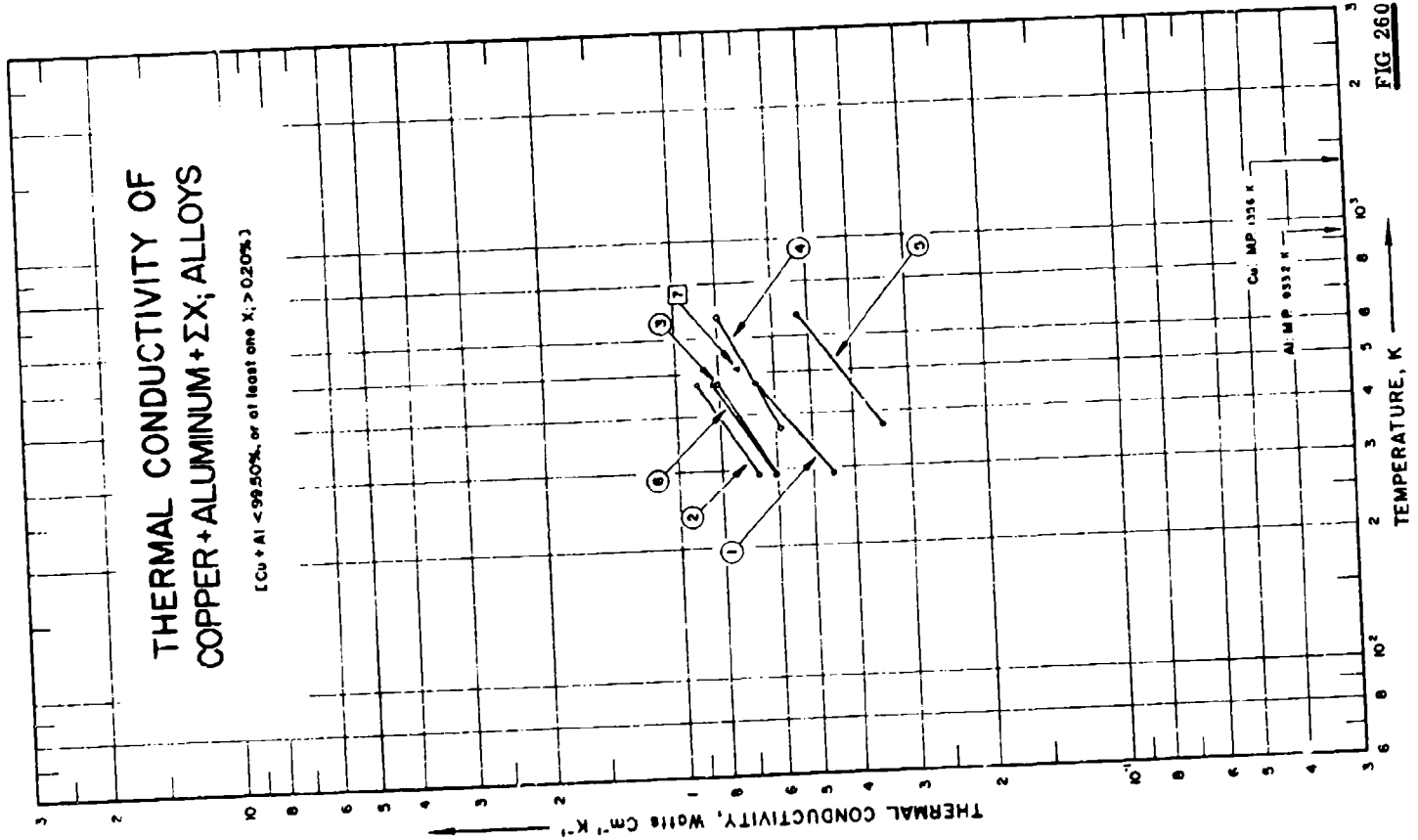


FIG 260

SPECIFICATION TABLE NO. 260 THERMAL CONDUCTIVITY OF [COPPER + ALUMINUM +  $\Sigma X_i$ ] ALLOYS(Cu + Al < 99.50% or at least one  $X_i > 0.20\%$ )

[ For Data Reported in Figure and Table No. 260 ]

Curve No.	ReL No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)							Composition (continued), Specifications and Remarks
							Cu	Al	Fe	Mn	Ni	Si	Sn	
1	135	L	1935	293, 473		Bar 67	89.08	5.11	0.08	4.98	0.74			Quenched from 850 C.
2	135	L	1935	293, 473		Bar 67 B	89.08	5.11	0.08	4.98	0.74			Annealed at 850 C.
3	135	L	1935	293, 473		Bar 49	89.38	9.41	0.52	0.31		0.38		Annealed at 750 C.
4	215	L	1939	373, 673	2.0	Aluminum bronze; 1	87.16	9.97	Trace	2.77				Cast.
5	215	L	1939	373, 673	2.0	Aluminum bronze; 2	81.69	9.77	2.96	1.95		3.71		Cast.
6	215	L	1939	373, 673	2.0	Aluminum bronze	89.56	8.66	Trace	1.75				Rolled.
7	224	L	1933	513		Aluminum bronze	89.87	9.09				0.47		



DATA TABLE NO. 260 THERMAL CONDUCTIVITY OF [COPPER + ALUMINUM +  $\Sigma X_i$ ] ALLOYS(Cu + Al < 99.50% or at least one  $X_i > 0.20\%$ )[Temperature, T, K; Thermal Conductivity, k, Watt  $\text{cm}^{-1}\text{K}^{-1}$ ]

T	k
<u>CURVE 1</u>	
293.00	0.448
473.00	0.661
<u>CURVE 2</u>	
293.00	0.661
473.00	0.895
<u>CURVE 3</u>	
293.00	0.603
473.00	0.803
<u>CURVE 4</u>	
373.00	0.586
673.00	0.799
<u>CURVE 5</u>	
373.00	0.343
763.00	0.527
<u>CURVE 6</u>	
373.00	0.603
673.00	0.824
<u>CURVE 7</u>	
513.00	0.728

SPECIFICATION TABLE NO. 261 THERMAL CONDUCTIVITY OF [COPPER + BERYLLIUM +  $\Sigma X_i$ ] ALLOYS Cu + Be +  $\Sigma X_i$   
 (Cu + Be < 99.50% or at least one  $X_i > 0.20\%$ )

Curve No.	Ref. Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)				Composition (continued), Specifications and Remarks
						Cu	Be	Fe	Ni	
1	L	1935	293, 473		Bar 140	97.49	2.24	0.06	0.27	Quenched from 815 C.
2	L	1935	293, 473		Bar 141	97.49	2.24	0.06	0.27	Quenched from 815 C and followed by reheating to 300 C.

DATA TABLE NO. 261 THERMAL CONDUCTIVITY OF [COPPER + BERYLLIUM +  $\Sigma X_i$ ] ALLOYS Cu + Be +  $\Sigma X_i$   
 (Cu + Be < 99.50% or at least one  $X_i > 0.20\%$ )

[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup> K<sup>-1</sup>]

T	k
<u>CURVE 1*</u>	
293.0	0.858
473.0	1.04
<u>CURVE 2*</u>	
293	1.03
473	1.17

\* No graphic presentation

SPECIFICATION TABLE NO. 262 THERMAL CONDUCTIVITY OF [COPPER + CADMIUM +  $\Sigma X_i$ ] ALLOYS Cu + Cd +  $\Sigma X_i$   
 (Cu + Cd < 99.50% or at least one  $X_i > 0.20\%$ )

Curve No.	Ref. Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)		Composition (continued), Specifications and Remarks	
						Cu	Cd		
1	L	1935	293, 473		Bar 70	98.41	1.07	0.02	0.02 Sn, 0.59 Sb; approx composition; annealed at 750 C for 1.5 hrs; electrical conductivity 32.74 and 23.30 x 10 <sup>6</sup> ohm <sup>-1</sup> cm <sup>-1</sup> at 20 and 200 C respectively.

DATA TABLE NO. 262 THERMAL CONDUCTIVITY OF [COPPER + CADMIUM +  $\Sigma X_i$ ] ALLOYS Cu + Cd +  $\Sigma X_i$   
 (Cu + Cd < 99.50% or at least one  $X_i > 0.20\%$ )

[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup> K<sup>-1</sup>]

T	k
CURVE 1*	
293.2	2.33
473.2	2.690

\* No graphical presentation

# THERMAL CONDUCTIVITY OF COPPER+COBALT + $\Sigma X_i$ ALLOYS

(Cu+Co  $\geq$  99.50%; impurity  $\leq$  0.20% each)

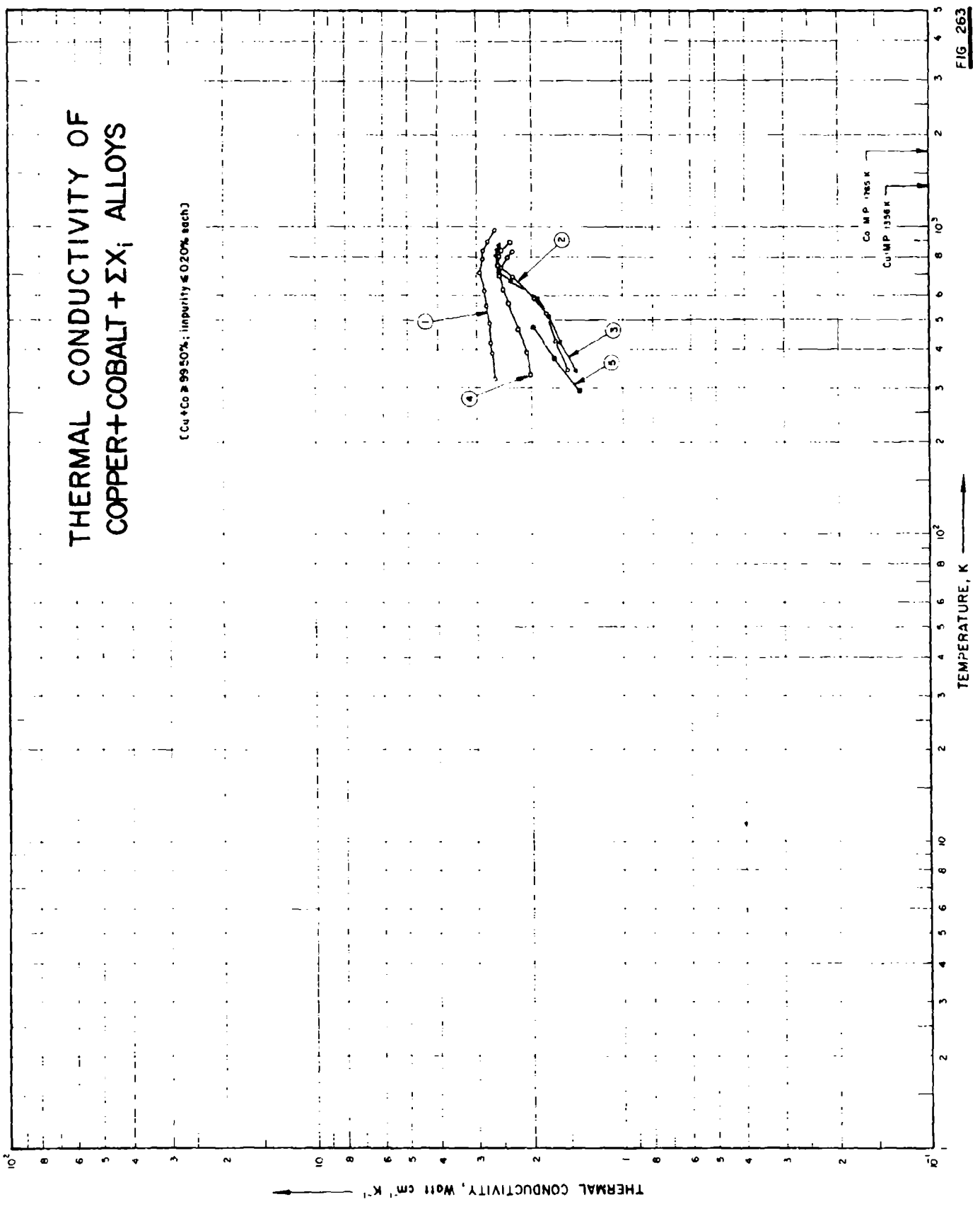


FIG 263

SPECIFICATION TABLE NO. 263 THERMAL CONDUCTIVITY OF [COPPER + COBALT +  $\Sigma X_i$ ] ALLOYS  
(Cu + Co = 99.50% of at least one;  $X_i > 0.20\%$ )

[ For Data Reported in Figure and Table No. 263 ]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)				Composition (continued), Specifications and Remarks
							Cu	Co	Be	Ag	
1	377		1957	321-975			99.4	0.3		0.3	Electrical conductivity 33.55, 28.12, 26.35, 23.18, 20.95, 18.9, 17.1, 15.33, 14.1, 12.68 and $10.95 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 47, 25, 115.9, 145.8, 214.8, 281.8, 281.8, 346.4, 436.6, 508.6, 567, 618.3 and 701.6 C, respectively.
2	377		1957	342-836			97.15	1.7	0.15	1.0	Electrical conductivity 17.48, 15.67, 13.96, 13.51, 13.75, 13.85, 12.25 and $11.30 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 68.6, 150.3, 247.5, 314, 416.8, 467, 523 and 562.5 C, respectively.
3	377		1957	341-373			99.29	0.3		0.27	0.14 Mo; electrical conductivity 17.15, 15.41, 14.0, 13.26, 14.5, 14.53, 13.12, 12.42 and $11.5 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 67.8, 146.7, 235.1, 313.6, 390, 438.3, 533.8, 570.6 and 600 C, respectively.
4	377		1957	330-692			97.2	2.2	<del>0.5</del> 0.5		Electrical conductivity 25.0, 21.8, 19.5, 17.12, 16.13, 14.9, 14.04, 13, 12.23 and $10.86 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 56.3, 117, 190, 291, 350, 421, 681.3, 530, 566.5 and 619 C, respectively.
5	588		1960	293-473			97.3	2.2	0.5		Normalized at 1000 C for 30 min.

DATA TABLE NO. 263 THERMAL CONDUCTIVITY OF [COPPER + COBALT +  $\Sigma X_i$ ] ALLOYS(Cu + Co < 99.50% or at least one  $X_i > 0.20\%$ )[Temperature, T, K; Thermal Conductivity, k, Watt  $\text{cm}^{-1} \text{K}^{-1}$ ]

T	k	T	k
<u>CURVE 1</u>			
320.5	2.62	694.2	2.54
389.1	2.68	754.5	2.57
419.0	2.71	803.2	2.56
488.0	2.74	839.7	2.51
555.0	2.82	892.2	2.36
619.6	2.85	<u>CURVE 5</u>	
709.8	2.95	293.2	1.40
781.8	2.90	373.2	1.70
840.2	2.89	473.2	1.91
891.5	2.79		
974.8	2.62		
<u>CURVE 2</u>			
341.8	1.53		
423.5	1.67		
520.7	1.79		
587.2	1.97		
690.0	2.32		
740.2	2.52		
795.2	2.40		
835.7	2.31		
<u>CURVE 3</u>			
341.0	1.44		
419.9	1.61		
508.3	1.76		
586.8	1.92		
663.2	2.35		
711.5	2.55		
807.0	2.61		
843.8	2.59		
873.2	2.56		
<u>CURVE 4</u>			
329.5	2.03		
390.2	2.08		
463.2	2.23		
564.2	2.39		
623.2	2.48		

SPECIFICATION TABLE NO. 264 THERMAL CONDUCTIVITY OF [COPPER + IRON +  $\Sigma X_i$ ] ALLOYS Cu + Fe +  $\Sigma X_i$   
 (Cu + Fe > 99.50% or at least one  $X_i > 0.20\%$ )

Curve No.	Ref. Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) Cu Fe Mn	Composition (continued), Specifications and Remarks
1	L	1935	293, 473		Bar 137	50.75 48.6 0.47	0.047 Si, 0.023 C; annealed at 800 C.

DATA TABLE NO. 264 THERMAL CONDUCTIVITY OF [COPPER + IRON +  $\Sigma X_i$ ] ALLOYS Cu + Fe +  $\Sigma X_i$   
 (Cu + Fe < 99.50% or at least one  $X_i > 0.20\%$ )

[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup> K<sup>-1</sup>]

T	k
<u>CURVE 1*</u>	
293.0	0.992
473.0	1.134

\* No graphical presentation

# THERMAL CONDUCTIVITY OF COPPER + LEAD+ΣX; ALLOYS

[Cu + Pb < 99.50%, or at least one X, > 0.20%]

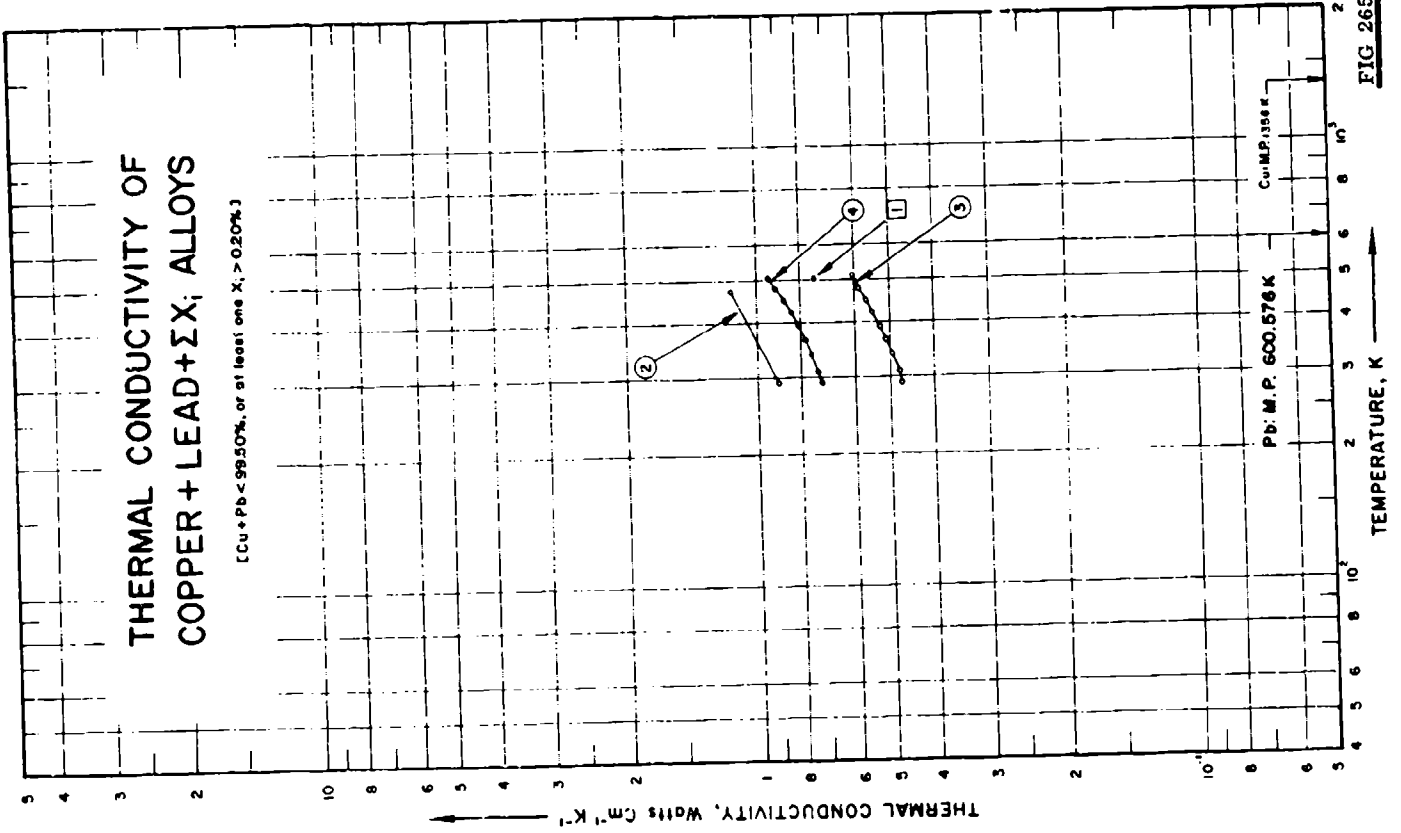


FIG 265



SPECIFICATION TABLE NO. 265 THERMAL CONDUCTIVITY OF [COPPER + LEAD +  $\Sigma X_i$ ] ALLOYS(Cu + Pb < 99.50% or at least one  $X_i > 0.20\%$ )

[ For Data Reported in Figure and Table No. 265 ]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Cu	Pb	Sn	Zn	Si	Fe	Sb	Composition (cont. ruled), Specifications and Remarks
1	224	L	1923	503.2		SAE Bearing alloy - No. 66	85.29	8.26	5.56	0.89				
2	135	L	1935	293, 473		Bar 99	88.07	3.83	3.77	3.7	0.6	0.03		Annealed, at 700 C for 2 hrs; electrical conductivity 11.53 and $9.80 \times 10^6 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 20 and 200 C, respectively.
3	529	C	1958	292-504	$\pm 5$	80-10-10	79.0	10.2	9.3	0.27	0.43	0.01	0.26	0.01 P; average composition: density 8.93 $\text{g cm}^{-3}$ , M. P. 929 C; electrical resistivity reported 17.01, 17.63, 18.00, 18.38, 18.76, 19.15, 19.53 and 19.92 $\mu\text{ohm cm}$ at 20, 66, 93, 121, 149, 177, 204, and 232 C, respectively.
4	529	C	1958	292-504	$\pm 5$	85-5-5-5	85.0	5.1	4.4	4.6	0.7	0.11	0.17	0.01 P; average composition; density 8.92 $\text{g cm}^{-3}$ , M. P. 1009 C; electrical resistivity reported 11.43, 11.87, 12.13, 12.42, 12.70, 12.99, 13.28 and 13.59 $\mu\text{ohm cm}$ at 20, 66, 93, 121, 149, 177, 204 and 232 C, respectively.

DATA TABLE NO. 265 THERMAL CONDUCTIVITY OF [COPPER + LEAD +  $\Sigma X_i$ ] ALLOYS(Cu + Pb  $\leq$  99.50% or at least one  $X_i \leq 0.20\%$ )[Temperature, T, K, Thermal Conductivity, k, Watt  $\text{cm}^{-1}\text{K}^{-1}$ ]

T	k
<u>CURVE 1</u>	
503.2	0.741
<u>CURVE 2</u>	
293.2	0.900
473.2	1.153
<u>CURVE 3</u>	
292.2	0.476
309.9	0.479
337.7	0.498
365.5	0.514
393.3	0.530
421.1	0.549
448.9	0.568
476.6	0.588
504.4	0.613
<u>CURVE 4</u>	
292.2	0.750
309.9	0.734
337.7	0.738
365.5	0.742
393.3	0.810
421.1	0.839
448.9	0.871
476.6	0.909
504.4	0.945

# THERMAL CONDUCTIVITY OF COPPER + MANGANESE + ΣX; ALLOYS

(Cu + Mn < 99.50%, or at least one X; > 0.20%)

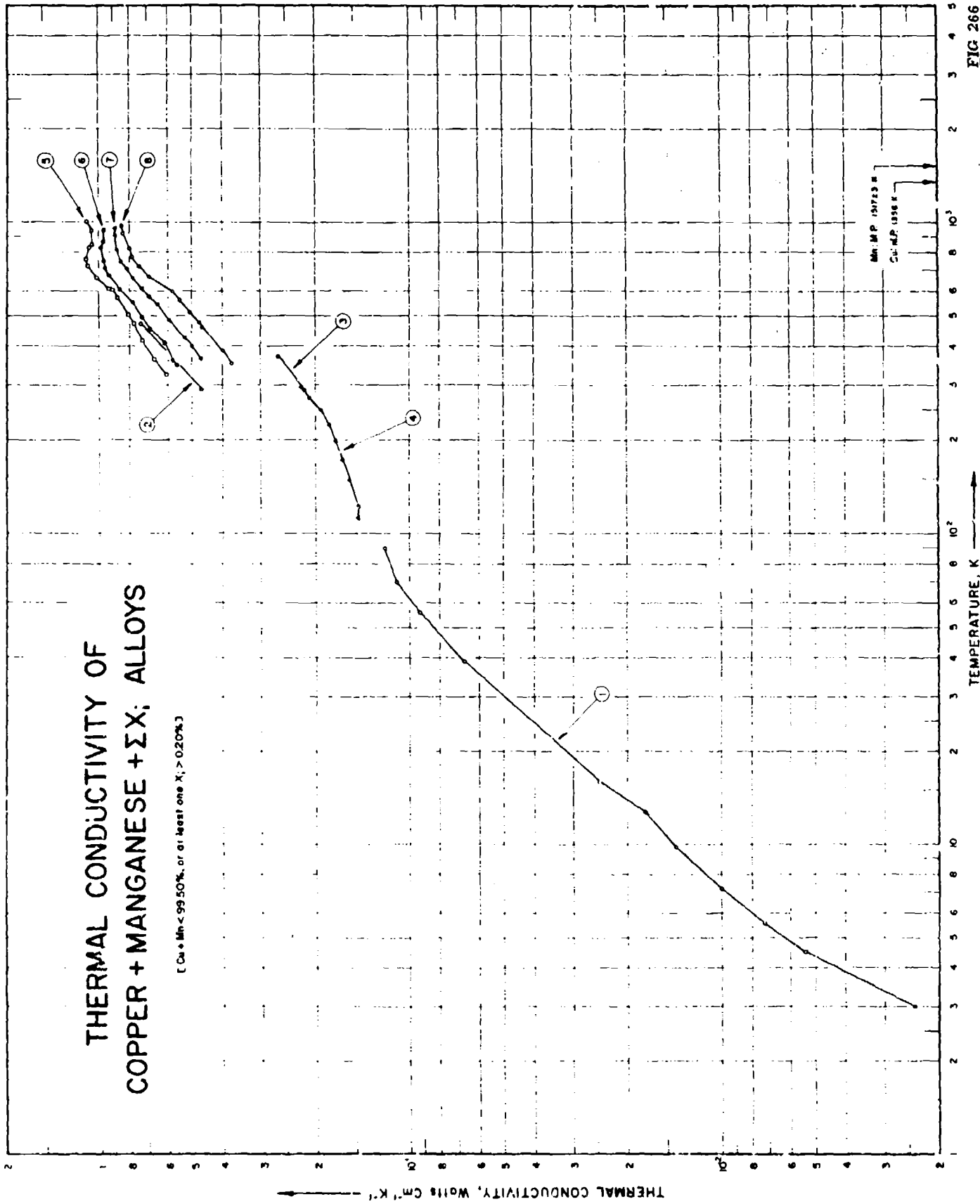


FIG 266

SPECIFICATION TABLE NO. 265 THERMAL CONDUCTIVITY OF [COPPER + MANGANESE + EX<sub>1</sub>] ALLOYS(Cu + Mn < 99.50% or at least one X<sub>1</sub> > 0.20%)

[For Data Reported in Figure and Table No. 266.]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)				Composition (continued), Specifications and Remarks
							Cu	Mn	Ni	Be	
1	154	L	1956	3.0-90	±5	Manganin NM M15	85.0	12.0	3.0		NMIMTs manganin; specimen 3 mm in dia; unannealed.
2	135	L	1955	293-473		Bar 64	95.61	4.51			0.11 Fe; annealed at 700 C; electrical conductivity 5.85 and 5.793 x 10 <sup>4</sup> ohm <sup>-1</sup> cm <sup>-1</sup> at 20 and 200 C, respectively.
3	77	E	1960	291-373		Manganin	84.0	12.0	4.0		Specimen 1.806 cm in dia and 27 cm long; drawn; density 8.44 g cm <sup>-3</sup> .
4	86	L	1908	113-295		Manganin	84.0	12.0	4.0		Turned from a bar; density 8.42 g cm <sup>-3</sup> at 22 C.
5	586		1958	326-1014			95.45	3.65		0.9	Annealed after heating in vacuum at 300 C for 6 hrs.
6	586		1958	348-846			93.63	5.47		0.9	Similar to the above specimen.
7	586		1958	362-963			91.8	7.3		0.9	Similar to the above specimen.
8	586		1958	353-981			89.91	9.12			Similar to the above specimen.

DATA TABLE NO. 266 THERMAL CONDUCTIVITY OF COPPER - MANGANESE - ZINC ALLOYS

(Cu - Mn 99.50% or at least one X<sub>1</sub> = 0.50%)Temperature T, K; Thermal Conductivity k, Watt cm<sup>-2</sup>K<sup>-1</sup>

CURVE 1		CURVE 3 (cont.)		CURVE 5	
T	k	T	k	T	k
3.00	0.00230	507.2	0.503	553.2	0.372
4.00	0.00336	578.2	0.570	586.2	0.297
5.00	0.00524	650.2	0.599	603.2	0.464
7.20	0.00909	633.2	0.923	478.2	0.473
9.30	0.0148	663.2	1.01	516.2	0.506
12.5	0.0176	727.2	1.08	564.2	0.548
16.0	0.024	764.2	1.09	600.2	0.577
20.0	0.0364	828.2	1.07	652.2	0.600
26.0	0.0521	896.2	1.06	726.2	0.711
30.0	0.110	943.2	1.05	778.2	0.757
36.0	0.123	1014.2	1.01	828.2	0.793
				928.2	0.793
				981.2	0.815

CURVE 2		CURVE 6	
T	k	T	k
293.0	0.464	348.2	0.536
473.0	0.736	363.2	0.737
		412.2	0.813
		431.2	0.856
		493.2	0.925
		533.2	0.972
		612.2	0.958
		633.2	0.923
		717.2	0.931
		732.2	0.892
		833.2	0.93
		896.2	0.961
		936.2	0.962

CURVE 4		CURVE 7	
T	k	T	k
113.2	0.146	302.2	0.469
123.2	0.146	311.2	0.493
148.2	0.173	426.2	0.787
153.2	0.093	452.2	0.793
158.2	0.172	512.2	0.613
178.2	0.180	533.2	0.626
223.2	0.192	611.2	0.778
248.2	0.192	661.2	0.771
253.2	0.267	708.2	0.812
291.2	0.248	749.2	0.841
295.2	0.251	816.2	0.874
		901.2	0.887
		963.2	0.887

CURVE 8	
T	k
326.2	0.605
363.2	0.665
418.2	0.724
473.2	0.770

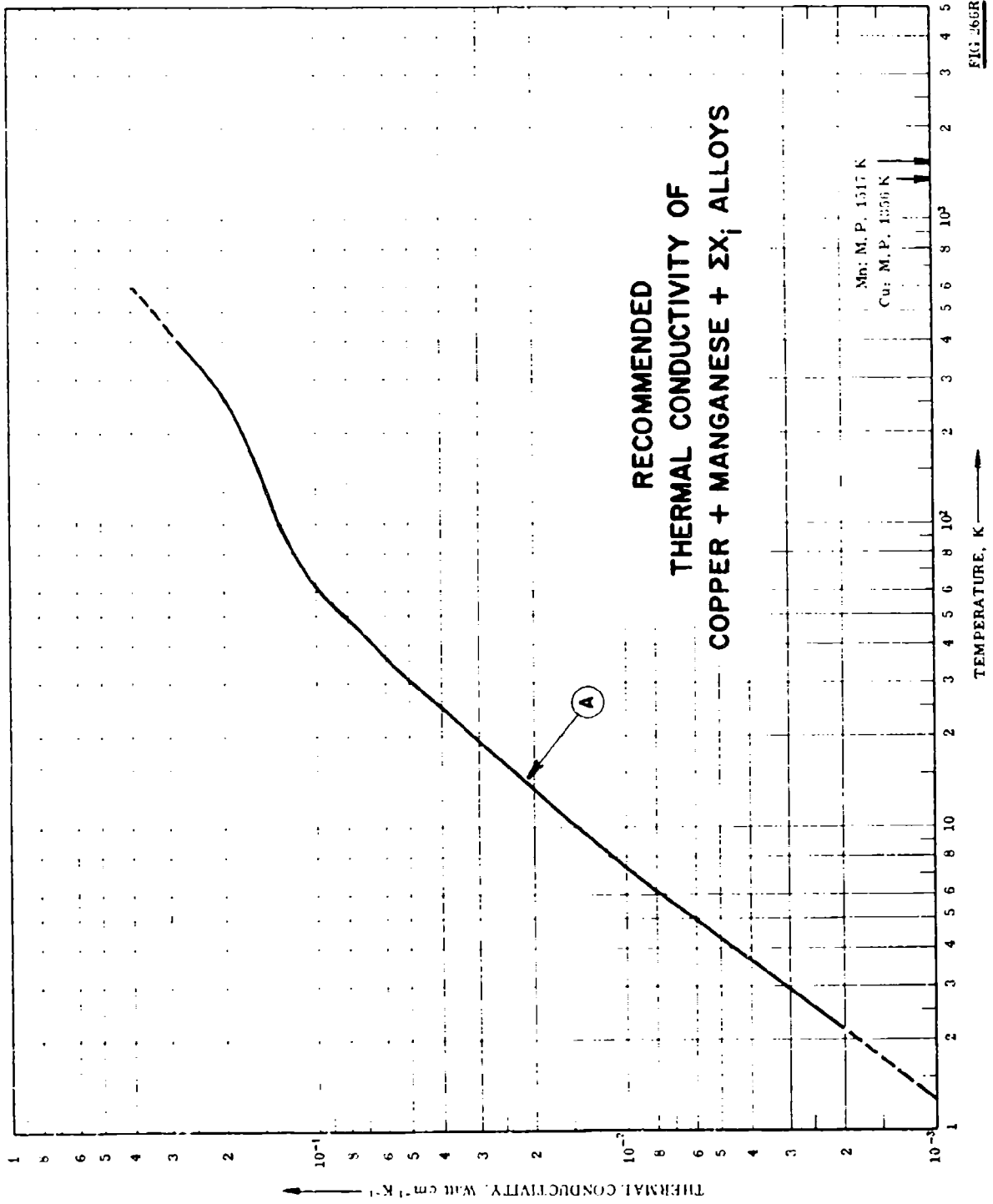


FIG. 266B

SPECIFICATION TABLE NO. 266R RECOMMENDED THERMAL CONDUCTIVITY OF [ COPPER + MANGANESE +  $\Sigma X_i$  ] ALLOYS

[ For Data Reported in Figure and Data Table No. 266R ]

Curve No.	Name and Designation	Nominal Composition (weight percent) and Remarks	T <sub>1</sub> in K		T <sub>2</sub> in F		T <sub>1</sub> in K		T <sub>2</sub> in F		T <sub>1</sub> in K		T <sub>2</sub> in F		Estimated Error	
			k <sub>1</sub>	k <sub>2</sub>	k <sub>1</sub>	k <sub>2</sub>	k <sub>1</sub>	k <sub>2</sub>	k <sub>1</sub>	k <sub>2</sub>	k <sub>1</sub>	k <sub>2</sub>	k <sub>1</sub>	k <sub>2</sub>		
A	Manganin	94 Cu, 12 Mn, and 4 Ni	±5% near room temperature and ±5 to ±10% at other temperatures.													
DATA TABLE NO. 266R RECOMMENDED THERMAL CONDUCTIVITY OF [ COPPER + MANGANESE + $\Sigma X_i$ ] ALLOYS																
[ Temperature, T <sub>1</sub> in K and T <sub>2</sub> in F; Thermal Conductivity, k <sub>1</sub> in Watt cm <sup>-1</sup> K <sup>-1</sup> and k <sub>2</sub> in Btu hr <sup>-1</sup> ft <sup>-1</sup> F <sup>-1</sup> ]																
T <sub>1</sub>	k <sub>1</sub>	k <sub>2</sub>	T <sub>2</sub>	T <sub>1</sub>	k <sub>1</sub>	k <sub>2</sub>	T <sub>2</sub>	T <sub>1</sub>	k <sub>1</sub>	k <sub>2</sub>	T <sub>2</sub>	T <sub>1</sub>	k <sub>1</sub>	k <sub>2</sub>	T <sub>2</sub>	
	CURVE A															
0	0	0	-459.7	15	0.0232	1.34	-432.7	100	0.133	7.64	-279.7	100	0.133	7.64	-279.7	
1	(0.0007)±	(0.0404)	-457.9	16	0.0250	1.44	-430.9	130	0.156	9.01	-199.7	130	0.156	9.01	-199.7	
2	(0.0018)	(0.104)	-456.1	18	0.0285	1.65	-427.3	200	0.172	9.94	-99.7	200	0.172	9.94	-99.7	
3	0.0031	0.179	-454.3	20	0.0322	1.86	-423.7	250	0.193	11.2	-	250	0.193	11.2	-	
4	0.0046	0.266	-452.5	25	0.0410	2.37	-414.1	273.2	0.206	11.9	32.0	273.2	0.206	11.9	32.0	
5	0.0062	0.354	-450.7	30	0.0497	2.87	-405.7	300	0.222	12.8	80.3	300	0.222	12.8	80.3	
6	0.0074	0.451	-448.9	35	0.0583	3.37	-396.7	350	0.250	14.4	176.3	350	0.250	14.4	176.3	
7	0.0095	0.549	-447.1	40	0.067	3.87	-387.7	400	(0.279)	(16.1)	260.3	400	(0.279)	(16.1)	260.3	
8	0.0111	0.641	-445.3	45	0.075	4.31	-378.7	500	(0.334)	(19.5)	440.3	500	(0.334)	(19.5)	440.3	
9	0.0124	0.740	-443.5	50	0.082	4.74	-369.7	600	(0.397)	(22.9)	620.3	600	(0.397)	(22.9)	620.3	
10	0.0145	0.838	-441.7	60	0.097	5.60	-351.7									
11	0.0162	0.936	-439.9	70	0.110	6.36	-333.7									
12	0.0180	1.04	-438.1	80	0.120	6.93	-315.7									
13	0.0197	1.14	-436.3	90	0.127	7.34	-297.7									
14	0.0215	1.24	-434.5													

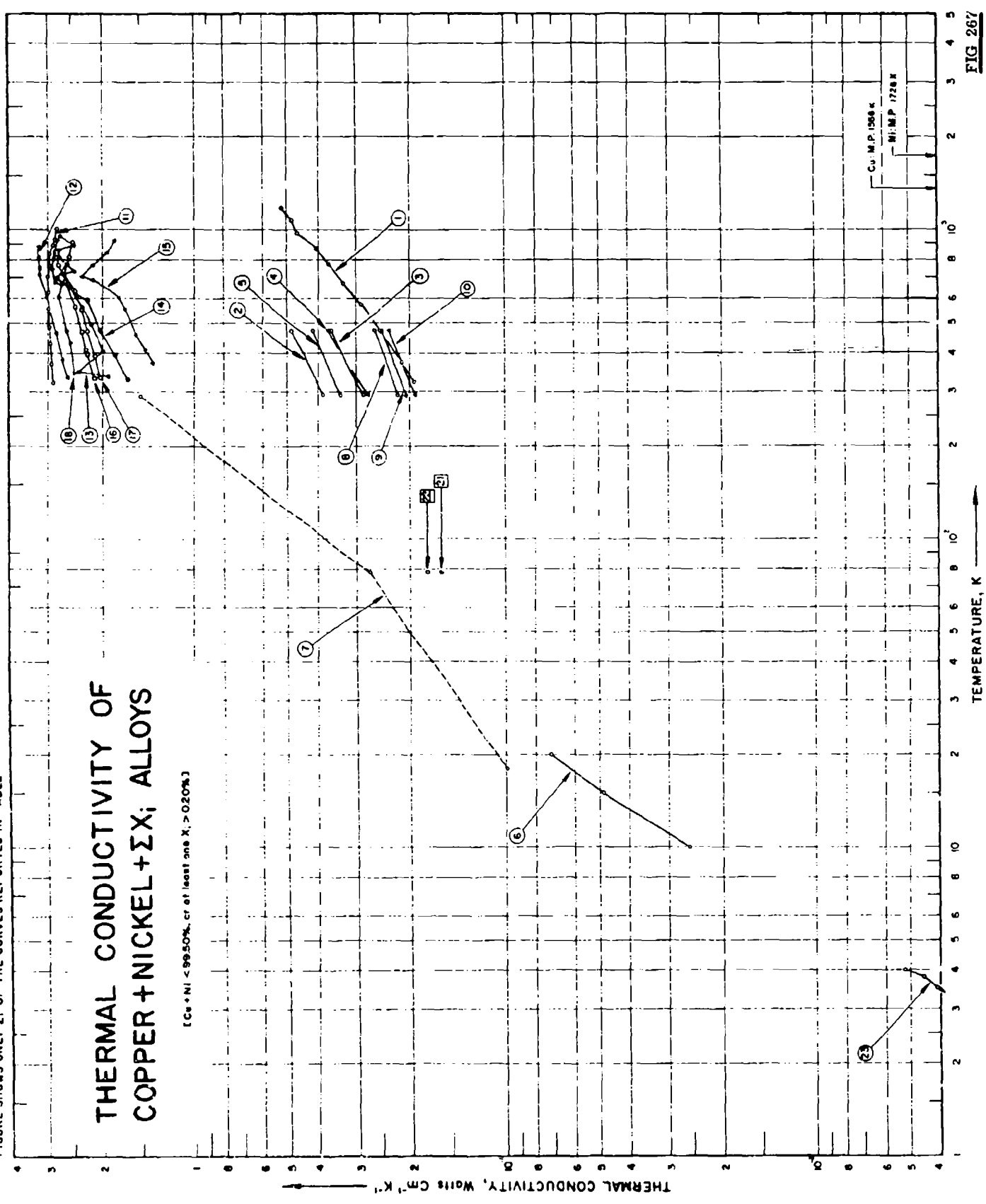
± Values in parentheses are extrapolated.

FIG 267

FIGURE SHOWS ONLY 21 OF THE CURVES REPORTED IN TABLE

# THERMAL CONDUCTIVITY OF COPPER + NICKEL + ΣX; ALLOYS

[Cu + Ni < 99.50%, or at least one X; > 0.20%]





SPECIFICATION TABLE NO. 267 THERMAL CONDUCTIVITY OF [COPPER + NICKEL +  $\Sigma X_i$ ] ALLOYS(Cu + Ni < 99.50% or at least one  $X_i > 0.20\%$ )

[For Data Reported in Figure and Table No. 267]

Curve No.	Ref. Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight per cent)										Zn	Sn	Composition (continued), Specifications and Remarks	
						Cu	Ni	C	Fe	Mn	Pb	Si	Pb	Sn	Zn				
1	131	C	1953	323-1173	2.0	Advance	54.79	44.04	0.035		1.2				0.003			Annealed at 900 C.	
2	135	L	1935	293-473		Bar 32	74.07	19.96		0.09	0.57							5.31	Annealed at 700 C.
3	135	L	1935	293-473		Bar 35	64.15	29.44		0.07	0.52							5.69	Annealed at 700 C.
4	135	L	1935	293-473		Bar 6	63.37	19.89		0.14	0.23			3.31				9.22	Sand-cast.
5	135	L	1935	293-473		Bar 23	64.14	18.38		0.023	0.19							17.06	Annealed at 750 C.
6	152	L	1949	10-20			63.0	20.0										17.0	Severely cold-worked.
7	193	L	1939	18-290		Cupronickel	77.44	20.48										1.99	
8	531	L	1936	293-473		5	62.16	20.22		0.012	0.05			0.003				17.44	0.005 S; cast and machined.
9	531	L	1936	293-473		6	61.96	25.56		0.02	0.07			0.004				12.13	0.005 S; cast and machined.
10	531	L	1936	293-473		7	62.02	29.77		0.019	0.09			0.003				7.93	0.002 S; cast and machined.
11	377		1937	321-1002			99.03	0.6											0.27 Zr, 0.1 P.
12	377		1957	334-884			98.99	0.6											0.26 Zr.
13	378		1957	336-946			99.0	0.8											0.2 Ti.
14	378		1957	329-774			98.85	0.9											0.25 P.
15	378		1957	370-920			96.5	1.2											
16	379		1957	331-815			98.73	0.8											0.33 Zr, 0.14 Be.
17	378		1957	333-910			98.53	1.0											0.33 Zr, 0.14 Be.
18	378		1957	345-923			99.25	0.55											0.17 Zr.
19	378		1957	326-974			99.13	0.62											0.25 Zr.
20	378		1957	333-855			99.3	0.28											0.24 Zr, 0.18 Be.
21	433	L	1940	78.2		6	50.153	49.491		0.06	0.05			trace					0.264 Co, 0.01 Al, 0.008 Sb, 0.004 S; calculated composition.
22	433	L	1940	78.2		7	60.08	39.6		0.066	0.02			trace					0.211 Co, 0.008 Al, 0.009 Sb, 0.004 S; calculated composition.
23	532	L	1960	0.28-4.0		Cupronickel	69.6	30.0		0.40									Nominal composition: annealed and drawn to a 30% reduction in area from super-nickel 702 supplied by Anaconda.

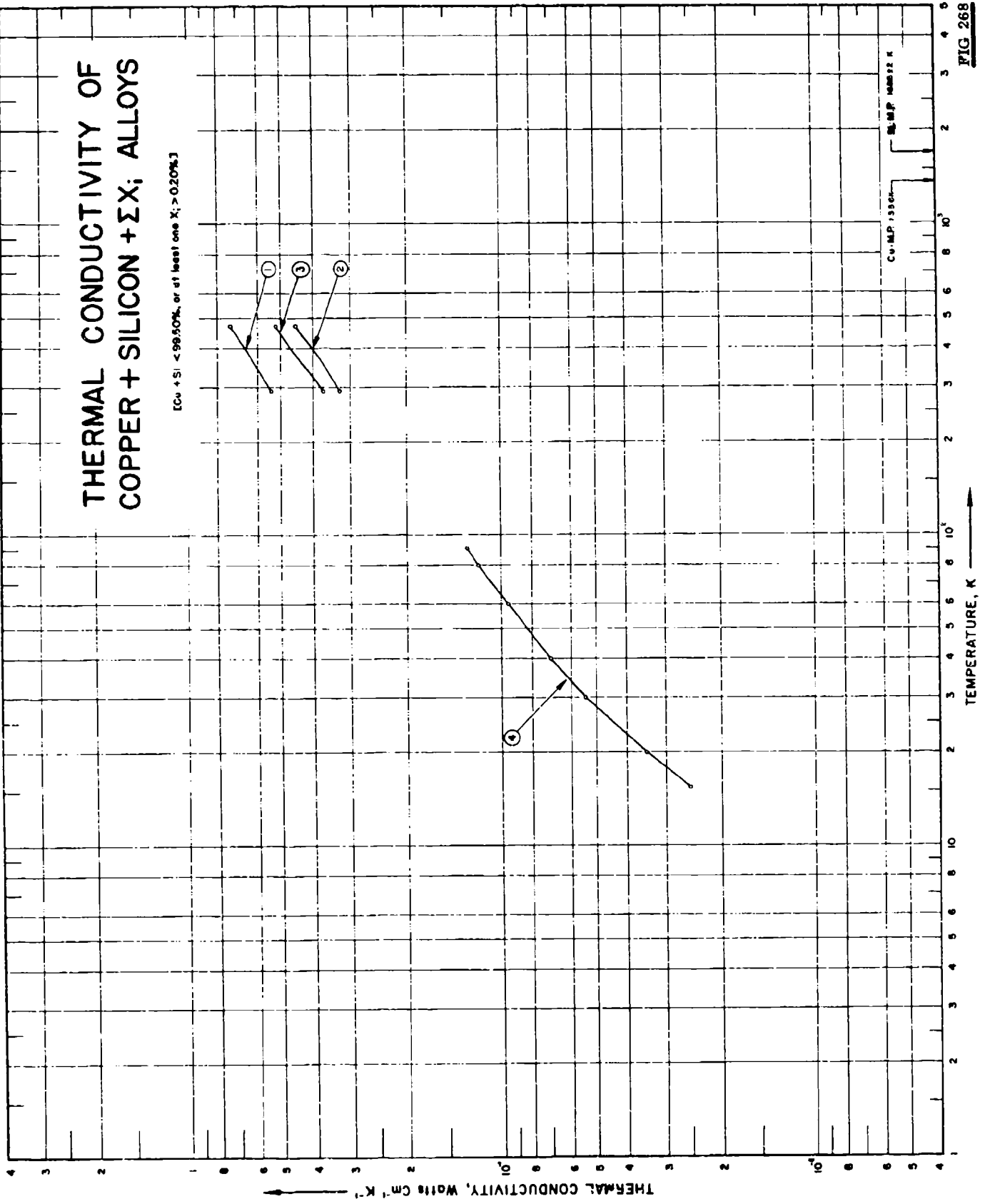
DATA TABLE NO. 267 THERMAL CONDUCTIVITY OF [COPPER + NICKEL +  $\Sigma X_i$ ] ALLOYS(Cu + Ni < 99.50% or at least one  $X_i > 0.20\%$ )

T	k	T	k	T	k	T	k	T	k	T	k
[Temperature, T, K; Thermal Conductivity, k, Watt cm <sup>-1</sup> K <sup>-1</sup> ]											
<u>CURVE 1</u>		<u>CURVE 8</u>		<u>CURVE 13</u>		<u>CURVE 17</u>		<u>CURVE 20(cont.)*</u>			
323.2	0.194	293.2	0.218	336.0	1.91	333.2	2.03	795.2	2.61		
373.2	0.212	473.2	0.254	346.8	2.38	390.2	2.11	854.8	2.54		
473.2	0.251	<u>CURVE 9</u>		404.1	1.99	469.0	2.22	<u>CURVE 21</u>			
573.2	0.289	293.2	0.205	490.7	2.16	555.2	2.31	78.2	0.162		
673.2	0.328	473.2	0.251	563.8	2.31	627.5	2.44	<u>CURVE 22</u>			
773.2	0.367	<u>CURVE 10</u>		713.5	2.66	715.2	2.65	78.2	0.178		
873.2	0.405	293.2	0.192	735.7	2.42	818.0	2.51	<u>CURVE 23</u>			
973.2	0.403	473.2	0.234	811.5	2.73	910.2	2.48	0.275	0.000171*		
1073.2	0.481	<u>CURVE 11</u>		853.5	2.75	<u>CURVE 18</u>		0.325	0.000240*		
1173.2	0.520	293.2	0.285	891.2	2.45	345.0	2.45	0.400	0.000295*		
<u>CURVE 2</u>		473.2	0.405	947.5	2.73	431.9	2.52	0.440	0.000310*		
<u>CURVE 3</u>		<u>CURVE 12</u>		<u>CURVE 14</u>		<u>CURVE 19*</u>		0.460	0.000360*		
293.2	0.285	334.2	2.57	329.2	1.64	369.8	1.36	0.475	0.000400*		
473.2	0.360	466.2	2.79	392.0	1.81	408.8	1.43	0.810	0.000695*		
<u>CURVE 4</u>		554.7	2.97*	474.2	2.03	458.0	1.54	0.930	0.000800*		
293.2	0.276	604.5	3.03	474.2	2.67	458.0	1.54	1.07	0.000990*		
473.2	0.368	715.2	3.13	664.2	2.74	498.6	2.61	1.51	0.00150*		
<u>CURVE 5</u>		817.2	3.18	674.6	2.82	498.6	2.61	1.70	0.00173*		
293.2	0.335	884.2	3.20	686.2	2.15	598.5	2.74	1.85	0.00195*		
473.2	0.418	<u>CURVE 15</u>		702.2	2.31	677.0	2.90	2.05	0.00210*		
<u>CURVE 6</u>		379.7	2.87	705.3	2.99	702.2	2.31	2.40	0.00255*		
10.0	0.0255	466.2	2.79	767.0	2.94	766.5	2.14	2.85	0.00310*		
15.0	0.0485	554.7	2.97*	833.7	2.94	807.1	2.97	3.50	0.00410		
20.0	0.0711	604.5	3.03	889.5	2.87	843.2	1.92	4.00	0.00520		
<u>CURVE 7</u>		715.2	3.13	928.3	2.84	908.9	2.93				
18.0	0.0987	817.2	3.19	1007.2	2.77	974.3	2.85				
78.0	0.272	<u>CURVE 16</u>		<u>CURVE 19*</u>		<u>CURVE 20*</u>					
290.0	1.490	331.2	2.12	369.8	1.36	333.0	2.15				
		396.2	2.23	408.8	1.43	392.7	2.21				
		409.2	2.25	458.0	1.54	490.0	2.35				
		469.0	2.31	552.2	1.67	575.8	2.45				
		563.5	2.43	606.3	1.76	656.2	2.49				
		678.8	2.60	686.2	2.15	728.3	2.57				
		766.2	2.75	702.2	2.31	815.2	2.79				
		815.2	2.79	766.5	2.14						
		919.8	1.83	843.2	1.92						
				974.3	2.85						

\* Not shown on plot

# THERMAL CONDUCTIVITY OF COPPER + SILICON + $\Sigma X$ ; ALLOYS

$Cu + Si < 99.50\%$ , or at least one  $X$ ;  $> 0.20\%$



SPECIFICATION TABLE NO. 268 THERMAL CONDUCTIVITY OF [COPPER + SILICON +  $\Sigma X_i$ ] ALLOYS(Cu + Si < 99.50% or at least one  $X_i > 0.20\%$ )

[For Data Reported in Figure and Table No. 268]

Curve No.	Ref. Method No. Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)				Zn	Composition (continued), Specifications and Remarks
						Cu	Si	Fe	Mn		
1	135	L	1935	293.473	Bar 136	98.1	1.5	0.06	0.3		Annealed at 700 C for 2 hrs; electrical conductivity 6.611 and $6.022 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 20 and 200 C, respectively.
2	135	L	1935	293.473	Bar 72	95.69	3.23	0.16	0.99		Annealed at 750 C for 1-1/2 hrs; electrical conductivity 3.773 and $3.611 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 20 and 200 C, respectively.
3	135	L	1935	293.473	Bar 135	95.83	3.11	0.02		1.12	Annealed at 700 C for 2 hrs; electrical conductivity 4.586 and $4.228 \times 10^4 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 20 and 200 C, respectively.
4	432	L	1957	16-91	Silicon bronze-A	≈94.72	3.15	0.001	1.13	1.0	0.001 each Cd, Cr, Ag, 0.001 > each B, Ca, Al, Pb, Sn; turned and ground from a hard temper rod.

## DATA TABLE NO. 268 THERMAL CONDUCTIVITY OF [COPPER + SILICON + EX.] ALLOYS

(Cu + Si = 99.50% or at least one  $X_1 = 0.20\%$ )[Temperature, T, K; Thermal Conductivity, k, Watt  $\text{cm}^{-1} \text{K}^{-1}$ ]

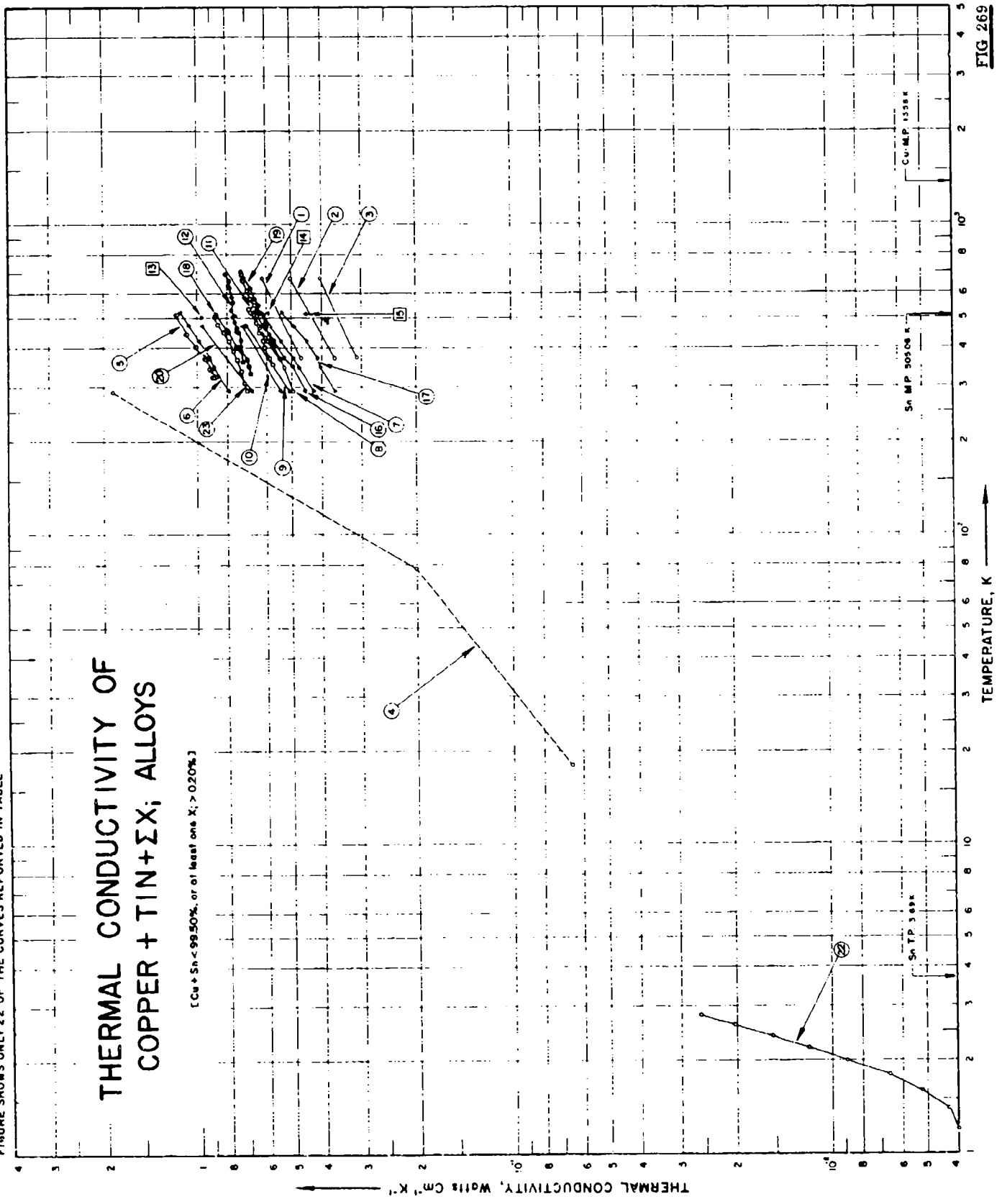
T k

CURVE 1293.2 0.540  
473.2 0.732CURVE 2293.2 0.226  
473.2 0.448CURVE 3293.2 0.372  
473.2 0.518CURVE 415.5 0.0250  
20.0 0.0345  
30.0 0.0540  
40.0 0.0695  
60.0 0.095  
80.0 0.117  
91.0 0.128

FIGURE SHOWS ONLY 22 OF THE CURVES REPORTED IN TABLE

# THERMAL CONDUCTIVITY OF COPPER + TIN+ΣX; ALLOYS

[Cu + Sn < 99.50% or of least one X; > 0.20%]



## SPECIFICATION TABLE NO. 269 THERMAL CONDUCTIVITY OF [COPPER + TIN + EX.] ALLOYS

(Cu + Sn  $\leq$  99.50% or at least one X<sub>i</sub>  $\geq$  0.20%)

[For Data Reported in Figure and Table No. 269]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)					Zn	Composition (continued), Specifications and Remarks	
							Cu	Sn	Fe	Ni	P			Pb
1	215	L	1939	373, 673	2.0	1	87.93	5.44		4.96		1.65	Cast.	
2	215	L	1939	373, 673	2.0	2	73.23	16.06		9.67		0.99	Cast.	
3	215	L	1939	373, 673	2.0	3	58.0	30.71		10.29		1.0	Cast.	
4	193	L	1939	18-290		Phosphor Bronze	84.89	13.07			0.013	0.16	1.86	Annealed at 650 C.
5	134	L	1931	324-512	2.0	42	94.02	4.88	0.03		0.06	1.16		
6	55	F	1928	293-523		Bronze; bar 1	92.8	5.0			0.15		2.0	
7	55	F	1928	293-523		Bronze; bar 2	87.8	10.0			0.15		2.0	
8	55	F	1928	293-523		Bronze; bar 5	88.0	10.0					2.0	
9	135	F	1935	293, 473		Bar 15 A	88.39	9.55	0.07				1.9	Sand-cast.
10	135	F	1935	293, 473		Bar 9	87.86	8.87	0.03				3.05	Sand-cast.
11	30	L	1925	357-692	<2.0	Admiralty Gun-Metal	87.24	10.02	0.21			0.35	2.19	Cast.
12	30	L	1925	361-692	<2.0	Gun-Metal (ordinary)	85.95	8.72	0.21			0.98	5.04	Cast.
13	224	L	1923	506		SAE Bearing Alloy No. 40	84.93	5.14				5.01	4.92	
14	224	L	1923	521		SAE Bearing Alloy No. 62	86.6	10.55				0.04	2.81	
15	224	L	1923	518		SAE Bearing Alloy No. 64	79.04	10.83			0.3	9.55		Annealed at 650 C.
16	55	F	1928	293-523		Phosphor bronze; bar 3	91.7	8.0			0.3			
17	55	F	1928	293-523		Bronze; bar 6	87.2	12.4			0.4			
18	134	L	1931	331-514	2.0	40	95.56	4.18	0.01		0.33	0.04		
19	30	L	1925	368-704	2.0	Phosphor Bronze	87.82	11.28	0.17		0.35			Cast.
20	516	L	1941	293, 473		Phosphor bronze; 2	96.5	3.09	0.01	0.01	0.39	<0.005		<0.005 Sb; cast, after air cooling annealed at 625 C, cold-rolled; machined.
21	516	L	1941	293, 473		Phosphor bronze; 4	92.2	7.41	0.02		0.38	0.01		<0.005 Sb; cast, after air cooling annealed at 625 C, hot-rolled at 300 C and annealed for 2.5 hrs at 625 C, and again hot rolled at 300 C and annealed for 2.5 hrs at 625 C, and then cold-rolled and machined.
22	530	L	1954	1.2-2.9	<10	Phosphor Bronze								Composition not reported; in wire form, 40 $\mu$ in dia.

SPECIFICATION TABLE NO. 269 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)					Composition (continued), Specifications and Remarks		
							Cu	Sn	Fe	Ni	P		Pb	Zn
23	529	C	1958	292-504	± 5.0	Navy "M"	68.0	5.7	0.07	0.6	0.01	1.4	4.4	0.13 S <sub>2</sub> ; average composition; density 8.64 g cm <sup>-3</sup> , M. P. 988 C; electrical resistivity 12.04, 12.46, 12.71, 12.97, 13.22, 13.48, 13.74, and 13.99 μ ohm cm at 20, 66, 93, 121, 149, 177, 204, and 232 C, respectively.



DATA TABLE NO. 269 THERMAL CONDUCTIVITY OF [COPPER + Tm + EX<sub>1</sub>] ALLOYS

(Cu + Sn < 99.50% or at least one X<sub>i</sub> > 0.20%)

[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup> K<sup>-1</sup>]

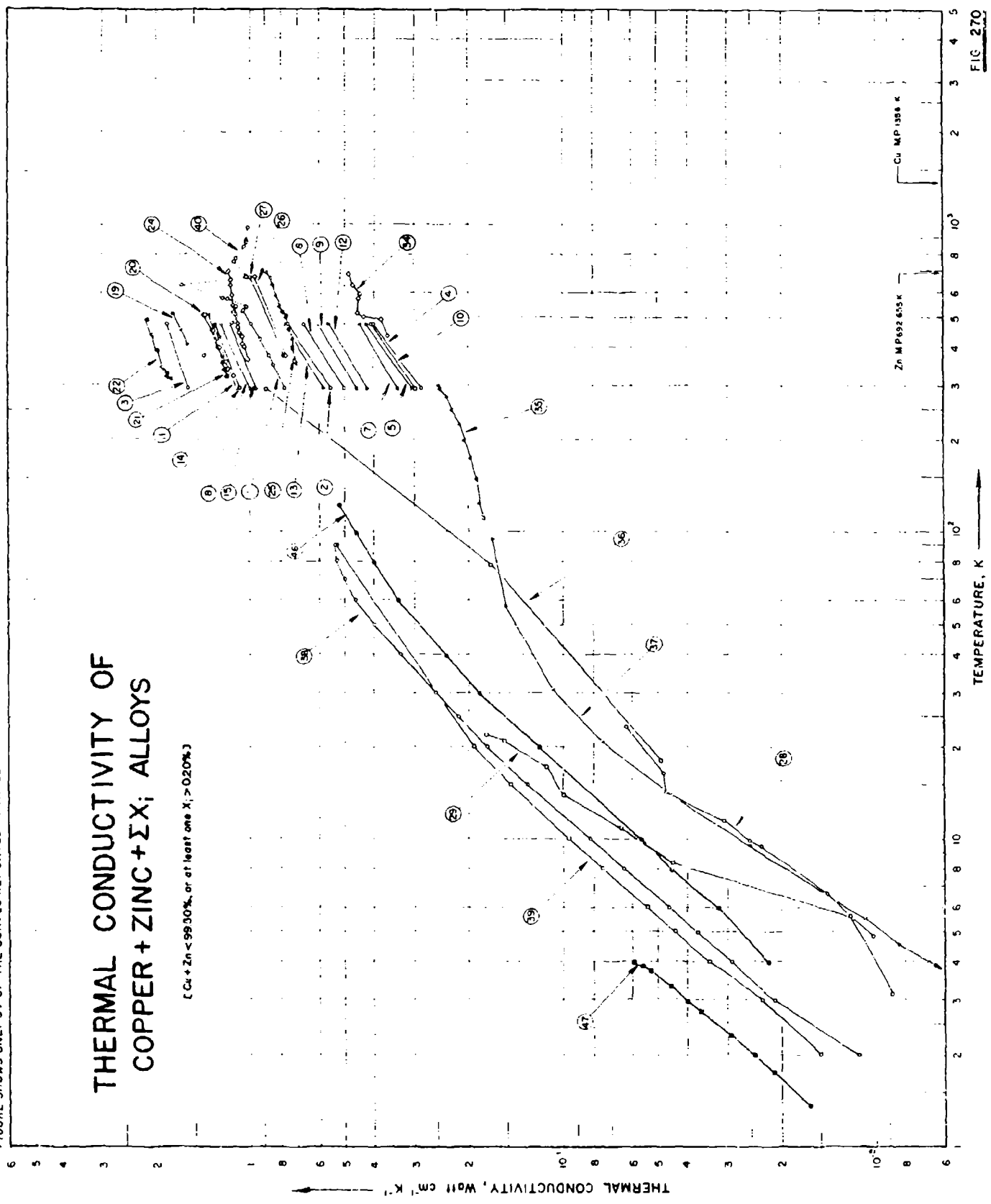
T	k	T	k	T	k	T	k	T	k
<u>CURVE 1</u>		<u>CURVE 6(cont.)</u>		<u>CURVE 11(cont.)</u>		<u>CURVE 17</u>		<u>CURVE 21*</u>	
373.2	0.464	523.2	1.138	578.2	0.678	293.2	0.364	293.2	0.460
673.2	0.619	<u>CURVE 7</u>		621.7	0.690	343.2	0.402	473.2	0.628
<u>CURVE 2</u>		293.2	0.427	660.2	0.715	373.2	0.414	<u>CURVE 22</u>	
373.2	0.364	348.2	0.473	691.7	0.720	423.2	0.448	1.2	0.0040
673.2	0.502	373.2	0.494	<u>CURVE 12</u>		473.2	0.490	1.4	0.0043
<u>CURVE 3</u>		423.2	0.540	361.2	0.715	523.2	0.536	1.6	0.0052
373.2	0.310	473.2	0.582	391.7	0.724	<u>CURVE 18</u>		1.8	0.0066
673.2	0.402	523.2	0.626	406.7	0.728	391.2	0.678	2.0	0.0089
<u>CURVE 4</u>		<u>CURVE 8</u>		423.7	0.724	350.2	0.686	2.2	0.0118
373.2	0.310	293.2	0.494	450.7	0.732	369.2	0.695	2.4	0.0154
673.2	0.402	348.2	0.527	455.7	0.749	404.2	0.736	2.6	0.0202
18	0.0653	373.2	0.548	467.2	0.741	405.2	0.749	2.8	0.0260
73	0.201	423.2	0.594	488.7	0.753	454.2	0.795	<u>CURVE 23</u>	
290	1.883	473.2	0.636	508.7	0.757	456.2	0.799	292.2	0.696
<u>CURVE 5</u>		523.2	0.678	529.7	0.774	514.2	0.864	309.9	0.704
324.2	0.883	<u>CURVE 9</u>		561.7	0.762	<u>CURVE 19</u>		337.7	0.722
324.2	0.887	293.2	0.502	585.2	0.770	367.7	0.540	365.5	0.741
325.2	0.666	473.2	0.695	629.7	0.782	403.2	0.569	393.3	0.762
341.2	0.908	<u>CURVE 10</u>		659.7	0.787	415.2	0.577	421.1	0.787
342.2	0.912	293.2	0.540	691.7	0.808	425.2	0.577	448.8	0.819
342.2	0.991	473.2	0.715	<u>CURVE 13</u>		445.7	0.586	476.6	0.856
359.2	0.937	506.0	0.982	506.0	0.982	463.2	0.603	504.4	0.872
360.2	0.941	<u>CURVE 11</u>		<u>CURVE 14</u>		477.2	0.598		
361.2	0.921	293.2	0.573	521.0	0.594	488.2	0.603		
404.2	1.000	473.2	0.715	<u>CURVE 15</u>		514.2	0.615		
404.2	1.004	356.7	0.573	373.7	0.553	519.7	0.636		
454.2	1.075	402.2	0.607	402.2	0.607	537.2	0.649		
512.2	1.172	425.7	0.615	318.0	0.456	550.2	0.636		
512.2	1.167	433.7	0.603	<u>CURVE 16</u>		576.2	0.657		
<u>CURVE 6</u>		449.7	0.632	293.2	0.452	628.2	0.674		
293.2	0.791	466.2	0.636	348.2	0.498	704.2	0.728		
348.2	0.970	487.2	0.649	373.2	0.523	<u>CURVE 20</u>		293.2	0.669
373.2	0.908	512.2	0.649	423.2	0.569	293.2	0.669	473.2	0.962
423.2	0.963	514.7	0.657	473.2	0.615				
473.2	1.059	534.2	0.686	523.2	0.667				
		551.2	0.657						

\* Not shown on plot

FIGURE SHOWS ONLY 34 OF THE CURVES REPORTED IN TABLE

# THERMAL CONDUCTIVITY OF COPPER + ZINC + ΣX; ALLOYS

(Cu + Zn < 99.90%, or at least one X; > 0.20%)





SPECIFICATION TABLE NO. 270 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)							Composition (continued), Specifications and Remarks					
							Cu	Zn	Al	Fe	Mn	Ni	Pb		Sn				
32	37	C	1951	366-589	4.0	Brass	61.5	35.5											
33	77	E	1900	291,373		German red brass	85.7	7.15											
34	6	L	1931	432-688		Platinoid	62.0	22.0											Density at 22 C 8.66 g cm <sup>-3</sup> .
35	88	L	1908	110-298		Platinoid	62.0	22.0											Density at 22 C 8.66 g cm <sup>-3</sup> .
36	193	L	1939	18-290		Argentan	47.0	40.45											
37	9	L	1951	2.1-94		German Silver	47.0	41.0											As received.
38	229	L	1955	2.0-90		B.S. 249 Brass	55.0/ 60.0	Bal											0.75 impurities; as received.
39	229	L	1955	2.0-90		B.S. 249 Brass	55.0/ 60.0	Bal											0.75 impurities; heated to a dull red; heated for ten min and cooled slowly.
40	175	P	1936	373-373			71.00	28.43											Trace Si; annealed at 650 C for 1.5 hrs.
41	2	L	1948	2-4		Nickel Silver	45.9	42.1											Machined nearly to size, then annealed in a.r., followed by final light cut.
42	531	L	1936	293,473		Nickel Silver 1	62.62	27.14											Trace 0.007 C, 0.003 S; 0.009 Si; cast and machined.
43	531	L	1936	293,473		Nickel Silver 2	63.17	24.31											Trace 0.007 C, 0.003 S; 0.009 Si; cast and machined.
44	531	L	1936	293,473		Nickel Silver 3	62.43	22.08											Trace 0.014 C, 0.004 S; 0.003 Si; cast and machined.
45	531	L	1936	293,473		Nickel Silver 4	62.05	19.36											Trace 0.008 C, 0.003 S; 0.004 Si; cast and machined.
46	432	L	1957	4-124		Free-cutting leaded brass	60.01	35.7	<0.001	0.01									0.01 Bi, 0.01 Cd, 0.01 Ag, 0.001 Sb, 0.001 In, <0.001 As, <0.001 Co, and <0.0001 Mg, 0.001 Si; hard tempered and ground.
47	518	L	1960	1.4-4.0		Z10	89.02	8.55											Annealed for 17 hrs at 500 C at ordinary atmosphere.
48	618	C	1960	313-328	±3.0	Yellow brass	72	22											Specimen 20 mm in dia and 18 mm long; steel used as comparative material.
49	618	C	1960	305-325	±3.0	Yellow brass	72	22											Specimen 20 mm in dia and 18 mm long; pure Ni used as comparative material.

SPECIFICATION TABLE NO. 270 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)							Composition (continued), Specifications and Remarks	
							Cu	Zn	Al	Fe	Mn	Ni	Pb		Sh
50	618	C	1969	301-320	+3.0	Yellow brass	72	22					4	2	Specimen 20 mm in dia and 18 mm long; yellow brass used as comparative material.
51	618	C	1969	306-318	+3.0	Yellow brass	72	22					4	2	Specimen 20 mm in dia and 18 mm long; Al used as comparative material.
52	511		1918	296.3		Red brass	~80/ 82	~12/ 7					5/~7	3/~4	Specimen 1.400 cm in dia; manufactured by Erba Co.
53	765	C	1957	298.2		Brass									Thermal comparator loaded with 100 gram weight applied on the plane lapped surface of the specimen.
54	851	L	1960	80-275	10	Brass	62.0	35.0					3.0		Free cutting yellow brass; 0.1877 in. dia and 2.224 in. long; turned down from a 0.375 in. dia rod obtained from commercial stock of J. M. Tull Metal and Supply Co.; data corrected for rise in temperature during measurement.
55	970	L	1959	85-118	<0.5	Brass	62.0	35.0					3.0		Specimen 0.25 in. in dia and 7.875 in. long; prepared from half-hard tempered drawn brass.

DATA TABLE NO. 270 THERMAL CONDUCTIVITY OF COPPER + ZINC +  $\Sigma X_i$  ALLOYS(Cu + Zn < 99.50% or at least one  $X_i > 0.20\%$ )[Temperature, T, K; Thermal Conductivity, k, Watt  $\text{cm}^{-1}\text{K}^{-1}$ ]

T	k	T	k	T	k	T	k	T	k	T	k	T	k	T	k	T	k	T	k
CURVE 1																			
293.00	0.787	293.00	0.297	320.20	1.121	584.70	0.837	323.20	0.977	2.10	0.00267 <sup>a</sup>	373.20	1.414	CURVE 40					
348.00	0.849	473.00	0.410	321.40	1.121	532.20	0.858	333.20	1.036	2.33	0.00314 <sup>a</sup>	473.20	1.155	CURVE 37					
373.00	0.879	CURVE 11		338.00	1.142	665.20	0.870	CURVE 32*		2.66	0.00377 <sup>a</sup>	479.20	1.143	CURVE 41					
423.00	0.937	CURVE 20		338.50	1.151	697.70	0.895	CURVE 36		3.23	0.00480 <sup>a</sup>	538.20	1.030	CURVE 42*					
473.00	0.996	321.00	1.136	353.80	1.163	CURVE 24		366.50	1.157	3.69	0.00582 <sup>a</sup>	573.20	1.188	CURVE 43*					
523.00	1.054	321.00	1.136	356.50	1.180	373.00	0.774	422.00	1.215	4.04	0.00677 <sup>a</sup>	577.20	1.231	CURVE 44*					
CURVE 2		321.00	1.136	392.40	1.230	673.00	0.971	477.50	1.272	4.55	0.00743	635.20	1.671	CURVE 45*					
293.00	0.557	357.20	1.192	393.70	1.243	CURVE 27		533.20	1.332	5.53	0.0107	673.20	1.033						
473.00	0.762	398.80	1.253	442.80	1.297	444.20	1.510	588.70	1.389	9.50	0.0254	677.20	1.028						
CURVE 3		398.80	1.253	444.20	1.510	499.00	1.372	CURVE 33*		14.10	0.0455	751.20	1.123						
293.00	0.423	450.10	1.326	500.50	1.385	373.00	0.787	291.20	0.397	20.70	0.0751	773.20	1.109						
473.00	0.565	507.40	1.406	507.70	1.414	673.00	1.000	373.20	0.710	30.70	0.107	842.20	1.049						
CURVE 4		293.00	0.586	363.20	1.013	CURVE 28		CURVE 34		57.30	0.152	873.20	1.033						
473.00	1.858	399.20	1.046	408.20	1.054	4.85	0.0102	CURVE 35		94.50	0.167	888.20	1.030						
CURVE 5		320.50	1.197	431.70	1.059	6.65	0.0142	CURVE 36		2.00	0.013	973.20	1.013						
293.00	0.285	337.90	1.218	446.20	1.084	9.44	0.0231	CURVE 37		3.00	0.021	1.40	0.00100						
473.00	0.402	338.00	1.201	463.70	1.096	9.84	0.0253	CURVE 38		5.00	0.029	1.75	0.00351						
CURVE 6		356.00	1.222	474.23	1.096	11.50	0.0307	CURVE 39		5.00	0.037	1.95	0.00460						
293.00	0.305	395.40	1.259	501.20	1.105	14.20	0.0471	CURVE 40		6.00	0.046	2.20	0.00540						
473.00	0.427	395.40	1.255	510.70	1.113	16.40	0.0480	CURVE 41		8.00	0.064	2.45	0.00760						
CURVE 7		401.40	1.239	517.20	1.134	23.10	0.0627	CURVE 42*		10.00	0.082	3.40	0.00211						
293.00	0.962	447.60	1.318	540.70	1.117	CURVE 29		CURVE 43*		15.00	0.130	4.00	0.00345						
473.00	1.159	448.50	1.314	541.70	1.138	3.15	0.00869	CURVE 35		20.00	0.175	CURVE 42*							
CURVE 8		456.60	1.318	587.70	1.146	5.62	0.0120	CURVE 36		25.00	0.215	CURVE 43*							
293.00	0.502	504.70	1.381	634.20	1.159	8.36	0.0449	CURVE 37		30.00	0.255	CURVE 44*							
473.00	0.678	506.10	1.377	659.70	1.159	10.88	0.0653	CURVE 38		40.00	0.330	CURVE 45*							
CURVE 9		517.20	1.381	702.20	1.172	13.90	0.100	CURVE 39		50.00	0.400								
293.00	0.339	CURVE 22		354.70	0.716	17.080	0.113	CURVE 40		60.00	0.460								
473.00	0.448	316.80	1.812	390.70	0.732	20.80	0.153	CURVE 41		70.00	0.500								
CURVE 10		320.50	1.866	397.20	0.732	21.80	0.176	CURVE 42*		80.00	0.530								
293.00	1.008	329.70	1.862	423.70	0.736	CURVE 30*		CURVE 43*		90.00	0.540								
473.00	1.234	333.00	1.895	445.20	0.749	293.20	1.084	CURVE 39		2.00	0.015								
CURVE 11		343.30	1.912	454.70	0.757	323.20	1.132	CURVE 40		3.00	0.023								
293.00	1.138	346.00	1.925	469.70	0.753	353.20	1.180	CURVE 41		4.00	0.034								
473.00	1.297	389.80	1.975	496.20	0.778	CURVE 31*		CURVE 42*		5.00	0.044								
CURVE 12		389.80	2.000	503.70	0.770	18.00	0.0490	CURVE 43*		6.00	0.054								
293.00	0.460	438.00	2.067	518.70	0.795	78.00	0.170	CURVE 44*		8.00	0.075								
473.00	0.611	438.10	2.084	535.20	0.812	293.20	0.922	CURVE 45*		10.00	0.096								
CURVE 13		490.40	2.172	552.70	0.812	CURVE 40		CURVE 46*		15.00	0.146								
405.00	1.603	490.50	2.146			CURVE 41		CURVE 47*		20.00	0.193								

<sup>a</sup> Not shown on plot

DATA TABLE NO. 270 (continued)

$\bar{r}$	k	T	k	T
<u>CURVE 46</u>				
4.0	0.022	313.5	1.119	
6.0	0.032	316.4	1.133	
8.0	0.043	319.7	1.129	
10.0	0.056	320.0	1.140	
20.0	0.119			
30.0	0.185	<u>CURVE 51*</u>		
40.0	0.235	305.7	1.110	
60.0	0.335	308.7	1.116	
80.0	0.400	311.7	1.110	
100.0	0.460	314.7	1.127	
124.0	0.520	317.7	1.135	
<u>CURVE 47</u>				
1.37	0.0164	<u>CURVE 52*</u>		
1.76	0.0211	296.3	1.095	
2.0	0.0245	<u>CURVE 53*</u>		
2.31	0.0294	298.2	1.14	
2.79	0.0363	<u>CURVE 54*</u>		
3.0	0.0400	79.88	0.434	
3.38	0.0455	92.40	0.470	
3.75	0.0529	113.36	0.542	
3.88	0.0557	133.44	0.610	
4.0	0.0592	158.78	0.708	
<u>CURVE 48*</u>				
312.7	1.097	177.60	0.774	
320.2	1.109	198.01	0.855	
327.7	1.119	217.34	0.940	
<u>CURVE 49*</u>				
305.0	1.069	240.84	1.05	
310.7	1.086	256.64	1.11	
317.9	1.102	275.36	1.18	
325.1	1.116	<u>CURVE 55*</u>		
<u>CURVE 50*</u>				
84.67	0.406	88.73	0.498	
98.90	0.523	109.01	0.584	
118.24	0.605			
301.2	1.074			
303.7	1.099			
306.9	1.110			
309.7	1.148			
310.2	1.132			
313.4	1.122			

\* Not shown on plot

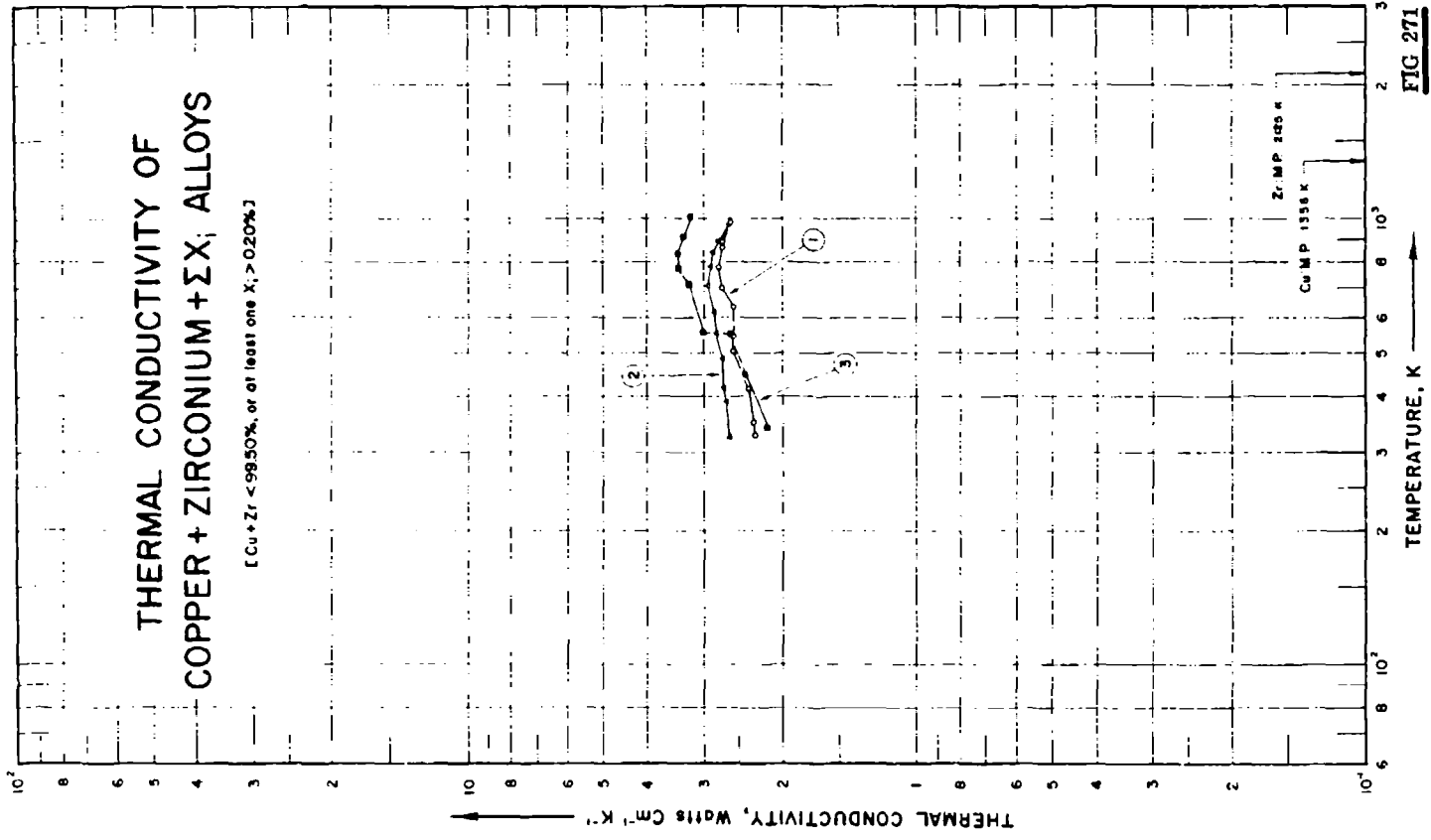


FIG 271



SPECIFICATION TABLE NO. 271 THERMAL CONDUCTIVITY OF [COPPER + ZIRCONIUM +  $\Sigma X_j$ ] ALLOYS(Cu + Zr = 99.50% or at least one  $X_j > 0.20\%$ )

(For Data Reported in Figure and Table No. 271.)

Curve No.	Ref. Method No. Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)				Composition (continued), Specifications and Remarks	
						Cu	Zr	Co	Al		Cr
1	377	1957	328-982			99.26	0.29	0.25	0.2		Electrical conductivity 28.34, 26.94, 23.25, 20.75, 19.10, 16.5, 15.85, 14.45, 12.93, 12.55 and 10.83 x 10 <sup>4</sup> ohm <sup>-1</sup> cm <sup>-1</sup> at 54.6, 77.1, 142.8, 234.1, 273.0, 366.6, 427.1, 507.1, 587.6, 618.6 and 708.6 C, respectively.
2	377	1957	321-975			99.4	0.2	0.3			Electrical conductivity 33.55, 28.12, 26.35, 23.18, 20.95, 18.9, 17.1, 15.33, 14.10, 12.68 and 10.95 x 10 <sup>4</sup> ohm <sup>-1</sup> cm <sup>-1</sup> at 47.25, 115.9, 145.8, 214.8, 281.8, 346.4, 416.6, 508.6, 567, 618.3 and 701.6 C, respectively.
3	378	1957	346-1014			72.8	27.0		0.2		Electrical resistivity 4.0, 4.77, 5.32, 5.5, 5.56, 5.79, 6.26, 6.9 and 7.95 $\mu$ ohm cm at 67, 176.4, 277.2, 283.8, 339.5, 506, 565, 637.1 and 740.6 C, respectively.

DATA TABLE NO. 271 THERMAL CONDUCTIVITY OF [COPPER + ZIRCONIUM +  $\Sigma X_i$ ] ALLOYS(Cu + Zr + 99.50% or at least one  $X_i > 0.20\%$ )[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup> K<sup>-1</sup>]

T k

CURVE 1

327.8	2.31
350.3	2.33
416.0	2.38
507.3	2.58
546.8	2.57
639.8	2.58
700.3	2.74
780.3	2.79
860.8	2.73
991.8	2.74
981.8	2.60

CURVE 2

320.5	2.62
389.1	2.68
419.0	2.71
488.0	2.74
555.0	2.82
619.6	2.85
709.8	2.95
781.8	2.90
840.2	2.87
891.5	2.79
974.8	2.62

CURVE 3

340.2	2.16
449.6	2.44
550.4	2.62
557.0	3.03
712.7	3.25
779.2	3.41
838.2	3.42
910.3	3.34
1013.9	3.22

# THERMAL CONDUCTIVITY OF LANTHANUM + NEODYMIUM + ΣX<sub>i</sub> ALLOYS

[La + Nd < 99.50%, or at least one X<sub>i</sub> > 0.20%]

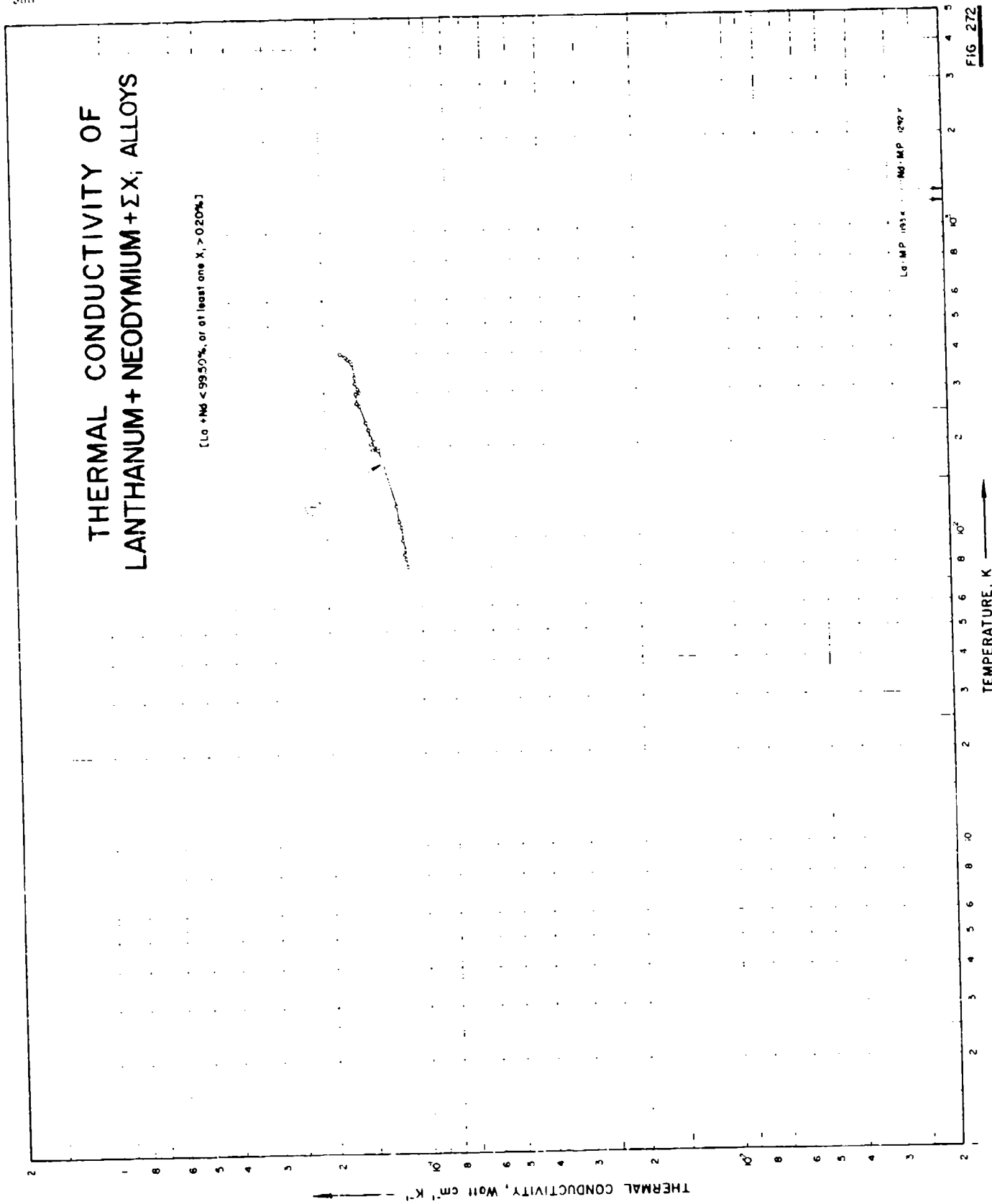


FIG 272

SPECIFICATION TABLE NO. 272 THERMAL CONDUCTIVITY OF [LANTHANUM + NEODYMIUM +  $\Sigma X_i$ ](La + Nd) 99.50% or at least one  $X_i > 0.20\%$ 

[ For Data Reported in Figure and Table No. 272 ]

Curve No.	ReL No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	La	Nd	Ca	Ce	Cu	Fe	Ta	Composition (continued), Specifications and Remarks
1	932.933	L	1966	81-401	$\pm 3$ to $\pm 5$			0.5	0.25	0.1	0.005	0.04	0.15	Polycrystalline with a hexagonal structure; obtained from lot 499; electrical resistivity reported as 39.2, 50.5, 59.2, 68.0, 74.0, 79.7, 84.6, and 98.8 $\mu\text{ohm cm}$ at 99, 152, 203, 253, 301, 348, 399, and 448 K, respectively; measured in a vacuum of $10^{-4}$ $\sim 10^{-3}$ mm Hg.

DATA TABLE NO. 272 THERMAL CONDUCTIVITY OF [LANTHANUM + NEODYMIUM +  $\Sigma X_i$ ] ALLOYS(La + Nd < 99.50% or at least one  $X_i > 0.20\%$ )[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup> K<sup>-1</sup>]

T	k
81	0.111
86	0.112
99	0.114
92	0.113
99	0.115
109	0.116
114	0.118
128	0.120
195	0.136
196	0.141
199	0.136
201	0.139*
205	0.142
208	0.139
216	0.144
227	0.145
241	0.147
273	0.154
277	0.157
279	0.153*
299	0.159
301	0.154
305	0.157*
309	0.156
314	0.155*
322	0.159
339	0.159
374	0.162
376	0.166*
380	0.164
386	0.167
393	0.171
401	0.177

\* Not shown on plot

SPECIFICATION TABLE NO. 273 THERMAL CONDUCTIVITY OF [LEAD + ANTIMONY +  $\Sigma X_1$ ] ALLOYS Pb + Sb +  $\Sigma X_1$   
 (Pb + Sb < 99.50% or at least one  $X_1 > 0.20\%$ )

Curve No.	Ref. Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)	Composition (continued), Specifications and Remarks
1	224	L	1923	34 <sup>s</sup>	SAF bearing Ay. No. 12	63.94 Pb, 28.84 Sb, and 7.07 Cu.	

DATA TABLE NO. 273 THERMAL CONDUCTIVITY OF [LEAD + ANTIMONY +  $\Sigma X_1$ ] ALLOYS Pb + Sb +  $\Sigma X_1$   
 (Pb + Sb < 99.50% or at least one  $X_1 > 0.20\%$ )

[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup> K<sup>-1</sup>]

T            k  
 CURVE 1\*  
 348.0        0.318

\* No graphical presentation

# THERMAL CONDUCTIVITY OF LITHIUM+BORON+ΣX; ALLOYS

[Li + B < 99.50%, or at least one X, > 0.20%]

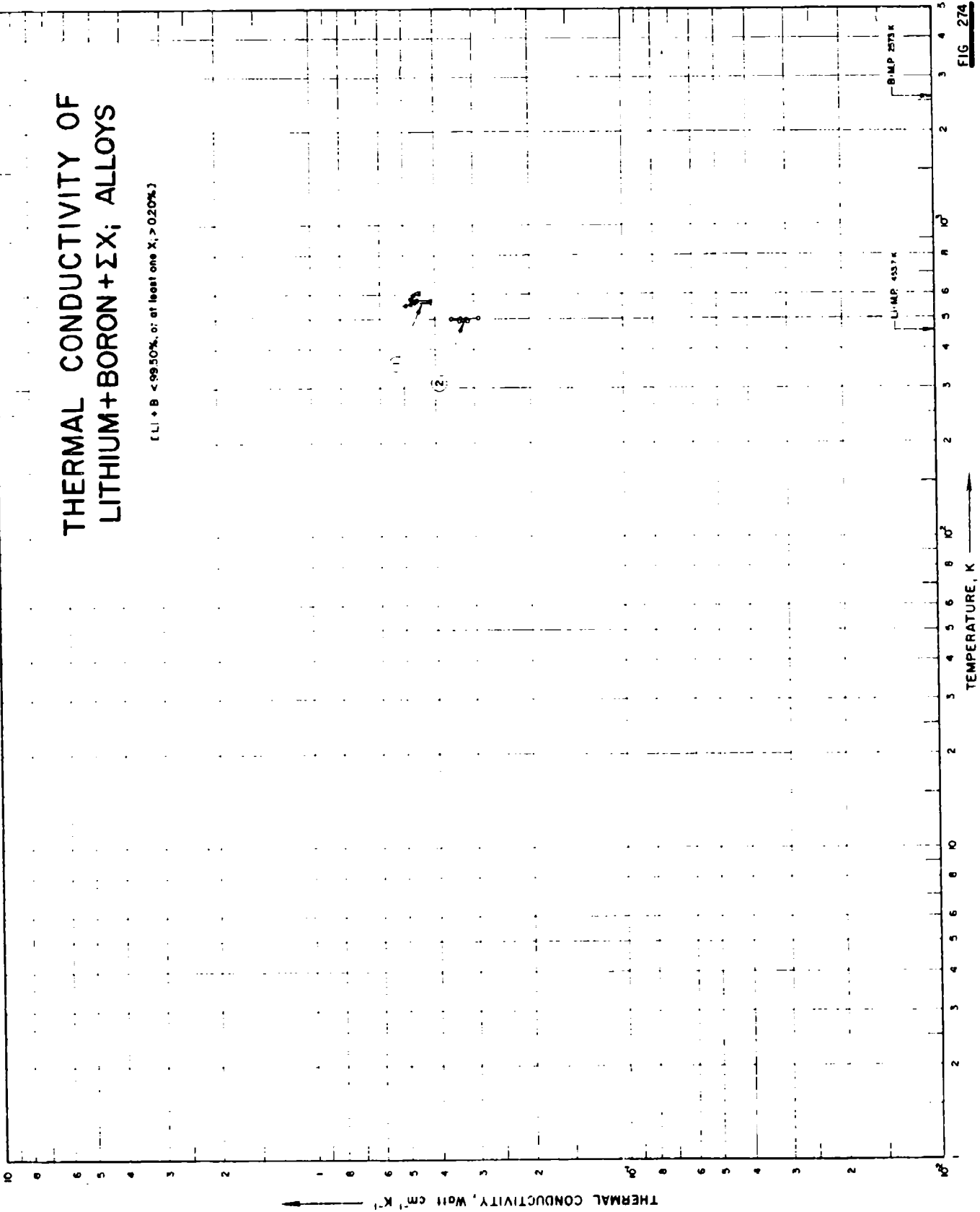


FIG. 274

SPECIFICATION TABLE NO. 274 THERMAL CONDUCTIVITY OF [LITHIUM + BORON + EX<sub>1</sub>] ALLOYS(L<sub>1</sub> + B < 99.50% or at least one X<sub>1</sub> > 0.20%)

[ For Data Reported in Figure and Table No. 274 ]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) L <sub>1</sub>	Composition (weight percent) B	Composition (continued), Specifications and Remarks
1	243	L	1950	553-606	±10		99.22	0.2	0.1 Ca, <0.1 HG, <0.1 P, 0.1 Al, 0.02 Cr, 0.04 Cu <0.01 K, <0.1 Na, <0.01 Ni; measured in liquid state, apparatus in open air at room temperature.
2	243	L	1950	491-505	±10		99.22	0.2	0.1 Ca, <0.1 HG, <0.1 P, 0.1 Al, 0.02 Cr, 0.04 Cu, <0.01 K, <0.1 Na, <0.01 Ni; measured in liquid state, apparatus in heated oven.



DATA TABLE NO. 274 THERMAL CONDUCTIVITY OF [LITHIUM + BORON + EX<sub>1</sub>] ALLOYS

(Li + B < 99.50% or at least one X<sub>1</sub> > 0.20%)

[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup> K<sup>-1</sup>]

T	k
<u>CURVE 1</u>	
553.2	0.494
556.2	0.477
566.2	0.464
568.2	0.414
568.2	0.414*
568.2	0.414*
572.2	0.414
572.2	0.473
573.2	0.456
584.2	0.477
598.2	0.464
605.2	0.452
605.2	0.448
<u>CURVE 2</u>	
491.2	0.335
493.2	0.314
493.2	0.318*
501.2	0.356
501.2	0.331
501.2	0.318
501.2	0.322*
501.2	0.318*
505.2	0.289

\* Not shown on plot

# THERMAL CONDUCTIVITY OF LITHIUM + SODIUM + ΣX; ALLOYS

(Li + Na < 99.50%, or at least one X; > 0.20%)

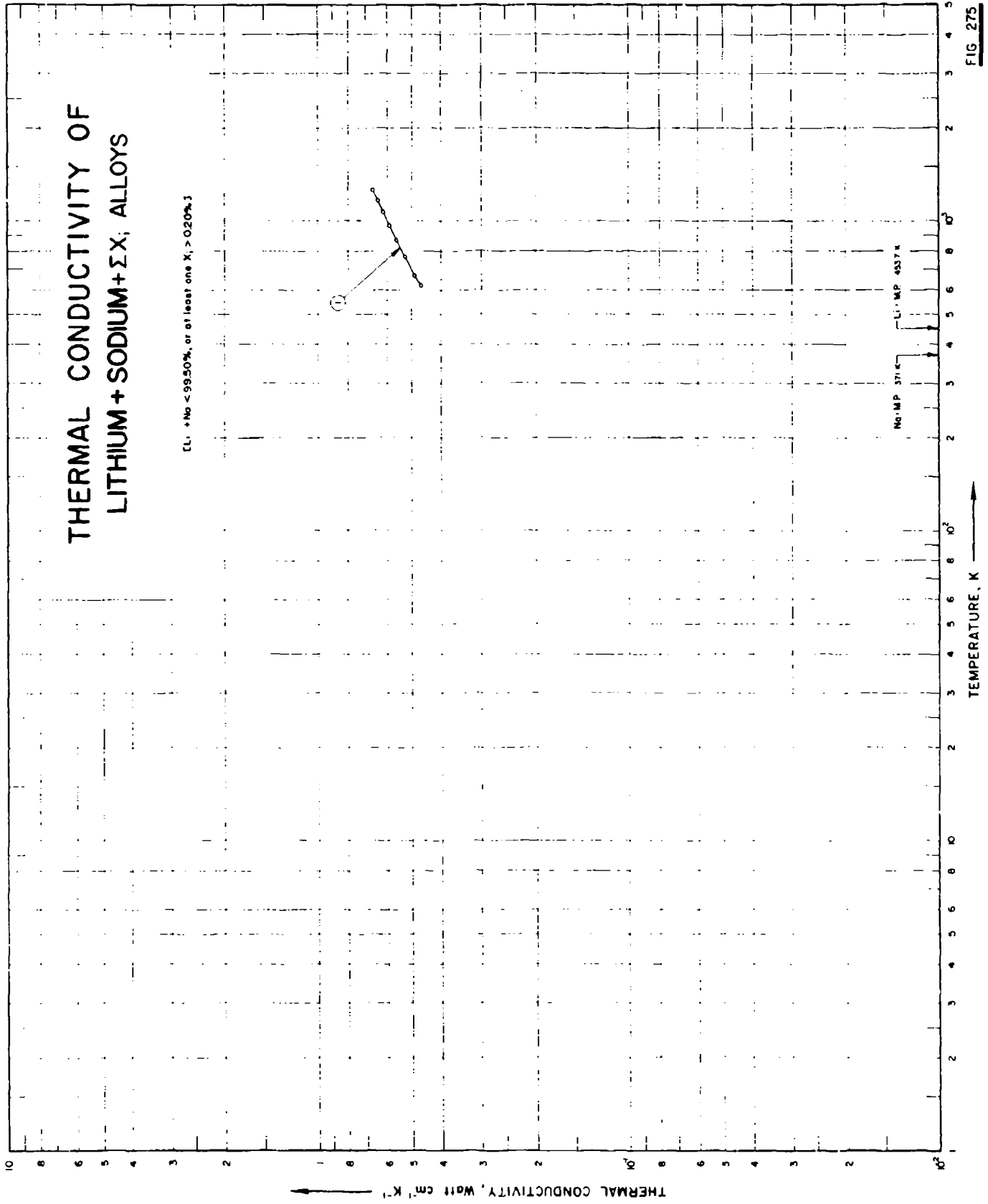


FIG 275

SPECIFICATION TABLE NO. 275 THERMAL CONDUCTIVITY OF [LITHIUM + SODIUM + EX<sub>1</sub>] ALLOYS  
 (Li + Na > 99.50% or at least one X<sub>1</sub> > 0.20%)

For Data Reported in Figure and Table No. 275.

Curve No.	Ref. Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) Li Na	Composition (continued), Specifications and Remarks
1	769, 866 P	1962	623-1273			99.98 0.27	0.19 Fe, 0.13 Mg, 0.08 C, 0.06 Cu, 0.052 Ni, 0.05 Cr, 0.032 Pb, 0.023 Sn, 0.016 Ti, 0.01 Ce, <0.01 Sb, <0.01 Zr, 0.006 Ba, 0.058 Mo, 0.0046 Ca, 0.0044 N, 0.0042 V, 0.0037 Al, 0.003 K, 0.0029 Mn, 0.002 Bi, <0.001 Re, 0.001 Cd, 0.001 In and <0.003 Ag; filtered through a capillary with an I. D. of 15 mm, poured in a vacuum of $\sim 1 \times 10^{-2}$ mm Hg into a thin-walled steel (1 Kh 18N 9T) tube 8.6 mm in dia., 230 mm long and 0.2 mm wall thickness; measured in vacuum.

DATA TABLE NO. 275 THERMAL CONDUCTIVITY OF (LITHIUM + SODIUM +  $\Sigma X_i$ ) ALLOYS(Li + Na < 99.50% or at least one  $X_i > 0.20\%$ )[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup> K<sup>-1</sup>]

T	k
623.2	0.469
673.2	0.490
773.2	0.527
873.2	0.561
973.2	0.594
1073.2	0.621
1173.2	0.644
1273.2	0.669

FIGURE SHOWS ONLY 7 OF THE CURVES REPORTED IN TABLE

**THERMAL CONDUCTIVITY OF  
MAGNESIUM + ALUMINUM + ΣX<sub>i</sub> ALLOYS**

(Mg + Al < 99.50%, or at least one X<sub>i</sub> > 0.20%)

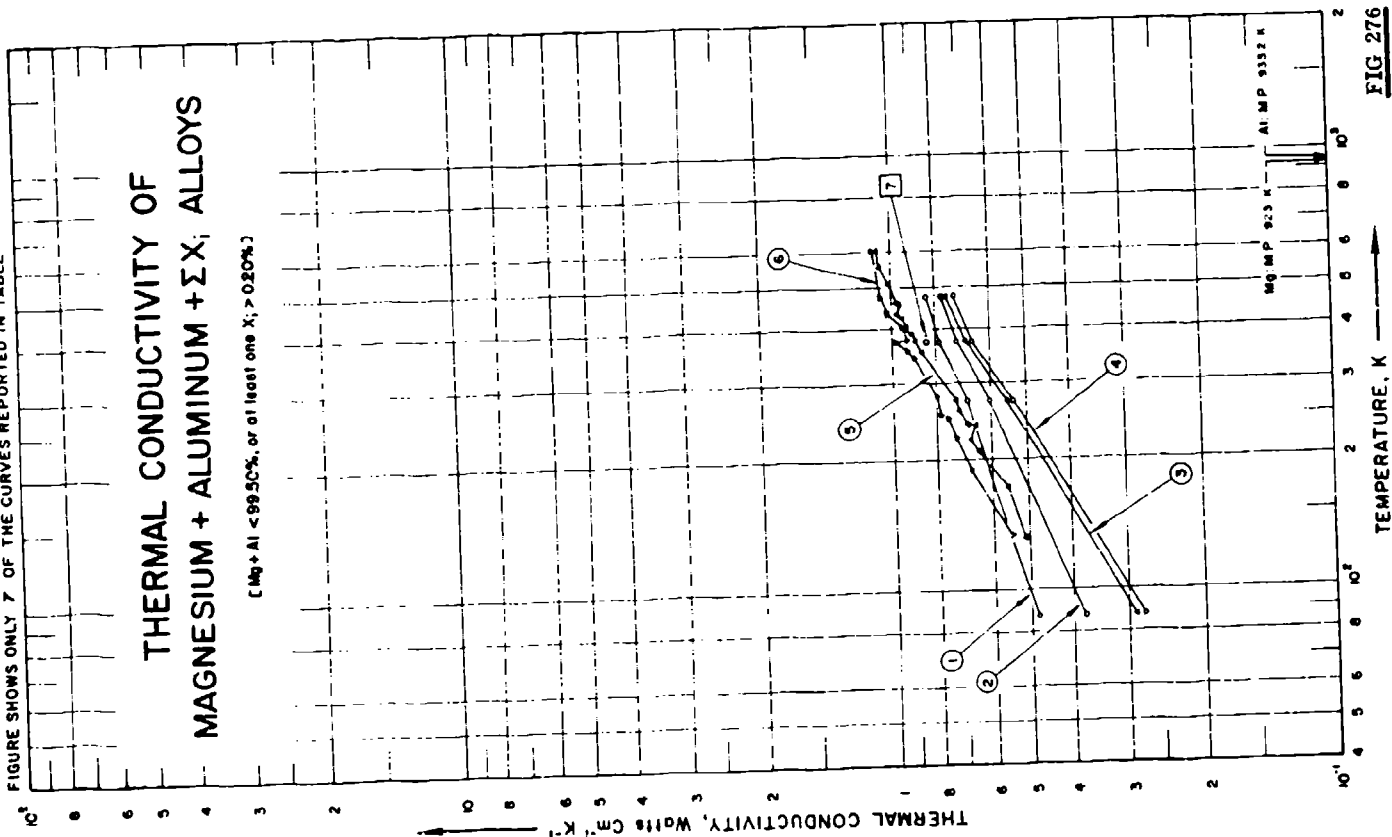


FIG 276

## SPECIFICATION TABLE NO. 276 THERMAL CONDUCTIVITY OF [MAGNESIUM + ALUMINUM + Sn] ALLOYS

(Mg + Al  $\leq$  99.50% or at least one  $X_i > 0.20\%$ )

[ For Data Reported in Figure and Table No. 276. ]

Curve No.	ReL No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)						Composition (continued), Specifications and Remarks		
							Mg	Al	Cu	Fe	Mn	Ni		Si	Zn
1	93,850	L	1931, 1929	87-476	3-4		92.0	9.0					2.0		As cast; electrical conductivity 9.98, 7.24, 6.31 and 5.75 x 10 <sup>4</sup> ohm <sup>-1</sup> cm <sup>-1</sup> at 87, 273, 373 and 476 K, respectively.
2	93,850	L	1931, 1929	87-476	3-4		90.0	9.0					2.0		As cast; electrical conductivity 9.15, 6.42, 5.53 and 5.19 x 10 <sup>4</sup> ohm <sup>-1</sup> cm <sup>-1</sup> at 87, 273, 373 and 476 K, respectively.
3	93,850	L	1931, 1929	87-476	3-4		88.0	10.0					2.0		As cast; electrical conductivity 8.83, 6.28, 5.16 and 4.70 x 10 <sup>4</sup> ohm <sup>-1</sup> cm <sup>-1</sup> at 87, 273, 373 and 476 K, respectively.
4	93,850	L	1931, 1929	87-476	3-4		86.0	12.0					2.0		As cast; electrical conductivity 8.71, 6.20, 5.10 and 4.62 x 10 <sup>4</sup> ohm <sup>-1</sup> cm <sup>-1</sup> at 87, 273, 373 and 476 K, respectively.
5	91	C	1951	130-605		AN-M-29	Bal	2.5/ 3.5	0.05	0.005	0.2	0.005	0.3	0.7/ 1.3	0.3 total other impurities; specimen 2 cm in dia and 15 cm long; hot-rolled parallel to the heat flow direction; annealed for 1 hr at 600 C; Armo iron used as comparative material.
6	91	C	1951	134-607		AN-M-29	Bal	2.5/ 3.5	0.05	0.005	0.2	0.005	0.3	0.7/ 1.3	0.3 total other impurities; specimen 2 cm in dia and 15 cm long; hot-rolled perpendicular to the heat flow direction; annealed for 1 hr at 600 C; Armo iron used as comparative material.
7	53	E	1927	373	± 1.0		94.0	4.0	2.0						Forged.
8	673	E	1932	295.2	± 1.3	Fickton 2	93.0	4.0						1.0	1.0 Cd, 1.0 Sn.
9	673	E	1932	301.6	± 1.3	Dow metal 3	90.6	6.0	0.4					3.0	
10	673	E	1932	295.3	± 1.3	Dow metal 4	92.5	4.0						0.5	2.0 Cd, 1.0 Sn.
11	673	E	1932	305.3	± 1.3	Dow metal 5	92.0	4.0							3.0 Cd, 1.0 Sn.
12	673	E	1932	303.3	± 1.3	Dow metal 6	92.0	4.0							2.0 Cd, 2.0 Sn.

DATA TABLE NO. 276 THERMAL CONDUCTIVITY OF [MAGNESIUM + ALUMINUM +  $\Sigma X_i$ ] ALLOYS(Mg + Al)  $\leq$  99.50% or at least one  $X_i > 0.20\%$ [Temperature, T, K; Thermal Conductivity, k, Watt  $\text{cm}^{-1} \text{K}^{-1}$ ]

T	k	T	k	T	k
<u>CURVE 1</u>		<u>CURVE 5 (cont.)</u>		<u>CURVE 11*</u>	
87.00	0.481	431.10	0.975	305.3	0.695
273.00	0.646	454.60	0.982	<u>CURVE 12*</u>	
373.00	0.787	458.00	0.975	303.3	0.556
476.00	0.837	506.80	1.013		
<u>CURVE 2</u>		604.80	1.071		
87.00	0.377	<u>CURVE 6</u>			
273.00	0.607	133.90	0.548		
373.00	0.720	189.80	0.674		
476.00	0.778	223.60	0.728		
<u>CURVE 3</u>		248.20	0.757		
87.00	0.289	252.30	0.742		
273.00	0.552	290.00	0.799		
373.00	0.682	311.40	0.895		
476.00	0.753	356.40	0.929		
<u>CURVE 4</u>		371.60	0.943		
87.00	0.275	376.60	0.925		
273.00	0.536	401.50	0.950		
373.00	0.661	435.70	1.033		
476.00	0.728	438.80	1.033		
<u>CURVE 5</u>		470.70	1.063		
130.80	0.510	606.80	1.100		
171.50	0.552	<u>CURVE 7</u>			
207.70	0.644	373.00	0.837		
222.30	0.674	<u>CURVE 8*</u>			
239.90	0.653	295.2	0.556		
240.10	0.686	<u>CURVE 9</u>			
257.00	0.711	301.6	0.611*		
273.20	0.728	<u>CURVE 10*</u>			
354.90	0.858	295.3	0.632		
374.40	0.879				
377.90	0.900				
400.00	0.941				
403.50	0.925				
421.60	0.946				

\* Not shown on plot

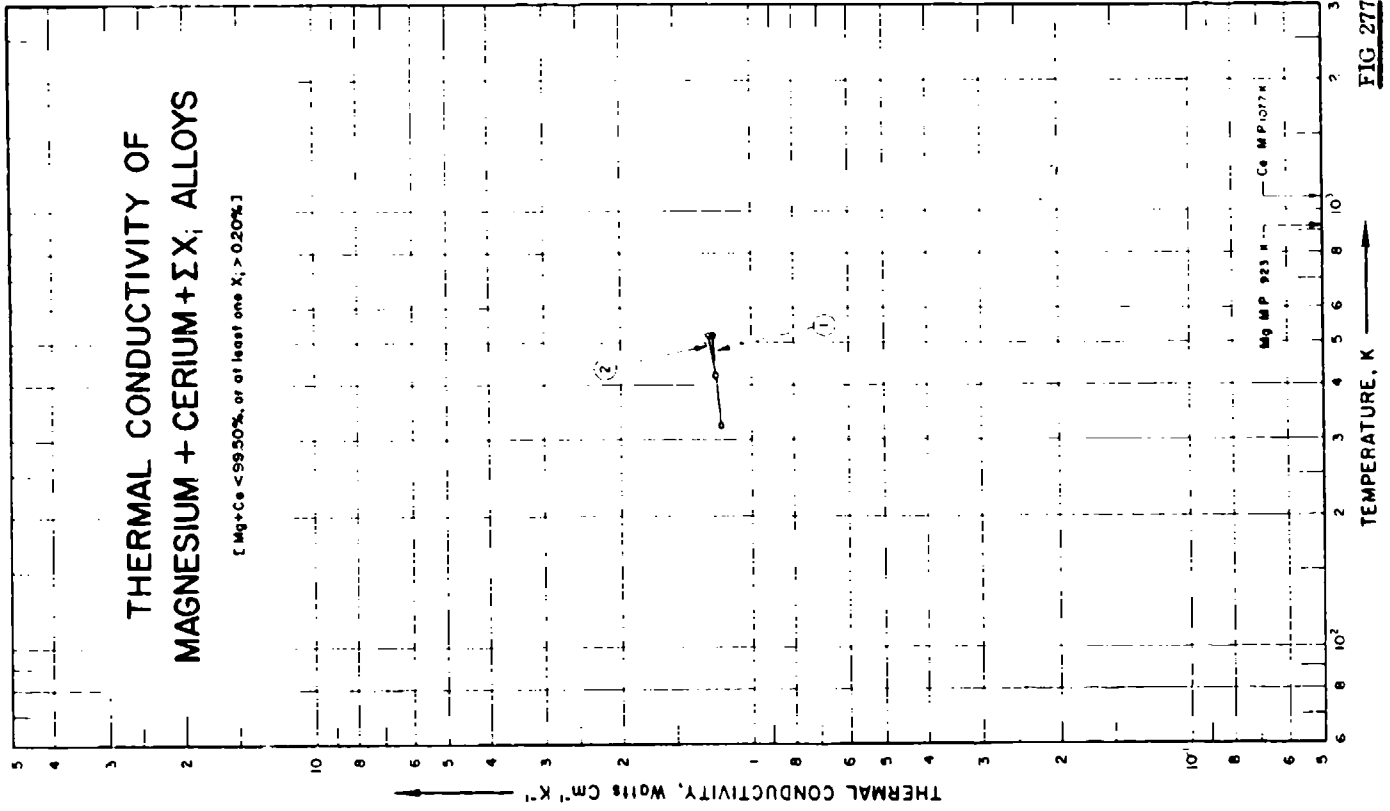


FIG 277



## SPECIFICATION TABLE NO. 277 THERMAL CONDUCTIVITY OF MAGNESIUM-CERUM-5% ALLOYS

(Mg-Ce = 50.50% or at least one Ni = 0.20%)

For Data Reported in Figure and Table No. 277.

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)			Composition (continued), Specifications and Remarks	
							Mg	Ce	Al		
1	397	L	1939	323-523	±3	W. 1662	94.96	3.17	1.7	Mg containing 0.07 impurities; specimen ~30 cm long and 1.4 cm in dia; density 1.78 g cm <sup>-3</sup> ; forged at elevated temperature; electrical resistivity 6.25, 6.8, 8.5 and 10.25 μhm cm at 20, 50, 150 and 250 C, respectively.	
2	397	L	1939	323-523	±3	N. P. L. P2	87.44	9.0	0.5	3.0	Similar to the above specimen except density 1.97 g cm <sup>-3</sup> and electrical resistivity 6.1, 6.65, 8.5 and 10.4 μhm cm at 20, 50, 150 and 250 C, respectively.

DATA TABLE NO. 277 THERMAL CONDUCTIVITY OF [MAGNESIUM + CERIUM + EX<sub>1</sub>] ALLOYS(Mg + Ce < 99.50% or at least one X<sub>1</sub> > 0.20%)[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup> K<sup>-1</sup>]

T k

CURVE 1

323.2	1.172
423.2	1.213
523.2	1.234

CURVE 2

323.2	1.172
423.2	1.213
523.2	1.255

SPECIFICATION TABLE NO. 278 THERMAL CONDUCTIVITY OF [MAGNESIUM + COBALT +  $\Sigma X_i$ ] ALLOYS Mg + Co +  $\Sigma X_i$   
 (Mg + Co < 99.50% or at least one  $X_i > 0.20\%$ )

Curve No.	Ref. Method Used	Year	Temp. 'ge. K	Reported Error. %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
1	397 L	1939	23-523	< 3.0	W-1702	93.73 Mg, 2.4 Co, 2.2 Ce, 1.6 M, and 0.07 impurities; forged at elevated temp.

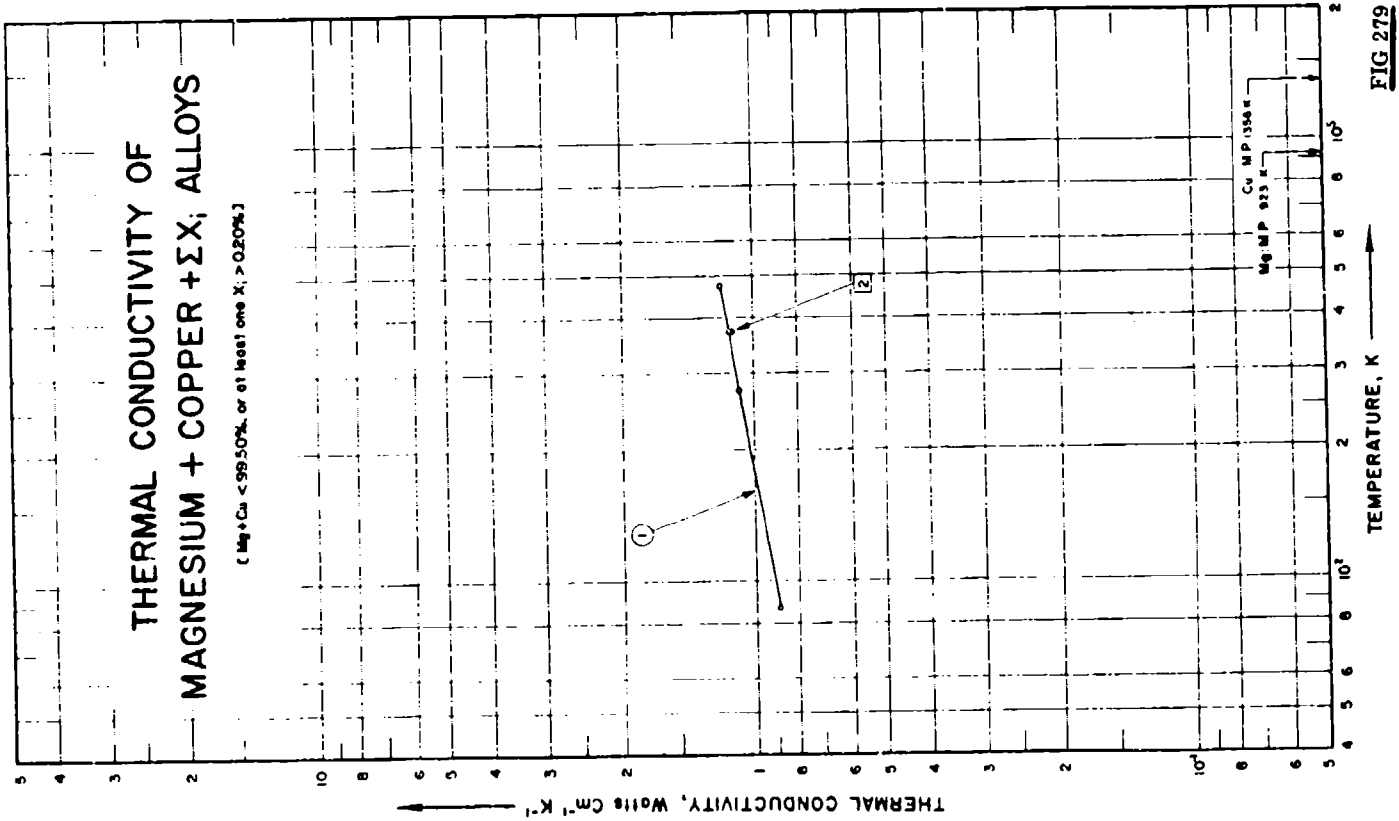
DATA TABLE NO. 278 THERMAL CONDUCTIVITY OF [MAGNESIUM + COBALT +  $\Sigma X_i$ ] ALLOYS Mg + Co +  $\Sigma X_i$

(Mg + Co < 99.50% or at least one  $X_i > 0.20\%$ )

[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup> K<sup>-1</sup>]

T	k
CURVE 1*	
323.2	1.32
423.2	1.32
523.2	1.32

\* No graphical presentation



SPECIFICATION TABLE NO. 279 THERMAL CONDUCTIVITY OF [MAGNESIUM + COPPER +  $\Sigma X_i$ ] ALLOYS(Mg + Cu < 99.50% or at least one  $X_i > 0.20\%$ )

[For Data Reported in Figure and Table No. 279]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)				Si	Composition (continued), Specifications and Remarks	
							Mg	Cu	Al	Fe			Mn
1	93	L	1931	87-476	3.0-4.0		77	20				3	As cast.
2	55	E	1927	373	1.0		94	4	2				Forged.

DATA TABLE NO. 279 THERMAL CONDUCTIVITY OF [MAGNESIUM + COPPER +  $\Sigma X_i$ ] ALLOYS(Mg + Cu < 99.50% or at least one  $X_i$  > 0.20%)[Temperature, T, K; Thermal Conductivity, k, Watt  $\text{cm}^{-1}\text{K}^{-1}$ ]

T	k
<u>CURVE 1</u>	
87	0.891
273	1.079
373	1.146
476	1.192
<u>CURVE 2</u>	
373	1.130

SPECIFICATION TABLE NO. 280 THERMAL CONDUCTIVITY OF [MAGNESIUM + NICKEL +  $\Sigma X_i$ ] ALLOYS Mg + Ni +  $\Sigma X_i$   
 (Mg + Ni < 99.50% or at least one  $X_i$  > 0.20%)

Curve No.	Ref. Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) Mg Ni Cc	Composition (continued), Specifications and Remarks
1	387	L	1939	323-523	±3.0	W-1648	91.93 5.36 2.65 0.06 impurities; forged at elevated temp; density 1.87 g cm <sup>-3</sup>

DATA TABLE NO. 280 THERMAL CONDUCTIVITY OF [MAGNESIUM + NICKEL +  $\Sigma X_i$ ] ALLOYS Mg + Ni +  $\Sigma X_i$   
 (Mg + Ni < 99.50% or at least one  $X_i$  > 0.20%)

[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup> K<sup>-1</sup>]

T	k
<u>CURVE 1*</u>	
323.2	1.30
423.2	1.30
523.2	1.30

\* No graphical presentation

FIGURE SHOWS ONLY 8 OF THE CURVES REPORTED IN TABLE

### THERMAL CONDUCTIVITY OF MANGANESE + IRON + ΣX<sub>i</sub> ALLOYS

(Mn + Fe < 99.50%, or at least one X<sub>i</sub> > 0.20%)

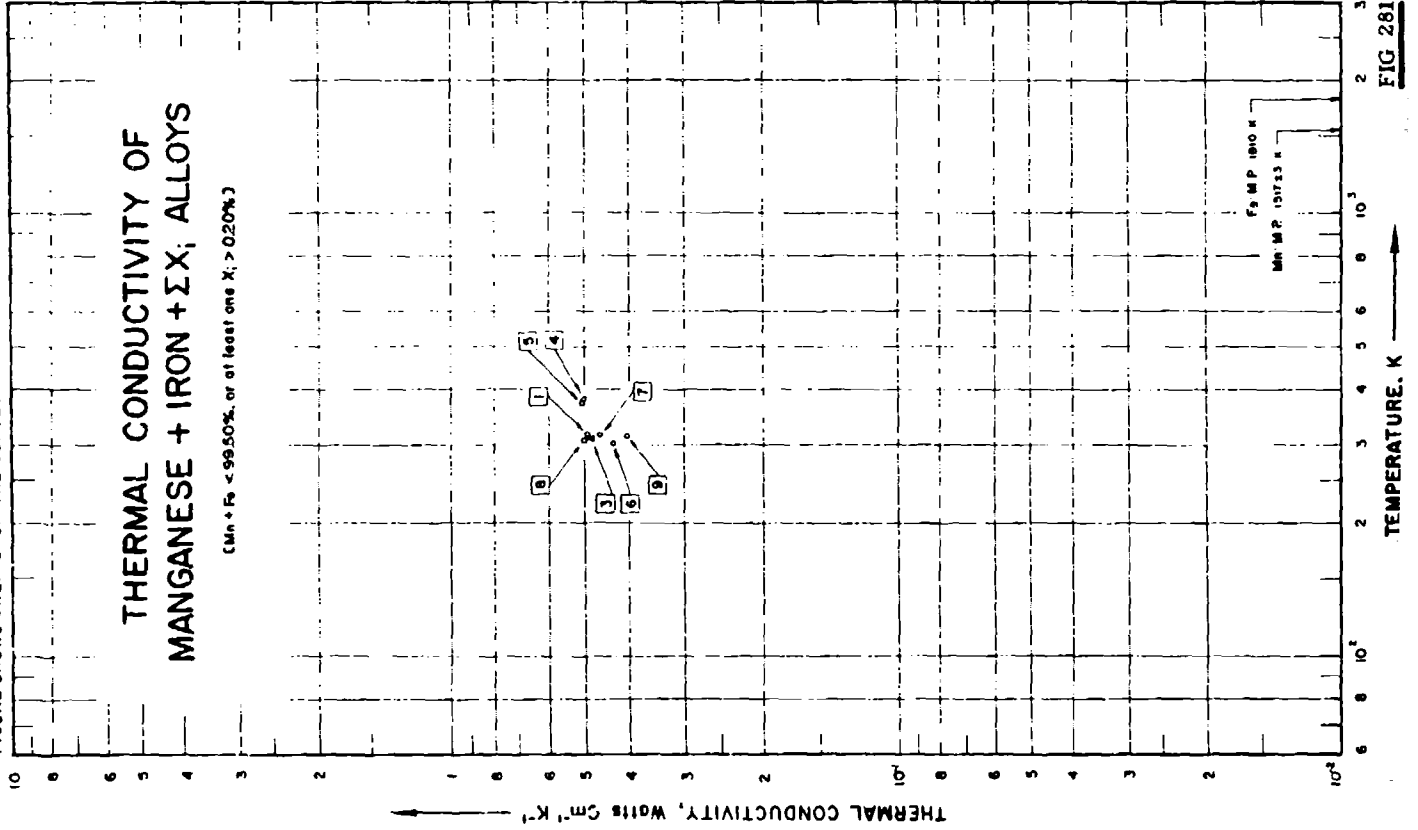


FIG 281



SPECIFICATIONS TABLE NO. 281 THERMAL CONDUCTIVITY OF [MANGANESE + IRON + Σ X<sub>i</sub>] ALLOY(Mn + Fe < 99.50% or at least one X<sub>i</sub> > 0.20%)

[For Data Reported in Figure and Table No. 281]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)				Composition (continued), Specifications and Remarks
							Mn	Fe	C	Si	
1	204	L	1937	317.7		Ferromanganese, 12	70.54	25.28	1.12	3.06	
2	204	L	1937	316.4		Ferromanganese, 13	79.0	17.89	2.36	1.75	
3	204	L	1937	314.6		Ferromanganese, 14	78.18	17.46	1.38	2.98	
4	204	L	1937	382.9		Ferromanganese, 15	78.0	15.5	4.7	1.8	From blast furnace.
5	204	L	1937	375.1		Ferromanganese, 16	73.0	26.48		0.52	From blast furnace.
6	204	L	1937	303.3		Ferromanganese, 17	77.33	13.73	6.92	2.02	Heat flow perpendicular to the thickness.
7	204	L	1937	318.8		Ferromanganese, 18	73.02	16.67	6.62	1.69	
8	204	L	1937	308.0		Ferromanganese, 19	77.3	15.36	6.65	1.69	
9	204	L	1937	315.1		Silicomanganese, 21	64.02	18.09	0.55	17.54	Highly crystalline specimen.

DATA TABLE NO. 291 THERMAL CONDUCTIVITY OF [MANGANESE + IRON +  $\Sigma X_i$ ] ALLOY  
 (Mn + Fe < 99.50% or at least one  $X_i > 0.20\%$ )

[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup>K<sup>-1</sup>]

<u>CURVE 1</u>	
317.7	0.493
<u>CURVE 2 *</u>	
318.4	0.494
<u>CURVE 3</u>	
314.6	0.486
<u>CURVE 4</u>	
382.9	0.500
<u>CURVE 5</u>	
375.1	0.504
<u>CURVE 6</u>	
303.3	0.434
<u>CURVE 7</u>	
318.8	0.464
<u>CURVE 8</u>	
308.0	0.500
<u>CURVE 9</u>	
315.1	0.404

\* Not shown on plot

SPECIFICATION TABLE NO. 282 THERMAL CONDUCTIVITY OF [MANGANESE + SILICON +  $\Sigma X_i$ ] ALLOYS Mn + Si +  $\Sigma X_i$   
 (Mn + Si < 99.50% or at least one  $X_i$  > 0.20%)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
1	204	L	1937	310.2		Silicomanganese, 20	63.7 Mn, 17.8 Si, and 1.05 C.

DATA TABLE NO. 282 THERMAL CONDUCTIVITY OF [MANGANESE + SILICON +  $\Sigma X_i$ ] ALLOYS Mn + Si +  $\Sigma X_i$   
 (Mn + Si < 99.50% or at least one  $X_i$  > 0.20%)

[Temperature, T, K; Thermal Conductivity, k, Watt  $\text{cm}^{-1} \text{K}^{-1}$ ]

T k

CURVE 1\*

310.2 0.469

\* No graphical presentation

SPECIFICATION TABLE NO. 283 THERMAL CONDUCTIVITY OF [MOLYBDENUM + IRON +  $\Sigma X_i$ ] ALLOYS Mo + Fe +  $\Sigma X_i$   
 (Mo + Fe  $\geq$  99.50% or at least one  $X_i > 0.20\%$ )

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Mo	Fe	Si	C	Composition (continued), Specifications and Remarks
1	204	L	1937	332.8		Ferromolybdenum, 34	81	14.59	2.41	2.0	2.41 Si.
2	204	L	1937	380.4		Ferromolybdenum, 35	67.77	31.88	0.26	0.09	0.021 Ti, 0.013 Cu, and 0.0003 Cr; after test the contents of Si and C changed to 0.063 and 0.008, respectively; hollow cylindrical specimen of 2 in. O.D. and 0.375 in. I.D.; supplied by Climax Molybdenum Co.; arc-melted unalloyed; density 10.22 g cm <sup>-3</sup> .
3	118	R	1956	1080-2795	5		0.25	0.073	0.007		

DATA TABLE NO. 283 THERMAL CONDUCTIVITY OF [MOLYBDENUM + IRON +  $\Sigma X_i$ ] ALLOYS Mo + Fe +  $\Sigma X_i$   
 (Mo + Fe  $\geq$  99.50% or at least one  $X_i > 0.20\%$ )

(Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup> K<sup>-1</sup>)

T k

CURVE 1<sup>c</sup>

332.8 0.585

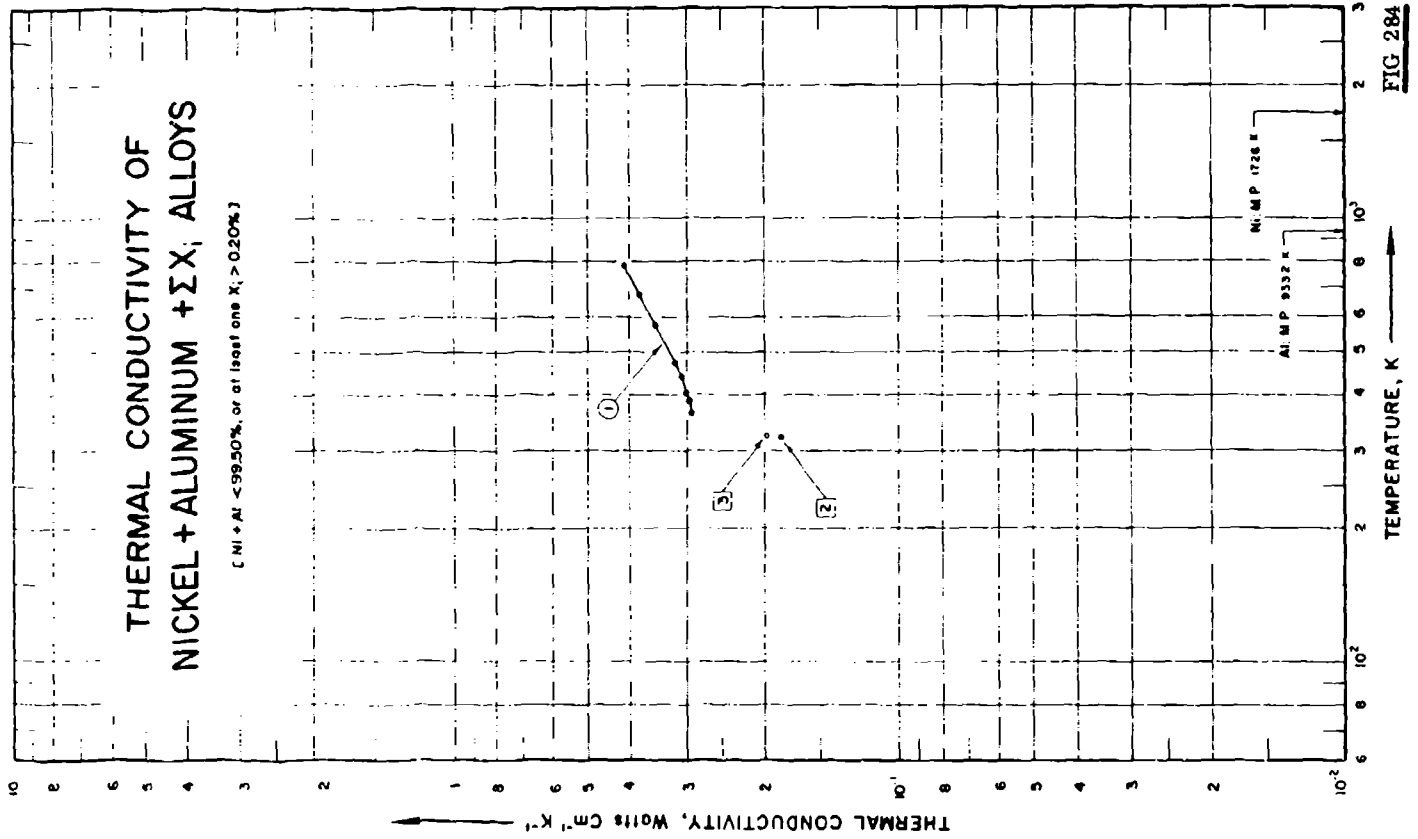
CURVE 2<sup>d</sup>

380.4 0.596

CURVE 3<sup>e</sup>

1080.3 1.007  
 1222.6 1.026  
 1359.6 0.929  
 1474.2 1.006  
 1612.1 0.954  
 2137.6 0.897  
 2319.7 0.879  
 2495.2 0.936  
 2651.5 0.835  
 2795.3 0.865

<sup>c</sup> No graphical presentation



SPECIFICATION TABLE NO. 284 THERMAL CONDUCTIVITY OF [NICKEL + ALUMINUM +  $\Sigma X_i$ ] ALLOYS  
 (Ni + Al = 99.50% or at least one  $X_i \geq 6.20\%$ )

(For Data Reported in Figure and Table No. 284)

Curve No.	ReL No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Ni	Al	Composition (weight percent)			Composition (continued), Specifications and Remarks		
									Mn	Si	Fe			
1	129	C	1933	363-773	3-5	Ni-Alumel	95.0	2.0	2.0	1.0		Specimen 2 cm in dia and 15 cm long; supplied by Hoskins manufacturing Co.; machined and hot-rolled to 3/4 in. in dia; lead used as comparative material.		
2	218		1956	323.2		Duranickel	93.0 Min.	4.4/ 4.75	0.5	1.0	0.6	0.25	0.25 - 1.0 Ti, 0.01 S; wrought, annealed; density 8.26 g cm <sup>-3</sup> ; electrical resistivity 290 ohms per cir mil ft.	
3	218		1956	323.2		Duranickel	93.0 Min.	4.4/ 4.75	0.5	1.0	0.6	0.3	0.25	0.25 - 1.0 Ti, 0.01 S; wrought, age-hardened; electrical resistivity 260 ohms per cir mil ft.

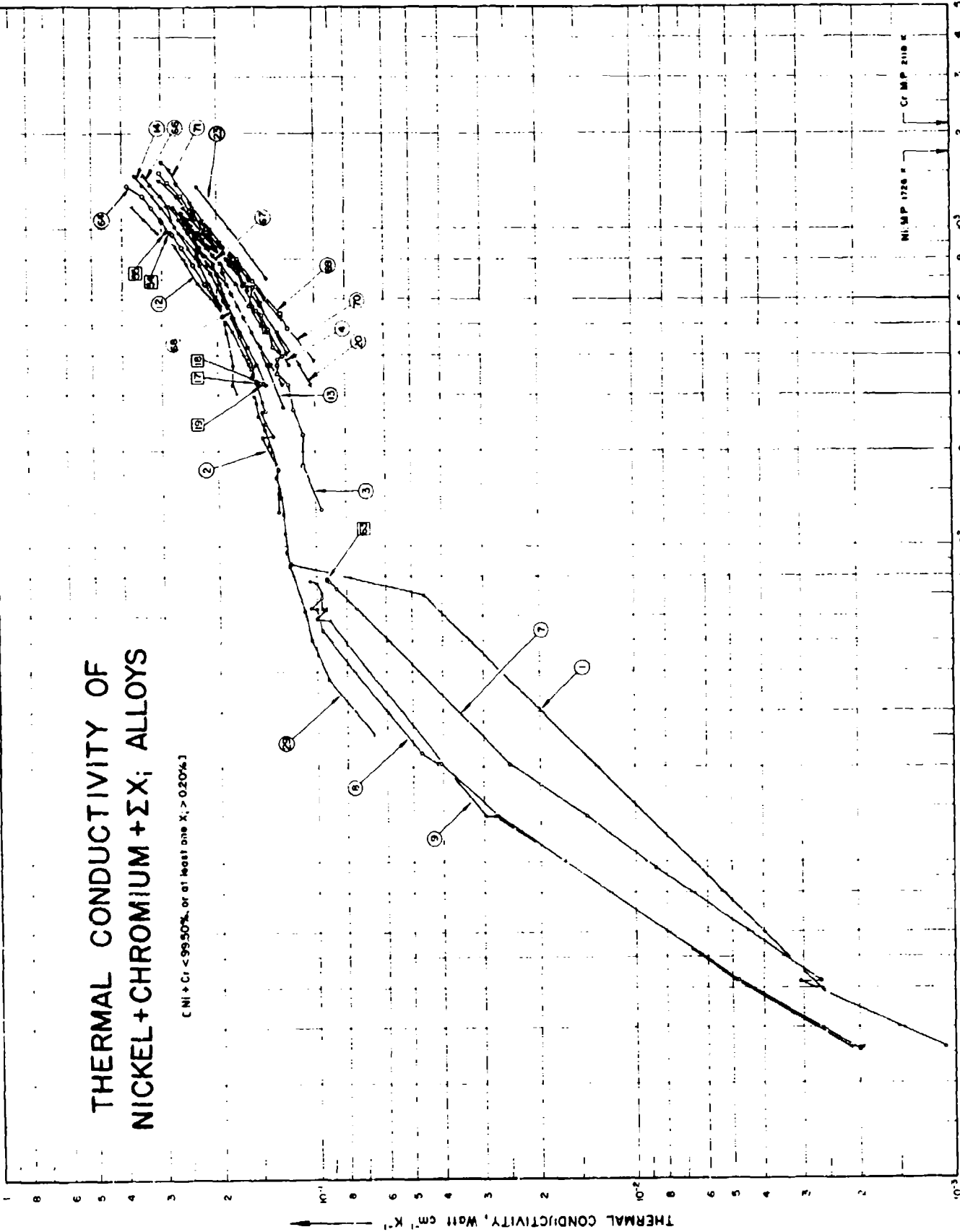
DATA TABLE NO. 284 THERMAL CONDUCTIVITY OF NICKEL + ALUMINUM +  $\Sigma X_i$  ALLOYS(N) + Al - 99.50% or at least one  $X_i$   $\geq 0.20\%$ [Temperature, T, K; Thermal Conductivity, k, Watt  $\text{cm}^{-1} \text{K}^{-1}$ ]

T	k
<u>CURVE 1</u>	
363.4	0.293
389.2	0.297
405.1	0.301
441.3	0.307
473.2	0.318
573.2	0.359
673.2	0.381
773.2	0.412
<u>CURVE 2</u>	
323.2	0.185
<u>CURVE 3</u>	
323.2	0.198

FIGURE SHOWS ONLY 26 OF THE CURVES REPORTED IN TABLE

# THERMAL CONDUCTIVITY OF NICKEL+CHROMIUM +ΣX; ALLOYS

[ Ni > Cr < 99.50%, or at least one X; > 0.20% ]



TEMPERATURE, K

THERMAL CONDUCTIVITY, Watt cm<sup>-1</sup> K<sup>-1</sup>



SPECIFICATION TABLE NO. 285 THERMAL CONDUCTIVITY OF [NICKEL + CHROMIUM + ΣX<sub>i</sub>] ALLOYS(Ni + Cr < 99.50% or at least one X<sub>i</sub> > 0.20%)

[ For Data Reported in Figure and Table No. 285 ]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)							Composition (continued), Specifications and Remarks	
							Ni	Cr	Al	C	Cu	Fe	Mn		Si
1	213	L	1950	3.9-88	2.0	Chroman	61.4	18.5				14.5	3.0	0.6	2.0 Mo.
2	91	C	1951	128-1189		Inconel	73.92	14.62		0.09	0.12	5.3	0.23	0.19	0.007 S; hot-rolled; annealed at 871-992 C for 3 hrs.
3	91	C	1951	131-1170		Inconel X	72.94	14.65	0.93	0.03	0.02	6.97	0.54	0.46	0.007 S; 1.01 Nb, and 2.44 Ti; hot-rolled; solution-treated at 1149 C for 3 hrs; aged at 843 C for 24 hrs and at 704 C for 20 hrs and air cooled.
4	214	L	1952	373-1173		Nimonic 80	72.8	21.4	0.38			3.08			2.34 Ti; heated to 1080 C for 8 hrs and aged at 700 C.
5	37	C	1951	377-845	4.0	Inconel X	73.4	14.6	0.7	0.05		6.9			2.3 Ti, 1.0 Nb.
6	37	C	1951	376-837	4.0	Nimonic 80	74.2	21.2	0.63	0.04					2.4 Ti.
7	155	L	1951	2.6-77		Inconel; i									Commercial Inconel; hard-drawn tubing.
8	155	L	1951	2.6-77		Inconel; 2									Commercial Inconel; annealed tubing.
9	155	L	1951	2.6-78		Inconel; 3									Commercial Inconel; hot-rolled rod.
10	43	L	1956	876-1348	5.0	Hastalloy C	56.07	15.83		0.07		4.94			4.41 W, 14.57 Mo.
11	215	L	1939	373,473	2.0	Inconel	80.09	12.97		0.67	0.18	6.31	0.24	0.15	Hot-rolled.
12	216		1959	323-1173		Inconel X	69.0 Min	14.0/ 17.0	0.4/ 1.0	0.08 Max	0.5 Max	5.0/ 9.0	1.0 Max	0.5	2.25-2.75 Ti, 0.7-1.2 Nb, 0.61(Max) S, 1.0 (Max) Co.
13	162	C	1936	273-1073		Nichrome	77.28	20.99		0.12		0.59	0.65	0.38	Forged and drawn.
14	163	L	1936	303-1473		German chromin; 20	82.23	12.98		0.09		3.05	0.88	0.75	Rolled.
15	163	L	1936	303-1373		German chromin; 21	75.15	19.93		0.04		1.94	1.81	1.13	Rolled.
16	217	C	1959	405-1044	4.0	Inconel	75.92	15.38				8.70			Obtained from commercial source in wrought form.
17	218		1956	323		Inconel	72.0 Min	14.0/ 17.0		0.15	0.5	6.0/ 10.0	1.0	0.5	0.015 S; annealed.
18	218		1956	323		Inconel X	70.0 Min	14.0/ 17.0	0.4/ 1.0	0.08	0.5	5.0/ 9.0	1.0	0.5	2.25-2.75 Ti, 0.01 S; age-hardened.

SPECIFICATION TABLE NO. 285 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)							Composition (continued), Specifications and Remarks	
							Ni	Cr	Al	C	Cu	Fe	Mn		Si
19	218		1956	323		Inconel	72.0 Min	14.0/17.0		0.15	0.5	6.0/10.0	1.0	0.5	0.015 S, as cast.
20	131	C	1953	323-1173	2.0	Nichrome V	77.94	19.87		0.036			0.06	1.44	Annealed at 950 C.
21	131	C	1953	323-1173	2.0	Inconel	78.13	13.94		6.33			0.32	0.33	0.30 Co; annealed at 1050 C.
22	91	C	1951	373-1073		Inconel									Commercial alloy produced by INCO.
23	195	C	1958	700, 1367	10.0	80 Ni-20 Cr	79.52	19.33		0.31		0.17	0.03	0.64	Trace P; density 8.35 g cm <sup>-3</sup> .
24	39	R	1958	536-1544	5.0	Hastelloy R-235	62.34	15.5	2.0	0.16		10.0	1.0		5.5 Mo, 2.5 Co, 1.0 Si; nominal composition; specimen composed of 15 disks) 0.625 in. I.D., 3.0 in. O.D. and 9 in. long.
25	187		1955	373, 1173		French nimonic 75	≈74	≈20		0.08/0.15	0.5 Max	≈2.4	1.0 Max		0.2-0.6 Ti.
26	187		1955	373, 1173		French nimonic 80/80A	≈71	≈2	0.51	0.1 Max	0.2 Max	≈5.0	1.0 Max		1.8-2.7 Ti.
27	187		1955	373, 1173		Nimonic 90	≈52	≈20	0.8/1.8	0.1 Max	0.2 Max	≈5.0	1.0 Max		15-20 Co, 1.8-2.7 Ti.
28	187		1955	373, 1173		Nimonic 95	≈56	≈19	1.6/1.85	0.12 Max	0.2 Max	1.0	0.3 Max		15-20 Co, 2.75-2.95 Ti.
29	219	L	1951	26-295	<2.5	Inconel	80	14							Provided by International Nickel Co.
30	533	L	1955	373-973	<2.0	Inconel									Composition not reported.
31	492	C	1960	323-1073		Nimonic 75	77.87	20.53		0.126	0.06	0.12	0.27	0.79	0.23 Ti.
32	492	C	1960	323-1173		Nimonic 80	73.66	21.0	1.20	0.04		0.50	0.60	0.50	2.5 Ti; nominal composition.
33	492	C	1960	323-1073		Nimonic 90	58.96	19.5	1.40	0.06	0.14	0.41	0.03	0.65	2.45 Ti, 16.5 Co.
34	492	C	1960	323-1073		Nimonic 95	57.71	19.1	1.99	0.10	0.06	0.36	0.06	0.65	2.91 Ti, 16.5 Co.
35	534	E	1958	373-973	3.0	Russian Kh80T	74.5	20.9	0.4	0.05			0.46	0.70	2.28 Ti.
36	534	E	1958	373-973	3.0	Russian Kh80T	74.5	20.9	0.4	0.05			0.46	0.70	2.28 Ti; tempered at 850 C for 100 hrs.
37	534	E	1958	373-973	3.0	Russian Kh80T	74.5	20.9	0.4	0.05			0.46	0.70	2.26 Ti; tempered at 830 C for 2000 hrs.
38	534	E	1958	973.2	3.0	El-607	80.95	15.4	0.55	0.02			0.50	0.42	0.49 Ti, 1.67 Nb; heat at 1100 C for 5 hrs and water-quenched.

SPECIFICATION TABLE NO. 285 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)							Composition (continued), Specifications and Remarks
							Ni	Cr	Al	C	Cu	Fe	Mn	
39	534	E	1958	973.2	3.0	EI-607	80.95	15.4	0.55	0.02		0.5	0.42	0.49 Ti, 1.67 Nb; the above specimen heated again at 1000 C for 2 hrs and air-cooled.
40	534	E	1958	973.2	3.0	EI-607	80.95	15.4	0.55	0.02		0.5	0.42	0.49 Ti, 1.67 Nb; the above specimen heated again at 900 C for 1 hr and again at 800 C for 2 hrs.
41	534	E	1958	973.2	3.0	EI-607	80.95	15.4	0.55	0.02		0.5	0.42	0.49 Ti, 1.67 Nb; the above specimen heated again at 750 C for 20 hrs.
42	534	E	1958	973.2	3.0	EI-607	80.95	15.4	0.55	0.02		0.5	0.42	0.49 Ti, 1.67 Nb; the above specimen heated again at 700 C for 48 hrs.
43	534	E	1958	373-973	3.0	EI-607	80.95	15.4	0.55	0.02		0.5	0.42	0.49 Ti, 1.67 Nb; the above specimen.
44	534	E	1958	373-973	3.0	EI-607	80.95	15.4	0.55	0.02		0.5	0.42	0.49 Ti, 1.67 Nb; the above specimen tempered at 700 C for 50 hrs.
45	534	E	1958	373-973	3.0	EI-607	80.95	15.4	0.55	0.02		0.5	0.42	0.49 Ti, 1.67 Nb; the above specimen tempered at 700 C for 200 hrs.
46	534	E	1958	373-973	3.0	EI-607	80.95	15.4	0.55	0.02		0.5	0.42	0.49 Ti, 1.67 Nb; the above specimen tempered at 700 C for 1000 hrs.
47	534	E	1958	373-973	3.0	EI-607	80.95	15.4	0.55	0.02		0.5	0.42	0.49 Ti, 1.67 Nb; the above specimen tempered at 700 C for 2000 hrs.
48	534	E	1958	973.2	3.0	EI-607	80.95	15.4	0.55	0.02		0.5	0.42	0.49 Ti, 1.67 Nb; tempered at 650 C for 1000 hrs.
49	534	E	1958	973.2	3.0	EI-607	80.95	15.4	0.55	0.02		0.5	0.42	0.49 Ti, 1.67 Nb; tempered at 650 C for 2000 hrs.
50	534	E	1958	973.2	3.0	EI-607	80.95	15.4	0.55	0.02		0.5	0.42	0.49 Ti, 1.67 Nb; tempered at 700 C for 1000 hrs.
51	534	E	1958	973.2	3.0	EI-607	80.95	15.4	0.55	0.02		0.5	0.42	0.49 Ti, 1.67 Nb; tempered at 700 C for 2000 hrs.

SPECIFICATION TABLE NO. 285 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)							Composition (continued), Specifications and Remarks
							Ni	Cr	Al	C	Cu	Fe	Mn	
52	534	E	1958	973.2	3.0	EI-607	80.95	15.4	0.55	0.02		0.5	0.42	0.49 Ti, 1.67 Nb; tempered at 750 C for 1000 hrs.
53	534	E	1958	973.2	3.0	EI-607	80.95	15.4	0.55	0.02		0.5	0.42	0.49 Ti, 1.67 Nb; tempered at 750 C for 2000 hrs.
54	534	E	1958	973.2	3.0	EI-607	80.95	15.4	0.55	0.02		0.5	0.42	0.49 Ti, 1.67 Nb; tempered at 800 C for 1000 hrs.
55	534	E	1958	973.2	3.0	EI-607	80.95	15.4	0.55	0.02		0.5	0.42	0.49 Ti, 1.67 Nb; tempered at 800 C for 2000 hrs.
56	467		1953	317-407	±5.0	Inconel	80.0	15.0			5.0			Nominal composition from Metals Handbook.
57	467		1953	317-417	±5.0	Inconel	80.0	15.0			5.0			Nominal composition from Metals Handbook.
58	535	L	1958	818-1149		Inconel	80.0	15.0			5.0			Nominal composition from Metals Handbook.
59	536	L	1955	366-1033	±7.0	Inconel	75.99	14.42		0.02	0.22	0.28	0.17	0.007 S; annealed at 2050 F followed by cooling in quiescent air; Rockwell superficial hardness (15 T scale) = 78.
60	536	L	1955	366-1033	±5.0	Inconel	76.45	14.96		0.07	0.15	0.26	0.19	0.007 S; annealed at 2050 F followed by cooling in quiescent air; Rockwell superficial hardness (15 T scale) = 80.
61	536	L	1955	366-1033	±5.0	Inconel	75.64	15.32		0.11	0.19	0.33	0.21	0.007 S; annealed at 2050 F followed by cooling in quiescent air; Rockwell superficial hardness (15 T scale) = 83.
62	490	F	1950	273-1173		Inconel	73.19	14.38	0.83	0.03	0.03	0.47	0.39	0.007 S; cylindrical bar 0.3175 cm in radius and 30 cm long.
63	537		1940	78.2		Nichrome	76.98	≈20						Others are Mn, C, Si and Fe; forged; commercial heat resistant alloy; measured in the boiling nitrogen bath.
64	163	L	1936	303-1373			62.85	16.95		0.12		1.00	0.51	0.007 S; annealed at 2050 F followed by cooling in quiescent air; Rockwell superficial hardness (15 T scale) = 80.
65	673	E	1932	309.6	±1.3		70	18						Others are Mn, C, Si and Fe; forged; commercial heat resistant alloy; measured in the boiling nitrogen bath.

SPECIFICATION TABLE NO. 285 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)							Composition (continued), Specifications and Remarks	
							Ni	Cr	Al	C	Cu	Fe	Mn		Si
66	596	L	1962	473-1473		Inconel 702	79.3	17.0	2.5	0.066	0.14	0.36	0.04	0.19	0.59 Ti and 0.08 Co; machined at NES from the solution annealed hot-rolled plate into the form of a right circular cylinder 2.539 cm in dia and 7.5 cm long with recesses at either end; the solution annealing for this alloy is to hold the material at 1080 C for one hr, followed by rapid air cooling.
67	686		1963	323-973	+2.0	OKh 20 N 60 B	59.64	20.4		0.06		17.7	1.59	0.25	0.58 Nb and 0.004 Si; quenched in water from 1050 C and then tempered in air at 720 C for one hr.
68	686		1963	323-1123	+2.0	EI-435	77.33	21.1		0.06	Trace	0.56	0.49	0.32	0.23 Ti, 0.006 S and 0.005 P; quenched in water from 1100 C.
69	614	R	1961	486-1501	<5	INCO 713C	71.53	11.0	6.5	0.20		5.0	1.0	1.0	1.0 Nb + Ta, 3.5 Mo, 0.25 Ti; specimen contained 5 one-in. dia disks.
70	614	R	1961	386-1427	<5	M252	57.15	18.65	1.17	0.12	9.75	<0.30	0.07	0.06	9.98 Mo, 2.74 Ti; specimen contained 5 one-in. dia disks.
71	614	R	1961	406-1617	<5	Rene 41	54.60	18.60	1.49	0.11	10.73	1.54	0.08	0.07	9.63 Mo, 3.14 Ti; specimen contained 5 one-in. dia disks.
72	973	L	1966	326-513	<6	Nimocast 713C	Bal	13.5	6.04	0.11					0.008 B, 4.65 Mo, 2.3 Nb, 0.95 Ti, and 0.10 Zr; specimen 1.27 cm in dia and 15 cm long; as cast; electrical resistivity 143.8. 144.1, 146.2, 149.3 and 151 $\mu$ ohm cm at 20, 50, 100, 200 and 250 C, respectively.

DATA TABLE NO. 285 THERMAL CONDUCTIVITY OF [NICKEL + CHROMIUM +  $\Sigma X_i$ ] ALLOYS(Ni + Cr < 99.50% or at least one  $X_i > 0.20\%$ )[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup> K<sup>-1</sup>]

T	k	T	k	T	k	T	k	T	k	T	k	T	k	T	k
<u>CURVE 1</u>		<u>CURVE 5 (concl.)</u>		<u>CURVE 9 (concl.)</u>		<u>CURVE 13 (concl.)</u>		<u>CURVE 16 (concl.)</u>		<u>CURVE 22*</u>		<u>CURVE 28*</u>			
3.90	0.0026	644.3	0.216	63.3	0.0935	973.2	0.247	637.8	0.192	373.2	0.163	373.2	0.125	373.2	0.289
70.00	0.046	699.8	0.225	63.3	0.0962	1073.2	0.264	724.4	0.205	473.2	0.172	1173.2	0.289		
88.00	0.120	755.4	0.234	70.6	0.0865			725.7	0.201	573.2	0.180				
		810.9	0.241	76.2	0.100	<u>CURVE 14</u>		804.1	0.230	673.2	0.188	<u>CURVE 29</u>			
		836.5	0.244	77.8	0.107			830.1	0.222	773.2	0.201				
<u>CURVE 2</u>		<u>CURVE 7</u>		<u>CURVE 10*</u>		<u>CURVE 17</u>		<u>CURVE 23</u>		<u>CURVE 29</u>		<u>CURVE 30*</u>			
128.3	0.130	2.6	0.00108	676.0	0.192	303.2	0.176	323.2	0.180	805.4	0.255	373.2	0.209	25.48	0.0657
175.2	0.138	4.2	0.00306	1005.0	0.254	323.2	0.180	373.2	0.188	824.9	0.247	373.2	0.218	37.94	0.0920
220.5	0.146	4.2	0.00265	1006.7	0.254	473.2	0.188	473.2	0.197	1043.6	0.272	1073.2	0.218	50.59	0.104
221.5	0.134	4.2	0.00265	1072.3	0.277	573.2	0.197	573.2	0.213					62.18	0.108
243.1	0.142	9.6	0.0087	1072.3	0.277	673.2	0.226	673.2	0.213					86.20	0.120
265.6	0.146	14.0	0.0143	1072.3	0.277	773.2	0.247	773.2	0.226					95.18	0.122
265.8	0.142	14.0	0.0143	1072.3	0.277	873.2	0.264	873.2	0.247					109.88	0.124
287.6	0.146	20.4	0.0248	1150.4	0.284	973.2	0.280	973.2	0.264	699.9	0.140			126.97	0.126
327.3	0.151	63.3	0.0744	1229.0	0.292	1073.2	0.301	1073.2	0.280	1366.5	0.231			142.94	0.127
38.9	0.155	73.4	0.0866	1348.3	0.299	1173.2	0.322	1173.2	0.301					153.44	0.129
70.1	0.155	77.0	0.0910			1273.2	0.322	1273.2	0.322					164.92	0.131
390.8	0.159			<u>CURVE 11*</u>		1373.2	0.343			336.2	0.111			174.49	0.130
459.1	0.172			<u>CURVE 12</u>		1473.2	0.364			378.8	0.116			181.06	0.132
593.7	0.192	2.55	0.00200	373.2	0.155					452.1	0.127			199.87	0.135
599.5	0.192	4.25	0.00483	473.2	0.168					598.3	0.152			214.13	0.137
725.6	0.209	14.0	0.0270			<u>CURVE 15*</u>				685.1	0.166			234.27	0.142
822.2	0.230	20.5	0.0407	303.2	0.151	303.2	0.151			846.8	0.185			256.92	0.149
841.1	0.230	20.5	0.0415	373.2	0.163	373.2	0.163			1073.5	0.214			281.59	0.151
1015.7	0.272	20.5	0.0470	473.2	0.176	473.2	0.176			1237.7	0.251			295.38	0.154
1189.1	0.289	22.1	0.0470	573.2	0.188	573.2	0.188			1416.8	0.272				
		54.0	0.0963	373.2	0.16										
		559.1	0.197	473.2	0.18										
		654.2	0.218	573.2	0.20										
		677.4	0.216	573.2	0.23										
		692.4	0.222	773.2	0.25										
		733.7	0.233	873.2	0.27										
		751.6	0.229	973.2	0.31										
		772.2	0.239	1073.2	0.34										
		786.1	0.240	1173.2	0.272										
		837.8	0.256	1273.2	0.297										
		844.9	0.249	1373.2	0.331										
		<u>CURVE 8*</u>		<u>CURVE 13</u>		<u>CURVE 16*</u>		<u>CURVE 21*</u>		<u>CURVE 25*</u>		<u>CURVE 26*</u>		<u>CURVE 31*</u>	
131.1	0.0962	2.59	0.00196	273.2	0.126	404.8	0.155	323.2	0.111	373.2	0.121	373.2	0.126	373.2	0.132
180.1	0.109	2.59	0.00213	473.2	0.140	440.3	0.155	473.2	0.119	473.2	0.136	473.2	0.157	473.2	0.159
226.0	0.109	4.25	0.00489	373.2	0.126	479.5	0.163	573.2	0.136	573.2	0.152	573.2	0.175	573.2	0.184
269.9	0.117	10.1	0.0168	473.2	0.157	516.2	0.176	673.2	0.169	673.2	0.185	673.2	0.191	673.2	0.191
324.6	0.121	14.0	0.0297	573.2	0.174	561.3	0.184	773.2	0.202	773.2	0.202	773.2	0.210	773.2	0.210
349.5	0.130	14.0	0.0297	673.2	0.188	566.8	0.184	873.2	0.218	873.2	0.218	873.2	0.226	873.2	0.226
373.5	0.130	58.7	0.0997	773.2	0.213	617.5	0.188	973.2	0.235	973.2	0.235	973.2	0.240	973.2	0.240
390.4	0.130	63.0	0.0935	873.2	0.230			1073.2	0.251	1073.2	0.251	1073.2	0.251	1073.2	0.251
397.7	0.126														
426.9	0.134														
485.4	0.138														
551.8	0.151														
573.7	0.146														
573.8	0.159														
573.6	0.155														
664.4	0.167														

\* Not shown on plot



DATA TABLE NO. 285 (continued)

T k

CURVE 70 (cont.)

765.9	0.173
872.6	0.194*
992.1	0.217
1106.5	0.241
1255.9	0.267
1427.1	0.305

CURVE 71

405.9	0.122
585.4	0.147
753.2	0.168
907.6	0.196
994.3	0.210
1060.4	0.220
1212.1	0.241
1392.1	0.267
1554.8	0.280
1617.1	0.300

CURVE 72\*

326.2	0.108
330.2	0.108
330.2	0.107
331.2	0.107
340.2	0.108
342.2	0.110
345.2	0.110
355.2	0.108
383.2	0.112
388.2	0.114
398.2	0.114
403.2	0.116
408.2	0.115
436.2	0.120
439.2	0.120
+64.2	0.124
513.2	0.130

\* Not shown on plot



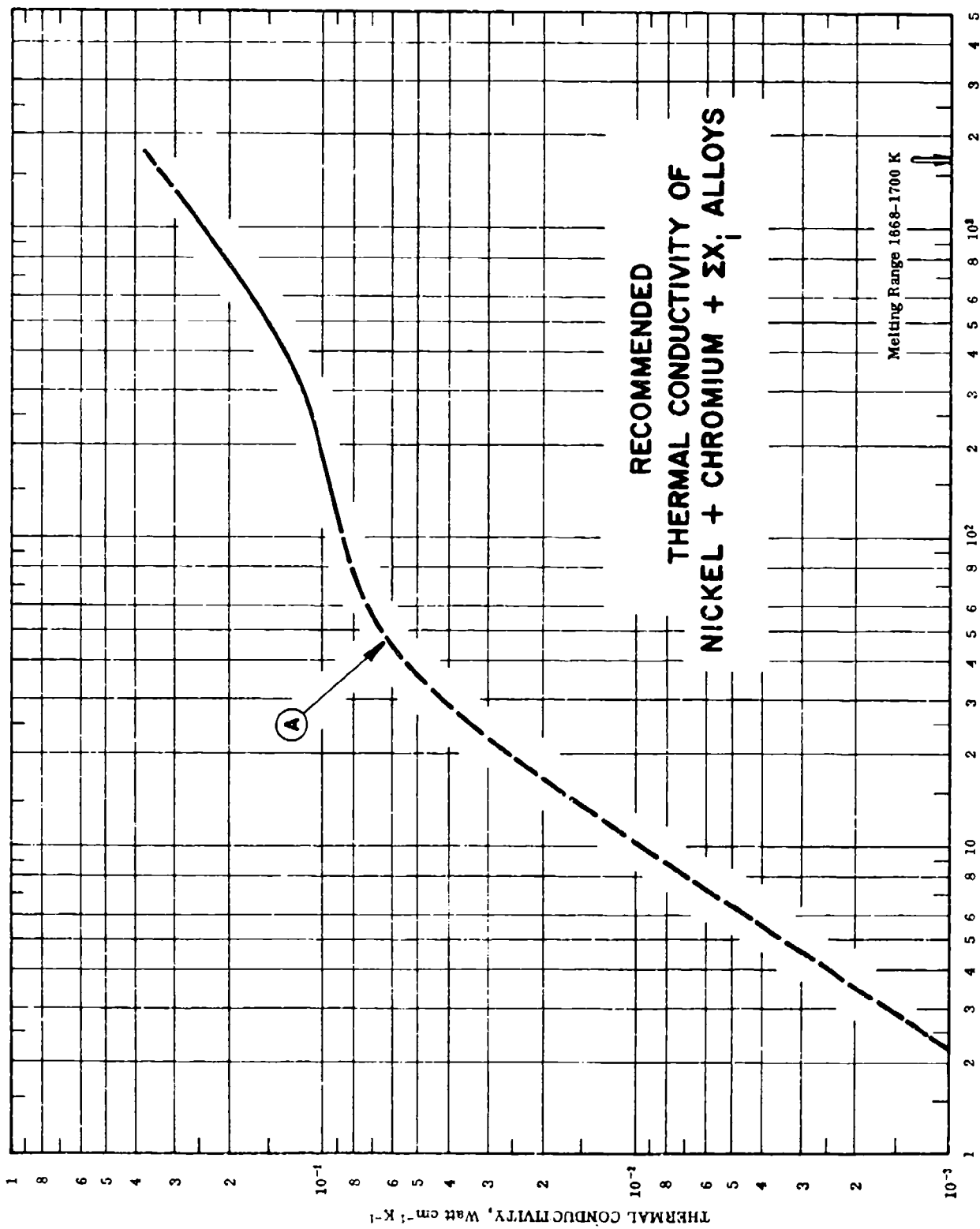


FIG 285R

SPECIFICATION TABLE NO. 285R RECOMMENDED THERMAL CONDUCTIVITY OF [NICKEL + CHROMIUM + EX<sub>1</sub>] ALLOYS

[For Data Reported in Figure and Data Table No. 285R]

Curve No.	Name and Designation	Nominal Composition (weight percent) and Remarks	Estimated Error
A	Inconel X-750 (previously designated as Inconel X)	73.0 Ni, 15 Cr, 6.75 Fe, 2.50 Ti, 0.85 Nb, 0.80 Al, 0.70 Mn, 0.30 Si, 0.05 Cu, 0.04 C and 0.007 S; fully heat treated.	+ 10% below 100 K, ± 5% from 200 to 1000 K, and ± 10% above 1400 K.

DATA TABLE NO. 285R RECOMMENDED THERMAL CONDUCTIVITY OF [NICKEL + CHROMIUM + EX<sub>1</sub>] ALLOYS[Temperature, T<sub>1</sub> in K and T<sub>2</sub> in °F; Thermal Conductivity, k<sub>1</sub> in Watt cm<sup>-1</sup>K<sup>-1</sup> and k<sub>2</sub> in Btu hr<sup>-1</sup>ft<sup>-1</sup>F<sup>-1</sup>]

T <sub>1</sub>	k <sub>1</sub>	k <sub>2</sub>	T <sub>2</sub>	T <sub>1</sub>	k <sub>1</sub>	k <sub>2</sub>	T <sub>2</sub>	T <sub>1</sub>	k <sub>1</sub>	k <sub>2</sub>	T <sub>2</sub>	CURVE A (cont.)	
												CURVE A	CURVE A (cont.)
0	0	0	-459.7	300	0.117	6.75	80.3	1500	0.325	(19.0)	2240		
1	(0.00924)†	(0.0139)	-437.9	350	0.126	7.28	170.3	1600	0.340	(20.0)	2420		
3	(0.0145)	(0.0218)	-434.3	400	0.135	7.80	260.3	1565	0.354	(20.7)	2537		
5	(0.0203)	(0.0311)	-450.7	450	0.143	8.26	356.3						
10	(0.0296)	(0.0455)	-441.7	500	0.152	8.78	440.3						
25	(0.041)	(0.061)	-414.7	600	0.170	9.82	620.3						
50	(0.066)	(0.096)	-369.7	700	0.188	10.9	800.3						
75	(0.090)	(0.132)	-324.7	800	0.205	11.8	980.3						
100	(0.097)	(0.142)	-279.7	900	0.223	12.9	1160						
150	0.096	0.140	-189.7	1000	0.240	13.9	1340						
200	0.105	0.150	-99.7	1100	0.258	14.9	1520						
250	0.110	0.155	-9.7	1200	0.276	15.9	1700						
273.2	0.113	0.158	32.0	1300	(0.293)	(16.9)	1880						
				1400	(0.311)	(18.0)	2060						

† Values in parentheses are extrapolated.

# THERMAL CONDUCTIVITY OF NICKEL + COBALT + $\Sigma X$ ; ALLOYS

[Ni + Co  $\geq$  99.50%; impurity  $\leq$  0.20% each]

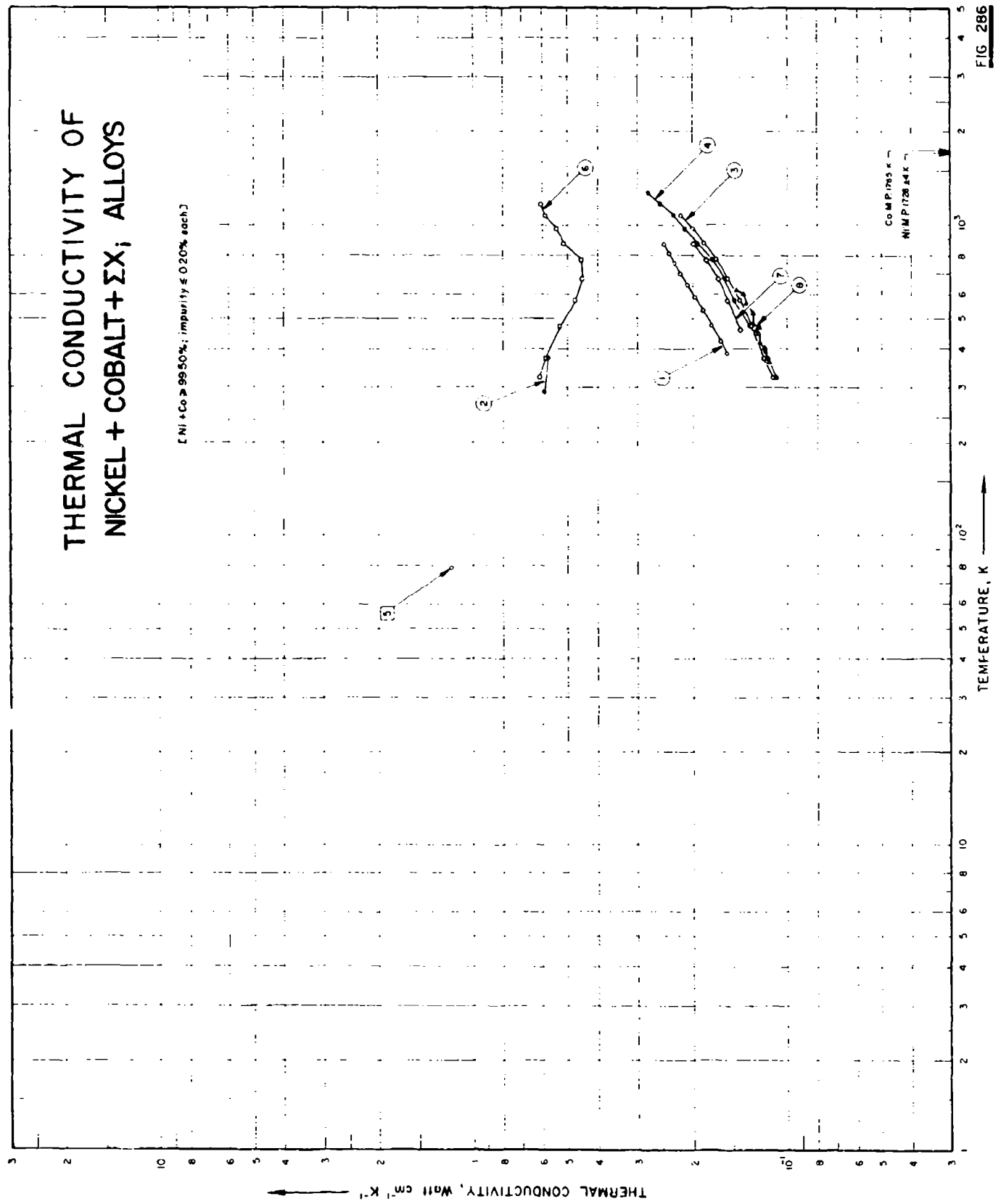


FIG. 286

SPECIFICATION TABLE NO. 286 THERMAL CONDUCTIVITY OF [NICKEL + COBALT + EX<sub>1</sub>] ALLOYS

(Ni + Co + EX<sub>1</sub> = 99.50% or at least one X<sub>1</sub> > 0.20%)

(For Data Reported in Figure and Table No. 286 )

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Ni	Composition (weight percent)							T <sub>1</sub>	Composition (continued), Specifications and Remarks	
								Co	Al	Cr	Fe	Mn	Mo	Ti			
1	37	C	1951	385-865	4.0	Refalloy 26	37.0	29.9	0.3	18.0	18.67					3.0	0.03 C.
2	77	F	1900	291-373			97.0	1.4			0.4	1.0					0.1 Si, 0.1 Cu.
3	492	C	1960	323-1073		Nimonic 100	Bal	29.0	5.22	11.1	0.18	0.03				1.07	0.28 Si, 0.24 C, 0.04 Cu.
4	492	C	1960	323-1273		Nimonic 105	Bal	18.0/ 22.0	4.0/ 6.0	14.0/ 16.0	3.0	1.0	Max	Max		0.5/ 2.5	1.0 (Max) Si, 0.3 (Max) C, 0.5 (Max) Cu.
5	433	L	1940	78.2			99.4	0.53	0.02		0.05						Electrolytic.
6	131	C	1953	344-1173	+2.0	AN <sub>1</sub>	98.19	0.746			0.26					0.705	0.053 Cu, 0.036 P; annealed at 900 C.
7	616		1947	473-873		Haynes Stellite No. 27	Bal	30 Min		23/ 29	2	Max				5/ 7	0.35-0.50 C; density 8.21 g cm <sup>-3</sup> .
8	973	L	1966	363-616	±6	Nimonic 115	Bal	14.5	4.82	14.0	0.27	0.05				4.07	0.032 B, 0.17 C, 0.03 Cu, 0.16 Si, 0.04 Zr; specimen 1.27 cm in dia and 5 cm long; fully heat treated; electrical resistivity 141.2, 142.9, 146.1, 149.2 and 151 μ ohm cm at 50, 100, 200, 300 and 350 C, respectively.

DATA TABLE NO. 286 THERMAL CONDUCTIVITY OF [NICKEL + COBALT + EX<sub>1</sub>] ALLOYS  
 (Ni + Co + EX<sub>1</sub> < 99.50% or at least one X<sub>i</sub> > 0.20%)

[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup>K<sup>-1</sup>]

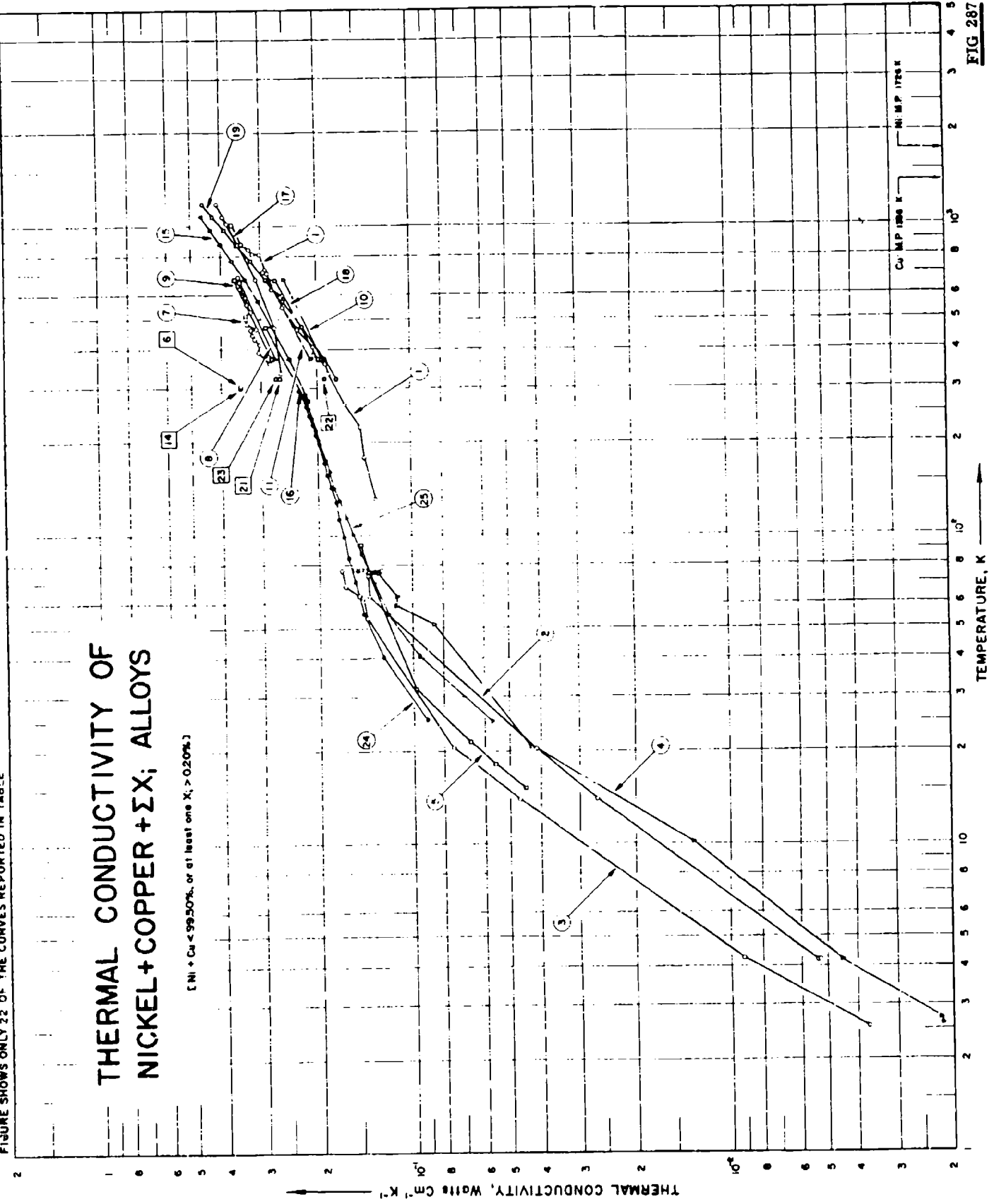
T	k	T	k
<u>CURVE 1</u>			
385.4	0.156	<u>CURVE 5</u>	
422.0	0.164	78.2	1.180
477.6	0.175	<u>CURVE 6</u>	
533.2	0.186	323.2	0.610
588.7	0.197	373.2	0.582
644.3	0.208	473.2	0.525
699.8	0.218	573.2	0.470
755.4	0.228	673.2	0.446
810.9	0.237	773.2	0.447
865.4	0.246	873.2	0.509
<u>CURVE 2</u>			
291.2	0.594	973.2	0.540
373.2	0.579	1073.2	0.587
		1173.2	0.604
<u>CURVE 3</u>			
323.2	0.112	473.2	0.141
373.2	0.114	573.2	0.155
473.2	0.130	673.2	0.166
573.2	0.142	773.2	0.191
673.2	0.155	873.2	0.198
773.2	0.169	<u>CURVE 8</u>	
873.2	0.184	363.2	0.114
973.2	0.200	402.2	0.117
1073.2	0.218	417.2	0.122
<u>CURVE 4</u>			
323.2	0.108	448.2	0.123
373.2	0.116	459.2	0.125
473.2	0.131	465.2	0.122
573.2	0.147	480.2	0.128
673.2	0.159	523.2	0.138
773.2	0.174	559.2	0.135
873.2	0.193	598.2	0.138
973.2	0.212	593.2	0.139*
1073.2	0.230	616.2	0.145
1173.2	0.253	616.2	0.147*
1273.2	0.276		

\* Not shown on plot

FIGURE SHOWS ONLY 22 OF THE CURVES REPORTED IN TABLE

# THERMAL CONDUCTIVITY OF NICKEL+COPPER +ΣX; ALLOYS

[ Ni + Cu < 99.50%, or at least one X; > 0.20% ]



SPECIFICATION TABLE NO. 287 THERMAL CONDUCTIVITY OF (NICKEL + COPPER + EX<sub>1</sub>) ALLOYS  
(Ni + Cu < 99.50% or at least one X<sub>1</sub> > 0.20%)

(For Data Reported in Figure and Table No. 287)

Curve No.	Ref. Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight per cent)						Composition (continued) Specifications and Remarks	
						Ni	Cu	Al	C	Fe	Mn		Si
1	91	C	1951	132-1186	"K" Monel	65.51	29.23	3.02	0.13	0.86	0.60	0.09	0.005 S; hot-rolled, annealed at 1650 F for 1 hr and water-quenched.
2	155	L	1951	4.2- 77	Monel; 1					2.0			Commercial Monel; hard-drawn tubing.
3	155	L	1951	2.6- 77	Monel; 2					2.68			Commercial Monel; annealed tubing.
4	155	L	1951	2.6- 77	Monel; 3					2.68			Commercial Monel; hard-drawn rod.
5	104	L	1951	15- 93	Monel	67.0	30.2						And other alloying elements; as forged.
6	186	P	1928	305	Monel	70.0	28.0						As forged.
7	50	L	1925	361- 688	Monel	67.05	29.07						As cast.
8	215	L	1939	373, 673	Monel	67.05	29.07						Hot-rolled.
9	215	L	1939	373, 673	Monel	67.34	29.46		0.21	1.79	1.11	0.07	As cast.
10	215	L	1938	373, 673	"K" Monel	66.73	29.76	2.50	0.20	0.35	0.21	0.25	As cast.
11	215	L	1939	373, 673	"SI" Monel	65.39	28.71		0.41	1.86	1.54	2.09	As cast.
12	215	L	1938	373, 673	"SI" Monel	64.04	28.32		0.38	2.15	1.74	3.37	As cast.
13	215	L	1938	373, 673	"NI" Bronze	51.37	37.86						9.91 Sn, 0.86 Zn; as cast.
14	178	E	1918	303	17a	92.3	4.2		0.29	2.45	0.35	0.15	0.24 Co, 0.017 S; annealed at 900 C.
15	162	C	1936	373-1073	Monel	67.10	29.18	0.04	0.16	1.72	0.98	0.01	0.33 Co, 0.014 S, 0.13 Mg, 0.024 P; hot-rolled, black surface.
16	162	C	1936	373, 473	Corronil	66.41	28.94	0.16	0.1	0.6	1.17	0.4	Forged and drawn.
17	220		1960	337-1073	R-Monel	67.0	30.0		0.15	1.40	1.0	0.01	0.035 S; (Nominal composition).
18	221		1954	373, 673	S-Monel	62.0/ 68.0	Bal		0.25 <3.5		0.5/ 1.3	3.5/ 3.0	(Nominal composition.)
19	131	C	1953	323-1173	Monel	66.2	30.0		0.4	1.88	0.919	3.135	0.407 Co, 0.032 Mg; annealed at 950 C.
20	222		1954	373, 673	Cast Monel	62.0/ 68.0	Bal		0.35 Max	>2.5	0.5/ 1.5	1.0/ 2.24	
21	218		1956	323	Monel	63.0/ 70.0	Bal		0.3	2.5	2.0	0.5	0.01 S; hot-rolled.
22	218		1956	323	"K" Monel	63.0/ 70.0	Bal	2.0/ 4.0	0.25	2.0	1.5	1.0	0.25 - 1.0 Ti, 0.01 S; wrought; age-hardened.
23	218		1956	323	"H" Monel	61.0/ 68.0	Bal		0.3	2.5	0.5/ 1.5	2.7/ 3.7	0.05 S; as cast.
24	219	L	1951	26- 295	Monel	67.0	30.0		0.15	1.4	1.0	0.1	0.01 S; hot-rolled.
25	219	L	1951	26- 299	Monel	67.0	30.0		0.15	1.4	1.0	0.1	0.01 S; cold-rolled; provided by International Nickel Co.

SPECIFICATION TABLE NO. 287 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)				Si	Composition (continued), Specifications and Remarks
							Ni	Cu	Al	C		
26	433	L	1940	78		No. 0	94.77	4.36	0.02	0.08	0.26	0.51 Co, 0.01 Sb, trace S, trace Pb; calculated composition.
27	433	L	1940	78		No. 1	90.43	8.85	0.02	0.09	0.13	0.48 Co, 0.001 Sb, trace S, trace Pb; calculated composition.
28	433	L	1940	78		No. 2	85.62	13.71	0.017	0.094	0.10	0.46 Co, 0.002 Sb, 0.001 S, trace Pb; calculated composition.
29	433	L	1940	78		No. 3	77.73	21.69	0.015	0.091	0.05	0.414 Co, 0.003 Sb, 0.002 S, trace Pb; calculated composition.
30	433	L	1940	78		No. 4	69.14	30.35	0.014	0.068	0.05	0.37 Co, 0.005 Sb, 0.002 S, trace Pb; calculated composition.
31	433	L	1940	78		No. 5	58.98	40.53	0.012	0.104	0.04	0.314 Co, 0.006 Sb, 0.003 S, trace Pb; calculated composition.

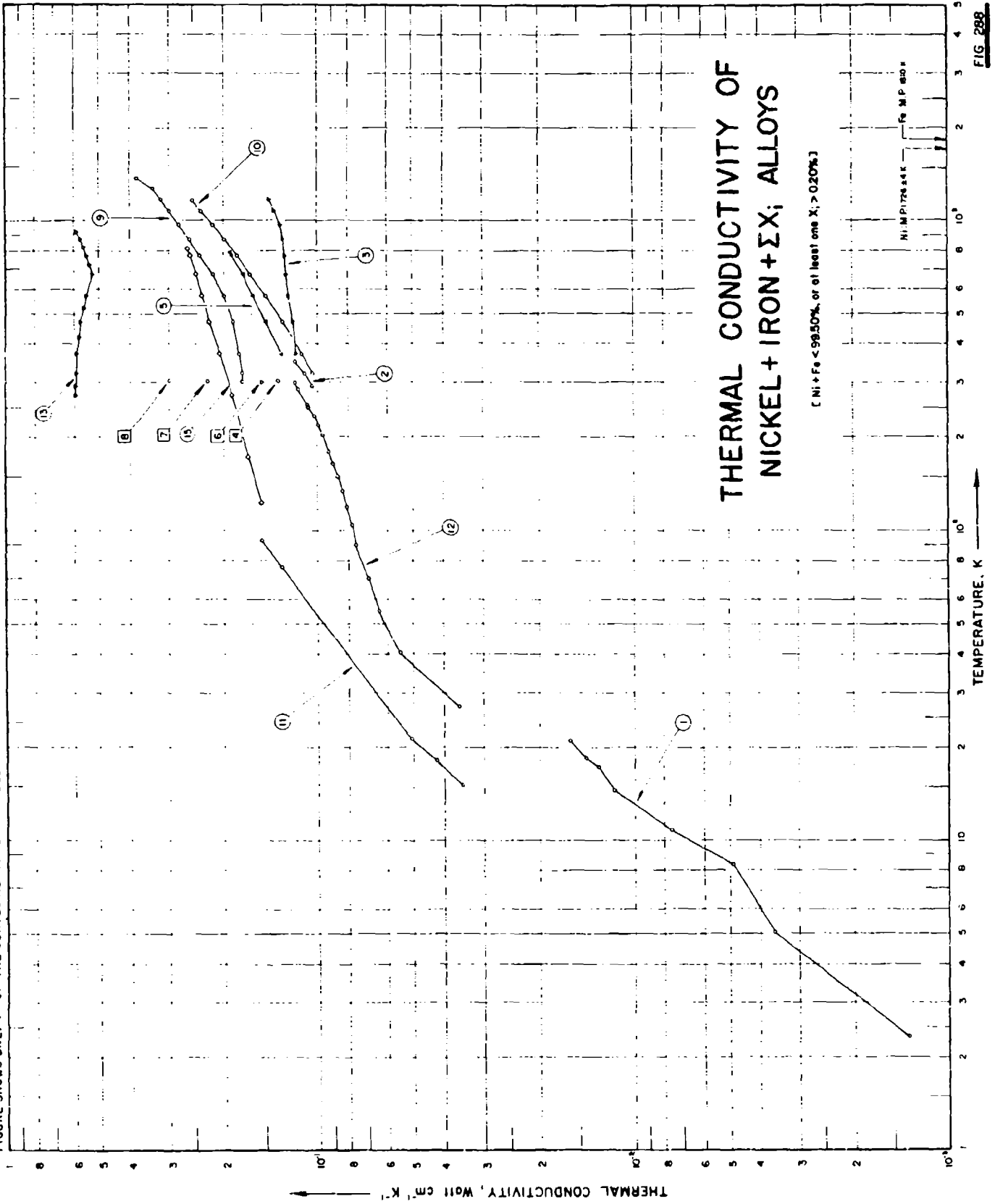


DATA TABLE NO. 287 THERMAL CONDUCTIVITY OF NICKEL + COPPER + EX<sub>1</sub> ALLOYS(Ni + Cu < 99.50% or at least one X<sub>i</sub> > 0.20%)(Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup> K<sup>-1</sup>)

T	k	T	k	T	k	T	k	T	k	T	k				
<u>CURVE 1</u>															
131.60	0.130	20.50	0.0745	484.70	0.331	273.20	0.213	373.20	0.268	101.15	0.154				
179.30	0.142	54.60	0.140	500.20	0.335	373.20	0.243	673.20	0.352	115.14	0.161				
224.30	0.146	63.00	0.144	514.20	0.335	473.20	0.276			130.22	0.170				
266.50	0.163	63.70	0.147	531.20	0.322	573.20	0.305	<u>CURVE 21</u>			143.20	0.177			
334.90	0.184	58.40	0.164	560.70	0.331	673.20	0.325	323.20	0.260	160.39	0.183				
360.40	0.188	77.00	0.168	591.20	0.335	773.20	0.363	<u>CURVE 22</u>			175.46	0.188			
384.70	0.205	<u>CURVE 4</u>										204.02	0.197		
407.90	0.205	2.60	0.00216	660.70	0.347	873.20	0.398	323.20	0.188	230.21	0.201				
465.30	0.230	2.70	0.00218	888.20	0.351	1073.20	0.429	323.20	0.268	239.98	0.208				
468.10	0.230	4.20	0.0045	<u>CURVE 16</u>								258.13	0.213		
548.10	0.255	10.10	0.0131	373.00	0.268	273.20	0.213	<u>CURVE 23</u>			288.14	0.218			
592.30	0.259	20.40	0.0406	673.00	0.351	473.20	0.289	323.20	0.268	<u>CURVE 26*</u>					
597.20	0.255	20.60	0.0428	<u>CURVE 9</u>								78.2	0.464		
706.20	0.285	51.60	0.0855	<u>CURVE 17</u>								<u>CURVE 27*</u>			
711.30	0.283	59.00	0.112	373.00	0.276	336.70	0.259	25.43	0.0908	<u>CURVE 28*</u>					
724.30	0.289	63.30	0.111	673.00	0.364	473.20	0.272	40.44	0.124	78.2	0.333				
811.30	0.301	75.50	0.128	<u>CURVE 10</u>								55.54	0.143		
813.60	0.318	77.00	0.148	373.00	0.193	673.20	0.310	70.52	0.152	<u>CURVE 29*</u>					
841.90	0.336	<u>CURVE 5</u>										84.36	0.160		
876.60	0.343	15.17	0.0444	673.00	0.251	<u>CURVE 18</u>								98.90	0.165
1011.00	0.368	18.16	0.0553	373.20	0.197	113.26	0.171	127.12	0.175	<u>CURVE 29*</u>					
1024.10	0.381	21.47	0.0661	673.20	0.268	673.20	0.197	142.56	0.181	78.2	0.213				
1185.70	0.410	31.80	0.0977	<u>CURVE 19</u>								<u>CURVE 30*</u>			
<u>CURVE 2</u>															
4.20	0.00526	373.00	0.209	673.20	0.280	323.20	0.173	229.03	0.266	78.2	0.182				
4.20	0.00533	673.00	0.280	<u>CURVE 12*</u>								<u>CURVE 31*</u>			
14.00	0.0264	305.00	0.348	373.20	0.189	373.20	0.189	245.02	0.211	78.2	0.150				
20.40	0.0412	<u>CURVE 7</u>										<u>CURVE 25</u>			
63.30	0.137	373.00	0.197	473.20	0.222	473.20	0.222	263.14	0.217	25.98	0.0565				
73.70	0.138	673.00	0.251	573.20	0.255	573.20	0.255	282.81	0.223	40.92	0.0950				
77.90	0.125	<u>CURVE 3*</u>										55.32	0.118		
77.00	0.141	361.20	0.280	673.20	0.284	673.20	0.284	295.20	0.226	70.02	0.133				
77.30	0.146	392.20	0.305	773.20	0.321	773.20	0.321	<u>CURVE 25</u>			86.51	0.144			
77.30	0.153	416.20	0.301	873.20	0.355	873.20	0.355	283.20	0.355	<u>CURVE 32</u>					
<u>CURVE 3</u>															
2.55	0.00369	373.00	0.209	1073.20	0.422	1073.20	0.422	25.98	0.0565	<u>CURVE 33*</u>					
4.25	0.00911	673.00	0.276	1173.20	0.455	1173.20	0.455	40.92	0.0950	<u>CURVE 34</u>					
14.00	0.0464	303.20	0.349	<u>CURVE 14</u>								70.02	0.133		
<u>CURVE 14</u>															
467.70	0.310	<u>CURVE 14</u>										86.51	0.144		

\* Not shown on plot

FIGURE SHOWS ONLY 14 OF THE CURVES REPORTED IN TABLE



SPECIFICATION TABLE NO. 288 THERMAL CONDUCTIVITY OF [NICKEL + IRON +  $\Sigma X_i$ ] ALLOYS(Ni + Fe < 99.50% or at least one  $X_i > 0.20\%$ )

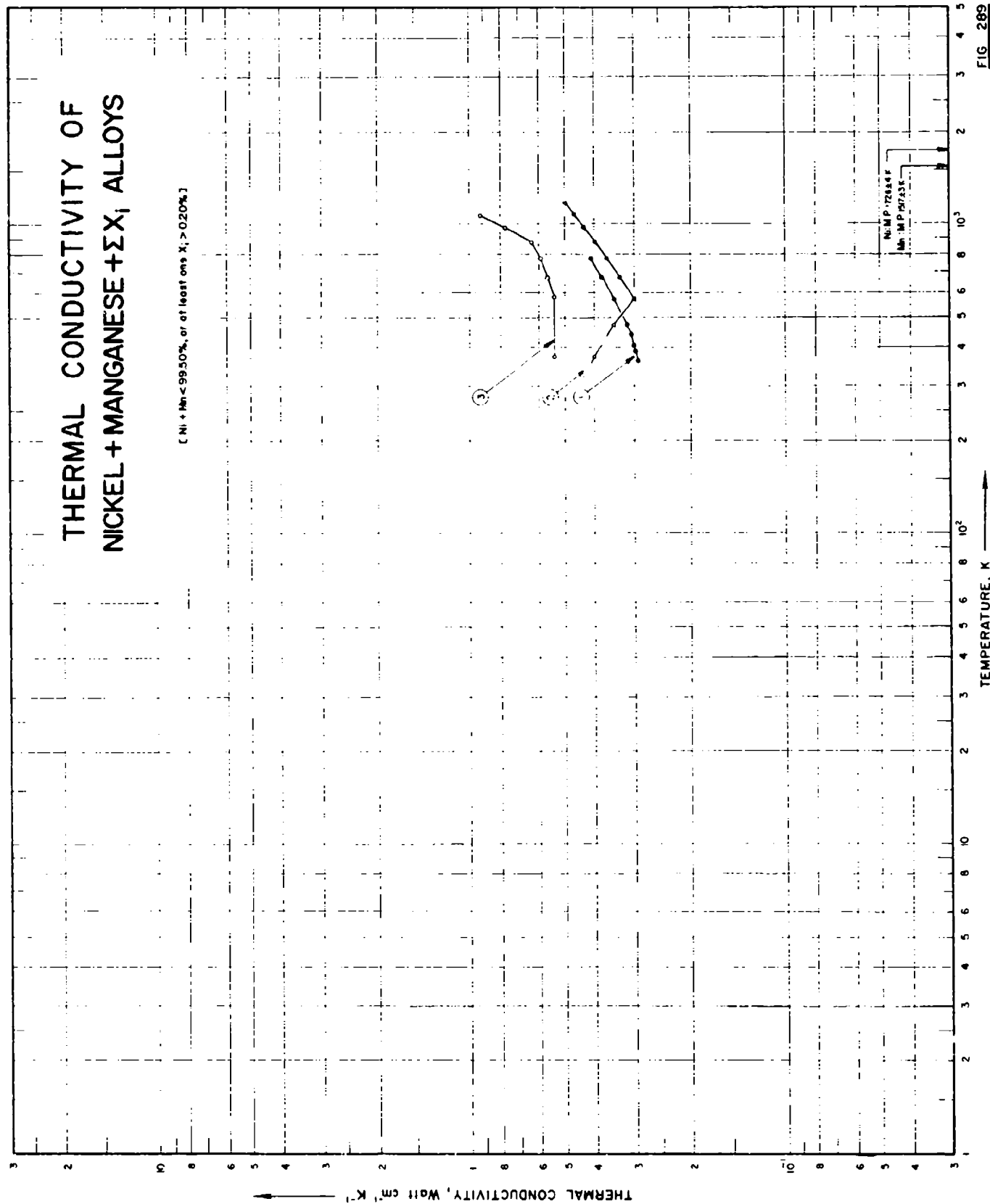
(For Data Reported in Figure and Table No. 288)

Cu's A. No.	Ref. Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Ni	Fe	C	Co	Cr	Cu	Mn	Mo	Si	Composition(continued), Specifications and Remarks
1	81	L	1939	2.3- 21	Contractid	60.0	16.0		15.0				7.0		
2	17	L	1958	293- 353	Contractid B7M	59.17	23.16		14.92			2.02	0.73	Trace	
3	169	L	1932	373-1173		61.0	20.0		15.0			4.0			
4	186	P	1928	305	Nichrome	62.0	26.0		12.0						
5	129	C	1933	373- 795	3.0-5.0	Chromel C	61.0	23.0		16.0					Hot-rolled.
6	178	E	1918	303	14a	55.4	41.14	0.208	0.142		2.61	0.331			0.133 0.012 P, 0.021 S; annealed at 900 C.
7	178	E	1918	303	15a	73.8	21.91	0.227	0.169		3.37	0.338			0.141 0.006 P, 0.019 S; annealed at 900 C.
8	178	E	1918	303	16a	83.0	12.27	0.266	0.213		3.75	0.341			0.144 0.003 P, 0.018 S; annealed at 900 C.
9	163	L	1936	303-1173	18	62.85	18.57	0.12		16.95		1.0			0.51 Rolled.
10	131	C	1953	323-1173	2.0	Hastelloy A	57.1	21.4	0.072			2.5	19.0		Annealed at 1150 C.
11	104	L	1951	15- 92	5277	57.5	40.71	0.34		14.74		1.31		0.14	As forged.
12	219	L	1951	28- 300	≤2.5	Contractid	60.05	15.82	0.05			2.14	7.20		Provided by G. E. Co.
13	73	P	1955	273-923		(min)99.0	0.3	0.17			0.2	0.27			0.18 Mg, 0.04 S, 0.14 SiO <sub>2</sub> ; annealed.
14	218		1956	323			0.4	0.15			0.25	0.35		0.35	0.01 S; annealed.
15	937	L	1961	123-813	HyMu-80	79.24	15.283	0.049		0.08		0.71	4.20	0.19	Specimen 2.54 cm in dia and 37 cm long; packed in powder and annealed in hydrogen for 5 hrs at 922.1 K, 5 hrs at 1449.8 K, furnace cooled to 699.8 K, cooled in hydrogen; specimen, chemical composition and heat-treatment history provided by the International Nickel Co.; data presented as a smooth curve.

DATA TABLE NO. 288 THERMAL CONDUCTIVITY OF [NICKEL + IRON + EX<sub>1</sub>] ALLOYS(Ni + Fe < 99.50% or at least one X<sub>1</sub> > 0.20%)T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup> K<sup>-1</sup>

T	k	T	k	T	k	T	k	T	k
<u>CURVE 1</u>									
2.34	0.00135	303.20	0.236	27.01	0.0360	173.2	0.167		
5.07	0.00360			40.28	0.0552	273.2	0.189		
8.36	0.00489	<u>CURVE 6</u>		55.16	0.0644	373.2	0.204		
10.70	0.00756			69.86	0.0695	473.2	0.222		
14.50	0.0116	303.20	0.304	89.88	0.0766	573.2	0.234		
17.26	0.0129	<u>CURVE 9</u>		104.93	0.0787	673.2	0.245		
18.50	0.0142			119.77	0.0820	773.2	0.255		
21.00	0.0160			134.83	0.0845	813.2	0.260		
<u>CURVE 2</u>									
		303.20	0.176	149.90	0.0874				
		323.20	0.176	154.97	0.0904				
		373.20	0.180	180.05	0.0933				
293.20	0.105	473.20	0.188	204.13	0.0971				
323.20	0.112	573.20	0.201	219.69	0.100				
353.20	0.120	673.20	0.218	235.02	0.104				
<u>CURVE 3</u>									
		773.20	0.239	250.04	0.108				
		873.20	0.259	255.23	0.109				
		973.20	0.280	286.58	0.117				
373.20	0.119	1073.20	0.301	300.28	0.120				
473.20	0.122	1173.20	0.322						
573.20	0.126	1273.20	0.343						
673.20	0.128	1373.20	0.385						
<u>CURVE 4</u>									
				273.20	0.597				
				293.20	0.597				
				323.20	0.596				
				373.20	0.592				
				423.20	0.584				
				473.20	0.576				
				523.20	0.563				
				573.20	0.549				
				623.20	0.528				
				673.20	0.537				
				723.20	0.537				
				773.20	0.549				
				823.20	0.564				
				873.20	0.578				
				923.20	0.593				
395.00	0.136	<u>CURVE 13</u>							
<u>CURVE 5</u>									
		273.20	0.597						
		323.20	0.597						
		373.20	0.592						
		423.20	0.584						
		473.20	0.576						
		523.20	0.563						
		573.20	0.549						
		623.20	0.528						
		673.20	0.537						
		723.20	0.537						
		773.20	0.549						
		823.20	0.564						
		873.20	0.578						
		923.20	0.593						
373.20	0.132	<u>CURVE 14*</u>							
473.20	0.147	323.20	0.606						
573.20	0.161								
673.20	0.175								
773.20	0.190								
794.90	0.193								
<u>CURVE 6</u>									
		123.2	0.153						
303.20	0.151								

\* Not shown on plot



SPECIFICATION TABLE NO. 289 THERMAL CONDUCTIVITY OF [NICKEL + MANGANESE + EX<sub>1</sub>] ALLOYS

(Ni + Mn < 99.50% or at least one X<sub>1</sub> > 0.20%)

[ For Data Reported in Figure and Table No. 289 ]

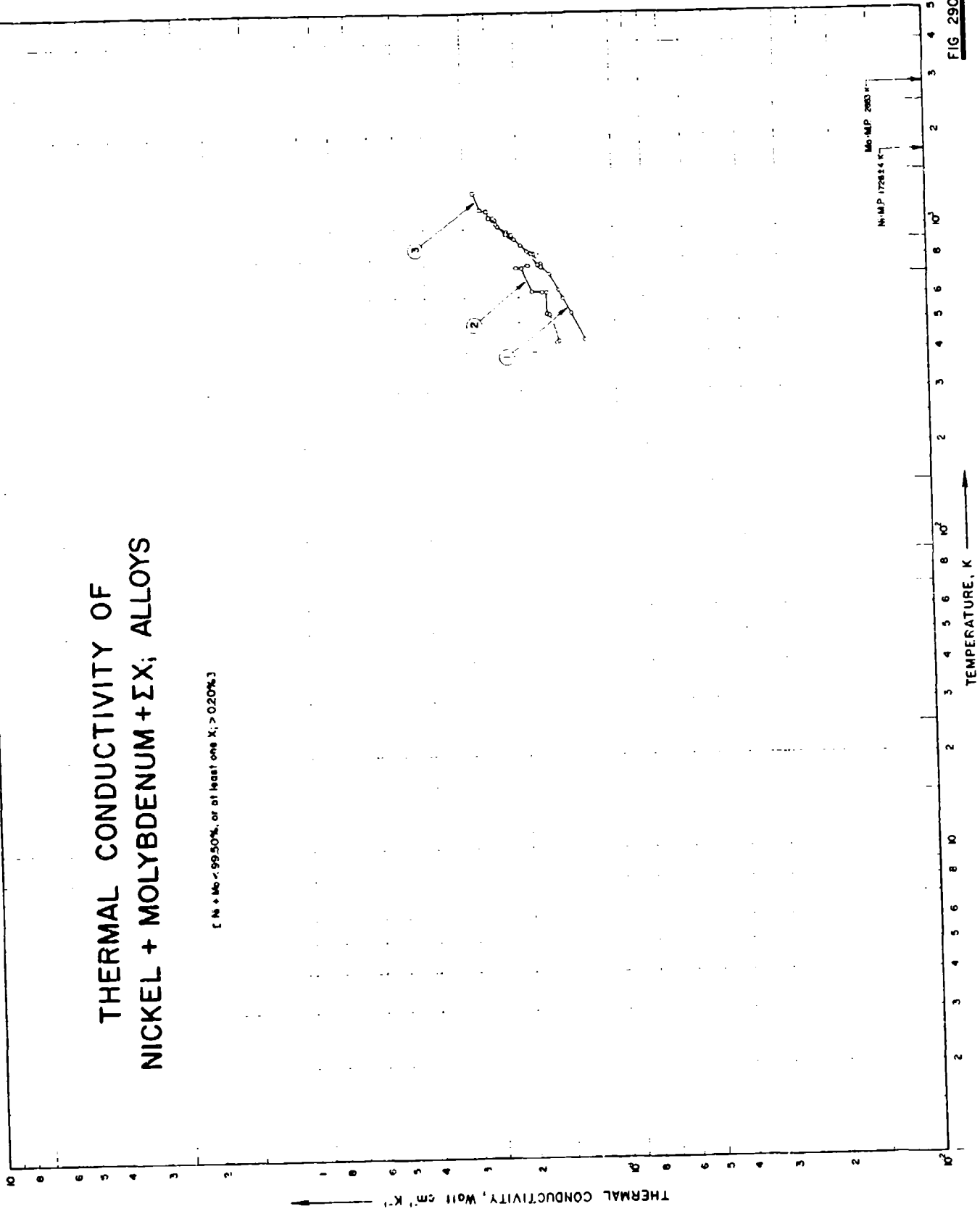
Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)							Composition (continued), Specifications and Remarks			
							Ni	Mn	Al	C	Co	Cr	Fe		Si		
1	129	C	1933	363-773	3.0-5.0	Alumel	94.943	2.0	2.0								Hot-rolled.
2	131	C	1933	323-1173	2.0	D Nickel	92.79	4.55		0.158	1.27		1.35	0.06			Annealed at 900 C.
3	439	L	1935	373-107C	2.0		97.5	1.9					0.4	0.14			
4	26	C	1954	423-910	6.0-19	Nickel A	99.542	0.25		0.034			0.068	0.03			0.034 Mg, 0.02 Ti, traces of Cu, Al, B, Ca, and Cr, cylindrical bar 2 cm in dia and ~15 cm long; Armco iron used as comparative material.

DATA TABLE NO. 289 THERMAL CONDUCTIVITY OF [NICKEL + MANGANESE +  $\Sigma X_i$ ] ALLOYS(Ni + Mn < 99.50% or at least one  $X_i > 0.20\%$ )[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup>K<sup>-1</sup>]

T	k
<u>CURVE 1</u>	
363.40	0.293
389.20	0.397
405.10	0.301
441.30	0.307
473.20	0.318
573.20	0.350
673.20	0.381
773.20	0.412
<u>CURVE 2</u>	
323.20	0.430
373.20	0.403
473.20	0.350
573.20	0.300
673.20	0.324
773.20	0.367
873.20	0.399
973.20	0.431
1073.20	0.464
1173.20	0.497
<u>CURVE 3</u>	
373.2	0.540
583.2	0.540
673.2	0.565
773.2	0.594
873.2	0.636
973.2	0.778
1073.2	0.925
<u>CURVE 4</u>	
422.7	0.713
521.4	0.591
597.4	0.464
618.6	0.435
656.3	0.473
678.9	0.479
739.2	0.468
800.0	0.491
910.0	0.578

# THERMAL CONDUCTIVITY OF NICKEL + MOLYBDENUM + ΣX; ALLOYS

(Ni + Mo < 99.50%, or at least one X; > 0.20%)





SPECIFICATION TABLE NO. 290 THERMAL CONDUCTIVITY OF [NICKEL + MOLYBDENUM +  $\Sigma X_i$ ] ALLOYS(Ni + Mo  $\geq$  99.50% or at least one  $X_i \geq 0.20\%$ )

[ For Data Reported in Figure and Table No. 290 ]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)					Composition (continued), Specifications and Remarks		
							Ni	Mo	Cr	Fe	Mn		Si	Al
1	568	R	1962	439-1090	$\pm 3$	INOR-8	71.1	16.2	7.25	4.6	0.36	0.34	0.22	0.17 W, 0.10 Ti, 0.094 C, 0.006 S, 0.003 P; specimen (composed of 18 disks, 9 in. long) 0.5 in. thick, 3 in. O. D. and 0.625 in. I. D.; prepared from a hot forged bar identified by Westinghouse as heat M1669-4.
2	569	C	1960	431-761	$\pm 5$	INOR-8	Bal.	17.0	7.0	5.0	0.8			0.06 C, 0.5 Max Al + Ti; nominal composition; specimen 6 in. long and 1 in. in dia.; Armco Iron used as comparative material.
3	43	L	1957	823-1319	$\pm 5$	Hastelloy B	65.57	23.78		5.05				0.02 C; specimen 6.75 in. in dia and 1.5 in. thick; "as received".

DATA TABLE NO. 290 THERMAL CONDUCTIVITY OF [NICKEL + MOLYBDENUM +  $\Sigma X_i$ ] ALLOYS(Ni + Mo) < 99.50% or at least one  $X_i > 0.20\%$ [Temperature,  $T$ , K; Thermal Conductivity,  $k$ , Watt  $\text{cm}^{-1} \text{K}^{-1}$ ]

T	k	CURVE 2 (cont.)	
		T	k
		<u>CURVE 1</u>	
438.7	0.123	527.6	0.183
439.2	0.124*	748.2	0.196
535.7	0.136	748.2	0.203
537.7	0.137*	760.9	0.188
589.2	0.145		
637.0	0.150	<u>CURVE 3</u>	
712.2	0.160	822.5	0.175
713.2	0.161*	825.2	0.177
759.9	0.171	962.8	0.215*
769.0	0.172	962.9	0.217
769.0	0.173*	1094.1	0.247
769.0	0.175	1148.4	0.253
772.4	0.174*	1254.5	0.265
842.4	0.188	1319.2	0.279
843.2	0.187*		
843.8	0.186*		
882.2	0.196		
883.2	0.195*		
883.7	0.194*		
923.8	0.205		
924.0	0.204*		
933.2	0.209		
953.4	0.207*		
954.0	0.206*		
974.8	0.217		
1013.2	0.231		
1038.2	0.235		
1039.6	0.235*		
1075.2	0.236		
1087.2	0.240		
1088.2	0.239*		
1089.7	0.240*		
1090.2	0.241*		
		<u>CURVE 2</u>	
430.9	0.154		
432.1	0.150		
530.4	0.161		
532.1	0.164		
622.1	0.165		
624.8	0.170		

\* Not shown on plot

SPECIFICATION TABLE NO. 291 THERMAL CONDUCTIVITY OF [NICKEL +  $\Sigma X_i$ ] ALLOYS Ni +  $\Sigma X_i$ 

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) Ni	Composition (continued), Specifications and Remarks
1	127	L	1922	572-1006			99.2	Specimen cast in 3 in. mold, hot-rolled to 1.5 in. dia., reheated and rolled to 0.875 in. dia., close-annealed at 800 C, cold drawn to 0.8125 in. dia., re-annealed and then drawn to 0.75 in. dia., and finally annealed at 750 C and 800 C; density 8.79 g cm <sup>-3</sup> at 21 C.
2	34	L	1927	80-273			99.9	Electrical conductivity 90.2 and 13.95 x 10 <sup>4</sup> ohm <sup>-1</sup> cm <sup>-1</sup> at 80 and 273 K, respectively.
3	40	L	1956	78-1616	5.0	Grade A		Grade "A" nickel; specimen 7 in. in dia and 1.5 in. thick; density 8.844 g cm <sup>-3</sup> .
4	104	L	1951	15-93			99.4	As forged.
5	88	L	1908	113-301			99.0	
6	276	C	1953	343.2	3.0	Grade A		Grade "A" nickel (nominally 99.4 Ni + Co); density 8.8 g cm <sup>-3</sup> ; Armco iron used as comparative material.
7	163	L	1936	348-473		I	99.23	Data determined by using D. Hattori's method.
8	163	L	1936	588-1428		I		The above specimen using the method of K. Honda and T. Simidu; measured while increasing temp.
9	163	L	1936	468-1073		I		The above specimen; measured while reducing temp.
10	71	L	1917	299-1130			96.8	

DATA TABLE NO. 291 THERMAL CONDUCTIVITY OF NICKEL +  $\Sigma_{i=1}^n$  ALLOYS Ni +  $\Sigma_{i=1}^n$ [Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup> K<sup>-1</sup>]

T	k	T	k	T	k
<u>CURVE 1*</u>					
372.4	0.611	15.12	0.180	693.2	0.477
374.0	0.607	18.15	0.221	803.2	0.515
414.8	0.598	21.50	0.274	943.2	0.567
414.8	0.598	77.10	0.610	1078	0.617
414.8	0.607	93.10	0.661	1178	0.661
439.7	0.589	<u>CURVE 3*</u>			
469.2	0.589	1293	0.682	1273	0.676
469.4	0.577	1313	0.686	1313	0.686
469.6	0.596	113.2	0.540	1338	0.690
469.9	0.552	123.2	0.543	1373	0.686
561.6	0.540	148.2	0.552	1428	0.690
562.4	0.540	171.2	0.560	<u>CURVE 9*</u>	
562.9	0.596	198.2	0.569	468.2	0.653
563.4	0.536	223.2	0.573	698.2	0.494
566.9	0.556	248.2	0.582	863.2	0.527
626.8	0.540	273.2	0.586	1073	0.611
626.2	0.537	291.2	0.596	<u>CURVE 10*</u>	
630.4	0.536	301.2	0.577	299.2	0.552
630.5	0.536	<u>CURVE 5*</u>			
764.2	0.531	343.2	0.636	346.2	0.552
764.6	0.540	<u>CURVE 7*</u>			
976.5	0.590	348.2	0.782	423.2	0.552
1906.2	0.615	378.2	0.732	482.2	0.544
<u>CURVE 2*</u>					
80	1.114	433.2	0.678	524.2	0.527
273	0.539	473.2	0.653	568.2	0.521
<u>CURVE 3*</u>					
778	0.458	628.2	0.498	628.2	0.498
815	0.505	705.2	0.494	705.2	0.494
942	0.504	796.2	0.527	796.2	0.527
920	0.551	823.2	0.536	823.2	0.536
1088	0.583	856.2	0.531	856.2	0.531
1295	0.600	871.2	0.540	871.2	0.540
1386	0.651	918.2	0.61	918.2	0.61
1460	0.690	972.2	0.556	972.2	0.556
1616	0.719	982.2	0.577	982.2	0.577
<u>CURVE 4*</u>					
<u>CURVE 5*</u>					
<u>CURVE 6*</u>					
<u>CURVE 7*</u>					
<u>CURVE 8*</u>					
<u>CURVE 9*</u>					
<u>CURVE 10*</u>					

\* No graphical presentation

# THERMAL CONDUCTIVITY OF NIOBIUM + MOLYBDENUM + $\Sigma X_i$ ALLOYS

[ Nb + Mo < 99.50%, or at least one  $X_i$ ; > 0.20% ]

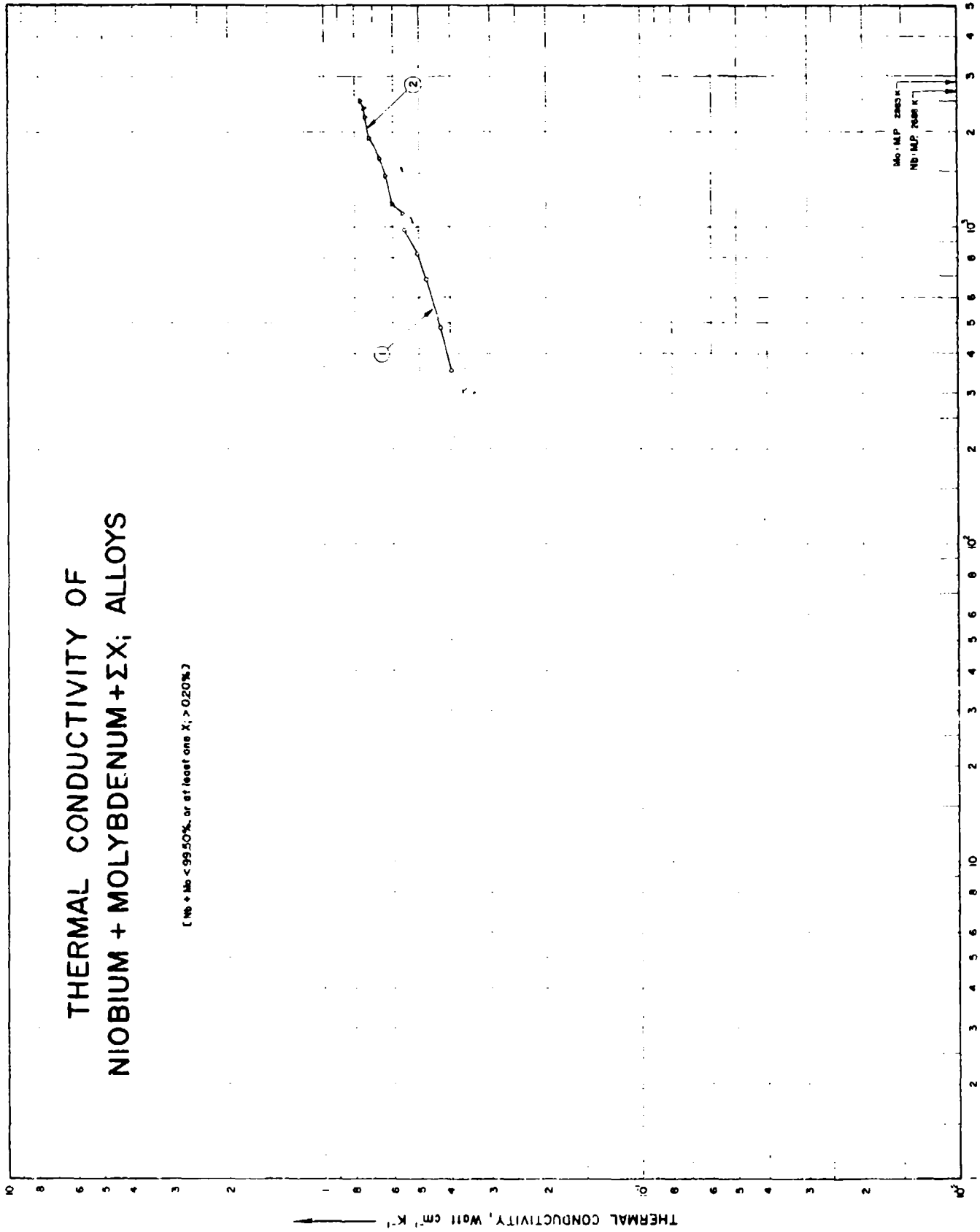


FIG 292

SPECIFICATION TABLE NO. 292 THERMAL CONDUCTIVITY OF [NIOBIUM - MOLYBDENUM + ΣX<sub>i</sub>] ALLOYS  
(Nb + Mo = 99.1% or at least one X<sub>i</sub> ≥ 0.20%)

(For Data Reported in Figure and Table No. 292.)

Curve No.	Ref. Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)			Composition (continued), Specifications and Remarks
						Nb	Mo	Zr	
1	C	1963	353-983	±4	Nb-5Mo-5V-1Zr	88.77	5.03	1.13	0.028 C, 0.0136 N, 0.0093 O; specimen 2 in. in dia. and 1 in. long; density 8.62 g cm <sup>-3</sup> ; measured in helium atmosphere; Armco iron used as comparative material.
2	P	1963	1103-2508	±4	Nb-5Mo-5V-1Zr	88.77	5.03	1.13	0.028 C, 0.0136 N, 0.0093 O; the above specimen measured by another method; thermal conductivity values were calculated from the measurement of thermal diffusivity, specific heat and density.

DATA TABLE NO. 232 THERMAL CONDUCTIVITY OF [NIOBIUM + MOLYBDENUM +  $\Sigma X_i$ ] ALLOYS(Nb + Mo < 99.50% or at least one  $X_i > 0.20\%$ )[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup> K<sup>-1</sup>]

T	k
<u>CURVE 1</u>	
352.6	0.398
483.2	0.431
688.7	0.476
822.1	0.504
983.1	0.550
<u>CURVE 2</u>	
1102.6	0.561
1193.2	0.604
1466.5	0.639
1660.9	0.665
1922.1	0.720
2230.4	0.739
2399.8	0.746
2508.2	0.765

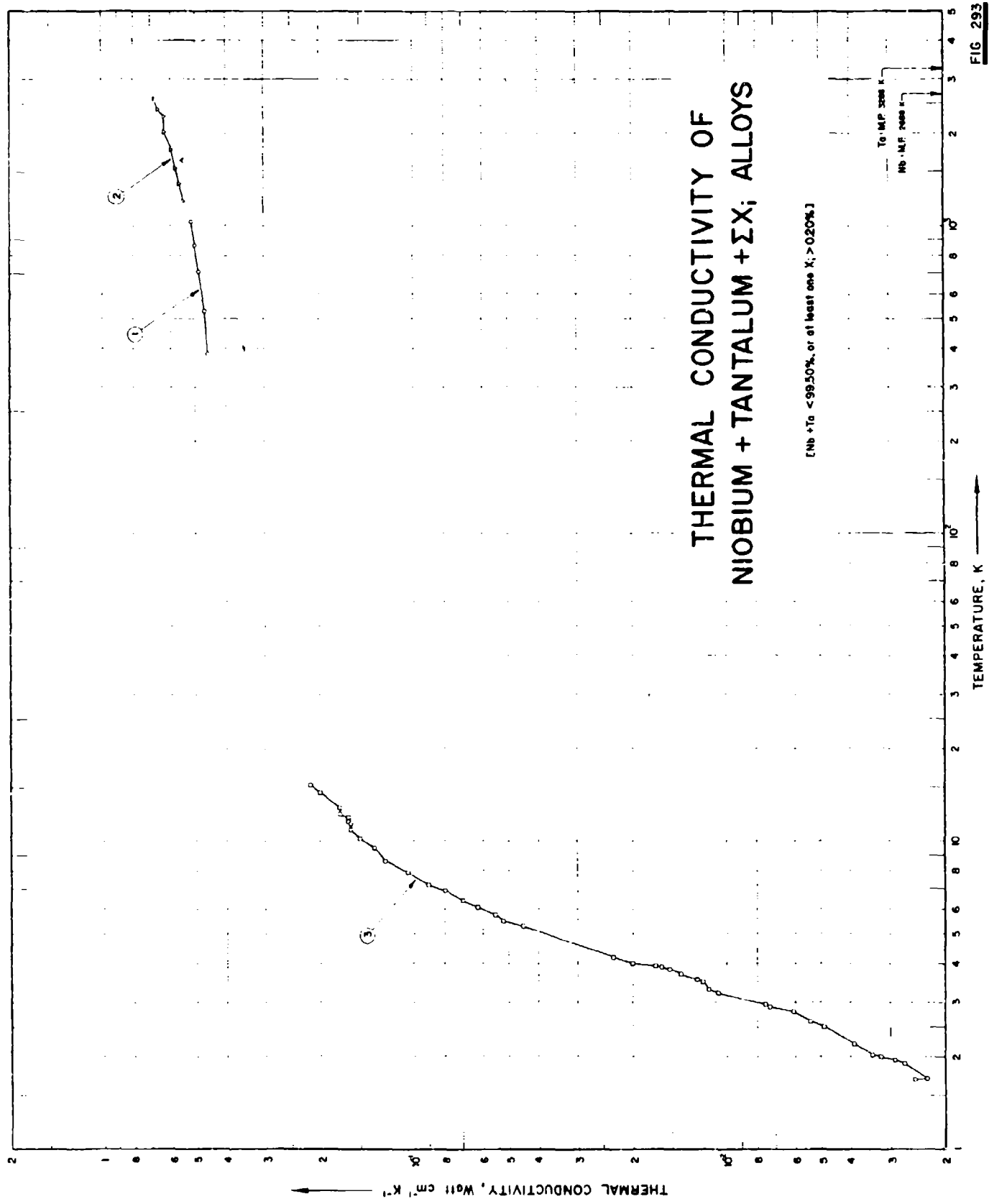


FIG. 293



SPECIFICATION TABLE NO. 293 THERMAL CONDUCTIVITY OF [Niobium + Tantalum + ΣX<sub>i</sub>] ALLOYS(Nb + Ta < 99.50% or at least one X<sub>i</sub> > 0.20%)

[For Data Reported in Figure and Table No. 293]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)					Composition (continued), Specifications and Remarks	
							Nb	Ta	W	Zr	Fe		Si
1	583	C	1963	386-1044	±4	Nb-27Ta-12W-0.2Zr	60.8	27.84	10.40	0.92	0.007	0.01	0.009 Ni, 0.005 Ti, 0.005 O, 0.004 C, and 0.002 N; specimen 2 in. in dia and 1 in. long with ends ground flat and parallel; density 10.72 g cm <sup>-3</sup> ; measured in helium atmosphere; Armco iron used as comparative material.
2	583	P	1963	1208-2592	±4		50.8	27.84	10.40	0.92	0.007	0.01	The above specimen measured by another method.
3	704	L	1963	1.7-15	5		Bal.	1.0		0.03	0.02	0.005 Sn, 0.003 Pb, 0.0005 Cr, 0.0005 Ni, and 0.0001 Mg; specimen 10 cm long and 0.474 cm in dia; measured in "as received" condition; specimens from Johnson, Matthey and Co., Ltd.	

DATA TABLE NO. 293 THERMAL CONDUCTIVITY OF [NIOBIUM + TANTALUM +  $\Sigma X_i$ ] ALLOYS(Nb + Ta < 99.50% or at least one  $X_i > 0.20\%$ )[Temperature, T, K; Thermal Conductivity, k, Watt  $\text{cm}^{-1} \text{K}^{-1}$ ]

T	k	T	k
<u>CURVE 1</u>			
386.0	0.459	5.36	0.0450
530.4	0.467	5.50	0.0520
710.9	0.486	5.75	0.0550
869.3	0.502	6.10	0.0625
1044.3	0.517	6.40	0.0700
		6.90	0.0900
		7.20	0.0900
		7.90	0.105
		8.60	0.124
		9.50	0.134
		9.60	0.134*
		10.20	0.150
		10.25	0.149*
		10.95	0.162
		11.40	0.161
		11.60	0.164
		11.95	0.164
		12.20	0.174
		12.80	0.174
		14.40	0.199
		15.20	0.215
<u>CURVE 2</u>			
1208.2	0.547		
1379.3	0.566		
1545.4	0.578		
1780.4	0.594		
2027.6	0.630		
2227.6	0.630		
2399.8	0.659		
2591.5	0.675		
<u>CURVE 3</u>			
1.69	0.00250		
1.70	0.00228		
1.90	0.00270		
1.96	0.00290		
2.00	0.00320		
2.03	0.00340		
2.20	0.00390		
2.50	0.00490		
2.60	0.00540		
2.80	0.00610		
2.90	0.00730		
2.94	0.00750		
3.20	0.0107		
3.30	0.0115		
3.50	0.0120		
3.55	0.0125		
3.70	0.0140		
3.84	0.0153		
3.90	0.0163		
3.95	0.0170		
4.00	0.0200		
4.04	0.0201*		
4.20	0.0230		

\* Not shown on plot

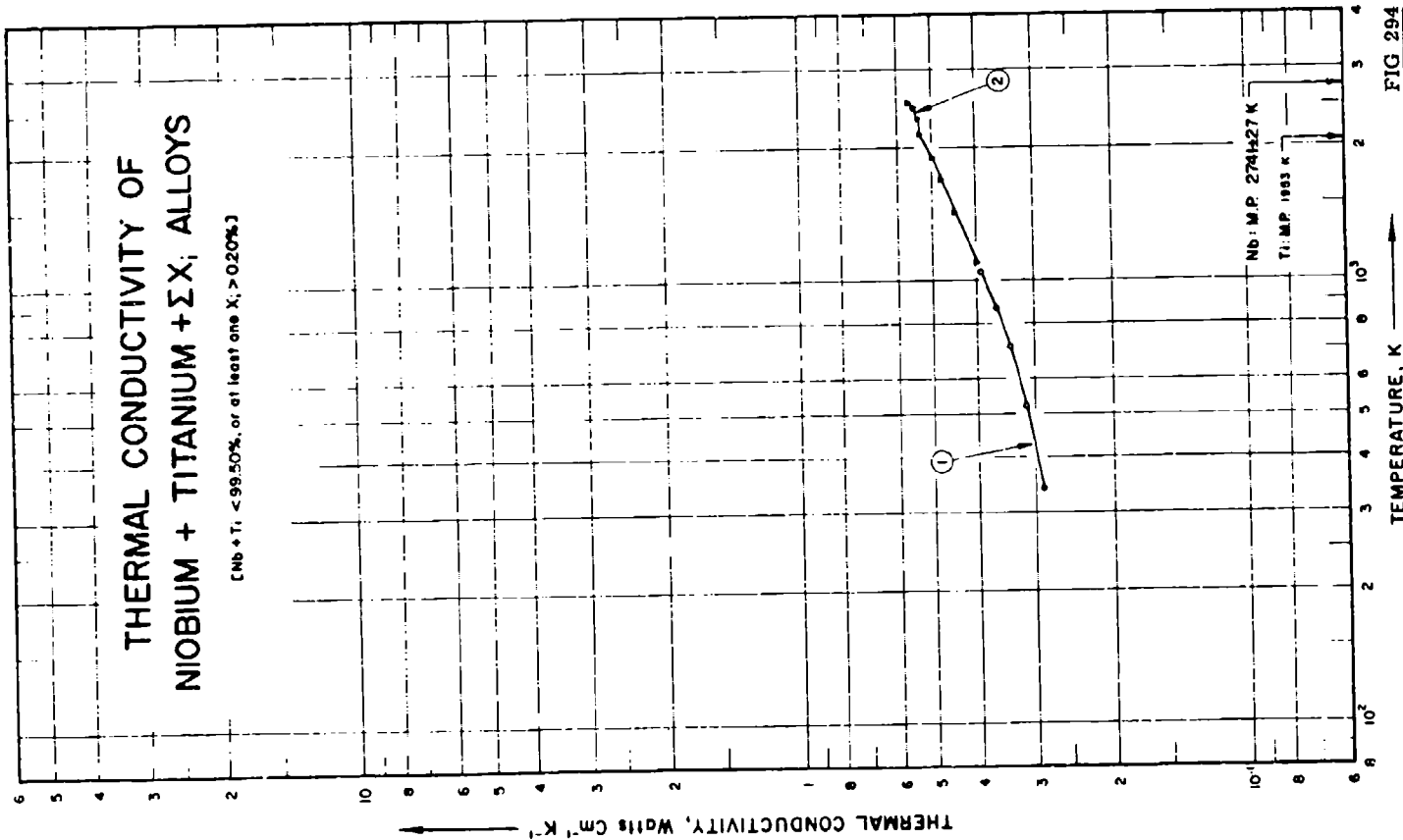


FIG. 294

SPECIFICATION TABLE NO. 294 THERMAL CONDUCTIVITY OF [NiOBiUM + TiTANIUM + EX<sub>1</sub>] ALLOYS(Nb + Ti < 99.50% or at least one X<sub>1</sub> ± 0.20%)

[ For Data Reported in Figure and Table No. 294 ]

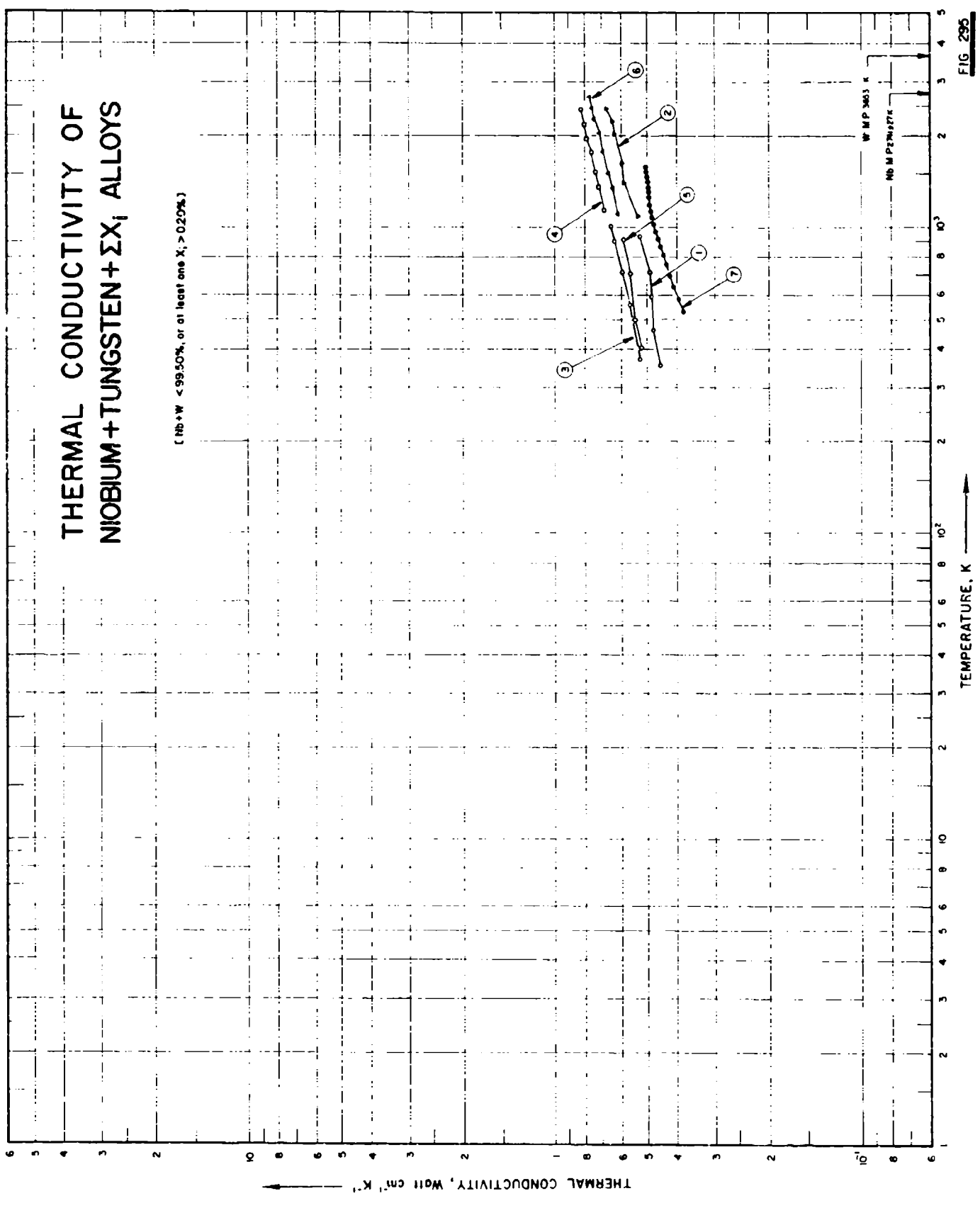
Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)		Composition (continued), Specifications and Remarks
							Nb	Ti	
1	583	C	1963	342-1053	±4	Nb-10Ti-5Zr	83.96	10.5	0.0249 O, 0.0071 C, 0.0027 N and 0.0009 H; specimen 2 in. in dia and 1 in. long; density 7.77 g cm <sup>-3</sup> ; measured in helium atmosphere; Armco iron used as comparative material.
2	583	P	1963	1105-2544	±4	Nb-10Ti-5Zr	83.96	10.5	0.0249 O, 0.0071 C, 0.0027 N and 0.0009 H; the above specimen measured by another method; thermal conductivity values calculated from measurements of thermal diffusivity, specific heat and density.

DATA TABLE NO. 294 THERMAL CONDUCTIVITY OF [NIOBIUM + TITANIUM +  $\Sigma X_i$ ] ALLOYS(Nb + Ti < 99.50% or at least one  $X_i \geq 0.20\%$ )[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup> K<sup>-1</sup>]

T	k
<u>CURVE 1</u>	
341.5	0.287
523.2	0.312
712.6	0.339
972.1	0.362
1053.2	0.393
<u>CURVE 2</u>	
1105.4	0.398
1447.1	0.443
1702.6	0.471
1916.5	0.498
2163.7	0.530
2349.8	0.533
2483.2	0.547
2544.3	0.561

# THERMAL CONDUCTIVITY OF NIOBIUM + TUNGSTEN + $\Sigma X_i$ ALLOYS

(Nb + W < 99.50%, or at least one  $X_i$  > 0.20%)



SPECIFICATION TABLE NO. 295 THERMAL CONDUCTIVITY OF [NIOBIUM + TUNGSTEN +  $\Sigma X_i$ ] ALLOYS(Nb + W < 99.50% or at least one  $X_i > 0.20\%$ )

[For Data Reported in Figure and Table No. 295]

Curve No.	Ref. Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)			Composition (continued), Specifications and Remarks
						Nb	W	Zr	
1	583 C	1963	355-949	±4	Nb-10W-5Zr	87.3	9.88	2.8	0.008 O, 0.004 N, 0.002 C and 0.0011 H; specimen 2 in. in dia and 1 in. long; density 9.16 g cm <sup>-3</sup> ; measured in helium atmosphere; Armco iron used as comparative material.
2	583 P	1963	1098-2461	±4	Nb-10W-5Zr	87.3	9.88	2.8	0.008 O, 0.004 N, 0.002 C and 0.0011 H; the above specimen measured by another method; thermal conductivity values were calculated from the measurement of thermal diffusivity, specific heat and density.
3	583 C	1963	372-1011	±4	Nb-10W-1Zr-0.1C	89.39	9.6	0.95	0.051 C, 0.0053 O, 0.0033 N and 0.0003 H; specimen 2 in. in dia and 1 in. long; density 9.03 g cm <sup>-3</sup> ; measured in helium atmosphere; Armco iron used as comparative material.
4	583 P	1963	1150-2455	±4	Nb-10W-1Zr-0.1C	89.39	9.6	0.95	0.051 C, 0.0053 O, 0.0033 N and 0.0003 H; the above specimen measured by another method; thermal conductivity values were calculated from the measurement of thermal diffusivity, specific heat and density.
5	583 C	1963	405-915	±4	Nb-15W-5Mo-1Zr-0.05C	78.78	15.6	0.84	0.01 Ta, 0.0489 C, 0.0163 O, 0.002 N and 0.0005 H; specimen 2 in. in dia and 1 in. long; density 9.6 g cm <sup>-3</sup> ; measured in helium atmosphere; Armco iron used as comparative material.
6	583 P	1963	1117-2678	±4	Nb-15W-5Mo-1Zr-0.05C	78.78	15.6	0.84	0.01 Ta, 0.0489 C, 0.0163 O, 0.002 N and 0.0005 H; the above specimen measured by another method; thermal conductivity values were calculated from the measurement of thermal diffusivity, specific heat and density.
7	968	1963	533-1589	±10	Haynes alloy Nb-752	87.5	10	2.5	No details reported.

DATA TABLE NO. 295 THERMAL CONDUCTIVITY OF NIOBIUM + TUNGSTEN +  $\Sigma X_i$  ALLOYS(Nb + W < 99.50% or at least one  $X_i > 0.20\%$ )[Temperature, T, K; Thermal Conductivity, k, Watt  $\text{cm}^{-1} \text{K}^{-1}$ ]

T	k	T	k
<u>CURVE 1</u>			
355.4	0.453	1516.5	0.668
462.7	0.478	1788.2	0.652
594.3	0.486	2044.3	0.717
719.3	0.490	2283.2	0.742
938.7	0.523	2483.2	0.758
		2677.6	0.775
<u>CURVE 2</u>			
1097.6	0.533	<u>CURVE 7</u>	
1406.5	0.588	533.2	0.381
1643.2	0.599	588.7	0.398
2027.6	0.643	644.3	0.412
2244.3	0.640	699.8	0.424
2461.0	0.675	755.4	0.436
		810.9	0.447
<u>CURVE 3</u>			
372.1	0.528	866.5	0.457
560.9	0.566	922.1	0.464
716.5	0.597	977.6	0.471
905.4	0.632	1033.2	0.478
1010.9	0.652	1088.7	0.485
		1144.3	0.490
<u>CURVE 4</u>			
1148.8	0.639	1199.8	0.493
1363.4	0.715	1255.4	0.499
1522.1	0.736	1310.9	0.497
1766.5	0.751	1366.5	0.498
1960.9	0.787	1422.1	0.500
2183.2	0.800	1477.6	0.502
2455.4	0.824	1533.2	0.504
		1588.7	0.505
<u>CURVE 5</u>			
405.4	0.521		
499.8	0.549		
708.2	0.564		
914.8	0.590		
<u>CURVE 6</u>			
1116.5	0.620		
1352.6	0.644		



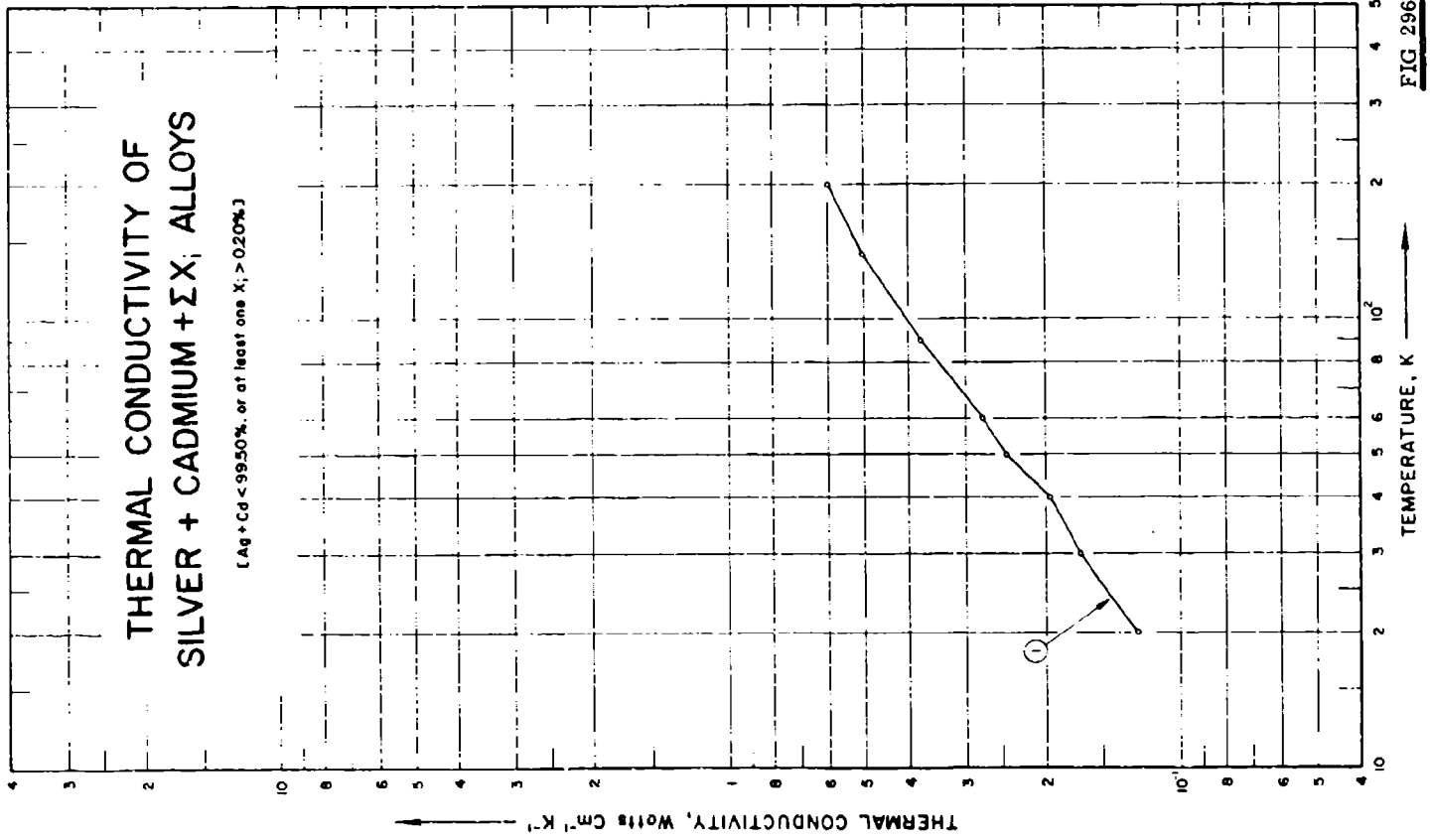


FIG. 296

SPECIFICATION TABLE NO. 296 THERMAL CONDUCTIVITY OF SILVER-CADMIUM-CAD<sub>3</sub> ALLOYS  
(Ag + Cd = 99.50% or at least one N<sub>1</sub> > 0.20%)

(For Data Reported in Figure and Table No. 296.)

Curve No.	Ref. Method No. Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)				Composition (continued), Specifications and Remarks
						Ag	Cd	Cu	Zn	
1	544 L	1953	20-240	5.0	Easy-Flow Silver solder	50.0	14.0	15.5	16.5	Drawn and annealed.

DATA TABLE NO. 296 THERMAL CONDUCTIVITY OF [SILVER + CADMIUM +  $\Sigma X_i$ ] ALLOYS(Ag + Cd < 99.50% or at least one  $X_i > 0.20\%$ )[Temperature, T, K; Thermal Conductivity, k, Watt  $\text{cm}^{-1}\text{K}^{-1}$ ]

T	k
20	0.126
30	0.167
40	0.196
50	0.243
60	0.275
90	0.376
140	0.503
200	0.600

CURVE 1

SPECIFICATION TABLE NO. 297 THERMAL CONDUCTIVITY OF [SILVER +  $\Sigma X_i$ ] ALLOYS Ag +  $\Sigma X_i$

Curv No.	Ref. Method No. Used	Year	Temp. Rangr. K	Reported Error. %	Name and Specimen Designation	Composition (weight percent) Ag	Composition (continued), Specifications and Remarks
1	37	C	1951	310-810	4.0	99.4	National Bureau of Standards' melting-point standard lead used as comparative material.

DATA TABLE NO. 297 THERMAL CONDUCTIVITY OF [SILVER +  $\Sigma X_i$ ] ALLOYS Ag +  $\Sigma X_i$

[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup> K<sup>-1</sup>]

T	k
<u>CURVE 1*</u>	
310	4.195
367	4.170
422	4.075
478	3.986
533	3.900
588	3.805
644	3.720
700	3.630
755	3.540
810	3.453

\* No graphical representation

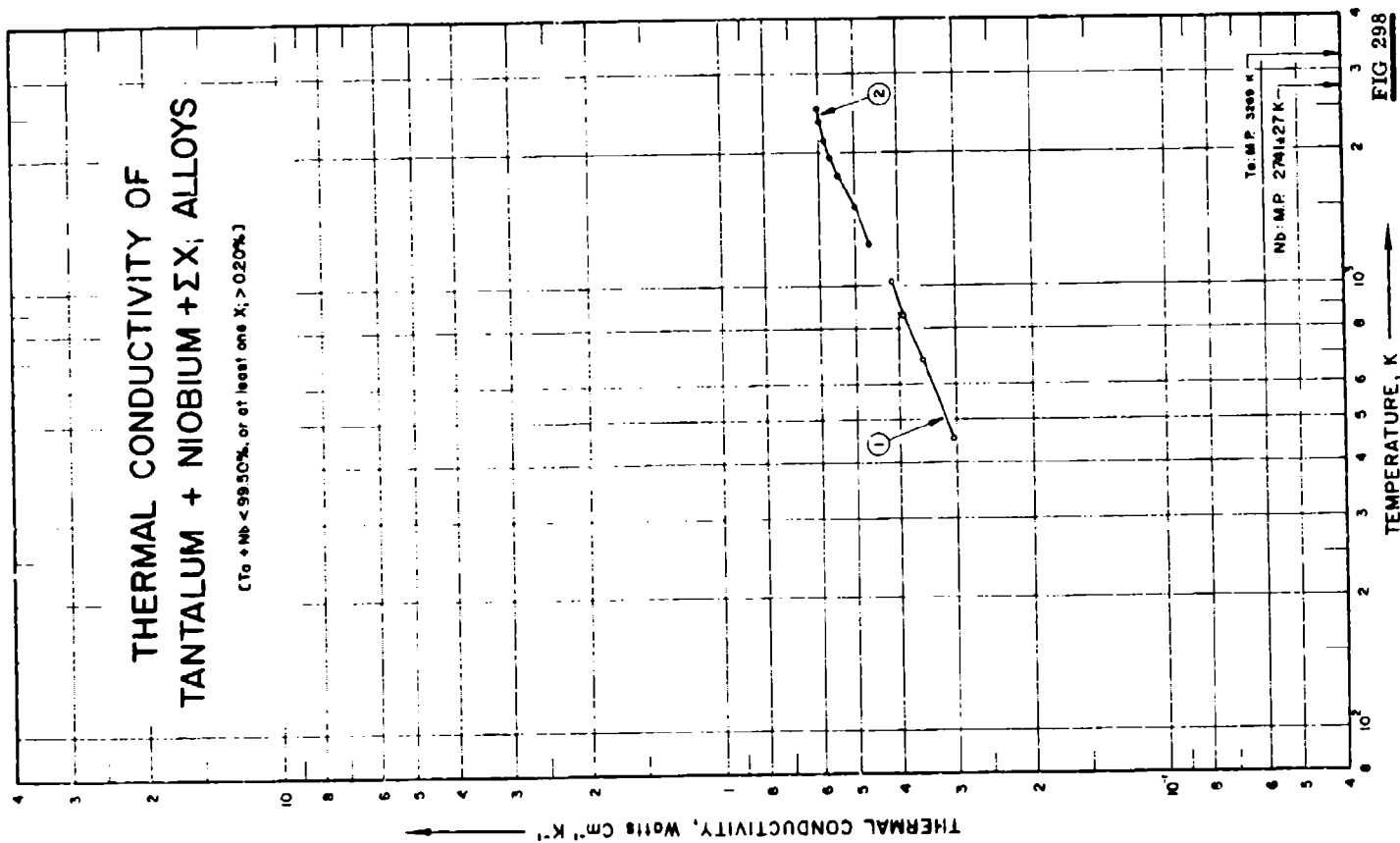


FIG 298

SPECIFICATION TABLE NO. 298 THERMAL CONDUCTIVITY OF [TANTALUM + NIOBIUM +  $\Sigma X_i$ ] ALLOYS(Ta + Nb < 99.50% or at least one  $X_i > 0.20\%$ )

[For Data Reported in Figure and Table No. 298]

Curve No.	Seif. Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)			Composition (continued), Specifications and Remarks	
						Ta	Nb	C		
1	C	1963	454-1031	±4	Ta-30Nb-7.5V	62.12	30.3	7.47	0.09	0.015 O, 0.0065 N; specimen 2 in. in dia and 1 in. long; density 11.55 g cm <sup>-3</sup> ; measured in helium atmosphere; Armco iron used as comparative material.
2	P	1963	1246-2511	±4	Ta-30Nb-7.5V	62.12	30.3	7.47	0.09	0.015 O, 0.0065 N; the above specimen measured by another method; thermal conductivity values were calculated from the measurement of thermal diffusivity, specific heat and density.

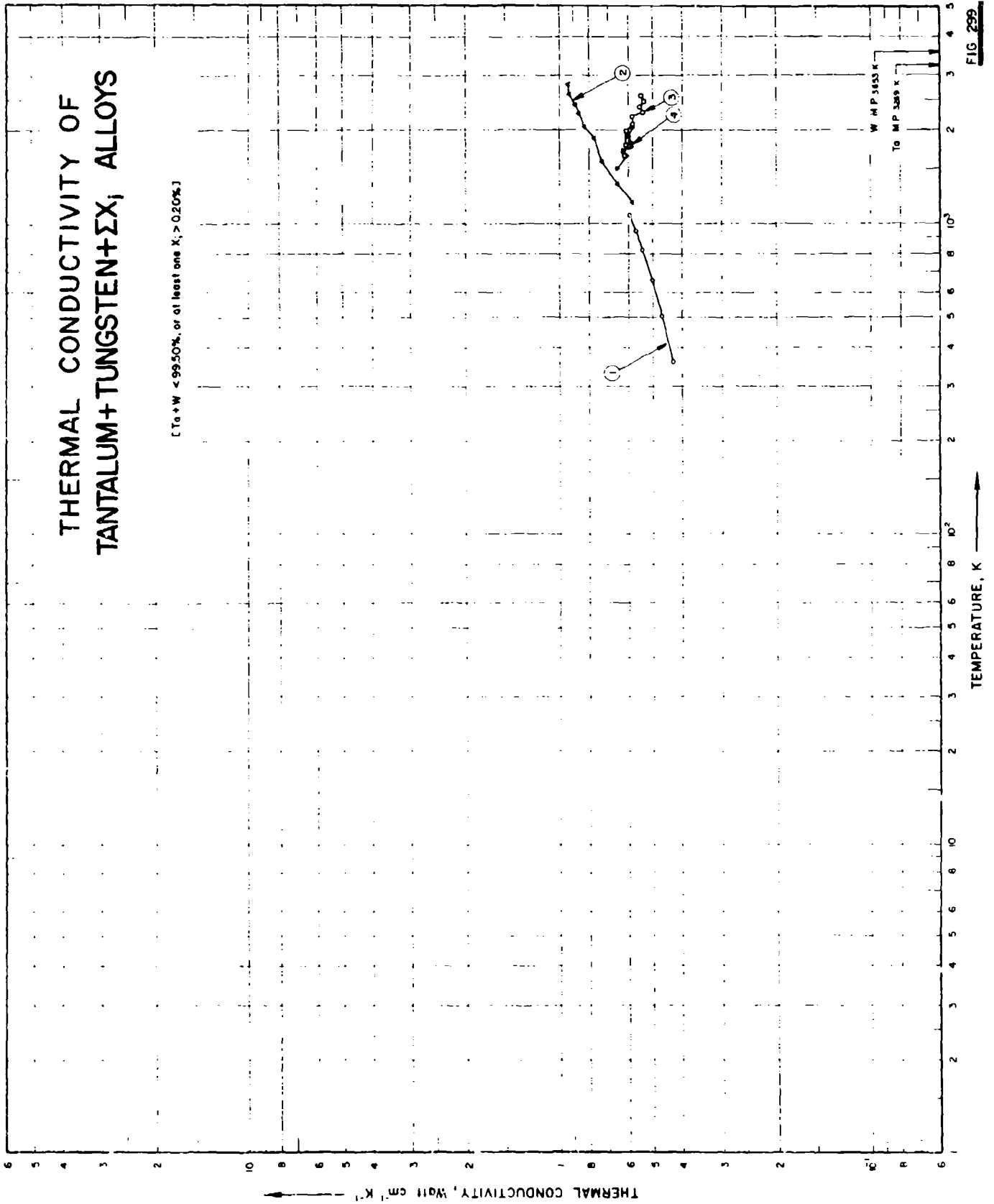
DATA TABLE NO. 298 THERMAL CONDUCTIVITY OF [TANTALUM + NIOBIUM +  $\Sigma X_i$ ] ALLOYS  
 (Ta + Nb < 99.50% or at least one  $X_i \geq 0.20\%$ )

[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup> K<sup>-1</sup>]

T	k
<u>CURVE 1</u>	
454.3	0.305
688.7	0.353
860.9	0.391
1039.9	0.415
<u>CURVE 2</u>	
1245.9	0.467
1508.2	0.497
1772.1	0.545
1954.2	0.569
2133.7	0.543
2352.6	0.598
2511.0	0.604

# THERMAL CONDUCTIVITY OF TANTALUM + TUNGSTEN + $\Sigma X_i$ ALLOYS

[Ta + W < 99.50%, or at least one  $X_i$ ; > 0.20%]



TEMPERATURE, K →

FIG. 299



## SPECIFICATION TABLE NO. 299 THERMAL CONDUCTIVITY OF [TANTALUM + TUNGSTEN + EX.] ALLOYS

(Ta + W < 99.50% or at least one X<sub>i</sub> > 0.20%)

[ For Data Reported in Figure and Table No. 299 ]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition Ta	Composition W	Composition (weight percent) HF	Composition (continued), Specifications and Remarks
1	583	C	1963	361-1072	± 4.0	Ta-RW-2HF	88.79	9.0	2.2	0.0041 C, 0.0040 O, 0.0023 N; specimen 2 in. in dia and 1 in. in length; end-ground flat and parallel; density 16.95 g cm <sup>-3</sup> ; measured in helium atmosphere; Armco iron used as comparative material.
2	583	P	1963	1172-2303	± 4.0	Ta-RW-2HF	88.79	9.0	2.2	The above specimen measured by another method.
3	849	-	1966	1660-2599		Tzz; No. 1	Bal	8.5	2.5	0.0095 C, 0.0012 H and 0.0006 O impurities; specimen 2.2524 cm in dia and 0.332 cm long; heated in high vacuum (10 <sup>-4</sup> mm Hg) by high frequency induction to 1000-3000 C; localized heating within 0.003 in. of the surface at current frequency of 50000 cps, heat lost only by radiation, the cylindrical surface being assumed isothermal, and the temperature gradient along the radius was analytically correlated to the thermal conductivity; density 16.81 g cm <sup>-3</sup> .
4	849	-	1966	1509-2111		Tzz; No. 2	88.9887	8.5	2.5	0.0095 C, 0.0012 H and 0.0006 O impurities; similar to the above specimen.

DATA TABLE NO. 299 THERMAL CONDUCTIVITY OF [TANTALUM + TUNGSTEN +  $\Sigma X_i$ ] ALLOYS(Ta + W < 99.50% or at least one  $X_i$  - 0.20%)[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup>K<sup>-1</sup>]

T k

CURVE 1

361.0	0.433
505.4	0.409
650.9	0.502
922.1	0.542
949.8	0.569
1072.1	0.537

CURVE 2

1172.1	0.588
1346.5	0.654
1580.4	0.732
1888.7	0.772
2063.7	0.832
2263.7	0.867
2427.5	0.890
5608.2	0.933
2802.6	0.938

CURVE 3

1660	0.618
1798	0.613
1990.5	0.611
2110.5	0.580
2231	0.584
2291.5	0.540
2390	0.553
2486.5	0.536
2598.5	0.545

CURVE 4

1509	0.650
1662	0.609
1716	0.621
1784	0.585
1842	0.581
1825	0.596
1933	0.587*
2022	0.581
2111	0.584*

\* Not shown on plot

SPECIFICATION TABLE NO. 300 THERMAL CONDUCTIVITY OF [TELLURIUM + ARSENIC +  $\Sigma X_i$ ] ALLOYS Te + As +  $\Sigma X_i$ (Te + As < 99.50% or at least one  $X_i > 0.20\%$ )

Curve No.	Ref. Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)			Composition (continued), Specifications and Remarks
						Te	As	Se	
1	1006 L	1963	313.2		S	40	30	30	52 mm dia x 2.5 mm thick; obtained from Servo Corp. of America; ground and polished; electrical resistivity $2.5 \times 10^6$ ohm cm at room temperature.
2	1006 L	1963	313.2		8	40	40	20	Similar to the above specimen except electrical resistivity $2.0 \times 10^6$ ohm cm at room temperature.

DATA TABLE NO. 300 THERMAL CONDUCTIVITY OF [TELLURIUM + ARSENIC +  $\Sigma X_i$ ] ALLOYS Te + As +  $\Sigma X_i$ (Te + As < 99.50% or at least one  $X_i > 0.20\%$ )[ Temperature, T, K; Thermal Conductivity, k, Watt  $\text{cm}^{-1} \text{K}^{-1}$  ]

T k

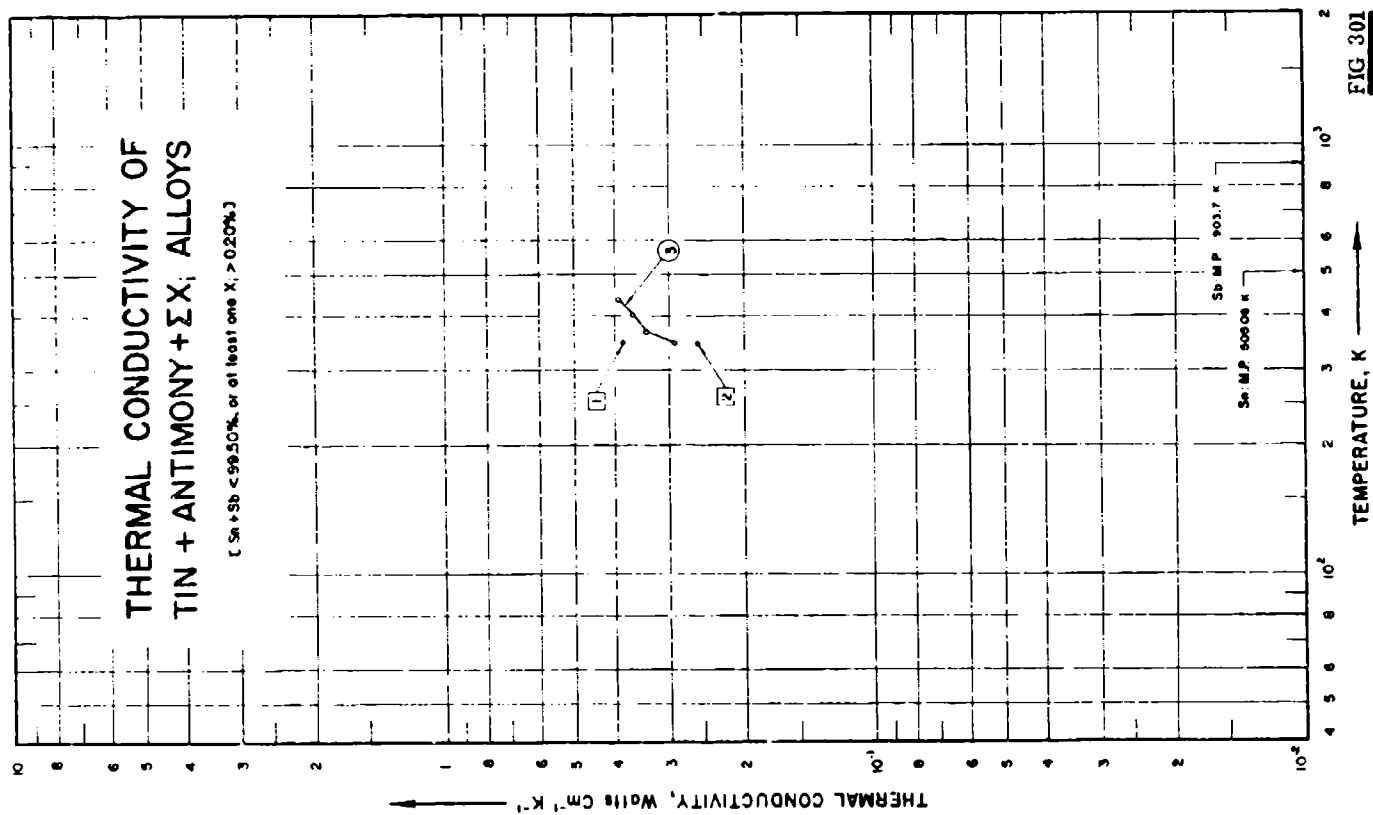
CURVE 1\*

313.2 0.00227

CURVE 2\*

313.2 0.00220

\* No graphical presentation



SPECIFICATION TABLE NO. 301 THERMAL CONDUCTIVITY OF (TiN + ANTIMONY + SX<sub>1</sub>) ALLOYS(Sn + Sb = 99.50% or at least one X<sub>1</sub> > 0.20%)

[For Data Reported in Figure and Table No. 300]

Curve No.	Ref. Method No. Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)			Remarks	
						Sn	Sb	Cu		
1	224	L	1923	348.2	SAE Bearing Alloy no. 10	92.49	4.74	3.58	0.19	
2	224	L	1923	348.2	SAE Bearing Alloy no. 11	86.92	7.96	5.16	0.12	
3	30	L	1925	350-441	White B.M.	87.8	7.73	4.0	0.12	0.14 Fe; specimen machined from a dry sand cast bar, (this alloy referred to as "white bearing metal").

DATA TABLE NO. 301 THERMAL CONDUCTIVITY OF [TIN + ANTIMONY +  $\Sigma X_i$ ] ALLOYS(Sn + Sb) > 99.50% or at least one  $X_i > 0.20\%$ [Temperature, T, K; Thermal Conductivity, k Watt  $\text{cm}^{-1}\text{K}^{-1}$ ]

T	k
<u>CURVE 1</u>	
348.2	0.385
<u>CURVE 2</u>	
348.2	0.259
<u>CURVE 3</u>	
348.7	0.293
370.7	0.339
402.7	0.364
440.7	0.393

SPECIFICATION TABLE NO. 302 THERMAL CONDUCTIVITY OF [TIN + COPPER +  $\Sigma X_i$ ] ALLOYS Sn + Cu +  $\Sigma X_i$   
 (Sn + Cu > 99.50% or at least one  $X_i$  > 0.20%)

Curve No.	Ref. Method No. Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) Sn Cu Ni	Composition (continued), Specifications and Remarks
1	215	L	1939	373, 473 ±2.0		51.39 37.86 9.94	0.85 Zn; cast.

DATA TABLE NO. 302 THERMAL CONDUCTIVITY OF [TIN + COPPER +  $\Sigma X_i$ ] ALLOYS Sn + Cu +  $\Sigma X_i$   
 (Sn + Cu < 99.50% or at least one  $X_i$  > 0.20%)

[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup> K<sup>-1</sup>]

T k

CURVE 1\*

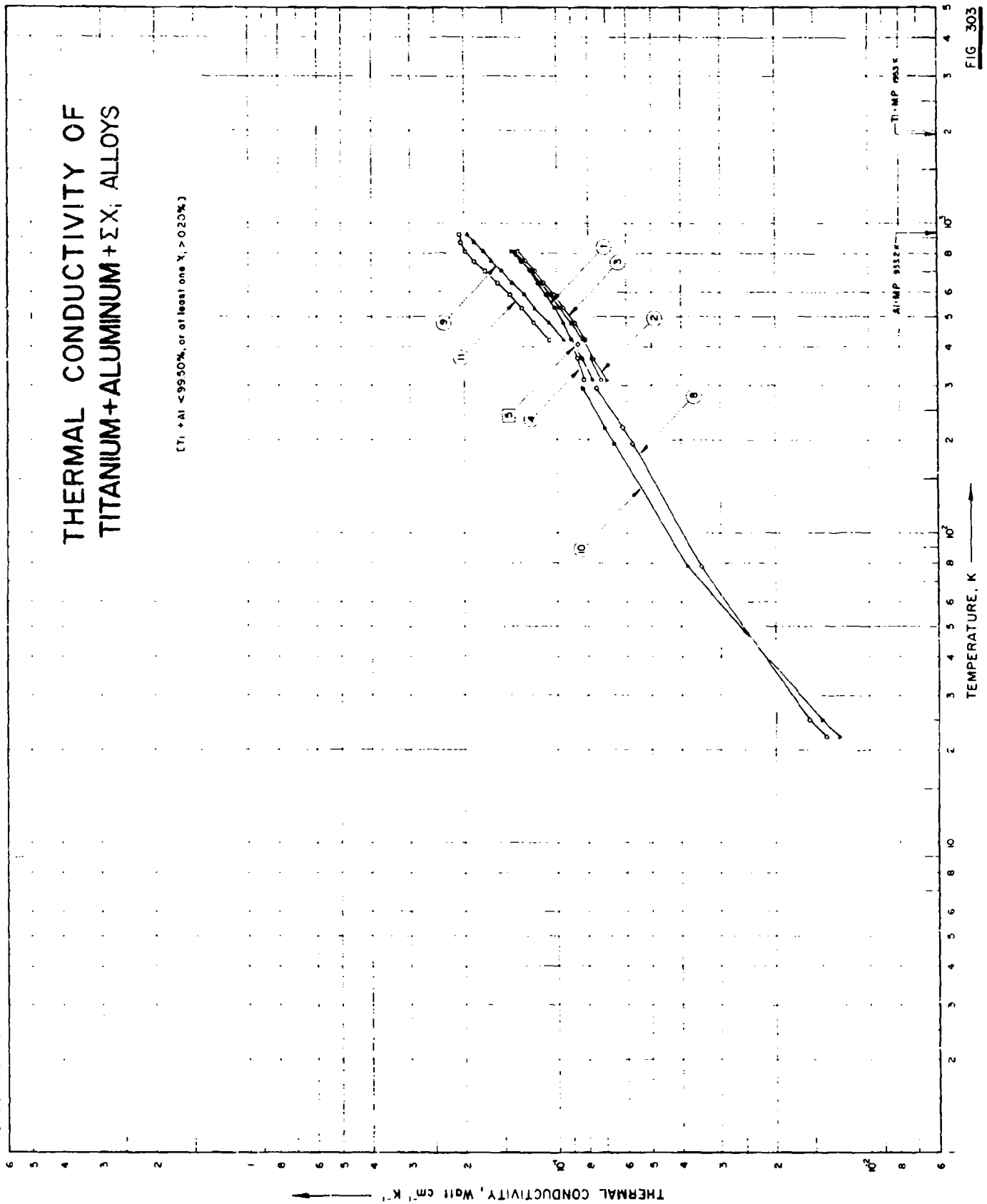
373.2 0.209  
 473.2 0.276

\* No graphical presentation

# THERMAL CONDUCTIVITY OF TITANIUM+ALUMINUM+ΣX<sub>i</sub> ALLOYS

(Ti + Al < 99.50%, or at least one X<sub>i</sub> > 0.20%)

FIGURE SHOWS ONLY 9 OF THE CURVES REPORTED IN TABLE





SPECIFICATION TABLE NO. 303 THERMAL CONDUCTIVITY OF [TITANIUM + ALUMINUM + EX<sub>1</sub>] ALLOYS(Ti + Al < 99.50% or at least one X<sub>1</sub> > 0.20%)

[For Data Reported in Figure and Table No. 303]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Ti	Al	V	Cr	Fe	Mo	Su	Mn	Composition (continued), Specifications and Remarks
1	231	C	1958	311-811	<5	A-110 AT	Bal.	5.0					2.5		Nominal composition; specimen in mill-annealed condition; lead used as comparative material.
2	231	C	1958	311-811	<5	C-130 AM	Bal.	4.0						4.0	Nominal composition; formerly designated as RC-130B; in mill-annealed condition.
3	231	C	1958	311-811	<5	Ti-6Al-4V	Bal.	6.0	4.0						Nominal composition; in mill-annealed condition.
4	231	C	1958	311-811	<5	Ti-155A	Bal.	5.0	1.4	1.5	1.2				Nominal composition; in mill-annealed condition.
5	555		1956	408.2			Bal.	7.0							0.5 Si.
6	831	L	1963	23-299	±5	6Al-4V	Bal.	5.89	3.87		0.15				0.02 C, 0.14 N, 0.005 H; the alloy produced by Nucor-Sharon Metals Corp. and heat-treated at 1199.8 K for 20 min and aged at 755 K for 4 hr and air cooled; measured under vacuum (<10 <sup>-6</sup> mm Hg).
7	831	L	1963	25-300	±5	4Al-3Mo-1V	Bal.	4.4	1.0	0.10	3.0				0.03 C, 0.011 N, 0.0057 H; the alloy produced by Crucible Steel Co.; heat-treated at 1175 K, aged at 769 K for 12 hrs and measured under vacuum (<10 <sup>-6</sup> mm Hg).
8	939	L	1962	22-294		6Al-1V	Bal.	5.89	3.87		0.15				0.2 C, 0.015 N; specimen 4 in. x 0.375 in. x 0.125 in.; supplied by Reactive Metals, Inc.; solution heat-treated at 1200 K for 20 min and aged at 755 K for 4 hrs; density 4.4 g cm <sup>-3</sup> .
9	939	L	1962	422-922		6Al-4V	Bal.	5.89	3.87		0.15				0.2 C, 0.015 N; specimen 10 in. x 0.5 in. x 0.125 in.; supplied by Reactive Metals Inc., Niles, Ohio; solution heat-treated at 1200 K for 20 min and aged at 755 K for 4 hrs; density 4.4 g cm <sup>-3</sup> .
10	939	L	1962	22-294		4Al-3Mo-1V	Bal.	1.0		0.10	3.0				0.03 C, 0.011 N; specimen 4 in. x 0.125 in.; supplied by Crucible Steel Co. of America, Pittsburgh, Penn.; solution heat-treated at 1175 K for 15-30 min and aged at 769 K for 12 hrs; density 4.51 g cm <sup>-3</sup> .

SPECIFICATION TABLE NO. 303 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)					Composition (continued), Specifications and Remarks	
							Ti	Al	V	Cr	Fe		Mo
11	939	L	1962	422-922		4Al-3Mo-IV	Bal.	4.4	1.0	0.10	3.0		0.03 C, 0.011 N; specimen 10 in. x 0.5 in. x 0.125 in.; supplied by Crucible Steel Co. of America, Pittsburgh, Penn.; solution heat-treated at 1175 K for 15-30 min and aged at 769 K for 12 hrs; density 4.51 g cm <sup>-3</sup> .



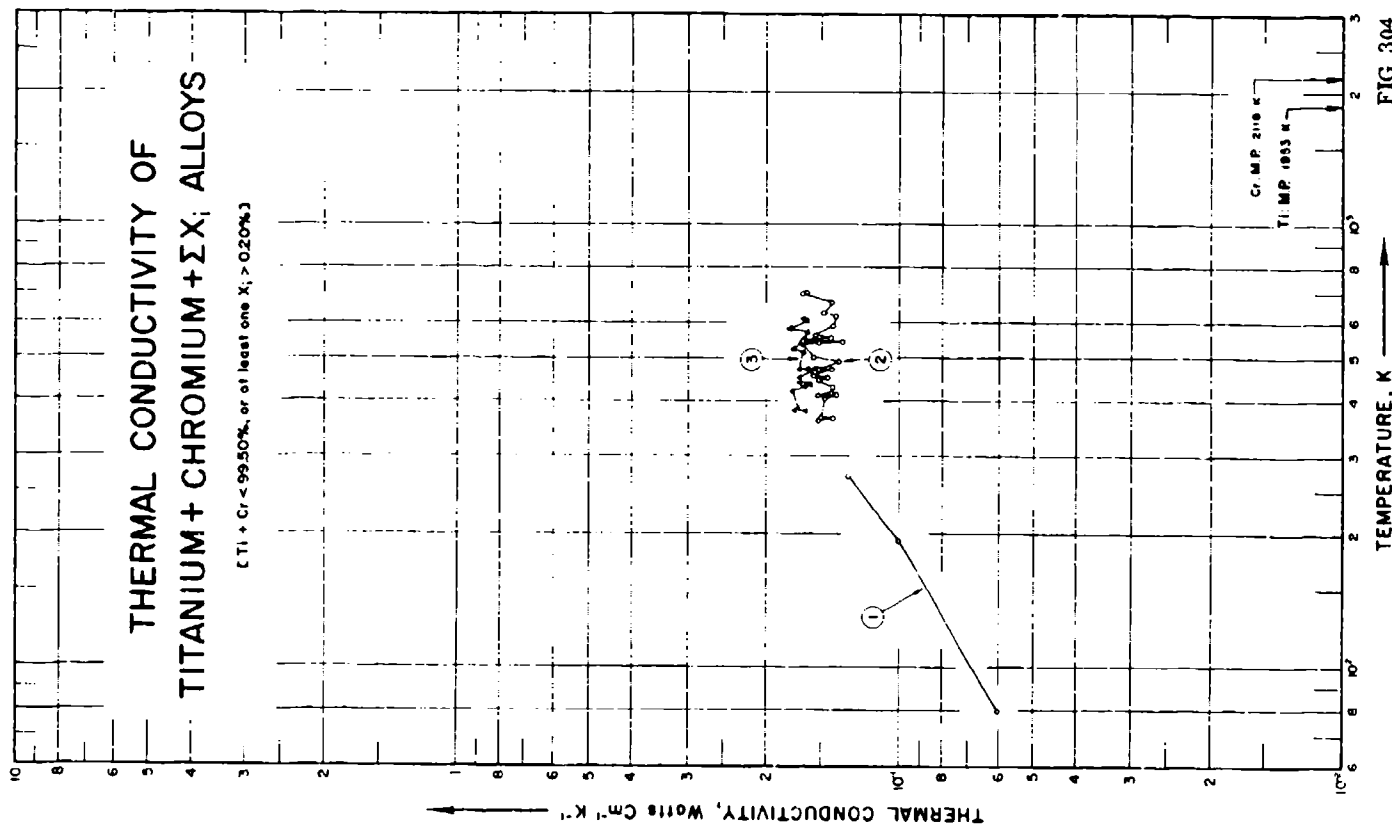


FIG 304

SPECIFICATION TABLE NO. 304 THERMAL CONDUCTIVITY OF [TITANIUM + CHROMIUM +  $\Sigma X_1$ ] ALLOYS(Ti + Cr < 99.50% or at least one  $X_1 > 0.20\%$ )

[For Data Reported in Figure and Table No. 304]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)				Composition (continued), Specifications and Remarks		
							Ti	Cr	Fe	O	Mo	C	
1	119, 718	L	1951	87-278	5-10	Ti 150A	96.2	2.8	1.3			0.02 N, and trace O ; specimen ~ 8 mm in dia and 72 mm long.	
2	340	L	1956	364-705	10	Ti 150A(2)	95.65	2.71	1.4	0.105	0.05	0.076 N, 0.0092 H; specimen 0.75 in. in dia; supplied by Watertown Arsenal.	
3	340	L	1956	382-615	10	Cr-Mo	96.3	3.38	0.13	0.131	2.10	0.02	0.032 N, 0.0077 H; specimen 0.75 in. in dia; supplied by Watertown Arsenal.

DATA TABLE NO. 304 THERMAL CONDUCTIVITY OF [TITANIUM + CHROMIUM + EX.] ALLOYS  
 (Ti + Cr < 99.50% or at least one X<sub>i</sub> > 0.20%)

[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup>K<sup>-1</sup>]

T	k	T	k
<u>CURVE 1</u>			
87	0.062	438.7	0.159
194	0.103	440.7	0.168
278	0.131	452.6	0.168
		477.6	0.156
		477.6	0.161
<u>CURVE 2</u>			
363.7	0.152	477.6	0.167
369.3	0.141	518.4	0.165
370.7	0.150	526.8	0.174
406.8	0.147	546.2	0.166
412.3	0.138	552.6	0.164
414.6	0.152	574.0	0.161
419.3	0.142	588.7	0.176
433.2	0.141	610.1	0.161
447.1	0.152	514.6	0.165
452.6	0.145		
459.6	0.156		
474.6	0.141		
474.8	0.152		
492.9	0.136		
503.4	0.156		
539.6	0.165		
542.9	0.152		
549.0	0.134		
549.0	0.165		
560.1	0.143		
568.4	0.155		
596.2	0.141		
625.7	0.138		
634.6	0.148		
672.1	0.141		
700.7	0.165		
704.6	0.161		
<u>CURVE 3</u>			
381.8	0.163		
383.2	0.173		
387.3	0.168		
426.2	0.175		
434.6	0.164		

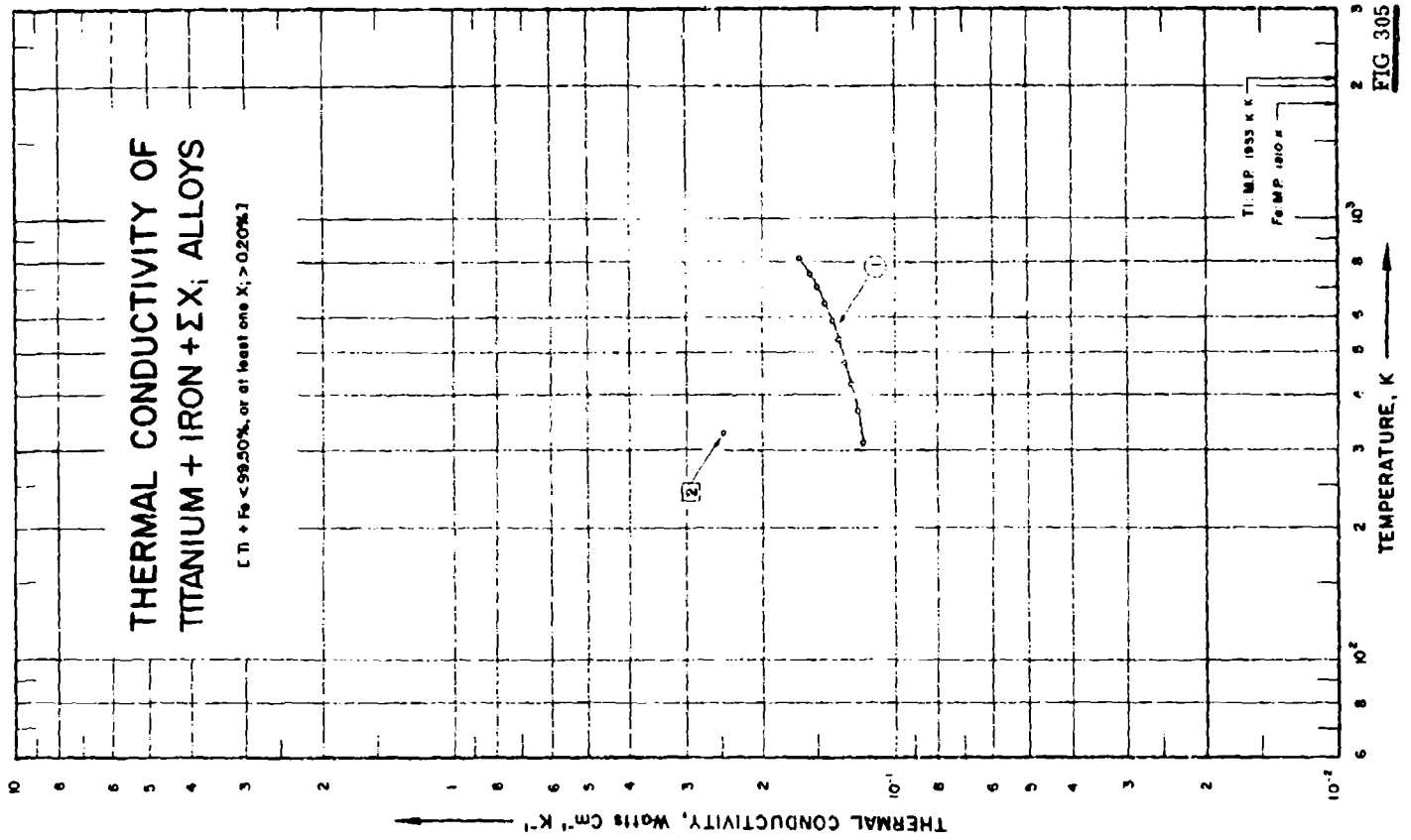


FIG. 305

SPECIFICATION TABLE NO. 305 THERMAL CONDUCTIVITY OF [TITANIUM + IRON +  $\Sigma X_i$ ] ALLOYS(Ti + Fe < 99.50% or at least one  $X_i > 0.20\%$ )

[For Data Reported in Figure and Table No. 305]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)					Composition (continued), Specifications and Remarks
							Ti	Fe	C	Cr	Mo	
1	231	C	1958	311-811	< 5	Ti-140A	53.7	2.2	2.1	2.0		Specimen in a mill-annealed condition; measured in vacuum of $\sim 2 \times 10^{-5}$ m Hg; electrical resistivity 79, 86, 95, 103, 111, 119, 125, 132, 138 and 143 $\mu\text{ohm cm}$ at 311, 366, 422, 477, 533, 589, 644, 700, 755 and 811 K respectively. Lead used as comparative material.
2	204	L	1937	327.4		Russian ferrocarbotitanium	45	34	13.5		7.5	Average composition of analysis.



DATA TABLE NO. 305 THERMAL CONDUCTIVITY OF [TITANIUM + IRON +  $\Sigma X_i$ ] ALLOYS(Ti + Fe < 99.50% or at least one  $X_i$  - 0.20%)[Temperature, T, K; Thermal Conductivity, k, Watts  $\text{cm}^{-1}\text{K}^{-1}$ ]

T	k
<u>CURVE 1</u>	
311.00	0.119
366.00	0.123
422.00	0.126
477.00	0.130
533.00	0.134
589.00	0.139
644.00	0.145
700.00	0.151
755.00	0.158
811.00	0.166
<u>CURVE 2</u>	
327.4	0.248

# THERMAL CONDUCTIVITY OF TITANIUM + MANGANESE + $\Sigma X_i$ ALLOYS

[Ti + Mn < 99.50%, or at least one  $X_i$  > 0.20%]

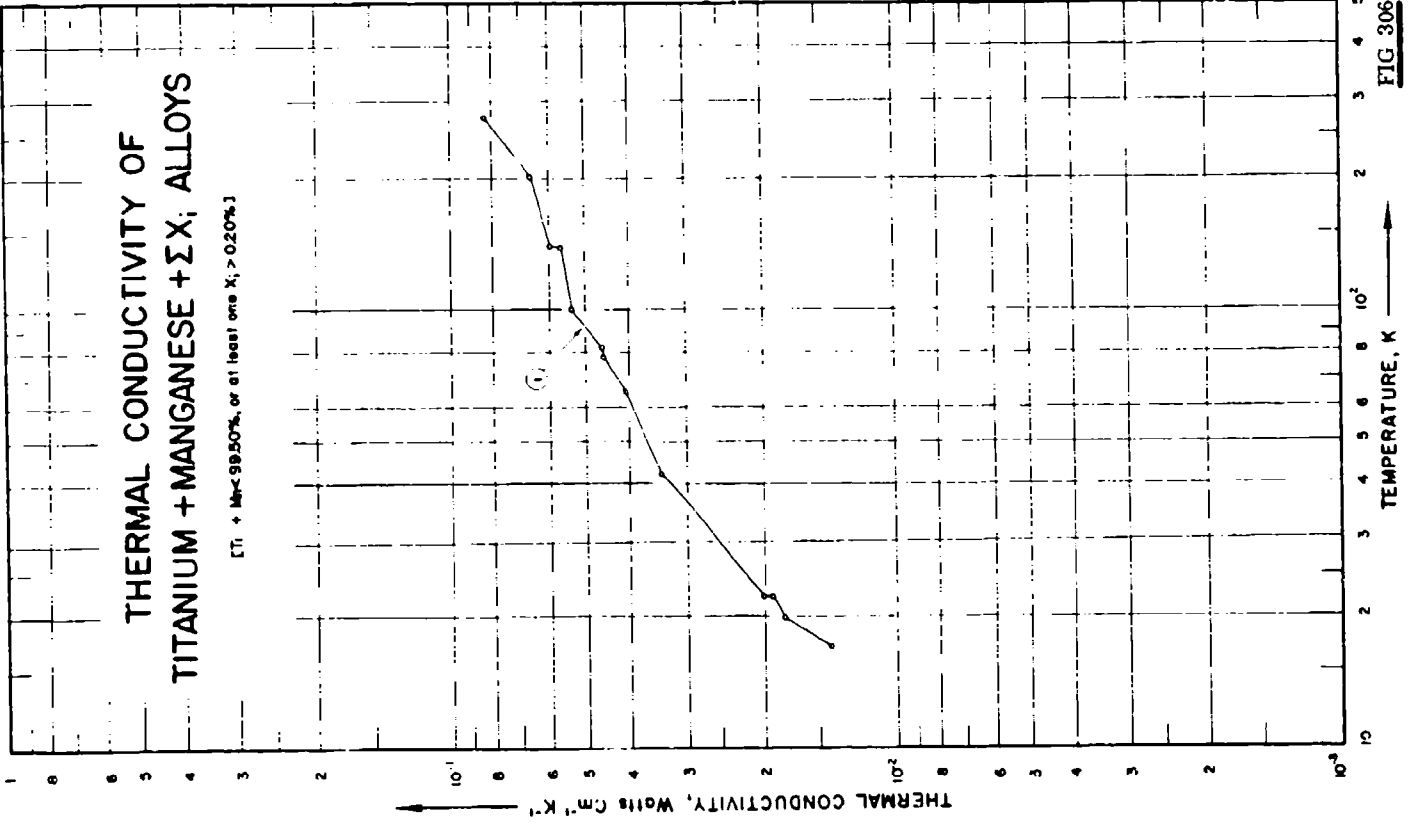


FIG 306

SPECIFICATION TABLE NO. 306 THERMAL CONDUCTIVITY OF [TITANIUM + MANGANESE +  $\Sigma X_i$ ] ALLOYS(Ti + Mn)  $\leq$  99.50% or at least one  $X_i > 0.20\%$ 

[ For Data Reported in Figure and Table No. 306 ]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)			Composition (continued), Specifications and Remarks	
							Ti	Mn	C		
1	159	L	1953	17-273	10.0	RC-1304	91.17	4.7	3.99	0.14	

DATA TABLE NO. 306 THERMAL CONDUCTIVITY OF TITANIUM - MANGANESE -  $\text{EX}_1$  ALLOYS  
 (Ti - Mn - 99.50% or at least one  $X_1 > 0.05\%$ )

[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup>K<sup>-1</sup>]

T	k	CURVE 1
17.15	0.0142	
19.91	0.0190	
23.35	0.0193	
23.40	0.0201	
42.37	0.0339	
65.25	0.0406	
74.76	0.0456	
82.59	0.0460	
100.60	0.0536	
141.20	0.0561	
141.30	0.0595	
204.10	0.0651	
278.00	0.0537	

FIGURE SHOWS ONLY 4 OF THE CURVES REPORTED IN TABLE

### THERMAL CONDUCTIVITY OF TITANIUM + VANADIUM + X, ALLOYS

(Ti + V < 99.50%, or at least one X, > 0.20%)

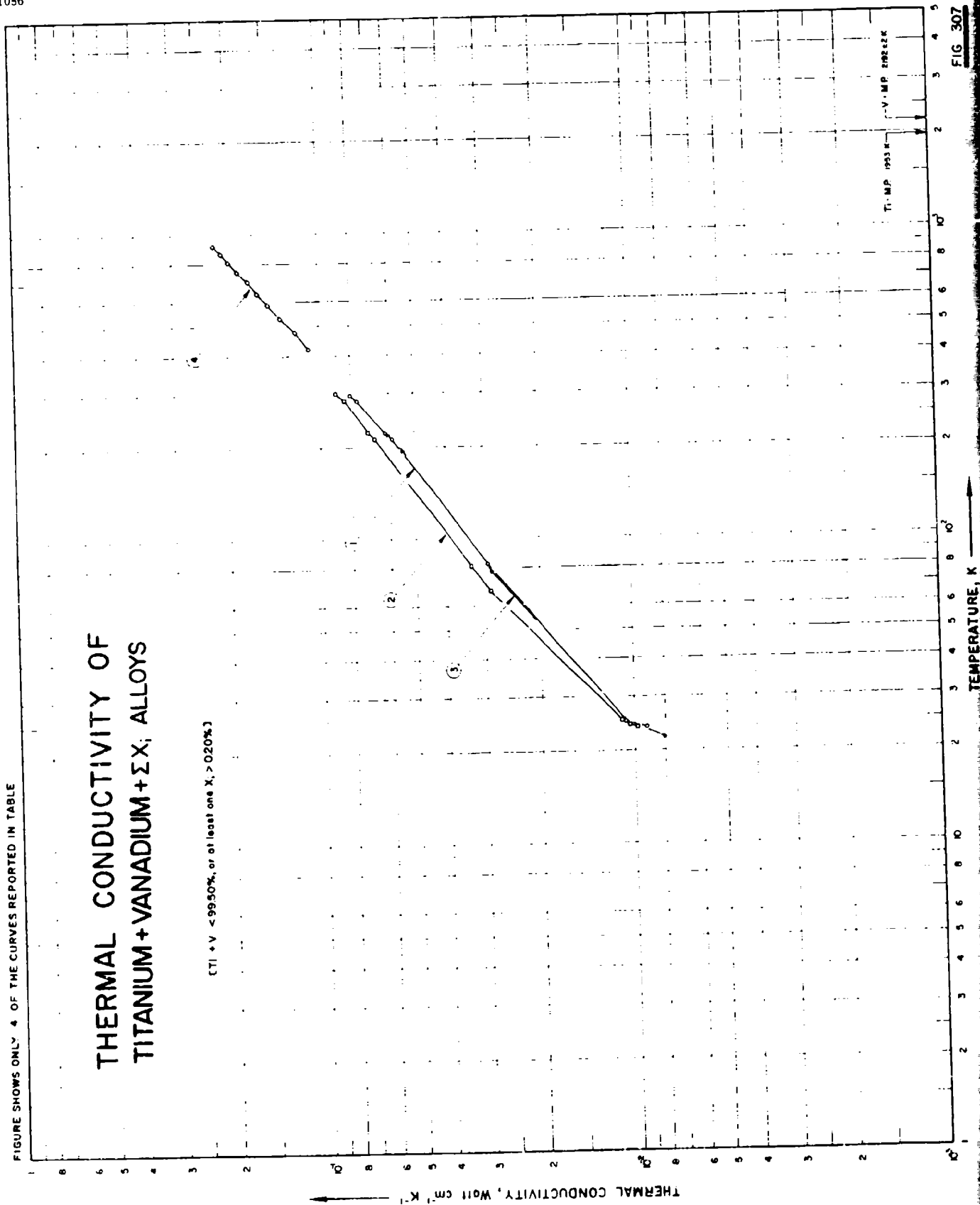


FIG 307

SPECIFICATION TABLE NO. 307 TITANIUM + VANADIUM + EX<sub>1</sub> ALLOYS(Ti + V + EX<sub>1</sub> < 99.50% or at least one X<sub>i</sub> > 0.20%)

[ For Data Reported in Figure and Table No. 307 ]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)						Composition (continued), Specifications and Remarks		
							Ti	V	Al	Cr	Fe	C		N <sub>2</sub>	H
1	831	L	1963	24-297	±5	13V-11Cr-3Al	Bal	13.9	3.5	10.4	0.25	0.04	0.025	0.0114	Solution heat-treated at 1061 K for 20 min, air-cooled; aged at 755.4 K for 60 hrs, air-cooled; measurements done in high vacuum (<10 <sup>-5</sup> mm Hg); specimen produced by Crucible Steel Co.
2	831	L	1963	24-301	+5	2.5Al-16V	Bal	14.95	2.75		0.21	0.03	0.015	0.0066	Solution heat-treated at 1038.7 K for 30 min; aged at 805.4 K for 4 hrs; measurements done in high vacuum (<10 <sup>-5</sup> mm Hg); specimen produced by Mallory-Sharon Metals Corporation.
3	939	L	1962	22-294		120VCA		13.9	3.5	10.4	0.25	0.04	0.025		Specimen 4 x 0.375 x 0.125 in.; supplied by Crucible Steel Co. of America; solution heat-treated at 1061 K for 20 min and aged at 755 K for 60 hrs; density 4.62 g cm <sup>-3</sup> .
4	939	L	1962	422-922		120VCA		13.9	3.5	10.4	0.25	0.04	0.025		Specimen 10 x 0.5 x 0.125 in.; supplied by Crucible Steel Co. of America; solution heat-treated at 1061 K for 20 min and aged at 755 K for 60 hrs; density 4.62 g cm <sup>-3</sup> .
5	939	L	1962	22-294		2.5Al-16V		14.95	2.75		0.21	0.3	0.015		Specimen 4 x 0.375 x 0.125 in.; supplied by Reactive Metals, Inc.; solution heat-treated at 1039 K for 30 min and aged at 805 K for 4 hrs; density 4.65 g cm <sup>-3</sup> .
6	939	L	1962	422-922		2.5Al-16V		14.95	2.75		0.21	0.3	0.015		Specimen 10 x 0.5 x 0.125 in.; supplied by Reactive Metals, Inc.; solution heat-treated at 1039 K for 30 min and aged at 805 K for 4 hrs; density 4.65 g cm <sup>-3</sup> .

DATA TABLE NO. 307 THERMAL CONDUCTIVITY OF [TITANIUM + VANADIUM +  $\Sigma X_i$ ] ALLOYS(Ti + V) 99.50% or at least one  $X_i > 0.20\%$ [Temperature, T, K; Thermal Conductivity, k, Watt  $\text{cm}^{-1} \text{K}^{-1}$ ]

T	k	T	k
<u>CURVE 1</u>			
21.89	0.0092	422	0.109
23.91	0.0092*	478	0.121
24.96	0.0108	533	0.135
24.96	0.0109*	589	0.147
25.34	0.0109	644	0.159
25.37	0.0111*	700	0.171
62.06	0.0294	755	0.185
212.78	0.0590	811	0.197
223.90	0.0623	867	0.209
283.96	0.0767	922	0.221
296.65	0.0802		
<u>CURVE 2</u>			
23.90	0.0098	22	0.0085
23.92	0.0099*	25	0.0107
24.38	0.0104	78	0.0320
24.41	0.0104*	194	0.0633
25.02	0.0109*	219	0.0696
25.08	0.0110	294	0.0869
25.37	0.0108*		
25.40	0.0108*	<u>CURVE 5*</u>	
67.33	0.0281*	422	0.109
81.59	0.0327*	478	0.121
81.65	0.0332	533	0.135
212.70	0.0673	589	0.147
224.92	0.0707	644	0.159
284.61	0.0839	700	0.171
301.12	0.0889*	755	0.185
301.13	0.0891	811	0.197
		867	0.209
		922	0.221
<u>CURVE 3</u>			
22	0.0080		
25	0.0107*		
78	0.0236		
194	0.0545		
219	0.0606		
294	0.0796*		

\* Not shown on plot.

SPECIFICATION TABLE NO. 308 THERMAL CONDUCTIVITY OF [TITANIUM +  $\Sigma X_1$ ] ALLOYS T1 +  $\Sigma X_1$

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)	Specifications and Remarks
1	340	L	1956	418-927	10.0	T1 150A(1)	Composition unknown.	

DATA TABLE NO. 308 THERMAL CONDUCTIVITY OF [TITANIUM +  $\Sigma X_1$ ] ALLOYS T1 +  $\Sigma X_1$

[Temperature, T, K; Thermal Conductivity, k, Watt  $\text{cm}^{-1} \text{K}^{-1}$ ]

T	k	CURVE 1*	
		T	k
418.44	0.166	815.7	0.163
421.2	0.169	917.9	0.169
493.4	0.171	926.8	0.174
497.1	0.169		
516.5	0.161		
524.8	0.163		
535.9	0.177		
610.1	0.164		
621.2	0.168		
653.4	0.156		
663.7	0.162		
731.8	0.170		
746.2	0.175		
801.8	0.166		

\* No graphical presentation



SPECIFICATION TABLE NO. 309 THERMAL CONDUCTIVITY OF [TUNGSTEN + IRON +  $\Sigma X_i$ ] ALLOYS  $W + Fe + \Sigma X_i$   
 (W + Fe < 99.50% or at least one  $X_i > 0.20\%$ )

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)			Composition: (continued), Specifications and Remarks
							W	Fe	C	
1	204	L	1937	340.7		Ferrotungsten, 32	80.5	12.02	0.48	

DATA TABLE NO. 309 THERMAL CONDUCTIVITY OF [TUNGSTEN + IRON +  $\Sigma X_i$ ] ALLOYS  $W + Fe + \Sigma X_i$

(W + Fe < 99.50% or at least one  $X_i > 0.20\%$ )

[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup> K<sup>-1</sup>]

T k  
 CURVE 1  
 340.7 0.468

\* No graphical presentation

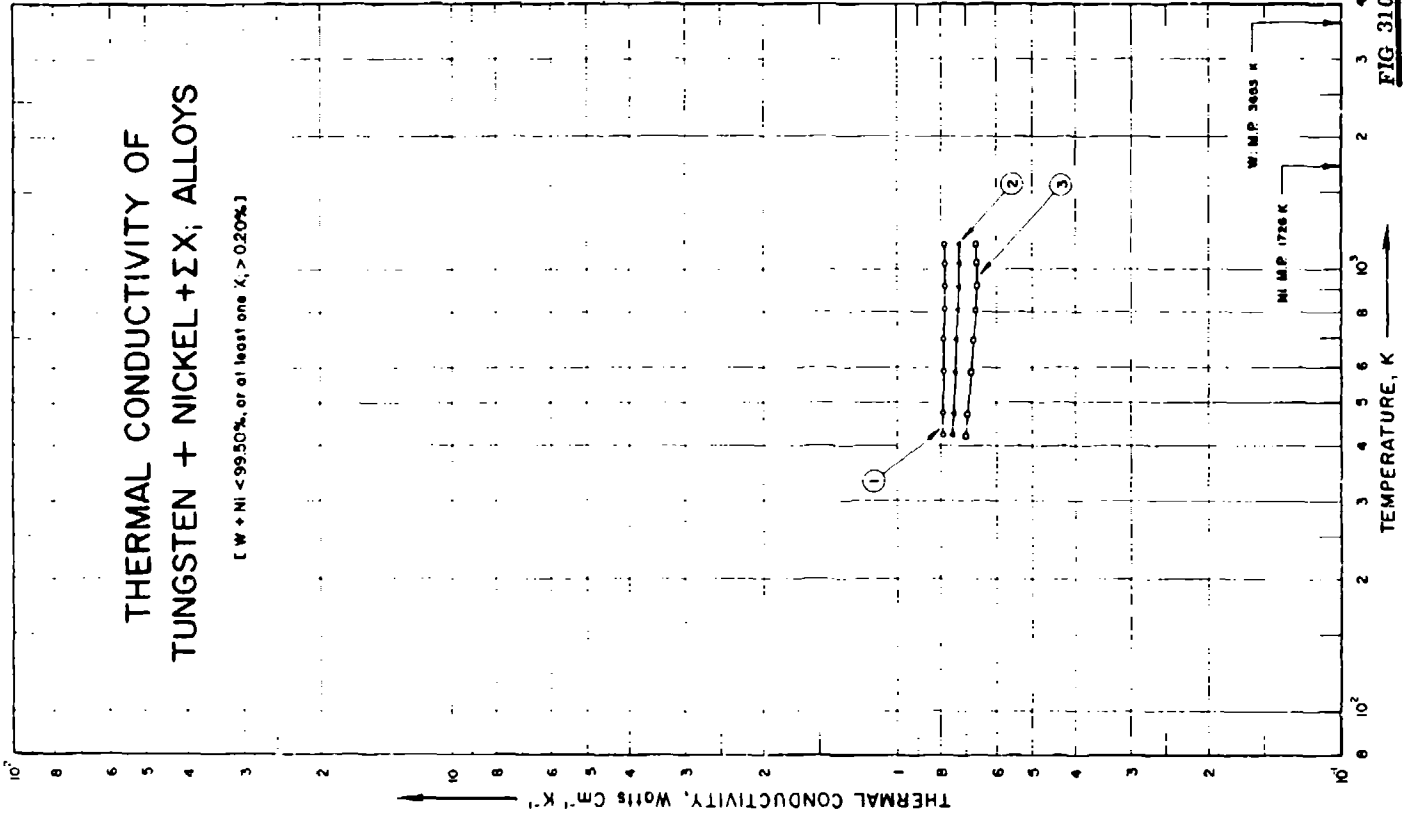


FIG. 310

SPECIFICATION TABLE NO. 310 THERMAL CONDUCTIVITY OF [TUNGSTEN + NICKEL +  $\Sigma X_i$ ] ALLOYS(W + Ni < 99.50% or at least one  $X_i > 0.20\%$ )

[ For Data Reported in Figure and Table No. 310 ]

Curve No.	Ref. Method No. Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)				Composition (continued), Specifications and Remarks
						W	Ni	Cu	B	
1	595	1961	422-1144			90	6	4		
2	595	1961	422-1144			90.4	6.3	2.0	0.15	
3	595	1961	422-1144			90.4	6.3	2.0	0.15	

DATA TABLE NO. 310 THERMAL CONDUCTIVITY OF [TUNGSTEN + NICKEL +  $\Sigma X_i$ ] ALLOYS  
 (W + Ni < 99.50% or at least one  $X_i > 0.20\%$ )

[Temperature, T, K; Thermal Conductivity,  $\lambda$ , Watt cm<sup>-1</sup>k<sup>-1</sup>]

T k

CURVE 1

422.1	0.790
477.6	0.790
586.7	0.790
699.8	0.788
810.9	0.787
922.1	0.786
1033.2	0.785
1144.3	0.784

CURVE 2

422.1	0.753
477.6	0.748
586.7	0.741
699.8	0.738
810.9	0.732
922.1	0.729
1033.2	0.728
1144.3	0.729

CURVE 3

422.1	0.701
477.6	0.697
586.7	0.684
699.8	0.675
810.9	0.670
922.1	0.667
1033.2	0.665
1144.3	0.665

FIGURE SHOWS ONLY 6 OF THE CURVES REPORTED IN TABLE

# THERMAL CONDUCTIVITY OF URANIUM + MOLYBDENUM + ΣX; ALLOYS

[ U + Mo < 99.50%, or at least one X; > 0.20% ]

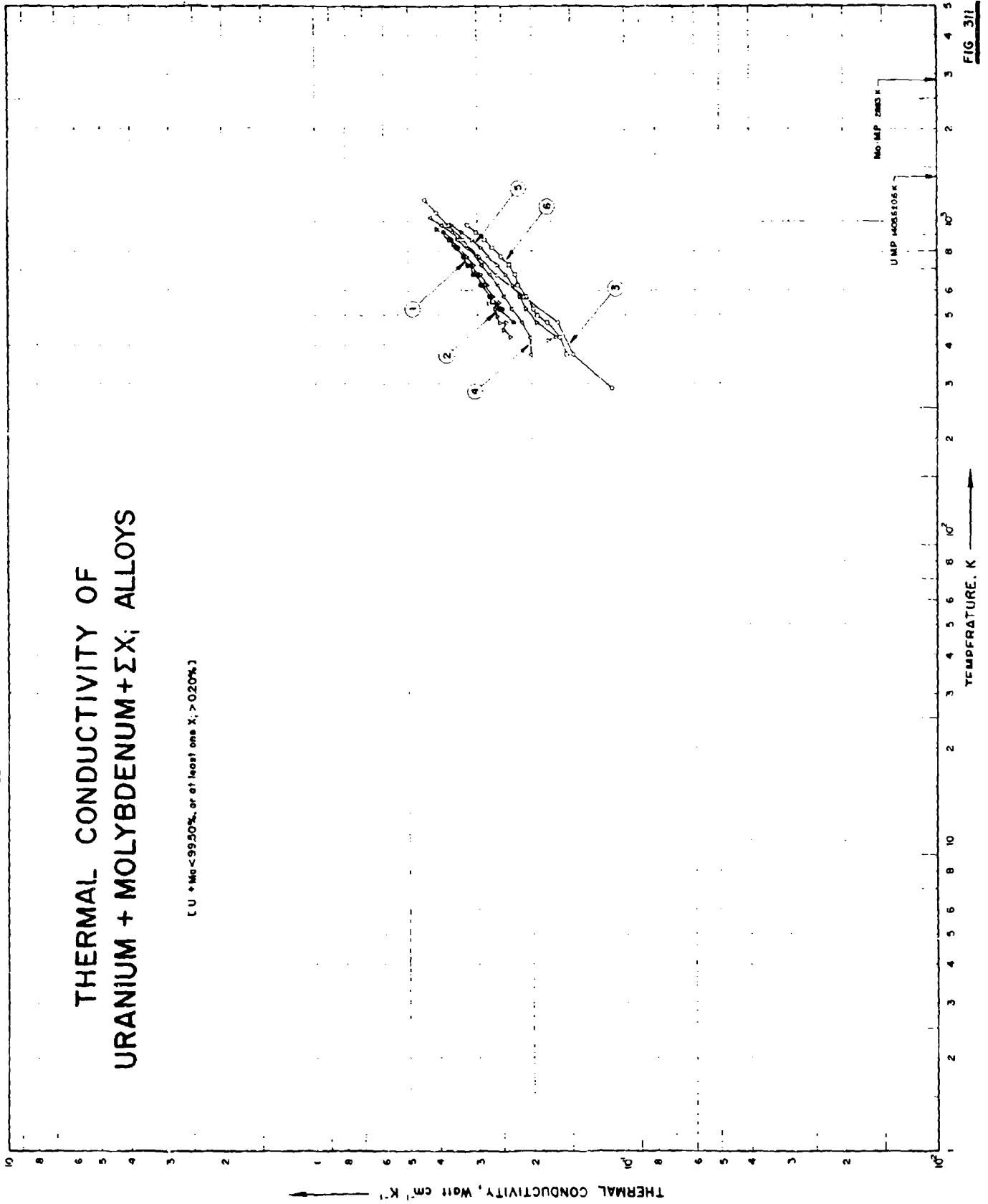


FIG 311

SPECIFICATION TABLE NO. 311 THERMAL CONDUCTIVITY OF [URANIUM + MOLYBDENUM +  $\Sigma X_i$ ] ALLOYS

(U + Mo < 99.50% or at least one  $X_i > 0.20\%$ )

[For Data Reported in Figure and Table No. 311]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)						Composition (continued), Specifications and Remarks
							U	Mo	Ru	Rh	Pd	Zn	
1	217	C	1959	473-923	4	U-3% FS alloy	96.974	1.5	1.2	0.16	0.12	0.04	0.006 Nb; specimen 15 cm long and 2 cm in dia; melt in vacuum and cast into a water-cooled copper mold; Inconel used as comparative material.
2	217	C	1959	423-948	4	U-3% FS alloy	96.974	1.5	1.2	0.16	0.12	0.04	0.006 Nb; the above specimen used Armco iron as comparative material.
3	538	L	1956	293-1173	±5	Fissionium-type alloy	94.95	2.56	1.55	0.30	0.54	0.10	Melt and cast; forged at 870 C, rolled from a helium atmosphere furnace at 675 C then machined; measured in vacuum of $\sim 1 \times 10^{-6}$ mm Hg.
4	421	C	1961	373-1023	4	U-5% FS alloy, casting No. 870	95.676	2.06	1.86	0.195	0.136	0.063	0.01 Nb; cast; Armco iron used as comparative material.
5	421	C	1961	413-973	4	U-8% FS alloy, casting No. 743	92.342	3.73	3.12	0.415	0.28	0.096	0.017 Nb; cast; Armco iron used as comparative material.
6	421	C	1961	373-973	4	U-10% FS alloy, casting No. 744	90.334	4.63	4.00	0.54	0.358	0.118	0.02 Nb; cast; Armco iron used as comparative material.



SPECIFICATION TABLE NO. 312 THERMAL CONDUCTIVITY OF [URANIUM + ZIRCONIUM +  $\Sigma X_i$ ] ALLOYS U + Zr +  $\Sigma X_i$ (U + Zr < 99.50% or at least one  $X_i > 0.20\%$ )

Curve No.	Ref. Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) Zr Mo Nb	Correlation (continued), Specifications and Remarks
1	421 C	1961	423-973		U-5 W/O Fs alloy; casting No. 896	2.54 1.85 0.01	0.135 Pd, 0.189 Rh, 1.76 Ru; specimen 2 cm in dia and 15 cm long; machined from the cast ingot prepared by vacuum induction melting of binary alloys of uranium with fissile elements in the form of buttons (prepared by arc melting) in thorium-coated graphite crucible, and then bottom pouring into a water-cooled copper mold at approx 1200 C; measured in vacuum; Armco iron used as comparative material.

DATA TABLE NO. 312 THERMAL CONDUCTIVITY OF [URANIUM + ZIRCONIUM +  $\Sigma X_i$ ] ALLOYS U + Zr +  $\Sigma X_i$ (U + Zr < 99.50% or at least one  $X_i > 0.20\%$ )

T	k	CURVE, I*
423.2	0.211	
448.2	0.226	
473.2	0.228	
523.2	0.243	
573.2	0.253	
623.2	0.265	
673.2	0.279	
698.2	0.289	
723.2	0.289	
773.2	0.303	
798.2	0.326	
823.2	0.315	
873.2	0.332	
923.2	0.353	
948.2	0.379	
973.2	0.377	

\* No graphical presentation



SPECIFICATION TABLE NO. 313 THERMAL CONDUCTIVITY OF [ZINC + ALUMINUM +  $\Sigma X_i$ ] ALLOYS Zn + Al +  $\Sigma X_i$ (Zn + Al < 99.50% or at least one  $X_i$  > 0.20%)

Curve No.	Ref. Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)			Composition (continued), Specifications and Remarks
						Zn	Al	Cu	
1	L	1958	293-353	±1.0	Zamak Nr410	95.18	3.90	0.87	0.008 Fe, 0.042 Mg, 0.024 Si, trace Bi, Cd, Pb, Sn and Ti.
2	L	1958	293-353	±1.0	Zamak Nr430	92.33	4.88	2.71	0.010 Fe, 0.043 Mg, 0.022 Si, trace Bi, Cd, Pb, Sn and Ti.

DATA TABLE NO. 313 THERMAL CONDUCTIVITY OF [ZINC + ALUMINUM +  $\Sigma X_i$ ] ALLOYS Zn + Al +  $\Sigma X_i$ (Zn + Al < 99.50% or at least one  $X_i$  > 0.20%)[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup> K<sup>-1</sup>]

T	k
<u>CURVE 1*</u>	
293.2	1.13
323.2	1.14
353.2	1.15
<u>CURVE 2*</u>	
293.2	1.12
323.2	1.13
353.2	1.14

No graphical presentation

SPECIFICATION TABLE NO. 314 THERMAL CONDUCTIVITY OF ZINC + LEAD +  $\Sigma X_i$  ALLOYS Zn + Pb +  $\Sigma X_i$   
 (Zn + Pb > 99.50% or at least one  $X_i > 0.20\%$ )

Curve No.	Ref. Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)			Composition (continued), Specifications and Remarks
						Zn	Pb	Cu	
1	E	1900	291, 373		Commercial zinc; Zn 1	98.6	1.1	0.03	0.25 Cu, 0.03 Fe.

DATA TABLE NO. 314 THERMAL CONDUCTIVITY OF [ZINC + LEAD +  $\Sigma X_i$ ] ALLOYS Zn + Pb +  $\Sigma X_i$   
 (Zn + Pb < 99.50% or at least one  $X_i > 0.20\%$ )

[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup> K<sup>-1</sup>]

T	k
<u>CURVE 1*</u>	
291.2	1.115
373.2	1.108

\* No graphical presentation

SPECIFICATION TABLE NO. 315 THERMAL CONDUCTIVITY OF [ZIRCONIUM + ALUMINUM +  $\Sigma X_i$ ] ALLOYS Zr + Al +  $\Sigma X_i$   
 (Zr + Al = 99.50% or at least one  $X_i > 0.20\%$ )

Curve No.	Ref. Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) Al Mo Sn	Composition (continued), Specifications and Remarks
1	956 R	1960	747-867			1.5 1.5 1.5	Prepared from reactor-grade Zr sponge by double-arc melting; cast into 6 in. dia, 35 lb ingots, and hard-rolled to 0.125 in. strip then vacuum annealed at 788 C for 24 hrs.
2	956 R	1960	696-871			3.0 1.5	Similar to the above specimen.
3	956 R	1960	724-869			3.0 3.0	Similar to the above specimen.

DATA TABLE NO. 315 THERMAL CONDUCTIVITY OF [ZIRCONIUM + ALUMINUM +  $\Sigma X_i$ ] ALLOYS Zr + Al +  $\Sigma X_i$

(Zr + Al = 99.50% or at least one  $X_i > 0.20\%$ )

[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup> K<sup>-1</sup>]

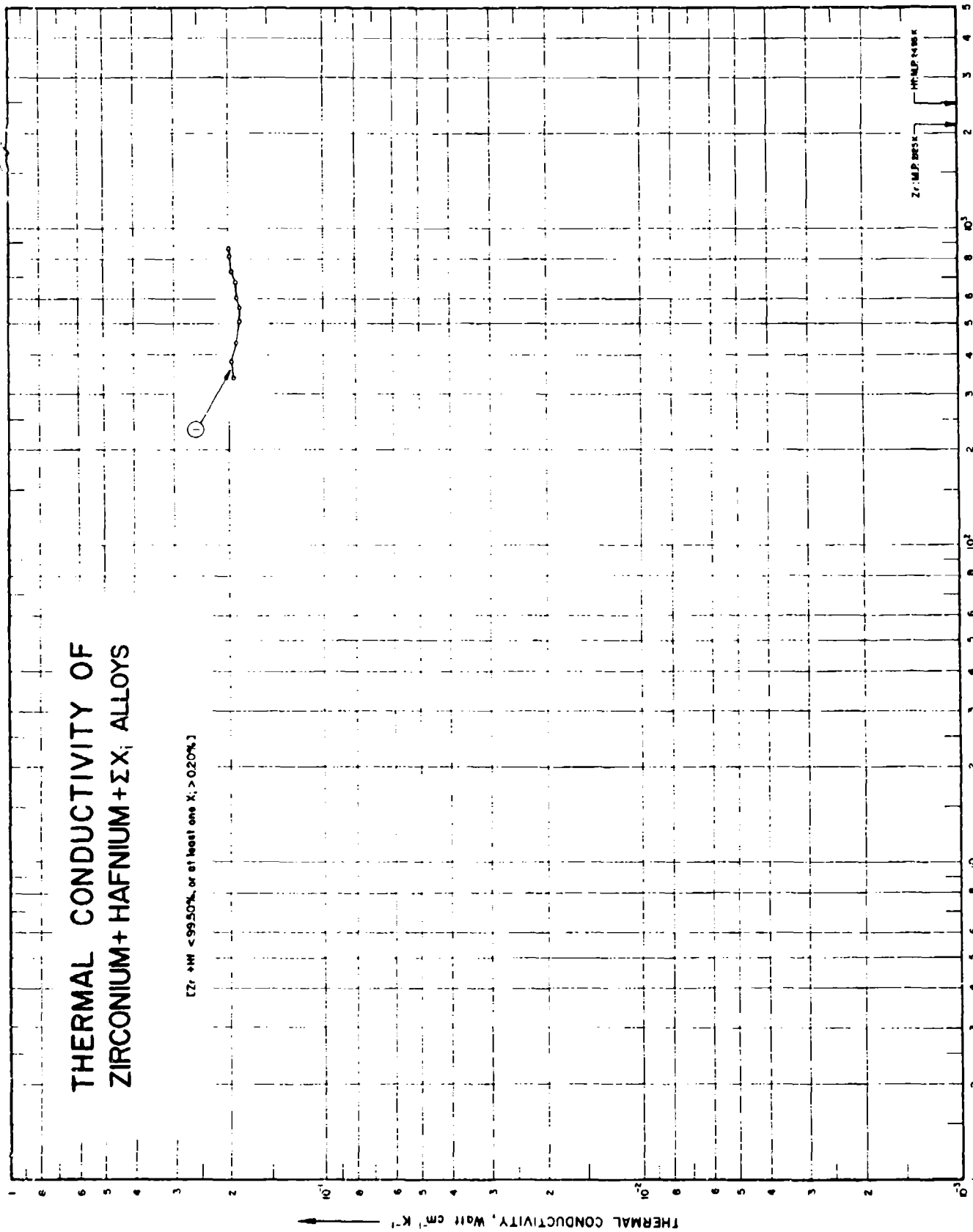
T	k	T	k
<u>CURVE 1*</u>			
747	0.166	724	0.140
761	0.168	760	0.144
808	0.171	792	0.147
868	0.185	825	0.151
<u>CURVE 2*</u>			
696	0.125	869	0.159
740	0.128		
820	0.137		
871	0.144		

\* No graphical presentation

FIG. 316

# THERMAL CONDUCTIVITY OF ZIRCONIUM + HAFNIUM + $\Sigma X_i$ ALLOYS

[Zr + Hf < 99.50%, or at least one  $X_i$  > 0.20%]



SPECIFICATION TABLE NO. 316 THERMAL CONDUCTIVITY OF [ZIRCONIUM + HAFNIUM +  $\Sigma X_i$ ] ALLOYS(Zr + Hf < 99.50% or at least one  $X_i > 0.20\%$ )

[For Data Reported in Figure and Table No. 316]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)		Composition (continued), Specifications and Remarks	
							Zr	Hf		
1	441	L	1957	340-863			98.73	0.97	0.3	Electrical resistivity 62.1, 69.4, 77.5, 89.3, 97.1, 101.0, 108.6, 113.6, 123.4, and 128.2 $\mu$ ohm cm at 66.3, 110.5, 165.1, 237.3, 292.8, 336.3, 403.0, 457.6, 546.0, and 589.5 C, respectively.

DATA TABLE NO. 316 THERMAL CONDUCTIVITY OF [ZIRCONIUM + HAFNIUM +  $\Sigma X_i$ ] ALLOYS(Zr + Hf < 99.50% or at least one  $X_i > 0.20\%$ )[Temperature, T, K; Thermal Conductivity, k, Watt  $\text{cm}^{-1} \text{K}^{-1}$ ]

T	k
<u>CURVE 1</u>	
339.5	0.194
383.7	0.196
438.3	0.189
510.5	0.185
566.0	0.185
609.5	0.189
676.2	0.190
730.8	0.196
819.2	0.199
862.7	0.200

SPECIFICATION TABLE NO. 317 THERMAL CONDUCTIVITY OF [ZIRCONIUM + MOLYBDENUM +  $\Sigma X_i$ ] ALLOYS Zr + Mo +  $\Sigma X_i$   
 (Zr + Mo > 99.50% or at least one  $X_i$  > 0.20%)

Curve No.	Ref. Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)			Composition (continued), Specifications and Remarks
						Mo	Al	Sn	
1	956	R	1960	747-868		1.5	1.5	1.5	Prepared from reactor-grade Zr sponge by double-arc melting; cast into 6 in. dia., 35 lb ingots; hard-rolled to 0.125 in. strip then vacuum annealed at 1061 K for 24 hrs.

DATA TABLE NO. 317 THERMAL CONDUCTIVITY OF [ZIRCONIUM + MOLYBDENUM +  $\Sigma X_i$ ] ALLOYS Zr + Mo +  $\Sigma X_i$

(Zr + Mo < 99.50% or at least one  $X_i$  > 0.20%)

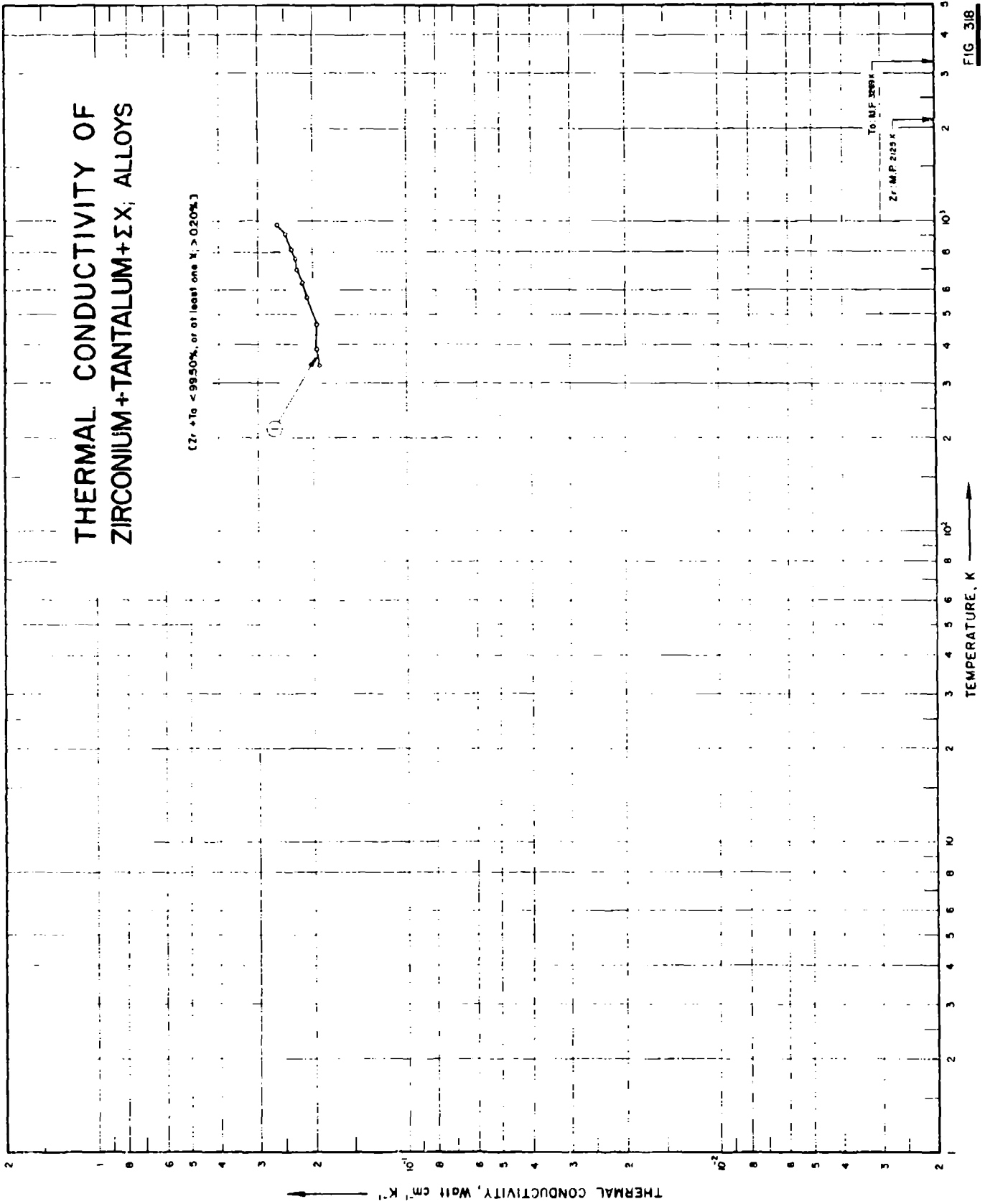
[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup> K<sup>-1</sup>]

T	k
<u>CURVE 1*</u>	
747	0.166
761	0.168
808	0.171
868	0.185

\* No graphical presentation

# THERMAL CONDUCTIVITY OF ZIRCONIUM + TANTALUM + ΣX<sub>i</sub> ALLOYS

(Zr + Ta < 99.50%, or at least one X<sub>i</sub> > 0.20%)



TEMPERATURE, K

FIG 318



SPECIFICATION TABLE NO. 318 THERMAL CONDUCTIVITY OF [ZIRCONIUM + TANTALUM +  $\Sigma X_i$ ] ALLOYS(Zr + Ta  $\leq$  99.50% or at least one  $X_i > 0.20\%$ )

[For Data Reported in Figure and Table No. 318]

Curve No.	Rel. Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)			Composition (continued), Specifications and Remarks	
						Zr	Ta	C		
1	441	L	1957	344-970		97.75	0.98	0.3	0.97	Electrical resistivity 58.4, 66.2, 79.3, 92.5, 100.2, 106.3, 113.6, 119.0, 126.5, and 129.8 $\mu$ ohm cm at 70.5, 114.0, 192.0, 294.8, 253.8, 422.0, 479.8, 535.8, 635.0, and 697 C respectively.

DATA TABLE NO. 318 THERMAL CONDUCTIVITY OF [ZIRCONIUM + TANTALUM +  $\Sigma X_i$ ] ALLOYS

(Zr + Ta < 99.50% or at least one  $X_i > 0.20\%$ )

(Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup> K<sup>-1</sup>)

T	k
343.7	0.1925
387.2	0.1971
465.2	0.1962
568.0	0.2100
633.0	0.2184
695.2	0.2264
753.0	0.2297
808.7	0.2360
908.2	0.2464
970.2	0.2628

CURVE 1

# THERMAL CONDUCTIVITY OF ZIRCONIUM+TIN+ΣX; ALLOYS

(Zr + Sn < 99.50%, or of least one X, > 0.20%)

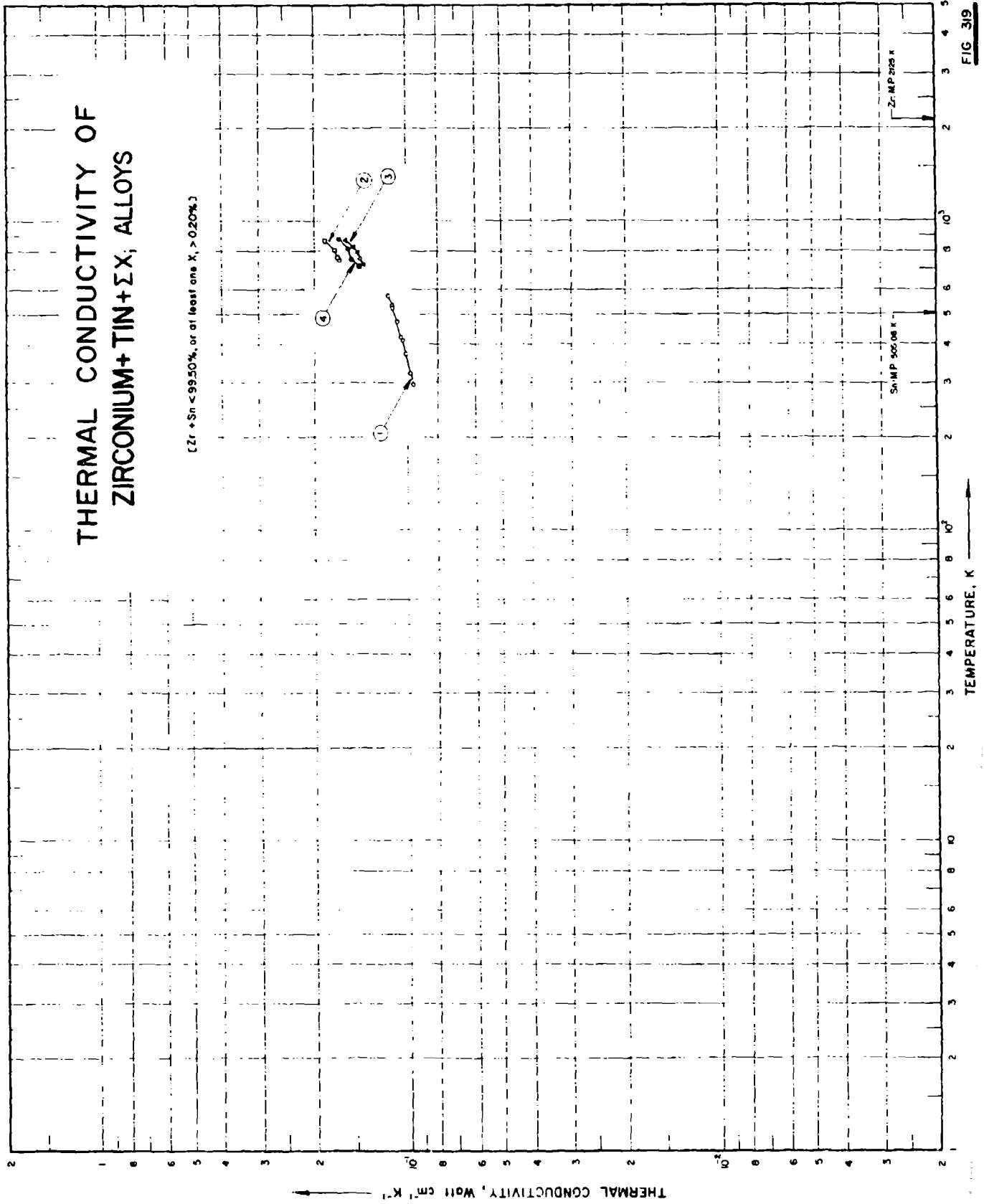


FIG 319

SPECIFICATION TABLE NO. 319 THERMAL CONDUCTIVITY OF (ZIRCONIUM + TIN +  $\Sigma X_i$ ) ALLOYS(Zr + Sn < 99.50% or at least one  $X_i > 0.50\%$ )

[For Data Reported in Figure and Table No. 319]

Curve No.	ReL No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)							Composition (continued), Specifications and Remarks			
							Zr	Sn	Al	C	Fe	HI	N		Ni		
1	442	C	1951	298-573	±3.0		Bal.	4.95	0.009	0.0125	0.235	0.024	0.0075	0.0055	0.014 Ti.	1.5 Mo; prepared from reactor-grade Zr sponge by double-arc melting; cast into 6 in. dia, 35 lb ingots; hard-rolled to 0.125 in. strip, then vacuum annealed at 878 C for 24 hrs.	
2	956	R	1960	747-868			Bal.	1.5	1.5								
3	956	R	1960	722-869			Bal.	3.0	3.0								Prepared from reactor-grade Zr sponge by double-arc melting; cast into 6 in. dia, 35 lb ingots; hard-rolled to 0.125 in. strip, then vacuum annealed at 878 C for 24 hrs.
4	956	R	1960	716-875			Bal.	3.0	1.5								

Similar to the above specimen.

DATA TABLE NO. 319 THERMAL CONDUCTIVITY OF [ZIRCONIUM + TIN + EX<sub>1</sub>] ALLOYS(Zr + Sn < 99.50% or at least one X<sub>1</sub> > 0.20%)[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup> K<sup>-1</sup>]

T	k
<u>CURVE 1</u>	
297.2	0.096
323.2	0.098
373.2	0.101
413.2	0.104
423.2	0.105
473.2	0.108
523.2	0.112
533.2	0.112
573.2	0.116
<u>CURVE 2</u>	
747	0.168
751	0.168
808	0.171
868	0.185
<u>CURVE 3</u>	
728	0.140
760	0.144
792	0.147
825	0.151
869	0.159
<u>CURVE 4</u>	
716	0.144
753	0.151
818	0.156
875	0.166

SPECIFICATION TABLE NO. 320 THERMAL CONDUCTIVITY OF [ZIRCONIUM + URANIUM +  $\Sigma X_i$ ] ALLOYS Zr + U +  $\Sigma X_i$   
 (Zr + U < 99.50% or at least one  $X_i > 0.20\%$ )

Curve No.	Ref. Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)	Composition (continued), Specifications and Remarks
						Zr U B	
1	27	C	1953	323-673	513	Bal 4.34 9.04	0.09 Cr, 0.125 Fe, 0.013 Ni, 0.027 Ni, and 1.33 Sn; arc-melted, forged, and rolled at 871 C.

DATA TABLE NO. 320 THERMAL CONDUCTIVITY OF [ZIRCONIUM + URANIUM +  $\Sigma X_i$ ] ALLOYS Zr + U +  $\Sigma X_i$

(Zr + U < 99.50% or at least one  $X_i > 0.20\%$ )

[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup> K<sup>-1</sup>]

T	k
323.20	0.145
373.20	0.141
423.20	0.138
473.20	0.137
523.20	0.137
573.20	0.137
623.20	0.138
673.20	0.139

\* No graphical presentation

SPECIFICATION TABLE NO. 321 THERMAL CONDUCTIVITY OF [ZIRCONIUM + EX<sub>1</sub>] ALLOYS Zr + EX<sub>1</sub>

Curve No.	Ref. Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) Zr	Composition (continued), Specifications and Remarks
1	97	L	1952	2.3-27	2-3	Zr 1	n=98 Polycrystalline; annealed.

DATA TABLE NO. 321 THERMAL CONDUCTIVITY OF [ZIRCONIUM + EX<sub>1</sub>] ALLOYS Zr + EX<sub>1</sub>[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup> K<sup>-1</sup>]

T	k
CURVE 1*	
2.32	0.0292
3.27	0.0396
4.30	0.0602
7.23	0.0998
10.75	0.138
14.11	0.179
16.90	0.206
20.86	0.237
21.85	0.246
23.14	0.255
24.82	0.262
27.30	0.268

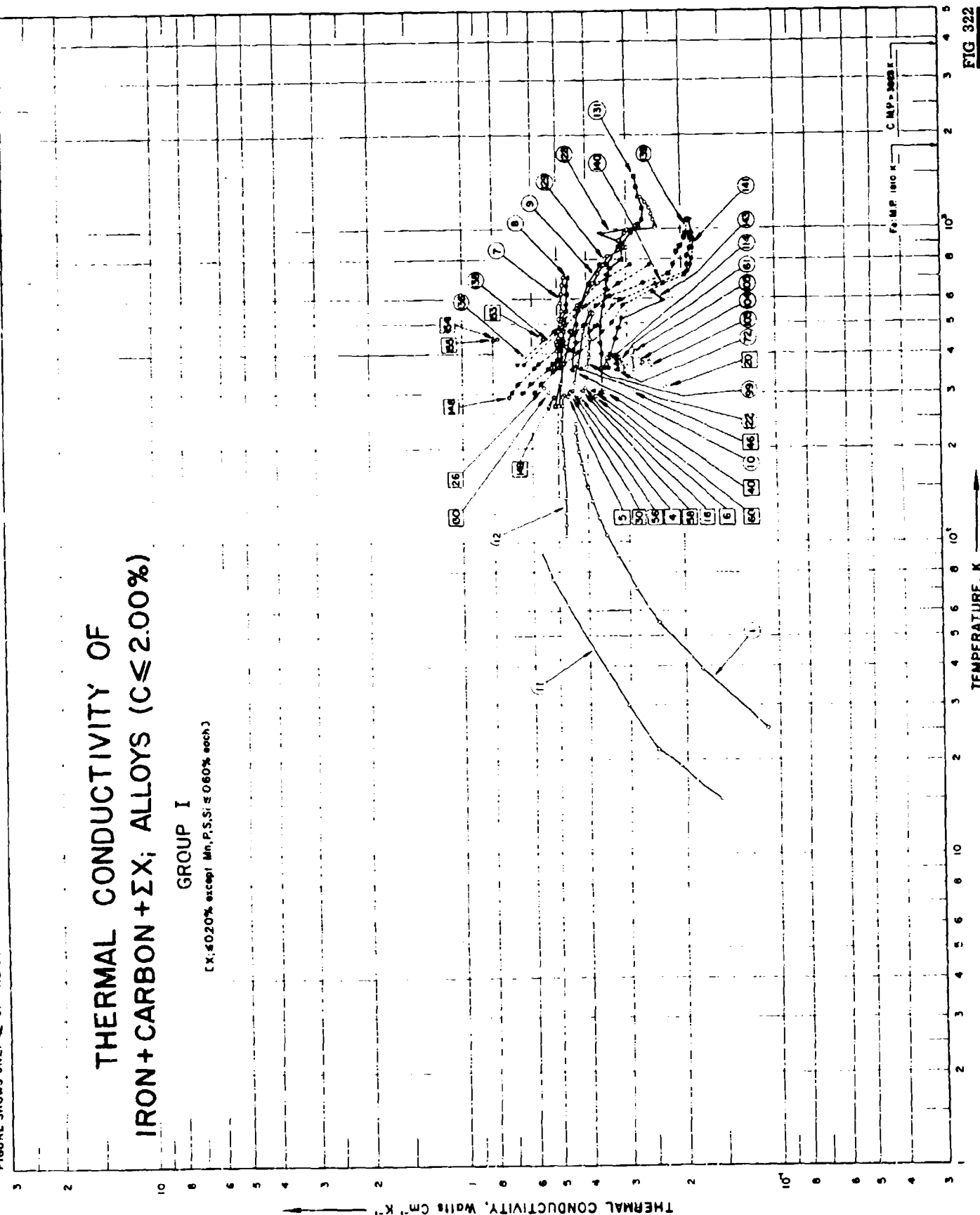
\* No graphical presentation

FIGURE SHOWS ONLY 42 OF THE CURVES REPORTED IN TABLE

# THERMAL CONDUCTIVITY OF IRON + CARBON + ΣX; ALLOYS (C ≤ 2.00%)

## GROUP I

(X: ≤ 0.20% except Mn, P, S; ≤ 0.60% each)



TEMPERATURE, K



SPECIFICATION TABLE NO. 322 THERMAL CONDUCTIVITY OF [IRON + CARBON + 2X<sub>j</sub>] ALLOYS (C ≤ 2.00%) GROUP 1(X<sub>j</sub> ≤ 0.20% except Mn, P, S, Si ≤ 0.60% each)

[For Data Reported in Figure and Table No. 322]

Curve Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight per cent)					Composition (continued), Specifications and Remarks	
						C	Mn	P	S	Si		
1	L	1951	26-240		SAE 1095	0.93	0.34			0.26	0.10 Cr; 0.10 Ni; < 0.05 Mn.	
2	C	1933	373-618	3.0-5.0	S1	0.83	0.27	0.017	0.015	0.16	Annealed at 800 C for 2 hrs and cooled in furnace.	
3	C	1933	364-773	3.0-5.0	S1 <sub>Q</sub>	0.83	0.27	0.017	0.015	0.16	The above specimen normalized at 900 C for 10 min and air-cooled, reheated to 800 C for 10 min and quenched in water, and reheated to 250 C for 1 hr and furnace cooled.	
4	167	1935	293			0.6	≈0.5			≈0.2		
5	167	1935	293			1.0	≈0.5			≈0.2		
6	167	1935	293			1.5	≈0.5			≈0.2		
7	31	L	1933	364-705	2.0	British steel: 4	0.92	0.56	0.032	0.039	0.177	Normalized at 900 C.
8	31	L	1933	363-703	2.0	British steel: 5	1.09	0.46	0.034	0.023	0.058	Normalized at 900 C.
9	169	R	1936	273-773		Steel 3	0.57	0.55			0.38	
10	169	R	1936	326-554		Steel 5	1.00	0.50			0.15	
11	104	L	1951	15- 93		1164 A/4	0.14	0.07			0.08	Heated to 800 C and cooled in furnace.
12	86	L	1908	105-296		Silver steel	1.0					Density = 7.84 at 24 C.
13	170	L	1926	313		4.1	0.41	0.014	0.015	0.28	0.28	Annealed.
14	170	L	1926	313		4.2	0.41	0.014	0.015	0.28	0.28	Forged.
15	170	L	1926	313		5.1	0.44	0.014	0.01	0.30	0.30	Annealed.
16	170	L	1926	313		5.2	0.44	0.014	0.01	0.30	0.30	Forged.
17	170	L	1926	313		5.3b	0.44	0.014	0.01	0.30	0.30	Annealed and then hardened at 800 C.
18	170	L	1926	313		6.1	0.54	0.014	0.015	0.25	0.25	Annealed.
19	170	L	1926	313		6.2	0.54	0.014	0.015	0.26	0.26	Forged.
20	170	L	1926	313		6.3b	0.54	0.014	0.015	0.26	0.26	Annealed and then hardened at 800 C.
21	170	L	1926	313		7.1	0.29	0.013	0.02	0.12	0.12	Annealed.
22	170	L	1926	313		7.2	0.29	0.013	0.02	0.12	0.12	Forged.
23	170	L	1926	313		7.3b	0.29	0.013	0.02	0.12	0.12	Annealed and then hardened at 800 C.
24	170	L	1926	313		8.1	0.29	0.013	0.03	0.08	0.08	Annealed.
25	170	L	1926	313		8.2	0.29	0.013	0.03	0.08	0.08	Forged.
26	170	L	1926	313		A	0.32			0.24	0.24	
27	170	L	1926	313		B	0.71			0.18	0.18	Annealed at 900 C and slowly cooled.
28	176	E	1920	303		1a	0.60					Annealed at 1:00 C and quickly cooled.
29	176	E	1920	303		1b	0.60					Annealed at 900 C and slowly cooled.
30	188	E	1919	303		1a	0.30					Annealed at 900 C and slowly cooled.

SPECIFICATION TABLE NO. 322 (continued)

Curve No.	Ref. Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	C	Composition (weight per cent)							Composition (continued), Specifications and Remarks	
							Cu	Mn	P	S	Si				
31	188	E	1919	303	1b	0.60									Annealed at 900 C and slowly cooled.
32	171	E	1917	303	Krupp steel:11a	0.56	0.09	0.30	0.016	0.034	0.18	0.034	0.18		Forged.
33	171	E	1917	303	Krupp steel:11b	0.56	0.09	0.30	0.016	0.034	0.18	0.034	0.18		Annealed at 900 C for 1 hr in vacuum.
34	171	E	1917	303	Krupp steel:11c	0.56	0.09	0.30	0.016	0.034	0.18	0.034	0.18		Oil-quenched from 900 C.
35	171	E	1917	303	Krupp steel:12a	0.64	0.09	0.27	0.10	0.028	0.06	0.028	0.06		Forged.
36	171	E	1917	303	Krupp steel:12b	0.64	0.09	0.27	0.10	0.028	0.06	0.028	0.06		Annealed at 900 C for 1 hr in vacuum.
37	171	E	1917	303	Krupp steel:12c	0.64	0.09	0.27	0.10	0.028	0.06	0.028	0.06		Oil-quenched from 900 C.
38	171	E	1917	303	Krupp steel:15a	0.75	0.07	0.35	0.010	0.018	0.10	0.018	0.10		Forged.
39	171	E	1917	303	Krupp steel:15b	0.75	0.07	0.35	0.010	0.018	0.10	0.018	0.10		Annealed at 900 C for 1 hr in vacuum.
40	171	E	1917	303	Krupp steel:15c	0.75	0.07	0.35	0.010	0.018	0.10	0.018	0.10		Oil-quenched from 900 C.
41	171	E	1917	303	Krupp steel:16a	0.80	0.08	0.30	0.012	0.016	0.10	0.016	0.10		Forged.
42	171	E	1917	303	Krupp steel:16b	0.80	0.08	0.30	0.012	0.016	0.10	0.016	0.10		Annealed at 900 C for 1 hr in vacuum.
43	171	E	1917	303	Krupp steel:16c	0.80	0.08	0.30	0.012	0.016	0.10	0.016	0.10		Oil-quenched from 900 C.
44	171	E	1917	303	Krupp steel:18a	0.94	0.08	0.25	0.016	0.016	0.13	0.016	0.13		Forged.
45	171	E	1917	303	Krupp steel:18b	0.94	0.08	0.35	0.016	0.018	0.13	0.018	0.13		Annealed at 900 C for 1 hr in vacuum.
46	171	E	1917	303	Krupp steel:18c	0.94	0.08	0.35	0.016	0.018	0.13	0.018	0.13		Oil-quenched from 900 C.
47	171	E	1917	303	Krupp steel:20a	1.02	0.05	0.36	0.013	0.014	0.08	0.014	0.08		Forged.
48	171	E	1917	363	Krupp steel:20b	1.02	0.05	0.36	0.013	0.014	0.08	0.014	0.08		Annealed at 900 C for 1 hr in vacuum.
49	171	E	1917	303	Krupp steel:20c	1.02	0.05	0.36	0.013	0.014	0.08	0.014	0.08		Oil-quenched from 900 C.
50	171	E	1917	303	Krupp steel:26a	1.30	0.06	0.40	0.046	0.018	0.08	0.018	0.08		Forged.
51	171	E	1917	303	Krupp steel:26b	1.30	0.06	0.40	0.046	0.018	0.08	0.018	0.08		Annealed at 900 C for 1 hr in vacuum.
52	171	E	1917	303	Krupp steel:26c	1.30	0.06	0.40	0.046	0.018	0.08	0.018	0.08		Oil-quenched from 900 C.
53	171	E	1917	303	Krupp steel:30a	1.50	0.05	0.36	0.020	0.020	0.05	0.020	0.05		Forged.
54	171	E	1917	303	Krupp steel:30b	1.50	0.05	0.36	0.020	0.020	0.05	0.020	0.05		Annealed at 900 C for 1 hr in vacuum.
55	171	E	1917	303	Krupp steel:30c	1.50	0.05	0.36	0.020	0.020	0.05	0.020	0.05		Oil-quenched from 900 C.
56	172	E	1927	307	3	0.70		0.21	0.023	0.023	0.27	0.023	0.27		
57	172	E	1927	307	4	0.885		0.19	0.028	0.015	0.34	0.015	0.34		
58	172	E	1927	307	5	1.015		0.20	0.027	0.020	0.27	0.020	0.27		
59	172	E	1927	307	6	1.185		0.22	0.017	0.031	0.28	0.031	0.28		
60	172	E	1927	307	7	1.480		0.18	0.026	0.020	0.24	0.020	0.24		
61	189	L	1934	356-390	2.0-5.0 Tool steel: A <sub>9</sub>	1.41	0.01	0.23	0.037	0.006	0.158	0.006	0.158		Water-quenched from 775 C.

SPECIFICATION TABLE NO. 322 (continued)

Curve No.	Ref. Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight per cent)							Composition (continued), Specifications and Remarks
						C	Cu	Mn	P	S	Sh		
62	189	L	1934	361-390	2.0-5.0	A.T. 150 C	1.41	0.01	0.23	0.037	0.006	0.158	Tempered for 30 min at 150 C and air-cooled.
63	189	L	1934	366-427	2.0-5.0	A.T. 200 C	1.41	0.01	0.23	0.037	0.006	0.158	Tempered for 30 min at 200 C and air-cooled.
64	189	L	1934	364-468	2.0-5.0	A.T. 250 C	1.41	0.01	0.23	0.037	0.006	0.158	Tempered for 30 min at 250 C and air-cooled.
65	189	L	1934	366-509	2.0-5.0	A.T. 300 C	1.41	0.01	0.23	0.037	0.006	0.158	Tempered for 30 min at 300 C and air-cooled.
66	189	L	1934	370-537	2.0-5.0	A.T. 350 C	1.41	0.01	0.23	0.037	0.006	0.158	Tempered for 30 min at 350 C and air-cooled.
67	189	L	1934	376-539	2.0-5.0	A.T. 400 C	1.41	0.01	0.23	0.037	0.006	0.158	Tempered for 30 min at 400 C and air-cooled.
68	189	L	1934	377-547	2.0-5.0	A.T. 500 C	1.41	0.01	0.23	0.037	0.006	0.158	Tempered for 30 min at 500 C and air-cooled.
69	189	L	1934	382-538	2.0-5.0	A.T. 600 C	1.41	0.01	0.23	0.037	0.006	0.158	Tempered for 30 min at 600 C and air-cooled.
70	189	L	1934	370-492	2.0-5.0	A.T. 700 C	1.41	0.01	0.23	0.037	0.006	0.158	Tempered for 30 min at 700 C and air-cooled.
71	189	L	1934	372-544	2.0-5.0	AA	1.41	0.01	0.23	0.037	0.006	0.158	Annealed at 775 C.
72	189	L	1934	353-389	2.0-5.0	B1 <sub>Q</sub>	1.14		0.207	0.020	0.026	0.117	Water-quenched from 780 C.
73	189	L	1934	353-387	2.0-5.0	B1 <sub>T</sub> , 150 C	1.14		0.207	0.020	0.026	0.117	Tempered for 30 min at 150 C and air-cooled.
74	189	L	1934	366-421	2.0-5.0	B1 <sub>T</sub> , 200 C	1.14		0.207	0.020	0.026	0.117	Tempered for 30 min at 200 C and air-cooled.
75	189	L	1934	371-469	2.0-5.0	B1 <sub>T</sub> , 250 C	1.14		0.207	0.020	0.026	0.117	Tempered for 30 min at 250 C and air-cooled.
76	189	L	1934	366-520	2.0-5.0	B1 <sub>T</sub> , 300 C	1.14		0.207	0.020	0.026	0.117	Tempered for 30 min at 300 C and air-cooled.
77	189	L	1934	367-540	2.0-5.0	B1 <sub>T</sub> , 350 C	1.14		0.207	0.020	0.026	0.117	Tempered for 30 min at 350 C and air-cooled.
78	189	L	1934	368-545	2.0-5.0	B1 <sub>T</sub> , 400 C	1.14		0.207	0.020	0.026	0.117	Tempered for 30 min at 400 C and air-cooled.
79	189	L	1934	371-541	2.0-5.0	B1 <sub>T</sub> , 500 C	1.14		0.207	0.020	0.026	0.117	Tempered for 30 min at 500 C and air-cooled.
80	189	L	1934	358-534	2.0-5.0	B1 <sub>T</sub> , 600 C	1.14		0.207	0.020	0.026	0.117	Tempered for 30 min at 600 C and air-cooled.
81	189	L	1934	361-530	2.0-5.0	B1 <sub>T</sub> , 700 C	1.14		0.207	0.020	0.026	0.117	Tempered for 30 min at 700 C and air-cooled.
82	189	L	1934	363-536	2.0-5.0	B1 <sub>A</sub>	1.14		0.207	0.020	0.026	0.117	Annealed at 780 C.
83	189	L	1934	355-388	2.0-5.0	B2 <sub>Q</sub>	1.14		0.207	0.020	0.026	0.117	Water-quenched from 940 C.
84	189	L	1934	359-389	2.0-5.0	B2 <sub>T</sub> , 150 C	1.14		0.207	0.020	0.026	0.117	Tempered for 30 min at 150 C and air-cooled.
85	189	L	1934	361-421	2.0-5.0	B2 <sub>T</sub> , 200 C	1.14		0.207	0.020	0.026	0.117	Tempered for 30 min at 200 C and air-cooled.
86	189	L	1934	363-469	2.0-5.0	B2 <sub>T</sub> , 250 C	1.14		0.207	0.020	0.026	0.117	Tempered for 30 min at 250 C and air-cooled.
87	189	L	1934	367-502	2.0-5.0	B2 <sub>T</sub> , 300 C	1.14		0.207	0.020	0.026	0.117	Tempered for 30 min at 300 C and air-cooled.
88	189	L	1934	361-534	2.0-5.0	B2 <sub>T</sub> , 350 C	1.14		0.207	0.020	0.026	0.117	Tempered for 30 min at 350 C and air-cooled.
89	189	L	1934	372-532	2.0-5.0	B2 <sub>T</sub> , 400 C	1.14		0.207	0.020	0.026	0.117	Tempered for 30 min at 400 C and air-cooled.
90	189	L	1934	368-535	2.0-5.0	B2 <sub>T</sub> , 500 C	1.14		0.207	0.020	0.026	0.117	Tempered for 30 min at 500 C and air-cooled.
91	189	L	1934	365-543	2.0-5.0	B2 <sub>T</sub> , 600 C	1.14		0.207	0.020	0.026	0.117	Tempered for 30 min at 600 C and air-cooled.

SPECIFICATION TABLE NO. 322 (continued)

Curve No.	Ref. Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight per cent)							Composition (continued), Specifications and Remarks	
						C	Cu	Mn	P	S	Si			
92	189	L	1934	368-540	2.0-5.0	B2, T. 700	1.14		0.207	0.020	0.026	0.117	0.117	Tempered for 30 min at 700 C and air-cooled.
93	189	L	1934	362-369	2.0-5.0	B3, Q	1.14		0.207	0.020	0.026	0.117	0.117	Water-quenched from 900 C.
94	189	L	1934	363-390	2.0-5.0	B3, T. 150 C	1.14		0.207	0.020	0.026	0.117	0.117	Tempered for 30 min at 150 C and air-cooled.
95	189	L	1934	370-426	2.0-5.0	B3, T. 200 C	1.14		0.207	0.020	0.026	0.117	0.117	Tempered for 30 min at 200 C and air-cooled.
96	189	L	1934	360-463	2.0-5.0	B3, T. 250 C	1.14		0.207	0.020	0.026	0.117	0.117	Tempered for 30 min at 250 C and air-cooled.
97	189	L	1934	361-493	2.0-5.0	B3, T. 300 C	1.14		0.207	0.020	0.026	0.117	0.117	Tempered for 30 min at 300 C and air-cooled.
98	189	L	1934	372-544	2.0-5.0	B3, T. 350 C	1.14		0.207	0.020	0.026	0.117	0.117	Tempered for 30 min at 350 C and air-cooled.
99	189	L	1934	366-540	2.0-5.0	B3, T. 400 C	1.14		0.207	0.020	0.026	0.117	0.117	Tempered for 30 min at 400 C and air-cooled.
100	189	L	1934	364-544	2.0-5.0	B3, T. 500 C	1.14		0.207	0.020	0.026	0.117	0.117	Tempered for 30 min at 500 C and air-cooled.
101	189	L	1934	368-540	2.0-5.0	B3, T. 600 C	1.14		0.207	0.020	0.026	0.117	0.117	Tempered for 30 min at 600 C and air-cooled.
102	189	L	1934	371-537	2.0-5.0	B3, T. 700 C	1.14		0.207	0.020	0.026	0.117	0.117	Tempered for 30 min at 700 C and air-cooled.
103	189	L	1934	360-387	2.0-5.0	B4, Q	1.14		0.207	0.020	0.026	0.117	0.117	Water-quenched from 1000 C.
104	189	L	1934	368-390	2.0-5.0	B4, T. 150 C	1.14		0.207	0.020	0.026	0.117	0.117	Tempered for 30 min at 150 C and air-cooled.
105	189	L	1934	358-419	2.0-5.0	B4, T. 200 C	1.14		0.207	0.020	0.026	0.117	0.117	Tempered for 30 min at 200 C and air-cooled.
106	189	L	1934	363-467	2.0-5.0	B4, T. 250 C	1.14		0.207	0.020	0.026	0.117	0.117	Tempered for 30 min at 250 C and air-cooled.
107	189	L	1934	366-508	2.0-5.0	B4, T. 300 C	1.14		0.207	0.020	0.026	0.117	0.117	Tempered for 30 min at 300 C and air-cooled.
108	189	L	1934	373-543	2.0-5.0	B4, T. 350 C	1.14		0.207	0.020	0.026	0.117	0.117	Tempered for 30 min at 350 C and air-cooled.
109	189	L	1934	365-547	2.0-5.0	B4, T. 400 C	1.14		0.207	0.020	0.026	0.117	0.117	Tempered for 30 min at 400 C and air-cooled.
110	189	L	1934	367-527	2.0-5.0	B4, T. 500 C	1.14		0.207	0.020	0.026	0.117	0.117	Tempered for 30 min at 500 C and air-cooled.
111	189	L	1934	368-533	2.0-5.0	B4, T. 600 C	1.14		0.207	0.020	0.026	0.117	0.117	Tempered for 30 min at 600 C and air-cooled.
112	189	L	1934	372-544	2.0-5.0	B4, T. 700 C	1.14		0.207	0.020	0.026	0.117	0.117	Tempered for 30 min at 700 C and air-cooled.
113	189	L	1934	362-388	2.0-5.0	C, Q	0.931	0.038	0.157	0.014	0.007	0.109	0.109	Water-quenched from 790 C.
114	189	L	1934	361-390	2.0-5.0	C, T. 150 C	0.931	0.038	0.157	0.014	0.007	0.109	0.109	Tempered for 30 min at 150 C and air-cooled.
115	189	L	1934	363-415	2.0-5.0	C, T. 200 C	0.931	0.038	0.157	0.014	0.007	0.109	0.109	Tempered for 30 min at 200 C and air-cooled.
116	189	L	1934	368-470	2.0-5.0	C, T. 250 C	0.931	0.038	0.157	0.014	0.007	0.109	0.109	Tempered for 30 min at 250 C and air-cooled.
117	189	L	1934	378-506	2.0-5.0	C, T. 300 C	0.931	0.038	0.157	0.014	0.007	0.109	0.109	Tempered for 30 min at 300 C and air-cooled.
118	189	L	1934	364-543	2.0-5.0	C, T. 350 C	0.931	0.038	0.157	0.014	0.007	0.109	0.109	Tempered for 30 min at 350 C and air-cooled.
119	189	L	1934	370-497	2.0-5.0	C, T. 400 C	0.931	0.038	0.157	0.014	0.007	0.109	0.109	Tempered for 30 min at 400 C and air-cooled.
120	189	L	1934	368-548	2.0-5.0	C, T. 500 C	0.931	0.038	0.157	0.014	0.007	0.109	0.109	Tempered for 30 min at 500 C and air-cooled.
121	189	L	1934	369-543	2.0-5.0	C, T. 600 C	0.931	0.038	0.157	0.014	0.007	0.109	0.109	Tempered for 30 min at 600 C and air-cooled.

SPECIFICATION TABLE NO. 322 (continued)

Curve No.	Ref. Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)							Remarks
						C	Fe	Cu	Mn	P	S	Si	
122	189	L	1934	370-551	2.0-5.0	C.T. 700 C	0.931	0.038	0.157	0.014	0.007	0.109	Tempered for 30 min at 700 C and air-cooled.
123	189	L	1934	377-545	2.0-5.0	C.A	0.931	0.038	0.157	0.014	0.007	0.109	Annealed at 790 C.
124	71	L	1917	303-1199		12	0.64	0.09	0.27	0.010	0.028	0.06	Forged.
125	71	L	1917	303-1170		16	0.80	0.08	0.30	0.012	0.016	0.10	Forged.
126	71	L	1917	303-1159		20	1.02	0.05	0.36	0.013	0.014	0.08	Forged.
127	71	L	1917	303-1147		26	1.30	0.06	0.40	0.046	0.018	0.08	Forged.
128	71	L	1917	303-1048		30	1.50	0.05	0.36	0.020	0.020	0.05	Forged.
129	190	C, R	1946	273-1273		British steel; 7	0.80	0.070	0.32	0.008	0.009	0.13	0.13 Ni, 0.11 Cr, 0.021 As, <0.01 Mo, 0.004 Al; annealed at 860 C.
130	165	E	1919	303-2		1	0.206		0.11	0.05	0.04	0.06	
131	163	L	1936	303-1473		German steel; PD1	0.45		0.07	0.010	0.014	0.09	Annealed.
132	166	C	1939	273-623		British steel; 7	0.80	0.070	0.32	0.008	0.009	0.13	0.13 Ni, 0.11 Cr, 0.021 As, <0.01 Mo, 0.004 Al; annealed at 860 C.
133	166	C	1939	273-423		8	1.22	0.077	0.35	0.009	0.015	0.16	0.13 Ni, 0.11 Cr, 0.025 As, 0.01 Mo, 0.006 Al; annealed at 800 C.
134	191	<del>LR</del>	1926	310-320		A	0.50		0.32			0.24	
135	191	<del>LR</del>	1926	313-4		B	0.71		0.18			0.24	
136	160	F	1938	373-773		12	0.85		0.65				Nominal composition; annealed.
137	160	F	1938	373-773		13	1.10		0.55				Nominal composition; annealed.
138	160	F	1938	373-773		14	1.40		0.53				Nominal composition; annealed.
139	175	P	1936	303-1073		Carbon steel; 2	0.31	0.134	0.14	0.011	0.014	0.040	Annealed at 850 C for 2.5 hrs.
140	175	P	1936	303-1073		Carbon steel; 3	0.65	0.150	0.16	0.008	0.012	0.023	Annealed at 850 C for 2.5 hrs.
141	175	P	1936	303-1073		Carbon steel; 4	0.88	0.135	0.16	0.009	0.011	0.049	Annealed at 850 C for 2.5 hrs.
142	177	C	1936	298-2	10.0	Russian steel; U-9	0.91		0.35	0.030		0.35	Annealed.
143	192	L	1954	365-632			1.13		0.43			0.4	1.43 FeO, 1.13 Fe <sub>2</sub> O <sub>3</sub> ; porosity 19%; without heat treatment.
144	560	E	1953	350-814	±3.0		0.32		0.30			0.30	1.02 O; sintered at 1150 C for 1 1/2 hrs; porosity 10.4%.
145	560	E	1953	342-711	±3.0		0.88		0.27			0.34	0.20 O; sintered at 1150 C for 1 1/2 hrs; porosity 9.5%.
146	560	E	1953	319-733	±3.0		0.88		0.27			0.34	0.22 O; sintered at 1150 C for 1 1/2 hrs; porosity 9.5%.
147	560	E	1953	358-750	±3.0		1.62		0.40			0.40	94.64 Fe, 0.4 Fe <sub>2</sub> O <sub>3</sub> ; 0.36 FeO; sintered at 1150 C for 1 1/2 hrs; porosity 10.6%.
148	435	L	1900	291.2		FeWA14	0.105	0.05	0.06	0.03	0.015	0.015	
149	435	L	1900	291.2		FeWA 2	0.57	0.03	0.12	0.012	0.042	0.21	

SPECIFICATION TABLE NO. 322 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)							Remarks
							C	Cu	Mn	P	S	Si		
150	435	L	1900	291.2		Fe-W-A 3	0.99	0.035	0.12	0.005	0.025	0.06	Composition (continued), Specifications and Remarks  0.13 Ni.  0.18 Ni and 0.06 Cr. 0.08 Cr.; annealed. 0.04 Cr.; cooled in oil from 850 C. Nominal Composition. Nominal Composition from Mark's handbook.	
151	435	L	1900	291.2		Fe-W-A 4	1.50	0.03	0.19	0.01	0.025	0.05		
152	561	C	1925	448.2			1.00	0.28				0.34		
153	561	C	1925	450.6		High C steel	1.28		0.07			0.12		
154	561	C	1925	449.0			0.05		0.04			0.01		
155	561	C	1925	445.7		Fish-plate	0.21	0.068	0.13	0.027	0.025	0.09		
156	539	L	1938	329-1105		Carbon steel	1.01		0.47	0.02	0.01	0.16		
157	539	L	1938	337-787		Carbon steel	1.01		0.47	0.02	0.01	0.16		
158	562	L	1949	26-240		1095								
159	504	P	1961	295.2	±5.0	1020	0.18/ 0.22		0.3/ 0.60	0.04 Max	0.05 Max			

DATA TABLE NO. 322 THERMAL CONDUCTIVITY OF [IRON + CARBON + EX<sub>1</sub>] ALLOYS (C ≤ 2.00%) GROUP I

(X<sub>1</sub> ≤ 0.20% except Mn, P, S, Si ≤ 0.60% each)

[Temperature, T, K; Thermal Conductivity, k, Watts cm<sup>-1</sup>K<sup>-1</sup>]

T	k	T	k	T	k	T	k	T	k	T	k	T	k
<u>CURVE 1</u>													
25.52	0.109	364.20	0.506	375.80	0.439	313.20	0.272	303.20	0.418	303.20	0.352	303.20	0.322
39.33	0.177	386.20	0.498	373.20	0.427	313.20	0.312	<u>CURVE 28*</u>		<u>CURVE 40</u>		<u>CURVE 52*</u>	
55.13	0.241	411.70	0.502	473.20	0.407	<u>CURVE 18</u>		<u>CURVE 29*</u>		<u>CURVE 41*</u>		<u>CURVE 53*</u>	
70.98	0.284	430.70	0.498	553.90	0.389	<u>CURVE 11</u>		303.20	0.411	303.20	0.423	303.20	0.360
90.60	0.325	438.70	0.502	<u>CURVE 19*</u>		313.20	0.402	<u>CURVE 30</u>		<u>CURVE 42*</u>		<u>CURVE 54*</u>	
105.29	0.350	450.70	0.494	15.00	0.154	<u>CURVE 20</u>		303.20	0.457	303.20	0.423	303.20	0.347
119.91	0.369	472.20	0.498	17.16	0.186	<u>CURVE 21*</u>		<u>CURVE 31*</u>		<u>CURVE 43*</u>		<u>CURVE 55*</u>	
135.78	0.386	474.20	0.494	19.22	0.209	313.20	0.414	303.20	0.422	303.20	0.385	303.20	0.297
152.06	0.400	497.70	0.490	21.66	0.244	<u>CURVE 22*</u>		<u>CURVE 32*</u>		<u>CURVE 44*</u>		<u>CURVE 56</u>	
167.15	0.412	519.70	0.494	29.90	0.306	313.20	0.510	303.20	0.418	303.20	0.393	307.20	0.447
179.62	0.419	548.70	0.485	93.00	0.568	<u>CURVE 23*</u>		<u>CURVE 33*</u>		<u>CURVE 45*</u>		<u>CURVE 57*</u>	
195.39	0.428	582.70	0.490	105.20	0.473	313.20	0.502	303.20	0.410	303.20	0.398	307.20	0.417
209.88	0.432	625.70	0.485	113.20	0.473	<u>CURVE 24*</u>		<u>CURVE 34*</u>		<u>CURVE 46</u>		<u>CURVE 58</u>	
225.36	0.436	665.70	0.481	148.20	0.473	313.20	0.511	303.20	0.389	303.20	0.289	307.20	0.416
239.84	0.441	704.70	0.477	173.20	0.477	<u>CURVE 25*</u>		<u>CURVE 35*</u>		<u>CURVE 47*</u>		<u>CURVE 59*</u>	
240.16	0.443	<u>CURVE 8</u>		223.20	0.485	313.20	0.502	303.20	0.439	303.20	0.431	307.20	0.387
<u>CURVE 2*</u>													
373.20	0.458	362.70	0.498	296.20	0.477	<u>CURVE 26</u>		<u>CURVE 36*</u>		<u>CURVE 48*</u>		<u>CURVE 60</u>	
473.20	0.435	394.70	0.494	313.20	0.473	313.20	0.531	<u>CURVE 37*</u>		<u>CURVE 49*</u>		<u>CURVE 61</u>	
573.20	0.413	409.70	0.490	313.20	0.473	313.20	0.511	303.20	0.431	303.20	0.381	307.20	0.384
673.20	0.390	428.20	0.481	313.20	0.473	<u>CURVE 27*</u>		<u>CURVE 38*</u>		<u>CURVE 50*</u>		<u>CURVE 62*</u>	
773.20	0.367	436.20	0.490	313.20	0.444	313.20	0.511	303.20	0.418	303.20	0.318	355.70	0.305
818.00	0.352	447.70	0.481	313.20	0.444	<u>CURVE 28*</u>		<u>CURVE 39*</u>		<u>CURVE 51*</u>		374.20	0.310
<u>CURVE 3*</u>													
364.40	0.411	470.70	0.477	273.20	0.485	313.20	0.502	303.20	0.414	303.20	0.360	390.20	0.318
373.20	0.412	494.20	0.473	291.20	0.481	<u>CURVE 29*</u>		<u>CURVE 40*</u>		<u>CURVE 52*</u>		360.70	0.322
473.20	0.408	515.70	0.473	296.20	0.477	313.20	0.510	303.20	0.393	303.20	0.350	376.70	0.322
573.20	0.396	519.20	0.485	313.20	0.473	<u>CURVE 30*</u>		<u>CURVE 41*</u>		<u>CURVE 53*</u>		389.70	0.326
673.20	0.381	547.70	0.469	313.20	0.460	313.20	0.511	303.20	0.381	303.20	0.360	<u>CURVE 63*</u>	
773.20	0.364	580.20	0.469	313.20	0.460	313.20	0.511	303.20	0.381	303.20	0.360	366.20	0.331
<u>CURVE 4</u>													
293.20	0.464	702.70	0.460	313.20	0.339	<u>CURVE 31*</u>		<u>CURVE 42*</u>		<u>CURVE 54*</u>		401.70	0.339
<u>CURVE 5</u>													
293.20	0.423	373.20	0.510	373.20	0.477	<u>CURVE 32*</u>		<u>CURVE 43*</u>		<u>CURVE 55*</u>		421.70	0.335
<u>CURVE 6</u>													
293.20	0.377	373.20	0.477	373.20	0.448	<u>CURVE 33*</u>		<u>CURVE 44*</u>		<u>CURVE 56</u>		424.70	0.372
<u>CURVE 64*</u>													
<u>CURVE 65*</u>													
<u>CURVE 66*</u>													
<u>CURVE 67*</u>													
<u>CURVE 68*</u>													
<u>CURVE 69*</u>													
<u>CURVE 70*</u>													
<u>CURVE 71*</u>													
<u>CURVE 72*</u>													
<u>CURVE 73*</u>													
<u>CURVE 74*</u>													
<u>CURVE 75*</u>													
<u>CURVE 76*</u>													
<u>CURVE 77*</u>													
<u>CURVE 78*</u>													
<u>CURVE 79*</u>													
<u>CURVE 80*</u>													
<u>CURVE 81*</u>													
<u>CURVE 82*</u>													
<u>CURVE 83*</u>													
<u>CURVE 84*</u>													
<u>CURVE 85*</u>													
<u>CURVE 86*</u>													
<u>CURVE 87*</u>													
<u>CURVE 88*</u>													
<u>CURVE 89*</u>													
<u>CURVE 90*</u>													
<u>CURVE 91*</u>													
<u>CURVE 92*</u>													
<u>CURVE 93*</u>													
<u>CURVE 94*</u>													
<u>CURVE 95*</u>													
<u>CURVE 96*</u>													
<u>CURVE 97*</u>													
<u>CURVE 98*</u>													
<u>CURVE 99*</u>													
<u>CURVE 100*</u>													

\* Not shown on plot

DATA TABLE NO. 322 (continued)

T	k	T	k	T	k	T	k	T	k	T	k	T	k		
<u>CURVE 68*</u>															
376.70	0.393	521.20	0.414	355.70	0.297	372.20	0.423	363.70	0.285	495.70	0.393	358.70	0.285		
422.70	0.393	544.70	0.406	374.20	0.297	399.40	0.431	376.70	0.285	540.20	0.381	395.70	0.289		
435.20	0.402	<u>CURVE 79*</u>													
456.20	0.398	<u>CURVE 80*</u>													
489.70	0.393	370.70	0.414	359.20	0.318	370.20	0.427	370.20	0.297	364.70	0.427	363.70	0.285		
518.20	0.398	392.20	0.418	374.70	0.314	532.20	0.414	390.20	0.301	422.20	0.414	392.70	0.381		
546.70	0.389	410.70	0.410	389.20	0.318	532.20	0.414	426.20	0.301	446.20	0.410	418.70	0.377		
<u>CURVE 69*</u>															
361.70	0.398	445.20	0.410	389.20	0.318	532.20	0.414	426.20	0.301	470.70	0.410	452.70	0.372		
409.20	0.402	474.20	0.406	368.20	0.423	368.20	0.423	368.20	0.368	503.70	0.414	467.20	0.377		
436.20	0.393	511.20	0.414	397.70	0.427	397.70	0.427	360.70	0.364	543.70	0.406	<u>CURVE 107*</u>			
463.70	0.402	540.70	0.410	427.70	0.427	427.70	0.427	391.70	0.368	368.70	0.435	366.20	0.393		
489.20	0.398	<u>CURVE 81*</u>													
512.20	0.402	358.20	0.427	391.20	0.335	452.20	0.431	422.20	0.364	394.70	0.431	415.70	0.398		
538.20	0.393	391.20	0.423	420.70	0.343	477.70	0.423	462.70	0.368	429.20	0.423	444.20	0.389		
<u>CURVE 70*</u>															
369.70	0.406	419.20	0.423	363.70	0.385	516.20	0.435	493.20	0.396	477.20	0.418	479.20	0.393		
395.70	0.398	457.70	0.414	387.20	0.381	365.70	0.435	261.20	0.398	506.20	0.427	508.20	0.385		
431.20	0.406	473.70	0.423	413.20	0.389	399.20	0.427	385.50	0.402	539.70	0.423	<u>CURVE 108*</u>			
452.20	0.410	515.20	0.410	439.20	0.385	429.20	0.431	415.20	0.392	371.20	0.427	373.70	0.398		
478.20	0.402	534.20	0.414	469.20	0.385	460.20	0.435	445.70	0.389	394.20	0.423	416.70	0.398		
508.20	0.402	<u>CURVE 82*</u>													
491.70	0.398	360.70	0.431	367.20	0.398	494.70	0.427	464.70	0.402	431.70	0.423	492.70	0.398		
<u>CURVE 71*</u>															
371.70	0.402	400.70	0.427	395.70	0.406	516.20	0.435	493.20	0.396	474.20	0.418	514.70	0.389		
391.20	0.410	432.70	0.423	427.70	0.402	543.20	0.427	372.20	0.402	509.20	0.423	543.20	0.391		
429.20	0.402	450.20	0.427	457.20	0.402	368.20	0.435	394.20	0.398	536.70	0.410	<u>CURVE 109*</u>			
462.20	0.406	482.70	0.423	479.70	0.398	445.20	0.423	477.70	0.389	360.20	0.427	365.20	0.402		
491.20	0.410	509.20	0.431	501.70	0.398	481.20	0.427	515.20	0.398	377.20	0.251	416.70	0.398		
516.70	0.402	529.70	0.427	<u>CURVE 83*</u>											
543.70	0.398	363.20	0.431	361.70	0.402	540.20	0.418	544.20	0.402	386.70	0.255	442.70	0.402		
<u>CURVE 72*</u>															
357.70	0.322	396.20	0.423	399.70	0.398	362.70	0.423	362.70	0.398	362.70	0.398	391.70	0.393		
374.70	0.322	428.20	0.431	421.20	0.406	374.70	0.264	374.70	0.393	368.20	0.272	410.70	0.393		
389.20	0.326	452.70	0.423	451.20	0.402	389.20	0.402	389.20	0.402	375.70	0.268	447.20	0.389		
		475.70	0.423	474.20	0.406	534.20	0.402	476.70	0.393	390.20	0.272	476.70	0.393		
		508.20	0.427	500.20	0.396										
		536.20	0.416	534.20	0.402										

\* Not shown on plot



DATA TABLE NO. 322 (continued)

T	k	T	k	T	k	T	k	T	k	T	k					
<u>CURVE 110*</u>																
367.20	0.427	368.20	0.398	1100.20	0.305	407.20	0.368	673.20	0.381	573.20	0.414					
391.20	0.423	393.70	0.439	1134.20	0.289	460.20	0.368	723.20	0.366	623.20	0.401					
428.20	0.427	425.70	0.439	1199.20	0.318	487.20	0.364	773.20	0.354	373.20	0.580					
451.20	0.414	458.20	0.431	<u>CURVE 125*</u>			541.20	0.369	823.20	0.341	<u>CURVE 133*</u>					
480.20	0.423	485.70	0.435	303.20	0.423	595.20	0.368	873.20	0.299	422.20	0.498					
506.20	0.414	509.70	0.427	363.20	0.435	638.20	0.343	923.20	0.316	273.20	0.451					
527.20	0.427	543.20	0.431	452.20	0.423	693.20	0.356	973.20	0.303	323.20	0.448					
<u>CURVE 117*</u>																
378.20	0.410	<u>CURVE 122</u>			760.20	0.343	1013.20	0.239	1013.20	0.448	423.20	0.448				
400.70	0.406	369.70	0.448	515.20	0.423	803.20	0.326	1023.20	0.259	673.20	0.285					
429.20	0.402	411.20	0.444	593.20	0.423	863.20	0.318	1073.20	0.243	723.20	0.217					
451.20	0.402	442.20	0.444	642.20	0.398	910.20	0.310	1123.20	0.247	773.20	0.211					
481.70	0.406	439.20	0.435	689.20	0.402	973.20	0.414	1173.20	0.251	850.20	0.206					
444.70	0.414	464.70	0.439	743.20	0.393	1022.20	0.276	1223.20	0.259	873.20	0.200					
473.20	0.410	499.20	0.431	781.20	0.385	1102.20	0.268	1273.20	0.268	941.20	0.196					
507.70	0.414	530.70	0.435	816.20	0.364	1147.20	0.280	<u>CURVE 135*</u>			973.20	0.194				
532.70	0.410	550.70	0.427	902.20	0.348	<u>CURVE 128</u>			313.40	0.519	1073.20	0.192				
<u>CURVE 118*</u>																
364.70	0.410	364.70	0.410	954.20	0.418	<u>CURVE 129</u>			303.20	0.543	<u>CURVE 140</u>					
396.20	0.414	396.20	0.414	1030.20	0.305	303.20	0.360	303.20	0.360	303.20	0.647	303.20	0.647			
428.20	0.410	428.20	0.410	1108.20	0.314	361.20	0.360	402.20	0.360	373.20	0.669	323.20	0.606			
455.20	0.410	455.20	0.410	1170.20	0.322	437.20	0.364	492.20	0.364	473.20	0.510	360.20	0.537			
487.20	0.414	416.20	0.439	<u>CURVE 126*</u>			576.20	0.356	323.20	0.406*	573.20	0.427	373.20	0.517		
511.70	0.410	442.20	0.444	303.20	0.431	385.20	0.423	650.20	0.347	373.20	0.406*	416.20	0.451	416.20	0.451	
543.20	0.402	475.20	0.444	418.20	0.423	418.20	0.423	691.20	0.347	473.20	0.406*	473.20	0.390	473.20	0.390	
484.70	0.418	500.70	0.439	470.20	0.431	527.20	0.431	719.20	0.343	573.20	0.402*	491.20	0.375	491.20	0.375	
516.20	0.423	525.20	0.444	527.20	0.431	582.20	0.414	753.20	0.339	673.20	0.377*	573.20	0.331	573.20	0.331	
544.20	0.410	544.70	0.435	641.20	0.410	633.20	0.418	800.20	0.310	773.20	0.343	599.20	0.312	599.20	0.312	
<u>CURVE 119*</u>																
370.20	0.435	370.20	0.435	680.20	0.414	904.20	0.310	973.20	0.289	473.20	0.653	673.20	0.238	673.20	0.238	
401.70	0.431	432.20	0.427	752.20	0.402	972.20	0.372	1073.20	0.264	730.20	0.498	730.20	0.193	730.20	0.193	
432.20	0.427	459.20	0.423	817.20	0.398	1030.20	0.272	1073.20	0.272	573.20	0.418	773.20	0.193	773.20	0.193	
459.20	0.423	495.20	0.427	898.20	0.372	1048.20	0.276	1373.20	0.276	673.20	0.335	829.20	0.193	829.20	0.193	
495.20	0.423	520.70	0.423	960.20	0.423	<u>CURVE 127*</u>			1473.20	0.280	773.20	0.272	873.20	0.187	873.20	0.187
520.70	0.423	496.70	0.427	1031.20	0.310	273.20	0.498	323.20	0.494	373.20	0.496	949.20	0.187	949.20	0.187	
543.20	0.427	544.70	0.435	1055.20	0.326	323.20	0.483	373.20	0.483	473.20	0.496	1073.20	0.187	1073.20	0.187	
<u>CURVE 120*</u>																
368.20	0.439	368.20	0.439	1159.20	0.322	423.20	0.469	473.20	0.452	373.20	0.636	373.20	0.636	373.20	0.636	
399.70	0.427	399.70	0.427	1088.20	0.322	473.20	0.452	523.20	0.431	473.20	0.636	473.20	0.636	473.20	0.636	
426.70	0.431	426.70	0.431	1159.20	0.322	523.20	0.431	573.20	0.414	573.20	0.636	573.20	0.636	573.20	0.636	
451.20	0.435	451.20	0.435	303.20	0.360	573.20	0.414	623.20	0.398	623.20	0.636	623.20	0.636	623.20	0.636	
478.70	0.427	478.70	0.427	363.20	0.360	623.20	0.398	623.20	0.398	623.20	0.636	623.20	0.636	623.20	0.636	
507.70	0.431	507.70	0.431	363.20	0.360	623.20	0.398	623.20	0.398	623.20	0.636	623.20	0.636	623.20	0.636	
548.20	0.423	548.20	0.423	363.20	0.360	623.20	0.398	623.20	0.398	623.20	0.636	623.20	0.636	623.20	0.636	

\* Not shown on plot

DATA TABLE NO. 322 (continued)

T	k	T	k	T	k	T	k
<u>CURVE 141 (cont.)</u>		<u>CURVE 145*</u>		<u>CURVE 151*</u>		<u>CURVE 159*</u>	
473.2	0.357	341.5	0.289	291.2	0.498	25.5	0.109
507.2	0.334	380.6	0.318	<u>CURVE 152*</u>		39.3	0.178
573.2	0.308	456.0	0.343	448.2	0.531	55.1	0.240
589.2	0.299	517.1	0.347	<u>CURVE 153</u>		71.0	0.285
673.2	0.226	533.9	0.347	450.6	0.550	90.5	0.326
719.2	0.194	606.8	0.351	<u>CURVE 154</u>		105.2	0.351
773.2	0.187	657.6	0.351	449.0	0.777	119.8	0.369
807.2	0.185	711.2	0.343	<u>CURVE 155</u>		135.7	0.386
873.2	0.185	<u>CURVE 146*</u>		445.7	0.798	151.4	0.400
931.2	0.186	318.6	0.310	<u>CURVE 156*</u>		152.0	0.400
973.2	0.187	373.2	0.318	329.2	0.488	167.1	0.411
1073.2	0.190	424.6	0.351	376.2	0.477	179.7	0.420
<u>CURVE 142*</u>		499.0	0.351	<u>CURVE 157*</u>		195.2	0.429
298.2	0.452	530.5	0.351	337.2	0.454	211.7	0.432
<u>CURVE 143</u>		603.2	0.351	384.2	0.440	225.2	0.436
<u>CURVE 144*</u>		695.2	0.356	449.2	0.439	239.8	0.439
365.0	0.351	733.2	0.347	674.2	0.382	240.1	0.441
400.0	0.339	<u>CURVE 147*</u>		787.2	0.377	<u>CURVE 159*</u>	
448.3	0.326	358.4	0.301	<u>CURVE 148</u>		285.2	0.502
484.4	0.318	393.4	0.297	291.2	0.715	<u>CURVE 159*</u>	
521.1	0.310	450.0	0.289	<u>CURVE 149</u>		<u>CURVE 159*</u>	
594.1	0.229	519.2	0.276	291.2	0.519	<u>CURVE 159*</u>	
631.8	0.245	584.1	0.268	<u>CURVE 150*</u>		<u>CURVE 159*</u>	
<u>CURVE 144*</u>		626.8	0.255	291.2	0.515	<u>CURVE 159*</u>	
349.7	0.285	750.0	0.238	<u>CURVE 151*</u>		<u>CURVE 159*</u>	
388.5	0.276	<u>CURVE 148</u>		337.2	0.454	<u>CURVE 159*</u>	
429.3	0.276	291.2	0.715	384.2	0.440	<u>CURVE 159*</u>	
475.0	0.272	<u>CURVE 149</u>		449.2	0.439	<u>CURVE 159*</u>	
544.7	0.259	291.2	0.519	568.2	0.412	<u>CURVE 159*</u>	
648.7	0.234	<u>CURVE 150*</u>		674.2	0.382	<u>CURVE 159*</u>	
700.2	0.230	291.2	0.515	787.2	0.377	<u>CURVE 159*</u>	
738.4	0.226	<u>CURVE 151*</u>		<u>CURVE 159*</u>			
814.3	0.209	337.2	0.454	<u>CURVE 159*</u>			
		384.2	0.440	<u>CURVE 159*</u>			
		449.2	0.439	<u>CURVE 159*</u>			
		568.2	0.412	<u>CURVE 159*</u>			
		674.2	0.382	<u>CURVE 159*</u>			
		787.2	0.377	<u>CURVE 159*</u>			

\* Not shown on plot

FIGURE SHOWS ONLY 15 OF THE CURVES REPORTED IN TABLE

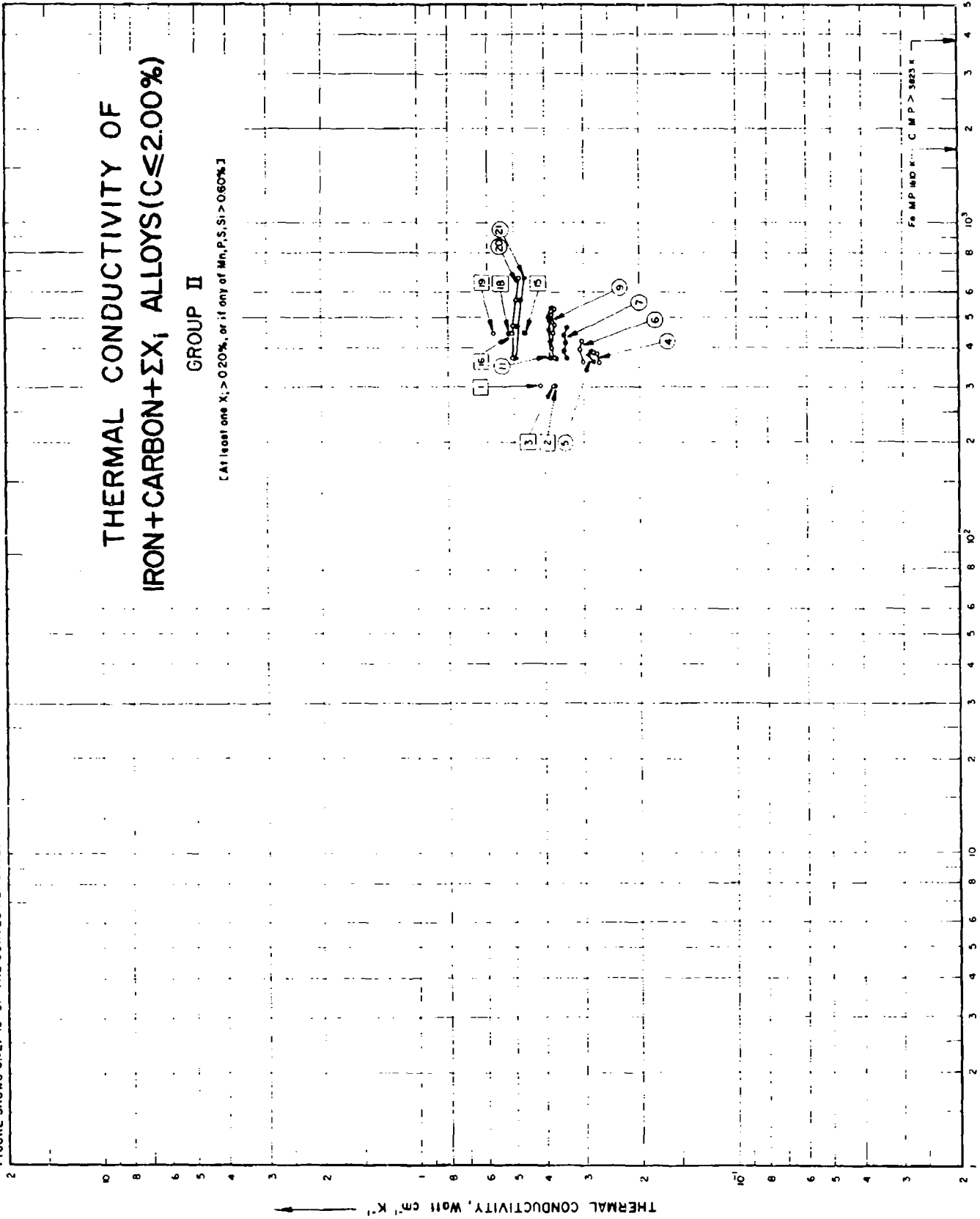


FIG. 363

(At least one  $X_i > 0.20\%$  or if any of Mn, P, S, Si  $> 0.60\%$ )

[For Data Reported in Figure and Table No. 323]

Curve No.	Ref. Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight per cent)							Composition (continued), Specifications and Remarks	
						C	Cr	Mn	P	S	Si			
1	176	E	1920	303	2a	0.6	0.5							Annaled at 900 C and slowly cooled.
2	176	E	1920	303	2b	0.6	0.5							Annaled at 1100 C and quickly cooled.
3	188	E	1919	303	2b	0.6								0.5 W; annealed at 900 C and slowly cooled.
4	189	L	1934	360-389	Oil-hardening non-deforming steel S <sub>Q</sub>	0.789	0.645	0.592	0.022	0.020	0.177			Oil-quenched from 790 C.
5	189	L	1934	363-391	Oil-hardening non-deforming steel S <sub>T</sub> 150 C	0.789	0.645	0.592	0.022	0.020	0.177			Tempered for 30 min at 150 C and air-cooled.
6	189	L	1934	360-421	Oil-hardening non-deforming steel S <sub>T</sub> 200 C	0.789	0.645	0.592	0.022	0.020	0.177			Tempered for 30 min at 200 C and air-cooled.
7	189	L	1934	371-466	Oil-hardening non-deforming steel S <sub>T</sub> 250 C	0.789	0.645	0.592	0.022	0.020	0.177			Tempered for 30 min at 250 C and air-cooled.
8	189	L	1934	367-499	Oil-hardening non-deforming steel S <sub>T</sub> 300 C	0.789	0.645	0.592	0.022	0.020	0.177			Tempered for 30 min at 300 C and air-cooled.
9	189	L	1934	370-532	Oil-hardening non-deforming steel S <sub>T</sub> 350 C	0.789	0.645	0.592	0.022	0.020	0.177			Tempered for 30 min at 350 C and air-cooled.
10	189	L	1934	371-545	Oil-hardening non-deforming steel S <sub>T</sub> 400 C	0.789	0.645	0.592	0.022	0.020	0.177			Tempered for 30 min at 400 C and air-cooled.
11	189	L	1934	372-540	Oil-hardening non-deforming steel S <sub>T</sub> 500 C	0.789	0.645	0.592	0.022	0.020	0.177			Tempered for 30 min at 500 C and air-cooled.
12	189	L	1934	371-535	Oil-hardening non-deforming steel S <sub>T</sub> 600 C	0.789	0.645	0.592	0.022	0.020	0.177			Tempered for 30 min at 600 C and air-cooled.
13	189	L	1934	371-538	Oil-hardening non-deforming steel S <sub>T</sub> 700 C	0.789	0.645	0.592	0.022	0.020	0.177			Tempered for 30 min at 700 C and air-cooled.
14	189	L	1934	371-544	Oil-hardening non-deforming steel S <sub>A</sub>	0.789	0.645	0.592	0.022	0.020	0.177			Annaled at 730 C.

SPECIFICATION TABLE NO. 123 (continued)

Carve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)							Composition (continued), Specifications and Remarks	
							C	Cr	Mn	P	S	Si	Ni		
15	561	C	1925	445			0.02		0.22				0.29	0.27 Ni	
16	561	C	1925	450			1.50		0.24				0.35	0.24 Ni	
17	561	C	1925	440		Carbon steel	0.95		0.22				0.24	0.39 Ni	quenched in NaOH solution from 850 C.
18	561	C	1925	441		Carbon steel	0.95		0.22				0.24	0.39 Ni	tempered at 450 C.
19	561	C	1925	447		Carbon steel	0.95		0.22				0.24	0.39 Ni	annealed at 850 C.
20	976	I	1933	373-673		Soft steel	0.92		0.56		0.032		0.15		No details reported.
21	976	I	1933	373-673		Soft steel	1.09		0.46		0.034		0.06		No details reported.

DATA TABLE NO. 323 THERMAL CONDUCTIVITY OF [IRON + CARBON +  $\Sigma X_i$ ] ALLOYS (C  $\approx$  0.00%) GROUP II(At least one  $N_i > 0.20\%$  or if any of Mn, P, S, Si  $> 0.60\%$ )[Temperature, T, K; Thermal Conductivity, k, Watt  $\text{cm}^{-1} \text{K}^{-1}$ ]

T	k	T	k	T	k	T	k
<u>CURVE 1</u>		<u>CURVE 2</u>		<u>CURVE 13 (cont.)</u>		<u>CURVE 21 (cont.)</u>	
303.20	0.416	369.70	0.368	488.70	0.389	473.2	0.485
<u>CURVE 2</u>		399.70	0.381	506.20	0.385	573.2	0.473
371.70	0.372	450.20	0.377	538.20	0.393	673.2	0.464
303.20	0.371	474.70	0.372	<u>CURVE 14<sup>s</sup></u>			
<u>CURVE 3</u>		511.70	0.381	370.70	0.381		
303.20	0.374	532.20	0.372	396.70	0.385		
<u>CURVE 4</u>		<u>CURVE 10<sup>s</sup></u>		475.20	0.389		
359.70	0.268	370.70	0.368	488.70	0.389		
374.70	0.272	393.20	0.377	483.20	0.381		
389.20	0.280	410.70	0.372	517.70	0.385		
<u>CURVE 5</u>		449.70	0.377	543.70	0.389		
359.70	0.401	476.70	0.368	<u>CURVE 15</u>			
394.70	0.310	515.20	0.372	444.8	0.464		
420.70	0.305	547.70	0.364	<u>CURVE 16</u>			
<u>CURVE 6</u>		<u>CURVE 11</u>		<u>CURVE 17<sup>c</sup></u>			
371.70	0.385	371.70	0.385	449.7	0.510		
391.20	0.387	402.70	0.385	<u>CURVE 18</u>			
416.70	0.383	428.20	0.385	440.3	0.397		
439.70	0.387	451.70	0.381	<u>CURVE 19</u>			
466.20	0.339	477.70	0.385	440.7	0.518		
<u>CURVE 7</u>		503.70	0.381	<u>CURVE 20</u>			
370.70	0.339	535.20	0.372	373.2	0.502		
391.20	0.387	<u>CURVE 13</u>		473.2	0.498		
416.70	0.383	371.20	0.385	573.2	0.490		
439.70	0.387	401.70	0.389	673.2	0.481		
466.20	0.339	459.20	0.372	<u>CURVE 21</u>			
<u>CURVE 8<sup>c</sup></u>		482.70	0.377	373.2	0.494		
366.70	0.372	498.70	0.381				
394.20	0.385						
424.20	0.377						
459.20	0.372						
482.70	0.377						
498.70	0.381						

<sup>s</sup> Not shown on plot

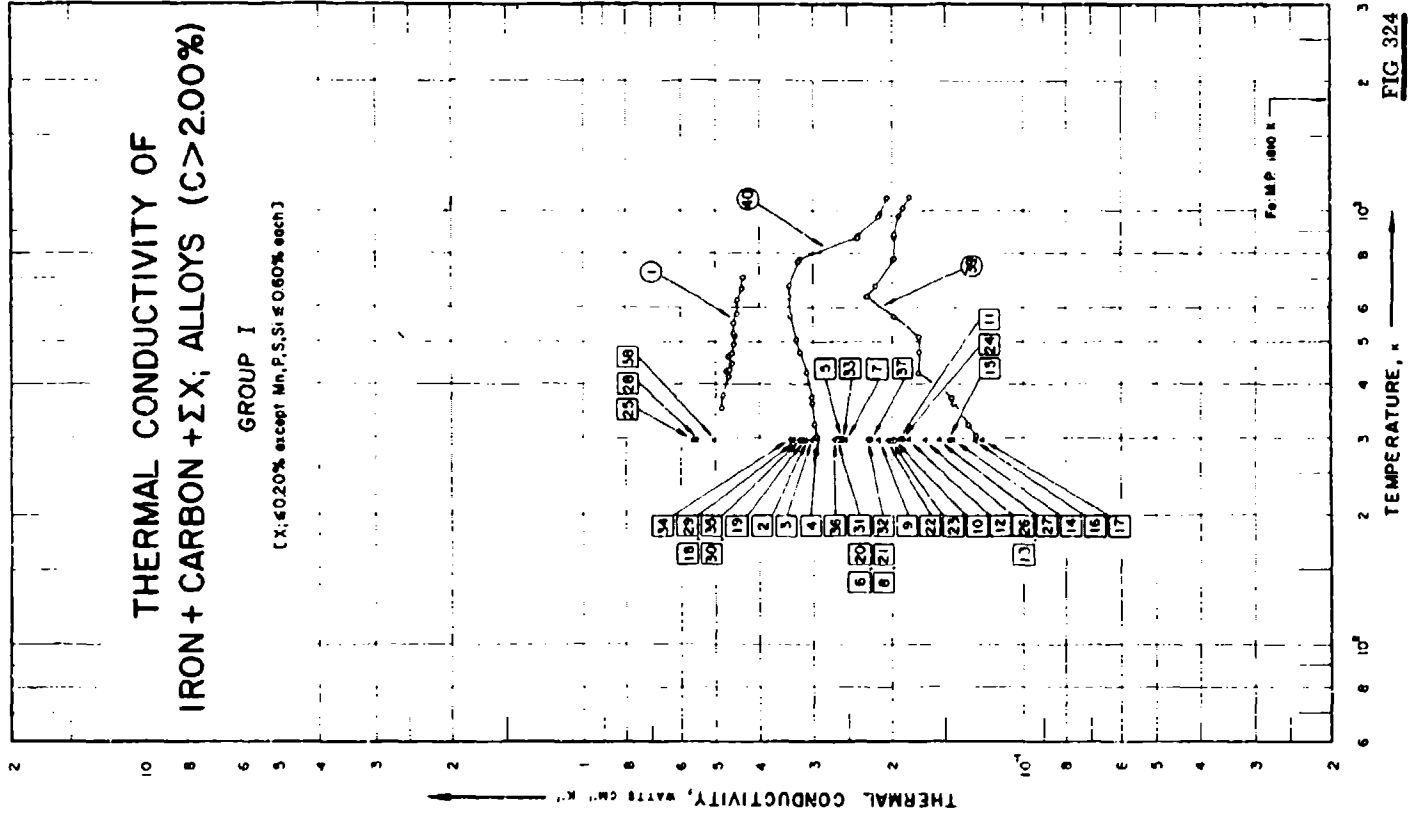


FIG 324

SPECIFICATION TABLE NO. 324 THERMAL CONDUCTIVITY OF [IRON + CARBON + SX] ALLOYS (C > 2.00%) GROUP I

(X<sub>1</sub> = 0.20% except Mn, P, S, Si = 0.60% each)

[For Data Reported in Figure and Table No. 324]

Curve No.	Ref. No. Used	Method Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	C	Composition (weight per cent)					Composition (continued), Specifications and Remarks
							Cu	Mn	P	S	Si	
1	31	L	1933	354-704	2.0	WM	2.80	0.19	0.061	0.093	0.39	[0.76 graphitic carbon, 2.0% combined carbon].
2	172	E	1927	298	Cast iron: 1a	4.41	Trace	0.05	0.036	0.093	0.12	As cast.
3	172	E	1927	298	2a	2.53	Trace	0.02	0.014	0.029	0.05	As cast.
4	172	E	1927	298	3a	2.67	Trace	0.02	0.033	0.048	0.11	As cast.
5	172	E	1927	298	4a	3.12	Trace	0.05	0.024	0.055	0.06	As cast.
6	172	E	1927	298	5a	3.14	Trace	0.03	0.019	0.030	0.01	As cast.
7	172	E	1927	298	6a	3.17	Trace	0.08	0.040	0.037	0.21	As cast.
8	172	E	1927	298	7a	3.53	Trace	0.05	0.009	0.032	0.04	As cast.
9	172	E	1927	298	8a	3.64	Trace	0.04	0.021	0.024	0.16	As cast.
10	172	E	1927	298	9a	3.93	Trace	0.04	0.020	0.049	0.15	As cast.
11	172	E	1927	298	10a	3.96	Trace	0.06	0.011	0.021	0.20	As cast.
12	172	E	1927	298	11a	4.13	Trace	0.03	0.017	0.023	0.10	As cast.
13	172	E	1927	298	12a	4.26	Trace	0.03	0.019	0.020	0.10	As cast.
14	172	E	1927	298	13a	4.35	Trace	0.08	0.022	0.023	0.35	As cast.
15	172	E	1927	298	14a	4.40	Trace	0.03	0.019	0.075	0.34	As cast.
16	172	E	1927	298	15a	4.61	Trace	0.03	0.017	0.040	0.37	As cast.
17	172	E	1927	298	16a	4.63	Trace	0.08	0.020	0.074	0.54	As cast.
18	172	E	1927	298	1b	2.41	Trace	0.05	0.036	0.093	0.12	Cast; annealed at 1000 C for 2 hrs.
19	172	E	1927	298	3b	2.67	Trace	0.02	0.033	0.048	0.11	Cast; annealed at 1000 C for 2 hrs.
20	172	E	1927	298	6b	3.17	Trace	0.08	0.040	0.057	0.21	Cast; annealed at 1000 C for 2 hrs.
21	172	E	1927	298	8b	3.64	Trace	0.04	0.021	0.024	0.16	Cast; annealed at 1000 C for 2 hrs.
22	172	E	1927	298	10b	3.96	Trace	0.06	0.011	0.021	0.20	Cast; annealed at 1000 C for 2 hrs. Small graphite appeared in granular form; annealed at 1000 C for 2 hrs.
23	172	E	1927	298	11b	4.13	Trace	0.03	0.017	0.023	0.10	Annealed for 3 hrs at 650 C.
24	172	E	1927	298	12b	4.26	Trace	0.03	0.019	0.020	0.10	Annealed for 3 hrs at 650 C.
25	172	E	1927	298	13b	4.35	Trace	0.08	0.022	0.023	0.35	3.89 graphite in granular form; annealed at 1000 C for 2 hrs.
26	172	E	1927	298	14b	4.40	Trace	0.03	0.019	0.075	0.34	Annealed for 3 hrs at 650 C.
27	172	E	1927	298	15b	4.61	Trace	0.03	0.017	0.040	0.37	Annealed for 3 hrs at 650 C.
28	172	E	1927	298	16b	4.63	Trace	0.08	0.020	0.074	0.54	4.55 graphite in granular form; annealed at 1000 C for 2 hrs.
29	172	E	1927	298	1c	2.41	Trace	0.05	0.036	0.093	0.12	Annealed first at 1000 C for 2 hrs and second at 1000 C for 4 hrs.



SPECIFICATION TABLE NO. 321 (continued)

Curve No.	Ref. Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight per cent)						Composition (continued), Specifications and Remarks
						C	Cu	Mn	P	S	Si	
30	172	E	298		Cast iron; 3c	2.67	Trace	0.02	0.033	0.048	0.11	Annealed first at 1000 C for 2 hrs and second at 1000 C for 4 hrs.
31	172	E	298		6c	3.17	Trace	0.08	0.040	0.057	0.21	Annealed first at 1000 C for 2 hrs and second at 1000 C for 4 hrs.
32	172	E	298		8c	3.64	Trace	0.04	0.021	0.024	0.16	Annealed first at 1000 C for 2 hrs and second at 1000 C for 4 hrs.
33	172	E	298		10c	3.96	Trace	0.06	0.011	0.021	0.20	1. 10 graphite in granular form; annealed first at 1000 C for 2 hrs and second at 1000 C for 4 hrs.
34	172	E	298		1d	2.41	Trace	0.05	0.036	0.093	0.12	Annealed first at 1000 C for 2 hrs and second at 1000 C for 4 hrs and third at 1090 C for 2 hrs.
35	172	E	298		3d	2.67	Trace	0.02	0.033	0.048	0.11	Annealed first at 1000 C for 2 hrs and second at 1000 C for 4 hrs and third at 1090 C for 2 hrs.
36	172	E	298		6d	3.17	Trace	0.08	0.040	0.057	0.21	Annealed first at 1000 C for 2 hrs and second at 1000 C for 4 hrs and third at 1090 C for 2 hrs.
37	172	E	298		8d	3.64	Trace	0.04	0.021	0.024	0.16	Annealed first at 1000 C for 2 hrs and second at 1000 C for 4 hrs and third at 1090 C for 2 hrs.
38	172	E	298		10d	3.96	Trace	0.06	0.011	0.021	0.20	3. 02 graphite in granular form; annealed first at 1000 C for 2 hrs and second at 1000 C for 4 hrs and third at 1090 C for 2 hrs.
39	175	P	303-1073		White cast iron; 1	3.02	0.089	0.53	0.567	0.074	0.57	
40	175	P	303-1073		Grey cast iron; 4	3.08	0.136	0.44	0.540	0.074	0.58	

GROUP 1

(C &gt; 2.00%)

## DATA TABLE NO. 324 THERMAL CONDUCTIVITY OF [IRON + CARBON + ΣX] ALLOYS

(X<sub>i</sub> ≤ 0.20% except Mn, P, S, Si ≤ 0.60% each)[Temperature, T, K; Thermal Conductivity, k, Watts cm<sup>-1</sup>K<sup>-1</sup>]

T	k	T	k	T	k	T	k
<u>CURVE 1</u>							
353.70	0.485						
376.70	0.481						
418.20	0.469						
429.70	0.473						
431.70	0.469						
447.20	0.460						
464.70	0.469						
472.20	0.460						
496.20	0.456						
515.70	0.452						
527.70	0.456						
552.20	0.456						
581.20	0.448						
625.70	0.448						
666.70	0.439						
703.70	0.435						
<u>CURVE 2</u>							
298.20	0.319						
<u>CURVE 3</u>							
298.20	0.303						
<u>CURVE 4</u>							
298.20	0.294						
<u>CURVE 5</u>							
298.20	0.260						
<u>CURVE 6</u>							
298.20	0.264						
<u>CURVE 7</u>							
298.20	0.254						
<u>CURVE 8</u>							
298.20	0.226						
<u>CURVE 9</u>							
298.20	0.215						
<u>CURVE 10</u>							
298.20	0.196						
<u>CURVE 11</u>							
298.20	0.191						
<u>CURVE 12</u>							
298.20	0.182						
<u>CURVE 13</u>							
298.20	0.167						
<u>CURVE 14</u>							
298.20	0.145						
<u>CURVE 15</u>							
298.20	0.146						
<u>CURVE 16</u>							
298.20	0.129						
<u>CURVE 17</u>							
298.20	0.124						
<u>CURVE 18</u>							
298.20	0.335						
<u>CURVE 19</u>							
298.20	0.318						
<u>CURVE 20</u>							
298.20	0.265						
<u>CURVE 21</u>							
298.20	0.226						
<u>CURVE 22</u>							
298.20	0.205						
<u>CURVE 23</u>							
298.20	0.200						
<u>CURVE 24</u>							
298.20	0.189						
<u>CURVE 25</u>							
298.20	0.167						
<u>CURVE 26</u>							
298.20	0.167						
<u>CURVE 27</u>							
298.20	0.155						
<u>CURVE 28</u>							
298.20	0.557						
<u>CURVE 29</u>							
298.20	0.334						
<u>CURVE 30</u>							
298.20	0.321						
<u>CURVE 31</u>							
298.20	0.265						
<u>CURVE 32</u>							
298.20	0.226						
<u>CURVE 33</u>							
298.20	0.256						
<u>CURVE 34</u>							
298.20	0.339						
<u>CURVE 35</u>							
298.20	0.322						
<u>CURVE 36</u>							
298.20	0.322						
<u>CURVE 37</u>							
298.20	0.268						
<u>CURVE 38</u>							
298.20	0.503						
<u>CURVE 39</u>							
303.20	0.128						
323.20	0.133						
362.20	0.143						
373.20	0.145						
425.20	0.173						
473.20	0.172						
512.20	0.172						
573.20	0.197						
636.20	0.228						
673.20	0.218						
773.20	0.199						
775.20	0.198						
873.20	0.198						
884.20	0.198						
973.20	0.192						
1015.20	0.188						
1073.20	0.181						
<u>CURVE 40</u>							
303.20	0.295						
323.20	0.297						
361.20	0.300						
373.20	0.301						
427.20	0.311						
473.20	0.322						
504.20	0.329						
573.20	0.338						
634.20	0.341						
673.20	0.341						
769.20	0.325						
773.20	0.323						
873.20	0.238						
876.20	0.238						
973.20	0.215						
987.20	0.214						
1073.20	0.205						

FIGURE SHOWS ONLY 46 OF THE CURVES REPORTED IN TABLE

**THERMAL CONDUCTIVITY OF  
IRON + CARBON + ΣX; ALLOYS (C > 2.00%)**

**GROUP II**

[At least one X; > 0.20%, or if any of Mn, P, S, Si > 0.60%]

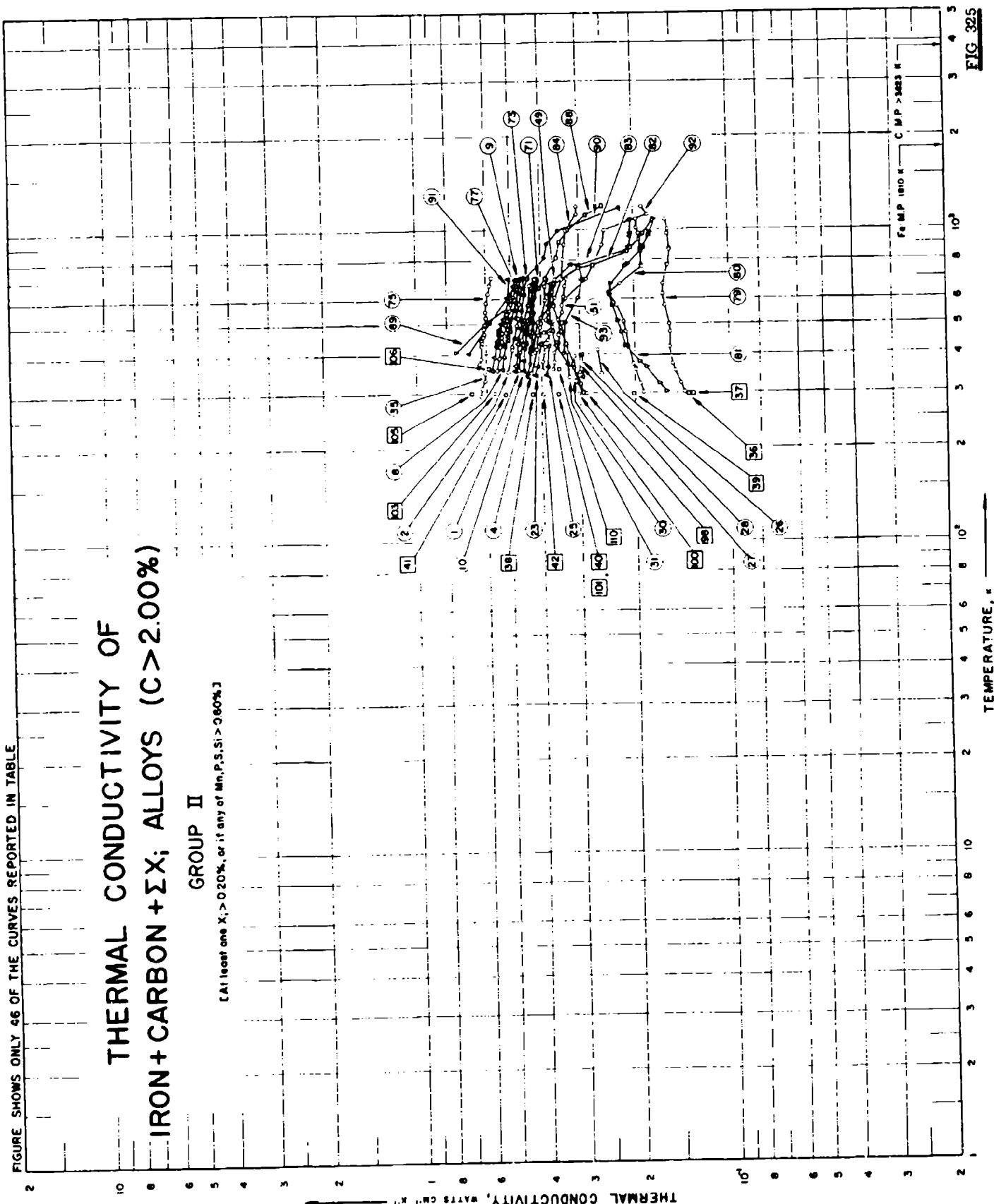


FIG 325

SPECIFICATION TABLE NO. 325 THERMAL CONDUCTIVITY OF [IRON + CARBON + ΣX<sub>i</sub>] ALLOYS (C > 2.00%) GROUP II

(At least one X<sub>i</sub> > 0.20% or if any of Mn, P, S, Si > 0.60%)

[For data Reported in Figure and Table 325]

Curve No.	Ref. Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight per cent)										Composition (continued), Specifications and Remarks	
						C	Cr	Cu	Mn	Ni	P	S	Si				
1	179	L	1939	357-698	2.0	Cu cast iron	3.18		1.58	0.69						1.58	Cast in mold.
2	179	L	1939	352-707	2.0	Cr-Mo cast iron	3.12	0.54		0.38						2.31	0.77 Mo; cast in mold.
3	179	L	1939	356-709	2.0	Mo cast iron	2.56			0.63						2.20	0.58 Mo; cast in mold.
4	179	L	1939	349-696	2.0	Ni-Tensyl cast iron	2.80	0.54		0.66	1.71					2.31	Cast in mold.
5	179	L	1939	355-710	2.0	Ni-Cr cast iron	3.41	0.54		0.65	1.49					1.03	Cast in mold.
6	196	L	1928	361-701	2.0	HD cast iron	3.25			1.87				0.69	0.027	1.56	[2.51 graphitic carbon, 0.74 combined carbon]; as cast.
7	196	L	1928	367-693	2.0	Gray cast iron	3.32			2.43				0.71	0.014	1.52	[2.55 C. C., 0.77 C. C.]; as cast.
8	196	L	1928	359-701	2.0	Cr cast iron.1	3.19	0.198		0.96				0.70	0.049	1.42	[2.49 G. C., 0.70 C. C.]; as cast.
9	196	L	1928	358-697	2.0	Cr cast iron.2	3.17	0.392		0.97				0.69	0.40	1.40	[2.24 G. C., 0.93 C. C.]; as cast.
10	196	L	1928	356-703	2.0	Ni cast iron	3.16			0.94	0.746			0.67	0.095	1.56	[2.50 G. C., 0.67 C. C.]; as cast.
11	196	L	1928	358-707	2.0	V cast iron	3.19			0.99				0.69	0.084	1.45	0.124 V; [2.49 G. C., 0.70 C. C.]; as cast.
12	196	L	1928	353-706	2.0	W cast iron	3.02			0.76				0.68	0.064	1.89	0.475 W; [2.24 C. C., 0.78 C. C.]; as cast.
13	196	L	1928	353-702	2.0	M-4-1	3.34			2.43				0.71	0.014	1.52	[3.08 G. C., 0.26 C. C.]; annealed at 550 C for 200 hrs.
14	196	L	1928	359-702	2.0	Cr-2-1	3.21	0.392		0.97				0.69	0.40	1.40	[2.72 G. C., 0.49 C. C.]; annealed at 550 C for 200 hrs.
15	196	L	1928	362-703	2.0	Ni cast iron.1	3.15			0.94	0.746			0.67	0.095	1.56	[3.08 G. C., 0.07 C. C.]; annealed at 550 C for 200 hrs.
16	196	L	1928	361-701	2.0	W cast iron.1	3.05			0.76				0.68	0.064	1.89	0.475 W; [2.45 G. C., 0.60 C. C.]; annealed at 550 C for 200 hrs.
17	197	L	1940	441-668		Cu-33	3.15			1.45	0.58			0.23	0.11	1.58	
18	197	L	1940	367-712		Cu-34	3.18			1.98	0.58			0.23	0.11	1.49	
19	197	L	1940	370-697		Cu-35	3.16			3.10	0.58			0.23	0.11	1.44	
20	198	L	1939	337-437		Cu cast iron.C2a	3.47			1.36	0.468			0.68	0.097	2.03	
21	198	L	1939	335-437		Cu cast iron.C2b	3.47			1.36	0.468			0.68	0.097	2.03	Specimen 35 mm in diameter.
22	198	L	1939	345-448		Cu cast iron.C5a	2.37			1.16	0.447			1.43	0.093	2.05	Specimen 75 mm in diameter.
23	198	L	1939	342-437		Cu cast iron.C5b	2.37			1.16	0.447			1.43	0.093	2.05	Specimen 35 mm in diameter.
24	198	L	1939	347-446		Cu cast iron.C6a	3.31			1.26	0.468			1.56	0.102	2.00	Specimen 75 mm in diameter.
25	198	L	1939	341-429		Cu cast iron.C6b	3.31			1.26	0.468			1.56	0.102	2.00	Specimen 75 mm in diameter.
26	198	L	1939	347-415		Mn cast iron.P2	2.6/2.9			1.20				1.96	0.06/2.0		Specimen 35 mm in diameter.
27	198	L	1939	351-399		Mn cast iron.P5	2.6/2.9			1.20				1.06	0.06/2.0		
28	198	L	1939	338-399		Mn cast iron.P9	2.6/2.9			1.20				0.92	0.06/2.0		

SPECIFICATION TABLE NO. 325 (continued)

Curve No.	Ref. Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight per cent)										Composition (continued), Specifications and Remarks	
						C	Cr	Cu	Mn	Ni	P	S	Si				
29	198	L	1939	342-408	Mncast iron P10	2.6/2.9				1.20				0.92	0.06/0.10	2.0	
30	198	L	1939	337-394	Mncast iron P11	2.6/2.9				1.20				1.16	0.06/0.10	2.0	
31	198	L	1939	338-405	Mncast iron P12	2.6/2.9				1.20				1.16	0.06/0.10	2.0	
32	197	L	1940	436-661	30	3.16				0.57				0.22	0.11	1.54	
33	197	L	1940	433-672	31	3.2				0.53	0.57			0.22	0.11	1.50	
34	197	L	1940	433-672	32	3.18				0.99	0.58			0.23	0.11	1.59	
35	199	L	1899	294-348		3.40/3.60				0.05/0.055	0.50/0.55			0.053/0.058	0.106	1.40	
36	172	E	1927	298	17a	3.82				Trace	0.09			0.009	0.056	1.24	No graphite, cast in iron mold.
37	172	E	1927	298	18a	3.81				Trace	0.05			0.003	0.05	1.96	No graphite, cast in iron mold.
38	172	E	1927	298	19a	3.84				Trace	0.06			0.004	0.021	1.98	3.26 graphite in flaky form; cast in sand mold.
39	172	E	1927	298	17b	3.82				Trace	0.09			0.009	0.056	1.24	Cast, and annealed at 800 C for 1 hr.
40	172	E	1927	298	18b	3.81				Trace	0.05			0.003	0.050	1.96	Cast, and annealed at 800 C for 1 hr.
41	172	E	1927	298	19b	3.84				Trace	0.06			0.004	0.012	1.98	Cast, and annealed at 800 C for 1 hr.
42	172	E	1927	298	17c	3.82				Trace	0.09			0.009	0.056	1.24	Cast, and annealed at 800 C for 1 hr.
43	198	L	1939	339-437	C1a	3.42				0.15	0.447			0.66	0.10	2.10	Annealed first at 800 C for 1 hr, then at 1000 C for 1 hr.
44	198	L	1939	342-436	C1b	3.42				0.15	0.447			0.66	0.10	2.10	Specimen 75 mm in diameter.
45	198	L	1939	334-425	C3a	3.37				0.45	0.436			0.82	0.093	2.07	Specimen 35 mm in diameter.
46	198	L	1939	336-419	C3b	3.37				0.45	0.436			0.82	0.093	2.07	Specimen 75 mm in diameter.
47	198	L	1939	337-436	C4a	3.18				0.15	0.468			1.20	0.10	2.10	Specimen 35 mm in diameter.
48	198	L	1939	341-452	C4b	3.18				0.15	0.468			1.20	0.10	2.10	Specimen 75 mm in diameter.
49	200	L	1932	356-698	Ingot iron; 1	3.77				0.43				0.037	0.049	1.91	Iron of ingot mold before service.
50	200	L	1932	349-699	2	3.48/3.49				0.73/0.74				0.042	0.039	2.05/2.08	[2.70 - 2.74 G. C., 0.74 - 0.79 C. C.]; iron of ingot mold before service.
51	200	L	1932	358-700	3T	3.43/3.51				0.78				0.029	0.083	2.44/2.46	[2.60 - 2.90 G. C., 0.61 - 0.83 C. C.]; iron from the top of a used ingot mold having a life of 65 casts.
52	200	L	1932	349-701	3B	3.57/3.58				0.74				0.039	0.032	2.20/2.23	[2.80 - 2.91 G. C., 0.67 - 0.77 C. C.]; iron from the bottom of a used ingot mold having a life of 65 casts.
53	200	L	1932	348-698	1A	3.12				0.43				1.86			[2.48 G. C., 0.64 C. C.]; iron from a used ingot mold having a life of 65 casts and being made of the same metal as that of sample 1 above.

SPECIFICATION TABLE NO. 325 (continued)

Curve No.	Ref. Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight per cent)							S	Si	Composition (continued), Specifications and Remarks
						C	Cr	Cu	Mn	Ni	P	S			
54	200	L	1932	342-700	2A	3.36			0.76				2.12		[2.87 G. C., 0.49 C. C.], iron from a used ingot mold from the same metal as that of sample 2 above.
55	179	L	1939	358-700	2.0 1	3.20			0.72				1.56		As cast.
56	179	L	1939	360-700	2.0 2	3.11			0.39				2.26		As cast.
57	179	L	1939	349-698	2.0 HD cast iron	2.61			0.45				2.46		As cast.
58	129	C	1933	373-773	2.0-5.0 C3	3.93			0.63			0.134	0.077	1.40	[3.34 G. C., 0.59 C. C.]; as cast.
59	129	C	1933	373-773	2.0-5.0 C4	4.16			0.79			0.120	0.040	1.35	[3.50 G. C., 0.66 C. C.]; as cast.
60	196	L	1928	361-706	2.0 Grey cast iron, S6	3.08			0.94			0.36	0.08	1.24	[2.28 G. C., 0.79 C. C.]; as cast.
61	196	L	1928	358-703	2.0 S8	3.16			0.97			0.70	0.054	1.48	[2.48 G. C., 0.68 C. C.]; as cast.
62	196	L	1928	350-691	2.0 S9	3.25			0.97			0.81	0.066	1.91	[2.65 G. C., 0.60 C. C.]; as cast.
63	196	L	1928	365-700	2.0 S-8-1	3.13			0.97			0.70	0.054	1.48	[3.01 G. C., 0.12 C. C.]; tempered at 550 C for 40 hrs.
64	196	L	1928	356-665	2.0 S-8-2	3.16			0.97			0.70	0.054	1.48	[3.05 G. C., 0.11 C. C.]; tempered at 550 C for 80 hrs.
65	196	L	1928	350-655	2.0 S-8-3	3.15			0.97			0.70	0.054	1.48	[3.06 G. C., 0.09 C. C.]; tempered at 550 C for 120 hrs.
66	196	L	1928	357-665	2.0 S-8-4	3.15			0.97			0.70	0.054	1.48	[3.03 G. C., 0.12 C. C.]; tempered at 550 C for 160 hrs.
67	196	L	1928	356-697	2.0 S-8-5	3.14			0.97			0.70	0.054	1.48	[3.02 G. C., 0.12 C. C.]; tempered at 550 C for 200 hrs.
68	196	L	1928	351-705	2.0 S-6-1	3.07			0.94			0.36	0.08	1.24	[2.77 G. C., 0.30 C. C.]; tempered at 550 C for 200 hrs.
69	196	L	1928	356-696	2.0 S-9-1	3.28			0.97			0.81	0.066	1.91	[3.23 G. C., 0.05 C. C.]; tempered at 550 C for 200 hrs.
70	31	L	1933	346-680	2.0 1	2.89			0.32			0.27	0.046	1.87	
71	31	L	1933	341-684	2.0 3	2.87			0.28			0.28	0.045	2.81	
72	31	L	1933	346-704	2.0 P1	3.34	0.33		0.76			0.18	0.065	1.90	
73	31	L	1933	356-701	2.0 P2	3.40	0.30		0.92			0.59	0.060	1.90	
74	31	L	1933	352-709	2.0 P3	3.30	0.31		1.00			0.95	0.050	2.00	
75	31	L	1933	373-703	2.0 BM	2.96			0.125			0.135	0.080	1.03	[0.13 G. C., 2.23 C. C.].
76	196	L	1928	363-703	2.0 Gray hot mold, S1	3.35			0.85			0.17	0.12	0.65	[2.44 G. C., 0.91 C. C.]; as cast.
77	196	L	1928	366-695	2.0 S-1-1	3.34			0.85			0.17	0.12	0.65	[2.62 G. C., 0.72 C. C.]; tempered at 550 C for 200 hrs.
78	201	R	1952	468-815	Gray soft	3.5			0.64				2.19		Soft cast iron.
79	175	P	1936	303-1073	White cast iron 2	3.16			0.059	0.34		0.329	0.045	0.46	0.34 Mo.
80	175	P	1936	302-1073	White cast iron 3	3.14			0.057	0.37		0.605	0.042	0.47	0.37 Mo.
81	175	P	1936	303-1073	White cast iron 3	2.37			0.071	0.31		0.776	0.036	0.54	0.55 Mo.
82	175	P	1936	303-1073	Grey cast iron 5	3.06			0.087	0.32		0.607	0.049	0.47	0.36 Mo.
83	175	P	1936	303-1073	Grey cast iron 6	3.09			0.184	0.29		0.610	0.048	0.45	0.55 Mo.

SPECIFICATION TABLE NO. 325 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight per cent)										Composition (continued), Specifications and Remarks
							C	Cr	Cu	Mn	Ni	P	S	Si			
84	202	R	1956	408-1193	<+5.0	Cast iron: A	2.98	0.490		1.5	0.130	0.456	0.077	1.94			[0.75 combined carbon]; hypoeutectoid with pearlitic base.
85	202	R	1956	408-693	<+5.0	Cast iron: A	2.98	0.490		1.5	0.130	0.456	0.077	1.94			[0.75 combined carbon]; hypoeutectoid with ferritic base.
86	202	R	1956	393-1193	<+5.0	Cast iron: B	3.95	0.420		0.91	0.090	0.509	0.083	1.82			0.65 W; [0.85 combined carbon]; hypoeutectoid with pearlitic base.
87	202	R	1956	403-703	<+5.0	Cast iron: B	3.05	0.420		0.91	0.090	0.509	0.083	1.82			0.65 W; [0.85 combined carbon]; hypoeutectoid with ferritic base.
88	202	R	1956	408-1193	<+5.0	Cast iron: V	3.82	0.19		0.95	0.15	0.484	0.090	2.02			[0.65 combined carbon]; hypereutectoid with pearlitic base.
89	202	R	1956	408-688	<+5.0	Cast iron: V	3.82	0.19		0.95	0.15	0.484	0.090	2.02			[0.65 combined carbon]; hypereutectoid with ferritic base.
90	202	R	1956	403-1173	<+5.0	Cast iron: G	3.20	0.09		0.88	0.10	0.143	0.116	2.25			[0.61 combined carbon]; eutectic with pearlitic base.
91	202	R	1956	403-698	<+5.0	Cast iron: G	3.20	0.09		0.88	0.10	0.143	0.116	2.25			[0.61 combined carbon]; eutectic with ferritic base.
92	202	R	1956	403-1198	<+5.0	Cast iron: D pl. 25	3.13	0.02		0.94	0.15	0.122	0.01	2.54			0.07 Mg; [0.61 combined carbon]; with pearlitic base.
93	202	R	1956	403-698	<+5.0	Cast iron: D pl. 26	2.95	0.02		1.08	0.18	0.126	0.01	2.90			0.08 Mg; [0.75 combined carbon]; with pearlitic base.
94	202	R	1956	403-693	<+5.0	Cast iron: D pl. 26	2.95	0.02		1.08	0.18	0.126	0.01	2.90			0.08 Mg; [0.75 combined carbon]; with ferritic base.
95	202	R	1956	408-1198	<+5.0	Cast iron: E	2.33	0.06		0.36	0.10	0.143	0.069	1.20			[1.20 combined carbon]; with pearlitic base.
96	202	R	1956	408-693	<+5.0	Cast iron: E	2.33	0.06		0.36	0.10	0.143	0.069	1.20			[1.20 combined carbon]; with ferritic base.
97	202	R	1956	408-708	<+5.0	Cast iron: Zh	3.69	0.05		0.69	0.02	0.28	0.079	1.01			[0.83 combined carbon]; with pearlitic base.
98	203	L	1957	300	+1.5	Cast iron: 2128	≈4.0			0.32				2.4			0.063 Mg; with spherical graphite; annealed at 900 C for 12 hrs and furnace-cooled.
99	203	L	1957	300	+1.5	Cast iron: 2078	≈4.0			0.27	1.0			2.1			0.088 Mg; trace pearlite; with spherical graphite; annealed at 900 C for 12 hrs and furnace-cooled.
100	203	L	1957	300	+1.5	Cast iron: 2131	≈4.0			0.22	1.1			2.1			0.065 Mg; graphite in compact mixed form; annealed at 900 C for 12 hrs and furnace-cooled.
101	203	L	1957	300	+1.5	Cast iron: 2100	≈4.0			0.21	1.1			2.1			0.066 Mg; 40.0 > pearlite; with short lamellar graphite; annealed at 900 C for 3 hrs and furnace-cooled.
102	203	L	1957	300	+1.5	Cast iron: 2076	≈4.0			0.23				2.2			With lamellar graphite; annealed at 900 C for 12 hrs and furnace-cooled.
103	203	L	1957	300	+1.5	Cast iron: 2077	≈4.0			0.20	1.1			2.2			With lamellar graphite; annealed at 900 C for 12 hrs and furnace-cooled.
104	203	L	1957	300	+1.5	Cast iron: 2079	≈4.0			0.26	1.1			2.5			With lamellar graphite; annealed at 900 C for 12 hrs and furnace-cooled.

SPECIFICATION TABLE NO. 325 (continued)

Curve No.	Ref. Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight per cent)										Composition (continued), Specifications and Remarks	
						C	Cr	Cu	Mn	Ni	P	S	Si				
105	203	L	1957	300	+ 1.5 - 1.5	Cast iron, 2132	≈4.0			0.22	1.1					2.2	With lamellar graphite; annealed at 900 C for 12 hrs and furnace-cooled.
106	204	L	1927	362		Pearlitic pig iron, 41	3.12		2.50							1.26	
107	205	C	1953	358		Nodular iron	3.57		0.33	1.33	0.035	0.004	1.12				0.06 Mg; cast.
108	205	C	1953	358		Nodular iron	3.56		0.33	1.30	0.025	0.010	2.27				0.06 Mg; cast.
109	205	C	1953	358		Nodular iron	3.47		0.29	1.30	0.030	0.012	3.53				0.06 Mg; cast.
110	205	C	1953	358		Nodular iron	3.36		0.40	1.23	0.030	0.010	4.34				0.06 Mg; cast.
111	205	C	1953	358		Nodular iron	3.33		0.50	1.12	0.055	0.010	2.28				0.06 Mg; cast.
112	976	L	1933	373-673		Black temper cast	2.36		0.13		0.135	0.080	1.03				94.5 graphite in C.
113	976	L	1933	373-673		White temper cast	2.80		0.10		0.061	0.093	0.39				72.9 graphite in C.
114	976	L	1933	373-673		Cast iron	2.89		0.32		0.27	0.046	1.87				
115	976	L	1933	373-673		Cast iron	2.87		0.28		0.28	0.045	2.81				
116	976	L	1933	373-673		Cast iron	3.34		0.76		0.18	0.065	1.90				
117	976	L	1933	373-673		Cast iron	3.40		0.92		0.59	0.060	1.90				
118	976	L	1933	373-673		Cast iron	3.30		1.00		0.95	0.050	2.00				



DATA TABLE NO. 325 THERMAL CONDUCTIVITY OF [IRON + CARBON +  $\Sigma X_i$ ] ALLOYS (C > 2.00%) GROUP II

(At least one  $X_i > 0.20\%$  or if any of Mn, P, S, Si > 0.60%)

(Temperature, T, K; Thermal Conductivity, k, Watts  $\text{cm}^{-1}\text{K}^{-1}$ )

T	k	T	k	T	k	T	k	T	k	T	k	T	k	T	k	T	k
<u>CURVE 1</u>																	
356.70	0.473	490.20	0.460	506.70	0.456	361.70	0.502	367.30	0.456	346.70	0.421						
407.70	0.456	506.20	0.452	522.20	0.460	397.20	0.494	404.60	0.427	375.70	0.408						
430.70	0.456	523.20	0.460	532.70	0.464	422.70	0.490	435.10	0.448	445.70	0.396						
444.70	0.460	551.70	0.448	566.20	0.452	430.20	0.485	443.90	0.414								
<u>CURVE 5*</u>																	
493.20	0.452	587.20	0.444	600.70	0.436	446.70	0.490	449.90	0.444								
514.70	0.439	624.70	0.435	636.70	0.444	467.70	0.481	482.70	0.431								
518.70	0.432	660.70	0.427	677.20	0.444	476.70	0.477	509.10	0.406	344.20	0.388						
575.20	0.439	693.20	0.414	706.20	0.431	486.70	0.481	532.90	0.402	365.70	0.363						
664.20	0.435	484.70	0.444	505.20	0.439	504.70	0.473	588.60	0.389	428.70	0.379						
697.70	0.418	519.20	0.439	521.70	0.435	524.70	0.469	669.10	0.385								
<u>CURVE 8</u>																	
497.70	0.460	521.70	0.435	558.20	0.469	558.20	0.469	712.40	0.389								
516.20	0.464	549.70	0.431	593.70	0.460	593.70	0.460										
519.70	0.456	601.20	0.427	617.70	0.498	627.20	0.456										
581.70	0.452	640.70	0.427	647.70	0.448	666.70	0.448										
667.70	0.444	681.20	0.423	703.20	0.418	702.70	0.439										
710.20	0.439	458.70	0.510	470.20	0.481	445.20	0.490										
<u>CURVE 11*</u>																	
476.70	0.502	476.70	0.485	486.20	0.481	476.70	0.481	370.50	0.448								
487.20	0.506	508.20	0.502	502.70	0.473	502.70	0.473	411.50	0.448								
508.20	0.502	521.20	0.498	521.70	0.477	521.70	0.477	438.60	0.444								
394.70	0.494	523.70	0.494	531.70	0.498	537.20	0.452	452.80	0.448								
426.20	0.485	557.20	0.485	557.20	0.485	557.20	0.485	494.20	0.448								
426.70	0.485	588.20	0.477	632.70	0.473	635.70	0.435	338.70	0.294								
451.70	0.485	451.70	0.485	450.70	0.494	460.70	0.494	399.20	0.295								
473.20	0.481	473.20	0.481	468.70	0.481	477.20	0.490										
474.20	0.481	474.20	0.481	470.70	0.485	486.70	0.490										
493.70	0.473	493.70	0.473	487.20	0.481	486.70	0.490										
510.70	0.473	509.70	0.473	509.70	0.473	506.20	0.485	342.20	0.301								
524.20	0.464	516.20	0.473	516.20	0.473	515.20	0.481	368.70	0.494								
526.70	0.464	524.20	0.464	524.20	0.469	522.70	0.485	408.20	0.295								
560.70	0.452	560.20	0.456	569.20	0.448	547.20	0.473										
599.20	0.448	594.20	0.448	594.20	0.448	587.20	0.469										
640.20	0.439	635.20	0.444	635.20	0.444	634.20	0.460	335.00	0.490								
674.20	0.435	678.70	0.435	678.70	0.435	668.20	0.452	364.70	0.476								
701.20	0.427	440.70	0.536	446.70	0.544	446.70	0.544	427.20	0.459								
455.20	0.536	455.20	0.536	467.70	0.544	467.70	0.544										
469.20	0.531	469.20	0.531	475.70	0.536	475.70	0.536										
470.70	0.519	470.70	0.519	483.70	0.540	483.70	0.540										
498.20	0.519	498.20	0.519	507.20	0.536	507.20	0.536	338.70	0.326								
518.20	0.494	518.20	0.494	519.70	0.527	519.70	0.527	405.70	0.330								
523.70	0.515	523.70	0.515	525.70	0.485	525.70	0.485										
555.20	0.502	555.20	0.502	559.20	0.481	559.20	0.481										
430.70	0.418	435.70	0.481	435.70	0.481	435.70	0.481	344.20	0.448								
435.70	0.473	435.70	0.473	451.20	0.477	451.20	0.477	515.80	0.431								
453.20	0.469	453.20	0.469	461.70	0.490	461.70	0.490	597.60	0.418								
467.70	0.402	467.70	0.402	467.70	0.402	467.70	0.402	681.10	0.414								
501.70	0.398	468.70	0.469	473.70	0.477	473.70	0.477										
525.20	0.402	474.20	0.473	474.20	0.473	474.20	0.473										

\* Not shown on plot

DATA TABLE NO. 325 (continued)

T	k	T	k	T	k	T	k	T	k	T	k	T	k	T	k	T	k
<u>CURVE 33*</u>		<u>CURVE 43*</u>		<u>CURVE 49 (cont.)</u>		<u>CURVE 52*</u>		<u>CURVE 54 (cont.)</u>		<u>CURVE 58*</u>		<u>CURVE 61 (cont.)</u>		<u>CURVE 64*</u>			
432.70	0.427	339.00	0.512	519.20	0.389	349.20	0.410	477.70	0.435	373.20	0.550	562.20	0.469	356.20	0.561		
510.80	0.414	371.20	0.495	546.20	0.381	384.70	0.406	509.70	0.427	473.20	0.519	588.70	0.460	423.20	0.548		
590.20	0.410	436.70	0.467	582.70	0.377	412.20	0.398	518.20	0.431	573.20	0.488	632.70	0.452	502.20	0.523		
672.10	0.398			623.70	0.372	424.20	0.393	544.70	0.427	673.20	0.458	671.20	0.448	664.70	0.485		
<u>CURVE 34*</u>		<u>CURVE 44*</u>		<u>CURVE 50*</u>		<u>CURVE 53*</u>		<u>CURVE 55*</u>		<u>CURVE 59*</u>		<u>CURVE 62*</u>		<u>CURVE 65*</u>			
432.90	0.439	342.20	0.455	474.70	0.398	474.70	0.398	493.70	0.385	373.20	0.528	349.20	0.464	349.70	0.565		
510.30	0.427	370.20	0.448	493.70	0.381	493.70	0.381	511.20	0.381	473.20	0.502	379.20	0.454	413.70	0.548		
590.30	0.414	437.70	0.432	523.70	0.389	523.70	0.389	547.70	0.381	573.20	0.476	404.20	0.456	492.20	0.523		
671.90	0.410			582.70	0.381	582.70	0.381	619.20	0.381	673.20	0.449	414.20	0.452	655.20	0.481		
<u>CURVE 35</u>		<u>CURVE 45*</u>		<u>CURVE 51*</u>		<u>CURVE 54*</u>		<u>CURVE 56*</u>		<u>CURVE 60*</u>		<u>CURVE 63*</u>		<u>CURVE 66*</u>			
294.10	0.625	333.60	0.486	426.20	0.427	426.20	0.427	482.70	0.418	357.70	0.510	451.20	0.448	357.70	0.548		
312.10	0.521	361.20	0.474	427.20	0.418	427.20	0.418	464.20	0.418	387.20	0.494	463.20	0.460	423.70	0.531		
329.40	0.608	424.70	0.460	455.70	0.414	455.70	0.414	468.70	0.381	425.70	0.515	475.20	0.444	503.20	0.515		
347.90	0.605			468.20	0.418	468.20	0.418	498.20	0.481	468.70	0.502	490.20	0.448	665.70	0.473		
<u>CURVE 36</u>		<u>CURVE 46*</u>		<u>CURVE 52*</u>		<u>CURVE 55*</u>		<u>CURVE 57*</u>		<u>CURVE 61*</u>		<u>CURVE 64*</u>		<u>CURVE 67*</u>			
298.20	0.136	336.20	0.462	470.20	0.410	470.20	0.410	498.20	0.481	361.20	0.536	509.20	0.458	356.70	0.552		
298.20	0.130	360.20	0.457	509.20	0.406	509.20	0.406	520.20	0.481	390.20	0.527	510.70	0.452	393.70	0.544		
298.20	0.136	419.20	0.441	548.70	0.406	548.70	0.406	590.20	0.464	419.20	0.523	543.20	0.444	421.70	0.536		
<u>CURVE 37</u>		<u>CURVE 47*</u>		<u>CURVE 53*</u>		<u>CURVE 56*</u>		<u>CURVE 59*</u>		<u>CURVE 63*</u>		<u>CURVE 66*</u>		<u>CURVE 69*</u>			
298.20	0.130	336.80	0.485	447.70	0.398	447.70	0.398	466.20	0.410	359.70	0.490	510.70	0.490	424.70	0.540		
298.20	0.130	367.70	0.475	466.20	0.410	466.20	0.410	482.20	0.444	411.70	0.456	516.20	0.456	440.20	0.531		
298.20	0.136	436.20	0.446	508.20	0.410	508.20	0.410	520.70	0.448	432.20	0.490	538.20	0.490	458.20	0.527		
<u>CURVE 38</u>		<u>CURVE 48*</u>		<u>CURVE 54*</u>		<u>CURVE 57*</u>		<u>CURVE 60*</u>		<u>CURVE 64*</u>		<u>CURVE 67*</u>		<u>CURVE 70*</u>			
298.20	0.429	341.20	0.486	405.70	0.364	405.70	0.364	431.20	0.452	358.70	0.485	365.70	0.569	474.70	0.519		
298.20	0.429	377.20	0.434	422.20	0.356	422.20	0.356	431.20	0.452	588.70	0.485	395.20	0.556	475.70	0.519		
298.20	0.203	451.70	0.418	438.60	0.356	438.60	0.356	490.70	0.452	594.20	0.477	420.70	0.548	499.20	0.510		
<u>CURVE 39</u>		<u>CURVE 49</u>		<u>CURVE 55*</u>		<u>CURVE 58*</u>		<u>CURVE 61*</u>		<u>CURVE 65*</u>		<u>CURVE 68*</u>		<u>CURVE 71*</u>			
298.20	0.203	356.20	0.398	460.70	0.352	460.70	0.352	477.20	0.531	357.70	0.515	480.20	0.536	624.70	0.477		
298.20	0.353	380.70	0.398	466.20	0.352	466.20	0.352	489.20	0.381	390.20	0.515	517.70	0.527	660.20	0.469		
298.20	0.353	407.70	0.393	506.70	0.347	506.70	0.347	521.70	0.460	420.20	0.502	521.70	0.523	697.20	0.460		
298.20	0.523	434.20	0.398	518.20	0.352	518.20	0.352	545.70	0.448	425.70	0.502	586.20	0.515	351.20	0.565		
298.20	0.523	446.20	0.381	545.70	0.343	545.70	0.343	583.70	0.398	448.20	0.502	586.20	0.515	385.20	0.552		
298.20	0.397	469.20	0.377	576.20	0.343	576.20	0.343	620.20	0.439	462.20	0.494	623.70	0.498	418.20	0.544		
298.20	0.397	490.20	0.368	615.70	0.339	615.70	0.339	663.20	0.431	471.70	0.498	668.70	0.485	422.70	0.548		
298.20	0.397	509.70	0.368	662.70	0.335	662.70	0.335	697.70	0.381	485.70	0.495	700.20	0.473	438.70	0.536		
				700.20	0.331	700.20	0.331			505.20	0.491			452.70	0.536		
										523.70	0.477						
										524.20	0.485						

\* Not shown on plot



DATA TABLE NO. 325 (continued)

T	k	T	k	T	k	T	k
<u>CURVE 92</u>							
408.20	0.345*	608.20	0.502	358.00	0.362	373.2	0.481
508.20	0.351*	708.20	0.473	<u>CURVE 109*</u>			
608.20	0.335	<u>CURVE 110</u>				473.2	0.473
693.20	0.285	<u>CURVE 98</u>				573.2	0.460
893.20	0.251	300.00	0.287	358.00	0.351	673.2	0.443
993.20	0.247	<u>CURVE 99*</u>				<u>CURVE 118*</u>	
1128.20	0.184	300.00	0.292	358.00	0.357	373.2	0.464
1198.20	0.188	<u>CURVE 100</u>				473.2	0.456
<u>CURVE 93</u>							
408.20	0.329	300.00	0.301	373.2	0.628	573.2	0.444
508.20	0.335	<u>CURVE 101</u>				673.2	0.431
608.20	0.301	300.00	0.353	<u>CURVE 113*</u>			
698.20	0.293	<u>CURVE 102*</u>				373.2	0.481
<u>CURVE 94*</u>							
408.20	0.347	300.00	0.628	473.2	0.464	573.2	0.452
508.20	0.343	<u>CURVE 103</u>				673.2	0.439
633.20	0.301	300.00	0.565	<u>CURVE 114*</u>			
<u>CURVE 95*</u>							
408.20	0.469	300.00	0.628	373.2	0.469	473.2	0.460
508.20	0.431	<u>CURVE 104*</u>				573.2	0.448
608.20	0.381	300.00	0.628	673.2	0.439	<u>CURVE 115*</u>	
698.20	0.347	<u>CURVE 105</u>				373.2	0.439
818.20	0.297	300.00	0.665	473.2	0.431	473.2	0.423
893.20	0.276	<u>CURVE 106</u>				573.2	0.410
993.20	0.218	361.50	0.587	673.2	0.410	<u>CURVE 116*</u>	
1123.20	0.180	<u>CURVE 107*</u>				373.2	0.490
1198.20	0.176	358.00	0.377	473.2	0.477	473.2	0.477
<u>CURVE 96*</u>							
408.20	0.502	<u>CURVE 108*</u>				573.2	0.469
508.20	0.494	358.00	0.372	673.2	0.460	<u>CURVE 109*</u>	
608.20	0.431	<u>CURVE 97 (cont.)</u>				373.2	0.490
693.20	0.347	608.20	0.502	473.2	0.477	473.2	0.477
<u>CURVE 97*</u>							
408.20	0.690	708.20	0.473	573.2	0.469	573.2	0.469
518.20	0.598	808.20	0.448	673.2	0.460	673.2	0.460

\* Not shown on plot

# THERMAL CONDUCTIVITY OF IRON + ALUMINUM + ΣX; ALLOYS

## GROUP I

[X: 4.0-20% except C ≤ 2.00% and Mn, P, S, Si ≤ 0.60% each]

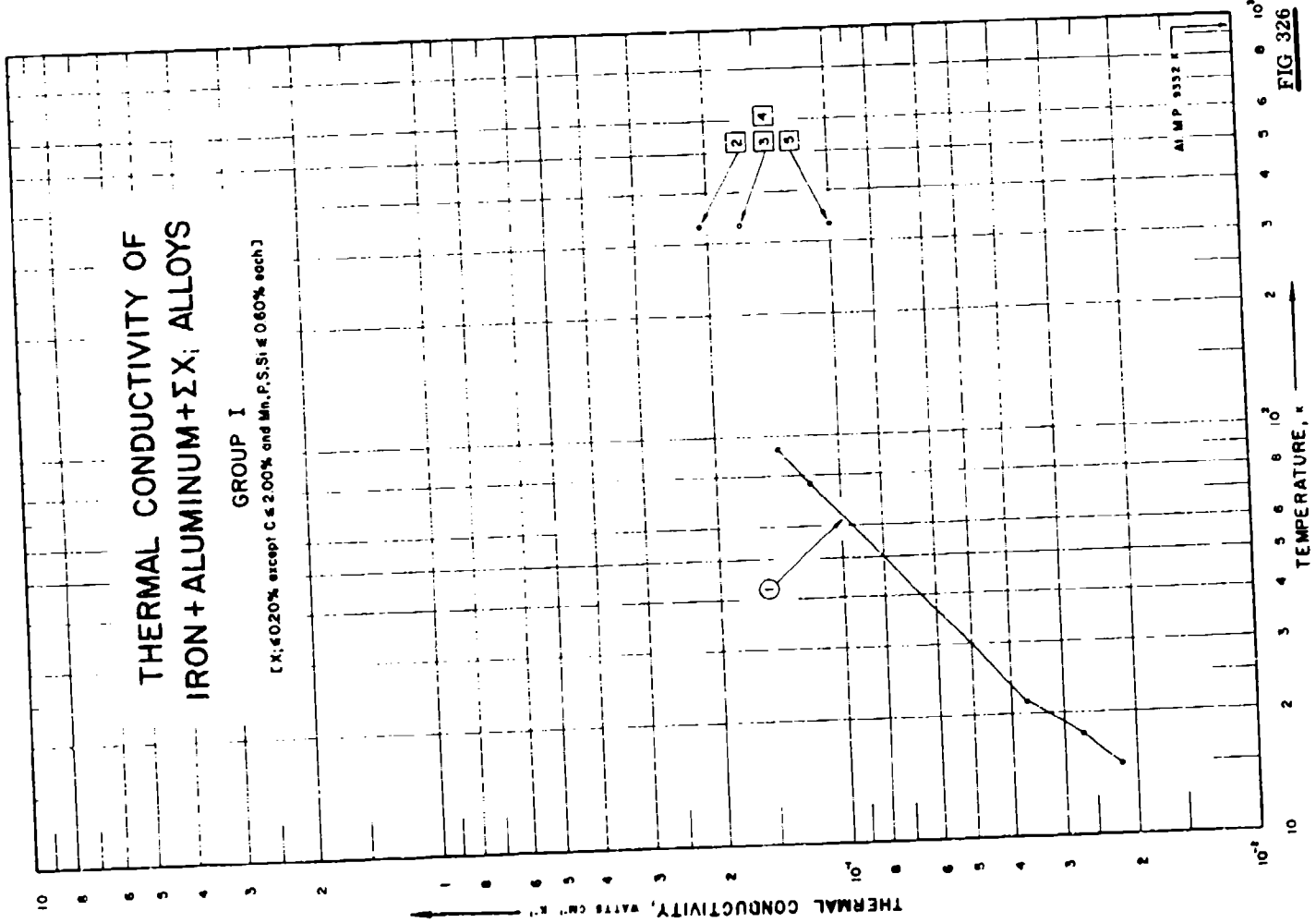


FIG 326

SPECIFICATION TABLE NO. 326 THERMAL CONDUCTIVITY OF [IRON + ALUMINUM +  $\Sigma X_i$ ] ALLOYS GROUP I

( $X_i \leq 0.20\%$  except C  $\leq 2.00\%$  and Mn, P, S, Si  $\leq 0.60\%$  each)

[ For Data Reported in Figure and Table No. 326 ]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)					Composition (continued), Specifications and Remarks		
							Al	C	Mn	P	S		Si	
1	104	L	1951	15-93		3792	Bal	4.11	0.03	0.08	0.017	0.006	0.13	Specimen heated to 800 C and furnace cooled; measured in a vacuo of $5 \times 10^{-4}$ mm Hg. Sand cast.
2	207	F	1934	333		Bal	11.18	<0.1					Sand cast.	
3	207	F	1934	333		Bal	12.39	<0.1					Sand cast.	
4	207	F	1934	333		Bal	14.36	<0.1					Sand cast.	
5	207	F	1934	333		Bal	16.07	<0.1					Sand cast.	

DATA TABLE NO. 326 THERMAL CONDUCTIVITY OF [IRON + ALUMINUM +  $\Sigma X_i$ ] ALLOYS GROUP I(X<sub>i</sub> ≤ 0.20% except C ≤ 2.00% and Mn, P, S, Si ≤ 0.60% each)[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup> K<sup>-1</sup>]

T	k
<u>CURVE 1</u>	
14.97	0.0218
17.71	0.0269
21.50	0.0363
76.60	0.118
93.10	0.142
<u>CURVE 2</u>	
333.20	0.209
<u>CURVE 3</u>	
333.20	0.167
<u>CURVE 4</u>	
333.20	0.167
<u>CURVE 5</u>	
333.20	0.100

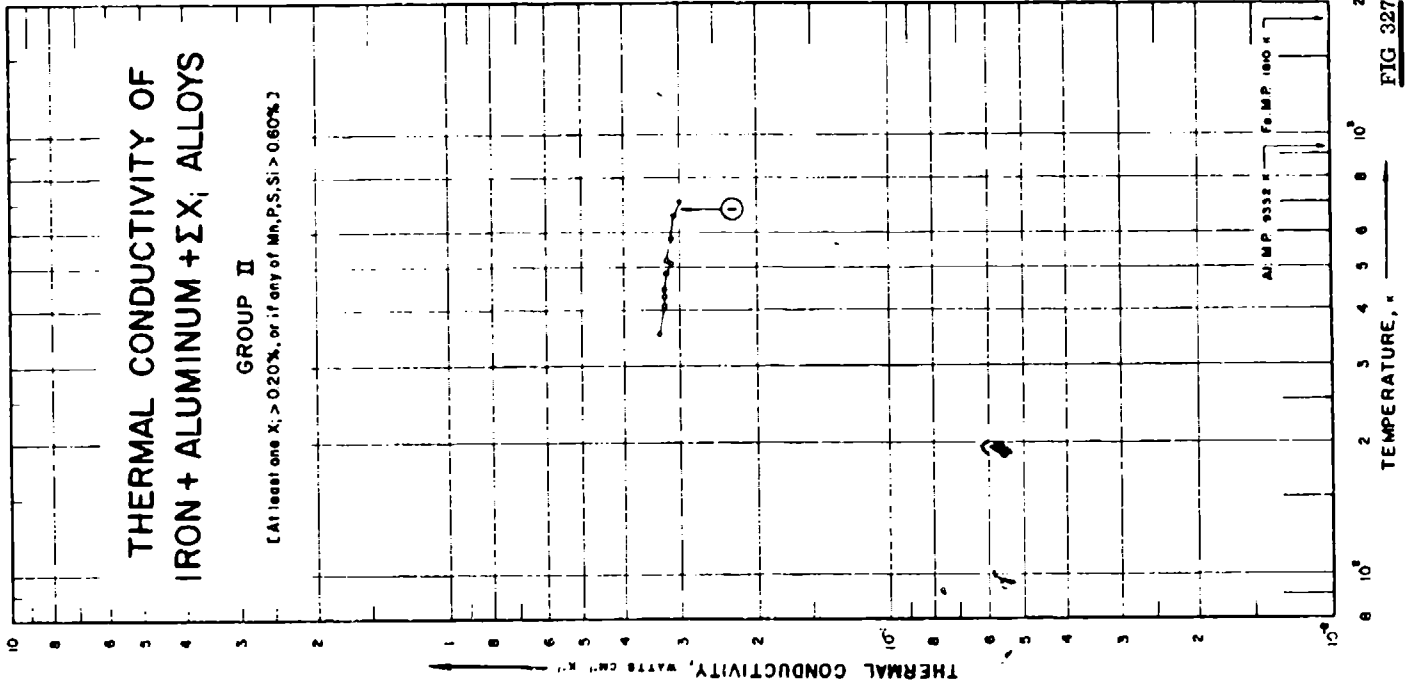


FIG. 327



SPECIFICATION TABLE NO. 327 THERMAL CONDUCTIVITY OF [IRON + ALUMINUM + EX<sub>1</sub>] ALLOYS GROUP II(At least one X<sub>1</sub> > 0.20% or if any of Mn, P, S, Si > 0.60%)

[ For Data Reported in Figure and Table No. 327 ]

Curve No.	Ref. Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)					Composition (continued), Specifications and Remarks	
						Fe	Al	C	Cr	Mn		Si
1	L	1939	357-703	2.0	Heat resistant cast iron	Bal	7.00	2.70	0.95	0.58	0.96	Specimen 0.75 in. in dia and 15.5 in. long; cast.

DATA TABLE NO. 327 THERMAL CONDUCTIVITY OF [IRON + ALUMINUM +  $\Sigma X_i$ ] ALLOYS GROUP II

(At least one  $X_i > 0.20\%$  or if any of Mn, P, S, Si  $> 0.60\%$ )

[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup> K<sup>-1</sup>]

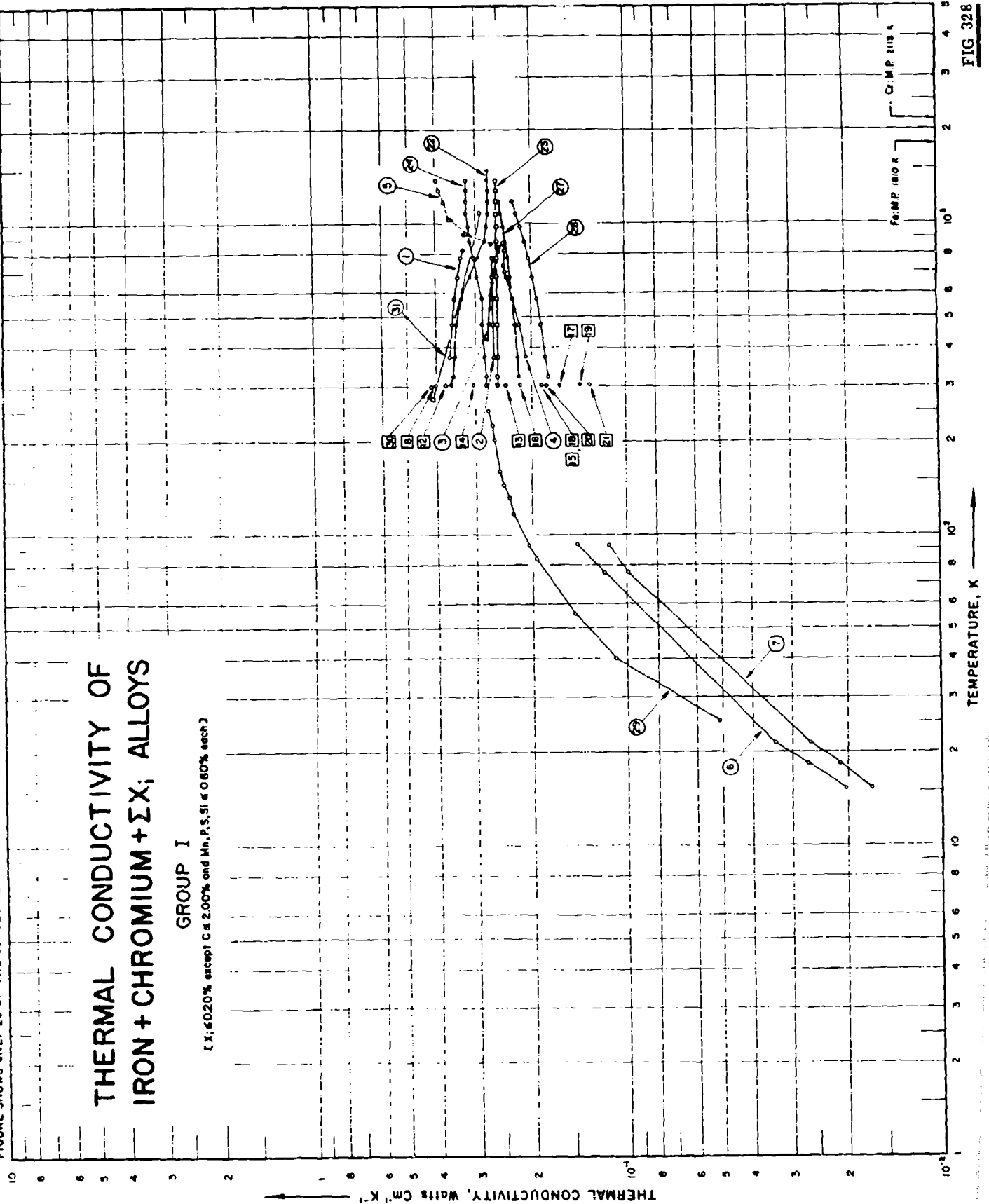
T	k
356.70	0.335
409.70	0.326
432.20	0.326
447.20	0.326
485.70	0.322
511.70	0.314
523.20	0.322
585.20	0.314
659.20	0.310
703.20	0.301

FIGURE SHOWS ONLY 26 OF THE CURVES REPORTED IN TABLE

# THERMAL CONDUCTIVITY OF IRON + CHROMIUM + X; ALLOYS

## GROUP I

[X: 0.20% except C is 2.00% and Mn, P, S, Si is 0.60% each]



TEMPERATURE, K

FIG 328

SPECIFICATION TABLE NO. 328 THERMAL CONDUCTIVITY OF [IRON + CHROMIUM +  $\Sigma X_i$ ] ALLOYS GROUP I(X<sub>i</sub> = 0.20% except C = 2.00% and Mn, P, S, Si  $\leq$  0.60% each)

[ For Data Reported in Figure and Table No. 328 ]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Fe	Cr	C	Mn	Ni	P	S	Si	Composition (continued), Specifications and Remarks
1	129	C	1933	373-821	3.0-5.0	S5	Bal	5.15	0.10	0.45	0.013	0.017	0.18	Specimen 2 cm in dia and 15 cm long; hot rolled; annealed; lead used as comparative material; (thermal conductivity value of 0.352 Watt cm <sup>-1</sup> deg <sup>-1</sup> at 0 C assumed).	
2	129	C	1933	373-773	3.0-5.0	A <sub>1</sub>	Bal	15.19	0.08	0.35	0.05	0.02	0.17	0.20	Specimen 2 cm in dia and 15 cm long; cast at 1490 C, cogged, and rolled; annealed at 845 C; lead used as comparative material (k = 0.352 Watt cm <sup>-1</sup> deg <sup>-1</sup> at 0 C).
3	129	C	1933	373-865	3.0-5.0	A <sub>4</sub>	Bal	26.0	0.10	0.40	0.18	0.013	0.008	0.45	Similar to the above specimen except cast at 1500 C.
4	37	C	1951	426-843	4.0	AISI 403 stainless	Bal	12.0	0.15						Lead on a sample calibrated against lead used as comparative material.
5	43	L	1958	852-1380	5.0	AISI 446 stainless	70.55	27.61	0.086						0.01 Mo; specimen 6.75 in. in dia and 1.5 in. thick.
6	104	L	1951	15-93		3632 A	Bal	13.57	0.36	0.13				0.22	Specimen heated to 800 C and furnace cooled; measured in a vacuo of 5 x 10 <sup>-4</sup> mm Hg.
7	104	L	1951	15-92		3632 B	Bal	13.57	0.36	0.13				0.22	Specimen heated to 950 C and oil quenched; measured in a vacuo of 5 x 10 <sup>-4</sup> mm Hg.
8	176	E	1920	303		3a	Bal	1.0	0.6						Annealed at 900 C and cooled slowly.
9	176	E	1920	303		3b	Bal	1.0	0.6						Annealed at 1100 C and cooled quickly.
10	176	E	1920	303		4a	Bal	2.0	0.6						Annealed at 900 C and cooled slowly.
11	176	E	1920	303		4b	Bal	2.0	0.6						Annealed at 1100 C and cooled quickly.
12	176	E	1920	303		5a	Bal	3.0	0.6						Annealed at 900 C and cooled slowly.
13	176	E	1920	303		5b	Bal	3.0	0.6						Annealed at 1100 C and cooled quickly.
14	176	E	1920	303		6a	Bal	5.0	0.6						Annealed at 900 C and cooled slowly.
15	176	F	1920	303		6b	Bal	5.0	0.6						Annealed at 1100 C and cooled quickly.
16	176	E	1920	303		7a	Bal	10.0	0.6						Annealed at 900 C and cooled slowly.

SPECIFICATION TABLE NO. 328 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Fe	Cr	C	Mn	Ni	P	S	Si	Composition (continued), Specifications and Remarks
17	176	E	1920	303-2		7b	Bal	8.5	0.6						Annealed at 1100 C and cooled quickly.
18	176	F	1920	303-2		8a	Bal	15.0	0.6						Annealed at 900 C and cooled slowly.
19	176	E	1920	303-2		8b	Bal	13.0	0.6						Annealed at 1100 C and cooled quickly.
20	176	E	1920	303-2		9a	Bal	20.0	0.6						Annealed at 900 C and cooled slowly.
21	176	E	1920	303-2		9b	Bal	17.0	0.6						Annealed at 1100 C and cooled quickly.
22	163	L	1936	303-1473		1	Bal	4.98	0.01	0.04				0.02	Forged.
23	163	L	1936	303-1373		6	Bal	13.08	0.07	0.04				0.02	Forged.
24	163	L	1936	303-1373		7	Bal	13.10	1.52	0.38				0.38	Forged.
25	163	L	1936	303-1273		9	Bal	20.63	0.07	0.06				0.03	Usual heat treatment.
26	166	C	1939	273-623		16	Bal	12.95	0.13	0.25	0.14	0.018	0.024	0.17	0.034 Al, 0.060 Cu, 0.012 V, 0.015 As; specimen 8 in. long; heated at 960 C in air, tempered at 750 C for 2 hrs and air cooled; iron used as comparative material; measured in a vacuum of ~0.2 mm Hg.
27	131	C	1953	323-1173	2	AISI 430 stainless	82.4	17.2	0.102	0.254			0.035		Annealed at 1050 C; lead used as comparative material.
28	131	C	1953	323-1173	2	AISI 446 stainless	76.44	23.53	0.152	0.043			0.021		Annealed at 900 C; lead used as comparative material.
29	115	L	1951	25-250	~2	AISI 410 stainless	Bal	12.60	9.99	0.32	0.12	0.012	0.011	0.36	0.06 Cu, 0.03 N.
30	177	C	1936	298	10		Bal	3.53	0.14	0.23				0.26	Normalized.
31	563	L	1935	273, 1073		.. Russian steel	Bal	1.15	0.32	0.63	0.10	0.028	0.023	0.31	0.17 Mo.
32	490	F	1950	273-1173			Bal	26.00	0.13	0.56	0.10	0.012	0.007	0.50	0.14 N.
33	564	C	1958	463-953		AISI 430 stainless	Bal	14.00/18.00	0.12/Max						Nominal composition from Metal's Handbook; alumina (Body Al-300) used as comparative material.
34	564	C	1958	463-953		AISI 430 stainless	Bal	14.00/18.00	0.12/Max						Nominal composition from Metal's Handbook; the above specimen measured using different alumina (Body Al-300) as comparative material.

DATA TABLE NO. 328 THERMAL CONDUCTIVITY OF [IRON + CHROMIUM +  $\Sigma X_i$ ] ALLOYS GROUP I

( $X_i \leq 0.20\%$  except C  $\leq 2.00\%$  and Mn, P, S, Si  $\leq 0.60\%$  each)

[Temperature, T, K. Thermal Conductivity, k, Watts  $\text{cm}^{-1}\text{K}^{-1}$ ]

T	k	T	k	T	k	T	k	T	k	T	k	T	k
<u>CURVE 1</u>													
373.20	0.366	1030.10	0.360	303.20	0.305	303.20	0.255	273.20	0.270	202.21	0.261		
473.20	0.358	1170.10	0.377	323.20	0.255	323.20	0.272	323.20	0.272	225.28	0.265		
573.20	0.351	1171.70	0.377	<u>CURVE 15</u>				373.20	0.276	250.12	0.272		
673.20	0.343	1270.80	0.391	303.20	0.186	473.20	0.255	423.20	0.276	<u>CURVE 30</u>			
773.20	0.336	1270.80	0.389	303.20	0.186	573.20	0.255	473.20	0.278				
821.20	0.330	1380.40	0.395	<u>CURVE 16</u>				673.20	0.278	298.20	0.414		
<u>CURVE 2</u>													
373.20	0.361	<u>CURVE 6</u>				773.20	0.255	573.20	0.280	<u>CURVE 31</u>			
473.20	0.262	15.26	0.0204	303.20	0.218	873.20	0.255	623.20	0.280				
573.20	0.262	16.37	0.0268	<u>CURVE 17</u>				1073.20	0.255	273.2	0.406		
673.20	0.262	21.53	0.0339	303.20	0.162	1073.20	0.255	1273.20	0.255	1073.2	0.290		
773.20	0.263	75.40	0.118	<u>CURVE 18</u>				1373.20	0.255	<u>CURVE 32*</u>			
<u>CURVE 3</u>													
373.20	0.209	<u>CURVE 7</u>				303.20	0.186	303.20	0.276	273.2	0.226		
473.20	0.219	15.30	0.0168	303.20	0.186	323.20	0.276	323.20	0.233	373.2	0.238		
573.20	0.229	18.29	0.0212	<u>CURVE 19</u>				373.20	0.244	473.2	0.255		
673.20	0.238	21.38	0.0264	303.20	0.140	473.20	0.285	473.20	0.248	573.2	0.272		
700.10	0.240	75.86	0.099	303.20	0.140	573.20	0.297	1073.20	0.248	673.2	0.285		
735.40	0.242	92.40	0.114	<u>CURVE 20</u>				1173.20	0.252	773.2	0.301		
818.10	0.241	<u>CURVE 8</u>				773.20	0.305	773.20	0.305	873.2	0.318		
864.90	0.244	303.20	0.402	303.20	0.179	873.20	0.314	873.20	0.314	973.2	0.335		
<u>CURVE 4</u>													
426.00	0.273	<u>CURVE 9*</u>				1073.20	0.322	323.20	0.176	1073.2	0.351		
477.60	0.270	303.20	0.369	303.20	0.179	1073.20	0.322	373.20	0.180	1173.2	0.364		
533.20	0.268	<u>CURVE 21</u>				1173.20	0.322	473.20	0.186	<u>CURVE 33*</u>			
588.70	0.267	303.20	0.130	303.20	0.130	1273.20	0.322	573.20	0.192	463.2	0.197		
644.30	0.262	<u>CURVE 22</u>				1373.20	0.322	673.20	0.198	524.2	0.199		
699.80	0.259	303.20	0.360	<u>CURVE 23*</u>				773.20	0.204	633.2	0.208		
755.40	0.256	323.20	0.356	303.20	0.360	303.20	0.234	873.20	0.210	658.2	0.220		
810.90	0.252	373.20	0.352	323.20	0.356	323.20	0.234	973.20	0.216	759.2	0.224		
842.60	0.249	473.20	0.347	473.20	0.347	473.20	0.234	1073.20	0.222	850.2	0.236		
<u>CURVE 5</u>													
303.20	0.364	573.20	0.355	573.20	0.355	573.20	0.234	1173.20	0.228	953.2	0.277		
303.20	0.364	673.20	0.314	673.20	0.314	673.20	0.239	<u>CURVE 29</u>					
303.20	0.364	773.20	0.297	773.20	0.297	773.20	0.243	25.14	0.0510	463.2	0.201		
303.20	0.364	873.20	0.280	873.20	0.280	873.20	0.247	40.26	0.109	524.2	0.203		
303.20	0.364	973.20	0.272	973.20	0.272	973.20	0.251	55.64	0.147	633.2	0.209		
303.20	0.364	1073.20	0.272	1073.20	0.272	1073.20	0.264	83.62	0.193	658.2	0.214		
303.20	0.364	1173.20	0.272	1173.20	0.272	1173.20	0.276	92.60	0.205	759.2	0.234		
303.20	0.364	1273.20	0.272	1273.20	0.272	1273.20	0.293	117.48	0.230	850.2	0.247		
303.20	0.364	1373.20	0.272	1373.20	0.272	1373.20	0.272	132.17	0.238	953.2	0.261		
303.20	0.364	1473.20	0.272	1473.20	0.272	1473.20	0.272	160.09	0.245				

\* Not shown on plot

FIGURE SHOWS ONLY 25 OF THE CURVES REPORTED IN TABLE

# THERMAL CONDUCTIVITY OF IRON+CHROMIUM + ΣX<sub>i</sub> ALLOYS

## GROUP II

[At least one X<sub>i</sub> > 0.20%, or if any of Mn, P, S, Si > 0.60%]

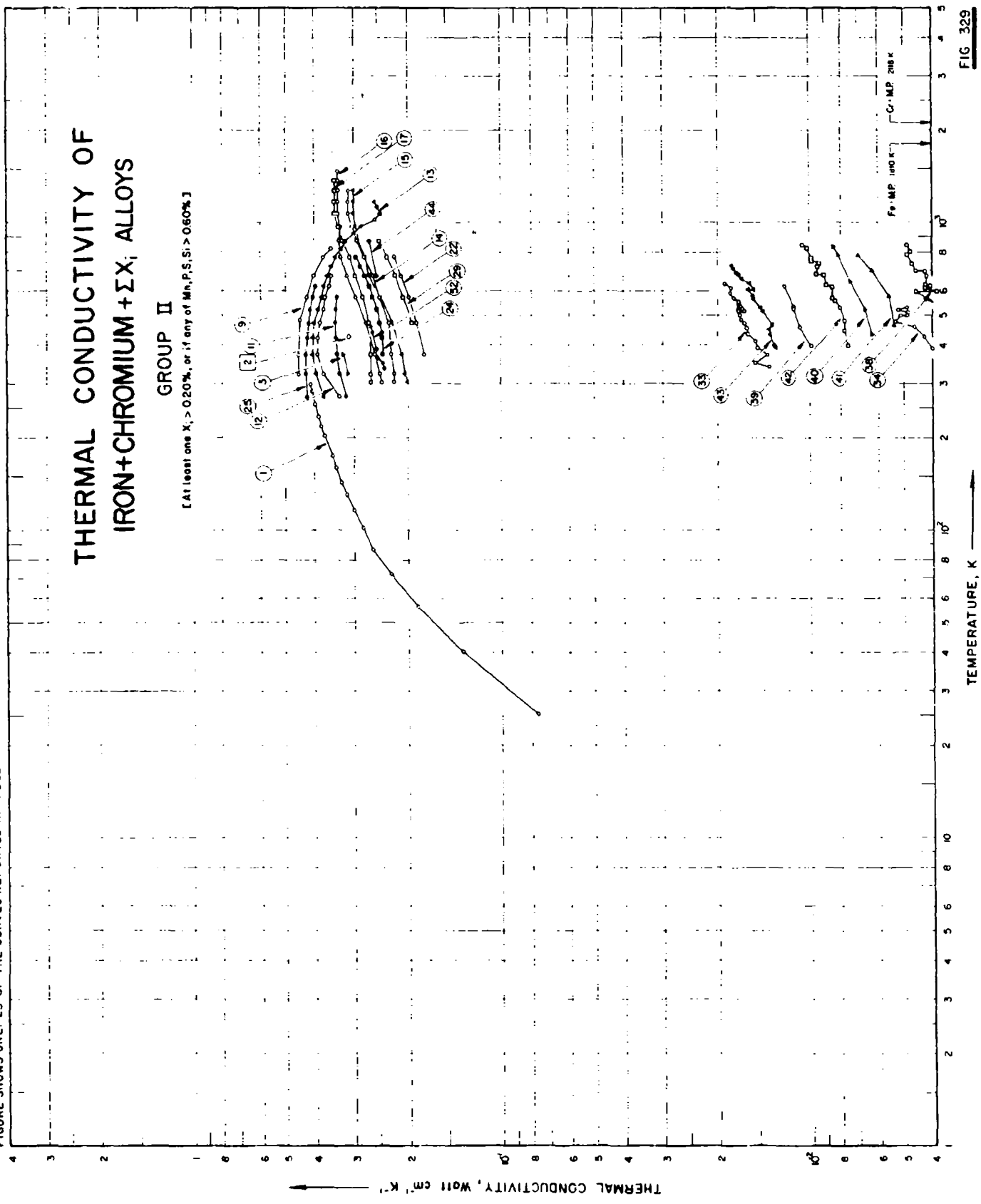


FIG. 329

SPECIFICATION TABLE NO. 329 THERMAL CONDUCTIVITY OF IRON + CHROMIUM +  $\Sigma X_i$  ALLOYS (C < 2.00%) GROUP II(At least one  $X_i > 0.20\%$  or if any of Mn, P, S, Si  $> 0.60\%$ )

[For Data Reported in Figure and Table No. 329]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)							Composition (continued), Specifications and Remarks	
							Fe	Cr	C	Mn	Mo	Ni	Si		S
1	115	L	1951	25-298	-2	SAE4130	bal.	0.99	0.33	0.52	0.22	0.17	0.20	0.019	Specimen supplied by Carnegie Illinois Steel Corp.
2	180	C	1950	428	2		bal.	6.65	0.16	0.69	0.52	0.44	0.020	0.20 Cu, 0.014 P; annealed at 900 C for 1 hr, cooled in still air, annealed at 750 C for 1 hr, and cooled in still air. Ni used as comparative material.	
3	173	C	1956	323-373	-	En19a	bal.	1.15	0.42	0.59	0.22	0.33	0.23	0.019	0.046 P; oil-quenched from 850 C, supplied by Meassta. Brown Bayley Steels Ltd.; specimen (tube) 1 in. O. D., 0.75 in. I. D.
4	173	C	1956	323-373		En19b	bal.	1.15	0.42	0.59	0.22	0.33	0.23	0.019	0.046 P; the above specimen tempered for 3 hr at 150 C.
5	173	C	1956	323-473		En19c	bal.	1.15	0.42	0.59	0.22	0.33	0.23	0.019	0.046 P; the above specimen tempered for 3 hr at 350 C.
6	173	C	1956	323-473		En19d	bal.	1.15	0.42	0.59	0.22	0.33	0.23	0.019	0.046 P; the above specimen tempered for 3 hr at 550 C.
7	173	C	1956	323-473		En19e	bal.	1.15	0.42	0.59	0.22	0.33	0.23	0.019	0.046 P; the above specimen tempered for 3 hr at 650 C.
8	173	C	1956	323-473		En19f	bal.	1.15	0.42	0.59	0.22	0.33	0.23	0.019	0.046 P; the above specimen annealed for 1 hr at 850 C.
9	173	C	1956	323-823		En19g	bal.	1.15	0.42	0.59	0.22	0.33	0.23	0.019	0.046 P; the above specimen reheated to 650 C for 120 hr.
10	173	C	1956	273-373		En31a	bal.	1.5	1.05	0.73	0.21	0.23	0.23	0.028	0.030 P; specimen 1 in. dia; oil-quenched from 830 C; tempered at 140 C for 6 hr.
11	173	C	1956	273-573		En31b	bal.	1.5	1.05	0.73	0.21	0.23	0.23	0.028	0.030 P; the above specimen tempered at 150 C for 2 hr.
12	173	C	1956	273-673		En31c	bal.	1.5	1.05	0.73	0.21	0.23	0.23	0.028	0.030 P; the above specimen tempered at 350 C for 3 hr.
13	173	C	1956	273-1173		En31d	bal.	1.5	1.05	0.73	0.21	0.23	0.23	0.028	0.030 P; the above specimen tempered at 550 C for 4 hr.
14	163	L	1936	303-1273		2	bal.	8.52	0.41	0.85			3.36	0.27 V; usual heat treatment.	
15	163	L	1936	303-1273		3	bal.	2.04	0.41	0.33			4.33	Usual heat treatment.	
16	163	L	1936	303-1473		4	bal.	2.05	8.39	0.48			3.36	Usual heat treatment.	



SPECIFICATION TABLE NO. 329 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Fe	Cu	C	Al	Mo	Ni	Si	S	Composition (continued), Specifications, and Remarks
17	163	L	1936	303-1373		5	bal.	2.91	0.45	0.36			3.62		1.43 Co; usual heat treatment.
18	163	L	1936	303-1373		S	bal.	15.35	1.36	0.47	1.13		0.46		1.50 Co; usual heat treatment.
19	181	+C	1953	373-973		H-20	bal.	2.7	0.23		0.5	0.5			0.75 V; 0.5 W; measured in vacuo.
20	181	+C	1953	373-973		H-27	bal.	3.0	0.4	0.6	0.8		0.3		0.2 V; measured in vacuo.
21	181	+C	1953	373-1073		H-46	bal.	11.0	0.2	0.4	0.5		0.3		0.7 V; 0.15 Ni; measured in vacuo.
22	129	C	1933	373-773	+4	A <sub>5</sub>	bal.	17.12	1.10	0.30		0.55	0.45		1.55 Al; annealed at 900 C; specimen 2 cm in. dia and 15 cm long; lead used as comparative material.
23	169	R	1936	273-773		4		0.86	0.60	0.54			0.29		0.21 V.
24	182	L	1939	373-773	5	AISI410-C	bal.	16.15	0.95		0.75	max			Nominal composition: specimen 7.5 m. dia; 7.5 in. long; measured in a vacuo of $10^{-5}$ mm of Hg.
25	166	C	1939	273-623		20	bal.	9.88	0.35	0.59	0.20	0.26	0.21	0.041	0.12 Cu, 0.079 As, 0.028 P, 0.001 Al; annealed at 500 C; reheated to 610 C and furnace cooled; density 7.845 g cm <sup>-3</sup> ; iron used as comparative material.
26	129, 151	C	1953	373-573	+4	A <sub>2</sub>	bal.	12.0	0.67	0.09		0.23	0.99	0.01	0.015 P; specimen 2 cm dia; 15 cm long; annealed at 845 C; lead used as comparative material.
27	166	C	1939	273-573		19	bal.	1.09	0.315	0.69	0.012	0.673	0.20	0.036	0.065 Cu, 0.039 P, 0.628 As, 0.005 Al; annealed at 500 C; density 7.812 g cm <sup>-3</sup> ; iron used as comparative material.
28	166	C	1939	273-573		17	bal.	13.69	0.27	0.28	0.01	0.29	0.18	0.022	0.25 W, 0.074 Cu, 0.03 Al, 0.022 P, 0.922 V, 0.003 As; heated at 960 C in air; tempered 2 hr at 750 C and air cooled; density 7.741 g cm <sup>-3</sup> ; iron used as comparative material.
29	613	C	1958	473-573		AISI309 stainless	bal.	14.18	0.12	1			1		Nominal composition: alumina used as comparative material (Wesgo Al-30).
30	183	C	1956	303-353		20Q	bal.	0.48	0.35	0.59	0.29	0.26	0.21	0.031	0.12 Cu, 0.039 As, 0.028 P, 0.004 Al; trace V; quenched from 1000 C; annealed at 100 C for several hrs; iron used as comparative material.
31	183	C	1956	323-1073		E-61	bal.	1.46	1.06	0.045		0.31	0.24	0.013	0.017 P; annealed; iron used as comparative material.

SPECIFICATION TABLE NO. 329 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range K	Reported Error %	Name and Specimen Designation	Fe	Cf	C	Alr	Mo	Ni	Si	S	Composition (continued), Specifications and Remarks
32	616		1941	473-773		SAE 4140	bal.	0.50 1.10	0.35 0.45	1.50 0.25			0.20/ 0.35	0.01 max	0.040 (max) P; nominal composition from Metals Handbook.
33	995	R	1951	340-631		AISI446-stainless bal.		21/ 27	0.35 max	1.50 max			1.00 max	0.03 max	0.25 (max) N, 0.040 (max) P; powder (0.000583 ft dia); measured with 0.53 volume fraction He; nominal composition from Metals Handbook.
34	995	P	1951	390-532		AISI446-stainless bal.		21/ 27	0.35 max	1.50 max			1.00 max	0.03 max	0.25 (max) N, 0.040 (max) P, (0.000583 ft dia); measured with 0.53 volume fraction air; nominal composition from Metals Handbook.
35	995	R	1951	366-579		AISI446-stainless bal.		21/ 27	0.35 max	1.50 max			1.00 max	0.03 max	0.25 (max) N, 0.040 (max) P; powder (0.000583 ft dia); measured with 0.53 volume fraction argon; nominal composition from Metal Handbook.
36	327	C	1941	673.2		bal.		1.35	0.31						0.20 V; annealed 2 hrs at 700 C, then oil-quenched.
37	327	C	1941	673.2		bal.		1.59	0.32						0.26 V; annealed 2 hrs at 700 C, then oil-quenched.
38	5-1	R	1958	377-845		AISI446-stainless bal.		21/ 27	0.35 max	1.50 max			1.00 max	0.03 max	0.25 (max) N, 0.040 (max) P; nominal composition; powder (mean dia 0.000583 ft) with 0.5 volume fraction argon.
39	5-1	R	1958	397-622		AISI446-stainless bal.		21/ 27	0.35 max	1.50 max			1.00 max	0.03 max	0.25 (max) N, 0.040 (max) P; nominal composition; powder (mean dia 0.000583 ft) with 0.5 volume fraction containing mixture of helium and argon (He: A = 2:333).
40	5-1	R	1958	432-838		AISI446-stainless bal.		21/ 27	0.35 max	1.50 max			1.00 max	0.03 max	0.25 (max) N, 0.040 (max) P; nominal composition; powder (mean dia 0.000583 ft) with 0.5 volume fraction containing a mixture of neon and argon (Ne: A = 1:941).

SPECIFICATION TABLE NO. 329 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Fe	Cr	C	Mn	Mo	Ni	Si	S	Composition (cont'd), Specifications and Remarks
41	581	R	1958	464-786		AISI446 stainless	bal.	23/ 27	0.35 max	1.50 max			1.00 max	0.03 max	0.25 (max) Ni; 0.040 (max) P; nominal composition; powder (mean dia 0.000583 ft) with 0.5 volume fraction containing a mixture of neon and argon (Ne:Ar = 1.000).
42	581	R	1958	39-549		AISI446 stainless	bal.	23/ 27	0.35 max	1.50 max			1.00 max	0.03 max	0.25 (max) Ni, 0.040 (max) P; nominal composition; powder (mean dia 0.000583 ft) with 0.5 volume fraction neon.
43	581	R	1958	382-723		AISI446 stainless	bal.	23/ 27	0.35 max	1.50 max			1.00 max	0.03 max	0.25 (max) Ni; 0.040 (max) P; nominal composition; powder (mean dia 0.000583 ft) with 0.5 volume fraction helium.
44	977, 978	E	1962	473-873		EI 802 steel		14.81	0.16	0.63	0.59	0.42	0.21	0.014	0.01 P, 1.03 W, and 0.24 V; heated to 1000 C, oil quenched; electrical resistivity 0.127, 0.105, and 0.0976 milliohm cm at 200, 400, and 600 C, respectively.
45	977, 978	E	1962	473-873		EI 802 steel									Same composition as the above specimen; heated to 1000 C, oil quenched, tempered at 680 C for 10 hrs; electrical resistivity 0.139, 0.113, and 0.0956 milliohm cm at 200, 400, and 600 C, respectively.
46	977, 978	E	1962	473-873		EI 802 steel									Same composition as the above specimen; heated to 1000 C, oil quenched, tempered at 680 C for 10 hrs, and at 650 C, for 1000 hrs; electrical resistivity 0.141, 0.111, and 0.0867 milliohm cm at 200, 400, and 600 C, respectively.
47	977, 978	E	1962	473-873		EI 802 steel									Same composition as the above specimen; heated to 1000 C, oil quenched, tempered at 680 C for 10 hrs, and at 600 C, for 1000 hrs; electrical resistivity 0.148, 0.115, and 0.101 at 200, 400, and 600 C, respectively.

SPECIFICATION TABLE NO. 329 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Fe	Cr	C	Mn	Mo	Ni	Si	S	Composition (continued), Specifications and Remarks
48	977, 979	E	1962	473-873		EI 802 steel									Same composition as the above specimen; heated to 1000 C, oil quenched, tempered at 700 C for 19 hrs; electrical resistivity 0.140, 0.114, and 0.0951 milliohm cm at 200, 400, and 600 C, respectively.
49	977, 979	E	1962	473-873		EI 802 steel	11.81	0.6	0.63	0.59	0.42	0.21	0.014		0.01 P, 1.03 W, and 0.24 V; heated to 1100 C, oil quenched; electrical resistivity 0.137, 0.114, and 0.0981 milliohm cm at 200, 400, and 600 C, respectively.
50	977, 978	E	1962	473-873		EI 802 steel									Same composition as the above specimen; heated to 1150 C, oil quenched, tempered at 700 C for 19 hrs; electrical resistivity 0.140, 0.116, and 0.0965 milliohm cm at 200, 400, and 600 C, respectively.
51	973	L	1966	312-436	<6	T 12	12.4	6.13	1.04	0.10	0.25	0.56	0.036		0.031 P; specimen 1.27 cm in dia and 15 cm long; annealed at 950 C, oil quenched from 950 C and tempered at 760 C; cast condition; electrical resistivity 61.2, 63.4, 65.4, 87.3, 69.7, and 72.7 microhm cm at 20, 45, 62, 91, 123, and 159 C respectively.
52	973	L	1966	324-459	<6	T 12									Similar to the above specimen except in wrought condition and electrical resistivity 61.8, 65.6, 67.3, 69.3, 73.6, and 75.1 microhm cm, at 22, 66, 92, 114, 170, and 187 C respectively.

DATA TABLE NO. 329 THERMAL CONDUCTIVITY OF IRON · CHROMIUM ·  $\Sigma X_j$  ALLOYS (C < 2.00%) GROUP II(At least one  $X_j > 0.20\%$  or if any of Mn, P, S, Si > 0.60%)(Temperatures, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup>K<sup>-1</sup>)

T	k	T	k	T	k	T	k	T	k	T	k	T	k	T	k
<u>CURVE 1</u>															
25.34	0.077	323.20	0.440	523.20	0.375	303.20	0.201	1073.20	0.347	373.20	0.272	423.20	0.422	303.20	0.345
40.24	0.104	373.20	0.455	573.20	0.370	323.20	0.205	1173.20	0.347	473.20	0.272	473.20	0.418	353.20	0.350
56.06	0.187	473.20	0.460	623.20	0.360	373.20	0.209	1273.20	0.347	573.20	0.276	523.20	0.414	<u>CURVE 31*</u>	
72.02	0.228	<u>CURVE 8*</u>		673.20	0.355	473.20	0.226	1373.20	0.347	673.20	0.276	573.20	0.405		
86.76	0.259			573.20	0.347	573.20	0.247	<u>CURVE 18*</u>		773.20	0.276	623.20	0.397		
101.74	0.280			673.20	0.248	673.20	0.268	<u>CURVE 26*</u>		873.20	0.276	373.20	0.415		
116.56	0.301			773.20	0.280	773.20	0.280	<u>CURVE 22</u>		973.20	0.276	423.20	0.410		
131.22	0.318			873.20	0.283	873.20	0.283	<u>CURVE 27*</u>		1073.20	0.276	473.20	0.405		
143.92	0.331			973.20	0.297	973.20	0.297	<u>CURVE 23*</u>		1173.20	0.272	523.20	0.375		
150.90	0.344			1073.20	0.301	1073.20	0.301	<u>CURVE 24</u>		1273.20	0.272	573.20	0.355		
158.18	0.353			1173.20	0.301	1173.20	0.301	<u>CURVE 19*</u>		1373.20	0.272	623.20	0.325		
169.22	0.372			1273.20	0.301	1273.20	0.301	<u>CURVE 20</u>		1473.20	0.272	673.20	0.325		
181.40	0.391			1373.20	0.300	1373.20	0.300	<u>CURVE 21*</u>		1573.20	0.272	723.20	0.300		
195.03	0.390			1473.20	0.300	1473.20	0.300	<u>CURVE 25</u>		1673.20	0.272	773.20	0.285		
206.88	0.399			1573.20	0.305	1573.20	0.305	<u>CURVE 28*</u>		1773.20	0.272	823.20	0.268		
224.04	0.408			1673.20	0.305	1673.20	0.305	<u>CURVE 29</u>		1873.20	0.272	873.20	0.268		
239.38	0.412			1773.20	0.305	1773.20	0.305	<u>CURVE 30*</u>		1973.20	0.272	923.20	0.240		
<u>CURVE 2</u>															
427.50	0.302			2073.20	0.300	2073.20	0.300	<u>CURVE 17 (cont.)</u>		2173.20	0.272	973.20	0.240		
<u>CURVE 3</u>															
323.20	0.315			2273.20	0.295	2273.20	0.295	<u>CURVE 18 (cont.)</u>		2373.20	0.272	1073.20	0.240		
373.20	0.225			2373.20	0.285	2373.20	0.285	<u>CURVE 19 (cont.)</u>		2473.20	0.272	1173.20	0.240		
<u>CURVE 4</u>															
323.20	0.330			2473.20	0.285	2473.20	0.285	<u>CURVE 20 (cont.)</u>		2573.20	0.272	1273.20	0.240		
373.20	0.335			2573.20	0.285	2573.20	0.285	<u>CURVE 21 (cont.)</u>		2673.20	0.272	1373.20	0.240		
<u>CURVE 5</u>															
223.20	0.350			2673.20	0.285	2673.20	0.285	<u>CURVE 22 (cont.)</u>		2773.20	0.272	1473.20	0.240		
373.20	0.370			2773.20	0.285	2773.20	0.285	<u>CURVE 23 (cont.)</u>		2873.20	0.272	1573.20	0.240		
473.20	0.357			2873.20	0.285	2873.20	0.285	<u>CURVE 24 (cont.)</u>		2973.20	0.272	1673.20	0.240		
<u>CURVE 6*</u>															
323.20	0.355			2973.20	0.285	2973.20	0.285	<u>CURVE 25 (cont.)</u>		3073.20	0.272	1773.20	0.240		
373.20	0.390			3073.20	0.314	3073.20	0.314	<u>CURVE 26 (cont.)</u>		3173.20	0.272	1873.20	0.240		
423.20	0.390			3173.20	0.314	3173.20	0.314	<u>CURVE 27 (cont.)</u>		3273.20	0.272	1973.20	0.240		
473.20	0.400			3273.20	0.314	3273.20	0.314	<u>CURVE 28 (cont.)</u>		3373.20	0.272	2073.20	0.240		

Not shown on plot



FIGURE SHOWS ONLY 16 OF THE CURVES REPORTED IN TABLE

# THERMAL CONDUCTIVITY OF IRON + CHROMIUM + NICKEL + ΣX; ALLOYS

## GROUP I

(X<sub>1</sub> ≤ 0.20% except C ≤ 2.00% and Mn, P, S, Si ≤ 0.60% each)

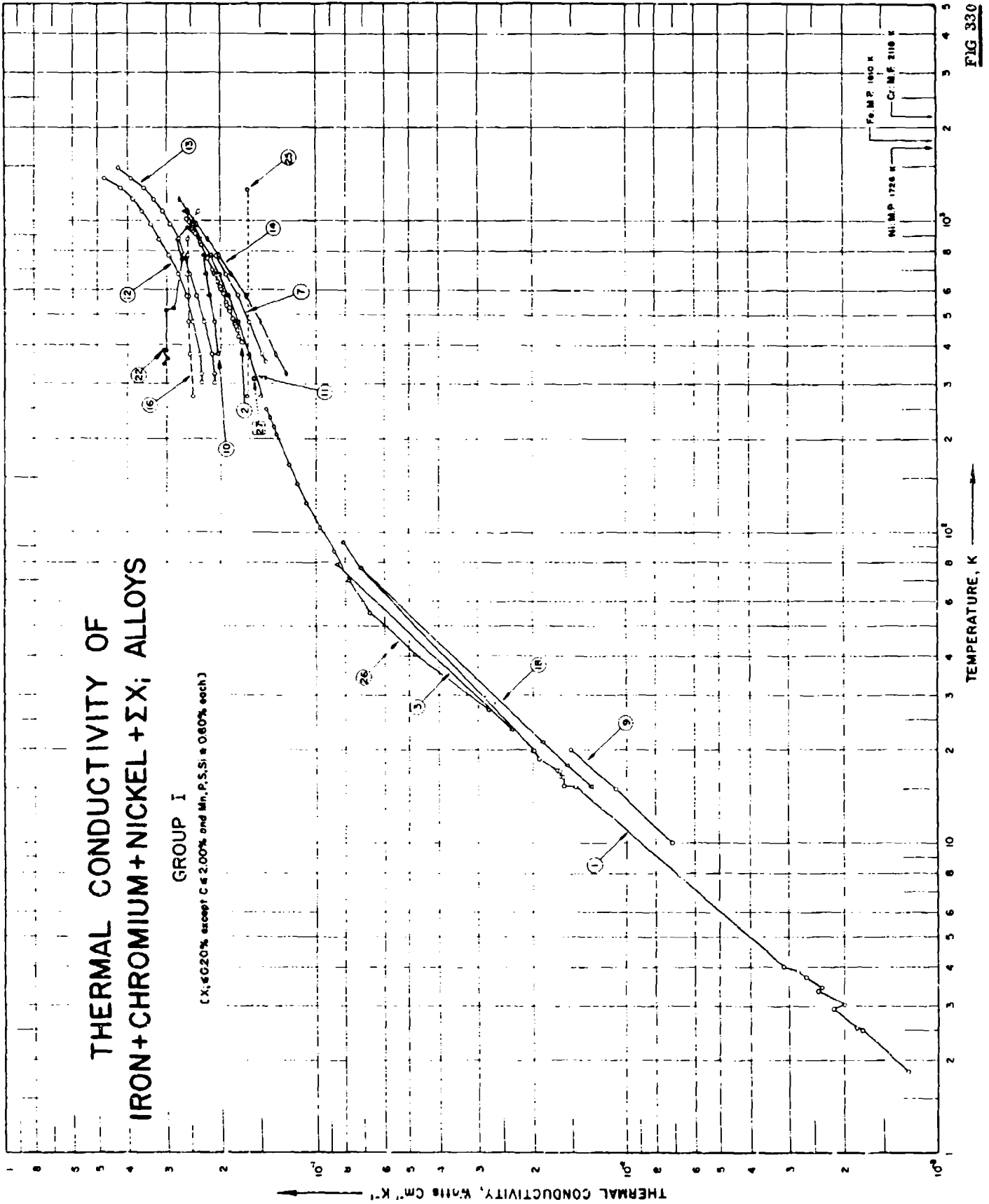


FIG 330

SPECIFICATION TABLE NO. 130 THERMAL CONDUCTIVITY OF IRON + CHROMIUM + NICKEL + Si, Al ALLOYS GROUP I

(X<sub>1</sub> = 0.20% except C = 2.06% and Mn, P, S, Si = 0.60% each)

For Data Reported in Figure and Table No. 130

Curve No.	Ref. Method Used	Year	Temperature Range, K.	Reported Error, %	Specimen Designation	Composition (weight per cent)					Composition (continued), Specifications and Remarks
						Cr	Ni	C	Mn	Si	
1	157	L	18-93		37-4	18.8	8.10	0.12	0.24	0.43	Quenched in water from 1150 C.
2	158	L	406-1008		AISI 304 stainless	17.0/	7.0/	0.11			Nominal composition.
3	159	L	23-79	<10.0	AISI 304 stainless	17.0/	7.0/	0.11			Nominal composition.
4	129	C	373-782	3.0-5.0	A7	18.6	9.10	0.07	0.27		Heated to 735 C for 8 hrs and cooled in diatomaceous earth.
5	129	C	373-773	3.0-5.0	A6	18.3	9.21	0.11	0.19		Heated to 1120 C and quenched in cold water.
6	129	C	373-773	3.0-5.0	A9	19.6	8.96	0.21	0.37		Heated to 735 C for 8 hrs and cooled in diatomaceous earth.
7	129	C	359-773	3.0-5.0	A10	19.6	7.99	0.21	0.28		Heated to 1120 C and quenched in water.
8	161	C	373-773	2.0	A10A	19.6	7.99	0.24	0.28		The above specimen A10 heated again for 8 hrs at 735 C and cooled in furnace.
9	152	L	10-20		Stainless						Composition unknown.
10	160	F	373-773		AS21	27.0	18.9	0.12			Quenched
11	162	C	273-1073		Stavbrite	17.87	8.04	0.15	0.26	0.19	Softened at 1150 to 1200 C.
12	163	L	303-1173		12	17.50	8.85	0.11	0.33	0.48	Rolled.
13	163	L	303-1173		11	24.30	4.22	0.19	0.51	0.50	Rolled.
14	131	C	323-1173	2.0	AISI 302 stainless	18.40	9.66	0.116		0.130	71.60 Fe, 0.021 P, 0.013 S, annealed at 1100 C.
15	129	C	373-773	3.0-5.0	A3	14.60	0.70	0.14	0.19	0.12	0.020 P, 0.015 S, annealed at 845 C.
16	162	C	273-1073		F. H. stainless	13.65	0.37	0.27	0.29	0.27	Air hardened from 940 C and tempered at 725 C.
17	163	L	303-1373		11	13.30	0.50	0.16	0.46	0.26	Rolled.
18	104	L	19-93		37-4	18.80	8.10	0.12	0.21	0.43	Heated to 1150 C and quenched in oil.
19	536	L	336-1050			17.30	8.61	0.26	0.30	0.20	0.017 P, 0.002 S, heated at 1100 C and quenched in water.
20	539	L	326-579			17.30	8.61	0.26	0.30	0.20	0.017 P, 0.002 S, heated at 1180 C and quenched in water.
21	539	L	344-1052			17.30	8.61	0.26	0.30	0.20	0.017 P, 0.002 S, heated at 1180 C and quenched in water, then reheated to 306 C.
22	539	L	348-1069			13.29	0.51	0.12	0.33	0.19	0.017 P, 0.016 S, annealed at 900 C and furnace cooled.
23	539	L	331-584			13.29	0.51	0.12	0.33	0.19	0.017 P, 0.016 S, water-quenched from 1180 C.
24	539	L	340-1074			13.29	0.51	0.12	0.33	0.19	0.017 P, 0.016 S, water-quenched from 1180 C, and reheated to 310 C.
25	563	L	273-1273		Russian stainless	18.9	7.5	0.20	0.47	0.11	0.022 P.
26	565	L	27-250		18-8 type 304 stainless						Nominal composition.



SPECIFICATION TABLE NO. 330 (continued)

Curve No.	Ref. Method Used	Year	Temperature Range, K.	Reported Error, %	Name and Specimen Designation	Cr	Ni	C	Mn	Si	Composition (continued), Specifications and Remarks
27	566	C	1956	313.2	18-8 stainless	19.32	9.55	0.08	0.26	0.6	Nominal composition.
28	439	L	1935	373-373	18-8 stainless	13.10	0.50	0.39	0.18	0.11	Heat-treated at 1050 C and water-quenched.
29	614	R	1961	485-1632	420 stainless						0.12 Cu, 0.06 Mo, 0.02 P and 0.011 S; specimen made up of 1 in. dia disks.



FIGURE SHOWS ONLY 33 OF THE CURVES REPORTED IN TABLE

# THERMAL CONDUCTIVITY OF IRON + CHROMIUM + NICKEL + ΣX; ALLOYS

## GROUP II

[At least one X, > 0.20%, or if any of Mn, P, S, Si > 0.60%]

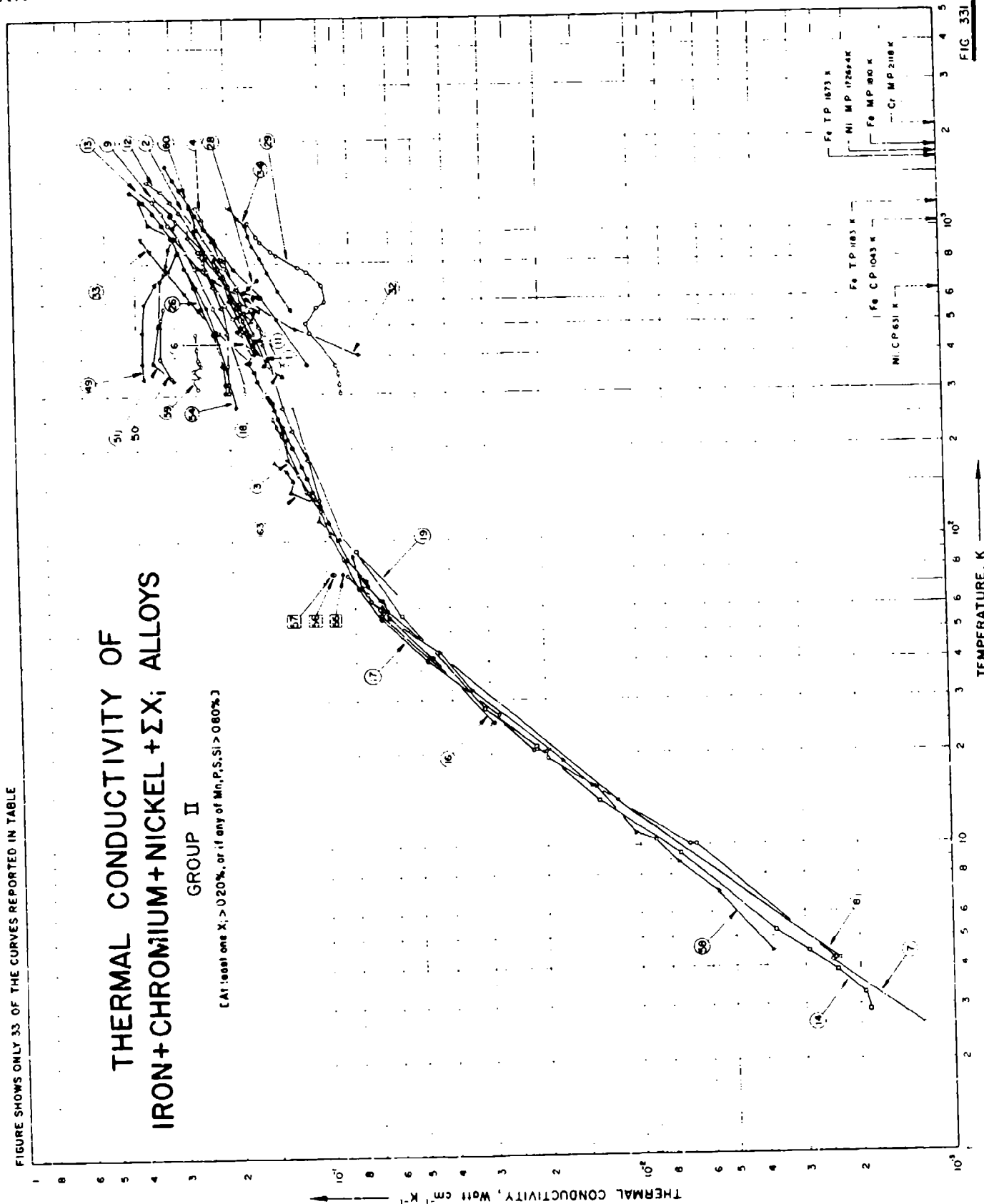


FIG. 33I

SPECIFICATION TABLE NO. 331 THERMAL CONDUCTIVITY OF IRON + CHROMIUM + NICKEL +  $\Sigma X_i$  ALLOYS GROUP II

(At least one  $X_i > 0.20\%$  or if any of Mn, P, S, Si  $> 0.60\%$ )

For Data Reported in Figure and Table No. 331.

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)				Composition (continued)		Specifications and Remarks
							Cr	Ni	C	Cu	Mn	Mo	
1	181	L	1953	373-1073		R 20	19.0	14.0	0.1		0.8		1.7 Nb, 0.3 Si.
2	181	L	1953	373-1173		G 15 B	13.0	13.0	0.4		0.8	2.0	10.0 Co, 3.0 Nb, 2.5 W, 1.0 Si.
3	91	C	1951	137-1217		AISI 301 stainless	16.0/ 18.0	6.0/ 8.0	0.18/ 0.20		2.0		Hot rolled; annealed at 1093 C for 1 hr.
4	91	C	1951	136-1220		AISI 316 stainless	16.82	11.66	0.108		1.59	2.18	0.26 Si, 0.023 S; hot rolled, annealed, and quenched in water.
5	91	C	1951	138-1174		AISI 347 stainless	17.65	10.94	0.06	0.09	1.64	0.02	0.73 Nb, 0.58 Si, 0.017 S; hot rolled, annealed, and quenched in water.
6	37	C	1951	460-851	4.0	Multimet N-155	20.0	20.0	0.2		3.25		20.0 Co, 2.5 W, 1.1 Nb.
7	155	L	1951	2.6- 78		AISI 303 stainless	17.0/ 19.0	8.0/ 10.0	0.15/ 0.15		2.0	0.60	0.60 (Max) Zr, 0.07 (Min) Se, 0.07 S, Min.
8	155	L	1951	4.3- 76		AISI 347 stainless	17.0/ 19.0	9.0/ 12.0	0.08/ 0.08		0.60		10 x C = Nb.
9	43	L	1958	877-1305	5.0	Stainless 17-7	17.30	7.06	0.074		0.60		72.21 Fe, 1.11 Al, 0.49 Si.
10	182	L	1959	373- 773	5.0	AISI 304 stainless	18.51	9.09	0.053		0.67		0.53 Si, 0.028 S.
11	182	L	1959	423- 823	5.0	AISI 304 stainless	18.51	9.09	0.053		0.67		0.53 Si, 0.028 S; tested in high vacuum.
12	163	L	1936	303-1473		17	12.85	12.23	0.48		1.44		2.88 W, 1.22 Si, usual heat treatment.
13	163	L	1936	303-1373		18	13.34	11.50	0.53		1.17		3.92 W, 1.07 Si, usual heat treatment.
14	9	L	1951	2.9- 92		Austenitic steel, 181S	18.9	7.9	0.1				1.0 Ti, 0.7 Si.
15	38	L	1952	473- 923		AISI 304 stainless	18.0/ 20.0	8.0/ 11.0	0.08/ 0.08		2.0		
16	115	L	1951	25- 295		AISI 347 stainless	17.88	10.28	0.05	0.26	1.24		0.85 Nb, 0.57 Si, 0.027 N, 0.023 total P and S.
17	115	L	1951	27- 250		AISI 304 stainless	18.68	8.84	0.05	0.06	1.12		0.43 Si, 0.031 N, 0.04 total P and S.
18	687	R	1966	344-482	~5.0	AISI 304 stainless	19.0/ 20.0	8.0/ 12.0	0.08/ 0.08		2.00		1.00 (Max) Si, 0.030 (Max) S, nominal composition; disklike specimen of 2.75 in. dia.; accuracy $\pm 0.008$ W cm <sup>-1</sup> C <sup>-1</sup> .

SPECIFICATION TABLE NO. 331 (continued)

Curve No.	Rel. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Cr	Ni	C	Cu	Mn	Mo	P	Composition (continued), Specifications and Remarks
19	1-9	L	1953	600-270	10.0	AISI 316 stainless	16.0-18.0	10.0-14.0	0.1-0.13			2.0-3.0		Annealed.
20	163	L	1936	300-1473		15	20.25	12.66	0.13		0.71			2.20 Si; rolled.
21	163	L	1936	303-1473		16	22.75	20.30	0.12		1.40			2.38 Si; rolled.
22	166	C	1939	273-623		15	19.13	8.14	0.08	0.000	0.37		0.022	0.68 Si, 0.60 W, 0.025 As, 0.011 S, 0.004 Al; heated to 1100 C and cooled in water.
23	39	R	1975	330-1306	2.0	AISI 316 stainless	16.0-18.0	10.0-14.0	0.10-0.15			2.0-3.0		
24	79	R	1958	330-1513	2.0	AISI 347 stainless	17.0-19.0	9.0-12.0	0.08-0.15					10 x C-Nb.
25	161	C	1934	373-773	2.0	A11	18.08	9.12	0.07		0.39		0.013	0.47 Si, 0.34 Ti, 0.003 S; annealed.
26	163	L	1936	303-1473		13	27.53	1.73	0.15		0.80			0.84; rolled.
27	193	L	1939	180-290		EYA-2 steel	16.05	9.89	0.20		0.66		0.024	0.88 Si, 0.002 S; annealed at 1050 C for 2 hrs and quenched in water.
28	194	L	1951	460-710	3.0	AISI 347 stainless	18.00	11.12	0.07		1.77		0.016	0.86 Nb, 0.007 S.
29	175	P	1936	303-1073		Stainless steel	21.04	9.02	0.19	0.108	0.35		0.015	1.018 Si, 0.007 S; annealed at 850 C for 2.5 hrs.
30	177	C	1936	298.2	10.0	WF 109	13.89	13.10	0.34		1.03		0.006	2.89 Si, 1.54 W; forged.
31	195	C	1938	700-1367	20.0	Stainless 17-7 PH	17.08	7.21	0.70		0.71		0.024	1.19 Al, 0.45 Si, 0.017 S; density 7.43 g/cm <sup>3</sup> .
32	664	E	1957	400-1273	1.1	Russian steel E1257 15	15	0.15			0.7	0.4	0.035	2.75 W, 0.8 Si, 0.030 S; specimen 4 mm dia and 120 mm long; austenitic; tempering at 1175 C (cooling medium is water) and aging at 750 C for 10 hrs.
33	664	F	1957	380-963	1.1	Russian steel Cr 19, N49	19	0.9						Other components unknown; specimen 4 mm dia and 120 mm long; austenitic; forging from 1150 C to 950 C, tempering at 1150 C (cooling medium is water) and aging at 700 C for 50 hrs.

SPECIFICATION TABLE NO. 331 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)					P	Composition (continued), Specifications and Remarks
							Cr	Ni	C	Cu	Mn		
34	427	L	1960	561-1203		AISI 316 stainless	24.07 26.0	19.07 22.0	0.25 MIS	0.25	2.0 Max	1.5 Si, Max.	
35	534	E	1958	414-511	3.0	1-S8 stainless; YAIT-1	18	10	0.07		1.4	0.5 Ti, 0.4 Si, in form of rod.	
36	534	E	1958	347-928	3.0	1-S8 stainless; YAIT-2	17.88	9.32	0.11		1.25	0.45 Si, 0.5 Ti, in form of rod.	
37	534	E	1958	973-928	3.0	1-S8 stainless; YAIT-3	18.6	8.0	0.10		1.25	0.5 Ti, 0.4 Si, in form of rod.	
38	534	E	1958	418-607	3.0	1-S8 stainless; YAIT-4	17.2	10.0	0.12		1.21	0.49 Si, 0.5 Ti, in form of tube.	
39	534	E	1958	448-607	3.0	1-S8 stainless; YAIT-5	17.79	9.8	0.1		1.74	0.46 Ti, 0.8 Si, in form of tube.	
40	534	E	1958	442-1225	3.0	1-S8 stainless; YAIT-6	19.0	9.8	0.09		1.25	0.54 Si, 0.46 Ti, from the melt.	
41	534	E	1958	373-973	3.0	1-S8 stainless; EI-572	20.6	11.0	0.36		1.2	1.5 0.70 Si, 0.55 Ti, 0.3 Nb.	
42	534	E	1958	373-973	3.0	1-S8 stainless; EI-572	20.6	11.0	0.36		1.2	1.5 0.70 Si, 0.55 Ti, 0.3 Nb, tempered at 700°C for 900 hrs.	
43	534	E	1958	373-973	3.0	1-S8 stainless; EI-572	20.6	11.0	0.36		1.2	1.5 0.70 Si, 0.55 Ti, 0.3 Nb, tempered at 700°C for 1600 hrs.	
44	534	E	1958	373-973	3.0	1-S8 stainless; EI-572	20.6	11.0	0.36		1.2	1.5 0.70 Si, 0.55 Ti, 0.3 Nb, tempered at 700°C for 2600 hrs.	
45	534	E	1958	373-973	3.0	1-S8 stainless; EI-572	20.6	11.0	0.36		1.2	1.5 0.70 Si, 0.55 Ti, 0.3 Nb, tempered at 700°C for 1000 hrs.	
46	534	F	1968	446-573	3.0	1-S8 stainless; EI 606-1	18.0	8.0	0.09		1.0	2.0 C, 1.5 Si.	
47	534	E	1958	410-43	3.0	1-S8 stainless; EI 606-2	18.0	8.0	0.09		1.0	2.0 C, 1.5 Si.	
48	534	E	1958	437-773	3.0	1-S8 stainless; EI 606-3	18.0	8.0	0.09		1.0	2.0 C, 1.5 Si, aged at 110°C for 1000 hrs.	
49	539	L	1958	340-1003		Ni-Cr steel	2.82	2.72	0.28		0.51	0.31 Si, 0.003 S, cooled in air from 900°C and tempered at 600°C.	
50	539	L	1958	342-571		Ni-Cr steel	2.82	2.72	0.28		0.41	0.31 Si, 0.003 S, quenched in oil from 550°C.	

SPECIFICATION TABLE NO. 331 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)				P	Composition (continued), Specifications and Remarks		
							Cr	Ni	C	Cu		Mn	Mo	Si
51	539	L	1938	325-984		Ni - Cr steel	2.42	2.72	0.28		0.51	0.56	0.013	0.34 Si, 0.003 S; quenched in oil from 850 C and tempered at 300 C.
52	490	F	1950	273-1173		AISI 303 stainless	18.42	8.97	0.17		0.61			0.51 Si.
53	490	F	1950	273-1173		AISI 310 stainless	25.34	20.68	0.10		1.83		0.025	50.98 Fe, 0.84 Si, 0.005 S.
54	490	F	1950	273-1173		AISI 347 stainless	18.10	11.20	0.069		1.80		0.021	0.77 Nb, 0.70 Si, 0.007 S.
55	537		1940	78.2		18/8 stainless; 1	18.45	8.20	0.04					0.20 Ti; forged; commercial heat-resistant alloy; measured in the boiling nitrogen bath.
56	537		1940	78.2		18/8 stainless; 2	17.95	8.30	0.04					Forged; commercial heat-resistant alloy; measured in the boiling nitrogen bath.
57	537		1940	78.2		18/8 stainless; 3	17.80	9.19	0.04	1.50		3.00		Forged; commercial heat-resistant alloy; measured in the boiling nitrogen bath.
58	587, 154	L	1956	4, 5, 88	± 5.0	SS K1118N3T	17/20	8/11	0.14/Max		2.0/Max			Max 0.8 Sn, max 0.8 Ti, max 0.63 Si, max 0.035 S; unannealed.
59	615	C	1962	317-471		416 Stainless	12/14	1.25/2.50	0.15/Max		1.25/Max			1.00 Si max, min 0.15 S.
60	614	R	1961	375-1668	± 5	Ph17-4(H900) stainless	16.4	4.2	0.07	4.1	1.00		0.04	1.00 Si, 0.30 Nb + Ta; specimen composed of 5 one-inch dia disks.
61	614	R	1961	317-1623	± 5	AM353 stainless	15.66	4.27	0.12		0.94	2.82	0.02	0.05 Si, specimen composed of 5 one-inch dia disks.
62	614	R	1961	373-1498	± 5	Crucible HNM	18.5	9.5	0.30		3.50		0.22	0.05 Si, trace Mo, Al, and W; specimen composed of 5 one-inch dia disks.
63	685	L	1963	100-150		AISI 304 Stainless	18/20	8/12	0.98/Max		2.00/Max		0.045 Max	1.00 max Si, 0.030 max S; nominal composition; cross sectional area 0.105 cm <sup>2</sup> and 2.55 cm long.

SPECIFICATION TABLE NO. 331 (continued)

Curve No.	ReL No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)					Composition (continued), Specifications and Remarks		
							Cr	Ni	C	Cu	Mn		Mo	P
64	982	C	1965	274-1185	±5	ALSI 316; 3A	17.45	12.60	0.063	0.09	1.59	2.55	0.023	0.60 Si, 0.19 Co, 0.010S, machined specimen 0.900 ± 0.001 in. in dia, 5.713 ± 0.006 in. long in the measuring section and 1.125 ± 0.003 in. in dia and 3.375 ± 0.006 in. long in the beater section; hot-rolled, annealed and pickled; density 7.95 g cm <sup>-3</sup> at 293.2 K; hardness, Rockwell B 78-79; measurements done in a vacuum of 2 x 10 <sup>-3</sup> Torr, electrical resistivity during thermal conductivity measurements, 75.4, 80.1, 80.6, 83.9, 84.1, 85.5, 86.1, 86.3, 86.9, 90.0, 88.4, 90.8, 90.0, 92.3, 93.4, 95.2, 96.7, 98.2, 99.5, 100, 100, 104, 106, 105, 106, 109, 113, 111, 114, and 120 μhm cm at 273.7, 334.8, 343.7, 386.7, 391.5, 413.7, 417.6, 430.9, 431.5, 457.1, 460.9, 491.5, 500.9, 514.3, 520.4, 589.8, 605.4, 613.7, 659.3, 670.4, 679.8, 712.6, 722.1, 799.8, 800.4, 897.1, 958.2, 982.6, 1113.2, and 1185.9 K, respectively; electrical resistivity before thermal conductivity measurements 75.4 μhm cm at 293.9 K; Armco iron from Battelle Memorial Institute used as comparative material; specimen supplied by NASA-Lewis; results reported to be about 6-11% higher than values previously reported by BMI.



SPECIFICATION TABLE NO. 331 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)					P	Composition (continued), Specifications and Remarks	
							Cr	Ni	C	Cu	Mn			Mo
65	982	C	1965	273-1219	±5	AISI 316; 4A	17.45	12.62	0.063		1.88	2.70	0.021	0.54 Si, 0.014 S; machined specimen 0.900 ± 0.001 in. in dia, 6.715 ± 0.006 in. long in the measuring section and 1.125 ± 0.003 in. in dia, 3.375 ± 0.006 in. long in the heater section; hot-rolled, annealed and pickled; density 7.95 g cm <sup>-3</sup> at 293.2 K; hardness, Rockwell B 77; measurements done in a vacuum of 2 × 10 <sup>-6</sup> Torr; electrical resistivity, during thermal conductivity measure- ments, 75.7, 80.3, 81.6, 82.7, 84.7, 86.0, 87.3, 87.5, 87.8, 89.7, 91.5, 92.0, 93.1, 95.1, 97.2, 99.0, 100, 102, 107, 110, 108, 108, 110, 114, 115, 115, 115, 115, 115, at 272.6, 272.6, 272.6, 272.6, 272.6, 406.7, 433.7, 460.7, 489.5, 459.3, 472.6, 454.5, 507.1, 558.2, 597.6, 600.9, 657.6, 662.6, 667.6, 703.7, 743.7, 792.1, 796.5, 822.1, 892.6, 952.6, 1012.1, 1140.9, and 1218.7 K respectively; electrical resistivity before thermal conductivity measure- ments 77.4 μohm cm at 294.3 K and after thermal conductivity measure- ments reported 78.6 μohm cm at 300.9 K; Armco Iron from Battelle Memorial Institute used as reference material; specimen supplied by NASA- Lewis; results reported to be about 6-11% higher than the values previously reported by BMI.

DATA TABLE NO. 331 THERMAL CONDUCTIVITY OF [IRON + CHROMIUM + NICKEL + EX<sub>1</sub>] ALLOYS GROUP II(At least one X<sub>1</sub> > 0.20% or if any of Mn, P, S, Si > 0.60%)[Temperature, T, K; Thermal Conductivity, k, Watts cm<sup>-1</sup> K<sup>-1</sup>]

T	k	T	k	T	k	T	k	T	k	T	k	T	k	T	k	T	k
<b>CURVE 1</b>																	
373.20	0.151	135.7	0.105	919.1	0.234	4.25	0.00234	303.2	0.201	28.80	0.0312	104.92	0.0970				
473.20	0.167	184.4	0.113	996.9	0.234	14.00	0.0117	323.2	0.201	56.50	0.0575	125.06	0.107				
573.20	0.184	228.9	0.126	1079.9	0.247	58.59	0.0638	373.2	0.201	92.00	0.0795	144.81	0.114				
673.20	0.201	270.0	0.134	1173.6	0.259	63.20	0.0660	473.2	0.201	<b>CURVE 1<sub>1</sub></b>			165.15	0.121			
773.20	0.218	367.5	0.146	63.30	0.0570	63.30	0.0570	573.2	0.209	206.64	0.132	206.64	0.132				
873.20	0.234	433.8	0.155	70.70	0.0732	70.70	0.0732	673.2	0.222	220.02	0.135	220.02	0.135				
973.20	0.251	482.1	0.151	76.20	0.0772	76.20	0.0772	773.2	0.234	235.13	0.140	235.13	0.140				
1073.20	0.268	497.8	0.159	400.4	0.170	<b>CURVE 9</b>			873.2	0.195	250.10	0.144					
<b>CURVE 2</b>																	
373.2	0.134	660.7	0.188	422.0	0.174	477.6	0.184	573.2	0.268	673.2	0.208						
473.2	0.153	759.8	0.201	533.2	0.194	876.8	0.237	1073.2	0.268	773.2	0.223						
573.2	0.174	812.6	0.205	568.7	0.204	880.1	0.243	1173.2	0.285	873.2	0.239						
673.2	0.188	823.6	0.209	644.3	0.214	1046.3	0.292	1273.2	0.305	923.2	0.250						
773.2	0.205	944.7	0.222	695.8	0.222	1156.7	0.305	1373.2	0.326								
873.2	0.222	953.8	0.218	735.4	0.231	1160.1	0.303	1473.2	0.360								
973.2	0.238	1031.5	0.234	810.9	0.238	1309.2	0.331	<b>CURVE 13</b>									
1073.2	0.257	1114.0	0.238	851.0	0.245	1309.2	0.332	303.2	0.190	25.26	0.0293	344.3	0.194				
1173.2	0.276	1220.4	0.251	<b>CURVE 7</b>				323.2	0.180	41.10	0.0460	372.1	0.155				
<b>CURVE 3</b>																	
135.8	0.109	137.7	0.117	2.59	0.00125	373.2	0.165	373.2	0.188	55.88	0.0628	388.7	0.151				
154.0	0.130	185.4	0.130	4.20	0.00247	423.2	0.173	473.2	0.201	70.80	0.0741	410.9	0.163				
226.0	0.134	228.6	0.142	4.25	0.00240	473.2	0.180	573.2	0.222	100.23	0.0912	415.4	0.165				
266.6	0.142	269.2	0.146	10.10	0.00663	523.2	0.188	673.2	0.239	115.00	0.0975	445.4	0.167				
329.8	0.163	361.9	0.151	10.10	0.00693	573.2	0.195	773.2	0.255	129.89	0.110	482.1	0.176				
355.5	0.167	427.9	0.159	19.40	0.0198	623.2	0.202	873.2	0.272	150.16	0.114						
380.3	0.176	429.5	0.167	20.40	0.0199	673.2	0.208	973.2	0.289	160.16	0.118						
405.0	0.167	490.8	0.167	20.60	0.0220	723.2	0.214	1073.2	0.310	175.11	0.118						
536.4	0.188	540.7	0.184	26.60	0.0220	773.2	0.221	1173.2	0.326	201.62	0.126						
602.5	0.197	550.1	0.175	56.90	0.0657	823.2	0.231	1273.2	0.347	214.42	0.130						
722.3	0.209	552.1	0.167	58.30	0.0664	873.2	0.245	1373.2	0.385	234.94	0.134						
831.8	0.226	574.5	0.188	59.50	0.0672	923.2	0.268	<b>CURVE 14</b>									
887.8	0.234	662.3	0.188	63.30	0.0719	973.2	0.285	2.88	0.00185	303.2	0.176	303.2	0.176				
1032.4	0.255	741.5	0.205	66.80	0.0733	1073.2	0.310	473.2	0.175	287.90	0.148	373.2	0.176				
1038.8	0.251	768.9	0.201	66.80	0.0733	1173.2	0.326	523.2	0.180	294.72	0.150	473.2	0.180				
1216.7	0.264	801.0	0.213	77.60	0.0823	1273.2	0.347	573.2	0.188	673.2	0.201	573.2	0.188				
<b>CURVE 5</b>																	
137.7	0.117	137.7	0.117	423.2	0.165	473.2	0.175	2.88	0.00185	773.2	0.222	773.2	0.222				
185.4	0.130	185.4	0.130	473.2	0.173	523.2	0.180	3.27	0.00192	873.2	0.222	873.2	0.222				
228.6	0.142	228.6	0.142	523.2	0.188	573.2	0.188	3.86	0.00237	973.2	0.234	973.2	0.234				
269.2	0.146	269.2	0.146	623.2	0.194	673.2	0.194	4.48	0.00292	1073.2	0.247	1073.2	0.247				
361.9	0.151	361.9	0.151	673.2	0.200	723.2	0.214	5.23	0.00371	1173.2	0.255	1173.2	0.255				
355.5	0.167	427.9	0.159	723.2	0.206	773.2	0.213	9.40	0.00742	1273.2	0.268	1273.2	0.268				
380.3	0.176	429.5	0.167	773.2	0.213	823.2	0.219	14.10	0.0135	1373.2	0.285	1373.2	0.285				
405.0	0.167	490.8	0.167	823.2	0.219	823.2	0.219	21.20	0.0216								

Not shown on plot

DATA TABLE NO. 331 (continued)

T	k	T	k	T	k	T	k	T	k	T	k	T	k
<u>CURVE 21<sup>a</sup></u>		<u>CURVE 24<sup>a</sup> (cont.)</u>		<u>CURVE 28 (cont.)</u>		<u>CURVE 33</u>		<u>CURVE 38</u>		<u>CURVE 45<sup>a</sup></u>		<u>CURVE 51 (cont.)</u>	
303.2	0.176	826.1	0.211	617.1	0.166	380.4	0.172	447.9	0.172	373.2	0.172	706.2	0.334
323.2	0.176	1043.4	0.245	622.1	0.160	475.7	0.214	564.8	0.186	573.2	0.195	793.2	0.318
373.2	0.176	1284.6	0.278	643.2	0.177	567.7	0.248	606.9	0.203	773.2	0.232	895.2	0.312
473.2	0.180	1388.7	0.283	669.3	0.173	639.5	0.277			973.2	0.272	984.2	0.299
573.2	0.188	1542.9	0.300	709.9	0.160	767.5	0.320	<u>CURVE 39<sup>a</sup></u>		<u>CURVE 46<sup>a</sup></u>			
673.2	0.197			884.2	0.158	884.2	0.358	447.9	0.172				
773.2	0.209	<u>CURVE 25<sup>a</sup></u>		963.2	0.382			504.8	0.186	446.6	0.158	273.2	0.142
873.2	0.222	373.2	0.161	303.2	0.0870	<u>CURVE 34</u>		606.9	0.203	585.1	0.182	373.2	0.151
973.2	0.239	473.2	0.176	324.2	0.0874	561.2	0.126	<u>CURVE 40<sup>a</sup></u>		973.2	0.241	473.2	0.159
1073.2	0.255	573.2	0.191	349.2	0.0883	647.2	0.137	441.6	0.181	<u>CURVE 47<sup>a</sup></u>		673.2	0.197
1173.2	0.272	673.2	0.206	373.2	0.0900	746.2	0.149	642.1	0.196	439.6	0.145	773.2	0.213
1273.2	0.289	773.2	0.221	473.2	0.110	822.2	0.158	720.0	0.224	566.5	0.177	873.2	0.230
1373.2	0.314	<u>CURVE 26</u>		510.2	0.112	907.2	0.168	1037.4	0.278	843.3	0.220	973.2	0.247
1473.2	0.339	303.2	0.205	594.2	0.0979	979.2	0.172	1209.0	0.316	<u>CURVE 48<sup>a</sup></u>		1073.2	0.268
		373.2	0.205	673.2	0.100	1061.2	0.176	1224.8	0.306			1173.2	0.255
		473.2	0.209	746.2	0.111	1120.2	0.186	<u>CURVE 41<sup>a</sup></u>		<u>CURVE 49</u>		<u>CURVE 53<sup>a</sup></u>	
		573.2	0.222	846.2	0.138	1293.2	0.196	373.2	0.138	457.1	0.147	273.2	0.130
		673.2	0.243	873.2	0.144	<u>CURVE 35<sup>a</sup></u>		573.2	0.176	601.2	0.183	373.2	0.142
		773.2	0.259	939.2	0.156	413.7	0.172	773.2	0.215	778.8	0.203	473.2	0.163
		873.2	0.276	973.2	0.161	520.6	0.181	973.2	0.251	<u>CURVE 49</u>		573.2	0.184
		973.2	0.305	1073.2	0.174	532.1	0.190	<u>CURVE 42<sup>a</sup></u>		340.2	0.380	673.2	0.201
		1073.2	0.326			742.1	0.233	381.2	0.380	482.2	0.380	773.2	0.226
		1173.2	0.347	<u>CURVE 30<sup>a</sup></u>		743.2	0.266	593.2	0.376	593.2	0.346	973.2	0.301
		1273.2	0.372	298.2	0.146	<u>CURVE 36<sup>a</sup></u>		687.2	0.201	765.2	0.312	1073.2	0.301
		1373.2	0.410	1366.5	0.273	346.9	0.163	773.2	0.234	862.2	0.295	<u>CURVE 54</u>	
				18	0.0188	431.1	0.185	973.2	0.273	1061.2	0.301	273.2	0.192
				78	0.0824	460.6	0.190	<u>CURVE 43</u>		<u>CURVE 50</u>		373.2	0.205
				290	0.469	543.7	0.181	373.2	0.176	342.2	0.310	473.2	0.222
						927.9	0.263	573.2	0.205	573.2	0.336	573.2	0.234
				400.2	0.0753			973.2	0.278	395.2	0.336	673.2	0.255
				489.2	0.121	973.2	0.272	<u>CURVE 44</u>		529.2	0.333	773.2	0.276
				610.2	0.176			373.2	0.169	574.2	0.327	873.2	0.283
				703.2	0.232			573.2	0.195	<u>CURVE 51</u>		973.2	0.314
				776.3	0.247			773.2	0.218	1073.2	0.339	1073.2	0.339
				878.0	0.272			973.2	0.228	1173.2	0.364	<u>CURVE 55</u>	
				974.2	0.295			335.2	0.302	78.2	0.0879		
				1083.2	0.360			382.2	0.352				
				1272.5	0.383			504.2	0.339				

Not shown on plot

DATA TABLE NO. 331 (continued)

T	k	T	k	T	k	T	k	T	k
<u>CURVE 56</u>		<u>CURVE 60 (cont.)</u>		<u>CURVE 64 (cont.)</u> *		<u>CURVE 65 (cont.)</u> *			
78.2	0.0941	1242.7	0.263	343.7	0.146	600.9	0.187		
<u>CURVE 57</u>		1405.9	0.245	388.7	0.154	657.6	0.192		
78.2	0.0950	1503.2	0.297	391.5	0.154	662.6	0.192		
		1668.2	0.315	413.7	0.158	667.6	0.194		
<u>CURVE 58</u>		<u>CURVE 61</u> †		417.6	0.157	703.7	0.196		
4.50	0.09380	316.5	0.154	430.9	0.165	743.7	0.211		
7.00	0.09569	407.1	0.167	431.5	0.159	792.1	0.208		
8.80	0.09733	514.3	0.178	457.1	0.163	796.5	0.211		
10.40	0.09996	631.7	0.191	460.9	0.165	822.1	0.223		
11.00	0.10103	795.4	0.210	481.5	0.171	892.6	0.228		
15.60	0.1138	967.6	0.238	500.9	0.173	952.6	0.237		
15.70	0.1142	1131.5	0.257	514.3	0.175	1012.1	0.251		
19.00	0.1177	1172.6	0.263	520.4	0.175	1140.9	0.282		
20.70	0.1197	1336.5	0.296	589.8	0.189	1218.7	0.291		
32.20	0.1343	1623.2	0.305	605.4	0.190				
43.00	0.1435	<u>CURVE 62</u>		613.7	0.189				
60.00	0.1678	372.6	0.152	659.3	0.196				
70.00	0.1762	498.2	0.172	670.4	0.195				
88.50	0.1816	627.1	0.200	679.8	0.199				
<u>CURVE 59</u>		801.7	0.229	712.6	0.200				
316.5	0.231	953.7	0.253	722.1	0.204				
338.7	0.262	1037.1	0.278	799.8	0.221				
340.4	0.246	1170.4	0.293	800.4	0.220				
359.8	0.260	1339.3	0.326	897.1	0.236				
372.1	0.244	1457.1	0.343	958.2	0.244				
384.3	0.248	1498.2	0.348	982.6	0.246				
388.7	0.251	<u>CURVE 63</u>		1113.2	0.265				
422.1	0.252	100	0.090	1185.9	0.277				
473.2	0.254	116	0.106	<u>CURVE 65*</u>					
<u>CURVE 60</u>		131	0.103	272.6	0.133				
375.4	0.113	144	0.128	335.4	0.146				
528.2	0.138	156	0.126	346.5	0.148				
618.7	0.158	168	0.132	375.4	0.157				
765.4	0.191	180	0.145	394.3	0.157				
930.4	0.219	<u>CURVE 64*</u>		408.7	0.154				
1039.3	0.236	273.7	0.130	433.7	0.166				
1162.6	0.253	334.8	0.142	438.2	0.160				
				439.8	0.165				
				459.3	0.170				
				472.6	0.169				
				494.3	0.170				
				507.1	0.170				
				558.2	0.179				
				597.6	0.182				

Not shown on plot

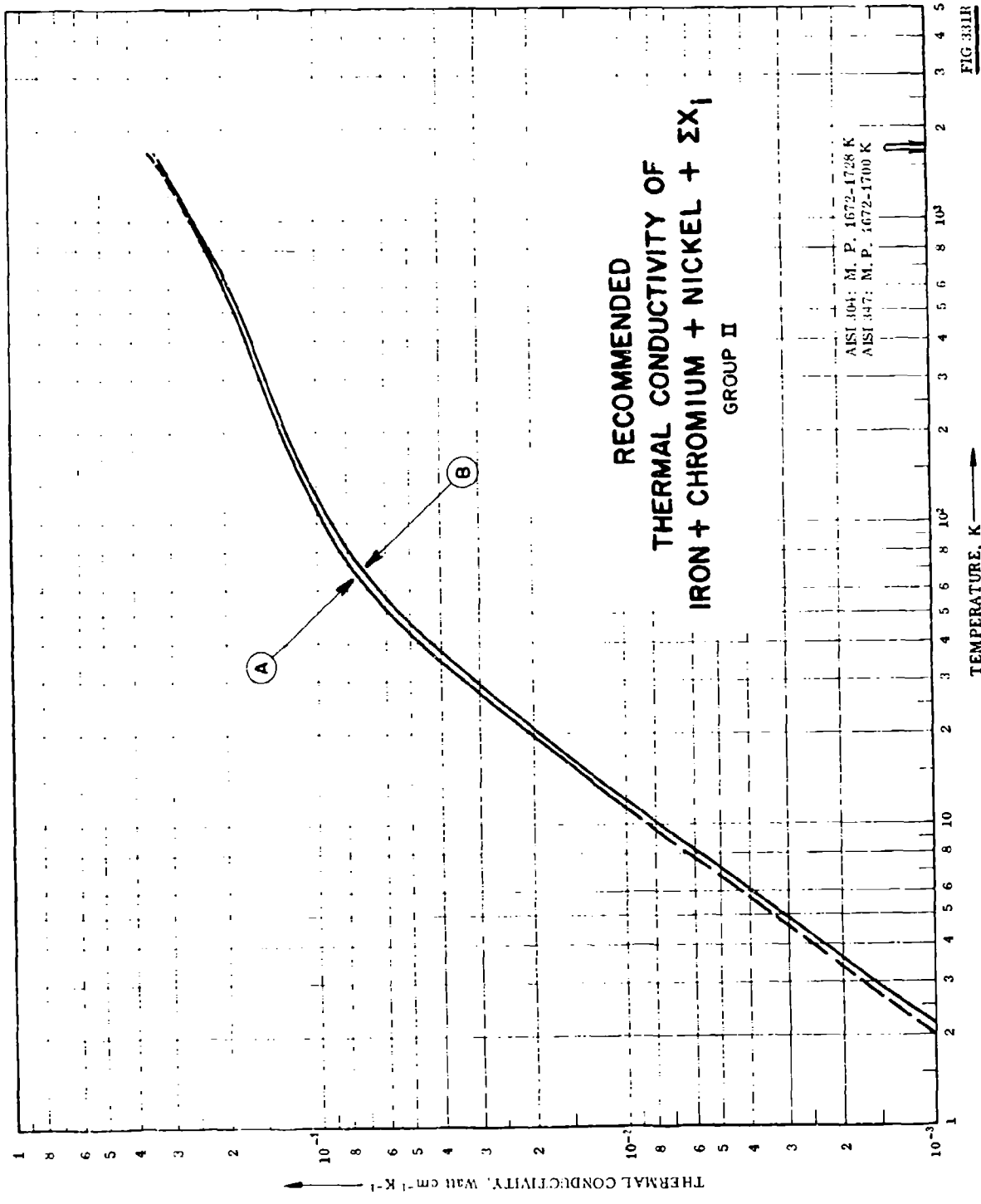


FIG 3B1R

SPECIFICATION TABLE NO. 331R RECOMMENDED THERMAL CONDUCTIVITY OF [IRON + CHROMIUM + NICKEL + ΣX<sub>i</sub>] GROUP II

[For Data Reported in Figure and Data Table No. 331R.]

Curve No.	Name and Designation	Nominal Composition (weight percent) and Remarks	Estimated Error
A	Stainless steel 304	18.00-20.00 Cr, 8.00-12.00 Ni, 2.00(max)Mn, 1.00(max)Si, and 0.08(max)C.	±10% below 100 K, ± 5% from 300 to 900 K, and ± 10% above 1400 K.
B	Stainless steel 347	17.00-19.00 Cr, 9.00-13.00 Ni, 2.00(max)Mn, 1.00(max)Si, 0.08(max)C, and 10 x C(min)Nb-Ta.	Same as above.

DATA TABLE NO. 331R RECOMMENDED THERMAL CONDUCTIVITY OF [IRON + CHROMIUM + NICKEL + ΣX<sub>i</sub>] GROUP II

[Temperature, T<sub>1</sub> in K and T<sub>2</sub> in F; Thermal Conductivity, k<sub>1</sub> in Watt cm<sup>-1</sup> K<sup>-1</sup> and k<sub>2</sub> in Btu hr<sup>-1</sup>(ft<sup>2</sup>F<sup>-1</sup>)]

T <sub>1</sub>	k <sub>1</sub>	k <sub>2</sub>	T <sub>2</sub>	T <sub>1</sub>	T <sub>2</sub>	k <sub>1</sub>	k <sub>2</sub>	CURVE A (cont.)		CURVE B		CURVE B (cont.)	
								T <sub>1</sub>	T <sub>2</sub>	T <sub>1</sub>	T <sub>2</sub>	T <sub>1</sub>	T <sub>2</sub>
0	0	0	-459.7	500	0.184	10.6	440.3	0	0	0.179	10.3	500	440.3
1	(0.00039)†	(0.0225)	-457.9	600	0.198	11.4	620.3	(0.00035)	(0.0202)	0.192	11.1	600	620.3
5	(0.0034)	(0.196)	-450.7	700	0.212	12.2	800.3	0.0031	0.179	0.205	11.4	700	800.3
10	(0.0085)	(0.491)	-441.7	800	0.225	13.0	980.3	0.0078	0.451	0.219	12.7	800	980.3
25	0.027	1.56	-414.7	900	0.239	13.8	1166	0.025	1.44	0.232	13.4	900	1166
50	0.058	3.35	-369.7	1000	(0.253)	(14.6)	1340	0.054	3.12	0.246	14.2	1000	1340
75	0.080	4.62	-324.7	1100	(0.267)	(15.4)	1520	0.076	1.39	0.259	15.0	1100	1520
100	0.095	5.49	-279.7	1200	(0.281)	(16.2)	1700	0.091	5.26	0.273	15.8	1200	1700
150	0.115	6.64	-189.7	1300	(0.295)	(17.0)	1880	0.111	6.41	0.286	16.5	1300	1880
200	0.130	7.51	-99.7	1400	(0.309)	(17.9)	2060	0.126	7.28	0.300	17.3	1400	2060
250	0.142	8.20	-9.7	1500	(0.323)	(18.7)	2240	0.138	7.97	0.313	18.1	1500	2240
273.2	0.147	8.49	32.0	1600	(0.337)	(19.5)	2420	0.143	8.26	(0.327)	(18.9)	1600	2420
300	0.152	8.73	80.3	1665	(0.347)	(20.0)	2537	0.148	8.55	(0.335)	(19.4)	1665	2537
350	0.162	9.36	170.3					0.157	9.07				
400	0.170	9.82	260.3					0.165	9.53				
450	0.177	10.2	350.3					0.172	9.94				

† Values in parentheses are extrapolated.

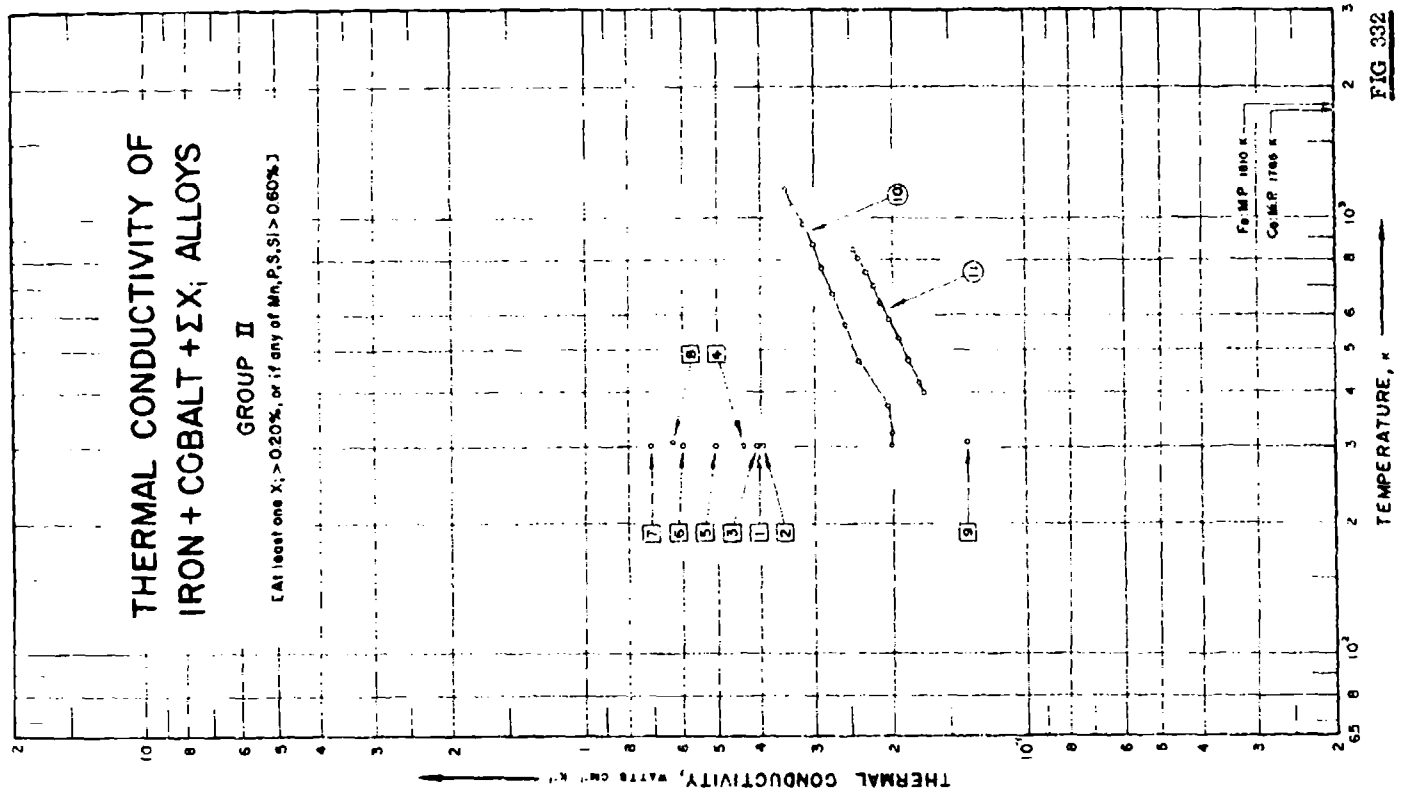


FIG 332

SPECIFICATION TABLE NO. 332 THERMAL CONDUCTIVITY OF [IRON + COBALT +  $\Sigma X_i$ ] ALLOYS GROUP II(At least one  $X_i > 0.20\%$  or if any of Mn, P, S, Si  $> 0.60\%$ )

[ For Data Reported in Figure and Table No. 332 ]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)							Composition (continued), Specifications and Remarks		
							Co	C	Cr	Cu	Mn	Ni	P		S	Si
1	208	E	1919	303.2		2	4.86	0.098		0.27	0.295	0.06	0.029	0.025	0.11	3.75-7.0 W; annealed at 1080 C and slowly cooled. 3.75-7.0 W; heated to 900 C for 30 min and quenched in oil, then requenched from 950 C. Forged. 2.5 W, 1. J Nb, 3.25 Mo.
2	208	E	1919	303.2		3	9.71	0.105		0.26	0.279	0.11	0.027	0.023	0.11	
3	208	E	1919	303.2		4	14.6	0.115		0.25	0.264	0.17	0.026	0.022	0.12	
4	208	E	1919	303.2		5	19.4	0.12		0.23	0.248	0.22	0.024	0.021	0.12	
5	208	E	1919	303.2		6	29.1	0.135		0.20	0.217	0.33	0.021	0.018	0.12	
6	208	E	1919	303.2		7	38.3	0.15		0.17	0.186	0.44	0.018	0.015	0.12	
7	208	E	1919	303.2		8	48.6	0.165		0.14	0.155	0.55	0.015	0.013	0.125	
8	172	E	1927	309.2		K. S. Magnet steel	35.0/ 41.0	0.90 5.75	3.50/ 5.75		0.30/ 0.85					
9	172	E	1927	309.2		K. S. Magnet steel	35.0/ 41.0	0.90 5.75	3.50/ 5.75		0.30/ 0.85					
10	163	L	1936	303-1173		10	26.00	0.07	20.47		0.42				0.51	
11	37	C	1951	409-851	4.0	Haynes alloy N-155	20.00	0.2	20.00		20.0					



DATA TABLE NO. 332 THERMAL CONDUCTIVITY OF [IRON + COBALT +  $\Sigma X_i$ ] ALLOYS GROUP II(At least one  $X_i \geq 0.20\%$  or if any of Mn, P, S, Si  $\geq 0.60\%$ )[Temperature, T, K Thermal Conductivity, k, Watt cm<sup>-1</sup> K<sup>-1</sup>]

T	k	T	k
<u>CURVE 1</u>			
303.20	0.402	773.20	0.239
		873.20	0.301
<u>CURVE 2</u>			
303.20	0.395	973.20	0.318
		1073.20	0.335
		1173.20	0.347
<u>CURVE 3</u>			
303.20	0.409	400.40	0.170
		422.00	0.174
<u>CURVE 4</u>			
303.20	0.435	477.60	0.184
		533.20	0.194
		588.70	0.204
		644.30	0.214
		699.80	0.222
		755.40	0.231
303.20	0.503	810.90	0.238
		851.00	0.245
<u>CURVE 5</u>			
303.20	0.598		
<u>CURVE 6</u>			
303.20	0.711		
<u>CURVE 7</u>			
309.20	0.634		
<u>CURVE 8</u>			
309.20	0.137		
<u>CURVE 9</u>			
303.20	0.201		
323.20	0.201		
373.20	0.205		
473.20	0.219		
573.20	0.255		
673.20	0.312		

# THERMAL CONDUCTIVITY OF IRON + COPPER + ΣX<sub>i</sub> ALLOYS

## GROUP I

[X<sub>i</sub>: 0.20% except C is 2.00% and Mn, P, S, Si is 0.00% each]

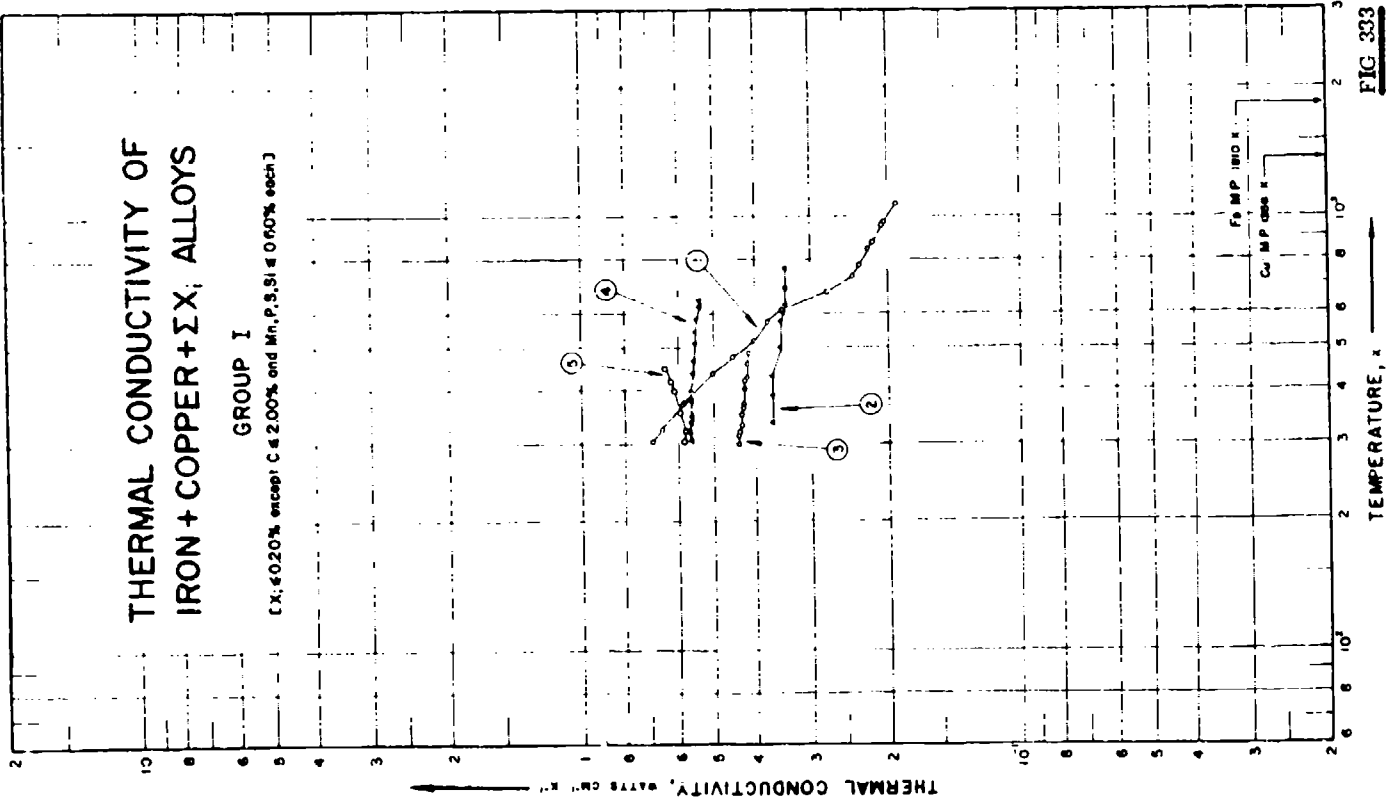


FIG 333

SPECIFICATION TABLE NO. 333 THERMAL CONDUCTIVITY OF IRON + COPPER +  $\Sigma$  ALLOYS GROUP I(X<sub>1</sub> = 0.20% except C = 2.00% and Mn, P, S, Si = 0.60% each)

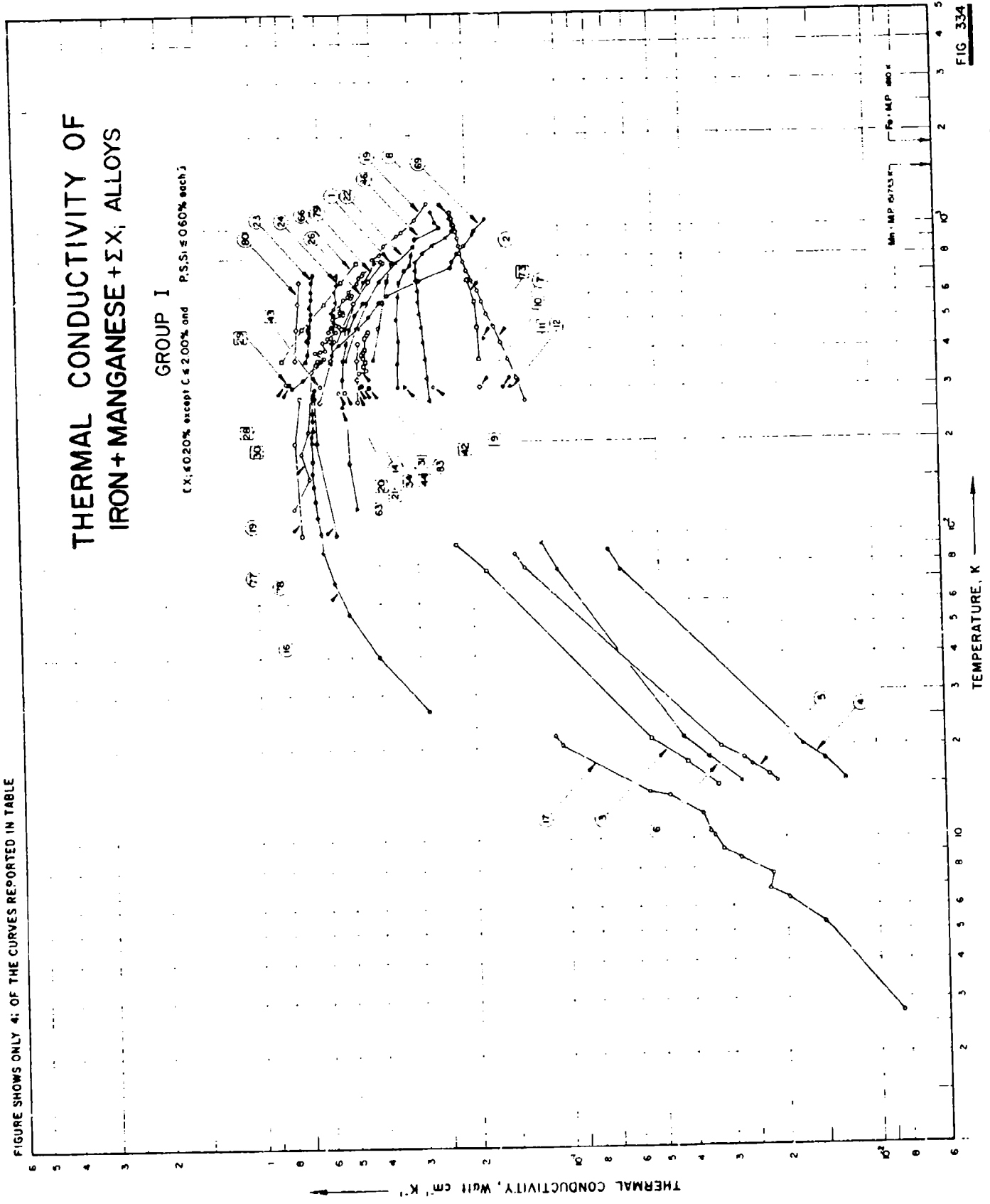
[For Data Reported in Figures and Table No. 333]

Curve No.	Ref. Method Used	Year	Temp. Range, K	Reported Error, %	Name and Designation of Specimen	Composition (weight percent)					Composition (continued), Specifications and Remarks	
						Cu	C	Mn	P	S		Si
1	173	P	303-1073		Carbon steel	0.638	0.22	0.38	0.041	0.067	0.030	Annealed at 850 C for 2.5 hrs. FeO + Fe <sub>2</sub> O <sub>3</sub> = 9.30, porosity = 17%; without heat treatment. Porosity = 10%; without heat treatment. Porosity = 10.5%; without heat treatment. Porosity = 10.7%; without heat treatment.
2	192	E	315-762			5.0	1.66	0.44			0.35	
3	192	E	293-485			10.54	1.6	0.42			0.29	
4	192	E	305-646			19.86	1.62	0.39			0.31	
5	192	E	305-448			40.69	1.51	0.40			0.29	

DATA TABLE NO. 333 THERMAL CONDUCTIVITY OF [IRON + COPPER +  $\Sigma X_i$ ] ALLOYS GROUP I(X<sub>i</sub> ≤ 0, 20% except C ≤ 2, 00% and Mn, P, S, Si ≤ 0, 60% each)[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup> K<sup>-1</sup>]

T	k	T	k
<u>CURVE 1</u>			
303.20	0.688	426.70	0.418
323.20	0.654	458.90	0.418
370.20	0.587	484.60	0.414
373.20	0.583	<u>CURVE 4</u>	
435.20	0.501	304.70	0.561
473.20	0.451	317.60	0.565
500.20	0.418	336.10	0.561
573.20	0.374	347.20	0.561
612.20	0.346	377.70	0.561
673.20	0.275	398.40	0.561
730.20	0.239	436.20	0.556
773.20	0.231	468.50	0.552
841.20	0.220	512.40	0.548
873.20	0.215	544.20	0.548
954.20	0.206	581.20	0.544
973.20	0.203	621.10	0.536
1073.20	0.190	646.20	0.536
<u>CURVE 2</u>			
335.40	0.364	<u>CURVE 5</u>	
388.30	0.364	304.80	0.582
430.50	0.364	320.90	0.577
501.10	0.347	354.20	0.594
576.50	0.347	397.20	0.611
601.00	0.343	417.90	0.623
628.20	0.339	448.20	0.640
684.70	0.339	<u>CURVE 3</u>	
761.70	0.339	298.40	0.439
<u>CURVE 3</u>			
312.20	0.439	319.10	0.435
330.50	0.431	350.00	0.431
350.00	0.431	350.20	0.431
363.20	0.427	381.40	0.427
381.40	0.427	401.50	0.423
420.40	0.423	420.40	0.423

FIGURE SHOWS ONLY 4; OF THE CURVES REPORTED IN TABLE



SPECIFICATION TABLE NO. 334 THERMAL CONDUCTIVITY OF [IRON + MANGANESE + ΣX<sub>i</sub>] ALLOYS GROUP I

(X<sub>i</sub> ≤ 0.20% except C ≤ 2.00% and P, S, Si ≤ 0.60% each)

[For Data Reported in Figure and Table No. 334]

Curve No.	Ref. Method	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight per cent)										Composition (continued), Specifications and Remarks		
						Mn	Al	C	Cr	Cu	Ni	P	S	Si				
1	129	C	1933	373-773	3.0-5.0	Low Mn steel	1.65		0.51				0.10	0.016	0.023	0.24	Normalized at 900 C.	
2	160	F	1938	373-573		Austenitic steel	13.50		1.12									Nominal composition: austenitic.
3	104	L	1951	15-93		53	2.23		0.41							0.07		Heated to 800 C and cooled in furnace.
4	104	L	1951	16-88		1010	12.69		1.27							0.12		Heated to 1000 C and quenched in water.
5	104	L	1951	15-86		1379 E	12.95		0.09				0.050	0.103	0.12			Heated to 1000 C and quenched in water.
6	104	L	1951	15-93		1379 H	38.9		0.20				0.36	0.055	0.70			Heated to 1000 C and quenched in water.
7	164	C	1946	273-1123		13	13.0	0.004	1.22	0.03	0.070	0.07	0.038	0.010	0.22			0.038 As. heated to 1050 C and cooled in air.
8	164	C	1946	273-1123		G	13.0	0.004	1.22	0.03	0.070	0.07	0.039	0.010	0.22			0.038 As. heated to 1050 C and air cooled, then heat-treated at 450 C for 102 days.
9	165	E	1919	303.2		8	5.0	0.056	0.202				0.044	0.034	0.072			
10	165	E	1919	303.2		9	5.0	0.095	0.199				0.047	0.034	0.080			
11	165	E	1919	303.2		10	7.9	0.102	0.195				0.046	0.037	0.092			
12	165	E	1919	303.2		11	8.8	0.170	0.194				0.046	0.036	0.096			
13	165	E	1919	303.2		12	9.8	0.189	0.192				0.045	0.036	0.10			
14	166	C	1939	273-423		4	1.50	0.015	0.23	0.06	0.105	0.04	0.037	0.038	0.12			0.033 As. 0.025 Mo; annealed at 860 C.
15	166	C	1939	273-623		13	13.0	0.004	1.22	0.03	0.070	0.07	0.038	0.010	0.22			0.036 As. heated to 1050 C and cooled in air.
16	115	L	1951	27-299		SAE 1020 steel	0.33		0.18						0.014			
17	81	L	1939	2.7-22			0.50		0.4				0.03	0.03	0.20			
							0.70								0.35			
18	17	L	1958	293-363	1.0	42.11 C steel	0.68		0.31				0.021	0.034	0.28			
19	91	C	1951	123-1194		SAE 1010 steel	0.42		0.10				0.008	0.28				Hot rolled.
20	167		1935	293.2			≤0.5		0.10						≤0.2			
21	167		1935	293.2			≤0.5		0.3						≤0.2			
22	168	L	1932	373-473	5.0		0.5		0.12	0.05			0.05	0.05	0.12			
23	31	L	1933	570-710	2.0	CS 1	0.34		0.10				0.031	0.041	0.071			Normalized at 900 C.
24	31	L	1933	360-705	2.0	CS 2	0.61		0.26				0.025	0.053	0.1			Normalized at 900 C.
25	31	L	1933	364-706	2.0	CS 3	0.67		0.44				0.024	0.037	0.102			Normalized at 900 C.
26	169	<del>+</del> R	1936	273-773		Steel 1	0.17		0.04						0.02			
27	169	<del>+</del> R	1936	273-516		Steel 2	0.37		0.10						0.17			
28	170	L	1926	313.2		1.1	0.13		0.08				0.009	0.005	0.03			Annealed.
29	170	L	1926	313.2		1.2	0.13		0.08				0.009	0.005	0.03			Forged.
30	170	L	1926	313.2		1.3 h	0.13		0.08				0.009	0.005	0.03			Annealed, and then quenched from 800 C.

SPECIFICATION TABLE NO. 334 (continued)

Curve No.	Ref. Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Fe	Mn	Al	Composition (weight per cent)					S	SI	Composition (continued), Specifications and Remarks
									C	Cr	Cu	Ni	P			
31	171	E	1917	303.2		Krupp steel, 2a	0.39		0.14		0.10		0.015	0.030	0.31	Forged.
32	171	E	1917	303.2		Krupp steel, 2b	0.39		0.14		0.10		0.015	0.030	0.31	Annealed in vacuum at 900 C for 1 hr. Quenched in oil from 900 C.
33	171	E	1917	303.2		Krupp steel, 2c	0.39		0.14		0.10		0.015	0.030	0.31	Quenched in oil from 900 C.
34	171	E	1917	303.2		Krupp steel, 4a	0.34		0.18		0.10		0.051	0.044	0.10	Forged.
35	171	E	1917	303.2		Krupp steel, 4b	0.34		0.18		0.10		0.051	0.044	0.10	Annealed at 900 C for 1 hr in vacuum.
36	171	E	1917	303.2		Krupp steel, 4c	0.34		0.18		0.10		0.051	0.044	0.10	Quenched in oil from 900 C.
37	171	E	1917	303.2		Krupp steel, 6a	0.65		0.31		0.13		0.013	0.028	0.23	Forged.
38	171	E	1917	303.2		Krupp steel, 6b	0.65		0.31		0.13		0.013	0.028	0.23	Annealed at 900 C for 1 hr in vacuum.
39	171	E	1917	303.2		Krupp steel, 6c	0.65		0.31		0.13		0.013	0.028	0.23	Quenched in oil from 900 C.
40	171	E	1917	303.2		Krupp steel, 8a	0.67		0.44		0.09		0.043	0.043	0.34	Forged.
41	171	E	1917	303.2		Krupp steel, 8b	0.67		0.44		0.09		0.043	0.043	0.34	Annealed at 900 C for 1 hr in vacuum.
42	171	E	1917	303.2		Krupp steel, 8c	0.67		0.44		0.09		0.043	0.043	0.34	Quenched in oil from 300 C.
43	172	E	1927	307.2		1	0.36		0.095				0.021	0.021	Trace	
44	172	E	1927	307.2		2	0.43		0.395				0.023	0.021	0.33	
45	71	L	1917	303-1199		4	0.34		0.18		0.10		0.051	0.044	0.10	Forged.
46	71	L	1917	303-1135		8	0.67		0.44		0.09		0.043	0.043	0.34	Forged.
47	173	C	1956	323-1123		En 8a	1.05		0.39			0.12	0.032	0.043	0.14	Normalized.
48	173	C	1956	323.373		En 9b	1.05		0.39			0.12	0.032	0.043	0.14	Quenched in oil from 850 C.
49	173	C	1956	323.373		En 8c	1.05		0.39			0.12	0.032	0.043	0.14	Tempered 3 hrs at 150 C after being quenched in oil from 850 C.
50	173	C	1956	323-473		En 8d	1.05		0.39			0.12	0.032	0.043	0.14	Tempered 3 hrs at 350 C after being quenched in oil from 850 C.
51	173	C	1956	323-473		En 8e	1.05		0.39			0.12	0.032	0.043	0.14	Tempered 3 hrs at 550 C after being quenched in oil from 850 C.
52	173	C	1956	323-473		En 8f	1.05		0.39			0.12	0.032	0.043	0.14	Tempered 3 hrs at 650 C after being quenched in oil from 850 C.
53	173	C	1956	323-473		En 8g	1.05		0.39			0.12	0.032	0.043	0.14	Annealed 1 hr at 850 C after being quenched in oil from 850 C.
54	173	C	1956	323-873		En 8h	1.05		0.39			0.12	0.032	0.043	0.14	Reheated at 650 C for 120 hrs after being quenched in oil from 850 C.
55	165	E	1919	303.2		2	0.31	0.004	0.206				0.05	0.04	0.061	
56	165	E	1919	303.2		3	6.6	0.010	0.295				0.05	0.04	0.062	
57	165	E	1919	303.2		4	0.8	0.013	0.205				0.05	0.04	0.063	

SPECIFICATION TABLE NO. 334 (continued)

Curve No.	ReL No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (Weight Percent)										Composition (continued), Specifications and Remarks	
							Mn	Al	C	Cr	Cu	P	S	Si	S	Si		
58	165	E	1919	303		5	1.1	0.019	0.205						0.049	0.040	0.064	
59	165	E	1919	303		6	1.6	0.029	0.204						0.049	0.039	0.066	
60	165	E	1919	303		7	2.0	0.017	0.203						0.049	0.039	0.068	
61	174	F	1954	298-1055	5.0	AISI 1010	0.42	0.08							0.015	0.020	0.19	Basic C-1010 fully aluminum-killed steel; annealed at 800 C.
62	166	C	1939	273-623		2	0.32	0.002	0.08	0.045	Trace	0.07	0.029	0.050	0.08			0.032 As; 0.020 Mo; annealed at 830 C.
63	166	C	1939	273-573		3	0.635	0.010	0.23	Trace	0.13	0.074	0.034	0.034	0.11			0.036 As; annealed at 830 C.
64	166	C	1939	273-623		5	0.643	0.006	0.415	Trace	0.12	0.063	0.031	0.029	0.11			0.033 As; annealed at 860 C.
65	166	C	1939	273-623		6	0.69	0.006	0.425	0.03	0.060	0.04	0.037	0.038	0.20			0.024 As; 0.01 Mo; annealed at 860 C.
66	160	F	1938	373-773		1	0.40		0.065									Trace Nominal composition.
67	160	F	1938	373-773		10	0.94		0.29									Nominal composition.
68	160	F	1938	373-773		11	0.63		0.52									Nominal composition.
69	175	P	1936	303-1073		1	0.16		0.04		0.150		0.007	0.016	0.031			Annealed at 850 C for 2.5 hrs.
70	175	P	1936	303-1073		Carbon steel	0.45		0.28		0.228		0.016	0.056	0.014			Annealed at 850 C for 2.5 hrs.
71	175	P	1936	303-1073		Carbon steel	0.37		0.12		0.148		0.018	0.016	0.009			Annealed at 850 C for 2.5 hrs.
72	561	C	1925	442.2			0.58		0.39			0.20						Quenched in oil from 1200 C.
73	561	C	1925	431.4		12			1.2									
74	561	C	1925	449.7		CS, Japanese	0.48		0.33									
75	539	L	1938	323-1122		Carbon steel	0.45		0.18	0.07			0.019	0.007	0.24			0.13 Mo; annealed.
76	31	L	1933	369-715	2.0	Wrought iron	0.20		Trace				0.007	0.014	0.092			Commercially pure.
77	671	L	1961	100-280		2	0.37											Original material remelted and rolled into bars with a cross-section of about 15 mm <sup>2</sup> and a length of 100 mm; after a short rolling annealed at 1373 K for 2 hrs in evacuated silica tubes, rolled to final form and annealed at about 773 K for 10 hrs; electrical resistivity 3.1, 7.5 and 12.2 μohm cm at 90, 193, and 290 K respectively; original material supplied by Heraeus, A. G. Inc., Hanau, Germany.





SPECIFICATION TABLE NO. 334 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)							Composition (continued), Specifications and Remarks			
							Mn	Al	C	Cr	Cu	Ni	P		S	Si	
91	976	L	1933	373-673		British steel	0.61		0.26					0.025	0.053	0.14	
92	976	L	1933	373-673		British steel	0.67		0.44					0.024	0.037	0.11	

DATA TABLE NO. 334 THERMAL CONDUCTIVITY OF [IRON + MANGANESE + EX<sub>i</sub>] ALLOYS

GROUP I

(X<sub>i</sub> ≤ 0.20% except C ≤ 2.00% and P, S, Si ≤ 0.60% each)[Temperature, T, K; Thermal Conductivity, k, Watts cm<sup>-1</sup>K<sup>-1</sup>]

T	k	T	k	T	k	T	k	T	k	T	k	T	k	T	k
CURVE 1															
373.20	0.403	273.20	0.132	303.20	0.184	204.43	0.650	431.50	0.544	370.20	0.678	500.70	0.519	303.20	0.423
473.20	0.389	323.20	0.140	211.51	0.652	211.51	0.652	439.40	0.573	391.20	0.669	522.20	0.515	303.20	0.423
573.20	0.376	373.20	0.149	220.71	0.648	220.71	0.648	463.60	0.536	431.70	0.661	534.70	0.527	303.20	0.423
673.20	0.363	423.20	0.157	238.33	0.648	238.33	0.648	465.30	0.569	436.70	0.669	553.70	0.527	303.20	0.423
773.20	0.349	473.20	0.165	256.07	0.647	256.07	0.647	479.60	0.552	448.20	0.657	585.20	0.519	303.20	0.423
CURVE 2															
523.20	0.174	523.20	0.174	284.38	0.648	284.38	0.648	482.20	0.523	462.70	0.657	626.20	0.515	303.20	0.423
573.20	0.180	573.20	0.180	290.23	0.641	290.23	0.641	511.60	0.548	471.70	0.661	665.20	0.506	303.20	0.423
623.20	0.186	623.20	0.186	298.54	0.634	298.54	0.634	527.80	0.506	484.20	0.653	705.70	0.502	303.20	0.423
673.20	0.185	673.20	0.185	CURVE 17											
723.20	0.187	723.20	0.201	533.90	0.519	533.90	0.519	535.90	0.506	531.70	0.649	CURVE 25*			
773.20	0.190	773.20	0.205	552.70	0.523	552.70	0.523	588.10	0.506	588.10	0.644	523.20	0.644	303.20	0.423
873.20	0.200	873.20	0.209	588.10	0.506	588.10	0.506	596.60	0.477	590.70	0.644	573.20	0.551	303.20	0.423
CURVE 3															
14.94	0.0350	973.20	0.213	2.70	0.00844	2.70	0.00844	602.20	0.490	627.20	0.644	573.20	0.515	303.20	0.423
17.82	0.0417	1023.20	0.216	5.30	0.0151	5.30	0.0151	609.40	0.473	672.70	0.640	573.20	0.468	303.20	0.423
21.33	0.0541	1073.20	0.222	6.38	0.0196	6.38	0.0196	625.40	0.485	709.70	0.632	573.20	0.418	303.20	0.423
26.00	0.081	1123.20	0.224	7.64	0.0222	7.64	0.0222	664.50	0.456	CURVE 24					
92.60	0.226	CURVE 18*													
CURVE 4															
15.56	0.0128	273.20	0.268	10.56	0.0351	10.56	0.0351	719.70	0.431	366.20	0.561	523.20	0.785	303.20	0.423
18.09	0.0148	323.20	0.272	12.04	0.0471	12.04	0.0471	779.50	0.377	390.70	0.556	523.20	0.785	303.20	0.423
21.06	0.0174	373.20	0.278	13.93	0.0573	13.93	0.0573	792.50	0.402	427.70	0.556	573.20	0.502	303.20	0.423
25.80	0.0260	423.20	0.282	14.29	0.0600	14.29	0.0600	818.90	0.381	433.70	0.556	573.20	0.473	303.20	0.423
85.20	0.0725	473.20	0.287	20.30	0.105	20.30	0.105	873.90	0.368	434.70	0.552	536.40	0.464	303.20	0.423
CURVE 5															
15.31	0.0211	523.20	0.289	21.80	0.111	21.80	0.111	942.20	0.335	451.20	0.548	CURVE 26			
16.11	0.0225	573.20	0.291	21.80	0.111	21.80	0.111	967.80	0.322	472.20	0.548	313.20	0.787	303.20	0.423
17.48	0.0255	623.20	0.289	293.20	0.463	293.20	0.463	1062.30	0.293	496.70	0.544	313.20	0.785	303.20	0.423
18.29	0.0271	673.20	0.291	323.20	0.468	323.20	0.468	1198.00	0.268	523.20	0.544	313.20	0.785	303.20	0.423
77.80	0.136	723.20	0.293	353.20	0.468	353.20	0.468	CURVE 19 (cont.)							
86.10	0.146	773.20	0.293	293.20	0.531	293.20	0.531	431.50	0.544	431.50	0.544	500.70	0.519	303.20	0.423
CURVE 6															
15.35	0.0276	823.20	0.276	293.20	0.531	293.20	0.531	439.40	0.573	439.40	0.573	522.20	0.515	303.20	0.423
18.40	0.0373	873.20	0.257	361.70	0.628	361.70	0.628	463.60	0.536	463.60	0.536	534.70	0.527	303.20	0.423
21.39	0.0425	923.20	0.232	373.20	0.500	373.20	0.500	465.30	0.569	465.30	0.569	553.70	0.527	303.20	0.423
76.60	0.107	973.20	0.220	373.20	0.500	373.20	0.500	479.60	0.552	479.60	0.552	585.20	0.519	303.20	0.423
93.10	0.119	1023.20	0.218	373.20	0.500	373.20	0.500	482.20	0.523	482.20	0.523	626.20	0.515	303.20	0.423
CURVE 7															
15.35	0.0276	1073.20	0.222	373.20	0.500	373.20	0.500	484.20	0.506	484.20	0.506	665.20	0.506	303.20	0.423
18.40	0.0373	1123.20	0.224	373.20	0.500	373.20	0.500	511.60	0.548	511.60	0.548	705.70	0.502	303.20	0.423
21.39	0.0425	CURVE 20													
76.60	0.107	26.54	0.282	122.50	0.749	122.50	0.749	373.20	0.500	365.20	0.544	CURVE 31			
93.10	0.119	40.03	0.404	154.00	0.669	154.00	0.669	473.20	0.531	388.20	0.540	303.20	0.423	307.20	0.605
CURVE 8															
15.35	0.0276	55.29	0.506	186.30	0.707	186.30	0.707	473.20	0.531	428.20	0.531	303.20	0.423	307.20	0.605
18.40	0.0373	70.12	0.565	218.50	0.669	218.50	0.669	473.20	0.531	435.70	0.536	303.20	0.423	307.20	0.605
21.39	0.0425	86.53	0.611	345.50	0.644	345.50	0.644	473.20	0.531	435.70	0.536	303.20	0.423	307.20	0.605
76.60	0.107	101.69	0.620	361.70	0.628	361.70	0.628	473.20	0.531	435.70	0.536	303.20	0.423	307.20	0.605
93.10	0.119	114.94	0.632	371.80	0.619	371.80	0.619	473.20	0.531	435.70	0.536	303.20	0.423	307.20	0.605
CURVE 9															
15.35	0.0276	144.88	0.648	378.70	0.586	378.70	0.586	473.20	0.531	435.70	0.536	303.20	0.423	307.20	0.605
18.40	0.0373	160.01	0.653	395.60	0.628	395.60	0.628	473.20	0.531	435.70	0.536	303.20	0.423	307.20	0.605
21.39	0.0425	175.14	0.650	399.50	0.603	399.50	0.603	473.20	0.531	435.70	0.536	303.20	0.423	307.20	0.605
76.60	0.107	203.60	0.632	400.60	0.582	400.60	0.582	473.20	0.531	435.70	0.536	303.20	0.423	307.20	0.605
93.10	0.119	203.60	0.632	427.50	0.594	427.50	0.594	473.20	0.531	435.70	0.536	303.20	0.423	307.20	0.605

\* Not shown on plot



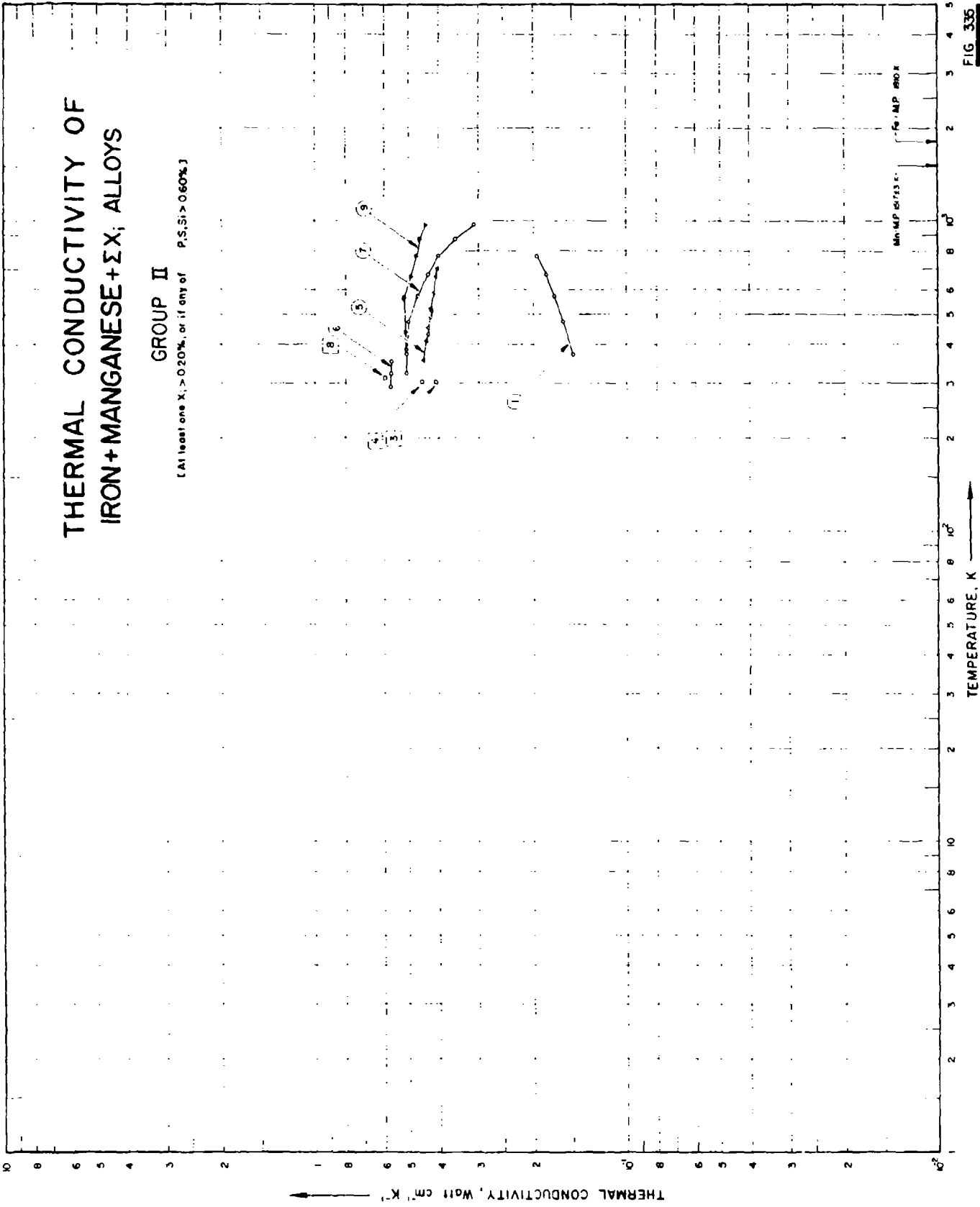
DATA TABLE NO. 334 (continued)

T	k
<u>CURVE 85*</u>	
373.2	0.435
<u>CURVE 87*</u>	
373.2	0.485
<u>CURVE 88*</u>	
373.2	0.50
<u>CURVE 89†</u>	
373.2	0.485
<u>CURVE 90*</u>	
293.20	0.430
323.20	0.437
353.50	0.445
<u>CURVE 91*</u>	
373.2	0.561
473.2	0.552
573.2	0.544
673.2	0.536
<u>CURVE 92†</u>	
373.2	0.540
473.2	0.527
573.2	0.519
673.2	0.506

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 † Not shown on plot

FIGURE SHOWS ONLY 8 OF THE CURVES REPORTED IN TABLE



SPECIFICATION TABLE NO. 335 THERMAL CONDUCTIVITY OF [IRON + MANGANESE +  $\Sigma X_i$ ] ALLOYS GROUP II(At least one  $X_i > 0.20\%$  or if any of P, S, Si  $> 0.60\%$ )

[ For Data Reported in Figure and Table No. 335 ]

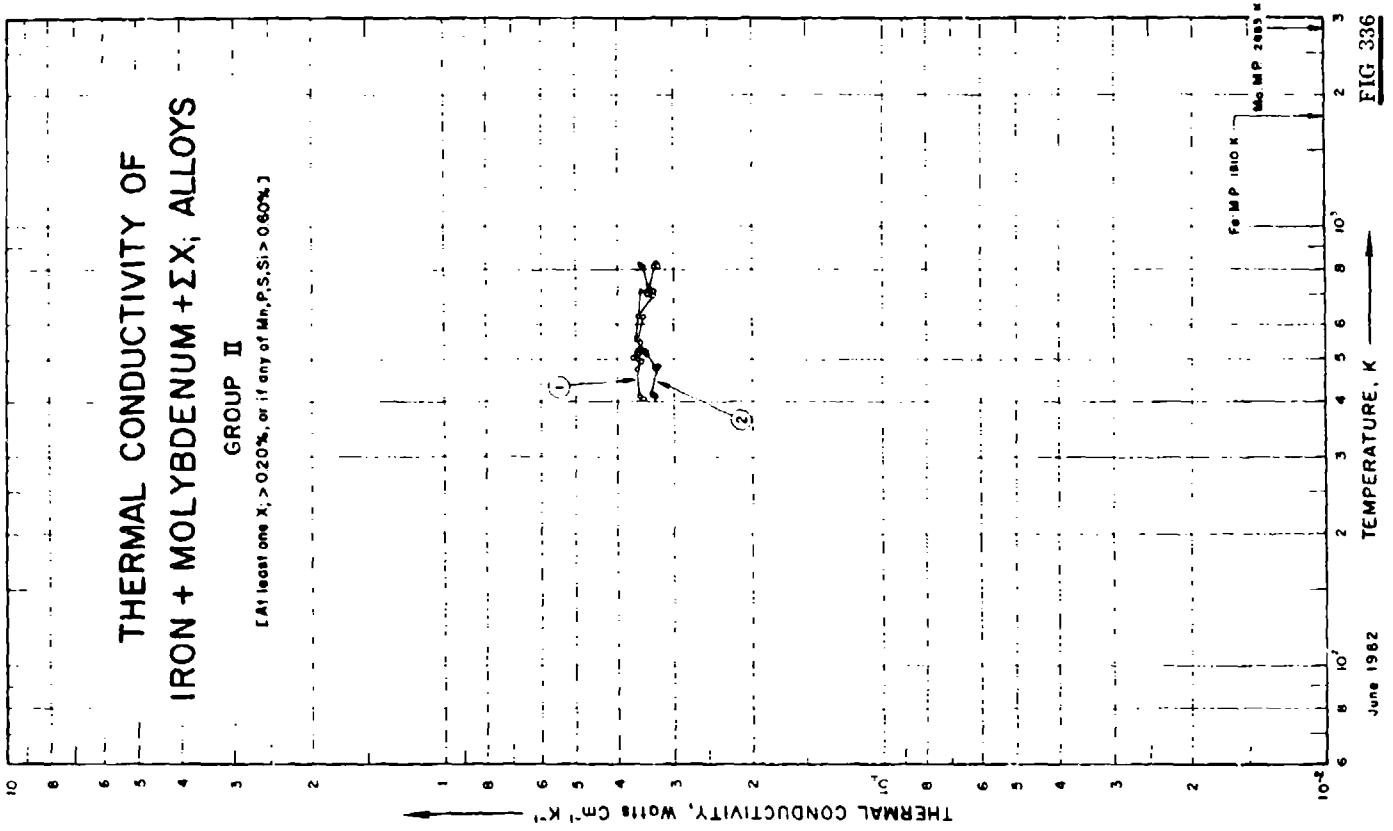
Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Mn	C	Cr	Cu	Mo	Ni	P	S	Si	Composition (continued), Specifications and Remarks
1	129	C	1933	373-773	3.0-5.0	Mn-Ni steel	12.0/ 13.0	0.70/ 0.80				3.0				Commercially pure.
2	178	E	1918	303		1a	0.31	0.09		0.288			0.03	0.026	0.11	Annealed at 900 C.
3	178	E	1918	303		1b	0.31	0.05		0.288			0.03	0.026	0.11	Cooled once to -190 C.
4	208	E	1919	303		1	0.31	0.09		0.288			0.03	0.026	0.11	
5	179	L	1939	355-706	2.0	Mn-Ni cast iron	3.11	3.10				1.00			2.51	Cast in mold.
6	17	L	1958	293-353	1.0	42.11 a	0.40	0.33	0.08	0.25	0.03				< 0.3	< 0.05 Ti, < 0.03 Mg.
7	209	C	1956	323-573	2.9	B.S. 970 En32A; BGK 1	0.42	0.10	0.07			0.23	0.018	0.036	0.21	
8	204	L	1937	312		Russian alloy 22	32.50	4.55					0.87		0.037	
9	664	E	1957	387-375		Russian steel 12 MKh	0.7	0.16	0.6	0.25	0.6	0.3	0.04	0.04	0.3	Specimen 4 mm in dia and 120 mm long; tempered at 930-940 C (cooling medium is water) and aged at 680-690 C for 2 hrs.

DATA TABLE NO. 335 THERMAL CONDUCTIVITY OF [IRON + MANGANESE + EX<sub>1</sub>] ALLOYS GROUP II(At least one X<sub>1</sub> > 0.20% or if any of P, S, Si > 0.60%)Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup> K<sup>-1</sup>

T	k	T	k
<u>CURVE 1</u>			
373.20	0.148	673.20	0.435
473.20	0.160	773.20	0.400
573.20	0.171	873.20	0.365
673.20	0.183	973.20	0.310
773.20	0.195		
<u>CURVE 2*</u>			
302.2	0.410	311.9	0.597
<u>CURVE 3</u>			
202.2	0.410	396.7	0.510
		439.7	0.512
		565.2	0.516
<u>CURVE 4</u>			
303.2	0.456	664.2	0.492
		771.2	0.472
		876.2	0.460
		974.5	0.441
<u>CURVE 5</u>			
355.70	0.448		
413.70	0.439		
431.70	0.435		
455.20	0.435		
490.20	0.427		
513.70	0.423		
522.70	0.427		
589.70	0.414		
662.76	0.410		
706.20	0.402		
<u>CURVE 6</u>			
293.20	0.575		
323.2	0.573		
353.2	0.571		
<u>CURVE 7</u>			
323.20	0.510		
373.20	0.510		
473.20	0.500		
573.20	0.470		

\* Not shown on plot





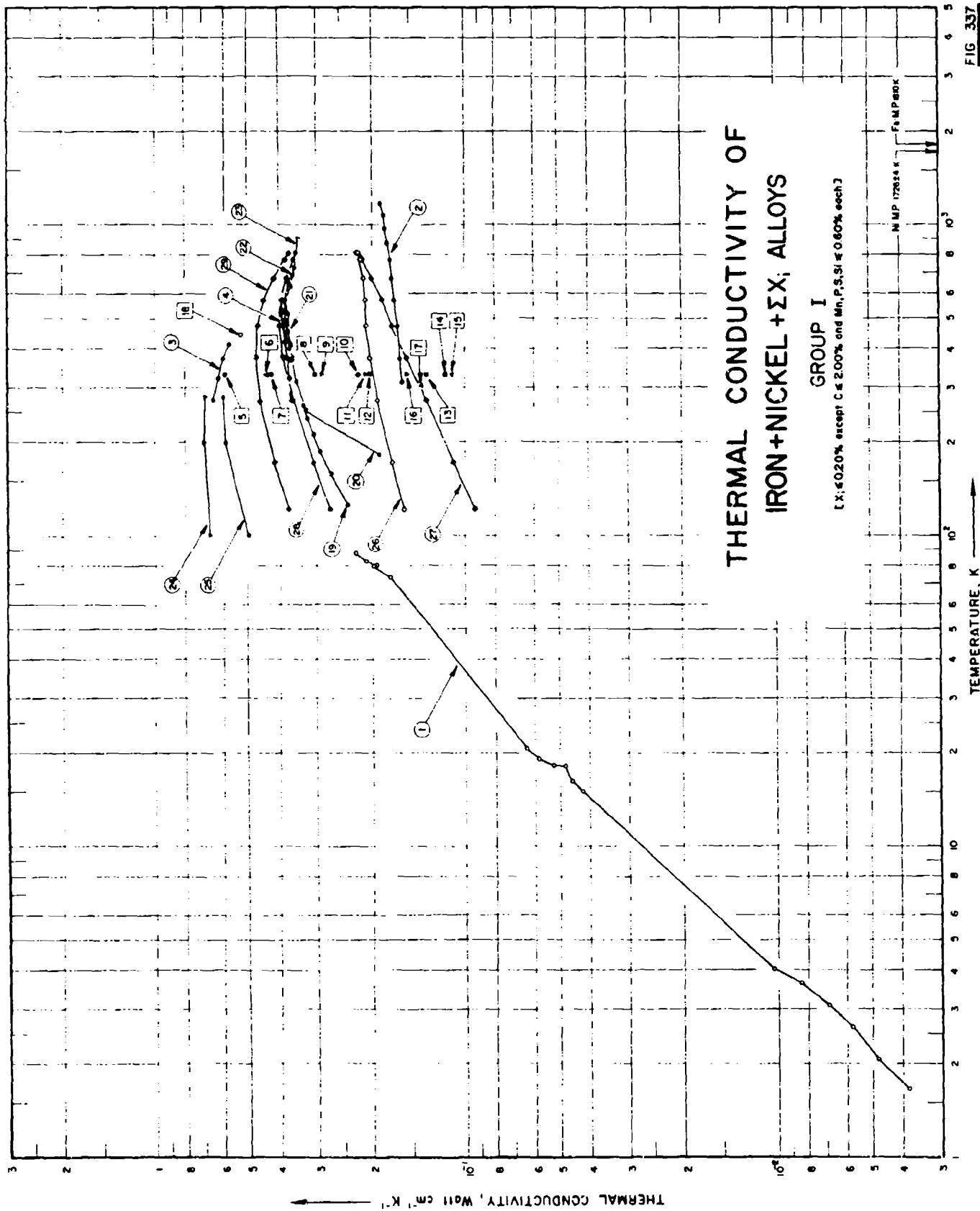
SPECIFICATION TABLE NO. 336 THERMAL CONDUCTIVITY OF [IRON + MOLYBDENUM +  $\Sigma X_i$ ] ALLOYS GROUP I  
 (At least one  $X_i > 0.20\%$  or if any of Mn, P, S, Si  $> 0.60\%$ )

[For Data Reported in Figure and Table No. 336]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)					V	W	Composition (continued), Specifications and Remarks
							Mo	Al	C	Cr	Mn			
1	340	L	1956	408-822	7.0	High speed steel; M-1	8.5		4.0			1.0	1.5	Annealed.
2	340	L	1956	413-820	7.0	High speed steel; M-10	8.0		4.0			2.0		Annealed.

DATA TABLE NO. 336 THERMAL CONDUCTIVITY OF [IRON + MOLYBDENUM +  $\Sigma X_i$ ] ALLOYS GROUP II(At least one  $X_i > 0.20\%$  or if any of Mn, P, S, Si  $> 0.60\%$ )[Temperature, T, K; Thermal Conductivity, k, Watt  $\text{cm}^{-1} \text{K}^{-1}$ ]

T	k
<u>CURVE 1</u>	
407.6	0.353
411.9	0.359
474.1	0.365
494.3	0.356
499.1	0.362
503.1	0.373
516.3	0.362
522.4	0.365
547.4	0.359
554.7	0.361
675.6	0.352
628.3	0.362
700.2	0.335
701.9	0.345
712.8	0.345
716.5	0.335
716.5	0.342
816.0	0.334
817.6	0.327
821.5	0.330
<u>CURVE 2</u>	
413.1	0.332
417.0	0.333
417.4	0.337
478.9	0.330
484.4	0.328
484.4	0.334
517.3	0.346
522.4	0.347
522.4	0.353
557.6	0.366
557.6	0.364
717.8	0.356
720.1	0.345
818.2	0.352
820.4	0.358



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SPECIFICATION TABLE NO. 337 THERMAL CONDUCTIVITY OF [IRON + NICKEL +  $\Sigma X_i$ ] ALLOYS GROUP I(X<sub>i</sub> ≤ 0.20% except C ≤ 2.00% and Mn, P, S, Si ≤ 0.60% each)

[ For Data Reported in Figure and Table No. 337 ]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)							Composition (continued), Specifications and Remarks				
							Fe	Ni	C	Cr	Mn	Mo	P		S	Si		
1	157	L	1959	1.6-86		3703	Bal.	5.10	0.11	0.34	0.041	0.04	0.16	Annealed.				
2	131	C	1953	323-1173	2.0	47% Ni-iron	55.8	43.91	0.050	0.22		0.003		Annealed at 950 C.				
3	166	C	1959	273-423		1	Bal.	0.55	0.06	0.022	0.38	0.030	0.017	0.035	0.01	0.001 Al, 0.039 As, 0.08 Cu; annealed at 930 C.		
4	166	C	1939	273-573		9	Bal.	3.47	0.325	0.17	0.55	0.04	0.032	0.034	0.18	0.006 Al, 0.023 As, 0.088 Cu, 0.01 V; annealed at 860 C.		
5	206	L	1920	330			Bal.	0	<0.1							Electrolytic.		
6	206	L	1920	330			Bal.	1.07	<0.1								Electrolytic.	
7	206	L	1920	330			Bal.	1.93	<0.1								Electrolytic.	
8	206	L	1920	330			Bal.	7.05	<0.1								Electrolytic.	
9	206	L	1920	330			Bal.	10.20	<0.1								Electrolytic.	
10	206	L	1920	330			Bal.	13.1	<0.1								Electrolytic.	
11	206	L	1920	330			Bal.	19.2	<0.1								Electrolytic.	
12	206	L	1920	330			Bal.	22.1	<0.1								Electrolytic.	
13	206	L	1920	330			Bal.	25.2	<0.1								Electrolytic.	
14	206	L	1920	330			Bal.	28.4	<0.1								Electrolytic.	
15	206	L	1920	330			Bal.	35.1	<0.1								Electrolytic.	
16	206	L	1920	330			Bal.	47.1	<0.1								Electrolytic.	
17	186	P	1928	305		Climax	Bal.	30.0										
18	561	C	1925	446		Japanese steel	Bal.	3.41	0.45									
19	537	L	1961	125-263		AISI 2515	Bal.	4.91	0.14		0.52				0.33		Specimen about 2.54 cm in dia and about 37 cm long; normalized at 1144.3 K, tempered at 866.5 K; furnished by International Nickel Co., chemical composition and heat treatment history provided by the International Nickel Co.	
20	937	L	1961	183-482		AISI 2515												The above specimen, run 2.
21	937	L	1961	372-573		AISI 2515												The above specimen, run 3.
22	937	L	1961	400-696		AISI 2515												The above specimen, run 4.



SPECIFICATION TABLE NO. 337 (continued)

Curve No.	ReL No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)							Composition (continued), Specifications and Remarks
							C	Cr	Mn	Mo	P	S	Si	
28	937	L	1961	123-813		ALSI 2315	95.483	3.46	0.16	0.54			0.32	Specimen 2.54 cm in dia and 37 cm long; normalized at 1172.5 K and tempered at 866.5 K; specimen furnished by, and chemical composition, heat-treatment history provided by International Nickel Co.; data presented as a smooth curve.
29	937	L	1961	123-813		1% Ni	97.984	1.04	0.126	0.56			0.27	Specimen 2.54 cm in dia and 37 cm long; normalized at 1200 K, tempered at 866.5 K; specimen furnished by, and chemical composition, heat-treatment history provided by International Nickel Co.; data presented as a smooth curve.

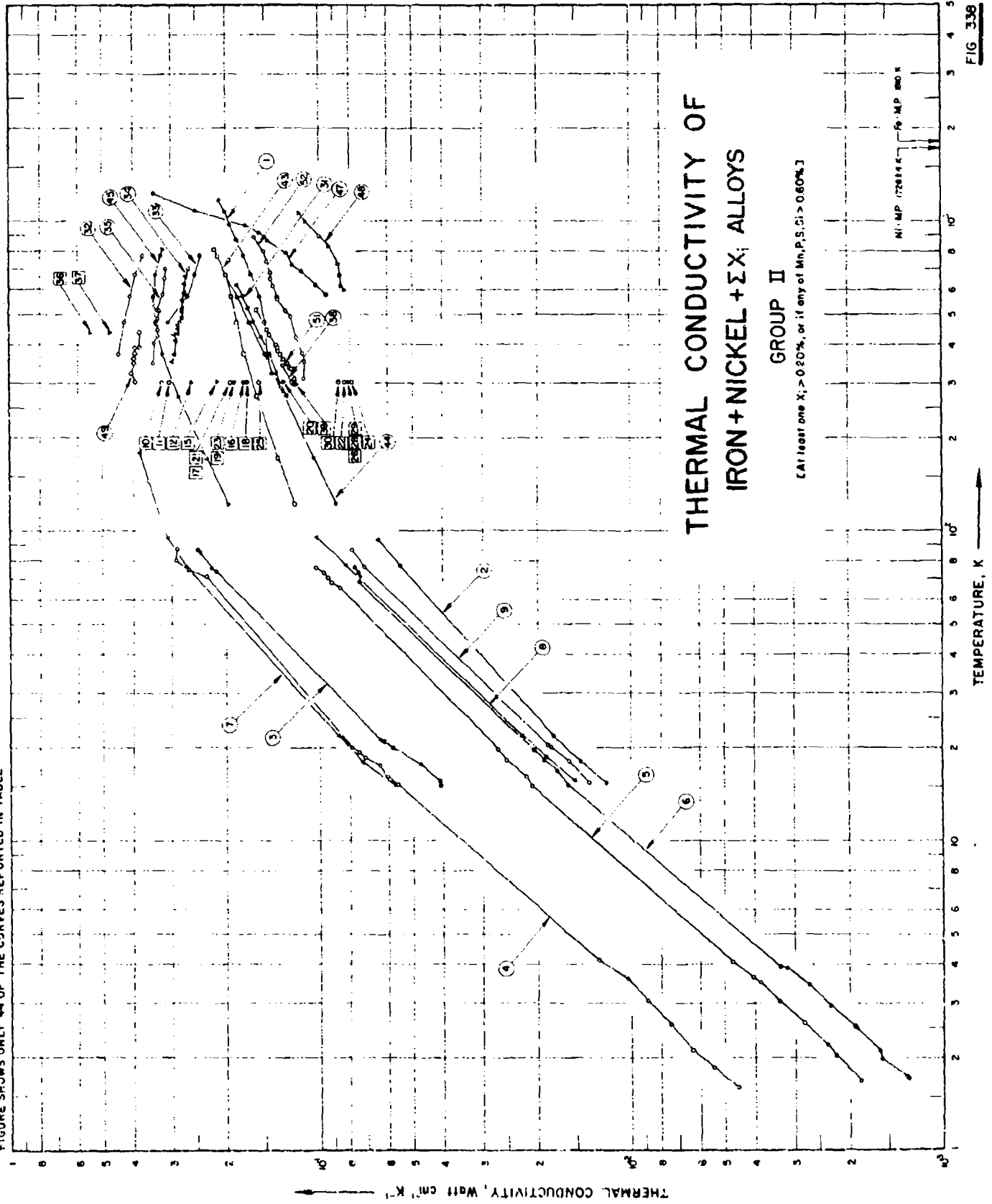
DATA TABLE NO. 337 THERMAL CONDUCTIVITY OF [IRON + NICKEL +  $\Sigma X_i$ ] ALLOYS GROUP I(X<sub>i</sub> ≤ 0.20% except C ≤ 2.00% and Mn, P, S, Si ≤ 0.60% each)[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup> K<sup>-1</sup>]

T	k	T	k	T	k	T	k	T	k	T	k
<u>CURVE 1</u>											
1.66	0.00381	423.20	0.361	330.00	0.110	458.2	0.376*	373.2	0.154	<u>CURVE 23 (cont.)</u>	
2.06	0.00481	473.20	0.389	<u>CURVE 16</u>		518.2	0.378	473.2	0.171	123.2 0.271	
2.62	0.00580	523.20	0.393	330.00	0.154	576.2	0.382*	573.2	0.186	173.2 0.307	
3.07	0.00689	573.20	0.393	<u>CURVE 17</u>		636.2	0.369	673.2	0.201	273.2 0.360	
3.63	0.00843	<u>CURVE 5</u>		330.00	0.136	696.2	0.362	773.2	0.216	373.2 0.388	
4.04	0.01026	330.00	0.598	<u>CURVE 18</u>		<u>CURVE 23</u>		813.2	0.223	473.2 0.397	
15.04	0.0422	<u>CURVE 6</u>		305.00	0.136	423.2	0.379*	<u>CURVE 28</u>		573.2 0.392*	
16.16	0.0459	<u>CURVE 7</u>		<u>CURVE 19</u>		493.2	0.376*	123.2	0.271	673.2 0.378	
18.14	0.0481	<u>CURVE 8</u>		330.00	0.435	573.2	0.376*	173.2	0.307	773.2 0.359	
16.15	0.0525	<u>CURVE 9</u>		445.8	0.527	652.2	0.370	273.2	0.360	813.2 0.351	
13.11	0.0596	<u>CURVE 10</u>		<u>CURVE 20</u>		730.2	0.359	373.2	0.388		
20.60	0.0638	330.00	0.423	125.2	0.242	908.2	0.346	473.2	0.397		
73.76	0.175	<u>CURVE 11</u>		<u>CURVE 21</u>		<u>CURVE 24</u>		573.2	0.392*		
73.80	0.199	<u>CURVE 12</u>		156.2	0.270	100	0.67	773.2	0.378		
80.05	0.195	<u>CURVE 13</u>		186.2	0.293	200	0.70	813.2	0.359		
82.92	0.211	<u>CURVE 14</u>		212.2	0.304	280	0.695	<u>CURVE 25</u>			
87.64	0.227	330.00	0.304	239.2	0.321	<u>CURVE 26</u>		123.2	0.375		
<u>CURVE 2</u>											
323.20	0.159	<u>CURVE 15</u>		263.2	0.324	100	0.497	173.2	0.173		
373.20	0.161	<u>CURVE 16</u>		<u>CURVE 20</u>		200	0.59	273.2	0.408		
473.20	0.165	330.00	0.283	183.2	0.290	280	0.60	373.2	0.452		
573.20	0.169	<u>CURVE 17</u>		253.2	0.325	<u>CURVE 25</u>		473.2	0.468		
673.20	0.171	<u>CURVE 18</u>		313.2	0.351	<u>CURVE 26</u>		573.2	0.441		
773.20	0.174	<u>CURVE 19</u>		373.2	0.363	123.2	0.158	673.2	0.412		
873.20	0.176	<u>CURVE 10</u>		428.2	0.377	173.2	0.173	773.2	0.352		
973.20	0.181	<u>CURVE 11</u>		483.2	0.382	273.2	0.194	813.2	0.371		
1073.20	0.184	<u>CURVE 12</u>		<u>CURVE 21</u>		373.2	0.205	<u>CURVE 27</u>			
1173.20	0.188	<u>CURVE 13</u>		372.2	0.366	475.2	0.210	123.2	0.158		
<u>CURVE 3</u>											
273.20	0.653	<u>CURVE 14</u>		413.2	0.370	573.2	0.211	173.2	0.173		
323.20	0.628	<u>CURVE 15</u>		453.2	0.377	673.2	0.218	273.2	0.408		
373.20	0.603	<u>CURVE 16</u>		49.2	0.382	813.2	0.222	373.2	0.452		
423.20	0.577	<u>CURVE 17</u>		51.2	0.381	<u>CURVE 28</u>		473.2	0.468		
<u>CURVE 4</u>											
273.20	0.364	<u>CURVE 18</u>		51.2	0.381	<u>CURVE 29</u>		573.2	0.441		
323.20	0.372	<u>CURVE 19</u>		51.2	0.381	<u>CURVE 30</u>		673.2	0.412		
373.20	0.377	<u>CURVE 20</u>		400.2	0.371	<u>CURVE 31</u>		773.2	0.352		
<u>CURVE 22</u>											
123.2	0.993	<u>CURVE 21</u>		<u>CURVE 32</u>		<u>CURVE 33</u>		<u>CURVE 34</u>			
173.2	0.109	<u>CURVE 22</u>		<u>CURVE 35</u>		<u>CURVE 36</u>		<u>CURVE 37</u>			
273.2	0.134	<u>CURVE 23</u>		<u>CURVE 36</u>		<u>CURVE 37</u>		<u>CURVE 38</u>			
<u>CURVE 27</u>											
123.2	0.993	<u>CURVE 24</u>		<u>CURVE 37</u>		<u>CURVE 38</u>		<u>CURVE 39</u>			
173.2	0.109	<u>CURVE 25</u>		<u>CURVE 38</u>		<u>CURVE 39</u>		<u>CURVE 40</u>			
273.2	0.134	<u>CURVE 26</u>		<u>CURVE 39</u>		<u>CURVE 40</u>		<u>CURVE 41</u>			
<u>CURVE 28</u>											
123.2	0.993	<u>CURVE 27</u>		<u>CURVE 40</u>		<u>CURVE 41</u>		<u>CURVE 42</u>			
173.2	0.109	<u>CURVE 28</u>		<u>CURVE 41</u>		<u>CURVE 42</u>		<u>CURVE 43</u>			
273.2	0.134	<u>CURVE 29</u>		<u>CURVE 42</u>		<u>CURVE 43</u>		<u>CURVE 44</u>			

\* Not shown on plot



FIGURE SHOWS ONLY 44 OF THE CURVES REPORTED IN TABLE



SPECIFICATION TABLE NO. 338 THERMAL CONDUCTIVITY OF [IRON + NICKEL + EX<sub>1</sub>] ALLOYS GROUP II(At least one X<sub>1</sub> > 0.20% or if any of Mn, P, S, Si > 0.60%)

[For Data Reported in Figure and Table No. 338]

Curve No.	Ref. Method Used	Year	Temp. Range, K	Reported Error, %	Name and Designation of Specimen	Composition (weight per cent)										Composition (continued), Specifications and Remarks			
						Ni	C	Co	Cr	Cu	Mn	P	S	Si					
1	13	C	1953	323-373	2.0	Kovar	28.75	0.017	17.15					0.47					53.7 Fe; annealed at 900 C.
2	104	L	1951	15- 94		1414B	24.30	1.18						6.05					Heated to 1050 C and quenched in water.
3	104	L	1951	15- 87		3975	2.61	0.27	0.49					0.45	0.029	0.011	0.11		0.75 Mo; heated to 950 C and quenched in oil, then reheated to 650 C and quenched in water.
4	157	L	1959	1.6- 66		1287 D	1.92	0.14						0.72			0.21		Annealed.
5	157	L	1959	1.7- 76		1287 I	11.39	0.16						0.93			0.22		Annealed.
6	157	L	1959	1.7- 76		1798 H	19.64	0.43						1.06					Annealed.
7	134	L	1951	15- 100		1287 D	1.92	0.14						0.72			0.21		Heated to 800 C and cooled in furnace.
8	104	L	1951	15- 96		1449 A	31.4	0.70						0.82					Heated to 800 C and cooled in furnace.
9	104	L	1951	15- 87		3450-3	36.17	0.16						0.92	0.09				Heated to 1050 C and quenched in water.
10	178	E	1918	303		2a	4.6	0.10	0.012					0.48	0.31	0.028	0.026	0.11	Approximate composition; annealed at 900 C.
11	178	E	1918	303		2b	4.6	0.10	0.012					0.48	0.31	0.028	0.026	0.11	Approximate composition; cooled once to -190 C in liquid air.
12	178	E	1918	303		3a	9.2	0.11	0.024					0.67	0.32	0.027	0.025	0.11	Approximate composition; annealed at 900 C.
13	178	E	1918	303		3b	9.2	0.11	0.024					0.67	0.32	0.027	0.025	0.11	Approximate composition; cooled once to -190 C in liquid air.
14	178	E	1918	303		4a	13.6	0.12	0.035					0.67	0.32	0.025	0.025	0.12	Approximate composition; annealed at 900 C.
15	178	E	1918	303		4b	12.8	0.12	0.035					0.67	0.32	0.025	0.025	0.12	Approximate composition; cooled once to -190 C in liquid air.
16	178	E	1918	303		5a	18.5	0.13	0.048					1.06	0.32	0.024	0.024	0.12	Approximate composition; annealed at 900 C.
17	178	E	1918	303		5b	18.5	0.13	0.048					1.05	0.32	0.024	0.024	0.12	Approximate composition; cooled once to -190 C in liquid air.
18	178	E	1918	303		6a	21.2	0.135	0.05					1.17	0.32	0.023	0.024	0.12	Approximate composition; annealed at 900 C.
19	178	E	1918	303		6b	21.2	0.135	0.05					1.17	0.32	0.023	0.024	0.12	Approximate composition; cooled once to -190 C in liquid air.
20	178	E	1918	303		7a	23.6	0.14	0.061					1.27	0.32	0.022	0.024	0.12	Approximate composition; annealed at 900 C.
21	178	E	1918	303		7b	23.6	0.14	0.061					1.27	0.32	0.022	0.024	0.12	Approximate composition; cooled once to -190 C in liquid air.
22	178	E	1918	303		9a	27.7	0.15	0.071					1.44	0.32	0.021	0.023	0.12	Approximate composition; annealed at 900 C.
23	178	E	1918	303		9b	27.7	0.15	0.071					1.44	0.32	0.021	0.023	0.12	Approximate composition; cooled once to -190 C in liquid air.
24	178	E	1918	303		10a	29.1	0.15	0.075					1.51	0.32	0.021	0.023	0.12	Approximate composition; annealed at 900 C.

SPECIFICATION TABLE NO. 338 (continued)

Curve No.	Ref. Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight per cent)										SI	S	Composition (continued). Specifications and Remarks
						NI	C	Co	Cr	Cu	Mn	P	S					
25	178	E	1918	303	10b	29.1	0.15	0.075		1.51	0.32	0.021	0.023	0.12		Approximate composition; cooled once to -190 C in liquid air.		
26	178	E	1918	303	11a	30.5	0.155	0.078		1.56	0.32	0.020	0.023	0.12		Approximate composition; annealed at 900 C.		
27	178	E	1918	303	11b	30.5	0.155	0.078		1.56	0.32	0.020	0.023	0.12		Approximate composition; cooled once to -190 C in liquid air.		
28	178	E	1918	303	12a	32.8	0.16	0.084		1.65	0.33	0.019	0.023	0.12		Approximate composition; annealed at 900 C.		
29	178	E	1918	303	12b	32.8	0.16	0.084		1.65	0.33	0.019	0.023	0.12		Approximate composition; cooled once to -190 C in liquid air.		
30	178	E	1918	303	13a	36.9	0.17	0.095		1.83	0.32	0.018	0.022	0.13		Approximate composition; annealed at 900 C.		
31	166	C	1939	273-623	14	28.37	0.28		Trace	0.030	0.89	0.009	0.003	0.15		0.027 As; 0.012 Al; heated to 550 C and cooled in water.		
32	129	C	1933	373-773	32	1.37	0.35		0.46		0.56	0.015	0.02	0.02		Normalized at 900 C.		
33	160	F	1938	473-773	16	1.5	0.30		0.5		0.6			0.35		Annealed.		
34	31	L	1933	352-704	Microsil	18.65	1.81		2.02					6.42				
35	178	L	1939	347-702	NI-Resist Cast Iron	13.70	2.41		3.37	6.41	0.62			1.80		Cast in mold.		
36	561	C	1925	439.4	Crucible steel	2.58	0.22		0.22	0.140	0.60	0.14	0.009	0.23		0.31 W.		
37	561	C	1925	439.2	Crucible steel	2.36	0.37		0.82	0.152	1.04	0.016	0.13	0.37				
38	493	L	1922	299-344		30.40	0.26				0.84			0.14		Forged and worked.		
39	493	L	1922	299.2		30.4	0.26				0.84			0.14		The above specimen annealed for 3 hrs at 690-700 C.		
40	493	L	1922	299.2		30.4	0.26				0.84			0.14		The above specimen cooled from 31 to 29 C in one hr and then cooled to 26 C in one hr.		
41	493	L	1922	299.2		30.4	0.26				0.84			0.14		The above specimen heated to 78 C and cooled to 76 C in 2 hrs and to 70 C in 2 hrs and then cooled very slowly below 70 C in 3 hrs.		
42	493	L	1922	299.2		30.4	0.26				0.84			0.14		The above specimen heated to 185 C and cooled to 165 C in one hour and cooled below 165 C in 2 hrs.		

SPECIFICATION TABLE NO. 338 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)							Composition (continued), Specifications and Remarks
							Ni	C	Co	Cr	Cu	Mn	P	
43	937	L	1961	123-813		Low-exp-42	42.11	0.045	0.09	0.97	0.16	0.16	0.16	56.303 Fe; specimen 2.54 cm in dia and 37 cm long; annealed for 30 min at 1088.7 K, furnace cooled; specimen furnished by, and chemical composition, heat-treatment history supplied by International Nickel Co.; data presented as a smooth curve.
44	937	L	1961	123-813		Free cut Invar	35.84	0.08	0.12	0.81	0.34	0.34	0.34	62.233 Fe; specimen 2.54 cm in dia and 37 cm long; annealed for 30 min at 1102.6 K, water quenched, 1 hr at 588.7 K, air cooled, 48 hrs at 369.3 K, air cooled; specimen furnished by, and chemical composition, heat-treatment history provided by International Nickel Co.; data presented as a smooth curve.
45	937	L	1961	123-813		9% Ni	8.56	0.10		0.77	0.28	0.28	0.28	90.29 Fe; specimen 2.54 cm in dia and 37 cm long; normalized at (1650 + 1450 F) (1172 + 1061 K), tempered at 833.7 K; specimen furnished by, and chemical composition, heat-treatment history provided by International Nickel Co.; data presented as a smooth curve.
46	976	L	1933	373-673		Cast iron	18.65	1.81	2.02		Trace	Trace	0.13	Specimen 4 mm in dia and 100 mm long; prepared by powder metallurgy method (Fe type PZHM of GOST 9849-61 supplied by Sulinski Metallurgical Works, Cu type PM-2 of GOST 4960-49 supplied by Pysmensky Electrolytic Copper Works, Ni type PNE of GOST 9722-61 and C type TKA of GOST 4340-54); sintered at 1413-1433 K for 1.5 hrs in H <sub>2</sub> atmosphere then sintered in cracked ammonia atmosphere; electrical conductivity 0.76, 0.73, 0.68, 0.72, 0.70, 0.60, 0.58, 0.57, 0.55, 0.55, and 0.58 x 10 <sup>4</sup> ohm <sup>-1</sup> cm <sup>-1</sup> at 305, 344, 414, 450, 517, 586, 644, 694, 805, 817, and 957 C, respectively.
47	983, 984	E	1965	578-1230		Alloy No. 3	21.4	0.98		9.9	0.18	Trace	Trace	

SPECIFICATION TABLE NO. 338 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)							Composition (continued), Specifications and Remarks		
							Ni	C	Co	Cr	Cu	Mn	P		S	Si
48	983, 984	E	1965	600-1068	<6	Alloy No. 4	29.8	1.02			9.87	0.15	Trace	Trace	0.14	Similar to the above specimen except electrical conductivity 0.66, 0.64, 0.59, 0.52, 0.47, 0.45, 0.44, and $0.43 \times 10^4$ ohm <sup>-1</sup> cm <sup>-1</sup> at 327, 345, 395, 456, 558, 623, 723 and 795 C, respectively.
49	973	L	1966	304-440	<6	M8	1.65	0.32	0.69		0.76	0.013	0.015	0.40	0.32 Mo; specimen 1.27 cm in dia and 15 cm long; annealed at 900 C, oil quenched from 870 C and tempered at 600 C; cast condition; electrical resistivity 27.1, 30.7, 33.3, and 36.6 $\mu$ ohm cm, at 19, 72, 110, and 160 C, respectively.	
50	973	L	1966	331-453	<6	M8										Similar to the above specimen except in wrought condition and electrical resistivity 25.9, 27.5, 28.5, 32.3, 34.0, and 36.2 $\mu$ ohm cm at 22, 56, 66, 121, 150, and 179, respectively.
51	973	L	1966	309-519	<6	Nimonic PE7	37.2	0.1	<0.2	16.1	0.10			0.22	1.25 Al, 1.19 Ti; specimen 1.27 cm in dia and 15 cm long; fully heat treated; electrical resistivity 107.4, 108.8, 110.9, 114.9, and 116.9 $\mu$ ohm cm at 20, 50, 100, 200, and 250 C, respectively.	
52	985	E	1963	311-887			15.4	1.05		9.63	0.19			0.15	Traces of P and S; specimen 4 mm in dia and 100 mm long; sintered for 1.5 hrs at 1140-1150 C in hydrogen atmosphere; electrical resistivity 1.13, 1.23, 1.35, 1.39, 1.49, 1.51, and 1.67 $\mu$ ohm cm at 37.6, 102.2, 219.1, 301.3, 406.3, 527.4, and 613.5 C, respectively.	



DATA TABLE NO. 318 (continued)

T	k	T	k
<u>CURVE 48 (cont.)</u>			
618.2	0.0854	470.2	0.153*
668.2	0.0862	485.2	0.153*
729.2	0.0870	519.2	0.159
831.2	0.0929	<u>CURVE 52</u>	
896.2	0.0992	310.8	0.1117
996.2	0.110	353.0	0.1113
1068.2	0.116	375.4	0.1136
<u>CURVE 49</u>			
304.2	0.393	422.8	0.1179
322.2	0.403	492.3	0.1243
347.2	0.396	515.1	0.1264
360.2	0.398	563.7	0.1356
374.2	0.394	574.5	0.1360
391.2	0.394	618.7	0.1406
392.2	0.381	652.3	0.1439
440.2	0.381	679.5	0.1439
<u>CURVE 50*</u>			
331.2	0.400	743.5	0.1477
336.2	0.405	800.6	0.1510
336.2	0.394	886.7	0.1623
350.2	0.399	<u>CURVE 51</u>	
396.2	0.390	309.2	0.120
421.2	0.385	334.2	0.123
453.2	0.382	316.2	0.125
<u>CURVE 51</u>			
309.2	0.120	336.2	0.127*
334.2	0.123	348.2	0.127
316.2	0.125	348.2	0.129*
336.2	0.127*	359.2	0.131
348.2	0.127	363.2	0.132*
348.2	0.129*	372.2	0.134
359.2	0.131	379.2	0.134*
363.2	0.132*	383.2	0.136
372.2	0.134	394.2	0.136
379.2	0.134*	399.2	0.137
383.2	0.136	430.2	0.145
394.2	0.136	445.2	0.148
399.2	0.137		
430.2	0.145		
445.2	0.148		

\* Not shown on plot

# THERMAL CONDUCTIVITY OF IRON + NICKEL + CHROMIUM + ΣX; ALLOYS

## GROUP I

[X; ≤ 0.20% except C ≤ 2.00% and Mn, P, S, Si ≤ 0.60% each]

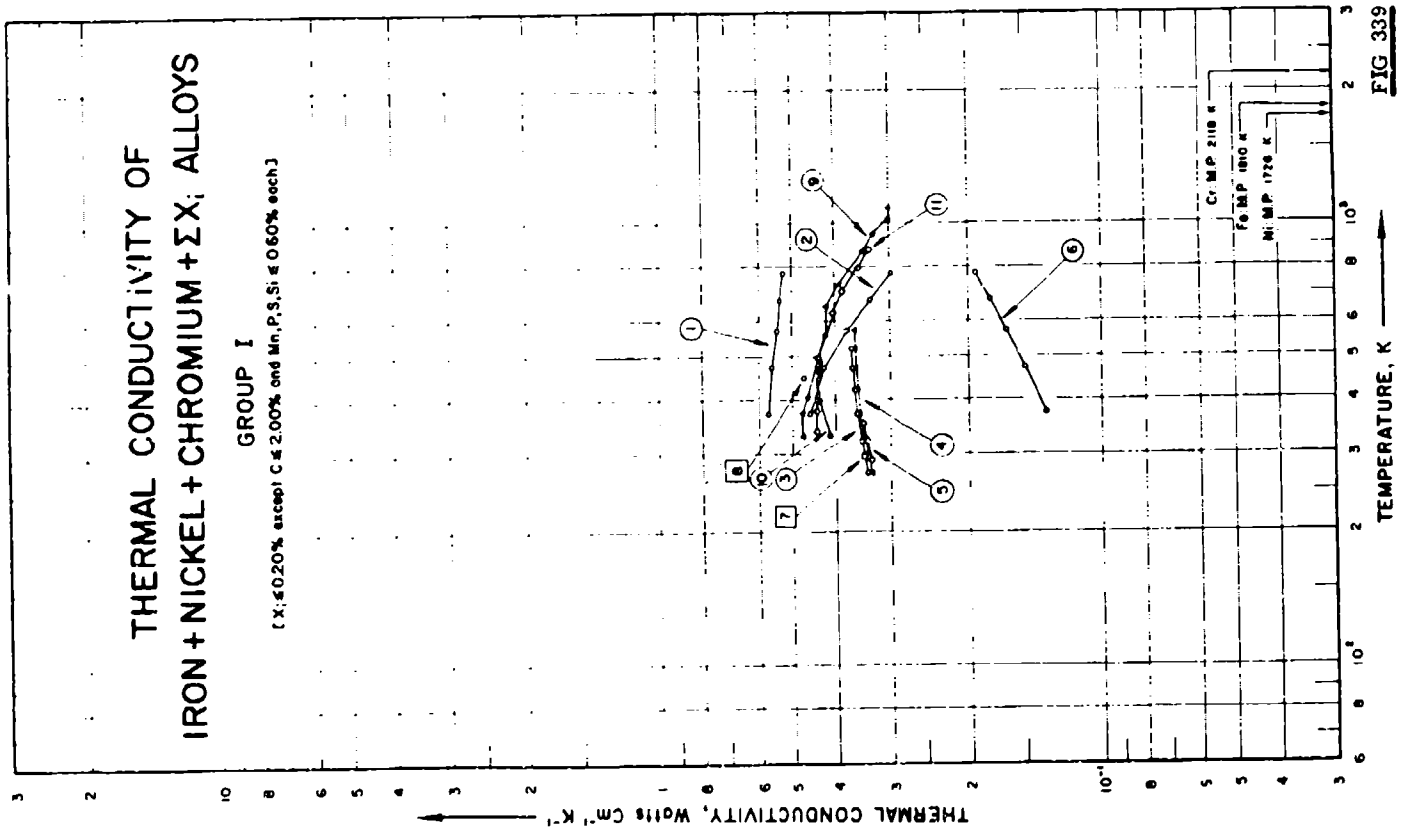


FIG 339



SPECIFICATION TABLE NO. 339 THERMAL CONDUCTIVITY OF [IRON + NICKEL + CHROMIUM +  $\Sigma X_i$ ] ALLOYS GROUP I

$\Sigma X_i \leq 0.20\%$  except C  $\leq 2.00\%$  and Mn, P, S, Si  $\leq 0.60\%$  each

[For Data Reported in Figure and Table No. 339]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)							Composition (continued), Specifications and Remarks			
							Ni	Cr	Al	C	Cu	Mn	Mo		P	Si	
1	160	F	1938	373-773		15	4.5	1.1		0.13		0.4				0.35	Annealed.
2	160	F	1938	373-773		17	4.5	1.3		0.35		0.6				0.35	Annealed.
3	166	C	1939	273-523		10	3.38	0.80	0.006	0.33	0.053	0.53	0.07	0.031	0.17		0.01 V, 0.028 As, 0.033 S; annealed at 860 C; reheated to 640 C and cooled in furnace.
4	166	C	1939	273-573		11	3.41	0.71	0.008	0.325	0.120	0.55	0.06	0.018	0.25		0.01 V, 0.023 As, 0.025 S; annealed at 860 C; reheated to 640 C and cooled in furnace.
5	17	L	1958	293-353	1.0	AMS 2713	1.7	0.7		0.55		0.6	0.2		0.3		0.1 V; annealed and quenched.
6	129	C	1933	373-773	3.0-5.0	Chromel 502	34.0	10.0									Hot rolled.
7	177	C	1936	298.2	10.0	K112N	2.89	1.02		0.27							Normalized.
8	561	C	1925	449.6		Japanese steel	3.17	0.45		0.33							0.012 S; quenched in oil from 850 C, then tempered at 600 C and quenched in water.
9	539	L	1938	330-1073		Ni - Cr steel	2.92	0.72		0.29		0.42					0.012 S; quenched in oil from 850 C, then tempered at 600 C and quenched in water.
10	539	L	1938	334-492			2.92	0.72		0.29		0.42					0.012 S; quenched in oil from 850 C.
11	539	L	1938	340-876			2.92	0.72		0.29		0.42					0.012 S; quenched in oil from 850 C, and tempered at 300 C

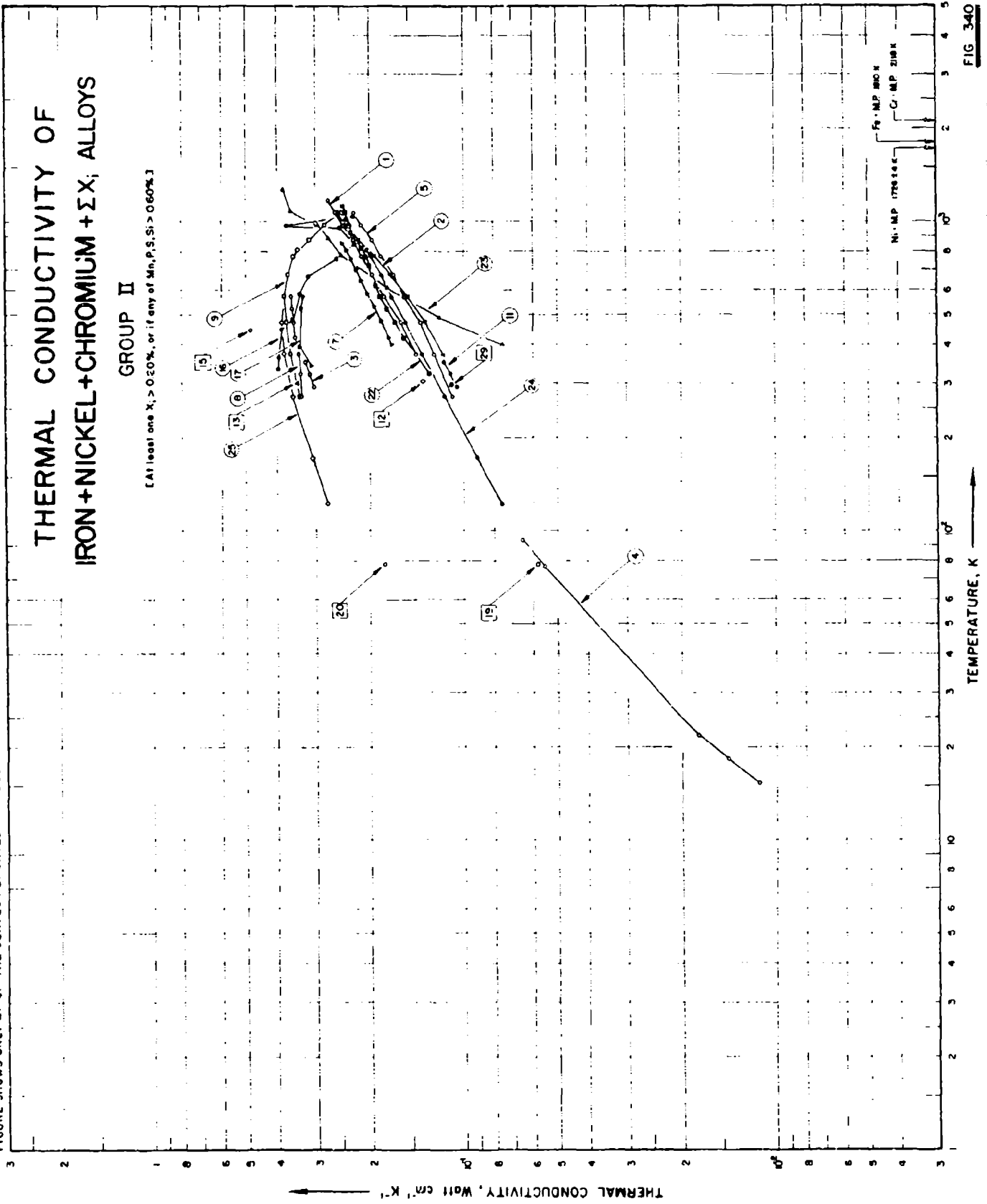


FIGURE SHOWS ONLY 21 OF THE CURVES REPORTED IN TABLE

# THERMAL CONDUCTIVITY OF IRON+NICKEL+CHROMIUM + $\Sigma X_i$ ALLOYS

## GROUP II

[At least one  $X_i > 0.20\%$ , or if any of Mn, P, S, Si  $> 0.60\%$ ]



SPECIFICATION TABLE NO. 340 THERMAL CONDUCTIVITY OF [IRON + NICKEL + CHROMIUM + ΣX<sub>i</sub>] ALLOYS GROUP II

(At least one X<sub>i</sub> > 0.20% or if any of Mn, P, S, Si > 0.80%)

[For Data Reported in Figure and Table No. 340]

Curve No.	Ref. Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight per cent)										SI	Composition (continued), Specifications and Remarks
						Ni	Cr	C	Mn	Mo	P	S	Si	Composition (continued), Specifications and Remarks			
1	181	L	1953	373-1173	Rex 78	18.0	14.0	0.1	0.5	3.5					0.5	3.5 Cu, 0.75 Ti.	
2	181	L	1953	373-1073	Jessop G.17	25.0	13.0	0.4	0.8	2.0					1.5	2.5 W.	
3	17	L	1958	293-353	AMS 2714	1.7	1.0	0.55	0.7	0.5					0.3	0.1 V; annealed and quenched.	
4	104	L	1951	15-93	Era ATV steel; 3731	27.3	14.6	0.44	1.34						1.62	3.5 W; heated to 1000 C and quenched in water.	
5	162	C	1936	273-1073	Era ATV steel	26.86	15.20	0.46	1.18		0.018	0.014	1.30		52.2 Fe; 2.77 W; forged.		
6	181	L	1953	373-1173	C 18 B	13.0	13.0	0.4	0.8	2.0			1.0		10.0 Co; 3.0 Nb; 2.5 W.		
7	37	C	1951	400-851	N-155	20.0	20.0	0.2		3.25					20.0 Co; 2.5 W; 1.1 Nb.		
8	166	C	1939	273-573	12	3.53	0.78	0.34	0.55	0.39	0.024	0.003	0.27		0.050 Co; 0.037 As; 0.007 Al; annealed at 860 C, then reheated to 640 C and cooled in furnace.		
9	162	C	1936	273-1073	F. N. C. T. steel	3.55	0.85	0.39	0.64				0.21		Oil-hardened from 830 C and tempered at 650 C.		
10	185	C, L	1960	273-1073	Macloy C steel	36.5	16.75	0.49	0.58		0.012	0.016	1.9		Heated to 1050 C and cooled in air.		
11	17	L	1958	293-353	Vacromin F	32.31	20.03		1.42	0.10					Trace		
12	186	P	1928	305	Climax	29.0	2.0		1.0						Trace		
13	177	C	1936	298	5 ZA 2 steel	4.11	1.54	0.32	0.37		0.018				0.83 W; normalized.		
14	187		1955	373, 1173	Nimonic DS	≈37	≈18	0.15	1.5	Max					2.5 Max		
15	561	C	1925	448.1	Cruminic steel	2.23	0.88	0.26	0.32		0.013	0.26			0.38 V.		
16	539	L	1938	332-1049	Ni - Cr steel	4.33	1.12	0.31	0.38	0.37	0.021	0.006	0.15		0.27 W; quenched in oil from 850 C and then tempered at 600 C.		
17	539	C	1938	342-574	Ni - Cr steel	4.33	1.12	0.31	0.38	0.37	0.021	0.006	0.15		Quenched in oil from 850 C.		
18	539	C	1938	335-736	Ni - Cr steel	4.33	1.12	0.31	0.58	0.37	0.021	0.006	0.15		The above specimen tempered again at 320 C.		
19	537		1940	78.2	Ni + Cr + W Steel	26.56	14.13	0.42	1.06				1.50		2.21 W, P and S unknown; commercial heat resistant alloy; forged; measured in a boiling nitrogen bath.		
20	537		1940	78.2	Low alloy steel	3.00	0.99	0.28	0.8						Si unknown; commercial heat resistant alloy; oil hardened from 830 C and then tempered at 600 C; measured in a boiling nitrogen bath.		
21	504	P	1961	295.2	AMS 4040 steel	1.65/2.60	0.76/0.90	0.38/0.43	0.6/0.8	0.2/0.3	0.04	0.04	0.2/0.35		Nominal composition from Mark's handbook.		

SPECIFICATION TABLE NO. 340 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)							Composition (continued), Specifications and Remarks
							Ni	Cr	C	Mn	Mo	P	S	
22	666		1963	323-1123	2.0	EI-855	36.55	15.5	0.08	0.46	0.0125	0.047	0.55	2.88 W and 0.31 Ti; quenched in air from 1100 C.
23	664	E	1957	400-1273	±4	Russian steel; EI-257	15	15	0.15	0.7	0.4	0.035	0.9	2.75 W; austenitic; specimen 4 mm in dia and 120 mm long; tempered at 1175 C (cooling medium: water) and aged at 750 C for 10 hrs.
24	937	L	1961	123-813		NI-Span-C	41.95	5.31	0.03	0.42				49.332 Fe, 2.51 Ti, 0.38 Al; specimen 2.54 cm in dia and 37 cm long, after machining at NBS age hardened for 6 hrs in a vacuum furnace at 955.4 K; furnace cooled; specimen, chemical composition, heat-treatment history provided by the International Nickel Co.; data presented as a smooth curve.
25	937	L	1961	123-813		AISI 4340QT	1.87	0.74	0.40	0.68	0.25		0.28	95.747 Fe; specimen 2.54 cm in dia and 37 cm long; normalized at 1144.3 K, oil-quenched at 1088.7 K, tempered at 866.5 K; specimen, chemical composition, heat-treatment history provided by the International Nickel Co.; data presented as a smooth curve.
26	937	L	1961	123-813		AISI 4340NT	1.87	0.74	0.40	0.68	0.25		0.28	Similar to the above specimen except only normalized at 1144.3 K and tempered at 866.5 K.
27	765	C	1957	298.2		Alloy steel	3.4	0.8	0.3					Thermal comparator applied on the machined curved surface of the 1 in. dia bar specimen.
28	765	C	1957	298.2		High alloy steel	36.5	16.75	0.49				1.90	Thermal comparator applied on the machined curved surface of the 1 in. dia bar specimen.
29	765	C	1957	298.2		High alloy steel	36.5	16.75	0.49				1.90	Thermal comparator loaded with 100 gram weight applied on the plane lapped surface of the specimen.



GROUP I

SPECIFICATION TABLE NO. 341 THERMAL CONDUCTIVITY OF [IRON + PHOSPHORUS +  $\Sigma X_i$ ] ALLOYS  
 ( $X_i \leq 0.20\%$  except C  $\leq 2.00\%$  and Mn, P, S, Si  $\leq 0.60\%$  each)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
1	204	L	1937	387		40	21 P; S, Si.

GROUP I

DATA TABLE NO. 341 THERMAL CONDUCTIVITY OF [IRON + PHOSPHORUS +  $\Sigma X_i$ ] ALLOYS  
 ( $X_i \leq 0.20\%$  except C  $\leq 2.00\%$  and Mn, P, S, Si  $\leq 0.60\%$  each)  
 [Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup> K<sup>-1</sup>]

T k  
CURVE 1<sup>o</sup>  
 386.7 0.363

<sup>o</sup> No graphical presentation

# THERMAL CONDUCTIVITY OF IRON + SILICON + X, ALLOYS

## GROUP I

(X, = 0.20% except C ≤ 2.00% and Mn, P, S, Si ≤ 0.60% each)

FIGURE SHOWS ONLY 33 OF THE CURVES REPORTED IN TABLE

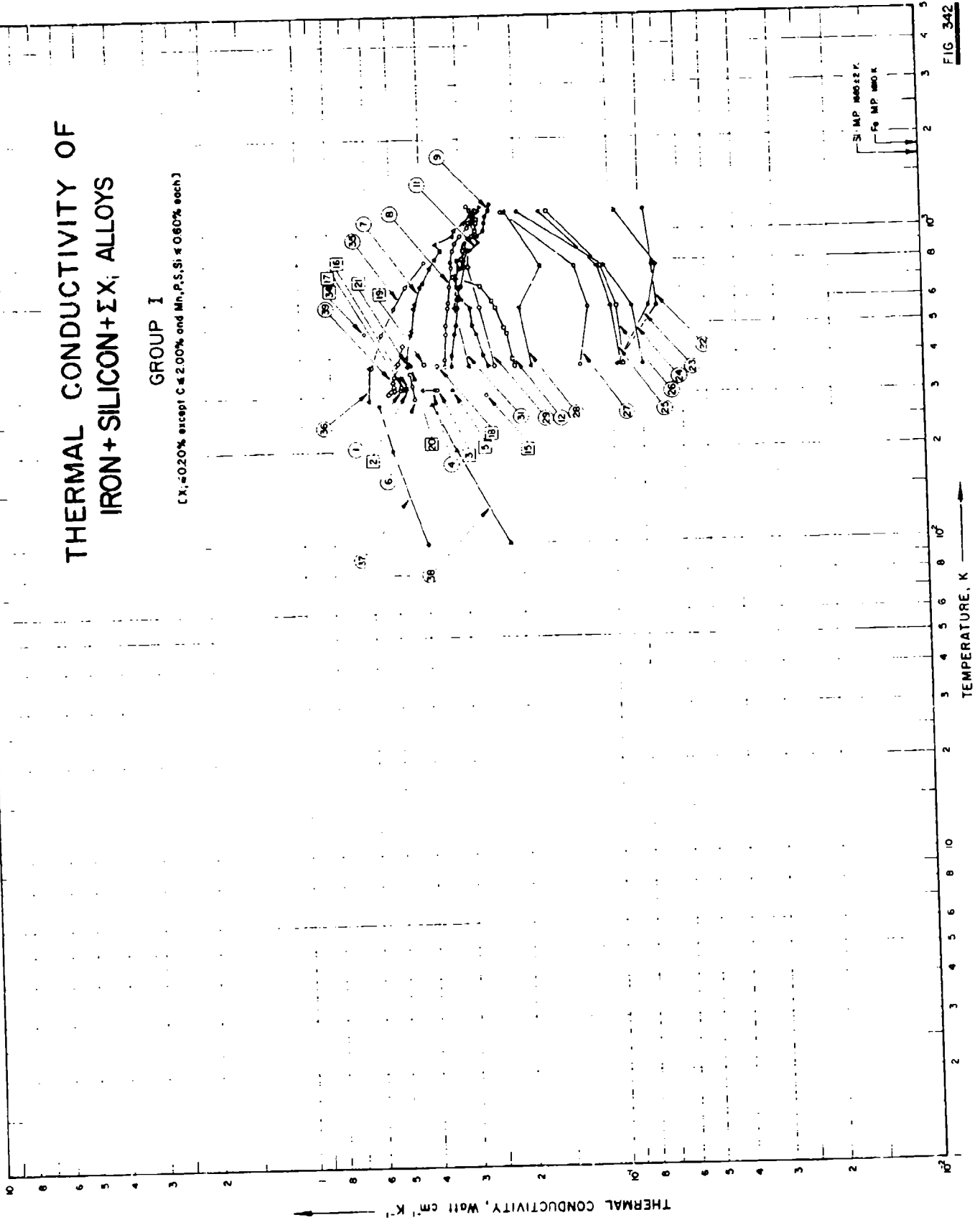


FIG. 342



## SPECIFICATION TABLE NO. J42 THERMAL CONDUCTIVITY OF [IRON + SILICON + SX.] ALLOYS GROUP I

(X<sub>i</sub> ≤ 0.20% except C ≤ 2.00% and Mn, P, S ≤ 0.60% each)

[For Data Reported in Figure and Table No. 342.]

Curve No.	Ref. Method Used	Year	Temp. Range, K	Reported Error, %	Num. and Specimen Designation	Composition (weight percent)							Composition (continued) Specifications and Remarks
						Si	Al	C	Cr	Mn	P	S	
1	170	L	1926	313.2	2.1	0.65		0.45		0.35	0.015	0.02	Annealed.
2	170	L	1926	313.2	2.2	0.65		0.45		0.35	0.015	0.02	Forged.
3	170	L	1926	313.2	2.3h	0.65		0.45		0.35	0.015	0.02	Annealed and then quenched from 800 C.
4	170	L	1926	313.2	3.2	0.86		0.55		0.41	0.014	0.02	Forged.
5	170	L	1926	313.2	3.3h	0.86		0.55		0.44	0.014	0.02	Annealed and then quenched from 800 C.
6	17	L	1958	293-353	1.0	0.5		0.37	0.1	0.5			
7	210	R	1956	373-1173	4.0	1.90		0.07		0.25	0.024	0.026	Unfinished.
8	210	R	1956	373-1173	4.0	1.23	0.01	0.09		0.29	0.047	0.029	0.35 mm foil.
9	210	R	1956	373-1235	4.0	1.80	0.01	0.09		0.32	0.038	0.023	0.35 mm foil.
10	210	R	1956	373-1203	4.0	2.20							The rest not determined, billet specimen.
11	210	R	1956	373-1203	4.0	2.78	0.06	0.09		0.35	0.034	0.023	Billet specimen.
12	210	R	1956	373-1213	4.0	3.94	0.09	0.08		0.27	0.027	0.008	0.35 mm foil.
13	210	R	1956	373-1183	4.0	4.28	0.05	0.06		0.08	0.012	0.006	Billet specimen.
14	210	R	1956	373-1183	4.0	4.38	0.05	0.07		0.20	0.015	0.008	Unfinished.
15	203	L	1957	300	±1.5	2.4		<0.1		0.39			(5.0 pearlite); annealed at 900 C for 12 hrs and furnace-cooled.
16	204	L	1937	319.5		47.2							Heat-flow parallel to thickness.
17	204	L	1937	320.7		44.0							Heat-flow parallel to thickness.
18	204	L	1937	374.7		16.0							
19	204	L	1937	379.5		10.01							
20	204	L	1937	376.6		10.30							
21	204	L	1937	382.4		12.92							
22	356	R	1956	373-1173	±7.0	45.0							
23	356	R	1956	373-1173	±7.0	38.0							
24	356	R	1956	373-1173	±7.0	32.0							
25	356	R	1956	373-1173	±7.0	29.0							
26	356	R	1956	373-1173	±7.0	25.0							
27	356	R	1956	373-1173	±7.0	17.0							
28	356	R	1956	373-1173	±7.0	10.0							
29	356	R	1956	373-1173	±7.0	4.4							

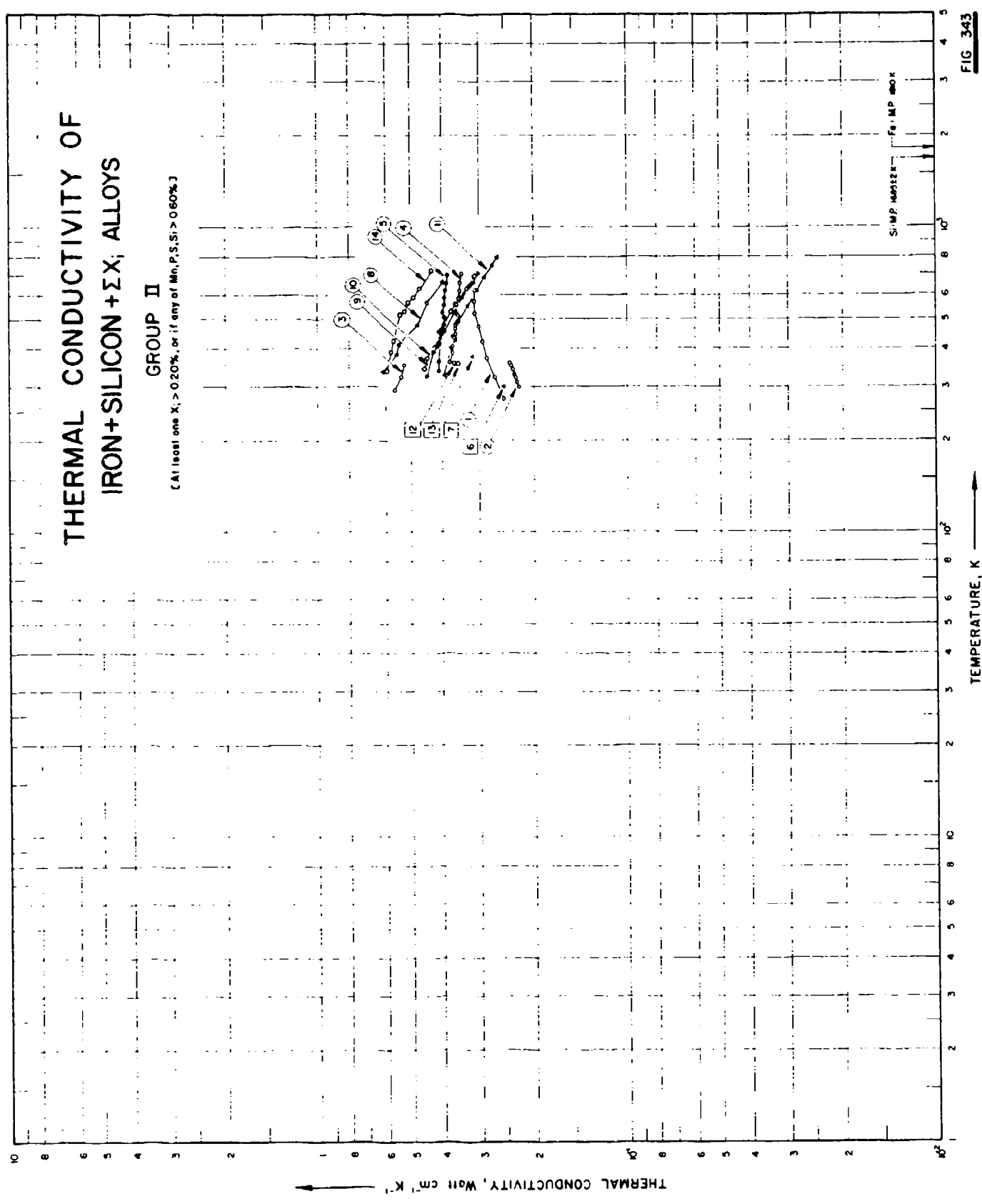




# THERMAL CONDUCTIVITY OF IRON+SILICON+ΣX; ALLOYS

## GROUP II

(At least one X; > 0.20%, or if any of Mn, P, S; > 0.60%)



TEMPERATURE, K

SPECIFICATION TABLE NO. 343 THERMAL CONDUCTIVITY OF [IRON + SILICON +  $\Sigma X_i$ ] ALLOYS GROUP II(At least one  $X_i > 0.20\%$  or if any of Mn, P, S  $> 0.60\%$ )

[For Data Reported in Figure and Table No. 343]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Si	C	Cr	Cu	Mn	Ni	P	S	Composition (continued), Specifications and Remarks
1	166	C	1939	273-623		21	1.08	6.485	0.04	0.637	0.90	0.156	0.044	0.047	0.007 Al, 0.029 As; annealed at 930 C.
2	17	<del>PL</del>	1958	299-360	1.0	Cast iron Nr 1510	3.53	2.16	0.05	0.04	0.3	±1.0			0.06 Mg, 0.05 V.
3	17	<del>PL</del>	1958	293-353	1.0	Cast iron Nr 1520	2.99	2.66	0.05	0.04	0.2	±0.3			
4	31	L	1933	360-703	2.0	Sisal	6.49	2.75							
5	31	L	1933	339-700	2.0	Cast iron; 5	4.20	3.62			0.28		0.30	0.043	(Trace pearlite); annealed at 900 C for 12 hrs and furnace-cooled.
6	203	L	1957	300	± 1.5	2165	2.8	0.1			0.37	1.0			7.0 Zr.
7	204	L	1937	377		Russian alloy; # 35									38.4 Fe, 0.59 O <sub>2</sub> (contained in 0.65 FeO and 1.5 Fe <sub>2</sub> O <sub>3</sub> ); 5.1% porosity; sintered 1 hr and 30 min at 1150 C and annealed 30 min at 825 C.
8	211	E	1953	336-660	± 3.0		0.60	0.06							
9	211	E	1953	325-703	± 3.0		0.60	0.18			0.35				98.11 Fe, 0.76 O <sub>2</sub> (contained in 0.79 FeO and 1.8 Fe <sub>2</sub> O <sub>3</sub> ); 11.4% porosity; sintered 1 hr and 30 min at 1150 C and annealed 30 min at 825 C.
10	211	E	1953	343-693	± 3.0		0.70	0.097			0.35				98.183 Fe, 0.67 O <sub>2</sub> (contained in 0.4 FeO and 1.9 Fe <sub>2</sub> O <sub>3</sub> ); 14.3% porosity; sintered 1 hr and 30 min at 1150 C and annealed 30 min at 825 C.
11	211	E	1953	328-795	± 3.0		0.59	0.087			0.35				98.263 Fe, 0.71 O <sub>2</sub> (contained in 0.6 FeO and 1.9 Fe <sub>2</sub> O <sub>3</sub> ); 17% porosity; sintered 1 hr and 30 min at 1150 C and annealed 30 min at 825 C.
12	205	C	1953	358		Nodular Iron	3.53	2.47			0.29	1.3	0.03	0.012	0.06 Mg; cast.
13	205	C	1953	358		Nodular Iron	4.34	3.36			0.4	1.23	0.03	0.01	0.06 Mg; cast.
14	560	E	1953	340-723	± 3.0		0.4	0.058			0.30				0.56 O; sintered at 1150 C and kept in this condition for 1 hr 30 min.

SPECIFICATION TABLE NO. 343 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)					Composition (continued), Specifications and Remarks		
							Si	C	Cr	Cu	Mn		Ni	P
15	975	L	1933	373-673		Cast iron	4.24	3.02			0.29	0.30	0.043	
16	976	L	1933	373-673		Sial iron	6.49	2.75						

DATA TABLE NO. 343 THERMAL CONDUCTIVITY OF [IRON + SILICON +  $\Sigma X_i$ ] ALLOYS GROUP II(At least one  $X_i > 0.20\%$  or if any of Mn, P, S  $> 0.60\%$ )[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup> K<sup>-1</sup>]

T	k	T	k	T	k	T	k	T	k
<u>CURVE 1</u>		<u>CURVE 5</u>		<u>CURVE 10</u>		<u>CURVE 15 (cont.)*</u>			
273.20	0.251	339.20	0.410	342.60	0.452	573.2	0.353		
323.20	0.269	363.70	0.410	371.00	0.444	673.2	0.365		
373.20	0.285	415.70	0.406	455.10	0.402	<u>CURVE 16*</u>			
423.20	0.293	471.20	0.410	529.00	0.372	373.2	0.372		
473.20	0.301	431.20	0.402	557.10	0.360	473.2	0.364		
523.20	0.310	449.70	0.402	631.20	0.326	573.2	0.351		
573.20	0.312	453.20	0.402	692.50	0.310	673.2	0.343		
623.20	0.310	465.20	0.397	<u>CURVE 11</u>					
<u>CURVE 2</u>		480.20	0.383	327.90	0.393				
299.20	0.224	507.20	0.391	401.70	0.372				
310.60	0.227	527.50	0.397	429.50	0.360				
321.70	0.229	557.70	0.393	492.40	0.351				
326.20	0.231	620.20	0.389	550.90	0.326				
327.70	0.231	653.70	0.385	623.10	0.305				
340.90	0.235	699.70	0.381	687.30	0.289				
351.00	0.238	<u>CURVE 6</u>		752.30	0.272				
360.20	0.240	300.00	0.251	795.10	0.264				
<u>CURVE 3</u>		<u>CURVE 7</u>		<u>CURVE 12</u>					
293.20	0.561	377.20	0.317	359.00	0.362				
323.20	0.508	<u>CURVE 8</u>		<u>CURVE 13</u>					
353.20	0.526	335.70	0.615	358.00	0.351				
<u>CURVE 4</u>		384.60	0.552	<u>CURVE 14</u>					
360.20	0.377	411.10	0.544	339.5	0.598				
386.70	0.368	478.80	0.477	389.0	0.577				
407.70	0.368	567.40	0.444	425.8	0.565				
434.20	0.364	659.70	0.397	516.6	0.540				
438.70	0.368	<u>CURVE 9</u>		530.1	0.523				
446.20	0.360	325.20	0.449	566.7	0.506				
462.20	0.360	388.70	0.427	590.7	0.490				
476.70	0.360	403.50	0.414	630.5	0.469				
483.20	0.356	452.00	0.393	723.2	0.431				
512.20	0.351	521.30	0.364	<u>CURVE 15*</u>					
520.20	0.356	567.50	0.343	373.2	0.406				
557.70	0.356	641.00	0.322	473.2	0.397				
583.70	0.351	658.60	0.314						
618.20	0.347	703.20	0.301						
656.70	0.347								
702.70	0.343								

\* Not shown on plot

GROUP I

SPECIFICATION TABLE NO. 344 THERMAL CONDUCTIVITY OF [IRON + TITANIUM + ΣX<sub>i</sub>] ALLOYS

(X<sub>i</sub> ≤ 0.20% except C ≤ 2.00% and Mn, P, S, Si ≤ 0.60% each)

Curve No.	Rel. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
1	204	L	1937	393		Ferrotitanium, 37	19.70 Ti and 0.09 C.

GROUP I

DATA TABLE NO. 344 THERMAL CONDUCTIVITY OF [IRON + TITANIUM + ΣX<sub>i</sub>] ALLOYS

(X<sub>i</sub> ≤ 0.20% except C ≤ 2.00% and Mn, P, S, Si ≤ 0.60% each)

[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup> K<sup>-1</sup>]

T k  
 CURVE 1\*  
 393.2 0.628

\* No graphical presentation



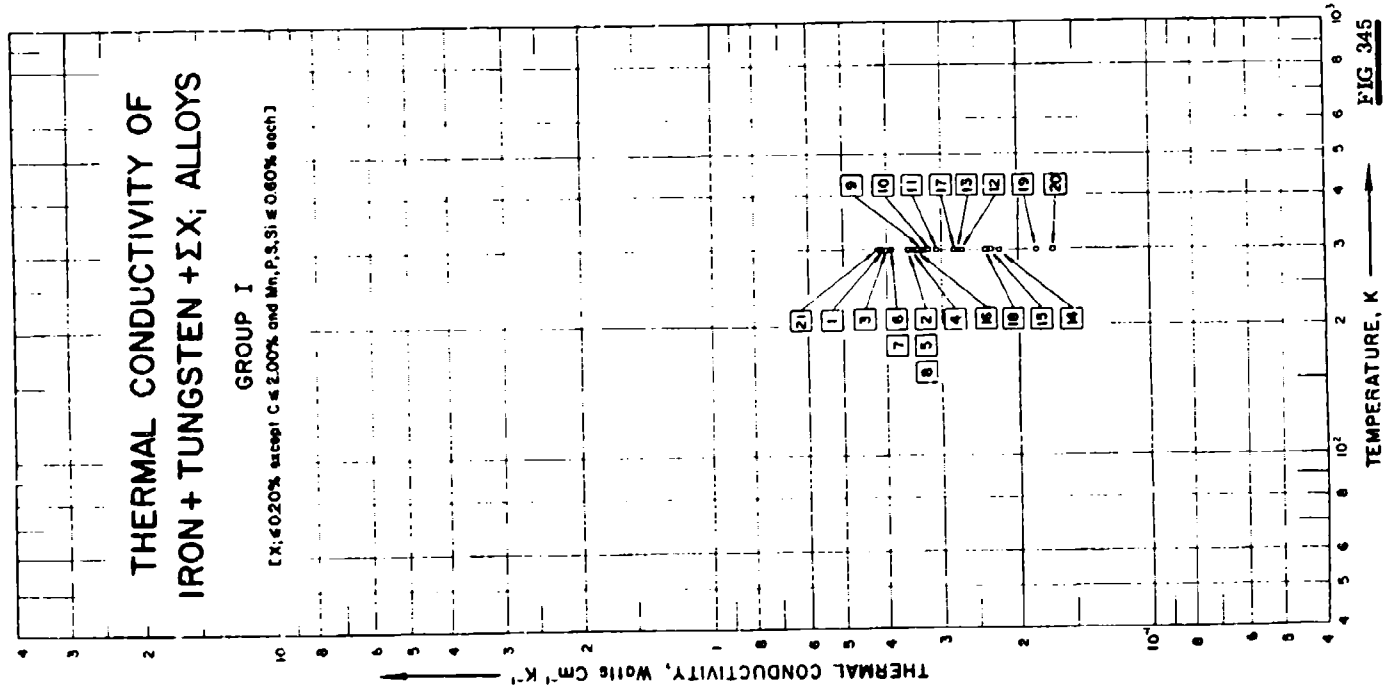


FIG. 345

SPECIFICATION TABLE NO. 345 THERMAL CONDUCTIVITY OF [IRON + TUNGSTEN +  $\Sigma X_i$ ] ALLOYS GROUP I(X<sub>i</sub> < 0.20% except C = 2.00% and Mn, P, S, Si = 0.60% each)

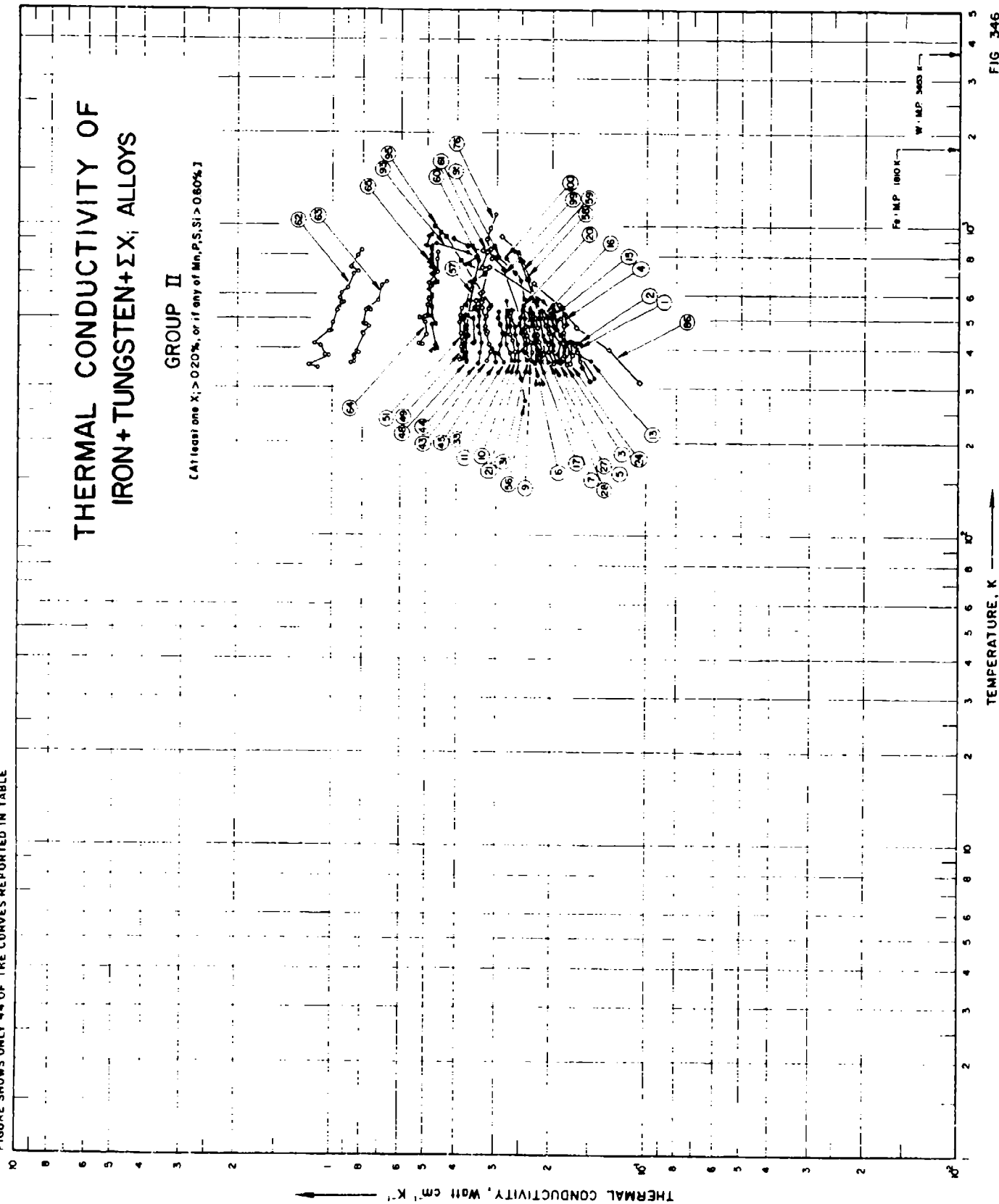
[For Data Reported in Figure and Table No. 345]

Curve No.	Ref. Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight per cent)					Composition (continued), Specifications and Remarks
						W	C	Mn	P	S	
1	188	E	1919	303	3a	1.0	0.3				Annealed at 900 C and slowly cooled.
2	188	E	1919	303	3b	1.0	0.6				Annealed at 900 C and slowly cooled.
3	188	E	1919	303	4a	2.0	0.3				Annealed at 900 C and slowly cooled.
4	188	E	1919	303	4b	2.0	0.6				Annealed at 900 C and slowly cooled.
5	188	E	1919	303	5b	3.0	0.6				Annealed at 900 C and slowly cooled.
6	188	E	1919	303	6a	5.0	0.3				Annealed at 900 C and slowly cooled.
7	188	E	1919	303	7a	6.0	0.3				Annealed at 900 C and slowly cooled.
8	188	E	1919	303	7b	6.0	0.6				Annealed at 900 C and slowly cooled.
9	188	E	1919	303	8b	10.0	0.6				Annealed at 900 C and slowly cooled.
10	188	E	1919	303	9a	15.0	0.3				Annealed at 900 C and slowly cooled.
11	188	E	1919	303	9b	15.0	0.6				Annealed at 900 C and slowly cooled.
12	188	E	1919	303	10a	20.0	0.3				Annealed at 900 C and slowly cooled.
13	188	E	1919	303	10b	20.0	0.6				Annealed at 900 C and slowly cooled.
14	188	E	1919	303	11a	25.0	0.3				Annealed at 900 C and slowly cooled.
15	188	E	1919	303	11b	25.0	0.6				Annealed at 900 C and slowly cooled.
16	188	E	1919	303	12	1.0	0.6				Annealed at 1100 C and quickly cooled.
17	188	E	1919	303	13	3.0	0.6				Annealed at 1100 C and quickly cooled.
18	188	E	1919	303	14	6.0	0.6				Annealed at 1100 C and quickly cooled.
19	188	E	1919	303	15	20.0	0.6				Annealed at 1100 C and quickly cooled.
20	188	E	1919	303	16	25.0	0.6				Annealed at 1100 C and quickly cooled.
21	188	E	1919	303	2a	0.5	0.3				Annealed at 500 C and slowly cooled.

DATA TABLE NO. 345 THERMAL CONDUCTIVITY OF [IRON + TUNGSTEN +  $\Sigma X_i$ ] ALLOYS GROUP I(X<sub>i</sub> = 0.20% except C = 2.00% and Mn, P, S, Si = 0.60% each)[Temperature, T, K; Thermal Conductivity, k, Watts cm<sup>-1</sup>K<sup>-1</sup>]

T	k	T	k
<u>CURVE 1</u>	<u>CURVE 13</u>		
303.20	0.411	303.20	0.276
<u>CURVE 2</u>	<u>CURVE 14</u>		
303.20	0.360	303.20	0.221
<u>CURVE 3</u>	<u>CURVE 15</u>		
303.20	0.402	303.20	0.231
<u>CURVE 4</u>	<u>CURVE 16</u>		
303.20	0.349	303.20	0.343
<u>CURVE 5</u>	<u>CURVE 17</u>		
303.20	0.356	303.20	0.280
<u>CURVE 6</u>	<u>CURVE 18</u>		
303.20	0.590	303.20	0.238
<u>CURVE 7</u>	<u>CURVE 19</u>		
303.20	0.387	303.20	0.182
<u>CURVE 8</u>	<u>CURVE 20</u>		
303.20	0.356	303.20	0.167
<u>CURVE 9</u>	<u>CURVE 21</u>		
303.20	0.333	303.20	0.419
<u>CURVE 10</u>			
303.20	0.324		
<u>CURVE 11</u>			
303.20	0.309		
<u>CURVE 12</u>			
303.20	0.270		

FIGURE SHOWS ONLY 44 OF THE CURVES REPORTED IN TABLE



## SPECIFICATION TABLE NO. 346 THERMAL CONDUCTIVITY OF IRON + TUNGSTEN + EX. ALLOYS GROUP II

(At least one  $X_i > 0.20\%$  or if any of Mn, P, S, Si  $> 0.60\%$ )

(For Data Reported in Figure and Table No. 346)

Curve No.	Ref. Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight per cent)							Composition (continued), Specifications and Remarks		
						W	C	Cr	Cu	Mn	Mo	P		S	Si
1	189	L	366-411	2.0-5.0	High speed steel, C <sub>q</sub>	18.65	0.734	4.07	0.066	0.242	0.730	Trace	0.037	0.272	4.90 Co, trace Ti, 1.93 V; oil-quenched from 1300 C.
2	189	L	368-420	2.0-5.0	High speed steel, C <sub>T</sub> 200 C	18.65	0.734	4.07	0.066	0.242	0.730	Trace	0.037	0.272	4.90 Co, trace Ti, 1.93 V; tempered at 200 C for 30 min and cooled in air.
3	189	L	368-503	2.0-5.0	High speed steel, C <sub>T</sub> 300 C	18.65	0.734	4.07	0.066	0.242	0.730	Trace	0.037	0.272	4.90 Co, trace Ti, 1.93 V; tempered at 300 C for 30 min and cooled in air.
4	189	L	367-526	2.0-5.0	High speed steel, C <sub>T</sub> 400 C	18.65	0.734	4.07	0.066	0.242	0.730	Trace	0.037	0.272	4.90 Co, trace Ti, 1.93 V; tempered at 400 C for 30 min and cooled in air.
5	189	L	363-527	2.0-5.0	High speed steel, C <sub>T</sub> 500 C	18.65	0.734	4.07	0.066	0.242	0.730	Trace	0.037	0.272	4.90 Co, trace Ti, 1.93 V; tempered at 500 C for 30 min and cooled in air.
6	189	L	362-535	2.0-5.0	High speed steel, C <sub>T</sub> 550 C	18.65	0.734	4.07	0.066	0.242	0.730	Trace	0.037	0.272	4.90 Co, trace Ti, 1.93 V; tempered at 550 C for 30 min and cooled in air.
7	189	L	369-544	2.0-5.0	High speed steel, C <sub>T</sub> 600 C	18.65	0.734	4.07	0.066	0.242	0.730	Trace	0.037	0.272	4.90 Co, trace Ti, 1.93 V; tempered at 600 C for 30 min and cooled in air.
8	189	L	365-533	2.0-5.0	High speed steel, C <sub>T</sub> 650 C	18.65	0.734	4.07	0.066	0.242	0.730	Trace	0.037	0.272	4.90 Co, trace Ti, 1.93 V; tempered at 650 C for 30 min and cooled in air.
9	189	L	373-538	2.0-5.0	High speed steel, C <sub>T</sub> 700 C	18.65	0.734	4.07	0.066	0.242	0.730	Trace	0.037	0.272	4.90 Co, trace Ti, 1.93 V; tempered at 700 C for 30 min and cooled in air.
10	189	L	368-533	2.0-5.0	High speed steel, C <sub>T</sub> 800 C	18.65	0.734	4.07	0.066	0.242	0.730	Trace	0.037	0.272	4.90 Co, trace Ti, 1.93 V; tempered at 800 C for 30 min and cooled in air.
11	189	L	368-537	2.0-5.0	High speed steel, C <sub>A</sub>	18.65	0.734	4.07	0.066	0.242	0.730	Trace	0.037	0.272	4.90 Co, trace Ti, 1.93 V; annealed at 830 C.
12	189	L	370-419	2.0-5.0	High speed steel, E <sub>q</sub>	19.22	0.674	3.45	0.072	0.165	0.275	Trace	0.010	0.535	0.736 Co, trace Ti, 0.848 V; oil-quenched from 1300 C.
13	189	L	368-416	2.0-5.0	High speed steel, E <sub>T</sub> 200 C	19.22	0.674	3.45	0.072	0.165	0.275	Trace	0.010	0.535	0.736 Co, trace Ti, 0.848 V; tempered at 200 C for 30 min and cooled in air.
14	189	L	365-499	2.0-5.0	High speed steel, E <sub>T</sub> 300 C	19.22	0.674	3.45	0.072	0.165	0.275	Trace	0.010	0.535	0.736 Co, trace Ti, 0.848 V; tempered at 300 C for 30 min and cooled in air.
15	189	L	366-540	2.0-5.0	High speed steel, E <sub>T</sub> 400 C	19.22	0.674	3.45	0.072	0.165	0.275	Trace	0.010	0.535	0.736 Co, trace Ti, 0.848 V; tempered at 400 C for 30 min and cooled in air.
16	189	L	371-535	2.0-5.0	High speed steel, E <sub>T</sub> 500 C	19.22	0.674	3.45	0.072	0.165	0.275	Trace	0.010	0.535	0.736 Co, trace Ti, 0.848 V; tempered at 500 C for 30 min and cooled in air.

SPECIFICATION TABLE NO. 346 (continued)

Curve No.	Ref. Method Used	Year	Temp. Rang., K	Reported Error, %	Name and Specimen Designation	Composition (weight per cent)							Composition (continued), Specifications and Remarks			
						W	C	Cr	Cu	Mn	Mo	P		S	Si	
17	189	L	1934	366-526	2.0-5.0	High speed steel; E.T., 550 C	19.22	0.674	3.45	0.072	0.165	0.275	Trace	0.010	0.535	0.736 Co, trace Ti, 0.848 V; tempered at 550 C for 30 min and cooled in air.
18	189	L	1934	359-533	2.0-5.0	High speed steel; L.T., 600 C	19.22	0.674	3.45	0.072	0.165	0.275	Trace	0.010	0.535	0.736 Co, trace Ti, 0.848 V; tempered at 600 C for 30 min and cooled in air.
19	189	L	1934	368-531	2.0-5.0	High speed steel; E.T., 650 C	19.22	0.674	3.45	0.072	0.165	0.275	Trace	0.010	0.535	0.736 Co, trace Ti, 0.848 V; tempered at 650 C for 30 min and cooled in air.
20	189	L	1934	365-542	2.0-5.0	High speed steel; E.T., 700 C	19.22	0.674	3.45	0.072	0.165	0.275	Trace	0.010	0.535	0.736 Co, trace Ti, 0.848 V; tempered at 700 C for 30 min and cooled in air.
21	189	L	1934	369-535	2.0-5.0	High speed steel; E.T., 800 C	19.22	0.674	3.45	0.072	0.165	0.275	Trace	0.010	0.535	0.736 Co, trace Ti, 0.848 V; tempered at 800 C for 30 min and cooled in air.
22	189	L	1934	370-547	2.0-5.0	High speed steel; E.A.	19.22	0.674	3.45	0.072	0.165	0.275	Trace	0.010	0.535	0.736 Co, trace Ti, 0.848 V; annealed at 830 C.
23	189	L	1934	359-425	2.0-5.0	High speed steel; H <sub>Q</sub>	15.53	0.605	3.29	Trace	Trace	Trace	0.028	0.073	0.075	1.13 V; quenched from 1300 C.
24	189	L	1934	361-425	2.0-5.0	High speed steel; H.T., 200 C	15.53	0.605	3.29	Trace	Trace	Trace	0.028	0.073	0.075	1.13 V; tempered at 200 C for 30 min and cooled in air.
25	189	L	1934	368-499	2.0-5.0	High speed steel; H.T., 300 C	15.53	0.605	3.29	Trace	Trace	Trace	0.028	0.073	0.075	1.13 V; tempered at 300 C for 30 min and cooled in air.
26	189	L	1934	365-526	2.0-5.0	High speed steel; H.T., 400 C	15.53	0.605	3.29	Trace	Trace	Trace	0.028	0.073	0.075	1.13 V; tempered at 400 C for 30 min and cooled in air.
27	189	L	1934	360-535	2.0-5.0	High speed steel; H.T., 500 C	15.53	0.605	3.29	Trace	Trace	Trace	0.028	0.073	0.075	1.13 V; tempered at 500 C for 30 min and cooled in air.
28	189	L	1934	365-535	2.0-5.0	High speed steel; H.T., 550 C	15.53	0.605	3.29	Trace	Trace	Trace	0.028	0.073	0.075	1.13 V; tempered at 550 C for 30 min and cooled in air.
29	189	L	1934	370-537	2.0-5.0	High speed steel; H.T., 600 C	15.53	0.605	3.29	Trace	Trace	Trace	0.028	0.073	0.075	1.13 V; tempered at 600 C for 30 min and cooled in air.
30	189	L	1934	369-534	2.0-5.0	High speed steel; H.T., 650 C	15.53	0.605	3.29	Trace	Trace	Trace	0.028	0.073	0.075	1.13 V; tempered at 650 C for 30 min and cooled in air.
31	189	L	1934	357-534	2.0-5.0	High speed steel; H.T., 700 C	15.53	0.605	3.29	Trace	Trace	Trace	0.028	0.073	0.075	1.13 V; tempered at 700 C for 30 min and cooled in air.
32	189	L	1934	366-538	2.0-5.0	High speed steel; H.T., 800 C	15.53	0.605	3.29	Trace	Trace	Trace	0.028	0.073	0.075	1.13 V; tempered at 800 C for 30 min and cooled in air.

SPECIFICATION TABLE NO. 346 (continued)

Curve No.	Ref. Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	W	C	Cr	Cu	Mn	Mo	P	S	Si	Composition (continued), Specifications and Remarks
33	189	L	363-530	2.0-5.0	High speed steel, H <sub>A</sub>	15.53	0.605	3.29	Trace	Trace	Trace	0.028	0.073	0.075	1.13 V; annealed at 830 C.
34	189	L	360-413	2.0-5.0	High speed steel, I <sub>Q</sub>	15.05	0.705	3.24	0.047	0.169	Trace	Trace	0.026	0.166	Trace Ti, 0.864 V; oil-queached from 1300 C.
35	189	L	357-417	2.0-5.0	High speed steel, I <sub>T, 200 C</sub>	15.05	0.705	3.24	0.047	0.169	Trace	Trace	0.026	0.166	Trace Ti, 0.864 V; tempered at 200 C for 30 min and cooled in air.
36	189	L	364-496	2.0-5.0	High speed steel, I <sub>T, 300 C</sub>	15.05	0.705	3.24	0.047	0.169	Trace	Trace	0.026	0.166	Trace Ti, 0.864 V; tempered at 300 C for 30 min and cooled in air.
37	189	L	366-477	2.0-5.0	High speed steel, I <sub>T, 400 C</sub>	15.05	0.705	3.24	0.047	0.169	Trace	Trace	0.026	0.166	Trace Ti, 0.864 V; tempered at 400 C for 30 min and cooled in air.
38	189	L	366-529	2.0-5.0	High speed steel, I <sub>T, 500 C</sub>	15.05	0.705	3.24	0.047	0.169	Trace	Trace	0.026	0.166	Trace Ti, 0.864 V; tempered at 500 C for 30 min and cooled in air.
39	189	L	368-532	2.0-5.0	High speed steel, I <sub>T, 550 C</sub>	15.05	0.705	3.24	0.047	0.169	Trace	Trace	0.026	0.166	Trace Ti, 0.864 V; tempered at 550 C for 30 min and cooled in air.
40	189	L	367-530	2.0-5.0	High speed steel, I <sub>T, 600 C</sub>	15.05	0.705	3.24	0.047	0.169	Trace	Trace	0.026	0.166	Trace Ti, 0.864 V; tempered at 600 C for 30 min and cooled in air.
41	189	L	362-531	2.0-5.0	High speed steel, I <sub>T, 650 C</sub>	15.05	0.705	3.24	0.047	0.169	Trace	Trace	0.026	0.166	Trace Ti, 0.864 V; tempered at 650 C for 30 min and cooled in air.
42	189	L	364-537	2.0-5.0	High speed steel, I <sub>T, 700 C</sub>	15.05	0.705	3.24	0.047	0.169	Trace	Trace	0.026	0.166	Trace Ti, 0.864 V; tempered at 700 C for 30 min and cooled in air.
43	189	L	369-543	2.0-5.0	High speed steel, I <sub>T, 800 C</sub>	15.05	0.705	3.24	0.047	0.169	Trace	Trace	0.026	0.166	Trace Ti, 0.864 V; tempered at 800 C for 30 min and cooled in air.
44	189	L	358-533	2.0-5.0	High speed steel, I <sub>A</sub>	15.05	0.705	3.24	0.047	0.169	Trace	Trace	0.026	0.166	Trace Ti, 0.864 V; annealed at 830 C.
45	189	L	363-389	2.0-5.0	T <sub>Q</sub>	1.02	1.03	0.102		0.997	0.033	0.042	0.215	0.110	0.110 Ni; water-quenched from 840 C.
46	189	L	362-391	2.0-5.0	T <sub>T, 150 C</sub>	1.02	1.03	0.102		0.997	0.033	0.042	0.215	0.110	0.110 Ni; tempered at 150 C for 30 min and cooled in air.
47	189	L	377-430	2.0-5.0	T <sub>T, 200 C</sub>	1.02	1.03	0.102		0.997	0.033	0.042	0.215	0.110	0.110 Ni; tempered at 200 C for 30 min and cooled in air.
48	189	L	365-468	2.0-5.0	T <sub>T, 250 C</sub>	1.02	1.03	0.102		0.997	0.033	0.042	0.215	0.110	0.110 Ni; tempered at 250 C for 30 min and cooled in air.

SPECIFICATION TABLE NO. 346 (continued)

Curve No.	Ref. Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight per cent)										Composition (continued), Specifications and Remarks	
						W	C	Cr	Cu	Ni	Mn	P	S	Si			
49	189	L	1934	370-493	2.0-5.0	T. 300 C	1.02	1.03	0.102		0.997	0.033	0.042	0.215	0.110	Ni; tempered at 300 C for 30 min and cooled in air.	
50	189	L	1934	368-544	2.0-5.0	T. 350 C	1.02	1.03	0.102		0.997	0.033	0.042	0.215	0.110	Ni; tempered at 350 C for 30 min and cooled in air.	
51	189	L	1934	368-546	2.0-5.0	T. 400 C	1.02	1.03	0.102		0.997	0.033	0.042	0.215	0.110	Ni; tempered at 400 C for 30 min and cooled in air.	
52	189	L	1934	374-548	2.0-5.0	T. 500 C	1.02	1.03	0.102		0.997	0.033	0.042	0.215	0.110	Ni; tempered at 500 C for 30 min and cooled in air.	
53	189	L	1934	377-543	2.0-5.0	T. 600 C	1.02	1.03	0.102		0.997	0.033	0.042	0.215	0.110	Ni; tempered at 600 C for 30 min and cooled in air.	
54	189	L	1934	372-547	2.0-5.0	T. 700 C	1.02	1.03	0.102		0.997	0.033	0.042	0.215	0.110	Ni; tempered at 700 C for 30 min and cooled in air.	
55	189	L	1934	371-545	2.0-5.0	T <sub>A</sub>	1.02	1.03	0.102		0.997	0.033	0.042	0.215	0.110	Ni; annealed at 780 C.	
56	166	C	1939	273-573		High speed steel; 18	18.45	0.715	4.26	0.064	0.25	Trace	0.018	0.028	0.30	0.067	Ni, 0.004 Al, 0.035 as, 1.075 V; annealed at 830 C.
57	129	C	1933	373-773	3.0-5.0	S <sub>4</sub>	1.04	0.35	0.61		0.75	0.035	0.028	0.22	0.17	Ni; normalized at 900 C.	
58	212	L	1941	423-823		High speed steel; IM-1	5.0/6.0	0.83	1.0		3.5/5.5						1.40 - 1.75 V.
59	212	L	1941	423-823		High speed steel; 18-4-1	16.00	0.70	4.0		0.30				0.25	1.0 V; annealed.	
60	340	L	1956	420-796	7.0	High speed steel; T-1	18.0		4.0								1.0 V; annealed.
61	340	L	1956	454-841	7.0	High speed steel; M-2	6.0		4.0								2.0 V; annealed.
62	340	L	1956	348-833	7.0	Tool material; CA-4											Composition unknown; cast iron grade; containing tungsten carbide and a small amount of cobalt binder.
63	340	L	1956	350-660	7.0	Tool material; K-6											Same as the above specimen.
64	340	L	1956	415-819	7.0	Tool material; CA-2											Composition unknown; steel cutting grade; containing tungsten carbide, titanium carbide and tantalum carbide, and an increased amount of cobalt than the above two specimens, CA-4 and K-6.
65	340	L	1956	395-935	7.0	Tool material; K-2a											Same as the above specimen.



SPECIFICATION TABLE NO. 346 (continued)

Curve No.	Ref. Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight per cent)							Composition (continued), Specifications and Remarks			
						W	C	Cr	Cu	Mn	Mo	P		S	Si	
66	327	1943	673.2		IV a	3.14	0.28	1.67			1.55				Cast, forged, annealed at 700 C for 2 hrs, quenched in oil at 900 up to 1200 C in steps of 50 C, and then annealed again from 400 C to 650 C in 30 min.	
67	327	1943	673.2		IV b	4.90	0.29	1.50			1.57				Same as the above specimen.	
68	327	1943	673.2		V a	3.00	0.26	1.45							0.62 V; preparation same as the above specimen.	
69	327	1943	673.2		V b	5.02	0.27	1.50							0.60 V; preparation same as the above specimen.	
70	327	1943	673.2		VII a	2.85	0.28	1.51			0.90				0.54 V; preparation same as the above specimen.	
71	327	1943	673.2		VII b	3.11	0.28	1.52			1.94				0.54 V; preparation same as the above specimen.	
72	327	1943	673.2		VII c	3.09	0.26	1.41			2.90				0.54 V; preparation same as the above specimen.	
73	327	1943	673.2		VIII a	5.02	0.25	1.42			1.04				0.57 V; preparation same as the above specimen.	
74	327	1943	673.2		VIII b	4.73	0.25	1.43			1.96				0.54 V; preparation same as the above specimen.	
75	327	1943	673.2		VIII c	5.14	0.27	1.35			2.89				0.41 V; preparation same as the above specimen.	
76	539	1938	356-1092		HS steel	18.52	0.70	4.40			0.32	0.51	0.027	0.004	0.11	0.84 V, 0.06 Co; annealed.
77	539	1938	338-823			18.52	0.70	4.40			0.32	0.51	0.027	0.004	0.11	0.84 V, 0.06 Co; the above specimen heated to 1320 C and quenched in oil.
78	539	1938	333-811			18.52	0.70	4.40			0.32	0.51	0.027	0.004	0.11	0.84 V, 0.06 Co; the above specimen tempered at 550 C for 40 min and cooled in furnace.
79	539	1938	340-804			18.52	0.70	4.40			0.32	0.51	0.027	0.004	0.11	0.84 V, 0.06 Co; the above specimen tempered again at 550 C for 40 min.
80	539	1938	342-1099			18.52	0.70	4.40			0.32	0.51	0.027	0.004	0.11	0.84 V, 0.06 Co; the above specimen tempered again at 550 C for 40 min.
81	539	1938	330-1099		High speed steel	19.31	0.77	4.35			0.23	0.91	0.026	0.003	0.14	2.03 V, 5.58 Co; annealed.
82	539	1938	338-590			19.31	0.77	4.35			0.23	0.91	0.026	0.003	0.14	2.03 V, 5.58 Co; the above specimen heated to 1330 C and quenched in oil.
83	539	1938	333-800			19.31	0.77	4.35			0.23	0.91	0.026	0.003	0.14	2.03 V, 5.58 Co; the above specimen tempered at 550 C for 40 min and cooled in furnace.
84	539	1938	328-790			19.31	0.77	4.35			0.23	0.91	0.026	0.003	0.14	2.03 V, 5.58 Co; the above specimen tempered again at 550 C for 40 min.
85	539	1938	351-1068			19.31	0.77	4.35			0.23	0.91	0.026	0.003	0.14	2.03 V, 5.58 Co; the above specimen tempered again at 550 C for 40 min.

SPECIFICATION TABLE NO. 346 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	W	C	Cr	Cu	Mn	P	S	Si	Composition (weight percent) Specifications and Remarks
86	986, 987	F	1963	312-920	< 3	R15									Quenched and tempered; measured during heating; electrical conductivity 1.09, 1.04, 1.00, 0.951, 0.902, 0.852, 0.829, 0.810, 0.799, and $0.776 \times 10^{-4} \text{ ohm}^{-1} \text{ cm}^{-1}$ at 37, 126, 200, 282, 381, 445, 500, 558, 594, and 654 C, respectively.
87	986, 987	E	1963	347-920	< 3	R15									The above specimen measured during cooling; electrical resistivity 1.63, 1.47, 1.36, 1.31, 1.23, 1.10, 0.971, 0.923, 0.877, 0.826, and $0.776 \times 10^{-4} \text{ ohm}^{-1} \text{ cm}^{-1}$ at 72, 153, 204, 235, 281, 373, 457, 500, 552, 600, and 654 C, respectively.
88	986, 987	E	1963	368-916	< 3	R15Kh3	15.06	0.81	2.80						1.55 V; quenched at 1250 C; tempered three times at 560 C; measured during heating; electrical resistivity 1.41, 1.39, 1.30, 1.11, 1.08, and $0.973 \times 10^{-4} \text{ ohm}^{-1} \text{ cm}^{-1}$ at 47, 134, 234, 429, 546, and 652 C, respectively.
89	986, 987	E	1963	398-916	< 3	R15Kh3									The above specimen measured during cooling; electrical resistivity 1.90, 1.50, 1.32, 1.11, 1.05, and $0.973 \times 10^{-4} \text{ ohm}^{-1} \text{ cm}^{-1}$ at 154, 325, 436, 556, 597, and 652 C, respectively.
90	986, 987	E	1963	399-671	< 3	R15Kh3									Same composition and heat treatment as the above specimen.
91	986, 987	E	1963	316-909	< 3	R15Kh3K5	15.13	0.85	2.96						5.05 Co, 1.40 V; quenched at 1250 C, tempered at 530 C.
92	986, 987	E	1962	305-783	< 3	R15Kh3K5									Similar to the above specimen.
93	986, 987	E	1963	323-877	< 3	R15Kh3K10	15.16	0.78	2.85						9.50 Co, 1.45 V; quenched at 1250 C, tempered at 580 C.
94	986, 987	E	1963	230-995	< 3	R15Kh3K12	14.90	0.77	3.3						11.74 Co, 1.59 V; quenched at 1250 C, tempered at 580 C; measured during heating.

SPECIFICATION TABLE NO. 346 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	W	C	Cr	Cu	Mn	Mo	P	S	Si	Composition (continued), Specifications and Remarks
95	986, 987	E	1963	417-995	<3	R15Kb3K12										The above specimen measured during cooling.
96	986, 987	E	1963	348-840	<3	R15Kb3K12										Same composition and heat treatment as the above specimen.
97	986, 987	E	1963	361-974	<3	R15Kb4	14.76	0.83	4.13							1.59 V; quenched at 1250 C, tempered three times at 560 C; electrical resistivity 1.93, 1.80, 1.66, 1.57, 1.40, 1.29, 1.18, 1.10, 1.01, and 0.972 x 10 <sup>-4</sup> ohm <sup>-1</sup> cm <sup>-1</sup> at 38, 91, 153, 200, 306, 381, 450, 515, 569, and 649 C, respectively.
98	986, 987	E	1963	314-753	<3	R7	7.94	0.79	4.10							1.69 V; quenched at 1200 C, tempered three times at 560 C; electrical resistivity 2.14, 1.82, 1.68, 1.23, and 1.22 x 10 <sup>-4</sup> ohm <sup>-1</sup> cm <sup>-1</sup> at 39, 153, 215, 481, and 508 C, respectively.
99	986, 987	E	1963	309-794	<3	R10	9.52	0.81	4.28							1.40 V; quenched at 1220 C, tempered three times at 560 C; electrical resistivity 2.08, 1.82, 1.57, 1.39, 1.22, and 1.23 x 10 <sup>-4</sup> ohm <sup>-1</sup> cm <sup>-1</sup> at 46, 141, 249, 337, 457, and 523 C, respectively.
100	986, 987	E	1963	308-790	<3	R12	12.25	0.84	4.13							1.55 V; quenched at 1240 C, tempered three times at 560 C; electrical resistivity 2.13, 1.71, 1.47, and 1.15 x 10 <sup>-4</sup> ohm <sup>-1</sup> cm <sup>-1</sup> at 29, 214, 318, and 525 C, respectively.
101	986, 987	E	1963	311-668	<3	R18	18.80	0.75	3.98							1.09 V; quenched at 1280 C, tempered three times at 560 C; electrical resistivity 1.85, 1.73, 1.44, and 1.21 x 10 <sup>-4</sup> ohm <sup>-1</sup> cm <sup>-1</sup> at 43, 97, 243, and 400 C, respectively.

DATA TABLE NO. 34E THERMAL CONDUCTIVITY OF (IRON + TUNGSTEN + EX<sub>1</sub>) ALLOYS GROUP II

(At least one X = 0% or if any of Mn, P, S, Si > 0.60%)

[Temperature, T, K; Thermal Conductivity, k, Watts cm<sup>-1</sup>K<sup>-1</sup>]

CURVE 1		CURVE 6 (cont.)		CURVE 11		CURVE 16 (cont.)		CURVE 21		CURVE 26 (cont.)		CURVE 31		CURVE 36 (cont.)	
T	k	T	k	T	k	T	k	T	k	T	k	T	k	T	k
366.20	0.163	484.20	0.230	368.20	0.268	453.20	0.201	368.70	0.255	443.70	0.197	357.20	0.239	414.20	0.205
390.70	0.157	502.70	0.230	390.20	0.272	480.20	0.205	393.20	0.259	466.20	0.201	398.70	0.247	446.20	0.201
411.20	0.167	534.70	0.234	417.20	0.280	513.70	0.205	416.20	0.259	497.20	0.197	420.20	0.243	474.20	0.205
CURVE 2		CURVE 7		CURVE 12*		CURVE 17		CURVE 22*		CURVE 27		CURVE 32*		CURVE 37*	
368.20	0.163	365.20	0.222	370.20	0.163	441.70	0.222	370.20	0.264	474.70	0.209	366.20	0.278	453.70	0.209
395.20	0.167	393.70	0.218	400.70	0.163	471.20	0.218	391.20	0.272	474.70	0.209	391.20	0.272	476.70	0.209
419.70	0.163	420.70	0.226	416.70	0.167	506.70	0.226	427.20	0.276	506.20	0.213	424.70	0.276	496.20	0.213
CURVE 3		CURVE 8*		CURVE 13		CURVE 18*		CURVE 23*		CURVE 28*		CURVE 33		CURVE 38	
338.20	0.180	365.20	0.222	367.70	0.159	359.20	0.218	365.20	0.272	500.70	0.272	471.20	0.280	366.20	0.205
384.20	0.184	382.70	0.226	382.20	0.163	422.70	0.218	390.20	0.276	546.70	0.285	471.20	0.280	391.20	0.213
416.70	0.180	425.70	0.230	418.20	0.167	423.20	0.218	423.20	0.276	546.70	0.285	504.70	0.272	391.20	0.213
455.20	0.184	454.20	0.234	454.20	0.184	446.70	0.218	446.70	0.276	546.70	0.285	538.20	0.272	418.20	0.218
503.20	0.188	484.20	0.226	484.20	0.184	473.70	0.226	473.70	0.276	534.70	0.234	538.20	0.272	442.20	0.213
CURVE 4		CURVE 9		CURVE 14*		CURVE 19*		CURVE 24		CURVE 29*		CURVE 34*		CURVE 39*	
366.70	0.184	372.70	0.234	365.20	0.184	365.20	0.226	361.20	0.176	370.20	0.226	370.20	0.226	368.20	0.234
391.20	0.176	391.70	0.230	393.70	0.188	393.70	0.226	385.70	0.172	385.70	0.230	398.70	0.230	387.20	0.239
423.20	0.180	423.20	0.234	416.20	0.180	416.20	0.226	425.20	0.176	425.20	0.234	398.70	0.230	420.70	0.243
446.20	0.188	444.20	0.238	444.20	0.184	444.20	0.226	438.70	0.172	438.70	0.230	417.20	0.226	420.70	0.243
484.20	0.180	481.20	0.232	481.20	0.188	481.20	0.226	481.20	0.176	481.20	0.230	417.20	0.226	442.20	0.243
503.70	0.188	503.70	0.236	499.20	0.184	499.20	0.226	495.70	0.176	515.20	0.234	413.20	0.188	479.70	0.247
525.70	0.184	525.70	0.240	525.70	0.188	525.70	0.226	532.70	0.176	534.70	0.234	413.20	0.188	508.20	0.243
CURVE 5		CURVE 10		CURVE 15		CURVE 20		CURVE 25*		CURVE 30*		CURVE 35*		CURVE 40*	
363.20	0.188	372.70	0.234	368.20	0.188	368.20	0.230	361.20	0.188	370.20	0.230	370.20	0.230	368.20	0.234
387.70	0.193	391.70	0.230	408.70	0.184	408.70	0.226	385.70	0.172	398.70	0.230	398.70	0.230	387.20	0.239
421.20	0.188	421.20	0.234	435.20	0.184	435.20	0.226	425.20	0.176	417.20	0.226	417.20	0.226	420.70	0.243
447.70	0.197	447.70	0.238	464.70	0.183	464.70	0.226	445.20	0.234	445.20	0.234	417.20	0.226	442.20	0.243
475.70	0.197	475.70	0.242	497.20	0.193	497.20	0.226	473.20	0.230	478.20	0.234	360.20	0.184	442.20	0.243
504.20	0.201	504.20	0.246	539.70	0.197	539.70	0.226	509.20	0.234	509.20	0.230	394.20	0.180	479.70	0.247
526.70	0.201	526.70	0.250	539.70	0.197	539.70	0.226	530.70	0.188	537.20	0.239	413.20	0.188	508.20	0.243
CURVE 6		CURVE 11		CURVE 16		CURVE 21		CURVE 26*		CURVE 31		CURVE 36*		CURVE 41*	
362.20	0.226	368.20	0.264	365.20	0.230	365.20	0.230	368.70	0.197	368.70	0.239	368.70	0.243	366.70	0.243
401.20	0.222	394.20	0.268	389.20	0.239	389.20	0.239	404.70	0.234	404.70	0.239	330.20	0.239	330.20	0.239
430.20	0.222	428.70	0.272	446.70	0.272	446.70	0.239	419.20	0.239	419.20	0.239	423.70	0.239	423.70	0.239
449.20	0.226	446.70	0.276	467.70	0.247	467.70	0.247	452.70	0.239	452.70	0.239	446.20	0.243	446.20	0.243
		512.70	0.272	507.70	0.243	507.70	0.243	472.20	0.234	472.20	0.234	477.20	0.247	477.20	0.247
		532.70	0.276	511.70	0.243	511.70	0.243	500.70	0.243	500.70	0.243	496.70	0.243	496.70	0.243
		449.20	0.226	449.20	0.247	449.20	0.247	423.20	0.193	423.20	0.193	394.70	0.197	394.70	0.197

\* Not shown on plot



DATA TABLE NO. 346 (continued)

T	k	T	k	T	k	T	k	T	k	T	k
<u>CURVE 79*</u>		<u>CURVE 84*</u>		<u>CURVE 88 (cont.)*</u>		<u>CURVE 93 (cont.)</u>		<u>CURVE 98 (cont.)</u>			
340.2	0.241	326.2	0.228	506.2	0.191	814.2	0.364	504.2	0.315		
379.2	0.244	377.2	0.238	697.2	0.254	877.2	0.466	687.2	0.343		
482.2	0.272	509.2	0.277	813.2	0.289	<u>CURVE 94*</u>		753.2	0.376		
572.2	0.290	580.2	0.274	916.2	0.341	330.2	0.134	<u>CURVE 99</u>			
665.2	0.275	670.2	0.276	<u>CURVE 89*</u>		348.2	0.151	309.2	0.213		
762.2	0.265	730.2	0.264	388.2	0.197	368.2	0.159	523.2	0.25**		
804.2	0.272	790.2	0.255	453.2	0.213	458.2	0.177	723.2	0.286		
<u>CURVE 80*</u>		<u>CURVE 85*</u>		598.2	0.248	523.2	0.196	794.2	0.308		
342.2	0.234	351.2	0.229	708.2	0.300	752.2	0.298	<u>CURVE 100</u>			
380.2	0.269	401.2	0.252	825.2	0.308	828.2	0.356	308.2	0.223		
596.2	0.292	489.2	0.272	868.2	0.341	895.2	0.410	481.2	0.236*		
660.2	0.286	576.2	0.275	916.2	0.341	950.2	0.451	582.2	0.244		
803.2	0.293	690.2	0.285	<u>CURVE 90*</u>		995.2	0.472	669.2	0.251		
865.2	0.276	783.2	0.267	399.2	0.211	<u>CURVE 95</u>		790.2	0.280		
975.2	0.258	818.2	0.273	473.2	0.231	417.2	0.275*	<u>CURVE 101*</u>			
1099.2	0.254	928.2	0.260	564.2	0.255	552.2	0.314	311.2	0.193		
<u>CURVE 81*</u>		<u>CURVE 86</u>		671.2	0.277	723.2	0.434	365.2	0.201		
330.2	0.318	312.2	0.106	<u>CURVE 91</u>		950.2	0.451	511.2	0.213		
381.2	0.312	372.2	0.131	316.2	0.151	995.2	0.472	608.2	0.244		
518.2	0.313	399.2	0.167	349.2	0.159	<u>CURVE 96*</u>					
858.2	0.312	473.2	0.167	421.2	0.172	348.2	0.210				
698.2	0.303	555.2	0.189	704.2	0.261	435.2	0.232				
798.2	0.294	652.2	0.226	793.2	0.297	569.2	0.300				
979.2	0.294	772.2	0.246*	857.2	0.364	840.2	0.369				
1008.2	0.292	824.2	0.258*	909.2	0.410	<u>CURVE 97*</u>					
1088.2	0.277	920.2	0.288	<u>CURVE 92*</u>		<u>CURVE 97*</u>					
<u>CURVE 82*</u>		<u>CURVE 87*</u>		<u>CURVE 93</u>		<u>CURVE 98*</u>					
338.2	0.176	347.2	0.105	305.2	0.214	361.2	0.206				
394.2	0.218	422.2	0.167	386.2	0.251	473.2	0.217				
502.2	0.235	483.2	0.173	446.2	0.261	573.2	0.249				
590.2	0.251	554.2	0.197	545.2	0.289	721.2	0.282				
<u>CURVE 83*</u>		<u>CURVE 88*</u>		673.2	0.312	833.2	0.324				
333.2	0.218	772.2	0.262	783.2	0.354	974.2	0.367				
403.2	0.269	950.2	0.271	<u>CURVE 94</u>		<u>CURVE 98*</u>					
497.2	0.269	920.2	0.288	323.2	0.146	314.2	0.214				
660.2	0.268	<u>CURVE 88*</u>		378.2	0.177*	410.2	0.266				
735.2	0.259	368.2	0.148	569.2	0.213	423.2	0.269				
800.2	0.256	408.2	0.179	720.2	0.286*	482.2	0.276				

\* Not shown on plot

# THERMAL CONDUCTIVITY OF $Sb_2Te_3$ INTERMETALLIC COMPOUND

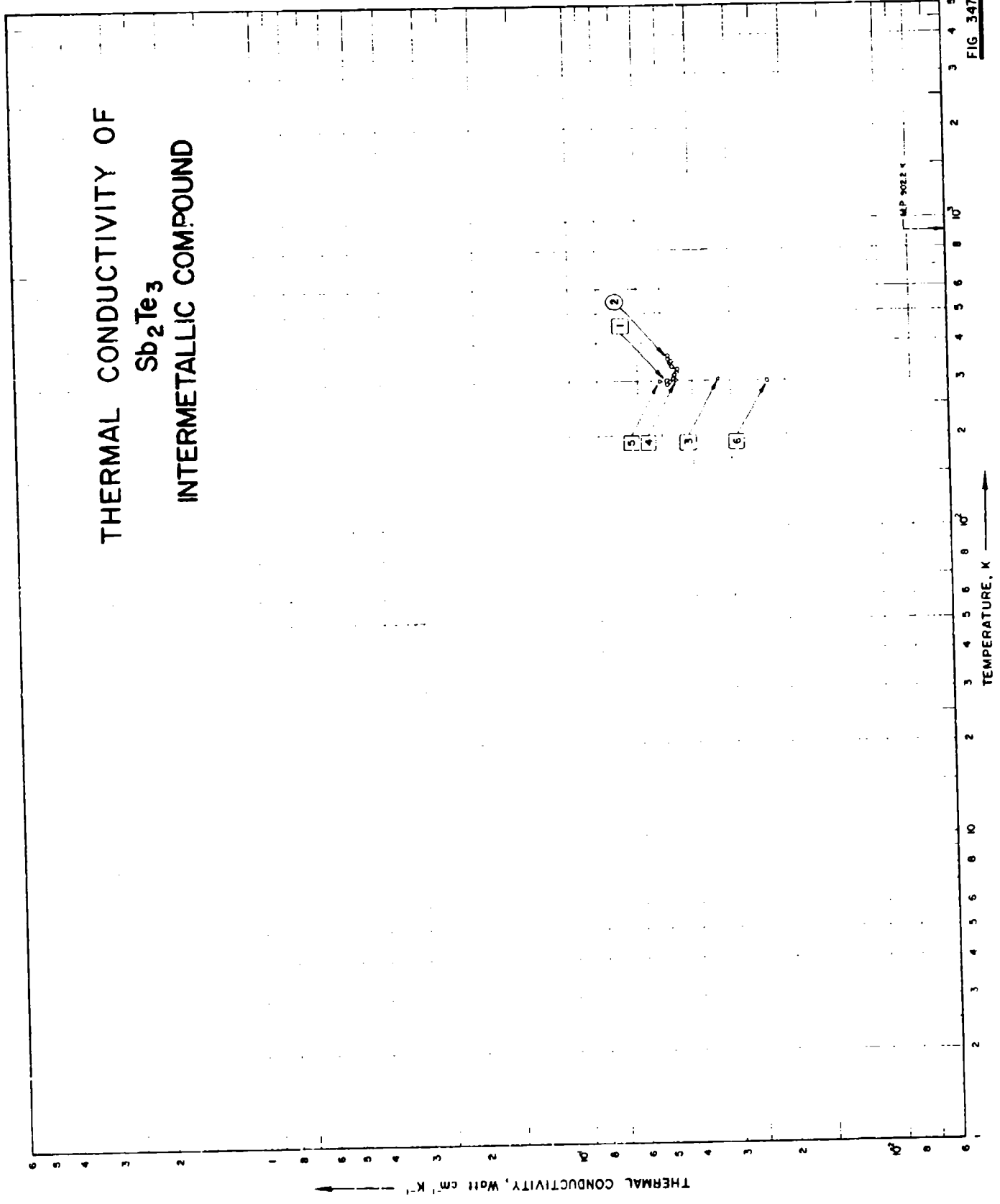


FIG 347

SPECIFICATION TABLE NO. 347 THERMAL CONDUCTIVITY OF  $Sb_2Te_3$  INTERMETALLIC COMPOUNDS

[For Data Reported in Figure and Table No. 347]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)	Specifications and Remarks
1	521	C	1957	300		$Sb_2Te_3$	p-type, excess Sb; rhombohedral crystal; hole concentration $70 \times 10^{18} \text{ cm}^{-3}$ ; prepared by sintering, annealing, and crystallizing by Bridgeman technique.	
2	595	L	1961	293-363		$Sb_2Te_3$	Polycrystal; cylindrical specimen; current carriers $10^{19} \text{ cm}^{-3}$ .	
3	832		1962	303.2		$Sb_2Te_3$	p-type; prepared from 99.99% pure elements by cold-pressing -60 mesh powder at 35 tsi and sintering at 450 C for 3 hrs in Argon; heat flow perpendicular to pressing direction; electrical resistivity $0.78 \times 10^{-3} \text{ ohm cm}$ and $1.09 \times 10^{-3} \text{ ohm cm}$ at 100-400 C and 150-450 C, respectively.	
4	833		1962	303.2		$Sb_2Te_3$	p-type; prepared from 99.99% pure elements by cold-pressing -60 mesh powder at 35 tsi and sintering at 500 C for 3 hrs in Argon; heat flow perpendicular to pressing direction; electrical resistivity $0.21 \times 10^{-3} \text{ ohm cm}$ and $0.61 \times 10^{-3} \text{ ohm cm}$ at 30 C and 150-450 C, respectively.	
5	936	T	1965	298.2		$Sb_2Te_3$	2.58 Te excess (calculated); p-type; $0.5 \times 0.5 \times 1 \text{ cm}$ ; prepared from 99.999 Sb supplied by Consolidated Mining and Smelting Co., and from 99.97 Te, supplied by Canadian Copper Refiners, Ltd.; materials weighed out, crushed, sealed in an ampule in a vacuum of $10^{-5}$ Torr, heated at 900 C for 20 hrs, rocked, cooled, zone-melted at a rate of 0.07-0.25 in. hr <sup>-1</sup> , then cooled and cut; thermal conductivity data calculated from measured values of figure of merit, Seebeck coefficient, and electrical conductivity; electrical conductivity reported as $4.33 \times 10^3 \text{ ohm}^{-1}\text{cm}^{-1}$ at room temperature. Cut from the same ingot as the above specimen; electrical conductivity reported as $4.33 \times 10^3 \text{ ohm}^{-1}\text{cm}^{-1}$ at room temperature.	
6	936	T	1965	298.2		$Sb_2Te_3$		



DATA TABLE NO. 347 THERMAL CONDUCTIVITY OF Sb, Te, INTERMETALLIC COMPOUNDS  
 [Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup> K<sup>-1</sup>]

T	k
<u>CURVE 1</u>	
300	0.048
<u>CURVE 2</u>	
293.2	0.0481
298.2	0.0473
300.7	0.0464*
305.7	0.0456
315.2	0.0452
320.2	0.0445
328.2	0.0445
335.7	0.0460
343.2	0.0469
348.2	0.0464
355.7	0.0477
363.2	0.0477
<u>CURVE 3</u>	
303.2	0.033
<u>CURVE 4</u>	
303.2	0.0449
<u>CURVE 5</u>	
298.2	0.0505
<u>CURVE 6</u>	
298.2	0.0229

\* Not shown on plot

SPECIFICATION TABLE NO. 348    THERMAL CONDUCTIVITY OF  $As_2Te_3$  INTERMETALLIC COMPOUNDS

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
1	521	C	1957	300		$As_2Te_3$	N-type (excess Te, I); electron concentration $18 \times 10^{18} \text{ cm}^{-3}$ at 300 K; monoclinic structure $a = 14.4 \text{ \AA}$ , $b = 4.05 \text{ \AA}$ , $c = 9.92 \text{ \AA}$ , and $\beta = 97$ ; prepared by verticils zone melting.
2	521	C	1957	300		$As_2Te_3$	Similar to the above specimen except p-type (excess As); hole concentration $40 \times 10^{18} \text{ cm}^{-3}$ at 300 K.

DATA TABLE NO. 348    THERMAL CONDUCTIVITY OF  $As_2Te_3$  INTERMETALLIC COMPOUNDS

[ Temperature, T, K, Thermal Conductivity, k, Watt  $\text{cm}^{-1}\text{K}^{-1}$  ]

T	k
<u>CURVE 1*</u>	
300	0.025
<u>CURVE 2*</u>	
300	0.027

\* No graphical presentation

SPECIFICATION TABLE NO. 349 THERMAL CONDUCTIVITY OF Ba<sub>2</sub>Pb INTERMETALLIC COMPOUNDS

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)	Specifications and Remarks
1	548	L	1961	298.2		Ba <sub>2</sub> Pb; No. 1		Seebeck coeff. $14.6 \mu\text{V K}^{-1}$ at 25 C; electrical resistivity $2.46 \times 10^{-4}$ ohm cm at 25 C; figure of merit $0.305 \times 10^{-4} \text{ K}^{-1}$ at 25 C.
2	548	L	1961	298.2		Ba <sub>2</sub> Pb; No. 3		Seebeck coeff. $133.5 \mu\text{V K}^{-1}$ at 25 C; electrical resistivity $1.37 \times 10^{-2}$ ohm cm at 25 C; figure of merit $0.893 \times 10^{-4} \text{ K}^{-1}$ at 25 C.
3	548	L	1961	296.2		Ba <sub>2</sub> Pb; No. 4		Seebeck coeff. $37.7 \mu\text{V K}^{-1}$ at 25 C; electrical resistivity $1.54 \times 10^{-3}$ ohm cm at 25 C; figure of merit $0.83 \times 10^{-4} \text{ K}^{-1}$ at 25 C.

DATA TABLE NO. 349 THERMAL CONDUCTIVITY OF Ba<sub>2</sub>Pb INTERMETALLIC COMPOUNDS[ Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup>K<sup>-1</sup> ]

T k

CURVE 1\*

298.2 0.0294

CURVE 2\*

298.2 0.0147

CURVE 3\*

298.2 0.0111

\* No graphical presentation

SPECIFICATION TABLE NO. 350 THERMAL CONDUCTIVITY OF Ba<sub>2</sub>Sn INTERMETALLIC COMPOUNDS

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent). Specifications and Remarks
1	548	L	1961	298.2		Ba <sub>2</sub> Sn; No. 3	Seebeck coeff. $14.2 \mu\text{V/K}^{-1}$ at 25 C; electrical resistivity $1.70 \times 10^{-3}$ ohm cm at 25 C; figure of merit $0.27 \times 10^{-4} \text{K}^{-1}$ at 25 C.
2	548	L	1961	298.2		Ba <sub>2</sub> Sn; No. 4	1.0 mole percent excess Ba; Seebeck coeff. $19.7 \mu\text{V/K}^{-1}$ at 25 C; electrical resistivity $1.40 \times 10^{-3}$ ohm cm at 25 C; figure of merit $0.303 \times 10^{-4} \text{K}^{-1}$ at 25 C.

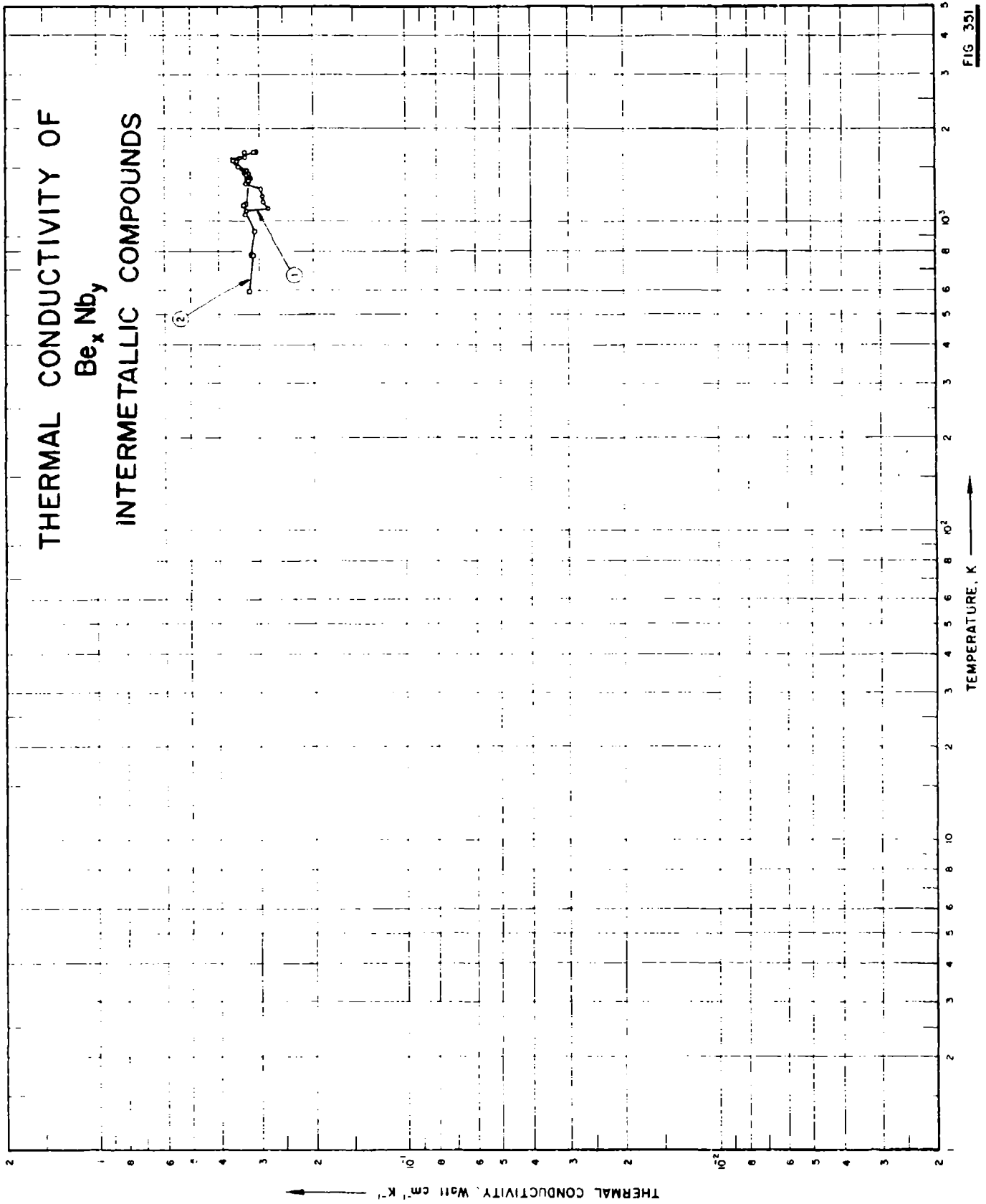
DATA TABLE NO. 350 THERMAL CONDUCTIVITY OF Ba<sub>2</sub>Sn INTERMETALLIC COMPOUNDS

‡ Temperature, T, K; Thermal Conductivity,  $\kappa$ , Watt  $\text{cm}^{-1}\text{K}^{-1}$

T	†
<u>CURVE 1</u>	
298.2	0.00437
<u>CURVE 2</u>	
298.2	0.00912

No graphical presentation

THERMAL CONDUCTIVITY OF  
 $\text{Be}_x\text{Nb}_y$   
INTERMETALLIC COMPOUNDS



SPECIFICATION TABLE NO. 351 THERMAL CONDUCTIVITY OF  $\text{Be}_x\text{Nb}_y$  INTERMETALLIC COMPOUNDS

[For Data Reported in Figure and Data Table No. 351]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent). Specifications and Remarks
1	446	R	1959	1052-1691		$\text{Be}_2\text{Nb}$	Prepared from Be (99.3 pure) supplied by Brush Beryllium Co. and Nb (< 99.5 pure) supplied by Kaweco Chemical Co.; cylindrical specimen 2.5 in. dia, 2.5 in. high; prepared by hot pressing powdered $\text{NbBe}_2$ at 2000 psi and 2700 F for 1 to 2 hrs.
2	938	R	1962	596-1699	5	$\text{Be}_{17}\text{Ni}_3$	Specimen consists of five vertically stacked cylinders, each 2.625 in. O. D. and 1 in. high with a 0.25 in. bore concentric with the axis; fabricated by cold pressing and sintering from $\text{Be}_{17}\text{Ni}_3$ powder; density 3.23 g cm <sup>-3</sup> .

DATA TABLE NO. 351 THERMAL CONDUCTIVITY OF  $B_e-Nb_y$  INTERMETALLIC COMPOUNDS[Temperature, T, K; Thermal Conductivity, k, Watt  $cm^{-1}K^{-1}$ ]

T	k
<u>CURVE 1</u>	
1051.5	0.325
1085.9	0.322
1104.3	0.282
1165.4	0.292
1203.2	0.294
1283.2	0.298
1345.9	0.334
1422.1	0.320
1430.4	0.325
1475.4	0.308
1514.3	0.350
1579.3	0.355
1634.3	0.324
1690.9	0.325

<u>CURVE 2</u>	
596	0.324
774	0.315
775	0.317
930	0.313
1144	0.336
1146	0.332
1367	0.327
1368	0.322
1368	0.327
1596	0.350
1597	0.360
1699	0.310
1699	0.312

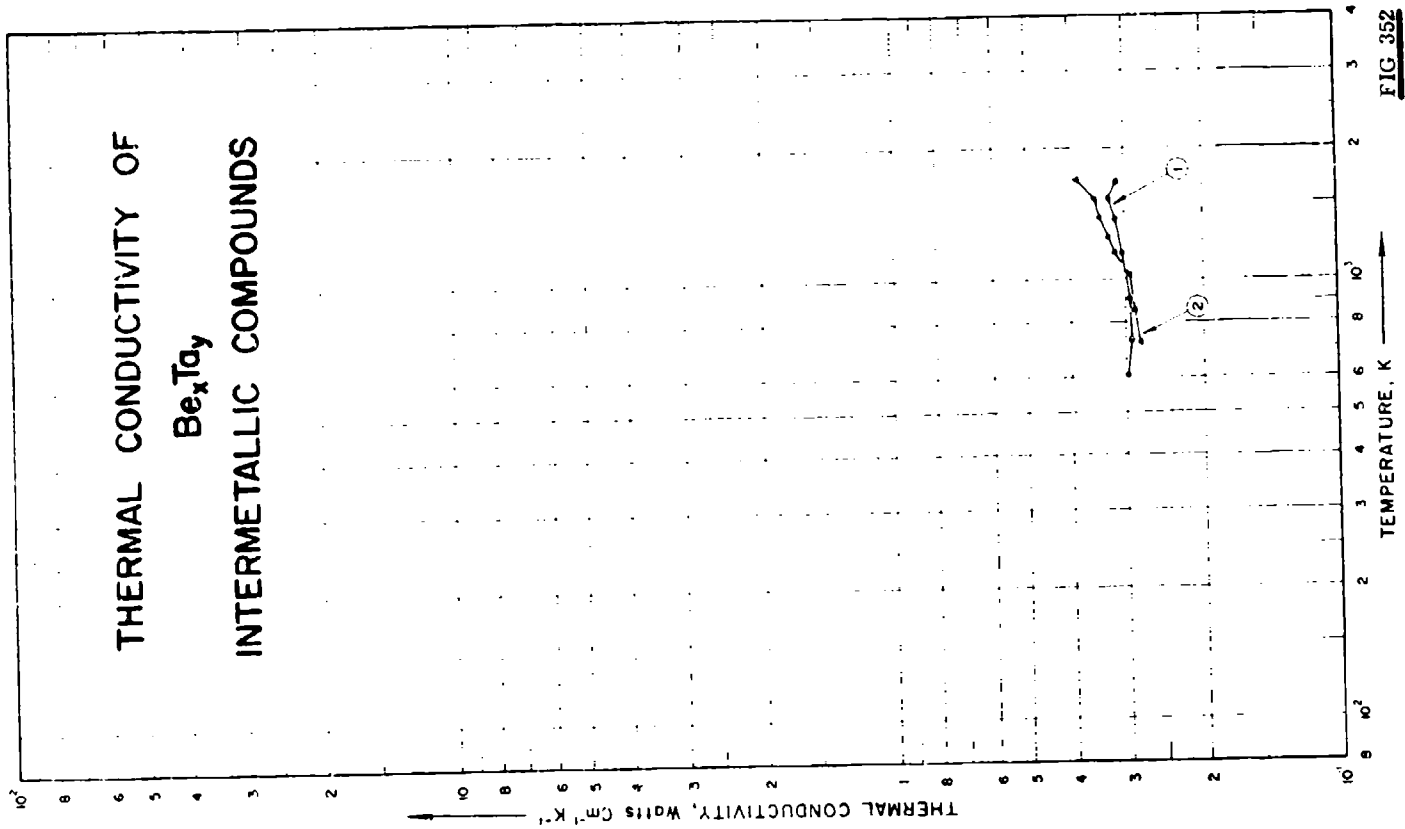


FIG 352



SPECIFICATION TABLE NO. 352 THERMAL CONDUCTIVITY OF  $\text{Be}_x\text{Ta}_y$  INTERMETALLIC COMPOUNDS

(For Data Reported in Figure and Table No. 352.)

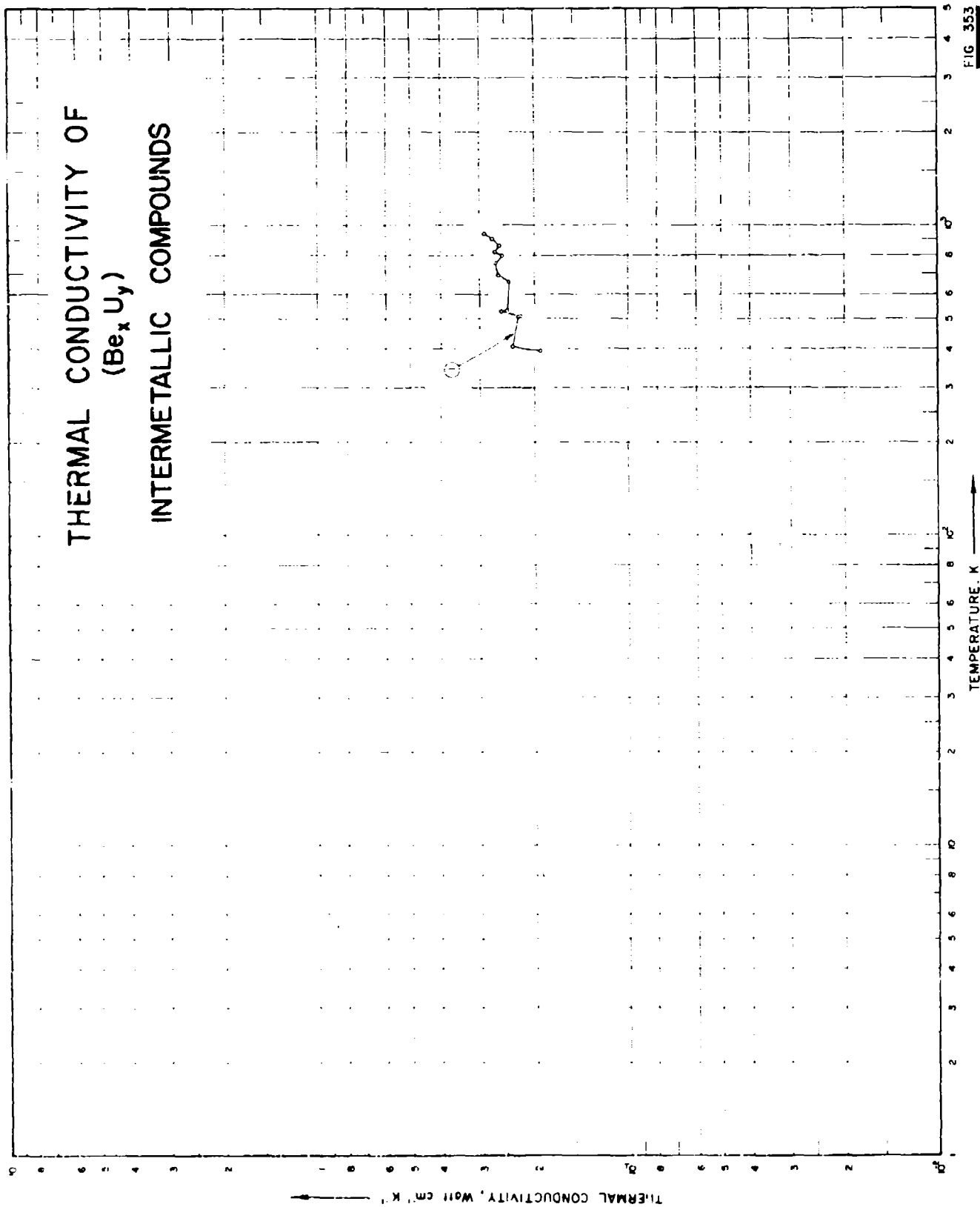
Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent). Specifications and Remarks
1	583	R	1962	608-1689	5.0	$\text{Ta}_2\text{Be}_{17}$	Single phase; 98% of absolute density.
2	594	R	1961	721-1698	5.0	$\text{TaBe}_{12}$	Specimen consisted of 5 hollow cylinders, each 2-5/8 in. O. D., 1/4 in. I. D. and 1 in. high.

DATA TABLE NO. 352 THERMAL CONDUCTIVITY OF  $BxTa_y$  INTERMETALLIC COMPOUNDS[Temperature, T, K. Thermal Conductivity,  $k$ , Watt  $cm^{-1} K^{-1}$ ]

T	k
<u>CURVE 1</u>	
608.2	0.298
609.3	0.304 <sup>a</sup>
732.1	0.291
742.6	0.291 <sup>a</sup>
905.9	0.292
1156.5	0.305
1156.5	0.333 <sup>a</sup>
1377.6	0.315
1378.2	0.315 <sup>a</sup>
1547.5	0.325
1548.2	0.324 <sup>a</sup>
1688.7	0.312
1689.3	0.312
<u>CURVE 2</u>	
726.9	0.279
724.8	0.279
857.6	0.287
862.6	0.287 <sup>a</sup>
1021.5	0.292
1031.2	0.292 <sup>a</sup>
1162.1	0.318
1166.5	0.315 <sup>a</sup>
1253.4	0.329
1257.6	0.325 <sup>a</sup>
1383.9	0.343
1397.1	0.317 <sup>a</sup>
1397.1	0.343 <sup>a</sup>
1533.2	0.351
1544.3	0.357 <sup>a</sup>
1697.6	0.386

Not shown on plot

THERMAL CONDUCTIVITY OF  
(Be<sub>x</sub>U<sub>y</sub>)  
INTERMETALLIC COMPOUNDS



SPECIFICATION TABLE NO. 353 THERMAL CONDUCTIVITY OF  $Bc_xU_y$  INTERMETALLIC COMPOUNDS

(For Data Reported in Figure and Table No. 353)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
1	556	C	1959	295-940		UBe <sub>13</sub>	Solid-Solid reaction of UH <sub>3</sub> and powdered Be in an induction furnace under 1 argon atmosphere at 1559 C and sintered; x-ray density 4.37 g cm <sup>-3</sup> .

DATA TABLE NO. 353 THERMAL CONDUCTIVITY OF  $\text{Be}_x\text{U}_y$  INTERMETALLIC COMPOUNDS

[Temperature, T, K; Thermal Conductivity, k, Watt  $\text{cm}^{-1}\text{K}^{-1}$ ]

T	k
395.2	0.192
406.2	0.236
506.2	0.225
510.7	0.220
528.2	0.258
533.2	0.246
659.2	0.242
693.2	0.262
756.2	0.267
798.2	0.250
823.2	0.269
961.2	0.260
905.2	0.274
940.2	0.290

CURVE 1

SPECIFICATION TABLE NO. 354 THERMAL CONDUCTIVITY OF Be<sub>13</sub>Zr INTERMETALLIC COMPOUNDS

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)	Specifications and Remarks
1	446	R	1959	983-1657		Be <sub>13</sub> Zr		Prepared from Be (99.3 pure) supplied by Brush Beryllium Co. and ZrH <sub>2</sub> (98.5% pure) supplied by Metal Hydrides Inc.; cylindrical specimen 2.5 in. dia, 2.5 in. long; prepared by hot pressing powdered ZrBe <sub>13</sub> at 2000 psi and 2700 to 2800 F for 2 to 2.50 hr.

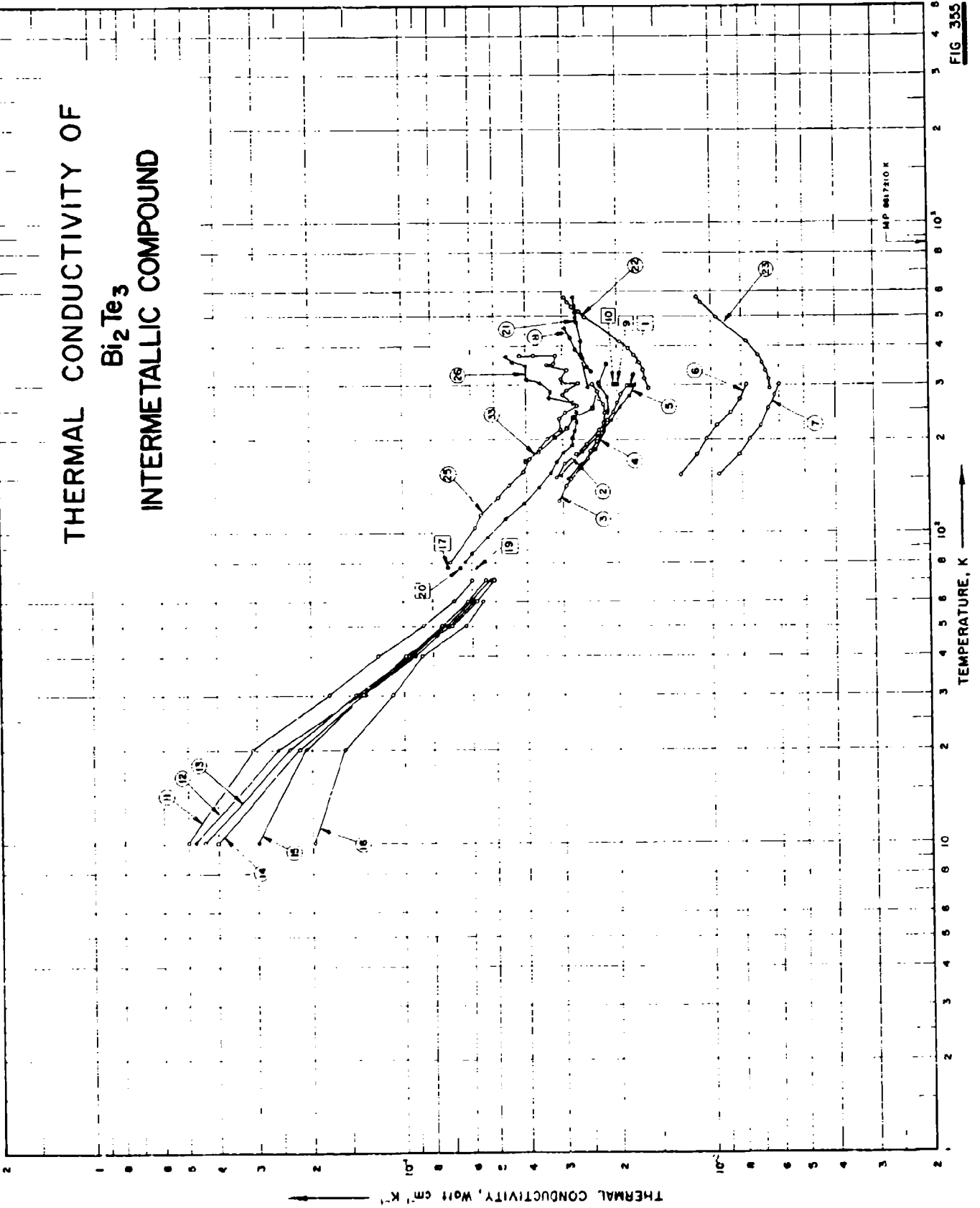
DATA TABLE NO. 354 THERMAL CONDUCTIVITY OF Be<sub>13</sub>Zr INTERMETALLIC COMPOUNDS[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup>K<sup>-1</sup>]

T	CURVE 1*		T	CURVE 1 (cont.)*	
	k	k		k	k
983.1	0.402		1614.3	0.350	
1025.4	0.391		1657.1	0.343	
1045.4	0.337				
1089.8	0.318				
1158.2	0.358				
1198.2	0.391				
1210.4	0.381				
1237.1	0.379				
1289.8	0.405				
1299.3	0.322				
1348.2	0.368				
1380.9	0.360				
1393.2	0.369				
1395.4	0.331				
1395.9	0.346				
1459.3	0.341				
1517.6	0.382				
1540.5	0.370				
1573.2	0.382				
1610.9	0.306				

\* No graphical presentation

THERMAL CONDUCTIVITY OF  
 $\text{Bi}_2\text{Te}_3$   
 INTERMETALLIC COMPOUND

FIGURE SHOWS ONLY 26 OF THE CURVES REPORTED IN TABLE



SPECIFICATION TABLE NO. 355 THERMAL CONDUCTIVITY OF  $\text{Bi}_2\text{Te}_3$  INTERMETALLIC COMPOUNDS

(For Data Reported in Figure and Table No. 355)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)	Specifications and Remarks
1	328	L, C	1954	298-2	0.2-3.0	$\text{Bi}_2\text{Te}_3$		Fine crystalline structure; extruded; current carriers $6.8 \times 10^{18} \text{ cm}^{-3}$ ; 15 mm dia. 1 to 3 mm thick; electrical conductivity $1500 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 25°C.
2	332	L	1956	150-300	~2	$\text{Bi}_2\text{Te}_3$ , No. 1		n-type; cut from impure end of zone refined bar; rectangular specimen; heat flow parallel to zoning direction and the cleavage plane; electrical conductivity $750 \text{ ohm}^{-1} \text{ cm}^{-1}$ at room temperature.
3	332	L	1956	127-300	~2	$\text{Bi}_2\text{Te}_3$ , No. 2		n-type; intrinsic; zone refined; rectangular specimen; cut from the same bar as the above specimen; heat flow and zoning direction parallel to the cleavage plane; electrical conductivity $200 \text{ ohm}^{-1} \text{ cm}^{-1}$ at room temperature.
4	332	L	1956	150-293	~2	$\text{Bi}_2\text{Te}_3$ , No. 3		Similar to the above specimen (adjacent specimen).
5	332	L	1956	152-300	~2	$\text{Bi}_2\text{Te}_3$ , No. 4		p-type single crystal; heat flow parallel to the cleavage plane; electrical conductivity $300 \text{ ohm}^{-1} \text{ cm}^{-1}$ at room temperature.
6	332	L	1956	154-300	~2	$\text{Bi}_2\text{Te}_3$ , No. 5		p-type single crystal; heat flow perpendicular to the cleavage plane; electrical conductivity $500 \text{ ohm}^{-1} \text{ cm}^{-1}$ at room temperature.
7	332	L	1956	154-300	~2	$\text{Bi}_2\text{Te}_3$ , No. 6		Similar to the above specimen.
8	326	C	1960	333-457		$\text{Bi}_2\text{Te}_3$		p-type polycrystalline specimen; prepared from Asarco 99.999% pure Bi and 99.999% Te; prepared by fusing in vacuo; hexagonal unit cell with constants of: $a = 4.37 \text{ \AA}$ and $c = 30.62 \text{ \AA}$ ; specimen 12 mm dia and 6 mm thick; electrical conductivity $\sim 10^5 \text{ ohm}^{-1} \text{ cm}^{-1}$ ; Armeto iron used as reference; measured in vacuo of $\sim 10^{-6} \text{ mm Hg}$ .
9	325	T	1959	300	1	$\text{Bi}_2\text{Te}_3$ , 3b		p-type; electrical resistivity $1.90 \times 10^{-3} \text{ ohm cm}$ .
10	325	T	1959	300	1	$\text{Bi}_2\text{Te}_3$ , 17a		p-type; electrical resistivity $1.07 \times 10^{-3} \text{ ohm cm}$ .
11	311	L	1960	10-70	5	$\text{Bi}_2\text{Te}_3$ , SBTC 18		p-type single crystal; form factor (F = length/cross-sectional area) = 30.4; measured in a vacuo of $5 \times 10^{-5} \text{ mm Hg}$ .
12	311	L	1960	10-70	5	$\text{Bi}_2\text{Te}_3$ , SBTC 19		p-type single crystal; doped with I, final concentration 0.027; F = 25.3; measured in a vacuo of $5 \times 10^{-5} \text{ mm Hg}$ .
13	311	L	1960	10-70	5	$\text{Bi}_2\text{Te}_3$ , SBTC 27		n-type single crystal; doped with I, final concentration 0.037; F = 30.0; measured in a vacuo of $5 \times 10^{-5} \text{ mm Hg}$ .
14	311	L	1960	10-70	5	$\text{Bi}_2\text{Te}_3$ , SBTC 16		n-type single crystal; doped with I, final concentration 0.046; F = 24.0; measured in a vacuo of $5 \times 10^{-5} \text{ mm Hg}$ .
15	311	L	1960	10-60	5	$\text{Bi}_2\text{Te}_3$ , SBTC 15		n-type single crystal; doped with I, final concentration 0.059; F = 52.8; measured in a vacuo of $5 \times 10^{-5} \text{ mm Hg}$ .
16	311	L	1960	10-60	5	$\text{Bi}_2\text{Te}_3$ , SBTC 10		n-type single crystal; doped with I, final concentration 0.124; F = 34.0; measured in a vacuo of $5 \times 10^{-5} \text{ mm Hg}$ .
17	386	L	1958	77		$\text{Bi}_2\text{Te}_3$ , S-22		p-type single crystal; undoped; measured in a magnetic field parallel to the crystal axis; heat flow perpendicular to the crystal axis.



SPECIFICATION TABLE NO. 355 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
18	386	L	1958	77		$\text{Bi}_2\text{Te}_3$ , S 21	Similar to the above specimen.
19	386	L	1958	77		$\text{Bi}_2\text{Te}_3$ , S 17	n-type single crystal, doped with 0.055 I (nominal); magnetic field parallel to the crystal axis, heat flow perpendicular to the crystal axis.
20	386	L	1958	77		$\text{Bi}_2\text{Te}_3$ , S 15	Similar to the above specimen except doped with 0.09 I.
21	387	P	1960	293-573	$\pm 20$	$\text{Bi}_2\text{Te}_3$ , 68	p-type, measured in a 1.95 MEV electron beam, heat flow parallel to crystal planes; electrical resistivity $\sim 5.1 \times 10^{-4}$ ohm cm at room temperature.
22	387	F	1960	293-573	$\pm 20$	$\text{Bi}_2\text{Te}_3$ , 64	p-type, measured in a 1.95 MEV electron beam; heat flow perpendicular to the crystal planes; $\rho = 9.6 \times 10^{-4}$ ohm cm at room temperature.
23	387	P	1960	293-573	$\pm 20$	$\text{Bi}_2\text{Te}_3$ , 59	Similar to the above specimen except $\rho = 13.8 \times 10^{-4}$ ohm cm at room temperature.
24	387	P	1960	293-573	$\pm 20$	$\text{Bi}_2\text{Te}_3$ , 69	n-type; measured in a 1.95 MEV electron beam; heat flow perpendicular to the crystal planes; $\rho = 6.5 \times 10^{-4}$ ohm cm at room temperature.
25	388	L	1957	80-370		$\text{Bi}_2\text{Te}_3$	p-type single crystal; specimen $0.5 \times 0.5 \times 2.5$ cm, prepared by zone melting from Bi (99.999 pure, supplied by Cerro de Páco Corp.) and Te (99.999 pure, supplied by American Smelting and Refining Co.); $2 \times 10^{19}$ excess holes $\text{cm}^{-3}$ .
26	388	L	1957	80-370		$\text{Bi}_2\text{Te}_3$ , D-13	n-type single crystal; specimen $0.5 \times 0.5 \times 2.5$ cm, prepared from Bi (99.999 pure, supplied by Cerro de Páco Corp.) and Te (99.999 pure, supplied by American Smelting and Refining Co.); $3 \times 10^{17}$ excess electrons $\text{cm}^{-3}$ .
27	549	L	1969	311.2	$\pm 10$	$\text{Bi}_2\text{Te}_3$	Supplied by Electronic System Laboratory, M. I. T.
28	521	C	1957	300		$\text{Bi}_2\text{Te}_3$	n-type (excess Te, I); electron concentration $5 \times 10^{18} \text{ cm}^{-3}$ at 300 K; rhombohedral structure ( $a_0 = 10.45 \text{ \AA}$ and $\alpha = 24^\circ 8'$ ); prepared by zone melting, quenching and annealing.
29	521	C	1957	300		$\text{Bi}_2\text{Te}_3$	p-type (excess Bi, Pb); hole concentration $8 \times 10^{18} \text{ cm}^{-3}$ at 300 K; structure and specimen similar to the above.
30	386	L	1958	77		$\text{Bi}_2\text{Te}_3$ , S 11	n-type single crystal; doped with (nominal) 0.40 I; heat flow perpendicular to the crystal axis, a magnetic field parallel to the crystal axis.
31	936	T	1965	298.2		$\text{Bi}_2\text{Te}_3$	2.03 Te excess (calculated); n-type, $0.5 \times 0.5 \times 1$ cm; prepared from 99.999 pure Bi supplied by Consolidated Mining and Smelting Co., and from 99.9% pure Te supplied by Canadian Copper Refiners, Ltd.; materials weighed out, crushed, sealed in an ampule in a vacuum of $10^{-5}$ Torr, heated at 900 C for 20 hrs., rocked, cooled, zone-melted at a rate of 0.07-0.28 in./hr., then cooled and cut; thermal conductivity data calculated from measured values of figure of merit, Seebeck coefficient, and electrical conductivity; electrical conductivity reported as $1.78 \times 10^3 \text{ ohm}^{-1} \text{ cm}^{-1}$ at room temperature.
32	936	T	1965	298.2		$\text{Bi}_2\text{Te}_3$	Cut from the same ingot as the above specimen; electrical conductivity reported as $1.75 \times 10^3 \text{ ohm}^{-1} \text{ cm}^{-1}$ at room temperature.
33	936	T	1965	170-351		$\text{Bi}_2\text{Te}_3$	Similar to the above specimen except electrical resistivity reported as 244, 267, 309, 360, 389, 454, 517, 582, and 721 $\mu\text{ohm cm}$ at 171, 183, 203, 219, 239, 256, 281, 304, and 350 K, respectively.

SPECIFICATION TABLE NO. 355 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent). Specifications and Remarks
34	936	T	1965	298.2		$\text{Bi}_2\text{Te}_3$	Similar to the above specimen except electrical conductivity reported as $1.29 \times 10^3 \text{ ohm}^{-1} \text{ cm}^{-1}$ at room temperature.
35	936	T	1965	298.2		$\text{Bi}_2\text{Te}_{3.04}$	0.95 Te excess (calculated); n-type; same fabrication method and measuring method as the above specimen; electrical conductivity reported as $0.24 \times 10^3 \text{ ohm}^{-1} \text{ cm}^{-1}$ at room temperature.

DATA TABLE NO. 355 THERMAL CONDUCTIVITY OF Bi, Te<sub>3</sub> INTERMETALLIC COMPOUNDS[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup> K<sup>-1</sup>]

T	k	T	k	T	k	T	k	T	k	T	k	T	k	T	k	T	k
<u>CURVE 1</u>		<u>CURVE 5</u>		<u>CURVE 10</u>		<u>CURVE 15</u>		<u>CURVE 22</u>		<u>CURVE 25 (cont.)</u>		<u>CURVE 27*</u>					
298.2	0.0180	152.6	0.0307	300	0.0206	10	0.299	293.2	0.0159	185.9	0.036	311.2	0.016				
<u>CURVE 2</u>		<u>CURVE 6</u>		<u>CURVE 11</u>		<u>CURVE 16</u>		<u>CURVE 23</u>		<u>CURVE 26</u>		<u>CURVE 28*</u>					
150.0	0.0315	181.9	0.0261	<u>CURVE 11</u>		20	0.208	313.2	0.0163	200.0	0.034	300	0.021				
166.7	0.0295	205.9	0.0234	30	0.136*	30	0.136*	333.2	0.0165	213.7	0.0305	<u>CURVE 28*</u>					
178.9	0.0271	230.0	0.0210	40	0.091	40	0.091	353.2	0.0170	232.6	0.0308	300	0.021				
192.2	0.0253	250.4	0.0196	50	0.071	50	0.071	373.2	0.0176	243.9	0.0295	<u>CURVE 29*</u>					
213.0	0.0232	276.3	0.0182	60	0.059	60	0.059	393.2	0.0184	255.1	0.027	300	0.021				
230.0	0.0216	300.0	0.0176	70	0.046	70	0.046	413.2	0.0197	277.8	0.030	300	0.021				
245.9	0.0207	<u>CURVE 6</u>		<u>CURVE 12</u>		<u>CURVE 17</u>		<u>CURVE 24*</u>		<u>CURVE 29</u>		<u>CURVE 30*</u>					
261.5	0.0200	153.7	0.0126	10	0.440	10	0.196	293.2	0.00650	285.7	0.0290	300	0.021				
278.5	0.0195	177.8	0.0112	20	0.235	20	0.156	313.2	0.00653	303.0	0.030	300	0.021				
298.9	0.0185	200.0	0.0104	30	0.143	30	0.109	333.2	0.00730	312.5	0.0305	300	0.021				
<u>CURVE 3</u>		<u>CURVE 7</u>		<u>CURVE 13</u>		<u>CURVE 18*</u>		<u>CURVE 24*</u>		<u>CURVE 31*</u>		<u>CURVE 32*</u>					
127.4	0.0309	221.1	0.00956	40	0.083	40	0.063	353.2	0.0107	347.2	0.0340	298.2	0.0234				
141.1	0.0292	244.4	0.00869	50	0.075	50	0.055	373.2	0.0111	349.7	0.0322	298.2	0.0234				
148.1	0.0282	265.3	0.00804	60	0.062	60	0.055	393.2	0.0111	370.4	0.0318	298.2	0.0234				
161.9	0.0264	300.0	0.00771	70	0.054	70	0.054	413.2	0.0111	370.4	0.0308	298.2	0.0234				
173.0	0.0251	<u>CURVE 7</u>		<u>CURVE 14</u>		<u>CURVE 19</u>		<u>CURVE 25</u>		<u>CURVE 32*</u>		<u>CURVE 33</u>					
184.4	0.0246	153.7	0.00949	10	0.476	10	0.072	293.2	0.00272	80.1	0.0635	170.4	0.0402				
196.3	0.0235	177.8	0.00818	20	0.256	20	0.072	313.2	0.00710	85.5	0.060	182.1	0.0364				
209.6	0.0226	200.0	0.00753	30	0.140	30	0.072	333.2	0.00710	96.6	0.053	202.0	0.0319				
225.2	0.0223	224.1	0.00695	40	0.094	40	0.072	353.2	0.0107	110.9	0.0465	217.9	0.0290				
243.7	0.0221	251.9	0.00655	50	0.073	50	0.072	373.2	0.0111	123.8	0.0410	236.4	0.0278				
259.5	0.0224	275.9	0.00622	60	0.061	60	0.066	393.2	0.0111	140.1	0.0365	253.2	0.0243				
285.2	0.0214	300.0	0.00600	70	0.052	70	0.066	413.2	0.0111	156.3	0.033	281.7	0.0235*				
300.0	0.0242	<u>CURVE 8</u>		<u>CURVE 14</u>		<u>CURVE 21</u>		<u>CURVE 25</u>		<u>CURVE 34*</u>		<u>CURVE 35*</u>					
<u>CURVE 4</u>		333.0	0.0243	10	0.400	10	0.0250	293.2	0.0250	264.6	0.0278	298.2	0.0240				
149.6	0.0287	341.3	0.0251	20	0.220	20	0.0250	313.2	0.0250	271.7	0.0270	298.2	0.0240				
166.7	0.0261	365.0	0.0259	30	0.137	30	0.0250	333.2	0.0250	285.7	0.0335	298.2	0.0240				
185.6	0.0238	427.4	0.0282	40	0.098	40	0.0250	353.2	0.0250	295.7	0.0335	298.2	0.0240				
211.1	0.0225	456.6	0.0293	50	0.070	50	0.0250	373.2	0.0250	303.0	0.0360	298.2	0.0240				
247.0	0.0215	<u>CURVE 9</u>		60	0.058	60	0.0250	393.2	0.0250	310.6	0.0398	298.2	0.0240				
272.2	0.0216	300	0.0199	70	0.051	70	0.0250	413.2	0.0250	344.8	0.0340	298.2	0.0240				
293.3	0.0228	<u>CURVE 9</u>		80	0.051	80	0.0250	433.2	0.0250	350.9	0.044	298.2	0.0240				
		427.4	0.0282	90	0.051	90	0.0250	453.2	0.0250	370.4	0.046	298.2	0.0240				
		456.6	0.0293	100	0.051	100	0.0250	473.2	0.0250	370.4	0.046	298.2	0.0240				

\* Not shown on plot

SPECIFICATION TABLE NO. 356 THERMAL CONDUCTIVITY OF  $B_xSi_y$  INTERMETALLIC COMPOUNDS

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)	Specifications and Remarks
1	962	C	1963	339-441	± 4	SiB <sub>4</sub>	0.5 Ca, 0.5 Cu, 0.430 O, 0.2 Al, 0.2 Fe, 0.2 W, 0.1 Co, 0.1 Ni, 0.072 H, 0.07 Au, 0.04 Cr, 0.02 Mg, 0.014 N, 0.01 Mn, and 0.01 Ti; polycrystalline; 0.5 in. $\times$ 0.875 in. long; prepared from amorphous boron of grade A-200 mesh supplied by Cooper Metallurgical Associates and 99.97 pure silicon of -200 mesh supplied by Union Carbide Corp., materials (2 Si to 1 B by weight) reacted in a dry argon atmosphere at a temperature not exceeding 1330 C for 4 hrs, the boride powders were cold-pressed at 100,000 psi, then hot-pressed for 3-4 hrs at 1350 C and 6000 psi in a graphite die with carbon impregnated graphite plungers, machined into rectangular bar; bulk density 1.94 g cm <sup>-3</sup> , porosity 20%; electrical conductivity reported as 0.437, 0.617, 0.904, 1.32, 1.50, 3.93, 6.49, 8.75, 15.4, 17.7, 21.1, and 28.2 ohm <sup>-1</sup> cm <sup>-1</sup> at 386, 407, 433, 478, 490, 610, 699, 756, 885, 962, 990, and 1075 K, respectively; polycrystalline alumina with 8 to 10% porosity used as comparative material; data corrected to zero porosity.	
2	962	C	1963	336-437		SiB <sub>4</sub>	0.410 O, 0.2 Al, 0.2 Ca, 0.2 Cu, 0.064 H, 0.02 Fe, 0.015 N, 0.01 Ni, 0.01 Ti, 0.01 Mn, and 0.005 Mg; polycrystalline; 0.5 in. $\times$ 0.875 in. long; prepared from amorphous boron of grade A-200 mesh supplied by Cooper Metallurgical Associates and 99.97 pure silicon of -200 mesh supplied by Union Carbide Corp., materials (2 Si to 1 B by weight) blended in a tungsten carbide mortar and pestle and pressed at 29,000 psi into cylindrical compacts without binder, then reacted in closed boron nitride crucibles containing a maximum of 1.5% B <sub>2</sub> O <sub>3</sub> and within a graphite induction furnace at 1630 C for 4 hrs, dry argon was fed into the furnace chamber during reaction, the boride powders were prepressed at 100,000 psi, then hot-pressed for 2 hrs at 1500 C and 6000 psi, machined into rectangular bar; bulk density 2.12 g cm <sup>-3</sup> , porosity 15%; electrical conductivity reported as 6.46, 18.1, 35.7, 60.3, 80.5, and 91.6 ohm <sup>-1</sup> cm <sup>-1</sup> at 299, 382, 429, 532, 575, and 585 K, respectively; polycrystalline alumina with 8 to 10% porosity used as comparative material; data corrected to zero porosity.	

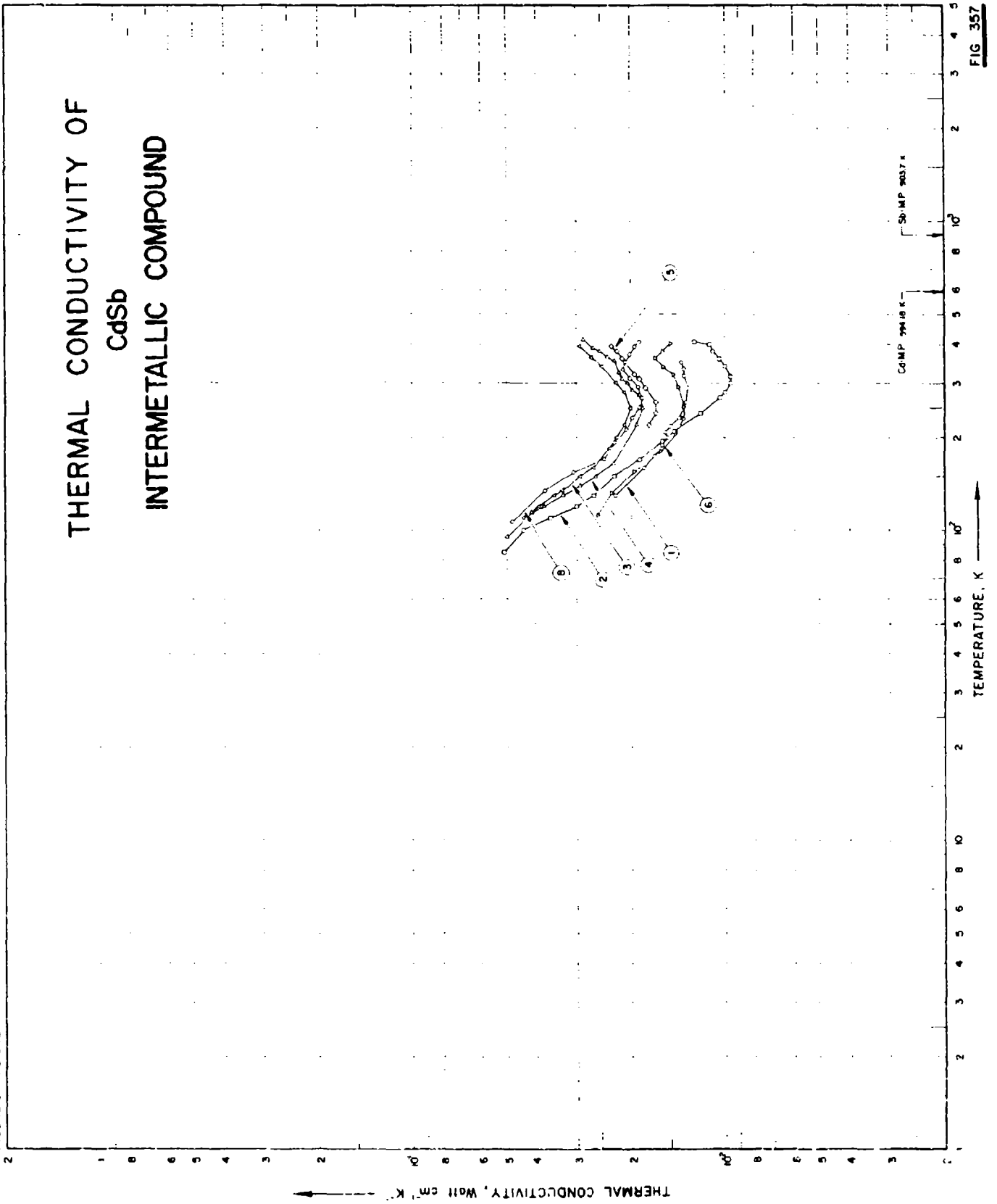
DATA TABLE NO. 356 THERMAL CONDUCTIVITY OF  $B_4Si_4$  INTERMETALLIC COMPOUNDS  
 [ Temperature, T, K; Thermal Conductivity, k, Watt  $cm^{-1}K^{-1}$  ]

T	k
	<u>CURVE 1<sup>a</sup></u>
339.2	0.0946
378.2	0.0904
411.2	0.0837
441.2	0.0768
	<u>CURVE 2<sup>a</sup></u>
336.2	0.0954
373.2	0.0925
412.2	0.0849
437.2	0.0766

<sup>a</sup> No graphical presentation

# THERMAL CONDUCTIVITY OF CdSb INTERMETALLIC COMPOUND

FIGURE SHOWS ONLY 7 OF THE CURVES REPORTED IN TABLE



SPECIFICATION TABLE NO. 357 THERMAL CONDUCTIVITY OF CuSb INTERMETALLIC COMPOUNDS

(For Data Reported in Figure and Table No. 357)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
1	823		1963	130-350		CuSb	No details reported.
2	824	L	1962	85-410		CuSb, No. 1	p-type; polycrystalline specimen; electrical conductivity $0.6 \text{ ohm}^{-1} \text{ cm}^{-1}$ at room temperature.
3	824	L	1962	110-395		CuSb, No. 2	p-type; single crystal grown by zone recrystallization; electrical conductivity $0.71 \text{ ohm}^{-1} \text{ cm}^{-1}$ at room temperature; heat flow normal to the [112] plane of the crystal.
4	824	L	1962	95-415		CuSb, No. 3	As above but the electrical conductivity, $0.68 \text{ ohm}^{-1} \text{ cm}^{-1}$ at room temperature.
5	824	L	1962	220-395		CuSb, No. 4	As above but the electrical conductivity, $6.54 \text{ ohm}^{-1} \text{ cm}^{-1}$ at room temperature.
6	940, 941	L	1964	113-491		CuSb, 1	p-type single crystal; high purity Cd and Sb obtained by triple vacuum fractional distillation followed by zone refining; specimen ingot grown by the zone recrystallization method; measured in a vacuum of $10^{-3} - 10^{-4}$ mm Hg; heat flow parallel to the [100] direction.
7	940, 941	L	1964	112-405		CuSb, 2	Cut from the same ingot as for above specimen; heat flow parallel to the [010] direction.
8	940, 941	L	1964	107-408		CuSb, 3	Cut from the same ingot as the above specimen; heat flow parallel to the [001] direction.
9	940, 941	L	1964	106-413		CuSb, 4	Similar to the above specimen; heat flow parallel to the (100) direction.
10	940, 941	L	1964	106-391		CuSb, 5	Cut from the same ingot as the above specimen; heat flow parallel to the [010] direction.
11	940, 941	L	1964	106-411		CuSb, 6	Cut from the same ingot as the above specimen; heat flow parallel to the [001] direction.



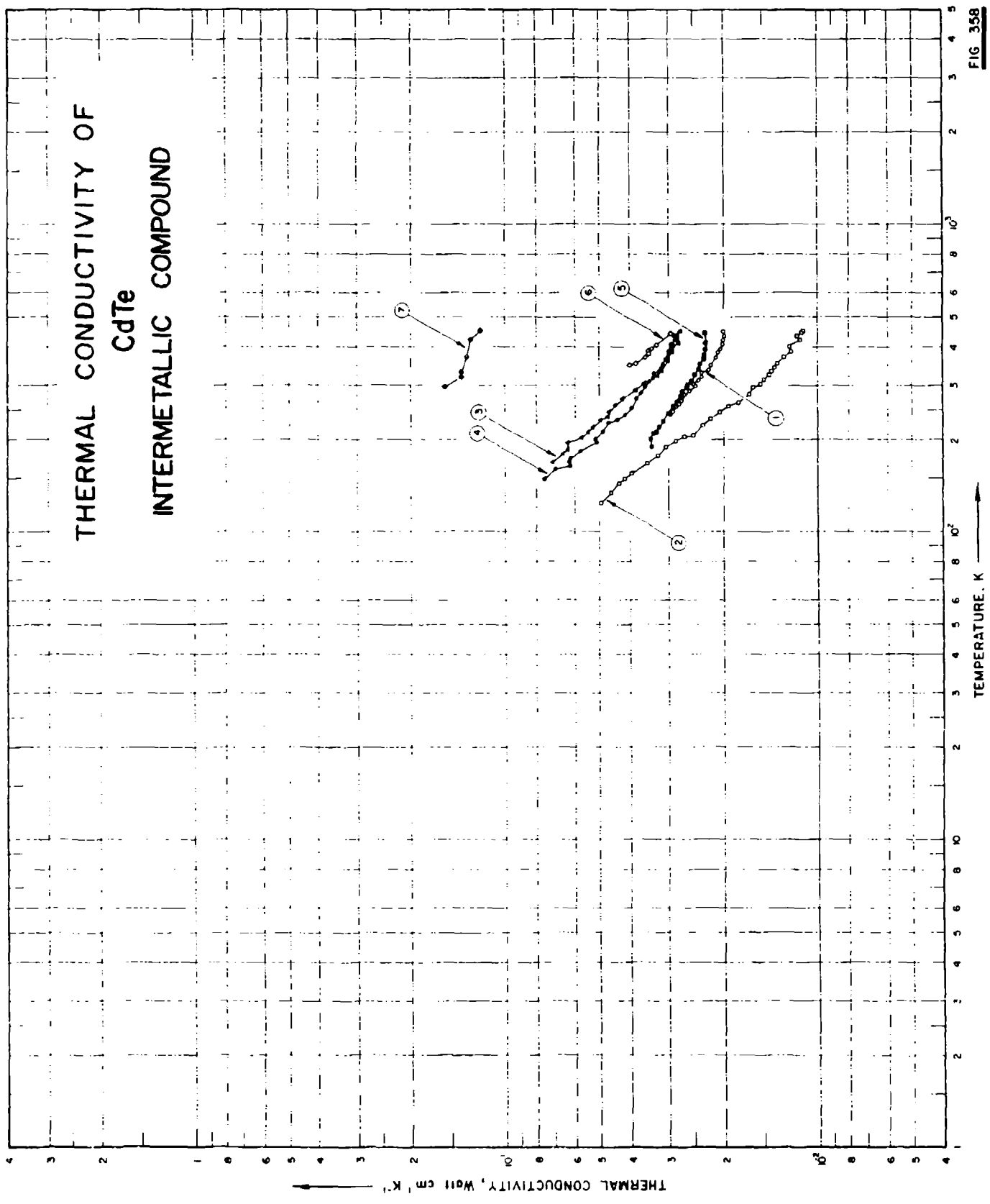


## SPECIFICATION TABLE NO. 357 THERMAL CONDUCTIVITY OF CdSb INTERMETALLIC COMPOUNDS

[For Data Reported in Figure and Table No. 357]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
1	823		1963	130-350		CdSb	No details reported.
2	824	L	1962	85-410		CdSb, No. 1	p-type; polycrystalline specimen, electrical conductivity $0.6 \text{ ohm}^{-1} \text{ cm}^{-1}$ at room temperature.
3	824	L	1962	110-395		CdSb, No. 2	p-type; single crystal grown by zone recrystallization; electrical conductivity $0.71 \text{ ohm}^{-1} \text{ cm}^{-1}$ at room temperature; heat flow normal to the [112] plane of the crystal.
4	824	L	1962	95-415		CdSb, No. 3	As above but the electrical conductivity, $0.68 \text{ ohm}^{-1} \text{ cm}^{-1}$ at room temperature.
5	824	L	1962	220-395		CdSb, No. 4	As above but the electrical conductivity, $0.54 \text{ ohm}^{-1} \text{ cm}^{-1}$ at room temperature.
6	940, 941	L	1964	113-401		CdSb, 1	p-type single crystal, high purity Cd and Sb, obtained by triple vacuum fractional distillation followed by zone refining; specimen ingot grown by the zone recrystallization method; measured in a vacuum of $10^{-3} - 10^{-4}$ mm Hg, heat flow parallel to the [100] direction.
7	940, 941	L	1964	112-406		CdSb, 2	Cut from the same ingot as the above specimen; heat flow parallel to the [010] direction.
8	940, 941	L	1964	107-408		CdSb, 3	Cut from the same ingot as the above specimen; heat flow parallel to the [001] direction.
9	940, 941	L	1964	106-413		CdSb, 4	Similar to the above specimen; heat flow parallel to the (100) direction.
10	940, 941	L	1964	106-391		CdSb, 5	Cut from the same ingot as the above specimen; heat flow parallel to the (010) direction.
11	940, 941	L	1964	106-411		CdSb, 6	Cut from the same ingot as the above specimen; heat flow parallel to the [001] direction.

THERMAL CONDUCTIVITY OF  
CdTe  
INTERMETALLIC COMPOUND

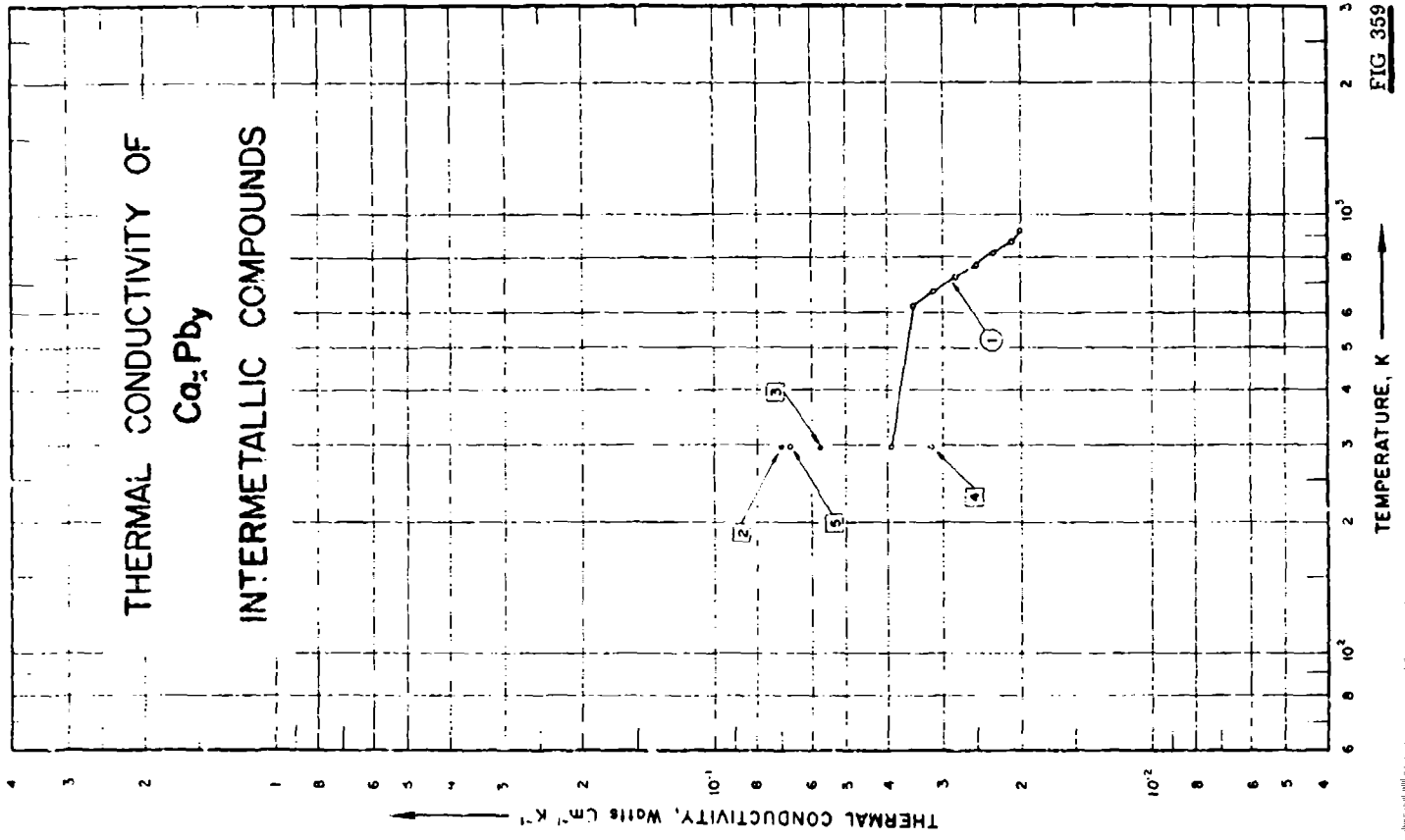


## SPECIFICATION TABLE NO. 358 THERMAL CONDUCTIVITY OF CuTe INTERMETALLIC COMPOUNDS

[For Data Reported in Figure and Table No. 358]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
1	942	L	1965	240-449	7	CdTe:3.2	Stoichiometric single crystal, p-type; prepared by melting 99.999 pure Te and Cd, mixed in a 5-6 mm dia quartz phial with 1 mm dia x 15 mm long capillary extended at bottom and annealed at 1150 C for 48 hrs; electrical resistivity reported at 228 ~ 495 K was $17.9 \times 10^4 \sim 123$ ohm cm; measured in a vacuum of $10^{-4}$ Torr.
2	942	L	1965	124-451	7	CdTe:7	Stoichiometric polycrystalline, p-type; same production method and measuring condition as the above specimen, electrical resistivity reported as 589 ~ 479 ohm cm at 134 ~ 485 K.
3	942	L	1965	168-449	7	CdTe:6	Single crystal, p-type; same production method and measuring condition as the above specimen; electrical resistivity reported as $41.3 \times 10^4 \sim 92.1$ ohm cm at 166 ~ 483 K.
4	942	L	1965	148-436	7	CdTe:9	Similar to the above specimen except electrical resistivity reported as $45.5 \times 10^4 \sim 0.273 \times 10^4$ ohm cm at 167 ~ 441 K.
5	942	L	1965	189-443	7	CdTe:8.2	p-type single crystal, same production method and measuring condition as the above specimen; electrical resistivity reported as $1.8 \times 10^4 \sim 337$ ohm cm at 190 ~ 444 K.
6	942	L	1965	348-441	7	CdTe:10.2	Similar to the above specimen except electrical resistivity reported as $42.7 \times 10^4 \sim 1.18 \times 10^4$ ohm cm at 346 ~ 433 K.
7	94	L	1963	297-450		CuTe	p-type; specimen 8 mm in dia, 12-14 mm long; synthesized in evacuated quartz ampule at $10^7$ mm Hg, heated to above the melting point of the component with higher melting point for 2 hrs, then heated to the melting point of the compound for 8 hrs; annealed at 700-800 C for several hrs and then cooled to room temperature.





SPECIFICATION TABLE NO. 359 THERMAL CONDUCTIVITY OF  $\text{Ca}_x\text{Pb}_y$  INTERMETALLIC COMPOUNDS

{ For Data Reported in Figure and Table No. 359 }

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition ( weight percent), Specifications and Remarks
1	548	L	1961	298-923		$\text{Ca}_2_{10}\text{Pb}$ ; No. 7	Seebeck coefficient (25C) $177.5 \mu\text{VK}^{-1}$ ; electrical resistivity (25C) $1.17 \times 10^{-3}$ ohm cm; figure of merit (25C) $0.68 \times 10^{-4} \text{K}^{-1}$ .
2	548	L	1961	298.2		$\text{Ca}_2\text{Pb}$ ; No. 4	Seebeck coefficient (25C) $102.3 \mu\text{VK}^{-1}$ ; electrical resistivity (25C) $5.75 \times 10^{-3}$ ohm cm; figure of merit (25C) $0.258 \times 10^{-4} \text{K}^{-1}$ .
3	548	L	1961	298.2		$\text{Ca}_2_{16}\text{Pb}$ ; No. 8	Seebeck coefficient (25C) $228.5 \mu\text{VK}^{-1}$ ; electrical resistivity (25C) $1.24 \times 10^{-2}$ ohm cm; figure of merit (25C) $0.718 \times 10^{-4} \text{K}^{-1}$ .
4	548	L	1961	298.2		$\text{Ca}_2\text{Pb}$ ; No. 10	Seebeck coefficient (25C) $109.6 \mu\text{VK}^{-1}$ ; electrical resistivity (25C) $6.53 \times 10^{-3}$ ohm cm; figure of merit (25C) $0.575 \times 10^{-4} \text{K}^{-1}$ .
5	548	L	1961	298.2		$\text{Ca}_2_{18}\text{Pb}$ ; No. 12	Seebeck coefficient (25C) $84.8 \mu\text{VK}^{-1}$ ; electrical resistivity (25C) $4.28 \times 10^{-3}$ ohm cm; figure of merit (25C) $0.247 \times 10^{-4} \text{K}^{-1}$ .

DATA TABLE NO. 359 THERMAL CONDUCTIVITY OF  $\text{Ca}_3\text{P}_2$ , INTERMETALLIC COMPOUNDS  
 [Temperature, T, K; Thermal Conductivity, k, Watt  $\text{cm}^{-1}\text{K}^{-1}$ ]

T	k
<u>CURVE 1</u>	
298.2	0.0394
523.2	0.035
673.2	0.0315
723.2	0.028
773.2	0.025
823.2	0.021
873.2	0.021
923.2	0.02
<u>CURVE 2</u>	
298.2	0.0702
<u>CURVE 3</u>	
298.2	0.0573
<u>CURVE 4</u>	
298.2	0.032
<u>CURVE 5</u>	
298.2	0.0677

SPECIFICATION TABLE NO. 360 THERMAL CONDUCTIVITY OF  $\text{Ca}_2\text{Sn}$  INTERMETALLIC COMPOUNDS

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
1	548	L	1961	298.2		$\text{Ca}_2\text{Sn}$ No. 9	Synthesized; seebeck coeff. $37 \mu\text{V K}^{-1}$ at 25 C; electrical resistivity $4.61 \times 10^4 \text{ ohm cm}$ at 25 C; figure of merit $0.742 \times 10^3 \text{ K}^{-1}$ at 25 C.
2	548	L	1961	298.2		$\text{Ca}_2\text{Sn}$ No. 12	Synthesized; seebeck coeff. $17.4 \mu\text{V K}^{-1}$ at 25 C; electrical resistivity $4.29 \times 10^4 \text{ ohm cm}$ at 25 C; figure of merit $0.280 \times 10^4 \text{ K}^{-1}$ at 25 C.
3	548	L	1961	298.2		$\text{Ca}_2\text{Sn}$ No. 13	Synthesized; seebeck coeff. $17.7 \mu\text{V K}^{-1}$ at 25 C; electrical resistivity $8.38 \times 10^4 \text{ ohm cm}$ at 25 C; figure of merit $0.114 \times 10^4 \text{ K}^{-1}$ at 25 C.

DATA TABLE NO. 360 THERMAL CONDUCTIVITY OF  $\text{Ca}_2\text{Sn}$  INTERMETALLIC COMPOUNDS[Temperature, T, K; Thermal Conductivity, k,  $\text{Watt cm}^{-1}\text{K}^{-1}$ ]

T	k
<u>CURVE 1*</u>	
298.2	0.00250
<u>CURVE 2†</u>	
298.2	0.0403
<u>CURVE 3*</u>	
298.2	0.0308

No graphical presentation



## SPECIFICATION TABLE NO. 361 THERMAL CONDUCTIVITY OF CoSi INTERMETALLIC COMPOUNDS

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
1	549	1.	1959	326.2	± 10	CoSi	Single crystal, supplied by Transiron Electronics.

## DATA TABLE NO. 361 THERMAL CONDUCTIVITY OF CoSi INTERMETALLIC COMPOUNDS

[ Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>2</sup>K<sup>-1</sup> ]

T            k  
CURVE 1\*  
 326.2      0.097

\* No graphical presentation

SPECIFICATION TABLE NO. 362 THERMAL CONDUCTIVITY OF  $\text{Cu}_2\text{S}_2\text{Se}_2$  INTERMETALLIC COMPOUNDS

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)	Specifications and Remarks
1	944		1962	298.2		$\text{Cu}_2\text{S}_2\text{Se}_2$		p-type; specimen obtained by fusing ASARCO 99.999 pure elements in carbon-coated quartz tube with agitation, cooling, crushing, recasting in 6 mm uncoated quartz, and zone-leveling; electrical resistivity 1.2 - 1.8 ohm cm; melting point (with decomposition) 485 K, (measuring temperature assumed 25 C).

DATA TABLE NO. 362 THERMAL CONDUCTIVITY OF  $\text{Cu}_2\text{S}_2\text{Se}_2$  INTERMETALLIC COMPOUNDS

(Temperature, T, K; Thermal Conductivity, k, Watt  $\text{cm}^{-1}\text{K}^{-1}$ )

T            k

CURVE 1\*

298.2    0.0170

\* No graphical presentation

SPECIFICATION TABLE NO. 363 THERMAL CONDUCTIVITY OF  $\text{Cu}_3\text{Se}_2$  INTERMETALLIC COMPOUNDS

Curve No.	Ref. No	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
1	544		1962	298.2		$\text{Cu}_3\text{Se}_2$	p-type; specimen obtained by fusing ASARCO 99.995 pure elements in carbon-coated 15 mm outside dia quartz tube with agitation, cooling, crushing, recasting in 8 mm uncoated quartz, and zone-leveling; electrical resistivity 0.1 milliohm cm; melting point 355 K (decomposition), (measuring temperature assumed 25 C).

DATA TABLE NO. 363 THERMAL CONDUCTIVITY OF  $\text{Cu}_3\text{Se}_2$  INTERMETALLIC COMPOUNDS[ Temperature, T, K; Thermal Conductivity, k,  $\text{Watt cm}^{-1}\text{K}^{-1}$  ]

T k

CURVE 1\*

298.2 0.0240

No graphical presentation



## SPECIFICATION TABLE NO. 364 THERMAL CONDUCTIVITY OF GaAs INTERMETALLIC COMPOUNDS

[ For Data Reported in Figure and Table No. 364 ]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition ( weight percent).	Specifications and Remarks
1	519	L	1958	90-100	2-3	GaAs		Impurities of the order of 0.0001%; p-type polycrystalline specimen ~5 mm in dia; current concentration of the order of $10^{18}$ cm <sup>-3</sup> .

DATA TABLE NO. 364 THERMAL CONDUCTIVITY OF GaAs INTERMETALLIC COMPOUNDS  
[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup> K<sup>-1</sup>]

T	k
90	0.397
100	0.391
110	0.384
120	0.377
130	0.372
140	0.368
150	0.364
160	0.361
170	0.358
180	0.356
190	0.353
200	0.351
210	0.349
220	0.346
230	0.344
240	0.343
250	0.341
260	0.340
270	0.337
280	0.336
290	0.335
300	0.330

## SPECIFICATION TABLE NO. 365 THERMAL CONDUCTIVITY OF GeTe INTERMETALLIC COMPOUNDS

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
1	947	L	1960	300		GeTe	Pure; electrical resistivity $1.43 \times 10^{-4}$ ohm cm.
2	948		1961	300		GeTe	Specimen 6 mm in dia and 4 mm thick; prepared from semi-conducting grade Ge (electrical resistivity 40 ohm cm) supplied by Eagle Pichard Co. and semi-conducting grade Te supplied by American Smelting and Refining Co.; electrical resistivity $1.39 \mu$ ohm cm at 300 K.

## DATA TABLE NO. 365 THERMAL CONDUCTIVITY OF GeTe INTERMETALLIC COMPOUNDS

[ Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup>K<sup>-1</sup> ]

T	k
<u>CURVE 1*</u>	
300	0.0687
<u>CURVE 2*</u>	
300	0.069

\* No graphical presentation

THERMAL CONDUCTIVITY OF  
 $Au_xCu_y$   
 INTERMETALLIC COMPOUNDS

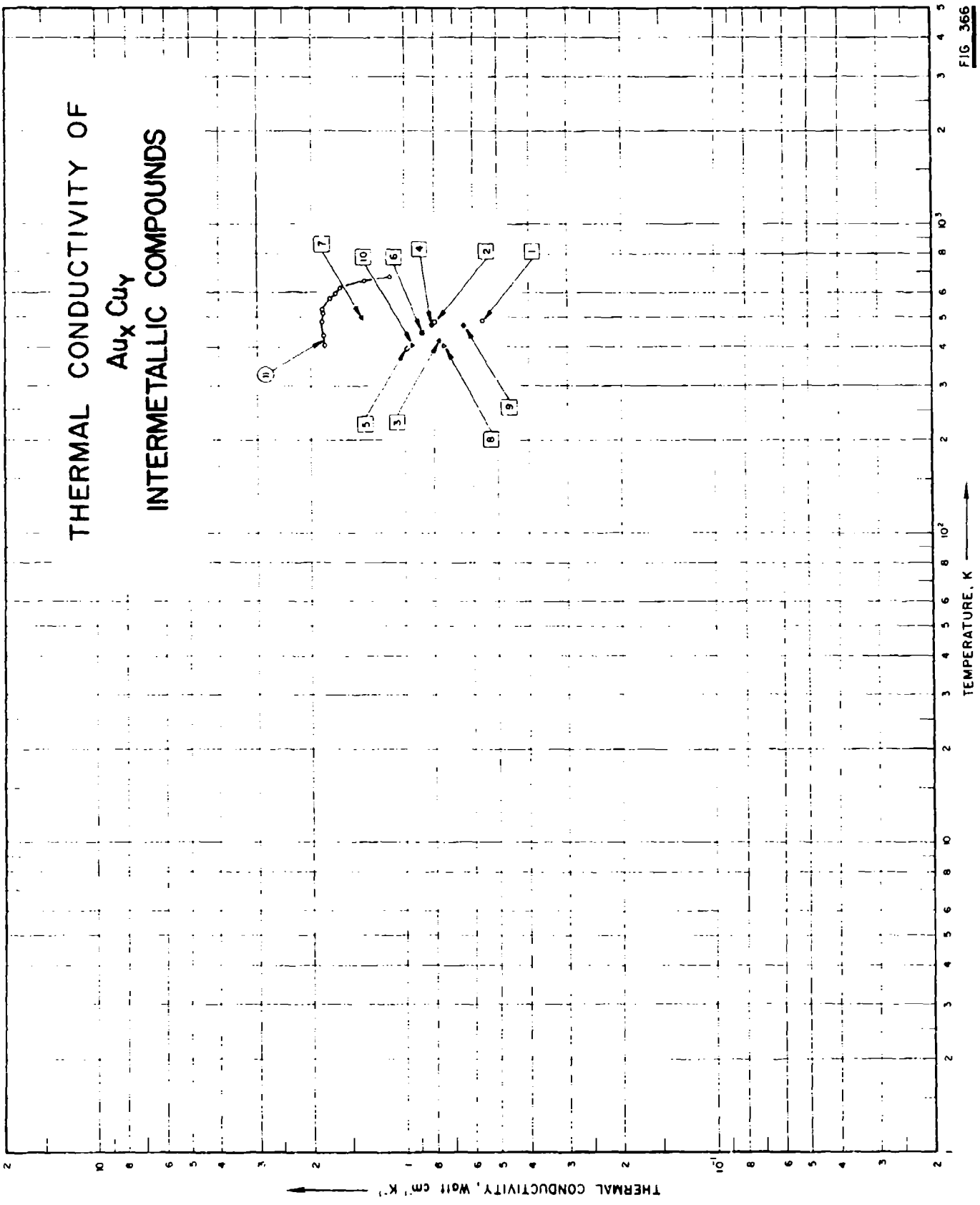


FIG. 366



SPECIFICATION TABLE NO. 366 THERMAL CONDUCTIVITY OF  $Au_xCu_y$  INTERMETALLIC COMPOUNDS

[For Data Reported in Figure and Table No. 366]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)	Specifications and Remarks
1	232	L	1957	488.7		CuAu:IV	Cast; 1.30 cm long; 0.63 cm <sup>2</sup> cross sectional area; density 18.34 g cm <sup>-3</sup> .	
2	232	L	1957	483.2		CuAu:IV	The above specimen annealed for 10 hr at 200 C.	
3	232	L	1957	420.7		CuAu:IV	The above specimen annealed for 20 hr at 200 C.	
4	232	L	1957	473.7		CuAu:IV	The above specimen annealed for 30 hr at 200 C.	
5	232	L	1957	395.2		CuAu:IV	The above specimen annealed for 40 hr at 200 C.	
6	232	L	1957	415.7		Cu <sub>2</sub> Au:II	Cast; 1.49 cm long, 0.63 cm <sup>2</sup> cross sectional area; density 15.05 g cm <sup>-3</sup> .	
7	232	L	1957	493.2		Cu <sub>2</sub> Au:II	The above specimen annealed for 10 hr at 200 C.	
8	232	L	1957	401.7		Cu <sub>2</sub> Au:II	The above specimen annealed for 20 hr at 200 C.	
9	232	L	1957	470.2		Cu <sub>2</sub> Au:II	The above specimen annealed for 30 hr at 200 C.	
10	232	L	1957	403.7		Cu <sub>2</sub> Au:II	The above specimen annealed for 40 hr at 200 C.	
11	477	L	1962	407-680		Cu <sub>2</sub> Au	0.1858 in. dia x 2.41 in. long; successively annealed at 360 C for 90 hrs, 240 C for 110 hrs, and 220 C for 600 hrs; critical temperature lies between 387.5 and 388.2 C; electrical resistivity reported as 4.2582, 4.3864, 4.8367, 5.2834, 5.6889, 6.2509, 6.6710, 7.2362, 8.2142, 9.3038, 10.6252, 10.8953, 11.3171, 12.1987, 13.6671, 14.0257, 14.0335, 14.0752, 14.1094 and 14.2959 $\mu$ ohm at 33.30, 43.74, 83.38, 124.04, 160.92, 211.71, 248.80, 278.71, 311.98, 345.78, 373.61, 377.93, 382.60, 385.80, 387.54, 388.19, 390.97, 395.25, 404.20, and 419.77 C respectively (selected from 76 points reported by the authors).	

DATA TABLE NO. 366 THERMAL CONDUCTIVITY OF  $Au_xCu_y$  INTERMETALLIC COMPOUNDS  
 (Temperature, T, K; Thermal Conductivity, k, Watt  $cm^{-1} K^{-1}$ )

T	k	T	k
<u>CURVE 1</u>			
488.7	0.561	437.4	1.849
<u>CURVE 2</u>			
483.2	0.803	445.7	1.854*
<u>CURVE 3</u>			
420.7	0.774	455.3	1.858*
<u>CURVE 4</u>			
473.7	0.816	467.1	1.854*
<u>CURVE 5</u>			
395.2	0.961	473.5	1.856*
<u>CURVE 6</u>			
445.7	0.879	482.0	1.854
<u>CURVE 7</u>			
493.2	1.339	490.3	1.862*
<u>CURVE 8</u>			
401.7	0.745	497.2	1.858*
<u>CURVE 9</u>			
470.2	0.649	506.3	1.849*
<u>CURVE 10</u>			
403.7	0.946	512.1	1.841*
<u>CURVE 11</u>			
406.8	1.833	512.2	1.845
414.5	1.841 <sup>c</sup>	514.1	1.845*
414.7	1.833 <sup>b</sup>	520.5	1.858*
420.8	1.841 <sup>c</sup>	520.6	1.862*
430.9	1.841 <sup>a</sup>	523.5	1.845*
		530.1	1.854
		538.2	1.849*
		547.2	1.841*
		555.7	1.820*
		564.0	1.778*
		572.5	1.753
		572.7	1.753*
		580.6	1.732*
		588.1	1.707*
		588.4	1.695*
		595.5	1.699
		603.3	1.665*
		609.3	1.657*
		608.7	1.657*
		614.0	1.648*
		621.2	1.632
		629.4	1.602*
		637.0	1.582*
		644.0	1.523*
		651.0	1.431*
		655.3	1.356
		662.6	1.121*
		664.1	1.121*
		670.2	1.121*
		672.2	1.121*
		674.4	1.123
		679.4	1.134*

Not shown on plot

# THERMAL CONDUCTIVITY OF HfB<sub>2</sub> INTERMETALLIC COMPOUND

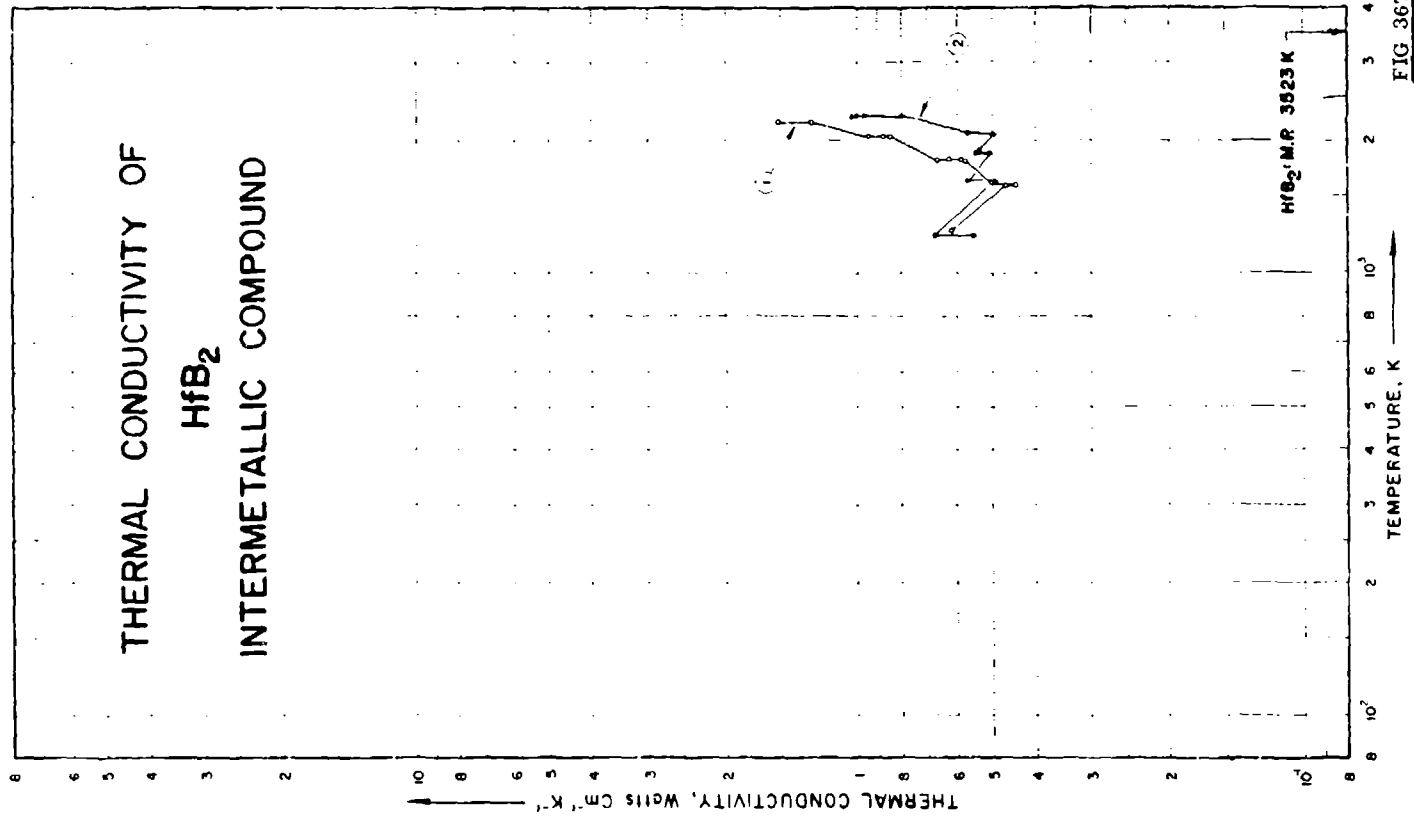


FIG. 367

SPECIFICATION TABLE NO. 367 THERMAL CONDUCTIVITY OF  $HfB_2$  INTERMETALLIC COMPOUNDS

( For Data Reported in Figure and Table No. 367 )

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition ( weight percent), Specifications and Remarks
1	544	R	1963	1267-2210	5-7.0	$HfB_2$ ; 1	Specimen 0.75 in. O.D. and 0.25 in. I.D., 0.75 in. long; heat soaked at 3200 F to 3450 F, ground and polished; specimen found cracked on post inspection.
2	544	R	1963	1233-2297	5-7.0	$HfB_2$ ; 2	Similar to the above specimen but the specimen broke during experiment.

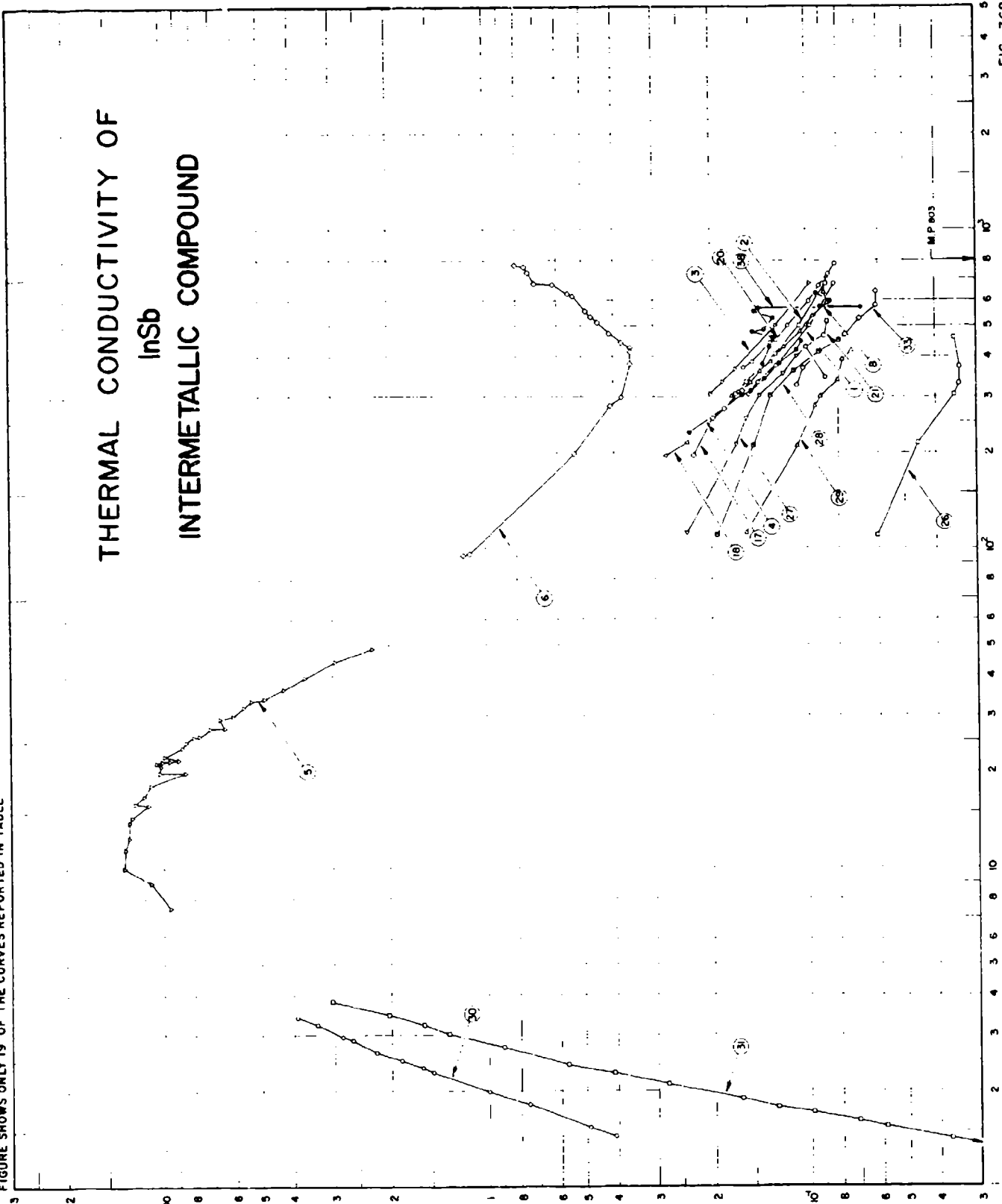
DATA TABLE NO. 367 THERMAL CONDUCTIVITY OF  $\text{HfB}_2$  INTERMETALLIC COMPOUNDS[Temperature, T, K; Thermal Conductivity, k, Watt  $\text{cm}^{-1}\text{K}^{-1}$ ]

T	k
<u>CURVE 1</u>	
1266.5	0.621
1266.5	0.617
1266.5	0.624 <sup>a</sup>
1592.1	0.467
1595.4	0.443
1599.3	0.501
1803.7	0.574
1805.4	0.585
1807.6	0.628
1808.7	0.660
2055.4	0.845
2055.9	0.874
2058.7	0.941
2205.4	1.27
2206.5	1.28 <sup>a</sup>
2210.4	1.52
<u>CURVE 2</u>	
1233.2	0.550
1233.2	0.555
1233.2	0.675
1624.3	0.492
1631.5	0.500 <sup>a</sup>
1537.6	0.568
1680.9	0.502
1881.5	0.545
1882.6	0.505
2082.1	0.495
2087.1	0.562
2087.9	0.555
2285.9	0.794
2294.3	0.956
2296.5	1.00
2297.1	1.02

Not shown on plot

# THERMAL CONDUCTIVITY OF InSb INTERMETALLIC COMPOUND

FIGURE SHOWS ONLY 19 OF THE CURVES REPORTED IN TABLE



## SPECIFICATION TABLE NO. 368 THERMAL CONDUCTIVITY OF InSb INTERMETALLIC COMPOUNDS

(For Data Reported in Figure and Table No. 368.)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)	Specifications and Remarks
1	309	C	1957	305-678	±10	InSb; A		p-type indium antimonide; single crystal; $2 \times 10^{15}$ extrinsic electrons $\text{cm}^{-3}$ , weakly degenerate over the temperature range; Fifth Brown F. H. steel used as reference material.
2	309	C	1957	305-678	±10	InSb; B		n-type, single crystal; $10^{17}$ extrinsic electrons $\text{cm}^{-3}$ , weakly degenerate over the temperature range; Fifth Brown F. H. steel used as reference material.
3	309	C	1957	305-678	±10	InSb; C		n-type, single crystal; $2 \times 10^{18}$ extrinsic electrons $\text{cm}^{-3}$ , strongly degenerate near room temperature; Fifth Brown F. H. steel used as reference material.
4	307	P	1959	232-446		InSb		Single crystal; $3 \times 10^{17}$ electrons $\text{cm}^{-3}$ , specific heat 12 cal $\text{mol}^{-1}$ .
5	303	L	1959	7.5-49		InSb		p-type single crystal; impurity concentration $2.89 \times 10^{17} \text{ cm}^{-3}$ (from thermal conductivity data); electrical conductivity $14.1 \text{ ohm}^{-1} \text{ cm}^{-1}$ at 78 K.
6	302		1954	95-773		InSb		No details reported.
7			1958	330-770		InSb		No details reported.
8	590	P	1960	300-715		InSb; IS-194		Single crystal; rectangular parallelepiped dimensions $1/2 \times 3/8 \times 3/8$ in.
9	542	C	1962	308-443	±20	InSb; E-1		n-type single crystal; donor concentration $10^{16} \text{ cm}^{-3}$ ; specimen 12.3 mm dia., 7.2 mm long, measured in vacuo of $5 \times 10^{-5}$ mm Hg.
10	542	C	1962	445-573	±20	InSb; S-1		Similar to the above specimen.
11	542	C	1962	406-575	±20	InSb; S-1		The above specimen measured in a magnetic field of 1000 gauss.
12	542	C	1962	406-575	±20	InSb; S-1		The above specimen measured in a magnetic field of 2000 gauss.
13	542	C	1962	406-575	±20	InSb; S-1		The above specimen measured in a magnetic field of 3000 gauss.
14	542	C	1962	406-575	±20	InSb; S-1		The above specimen measured in a magnetic field of 5000 gauss.
15	542	C	1962	444-575	±20	InSb; S-1		The above specimen measured in a magnetic field of 7000 gauss.
16	542	C	1962	406-575	±20	InSb; S-3		The above specimen measured in a magnetic field of 8000 gauss.
17	611	L	1959	196-667	±4	InSb; We		p-type single crystal; dislocation concentration $1.6 \times 10^{15} \text{ cm}^{-2}$ , measured in vacuo of $5 \times 10^{-5}$ torr.
18	611	L	1959	196-667	±4	InSb; Wd		n-type single crystal; dislocation concentration $1.2 \times 10^{16} \text{ cm}^{-2}$ , measured in vacuo of $5 \times 10^{-5}$ torr.
19	611	L	1959	329-526	±4	InSb; Wb		p-type single crystal; dislocation concentration $3.3 \times 10^{15} \text{ cm}^{-2}$ , measured in vacuo of $5 \times 10^{-5}$ torr.
20	825		1958	368-781		InSb		No details reported.

## SPECIFICATION TABLE NO. 365 THERMAL CONDUCTIVITY OF InSb INTERMETALLIC COMPOUNDS

For Data Reported in Figure and Table No. 368

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)	Specifications and Remarks
21	949	F	1965	345-518	12	InSb		n-type single crystal; donor concentration $5 \times 10^{16}$ cm <sup>-3</sup> ; specimen 2.51 cm in dia and 0.27 cm thick; measured in vacuum of $\approx 5 \times 10^{-5}$ torr; measured in a magnetic field of zero gauss.
22	949	F	1965	345-518	12	InSb		The above specimen measured in a magnetic field of 1000 gauss.
23	949	F	1965	345-518	12	InSb		The above specimen measured in a magnetic field of 2000 gauss.
24	949	F	1965	345-518	12	InSb		The above specimen measured in a magnetic field of 5000 gauss.
25	949	F	1965	345-518	12	InSb		The above specimen measured in a magnetic field of 8000 gauss.
26	950, 951	L	1963	110-459		InSb		Large crystals; prepared from highly pure indium and Sb-4000 antimony purified by multiple-zone recrystallization, synthesized in evacuated ( $10^{-3}$ mm Hg) quartz ampoule, slowly cooled; carrier concentration $10^{16}$ cm <sup>-3</sup> .
27	950, 951	L	1963	113-459		InSb		0.26 In <sub>2</sub> Te <sub>3</sub> ; prepared from highly pure indium, Sb-5000 antimony, and tellurium purified by multiple-zone recrystallization, synthesized in evacuated (10 <sup>-3</sup> mm Hg) quartz ampoule, slowly cooled.
28	950, 951	L	1963	111-452		InSb		0.77 In <sub>2</sub> Te <sub>3</sub> ; same fabrication method as the above specimen.
29	950, 951	L	1963	112-419		InSb		1.28 In <sub>2</sub> Te <sub>3</sub> ; same fabrication method as the above specimen.
30	952	L	1962	1.5-3.4	$\pm 2$	InSb:N		Tellurium added as impurity; n-type; carrier concentration reported at 77 K as $1.4 \times 10^{17}$ per cm <sup>3</sup> ; specimen 1.8 x 3.95 mm in. cross-section; cut from single crystals with axis normal to direction of growth; sandblasted.
31	952	L	1962	1.4-3.8	$\pm 2$	InSb:p		Germanium added as impurity; p-type; hole concentration reported at 77K as $2 \times 10^{16}$ per cm <sup>3</sup> ; specimen 2.3 x 3.55 mm in. cross-section; cut from single crystal with axis normal to direction of growth; sandblasted.
32	953	C	1965	327, 416		InSb;E-2		$5 \times 10^{18}$ Zr and $5 \times 10^{19}$ Cd impurity atoms per cm <sup>3</sup> ; p-type; single crystal; specimen 12 mm in. dia and 7 mm - 12 mm high; supplied by Exotic Materials Inc.; first set.
33	953	C	1965	325-640		InSb;E-2		Similar to the above specimen; second set.
34	953	C	1965	328-633		InSb;S-5		Similar to the above specimen except for $2 \times 10^{18}$ Zr impurity atoms per cm <sup>3</sup> ; first set.
35	953	C	1965	325-494		InSb;S-5		Similar to the above specimen; second set.
36	953	C	1965	323-634		InSb;S-4A		Similar to the above specimen except for $1.3 \times 10^{18}$ Zr impurity atoms per cm <sup>3</sup> ; first set.
37	953	C	1965	325-635		InSb;S-4A		Similar to the above specimen; second set.
38	953	C	1965	379-621		InSb;S-1		$3.32 \times 10^{18}$ Te impurity atoms per cm <sup>3</sup> ; n-type; single crystal; specimen 12 mm in. dia and 7 mm - 12 mm high; supplied by Merck and Co., Rahway, New Jersey; run 1.
39	953	C	1965	318-600		InSb;S-1		Similar to the above specimen; run 2.



## DATA TABLE NO. 368 THERMAL CONDUCTIVITY OF 1788 INTERMETALLIC COMPOUNDS

[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup> K<sup>-1</sup>]

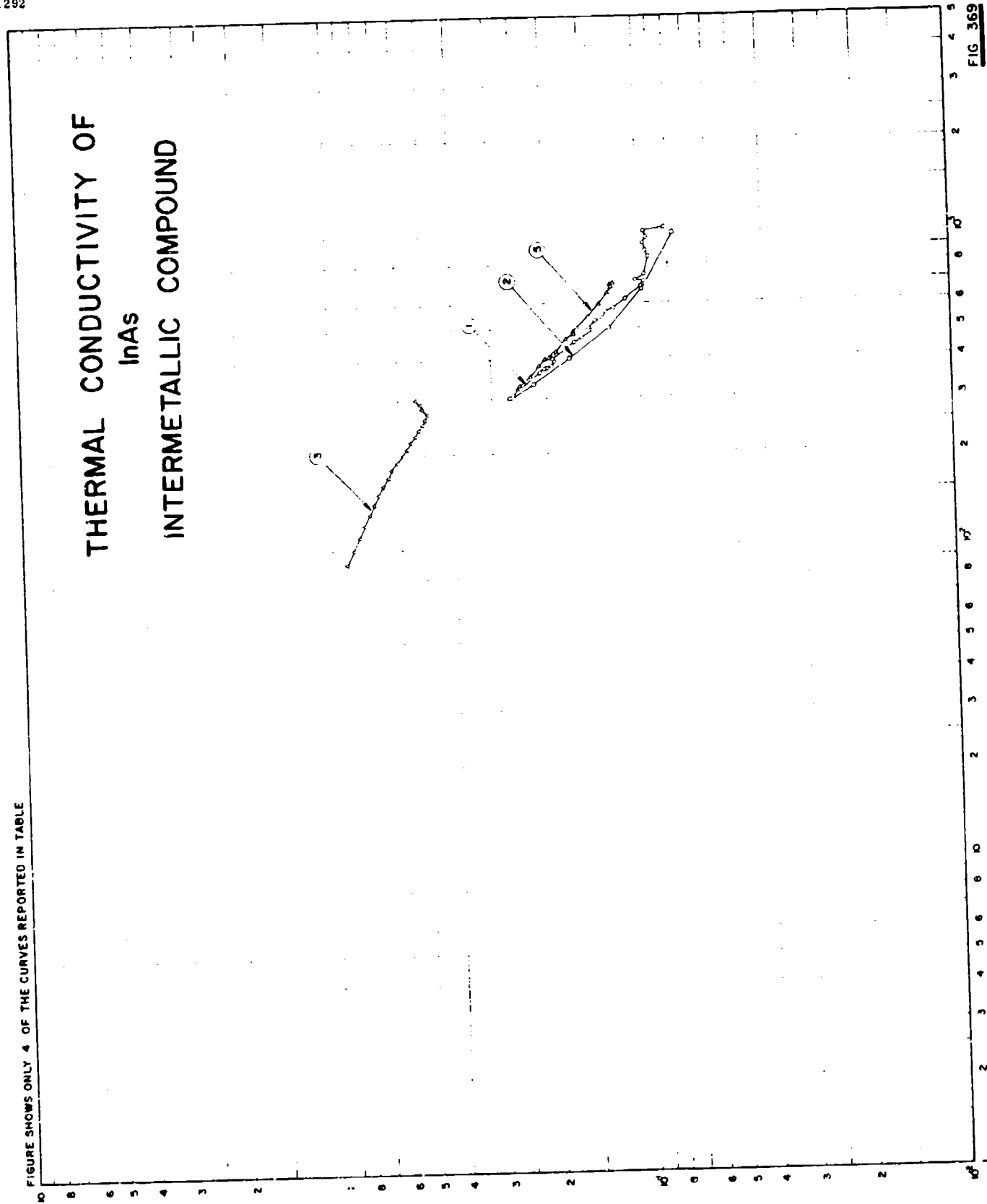
T	k	T	k	T	k	T	k	T	k	T	k	T	k
<b>CURVE 1</b>													
305	0.154	18.3	10.99	733.2	0.736	406	0.122	370.4	0.133*	345.2	0.087	113	0.238
333	0.142	20.0	8.47	761.2	0.753	473	0.0931	400.0	0.126*	428.2	0.100	212	0.167
375	0.126	20.0	10.3	773.2	0.803	575	0.0907	425.5	0.118*	468.2	0.088	258	0.156
500	0.0975	21.0	10.2	<b>CURVE 7</b>		<b>CURVE 11<sup>f</sup></b>		<b>CURVE 17 (cont.)</b>		<b>CURVE 21</b>		<b>CURVE 27</b>	
678	0.0820	21.5	10.6	300	0.150	406	0.121	476.2	0.109 <sup>g</sup>	518.2	0.086	302	0.141
<b>CURVE 2</b>													
21.7	10.1	21.7	9.52	330	0.130	473	0.0922	512.8	0.100	<b>CURVE 22*</b>		355	0.119
305	0.167	21.8	8.93	400	0.106	575	0.0903	555.6	0.0920 <sup>h</sup>	345.2	0.0867	403	0.109
333	0.155	22.5	9.90	500	0.086	<b>CURVE 12<sup>f</sup></b>		588.2	0.0900 <sup>i</sup>	428.2	0.0992	433	0.0943*
407	0.126	24.0	8.70	670	0.075	<b>CURVE 13<sup>f</sup></b>		625.0	0.0879 <sup>j</sup>	468.2	0.0871	459	0.0877*
500	0.106	25.0	8.47	770	0.075	<b>CURVE 14</b>		666.7	0.0837 <sup>k</sup>	518.2	0.0849	<b>CURVE 23</b>	
678	0.087	26.0	7.69	<b>CURVE 8</b>		406	0.120	<b>CURVE 14</b>		111	0.192	211	0.197
<b>CURVE 3</b>													
27.5	7.14	300	0.172	448	0.106	448	0.103	196.1	0.276	<b>CURVE 23*</b>		303	0.130
27.5	6.45	300	0.154 <sup>l</sup>	473	0.0913	473	0.0887	215.1	0.236	345.2	6.0863	361	0.110
29.5	6.67	330	0.149	575	0.0887	<b>CURVE 15<sup>f</sup></b>		428.2	0.0920 <sup>m</sup>	428.2	0.0999	417	0.0909
30.0	6.06	330	0.130	<b>CURVE 9</b>		<b>CURVE 16<sup>f</sup></b>		666.7	0.0900 <sup>n</sup>	468.2	0.0870	452	3.0794
305	0.201	348	0.150 <sup>o</sup>	308	0.148	406	0.117	<b>CURVE 19<sup>f</sup></b>		518.2	0.0850	<b>CURVE 29</b>	
365	0.184	32.0	5.62	406	0.105 <sup>p</sup>	448	0.103	339.0	0.149	345.2	9.0866	112	0.136
500	0.126	33.5	5.32	448	0.105 <sup>q</sup>	473	0.0891	350.9	0.142	428.2	0.0984	211	0.108
678	0.0975	34.0	4.85	473	0.0863	575	0.0879	408.2	0.121	468.2	0.0864	280	0.0943
<b>CURVE 4</b>													
39.5	3.64	427	0.118	500	0.100	<b>CURVE 17</b>		454.5	0.109	518.2	0.0856	301	0.0901
36.5	4.24	477	0.105	535	0.0952	448	0.103	512.8	0.105	192	0.0775	338	0.0960
39.5	3.64	48.5	2.94	590	0.0856	473	0.0879	526.3	0.0983	419	0.0725	398	0.0960
44.5	2.94	<b>CURVE 6</b>		635	0.0880	575	0.0843	<b>CURVE 20</b>		<b>CURVE 25*</b>		<b>CURVE 30</b>	
48.5	2.24	95.2	1.17	680	0.0830 <sup>r</sup>	<b>CURVE 18<sup>f</sup></b>		345.2	0.0863	345.2	0.0863	1.45	0.4102
<b>CURVE 5</b>													
231.9	0.233	96.2	1.12	715	0.0864	406	0.117	368	0.158	428.2	0.0977	1.55	0.4943
303.4	0.159	198.2	0.527	635	0.0880	448	0.103	345	0.148	468.2	0.0754	1.92	0.7586
313.0	0.149	279.2	0.410	680	0.0830 <sup>r</sup>	473	0.0879	442	0.130	518.2	0.0827	1.98	1.069
341.9	0.136	288.2	0.377	715	0.0864	575	0.0843	450	0.125	<b>CURVE 26</b>		2.30	1.496
381.8	0.123	378.2	0.356	<b>CURVE 10<sup>f</sup></b>		<b>CURVE 17</b>		500	0.115	110	0.0610	2.37	1.603
420.3	0.108	415.2	0.356	308	0.148	406	0.117	448	0.122	214	0.0455	2.50	1.866
446.0	0.104	426.2	0.356	443	0.108	448	0.103	473	0.0872	305	0.0253	2.65	2.339
<b>CURVE 6</b>													
95.2	1.17	443.2	0.377	443	0.108	575	0.0832	595	0.098	330	0.0340	2.89	2.636
303.4	0.159	443.2	0.377	<b>CURVE 10<sup>f</sup></b>		<b>CURVE 17</b>		667	0.091	373	0.0339	2.97	2.831
313.0	0.149	473.2	0.410	445	0.110	196.1	0.266	725	0.085	459	0.0353	3.25	3.412
341.9	0.136	513.2	0.444	473	0.0943	256.4	0.197	781	0.081	<b>CURVE 28</b>		3.43	3.908
381.8	0.123	533.2	0.469	575	0.0832	274.0	0.182	<b>CURVE 28</b>		<b>CURVE 28</b>		<b>CURVE 28</b>	
420.3	0.108	535.2	0.485	575	0.0832	312.5	0.159 <sup>s</sup>	<b>CURVE 28</b>		<b>CURVE 28</b>		<b>CURVE 28</b>	
446.0	0.104	619.2	0.552	575	0.0832	344.8	0.144 <sup>s</sup>	<b>CURVE 28</b>		<b>CURVE 28</b>		<b>CURVE 28</b>	
<b>CURVE 7</b>													
7.5	9.52	673.2	0.703	<b>CURVE 28</b>		<b>CURVE 28</b>		<b>CURVE 28</b>		<b>CURVE 28</b>		<b>CURVE 28</b>	
9.0	10.99	<b>CURVE 28</b>		<b>CURVE 28</b>		<b>CURVE 28</b>		<b>CURVE 28</b>		<b>CURVE 28</b>		<b>CURVE 28</b>	
10.0	13.3	<b>CURVE 28</b>		<b>CURVE 28</b>		<b>CURVE 28</b>		<b>CURVE 28</b>		<b>CURVE 28</b>		<b>CURVE 28</b>	
11.5	13.2	<b>CURVE 28</b>		<b>CURVE 28</b>		<b>CURVE 28</b>		<b>CURVE 28</b>		<b>CURVE 28</b>		<b>CURVE 28</b>	
12.5	12.8	<b>CURVE 28</b>		<b>CURVE 28</b>		<b>CURVE 28</b>		<b>CURVE 28</b>		<b>CURVE 28</b>		<b>CURVE 28</b>	
14.0	12.7	<b>CURVE 28</b>		<b>CURVE 28</b>		<b>CURVE 28</b>		<b>CURVE 28</b>		<b>CURVE 28</b>		<b>CURVE 28</b>	
14.5	12.5	<b>CURVE 28</b>		<b>CURVE 28</b>		<b>CURVE 28</b>		<b>CURVE 28</b>		<b>CURVE 28</b>		<b>CURVE 28</b>	
15.8	11.1	<b>CURVE 28</b>		<b>CURVE 28</b>		<b>CURVE 28</b>		<b>CURVE 28</b>		<b>CURVE 28</b>		<b>CURVE 28</b>	
16.0	12.2	<b>CURVE 28</b>		<b>CURVE 28</b>		<b>CURVE 28</b>		<b>CURVE 28</b>		<b>CURVE 28</b>		<b>CURVE 28</b>	
16.9	11.5	<b>CURVE 28</b>		<b>CURVE 28</b>		<b>CURVE 28</b>		<b>CURVE 28</b>		<b>CURVE 28</b>		<b>CURVE 28</b>	

\* Not shown on plot



THERMAL CONDUCTIVITY OF  
InAs  
INTERMETALLIC COMPOUND

FIGURE SHOWS ONLY 4 OF THE CURVES REPORTED IN TABLE



## SPECIFICATION TABLE NO. 369 THERMAL CONDUCTIVITY OF InAs INTERMETALLIC COMPOUNDS

[For Data Reported in Figure and Table No. 369]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
1	300	C	1954	301-1054	10	InAs	n-type; polycrystal; doped with extrinsic carriers of $\sim 10^{18} \text{ cm}^{-3}$ , the doping agent (sulphur added in the form of an In + S alloy; high grade dense alumina (calibrated against Armo iron) used as the comparative material; data obtained from two specimens of the same specifications.
2	300	C	1958	302-1000	10	InAs	Similar to the above specimens but doped with $5 \times 10^{16}$ extrinsic carriers.
3	519	L	1958	90-300	2-3	InAs	p-type; concentration of current carriers of the order of $10^{18} \text{ cm}^{-3}$ ; specimen $\sim 5 \text{ mm}$ in dia; synthesized from original materials in double evacuated quartz ampoules, re-melted to obtain the right form with heating rate of 100 C per hr; purified by zone melting (impurity $\sim 0.0001\%$ ).
4	578	C	1960	309-691	$\pm 5$	Indium Arsenide	Pure, polycrystal, electron concentration $\sim 3 \times 10^{16} \text{ cm}^{-3}$ ; F.H. stainless steel (checked by Armo iron) used as comparative material; data from three specimens.
5	578	C	1960	308-693	$\pm 5$	Indium Arsenide	Similar to the above specimens but sulphur doped to give an electron concentration of $\sim 10^{19} \text{ cm}^{-3}$ .



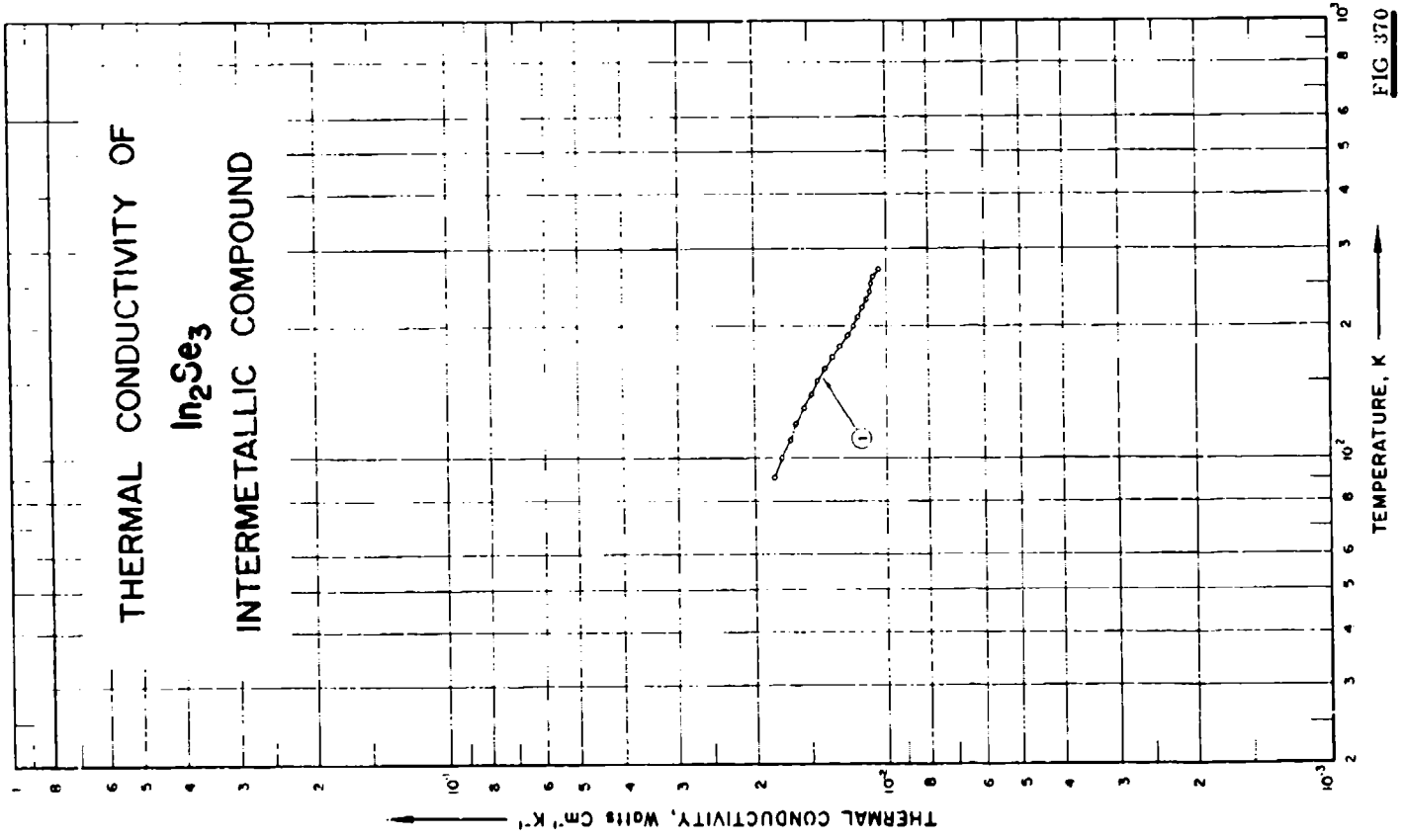


FIG. 370

SPECIFICATION TABLE NO. 370 THERMAL CONDUCTIVITY OF  $\text{In}_2\text{Se}_3$  INTERMETALLIC COMPOUNDS

[For Data Reported in Figure and Table No. 370]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
1	519		1958	90-270	~2.5	$\text{In}_2\text{Se}_3$	Polycrystal; specimen ~5.5 mm dia; prepared by zone melting from pure Se and In.

DATA TABLE NO. 370 THERMAL CONDUCTIVITY OF  $\text{In}_2\text{Se}_3$  INTERMETALLIC COMPOUNDS[Temperature, T, K; Thermal Conductivity, k, Watt  $\text{cm}^{-1} \text{K}^{-1}$ ]

T	k
<u>CURVE 1</u>	
90	0.0182
100	0.0175
110	0.0167
120	0.0162
130	0.0155
140	0.0149
150	0.0144
160	0.0138
170	0.0133
180	0.0128
190	0.0123
200	0.0120
210	0.0117
220	0.0115
230	0.0112
240	0.0110
250	0.0109
260	0.0108
270	0.0105



THERMAL CONDUCTIVITY OF  
 $\text{In}_2\text{Te}_3$   
INTERMETALLIC COMPOUND

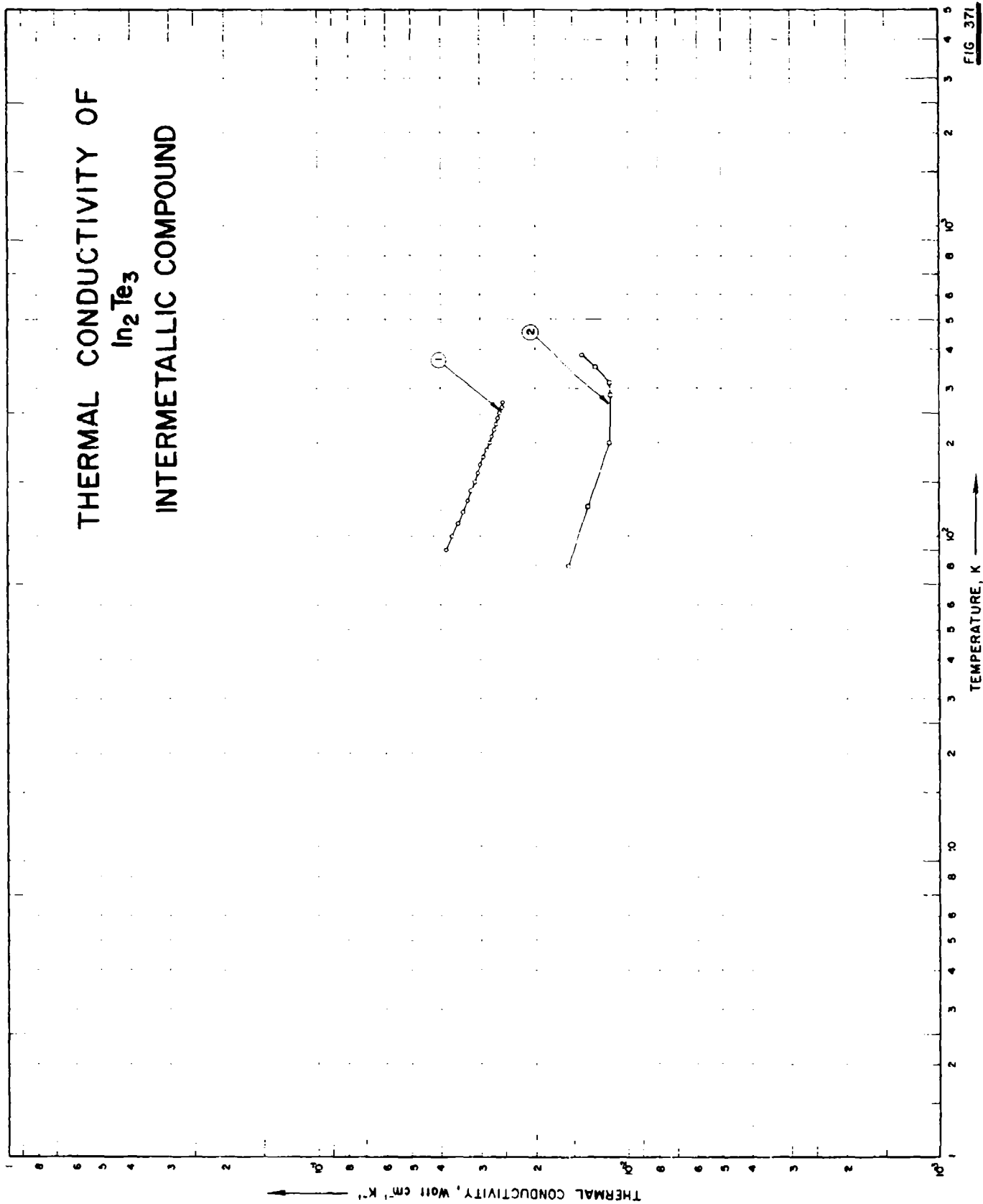


FIG. 371

TEMPERATURE, K

SPECIFICATION TABLE NO. 371 THERMAL CONDUCTIVITY OF  $\text{In}_2\text{Te}_3$  INTERMETALLIC COMPOUNDS

[For Data Reported in Figure and Table No. 371]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
1	519		1954	90-270	2-3	$\text{In}_2\text{Te}_3$	Polycrystal; zone purified $10^{-4}$ carriers $\text{cm}^{-3}$ ; specimen $\sim 5.5$ mm dia.
2	825		1958	80-385		$\text{In}_2\text{Te}_3$	Zone refined.

DATA TABLE NO. 371 THERMAL CONDUCTIVITY OF  $\text{In}_2\text{Te}_3$  INTERMETALLIC COMPOUNDS[Temperature, T, K; Thermal Conductivity, k, Watt  $\text{cm}^{-1}\text{K}^{-1}$ ]

T	k
<u>CURVE 1</u>	
90	0.0385
100	0.0368
110	0.0351
120	0.0339
130	0.0328
140	0.0320
150	0.0312
160	0.0305
170	0.0290
180	0.0283
190	0.0286
200	0.0280
210	0.0276
220	0.0271
230	0.0267
240	0.0264
250	0.0260
260	0.0256
270	0.0254
<u>CURVE 2</u>	
80	0.0155
125	0.0135
200	0.0116
286	0.0115
313	0.0116
351	0.0127
385	0.0140

THERMAL CONDUCTIVITY OF  
LaSe  
INTERMETALLIC COMPOUND

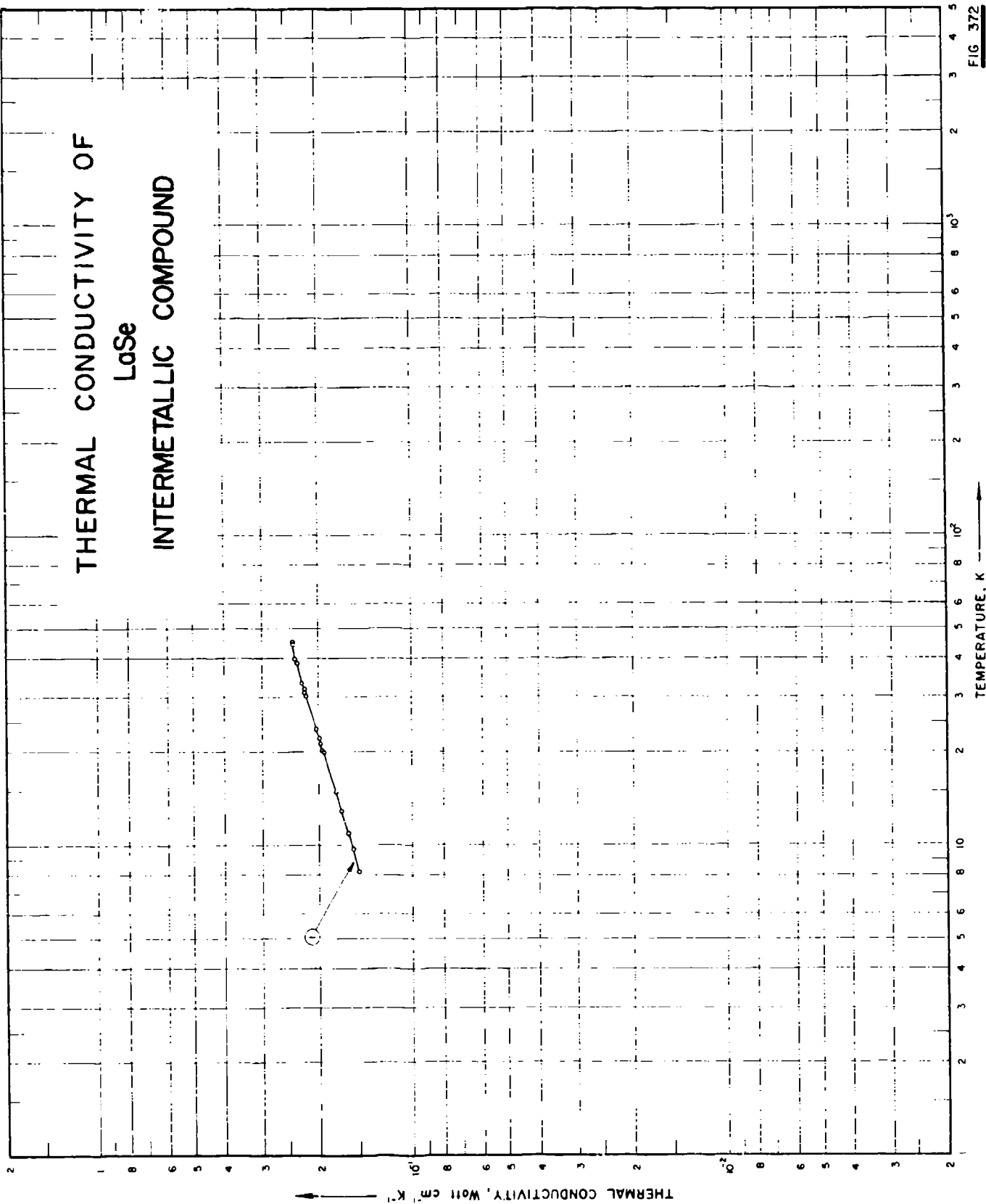


FIG 372

SPECIFICATION TABLE NO. 372 THERMAL CONDUCTIVITY OF  $\text{LaSc}$  INTERMETALLIC COMPOUNDS

For Data Reported in Figure and Table No. 372j

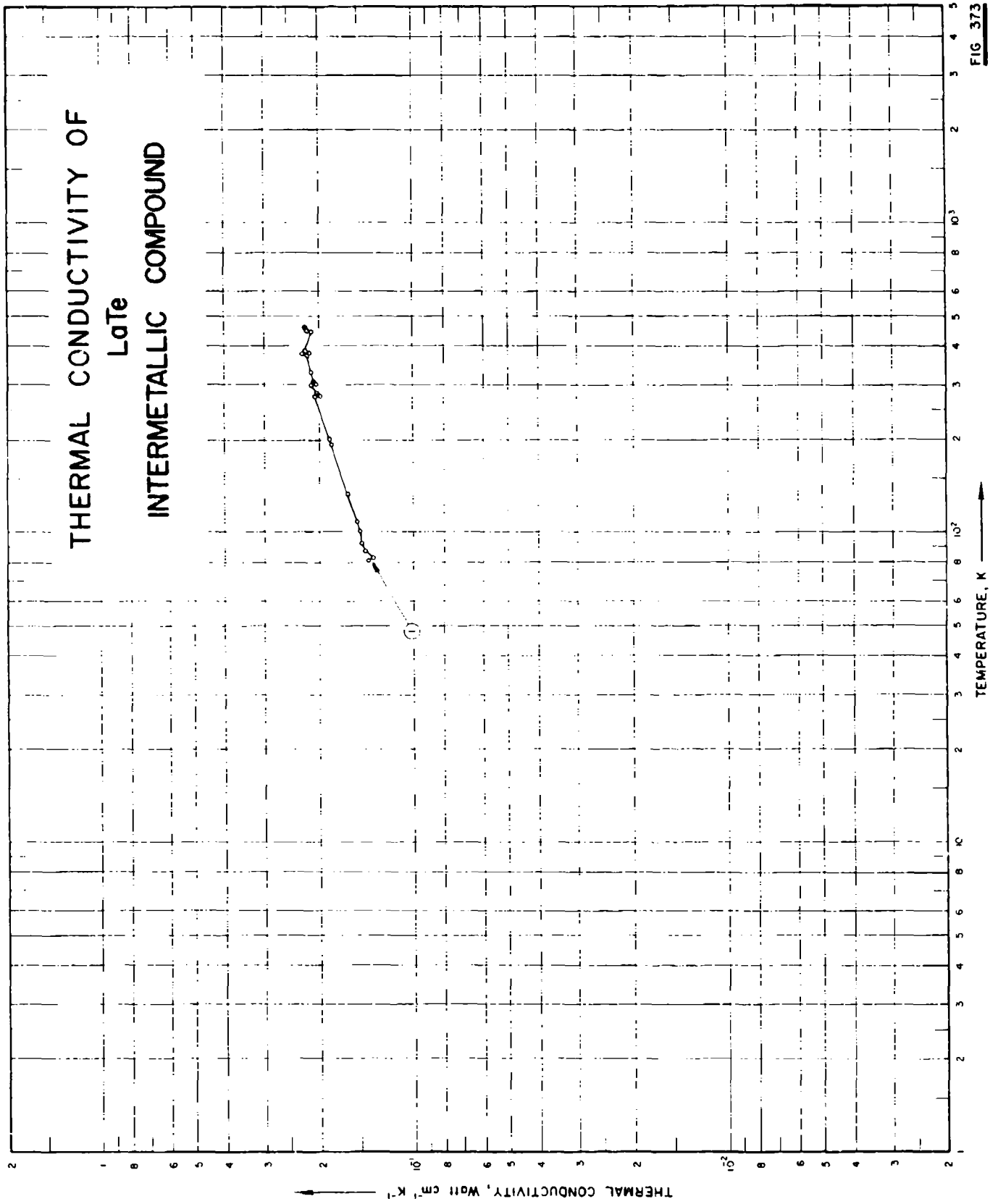
Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
1	932, 933	I.	1966	82-419	±5 - 5.5	$\text{LaSc}$	$\text{NaCl}$ type compound with ionic-metallic type bending; prepared by pressing powders of the compound under a pressure of about $800 \text{ kg cm}^{-2}$ , sintering in a vacuum of $\sim 10^{-5}$ Torr for 1 to 2 hrs. at 1600 to 1800 C; electrical resistivity reported as 2.12 ~ 5.63 $\mu\text{ohm cm}$ in the range 5. - 461 K; measured in a vacuum of $10^{-4}$ - $10^{-5}$ mm Hg.

## DATA TABLE NO. 372 THERMAL CONDUCTIVITY OF LaSe INTERMETALLIC COMPOUNDS

[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup> K<sup>-1</sup>]

T	k
82	0.150
97	0.156
109	0.161
128	0.169
148	0.176
199	0.192
201	0.195
213	0.197
221	0.199
236	0.204
302	0.220
310	0.223
318	0.223
331	0.228
389	0.235
399	0.239
445	0.241
449	0.241

CURVE 1



SPECIFICATION TABLE NO. 373 THERMAL CONDUCTIVITY OF  $\text{LaTe}$  INTERMETALLIC COMPOUNDS

[For Data Reported in Figure and Table . 373]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
1	932, 933	L	1966	81-457	$\pm 3-5$	$\text{LaTe}$	NaCl type compared with ionic-metallic type bending; prepared by pressing powders of the compound under a pressure of about $8000 \text{ kg cm}^{-2}$ , sintering in a vacuum of $\sim 10^{-4}$ Torr for 1 to 2 hrs at 1600 to 1800 C; electrical resistivity reported as $1.70 \sim 5.29 \mu \text{ ohm cm}$ in the range $14 \sim 463 \text{ K}$ ; measured in vacuum of $10^{-4} \sim 10^{-5} \text{ mm Hg}$ .



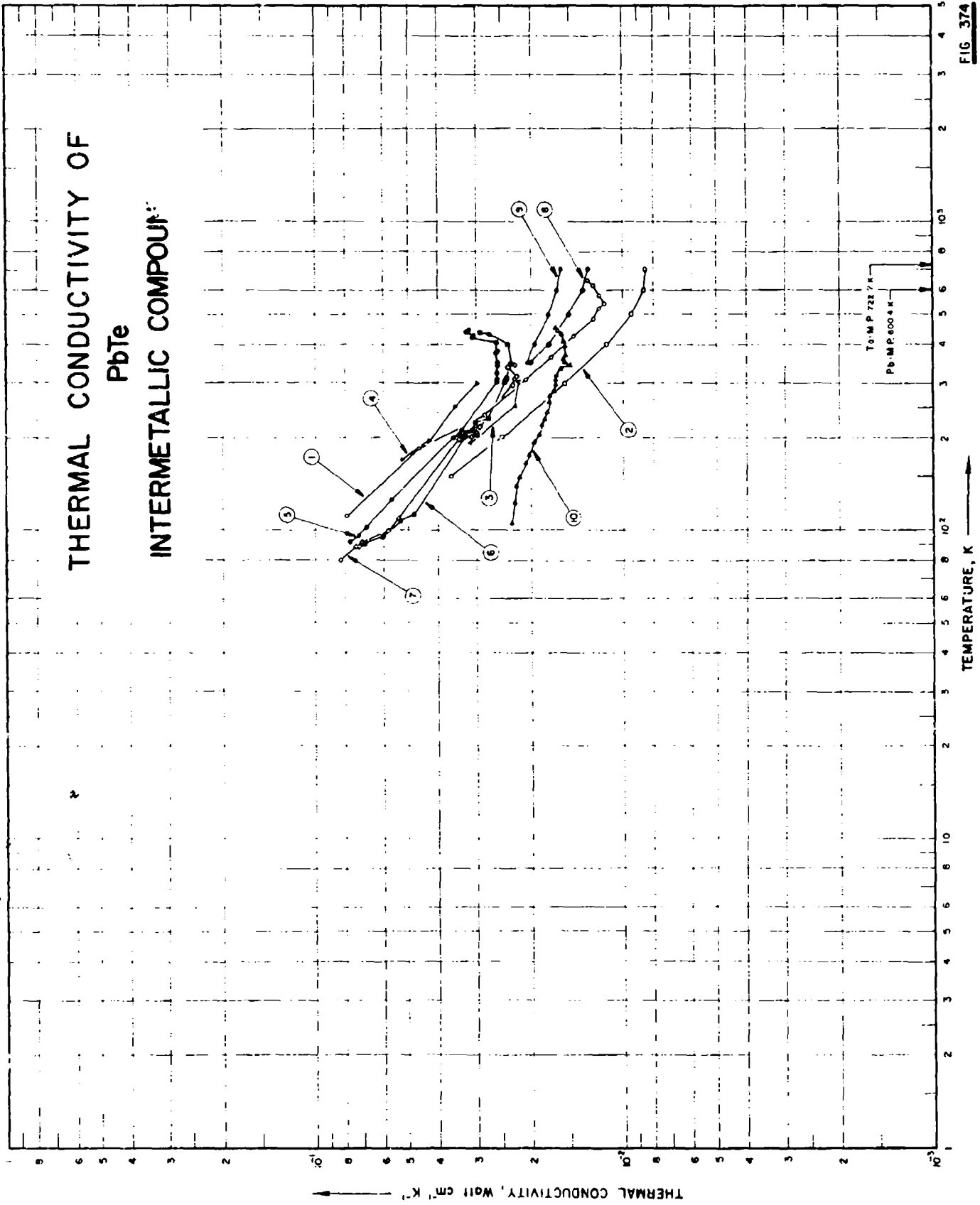
## DATA TABLE NO. 373 THERMAL CONDUCTIVITY OF LaTe INTERMETALLIC COMPOUNDS

[Temperature, T, K; Thermal Conductivity,  $k$ , Watt  $\text{cm}^{-1} \text{K}^{-1}$ ]

T	$k$
81	0.141
83	0.137
87	0.144
92	0.147
101	0.149
108	0.152
133	0.162
192	0.184
200	0.186
275	0.207
276	0.199
279	0.201
281	0.205
299	0.213
301	0.206
301	0.208
308	0.210
328	0.213
374	0.220
376	0.228
378	0.215
383	0.223
447	0.214
449	0.220
453	0.222
457	0.223

CURVE 1

**THERMAL CONDUCTIVITY OF  
PbTe  
INTERMETALLIC COMPOUND**



SPECIFICATION TABLE NO. 374 THERMAL CONDUCTIVITY OF  $PbTe$  INTERMETALLIC COMPOUNDS

(For Data Reported in Figure and Table No. 374)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
1	307	P	1959	112-647		PbTe	Single crystal; hole concentration $2 \times 10^{19} \text{ cm}^{-3}$ ; thermal conductivity calculated from measurements of thermal diffusivity using specific heat values of Parkinson and Quarrington at temperatures higher than room condition, and using $12 \text{ cal mole}^{-1}$ for temperatures less than room condition.
2	380	P	1958	150-700		PbTe	Data determined by the same method (improved) as the above.
3	381, 329	L	1957	193-302		PbTe	Polycrystal; electrical conductivity at room temperature, $\sigma = 60 \text{ ohm}^{-1} \text{ cm}^{-1}$ .
4	381, 329	L	1957	170-300		PbTe	Similar to the above specimen but $\sigma = 1740 \text{ ohm}^{-1} \text{ cm}^{-1}$ .
5	381, 329	L	1957	92-446		PbTe	Similar to the above specimen but $\sigma = 1200 \text{ ohm}^{-1} \text{ cm}^{-1}$ .
6	381, 329	L	1957	90-440		PbTe	Similar to the above specimen but $\sigma = 430 \text{ ohm}^{-1} \text{ cm}^{-1}$ .
7	381, 329	L	1957	79-344		PbTe	Similar to the above specimen but $\sigma = 60 \text{ ohm}^{-1} \text{ cm}^{-1}$ .
8	548	L	1961	350-700		PbTe	p-type; density $8.15 \text{ g cm}^{-3}$ .
9	548	L	1961	350-700		PbTe	n-type; density $8.15 \text{ g cm}^{-3}$ .
10	554	L	1966	106-451		PbTe	p-type stoichiometric single crystal; prepared by melting 99.999 pure Pb and Te in a 6 mm dia quartz phial (to which a 1 mm dia x 15 mm long capillary was attached) under a vacuum of $10^{-2}$ Torr at 1000 C for 100 hrs, annealed at 200 C for 4 to 8 hrs; electrical resistivity reported as 0.6114, 0.0123, 0.0142, 0.0160, 0.0187, 0.0230, 0.0270, 0.0324, 0.0476, 0.0622, 0.0733, 0.0863, 0.119, 0.146, 0.174, 0.220, 0.277, 0.312, 0.365, 0.407, 0.406, and 0.219 ohm cm at 113, 122, 132, 153, 175, 191, 211, 234, 275, 301, 313, 336, 357, 382, 394, 415, 450, 478, 493, 508, 658, and 794 K respectively.



SPECIFICATION TABLE NO. 375 THERMAL CONDUCTIVITY OF  $Mg_2Sb_2$  INTERMETALLIC COMPOUNDS

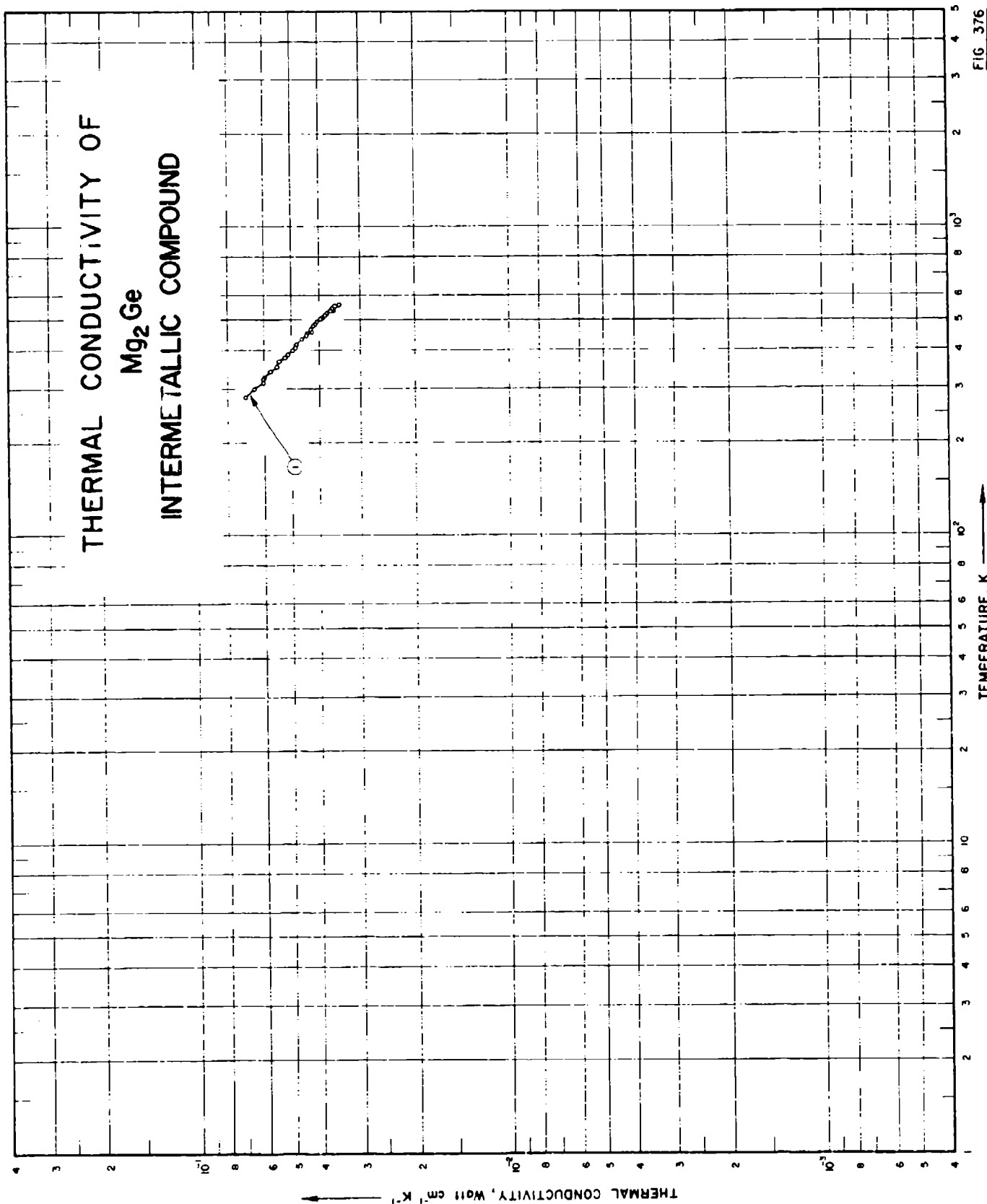
Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition ( weight percent). Specifications and Remarks
1	547		1948	333.2	± 6	$Mg_2Sb_2$ ; 1	Specimen prepared by fusing in an inert atmosphere, Mg (electrical conductivity $\sigma$ , $22.4 \times 10^4 \text{ ohm}^{-1}\text{cm}^{-1}$ at 20 C and thermal conductivity $k$ , $1.548 \text{ watt cm}^{-1}\text{K}^{-1}$ at 25 C) and Sb ( $\sigma = 2.57 \times 10^4 \text{ ohm}^{-1}\text{cm}^{-1}$ at 20 C and $k = 0.184 \text{ watt cm}^{-1}\text{K}^{-1}$ at 25 C); annealed at 500-600 C for 10-12 hrs; electrical conductivity $0.77 \times 10^4 \text{ ohm}^{-1}\text{cm}^{-1}$ at 20 C.
2	547		1948	333.2	± 6	$Mg_2Sb_2$ ; 2	As above except the electrical conductivity, $0.36 \times 10^4$ at 20 C.
3	547		1948	333.2	± 6	$Mg_2Sb_2$ ; 3	As above except the electrical conductivity, $0.30 \times 10^4$ at 20 C.
4	547		1949	333.2	± 6	$Mg_2Sb_2$ ; 4	As above except the electrical conductivity, $0.22 \times 10^4$ at 20 C.

DATA TABLE NO. 375 THERMAL CONDUCTIVITY OF  $Mg_2Sb_2$  INTERMETALLIC COMPOUNDS[ Temperature, T, K; Thermal Conductivity, k, Watt  $\text{cm}^{-1}\text{K}^{-1}$  ]

T	k
<u>CURVE 1<sup>a</sup></u>	
333.2	0.0502
<u>CURVE 2<sup>a</sup></u>	
333.2	0.0360
<u>CURVE 3<sup>a</sup></u>	
333.2	0.0209
<u>CURVE 4<sup>a</sup></u>	
333.2	0.0109

<sup>a</sup> No graphical presentation

THERMAL CONDUCTIVITY OF  
 $Mg_2Ge$   
INTERMETALLIC COMPOUND



SPECIFICATION TABLE NO. 376 THERMAL CONDUCTIVITY OF Mg<sub>2</sub>Ge INTERMETALLIC COMPOUNDS

[For Data Reported in Figure and Table No. 376]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
1	955	I.	1963	280-561	± 6	Mg <sub>2</sub> Ge	Stoichiometric composition; polycrystalline; 2.34 cm <sup>2</sup> x 1.61 cm long; prepared by fusion of the transistor grade Ge obtained from Eagle Picher Company and 99.99+ pure sublimed Mg from Dow Metal Products Company, materials weighed and loaded into a graphite crucible, put into a stainless steel tube, evacuated, heated to 500-600 C in an induction furnace, cooled back to room temperature, filled the system with argon to a pressure of 20 psia, heated rapidly to 30-40 C above the melting temperature of compound, transferred to a resistance furnace (preheated to the same temperature), soaked for about 5 min, placed the system near the end of the furnace which was then cooled at 15 C hr <sup>-1</sup> to a temperature 15 C below the freezing temperature, specimen held at 30 C below the freezing temperature for about 12 hrs, then cooled to room temperature at about 100 C hr <sup>-1</sup> .

DATA TABLE NO. 376 THERMAL CONDUCTIVITY OF Mg<sub>2</sub>Ge INTERMETALLIC COMPOUNDS[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup> K<sup>-1</sup>]

T	k
280	0.0699
299	0.0646
312	0.0608
324	0.0608
326	0.0601
340	0.0580
352	0.0551
366	0.0543
377	0.0525
386	0.0512
398	0.0498
409	0.0488
417	0.0484
431	0.0461
442	0.0448
450	0.0441
458	0.0427
467	0.0430
478	0.0422
484	0.0413
491	0.0410
502	0.0398
509	0.0393
517	0.0383
524	0.0379
534	0.0369
540	0.0362
547	0.0362
553	0.0356
561	0.0348





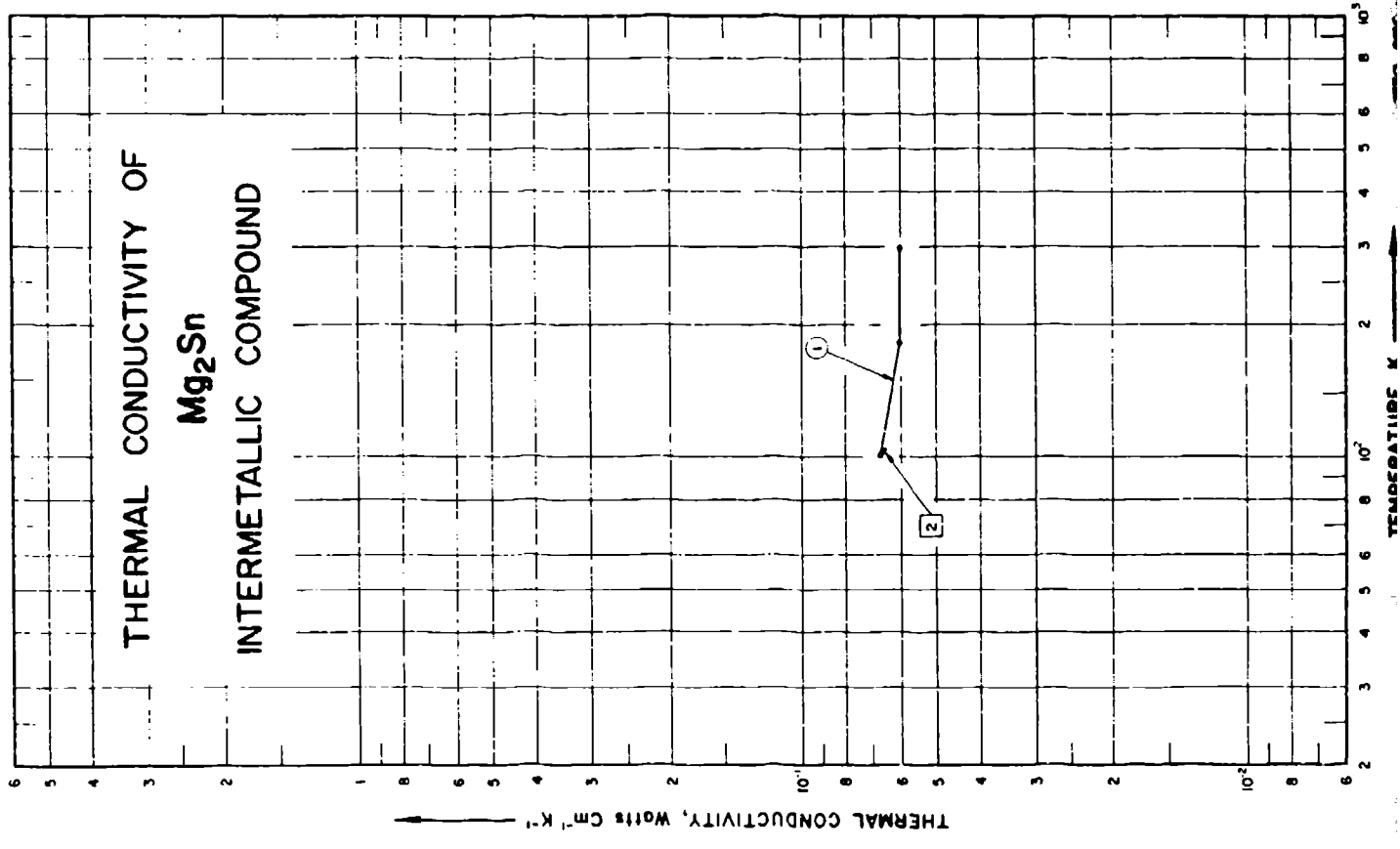
SPECIFICATION TABLE NO. 377 THERMAL CONDUCTIVITY OF  $Mg_2Si$  INTERMETALLIC COMPOUND

[For Data Reported in Figure and Table No. 377]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)	Specifications and Remarks
1	955	L	1963	281-580	± 6	$Mg_2Si$		Stoichiometric composition; polycrystalline; 1.57 cm <sup>2</sup> x 1.61 cm long; prepared by fusion of the transistor grade Si from Allegheny Electronic Chemicals Company and 99.99+ pure sublimed Mg from Dow Metal Products Company. materials weighed (with 1% excess of Si) and loaded into a graphite crucible, put into a stainless steel tube, evacuated, heated to 500 ~ 600 C in an induction furnace, cooled back to room temperature, filled the system with argon to a pressure of 20 psia, heated rapidly to 30 ~ 40 C above the melting temperature of compound, transferred to a resistance furnace (preheated to the same temperature), soaked for about 5 min, placed the system near the end of the furnace, which was then cooled at 15 C hr <sup>-1</sup> to a temperature 15 C below the freezing temperature, specimen held at 30 C below the freezing temperature for about 12 hrs, then cooled to room temperature at about 100 C hr <sup>-1</sup> .

DATA TABLE NO. 377 THERMAL CONDUCTIVITY OF Mg<sub>2</sub>Si INTERMETALLIC COMPOUNDS[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup> K<sup>-1</sup>]

T	k
281	0.0836
294	0.0779
307	0.0793
317	0.0747
328	0.0791
339	0.0704
349	0.0680
357	0.0666
365	0.0647
373	0.0633
381	0.0610
389	0.0596
398	0.0577
403	0.0580
409	0.0539
420	0.0559
427	0.0542
433	0.0532
462	0.0507
469	0.0507
479	0.0491
489	0.0494
496	0.0480
504	0.0472
511	0.0454
519	0.0451
529	0.0451
534	0.0431
541	0.0414
548	0.0420
552	0.0412
557	0.0418
565	0.0411
571	0.0401
575	0.0394
580	0.0398



SPECIFICATION TABLE NO. 378 THERMAL CONDUCTIVITY OF Mg<sub>2</sub>Sn INTERMETALLIC COMPOUNDS

[ For Data Reported in Figure and Table No. 378 ]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition ( weight percent), Specifications and Remarks
1	542	C	1962	101-298	±20	P1-2; Mg <sub>2</sub> Sn	n-type single crystal; specimen in circular form 12.3 mm in dia and 7.2 mm long; measured in a pressure of 3 x 10 <sup>-5</sup> mm Hg; Inconel 702 used as comparative material.
2	542	C	1962	103	±20	P1-2; Mg <sub>2</sub> Sn	The above specimen measured with magnetic induction of 8000 gauss; Inconel 702 used as comparative material.

DATA TABLE NO. 378 THERMAL CONDUCTIVITY OF  $Mg_2Sn$  INTERMETALLIC COMPOUNDS[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup> K<sup>-1</sup>]

T	k
<u>CURVE 1</u>	
101	0.0671
181	0.0697
298	0.0604
<u>CURVE 2</u>	
103	0.066

## SPECIFICATION TABLE NO. 379 THERMAL CONDUCTIVITY OF HgSe INTERMETALLIC COMPOUNDS

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
1	368	C.I.	1954	298.2	0.2-3.0	Hg Se	Electrical conductivity 1840 and 1870 $\text{ohm}^{-1}\text{cm}^{-1}$ at 25 and $\infty$ respectively.

## DATA TABLE NO. 379 THERMAL CONDUCTIVITY OF HgSe INTERMETALLIC COMPOUNDS

[Temperature, T, K; Thermal Conductivity, k,  $\text{Watt cm}^{-1}\text{K}^{-1}$ ]

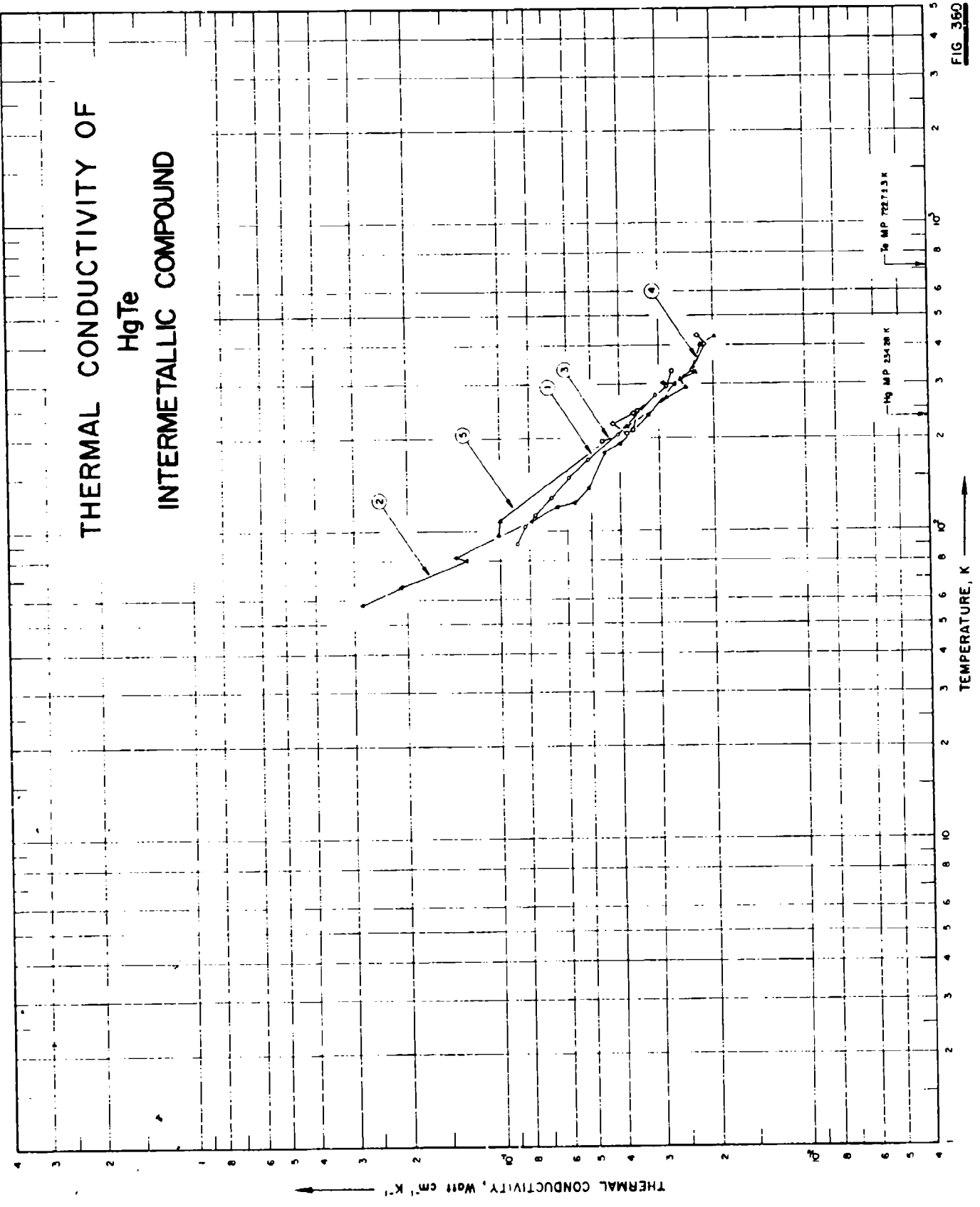
T k

CURVE 1\*

298.2 0.0176

\* No graphical presentation

THERMAL CONDUCTIVITY OF  
HgTe  
INTERMETALLIC COMPOUND





## SPECIFICATION TABLE NO. 380 THERMAL CONDUCTIVITY OF HgTe INTERMETALLIC COMPOUNDS

[ For Data Reported in Figure and Table No. 380 ]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
1	520	L	1960	91-276	±5	HgTe; No. 18	Single crystal; specimen $\sim 0.1 \times 0.1 \times 0.1$ in.
2	915	L	1958	58-302		HgTe	p-type polycrystalline specimen $2 \times 4 \times 10$ mm, prepared from 99.99% Hg and 99.999% Te; supplied by American Smelting and Refining Co.; acceptor concentration about $1.5 \times 10^{18} \text{ cm}^{-3}$ ; melting point $\sim 680$ C.
3	957, 958	L	1965	196-330	6		p-type; impurity concentration $\sim 3 \times 10^{17} \text{ cm}^{-3}$ ; measured in a vacuum of $\sim 10^{-4}$ mm Hg.
4	957, 958	L	1965	207-431	6		p-type; impurity concentration $\sim 5 \times 10^{17} \text{ cm}^{-3}$ ; measured in a vacuum of $\sim 10^{-4}$ mm Hg.
5	937, 958	L	1965	97-427	6		p-type; impurity concentration $\sim 10^{19} \text{ cm}^{-3}$ ; measured in a vacuum of $\sim 10^{-4}$ mm Hg.

## DATA TABLE NO. 380 THERMAL CONDUCTIVITY OF HfTe INTERMETALLIC COMPOUNDS

[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup> K<sup>-1</sup>]

T	k	T	k
<u>CURVE 1</u>		<u>CURVE 5</u>	
90.9	0.0879	97	0.1012
103.1	0.0828	108	0.1001
113.4	0.0763	219	0.0380
128.2	0.0678	271	0.0282
148.6	0.0592	292	0.0244
170.9	0.0512	309	0.0254
206.2	0.0406	328	0.0227
276.2	0.0309	337	0.0230
		402	0.0219
		427	0.0197
<u>CURVE 2</u>			
58	0.283		
66	0.210		
80	0.129		
82	0.140		
108	0.0790		
120	0.0650		
123.5	0.0570		
139.5	0.0510		
180	0.0450		
192	0.0400		
239	0.0325		
300	0.0265		
301.5	0.0290		
<u>CURVE 3</u>			
196	0.0456		
212	0.0364		
246	0.0353		
295	0.0282		
330	0.0271		
<u>CURVE 4</u>			
207	0.0380		
223	0.0421		
241	0.0361		
291	0.0282*		
334	0.0233		
404	0.0213		
431	0.0224		

\* Not shown on plot

FIGURE SHOWS ONLY 5 OF THE CURVES REPORTED IN TABLE

# THERMAL CONDUCTIVITY OF $\text{MoSi}_2$ INTERMETALLIC COMPOUND

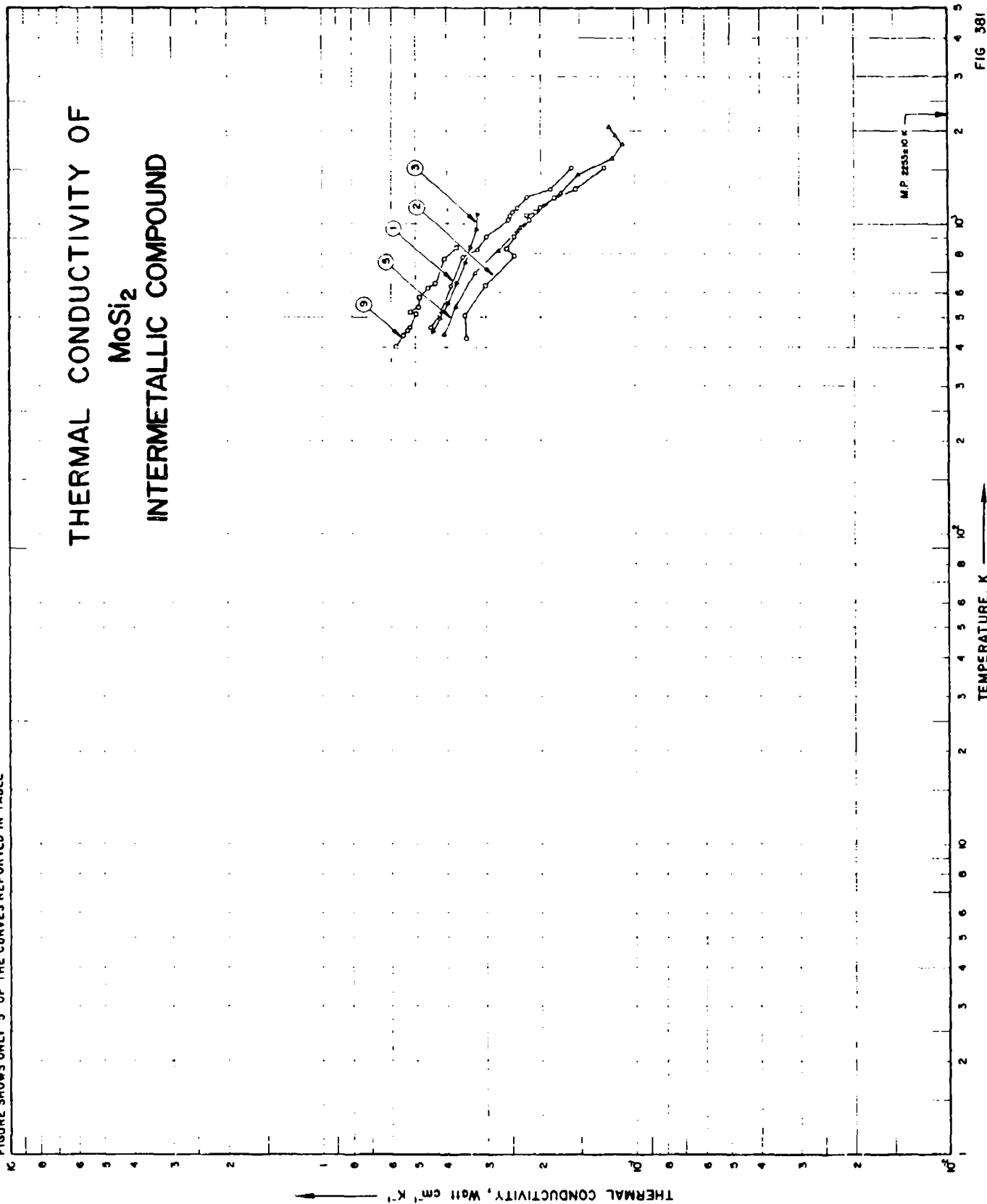


FIG 391

SPECIFICATION TABLE NO. 381 THERMAL CONDUCTIVITY OF MoSi<sub>2</sub> INTERMETALLIC COMPOUNDS

[For Data Reported in Figure and Table No. 381]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
1	575		1954	424-1517		MoSi <sub>2</sub>	Supplied by General Electric Co.; run No. 1.
2	575		1954	426-1513		MoSi <sub>2</sub>	The above specimen; run No. 2.
3	576	C	1954	449-1074	<5	MoSi <sub>2</sub> ; 1-4c	Hot pressed; 0.5 in. dia; supplied by NACA; density 6.08 g cm <sup>-3</sup> ; stainless steel used as standard.
4	576	C	1954	471-1097	<5	MoSi <sub>2</sub> ; 2-4c	Similar to the above except density 6.02 g cm <sup>-3</sup> .
5	614	R	1961	437-2058	<5	MoSi <sub>2</sub>	Slip cast; 1 in. dia, 1 in. thick; density 5.80 g cm <sup>-3</sup> ; measured in a He atmosphere.
6	576	C	1954	449-1074	<5	MoSi <sub>2</sub> ; 1-4c	Molybdenum disilicide; 62.0 Mo, 36.3 Si, 1.0 Fe, and 0.8 O; furnished by NACA; hot-pressed; density 6.08 g cm <sup>-3</sup> (97.5% of theoretical density); stainless steel used as comparative material.
7	576	C	1954	471-1097	<5	MoSi <sub>2</sub> ; 2-4c	Similar to the above specimen except density 6.02 g cm <sup>-3</sup> (96.5% of theoretical density).
8	571	L	1954	607-1062	10	MoSi <sub>2</sub> ; 5-4c	Molybdenum disilicide; 62.0 Mo, 36.3 Si, 0.8 Fe, 0.5 O, 0.34 N, and 0.17 C; specimen 3.5 in. long and 0.5 in. in dia; supplied by Lewis Flight Propulsion Lab; density 6.12 g cm <sup>-3</sup> (98% of theoretical density).
9	572	C	1950	404-840		MoSi <sub>2</sub>	Molybdenum disilicide; 62.4-62.8 Mo, 36.6-36.7 Si, and 0.5-1.0 Te; estimated composition; specimen 0.5 in. in dia and 7 in. long; hot-pressed from nonuniform powder and etched with a solution containing 17% HCl and 8% HNO <sub>3</sub> for 70 min; density ~5.91 g cm <sup>-3</sup> ; nickel used as comparative material.

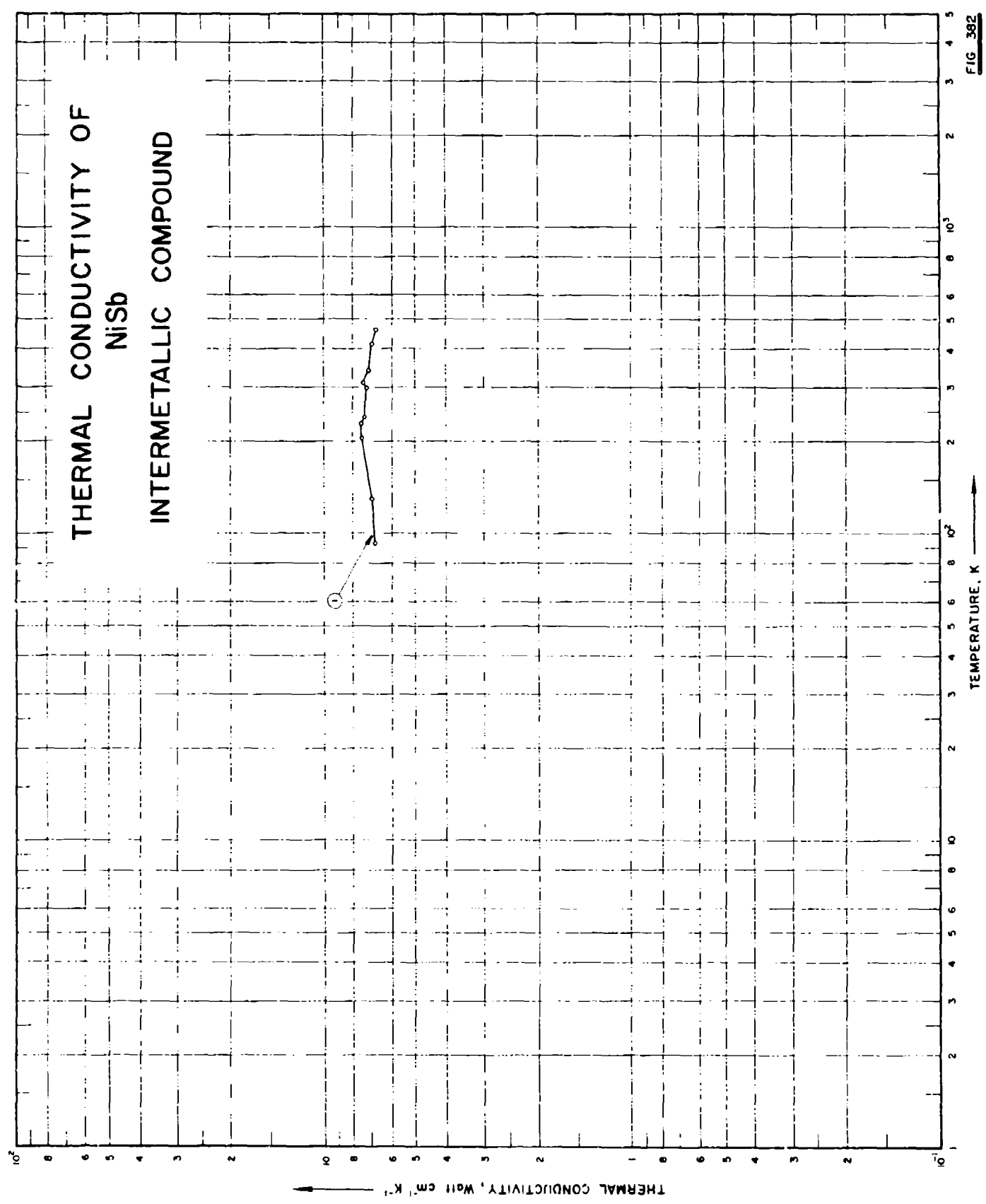
DATA TABLE NO. 381 THERMAL CONDUCTIVITY OF  $\text{MoSi}_2$  INTERMETALLIC COMPOUNDS[Temperature, T, K; Thermal Conductivity, k, Watt  $\text{cm}^{-1} \text{K}^{-1}$ ]

T	k	T	k	T	k
<u>CURVE 1</u>					
424.2	0.450	470.6	0.423	676.1	0.355
506.2	0.423	517.4	0.402	778.7	0.346
630.7	0.387	580.5	0.384	869.3	0.332
781.2	0.356	676.1	0.356	983.2	0.321
825.2	0.318	778.7	0.346	1097.0	0.317
905.2	0.297	869.3	0.332	<u>CURVE 8*</u>	
1033.2	0.255	983.2	0.321	607.3	0.389
1061.2	0.252	1097.0	0.317	613.4	0.384
<u>CURVE 5</u>					
1093.2	0.248	437.1	0.407	655.1	0.369
1129.2	0.238	542.1	0.375	664.5	0.368
1216.2	0.222	635.9	0.324	666.1	0.364
1297.2	0.186	819.3	0.275	703.2	0.354
1517.2	0.159	967.6	0.235	706.6	0.356
<u>CURVE 2</u>					
426.2	0.345	1104.3	0.203 <sup>f</sup>	712.1	0.352
506.2	0.349	1255.9	0.173	787.9	0.329
631.7	0.299	1441.5	0.153	799.8	0.326
788.2	0.244	1623.2	0.118	801.3	0.315
832.2	0.256	1812.6	0.109	853.2	0.322
905.2	0.244	1933.7	0.115	855.5	0.314
1036.2	0.218	2058.2	0.121	956.0	0.294
1067.2	0.222	<u>CURVE 6*</u>		961.6	0.297
1095.2	0.208	448.8	0.443	1048.9	0.306
1127.2	0.201	491.3	0.422	1061.6	0.297
1213.2	0.180	552.5	0.397	<u>CURVE 9</u>	
1298.2	0.154	643.8	0.371	404.0	0.577
1513.2	0.126	749.9	0.348	436.0	0.545
<u>CURVE 3</u>					
448.8	0.443	835.8	0.336	451.2	0.5287
491.3	0.422	957.1	0.320	464.3	0.524
552.5	0.397	1073.8	0.317	514.7	0.4993
643.8	0.371	<u>CURVE 7*</u>		520.3	0.520
749.9	0.348	470.6	0.423	539.4	0.490
839.8	0.336	517.4	0.402	583.2	0.485
957.1	0.320	580.5	0.384	620.4	0.4565
1073.8	0.317			643.2	0.434
				766.5	0.407
				839.8	0.371

\* Not shown on plot

FIG. 382

# THERMAL CONDUCTIVITY OF NiSb INTERMETALLIC COMPOUND



## SPECIFICATION TABLE NO. 382 THERMAL CONDUCTIVITY OF NiSi INTERMETALLIC COMPOUNDS

[For Data Reported in Figure and Table No. 382]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)	Specifications and Remarks
1	959		1966	93-166		NiSi	Specimen 7 x 7 x 30 mm; electrical resistivity 3.45, 8.01, 8.78, 9.97, 10.75, 10.42, 11.55, 10.06, 14.80, 14.76, 16.15, and 16.68 $\mu$ ohm cm at 79, 214, 241, 274, 281, 289, 325, 376, 416, 424, 474 and 483 K respectively.	

## DATA TABLE NO. 382 THERMAL CONDUCTIVITY OF NiSD INTERMETALLIC COMPOUNDS

(Temperature, T, K; Thermal Conductivity,  $k$ , Watt  $\text{cm}^{-1} \text{K}^{-1}$ )

T	$k$
93	0.682
130	0.697
205	0.749
229	0.751
240	0.731
298	0.724
311	0.741
340	0.712
412	0.694
466	0.678



SPECIFICATION TABLE NO. 383 THERMAL CONDUCTIVITY OF  $\text{Re}_3\text{As}_7$  INTERMETALLIC COMPOUNDS

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
1	560		1961	303.2		$\text{Re}_3\text{As}_7$	Electrical resistivity $1.10 \times 10^{-3}$ ohm cm at 30 C.

DATA TABLE NO. 383 THERMAL CONDUCTIVITY OF  $\text{Re}_3\text{As}_7$  INTERMETALLIC COMPOUNDS(Temperature, T, K; Thermal Conductivity, k, Watt  $\text{cm}^{-1}\text{K}^{-1}$ )

T            k  
CURVE 1  
 303.2      0.024

No graphical presentation

SPECIFICATION TABLE NO. 384 THERMAL CONDUCTIVITY OF  $\text{Re}_x\text{Ge}_y$  INTERMETALLIC COMPOUNDS

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)	Specifications and Remarks
1	574		1961	393-563		ReGe		Hot pressed at 1700 F; density 8.68 g cm <sup>-3</sup> ; electrical resistivity 1079, 1601, 1721, 1761, 1810, 1763, 1717, 1774, 1736, 1794, 1743, and 1695 $\mu\text{ohm cm}$ at 25, 44, 79, 90, 120, 143, 165, 170, 204, 224, 259, and 283 C respectively.
2	960		1961	303.2		$\text{ReGe}_2$		Electrical resistivity $1.55 \times 10^{-3}$ ohm cm at 30 C.

DATA TABLE NO. 384 THERMAL CONDUCTIVITY OF  $\text{Re}_x\text{Ge}_y$  INTERMETALLIC COMPOUNDS(Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup> K<sup>-1</sup>)

T k

CURVE 1<sup>a</sup>

393.2 0.050  
 438.2 0.055  
 483.2 0.057  
 563.2 0.060  
 563.2 0.063

CURVE 2<sup>b</sup>

303.2 0.072

No graphical presentation

# THERMAL CONDUCTIVITY OF ReSe<sub>2</sub> INTERMETALLIC COMPOUND

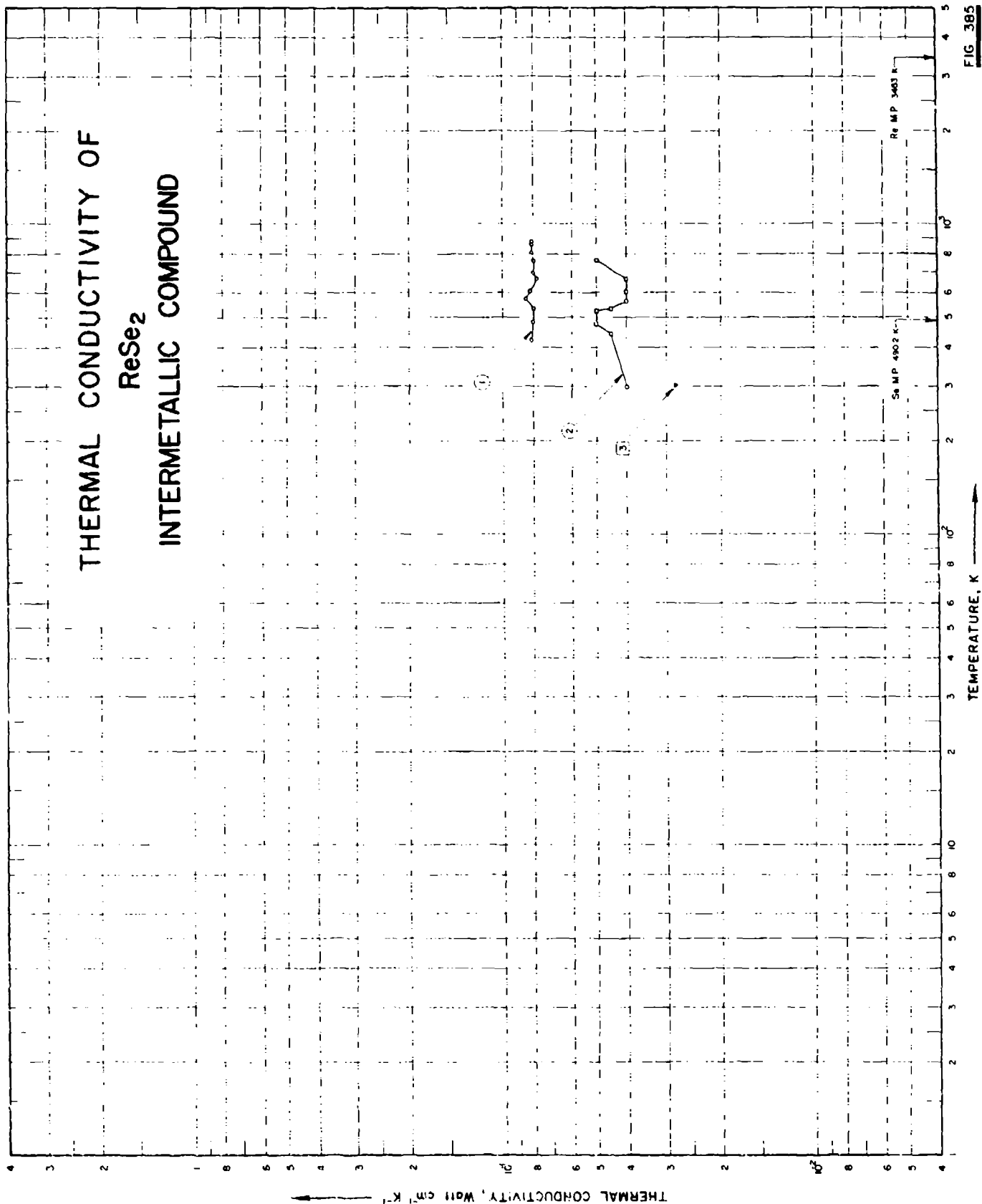


FIG 385

SPECIFICATION TABLE NO. 355 THERMAL CONDUCTIVITY OF  $\text{ReSe}_3$  INTERMETALLIC COMPOUNDS

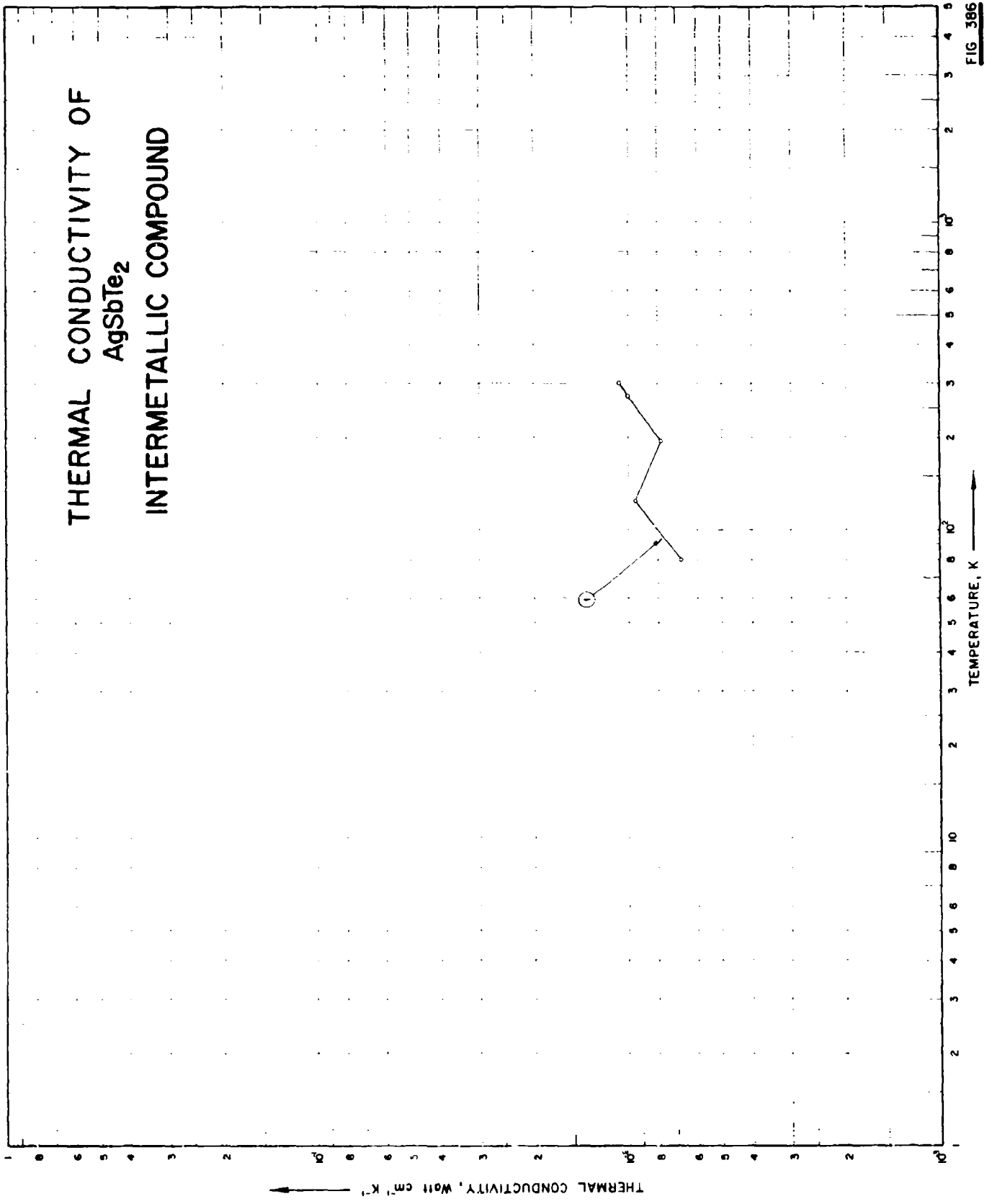
For Data Reported in Figure and Table No. 3557

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)	Specifications and Remarks
1	574		1961	423-873		$\text{ReSe}_3$		Very faint trace of Si; electrical resistivity reported as 1308, 816, 686, 471, 287, 185, 96, 79, 61, 33, 33, 23, 25, 27, 26, 39, 22, 28, 27, 27, 25 ohm cm at 72, 96, 112, 141, 161, 189, 221, 245, 255, 296, 328, 353, 383, 418, 442, 465, 486, 522, 563, 616, 642, and 681 C respectively.
2	574		1961	298-763		$\text{ReSe}_3$		Very faint trace of Si; electrical resistivity reported as 470, 269, 110, 8, 39, 53, 62, 75, 81, 71, 63, 49, 31, 31, 22, 13, 6, and 6 ohm cm at 25, 38, 59, 70, 78, 86, 99, 127, 148, 171, 197, 239, 278, 309, 359, 410, 439, 469, and 500 C respectively.
3	960		1961	303.2		$\text{ReSe}_3$		Electrical resistivity 2000 ohm cm at 30 C.

DATA TABLE NO. 355 THERMAL CONDUCTIVITY OF  $\text{Re}_2\text{S}_7$  INTERMETALLIC COMPOUNDS(Temperature, T, K; Thermal Conductivity,  $\kappa$ ,  $\text{Watt cm}^{-1}\text{K}^{-1}$ )

T	$\kappa$
<u>CURVE 1</u>	
423.2	0.081
483.2	0.080
533.2	0.080
573.2	0.085
603.2	0.082
663.2	0.078
693.2	0.080
763.2	0.080
803.2	0.081
853.2	0.081
873.2	0.081
<u>CURVE 2</u>	
298.2	0.040
443.2	0.045
473.2	0.050
523.2	0.050
533.2	0.045
563.2	0.040
608.2	0.040
665.2	0.040
763.2	0.050
<u>CURVE 3</u>	
303.2	0.028

THERMAL CONDUCTIVITY OF  
AgSbTe<sub>2</sub>  
INTERMETALLIC COMPOUND



SPECIFICATION TABLE NO. 386 THERMAL CONDUCTIVITY OF  $\text{Ag}_2\text{SbTe}_3$  INTERMETALLIC COMPOUNDS

[For Data Reported in Figure and Table No. 386]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)	Specifications and Remarks
1	553, 756	L	1959	80-300		$\text{Ag}_2\text{SbTe}_3$		p-type, prepared from elements in an evacuated quartz ampoule by heating to 700 C, melting and cooling at $\sim 5 \text{ C min}^{-1}$ ; electrical resistivity reported as 5.04, 5.00, 4.56, 5.14, 4.44, 4.63, 4.56, 3.94, 4.15, 3.91, 4.35, 3.83, 4.24, and 4.07 milliohm cm at 25, 85, 110, 168, 182, 224, 260, 260, 275, 336, 361, 403, 410, and 478 C, respectively.

DATA TABLE NO. 386 THERMAL CONDUCTIVITY OF  $\text{Ag}_3\text{SnTe}_2$  INTERMETALLIC COMPOUNDS  
[Temperature, T, K; Thermal Conductivity,  $k$ ,  $\text{Watt cm}^{-1} \text{K}^{-1}$ ]

T	k
<u>CURVE 1</u>	
80	0.0068
125	0.0095
195	0.0078
273	0.0100
300	0.0107



## SPECIFICATION TABLE NO. 387 THERMAL CONDUCTIVITY OF Ag Cu INTERMETALLIC COMPOUNDS

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)	Specifications and Remarks
1	230	L	1925	335.2				Specimen ~ 5 cm long with cross section 0.3 cm <sup>2</sup> ; made from Cu (<0.63 of total impurity) supplied by Baker, fused with Ag (99.9 pure); electrical conductivity 4.46 x 10 <sup>8</sup> ohm <sup>-1</sup> cm <sup>-1</sup> at 25 C.

## DATA TABLE NO. 387 THERMAL CONDUCTIVITY OF Ag Cu INTERMETALLIC COMPOUNDS

[ Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup>K<sup>-1</sup>]

T k

CURVE 1\*

335.2 3.142

\* No graphical presentation



SPECIFICATION TABLE NO. 388 THERMAL CONDUCTIVITY OF  $\text{Ag}_2\text{Se}$  INTERMETALLIC COMPOUNDS

[ For Data Reported in Figure and Table No. 388 ]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition ( weight percent), Specifications and Remarks
1	551	L	1960	288-373		$\text{Ag}_2\text{Se}$	Stoichiometric crystalline; 12 mm dia x 19.1 mm long; prepared from 99.999 + Se and 99.99+ Ag from American Smelting and Refining Co., melted in carbon coated quartz tube, evacuated at $\sim 10^{-5}$ mm Hg heated at 1050-1100C for 16 hrs; electrical resistivity $10.2 \text{ ohm cm}$ .
2	963	L	1962	199-385	5	$\text{Ag}_2\text{Se}; \text{S1}$	Polycrystalline; prepared by melting appropriate amounts of pure elements in a quartz tube, floating zone melting in a graphite boat under controlled vapor pressure at a speed of $0.5 \text{ mm min}^{-1}$ for about 25 passages, slowly cooled $15 \sim 10 \text{ degree h}^{-1}$ , to 150 C, then cooled at $1 \sim 2 \text{ degree h}^{-1}$ through the transition temperature (133 C), below the transition temperature the specimen annealed for several hrs; carrier concentration $1.2 \times 10^{18} \text{ cm}^{-3}$ ; electrical resistivity reported as $0.468 \sim 1.98 \text{ milliohm cm}$ in the range 85.8-621.1 K.
3	963	L	1962	84-389	5	$\text{Ag}_2\text{Se}; \text{H1}$	Polycrystalline; same fabrication method as the above specimen; carrier concentration $1.5 \times 10^{18} \text{ cm}^{-3}$ ; electrical resistivity reported as $0.462 \sim 1.33 \text{ milliohm cm}$ in the range 85.6-632.9 K.
4	963	L	1962	80-585	5	$\text{Ag}_2\text{Se}; \text{H2}$	Polycrystalline; same fabrication method as the above specimen; carrier concentration $5.7 \times 10^{17} \text{ cm}^{-3}$ ; electrical resistivity reported as $0.533 \sim 2.89 \text{ milliohm cm}$ in the range 80.4-645.2 K.
5	963	L	1962	112-588	5	$\text{Ag}_2\text{Se}; \text{H3}$	Polycrystalline; same fabrication method as the above specimen; carrier concentration $1.2 \times 10^{18} \text{ cm}^{-3}$ .
6	963	L	1962	80-395	5	$\text{Ag}_2\text{Se}; \text{F1}$	Polycrystalline; same fabrication method as the above specimen, carrier concentration $1.0 \times 10^{18} \text{ cm}^{-3}$ .
7	943	L	1963	315-451		$\text{Ag}_2\text{Se}$	n-type; specimen 8 mm in dia and 12-14 mm long; synthesized in evacuated quartz ampule at $10^{-4}$ mm Hg, heated to above the melting point of the component with higher melting point for 2 hrs, then heated to the melting point of the compound for 8 hrs; annealed at 700-800C for several hrs and then cooled to room temperature.

DATA TABLE NO. 388 THERMAL CONDUCTIVITY OF  $A_{1-x}S_{2x}Se$  INTERMETALLIC COMPOUNDS[Temperature, T, K; Thermal Conductivity, k, Watt  $cm^{-1} K^{-1}$ ]

T	k	T	k
<u>CURVE 1</u>			
288	0.0089	112.1	0.0188
313	0.0070	116.8	0.0139
333	0.0078	328.9	0.0214
353	0.0089	389.1	0.0301
373	0.010	446.4	0.0301
<u>CURVE 2</u>			
199.2	0.0118	456.6	0.0282
256.4	0.0146	487.8	0.0278
274.7	0.0151	537.6	0.0279
303.0	0.0182	546.4	0.0287
324.7	0.0199	588.2	0.0263
342.5	0.0208		
367.6	0.0225		
394.6	0.0249		
<u>CURVE 3</u>			
83.5	0.0177	79.9	0.00870
197.6	0.0136	194.9	0.0102
257.7	0.0151	257.7	0.0115
277.0	0.0159	275.5	0.0119
312.5	0.0166	310.6	0.0146
330.0	0.0179	332.2	0.0164
348.4	0.0189	348.4	0.0189*
366.3	0.0205	373.1	0.0214
377.4	0.0221	393.7	0.0255
389.1	0.0236	395.3	0.0260
<u>CURVE 4</u>			
79.9	0.0120		
196.1	0.0127		
256.4	0.0131		
277.0	0.0137		
314.5	0.0183		
334.4	0.0207		
357.1	0.0230		
380.2	0.0245		
387.6	0.0263		
427.4	0.0269		
446.4	0.0266		
473.9	0.0261		
531.9	0.0239		
584.8	0.0210		
<u>CURVE 5</u>			
<u>CURVE 6</u>			
<u>CURVE 7</u>			
315	0.0285	404	0.0216
334	0.0223	412	0.0231
354	0.0194	438	0.0280
374	0.0187	451	0.0303

\* Not shown on plot

SPECIFICATION TABLE NO. 389 THERMAL CONDUCTIVITY OF  $Ag_xTe_y$  INTERMETALLIC COMPOUNDS

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
1	552	L	1961	298.2		$Ag_{1-x}Te$	p-type hexagonal crystal by increasing the tellurium content to about 3 atomic % excess; specimen prepared by 99.999+ pure silver and 99.999+ pure tellurium reacted and either zone leveled or zone fired in carbon boat in vacuo; measured at room temperature vs. the electric conductivity $\sigma$ .
2	552	L	1961	298.2		$Ag_1Te$	Intrinsic and n-type crystal; prepared by 99.999+ pure silver and 99.999+ pure tellurium reacted and either zone leveled or zone fired in carbon boat in vacuo; measured at room temperature vs. the electric conductivity $\sigma$ .
3	552	L	1961	298.2		$Ag_1Te$	The above specimen doped with AgI of the order of $10^{19}$ atom/cc resulting excess silver of the order of 0.5 atomic % in $Ag_1Te$ , measured at room temperature vs. electric conductivity $\sigma$ .

DATA TABLE NO. 389 THERMAL CONDUCTIVITY OF  $Ag_xTe_y$  INTERMETALLIC COMPOUNDS[ Temperature, T, K; Thermal Conductivity, k, Watt  $cm^{-1}K^{-1}$  ]

$\sigma$ ( $ohm\ cm$ ) $^{-1}$ k	$\sigma$ ( $ohm\ cm$ ) $^{-1}$ k
$\frac{CURVE\ 1^*}{T = 298.2}$	$\frac{CURVE\ 3^*}{T = 298.2}$
140 0.0079	537 0.0125
202 0.00915	1247 0.0159
	1495 0.0192
$\frac{CURVE\ 2^*}{T = 298.2}$	1766 0.0172
258 0.0081	
274 0.0065	
258 0.00809	
720 0.0125	
1050 0.0165	
1370 0.01723	
1400 0.0182	
1497 0.0164	

\* No graphical presentation

SPECIFICATION TABLE NO. 390 THERMAL CONDUCTIVITY OF  $\text{Sr}_2\text{Si}$  INTERMETALLIC COMPOUNDS

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (wt.-%)	Specifications and Remarks
1	548	L	1961	298.2		$\text{Sr}_2\text{Si}$ ; No. 1		Synthesized; seebeck coeff. $21.1 \mu\text{V/K}^\circ$ at 25 C; electrical resistivity $2.42 \times 10^{-3}$ ohm cm at 25 C; figure of merit $0.289 \times 10^{-4} \text{K}^{-1}$ at 25 C.

DATA TABLE NO. 390 THERMAL CONDUCTIVITY OF  $\text{Sr}_2\text{Si}$  INTERMETALLIC COMPOUNDS[ Temperature, T, K; Thermal Conductivity, k, Watt  $\text{cm}^{-1}\text{K}^{-1}$  ]

T k

CURVE 1\*

298.2 0.00636

\* No graphical presentation

SPECIFICATION TABLE NO. 391 THERMAL CONDUCTIVITY OF  $\text{Sr}_2\text{Sn}$  INTERMETALLIC COMPOUNDS

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)	Specifications and Remarks
1	548	I.	1961	298.2		$\text{Sr}_2\text{Sn}$ , No. 1		Synthesized; seebeck coeff. $24.4 \mu\text{V K}^{-1}$ at 25 C; electrical resistivity $3.5 \times 10^4$ ohm cm at 25 C; figure of merit $0.692 \times 10^4 \text{ K}^{-1}$ at 25 C.

DATA TABLE NO. 391 THERMAL CONDUCTIVITY OF  $\text{Sr}_2\text{Sn}$  INTERMETALLIC COMPOUNDS[Temperature, T, K; Thermal Conductivity, k,  $\text{Watt cm}^{-1}\text{K}^{-1}$ ]

T k

CURVE 1

298.2 0.0238

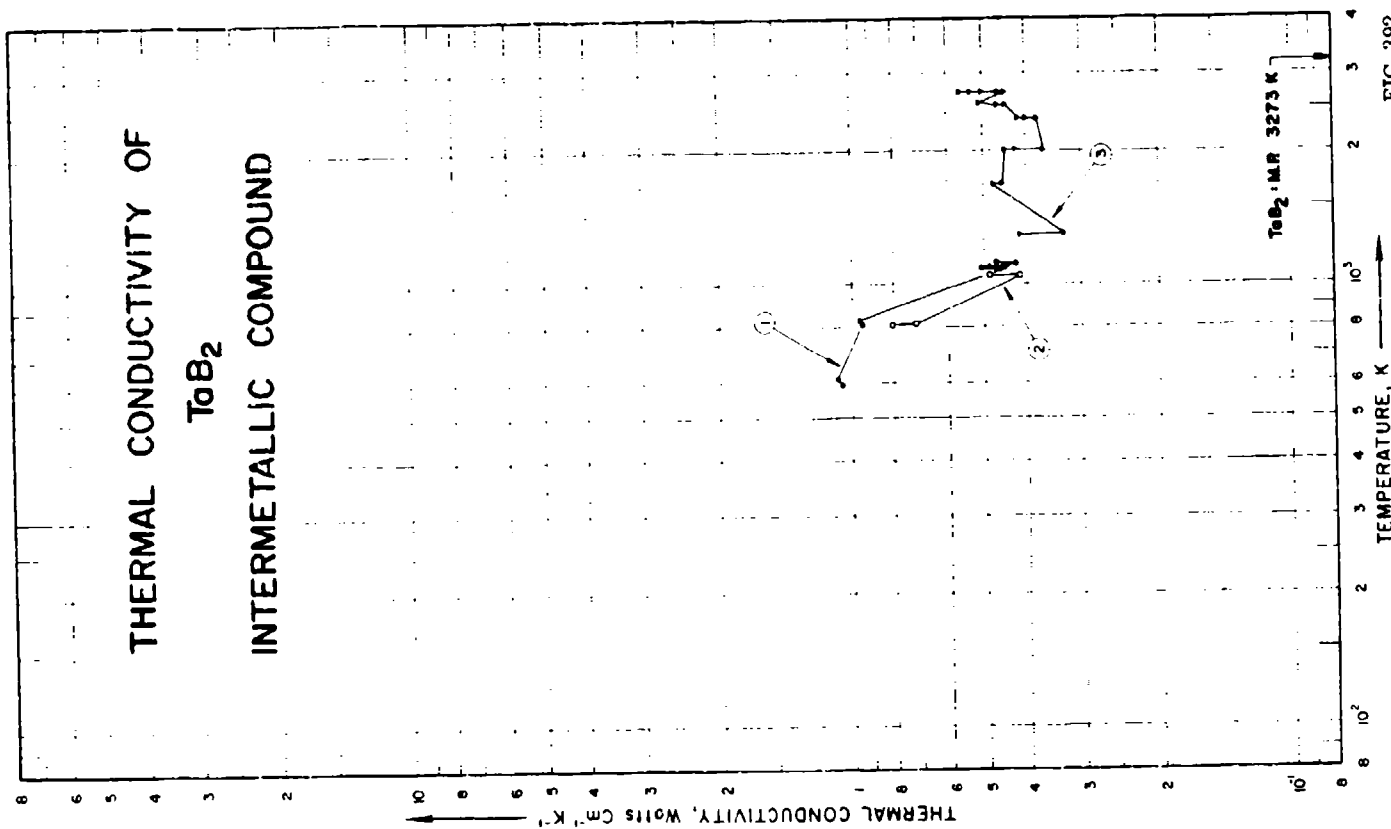


FIG 392



SPECIFICATION TABLE NO. 392 THERMAL CONDUCTIVITY OF  $TaB_3$  INTERMETALLIC COMPOUNDS

[ For Data Reported in Figure and Table No. 392 ]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition ( weight percent), Specifications and Remarks
1	603	R	1962	595-1174	6.5	$TaB_3$	Specimen 3/4 in. in dia and 3/4 in. long; pressed and sintered; maximum exposure temperature 2816 K; density 12.11 g cm <sup>-3</sup> .
2	603	R	1962	810-1059	6.5	$TaB_3$	Similar to the above specimen.
3	603	k	1962	1300-2750	6.5	$TaB_3$	Similar to the above specimen except specimen found to show incipient melting after measurements.

DATA TABLE NO. 302 THERMAL CONDUCTIVITY OF Tl<sub>2</sub>B<sub>2</sub> INTERMETALLIC COMPOUNDS[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup> K<sup>-1</sup>]

T	k
<u>CURVE 1</u>	
594.8	1.06
603.2	1.07*
615.4	1.04
617.6	0.943
636.5	0.961
640.9	0.966*
1084.3	0.460
1090.4	0.509
1092.1	0.490
1092.6	0.450
1113.7	0.425
1174.3	0.470
<u>CURVE 2</u>	
810.4	0.805
813.2	0.596*
813.7	0.718
1054.8	0.415
1058.7	0.487
<u>CURVE 3</u>	
1299.8	0.514
1305.4	0.327
1306.2	0.336*
1683.2	0.476
1688.7	0.453
2011.0	0.446
2011.0	0.421
2011.0	0.363
2369.3	0.376
2369.3	0.400
2372.1	0.418
2558.2	0.444
2558.2	0.464
2563.7	0.505
2733.2	0.447
2738.7	0.462
2741.5	0.500
2747.1	0.532
2749.8	0.562

\* Not shown on plot

SPECIFICATION TABLE NO. 393 THERMAL CONDUCTIVITY OF TaGe<sub>2</sub> INTERMETALLIC COMPOUNDS

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)	Specifications and Remarks
1	360		1961	303.2		TaGe <sub>2</sub>		Electrical resistivity $6.5 \times 10^{-4}$ ohm cm.

DATA TABLE NO. 393 THERMAL CONDUCTIVITY OF TaGe<sub>2</sub> INTERMETALLIC COMPOUNDS[ Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup>K<sup>-1</sup> ]

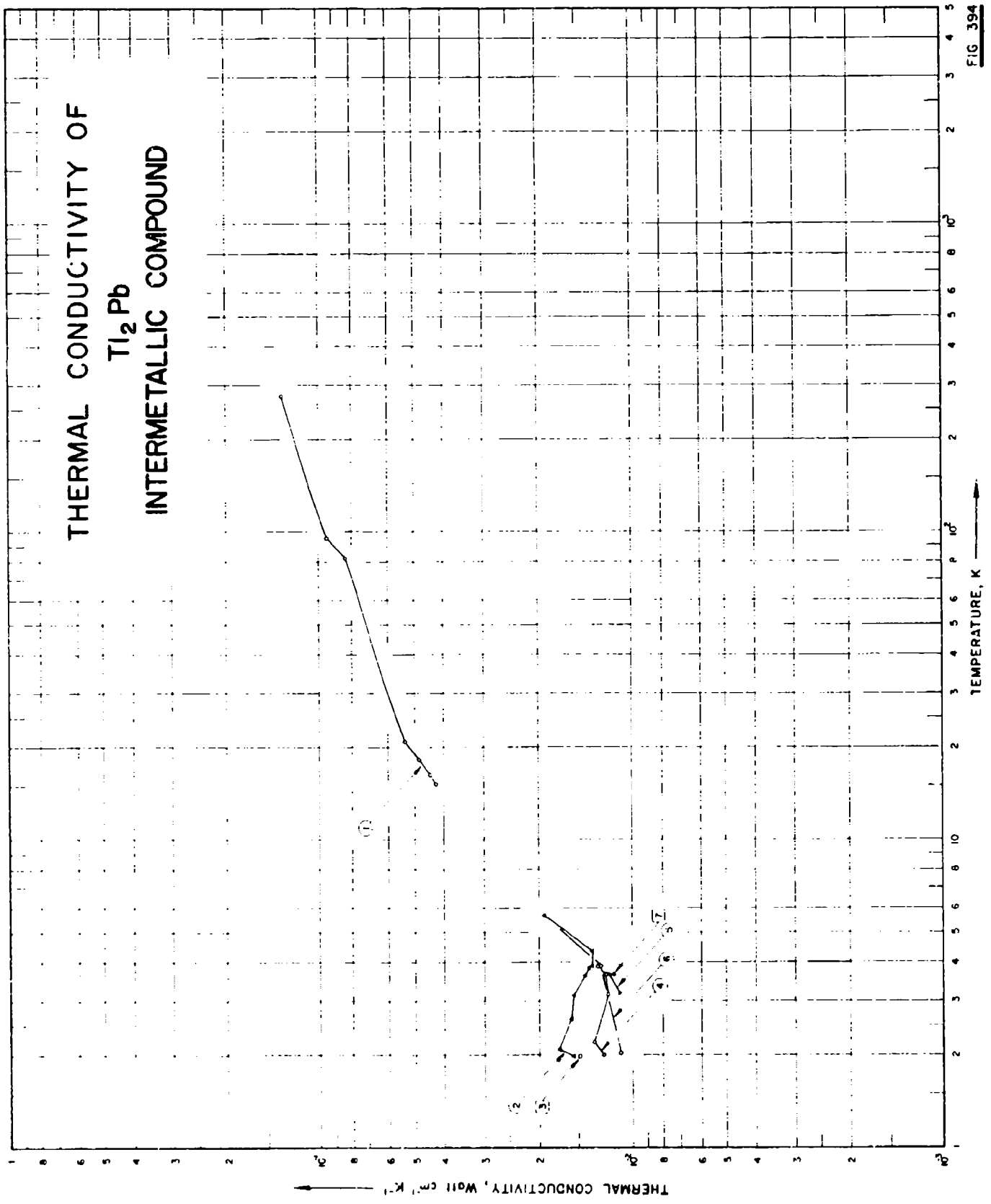
T K

CURVE 1\*

303.2 0.043

\* No graphical presentation

THERMAL CONDUCTIVITY OF  
 $Tl_2Pb$   
 INTERMETALLIC COMPOUND



SPECIFICATION TABLE NO. 394 THERMAL CONDUCTIVITY OF  $Tl_2Pb$  INTERMETALLIC COMPOUNDS

[ For Data Reported in Figure and Table No. 394 ]

Curve No.	Ref. No. Used	Method	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) Tl	Pb	Composition (continued), Specifications and Remarks
1	337	I	1932	15-276		$Tl_2Pb$	66.3	33.7	Specimen 9 mm long, 7.4 mm dia; annealed for 180 hrs at ~melting point.
2	337	L	1932	2.0-5.7		$Tl_2Pb$	66.3	33.7	The above specimen in superconducting state.
3	337	I	1932	1.97		$Tl_2Pb$	66.3	33.7	The above specimen measured in a 214 gauss magnetic field.
4	337	I	1932	2.0-3.9		$Tl_2Pb$	66.2	33.7	The above specimen measured in a 481 gauss magnetic field.
5	337	I	1932	3.2, 3.7		$Tl_2Pb$	66.3	33.7	The above specimen measured in a 642 gauss magnetic field.
6	337	L	1932	2.0-5.7		$Tl_2Pb$	66.3	33.7	The above specimen measured in a 7' ss magnetic field.
7	337	I	1932	3.66		$Tl_2Pb$	66.3	33.7	The above specimen measured in a 835 gauss magnetic field.

DATA TABLE NO. 394 THERMAL CONDUCTIVITY OF Ti,Pb INTERMETALLIC COMPOUNDS

[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup>K<sup>-1</sup>]

T	k	T	k
<u>CURVE 1</u>			
15.1	0.0423	3.56	0.0116
16.2	0.0446		
18.2	0.0487		
20.9	0.0538		
82	0.0926		
95	0.0946		
27.6	0.1133		
<u>CURVE 2</u>			
1.97	0.0156		
2.17	0.0173		
2.61	0.0158		
3.12	0.0156		
3.63	0.0143		
3.84	0.0139		
3.97	0.0136		
4.34	0.0136		
5.69	0.0193		
<u>CURVE 3</u>			
1.97	0.0147		
<u>CURVE 4</u>			
2.00	0.0124		
2.21	0.0133		
3.155	0.0120		
3.65	0.0122		
3.88	0.0123		
<u>CURVE 5</u>			
3.17	0.0110		
3.66	0.0119		
<u>CURVE 6</u>			
2.02	0.0109		
3.88	0.0127		
5.69	0.0194		

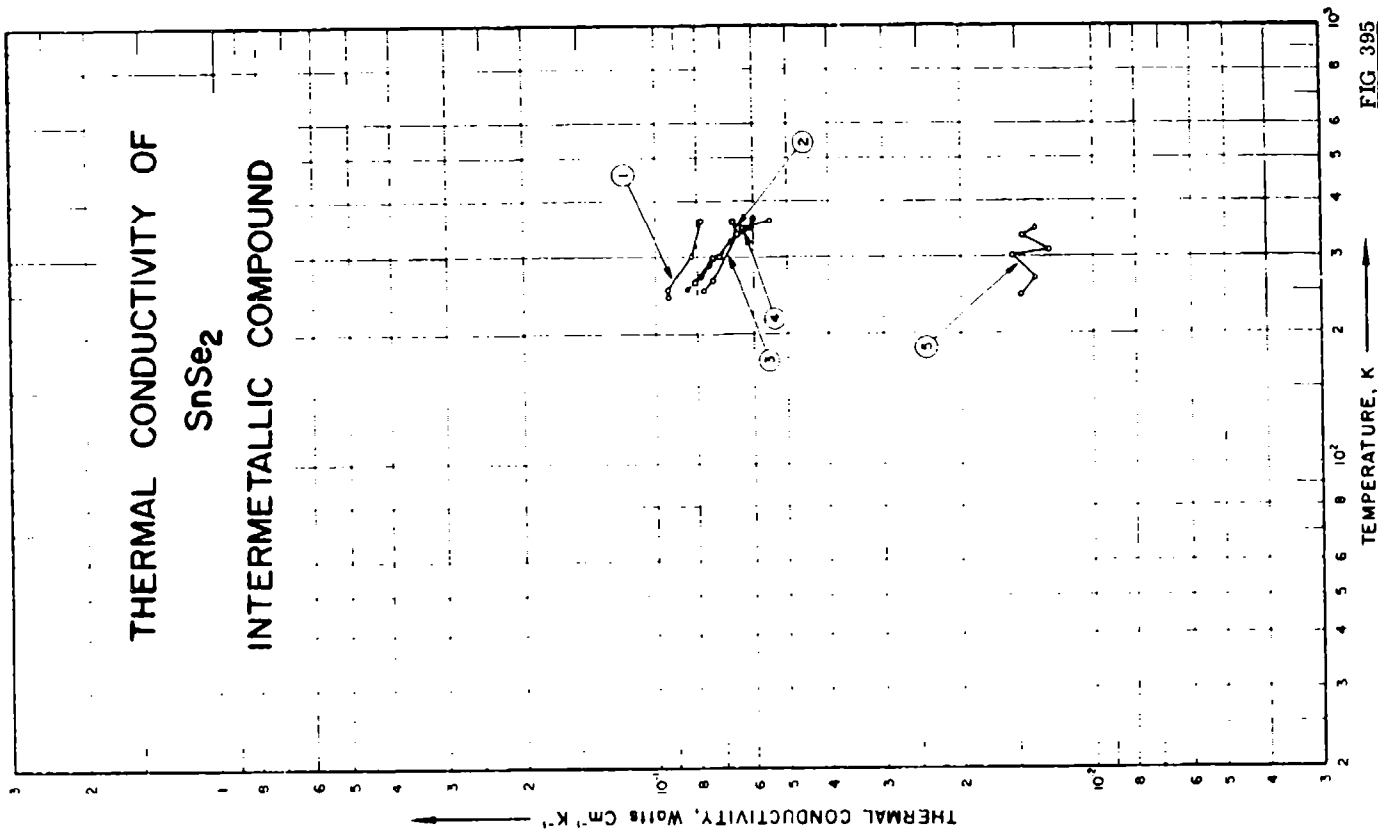


FIG. 395

SPECIFICATION TABLE NO. 795 THERMAL CONDUCTIVITY OF  $\text{SnSe}_2$  INTERMETALLIC COMPOUNDS

[For Data Reported in Figure and Table No. 395]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
1	545	L	1961	241-369	$\pm 10$	$\text{Sn Se}_2$	Polycrystal; $n_s = 2 \times 10^{15} \text{ cm}^{-2}$ ; micro hardness 30 kg mm $^{-2}$ ; specimen 8 x 8 x 8 mm.
2	545	L	1961	261-359	$\pm 10$	$\text{Sn Se}_2$	Single crystal; prepared by the Bridgeman method from spectroscopically pure elements; specimen 8 x 8 x 8 mm; freezing point $629 \pm 5 \text{ C}$ ; $n_s = 2 \times 10^{17} \text{ cm}^{-2}$ ; crystal structure, a = 3.811 Å, c = 6.137 Å; heat flow perpendicular to c-axis of crystal.
3	545	L	1961	249-380	$\pm 10$	$\text{Sn Se}_2$	Similar to the above specimen except $n_s = 6 \times 10^{15} \text{ cm}^{-2}$ .
4	545	L	1961	251-363	$\pm 10$	$\text{Sn Se}_2$	Similar to the above specimen.
5	545	L	1961	244-345	$\pm 10$	$\text{Sn Se}_2$	Similar to the above specimen except heat flow parallel to c-axis.



DATA TABLE NO. 395 THERMAL CONDUCTIVITY OF  $\text{SnSi}_2$  INTERMETALLIC COMPOUNDS[Temperature, T, K; Thermal Conductivity, k,  $\text{Watt cm}^{-1} \text{K}^{-1}$ ]

T k

CURVE 1

241.0	0.0934
250.9	0.0940
294.5	0.0823
360.4	0.0784

CURVE 2

260.6	0.0810
296.3	0.0739
298.5	0.0711
336.1	0.0655
359.1	0.0669

CURVE 3

249.1	0.0772
263.2	0.0739
347.4	0.0633
360.4	0.0550

CURVE 4

250.6	0.0847
269.5	0.0784
284.9	0.0749
300.8	0.0711
303.0	0.0701
326.8	0.0659
347.9	0.0611
363.6	0.0600

CURVE 5

243.9	0.0146
265.6	0.0136
294.5	0.0153
300.5	0.0126
333.3	0.0146
344.8	0.0136

\* Not shown on plot



## SPECIFICATION TABLE NO. 396 THERMAL CONDUCTIVITY OF SnTe INTERMETALLIC COMPOUNDS

[For Data Reported in Figure and Table No. 396]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
1	553	L	1959	300		SnTe	No information reported.
2	964	I	1966	116-440		SnTe	Stoichiometric single crystal; prepared by melting 99.999 pure Sn and Te in a 5-6 mm dia quartz tube (to which a 1 mm dia x 14 mm long capillary was attached) in a vacuum of 10 <sup>-2</sup> Torr, heated at 300 C higher than the melting point for 100 hrs, pulling-processed through a double melting oven (upper oven 30-50 C above, and the lower oven 30-50 C below, the melting temperature) with a velocity 10 mm hr <sup>-1</sup> , annealed at 300 C for 4-6 hrs, electrical resistivity reported as 589-762 $\mu$ ohm cm in the range 85.2-483 K.

DATA TABLE NO. 396 THERMAL CONDUCTIVITY OF  $\text{SnTe}$  INTERMETALLIC COMPOUNDS[Temperature, T, K; Thermal Conductivity, k,  $\text{Watt cm}^{-1} \text{K}^{-1}$ ]

T	k
<u>CURVE 1</u>	
300	0.0615
<u>CURVE 2</u>	
116	0.0537
136	0.0548
136	0.0544
150	0.0552
160	0.0548
171	0.0545
185	0.0543
202	0.0540
209	0.0540
223	0.0548
252	0.0566
264	0.0563
274	0.0561
289	0.0591
297	0.0592
306	0.0617
314	0.0619
324	0.0593
330	0.0611
338	0.0595
342	0.0600
351	0.0604
354	0.0604
360	0.0615
366	0.0631
379	0.0630
399	0.0624
406	0.0617
416	0.0641
426	0.0634
440	0.0635

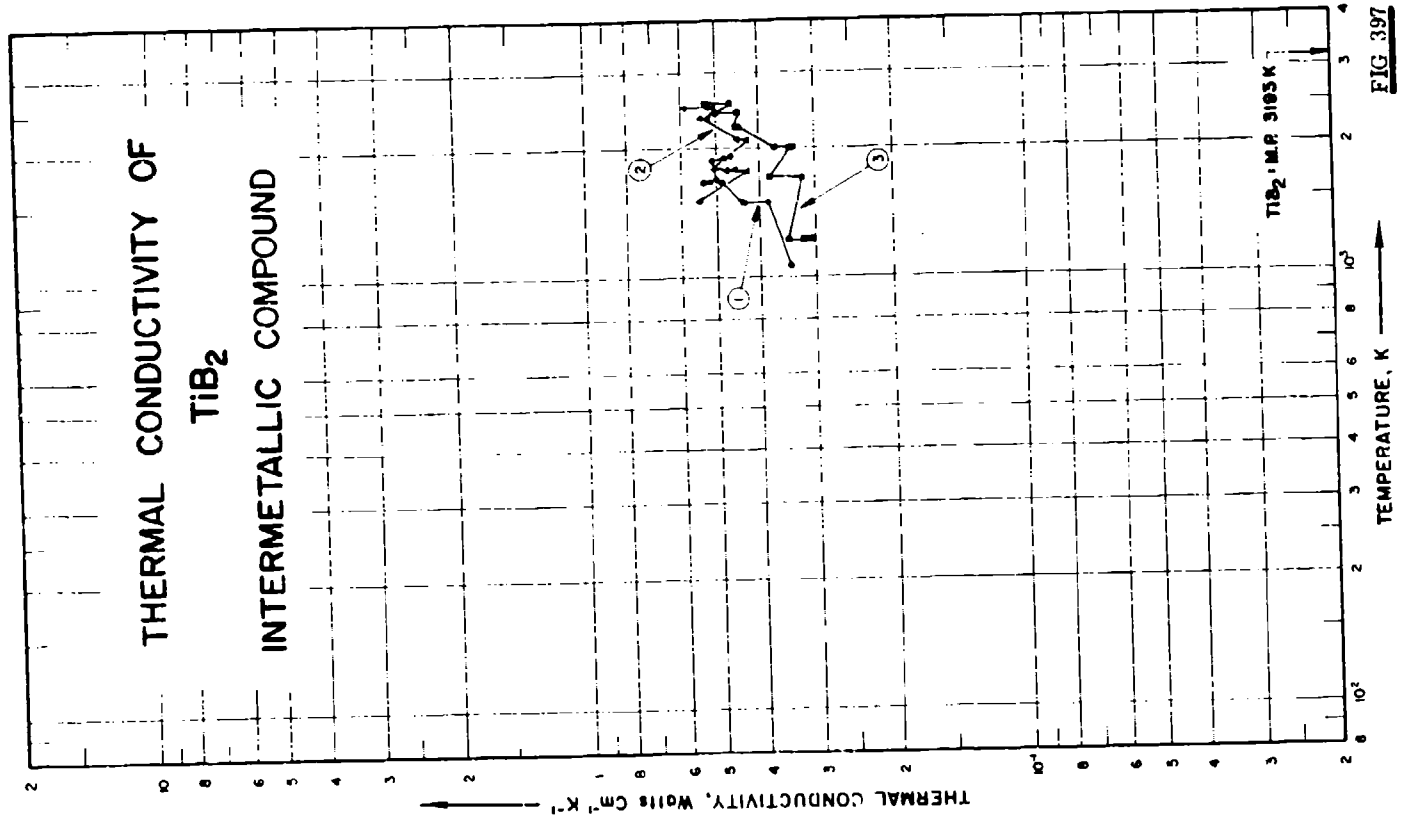


FIG 397

SPECIFICATION TABLE NO. 397 THERMAL CONDUCTIVITY OF TiB<sub>2</sub> INTERMETALLIC COMPOUNDS

[For Data Reported in Figure and Table No. 397.]

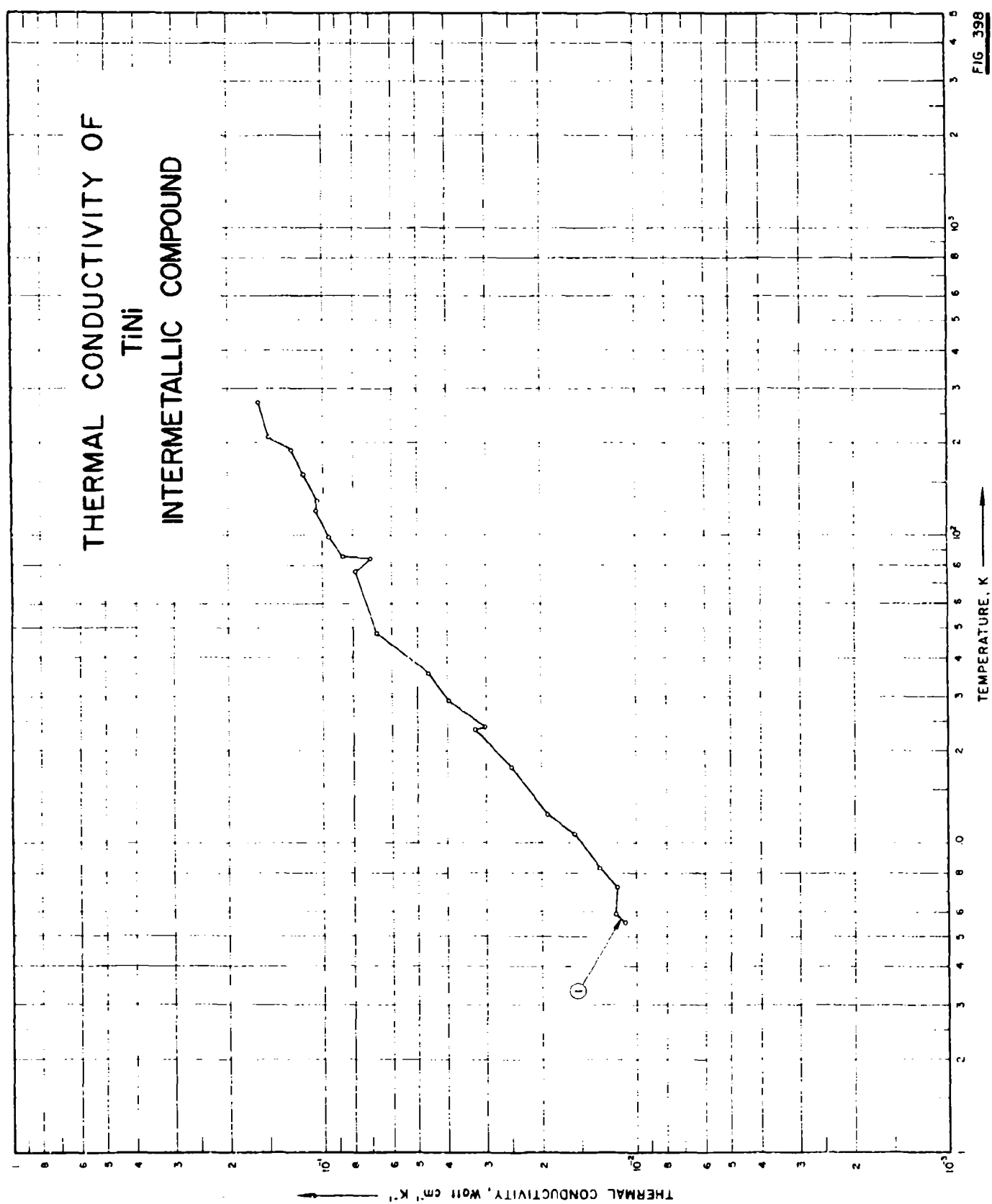
Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
1	544	R	1963	1075-1937	5-7	TiB <sub>2</sub> , 1	Specimen 3/4 in. long, 1/4 in. O.D. and 1/4 in. I.D.; ground and polished; specimen found cracked on post inspection, heat-soaked at 3500 F.
2	544	R	1963	1542-2516	5-7	TiB <sub>2</sub> , 3	Similar to the above specimen except heat-soaked at 3509 F; specimen found cracked on the post inspection.
3	544	R	1963	1239-2578	5-7	TiB <sub>2</sub> , 4	Similar to the above specimen.

DATA TABLE NO. 397 THERMAL CONDUCTIVITY OF  $TiB_2$  INTERMETALLIC COMPOUNDS[ Temperature, T, K; Thermal Conductivity, k, Watt  $cm^{-1} K^{-1}$  ]

T	k	T	k
<u>CURVE 1</u>			
1074.8	0.341	2456.5	0.447
1536.5	0.381	2456.5	0.504
1536.5	0.432	2569.3	0.466
1537.1	0.438	2570.4	0.466 <sup>a</sup>
1594.3	0.486	2577.6	0.530
1695.9	0.537		
1703.2	0.530		
1890.4	0.511		
1910.9	0.480		
1937.1	0.463		
<u>CURVE 2</u>			
1541.5	0.544		
1544.3	0.546		
1545.4	0.541		
1793.2	0.424		
1797.1	0.476		
1798.7	0.453		
1803.7	0.507		
2103.7	0.424		
2109.8	0.423 <sup>a</sup>		
2110.9	0.447		
2370.4	0.544		
2372.6	0.522		
2504.3	0.506		
2511.5	0.520		
2515.9	0.550		
<u>CURVE 3</u>			
1238.7	0.306		
1238.7	0.316		
1238.7	0.319		
1239.8	0.345		
1720.4	0.420		
1720.4	0.320 <sup>a</sup>		
1724.3	0.376		
2049.3	0.340		
2024.2	0.335		
2024.3	0.369		
2262.1	0.446		
2262.1	0.454		
2263.2	0.448 <sup>a</sup>		

<sup>a</sup> Not shown on plot.

# THERMAL CONDUCTIVITY OF TlNi INTERMETALLIC COMPOUND



TEMPERATURE, K



## SPECIFICATION TABLE NO. 398 THERMAL CONDUCTIVITY OF TiNi INTERMETALLIC COMPOUNDS

[ For Data Reported in Figure and Table No. 398 ]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition ( weight percent), Specifications and Remarks
1	965	L	1964	5.5 - 272	5	TiNi	Stoichiometric; prepared from Monel nickel shot (99.9% pure) and DuPont high purity sponge; the titanium contained up to: 0.08 Mg, 0.07 Fe, 0.05 Mn, 0.04 Si, and 0.15 other impurities; specimen rod was hot swaged and furnace cooled from homogenized buttons and machined into cylinder of 0.4 cm dia x 3 cm long; grain size ~ 42 $\mu$ ; electrical resistivity reported range from 23.0 ~ 66.8 $\mu$ ohm cm at 1.32 ~ 302.7 K, respectively.

## DATA TABLE NO. 398 THERMAL CONDUCTIVITY OF T-Ni INTERMETALLIC COMPOUNDS

(Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup> K<sup>-1</sup>)

T	k
5.53	0.0109
5.97	0.0117
7.26	0.0116
8.34	0.0131
10.7	0.0157
12.5	0.0192
17.7	0.0249
23.5	0.0327
23.8	0.0302
29.2	0.0397
35.6	0.0462
48.2	0.0671
75.4	0.0748
84.0	0.0703
85.5	0.0877
89.1	0.0964
121.6	0.106
130.0	0.105
158.1	0.116
190.4	0.127
210.4	0.151
271.6	0.163

CURVE 1

SPECIFICATION TABLE NO. 389 THERMAL CONDUCTIVITY OF W<sub>3</sub>As<sub>7</sub>; INTERMETALLIC COMPOUNDS

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)	Specifications and Remarks
1	960		1961	303.2		W <sub>3</sub> As <sub>7</sub>		Electrical resistivity $2.6 \times 10^{-4}$ ohm cm at 30 C.

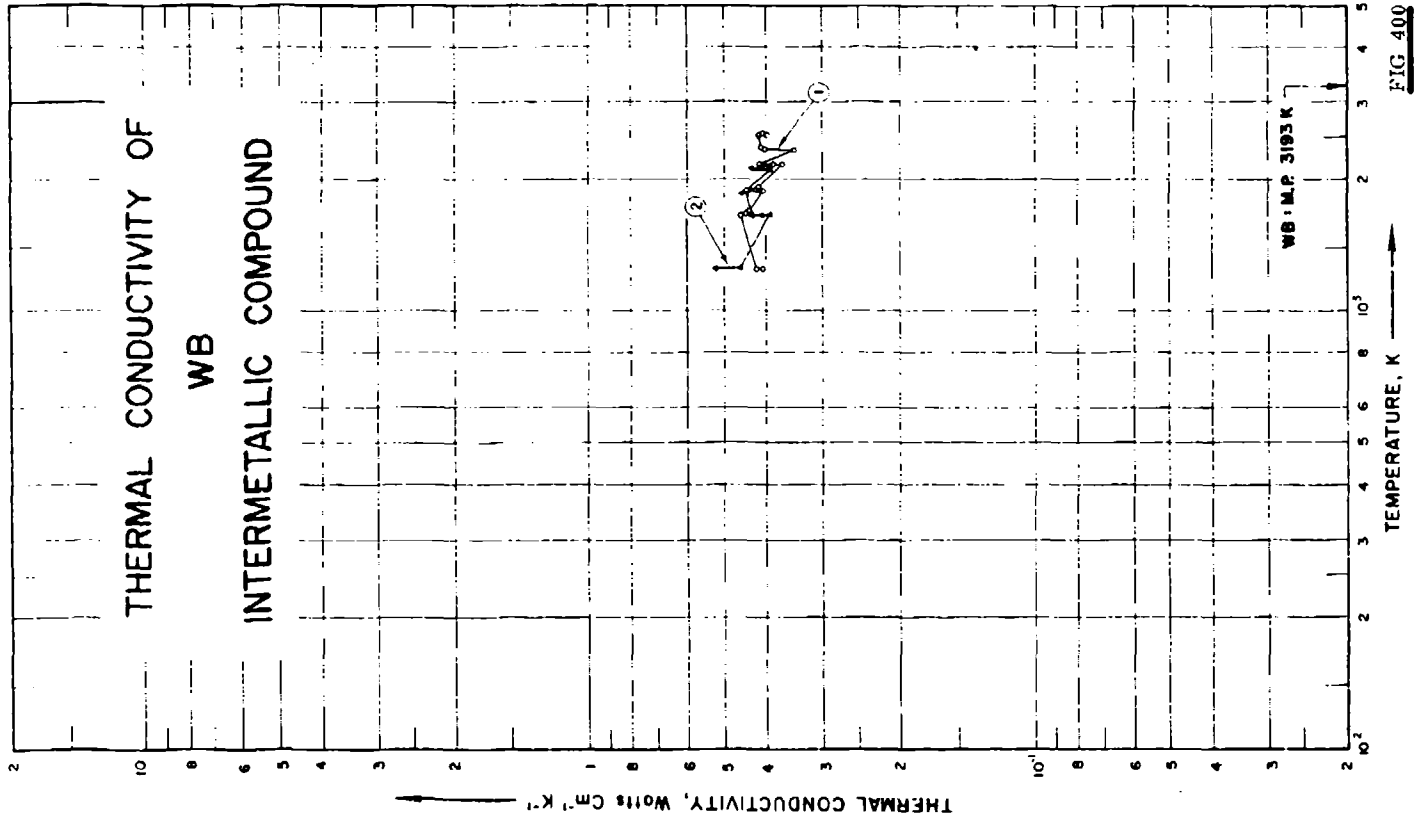
DATA TABLE NO. 399 THERMAL CONDUCTIVITY OF W<sub>3</sub>As<sub>7</sub>; INTERMETALLIC COMPOUNDS  
[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup>K<sup>-1</sup>]

T k

CURVE 1'

303.2 0.032

No graphical presentation



## SPECIFICATION TABLE NO. 400 THERMAL CONDUCTIVITY OF WB INTERMETALLIC COMPOUNDS

(For Data Reported in Figure and Table No. 400)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and R. marks
1	544	R	1963	1255-2540	5-7	WB.1	Specimen 3/4 in. long, 3/4 in. O.D. and 1/41 D.; ground and polished, specimen found broken on post inspection; heat-soaked at 3300 F.
2	544	R	1963	1253-2150	5-7	WB.2	Similar to the above specimen except heat-soaked at 3350 F.; specimen found cracked on post inspection.

## DATA TABLE NO. 400 THERMAL CONDUCTIVITY OF WB INTERMETALLIC COMPOUNDS

[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup>K<sup>-1</sup>]

T	k
<u>CURVE 1</u>	
1255.4	0.426 <sup>s</sup>
1255.4	0.410
1255.4	0.426
1654.8	0.460
1674.8	0.448
1687.1	0.441
1896.5	0.409
1899.3	0.448
1905.4	0.416
2162.1	0.369
2162.6	0.370
2166.5	0.387
2172.1	0.418
2344.3	0.348
2349.3	0.403
2355.4	0.413
2526.5	0.420
2529.3	0.398
2540.4	0.412
<u>CURVE 2</u>	
1252.6	0.526
1252.6	0.461
1252.6	0.462 <sup>s</sup>
1658.7	0.395
1659.3	0.410
1660.9	0.433
1887.1	0.442
1887.6	0.458
1888.7	0.451
1888.7	0.453
2132.6	0.393
2137.1	0.436
2150.4	0.389

Not shown on plot

SPECIFICATION TABLE NO. 401 THERMAL CONDUCTIVITY OF  $WS_2$  INTERMETALLIC COMPOUNDS

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
1	960		1961	303.2		$WS_2$	Electrical resistivity 95 ohm cm at 30 C.

DATA TABLE NO. 401 THERMAL CONDUCTIVITY OF  $WS_2$  INTERMETALLIC COMPOUNDS(Temperature, T, K; Thermal Conductivity,  $\lambda$ ,  $W/m^2 K^{-1}$ )

T K

CURVE 1

303.2 0.034

No graphical presentation

SPECIFICATION TABLE NO. 402 THERMAL CONDUCTIVITY OF  $W_{52}$  INTERMETALLIC COMPOUNDS

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
1	594	R	1961	733-1793		$W_{52}$	No details reported.

DATA TABLE NO. 402 THERMAL CONDUCTIVITY OF  $W_{52}$  INTERMETALLIC COMPOUNDS[Temperature, T, K; Thermal Conductivity, k, Watt  $cm^{-1} K^{-1}$ ]

T	k	T	k
CURVE 1*			
CURVE 1 (cont.) <sup>†</sup>			
733.2	0.393	1789.3	0.391
735.4	0.393	1790.9	0.386
749.8	0.381	1792.6	0.400
753.2	0.374		
874.8	0.339		
883.2	0.343		
893.2	0.348		
895.4	0.337		
895.4	0.336		
1038.7	0.310		
1041.5	0.312		
1210.9	0.294		
1211.5	0.299		
1368.7	0.305		
1373.2	0.299		
1473.2	0.318		
1481.5	0.322		
1537.1	0.296		
1539.6	0.308		
1645.4	0.325		
1654.3	0.327		
1717.6	0.313		
1720.4	0.315		
1765.5	0.287		

\* No graphical presentation



SPECIFICATION TABLE NO. 403 THERMAL CONDUCTIVITY OF  $WTe_2$  INTERMETALLIC COMPOUNDS

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent). Specifications and Remarks
1	960		1961	303.2		$WTe_2$	Electrical resistivity $3.5 \times 10^{-2}$ ohm cm at 30 C.

DATA TABLE NO. 403 THERMAL CONDUCTIVITY OF  $WTe_2$  INTERMETALLIC COMPOUNDS

[ Temperature, T, K; Thermal Conductivity, k, Watt  $cm^{-1}K^{-1}$  ]

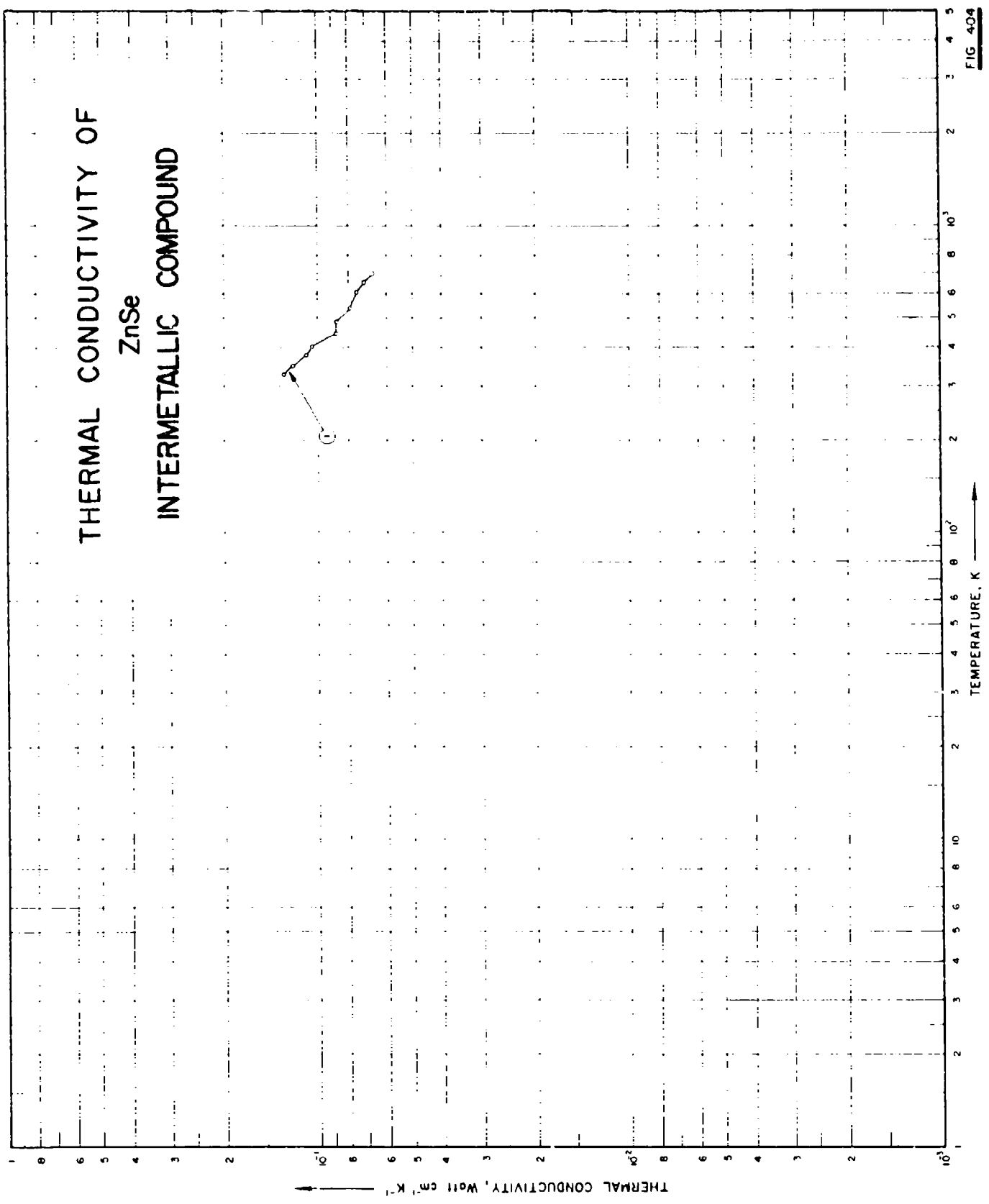
T      k

CURVE 1:

303.2    0.020

No graphical presentation

THERMAL CONDUCTIVITY OF  
ZnSe  
INTERMETALLIC COMPOUND



## SPECIFICATION TABLE NO. 404 THERMAL CONDUCTIVITY OF ZnS: INTERMETALLIC COMPOUNDS

[For Data Reported in Figure and Table No. 404]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
1	966		1961	327-695		ZnSe	Hot-pressed; density 5.267 g cm <sup>-3</sup> .

DATA TABLE NO. 405 THERMAL CONDUCTIVITY OF  $ZnSc$  INTERMETALLIC COMPOUNDS[Temperature, T, K; Thermal Conductivity, k, Watt  $cm^{-1} K^{-1}$ ]

T	k
<u>CURVE 1</u>	
327.2	0.130
348.2	0.121
376.2	0.109
402.2	0.105
444.2	0.0879
489.2	0.0879
536.2	0.0795
572.2	0.0796
608.2	0.0753
650.2	0.0711
695.2	0.0669

SPECIFICATION TABLE NO. 405 THERMAL CONDUCTIVITY OF  $ZnSiAs_2$  INTERMETALLIC COMPOUNDS

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)	Specifications and Remarks
1	967	L	1966	298.2		$ZnSiAs_2$ ; S		Stoichiometric p-type polycrystalline; 12 x 4 x 4 mm; prepared by melting appropriate amounts of 99.999% pure Zn, p-type Si ( $\sim 500 \text{ ohm cm}$ ), and 99.999 pure As in a carbon coated fused quartz tube which was filled with argon at 1 atm, heated at 75 C $hr^{-1}$ until specimen completely molten, annealed at $\sim 1136 \text{ C}$ for 1 J, vibrated at 100 cps from about 400 C to the maximum temperature to remove blow holes and voids and insure complete mixing, then the Bridgman process was performed at a lowering rate of 3 mm $hr^{-1}$ until room temperature reached; melting temperature 1096 C; electrical resistivity $4.5 \times 10^2 \text{ ohm cm}$ at room temperature.

DATA TABLE NO. 405 THERMAL CONDUCTIVITY OF  $ZnSiAs_2$  INTERMETALLIC COMPOUNDS

[Temperature, T, K; Thermal Conductivity, k,  $\text{Watt cm}^{-1}\text{K}^{-1}$ ]

T k  
 CURVE 1<sup>†</sup>  
 298.2 0.14

<sup>†</sup> No graphical presentation

# THERMAL CONDUCTIVITY OF Zr B INTERMETALLIC COMPOUND

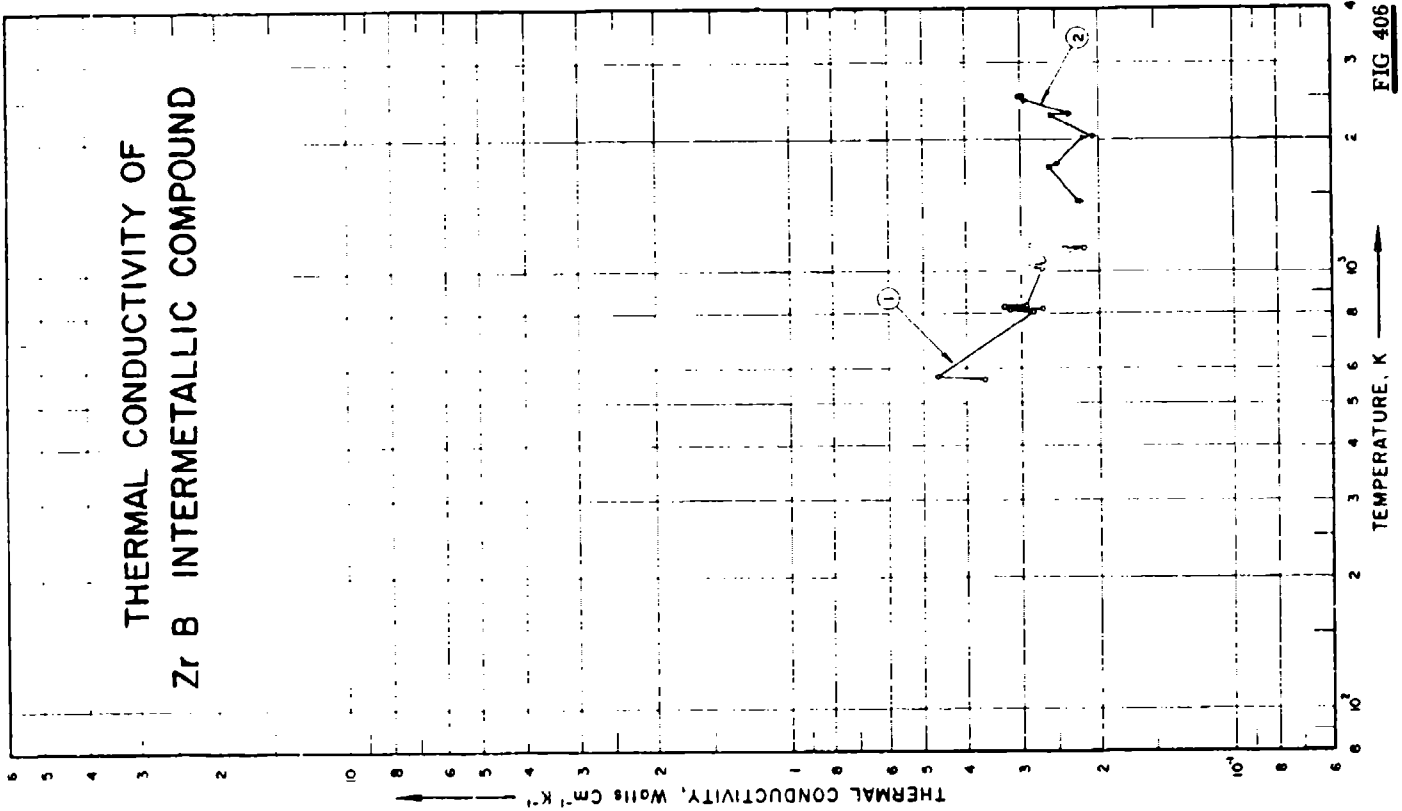


FIG 406

SPECIFICATION TABLE NO. 406 THERMAL CONDUCTIVITY OF ZrB INTERMETALLIC COMPOUNDS

[ For Data Reported in Figure and Table No. 406 ]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition ( weight percent), Specifications and Remarks
1	603	R	1962	568-1134	3	Zr B; C-52	78.7 Zr, 17.6 B, and 0.30 C, balance Ti, Fe, Ni, Ca, Al, and Si; specimen composed of Zr B plus Zr B <sub>2</sub> ; dimensions 0.75 in. O. D., 0.25 in. I. D., 0.75 in. long; supplied by Norton Company; hot pressed; density 4.13 g cm <sup>-3</sup> (70-75% of theoretical density). Similar to the above specimen except found melted after test.
2	603	R	1962	1442-2528	3	Zr B; C-60	

## DATA TABLE NO. 406 THERMAL CONDUCTIVITY OF ZrB INTERMETALLIC COMPOUNDS

[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup>K<sup>-1</sup>]

T k

CURVE 1

567.5	0.365
575.4	0.467
805.9	0.283
811.5	0.315
818.7	0.268
822.1	0.326
824.3	0.294
1012.1	0.270
1025.4	0.294
1035.9	0.264
1107.1	0.274
1116.5	0.232
1126.5	0.216
1134.3	0.237

CURVE 2

1441.5	0.219
1444.3	0.224
1727.6	0.257
1731.8	0.255*
1756.5	0.247
2038.7	0.216
2038.7	0.212*
2038.7	0.205
2272.1	0.255
2274.8	0.231
2456.5	0.294
2516.5	0.304
2527.6	0.296

\* Not shown on plot



SPECIFICATION TABLE NO. 407 THERMAL CONDUCTIVITY OF (Sb<sub>2</sub>Se<sub>3</sub> + Ag<sub>2</sub>Se + PbSe) MIXTURES

(Sb<sub>2</sub>Se<sub>3</sub> + Ag<sub>2</sub>Se + PbSe : 95.0%; Impurity < 2.0% each)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
1	971		1962	298.2		Ag <sub>2</sub> Sb <sub>2</sub> PbSe <sub>3</sub>	55.19 Sb <sub>2</sub> Se <sub>3</sub> , 33.85 Ag <sub>2</sub> Se, and 10.96 PbSe; p-type; specimen obtained by fusing ASARCO 99.999 pure elements in carbon coated quartz tube with agitation, cooling, crushing, recasting in 8 mm uncoated quartz, and zone-leveling; electrical resistivity 0.21 ohm cm, (measuring temperature assumed 25 C).

DATA TABLE NO. 407 THERMAL CONDUCTIVITY OF (Sb<sub>2</sub>Se<sub>3</sub> + Ag<sub>2</sub>Se + PbSe) MIXTURES

(Sb<sub>2</sub>Se<sub>3</sub> + Ag<sub>2</sub>Se + PbSe : 95.0%; impurity < 2.0% each)

[ Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup>K<sup>-1</sup> ]

T k  
 CURVE 1<sup>o</sup>  
 298.2 0.0058

No graphical presentation

# THERMAL CONDUCTIVITY OF (Sb<sub>2</sub>Te<sub>3</sub> + Bi<sub>2</sub>Te<sub>3</sub>) MIXTURES

FIGURE SHOWS ONLY 22 OF THE CURVES REPORTED IN TABLE

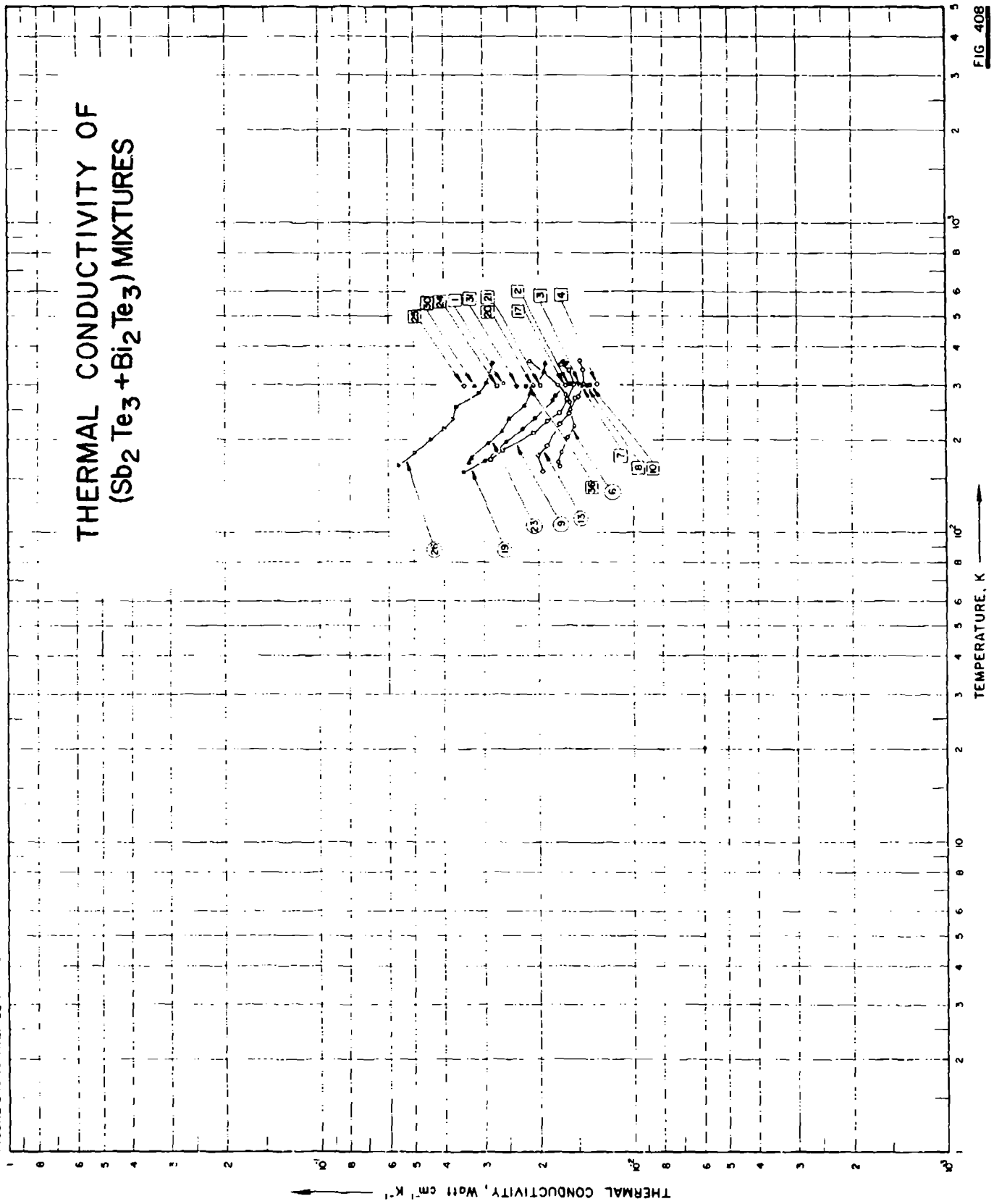


FIG 408

SPECIFICATION TABLE NO. 408 THERMAL CONDUCTIVITY OF  $(\text{Sb}_2\text{Te}_3 + \text{Bi}_2\text{Te}_3)$  MIXTURES $(\text{Sb}_2\text{Te}_3 + \text{Bi}_2\text{Te}_3) > 95.0\%$ ; impurity  $< 2.0\%$  each

[ For Data Reported in Figure and Table No. 408 ]

Curve No.	Method Used	Year	Temp. Range, K	Temp. Reported Error, %	Name and Specimen Designation	Composition $\text{Sb}_2\text{Te}_3$	Composition (weight percent) $\text{Bi}_2\text{Te}_3$	Composition (continued), Specifications and Remarks
1	832	1962	303.2		$\text{Sb}_{1.8}\text{Bi}_{0.2}\text{Te}_3$	87.6	12.4	p-type; prepared by using the powder metallurgical techniques, pressed and sintered; heat flow perpendicular to the pressing direction.
2	832	1962	303.2		$\text{Sb}_{1.7}\text{Bi}_{0.3}\text{Te}_3$	80.1	19.9	Similar to the above specimen.
3	832	1962	303.2		$\text{Sb}_{1.6}\text{Bi}_{0.4}\text{Te}_3$	75.8	24.2	Similar to the above specimen.
4	832	1962	303.2		$\text{Sb}_{1.5}\text{Bi}_{0.5}\text{Te}_3$	70.1	29.9	Similar to the above specimen.
5	936	T	298.2		$\text{Sb}_{1.2}\text{Bi}_{0.8}\text{Te}_{3.13}$	52.73	44.94	2.33 Te (calculated) p-type; specimen size $0.5 \times 0.5 \times 1$ cm; prepared from 99.999 pure Bi and Sb supplied by consolidated Mining and Smelting Co., and from 99.97 pure Te supplied by Canadian Copper Refiners Ltd., materials weighed out, crushed, sealed in an ampule in a vacuum of $10^{-5}$ Torr., heated at 900 C for 20 hrs., rocked, cooled, then zone-melted at a rate of $0.07 \sim 0.28$ in. $\text{hr}^{-1}$ , cooled and cut; thermal conductivity data calculated from measured values of figure of merit, Seebeck coefficient, and electrical conductivity; electrical conductivity reported as $0.83 \times 10^3 \text{ ohm}^{-1}\text{cm}^{-1}$ at room temperature.
6	936	T	1965	164-337	$\text{Sb}_{1.2}\text{Bi}_{0.8}\text{Te}_{3.13}$	52.73	44.94	p-type; same fabrication method and measuring method as 2.33 Te (calculated); the above specimen; electrical resistivity reported as 0.116, 0.121, 0.135, 0.156, 0.185, 0.225, 0.259, 0.284, 0.315, 0.352, and 0.356 milliohm cm at 166, 172, 185, 203, 221, 245, 267, 282, 301, 333, and 357 K, respectively.
7	936	T	1965	298.2	$\text{Sb}_{1.33}\text{Bi}_{0.67}\text{Te}_{3.13}$	59.56	38.07	2.37 Te (calculated); p-type; same fabrication method and measuring method as the above specimen; electrical conductivity reported as $0.77 \times 10^3 \text{ ohm}^{-1}\text{cm}^{-1}$ at room temperature.
8	936	T	1965	298.2	$\text{Sb}_{1.33}\text{Bi}_{0.67}\text{Te}_{3.13}$			Cut from the same ingot as the above specimen; electrical conductivity reported on $0.69 \times 10^3 \text{ ohm}^{-1}\text{cm}^{-1}$ at room temperature.
9	936	T	1965	172-351	$\text{Sb}_{1.33}\text{Bi}_{0.67}\text{Te}_{3.13}$			Same composition, fabrication method, and measuring method as the above specimen; electrical resistivity 0.472, 0.509, 0.661, 0.791, 0.910, 1.10, 1.41, 1.76, 1.95, and 2.04 milliohm cm at 173, 185, 212, 232, 248, 269, 301, 335, 351, and 360 K, respectively.
10	936	T	1965	293.2	$\text{Sb}_{1.4}\text{Bi}_{0.6}\text{Te}_{3.13}$	63.06	34.55	2.39 Te (calculated); p-type; same fabrication method and measuring method as the above specimen; electrical conductivity reported as $0.83 \times 10^3 \text{ ohm}^{-1}\text{cm}^{-1}$ at room temperature.

SPECIFICATION TABLE NO. 404 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (wt percent) $Sb_2Te_3$	Composition (wt percent) $Bi_2Te_3$	Composition (continued), Specifications and Remarks
11	936	T	1965	298.2		$Sb_{1.4}Bi_{0.6}Te_{3.13}$			Cut from the same ingot as the above specimen; electrical conductivity reported as $0.89 \times 10^2 \text{ ohm}^{-1}\text{cm}^{-1}$ at room temperature.
12	936	T	1965	298.2		$Sb_{1.4}Bi_{0.6}Te_{3.13}$			Similar to the above specimen except electrical conductivity reported as $0.87 \times 10^2 \text{ ohm}^{-1}\text{cm}^{-1}$ at room temperature.
13	936	T	1965	158-360		$Sb_{1.4}Bi_{0.6}Te_{3.13}$			Similar to the above specimen except electrical resistivity reported as 0.354, 0.404, 0.467, 0.637, 0.776, 0.917, 1.17, 1.45, and 1.65 milliohm cm at 159, 178, 192, 224, 251, 273, 305, 337, and 357 K, respectively.
14	936	T	1965	298.2		$Sb_{1.5}Bi_{0.5}Te_{3.13}$	68.42	29.16	2.42 Te (calculated); p-type; same fabrication method and measuring method as the above specimen; electrical conductivity reported as $1.13 \times 10^3 \text{ ohm}^{-1}\text{cm}^{-1}$ at room temperature.
15	936	T	1965	298.2		$Sb_{1.5}Bi_{0.5}Te_{3.13}$			Another run of the above specimen; electrical conductivity reported as $1.39 \times 10^3 \text{ ohm}^{-1}\text{cm}^{-1}$ at room temperature.
16	936	T	1965	298.2		$Sb_{1.5}Bi_{0.5}Te_{3.13}$	68.42	29.16	2.42 Te (calculated); cut from the same ingot as the above specimen; electrical conductivity reported as $1.18 \times 10^3 \text{ ohm}^{-1}\text{cm}^{-1}$ at room temperature.
17	936	T	1965	298.2		$Sb_{1.5}Bi_{0.5}Te_{3.13}$	68.42	29.16	2.42 Te (calculated); similar to the above specimen except electrical conductivity reported as $1.22 \times 10^3 \text{ ohm}^{-1}\text{cm}^{-1}$ at room temperature.
18	936	T	1965	298.2		$Sb_{1.5}Bi_{0.5}Te_{3.13}$			Similar to the above specimen except electrical conductivity reported as $1.17 \times 10^3 \text{ ohm}^{-1}\text{cm}^{-1}$ at room temperature.
19	936	T	1965	158-358		$Sb_{1.5}Bi_{0.5}Te_{3.13}$	68.42	29.16	2.42 Te (calculated); p-type; same fabrication method and measuring method as the above specimen; electrical resistivity reported as 0.269, 0.306, 0.376, 0.461, 0.539, 0.689, 0.746, 0.859, 1.00, and 1.19 milliohm cm at 159, 173, 197, 219, 237, 269, 281, 301, 329, and 352 K, respectively.
20	936	T	1965	298.2		$Sb_{1.5}Bi_{0.5}Te_{3.13}$	73.92	23.63	2.45 Te (calculated); p-type; same fabrication method and measuring method as the above specimen; electrical conductivity reported as $1.73 \times 10^3 \text{ ohm}^{-1}\text{cm}^{-1}$ at room temperature.

SPECIFICATION TABLE NO. 408 (continued)

Curve No.	Ref. No. Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition $\text{Sb}_2\text{Te}_3$	Composition (weight percent) $\text{Bi}_2\text{Te}_3$	Composition (continued), Specifications and Remarks
21	936	T	298.2		$\text{Sb}_{1.6}\text{Bi}_{0.4}\text{Te}_3$ , 13	73.92	24.63	2.45 Te (calculated); cut from the same ingot as the above specimen; electrical conductivity reported as $1.92 \times 10^3 \text{ ohm}^{-1} \text{ cm}^{-1}$ at room temperature.
22	936	T	298.2		$\text{Sb}_{1.6}\text{Bi}_{0.4}\text{Te}_3$ , 13	73.92	24.63	2.45 Te (calculated); similar to the above specimen except electrical conductivity reported as $1.82 \times 10^3 \text{ ohm}^{-1} \text{ cm}^{-1}$ at room temperature.
23	936	T	165-353		$\text{Sb}_{1.6}\text{Bi}_{0.4}\text{Te}_3$ , 13	73.92	24.63	2.45 Te (calculated); similar to the above specimen except electrical resistivity reported as 0.210, 0.226, 0.255, 0.293, 0.337, 0.400, 0.462, 0.513, 0.618, and 0.692 milliohm cm at 169, 176, 195, 213, 232, 256, 282, 300, 335, and 354 K, respectively.
24	936	T	298.2		$\text{Sb}_{1.6}\text{Bi}_{0.4}\text{Te}_3$ , 13	83.36	12.13	2.51 Te (calculated); p-type; same fabrication method and measuring method as the above specimen; electrical conductivity reported as $2.44 \times 10^3 \text{ ohm}^{-1} \text{ cm}^{-1}$ at room temperature.
25	936	T	298.2		$\text{Sb}_{1.6}\text{Bi}_{0.4}\text{Te}_3$ , 13			Cut from the same ingot as the above specimen; electrical conductivity reported as $3.12 \times 10^3 \text{ ohm}^{-1} \text{ cm}^{-1}$ at room temperature.
26	936	T	167-356		$\text{Sb}_{1.6}\text{Bi}_{0.4}\text{Te}_3$ , 13			Similar to the above specimen; electrical resistivity reported as 0.136, 0.150, 0.168, 0.188, 0.207, 0.229, 0.272, 0.304, 0.356, and 0.385 milliohm cm at 168, 183, 200, 217, 233, 255, 284, 304, 337, and 354 K, respectively.
27	936	T	298.2		$\text{Sb}_{1.4}\text{Bi}_{0.6}\text{Te}_3$ , 15	63.48	35.00	1.12 Te (calculated); p-type; same fabrication method and measuring method as the above specimen; electrical conductivity reported as $1.15 \times 10^3 \text{ ohm}^{-1} \text{ cm}^{-1}$ at room temperature.
28	936	T	298.2		$\text{Sb}_{1.4}\text{Bi}_{0.6}\text{Te}_3$ , 15	62.37	34.16	3.45 Te (calculated); p-type; same fabrication method and measuring method as the above specimen; electrical conductivity reported as $0.85 \times 10^3 \text{ ohm}^{-1} \text{ cm}^{-1}$ at room temperature.

SPECIFICATION TABLE NO. 404 (continued)

Curve No.	Ref. Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) $Sb_2Te_3$	Composition (weight percent) $Bi_2Te_3$	Composition (continued), Specifications and Remarks
29	936 T	1965	298.2		$Sb_{1.4}Bi_{0.5}Te_{3.26}$	61.59	23.75	4.66 Te (calculated); p-type; same fabrication method and measuring method as the above specimen; electrical conductivity reported as $0.88 \times 10^3 \text{ ohm}^{-1} \text{ cm}^{-1}$ at room temperature.
30	936 T	1965	298.2		$Sb_{1.5}Bi_{0.5}Te_3$	70.12	29.88	p-type; same fabrication method and measuring method as the above specimen; electrical conductivity reported as $3.22 \times 10^3 \text{ ohm}^{-1} \text{ cm}^{-1}$ at room temperature.
31	936 T	1965	298.2		$Sb_{1.5}Bi_{0.5}Te_{3.06}$	69.32	29.55	1.13 Te (calculated); p-type; same fabrication method and measuring method as the above specimen; electrical conductivity reported as $1.72 \times 10^3 \text{ ohm}^{-1} \text{ cm}^{-1}$ at room temperature.
32	936 T	1965	298.2		$Sb_{1.5}Bi_{0.5}Te_{3.19}$	67.67	28.84	3.49 Te (calculated); p-type; same fabrication method and measuring method as the above specimen; electrical conductivity reported as $1.24 \times 10^3 \text{ ohm}^{-1} \text{ cm}^{-1}$ at room temperature.
33	936 T	1965	298.2		$Sb_{1.5}Bi_{0.5}Te_{3.26}$	66.31	28.47	4.72 Te (calculated); p-type; same fabrication method and measuring method as the above specimen; electrical conductivity reported as $1.19 \times 10^3 \text{ ohm}^{-1} \text{ cm}^{-1}$ at room temperature.
34	936 T	1965	298.2		$Sb_{1.6}Bi_{0.4}Te_{3.06}$	74.91	23.94	1.15 Te (calculated); p-type; same fabrication method and measuring method as the above specimen; electrical conductivity reported as $1.52 \times 10^3 \text{ ohm}^{-1} \text{ cm}^{-1}$ at room temperature.
35	936 T	1965	298.2		$Sb_{1.6}Bi_{0.4}Te_{3.14}$	73.10	23.36	3.54 Te (calculated); p-type; same fabrication method and measuring method as the above specimen; electrical conductivity reported as $1.74 \times 10^3 \text{ ohm}^{-1} \text{ cm}^{-1}$ at room temperature.
36	936 T	1965	298.2		$Sb_{1.6}Bi_{0.4}Te_{3.26}$	72.16	23.06	4.78 Te (calculated); p-type; same fabrication method and measuring method as the above specimen; electrical conductivity reported as $1.78 \times 10^3 \text{ ohm}^{-1} \text{ cm}^{-1}$ at room temperature.

DATA TABLE NO. 408 THERMAL CONDUCTIVITY OF ( $Sb_2Te_3 + Bi_2Te_3$ ) MIXTURES( $Sb_2Te_3 + Bi_2Te_3$ : 95.0%; impurity  $\pm 2.0\%$  each)(Temperature, T, K, Thermal Conductivity, k, Watt  $cm^{-1}K^{-1}$ )

T	k	T	k	T	k	T	k	T	k
<u>CURVE 1</u>		<u>CURVE 9</u>		<u>CURVE 16*</u>		<u>CURVE 23 (cont.)</u>		<u>CURVE 31</u>	
303.2	0.026	172.1	0.0246	298.2	0.0158	232.0	0.0250	298.2	0.0236
<u>CURVE 2</u>		184.2	0.0261	<u>CURVE 17</u>		257.1	0.0222	<u>CURVE 32*</u>	
303.2	0.016	211.9	0.0207	298.2	0.0166	281.7	0.0213	298.2	0.0181
<u>CURVE 3</u>		230.4	0.0187	<u>CURVE 18*</u>		301.2	0.0206*	<u>CURVE 33*</u>	
303.2	0.015	246.9	0.0173	298.2	0.0156	331.1	0.0193	298.2	0.0181
<u>CURVE 4</u>		269.5	0.0163	301.2	0.0156	353.4	0.0190	<u>CURVE 34*</u>	
303.2	0.013	335.6	0.0161	298.2	0.0156	<u>CURVE 24</u>		298.2	0.0181
<u>CURVE 5*</u>		350.9	0.0170	<u>CURVE 19</u>		298.2	0.0272	<u>CURVE 35*</u>	
298.2	0.0182	298.2	0.0137	157.7	0.0348	<u>CURVE 25</u>		298.2	0.0192
<u>CURVE 6</u>		<u>CURVE 11*</u>		171.8	0.0298	298.2	0.0346	<u>CURVE 36</u>	
164.2	0.0172	298.2	0.0145	196.1	0.0255	<u>CURVE 26</u>		298.2	0.0164
170.6	0.0174	<u>CURVE 12*</u>		217.9	0.0226	166.7	0.0560	<u>CURVE 27*</u>	
183.8	0.0169	298.2	0.0141	235.9	0.0205	181.8	0.0496	298.2	0.0219
202.0	0.0162	<u>CURVE 13</u>		268.8	0.0181	200.0	0.0440	<u>CURVE 28*</u>	
221.2	0.0155	157.7	0.0194	278.6	0.0178	217.4	0.0399	298.2	0.0160
249.4	0.0160	177.0	0.0200	299.4	0.0168*	232.6	0.0374	<u>CURVE 29*</u>	
264.6	0.0160	221.2	0.0155	327.9	0.0167*	255.7	0.0366	298.2	0.0160
280.9	0.0165	249.4	0.0160	352.1	0.0164	283.3	0.0309	<u>CURVE 30</u>	
301.2	0.0175	271.6	0.0154	358.4	0.0169	304.0	0.0295	298.2	0.0320
357.1	0.0212	275.5	0.0150	<u>CURVE 20</u>		355.8	0.0280	<u>CURVE 31</u>	
<u>CURVE 7</u>		303.0	0.0144	298.2	0.0208	<u>CURVE 27*</u>		298.2	0.0152
298.2	0.0146	336.7	0.0144	<u>CURVE 21</u>		298.2	0.0152	<u>CURVE 28*</u>	
<u>CURVE 8</u>		359.7	0.0148	298.2	0.0198	<u>CURVE 22*</u>		298.2	0.0160
298.2	0.0141	<u>CURVE 14*</u>		298.2	0.0202	<u>CURVE 29*</u>		298.2	0.0160
<u>CURVE 9</u>		298.2	0.0147	<u>CURVE 23</u>		168.4	0.0334	<u>CURVE 30</u>	
298.2	0.0141	175.1	0.0327	168.4	0.0334	175.1	0.0327	298.2	0.0320
<u>CURVE 10</u>		194.2	0.0291	175.1	0.0327	194.2	0.0291	<u>CURVE 31</u>	
298.2	0.0178	212.3	0.0264	212.3	0.0264	212.3	0.0264	<u>CURVE 32*</u>	

\* Not shown on plot

SPECIFICATION TABLE NO. 409 THERMAL CONDUCTIVITY OF  $(\text{Sb}_2\text{Te}_3 + \text{In}_2\text{Te}_3)$  MIXTURES  
 $(\text{Sb}_2\text{Te}_3 + \text{In}_2\text{Te}_3 = 95.0\%$ ; impurity =  $2.0\%$  each)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent). Specifications and Remarks
1	833		1962	303.2		$\text{Sb}_2\text{Te}_3 - \text{In}_2\text{Te}_3$ ; No. 338	10 mole % $\text{In}_2\text{Te}_3$ ; p-type alloy; pellet was prepared by powder metallurgical techniques; annealed for 8 hrs at 475 C; pellet size = .50 in. long and .50 in. in dia; type of heat - Vycor tube.
2	833		1962	303.2		$\text{Sb}_2\text{Te}_3 - \text{In}_2\text{Te}_3$ ; No. 76	18 mole % $\text{In}_2\text{Te}_3$ ; the others are the same as that of the above specimen.
3	833		1962	303.2		$\text{Sb}_2\text{Te}_3 - \text{In}_2\text{Te}_3$ ; No. 59	25 mole % $\text{In}_2\text{Te}_3$ ; the others are the same as that of the above specimen.
4	833		1962	303.2		$\text{Sb}_2\text{Te}_3 - \text{In}_2\text{Te}_3$ ; No. 10	33.3 mole % $\text{In}_2\text{Te}_3$ ; the others are the same as that of the above specimen.
5	833		1962	303.2		$\text{Sb}_2\text{Te}_3 - \text{In}_2\text{Te}_3$ ; No. 3384	5 mole % $\text{In}_2\text{Te}_3$ ; p-type alloy; pellet was prepared by powder metallurgical techniques; cold pressed and sintered; pellet size = .50 in. long and .50 in. in dia; type of heat - Balzers furnace.
6	833		1962	303.2		$\text{Sb}_2\text{Te}_3 - \text{In}_2\text{Te}_3$ ; No. 3421	6 mole % $\text{In}_2\text{Te}_3$ ; the others are the same as that of the above specimen.
7	833		1962	303.2		$\text{Sb}_2\text{Te}_3 - \text{In}_2\text{Te}_3$ ; No. 3423	7 mole % $\text{In}_2\text{Te}_3$ ; the others are the same as that of the above specimen.
8	833		1962	303.2		$\text{Sb}_2\text{Te}_3 - \text{In}_2\text{Te}_3$ ; No. 3386	10 mole % $\text{In}_2\text{Te}_3$ ; the others are the same as that of the above specimen.
9	833		1962	303.2		$\text{Sb}_2\text{Te}_3 - \text{In}_2\text{Te}_3$ ; No. 3330	15 mole % $\text{In}_2\text{Te}_3$ ; the others are the same as that of the above specimen.



DATA TABLE NO. 409 THERMAL CONDUCTIVITY OF (Sb<sub>2</sub>Te<sub>3</sub> + In<sub>2</sub>Te<sub>3</sub>) MIXTURES  
 (Sb<sub>2</sub>Te<sub>3</sub> + In<sub>2</sub>Te<sub>3</sub> = 95.0%; impurity = 2.0% each)

[ Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup>K<sup>-1</sup> ]

T	k
<u>CURVE 1<sup>a</sup></u>	
303.2	0.0152
<u>CURVE 2<sup>a</sup></u>	
303.2	0.0094
<u>CURVE 3<sup>a</sup></u>	
303.2	0.0083
<u>CURVE 4<sup>a</sup></u>	
303.2	0.0083
<u>CURVE 5<sup>a</sup></u>	
303.2	0.0175
<u>CURVE 6<sup>a</sup></u>	
303.2	0.0159
<u>CURVE 7<sup>a</sup></u>	
303.2	0.0153
<u>CURVE 8<sup>a</sup></u>	
303.2	0.0112
<u>CURVE 9<sup>a</sup></u>	
303.2	0.0107

No graphical presentation

# THERMAL CONDUCTIVITY OF (Bi<sub>2</sub>Te<sub>3</sub>+Sb<sub>2</sub>Te<sub>3</sub>) MIXTURES

FIGURE SHOWS ONLY 9 OF THE CURVES REPORTED IN TABLE

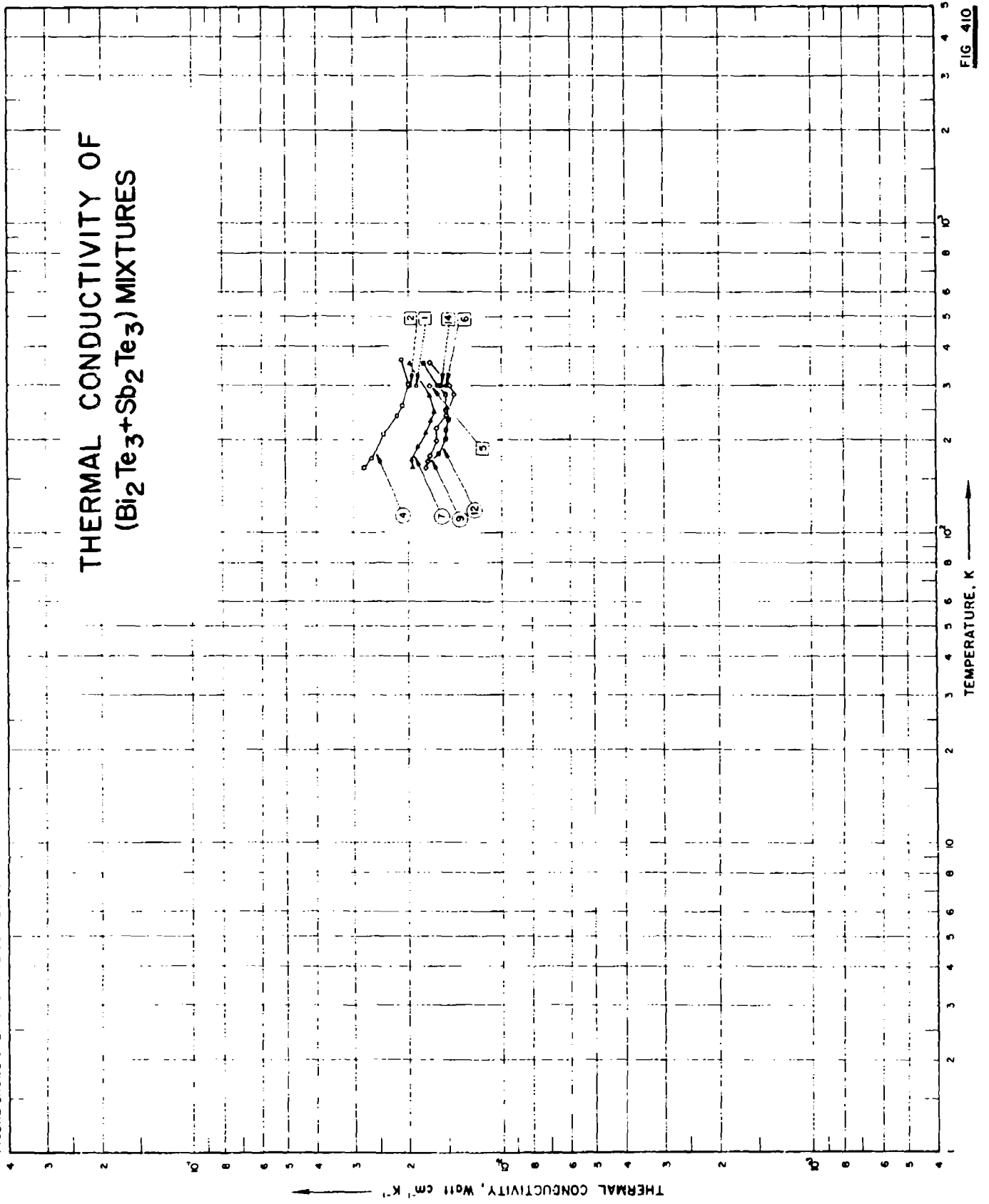


FIG 410

SPECIFICATION TABLE NO. 410 THERMAL CONDUCTIVITY OF (Bi<sub>2</sub>Te<sub>3</sub> + Sb<sub>2</sub>Te<sub>3</sub>) MIXTURES(Bi<sub>2</sub>Te<sub>3</sub> + Sb<sub>2</sub>Te<sub>3</sub> = 95.0%; impurity = 2.0% each)

(For Data Reported in Figure and Table No. 410)

Curve No.	Ref. No. Used	Method	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) Bi <sub>2</sub> Te <sub>3</sub>	Composition (weight percent) Sb <sub>2</sub> Te <sub>3</sub>	Composition (continued), Specifications and Remarks
1	936	T	1965	298.2		Bi <sub>1.35</sub> Sb <sub>0.65</sub> Te <sub>3.10</sub>	88.07	9.84	2.09 Te (calculated); n-type; $0.5 \times 0.5 \times 1$ cm; prepared from 99.999 pure Bi and Sb supplied by Consolidated Mining and Smelting Co., and 99.97 pure Te supplied by Canadian Copper Refiners Ltd.; materials weighed out, crushed, sealed in an ampule in a vacuum of $10^{-5}$ Torr, heated at 900 C for 20 hrs., rocked, cooled, zone-melted at a rate of 0.07-0.28 in./hr <sup>-1</sup> , cooled and cut; thermal conductivity data calculated from measured values of figure of merit, Seebeck coefficient, and electrical conductivity; electrical conductivity reported as $1.19 \times 10^3$ ohm <sup>-1</sup> cm <sup>-1</sup> at room temperature.
2	936	T	1965	298.2		Bi <sub>1.35</sub> Sb <sub>0.65</sub> Te <sub>3.10</sub>			Cut from the same ingot as the above specimen; electrical conductivity reported as $1.39 \times 10^3$ ohm <sup>-1</sup> cm <sup>-1</sup> at room temperature.
3	936	T	1965	298.2		Bi <sub>1.35</sub> Sb <sub>0.65</sub> Te <sub>3.10</sub>			Similar to the above specimen except electrical conductivity reported as $1.20 \times 10^3$ ohm <sup>-1</sup> cm <sup>-1</sup> at room temperature.
4	936	T	1965	163-362		Bi <sub>1.35</sub> Sb <sub>0.65</sub> Te <sub>3.10</sub>			Similar to the above specimen except electrical resistivity reported as 311, 340, 423, 524, 379, 721, and 912 $\mu$ ohm cm at 163, 175, 209, 240, 259, 302, and 361 K, respectively.
5	936	T	1965	298.2		Bi <sub>1.35</sub> Sb <sub>0.65</sub> Te <sub>3.10</sub>	77.62	20.24	2.14 Te (calculated); n-type; same fabrication method and measuring method as the above specimen; electrical conductivity reported as $0.7 \times 10^3$ ohm <sup>-1</sup> cm <sup>-1</sup> at room temperature.
6	936	T	1965	298.2		Bi <sub>1.35</sub> Sb <sub>0.65</sub> Te <sub>3.10</sub>			Cut from the same ingot as the above specimen; electrical conductivity reported as $0.75 \times 10^3$ ohm <sup>-1</sup> cm <sup>-1</sup> at room temperature.
7	936	T	1965	162-353		Bi <sub>1.35</sub> Sb <sub>0.65</sub> Te <sub>3.10</sub>			Similar to the above specimen except electrical resistivity reported as 0.622, 0.643, 0.736, 0.843, 0.937, 1.03, 1.20, 1.31, and 1.55 milliohm cm at 169, 173, 191, 212, 242, 248, 278, 299, and 352 K, respectively.
8	936	T	1965	298.2		Bi <sub>1.35</sub> Sb <sub>0.65</sub> Te <sub>3.10</sub>	70.92	27.50	2.18 Te (calculated); n-type; same fabrication method and measuring method as the above specimen; electrical conductivity reported as $0.70 \times 10^3$ ohm <sup>-1</sup> cm <sup>-1</sup> at room temperature.

SPECIFICATION TABLE NO. 410 (continued)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) $\text{Bi}_2\text{Te}_3$	Composition (weight percent) $\text{Sb}_2\text{Te}_3$	Composition (continued), Specifications and Remarks
9	936	T	1965	164-356		$\text{Bi}_{1.33}\text{Sb}_{0.67}\text{Te}_{3.13}$			Similar to the above specimen except electrical resistivity reported as 0.689, 0.766, 0.853, 0.946, 1.10, 1.15, 1.32, 1.43, and 1.71 milliohm cm at 164, 179, 200, 220, 240, 252, 280, 301, and 354 K, respectively.
10	936	T	1965	298.2		$\text{Bi}_2\text{SbTe}_{3.13}$	54.84	42.89	2.27 Te (calculated); n-type; same fabrication method and measuring method as the above specimen; electrical conductivity reported as $0.43 \times 10^3 \text{ ohm}^{-1}\text{cm}^{-1}$ at room temperature.
11	936	T	1965	298.2		$\text{Bi}_2\text{SbTe}_{3.13}$			Another run of the above specimen.
12	936	T	1965	170-355		$\text{Bi}_2\text{SbTe}_{3.17}$			Similar to the above specimen except electrical resistivity reported as 1.27, 1.32, 1.49, 1.63, 1.80, 1.95, 2.24, 2.38, and 2.56 milliohm at 171, 181, 203, 218, 233, 250, 280, 302, and 356 K, respectively.
13	936	T	1965	298.2		$\text{Bi}_{1.33}\text{Sb}_{0.35}\text{Te}_{3.11}$	88.50	9.89	1.61 Te (calculated); n-type; same fabrication method and measuring method as the above specimen; electrical conductivity reported as $0.93 \times 10^3 \text{ ohm}^{-1}\text{cm}^{-1}$ at room temperature.
14	936	T	1965	298.2		$\text{Bi}_{1.33}\text{Sb}_{0.35}\text{Te}_{3.13}$	97.24	9.75	3.11 Te (calculated); n-type; same fabrication method and measuring method as the above specimen; electrical conductivity reported as $1.41 \times 10^3 \text{ ohm}^{-1}\text{cm}^{-1}$ at room temperature.
15	936	T	1965	298.2		$\text{Bi}_{1.33}\text{Sb}_{0.35}\text{Te}_{3.26}$	86.28	9.64	4.08 Te (calculated); n-type; same fabrication method and measuring method as the above specimen; electrical conductivity reported as $1.46 \times 10^3 \text{ ohm}^{-1}\text{cm}^{-1}$ at room temperature.



SPECIFICATION TABLE NO. 411 THERMAL CONDUCTIVITY OF  $(\text{Bi}_2\text{Te}_3 + \text{Sb}_2\text{Te}_3 + \text{Sb}_2\text{Se}_3)$  MIXTURES  
 $(\text{Bi}_2\text{Te}_3 + \text{Sb}_2\text{Te}_3 + \text{Sb}_2\text{Se}_3 - 95.0\%$ ; impurity  $\approx 2.0\%$  each)

Curve No.	Ref. Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)			Composition (continued), Specifications and Remarks
						$\text{Bi}_2\text{Te}_3$	$\text{Sb}_2\text{Te}_3$	$\text{Sb}_2\text{Se}_3$	
1	947	I.	1960	300		50	40	10	p-type; doped with excess Bi; electrical resistivity 0.8 milliohm cm at room temperature.

DATA TABLE NO. 411 THERMAL CONDUCTIVITY OF  $(\text{Bi}_2\text{Te}_3 + \text{Sb}_2\text{Te}_3 + \text{Sb}_2\text{Se}_3)$  MIXTURES  
 $(\text{Bi}_2\text{Te}_3 + \text{Sb}_2\text{Te}_3 + \text{Sb}_2\text{Se}_3 - 95.0\%$ ; impurity  $\approx 2.0\%$  each)

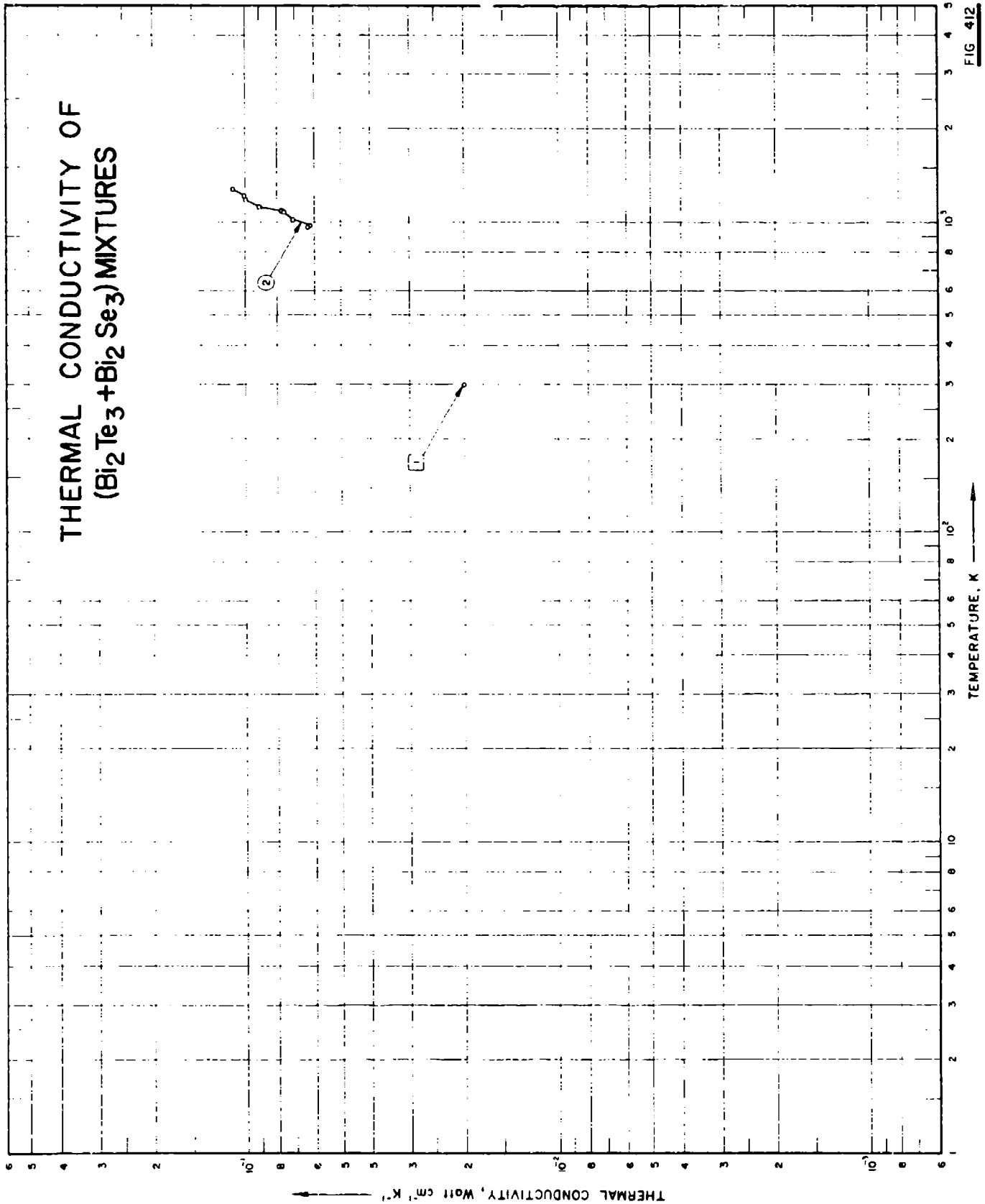
[ Temperature, T, K; Thermal Conductivity, k, Watt  $\text{cm}^{-1}\text{K}^{-1}$  ]

T k

CURVE 1<sup>2</sup>

300 0.015

THERMAL CONDUCTIVITY OF  
(Bi<sub>2</sub>Te<sub>3</sub> + Bi<sub>2</sub>Se<sub>3</sub>) MIXTURES



SPECIFICATION TABLE NO. 412 THERMAL CONDUCTIVITY OF  $(\text{Bi}_2\text{Te}_3 + \text{Bi}_2\text{Se}_3)$  MIXTURES $(\text{Bi}_2\text{Te}_3 + \text{Bi}_2\text{Se}_3)$  95.0%; impurity  $\pm 2.0\%$  each

[For Data Reported in Figure and Table No. 412.]

Curve No.	Ref. No. Used	Method	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition $(\text{Bi}_2\text{Te}_3 + \text{Bi}_2\text{Se}_3)$	Composition (continued), Specifications and Remarks	
1	947	I.	1960	300			78.6	21.4	n-type; doped with CuBr; electrical resistivity 0.6 milliohm cm at room temp.
2	974, 975	C	1965	971-1280	$\pm 12$		-	-	p0 $\text{Bi}_2\text{Te}_3$ ; 20 $\text{Bi}_2\text{Se}_3$ (weight percent or mole percent not specified); the apparatus comprises two coaxial graphite cylinders 20 mm I.D. and 50 mm O.D., separated by a gap 3 mm wide into which the powder specimen is poured, and the annular gap itself 140 mm long; before measurement of the apparatus was sealed up and heated at 600 K for 18-20 hrs then the entire system was placed in vacuum; bismuth used as comparative material.



DATA TABLE NO. 412 THERMAL CONDUCTIVITY OF  $(\text{Bi}_2\text{Te}_3 + \text{Bi}_2\text{Se}_3)$  MIXTURES $(\text{Bi}_2\text{Te}_3 + \text{Bi}_2\text{Se}_3 : 95.0\% \text{; impurity } \leq 2.0\% \text{ each})$ [Temperature, T, K; Thermal Conductivity, k, Watt  $\text{cm}^{-1}\text{K}^{-1}$ ]

T	k
<u>CURVE 1</u>	
300	0.020
<u>CURVE 2</u>	
971	0.0635
977	0.0621
1021	0.0712
1076	0.0763
1088	0.0772
1137	0.0902
1195	0.100
1227	0.102
1280	0.111

SPECIFICATION TABLE NO. 413 THERMAL CONDUCTIVITY OF  $(\text{Cd}_x\text{As}_2 + \text{Zn}_y\text{As}_2)$  MIXTURES $(\text{Cd}_x\text{As}_2 + \text{Zn}_y\text{As}_2 \sim 95.0\%$ ; impurity  $\sim 2.0\%$  each)

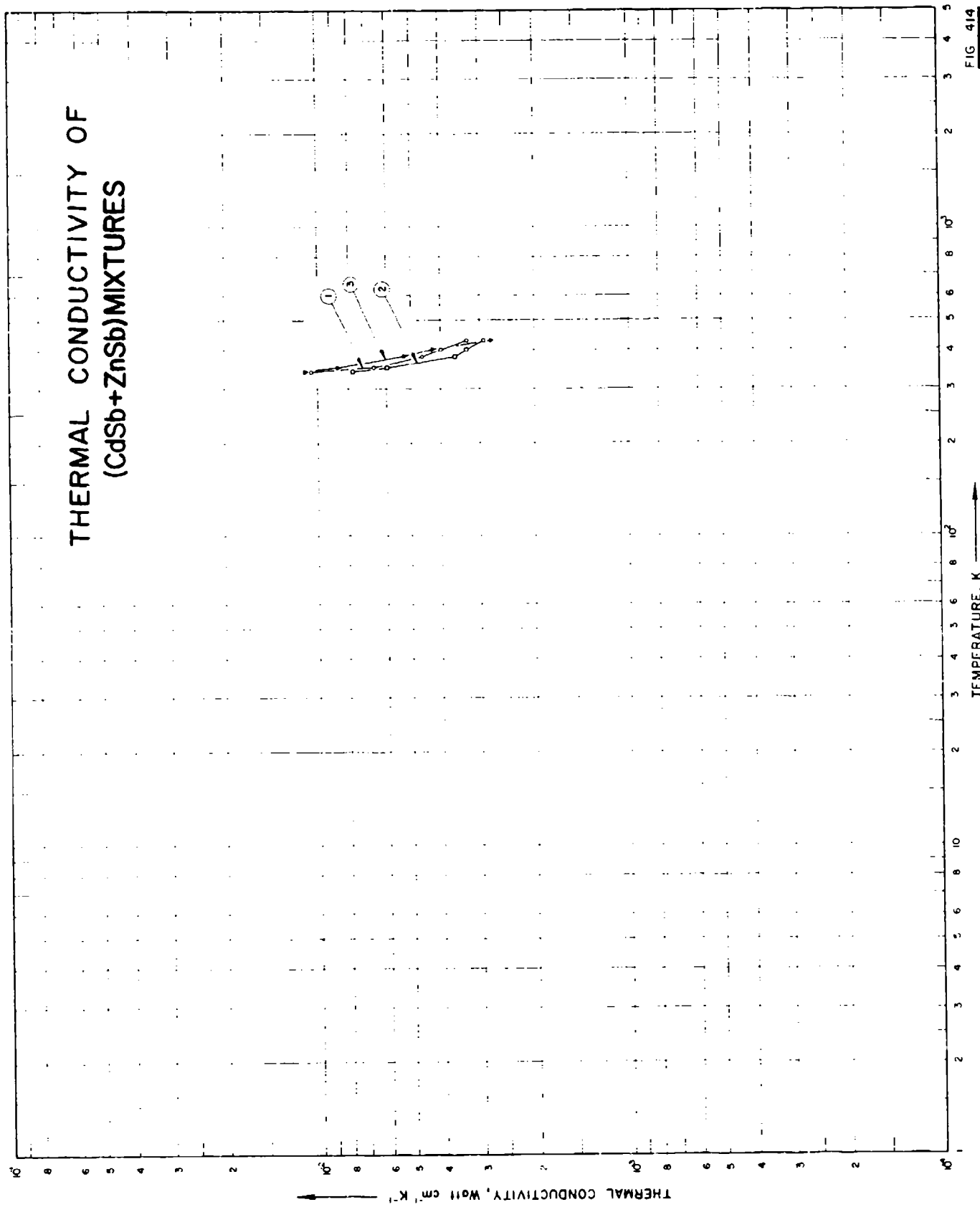
Curve No.	Ref. Method No. Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) $\text{CdAs}_2$	$\text{ZnAs}_2$	Composition (continued), Specifications and Remarks
1	972	1965	298.2		$\text{Cd}_{0.8}\text{Zn}_{0.2}\text{As}_2$ ; 1	87.6	12.4	Single crystal; 12 mm dia x 56 mm long; prepared by loading 99.9999 pure Cd, 99.9998 pure Zn, and 9.9995 pure As, in stoichiometric quantities, into a carbon-coated cylindrical quartz crucible, evacuated and sealed, heated till materials completely molten, rocked for a few hours, cooled at a rate of 15 C hr <sup>-1</sup> ; electrical resistivity 0.32 milliohm cm.
2	972	1965	298.2		$\text{Cd}_{0.5}\text{Zn}_{0.5}\text{As}_2$ ; 2	87.6	12.4	0.05 Cu; same dimensions and fabrication method as the above specimen; electrical resistivity 1.9 milliohm cm.
3	972	1965	298.2		$\text{Cd}_2\text{ZnAs}_2$ ; 3	73.7	26.3	Same dimensions and fabrication method as the above specimen; electrical resistivity 4.6 milliohm cm.
4	972	1965	298.2		$\text{Cd}_{1.6}\text{Zn}_{1.4}\text{As}_2$ ; 4	61.6	38.4	Similar to the above specimen; electrical resistivity 12 milliohm cm.

DATA TABLE NO. 413 THERMAL CONDUCTIVITY OF  $(\text{Cd}_x\text{As}_2 + \text{Zn}_y\text{As}_2)$  MIXTURES $(\text{Cd}_x\text{As}_2 + \text{Zn}_y\text{As}_2 \sim 95.0\%$ ; impurity  $\sim 2.0\%$  each)| Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup>K<sup>-1</sup>|

T	k	T	k
<u>CURVE 1</u>		<u>CURVE 3</u>	
298.2	0.023	298.2	0.012
<u>CURVE 2</u>		<u>CURVE 4</u>	
298.2	0.016	298.2	0.011

No graphical presentation

# THERMAL CONDUCTIVITY OF (CdSb+ZnSb) MIXTURES



SPECIFICATION TABLE NO. 414 THERMAL CONDUCTIVITY OF (CdSb + ZnSb) MIXTURES  
(CdSb + ZnSb  $\geq$  95.0%, impurity  $\leq$  2.0% each)

[For Data Reported in Figure and Table No. 414]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)		Composition (continued), Specifications and Remarks
							CdSb	ZnSb	
1	979, 980	C	1960	343-433		CdSb - ZnSb	55.6	44.4	Prepared from spectroscopically pure Zn, Sb, and Cd obtained by repeated vacuum distillation, materials weighed and melted in evacuated Pyrex containers at 620-630 C., vibrated, annealed at 450-550 C for 5 hrs and at 250 C for 8 hrs; electrical resistivity reported as 0.581 and 0.231 ohm cm at 70 and 130 C, respectively; measured in high vacuum.
2	979, 980	C	1960	343-433		3CdSb - 2ZnSb	65.3	34.7	Same fabrication method as the above specimen; electrical resistivity reported as 0.188 and 0.104 ohm cm at 70 and 130 C, respectively; measured in high vacuum.
3	979, 980	C	1960	343-433		7CdSb - 3ZnSb	74.5	25.5	Same fabrication method as the above specimen; electrical resistivity reported as 0.0749 and 0.0425 ohm cm at 70 and 130 C, respectively; measured in high vacuum.

## DATA TABLE NO. 414 THERMAL CONDUCTIVITY OF (CdSb + ZnSb) MIXTURES

(CdSb + ZnSb = 95.0%; Impurity = 2.0% each)

[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup>K<sup>-1</sup>]

T	k
<u>CURVE 1</u>	
343.2	0.0106
353.2	0.00651
383.2	0.00464
403.2	0.00402
433.2	0.00335
<u>CURVE 2</u>	
343.2	0.00782
353.2	0.00502
383.2	0.00364
403.2	0.00335
433.2	0.00293
<u>CURVE 3</u>	
343.2	0.0111
353.2	0.00874
383.2	0.00531
403.2	0.00427
433.2	0.00276

SPECIFICATION TABLE NO. 415 THERMAL CONDUCTIVITY OF  $(\text{CuSbSe}_2 + \text{Cu}_3\text{Se}_2)$  MIXTURES $(\text{CuSbSe}_2 + \text{Cu}_3\text{Se}_2)$ : 95.0%; impurity  $\pm 2.0\%$  each

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
1	971		1962	298.2		$(\text{CuSbSe}_2)_0.6(\text{Cu}_3\text{Se}_2)_0.4$	p-type; specimen obtained by fusing ASARCO 99.999 pure elements in carbon-coated quartz tube, with agitation, cooling, crushing, recasting in 8 mm uncoated quartz, and zone-leveling; electrical resistivity 9~11 ohm cm; melting point (with decomposition) 450 K (measuring temperature not reported, assumed room temperature).
2	971		1962	298.2		$(\text{CuSbSe}_2)_0.4(\text{Cu}_3\text{Se}_2)_0.6$	p-type; same fabrication method as the above specimen; electrical resistivity 10~16 ohm cm; melting point 440 (eutectic) to 475 K (measuring temperature assumed 25 C).
3	944		1962	298.2		$(\text{CuSbSe}_2)_0.3(\text{Cu}_3\text{Se}_2)_0.7$	n-type; same fabrication method as the above specimen; electrical resistivity 6.5~9 ohm cm; melting point 450 K (measuring temperature assumed 25 C).
4	944		1962	298.2		$(\text{CuSbSe}_2)_0.2(\text{Cu}_3\text{Se}_2)_0.8$	p-type; same fabrication method as the above specimen; electrical resistivity 2 ohm cm; melting point 450 K (measuring temperature assumed 25 C).

DATA TABLE NO. 415 THERMAL CONDUCTIVITY OF  $(\text{CuSbSe}_2 + \text{Cu}_3\text{Se}_2)$  MIXTURES $(\text{CuSbSe}_2 + \text{Cu}_3\text{Se}_2)$ : 95.0%; impurity  $\pm 2.0\%$  each[ Temperature, T, K; Thermal Conductivity, k, Watt  $\text{cm}^{-1}\text{K}^{-1}$  ]

T	k	T	k
<u>CURVE 1*</u>		<u>CURVE 4*</u>	
298.2	0.0188	298.2	0.0184
<u>CURVE 2*</u>			
298.2	0.0225		
<u>CURVE 3*</u>			
298.2	0.0141		

\* No graphical presentation

SPECIFICATION TABLE NO. 416 THERMAL CONDUCTIVITY OF  $(\text{Cu}_2\text{S}_2 + \text{CuSiS}_2)$  MIXTURES  
( $\text{Cu}_2\text{S}_2 + \text{CuSiS}_2$  95.0%; impurity = 2.0% each)

Curve No.	Ref. No. Used	Method	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition $\text{Cu}_2\text{S}_2$	Composition (weight percent) $\text{CuSiS}_2$	Composition (continued), Specifications and Remarks
1	944		1962	298.2		$(\text{CuSiS}_2)_{0.25}(\text{Cu}_2\text{S}_2)_{0.75}$	75.29	24.71	p-type; specimen obtained by fusing ASARCO 99.999 pure elements in carbon-coated quartz tube with agitation, cooling, crushing, re-annealing in 4 mm uncoated quartz, and zone-leveling; electrical resistivity 0.4 to 1.65 milliohm cm; melting point 460 K (measuring temperature assumed 25 C).
2	944		1962	298.2		$(\text{CuSiS}_2)_{0.25}(\text{Cu}_2\text{S}_2)_{0.75}$	80.25	19.75	p-type; same fabrication method as the above specimen; electrical resistivity 0.35 milliohm cm; melting point 1049 K; phase changes at 120 and 569 K (measuring temperature assumed 25 C).
3	944		1962	298.2		$(\text{CuSiS}_2)_{0.25}(\text{Cu}_2\text{S}_2)_{0.75}$	90.14	9.86	p-type; same fabrication method as the above specimen; electrical resistivity 0.35 milliohm cm; melting point 470 K; phase changes at 360 and 435 K (measuring temperature assumed 25 C).
4	944		1962	298.2		$(\text{CuSiS}_2)_{0.25}(\text{Cu}_2\text{S}_2)_{0.75}$	50.39	49.61	p-type; same fabrication method as the above specimen; electrical resistivity 12 milliohm cm; melting point 160 K (measuring temperature assumed 25 C).
5	944		1962	298.2		$(\text{CuSiS}_2)_{0.4}(\text{Cu}_2\text{S}_2)_{0.6}$	60.37	39.63	p-type; same fabrication method as the above specimen; electrical resistivity 11 milliohm cm; melting point 460 K (measuring temperature assumed 25 C).
6	944		1962	298.2		$(\text{CuSiS}_2)_{0.3}(\text{Cu}_2\text{S}_2)_{0.7}$	67.34	32.66	p-type; same fabrication method as the above specimen; electrical resistivity 9.5 to 10.5 milliohm cm; melting point 460 K (measuring temperature assumed 25 C).
7	944		1962	298.2		$(\text{CuSiS}_2)_{0.4}(\text{Cu}_2\text{S}_2)_{0.6}$	70.32	29.68	p-type; same fabrication method as the above specimen; electrical resistivity 1.2 milliohm cm; melting point 460 K (measuring temperature assumed 25 C).

DATA TABLE NO. 416 THERMAL CONDUCTIVITY OF (Cu<sub>2</sub>Se<sub>2</sub> + CuSiAsSe<sub>2</sub>) MIXTURES(Cu<sub>2</sub>Se<sub>2</sub> + CuSiAsSe<sub>2</sub> = 95.0%; impurity < 2.0% each)[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup>K<sup>-1</sup>]

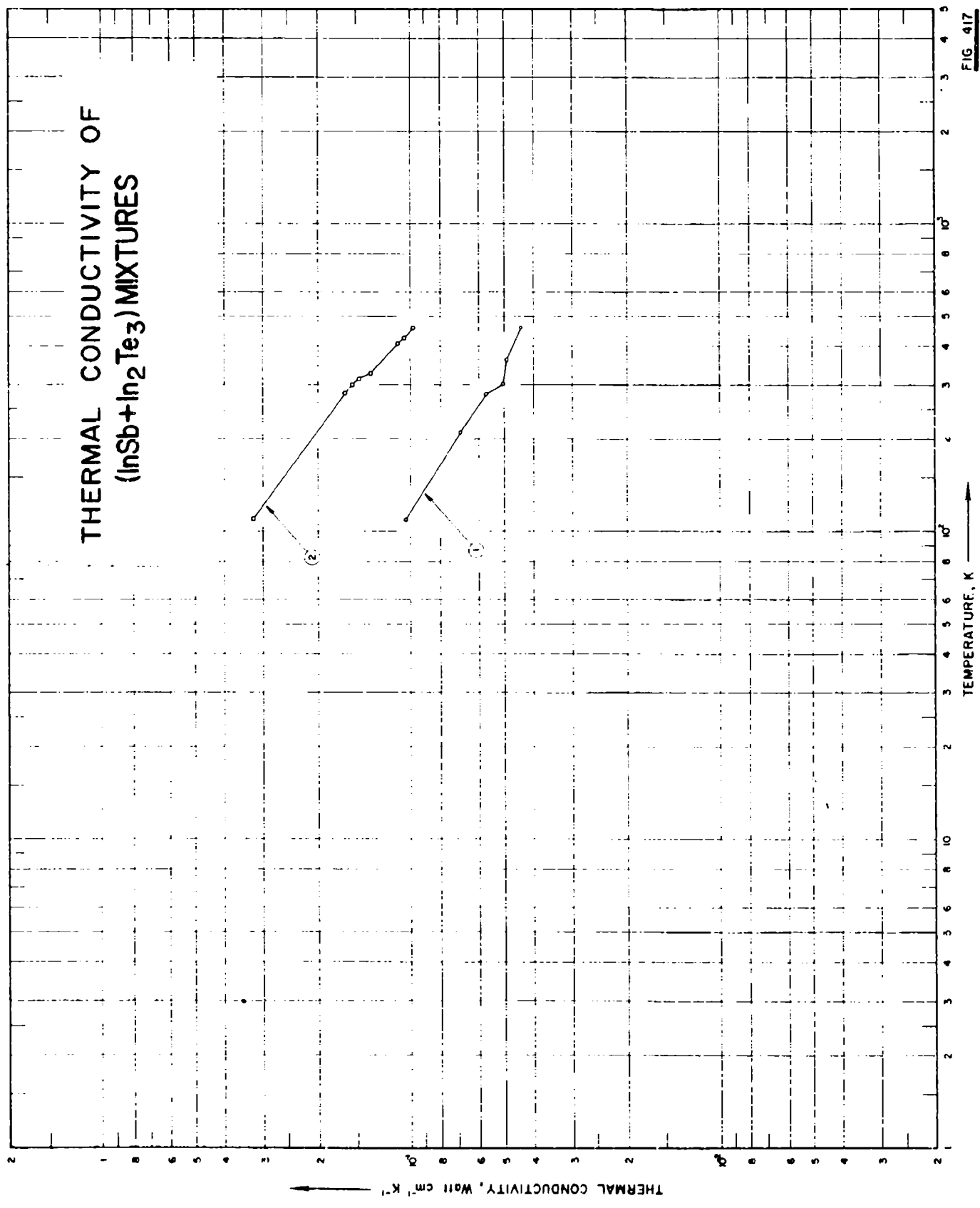
T	k
<u>CURVE 1<sup>a</sup></u>	
298.2	0.0220
<u>CURVE 2<sup>a</sup></u>	
298.2	0.0195
<u>CURVE 3<sup>a</sup></u>	
298.2	0.0190
<u>CURVE 4<sup>a</sup></u>	
298.2	0.0190
<u>CURVE 5<sup>a</sup></u>	
298.2	0.0190
<u>CURVE 6<sup>a</sup></u>	
298.2	0.0250
<u>CURVE 7<sup>a</sup></u>	
298.2	0.0200

<sup>a</sup> No graphical presentation



FIG. 417

# THERMAL CONDUCTIVITY OF (InSb+In<sub>2</sub>Te<sub>3</sub>) MIXTURES



SPECIFICATION TABLE NO. 417 THERMAL CONDUCTIVITY OF (InSb + In<sub>2</sub>Te<sub>3</sub>) MIXTURE(InSb + In<sub>2</sub>Te<sub>3</sub>) 95.0%; impurity < 2.0% each

[ For Data Reported in Figure and Table No. 417 ]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)		Composition (continued), Specifications and Remarks
							InSb	In <sub>2</sub> Te <sub>3</sub>	
1	950, 951	L	1963	109-400			97.45	2.55	Prepared from highly pure indium, Sb-000 antimony tellurium by multiple-zone recrystallization, synthesized in evacuated (10 <sup>-3</sup> mm Hg) quartz ampoule, slowly cooled. Same fabrication method as the above specimen.
2	950, 951	L	1963	111-401			98.01	11.99	

DATA TABLE NO. 417 THERMAL CONDUCTIVITY OF (InSb + In<sub>2</sub>Te<sub>3</sub>) MIXTURES(InSb + In<sub>2</sub>Te<sub>3</sub> ~ 95.0%; impurity ~ 2.0% each)[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup>K<sup>-1</sup>]

T k

CURVE 1

109	0.104
211	0.0690
280	0.0571
302	0.0505
362	0.0493
463	0.0442

CURVE 2

111	0.323
282	0.164
300	0.154
314	0.147
328	0.135
409	0.110
427	0.105
461	0.0980

SPECIFICATION TABLE NO. 418 THERMAL CONDUCTIVITY OF  $(\text{In}_2\text{Te}_3 + \text{Cu}_2\text{Te} + \text{Ag}_2\text{Te})$  MIXTURES

( $\text{In}_2\text{Te}_3 + \text{Cu}_2\text{Te} + \text{Ag}_2\text{Te} = 95.0\%$ ; impurity = 2.0% each)

Curve No.	Ref. Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) $\text{In}_2\text{Te}_3$ $\text{Cu}_2\text{Te}$ $\text{Ag}_2\text{Te}$	Composition (continued), Specifications and Remarks
1	55H	TE	1960	272.9	$\text{Ag}_{.25}\text{Cu}_{.75}\text{InTe}_3$	68.87    21.48    9.65	Quadruple covalent semiconductors.

DATA TABLE NO. 418 THERMAL CONDUCTIVITY OF  $(\text{In}_2\text{Te}_3 + \text{Cu}_2\text{Te} + \text{Ag}_2\text{Te})$  MIXTURES

( $\text{In}_2\text{Te}_3 + \text{Cu}_2\text{Te} + \text{Ag}_2\text{Te} = 95.0\%$ ; impurity = 2.0% each)

[ Temperature, T, K; Thermal Conductivity, k,  $\text{Watt cm}^{-1} \text{K}^{-1}$  ]

T            k  
CURVE 1\*  
272.9    0.0125

No graphical presentation

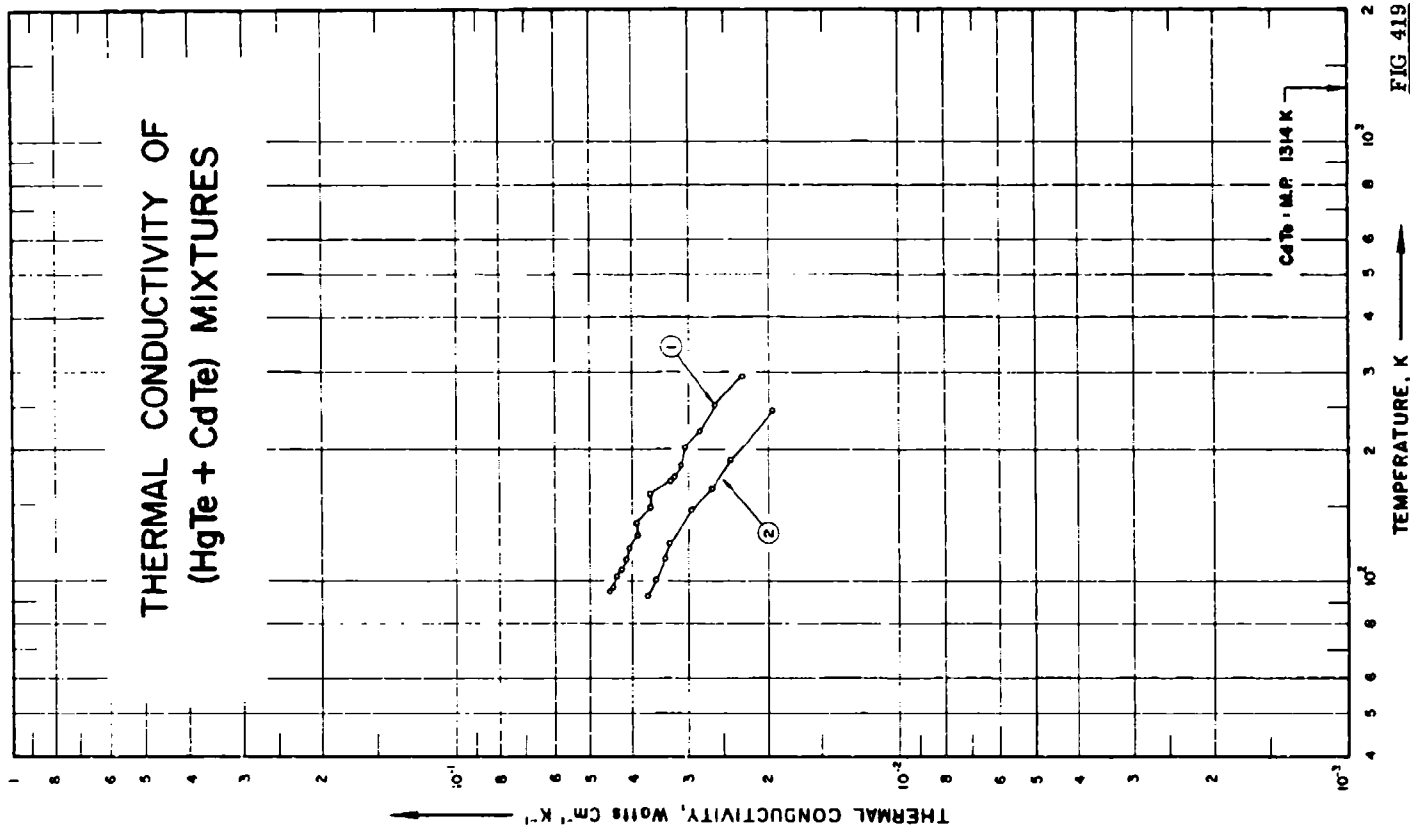


FIG. 419

## SPECIFICATION TABLE NO. 419 THERMAL CONDUCTIVITY OF (HgTe + CdTe) MIXTURES

(HgTe + CdTe : 95.0% impurity  $\pm$  2.0% each)

[ For Data Reported in Figure and Table No. 419 ]

Curve No.	Ref. No.	Method Used	Year	Temp Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)		Composition (continued), Specifications and Remarks
							HgTe	CdTe	
1	520	I.	1960	95-295	$\pm$ 5	Cd <sub>0.04</sub> Hg <sub>0.96</sub> Te; 9	97.04	2.96	Dilute solution of cadmium telluride in mercury telluride.
2	520	I.	1960	94-245	$\pm$ 5	Cd <sub>0.07</sub> Hg <sub>0.93</sub> Te; 23	94.78	5.22	Dilute solution of cadmium telluride in mercury telluride.

DATA TABLE NO. 419 THERMAL CONDUCTIVITY OF (HgTe + CdTe) MIXTURES  
(HgTe + CdTe  $\approx$  95.0%; impurity  $\leq$  2.0% each)

[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup>K<sup>-1</sup>]

T	k
<u>CURVE 1</u>	
96.2	0.0451
97.8	0.0443
102.6	0.0435
107.5	0.0423
113.6	0.0415
120.2	0.0403
124.2	0.0390
137.0	0.0391
148.1	0.0365
159.5	0.0368
170.1	0.0331
174.8	0.0326
185.9	0.0314
203.3	0.0306
222.2	0.0284
253.2	0.0262
295.9	0.0227
<u>CURVE 2</u>	
93.9	0.0374
101.9	0.0357
113.4	0.0342
122.4	0.0333
146.6	0.0296
163.4	0.0266
190.8	0.0242
245.7	0.0195

SPECIFICATION TABLE NO. 420 THERMAL CONDUCTIVITY OF (AgSbTe<sub>2</sub> + SnTe) MIXTURES(AgSbTe<sub>2</sub> + SnTe) ≥ 95.0%; Impurity ≤ 2.0% each

Curve No.	Ref. Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) AgSbTe <sub>2</sub>	Composition (weight percent) SnTe	Composition (continued), Specifications and Remarks
1	553 L	1959	300		AgSbTe <sub>2</sub> -SnTe	56.75	43.25	60% of SnTe by mole added to p-type AgSbTe <sub>2</sub> .
2	553 L	1959	300		AgSbTe <sub>2</sub> -SnTe	85.52	14.48	25 m/o SnTe.

DATA TABLE NO. 420 THERMAL CONDUCTIVITY OF (AgSbTe<sub>2</sub> + SnTe) MIXTURES(AgSbTe<sub>2</sub> + SnTe) ≥ 95.0%; impurity ≤ 2.0% each)[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup>K<sup>-1</sup>]

T	k
<u>CURVE 1*</u>	
300	0.018
<u>CURVE 2*</u>	
300	0.01

\* No graphical presentation



SPECIFICATION TABLE NO. 421 THERMAL CONDUCTIVITY OF (SnTe + AgSbTe<sub>2</sub>) MIXTURES(SnTe + AgSbTe<sub>2</sub> = 95.0%; impurity ≤ 2.0% each)

Curve No.	Ref. Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) SnTe	Composition (weight percent) AgSbTe <sub>2</sub>	Composition (continued), Specifications and Remarks
1	553	I.	1959	300	AgSbTe <sub>2</sub> -SnTe	67.02	32.98	80% of SnTe by mole added to p-type AgSbTe <sub>2</sub> .

DATA TABLE NO. 421 THERMAL CONDUCTIVITY OF (SnTe + AgSbTe<sub>2</sub>) MIXTURES(SnTe + AgSbTe<sub>2</sub> = 95.0%; impurity ≤ 2.0% each)[Temperature, T, K; Thermal Conductivity k, Watt cm<sup>-1</sup>K<sup>-1</sup>]

T k

CURVE 1\*

300 0.026

\* No graphical presentation

# THERMAL CONDUCTIVITY OF (ZnSb + CdSb) MIXTURES

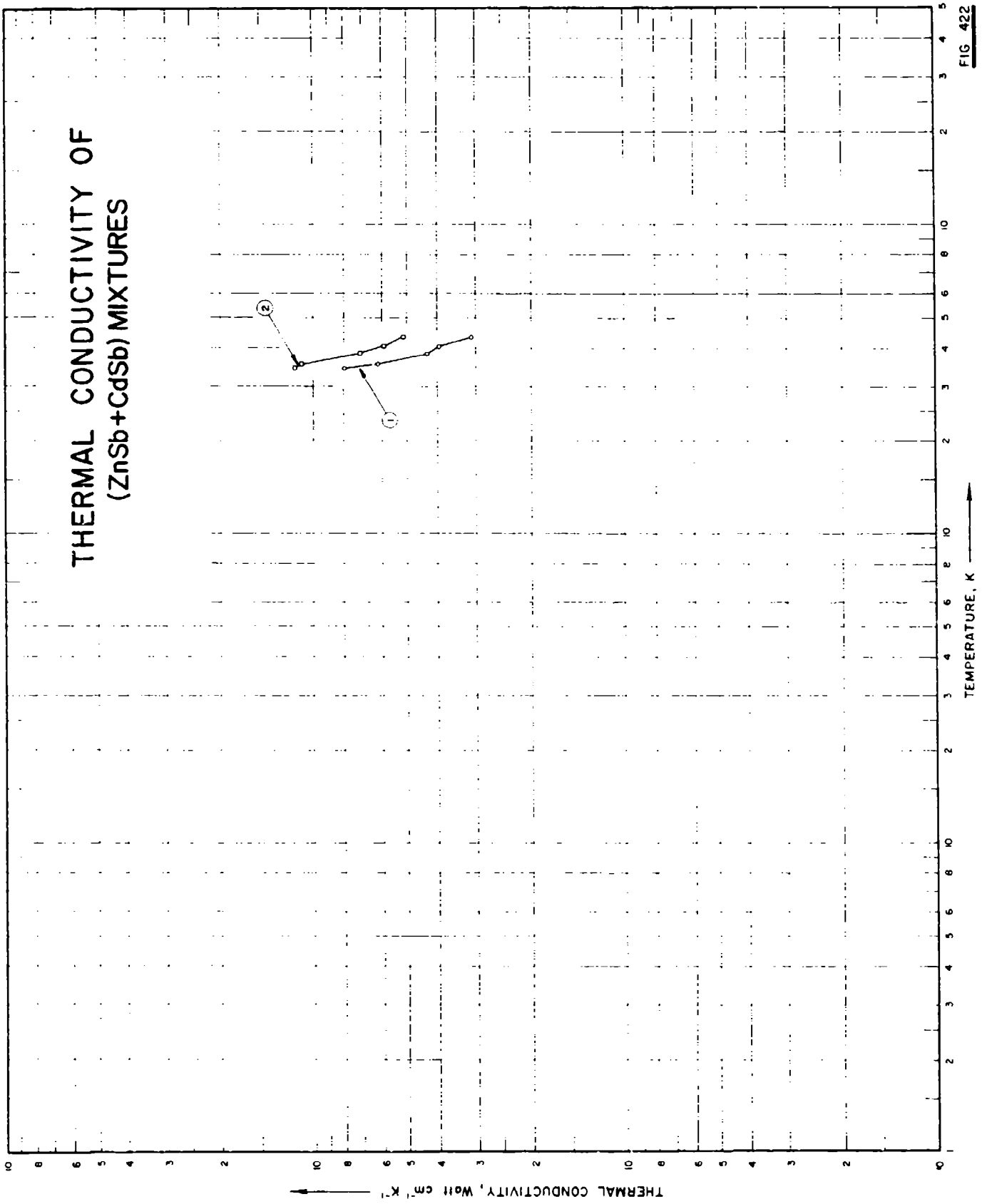


FIG 422

SPECIFICATION TABLE J. 422 THERMAL CONDUCTIVITY OF (ZnSb + CuSb) MIXTURES

(ZnSb + CuSb) 95.0% impurity 2.0% each

[For Data Reported in Figure and Table No. 422]

Curve No.	Rel. Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) ZnSb	Composition (weight percent) CuSb	Composition (continued), Specifications and Remarks
1	979, C 980	1966	343-433		2(CdSb) + 3(ZnSb)	54.5	45.5	Prepared from spectroscopically pure Zn, Sb, and Cd weighed and melted in evacuated Pyrex container at 620-630 C, vibrated and annealed at 450-550 C for 5 hrs and at 250 C for 8 hrs; electrical resistivity reported as 0.285 and 0.155 ohm cm at 70 and 130 C, respectively; measured in high vacuum.
2	979, C 980	1960	343-433		3(CdSb) + 7(ZnSb)	65.1	34.9	Same fabrication method as the above specimen; electrical resistivity reported as 0.0715 and 0.0523 ohm cm at 70 and 130 C, respectively; measured in high vacuum.

## DATA TABLE NO. 422 THERMAL CONDUCTIVITY OF (ZnSb + CdSb) MIXTURES

(ZnSb + CdSb .95.0%: impurity  $\pm$  2.0% each)[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup>K<sup>-1</sup>]

T	k
<u>CURVE 1</u>	
343.2	0.00803
353.2	0.00628
383.2	0.00431
403.2	0.00397
433.2	0.00314
<u>CURVE 2</u>	
343.2	0.0116
353.2	0.0110
383.2	0.00711
403.2	0.00594
433.2	0.00519

SPECIFICATION TABLE NO. 423 THERMAL CONDUCTIVITY OF (Bi<sub>2</sub>Te<sub>3</sub> + Te) MIXTURES  
(Bi<sub>2</sub>Te<sub>3</sub> + Te = 95.0%; impurity = 2.0% each)

Curve No.	Ref. Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) Bi <sub>2</sub> Te <sub>3</sub>	Te	Composition (continued), Specifications and Remarks
1	936 T	1965	298.2		Bi <sub>2</sub> Te <sub>3</sub> .45	97.06	2.94	n-type; prepared from 99.995 pure bismuth supplied by Consolidated Mining and Smelting Co. and from 99.97 pure tellurium supplied by Canadian Copper Refiners Ltd.; materials weighed and out, crushed, sealed in an ampule in a vacuum of 10 <sup>-5</sup> Torr, heated at 900 C for 20 hrs, rocked, cooled, zone-melted at a rate of 0.07-0.25 in. hr <sup>-1</sup> , then cooled and cut; thermal conductivity data calculated from measured values of the figure of merit, Seebeck coefficient, and electrical conductivity; electrical conductivity reported as 1.93 x 10 <sup>3</sup> ohm <sup>-1</sup> cm <sup>-1</sup> at room temperature.
2	936 T	1965	298.2		Bi <sub>2</sub> Te <sub>3</sub> .26	96.02	3.98	n-type; same fabrication method and measuring method as the above specimen; electrical conductivity reported as 2.26 x 10 <sup>3</sup> ohm <sup>-1</sup> cm <sup>-1</sup> at room temperature.

DATA TABLE NO. 423 THERMAL CONDUCTIVITY OF (Bi<sub>2</sub>Te<sub>3</sub> + Te) MIXTURES

(Bi<sub>2</sub>Te<sub>3</sub> + Te = 95.0%; impurity = 2.0% each)

[ Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup>K<sup>-1</sup> ]

T	k
<u>CURVE 1*</u>	
298.2	0.0279
<u>CURVE 2*</u>	
298.2	0.0281

\* No graphical presentation

# THERMAL CONDUCTIVITY OF Be + BeO

[Be + BeO > 99.50%; impurity < 0.20% each]

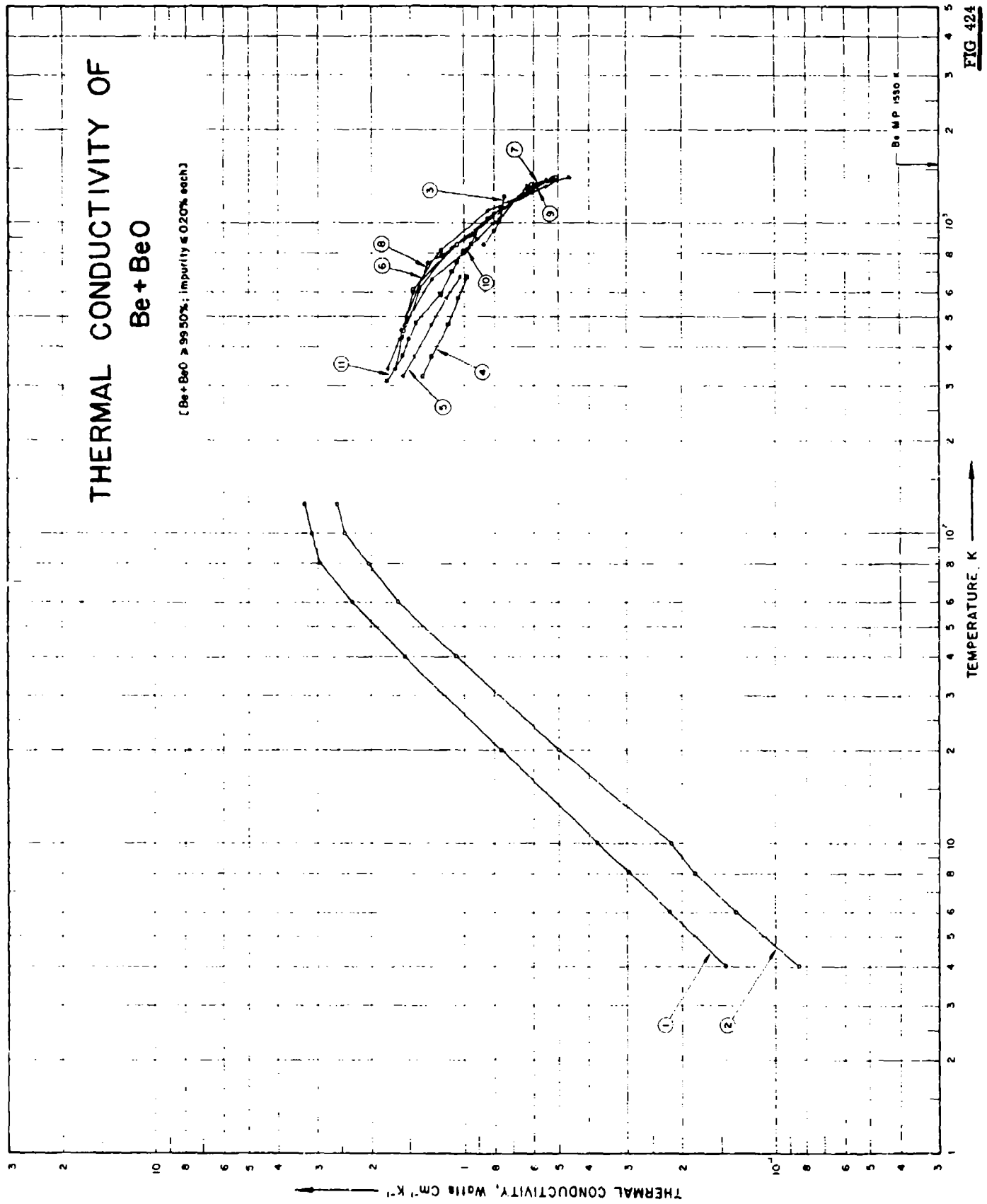


FIG 424

## SPECIFICATION TABLE NO. 424 THERMAL CONDUCTIVITY OF (BERYLLIUM + BERYLLIUM OXIDE) ALLOYS

(Be + BeO = 99.50%; impurity &lt; 0.20% each)

(For Data Reported in Figure and Table No. 424)

Curve No.	Rel. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition Be	Composition (weight percent) BeO	Composition (continued), Specifications and Remarks
1	494		1960	4.0-125			98.700	1.2	0.936 Al, 0.015 Si, 0.016 Ni, 0.010 Mn, trace B and Li; specimen axis parallel to the pressing axis
2	494		1960	4.0-125			98.700	1.15	0.044 Al, 0.014 Ni, 0.009 Mn, trace B and Li; specimen axis perpendicular to the pressing axis.
3	43	L	1958	850-1238	5.0	YB-9052	99.160	0.84	0.13 Al, 0.18 Fe, 0.03 Cu, 0.05 Cl, the rest BeO and other impurities; chill-cast.
4	111	C	1953	323-673		IV A. R.	98.5		The above specimen after heat treatment at 700 C.
5	111	C	1953	323-673		IV H. T.			0.003 Mg, 0.015 Al, 0.01 Si, 0.001 Ca, 0.002 Ti, 0.008 Cr, 0.005 Mn, 0.015 Fe, and 0.004 Cu.
6	513		1960	340-1400		Y 6825	98.322	1.45	0.006 Mg, 0.05 Al, 0.008 Si, 0.002 Ca, 0.004 Ti, 0.01 Cr, 0.008 Mn, 0.15 Fe, 0.015 Ni, 0.01 Cu.
7	513		1960	450-1400		Y 9384	98.892	0.845	0.01 Mg, 0.03 Al, 0.02 Si, 0.002 Ca, 0.002 Ti, 0.01 Cr, 0.006 Mn, 0.15 Fe, 0.015 Ni, 0.01 Cu.
8	513		1960	450-1375		Y 6836	98.554	1.292	0.015 Mg, 0.03 Al, 0.008 Si, 0.002 Ca, 0.002 Ti, 0.01 Cr, 0.01 Mn, 0.15 Fe, 0.02 Ni, 0.015 Cu.
9	513		1960	340-1390		YB 1900	98.509	1.229	0.02 Mg, 0.04 Al, 0.04 Si, 0.002 Ca, 0.004 Ti, 0.02 Cr, 0.008 Mn, 0.20 Fe, 0.02 Ni, and 0.01 Cu.
10	513		1960	480-1400		LYB 1102	98.641	0.992	0.015 Mg, 0.03 Al, 0.01 Si, 0.001 Ca, 0.003 Ti, 0.015 Cr, 0.008 Mn, 0.20 Fe, 0.015 Ni, and 0.015 Cu.
11	513		1960	310-1035		BMI 5	99.084	0.609	





## SPECIFICATION TABLE NO. 425 THERMAL CONDUCTIVITY OF [CHROMIUM + ALUMINUM OXIDE] ALLOYS

(Cr + Al<sub>2</sub>O<sub>3</sub> ± 95.0%; impurity ± 2.0% each)

Curve No.	Ref. Method Used	Year	Temp. Range, °K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) Cr	Composition (weight percent) Al <sub>2</sub> O <sub>3</sub>	Composition (continued), Specifications and Remarks
1	981	1957	293.2			80	20	

## DATA TABLE NO. 425 THERMAL CONDUCTIVITY OF [CHROMIUM + ALUMINUM OXIDE] ALLOYS

(Cr + Al<sub>2</sub>O<sub>3</sub> ± 95.0%; impurity ± 2.0% each)[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup> K<sup>-1</sup>]

T k

CURVE 1\*

293.2 0.460

\* No graphical presentation

# THERMAL CONDUCTIVITY OF (Cu+BeCo) ALLOYS

(Cu+BeCo 99.50%, impurity  $\leq$  0.20% each)

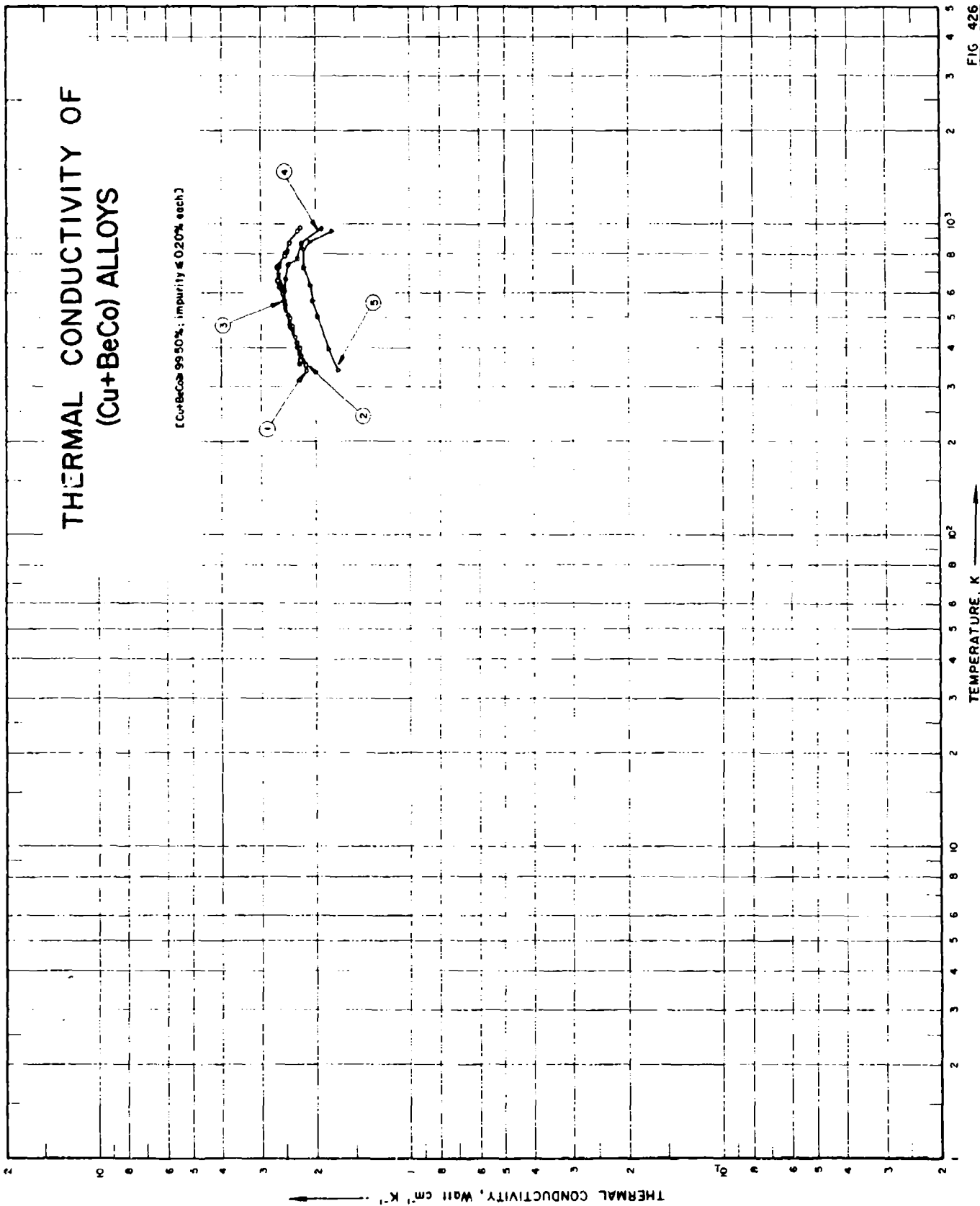


FIG. 426

SPECIFICATION TABLE NO. 426 THERMAL CONDUCTIVITY OF [Cu + BeCo] ALLOYS  
(Cu + BeCo: 99.50%, impurity  $\leq 0.20\%$  each)

[ For Data Reported in Figure and Table No. 426 ]

Curve No.	Ref. Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) Cu	Composition (weight percent) BeCo	Composition (continued), Specifications and Remarks
1	541	1959	358-968			98.49	1.51	(1.31 Co, 0.2 Be); specimen quenched then tempered at 740 C in radial section.
2	541	1959	350-791			97.03	2.92	(2.52 Co, 0.4 Be); specimen quenched then tempered at 740 C in radial section.
3	541	1959	376-810			96.23	3.77	(3.27 Co, 0.5 Be); specimen quenched then tempered at 700 C in radial section.
4	541	1959	355-960			94.72	5.28	(4.58 Co, 0.7 Be); specimen quenched then tempered at 700 C in radial section.
5	541	1959	340-947			93.21	6.79	(5.89 Co, 0.9 Be); specimen quenched then tempered at 700 C in radial section.

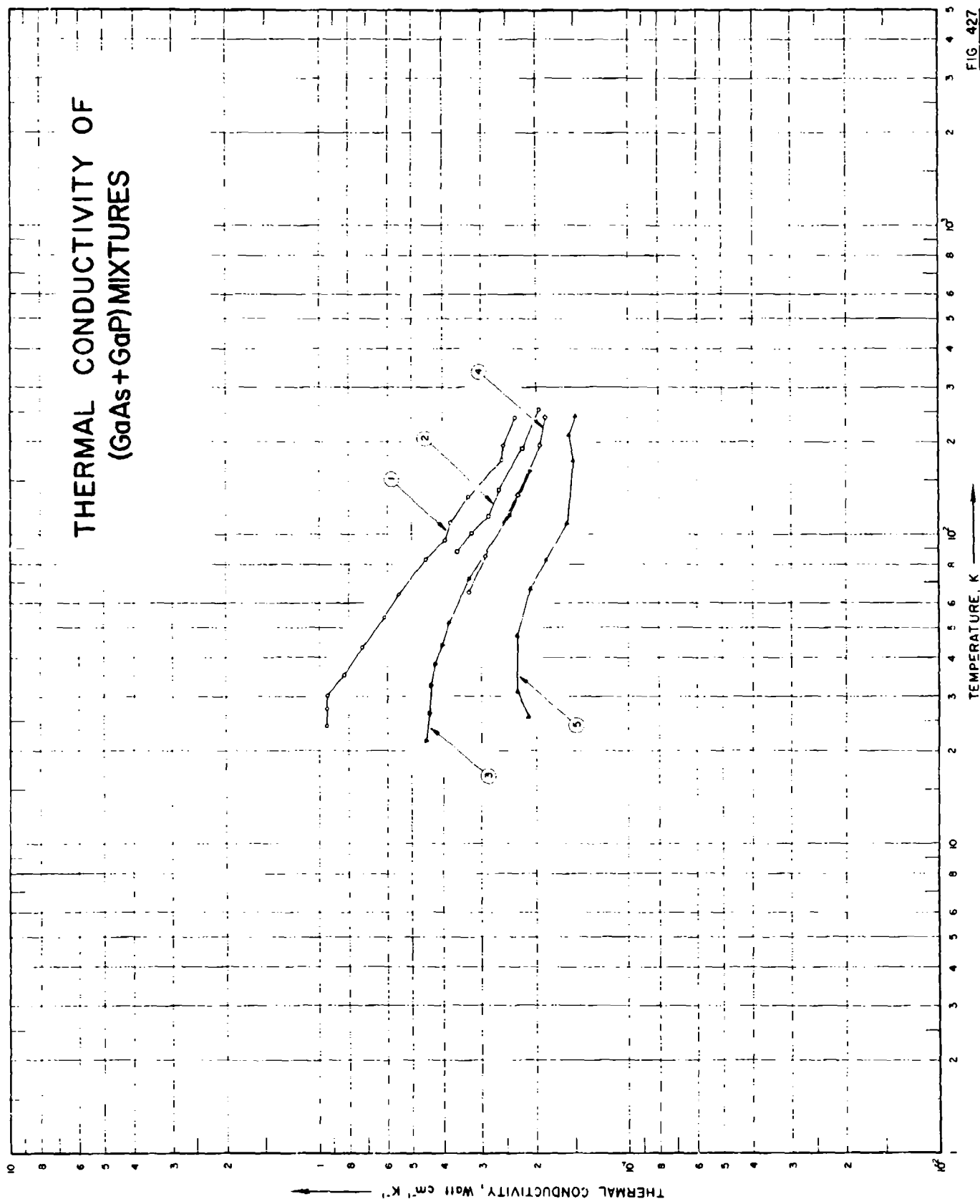
## DATA TABLE NO. 426 THERMAL CONDUCTIVITY OF (Cu + BeCo) ALLOYS

(Cu + BeCo) 99.50%; impurity &lt;math&gt;\leq 0.20\%&lt;/math&gt; each

[ Temperature, T, K; Thermal Conductivity,  $\kappa$ , Watt  $\text{cm}^{-1}\text{K}^{-1}$  ]

T	$\kappa$	T	$\kappa$
<u>CURVE 1</u>			
338.2	2.16	968.2	2.24
432.2	2.34	960.2	1.93
498.2	2.44	<u>CURVE 5</u>	
567.2	2.52	340.2	1.70
629.2	2.65	398.2	1.83
721.2	2.65	501.2	1.98
805.2	2.52	563.2	2.06
861.2	2.46	638.2	2.09
948.2	2.30	728.2	2.20
968.2	2.26	834.2	2.20
<u>CURVE 2</u>			
350.2	2.17	870.2	2.11
399.2	2.27	947.2	1.78
467.2	2.41	<u>CURVE 3</u>	
597.2	2.57	376.2	2.24
658.2	2.68	413.2	2.30
728.2	2.67	469.2	2.41
791.2	2.54	508.2	2.46
<u>CURVE 4</u>			
563.2	2.55	628.2	2.57
713.2	2.68	713.2	2.68
810.2	2.52	810.2	2.52
<u>CURVE 4</u>			
355.2	2.26	403.2	2.30
473.2	2.43	550.2	2.51
608.2	2.55	608.2	2.55
675.2	2.52	738.2	2.47
738.2	2.47	771.2	2.30

# THERMAL CONDUCTIVITY OF (GaAs + GaP) MIXTURES



## SPECIFICATION TABLE NO. 427 THERMAL CONDUCTIVITY OF (GaAs + GaP) MIXTURES

(GaAs + GaP ~95.0%; impurity ~2.0% each)

[For Data Reported in Figure and Table No. 427]

Curve No.	Ref. Method No. Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)		Composition (continued), Specifications and Remarks
						GaAs	GaP	
1	946 L	1965	24-239	+10	GaAs <sub>93.9</sub> P <sub>0.1</sub> : FH112	92.8	7.2	n-type Te-doped polycrystalline; 1.0 x 1.5 x 8 mm; prepared by a closed tube vapor transport method using PbCl <sub>2</sub> as the transport agent, the sealed quartz tube moved slowly through a stationary temperature gradient; carrier concentration $4 \times 10^{18} \text{ cm}^{-3}$ .
2	946 L	1965	88-254	±10	GaAs <sub>85.2</sub> P <sub>0.3</sub> : G302	85.2	14.8	n-type Si-doped polycrystalline; same dimensions and fabrication method as the above specimen; carrier concentration $2 \times 10^{18} \text{ cm}^{-3}$ .
3	946 L	1965	22-161	+10	GaAs <sub>74.0</sub> SiP <sub>0.33</sub> : GM50	74.0	26.0	n-type Se-doped polycrystalline; same dimensions and fabrication method as the above specimen; carrier concentration $2 \times 10^{18} \text{ cm}^{-3}$ .
4	946 L	1965	66-241	±10	GaAs <sub>72.7</sub> SiP <sub>0.35</sub>	72.7	27.3	n-type Te-doped polycrystalline; same dimensions; fabrication method, and carrier concentration as the above specimen.
5	946 L	1965	26-243	±10	GaAs <sub>59.0</sub> SiP <sub>0.5</sub> : G407	59.0	41.0	Similar to the above specimen, except carrier concentration $3 \times 10^{18} \text{ cm}^{-3}$ .

## DATA TABLE NO. 427 THERMAL CONDUCTIVITY OF (GaAs + GaP) MIXTURES

(GaAs + GaP : 95.0% impurity &lt; 2.0% each)

[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup>K<sup>-1</sup>]

T	k	T	k
<u>CURVE 1</u>			
24.2	0.944	25.7	0.212
27.4	0.944	31.1	0.230
30.2	0.944	47.0	0.230
35.3	0.928	66.9	0.209
43.2	0.730	82.8	0.185
54.0	0.619	108.1	0.157
64.3	0.551	174.6	0.151
83.4	0.451	211.8	0.156
96.2	0.396	242.8	0.148
109.1	0.381		
133.4	0.332		
175.0	0.257		
196.8	0.255		
239.4	0.232		

<u>CURVE 2</u>	
88.1	0.351
101.2	0.322
114.0	0.284
139.3	0.263
191.9	0.221
254.2	0.196

<u>CURVE 3</u>	
21.5	0.451
26.4	0.441
32.5	0.435
38.2	0.421
43.9	0.403
51.7	0.382
71.8	0.328
115.6	0.243
161.1	0.207

<u>CURVE 4</u>	
65.5	0.330
84.6	0.290
134.0	0.227
194.1	0.194
240.5	0.186

# THERMAL CONDUCTIVITY OF (InAs+InP) MIXTURE

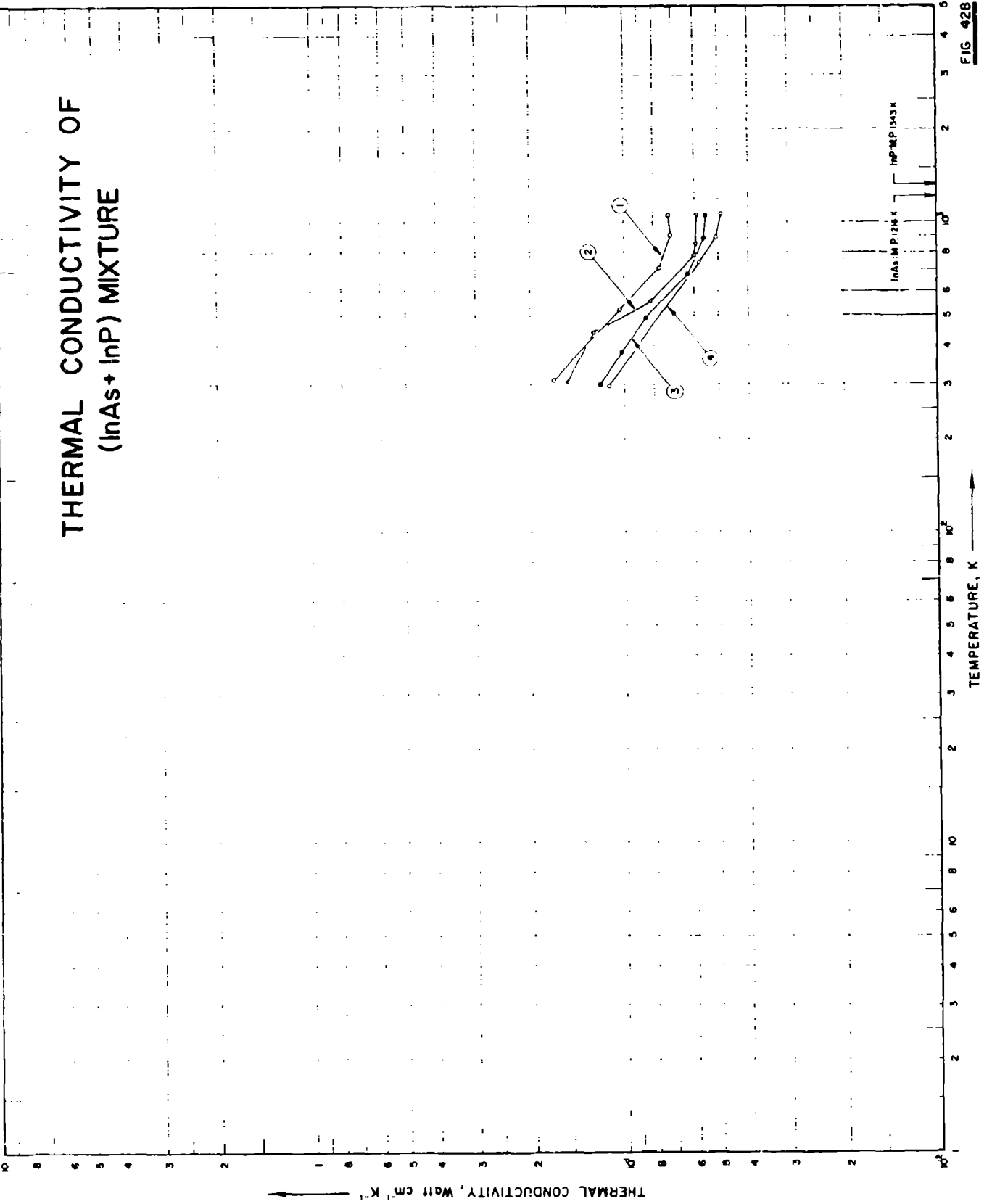


FIG. 428



SPECIFICATION TABLE NO. 428 THERMAL CONDUCTIVITY OF (InAs + InP) MIXTURES

(InAs + InP ~ 95.0% ; impurity ~ 2.0% each)

[For Data Reported in Figure and Table No. 428]

Curve No.	Ref. No. Used	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)		Composition (continued), Specifications and Remarks
							InAs	InP	
1	567	C	1959	310-1063	10	InAs <sub>95.35</sub> P <sub>0.35</sub>	96.11	3.89	n-type; free from group VI doping agent; extrinsic carrier concentration $4 \times 10^{17} \text{ cm}^{-3}$ ; specimen made from 99.999 pure indium, 99.999 pure arsenic, and commercial white phosphorus purified by repeated steam distillation; specimen prepared in a two-zone tube furnace with carefully controlled phosphorus vapor pressure; FH stainless steel (checked by Armco Iron) used as the comparative material.
2	567	C	1959	305-1066	10	InAs <sub>92.13</sub> P <sub>0.1</sub>	92.13	7.87	Similar to the above specimen but prepared by diluting specimens of higher phosphorus content with the proper amount of indium and arsenic, sealing off under vacuum, reacting and lowering at a rate of about $2.5 \text{ cm hr}^{-1}$ through a temperature gradient at the melting point; extrinsic carrier concentration $6 \times 10^{16} \text{ cm}^{-3}$ .
3	567	C	1959	303-1058	10	InAs <sub>83.4</sub> P <sub>0.2</sub>	83.89	16.11	Similar to the specimen InAs <sub>95.35</sub> P <sub>0.35</sub> but with a carrier concentration of $10^{17} \text{ cm}^{-3}$ .
4	567	C	1959	298-1075	10	InAs <sub>66.13</sub> P <sub>0.4</sub>	66.13	33.87	Similar to the above specimen but with a carrier concentration of $2 \times 10^{17} \text{ cm}^{-3}$ .

## DATA TABLE NO. 428 THERMAL CONDUCTIVITY OF (InAs + InP) MIXTURES

(InAs + InP) : 95.0%; impurity : 2.0% each)

T: temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup>K<sup>-1</sup>

T k

CURVE 1

310	0.169
524	0.103
713	0.077
913	0.071
1063	0.072

CURVE 2

305	0.152
445	0.125
551	0.082
783	0.060
853	0.0595
1066	0.0590

CURVE 3

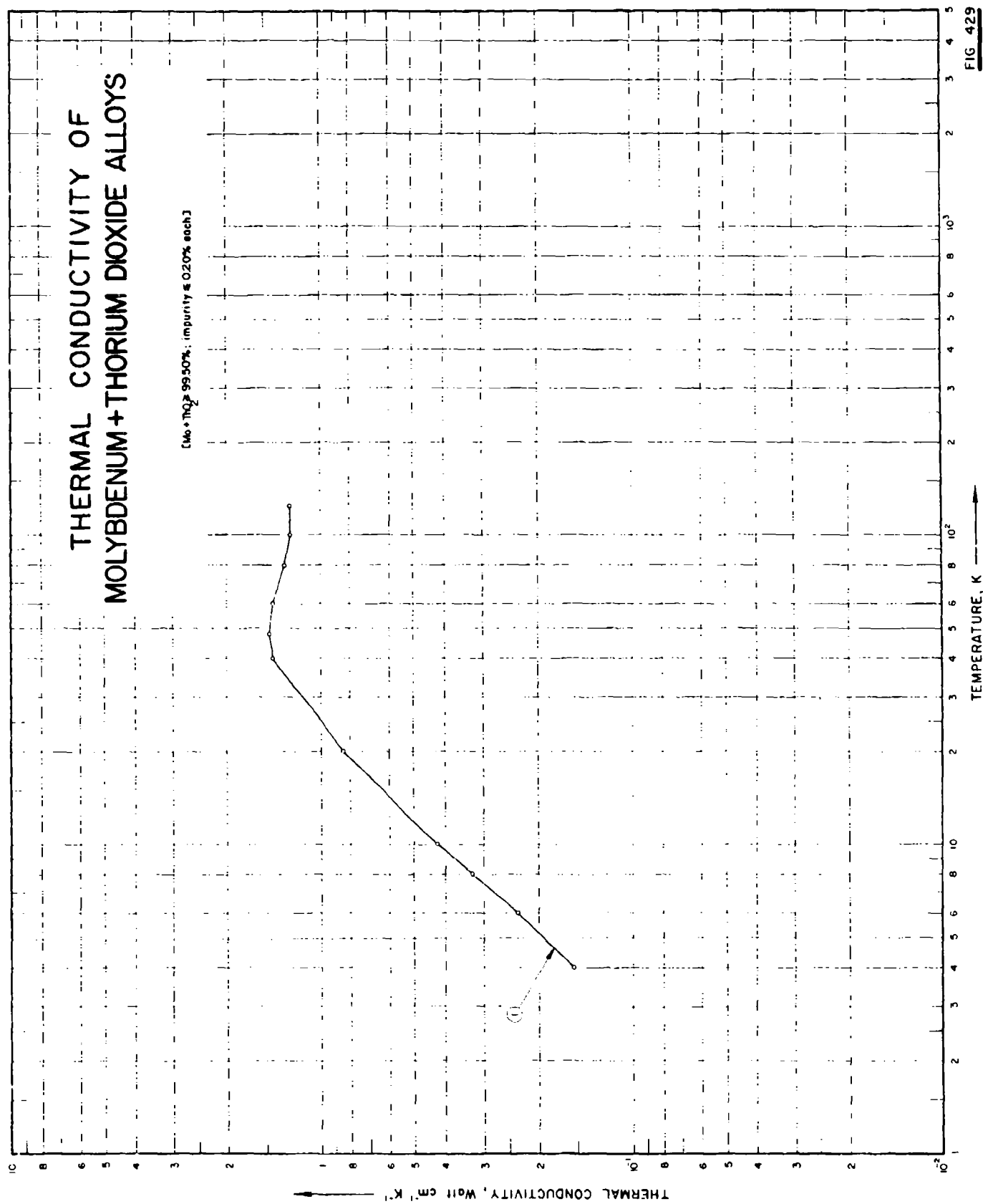
303	0.119
383	0.102
493	0.085
633	0.063
895	0.056
1078	0.055

CURVE 4

298	0.112
743	0.055
900	0.051
1075	0.049

# THERMAL CONDUCTIVITY OF MOLYBDENUM + THORIUM DIOXIDE ALLOYS

(Mo + ThO<sub>2</sub> 99.50%, impurity  $\leq$  0.20% each)



## SPECIFICATION TABLE NO. 429 THERMAL CONDUCTIVITY OF [MOLYBDENUM + THORIUM DIOXIDE] ALLOYS

(Mo + ThO<sub>2</sub> = 99.50%; impurity < 0.20% each)

[For Data Reported in Figure and Table No. 429]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
1	494		1950	4-125			0.001-0.01 Al, 0.001-0.01 Cu, 0.01-0.1 Fe, 0.001-0.01 Nb, 0.01-0.1 Si, trace Ca, Cr and Mg; doped by ThO <sub>2</sub> 1-2 by volume; ground down to a rod with 3.67 mm dia and 13 cm long.

## DATA TABLE NO. 429 THERMAL CONDUCTIVITY OF CARBON FIBER MAT - TAPERED DISC (D) ALLECS

Alloy: T109 98.50% graphite, 0.50% epoxy

Temperature: T, K Thermal Conductivity: k, W/m<sup>2</sup>°K<sup>-1</sup>

T	k
4.0	0.155
6.0	0.215
8.0	0.325
10.0	0.435
20.0	0.845
40.0	1.44
48.0	1.46
60.0	1.43
80.0	1.30
100.0	1.25
125.0	1.25

CURVE 1

SPECIFICATION TABLE NO. 430 THERMAL CONDUCTIVITY OF [SODIUM + DISODIUM OXIDE] ALLOYS

(Na + Na<sub>2</sub>O ≥ 99.50%; impurity ≤ 0.20% each)

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)	Specifications and Remarks
1	868,867	L	1965	328	≤ 15		4.9 Na <sub>2</sub> O.	
2	868,867	L	1965	328	≤ 15		5.3 Na <sub>2</sub> O.	
3	868,867	L	1965	328	≤ 15		7.1 Na <sub>2</sub> O.	
4	868,867	L	1965	328	≤ 15		7.6 Na <sub>2</sub> O.	
5	868,867	L	1965	328	≤ 15		7.6 Na <sub>2</sub> O.	
6	868,867	L	1965	328	≤ 15		23.0 Na <sub>2</sub> O.	
7	868,867	L	1965	328	≤ 15		25.1 Na <sub>2</sub> O.	
8	868,867	L	1965	328	≤ 15		28.2 Na <sub>2</sub> O.	
9	868,867	L	1965	328	≤ 15		44.4 Na <sub>2</sub> O.	
10	868,867	L	1965	328	≤ 15		47.7 Na <sub>2</sub> O.	

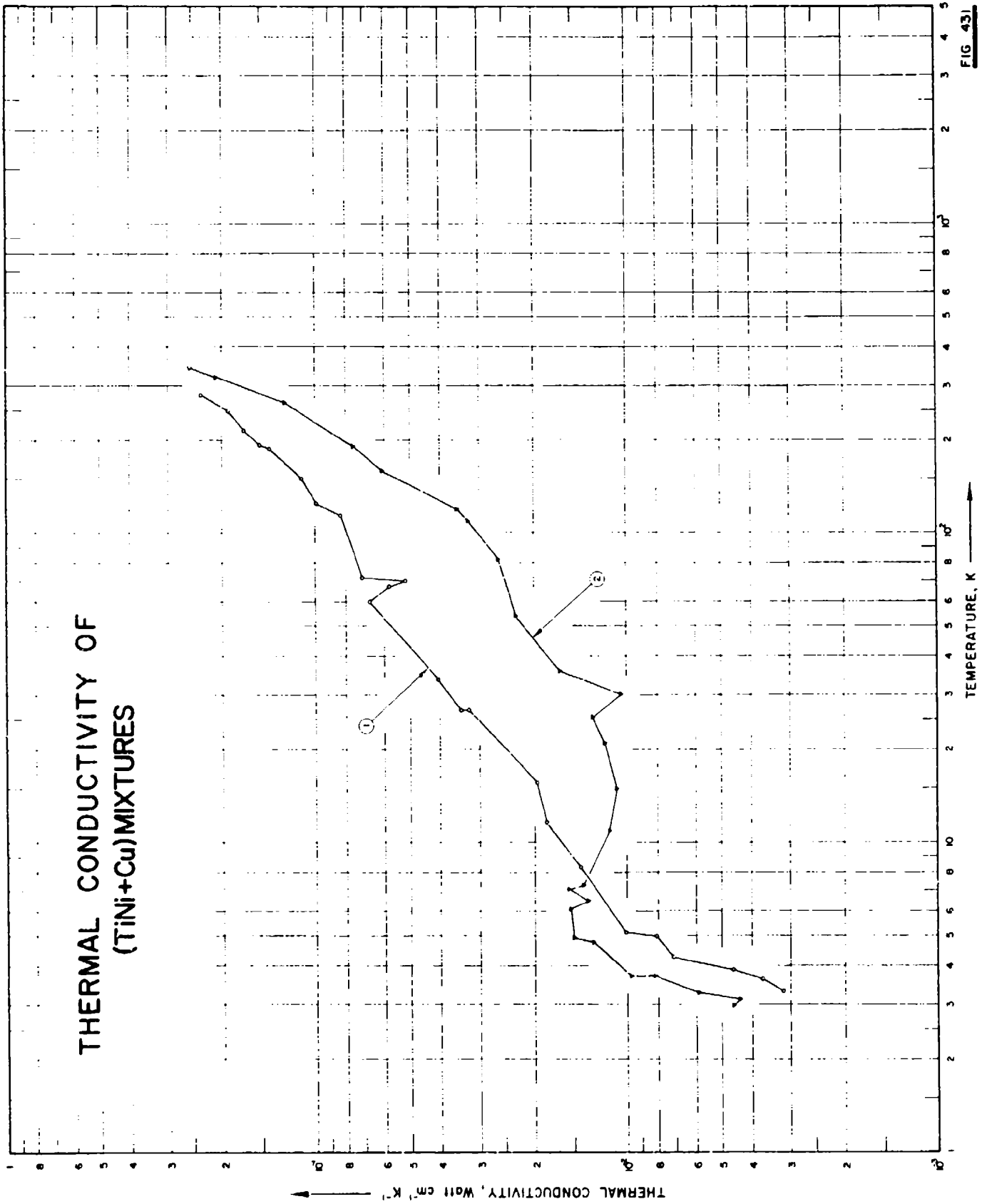
DATA TABLE NO. 430 THERMAL CONDUCTIVITY OF [SODIUM + DISODIUM OXIDE] ALLOYS

(Na + Na<sub>2</sub>O ≥ 99.50%; impurity ≤ 0.20% each)[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup>K<sup>-1</sup>]

T	k	T	k	T	k	T	k
<u>CURVE 1*</u>		<u>CURVE 4*</u>		<u>CURVE 7*</u>		<u>CURVE 10*</u>	
328	1.08	328	1.14	328	0.719	328	0.540
<u>CURVE 2*</u>		<u>CURVE 5*</u>		<u>CURVE 8*</u>		<u>CURVE 9*</u>	
328	1.00	328	1.02	328	0.708		
328	1.14	328	0.778	328	0.494		

\* No graphical presentation

# THERMAL CONDUCTIVITY OF (TiNi+Cu) MIXTURES



## SPECIFICATION TABLE NO. 431 THERMAL CONDUCTIVITY OF (TiNi + Cu) MIXTURES

(TiNi + Cu) &gt; 95.0%; impurity &lt; 2.0% each

[For Data Reported in Figure and Table No. 431]

Curve No.	Ref. Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)		Composition (continued), Specifications and Remarks	
						TiNi	Cu		
1	965	L	1964	3.3-279	5	TiNi + 2Cu	98.0	2.0	Prepared from Mond nickel shot (99.9% pure) and DuPont high purity sponge, the titanium contained up to: 0.03 Mg, 0.07 Fe, 0.05 Mn, 0.04 Si, and 0.15 other impurities; rods were hot swaged and furnace cooled from homogenized buttons and machined into cylinder of 0.4 cm dia x 3 cm long; grain size ~46 $\mu$ ; electrical resistivity reported as 41.0, 41.0, 41.0, 40.7, 41.3, 41.0, 41.2, 41.4, 43.2, 49.2, 52.1, 55.0, 60.3, 67.3, 58.2, 79.6, 78.7, and 80.2 $\mu\text{ohm cm}$ at 2.90, 3.29, 3.67, 4.05, 8.61, 12.7, 21.8, 32.0, 48.5, 78.2, 95.1, 118.9, 158.9, 204.6, 212.3, 291.1, 301.3, and 304.8 K, respectively.
2	963	L	1964	2.9-342	5	TiNi + 8Cu	92.0	8.0	Same fabrication method and dimensions as the above specimens; grain size ~54 $\mu$ ; electrical resistivity reported as 0.111, 0.111, 0.112, 0.112, 0.111, 0.112, 0.111, 0.111, 0.111, 0.111, 0.107, 0.103, 0.0942, and 0.0877 $\text{m}\mu\text{ohm cm}$ at 2.22, 3.66, 4.11, 4.97, 7.60, 15.7, 30.9, 53.0, 77.5, 91.6, 115.3, 146.9, 204.6, and 300.6 K, respectively.



DATA TABLE NO. 411 THERMAL CONDUCTIVITY OF (TINi + Cu) MIXTURES  
(TINi + Cu : 95.0%; impurity : 2.0% each)

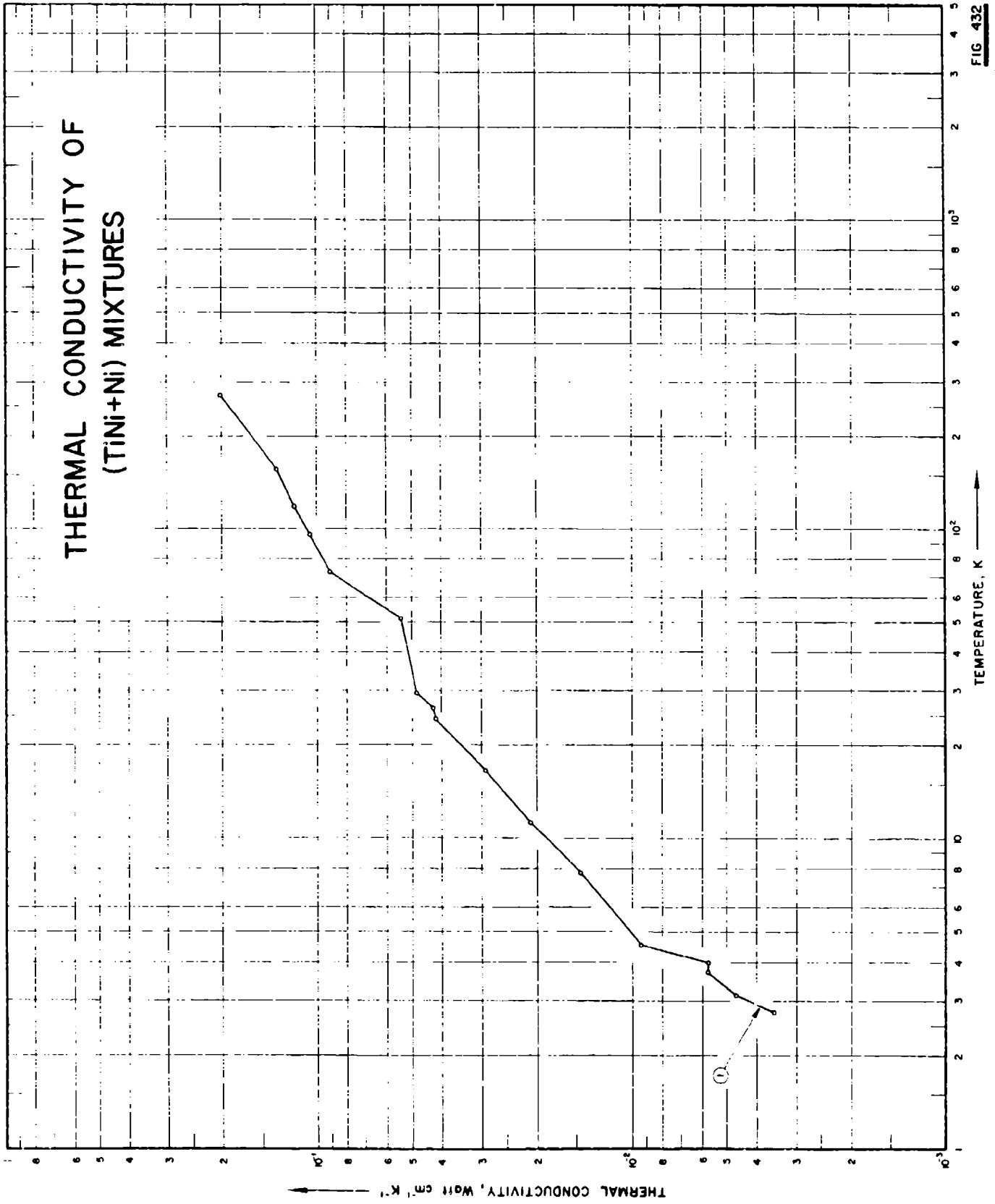
(Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup>K<sup>-1</sup>)

T	k	T		k
		CURVE 1	CURVE 2 (cont.)	
3.31	0.00316	25.5	0.0130	
3.63	0.00370	30.1	0.0104	
3.88	0.00460	35.7	0.0164	
4.28	0.00715	53.8	0.0229	
4.96	0.00813	82.4	0.0260	
5.11	0.0101	109.6	0.0325	
6.30	0.0143	119.4	0.0350	
11.6	0.0182	153.9	0.0608	
15.6	0.0196	191.4	0.0785	
26.8	0.0324	264.9	0.126	
26.8	0.0342	319.2	0.208	
33.6	0.0406	342.0	0.252	
39.6	0.0573			
67.3	0.0581			
69.5	0.0516			
71.5	0.0710			
114.3	0.0832			
125.3	0.0991			
150.0	0.111			
187.1	0.140			
193.2	0.151			
215.8	0.170			
247.7	0.191			
279.3	0.232			

CURVE 2	
2.92	0.00457
3.12	0.00439
3.28	0.00589
3.70	0.00819
3.70	0.00973
4.76	0.0129
4.94	0.0150
6.14	0.0152
6.46	0.0135
7.05	0.0155
7.26	0.0139
10.9	0.0114
14.9	0.0108
20.9	0.0118

# THERMAL CONDUCTIVITY OF (TiNi+Ni) MIXTURES



## SPECIFICATION TABLE NO. 432 THERMAL CONDUCTIVITY OF (TiNi + Ni) MIXTURES

(TiNi + Ni) 95.0%; impurity &lt; 2.0% each)

[For Data Reported in Figure and Table No. 432]

Curve No.	Ref. Method No. Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)		Composition (continued), Specifications and Remarks
						TiNi	Ni	
1	965 L	1964	2.6-271	5	TiNi + 18Ni	82	18	Prepared by mixing Mond nickel shots (99.9% pure) and DuPont high purity sponge; the titanium contained up to 0.08 Mg, 0.07 Fe, 0.05 Mn, 0.04 Si, and 0.15 other impurities; specimen rod were hot swaged and furnace cooled from homogenized buttons and machined into a cylinder of 0.4 cm dia x 3 cm long; electrical resistivity reported as 27.5, 27.2, 27.6, 27.5, 28.0, 27.8, 28.8, 29.9, 31.0, 34.7, 35.8, 39.0, 42.4, 44.2, 52.2, 59.7, 63.0, and 69.3 $\mu\text{ohm cm}$ at 2.37, 4.13, 5.50, 7.26, 11.8, 19.3, 27.7, 45.1, 59.3, 74.0, 79.4, 95.1, 111.7, 130.6, 167.9, 209.9, 248.3, and 302.0 K, respectively.

DATA TABLE NO. 432 THERMAL CONDUCTIVITY OF (TiNi + Ni) MIXTURES  
 (TiNi + Ni) 95.0%: impurity = 2.0% each

[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup>K<sup>-1</sup>]

T	k
2.75	0.00353
3.13	0.00466
3.71	0.00579
3.99	0.00574
4.53	0.00933
7.73	0.0145
11.3	0.0210
16.6	0.0292
24.4	0.0420
26.4	0.0428
29.5	0.0482
51.5	0.0542
72.9	0.0916
96.6	0.107
119.4	0.119
157.0	0.136
271.0	0.204

# THERMAL CONDUCTIVITY OF TUNGSTEN + THORIUM DIOXIDE ALLOYS

[ W = ThO<sub>2</sub> = 99.50%; impurity  $\leq$  0.20% each ]

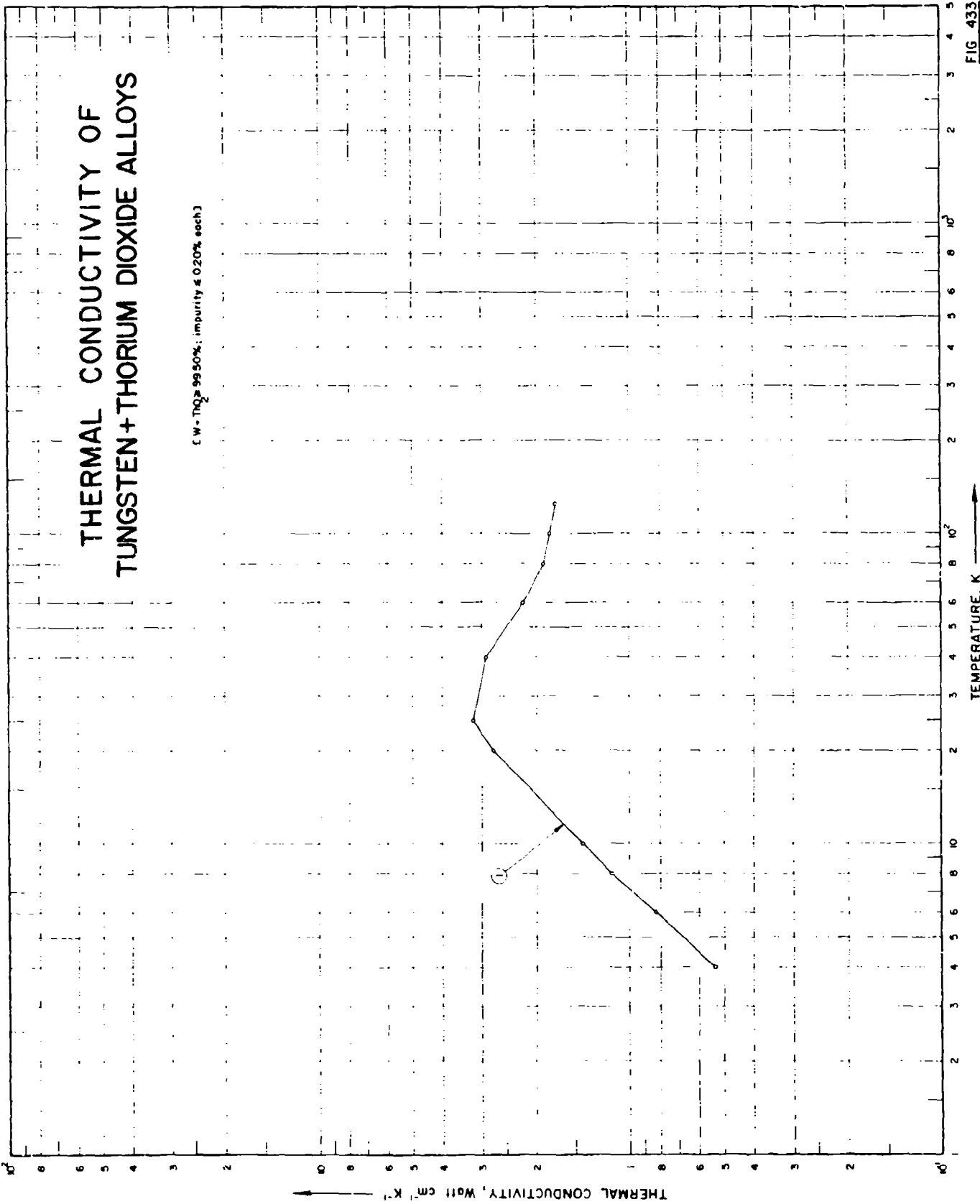


FIG. 433

## SPECIFICATION TABLE NO. 433 THERMAL CONDUCTIVITY OF [TUNGSTEN - THORIUM DIOXIDE] ALLOYS

(W + ThO<sub>2</sub> : 99.50%; impurity ± 0.20% each)

[ For Data Reported in Figure and Table No. 433 ]

Curve No.	Ref. No.	Method Used	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition ( weight percent), Specifications and Remarks
1	494		1960	4-125			0.001-0.1 Fe, 0.091-0.01 Nb, 0.01-0.1 Si, trace Al, Cr, Cu, Mg, Mn, Mo; doped by ThO <sub>2</sub> 1-2 by volume; ground to a rod with 3.67 mm dia and 13 cm long.

## DATA TABLE NO. 433 THERMAL CONDUCTIVITY OF [TUNGSTEN + THORIUM DIOXIDE] ALLOYS

(W + ThO<sub>2</sub> > 99.50%; impurity ≤ 0.20% each)[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup>K<sup>-1</sup>]

T	k
4.0	0.535
6.0	0.83
8.0	1.15
10.0	1.43
20.0	2.75
25.0	3.20
40.0	2.90
60.0	2.20
80.0	1.90
100.0	1.80
125.0	1.73

SPECIFICATION TABLE NO. 434 THERMAL CONDUCTIVITY OF URANIUM + URANIUM DIOXIDE; ALLOYS  
(U + UO<sub>2</sub> = 99.50%; impurity < 0.20%, each)

Curve No.	Ref. Method No.	Year	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent)		Composition (continued), Specifications and Remarks
						U	UO <sub>2</sub>	
1	843	1966	298.2			98.3	1.7	Spherical uranium powder obtained from National Lead Co. containing impurities: 0.05 Fe, 0.01 Mg, 0.008 Mo, 0.005 Si, < 0.005 K, < 0.005 P, < 0.005 Ti, < 0.005 Zn, < 0.002 Ca, < 0.001 As, < 0.001 Na, 0.0005 Ni, < 0.0005 Al, < 0.0005 Co, < 0.0005 Sn, 0.0004 Mn, 0.0002 Cu, 0.0001 Pb, traces of Ag, Bi, Cr, Li, Sb, Be, and B; oxidized to desired percentage by spreading over the bottom of a Petri dish and placed in an oven at 150 C; specimen contained in a 0.75 in. dia x 2 in. long cylindrical cell; mesh size -70 +80; thermal conductivity measured by using the transient line source method, the heat source was a 36-gauge constantan wire contained in a 0.025 in. O.D. hypodermic tube soldered along the axis of the cylindrical cell, data calculated from the measured line temperature at two certain times; measured in nitrogen at 1 atm.
2	843	1966	298.2			98.3	1.7	Similar to the above specimen; measured in nitrogen under pressure in the range $1.62 \times 10^{-3} \sim 5.433 \times 10^3$ mm Hg.
3	843	1966	298.2			94.9	5.1	Similar to the above specimen; measured in nitrogen at 1 atm.
4	843	1966	298.2			94.9	5.1	Similar to the above specimen; measured in nitrogen under pressure in the range $7.00 \times 10^3 \sim 4.842 \times 10^3$ mm Hg.
5	843	1966	298.2			96.6	13.4	Similar to the above specimen; measured in nitrogen at 1 atm.
6	843	1966	298.2			96.6	13.4	Similar to the above specimen; measured in nitrogen under pressure in the range $4.03 \times 10^3 \sim 5370$ mm Hg.
7	843	1966	298.2			91.6	8.4	Same impurities, source, and measuring method as the above specimen; mesh size -230 +325; measured in nitrogen at 1 atm.
8	843	1966	298.2			91.6	8.4	Similar to the above specimen; measured in nitrogen under pressure in the range $7.00 \times 10^3 \sim 4.955 \times 10^3$ mm Hg.
9	843	1966	298.2			77.8	22.2	Similar to the above specimen; measured in nitrogen at 1 atm.



DATA TABLE NO. 434 THERMAL CONDUCTIVITY OF [URANIUM + URANIUM DIOXIDE] ALLOYS

(U + UO<sub>2</sub>) = 99.50%; Impurity ≤ 0.20% each)[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup>K<sup>-1</sup>]

T	k	ρ (mm Hg)	k
<u>CURVE 1*</u>			
298.2		0.00415	
<u>CURVE 2*</u>			
ρ (mm Hg)		k	
<u>CURVE 3*</u>			
T		k	
<u>CURVE 4*</u>			
298.2		0.00390	
<u>CURVE 5*</u>			
ρ (mm Hg)		k	
<u>CURVE 6*</u>			
T		k	
<u>CURVE 7*</u>			
298.2		0.00314	
<u>CURVE 8*</u>			
T		k	
<u>CURVE 9*</u>			
298.2		0.00205	

\* No graphical presentation

## SPECIFICATION TABLE NO. 435 THERMAL CONDUCTIVITY OF [ZIRCONIUM + ZIRCONIUM DIOXIDE] ALLOYS

(Zr + ZrO<sub>2</sub> > 99.50%; impurity ≤ 0.20% each)

Curve No.	Ref. Method No.	Year Used	Temp. Range, K	Reported Error, %	Name and Specimen Designation	Composition (weight percent) Zr	Composition (weight percent) ZrO <sub>2</sub>	Composition (continued), Specifications and Remarks
1	843	-	1966	298.2		84.4	15.6	Powder specimen contained in a 0.75 in. dia x 2 in. long stainless steel cylindrical cell; mesh size -70 +80; thermal conductivity measured by using the transient line source method; the heat source was a 36 gauge constantan wire contained in a 0.025 in. O. D. hypodermic tube soldered along the axis of the cylindrical cell; data calculated from measured line temperatures at two certain times, measured in nitrogen at 1 atm.

## DATA TABLE NO. 435 THERMAL CONDUCTIVITY OF [ZIRCONIUM + ZIRCONIUM DIOXIDE] ALLOYS

(Zr + ZrO<sub>2</sub> > 99.50%; impurity ≤ 0.20% each)[Temperature, T, K; Thermal Conductivity, k, Watt cm<sup>-1</sup>K<sup>-1</sup>]

T k

CURVE 1\*

298.2 0.00293

\* No graphical presentation

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$Be_{13}Zr$	1	1256	$Bi_2Te_3 + Te$	1	1415
			$Bi_2Te_{3,19}$	1	1415
			$Bi_2Te_{3,26}$	1	1415

Material Name	Vol.	Page	Material Name	Vol.	Page
Bitter spar (see dolomite)			Brass (specific types) (continued)		
Bitumen	2	1155	Cast	1	980
Bitumin concrete	2	863	High (see yellow brass)		
Bituminous concrete aggregate, blended	2	863	High tensile	1	980
Black temper cast iron	1	1137	Leaded free cutting	1	981
Bone char	2	1156	MS 58	1	980
Bone fat	2	1072	MS 76/22/2	1	980
Boralloy (see boron nitride)			Red	1	591
Boric anhydride [ $B_2O_3$ ] (see boron oxide)			Red, German	1	981
Boric oxide [ $B_2O_3$ ] (see boron oxide)			Rolled	1	980
Boron	1	41	Yellow	1	981, 982
Boron - silicon intermetallic compounds	2	1	Brazil beryl	2	801
$SiB_4$	1	1262	Brazil topaz	2	252
$SiB_2$	1	1262	Brazil tourmaline	2	855
(tetra)Boron carbide ( $B_4C$ )	2	572	Bricks	2	859
(tetra)Boron carbide + Sodium metasilicate	2	541	Alumina fused	2	897
(tetra)Boron carbide - aluminum cermets	2	717	Aluminous fire clay	2	500
Boron trifluoride ( $BF_3$ )	3	99	Bauxite	2	329, 901, 902
Boron nitride (BN)	2	656	Carbofrax	2	897
Boron oxide ( $B_2O_3$ )	2	138	Carbofrax carborundum	2	895
Boron sesquioxide [ $B_2O_3$ ] (see boron oxide)			Carbon	2	890, 896
(di)Boron trioxide [ $B_2O_3$ ] (see boron oxide)			Carsiat carborundum	2	895
Boron silicides (see boron - silicon intermetallic compounds)			Cement porous	2	890
Boronated graphite	2	61	Ceramic	2	890
Borosilicate glass	2	923, 924	Chamotte	2	890
Borosilicate 3235 glass	2	923	Chrome	2	454, 897, 898
Borosilicate crown glass	2	923	Chrome fire brick	2	897
Boxwood	2	1061	Chrome magnesite	2	890
Brass	1	591, 592, 980, 981, 982	Chromite	2	473, 899
Brass (specific types)			Chromomagnesite	2	481
70/30	1	590	Common	2	492, 897
B. S. 249	1	981			

Material Name	Vol.	Page	Material Name	Vol.	Page
Bricks (continued)			Bricks (continued)		
Corundum	2	454, 905	Magnesia	2	485, 897, 898, 899
Dense	2	443, 904	Magnesite	2	478, 485, 892, 895, 905
Dense fireclay	2	403	Magnesite fire	2	897
Diatomaceous	2	890, 891	Magnezit	2	899, 902
Diatomaceous insulating	2	906, 907	Marksa	2	893
Dmas	2	891	Metallurgical	2	892, 893
Egyptian fire clay	2	491, 901	Metallurgical porous	2	893
Fire	2	491, 891, 895, 902, 903	Mica	2	892
Fireclay	2	403, 404, 490, 491, 896, 901, 903	Missouri fire	2	492, 905
Fire clay, dense	2	903	Normal	2	488, 489, 900, 901
Fire clay, superduty	2	890	Orzhomkide	2	899
Georgia fire	2	896	P. on, fire	2	905
Hand-burned face	2	891	P. ous	2	894
High temp. insulating	2	891	Porous concrete	2	894
High temp. insulating blast furnace	2	488, 899	Porous fire (Italy)	2	895
Hytex hydraulic pressed building	2	896	Red	2	405, 492, 898
Insulating	2	443, 891, 904	Red, hard burned	2	896
Insulating fire	2	891	Red, soft burned	2	896
Kaolin fire	2	404, 405, 904	Red shamotte	2	405
Kaolin insulating refractory	2	895	Refractory insulating	2	892
Light weight	2	488, 489, 892, 899, 900	Refractory insulating common chamotte	2	892
Lime sand	2	892	Shamotte	2	492, 894, 898
			Shamotte, white	2	405
			Silica	2	408, 489, 492, 502, 894, 896

Material Name	Vol.	Page	Material Name	Vol.	Page
Bricks (continued)			British Y-1	1	900
Silica (continued)	2	897, 898, 900, 902, 904, 906	British Y-2	1	900
			British steel	1	1114, 1118, 1187
Silica fire	2	894, 895, 905	Brom-graphite	2	768
Silica refractory	2	185	Bromine	3	13
Silicon carbide	2	555, 586, 895	Bromyride (see silver bromide)		
Silicon carbide, refrax	2	586, 906	Bronze	1	585, 586, 976, 980
Silicious	2	492, 902	Bronze, aluminum	1	531, 532, 953
Sillimanite	2	329, 902	Bronze, beryllium	1	539
Sillimanite refractory	2	329, 403, 902, 903	Bronze, phosphor	1	585, 586, 976
Sil-O-Cei	2	896	Bronze, silicon	1	973
Sil-O-Cel, calcined	2	896	Bronze, silver	1	579, 980
Sil-O-Cel, natural	2	896	B <sub>4</sub> Si	1	1262
Sil-O-Cel, special	2	896	B <sub>5</sub> Si	1	1262
Sil-O-Cel, super	2	896	Butane, 1-(i-C <sub>4</sub> H <sub>10</sub> )	3	139
Slag	2	898	Butane, n-(n-C <sub>4</sub> H <sub>10</sub> )	3	141
Spinel fire	2	905	Butaprene E rubber	2	982
Star-brand	2	185	Butter of zinc (see zinc dichloride)		
Tripolite	2	894	Cadmium	1	45
Vermiculite	2	894	Cadmium + Antimony	1	514
Zirconia	2	535, 895, 905	Cadmium - antimony intermetallic compound CdSb	1	1264
Brimstone (see sulfur)			Cadmium + Bismuth	1	517
British 2L-11	1	900	Cadmium + Bismuth + ΣX <sub>1</sub>	1	941
British C-32	1	948	Cadmium - tellurium intermetallic compound CdTe	1	267
British carbon steel	1	1186	Cadmium + Thallium	1	520
British L-5	1	925	Cadmium + Tin	1	521
British L-8	1	899	Cadmium + Zinc	1	524
			Cadmium antimonide [CdSb] (see cadmium - antimony intermetallic compound)		

Material Name	Vol.	Page	Material Name	Vol.	Page
Cadmium germanium phosphide (CdGeP <sub>2</sub> )	2	758	Ca <sub>2,02</sub> Pb	1	1271
Cadmium telluride [CdTe] (see cadmium - tellurium intermetallic compound)			Ca <sub>2,16</sub> Pb	1	1271
Caesia (see calcium oxide)			Ca <sub>2,18</sub> Pb	1	1271
Calcite	2	761	Carbofrax brick	2	897
Calcium - lead intermetallic compounds			Carbofrax carborundum brick	2	895
Ca <sub>x</sub> Pb <sub>y</sub>	1	1270	Carbon	2	5
Ca <sub>2</sub> Pb	2	1271	Diamond	2	9
Ca <sub>2,3</sub> Pb	1	1271	Graphite (see each individual graphite)		
Ca <sub>2,15</sub> Pb	1	1271	Lampblack	2	6
Ca <sub>2,15</sub> Pb	1	1271	Petroleum coke	2	6
Calcium - tin intermetallic compound			Carbon + Oxygen	2	764
Ca <sub>2</sub> Sn	1	1273	Carbon + Volatile materials	2	765
Calcium carbonate (CaCO <sub>3</sub> )	2	759	Carbon brick	2	890, 896
Calcium carbonate (CaCO <sub>3</sub> )			Carbon tetrachloride (CCl <sub>4</sub> )	3	156
Black marble	2	761	Carbon monoxide (CO)	3	151
Brown marble	2	761	Carbon monoxide - hydrogen system	3	405
Calcite	2	761	Carbon dioxide (CO <sub>2</sub> )	3	145
Marble	2	760, 761	Carbon dioxide and ethylene system	3	389
Marble powder	2	760, 761	Carbon dioxide - hydrogen system	3	391
Natural (see limestone)			Carbon dioxide - nitrogen system	3	396
White marble	2	761	Carbon dioxide - oxygen system	3	401
White Alabama marble	2	761	Carbon dioxide - propane system	3	403
Calcium difluoride (CaF <sub>2</sub> )	2	630	Carbon steel	1	1118, 1119, 1126, 1180, 1185
Calcium oxide (CaO)	2	141	Carbon steel, British	1	1186
Calcium phosphate + Lithium carbonate + Magnesium carbonate	2	763	Carbon steel, Japanese	1	1185
Calcium stannate (CaSnO <sub>3</sub> )	2	264	Carborandum	2	553, 555, 596
Calcium stannide [Ca <sub>2</sub> Sn] (see calcium - tin intermetallic compound)			Carboxy nitrile rubber	2	982
Calcium metatitanate (CaTiO <sub>3</sub> )	2	267	Cardboard	2	1109
Calcium tungstate (CaWO <sub>4</sub> )	2	270	Carsiat carborundum brick	2	895
Calcium wolframate [CaWO <sub>4</sub> ] (see calcium tungstate)			Cartridge brass 70% (see brass 70/30)		
Canadian natural graphite	2	54	Ca <sub>2</sub> Sn	1	1273
Ca <sub>2</sub> Pb	1	1271			

Material Name	Vol.	Page	Material Name	Vol.	Page
Cassiopeium (see lutetium)			Cellular glass	2	923
Cast iron	1	1129, 1130, 1133, 1134, 1136, 1137, 1205, 1222	Cellulose fiberboard	1	1110
Cast irons (specific types)			Celtium (see hafnium)		
Black temper	1	1137	Cement		
Gray	1	1130, 1135	Hydraulic (see Portland cement)	2	861
Heat resistant	1	1146	Portland	2	861
High duty	1	1133, 1137	Slag	2	861
Hot mold, gray	1	1135	Slag - Portland	2	861
Nickel-resist	1	1204	Cement porous brick	2	890
Nr 1510, spherical	1	1222	Ceramic brick	2	890
Nr 1520, pearlitic matrix	1	1222	Ceramics, miscellaneous	2	915
Soft, gray	1	1135	Cerium	1	50
White	1	1130, 1135	Cerium dioxide (CeO <sub>2</sub> )	2	144
White temper	1	1137	Cerium dioxide + Magnesium oxide	2	150
Cd <sub>3</sub> As <sub>2</sub> + Zn <sub>3</sub> As <sub>2</sub>	1	1396	Cerium dioxide + Uranium dioxide	2	153
Cd <sub>1.94</sub> Hg <sub>0.33</sub> Te	1	1408	Cerium sulfides		
Cd <sub>0.97</sub> Hg <sub>0.33</sub> Te	1	1408	CeS	2	697
CdSb	1	1264	Ce <sub>2</sub> S <sub>3</sub>	2	698
CdSb + ZnSb	1	1397	Cermets (see each individual cermet)		
CdSb + ZnSe	1	1398	Cesium	1	54
2CdSb + 3ZnSb	1	1413	Cesium bromide (CsBr)	2	565
3CdSb + 2ZnSb	1	1398	Cesium iodide (CsI)	2	561
3CdSb + 7ZnSb	1	1413	Chamotte brick	2	890
7CdSb + 3ZnSb	1	1398	Chamotte clay	2	804
CdTe	1	1267	Channel carbon black	2	764
Cd <sub>1.4</sub> Zn <sub>1.4</sub> As <sub>2</sub>	1	1396	Charcoal	2	1157
Cd <sub>2</sub> ZnAs <sub>2</sub>	1	1396	Chlorine	3	17
Cd <sub>2.5</sub> Zn <sub>2.5</sub> As <sub>2</sub>	1	1396	Chlorodifluoromethane (ClCHF <sub>2</sub> ) (see Freon 22)		
Cedar	2	1062	Chloroform (CHCl <sub>3</sub> )	3	461
Ceiba (see kapok)			Chloroform + ethyl ether system	3	479
			Chloromethane (CH <sub>3</sub> Cl) (see methylene chloride)		
			Chloroprene rubber	2	983
			Chlorotrifluoromethane (ClCF <sub>3</sub> ) (see Freon 13)		

Material Name	Vol.	Page	Material Name	Vol.	Page
Chroman	1	1015	Cobalt	1	64
Chrome brick	2	454, 897, 898	Cobalt + Carbon	1	526
Chrome fire brick	2	897	Cobalt + Chromium	1	527
Chrome magnesite brick	2	890	Cobalt + Chromium + $\Sigma X_1$	1	947
Chromel 502	1	1210	Cobalt + Iron + $\Sigma X_1$	1	950
Chromel A	1	698	Cobalt + Nickel	1	528
Chromel C	1	1036	Cobalt + Nickel + $\Sigma X_1$	1	951
Chromel P	1	698	Cobalt - silicon intermetallic compound		
Chromite brick	2	473, 899	CoSi	1	1274
Chromium	1	60	Cobalt alloys (specific types)		
Chromium + Aluminum oxide	1	1419	British C-32	1	948
Chromium + Iron + $\Sigma X_1$	1	944	Haynes stellite 21	1	948
Chromium - Nickel	1	525	Haynes stellite 23	1	948
Chromium alloy, ferrochromium	1	945	S 816	1	948
(in) Chromium trioxide + Magnesium oxide + $\Sigma X_1$	2	473	WI 52	1	948
Cinder aggregate concrete	2	869, 870	X-40	1	948
Clays	2	803	(tri) Cobalt strontium metatitanate ( $Co_3SrTiO_3$ )	2	271
Ashdhabad	2	804, 805	Cobalt zinc ferrate ( $CoZnFe_2O_4$ )	2	272
Beskudnikov	2	804	Coke, petroleum	2	765
Chamotte	2	804	Colloidal aggregate polystyrene	2	965
Fire clay	2	804	Colorless glass	2	924
Kuehn	2	804	Columbium (see niobium)		
Sandy clay	2	805	Columbium alloys (see niobium alloys)		
Clay aggregate concrete, expanded burned	2	870	Commercial castable concrete	2	871, 875, 876, 877, 878
Climax	1	1198, 1213	Common brick	2	492, 897
Coal	2	807	Concretes	2	862
Angren brown coal	2	805	Asphaltic bituminous	2	863
Donets anthracite	2	808	Barytes	2	871
Donets gas coal	2	808	Bitumin	2	863
Coal tar fractions	2	1158	Bituminous aggregate, blended	2	863
Coatings, applied (nonmetallic)	2	1009	Cinder aggregate	2	869, 870
			Clay aggregate, expanded burned	2	870



Material Name	Vol.	Page	Material Name	Vol.	Page
Concretes (continued)			Copper, electrolytic tough pitch	1	70, 72
Commercial castable	2	871, 875, 876, 877, 878	Copper, free-cutting	1	582
Diatomaceous aggregate	2	874	Copper, oxygen-free high-conducting	1	69, 74
Haydite aggregate	2	870	Copper, phosphorus deoxidized	1	72
Leuna slag	2	864	Copper-126, leaded	1	555
Light weight	2	874	Copper + Aluminum	1	530
Light weight, foamed	2	881	Copper + Aluminum + $\Sigma X_1$	1	952
Limestone aggregate	2	869	Copper + Antimony	1	534
Limestone gravel	2	864, 865	Copper - antimony - selenium intermetallic compound $CuSbSe_2$	1	1275
Lumnite cement	2	871	Copper + Arsenic	1	535
Metallurgical pumice	2	863, 864	Copper + Beryllium	1	538
Paraffin	2	863	Copper + Beryllium + $\Sigma X_1$	1	955
Portland cement	2	871	Copper + Cadmium	1	541
Sand cement	2	874	Copper + Cadmium + $\Sigma X_1$	1	956
Sand and gravel aggregate	2	868, 869	Copper + Chromium	1	542
Slag	2	864, 880, 881	Copper + Cobalt	1	545
Slag, direct process	2	864	Copper + Cobalt + $\Sigma X_1$	1	957
Slag, expanded	2	878, 879	Copper + Gold	1	548
Slag aggregate, limestone treated	2	870	Copper + Iron	1	551
Cond-Al	1	906	Copper + Iron + $\Sigma X_1$	1	960
Constantan	1	564	Copper + Lead	1	554
Contraacid	1	1036	Copper + Lead + $\Sigma X_1$	1	961
Contraacid B 7 M	1	1036	Copper + Manganese	1	557
Copoly(chloroethylene-vinyl-acetate)	2	943	Copper + Manganese + $\Sigma X_1$	1	964
Copoly-[1,1-difluoro-ethylene-hexafluoro-propene], Viton A rubber (see Viton rubber)			Copper + Nickel	1	561
Copoly(tormaldehyde-urea)	2	944	Copper + Nickel + $\Sigma X_1$	1	969
Copper	1	68	Copper + Palladium	1	568
Copper, coalesced	1	69, 72	Copper + Phosphorus	1	571
Copper, electrolytic	1	72, 73	Copper + Platinum	1	574
			Copper - selenium intermetallic compound $Cu_3Se_2$	1	1276
			Copper + Silicon	1	575
			Copper + Silicon + $\Sigma X_1$	1	972
			Copper + Silver	1	578

Material Name	Vol.	Page	Material Name	Vol.	Page
Copper + Tellurium	1	581	Copper alloys (specific types) (continued)		
Copper + Tin	1	584	Cupralloy type 5, Russian	1	543
Copper + Tin + $\Sigma X_1$	1	975	Cupro nickel	1	976
Copper + Zinc	1	588	Cupro nickel, NM-81, Russian	1	562
Copper + Zinc + $\Sigma X_1$	1	979	Eureka	1	563
Copper + Zirconium + $\Sigma X_1$	1	985	German silver	1	980, 981
Copper alloys (specific types)			Gun-metal, admiralty	1	976
Advance	1	970	Gun-metal, ordinary	1	976
ASTM B301-58T	1	582	Lohn	1	564
Beryllium copper	1	539	Manganin	1	965
Brass	1	591, 592, 980, 981, 982	Manganin NM Mts	1	965
Brass 70-30	1	570	Navy M	1	977
Brass B.S. 249	1	981	Nickel silver	1	981
Brass, cast	1	980	SAE bearing alloy 40	1	976
Brass, high tensile	1	989	SAE bearing alloy 62	1	976
Brass, leaded free cutting	1	981	SAE bearing alloy 64	1	976
Brass MS 58	1	980	SAE bearing alloy 66	1	962
Brass MS 76-22-2	1	980	Copper glance - see (di)copper sulfide,		
Brass, red, German	1	591, 981	Copper iodide (CuI)	2	562
Brass, rolled	1	986	Copper hemioxide (Cu <sub>2</sub> O) - see (di)copper oxide		
Brass, yellow	1	981, 982	(di)Copper oxide (Cu <sub>2</sub> O)	2	147
Bronze	1	585, 586, 976, 980	Copper protoxide (Cu <sub>2</sub> O) - see (di)copper oxide		
Bronze, aluminum	1	531, 532, 953	Copper selenide (Cu <sub>3</sub> Se <sub>2</sub> , (see copper - selenium intermetallic compound)		
Bronze, beryllium	1	539	(di)Copper sulfide (Cu <sub>2</sub> S)	2	699
Bronze, phosphor	1	585, 586, 976	(di)Copper sulfide + Iron sulfide + (tri)Nickel disulfide	2	700
Bronze, silicon	1	973	(di)Copper sulfide + (tri)Nickel disulfide	2	701
Bronze, silver	1	579, 980	Copperous oxide (Cu <sub>2</sub> O) - see (di)copper oxide		
Constantan	1	564	Copperous sulfide (Cu <sub>2</sub> S) - see (di)copper sulfide		
			Cordierite	2	918
			Cordierite 202	2	919
			Rutgers	2	919
			Steatite	2	919

Material Name	Vol.	Page	Material Name	Vol.	Page
Cork	2	1063	$(\text{CuSbSe}_2)_{0.8}(\text{Cu}_3\text{Se}_2)_{0.2}$	1	1400
Corning 0080 glass	2	511, 928	$(\text{CuSbSe}_2)_{0.9}(\text{Cu}_3\text{Se}_2)_{0.1}$	1	1400
Corning 7740 glass	2	933	$\text{Cu}_3\text{Se}_2$	1	1276
Cornstalk wallboard	2	1111	$\text{Cu}_3\text{Se}_2 + \text{CuSbSe}_2$	1	1401
Corronil	1	1032	"D" nickel	1	1039
Corundum	2	94, 99	Decane, n- $(\text{C}_{10}\text{H}_{22})$	3	164
Corundum brick	2	454, 905	Dense brick	2	443, 904
Cotton	2	1068	Deuterium	3	21
Waste	2	1070	Deuterium - hydrogen system	3	407
Medical	2	1069, 1070	Deuterium - nitrogen system	3	410
Cotton fabric	2	1093	Diamond	2	9
Cotton silicate felt fabric	2	1094	Type I	2	10
Cotton wool	2	1096	Type II	2	10
Crucible HNM	1	1168	Diatomaceous aggregate concrete	2	874
Crucible steel, Japanese	1	1204	Diatomaceous brick	2	890, 891
Cu + BeCo	1	1420	Diatomaceous earth	2	814
CuAu	1	1281	Diatomaceous insulating brick	2	906, 907
$\text{Cu}_3\text{Au}$	1	1281	Diatomite (see diatomaceous earth)		
Cupralloy, Russian, type 5	1	543	Diatomite aggregate	2	1112
Cupronickel	1	970	Sil-O-Cel coarse grade	2	1112
Cupronickel, Russian, NM-81	1	562	Dichlorodifluoromethane $(\text{Cl}_2\text{CF}_2)$ (see Freon 12)		
Cupram (see copper)			Dichlorofluoromethane $(\text{Cl}_2\text{CHF})$ (see Freon 21)		
$\text{CuSbSe}_2$	1	1275	1, 2-Dichloro-1, 1, 2, 2-tetrafluoroethane $(\text{CCl}_2\text{F}_2\text{CClF}_2)$ (see Freon 114)		
$\text{CuSbSe}_2 + \text{Cu}_3\text{Se}_2$	1	1400	Diethylamine - ethyl ether system	3	472
$(\text{CuSbSe}_2)_{0.1}(\text{Cu}_3\text{Se}_2)_{0.9}$	1	1401	Dimethyl ketone $(\text{CH}_3)_2\text{CO}$ (see acetone)		
$(\text{CuSbSe}_2)_{0.2}(\text{Cu}_3\text{Se}_2)_{0.8}$	1	1401	Dimethyl methane $(\text{C}_2\text{H}_6)$ (see propane)		
$(\text{CuSbSe}_2)_{0.3}(\text{Cu}_3\text{Se}_2)_{0.7}$	1	1401	Dinas brick	2	891
$(\text{CuSbSe}_2)_{0.4}(\text{Cu}_3\text{Se}_2)_{0.6}$	1	1401	Diphenyl $(\text{C}_6\text{H}_5)_2\text{C}$	2	989
$(\text{CuSbSe}_2)_{0.5}(\text{Cu}_3\text{Se}_2)_{0.5}$	1	1401	Diphenylamine $(\text{C}_6\text{H}_5)_2\text{NH}$	2	991
$(\text{CuSbSe}_2)_{0.6}(\text{Cu}_3\text{Se}_2)_{0.4}$	1	1400	Diphenylmethane + Naphthalene	2	994
$(\text{CuSbSe}_2)_{0.7}(\text{Cu}_3\text{Se}_2)_{0.3}$	1	1400	Diphenyl oxide $(\text{C}_6\text{H}_5)_2\text{O}$	2	990
			Dolomite	2	810

Material Name	Vol.	Page	Material Name	Vol.	Page
Dolomite (continued)			Enamel (continued)		
NTS dolomite	2	811	Silicon	2	921
Domestic graphite, Japan	2	56	Erbium	1	86
Donets anthracite coal	2	808	Ethane (C <sub>2</sub> H <sub>6</sub> )	3	167
Donets gas coal	2	808	Ethanol [C <sub>2</sub> H <sub>5</sub> OH] (see ethyl alcohol)		
Dow metal	1	999	<del>ethanol</del> - argon system Dimethyl ether	3	454
Duralumin	1	896	<del>ethanol</del> - methyl formate system Dimethyl ether	3	474
Duranickel	1	1010	<del>ethanol</del> - propane system Dimethyl ether	3	456
Duranickel alloy 301 (see duranickel)			Ethyl alcohol (C <sub>2</sub> H <sub>5</sub> OH)	3	169
Duroid 5600	2	968	Ethyl ether (C <sub>2</sub> H <sub>5</sub> ) <sub>2</sub> O	3	179
Dyna quartz fiber	2	1144	Ethylene (CH <sub>2</sub> CH <sub>2</sub> )	3	173
Dysprosium	1	82	Ethylene - hydrogen system	3	413
Earth	2	813	Ethylene - methane system	3	415
Diatomaceous	2	814	Ethylene - nitrogen system	3	417
Kieselguhr	2	814	Ethylene glycol (CH <sub>2</sub> OHCH <sub>2</sub> OH)	3	177
Kieselguhr, ignited	2	814	Eureka	1	563
Kieselguhr, ordinary	2	814	Europium	1	90
Easy-Flow silver solder silver alloy	1	1059	Excellstor	2	1113
Ebonite rubber	2	971	Fat	2	1072
Egyptian fire clay brick	2	491, 901	Beef	2	1072
EI-257, Russian	1	1166, 1214	Bone	2	1073
EI-415, Russian	1	1022	Pig	2	1073
EI-573, Russian	1	1167	Ferrocobaltitium, Russian	1	1081
EI-609, Russian	1	1167	Ferrocromium, Russian	1	945
EI-607, Russian	1	1019, 1020, 1021	Ferromanganese, Russian	1	684, 1010
EI-802, Russian	1	1156, 1157	Ferromanganese, low carbon, Russian	1	1010
EI-855, Russian	1	1214	Ferromanganese, normal, Russian	1	1010
Elastomer rubber	2	974	Ferromolybdenum, Russian	1	690, 1013
Elekton 2	1	999	Ferrosilicon, Russian	1	765
Electrical porcelain	2	937	Ferrosilicon 45%, Russian	1	1218
Electrolytic iron	1	157, 159	Ferrotitanium, Russian	1	1225
Enamel	2	921	Ferrotungsten, Russian	1	1090
			Ferrovandium, Russian	1	875
			Ferrum (see iron)		

Material Name	Vol.	Page	Material Name	Vol.	Page
Fiberglass	2	1115	Fused quartz [see silicon dioxide (fused)]		
Fiberite	2	1052	GaAs	1	1277
Fir	2	1073	GaAs + GaP	1	1423
Fir plywood	2	1114	GaAs <sub>0.5</sub> P <sub>0.5</sub>	1	1424
Fire brick	2	491, 591, 895, 902, 903	GaAs <sub>0.55</sub> P <sub>0.36</sub>	1	1424
			GaAs <sub>0.67</sub> P <sub>0.33</sub>	1	1424
			GaAs <sub>0.8</sub> P <sub>0.2</sub>	1	1424
			GaAs <sub>0.8</sub> P <sub>0.1</sub>	1	1424
Fire clay	2	804	Gabbro	2	816
Fire clay, Aluminous	2	489	Gadolinium	1	93
Fire clay, light weight	2	403, 404	Gadolinium oxide + Samarium oxide	2	356
Fire clay, pressed	2	403	Gallium	1	87
Fire clay brick	2	403, 404, 490, 491, 590, 901, 904	Gallium - arsenic intermetallic compound GaAs	1	1277
			Gallium arsenide [GaAs] (see gallium - arsenic intermetallic compound)		
			Garnet [M <sub>1</sub> <sup>II</sup> M <sub>2</sub> <sup>III</sup> (SiO <sub>4</sub> ) <sub>3</sub> ]	2	278
			Genetron 11 [Cl <sub>2</sub> CF] (see Freon 11)		
			Genetron 12 [Cl <sub>2</sub> CF <sub>2</sub> ] (see Freon 12)		
Fire clay brick, aluminous	2	900	Genetron 13 [ClCF <sub>3</sub> ] (see Freon 13)		
Fire clay brick, dense	2	901	Genetron 22 [ClCHF <sub>2</sub> ] (see Freon 22)		
Fire clay brick, superduty	2	890	Genetron 113 [CCl <sub>2</sub> FCClF <sub>2</sub> ] (see Freon 113)		
Fission alloy	1	1095	Genetron 114 [CClF <sub>2</sub> CClF <sub>2</sub> ] (see Freon 114)		
Flowers of tin (see tin dioxide)			Georgia fire brick	2	896
Fluorine	3	26	German chromite	1	1018
Foam glass	2	924, 925	German silver	1	980, 981
Forsterite (Mg <sub>2</sub> SiO <sub>4</sub> )	2	275	German steel	1	1118
Freon 10 [CCl <sub>4</sub> ] (see carbon tetrachloride)			German Y alloy	1	896, 898
Freon 11 (Cl <sub>2</sub> CF)	3	183	Germanium	1	108
Freon 12 (Cl <sub>2</sub> CF <sub>2</sub> )	3	187	Germanium + Silicon	1	597
Freon 13 (ClCF <sub>3</sub> )	3	191	Germanium - tellurium intermetallic compound GeTe	1	1280
Freon 20 [CHCl <sub>3</sub> ] (see chloroform)			Germanium 74, enriched	1	112
Freon 21 (Cl <sub>2</sub> CHF)	3	193	Germanium telluride [GeTe] (see germanium- tellurium intermetallic compound)		
Freon 22 (ClCHF <sub>2</sub> )	3	197	GeTe	1	1280
Freon 113 (CCl <sub>2</sub> FCClF <sub>2</sub> )	3	201			
Freon 114 (CClF <sub>2</sub> CClF <sub>2</sub> )	3	205			
Fuel-filled graphite	2	545, 548, 555			

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Glasses	2	922	Glasses (continued)		
Aluminate silicate 723	2	923	Soda-lime silica	2	511, 924, 927
Amber	2	924	Soda-lime silica plate 9330	2	923
Borosilicate	2	923, 924	Soft	2	511
Borosilicate 3235	2	923	Solex 2808 plate	2	923
Borosilicate crown	2	923	Solex 2808 X	2	925
Cellular	2	923	Solex "S"	2	925
Colorless	2	924	Soldex "S" plate	2	923
Corning 9050	2	511, 926	Thuringian	2	923, 924
Foam	2	924, 925	Vycor-brand	2	926
Golden plate (see amber glass)			White plate	2	923, 925
Green	2	923	Window	2	923, 924
Jena Geräte	2	924	X-ray protection	2	924
Lead	2	923	Glass fiber blankets (same as fiberglass)	2	1115
Monax	2	924	Insulation	2	1117
Phoenix	2	924	Superfine	2	1116
Plate	2	923, 924, 925, 926	Glass fiber board	2	1124
Pyrex	2	499, 923, 924, 926, 927	Glucinum (see beryllium)		
Pyrex 7740	2	499, 923, 924, 925, 926	Glycerol ( $\text{CH}_2\text{OHCHOHCH}_2\text{OH}$ )	3	209
Quartz	2	923, 924	Gnome salt	2	832
Silica	2	923, 925, 926	Gold	1	132
Silica, fused	2	925	Gold + Cadmium	1	600
Silicate	2	511	Gold + Chromium	1	603
Soda	2	923	Gold + Cobalt	1	606
Soda-lime	2	926	Gold + Copper	1	609
Soda-lime plate	2	926	Gold - copper intermetallic compounds		
			$\text{Au}_x\text{Cu}_y$	1	1281
			$\text{CuAu}$	1	1282
			$\text{Cu}_3\text{Au}$	1	1282
			Gold + Palladium	1	614
			Gold + Platinum	1	617
			Gold + Silver	1	620

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Gold + Zinc	1	623	Graphite (continued)		
Golden plate glass (see amber glass)			Grade CEQ	2	63, 65
Government rubber-styrene rubber	2	977	Grade CFW	2	67
Granite	2	517	Grade CFZ	2	67, 71, 72
NTS granite	2	518			
Graphite	2	53	Grade CS	2	54, 55, 56, 64
Acheson	2	73			
Boronated	2	61	Grade CS-112	2	63
British reactor grade A	2	69	Grade CS-212	2	63
British reactor grade carbon	2	69, 70	Grade CSF	2	55
Brom-graphite	2	768	Grade CSF-MTR	2	63
Brookhaven	2	26	Grade EY 9	2	69, 70, 71
Canadian natural graphite	2	54	Grade EY 9A	2	70
Carbon resistor	2	73	Grade G-5	2	60, 61
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Domestic, Japan	2	56	Grade GBE	2	54, 55
Fuel-filled	2	545, 548, 558	Grade GBH	2	55
Grade 875 S	2	45	Grade H4LM	2	61
Grade 890 S	2	49	Grade JTA	2	70, 72
Grade AGA	2	64	Grade L-117	2	63
Grade AGHT	2	57	Grade MH4LM	2	70
Grade AGOT	2	13	Grade P1	2	35
Grade AGOT-KC	2	17	Grade R-0008	2	60
Grade AGOT-CSF-MTR	2	17	Grade R0025	2	71
Grade AGSR	2	57, 58, 63, 64	Grade RT-0003	2	54
Grade AGSX	2	64	Grade RVA	2	66, 67
Grade ATJ	2	20	Grade RVD	2	67
Grade ATL	2	64	Grade SA-25	2	42
Grade ATL-82	2	71	Grade TS-148	2	59
Grade AUC	2	63, 64, 65	Grade TS-160	2	59
Grade AWG	2	24			
Grade CDG	2	65			

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Graphite (continued)			Greenheart	2	1074
Nuclear grade TSP	2	60	Gulton HS. B aluminum oxide	2	103
Grade ZT	2	60, 61, 71	Gun metal, admiralty	1	976
Grade ZTA	2	65, 66, 70	Gun <del>metal</del> <sup>metal</sup> , ordinary	1	976
Grade ZTB	2	66	"H" Monel	1	1032
Grade ZTC	2	66	Hafnia (see hafnium oxide)		
Grade ZTD	2	66	Hafnium	1	138
Grade ZTE	2	66	Hafnium - boron intermetallic compound HfB <sub>2</sub>	1	1254
Grade ZTF	2	66	Hafnium + Zirconium	1	624
Graphitized carbon black	2	60	Hafnium carbide (HfC)	2	575
Karbate	2	59	Hafnium nitride (HfN)	2	659
Korite	2	55	Hafnium oxide (HfO <sub>2</sub> )	2	150
Moderator graphite	2	70	Hair felt	2	1099
Natural Ceylon block	2	55	Hand-burned face brick	2	891
Ohmite	2	73	Hardwood	2	1075
Pencil lead graphite	2	65	Hastelloy A	1	1036
Porous-40	2	63	Hastelloy B	1	1042
Porous-60	2	63	Hastelloy C	1	1018
Pyrolytic	2	32	Hastelloy R-235	1	1019
Pyrolytic graphite filament	2	32	Haydite aggregate concrete	2	870
Reactor grade carbon stock	2	73	Haynes alloy N-155	1	1177
Spektral Kohle 1	2	54	Haynes alloy Nb-752	1	1056
Supertemp pyrolytic	2	72	Haynes stellite alloy 21	1	948
U. B. carbon	2	62	Haynes stellite alloy 23	1	948
U. B. graphite	2	62	Haynes stellite alloy 27	1	1029
Graphite + Bromine	2	767	Haynes stellite alloy 31 (same as cobalt alloy X40)	1	948
Graphite + Thorium dioxide	2	544	Heavy hydrogen (see deuterium, or tritium)		
Graphite + Uranium dicarbide	2	770	Helium	3	29
Graphite + Uranium dioxide	2	547	Helium - air system	3	318
Gray cast iron	1	1130, 1135	Helium - argon - krypton system	3	481
Gray cast iron, hot moid	1	1135	Helium - argon - nitrogen system	3	486
Green glass	2	923	Helium - argon - xenon system	3	479
			Helium - n-butane system	3	320
			Helium - carbon dioxide system	3	322



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Helium - cyclopropane system	3	325	Hydrargyrum (see mercury)		
Helium - deuterium system	3	327	Hydriodic acid [HI] (see hydrogen iodide)		
Helium - ethane system	3	329	Hydrochloric acid [HCl] (see hydrogen chloride)		
Helium - ethylene system	3	331	Hydrogen	3	41
Helium - hydrogen system	3	333	Hydrogen - oxygen system	3	429
Helium - krypton system	3	276	Hydrogen - nitrogen system	3	419
Helium - krypton - xenon system	3	480	Hydrogen - nitrogen - ammonia system	3	500
Helium - methane system	3	338	Hydrogen - nitrogen - oxygen system	3	498
Helium - neon system	3	271	Hydrogen - nitrous oxide system	3	427
Helium - neon - deuterium system	3	489	Hydrogen chloride (HCl)	3	101
Helium - neon - xenon system	3	482	Hydrogen iodide (HI)	3	103
Helium - nitrogen system	3	340	Hydrogen sulfide (H <sub>2</sub> S)	3	104
Helium - nitrogen - methane system	3	487	Hypalon S2 rubber	2	983
Helium - oxygen system	3	343	Hypo (see sodium thiosulfate)		
Helium - oxygen - methane system	3	484	Hytex hydraulic pressed building brick	2	896
Helium - propane system	3	345	Ignited alumina	2	106
Helium - propylene system	3	347	Illinium (see promethium)		
Helium - xenon system	3	280	InAs	1	1292
Heptane, n-(C <sub>7</sub> H <sub>16</sub> )	3	211	InAs + InP	1	1426
Hevea rubber	2	983	InAs <sub>0.6</sub> P <sub>0.4</sub>	1	1427
Hexane, n-(C <sub>6</sub> H <sub>14</sub> )	3	214	InAs <sub>0.8</sub> P <sub>0.2</sub>	1	1427
HfB <sub>2</sub>	1	1284	InAs <sub>0.9</sub> P <sub>0.1</sub>	1	1427
HgSe	1	1320	InAs <sub>0.96</sub> P <sub>0.06</sub>	1	1427
HgTe	1	1321	Inco "713 C"	1	1022
HgTe + CdTe	1	1407	Inconel	1	1018, 1019, 1021
Hi alumina	2	99	Inconel alloy 600 (see inconel)		
High carbon steel, Japanese	1	1119	Inconel alloy 702	1	1022
High-perm-49	1	1199	Inconel alloy 713 (see Inco "713 C")		
High temp. insulating brick	2	891	Inconel alloy X-750 (see Inconel X)		
High temp. insulating blast furnace brick	2	899	Inconel X	1	1018
High zircon porcelain	2	937	India beryl	2	801
Holmium	1	142	Indiana limestone	2	821
Honeycomb structures (metallic - nonmetallic)	2	1015	Indium	1	146
Honeycomb structures (nonmetallic)	2	1010			
Hutchins alloy	1	512			

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Indium - antimony intermetallic compound InSb	1	1287	Iron	1	156
Indium - arsenic intermetallic compound InAs	1	1292	Iron + Aluminum + $\Sigma X_1$ (I)	1	1142
Indium + Lead	1	627	Iron + Aluminum + $\Sigma X_1$ (II)	1	1145
Indium - selenium intermetallic compound In <sub>2</sub> Se <sub>3</sub>	1	1295	Iron + Carbon + $\Sigma X_1$ (I) (C $\leq$ 2.00%)	1	1113
Indium - tellurium intermetallic compound In <sub>2</sub> Te <sub>3</sub>	1	1298	Iron + Carbon + $\Sigma X_1$ (II) (C $\leq$ 2.00%)	1	1124
Indium + Thallium	1	630	Iron + Carbon + $\Sigma X_1$ (I) (C > 2.00%)	1	1128
Indium + Tin	1	634	Iron + Carbon + $\Sigma X_1$ (II) (C > 2.00%)	1	1132
Indium antimonide [InSb] (see indium - antimony intermetallic compound)			Iron + Chromium + $\Sigma X_1$ (I)	1	1148
Indium arsenide [InAs] (see indium - arsenic intermetallic compound)			Iron + Chromium + $\Sigma X_1$ (II)	1	1152
Indium oxide (InO)	2	153	Iron + Chromium + Nickel + $\Sigma X_1$ (I)	1	1160
Indium selenide [In <sub>2</sub> Se <sub>3</sub> ] (see indium - selenium intermetallic compound)			Iron + Chromium + Nickel + $\Sigma X_1$ (II)	1	1164
Indium telluride [In <sub>2</sub> Te <sub>3</sub> ] (see indium - tellurium intermetallic compound)			Iron + Cobalt + $\Sigma X_1$ (II)	1	1176
Ingot iron	1	1134	Iron + Copper + $\Sigma X_1$ (I)	1	1179
InSb	1	1287	Iron + Manganese + $\Sigma X_1$ (I)	1	1182
InSb + In <sub>2</sub> Te <sub>3</sub>	1	1403	Iron + Manganese + $\Sigma X_1$ (II)	1	1191
In <sub>2</sub> Se <sub>3</sub>	1	1295	Iron + Molybdenum + $\Sigma X_1$ (II)	1	1194
Insulating brick	2	443, 891, 904	Iron + Nickel + $\Sigma X_1$ (I)	1	1197
Insulating fire brick	2	891	Iron + Nickel + $\Sigma X_1$ (II)	1	1202
Insulation fiberglass	2	1117	Iron + Nickel + Chromium + $\Sigma X_1$ (I)	1	1209
Insurok	2	1023, 1024	Iron + Nickel + Chromium + $\Sigma X_1$ (II)	1	1212
In <sub>2</sub> Te <sub>3</sub>	1	1298	Iron + Phosphor + $\Sigma X_1$ (I)	1	1216
In <sub>2</sub> Te <sub>3</sub> + Cu <sub>2</sub> Te + Ag <sub>2</sub> Te	1	1406	Iron + Silicon + $\Sigma X_1$ (I)	1	1217
Intermetallic compounds (see each individual intermetallic compound)			Iron + Silicon + $\Sigma X_1$ (II)	1	1221
Invar	1	1199	Iron + Titanium + $\Sigma X_1$ (I)	1	1225
Invar, free cut	1	1205	Iron + Tungsten + $\Sigma X_1$ (I)	1	1226
Iodine	2	83	Iron + Tungsten + $\Sigma X_1$ (II)	1	1229
Iodyride [AgI] (see silver iodide)			Iron, Armco	1	157, 158, 159, 160, 161, 163
Ionium (see thorium)			Iron, electrolytic	1	157, 159
Iridium	1	152	Iron, nodular	1	1222
			Iron, silal	1	1222, 1223
			Iron, Swedish	1	158

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Iron, wrought	1	1185, 1219	Kieselguhr earth	2	814
(tri)iron carbide ( $Fe_3C$ )	2	578	Kieselguhr earth, ignited	2	814
(tri)iron tetraoxide ( $Fe_3O_4$ )	2	154	Kieselguhr earth, ordinary	2	814
Iron oxide, magnetic [ $Fe_3O_4$ ] (see (tri)iron tetraoxide)			Knapic	1	327
Isotron 11 (see Freon 11)			Koldboard	2	1125
Isotron 12 (see Freon 12)			Korite graphite	2	55
Isotron 13 (see Freon 13)			Kovar	1	1203
Isotron 22 (see Freon 22)			Krupp steel	1	1115, 1184
Isotron 113 (see Freon 113)			Krypton	3	50
Isotron 114 (see Freon 114)			Krypton - deuterium system	3	349
Ivory	2	1076	Krypton - hydrogen system	3	351
African	2	1076	Krypton - neon system	3	284
Japanese 2E-8	1	899	Krypton - nitrogen system	3	354
Japanese fish-plate	1	1119	Krypton - oxygen system	3	356
Japanese M-1	1	899	Krypton - xenon system	3	288
Japanese steel	1	1195, 1210	Kuchin clay	2	804
Jena Geräte glass	2	924	"L" nickel	1	238, 239
Jodium (see iodine)			Lamicoid	2	1023, 1024
"K" Monel	1	1032	Laminates (metallic - nonmetallic)	2	1036
K. S. alloy 245	1	920	Laminates (nonmetallic)	2	1021
K. S. alloy 250	1	920	Armalen	2	1032
K. S. alloy special	1	902	Astrolite	2	1029, 1030
K. S. magnet steel	1	1177	Insurok	2	1023, 1024
Kalium (see potassium)			Lamicoid	2	1023, 1024
Kaolin fire brick	2	404, 405, 904	Scotchply	2	1025
Kaolin insulating refractory brick	2	895	Laminate, epoxy resin (see scotch ply laminate)		
Kapok	2	1077	Lamphlack	2	6
Karbate graphite	2	59	Lanthanum	1	171
Kel-F	2	970	Lanthanum + Neodymium + $\Sigma N_3$	1	988
Kel-F 3700	2	983	Lanthanum - selenium intermetallic compound		
Kennametals K161B	2	728	LaSe	1	1301
Ketopropane [ $(CH_3)_2CO$ ] (see acetone)					
Kh80 T, Russian	1	1019			

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Lanthanum - tellurium intermetallic compound			Lignum Vitae	2	1079
LaTe	1	1304	Lime sand brick	2	892
Lanthanum trifluoride (LaF <sub>3</sub> )	2	633	Limestone	2	820
Lanthanum selenide [LaSe] (see lanthanum - selenium intermetallic compound)			Indiana	2	821
Lanthanum sulfide (LaS)	2	702	Queenstone grey	2	821
Lanthanum telluride [LaTe] (see lanthanum - tellurium intermetallic compound)			Rama	2	821
LaSe	1	1301	Limestone aggregate concrete	2	869
LaTe	1	1304	Limestone gravel concrete	2	864, 865
Laughing gas (see nitrous oxide)			Lipowitz alloy	1	939
Lead	1	175	Lithia (see lithium oxide)		
Lead, pyrometric standard	1	183, 184	Lithium	1	192
Lead + Antimony	1	637	Lithium + Boron + $\Sigma X_1$	1	992
Lead + Antimony + $\Sigma X_1$	1	991	Lithium + Sodium	1	955
Lead + Bismuth	1	640	Lithium + Sodium + $\Sigma X_1$	1	995
Lead + Indium	1	643	Lithium fluoride (LiF)	2	636
Lead + Silver	1	646	Lithium fluoride + Potassium fluoride + $\Sigma X_1$	2	641
Lead - tellurium intermetallic compound			Lithium hydride (LiH)	2	773
PbTe	1	1307	Lithium oxide (Li <sub>2</sub> O)	2	157
Lead + Thallium	1	649	Loam	1	564
Lead + Tin	1	652	Low alloy steel	1	1213
Lead alloy, SAE bearing alloy 12	1	991	Low-exp-42	1	1205
Lead glass	2	923	Lowell sand	2	834, 835
Lead oxide + Silicon dioxide	2	359	Lucalox	2	106
Lead oxide + Silicon dioxide + $\Sigma X_1$	2	474	Lummit cement concrete	2	871
Lead telluride [PbTe] (see lead - tellurium intermetallic compound)			Lutetium	1	198
Lead metatitanate (PbTiO <sub>3</sub> )	2	279	Macloy G steel	1	1213
Lead zirconate (PbZrO <sub>3</sub> )	2	282	Magnalium	1	478
Light weight brick	2	485, 489, 892, 899, 900	Magnesia (see magnesium oxide)		
Light weight concrete	2	874	Magnesia brick	2	485, 897, 898, 899
Light weight concrete, foamed	2	881	Magnesite brick	2	478, 483, 892, 895, 905

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Magnesite fire brick	2	597	Magnesium aluminates (continued)		
Magnesium	1	202	Natural ruby spinel	2	284
Magnesium + Aluminum	1	658	Spinel	2	284
Magnesium + Aluminum + $\Sigma X_1$	1	998	Synthetic spinel	2	287
Magnesium - antimony intermetallic compound			Magnesium aluminate + Magnesium oxide	2	362
$Mg_3Sb_2$	1	1316	Magnesium aluminate + Silicon dioxide	2	365
Magnesium + Cadmium	1	661	Magnesium aluminate + (di)Sodium oxide	2	368
Magnesium + Calcium	1	662	Magnesium metaaluminate [ $MgAl_2O_4$ ] (see magnesium aluminate)		
Magnesium + Cerium	1	663	Magnesium antimonide [ $Mg_3Sb_2$ ] (see magnesium - antimony intermetallic compound)		
Magnesium + Cerium + $\Sigma X_1$	1	1001	Magnesium carbonate ( $MgCO_3$ )	2	776
Magnesium + Cobalt + $\Sigma X_1$	1	1004	Magnesium oxide ( $MgO$ )	2	158
Magnesium + Copper	1	666	Magnesium oxide + Beryllium oxide	2	371
Magnesium + Copper + $\Sigma X_1$	1	1005	Magnesium oxide + Calcium oxide + $\Sigma X_1$	2	477
Magnesium - germanium intermetallic compound			Magnesium oxide + (di)Chromium trioxide + $\Sigma X_1$	2	480
$Mg_2Ge$	1	1311	Magnesium oxide + Clay	2	374
Magnesium + Manganese	1	669	Magnesium oxide + (di)Iron trioxide + $\Sigma X_1$	2	483
Magnesium + Nickel	1	672	Magnesium oxide + Magnesium aluminate	2	375
Magnesium + Nickel + $\Sigma X_1$	1	1008	Magnesium oxide + Magnesium orthosilicate	2	378
Magnesium + Silicon	1	675	Magnesium oxide + Nickel oxide	2	381
Magnesium - silicon intermetallic compound			Magnesium oxide + Silicon dioxide	2	384
$Mg_2Si$	1	1314	Magnesium oxide + Silicon dioxide + $\Sigma X_1$	2	484
Magnesium + Silver	1	678	Magnesium oxide + Tale	2	550
Magnesium + Tin	1	679	Magnesium oxide + Tin dioxide	2	387
Magnesium - tin intermetallic compound			Magnesium oxide + Uranium dioxide	2	390
$Mg_2Sn$	1	1317	Magnesium oxide + Zinc oxide	2	391
Magnesium + Zinc	1	680	Magnesium silicate (see Forsterite)		
Magnesium alloys (specific types)			Magnesium orthosilicate + Magnesium oxide	2	394
AN-M-29	1	999	Magnesium silicide [ $Mg_2Si$ ] (see magnesium - silicon intermetallic compound)		
AZ 31 A (see magnesium alloy, AN-M-29)			Magnesium stannate ( $MgSnO_3$ )	2	289
Dow metal	1	999	Magnesium stannide [ $Mg_2Sn$ ] (see magnesium - tin intermetallic compound)		
Elekton 2	1	999	Magnesium titanate porcelain	2	937
Magnesium aluminates			Magnezit	2	385, 481
$MgO \cdot Al_2O_3$	2	283			
$MgO \cdot 3.5Al_2O_3$	2	286			

Material Name	Vol.	Page	Material Name	Vol.	Page
Magnezit brick	2	899, 902	Marsh gas (see methane)		
Mahogany	2	1080	Marksa brick	2	899
Manganese	1	208	Medical cotton	2	1059, 1070
Manganese + Copper	1	683	Mercury	1	212
Manganese + Iron	1	684	Mercury - selenium intermetallic compound		
Manganese + Iron + $\Sigma X_1$	1	1009	HgSe	1	1320
Manganese + Nickel	1	685	Mercury + Sodium	1	686
Manganese + Silicon + $\Sigma X_1$	1	1012	Mercury - tellurium intermetallic compound		
Manganese alloys (specific types)			HgTe	1	1321
Ferromanganese, Russian	1	684, 1010	Mercury selenide [HgSe] (see mercury - selenium intermetallic compound)		
Silicomanganese, Russian	1	1010, 1012	Mercury telluride [HgTe] (see mercury - tellurium intermetallic compound)		
Manganese ferrate ( $MnFe_2O_4$ )	2	292	Metallurgical brick	2	892, 893
Manganese oxides			Metallurgical porous brick	2	893
MnO	2	168	Metallurgical pumice concrete	2	863, 864
$Mn_3O_4$	2	170	Methacrylate rubber	2	983
Manganese monoxide [MnO] (see manganese oxides)			Methane ( $CH_4$ )	3	218
(di)Manganese trioxide + Aluminum oxide	2	397	Methane - propane system	3	432
(di)Manganese trioxide + Magnesium oxide	2	398	Methanol [ $CH_3OH$ ] (see methyl alcohol)		
(di)Manganese trioxide + Silicon dioxide	2	399	Methanol - argon system	3	458
(tri)Manganese tetraoxide [ $Mn_3O_4$ ] (see manganese oxides)	2	170	Methanol - hexane system	3	460
Manganese zinc ferrate [ $Mn(Zn)Fe_2O_4$ ]	2	295	Methyl alcohol ( $CH_3OH$ )	3	223
Manganin	1	965	Methyl chloride ( $CH_3Cl$ )	3	227
Manganin NM Mts, Russian	1	965	Methyl formate - propane system	3	402
Manganomanganic oxide [ $Mn_3O_4$ ] (see (tri)manganese tetraoxide)			$Mg_2Ge$	1	1311
Maple	2	1051	$Mg_3Sb_2$	1	1310
Marbles			$Mg_2Si$	1	1314
Black	2	761	$Mg_2Sn$	1	1317
Brown	2	761	Mica	2	823, 892
Powder	2	760, 761	Canadian phlogopites	2	824, 825
White	2	761	Granulated vermiculite	2	825
White Alabama	2	761	Madagascan phlogopites	2	824

Material Name	Vol.	Page	Material Name	Vol.	Page
Mica (continued)			Monel alloy 505 (see "S" monel)		
Synthetic	2	825	Monel alloy 506 (see "H" monel)		
Mica, bonded	2	825	Monel alloy K-500 (see "K" monel)		
Micanite	2	1138	Monel alloy R-405 (see "R" monel)		
Mild steel	1	1186	Monolithic wall	2	1126
Mineral cotton (see mineral wool)			MoSi <sub>2</sub>	1	1324
Mineral fiber	2	1139	MSM-4Al-4Mn (see titanium alloy C-130 AM or titanium alloy RC-1308)		
Mineral wool	2	1147	MSM-6Al-4V (see Ti-6Al-4V)		
Mineral wood, processed	2	1140	MST-6Al-4V (see Ti-6Al-4V)		
Board	2	1141	MST-8Mn (see Ti-8Mn)		
Felt	2	1141	Mullite	2	254, 934
Mipora	2	944	Mullite + Alumina	2	335
Missouri firebrick	2	905	Multimet N-155	1	1165
Moderator graphite	2	70	Mystic slag	2	1150
Molybdenum	1	222	N. S. nickel	1	708
Molybdenum + Iron	1	690	Naphthalene (C <sub>10</sub> H <sub>8</sub> )	2	995
Molybdenum + Iron + Σ <sub>1</sub>	1	1013	Naphthalin [C <sub>10</sub> H <sub>8</sub> ] (see naphthalene)		
Molybdenum - silicon intermetallic compound			Naphthol (C <sub>10</sub> H <sub>7</sub> OH)	2	998
MoSi <sub>2</sub>	1	1324	Natrium (see sodium)		
Molybdenum + Thorium dioxide	1	1429	Natural Ceylon graphite	2	55
Molybdenum + Titanium	1	691	Navy M	1	977
Molybdenum + Tungsten	1	694	Nelson - Kebbenleg 10	1	896
Molybdenum alloy, ferromolybdenum, Russian	1	690, 1013	Neodymium	1	230
(di)Molybdenum carbide (Mo <sub>2</sub> C)	2	579	Neon	3	56
Molybdenum disilicide [MoSi <sub>2</sub> ] (see molybdenum - silicon intermetallic compound)			Neon - argon - deuterium system	3	490
			Neon - argon - hydrogen - nitrogen system	3	509
Monax glass	2	924	Neon - argon - krypton system	3	478
Monel	1	1032	Neon - argon - krypton - xenon system	3	504
Monel, cast	1	1032	Neon - carbon dioxide system	3	356
Monel, "H"	1	1032	Neon - deuterium system	3	360
Monel, "K"	1	1032	Neon - hydrogen system	3	362
Monel, "R"	1	1032	Neon - hydrogen - nitrogen system	3	494
Monel, "S"	1	1032	Neon - hydrogen - oxygen system	3	492
Monel alloy 400 (see monel)			Neon - krypton - deuterium system	3	491

Material Name	Vol.	Page	Material Name	Vol.	Page
Neon - nitrogen system	3	365	Nickel + Iron	1	707
Neon - nitrogen - oxygen system	3	495	Nickel + Iron + $\Sigma X_1$	1	1035
Neon - oxygen system	3	368	Nickel + Manganese	1	710
Neon - xenon system	3	291	Nickel + Manganese + $\Sigma X_1$	1	1038
Neptunium	1	234	Nickel + Molybdenum + $\Sigma X_1$	1	1041
80 Ni-20 Cr (see chromel A)			Nickel + $\Sigma X_1$	1	1044
Ni-Cr steel	1	1167, 1168, 1210, 1213	Nickel alloys (specific types)		
Nickrom (see chromel A)			"A" nickel	1	711
Nichrome	1	1018, 1019, 1021, 1036	Alumel	1	1015, 1039
Nichrome N	1	698	Chroman	1	1018
Nichrome V (see chromel A)			Chromel A	1	698
Nickel	1	237	Chromel C	1	1036
Nickel, "A"	1	239, 241, 1029, 1039	Chromel P	1	698
Nickel, "D"	1	1039	Contraacid	1	1036
Nickel, electrolytic	1	238, 239, 240	Contraacid B7M	1	1036
Nickel, "L"	1	238, 239	Corronil	1	1032
Nickel, "O"	1	239	"D" nickel	1	1039
Nickel, "Z" (see duranickel)			Duranickel	1	1015
Nickel 200 (see nickel, A)			EI-435, Russian	1	1022
Nickel 211 (see nickel, D)			EI-607, Russian	1	1019, 1020, 1021
Nickel + Aluminum + $\Sigma X_1$	1	1014	German chromin	1	1018
Nickel - antimony intermetallic compound			Grade A	1	711, 1044
NiSb	1	1027	H monel	1	1032
Nickel + Chromium	1	697	Hastelloy A	1	1036
Nickel + Chromium + $\Sigma X_1$	1	1017	Hastelloy B	1	1042
Nickel + Cobalt	1	700	Hastelloy C	1	1018
Nickel + Cobalt + $\Sigma X_1$	1	1028	Hastelloy R-235	1	1019
Nickel + Copper	1	703	Haynes steellite 27	1	1029
Nickel + Copper + $\Sigma X_1$	1	1031	HyMn-80	1	1030
			INCO "713 C"	1	1022
			Inconel	1	1016, 1019, 1021
			Inconel 702	1	1022



Material Name	Vol.	Page	Material Name	Vol.	Page
Nickel alloys (specific types) (continued)			Nickel alloys (specific types) (continued)		
Inconel alloy 713 (see Inco "713C")			Refralloy 26	1	1029
Inconel X	1	1018	Rene 41	1	1022
Inconel X-750 (see inconel X)			"S" monel	1	1032
INOR-8	1	1042	Silicon monel	1	1032
K monel	1	1032	"Z" nickel (see duranickel)		
Kh50T, Russian	1	1019	Nickel antimonide [NiSb] (see nickel - antimony intermetallic compound)		
"L" nickel	1	238, 239	Nickel bronze	1	1032
M 252	1	1022	Nickel oxide (NiO)	2	171
Monel	1	1032	Nickel silver	1	981
Monel, cast	1	1032	Nickel silver 12% (see german silver)		
Monel alloy 400 (see monel)			(tri)Nickel disulfide (Ni <sub>3</sub> S <sub>2</sub> )	2	705
Monel alloy 505 (see "S" monel)			Nickel zinc ferrate [Ni(Zn)Fe <sub>2</sub> O <sub>4</sub> ]	2	298
Monel alloy K-500 (see "K" monel)			Nicrosilal, British	1	1204
60Ni-20Cr	1	1019	Nigrine (see rutile)		
Nichrome	1	1018, 1019, 1021, 1036	Nil alba (see zinc oxide)		
Nichrome N	1	698	Nimocast 713 C	1	1022
Nickel bronze	1	1032	Nimonic 75	1	1019
Nimocast 713 C	1	1022	Nimonic 75, French	1	1019
Nimonic 75	1	1019	Nimonic 80	1	1018
Nimonic 75, French	1	1019	Nimonic 80/80 A, French	1	1019
Nimonic 80	1	1018, 1019	Nimonic 90	1	1019
Nimonic 80/80A, French	1	1019	Nimonic 95	1	1019
Nimonic 90	1	1019	Nimonic 100	1	1029
Nimonic 95	1	1019	Nimonic 105	1	1029
Nimonic 100	1	1029	Nimonic 115	1	1029
Nimonic 105	1	1029	Nimonic D6, French	1	1213
Nimonic 115	1	1029	Nimonic PE 7	1	1206
N. S. nickel	1	708	Niobium	1	245
"O" nickel	1	239	Niobium + Molybdenum + ΣX <sub>1</sub>	1	1046
OKh 20N 60B	1	1022	Niobium + Tantalum + ΣX <sub>1</sub>	1	1049
"R" monel	1	1032	Niobium + Titanium + ΣX <sub>1</sub>	1	1052
			Niobium + Tungsten + ΣX <sub>1</sub>	1	1055
			Niobium + Uranium	1	713

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Niobium + Zirconium	1	716	Nylon	2	945
Niobium alloys (specific types)			Nylon 6 (see polyhexahydro-2H-azepin-2-one)		
D-96 (see niobium alloy Nb-10W-5Zr)			"O" nickel	1	239
Haynes alloy Nb-752	1	1056	Oak	2	1082
Nb-5Mo-5V-1Zr	1	1047	White	2	1082
Nb-27Ta-12W-0.2Zr	1	1050	Octane, n-(C <sub>8</sub> H <sub>18</sub> )	3	233
Nb-10Ti-5Zr	1	1053	Ohmite graphite	2	73
Nb-15W-5Mo-1Zr-0.05C	1	1056	OKh20 N60 B, Russian	1	1022
Nb-10W-1Zr-0.1C	1	1056	Olivine (see forsterite)		
Nb-10W-5Zr	1	1056	Olivine basalt	2	798
Nb-0.5Zr	1	717	Ordzhonikidze brick	2	899
Niobium carbide (NbC)	2	582	Osmium	1	254
NiSb	1	1327	Oxygen	3	76
Niton (see radon)			Palladium	1	258
Nitric oxide (NO)	3	106	Palladium + Copper	1	720
Nitrile rubber	2	982	Palladium + Gold	1	723
Nitrogen	3	64	Palladium + Platinum	1	726
Nitrogen - oxygen system	3	434	Palladium + Silver	1	727
Nitrogen - oxygen - carbon dioxide system	3	497	Paper	2	1127
Nitrogen - propane system	3	438	Paraffin concrete	2	863
Nitrogen dioxide [NO <sub>2</sub> ] (see nitrogen peroxide)			PbTe	1	1307
Nitrogen peroxide (NO <sub>2</sub> )	3	108	Pearlitic matrix cast iron, Nr. 1520	1	1222
Nitrogen monoxide [N <sub>2</sub> O] (see nitrous oxide)			Pearlitic pig iron, Russian	1	1137
Nitrophenol (NO <sub>2</sub> C <sub>6</sub> H <sub>4</sub> OH)	2	1001	Pencil lead graphite	2	65
Nitrous oxide (N <sub>2</sub> O)	3	114	Penn. fire brick	2	905
Nivac	1	238	Pentane, n-(C <sub>5</sub> H <sub>12</sub> )	3	236
Nodular iron	1	1137, 1222	Periclase	2	160
Nonane, n-(C <sub>9</sub> H <sub>20</sub> )	3	230	Perlite	2	827
Normal brick	2	488, 489, 900, 901	Petalite	2	935
NTS basalt	2	798	Petroleum coke	2	765
NTS dolomite	2	811	Phenanthrene (C <sub>14</sub> H <sub>10</sub> )	2	1004
NTS granite	2	818	Phenanthrin [C <sub>14</sub> H <sub>10</sub> ] (see phenanthrene)		
			Phenyl ether [(C <sub>6</sub> H <sub>5</sub> ) <sub>2</sub> O] (see diphenyloxide)		
			Phoenix glass	2	924

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Phosphor bronze	1	585, 586, 976	Polyethylene	2	956
Phosphorus	2	86	Polyethylene, chlorosulfonated (see rubber, hypalon)		
Pig fat	2	1073	Polyhexahydro-2H-azepin-2-one, silon	2	959
Pines	2	1083	Poly(methyl methacrylate) [same as plexiglas]	2	960
Pitch	2	1083	AN-P-44A	2	961
White	2	1083	Perspex	2	961
Pitch pines	2	1083	Polystyrene	2	963
Pladuram	1	416	Colloidal aggregate	2	965
Plaster	2	887	Styrofoam	2	965
Plate glass	2	923, 924, 925, 926	Polysulfide rubber (see rubber, Thiokol)		
Platinoid	1	981	Polytetrafluoroethylene (same as Teflon)	2	967
Platinum	1	262	Polytrifluorochloroethylene	2	970
Platinum + Copper	1	730	Polyurethane (see rubber, Adiprene)		
Platinum + Gold	1	733	Polyvinyl chloride	2	953
Platinum + Iridium	1	734	Porcelains	2	936
Platinum + Palladium	1	737	Alumina	2	937
Platinum + Rhodium	1	738	Electrical	2	937
Platinum + Ruthenium	1	743	High zircon	2	937
Platinum + Silver	1	745	MgTiO <sub>3</sub> porcelain	2	937
Plexiglas	2	960	Porcelain 576	2	937
Plexiglas AN-P-44A	2	961	Wet process	2	937
Polfoam	2	950	Porous brick	2	894
Pluton cloth	2	1100	Porous concrete brick	2	894
Plutonium	1	270	Porous fire brick (Italy)	2	895
Plutonium, $\alpha$ -	1	271	Portland cement	2	861
Plutonium + Aluminum	1	746	Portland cement concrete	2	871
Plutonium + Iron	1	747	Potassium	1	274
Plutonium alloy, delta-stabilized	1	746	Potassium + Sodium	1	748
Polychloroethylene (polyvinyl chloride)	2	953	Potassium acid phosphate [KH <sub>2</sub> PO <sub>4</sub> ] (see potassium dihydrogen phosphate)		
Polychloroethylene (polyvinyl chloride), plasticized	2	954	Potassium bromide (KBr)	2	566
Polychlorotrifluoroethylene (see polytrifluorochloroethylene)			Potassium bromide + Potassium chloride	2	779
			Potassium chloride (KCl)	2	613
			Potassium chloride + Potassium bromide	2	782

Material Name	Vol.	Page	Material Name	Vol.	Page
Potassium chrome alum salt	2	689	Quartz fiber	2	1143
Potassium chromium sulfate [ $KCr(SO_4)_2 \cdot 12H_2O$ ]	2	688	Dyna	2	1144
Potassium <del>dideuterium</del> <sup>dideuterium</sup> phosphate ( $KD_2PO_4$ )	2	680	Quartz glass	2	187, 188, 923, 924
Potassium dihydrogen arsenate ( $KH_2AsO_4$ )	2	785	Quartz sand	2	834, 835, 836, 837
Potassium dihydrogen phosphate ( $KH_2PO_4$ )	2	684	Queenstone grey limestone	2	821
Potassium hydrogen sulfate ( $KHSO_4$ )	2	691	Quick silver (see mercury)		
Potassium nitrate ( $KNO_3$ )	2	647	"R" monel	1	1032
Potassium phosphate, monobasic [ $KH_2PO_4$ ] (see potassium dihydrogen phosphate)			Radon	3	84
Potassium biphosphate [ $KH_2PO_4$ ] (see potassium dihydrogen phosphate)			Rama limestone	2	821
Potassium diphosphate [ $KH_2PO_4$ ] (see potassium dihydrogen phosphate)			RCA N91	1	701
Potassium rhodanide [ $KSCN$ ] (see potassium thiocyanate)			RCA N97	1	701
Potassium sulfocyanate [ $KSCN$ ] (see potassium thiocyanate)			$Re_3As_7$	1	1330
Potassium sulfocyanide [ $KSCN$ ] (see potassium thiocyanide)			Red brass	1	591
Potassium thiocyanate ( $KSCN$ )	2	788	Red brass, German	1	981
Powders (nonmetallic)	2	1040	Red brick	2	405, 492, 898
Praseodymium	1	281	Red brick, hard burned	2	896
Promethium	1	285	Red brick, soft burned	2	896
Propane ( $C_3H_8$ )	3	240	Redwood	2	1084
2-Propanone [ $(CH_3)_2CO$ ] (see acetone)			Bark	2	1084
Pseudo balsa	2	1060	Red wood fiber	2	1091
Pyrex	2	499, 923, 924, 926, 927	Refractory insulating brick	2	892
Pyrex 7740	2	499, 923, 924, 925, 926	Refractory insulating common chamotte brick	2	892
Pyroacetic acid (see acetone)			Refralloy 26	1	1029
Pyroceram 9606	2	940	Refrax	2	586
Pyroceram brand glass-ceramic	2	939	ReGe	1	1331
Pyrolytic graphite	2	30	ReGe <sub>2</sub>	1	1331
Quartz [see silicon dioxide (crystalline)]			Rene 41	1	1022
			Rene 41 cloth	2	1102
			ReSe <sub>2</sub>	1	1332
			Rex 78	1	1213

Material Name	Vol.	Page	Material Name	Vol.	Page
Rhenium	1	288	Rubbers (continued)		
Rhenium - arsenic intermetallic compound			Nitrile	2	982
$Re_3As_7$	1	1330	Poly(ethyl acrylate)	2	983
Rhenium - germanium intermetallic compounds			Polysulfide (see rubber, Thiokol)		
ReGe	1	1331	Resin-cured butyl	2	983
$ReGe_2$	1	1331	Rubutex	2	981
Rhenium - selenium intermetallic compound			Rubutex R203-H (same as Buna-N foam)	2	981
$ReSe_2$	1	1332	Silicone	2	983
<del>Rhenium</del> selenide [ $ReSe_2$ ] (see rhenium selenium intermetallic compound)			Tellurace-cured butyl	2	983
Rhodium	1	292	Thiokol ST	2	982
Rock	2	828	Viton	2	983
Rock cork	2	1146	X-ray protective	2	981
Rock wool	2	1148	Rubidium	1	296
Rose metal	1	939	Rubidium + Cesium	1	751
Rubutex rubber	2	981	Russian alloy	1	1192, 1218, 1222
Rubutex R203-H rubber	2	981	Russian cupralloy, type 5	1	543
Rubbers	2	980	Russian cupro nickel, NM-81	1	562
Acrylate	2	982	Russian stainless steel (see stainless steel)		
Acrylic	2	982	Russian steel	1	1118
Adiprene	2	982	Rutgers cordierite	2	919
Buna-N foam (see rubber, Rubutex R203-H)			Ruthenium	1	300
Butaprene E	2	982	Rutile	2	203
Carboxy nitrile	2	982	"S" monel	1	1032
Chloroprene	2	983	SAE 1010	1	1183
Dibenzo GMF-cured butyl	2	983	SAE 1015 (see AISI C 1015)		
Ebonite	2	971	SAE 1020	1	1183
Elastomer	2	974	SAE 1095	1	1114
Government rubber-styrene	2	977	SAE 4130	1	1153
Hard	2	972, 981	SAE 4140	1	1155
Hevea	2	983	SAE 4340 (see AISI 4340)		
Hypalon S2	2	983	SAE bearing alloy 10	1	1070
Kel-F 3700	2	983	SAE bearing alloy 11	1	1070
Methacrylate	2	983	SAE bearing alloy 12	1	991
			SAE bearing alloy 40	1	976

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SAE bearing alloy 62	1	976	$Sb_{1.4}Bi_{0.6}Te_{3.13}$	1	1381
SAE bearing alloy 64	1	976	$Sb_{1.4}Bi_{0.6}Te_{3.19}$	1	1383
SAE bearing alloy 66	1	962	$Sb_{1.4}Bi_{0.6}Te_{3.26}$	1	1384
Salt, gnome	2	832	$Sb_{1.6}Bi_{0.5}Te_3$	1	1381
Samarium	1	305	$Sb_{1.5}Bi_{0.6}Te_{3.06}$	1	1384
Sand	2	833	$Sb_{1.5}Bi_{0.6}Te_{3.13}$	1	1382
Lowell	2	834, 835	$Sb_{1.5}Bi_{0.5}Te_{3.19}$	1	1384
Quartz	2	834, 835, 836, 837	$Sb_{1.5}Bi_{0.5}Te_{3.26}$	1	1384
Silica	2	441, 837	$Sb_{1.6}Bi_{0.4}Te_3$	1	1381
Sand cement concrete	2	874	$Sb_{1.6}Bi_{0.4}Te_{3.19}$	1	1384
Sand and gravel aggregate concrete	2	868, 869	$Sb_{1.6}Bi_{0.4}Te_{3.26}$	1	1384
Sandstone	2	840	$Sb_{1.7}Bi_{0.3}Te_3$	1	1381
Berea	2	841, 842	$Sb_{1.8}Bi_{0.2}Te_3$	1	1381
Berkeley	2	841, 842	$Sb_2Se_3 + Ag_2Se + PbSe$	1	1379
St. Peters	2	841	$Sb_2Te_3$	1	1241
Teapot	2	842	$Sb_2Te_3 + Bi_2Te_3$	1	1380
Tensleep	2	841, 842	$Sb_2Te_3 + In_2Te_3$	1	1386
Tripolite	2	842	Scandium	1	309
Sandwiches (nonmetallic)	2	1044	Scotchply laminate (nonmetallic)	2	1029
Sandwiches (metallic - nonmetallic)	2	1047	Sea-weed product	2	1128
Sandy clay	2	805	Selenium	1	313
Santowax R	2	1005	Selenium + Bromine	1	754
Sapphire	2	93	Selenium + Cadmium	1	755
Sapphire, synthetic	2	95	Selenium + Chlorine	1	756
Sapphire, Linde synthetic	2	94	Selenium + Iodine	1	757
Eatin walnut	2	1089	Selenium + Thallium	1	758
Sawdust	2	1085	Shamotte brick	2	894, 898
$Sb_{1.2}Bi_{0.8}Ti_{3.11}$	1	1381	Sheep wool	2	1092
$Sb_{1.13}Bi_{0.87}Te_{3.11}$	1	1381	Silat iron	1	1222, 1223
$Sb_{1.4}Bi_{0.6}Te_{3.08}$	1	1383	Silica (see silicon dioxide)		

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Silica brick	2	408,	Silicon dioxide (SiO <sub>2</sub> )					
		489,						
		492,				Crystalline	2	174
		502,				Domestic (USA)	2	175
		894,				Foamed fused silica	2	184
		896,				Fused	2	183
		897,				Linde silica	2	184
		898,				Slip 10	2	189
		900,				Slip 1s	2	188
		902,				Quartz glass	2	187,
		904,					2	188
906								
Silica fire brick	2	894,	Silica gel	2	185			
		895,	Silica refractory brick	2	185			
		905	Slip cast fused silica	2	184			
Silica glass	2	923,	Star-brand brick	2	185			
		925,	Vitreous	2	184,			
		926		2	185,			
Silica glass, fused	2	925		2	187			
Silica sand	2	837	Silicon dioxide + Aluminum oxide	2	402			
Silicate glass	2	511	Silicon dioxide + Aluminum oxide + ΣX <sub>1</sub>	2	487			
Silicious brick	2	492,	Silicon dioxide + Barium oxide + ΣX <sub>1</sub>	2	495			
		902	Silicon dioxide + Boron oxide + ΣX <sub>1</sub>	2	498			
Silicomanganese, Russian	1	1010,	Silicon dioxide + Calcium oxide	2	407			
		1012	Silicon dioxide + Calcium oxide + ΣX <sub>1</sub>	2	501			
Silicon	1	326	Silicon dioxide + (di)Iron trioxide	2	410			
Silicon + Germanium	1	761	Silicon dioxide + Lead oxide + ΣX <sub>1</sub>	2	504			
Silicon + Iron	1	764	Silicon dioxide + (di)Potassium oxide + ΣX <sub>1</sub>	2	507			
Silicon alloy, ferrosilicon, Russian	1	765	Silicon dioxide + (di)Sodium oxide + ΣX <sub>1</sub>	2	510			
Silicon bronze	1	973	Silicone rubber	2	983			
Silicon carbide (SiC)	2	585	Silk fabric	2	1105			
Crystolon SiC	2	586	Sillimanite	2	454,			
SiC brick, refrax	2	586		2	845			
Silicon carbide, refractory (see refrax)			Sillimanite brick	2	902			
Silicon carbide + Graphite	2	789	Sillimanite refractory brick	2	902,			
Silicon carbide - silicon cermets	2	718		2	903			
Silicon carbide + Silicon dioxide	2	553	Sil-O-Cel brick	2	896			
Silicon carbide + Silicon dioxide + ΣX <sub>1</sub>	2	554	Sil-O-Cel brick, calcined	2	896			
Silicon carbide brick	2	895	Sil-O-Cel brick, natural	2	896			
Silicon carbide brick, refrax	2	586,						
		906						
Silicon enamel	2	921						
Silicon metal	1	1032						
(tri)Silicon tetranitride (Si <sub>3</sub> N <sub>4</sub> )	2	662						

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Sil-O-Cel brick, special	2	896	Silver chloride (AgCl)	2	620
Sil-O-Cel brick, super	2	896	Silver iodide (AgI)	2	563
Sil-O-Cel coarse grade diatomite aggregate	2	1112	Silver nitrate (AgNO <sub>3</sub> )	2	650
Silon	2	959	Silver selenide [Ag <sub>2</sub> Se] (see silver - selenium intermetallic compound)		
Silumin, sodium modified	2	920	Silver solder, Easy-Flo	1	1059
γ-Silumin, modified	1	920	Silver steel	1	1114
Silver	1	340	Silver telluride [Ag <sub>2</sub> Te] (see silver - tellurium intermetallic compound)		
Silver + Antimony	1	767	Slag aggregate concrete, limestone treated	2	870
Silver - antimony - tellurium intermetallic compound			Slag brick	2	898
AgSbTe <sub>2</sub>	1	1335	Slag cement	2	861
Silver + Cadmium	1	770	Slag concrete	2	864, 880, 881
Silver + Cadmium + ΣX <sub>1</sub>	1	1058	Slag concrete, direct process	2	864
Silver + Copper	1	773	Slag concrete, expanded	2	878, 879
Silver - copper intermetallic compound			Slag concrete, Leuna	2	864
AgCu	1	1338	Slag-Portland cement	2	861
Silver + Gold	1	774	Slag wool (same as mineral wool)	2	1151
Silver + Indium	1	777	Slate	2	846
Silver + Lead	1	780	SnSe <sub>2</sub>	1	1352
Silver + Manganese	1	783	SnTe	1	1355
Silver + Palladium	1	786	SnTe + AgSbTe <sub>2</sub>	1	1411
Silver + Platinum	1	790	Soapstone	2	853
Silver - selenium intermetallic compound			Soda glass	2	923
Ag <sub>2</sub> Se	1	1339	Soda-lime glass	2	926
Silver - tellurium intermetallic compounds			Soda-lime plate glass	2	926
Ag <sub>2-x</sub> Te	1	1342	Soda-lime silica glass	2	511, 924, 927
Ag <sub>2</sub> Te	1	1342	Soda-lime silica plate glass, 9330	2	923
Silver + Tin	1	791	Sodium	1	349
Silver + Zinc	1	792	Sodium + Mercury	1	795
Silver + ΣX <sub>1</sub>	1	1061	Sodium + Potassium	1	798
Silver alloy, silver solder, Easy-Flo	1	1059	Sodium + (di)Sodium oxide	1	1432
Silver antimony telluride [AgSbTe <sub>2</sub> ] (see silver - antimony - tellurium intermetallic compound)			Sodium acetate (NaC <sub>2</sub> H <sub>3</sub> O <sub>2</sub> · 3H <sub>2</sub> O)	2	1006
Silver bromide (AgBr)	2	569			
Silver bronze	1	579, 980			



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Sodium chloride (NaCl)	2	621	Stainless steels (specific types)		
Sodium fluoride (NaF)	2	642	1 Kh 18 N9T (Russian)	1	1168
Sodium fluoride + Beryllium difluoride	2	645	15 Kh 12 VMF, Russian (see steel EI 502, Russian)		
Sodium fluoride + Zirconium tetrafluoride + $\Sigma X_1$	2	646	17-4 PH	1	1168
Sodium hydrate [NaOH] (see sodium hydroxide)			17-7	1	1165
Sodium hydrogen sulfate (NaHSO <sub>4</sub> )	2	692	17-7 PH	1	1166
Sodium hydroxide (NaOH)	2	790	18-8	1	1161, 1162, 1167, 1168
Sodium nitrate (NaNO <sub>3</sub> )	2	651	416	1	1168
(d) Sodium oxide - sodium cermets	2	721	3754	1	1161
Sodium hyposulfite [Na <sub>2</sub> S <sub>2</sub> O <sub>3</sub> · 5H <sub>2</sub> O] (see sodium thiosulfate)			AISI 301	1	1165
Sodium thiosulfate (Na <sub>2</sub> S <sub>2</sub> O <sub>3</sub> · 5H <sub>2</sub> O)	2	693	AISI 302	1	1161
Sodium tungsten bronze (Na <sub>2</sub> WO <sub>3</sub> )	2	301	AISI 303	1	1165, 1168
Sodium tungsten oxide [Na <sub>2</sub> WO <sub>3</sub> ] (see sodium tungsten bronze)			AISI 304	1	1161, 1165, 1168
Soft cast iron, gray	1	1135	AISI 310	1	1168
Soft glass	2	511	AISI 316	1	1165, 1166, 1169, 1170
Soft steel	1	1126	AISI 347	1	1165, 1166, 1168
Soil	2	847			
Solder, soft	1	840	AISI 403	1	1149
Solex 2805 plate glass	2	923	AISI 410	1	1150
Solex 2808 X glass	2	925	AISI 420	1	1162
Solex "S" glass	2	925	AISI 430	1	1150, 1154
Solex "S" plate glass	2	923	AISI 440 C	1	1154
Spektral Kohle 1	2	54	AISI 446	1	1149, 1150, 1155, 1156
Spherical cast iron, Nr 1510	1	1222	AM 355 (Russian)	1	1168
Spinel	2	284, 369, 848	AS 21	1	1161
Spinel, natural ruby	2	284	Austenitic	1	1165, 1183
Spinel firebrick	2	905	Crucible HNM	1	1168
Spodumene	2	851	EI 572, Russian (same as stainless steel 18-8)	1	1168
Spruce	2	1086			
Sr <sub>2</sub> Si	1	1343			
Sr <sub>2</sub> Sn	1	1344			

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EY a 1 T (see stainless steel 1 Kh 18 N9 T)			AISI C 1010 (see steel SAE 1010)		
F. H. (British)	1	1161	AISI C 1015 (same as steel SAE 1015)	1	1186
Russian	1	1150, 1161	AISI C 1020 (same as steel SAE 1020)		
SF 11, British (see stainless steel AISI 403)			Alloy steel	1	1214
Staybrite	1	1161	Alloy steel, high	1	1214
Stannic anhydride [SnO <sub>2</sub> ] (see tin dioxide)			AMS 2713	1	1210
Stannic selenide [SnSe <sub>2</sub> ] (see tin - selenium intermetallic compound)			AMS 2714	1	1213
Stannous telluride [SnTe] (see tin - tellurium intermetallic compound)			Haynes alloy N-155	1	1177
Stannum (see tin)			High carbon, Japanese	1	1119
Staybrite steel, British	1	1161	High-perm-49	1	1199
Steam - air system	3	464	High speed	1	1230, 1231, 1232, 1234
steam - carbon dioxide system	3	466	High speed, 18	1	1233
steam - nitrogen system	3	468	High speed, 18-4-1	1	1233
Steam bronze (see navy M)			High speed, M1	1	1195
Steatite	2	852	High speed, M2	1	1233
10 B 2	2	853	High speed, M10	1	1195
12 C 2	2	853	High speed, T1	1	1233
228	2	853	Invar	1	1199
Soapstone	2	853	Invar, free cut	1	1205
Steatite cordierite	2	919	Japanese	1	1195, 1210
Steels (specific types)			Jessop G 17, British	1	1213
1 Kh 14 N 14 V2M (see steel EI 257)			Kh Zn (Russian)	1	1210
5 ZA 2, Russian	1	1213	Kovar	1	1203
12 MKH, Russian	1	1192	Krupp	1	1115, 1184
AISI 1010	1	1185	K. S. magnet	1	1177
AISI 1095 (see steel SAE 1095)			Low alloy	1	1213
AISI 2515	1	1198, 1199, 1200	Low-exp-42	1	1205
AISI 4130 (see steel SAE 4130)			Low Mn	1	1183
AISI 4140 (see steel SAE 4140)			Macloy G	1	1213
AISI 4340	1	1213, 1214	Mild steel	1	1186
			Ni-Cr steel	1	1167, 1168

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Ni-Spaa C	1	1214	EI-257 (Russian)	1	1166, 1214
Nichrom	1	1210, 1213	EI-606 (Russian)	1	1168
Nicrosilal, British	1	1204	EI-802 (Russian)	1	1156
Nimonic DS, French	1	1213	EI-855 (Russian)	1	1214
Nimonic PE7	1	1206	En8 (CMK), British	1	1184, 1186
Oil-hardening non-deforming	1	1125	En 19 (British)	1	1153
R7 (Russian)	1	1236	En 31 (British)	1	1153, 1154
R10 (Russian)	1	1236	En 32 A (BGKI), British	1	1192
R12 (Russian)	1	1236	Era ATV (British)	1	1213
R15 (Russian)	1	1235	EYA-2	1	1166
R18 (Russian)	1	1236	Ferrosilcon 45%, Russian	1	1218
R15 Kh 3 (Russian)	1	1235	Ferrotitanum, Russian	1	1225
R15 Kh 3 K 5 (Russian)	1	1235	Fish-plate, Japanese	1	1119
R15 Kh 3 K 10 (Russian)	1	1235	FNCT	1	1213
R15 Kh 3 K 12 (Russian)	1	1235, 1236	G 18B, British	1	1165, 1213
R15 Kh 4 (Russian)	1	1236	German	1	1118
R20, British	1	1165	H. 20, British	1	1154
Rex 75	1	1213	H. 27, British	1	1154
Russian	1	1118, 1166	H. 46, British	1	1154
Russian alloy	1	1192, 1218, 1222	SAE 1020	1	1183
SAE 1010	1	1183	SAE 1095	1	1114
SAE 1015 (see steel AISI C 1015)			SAE 4130	1	1153
British	1	1114, 1118, 1187	SAE 4140	1	1155
Carbon	1	1118, 1119, 1126, 1150, 1185	SAE 4340 (see steel AISI 4340)		
Carbon, British	1	1186	Silver steel	1	1114
Carbon, Japanese	1	1185	Soft	1	1126
Chromel 502	1	1210	St 42. 11 (German)	1	1186, 1218
Climax	1	1198, 1213	Stainless steels (see separate entries under stainless steels)		
			Tool steel	1	1115

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Vacromin F	1	1213	Ta <sub>2</sub> Be <sub>12</sub> (see beryllium - tantalum interm. comp.)		
WF 100 (Russian)	1	1166	TaGe <sub>2</sub>	1	1348
Stibium (see antimony)			Tantalum	1	355
Strontia (see strontium oxide)			Tantalum - boron intermetallic compound		
Strontium - silicon intermetallic compound			TaB <sub>2</sub>	1	1345
Sr <sub>2</sub> Si	1	1343	Tantalum - germanium intermetallic compound		
Strontium - tin intermetallic compound			TaGe <sub>2</sub>	1	1348
Sr <sub>2</sub> Sn	1	1344	Tantalum + Niobium	1	891
Strontium difluoride + ΣX <sub>1</sub>	2	791	Tantalum + Niobium + ΣX <sub>1</sub>	1	1062
Strontium oxide (SrO)	2	194	Tantalum + Tungsten	1	802
Strontium oxide + Lithium aluminate + ΣX <sub>1</sub>	2	513	Tantalum + Tungsten + ΣX <sub>1</sub>	1	1065
Strontium oxide + Lithium zirconium silicate + ΣX <sub>1</sub>	2	514	Tantalum alloys (specific types)		
Strontium oxide + Titanium dioxide + ΣX <sub>1</sub>	2	517	T 222	1	1066
Strontium oxide + Zinc oxide + ΣX <sub>1</sub>	2	520	Ta-30Nb-7, 5V	1	1063
Strontium silicide (Sr <sub>2</sub> Si <sup>2</sup> ) (see strontium - silicon intermetallic compound)			Ta-8W-2Hf	1	1066
Strontium stannide (Sr <sub>2</sub> Sn <sup>2</sup> ) (see strontium - tin intermetallic compound)			Tantalum boride (TaB <sub>2</sub> ) (see tantalum - boron intermetallic compound)		
Strontium metatitanate (SrTiO <sub>3</sub> )	2	304	Tantalum carbide (TaC)	2	589
Strontium metatitanate - cobalt cermets	2	722	Tantalum nitride (Ta <sub>3</sub> N <sub>5</sub> )	2	665
Strontium zirconate (SrZrO <sub>3</sub> )	2	307	Teak	2	1087
Styrofoam polystyrene	2	965	Technetium	1	363
Sulfothionine [Na <sub>2</sub> S <sub>2</sub> O <sub>3</sub> · 5H <sub>2</sub> O] (see sodium thiosulfate)			Teflon	2	967
Sulfur	2	89	Teflon, Duronid 5000	2	968
Sulfur dioxide (SO <sub>2</sub> )	3	116	Tellurium	1	366
Sulfurous acid anhydride (SO <sub>2</sub> ) (see sulfur dioxide)			Tellurium + Arsenic + ΣX <sub>1</sub>	1	1068
Supertemp pyrolytic graphite	2	72	Tellurium + Selenium	1	805
Swedish iron	1	158	Tellurium + Thallium	1	808
Systems, miscellaneous (metallic - non-metallic)	2	1055	Terbium	1	372
Systems, miscellaneous (nonmetallic)	2	1051	Thallium	1	376
Ta-30Nb-7, 5V	1	1063	Thallium + Cadmium	1	811
Ta-8W-2Hf	1	1066	Thallium + Indium	1	812
TaB <sub>2</sub>	1	1345	Thallium + Lead	1	815

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Thallium - lead intermetallic compound			TiB <sub>2</sub>	1	1358
Tl <sub>2</sub> Pb	1	1349	Tin	1	389
Thallium + Tellurium	1	815	Tin + Aluminum	1	823
Thallium + Tin	1	821	Tin + Antimony	1	824
Thallium bromide (TlBr)	2	570	Tin + Antimony + $\Sigma X_1$	1	1069
Thallium carbide (TlC)	2	625	Tin + Bismuth	1	827
Thiokol SF rubber	2	982	Tin + Cadmium	1	830
Thoria (see thorium dioxide)			Tin + Cepper	1	833
Thorium	1	381	Tin + Copper + $\Sigma X_1$	1	1072
Thorium + Uranium	1	822	Tin + Indium	1	834
Thorium carbides			Tin + Lead	1	839
ThC	2	592	Tin + Mercury	1	842
ThC <sub>2</sub>	2	593	Tin - selenium intermetallic compound		
Thorium dioxide (ThO <sub>2</sub> )	2	195	SnSe <sub>2</sub>	1	1352
Thorium dioxide + Graphite	2	557	Tin + Silver	1	845
Thorium dioxide + Uranium dioxide	2	413	Tin - tellurium intermetallic compound		
Thoron (see radon)			SnTe	1	1355
Thulium	1	355	Tin + Thallium	1	846
Thuringian glass	2	923, 924	Tin + Zinc	1	847
Ti-130 A	1	850	Tin alloys (specific types)		
Ti-140 A	1	1081	SAE bearing alloy 10	1	1070
Ti-150 A	1	1078, 1059	SAE bearing alloy 11	1	1070
Ti-155 A	1	1074	Soft solder	1	840
Ti-2.5 Al-16V	1	1087	White bearing metal	1	1070
Ti-3Al-11Cr-13V	1	1087	Tin anhydride [SnO <sub>2</sub> ] (see tin dioxide)		
Ti-4Al-4Mn (see titanium alloy C-130 AM, or titanium alloy RC-1308)			Tin ash [SnO <sub>2</sub> ] (see tin dioxide)		
Ti-4Al-3Mo-1V	1	1074, 1075	Tin dioxide (SnO <sub>2</sub> )	2	199
Ti-5Al-1.4Cr-1.5Fe-1.2Mo (see Ti-155 A)			Tin dioxide + Magnesium oxide	2	416
Ti-5Al-2.5Sn (see titanium alloy A-110 AT)			Tin dioxide + Magnesium oxide + $\Sigma X_1$	2	523
Ti-6Al-4V	1	1074	Tin dioxide + Zinc oxide	2	419
Ti-2Cr-2Fe-2Mo (see Ti-140 A)			Tin dioxide + Zinc oxide + $\Sigma X_1$	2	524
Ti-5Mn	1	850	Tin peroxide [SnO <sub>2</sub> ] (see tin dioxide)		
Ti-13V-11Cr-3Al	1	1087	TiNi	1	1361
			TiNi + Cu	1	1433
			TiNi + Ni	1	1436

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Titanic acid anhydride [TiO <sub>2</sub> ] (see titanium dioxide)			C-130 AM	1	1074
Titanic anhydride [TiO <sub>2</sub> ] (see titanium dioxide)			C-110 M (see Ti-8Mn)		
Titanic oxide [TiO <sub>2</sub> ] (see titanium dioxide)			MSM-4Al-4Mn (see titanium alloy C-130 AM, or titanium alloy RC-130S)		
Titanium	1	410	MSM-6Al-4V (see titanium alloy Ti-6Al-4V)		
Titanium, iodide	1	411	MST-6Al-4V (see titanium alloy Ti-6Al-4V)		
Titanium + Aluminum	1	848	MST-8Mn (see titanium alloy Ti-8Mn)		
Titanium + Aluminum + ΣX <sub>1</sub>	1	1073	RC-130S	1	1084
Titanium - boron intermetallic compound			Ti-130 A	1	850
TiB <sub>2</sub>	1	1358	Ti-140 A	1	1081
Titanium + Chromium + ΣX <sub>1</sub>	1	1077	Ti-150 A	1	1078, 1089
Titanium + Iron + ΣX <sub>1</sub>	1	1080	Ti-155 A	1	1074
Titanium + Manganese	1	849	Ti-2.5Al-10V	1	1081
Titanium + Manganese + ΣX <sub>1</sub>	1	1083	Ti-3Al-11Cr-13V	1	1087
Titanium - nickel intermetallic compound			Ti-4Al-4Mn (see titanium alloy C-130 AM, or titanium alloy RC-130S)		
TiNi	1	1361	Ti-4Al-3Mo-1V	1	1074, 1075
Titanium + Oxygen	1	852	Ti-5Al-1.4Cr-1.5Fe-1.2Mo (see titanium alloy Ti-155 A)		
Titanium + Vanadium + ΣX <sub>1</sub>	1	1086	Ti-5Al-2.5Sn (see titanium alloy A-110 AT)		
Titanium + ΣX <sub>1</sub>	1	1089	Ti-6Al-4V	1	1074
Titanium alloys (specific types)			Ti-2Cr-2Fe-2Mo (see titanium alloy Ti-140 A)		
120 VCA	1	1087	Ti-8Mn	1	850
A-110 AT	1	1074	Ti-13V-11Cr-3Al	1	1087
AMS 4908 (see titanium alloys Ti-8Mn)			Titanium boride [TiB <sub>2</sub> ] (see titanium - boron intermetallic compound)		
AMS 4925 A (see titanium alloys C-130 AM, or titanium alloys RC-130S)			Titanium carbide (TiC)	2	594
AMS 4926 (see titanium alloys A-110 AT)			Titanium carbide - cobalt cermets	2	725
AMS 4928 (see titanium alloys Ti-6Al-4V)			Titanium carbide - cobalt - niobium carbide cermets	2	726
AMS 4929 (see titanium alloys Ti-155 A)			Titanium carbide - nickel - molybdenum - niobium carbide cermets	2	727
AMS 4969 (see titanium alloys Ti-155 A)			Titanium carbide - nickel - niobium carbide cermets	2	730
ASTM B 265-58 T, grade 6 (see titanium alloy A-110 AT)					
ASTM 265-58 T, grade 7 (see titanium alloy Ti-8Mn)					

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Titanium dioxide (TiO <sub>2</sub> )	2	202	WSe <sub>2</sub>	1	1368
Dense titania	2	204	Tungsten - silicon intermetallic compound		
Rutile	2	203	WSi <sub>2</sub>	1	1369
Ti <sub>2</sub> (Pb)	1	1349	Tungsten - tellurium intermetallic compound		
Toluene (C <sub>6</sub> H <sub>5</sub> CH <sub>3</sub> )	3	242	WTe <sub>2</sub>	1	1370
Tool steel	1	1115, 1233	Tungsten + Thorium dioxide	1	1439
Tool steel, M1 high-speed	1	1195	Tungsten alloy, ferrotungsten (Russian)	1	1090
Tool steel, M10 high-speed	1	1195	Tungsten boride [WB] (see tungsten - boron intermetallic compound)		
Topaz	2	251	Tungsten carbide (WC)	2	598
Tourmaline	2	855	Tungsten trioxide (WO <sub>3</sub> )	2	209
Tourmaline, Brazil	2	855	Tungsten trioxide + Zinc oxide	2	422
Transite	2	1107	Tungsten diselenide [WSe <sub>2</sub> ] (see tungsten - selenium intermetallic compound)		
Triangle beryllia	2	126	Tungsten disulfide [WSi <sub>2</sub> ] (see tungsten - silicon intermetallic compound)		
Trichlorofluoromethane [Cl <sub>2</sub> CF] (see Freon 11)			Tungsten ditelluride [WTe <sub>2</sub> ] (see tungsten - tellurium intermetallic compound)		
Trichloromethane [CHCl <sub>3</sub> ] (see chloroform)			Tungstic acid anhydride [WO <sub>3</sub> ] (see tungsten trioxide)		
Trichlorotrifluoroethane [CCl <sub>2</sub> FCF <sub>2</sub> ] (see Freon 113)			Tungstic anhydride [WO <sub>3</sub> ] (see tungsten trioxide)		
Trifluoroborane [BF <sub>3</sub> ] (see boron trifluoride)			Tungstic oxide [WO <sub>3</sub> ] (see tungsten trioxide)		
Trifluorotrichloroethane [CCl <sub>2</sub> FCF <sub>2</sub> ] (see Freon 113)			UBe <sub>13</sub> (see beryllium - uranium intermetallic compound)		
Trinitrotoluene [CH <sub>2</sub> C <sub>6</sub> H <sub>2</sub> (NO <sub>2</sub> ) <sub>3</sub> ]	2	1007	Uranic oxide [UO <sub>2</sub> ] (see uranium dioxide)		
Tripolite brick	2	894	Uranium	1	429
Tritium	3	87	Uranium + Aluminum	1	858
Tuballoy (same as uranium)	1	429	Uranium + Chromium	1	859
Tuff	2	856	Uranium + Iron	1	862
Tungsten	1	415	Uranium + Magnesium	1	863
Tungsten - arsenic intermetallic compound			Uranium + Molybdenum	1	864
W <sub>3</sub> As <sub>7</sub>	1	1364	Uranium + Molybdenum + ΣX <sub>1</sub>	1	1094
Tungsten - boron intermetallic compound			Uranium + Niobium	1	867
WB	1	1365	Uranium + Silicon	1	868
Tungsten + Iron + ΣX <sub>1</sub>	1	1090	Uranium + Uranium dioxide	1	1442
Tungsten + Nickel + ΣX <sub>1</sub>	1	1091	Uranium + Zirconium	1	871
Tungsten + Rhenium	1	855			

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Uranium + Zirconium + $\Sigma X_1$	1	1097	Vermiculite brick	2	894
Uranium carbides			Vermiculite mica, granulated	2	825
UC	2	601	Vitalium type alloy (see Haynes stellite alloy 21)		
UC <sub>2</sub>	2	605	Viton rubber	2	953
Uranium carbide - uranium cermets	2	731	Vitreous silica	2	184, 185, 187
Uranium - 3% fissium alloy	1	1095	Volcanic ash (see tuff)		
Uranium - 5% fissium alloy	1	1095, 1097	Vulcanized fiber	2	1088
Uranium - 5% fissium alloy	1	1095	Vycor-brand glass	2	926
Uranium - 10% fissium alloy	1	1095	W-2 enamelloy (see molybdenum - silicon intermetallic compound)		
Uranium nitride (UN)	2	672	Wallboard	2	1131
Uranium oxides			Walnut	2	1089
UO <sub>2</sub>	2	210	W <sub>3</sub> As <sub>7</sub>	1	1364
U <sub>3</sub> O <sub>8</sub>	2	237	Water (H <sub>2</sub> O)	3	120
Uranium dioxide (UO <sub>2</sub> )	2	210	WB	1	1365
Uranium dioxide + Beryllium oxide	2	423	White bearing metal	1	1070
Uranium dioxide + Calcium oxide	2	426	White cast iron	1	1130, 1135
Uranium dioxide - chromium cermets	2	732	White oak	2	1082
Uranium dioxide - molybdenum cermets	2	735	White pines	2	1083
Uranium dioxide - niobium cermets	2	738	White plate glass	2	923, 925
Uranium dioxide + (di)Niobium pentoxide	2	427	White temper cast iron	1	1137
Uranium dioxide - stainless steel cermets	2	741	White wood	2	1090
Uranium dioxide - uranium cermets	2	744	Winchester crushed trap rock	2	829, 830
Uranium dioxide + Yttrium oxide	2	428	Window glass	2	923, 924
Uranium dioxide - zirconium cermets	2	746	Wolfram (see tungsten)		
Uranium dioxide + Zirconium dioxide	2	429	Wolframic acid, anhydrous (WO <sub>3</sub> ) (see tungsten trioxide)		
(tri)Uranium octoxide (U <sub>3</sub> O <sub>8</sub> )	2	237	Wolframite (WO <sub>3</sub> ) (see tungsten trioxide)		
Uranous uranic oxide (U <sub>3</sub> O <sub>8</sub> ) (see (tri)Uranium octoxide)			Wollastonite	2	859
Vacromin F	1	1213	Wood felt	2	1133
Valve bronze (see navy M)			Wood fibers	2	1091
Vanadium	1	441	Wood's metal	1	939
Vanadium + Iron	1	874	Wood products	2	1132
Vanadium + Yttrium	1	577			
Vanadium alloy, ferrovanadium (Russian)	1	875			
Vanadium carbide (VC)	2	606			
Vegetable fiberboards	2	1129			



Material Name	Vol.	Page	Material Name	Vol.	Page
Wool	2	1092	Zinc - silicon - arsenic intermetallic compound		
Angora	2	1092	ZnSiAs <sub>2</sub>	1	1374
Sheep	2	1092	Zinc alloys (specific types)		
Wrought iron	1	1185, 1219	Zamak Nr 400	1	880
WSe <sub>2</sub>	1	1368	Zamak Nr 410	1	1098
WSi <sub>2</sub>	1	1369	Zamak Nr 430	1	1098
WTe <sub>2</sub>	1	1370	Zinc dichloride (ZnCl <sub>2</sub> )	2	626
X-metal (see uranium)			Zinc ferrate (ZnFe <sub>2</sub> O <sub>4</sub> )	2	314
X-ray protection glass	2	924	Zinc germanium phosphide (ZnGeP <sub>2</sub> )	2	792
Xenon	3	88	Zinc oxide (ZnO)	2	243
Xenon - deuterium system	3	371	Zinc oxide + Magnesium oxide	2	435
Xenon - hydrogen system	3	374	Zinc oxide + Strontium oxide + ΣX <sub>1</sub>	2	527
Xenon - nitrogen system	3	377	Zinc oxide + Tin dioxide	2	438
Xenon - oxygen system	3	379	Zinc oxide + Tin dioxide + ΣX <sub>1</sub>	2	528
Yellow brass	1	981, 982	Zinc selenide - ZnSe (see zinc - selenium intermetallic compound)		
Ytterbium	1	446	Zinc selenium arsenide [ZnSiAs <sub>2</sub> ] (see zinc - selenium - arsenic intermetallic compound)		
Yttria (see yttrium oxide)			Zinc sulfate heptahydrate (ZnSO <sub>4</sub> · 7H <sub>2</sub> O)	2	694
Yttrium	1	449	Zircaloy -2	1	888
Yttrium aluminate (Y <sub>3</sub> Al <sub>2</sub> O <sub>12</sub> )	2	308	Zircaloy -4	1	888
Yttrium ferrate [Y <sub>3</sub> Fe <sub>2</sub> (FeO <sub>4</sub> ) <sub>3</sub> ]	2	311	Zircon, Brazil	2	318
Yttrium iron garnet (see yttrium ferrate)			Zircon 475	2	318
Yttrium oxide (Y <sub>2</sub> O <sub>3</sub> )	2	240	Zirconia (see zirconium dioxide)		
Yttrium oxide + Uranium dioxide	2	432	Zirconia, stabilized	2	522
"Z" nickel (see duranickel)			Zirconia brick	2	535, 895, 905
Zamak Nr 400	1	880	Zirconium	1	461
Zamak Nr 410	1	1098	Zirconium, iodide	1	462, 463
Zamak Nr 430	1	1098	Zirconium + Aluminum	1	882
Zinc	1	453	Zirconium + Aluminum + ΣX <sub>1</sub>	1	1100
Zinc + Aluminum	1	880	Zirconium - boron intermetallic compound		
Zinc + Aluminum + ΣX <sub>1</sub>	1	1098	ZrB	1	1375
Zinc + Cadmium	1	881	Zirconium + Hafnium	1	883
Zinc + Lead + ΣX <sub>1</sub>	1	1099			
Zinc - selenium intermetallic compound					
ZnSe	1	1371			

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Zirconium + Hafnium + $\Sigma X_1$	1	1101	Zirconium orthosilicate ( $ZrSiO_4$ ) (continued)		
Zirconium + Molybdenum + $\Sigma X_1$	1	1104	Zircon	2	318
Zirconium + Niobium	1	886	Zircon tam	2	318
Zirconium + Tantalum + $\Sigma X_1$	1	1105	ZnSb + CdSb	1	1412
Zirconium + Tin	1	887	ZnSe	1	1371
Zirconium + Tin + $\Sigma X_1$	1	1108	ZnSiAs <sub>2</sub>	1	1374
Zirconium + Titanium	1	890	ZrB	1	1375
Zirconium + Uranium	1	891			
Zirconium + Uranium + $\Sigma X_1$	1	1111			
Zirconium + Zirconium dioxide	1	1444			
Zirconium + $\Sigma X_1$	1	1112			
Zirconium alloys (specific types)					
Zircaloy-2	1	888			
Zircaloy-4	1	888			
Zirconium boride [ZrB] (see zirconium - boron intermetallic compound)					
Zirconium carbide (ZrC)	2	609			
Zirconium hydride (ZrH)	2	793			
Zirconium nitride (ZrN)	2	675			
Zirconium dioxide (ZrO <sub>2</sub> )	2	246			
Zirconium dioxide + Aluminum oxide	2	441			
Zirconium dioxide + Calcium oxide	2	442			
Zirconium dioxide + Calcium oxide + $\Sigma X_1$	2	531			
Zirconium dioxide + Magnesium oxide	2	446			
Zirconium dioxide + Silicon dioxide + $\Sigma X_1$	2	534			
Zirconium dioxide - titanium cermets	2	749			
Zirconium dioxide + Yttrium oxide	2	449			
Zirconium dioxide + Yttrium oxide + $\Sigma X_1$	2	537			
Zirconium dioxide - yttrium oxide - zirconium cermets	2	753			
Zirconium dioxide - zirconium cermets	2	752			
Zirconium silicate [ $ZrSiO_4$ ] (see zirconium orthosilicate)					
Zirconium silicate, natural (see zircon)					
Zirconium orthosilicate ( $ZrSiO_4$ )	2	317			
Brazil zircon	2	318			